The importance of freshwater systems to the net exchange of atmospheric 1

2 carbon dioxide and methane with rapidly changing high Arctic landscapes

- 3 Craig A. Emmerton, Vincent L. St. Louis, Igor Lehnherr, Jennifer A. Graydon, Jane L. Kirk, Kimberly J. Rondeau
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Supporting Information 5

6 Numerical analysis

7 We used hierarchical clustering analysis (IBM SPSS Statistics 23) to organize ponds and 8 lakes into similar categories based on concurrent greenhouse gases and chemistry analyses (10 9 sites; n=62). We used between-group linkage and squared Euclidean distances to group similar 10 sites together and delineate distinct high Arctic freshwater types. We then used linear-mixed models (SPSS) to quantify differences in greenhouse gases concentrations and fluxes between 11 12 different high Arctic freshwater types. Linear-mixed models are ideal for analysing non-13 independent, repeated measures data as they integrate inherent errors in repeated sampling 14 designs to more clearly distinguish statistical differences between groups. Linear mixed model 15 details included: use of an auto-regressive moving average (1,1) repeated covariance model; use 16 of a Maximum Likelihood estimation method; and variables organized by freshwater type (fixed) 17 and year (random).

18 Bottle – Automated system dissolved CO₂ concentration comparison

19 Though bottle (time-series) and automated system (diurnal) dissolved CO₂ concentrations 20 were not directly compared in this study, the concentrations measured in Pond 01 and Skeleton 21 Lake using each approach were near identical in most cases (Figure S1). In 2008 (Skeleton Lake) 22 and 2009 (Pond 01), however, calibration of the automated systems appeared to have shifted 23 during transit, causing a slight step difference in measured CO_2 concentrations between the two 24 approaches.

25 **Ebullition fluxes**

26 Ebullition can also liberate CO_2 and CH_4 from freshwater systems. We used submerged 27 inverted 30-cm plastic funnels with a bubble collection chamber to quantify ebullition fluxes of 28 CO_2 and CH_4 from the surface of Skeleton Lake and Pond 01. Traps were deployed continuously 29 at both sites during the 2007 and 2008 summers and checked weekly for bubble volume 30 accumulation. Ebullition volume was measured by drawing into a syringe, through a rubber 31 septum in the collection chamber, the accumulated gas. However, we did not measure gas 32 concentrations of this trapped gas because CO₂ and CH₄ can diffuse back into surface waters 33 while sitting in the trap. Instead, fresh bubbles were collected for CO₂ and CH₄ analyses by 34 probing the sediments and collecting them into a hand held bubble trap. Samples were then 35 immediately transferred to evacuated, stoppered 30 ml Wheaton bottles and analyzed for CO_2 36 and CH₄ concentrations on the GC in a manner similar to that described in the main manuscript 37 for water samples. Bubble CO_2 and CH_4 concentrations were multiplied by bubble volume

collected over the weeklong period to determine ebullition fluxes. 38

39 **Tables**

40 **Table S1:** Mean weather conditions measured by a polar semidesert eddy

41 covariance/meteorological station about 1 km from base camp during the growing season

42 between 2008 and 2012 in the Lake Hazen watershed.

	Air		Wind		Air	Soil	Soil
Month	temp. $(^{\circ}C)$	Rainfall (mm)	speed	$\frac{PAR}{(umol m^{-2} s^{-1})}$	pressure (kPa)	moisture $(m^3 m^{-3})$	temp. $(^{\circ}C)$
	(\mathbf{U})	(mm)	(крп)	(µmorm s)	(KPa)	((\mathbf{U})
June	5.0	3.8	2.9	699	99.4	0.16	6.9
July	8.4	19.3	2.9	596	98.8	0.15	12.0
August	5.7	10.5	2.7	384	98.7	0.14	7.4

- 43 **Table S2** Sampling years and dates for greenhouse gases concentrations collected using bottles
- (B) or automated systems (AS), and general chemical analyses (C) of several freshwater systems
 throughout the Lake Hazen watershed.

