1	Conversion of tropical forests to smallholder rubber and oil palm
2	plantations impacts nutrient leaching losses and nutrient retention
3	efficiency in highly weathered soils
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Abstract. Conversion of forest to rubber and oil palm plantations is widespread in Sumatra, 15 16 Indonesia, and it is largely unknown how such land-use conversion affects nutrient leaching losses. Our study aimed to quantify nutrient leaching and nutrient retention efficiency in the 17 soil after land-use conversion to smallholder rubber and oil palm plantations. In Jambi province, 18 Indonesia, we selected two landscapes on highly weathered Acrisol soils that mainly differed 19 in texture: loam and clay. Within each soil type, we compared two reference land uses: lowland 20 21 forest and jungle rubber (defined as rubber trees interspersed in secondary forest) with two converted land uses, smallholder rubber and oil palm plantations. Within each soil type, the 22 first three land uses were represented by four replicate sites and the oil palm by three sites, 23 24 totaling to 30 sites. We measured leaching losses using suction cup lysimeters, sampled biweekly to monthly from February to December 2013. Forests and jungle rubber had low 25 solute concentrations in drainage water, suggesting low internal inputs of rock-derived nutrients 26 27 and efficient internal cycling of nutrients. These reference land uses on the clay Acrisol soils had lower leaching of dissolved N and base cations (P = 0.01-0.06) and higher N and base 28 29 cation retention efficiency (P < 0.01-0.07) than those on the loam Acrisols. In the converted land uses, particularly on the loam Acrisol, the fertilized area of oil palm plantations showed 30 higher leaching of dissolved N, organic C and base cations (P < 0.01-0.08) and lower N and 31 base cation retention efficiency compared to all the other land uses (P < 0.01-0.06). The 32 unfertilized rubber plantations, particularly on the loam Acrisol, showed lower leaching of 33 dissolved P (P = 0.08) and organic C (P < 0.01) compared to forest or jungle rubber, reflecting 34 decreases in soil P stocks and C inputs to the soil. Our results suggest that land-use conversion 35 to rubber and oil palm causes disruption of initially efficient nutrient cycling, which decreases 36 nutrient availability. Over time, smallholders will likely be increasingly reliant on fertilization, 37 with the risk of diminishing water quality due to increased nutrient leaching. Thus, there is a 38 need to develop management practices to minimize leaching while sustaining productivity. 39

40 **1 Introduction**

Rainforests play an important role in maintaining ground water quality in tropical regions; 41 however, in some regions their effectiveness may be decreasing as a consequence of forest 42 conversion to agriculture. From 1990 to 2010, the deforestation rate in South and Southeast 43 Asia was approximately 3 million ha yr⁻¹, of which 1.2 million ha yr⁻¹ occurred in Indonesia 44 (FAO, 2010). During these two decades, the forest loss in the whole of Sumatra was 7.5 million 45 46 ha, of which 1.1 million ha occurred in Jambi province (Margono et al., 2012). The two most common land uses replacing forests in Jambi province are oil palm and rubber plantations. From 47 2000 to 2010, the area of rubber plantations in Jambi increased by about 19 % while oil palm 48 49 plantations increased by 85 % (Luskin et al., 2013). The expansion of rubber and oil palm plantations has increased the income of Jambi, in particular the smallholder farmers (Clough et 50 al., 2016; Rist et al., 2010), which account 99 % of rubber and 62 % of oil palm landholdings 51 52 in the Jambi Province. In the whole of Indonesia, 85 % of rubber and 40 % of oil palm plantations are smallholders (DGEC, 2017). However, forest conversion to rubber and oil palm 53 54 plantations has shown high ecological costs: losses in biodiversity (Clough et al., 2016), decreases in above- and below-ground organic carbon (C) stocks (Kotowska et al., 2015; van 55 Straaten et al., 2015), reduction in soil nitrogen (N) availability (Allen et al., 2015), decrease in 56 57 uptake of methane (CH₄) from the atmosphere into the soil (Hassler et al., 2015), and increase in soil N₂O emission following N fertilization (Hassler et al., 2017). 58

59 Under similar climatic conditions and soil types, the two major factors that influence 60 nutrient leaching losses from forest conversion are soil texture and management practices. Soil 61 texture affects nutrient leaching through its control on soil fertility (e.g., cation exchange 62 capacity, decomposition, and nutrient cycling) and soil water-holding capacity. Fine-textured 63 soils have higher cation exchange capacity, decomposition and soil-N cycling rates, which 64 result in higher soil fertility than coarse-textured soils (Allen et al., 2015; Silver et al., 2000;

Sotta et al., 2008). Soil texture also influences water-holding capacity and drainage through its 65 66 effects on porosity, pore size distribution, and hydraulic conductivity (Hillel, 1982). Clay soils can hold a large amount of water and are dominated by small pores, which have low hydraulic 67 conductivity in high moisture conditions. In contrast, coarse-textured soils have low water-68 holding capacity and are dominated by large pores, which conduct water rapidly in high 69 moisture conditions, and therefore have high potential for leaching of dissolved solutes (Fujii 70 71 et al., 2009; Lehman and Schroth, 2002). Thus, in heavily weathered soils, such as Acrisols, which dominate the converted lowland landscapes in Jambi, Indonesia (FAO et al., 2012), 72 retention of their inherently low exchangeable base cations in the soil and maintenance of 73 74 efficient soil-N cycling are largely influenced by soil texture (Allen et al., 2015).

Soil management practices (e.g., fertilizer and lime applications) in converted land uses 75 also play an important role in influencing nutrient leaching, as the magnitude of dissolved 76 77 nutrients moving downward with water is predominantly driven by the levels of those nutrients in the soil (Dechert et al., 2005, 2004). Without fertilization, nutrient leaching losses in 78 79 agricultural land usually decrease with years following forest conversion (Dechert et al., 2004). This may be the case for the smallholder rubber plantations in our present study, as these have 80 not been fertilized since conversion from forest (Allen et al., 2015; Hassler et al., 2017, 2015). 81 82 However, soils in oil palm plantations are very often supplemented with chemical fertilizer and lime applications (Allen et al., 2015; Goh et al., 2003; Hassler et al., 2017, 2015). In cases 83 where oil palm plantations are regularly fertilized, nutrient leaching losses in older plantations 84 may be higher than in younger ones, as the applied nutrients accumulate in the subsoil over 85 time (Goh et al., 2003; Omoti et al., 1983). Consequently, nutrient leaching in regularly 86 fertilized oil palm plantations will likely be higher than in the original forest. Moreover, in our 87 earlier study conducted in smallholder oil palm plantations, fertilization was shown to decrease 88

microbial N immobilization due to decreases in microbial biomass (Allen et al., 2015), which
could lead to decrease in retention of N in the soil.

Despite a growing body of information on the effects of deforestation on soil properties 91 and processes, there is a lack of information on how forest conversion to rubber and oil palm 92 influences nutrient leaching and the efficiency with which nutrients are retained in the soil. This 93 lack is especially notable for nutrients other than N, as previous leaching studies commonly 94 95 focus on this. Here, we present leaching losses of the full suite of major nutrients using a largescale replicated design in a region affected by widespread land-use conversion to rubber and 96 oil palm plantations. Our study aimed to assess: 1) how soil physical and biochemical 97 98 characteristics affect nutrient leaching in highly weathered soils, and 2) the impact of land-use 99 conversion to smallholder rubber and oil palm plantations on nutrient leaching and on N and base cation retention efficiency in the soil. We hypothesized that: 1) lowland forest and jungle 100 101 rubber (rubber trees planted in secondary forest), which were the previous land uses before conversion, will have lower leaching losses and higher nutrient retention in clay Acrisol soil 102 103 than in loam Acrisol soil, and 2) smallholder oil palm plantations with fertilizer and lime applications will have the highest nutrient leaching losses (lowest nutrient retention) whereas 104 105 smallholder rubber plantations with no fertilizer input will have the lowest nutrient leaching 106 losses.

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108 2 Materials and methods

109 2.1 Study sites and experimental design

Our study is part of the on-going multidisciplinary research project, EFForTS (http://www.unigoettingen.de/en/310995.html), investigating the ecological and socioeconomic impact of conversion of lowland forest to rubber and oil palm plantations. The detailed experimental design and locations of the study sites were reported earlier (e.g., Allen et al., 2015; Hassler et

al., 2017, 2015). In short, our study region is located in Jambi province, Indonesia (2° 0' 57" S, 114 103° 15' 33" E, 35 - 95 m elevation). The area has a mean annual air temperature of 26.7 ± 0.1 115 $^{\circ}$ C and a mean annual precipitation of 2235 ± 385 mm (1991–2011; data from a climate station 116 117 at the Jambi Sultan Thaha airport from the Indonesian Meteorological, Climatological and Geophysical Agency). The dry season (<100 mm month⁻¹) is from May to September, and the 118 wet season is from October to April. We selected two landscapes within our study region; while 119 120 both were located on highly weathered Acrisol soils, one has clay-textured soils and the other as loam-textured soils (hereafter we refer to them as clay Acrisol and loam Acrisol soils). 121 Detailed soil characteristics of these classifications are reported by Allen et al. (2016). In 122 123 summary, the soil textural difference leads to inherent differences in soil fertility (e.g., higher effective cation exchange capacity, base saturation, Bray-extractable P and lower Al saturation) 124 in the clay than the loam Acrisols under forest and jungle rubber (Appendix Table A1). Within 125 126 each soil type, we selected four land uses: lowland forest, jungle rubber, and smallholder plantations of rubber and oil palm (Appendix Table A2). Within each soil type, we had 15 sites 127 (see Allen et al. 2015 for the map of these sites in the study region): four forest, four jungle 128 rubber, four rubber plantations, and three oil palm plantations. We started with four oil palm 129 130 sites at each soil type, but one plantation was sold and the new owner did not continue the 131 collaboration with our research and in another site the instruments for leaching sampling were damaged. In our experimental design, land-use types (including the soil management practices 132 typical for smallholders in the region) were the treatment and the sites were the replications. At 133 134 each site, we established a plot of 50 m x 50 m. All plots were on the well-drained position of the landscape with slopes ranging from 3-10 % across all plots. 135

Based on our interviews with the smallholders, their plantations were established after clearing and burning of either forest or jungle rubber and hence these latter land uses served as the reference with which the converted plantations were compared. Additionally, the

comparability of the initial soil conditions between the reference and converted land uses was 139 140 tested using a land use-independent soil characteristic, i.e., clay content at 1-2 m depth (van Straaten et al., 2015); this did not statistically differ among land uses within each soil type 141 (Appendix Table A1; Allen et al., 2015; Hassler et al., 2015). Thus, changes in nutrient leaching 142 can be attributed to land-use conversion with its inherent soil management practices. These first 143 generation rubber and oil palm plantations were between 7 and 17 years of age. Tree density, 144 145 height, basal area, and tree species abundance were higher in the reference land uses than the smallholder plantations (Appendix Table A2; Allen et al., 2015; Hassler et al., 2015; Kotowska 146 et al., 2015). 147

148 Soil management practices in smallholder oil palm plantations are inherently varied (e.g., fertilization rate), as this depended on financial resources of the smallholders. Fertilization 149 rates were 48 kg N, 21 kg P and 40 kg K ha⁻¹ yr⁻¹ in the clay Acrisol soil, whereas these were 150 88 kg N, 38 kg P and 157 kg K ha⁻¹ yr⁻¹ (accompanied by Cl input of 143 kg Cl ha⁻¹ yr⁻¹) in the 151 loam Acrisol soil. Lime (e.g., CaMg(CO₃)₂), kieserite (MgSO₄.H₂O) and borate 152 (Na₂B₄O₂.5H₂O) were also occasionally applied. These fertilization rates are typical of the 153 smallholder farms in the region. Soil amendments were applied by hand around each palm tree 154 at 0.8-1.5 m from the stem base. A combination of manual weeding and herbicides was 155 156 practiced. Old oil palm fronds were regularly cut and stacked at 4–4.5 m from the palm rows (row spacing was about 9 m). The rubber plantations were not fertilized but were weeded both 157 manually and with herbicides. 158

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160 2.2 Lysimeter installation and soil water sampling

For measuring nutrient leaching, we sampled soil water using lysimeters, which were installed at two randomly chosen locations per replicate plot of the forest, jungle rubber and rubber plantations. In the oil palm plantations, the lysimeters were deployed according to the spatial

structure of the soil management practices: one lysimeter was installed between 1.3-1.5-m 164 distance from the tree stem where fertilizers were applied, and another lysimeter was installed 165 between 4–4.5-m distance from the tree stem where the cut fronds were stacked. These suction 166 cup lysimeters (P80 ceramic, maximum pore size 1 µm; CeramTec AG, Marktredwitz, 167 Germany) were inserted into the soil down to 1.5-m depth. This depth was based from our 168 previous work in a lowland forest on highly weathered Ferralsol soil, where leaching losses 169 170 were measured at various depth intervals down to 3 m and from which we found that leaching fluxes did not change below 1 m (Schwendenmann and Veldkamp, 2005). Moreover, this 1.5-171 m depth of lysimeter installation at our sites was well below the rooting depth, as determined 172 173 from the fine-root biomass distribution with depths (Appendix Fig. B1; Kurniawan, 2016).

