

Interactive comment on “Phytoplankton growth responses to Asian dust additions in the Northwest Pacific Ocean versus the Yellow Sea” (doi: 10.5194/bg-2017-176) by Chao Zhang et al.

Dear Prof. Manmohan Sarin,

Enclosed is a copy of the revised manuscript, “Phytoplankton growth responses to Asian dust additions in the Northwest Pacific Ocean versus the Yellow Sea” (doi: 10.5194/bg-2017-176) by Chao Zhang et al.

In this revision, we have finished a revision by addressing all comments and suggestions from the three reviewers and your suggestions. In addition, we took this opportunity to thoroughly go through the whole MS to improve the presentation.

We thank you again for your consideration of this manuscript.

Sincerely,



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Chao Zhang et al.

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Additional changes we made to improve the presentation:

Line 52: Changed ‘worthy’ to ‘worthy of’;

Line 163: Changed ‘Selection of study period’ to ‘Protocol of data analysis’;

Line 200-201: Changed ‘For the S5 station in the YS, although the nutrient levels were comparable to those at S1, Chl *a* concentrations were as high as 2.74 $\mu\text{g L}^{-1}$, close to the conditions of spring bloom in the YS’ to ‘Although the nutrient levels at S5 in the YS were comparable to those at S1, the Chl *a* concentration of 2.74 $\mu\text{g L}^{-1}$ was close to the values during spring blooms in the YS’;

Line 204-205: Changed ‘S1 and S2 (< 20 cells mL^{-1}) to 79 cells mL^{-1} at S3 and 35 cells mL^{-1} at S4’ to ‘< 20 cells mL^{-1} at S1 and S2 to 35 cells mL^{-1} at S4 and 79 cells mL^{-1} at S3’;

Line 209-210: Changed ‘As shown in Table 3, the concentration of dissolved inorganic nitrogen (DIN, i.e. $\text{NO}_3^- + \text{NO}_2^- + \text{NH}_4^+$) in the AM-dust was 577 $\mu\text{mol g}^{-1}$, which is four times higher than that in the untreated dust.’ to ‘The concentration of dissolved inorganic nitrogen (DIN, i.e. $\text{NO}_3^- + \text{NO}_2^- + \text{NH}_4^+$) in the AM-dust was 577 $\mu\text{mol g}^{-1}$ (Table 3) and increased by a factor of four against that in the untreated dust’;

Line 213-214: Changed ‘The N:P ratio in the AM-dust was ~134, far greater than 16 (Redfield Ratio), and similar to those reported for Asian dust aerosols in previous studies’ to ‘The N:P ratio of ~134 in the AM-dust was far greater than 16 (Redfield Ratio), and similar to those in Asian dust aerosols previously reported’.

Line 275: Changed ‘sp.’ to ‘spp.’.

Referee #1

General comments:

*This study describes the results of several experiments in which surface plankton communities from the Yellow Sea and the NW Pacific ocean were amended with atmospheric dust and different nutrients added alone and in various combinations. The responses studied included the numerical abundance of diatoms and dinoflagellates, size-fractionated chl *a* concentration, and nutrient concentration. The main strength of the study is that parallel incubations, in which inorganic nutrients were added in different combinations, allowed the authors to gain insight into the causative mechanisms underlying the phytoplankton responses to dust. However, in some cases (as in the case of *P* availability, see below) the authors seem to over-interpret the available evidence. A limitation of the study is that only standing stocks were examined; no metabolic rate measurements were included, and therefore it is not possible to ascertain the dominant type of nutrient limitation (Blackman versus Liebig). The authors rely heavily on the use of chl *a* as a proxy for phytoplankton biomass. However, variability in C:Chl *a* ratios should be taken into account. The extent to which the simulated process of atmospheric transformation of dust yields materials that are realistic in terms of nutrient content and solubility should be discussed. Finally, some sections of the Discussion are speculative and based on tenuous assumptions. All these limitations should be addressed before publication is recommended in *Biogeosciences*. Some suggestions as to data presentation and analysis are also given below.*

Response:

We would like to thank the referee very much for the valuable comments which enabled us to improve the quality of the manuscript. We have revised the manuscript accordingly to address the comments.

Regarding concern over nutrient limitation status: The standing stocks such as Chl *a* and biomass have been widely used for ascertaining the dominant type of nutrient limitation in the ocean. However, indeed, we acknowledge that metabolic rate has a better representation than standing stocks in ascertaining nutrient limitation. We will add metabolic rate measurements for future incubation experiments.

Regarding C:Chl *a* ratios, we have illustrated that our results are based on Chl *a* and cell abundance rather than biomass, and also considered the uncertainty in the new manuscript (refer to Q4).

We have provided additional descriptions about the simulating processes in Text S1 (refer to Q1).

Regarding concerns over *P*, we have made a detailed response to Q5.

Specific comments

1. Simulation of atmospheric transformation of dust. How do these AM-dust materials compare, in terms of nutrient composition and solubility, with real dust samples collected in situ? This is critical to assess if the results observed are representative of real responses at sea. Table in supp. info. shows that N concentration is increased 4 orders of magnitude relative to N concentration in collected rain. Does this mean that the potential for nutrient supply is grossly overestimated in these artificially treated materials?

Response:

It is well known that the dust deposition can supply bioavailable nutrients such as N, P, and Fe to support the growth of phytoplankton. Therefore, we mainly focused on the effect of N, P, and Fe supplied by AM-dust on phytoplankton growth in this study. During a strong Asian dust event, the loading of inorganic nitrogen ($\text{NO}_3^- + \text{NH}_4^+$), soluble P and Fe in atmospheric particles collected in the Yellow Sea was $\sim 714 \mu\text{mol g}^{-1}$, $\sim 4.3 \mu\text{mol g}^{-1}$ and $\sim 550 \mu\text{g g}^{-1}$, respectively (Shi et al., 2012). The values in AM-dust used in this study are $577 \mu\text{mol g}^{-1}$, $4.3 \mu\text{mol g}^{-1}$ and $473 \mu\text{g g}^{-1}$, respectively (Table 3 in the origin version), which were highly comparable to those observed by Shi et al. (2012). The authors, however, agree that the loadings of these components in atmospheric dust particles could highly vary in different cases. For instance, the content of DIN and soluble P in Asian dust aerosols after a long-range transport varied over a range of two-three orders of magnitude, e.g., $11\text{-}3253 \mu\text{mol g}^{-1}$ for DIN and $0.26\text{-}18.86 \mu\text{mol g}^{-1}$ for soluble P (Liu et al., 2013; Meng, et al., 2016; Qi, et al., 2017). Besides, trace metals Fe, Mn, and Co were mainly originated from mineral aerosols, thus we can observe noticeably enhanced solubility of Fe, Mn, and Co in the AM-dust relative to untreated dust. The contents of other soluble trace metals such as Cu, Pb, and Zn in the dust were mainly affected by anthropogenic factor such as automobile exhaust and coal combustion, which were not reflected well in our study. To this point, we have added an illustration at the end of the conclusion. (Line 497-500)

We have added a detailed description of preparing AM-dust in the Text S1 to illustrate the difference in N content in the AM-dust and the collected rain (Line 755-765):

In this study, the aging process of dust followed Guieu's (2010) method and aimed at stimulating the cloud reaction between dust and synthetic evaporating cloud water. The pH around dust in the cloud process (i.e. mix with evaporating cloud water) was found to be as low as ~ 1 during their transport to the Yellow Sea (Meskhidze et al., 2003), whereas the typical pH in rainwater is 5 (Watanabe et al. 2001, Sasakawa and Uematsu, 2002, Wang et al. 2002, Sakihama et al. 2008, Zhang et al. 2011), meaning that a dilution by a factor of 10^4 . In consequent, in order to reproduce an evaporating cloud, we have used a concentration that is 10 000-fold larger in our experiments than the typical concentrations found in rainwater. Considering the typical concentrations of dust in rainwaters was 10 mg L^{-1} (Ridame et al., 2002), the dust loading in evaporating cloud water could reach 100 g L^{-1} . As a consequence,

all of the concentrations in evaporating cloud water were around 10000-fold larger (i.e. 4 orders of magnitude larger) than those in natural rainwater. Table S1 summarized the primary chemical composition of rains in the Eastern Asian regions and the evaporating cloud water used for our simulation.

Shi, J., Gao, H., Zhang, J., Tan, S., Ren, J., Liu, C., Liu, Y., and Yao, X.: Examination of causative link between a spring bloom and dry/wet deposition of Asian dust in the Yellow Sea, China, *Journal of Geophysical Research: Atmospheres*, 117, doi:10.1029/2012JD017983, 2012.

Liu, Y., Zhang, T., Shi, J., Gao, H., and Yao, X.: Responses of chlorophyll a to added nutrients, Asian dust, and rainwater in an oligotrophic zone of the Yellow Sea: Implications for promotion and inhibition effects in an incubation experiment, *Journal of Geophysical Research: Biogeosciences*, 118, 1763-1772, doi: 10.1002/2013JG002329, 2013.

Meng, X., Chen, Y., Wang, B., Ma, Q., and Wang, F.: Responses of phytoplankton community to the input of different aerosols in the East China Sea, *Geophysical Research Letters*, 43, 7081-7088, doi: 10.1002/2016GL069068, 2016.

Qi, J., Zhang, R., Chen, X., Lin, X., Gao, H., and Liu, R.: The concentration, source apportionment and deposition flux of atmospheric particulate inorganic nitrogen during dust events. *Atmospheric Chemistry & Physics. Discussions*, doi: 10.5194/acp-2016-1183, 2017.

Guieu, C., Dulac, F., Desboeufs, K., Wagener, T., Pulido-Villena, E., Grisoni, J.-M., Louis, F., Ridame, C., Blain, S., and Brunet, C.: Large clean mesocosms and simulated dust deposition: a new methodology to investigate responses of marine oligotrophic ecosystems to atmospheric inputs, *Biogeosciences*, 7, 2765-2784, doi: 10.5194/bg-7-2765-2010, 2010.

Meskhidze, N., Chameides, W. L., Nenes, A., and Chen, G.: Iron mobilization in mineral dust: Can anthropogenic SO₂ emissions affect ocean productivity?. *Geophysical Research Letters*, 30(21), doi: 10.1029/2003GL018035, 2003.

Watanabe, K., Ishizaka, Y., and Takenaka, C.: Chemical characteristics of cloud water over the Japan Sea and the Northwestern Pacific Ocean near the central part of Japan: airborne measurements. *Atmospheric Environment*, 35(4), 645-655, doi: 10.1016/S1352-2310(00)00358-7, 2001.

Sasakawa, M., and Uematsu, M.: Chemical composition of aerosol, sea fog, and rainwater in the marine boundary layer of the northwestern North Pacific and its marginal seas. *Journal of Geophysical Research: Atmospheres*, 107(D24), doi: 10.1029/2001JD001004, 2002.

Wang, Z., Akimoto, H., and Uno, I.: Neutralization of soil aerosol and its impact on the distribution of acid rain over east Asia: Observations and model results. *Journal of Geophysical Research: Atmospheres*, 107(D19), doi: 10.1029/2001JD001040, 2002.

Sakihama, H., Ishiki, M., and Tokuyama, A.: Chemical characteristics of precipitation in Okinawa Island, Japan. *Atmospheric Environment*, 42(10), 2320-2335, doi: 10.1016/j.atmosenv.2007.12.026, 2008.

Zhang, J., Zhang, G. S., Bi, Y. F., and Liu, S. M.: Nitrogen species in rainwater and aerosols of the Yellow and East China seas: Effects of the East Asian monsoon and anthropogenic emissions and relevance for the NW Pacific

Ocean. *Global Biogeochemical Cycles*, 25(3), doi: 10.1029/2010GB003896, 2011.

Ridame, C., and Guieu, C.: Saharan input of phosphate to the oligotrophic water of the open western Mediterranean Sea. *Limnology and Oceanography*, 47(3), 856-869, doi: 10.4319/lo.2002.47.3.0856, 2002.

2. *Section 3.2. This section should present first the changes in nutrient concentration, and then those of chl a concentration. In both cases, the actual increases (absolute values) should be described (e.g. nutrient or chl a concentration increased by xx umol/L or ug/L), rather than just the relative increases (xx-fold). It is important to describe the chl a responses in terms of absolute value of increase, so that they can be compared with the amount of nutrient released from the dust or provided by the nutrient amendments.*

Response:

Agree. We have revised accordingly through the Section 3.2. (Line 219-249)

3. *Conversion efficiency index. This index should be described in the Methods section.*

Response:

Agree. It has been moved to the Materials and Methods Section 2.6. (Line 170-181)

4. *It is unclear why the index is formulated in this way. Why not use just final minus initial chl a concentration, as is done for N? It does not make sense to sum consecutive differences over time in chl a concentration between treatments and control. In addition, the index has a potential flaw, because C:Chl a ratios are bound to be different in the different sites (due, for instance, to differences in nutrient and/or light availability). So the same response in terms of % increase in biomass (carbon) will yield higher chl a concentration (in absolute values), and thus higher conversion efficiency, in waters with low phytoplankton C:Chl a values. The limitations of using Chl a as a proxy for biomass should be acknowledged and discussed. Finally, when reporting the values of this index in the text, its units should be indicated.*

Response:

In the revision, we have clarified the net conversion efficiency index (NCEI) proposed in this study to be an approximate estimation for the utilization of N for the growth of phytoplankton. Therefore, the capacity to synthesize Chl *a* per unit concentration of nitrogen (N) in different treatments can be compared. We agree that the sum consecutive differences over time in Chl *a* concentration between treatments and control will lead to an overestimation of the real net conversion efficiency because of the accumulation effect. Theoretically, the use of the maximum difference will lead to an underestimation of the real net conversion efficiency because of degradation of Chl *a* in the growth period. The real net conversion efficiency should be between those calculated by the two approaches. This has been clarified in the revision. (Line 179-181)

We agree that C:Chl *a* ratios can be different at the different sites and Chl *a* regarding as a proxy for biomass is not appropriate. Thus, we have illustrated that our results are based on Chl *a* and cell abundance rather than biomass, and also added a discussion illustrating the varying C:Chl *a* ratios at different sites in the revision. (Line 373-376)

The unit of NCEI in the text have been added in the revision. (Line 340-363)

5. Section 4.3. This section is speculative and difficult to follow. It is unclear how the 'increase in bioavailable P concentration following AM-dust addition' has been identified. The relationship between N:P ratios in supply vs demand is tentative at best, since actual supply N:P ratios were not measured. The paragraph on lines 395-406 starts with an untenable assumption, namely that 'C_N:P in AM-dust treatments is equal to that in N treatments'. To the extent that dust additions and N additions create distinct nutrient environments, it is most unlikely that consumption N:P ratios will be the same. In fact, the previous paragraph has argued that consumption N:P ratio is lower in dust treatments than in N treatments. Thus the subsequent calculations and conclusions have no use. This sub-section (l. 395-406) should be deleted. In the subsequent paragraph, the basis for the need for additional P supply is unclear.

Response:

We have determined the soluble nutrients leached from AM-dust in the seawater. The details can be seen in Methods Section 2.4 (Line 139-146). The content of bioavailable nutrients from AM-dust has been listed in Table 3. As the added concentration of AM-dust was 2 mg L⁻¹, thus, we can get the theoretical amounts of N and P nutrients supplied by AM-dust in the incubation system. The time interval of adding materials to the incubation bottles and sampling seawater for nutrient measurement was 1-2 hr. Microbial uptake, scavenging by cell surface and bottle wall, etc., possibly decreased the concentrations of nutrients at the 1-2 hr, leading to the measured values smaller than the theoretical values, which has been proved by the previous incubation experiments (Liu et al., 2013). When the concentrations of NO₃⁻+NO₂⁻ and PO₄³⁻ decreased in the seawater, those absorbed by cell surface and bottle wall had the potential to be released into the solution again for reaching equilibrium. Therefore, the calculated theoretical amounts of N and P nutrients supplied by AM-dust in the incubation system is close to the actual values relative to those measured directly on day 0.

We have corrected the inappropriate expression of 'C_{N:P} in AM-dust treatments is equal to that in N treatments' and changed the equation (2) into inequation (2). (Line 441)

The last paragraph in the Discussion Section 4.3 is particularly important because it provides a new insight to analyze the budget of bioavailable P in the AM-dust addition incubation experiments. Unfortunately, the original presentation seemed to be unclear and didn't service the target well. We have rewritten this paragraph to make our thoughts understandable. (Line 453-466)

Liu, Y., Zhang, T., Shi, J., Gao, H., and Yao, X.: Responses of chlorophyll a to added nutrients, Asian dust, and rainwater in an oligotrophic zone of the Yellow Sea: Implications for promotion and inhibition effects in an incubation experiment, *Journal of Geophysical Research: Biogeosciences*, 118, 1763-1772, doi: 10.1002/2013JG002329, 2013.

Minor comments

1. lines 53-54: Rewrite sentence: *'The N:P ratio of dust deposition is much higher than the Redfield ratio (N:P=16)...'*

Response:

Corrected. (Line 54)

2. line 110. *Were attenuation filters used? PAR levels should be given.*

Response:

We did not use attenuation filters in this study. The surface seawater (2-5 m) was collected and incubated under natural light (Line 110). In this study, we primarily made a comparison between groups at each station rather than that between stations. Since the experiments at the one station experienced the same conditions, the effect of different PAR levels on our comparison for the results could be reduced as much as possible. We thank the comment and will add the attenuation filters for future incubation experiments.

3. line 139. *The ultrasonic method should be described briefly. The use of ultrasounds maximises the extraction of nutrients but it probably overestimates the amount of nutrients that is actually released in real conditions at sea.*

Response:

We have made a brief description of the ultrasonic method in the revision. (Line 139-141)

In general, more than 30 minute of ultra-sonication treatment will increase the temperature and may enhance the leaching of nutrients and thus overestimate. In our study, the time duration used for ultra-sonication is 30 minute. We used the ice pack to keep the temperature of water bath stable at $\sim 0^{\circ}\text{C}$. Please also refer to 'Specific comments No. 3' for Referee #3.

