Author response:

We thank the Associate Editor for her comments and have edited the manuscript (yellow highlight) in accordance with the her suggestion. We clarify that we are not hypothesizing that contamination during or after sample collection is what occurred, but rather we are arguing that it is a valid option (along with environmental release), which procedural quality control measures would exclude or identify. We have also added an Abstract, cut from our original cover letter, which should also help to make this position clear while also summarizing the comment.

We have also added reference to the ¹⁴C-CH₄ methodology for natural waters developed by Dean et al. (2017), which we neglected to include in the initial submission.

Lastly, in the third paragraph, we include a sentence cut from our response to RC1, which clearly points out that because a data table containing methane concentration, stable isotope, and radiocarbon information for each sample is not in the article, readers are again blocked from drawing independent conclusions related to this data.

Thank you,

Katy Sparrow and John Kessler

1 Comment on "The origin of methane in the East Siberian Arctic Shelf unraveled with triple 2 isotope analysis," by Sapart et al. (2017) 3 Katy J. Sparrow and John D. Kessler 4 Department of Earth and Environmental Sciences, University of Rochester, Rochester, New York 5 Correspondence email: katysparrow@gmail.com (K.J.S.); john.kessler@rochester.edu (J.D.K.) 6 7 **Abstract** 8 9 In this comment, we outline two major concerns regarding some of the key data presented in this 10 paper. Both of these concerns are associated with the natural abundance radiocarbon-methane 11 (¹⁴C-CH₄) data. First, no systematic methodology is presented, nor previous peer-reviewed 12 publication referenced, for how these samples were collected, prepared, and ultimately analyzed 13 for ¹⁴C-CH₄. Not only are these procedural details missing, but the critical evaluation of them 14 using gaseous and aqueous blanks and standards was omitted although these details are essential 15 for any reader to evaluate the quality of data and subsequent interpretations. Second, due to the 16 lack of methodological details, the source of the sporadic anthropogenic contamination cannot be 17 determined and thus it is premature for the authors to suggest it was in the natural environment 18 prior to sample collection. As the natural ¹⁴C-CH₄ data are necessary for the authors' stated 19 scientific objectives of understanding the origin of methane in the East Siberian Arctic Shelf, our 20 comment serves to highlight that the study's objectives have not been met. 21 22 23

In the article titled, "The origin of methane in the East Siberian Arctic Shelf unraveled with triple

25 isotope analysis," (5 May, p. 2283, doi:10.5194/bg-14-2283-2017), Célia Sapart and coauthors

present natural abundance radiocarbon-methane (14C-CH₄) measurements from Laptev Sea

27 sediments and waters alongside methane concentration and methane stable isotope

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28 measurements. The authors then draw conclusions about methane source-sink dynamics

operating in this arctic shelf sea based upon these methane data. Two concerns with the ¹⁴C-CH₄

data lead us to question whether these data should be used to interpret this natural system.

The first issue is that the method used to collect and prepare the ¹⁴C-CH₄ samples is inadequately described by Sapart et al. and there is no quality control data presented. Radiocarbon-methane is not a routine measurement in natural waters because of the challenges associated with sampling and preparing a trace isotope of a trace gas. In the methods section of the article, the authors cite two techniques that relate only to the ¹⁴C-accelerator mass spectrometry (AMS) analysis, while the methodologies used for the sample collection and preparation steps leading up to the ¹⁴C analyses of sediment and seawater samples are absent. The natural ¹⁴C-CH₄ content of a sample can be affected by carbon and CH₄ added from the materials it encounters and by any contact with the atmosphere, so quality control measures are necessary to ensure that a sample is not significantly contaminated prior to analysis and that any minor contamination (i.e. blank addition) is accounted for in the final results. In the supplement, the authors write that, "None of the reference and blank measurements were abnormal," without presenting any descriptions of or data stemming from these tests. Refereed techniques for collecting and preparing ¹⁴C-CH₄ samples from natural waters (Dean et al., 2017; Elder et al., 2018; Kessler and Reeburgh, 2005; Pack et al., 2015; Pohlman et al., 2000; Sparrow and Kessler, 2017) include detailed qualitative and quantitative descriptions of the measures taken to validate their methodologies. These measures include processing blank (methane-free) waters and treating methane-free gas and methane of known ¹⁴C-CH₄ content in the same way as samples. As the ¹⁴C-AMS measurement error is typically very low relative to ¹⁴C-CH₄ collection and preparation procedures, we can only assume that the error associated with the processes that most greatly affects the precision,

