

1 **Phosphorus addition mitigates N₂O and CH₄ emissions in N-**
2 **saturated subtropical forest, SW China**

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11 **Abstract**

12 Chronically elevated nitrogen (N) deposition has led to severe nutrient imbalance in forest soils.
13 Particularly in tropical and subtropical forest ecosystems, increasing N loading has aggravated
14 phosphorus (P) limitation of biomass production, and has resulted in elevated emissions of
15 nitrous oxide (N₂O) and reduced uptake of methane (CH₄), both of which are important
16 greenhouse gases. Yet, the interactions of N and P and their effects on GHG emissions remain
17 understudied. Here, we report N₂O and CH₄ emissions together with soil N and P data for a
18 period of 18 months following P addition (79 kg P ha⁻¹ yr⁻¹, applied as NaH₂PO₄ powder) to an
19 N-saturated, Masson pine-dominated forest at TieShanPing (TSP), Chongqing, SW China. We
20 observed a significant decline both in NO₃⁻ concentrations in soil water (at 5- and 20-cm depths)
21 and in N₂O emissions, the latter by 3 kg N ha⁻¹ yr⁻¹. We hypothesize that enhanced N uptake by
22 plants and soil microbes in response to P addition, results in less available NO₃⁻ for
23 denitrification. By contrast to most other forest ecosystems, TSP is a net source of CH₄. As for
24 N₂O, P addition significantly decreased CH₄ emissions, turning the soil into a net sink. Based on
25 our observation and previous studies in South America and China, we believe that P addition
26 relieves N-inhibition of CH₄ oxidation. Within the 1.5 years after P addition, no significant
27 increase of forest growth was observed at TSP, but we cannot exclude that understory vegetation
28 increased. Our study indicates that P fertilization of N-saturated, subtropical forest soils mitigates
29 N₂O and CH₄ emissions, in addition to alleviating nutrient imbalances and reducing losses of N
30 through NO₃⁻ leaching.

31 **Key Word:** N₂O and CH₄ emission, N saturation, Phosphate fertilization, soil CH₄ uptake, acid
32 forest soil.

33 **1 Introduction**

34 Anthropogenic activities have transformed the terrestrial biosphere into a net source of CH₄, N₂O
35 and CO₂, leading to increased radiative forcing (Montzka et al., 2011; Tian et al., 2016). During
36 the last decade, atmospheric concentrations of CO₂, CH₄, N₂O have increased at rates of 1.9 ppm
37 yr⁻¹, 4.8 and 0.8 ppb yr⁻¹, respectively (Hartmann et al., 2013). In China, the exponential increase
38 of reactive nitrogen (N) input into the biosphere since the 1970s has likely led to more carbon (C)
39 being sequestered in the biosphere (Cui et al., 2013; Shi et al., 2015). However, enhanced
40 emissions of N₂O and CH₄ due to chronic N pollution potentially offset the cooling effect by C
41 sequestration (Liu and Greaver, 2009; Tian et al., 2011).

42 Microbial nitrification and denitrification in soils account for about 60% of N₂O emissions
43 globally (Ciais et al., 2013; Hu et al., 2015). Although, microbial activity is often restricted in
44 low pH soils of unproductive forests, surprisingly large N₂O emissions have been reported from
45 acid, upland forest soils in South China (Zhu et al., 2013b). Reported average N₂O fluxes in
46 humid, subtropical forests range from 2.0 to 5.4 kg ha⁻¹ yr⁻¹ (Fang et al., 2009; Tang et al., 2006;
47 Zhu et al., 2013b), which by far exceeds global averages for temperate or tropical forest
48 ecosystems (Werner et al., 2007; Zhuang et al., 2012). This has been attributed to frequently
49 shifting aeration conditions during monsoonal summers, promoting both nitrification and
50 denitrification (Zhu et al., 2013b) and to large soil NO₃⁻ concentrations due to efficient cycling of
51 deposited N in acid subtropical soils (Yu et al., 2016).

52 Chronically elevated rates of N deposition (30-65 kg ha⁻¹ yr⁻¹; Xu et al., 2015) have resulted in
53 strong nutrient imbalances in southern Chinese forests, aggravating phosphorus (P) limitation
54 (Du et al., 2016). Phosphorous deficiency in N-saturated forests restricts forest growth and thus

55 constrains its capability to retain N (Huang et al., 2015; Li et al., 2016), resulting in ample
56 amounts of mineral N (NH_4^+ and NO_3^-) being present in the soil solution. Accordingly, Hall &
57 Matson (1999) observed larger N_2O emission in P-limited than in N-limited tropical forests after
58 one year of repeated N addition. Likewise, previous N manipulation studies in forests of South
59 China reported pronounced stimulation of N_2O emissions by N addition (Chen et al., 2016;
60 Wang et al., 2014; Zheng et al., 2016), supporting the idea that P limitation causes forests to be
61 more susceptible to N saturation and N_2O -N loss. In an N-limited tropical montane forest in
62 southern Ecuador, P addition alone ($10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$) had no effect on N_2O emissions during the
63 first two years. However, N_2O emission was smaller when P was added together with N (50 kg N
64 $\text{ha}^{-1} \text{ yr}^{-1}$) than in treatments with N addition alone (Martinson et al., 2013). After continued
65 fertilization for three years, also P addition alone reduced N_2O emissions at these sites (Müller et
66 al., 2015). In tropical China, with high N deposition ($\sim 36 \text{ kg ha}^{-1} \text{ yr}^{-1}$; Mo et al., 2008), P
67 addition ($150 \text{ kg P ha}^{-1} \text{ yr}^{-1}$) to an old-growth forest revealed a similar pattern, with no initial
68 effect on N_2O emissions (0-2 years) but a significant longer term effect (3 to 5 years) on N_2O
69 emissions (Chen et al., 2016; Zheng et al., 2016). In a secondary tropical forests in South China,
70 Wang et al. (2014) found no effect on N_2O emissions of P alone ($100 \text{ kg P ha}^{-1} \text{ yr}^{-1}$), and in
71 treatments combining P with N ($100 \text{ kg N ha}^{-1} \text{ yr}^{-1}$), N_2O emissions increased during the wet
72 season. Meanwhile, they observed a significant increase in soil microbial biomass after P
73 addition, which is in line with previous findings in tropical forest soils of South China (Liu et al.,
74 2012). Thus, they attributed the stimulating effect of P addition on N_2O emissions to the larger
75 nitrification and denitrification potential of the increased soil microbial biomass. This was also
76 proposed by Mori et al. (2014), based on results from a short-term incubation study with P
77 addition, excluding plant roots.

