



- 1 A new procedure for processing eddy-covariance data to better
- 2 quantify atmosphere-aquatic ecosystem CO<sub>2</sub> exchanges
- 3
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- 11 Abstract. The capture of carbon by aquatic ecosystems and its sequestration in
- 12 sediments has been studied as a potential method for mitigating the adverse effects of
- 13 climate change. However, the evaluation of in situ atmospheric CO<sub>2</sub> fluxes is
- 14 challenging because of the difficulty in making continuous measurements over areas
- 15 and for periods of time that are environmentally relevant. The eddy covariance (EC)
- 16 method is the most promising approach to address this concern with the measurement of
- 17 atmospheric CO<sub>2</sub> fluxes. However, methods to process the data obtained from EC
- 18 measurements are still being developed, and the estimated air-water CO<sub>2</sub> fluxes have
- 19 large uncertainties and differ from those obtained using conventional methods. In this
- 20 study, we improved the post-processing procedure for the EC method to reduce the
- 21 uncertainty in the measured air-water CO<sub>2</sub> fluxes. Our new procedure efficiently
- 22 removes erroneous fluxes using a combination of filtering methods based on the
- 23 received signal strength indicator of the EC sensor, the normalized standard deviation of





- atmospheric CO<sub>2</sub> and water vapor concentrations, and a high-pass filter. Our procedure
- 25 is easier to apply to EC measurements than existing correction methods. The improved
- 26 EC fluxes did not always agree with those obtained by using conventional methods
- 27 (e.g., the bulk formula method), but this difference was attributable to the difference of
- 28 measurement heights and the effect on the measured fluxes of the physical and
- 29 biological properties of the water surface (e.g., the presence of vegetation on the water
- 30 surface and the temperature gradient in the overlying atmospheric layer). Because the
- 31 measurement height and the spatiotemporal scales of the flux measurement depend on
- 32 the applied method, it is essential to select the appropriate method for studies related to
- 33 CO<sub>2</sub> fluxes and to the determination of ecosystem-atmospheric CO<sub>2</sub> interactions and the
- role of aquatic ecosystems in mitigating the adverse effects of climate change.

### 35 1 Introduction

36	Aquatic environments are considered critical to the mitigation of adverse climate
37	change effects because of their ability to store atmospheric CO2. Previous studies have
38	estimated that the ocean absorbs approximately one-fourth of the CO <sub>2</sub> emitted by
39	anthropogenic activities (IPCC, 2013). However, the effect of shallow aquatic
40	ecosystems on atmospheric CO2 remains a controversial topic. Several previous studies
41	have concluded that shallow aquatic ecosystems are sources of atmospheric CO2 after
42	taking account of carbon inputs from land (e.g. Gazeau et al., 2005; Borges et al., 2006;
43	Chen et al., 2013). In contrast, some autotrophic, shallow aquatic ecosystems have been
44	reported to be net sinks for atmospheric CO <sub>2</sub> (e.g. Schindler et al., 1997; Tokoro et al.,
45	2014).

46 In situ measurements of atmospheric CO<sub>2</sub> fluxes are necessary for precise analysis





- 47 of carbon cycling in aquatic environments. CO<sub>2</sub> fluxes in aquatic environments are
- 48 difficult to determine because of the variability of several factors, including
- 49 concentrations of CO<sub>2</sub> in the water and air and the physical characteristics of the
- 50 atmosphere and water surface. Several methods have been proposed for measuring in
- 51 situ CO<sub>2</sub> fluxes. Because each of these methods works best at a different combination of
- 52 spatial and temporal scales and is associated with different costs and technical
- 53 difficulties, a variety of methods have been applied to different aquatic environments
- 54 (e.g. oceans, estuaries, and lakes) to assess rates of aquatic carbon cycling.
- 55 Methods of estimating air-water CO<sub>2</sub> fluxes can be assigned to one of two
- 56 categories: (1) indirect estimations based on CO<sub>2</sub> concentration gradients just below the
- 57 water surface (Lewis and Whitman, 1924) or from the renewal rate of a very small body
- 58 of water (Danckwerts, 1951), and (2) direct estimations. With either of the indirect
- 59 methods, the  $CO_2$  flux is calculated from the product of the difference in the  $CO_2$
- 60 fugacity (fCO<sub>2</sub>) between air and water, the CO<sub>2</sub> solubility, and a physically regulated
- 61 parameter called the transfer velocity. Because the transfer velocity cannot be estimated
- 62 directly, empirical and hydrodynamic models for estimating transfer velocity have been
- 63 proposed (Garbe et al., 2013).
- 64 At the present time, the empirical model is primarily used for evaluating aquatic
- 65 CO<sub>2</sub> fluxes because of the difficulty in applying the hydrodynamic model. In the
- 66 empirical model, the regulating factor for transfer velocity has been identified from
- 67 several direct CO<sub>2</sub> measurements by using tracers such as <sup>14</sup>C and SF<sub>6</sub> (e.g. Broecker
- and Peng 1982; Ho et al., 2014) or water-tank experiments (e.g. Komori et al., 1993).
- 69 Based on these results, several empirical equations have been formulated. The wind
- speed above the water surface is a metric of one regulating factor (e.g. Liss and





- 71 Merlivat, 1986; Wanninkhof, 1992; Ho et al., 2006). In the case of shallow systems,
- vater velocity fields and depths have been used to estimate the gas transfer velocity
- 73 (O'Conner and Dobbins, 1958; Borges et al., 2004). In this study, the method using such
- ra empirical model for estimating aquatic CO<sub>2</sub> flux is defined as the Bulk Formula method
- 75 (BF method).
- 76 However, application of the BF method is limited because of its poor temporal and