174 Prior to installation, lysimeters, tubes and collection containers were acid-washed and rinsed with deionized water. Lysimeters were installed in the field three months prior to the first 175 176 sampling. The collection containers (dark glass bottles) were placed in plastic buckets with lids and buried in the ground approximately 2 m away from the lysimeters. Soil water was sampled 177 biweekly to monthly, depending on the frequency of rainfall, from February to December 2013. 178 Soil water was withdrawn by applying a 40 kPa vacuum on the sampling tube (Dechert et al., 179 180 2005; Schwendenmann and Veldkamp, 2005). The collected soil water was then transferred 181 into clean 100-mL plastic bottles. Upon arrival at the field station, a subsample of 20 mL was set aside for pH measurement while the remaining sample was frozen. All frozen water samples 182 were transported to the University of Goettingen, Germany and were kept frozen until analysis. 183 184 The total dissolved N (TDN), NH₄⁺, NO₃⁻ and Cl⁻ concentrations were measured using continuous flow injection colorimetry (SEAL Analytical AA3, SEAL Analytical GmbH, 185 Norderstedt, Germany). TDN was determined by ultraviolet-persulfate digestion followed by 186 hydrazine sulfate reduction (Autoanalyzer Method G-157-96); NH4⁺ was analyzed by salicylate 187

and dicloroisocyanuric acid reaction (Autoanalyzer Method G-102-93); NO₃⁻ by cadmium

reduction method with NH₄Cl buffer (Autoanalyzer Method G-254-02); and Cl⁻ was determined 189 190 with an ion strength adjustor reagent that is pumped through an ion selective chloride electrode with an integrated reference electrode (Auto analyzer Method G-329-05). Dissolved organic N 191 (DON) is the difference between TDN and mineral N ($NH_4^+ + NO_3^-$). Dissolved organic C 192 (DOC) was determined using a Total Organic Carbon Analyzer (TOC-Vwp, Shimadzu Europa 193 GmbH, Duisburg, Germany). DOC was analyzed by pre-treating the samples with H₃PO₄ 194 solution (to remove inorganic C) followed by ultraviolet-persulfate oxidation of organic C to 195 CO₂, which is determined by an infrared detector. Base cations (Na, K, Ca, Mg), total Al, total 196 Fe, total Mn, total S, total P, and total Si in soil water were analyzed using inductively coupled 197 198 plasma-atomic emission spectrometer (iCAP 6300 Duo View ICP Spectrometer, Thermo Fischer Scientific GmbH, Dreieich, Germany). Instruments' detection limits were: 6 µg NH4⁺-199 N L⁻¹, 5 μg NO₃⁻-N L⁻¹, 2 μg TDN L⁻¹, 4 μg DOC L⁻¹, 30 μg Na L⁻¹, 50 μg K L⁻¹, 3 μg Ca L⁻¹, 200 $3 \mu g M g L^{-1}$, $2 \mu g A l L^{-1}$, $3 \mu g F e L^{-1}$, $2 \mu g M n L^{-1}$, $10 \mu g P L^{-1}$, $10 \mu g S L^{-1}$, $1 \mu g S i L^{-1}$ and 30201 $\mu g \operatorname{Cl} L^{-1}$. 202

203 Partial cation-anion charge balance of the major solutes (i.e., those with concentrations $>0.03 \text{ mg } \text{L}^{-1}$) in soil water was done by expressing solute concentrations in μ mol_c L⁻¹ (molar 204 concentration multiplied by the equivalent charge of each solute). Contributions of organic 205 206 acids (RCOO⁻) and bicarbonate (HCO₃⁻) were calculated, together with S (having very low concentrations), from the difference between cations and anions. Charge contributions of total 207 Al were assumed to be 3⁺, whereas solutes that had very low concentrations (i.e., total Fe, Mn 208 and P), and thus had minimal charge contribution, as well as the total dissolved Si (commonly 209 in a form of monosilicic acid $(H_4SiO_4^0)$ that has no net charge) were excluded (similar to the 210 method used by Hedin et al., 2003). 211

213 **2.3** Soil water modelling and calculation of nutrient leaching fluxes

214 Drainage water fluxes were estimated using the soil water module of the Expert-N model (Priesack, 2005), which has been used in our earlier work on nutrient leaching losses in 215 Sulawesi, Indonesia (Dechert et al., 2005). The model was parameterized with the 216 characteristics measured at our sites, namely climate data, leaf area index, rooting depth, and 217 soil characteristics. The climate variables included daily air temperature (minimum, maximum 218 219 and average), relative humidity, wind speed, solar radiation, and precipitation. For the loam Acrisol soil, the climate data were taken from a climate station at the Harapan Forest Reserve, 220 which was located 10-20 km from our sites. For the clay Acrisol soil, the climate data were 221 222 taken from the climate stations at the villages of Lubuk Kepayang and Sarolangun, which were respectively 10 km and 20 km from our sites. The leaf area indices measured in our forest, 223 jungle rubber, rubber and oil palm sites in the loam Acrisol soil were 5.8, 4.8, 3.5, and 3.9 m^2 224 m⁻², respectively, and in the clay Acrisol soil were 6.2, 4.5, 2.8 and 3.1 m² m⁻², respectively 225 226 (Rembold et al., unpublished data). Our measured fine root biomass distribution (Appendix Fig. 227 B1; Kurniawan, 2016) was used to partition root water uptake at various soil depths. Soil characteristics included soil bulk density, texture (Appendix Table A1) and the water retention 228 229 curve. The latter was determined using the pressure plate method for which intact soil cores (250 cm³), taken at five soil depths (0.05, 0.2, 0.4, 0.75 and 1.25 m) from each land use within 230 each soil type, were measured for water contents at pressure heads of 0, 100, 330 and 15000 231 hPa. 232

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Calculation of drainage water fluxes followed the water balance equations:

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 $\Delta W + D = P - R - ET$ and ET = I + E + T

in which ΔW = change in soil water storage, D = drainage water below rooting zone, P = precipitation, R = runoff, ET = evapotranspiration, I = interception of water by plant foliage, E = evaporation from soil, and T = transpiration by plants. The Expert-N model calculates actual

evapotranspiration using the Penman-Monteith method, runoff based on the sites' slopes, and 238 239 vertical water movement using the Richards equation, of which the parameterization of the hydraulic functions were based on our measured soil texture and water retention curve 240 241 (Mualem, 1976; Van Genuchten, 1980). To validate the output of the water model, we compared the modelled and measured soil matrix potential (Appendix Fig. B2). Soil matrix 242 potential was measured biweekly to monthly from February to December 2013, using 243 tensiometers (P80 ceramic, maximum pore size 1 µm; CeramTec AG, Marktredwitz, 244 Germany), which were installed at the depths of 0.3 m and 0.6 m in two replicate plots per land 245 use within each soil type. 246

Modelled daily drainage water fluxes at a depth of 1.5 m were summed to get the biweekly or monthly drainage fluxes. Nutrient leaching fluxes were calculated by multiplying the element concentrations from each of the two lysimeters per replicate plot with the total biweekly or monthly drainage drainage water flux. The annual leaching flux was the sum of biweekly to monthly measured leaching fluxes from February to December 2013, added with the interpolated value for the unmeasured month of January 2013.

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254 **2.4 Nutrient retention efficiency**

255 To evaluate the efficiency with which nutrients are retained in soil, we calculated the N and base cation retention efficiency as follows: 1 – (nutrient leaching loss/soil available nutrient) 256 (Hoeft et al., 2014). For the oil palm plantations, we took the average leaching fluxes in the 257 fertilized and frond-stacked areas of each plot for calculating the nutrient retention efficiency. 258 This is because these sampling locations may contribute equally in terms of area as both the 259 vertical and lateral flows in the soil profile could influence the sampled drainage water, and 260 thus a wider area may contribute to the sampled drainage water than just the categorized 261 sampling locations. For N retention efficiency calculation, TDN leaching flux was ratioed to 262

263 gross N mineralization rate as the index of soil available N, with both terms expressed in mg N 264 $m^{-2} d^{-1}$. For calculation of base cation retention efficiency, base cation leaching flux was the 265 sum of K, Na, Mg and Ca in units of mol_{charge} $m^{-2} yr^{-1}$ and soil available base cations was the 266 sum of these exchangeable cations in units of mol_{charge} m^{-2} . We used the measurements of gross 267 N mineralization rate in the top 0.05-m depth and the stocks of exchangeable bases in the top 268 0.1-m depth (Appendix Table A1, reported by Allen et al., 2015).

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270 **2.5 Supporting parameter: nutrient inputs through bulk precipitation**

In each landscape, we installed two rain samplers in an open area at 1.5 m above the ground. 271 Rain samplers consisted of 1-liter high-density polyethylene bottles with lids attached to 272 funnels that were covered with a 0.5-mm sieve, and were placed inside polyvinyl chloride tubes 273 (to shield from sunlight and prevent algal growth). Rain samplers were washed with acid and 274 275 rinsed with deionized water after each collection. Rain was sampled during the same sampling period as the soil water. Each rain sample was filtered through prewashed filter paper (4 µm 276 277 pore size) into a 100 mL plastic bottle and stored frozen for transport to the University of Goettingen, Germany. The element analyses were the same as those described for soil water. 278 The element concentrations in rainwater were weighted with the rainfall volume during the two-279 week or 1-month collection period to get volume-weighted concentrations. The annual element 280 inputs from bulk precipitation were calculated by multiplying the volume-weighted average 281 282 element concentrations in a year with the annual rainfall in each landscape.

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284 **2.6 Statistical analysis**

Each replicate plot was represented by the average of two lysimeters, except for the oil palm plantations where lysimeters in fertilized and frond-stacked areas were analyzed separately. Tests for normality (Shapiro-Wilk's test) and homogeneity of variance (Levene's test) were

conducted for each variable. Logarithmic or square-root transformation was used for variables 288 289 that showed non-normal distribution and/or heterogeneous variance. We used linear mixed effects (LME) models (Crawley, 2009) to (1) assess differences between the two soil types for 290 291 the reference land uses (to answer objective 1), and (2) assess differences among land-use types within each soil type (to answer objective 2). The latter was analyzed for each landscape 292 because the fertilization rates applied to the smallholder oil palm plantations inherently differed 293 294 between the two landscapes. For element concentrations, the LME model had soil type or land use as the fixed effect with spatial replication (plot) and time (biweekly or monthly 295 measurements) as random effects. For the annual leaching fluxes, the LME model had soil type 296 297 or land use as the fixed effect with spatial replication (plot) as a random effect. If they improved 298 the relative goodness of the model fit (based on the Akaike information criterion), we extended 299 the LME model to include (1) a variance function that allows different variances of the fixed 300 effect, and/or (2) a first-order temporal autoregressive process that assumes that correlation between measurement periods decreases with increasing time intervals. Fixed effects were 301 302 considered significant based on analysis of variance at $P \le 0.05$, and differences between soil types or land uses were assessed using Fisher's least significant difference test at $P \leq 0.05$. 303 Given the inherent spatial variability in our experimental design, we also considered P values 304 305 of $> 0.05 \le 0.09$ as marginal significance, mentioned explicitly for some variables. To support 306 the partial charge balance of dissolved cations and anions, we used Pearson correlation analysis to assess the relationships between solute cations and anions, using the monthly average (n =307 308 12) of the four replicate plots per land use within each soil type. We also used Pearson 309 correlation analysis to test the modelled and measured soil matrix potential, using the monthly average (n = 12) of the measured two replicate plots per land use within each soil type. To 310 assess how the soil physical and biochemical characteristics (Table A1) influence the annual 311 nutrient leaching fluxes, we conducted Spearman's rank correlation test for these variables, 312

separately for the reference land uses and the converted land uses across both soil types (n = 16). All statistical analyses were conducted using R 3.0.2 (R Development Core Team, 2013).

316 **3 Results**

317 **3.1** Water balance and nutrient input from bulk precipitation

The modelled and measured soil matric potential were highly correlated (R = 0.79 to 0.98, n =318 319 12, P < 0.01) (Appendix Fig. B2). In forest and jungle rubber, modelled annual ET was 36-47 %, runoff was 16-27 %, and drainage was 32-44 % of annual precipitation. In rubber and oil 320 palm, modelled annual ET was 30-32 %, runoff was 22-31 %, and drainage was 37-47 % of 321 annual precipitation (Table 1). In both landscapes, annual input from bulk precipitation was 322 dominated by DOC (58 % of total element deposition), followed by Na, Cl, TDN, Ca, K and 323 total S (Table 2). We compared the chlorinity ratios of elements in the bulk precipitation at our 324 325 sites to those of seawater to infer anthropogenic influence. The average chlorinity ratios from both landscapes were 1.13 ± 0.05 for Na:Cl, 0.05 ± 0.01 for Mg:Cl, 0.20 ± 0.02 for Ca:Cl and 326 327 0.13 ± 0.04 for K:Cl, which were higher, except for Mg:Cl, than seawater chlorinity ratios (0.56 for Na:Cl, 0.07 for Mg:Cl, 0.02 for Ca:Cl and 0.02 for K:Cl; p. 349, Schlesinger and Bernhardt, 328 2013). 329

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331 **3.2 Element concentrations in soil water**

For forest, the loam Acrisol had higher dissolved Na, Mg, total Al (all $P \le 0.05$), NH₄⁺-N, DON, total Fe and Cl concentrations (all $P \le 0.09$) than the clay Acrisol (Table 3). For jungle rubber, the loam Acrisol had higher dissolved NO₃⁻-N ($P \le 0.05$) and lower total Si concentrations ($P \le 0.09$) than the clay Acrisol (Table 3). The ionic charge concentration of soil solution in the forest sites was higher in the loam (274 ± 19 µmol_{charge} L⁻¹) than in the clay Acrisols (203 ± 20 µmol_{charge} L⁻¹) (P = 0.01; Fig. 1), whereas in the jungle rubber these were comparable (loam Acrisols: $199 \pm 31 \ \mu \text{mol}_{charge} \ \text{L}^{-1}$, clay Acrisols: $207 \pm 24 \ \mu \text{mol}_{charge} \ \text{L}^{-1}$; Fig. 1). Correlation analysis of dissolved cations and anions in forest and jungle rubber showed that NH_4^+ -N, Na, K, Ca, Mg and total Al were positively correlated with DON, DOC, Cl, NO_3^- -N and total S (Appendix Tables A3 and A4).

