

1 **Effect of elevated  $p\text{CO}_2$  on trace gas production during an**  
2 **ocean acidification mesocosm experiment**

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27

28 **Abstract**

29 A mesocosm experiment was conducted in Wuyuan Bay (Xiamen), China to investigate the effects of elevated  
30  $p\text{CO}_2$  on the phytoplankton species *Phaeodactylum tricornutum* (*P. tricornutum*), *Thalassiosira weissflogii* (*T.*  
31 *weissflogii*) and *Emiliania huxleyi* (*E. huxleyi*) and their production ability of dimethylsulfide (DMS),  
32 dimethylsulfoniopropionate (DMSP), as well as four halocarbon compounds bromodichloromethane ( $\text{CHBrCl}_2$ ),  
33 methyl bromide ( $\text{CH}_3\text{Br}$ ), dibromomethane ( $\text{CH}_2\text{Br}_2$ ) and iodomethane ( $\text{CH}_3\text{I}$ ). Over a period of 5 weeks, *P.*  
34 *tricornutum* outcompeted *T. weissflogii* and *E. huxleyi*, comprising more than 99% of the final biomass. During  
35 the logarithmic growth phase (phase I), mean DMS concentration in high  $p\text{CO}_2$  mesocosms (1000  $\mu\text{atm}$ ) was 28%  
36 lower than that in low  $p\text{CO}_2$  mesocosms (400  $\mu\text{atm}$ ). Elevated  $p\text{CO}_2$  led to a delay in DMSP-consuming bacteria  
37 concentrations attached to *T. weissflogii* and *P. tricornutum* and finally resulted in the delay of DMS concentration  
38 in the high  $p\text{CO}_2$  treatment. Unlike DMS, the elevated  $p\text{CO}_2$  did not affect DMSP production ability of *T.*  
39 *weissflogii* or *P. tricornutum* throughout the 5 weeks culture. A positive relationship was detected between  $\text{CH}_3\text{I}$   
40 and *T. weissflogii* and *P. tricornutum* during the experiment, and there was a 40% reduction in mean  $\text{CH}_3\text{I}$   
41 concentration in the high  $p\text{CO}_2$  mesocosms.  $\text{CHBrCl}_2$ ,  $\text{CH}_3\text{Br}$ , and  $\text{CH}_2\text{Br}_2$  concentrations did not increase with  
42 elevated chlorophyll *a* (Chl *a*) concentrations compared with DMS(P) and  $\text{CH}_3\text{I}$ , and there were no major peaks  
43 both in the high  $p\text{CO}_2$  or low  $p\text{CO}_2$  mesocosms. In addition, no effect of elevated  $p\text{CO}_2$  was identified for any of  
44 the three bromocarbons.

45 **Keywords:** ocean acidification, dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), halocarbons,  
46 phytoplankton, bacteria

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## 48 **1. Introduction**

49 Anthropogenic emissions have increased the fugacity of atmospheric carbon dioxide ( $p\text{CO}_2$ ) from  
50 the pre-industrial value of 280  $\mu\text{atm}$  to the present-day value of over 400  $\mu\text{atm}$ , and these values  
51 will further increase to 800–1000  $\mu\text{atm}$  by the end of this century (Gattuso et al., 2015). The  
52 dissolution of this excess  $\text{CO}_2$  into the surface of the ocean directly affects the carbonate system  
53 and has lowered the pH by 0.1 units, from 8.21 to 8.10 over the last 250 years. Further decreases  
54 of 0.3–0.4 pH units are predicted by the end of this century (Doney et al., 2009; Orr et al., 2005;  
55 Gattuso et al., 2015), which is commonly referred to as ocean acidification. The physiological and  
56 ecological aspects of the phytoplankton response to this changing environment can potentially  
57 alter marine phytoplankton community composition, community biomass, and feedback to  
58 biogeochemical cycles (Boyd and Doney, 2002). These changes simultaneously have an impact on  
59 some volatile organic compounds produced by marine phytoplankton (Liss et al., 2014; Liu et al.,  
60 2017), including the climatically important trace gas dimethylsulfide (DMS) and a number of  
61 volatile halocarbon compounds.

62 DMS is the most important volatile sulfur compound produced from  
63 dimethylsulfoniopropionate (DMSP), which is ubiquitous in marine environments, mainly  
64 synthesized by marine microalgae (Stefels et al., 2007), a few angiosperms, some corals (Raina et  
65 al., 2016), and several heterotrophic bacteria (Curson et al., 2017) through complex biological  
66 interactions in marine ecosystems. Although it remains controversial, DMS and its by-products,  
67 such as methanesulfonic acid and non-sea-salt sulfate, are suspected to have a prominent part in  
68 climate feedback (Charlson et al., 1987; Quinn and Bates, 2011). The conversion of DMSP to  
69 DMS is facilitated by several enzymes, including DMSP-lyase and acyl CoA transferase

70 (Kirkwood et al., 2010; Todd et al., 2007); these enzymes are mainly found in phytoplankton,  
71 macroalgae, symbiodinium, bacteria and fungi (de Souza and Yoch, 1995; Stefels and Dijkhuizen,  
72 1996; Steinke and Kirst, 1996; Bacic and Yoch, 1998; Yost and Mitchelmore, 2009). Several  
73 studies have shown a negative impact of decreasing pH on DMS-production capability (Hopkins  
74 et al., 2010; Avgoustidi et al., 2012; Archer et al., 2013; Webb et al., 2016), while others have  
75 found either no effect or a positive effect (Vogt et al., 2008; Hopkins and Archer, 2014). Several  
76 assumptions have been presented to explain these contrasting results and attributed the  
77 pH-induced variation in DMS-production capability to altered physiology of the algae cells or of  
78 bacterial DMSP degradation (Vogt et al., 2008; Hopkins et al., 2010, Avgoustidi et al., 2012;  
79 Archer et al., 2013; Hopkins and Archer, 2014; Webb et al., 2015).