4. line 141. *Re-write sentence, '...and filtrates were stored...'*

Response:

Corrected. (Line 142)

5. line 153: *'...enumeration of...'*

Response:

Corrected. (Line 153)

6. line 171: *delete 'evidently'.*

Response:

Corrected. (Line 197)

7. line 174: *'trophic level' means something else. Replace by appropriate phrase.*

Response:

We have replaced 'trophic level' with 'trophic state'. (Line 200)

8. line 185: *Why is P gained during the treatment?*

Response:

The acid process can enhance the solubility of P in the mineral dust.

9. line 189 *Here cite Fig. 3, otherwise the reader does not know where is that increase reported. Tables do not report the nutrient increases observed in the treatments.*

Response:

Corrected. (Line 215)

10. line 194 and elsewhere (including Fig. legends). *The phrase 'successive increase' should be omitted. Sentence should read simply: 'During the incubation...'*

Response:

The duration of the incubation experiment in this study is 9~10 days, but we mainly analysed the data on the initial 3-5 days, in which the Chl *a* concentration showed a successive increase. Thus, we used the 'successive increase' to distinguish the initial 3-5 days from the whole incubation period (9-10 days). Please refer to the Methods Section 2.6. (Line 164-168)

11. line 271. *Remove 'certain amount of'.*

Response:

Corrected. (Line 305)

12. line 300. *This sentence seems to assume that all N present in the dust becomes bioavailable, because the concentrations referred to are those given in Table 3 (which correspond to concentrations in the dust, not in seawater).*

Response:

We have rewritten the sentence. The contents of N and P in the untreated and AM-dust listed in Table 3 were determined in seawater. The details can be seen in the description of 'ultrasonic bath'. Please also refer to 'Minor comments No. 15' for Referee #1. (Line 328)

13. line 448-449. *Here the authors are deriving biogeochemical conclusions on the functioning of the biological pump, but their data consider just phytoplankton. Without information on how the metabolic activity of heterotrophs, bacteria in particular, changes in response to nutrient/dust additions, the ultimate effect on the biological pump*

remains unknown.

Response:

Thank you for your thoughtful suggestion. We have rewritten the sentence and illustrated that the enhanced biological pump was only a possibility. (Line 495-497)

14. Table 1. Silicate measurements are missing – they would have been helpful to constrain the stoichiometry of diatom blooms in response to nutrient/dust amendments.

Response:

In the original experimental settings, we aimed at exploring the relationship between phytoplankton growth and supplied bioavailable nutrients (N, P, and Fe) by AM-dust additions. Thus, we did not determine silicate (Si) concentrations. For the Yellow Sea, Si is generally not a limiting nutrient for the growth of phytoplankton in spring because of the influence of riverine input (Wang et al., 2003). Si may become a limiting nutrient during the diatom blooms in the open ocean of the northwest Pacific, but it will not influence the result of this study, which was mainly concerned with N, P, and Fe nutrients. We thank the comment and will add the measurement of silicate for future incubation experiments.

Wang, B. D., Wang, X. L., and Zhan, R. Nutrient conditions in the Yellow Sea and the East China Sea. *Estuarine, Coastal and Shelf Science*, 58, 127-136, doi: 10.1016/S0272-7714(03)00067-2, 2003.

15. Table 3. The data labelled ‘increased concentrations’ are theoretical or expected concentrations, assuming 100% of the nutrients in the dust becomes dissolved. This should be explicitly acknowledged in the Table legend. Is there any evidence to support the tenet that, upon dust deposition onto the ocean’s surface, all nutrients become dissolved and bioavailable?

Response:

We have added ‘theoretically’ in the Table legend. (Line 696)

The $\text{NO}_3^- + \text{NO}_2^-$ in the AM-dust would dissolve almost thoroughly once exposing to the filtered seawater (Ridame et al., 2014) while the dissolution of PO_4^{3-} would continue over multiple days because the dissolution of less labile (but still soluble) P compounds would take some time (generally lower than 72 hr) to dissolve in seawater (Mackey et al., 2012). Over the duration of the incubation experiments, phytoplankton in the incubated bottles can absorb the soluble $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} from AM-dust if they need.

Ridame, C., Dekaezemacker, J., Guieu, C., Bonnet, S., L’Helguen, S., and Malien, F.: Contrasted Saharan dust events in LNLC environments: impact on nutrient dynamics and primary production, *Biogeosciences*, 11, 4783-4800, doi: 10.5194/bg-11-4783-2014, 2014.

Mackey, K. R., Roberts, K., Lomas, M. W., Saito, M. A., Post, A. F., and Paytan, A.: Enhanced solubility and ecological impact of atmospheric phosphorus deposition upon extended seawater exposure, *Environmental science*

& technology, 46, 10438-10446, doi: 10.1021/es3007996, 2012.

16. *Figure legends: The phrase 'successive increase' is awkward. Delete in all fig. legends. It should be simply: 'changes in xxxx during the incubation period at each station'.*

Response:

Please see 'Minor comments No.10'.

17. *Fig. 2. Y-axis intervals should be regular (e.g., 0.5 or 1.0 ug/L) and consistent in all plots. Minor ticks should be included, to help the reader ascertain the magnitude of responses.*

Response:

Corrected. (Line 715)

18. *Fig. 3. Symbol for control should be more visible (it is often masked by other symbols).*

Response:

Corrected. (Line 710)

19. *Fig. 5. Revise species names spelling (e.g. Skeletonema). Species names and genera should be written in italics (but not 'spp.').*

Response:

Corrected. (Line 725)

20. *Figs. 6 and 7. The index values shown here result from the subtraction and division of variables measured independently, each with its own error. Therefore the error bars shown should be computed using the error propagation formulae for addition and division.*

Response:

Thanks for the valuable suggestion. We have used the error propagation formulae to calculate the error bars and clarified in the revision. Please see Figs. 6 and 7. (Line 340-363, 739-741)

21. *Fig. 8. These N:P ratios should be defined in the Methods section. Strictly speaking, the N:P supply ratio is not known, since no solubility experiments were conducted.*

Response:

We have moved the definitions of N:P ratios to the Methods Section 2.6. The dissolved concentrations of N and P from AM-dust in the seawater have been determined in the laboratory (Table 3). Please also refer to 'Minor comments No.12 and 15'. (Line 183-186)

Referee #2

General comments

1. *Biological productivity of the open ocean regions, especially oligotrophic parts, have attracted the attention of several researchers. Our present understanding of the role of mineral dust in enhancing the primary productivity of the oligotrophic ocean by supplying bio-available nutrients is still in its infancy. The present manuscript aims at advancing such an understanding of enhancement of phytoplankton growth and the phytoplankton community structure by the nutrient supply from the Asian in the subtropical gyre, Kuroshio Extension of Northwest Pacifica Ocean and Yellow Sea region. The authors attempt this by incubation experiment onboard R/V Dongfanghong II during spring. For incubation experiment, the mineral dust collected from Gobi Desert region was artificially modified and phytoplankton assemblages from subtropical gyre, Northwest Pacific ocean and Yellow Sea region were used. Each of the five microcosm experiment lasted for 9 to 10 days. Using a net conversion efficiency index, proposed by the authors, of nitrogen conversion to chlorophyll a the authors explore the role of bio-available nutrients from mineral dust in the primary production at the above mentioned three regions.*

2. *The subject matter of the manuscript addresses an important aspect of phytoplankton growth by “bio-available nutrients” from “treated soil from Gobi desert” which is “expected to” simulate the natural mineral dust. Though there are several concerns that have been listed under specific comments, in my opinion, the results are publishable; but only after adequately addressing the concerns.*

Response:

We greatly appreciate the referee for the thoughtful comments and have revised our manuscript accordingly.

Major concerns:

1. *The major concern is that the authors have not succeeded in unambiguously resolving the issue of quantification of bioavailability of nutrients form artificially modified mineral dust, especially phosphorous, for phytoplankton growth, which is the central theme of the manuscript. See for example, lines 334-341. The ambiguity regarding the ‘missing N and P’.*

Response:

We are sorry that the word of ‘missing’ in the manuscript confused the referee. The ‘missing N and P’ and ‘missing parts’ in the text meant the N and P might be adsorbed on the bottle walls, suspended particles, and phytoplankton in the solution, while ‘missing sources’ meant there might be other sources to supply bioavailable P. Thus, we will change ‘missing sources’ to ‘other sources’ to express them clearly (Line 482). Considering the comment and the similar comments from other two referees, we have rewritten the part related to the budget of P in incubation experiments. Please see our revised manuscript. (Line 419-466)

2. *How would the authors differentiate the phytoplankton growth-response due to N/P/Fe and that due to Mn and Zn (see for example, Saito et al., 2008; Sunda 2012).*

Response:

We have replaced “Zn” with “Co” in the text because the concentration of suite of trace metals in AM-dust are very low except Fe, Mn and Co (Table 3). Please also see Specific Comments No. 4 for Referee #3. (Line 359)

We used the various nutrient addition experiments to illustrate which nutrients (i.e. N, P, and Fe) limit phytoplankton growth. As we did not conduct Mn or Co addition incubation experiments and determine the concentration of Mn and Co in the seawater, it is difficult for us to accurately see whether phytoplankton were under limitation of Mn and/or Co. However, through making a comparison of the net conversion efficiency index (NCEI) in various nutrient addition treatments, we can roughly determine whether there were other nutrients except N, P, and Fe dissolved from AM-dust to stimulate phytoplankton growth. For instance, the NCEI value in AM-dust treatment was significantly higher than that in N+P+Fe treatments, indicating the N, P, and Fe supplied by AM-dust cannot explain the corresponding phytoplankton growth. Besides, As reported in literature, trace metals such as Mn and Co can affect phytoplankton growth rate (Sunda, 2012), which might lead to higher NCEI. The effect of Mn and Co on phytoplankton growth is just a hypothesis, which have been clarified in the revision. (Line 359-360)

Sunda, W.: Feedback interactions between trace metal nutrients and phytoplankton in the ocean, *Frontiers in microbiology*, 3, 204, doi: 10.3389/fmicb.2012.00204, 2012.

3. *The authors do not observe any community shift in the phytoplankton in their study. There is no discussion on this aspect and authors need to address this.*

Response:

We thank the comment and have added some discussions about community shift of phytoplankton in Discussion Section 4.2, as well as relative abundance in Fig. 5. (Line 402-412, Line 725)

4. *Based on the information provided, it is hard to see how closely the artificially modified soil collected from Gobi desert mimics the nature. Some more robust information on the atmospheric (chemical) processing of the mineral dust during the long-range transport from source to the proposed study site during spring is needed along with the upper air wind vectors. What are the chemical constituents of such atmospheric processed mineral dust in presence of anthropogenic aerosol is not clear.*

Response:

It is well known that the dust deposition can supply bioavailable nutrients such as N, P, and Fe to support the growth of phytoplankton. Therefore, we mainly focused on the effect of N, P, and Fe supplied by AM-dust on phytoplankton growth in this study. During a strong Asian dust event, the loading of inorganic nitrogen ($\text{NO}_3^- + \text{NH}_4^+$), soluble P and Fe in atmospheric particles collected in the Yellow Sea was $\sim 714 \mu\text{mol g}^{-1}$, $\sim 4.3 \mu\text{mol g}^{-1}$ and $\sim 550 \mu\text{g g}^{-1}$,

respectively (Shi et al., 2012). The values in AM-dust used in this study are $577 \mu\text{mol g}^{-1}$, $4.3 \mu\text{mol g}^{-1}$ and $473 \mu\text{g g}^{-1}$, respectively (Table 3 in the origin version), which were highly comparable to those observed by Shi et al. (2012). The authors, however, agree that the loadings of these components in atmospheric dust particles could highly vary in different cases. For instance, the content of DIN and soluble P in Asian dust aerosols after a long-range transport varied over a range of two-three orders of magnitude, e.g., $11\text{-}3253 \mu\text{mol g}^{-1}$ for DIN and $0.26\text{-}18.86 \mu\text{mol g}^{-1}$ for soluble P (Liu et al., 2013; Meng, et al., 2016; Qi, et al., 2017). Besides, trace metals Fe, Mn, and Co were mainly originated from mineral aerosols, thus we can observe noticeably enhanced solubility of Fe, Mn, and Co in the AM-dust relative to untreated dust. The contents of other soluble trace metals such as Cu, Pb, and Zn in the dust were mainly affected by anthropogenic factor such as automobile exhaust and coal combustion, which were not reflected well in our study. To this point, we have added an illustration at the end of the conclusion. (Line 497-500)

We have also added a description of preparing AM-dust in the Text S1 to illustrate atmospheric processing of the mineral dust during the long-range transport (Line 755-765):

In this study, the aging process of dust followed Guieu's (2010) method and aimed at stimulating the cloud reaction between dust and synthetic evaporating cloud water. The pH around dust in the cloud process (i.e. mix with evaporating cloud water) was found to be as low as ~ 1 during their transport to the Yellow Sea (Meskhidze et al., 2003), whereas the typical pH in rainwater is 5 (Watanabe et al. 2001, Sasakawa and Uematsu, 2002, Wang et al. 2002, Sakihama et al. 2008, Zhang et al. 2011), meaning that a dilution by a factor of 10^4 . In consequent, in order to reproduce an evaporating cloud, we have used a concentration that is 10 000-fold larger in our experiments than the typical concentrations found in rainwater. Considering the typical concentrations of dust in rainwaters was 10 mg L^{-1} (Ridame et al., 2002), the dust loading in evaporating cloud water could reach 100 g L^{-1} . As a consequence, all of the concentrations in evaporating cloud water were around 10000-fold larger (i.e. 4 orders of magnitude larger) than those in natural rainwater. Table S1 summarized the primary chemical composition of rains in the Eastern Asian regions and the evaporating cloud water used for our simulation.

Shi, J., Gao, H., Zhang, J., Tan, S., Ren, J., Liu, C., Liu, Y., and Yao, X.: Examination of causative link between a spring bloom and dry/wet deposition of Asian dust in the Yellow Sea, China, *Journal of Geophysical Research: Atmospheres*, 117, doi:10.1029/2012JD017983, 2012.

Liu, Y., Zhang, T., Shi, J., Gao, H., and Yao, X.: Responses of chlorophyll a to added nutrients, Asian dust, and rainwater in an oligotrophic zone of the Yellow Sea: Implications for promotion and inhibition effects in an incubation experiment, *Journal of Geophysical Research: Biogeosciences*, 118, 1763-1772, doi: 10.1002/2013JG002329, 2013.

Meng, X., Chen, Y., Wang, B., Ma, Q., and Wang, F.: Responses of phytoplankton community to the input of different aerosols in the East China Sea, *Geophysical Research Letters*, 43, 7081-7088, doi: 10.1002/2016GL069068, 2016.

Qi, J., Zhang, R., Chen, X., Lin, X., Gao, H., and Liu, R.: The concentration, source apportionment and deposition

flux of atmospheric particulate inorganic nitrogen during dust events. *Atmospheric Chemistry & Physics. Discussions*, doi: 10.5194/acp-2016-1183, 2017.

Guieu, C., Dulac, F., Desboeufs, K., Wagener, T., Pulido-Villena, E., Grisoni, J.-M., Louis, F., Ridame, C., Blain, S., and Brunet, C.: Large clean mesocosms and simulated dust deposition: a new methodology to investigate responses of marine oligotrophic ecosystems to atmospheric inputs, *Biogeosciences*, 7, 2765-2784, doi: 10.5194/bg-7-2765-2010, 2010.

Meskhidze, N., Chameides, W. L., Nenes, A., and Chen, G.: Iron mobilization in mineral dust: Can anthropogenic SO₂ emissions affect ocean productivity?. *Geophysical Research Letters*, 30(21), doi: 10.1029/2003GL018035, 2003.

Watanabe, K., Ishizaka, Y., and Takenaka, C.: Chemical characteristics of cloud water over the Japan Sea and the Northwestern Pacific Ocean near the central part of Japan: airborne measurements. *Atmospheric Environment*, 35(4), 645-655, doi: 10.1016/S1352-2310(00)00358-7, 2001.

Sasakawa, M., and Uematsu, M.: Chemical composition of aerosol, sea fog, and rainwater in the marine boundary layer of the northwestern North Pacific and its marginal seas. *Journal of Geophysical Research: Atmospheres*, 107(D24), doi: 10.1029/2001JD001004, 2002.

Wang, Z., Akimoto, H., and Uno, I.: Neutralization of soil aerosol and its impact on the distribution of acid rain over east Asia: Observations and model results. *Journal of Geophysical Research: Atmospheres*, 107(D19), doi: 10.1029/2001JD001040, 2002.

Sakihama, H., Ishiki, M., and Tokuyama, A.: Chemical characteristics of precipitation in Okinawa Island, Japan. *Atmospheric Environment*, 42(10), 2320-2335, doi: 10.1016/j.atmosenv.2007.12.026, 2008.

Zhang, J., Zhang, G. S., Bi, Y. F., and Liu, S. M.: Nitrogen species in rainwater and aerosols of the Yellow and East China seas: Effects of the East Asian monsoon and anthropogenic emissions and relevance for the NW Pacific Ocean. *Global Biogeochemical Cycles*, 25(3), doi: 10.1029/2010GB003896, 2011.

Ridame, C., and Guieu, C.: Saharan input of phosphate to the oligotrophic water of the open western Mediterranean Sea. *Limnology and Oceanography*, 47(3), 856-869, doi: 10.4319/lo.2002.47.3.0856, 2002.