78 As the sole biogenic sink for CH₄, upland soils play an important role in balancing terrestrial
79 CH₄ emissions (Ciais et al., 2013; Dutaur and Verchot, 2007). Atmospheric CH₄ uptake in soil is
80 mediated by the activity of methanotrophic bacteria, which oxidize CH₄ to CO₂ to gain energy
81 for growth. Well-drained forest and grassland soils are dominated by yet uncultured, high-
82 affinity methanotrophs residing in the upper soil layers (Le Mer and Roger, 2010). In addition to
83 edaphic factors (pH and nutrients), parameters affecting the diffusion of CH₄ into the soil (soil
84 structure, moisture, temperature) are believed to be the major controllers for CH₄ uptake (Smith
85 et al., 2003). A number of studies have shown that excess N affects CH₄ fluxes in forest soils
86 (Liu and Greaver, 2009; Veldkamp et al., 2013; Zhang et al., 2008b). In general, N addition
87 promotes CH₄ uptake in N-limited soils by enhancing growth and activity of methanotrophs,
88 whereas excessive N input and N saturation inhibit CH₄ oxidation on an enzymatic level
89 (Aronson and Helliker, 2010; Bodelier and Laanbroek, 2004). P addition experiments in N-
90 enriched soils have shown positive effects on CH₄ uptake (Mori et al., 2013a; Zhang et al., 2011),
91 but the underlying mechanisms, i.e. whether P addition affects the methanotrophic community in
92 soils directly or alleviates the N-inhibition effect on CH₄ oxidation through enhanced N uptake
93 (Mori et al., 2013b; Veraart et al., 2015), remain unresolved.

94 Subtropical forests in South China show strong signs of N saturation, with exceedingly high
95 NO₃⁻ concentrations in soil water (Larssen et al., 2011; Zhu et al., 2013b). Little is known about
96 how P addition affects N cycling and N₂O emission in these acidic, nutrient-poor soils. Likewise,
97 the importance of increased mineral N concentrations for soil-atmosphere exchange of CH₄, and
98 how this is affected by P fertilization remain to be elucidated for soils of the subtropics. Here, we
99 assessed N₂O and CH₄ fluxes in an N-saturated subtropical forest in SW China under ambient N
100 deposition and studied the effects of P addition on emission rates, nutrient availability and tree

101 growth. The objectives were i) to quantify N₂O and CH₄ emissions, ii) to investigate the effect of
102 P addition on N₂O and CH₄ emission.

103 **2 Materials and Methods**

104 **2.1 Site description**

105 The study site “TieShanPing” (TSP) is a 16.2 ha subtropical forest (29° 38' N, 106° 41' E; 450
106 m a.s.l.), about 25 km northeast of Chongqing, SW China. TSP is a naturally regenerated,
107 secondary mixed coniferous-broadleaf forest, which developed after clear cutting in 1962
108 (Larssen et al., 2011). The forest stand is dominated by Masson pine (*Pinus massoniana*) and has
109 a density of about 800 stems ha⁻¹ (Huang et al., 2015). TSP has a monsoonal climate, with mean
110 annual precipitation of 1028 mm, and a mean annual temperature of 18.2 °C (Chen and Mulder,
111 2007a). Most of the precipitation (> 70%) occurs during summer periods (April to September).
112 Soils are predominantly well-drained, loamy yellow mountain soil, classified as Haplic Acrisol
113 (WRB 2014), with a thin O horizon (< 2 cm). In the O/A horizon, soil pH is around 3.7, and the
114 mean C/N and N/P ratios are 17 and 16, respectively. In the AB horizon, which has a slightly
115 higher pH, mean C/N is well above 20. The soil bulk density of the O/A horizon (~ 5 cm) is
116 about 0.75 g cm⁻³. Soil water-filled pore space (10 cm) at TSP hilltop generally ranges from 50
117 to 70% (mean ~ 60%; Zhu et al., 2013b). More details on soil properties are given in Table 1.

118 Annual N deposition at TSP measured in throughfall varies between 40 and 65 kg ha⁻¹
119 (dominated by NH₄⁺; Yu et al., 2016), while the annual bulk N deposition is from 20 to 30 kg ha⁻¹
120 (Chen and Mulder, 2007b). According to regional data, annual P deposition via throughfall is <
121 0.40 kg ha⁻¹ (Du et al., 2016). Strong soil acidification at TSP has resulted in severe decline in
122 forest growth (Li et al., 2014; Wang et al., 2007), and in decreased abundance and diversity of
123 ground vegetation (Huang et al., 2015). Pronounced N saturation with strong NO₃⁻ leaching from
124 the top soil has aggravated P deficiency (Huang et al., 2015). The total P content in the O/A

125 horizon is $\sim 300 \text{ mg kg}^{-1}$, while ammonium lactate-extractable P is smaller than 5 mg kg^{-1} (Table
126 1).

127 **2.2 Experimental Design**

128 Three blocks, each having two $20 \text{ m} * 20 \text{ m}$ plots, were established on well drained soils of a
129 gently sloping hillside. Adjacent plots were separated by at least 10-m buffer zone. In each block,
130 plots were randomly assigned to a Reference and a P treatment. On 4 May 2014, a single dose of
131 P fertilizer was applied as solid $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$, at a rate of $79.5 \text{ kg P ha}^{-1}$. The amount of P
132 added was estimated from P adsorption isotherms (Supplementary Materials, Table S1 and
133 Figure S1), to ensure significantly increase in soil available P. To apply P fertilizer evenly, we
134 divided each plot into a $5 \text{ m} * 5 \text{ m}$ grid and broadcasted the powdered fertilizer by hand in each
135 grid cell. The P dose applied at TSP was intermediate as compared to the $10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$
136 applied by Müller et al. (2015) to a mountain forest in Ecuador and the $150 \text{ kg P ha}^{-1} \text{ yr}^{-1}$ applied
137 by Zheng et al. (2016) to a subtropical forest in South China.

138 The addition of $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$ at the P-treated plots also resulted in an input of 59.0 kg ha^{-1} of
139 sodium (Na). One month after the fertilizer application, Na^+ concentrations in soil water of the P
140 treatments were about 5 mg L^{-1} at 5-cm depth and 3 mg L^{-1} at 20-cm depth (Table S2). Although
141 somewhat larger than in the reference plots, the Na^+ concentration in soil water of the P
142 treatments are unlikely to have exerted a strong negative impact on plant and microbial activities.

143 **2.3 Sample collection and analyses**

144 Within each plot, three ceramic lysimeters (P80; Staatliche Porzellanmanufaktur, Berlin) were
145 installed at 5- and 20-cm soils near the plot centre in August 2013. To obtain water samples,

146 350-ml glass bottles with rubber stoppers were pre-evacuated, using a paddle pump, and
147 connected to the lysimeters for overnight sampling. Between November 2013 and October 2015,
148 we sampled soil pore water bi-monthly in the dry and dormant season and monthly during the
149 growing season. All water samples were kept frozen during storage and transport. Concentrations
150 of NH_4^+ , NO_3^- , potassium (K^+), calcium (Ca^{2+}), and magnesium (Mg^{2+}) in soil water were
151 measured at the Research Center for Eco-Environmental Sciences (RCEES), Chinese Academy
152 of Sciences, Beijing, using ion chromatography (DX-120 for cations and DX-500 for anions).

153 In August 2013, soils from the O/A (0-3 cm), AB (3-8 cm) and B (8-20 cm) horizons were
154 sampled near the lysimeters for soil analysis. Total P and plant-available P contents were
155 monitored in samples collected from the O/A horizons every six months, starting two days
156 before P addition. Soil samples were kept cold ($< 4\text{ }^\circ\text{C}$) during transport and storage. Before
157 analysis, soil samples were air dried and sieved (2 mm). Soil pH was measured in soil
158 suspensions (10 g dry soil and 50 ml deionized water) using a pH meter (PHB-4, Leici, China).
159 Total soil C and N contents were determined on dried and milled samples, using a LECO
160 elemental analyzer (TruSpec[®]CHN, USA). To measure total P, 1 g dry soil was digested with 5
161 ml of 6 M H_2SO_4 (Singh et al., 2005) and measured as ortho-phosphate by the molybdenum blue
162 method (Murphy and Riley, 1962). Ammonium lactate (0.01 M)-extractable P and H_2O -
163 extractable P (P_{Al} and $\text{P}_{\text{H}_2\text{O}}$, respectively) were measured as ortho-phosphate after extraction (1.5
164 g dry soil in 50 ml solution) (Singh et al., 2005). Ammonium oxalate (0.2 M)-extractable Fe, Al
165 and P were measured by inductive coupled plasma (7500; Agilent) after extraction (1.5 g dry soil
166 in 50 ml solution).