80 Halocarbons also play a significant role in the global climate because they are linked to  
81 tropospheric and stratospheric ozone depletion and a synergistic effect of chlorine and bromine  
82 species has been reported accounting for approximately 20% of the polar stratospheric ozone  
83 depletion (Roy et al., 2011). In addition, iodocarbons can release atomic iodine quickly through  
84 photolysis in the atmospheric boundary layer and iodine atoms are very efficient in the catalytic  
85 removal of O<sub>3</sub>, which governs the lifetime of many climate relevant gases including methane and  
86 DMS (Jenkins et al., 1991). Compared with DMS, limited attention was received about the effect  
87 of ocean acidification on halocarbon concentrations. Hopkins et al. (2010) and Webb et al. (2015)  
88 measured lower concentrations of several iodocarbons, while bromocarbons were unaffected by  
89 elevated *p*CO<sub>2</sub> in two acidification experiments. In addition, another mesocosm study did not elicit  
90 significant differences from any halocarbon compounds at up to 1,400  $\mu$ atm *p*CO<sub>2</sub> (Hopkins et al.,  
91 2013).

92 Taken together, the data indicate that the response of DMS and halocarbon release to elevated  
93  $p\text{CO}_2$  is complex and controversial. DMS and halocarbons play a significant role in the global  
94 climate and will perhaps act to a greater extent in the future. An intermediate step between  
95 laboratory and natural community field experiments was designed in this study to understand the  
96 response of the release of DMS and halocarbon to ocean acidification in Chinese coastal seas  
97 using isolates of non-axenic phytoplankton added to filtered natural water. We hypothesized that  
98 the response of DMS and halocarbon release to elevated  $p\text{CO}_2$  in natural seawater can be better  
99 presented after minimizing the shifting composition of the natural phytoplankton and microbial  
100 communities.

## 101 **2. Experimental method**

### 102 *2.1 Experimental setup*

103 To investigate the response of DMS and halocarbon release to ocean acidification, a mesocosm  
104 experiment was carried out on a floating platform (set in seawater, about 150 m from the shore) at  
105 the Facility for Ocean Acidification Impacts Study of Xiamen University (FOANIC-XMU,  
106 24.52 N, 117.18 E) (for full technical details of the mesocosms, see Liu et al. 2017). Six  
107 cylindrical transparent thermoplastic polyurethane bags with domes were deployed along the  
108 south side of the platform. The width and depth of each mesocosm bag was 1.5 m and 3 m,  
109 respectively. Filtered (0.01  $\mu\text{m}$  ultrafiltration water purifier, MU801-4T, Midea, Guangdong,  
110 China) in situ seawater was pumped into the six bags simultaneously within 24 hrs. A known  
111 amount of NaCl solution was added to each bag to calculate the exact volume of seawater in the  
112 bags, according to a comparison of the salinity before and after adding salt (Czerny et al., 2013).  
113 The initial in situ  $p\text{CO}_2$  was about 650  $\mu\text{atm}$ . To set the low (400  $\mu\text{atm}$ ) and high  $p\text{CO}_2$  (1000  $\mu\text{atm}$ )

114 levels, we added Na<sub>2</sub>CO<sub>3</sub> solution and CO<sub>2</sub> saturated seawater to the mesocosm bags to alter total  
115 alkalinity and dissolved inorganic carbon (Gattuso et al., 2010; Riebesell et al., 2013).  
116 Subsequently, during the whole experimental process, air at the ambient (400 µatm) and elevated  
117 pCO<sub>2</sub> (1000 µatm) concentrations was continuously bubbled into the mesocosm bags using a CO<sub>2</sub>  
118 Enricher (CE-100B, Wuhan Ruihua Instrument & Equipment Ltd., Wuhan, China). Seawater taken  
119 from the coastal environment was first filtered to remove algae and their attached bacteria before  
120 usage in mesocosm bags. Bacterial abundance in the pre-filtered water was less than 10<sup>3</sup> cell mL<sup>-1</sup>,  
121 which was three magnitudes lower than the bacterial abundance in the natural water and close to  
122 the detection limit of the flow cytometer. The trace gases, including DMS, bromodichloromethane  
123 (CHBrCl<sub>2</sub>), methyl bromide (CH<sub>3</sub>Br), dibromomethane (CH<sub>2</sub>Br<sub>2</sub>), and iodomethane (CH<sub>3</sub>I)  
124 produced in the environment did not affect the mesocosm trace gas concentrations after the bags  
125 were sealed.

## 126 2.2 Algal strains

127 Before being introduced into the mesocosms, the three phytoplankton species *Phaeodactylum*  
128 *tricornutum* (*P. tricornutum*), *Thalassiosira weissflogii* (*T. weissflogii*) and *Emiliana huxleyi* (*E.*  
129 *huxleyi*) were cultured in autoclaved, pre-filtered seawater from Wuyuan Bay at 16 °C (similar to  
130 the in situ temperature of Wuyuan Bay) without any addition of nutrients. Cultures were  
131 continuously aerated with filtered ambient air containing 400 µatm of CO<sub>2</sub> within plant chambers  
132 (HP1000G-D, Wuhan Ruihua Instrument & Equipment, China) at a constant bubbling rate of 300  
133 mL min<sup>-1</sup>. The culture medium was renewed every 24 hrs to maintain the cells of each  
134 phytoplankton species in exponential growth. When the experiment began, these three  
135 phytoplankton species were inoculated into the mesocosm bags, with an initial

136 diatom/coccolithophorid cell ratio of 1:1. The initial concentrations of *P. tricornutum*, *T.*  
137 *weissflogii*, and *E. huxleyi* inoculated into the mesocosm were 10, 10, and 20 cells mL<sup>-1</sup>,  
138 respectively. *P. tricornutum* and *T. weissflogii* were obtained from the Center for Collections of  
139 Marine Bacteria and Phytoplankton of the State Key Laboratory of Marine Environmental Science  
140 (Xiamen University). *P. tricornutum* was originally isolated from the South China Sea in 2004  
141 and *T. weissflogii* was isolated from Daya Bay in the coastal South China Sea. *E. huxleyi* was  
142 originally isolated in 1992 from the field station of the University of Bergen (Raunefjorden;  
143 60°18'N, 05°15'E).

