

1 **Effect of elevated $p\text{CO}_2$ on trace gas production during an**
2 **ocean acidification mesocosm experiment**

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27

28 **Abstract**

29 A mesocosm experiment was conducted in Wuyuan Bay (Xiamen), China to investigate the effects of elevated
30 $p\text{CO}_2$ on the phytoplankton species *Phaeodactylum tricornutum* (*P. tricornutum*), *Thalassiosira weissflogii* (*T.*
31 *weissflogii*) and *Emiliania huxleyi* (*E. huxleyi*) and their production ability of dimethylsulfide (DMS),
32 dimethylsulfoniopropionate (DMSP), as well as four halocarbon compounds bromodichloromethane (CHBrCl_2),
33 methyl bromide (CH_3Br), dibromomethane (CH_2Br_2) and iodomethane (CH_3I). Over a period of 5 weeks, *P.*
34 *tricornutum* outcompeted *T. weissflogii* and *E. huxleyi*, comprising more than 99% of the final biomass. During
35 the logarithmic growth phase (phase I), mean DMS concentration in high $p\text{CO}_2$ mesocosms (1000 μatm) was 28%
36 lower than that in low $p\text{CO}_2$ mesocosms (400 μatm). Elevated $p\text{CO}_2$ led to a delay in DMSP-consuming bacteria
37 concentrations attached to *T. weissflogii* and *P. tricornutum* and finally resulted in the delay of DMS concentration
38 in the high $p\text{CO}_2$ treatment. Unlike DMS, the elevated $p\text{CO}_2$ did not affect DMSP production ability of *T.*
39 *weissflogii* or *P. tricornutum* throughout the 5 weeks culture. A positive relationship was detected between CH_3I
40 and *T. weissflogii* and *P. tricornutum* during the experiment, and there was a 40% reduction in mean CH_3I
41 concentration in the high $p\text{CO}_2$ mesocosms. CHBrCl_2 , CH_3Br , and CH_2Br_2 concentrations did not increase with
42 elevated chlorophyll *a* (Chl *a*) concentrations compared with DMS(P) and CH_3I , and there were no major peaks
43 both in the high $p\text{CO}_2$ or low $p\text{CO}_2$ mesocosms. In addition, no effect of elevated $p\text{CO}_2$ was identified for any of
44 the three bromocarbons.

45 **Keywords:** ocean acidification, dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), halocarbons,
46 phytoplankton, bacteria

47

48 **1. Introduction**

49 Anthropogenic emissions have increased the fugacity of atmospheric carbon dioxide ($p\text{CO}_2$) from
50 the pre-industrial value of 280 μatm to the present-day value of over 400 μatm , and these values
51 will further increase to 800–1000 μatm by the end of this century (Gattuso et al., 2015). The
52 dissolution of this excess CO_2 into the surface of the ocean directly affects the carbonate system
53 and has lowered the pH by 0.1 units, from 8.21 to 8.10 over the last 250 years. Further decreases
54 of 0.3–0.4 pH units are predicted by the end of this century (Doney et al., 2009; Orr et al., 2005;
55 Gattuso et al., 2015), which is commonly referred to as ocean acidification. The physiological and
56 ecological aspects of the phytoplankton response to this changing environment can potentially
57 alter marine phytoplankton community composition, community biomass, and feedback to
58 biogeochemical cycles (Boyd and Doney, 2002). These changes simultaneously have an impact on
59 some volatile organic compounds produced by marine phytoplankton (Liss et al., 2014; Liu et al.,
60 2017), including the climatically important trace gas dimethylsulfide (DMS) and a number of
61 volatile halocarbon compounds.

62 DMS is the most important volatile sulfur compound produced from
63 dimethylsulfoniopropionate (DMSP), which is ubiquitous in marine environments, mainly
64 synthesized by marine microalgae (Stefels et al., 2007), a few angiosperms, some corals (Raina et
65 al., 2016), and several heterotrophic bacteria (Curson et al., 2017) through complex biological
66 interactions in marine ecosystems. Although it remains controversial, DMS and its by-products,
67 such as methanesulfonic acid and non-sea-salt sulfate, are suspected to have a prominent part in
68 climate feedback (Charlson et al., 1987; Quinn and Bates, 2011). The conversion of DMSP to
69 DMS is facilitated by several enzymes, including DMSP-lyase and acyl CoA transferase

70 (Kirkwood et al., 2010; Todd et al., 2007); these enzymes are mainly found in phytoplankton,
71 macroalgae, symbiodinium, bacteria and fungi (de Souza and Yoch, 1995; Stefels and Dijkhuizen,
72 1996; Steinke and Kirst, 1996; Bacic and Yoch, 1998; Yost and Mitchelmore, 2009). Several
73 studies have shown a negative impact of decreasing pH on DMS-production capability (Hopkins
74 et al., 2010; Avgoustidi et al., 2012; Archer et al., 2013; Webb et al., 2016), while others have
75 found either no effect or a positive effect (Vogt et al., 2008; Hopkins and Archer, 2014). Several
76 assumptions have been presented to explain these contrasting results and attributed the
77 pH-induced variation in DMS-production capability to altered physiology of the algae cells or of
78 bacterial DMSP degradation (Vogt et al., 2008; Hopkins et al., 2010, Avgoustidi et al., 2012;
79 Archer et al., 2013; Hopkins and Archer, 2014; Webb et al., 2015).