5. It is not clear from the manuscript, whether the ocean atmospheric conditions during the 3-month period (March-May 2014) at each of the sampling locations could be considered as a part of the same season where ocean and atmosphere represents similar conditions. The data from the Table 1 do not support this. For example, the average temperature (is it SST?) at S4 is quite different from the rest of the stations, which was sampled in May 2014. Similar, the MLD also is quite different. The authors need to address these issues.

Response:

We are sorry that we made a mistake of sea surface temperature (SST) and mixed layer depth (MLD) at S5 and have corrected them in the revision. (Line 697)

As a general idea on season, those stations were sampled almost in the same season. It is the location not the time that induces a low temperature at S4 and S5 but a high temperature at the other three stations (S1-S3).

Stations S4 and S5 are located in a shallow coastal area with a depth less than 100 m. The winter cooling induces a vertically homogenous water temperature with a value around 12 °C (Yeh & Kim, 2010). The heating process usually starts in early April and there is no much time to increase the water temperature before our observations. Consequently, we observed a low temperature at the two stations in April and May respectively (the difference between S4 and S5 is also caused by location: S4 is north to S5 and therefore has a lower temperature than S5 in winter; S4 is close to the Bohai Strait and therefore has a strong vertical mixing than S5 that prevents increasing of SST after April.)

On the other hand, stations S1-S3 are located at the open water with a depth larger than 2000 m. The winter cooling cannot easily reduce its water temperature because of the heat supplying from bottom of mixed layer (the sea bed in coastal water has not such function). In addition, the *Kuroshio* and mesoscale eddies also bring warm water into the observation area. Such lateral supply of heat can also prevent the decreasing of surface temperature in winter. This is why we observed a temperature around 19 °C.

The MLD depends on wind stresses and heat fluxes over the sea. The surrounding land induces weaker winds over stations S4 and S5 than over stations S1-S3. Consequently, the mixed layer is shallower at stations S4 and S5 than at station S1-S3. In addition, the larger loss of latent and sensible heats over stations S1-S3 (due to higher SST) than over stations S4 and S5 is also likely responsible for the deeper mixed layer at stations S1-S3 than at station S4 and S5.

In this study, we primarily made a comparison between groups at each station rather than that between stations. Since the experiments at the one station experienced the same conditions, the different oceanic and atmospheric conditions between stations would not influence our comparison for the results.

We have added a description of SST and MLD conditions at the sampling locations in Results Section 3.1. (Line 190-195)

Yeh S W and Kim C H: Recent warming in the Yellow/East China Sea during winter and the associated atmospheric circulation, *Continental Shelf Research*, 30(13): 1428-1434, doi: 10.1016/j.csr.2010.05.002, 2010.

7. *Most of the results obtained from the incubation experiment, such as co-limitation of nutrients, response of phytoplankton biomass and structure, are largely known as could be seen from the literature cited in the manuscript. For example, co-limitation in the south China Sea and its response to aeolian input (Gao et al., 2012), Fu et al. (2009) on N:P ratios during spring, Fu et al (2009) study on the phytoplankton biomass and structure in South China Sea. Nishibe et al. (2015) work in the Kuroshio Extension during spring. See also under Minor comments (8).*

Response:

We recognize that previous studies have identified the nutrient limitation status in the study area. We are content that our nutrient addition results are consistent with theirs, as this confirms that the conditions we were studying were representative to the regions.

The similar results compared to other studies aimed at illustrating the following ideas:

The main strength of the study is to gain insight into the causative mechanisms underlying the phytoplankton responses to dust through parallel incubations, in which inorganic nutrients were added in different combinations. The added amount of inorganic nutrients (e.g. N: 2 $\mu\text{mol L}^{-1}$, P: 0.2 $\mu\text{mol L}^{-1}$, Fe: 2 nmol L^{-1}) was not equal to that dissolved from AM-dust, which leads to incomparable increases of Chl *a* concentrations in AM-dust and various nutrient treatments. Thus, we proposed the conversion efficiency index (NCEI) to quantify the role of N, P, and Fe dissolved from AM-dust played in stimulating phytoplankton growth, based on making a comparison with the results concluded from various nutrient addition treatments. Finally, we used the correlation of $S_{\text{N:P}}$ and $C_{\text{N:P}}$ to highlight increased bioavailability of P in AM-dust addition experiments.

Minor concerns:

8. *Lines 310-311: Authors need to at least briefly state what are those “Complex hydrographic conditions”.*

Response:

We have added a brief state of complex hydrographic conditions in the revision: “Complex hydrographic conditions e.g., various riverine inputs and atmospheric deposition, create a large spatiotemporal variation in nutrient concentrations in the YS”. (Line 313)

9. *Lines 339-342: This is purely speculative and needs further substantiation.*

Response:

We have rewritten this sentence to express it clearly. (Line 328-337)

In the field incubation experiments, the time interval of adding materials (e.g. AM-dust and various nutrients) into the incubation bottles and seawater sampling for nutrient measurement (i.e. measurement on day 0) was around 1-2 hr. During this period, microbial uptake, scavenging by cell surface and bottle wall are all possible to influence the measurement of added N and P nutrients from AM-dust, which lead to the concentrations determined on day 0 were lower than the theoretical adding amounts. When the concentrations of $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} decreased in the seawater, those absorbed by cell surface and bottle wall had the potential to be released into the solution again for reaching equilibrium.

10. Lines 307-308: *So what is new/different from the work of Nishibe et al. (2015).*

Response:

Please see Major concerns No.7.

11. Line 143: *Expand SPSS*

Response:

Corrected. (Line 129)

12. Table 3 : *2ns foot note (b) is missing in the Table*

Response:

Corrected. (Line 700)

13. *Also explain “E-3, E-4”*

Response:

We have added an explanation of ‘E-3’ and ‘E-4’ in Table 3. (Line 700)

Referee #3

General comments

This paper present a set of microcosm experiments performed on-board using sea water collected from two distinct oceanic regime: 1) Oligotrophic waters (Northwest Pacific Ocean) and 2) nutrient rich waters (Yellow Sea), to understand the impact of atmospheric dust (or processed dust) and other nutrients (N, P, Fe e.t.c) on the phytoplankton productivity in terms of increase in chlorophyll a (Chl a) and abundance of phytoplankton in various size fractions (e.g. micro, nano, pico e.t.c). The artificially modified atmospheric dust (AM-dust) is prepared using surface soil collected from Gobi Desert. The set of experiments lasted for 9-10 days and clearly indicate an overall increase in Chl a concentrations due to addition of various combination of nutrients including AM-dust. Authors have proposed a “new” net conversion efficiency index (NCEI) to better understand the impact of specific nutrients (N, P, Fe, and AM-dust) on primary productivity at the sampled locations. The presented case study makes an important contribution towards improving our understanding on impact of aeolian deposition (external source of nutrients) on productivity and ocean biogeochemistry. The results obtained from set of experiments are discussed well, manuscript is easy to read (except few sections, see in specific comments) and should be of great interest to the Biogeoscience community. So, I recommend this paper for publication in Biogeosciences, but after addressing

some of the concerns detailed in specific comments.

Response:

We greatly appreciate the referee for the constructive comments and have revised our manuscript accordingly.

Specific comments:

1 Section 2.1: What is the size distribution of soil samples used for preparing AM-dust? This information is important because if, majority of collected soil particles are in coarse fraction (e.g. more than 30 microns), most of them gets deposited at the source region and hardly get transported to the Pacific. So, the soil used for AM-dust preparation is not at all a representative one undergoing long-range transport and depositing on surface waters. The fine fraction (less than 5 microns) or typically clay fraction of the soil is a more representative dust which can be artificially processed to mimic the processed aeolian dust.

Response:

Recently, the transport routes of Asian dust move the northward. We had practical difficulty to collect the sufficient amount of ambient dust samples for incubation experiments. Alternatively, we used AM-dust for experiments as those reported by Guieu and Ridame, etc (Guieu et al., 2010; Ridame et al., 2014). In this study, we did not determine the size distribution of soil samples, but only the fraction less than 20 μm was used for preparing AM-dust in this study. Fe and P composition in PM_{10} and PM_{20} generated from the same soils were reported to be quite similar (Shi et al., 2011; Nenes et al., 2011). We agree that finer dust, e.g., less than 5 microns, should be more representative of those in the aeolian dust transported to the sea. However, it is also practically difficult to gain the sufficient amount of the fine dust for modification. Thus, it is a practical compromise by using artificially modified PM_{20} .

Guieu, C., Dulac, F., Desboeufs, K., Wagener, T., Pulido-Villena, E., Grisoni, J.-M., Louis, F., Ridame, C., Blain, S., and Brunet, C.: Large clean mesocosms and simulated dust deposition: a new methodology to investigate responses of marine oligotrophic ecosystems to atmospheric inputs, *Biogeosciences*, 7, 2765-2784, doi: 10.5194/bg-7-2765-2010, 2010.

Ridame, C., Dekazemacker, J., Guieu, C., Bonnet, S., L'Helguen, S., and Malien, F.: Contrasted Saharan dust events in LNLC environments: impact on nutrient dynamics and primary production, *Biogeosciences*, 11, 4783-4800, doi: 10.5194/bg-11-4783-2014, 2014.

Nenes, A., Krom, M. D., Mihalopoulos, N., Cappellen, P., Shi, Z., Bougiatioti, A., Zampas, P., and Herut, B.: Atmospheric acidification of mineral aerosols: a source of bioavailable phosphorus for the oceans, *Atmospheric Chemistry and Physics*, 11(13): 6265-6272, doi: 10.5194/acp-11-6265-2011, 2011.

Shi, Z., Bonneville, S., Krom, M. D., Carslaw, K.S., Jickells, T. D., Baker, A.R., and Benning L.G.: Iron dissolution kinetics of mineral dust at low pH during simulated atmospheric processing, *Atmospheric Chemistry and Physics*, 11(3): 995-1007, doi: 10.5194/acp-11-995-2011, 2011.

2. Section 2.2: Line 110-112: *Why Day 1 was not sampled?*

Response:

In our previous incubation experiments, there was no distinct difference of Chl *a* concentrations on day 1 and day 0 in all cases (Liu et al., 2013). Thus, we did not take a sample on day 1 in this study. We agree that the data on day 1 may be valuable and will change back our sampling protocol for future incubation experiments.

Liu, Y., Zhang, T., Shi, J., Gao, H., and Yao, X.: Responses of chlorophyll *a* to added nutrients, Asian dust, and rainwater in an oligotrophic zone of the Yellow Sea: Implications for promotion and inhibition effects in an incubation experiment, *Journal of Geophysical Research: Biogeosciences*, 118, 1763-1772, doi: 10.1002/2013JG002329, 2013.

3 Section 2.4: *The ultrasonic bath treatment may overestimate the nutrient concentration. What was the time duration used for ultra-sonication? More than 30 minute of treatment will increase the temperature and may enhance the leaching of nutrients and thus overestimate. Usually, the treatment is done for aerosol samples collected on filter substrate to loosen the particles from matrix.*

Response:

The time duration used for ultra-sonication is 30 minute. We used the ice pack to keep the temperature of water bath stable at ~0°C. We have added a detailed description of ultrasonic bath treatment in the text. (Line 139-141)

4. Section 4.1: Line 328-329: *This is a speculative statement and of course, it need further investigation. Except here, the role of trace metals (as nutrient or toxicant) is nowhere discussed. The concentration of suite of trace metals (in AM-dust) are very low except Fe and Mn. This once again indicate that, AM-dust is not the best representative for processed dust.*

Response:

The sentence has been revised as “Apart from N, P, and Fe, AM-dust also provided considerable other nutrients, e.g., Mn and Co (Table 3), which may have contributed to the phytoplankton growth in the incubations at S2 (Coale, 1991; Jakuba et al., 2008; Saito et al., 2008; Sunda, 2012).” The concentration of Co in the surface seawater of the Pacific generally maintained at picomolar levels (Sunda, 2012; Biller and Bruland, 2012), and the supplied Co by AM-dust in this study reached ~90 pmol L⁻¹, which cannot be negligible compared to the baseline values. (Line 359-361)

We have also revised the part related to the role of trace metals (as nutrient or toxicant) other than Fe, Mn and Co. As showed in our Table 3, most of trace metals are indeed negligible. (Line 458)

The aging process of dust in this study focused on the reaction between dust and inorganic acids (H₂SO₄ and HNO₃, details can be seen in Text S1). The distinguishing characteristic of AM-dust relative to untreated dust mainly

reflects in the increased contents of soluble N, P, and trace metals. Trace metals Fe, Mn, and Co were mainly originated from mineral aerosols, thus we can observe noticeably enhanced solubility of Fe, Mn, and Co in the AM-dust relative to untreated dust. The contents of other soluble trace metals such as Cu, Pb, and Zn in the dust were mainly affected by anthropogenic factor such as automobile exhaust and coal combustion, which were not reflected well in our study. To this point, we have added an illustration at the end of the conclusion. (Line 497-500)

Sunda, W.: Feedback interactions between trace metal nutrients and phytoplankton in the ocean, *Frontiers in microbiology*, 3, 204, doi: 10.3389/fmicb.2012.00204, 2012.

Billler, D. V., and Kenneth W. B.: Analysis of Mn, Fe, Co, Ni, Cu, Zn, Cd, and Pb in seawater using the Nobias-chelate PA1 resin and magnetic sector inductively coupled plasma mass spectrometry (ICP-MS), *Marine Chemistry*, 130, 12-20, doi: 10.1016/j.marchem.2011.12.001, 2012.

5. Section 4.2: *It is very difficult to follow the proposed conversion efficiency index (NCEI). It may be a good tool to specifically understand the role of nutrients on nitrogen consumption or productivity, but need to be elaborated more. It is not clear, why summation of differences of treatment and control for consecutive days are used?*

The proposed conversion efficiency index (NCEI).

Response:

In the revision, we have clarified the net conversion efficiency index (NCEI) proposed in this study to be an approximate estimation for the utilization of N for the growth of phytoplankton. Therefore, the capacity to synthesize Chl *a* per unit concentration of nitrogen (N) in different treatments can be compared. We agree that the sum consecutive differences over time in Chl *a* concentration between treatments and control will lead to an overestimation of the real net conversion efficiency because of the accumulation effect. Theoretically, the use of the maximum difference will lead to an underestimation of the real net conversion efficiency because of degradation of Chl *a* in the growth period. The real net conversion efficiency should be between those calculated by the two approaches.

We have added a more elaborated description of NCEI in the Methods Section 2.6. (Line 174-181)

6. Section 4.3: *Line 408-419: This paragraph is mostly speculative and difficult to follow, although authors have concluded the importance of DOP determination in seawater.*

Response: The part is particularly important because it provides a new insight to analyse the budget of bioavailable P in the AM-dust addition incubation experiments. Unfortunately, the original presentation seemed to be unclear and didn't service the target well. We have rewritten this paragraph to make our thoughts understandable. (Line 453-466)

Minor comments:

7. *Line 181: should be Table 1.*

Response:

Corrected. (Line 206)

8. *The legends used in Fig. 2, 3 and 4 for nutrients (other than AM-dust) are in same colors and very hard to make out. Most of them are superimposed. It will be useful for reader if different coloured legends with connecting lines can be used.*

Response:

Corrected. (Line 710-720)

Phytoplankton growth responses to Asian dust additions in the Northwest Pacific Ocean versus the Yellow Sea

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10 **Abstract.** In this study, five on-board microcosm experiments were performed in the subtropical gyre, *Kuroshio* Extension region of the Northwest Pacific Ocean (NWPO) and the Yellow Sea (YS), in order to investigate phytoplankton growth following the addition of artificially modified mineral dust (AM-dust) and various nutrients (nitrogen-N, phosphorus-P, iron-Fe, N+P, and N+P+Fe). The two experiments carried out with AM-dust addition in the subtropical gyre showed a maximum chlorophyll *a* (Chl *a*) concentration increase of 1.7- and 2.8-fold, while the cell abundance of large-sized phytoplankton (>5
15 μm) showed a 1.8- and 3.9-fold increase, respectively, relative to the controls. However, in the *Kuroshio* Extension region and the YS, the increases in maximum Chl *a* and cell abundance of large-sized phytoplankton following AM-dust addition were at most 1.3-fold and 1.7-fold larger than those in the controls, respectively. A net conversion efficiency index (NCEI) newly proposed in this study, size-fractionated Chl *a*, and the abundance of large-sized phytoplankton were analysed to determine which nutrients contribute to support phytoplankton growth. Our results demonstrate that a combination of nutrients, NP or
20 NPFe as well as other micro-constituents, are responsible for phytoplankton growth in the subtropical gyre following AM-dust addition. Single nutrient addition, i.e., N in the *Kuroshio* Extension region and P/N in the YS, controls the phytoplankton growth following AM-dust addition. In the AM-dust-addition experiments, wherein the increased NP or P were identified to determine phytoplankton growth, the dissolved inorganic P from AM-dust (8.6 nmol L^{-1}) was much lower than the theoretically

25 estimated minimum P demand ($\sim 20 \text{ nmol L}^{-1}$) for phytoplankton growth. These observations suggest that additional supply
augments the bioavailable P stock in incubated seawater with AM-dust addition, most likely due to an enhanced solubility of
P from AM-dust or re-mineralization of the dissolved organic P.