The rubber plantations in the loam Acrisol had lower NO₃⁻-N, DON, DOC, Na, Ca, Cl 342 (all $P \le 0.05$), total P and total S concentrations (both $P \le 0.08$) than either forest or jungle 343 rubber (Table 3). This resulted in lower ionic charge concentration of soil solution in rubber 344 plantation (200 \pm 21 µmol_{charge} L⁻¹) than in forest (P < 0.01; Fig. 1). In the clay Acrisol, only 345 dissolved Na was lower in rubber plantations than in jungle rubber ($P \le 0.01$; Table 3), and 346 hence the ionic charge concentration of soil solution in rubber plantation (189 \pm 23 μ mol_{charge} 347 L⁻¹) were comparable to those in the reference land uses (Fig. 1). In contrast to the reference 348 land uses, unfertilized rubber plantations showed strong positive correlations of dissolved 349 350 cations (NH₄⁺-N, Na, K, Ca, Mg and total Al) with Cl and only weaker positive correlations with DOC or total S (Appendix Tables A3 and A4). 351

The fertilized areas of oil palm plantations had higher NO₃⁻-N, Na, Ca, Mg, total Al, Cl 352 (all $P \le 0.05$) and lower soil solution pH (P = 0.07) than in the reference land uses within the 353 loam Acrisol soil (Table 3). In the clay Acrisol soil, the fertilized areas of oil palm plantations 354 had higher soil solution pH and dissolved Na (both $P \le 0.05$) whereas DON was lower (P =355 0.08) than the reference land uses (Table 3). Ionic charge concentrations of soil solutions in the 356 fertilized areas of oil palm plantations (648 \pm 306 μ mol_{charge} L⁻¹ for loam Acrisol and 317 \pm 83 357 μ mol_{charge} L⁻¹ for clay Acrisol) were higher than in frond-stacked areas (190 ± 23 μ mol_{charge} L⁻¹ 358 for loam Acrisol and $173 \pm 37 \,\mu \text{mol}_{\text{charge}} \, \text{L}^{-1}$ for clay Acrisol) and in other land uses (P < 0.01; 359 Fig. 1). In the fertilized areas of the loam Acrisol, dissolved NO₃-N was positively correlated 360 with total Al (Table A3) and both were negatively correlated with soil solution pH (R = -0.57361 to -0.76, $n = 12, P \le 0.05$). The fertilized areas showed strong positive correlations of dissolved 362

cations (Na, K, Ca, Mg and total Al) with total S or Cl and only weaker positive correlations
with DOC (Appendix Tables A3 and A4). The frond-stacked areas showed positive correlations
of these dissolved cations largely with Cl (Appendix Tables A3 and A4).

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367 **3.3 Annual leaching flux and nutrient retention efficiency**

For forest, annual leaching fluxes of Na, Ca, Mg, total Al, Cl (all $P \le 0.05$), NH₄⁺-N, DON, 368 369 total Si ($P \le 0.09$) were larger in the loam than in the clay Acrisols, whereas in jungle rubber only annual NO₃⁻-N leaching flux was larger ($P \le 0.05$) (Table 4). Across all forest and jungle 370 rubber sites, annual leaching fluxes of anions (DON and NO3⁻-N) were negatively correlated 371 372 with indicators of soil exchangeable cations (base saturation, effective cation exchange capacity (ECEC), exchangeable Al; Spearman's $\rho = -0.51$ to -0.61, n = 16, $P \le 0.05$), while annual NH₄⁺-373 N leaching flux was negatively correlated (Spearman's $\rho = -0.53$, n = 16, P = 0.04) with soil 374 organic C (Table A1). For both reference land uses, the higher leaching in loam than in clay 375 Acrisols was mirrored by decreases in N and base cation retention efficiency in the soil (Table 376 377 5). Across all reference sites, N and base cation retention efficiency in the soil were positively correlated with base saturation, ECEC and soil organic C (Spearman's $\rho = 0.52$ to 0.70, n = 16, 378 $P \le 0.04$) which, in turn, were positively correlated with clay content (Spearman's $\rho = 0.55$ to 379 0.59, n = 12 sites analyzed for clay content, $P \le 0.05$). 380

The rubber plantations had lower annual P leaching flux than forests (P = 0.08) and lower annual DOC leaching flux than jungle rubber in the loam Acrisol (P < 0.01) (Table 4). N and base cation retention efficiency in the soil of rubber plantations were comparable with the reference land uses in both soil types (Table 5). In oil palm plantations of the loam Acrisol soil, the fertilized areas had higher annual leaching fluxes of NO₃⁻, TDN, DOC, Na, Ca, Mg, total Al, total S and Cl (all $P \le 0.05$) than in the unfertilized rubber plantations or the reference land uses, whereas the frond-stacked areas showed comparable leaching fluxes with the other land

uses (Table 4). In the loam Acrisol, oil palm plantations had lower N and base cation retention 388 389 efficiency in the soil than the other land uses ($P \le 0.01 - 0.06$; Table 5). In the clay Acrisol soil, where leaching fluxes were small (Table 4), there were no differences observed in soil N and 390 base cation retention efficiency among land uses (Table 5). Across all rubber and oil palm sites, 391 annual NH4⁺-N and DON leaching fluxes were negatively correlated with ECEC and clay 392 content (Spearman's $\rho = -0.50$ to -0.64, $n \le 16$, P = 0.03 - 0.07). Moreover, base cation retention 393 efficiency in the soil was positively correlated with ECEC, soil organic C and clay content 394 (Spearman's $\rho = 0.68$ to 0.91, $n \le 16$, $P \le 0.01 - 0.02$) which, in turn, were correlated with each 395 other (*Spearman's* $\rho = 0.87$ to 0.90, n = 12 sites analyzed for clay content, $P \le 0.01$). 396

397

398 4 Discussion

399 4.1 Water balance and nutrient input from bulk precipitation

400 Our modelled water balance was generally comparable with the estimates from other studies in Indonesia. When compared to a forest at 200-500 m elevation on a clay loam soil in Kalimantan 401 (with 28-47 % ET and 40-55 % runoff of 3451 mm yr⁻¹ precipitation; Survatmojo et al., 2013), 402 our estimated ET in the forest sites was comparable, although our modelled runoff was lower 403 (Table 1). However, our runoff estimates were similar to the modelled runoff in oil palm and 404 rubber plantations in Jambi province (10-20 % of rainfall; Tarigan et al., 2016). Our values for 405 runoff and drainage flux in oil palm plantations (Table 1) were similar to oil palm plantations 406 407 at 130 m elevation on Andisol soils in Papua New Guinea (with 37-57 % ET, 0-44 % runoff, and 38-59 % drainage of 2398-3657 mm yr⁻¹ precipitation; Banabas et al., 2008). Additionally, 408 our estimated daily ET in oil palm $(2.4 \pm 0.1 \text{ and } 2.2 \pm 0.1 \text{ mm d}^{-1} \text{ in the loam and clay Acrisols,}$ 409 respectively) was similar to the measurements of Niu et al. (2015) $(2.6 \pm 0.7 \text{ mm d}^{-1})$ in the 410 same oil palm plantations included in our study. Finally, the high correlations between modelled 411

and measured matric potential (0.3-m depth; Appendix Fig. B2) suggest that our modelleddrainage fluxes closely approximated those in the studied land uses.

The chemical composition of bulk precipitation in our study area was clearly influenced 414 by biomass burning and terrigenous dust from agriculture. This is evident from the high DOC, 415 TDN, Na:Cl, K:Cl and Ca:Cl ratios in bulk precipitation, which were comparable to values of 416 bulk precipitation impacted by such anthropogenic activities in Southeast Asia as well as in 417 418 Latin America (Balasubramanian et al., 1999; Corre et al., 2010; Eklund et al., 1997). From a peatland in Kalimantan, influenced by land-clearing fires, throughfall nutrient depositions (19-419 22 kg N, 6-11 kg P, 25-44 kg S ha⁻¹ yr⁻¹) are larger than those from bulk precipitation, indicating 420 421 large contribution from dry deposition (Ponette-Gonzales et al., 2016). Total (wet + dry) nutrient depositions in our study region could be larger than the values from bulk precipitation. 422 High atmospheric nutrient deposition may have fertilizing or polluting effect, depending on 423 424 whether or not the receiving ecosystem is a sink and able to buffer its other cascading effects (e.g., acidification). Additionally, atmospheric redistribution of nutrients in areas with 425 426 widespread land-use conversion and intensification may have unforeseen effects on down-wind and down-stream ecosystems (e.g., Bragazza et al., 2016; Sundarambal et al., 2010). 427

428

429 **4.2** Leaching fluxes and nutrient retention efficiency in the reference land uses

Highly weathered soils (e.g., Acrisols and Ferralsols) are characterized by low solute concentrations in drainage and stream waters due to minimal internal input of rock-derived nutrients via weathering (Hedin et al., 2003; Markewitz et al., 2001). Our reference land uses on Acrisol soils exhibited comparably low ionic charge concentration with high dissolved Al (Fig. 1) as those reported by these authors. Soil nutrients of such highly weathered soils are conserved through efficient cycling between the soil and vegetation, for which soil texture is one important controlling factor. For example, fine-textured Acrisol and Ferralsol soils show

higher nutrient- and water-holding capacity, higher soil N availability, decomposition rate and 437 438 plant productivity than the coarse-textured soils of the same groups (e.g., Ohta et al., 1993; Silver et al., 2000; Sotta et al., 2008). Our measured nutrient leaching losses concurred to these 439 findings. The lower annual nutrient leaching fluxes in clay as compared to loam Acrisols (i.e., 440 TDN, Na, Ca, Mg; Table 4) were paralleled by higher gross rates of NH₄⁺ production and 441 immobilization (Allen et al., 2015), soil N stocks, ECEC, base saturation; Appendix Table A1) 442 443 and water-holding capacity (Hassler et al., 2015). Nutrient demand of vegetation may not be the dominant control on leaching fluxes, as the vegetation structure of the reference land uses 444 (tree density, basal area, root biomass; Appendix Table A2) even seemed larger in the loam 445 446 than the clay Acrisols. Similarly, the differences in tree species compositions between the loam and clay Acrisol soils (Appendix Table A2) may not have influenced the nutrient leaching 447 fluxes, as supported by the comparable net primary production of the reference land uses 448 449 between soil types (Kotowska et al., 2015). Our findings showed that soil texture was the main factor regulating nutrient leaching losses and soil fertility (e.g., nutrient stocks and N-cycling 450 451 rates) in these highly weathered Acrisol soils.

The influenced of soil texture on soil biochemical characteristics also linked to the 452 leaching losses or, conversely, nutrient retention efficiency. This was shown by the negative 453 454 correlations of annual DON and NO_3^- -N leaching losses with soil base saturation, ECEC and exchangeable Al across all the reference sites. The higher the N and cation leaching (as in the 455 loam Acrisol), the lower were the cation stocks and ECEC in the soil (Appendix Table A1). 456 457 Similarly, the negative correlation of annual NH₄⁺-N leaching losses with soil organic C suggest high retention of NH₄⁺ in the clay Acrisol that has higher soil organic C (Appendix Table A1), 458 higher soil microbial biomass and higher gross rates of NH₄⁺ cycling than in the loam Acrisol 459 (Allen et al., 2015). These all led to the higher N and base cation retention efficiency in clay 460 than in loam Acrisols (Table 5). The positive correlations of N and base cation retention 461

efficiency with soil base saturation, ECEC, organic C and clay content across all the referencesites suggest efficient cycling of nutrients between the soil and vegetation in the clay Acrisol.

464

465 **4.3. Land-use change effects on leaching fluxes and nutrient retention efficiency**

Land-use conversion (i.e., slashing and burning of the previous vegetation) causes a large 466 portion of nutrients in biomass to be lost during burning (Kaufmann et al., 1995; Mackensen et 467 468 al., 1996) and, after the initial pulse of nutrient release from ashes and decomposition, nutrient leaching continuously decreases with time (Klinge et al., 2004). Our smallholder rubber 469 plantations were not fertilized during our study years, in part because the price of rubber was 470 471 low during those years (Clough et al., 2015). Without soil amendments, soil nutrient levels can decrease significantly years after land-use conversion (e.g., decreases in exchangeable bases 472 (Dechert et al. 2004), P availability (Ngoze et al., 2008), soil-N cycling rate and microbial N 473 474 (Allen et al., 2015; Corre et al., 2006; Davidson et al., 2007)). This was evident in our unfertilized rubber plantations with low ionic charge concentrations of soil solutions in the loam 475 476 Acrisol soil (Fig. 1; Table 3). Annual P leaching flux in rubber decreased (Table 4) and was reflected by a decrease in Bray-extractable P compared to forest (Allen et al., 2016). Compared 477 478 to forest, the decrease in annual DOC leaching flux in rubber (Table 4) was mirrored by 479 decreases in microbial C (Allen et al., 2015), litterfall and root production (Kotowska et al., 2015) in the same rubber plantations, and the overall decrease in soil organic C stocks in 480 smallholder rubber plantations in the same study region (van Straaten et al., 2015). Decreases 481 482 in DOC concentrations of soil solutions were possibly the reason why cations in the soil solutions of the rubber plantations were strongly correlated with Cl and only weakly correlated 483 with organic-associated anions (DOC or total S; Appendix Tables A3 and A4). Our results 484 showed that disruption of nutrient cycling between the soil and vegetation brought about by 485 land-use conversion to rubber plantations, combined with the absence of soil amendments, had 486

decreased nutrient leaching (Tables 3 and 4) as well soil nutrient availability (i.e., P stocks,
microbial N, gross N mineralization rates; Allen et al., 2015; Allen et al., 2016).

