

***Authors' response on “Inverse-model estimates of the ocean's coupled phosphorus, silicon, and iron cycles” by Benoît Pasquier and Mark Holzer***

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July 24, 2017

Submitted manuscript for discussion available at:  
<https://doi.org/10.5194/bg-2017-122>

Dear Dr. Slomp,

Below please find the comments from the two Reviewers reproduced in bold followed by our point-by-point replies along with the corresponding changes to the manuscript in blue (page numbers refer to the revised version).

The manuscript version with tracked changes appended here shows the deleted parts struck through and in red and the revised parts in blue.

In addition to the revisions in response to the Reviewers, we have made minor revisions to help the reader interpret Figures 8 and 10: We have added fits to simple theoretical curves with some corresponding discussion on the efficiency of each iron type in supporting export (pages 28–34). We believe this strengthens one of the main points of our manuscript, namely that per source-injected molecule aeolian iron supports more phosphorus and opal export than sedimentary and hydrothermal iron. We have correspondingly edited the abstract and the conclusions.

## Reply to comments by Reviewer #1

We appreciate the referee's comments. The referee's main points concern our statements on carbon export, specifically the lack of DOC and constant C:P stoichiometry. Before we address these in detail, we would like to emphasize that we model the phosphorus, silicon, and iron cycles and not the carbon cycle. Carbon export is plotted merely to show the phosphorus export in convenient units that make it easy to compare to other model and observational estimates. We therefore used a constant C:P ratio as a convenient unit conversion not intended to quantitatively estimate carbon export. The advantage of a constant Redfield ratio is that it does not distort the pattern of the phosphorus export. While we did point out that the carbon export flux is merely meant as a conversion of the phosphorus export after Eq. (27), we should have been more explicit that we are discussing phosphorus export in the abstract and in some of the figure discussions. [In response, we have revised the manuscript \(especially the abstract\) to make clear that we quantify phosphorus export and to mention throughout that the plots of carbon export show phosphorus export expressed in carbon units.](#)

The referee's comments raise the interesting questions as to how much dissolved organic matter transport is implicitly captured by our approach, and how different the inferred global carbon export would be if one used more careful estimates of the C:P ratio for the export of particulate and dissolved organic matter. We will address these issues in our point-by-point responses below.

**This manuscript presents the formulation of a global biogeochemistry-ocean circulation model that considers the phosphorus, silica, and iron cycles. Results are presented from a family of solutions that fit the data (dissolved phosphate, silicate, iron, and phytoplankton distributions) equally well but explore the sensitivity to the unconstrained external iron sources to the ocean. Metrics related to global carbon and opal export, limiting nutrients, and iron based export production patterns are presented. The presented modeling framework is at the state-of-the-art for building a 3D global biogeochemical model with the solution computed in offline mode and is of high interest to the ocean modeling and marine biogeochemistry communities. The main advance of the work is to show that the global biogenic carbon and opal exports are well constrained by the available nutrient and satellite phytoplankton**

data even though the external and internal ocean iron fluxes are not. The family of most probable model solutions given the sensitivity in assumptions on iron cycling mostly converge on 10 Pg C yr<sup>-1</sup> and 170 Tmol Si yr<sup>-1</sup> global export.

My two main comments concern the sensitivity of their calculated global carbon export flux to their omission of DOM cycling and variable C:P stoichiometry in organic matter production/export. DOC has been estimated to contribute 20% (Hansell et al., 2009, Oceanography) to 25% (Letscher et al., 2015, Biogeosciences) of global carbon export production. In the model presented by the authors they chose to omit DOP cycling, with their argument being that DOP cycling represents a small to negligible contribution to the biological phosphorus cycle. They also rationalize that DOP typically has lifetimes <1 year in surface waters such that it is not significantly advected with the ocean circulation and can instead focus on vertical redistribution of particles as the dominant export process in their model. However DOC has longer lifetimes in surface waters on the order of a couple years and does accumulate to large enough quantities to be an important part of the carbon export term. Can the authors address the sensitivity of their calculated global carbon export flux to this omission of DOM cycling in their model? Should DOC export be considered as an addition to the computed  $\sim 10$  Pg C yr<sup>-1</sup> flux? Or is the DOC export flux somehow already included in their computations from their model solution?

Our estimates of phosphorus export effectively capture the export of both POP and DOP, despite DOP not being explicitly represented as a separate tracer. This is because the phosphate export of our model is set by the strength of PO<sub>4</sub> uptake and by the Martin exponent  $b$  of the remineralization profile, both of which are optimized by minimizing the mismatch between modelled and observed PO<sub>4</sub> concentrations. Because the real ocean's (regenerated) PO<sub>4</sub> concentrations result from both POP and DOP export, the effect of DOP export on the remineralization field is implicitly accounted for in our optimized states. [In response, we have added a short statement on the fact that the effects of DOP are captured by our model on page 5.](#)

We agree that the lack of explicit representation of DOC results in an underestimation of the carbon export as inferred from the phosphorus export. We estimate the “missing” carbon export by explicitly calculating both POP and

(semilabile) DOP export using the data-assimilated phosphorus cycle of Primeau et al. (JGR, 2013), which carries explicit  $\text{PO}_4$  and DOP tracers and uses the same circulation as in our manuscript. (The DOP export was computed as the aphotic remineralization rate of DOP produced in the euphotic layer.) We find a DOP export of  $0.70 \text{ Tmol P yr}^{-1}$  and a total phosphorus export of  $5.79 \text{ Tmol P yr}^{-1}$ . If one applies C:P=106:1 to the total phosphorus export, this converts to a total carbon export of  $7.37 \text{ Pg C yr}^{-1}$ . However, if one uses C:P=225:1 for DOP (as determined by the DOM OPT simulation of Letscher et al. (2015) for semilabile DOM), the total carbon export becomes  $8.37 \text{ Pg C yr}^{-1}$ . Thus, by not representing DOC explicitly, one “misses”  $\sim 12\%$  of the total carbon export. (In this calculation, the DOC export represents 23% of the total carbon export, consistent with the estimate of Letscher et al., (2015).) *In response, we have added a paragraph briefly discussing the fact that phosphorus export converted to carbon units underestimates carbon export by roughly 12% due to the missing DOC export after Equation (27), page 23–24.*

While we agree that accounting for DOC export gives more realistic estimates for the total carbon export, we reiterate that we use our model to calculate phosphorus export and merely convert to carbon units when we think this is useful (e.g., for comparison to the results of Primeau et al., 2013).

**Secondly, recent global datasets and model inversions of nutrient data have shown/predicted that the production and export of organic matter from the surface ocean is not constant and exhibits latitudinal and ocean biome-level variability (e.g. Martiny et al., 2013, Nature Geoscience; Teng et al., 2014, Nature Geoscience; Devries & Deutsch, 2014, Nature Geoscience; Galbraith & Martiny, 2015, PNAS). The authors chose to calculate all of their carbon export metrics using a constant Redfield ratio of 106:1 C:P to get carbon units from their model which is in phosphorus units. How much would their global estimates of carbon export change if a variable C:P of organic matter production/export were used? For example, a regionally variable C:P could be computed using the relationship predicted with surface phosphate conditions from Galbraith & Martiny, 2015 (PNAS) using the model simulated phosphate fields. Alternatively, the twelve-biome inferred export C:P ratios from Teng et al., 2014 (Nature Geoscience) could be used to calculate the regionally variable C:P of export from the authors model. It seems given what we now know vis-à-vis regionally variable organic matter stoichiometry, it would be remiss not to include that knowledge to update the global C export flux from the authors’ model solution.**



We agree with the referee that if the objective were to provide an accurate estimate of carbon export, using recently constrained regionally varying C:P ratios would be more realistic. However, using the regionally varying C:P ratios suggested by the referee has only modest effects on the globally integrated carbon export:

- (i) Applying the P:C relation of Galbraith and Martiny (2015) to the phosphate export of our typical state gives a carbon export of  $8.5 \pm 0.4 \text{ Pg C yr}^{-1}$  or  $9.4 \pm 0.9 \text{ Pg C yr}^{-1}$  when we use their log-binned parameter values.
- (ii) Applying the regional C:P inverse-model estimates of Teng et al. (2014) gives a carbon export of  $10. \pm 2. \text{ Pg C yr}^{-1}$ . Both this estimate and the one based on the log-binned regression agree within their uncertainties with our simple unit-conversion value of  $10.3 \pm 0.4 \text{ Pg C yr}^{-1}$ .

In response, we have added some discussion of these calculations to the caveats section where we discuss regionally variable C:P ratios (page 36). We have also revised the manuscript to make explicit throughout that we compute phosphorus export converted to carbon units.

#### Other comments:

**Pg 20 L1-5: The authors blame phytoplankton biomass mismatches between the model and satellite observations based on a lack of seasonality in the model but aren't these steady-state satellite climatologies they are comparing against, and therefore seasonality is averaged over?**

We are alluding to the fact that there is a difference between the fields produced by a steady-state model and the long-term average of a seasonally varying field. (For example, the circulation we use is *not* a simple annual-mean circulation, but rather a steady-state circulation whose transport reproduces the annually averaged tracers used in the assimilation.) However, we agree that this passage could have been clearer. In response, concerning the mismatch at  $\sim 60^\circ\text{S}$ , we have revised the manuscript to point out that the strong seasonality of the Southern Ocean, with its large seasonal cycle in sea-ice coverage, is not captured by our model and that this could conceivably be the reason for the mismatches in that region (page 43). We have deleted the second mention of seasonality in this paragraph, which was unintentional repetition.

**Pg 23 L20-25: The authors argue that the sharper meridional gradient in C export is more realistic because there is a sharp gradient in satellite NPP. But they don't include DOC export. DOC export is estimated to be 1/5 to 1/4 of total global C export with its larger contribution occurring in the subtropical gyre systems due to large-scale downwelling.**

We agree that the inclusion of DOC would result in more realistic estimates of carbon export. However, in this passage we are comparing our phosphorus export (converted to C units) to POC export as estimated from satellite-derived NPP, and by definition POC export does not include DOC export. [In response, we have revised this passage to make clear that we are comparing phosphorus export converted to carbon units, and to state that sharp meridional gradients can also be seen in the satellite-derived NPP \(pages 23–24\). We have deleted the statement about the sharp gradients being “more realistic” to avoid confusion.](#)

**Pg 27 L24-30. One statement says the dominant Fe sink is from POP scavenging. The very next statement says that opal scavenging accounts for half of Fe sinks. The next statement says that dust scavenging is negligible for Fe sinks. Why not just say that POP and opal scavenging account about equally to Fe sinks?**

Thank you for catching this. (There was some left-over wording from a previous iteration.) [We have revised as suggested \(page 26\).](#)

## Reply to comments by Reviewer #2

Reviewer 2 raises some interesting points (reproduced here in bold) to which we respond in detail below. The Reviewer also points out that our Introduction section did not explicitly articulate the main science questions addressed by our study, which can be easily remedied in the revised manuscript. Reviewer 2 then asks if our model can be used to constrain class-dependent Fe quotas and the scavenging efficiency of different particle types. However, these quantities are not robustly constrainable within our framework. Moreover, these questions are tangential to our study whose main goal is to obtain data-constrained estimates of the coupled iron–macronutrient cycles and to elucidate the relative importance of the different iron sources for supporting export production. The Reviewer also makes some “minor” points regarding detrital fractions and a minimum iron quota. Thinking about these helped us deepen our understanding and improve our manuscript.

**The submission by Pasquier and Holzer uses a new ocean biogeochemical and ecosystem model, embedded within a data-constrained steady-state circulation model, to explore linkages between the phosphorous, silicon, and iron cycles. The model uses interesting methods to represent nutrient uptake by multiple-species phytoplankton communities without explicitly resolving their biomass, allowing for efficient simulations and parameter optimization. Based on previous work, the authors understand that no single optimal solution for the Fe cycle can be obtained because certain source and sink processes have overlapping effects on the Fe distribution. They therefore explore a “family” of solutions with different source strengths, which are independently optimized and then compiled into a “typical” solution and uncertainty range. There are a number of interesting outcomes that are robust across the family of solutions, for example the patterns of phosphorous export supported by each iron source and the “efficiency” of each source at supporting export. Atmospheric Fe supports most export relative to the magnitude of its source, followed by benthic and then hydrothermal Fe.**

We would like to point out that our work additionally shows that part of the uncertainty in the iron sources comes from the fact that, for very similar overall mismatch with observations, the three sources of iron can compensate for

each other. [We now explicitly make this point in the revised Introduction section.](#)

**I think this paper takes an interesting approach and has the potential to be a valuable contribution to the literature. Nevertheless, I have two main critiques of the paper in its current form. First, the paper lacks a clear direction from the outset. The introduction does not lay out any specific questions or hypotheses that the new model is designed to address, nor does it identify the particular gaps in our understanding of the Fe cycle that the authors aim to close. Instead, the goal is simply states as “to constrain a model of the coupled nutrient cycles by optimizing the biogeochemical parameters against available observations”, which does not seem like a strong motivation. The purpose of an inverse model should be to extract new information from the available observations, not just to match the observations. The authors should begin by clearly stating what new information they aim to extract by explicitly simulating the coupling of Si, P and Fe, relative to their previous work.**

We agree that the Introduction section could have more clearly articulated the key points of our paper. The purpose of the inverse model is definitely not to build a model and compare to observations. [In response, we have revised the Introduction section, which now clearly states that our paper makes the following advances:](#)

A. We are building an inverse model of the coupled Fe, P, and Si cycles so that the macronutrients and organic matter export can respond to changes in the iron supply. This is a key advance over the work of *Frants et al.* [2016] where the phosphate cycle was prescribed. Our new inverse model provides, for the first time, a family of data-constrained state estimates of the coupled Fe-P-Si cycles for a wide range of not only aeolian, but also hydrothermal and sedimentary sources. Analysis of this family of estimates allows us to show that the uncertainty in the iron sources stems not only from compensation between sources and scavenging sinks, but also from the fact that the different types of iron sources (aeolian, hydrothermal, and sedimentary) can compensate each other.

B. We use our state estimates to address an important open question about the marine iron cycle: What are the relative contributions of the different iron sources to supporting the world ocean’s export production? While there have

been perturbation experiments with forward models where one type of source (e.g., hydrothermal or sedimentary) was shut down to assess the importance of dFe from the missing source [e.g., *Tagliabue et al.*, 2009, 2010, 2014], such experiments cannot quantify the true contribution of hydrothermal or sedimentary iron to biological production because of the nonlinearities of the iron cycle [*Holzer et al.*, 2016]. Moreover such experiments were conducted with definite but highly uncertain choices of the iron sources, and the models were not objectively constrained by the available observational data. Thus, in addition to presenting the first inverse model of the coupled Fe-P-Si cycle, the central scientific objectives of our study are to answer the following key questions:

1. How well can the observed dFe, PO<sub>4</sub>, and Si(OH)<sub>4</sub> concentrations be fitted to observations for widely differing iron sources, and are there limits on the iron source strengths that are consistent with the observed dFe concentrations?
2. What are the limitation patterns that emerge from the data-constrained estimates of the coupled nutrient cycles, given that direct observational data on these patterns is very sparse?
3. How well constrained are the phosphorus and opal exports for optimized state estimates with widely different iron sources?
4. What fractions of phosphorus and opal export are supported by aeolian, hydrothermal, and sedimentary iron, and how do these fractions vary with the relative iron source strengths?

We think that the underlying theme and science objectives of our work should now be clear to the reader.

In addition, we have added a “road map” paragraph at the end of the Introduction section, so that the reader knows what to expect for the rest of the paper even without looking at the section headings.

**This same mentality extends throughout the paper, where numerous model-data comparisons are presented without properly highlighting what new has been learned in the process. For example, one of the key benefits of this coupled model is the ability to assess the relative Fe-scavenging efficiency of different particle types (organic, silica, dust), which remains an open question in Fe biogeochemistry. While this result is part of the**

**model solution, it receives very little attention in the text — it is briefly noted that on a global basis, organic matter and silica are equally responsible for Fe removal from the ocean, and a figure is shown in the Appendix. But the authors should discuss which particle type is the stronger Fe scavenger on a per-gram basis, whether this is robust across the family of solutions. This would be a new interesting result of this study.**

We would like to point out that we showed only three figures (joint pdfs, mean iron profiles, and phytoplankton concentrations) that compare model against observations, and we note that the Reviewer requests a yet more detailed comparison with GEOTRACES sections below. These comparisons are not made to answer new science questions per se, only to quantify the degree to which the dFe and macronutrient concentrations can be matched to the observations.

Regarding the suggestion to delve further into the scavenging efficiency of different types of particles (POP, opal, and dust), the partitioning of the scavenging among the different particle types is not something that can be constrained robustly from our inverse model. The scavenging by one particle type can be compensated by another particle type because of overlap in the spatial pattern of their fluxes. The nutrient and phytoplankton data used do not provide separate constraints on the scavenging by each particle type, only on the total amount of scavenging. For these reasons this aspect of our model is not a focus in our manuscript.

Nevertheless, one may of course ask how the partition among particles types varies across our family of estimates. We find that scavenging by dust is negligible for all our state estimates, while the fraction scavenged by POP ranges from ~10% for the lowest iron source cases and saturates near ~100% for the highest iron sources considered. (The remaining fraction is due to opal scavenging.)

[In response to this comment, we have added a brief discussion that states that the partition among particle types is likely not robust, and that details the systematic dependence on the total iron source strength \(page 27\).](#)

**As another example, the new model seems to be the ideal tool for examining differences in Fe quotas among phytoplankton types — another open question in Fe cycle research. The authors briefly mention that they ex-**

perimented with different Fe quotas, but abandoned the approach when the parameters converged to similar values. If the model selects similar Fe quotas for all plankton groups, and this is robust across the whole family of solutions, it would be an interesting result indeed and worth of some attention in the paper! Especially if the authors could demonstrate that there is no evidence for enhanced Fe quotas in subtropical gyres where diazotrophic plankton are common, given that there is ongoing debate about the relative Fe requirements of N-fixing and non-fixing plankton.

Reviewer 2 is correct that we do not distinguish the Fe:P parameters of different functional classes based on experiments where we optimized class-dependent Fe:P parameters. While we agree that establishing any differences in Fe quotas among N-fixing and non-fixing plankton is an interesting open question, we would like to remind the Reviewer that we do not model the nitrogen cycle at all so that this issue is beyond the scope of our paper.

We cannot use our inverse model to robustly constrain different values of  $R_0^{\text{Fe:P}}$  and  $k_{\text{Fe:P}}$  for each class because these six parameters (two per class) would directly compensate for one another in the global Fe export. We therefore decided to use only the two class-independent parameters,  $R_0^{\text{Fe:P}}$  and  $k_{\text{Fe:P}}$ , and optimized them sequentially. [Constraining class-dependent Fe quota is beyond the scope of what is possible with our inverse model, which we now explicitly point out in the revised manuscript \(page 9\).](#)

My second main critique of the paper is that it doesn't present the model-data comparisons for Fe that would be best suited to support the conclusions. If one of the main goals of the paper is to understand the relative contribution of each Fe source to organic matter export, one would want to show that the model accurately reproduces the locations and transport trajectories of the sources. By design, many of the GEOTRACES transects sampled different source regions of Fe, and show clear signatures of these sources and their transport across basins. For example, GA03 and GP16 both traverse benthic and hydrothermal source regions. Plotting cross-sections of modeled and observed Fe along these transects would give a clearer visual impression of the model's performance than the summary statistics and basin-wide profiles that are presented. The reader would want to ensure that these source signa-

**tures and transport trajectories are well reproduced, before considering the export contribution analysis.**

The Reviewer is correct that a key point of our analysis is to quantify the relative contribution of each iron type to organic matter export and to explore the systematics of these contributions over a wide range of iron source strengths. While source patterns and transport are important for this, we do not think one can reasonably expect our model to reproduce the GEOTRACES transects with complete fidelity. First, we use a coarse-resolution, steady-state inverse model, while the GEOTRACES sections provide snapshots in space and time. Therefore, as we point out in our manuscript, our model cannot capture any transient plumes (e.g., from an African dust event) that are highly localized and episodic. Our state estimates can only capture the long-term average concentration, coarse-grained to  $2^\circ \times 2^\circ$  resolution. Second, we had only the Intermediate Data Product available when developing this model. Pacific features are therefore only constrained from the older data compilation by *Tagliabue et al.* [2012]. In terms of capturing hydrothermal plumes, we note that our model uses a data-assimilated circulation, but this circulation only assimilated  $T$ ,  $S$ ,  $\text{PO}_4$ , and  $^{14}\text{C}$  but not  $^3\text{He}$ . Therefore, there are likely still some biases in the abyssal circulation, which contribute to the fact that we do not perfectly match the observed hydrothermal iron plumes. However, what matters for our analysis is the large-scale transport into the euphotic zone, particularly the transport into iron-limited regions such as the Southern Ocean. We have no reason to think that this large-scale transport is suspect as evidenced by realistic large-scale patterns of production that are robust across a family of states with widely varying iron source strengths. We emphasize that production in our inverse model is mechanistically driven by dFe and macronutrient availability.

Of course, we are happy to show a direct comparison with the GEOTRACES sections subject to the caveats discussed above. Figure 1 herein compares the main transects included in the Intermediate Data Product with our typical state estimate. The coarse resolution model does capture the large-scale features, but localized high concentrations cannot be captured at our resolution.

We respectfully disagree that we did not present the model–data comparison best suited to supporting our conclusions. We think that Figures 1d, 2c, and 3 of our manuscript are the most relevant and appropriate *quantitative* comparisons between estimated and observed dFe, given that essentially raw bottle data is compared with a coarse-resolution steady-state model. Figure 1d plots the



RMS cost-weighted mismatch for the whole family of estimates. This shows how varying iron source strengths affects our ability to match dFe observations. Figure 2c plots the cost-weighted joint distribution of modelled and observed dFe for our typical estimate. This shows just how difficult it is to match the sparse dFe observations, at least compared to the macronutrients for which gridded climatologies are available. Figure 3 shows the basin dFe profiles of each family member. Because these profiles average a large number of observations, they provide a robust metric for assessing the model's ability to capture the large-scale vertical gradients and a realistic nutricline.

In response, we have added an appendix showing the GEOTRACES comparison of Figure 1 herein, plus a brief discussion on what features one should not expect to be captured by our coarse-resolution steady-state estimates (Appendix G, pages 45–46).

**In addition, I have the following minor comments:**

**1. I agree with Reviewer #1 that caveats of neglecting DOP cycling need to be more carefully considered. Ignoring DOP will not only bias the total estimated export, but also its pattern and therefore potentially the contribution of different Fe sources to export. Particularly, DOP convergence is thought to provide a significant P supply to subtropical gyres, and essentially “relocates” export downstream, from tropical and coastal upwelling zones into the gyres. Given that benthic Fe supports most export in upwelling zones and atmospheric Fe supports most export in gyres, relocating exporting between those two regimes seems important.**

We agree that not carrying DOP has (minor) implications for converting our phosphorous (P) export to carbon export as discussed in response to Reviewer 1. However, our optimized P export effectively includes DOP effects in spite of DOP not being explicitly represented. The optimization of the P-cycle parameters implicitly accounts for all P export by minimizing the mismatch with the observed distribution of  $\text{PO}_4$ , which in the real ocean is determined by the remineralization of DOP (and possibly also of POP directly). We are therefore not concerned about missing or “relocated” export due to lateral transport

and utilization of DOP. This reasoning is supported by the fact that our POP export averaged over the subtropical gyres matches the estimates of *Letscher et al.* [Nature Geoscience, 2016] when we use the same masks (interpolated to our grid) to define the subtropical gyres. *Letscher et al.* [2016] explicitly model DOP transport and utilization and find a mean subtropical-gyre POP export of  $10 \pm 2 \text{ mmol P m}^{-2} \text{ yr}^{-1}$ , which agrees with our corresponding export of  $10 \pm 1 \text{ mmol P m}^{-2} \text{ yr}^{-1}$  (mean and standard deviation across our family of estimates). Our estimates agree with *Letscher et al.*'s despite the fact that we do not have DOP contributing to biological production, which underscores that our estimates of POP export implicitly account for DOP effects.

Regarding the patterns of dFe-supported P export, it is true that aeolian dFe support is more important than sedimentary dFe support in the subtropical gyres, although both aeolian and benthic dFe are important in upwelling regions (Figure 9 of our manuscript). However, as discussed above, we capture effects due to DOP implicitly and there is no reason to think that export has been relocated out of the subtropical gyres. Our estimates of the Fe-type-supported export patterns are by construction consistent with the available nutrient data and we showed that they are robust for a range of Fe sources with widely different ratios of the benthic to aeolian source strengths.

In response to this comment, and in addition to our revisions in response to Referee 1, we have added some discussion to the manuscript about the agreement with the subtropical POP exports of *Letscher et al.* [2016] when we more fully discuss our choice of omitting an explicit representation of DOP (page 24).

**2. What is the justification for choosing such widely different export ratios between plankton types (page 5, line 13)? The authors cite Dunne 2005, but there have been other studies since (e.g., Richardson 2007) that suggest small plankton contribute as much export, relative to their NPP, as large plankton.**

Reviewer 2 is correct that the values of our detrital fractions,  $f_c$ , are the optimized values from the work of *Dunne et al.* [2005], that is,  $f_{\text{sml}} = 0.14$  and  $f_{\text{lr}} = f_{\text{dia}} = 0.74$ . This means that export due to large phytoplankton is  $\sim 5$  times larger than export due to small phytoplankton for the same amount of production. We prescribed the  $f_c$  parameters because they are not constrainable from the nutrient and plankton concentration data alone. The  $f_c$  values set the fractional

export of each phytoplankton class. However, the nutrient observations constrain only the total export of all classes, while the phytoplankton data [Kostadinov *et al.*, 2016] constrain only the concentration of the different functional classes and thus their respective *uptake*, but not their respective *export*. We think that prescribing  $f_c$  to take the values from the study of Dunne *et al.* [2005] was an appropriate choice, and we will argue below that this is not contradicted by the findings of Richardson and Jackson [2007], although we acknowledge that the precise  $f_c$  values are uncertain.