1 Introduction

30 Aeolian dust deposition can supply bioavailable nutrients such as nitrogen (N), phosphorus (P), and iron (Fe) to the upper
ocean layers (Duce et al., 1991; Jickells et al., 2005; Kanakidou et al., 2012). Observational and modelling studies have
demonstrated that external nutrient input can stimulate the primary productivity, strengthen the nitrogen fixation, alter the
phytoplankton size, and potentially enhance the carbon sequestration by the biological pump in the ocean (Mills et al., 2004;
Maranon et al., 2010; Liu et al., 2013). A well-recognized aspect of dust deposition is the iron fertilization effect in high-
nutrient low-chlorophyll (HNLC) regions (Martin, 1991; Boyd et al., 2007). Recently, many studies have attempted to explore
35 the primary chemicals that promote phytoplankton growth following dust deposition in low-nutrient low-chlorophyll (LNLC)
regions, which cover $\sim 60\%$ of the ocean area worldwide (Moore et al., 2008; Guo et al., 2012; Ridame et al., 2014; Li et al.,
2015). For example, dissolved Fe and P from deposited dust were reported to stimulate nitrogen fixation in the oligotrophic
region of the eastern tropical North Atlantic Ocean (Mills et al., 2004). In the oligotrophic region of the western North Atlantic
Ocean off Barbados, a greater concentration of dissolved N and Fe relative to P, arising from the deposited dust, likely favoured
40 the growth of *Prochlorococcus*, but limited the activity of diazotrophs (Chien et al., 2016). Moreover, the reported positive
effect of dust deposition on the primary production in the central Atlantic Ocean decreased with increasing oligotrophy of the
seawater (Maranon et al., 2010). However, studies in areas other than the North Atlantic Ocean and the Mediterranean Sea are
scarce.

45 Dust particles frequently mix with anthropogenic aerosols on their transport pathway to the oceans (Guieu et al., 2010; Shi et
al., 2012; Herut et al., 2016). The response of phytoplankton to the added dust particles mixed with anthropogenic aerosols

appeared to be more sensitive in oligotrophic waters than in moderately nutrient-enriched waters (Guo et al., 2012). Unlike in HNLC and LNLC regions of the oceans, the response of phytoplankton to dust deposition in coastal seas that receive a relatively large quantity of nutrients from rivers is poorly understood. A few studies showed that dissolved N from the added dust likely stimulated phytoplankton growth in the Yellow Sea (YS) (Liu et al., 2013). Added Fe instead of other dissolved nutrients from atmospheric deposition played an important role in stimulating phytoplankton growth in the Eastern China Sea (Meng et al., 2016). The complex responses of phytoplankton growth to dust deposition are worthy of investigation.

The N:P ratio of dust deposition is much higher than the Redfield ratio (N:P=16) (Baker et al., 2003; Guo et al., 2012), which reflects the average cellular N:P stoichiometry for oceanic phytoplankton (Arrigo, 2005). Although the deposited P was deficient relative to the demands of the resident phytoplankton in the ocean surface, a few studies showed that the input of dust could compensate for P deficiency to some extent by stimulating the biogenic conversion of dissolved organic P (DOP) to dissolved inorganic P (DIP), or a slow release of DIP from dust (Ridame and Guieu, 2002; Mackey et al., 2012; Krom et al., 2016). This supply of bioavailable P seemingly varies substantially across different oceanic regions and is affected by many factors, including dust sources and its mixing with anthropogenic pollutants, P demand, and uptake of nutrients that are co-limiting for phytoplankton in the seawater (Mackey et al., 2012). It remains a challenge to accurately estimate the supply of bioavailable P for phytoplankton growth, induced by dust deposition in specific environments.

The arid regions of eastern Asia are the most important mineral dust source to the western Pacific Ocean (Shao et al., 2011). In spring, strong westerly winds can carry large amounts of mineral dust from Asian continent along a corridor between 25°N to 45°N, and bring significant amounts of nutrients to downwind areas, including the coastal seas of China and Japan, and the Northwest Pacific Ocean (NWPO) (Shao et al., 2011). During this long-range transport, the contents of bioavailable nutrients in the Asian dust might increase through mixing with air pollutants such as sulphur dioxide and nitrogen oxides (Formenti et al., 2011). The subtropical gyre in the NWPO is a LNLC region with $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} concentrations generally maintained at nanomolar levels (Hashihama et al., 2009). Seawater in the *Kuroshio* Extension region, however, show flexible characteristics in nutrients, which is ascribed to the confluence of the *Oyashio* current with replete nutrient stocks, and that of

the *Kuroshio* current with impoverished nutrient stocks (Measures et al., 2006; Kitajima et al., 2009). On the other hand, the YS is a semi-enclosed continental shelf region, which is located in the margin of the NWPO surrounded by the Chinese mainland and the Korean Peninsula. Seawater in the YS exhibits high nutrient levels but varying nutrient-limiting conditions, e.g., N or P limitation (Liu et al., 2003). Thus, the NWPO and YS are the ideal zones to explore how Asian dust deposition influences phytoplankton growth and community size shifts under varying nutrient levels in seawaters.

In this study, we conducted on-board bioassay incubations using artificially modified mineral dust (AM-dust) collected from the Gobi Desert of China and native phytoplankton assemblages collected from the NWPO and YS, to investigate the effect of atmospheric nutrients and trace metal inputs on phytoplankton growth. The measured concentrations of Chl *a* in the nutrient treatments (i.e., N, P, Fe, N+P, and N+P+Fe additions) against the control were used to determine the nutrient limitation in the subtropical gyre, the *Kuroshio* Extension region of the NWPO and the YS, respectively. The concentrations of total and size-fractionated Chl *a*, as well as the abundances and community structures of large-sized phytoplankton (> 5 μm) were determined to study the responses of different-sized phytoplankton to the nutrients supplied by AM-dust. A net conversion efficiency index (NCEI) is proposed to identify the key nutrient(s) in determining the increase of Chl *a* concentrations, following AM-dust addition at different incubation stations. Finally, we analysed the minimum P consumption that supports phytoplankton growth following AM-dust addition, to estimate the P budget of the added dust.

2 Materials and Methods

2.1 Preparation of AM-dust

We collected surface soil samples (42.37°N, 112.97°E) from the Gobi Desert, one of the most important sources of dust events crossing over the YS and NWPO (Ooki and Uematsu, 2005; Shao et al., 2011). The soil samples were crushed, sieved to less than 20- μm particle size, and freeze-dried (Shi et al., 2009). To account for the aging of dust particles in the atmosphere, we followed Guieu et al. (2010) by mixing the soil with synthetic cloud water in a cleaned polypropylene bottle, then spread the solution on a polystyrene tray and evaporated the aqueous phase under clean air flow in a fume hood. A more detailed modification protocol is presented in the Supplementary Information (Text S1 & Table S1). These treated soil dust particles

are referred to as ‘AM-dust’. Clean plastic or plastic-coated materials were used for the preparation of AM-dust to avoid metal contamination.

2.2 Experimental design

100 Five AM-dust addition bioassay experiments were carried out on-board the *R/V Dongfanghong II* during two cruises in 2014 at stations Ar4, G7, and K4 in the NWPO (Cruise I: March–April), and B7 and H10 in the YS (Cruise II: April–May) (Fig. 1 & Table 1). Based on the baseline Chl *a* concentrations (Table 1), the five stations were redefined as S1 (Ar4), S2 (G7), S3 (K4), S4 (B7), and S5 (H10), in the ascending order. Surface seawater (2–5 m) was collected using acid-washed Teflon-coated Go-Flo bottles mounted on a SeaBird CTD assembly (SBE 9/11, USA) and filtered through a 200- μm acid-washed mesh to
105 remove larger grazers. Filtered seawater with nine different additions (true replicates, detailed in Table 2) was randomly dispensed into 18 pre-acid-washed and sample-rinsed (three times) Nalgene polycarbonate bottles (20-L each). Water samples in the incubated bottles were collected before the additions, for characterizing the baseline seawater samples, and immediately after the additions, for characterizing the amended seawater (day 0). Surface seawater was pumped into the microcosm equipment, i.e., three large plastic vessels, to stabilize the temperature of the incubation systems (Liu et al., 2013). The
110 incubation experiments were processed over 9–10 days under natural light. Water samples were collected from incubated bottles at 08:00 am every day except day 1 (i.e., day 2 through day 9/10), for determining Chl *a* and nutrient concentrations, and on certain days for phytoplankton identification and enumeration.

Treatments included AM-dust and single nutrient (N, P, or Fe) additions as well as N+P and N+P+Fe (Table 2). Previous
115 studies showed that annual deposition flux of Asian dust in the YS and NWPO regions ranged from 10 to 80 $\text{g m}^{-2} \text{yr}^{-1}$ (Gao et al., 1992; Brown et al., 2005). An extreme dust storm event accompanied by wet precipitation in the YS can lead to a considerable dust deposition flux in this range (Shi et al., 2012). Considering these results, we added 2 mg L^{-1} of the AM-dust to the incubated seawater to simulate deposition due to a strong dust event (20 g m^{-2}) in the upper 10 m water layer in the YS (Liu et al., 2013). We added the same amount of AM-dust for incubation experiments in the NWPO for comparison. This

120 amount of added dust has been widely used in other studies as well (Mills et al., 2004; Maranon et al., 2010). Based on the N-
deposition flux in the NWPO (Kim et al., 2014) and an estimate of the N addition by a dust event in the surface water of the
YS (Shi et al., 2012), we added 40 $\mu\text{mol NaNO}_3$ into the 20-L bottles for the N-related treatments (N, N+P, and N+P+Fe
additions), to increase the N concentration by 2 $\mu\text{mol L}^{-1}$. The added P and Fe in incubation systems were 0.2 $\mu\text{mol L}^{-1}$ and 2
nmol L^{-1} , respectively, based on the nutrient-addition experiments conducted in the NWPO and YS (Noiri et al., 2005; Liu et
125 al., 2013) (Table 2).

One-way analysis of variance (ANOVA) was used to assess significant differences in the mean values of the selected
parameters among various treatments, and then Dunnett's test was used to compare these treatments with the control using
SPSS (Statistical Product and Service Solutions) software.

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2.3 Chl *a* concentration

From each bottle, a 300-mL seawater sample was sequentially filtered through 20-, 2-, and 0.2- μm Whatman polycarbonate
filters to determine size-fractionated Chl *a* concentrations, i.e., pico- (0.2–2 μm), nano- (2–20 μm), and micro-sized (> 20 μm)
Chl *a*. The pigments on the filters were extracted in 90% acetone at -20 °C over 24h in the dark, and measured on a Turner
135 Designs Trilogy fluorometer (Strickland and Parsons, 1972). The total Chl *a* concentration was calculated as the sum of the
three size-fractionated values.

2.4 Inorganic nutrients and trace metals

To determine the soluble nutrients such as NO_3^- , NO_2^- , NH_4^+ , and PO_4^{3-} leached from AM-dust, oligotrophic seawater was pre-
140 filtered by 0.2 μm glass fiber filters and used for extracting these ions at $\sim 0^\circ\text{C}$ for 30 minutes in an ultrasonic bath filled with
ice and water. A 200-mL seawater sample retrieved from each of the incubation bottles was filtered through pre-acid-washed
cellulose acetate membranes and filtrates were stored immediately at -20 °C in acid-washed high-density polyethylene bottles

for nutrients analysis in the laboratory. Soluble nutrients leached from AM-dust and in the incubated seawaters were measured following the automated colorimetric technique described by Grasshoff et al. (1999) and Ridame et al. (2014), using a QuAAtro Continuous Flow Analyzer (SEAL Analytical). Detection limits of this instrument were defined as three times the standard deviation of the blank, which corresponds to 30 nM, 7 nM, 80 nM, and 40 nM for NO_3^- , NO_2^- , NH_4^+ , and PO_4^{3-} , respectively. Note that the NH_4^+ concentrations were determined only for the leached solution of AM-dust. We followed the method proposed by Hsu et al. (2010) to determine the concentrations of soluble trace metals leached from AM-dust using an ICP-MS. The recovery yield, accuracy, and detection limit are summarized in Table S2.

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2.5 Phytoplankton identification and enumeration

Seawater samples (300-mL volume) were collected from the baseline seawaters at each station and incubated bottles on day 4 from S1 and S2, day 3 from S3 and S5, and day 5 from S4 for the identification and enumeration of large-sized phytoplankton ($> 5 \mu\text{m}$). The sampling dates selected during the incubation experiment were either close to or corresponded exactly with the days that showed the maximum Chl *a* concentrations (Figure S1). The sampled seawater was fixed with 1% Lugol's iodine and stored in dark until microscopic observation in the laboratory (Burson et al., 2016). Before analysis, the preserved samples were settled for 48 h in the dark and then concentrated to 10-mL volume in glass cylinders. Large-sized phytoplankton was identified and enumerated using a Nikon ECLIPSE TE2000-U inverted microscope. To illustrate the responses of different phytoplankton to various additions, the detected phytoplankton species were divided into two major functional groups: diatoms and dinoflagellates. The dominant species of diatoms found in this study were *Nitzschia* spp., *Chaetoceros* spp., *Thalassiosira* spp., *Skeletonema* spp., *Cylindrotheca closterium*, and *Rhizosolenia setigera*.

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2.6 Protocol of data analysis

In this study, Chl *a* concentrations generally showed a bell-shape growth curve at all stations with the maximum concentration occurring around 2–5 days (Fig. S1). We focused on analysing the initial 2–5 days, which showed a successive increase in the

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incubation period. The duration of 2–5 days for incubation has been widely used in other microcosm experiments (Herut et al., 2005; Tanaka et al., 2011; Guo et al., 2012; Li et al., 2015) and is supposed to minimize the effects of bottle enclosure and a possible deviation from the natural environment (Mackey et al., 2012; Coelho et al., 2013).

170 A net conversion efficiency index (NCEI) was introduced to quantify the differences in N utilization efficiency among the AM-dust treatments and N, N+P, and N+P+Fe treatments. NCEI (unit: $\frac{\mu\text{g L}^{-1}}{\mu\text{mol L}^{-1}}$, i. e., g mol^{-1}) was calculated using the following equation:

$$\text{NCEI} = \frac{\sum_{i=0}^t (\text{Chl } a_{Ti} - \text{Chl } a_{Ci})}{\Delta\text{N}} \quad (1)$$

175 where $\text{Chl } a_{Ti}$ and $\text{Chl } a_{Ci}$ represent the Chl *a* concentrations on the i^{th} day, i.e., day 0, day 2–day 3/5, in the treatments and the control ($\mu\text{g L}^{-1}$), respectively. ΔN is the decreased N concentration in the treatment minus that in the control during the successive increase in the incubation period ($\mu\text{mol L}^{-1}$). The decreased N concentration equals to the difference in $\text{NO}_3^- + \text{NO}_2^-$ before and after the initial increase period. A large NCEI value represents a positive effect of the added N on phytoplankton growth, and a value close to zero represents no effect. Larger the NCEI value, higher is the N utilization efficiency by the phytoplankton. Theoretically, the maximum increase of Chl *a* in concentration can also be used to calculate NCEI. Considered 180 the accumulation and degradation of Chl *a*, the real net conversion efficiency should be among the NCEI values between the two approaches.

We also defined the consumed ratio of N:P (hereafter, $C_{\text{N:P}}$) and supply ratio of N:P (hereafter, $S_{\text{N:P}}$) to illustrate the changing stocks of bioavailable N and P in various incubation experiments: $C_{\text{N:P}}$ was the ratio of the difference in $\text{NO}_3^- + \text{NO}_2^-$ concentrations at the beginning and the end of the successive increase in incubation period to that of PO_4^{3-} ; $S_{\text{N:P}}$ was the amended ratio of $\text{NO}_3^- + \text{NO}_2^-$ to PO_4^{3-} concentrations in the incubated seawaters on day 0 in various treatments. 185

3 Results

3.1 Characteristics of baseline surface seawater and AM-dust

190 Stations S1, S2, and S3 are in the open ocean and were characterized by warm surface seawaters with Sea Surface Temperature (SST) larger than 18°C, and deep mixed layer with a depth larger than 80 m. On the other hand, stations S4 and S5 are in the YS and the SST and MLD therein were less than 13°C and 20 m, respectively (Table 3). The winter cooling induces a lower water temperature in the shallow YS than in the deep open ocean. Therefore, the short warming time before our observations during the springtime cannot reverse this situation. The weaker winds as well as lesser loss of heat flux over the land-surrounding YS than over the open ocean lead to the shallower mixed depth at S4 and S5 than at S1-S3.

195 Low concentrations of Chl *a* ($\leq 0.50 \mu\text{g L}^{-1}$) and nutrients ($\text{NO}_3^- + \text{NO}_2^- \leq 0.26 \mu\text{mol L}^{-1}$, $\text{PO}_4^{3-} = 0.05 \mu\text{mol L}^{-1}$) were observed at S1 and S2 in the subtropical gyre of the NWPO (Table 1), indicating oligotrophy (Hashihama et al., 2009). At S3 in the *Kuroshio* extension and S4 in the YS, Chl *a* and nutrient concentrations were at least 2-fold higher than those at S1 and S2, indicating an increased trophic state at S3 and S4. Although the nutrient levels at S5 in the YS were comparable to those at S1, the Chl *a* concentration of $2.74 \mu\text{g L}^{-1}$ was close to the values during spring blooms in the YS (Liu et al., 2003; Fu et al., 2009; Liu et al., 2013). The N:P ratios at S1, S2, S3, and S5 were far lower than the Redfield Ratio (16:1), while that at S4 was as high as 32 (Table 2). Pico-sized Chl *a* accounted for >70% of the total Chl *a* at S1 and S2, but it decreased to less than ~50% at S3, S4, and S5. Specifically, large-sized phytoplankton abundance showed an increasing trend from < 20 cells mL^{-1} at S1 and S2 to 35 cells mL^{-1} at S4 and 79 cells mL^{-1} at S3, and to the highest value of 314 cells mL^{-1} at S5. Diatoms dominated the large-sized phytoplankton community at S1, S3, and S5, and a codominance of dinoflagellates was noted at S2 and S4 (Table 1).