80 Halocarbons also play a significant role in the global climate because they are linked to
81 tropospheric and stratospheric ozone depletion and a synergistic effect of chlorine and bromine
82 species has been reported [accounting](#) for approximately 20% of the polar stratospheric ozone
83 depletion (Roy et al., 2011). In addition, iodocarbons can release atomic iodine quickly through
84 photolysis in the atmospheric boundary layer and [iodine](#) atoms are very efficient in the catalytic
85 removal of O₃, which governs the lifetime of many climate relevant gases including methane and
86 DMS (Jenkins et al., 1991). Compared with DMS, limited attention was received about the effect
87 of ocean acidification on halocarbon concentrations. Hopkins et al. (2010) and Webb et al. (2015)
88 measured lower concentrations of several iodocarbons, while bromocarbons were unaffected by
89 elevated *p*CO₂ [in](#) two acidification experiments. In addition, [another](#) mesocosm study did not elicit
90 significant differences from any halocarbon compounds at up to 1,400 μatm *p*CO₂ (Hopkins et al.,
91 2013).

92 Taken together, the data indicate that the response of DMS and halocarbon release to elevated
93 $p\text{CO}_2$ is complex and controversial. DMS and halocarbons play a significant role in the global
94 climate and will perhaps act to a greater extent in the future. An intermediate step between
95 laboratory and natural community field experiments was designed in this study to understand the
96 response of the release of DMS and halocarbon to ocean acidification in Chinese coastal seas
97 using isolates of non-axenic phytoplankton added to filtered natural water. We hypothesized that
98 the response of DMS and halocarbon release to elevated $p\text{CO}_2$ in natural seawater can be better
99 presented after minimizing the shifting composition of the natural phytoplankton and microbial
100 communities.

101 **2. Experimental method**

102 *2.1 Experimental setup*

103 To investigate the response of DMS and halocarbon release to ocean acidification, a mesocosm
104 experiment was carried out on a floating platform (set in seawater, about 150 m from the shore) at
105 the Facility for Ocean Acidification Impacts Study of Xiamen University (FOANIC-XMU,
106 24.52°N , 117.18°E) (for full technical details of the mesocosms, see Liu et al. 2017). Six
107 cylindrical transparent thermoplastic polyurethane bags with domes were deployed along the
108 south side of the platform. The width and depth of each mesocosm bag was 1.5 m and 3 m,
109 respectively. Filtered (0.01 μm ultrafiltration water purifier, MU801-4T, Midea, Guangdong,
110 China) in situ seawater was pumped into the six bags simultaneously within 24 h. A known
111 amount of NaCl solution was added to each bag to calculate the exact volume of seawater in the
112 bags, according to a comparison of the salinity before and after adding salt (Czerny et al., 2013).
113 The initial in situ $p\text{CO}_2$ was about 650 μatm . To set the low (400 μatm) and high $p\text{CO}_2$ (1000 μatm)

114 levels, we added Na₂CO₃ solution and CO₂ saturated seawater to the mesocosm bags to alter total
115 alkalinity and dissolved inorganic carbon (Gattuso et al., 2010; Riebesell et al., 2013).
116 Subsequently, during the whole experimental process, air at the ambient (400 µatm) and elevated
117 pCO₂ (1000 µatm) concentrations was continuously bubbled into the mesocosm bags using a CO₂
118 Enricher (CE-100B, Wuhan Ruihua Instrument & Equipment Ltd., Wuhan, China). Seawater taken
119 from the coastal environment was first filtered to remove algae and their attached bacteria before
120 usage in mesocosm bags. Bacterial abundance in the pre-filtered water was less than 10³ cell mL⁻¹,
121 which was three magnitudes lower than the bacterial abundance in the natural water and close to
122 the detection limit of the flow cytometer. The trace gases, including DMS, bromodichloromethane
123 (CHBrCl₂), methyl bromide (CH₃Br), dibromomethane (CH₂Br₂), and iodomethane (CH₃I)
124 produced in the environment did not affect the mesocosm trace gas concentrations after the bags
125 were sealed.

126 2.2 Algal strains

127 Before being introduced into the mesocosms, the three phytoplankton species *Phaeodactylum*
128 *tricornutum* (*P. tricornutum*), *Thalassiosira weissflogii* (*T. weissflogii*) and *Emiliana huxleyi* (*E.*
129 *huxleyi*) were cultured in autoclaved, pre-filtered seawater from Wuyuan Bay at 16°C (similar to
130 the in situ temperature of Wuyuan Bay) without any addition of nutrients. Cultures were
131 continuously aerated with filtered ambient air containing 400 µatm of CO₂ within plant chambers
132 (HP1000G-D, Wuhan Ruihua Instrument & Equipment, China) at a constant bubbling rate of 300
133 mL min⁻¹. The culture medium was renewed every 24 hrs to maintain the cells of each
134 phytoplankton species in exponential growth. When the experiment began, these three
135 phytoplankton species were inoculated into the mesocosm bags, with an initial

136 diatom/coccolithophorid cell ratio of 1:1. The initial concentrations of *P. tricornutum*, *T.*
137 *weissflogii*, and *E. huxleyi* inoculated into the mesocosm were 10, 10, and 20 cells mL⁻¹,
138 respectively. *P. tricornutum* and *T. weissflogii* were obtained from the Center for Collections of
139 Marine Bacteria and Phytoplankton of the State Key Laboratory of Marine Environmental Science
140 (Xiamen University). *P. tricornutum* was originally isolated from the South China Sea in 2004
141 and *T. weissflogii* was isolated from Daya Bay in the coastal South China Sea. *E. huxleyi* was
142 originally isolated in 1992 from the field station of the University of Bergen (Raunefjorden;
143 60°18'N, 05°15'E).

