



1 **Phosphorus addition mitigates N<sub>2</sub>O and CH<sub>4</sub> emissions in N-**  
2 **saturated subtropical forest, SW China**

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11 **Abstract**

12 Chronically elevated nitrogen (N) deposition has led to severe nutrient imbalance in forest soils.  
13 Particularly in tropical and subtropical forest ecosystems, increasing N loading has aggravated  
14 phosphorus (P) limitation of biomass production, and has resulted in elevated emissions of  
15 nitrous oxide (N<sub>2</sub>O) and reduced uptake of methane (CH<sub>4</sub>), both of which are important  
16 greenhouse gases. Yet, the interactions of N and P and their effects on GHG emissions remain  
17 understudied. Here, we report N<sub>2</sub>O and CH<sub>4</sub> emissions together with soil chemistry data for the a  
18 period of 18 months following P addition (79 kg P ha<sup>-1</sup> yr<sup>-1</sup>, applied as NaH<sub>2</sub>PO<sub>4</sub> powder) to a N-  
19 saturated, Masson pine-dominated forest at TieShanPing (TSP), Chongqing, SW China. We  
20 observed a significant decline both in NO<sub>3</sub><sup>-</sup> concentrations in soil water (at 5- and 20-cm depths)  
21 and in N<sub>2</sub>O emissions, the latter by 3 kg N ha<sup>-1</sup> yr<sup>-1</sup>. We hypothesize that enhanced N uptake by  
22 plants and soil microbes in response to P addition, results in less available NO<sub>3</sub><sup>-</sup> for  
23 denitrification. By contrast to most other forest ecosystems, TSP is a net source of CH<sub>4</sub>. As for  
24 N<sub>2</sub>O, P addition significantly decreased CH<sub>4</sub> emissions, turning the soil into a net sink. Based on  
25 our data and previous studies in South America and China, we believe that P addition relieves N-  
26 inhibition of CH<sub>4</sub> oxidation. Within the 1.5 years after P addition, no significant increase of  
27 forest growth was observed at TSP, but we cannot exclude that understory vegetation increased.  
28 Our study suggests that P fertilization of acid forest soils could mitigate GHG emissions in  
29 addition to alleviate nutrient imbalances and reduce losses of nitrogen through NO<sub>3</sub><sup>-</sup> leaching and  
30 N<sub>2</sub>O emission.

31 **Key Word:** N<sub>2</sub>O and CH<sub>4</sub> emission, N saturation, Phosphate fertilization, soil CH<sub>4</sub> uptake, acid  
32 forest soil.



## 33 **1 Introduction**

34 Anthropogenic activities have transformed the terrestrial biosphere into a net source of CH<sub>4</sub>, N<sub>2</sub>O  
35 and CO<sub>2</sub>, leading to increased radiative forcing (Montzka et al., 2011; Tian et al., 2016). During  
36 the last decade, atmospheric concentrations of CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O have increased at rates of 1.9 ppm  
37 yr<sup>-1</sup>, 4.8 and 0.8 ppb yr<sup>-1</sup>, respectively (Hartmann et al., 2013). In China, the exponential increase  
38 of reactive nitrogen (N) input into the biosphere since the 1970s has likely led to more carbon (C)  
39 being sequestered in the biosphere (Cui et al., 2013; Shi et al., 2015). However, enhanced  
40 emissions of N<sub>2</sub>O and CH<sub>4</sub> due to chronic N pollution potentially offset the cooling effect by C  
41 sequestration (Liu and Greaver, 2009; Tian et al., 2011).

42 Microbial nitrification and denitrification in soils account for about 60% of N<sub>2</sub>O emissions  
43 globally (Ciais et al., 2013; Hu et al., 2015). Although, microbial activity is often restricted in  
44 low pH soils of unproductive forests, surprisingly large N<sub>2</sub>O emissions have been reported from  
45 acid, upland forest soils in South China (Zhu et al., 2013b). Reported average N<sub>2</sub>O fluxes in  
46 humid, subtropical forests range from 2.0 to 5.4 kg ha<sup>-1</sup> yr<sup>-1</sup> (Fang et al., 2009; Tang et al., 2006;  
47 Zhu et al., 2013b), which by far exceeds global averages for temperate or tropical forest  
48 ecosystems (Werner et al., 2007; Zhuang et al., 2012). This has been attributed to frequently  
49 shifting aeration conditions during monsoonal summers, promoting both nitrification and  
50 denitrification (Zhu et al., 2013b) and to large soil NO<sub>3</sub><sup>-</sup> concentrations due to efficient cycling of  
51 deposited N in acid subtropical soils (Yu et al., 2016).

52 Chronically elevated rates of N deposition (30-65 kg ha<sup>-1</sup> yr<sup>-1</sup>; Xu et al., 2015) have resulted in  
53 strong nutrient imbalances in southern Chinese forests, aggravating phosphorus (P) limitation  
54 (Du et al., 2016). Phosphorous deficiency in N-saturated forests restricts forest growth and thus



55 constrains its capability to retain N (Huang et al., 2015; Li et al., 2016), resulting in ample  
56 amounts of mineral N ( $\text{NH}_4^+$  and  $\text{NO}_3^-$ ) being present in the soil solution. Accordingly, Hall &  
57 Matson (1999) observed larger  $\text{N}_2\text{O}$  emission in P-limited than in N-limited tropical forests after  
58 1 year of repeated N addition. Likewise, previous N manipulation studies in forests of South  
59 China reported pronounced stimulation of  $\text{N}_2\text{O}$  emissions by N addition (Chen et al., 2016;  
60 Wang et al., 2014; Zheng et al., 2016), supporting the idea that P limitation causes forests to be  
61 more susceptible to N saturation and  $\text{N}_2\text{O}$ -N loss. In an N-limited tropical montane forest in  
62 southern Ecuador, P addition alone ( $10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$ ) had no effect on  $\text{N}_2\text{O}$  emissions during the  
63 first two years. However,  $\text{N}_2\text{O}$  emission was smaller when P was added together with N ( $50 \text{ kg N}$   
64  $\text{ha}^{-1} \text{ yr}^{-1}$ ) than treatments with N addition alone (Martinson et al., 2013). After continued  
65 fertilization for three years, also P addition alone reduced  $\text{N}_2\text{O}$  emissions at these sites (Müller et  
66 al., 2015). In tropical China, with high N deposition ( $\sim 36 \text{ kg ha}^{-1} \text{ yr}^{-1}$ ; Mo et al., 2008), P  
67 addition ( $150 \text{ kg P ha}^{-1} \text{ yr}^{-1}$ ) to an old-growth forest revealed a similar pattern, with no initial  
68 effect on  $\text{N}_2\text{O}$  emissions (0-2 years) but a significant longer term effect (3 to 5 years) (Chen et al.,  
69 2016; Zheng et al., 2016). In a secondary tropical forests in South China, Wang et al. (2014)  
70 found no effect on  $\text{N}_2\text{O}$  emissions of P alone ( $100 \text{ kg P ha}^{-1} \text{ yr}^{-1}$ ), and in treatments combining P  
71 with N ( $100 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ ),  $\text{N}_2\text{O}$  emissions even increased during the wet season. Meanwhile,  
72 they observed a significant increase in soil microbial biomass after P addition, which is in line  
73 with previous findings in tropical forest soils of South China (Liu et al., 2012). Thus, they  
74 attributed the stimulating effect of P addition on  $\text{N}_2\text{O}$  emissions to the larger nitrification and  
75 denitrification potential of the increased soil microbial biomass. This was also proposed by Mori  
76 et al. (2014), based on results from a short-term incubation study with P addition, excluding plant  
77 roots.