200 The concentration of dissolved inorganic nitrogen (DIN, i.e. $\text{NO}_3^- + \text{NO}_2^- + \text{NH}_4^+$) in the AM-dust was $577 \mu\text{mol g}^{-1}$ (Table 3) and increased by a factor of four against that in the untreated dust. A notable increase in the PO_4^{3-} content ($4.3 \mu\text{mol g}^{-1}$) in the AM-dust was observed, relative to the untreated dust ($1.3 \mu\text{mol g}^{-1}$). The abundances of DIN, PO_4^{3-} , and soluble Fe in the AM-dust were generally consistent with the values observed in a strong dust event that occurred over the YS in the spring of 2007

(Shi et al., 2012). The N:P ratio of ~134 in the AM-dust was far greater than 16 (Redfield Ratio), and similar to those in Asian dust aerosols previously reported (Shi et al., 2012; Liu et al., 2013; Chien et al., 2016). The $\text{NO}_3^- + \text{NO}_2^-$ concentration in the incubated seawater after AM-dust addition increased up to $\sim 1.13 \mu\text{mol L}^{-1}$ (Table 3 & Fig. 2), larger than the baseline $\text{NO}_3^- + \text{NO}_2^-$ stocks at S1, S2, S3, and S5, but accounted only for approximately 1/3rd of the baseline stock at S4 (Table 2). However, the increase in PO_4^{3-} ($\sim 8.6 \text{ nmol/L}$) following AM-dust additions was negligible, compared with the baseline P stock at each station (Tables 2 & 3).

3.2 Variation in nutrient and total Chl *a* concentrations during incubation experiments

During the successive increase in the incubation period (initial 2–5 days of the incubations, Sect. 2.6), $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} concentrations generally showed a decline in the control and all treatments (Fig. 2). At each station, the most obvious decline of $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} in concentration in various treatments occurred when N+P or N+P+Fe was added together. The net declines were $1.49\text{--}4.66 \mu\text{mol L}^{-1}$ for $\text{NO}_3^- + \text{NO}_2^-$ and $0.18\text{--}0.27 \mu\text{mol L}^{-1}$ for PO_4^{3-} . The decline was $0.49\text{--}2.99 \mu\text{mol L}^{-1}$ for $\text{NO}_3^- + \text{NO}_2^-$ and $0.10\text{--}0.27 \mu\text{mol L}^{-1}$ for PO_4^{3-} followed by N and P additions, respectively. For P addition alone at S4, the decline of PO_4^{3-} was comparable to those in combined nutrient additions ($0.27 \mu\text{mol L}^{-1}$) and that of $\text{NO}_3^- + \text{NO}_2^-$ ($2.99 \mu\text{mol L}^{-1}$) was 1.3-fold larger relative to the control. Correspondingly, addition of N+P+Fe induced the largest increases in total Chl *a* (hereafter, Chl *a*) among all treatments, and the increases were significant against the controls (Fig. 3, $p < 0.05$) at all stations (Fig. 3). The maximum concentrations of Chl *a* in the N+P+Fe treatments reached $1.70 \mu\text{g L}^{-1}$ at S1 (day 4), $1.65 \mu\text{g L}^{-1}$ at S2 (day 5), $3.50 \mu\text{g L}^{-1}$ at S3 (day 3), $4.98 \mu\text{g L}^{-1}$ at S4 (day 4), and $5.08 \mu\text{g L}^{-1}$ at S5 (day 2), respectively. This was a 1.9- to 7.1-fold increase compared to the baseline and was 1.5- to 3.0-fold larger than the controls in the five incubation experiments. The time-series of Chl *a* in N+P and N+P+Fe treatments almost overlapped at all stations except S2, where the maximum value of Chl *a* in the N+P treatment ($1.15 \mu\text{g L}^{-1}$) was only 70% of that in the N+P+Fe treatment. Significant increases in Chl *a* relative to the control were also observed in N treatments at S1, S2, S3, and S5, and in P treatments at S2 and S4 ($p < 0.05$). The maximum concentrations of Chl *a* in the N treatments at S1 ($0.73 \mu\text{g L}^{-1}$ on day 3), S2 ($0.67 \mu\text{g L}^{-1}$ on day 2), S3 ($2.14 \mu\text{g L}^{-1}$ on day 3), S5 ($4.34 \mu\text{g L}^{-1}$ on day 2), and in the P treatments at S2 ($0.79 \mu\text{g L}^{-1}$ on day 2) and S4 ($4.46 \mu\text{g L}^{-1}$ on day 3) showed a 1.3- to 3.1-fold increase compared to the baseline and were 1.2- to 1.6-fold larger than the controls. There was no significant

increase in Chl *a* relative to the control in the rest of the additions at the five stations.

240 Followed the addition of AM-dust, $\text{NO}_3^- + \text{NO}_2^-$ concentrations declined by $1.06 \mu\text{mol L}^{-1}$ at S1, $0.88 \mu\text{mol L}^{-1}$ at S2, $1.36 \mu\text{mol L}^{-1}$ at S3, $2.93 \mu\text{mol L}^{-1}$ at S4, and $0.36 \mu\text{mol L}^{-1}$ at S5, respectively. The declines were generally larger than the values in the baseline (Fig. 2). PO_4^{3-} concentrations in AM-dust treatments declined by $0.04\text{-}0.11 \mu\text{mol L}^{-1}$ and were similar to those of the control during the successive increase in the incubation period at each station. Correspondingly, the maximum concentration of Chl *a* following AM-dust addition was $1.04 \mu\text{g L}^{-1}$ at S1 (day 5), $1.54 \mu\text{g L}^{-1}$ at S2 (day 5), $2.23 \mu\text{g L}^{-1}$ at S3 (day 2), $3.27 \mu\text{g L}^{-1}$ at S4 (day 5), and $3.13 \mu\text{g L}^{-1}$ at S5 (day 2), respectively (Fig. 3), which was 1.1- to 4.4-fold larger against the baseline.

245 The increases in Chl *a* were significant different ($p < 0.05$) from the control at S1, S2, S3, and S5. The maximum Chl *a* concentration following AM-dust addition was 1.7-fold, 2.8-fold, and 1.3-fold of those in the controls at S1, S2, and S3, respectively. At S4 and S5, the maximum Chl *a* concentration in the AM-dust treatments was comparable to those in the controls. However, the Chl *a* concentrations remained relatively high in the AM-dust treatments, while showing a decreasing trend in the controls (Fig. 3).

3.3 Variation in the size-fractionated Chl *a* concentrations during incubation experiments

When the size-fractionated Chl *a* was examined in various treatments at the five stations, a few notable changes were observed, as highlighted below.

255 a) In the controls, the dominant contributors to the total Chl *a* were pico-sized cells at S1, S2, S4, and nano-sized cells at S5, while they changed from pico-sized to micro-sized cells at S3 during the successive increase in the incubation period (Fig. 4). Addition of AM-dust increased the Chl *a* concentrations of all sizes and the dominant contributor was consistent with those in the controls at the five stations. However, the magnitude of Chl *a* increase was the highest for micro- or nano-sized cells following AM-dust additions. For example, the largest increase in maximum Chl *a* occurred in micro-sized cells at S1 (2.0-fold), S2 (4.4-fold), S4 (1.4-fold), and S5 (1.6-fold), and in nano-sized cells at S3 (1-3 fold), compared with the controls.

260 b) Size-fractionated Chl *a* in N+P and N+P+Fe additions showed the largest and similar increases at S1, S3, S4, and S5,

especially in the micro-sized range, where the maximum Chl *a* concentrations showed a 1.2- to 6.5-fold increase against the controls (Fig. 4). For station S2, the micro- and nano- sized Chl *a* showed higher increases in N+P+Fe treatments than in N+P treatments, whereas the inverse was true for the pico-sized Chl *a*. Addition of N or P alone also led to an increase in size-fractionated Chl *a* to some extents at all stations. The size-fractionated Chl *a* in N treatments at S1, S3, and S5 increased noticeably, e.g., the maximum Chl *a* showed the largest increase in nano-sized cells by a factor of 1.6 at S1, and 1.4 at S3, and in micro-sized cells by a factor of 2.2 at S5, relative to the controls. A clear increase was also observed in the P treatment at S2 and S4, e.g., the maximum Chl *a* concentrations showed a 1.8-fold (Nano-) and 1.5-fold (Micro-) increase against the controls (Fig. 4). There was no noticeable increase in any of the size-fractionated Chl *a* following Fe addition at S2 and S4.

c) The results of size-fractionated Chl *a* demonstrated that the micro- and nano-sized phytoplankton generally showed a stronger response than pico-sized cells following AM-dust and various nutrient additions, although the extent of increase varied among the stations.

3.4 Variation in large-sized phytoplankton abundance and community during incubation experiments

After 2–5 incubation days, diatoms generally dominated the large-sized phytoplankton community in all treatments at the five stations. *Chaetoceros* spp. and *Nitzschia* spp. comprised the largest fraction of diatoms in all treatments at S1, S2, and S3. For stations S4 and S5, the dominant species of diatoms did not change compared with those in the baseline seawaters (Fig. 5). Specifically, the cell abundance of large-sized phytoplankton in the controls increased to 37 cells mL⁻¹ at S1, 39 cells mL⁻¹ at S2, 422 cells mL⁻¹ at S3, 93 cells mL⁻¹ at S4, and 643 cells mL⁻¹ at S5. Similar to the response of Chl *a*, additions of N+P and N+P+Fe increased the phytoplankton abundances noticeably in all treatments by about 2.5- to 2.7-fold at S1 and S5, 3.1- to 3.3-fold at S3, and 2.7-fold at S4, relative to the controls. For single-nutrient treatments, N additions at S1, S3, and S5, and P additions at S4 induced the highest increases in cell abundance, with 1.5-, 1.6-, 2.3-, and 1.9-fold higher values, respectively, than those in the controls. The cell abundance of large-sized phytoplankton in AM-dust treatments relative to the controls increased by 1.8-fold at S1 and 1.7-fold at S3 and S5, but showed a negligible increase at S4. Moreover, the increases following AM-dust additions were comparable, and sometimes even larger, relative to those in the N treatments at S1, S3, and S4, but

285 lower than the latter at S5. At S2, the cell abundance with AM-dust addition increased by 3.9-fold against the controls, which was comparable to those in the N+P treatments (Fig. 5).

4 Discussion

4.1 Nutrient limitation in the NWPO and YS

Building on the results mentioned above (Fig. 2 and Table 2), we summarized the nutrient limiting status at the five stations.

290 At S1, phytoplankton growth was very likely co-limited by N and P, because: 1) N addition induced significant increase in Chl *a* against the controls ($p < 0.05$), 2) there were significant increases in Chl *a* following N+P additions, compared with the N treatments ($p < 0.05$), and 3) there were no significant differences in Chl *a* between N+P and N+P+Fe treatments ($p > 0.05$).

At S2, a significant increase in Chl *a* was observed in the P, N+P, and N+P+Fe treatments ($p < 0.05$) against the controls, and addition of N+P and N+P+Fe induced a significantly larger response, relative to the P treatments ($p < 0.05$). Besides, Chl *a*

295 concentrations in the N+P+Fe treatments were significantly larger than in the N+P treatments on days 4–5. Therefore, we concluded that phytoplankton at S2 were primarily co-limited by N, P, and Fe (Fig. 2). At S3, N addition induced a significant

increase in Chl *a* compared with the controls, while an even larger significant increase was observed upon N+P (+Fe) addition, relative to the N treatments. Thus, phytoplankton was primarily co-limited by N and P at S3. Accordingly, $\text{NO}_3^- + \text{NO}_2^-$ and

PO_4^{3-} concentrations in N+P (+Fe) treatments at the three stations decreased by 73%–100%, larger than the decline in $\text{NO}_3^- + \text{NO}_2^-$ concentrations after adding N alone, and PO_4^{3-} concentrations after adding P alone (43%–70%) (Fig. 3). Such co-limiting

300 conditions have been widely reported in the oceans, especially in oligotrophic regions such as Eastern Mediterranean Sea (Tanaka et al., 2011) and South China Sea (Guo et al., 2012). Moore et al. (2013) argued that low abundances of N and P in

oligotrophic environments are likely to simultaneously reach the limiting levels for phytoplankton growth. Note that phytoplankton at S3 in the *Kuroshio* Extension showed a greater response to N rather than P addition. The intrusion of

305 *Kuroshio–Oyashio* transition water from the north and *Kuroshio* water from the southwest, both carrying macronutrients typically with $\text{N:P} < 16$, likely changed the nutrient stocks of the seawaters at S3 (Whitney, 2011; Guo et al., 2012; Yatsu et

al., 2013). A low primary productivity in the *Kuroshio* Extension region during the spring time was also indicated by the low availability of nitrate in the seawater (Nishibe et al., 2015).

310 Similarly, we also found that P and N were the primary limiting nutrients at S4 and S5, respectively (Fig. 2). This is consistent with the baseline N:P ratios at S4 (>16) and S5 (<16) (Table 2). In fact, the PO_4^{3-} after adding P alone at S4, and $\text{NO}_3^- + \text{NO}_2^-$ after adding N alone at S5, decreased by 80% and 100%, respectively, during the successive increase in incubation period (Fig. 3). Complex hydrographic conditions, e.g., various riverine inputs and atmospheric deposition, create a large spatiotemporal variation in nutrient concentrations in the YS (Liu et al., 2003). The riverine input and atmospheric deposition with a high N:P (>16) lead to relatively P-deficient conditions, while the rapid uptake by phytoplankton characterized by a lower N:P (<16) during the bloom period likely accelerates the decline of P concentrations in the surface seawaters (Liu et al., 2003; Arrigo, 2005; Liu et al., 2013). In general, the N:P ratios during the spring time in the coastal waters (near S4) are higher than 16 and lower than 16 in the central waters (near S5) of the YS (Fu et al., 2009). The primary P- or N-limiting conditions in the surface seawaters of the YS have also been widely reported in previous studies (Wang et al., 2003; Liu et al., 2013).

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4.2 Positive effects of AM-dust on phytoplankton growth in the NWPO and YS

It was clear from the data that AM-dust addition increased the Chl *a* concentrations and large-sized phytoplankton abundance to varying extents at the five stations (Fig. 2, 4&5). However, the increased concentrations of nutrients (e.g., N, P, and Fe) in AM-dust additions were different from those in various nutrient additions. In this study, the NCEI was proposed to quantitatively compare the utilization of N for the growth of phytoplankton among treatments and infer which nutrients in the AM-dust primarily supported phytoplankton growth.

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$\text{NO}_3^- + \text{NO}_2^-$ was theoretically added in the AM-dust treatments at a concentration of $1.13 \mu\text{mol L}^{-1}$ (Table 3). However, the increase in $\text{NO}_3^- + \text{NO}_2^-$ concentration was only $0.59\text{--}0.85 \mu\text{mol L}^{-1}$, immediately after the addition and mixing steps (i.e. on day 0) at the five stations, which was about 52%–75% of the added $\text{NO}_3^- + \text{NO}_2^-$. Similar trends were also observed for $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} concentrations in the nutrient treatments (Fig. 3). The time interval of adding materials to the incubation bottles and sampling seawater for nutrient measurement was 1-2 hr. Microbial uptake, scavenging by cell surface and bottle

wall, etc., possibly decreased the concentrations of nutrients at the 1-2 hr, leading to the measured values smaller than the theoretical values. When the concentrations of $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} decreased in the seawater, those absorbed by cell surface and bottle wall had the potential to be released into the solution again for reaching equilibrium (Liu et al., 2013). Thus, the theoretically added N and P, plus the baseline concentrations in the seawaters, were considered the amended concentrations on day 0, which were used for calculating the NCEI.

The added N in AM-dust and N-related treatments generally stimulated phytoplankton growth to varying extents, as indicated by the large NCEI values (Fig. 6). Note that the NCEI in N treatments (hereafter NCEI_N) at S4 was around 0 g mol^{-1} , because the phytoplankton was primarily limited by P. Besides, the significant difference in NCEI among various nutrient treatments can also reflect the roles played by other nutrients such as P and Fe in affecting the N utilization efficiency. The NCEI in the N+P+Fe treatments were $1.6 \pm 0.1 \text{ g mol}^{-1}$ at S1, $1.5 \pm 0.1 \text{ g mol}^{-1}$ at S3, $2.7 \pm 0.2 \text{ g mol}^{-1}$ at S4, and $1.4 \pm 0.1 \text{ g mol}^{-1}$ at S5, comparable with those in N+P treatments (Fig. 6). This suggested that the effect of added Fe on the N-utilization efficiency was negligible at the four stations. Although the added Fe alone had no obvious influence on the Chl *a* concentration at S2 (Fig. 2), the higher value of NCEI_{N+P+Fe} ($1.8 \pm 0.0 \text{ g mol}^{-1}$) than the NCEI_{N+P} ($1.6 \pm 0.0 \text{ g mol}^{-1}$) therein demonstrated that Fe likely contributed to increase the N-conversion efficiency during Chl *a* generation (Twining & Baines, 2015). In addition, the values of NCEI_{N+P} and NCEI_{N+P+Fe} were generally higher than those of NCEI_N at S1, S2, S3, and S4 (Fig. 6), indicative of the fact that added P contributed to the increased N utilization efficiency at these stations. Indeed, P is an important part of the ribosome and can promote cell division (Arrigo, 2005), which likely increased the NCEI values to some extent. Liu et al. (2013) also reported the excessive P relative to N ($\text{N:P} < 16$) in incubated seawater could increase the biomass. There were no significant differences in the NCEI values between the various treatments at S5, demonstrating that the effects of other nutrients (i.e., P and Fe) on N-conversion efficiency were negligible compared with N.

The NCEI in the AM-dust treatments was $1.3 \pm 0.1 \text{ g mol}^{-1}$ at S1 and $1.4 \pm 0.6 \text{ g mol}^{-1}$ at S4, which was 1.7- and 16.8-fold higher than those in the N treatments, respectively. (Fig. 6). Considering the P-limiting conditions at both stations, we concluded that the external supply of P associated with the AM-dust input likely also played a role in increasing the NCEI.