Richardson and Jackson [2007] suggest that export efficiencies of large and small phytoplankton should be closer to each other, specifically that “the relative contributions of various phytoplankton size classes to carbon export are proportional to their contributions to total net primary production”. However, their data is consistent with  $f_c$  having different values for different classes. Figure 2 herein shows the local fractional uptake plotted versus the corresponding fractional export of each phytoplankton class for our typical estimate. The small class collapses to a compact curve because  $f_{\text{dia}} = f_{\text{lrg}}$ . If we chose different “detrital” fractions for the Diatoms and Large class, then the “small” curve would become a cloud of points whose boundaries are determined by the ratios  $f_{\text{lrg}}/f_{\text{sml}}$  and  $f_{\text{dia}}/f_{\text{sml}}$ . Specifically, if one of these ratios were reduced to 4 or 3, the “small” points would spread out somewhat toward the 1:1 line. Richardson and Jackson [2007] show that the export through mezozooplankton dominates the purely detrital export by about an order of magnitude so that our “detrital” fraction pertains to the export through the mezozooplankton pathway. Therefore, Figure 2 should be compared to Fig. 1B of Richardson and Jackson [2007] for mezozooplankton-driven export. Note that our Diatom and Large points in Figure 2 are broadly consistent with the measurements plotted by Richardson and Jackson [2007], while our compact curve for the Small class is very close to two of the Arabian-Sea picoplankton measurements. Given the extreme sparseness of the measurements and the fact that they are presented without error bars, we think the Richardson and Jackson [2007] paper provides no strong evidence that our  $f_c$  values should be drastically different from those of Dunne *et al.* [2005]. We acknowledge however that there is significant uncertainty in the precise values of the  $f_c$  parameters and that the small-to-large ratio may well be smaller than 5, although the Richardson and Jackson [2007] data certainly does not suggest that it should be unity.

In response, we have relegated Figure 7 of the manuscript, which shows the export production of each functional class, to an appendix because the partition of the export among functional classes depends strongly on our choice of the

prescribed, somewhat uncertain  $f_c$  parameters. (Former Figure 7 is now Figure F1 on page 45.) In addition, we have added a brief discussion on the uncertainty in the  $f_c$  values where they are defined (page 6).

**3. What is the justification for not prescribing a minimum Fe:P quota in equation 14? It is impossible to sustain phytoplankton growth with no Fe, so if the model is optimizing towards zero it means that model is straying into unrealistic parameter space, not that this parameter can be neglected. The authors should set a reasonable lower limit during the optimization (e.g. low end of the range shown in Moore et al 2013), rather than allowing the Fe quota to approach zero at low [Fe].**

The justification for not prescribing a minimum Fe:P quota in Equation (14) is that it turns out that none is required to obtain realistic state estimates for the following reasons: For very low dFe concentrations, our Fe:P cellular ratio would fall below a realistic cell quota, but this has no mechanistic consequence because for such low dFe concentrations there is essentially no uptake in our formulation. This is because our Fe:P ratio is proportional to a dFe Monod term, while phosphate uptake is proportional to the square of a dFe Monod term. Thus, as dFe becomes small, the uptake goes to zero faster than the Fe:P quota itself. Simply put, this means that when Fe:P is unrealistically small, it does not matter because there is no P or Fe uptake.

One should therefore regard our formulation to have a minimum Fe:P quota of zero, and we should not have stated on page 8 that we “ignored it for simplicity”. The fact that optimizing a minimum quota (constrained to be non-negative) resulted in an optimized minimum quota of zero means that a simple Monod factor suffices to capture the dFe dependence of the Fe:P ratio where there is significant uptake. In response, we have reworded these passages to make these points explicit (page 9).

We thank Reviewer 2 for making us revisit our formulation of the Fe:P ratio, which made us realize that there was an issue not with the minimum Fe:P ratio, but with the parameter  $R_0^{\text{Fe:P}}$ .  $R_0^{\text{Fe:P}}$  multiplies the dFe Monod term in the Fe:P ratio and is thus the maximum attainable Fe:P ratio at high dFe. For a small fraction of our state estimates, the optimization pushed  $R_0^{\text{Fe:P}}$  to near-zero values. This is very unrealistic as it means significant P uptake and export are maintained without Fe uptake. In response, we have now corrected

this by excluding cases for which the optimized  $R_0^{\text{Fe:P}} < 0.5 \text{ mmol Fe (mol P)}^{-1}$  from our family of state estimates. We have updated all our figures accordingly and note that removing these unphysical outliers makes no visual difference, although it narrows the range of Fe export across our family of estimates to  $0.87\text{--}5.6 \text{ Gmol Fe yr}^{-1}$ . Where we discuss our family of solutions (Sec. 3.4), we have added a brief discussion on the fact that we excluded state estimates with  $R_0^{\text{Fe:P}} < 0.5 \text{ mmol Fe (mol P)}^{-1}$  because they are unrealistic (page 18).

# Inverse-model estimates of the ocean's coupled phosphorus, silicon, and iron cycles

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**Abstract.** The ocean's nutrient cycles are important for the carbon balance of the climate system and for shaping the ocean's distribution of dissolved elements. Dissolved iron (dFe) is a key limiting micronutrient, but iron scavenging is observationally poorly constrained leading to large uncertainties in the external sources of iron and hence in the state of the marine iron cycle.

Here we build a steady-state model of the ocean's coupled phosphorus, silicon, and iron cycles embedded in a data-assimilated steady-state global ocean circulation. The model includes the redissolution of scavenged iron, parameterization of subgrid topography, and small, large, and diatom phytoplankton functional classes. Phytoplankton concentrations are implicitly represented in the parameterization of biological nutrient utilization through an equilibrium logistic model. Our ~~coupled nutrient model thus carries only three~~ formulation thus has only three coupled nutrient tracers whose three-dimensional ~~steady-state distributions can be found efficiently~~ distributions are found using a Newton solver. The very efficient numerics allow us to use the model in inverse mode to objectively constrain many biogeochemical parameters by minimizing the mismatch between modelled and observed nutrient and phytoplankton concentrations. Iron source and sink parameters cannot jointly be optimized because of local compensation between regeneration, recycling, and scavenging. We therefore consider a family of possible state estimates corresponding to a wide range of external iron source strengths. All ~~optimized~~ state estimates have a similar mismatch with the observed nutrient concentrations and very similar large-scale dFe distributions. However, the relative contributions of aeolian, sedimentary, and hydrothermal iron to the total dFe concentration differ widely depending on the sources.

Both the magnitude and pattern of ~~carbon~~ the phosphorus and opal exports are well constrained with global values of  $8.1 \pm 0.3$  Tmol P yr<sup>-1</sup> (or, in carbon units  $10.3 \pm 0.4$  Pg C yr<sup>-1</sup>) and  $171. \pm 3. \text{ Tmol Si yr}^{-1}$ . We diagnose the ~~carbon~~ phosphorus and opal exports supported by aeolian, sedimentary, and hydrothermal iron. The geographic patterns of the export supported by each iron type are well constrained across the family of state estimates. Sedimentary-iron-supported export is important in shelf and large-scale upwelling regions, while hydrothermal iron contributes to export mostly in the Southern Ocean. The ~~globally integrated fraction of the global~~ export supported by a given iron type varies systematically with ~~the fractional contribution of its source~~ its fractional contribution to the total iron source. Aeolian iron is most efficient in supporting export in the sense that its fractional contribution to export exceeds its fractional contribution to the total source ~~by as much as ~30% for carbon~~ and ~20% for opal export. Per source-injected molecule, aeolian iron supports  $3.1 \pm 0.8$  times more phosphorus export and

2.0 ± 0.5 times more opal export than the other iron types. Conversely, per injected molecule, sedimentary and hydrothermal iron ~~are less efficient with a fractional export that is less than their fractional sources. For the same fractional contribution to the total source, hydrothermal iron is less efficient than sedimentary iron for supporting carbon export but about equally efficient for supporting opal export~~ support  $2.3 \pm 0.6$  and  $3.7 \pm 2.3$  times less phosphorus export, and  $1.9 \pm 0.5$  and  $2.2 \pm 1.0$  times less opal export than the other iron types.

## 1 Introduction

The ocean's nutrient cycles control the primary productivity of the global marine ecosystem and the ocean's biological carbon pump, which are crucial components of the global carbon cycle that regulate atmospheric CO<sub>2</sub> concentrations. The nutrient cycling of the ocean is governed by the interplay of the ocean's advective-diffusive circulation, biological utilization, biogenic particle transport, and the external sources and sinks of nutrients. The cycles of macro and micronutrients are coupled through colimitation on biological uptake, and through the scavenging of micronutrients such as iron by sinking organic matter.

We focus on dissolved iron (dFe) as a key micronutrient because of its well-documented fundamental role in primary production (e.g., Boyd and Ellwood, 2010). Indeed, dFe was suggested to limit oceanic phytoplankton growth as early as the 1930's (e.g., Gran et al., 1931; Hart, 1934). Since then, numerous studies have reported that iron deficiency limits productivity over vast ~~regions of the ocean, particularly~~ areas, particularly the high-nutrient low-chlorophyll (HNLC) regions like the Southern Ocean (e.g., de Baar et al., 1995; Lundy et al., 1997; Martin and Fitzwater, 1988; Boyd et al., 2007; Boyd and Ellwood, 2010). Martin (1990) went as far as to suggest that perturbations in the iron cycle played a crucial role in past climate fluctuations. More recently, iron-enrichment field experiments (e.g., Boyd et al., 2007) and model simulations (e.g., Nickelsen and Oschlies, 2015) have demonstrated the importance of iron for the global biological pump.

We model phosphate (PO<sub>4</sub>) because it is essential to the metabolism of all living organisms (e.g., Smith, 1984; Howarth, 1988), which allows all biological production to be keyed to phosphate utilization (e.g., Kwon and Primeau, 2008; Primeau et al., 2013; Holzer and Primeau, 2013). Silicic acid (Si(OH)<sub>4</sub>) was considered because of the importance of diatoms in marine ecosystems, ~~particularly in the Southern Ocean,~~ (e.g., Nelson et al., 1995; Buesseler, 1998; Moore et al., 2004; Brzezinski et al., 2011b) and because the pronounced silicon trapping of the Southern Ocean (e.g., Holzer et al., 2014) might be sensitive to iron availability.

With a changing climate, we expect not only changes in the ocean circulation, but also changes in the winds, hydrological cycle, and land use, and hence in the aeolian iron supply. To understand how such changes impact the ~~global ocean nutrient cycles~~ ocean's nutrient cycles, it is necessary to model the coupling between the nutrients mechanistically. While global biogeochemistry models have been used extensively for this purpose (e.g., Tagliabue et al., 2014; Nickelsen and Oschlies, 2015), ~~here it is our goal to constrain a model of the coupled nutrient cycles by optimizing the biogeochemical parameters against none of these models have been objectively constrained by the~~ available observations. ~~To that end, we build a model of intermediate complexity that focuses on~~ Here, we formulate an inverse model of the ocean's ~~phosphorus, silicon, coupled macronutrient~~ and iron cycles ~~embedded in a data-assimilated global circulation. The biogeochemical parameters of the model are determined~~

by objectively minimizing the mismatch with observed nutrient and phytoplankton concentrations. To ensure the optimization problem remains tractable with a reasonable computational burden, we formulate a model of intermediate complexity.

The intercomparison of iron models by Tagliabue et al. (2016) showed that current models ~~of the iron cycle~~ contain significant uncertainties. Despite the fact that the models have iron source strengths that range over nearly two orders of magnitude, all models can be tuned to roughly the same mean dFe concentration with an inter-model variance of only 27 %. This is due to essentially unconstrained scavenging rates so that models are free to employ different scavenging strengths to balance the sources at roughly comparable dFe concentrations. All the models of the intercomparison are prognostic forward models that are computationally too ~~costly to explore~~ expensive for exploring the biogeochemical parameter space systematically, or ~~to compute for computing~~ the sensitivity with respect to multiple parameters (e.g., Kwon and Primeau, 2006).

~~Recently, Frants et al. (2016) designed a simple inverse~~ Here we aim to close this gap in our ability to constrain the iron cycle objectively by formulating a numerically highly efficient model of the ~~global iron cycle embedded in the steady iron cycle that is mechanistically coupled to key macronutrient cycles and that is embedded in a~~ data-assimilated ~~circulation of~~ Primeau et al. (2013). ~~A matrix representation of the associated transport operator and biogeochemical processes afforded numerically highly efficient solutions. This made it possible not only for biogeochemical parameters to be systematically optimized, but also for global circulation.~~

We build on the simple iron-only inverse model of Frants et al. (2016) for which biogeochemical parameters were optimized and which allowed novel diagnostics to be computed such as ~~a the~~ mean iron age and rigorous source attribution of dFe (Holzer et al., 2016). Consistent with the findings of the iron-model intercomparison, Frants et al. (2016) showed that current dFe observations cannot constrain the iron sources because of local compensation between sources and sinks. Frants et al. (2016) therefore explored a family of state estimates corresponding to a range of aeolian source strengths, all of which are consistent with the currently available dFe observations.

~~Here, we build on the iron model of Frants et al. (2016), but do not prescribe the phosphate cycle. Instead, we explicitly couple~~ However, in the study of Frants et al. (2016) the iron cycle could not interact with a prescribed phosphate cycle, and the silicon cycle was not considered at all. Moreover, for the family of state estimates of Frants et al. (2016) only the aeolian source was varied and it is unclear if the wide range of hydrothermal and sediment iron sources found in the literature is consistent with the dFe observations. The model of Frants et al. (2016) is therefore unsuitable for investigating the effects of iron-source perturbations on biological production or for exploring how much export is supported by the different iron sources. Here, we overcome these shortcomings by explicitly coupling the iron, phosphorus, and silicon cycles through their mutual colimitations so that the macronutrients can respond to changes in dFe. We furthermore refine the modelling of the sedimentary iron source, the representation of iron scavenging ~~that now includes explicit representation of to include~~ redissolution, and we model three phytoplankton functional classes ~~, whose concentrations are derived from a steady-state logistic equation (Dunne et al., 2005).~~

~~The model's biogeochemical parameters are optimized by minimizing the quadratic mismatch of the nutrient and phytoplankton concentrations with the available observations. Following Frants et al. (2016), we consider (e.g. Dunne et al., 2005). Through these advances we are able to produce, for the first time,~~ a family of ~~state estimates corresponding to a range of sources,~~



expanded here to a greater range of sediment and hydrothermal sources. The spread in key metrics (e.g., the global carbon export) across data-constrained state estimates of the coupled Fe-P-Si cycles for a wide range of not only aeolian, but also hydrothermal and sedimentary sources. We find that these state estimates are roughly equally consistent with the observed macronutrient and dFe concentrations regardless of the iron source strengths. Analysis of our family of state estimates is used as a measure of the metric's uncertainty shows that the uncertainty in the iron sources stems not only from compensation between overlapping sources and sinks, but also from the ability of the different iron source types (aeolian, hydrothermal, and sedimentary) to compensate for each other despite their different spatial distributions.

We use the model to establish the geographic patterns of nutrient limitation and colimitation, and to quantify the export of each nutrient by each phytoplankton functional class. By partitioning our inverse-model estimates of the coupled Fe-P-Si cycles to address an important open question about the marine iron cycle: What are the relative contributions of the different iron sources to supporting the world ocean's export production? While there have been perturbation experiments with forward models (Tagliabue et al., 2009, 2010, 2014) where one type of source (e.g., hydrothermal or sedimentary) was shut down to assess the importance of dFe into contributions from the aeolian from the missing source, such experiments cannot quantify the true contribution of hydrothermal or sedimentary iron to biological production because of the nonlinearity of the iron cycle (Holzer et al., 2016). Moreover, these numerical experiments were conducted with definite but highly uncertain choices of the iron sources, and the models were not objectively constrained by the observations. Thus, in addition to presenting the first inverse model of the coupled Fe-P-Si cycles, we address the following key scientific questions:

1. How well can the modelled dFe, sedimentary, and hydrothermal sources, we quantify the role and efficiency of each  $\text{PO}_4$ , and  $\text{Si}(\text{OH})_4$  concentrations be fitted to observations for widely differing iron sources, and are there limits on the iron source strengths that are consistent with the observed dFe type in supporting carbon and opal export, concentrations?
2. What are the nutrient limitation patterns that emerge from the data-constrained estimates of the coupled nutrient cycles, given that direct observational data on these patterns is very sparse?
3. How well constrained are the phosphorus and opal exports of optimized state estimates with widely different iron sources?
4. What fractions of phosphorus and opal exports are supported by aeolian, hydrothermal, and sedimentary iron, and how do these fractions vary with the iron-source strengths?

In the following, we detail the model formulation in Section 2 and the optimization strategy in Section 3. In Section 4 we quantify the fidelity of the family of state estimates to the nutrient observations. We examine nutrient limitation in Section 5, export production in Section 6, and iron-attributed export in Section 7. Caveats are discussed in Section 8 and we summarize and conclude in Section 9.

## 2 Biogeochemical model

We distinguish three phytoplankton functional groups: non-diatom small and large phytoplankton as well as diatoms, with a nominal separation between small and large at a cell diameter of 2  $\mu\text{m}$ . We denote the molar  $\text{PO}_4$  uptake rate per unit volume of each class by  $U_c$ , where the subscript  $c \in \{\text{lrg}, \text{sml}, \text{dia}\}$  identifies functional class. The uptake rates  $U_c$  are only non-zero in the model's upper 73.4 m (two layers), the model's euphotic zone.

We consider the three nutrients  $\text{PO}_4$ ,  $\text{Si}(\text{OH})_4$ , and dFe and denote their concentrations by  $\chi^i$ , with  $i \in \{\text{P}, \text{Si}, \text{Fe}\}$ . We write the steady-state tracer equations for these concentrations by keying all biological production to the uptake  $U_c$  of phosphate as follows:

$$\mathcal{T}\chi_{\text{P}} = \sum_c (\mathcal{S}_c^{\text{P}} - 1)U_c - \gamma_g(\chi_{\text{P}} - \bar{\chi}_{\text{P}}^{\text{obs}}) \quad , \quad (1)$$

$$15 \quad \mathcal{T}\chi_{\text{Si}} = (\mathcal{S}^{\text{Si}} - 1)R^{\text{Si:P}}U_{\text{dia}} - \gamma_g(\chi_{\text{Si}} - \bar{\chi}_{\text{Si}}^{\text{obs}}) \quad , \quad (2)$$

$$\begin{aligned} \mathcal{T}\chi_{\text{Fe}} = & \sum_c (\mathcal{S}_c^{\text{Fe}} - 1)R^{\text{Fe:P}}U_c \\ & + (\mathcal{S}^{\text{s,POP}} - 1)J_{\text{POP}} + (\mathcal{S}^{\text{s,bSi}} - 1)J_{\text{bSi}} - J_{\text{dst}} \\ & + s_{\text{A}} + s_{\text{S}} + s_{\text{H}} \quad . \end{aligned} \quad (3)$$

In (1)–(3),  $\mathcal{T}$  is the advection-eddy-diffusion operator, the operators  $\mathcal{S}_c^i$  model the biogenic transport and remineralization of nutrient  $i$  taken up by functional class  $c$ , and the operators  $\mathcal{S}^{\text{s,POP}}$  and  $\mathcal{S}^{\text{s,bSi}}$  model the particle transport of scavenged iron and its partial redissolution at depth as the scavenging particles remineralize or dissolve (details in section 2.2). The iron scavenging rates per unit volume are  $J_{\text{POP}}$  for scavenging by particulate organic phosphorus (POP),  $J_{\text{bSi}}$  for scavenging by opal particles, and  $J_{\text{dst}}$  for scavenging by mineral dust (details in section 2.4.2). The terms  $s_{\text{A}}$ ,  $s_{\text{S}}$ , and  $s_{\text{H}}$  are the aeolian, sediment, and hydrothermal iron sources (details in section 2.4.1). The factors  $R^{\text{Si:P}}$  and  $R^{\text{Fe:P}}$  are the stoichiometric uptake ratios that allow us to key all production to phosphorus. These ratios are functions of the nutrient concentrations as described in section 2.3.3.

The terms proportional to  $\gamma_g$  in (1)–(2) fix the global mean phosphate and silicic acid concentrations through weak relaxation to their observed global means  $\bar{\chi}_{\text{P}}^{\text{obs}}$  and  $\bar{\chi}_{\text{Si}}^{\text{obs}}$ . This is necessary because the phosphorus and silicon cycles have no external sources and sinks to set the global mean in steady state. (For phosphate and silicic acid, external sources, e.g., riverine input, and loss to sediment burial are neglected.) We choose the restoring timescale  $\gamma_g^{-1} = 10^6$  years (“geological” restoring); there is no sensitivity to the precise value of  $\gamma_g$ .

Equations (1)–(3) are coupled via the uptake of  $\text{PO}_4$ , which depends on the concentrations of all three nutrients, via the iron scavenging that depends on the export fluxes of organic matter and opal, and via the sedimentary release of dFe, which is keyed to the flux of organic matter onto the sediments (Elrod et al., 2004), as discussed in detail below.

### 2.1 Circulation

We use the data-assimilated, steady (non-seasonal) circulation of Primeau et al. (2013) which has a horizontal resolution of  $2^\circ \times 2^\circ$  and 24 vertical levels whose thickness increases with depth. Temperature, salinity, radiocarbon, CFC-11, and  $\text{PO}_4$

have been used as constraints in the data assimilation. The circulation is constrained dynamically and the data assimilation used the wind-stress climatology of Trenberth et al. (1989) and specified horizontal and vertical viscosities of  $5 \times 10^4 \text{ m}^2 \text{ s}^{-1}$  and  $10^{-4} \text{ m}^2 \text{ s}^{-1}$ , respectively. The circulation's advective-diffusive transport operator has fixed horizontal and vertical eddy diffusivities of  $10^3 \text{ m}^2 \text{ s}^{-1}$  and  $10^{-5} \text{ m}^2 \text{ s}^{-1}$ , respectively. We emphasize that the circulation effectively provides a ventilation-weighted transport because it has been optimized against  $\text{PO}_4$  and the ventilation tracers CFC-11 and radiocarbon. The steady model circulation, which has no seasonal cycle, thus does not bias estimates of preformed nutrients in the way an annual-average circulation would.

## 2.2 Biogenic transport

Organic matter sinks as POP, dissolves, and remineralizes at depth. Inverse models of the phosphorus cycle (Primeau et al., 2013; Holzer and Primeau, 2013; Pasquier and Holzer, 2016) suggest that dissolved organic phosphorus (DOP) represents a relatively small fraction of the total dissolved phosphorus that we neglect here for simplicity and numerical efficiency (no DOP tracer). We note that our estimates of phosphorus export effectively capture the export due to DOP, despite DOP not being explicitly represented as a separate tracer. This is because the optimization of the biogeochemical parameters minimizes the mismatch with the observed  $\text{PO}_4$  distribution, which in the real ocean is determined by the remineralization of all organic phosphorus, including DOP. Because the particle transport is much faster than the fluid transport across a grid box, we approximate particle transport and remineralization, which acts as an interior source of nutrients, as instantaneous. We model this process for each phytoplankton functional class by the “source” operator,  $\mathcal{S}_c^{\text{P}}$ , which reassigns a “detrital” fraction  $f_c$  of the uptake rate to a remineralization rate throughout the water column, while a fraction  $1 - f_c$  remineralizes in situ where the uptake occurred. We therefore express  $\mathcal{S}_c^{\text{P}}$  in terms of a biogenic redistribution operator  $\mathcal{B}^{\text{P}}$  as

$$\mathcal{S}_c^{\text{P}} = 1 - f_c + \mathcal{B}^{\text{P}} f_c \quad . \quad (4)$$

(The operator  $\mathcal{B}^{\text{P}}$  does not have a functional class subscript  $c$  because it redistributes a unit uptake with the same profile regardless of functional class.) ~~Following Dunne et al. (2005), we model the detrital fraction~~ We assume that the remineralization of organic matter releases dFe and phosphate in the same ratio with which they were taken up. Therefore,  $\mathcal{S}_c^{\text{Fe}} = \mathcal{S}_c^{\text{P}}$ .

The values of  $f_c$ , which set the export efficiency of each class, are not directly constrainable by the data used here. The nutrient concentrations constrain only the total export of all classes, while the phytoplankton concentrations constrain the uptake, but not the export, of each class. We therefore use the optimized detrital fractions from the work of Dunne et al. (2005): The detrital  $f_c$  fractions are modelled as decreasing with temperature  $T$  so that  $f_c = f_c^0 e^{-k_f T}$ , with  $k_f = 0.032^\circ \text{C}^{-1}$  independent of class,  $f_{\text{sm1}}^0 = 0.14$ ,  $f_{\text{lr1}}^0 = 0.74$ , and we assign  $f_{\text{dia}} = f_{\text{lr1}}$ . ~~We assume that the remineralization of organic matter releases and phosphate in the same ratio with which they were taken up. Therefore,  $\mathcal{S}_c^{\text{Fe}} = \mathcal{S}_c^{\text{P}}$ .~~ The large and diatom classes are thus  $\sim 5$  times more efficient at exporting organic matter than the small class. We acknowledge that Richardson and Jackson (2007) suggest that small and large phytoplankton have similar export efficiencies. However, their very sparse data does not provide strong evidence that  $f_{\text{sm1}} = f_{\text{lr1}}$ , only that their values are uncertain. Indeed, our state estimates using the  $f_c$  values of Dunne et al. (2005) are

consistent with the data presented by Richardson and Jackson (2007). Plots of the fractional uptake of each class versus its fractional export (not shown here) are broadly consistent with Figure 1B of Richardson and Jackson (2007).

Following Najjar et al. (1992), we assume that the detrital production rate is fluxed as POP through the base of the euphotic zone at  $z_e = 73.4$  m with  $\phi^{\text{POP}}(z_e) = \int_{z_e}^0 f_c U_c dz$ , and that the POP flux attenuates with depth according to the Martin power law  $\phi^{\text{POP}}(z) = \phi^{\text{POP}}(z_e)(z/z_e)^{-b}$  due to remineralization in the aphotic zone. The operator  $\mathcal{B}^{\text{P}}$  therefore injects  $\text{PO}_4$  with the divergence of  $\phi^{\text{POP}}$  into the aphotic water column. The flux into the ocean bottom is remineralized in the lowest grid box as in the work of Primeau et al. (2013). The exponent  $b$  was determined to be  $b = 0.82$  using a restoring-type phosphate-only model. (Most parameters were optimized for the full coupled model – for details of our optimization strategy see section 3.3.)

