



1 **Distinctive effects of allochthonous and autochthonous organic matter on CDOM spectra in a tropical lake**

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10 **Abstract**

11 Despite the increasing understanding about differences in carbon cycling between temperate and tropical freshwater systems,
12 our knowledge on the importance of organic matter (OM) pools on light absorption properties in tropical lakes is very scarce.
13 We performed a factorial mesocosm experiment in a tropical lake (Minas Gerais, Brazil) to evaluate the effects of increased
14 concentrations of allochthonous and autochthonous OM on the light absorption characteristics of colored dissolved organic
15 matter (CDOM), in addition to differences in light availability. We added inorganic nutrients (to stimulate OM
16 autochthonous production by phytoplankton growth) and allochthonous OM, and also introduced shading. Dissolved organic
17 carbon (DOC), Chl-*a*, nutrients, total suspended solid concentrations (TSM) and spectral CDOM absorption were measured
18 every three days to evaluate how they responded to these treatments. CDOM quality was characterized by spectral indices
19 ($S_{250-450}$, $S_{275-295}$, $S_{350-450}$, S_R and $SUVA_{254}$). The effect of carbon sources on the spectral CDOM absorption was investigated
20 through principal component (PCA) and redundancy (RDA) analyses. The two different OM sources affected CDOM quality
21 differently. Spectral indices ($S_{250-450}$ and S_R) were mostly affected by allochthonous OM addition. The PCA showed that
22 enrichment by allochthonous carbon had a strong effect on the CDOM spectra in the range between 300 and 400 nm, while
23 the increase of autochthonous carbon increased absorption at wavelengths below 350 nm. Our results show that small inputs
24 of allochthonous OM have much larger effects on the spectral characteristics on the lake CDOM, compared to large
25 production of autochthonous OM.

26 **Keywords:** dissolved organic matter; mesocosm; carbon cycling; tropical lake



27 1 Introduction

28 Organic matter (OM) consists of particulate organic matter (POM; organic compounds represented by aquatic communities
29 and detritus), and dissolved organic matter (DOM – in most of DOM studies it is the compounds smaller than 0.2 or 0.7 μm)
30 which is by far the largest pool of organic carbon in aquatic systems (Hedges, 1992). A better understanding of carbon
31 cycling in aquatic systems and its regional and global importance therefore requires knowledge of the biogeochemical
32 processes involved in the transformation, production and mineralization of DOM (Bertilsson and Tranvik, 2000;
33 Johannessen et al., 2007; Tranvik et al., 2009).

34 The main sources of DOM in aquatic ecosystems can be divided into two main categories: the allochthonous pool, which
35 comes from terrestrial vegetation and soil sources (Kieber et al., 2006; Miller et al., 2009), and the autochthonous pool which
36 is produced by aquatic primary producers (Kritzberg et al., 2004; Guillemette and Del Giorgio, 2012). These two DOM
37 pools have fundamental differences in their optical and chemical characteristics which in turn influence the mechanisms by
38 which DOM is degraded (Wetzel et al., 1995; Bertilsson and Tranvik, 2000). The allochthonous DOM is considered to be
39 more susceptible to photodegradation because it contains relatively large molecules with high numbers of aromatic
40 compounds which strongly absorb UV light (Amon and Benner, 1994; McKnight et al., 1994; Benner, 2002; Helms et al.,
41 2008). The autochthonous DOM originating from phytoplankton mainly consists of simple molecules (carbohydrates,
42 proteins, amino acids) of low molecular weight and is typically more labile for microbial community (Farjalla et al., 2009;
43 Fonte et al., 2013).

44 The rate of photodegradation depends on a combination of available sunlight and the chemical characteristics of DOM
45 (Benner, 2002), whereas the microbial degradation rate depends on the inherent DOM bioavailability and the utilization
46 efficiency of the bacterial community (Catalán et al., 2013; Asmala et al., 2014), and both are important processes that
47 transform and remove DOM in aquatic ecosystems (Roland et al., 2010; Mopper et al., 2015). Photodegradation is also
48 known to transform DOM to ammonia and other highly bioavailable compounds (Aarnos et al., 2012) which can be an
49 important nutrient supply for both phytoplankton (Hessen and Tranvik 1998) and heterotrophic bacterial communities
50 (Kieber et al., 1989; Miller et al., 2002; Lønborg et al., 2010). The microbial uptake of DOM by heterotrophic organisms
51 converts it to POM, which in turn can be assimilated by protozooplankton through the microbial loop (Azam et al., 1983).
52 Additionally, biodegradation of DOM can be stimulated by inorganic nutrients, mainly nitrogen and phosphorus, which
53 increase the bacterial growth efficiency (Zweifel et al., 1995; Asmala et al., 2013) by reducing the energetic cost of substrate
54 acquisition (Hopkinson et al., 1998). In tropical lakes, aquatic processes including mineralization of organic compounds
55 occur more rapidly than in temperate lakes due to high temperatures and light availability throughout the year (Marotta et al.,



2010). However, there are only few studies on the photochemical (Teixeira et al., 2013; Bittar et al., 2015) and bacterial (Farjalla et al., 2002; 2009; Roland et al., 2010) degradation of DOM in tropical environments compared to temperate freshwater systems and estuaries (Bertilsson and Tranvik, 2000; Anesio and Granéli, 2003; Boreen et al., 2008; Asmala et al., 2014; Attermeyer et al., 2015).

The drivers of DOM dynamics in temperate environments are different from those in tropical environments, because in the first one the allochthonous contribution is related to changes in temperature (such as by the flow of the melting water in the surroundings or by the destratification of lakes; Lindell et al., 2000) and in the tropics the seasonality of rainfall is the main driver (Suhett et al., 2006). Brazil has a variety of complex freshwater systems that behave in different ways regarding the temporal dynamics of DOM. For example, in most tropical rivers and lakes, the seasonal allochthonous contribution occurs via runoff in the rainy season (between September and April), raising humic carbon concentrations and water color (Farjalla et al., 2002). In some regions, such as the complexes of Brazilian rivers and Amazonian lakes, the contribution of allochthonous material is related to the hydrological pulse, which raises the level of the water invading the surrounding forests (Amado et al., 2006).