78 As the sole biogenic sink for CH<sub>4</sub>, upland soils play an important role in balancing terrestrial  
79 CH<sub>4</sub> emissions (Ciais et al., 2013; Dutaur and Verchot, 2007). Atmospheric CH<sub>4</sub> uptake in soil is  
80 mediated by the activity of methanotrophic bacteria, which oxidize CH<sub>4</sub> to CO<sub>2</sub> to gain energy  
81 for growth. Well-drained forest and grassland soils are dominated by yet uncultured, high-  
82 affinity methanotrophs residing in the upper soil layers (Le Mer and Roger, 2010). In addition to  
83 edaphic factors (pH and nutrients), other parameters affecting the diffusion of CH<sub>4</sub> into the soil  
84 (soil structure, moisture, temperature) are believed to be the major controllers for CH<sub>4</sub> uptake  
85 (Smith et al., 2003). A number of studies have shown that excess N affects CH<sub>4</sub> fluxes in forest  
86 soils (Liu and Greaver, 2009; Veldkamp et al., 2013; Zhang et al., 2008b). In general, N addition  
87 promotes CH<sub>4</sub> uptake in N-limited soils by enhancing growth and activity of methanotrophs,  
88 whereas excessive N input and N saturation inhibit CH<sub>4</sub> oxidation on an enzymatic level  
89 (Aronson and Helliker, 2010; Bodelier and Laanbroek, 2004). P addition experiments in N-  
90 enriched soils have shown positive effects on CH<sub>4</sub> uptake (Mori et al., 2013a; Zhang et al., 2011),  
91 but the underlying mechanisms, i.e. whether P addition affects the methanotrophic community in  
92 soils directly or alleviates the N-inhibition effect on CH<sub>4</sub> oxidation through enhanced N uptake  
93 (Mori et al., 2013b; Veraart et al., 2015), remain unresolved.

94 Subtropical forests in South China show strong signs of N saturation, with exceedingly high  
95 NO<sub>3</sub><sup>-</sup> concentrations in soil water (Larssen et al., 2011; Zhu et al., 2013b). Little is known about  
96 how P addition affects N cycling and N<sub>2</sub>O emission in these acidic, nutrient-poor soils. Likewise,  
97 the importance of increased mineral N concentrations for soil-atmosphere exchange of CH<sub>4</sub>, and  
98 how this is affected by P fertilization remain to be elucidated for soils of the subtropics. Here, we  
99 assessed N<sub>2</sub>O and CH<sub>4</sub> fluxes in an N-saturated subtropical forest in SW China under ambient N  
100 deposition and studied the effects of P addition on emission rates, nutrient availability and tree



101 growth. The objectives were i) to quantify ambient N<sub>2</sub>O and CH<sub>4</sub> emissions, ii) to test whether P  
102 affects N cycling in a highly N-saturated forest and iii) to investigate the effect of P addition on  
103 N<sub>2</sub>O and CH<sub>4</sub> emission.



## 104 **2 Materials and Methods**

### 105 **2.1 Site description**

106 The study site “TieShanPing” (TSP) is a 16.2 ha subtropical forest (29° 38' N, 106° 41' E; 450  
107 m a.s.l.), about 25 km northeast of Chongqing, SW China. TSP is a naturally regenerated,  
108 secondary mixed coniferous-broadleaf forest, which developed after clear cutting in 1962  
109 (Larssen et al., 2011). The forest stand is dominated by Masson pine (*Pinus massoniana*) and has  
110 a density of about 800 stems ha<sup>-1</sup> (Huang et al., 2015). Having a monsoonal climate, TSP has a  
111 mean annual precipitation of 1028 mm, and a mean annual temperature of 18.2 °C (Chen and  
112 Mulder, 2007). Most of precipitation (> 70%) occurs during the summer period (April to  
113 September). The soil is a loamy yellow mountain soil, classified as Haplic Acrisol (WRB 2014),  
114 with a thin O horizon (< 2 cm). In the O/A horizon, soil pH is around 3.7, and the mean C/N and  
115 N/P ratios are 17 and 16, respectively. In the AB horizon, which has a slightly higher pH, mean  
116 C/N is well above 20. More details on soil properties are presented in Table 1.

117 Annual N deposition at TSP measured in throughfall varies between 40 to 65 kg ha<sup>-1</sup> and is  
118 dominated by NH<sub>4</sub><sup>+</sup> (Yu et al., 2016). According to regional data, annual P deposition via  
119 throughfall is < 0.40 kg ha<sup>-1</sup> (Du et al., 2016). Strong soil acidification at TSP has resulted in  
120 severe decline in forest growth (Li et al., 2014; Wang et al., 2007), and in abundance and  
121 diversity of ground vegetation (Huang et al., 2015). Pronounced N saturation with strong NO<sub>3</sub><sup>-</sup>  
122 leaching from the top soil has aggravated P deficiency (Huang et al., 2015). The total P content  
123 in the O/A horizon is ~ 300 mg kg<sup>-1</sup>, while P<sub>AI</sub> is smaller than 5 mg kg<sup>-1</sup> (Table 1).

### 124 **2.2 Experimental Design**



125 Three blocks, each having two 20 m \* 20 m plots, were established near a hilltop on a gently  
126 sloping hillside. A 5-m buffer strip separated the two plots in each block. In each block, plots  
127 were assigned ad random to a reference (Ref) and a P treatment. On 4 May 2014, a single dose of  
128 P fertilizer was applied as solid  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$ , at a rate of  $79.5 \text{ kg P ha}^{-1}$ . The amount of P  
129 added was estimated from P adsorption isotherms (Supplementary Materials, Table S1 and  
130 Figure S1), to ensure significantly increased available P in TSP soil. To apply P fertilizer evenly,  
131 we divided each plot into a 5 m \* 5 m grid and broadcasted the powdered fertilizer by hand in  
132 each grid cell. The P dose applied at TSP was intermediate as compared to the  $10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$   
133 applied by Müller et al. (2015) to a mountain forest in Ecuador and the  $150 \text{ kg P ha}^{-1} \text{ yr}^{-1}$  applied  
134 by Zheng et al. (2016) to a subtropical forest in South China.

135 Together with the addition of phosphate, the P-treated plots also received  $59.0 \text{ kg ha}^{-1}$  of sodium  
136 (Na). One month after the fertilizer application,  $\text{Na}^+$  concentrations in soil water of the P  
137 treatments were about  $5 \text{ mg L}^{-1}$  at 5-cm depth and  $3 \text{ mg L}^{-1}$  at 20-cm depth (Table S2). Although  
138 somewhat larger than in the reference plots, the  $\text{Na}^+$  concentration in soil water of the P  
139 treatments are unlikely to have exerted a strong negative impact on plant and microbial activities.