The redistribution operator  $\mathcal{B}^{\text{Si}}$  similarly injects silicic acid into the aphotic water column with the divergence of the opal flux,  $\phi^{\text{bSi}}$ , which attenuates because of temperature dependent opal dissolution following Gnanadesikan (1999) and Holzer et al. (2014). For each latitude and longitude,  $\phi^{\text{bSi}}$  is computed as the solution to  $\partial_z \phi^{\text{bSi}}(z) = -(\kappa_{\text{Si}}^{\text{max}}/w_{\text{Si}}) \exp(-T_E/T(z)) \phi^{\text{bSi}}(z)$ , with the boundary condition  $\phi^{\text{bSi}}(z_e) = \int_{z_e}^0 R^{\text{Si:P}} f_{\text{dia}} U_{\text{dia}} dz$ . We use  $T_E = 11,481$  K as Gnanadesikan (1999) and the same detrital fraction  $f_{\text{dia}}$  for the opal export and diatom POP export. The parameter combination  $\kappa_{\text{Si}}^{\text{max}}/w_{\text{Si}}$  has nearly the same value as determined by Holzer et al. (2014), but was re-optimized here for a simple restoring-type model that takes subgrid topography into account (see below).

The scavenging operators  $\mathcal{S}^{\text{s,POP}}$  and  $\mathcal{S}^{\text{s,bSi}}$  act on  $J_{\text{POP}}$  and  $J_{\text{bSi}}$  to redistribute a fraction of the iron scavenged at every layer throughout the water column below the layer. In terms of the corresponding redistribution operators, we write

$$\mathcal{S}^{\text{s,POP}} = f^{\text{POP}} \mathcal{B}^{\text{s,POP}} \quad \text{and} \quad \mathcal{S}^{\text{s,bSi}} = f^{\text{bSi}} \mathcal{B}^{\text{s,bSi}} \quad , \quad (5)$$

where the fractions  $f^{\text{POP}}$  and  $f^{\text{bSi}}$  were both fixed at 0.9 (see Appendix D). The operators  $\mathcal{B}^{\text{s,POP}}$  and  $\mathcal{B}^{\text{s,bSi}}$  in effect “recycle” scavenged iron. They are very similar to  $\mathcal{B}^{\text{P}}$  and  $\mathcal{B}^{\text{Si}}$  but in addition to distributing scavenged iron from the euphotic zone to the aphotic zone, they also redistribute the scavenging rates of every aphotic layer to a source of redissolving iron with the divergence of the scavenging particle fluxes. The flux of scavenged iron into the bottom is assumed to be lost forever so that there would be iron loss even for 100 % efficient recycling of scavenged iron. (For details see Appendices A and B.)

To compute accurate particle fluxes for constructing all  $\mathcal{S}$  operators, we take sub-grid topography into account (as done by Moore and Braucher, 2008), using the high-resolution ETOPO2V2c data set (National Geophysical Data Center, 2006). This is done by calculating for each grid box the fractional area occupied by the sub-grid topography, which is also the fraction of the particle flux that is intersected by the sub-grid topography. For each grid box, the fraction of the flux intersected is instantly remineralized or dissolved (details in Appendix B).

### 2.3 Uptake rates

The  $\text{PO}_4$  uptake rate at a point is a function of the local temperature  $T$ , irradiance  $I$ , and nutrient concentrations. The uptake rate for functional class  $c$  is calculated as the product of its phytoplankton concentration,  $p_c$ , and its specific growth rate,  $\mu_c$ , as

$$U_c = \mu_c p_c = \frac{p_c^{\text{max}}}{\tau_c} e^{\kappa T} (F_{I,c} F_{N,c})^2 \quad , \quad (6)$$

where  $\tau_c$  is the timescale for growth,  $p_c^{\max}$  is the phytoplankton concentration under ideal conditions, and  $F_{I,c}$  and  $F_{N,c}$  are dimensionless factors in the interval  $[0, 1)$  that represent light and nutrient limitation, respectively, as defined below. We derive equation (6) similarly to Dunne et al. (2005) and Galbraith et al. (2010) as follows.

First,  $p_c$  is calculated diagnostically by assuming steady state between growth and mortality, which avoids the need to carry explicit plankton concentration tracers. This is justified by the coarse resolution of our model, which implies transport timescales across a grid box much larger than the typical timescales for phytoplankton growth. Based on Dunne et al. (2005)’s mortality formulation,  $p_c$  can be modelled by a logistic equation

$$\partial_t p_c = \mu_c p_c - \lambda \left( \frac{p_c}{p_c^*} \right) p_c, \quad (7)$$

where the  $p_c/p_c^*$  scales the specific mortality rate  $\lambda$ , and  $p_c^*$  has also been referred to as the “pivotal” population density (e.g., Dunne et al., 2005; Galbraith et al., 2010). Equation (7) has a nontrivial steady state, given by

$$p_c = \left( \frac{\mu_c}{\lambda} \right) p_c^*. \quad (8)$$

We assume that all phytoplankton classes share the same specific mortality rate  $\lambda$ , which depends only on temperature. For simplicity, we follow Galbraith et al. (2010) and approximate the  $T$  dependence of  $\lambda$  to be identical to that of the growth rate. We adopt a  $T$  dependence of the form  $e^{\kappa T}$  as determined by Eppley (1972) for the growth rate. Thus,  $\lambda = \lambda_0 e^{\kappa T}$ , where  $\lambda_0$  is a constant and  $T$  is in  $^{\circ}\text{C}$ .

Our formulation differs from that of Dunne et al. (2005) and Galbraith et al. (2010), who raise the ratio  $p_c/p_c^*$  to a power  $\alpha = 1$  or  $\alpha = 1/3$  to differentiate between their small and large phytoplankton classes. Here, we instead differentiate between classes by assigning them different half-saturation rates and maximum uptake rate constants similarly to the work of Matsumoto et al. (2008) (details in sections 2.3.1 and 2.3.3).

We model the specific growth rate  $\mu_c$  as multiplicatively colimited (Saito et al., 2008) by temperature, light, and nutrients:

$$\mu_c = \frac{1}{\tau_c} e^{\kappa T} F_{I,c} F_{N,c}, \quad (9)$$

where  $\tau_c$  is the growth timescale at  $0^{\circ}\text{C}$  under ideal conditions and the temperature dependence  $e^{\kappa T}$  (Eppley, 1972) is identical to that used for the mortality rate (e.g., Galbraith et al., 2010). To group parameters for more efficient optimization, we define  $p_c^{\max} = p_c^*/(\lambda_0 \tau_c)$ , so that diagnostic equation (8) for the phytoplankton concentration becomes

$$p_c = F_{I,c} F_{N,c} p_c^{\max}. \quad (10)$$

Substituting (9) and (10) into  $U_c^P = \mu_c p_c$  gives (6), which is similar to the uptake formulation of Doney et al. (2006) and Matsumoto et al. (2008).

We note that in the Sea of Japan the model’s circulation produces unrealistic nutrient trapping, likely due to under resolved currents. For this reason we set the specific growth rate in the Sea of Japan to zero, effectively removing it from the computational domain of the biogeochemical model.

### 2.3.1 Nutrient limitation

We model the limitation of functional class  $c$  by nutrient  $i$  by a Monod function (Monod, 1942) of the concentration,  $\chi_i/(\chi_i + k_c^i)$ , where  $k_c^i$  is the half-saturation constant that determines the scale on which the concentration influences uptake. (Because only diatoms take up silicon  $k_{\text{lr}g}^{\text{Si}} = 0$  and  $k_{\text{sm}l}^{\text{Si}} = 0$ .) For the colimitation of all three nutrients, we use the type-I multiplicative form (Saito et al., 2008)

$$F_{N,c} = \prod_i \frac{\chi_i}{\chi_i + k_c^i} . \quad (11)$$

We chose the Monod model over the arguably more realistic quota model (e.g., Flynn, 2003) for simplicity. Moreover, the shortcomings of the Monod formulation likely only come into play for rapidly evolving transient blooms, which our steady-state formulation does not attempt to capture.

Using a minimum over nutrient type  $i$  (Liebig's rule, e.g., de Baar, 1994), rather than the product (11) is thought to fit the observational data slightly better (e.g., Rhee, 1978; Droop, 2009). However, here we prefer the smoothness of the multiplicative formulation because differentiability is a theoretical requirement for Newton's method to converge (e.g., Kelley, 2003a). A product of  $\text{PO}_4$ , dFe, and irradiance Monod terms was also used by Parekh et al. (2005) and Dutkiewicz et al. (2006) in the uptake formulation of their coupled phosphorus-iron model.

### 2.3.2 Light limitation

We prescribe irradiance  $I$  and model light limitation with a simple Monod factor

$$F_{I,c} = \frac{I}{I + k_{I,c}} , \quad (12)$$

with half-saturation constant  $k_{I,c}$  for class  $c$  (e.g., Doney et al., 2006). We use an annual mean  $I$  derived from photosynthetically active radiation (PAR) measured over the period 2002–2015 by the Modis Aqua satellite (NASA Goddard Space Flight Center, 2014). The surface PAR at location  $(x, y)$ , denoted by  $I_0(x, y)$ , was converted to  $\text{W m}^{-2}$  using  $2.77 \times 10^{18} \text{ quanta s}^{-1} \text{ W}^{-1}$  (Morel and Smith, 1974). Irradiance is modelled as exponentially attenuated with depth  $z$  so that

$$I(x, y, z) = I_0(x, y) e^{-k_w z} , \quad (13)$$

with  $k_w^{-1} = 25 \text{ m}$ .

### 2.3.3 Elemental uptake ratios

Because we key all biological production to  $\text{PO}_4$  utilization, we must specify the Fe : P and Si : P elemental uptake ratios for the iron and silicon cycles. The Fe : P uptake ratio,  $R^{\text{Fe:P}}$ , is known to increase and saturate with increasing dFe concentration (e.g., Sunda and Huntsman, 1997). We follow Galbraith et al. (2010) and model the dFe dependence as a simple Monod term

$$R^{\text{Fe:P}} = R_0^{\text{Fe:P}} \frac{\chi_{\text{Fe}}}{\chi_{\text{Fe}} + k_{\text{Fe:P}}} , \quad (14)$$

where  $R_0^{\text{Fe:P}}$  is the maximal Fe : P uptake ratio. In principle,  $R_0^{\text{Fe:P}}$  and  $k_{\text{Fe:P}}$  could be different for different functional classes. However, ~~we find that when optimized they tend to be nearly equal for different classes, so that we chose for simplicity to use the same values of  $R_0^{\text{Fe:P}}$  constraining class-dependent Fe quotas is beyond the scope of what is possible with our inverse model: different values of  $R_0^{\text{Fe:P}}$  and  $k_{\text{Fe:P}}$  for all-classes each class would directly compensate for one another in the global Fe export.~~ As noted by Galbraith et al. (2010), this formulation ignores the effects of light limitation suggested by several studies (e.g., Sunda and Huntsman, 1997; Strzepek et al., 2012).

Equation (14) ~~also does not encode effectively encodes~~ a minimum iron requirement ~~of zero. Thus, for very low dFe concentrations, our Fe : P ratio  $R^{\text{Fe:P}}$  would fall below a realistic cell quota, but this has no mechanistic consequence because for such low dFe concentrations there is essentially no uptake in our formulation. This is because  $R^{\text{Fe:P}}$  is proportional to a dFe Monod term, while P uptake is proportional to the square of a dFe Monod term. Thus, as dFe becomes small, the uptake goes to zero faster than  $R^{\text{Fe:P}}$  itself. Simply put, this means that when the Fe : P ratio is unrealistically small, it does not matter because there is no P or Fe uptake.~~ When we introduced ~~such a minimum a nonzero minimum for  $R^{\text{Fe:P}}$~~ , it tended to be optimized to zero ~~and we therefore ignore it for simplicity.~~, which means that a simple Monod factor suffices to capture the dFe dependence of the Fe : P ratio where there is significant uptake.

The Monod formulation (14) does capture luxury iron uptake (e.g., Marchetti et al., 2009a) when the half-saturation constant of (14) exceeds the half-saturation constant of the iron limitation in (11), as made explicit by Galbraith et al. (2010). This is the case for our optimized value of  $k_{\text{Fe:P}}$  so that phytoplankton has the luxury to increase its iron uptake with increasing dFe concentration even when iron is not limiting.

Our representation of the  $R^{\text{Si:P}}$  uptake ratio takes into consideration field studies and iron enrichment experiments, which have indicated that in HNLC regions and upwelling regions iron limitation leads to increased diatom silicification, i.e., increased cellular Si : N and Si : P ratios (e.g., Takeda, 1998; Hutchins and Bruland, 1998; Franck et al., 2000; Brzezinski et al., 2003). However, there is no literature consensus on a mechanistic formulation of the iron dependence of silicic-acid uptake. For example, Matsumoto et al. (2013) assume a Si : N uptake ratio inversely proportional to the dFe concentration (capped at a minimum), while Jin et al. (2006) assume the Si : N ratio to depend only on the  $\text{Si(OH)}_4$  concentration. Others suggest that the dFe concentration only impacts the diatom growth rate and not the cellular Si : C ratio, while the  $\text{Si(OH)}_4$  concentration impacts the cellular Si : C ratio and not growth rate (e.g., Marchetti et al., 2009b; Brzezinski et al., 2011a). Here, we chose to retain the effects of increased silicification due to iron limitation and the impact of high  $\text{Si(OH)}_4$  concentration on silicification (Brzezinski 2016, personal communication). We model these effects with the formulation

$$R^{\text{Si:P}} = R_0^{\text{Si}} + (R_m^{\text{Si}} - R_0^{\text{Si}}) \frac{k_{\text{Si:P}}^{\text{Fe}}}{\chi_{\text{Fe}} + k_{\text{Si:P}}^{\text{Fe}}} \frac{\chi_{\text{Si}}}{\chi_{\text{Si}} + k_{\text{Si:P}}^{\text{Si}}} \quad (15)$$

The ratio involving the  $\chi_{\text{Fe}}$  produces increased silicification when iron is deficient, while the Monod term for  $\chi_{\text{Si}}$  produces increased silicification in silicon-replete environments: If  $\chi_{\text{Fe}} \rightarrow 0$  and  $\chi_{\text{Si}} \gg k_{\text{Si:P}}^{\text{Si}}$ , then  $R^{\text{Si:P}} \rightarrow R_m^{\text{Si}}$ , while if  $\chi_{\text{Fe}} \gg k_{\text{Si:P}}^{\text{Fe}}$  or  $\chi_{\text{Si}} \rightarrow 0$ , then  $R^{\text{Si:P}} \rightarrow R_0^{\text{Si}}$ . The minimum and maximum Si : P ratios  $R_0^{\text{Si}}$  and  $R_m^{\text{Si}}$ , as well as the constants  $k_{\text{Si:P}}^{\text{Fe}}$  and  $k_{\text{Si:P}}^{\text{Si}}$  were tuned rather than fully optimized to achieve the observation-based fractional uptake of each functional class (see section

3.3 on optimization for details). (Plots of the experimental data that show increased silicification under conditions of low dFe can be seen in Figure 6 of Franck et al. (2000) and in Figure 7 of Brzezinski et al. (2003).)

## 2.4 Iron model

### 2.4.1 Iron sources

- 5 The aeolian source,  $s_A$ , is based on the spatial pattern of the surface flux of atmospheric soluble iron of Luo et al. (2008), obtained from an atmospheric model for current climate conditions that includes size-partitioned mineral dust, biomass burning, and industrial emissions. Because the global strength  $\sigma_A \equiv \int s_A(\mathbf{r}) d^3\mathbf{r}$  of the aeolian source is highly uncertain (e.g., Tagliabue et al., 2016), we scale the global amplitude of this pattern to an initial guess of the global source strength that is then refined in our final optimization step (see sections 3.3 and 3.4 below). We note that the model of Luo et al. (2008) estimate a soluble aeolian iron flux into the ocean of  $\sigma_A \sim 6 \text{ Gmol yr}^{-1}$ . (The global source strength of iron type  $k$  is defined as
- 10  $\sigma_k \equiv \int s_k(\mathbf{r}) d^3\mathbf{r}$ .)

The sedimentary source,  $s_S$ , has the pattern of the POP flux reaching the sediments (Elrod et al., 2004; Frants et al., 2016) and accounts for both resolved and subgrid topography. The amplitude of this pattern is the global sediment iron source strength  $\sigma_S$ , which is an optimized parameter. The dependence of the sediment redox reaction on dissolved oxygen (e.g., Galbraith et al., 2010) is ignored here for simplicity and to avoid carrying oxygen as another tracer. Unlike in the model of Frants et al. (2016), the phosphorus cycle and POP flux are not prescribed but coupled to the iron and silicon cycles as described above.

To model the hydrothermal source,  $s_H$ , we use the  $^3\text{He}$  source pattern of the Ocean-Carbon Cycle Model Intercomparison Project (OCMIP) protocol (Dutay et al., 2004), and jointly optimize the hydrothermal iron source strengths  $\sigma_{H,ATL}$ ,  $\sigma_{H,PAC}$ ,  $\sigma_{H,IND}$ , and  $\sigma_{H,SO}$  of the Atlantic, Pacific, Indian, and Southern Ocean ridge systems, as in the work of Frants et al. (2016).

### 20 2.4.2 Iron sinks

Dissolved iron can be chelated by ligands or “free”. We assume that the scavenging acts only on the concentration  $\chi_{\text{Fe}'}$  of free iron so that chelation by ligands protects dFe from being scavenged. Scavenging is modelled as a first order process (e.g., Aumont et al., 2015) so that the scavenging rate is proportional to the product of  $\chi_{\text{Fe}'}$  and the concentration of the scavenging particles  $\chi^j$ , for  $j \in \{\text{POP}, \text{bSi}, \text{dst}\}$ , the three types of particles considered. For each particle type, the scavenging rate per

25 unit volume is thus modelled as

$$J_j = \kappa_{\text{scv}}^j \chi^j \chi_{\text{Fe}'} \quad , \quad (16)$$

where the scavenging rate constants  $\kappa_{\text{scv}}^{\text{POP}}$ ,  $\kappa_{\text{scv}}^{\text{bSi}}$ , or  $\kappa_{\text{scv}}^{\text{dst}}$  are optimizable parameters.

To compute the concentration of the scavenging particles, we use the fact that the flux divergences generated by the biogenic transport operators must be balanced by local remineralization or dissolution rates, that is,

$$30 \quad \mathcal{B}^P \sum_c f_c U_c = \kappa_P \chi^{\text{POP}} \quad (17)$$



and

$$\mathcal{B}^{\text{Si}} f_{\text{dia}} U_{\text{dia}} = \kappa_{\text{Si}}^{\text{max}} e^{-T_E/T} \chi^{\text{bSi}} . \quad (18)$$

Although we use the nominal values of  $\kappa_{\text{P}}$  and  $\kappa_{\text{Si}}^{\text{max}}$  listed in Table 1, note that these constants only enter the scavenging rates (16) through the combinations  $\kappa_{\text{scv}}^{\text{POP}}/\kappa_{\text{P}}$  and  $\kappa_{\text{scv}}^{\text{bSi}}/\kappa_{\text{Si}}$ , where  $\kappa_{\text{scv}}^{\text{POP}}$  and  $\kappa_{\text{scv}}^{\text{bSi}}$  are optimized. The concentration of dust particles is modelled as vertically uniform due to sinking dust particles that do not dissolve or re-suspend from sediments (e.g., Moore and Braucher, 2008). We use the geographic pattern of the dust mass flux into the ocean provided by Luo et al. (2008), which we convert to a particle concentration using a nominal sinking speed of  $w_{\text{dst}} = 50 \text{ m day}^{-1}$ . The exact value of  $w_{\text{dst}}$  does not matter because the dust scavenging rate depends only on  $\kappa_{\text{scv}}^{\text{dst}}/w_{\text{dst}}$  and  $\kappa_{\text{scv}}^{\text{dst}}$  is optimized.

The key control on shaping the free iron concentration, and hence the scavenging, is the ligand concentration  $L$ . Chemical equilibrium between ligands, total dFe, and free iron determines  $\chi_{\text{Fe}'}$  as a quadratic function of the (total) dFe concentration (see, e.g., Frants et al., 2016). We used the same ligand stability constant of  $K_L = 8 \times 10^{10} \text{ kg (mol Lig)}^{-1}$  as Frants et al. (2016). The ligand concentration itself is modelled to have a uniform background value  $L_b$  that can be enhanced in old waters (Misumi et al., 2013) and in hydrothermal plumes (e.g., Bennett et al., 2008; Hawkes et al., 2013), similar to the formulation of Frants et al. (2016). Specifically, we use

$$L = \max(L_{\text{H}} + L_{\text{sw}}, L_{\text{b}}) , \quad (19)$$

where the  $L_{\text{H}}$  and  $L_{\text{sw}}$  are the elevated hydrothermal and aged “sea water” ligand concentrations, which we model as follows. The hydrothermal ligand plumes are computed from the source-sink balance

$$\mathcal{T} L_{\text{H}} = -\frac{1}{\tau_{\text{b}}} L_{\text{H}} - \frac{\Delta_{\text{v}}}{\tau_{\text{v}}} (L_{\text{H}} - L_{\text{v}}) , \quad (20)$$

where  $\Delta_{\text{v}}$  is a mask that is unity for grid boxes containing vent sites (taken from the OCMIP  $^3\text{He}$  source (Dutay et al., 2004)) and zero elsewhere. The timescale  $\tau_{\text{v}} = 1 \text{ s}$  clamps the ligand concentration to  $L_{\text{v}}$  at the vents, and the timescale  $\tau_{\text{b}}$  controls the plume spread by setting the rate with which  $L_{\text{H}}$  decays away from the vents. The ligand concentration  $L_{\text{sw}}$  is enhanced in old waters according to

$$L_{\text{sw}}(\mathbf{r}) = \frac{\Gamma(\mathbf{r})}{\Gamma_{\text{max}}} L_{\text{max}} , \quad (21)$$

where  $\Gamma(\mathbf{r})$  is the ideal mean water age (easily computed for our model), we choose  $\Gamma_{\text{max}} = 1600 \text{ yr}$  following Frants et al. (2016), and  $L_{\text{max}}$  together with  $\tau_{\text{b}}$ ,  $L_{\text{v}}$ , and  $L_{\text{b}}$  are optimizable parameters.

As is the case for most iron models, there is no need to explicitly represent the chemical precipitation of dFe. This is because in most formulations the scavenging rates increase rapidly when dFe exceeds a certain threshold. For our model this threshold is set by the ligand concentration  $L$ : in chemical equilibrium  $\chi_{\text{Fe}'}$  and hence the scavenging rate rise rapidly when dFe concentrations exceed  $L$ .

### 3 Numerical method, parameter optimization, and family of state estimates

#### 3.1 Steady state solution

All three-dimensional fields (e.g., the concentrations  $\chi_i$ ) are discretized on our model grid and organized into column vectors (length  $n = 191\,169$  at our resolution). Linear operators such as  $\mathcal{T}$ ,  $\mathcal{S}_c^i$ , and  $\mathcal{S}^{s,j}$  are correspondingly organized into  $n \times n$  sparse matrices. The steady-state tracer equations (1)–(3) then become a system of  $3n \times 3n$  equations that are nonlinear because of the iron scavenging and the colimitation of the  $\text{PO}_4$  uptake.

The  $3n \times 3n$  system is solved efficiently using Newton’s method (e.g., Kelley, 2003a, b). Convergence of the Newton method depends on the initial guess for the solution and is not guaranteed. For the initial guess of  $\chi_P$  and  $\chi_{\text{Si}}$  we use the annual mean fields of the World Ocean Atlas (WOA13, Garcia et al., 2014) interpolated to our grid, and for the initial guess of  $\chi_{\text{Fe}}$  we use the dFe fields estimated by Frants et al. (2016). The Newton solver typically converges to numerical precision in  $\sim 10$  iterations.

#### 3.2 Cost function

We optimize the model parameters by systematically minimizing a quadratic cost function of the mismatch between modelled and observed fields. For  $\text{PO}_4$  and  $\text{Si}(\text{OH})_4$ , for which gridded climatologies are available, we define the weights based on the grid-box volumes, organized into vector  $\mathbf{v}$ , as

$$\mathbf{w}_P = \frac{\mathbf{v}}{(\overline{\chi}_P^{\text{obs}})^2 V} \quad , \quad \text{and} \quad \mathbf{w}_{\text{Si}} = \frac{\mathbf{v}}{(\overline{\chi}_{\text{Si}}^{\text{obs}})^2 V} \quad , \quad (22)$$

where we have normalized the weights by the total ocean volume  $V$  and the squared global mean observed concentrations. This non-dimensionalizes the quadratic cost terms and scales them to the same order of magnitude. For dFe, for which only sparse observations are available, we also define weights  $\mathbf{w}_{\text{Fe}}$  based on grid box volumes, but observations that are part of a vertical profile receive additional weight as detailed in Appendix C.

With diagonal weight matrix  $\mathbf{W}_i = \text{diag}(\mathbf{w}_i)$  for the  $i^{\text{th}}$  nutrient, its cost for the mismatch with observations is then given by

$$E_i = \delta\chi_i^T \mathbf{W}_i \delta\chi_i \quad , \quad (23)$$

where  $\delta\chi_i \equiv \chi_i - \chi_i^{\text{obs}}$ . For  $\chi_P^{\text{obs}}$  and  $\chi_{\text{Si}}^{\text{obs}}$  we use WOA13 fields interpolated to our grid, and for  $\chi_{\text{Fe}}^{\text{obs}}$  we used the GEO-TRACES intermediate data product (Mawji et al., 2015) and the data set compiled by Tagliabue et al. (2012).

The cost terms for the nutrient mismatch do not provide a strong constraint on the relative sizes of the phytoplankton class because the nutrients are determined by their combined export. We therefore include additional terms in our cost function that constrain the phytoplankton concentrations  $p_c$  to the recent satellite derived estimates of Kostadinov et al. (2009). These estimates provide phytoplankton concentrations for picophytoplankton ( $0.5\text{--}2\,\mu\text{m}$  in diameter), nanophytoplankton ( $2\text{--}20\,\mu\text{m}$ ), and microphytoplankton ( $20\text{--}50\,\mu\text{m}$ ), which we identify with our small, large, and diatom functional classes. We use the entire mission composite data set as the satellite climatology (Kostadinov et al., 2016).

Because of the large dynamic range of the phytoplankton concentrations, we consider mismatches in the log of the concentrations, that is,  $\delta\pi_c \equiv \log[(p_c + \epsilon_c)/p_0] - \log[(p_c^{\text{obs}} + \epsilon_c)/p_0]$ , where  $\epsilon_c = \overline{p_c^{\text{obs}}}$  is introduced to limit the logarithm where the phytoplankton concentration falls to zero. For each class, we construct normalized weight vectors

$$\mathbf{w}_c = \frac{\mathbf{v}}{[\log(\overline{p_c^{\text{obs}}}/p_0)]^2 V_{\text{eup}}} \quad , \quad (24)$$

5 where  $V_{\text{eup}}$  is the global euphotic volume, and  $p_0 = 1 \text{ mg C m}^{-3}$  nondimensionalizes the argument of the logarithm.