The lake system of the Middle Rio Doce is composed of about 300 natural lakes and is among the three most important in Brazil, behind the Amazonian and Pantanal basins (Maillard et al., 2012). Recent studies in this region have shown that the marked seasonality in the inputs of allochthonous material (nutrients and organic matter) during periods of rain (thermal stratification period, summer) plays a key role in the pattern observed for the optical characteristics of lakes, for example the transparency to photosynthetically active radiation (PAR) and ultraviolet (UV) (Gagliardi, 2015; Brandão et al., 2016). Contrary to expectations, greater transparency of these lakes is observed during the rainy season, since the allochthonous material remains trapped in the hypolimnium by temperature difference, until the water mixture (dry season, winter) redistributes it throughout the water column (Reynolds et al., 2009; Brandão et al., 2016). In this context, higher net phytoplankton production rates occur in the mixing periods, with lower solar radiation incidence and lower transparency (Brighenti et al., 2015). Bezerra-Neto et al. (2006) showed a strong negative influence on the concentration of CDOM (chromophoric dissolved organic matter) and transparency to the PAR radiation in a set of lakes of this lacustrine system, which emphasizes the importance of the chromophoric carbon from allochthonous origin for the physical and chemical conditions of the lakes, and consequently seasonal dynamics of phytoplankton and aquatic metabolism. In this way, the physical, chemical and ecological balance of these lakes is extremely related to the inputs of nutrients and organic matter from the catchment during rainy season. However, the frequency and intensity of precipitation events in this region has been altered (Roland et al., 2012). In addition, the Atlantic Forest is a threatened and extremely devastated biome (Myers et al.,



2000) and most of the lakes have already had the surrounding forest replaced by eucalyptus and pasture plantations. Land use transformation and the disruption of biogeochemical cycles may change the amount and quality of the inputs of dissolved nutrients (such as phosphate and nitrogen compounds) and organic matter into the lake (Vitousek et al., 1997; Pinheiro et al., 2015), changing the balance between allochthonous and autochthonous DOM sources in the systems and consequently DOM degradation ways.

Some recent studies have demonstrated that DOM transformations (such as the the effect of photodegradation and biodegradation on the absorption properties of CDOM) was not constant over the spectral range, thus influencing the shape of the absorption curve (Helms et al., 2013; Reader et al., 2015). As modifications in the spectral shape reflect underlying changes in the carbon compounds at the molecular level, studies on biological and chemical effects on CDOM spectra allows a better understanding of the DOM transformations and how this links to overall carbon cycling in aquatic ecosystems (Stubbins et al., 2014).

In order to investigate how the optical properties of the lakes change due to more common anthropic impacts (such as eutrophication, land use change) and changes in rainfall recently observed in the Middle Rio Doce region, we performed a mesocosm experiment in a tropical lake, manipulating nutrients, OM and light conditions. We expected that addition of nutrients will cause algal growth, which in turn would increase the production of autochthonous DOM (Schindler, 1977; Lean and Pick, 1981) until nutrients become exhausted. In contrast, the addition of extracted OM from leaves of the native forest was done to increase the concentration of allochthonous DOM. In this context, we hypothesize that OM from different sources will change the CDOM absorption spectra and the associated indices differently over time. We also investigated the effect of the two interacting OM sources on the optical characteristics of DOM. We have included different levels of exposure to sunlight in the mesocosms experiment as a third variable due to its important effect for both production (by phytoplankton, an autochthonous source) and degradation (photodegradation) of DOM.

106

2 Methods

2.1 Study area

This study was conducted in Carioca Lake (19°45'26.0"S; 42°37'06.2"W), one of the approximately 300 natural lakes of the Middle Rio Doce lacustrine system, the third most important of Brazil. It is located in the southern part of the State Park of Rio Doce (PERD, Minas Gerais, Brazil) which is the largest remnant of the Atlantic Forest in Minas Gerais (36000 ha forest) with lakes occupying 9.8% (3530 ha) of its total area. Carioca Lake is surrounded by secondary Atlantic Forest and is a warm-monomictic lake with a mixing period during the dry winter (May to August) and thermal stratification during the rest



of the year (September to April). It is a mesotrophic lake (total phosphorus: $5.6\text{--}21.4\ \mu\text{g L}^{-1}$, epilimnion annual average $13.9\ \mu\text{g L}^{-1}$; chlorophyll-*a*: $1.3\text{--}16.6\ \mu\text{g L}^{-1}$, epilimnion annual average $7.7\ \mu\text{g L}^{-1}$) (Petrucio et al., 2006; Brighenti, 2014), with $1718.6\ \text{m}$ of perimeter, $14.1\ \text{ha}$, $671 \times 10^3\ \text{m}^3$, and maximum depth of $11.8\ \text{m}$ and average depth of $4.8\ \text{m}$ (Bezerra-Neto et al., 2010). Carioca is one of the lakes that have been monitored for water quality and aquatic biota since 2000 through the Brasil-LTER Programme (PELD-CNPq Proc. 403698/2012-0). It is known that the input of DOM and nutrients in Carioca Lake occurs via runoff during the rainy season, but most of them remain below the thermocline due to temperature differences (Reynolds, 2009). During the mixing period, DOC, CDOM and nutrients become distributed in the water column, increasing their concentrations and availability in the surface layer, which in turn contributes to increased primary production and respiration rates (Brighenti et al., 2015).

2.2 Experimental design and measurements

To test the effect of organic matter inputs, sunlight, and nutrients on DOM degradation, we conducted an *in situ* experiment using a total of 16 cylindrical mesocosms (diameter $1.3\ \text{m}$, height $1.5\ \text{m}$ and volume $2\ \text{m}^3$) with eight different combinations (two replicates for each combination). The sampling was carried out between January 20th and February 1st 2015 with daily measurements occurring between 10:00 am and 12:00 pm. Water samples ($3\ \text{L}$ at $0.5\ \text{m}$ from the surface) from mesocosms were collected every three days. The mesocosms setup was based on a $2 \times 2 \times 2$ factorial design as follows: 1) with and without addition of organic matter extracted from leaves surrounding the lake, 2) with and without addition of nutrients (NaNO_3 , K_2HPO_4 , NH_4Cl) and with and without 50% of shading of solar radiation (Fig. 1).

The organic matter added to the mesocosms was a mixture of leaves, plant detritus and soil particles adhered to this material from the ground around the lake (4 cylinders, ca. $20\ \text{L}$ each). The material was placed in buckets with distilled water ($60\ \text{L}$) for decomposition and stored in the laboratory under room temperature (ca. $25\ ^\circ\text{C}$). After one week the water was filtered with $20\ \mu\text{m}$ mesh and $7.5\ \text{L}$ of the filtrate was added to each of the 8 units in order to increase allochthonous organic matter availability in these mesocosms. The initial organic matter concentration in each of the 8 carbon amended cylinders was $8.6 \pm 0.1\ \text{mg L}^{-1}$ (DOC, mean \pm standard deviation), $21.44 \pm 0.52\ \text{m}^{-1}$ ($a_{\text{CDOM}254}$, mean \pm standard deviation) and $0.66 \pm 0.04\ \text{m}^{-1}$ ($a_{\text{CDOM}440}$, mean \pm standard deviation). In the 8 units without organic matter addition the initial organic matter was $8.0 \pm 0.4\ \text{mg L}^{-1}$ (DOC, mean \pm standard deviation), $17.11 \pm 0.43\ \text{m}^{-1}$ ($a_{\text{CDOM}254}$, mean \pm standard deviation) and $0.43 \pm 0.04\ \text{m}^{-1}$ ($a_{\text{CDOM}440}$, mean \pm standard deviation). We performed a pilot mesocosm experiment in July 2014 to study the suitable OM addition levels for the experiment. In this pilot study, we noticed that this methodology and volume of extract was enough to modify the quality of organic matter and also light attenuation levels in the organic matter added treatment (with added OM: $a_{\text{CDOM}254}\ 22.31\ \text{m}^{-1}$, $K_{\text{dPAR}}\ 1.68\ \text{m}^{-1}$; without OM addition: $a_{\text{CDOM}254}\ 16.67\ \text{m}^{-1}$, $K_{\text{dPAR}}\ 0.99\ \text{m}^{-1}$).