The most important factor influencing nutrient leaching in the smallholder oil palm 489 plantations was fertilizer application. This was evident by the higher solute concentrations of 490 491 the fertilized area compared to the frond-stacked area and to the other land uses (Fig. 1; Table 3). In the fertilized area, the stronger correlations of dissolved cations with total S and Cl, rather 492 493 than with DOC, were because S and Cl are components of the applied fertilizers (see 2.1). The larger increases in solute concentrations of the fertilized areas in the loam Acrisol than the clay 494 Acrisol soils were attributed to the following: 1) higher fertilization rates of oil palm plantations 495 496 in the loam Acrisol (see 2.1), and 2) its lower clay contents that contributed to its lower waterholding capacity and nutrient retention (Tables 1 and 5). In fertilized areas of the loam Acrisol, 497 the correlations among dissolved NO_3^- , total Al and acidity were likely due to nitrification of 498 499 added N fertilizer and the low acid-buffering capacity of this loam Acrisol soil (i.e., low base saturation; Appendix Table A1; Allen et al. 2016). Soil extractable NO_3^- and NH_4^+ in these 500 501 smallholder plantations are elevated up to six weeks following fertilization (Hassler et al., 2017), during which time NO_3^- is susceptible to leaching. Nitrification-induced acidity may 502 503 have enhanced the Al acid-buffering reaction and led to the increases in dissolved Al and acidity 504 of soil solution (Table 3). Other studies in Indonesia and Malaysia have also reported increases in soil acidity due to N fertilization in oil palm plantations (Anuar et al., 2008; Comte et al., 505 2013). Even though occasional liming is practiced by smallholders in these oil palm plantations, 506 507 soil pH (Appendix Table A1) was still within the Al acid-buffering range (pH 3-5; Van Breemen et al., 1983). The acidic soil water and elevated dissolved Al concentration resulting 508 509 from N fertilization in these oil palm plantations may also have triggered the decrease in mycorrhizal colonization of fine roots and the increase in distorted root tips found at the same 510 sites (Sahner et al., 2015). 511

512	In the fertilized areas of oil palm plantations in the loam Acrisol, increased annual
513	leaching fluxes of Na, total S, Cl, NO ₃ ⁻ , TDN, Ca and Mg (Table 4) were due to applications of
514	Na-, S- and N-containing fertilizers and lime (see 2.1). The leaching losses in our oil palm
515	plantations were lower than those reported for oil palm plantations on Acrisol soils in Nigeria
516	(2.6 g Ca m ⁻² and 0.6 g Mg m ⁻² during a six-month period; Omoti et al., 1983) and Malaysia
517	(0.3-0.6 g N m ⁻² during a five-month period; Tung et al., 2009), and on Andisol soils in Papua
518	New Guinea (3.7-10.3 g N m ⁻² yr ⁻¹ during a fourteen-month period; Banabas et al., 2008), all
519	of which had larger fertilization rates than our smallholders. Moreover, the increased annual
520	DOC fluxes in fertilized areas of oil palm plantations (Table 4) suggests a reduction in the
521	retention of DOC in the soil. This, combined with the decreases in litterfall and root production,
522	harvest export (Kotowska et al., 2015), and decreases in soil CO ₂ emissions (Hassler et al.,
523	2015) from the same oil palm plantations, provided additional support for the decreases in soil
524	organic C stocks in smallholder oil palm plantations in the same study region (van Straaten et
525	al., 2015). Altogether, our results showed the overarching influence of soil texture on nutrient
526	retention or leaching in these converted land uses. This was shown by the positive correlations
527	of annual NH4 ⁺ -N leaching, annual DON leaching and base cation retention efficiency with
528	ECEC, soil organic C and clay content across all sites of the converted land uses.

The fertilization rates in our studied smallholder oil palm plantations were only 2-5 times lower than the nearby large-scale plantations, which were typically 230-260 kg N ha⁻¹ yr⁻¹. Our findings of increased TDN and base cation leaching or decreased retention efficiency, particularly in the loam Acrisol, despite the low fertilization rates (Tables 4 and 5), imply for a need to optimize fertilization rate in large-scale plantations, especially on coarse-texture soils which have low inherent nutrient retention, in order to minimize environmental effect while maintaining production.

537 **5 Conclusions**

538 The low solute concentrations in drainage water of the reference land uses signified low internal inputs of rock-derived nutrients in these highly-weathered soils, and suggest efficient internal 539 cycling of nutrients. Our findings of lower nutrient leaching losses and higher nutrient retention 540 efficiency in the reference land uses on the clay as compared to the loam Acrisol soils supported 541 our first hypothesis, and reflected the influence of soil texture on nutrient retention and water-542 543 holding capacity. The low nutrient leaching losses in the unfertilized rubber plantations and the high leaching in the fertilized oil palm plantations supported our second hypothesis. Reduced 544 P and DOC leaching in rubber plantations signaled reduction in nutrient levels, which may 545 546 influence how long these rubber plantations can remain before conversion to another land use. 547 Sustainability of oil palm plantations must take into account the long-term effect of chronic N fertilization on soil water acidity and Al solubility; the inherently low acid-buffering capacity 548 549 of Acrisol soils implies that the smallholders will be increasingly dependent on lime application, which entails additional capital input. Our results highlight the need to develop soil 550 551 management practices that conserve soil nutrients in unfertilized rubber plantations and increase nutrient retention efficiency in fertilized oil palm plantations. Management practices 552 553 to regulate leaching losses are possibly more pressing for large-scale oil palm plantations, which 554 have 2-5 times higher fertilization rates and may have a larger impact on ground water quality than the smallholders. Process-based models, used to predict yield and associated 555 environmental footprint of these tree cash crop plantations, should reflect the differences in soil 556 management (e.g., absence or low vs. high fertilization rates, weed control) between 557 smallholder and large-scale plantations. For valid large-scale extrapolation, quantification of 558 559 leaching losses in oil palm plantations should not only represent the spatial structure of management practices but also surface landforms, which influence water redistribution (e.g., 560

inclusion of riparian areas), and an improved water budget (e.g., estimates of evapotranspirationfrom inter-rows).

563 *Data availability*

564 Our data are deposited in the EFForTS-IS data repository (https://efforts-is.uni-goettingen.de),

an internal data-exchange platform, which is accessible to EFForTS members only. Based on

data sharing agreement within EFForTS, these data are currently not publicly accessible but

567 will be made available through a written request to the corresponding and senior authors.

568 Author contribution

569 SK, MDC, EV and SRU conceived and designed research. SK carried out field measurements.

570 MDC, EV and SRU supported the field research. SK and HSB modelled water budget with the

571 Expert N water module. SK, MDC and EV analyzed the data. SK, MDC, ALM, OvS and EV
572 wrote the manuscript.

573 *Competing interests*

574 All authors declare no conflict of interest.

575

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Water balance components	Forest	Jungle rubber	Rubber	Oil palm				
(mm yr ⁻¹)			plantations	plantations				
	loam Acrisol	soil (precipitation:	3418 mm yr ⁻¹)					
Evapotranspiration	1384	1224	1077	1027				
Transpiration	1033	815	594	437				
Evaporation	155	213	287	408				
Interception	196	196	196	182				
Water drainage	1483	1487	1544	1614				
Runoff	545	704	800	761				
	clay Acrisol soil (precipitation: 3475 mm yr ⁻¹)							
Evapotranspiration	1622	1271	1114	1071				
Transpiration	1284	861	402	446				
Evaporation	157	242	548	459				
Interception	181	168	164	166				
Water drainage	1117	1268	1280	1311				
Runoff	722	932	1070	1087				

Table 1. Simulated water balance during 2013 in different land uses within the loam and clay

782 Acrisol soils in Jambi, Sumatra, Indonesia.

784	Table 2. Mean (\pm SE, $n = 2$) volume-weighted element concentrations and annual inputs in
785	bulk precipitation, measured bi-weekly to monthly from February to December 2013 in the two
786	landscapes with loam and clay Acrisol soils in Jambi, Sumatra, Indonesia.

Elements	Volume-weigh	nted	Annual input		
	concentration (mg L ⁻¹)		$(g m^{-2} yr^{-1})$		
	loam Acrisol	clay Acrisol	loam Acrisol	clay Acrisol	
Ammonium (NH ₄ ⁺ -N)	0.17 (0.02)	0.20 (0.02)	0.58 (0.06)	0.69 (0.07)	
Nitrate (NO ₃ ⁻ -N)	0.04 (0.02)	0.07 (0.01)	0.13 (0.06)	0.26 (0.04)	
Dissolved organic nitrogen (N)	0.17 (0.01)	0.20 (0.04)	0.58 (0.02)	0.70 (0.14)	
Total dissolved nitrogen (N)	0.38 (0.00)	0.47 (0.07)	1.29 (0.01)	1.64 (0.26)	
Dissolved organic carbon (C)	8.15 (0.19)	7.44 (0.07)	27.84 (0.66)	25.86 (0.25)	
Sodium (Na)	1.84 (0.04)	1.90 (0.18)	6.30 (0.13)	6.61 (0.63)	
Potassium (K)	0.16 (0.04)	0.28 (0.14)	0.55 (0.15)	0.96 (0.49)	
Calcium (Ca)	0.32 (0.02)	0.36 (0.07)	1.09 (0.08)	1.24 (0.24)	
Magnesium (Mg)	0.07 (0.01)	0.09 (0.01)	0.24 (0.05)	0.30 (0.04)	
Total aluminum (Al)	0.02 (0.01)	0.01 (0.00)	0.05 (0.03)	0.04 (0.01)	
Total iron (Fe)	0.01 (0.00)	0.01 (0.00)	0.04 (0.01)	0.03 (0.01)	
Total manganese (Mn)	0.001 (0.00)	0.001 (0.00)	0.003 (0.00)	0.004 (0.00)	
Total phosphorus (P)	0.01 (0.00)	0.02 (0.00)	0.04 (0.01)	0.08 (0.01)	
Total sulfur (S)	0.26 (0.00)	0.30 (0.03)	0.90 (0.01)	1.04 (0.10)	
Total silica (Si)	0.02 (0.01)	0.03 (0.01)	0.06 (0.02)	0.09 (0.03)	
Chloride (Cl)	1.79 (0.25)	1.54 (0.30)	6.11 (0.84)	5.34 (1.06)	

788	Table 3. Mean (\pm SE, $n = 4$, except for oil palm $n = 3$) nutrient concentrations in soil solutions
789	from a depth of 1.5 m in different land uses within the loam and clay Acrisol soils in Jambi,
790	Sumatra, Indonesia. Means followed by different lowercase letters indicate significant
791	differences among land uses within each soil type and different uppercase letters indicate
792	significant differences between soil types for each reference land use (Linear mixed effects
793	models with Fisher's LSD test at $P \le 0.05$, and \dagger at $P \le 0.09$ for marginal significance).

Elements	Forest	Jungle rubber	Rubber	Oil palm	Oil palm frond-		
				fertilized area	stacked area		
	loam Acrisol soil						
pH	4.3 (0.0) _{a†}	4.3 (0.1) _{a†}	4.4 (0.0) _{a†}	4.1 (0.1) _{b†}	4.3 (0.0) _{a†}		
Ammonium	0.2 (0.0) _{A†}	0.3 (0.1)	0.2 (0.0)	0.2 (0.0)	0.2 (0.0)		
$(mg NH_4^+-N L^{-1})$							
Nitrate	0.1 (0.1) _b	0.1 (0.0) _{b A}	0.0 (0.0) c	0.3 (0.2) _a	0.1 (0.0) _b		
(mg NO ₃ ⁻ -N L ⁻¹)							
Dissolved organic N	0.2 (0.0) _{a A†}	0.1 (0.0) b	0.1 (0.0) _b	0.1 (0.0) _{ab}	0.1 (0.0) b		
(mg N L ⁻¹)							
Total dissolved N	0.5 (0.1) _{A†}	0.4 (0.1) _{A†}	0.2 (0.0)	0.6 (0.2)	0.3 (0.0)		
(mg N L ⁻¹)							
Dissolved organic C	3.7 (0.3) _{ab}	4.0 (0.5) _{ab}	3.1 (0.2) _c	4.2 (0.1) _a	3.6 (0.1) _b		
(mg C L ⁻¹)							
Sodium (mg Na L ⁻¹)	3.2 (0.1) _{b A}	2.4 (0.2) _c	2.2 (0.2) _c	7.2 (3.9) _a	2.3 (0.3) c		
Potassium (mg K L ⁻¹)	0.4 (0.0)	0.2 (0.1)	0.3 (0.1)	0.4 (0.1)	0.4 (0.1)		
Calcium (mg Ca L ⁻¹)	0.8 (0.0) b	0.7 (0.1) c	0.7 (0.1) c	2.7 (0.9) a	0.7 (0.1) c		
Magnesium	0.3 (0.0) _{b A}	0.2 (0.0) _c	0.3 (0.1) _b	0.5 (0.1) a	0.2 (0.0) c		
(mg Mg L ⁻¹)							

Total aluminum	0.4 (0.1) _{b A}	0.2 (0.0) c	0.3 (0.0) _b	1.2 (0.7) _a	0.1 (0.0) c
$(mg Al L^{-1})$					
Total iron (mg Fe L ⁻¹)	0.2 (0.1) _{A†}	0.0 (0.0)	0.0 (0.0)	0.0 (0.0)	0.1 (0.1)
Total manganese	0.02 (0.00)	0.01 (0.00)	0.01 (0.00)	0.01 (0.00)	0.01 (0.00) _B
(mg Mn L ⁻¹)					
Total phosphorus	0.008 (0.0) a†	0.004~(0.0) b†	0.003 (0.0) _{c†}	0.005 (0.0) _{ab†}	0.005 (0.0) _{ab†}
(mg P L ⁻¹)					
Total sulfur (mg S L ⁻¹)	0.16 (0.00) _{a†}	0.14~(0.00) bc†	0.10 (0.00) _{c†}	0.14 (0.00) _{ab†}	0.12 (0.00) _{b†}
Total silica (mg Si L ⁻¹)	0.5 (0.1)	0.3 (0.1) _{B†}	0.2 (0.1)	0.3 (0.1)	0.2 (0.0)
Chloride (mg Cl L ⁻¹)	8.9 (0.8) _{b A†}	6.6 (0.8) _c	6.7 (0.6) _c	21.0 (2.7) _a	6.2 (0.8) _c
		clay Acriso	l soil		
рН	4.3 (0.1) c	4.4 (0.1) bc	4.4 (0.0) c	4.6 (0.1) ab	4.6 (0.1) a
Ammonium	0.2 (0.0) в†	0.1 (0.0)	0.1 (0.0)	0.2 (0.0)	0.1 (0.0)
$(mg NH_4^+-N L^{-1})$					
Nitrate	0.1 (0.0)	0.0 (0.0) b	0.2 (0.1)	0.9 (0.9)	0.0 (0.0)
$(mg NO_3^N L^{-1})$					
Dissolved organic N	0.1 (0.0) _{a†B†}	0.1 (0.0) _a †	0.1 (0.0) _{ab} †	0.0~(0.0) _b †	0.0~(0.0) b [†]
(mg N L ⁻¹)					
Total dissolved N	0.3 (0.0) _{B†}	0.2 (0.0) _{B†}	0.4 (0.1)	1.1 (0.9)	0.2 (0.0)
(mg N L ⁻¹)					
Dissolved organic C	3.3 (0.4)	4.0 (0.3)	2.9 (0.1)	4.8 (0.9)	4.4 (1.1)
(mg C L ⁻¹)					
Sodium (mg Na L ⁻¹)	2.4 (0.2) _{bc B}	2.5 (0.1) b	2.0 (0.1) c	4.6 (1.2) _a	2.5 (0.5) _{bc}
Potassium (mg K L ⁻¹)	0.3 (0.0)	0.3 (0.1)	0.3 (0.0)	0.4 (0.1)	0.2 (0.1)
Calcium (mg Ca L ⁻¹)	0.7 (0.1)	0.7 (0.0)	0.7 (0.1)	0.8 (0.2)	0.5 (0.1)