57 spatial coverage. Moreover, in previous studies, air-water CO<sub>2</sub> fluxes have been

determined mostly as snapshots that did not account for diurnal changes or annual

79 cycles, the result being considerable uncertainty and bias (Kuwae et al., 2016). In

80 brackish environments in particular, temporal variability of water fCO<sub>2</sub> is significant,

81 whereas the carbonate buffer effect is weak, and the fluctuations of fCO<sub>2</sub> become very

82 large (Zeebe and Wolf-Gladrow, 2001). Use of BF methods to carry out a

83 comprehensive analysis of dynamic carbon cycling in aquatic environments with large

spatial and temporal variability would therefore be very costly and require much effort.

- 85 Another method for evaluating the air-water CO<sub>2</sub> fluxes is direct measurement of in
- situ fluxes, which involves use of a chamber floating on the water surface (e.g.

Frankignoulle, 1988; Tokoro et al., 2008) and eddy covariance devices (v)le infra). The

88 floating chamber method is used to determine the air-water CO<sub>2</sub> flux from continuous

- 89 measurements of CO<sub>2</sub> concentrations in the air inside a hollow, box-shaped device
- 90 floating on the water surface. Although this method is the easiest of the direct methods
- 91 to use in shallow coastal waters because of its relative simplicity, like the BF method, it
- 92 is poorly suited for obtaining long-term measurements over wide areas.
- 93 The eddy covariance (EC) method, which is commonly used to determine mass and
- heat fluxes in terrestrial environments, has recently been used to estimate air-water





fluxes of greenhouse gases [] Lee et al., 2004). The determination of the EC CO<sub>2</sub> flux 95is based on the micro-meteorological behavior of atmospheric eddy diffusion and is 9697 calculated from the covariance of atmospheric CO2 concentrations and vertical wind 98speeds measured at high frequency (m) than 10 Hz). Because EC measurements can 99 be performed automatically and represent the flux over a large area, the EC method can be used to obtain a detailed analysis of CO2 fluxes. 100 Despite the promise of EC measurements, application of the EC method in aquatic 101 102environments remains challenging (Tsukamoto et al., 2004; Rutgersson and Smedman, 2010; Vesala, 2012; Blomquist et al., 2013; Kondo et al., 2014; Ikawa and Oechel, 103 2014; Landwehr et al., 2014). The difficulty of making aquatic EC measurements is that 10) 105 the air-water CO<sub>2</sub> flux is small compared with the air-land CO<sub>2</sub> flux (Vesala, 2012; 106 Landwhehr et al., 2014). The main technical problem with the EC method is cross sensitivity, which reflects the interference between the atmospheric CO<sub>2</sub> and H<sub>2</sub>O 107 108 measurements caused by spectrometric error (Kohsiek et al., 2000; Prytherch et al., 109 2010; Kondo et al., 2014; Landwehr et al., 2014). A procedure based on the relationship 110 between atmospheric CO<sub>2</sub> concentration and relative humidity called the PKT 111 correction has been proposed to correct for the effects of cross sensitivity (Prytherch et 112al., 2010). However, the PKT correction is not always effective. Past studies have 113 revealed that cross sensitivity is reduced only after application of certain operational 114 procedures such as cleaning the optical lens of the sensor (Ikawa and Oechel, 2014; 115Kondo et al., 2014) and drying the sample gas (Landwhehr et al., 2014). 116 There are several problems in addition to cross sensitivity in using EC 117 measurements in aquatic environments. The uncertainty of EC measurements has been 118 attributed to the spatial and temporal heterogeneity of water (Mørk et al., 2014). The EC





119	flux is calculated as the average within a measurement area called the "footprint", which
120	can range from several hundred meters to several kilometers windward from the
121	measurement point (Schuepp et al., 1990). Therefore, EC fluxes at heterogeneous water
122	sites are different from the fluxes determined by methods that estimate the CO <sub>2</sub> flux in
123	an area of only several square meters (e.g., the BF method and floating chamber
124	method). The EC flux is an average flux over a certain time interval (approximately
125	several tens of minutes) (Lee et al., 2004), whereas the BF method estimates the flux at
$\mathcal{D}$	the time of sampling. Thus, EC fluxes estimated at sites where fluxes are temporally
127	variable also differ from fluxes obtained using other methods. Furthermore, the inflow
128	of terrestrial air into the measurement site can generate uncertainty in the flux
$\bigcirc$	measurement because the atmospheric CO <sub>2</sub> concentration over terrestrial vegetation
130	may differ significantly from the concentration over water. The inflow of terrestrial air
131	can cause unnatural temporal changes in the atmospheric CO2 concentration and spatial
132	heterogeneity at the measurement site. It is therefore necessary to account for the
133	characteristics of the aquatic environment and apply appropriate post-processing (PP)
134	procedures (Leinweber et al., 2009) to avoid large uncertainties or biases in EC flux
135	calculations.
136	In this study, we developed a PP procedure for EC aquatic measurements. This PP
137	procedure involves the exclusion of erroneous data and correction of unnatural changes
138	in the atmospheric CO <sub>2</sub> using a series of data-filtering steps. The new process is based
139	on the idea that cross sensitivity and environmental heterogeneity during flux
140	measurement cause spikes, drifts, offsets, and long-term variation in the CO2 and H2O
141	raw data. We compared the results calculated with our new PP procedure to those
142	obtained using conventional EC PP procedures along with BF flux data as an example