143 Mesocosms were submerged on the surface of the lake and filled with lake water. Mesocosms with reduced light availability
144 (SH) were shaded with spectrally neutral shading screens (50%) and only opened for quick samplings and measurements.
145 Every day, the mesocosms were gently stirred and measured for water temperature using a probe Hydrolab DS5 (Hach Inc.).

146 Water samples were filtered immediately after sampling for Chl-*a* and nutrients (0.47 µm filter) and TSM (AP040 filter).
147 The filters were kept frozen until analysis. Water samples were also filtered for analysis of DOC and CDOM (0.22 µm
148 Millipore filter) and stored in amber glass bottles (pre-washed with distilled water and hydrochloric acid 10%) at 4°C in the
149 dark. The Chl-*a* concentration corrected by pheophytin (µg L⁻¹) was obtained by acetone extraction (90%) measured in a
150 spectrophotometer (UV-VIS Shimadzu) at 665 and 750 nm and calculated using the protocol provided in APHA (1998). The
151 TSM (mg L⁻¹) were determined by the gravimetric method, considering the difference between the dry weights of AP40
152 Millipore filters (105 °C for 2 hours) before and after the filtration of water samples (APHA 1998). The DOC concentration
153 (mg L⁻¹) was obtained by catalytic oxidation method of high temperature using TOC Analyzer (Shimadzu TOC – 5000A).
154 Filtered water samples were taken for dissolved nutrients (ammonia, nitrate, nitrite and phosphate; µg L⁻¹) and frozen until
155 analyzes with an auto-analyzer (Metrohm 8000 IC-Plus).

156 2.3 CDOM optical properties

157 Absorption spectra of CDOM were obtained between 250 and 700 nm at 1 nm intervals with a spectrophotometer (UV-VIS
158 Shimadzu) using a 5 cm quartz cuvette and a Milli-Q water sample as blank reference. The absorption spectra of each
159 sample were measured in replicate (standard deviation < 0.01). The absorption coefficients ($a_{\text{CDOM}}(\lambda)$; m⁻¹) were derived
160 from absorbance measurements according to the equation $a_{\text{CDOM}}(\lambda) = 2.303A(\lambda)L^{-1}$, where $A(\lambda)$ is the absorbance measured
161 at wavelength λ and L is the optical path of the cuvette (in meters). Absorption coefficients were corrected for backscattering
162 by subtracting the value of the coefficient at 700 nm. The absorption coefficient at 254 nm ($a_{\text{CDOM}254}$) was used as an index
163 of CDOM UV-absorption and at 440 nm ($a_{\text{CDOM}440}$) as a CDOM PAR-absorption.

164 We used a simple exponential curve to model the decrease in absorption with increasing wavelength using the equation
165 (Jerlov, 1968; Bricaud et al., 1981; Stedmon and Markager, 2001): $a_{\text{CDOM}}(\lambda) = a_{\text{CDOM}}(\lambda_0) e^{-S(\lambda - \lambda_0)} + K$, where a_{CDOM} is the
166 absorption coefficient (m⁻¹), λ is the wavelength (nm), λ_0 is a reference wavelength (nm), K is a background constant (m⁻¹)
167 accounting for scatter in the cuvette and drift of the instrument and S is the spectral slope (nm⁻¹) that describes the
168 approximate exponential rate of decreasing absorption with increasing wavelength. Furthermore, we calculated the spectral
169 slope between 275-295 nm ($S_{275-295}$) and 350-400 nm ($S_{350-400}$). The slope ratio (S_R , Helms et al., 2008) was obtained by
170 dividing $S_{275-295\text{nm}}$ by $S_{350-400\text{nm}}$. These metrics were calculated using the *cdom* R package (Massicotte, 2016). We also



171 calculated $S_{250-450}$ as a proxy to measure changes in the entire spectrum including UV and PAR-absorbing (we limit to 450
172 nm because at higher wavelengths the signal to noise ratio was high). This metric was further used in the principal
173 component analysis. The specific UV absorbance ($SUVA_{254}$) ($m^2 mg^{-1} C$) was calculated dividing the value of the absorption
174 coefficient at 254 nm (m^{-1}) by the concentration of DOC ($mg L^{-1}$) (Weishaar et al., 2003).

175 2.4 Data analysis

176 The mesocosms were grouped in the figures as follows in order to show the differences between the two different sources of
177 OM (allochthonous versus autochthonous source): the group “OM addition” includes the OM, OMNUT, OMSH and
178 OMNUTSH treatments combined. The group “nutrients addition” includes the NUT, OMNUT, NUTSH, OMNUTSH
179 treatments and the group “with shade” include the SH, OMSH, NUTSH and OMNUTSH. The last three groups “without
180 OM addition”, “without NUT addition” and “full light” include the remaining four treatments, totaling six different groups
181 (see Table S1 in the Supplementary Material). In this context, the group “OM addition” represents the allochthonous source,
182 while the group “nutrients addition” represents the autochthonous OM source.

183 The relative changes (%) of the parameters over time were calculated dividing the value measured at the end of the
184 experiment by the value at the beginning (day 0) of the experiment, after subtracting this result from 1 and multiplying by
185 100 $[(1 - end/start) \times 100]$. Negative relative change values indicate decrease and positive values indicate increase compared
186 to initial values in the beginning of the experiment.

187 We performed a three-way ANOVA plus the second order interactions, in order to verify the effect of each factor (nutrients
188 and OM additions and shade) on the response variables (quantity [$a_{CDOM254}$, $a_{CDOM440}$ and DOC] and quality [$SUVA_{254}$ and
189 spectral slopes] of DOM and phytoplankton biomass [measured by Chl-*a* concentration]). We also partitioned the coefficient
190 of determination of each ANOVA in terms of its factors and interactions in order to estimate the relative importance of each
191 driver on the results for DOM optical properties (Lindeman et al., 1980). The interactions that were worth to mention we
192 compute the least-squares means, plus the 95% confidence interval, for specified factors combinations. All statistical
193 analysis were performed in software R (R Core Team, 2017), plus packages “lsmeans” (Lenth, 2016) and “relaimpo”
194 (Grömping, 2006). A threshold significance level of 5% was considered.