Magnesium	0.3 (0.0) _B	0.3 (0.0)	0.3 (0.0)	0.4 (0.1)	0.2 (0.1)
(mg Mg L ⁻¹)					
Total aluminum	0.2 (0.0) _B	0.2 (0.1)	0.3 (0.1)	0.2 (0.1)	0.1 (0.0)
(mg Al L ⁻¹)					
Total iron (mg Fe L ⁻¹)	0.0~(0.0) b†B†	0.0~(0.0) b†	0.0~(0.0) _{b†}	0.0~(0.0) b†	0.1 (0.0) _{a†}
Total manganese	0.01 (0.00)	0.01 (0.00)	0.01 (0.00)	0.08 (0.10)	0.02 (0.00)
(mg Mn L ⁻¹)					
Total phosphorus	0.010 (0.0)	0.004 (0.0)	0.004 (0.0)	0.004 (0.0)	0.010 (0.0)
$(mg P L^{-1})$					
Total sulfur (mg S L ⁻¹)	0.15 (0.00)	0.11 (0.00)	0.11 (0.00)	0.13 (0.00)	0.12 (0.00)
Total silica (mg Si L ⁻¹)	0.4 (0.0)	0.6 (0.1) _{A†}	0.3 (0.0)	1.0 (0.4)	0.7 (0.2)
Chloride (mg Cl L ⁻¹)	6.4 (0.6) _{B†}	6.8 (0.9)	5.7 (0.8)	7.2 (2.1)	4.6 (0.8)

795	Table 4. Mean (\pm SE, $n = 4$, except for oil palm $n = 3$) annual (2013) nutrient leaching fluxes
796	measured at a depth of 1.5 m in different land uses within the loam and clay Acrisol soils in
797	Jambi, Sumatra, Indonesia. Means followed by different lowercase letters indicate significant
798	differences among land uses within each soil type and different uppercase letters indicate
799	significant differences between soil types for each reference land use (Linear mixed effects
800	models with Fisher's LSD test at $P \le 0.05$, and \dagger at $P \le 0.09$ for marginal significance).

Elements	Forest	Jungle rubber	Rubber	Oil palm	Oil palm frond-
				fertilized area	stacked area
-		loam Acris	ol soil		
Ammonium	0.3 (0.0) _{ab A†}	0.5 (0.3) _a	0.2 (0.01) bc	0.3 (0.0) _{ab}	0.2 (0.0) c
$(g NH_4^+-N m^{-2} yr^{-1})$					
Nitrate	0.1 (0.1) ab	0.1 (0.1) ab A	0.0 (0.0) b	0.6 (0.3) _a	0.1 (0.0) ab
(g NO ₃ ⁻ -N m ⁻² yr ⁻¹)					
Dissolved organic N	0.2 (0.0) _{A†}	0.1 (0.0)	0.1 (0.0)	0.2 (0.1)	0.1 (0.0)
$(g N m^{-2} yr^{-1})$					
Total dissolved N	$0.6 (0.1) _{ab^{\dagger}A^{\dagger}}$	† 0.8 (0.3) _{ab} †	0.4~(0.0) _b †	1.1 (0.3) _{a†}	0.4~(0.1) _b †
(g N m ⁻² yr ⁻¹)					
Dissolved organic C	4.2 (0.5) bc	6.2 (1.5) _{ab}	3.9 (0.2) _c	7.3 (0.2) _a	4.2 (0.4) bc
$(g C m^{-2} yr^{-1})$					
Sodium	3.8 (0.4) _{b A}	3.7 (0.8) b	3.1 (0.3) b	13.1 (7.6) a	3.1 (0.5) b
(g Na m ⁻² yr ⁻¹)					
Potassium	0.4 (0.1)	0.4 (0.2)	0.4 (0.1)	0.7 (0.2)	0.4 (0.1)
$(g K m^{-2} yr^{-1})$					
Calcium	1.0 (0.1) _{b A}	1.2 (0.3) _b	0.9 (0.1) _b	4.6 (1.3) _a	1.0 (0.2) _b
(g Ca m ⁻² yr ⁻¹)					

Magnesium	0.4 (0.0) _{b A}	0.4 (0.1) _b	0.4 (0.1) _b	0.9 (0.2) _a	0.3 (0.1) _b
(g Mg m ⁻² yr ⁻¹)					
Total aluminum	0.4 (0.1) _{b A}	0.3 (0.1) b	0.4 (0.0) _b	2.3 (1.3) a	0.2 (0.0) b
(g Al m ⁻² yr ⁻¹)					
Total iron	0.20 (0.10)	0.02 (0.01)	0.03 (0.01)	0.04 (0.00)	0.10 (0.10)
$(g Fe m^{-2} yr^{-1})$					
Total manganese	0.02 (0.01)	0.03 (0.02)	0.01 (0.01)	0.03 (0.00)	0.01 (0.00)
(g Mn m ⁻² yr ⁻¹)					
Total phosphorus	0.01 (0.00) _{a†}	0.01 (0.00) abc†	0.00 (0.00) _{c†}	0.01 (0.0) _{ab†}	0.01 (0.00) _{bc†}
$(g P m^{-2} yr^{-1})$					
Total sulfur	0.20 (0.00) ab	0.20 (0.10) _{ab}	0.13 (0.01) _b	0.24 (0.0) a	0.15 (0.0) ab
$(g S m^{-2} yr^{-1})$					
Total silica	0.7 (0.2) _{A†}	0.6 (0.3)	0.4 (0.1)	0.4 (0.1)	0.3 (0.1)
$(g Si m^{-2} yr^{-1})$					
Chloride	10.5 (0.9) _{b A}	11.5 (2.4) _b	9.1 (0.6) _b	38.0 (6.7) _a	7.8 (1.2) _b
$(g \operatorname{Cl} m^{-2} yr^{-1})$					

	clay Acrisol soil				
Ammonium	0.2 (0.0) _{B†}	0.2 (0.0)	0.2 (0.0)	0.2 (0.0)	0.2 (0.0)
$(g NH_4^+-N m^{-2} yr^{-1})$					
Nitrate	0.1 (0.1)	0.0 (0.0) _B	0.3 (0.2)	1.1 (1.1)	0.0 (0.0)
$(g NO_3^N m^{-2} yr^{-1})$					
Dissolved organic N	0.1 (0.0) _{B†}	0.1 (0.0)	0.1 (0.0)	0.1 (0.0)	0.1 (0.0)
$(g N m^{-2} yr^{-1})$					
Total dissolved N	0.3 (0.1) _{В†}	0.3 (0.0)	0.6 (0.2)	1.4 (1.1)	0.3 (0.0)
$(g N m^{-2} yr^{-1})$					