### 140 **2.3 Sample collection and analyses**

141 Within each plot, triplicates of ceramic lysimeters (P80; Staatliche Porzellanmanufaktur, Berlin)  
142 were installed at 5- and 20-cm soil depths in August 2013. To obtain water samples, 350-ml  
143 glass bottles with rubber stoppers were pre-evacuated, using a paddle pump, and connected to the  
144 lysimeters for overnight sampling. Between November 2013 and October 2015, we sampled soil  
145 pore water bi-monthly in the winter season and monthly during the growing season. All water  
146 samples were kept frozen during storage and transport. Concentrations of  $\text{NH}_4^+$ ,  $\text{NO}_3^-$ , potassium



147 (K<sup>+</sup>, calcium (Ca<sup>2+</sup>), and magnesium (Mg<sup>2+</sup>) in soil water were measured at the Research Center  
148 for Eco-Environmental Sciences (RCEES), Chinese Academy of Sciences, Beijing, using ion  
149 chromatography (DX-120 for cations and DX-500 for anions).

150 In August 2013, soils from the O/A (0-3 cm), AB (3-8 cm) and B (8-20 cm) horizons were  
151 sampled near the lysimeters for soil analysis. Total P and plant-available P contents were  
152 monitored in samples collected from the O/A horizons every six months, starting two days  
153 before P addition. Soil samples were kept cold (< 4 °C) during transport and storage. Before  
154 analysis, soil samples were air dried and sieved (2 mm). Soil pH was measured in soil  
155 suspensions (10 g dry soil and 50 ml deionized water) using a pH meter (PHB-4, Leici, China).  
156 Total soil C and N contents were determined on dried and milled samples, using a LECO  
157 elemental analyzer (TruSpec@CHN, USA). To measure total P, 1 g dry soil was digested with 5  
158 ml of 6 M H<sub>2</sub>SO<sub>4</sub> (Singh et al., 2005) and measured as ortho-phosphate by the molybdenum blue  
159 method (Murphy and Riley, 1962). Ammonium lactate (0.01 M)-extractable P and H<sub>2</sub>O-  
160 extractable P (P<sub>Al</sub> and P<sub>H<sub>2</sub>O</sub>, respectively) were measured as ortho-phosphate after extraction (1.5  
161 g dry soil in 50 ml solution) (Singh et al., 2005). Ammonium oxalate (0.2 M)-extractable Fe, Al  
162 and P were measured by inductive coupled plasma (7500; Agilent) after extraction (1.5 g dry soil  
163 in 50 ml solution).

164 From August 2013 onwards, we measured N<sub>2</sub>O and CH<sub>4</sub> emissions in triplicate in micro-plots  
165 close to the lysimeters, using static chambers (Zhu et al., 2013b). To investigate the immediate  
166 effect of P addition on N<sub>2</sub>O emissions, we sampled the gas emissions once before (2 May) and  
167 three times (7, 10 and 12 May) after the P application. Gas samples (20 ml) were taken 1, 5, 15  
168 and 30 minutes after chamber deployment and injected into pre-evacuated glass vials (12 ml)  
169 crimp-sealed with butyl septa (Chromacol, UK), maintaining overpressure to avoid



170 contamination during sample transport. Mixing ratios of N<sub>2</sub>O, CO<sub>2</sub> and CH<sub>4</sub> were analyzed using  
171 a gas chromatograph (Model 7890A, Agilent, US) at RCEES, equipped with an ECD for  
172 detection of N<sub>2</sub>O (at 375 °C with 25 ml min<sup>-1</sup> Ar/CH<sub>4</sub> as make up gas), a FID for CH<sub>4</sub> (250 °C;  
173 20 ml min<sup>-1</sup> N<sub>2</sub> as make-up gas) and a TCD for CO<sub>2</sub>. Exchange rates between soil and  
174 atmosphere (emission/uptake) were calculated from measured concentration change in the  
175 chambers over time, applying linear or polynomial fits to the concentration data. Cumulative  
176 N<sub>2</sub>O emissions over time were estimated by linear interpolation between measurement dates  
177 (Zhu et al., 2013b).

178 From October 2013 onwards, litterfall was collected during the first week of every month in five  
179 replicates per plot. Litterfall collectors were made of 1 m<sup>2</sup> nylon nets (1 mm mesh size), held in  
180 place by four wooden poles 0.8 m above the ground. Fresh litter was dried at 65°C. In early  
181 November 2013 and 2014 (at the end of the growing season), we collected current-year pine  
182 needles from several branches of three trees in each plot. The collected needles were dried at  
183 65 °C and the dry weight of 500 needles was determined. A subsample was dried at 80 °C and  
184 finely milled prior to chemical analysis at the Chinese Academy of Forestry. Total C and N were  
185 measured using an elemental analyzer (FLASH 2000; Thermo Scientific; USA). The contents of  
186 K, Ca, Mg and P in the needles were determined by ICP-AES (IRIS Intrepid II; Thermo  
187 Scientific; USA) after digesting 0.25 g dry weight samples with 5 ml of ultra-pure nitric acid. In  
188 November 2013, and 2014, and in February of 2015, we measured the height and the diameter at  
189 breast height (DBH) of 6 to 10 Masson pines (only those with DBH > 5 cm) at each plot. These  
190 data were used to estimate the standing biomass of Masson pines based on standard allometric  
191 equations (Li et al., 2011; Zeng et al., 2008).



192 Daily average air temperature and sum of precipitation were monitored by a weather station  
193 (WeatherHawk 232, USA) placed on the roof at the local forest bureau, in about 1 km distance  
194 from the sampling site (Yu et al., 2016).

#### 195 **2.4 Statistical analyses**

196 Statistical analyses were performed with Minitab 16.2.2 (Minitab Inc., USA). All data were  
197 tested for normality (Kolmogorov-Smirnov's test) and homoscedasticity (Levene's test) before  
198 further analysis. If not normally distributed, the data were then normalized by logarithmic  
199 transformation. Due to heterogeneity between blocks, data on gas fluxes and mineral N  
200 concentrations are presented separately for each block. One-way ANOVA was used to evaluate  
201 differences in gas fluxes, as well as nutrient concentrations in soil, soil water and plants between  
202 treatments and blocks. Significance levels were set to  $p < 0.05$ , if not specified otherwise.



## 203 **3 Results**

### 204 **3.1 Nutrient concentrations in soil and soil water**

205 Addition of P resulted in a significant increase in soil P content in the O/A horizon, both as  $P_{AI}$   
206 and total P (Table 2). However, after 15 months, only  $P_{AL}$  indicated an enhanced P status, while  
207 total soil P did not differ significantly from background values at the reference sites. P addition  
208 had no significant effect on soil pH, or soil C and N content. The  $NO_3^-$  concentration in soil  
209 water collected at 5 cm depth varied seasonally, with significantly greater values (30-40 mg N L<sup>-1</sup>)  
210 towards the start of the growing season in 2015 (April, Fig. S2), but not in 2014, likely due to  
211 dilution by abundant precipitation in February to March 2014. Addition of P resulted in  
212 significantly smaller  $NO_3^-$  concentrations in soil water at 5 and 20 cm depth in blocks 2 and 3 but  
213 not in block 1 (Fig. 1). In general, the concentration of  $NH_4^+$  in soil water was small (< 0.6 mg L<sup>-1</sup>)  
214 and not affected by P addition (Fig. S3). At both depths, mean soil water concentrations of  
215  $Mg^{2+}$  and  $Ca^{2+}$  were significantly smaller in the P-treated than the reference plots, and the sum of  
216 charge of base cations declined significantly in response to P addition (Fig. S4).