195 A principal component analysis (PCA) was carried out using CDOM absorption spectra (with 1 nm interval between 250
196 and 450 nm) on a $n \times p$ matrix where n is the number of observations in the dataset ($n = 80$) and p is the wavelength number
197 ($250 \leq p \leq 450$). The PCA was performed on scaled data (zero mean and unit variance) as suggested by Borcard et al.,
198 (2011). This approach was used to summarize absorption spectra (Reader et al., 2015), which are difficult to summarize into



a single value observation such as other variables like Chl-*a*. Based on the Kaiser-Guttman criterion (Kaiser, 1960), the first two principal components were kept for subsequent analyzes. PCA scores of first two axes (PC1 and PC2) were correlated against environmental variables (categorical variables shading, nutrients and OM additions and the continuous variables Chl-*a*, TSM, DOC, S_R and $SUVA_{254}$) using a redundancy analysis (RDA) to help to understand their interpretation. All statistical analyzes were performed using R Software (R core development team 2011).

3 Results

3.1 Characteristics of the treatments

During the experiment, the water temperature of the mesocosms ranged between 28.4 and 31.3 °C (average 30.8 °C).

The results obtained by Three Way Analysis of Variance (Table 1) showed that Chl-*a* levels were significantly higher for treatments with addition of nutrients (average 11.9 $\mu\text{g L}^{-1}$), and also with addition of OM (average 9.2 $\mu\text{g L}^{-1}$) (Fig. 2A-B). Concentrations of DOC and absorption coefficients of $a_{CDOM254}$ and $a_{CDOM440}$ were significantly higher in all the treatments with addition of OM (averages DOC – 8.6 mg L^{-1} ; $a_{CDOM254}$ – 21.9 m^{-1} ; $a_{CDOM440}$ – 0.7 m^{-1}). Lower $S_{275-295}$ (average 0.02 nm^{-1}) and higher $SUVA_{254}$ (average 2.5 $\text{m}^2 \text{mg}^{-1} \text{C}$) were observed also in the treatments with OM added. Treatments shaded showed lower DOC concentrations (average 8.2 mg L^{-1}), higher $SUVA_{254}$ (average 2.4 $\text{m}^2 \text{mg}^{-1} \text{C}$) and lower $S_{275-295}$ (average 0.02 nm^{-1}). The results also showed significant effect for the interaction between nutrients and shade for $a_{CDOM254}$ and $a_{CDOM440}$ (Table 1 and Fig. S1 and S2 in the Supplementary Material).

The partitioned coefficient of determination (for $a_{CDOM254}$, $a_{CDOM440}$, $SUVA_{254}$ and $S_{275-295}$ were very high (>75%), indicating that the most important drivers for these variables were included in the experiment. For DOC and Chl-*a* the coefficient of determination showed intermediate values (~50%), indicating that other variables not considered in the experiment also played an important role. The difference between treatments with and without OM played the biggest role in explaining the variation of all response variables evaluated in the experiment, except Chl-*a*. As expected, the difference in nutrients had the biggest effect over Chl-*a* concentrations.

3.2 Temporal changes in the mesocosms

Average phytoplankton biomass (Chl-*a*) increased over time only in the treatments with addition of nutrients (from average of 3.9 on day 0 to 19.1 $\mu\text{g L}^{-1}$ on the 12th day) and OM (from minimum average 3.4 on day 0 to maximum of 12.1 $\mu\text{g L}^{-1}$ on 9th day) (Fig. 2A-B). Average DOC increased in the experimental units with added nutrients until day 9 (maximum 8.7 mg L^{-1}). In treatments with addition of OM, DOC was higher as expected with a minimum of 8.3 mg L^{-1} at day 3 and maximum of



226 8.9 mg L⁻¹ on the 9th day, while in the treatments without addition of OM the DOC had a maximum of 8.2 mg L⁻¹ in 6th day
227 (Fig. 2C-D). $S_{250-450}$ and slope ratio (S_R) had the same pattern in the treatments with and without nutrients addition, but
228 decreased consistently in treatments with OM addition (Fig. 2 E-H).

229 The relative changes in CDOM absorption along the spectral range were different for each sampling day (Fig. 3A-E). On the
230 initial day, only treatments with and without addition of OM had distinct absorption curves, especially in the UV range
231 below 400 nm, and the absorption spectra for each treatment group on day 0 are shown in Fig. 3A. To evaluate treatment
232 effects we determined the change in light absorption spectra for the other sampling days relative to the initial day (Fig. 3B-
233 E). On day 3, treatments with and without nutrients added were quite similar, while those with and without OM and with and
234 without shading showed opposing changes. Loss of absorption occurred only in treatments with full light (less than 5%
235 between 300-420 nm) and in those without OM addition (the loss of absorption increased with the increase of wavelength)
236 (Fig. 3B). On day 6 all treatments showed an increased absorption especially after 350 nm (higher increase with shade: ~
237 40% at 450 nm), except the treatments with full light that still had a loss of absorption (Fig. 3C). On day 9 and 12, all
238 treatments had a loss of CDOM absorption with increasing wavelengths, especially for the full light treatment (Fig. 3D-E).

239 The concentrations of DIP and DIN ($\mu\text{g L}^{-1}$) and the molar ratio between them (DIN:DIP) were higher in treatments with
240 added nutrients as expected (Fig. 4A-C). In the treatments without nutrients addition, DIP concentrations ranged between 3
241 and 9 $\mu\text{g L}^{-1}$ and DIN between 1.7 and 100.4 $\mu\text{g L}^{-1}$, with DIN:DIP ratios ranging between 7.6 (day 12) and 21.3 (day 0) and
242 below 13.9 after day 3 suggesting that phytoplankton community were mostly limited by nitrogen in these treatments (Redfield 1958; Reynolds 1999). All the dissolved nutrients decreased over the course of the experiment in all treatments
243 with nutrients addition (DIP decreased from an average of 153 to 59 $\mu\text{g L}^{-1}$ and DIN from 2600 to 400 $\mu\text{g L}^{-1}$). The DIN:DIP
244 molar ratio ranged from 55.3 (day 3) to 12.2 (day 12) in the mesocosms with addition of nutrients.

246 3.3 Principal component analysis results

247 The first principal component of the redundancy analysis (Fig. 5A) was mostly associated with availability of OM. Samples
248 presenting high scores on the first principal component furthermore tended to have high values of DOC and SUVA_{254} but
249 low values of S_R . High scores in the second principal component correlated positively with Chl-*a*, nutrients and TSM and
250 negatively with shading.