(g C m² yr¹)2.5 (0.4) b B3.2 (0.3) b2.5 (0.1) b6.3 (1.8) a3.3 (0.6) b(g Na m² yr¹) <th>Dissolved organic C</th> <th>3.4 (0.4) _c</th> <th>5.4 (0.7) _{ab}</th> <th>3.6 (0.2) _{bc}</th> <th>6.2 (1.4) _a</th> <th>5.6 (1.0) _{ab}</th>	Dissolved organic C	3.4 (0.4) _c	5.4 (0.7) _{ab}	3.6 (0.2) _{bc}	6.2 (1.4) _a	5.6 (1.0) _{ab}
(g Na m² yr¹)No. 2 (0.1)0.3 (0.1)0.3 (0.1)0.5 (0.1)0.2 (0.1)(g K m² yr¹) $$	$(g C m^{-2} yr^{-1})$					
Potassium $0.3 (0.0)$ $0.3 (0.1)$ $0.3 (0.1)$ $0.5 (0.1)$ $0.2 (0.1)$ (g K m ² yr ¹) $0.7 (0.1)$ b $0.9 (0.0)$ $0.8 (0.1)$ $1.0 (0.2)$ $0.7 (0.1)$ (g Ca m ² yr ¹) $0.2 (0.0)$ b $0.3 (0.0)$ b $0.3 (0.0)$ b $0.6 (0.1)$ a $0.2 (0.1)$ b(g Mg m ⁻² yr ⁻¹) $0.2 (0.0)$ b $0.3 (0.0)$ b $0.3 (0.0)$ b $0.6 (0.1)$ a $0.2 (0.1)$ b(g Mg m ⁻² yr ⁻¹) $0.2 (0.0)$ b $0.2 (0.1)$ $0.3 (0.1)$ $0.6 (0.1)$ a $0.2 (0.1)$ b(g Al m ⁻² yr ⁻¹) $0.2 (0.0)$ b $0.2 (0.1)$ $0.3 (0.1)$ $0.3 (0.1)$ $0.1 (0.0)$ (g Fe m ² yr ⁻¹) $0.1 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.00)$ (g Mn m ⁻² yr ⁻¹) $0.16 (0.0)$ ab $0.16 (0.0)$ b $0.16 (0.0)$ b $0.17 (0.0)$ a $0.17 (0.0)$ ab(g Mn m ⁻² yr ⁻¹) $0.16 (0.0)$ ab $0.15 (0.0)$ ab $0.14 (0.0)$ b $0.17 (0.0)$ a $0.17 (0.0)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ bb $0.7 (0.1)$ ab $0.3 (0.0)$ b $1.3 (0.6)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ bb $0.7 (0.1)$ ab $0.3 (0.0)$ b $1.3 (0.6)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ bb $0.7 (0.1)$ ab $0.3 (0.0)$ b $1.3 (0.6)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.15 (0.0)$ b $0.3 (0.0)$ b $0.3 (0.0)$ b $0.3 (0.0)$ b $0.3 (0.0)$ b(h and a b an ab	Sodium	2.5 (0.4) _{b B}	3.2 (0.3) b	2.5 (0.1) b	6.3 (1.8) _a	3.3 (0.6) b
(g K m² yr¹)<	(g Na m ⁻² yr ⁻¹)					
Calcium $0.7 (0.1)$ B $0.9 (0.0)$ $0.8 (0.1)$ $1.0 (0.2)$ $0.7 (0.1)$ (g Ca m² yr¹) $0.2 (0.0)$ B $0.3 (0.0)$ b $0.3 (0.0)$ D $0.6 (0.1)$ a $0.2 (0.1)$ b(g Mg m² yr¹) $0.2 (0.0)$ B $0.2 (0.1)$ $0.3 (0.1)$ $0.6 (0.1)$ a $0.2 (0.1)$ bTotal aluminum $0.2 (0.0)$ B $0.2 (0.1)$ $0.3 (0.1)$ $0.6 (0.1)$ a $0.2 (0.1)$ b(g Mg m² yr¹) $0.2 (0.0)$ B $0.2 (0.1)$ $0.3 (0.1)$ $0.3 (0.1)$ $0.3 (0.1)$ (g Al m² yr¹) $0.2 (0.00)$ $0.03 (0.00)$ $0.02 (0.00)$ $0.01 (0.00)$ $0.06 (0.05)$ (g Fe m² yr¹) $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.09 (0.07)$ $0.02 (0.00)$ (g Mn m² yr¹) $0.11 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g Mn m² yr¹) $0.11 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g Mn m² yr¹) $0.16 (0.0)$ $0.15 (0.0)$ $0.14 (0.0)$ $0.17 (0.0)$ $0.17 (0.0)$ (g S m² yr¹) $0.16 (0.0)$ $0.15 (0.0)$ $0.3 (0.0)$ $1.3 (0.6)$ $0.8 (0.3)$ (g S m² yr¹) $0.3 (0.1)$ $0.7 (0.1)$ $0.3 (0.0)$ $1.3 (0.6)$ $0.8 (0.3)$ (g S m² yr¹) $0.10 (0.3)$ $0.2 (1.3)$ $0.3 (0.0)$ $0.3 (0.0)$ $0.3 (0.0)$	Potassium	0.3 (0.0)	0.3 (0.1)	0.3 (0.1)	0.5 (0.1)	0.2 (0.1)
(g Ca m ⁻² yr ⁻¹)Nagnesium $0.2 (0.0)_{bB}$ $0.3 (0.0)_{b}$ $0.3 (0.0)_{b}$ $0.6 (0.1)_{a}$ $0.2 (0.1)_{b}$ (g Mg m ⁻² yr ⁻¹) $0.2 (0.0)_{B}$ $0.2 (0.1)$ $0.3 (0.1)$ $0.6 (0.1)_{a}$ $0.2 (0.1)_{b}$ Total aluminum $0.2 (0.0)_{B}$ $0.2 (0.1)$ $0.3 (0.1)$ $0.3 (0.1)$ $0.1 (0.0)$ (g Al m ⁻² yr ⁻¹) $0.02 (0.00)$ $0.03 (0.00)$ $0.02 (0.00)$ $0.01 (0.0)$ $0.01 (0.0)$ $0.06 (0.05)$ (g Fe m ⁻² yr ⁻¹) $0.10 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.09 (0.07)$ $0.02 (0.00)$ (g Mn m ⁻² yr ⁻¹) $0.10 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g P m ⁻² yr ⁻¹) $0.16 (0.0)_{ab}$ $0.15 (0.0)_{ab}$ $0.14 (0.0)_{b}$ $0.17 (0.0)_{a}$ $0.17 (0.0)_{ab}$ (g S m ⁻² yr ⁻¹) $0.3 (0.1) b_{B^{\dagger}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ (g S m ⁻² yr ⁻¹) $0.16 (0.0)_{ab}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	$(g K m^{-2} yr^{-1})$					
Magnesium $0.2 (0.0) bB$ $0.3 (0.0) b$ $0.3 (0.0) b$ $0.6 (0.1) a$ $0.2 (0.1) b$ $(g Mg m^2 yr^{-1})$ $0.2 (0.0) B$ $0.2 (0.1)$ $0.3 (0.1)$ $0.3 (0.1)$ $0.1 (0.0)$ $(g Al m^{-2} yr^{-1})$ $0.2 (0.00)$ $0.03 (0.00)$ $0.02 (0.00)$ $0.01 (0.0)$ $0.06 (0.05)$ $(g Fe m^{-2} yr^{-1})$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.00)$ $(g Fe m^{-2} yr^{-1})$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.00)$ $(g Mn m^2 yr^{-1})$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ $(g Mn m^2 yr^{-1})$ $0.16 (0.0) ab$ $0.15 (0.0) ab$ $0.14 (0.0) b$ $0.17 (0.0) a$ $0.17 (0.0) ab$ $(g S m^{-2} yr^{-1})$ $0.16 (0.0) ab$ $0.15 (0.0) ab$ $0.14 (0.0) b$ $0.17 (0.0) a$ $0.17 (0.0) ab$ $(g S m^{-2} yr^{-1})$ $0.3 (0.1) bBT$ $0.7 (0.1) ab$ $0.3 (0.0) b$ $1.3 (0.6) a$ $0.8 (0.3) ab$ $(g S i m^{-2} yr^{-1})$ $0.16 (0.0) BB $ $0.1 (0.0) b$ $0.1 (0.0) b$ $0.13 (0.0) b$ $0.15 (0.0) ab$ $(g S i m^{-2} yr^{-1})$ $0.15 (0.0) BB $ $0.3 (0.0) b$ $0.3 (0.0) b$ $0.3 (0.0) ab$ $0.3 (0.0) b$ $(0.10 m^{-1}) bBT $ $0.7 (0.1) ab$ $0.3 (0.0) b$ $0.3 (0.0) b$ $0.5 (0.6) b$	Calcium	0.7 (0.1) _B	0.9 (0.0)	0.8 (0.1)	1.0 (0.2)	0.7 (0.1)
(g Mg m² yr¹)	(g Ca m ⁻² yr ⁻¹)					
Total aluminum $0.2 (0.0)$ B $0.2 (0.1)$ $0.3 (0.1)$ $0.3 (0.1)$ $0.1 (0.0)$ (g Al m ⁻² yr ⁻¹) $0.02 (0.00)$ $0.03 (0.00)$ $0.02 (0.00)$ $0.01 (0.0)$ $0.06 (0.05)$ (g Fe m ⁻² yr ⁻¹) $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.09 (0.07)$ $0.02 (0.00)$ (g Mn m ⁻² yr ⁻¹) $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g P m ⁻² yr ⁻¹) $0.16 (0.0)$ ab $0.15 (0.0)$ ab $0.14 (0.0)$ b $0.17 (0.0)$ a $0.17 (0.0)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ b B† $0.7 (0.1)$ ab $0.3 (0.0)$ b $1.3 (0.6)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ b B† $0.7 (0.1)$ ab $0.9 (1.0)$ $0.8 (0.0)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ b B† $0.7 (0.1)$ ab $0.9 (1.0)$ $0.8 (0.0)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ b B† $0.7 (0.1)$ ab $0.9 (1.0)$ $0.8 (0.0)$ a $0.8 (0.3)$ ab(g S m ⁻² yr ⁻¹) $0.3 (0.1)$ b B† $0.7 (0.1)$ ab $0.9 (1.0)$ $0.8 (0.0)$ a $0.8 (0.3)$ ab	Magnesium	0.2 (0.0) _{b B}	0.3 (0.0) _b	0.3 (0.0) _b	0.6 (0.1) _a	0.2 (0.1) _b
$ (g Al m^{-2} yr^{-1}) $ Total iron $ 0.02 (0.00) 0.03 (0.00) 0.02 (0.00) 0.01 (0.0) 0.06 (0.05) $ $ (g Fe m^{-2} yr^{-1}) $ Total manganese $ 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.09 (0.07) 0.02 (0.00) $ $ (g Mn m^{-2} yr^{-1}) $ Total phosphorus $ 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.02 (0.01) $ $ (g P m^{-2} yr^{-1}) $ Total sulfur $ 0.16 (0.0)_{ab} 0.15 (0.0)_{ab} 0.14 (0.0)_{b} 0.17 (0.0)_{a} 0.17 (0.0)_{ab} $ $ (g S m^{-2} yr^{-1}) $ Total silica $ 0.3 (0.1)_{bB^{\dagger}} 0.7 (0.1)_{ab} 0.3 (0.0)_{b} 1.3 (0.6)_{a} 0.8 (0.3)_{ab} $ $ (g S i m^{-2} yr^{-1}) $ Chloride $ 6.0 (0.3)_{B} 8.2 (1.3) 6.9 (1.0) 9.8 (3.0) 5.6 (0.6) $	$(g Mg m^{-2} yr^{-1})$					
Total iron $0.02 (0.00)$ $0.03 (0.00)$ $0.02 (0.00)$ $0.01 (0.0)$ $0.06 (0.05)$ (g Fe m ⁻² yr ⁻¹)Total manganese $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.09 (0.07)$ $0.02 (0.00)$ (g Mn m ⁻² yr ⁻¹) $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g P m ⁻² yr ⁻¹) $0.16 (0.0)_{ab}$ $0.15 (0.0)_{ab}$ $0.14 (0.0)_{b}$ $0.17 (0.0)_{a}$ $0.17 (0.0)_{ab}$ (g S m ⁻² yr ⁻¹) $0.3 (0.1)_{b B^{\dagger}}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ (g Si m ⁻² yr ⁻¹) $0.16 (0.0)_{ab}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	Total aluminum	0.2 (0.0) _B	0.2 (0.1)	0.3 (0.1)	0.3 (0.1)	0.1 (0.0)
(g Fe m-2 yr-1) Total manganese 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.09 (0.07) 0.02 (0.00) (g Mn m ⁻² yr ⁻¹) Total phosphorus 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.01 (0.00) 0.02 (0.01) (g P m ⁻² yr ⁻¹) Total sulfur 0.16 (0.0) ab 0.15 (0.0) ab 0.14 (0.0) b 0.17 (0.0) a 0.17 (0.0) ab (g S m ⁻² yr ⁻¹) Total silica 0.3 (0.1) b B† 0.7 (0.1) ab 0.3 (0.0) b 1.3 (0.6) a 0.8 (0.3) ab (g Si m ⁻² yr ⁻¹) Chloride 6.0 (0.3) B 8.2 (1.3) 6.9 (1.0) 9.8 (3.0) 5.6 (0.6)	$(g Al m^{-2} yr^{-1})$					
Total manganese $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.09 (0.07)$ $0.02 (0.00)$ $(g Mn m^{-2} yr^{-1})$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ $(g P m^{-2} yr^{-1})$ $0.16 (0.0)_{ab}$ $0.15 (0.0)_{ab}$ $0.14 (0.0)_{b}$ $0.17 (0.0)_{a}$ $0.17 (0.0)_{ab}$ $(g S m^{-2} yr^{-1})$ $0.3 (0.1)_{b B^{\dagger}}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ $(g S m^{-2} yr^{-1})$ $0.16 (0.0)_{b}$ $0.17 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ $(g S m^{-2} yr^{-1})$ $0.10 (0.0)_{b}$ $0.9 (1.0)_{b}$ $0.8 (0.3)_{a}$ $0.8 (0.3)_{a}$	Total iron	0.02 (0.00)	0.03 (0.00)	0.02 (0.00)	0.01 (0.0)	0.06 (0.05)
(g Mn m ⁻² yr ⁻¹) $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g P m ⁻² yr ⁻¹) $0.16 (0.0)_{ab}$ $0.15 (0.0)_{ab}$ $0.14 (0.0)_{b}$ $0.17 (0.0)_{a}$ $0.17 (0.0)_{ab}$ (g S m ⁻² yr ⁻¹) $0.3 (0.1)_{b B^{\dagger}}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ (g Si m ⁻² yr ⁻¹) $0.01 (0.03)_{B}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	$(g \text{ Fe } m^{-2} \text{ yr}^{-1})$					
Total phosphorus $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.01 (0.00)$ $0.02 (0.01)$ (g P m ⁻² yr ⁻¹) $0.16 (0.0)_{ab}$ $0.15 (0.0)_{ab}$ $0.14 (0.0)_{b}$ $0.17 (0.0)_{a}$ $0.17 (0.0)_{ab}$ (g S m ⁻² yr ⁻¹) $0.3 (0.1)_{b B^{\dagger}}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ (g Si m ⁻² yr ⁻¹) $0.01 (0.03)_{B}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	Total manganese	0.01 (0.00)	0.01 (0.00)	0.01 (0.00)	0.09 (0.07)	0.02 (0.00)
(g P m ⁻² yr ⁻¹) $0.16 (0.0)_{ab} = 0.15 (0.0)_{ab} = 0.14 (0.0)_{b} = 0.17 (0.0)_{a} = 0.17 (0.0)_{ab}$ Total sulfur $0.16 (0.0)_{ab} = 0.15 (0.0)_{ab} = 0.14 (0.0)_{b} = 0.17 (0.0)_{a} = 0.17 (0.0)_{ab}$ (g S m ⁻² yr ⁻¹) $0.3 (0.1)_{b B^{\dagger}} = 0.7 (0.1)_{ab} = 0.3 (0.0)_{b} = 1.3 (0.6)_{a} = 0.8 (0.3)_{ab}$ (g Si m ⁻² yr ⁻¹) $0.3 (0.0)_{b} = 8.2 (1.3) = 0.9 (1.0) = 9.8 (3.0) = 5.6 (0.6)$	$(g \operatorname{Mn} m^{-2} yr^{-1})$					
Total sulfur $0.16 (0.0)_{ab}$ $0.15 (0.0)_{ab}$ $0.14 (0.0)_{b}$ $0.17 (0.0)_{a}$ $0.17 (0.0)_{ab}$ $(g S m^{-2} yr^{-1})$ $0.3 (0.1)_{b B^{\dagger}}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ $(g Si m^{-2} yr^{-1})$ $6.0 (0.3)_{B}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	Total phosphorus	0.01 (0.00)	0.01 (0.00)	0.01 (0.00)	0.01 (0.00)	0.02 (0.01)
$(g \ S \ m^{-2} \ yr^{-1})$ $0.3 \ (0.1) \ {}_{b \ B^{\dagger}}$ $0.7 \ (0.1) \ {}_{ab}$ $0.3 \ (0.0) \ {}_{b}$ $1.3 \ (0.6) \ {}_{a}$ $0.8 \ (0.3) \ {}_{ab}$ $(g \ Si \ m^{-2} \ yr^{-1})$ $-6.0 \ (0.3) \ {}_{B}$ $8.2 \ (1.3)$ $6.9 \ (1.0)$ $9.8 \ (3.0)$ $5.6 \ (0.6)$	$(g P m^{-2} yr^{-1})$					
Total silica $0.3 (0.1)_{b B^{\dagger}}$ $0.7 (0.1)_{ab}$ $0.3 (0.0)_{b}$ $1.3 (0.6)_{a}$ $0.8 (0.3)_{ab}$ (g Si m ⁻² yr ⁻¹)Chloride $6.0 (0.3)_{B}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	Total sulfur	0.16~(0.0) ab	0.15 (0.0) ab	0.14 (0.0) _b	0.17 (0.0) a	0.17 (0.0) _{ab}
(g Si m ⁻² yr ⁻¹)Chloride $6.0 (0.3)_{B}$ $8.2 (1.3)$ $6.9 (1.0)$ $9.8 (3.0)$ $5.6 (0.6)$	$(g S m^{-2} yr^{-1})$					
Chloride $6.0 (0.3)_{B} 8.2 (1.3) 6.9 (1.0) 9.8 (3.0) 5.6 (0.6)$	Total silica	0.3~(0.1) b B†	0.7 (0.1) _{ab}	0.3 (0.0) b	1.3 (0.6) a	0.8 (0.3) _{ab}
	$(g Si m^{-2} yr^{-1})$					
$(g Cl m^{-2} yr^{-1})$	Chloride	6.0 (0.3) _B	8.2 (1.3)	6.9 (1.0)	9.8 (3.0)	5.6 (0.6)
	$(g \operatorname{Cl} m^{-2} yr^{-1})$					

802	Table 5. Mean (\pm SE, n = 4, except for oil palm $n = 3$) nitrogen and base cation retention
803	efficiency in soils under different land uses within the loam and clay Acrisol soils in Jambi,
804	Sumatra, Indonesia. Mean followed by different lower case letters indicate significant
805	differences among land uses within each soil type and different upper case letters indicate
806	significant differences between soil types for each reference land use (Linear mixed effects
807	models with Fisher's LSD test at $P \le 0.05$, and † at $P = 0.07$ for marginal significance).

Characteristic	Forest	Jungle rubber	Rubber	Oil palm
	1	oam Acrisol soil		
N retention efficiency	0.997 (0.000) _{a B}	0.996 (0.001) _{a B†}	0.998 (0.000) _a	0.995 (0.001) _b
(mg N m ⁻² d ⁻¹ /mg N				
$m^{-2} d^{-1}$)				
Base cation retention	0.455 (0.094) _{a† B}	0.591 (0.088) _{a† B†}	0.699 (0.08259) _{a†}	0.280 (0.128) _{b†}
efficiency (mol _{charge}				
$m^{-2} yr^{-1} / mol_{charge} m^{-2}$)				
	(clay Acrisol soil		
N retention efficiency	0.999 (0.000) _A	0.999 (0.000) _{A†}	0.997 (0.001)	0.998 (0.001)
$(mg N m^{-2} d^{-1}/mg N$				
$m^{-2} d^{-1}$)				
Base cation retention	0.812 (0.084) _A	0.852 (0.083) _{A†}	0.841 (0.025)	0.894 (0.028)
efficiency (molcharge				
$m^{-2} yr^{-1} / mol_{charge} m^{-2}$)				

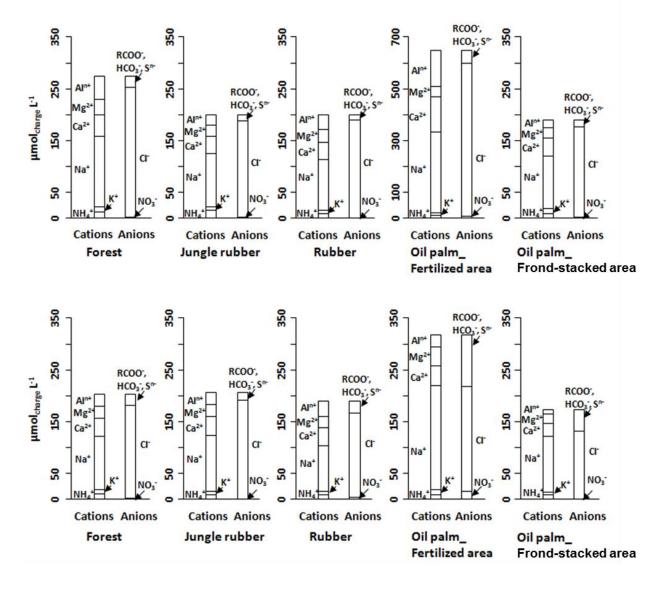


Figure 1. Partial cation-anion charge balance of the major solutes (with concentrations >0.03 mg L^{-1}) in soil water at a depth of 1.5 m in different land uses within the loam (top panel) and clay (bottom panel) Acrisol soils in Jambi, Sumatra, Indonesia. The y-axis scale of the oil palm fertilized area in the loam Acrisol soil is twice than the other land uses.