144 2.3 Sampling for DMS(P) and halocarbons

145 DMS(P) and halocarbon samples were taken from the above mentioned mesocosm bags at 9 a.m.,
146 then all collected samples were transported into a dark cool box back to the laboratory onshore for
147 analysis within 1 hr. For the DMS analysis, 2 mL sample was gently filtered through a 25 mm
148 GF/F (glass fiber) filter and transferred to a purge and trap system linked to a Shimadzu GC-2014
149 gas chromatograph (Tokyo, Japan) equipped with a glass column packed with 10% DEGS on
150 Chromosorb W-AW-DMCS (3 m × 3 mm) and a flame photometric detector (Zhang et al., 2014).
151 For total DMSP analysis, 10 mL water sample was fixed using 50 μL of 50 % H₂SO₄ and sealed
152 (Kiene and Slezak, 2006). After > 1 d preservation, DMSP samples were hydrolyzed for 24 h with
153 a pellet of KOH (final pH > 13) to fully convert DMSP to DMS. Then, 2 mL of the hydrolyzed
154 sample was carefully transferred to the purge and trap system mentioned above for extraction of
155 DMS. For halocarbons, 100 mL sample was purged at 40°C with pure nitrogen at a flow rate of
156 100 mL min⁻¹ for 12 min using another purge and trap system coupled to an Agilent 6890 gas
157 chromatograph (Agilent Technologies, Palo Alto, CA, USA) equipped with an electron capture

158 detector (ECD) as well as a 60 m DB-624 capillary column (0.53 mm ID; film thickness, 3 μm)
159 (Yang et al., 2010). The analytical precision for duplicate measurements of DMS(P) and
160 halocarbons was $> 10\%$.

161 *2.4 Measurements of chlorophyll a*

162 Chlorophyll *a* (Chl *a*) was measured in water samples (200–1,000 mL) collected every 2 d at 9
163 a.m. by filtering onto Whatman GF/F filters (25 mm). The filters were placed in 5 mL 100%
164 methanol overnight at 4°C and centrifuged at 5000 r min^{-1} for 10 min. The absorbance of the
165 supernatant (2.5 mL) was measured from 250 to 800 nm using a scanning spectrophotometer (DU
166 800, Beckman Coulter Inc., Brea, CA, USA). Chl *a* concentration was calculated according to the
167 equation reported by Porra (2002).

168 *2.5 Enumeration of DMSP-consuming bacteria*

169 The number of DMSP-consuming bacteria [in the mesocosms](#) was estimated using the most
170 probable number methodology. The medium consisted of a mixture (1:1 v/v) of sterile artificial
171 sea water and mineral medium (Visscher et al., 1991), 3 mL of which was dispensed [into](#) 6 mL test
172 tubes, which were closed by an over-sized cap, allowing gas exchange. Triplicate dilution series
173 were set up. All test tubes contained 1 mmol L^{-1} DMSP as the sole organic carbon source and
174 were kept at 30 °C in the dark. After 2 weeks, the presence/absence of bacteria in the tubes was
175 verified by DAPI staining (Porter and Feig, 1980). Three tubes containing 3 mL ASW without
176 substrate were used as controls.

177 *2.6 Statistical analysis*

178 One-way analysis of variance (ANOVA), Tukey's test, and the two-sample t-test were carried out
179 to demonstrate the differences between treatments. A p-value < 0.05 was considered significant.

180 Relationships between DMS(P), halocarbons and a range of other parameters were detected using
181 Pearson's correlation analysis via SPSS 22.0 for Windows (SPSS Inc., Chicago, IL, USA).

182 **3. Results and Discussion**

183 *3.1 Temporal changes in pH, Chl a, P. tricornutum, T. weissflogii, and E. huxleyi during the* 184 *experiment*

185 During the experiment, the seawater in each mesocosm was well **mixed**, and the temperature and
186 salinity **remained stable**, with means of 16°C and 29, **respectively**, in all mesocosm bags. We
187 observed significant differences in pH levels between the two CO₂ treatments on days 0–11, but
188 the differences disappeared with subsequent phytoplankton growth (Fig. 1). The phytoplankton
189 growth process was divided into three phases in terms of variations in Chl *a* concentrations in the
190 mesocosm experiments as described in Liu et al. (2017): i) the logarithmic growth phase (phase I,
191 days 0–13), ii) a plateau phase (phase II, days 13–23, bloom period), and iii) a secondary plateau
192 phase (phase III, days 23–33) attained after a decline in biomass from a maximum in phase II. The
193 initial chemical parameters of the mesocosm experiment are shown in Table 1. The initial mean
194 dissolved nitrate (including NO₃⁻ and NO₂⁻), NH₄⁺, PO₄³⁻ and silicate (SiO₃²⁻) concentrations were
195 54, 20, 2.6 and 41 μmol L⁻¹, **respectively** for the low pCO₂ treatment and 52, 21, 2.4 and 38 μmol
196 L⁻¹, **respectively** for the high pCO₂ treatment. The nutrient concentrations (NO₃⁻, NO₂⁻, NH₄⁺ and
197 phosphate) during phase I were consumed rapidly and their concentrations were below or close to
198 the detection limit during phase II (Table 1). **SiO₃²⁻ was detectable during the entire experimental**
199 **period, and was unlikely to be a limiting factor for phytoplankton growth during the experiment.**
200 In addition, although dissolved inorganic nitrogen (NH₄⁺, NO₃⁻, and NO₂⁻) and phosphate were
201 depleted, Chl *a* concentration in both treatments (biomass dominated by *P. tricornutum*)