167 From August 2013 onwards, we measured N_2O and CH_4 emissions in triplicate close to the
168 lysimeters, using static chambers (Zhu et al., 2013b). To investigate the immediate effect of P

169 addition on N₂O emissions, we sampled the gas emissions once before (2 May) and three times
170 (7, 10 and 12 May) after the P application. Gas samples (20 ml) were taken 1, 5, 15 and 30
171 minutes after chamber deployment and injected into pre-evacuated glass vials (12 ml) crimp-
172 sealed with butyl septa (Chromacol, UK), maintaining overpressure to avoid contamination
173 during sample transport. Mixing ratios of N₂O and CH₄ were analyzed using a gas
174 chromatograph (Model 7890A, Agilent, US) at RCEES, equipped with an ECD for detection of
175 N₂O (at 375 °C with 25 ml min⁻¹ Ar/CH₄ as make up gas), a FID for CH₄ (250 °C; 20 ml min⁻¹
176 N₂ as make-up gas) and a TCD for CO₂. Exchange rates between soil and atmosphere
177 (emission/uptake) were calculated from measured concentration change in the chambers over
178 time, applying linear or polynomial fits to the concentration data. Cumulative N₂O emissions
179 over time were estimated by linear interpolation between measurement dates (Zhu et al., 2013b).

180 From October 2013 onwards, litterfall was collected during the first week of every month in five
181 replicates per plot. Litterfall collectors were made of 1 m² nylon nets (1 mm mesh size), held in
182 place by four wooden poles 0.8 m above the ground. Fresh litter was dried at 65°C. In early
183 November 2013 and 2014 (at the end of the growing season), we collected current-year pine
184 needles from several branches of three trees in each plot. The collected needles were dried at
185 65 °C and the dry weight of 500 needles was determined. A subsample was dried at 80 °C and
186 finely milled prior to chemical analysis at the Chinese Academy of Forestry. Total C and N were
187 measured using an elemental analyzer (FLASH 2000; Thermo Scientific; USA). The contents of
188 K, Ca, Mg and P in the needles were determined by ICP-AES (IRIS Intrepid II; Thermo
189 Scientific; USA) after digesting 0.25 g dry weight samples with 5 ml of ultra-pure nitric acid. In
190 November 2013, and 2014, and in February of 2015, we measured the height and the diameter at
191 breast height (DBH) of 6 to 10 Masson pines (only those with DBH > 5 cm) at each plot. These

192 data were used to estimate the standing biomass of Masson pines based on standard allometric
193 equations (Li et al., 2011; Zeng et al., 2008).

194 Daily average air temperature and sum of precipitation were monitored by a weather station
195 (WeatherHawk 232, USA) placed on the roof at the local forest bureau, in about 1 km distance
196 from the sampling site (Yu et al., 2016).

197 **2.4 Statistical analyses**

198 Statistical analyses were performed using R version 3.3.1 (R Core Team, 2016). All data were
199 tested for normality (Kolmogorov-Smirnov's test) and homoscedasticity (Levene's test) before
200 further analysis. If not normally distributed, the data were normalized by logarithmic
201 transformation. Considering heterogeneity among blocks, temporal variabilities of NO_3^-
202 concentrations, N_2O and CH_4 fluxes were presented separately for each block. For time series
203 data, we used linear mixed-effect (LME) models, to account for both repeated measurements and
204 within-group variance of a stratification variable (block design). LME models were applied to
205 test the effects of P addition on soil N_2O and CH_4 fluxes, NH_4^+ , NO_3^- , K^+ , Ca^{2+} and Mg^{2+}
206 concentrations in soil water, as well as litterfall (Koehler et al., 2009; Müller et al., 2015). The
207 analysis was based on data for plot means (the average of 3 subplot replicates) from three blocks.
208 In LME models, treatments (Reference or P addition) were considered fixed effects, while
209 sampling time and plots were treated as random effects. We then assessed the significance of
210 fixed effects through analysis of variance for LME models. One-way analysis of variance
211 (ANOVA, Turkey post-hoc test) was conducted to examine the treatment effects on soil pH,
212 nutrient contents in organic matter, and data of tree growth. Significance levels were set to $p <$
213 0.05, if not specified otherwise.

214 **3 Results**

215 **3.1 Nutrient concentrations in soil and soil water**

216 Addition of P resulted in a significant increase in soil P content in the O/A horizon, both as P_{AI}
217 and total P (Table 2). However, after 15 months, only P_{AI} indicated an enhanced P status, while
218 total soil P did not differ significantly from background values at the reference sites. P addition
219 had no significant effect on soil pH, or soil C and N content. The NO_3^- concentration in soil
220 water collected at 5 cm depth varied seasonally, with significantly greater values (30-40 mg N L⁻¹)
221 towards the start of the growing season in 2015 (April, Fig. S2), but not in 2014, likely due to
222 dilution by abundant precipitation in February to March 2014. Addition of P resulted in
223 significantly smaller NO_3^- concentrations in soil water at both 5- and 20-cm depths (Fig. 1b). In
224 general, the concentration of NH_4^+ in soil water was small (< 0.5 mg L⁻¹) and not affected by P
225 addition (Fig. 1a). At both depths, mean soil water concentrations of Mg^{2+} and Ca^{2+} were
226 significantly smaller in the P-treated than the reference plots, and the sum of charge of dissolved
227 base cations declined significantly in response to P addition (Fig. S3).

228 **3.2 N₂O and CH₄ fluxes: effects of P addition**

229 During the experimental period, N₂O fluxes varied seasonally (Fig. 2), showing a significant
230 relationship with daily precipitation (Fig. S4a), but not with daily mean temperature (Fig. S4b).
231 In the Reference plots, mean N₂O fluxes were generally below 50 $\mu\text{g N m}^{-2} \text{hr}^{-1}$ in the dry, cool
232 season, but reached values of up to 600 $\mu\text{g N m}^{-2} \text{hr}^{-1}$ in the growing season (Fig. 2). Cumulative
233 N₂O emissions were estimated with seasonally averaged fluxes, and they differed greatly among
234 the three blocks (Fig. 3). The greatest annual N₂O emission was observed in the Reference plot
235 (7.9 kg N ha⁻¹) of block 2. Mean N₂O fluxes during the 1.5 years after P addition were

236 significantly smaller in the P treatment than in the Reference (Fig. 4). The P addition resulted in
237 a decrease in cumulative N₂O emission by about 3 kg N ha⁻¹ yr⁻¹ on average, which is a 50%
238 reduction (Fig. 3). No immediate effects (within days) of P addition on N₂O emission was
239 observed (Fig. S5).