- 143 of a frequently used method. We then discuss the differences between the conventional
- 144 BF and EC fluxes with respect to atmospheric and environmental characteristics.
- 145 2 Methods

#### 146 **2.1 Field measurements**

147	Continuous EC measurement data were used for the evaluation of the PP procedure and
148	the analysis of atmospheric-aquatic ecosystem CO2 exchange. The data were collected
149	from a brackish lagoon in Japan (the Furen Lagoon, Fig. 1) from 28 May to 21 October
150	2014, during which time the water surface was not frozen. Most of the study area (57.4
151	km <sup>2</sup> ) is covered by seagrass meadows (mainly Zostera marina). The water is shallow
152	(1-2  m), except in a channel that connects the eastern and western basins of the lagoon
153	(depth = approximately 5 m). Freshwater flows into the western basin through several
154	rivers that run through the surrounding grass farms, and seawater is exchanged through
155	the lagoon mouth, which opens to the Okhotsk Sea. A previous study has found that the
156	air-water CO <sub>2</sub> flux in the lagoon is affected by changes of salinity caused by the inflow
157	of river water and tides as well as by changes of dissolved inorganic carbon resulting
158	from biological processes such as photosynthesis (Tokoro et al., 2014). The
159	measurement platform was built at the same site used in the study of Tokoro et al.
160	(2014) (N43° 19.775', E145° 15.463'); the effects of photosynthesis and changes in
161	salinity are most notable at this location in the lagoon (Tokoro et al., 2014).
162	The EC devices used in this study were as follows. Atmospheric CO <sub>2</sub> concentrations
163	and water vapor were measured with an open-path sensor (LI-7500A, LI-COR, USA).
164	The three-dimensional (3D) wind velocity, air temperature, and atmospheric pressure
165	were measured with a 3D ustic Doppler velocimeter (CSAT-3, Campbell Scientific,





- 166 USA). The data were logged and managed by a SMARTFlux system (LI-COR). The
- 167 open-path sensor and the wind velocimeter were attached to the platform approximately
- 168 3.0–5.5 m above the water surface (the height varied with the tide). The sampling rate
- 169 for all data was 10 Hz, and the fluxes (CO<sub>2</sub>, water vapor, and heat) were calculated as
- averages over 30-min intervals. Batteries and solar panels were attached to the platform
- as a power source. Battery replacement, data collection, and device maintenance were
- 172 performed approximately every two weeks. Water temperature and salinity were
- 173 measured continuously with a conductivity-temperature sensor (Compact-CT, Alec,
- 174 Japan).

#### 175 **2.2** Calculation of fluxes using the conventional PP procedure (PP1)

The air-water CO<sub>2</sub> flux (*F*) was calculated every 30-min using the following equation:

178 
$$F = \overline{\rho_c' w'} \cdot F_1 + \mu \frac{\rho_c}{\rho_d} \overline{\rho_v' w'} \cdot F_1 + \rho_c \left(1 + \mu \frac{\rho_v}{\rho_d}\right) \frac{\overline{T_a' w'}}{T_a} \cdot F_2 , \qquad (1)$$

179

180 where the coefficients  $F_{I}$  and  $F_{2}$  are correction terms based on the transfer functions that 181 correct for the frequency attenuation of the air-sea CO<sub>2</sub> flux caused by the response time of the sensor, path-length averaging, sensor separation, signal processing, and flux-182183averaging time (Massman, 2000). The first term on the right-hand side of Eq. (1) is the 184 product of  $F_1$  and the uncorrected air-sea CO<sub>2</sub> flux calculated as the covariance of the  $CO_2$  density  $\rho_c$  and the vertical wind speed w (the bar and the prime indicate the mean 185186 and the deviation from the mean, respectively). The second and third terms are the 187 Webb-Pearman-Leuning correction of latent heat and sensible heat, respectively (Webb





188	et al., 1980). The other variables in Eq. (1) are defined as follows: $\rho_d = dry$ air density;
189	$\rho_v$ = water vapor density; $T_a$ = air temperature; and $\mu$ = ratio of the molar weight of dry
190	air to that of water vapor. The wind speed was corrected by using a double rotation to
191	make the average vertical wind speed zero during the 30-min time interval (Lee et al.,
192	2004). The footprint (measurement area) depends on several factors, including the
193	measurement height, wind speed, atmospheric stability, measurement site roughness
194	$(10^{-4} \text{ cm})$ (Schuepp et al., 1990). This footprint ranged from several hundred square
195	meters to several square kilometers on the windward side of the measurement site.
196	The deviation of each parameter in Eq. (1) from the 30-min average was calculated by
197	subtracting the 30-min average from the instantaneous data after deleting obviously
198	erroneous data (e.g., negative values of CO <sub>2</sub> or water vapor concentration). Other
199	corrections to the raw data included coordinate rotation of the 3D wind component
200	(double rotation; Lee et al., 2004), time lag of the measurement due to the separation of
201	the CO <sub>2</sub> sensor and the wind velocimeter (covariance maximization; Lee et al., 2000),
202	exclusion of wind data contaminated by the wind velocimeter flame, and correction of
203	the measurement noise (statistical tests of Vickers and Mahrt, 1997) based on the default
204	settings of the data management software (EddyPro 5.1.1, LI-COR).
205	2.3 Calculation of flux using our new PP procedure (PP2)

- 206 After calculating the EC flux using the conventional PP procedure (PP1) described in
- 207 Sect. 2.2, we recalculated the EC flux using our new procedure (PP2; Fig. 2). The PP2
- 208 procedure is mainly based on filtering and excluding erroneous data rather than on the
- 209 PKT method of data correction. The PP2 procedure is also focused on aquatic
- 210 environments in which the spatial and temporal variations in atmospheric CO<sub>2</sub> are large.