#### 144 2.3 Sampling for DMS(P) and halocarbons

145 DMS(P) and halocarbon samples were taken from the above mentioned mesocosm bags at 9 a.m.,  
146 then all collected samples were transported into a dark cool box back to the laboratory onshore for  
147 analysis within 1 hr. For the DMS analysis, 2 mL sample was gently filtered through a 25 mm  
148 GF/F (glass fiber) filter and transferred to a purge and trap system linked to a Shimadzu GC-2014  
149 gas chromatograph (Tokyo, Japan) equipped with a glass column packed with 10% DEGS on  
150 Chromosorb W-AW-DMCS (3 m × 3 mm) and a flame photometric detector (Zhang et al., 2014).  
151 For total DMSP analysis, 10 mL water sample was fixed using 50 µL of 50 % H<sub>2</sub>SO<sub>4</sub> and sealed  
152 (Kiene and Slezak, 2006). After > 1 d preservation, DMSP samples were hydrolyzed for 24 hrs  
153 with a pellet of KOH (final pH > 13) to fully convert DMSP to DMS. Then, 2 mL of the  
154 hydrolyzed sample was carefully transferred to the purge and trap system mentioned above for  
155 extraction of DMS. For halocarbons, 100 mL sample was purged at 40 °C with pure nitrogen at a  
156 flow rate of 100 mL min<sup>-1</sup> for 12 min using another purge and trap system coupled to an Agilent  
157 6890 gas chromatograph (Agilent Technologies, Palo Alto, CA, USA) equipped with an electron

158 capture detector (ECD) as well as a 60 m DB-624 capillary column (0.53 mm ID; film thickness, 3  
159  $\mu\text{m}$ ) (Yang et al., 2010). The analytical precision for duplicate measurements of DMS(P) and  
160 halocarbons was  $> 10\%$ .

#### 161 *2.4 Measurements of chlorophyll a*

162 Chlorophyll *a* (Chl *a*) was measured in water samples (200–1,000 mL) collected every 2 d at 9  
163 a.m. by filtering onto Whatman GF/F filters (25 mm). The filters were placed in 5 mL 100%  
164 methanol overnight at 4 °C and centrifuged at 5000  $\text{r min}^{-1}$  for 10 min. The absorbance of the  
165 supernatant (2.5 mL) was measured from 250 to 800 nm using a scanning spectrophotometer (DU  
166 800, Beckman Coulter Inc., Brea, CA, USA). Chl *a* concentration was calculated according to the  
167 equation reported by Porra (2002).

#### 168 *2.5 Enumeration of DMSP-consuming bacteria*

169 The number of DMSP-consuming bacteria in the mesocosms was estimated using the most  
170 probable number methodology. The medium consisted of a mixture (1:1 v/v) of sterile artificial  
171 sea water and mineral medium (Visscher et al., 1991), 3 mL of which was dispensed into 6 mL test  
172 tubes, which were closed by an over-sized cap, allowing gas exchange. Triplicate dilution series  
173 were set up. All test tubes contained 1  $\text{mmol L}^{-1}$  DMSP as the sole organic carbon source and  
174 were kept at 30 °C in the dark. After 2 weeks, the presence/absence of bacteria in the tubes was  
175 verified by DAPI staining (Porter and Feig, 1980). Three tubes containing 3 mL ASW without  
176 substrate were used as controls.

#### 177 *2.6 Statistical analysis*

178 One-way analysis of variance (ANOVA), Tukey's test, and the two-sample t-test were carried out  
179 to demonstrate the differences between treatments. A p-value  $< 0.05$  was considered significant.



180 Relationships between DMS(P), halocarbons and a range of other parameters were detected using  
181 Pearson's correlation analysis via SPSS 22.0 for Windows (SPSS Inc., Chicago, IL, USA).

### 182 **3. Results and Discussion**

#### 183 *3.1 Temporal changes in pH, Chl a, P. tricornutum, T. weissflogii, and E. huxleyi during the* 184 *experiment*

185 During the experiment, the seawater in each mesocosm was well mixed, and the temperature and  
186 salinity remained stable, with means of 16 °C and 29, respectively, in all mesocosm bags. We  
187 observed significant differences in pH levels between the two CO<sub>2</sub> treatments on days 0–11, but  
188 the differences disappeared with subsequent phytoplankton growth (Fig. 1). The phytoplankton  
189 growth process was divided into three phases in terms of variations in Chl *a* concentrations in the  
190 mesocosm experiments as described in Liu et al. (2017): i) the logarithmic growth phase (phase I,  
191 days 0–13), ii) a plateau phase (phase II, days 13–23, bloom period), and iii) a secondary plateau  
192 phase (phase III, days 23–33) attained after a decline in biomass from a maximum in phase II. The  
193 initial chemical parameters of the mesocosm experiment are shown in Table 1. The initial mean  
194 dissolved nitrate (including NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup>), NH<sub>4</sub><sup>+</sup>, PO<sub>4</sub><sup>3-</sup> and silicate (SiO<sub>3</sub><sup>2-</sup>) concentrations  
195 were 54, 20, 2.6 and 41 μmol L<sup>-1</sup>, respectively for the low pCO<sub>2</sub> treatment and 52, 21, 2.4 and 38  
196 μmol L<sup>-1</sup>, respectively for the high pCO<sub>2</sub> treatment. The nutrient concentrations (NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>,  
197 NH<sub>4</sub><sup>+</sup> and phosphate) during phase I were consumed rapidly and their concentrations were below  
198 or close to the detection limit during phase II (Table 1). SiO<sub>3</sub><sup>2-</sup> was detectable during the entire  
199 experimental period, and was unlikely to be a limiting factor for phytoplankton growth during the  
200 experiment. In addition, although dissolved inorganic nitrogen (NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup>, and NO<sub>2</sub><sup>-</sup>) and  
201 phosphate were depleted, Chl *a* concentration in both treatments (biomass dominated by *P.*