240 CH₄ fluxes varied greatly between blocks (Fig. 5). Net-emission of CH₄ was observed in summer
241 2013 (~ 80 µg C m⁻² hr⁻¹) in blocks 1 and 2, whereas block 3 showed CH₄ uptake. From spring
242 2014 until October 2015, CH₄ fluxes were less variable in all blocks, with values fluctuating
243 around zero. A longer period of net-emission was observed in block 3 during the dry season
244 2014. The fluxes did not correlate with either precipitation or air temperature (Fig. S5c&d). In
245 the 1.5 years following P addition, mean CH₄ fluxes indicated net CH₄ emission (~ +3.8 µg C m⁻²
246 hr⁻¹) in the Reference, whereas net CH₄ uptake (~ -6.5 µg C m⁻² hr⁻¹) was observed in the P
247 treatment (Fig. 6). The suppressing effect of P addition on CH₄ emission was in accordance with
248 what was found for NO₃⁻ concentration and N₂O emission.

249 **3.3 The effect of P addition on tree growth**

250 Throughout the 2-year experimental period, we observed no change in tree biomass (138 t ha⁻¹)
251 in response to P addition (Table S3). Likewise, there was no effect of P treatment on the 500-
252 needle weight (13 g on average). Between the two samplings in 2013 and 2014, we found
253 differences in chemical composition of the pine needles, but this effect was not linked to P
254 addition. Also, the C/N and N/P ratios of the needles (40 and 16, respectively) were hardly
255 affected by P addition. Monthly litterfall varied seasonally in both Reference and P treatment
256 (Fig. S6), but no significant difference was found between the two treatments.

257 **4 Discussion**

258 Background N₂O emission rates in the Reference plots were relatively large (Fig. 2), with mean
259 values close to 100 $\mu\text{g N m}^{-2} \text{hr}^{-1}$ (Fig. 4). This is within the range of N₂O emission rates
260 previously reported for well-drained hillslope soils at TSP (Zhu et al., 2013b), but greater than
261 the rates reported for other forests in South China. For instance, N₂O emission rates averaged to
262 37 $\mu\text{g N m}^{-2} \text{hr}^{-1}$ in unmanaged sites at Dinghushan (Fang et al., 2009; Tang et al., 2006) and 50
263 $\mu\text{g N m}^{-2} \text{hr}^{-1}$ in N-fertilized sites (Zhang et al., 2008a). TSP Reference plots emitted on average
264 5.3 $\text{kg N ha}^{-1} \text{yr}^{-1}$ (Fig. 3), which is about 10% of the annual N deposition (50 $\text{kg ha}^{-1} \text{yr}^{-1}$)
265 (Huang et al., 2015). These fluxes are well above average fluxes reported for tropical rainforests
266 (Werner et al., 2007). Large N₂O emissions at TSP are likely due to the large N deposition rates
267 (Huang et al., 2015), as suggested by the similar trends indicated by data from a wide range of
268 ecosystems (Liu et al., 2009). Also, warm-humid conditions during monsoonal summers may
269 stimulate N₂O emissions (Ju et al., 2011), as monsoonal rainstorms triggered peak fluxes (Pan et
270 al., 2003). The positive correlation between precipitation and N₂O emission peaks (Fig. S4a)
271 may indicate the importance of denitrification as the dominant N₂O source. This is supported by
272 recent ¹⁵N tracing experiments at TSP (Zhu et al., 2013a).

273 Addition of P caused a significant decline in soil mineral N (predominantly NO₃⁻; Fig. 2),
274 particularly during summers, when NO₃⁻ concentrations were relatively high (Fig. S2). At the
275 same time, annual N₂O emissions decreased by more than 50% (Figs. 3 and 4). These findings
276 are consistent with a number of previous studies (Baral et al., 2014; Hall and Matson, 1999; Mori
277 et al., 2014), which attributed the reduction of N₂O emissions in P-treated soils decreased
278 mineral N content, most likely as a consequence of stimulated plant uptake and/or microbial

279 assimilation. It is noteworthy that there was no significant correlation between N₂O emission
280 rates and soil water NO₃⁻ concentration in our study (Figs. 2 and S2), suggesting that the
281 suppressing effect of P on N₂O emissions was indirect, probably by affecting the competition for
282 mineral N between plant roots and microbes (Zhu et al., 2016). In contrast to our 1.5-year study,
283 P-addition experiments in South Ecuador (Martinson et al., 2013) and South China (at
284 Dinghushan Biosphere Reserve (Zheng et al., 2016) found no effect of a single P addition on
285 N₂O emission during the first two years after application. However, significant reduction in N₂O
286 emission was observed after three to five years with continuous P addition, both at the
287 Ecuadorian and the Chinese site (Chen et al., 2016; Müller et al., 2015). For the montane forest
288 site in Ecuador, the observed delay in N₂O emission response to P addition may be explained by
289 the moderate amount of P added (10 kg P ha⁻¹ yr⁻¹; Martinson et al., 2013). Moreover, the
290 experiments were conducted in a forest with low ambient N deposition (~ 10 kg N ha⁻¹ yr⁻¹) and
291 N₂O fluxes (~ 0.36 kg N ha⁻¹ yr⁻¹ in the Reference plots) (Martinson et al., 2013; Müller et al.,
292 2015). By contrast, the Dinghushan site in South China receives 28 kg N ha⁻¹ yr⁻¹ through wet
293 inorganic N deposition (Zheng et al., 2016), which is similar to the N deposition at our site (Chen
294 and Mulder, 2007b; Huang et al., 2015). They also observed larger N₂O emission rates (~ 0.88
295 kg N ha⁻¹ yr⁻¹ in the Reference plots) than in the Ecuadorian site. However, forests do not always
296 display a straightforward relationship between N deposition and N₂O emissions. Manipulation
297 experiments in the European NITREX project, for instance, revealed a much stronger correlation
298 of N₂O emissions with soil NO₃⁻ leaching than with N deposition (Gundersen et al., 2012).
299 Indeed, KCl-extractable mineral N at the Dinghushan site (~ 40 mg kg⁻¹; Zheng et al., 2016) is
300 several-fold smaller than at our site (> 100 mg kg⁻¹; Zhu et al., 2013b), indicating that
301 Dinghushan is less N-rich than TSP. This suggests that the response of N₂O emission to P

302 addition might depend on the N status of the soil. The fact that numerous studies found apparent
303 suppression of N₂O emission in short-term experiments (< 2 years) in N + P treatments, but not
304 in treatments with P alone, supports this idea (Müller et al., 2015; Zhang et al., 2014b; Zheng et
305 al., 2016).

306 Other studies have observed increased N₂O emissions upon P addition (Mori et al., 2013c; Wang
307 et al., 2014). In an *Acacia mangium* plantation, fertilized with P, Mori et al. (2013b&c) found
308 that N₂O emissions were stimulated in the short-term but reduced in the long-term. While
309 suppression of N₂O emission by P has been attributed to increased plant N uptake (Mori et al.,
310 2014), increased N₂O emission are generally explained by enhanced microbial biomass (Liu et
311 al., 2012) and denitrification activity (Ehlers et al., 2010; He and Dijkstra, 2015). N₂O emissions
312 measured shortly after P addition at our site in May 2014 were not different from fluxes in
313 untreated reference plots (Fig. S5). This may indicate that plant uptake at TSP is more important
314 for the effect of P addition on N₂O emissions than changes in microbial activity, which are
315 expected to occur more rapidly.