202 remained constant over days 12–22, and then declined over subsequent days. *T. weissflogii* was
203 found throughout the entire period in each bag, but the maximum concentration was 8,120 cells
204 mL⁻¹, which was far less than the concentration of *P. tricornutum* with a maximum density of
205 about 1.5 million cells mL⁻¹ (Liu et al., 2017). It is possible that *P. tricornutum* outcompeted *T.*
206 *weissflogii* because of its higher surface to volume ratio and/or species-specific physiology, which
207 would enhance the efficiency of nutrient uptake and related metabolism (Alessandrade et al.,
208 2007). *E. huxleyi* was only found in phase I and its maximal concentration reached 310 cells mL⁻¹
209 according to the results of Liu et al. (2017). Previous studies have reported that the maximum
210 specific growth rate of *T. weissflogii* and *P. tricornutum* is about 1.2 d⁻¹ (Li et al., 2014; Sugie and
211 Yoshimura, 2016), while that of *E. huxleyi* is about 0.8 d⁻¹ (Xing et al., 2015). This might be the
212 main reason why diatoms overwhelmingly outcompeted the coccolithophores during this
213 experiment.

214 3.2 Impact of elevated pCO₂ on DMS and DMSP production

215 DMSP concentrations in the high pCO₂ and low pCO₂ treatments increased significantly along
216 with the increase in Chl *a* concentrations and algal cells, and remained relatively constant over the
217 following days. A significant positive relationship was observed between DMSP and
218 phytoplankton in the experiment ($r = 0.961$, $p < 0.01$ for *P. tricornutum*, $r = 0.617$, $p < 0.01$ for *T.*
219 *weissflogii* in the low pCO₂ treatment, Table 2; $r = 0.954$, $p < 0.01$ for *P. tricornutum*, $r = 0.743$, p
220 < 0.01 for *T. weissflogii* in the high pCO₂ treatment, Table 3). DMS was maintained at a low level
221 during phase I (mean of 1.03 nmol L⁻¹ in the low pCO₂ and 0.74 nmol L⁻¹ in the high pCO₂
222 treatments, respectively) compared with DMSP. DMS concentrations began to increase rapidly on
223 day 15, peaked on day 25 in the low pCO₂ treatment (112.1 nmol L⁻¹) and on day 29 in the high

224 $p\text{CO}_2$ treatment ($101.9 \text{ nmol L}^{-1}$) respectively, and then decreased in the following days. A
225 moderate positive relationship was observed between DMS and *P. tricornuntum* ($r = 0.560$, $p <$
226 0.05 in the low $p\text{CO}_2$ treatment; $r = 0.635$, $p < 0.01$ in the high $p\text{CO}_2$ treatment), while no
227 relationship was observed between DMS and *T. weissflogii* (Table 2 and Table 3) during the
228 experiment. Similar to DMS, DMSP-consuming bacteria also maintained a low level during phase
229 I (mean of 0.57×10^6 and 0.40×10^6 cells mL^{-1} in the low $p\text{CO}_2$ and high $p\text{CO}_2$ treatments,
230 respectively). DMSP-consuming bacterial concentrations respectively peaked on days 19 ($11.65 \times$
231 10^6 cells mL^{-1}) and 21 (10.70×10^6 cells mL^{-1}) in the low $p\text{CO}_2$ and high $p\text{CO}_2$ treatments.

232 In this study, no difference in mean DMSP concentrations was observed between the two
233 treatments, indicating that elevated $p\text{CO}_2$ had no significant influence on DMSP production in *P.*
234 *tricornuntum* and *T. weissflogii*. However, significant reductions in mean DMS concentration
235 (28%) and DMSP-consuming bacteria (29%) were detected during phase I in the high $p\text{CO}_2$
236 treatment compared with those in the low $p\text{CO}_2$ treatment, indicating that elevated $p\text{CO}_2$ inhibited
237 DMSP-consuming bacteria and DMS production during the logarithmic growth phase. In addition,
238 the peak DMS concentration in the high $p\text{CO}_2$ treatment was delayed 4 days relative to that in the
239 low $p\text{CO}_2$ treatment during phase II (Fig. 2-A). This result has been observed in previous
240 mesocosm experiments and it was attributed to small scale shifts in community composition and
241 succession (Vogt et al., 2008; Webb et al., 2016). However, this phenomenon during the present
242 study can be explained in another straightforward way. Previous studies have shown that marine
243 bacteria play a key role in DMS production, and that the efficiency of bacteria converting DMSP
244 to DMS may vary from 2 to 100% depending on the nutrient status of the bacteria and the quantity
245 of dissolved organic matter (Simó et al., 2002, 2009; Kiene et al., 1999, 2000). In addition, a