814 Appendix A. Soil and vegetation characteristics, and Pearson correlations among solute

815 concentrations in each land use within each soil type

Table A1. Soil characteristics in the top 0.1 m of soil (except for clay content, which is for 1-2 m) in different land uses within the loam and clay Acrisol soils in Jambi, Sumatra, Indonesia. Mean (\pm SE, n = 4, except for clay content n = 3) followed by different lower case letters indicate significant differences among land uses within each soil type and different upper case letters indicate significant differences between soil types for each reference land use (Linear mixed effects models with Fisher's LSD test at $P \le 0.05$, and † at $P \le 0.09$ for marginal significance).

822 These soil characteristics were reported by Allen et al. (2015).

Forest	Jungle rubber	Rubber	Oil palm
		plantation	plantation
loam Ao	crisol soil		
1.0 (0.04) _{ab}	0.9 (0.03) _{b A}	1.1 (0.1) a	1.1 (0.1) a
4.3 (0.04) _{b†}	4.3 (0.03) _{b† B}	4.5 (0.1) _{ab†}	4.5 (0.1) _{a†}
2.6 (0.2)	2.7 (0.3) в	2.0 (0.3)	1.8 (0.2)
182.9 (10.8)	186.1 (11.0) в	172.6 (23.8)	145.0 (13.5)
14.3 (0.2) a	13.7 (0.8) a	11.7 (0.7) b	12.5 (0.5) ab
44.8 (5.0)	40.6 (7.6) в	46.0 (5.4)	39.5 (7.9)
10.6 (0.5) _{b† B}	16.0 (2.2) _{ab†}	21.1 (7.5) _{ab†}	27.9 (5.4) _{a†}
3.3 (0.3)	2.6 (0.2) _B	3.4 (0.8)	2.1 (0.8)
0.5 (0.1) _{c B}	1.5 (0.2) _{b B}	1.4 (0.1) _b	3.9 (1.1) _a
5.5 (2.0)	6.9 (0.8) _{B†}	14.5 (7.1)	18.5 (7.4)
1.8 (0.1)	2.0 (0.3) _B	3.4 (1.4)	1.7 (0.9)
33.1 (3.5)	29.6 (6.6) _B	30.7 (4.3)	23.5 (2.7)
	$\begin{array}{c} \text{loam Ac} \\ \hline 1.0 \ (0.04) \ _{ab} \\ \hline 4.3 \ (0.04) \ _{b\dagger} \\ \hline 2.6 \ (0.2) \\ \hline 182.9 \ (10.8) \\ \hline 14.3 \ (0.2) \ _{a} \\ \hline 44.8 \ (5.0) \\ \hline 10.6 \ (0.5) \ _{b\dagger} \ _{B} \\ \hline 3.3 \ (0.3) \\ \hline 0.5 \ (0.1) \ _{c} \ _{B} \\ \hline 5.5 \ (2.0) \\ \hline 1.8 \ (0.1) \\ \end{array}$	Ioam Acrisol soil $1.0 (0.04)_{ab}$ $0.9 (0.03)_{bA}$ $4.3 (0.04)_{b\dagger}$ $4.3 (0.03)_{b\dagger B}$ $2.6 (0.2)$ $2.7 (0.3)_{B}$ $182.9 (10.8)$ $186.1 (11.0)_{B}$ $14.3 (0.2)_{a}$ $13.7 (0.8)_{a}$ $44.8 (5.0)$ $40.6 (7.6)_{B}$ $10.6 (0.5)_{b\dagger B}$ $16.0 (2.2)_{ab\dagger}$ $3.3 (0.3)$ $2.6 (0.2)_{B}$ $0.5 (0.1)_{cB}$ $1.5 (0.2)_{bB}$ $5.5 (2.0)$ $6.9 (0.8)_{B\dagger}$ $1.8 (0.1)$ $2.0 (0.3)_{B}$	plantationloam Acrisol soil $1.0 (0.04)_{ab}$ $0.9 (0.03)_{bA}$ $1.1 (0.1)_{a}$ $4.3 (0.04)_{b\dagger}$ $4.3 (0.03)_{b\dagger B}$ $4.5 (0.1)_{ab\dagger}$ $2.6 (0.2)$ $2.7 (0.3)_{B}$ $2.0 (0.3)$ $182.9 (10.8)$ $186.1 (11.0)_{B}$ $172.6 (23.8)$ $14.3 (0.2)_{a}$ $13.7 (0.8)_{a}$ $11.7 (0.7)_{b}$ $44.8 (5.0)$ $40.6 (7.6)_{B}$ $46.0 (5.4)$ $10.6 (0.5)_{b\dagger B}$ $16.0 (2.2)_{ab\dagger}$ $21.1 (7.5)_{ab\dagger}$ $3.3 (0.3)$ $2.6 (0.2)_{B}$ $3.4 (0.8)$ $0.5 (0.1)_{cB}$ $1.5 (0.2)_{bB}$ $1.4 (0.1)_{b}$ $5.5 (2.0)$ $6.9 (0.8)_{B\dagger}$ $14.5 (7.1)$ $1.8 (0.1)$ $2.0 (0.3)_{B}$ $3.4 (1.4)$

Iron (g Fe m ⁻²)	0.8 (0.1) _{a B}	0.3 (0.02) _{bc B}	0.3 (0.1) c	0.5 (0.02) _{ab}
Manganese (g Mn m ⁻²)	0.3 (0.1)	0.4 (0.2) _B	0.8 (0.3)	0.5 (0.2)
Bray-extractable phosphorus	0.5 (0.1) _B	0.7 (0.1)	0.5 (0.1)	0.8 (0.1)
(g P m ⁻²)				
Clay at 1.0-1.5 m (%)	33.3 (7.6)	42.4 (9.9)	46.1 (9.9)	43.3 (2.8)
Clay at 1.5-2.0 m (%)	37.3 (8.7)	44.5 (10.0)	43.4 (6.5)	47.6 (4.5)
	clay Ac	crisol soil		
Bulk density (g cm ⁻³)	1.0 (0.1)	0.8 (0.1) в	0.9 (0.1)	0.9 (0.1)
pH (1:4 H ₂ O)	4.2 (0.04) _b	4.5 (0.04) _{a A}	4.5 (0.1) a	4.4 (0.04) a
Soil organic C	3.3 (0.5)	4.3 (0.4) _A	2.8 (0.4)	3.5 (0.2)
(kg C m^{-2})				
Total N (g N m ⁻²)	263.4 (67.1)	331.4 (34.1) _A	198.9 (32.5)	260.2 (22.6)
C:N ratio	13.1 (1.3)	13.0 (0.3)	14.3 (0.6)	13.5 (0.2)
Effective cation exchange	94.3 (40.8)	124.5 (25.5) _A	71.3 (22.3)	78.1 (8.4)
capacity (mmolc kg ⁻¹)				
Base saturation (%)	22.9 (5.6) _A	23.2 (5.8)	20.1 (2.6)	37.5 (7.1)
Potassium (g K m ⁻²)	9.4 (3.9)	9.6 (2.6) _A	4.2 (1.1)	4.8 (0.9)
Sodium (g Na m ⁻²)	3.6 (0.8) _A	4.2 (0.2) _A	3.7 (1.3)	1.9 (1.3)
Calcium (g Ca m ⁻²)	32.3(21.2)	33.3 (10.9) _{A†}	14.7 (2.8)	59.1 (19.5)
Magnesium (g Mg m ⁻²)	7.3 (3.9)	12.0 (4.1) _A	4.0 (0.9)	3.5 (0.8)
Aluminum (g Al m ⁻²)	50.9 (22.7)	76.6 (15.6) _A	47.2 (17.6)	34.4 (2.0)
Iron (g Fe m ⁻²)	3.7 (1.1) _{a A}	3.0 (0.4) _{a A}	2.3 (0.6) _a	0.7 (0.3) _b
Manganese (g Mn m ⁻²)	4.5 (3.1)	2.5 (0.7) _A	1.5 (0.4)	3.4 (1.3)
Bray-extractable phosphorus	1.4 (0.1) _{ab A}	0.8 (0.1) bc	0.4 (0.04) c	4.7 (1.5) a
(g P m ⁻²)				
Clay at 1.0-1.5 m (%)	39.0 (13.0)	62.8 (12.6)	40.8 (10.3)	62.8 (3.7)
Clay at 1.5-2.0 m (%)	41.3 (11.2)	46.6 (16.2)	36.5 (10.8)	63.3 (6.1)

824	Table A2. Mean (\pm SE, $n = 4$) tree density, diameter at breast height (DBH), basal area, height,
825	cumulative fine root mass in the top 1-m depth and the most common tree species with DBH \geq
826	0.10 m in different land uses within the loam and clay Acrisol soils in Jambi, Sumatra, Indonesia.
827	The vegetation characteristics (e.g. tree density, DBH, basal area, and height) were reported by
828	Kotowska et al. (2015); the five most numerous tree families with DBH \ge 0.10 m were based from
829	Rembold et al. (2017) and Rembold (pers. comm.). The fine root mass in the top 1-m soil depth
830	was measured in our present study. Mean of fine root mass followed by different lower case letters
831	indicate significant differences among land uses within each soil type (one-way ANOVA with
832	Fisher's LSD test at $P \le 0.05$, and † at $P \le 0.09$ for marginal significance).

Characteristics	Forest	Jungle rubber	Rubber	Oil palm
		loam Acr	isol soil	
Plantation age (years)	not determined (ND)	ND	14 – 17	12 – 16
Tree density (trees ha ⁻¹)	658 (26)	525 (60)	440 (81)	140 (4)
DBH (cm)	21.0 (0.5)	16.8 (0.5)	17.8 (1.2)	not applicable (NA)
Basal area (m ² ha ⁻¹)	30.7 (1.0)	16.6 (0.4)	12.2 (1.6)	NA
Tree height (m)	20.0 (0.6)	14.0 (0.2)	13.4 (0.5)	4.9 (0.6)
Fine root mass in the top 1-m soil depth (g m ⁻²)	290.2 (82.6) _{ab†}	143.9 (33.0) _b	188.2 (37.6) _b	356.8 (49.9) _a

Five most	Burseraceae,	Euphorbiaceae,	Hevea	Elaeis guineensis
numerous tree	Dipterocarpaceae,	Moraceae,	brasiliensis	
families	Sapotaceae,	Apocynaceae,		
	Phyllanthaceae,	Rubiaceae,		
	Euphorbiaceae	Fabaceae		
		clay Acri	sol soil	
Plantation age	ND	ND	7 – 16	9 - 13
(years)				
Tree density (trees	471 (31)	685 (72)	497 (15)	134 (6)
ha ⁻¹)				
DBH (cm)	23.0 (0.4)	17.3 (0.6)	15.2 (0.7)	NA
Basal area (m ² ha ⁻¹)	29.4 (1.7)	21.1 (1.4)	10.0 (1.4)	NA
Tree height (m)	17.0 (0.5)	15.2 (0.3)	13.4 (0.1)	4.0 (0.3)
Fine root mass in	140.4 (33.0) c	402.2 (65.9) _b	309.6 (16.0) bc	630.1 (86.2) a
the top 1-m soil				
depth (g m ⁻²)				
Five most	Phyllanthaceae,	Euphorbiaceae,	Hevea	Elaeis guineensis
numerous tree	Olacaceae,	Moraceae,	brasiliensis	
families	Fabaceae,	Fabaceae,		
	Meliaceae,	Apocynaceae,		
	Dipterocarpaceae	Ixonanthaceae		

Table A3. Pearson correlations among element concentrations (mg L⁻¹) in soil solution (1.5-m depth) of the different land uses within the loam Acrisol soil in Jambi, Sumatra, Indonesia. Correlations were carried out using monthly averages of four replicate plots per land use (n = 12monthly measurements in 2013). Elements that had concentrations < 0.03 mg L⁻¹ (total Fe, total Mn, and total P) and total Si (that did not show correlation with other elements) are not reported below.