- 211 The PP2 procedure combines a series of filtering methods based on the received signal
- 212 strength indicator (RSSI) of the EC sensor, the normalized standard deviations (nSDs)
- 213 of the atmospheric CO<sub>2</sub> and water vapor concentrations, and high-pass (HP) filtering.
- 214 The RSSI is one of the data concurrently obtained from the CO<sub>2</sub> sensor of the EC
- 215 measurement instrumentation and indicates the available signal strength of the sensor.
- 216 This parameter has been used for evaluating measurement validation of objective gases
- such as methane. In this study, we used the RSSI to filter the  $CO_2$  data because both
- CO<sub>2</sub> and methane absorb infrared radiation. First, data in the 30-min time series were
- excluded if their RSSI was low. The RSSI threshold for exclusion was set to 90 %,
- 220 because this percentage was sufficiently high to identify valid CO<sub>2</sub> concentrations and
- 221 wed most of the measurement data (~81 %) to be retained. This 90 % threshold was
- also recommended as the quality criterion for the LI-7700 methane analyzer (LI-COR),
- 223 which has been used to measure methane concentrations.
- (http://www.licor.com/env/newsline/2012/04/overcoming-the-challenges-of-open-path methane-measurements-with-the-li-7700/).
- Second, the criteria for excluding erroneous fluxes were identified. Erroneous fluxes were identified based on unnatural spikes, jumps, and shifts in the data. Outliers were excluded based on three statistical parameters: (1) the normalized standard deviation, nSD (i.e. the SD over a 30-min period divided by the average value for the entire measurement period after RSSI filtering; (2) skewness; and (3) the absolute value of kurtosis. Each of these threshold values was determined after the top four outliers for each parameter were excluded; the raw data (CO<sub>2</sub> concentration, wind speed) associated
- 233 with the top four outliers were assumed to be erroneous changes that would not happen
- anaturally.





- Finally, HP filtering was applied to the calculation of the deviations from the mean of each parameters in Eq. (1). This procedure corrected relatively long-term (several minutes to 30 minutes) variations in CO<sub>2</sub> or water vapor concentrations that were independent of eddy fluctuations and were caused by the temporal and spatial heterogeneity of the atmospheric mass. HP filtering is often applied to measurements in a complex environment; however, incorrect application of HP filtering results in underestimation of fluxes (Lee et al., 2004). HP filtering was applied by using an
- 242 exponential moving average as follows:

244 
$$\begin{array}{c} x_{i}' = A(x_{i} - x_{i-1}) + Ax_{i-1}' \\ A = \exp\left(-\frac{1}{\tau}, f\right) \end{array}, \qquad (2)$$

245

where  $x_i$  and  $x_i$ ' are an instantaneous datum and deviation from the mean at time *i*, respectively. The parameter the time constant of the exponential moving average, which was determined to be 150 s in a previous study (McMillan, 1988), and *f* is the sampling frequency (10 Hz). HP filtering was applied to all of the measured data (i.e., 3D wind velocity, air temperature, CO<sub>2</sub> and water vapor concentrations, and

atmospheric pressure).

### 252 **2.4 Measurement of fluxes using the BF method**

253 BF flux measurements were performed during the daytime on 29 May, 15 July, and 21

- 254 September 2014 for comparison with the EC measurements. In the BF method, flux is
- 255 calculated as follows:
- 256





$$257 \qquad F = kS(fCO_{2water} - fCO_{2air}),$$

(3)

258

where k is the transfer velocity, which was calculated using one or another of the 259260following three empirical equations. The first was the equation of Wanninkhof (1992) (W92). Because this equation was constructed using tracer methodology under oceanic 261262conditions, it might be inappropriate to our measurements because of the differences in 263fetch and water depth. However, this equation is very commonly used in a variety of 264oceanographic CO2 flux studies, including coastal measurements. We therefore used it 265for comparison. The second equation was that of Borges et al. (2004) (B04), which has 266 been applied in an estuarine study that involved use of the floating chamber method. 267 This equation makes use of the current velocity in addition to the wind speed. The third 268equation was that of Mørk et al., (2014) (M14), which was formulated to characterize 269the transfer velocity in a fjord for use with the EC method. Thus, the second and third 270 equations can be used to characterize coastal gas transfer velocity. The wind speed for 271the gas transfer velocity was measured by the EC device and normalized to a height of 27210 m from the water surface using a logarithm law (Kondo, 2000). The current speed 273and depth were measured by the sensors attached to the platform (Compact-TC and 274Compact-EM, Alec, Japan). The final CO<sub>2</sub> flux was calculated as the weighted average 275of the fluxes measured within the EC footprint (Schuepp et al., 1990). 276The parameter S is the dissolution coefficient of CO<sub>2</sub>, which was estimated from the 277water temperature and salinity (Weiss, 1974). The parameters fCO<sub>2water</sub> and fCO<sub>2air</sub> are 278the fugacity of water and atmospheric CO<sub>2</sub>, respectively. Water temperature and salinity were measured with a handheld conductivity-temperature-depth sensor (ACTD-DF, JFE 279Advantech, Japan). The water samples used to determine fCO<sub>2water</sub> were collected just 280