316 The Reference plots at TSP showed net CH₄ emission for extended periods of the year (Figs. 5
317 and 6). Also, long-term CH₄ fluxes sampled between 2012 and 2014 on hillslopes near-by (Fig.
318 S7; Zhu et al., unpublished data) showed net CH₄ emission. This is in contrast to the generally
319 reported CH₄ sink function of forested upland soils (Ciais et al., 2013; Dutaur and Verchot,
320 2007). For example, CH₄ uptake rates reported for well-drained, forest soils in South Chinese
321 forest range from 30 to 60 µg C m⁻² hr⁻¹ (Fang et al., 2009; Tang et al., 2006; Zhang et al., 2014a).
322 Since aerated upland soils typically provide favourable conditions for microbial CH₄ uptake (Le
323 Mer and Roger, 2010), the net emission observed in our sites is unlikely due to enhanced
324 production, but rather by suppressed consumption. One reason for the net CH₄ emission at TSP

325 could be inhibition of CH₄ oxidation by NH₄⁺, as reported previously (Bodelier and Laanbroek,
326 2004; Zhang et al., 2014a). The concentration of NH₄⁺ in the soil water was rather small (< 0.5 g
327 L⁻¹; Fig. 1), which does not preclude, however, that NH₄⁺ availability from the soil exchangeable
328 pool is high. Zhu et al. (2013b) found extraordinarily high KCL-extractable NH₄⁺ in TSP surface
329 soils, likely reflecting the large atmogenic NH₄⁺ input at the TSP site (Huang et al., 2015).

330 P addition had a significant impact on CH₄ fluxes, changing the soil from a net source to a net
331 sink on an annual basis (Fig. 6). However, the uptake rates of CH₄ in the P treatments remained
332 smaller than those reported for forest soils in tropical China (Tang et al., 2006; Zhang et al.,
333 2008b). The stimulating effect of P addition on CH₄ uptake is consistent with previous studies
334 (Mori et al., 2013a, 2013b; Zhang et al., 2011), and has been attributed to lessening the NH₄⁺
335 inhibition of methane oxidation. Unfortunately, we did not measure KCl-extractable NH₄⁺ in our
336 study, but a decline of available NH₄⁺, which is the substrate for nitrification, is likely as NO₃⁻
337 concentrations in soil water were significantly smaller with in the P-treatments (Fig. 1). P
338 addition may also result in a change of the taxonomic composition of the methane oxidizing
339 community (Mori et al., 2013a; Veraart et al., 2015). Alternatively, CH₄ oxidation may be
340 stimulated by increased CH₄ diffusion into the soil, due to enhanced root growth and increased
341 soil water loss due to transpiration in P-amended plots (Zhang et al., 2011). Given the strong N
342 enrichment of TSP forest (Huang et al., 2015), it is likely that the reason for the observed
343 reduction in CH₄ emissions in response to P fertilization is due to alleviating direct NH₄⁺
344 inhibition of methane monooxygenase (Veldkamp et al., 2013), rather than to P-stimulation of
345 methanotrophic activity (Veraart et al., 2015).

346 Shortly after fertilizer application, we observed a modest, albeit significant increase of Na⁺
347 concentration in soil water (Table S2). Other studies have documented the potential toxicity of

348 excess Na^+ in soil water to plant and microbial activities (Rengasamy et al., 2003; Wong et al.,
349 2008). However, the occurrence of Na^+ toxicity at the treated plots, affecting N turnover
350 processes, is unlikely, as Na^+ concentrations in soil water, within one month after application
351 (Table S2), did not exceed 5 mg L^{-1} , far smaller than the values commonly assumed to indicate
352 toxicity (40 to 100 mg L^{-1}) (Bernstein 1975). Frequent precipitation at TSP (Yu et al., 2016),
353 both prior and following the addition of $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$ (Fig. 2), apparently diluted and leached
354 Na^+ , thus preventing toxic effects.

355 P application significantly increased plant-available P in the P-limited TSP soil (Table 2).
356 Meanwhile, concentrations of leachable base cations (K^+ , Mg^{2+} , Ca^{2+}) in soil water decreased
357 (Fig. S3), as expected from the reduction of NO_3^- concentrations in the P-treatments, which
358 represent a major decline in mobile anions in the P-treated soils (Mochoge and Beese, 1986). We
359 observed no sign of stimulated forest growth or increased N uptake by plants within the
360 relatively short period of our study (Table S3 and Fig. S6), making it difficult to link the
361 observed reduction in mineral N in the soil solution (Fig. 1) to plant growth. When interpreting
362 the observed P effect on NO_3^- concentrations in soil water, several aspects need to be considered.
363 Firstly, two years of observation may be too short to detect any significant increase in tree
364 growth, due to NO_3^- uptake by plants, given the commonly large variabilities in tree biomass
365 estimates (Alvarez-Clare et al., 2013; Huang et al., 2015). Secondly, a significant proportion of
366 the added P, and of excess N, may have been assimilated by the understory biomass, which was
367 not assessed in this study. Previously, understory vegetation has been reported to quickly
368 respond to P addition (Fraterrigo et al., 2011). Thirdly, as long-term N saturation and
369 acidification at TSP have reduced forest health (Lu et al., 2010; Wang et al., 2007), we may not
370 expect immediate response of forest growth to P addition. Large needle N/P ratios (17-22, Table

371 S3) indicated that P limitation for tree growth was not relieved 1.5 years after P addition (Li et al.,
372 2016). Therefore, enhanced N uptake by understory growth and/or soil microbial biomass may
373 have been the main mechanisms responsible for observed NO_3^- decline in the P-treated soil (Hall
374 & Matson 1999).

375 Overall, our study suggests that N-saturated TSP soils act as a regional hotspot for N_2O and CH_4
376 emissions. Within the short experimental period of 1.5 years, P fertilization was shown to
377 significantly decrease NO_3^- concentrations in soil water and to reduce both N_2O and CH_4
378 emissions. These findings provide a promising starting point for improving forest management
379 towards GHG abatement targets, taking into account the P and N status of subtropical soils in the
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603 **Table 1** Background soil properties of the experimental plots at Tieshanping (TSP). Values are
 604 means and standard deviations in parenthesis (n = 6). Soils were sampled in August 2013.