246 positive relationship was observed between DMS and DMSP-consuming bacteria ($r = 0.643$, $p <$
247 0.01 in the low $p\text{CO}_2$ treatment; $r = 0.544$, $p < 0.01$ in the high $p\text{CO}_2$ treatment) during this
248 experiment. All of these observations point to the importance of bacteria in DMS and DMSP
249 dynamics. During the present mesocosm experiment, DMSP concentrations in the low $p\text{CO}_2$
250 treatment decreased slightly on day 23, while the slight decrease appeared on day 29 in the high
251 $p\text{CO}_2$ treatment (Fig. 2-B). In addition, the time that the DMSP concentration began to decrease
252 was very close to the time when the highest DMS concentration occurred in both treatments.
253 Similar to DMS, DMSP-consuming bacteria was also delayed in the high $p\text{CO}_2$ mesocosm
254 compared to that in the low $p\text{CO}_2$ mesocosm. Taken together, we inferred that the elevated $p\text{CO}_2$
255 first delayed growth of DMSP-consuming bacteria, then the delayed DMSP-consuming bacteria
256 postponed the DMSP degradation process, and eventually delayed the DMS concentration in the
257 high $p\text{CO}_2$ treatment. In addition, considering that algae and bacteria in natural seawater were
258 removed through a filtering process before the experiment (Huang et al., 2018), we further
259 concluded that the elevated $p\text{CO}_2$ controlled DMS concentrations mainly by affecting
260 DMSP-consuming bacteria attached to *T. weissflogii* and *P. tricornuntum*.

261 3.3 Impact of elevated $p\text{CO}_2$ on halocarbon compounds

262 The temporal development in CHBrCl_2 , CH_3Br , and CH_2Br_2 concentrations is shown in Fig. 3
263 (A–C) and the temporal changes of their concentrations were substantially different from those of
264 DMS, DMSP, *P. tricornuntum* and *T. weissflogii*. The mean concentrations of CHBrCl_2 , CH_3Br ,
265 and CH_2Br_2 for the entire experiment were 8.58, 7.85, and 5.13 pmol L^{-1} in the low $p\text{CO}_2$
266 treatment and 8.81, 9.73, and 6.27 pmol L^{-1} in the high $p\text{CO}_2$ treatment. The concentrations of
267 CHBrCl_2 , CH_3Br , and CH_2Br_2 did not increase with the Chl *a* concentration compared with those

268 of DMS and DMSP, and no major peaks were detected in the mesocosms. In addition, no effect of
269 elevated $p\text{CO}_2$ was identified for any of the three bromocarbons, which compared well with
270 previous mesocosm findings (Hopkins et al., 2010, 2013; Webb et al., 2016). No clear correlation
271 was observed between the three bromocarbons and any of the measured algal groups (table 2 and
272 table 3), indicating that *P. tricornutum* and *T. weissflogii* did not primarily release these three
273 bromocarbons during the mesocosm experiment. Previous studies reported that large-size
274 cyanobacteria, such as *Aphanizomenon flos-aquae*, could produce bromocarbons (Karlsson et al.,
275 2008). Significant correlations between the abundance of cyanobacteria and several bromocarbons
276 have been reported in the Arabian Sea (Roy et al., 2011). However, the filtration procedure led to
277 the loss of cyanobacteria in the mesocosms and finally resulted in low bromocarbon
278 concentrations during the experiment, although *P. tricornutum* and *T. weissflogii* abundances
279 were high.

280 The temporal dynamics of CH_3I in the high $p\text{CO}_2$ and low $p\text{CO}_2$ treatments are shown in Fig. 3-D.
281 The CH_3I concentrations in the low $p\text{CO}_2$ treatment varied from 0.38 to 12.61 pmol L^{-1} , with a
282 mean of 4.76 pmol L^{-1} . The CH_3I concentrations in the high $p\text{CO}_2$ treatment ranged between 0.44
283 and 8.78 pmol L^{-1} , with a mean of 2.88 pmol L^{-1} . The maximum CH_3I concentrations in the high
284 $p\text{CO}_2$ and low $p\text{CO}_2$ treatments were both observed on day 23. The range of CH_3I concentrations
285 during this experiment was similar to that measured in the mesocosm experiment ($< 1\sim 10$ pmol
286 L^{-1}) in Kongsfjorden conducted by Hopkins et al. (2013). In addition, the mean CH_3I
287 concentration in the low $p\text{CO}_2$ treatment was similar to that measured in the East China Sea, with
288 an average of 5.34 pmol L^{-1} in winter and 5.74 pmol L^{-1} in summer (Yuan et al., 2015).
289 Meanwhile, a positive relationship was detected between CH_3I and Chl *a*, *P. tricornutum* and *T.*

290 *weissflogii* ($r = 0.588$, $p < 0.01$ in the low $p\text{CO}_2$ treatment; $r = 0.834$, $p < 0.01$ in the low $p\text{CO}_2$
291 treatment for *P. tricornuntum*; $r = 0.680$ $p < 0.01$ in the low $p\text{CO}_2$ treatment; $r = 0.690$, $p < 0.01$ in
292 the high $p\text{CO}_2$ treatment for *Thalassiosira weissflogii*; $r = 0.717$, $p < 0.01$ in the low $p\text{CO}_2$
293 treatment; $r = 0.741$, $p < 0.01$ in the high $p\text{CO}_2$ treatment for Chl *a*). This result agrees with
294 previous mesocosm (Hopkins et al., 2013) and laboratory experiments (Hughes et al., 2013;
295 Manley and De La Cuesta, 1997) identifying diatoms as significant producers of CH_3I . Moreover,
296 similar to DMS, the maximum CH_3I concentration also occurred after the maxima of *P.*
297 *tricornuntum* and *T. weissflogii*, at about 4 d (Fig. 3-D). This result was similar to the conclusions
298 reported by Hopkins et al. (2010) and Wingenter et al. (2007) during two mesocosm experiments
299 conducted in Norway. Their results confirmed that iodocarbon gases generally occur after the Chl
300 *a* maxima. Furthermore, the mean CH_3I concentration measured in the high $p\text{CO}_2$ treatment was
301 significantly lower (40%) than that measured in the low $p\text{CO}_2$ treatment during the mesocosm
302 experiment. This result is in accordance with Hopkins et al. (2010) and Webb et al. (2015) who
303 also reported that elevated $p\text{CO}_2$ leads to a reduction in iodocarbon concentrations, but in contrast
304 to the findings of Hopkins et al. (2013) and Webb et al. (2016) who showed that elevated $p\text{CO}_2$
305 does not significantly affect the iodocarbon concentrations in the mesocosms. Considering that the
306 phytoplankton species did not show significant differences in the high $p\text{CO}_2$ and low $p\text{CO}_2$
307 treatments during the experiment, this reduction in the high $p\text{CO}_2$ treatment was likely not caused
308 by phytoplankton. Apart from direct biological production via methyl transferase enzyme activity
309 by both phytoplankton and bacteria (Amachi et al., 2001), CH_3I is produced from the breakdown
310 of higher molecular weight iodine-containing organic matter (Fenical, 1982) through
311 photochemical reactions between organic matter and light (Richter and Wallace, 2004). Both