251 Exploration of spectral PCA loadings (Fig. 5B-C) revealed that principal component 1 (PC1) had the strongest effect on the
252 shape of CDOM absorbance between 300 and 400 nm. Principal component 2 (PC2) loadings showed a quasi-linear decrease
253 with increasing wavelength suggesting that phytoplankton enrichment had a stronger effect at lower wavelengths.



254 Furthermore, loading values were negative after ~340 nm, indicating that phytoplankton was on average lowering CDOM
255 absorption after this threshold. Based on the redundancy analysis, PC1 was renamed “allochthonous carbon enrichment”
256 whereas PC2 was renamed “autochthonous carbon enrichment”.

257 4 Discussion

258 4.1 Nutrients, allochthonous matter and shade responses in DOM and Chl-*a*

259 Our study supports previous findings showing that the addition of allochthonous matter results in a DOM pool which is
260 dominated by more aromatic carbon with higher molecular weight (Bertilsson and Tranvik, 2000; Benner, 2002) and lower
261 spectral slopes (Helms et al., 2008; Fig. 2F, 2H). Addition of nutrients also affected DOM quantity and quality related to
262 autochthonous production by phytoplankton growth which can be an important source of DOM (Zhang et al., 2009; Zhang et
263 al., 2013; Brandão et al., 2016). In the treatments without addition of nutrients, the phytoplankton community was limited by
264 nitrogen since the beginning of the experiment ($\text{DIN} < 100 \mu\text{g L}^{-1}$; Reynolds, 1999). The DIN:DIP molar ratio in these
265 treatments indicates that nitrogen was consumed very fast and was limited after day 3 (ratio below 13.9; Redfield, 1958).
266 With the addition of nutrients, the concentrations of DIN and DIP were higher than what is considered a limiting condition
267 proposed by Reynolds (1999). However, the molar ratio values proposed by Redfield (1958) suggest limitation by
268 phosphorus (molar ratios between 21.3 and 55.3) during the experiment and only becoming limited by nitrogen following
269 day twelve (molar ratio 12.2). Moreover, these apparent limitations did not restrict the phytoplankton growth in mesocosms
270 with addition of nutrients. Mesocosms which did not receive nutrients, but which received allochthonous OM also
271 experienced an increase in Chl-*a* after 3 days (Fig. 2B). The light attenuation caused by enhanced light absorption from
272 added allochthonous OM (Kirk, 1994), may have favored growth of phytoplankton by reducing photoinhibition, known to
273 occur in Lake Carioca (Brighenti et al., 2015). The increase of phytoplankton in these treatments therefore suggests that algal
274 growth was stimulated by a combination of increases in nutrient availability due to degradation of the OM added (Hessen
275 and Tranvik, 1998). These results indicate that the seasonal allochthonous input in tropical lakes, in addition to the nutrient
276 increment, also controls the observed seasonality in water transparency, influencing the seasonal dynamics of phytoplankton.
277 It therefore corroborates previous studies that have suggested that aquatic metabolism and consequent effects on organisms
278 at higher trophic levels are indirectly controlled by terrestrial material during periods of higher rainfall (Brighenti et al.,
279 2015; Brandão et al., 2016).

280 Although additions of allochthonous OM and nutrients both contributed to higher DOC concentrations, divergent effects of
281 these additions were evident in the quality of carbon assessed by optical indices ($S_{250-450}$ and S_R).



282 $S_{250-450}$ and S_R decreased significantly after addition of allochthonous OM (Fig. 2F, H). The decrease in the slope $S_{250-450}$ was
283 related to the increase of a higher molecular weight carbon, which lowered the values of $S_{275-295}$ and consequently of S_R
284 (Helms et al., 2008). In this context, our results showed that small changes in the amount and quality of allochthonous
285 contribution to lakes, either due to changes in rainfall or land use change, may cause considerable changes in the optical
286 quality of the aquatic systems, as alterations in the water transparency by changes in the UV and PAR absorbing. This
287 corroborates other studies that claim that tropical lakes are highly sensitive to climate changes, causing serious modifications
288 of lake physical and chemical conditions (Jeppesen et al. 2014), with ultimate effects on lake productivity for example
289 (O'Reilly et al. 2003).

290 Addition of nutrients, however, had little effect on these metrics, which we interpreted as a consequence of autochthonous
291 production of DOM. This is likely because these indices are derived from slope intervals in the ultraviolet range (250–400
292 nm) known to be influenced by carbon with higher molecular weight and aromatic compounds capable of absorbing energy
293 at shorter wavelengths (Bertilsson and Tranvik, 2000; Benner, 2002; Helms et al., 2008). Moreover, nutrient additions only
294 increased $a_{CDOM254}$ and $a_{CDOM440}$ in the shaded treatments (Fig. S1 and S2 in the Supplementary Material), suggesting an
295 increase of PAR and UV absorbance due to less photodegradation and less photoinhibition for the phytoplankton in the
296 shaded treatments.

297 Manipulations of nutrients, allochthonous OM and light availability caused distinctive changes in the spectral curves of
298 CDOM over the sampling days (Fig. 3). Several studies have shown that aromatic organic carbon, typically of terrestrial
299 origin, has relatively higher absorption in the ultraviolet range (Bertilsson and Tranvik, 2000; Benner, 2002; Helms et al.,
300 2008). This can explain the initial (day 0) effects of allochthonous OM addition on elevated CDOM absorption primarily
301 below 350 nm (Fig. 3A). We interpret the following increase in the CDOM absorption (days 3 and 6, especially above 350
302 nm) for most treatments to result from autochthonous DOM related to phytoplankton growth, as increases in absorption in
303 the PAR range (Fig. 3B–C) is known to be related to increases in carbon of simple structures from algal origin (Amon and
304 Benner, 1994; McKnight et al., 1994; Benner, 2002; Helms et al., 2008). After day 9 (Fig. 3D–E) the absorption loss was
305 larger than the gain by the autochthonous production in all treatments, and such spectral changes with larger absorption
306 decreases in higher wavelengths are likely due to biological degradation of CDOM (Asmala et al., 2014). However, it is
307 important to note that the relative changes in the spectral curves shown in Fig. 3 reflect the net change from two
308 counteracting processes: autochthonous production and loss of absorption by photodegradation and/or biodegradation.
309 Treatments that were exposed to full light (orange solid lines) and the shaded treatments (orange dashed lines) presented
310 notable differences between each other in the relative changes in the CDOM absorption spectrum. This corroborates to our



311 results of the more aromatic carbon with larger molecular size (higher $SUVA_{254}$ and lower $S_{275-295}$, Table 1) observed in
312 shaded mesocosms, suggesting lower photodegradation in these units.