### 217 **3.2 N<sub>2</sub>O and CH<sub>4</sub> fluxes: effects of P addition**

218 During the experimental period, N<sub>2</sub>O fluxes varied seasonally (Fig. 2), showing a significant  
219 relationship with daily precipitation (Fig. S5a), but not with daily mean temperature (Fig. S4b).  
220 In the reference plots, mean N<sub>2</sub>O fluxes were generally below 50 μg N m<sup>-2</sup> hr<sup>-1</sup> in the dry, cool  
221 season, but reached up to 600 μg N m<sup>-2</sup> hr<sup>-1</sup> in the growing season (Fig. 2). Average and  
222 cumulative fluxes of N<sub>2</sub>O differed greatly between the three blocks (Figs. 3 and 4, respectively),  
223 with the greatest annual emission observed in the reference plot (7.9 kg N ha<sup>-1</sup>) of block 2. Mean  
224 N<sub>2</sub>O fluxes during the 1.5 years after P addition were smaller in the P treatments than in the



225 references, the differences being significant in blocks 2 and 3 (Fig. 3). Cumulative N<sub>2</sub>O  
226 emissions showed that P addition resulted in a decrease in N<sub>2</sub>O emission by about 3 kg N ha<sup>-1</sup> yr<sup>-1</sup>,  
227 which is a 57% reduction on average (Fig. 4). No immediate effects (within days) of P addition  
228 on N<sub>2</sub>O emission were observed (Fig. S6).

229 CH<sub>4</sub> fluxes varied greatly between blocks (Fig. 5). Net-emission of CH<sub>4</sub> was observed in  
230 summer 2013 (~ 80 µg C m<sup>-2</sup> hr<sup>-1</sup>) in blocks 1 and 2, whereas block 3 showed CH<sub>4</sub> uptake. From  
231 spring 2014 until October 2015, CH<sub>4</sub> fluxes were less variable in all blocks, with values  
232 fluctuating around zero. A longer period of net-emission was observed in block 3 during the dry  
233 season 2014. The fluxes did not correlate with either precipitation or air temperature (Fig.  
234 S5c&d). In the 1.5 years following P addition, mean CH<sub>4</sub> fluxes indicated net CH<sub>4</sub> emission (~  
235 +3.8 µg C m<sup>-2</sup> hr<sup>-1</sup>) in the reference plots (except for block 1), whereas net CH<sub>4</sub> uptake (~ -6.5 µg  
236 C m<sup>-2</sup> hr<sup>-1</sup>) was observed in all P- treated plots (Fig. 6). The suppressing effect of P addition on  
237 CH<sub>4</sub> emission was significant in blocks 2 and 3, similar to what was found for NO<sub>3</sub><sup>-</sup>  
238 concentrations and N<sub>2</sub>O fluxes.

### 239 3.3 The effect of P addition on tree growth

240 Throughout the 2-year experimental period, we observed no change in tree biomass (138 t ha<sup>-1</sup>)  
241 in response to P addition (Table S3). Likewise, there was no effect of P treatment on the 500-  
242 needle weight (13 g on average). Between the two samplings in 2013 and 2014, we found  
243 differences in chemical composition of the pine needles, but this effect was not linked to P  
244 addition. Also, the C/N and N/P ratios of the needles (40 and 16, respectively) were hardly  
245 affected by P addition. Monthly litterfall varied seasonally (Fig. S7), but no significant difference  
246 was found between the reference and the P treated plots.



## 247 4 Discussion

248 Background N<sub>2</sub>O emission rates in the reference plots were relatively large, with mean values of  
249 40 to 120  $\mu\text{g N m}^{-2} \text{hr}^{-1}$  (Fig. 3). This is within the range previously reported for well-drained  
250 hillslope soils at TSP (Zhu et al., 2013b), but greater than the rates reported for other forests in  
251 South China. For instance, N<sub>2</sub>O emission rates averaged 37  $\mu\text{g N m}^{-2} \text{hr}^{-1}$  in unmanaged sites at  
252 Dinghushan (Fang et al., 2009; Tang et al., 2006) and up to 50  $\mu\text{g N m}^{-2} \text{hr}^{-1}$  in N-fertilized sites  
253 (Zhang et al., 2008a). TSP reference plots emitted on average 5.3 kg N ha<sup>-1</sup> yr<sup>-1</sup> (Fig. 4), which is  
254 about 10% of the annual N deposition (50 kg ha<sup>-1</sup> yr<sup>-1</sup>) (Huang et al., 2015). These fluxes were  
255 well above average fluxes reported for tropical rainforests (Werner et al., 2007). High N<sub>2</sub>O  
256 emissions at TSP are likely due to the large N deposition rates (Huang et al. 2015), as suggested  
257 by the similar trends indicated by data from a wide range of ecosystems (Liu et al., 2009). Also,  
258 warm-humid conditions during monsoonal summers may stimulate N<sub>2</sub>O emissions (Ju et al.,  
259 2011), as monsoonal rainstorms triggered peak fluxes (Fig. S5a) (Pan et al., 2003). The positive  
260 correlation between precipitation and N<sub>2</sub>O emission peaks may indicate the importance of  
261 denitrification as the dominant N<sub>2</sub>O source. This is supported by recent <sup>15</sup>N tracing experiments  
262 at TSP (Zhu et al., 2013a; Yu et al., submitted).

263 Addition of P caused a significant decline in soil mineral N (predominantly NO<sub>3</sub><sup>-</sup>) in two of three  
264 blocks (Fig. 2), particularly during summers, when NO<sub>3</sub><sup>-</sup> concentrations were relatively high (Fig.  
265 S2). At the same time, annual N<sub>2</sub>O emissions decreased by more than 50% (Figs. 3 and 4). These  
266 findings are consistent with a number of previous studies (Baral et al., 2014; Hall and Matson,  
267 1999; Mori et al., 2014). The reduction of N<sub>2</sub>O emissions in P treated soils was attributed to  
268 decreased mineral N content, most likely due to stimulated plant uptake and/or microbial