Organizing mismatches and weights into vectors and diagonal matrices, we calculate the cost for the phytoplankton concentration mismatch as

$$E_{\text{plk}} = \sum_c \delta\pi_c^T \mathbf{W}_c \delta\pi_c \quad , \quad (25)$$

and combine the costs for the nutrient and plankton mismatches into the total cost

$$10 \quad E_{\text{tot}} = \omega_P E_P + \omega_{\text{Si}} E_{\text{Si}} + \omega_{\text{Fe}} E_{\text{Fe}} + \omega_{\text{plk}} E_{\text{plk}} \quad , \quad (26)$$

which we minimize to constrain our model parameters by the available observations. In (26) the  $\omega$  weights were chosen such that the four cost terms contribute roughly equally to the total cost for a typical member of our family of state estimates. This was achieved with  $(\omega_P, \omega_{\text{Si}}, \omega_{\text{Fe}}, \omega_{\text{plk}}) = (1, 0.47, 0.044, 0.30)$ , the smaller weight for dFe reflecting its larger root-mean-square (RMS) mismatch and hence much larger cost  $E_{\text{Fe}}$ .

### 15 3.3 Optimization strategy

Our model has  $\sim 50$  biogeochemical parameters that can in principle be determined through objective optimization given appropriate observational data. However, even with perfect data, some parameters can compensate for others (e.g., two parameters appearing as a ratio) so that not all parameters are independent. Other parameters cannot be optimized because the mismatch with available nutrient and phytoplankton data is not sensitive to their value. In practice, it therefore is not possible to optimize  
20 all parameters, and care is needed to optimize only those parameters that independently shape the nutrient and phytoplankton concentrations.

The parameters associated with the remineralization of phosphate and dissolution of opal are well constrained by the high-quality climatologies of  $\text{PO}_4$  and  $\text{Si}(\text{OH})_4$ . However, the iron cycle is relatively poorly constrained because the dFe data is much more sparse in both time and space, and estimates of the iron sources range over two orders of magnitude (e.g., Tagliabue  
25 et al., 2016). Moreover, the ligand field that determines the scavengable free iron is highly uncertain. Given these challenges, the recent inverse model of the iron cycle by Frants et al. (2016) considered a family of state estimates for a range of external source strengths, an approach we will follow here for our coupled model.

Another key consideration is computational cost. Even with the numerically efficient Newton Solver, optimization typically requires hundreds of solutions of equations (1)–(3) per optimized parameter. We therefore optimized no more than 13 pa-  
30 rameters at a time. We acknowledge that the minimum attained by sequentially optimizing groups of independent parameters

**Table 1.** Parameters that were prescribed from the literature, or that were separately optimized in a submodel.

Parameter	Description	Value	Unit	Source
$\kappa$	Growth and mortality temperature coefficient	0.063	$(^{\circ}\text{C})^{-1}$	Eppley (1972)
$k_w$	Irradiance attenuation coefficient	0.040	$\text{m}^{-1}$	Dutkiewicz et al. (2005)
$K_L$	Ligand stability constant	$8 \times 10^{10}$	$\text{kg}(\text{mol Lig})^{-1}$	Galbraith et al. (2010)
$\Gamma_{\max}$	Age coefficient for ligand parameterization	1600.	yr	Frants et al. (2016)
$f^{\text{POP}}$	Recyclable fraction of POP-scavenged dFe	0.90	–	Moore and Braucher (2008)
$f^{\text{bSi}}$	Recyclable fraction of opal-scavenged dFe	0.90	–	Moore and Braucher (2008)
$f_{\text{dia}}^0$	Diatom class detrital fraction at $0^{\circ}\text{C}$	0.74	–	Dunne et al. (2005)
$f_{\text{lr}}^0$	Large class detrital fraction at $0^{\circ}\text{C}$	0.74	–	Dunne et al. (2005)
$f_{\text{sml}}^0$	Small class detrital fraction at $0^{\circ}\text{C}$	0.14	–	Dunne et al. (2005)
$\kappa_{\text{P}}$	POP remineralization rate constant	0.03	$\text{d}^{-1}$	Kriest and Oschlies (2008)
$\kappa_{\text{Si}}^{\max}$	Opal dissolution rate coefficient	$13 \times 10^{15}$	$\text{d}^{-1}$	Gnanadesikan (1999)
$T_{\text{E}}$	Temperature scale for opal dissolution	11481.	K	Gnanadesikan (1999)
$k_{\text{dia}}^{\text{Si}}$	Diatom class $\text{Si}(\text{OH})_4$ half-saturation constant	1.0	$\text{mmol Si m}^{-3}$	Matsumoto et al. (2013)
$w_{\text{Si}}$	Opal sinking speed	40.	$\text{m d}^{-1}$	Submodel optimization
$b$	POP flux Martin exponent	0.82	–	Submodel optimization

is generally different than jointly optimizing all independent parameters, but computational and practical considerations demanded a sequential approach. We justify this a posteriori by the fact that we are able to achieve fits to the observed nutrient concentration fields with RMS mismatches similar to those of other recent data-constrained models (e.g., Primeau et al., 2013; Holzer et al., 2014; Frants et al., 2016). Given these considerations, we adopted the following strategy:

- 5 (i) Parameters that are measurable and considered well-known, as well as parameters that are unconstrainable by our cost function or whose value is not critical because they are strongly compensated by other parameters, were assigned values from the literature as collected in Table 1. The considerations that entered our choice of prescribed parameters are detailed in Appendix D.
- (ii) The parameters that set the phosphate remineralization and opal dissolution profiles were optimized by minimizing the mismatch with  $\text{PO}_4$  and  $\text{Si}(\text{OH})_4$  concentration data from the WOA13 using separate single-nutrient models. For the Si cycle, we used the model of Holzer et al. (2014) and verified that the opal sinking speed parameter  $w_{\text{Si}}$  was not affected by the inclusion of sub-grid topography (Appendix B). For the P cycle, we used a similar conditional restoring model without POP, but with subgrid topography, and optimized the Martin exponent  $b$ . The resulting values of  $w_{\text{Si}}$  and  $b$  (Table 1) were held fixed for all optimizations of the coupled nutrient cycling model.
- 10

**Table 2.** Optimized parameters and range across family of state estimates.

Parameter	Description	Initial Value	Optimized Value	Range	Unit
$k_{I,\text{dia}}$	Diatom class irradiance half-saturation rate	20.	8.1	–	$\text{W m}^{-2}$
$k_{I,\text{lrg}}$	Large class irradiance half-saturation rate	20.	9.0	–	$\text{W m}^{-2}$
$k_{I,\text{sml}}$	Small class irradiance half-saturation rate	20.	8.8	–	$\text{W m}^{-2}$
$R_{\text{m}}^{\text{Si}}$	Diatom maximum Si : P	160.	220.	–	$\text{mol Si (mol P)}^{-1}$
$R_0^{\text{Si}}$	Diatom minimum Si : P	8.0	13.	–	$\text{mol Si (mol P)}^{-1}$
$k_{\text{Si:P}}^{\text{Si}}$	Silicon half-saturation constant in Si : P	30.	4.0	–	$\text{mmol Si m}^{-3}$
$k_{\text{Si:P}}^{\text{Fe}}$	Iron hyperbolic constant in Si : P	1.0	0.077	–	$\text{nM Fe}$
$k_{\text{Fe:P}}$	Iron half-saturation constant in Fe : P	0.74	0.74	–	$\text{nM Fe}$
$k_{\text{dia}}^{\text{P}}$	Diatom class $\text{PO}_4$ half-saturation constant	0.39	0.72	–	$\text{mmol P m}^{-3}$
$k_{\text{lrg}}^{\text{P}}$	Large class $\text{PO}_4$ half-saturation constant	0.39	0.72	–	$\text{mmol P m}^{-3}$
$k_{\text{sml}}^{\text{P}}$	Small class $\text{PO}_4$ half-saturation constant	0.030	0.13	–	$\text{mmol P m}^{-3}$
$k_{\text{dia}}^{\text{Fe}}$	Diatom class dFe half-saturation constant	0.10	0.30	–	$\text{nM Fe}$
$k_{\text{lrg}}^{\text{Fe}}$	Large class dFe half-saturation constant	0.10	0.29	–	$\text{nM Fe}$
$k_{\text{sml}}^{\text{Fe}}$	Small class dFe half-saturation constant	0.010	0.11	–	$\text{nM Fe}$
$p_{\text{dia}}^{\text{max}}$	Diatom class maximum concentration	23.	42.	–	$\text{mg C m}^{-3}$
$p_{\text{lrg}}^{\text{max}}$	Large class maximum concentration	23.	61.	–	$\text{mg C m}^{-3}$
$p_{\text{sml}}^{\text{max}}$	Small class maximum concentration	23.	21.	–	$\text{mg C m}^{-3}$
$\tau_{\text{dia}}$	Maximal diatom growth timescale	6.0	0.65	–	d
$\tau_{\text{lrg}}$	Maximal large growth timescale	6.0	1.5	–	d
$\tau_{\text{sml}}$	Maximal small growth timescale	6.0	7.4	–	d
$R_0^{\text{Fe:P}}$	Maximum Fe : P uptake ratio	5.0	2.0	<del>0.00047</del> <u>0.52</u> –3.0	$\text{mmol Fe (mol P)}^{-1}$
$\kappa_{\text{scv}}^{\text{POP}}$	POP scavenging rate constant	0.13	1.0	0.015–7.9	$(\text{mmol POP m}^{-3})^{-1} \text{d}^{-1}$
$\kappa_{\text{scv}}^{\text{bSi}}$	Opal scavenging rate constant	3.1	1.3	0.85–13.	$(\text{mol bSi m}^{-3})^{-1} \text{d}^{-1}$
$\kappa_{\text{scv}}^{\text{dst}}$	Dust scavenging rate constant	10000.	9.4	8.5–10.	$(\text{g dust m}^{-3})^{-1} \text{d}^{-1}$
$L_{\text{b}}$	Background ligand concentration	1.0	0.51	0.40–0.72	$\text{nM Lig}$
$L_{\text{v}}$	Maximal hydrothermal vent ligand conc.	3.0	1.2	0.68–1.4	$\text{nM Lig}$
$\tau_{\text{b}}$	Hydrothermal vent plume restoring timescale	10.	5.7	3.0–7.5	yr
$L_{\text{max}}$	Maximal age-enhanced ligand conc.	2.3	0.97	0.82–1.3	$\text{nM Lig}$
$\sigma_{\text{A}}$	Aeolian source strength	1.9	5.3	0.63–22.	$\text{Gmol Fe yr}^{-1}$
$\sigma_{\text{S}}$	Sedimentary source strength	4.2	1.7	0.11–22.	$\text{Gmol Fe yr}^{-1}$
$\sigma_{\text{H,ATL}}$	Hydrothermal source strength, Atlantic	0.098	0.19	0.00013–0.50	$\text{Gmol Fe yr}^{-1}$
$\sigma_{\text{H,PAC}}$	Hydrothermal source strength, Pacific	0.21	0.42	0.035–2.9	$\text{Gmol Fe yr}^{-1}$
$\sigma_{\text{H,IND}}$	Hydrothermal source strength, Indian O.	0.066	0.13	0.011– <del>0.81</del> <u>0.80</u>	$\text{Gmol Fe yr}^{-1}$
$\sigma_{\text{H,SO}}$	Hydrothermal source strength, Southern O.	0.066	0.13	0.011–1.2	$\text{Gmol Fe yr}^{-1}$

(iii) The remaining parameters were optimized using our coupled model. We first assign initial values for all these parameters and then sequentially update these initial values by optimizing subsets of parameters as detailed in Appendix D. Both initial and final optimized parameter values are collected in Table 2. (For the parameters of the iron cycle, Table 2 gives the values of our typical state estimate and the range across a family of state estimates, different members of which have different external iron sources.)

### 3.4 Family of state estimates

Figure 1 shows the quality of the fit to nutrient and phytoplankton data for all our optimized state estimates, which span a wide range of source strengths. For ease of presentation, state estimates are divided at  $\sigma_H = 1 \text{ Gmol yr}^{-1}$  into low and high hydrothermal cases, with  $\sigma_H$  spanning a range from 0.073 to 11.  $\text{Gmol yr}^{-1}$ . For high  $\sigma_H$ , we focused on correspondingly higher aeolian and sedimentary source regimes. Source-parameter space was not explored uniformly because (i) the final step of our optimization adjusted our initial choice of sources, and because (ii) some source choices produced spurious numerical difficulties for the Newton solver.

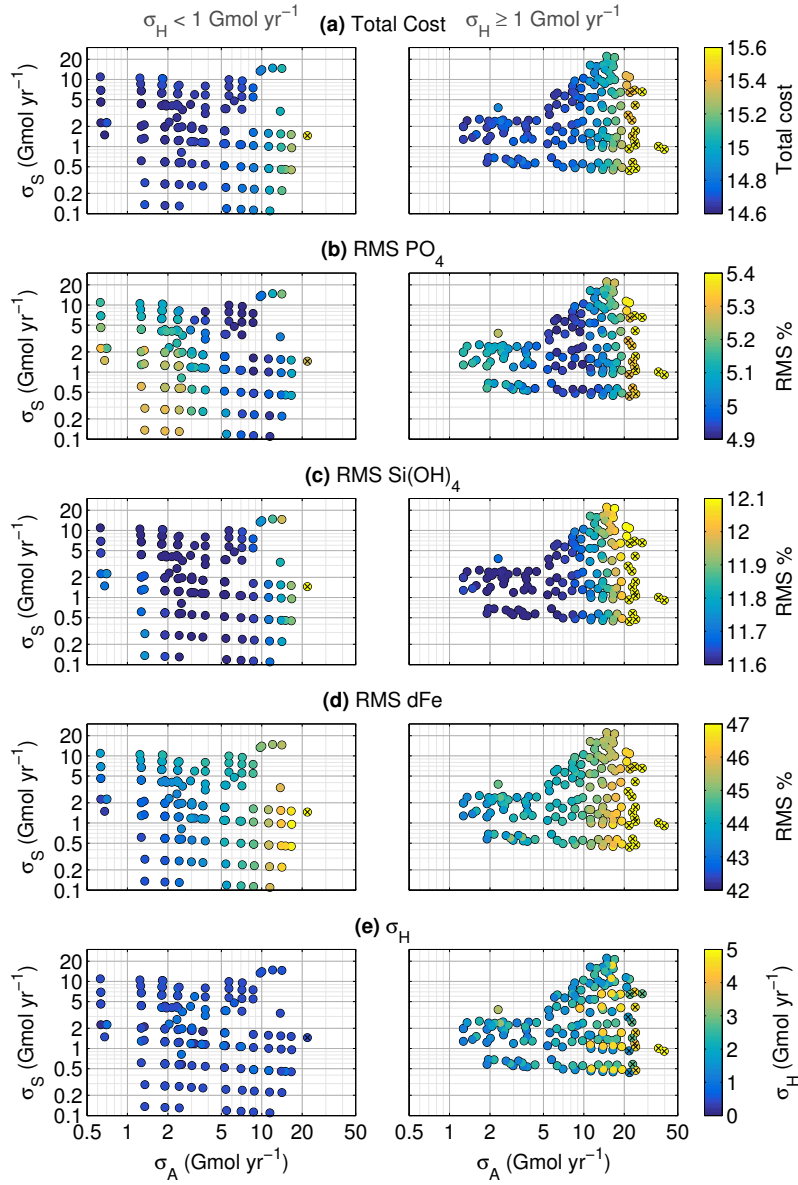
All state estimates fit the macronutrient fields about equally well, but the overall quality of fit as quantified by the square root of the quadratic mismatch (“total cost”, top panels of Figure 1) gets systematically worse with increasing aeolian source strength,  $\sigma_A$ , especially for high hydrothermal sources. This worsening fit for high  $\sigma_A$  is reflected in the mismatch of all three nutrients. We define our family of state estimates as the set of state estimates whose total cost remains within  $\sim 5\%$  of the smallest misfit (total cost less than 15.4), which essentially eliminates state estimates with  $\sigma_A \gtrsim 22 \text{ Gmol yr}^{-1}$  and  $\sigma_H \gtrsim 5 \text{ Gmol yr}^{-1}$  (black crosses in Figure 1). (If we include the “crossed-out” state estimates for plots of subsequent sections that show scatter across the family of state estimates, the visual impact is virtually imperceptible.) While it is clear from Figure 1 that high- $\sigma_A$  states are less likely, we hasten to add that the cost threshold for inclusion in the family is arbitrary as we do not have a formal error covariance to convert the cost into a likelihood.

For a small fraction of our state estimates, the optimization pushed the maximum possible Fe : P ratio  $R_0^{\text{Fe:P}}$  to near-zero values. These cases are unrealistic because zero  $R_0^{\text{Fe:P}}$  means significant P uptake and export are maintained without Fe uptake. We therefore also exclude cases for which the optimized  $R_0^{\text{Fe:P}} < 0.5 \text{ mmol Fe (mol P)}^{-1}$  from our family of state estimates.

Removing these unphysical outliers has negligible visual impact on plots that show the family of state estimates.

In terms of total cost, there is little sensitivity to the strength of the sedimentary source — scavenging can be optimized for a sedimentary source ranging over two orders of magnitude for an overall similar quality of fit. For low  $\sigma_H$ , there are small opposing RMS mismatches for  $\text{PO}_4$  and dFe, with a slightly better  $\text{PO}_4$  fit for higher sedimentary source and a slightly better dFe fit for lower sedimentary source, although the variation in the mismatch is less than 1 % of the global mean concentrations.

While the mismatch for dFe is substantial at  $\sim 45\%$  of the global mean dFe concentration, the smallest dFe mismatch occurs when all three sources are low. The dFe mismatch rapidly increases with  $\sigma_A$ , consistent with the findings of the much simpler model of Frants et al. (2016). The overall cost and the mismatch for each nutrient are insensitive to the strength of the hydrothermal source.



**Figure 1.** Total cost metric and RMS mismatch of the nutrient concentrations as a function of the aeolian, hydrothermal, and sedimentary iron source strengths ( $\sigma_A, \sigma_S, \sigma_H$ ) plotted for all our optimized state estimates. State estimates whose total cost exceed 15.4 are indicated by black crosses and were excluded from our family. Plots on the left show state estimates for which  $\sigma_H < 1 \text{ Gmol yr}^{-1}$ , while for plots on the right  $\sigma_H \geq 1 \text{ Gmol yr}^{-1}$ . (a) Square root of the total cost expressed as a nominal percentage representative of the mean RMS mismatch of the nutrient and phytoplankton concentrations. (b) RMS mismatch of the  $\text{PO}_4$  concentration as a percentage of the global mean  $\text{PO}_4$  concentration. (c) As (b) for  $\text{Si(OH)}_4$ . (d) As (b) for dFe. (e) The value of the hydrothermal source  $\sigma_H$  for each family member.

While Figure 1 shows some variations with the source strengths in the overall quality of the fit, it is clear that the iron sources and scavenging sinks are poorly constrained by the available nutrient and phytoplankton observational data. Given the uncertainties in the sources and the small cost differential between family members, it is not appropriate to single out the state estimate with the numerically lowest cost as the most realistic state estimate. We therefore use the entire family of state estimates below to assess the robustness of our results and to elucidate the systematic variations of the carbon and opal exports with the fractional source of each iron type (aeolian, sedimentary, hydrothermal). The uncertainty in the value of any metric is assigned from its spread across the family of state estimates.

As a typical representative of our family of state estimates, for which we plot patterns and typical results below, we selected the state with  $(\sigma_A, \sigma_S, \sigma_H) = (5.3, 1.7, 0.9) \text{ Gmol Fe yr}^{-1}$ . This state is typical in that it lies at the mode of the distribution of overall RMS misfit values and, for most quantities, tends to lie in the middle of the range across the family.

We emphasize that the variations across the family of state estimates explored here are variations of the *fully optimized* biogeochemical states. These variations cannot be used to infer the system’s response to dFe perturbations for which the other biogeochemical parameters would not change. Such perturbations, which are of great interest in themselves, are beyond the scope of this paper and will be examined in a separate publication.

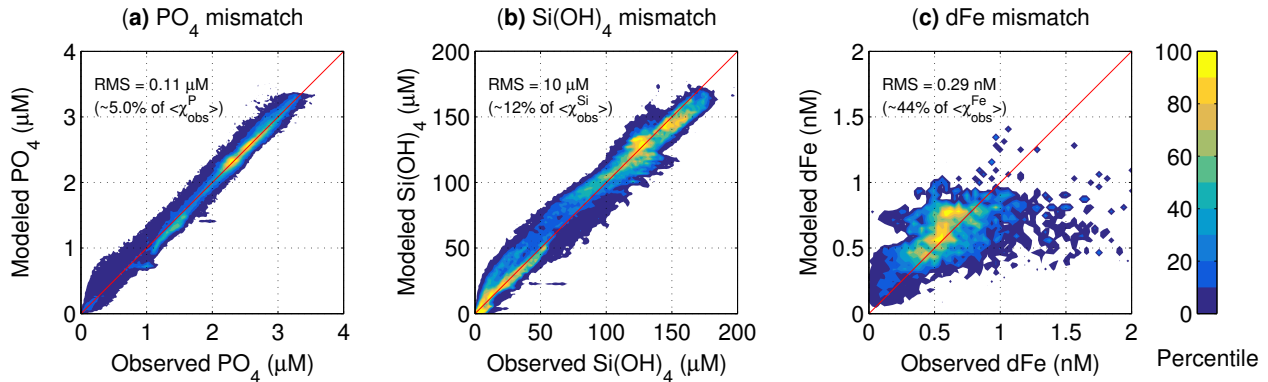
## 4 Fidelity to observations

We now examine in more detail how well our state estimates match the observations against which they were optimized. We focus here on the nutrient fields which contribute the bulk of the cost function. The phytoplankton fields contribute only  $\sim 10\%$  to the cost function and serve primarily to differentiate the uptake of small phytoplankton from that of large phytoplankton — a comparison between estimated and observed phytoplankton concentrations is provided in Appendix E. Where there is little variation across the family, we focus on our typical state estimate. For iron-related quantities that have by construction significant spread across the family, we will focus on the systematic variations of the optimized states with the dFe sources.

### 4.1 Nutrient concentrations

The nutrient concentrations are well constrained for all members of our family of state estimates. We quantify the overall fit of the modelled nutrient concentrations in terms of the joint probability density function (pdf) of the modelled and observed concentrations. This joint pdf may be thought of as the binned scatter plot of the modelled versus observed values for all grid boxes. The binning for a given nutrient was weighted by the corresponding weights of the associated mismatch cost. These joint pdfs are shown in Figure 2 for all three nutrients for our typical state estimate. Both the  $\text{PO}_4$  and  $\text{Si}(\text{OH})_4$  pdfs fall close to the 1 : 1 line, showing high fidelity to observations. For  $\text{PO}_4$  the cost-weighted RMS error is 5.0 % of its global mean of  $2.17 \mu\text{M}$ . In comparison, Primeau et al. (2013) achieved an RMS mismatch of 3 % by jointly optimizing the uptake rate of each grid box with the circulation. Silicic acid has a slightly larger RMS mismatch of 12. % relative to its global mean  $89.1 \mu\text{M}$ . This is similar to the 13 % RMS error reported by Holzer et al. (2014), who used the same circulation but a much simpler model of the silicon cycle.





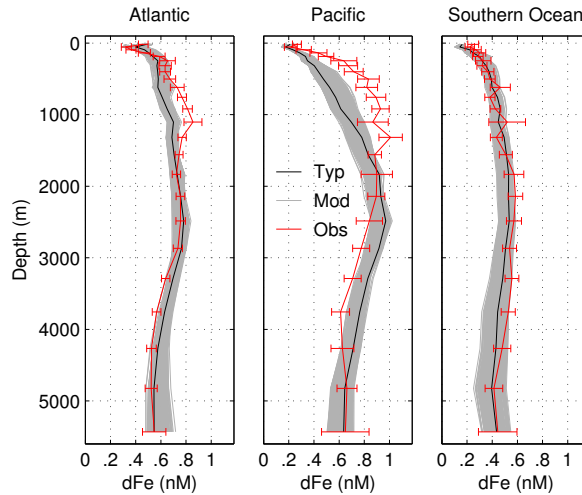
**Figure 2.** Joint distribution of the volume-weighted observed and modelled concentrations of  $\text{PO}_4$  (a),  $\text{Si(OH)}_4$  (b), and dFe (c). The percentiles of the cumulative distribution are defined such that  $x\%$  of the distribution lies outside the  $x$ -percentile contour. Large percentiles thus correspond to high densities. For  $\text{PO}_4$  and  $\text{Si(OH)}_4$ , the WOA13 observations were interpolated to our model grid. The dFe observations were interpolated to our model grid from the data compilation of Tagliabue et al. (2012) and the GEOTRACES data (Mawji et al., 2015).

The global mean dFe concentration is well constrained within the narrow range of 0.56–0.68 nM across the family of state estimates. For iron, the joint probability is by necessity computed using only those grid boxes that contain dFe observations. The scatter from the 1 : 1 line is much larger than for the macronutrients with a substantial RMS mismatch of 0.29 nM, or 44 % of the mean. This mismatch is comparable to that of other models (e.g., Tagliabue et al., 2016). Compared to the simpler model of Frants et al. (2016), the joint pdf shows that our dFe field has a wider, more realistic dynamic range. We note that while Frants et al. (2016) report an RMS mismatch of only 0.19 nM, they also employed different weights for the model-observation mismatch. If we re-compute the RMS mismatch of the optimized dFe field of Frants et al. (2016) using the weights of this work, we also obtain a 0.29 nM mismatch.

The relatively large mismatch for dFe not only quantifies model deficiencies, but to a large degree also reflects the fact that we are comparing snapshot observations against a steady-state coarse-resolution climatological model. The dFe observations have difficult-to-quantify temporal and spatial sampling biases, and dFe being a trace element it is sensitive to episodic events in the aeolian source (e.g., Croot et al., 2004), and possibly to internal episodic events such as submarine volcanism (e.g., Massoth et al., 1995).