202 *tricornutum*) remained constant over days 12–22, and then declined over subsequent days. *T.*  
203 *weissflogii* was found throughout the entire period in each bag, but the maximum concentration  
204 was 8,120 cells mL<sup>-1</sup>, which was far less than the concentration of *P. tricornutum* with a  
205 maximum density of about 1.5 million cells mL<sup>-1</sup> (Liu et al., 2017). It is possible that *P.*  
206 *tricornutum* outcompeted *T. weissflogii* because of its higher surface to volume ratio and/or  
207 species-specific physiology, which would enhance the efficiency of nutrient uptake and related  
208 metabolism (Alessandrade et al., 2007). *E. huxleyi* was only found in phase I and its maximal  
209 concentration reached 310 cells mL<sup>-1</sup> according to the results of Liu et al. (2017). Previous studies  
210 have reported that the maximum specific growth rate of *T. weissflogii* and *P. tricornutum* is about  
211 1.2 d<sup>-1</sup> (Li et al., 2014; Sugie and Yoshimura, 2016), while that of *E. huxleyi* is about 0.8 d<sup>-1</sup> (Xing  
212 et al., 2015). This might be the main reason why diatoms overwhelmingly outcompeted the  
213 coccolithophores during this experiment.

### 214 3.2 Impact of elevated pCO<sub>2</sub> on DMS and DMSP production

215 DMSP concentrations in the high pCO<sub>2</sub> and low pCO<sub>2</sub> treatments increased significantly along  
216 with the increase in Chl *a* concentrations and algal cells, and remained relatively constant over the  
217 following days. A significant positive relationship was observed between DMSP and  
218 phytoplankton in the experiment ( $r = 0.961$ ,  $p < 0.01$  for *P. tricornutum*,  $r = 0.617$ ,  $p < 0.01$  for *T.*  
219 *weissflogii* in the low pCO<sub>2</sub> treatment, Table 2;  $r = 0.954$ ,  $p < 0.01$  for *P. tricornutum*,  $r = 0.743$ ,  $p$   
220  $< 0.01$  for *T. weissflogii* in the high pCO<sub>2</sub> treatment, Table 3). DMS was maintained at a low level  
221 during phase I (mean of 1.03 nmol L<sup>-1</sup> in the low pCO<sub>2</sub> and 0.74 nmol L<sup>-1</sup> in the high pCO<sub>2</sub>  
222 treatments, respectively) compared with DMSP. DMS concentrations began to increase rapidly on  
223 day 15, peaked on day 25 in the low pCO<sub>2</sub> treatment (112.1 nmol L<sup>-1</sup>) and on day 29 in the high

224  $p\text{CO}_2$  treatment ( $101.9 \text{ nmol L}^{-1}$ ) respectively, and then decreased in the following days. A  
225 moderate positive relationship was observed between DMS and *P. tricornutum* ( $r = 0.560$ ,  $p <$   
226  $0.05$  in the low  $p\text{CO}_2$  treatment;  $r = 0.635$ ,  $p < 0.01$  in the high  $p\text{CO}_2$  treatment), while no  
227 relationship was observed between DMS and *T. weissflogii* (Table 2 and Table 3) during the  
228 experiment. Similar to DMS, DMSP-consuming bacteria also maintained a low level during phase  
229 I (mean of  $0.57 \times 10^6$  and  $0.40 \times 10^6$  cells  $\text{mL}^{-1}$  in the low  $p\text{CO}_2$  and high  $p\text{CO}_2$  treatments,  
230 respectively). DMSP-consuming bacterial concentrations respectively peaked on days 19 ( $11.65 \times$   
231  $10^6$  cells  $\text{mL}^{-1}$ ) and 21 ( $10.70 \times 10^6$  cells  $\text{mL}^{-1}$ ) in the low  $p\text{CO}_2$  and high  $p\text{CO}_2$  treatments.

232 In this study, no difference in mean DMSP concentrations was observed between the two  
233 treatments, indicating that elevated  $p\text{CO}_2$  had no significant influence on DMSP production in *P.*  
234 *tricornutum* and *T. weissflogii*. However, significant reductions in mean DMS concentration  
235 (28%) and DMSP-consuming bacteria (29%) were detected during phase I in the high  $p\text{CO}_2$   
236 treatment compared with those in the low  $p\text{CO}_2$  treatment, indicating that elevated  $p\text{CO}_2$  inhibited  
237 DMSP-consuming bacteria and DMS production during the logarithmic growth phase. In addition,  
238 the peak DMS concentration in the high  $p\text{CO}_2$  treatment was delayed 4 days relative to that in the  
239 low  $p\text{CO}_2$  treatment during phase II (Fig. 2-A). This result has been observed in previous  
240 mesocosm experiments and it was attributed to small scale shifts in community composition and  
241 succession (Vogt et al., 2008; Webb et al., 2016). However, this phenomenon during the present  
242 study can be explained in another straightforward way. Previous studies have shown that marine  
243 bacteria play a key role in DMS production, and that the efficiency of bacteria converting DMSP  
244 to DMS may vary from 2 to 100% depending on the nutrient status of the bacteria and the quantity  
245 of dissolved organic matter (Simó et al., 2002, 2009; Kiene et al., 1999, 2000). In addition, a

246 significant positive relationship was observed between DMS and DMSP-consuming bacteria ( $r =$   
247  $0.643$ ,  $p < 0.01$  in the low  $p\text{CO}_2$  treatment;  $r = 0.544$ ,  $p < 0.01$  in the high  $p\text{CO}_2$  treatment) during  
248 this experiment. All of these observations point to the importance of bacteria in DMS and DMSP  
249 dynamics. During the present mesocosm experiment, DMSP concentrations in the low  $p\text{CO}_2$   
250 treatment decreased slightly on day 23, while the slight decrease appeared on day 29 in the high  
251  $p\text{CO}_2$  treatment (Fig. 2-B). In addition, the time that the DMSP concentration began to decrease  
252 was very close to the time when the highest DMS concentration occurred in both treatments.  
253 Similar to DMS, DMSP-consuming bacteria was also delayed in the high  $p\text{CO}_2$  mesocosm  
254 compared to that in the low  $p\text{CO}_2$  mesocosm (Fig. 2-C). Taken together, we inferred that the  
255 elevated  $p\text{CO}_2$  first delayed growth of DMSP-consuming bacteria, then the delayed  
256 DMSP-consuming bacteria postponed the DMSP degradation process, and eventually delayed the  
257 DMS concentration in the high  $p\text{CO}_2$  treatment. In addition, considering that algae and bacteria in  
258 natural seawater were removed through a filtering process before the experiment (Huang et al.,  
259 2018), we further concluded that the elevated  $p\text{CO}_2$  controlled DMS concentrations mainly by  
260 affecting DMSP-consuming bacteria attached to *T. weissflogii* and *P. tricornutum*.