313 *4.2 Effect of allochthonous and autochthonous DOM on CDOM spectra*

314 The results obtained from the PCA and redundancy analysis demonstrated that the increase of allochthonous OM increased
315 absorption between the wavelengths 300 and 400 nm (PC1, Fig. 5B). Several studies have shown that photodegradation is
316 more pronounced at shorter wavelengths (300-400 nm) due to absorption of aromatic carbon compounds (Helms et al., 2008;
317 Helms et al., 2013) typically related to degradation of either terrestrial vegetation (Bertilsson and Tranvik, 2000; Benner,
318 2002; Helms et al., 2008) or aquatic macrophytes (Catalán et al., 2013). We noticed a decrease in the CDOM absorption
319 below 300 nm, suggesting a greater degradation by photodegradation in these compounds from allochthonous origin
320 affecting the absorption at shorter wavelengths and increasing the absorption between 300 and 400 nm. In contrast, the
321 increase of autochthonous OM from the phytoplankton growth is likely to have resulted in an increase of absorption in the
322 UV range and a loss of absorption at wavelengths beyond 350 nm (PC2; Fig. 5C). The loss of absorption above 350 nm
323 indicates degradation by microorganisms which have greater impact on the PAR absorption. Substances that absorb in this
324 range are typically non-aromatic compounds originating from algal sources with high lability for bacterial degradation
325 (Baines and Pace, 1990; Berggren et al., 2009).

326 **5 Conclusions**

327 Additions of terrestrial OM and inorganic nutrients to a tropical lake mesocosm caused fast changes in the production and
328 transformation of OM pools as well as distinct changes in the absorption spectra of CDOM. Increased production of
329 autochthonous OM caused an increase of CDOM absorption in the UV range. However, we found that CDOM absorption
330 was reduced in the PAR range, indicating bacterial degradation of highly labile algal material (Baines and Pace, 1990;
331 Berggren et al., 2009). In contrast, the additions of allochthonous OM caused increased absorption of CDOM, especially
332 between 300 and 400 nm. $S_{250-450}$ was an effective tool to evaluate the spectral changes in general from short to long
333 wavelengths, not restricting spectrum ranges where we can only perceive changes in some regions of UV-absorbing ($S_{275-295}$,
334 $S_{350-400}$). Although the non-shaded treatments showed apparent effects of photodegradation, changes in CDOM absorption
335 curves after day 9 suggest that biodegradation was overall responsible for the largest percentage of OM degradation in these
336 experiments. The 50% reduction of sunlight in some mesocosms accordingly had minor effects on overall changes in OM
337 concentration, but the effect of shading was significant to OM quality and was important for the effect of nutrients to the
338 absorbance at 254 and 440 nm.



339 5.1 Implications

340 The factorial design of the mesocosm experiment simulated primary production and water quality seasonal changes
341 commonly observed due to shift in temperature, stratification, and rains which affect the input of terrestrial OM and nutrients
342 (Brighenti et al., 2015; Brandão et al., 2016). Our results showed that small inputs of allochthonous OM have much larger
343 effects on the spectral characteristics on the lake CDOM, compared to large production of autochthonous OM. This means
344 that changes in the land use around natural lakes (with the replacement of native vegetation) alter its spectral quality,
345 affecting the chemical (concentration and carbon quality), physical (for example it alters the upper mixing layer thickness
346 due to changes in the absorption of PAR and UV radiation) and biological conditions (distribution of organisms according to
347 the availability of light, bacterial growth, etc). In addition, with the current scenario of climate change, it has already been
348 observed a drastic decline in the annual volume of rainfall in this lake system of the Middle Rio Doce since 2012. This
349 reduction in rainfall has been affecting the allochthonous contribution to the lakes, with consequences already demonstrated
350 in recent studies for aquatic metabolism, DOM dynamics and water transparency (Gagliardi, 2015; Brighenti et al., 2015;
351 Brandão, 2016).

352

353 Author contribution

354 Luciana Brandão designed the experiment, participated in the field work, laboratory analysis and writing the manuscript.

355 Ludmila Brighenti designed the experiment, participated in the field work and laboratory analysis, and reviewed the
356 manuscript.

357 Peter Anton Staehr designed the experiment, participated in the field work, writing and revision of the manuscript.

358 Eero Asmala participated in the writing and revision of the manuscript.

359 Philippe Massicotte participated in the writing and revision of the manuscript and the statistical analyzes.

360 Denise Tonetta participated in the field work and revision of the manuscript.

361 Francisco Barbosa designed the experiment, participated in the field work, and revision of the manuscript.

362 Diego Pujoni participated in the statistical analyzes, figures and revision of the manuscript.

363 José Fernandes Bezerra-Neto designed the experiment, participated in the field work, writing and revision of the manuscript.



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548 **Figure Captions**

549 **Fig. 1** Schematic figure of the factorial design of mesocosms experiments

550 **Fig. 2** Temporal variation in the mesocosms units with and without nutrients (left panel) and organic matter additions (right
551 panel) for Chl-*a* (A,B), DOC (C,D), $S_{250-450}$ (E,F), S_R (G,H) (center line-median, 5th/95th percentiles, 95% confidence level,
552 black circles-outlier)

553 **Fig. 3** Spectral absorption curves of CDOM in the different mesocosms units for the initial day (A) and the relative changes
554 for each sampling day compared with the initial day (3rd day – B, 6th day – C, 9th day – D, 12th day – E)

555 **Fig. 4** Temporal variation in the mesocosms units with and without nutrients additions for dissolved inorganic phosphorus-
556 DIP (A), dissolved inorganic nitrogen-DIN (B) and the molar ratio DIN/DIP (C) (center line-median, 5th/95th percentiles,
557 95% confidence level, black circles-outlier)



558 **Fig. 5** Results of redundancy analysis-RDA (A) and the first two principal components obtained from PCA analysis, PC1 (B)
559 and PC2 (C), plotted against wavelengths. Gray symbols represents the treatments without organic matter addition and the
560 black ones those with organic matter. Triangles represent those without added nutrients and balls those with nutrients
561 addition.

562

563 Table 1 - Results of the Three Way Analysis of Variance. The coefficient of determination of the analysis of variance
564 partition is represented in the last column by %R².