Moreover, the $NCEI_{AM-dust}$ of $2.4 \pm 0.1 \text{ g mol}^{-1}$ was higher than $NCEI_{N+P+Fe}$ ($1.8 \pm 0.0 \text{ g mol}^{-1}$), $NCEI_{N+P}$ ($1.6 \pm 0.0 \text{ g mol}^{-1}$), and $NCEI_N$ ($0.3 \pm 0.2 \text{ g mol}^{-1}$) at S2. Apart from N, P, and Fe, AM-dust also provided considerable other nutrients, e.g., Mn and Co (Table. 3), which may also have contributed to the phytoplankton growth in the incubations at S2 (Coale, 1991; Jakuba et al., 2008; Saito et al., 2008; Sunda, 2012). This potential synergistic effect is worthy of further investigation. At S3 and S5, the NCEI in AM-dust treatments, i.e., $1.2 \pm 0.1 \text{ g mol}^{-1}$ at S3, $1.1 \pm 0.5 \text{ g mol}^{-1}$ at S5, were similar to those in N treatments, i.e., $1.1 \pm 0.1 \text{ g mol}^{-1}$ at S3, $1.2 \pm 0.2 \text{ g mol}^{-1}$ at S5 (Fig. 6), indicating a negligible effect of the dissolved P from AM-dust on NCEI.

In recent years, more studies challenged to the use of ‘Redfield ratio’ to evaluate nutrient limiting conditions of phytoplankton (Klausmeier et al., 2004; Arrigo, 2005; Moore et al., 2013). Atmospheric aerosols characterized by mineral dust or anthropogenic pollutants contained complicated chemical components, their deposition to oceans can lead to complicated responses of phytoplankton (Guo et al., 2012; Paytan et al., 2009; Meng et al., 2016). The NCEI appears to have the advantage to reveal the possible mechanisms associated with the responses to some extents. For instance, phytoplankton at S1, S2, and S3 were under co-limiting conditions of various nutrients, while the ratios of N:P in the seawaters were lower than ‘Redfield ratio’ of 16 (indicating N limitation). The difference of NCEI values among AM-dust and various nutrient treatments (N, P, N+P, N+P+Fe) allows quantifying the role of N, P, and Fe in stimulating phytoplankton growth (Fig. 6). Moreover, the NCEI also reveals other nutrients which may have a promotion effect on phytoplankton at S2. Note that the calculation of NCEI was based on the change of Chl *a* rather than biomass. In general, phytoplankton biomass increased with Chl *a*, which is an essential component for photosynthesis. However, as C:Chl *a* ratios could vary for different phytoplankton assemblages (Sathyendranath et al., 2009), the relationship between biomass and N nutrients needs to be further investigated.

Fig. 7 shows the changes in the three size-fractionated Chl *a* concentrations in the treatments, relative to the corresponding controls. Generally, the relative changes in the size-fractionated Chl *a* in AM-dust treatments showed a decreasing trend with decreasing cell size, i.e., micro- \geq nano- \geq pico-sized cells, consistent with the pattern observed in N+P+Fe treatments. Previous incubation experiments have also showed that addition of Asian aerosols could shift the phytoplankton size towards larger cells (Guo et al., 2012; Liu et al., 2013). Indeed, micro- and nano-sized cells, as indicated by faster uptake rates for

nutrients and higher biomass-specific production rates, have a growth advantage relative to the pico-sized cells, when the added materials relieve the nutrient-limiting pressures (Cermeno et al., 2005; Maranon et al., 2007&2012). Besides, the relative changes in micro- and nano-sized Chl *a* in AM-dust treatments were generally higher than in the N treatments at S1, S2, and S4. Especially for station S2, the relative changes in micro- and nano-sized Chl *a* in the AM-dust treatments were even larger than in the N+P treatments, and comparable to those in the N+P+Fe treatments (Fig. 7). These results demonstrated the importance of increased bioavailability of P at S1 and S4, and that of P and Fe at S2, following AM-dust addition, which supported the growth of micro- and nano-sized phytoplankton.

When the cell abundances of large-sized phytoplankton were considered, we also found that the addition of AM-dust induced larger increases than N additions after certain incubation days at S1 and S2, as characterized by co-limitation of multiple nutrients. Especially at S2, the increase in large phytoplankton abundance was even larger than in the N+P treatments and lower than in the N+P+Fe treatments. However, for station S3, characterized by NP co-limitation, the changes in the cell abundance of large-sized phytoplankton in AM-dust treatments were comparable to those in the N treatments. These results indicated that the supply of NP and NPFe induced by AM-dust addition, likely contributed to the increase in large-sized cell abundance at S1 and S2, respectively, whereas the supplied N from AM-dust likely determined the growth of large-sized phytoplankton at S3, because the supply of P from the AM-dust was negligible relative to the baseline P stocks ($\sim 0.11 \mu\text{mol L}^{-1}$). For station S5, characterized by N limitation alone, the addition of AM-dust provided the limiting nutrient to increase the cell abundance of large-sized phytoplankton (Fig. 5).

Diatoms generally dominated the phytoplankton community in the control and treatments during the successive increase in the incubation period at S1, S2, S3, and S5 (Fig. 5). The addition of AM-dust and limited nutrients induced a notable increase in the diatom cell abundance at the four stations. Indeed, when the nutrient-limitation stress were relieved, diatoms tended to prosper rapidly, which was ascribed to their fast uptake rates and quick metabolic responses to enhanced nutrient availability (Fawcett and Ward, et al., 2011; Franz et al., 2012). The dominant species of diatoms (*Chaetoceros* spp., *Nitzschia* spp. in the open ocean and *Thalassiosira* spp., *Skeletonema* spp., *Cylindrotheca closterium*, *Rhizosolenia setigera*, *Nitzschia* spp. in the

YS) in this study were similar to those reported in the North Pacific and YS (Noiri et al., 2005; Aizawa et al., 2005; Sun et al., 2013). Although the dominated algae at S4 was diatoms in treatments that received various additions, a shift of community structure from diatoms to dinoflagellates was observed for AM-dust and N additions relative to the control. The contributions of dinoflagellates to the large-sized phytoplankton community in cell abundance reached 40%–48%, while there was only 18%–28% among the P, N+P, and N+P+Fe treatments (Fig. 5). This was closely related to the fast growth of diatoms as a result of the relief from P-limiting conditions after the addition of P, N+P, and N+P+Fe. Besides, it has been reported that dinoflagellates with lower demands and higher P absorption efficiency have an advantage of surviving in P-deficient conditions, compared with diatoms (Egge, 1998; Zhou et al., 2008). Hence, in contrast to P-related treatments, the supply of bioavailable P from AM-dust was not enough to support the growth of diatoms at S4 (Fig. 5).

4.3 Bioavailable P in the AM-dust addition experiment

The increase in bioavailable P concentration following AM-dust addition played an important role in stimulating phytoplankton growth at S1, S2, and S4, as mentioned above (Sect. 4.2). However, we did not detect a marked increase in the PO_4^{3-} concentrations in incubated seawaters, following AM-dust additions, during the successive increase in incubation period at the three stations (Fig. 3). This could be ascribed to the gradual bioavailable P released over time and its rapid uptake by phytoplankton with a high P demand (Mackey et al., 2012), and the easy adsorption of PO_4^{3-} by particles or bottle walls (Liu et al., 2013), which likely increases of PO_4^{3-} concentration in the incubated seawaters below the detection limit. In order to quantify the increased bioavailable P concentrations following AM-dust addition, we analysed the correlation of the $C_{\text{N:P}}$ and $S_{\text{N:P}}$ in AM-dust and N treatments in Fig 8.

$C_{\text{N:P}}$ significantly increased with $S_{\text{N:P}}$ (Fig. 8), and the same trend was observed at each station (Fig. S2). Klausmeier et al. (2004, 2008) pointed out that the phytoplankton in nutrient-rich environments tends to exhibit an exponential/bloomer growth with a requirement of N:P ratio in a relatively constant value, regardless of the varying nutrient supply ratios. However, when the phytoplankton tend to sustain a competitive equilibrium status in nutrient-limited environments, e.g., N or P, their

requirements of N:P ratios generally increase with the nutrient supply ratios. In our study, the phytoplankton at the five stations were limited by different nutrients, and thus, tended to sustain a competitive equilibrium status. As the provided DIN following the AM-dust addition was $1.15 \mu\text{mol N L}^{-1}$, lower than the amount of N when added alone ($2 \mu\text{mol N L}^{-1}$), the $S_{\text{N:P}}$ in AM-dust treatment was lower than that in N treatments. Thus, $C_{\text{N:P}}$ in both treatments at each station would follow the pattern: $C_{\text{N:P}}$ in AM-dust treatments $\leq C_{\text{N:P}}$ in N treatments. The consumed N in AM-dust treatments, as well as N and P in N treatments, can be determined on basis of nutrient measurements at the beginning and the end of the successive increase in the incubation period (Fig. 3).

Because of $C_{\text{N:P}}$ in AM-dust treatments $\leq C_{\text{N:P}}$ in N treatments, an inequality can be obtained as below

$$CP_{AD} \geq \frac{CN_{AD} \times CP_N}{CN_N} \quad (2)$$

where CP and CN are the consumed bioavailable P and $\text{NO}_3^- + \text{NO}_2^-$ concentrations ($\mu\text{mol L}^{-1}$) in the incubation systems during the successive increase in incubation period, respectively. Subscript means AM-dust ('AD') or N treatments.

The calculated result of the right term can be considered as the lower limit of the consumed P in the AM-dust treatments. They were 76 nmol L^{-1} at S1, 46 nmol L^{-1} at S2, 98 nmol L^{-1} at S3, 122 nmol L^{-1} at S4, and 20 nmol L^{-1} at S5. Deducting the P concentrations in the baseline seawaters (Table 2), the minimum bioavailable P supplied by AM-dust additions were estimated to be 22 nmol L^{-1} at S1, -5 nmol L^{-1} at S2, -6 nmol L^{-1} at S3, 18 nmol L^{-1} at S4, and -20 nmol L^{-1} at S5. Note that the dissolved P from AM-dust was as low as 8.6 nmol L^{-1} in the incubation bottles, which could not meet demands of phytoplankton at S1 and S4. The negative values implied the P concentrations in baseline seawaters were likely enough to support the growth of phytoplankton.

Thus, the supply other than the theoretically dissolved bioavailable P of 8.6 nmol L^{-1} induced by the AM-dust might exist. As reported in literature,

the dissolved organic phosphorus (DOP) is considered a significant portion of the dissolved P pool in

oceans, especially in surface seawater (Paytan and McLaughlin, 2007). Dust input has been reported to induce the biological

mineralization of DOP to DIP under P-deficient conditions, and consequently, increased P bioavailability. DIP mineralized from DOP in the seawater would even accumulate if the demand of phytoplankton for P was small (Mackey et al., 2012). Trace elements such as Co is an important component of the metalloenzyme alkaline phosphatase, which could facilitate the acquisition of bioavailable P from the DOP pool (Cembella et al., 1982; Jakuba et al., 2008). The concentration of Co nutrient in the surface seawater of the North Pacific generally maintained at picomolar levels (Sunda, 2012; Biller and Bruland, 2012), and the supplied Co of $\sim 90 \text{ pmol L}^{-1}$ by AM-dust addition (Table 3) likely played a role in increasing P bioavailability in this study. Note that the gradual enhanced solubility of atmospheric P with the duration of exposure in seawater was also reported to contribute additional bioavailable P for phytoplankton growth (Ridame and Guieu, 2002; Mackey et al., 2012). Unfortunately, no measurements of trace metal concentrations in seawaters as well as DOP and P contents in the phytoplankton cells were available in our study. The lack of data increased the uncertainty in data interpretation, which needs to be substantiated in a future study.

5 Conclusion

In this study, phytoplankton growth was found to be limited by two or more nutrients (i.e., NP or NPFe) in the NWPO, and by a single nutrient (i.e., N or P) in the YS. In the subtropical gyre of the NWPO, the addition of AM-dust provided NP or NPFe and micro-constituents to stimulate phytoplankton growth. In comparison with the controls, the maximum Chl *a* following AM-dust addition showed a 1.7- and 2.8-fold increase, while the cell abundance of large-sized phytoplankton showed a 1.8- and 3.9-fold increase. As the increased P from AM-dust was negligible in comparison with the baseline P stocks, the dissolved N from AM-dust, thus, primarily supported the phytoplankton growth in the *Kuroshio* extension. The maximum Chl *a* concentrations and cell abundance of large-sized phytoplankton following AM-dust addition were 1.3-fold and 1.7-fold larger, respectively, than those in the controls. In the YS, the increased P or N by AM-dust additions primarily contributed to the growth of phytoplankton. The Chl *a* concentrations in AM-dust treatments were generally higher than those in the controls, although the differences in maximum Chl *a* were negligible in both groups. The increase in the cell abundance of large-sized phytoplankton was < 1.7 -fold, compared with the controls. Comparing the difference in consumed N:P between the AM-dust

480 and N treatments, we found that the directly supplied bioavailable P by AM-dust in the incubated seawaters was not enough to support phytoplankton growth in the YS, which is characterized by P limitation, and in the subtropical gyre, which is characterized by NP co-limitation. This suggests that there are other sources of P in the incubations, which may be explained by the enhanced solubility of P from AM-dust and/or mineralization of DOP in the seawaters. Besides, the addition of AM-dust had a potential to shift the phytoplankton towards larger cells at all incubation stations, although it did not change the dominant taxa of phytoplankton assemblages. In general, larger positive responses of phytoplankton induced by combined nutrients than by single nutrient from the AM-dust were observed in our study. This is likely related to the varying nutrient levels, community structures of phytoplankton in the baseline seawaters, and the input of nutrients following AM-dust addition.

Our study proposes the importance of increase in bioavailable P stock for phytoplankton growth following AM-dust addition. This would help us better understand the effects of dust deposition on P cycles in the ocean. In view of the increasing anthropogenic N deposition in the NWPO and YS, due to continuously strong NO_x emissions in the eastern Asia (Kim et al., 2011; Kim et al., 2014), the increase in bioavailable P stock induced by dust deposition might be even more important in phytoplankton growth in the future. Moreover, further investigations are needed to better understand the differential effects of increase in bioavailable P stock as a result of the atmospheric deposition on phytoplankton growth in the coastal seas and open oceans. In terms of autotrophs, the stimulation of large-sized phytoplankton growth due to the input of AM-dust might enhance the carbon storage in the deep ocean, as the sinking rate of large-sized cells in the water column are higher than the pico-sized ones (Bach et al., 2012). On the other hand, the increasing anthropogenic activities in recent years can also increase the contents of heavy metal in the dust, which may inhibit phytoplankton growth (Miao et al., 2005; Paytan et al., 2009). However, the toxic effect of dust aerosol was not well reflected in our study. Therefore, further studies are also needed to illustrate the two-blade function of dust deposition to the marine ecosystems.

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690 **Table 1.** Experimental information and baseline conditions at the sampling stations

	Ar4 (S1)	(G7) S2	K4 (S3)	B7 (S4)	H10 (S5)
Sampling date	2014.3.23	2014.4.5	2014.4.14	2014.5.9	2014.4.29
Incubation time	10 days	9 days	9 days	10 days	9 days
Sampling location	29.5°N, 142.5°E	30.0°N, 148.5°E	34.0°N, 144.0°E	37.00°N, 123.17°E	35.0°N, 124.0°E
Sea surface temperature (SST, °C)	18.9	18.3	18.6	9.8	12.6
Mixed layer depth (MLD, m)	150	82	103	-	18
Chl a ($\mu\text{g L}^{-1}$)	0.24	0.50	1.28	1.58	2.74
NO₃⁻+NO₂⁻ ($\mu\text{mol L}^{-1}$)	0.26	0.04	0.51	3.20	0.28
PO₄³⁻ ($\mu\text{mol L}^{-1}$)	0.05	0.05	0.11	0.10	0.04
N:P ($\mu\text{mol}:\mu\text{mol}$)	5	<1	5	32	7
Micro-sized Chl a (%)	6	8	28	14	27
Nano-sized Chl a (%)	18	19	23	36	37
Pico-sized Chl a (%)	76	73	49	50	36
Large-sized phytoplankton abundance (cell mL⁻¹)	17	10	79	35	314
Diatoms (cell mL⁻¹)	12	6	74	16	286
Dinoflagellates (cell mL⁻¹)	5	4	5	19	28

Table 2. Treatments to the bioassay incubation experiments

Incubation No.	Treatments	Amended concentrations
1	Control	-
2	AM-dust	2 mg L ⁻¹
3	NaNO ₃	2 μmol L ⁻¹
4	NaH ₂ PO ₄	0.2 μmol L ⁻¹
5*	FeCl ₃	2 nmol L ⁻¹
6	NaNO ₃ +NaH ₂ PO ₄	2 μmol L ⁻¹ +0.2 μmol L ⁻¹
7	NaNO ₃ +NaH ₂ PO ₄ +FeCl ₃	2 μmol L ⁻¹ +0.2 μmol L ⁻¹ +2 nmol L ⁻¹

695 *Only at stations S2 and S4.

Table 3. Nutrient and soluble trace-metal contents in artificially modified and untreated dust, and **theoretically** increased corresponding concentrations in the incubated seawaters amended with AM-dust.