### 261 3.3 Impact of elevated $p\text{CO}_2$ on halocarbon compounds

262 The temporal development in  $\text{CHBrCl}_2$ ,  $\text{CH}_3\text{Br}$ , and  $\text{CH}_2\text{Br}_2$  concentrations is shown in Fig. 3-A,  
263 Fig. 3-B, and Fig. 3-C, respectively. The temporal changes of their concentrations were  
264 substantially different from those of DMS, DMSP, *P. tricornutum* and *T. weissflogii*. The mean  
265 concentrations of  $\text{CHBrCl}_2$ ,  $\text{CH}_3\text{Br}$ , and  $\text{CH}_2\text{Br}_2$  for the entire experiment were 8.58, 7.85, and  
266  $5.13 \text{ pmol L}^{-1}$  in the low  $p\text{CO}_2$  treatment and 8.81, 9.73, and  $6.27 \text{ pmol L}^{-1}$  in the high  $p\text{CO}_2$   
267 treatment. The concentrations of  $\text{CHBrCl}_2$ ,  $\text{CH}_3\text{Br}$ , and  $\text{CH}_2\text{Br}_2$  did not increase with the Chl *a*

268 concentration compared with those of DMS and DMSP, and no major peaks were detected in the  
269 mesocosms. In addition, no effect of elevated  $p\text{CO}_2$  was identified for any of the three  
270 bromocarbons, which compared well with previous mesocosm findings (Hopkins et al., 2010,  
271 2013; Webb et al., 2016). No clear correlation was observed between the three bromocarbons and  
272 any of the measured algal groups (table 2 and table 3), indicating that *P. tricornutum* and *T.*  
273 *weissflogii* did not primarily release these three bromocarbons during the mesocosm experiment.  
274 Previous studies reported that large-size cyanobacteria, such as *Aphanizomenon flos-aquae*, could  
275 produce bromocarbons (Karlsson et al., 2008). Significant correlations between the abundance of  
276 cyanobacteria and several bromocarbons have been reported in the Arabian Sea (Roy et al., 2011).  
277 However, the filtration procedure led to the loss of cyanobacteria in the mesocosms and finally  
278 resulted in low bromocarbon concentrations during the experiment, although *P. tricornutum* and  
279 *T. weissflogii* abundances were high.

280 The temporal dynamics of  $\text{CH}_3\text{I}$  in the high  $p\text{CO}_2$  and low  $p\text{CO}_2$  treatments are shown in Fig. 3-D.  
281 The  $\text{CH}_3\text{I}$  concentrations in the low  $p\text{CO}_2$  treatment varied from 0.38 to 12.61  $\text{pmol L}^{-1}$ , with a  
282 mean of 4.76  $\text{pmol L}^{-1}$ . The  $\text{CH}_3\text{I}$  concentrations in the high  $p\text{CO}_2$  treatment ranged between 0.44  
283 and 8.78  $\text{pmol L}^{-1}$ , with a mean of 2.88  $\text{pmol L}^{-1}$ . The maximum  $\text{CH}_3\text{I}$  concentrations in the high  
284  $p\text{CO}_2$  and low  $p\text{CO}_2$  treatments were both observed on day 23. The range of  $\text{CH}_3\text{I}$  concentrations  
285 during this experiment was similar to that measured in the mesocosm experiment (< 1~10  $\text{pmol}$   
286  $\text{L}^{-1}$ ) in Kongsfjorden conducted by Hopkins et al. (2013). In addition, the mean  $\text{CH}_3\text{I}$   
287 concentration in the low  $p\text{CO}_2$  treatment was similar to that measured in the East China Sea, with  
288 an average of 5.34  $\text{pmol L}^{-1}$  in winter and 5.74  $\text{pmol L}^{-1}$  in summer (Yuan et al., 2016).  
289 Meanwhile, a positive relationship was detected between  $\text{CH}_3\text{I}$  and Chl *a*, *P. tricornutum* and *T.*

290 *weissflogii* ( $r = 0.588$ ,  $p < 0.01$  in the low  $p\text{CO}_2$  treatment;  $r = 0.834$ ,  $p < 0.01$  in the low  $p\text{CO}_2$   
291 treatment for *P. tricornutum*;  $r = 0.680$ ,  $p < 0.01$  in the low  $p\text{CO}_2$  treatment;  $r = 0.690$ ,  $p < 0.01$   
292 in the high  $p\text{CO}_2$  treatment for *Thalassiosira weissflogii*;  $r = 0.717$ ,  $p < 0.01$  in the low  $p\text{CO}_2$   
293 treatment;  $r = 0.741$ ,  $p < 0.01$  in the high  $p\text{CO}_2$  treatment for Chl *a*). This result agrees with  
294 previous mesocosm (Hopkins et al., 2013) and laboratory experiments (Hughes et al., 2013;  
295 Manley and De La Cuesta, 1997) identifying diatoms as significant producers of  $\text{CH}_3\text{I}$ . Moreover,  
296 similar to DMS, the maximum  $\text{CH}_3\text{I}$  concentration also occurred after the maxima of *P.*  
297 *tricornutum* and *T. weissflogii*, at about 4 d (Fig. 3-D). This result was similar to the conclusions  
298 reported by Hopkins et al. (2010) and Wingenter et al. (2007) during two mesocosm experiments  
299 conducted in Norway. Their results confirmed that iodocarbon gases generally occur after the Chl  
300 *a* maxima. Furthermore, the mean  $\text{CH}_3\text{I}$  concentration measured in the high  $p\text{CO}_2$  treatment was  
301 significantly lower (40%) than that measured in the low  $p\text{CO}_2$  treatment during the mesocosm  
302 experiment. This result is in accordance with Hopkins et al. (2010) and Webb et al. (2015) who  
303 also reported that elevated  $p\text{CO}_2$  leads to a reduction in iodocarbon concentrations, but in contrast  
304 to the findings of Hopkins et al. (2013) and Webb et al. (2016) who showed that elevated  $p\text{CO}_2$   
305 does not significantly affect the iodocarbon concentrations in the mesocosms. Considering that the  
306 phytoplankton species did not show significant differences in the high  $p\text{CO}_2$  and low  $p\text{CO}_2$   
307 treatments during the experiment, this reduction in the high  $p\text{CO}_2$  treatment was likely not caused  
308 by phytoplankton. Apart from direct biological production via methyl transferase enzyme activity  
309 by both phytoplankton and bacteria (Amachi et al., 2001),  $\text{CH}_3\text{I}$  is produced from the breakdown  
310 of higher molecular weight iodine-containing organic matter (Fenical, 1982) through  
311 photochemical reactions between organic matter and light (Richter and Wallace, 2004). Both