269 assimilation. It is noteworthy, however, that there was no significant correlation between N<sub>2</sub>O  
270 emission rates and soil water NO<sub>3</sub><sup>-</sup> concentration in our study (Figs. 2 and S2), suggesting that  
271 the suppressing effect of P on N<sub>2</sub>O emissions was indirect, probably by affecting the competition  
272 for mineral N between plant roots and microbes (Zhu et al., 2016). In contrast to our study, P-  
273 addition experiments in South Ecuador (Martinson et al., 2013) and South China (at Dinghushan  
274 Biosphere Reserve (DHSBR); Zheng et al., 2016) found no effect of a single P addition on N<sub>2</sub>O  
275 emission during the first two years after application. However, significant reduction in N<sub>2</sub>O  
276 emission was observed after three to five years with continuous P addition, both at the  
277 Ecuadorian and the Chinese site (Chen et al., 2016; Müller et al., 2015). For the montane forest  
278 site in Ecuador, the observed delay in N<sub>2</sub>O emission response to P addition may be explained by  
279 the moderate amount of P added (10 kg P ha<sup>-1</sup> yr<sup>-1</sup>; Martinson et al., 2013). Moreover, the  
280 experiments were conducted in a forest with low ambient N deposition (~ 10 kg N ha<sup>-1</sup> yr<sup>-1</sup>) and  
281 N<sub>2</sub>O fluxes (~ 0.36 kg N ha<sup>-1</sup> yr<sup>-1</sup> in the reference plot) (Martinson et al., 2013; Müller et al.,  
282 2015). By contrast, the DHSBR site in South China receives 36 kg of atmospheric N ha<sup>-1</sup> yr<sup>-1</sup>,  
283 which is only slightly smaller than the N deposition at our site (Huang et al., 2015), and showed  
284 larger N<sub>2</sub>O emission rate than the Ecuadorian site (~ 0.88 kg N ha<sup>-1</sup> yr<sup>-1</sup> in the reference plot;  
285 Zheng et al., 2016). However, forests do not always display a straightforward relationship  
286 between N deposition and N<sub>2</sub>O emissions. Manipulation experiments in the European NITREX  
287 project, for instance, revealed a much stronger correlation of N<sub>2</sub>O emissions with soil NO<sub>3</sub><sup>-</sup>  
288 leaching than with N deposition (Gundersen et al., 2012). Indeed, KCl-extractable mineral N at  
289 the DHSBR site (~ 40 mg kg<sup>-1</sup>; Zheng et al., 2016) were several-fold smaller than at our site (>  
290 100 mg kg<sup>-1</sup>; Zhu et al., 2013b), indicating that DHSBR is less N-saturated than TSP. This  
291 suggests that the response of N<sub>2</sub>O emission to P addition might depend on the N status of the soil.



292 The fact that numerous studies found apparent suppression of N<sub>2</sub>O emission in short-term  
293 experiments (< 2 years) in N + P treatments, but not in treatments with P alone, supports this  
294 idea (Müller et al., 2015; Zhang et al., 2014b; Zheng et al., 2016).

295 Other studies have observed increased N<sub>2</sub>O emissions upon P addition (Mori et al., 2013c; Wang  
296 et al., 2014). In an *Acacia mangium* plantation, fertilized with P, Mori et al. (2013b&c) found  
297 that N<sub>2</sub>O emissions were stimulated in the short-term but reduced in the long-term. While  
298 suppression of N<sub>2</sub>O emission by P has been attributed to increased plant N uptake (Mori et al.,  
299 2014), increased N<sub>2</sub>O emission are generally explained by enhanced microbial biomass (Liu et  
300 al., 2012) and denitrification activities (Ehlers et al., 2010; He and Dijkstra, 2015). N<sub>2</sub>O  
301 emissions measured frequently after P addition at our site in May 2014 were not different from  
302 fluxes in untreated reference plots (Fig. S5). This may indicate that plant uptake at TSP is more  
303 important for the effect of P addition on N<sub>2</sub>O emissions than changes in microbial activity, which  
304 are expected to occur more rapidly.

305 Two of three reference plots at TSP showed net CH<sub>4</sub> emission for extended periods of the year  
306 (Figs. 5 and 6). Also, long-term CH<sub>4</sub> fluxes sampled between 2012 and 2014 on TSP hillslopes  
307 near-by (Fig. S8; Zhu et al., unpublished data) showed net CH<sub>4</sub> emission. This is in contrast to  
308 the generally reported CH<sub>4</sub> sink function of forested upland soils (Ciais et al., 2013; Dutaur and  
309 Verchot, 2007). For example, CH<sub>4</sub> uptake rates reported for South Chinese forest soils range  
310 from 30 to 60 µg C m<sup>-2</sup> hr<sup>-1</sup> (Fang et al., 2009; Tang et al., 2006; Zhang et al., 2014a). As CH<sub>4</sub>  
311 fluxes at our sites were not correlated with climatic factors (Fig. S5c and d), CH<sub>4</sub> emissions  
312 cannot be explained by transiently wet conditions. One reason for the net-CH<sub>4</sub> emission observed  
313 at TSP could be inhibition of CH<sub>4</sub> oxidation activity by NH<sub>4</sub><sup>+</sup>, as reported previously (Bodelier  
314 and Laanbroek, 2004; Zhang et al., 2014a). The concentration of NH<sub>4</sub><sup>+</sup> in the soil water was



315 rather small ( $< 0.5 \text{ g L}^{-1}$ ; Fig. S3), which does not preclude, however, that  $\text{NH}_4^+$  availability from  
316 the soil exchangeable pool is high. Zhu et al. (2013b) found extraordinarily high KCl-  
317 extractable  $\text{NH}_4^+$  in TSP surface soils, likely reflecting the large atmospheric  $\text{NH}_4^+$  input at our site  
318 (Huang et al., 2015).

319 P addition had a significant impact on  $\text{CH}_4$  fluxes, changing the soil from a net source to a net  
320 sink on an annual basis (Fig. 6). However, the uptake rates of  $\text{CH}_4$  in the P treatments remained  
321 smaller than those reported for forest soils in tropical China (Tang et al., 2006; Zhang et al.,  
322 2008b). The stimulating effect of P addition on  $\text{CH}_4$  uptake is consistent with previous studies  
323 (Mori et al., 2013a, 2013b; Zhang et al., 2011), and has been attributed to lessening the  $\text{NH}_4^+$   
324 inhibition of methane oxidation. Unfortunately, we did not measure KCl-extractable  $\text{NH}_4^+$  in our  
325 study, but a decline of available  $\text{NH}_4^+$ , which is the substrate for nitrification, is likely as  $\text{NO}_3^-$   
326 concentrations in soil water were significantly smaller with in the P-treatments (Fig. 2). P  
327 addition may also result in a change of the taxonomic composition of the methane oxidizing  
328 community (Mori et al., 2013a; Veraart et al., 2015). Alternatively,  $\text{CH}_4$  oxidation may be  
329 stimulated by increased  $\text{CH}_4$  diffusion into the soil, due to enhanced root growth and increased  
330 transpiration in P-amended plots (Zhang et al., 2011). Given the high degree of N saturation of  
331 TSP forest (Huang et al., 2015), it is likely that the reason for the observed reduction in  $\text{CH}_4$   
332 emissions in response to P fertilization was due to alleviating the  $\text{NH}_4^+$  inhibition of the methane  
333 monooxygenase enzyme (Veldkamp et al., 2013), rather than a direct P-stimulation of  
334 methanotrophic activity (Veraart et al., 2015).

335 Shortly after fertilizer application, we observed a modest, albeit significant increase of  $\text{Na}^+$   
336 concentration in soil water (Table S2). Other studies have documented the toxicity of excess  $\text{Na}^+$   
337 in soil water to plant and microbial activities (Rengasamy et al., 2003; Wong et al., 2008).



338 However, the occurrence of  $\text{Na}^+$  toxicity at the treated plots, affecting N turnover processes, is  
339 unlikely, as  $\text{Na}^+$  concentrations in soil water, within one month after application (Table S2), did  
340 not exceed  $5 \text{ mg L}^{-1}$ , far below the values commonly assumed to indicate the toxicity threshold  
341 ( $40$  to  $100 \text{ mg L}^{-1}$ ) (Bernstein 1975). The frequent precipitation in the humid forest of this study  
342 (Yu et al., 2016), both prior and following the addition of  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  (Fig. 2), efficiently  
343 diluted and leached  $\text{Na}^+$ , thus minimizing toxic effects.