312 bacterial methyl transferase enzyme activity and photochemical reaction could be responsible for
313 the reduction of CH₃I concentrations in the high pCO₂ treatment but further experiments are
314 needed to verify this result.

315 4. Conclusions

316 In this study, the effects of increased levels of pCO₂ on marine DMS(P) and halocarbons release
317 were studied in a controlled mesocosm facility. During the logarithmic growth phase, the elevated
318 pCO₂ led to a reduction in mean DMSP-consuming bacteria (29%) and DMS concentration (28%)
319 compared with those in the low pCO₂ treatment. In addition, a 4 d delay in DMS concentration
320 was observed in the high pCO₂ treatment due to the effect of elevated pCO₂ and we attribute this
321 delay in DMS concentration to the DMSP-consuming bacteria attached to *P. tricornutum* and *T.*
322 *weissflogii*. Affected by the filtration procedure, three bromocarbons compounds measured in this
323 study were not correlated with *P. tricornutum* and *T. weissflogii*, and Chl *a*. Besides, elevated
324 pCO₂ had no effect on any of the three bromocarbons. The temporal dynamics of CH₃I, combined
325 with strong correlations with *P. tricornutum* and *T. weissflogii*, and Chl *a*, indicate that *P.*
326 *tricornutum* and *T. weissflogii* play a critical role controlling CH₃I concentrations. In addition,
327 the production of CH₃I was sensitive to pCO₂, with a significant increase in CH₃I concentration at
328 higher pCO₂. However, without additional empirical measurements, it is unclear whether this
329 decrease was caused by bacterial methyl transferase enzyme activity or by photochemical
330 degradation at higher pCO₂.

331 Author contribution: Gui-Peng Yang and Kun-Shan Gao designed the experiments. Sheng-Hui
332 Zhang, Juan Yu and Qiong-Yao Ding carried out the experiments and prepared the manuscript.
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Figure captions

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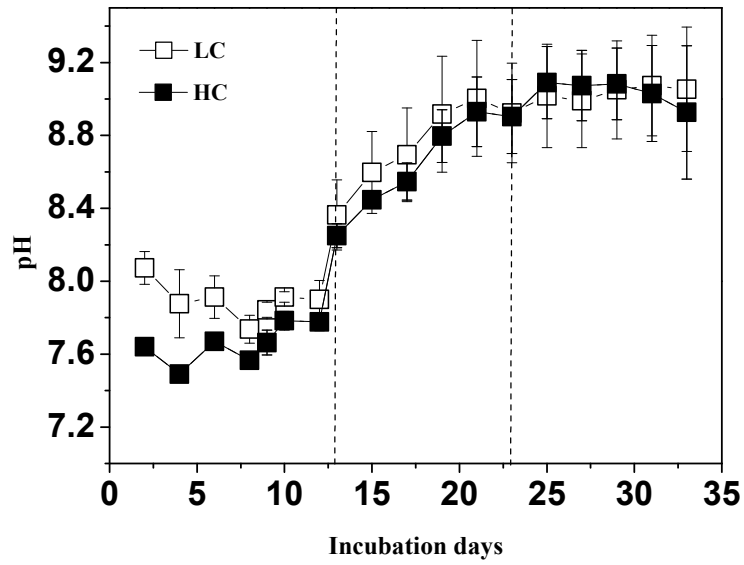
486 Fig. 1 Temporal **development** of pH in the high $p\text{CO}_2$ (1,000 μatm , solid squares) and low $p\text{CO}_2$
487 (400 μatm , white squares) mesocosms. Data are mean \pm standard deviation, $n = 3$ (triplicate
488 independent mesocosm bags) (Origin 8.0).

489 Fig. 2 Temporal **development** in dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP) and
490 DMSP-consuming bacteria concentrations in the high $p\text{CO}_2$ (1,000 μatm , black squares) and low
491 $p\text{CO}_2$ (400 μatm , white squares) mesocosms. Data are mean \pm standard deviation, $n = 3$ (triplicate
492 independent mesocosm bags).

493 Fig. 3 Temporal **development** in **bromodichloromethane (CHBrCl_2)**, **methyl bromide (CH_3Br)**,
494 **dibromomethane (CH_2Br_2)**, **iodomethane (CH_3I)** concentrations in the high $p\text{CO}_2$ (1,000 μatm ,
495 black squares) and low $p\text{CO}_2$ (400 μatm , white squares) mesocosms. Data are mean \pm standard
496 deviation, $n = 3$ (triplicate independent mesocosm bags).