Water body		2005	2007	2008	2009	2010	2011	2012
	В	6-21/7	24/6-21/7	6/7-4/8	29/6-22/7	16/6-20/7	6/7-30/7	-
Pond 01	AS	-	24/6-21/7	10/7-2/8	29/6-21/7	19/6-5/7	-	-
	С	6-23/7	28/6-18/7	9/7-2/8	2-22/7	28/6-20/7		-
Dond 02	В	6-21/7	8/7	9/7-2/8	-	10-20/7	6/7-30/7	-
Polid 02	С	6-22/7	6/7	9/7-2/8	-	10-20/7	-	-
Dond 02	В	10/7	13-14/7	29/7	-	12-17/7	6/7-30/7	-
Polid 05	С	-	14/7	-	-	12-17/7	-	-
Pond 07	В	15/7	9/7	29/7	-	13-18/7	6/7-30/7	-
Folia 07	С	-	10/7	-	-	13-18/7	-	-
Dond 10	В	15/7	-	-	-	13-18/7	6/7-30/7	-
Folia 10	С	-	-	-	-	13-18/7	-	-
Dond 11	В	15/7	-	-	-	12-17/7	4/7-30/7	4/7-31/7
FOIId 11	С	-	-	-	-	12-17/7	-	31/7
Dond 12	В	15/7	14-16/7	29/7	-	12-17/7	-	-
Folia 12	С	-	14/7	-	-	12-17/7	-	-
Dond 16	В	-	-	-	-	13-18/7	6/7-30/7	
Folia 10	С	-	-	-	-	13-18/7	-	
	В	-	25/6-19/7	6/7-3/8	29/6-22/7	18/6-19/7	4/7-30/7	4/7-31/7
Skeleton Lake	AS	-	-	8/7-3/8	1-21/7	25/6-20/7	-	-
	С	-	14/7	10/7-2/8	2-22/7	28/6-17/7	-	
Lake Hazen	В	4-20/7	24/6-21/7	6/7-4/8	29/6-22/7	22/6-20/7	6/7-30/7	-
shoreline	С	6-13/7	-	10/7-3/8	2-22/7	28/6-20/7	-	-

46 **Table S3** Empirical relationships for k (cm hr^{-1} ; Hamilton *et al.*, 1994) used in the mass flux 47 equation for greenhouse gases samples (Equation 1 in the manuscript).

if U<3 m s ⁻¹ :	$\begin{split} k_{600} &= 0.76U \\ k_{CO2} &= k_{600} * (600^{0.67} / \text{SC}_{\text{CO2}}^{0.67}) \\ k_{CH4} &= k_{600} * (600^{0.67} / \text{SC}_{\text{CO2}}^{0.67}) \end{split}$
if U≥3 m s ⁻¹ :	$k_{600} = 5.6U - 14.14$ $k_{CO2} = k_{600} * (600^{0.50} / \text{SC}_{CO2}^{0.50})$ $k_{CO2} = k_{CO2} * (600^{0.50} / \text{SC}_{CO2}^{0.50})$
$\mathbf{N}_{\mathbf{r}}$	$K_{CH4} - K_{600}$ (000 /SC _{CO2})

- 48 Notes: U is in-situ wind speed (m s⁻¹) measured at 1 m on automated systems or at a nearby meteorological station;
- 48 Notes: U is in-situ wind speed $(m s^{-1})$ me 49 k_{600} $(cm hr^{-1})$ is the exchange coefficient

- **Table S4** Pearson correlation coefficients of greenhouse gases and general chemistry of
- 51 freshwater types in the Lake Hazen watershed during the growing seasons of 2005, 2007-2012.

52 Statistical significance at α =0.05 indicated in **bold** (IBM SPSS Statistics 23).

	CO_2				CH_4			
	Evap.	Melt.	Shore.	LH Shore	Evap.	Melt.	Shore.	LH Shore
Air _P	.542	.050	.350	565*	803 *	.097	.144	429
Water _T	.131	409	.397	308	.258	483 *	.266	.187
CO_2	1	1	1	1	526	.630**	.530*	.418
CH_4	526	.630**	.530 *	.418	1	1	1	1
Ws	048	066	063	.624*	329	289	.287	276
DIC	.720	.005	.818**	.694**	724	567 *	.387	.291
$\mathrm{NH_4}^+$.453	.139	.783**	255	536	183	.409	142
$NO_3^{-}+NO_2^{-}$	314	003	.345	.689**	.884**	120	.661**	.158
TDN	.285	576*	.692**	.351	341	 527 [*]	.579**	.335
DON	.278	467	.600**	.297	333	235	.554**	.236
PN	071	093	.387	.475	001	346	192	.403
TP	.597	132	.033	.346	211	.175	117	118
TDP	.794 *	.014	.539**	.616*	285	.251	.397	025
PC	.669	141	.359	.526*	743	401	146	.613 *
DOC	.913**	670 **	.462 *	510	622	 552 [*]	.396	034
Cl	.227	.044	.458 *	.641*	680	303	.329	.149
SO_4^{2-}	.154	582 *	291	.668**	.004	706**	027	.282
Na^+	315	481 *	.654**	.715**	376	 573 [*]	.474 *	.331
\mathbf{K}^+	.156	.147	.793**	.712**	644	215	.467 *	.416
Ca^{2+}	.561	340	.587**	.703**	014	824**	.282	.200
Mg^{2+}	.349	364	.327	.718 ^{**}	478	751 **	.448 *	.380
Fe	.974**	 575 [*]	.74 1 ^{**}		483	029	.278	
Alkalinity	.746	015	.818**	.697**	 755 [*]	592**	$.487^{*}$.266
HCO ₃ ⁻	.783 *	015	.818**	.698**	744	591**	.487 *	.266
CO_{3}^{2}	472				287			
TDS	.436	150	.439 *	230	251	383	.360	.011
Chl-a	636	317	.277	068	.131	251	076	039
pН	854 [*]	392	169	261	.088	512 *	073	175