## 4.2 dFe Profiles

To quantify the spatial structure of the dFe mismatch, we examine vertical profiles for each basin. For both model and observations, we only use the grid boxes that contain observations and average horizontally over the basins using the  $E_{\text{Fe}}$  cost weights. The resulting profiles are shown in Figure 3. The family of model profiles generally overlaps with the observational uncertainties. The estimates are particularly close to the observations near the surface. In the abyssal oceans, the spread in the family of profiles is larger. The spread across the family is in part a reflection of the weights in our cost function. Most dFe



**Figure 3.** Basin-wide, cost-weighted average profiles of the (red) observed and (grey) modelled dFe concentrations for the Atlantic and Pacific oceans (both north of  $40^{\circ}\text{S}$ ), and the Southern Ocean (south of  $40^{\circ}\text{S}$ ). The profiles of our typical state estimate are highlighted in black. The error bars represent the combined standard error associated with the spatial standard deviation from the basin-mean profile and the observational standard deviation for each grid box. These were added in weighted quadrature using the weights for dFe mismatch from our cost function.

observations are available in the upper ocean, implying a small variance of the mean concentration and hence large weights, while deep observations tend to be sparser with smaller weights (for details on the weights see Appendix C).

Figure 3 also shows systematic biases in the inferred dFe concentrations. Biases are particularly strong in the Pacific where the observations tend to be underpredicted by as much as  $\sim 0.3\text{ nM}$  above  $\sim 1500\text{ m}$  and overpredicted by  $\sim 0.2\text{ nM}$  below  $\sim 2000\text{ m}$  depth. The typical estimated Pacific profiles is too linear in the upper  $1500\text{ m}$ , with vertical gradients that are too weak above  $\sim 300\text{ m}$  and too strong below  $\sim 1000\text{ m}$ . In the Atlantic, a smaller low bias of  $\sim 0.15\text{ nM}$  can be seen between  $\sim 500$  and  $\sim 1300\text{ m}$  depth.

These biases could be due to deficiencies in our model such as, for example, oversimplified ligand parameterization, but one must also keep in mind that there are hard-to-quantify biases in the observations. The observations are too sparse to form a reliable climatology, and it is remarkable that we can fit the available observations as well as we do. The larger biases in the Pacific could well be due to the absence of Pacific transects in the GEOTRACES Intermediate Data Product 2014, which means that mismatches in the Pacific incur a relatively smaller penalty in our cost function.

### 4.3 Phytoplankton distribution

~~Figure E1 shows a model-observation comparison of the phytoplankton concentration (plotted in units using a constant Redfield-ratio of  $C:P=106$ ). Although the distinction between our phytoplankton classes is functional and not determined by~~

size, we compare our Small, Large, and Diatom concentrations with the picophytoplankton (0.5–2), nanophytoplankton (2–20), and microphytoplankton (20–50) of the satellite-based estimates of Kostadinov et al. (2016), consistent with the construction of our  $E_{\text{plk}}$  cost function. Figures 1d, 2c, and 3 are appropriate quantitative comparisons between estimated and observed dFe, given that essentially raw bottle data is compared with a coarse-resolution steady-state model. For completeness, Appendix G also compares the main transects included in the GEOTRACES Intermediate Data Product with our typical state estimate.

The inverse-model estimates capture the broad global patterns of the phytoplankton concentrations reasonably well, although some biases are also evident. The observation-based Diatom and Large concentration has a minimum at  $\sim 60^\circ\text{S}$ , a feature our estimates do not capture. This may be related to seasonality issues, which our approach cannot capture. Our estimates for the Large and Small concentrations have higher concentrations in the subtropical gyres and weaker meridional gradients with lower high-latitude and tropical concentrations than observed. These discrepancies may again be related to the absence of seasonality. Another factor is that the phytoplankton mismatch carries less penalty in our cost than the combined misfit terms of the three nutrient fields.

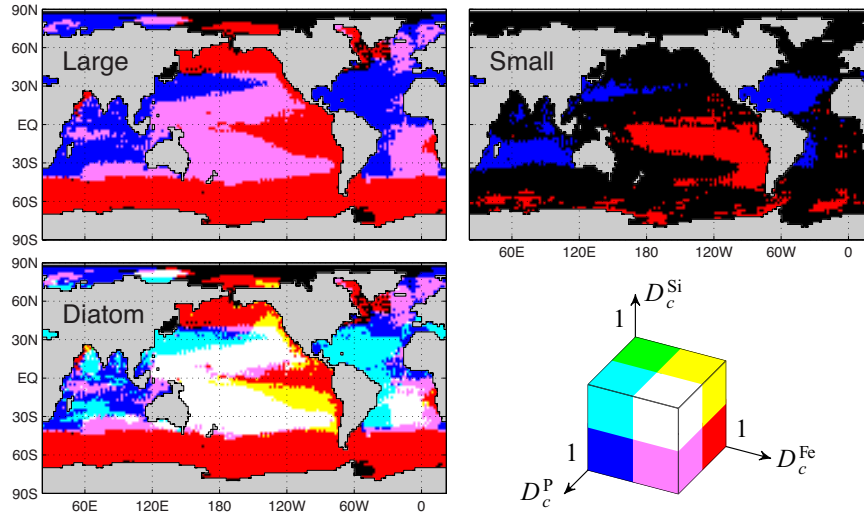
The global mean phytoplankton concentration of each functional class was remarkably stable across all members of our family of state estimates with ranges of  $\overline{p_{\text{sml}}} = (2.63 - 2.87)^{-3}$ ,  $\overline{p_{\text{lrg}}} = (6.16 - 6.53)^{-3}$ , and  $\overline{p_{\text{dia}}} = (2.63 - 2.87)^{-3}$ . This indicates that the satellite data provides a good constraint on the global-scale ecosystem composition.

## 5 Limiting nutrients

For a given phytoplankton functional class, different nutrients are known to limit biological production in different parts of the ocean (e.g., Moore et al., 2001). These geographic limitation patterns are a fundamental fingerprint of upper-ocean ecosystem dynamics. Knowledge of the limitation patterns is important for understanding how the global nutrient cycles operate in the current climate and for assessing possible future changes of the global ocean ecosystem.

Limiting nutrients can be determined observationally (e.g., Moore et al., 2013), and from biogeochemical models (e.g., Moore et al., 2004). Here, we estimate the limitation patterns from our optimized inverse-model state estimates. In our model, the biological uptake of each functional class (6) is limited through  $F_{N,c}$ , the product defined in (11) of three Monod terms, one for each nutrient. We define the deficiency  $D_c^i$  of functional class  $c$  in nutrient  $i$  as the complement of the corresponding Monod factor, i.e., as  $D_c^i \equiv 1 - \frac{\chi_i}{\chi_i + k_c^i}$ . We deem nutrient  $i$  to be “limiting” class  $c$  if  $D_c^i > 0.5$  or, equivalently, if  $\chi_i < k_c^i$ , i.e., if the nutrient concentration falls below its half-saturation value for uptake.

To display the pattern of the nutrient limitations, we could use the fact that we have three nutrients to define an RGB color as  $(D_c^{\text{P}}, D_c^{\text{Si}}, D_c^{\text{Fe}})$ . However, because the resulting colors vary continuously, it is hard to quantify the resulting patterns. We therefore define the limiting RGB color as  $(L_c^{\text{P}}, L_c^{\text{Si}}, L_c^{\text{Fe}})$ , where  $L_c^i = 1$  if  $D_c^i > 0.5$  and  $L_c^i = 0$  otherwise. This partitions the RGB color cube into eight possible colors that define and identify the eight nutrient limitation regimes shown in Figure 4. Specifically, the resulting colour is black (0,0,0) if all nutrients are available in sufficient quantities so that none are deemed limiting, white (1,1,1) if all three nutrients are limiting, red (1,0,0) if only dFe is limiting, green (0,1,0) if only  $\text{Si}(\text{OH})_4$  is limiting, and blue (0,0,1) if only  $\text{PO}_4$  is limiting. The remaining three possibilities correspond to two nutrients being



**Figure 4.** The patterns of limiting nutrients for each phytoplankton functional class. The colour cube at the bottom right shows the eight possible limitation regimes of our inverse model: Red corresponds to dFe limitation, blue to P limitation, and green to Si limitation. Cyan, yellow, and magenta correspond to colimitations of P and Si, dFe and Si, and P and dFe, respectively. White corresponds to colimitation of all three nutrients while black indicates no limitation. (See text for the definitions of the deficiencies  $D_c^P$ ,  $D_c^{Si}$ , and  $D_c^{Fe}$  of the cube axes.)

colimiting: magenta (1,0,1) if dFe and  $PO_4$  are colimiting, cyan (1,1,0) if  $PO_4$  and  $Si(OH)_4$  are colimiting, and yellow (0,1,1) if  $Si(OH)_4$  and dFe are colimiting. Only a few grid boxes in the Arctic are solely limited by silicic acid (green).

Figure 4 shows the limitation patterns of all three phytoplankton classes. The large and diatom classes have similar patterns of iron limitation in the Southern Ocean, eastern tropical Pacific, and North Pacific. For both classes, the Indian Ocean and North Atlantic are largely  $PO_4$  limited. The subtropical gyres of the Indian Ocean and North Atlantic are  $PO_4$  limited for the large class, and  $PO_4$ – $Si(OH)_4$  colimited for diatoms. The differences between the large and diatom classes come from the  $Si(OH)_4$  dependency of diatoms. Because the large class requires zero silicic acid, its limitation map show no areas where all three nutrients are limiting (white). The subtropical gyres of the Pacific and South Atlantic are dFe and  $PO_4$  colimited for the large class, while for diatoms the centre of these gyres are limited in all three nutrients. For diatoms, the eastern margins of the Pacific subtropical gyres show Si–Fe colimitation (yellow). The completely nutrient replete regions of the Arctic and Weddell Sea reflect the low biological utilization there driven in our model by light limitation through the prescribed PAR irradiance field.

The small phytoplankton class shows a much simpler limitation pattern. Limitation occurs primarily in the subtropical oceans with small patches of iron limitation also in the Southern Ocean and tropical Pacific. Iron limitation dominates the subtropical South Pacific, while  $PO_4$  limitation occurs primarily in the subtropical gyres of the south Indian Ocean and North Atlantic. The rest of the ocean is largely nutrient replete for the small functional class.

The broad features of the limitation pattern of Figure 4 are robust across all members of our family of state estimates, with areas of each type of limitation generally varying by  $\pm 5\%$  or less across the family of state estimates.

The general features seen in Figure 4 broadly agree with the observational data (in situ and bottle nutrient addition experiments) reported by Moore et al. (2013). Like our estimates, the observations show Fe limitation in the Southern Ocean, subpolar North Pacific, and eastern tropical Pacific. The observations also indicate Fe limitation in the North Atlantic, which for our state estimates is also present in small patches in the western subpolar North Atlantic and becomes slightly more pronounced for the family members with higher total iron source. Moore et al. (2013) report Si limitation in the Pacific sector of the Southern Ocean at its northern boundary, where the silicic acid concentration sharply decreases. This is consistent with our yellow region of joint Si and Fe limitation along the eastern edge of the Pacific subtropical gyres. Consistent with our estimates, the observations show  $\text{PO}_4$  limitation in the North Atlantic subtropical gyre and in the equatorial Atlantic.

Our limitation patterns can also be compared to those calculated for summer conditions in the BEC model of Moore et al. (2004). However, it must be kept in mind that (i) the BEC model has a different circulation and a different representation of the ocean's biogeochemical cycles (e.g., explicitly representing the nitrogen cycle and diazotrophs) and that (ii) Moore et al. (2004) define limitation in terms of the minimum Monod factor, while we use a threshold of  $1/2$  for the Monod factors and jointly consider three Monod terms to define the type of limitation. For diatoms, the Fe limitation pattern reported by Moore et al. (2004) is similar to ours including bands of Si limitation surrounding the tongue of Fe limitation in the tropical eastern Pacific. For non-diatom phytoplankton, there are also broad similarities, such as iron limitation in the eastern tropical Pacific, subpolar North Pacific, and Southern Ocean. In the BEC model, most of the Atlantic is phosphate or nitrate limited. While we do not model nitrate, nitrate limitation in our estimates would be reflected as phosphate limitation, and our limitation patterns show most of the Atlantic deficient in phosphate. The BEC model's small phytoplankton class shows nitrogen limitation surrounding the tropical tongue of iron limitation in the Pacific, while with our definitions there is very little  $\text{PO}_4$  limitation in the Pacific for the small class, which is iron limited or nutrient replete in most of the Pacific. Finally, we note that the annual-mean nature of our estimates is another possible reason for differences.

## 6 Export production

A key metric of the nutrient cycles is their export production, which determines the strength of the biological pump (e.g., Pasquier and Holzer, 2016). Export production is not directly available from satellite measurements, but observationally constrained estimates are easily calculated from our inverse model. The phosphorus export flux,  $\Phi^P$ , is simply the flux of organic phosphorus into the aphotic zone that is remineralized there, which we compute using the operators  $\mathcal{S}_c^P$  (sinking and remineralization) as

$$\Phi^P = \sum_{c=1}^3 \int_a dz \mathcal{S}_c^P U_c . \quad (27)$$

For plotting, we convert  $\Phi^P$  to a carbon export flux using a constant C : P ratio of 106 : 1. We [acknowledge that this simple unit conversion underestimates the true C export because we do not explicitly represent DOC. Semilabile DOC has a longer typical](#)

lifetime than DOP, resulting in effect in a larger C : P ratio for dissolved organic matter (DOM) than for particulate organic matter. Using the data-assimilated phosphorus cycle of Primeau et al. (2013), which explicitly carries both  $\text{PO}_4$  and DOP, and applying a C : P ratio of 225 : 1 for DOP (as determined by the DOM OPT simulation of Letscher et al. (2015) for semilabile DOM), we estimate that the simple unit conversion of the POP export (27) underestimates the carbon export by  $\sim 12\%$ .

5 We similarly calculate the opal export as

$$\Phi^{\text{Si}} = \int_a dz \mathcal{S}_{\text{dia}}^{\text{P}} R_{\text{dia}}^{\text{Si:P}} U_{\text{dia}} , \quad (28)$$

and the iron export associated with the remineralization of organic matter as

$$\Phi^{\text{Fe}} = \sum_{c=1}^3 \int_a dz \mathcal{S}_c^{\text{P}} R^{\text{Fe:P}} U_c , \quad (29)$$

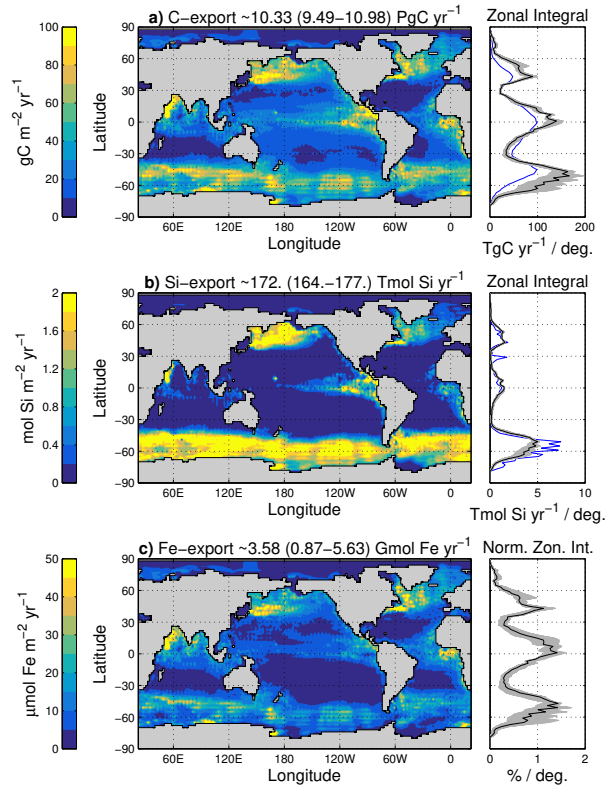
where the vertical integrals are over the model aphotic zone (bottom to 73.4 m depth).

10 Figure 5a shows a map of the ~~carbon export flux~~ phosphorus export flux (converted to carbon units), together with its zonal integral for each member of our family of state estimates. The spatial pattern shows some differences with the estimate of Primeau et al. (2013) (blue curve in Figure 5a). Our estimate of the carbon export has 1.5–2 times larger tropical and high latitude peaks, but is closer to the satellite-derived estimates of Dunne et al. (2007). Our estimate also has sharper meridional gradients, which ~~is arguably more realistic considering the sharp gradients can also be seen~~ in satellite-derived estimates of  
15 production (e.g., Frants et al., 2016). Our globally integrated ~~carbon export of~~ phosphorus export of 7.5–8.6  $\text{Tmol P yr}^{-1}$  (9.5–11. Pg C yr<sup>-1</sup>) is also larger than the  $7.4 \pm 2.5 \text{ Pg C yr}^{-1}$  estimate of Primeau et al. (2013).

The differences with the estimate of Primeau et al. (2013) are likely due to very different uptake parameterizations: Primeau et al. (2013) consider the phosphorus cycle in isolation and optimize a single spatially varying uptake timescale for each grid box, while we explicitly represent three phytoplankton functional classes with different, optimized globally uniform uptake  
20 timescales,  $\tau_c$ . We note that if we use the same growth timescale for each phytoplankton class, our model's ~~carbon~~ phosphorus export remains close to that of Primeau et al. (2013).

Our estimates of ~~the carbon export~~ export production compare well with the satellite-based estimates of  $9.7\text{--}12. \text{ Pg C yr}^{-1}$  by Gnanadesikan et al. (2004). Our estimates also lie within the wide range of  $9\text{--}28. \text{ Pg C yr}^{-1}$  of the Ocean-Carbon Cycle Model Intercomparison Project 2 (OCMIP-2, Najjar et al., 2007), and compare well the OCMIP-2 mean particle export of  
25  $13 \pm 3 \text{ Pg C yr}^{-1}$ . (Because our model does not carry ~~DOP, its particulate carbon export is given by its total carbon export~~ → DOM, we compare our phosphorus export in carbon units to both the export production and the particle export reported by OCMIP-2.)

Our estimates of phosphorus export in the subtropical gyres compare well to the POP exports of Letscher et al. (2016) in spite of the fact that we do not explicitly represent DOP. Using the same masks to define the subtropical gyres, we estimate  
30 a global mean subtropical phosphorus export of  $10. \pm 1. \text{ mmol P m}^{-2} \text{ yr}^{-1}$  (mean and standard deviation across our family of estimates), while the estimate of Letscher et al. (2016) is  $10. \pm 2. \text{ mmol P m}^{-2} \text{ yr}^{-1}$ . This underscores the fact that our inverse model implicitly captures the effects of DOP lateral transport and utilization, which Letscher et al. (2016) estimate to contribute  $29 \pm 9\%$  of the subtropical phosphorus export.



**Figure 5.** Local export production for each nutrient (maps on the left) and its zonal integral (curves on the right). Maps are shown for our typical state estimate, while we plot the zonal integral of each family member (scaled for dFe) in grey and the typical state estimate in black. (a) Phosphorus export, expressed in carbon units using C : P = 106 : 1. The blue zonal integral is the export production estimate of Primeau et al. (2013). (b) Opal export, where the blue zonal integral is the estimate of Holzer et al. (2014). (c) Iron export, with its zonal integrals expressed as a percentage of the global iron export.

Figure 5b shows a map of the opal export, together with its zonal integral. As expected, opal export is most pronounced at high latitudes, particularly in the Southern Ocean. In spite of our relatively complex formulation of silicic acid utilization in terms of colimitations, the spatial pattern of the opal export and its global total of 164–177 Tmol Si yr<sup>-1</sup> compare well with the estimates by Holzer et al. (2014) (171 ± 31 Tmol Si yr<sup>-1</sup>). Other estimates of the global opal export range from 69 to 185 Tmol Si yr<sup>-1</sup> (e.g., Moore et al., 2004; Sarmiento et al., 2007; Heinze et al., 2003).

There is very little spread in the carbon and opal export production across our family of state estimates as can be seen by the tightly clustered zonal integrals plotted in grey in Figures 5a,b. This shows that the carbon and opal exports are well constrained despite the wide range of iron inputs. Note, however, that export productions are sensitive to the parameters that set the biological uptake rate, which were optimized before generating the family of optimal state estimates with different iron



Figure 5c shows a map of the iron export associated with organic matter, but not including the iron export carried by scavenging particles. The phosphorus and iron exports have broadly similar patterns, with differences that reflect variations in the local Fe : P uptake ratio. In the iron-deficient Southern Ocean, the Fe : P ratio is smaller than its global mean, which results in Southern Ocean iron export that is less efficient than that of phosphorus (for iron, the peak Southern Ocean export relative to the tropical peak is lower than for phosphorus). As expected from the widely varying iron source strengths across our family of state estimates, the globally integrated iron export covers a wide range of  $9.6 \times 10^{-4}$ – $0.87$ – $5.6$  Gmol Fe yr<sup>-1</sup>. However, the geographic pattern of the iron export is robust across the family: the zonally integrated iron exports normalized by their global integrals collapse onto a well-defined cluster of curves. The spread in the thus normalized iron export is similar to the spread in the (un-normalized) ~~carbon~~-phosphorus export, but slightly larger due to variations in the Fe : P ratio.

All export fields of Figure 5 show near-zero export in the Weddell Sea, in contrast to what restoring-type models tend to show. For example, the opal export estimated by Holzer et al. (2014) has a local maximum in the Weddell Sea. The Weddell Sea minimum here is due to near-zero satellite measurements of photosynthetically active radiation in this region. This may well be an artifact of the satellite data, for which the irradiance in the Weddell Sea varies substantially depending on which years are averaged.

~~Figure F1 shows the carbon export partitioned according to each functional class. The bulk of the carbon export is accomplished by the Large and Diatom phytoplankton classes (~53 and 41, respectively), while the remainder (~6) is exported by the Small class. In the tropics, the Large class dominates the export, while in the high latitudes, and particularly in the Southern Ocean, the Diatom class dominates. This partition between the three phytoplankton classes is the result of the adjustments of the class-specific growth timescales,  $\tau_C$ , and of the parameters of the ratio to bring the fractional uptake by each class into alignment with the satellite-based estimates of Uitz et al. (2010). For our typical state estimate, these uptake fractions are 38, 42, and 30 for the Diatom, Large, and Small classes, respectively. This compares to 32, 44, and 24 for micro-, nano-, and picophytoplankton, respectively, in the satellite-based estimates. (We find that if we use only a single optimized timescale for all three classes, the Small phytoplankton class completely dominates the phosphorus export, underlining the need for class-specific growth timescales.)~~

~~Export by the Diatom class has the sharpest meridional gradients. This is due to the diatoms' larger half-saturation rates, larger maximum phytoplankton concentrations, and their limitation. Conversely, the carbon export by the Small class has the least spatial variation. This is consistent with satellite estimates of plankton concentration that show the least spatial variability for picophytoplankton (Kostadinov et al., 2016): Ubiquitous small phytoplankton are better adapted to nutrient-limited conditions, but do not benefit from nutrient abundance as much as larger phytoplankton.~~

## 30 7 Iron cycle

Here we document some of the key features of the iron cycle as constrained by our inverse model. Certain features such as the dFe concentration field are robustly constrained by the observations regardless of iron source strengths, while other features such as, e.g., the relative importance of hydrothermal iron, vary systematically with the source strengths.



## 7.1 Iron sources and sinks

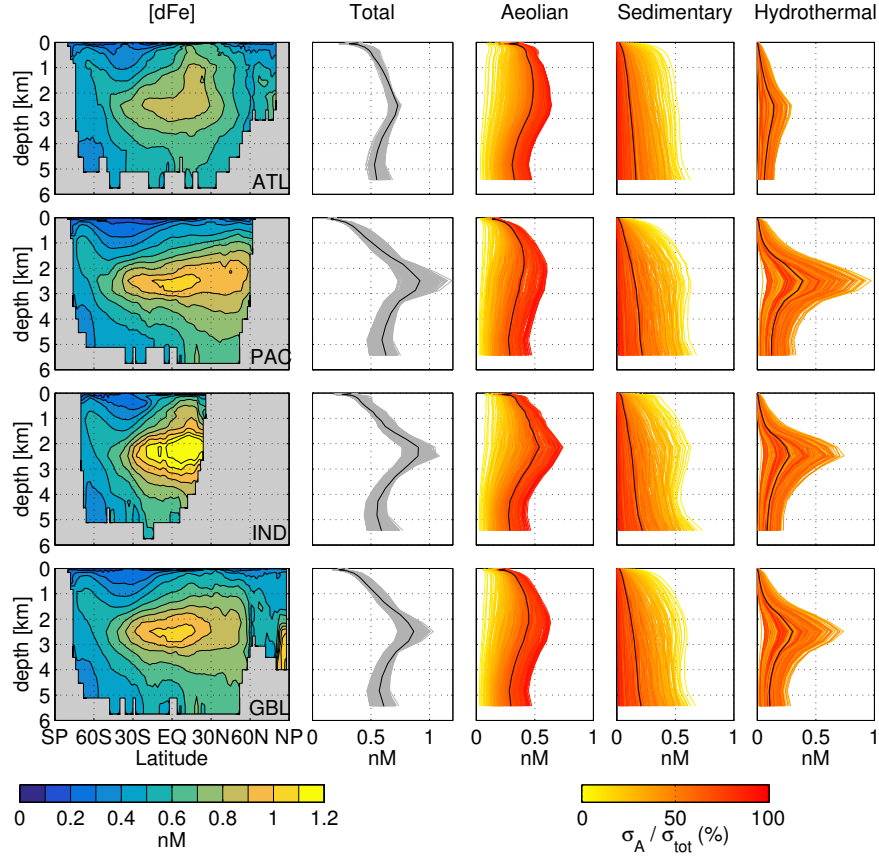
The pattern of the aeolian source is identical for all family members because we only vary its global source strength,  $\sigma_A$ . The sediment source is keyed to export production, which is well constrained across the family of state estimates. Therefore, the sedimentary iron source patterns are very similar across all state estimates, with only the global strength  $\sigma_S$  of the pattern varying among state estimates. The initial hydrothermal pattern is set by the OCMIP  $^3\text{He}$  source (Dutay et al., 2004), but for total hydrothermal sources larger than  $\sim 0.5 \text{ Gmol yr}^{-1}$ , the optimized contributions from each basin changed substantially. Across our family of state estimates the mean and standard deviations of the percentage contributions from each basin to the total hydrothermal source are  $(15 \pm 9) \%$  for the Atlantic,  $(52 \pm 6) \%$  for the Pacific,  $(16 \pm 2) \%$  for the Indian Ocean, and  $(17 \pm 3) \%$  for the Southern Ocean (south of  $40^\circ\text{S}$ ). For reference, the vertically integrated iron sources of our typical state estimate are plotted in Appendix H.