- 281 below the water surface (up to 20 cm below the water surface) to measure the
- 282 concentration of CO<sub>2</sub> where direct gas exchange with air occurs. The sampling was
- 283 performed within the EC footprint to minimize the effect of spatial heterogeneity when
- 284 comparing the BF and EC fluxes. The sampling points were determined from the wind
- 285 direction and the distance from the platform measured using a hand-held GPS unit
- 286 (Venture HC, Garmin, USA; see Table S1). The water fCO<sub>2</sub> was determined from the
- total alkalinity and the dissolved inorganic carbon content of the water sample using a
- 288 batch-type carbonate measurement system (ATT-05, Kimoto electrics, Japan) and the
- 289 CO2SYS program (Pierrot et al., 2006).
- 290
- 291 3 Results
- 292 3.1 PP1 data
- 293 During the measurement period, 4464 flux data points corresponding to 2232 hours
- were obtained; 1971 of those data points (44 %) were excluded as erroneous data after
- 295 PP1 application. The mean and SD of the EC CO<sub>2</sub> fluxes were -1.93 and 52.4  $\mu$ mol m<sup>-2</sup>
- $s^{-1}$ , respectively. Figure 3(a) shows the retained CO<sub>2</sub> flux data.
- The largest positive CO<sub>2</sub> flux (release to atmosphere) was 156.51  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup> at
- 298 2:00 on 23 June (day 56). The largest negative CO<sub>2</sub> flux (uptake of atmospheric CO<sub>2</sub>)
- 299) was  $-217.93 \mu$ mol m<sup>-2</sup> s<sup>-1</sup> at 22:00 on 4 October (day 129). These fluxes were more
- 300 than three orders of magnitude larger than the magnitude of the average of the measured
- 301 EC fluxes. Figure 4 shows the instantaneous atmospheric CO<sub>2</sub> concentration, water
- 302 vapor concentration, and the cumulative covariance between CO<sub>2</sub> and vertical wind
- 303 speed during the times when the CO<sub>2</sub> fluxes were most positive or most negative.





- 304 Spikes and discontinuities were observed in the atmospheric CO<sub>2</sub> and water vapor
- 305 concentrations, despite the prior de-spiking process applied by the data management
- 306 system. The cumulative covariance indicated that the covariance at certain periods (0-5
- 307 min) contributed significantly to the total cumulative covariance.
- 308 3.2 PP2 data
- 309 The EC CO<sub>2</sub> flux data subjected to PP2 (RSSI, nSD, and HP filtering) are shown in Fig.
- 310 3(b). Of the 2493 total data points remaining after PP1, approximately 234 (9 %) were
- 311 excluded by RSSI filtering. Subsequent nSD filtering removed 426 additional data
- 312 points (17%); approximately 73% of the measurement data remained after this
- filtering. The mean and SD of EC CO<sub>2</sub> flux after PP2 were -0.54 and 2.2 µmol m<sup>-2</sup> s<sup>-1</sup>,
- 314 respectively.
- The nSD threshold was calculate using the averages of the CO<sub>2</sub> and water vapor
- 316 concentrations for the entire measurement period after RSSI filtering (CO<sub>2</sub>: 16.02 mmol
- $m^{-3}$ , vapor: 548.10 mmol m<sup>-3</sup>). The nSD threshold value obtained from the average
- 318 values was 0.050. Other thresholds were 0.48 for skewness and 3.1 for the absolute
- 319 value of kurtosis. Among these comparisons, the nSD was determined to provide the
- 320 best threshold for the measurement data because the number of data remaining after
- 321 filtering was the largest for the nSD (nSD: 1661 [74 %], skewness: 422 [19 %], and the
- 322 absolute value of kurtosis: 445 [20 %]). The nSD filtering resulted in the exclusion of
- 137 (6 %) of the CO<sub>2</sub> data points and 569 (25 %) of the water vapor data points. Figure
   324 5 shows the three parameters for the EC CO<sub>2</sub> flux data.
- **326**

HP filtering decreased the absolute values of some CO<sub>2</sub> fluxes (Fig. 6). Most of the data were not changed very much by HP filtering; however, HP filtering decreased the





- 327 absolute values of some fluxes, but 7 shows an example of the results of filtering
- 328 atmospheric CO<sub>2</sub> concentrations as well as the cumulative covariance of the
- 329 atmospheric CO<sub>2</sub> concentrations and vertical wind speed (measured at 8:00 on August
- 330 21 [day 84]). These data were not excluded by the RSSI and nSD filtering (RSSI =
- 331 100 %, nSD of CO<sub>2</sub> =  $9.84 \times 10^{-3}$ , nSD of H<sub>2</sub>O =  $2.07 \times 10^{-2}$ ). The concentration of
- atmospheric CO<sub>2</sub> showed a trend over the 30-min time interval, the indication being that
- 333 the block average could not extract appropriate eddy movements from the time-series
- data. The cumulative covariance of the block averaging (BA) was unusually large; after
   HP filtering, the values were more reasonable because filtering successfully excluded
- deviations caused by the variation for about 10-min of atmospheric CO<sub>2</sub> concentrations.