	Nutrients ($\mu\text{mol g}^{-1}$)				Soluble trace metals ($\mu\text{g g}^{-1}$)						
	NO_3^-	NO_2^-	NH_4^+	PO_4^{3-}	Fe	Mn	Cu	Cd	Pb	Co	Zn
AM-dust	532.9	34.2	10.3	4.3	473.1	413.5	0.23	0.04	0.24	2.58	4.27
Untreated dust	133.2	8.6	10.5	1.3	22.7	1.66	ND ^a	ND	ND	ND	0.03
Increased concentrations amended with AM-dust (nmol L^{-1})^b	1066	68.4	20.6	8.6	16.90	15.05	$7.2\text{E-}3^c$	$7.1\text{E-}4$	$2.3\text{E-}3$	0.09	0.13

700 ^a 'ND' means no detection.

^b Calculated by dividing the added amounts of nutrient and soluble trace metals with AM-dust input by the incubation volume of 20 L.

^c 'E-number' indicates ' $\times 10^{\text{number}}$ '.

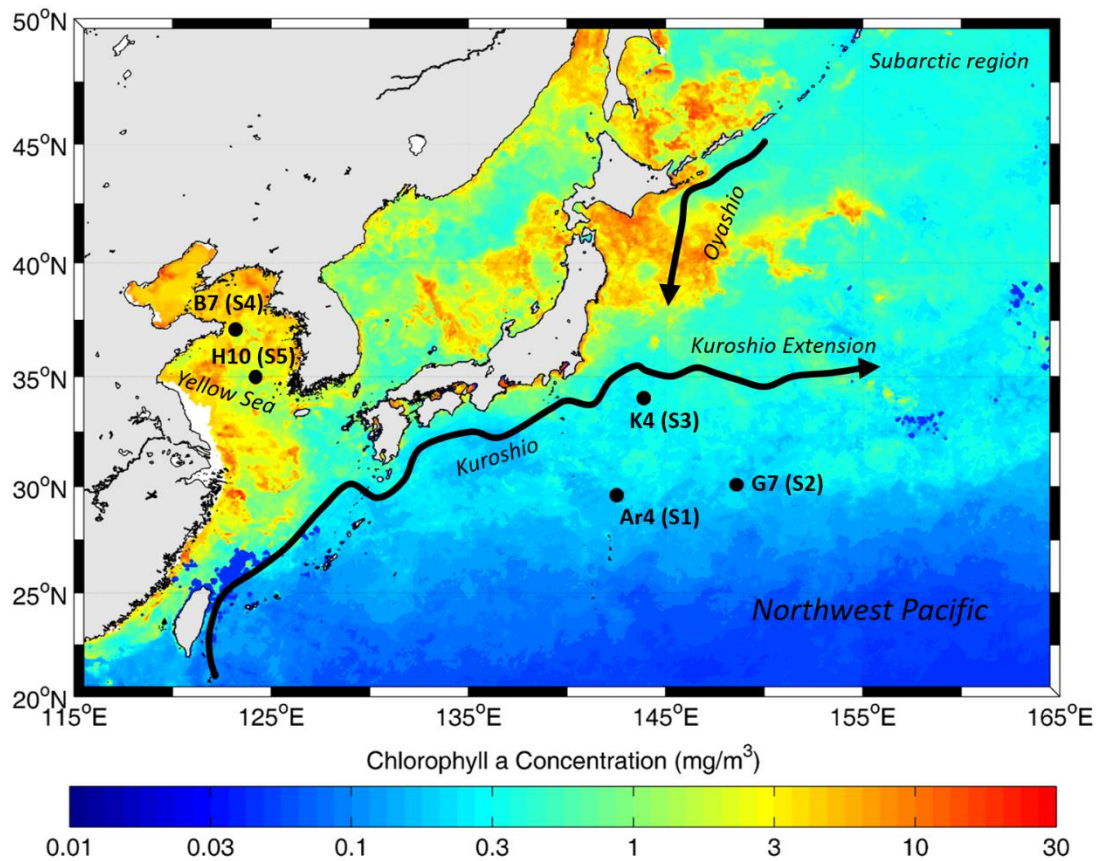


Figure 1. Locations of the five seawater-collection stations for the microcosm experiments (The base map reflects an average composite of Chl *a* concentrations in March-May, 2014 obtained from the NASA website (https://modis.gsfc.nasa.gov/data/dataproduct/chlor_a.php))

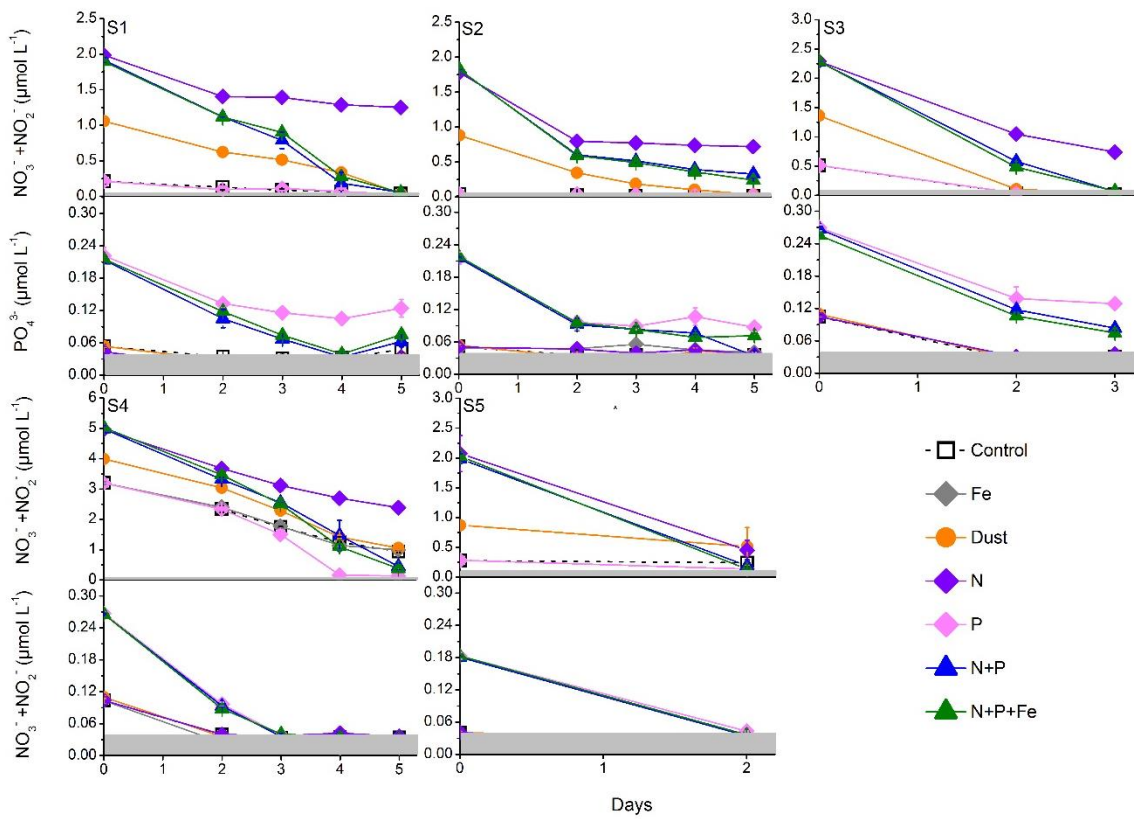


Figure 2. Changes in $\text{NO}_3^- + \text{NO}_2^-$ and PO_4^{3-} concentrations during the successive increase in the incubation period at each station. Shaded areas mean the values below the detection limit

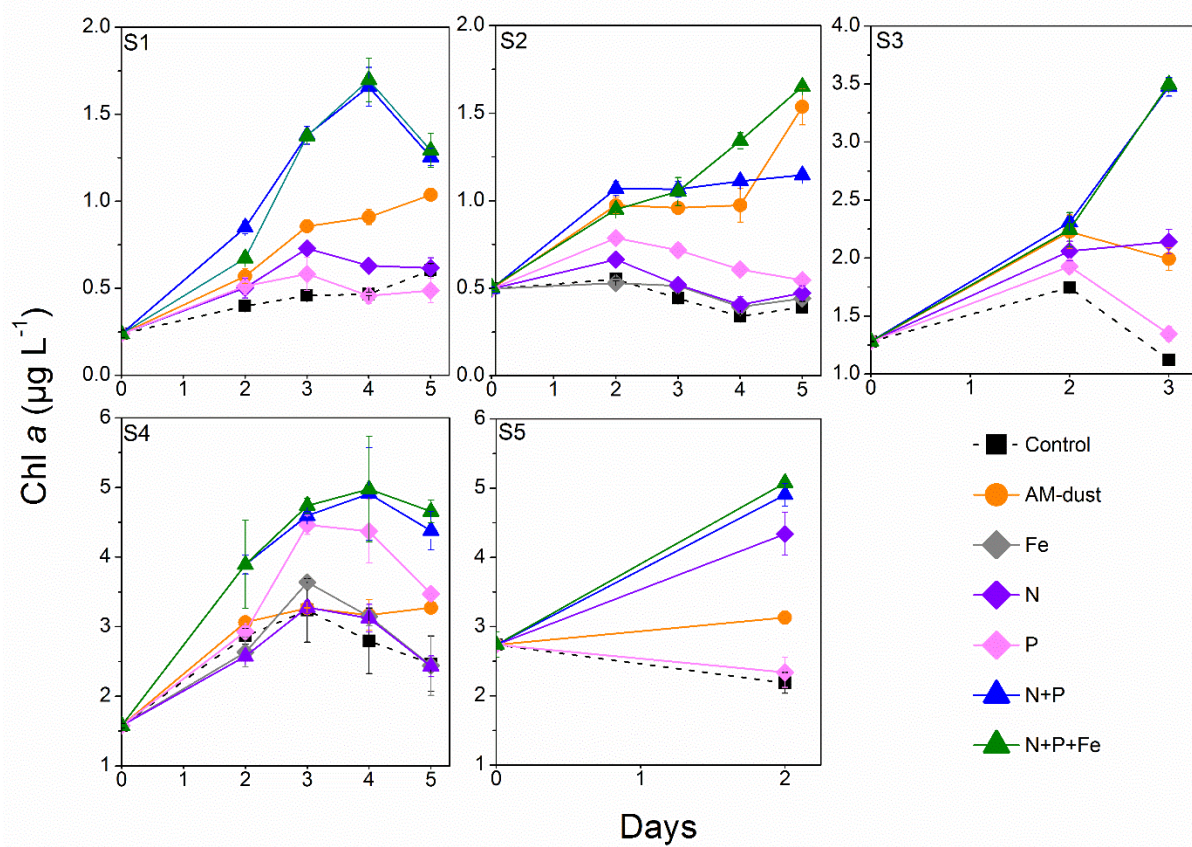
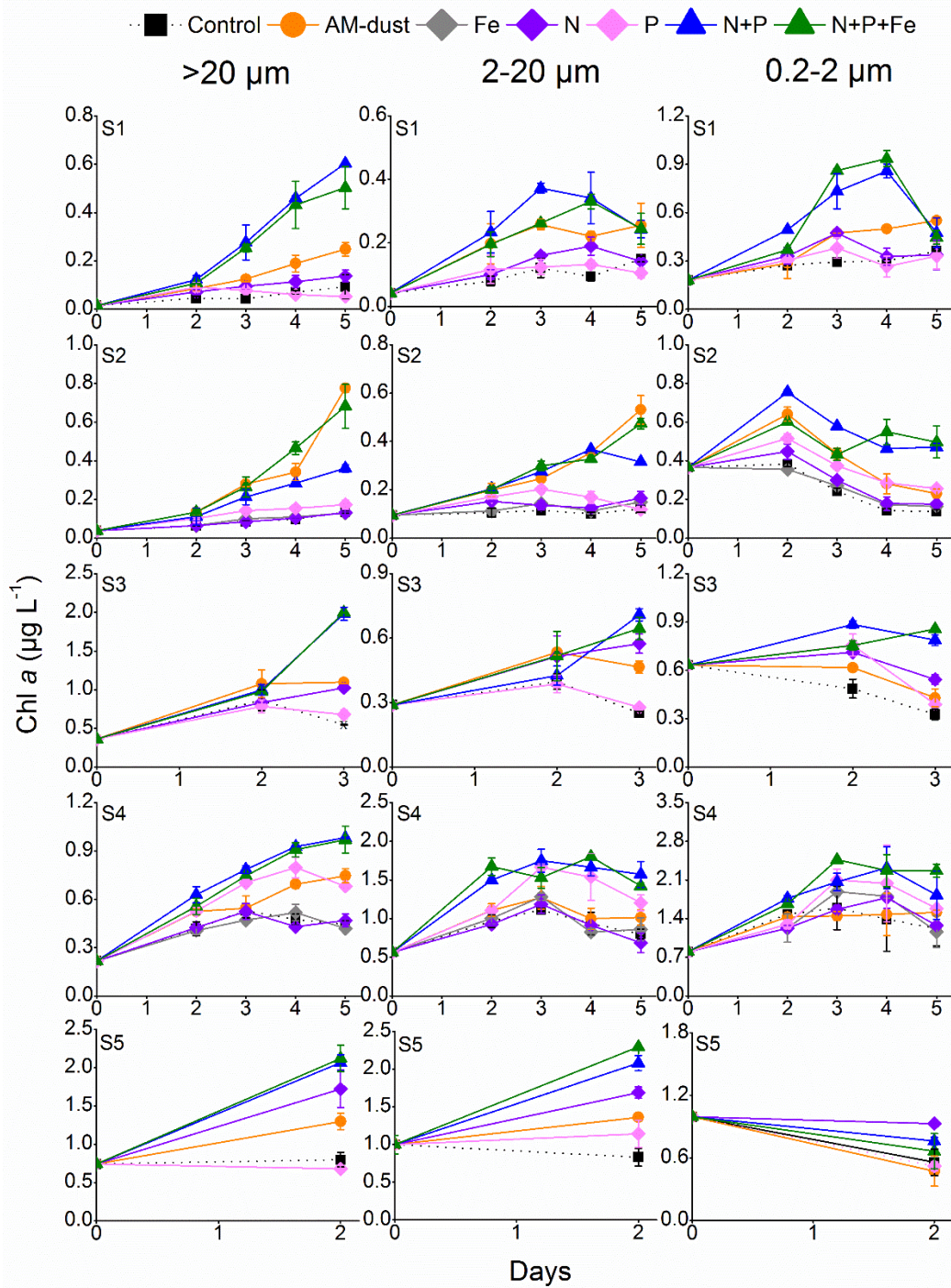
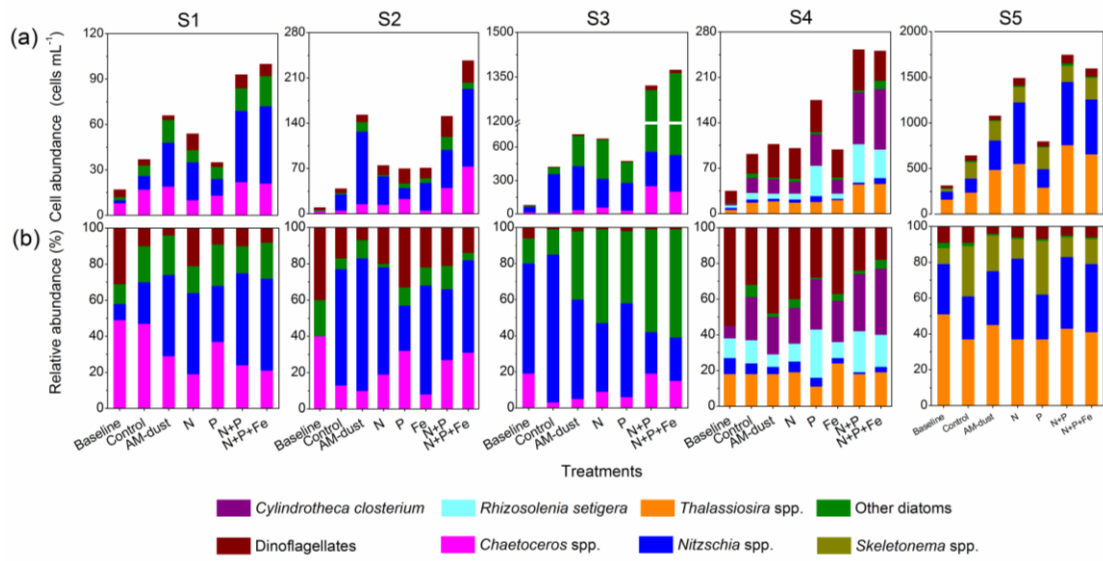


Figure 3. Changes in Chl *a* during the successive increase in the incubation period at each station



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Figure 4. Changes in size-fractionated Chl *a* during the successive increase in the incubation period at each station



725 **Figure 5.** Changes in the taxonomic structure of large-sized phytoplankton community before (i.e., ‘baseline’) and during (i.e., ‘control’ and various treatments) the successive increase in the incubation period at each station. (a) Cell abundance of large-sized phytoplankton. (b) Relative abundance, i.e., the contribution of specific species to the large-sized phytoplankton community in cell abundance

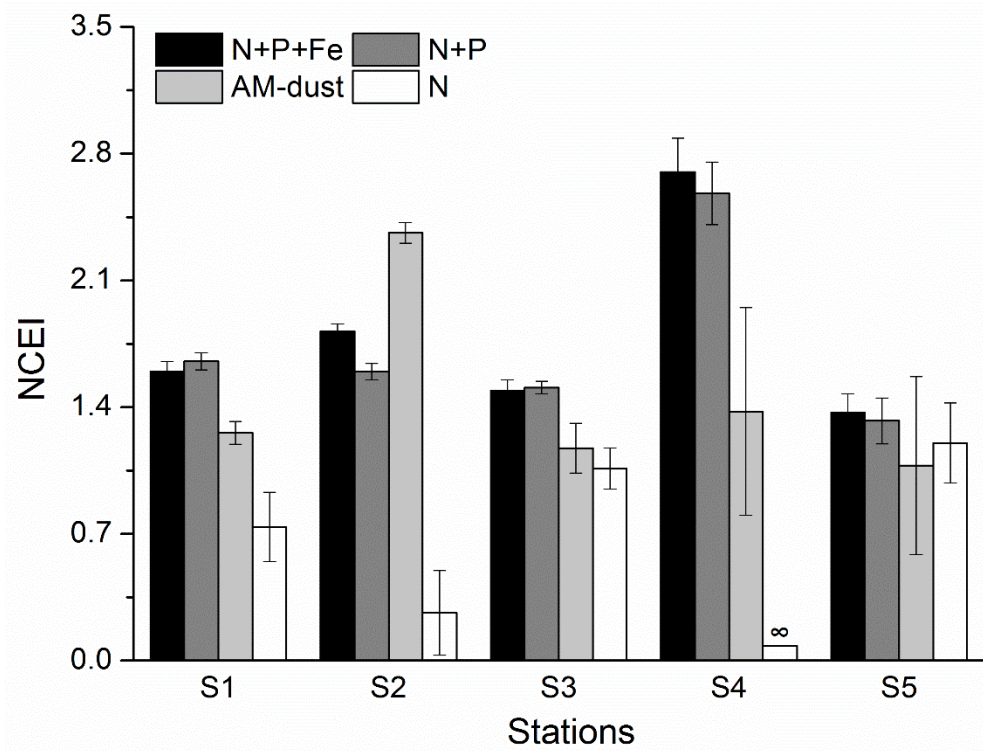


Figure 6. The net conversion efficiency index (NCEI) of N conversion to Chl *a* in AM-dust and various nutrient additions during the successive increase in the incubation period at each station. Error bars were calculated by propagating the standard deviation of NCEI in various treatments. "∞" indicates the standard deviation is at least threefold larger than the mean value of NCEI.