Because of the small variations in the source patterns, the vertically integrated total sinks of  $d\text{Fe}$ ,  $\int dz(1 - \mathcal{S}_s^P)J_{\text{POP}}$ ,  $\int dz(1 - \mathcal{S}_s^{\text{Si}})J_{\text{bSi}}$ , and  $\int dz J_{\text{dst}}$ , have patterns that also vary little across the family of state estimates (see Appendix H for plots of the vertically integrated sinks of our typical state estimate). Note that these sinks balance the total source exactly because we are in steady state. ~~The dominant iron sink is due to scavenging by organic particles (POP). The sink due to scavenging by opal accounts for roughly~~ For our typical state estimate, scavenging by POP and opal each account for about half of the total iron sink. The patterns of POP and opal scavenging are determined by the phosphorus and opal exports and by the concentration of free iron. Consequently, the POP scavenging sink is strongest in the tropics, and the opal scavenging sink is strongest in the Southern Ocean. The sink due to mineral dust scavenging reflects the pattern of the aeolian dust input modulated by the free iron concentration. However, for our family of state estimates, the sink due to dust scavenging is essentially negligible, being about three orders of magnitude smaller than the POP and opal scavenging.

We note that the partition of scavenging among the different particle types cannot be inferred robustly from our inverse model. This is because the nutrient and phytoplankton data used do not provide separate constraints on the scavenging by each particle type, only on the total amount of scavenging. Moreover, scavenging by one particle type can be compensated by another type because of overlap in their spatial patterns. However, the partition among particles types does vary systematically across our family of estimates. Scavenging by dust is negligible for all state estimates, while the fraction scavenged by POP ranges from  $\sim 10 \%$  for the lowest iron sources to saturation near  $100 \%$  for the highest iron sources. (The complementary fraction is due to opal scavenging.)

## 7.2 dFe Concentration and source attribution

Figure 6 shows our typical state estimate’s zonally averaged dFe concentration for each basin and for the global ocean. For each zonal average, we also show the corresponding profile of horizontally averaged dFe for each member of our family of state estimates. The profiles are tightly clustered showing that the large-scale features of the dFe field are well constrained despite the large variations of the iron sources. The inverse model fits the observed dFe field for widely different sources by



**Figure 6.** Estimates of the dFe concentration in each basin (ATL, PAC, IND) and globally (GBL). The zonal averages in latitude-depth space on the left show the total dFe field of our typical state estimate. The corresponding horizontally averaged profiles of total dFe are shown in grey for each family member and in black for the typical state estimate. The three columns of plots on the right show the source-partitioned dFe profiles, i.e., the profiles of the aeolian, sedimentary, and hydrothermal component of dFe for each family member. The individual source-partitioned profiles are colour coded according to the percent contribution of the aeolian iron source to the total iron source, with our typical state estimate in black.

adjusting the corresponding scavenging. While these adjustments keep the total dFe field close to the observations, the relative contributions from the aeolian, sediment, and hydrothermal sources are unconstrained and can vary widely.

We calculate dFe concentrations due to each source following Holzer et al. (2016) by replacing the dFe concentration tracer equation (3) by an equivalent linear diagnostic system that has the same solution. This linear system, corresponding to a given solution of the full nonlinear system, is obtained by replacing the iron uptake and scavenging by linear operators. Specifically, the dFe uptake  $R^{\text{Fe:P}}U_c$  is replaced with  $L_{U,c}\chi_{\text{Fe}}$  and the scavenging rate  $J_j$  with  $L_{J,j}\chi_{\text{Fe}}$ , where the linear operators, organized into matrix form, are simply specified from the uptake and scavenging rates of the nonlinear solution as

$\mathbf{L}_{U,c} = \text{diag}(R^{\text{Fe:P}} U_c / \chi_{\text{Fe}})$  and  $\mathbf{L}_{J,j} \chi_{\text{Fe}} \equiv \text{diag}(J_j / \chi_{\text{Fe}})$ . The dFe concentration  $\chi_{\text{Fe}}^k$  due to source  $s_k$  (with  $k \in \{\text{A}, \text{S}, \text{H}\}$ ) is then computed by replacing the total source in the linear equivalent system by the source  $s_k$  of the component of interest and computing the corresponding concentration.

Figure 6 also shows the profiles of the individual source components of dFe, colour coded according to the fractional strength of the aeolian source. In contrast to the profiles of the total dFe, these individual source components vary widely across the family of state estimates, but in such a way that the total concentration  $\chi_{\text{Fe}} = \chi_{\text{Fe}}^{\text{A}} + \chi_{\text{Fe}}^{\text{S}} + \chi_{\text{Fe}}^{\text{H}}$  is tightly constrained. For example, for low aeolian sources (yellow profiles in Figure 6), the concentration of aeolian iron  $\chi_{\text{Fe}}^{\text{A}}$  is relatively low, but largely compensated by a relatively larger sediment contribution  $\chi_{\text{Fe}}^{\text{S}}$ . The concentrations of hydrothermal iron vary less systematically with the aeolian source, but all family members have very similarly shaped hydrothermal dFe profiles. However, the amplitudes of the hydrothermal dFe profiles can be seen to vary by roughly an order of magnitude across the majority of state estimates, effectively fine tuning the total dFe concentration to be as close to the observations as possible.

### 7.3 Iron-type attributed export

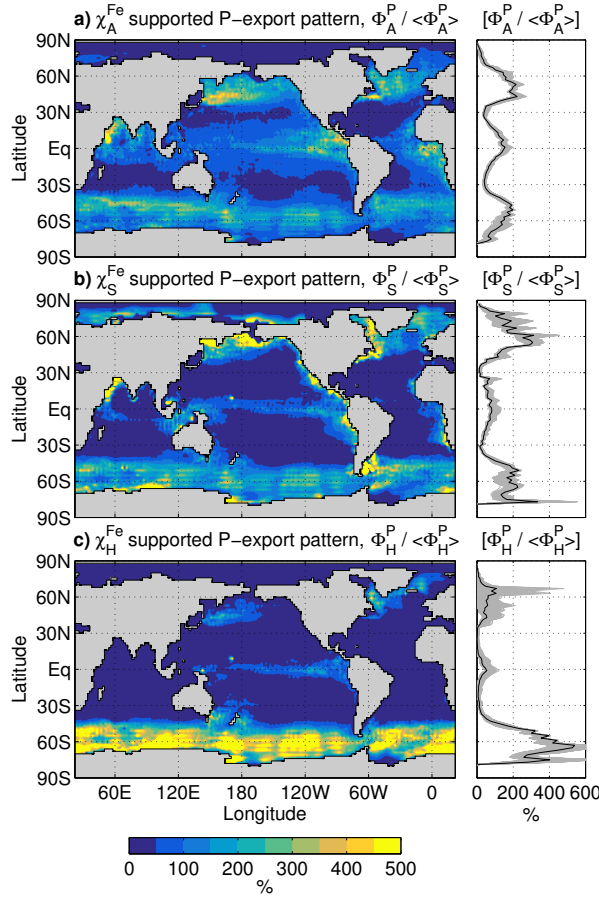
#### 7.3.1 Phosphorus export

We quantify the contribution of each iron type to the export production as follows. In our formulation, non-zero dFe is necessary for non-zero phosphate uptake  $U_c$ . The uptake  $U_c(\mathbf{r})$  inferred at point  $\mathbf{r}$  is supported by the dFe concentration at  $\mathbf{r}$ , which is a mixture of aeolian, sedimentary, and hydrothermal dFe. Thus, the uptake supported by iron type  $k$  is given by  $(\chi_{\text{Fe}}^k(\mathbf{r}) / \chi_{\text{Fe}}(\mathbf{r})) U_c(\mathbf{r})$ , that is, the local uptake supported by dFe of type  $k$  must be in proportion to the concentration fraction  $\chi_{\text{Fe}}^k / \chi_{\text{Fe}}$ . (Note that  $\sum_k \chi_{\text{Fe}}^k / \chi_{\text{Fe}} = 1$ .) For a given nonlinear solution, the phosphorus export production supported by iron type  $k$ , denoted by  $\Phi_k^{\text{P}}$  is therefore calculated by replacing the uptake  $U_c$  in (27) with  $(\chi_{\text{Fe}}^k / \chi_{\text{Fe}}) U_c(\mathbf{r})$ .

While the total export production is well constrained regardless of the chosen iron source strengths, the production supported by a given iron type varies substantially with the magnitude of the corresponding source. (Summing over the three iron types yields the well-constrained total.) However, regardless of the source amplitudes, the patterns of the export supported by each iron type is similar across the entire family of state estimates.

Figure 7 shows  $\widehat{\Phi}_k^{\text{P}} \equiv \Phi_k^{\text{P}} / \langle \Phi_k^{\text{P}} \rangle$ , which is the export flux supported by iron type  $k$  normalized by the global mean export  $\langle \Phi_k^{\text{P}} \rangle$ . The  $\widehat{\Phi}_k^{\text{P}}$  patterns are plotted for our typical state estimate, together with zonal averages of  $\widehat{\Phi}_k^{\text{P}}$  for all family members. The patterns can be seen to differ little among family members. Even the pattern of the export supported by hydrothermal dFe, which varies most across the family, has the broadly similar features of peak export in the Southern Ocean, with secondary peaks in the tropics and in the Northern Hemisphere subpolar oceans, for all family members.

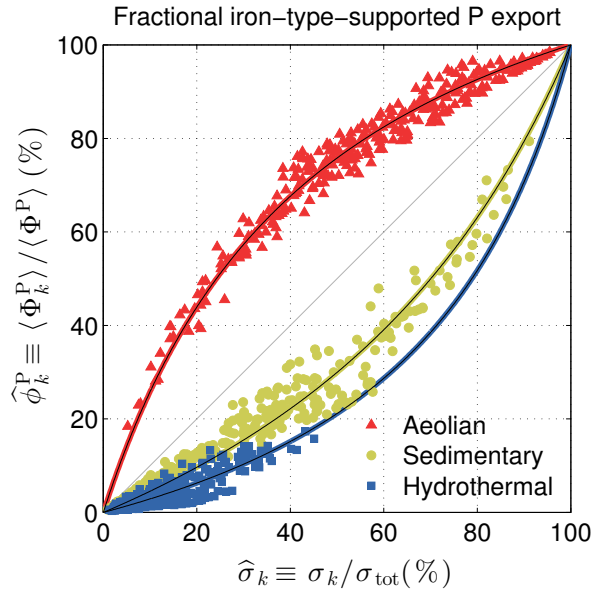
Figure 7a shows that aeolian iron supports export primarily in the tropics and in the subpolar oceans. The tropics receive direct input of fresh aeolian iron, while the subpolar oceans receive upwelling regenerated iron (Holzer et al., 2016). The aeolian-iron-supported export pattern is very similar to the pattern of the total export flux shown in Figure 5a. (Note that here we plot zonal averages, while Figure 5 shows zonal integrals.) For sedimentary dFe to support export it must be transported from the ocean bottom into the euphotic zone. Consequently, the pattern of the export supported by sedimentary dFe (Fig.



**Figure 7.** Phosphorus export supported by each iron type [aeolian (a), sedimentary (b), hydrothermal (c)] normalized by its global mean. The maps show our typical state estimate, while zonal averages of the normalized phosphorus export are shown for each family member in grey, with the typical state estimate in black.

7b) is dominated by regions of upwelling in the tropical and subpolar oceans and by regions of shallow depth (both resolved and subgrid) where there is high organic matter flux, such as the seas around Indonesia. The meridional gradients of the sedimentary-iron-supported export are much sharper than for the aeolian-iron-supported export, presumably because of the lack of large-scale direct surface input for sedimentary dFe. The pattern of export supported by hydrothermal dFe (Fig. 7c) is dominated by the Southern Ocean, where most of the density classes into which hydrothermal fluid is injected outcrop. Secondary regions of hydrothermal-iron-supported export are associated with upwelling in the tropics and in the subpolar oceans of the Northern Hemisphere.

Underscoring the similar source distribution of hydrothermal dFe and mantle  $^3\text{He}$ , the pattern of hydrothermal-iron-supported export production is similar to the pattern with which mantle  $^3\text{He}$  outgases to the atmosphere (e.g., Holzer et al., 2017). We



**Figure 8.** Percent global phosphorus export (equivalently carbon export) supported by each iron type (aeolian, sedimentary, hydrothermal) versus the corresponding fractional source of that iron type. The superposed lines are least-squares fits to theoretical relationships with fixed relative export-support efficiencies. (See text for details.)

do not expect an exact correspondence in the patterns because hydrothermal dFe is subject to scavenging losses, while  $^3\text{He}$  is not, and our ratio of hydrothermal dFe source to mantle  $^3\text{He}$  source is different for different basins. (The ranges of the ratio of the optimized hydrothermal iron source to the mantle  $^3\text{He}$  source across the family of inverse-model estimates were 0.00087–3.3, 0.097–0.098–8.2, 0.208–1.0, 0.2–15., and 0.025–2.8 in units of  $\text{Mmol Fe} (\text{mol } ^3\text{He})^{-1}$ , for the Atlantic, Pacific, Indian, and

5 Southern Ocean basins, respectively.)

While the total phosphorus export is well constrained and varies little across our family of state estimates, the magnitude of the iron-type-supported export production varies systematically with the relative source strength of each iron type. To quantify these systematic variations, Figure 8 plots the fraction  $\langle \Phi_k^P \rangle / \langle \Phi^P \rangle$  of the globally averaged iron-type- $k$ -supported export to the total global export as a function of the corresponding fractional global iron source strengths

10  $\hat{\sigma}_k / \hat{\sigma}_{\text{tot}} \hat{\sigma}_k \equiv \sigma_k / \sigma_{\text{tot}}$ , where  $\sigma_{\text{tot}} \equiv \sum_k \sigma_k$ . Note that if a given source strength  $\sigma_k$  constitutes 100 % of the total, then it must support 100 % of the export and that if  $\sigma_k = 0$  then it supports 0 % of the export. i.e., the relationship between  $\hat{\phi}_k^P$  and  $\hat{\sigma}_k$  must pass through the points (0,0) % and (100,100) %.

Figure 8 shows a number of interesting systematic features of our family of state estimates. First, aeolian dFe supports ~20–100 % of the global export, with the low end of the range corresponding to an aeolian source of only ~5 % of the total  
15 source. (We did not explore lower fractional aeolian sources.) Sedimentary iron supports ~0–80 % of the global export, with

the high end of the range corresponding to a sediment source as high as  $\sim 90\%$  of the total source. Hydrothermal iron supports the least export ranging from  $\sim 0\text{--}18\%$  for fractional hydrothermal sources as large as  $\sim 45\%$  of the total source.

The key point of Figure 8 is that aeolian iron can be considered to be the most **important-efficient** type of iron for supporting export production: For a given fraction of the total source, the fraction of export supported by aeolian iron is larger (i.e., the aeolian points all lie above the 1 : 1 line ~~).~~ ~~For the fractional aeolian sources between  $\sim 20$  and  $\sim 50$  by as much as  $30\%$ , the corresponding export fraction is  $\sim 30\%$ .~~ In other words, per source-injected dFe molecule, aeolian iron supports more export than the other iron types. Sedimentary and hydrothermal dFe make fractional contributions to export that are less than their fractional sources (the sedimentary and hydrothermal points lie below the 1 : 1 line by as much as  $\sim 30\text{--}20\%$  ~~greater than the fractional source. As the fractional aeolian source approaches  $100\%$  from below or  $0\%$  from above, it comes close to supporting an equal fraction of the export because the relation between fractional production and fractional source must pass through the points  $(0,0)\%$  and  $(100,100)\%$ .~~).

The fact that the scatter plots of  $\hat{\phi}_k$  versus  $\hat{\sigma}_k$  in Figure 8 are reasonably compact suggests a simple underlying relationship. If we define a given source-type's efficiency in supporting phosphorus export by  $\epsilon_k^P \equiv \hat{\phi}_k^P / \hat{\sigma}_k$ , we see from Figure 8 that  $\epsilon_k^P$  varies with  $\hat{\sigma}_k$ . However, one might expect the efficiency of source type  $k$  *relative* to the efficiency of the other sources to be more constant: This relative export-support efficiency is controlled by dFe transport pathways and by scavenging, in turn controlled by the well constrained organic-matter export. To investigate this possibility, we note that the export-support efficiency of the sources other than  $s_k$  is given by  $\bar{\epsilon}_k^P \equiv (1 - \hat{\phi}_k^P) / (1 - \hat{\sigma}_k)$  so that the export-support efficiency of source  $s_k$  relative to the other sources is  $e_k^P = \epsilon_k^P / \bar{\epsilon}_k^P$ . If  $e_k^P$  is constant, then it follows algebraically that  $\hat{\phi}_k^P = e_k^P \hat{\sigma}_k / [1 + (e_k^P - 1)\hat{\sigma}_k]$ , i.e., the relationship between  $\hat{\phi}_k^P$  and  $\hat{\sigma}_k$  is determined by the single parameter  $e_k^P$ . Note that  $e_k^P$  is the slope of this theoretical  $\hat{\phi}_k^P$  versus  $\hat{\sigma}_k$  relationship at the origin. Nonlinear least-squares fits of this functional form to the  $(\hat{\phi}_k^P, \hat{\sigma}_k)$  pairs of our family of states approximate the scatter plots well (lines in Figure 8) and result in relative export-support efficiencies of  $e_A^P = 3.1 \pm 0.8$ ,  $e_S^P = 0.4 \pm 0.2$ , and  $e_H^P = 0.3 \pm 0.1$ , where the uncertainty for each source type is the standard deviation of the corresponding residuals. Thus, per source-injected molecule, aeolian iron supports  $3.1 \pm 0.8$  times more phosphorus export than the other sources, while sedimentary and hydrothermal iron support  $1/e_S = 2.3 \pm 0.6$  and  $1/e_H = 3.7 \pm 2.3$  times less export than the other sources.

The ability of aeolian iron to make disproportionately large contributions to supporting ~~phosphate and carbon export~~ **organic-matter export**, quantified here by a relative export-support efficiency greater than unity, is presumably due to fresh aeolian iron being directly injected into the euphotic zone. ~~This view is~~ The less-than-unity relative export-support efficiencies of sedimentary and hydrothermal iron reflect the fact that iron from interior sources is generally subject to scavenging before it even reaches the euphotic zone. Because most large sedimentary sources are relatively shallow, a typical sedimentary dFe molecule will undergo less scavenging en route to the euphotic zone than a typical hydrothermal dFe molecule, which is quantified here by the lower relative export-support efficiency of hydrothermal iron. These arguments are supported by the fact that if we calculate the fractional export  $\hat{\phi}_k^P$  only for the Southern Ocean, where the aeolian source is small and most aeolian iron is supplied as upwelled regenerated iron (Holzer et al., 2016), the deviation of the fractional aeolian iron-supported export from the then all  $(\hat{\phi}_k^P, \hat{\sigma}_k)$  pairs lie closer to the 1 : 1 line is roughly halved.

Sedimentary and hydrothermal make fractional contributions to the export that are less than their fractional sources (the sedimentary and hydrothermal points lie below the 1 : 1 line by as much as  $\sim 20\%$ ). Over the range of the fractional hydrothermal source, the hydrothermal-iron-supported export fraction is roughly two thirds the corresponding sedimentary-iron-supported export fraction. The lower efficiency of hydrothermal iron presumably reflects the fact that most large sedimentary sources are relatively shallow and can thus supply iron more readily to the euphotic zone compared to the deep hydrothermal sources. This is consistent with the fact that hydrothermal and sedimentary iron have roughly equal efficiency for the fractional exports computed for the Southern Ocean only (not shown). (In terms of relative export-support efficiencies,  $e_A^P$  is reduced to  $2.0 \pm 0.5$ , while  $e_S^P$  and  $e_H^P$  are increased to  $0.5 \pm 0.1$  and  $0.6 \pm 0.2$ , respectively.)

### 7.3.2 Opal export

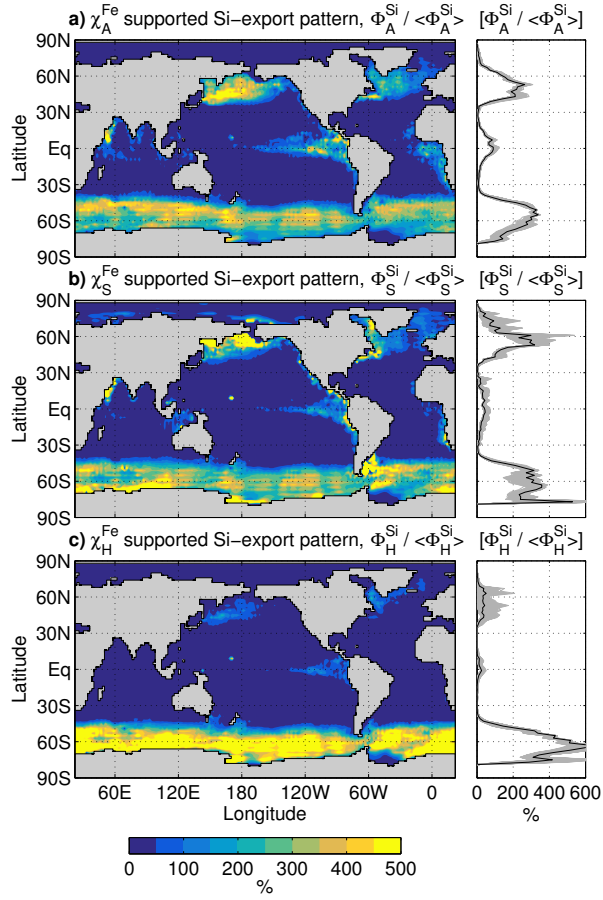
The opal export supported by each iron type can be calculated analogously, and the corresponding geographic patterns are shown in Figure 9. Similar to the total opal export (Figure 5b), the patterns of the opal export supported by each iron type emphasize regions with high diatom concentrations, namely the Southern Ocean and subpolar North Pacific and North Atlantic where there is also upwelling and/or vertical mixing. Aeolian-iron-supported opal export (Fig. 9a) is large in the Southern Ocean, but most pronounced in the subpolar North Pacific, where both diatom production is significant and aeolian input is high downwind from Asia's deserts. While tropical opal export is of secondary importance, the tropics are most pronounced for aeolian-supported export, again because of the direct source there. The pattern of sedimentary-iron-supported opal export (Fig. 9b) is broadly similar to that for aeolian dFe, but weaker in the tropics. The pattern of hydrothermal-iron-supported opal export (Fig. 9c) is dominated by the Southern Ocean, where diatom production is high and where most hydrothermal iron upwells. The patterns of iron-type-supported opal export have tightly clustered zonal means with a similar spread across the family as for the phosphorus export.

The amplitude of the opal-export patterns varies systematically with the iron source strength as summarized in Figure 10, which shows the fractional iron-type-supported opal export,  $\hat{\phi}_k^{\text{Si}} \equiv \langle \Phi_k^{\text{Si}} \rangle / \langle \Phi^{\text{Si}} \rangle$ , as a function of the corresponding fractional dFe source,  $\hat{\phi}_k$ . While aeolian dFe is still the most efficient iron type for supporting opal export, aeolian dFe is less efficient for opal export than for phosphorus export (the aeolian points fall closer to the 1 : 1 line by  $\sim 10\%$  fractional export). For a given fractional aeolian source strength between 20 % and 60 %, the supported fractional opal export is typically  $\sim 20\%$  larger, though there is significant scatter. Conversely, sedimentary iron is slightly more efficient in supporting opal export than in supporting phosphorus export, and hydrothermal dFe is only slightly less efficient than sedimentary dFe.

Similarly to our analysis of phosphorus export in the previous section, we define the relative opal export-support efficiencies by  $e_k^{\text{Si}} \equiv \epsilon_k^{\text{Si}} / \tilde{\epsilon}_k^{\text{Si}}$ , where  $\epsilon_k^{\text{Si}} \equiv \hat{\phi}_k^{\text{Si}} / \hat{\phi}_k$  and  $\tilde{\epsilon}_k^{\text{Si}} = (1 - \hat{\phi}_k^{\text{Si}}) / (1 - \hat{\phi}_k)$ . Nonlinear least-squares fit result in relative opal export-support efficiencies of  $e_A^{\text{Si}} = 2.3 \pm 0.5$ ,  $e_S^{\text{Si}} = 0.5 \pm 0.2$ , and  $e_H^{\text{Si}} = 0.5 \pm 0.2$ . Per source-injected molecule, aeolian iron is thus  $2.3 \pm 0.5$  times more efficient in supporting opal export than the other sources, while sedimentary and hydrothermal iron are  $1.9 \pm 0.5$  and  $2.2 \pm 1.0$  times less efficient, respectively, than the other sources.

The lower per-source efficiency of aeolian iron for supporting opal export is consistent with the fact that opal export occurs primarily in the Southern Ocean, where direct aeolian input is small. Similarly, the greater per-source efficiency of sedimentary



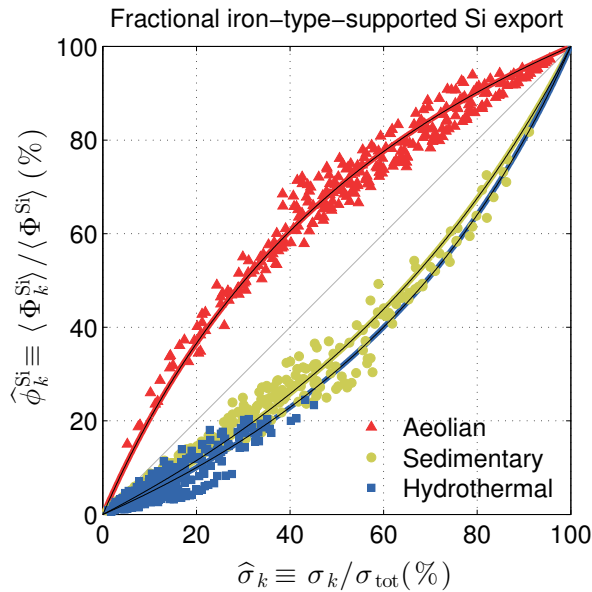


**Figure 9.** Opal export supported by each iron type [aeolian (a), sedimentary (b), hydrothermal (c)] normalized by its global mean. The map is for our typical state estimate, while zonal averages of the normalized opal export are shown for each family member in grey, with the typical state estimate in black.

and hydrothermal iron is consistent with the bulk of the opal export occurring in the upwelling regions of the Southern Ocean where access to deep iron sources is greatest. This is supported by the fact that ~~a plot of the fractional phosphorus export of the plots of  $\hat{\phi}_k^{\text{P}}$  and  $\hat{\phi}_k^{\text{Si}}$  for the Southern Ocean only versus  $\hat{\sigma}_k$  (not shown) versus the fractional iron sources is are both~~ nearly identical to the plot of fractional *global* opal export of Figure 10. (The relative Southern-Ocean-only opal export-support

5 efficiencies are  $e_A^{\text{Si}} = 1.9 \pm 0.5$ ,  $e_S^{\text{Si}} = 0.5 \pm 0.2$ , and  $e_H^{\text{Si}} = 0.7 \pm 0.2$ .)