565

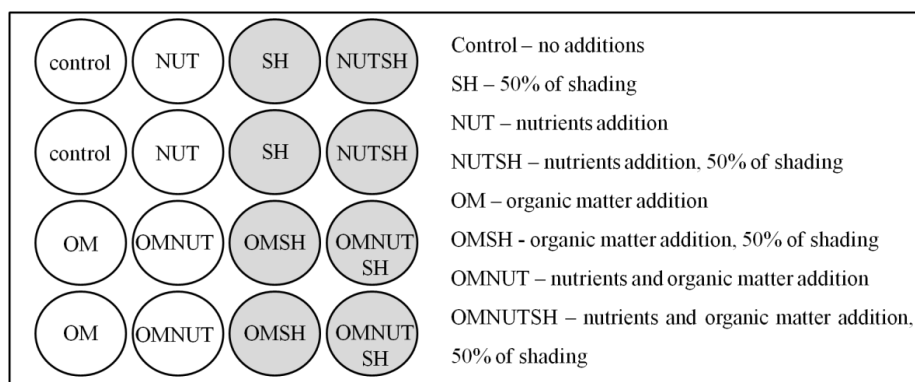


Parameters	Source of variation	DF	SS	MS	F	P	%R2
a ₂₅₄ (m ⁻¹)	OM	1	419.88	419.88	1427.44	<0.001	93.3%
	NUT	1	6.51	6.51	22.13	<0.001	1.4%
	SH	1	0.01	0.01	0.04	0.841	0.0%
	OM x NUT	1	0.52	0.52	1.76	0.189	0.1%
	OM x SH	1	0.47	0.47	1.59	0.211	0.1%
	NUT x SH	1	1.31	1.31	4.47	0.038	0.3%
	Residuals	73	21.47	0.29			0.0%
	Total	79	450.17				95.2%
a ₄₄₀ (m ⁻¹)	OM	1	1.33	1.33	230.38	<0.001	73.5%
	NUT	1	0.01	0.01	1.69	0.198	0.5%
	SH	1	0.02	0.02	2.99	0.088	1.0%
	OM x NUT	1	0.01	0.01	0.97	0.329	0.3%
	OM x SH	1	7.37E-07	7.37E-07	1.28E-04	0.991	0.0%
	NUT x SH	1	0.03	0.03	4.37	0.040	1.4%
	Residuals	73	0.42	0.01			0.0%
	Total	79	1.81				76.7%
DOC (mg L ⁻¹)	OM	1	7.07	7.07	61.63	<0.001	42.8%
	NUT	1	0.38	0.38	3.30	0.073	2.3%
	SH	1	0.70	0.70	6.07	0.016	4.2%
	OM x NUT	1	1.32E-03	1.32E-03	0.01	0.915	0.0%
	OM x SH	1	3.74E-04	3.74E-04	3.26E-03	0.955	0.0%
	NUT x SH	1	1.97E-03	1.97E-03	0.02	0.896	0.0%
	Residuals	73	8.38	0.11			0.0%
	Total	79	16.53				49.3%
SUVA ₂₅₄ (m ² mg ⁻¹ C)	OM	1	2.85	2.85	296.38	<0.001	78.0%
	NUT	1	0.02	0.02	2.37	0.128	0.7%
	SH	1	0.05	0.05	5.60	0.021	1.4%
	OM x NUT	1	0.01	0.01	0.98	0.325	0.3%
	OM x SH	1	0.01	0.01	0.67	0.417	0.2%
	NUT x SH	1	0.02	0.02	1.90	0.172	0.5%
	Residuals	72	0.69	0.01			0.0%
	Total	78	3.65				81.0%
S ₂₇₅₋₂₉₅ (nm ⁻¹)	OM	1	1.89E-04	1.89E-04	287.91	<0.001	76.4%
	NUT	1	2.00E-06	2.00E-06	3.05	0.085	0.8%
	SH	1	7.93E-06	7.93E-06	12.08	0.001	3.2%
	OM x NUT	1	2.95E-09	2.95E-09	4.50E-03	0.947	0.0%
	OM x SH	1	3.58E-08	3.58E-08	0.05	0.816	0.0%
	NUT x SH	1	4.11E-07	4.11E-07	0.63	0.431	0.2%
	Residuals	73	4.79E-05	6.56E-07			0.0%
	Total	79	2.47E-04				80.6%
Chl-a (µg L ⁻¹)	OM	1	92.60	92.60	4.46	0.038	3.3%
	NUT	1	1172.94	1172.94	56.53	<0.001	41.4%
	SH	1	6.61	6.61	0.32	0.574	0.2%
	OM x NUT	1	32.24	32.24	1.55	0.217	1.1%
	OM x SH	1	12.44	12.44	0.60	0.441	0.4%
	NUT x SH	1	0.29	0.29	0.01	0.906	0.0%
	Residuals	73	1514.56	20.75			0.0%
	Total	79	2831.67				46.5%