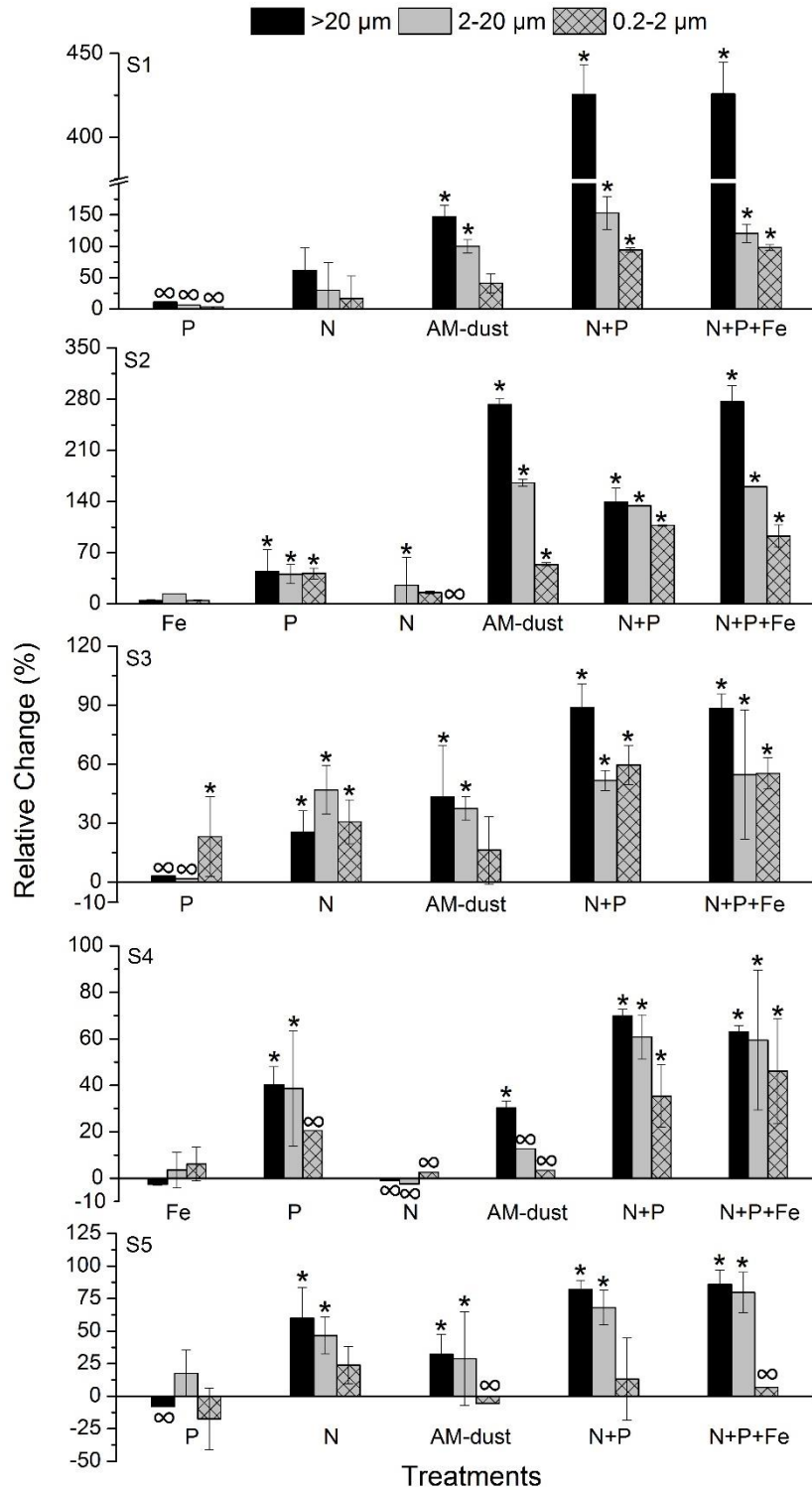
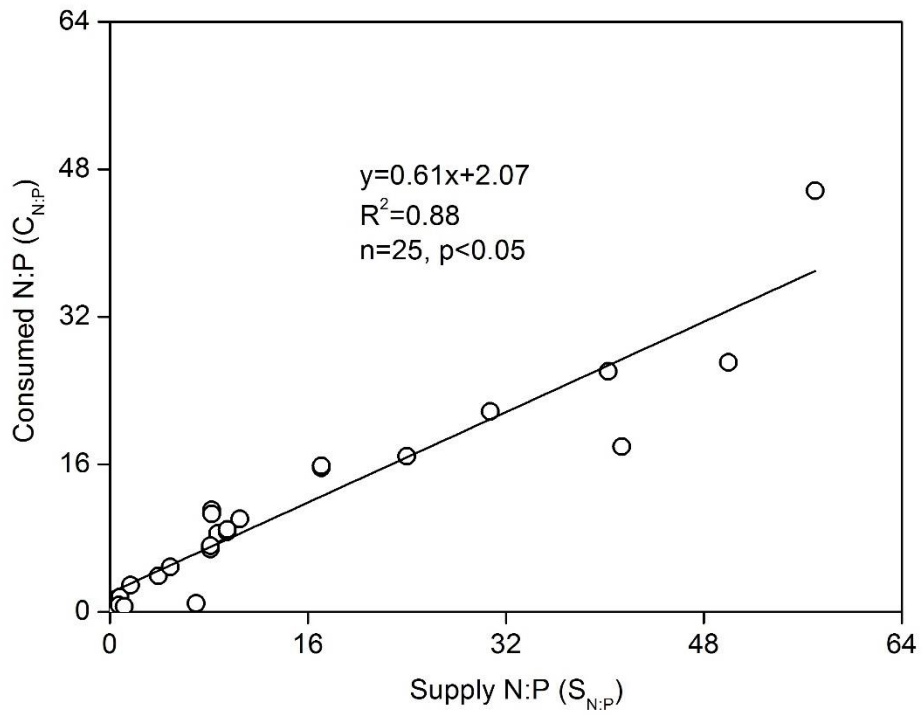


Figure 7. Relative change (%) in size-fractionated Chl *a* in AM-dust and various nutrient treatments during the successive increase in incubation period at each station. Relative change in this study was calculated as $100 \times ((\text{mean in treatments} - \text{mean in control}) / \text{mean in control})$. Error bars were calculated by propagating the standard deviation of mean values in the control and various treatments. “∞” indicates the standard deviation is at least threefold larger than the mean value of relative change. Asterisks indicate that the treatment induces a significant difference in the mean of Chl *a* compared with those in control ($p < 0.05$).

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Figure 8. The relationship between the consumed N:P ratio (C_{N:P}) and supply N:P ratio (S_{N:P}) in the control and the various nutrient treatments during the successive increase in the incubation period at all stations.

Supplementary information

750 **Text S1**

Simulation of Asian dust

When dust travels over the Asian continent through the atmosphere, it can experience mixing and deposition, as well as undergo chemical reactions (Formenti et al., 2011). The Asian continent suffers from air pollution to varying extents, with dynamically changing emissions of anthropogenic pollutants such as NO_x, SO₂, and NH₃ (Kim et al., 2014). The aging processes, i.e., the reactions of dust aerosols with anthropogenic pollutants, result in the Asian dust carrying a large amount of nutrients and bioavailable trace metals, prior to its deposition in the oceans. In this study, the aging process of dust followed Guieu's (2010) method and aimed at stimulating the cloud reaction between dust and synthetic evaporating cloud water. The pH around dust in the cloud process (i.e. mix with evaporating cloud water) was found to be as low as ~1 during their transport to the Yellow Sea (Meskhidze et al., 2003), whereas the typical pH in rainwater is 5 (Watanabe et al. 2001, Sasakawa and Uematsu, 2002, Wang et al. 2002, Sakihama et al. 2008, Zhang et al. 2011), meaning that a dilution by a factor of 10⁴. In consequent, in order to reproduce an evaporating cloud, we have used a concentration that is 10 000-fold larger in our experiments than the typical concentrations found in rainwater. Considering the typical concentrations of dust in rainwaters was 10 mg L⁻¹ (Ridame et al., 2002), the dust loading in evaporating cloud water could reach 100 g L⁻¹. As a consequence, all of the concentrations in evaporating cloud water were around 10000-fold larger (i.e. 4 orders of magnitude larger) than those in natural rainwater. Table S1 summarized the primary chemical composition of rains in the Eastern Asian regions and the evaporating cloud water used for our simulation. As the uptake of organic acidic gases during transport is complicated for Asian dust, we did not add oxalic acid, which was used for simulating the Saharan dust by Guieu et al. (2010), to simplify the reaction of dust surface and emphasize the importance of inorganic acids (H₂SO₄ and HNO₃) (Fan et al., 2006; Formenti et al., 2011; Shi et al., 2012).

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Table S1. Primary chemical composition of the rains in the eastern Asian region and the simulated eastern Asian cloud water.

	pH	NO ₃ ⁻ (M)	SO ₄ ²⁻ (M)
Reference eastern Asian rains*	3.89–7.61	10 ⁻⁵	10 ⁻⁵
Simulated cloud water	1**	10 ⁻¹	10 ⁻¹

*Sasakawa and Uematsu, 2002; Watanabe et al. 2001; Zhang et al. 2011; Sakihama et al. 2008; Wang et al. 2002.

** Meskhidze et al., 2003.

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Table S2. Recovery yield, accuracy, and detection limit for trace metal analysis

Metal	Detection limit ($\mu\text{g L}^{-1}$) [*]	Recovery (%)	RSD (%) ^{**}
Zn	0.012	90.6	3.17
Cu	0.226	95.2	2.09
Cd	0.016	88.5	0.87
Pb	0.019	93.2	2.93
Co	0.017	97.9	0.24
Fe	3.738	95.4	3.88
Mn	0.056	90.9	4.48

^{*} Detection limit was calculated as three times the standard deviation of the blank.

^{**} RSD means 'Relative Standard Deviation'.

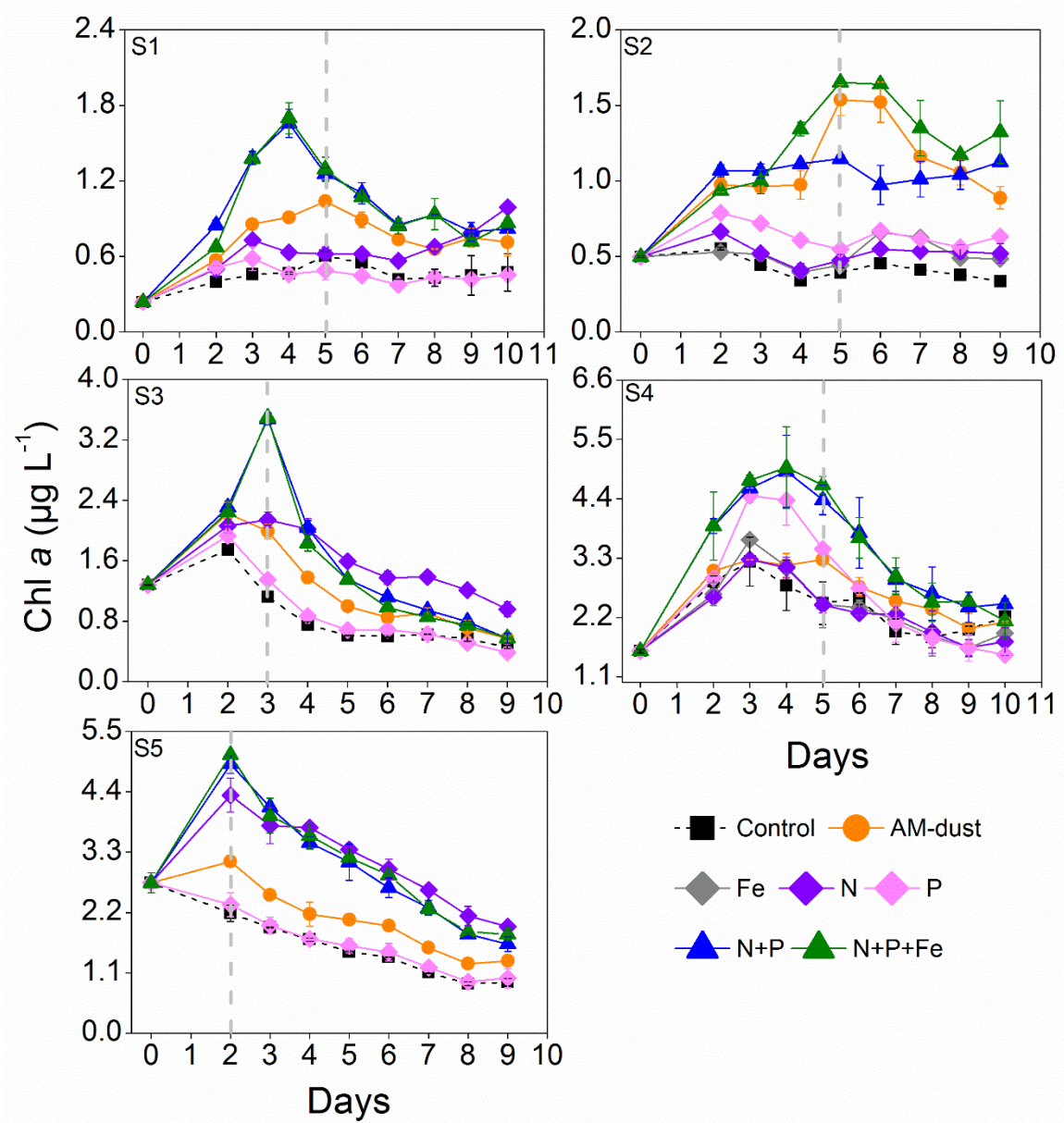
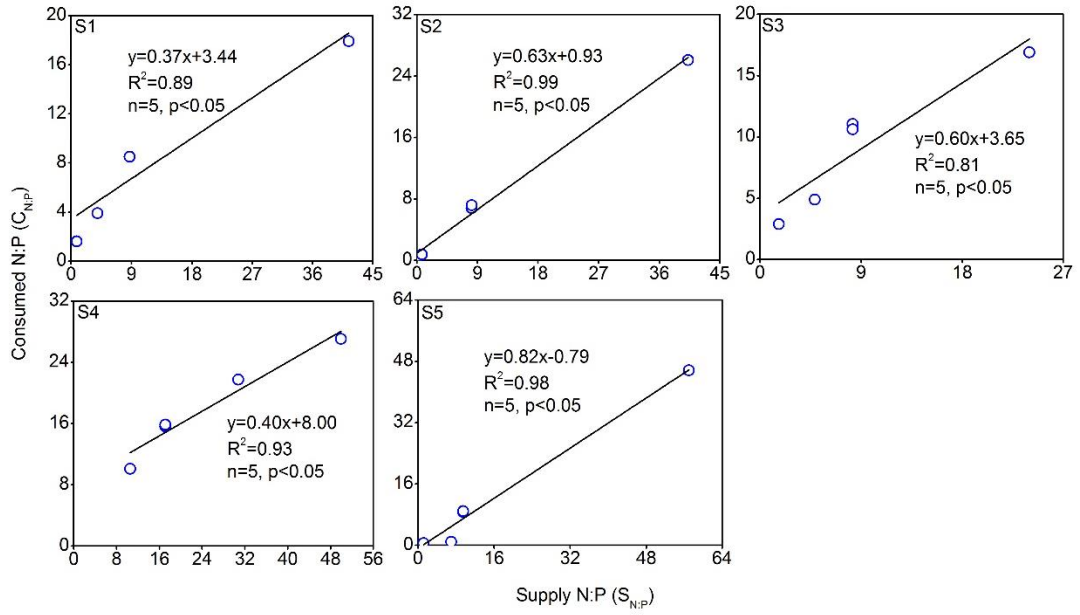


Figure S1. Changes in Chl *a* during the incubation experiments at each station. The successive increase during the incubation period in this study is identified by the dotted line.



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Figure S2. The relationship between the consumed N:P ratio ($C_{N:P}$) and supply N:P ratio ($S_{N:P}$) in the control and the various nutrient treatments during the successive increase in the incubation period at each station