**Figure 10.** Percent global opal export supported by each iron type (aeolian, sedimentary, hydrothermal) versus the corresponding fractional source of that iron type. [Lines represent fits to theoretical curves with fixed relative export-support efficiencies. \(See text for details.\)](#)

## 8 Discussion and caveats

Our approach has a number of limitations that must be kept in mind. Most importantly, inverse-model estimates are only as good as the data used to constrain them. The dFe observations are too sparse in space and time to construct a gridded annual mean climatology like those available for  $\text{PO}_4$  and  $\text{Si}(\text{OH})_4$ . We averaged the available dFe data to minimize observational biases, but in many places observations are only available for one time of the year and likely contain seasonal biases. Other biases are likely introduced when dFe measurements alias episodic source events such mineral dust downwind from the major deserts (e.g., Croot et al., 2004; Johnson et al., 2010). In the near future, GEOTRACES will release an expanded data product that will include Pacific transects that were not available for the intermediate data product used here. The additional dFe observations will help constrain the hydrothermal sources, particularly the strength of the Pacific source relative to that of the other basins.

Important non-nutrient observational fields for our inverse model are the satellite-measured photosynthetically active radiation (PAR) and ocean-colour-derived estimates of the size-partitioned phytoplankton concentrations. Small-scale features of the PAR field, e.g., in the Weddell Sea where ice and cloud cover play a role, are uncertain with the PAR for different time averages showing different features. The satellite-based estimates of phytoplankton concentrations also carry unquantified uncertainties due to a number of assumptions (Kostadinov et al., 2016). In our inverse model, these estimates provide crucial constraints on

how carbon-organic-matter export is partitioned among the different functional classes. The unquantified uncertainties warrant re-evaluation as independent satellite-derived estimates become available in the future.

Most biogeochemical parameters are determined through objective optimization against available observations, but the construction of the cost function, and the choice of which parameters are optimized and which are prescribed, necessarily involves subjective choices. For example, choosing a different set of weights ( $\omega_P, \omega_{Fe}, \omega_{Si}, \omega_{plk}$ ) to combine the four terms of the cost function would result in different optimal parameters. Similarly, assigning greater weight to dFe data measured as part of a vertical profile introduces another arbitrary weight. As for any nonlinear least-squares problem, it is also important to recognize that any minimum of the cost function found numerically is not guaranteed to be the global minimum and it is always possible that a better fit exists for a different set of parameters. Conversely, depending on the choice of initial state, the optimizer may find a local minimum that has grossly unrealistic features and must be rejected.

We addressed the uncertainty in our estimates that results from the nearly unconstrained strengths of the external iron sources. The uncertainty in key metrics (e.g., global carbon-phosphorus export) was quantified in terms of their spread across our family of state estimates and in terms of systematic variations with the iron source strengths. While our efficient numerics allow us to easily determine the linear sensitivities of any metric with respect to all parameters (from which one can also estimate uncertainty), we did not do so here because the spread in the metric across the family is more relevant. Given the large set of parameters  $x_j$  and several interesting metrics  $M_i$ , a detailed investigation of all the sensitivities  $\partial M_i / \partial x_j$  evaluated at the optimal states is beyond the scope of this study. In principle, one can estimate the uncertainty of the optimal parameters themselves using a Bayesian framework (e.g., Teng et al., 2014). However, this requires the construction of suitable covariances and is also beyond the scope of this study.

A key limitation of our approach is that seasonality is ignored and we use a steady circulation. This circulation is constructed so that its transport reproduces the annual-mean observed temperature, salinity, CFC-11, radiocarbon, and  $PO_4$  fields with minimal error. The circulation is hence not a simple average, but an effective ventilation-weighted mean. However, we acknowledge that effects due to the seasonal covariance of biological production and circulation cannot be captured.

Our model of the nutrient cycles makes a number of simplifying assumptions. We ignore external inputs of silicic acid and therefore also neglect permanent burial of opal in sediments. While this approximation has been shown to have negligible impact on particle fluxes (Sarmiento et al., 2007), we acknowledge that our estimates will miss features such as, e.g., silicic-acid plumes due to crustal fluid venting (Johnson et al., 2006). The uncertainty of the silicon cycle that is most difficult to quantify stems from our simple parameterization of opal dissolution, which does not account for partial frustule protection by decaying organic material or the effect of digestion by zooplankton. Another key uncertainty lies in our parameterization of the Si : P uptake ratio, particularly its dependence on dFe. While our empirical formulation captures known dependencies qualitatively, a first-principles derivation based on cell biology is currently lacking. These remarks apply equally to the Fe : P uptake ratio.

Although our model of the iron cycle includes an explicit representation of the redissolution of scavenged iron, effects of subgrid topography, and dynamic coupling to the phosphorus and silicon cycles, and is thus much more complex and mechanistic than the iron model of Frants et al. (2016), it was still necessary to make a number of simplifying assumptions.

Specifically, we do not model ligands dynamically, ignore colloidal iron (e.g., Fitzsimmons and Boyle, 2014), and do not represent some iron sources that may be locally important such as input from icebergs (Klunder et al., 2011, 2014). We also assume that  $\text{PO}_4$  and dFe are remineralized with the same Martin curve and in the same ratio in which they were utilized. The recent work by Twining et al. (2014) suggests that sinking diatoms release phosphorus higher in the water column than iron, but we do not have sufficient information to model these effects. Given the large uncertainties in the external iron sources, the neglected details are likely of second order for estimating the large-scale dFe concentration.

Other uncertainties concern the phosphorus cycle to which the uptake of the other elements is keyed. While the optimized phosphate fields have the smallest misfit with observations, our model of the phosphorus cycle makes several simplifying assumptions that should be kept in mind. The Martin exponent is assumed to be globally uniform although in reality it almost certainly varies spatially (Weber et al., 2016), potentially leading to underestimated gradients in our model. To avoid carrying an additional tracer, we approximated DOP to have zero lifetime. In reality, DOP has a wide range of lifetimes, and the lifetime of semi-labile DOP is typically assumed to be a fraction of a year (e.g., Primeau et al., 2013). However, the neglect of DOP is unlikely to seriously affect our estimates. DOP represents only a tiny fraction (less than 1 %) of the total phosphorus pool (e.g., Pasquier and Holzer, 2016) and by using a Martin exponent optimized for a restoring model without DOP, we were able to match  $\text{PO}_4$  concentrations to within 5 % of the observations.

We emphasize that the carbon export reported here was simply our estimate of the phosphorus export converted to carbon units. No effort was made to compute a more realistic carbon export such as could be achieved with an explicit representation of the carbon cycle (which would require additional tracers and was numerically too expensive) and the C : P export ratio was treated as globally uniform. While a globally uniform export ratio is acceptable for a unit conversion, the true C : P export ratio is now known to vary spatially (~~Teng et al., 2014~~). (e.g., Galbraith and Martiny, 2015; Teng et al., 2014). Although using the regionally varying C : P ratios would be more realistic, we find that variations in C : P have only modest effects on the globally integrated carbon export: (i) Applying the P : C relation of Galbraith and Martiny (2015) to the phosphorus export of our typical state gives a carbon export of  $8.5 \pm 0.4 \text{ Pg C yr}^{-1}$  or  $9.4 \pm 0.9 \text{ Pg C yr}^{-1}$  when we use their log-binned parameter values. (ii) Applying the regional C : P inverse-model estimates of Teng et al. (2014) gives a carbon export of  $10. \pm 2. \text{ Pg C yr}^{-1}$ . Both this estimate and the one based on the log-binned regression agree within their uncertainties with our simple unit-conversion value of  $10.3 \pm 0.4 \text{ Pg C yr}^{-1}$ .

## 9 Summary and Conclusions

We have formulated a steady-state model of the coupled phosphorus, silicon, and iron cycles that is embedded in a steady data-assimilated global circulation. The model is of intermediate complexity and couples the nutrient cycles through colimitations on biological uptake and through the scavenging of iron by organic particles. The concentrations of the small, large, and diatom phytoplankton functional classes are calculated diagnostically, which avoids the need for plankton concentration tracers. We explicitly represent iron scavenging by POP, opal, and mineral-dust particles, and the redissolution of POP- and opal-scavenged iron. Subgrid topography is parameterized for the sedimentary iron sources and intercepts all vertical fluxes. The relative

simplicity of the biogeochemical model and the matrix formulation of the steady-state advective-diffusive transport afford highly efficient numerics. Steady-state solutions are readily found using a Newton solver, which permits the model to be used in inverse mode to constrain many of the biogeochemical parameters through objective optimization. The optimization minimizes the mismatch with the observed nutrient concentrations and with satellite-derived estimates of phytoplankton concentrations.

- 5 Our estimates of the macronutrient concentrations closely match the observational WOA13 climatology with volume-weighted RMS errors of 5 % for phosphate and 12 % for silicic acid relative to the global mean. The modelled dFe concentration has a larger cost-weighted RMS mismatch of  $\sim 45$  % relative to the global mean. However, the cost-weighted basin-averaged vertical dFe profiles for the Atlantic and Southern Ocean generally lie within the observational uncertainties. The Pacific dFe profiles show systematic biases, in part because ~~the Pacific basin contains~~ there are relatively few dFe observations ~~in the~~  
10 ~~currently available data sets. The estimated phytoplankton concentrations have generally weaker meridional gradients than observed possibly because of our use of a globally uniform Martin exponent. However, the fraction of the global plankton biomass represented by each~~ for the Pacific, with no Pacific transect in the GEOTRACES Intermediate Data Product. The fractional global biomass of each phytoplankton functional class lies within 7 % of the ~~fraction in the corresponding size class as estimated by Kostadinov et al. (2009) from ocean colour observations~~ observation-based estimates by Kostadinov et al. (2009).
- 15 Because even the order of magnitude of the iron sources is uncertain, we produced a family of state estimates with a wide range of iron source strengths. Because different iron source strengths are compensated by optimally adjusting the scavenging parameters, each family member fits the observations with roughly the same fidelity. This means that the available observed dFe and phytoplankton concentrations by themselves are insufficient to constrain the sources. This conclusion can also be gleaned from the model intercomparison of Tagliabue et al. (2016) and was reached using an inverse model by Frants et al.  
20 (2016), who also considered a family of state estimates. However, while Frants et al. (2016) varied only the aeolian source, our estimates here explore a range of sedimentary and hydrothermal source strengths in addition to a much wider range of aeolian source strengths.

We partitioned the dFe concentration field into its aeolian, sedimentary, and hydrothermal components without perturbing the system using the approach of Holzer et al. (2016). While the individual source components vary widely depending on the source  
25 strengths, we find that the total dFe concentration given by the sum of the source components is well constrained. Variations in the aeolian component are compensated primarily by sedimentary dFe. Both the compensations between different iron types and between effective sources and sinks suggest that a more dense sampling of the ocean's dFe field by future measurement campaigns may not provide the information necessary for constraining the source strengths. The required information may ultimately have to come from better direct quantification of the source and/or scavenging processes themselves.

- 30 Nutrient limitation patterns were defined by jointly considering if the  $\text{PO}_4$ ,  $\text{Si}(\text{OH})_4$ , and dFe concentrations fell below their half-saturation values for uptake. Iron limitation was thus deemed to occur where only dFe fell below its half-saturation value, phosphate-iron colimitation where both  $\text{PO}_4$  and dFe fell below their half-saturation values, and so on. The resulting limitation patterns are robust across our family of state estimates and broadly consistent with direct observations (Moore et al., 2013) and with alternatively defined limitation patterns in the BEC model (Moore et al., 2004). The large and diatom functional  
35 classes show iron limitation in the Southern Ocean, eastern tropical Pacific and subpolar North Pacific, with  $\text{PO}_4$ -dFe and (for

diatoms)  $\text{PO}_4\text{-Si(OH)}_4\text{-dFe}$  colimitations in the Pacific and South Atlantic subtropical gyres. The Indian Ocean, tropical Atlantic, and North Atlantic are largely iron replete (i.e., not limited in the sense defined) with  $\text{PO}_4$  limitation and for diatoms  $\text{PO}_4\text{-Si(OH)}_4$  colimitation.

The export productions of phosphorus and opal are well constrained across our family of state estimates, in terms of both pattern and magnitude. Because we model three phytoplankton functional classes with distinct ~~-,~~ optimized uptake timescales, our ~~carbon export of~~ phosphorus export (expressed in carbon units) of  $9.5\text{--}11. \text{ Pg C yr}^{-1}$  is  $\sim 30\text{--}45\%$  larger than that estimated by Primeau et al. (2013) and closer in spatial pattern to the satellite-based estimates of Dunne et al. (2007). The opal export of  $164\text{--}177 \text{ Tmol Si yr}^{-1}$  overlaps with the estimate of Holzer et al. (2014), who used a simple restoring-type model of the silicon cycle uncoupled from other nutrients.

We ~~estimate the percentage of the global carbon export due the small, large, and diatom functional classes to be  $(6.4 \pm 0.3)\%$ ,  $(52 \pm 2)\%$ , and  $(41 \pm 3)\%$ , respectively. These percentages were not well constrained by the nutrient and phytoplankton concentration data and required joint tuning of the parameters of the uptake ratio and of the uptake timescales so that the fractional uptake by each class approximately matched the satellite-based estimates of Uitz et al. (2010).~~

We quantified the role of the iron cycle in shaping the carbon-phosphorus and opal export productions. We find that each iron source type (aeolian, sedimentary, hydrothermal) supports phosphorus and opal exports with a distinct geographic pattern that is robust across the family of state estimates. The export pattern supported by a given iron type reflects the nature of its source. Sedimentary and hydrothermal iron support carbon-phosphorus export that is dominantly shaped by the large-scale patterns of upwelling, which brings these iron types to the surface. Aeolian iron supports export that is shaped by both the pattern of direct aeolian input and by large-scale upwelling, which brings regenerated as well as scavenged and redissolved aeolian iron back into the euphotic zone. For opal export, the signature of each iron type is qualitatively similar, but compared to carbon-phosphorus export, the opal export patterns tend to be weaker in the tropics and stronger at high latitudes, especially in the Southern Ocean where diatom concentrations and silicon trapping are strongest.

The globally integrated export supported by a given iron type varies systematically with its fractional global source. These variations quantify the export-support efficiency of each iron type per source-injected molecule for supporting export. Aeolian iron is most efficient ~~in the sense that the percentage of the export it supports exceeds the percentage of the aeolian source by as much as  $\sim 30\%$  for carbon export and  $\sim 20\%$  for opal export. Globally, and supports a fraction of the global export that is larger than its fractional source, while~~ sedimentary and hydrothermal iron are ~~inefficient, supporting a fraction of export that is up to  $\sim 20\%$  less efficient, supporting fractions of the global export that are~~ less than their fractional sources. ~~For carbon export, hydrothermal iron is least efficient supporting about 1/3 less fractional export than sedimentary iron. When the effect of direct aeolian deposition is minimized by calculating the exports for the Southern Ocean only, the efficiency of a given~~ This is because dFe from deeper sources is more likely to be scavenged en route to the euphotic zone. The relation between the fractional export supported by each iron type and its fractional source reveals that the relative export-support efficiency of each iron type is similar regardless of whether carbon or opal export is considered. This reflects the similar oceanic transport paths of all iron types into the Southern Ocean euphotic zone, where the bulk of aeolian iron is supplied in regenerated form through upwelling robust across our family of estimates. Per source-injected molecule, aeolian iron supports  $3.1 \pm 0.8$  times more

phosphorus export and  $2.3 \pm 0.5$  times more opal export than the other iron types. Conversely, sedimentary and hydrothermal iron are respectively  $2.3 \pm 0.6$  and  $3.7 \pm 2.3$  times *less* efficient in supporting phosphorus export, and  $1.9 \pm 0.5$  and  $2.2 \pm 1.0$  times less efficient in supporting opal export, than the other iron types.

Our optimized model is ideally suited for investigating the response of the global ocean ecosystem to a variety of biogeochemical perturbations. In the future, we will report on the model's response to perturbations in the iron supply and on a more comprehensive analysis of the detailed workings of the iron cycle.

## 10 Data availability

The temperature, phosphate, and silicic-acid data used in this study are available from the World Ocean Atlas v2 2013 ([www.nodc.noaa.gov/OC5/woa13/woa13data.html](http://www.nodc.noaa.gov/OC5/woa13/woa13data.html)). The dFe data used in this study, including the TFE data set, are available from GEOTRACES ([www.bodc.ac.uk/geotraces/data](http://www.bodc.ac.uk/geotraces/data)). The satellite estimates of the concentrations of picophytoplankton, nanophytoplankton, and microphytoplankton are available from the PANGAEA data repository (doi:10.1594/PANGAEA.859005). The yearly irradiance data from NASA's MODIS Aqua PAR are available from the OceanColor website ([oceandata.sci.gsfc.nasa.gov](http://oceandata.sci.gsfc.nasa.gov)).

## Appendix A: Recycling operators for scavenged iron

The recycling operator for POP-scavenged iron,  $\mathcal{B}^{\text{s,POP}}$ , comprises two parts: For dFe scavenged in the euphotic layer, i.e., above  $z_e$ ,  $\mathcal{B}^{\text{s,POP}}$  acts identically to  $\mathcal{B}^{\text{P}}$ . For dFe scavenged below  $z_e$ , we solve the flux equation of scavenged iron for continuous  $z$ . We assume that iron scavenged by POP below the mixed layer continuously sinks and can be recycled in the same grid cell in which it was scavenged. (We assume that the mixed layer coincides with the euphotic zone.) As shown by Kriest and Oschlies (2008), the Martin curve can be simply modeled with a sinking speed linearly increasing with depth, an approach we follow here. The equation for the flux of iron,  $\phi_{\text{Fe}}^{\text{POP}}$ , that was scavenged by POP below the mixed layer is thus

$$20 \quad \frac{\partial \phi_{\text{Fe}}^{\text{POP}}}{\partial z} + \frac{b}{z} \phi_{\text{Fe}}^{\text{POP}} = -f^{\text{POP}} J_{\text{POP}} \quad , \quad (\text{A1})$$

where  $b$  is the Martin exponent of the POP flux, and with the condition that  $\phi_{\text{Fe}}^{\text{POP}} = 0$  for  $z > z_e$  because here we only consider dFe scavenged below  $z_e$ . The solution is given by

$$\phi_{\text{Fe}}^{\text{POP}}(z) = - \int_z^{z_e} \left( \frac{z}{z'} \right)^{-b} f^{\text{POP}} J_{\text{POP}}(z') dz' \quad . \quad (\text{A2})$$

And the rate per unit volume at which POP-scavenged dFe is recycled is thus given by

$$25 \quad \mathcal{B}^{\text{s,POP}} f^{\text{POP}} J_{\text{POP}} = \mathcal{B}^{\text{P}} f^{\text{POP}} J_{\text{POP}} + \frac{b}{z} \phi_{\text{Fe}}^{\text{POP}} \quad , \quad (\text{A3})$$

where the first term is for iron that was scavenged in the euphotic zone, and the second term for iron that was scavenged in the interior.

Similarly, the recycling operator for opal-scavenged iron,  $\mathcal{B}^{s, \text{bSi}}$ , has a euphotic part identical to  $\mathcal{B}^{\text{Si}}$ , and an aphotic interior part. In the interior, we solve the continuous equation for the flux of iron,  $\phi_{\text{Fe}}^{\text{bSi}}$ , that was scavenged by opal below the mixed layer. The flux obeys

$$\frac{\partial \phi_{\text{Fe}}^{\text{bSi}}}{\partial z} + \frac{\kappa_{\text{Si}}^{\text{max}}}{w_{\text{Si}}} e^{-T_E/T(z)} \phi_{\text{Fe}}^{\text{bSi}} = -f^{\text{bSi}} J_{\text{bSi}} \quad , \quad (\text{A4})$$

- 5 with the condition that  $\phi_{\text{Fe}}^{\text{bSi}} = 0$  for  $z > z_e$ . The solution is of the form

$$\phi_{\text{Fe}}^{\text{bSi}}(z) = - \int_z^{z_e} Q(z, z') f^{\text{bSi}} J_{\text{bSi}}(z') dz' \quad , \quad (\text{A5})$$

so that the flux of opal-scavenged iron at  $z$  integrates all the scavenging of dFe by opal that occurs above  $z$  (and below  $z_e$ ) and that is not recycled before reaching  $z$ . This is accomplished by  $Q(z, z')$ , given by

$$Q(z, z') = \exp \left( - \int_z^{z'} \frac{\kappa_{\text{Si}}^{\text{max}}}{w_{\text{Si}}} e^{-T_E/T(z'')} dz'' \right) \quad , \quad (\text{A6})$$

- 10 which removes all the recycling that occurs between the current depth  $z$  and the depth of scavenging  $z'$ . The rate per unit volume at which opal-scavenged dFe is recycled is thus given by

$$\mathcal{B}^{s, \text{bSi}} f^{\text{bSi}} J_{\text{bSi}} = \mathcal{B}^{\text{P}} f^{\text{bSi}} J_{\text{bSi}} + \frac{\kappa_{\text{Si}}^{\text{max}}}{w_{\text{Si}}} e^{-T_E/T(z)} \phi_{\text{Fe}}^{\text{bSi}} \quad , \quad (\text{A7})$$

where the first term is for iron that was scavenged in the euphotic zone, and the second term for iron that was scavenged in the aphotic interior.

## 15 Appendix B: Biogenic transport operators with subgrid topography

We follow Moore and Braucher (2008) to include the effects of the more realistic, high-resolution topographic data from the National Geophysical Data Center (2006). The subgrid topography parameterization must be reflected in the redistribution operators. Here we explain how this is done, based on the biogenic redistribution operators  $\mathcal{B}^{\text{P}}$  and  $\mathcal{B}^{\text{Si}}$ , which link biological production in the euphotic zone to remineralization or redissolution in the aphotic zone. The operators  $\mathcal{B}^{\text{P}}$  and  $\mathcal{B}^{\text{Si}}$  are related

- 20 to the divergence of the aphotic particle fluxes through

$$\mathcal{B}^{\text{P}} f_c U_c = \frac{\partial}{\partial z} [\Theta(z - z_b) \phi_{\text{POP}}(z)] \quad , \quad (\text{B1})$$

and

$$\mathcal{B}^{\text{Si}} f_{\text{dia}} U_{\text{dia}} = \frac{\partial}{\partial z} [\Theta(z - z_b) \phi_{\text{bSi}}(z)] \quad , \quad (\text{B2})$$

where  $z_e$  and  $z_b$  are the height coordinates at the base of the euphotic zone, and at the ocean bottom, respectively. The Heaviside

- 25 function,  $\Theta(z - z_b)$ , ensures that the fraction of POP that reaches  $z_b$  is redissolved and remineralized there (Primeau et al., 2013; Holzer et al., 2014).

The subgrid topography parameterization is implemented by applying equations (B1)–(B2) to the high-resolution topography. In practice, for each model grid cell, we calculate the fraction of the flux that should remineralize where it hits the topography. Note, to ensure that  $\mathcal{B}^P$  and  $\mathcal{B}^{Si}$  are conservative, special care is required where subgrid topography is present within euphotic grid cells. In this case, a corresponding fraction of sinking particles must remineralize in that cell. We similarly  
 5 implemented the same subgrid topography parameterization to the scavenging redistribution operators,  $\mathcal{B}^{s,POP}$  and  $\mathcal{B}^{s,bSi}$ .

### Appendix C: Weights for dFe mismatch

We use the dFe observations of both the global data set compiled by Tagliabue et al. (2012) and the GEOTRACES Intermediate Data Product v3 (Mawji et al., 2015). We combine both data sets and remove dFe observations above 2.71 nM which probably correspond to transient states with short timescales that cannot be captured by our steady-state model. In order to compensate  
 10 the fact that most dFe observations are close to the surface, we give more weight to observations that are part of a “profile”. (A dFe observation is deemed to belong to a “profile” if there are 10 or more observations at the same latitude and longitude, and if one of those was recorded deeper than 2000 m.) Because the dFe observations do not sample the seasonal cycle uniformly, we adopt an approach similar to Frants et al. (2016) to reduce potential sampling bias when we interpolate the data to our model grid: If multiple dFe observations lie in the same grid cell, we first take the seasonal averages, which we then averaged  
 15 again to estimate the annual mean.

As in equations (22) for  $PO_4$  and  $Si(OH)_4$ , we use volume weights to evaluate the dFe concentration mismatch with observations. However, because not all model grid cells contain dFe observations, we define a dFe-specific vector of grid-box volumes,  $\mathbf{v}_{Fe}^{all}$ , which has nonzero elements only for grid boxes that contain at least one dFe observation. We also define a dFe “profile-specific” vector,  $\mathbf{v}_{Fe}^{pro}$ , which is nonzero only for grid boxes that contain “profile” observations. The corresponding  
 20 weights are defined by

$$\mathbf{w}_{Fe}^{all} = \frac{\mathbf{v}_{Fe}^{all}}{(\bar{\chi}_{Fe}^{obs})^2 V_{Fe}^{all}} \quad \text{and} \quad \mathbf{w}_{Fe}^{pro} = \frac{\mathbf{v}_{Fe}^{pro}}{(\bar{\chi}_{Fe}^{obs})^2 V_{Fe}^{pro}}, \quad (C1)$$

where  $V_{Fe}^{all}$  is the total volume of grid cells which contain a dFe observation, and  $V_{Fe}^{pro}$  the total volume of grid cells containing “profile” observations. We define the total dFe weight vector,  $\mathbf{w}_{Fe}$ , for the mismatch with observations in equation (23), by

$$\mathbf{w}_{Fe} = \frac{1}{5} \mathbf{w}_{Fe}^{all} + \frac{4}{5} \mathbf{w}_{Fe}^{pro}, \quad (C2)$$

25 where we give extra weight to the “profile” observations. The 1 : 4 ratio was manually adjusted until “profile” observations were deemed to have sufficiently strong influence on the state estimates. We also tried different approaches to weight the model-observation dFe mismatch, including the use of inverse variances (Frants et al., 2016), but we found no significant difference in our results.