	Soil Layer	pH	Total C g kg ⁻¹	Total N g kg ⁻¹	Total P mg kg ⁻¹	C/N	N/P
Block 1	O/A (0-3 cm)	3.7 (0.1)	80.7 (32.3)	4.8 (1.7)	308 (57)	17.0 (2.5)	15.5 (5.7)
	AB (3-8 cm)	3.8 (0.0)	23.9 (9.3)	1.3 (0.6)	-*	20.0 (3.0)	-
	B (8-20 cm)	3.9 (0.2)	8.6 (1.2)	< 0.05	-	-	-
Block 2	O/A (0-3 cm)	3.6 (0.1)	77.6 (13.4)	4.7 (0.8)	297 (44)	16.7 (1.3)	15.7 (2.8)
	AB (3-8 cm)	3.7 (0.1)	20.2 (5.3)	1.0 (0.3)	-	21.4 (3.3)	-
	B (8-20 cm)	3.9 (0.1)	7.1 (1.6)	< 0.05	-	-	-
Block 3	O/A (0-3 cm)	3.6 (0.1)	67.0 (15.5)	3.8 (0.8)	223 (45)	17.4 (0.6)	17.2 (3.7)
	AB (3-8 cm)	3.6 (0.1)	21.0 (7.9)	1.1 (0.5)	-	24.5 (4.6)	-
	B (8-20 cm)	3.8 (0.1)	7.2 (1.5)	< 0.05	-	-	-
	Soil Layer	P _{H2O} mg kg ⁻¹	P _{Al} mg kg ⁻¹	Al _{ox} mg kg ⁻¹	Fe _{ox} mg kg ⁻¹	P _{ox} mg kg ⁻¹	P _{ox} / (Al _{ox} + Fe _{ox})
Block 1	O/A (0-3 cm)	< 5.0	5.8 (1.4)	1700 (513)	1933 (350)	85.8 (22.6)	0.025 (0.008)
	AB (3-8 cm)	< 5.0	2.1 (0.6)	1217 (243)	1692 (493)	47.1 (22.0)	0.016 (0.007)
	B (8-20 cm)	< 5.0	< 1.0	1083 (90)	1158 (249)	29.3 (28.6)	0.012 (0.011)
Block 2	O/A (0-3 cm)	< 5.0	5.9 (1.0)	1500 (238)	1792 (215)	79.2 (21.5)	0.024 (0.007)
	AB (3-8 cm)	< 5.0	1.6 (0.4)	925 (149)	1517 (320)	37.2 (10.7)	0.016 (0.006)
	B (8-20 cm)	< 5.0	< 1.0	892 (209)	1033 (413)	16.1 (10.5)	0.009 (0.007)
Block 3	O/A (0-3 cm)	< 5.0	4.1 (0.9)	1367 (180)	1667 (168)	50.7 (10.9)	0.017 (0.003)
	AB (3-8 cm)	< 5.0	4.4 (4.0)	1075 (128)	1350 (150)	24.8 (8.3)	0.010 (0.002)
	B (8-20 cm)	< 5.0	< 1.0	992 (130)	875 (138)	8.0 (2.0)	0.004 (0.001)

605 P_{H2O} = Water-extractable P, P_{Al} = Ammonium lactate-extractable P,

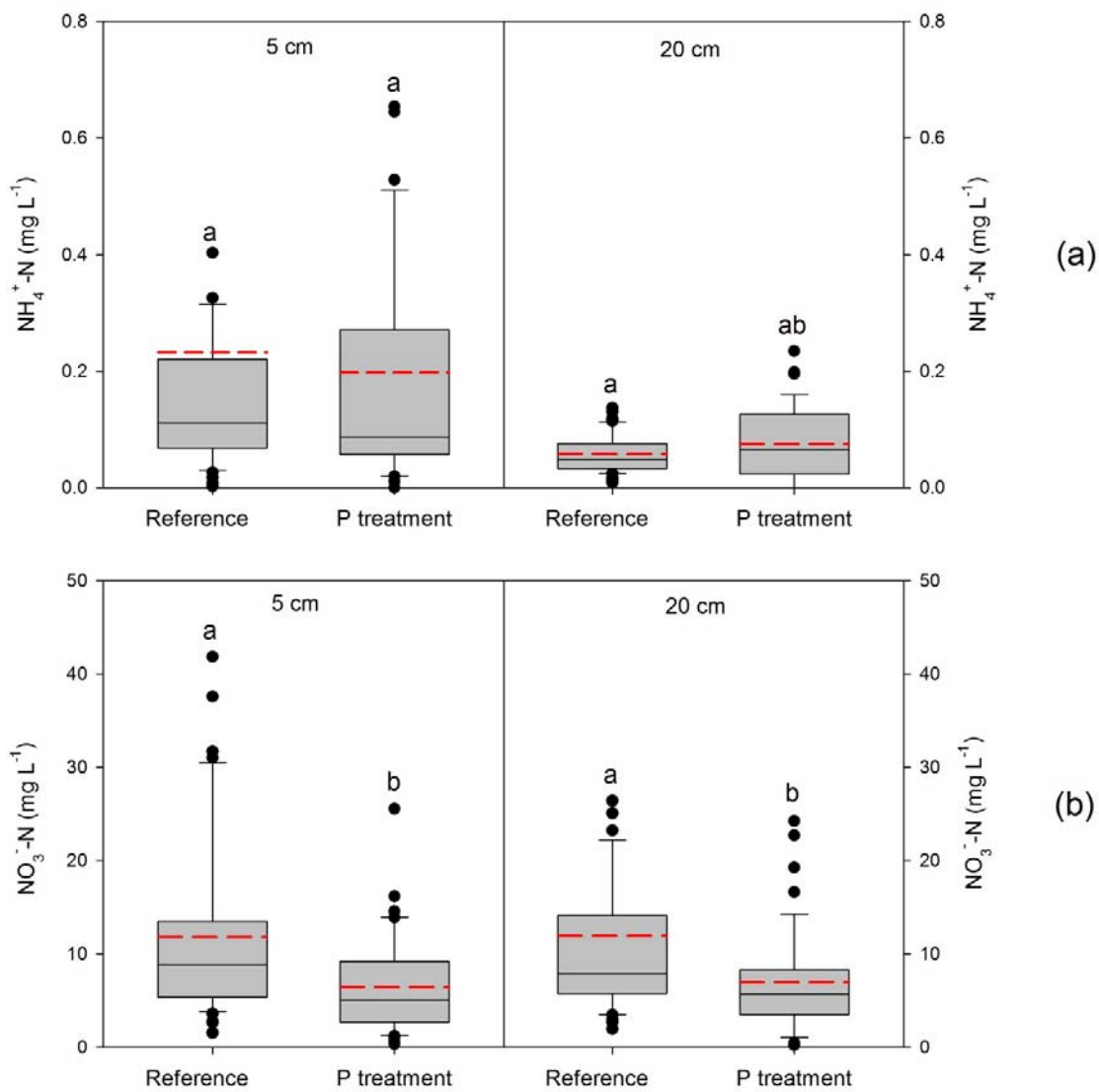
606 Al_{ox} = Oxalate extractable Al, Fe_{ox} = Oxalate extractable Fe, P_{ox} = Oxalate extractable P.

607 * Data not available

608 **Table 2** Soil pH, C, N and P contents in the O/A horizon (0-3 cm) in the References (Ref) and P
609 treatments. Values are means and standard deviations in parenthesis (n = 9). P addition was
610 conducted on 14/05/04, after the first two sampling dates.