Element	NH4 ⁺ -N	NO ₃ N	DOC	Na ⁺	\mathbf{K}^+	Ca ²⁺	Mg ²⁺	Total Al	Total S	Cl
				F	Forest					
DON	0.79 °	-0.24	0.77 ^c	0.36	0.43	0.80 °	0.77 ^c	0.84 ^c	-0.17	0.86 °
NH4 ⁺ -N		0.22	0.48	0.23	0.64 ^b	0.67 ^b	0.65 ^b	0.58 ^b	0.30	0.58 ^b
NO ₃ ⁻ -N			-0.12	-0.09	0.35	-0.26	-0.25	-0.45	0.63 ^b	-0.47
DOC				0.36	0.45	0.72 ^c	0.71 ^c	0.73 ^c	-0.02	0.68 ^b
Na^+					0.58 ^b	0.53 ^a	0.46	0.34	0.23	0.45
\mathbf{K}^+						0.51 ^a	0.45	0.29	0.71 ^c	0.33
Ca ²⁺							0.99 ^c	0.94 ^c	0.00	0.92 °
Mg^{2+}								0.95 °	-0.03	0.92 °
Total Al									-0.28	0.95 °
Total S										-0.23
				Jung	le rubbe	r				
DON	0.80 °	0.28	0.77 ^c	0.72 °	0.85 ^c	0.72 °	0.79 ^c	0.30	0.60 ^b	0.68 ^b

DON	0.80 °	0.28	0.77 ^c	0.72 ^c	0.85 ^c	0.72 °	0.79 °	0.30	0.60 ^b	0.68 ^b
NH4 ⁺ -N		0.32	0.73 °	0.35	0.77 ^c	0.53 ^a	0.67 ^b	0.55 ^b	0.17	0.79 °
NO ₃ ⁻ -N			0.35	0.17	0.20	0.65 ^b	0.62 ^b	0.61 ^b	-0.11	0.65 ^b
DOC				0.63 ^b	0.76 ^c	0.51 ^a	0.53 ^a	0.13	0.57^{b}	0.49 ^a

Na^+	0.80 °	0.58 ^b	0.55 ^b	-0.18	0.93 °	0.29
K^+		0.65 ^b	0.70 ^c	0.12	0.65 ^b	0.60 ^b
Ca ²⁺			0.97 ^c	0.56 ^b	0.32	0.84 ^c
Mg^{2+}				0.65 ^b	0.27	0.93 ^c
Total Al					-0.47	0.85 ^c
Total S						-0.02

				R	Rubber					
DON	-0.12	-0.32	0.53 ^a	0.04	0.65 ^b	0.37	0.65	0.67 ^b	-0.28	0.39
NH4 ⁺ -N		0.10	0.31	0.61 ^b	-0.05	0.17	-0.07	-0.41	0.65 ^b	-0.18
NO ₃ ⁻ -N			-0.25	0.25	-0.48	0.42	0.15	-0.09	0.26	0.31
DOC				0.50 ^a	0.46	0.51 ^a	0.50 ^a	0.29	0.30	0.34
Na ⁺					0.17	0.46	0.08	-0.34	0.85 ^c	0.00
\mathbf{K}^+						0.24	0.55 ^b	0.54 ^a	-0.15	0.38
Ca ²⁺							0.81 ^c	0.40	0.27	0.72 ^c
Mg^{2+}								0.84 ^c	-0.26	0.92 °
Total Al									-0.70 ^c	0.83 ^c
Total S										-0.35

Oil palm fertilized areas											
DON	-0.28	0.08	-0.18	-0.57 ^b	-0.12	0.16	0.31	0.50	-0.06	0.08	
NH4 ⁺ -N		0.54 ^a	-0.12	0.00	0.50	0.15	0.37	0.46	0.22	0.46	
NO ₃ ⁻ -N			-0.12	0.14	-0.02	-0.49	0.00	0.63 ^b	-0.38	0.10	
DOC				-0.22	0.08	0.02	0.29	-0.17	0.40	-0.47	
Na ⁺					-0.12	-0.45	-0.45	-0.37	-0.38	0.22	

K^+	0.58 ^b	0.43	-0.17	0.58 ^b	0.27
Ca^{2+}		0.48	-0.19	0.79 ^c	0.45
Mg^{2+}			0.40	0.72 ^c	0.41
Total Al				-0.16	0.27
Total S					0.30

-0.38	0.38	0.22							
		0.22	-0.38	0.24	-0.47	-0.16	0.47	-0.59 ^b	0.04
	0.07	0.23	0.40	0.25	0.04	0.08	-0.17	0.42	0.06
		0.61 ^b	0.12	0.56 ^b	-0.26	-0.21	0.11	0.20	0.02
			-0.10	0.57 ^b	-0.38	-0.55 ^b	-0.28	0.22	-0.42
				0.09	0.23	0.22	-0.35	0.61 ^b	0.09
					-0.27	-0.21	-0.07	0.29	-0.06
						0.83 ^c	0.30	-0.15	0.72 °
							0.63 ^b	-0.41	0.95 °
								-0.81 ^c	0.79 ^c
									-0.48
		- < 0.05 CD < 0.01	0.61 ^b	0.61 ^b 0.12 -0.10	0.61 ^b 0.12 0.56 ^b -0.10 0.57 ^b 0.09	0.61 ^b 0.12 0.56 ^b -0.26 -0.10 0.57 ^b -0.38 0.09 0.23 -0.27	0.61 ^b 0.12 0.56 ^b -0.26 -0.21 -0.10 0.57 ^b -0.38 -0.55 ^b 0.09 0.23 0.22 -0.27 -0.21 0.83 ^c	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.61 ^b 0.12 0.56 ^b -0.26 -0.21 0.11 0.20 -0.10 0.57 ^b -0.38 -0.55 ^b -0.28 0.22 0.09 0.23 0.22 -0.35 0.61 ^b -0.27 -0.21 -0.07 0.29 0.83 ^c 0.30 -0.15 0.63 ^b -0.41 -0.81 ^c

 ${}^{a}P \le 0.09, {}^{b}P \le 0.05, {}^{c}P \le 0.01.$

Table A4. Pearson correlations among element concentrations (mg L⁻¹) in soil solution (1.5-m depth) of the different land uses within the clay Acrisol soil in Jambi, Sumatra, Indonesia. Correlations were carried out using monthly averages of four replicate plots per land use (n = 12monthly measurements in 2013). Element that had concentrations < 0.03 mg L⁻¹ (total Fe, total Mn, and total P) and total Si (that did not show correlation with other elements) are not reported below.

Element	NH4 ⁺ -N	NO ₃ ⁻ -N	DOC	Na ⁺	K ⁺	Ca ²⁺	Mg ²⁺	Total Al	Total S	Cl
					Forest					
DON	0.10	-0.39	0.57 ^b	0.32	0.53 ^a	0.17	0.20	-0.28	0.25	-0.20
NH4 ⁺ -N		-0.48	0.81 ^c	0.63 ^b	0.23	0.51 ^a	0.28	-0.11	-0.27	0.09
NO ₃ ⁻ -N			-0.48	-0.24	-0.18	-0.05	-0.03	0.36	0.12	0.37
DOC				0.66 ^b	0.41	0.48	0.31	-0.25	-0.15	-0.06
Na ⁺					0.69 ^b	0.52 ^a	0.54 ^a	-0.22	-0.24	-0.10
\mathbf{K}^+						0.74 ^c	0.88 ^c	0.22	-0.17	0.26
Ca ²⁺							0.93 ^c	0.54 ^a	-0.29	0.70 ^c
Mg^{2+}								0.52 ^a	-0.34	0.59 ^b
Total Al									-0.15	0.94 ^c
Total S										-0.10
Jungle rubber										

DON	0.23	0.55 ^b	0.58 ^b	0.19	0.69 °	0.50 ^a	0.63 ^b	0.70 ^c	-0.22	0.49 ^a
NH4 ⁺ -N		0.01	0.36	0.35	0.35	0.29	0.29	0.16	0.31	0.18
NO ₃ ⁻ -N			0.32	0.30	0.49 ^a	0.51 ^a	0.50 ^a	0.35	0.13	0.42
DOC				-0.24	0.11	-0.14	-0.05	0.29	0.06	-0.20
Na ⁺					0.68 ^c	0.84 ^c	0.73 ^c	0.01	0.52 ^a	0.66 ^b

K^+	0.87 ^c	0.93 °	0.63 ^b	0.09	0.84 ^c
Ca^{2+}		0.97 ^c	0.50 ^a	0.09	0.95 °
Mg^{2+}			0.66 ^b	-0.04	0.97 ^c
Total Al				-0.62 ^b	0.68 ^b
Total S					-0.18

Rubber										
DON	-0.20	-0.18	0.21	-0.29	0.41	0.40	0.55 ^b	0.65 ^b	-0.57 ^b	0.48
NH4 ⁺ -N		0.22	0.81 ^c	0.85 °	0.47	0.19	0.10	-0.20	0.52 ^a	-0.06
NO ₃ ⁻ -N			-0.07	-0.16	-0.44	-0.68 ^b	-0.60 ^b	-0.38	0.05	-0.63 ^b
DOC				0.79 °	0.71 ^c	0.54 ^a	0.45	0.20	0.43	0.30
Na ⁺					0.61 ^b	0.38	0.21	-0.15	0.65 ^b	0.07
\mathbf{K}^+						0.67 ^b	0.66 ^b	0.46	0.08	0.64 ^b
Ca ²⁺							0.93 ^c	0.73 ^c	-0.16	0.83 ^c
Mg^{2+}								0.88 ^c	-0.39	0.93 ^c
Total Al									-0.58 ^b	0.89 °
Total S										-0.40

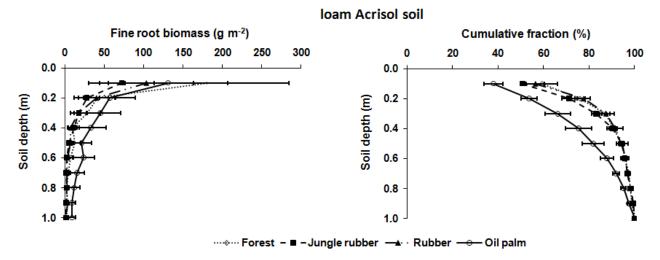
Oil palm fertilized areas										
DON	0.02	-0.09	0.49	0.70 ^b	0.69 ^b	0.67 ^b	0.42	0.45	0.54 ^a	0.63 ^b
NH4 ⁺ -N		0.08	0.15	0.39	0.37	0.16	0.06	0.06	0.46	-0.01
NO ₃ ⁻ -N			-0.18	0.03	0.46	0.51 ^a	-0.01	0.19	0.33	-0.49
DOC				0.52 ^a	0.66 ^b	0.56 ^a	0.50	0.56 ^a	0.25	0.70 ^b
Na ⁺					0.61 ^b	0.61 ^b	0.29	0.21	0.75 °	0.55 ^a
\mathbf{K}^+						0.85 °	0.74 ^c	0.78 ^c	0.52 ^a	0.59 ^b

Ca^{2+}	0.81 ^c	0.74 °	0.69 ^b	0.64 ^b
Mg^{2+}		0.95 °	0.26	0.74 ^c
Total Al			0.15	0.75 ^c
Total S				0.26

Oil palm frond-stacked areas										
DON	0.19	0.34	0.15	0.49 ^a	0.47	0.51 ^a	0.23	0.29	0.28	0.36
NH4 ⁺ -N		-0.07	0.27	0.21	0.38	0.11	0.06	0.07	0.13	0.09
NO ₃ ⁻ -N			-0.28	0.24	0.32	0.13	-0.13	0.09	0.56 ^b	-0.05
DOC				0.09	0.23	0.25	0.45	0.02	-0.46	0.19
Na ⁺					0.91 ^c	0.94 ^c	0.76 ^c	0.91 ^c	0.33	0.89 ^c
\mathbf{K}^+						0.88 ^c	0.74 ^c	0.80 °	0.21	0.79 ^c
Ca ²⁺							0.90 ^c	0.91 ^c	0.10	0.95 ^c
Mg^{2+}								0.81 ^c	-0.28	0.93 ^c
Total Al									0.16	0.92 ^c
Total S										-0.06

 ${}^{a}P \le 0.09, {}^{b}P \le 0.05, {}^{c}P \le 0.01.$

847 Appendix B. Fine root biomass and soil water model validation





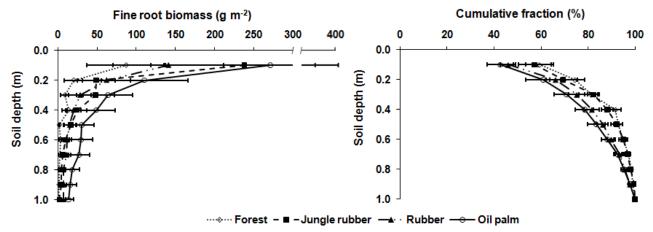


Figure B1. Fine root biomass (g m⁻²) and distribution (%) down to a depth of 1 m in different 849 land uses within the loam and clay Acrisol soils in Jambi, Sumatra, Indonesia. The root 850 measurement was conducted in each replicate plot by digging a pit (1 m x 1.5 m x 2-m depth) 851 at about 2.5-m distance from an oil palm or a tree with a diameter at breast height of ≥ 10 cm. 852 Root mass were sampled using a metal block (20 cm x 20 cm x 10 cm) at 10-cm depth interval 853 from the top down to 1 m. Roots were carefully separated from the soil by washing over a 2-854 mm mesh screen and the fine roots were collected in a basin placed underneath the mesh screen. 855 856 The roots were categorized into fine roots ($\leq 2 \text{ mm}$ diameter) and coarse roots (>2 mm diameter), dried in an oven at 70 °C for 5 days and weighed. 857

loam Acrisol soil

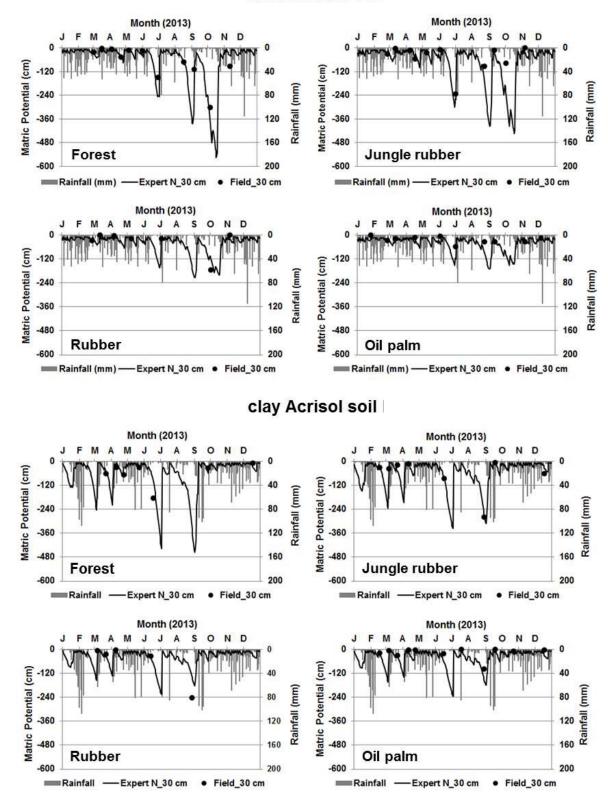


Figure B2. Validation between Expert N-modelled and field-measured matric potential at a
depth of 0.3 m in different land uses within the loam and clay Acrisol soils in Jambi, Sumatra,
Indonesia.