566 OM = organic matter NUT = nutrients SH = shade

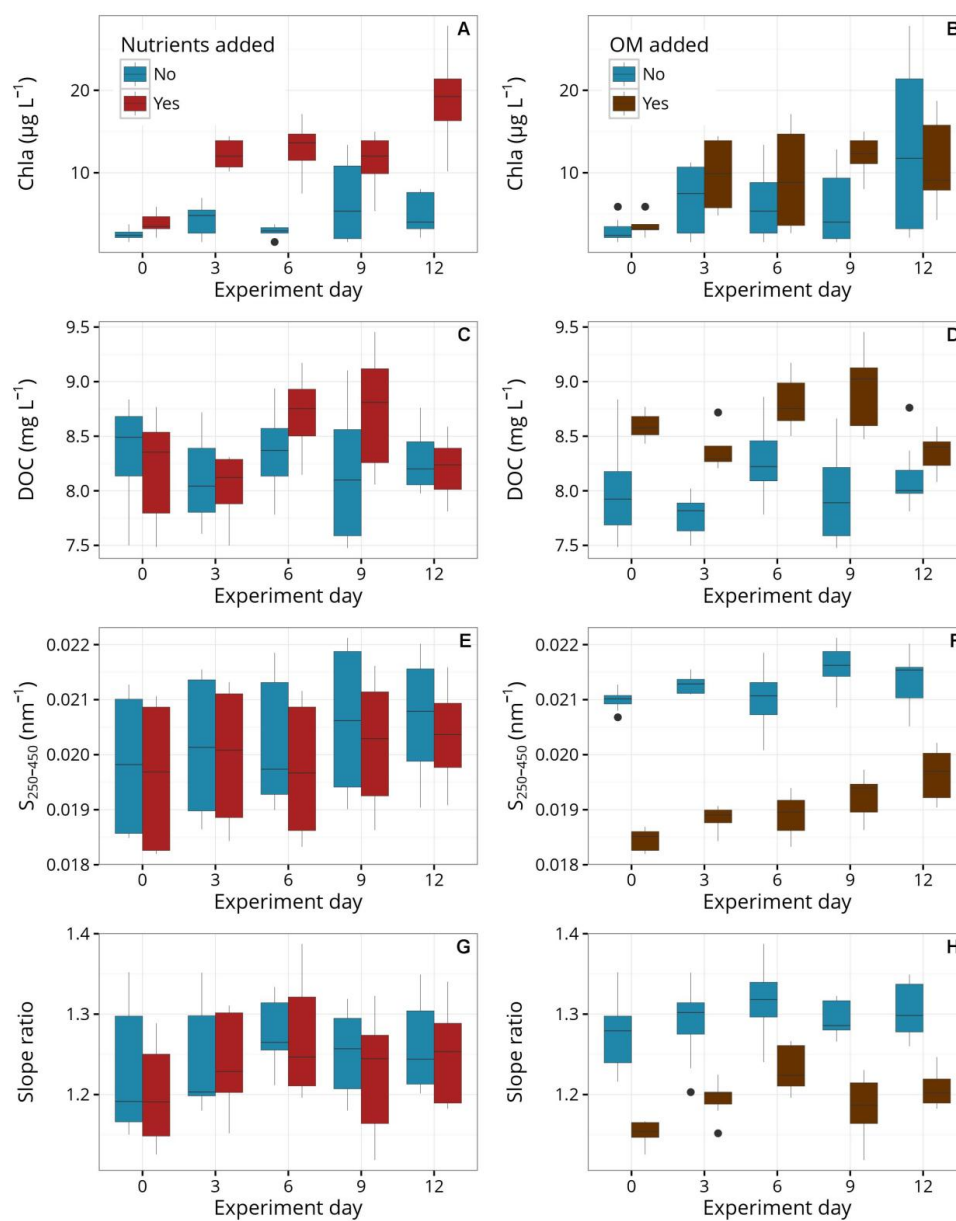


Fig 1





581 Fig 2



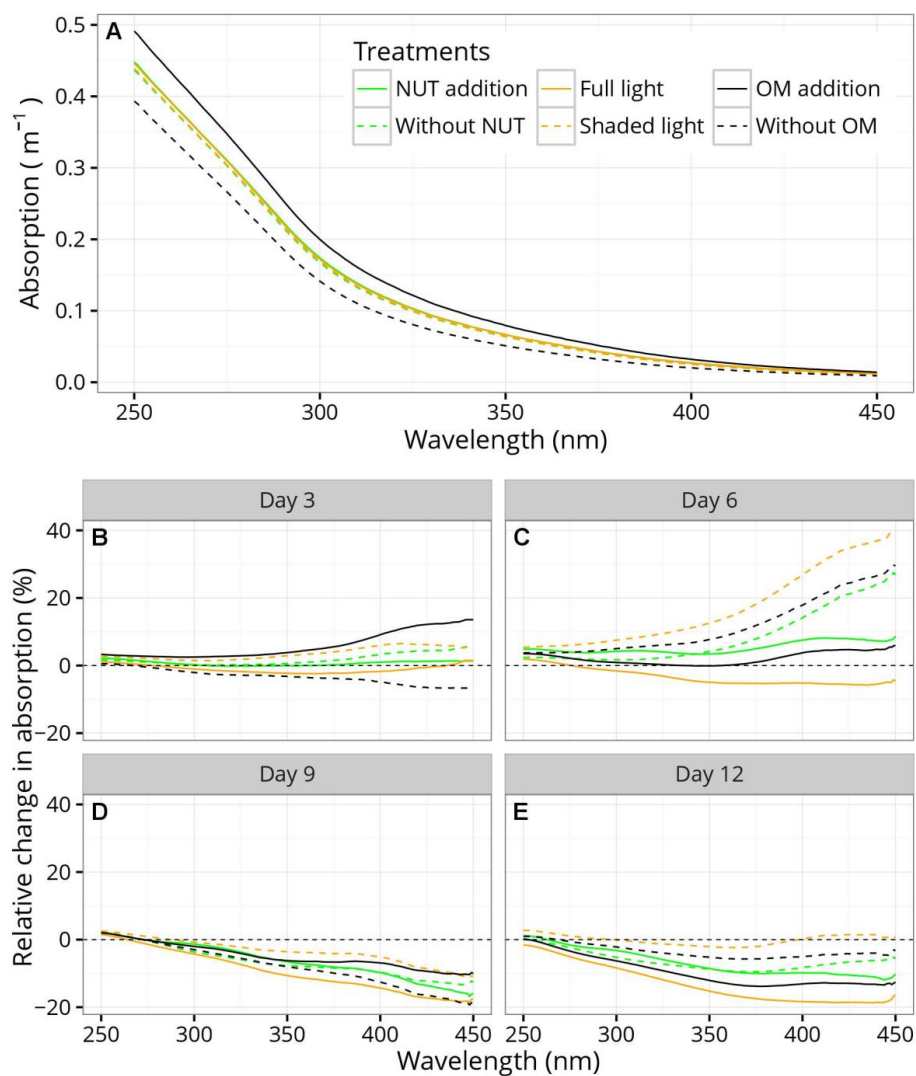
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585 Fig 3



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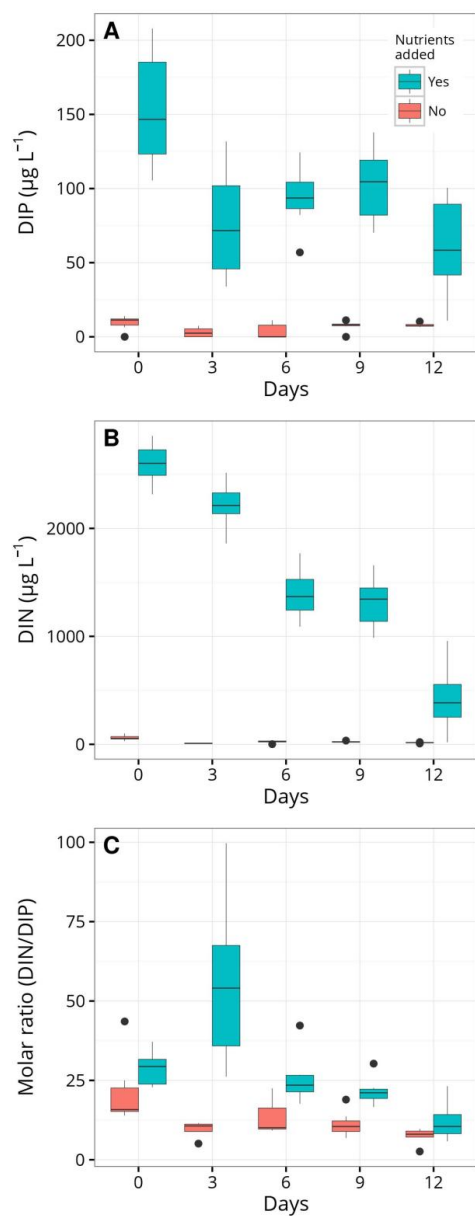
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591 Fig 4



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Fig 5