312 bacterial methyl transferase enzyme activity and photochemical reaction could be responsible for  
313 the reduction of CH<sub>3</sub>I concentrations in the high pCO<sub>2</sub> treatment but further experiments are  
314 needed to verify this result.

#### 315 **4. Conclusions**

316 In this study, the effects of increased levels of pCO<sub>2</sub> on marine DMS(P) and halocarbons release  
317 were studied in a controlled mesocosm facility. During the logarithmic growth phase, the elevated  
318 pCO<sub>2</sub> led to a reduction in mean DMSP-consuming bacteria (29%) and DMS concentration (28%)  
319 compared with those in the low pCO<sub>2</sub> treatment. In addition, a 4 d delay in DMS concentration  
320 was observed in the high pCO<sub>2</sub> treatment due to the effect of elevated pCO<sub>2</sub> and we attribute this  
321 delay in DMS concentration to the DMSP-consuming bacteria attached to *P. tricornutum* and *T.*  
322 *weissflogii*. Due to the loss of main bromocarbon-producing species affected by the filtration  
323 procedure, three bromocarbons compounds measured in this study were not correlated with *P.*  
324 *tricornutum* and *T. weissflogii*, and Chl *a*. Besides, elevated pCO<sub>2</sub> had no effect on any of the  
325 three bromocarbons. The temporal dynamics of CH<sub>3</sub>I, combined with strong correlations with *P.*  
326 *tricornutum* and *T. weissflogii*, and Chl *a*, indicate that *P. tricornutum* and *T. weissflogii* play a  
327 critical role controlling CH<sub>3</sub>I concentrations. In addition, the production of CH<sub>3</sub>I was sensitive to  
328 pCO<sub>2</sub>, with a significant increase in CH<sub>3</sub>I concentration at higher pCO<sub>2</sub>. However, without  
329 additional empirical measurements, it is unclear whether this decrease was caused by bacterial  
330 methyl transferase enzyme activity or by photochemical degradation at higher pCO<sub>2</sub>.

331 Author contribution: Gui-Peng Yang and Kun-Shan Gao designed the experiments. Sheng-Hui  
332 Zhang, Juan Yu and Qiong-Yao Ding carried out the experiments and prepared the manuscript.  
333 Hong-Hai Zhang and Da-Wei Pan revised the paper.

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### Figure captions

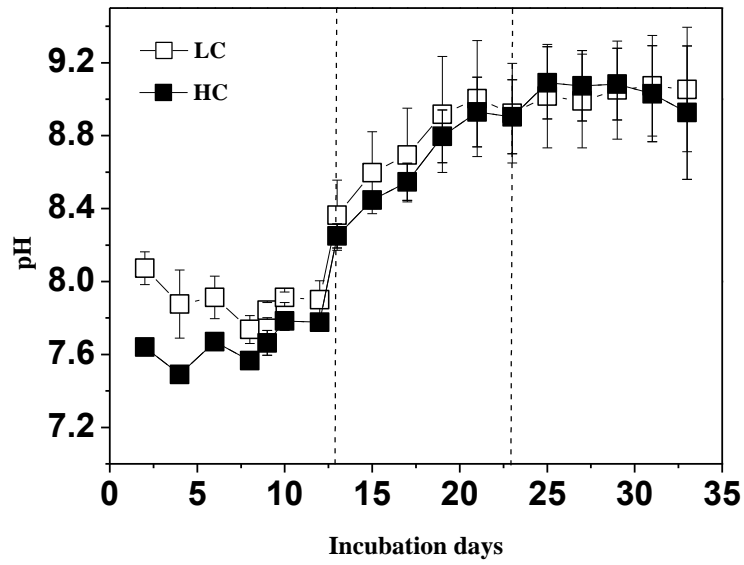
486 Fig. 1 Temporal development of pH in the high  $p\text{CO}_2$  (1,000  $\mu\text{atm}$ , solid squares) and low  $p\text{CO}_2$   
487 (400  $\mu\text{atm}$ , white squares) mesocosms. Data are mean  $\pm$  standard deviation,  $n = 3$  (triplicate  
488 independent mesocosm bags) (Origin 8.0).

489 Fig. 2 Temporal development in dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP) and  
490 DMSP-consuming bacteria concentrations in the high  $p\text{CO}_2$  (1,000  $\mu\text{atm}$ , black squares) and low  
491  $p\text{CO}_2$  (400  $\mu\text{atm}$ , white squares) mesocosms. Data are mean  $\pm$  standard deviation,  $n = 3$  (triplicate  
492 independent mesocosm bags).

493 Fig. 3 Temporal development in bromodichloromethane ( $\text{CHBrCl}_2$ ), methyl bromide ( $\text{CH}_3\text{Br}$ ),  
494 dibromomethane ( $\text{CH}_2\text{Br}_2$ ), iodomethane ( $\text{CH}_3\text{I}$ ) concentrations in the high  $p\text{CO}_2$  (1,000  $\mu\text{atm}$ ,  
495 black squares) and low  $p\text{CO}_2$  (400  $\mu\text{atm}$ , white squares) mesocosms. Data are mean  $\pm$  standard  
496 deviation,  $n = 3$  (triplicate independent mesocosm bags).