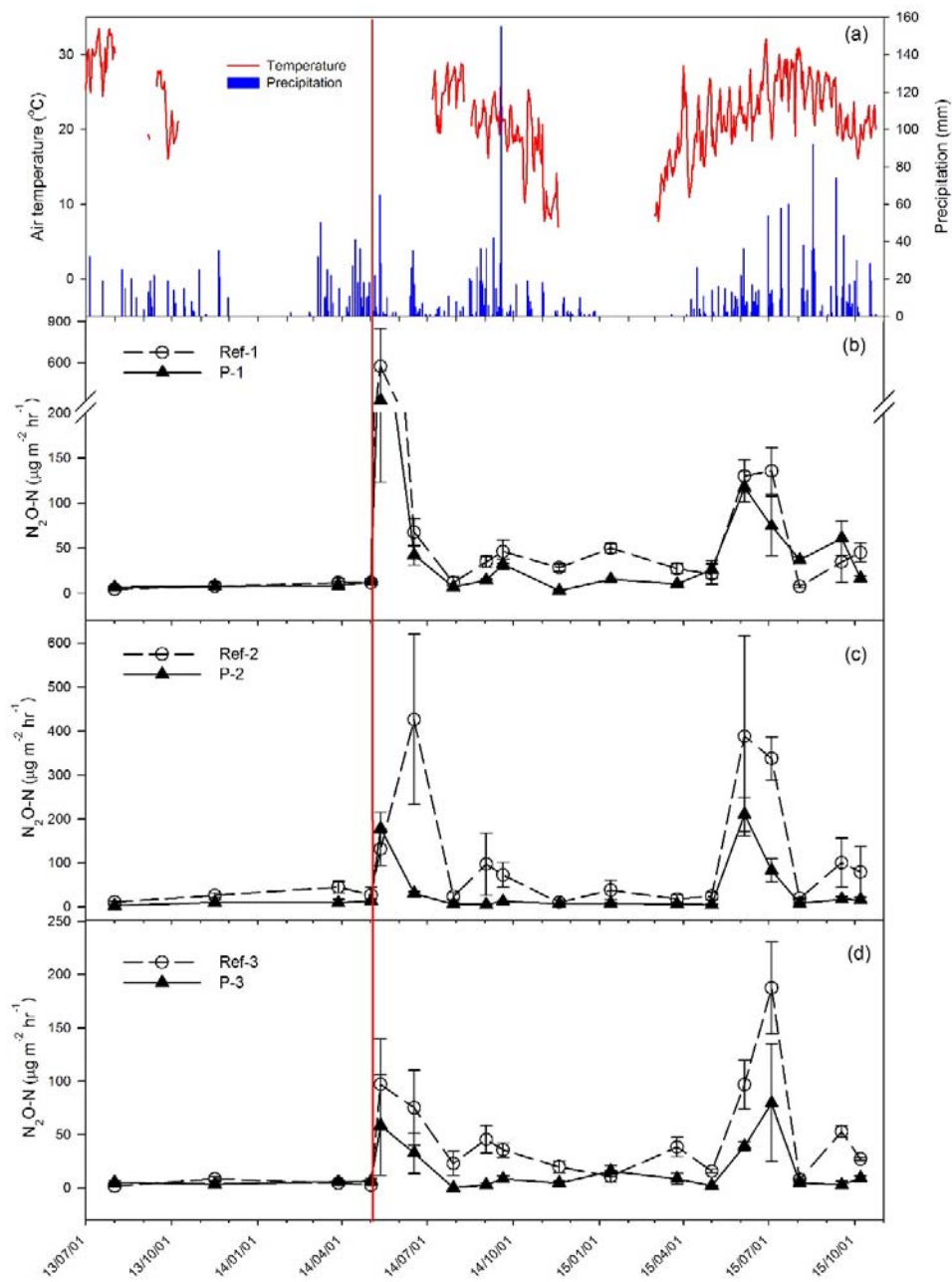
		pH	Total C g kg ⁻¹	Total N g kg ⁻¹	C/N	P _{AI} mg kg ⁻¹	Total P mg kg ⁻¹
13/08/02	Ref	3.7 (0.1) ^{bc†}	8.3 (2.3) ^{ab}	0.5 (0.1) ^{bcd}	16.9 (1.1) ^{bcd}	5.4 (1.4) ^c	292 (46) ^{bc}
	P	3.6 (0.1) ^c	6.7 (2.0) ^b	0.4 (0.1) ^{bd}	17.1 (2.1) ^{bc}	5.1 (1.3) ^c	260 (70) ^c
14/05/02	Ref	3.7 (0.1) ^{abc}	12.2 (4.2) ^a	0.9 (0.3) ^a	13.7 (1.5) ^e	19.0 (8.0) ^c	336 (65) ^{bc}
	P	3.8 (0.2) ^{abc}	9.0 (3.5) ^{ab}	0.7 (0.2) ^{abc}	14.2 (2.8) ^{de}	13.7 (5.2) ^c	270 (72) ^{bc}
14/05/10	Ref	3.8 (0.1) ^{abc}	9.9 (2.1) ^{ab}	0.7 (0.2) ^{ab}	14.0 (0.7) ^e	15.4 (7.0) ^c	304 (49) ^{bc}
	P	3.9 (0.3) ^{ab}	8.0 (1.9) ^{ab}	0.6 (0.1) ^{bcd}	14.3 (1.3) ^{cde}	174 (114) ^a	572 (242) ^a
14/12/02	Ref	3.8 (0.1) ^{abc}	10.5 (3.6) ^{ab}	0.7 (0.3) ^{ab}	14.5 (1.3) ^{cde}	14.2 (7.4) ^c	328 (102) ^{bc}
	P	3.9 (0.2) ^{abc}	9.5 (2.1) ^{ab}	0.7 (0.1) ^{abc}	14.0 (0.8) ^e	66 (24) ^{ab}	442 (106) ^{ab}
15/08/02	Ref	3.9 (0.2) ^{ab}	8.3 (2.2) ^{ab}	0.4 (0.1) ^{cd}	20.5 (2.5) ^a	13.4 (6.2) ^c	291 (61) ^{bc}
	P	4.0 (0.2) ^a	6.5 (1.9) ^b	0.3 (0.1) ^d	19.7 (2.2) ^{ab}	57 (36) ^{ab}	383 (136) ^{bc}

611 † Different letters indicate significant differences (p < 0.05).



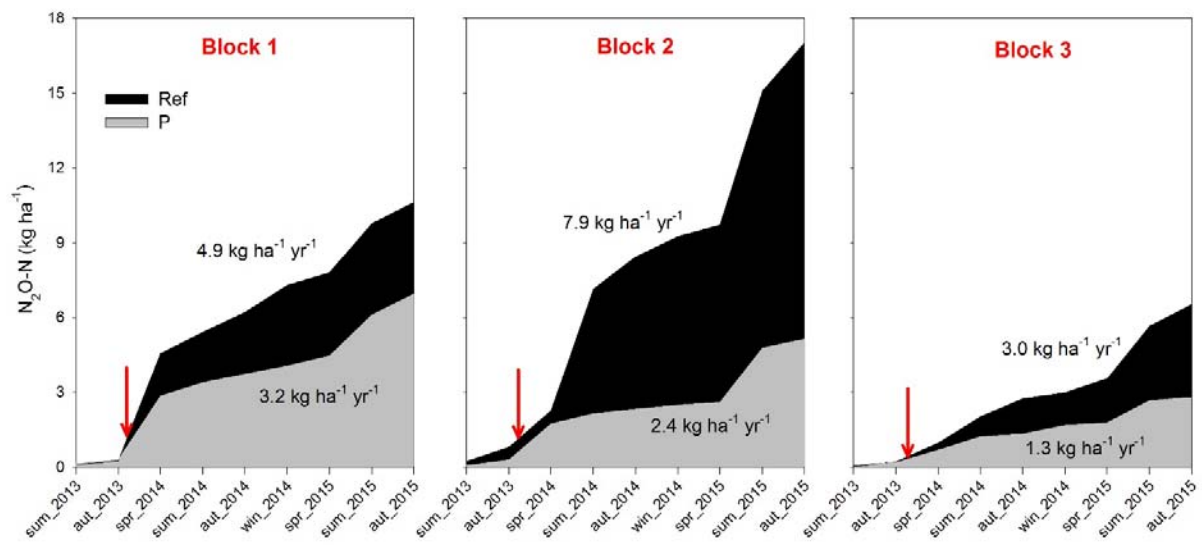
612

613 **Fig. 1** Box whisker plots of NH_4^+ (a) and NO_3^- (b) concentration in soil water at 5- and 20-cm
 614 depths in the References and P treatments, throughout 1.5 years after the P addition; red dashed
 615 lines indicate mean values; different letters indicate significant differences ($p < 0.05$).