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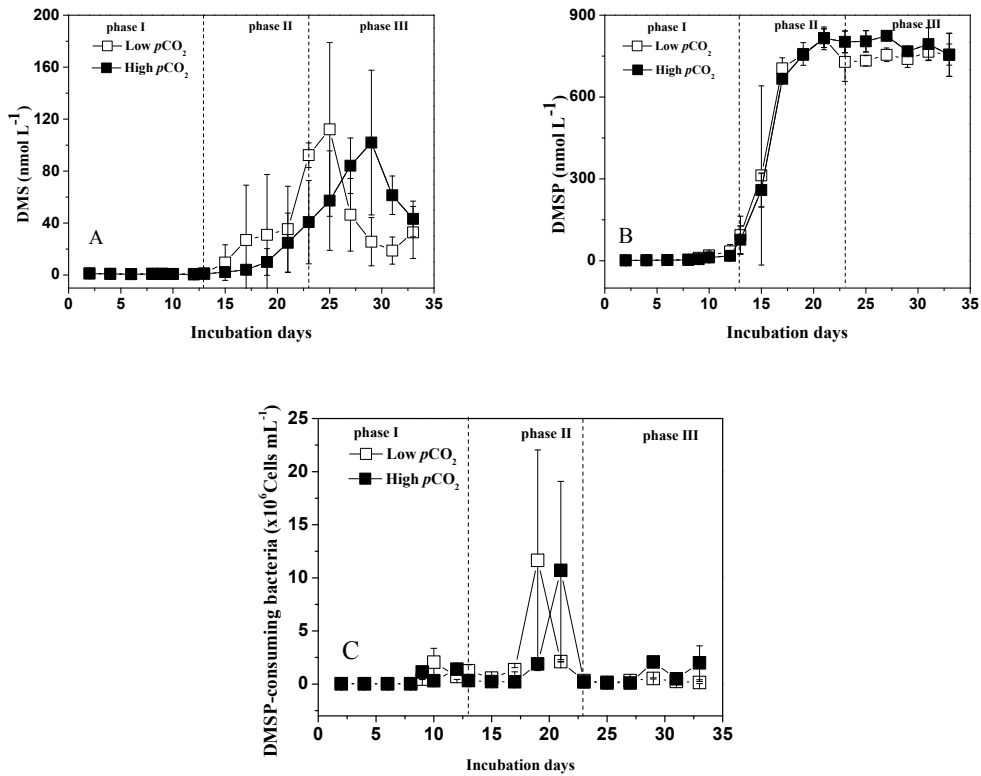
500 **Fig. 1.** Temporal changes of pH in the high $p\text{CO}_2$ (1,000 μatm , solid squares) and low $p\text{CO}_2$ (400 μatm , white

501 squares) mesocosms. Data are mean \pm standard deviation, $n = 3$ (triplicate independent mesocosm bags) (Origin

502 8.0).

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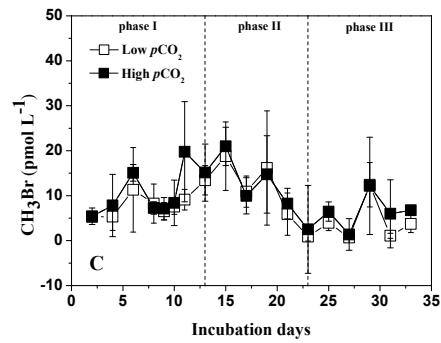
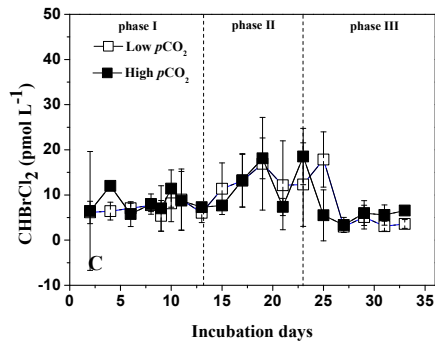
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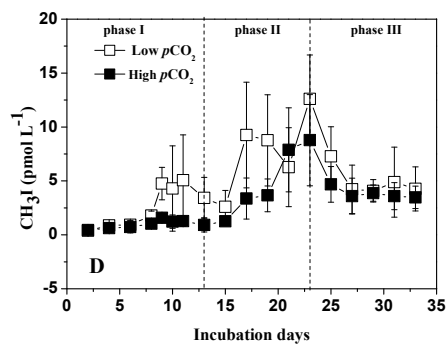
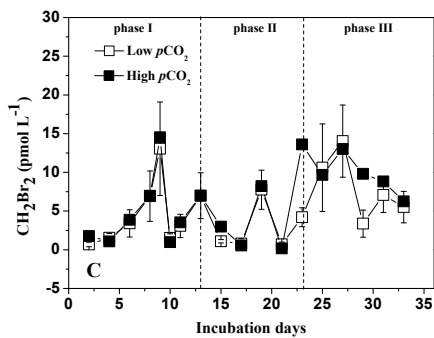
507 **Fig. 2** Temporal changes in **dimethylsulfide (DMS)**, **dimethylsulfoniopropionate (DMSP)**, DMSP-consuming
508 bacteria concentrations in the high $p\text{CO}_2$ (1,000 μatm , black squares) and low $p\text{CO}_2$ (400 μatm , white squares)
509 mesocosms. Data are mean \pm standard deviation, $n = 3$ (triplicate independent mesocosm bags) (Origin 8.0).