344 P application significantly increased plant-available P in the P-limited TSP soil (Table 2).  
345 Meanwhile, concentrations of leachable base cations ( $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ) in soil water decreased  
346 (Fig. S4), as expected from the reduction of  $\text{NO}_3^-$  concentrations in the P-treatments (Mochoge  
347 and Beese, 1986). We observed no sign of stimulated forest growth or increased N uptake by  
348 plants within the relatively short period of our study (Table S3 and Fig. S7), which makes it  
349 difficult to link the observed reduction in mineral N in the soil solution to plant growth (Fig. 2).  
350 When interpreting the observed P effect on  $\text{NO}_3^-$  concentrations in soil water, several aspects  
351 need to be considered. Firstly, two years of observation may be too short to detect any significant  
352  $\text{NO}_3^-$  uptake by plants, given the commonly large variabilities in tree biomass estimates  
353 (Alvarez-Clare et al., 2013; Huang et al., 2015). Secondly, a significant proportion of the added  
354 P, and of excess N, may have been assimilated by the understory biomass, which was not  
355 assessed in this study. Previously, understory vegetation has been reported to quickly respond to  
356 P addition (Fraterrigo et al., 2011). Thirdly, as long-term N saturation and acidification at TSP  
357 has reduced the forest health (Lu et al., 2010; Wang et al., 2007), we may not expect immediate  
358 response of forest growth to P addition. Large needle N/P ratios (17-22, Table S3) indicated that  
359 P limitation for tree growth was not relieved 1.5 years after P addition (Li et al., 2016). Therefore,



360 enhanced N uptake by understory growth and/or soil microbial biomass may have been the main  
361 mechanisms responsible for observed  $\text{NO}_3^-$  decline in the P-treated soil (Hall & Matson 1999).

362 Overall, our study demonstrates that chronically high N deposition has transformed TSP soils to  
363 a regional hotspot for  $\text{N}_2\text{O}$  and  $\text{CH}_4$  emission. Within the short experimental period of 1.5 years,  
364 P fertilization was shown to significantly decrease  $\text{NO}_3^-$  concentrations in soil water and to  
365 reduce both  $\text{N}_2\text{O}$  and  $\text{CH}_4$  emissions. These findings provide a promising starting point for  
366 improving forest management towards GHG abatement targets, taking into account the P and N  
367 status of subtropical soils in the region.



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375 **Reference**

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 584 **Table 1** Background soil properties of the experimental plots at Tieshanping (TSP). Values are

	Soil Layer	pH	Total C g kg <sup>-1</sup>	Total N g kg <sup>-1</sup>	Total P mg kg <sup>-1</sup>	C/N	N/P
Block 1	O/A (0-3 cm)	3.7 (0.1)	80.7 (32.3)	4.8 (1.7)	308 (57)	17.0 (2.5)	15.5 (5.7)
	AB (3-8 cm)	3.8 (0.0)	23.9 (9.3)	1.3 (0.6)	-*	20.0 (3.0)	-
	B (8-20 cm)	3.9 (0.2)	8.6 (1.2)	< 0.05	-	-	-
Block 2	O/A (0-3 cm)	3.6 (0.1)	77.6 (13.4)	4.7 (0.8)	297 (44)	16.7 (1.3)	15.7 (2.8)
	AB (3-8 cm)	3.7 (0.1)	20.2 (5.3)	1.0 (0.3)	-	21.4 (3.3)	-
	B (8-20 cm)	3.9 (0.1)	7.1 (1.6)	< 0.05	-	-	-
Block 3	O/A (0-3 cm)	3.6 (0.1)	67.0 (15.5)	3.8 (0.8)	223 (45)	17.4 (0.6)	17.2 (3.7)
	AB (3-8 cm)	3.6 (0.1)	21.0 (7.9)	1.1 (0.5)	-	24.5 (4.6)	-
	B (8-20 cm)	3.8 (0.1)	7.2 (1.5)	< 0.05	-	-	-
	Soil Layer	P <sub>H2O</sub> mg kg <sup>-1</sup>	P <sub>Al</sub> mg kg <sup>-1</sup>	Al <sub>ox</sub> mg kg <sup>-1</sup>	Fe <sub>ox</sub> mg kg <sup>-1</sup>	P <sub>ox</sub> mg kg <sup>-1</sup>	P <sub>ox</sub> / (Al <sub>ox</sub> + Fe <sub>ox</sub> )
Block 1	O/A (0-3 cm)	< 5.0	5.8 (1.4)	1700 (513)	1933 (350)	85.8 (22.6)	0.025 (0.008)
	AB (3-8 cm)	< 5.0	2.1 (0.6)	1217 (243)	1692 (493)	47.1 (22.0)	0.016 (0.007)
	B (8-20 cm)	< 5.0	< 1.0	1083 (90)	1158 (249)	29.3 (28.6)	0.012 (0.011)
Block 2	O/A (0-3 cm)	< 5.0	5.9 (1.0)	1500 (238)	1792 (215)	79.2 (21.5)	0.024 (0.007)
	AB (3-8 cm)	< 5.0	1.6 (0.4)	925 (149)	1517 (320)	37.2 (10.7)	0.016 (0.006)
	B (8-20 cm)	< 5.0	< 1.0	892 (209)	1033 (413)	16.1 (10.5)	0.009 (0.007)
Block 3	O/A (0-3 cm)	< 5.0	4.1 (0.9)	1367 (180)	1667 (168)	50.7 (10.9)	0.017 (0.003)
	AB (3-8 cm)	< 5.0	4.4 (4.0)	1075 (128)	1350 (150)	24.8 (8.3)	0.010 (0.002)
	B (8-20 cm)	< 5.0	< 1.0	992 (130)	875 (138)	8.0 (2.0)	0.004 (0.001)

 585 means and standard deviations in parenthesis (n = 6)<sup>†</sup>.

586

 587 P<sub>H2O</sub> = Water extractable P, P<sub>Al</sub> = Ammonium extractable P,

 588 Al<sub>ox</sub> = Oxalate extractable Al, Fe<sub>ox</sub> = Oxalate extractable Fe, P<sub>ox</sub> = Oxalate extractable P.

 589 <sup>†</sup> Soils were sampled in August 2013.

590 \* Data not available


 591 **Table 2** Soil pH, C, N and P contents in the O/A horizon (0-3 cm) in Reference and phosphate (P)