Evap.: Evaporative ponds; Melt.: Meltwater systems; Shore.: Shoreline ponds; LH Shore: Lake Hazen Shoreline; Air_p: baraometric pressure; Water_T: water temperature; dCO₂: dissolved carbon dioxide concentration; dCH₄: dissolved methane

concentration; W_S: wind speed; DIC: dissolved inorganic carbon; NH₄⁺: ammonium; NO₃⁻+NO₂⁻: nitrate + nitrite; TDN: total

dissolved nitrogen; DON: dissolved organic nitrogen; PN: particulate nitrogen, TP: total phosphorus; TDP: total dissolved

phosphorus; PC: particulate carbon; DOC: dissolved organic carbon; Cl⁻: chloride; $SO_4^{2^-}$: sulphate; Na⁺: sodium; K⁺: potassium; Ca²⁺: calcium; Mg²⁺: magnesium; Fe: total iron; HCO₃²⁻: bicarbonate; CO₃²⁻: carbonate; TDS: total dissolved solids; Chl-a:

 G_3^{-1} : calcium; Mg_3^{-1} : magnesium; Fe: total iron; HCO_3^{-1} : bicarbonate; CO_3^{-1} : carbonate; TDS: total dissolved solids; Chl-a: chlorophyll-a; pH: log[H⁺] 61 Table S5 Freshwater ebullition fluxes of carbon dioxide (CO_2) and methane (CH_4) during the

62 63 growing seasons of 2007 and 2008 in Skeleton Lake and Pond 01 in the Lake Hazen watershed.

Pond/Lake	Bubble volume (mL)	CO ₂ Bubble concentration (mgCO ₂ L ⁻¹)	CO ₂ Bubble flux (mgCO ₂ m ⁻² d ⁻¹)	CH ₄ Bubble concentration (mgCH ₄ L ⁻¹)	CH4 Bubble flux (mgCH4 m ⁻² d ⁻¹)
Pond 01					
30-Jun2007	1.76	0.60	0.06 ± 0.01	0.07	0.01 ± 0.00
10-Jul2007	2.01	0.33	0.00 ± 0.00	0.09	0.00 ± 0.00
17-Jul2007	1.07	18.53	0.13±0.01	60.7	0.43 ± 0.04
23-Jul2008	0.20	0.42	0.00 ± 0.00	0.18	0.00 ± 0.00
31-Jul2008	1.17	0.53	0.04 ± 0.03	0.52	0.03 ± 0.03
Skeleton L.					
04-Jul2007	5.51	0.46	0.01	-	-
10-Jul2007	6.53	0.28	0.01 ± 0.00	0.00	0.00 ± 0.00
17-Jul2007	5.86	0.29	0.01 ± 0.00	0.05	0.00 ± 0.00
23-Jul2008	2.50	0.64	0.09 ± 0.00	0.04	0.01 ± 0.00
31-Jul2008	2.00	0.61	0.07 ± 0.06	0.04	0.00 ± 0.00

64 Figures

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Figure S1 Comparison of dissolved CO₂ concentrations between manually-collected bottle
 samples and automated systems in Pond 01 and Skeleton Lake during the growing seasons of
 2007-2010.





71 freshwater systems in the Lake Hazen watershed in July 2010. Grouping of freshwater system

types followed those delineated in the hierarchical cluster analyses (see Results and Discussion).



74 Figure S3 Lake Hazen water level data during the summer growing seasons of 2005, 2007, and 2009-10 at Ruggles River (Water Survey of Canada, 2015). The range of water level when Pond 75

01 received Lake Hazen seepage water through its gravel berm is indicated and based on rapid 76

77 changes in greenhouse gases concentrations. Rapid dilution of methane (CH₄) concentrations and

78 field observations were used to determine the water level of pond breach and flushing.



Figure S4 Carbon dioxide (CO₂) and methane (CH₄) fluxes during the 2005, 2007-2012 growing seasons (June-August) from four different freshwater types in the Lake Hazen watershed. Fluxes calculated using empirical equations and site conditions including water temperature, wind speed, barometric pressure and gas concentrations in water (see Methods). Daily wind speed from nearest measurements indicated by grey points.



86 **Figure S5** Water temperature stratification in Skeleton Lake (Meltwater pond) during the

87 summer growing season of 2008.

88 **References**

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