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512



513

514 **Fig. 3** Temporal changes in bromodichloromethane (CHBrCl_2), methyl bromide (CH_3Br), dibromomethane
515 (CH_2Br_2), iodomethane (CH_3I) concentrations in the high $p\text{CO}_2$ (1,000 μatm , black squares) and low $p\text{CO}_2$ (400
516 μatm , white squares) mesocosms. Data are mean \pm standard deviation, $n = 3$ (triplicate independent mesocosm
517 bags) (Origin 8.0).

Table 1. Dissolved inorganic carbon (DIC), pH, $p\text{CO}_2$ and nutrient concentrations in the mesocosm experiments. “-” means that the values were below the detection limit.

		pH	DIC ($\mu\text{mol kg}^{-1}$)	$p\text{CO}_2$ (μatm)	$\text{NO}_3^- + \text{NO}_2^-$ ($\mu\text{mol L}^{-1}$)	NH_4^+ ($\mu\text{mol L}^{-1}$)	PO_4^{3-} ($\mu\text{mol L}^{-1}$)	SiO_3^{2-} ($\mu\text{mol L}^{-1}$)
day 0	Low $p\text{CO}_2$	8.0±0.1	2181±29	1170~1284	52~56	19~23	2.6±0.2	38~40
	High $p\text{CO}_2$	7.5±0.1	2333±34	340~413	51~55	19~23	2.5±0.2	38~39
Phase I	Low $p\text{CO}_2$	7.9~8.4	1825~2178	373~888	15~52	1.6~20	0.5~2.6	31~38
	High $p\text{CO}_2$	7.4~8.2	2029~2338	1295~1396	47~54	0.2~21	0.7~2.5	34~39
Phase II	Low $p\text{CO}_2$	8.4~8.5	1706~1745	46~749	~ 15.9	-	0.1~0.5	10~24
	High $p\text{CO}_2$	8.4~8.6	1740~1891	59~1164	1.1~25	-	~0.1	29~30
Phase III	Low $p\text{CO}_2$	8.5~8.8	1673~1706	30~43	-	-	-	10~16
	High $p\text{CO}_2$	8.6~8.7	1616~1740	34~110	-	-	~0.3	24~25

520 **Table 2.** Correlation between dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), chlorophyll a (Chl *a*), bromodichloromethane (CHBrCl₂), methyl bromide (CH₃Br),
 521 dibromomethane (CH₂Br₂), iodomethane (CH₃I), DMSP-consuming bacteria, *Thalassiosira weissflogii* (*T. weissflogii*) and *Phaeodactylum tricornutum* (*P. tricornutum*) concentrations in the
 522 low pCO₂ treatments.

	DMS	DMSP	Chl <i>a</i>	CHBrCl ₂	CH ₃ Br	CH ₂ Br ₂	CH ₃ I	DMSP-consuming bacteria	<i>T. weissflogii</i>	<i>P. tricornutum</i>
DMS	1									
DMSP	0.701**	1								
Chl <i>a</i>	0.597**	0.792**	1							
CHBrCl ₂	0.526	0.280	0.559	1						
CH ₃ Br	-0.413	-0.230	0.196	0.313	1					
CH ₂ Br ₂	0.310	0.180	0.001	-0.136	-0.308	1				
CH ₃ I	0.694**	0.654**	0.717**	0.596*	-0.151	0.129	1			
DMSP-consuming bacteria	0.643**	0.520*	0.522*	0.394	-0.268	-0.038	0.762**	1		
<i>T. weissflogii</i>	0.410	0.617**	0.899**	0.301	0.322	0.028	0.680**	0.399	1	
<i>P. tricornutum</i>	0.560*	0.961**	0.821**	0.528	-0.032	0.162	0.588**	0.334	0.685**	1

523 *. Correlation is significant at the 0.05 level (2-tailed).

524 **. Correlation is significant at the 0.01 level (2-tailed).

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528

529 **Table 3.** Correlation between dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), chlorophyll a (Chl *a*), bromodichloromethane (CHBrCl₂), methyl bromide (CH₃Br),
 530 dibromomethane (CH₂Br₂), iodomethane (CH₃I), DMSP-consuming bacteria, *Thalassiosira weissflogii* (*T. weissflogii*) and *Phaeodactylum tricorutum* (*P. tricorutum*) concentrations in the
 531 high *p*CO₂ treatments.

	DMS	DMSP	Chl <i>a</i>	CHBrCl ₂	CH ₃ Br	CH ₂ Br ₂	CH ₃ I	DMSP-consuming bacteria	<i>T. weissflogii</i>	<i>P. tricorutum</i>
DMS	1									
DMSP	0.752**	1								
Chl <i>a</i>	0.318*	0.738**	1							
CHBrCl ₂	0.324	0.094	0.326	1						
CH ₃ Br	-0.410	-0.349	0.065	0.076	1					
CH ₂ Br ₂	0.540*	0.352	0.142	0.233	-0.377	1				
CH ₃ I	0.694**	0.816**	0.741**	0.690*	-0.407	0.316	1			
DMSP-consuming bacteria	0.544*	0.522	0.549*	0.532	-0.311	0.368	0.851*	1		
<i>T. weissflogii</i>	0.355	0.743**	0.930**	0.304	0.076	0.233	0.690**	0.567	1	
<i>P. tricorutum</i>	0.635**	0.954**	0.803**	0.143	-0.257	0.267	0.834**	0.559	0.820**	1

532 *. Correlation is significant at the 0.05 level (2-tailed).

533 **. Correlation is significant at the 0.01 level (2-tailed).