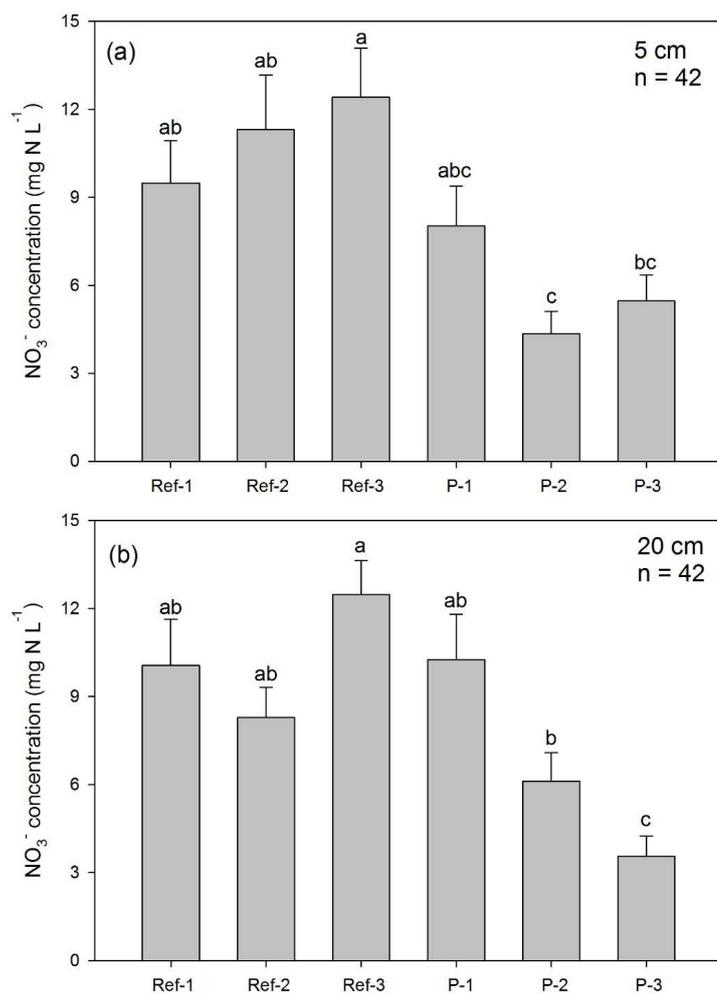
		pH	Total C g kg <sup>-1</sup>	Total N g kg <sup>-1</sup>	C/N	P <sub>AI</sub> mg kg <sup>-1</sup>	Total P mg kg <sup>-1</sup>
13/08/02	Ref	3.7 (0.1) <sup>bc†</sup>	8.3 (2.3) <sup>ab</sup>	0.5 (0.1) <sup>bcd</sup>	16.9 (1.1) <sup>bcd</sup>	5.4 (1.4) <sup>c</sup>	292 (46) <sup>bc</sup>
	P	3.6 (0.1) <sup>c</sup>	6.7 (2.0) <sup>b</sup>	0.4 (0.1) <sup>bd</sup>	17.1 (2.1) <sup>bc</sup>	5.1 (1.3) <sup>c</sup>	260 (70) <sup>c</sup>
14/05/02	Ref	3.7 (0.1) <sup>abc</sup>	12.2 (4.2) <sup>a</sup>	0.9 (0.3) <sup>a</sup>	13.7 (1.5) <sup>c</sup>	19.0 (8.0) <sup>c</sup>	336 (65) <sup>bc</sup>
	P	3.8 (0.2) <sup>abc</sup>	9.0 (3.5) <sup>ab</sup>	0.7 (0.2) <sup>abc</sup>	14.2 (2.8) <sup>de</sup>	13.7 (5.2) <sup>c</sup>	270 (72) <sup>bc</sup>
14/05/10	Ref	3.8 (0.1) <sup>abc</sup>	9.9 (2.1) <sup>ab</sup>	0.7 (0.2) <sup>ab</sup>	14.0 (0.7) <sup>c</sup>	15.4 (7.0) <sup>c</sup>	304 (49) <sup>bc</sup>
	P	3.9 (0.3) <sup>ab</sup>	8.0 (1.9) <sup>ab</sup>	0.6 (0.1) <sup>bcd</sup>	14.3 (1.3) <sup>cde</sup>	174 (114) <sup>a</sup>	572 (242) <sup>a</sup>
14/12/02	Ref	3.8 (0.1) <sup>abc</sup>	10.5 (3.6) <sup>ab</sup>	0.7 (0.3) <sup>ab</sup>	14.5 (1.3) <sup>cde</sup>	14.2 (7.4) <sup>c</sup>	328 (102) <sup>bc</sup>
	P	3.9 (0.2) <sup>abc</sup>	9.5 (2.1) <sup>ab</sup>	0.7 (0.1) <sup>abc</sup>	14.0 (0.8) <sup>c</sup>	66 (24) <sup>ab</sup>	442 (106) <sup>ab</sup>
15/08/02	Ref	3.9 (0.2) <sup>ab</sup>	8.3 (2.2) <sup>ab</sup>	0.4 (0.1) <sup>cd</sup>	20.5 (2.5) <sup>a</sup>	13.4 (6.2) <sup>c</sup>	291 (61) <sup>bc</sup>
	P	4.0 (0.2) <sup>a</sup>	6.5 (1.9) <sup>b</sup>	0.3 (0.1) <sup>d</sup>	19.7 (2.2) <sup>ab</sup>	57 (36) <sup>ab</sup>	383 (136) <sup>bc</sup>

592 treatments. Values are means and standard deviations in parenthesis (n = 9).

593

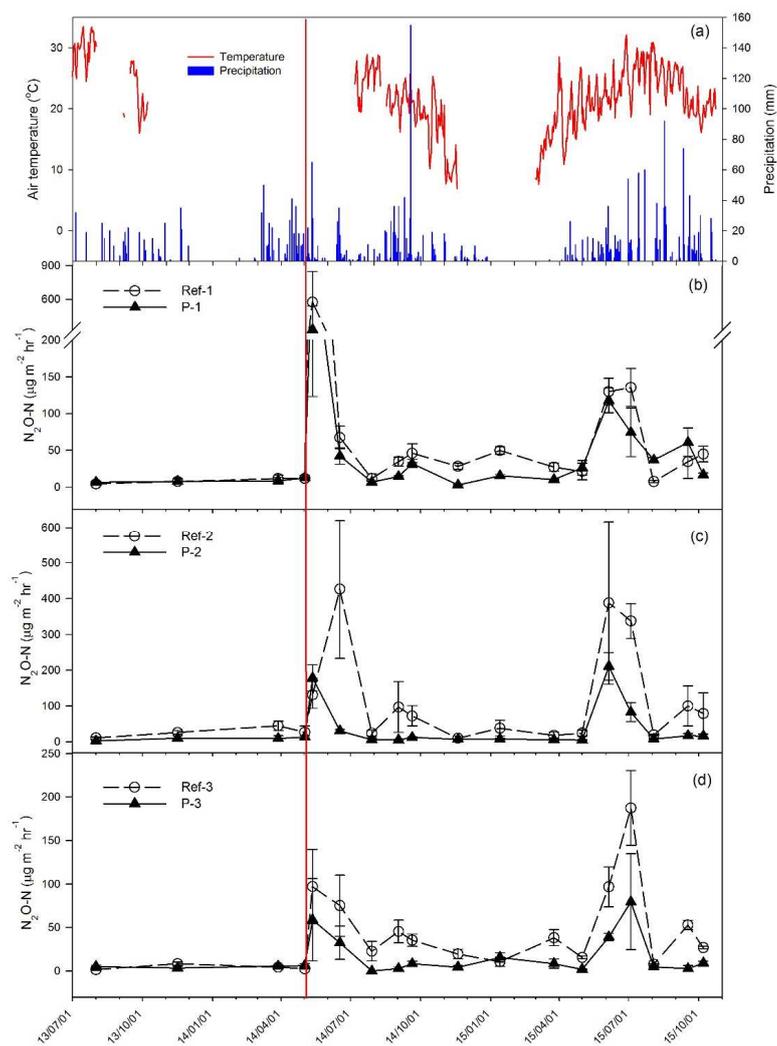
 594 <sup>0</sup>P addition was conducted on 14/05/04, after the first two sampling dates.

 595 <sup>†</sup> Different letters indicate significance in difference.