#### 337 3.3 BF data

338	The measured BF fluxes showed spatial and seasonal variations (see Table S1). The
339	means and SDs obtained by using the three gas transfer velocity equations to analyze
340	the weighted averages of the BF CO <sub>2</sub> flux data were $0.44 \pm 0.33 \mu\text{mol }\text{m}^{-2} \text{s}^{-1}$ ( <i>n</i> = 15)
341	on 29 May (day 1), $2.10 \pm 1.58 \ \mu\text{mol} \ \text{m}^{-2} \ \text{s}^{-1}$ ( <i>n</i> = 14) on 15 July (day 48), and $-0.11 \pm$
342	0.13 $\mu$ mol m <sup>-2</sup> s <sup>-1</sup> ( <i>n</i> = 9) on 21 September (day 115). The differences in the number of
343	data reflect problems during the measurements, such as inadequate water depth and
344	problems with the water-depth sensor. Except for the measurements on 15 July, the
345	difference between the results obtained with the three gas transfer velocity equations
346	were not significant.

#### 347 4 Discussion

348 Our new filtering method, PP2, successfully excluded erroneous outliers. The SD was





349	decreased by a factor of 24 (52.4 $\mu mol~m^{-2}~s^{-1}$ in PP1 to 2.2 $\mu mol~m^{-2}~s^{-1}$ in PP2). The
350	atmospheric CO <sub>2</sub> uptake rate calculated via PP1 measurements (–1.93 $\mu$ mol m <sup>-2</sup> s <sup>-1</sup> )
351	was reduced in magnitude by 72 % after PP2 to $-0.54~\mu mol~m^{-2}~s^{-1}.$ This uptake rate is
352	more consistent with the range of atmospheric CO2 uptake rates reported in previous
353	studies (e.g. Borges et al., 2005; Chen et al., 2013; Laruelle et al., 2013). The most
354	negative CO <sub>2</sub> flux ever reported was $-1.08 \ \mu mol \ m^{-2} \ s^{-1}$ during spring in the Baltic Sea
355	(Chen et al., 2013).
356	The exclusion of erroneous outliers by PP2 also contributed to the time-series
<b>§</b> 5)	analysis. The power spectra of the EC CO <sub>2</sub> fluxes after PP1 evidenced large, noise-like
358	fluctuations at high frequencies (Fig. 8), and thus any suggestion of peaks in the time
<mark>359</mark>	series was obscured. After PP2, however, the noise-like fluctuations were smaller, and
360	two peaks associated with semi-diurnal (~12.5 h) and diurnal (~24 h) time intervals
361	were apparent. The fCO <sub>2</sub> variations in the lagoon, which are among the parameters that
362	regulate air-water CO <sub>2</sub> fluxes, have been confirmed to be related to mixing of lagoon
363	water with freshwater coming from rivers and with biological processes such as
364	photosynthesis (Tokoro et al., 2014). Given that the former and latter phenomenon are
365	caused by the semi-diurnal tidal cycle and diel changes of irradiance, respectively, the
366	peaks in the power spectra are consistent with the results of Tokoro et al. (2014). This
367	consistency is a good demonstration of the utility of the PP2.
368	The EC data that were much different from the BF results were excluded by the PP2
369	(D). The remaining EC fluxes in May and September seemed to agree well with the
370	BF fluxes. We believe that one of the reasons that the results were comparable was the
371	improvement in the accuracy of the EC fluxes with the use of PP2. Another reason was
372	our strategy of making BF measurements at multiple points within the EC footprint to





- 373 filter out the noise associated with the spatial heterogeneity of the BF fluxes. In July, the
- BF fluxes estimated with the W92 seemed to be the most consistent with the EC fluxes.
- 375 This similarity indicates that the effect of currents in the B04 and the effect of early-
- breaking waves in the M14 were overestimated at our measurement site.
- 377 However, the EC fluxes estimated with PP2 did not always agree with the BF
- fluxes. Because the *fCO*<sub>2water</sub> is theoretically never negative, a theoretical maximum
- negative BF flux can be calculated by arbitrarily setting  $fCO_2$  equal to zero. The
- maximum negative flux calculated in this way (with M14) was  $-6.16 \mu$ mol m<sup>-2</sup> s<sup>-1</sup> at
- 381 15:00 on 30 May (day 2), when the maximum wind speed was recorded (11.9 m/s).
- 382 Forty-seven EC flux data points (3 % of all data) indicated even lower fluxes.
- 383 Moreover, the mean of the EC fluxes with PP2 ( $-0.54 \mu$ mol m<sup>-2</sup> s<sup>-1</sup>) was more negative
- or almost the same as the mean of the theoretical BF fluxes ( $-0.26 \mu mol m^{-2} s^{-1} in$ , -
- $385 \quad 0.55 \ \mu mol \ m^{-2} \ s^{-1}$ , and  $-0.43 \ \mu mol \ m^{-2} \ s^{-1}$  estimated with W92, B04, and M14,
- 386 respectively). Because actual BF fluxes include less negative as well as positive fluxes,
- these EC fluxes cannot be explained by only the BF fluxes.
- 388 Similar inconsistencies between air-water CO<sub>2</sub> fluxes calculated with the EC method
- and other conventional methods have been reported in several studies (e.g., Tsukamoto
- et al., 2004; Rutgersson et al., 2010). In the case of coastal measurements, water-side
- 391 convection due to the vertical temperature difference inside water has been postulated to
- enhance the gas transfer velocity (Rutgersson et al., 2010). However, such an
- 393 enhancement was not previously observed with direct flux measurement using a floating
- chamber at our site (Tokoro et al., 2014). Because of the very shallow water depth (less
- than 2 m) at our site, we suspect that water-side convection was weak and was not the
- 396 main reason for the inconsistency of the fluxes.