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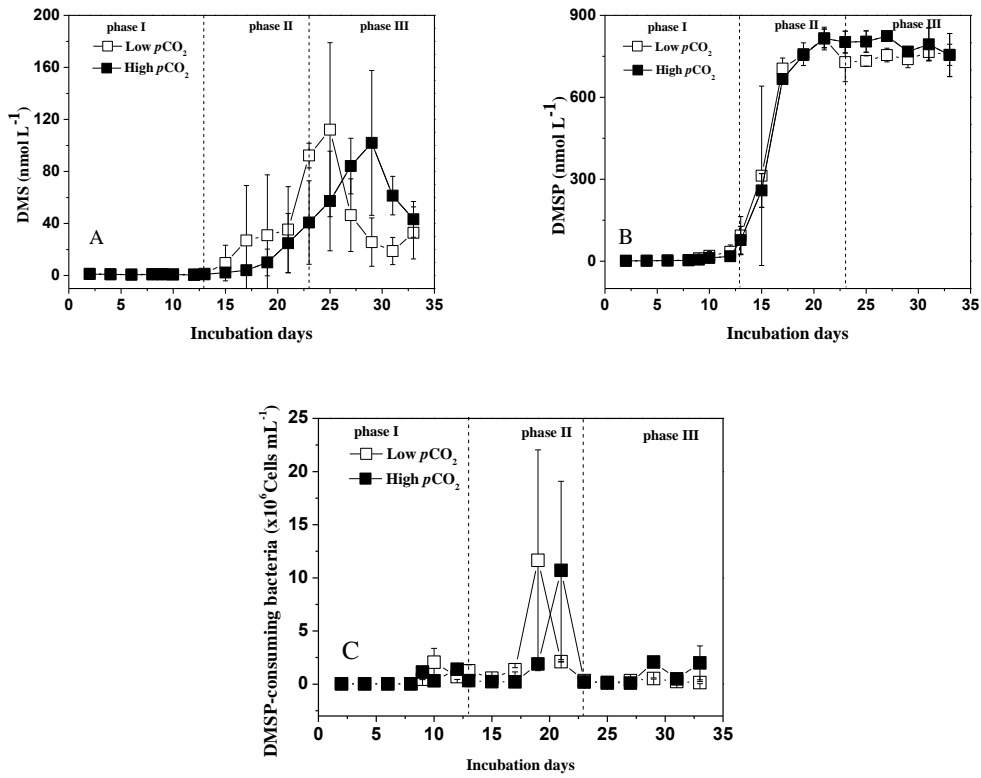
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500 **Fig. 1.** Temporal changes of pH in the high  $p\text{CO}_2$  (1,000  $\mu\text{atm}$ , solid squares) and low  $p\text{CO}_2$  (400  $\mu\text{atm}$ , white  
 501 squares) mesocosms. Data are mean  $\pm$  standard deviation,  $n = 3$  (triplicate independent mesocosm bags) (Origin  
 502 8.0).

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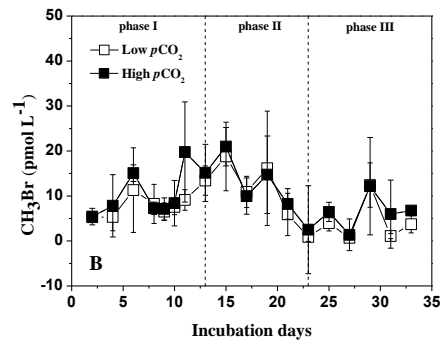
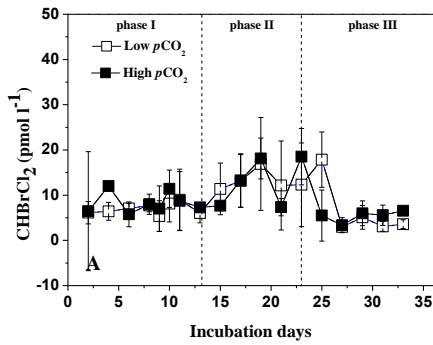
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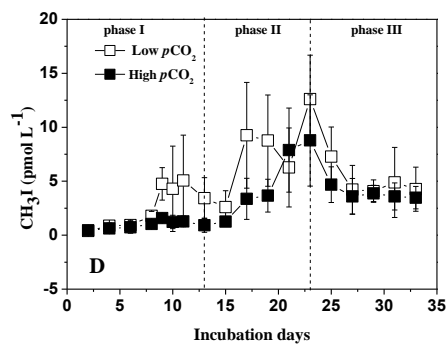
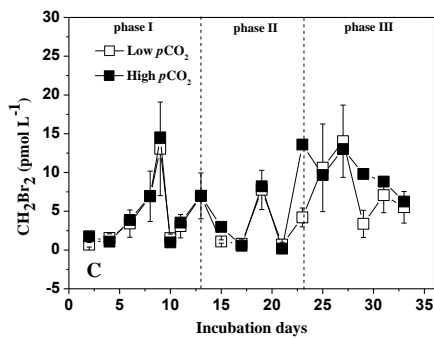
507 **Fig. 2** Temporal changes in dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), DMSP-consuming  
508 bacteria concentrations in the high pCO<sub>2</sub> (1,000 μatm, black squares) and low pCO<sub>2</sub> (400 μatm, white squares)  
509 mesocosms. Data are mean ± standard deviation, n = 3 (triplicate independent mesocosm bags) (Origin 8.0).

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514 **Fig. 3** Temporal changes in bromodichloromethane ( $\text{CHBrCl}_2$ ), methyl bromide ( $\text{CH}_3\text{Br}$ ), dibromomethane  
515 ( $\text{CH}_2\text{Br}_2$ ), iodomethane ( $\text{CH}_3\text{I}$ ) concentrations in the high  $p\text{CO}_2$  (1,000  $\mu\text{atm}$ , black squares) and low  $p\text{CO}_2$  (400  
516  $\mu\text{atm}$ , white squares) mesocosms. Data are mean  $\pm$  standard deviation,  $n = 3$  (triplicate independent mesocosm  
517 bags) (Origin 8.0).