The second issue that calls the integrity of this study's ¹⁴C-CH₄ data into question is the existence of super-modern sediment and water column samples (approaching 100 times above modern) in the dataset. As the authors correctly reference, elevated ¹⁴C-CH₄ has previously been documented in other ocean waters (Kessler et al., 2008), however, the values presented here are up to 27 times higher than any previously reported elevated value. It is suggested in the main text and in the supplement that the source of the "highly enriched ¹⁴C" is anthropogenic and that it existed in the natural environment prior to sampling. We argue that it is premature to suggest an origin of this enriched ¹⁴C, either environmental release or contamination (incurred during sample collection, processing, and/or analysis) when the ¹⁴C-CH₄ methodological details, with appropriate standards and blanks, are absent from the article. The possibility that the enriched ¹⁴C was derived from the sampling equipment, vessel, and/or land-based laboratories was largely dismissed by the authors, while we attest that it is a valid option. The authors discount the possibility that their samples were contaminated during the sampling process, "because no radioactive tracers were used during the sampling expeditions." This argument is untenable because the half-life of ¹⁴C is 5730 years, meaning any surface contamination will persist for tens of thousands of years—well beyond the

sensitivity, and accuracy of the reported ¹⁴C-CH₄ signature is unaccounted for by the authors.

67 specific project where it was used. In addition, the authors highlight that, for sediment samples, "the higher ¹⁴C values correspond to the lower CH₄ concentrations," to suggest that a small 68 amount of radioactive contamination in the environment was added to a variable background of 69 naturally occurring CH₄, which would most greatly affect the ¹⁴C signature of the smallest sized 70 71 (lowest CH₄ concentration) samples. This may be true, but another scenario that is also valid 72 using the same logic is that the contamination was added during the ¹⁴C-CH₄ sample collection and/or preparation processes. This relationship was noted for sediment samples, but we are not 73 74 informed in the article or supplement on the relationship between CH₄ concentration and ¹⁴C-CH₄ content for the seawater samples. The lack of a data table containing the specific triple-isotope 75 76 information for each CH₄ sample, in the article or in a data repository, has the effect of making this study unnecessarily opaque for a reader attempting to draw conclusions for themselves. The 77 78 authors clearly state that additional experiments are necessary to determine the unknown origin of 79 this isotopic enrichment, however, without that complimentary data, or at least data that proves it was in the sediments and waters prior to sample collection, its presence invalidates all ¹⁴C-CH₄ 80 81 data presented in this study from contributing to our understanding of methane dynamics in the 82 Arctic Ocean. 83 In a recently published study, we demonstrate how useful natural abundance ¹⁴C-CH₄ 84 measurements can be towards understanding the role of ancient sources of methane in arctic shelf seas (Sparrow et al., 2018). Importantly, in this study, we find that the stable isotope (δ^{13} C-CH₄) 85 and dissolved CH₄ concentration data, together, would suggest an entirely different (and, we 86 87 argue, incorrect) interpretation of this system, which attests to the importance of ¹⁴C-CH₄ measurements for investigations into the origins of methane. When conducting natural abundance 88 89 ¹⁴C-CH₄ studies, it is imperative that we do so using peer review published methods with 90 appropriate radiocarbon blanks and standards; otherwise, interpretations made from ¹⁴C-CH₄ data 91 are unverifiable and inconclusive. 92 93 94 95 96 97 98 99 100 101 102

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