397	Assuming that the EC fluxes obtained with PP2 are valid, the discrepancy between
398	the EC and BF fluxes was also postulated to reflect the limitations of the BF method
399	and/or the difference of the measurement height of the BF and EC methods; the former
400	and the latter are at the water surface and the height of the EC devices, respectively. As
401	for the BF method limitaitons, seagrass leaves, which reached the water surface during
402	low tide at the study site, might have affected the physical and chemical conditions at
403	the water surface (Watanabe and Kuwae, 2015). In BF theory, the CO <sub>2</sub> flux is caused by
404	the CO <sub>2</sub> concentration gradient just below the water surface. The BF method should
405	therefore not be applied when seagrass is present on the water surface. A previous study
406	that investigated the radiocarbon isotopic signatures of seagrass at the study site
407	indicated that of the total CO <sub>2</sub> assimilated by the seagrass, 0–40 % (mean = 17 %)
408	originated from the atmosphere and the rest from the water (Watanabe and Kuwae,
409	2015). The implication is that there is direct uptake of atmospheric $CO_2$ (rather than
410	uptake through the water column) by seagrass when seagrass leaves are on the water
411	surface. Atmospheric CO <sub>2</sub> is therefore directly taken up within a thin film of water over
412	the seagrass leaves, but this seagrass-driven CO2 flux is not included in the BF flux
413	calculations.
414	The fact that the $\mathrm{CO}_2$ flux was larger than the BF flux may have been partly caused
415	by the temperature and CO <sub>2</sub> gradients in the atmospheric layer between the EC
416	measurement height and the water surface. Vertical gradients in air temperature
417	frequently occur because of the large difference in temperature between the atmosphere
418	and water. Indeed, the difference between the temperature at the EC measurement
419	height and the water temperature ( $\Delta T$ ) ranged from +8 °C to -10 °C during the
420	measurement period (Fig. 10). The atmospheric CO <sub>2</sub> concentration was inversely





- 421 proportional to air temperature, mainly because of the ideal gas law. The inflow of air
- 422 masses with different temperatures and CO<sub>2</sub> concentrations from other regions
- 423 surrounding the lagoon (e.g., terrestrial environments and open sea) could also have
- 424 contributed to this inverse relationship. We therefore believe that the  $\Delta T$  caused a
- vertical gradient in atmospheric CO<sub>2</sub> concentration, the result being a vertical CO<sub>2</sub> flux
- 426 in accord with Fick's law. Indeed, the absolute values of the EC CO<sub>2</sub> fluxes were large
- 427 when  $\Delta T$  was large (Fig. 11b). For example, a large negative EC flux that was more than
- 428 10 times the maximum negative BF flux was observed when the  $\Delta T$  was negative and
- 429 large in magnitude. This phenomenon frequently occurs when a cold air mass overlies a
- 430 warm air mass (Fig. 11a). The opposite phenomenon is observed when a warm air mass
- 431 overlies a cold air mass (Fig. 11b). The positive temperature gradient did not result in a
- 432 large EC flux similar to the flux associated with the negative temperature gradient
- 433 because the positive temperature gradient produced stable stratification and thereby
- 434 prevented vertical eddy movement and CO<sub>2</sub> flux, a phenomenon observed in the case of
- 435 terrestrial EC fluxes at night (Aubinet and Feigenwinter, 2010). Nevertheless, a
- 436 temperature gradient-driven vertical CO<sub>2</sub> flux may explain the discrepancy between the
- 437 EC and BF fluxes.

In summary, we attribute the discrepancy between the EC and BF fluxes to the following four factors: (1) major technical uncertainties in both methods; (2) differences in measurement location and in the temporal scales of the measurements; (3) limitations of the BF method related to the presence of vegetation on the water surface; and (4) the gradient of atmospheric conditions between the height of the EC measurements and just above the water surface (Table 1). The latter two factors may cause the EC CO<sub>2</sub> flux to be larger than the BF flux in aquatic systems that have large amounts of vegetation or





445	are located where temperature gradients form in the atmospheric boundary layer.
446	Determination of the contribution of aquatic ecosystems to mitigating the adverse
447	effects of climate change will require consideration of all processes related to
448	atmosphere-aquatic ecosystem exchange. For this purpose, the EC $\mathrm{CO}_2$ flux should be a
449	more robust indicator than the BF flux, which includes only processes related to air-
450	water exchanges. Improving the EC method is therefore essential for a re-evaluation of
451	atmosphere-aquatic ecosystem CO2 gas exchanges and comprehensive analyses of the
452	contributions of aquatic environments to mitigating the adverse effects of climate
453	change.
454	
455	Competing interests. The authors declare that they have no conflict of interests.
456	
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463	Climate Change).

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	EC	BF
Major sources of uncertainty	<ul> <li>Cross sensitivity</li> <li>Long-term variation (minutes) of CO<sub>2</sub> and water vapor concentrations in air</li> </ul>	<ul> <li>Wind-dependent formula</li> <li>Heterogeneity of measurement site</li> </ul>
Measurement location and scales	- CO <sub>2</sub> flux at EC devices - 100 m to km	<ul> <li>CO<sub>2</sub> flux at the air-water surface</li> <li>&lt; 100 m</li> </ul>
Vegetation on the water surface	- effect included	- effect not included
Atmospheric gradient between the measurement height and water surface	- effect included	- effect not included

### 588 Table 1. Summary of the differences in fluxes calculated by the EC and BF methods

589 590





### 591 Figure captions

- 592 Figure 1. Location of the measurement site (Furen Lagoon, Hokkaido, Japan). The
- 593 lagoon is shallow (1–2 m). Shading indicates seagrass meadows. Eddy covariance (EC)
- measurements were performed on the platform in 2014.
- 595
- 596 Figure 2. Post-processing (PP) methods for EC flux calculation. The EC CO<sub>2</sub> flux was

597 calculated by using the conventional PP method (PP1) in EddyPro. The corrections

involved in PP1 have been described in previous publications (e.g., Lee et al., 2004).