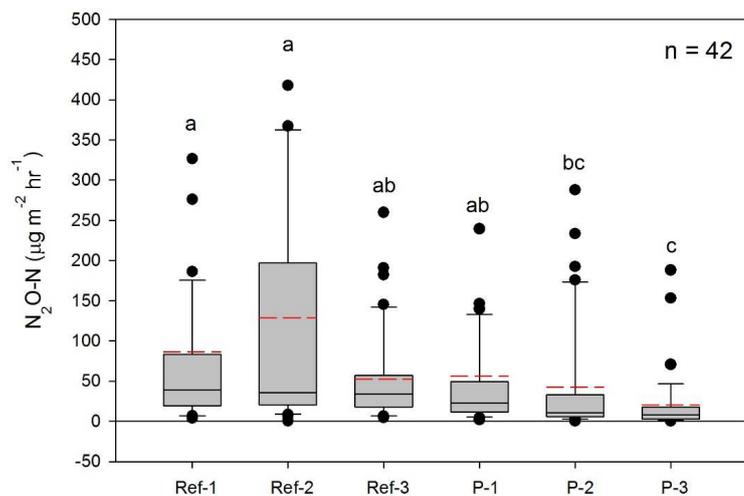
596

597 **Fig. 1** Mean  $\text{NO}_3^-$  concentrations in soil water at 5 (a) and 20 (b) cm depths from three blocks  
598 with Reference and P treatment, 1.5 years after P addition; different letters indicate significant  
599 differences between the treatments and blocks



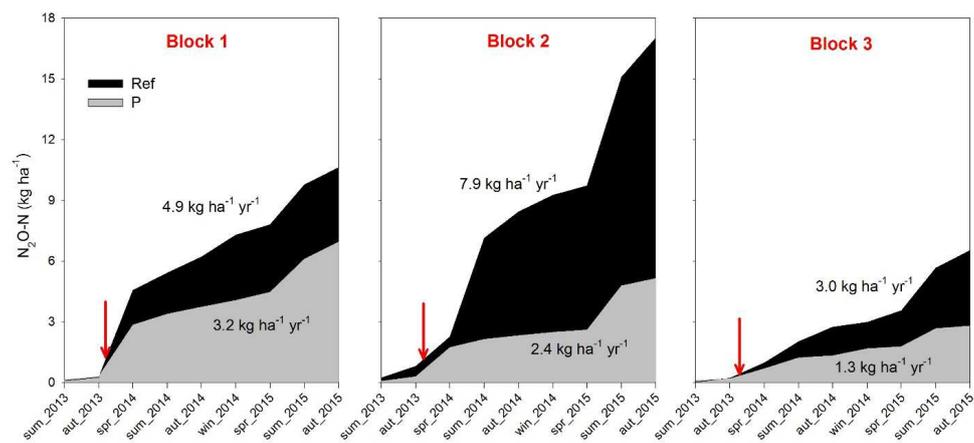
600

601 **Fig. 2** Daily mean air temperature and precipitation (a), and monthly mean N<sub>2</sub>O fluxes in  
602 Reference and P treatments in each of the three blocks (b-d); the red line gives the date of P  
603 addition.



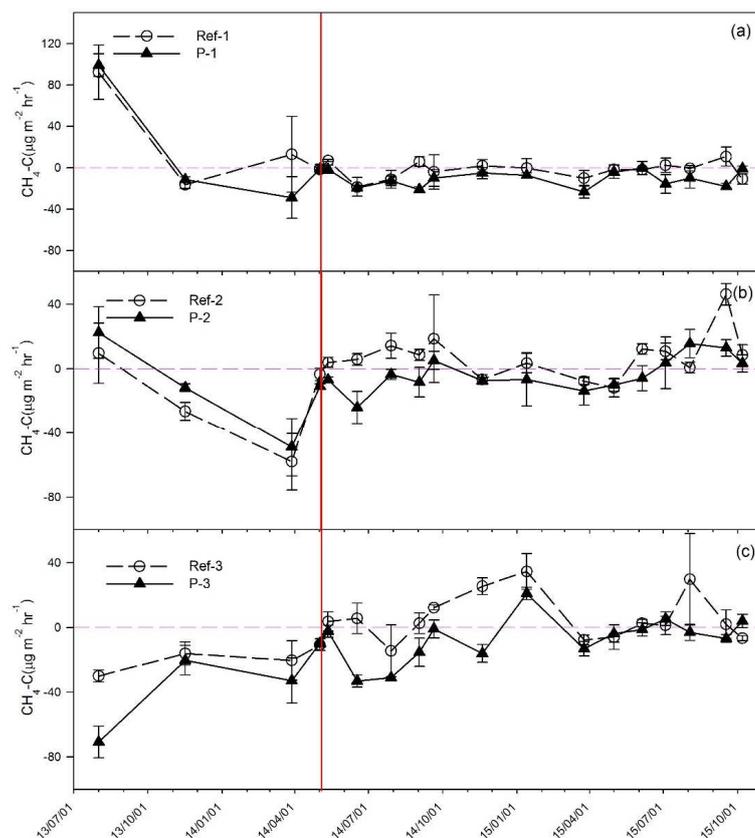
604

605 **Fig. 3** Box whisker plots for N<sub>2</sub>O fluxes in three blocks with Reference and P treatments  
 606 throughout 1.5 years after the P addition; red dash lines indicate mean values; different letters  
 607 indicate significant differences between treatments and blocks.



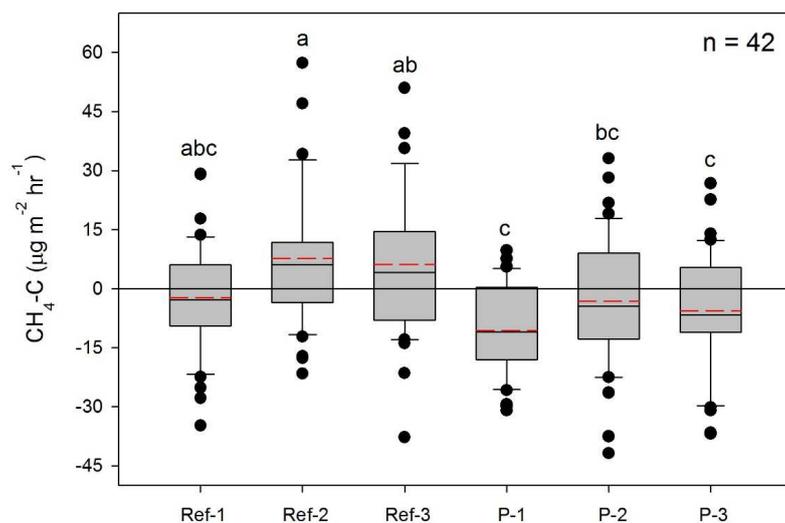
608

609 **Fig. 4** Cumulative N<sub>2</sub>O emissions for three blocks with Reference and P treatments during two  
610 years; the red arrows refer to the date when P addition was conducted.



611

612 **Fig. 5** Monthly mean CH<sub>4</sub> fluxes for three blocks (a-c) with Reference and P treatments during  
 613 two years; the horizontal broken line indicates zero flux the red line refers to the date of P  
 614 addition.



615

616 **Fig. 6** Box whisker plots of CH<sub>4</sub> fluxes for three blocks with Reference and P treatments 1.5  
617 years after the P addition; red dash lines indicate mean values; the small letters indicate the  
618 significance levels among the treatments and blocks.