## Appendix D: Optimization strategy details

### D0.1 Prescribed parameters

The following considerations determined which parameters were not optimized and how their values were chosen. The recyclable fractions of POP and opal scavenging,  $f^{\text{POP}}$  and  $f^{\text{bSi}}$ , compensate with the maximum Fe : P uptake ratio,  $R_0^{\text{Fe:P}}$ , and thus were prescribed at 90 % (Moore and Braucher, 2008). (This compensation results from the biological iron pump having almost the same effect as the combination of scavenging and recycling iron.) Similarly, the detrital fractions,  $f_c^0$ , which set the particle export ratio, are directly compensated by all the other parameters in the uptake formulation. We therefore followed Dunne et al. (2005) and assigned their “small” detrital fraction to  $f_{\text{sml}}^0$  and their “large” detrital fraction to both  $f_{\text{lrg}}^0$  and  $f_{\text{dia}}^0$ . When trying to optimize the silicon half-saturation rate  $k_{\text{dia}}^{\text{Si}}$ , starting from a value of  $1 \text{ mmol m}^{-3}$  (e.g., Matsumoto et al., 2013), we found that the optimal value always remained within a few percent of this initial value. This is in part due to the fact that in regions of high diatom concentration the Monod term for silicic acid is near saturation so that there is little sensitivity to the precise value of  $k_{\text{dia}}^{\text{Si}}$ . Moreover, there appears to be consistency across the literature that  $k_{\text{dia}}^{\text{Si}} = 1 \text{ mmol m}^{-3}$ . We therefore simply fixed  $k_{\text{dia}}^{\text{Si}}$  at this value for numerical efficiency.

### 15 D0.2 Choice of initial parameter values

We first chose an initial set of values for the remaining parameters as collected in Table 2. The parameters of the iron cycle were taken from the typical state estimate of Frants et al. (2016) except for the half-saturation constant of the Fe : P ratio, which was taken from the work of Galbraith et al. (2010), and the scavenging-rate parameters. The initial parameters for POP and opal scavenging,  $\kappa_{\text{scv}}^{\text{POP}}$  and  $\kappa_{\text{scv}}^{\text{bSi}}$ , were determined so that the globally integrated scavenging of each process was initially  $\sim 5 \text{ Gmol Fe yr}^{-1}$  (the typical total source/sink strength reported by Frants et al. (2016)). The initial value of the dust scavenging rate parameter,  $\kappa_{\text{dst}}^{\text{POP}}$ , was chosen so that the sink due to dust scavenging was  $\sim 10 \%$  of the total sink of the initial state.

The initial irradiance half-saturation constants were taken from the work of Doney et al. (2006). The initial uptake half-saturation constants  $k_c^i$  were taken from the work of Matsumoto et al. (2013). The uptake timescales  $\tau_c$  were set to an initial value of 6 days and optimized subject to the constraint  $\tau_{\text{sml}} \geq \tau_{\text{lrg}} \geq \tau_{\text{dia}}$ . The initial values of the maximum phytoplankton concentrations were calculated as  $p_c^{\text{max}} = p^* / (\lambda_0 \tau_c)$  using  $p^* = 0.018 \text{ mmol P m}^{-3}$  (Galbraith et al., 2010) and  $\lambda_0^{-1} = 5.26 \text{ d}$  (Dunne et al., 2005).

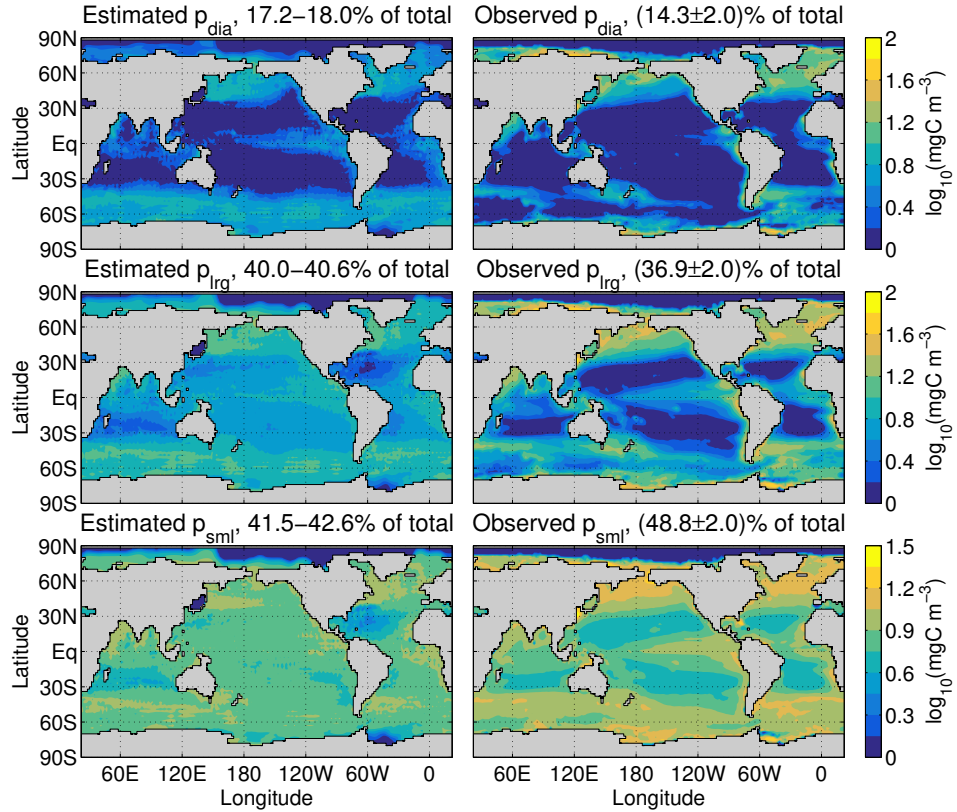
The initial values of the parameters of the Si : P ratio were set so that  $k_{\text{Si:P}}^{\text{Fe}}$  and  $k_{\text{Si:P}}^{\text{Si}}$  were on the order of typical dFe and  $\text{Si(OH)}_4$  concentrations, while  $R_0^{\text{Si}}$  and  $R_m^{\text{Si}}$  were based on corresponding Si : N uptake ratios found in the literature and converted using N : P = 16 : 1. Thus, in terms of Si : N units,  $R_0^{\text{Si}}$  was chosen to be on the order of the minimum Si : N uptake ratio used by Matsumoto et al. (2013) and  $R_m^{\text{Si}}$  was chosen to be on the order of the maximum Si : N uptake ratio reported by Franck et al. (2000) and Brzezinski et al. (2002).

### D0.3 Sequential optimization steps

- (a) We first optimized the hydrothermal iron source parameters  $\sigma_{H,ATL}$ ,  $\sigma_{H,PAC}$ ,  $\sigma_{H,IND}$ , and  $\sigma_{H,SO}$  forcing  $\sigma_{H,IND} = \sigma_{H,SO}$ . Although we adjusted the overall source strength of the hydrothermal iron source when generating our family of iron cycling estimates, we did not re-optimize the relative strength of the four basin amplitudes until the final step of our strategy.
- 5 (b) We jointly optimize the three irradiance half-saturations  $k_c^I$  and then keep these fixed because of potential compensation with the half-saturation constants  $k_c^i$ .
- (c) We jointly optimized the half-saturations  $k_c^i$  of the nutrient-limitation Monod terms.
- (d) We were not able to optimize the Si : P uptake ratio parameters, because they are not well constrained due to compensation with the parameters that set the uptake by diatoms. We therefore separately tuned the parameters  $R_m^{Si}$ ,  $R_0^{Si}$ ,  $k_{Si:P}^{Si}$ , and  $k_{Si:P}^{Fe}$
- 10 together with the three growth timescales  $\tau_c$  to match the fractions of total uptake by each phytoplankton class as estimated by Uitz et al. (2010).
- (e) Because of compensation with the maximum Fe : P uptake ratio, the associated half-saturation rate,  $k_{Fe:P}$ , was optimized on its own.
- (f) We then jointly re-optimized the 13 parameters  $k_c^P$ ,  $k_c^{Fe}$ ,  $k_{dia}^{Si}$ ,  $p_c^{max}$ , and  $\tau_c$ .
- 15 (g) Only the parameters of the iron cycle remain to be optimized. Iron source and sink parameters cannot jointly be optimized because of strong local compensation. (Although the aeolian source injects dFe at the surface, uptake and scavenging export iron to depth, thus creating an effective interior source.) To generate our family of estimates, we therefore first assigned the aeolian, sedimentary, and hydrothermal source-strength parameters (keeping the same ratio of basin hydrothermal source strengths to global hydrothermal source strength), and held these fixed while jointly optimizing the parameters determining the
- 20 iron scavenging, namely  $R_0^{Fe:P}$ ,  $\kappa_{scv}^{POP}$ ,  $\kappa_{scv}^{bSi}$ ,  $\kappa_{scv}^{dst}$ ,  $L_v$ ,  $L_{max}$ ,  $L_b$ , and  $\tau_b$ .
- (h) As a final step, we jointly optimized all source-strength parameters  $\sigma_A$ ,  $\sigma_S$ ,  $\sigma_{H,ATL}$ ,  $\sigma_{H,PAC}$ ,  $\sigma_{H,IND}$ , and  $\sigma_{H,SO}$  primarily to give the hydrothermal source pattern (relative strength in each basin) a chance to adjust from its initial state. We find that if the total hydrothermal source strength,  $\sigma_H$ , is below  $1 \text{ Gmol yr}^{-1}$ , the final optimization step hardly changes  $\sigma_H$ , while larger hydrothermal source strengths tend to be reduced by up to  $\sim 0.5 \text{ Gmol yr}^{-1}$ . If  $\sigma_H < 0.5 \text{ Gmol yr}^{-1}$ , the fractional
- 25 hydrothermal source strength of each basin tends to remain unchanged during this last step, while if  $\sigma_H \geq 0.5 \text{ Gmol yr}^{-1}$ , the fractional hydrothermal sources for the Pacific and Southern Ocean tend to increase by order 10 %, the fractional Indian Ocean source tends to decrease order 5 %, and the Atlantic source is typically reduced by order 10 % and for some state estimates to near zero. The aeolian iron source strength,  $\sigma_A$ , tends to be reduced by  $\sim 10 \%$ , while the sedimentary iron source strength,  $\sigma_S$ , stays within 10 % of its previous value for most family members but can more than double for cases with high hydrothermal
- 30 and aeolian sources.

### Appendix E: ~~Phytoplankton~~ Inferred versus observed phytoplankton distribution

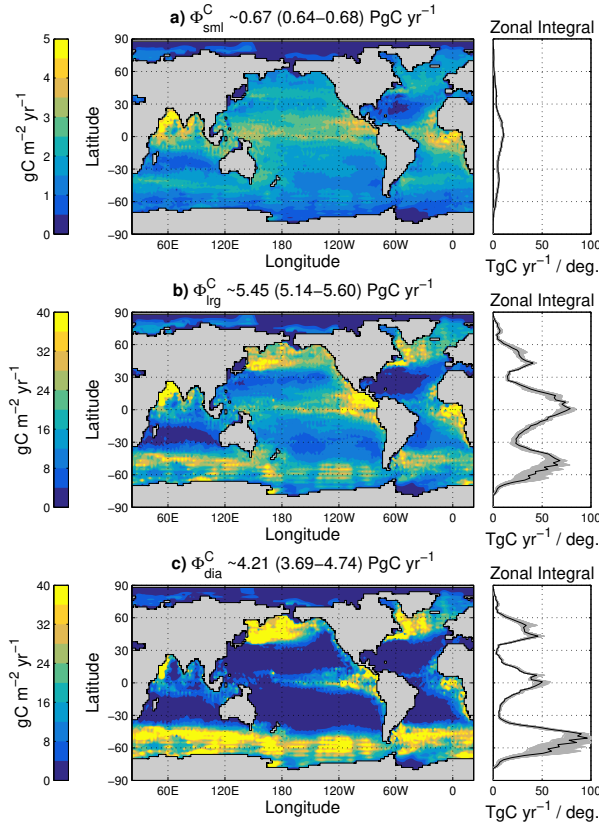
Figure E1 shows a model–observation comparison of the phytoplankton concentration (plotted in C units using a constant Redfield ratio of C : P = 106 : 1). Although the distinction between our phytoplankton classes is functional and not determined



**Figure E1.** Comparison of estimated (left) and observed (right) phytoplankton concentrations averaged vertically over the model euphotic zone. The diatom concentration is shown in the top row, the large phytoplankton concentration in the middle row, and the small phytoplankton concentration in the bottom row. Note the logarithmic colour scales with separate ranges for each functional class.

by size, we compare our small, large, and diatom concentrations with the picophytoplankton ( $0.5\text{--}2\mu\text{m}$ ), nanophytoplankton ( $2\text{--}20\mu\text{m}$ ), and microphytoplankton ( $20\text{--}50\mu\text{m}$ ) of the satellite-based estimates of Kostadinov et al. (2016), consistent with the construction of our  $E_{\text{plk}}$  cost function.

The inverse-model estimates capture the broad global patterns of the phytoplankton concentrations reasonably well, although some biases are also evident. The observation-based diatom and large concentration has a minimum at  $\sim 60^\circ\text{S}$ , a feature our estimates do not capture. This may be related to ~~seasonality issues, the strong seasonality of the Southern Ocean, with its large variability in sea-ice coverage, the effects of~~ which our approach cannot capture. Our estimates for the large and small concentrations have higher concentrations in the subtropical gyres and weaker meridional gradients with lower high-latitude and tropical concentrations than observed. ~~These discrepancies may again be related to the absence of seasonality.~~ Another factor is that the phytoplankton mismatch carries less penalty in our cost than the combined misfit terms of the three nutrient fields.



**Figure F1.** Local export production (maps on left) and its zonal integral (curves on the right) expressed in carbon units (using  $C : P = 106 : 1$ ). Maps are shown for our typical state estimate, while we plot the zonal integral of each family member in grey and the typical state estimate in black. The export productions are plotted for each phytoplankton functional class: Small (top plots, a), large (middle plots, b), and diatom (bottom plots, c). Note the different colour scale for the small class.

The global mean phytoplankton concentration of each functional class was remarkably stable across all members of our family of state estimates with ranges of  $\overline{p}_{\text{dia}} = (2.6\text{--}2.9) \text{ mg C m}^{-3}$ ,  $\overline{p}_{\text{lrg}} = (6.2\text{--}6.5) \text{ mg C m}^{-3}$ , and  $\overline{p}_{\text{sml}} = (6.5\text{--}6.8) \text{ mg C m}^{-3}$ . This indicates that the satellite data provides a good constraint on the global-scale ecosystem composition.

## Appendix F: Partition of export production by phytoplankton class

- 5 Figure F1 shows the ~~carbon export~~ phosphorus export, expressed in C units, partitioned according to each functional class. The ~~bulk of the carbon export is accomplished by the large and diatom phytoplankton classes ( $\sim 53$  and  $41$  export due to each functional class ( $6.5 \pm 0.3\%$ , respectively), while the remainder ( $\sim 6$  for small,  $53. \pm 1. \%$  for large, and  $41. \pm 1. \%$ ) is exported by the small class. In the tropics, the large class dominates the export, while in the high latitudes, and particularly~~

in the Southern Ocean, the diatom class dominates. This for diatoms for our state estimates), are poorly constrained because of uncertainty and lack of consensus about the values of the detrital fractions as discussed in Section 2.2 for phytoplankton, and uncertainty on the Si : P ratio for diatoms. However, this partition between the three phytoplankton classes is the result of the adjustments of the class-specific growth timescales,  $\tau_c$ , and of the parameters of the Si : P ratio to bring the fractional uptake by each class into alignment with the satellite-based estimates of Uitz et al. (2010). For our typical state estimate, these uptake fractions are 38 %, 42 %, and 30 % for the diatom, large, and small classes, respectively. This compares to 32 %, 44 %, and 24 % for micro-, nano-, and picophytoplankton, respectively, in the satellite-based estimates. (We find that if we use only a single optimized timescale for all three classes, the small phytoplankton class completely dominates the phosphorus export, underlining the need for class-specific growth timescales.)

## 10 Appendix G: **Iron source and sink patterns** Comparison with select transects of the GEOTRACES Intermediate Data Product

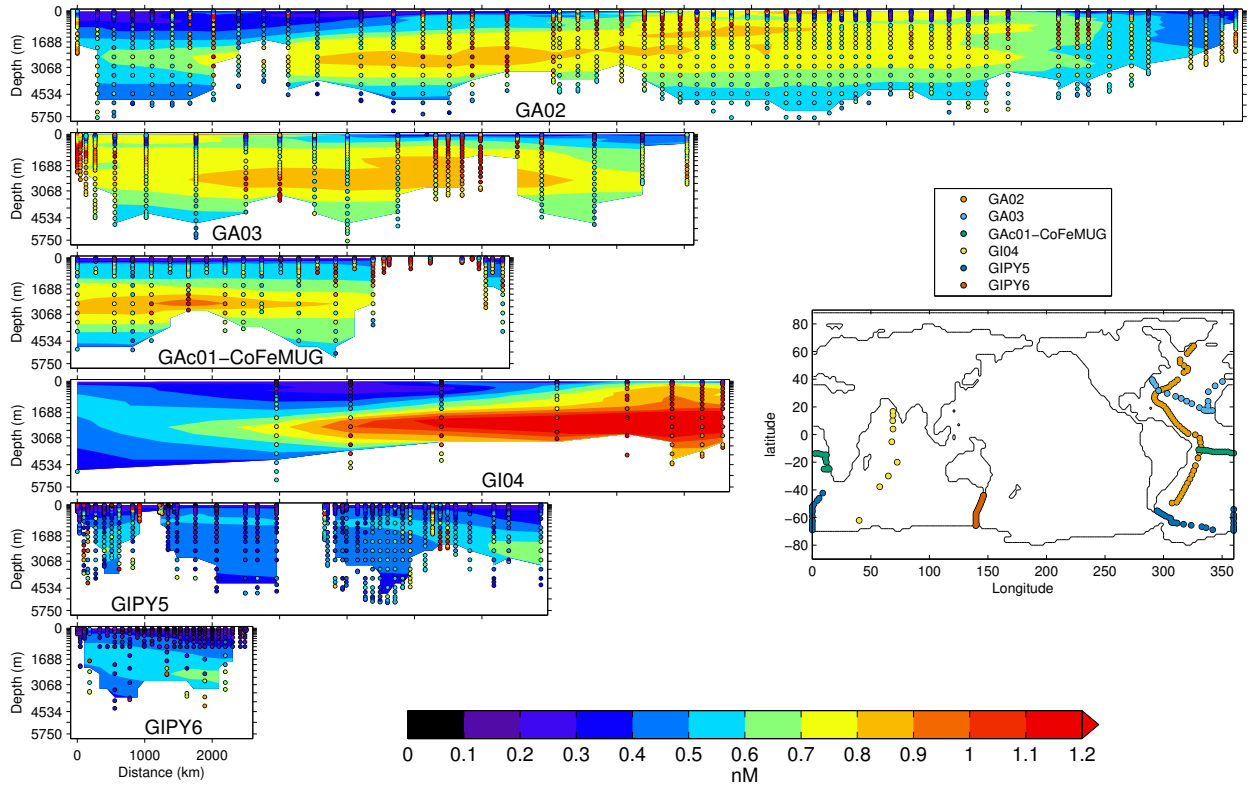
Figure G1 compares the main GEOTRACES transects included in the Intermediate Data Product with our typical state estimate. The coarse-resolution model captures the large-scale features, but localized high concentrations cannot be captured at our resolution.

15 We emphasize that a direct comparison with the GEOTRACES sections is subject to a number of caveats. We use a coarse-resolution, steady-state inverse model, while the GEOTRACES sections provide snapshots in space and time. Therefore, our model cannot capture any transient plumes (e.g., from an African dust event) that are highly localized and episodic. Our state estimates can only capture the long-term average concentrations, coarse-grained to  $2^\circ \times 2^\circ$  resolution. In terms of capturing hydrothermal plumes, we note that the data-assimilated circulation used only assimilated T, S,  $\text{PO}_4$ , and  $^{14}\text{C}$  but not  $^3\text{He}$ .  
20 Therefore, there are likely still some biases in the abyssal circulation (Holzer et al., 2017), which may contribute to the fact that we do not perfectly match the observed hydrothermal iron plumes.

However, what matters for our inverse model, whose biological production is mechanistically driven by dFe and macronutrient availability, is the large-scale transport into the euphotic zone, particularly the transport into iron-limited regions such as the Southern Ocean. We have no reason to think that this large-scale transport is suspect as evidenced by realistic large-scale patterns of production that are robust across our family of states with widely varying iron source strengths.

## Appendix H: **Iron source and sink patterns**

Figure H1 shows the vertically integrated sources of dFe with a logarithmic colour scale. The aeolian soluble iron deposition pattern is identical to that of the study of Luo et al. (2008), albeit limited to the oceans. The tropical Atlantic close to the Sahara, the Arabian Sea, and the Bay of Bengal are the regions of largest aeolian iron deposition. The hydrothermal iron sources follow the mid-ocean ridges with the pattern of the OCMIP protocol, but independently scaled for the Atlantic, Pacific, Indian, and Southern Ocean basins. Sedimentary iron is more intense where export production is large and in areas where oceans are

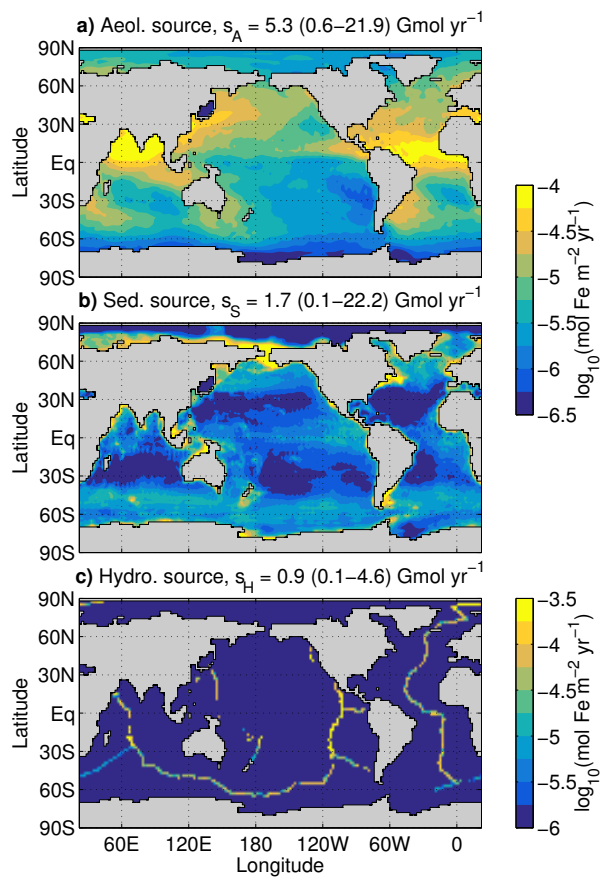


**Figure G1.** Dissolved iron concentrations of the typical state estimate (contours) compared to the GEOTRACES Intermediate Data Product (dots). The abscissa runs south to north or west to east along the transects (map).

shallower, because in both cases, a large flux of organic matter (or POP in our model) reaches the sediment. The sub-grid topography plays a significant role in the pattern of sedimentary iron, in particular for coastal regions and large underwater plateaus, e.g., near the Kerguelen islands or the Falkland islands. Because of unrealistic circulation features in the Sea of Japan, we zero all sources there consistent with our zeroing out production in the Sea of Japan.

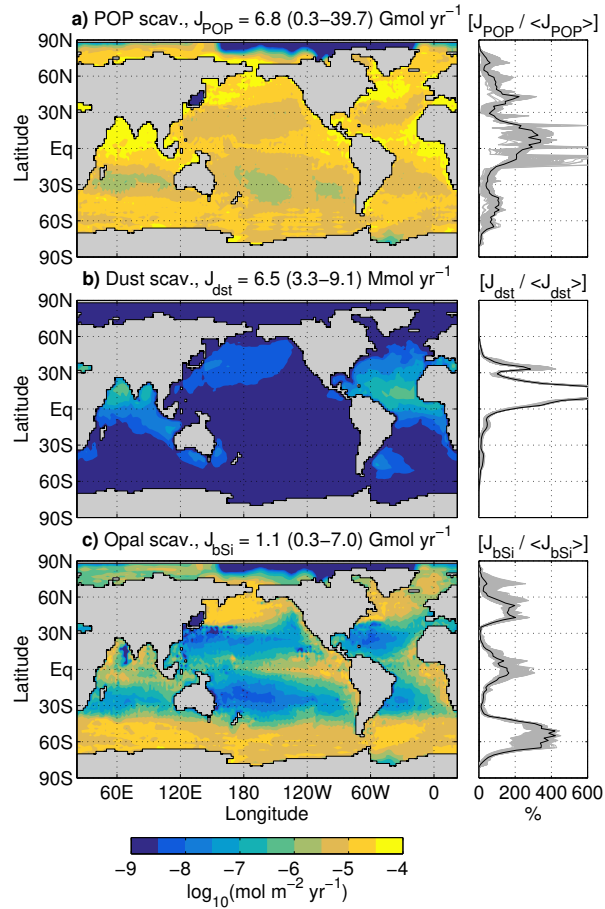
5 Figure H2 shows the vertically integrated sinks that balance the sources of Figure H1. The scavenging due to sinking mineral dust particles is about three orders of magnitude smaller than the sink due to organic and opal particle scavenging and could be neglected without changing our estimates appreciably. Although the pattern of the scavenging sinks has significant local variations among the members of our family of state estimates, the zonally averaged pattern (vertically integrated sink normalized by its global mean) is broadly similar across the family.

10 *Acknowledgements.* We thank François Primeau for making the data-assimilated circulation available to us, Natalie Mahowald for providing the aeolian iron flux estimate of Luo et al. (2008), and Marina Frants for discussions. This work was supported by ARC grant DP120100674



**Figure H1.** Vertically integrated sources of aeolian dFe (top, a), sedimentary dFe (middle, b), and hydrothermal dFe (bottom, c) for our typical state estimate. Note the logarithmic colour scales.

(MH). BP gratefully acknowledges scholarship support from the Government of Monaco, the Scientific Centre of Monaco, the Frères Louis et Max Principale Foundation, and the Cuomo Foundation.



**Figure H2.** Maps of the vertically integrated iron sinks of our typical state estimate due to (a) POP, (b) mineral dust, and (c) sinking opal particles. Plotted to the right are the zonal averages of the vertically integrated sinks normalized by their global mean, with each family member in grey and our typical state estimate in black.

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