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**Table 1.** Dissolved inorganic carbon (DIC), pH,  $p\text{CO}_2$  and nutrient concentrations in the mesocosm experiments. “-” means that the values were below the detection limit.

		pH	DIC ( $\mu\text{mol kg}^{-1}$ )	$p\text{CO}_2$ ( $\mu\text{atm}$ )	$\text{NO}_3^- + \text{NO}_2^-$ ( $\mu\text{mol L}^{-1}$ )	$\text{NH}_4^+$ ( $\mu\text{mol L}^{-1}$ )	$\text{PO}_4^{3-}$ ( $\mu\text{mol L}^{-1}$ )	$\text{SiO}_3^{2-}$ ( $\mu\text{mol L}^{-1}$ )
day 0	Low $p\text{CO}_2$	8.0 $\pm$ 0.1	2181 $\pm$ 29	1170~1284	52~56	19~23	2.6 $\pm$ 0.2	38~40
	High $p\text{CO}_2$	7.5 $\pm$ 0.1	2333 $\pm$ 34	340~413	51~55	19~23	2.5 $\pm$ 0.2	38~39
Phase I	Low $p\text{CO}_2$	7.9~8.4	1825~2178	373~888	15~52	1.6~20	0.5~2.6	31~38
	High $p\text{CO}_2$	7.4~8.2	2029~2338	1295~1396	47~54	0.2~21	0.7~2.5	34~39
Phase II	Low $p\text{CO}_2$	8.4~8.5	1706~1745	46~749	-- 15.9	-	0.1~0.5	10~24
	High $p\text{CO}_2$	8.4~8.6	1740~1891	59~1164	1.1~25	-	--0.1	29~30
Phase III	Low $p\text{CO}_2$	8.5~8.8	1673~1706	30~43	-	-	-	10~16
	High $p\text{CO}_2$	8.6~8.7	1616~1740	34~110	-	-	--0.3	24~25

520 **Table 2.** Correlation between dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), chlorophyll a (Chl *a*), bromodichloromethane (CHBrCl<sub>2</sub>), methyl bromide (CH<sub>3</sub>Br),  
 521 dibromomethane (CH<sub>2</sub>Br<sub>2</sub>), iodomethane (CH<sub>3</sub>I), DMSP-consuming bacteria, *Thalassiosira weissflogii* (*T. weissflogii*) and *Phaeodactylum tricornutum* (*P. tricornutum*) concentrations in the  
 522 low pCO<sub>2</sub> treatments.

	DMS	DMSP	Chl <i>a</i>	CHBrCl <sub>2</sub>	CH <sub>3</sub> Br	CH <sub>2</sub> Br <sub>2</sub>	CH <sub>3</sub> I	DMSP-consuming bacteria	<i>T. weissflogii</i>	<i>P. tricornutum</i>
DMS	1									
DMSP	0.701**	1								
Chl <i>a</i>	0.597**	0.792**	1							
CHBrCl <sub>2</sub>	0.526	0.280	0.559	1						
CH <sub>3</sub> Br	-0.413	-0.230	0.196	0.313	1					
CH <sub>2</sub> Br <sub>2</sub>	0.310	0.180	0.001	-0.136	-0.308	1				
CH <sub>3</sub> I	0.694**	0.654**	0.717**	0.596*	-0.151	0.129	1			
DMSP-consuming bacteria	0.643**	0.520*	0.522*	0.394	-0.268	-0.038	0.762**	1		
<i>T. weissflogii</i>	0.410	0.617**	0.899**	0.301	0.322	0.028	0.680**	0.399	1	
<i>P. tricornutum</i>	0.560*	0.961**	0.821**	0.528	-0.032	0.162	0.588**	0.334	0.685**	1

523 \*. Correlation is significant at the 0.05 level (2-tailed).

524 \*\*. Correlation is significant at the 0.01 level (2-tailed).

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530 **Table 3.** Correlation between dimethylsulfide (DMS), dimethylsulfoniopropionate (DMSP), chlorophyll a (Chl *a*), bromodichloromethane (CHBrCl<sub>2</sub>), methyl bromide (CH<sub>3</sub>Br),  
 531 dibromomethane (CH<sub>2</sub>Br<sub>2</sub>), iodomethane (CH<sub>3</sub>I), DMSP-consuming bacteria, *Thalassiosira weissflogii* (*T. weissflogii*) and *Phaeodactylum tricornutum* (*P. tricornutum*) concentrations in the  
 532 high *p*CO<sub>2</sub> treatments.

	DMS	DMSP	Chl <i>a</i>	CHBrCl <sub>2</sub>	CH <sub>3</sub> Br	CH <sub>2</sub> Br <sub>2</sub>	CH <sub>3</sub> I	DMSP-consuming bacteria	<i>T. weissflogii</i>	<i>P. tricornutum</i>
DMS	1									
DMSP	0.752**	1								
Chl <i>a</i>	0.318*	0.738**	1							
CHBrCl <sub>2</sub>	0.324	0.094	0.326	1						
CH <sub>3</sub> Br	-0.410	-0.349	0.065	0.076	1					
CH <sub>2</sub> Br <sub>2</sub>	0.540*	0.352	0.142	0.233	-0.377	1				
CH <sub>3</sub> I	0.694**	0.816**	0.741**	0.690*	-0.407	0.316	1			
DMSP-consuming bacteria	0.544*	0.522	0.549*	0.532	-0.311	0.368	0.851*	1		
<i>T. weissflogii</i>	0.355	0.743**	0.930**	0.304	0.076	0.233	0.690**	0.567	1	
<i>P. tricornutum</i>	0.635**	0.954**	0.803**	0.143	-0.257	0.267	0.834**	0.559	0.820**	1

533 \*. Correlation is significant at the 0.05 level (2-tailed).

534 \*\*. Correlation is significant at the 0.01 level (2-tailed).

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