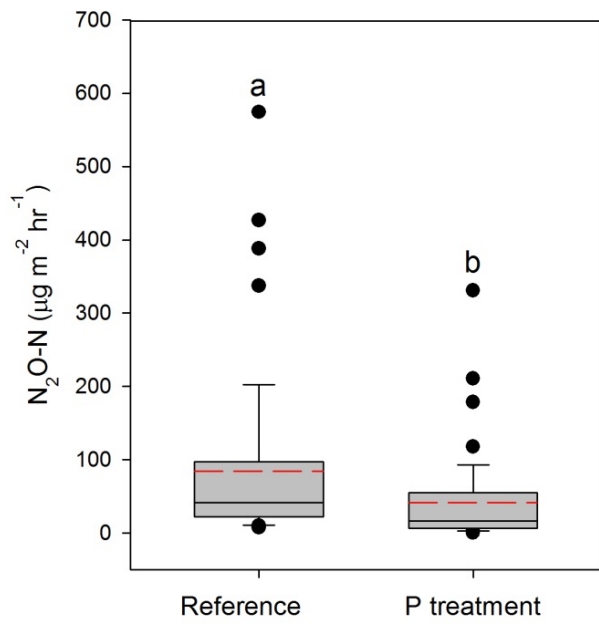
616

617 **Fig. 2** Daily mean air temperature and precipitation (a), and monthly mean N₂O fluxes (±SE) in
 618 the References (Ref) and P treatments in each of the three blocks (b-d); the red vertical line gives
 619 the date of P addition (4 May, 2014).



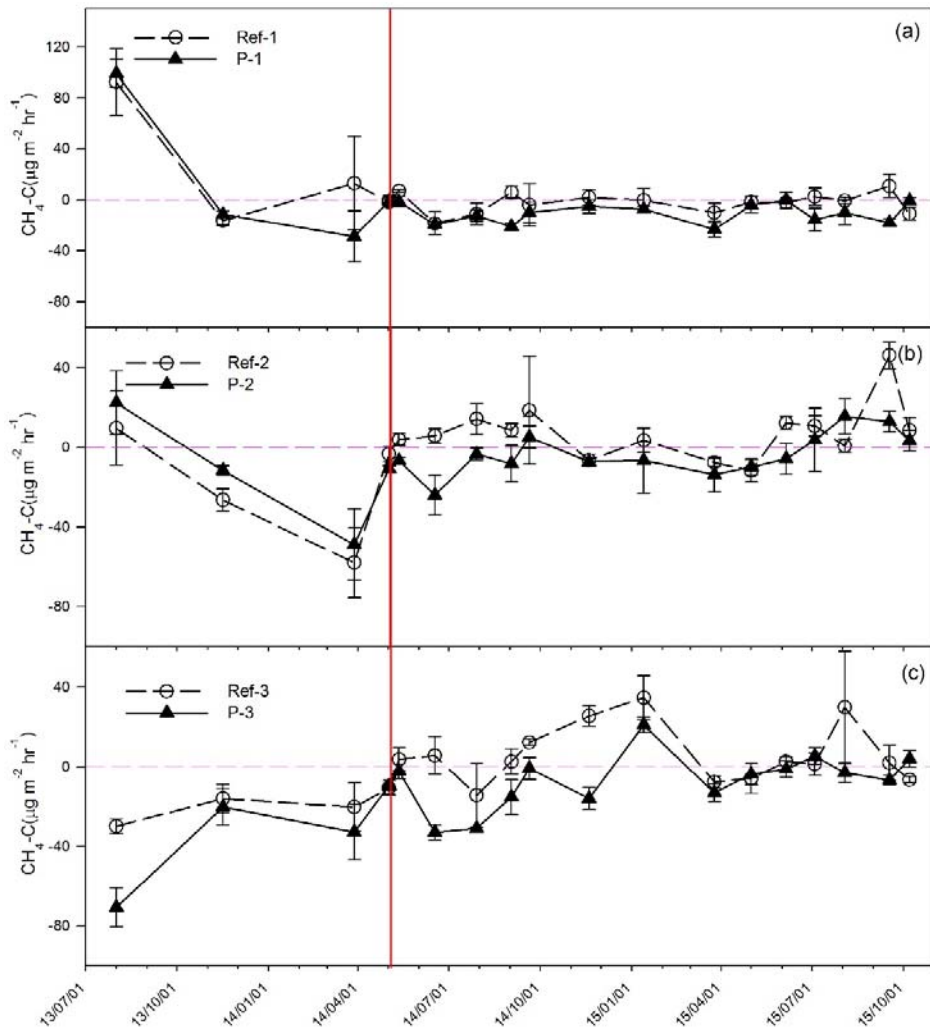
620

621 **Fig. 3** Cumulative N₂O emissions for three blocks in the References (Ref) and P treatments from
 622 summer 2013 to autumn 2015; the red arrows refer to the date of P addition (4 May, 2014).



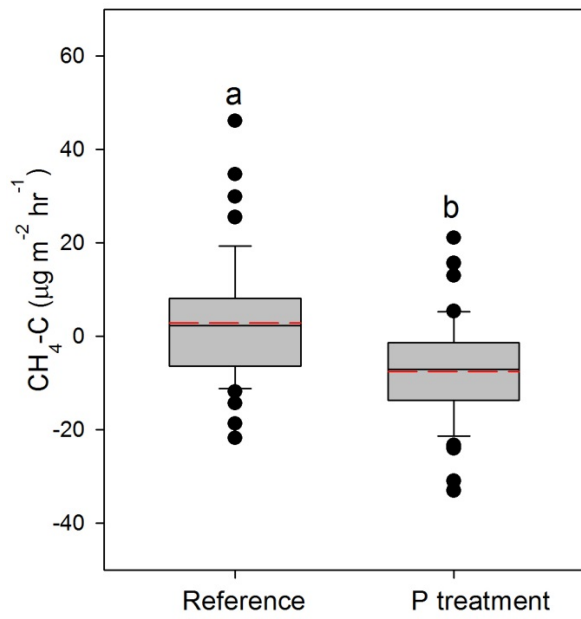
623

624 **Fig. 4** Box whisker plots for N₂O fluxes in the Reference and P treatment throughout 1.5 years
 625 after the P addition; red dashed lines indicate mean values; different letters indicate significant
 626 difference ($p < 0.05$).



627

628 **Fig. 5** Monthly mean CH₄ fluxes (±SE) in the References (Ref) and P treatments for three blocks
 629 (a-c); the horizontal broken line indicates zero flux the red vertical line refers to the date of P
 630 addition (4 May, 2014).



631

632 **Fig. 6** Box whisker plots of CH₄ fluxes in the Reference and P treatment throughout 1.5 years
 633 after the P addition; red dash lines indicate mean values; the different letters indicate significant
 634 difference ($p < 0.05$).