599 Detrending was performed by using block averaging (BA). Our new PP method (PP2)

600 included three data-filtering steps based on the received signal strength indication of the

601 CO<sub>2</sub> sensor and the standard deviation of the CO<sub>2</sub> and water vapor concentrations

602 divided by the corresponding average standard deviation during the measurement period

603 (nSD). Detrending in PP2 was performed by using high-pass filtering (Massman, 2000).

604

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Figure 3. EC CO<sub>2</sub> fluxes with (a) PP1 (mean: -1.93 \ \mu mol \ m^{-2} \ s^{-1}, SD: 52.4 \mu mol \ m^{-2}
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606 s<sup>-1</sup>, n = 2502) and with (b) PP2 (mean: -0.54 µmol m<sup>-2</sup> s<sup>-1</sup>, SD: 2.2 µmol m<sup>-2</sup> s<sup>-1</sup>, n =

607 1,833). Several data points in panel (a) are off the scale and not shown for comparison

with (b), in which all data are shown. The red line in (b) shows the theoretical maximum negative flux estimated from the BF method ( $-2.96 \text{ }\mu\text{mol }m^{-2}\text{ }s^{-1}$ ).

610

611 Figure 4. Instantaneous atmospheric CO<sub>2</sub> concentration (a), water vapor (atmospheric

- 612 H<sub>2</sub>O) concentration (b), and cumulative covariance of atmospheric CO<sub>2</sub> concentration
- and vertical wind speed calculated with PP1 when the CO<sub>2</sub> fluxes (c) showed the largest





- 614 positive value (156.5  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>; blue) and the largest negative value (-217.9  $\mu$ mol
- $m^{-2}$  s<sup>-1</sup>; red). Note that the covariance was not equal to the CO<sub>2</sub> flux because there was
- 616 no Webb-Pearman-Leuning correction.
- 617
- 618 **Figure 5.** Comparison between the effects of the three filtering parameters and the CO<sub>2</sub>
- 619 flux after the RSSI filtering procedure. The red circles indicate the top four outliers for
- 620 each parameter, which were determined as erroneous (not natural) fluxes. Each
- 621 threshold (broken red line) was determined so as to remove these data. (a) Normalized
- standard deviation (nSD = standard deviation over 30-min divided by the average
- during the entire measurement period; threshold = 0.05; 74 % of data retained). (b)
- 624 Skewness (threshold = 0.48; 19 % of data retained). (c) Absolute value of kurtosis
- 625 (threshold = 3.1; 20 % of data retained).
- 626
- 627 **Figure 6.** Comparison of CO<sub>2</sub> fluxes calculated by the block averaging (BA) method
- and EC flux with HP filtering. Most of the data lay on or close to the solid line (y = x)
- and were not much changed by HP filtering. However, in the case of the data in the

630 shaded area, HP filtering decreased the absolute value of fluxes by removing the long-

- term effect of CO<sub>2</sub> change (see Fig. 7).
- 632
- Figure 7. Examples of the deviation calculations of atmospheric CO<sub>2</sub> concentration (a)
  and the cumulative covariance of atmospheric CO<sub>2</sub> concentration and vertical wind
  speed (b).
- 636
- 637 Figure 8. Power spectra of CO<sub>2</sub> flux with PP1 and PP2. The spectra were normalized





- 638 using the covariances of CO<sub>2</sub> and vertical wind velocity for the entire measurement
- 639 period after PP1 and PP2. The shaded areas indicate the frequency of the 24-h diurnal
- 640 cycle (left) and 12.5-h tidal cycle (right). Average CO<sub>2</sub> flux data during the entire
- 641 measurement period were used to replace missing CO<sub>2</sub> flux data.
- 642
- 643 Figure 9. Comparison of BF flux, EC flux with PP1, and EC flux with PP2 in May (a),
- 644 July (b) and September (c).
- 645
- Figure 10. (a) Relationship between air temperatures and atmospheric CO<sub>2</sub>
- 647 concentrations. The gray circles indicate the data averaged over 30 min. The open
- diamonds and error bars indicate the binned averages every 1 °C and 1 SD, respectively.
- 649 The solid line indicates the slope estimated from the change in air volume assuming that
- 650 CO<sub>2</sub> behaves as an ideal gas. (b) Relationship and between temperature difference ( $\Delta T$ )
- and CO<sub>2</sub> flux. The gray circles indicate the EC flux data calculated every 30 min. The
- solid diamonds and error bars indicate the binned averages every 1 °C and 1 SD,
- respectively. Note that for clarity not all data are plotted on both graphs.
- 654
- **Figure 11.** Schematic diagram showing the relationships between negative temperature
- 656 gradient and negative EC CO<sub>2</sub> flux (a) and between positive temperature gradient and
- 657 positive EC CO<sub>2</sub> flux (b). The atmospheric CO<sub>2</sub> gradient reflects the air temperature
- gradient and assumes that CO<sub>2</sub> behaves like an ideal gas. The temperature gradient-
- driven vertical CO<sub>2</sub> flux may explain the discrepancy between the EC and BF fluxes.



































































