

1 **Inputs and processes affecting the distribution of particulate** 2 **iron in the North Atlantic along the GEOVIDE (GEOTRACES** 3 **GA01) section**

4
5
6 Arthur Gourain^{1,2}, H el ene Planquette¹, Marie Cheize^{1,3}, Nolwenn Lemaitre^{1,4}, Jan-Lukas
7 Menzel Barraqueta^{5, 6}, Rachel Shelley^{1, 7}, Pascale Lherminier⁸ and G eraldine Sarthou¹

8
9 1-UMR 6539/LEMAR/IUEM, CNRS, UBO, IRD, Ifremer, Technop ole Brest Iroise, Place Nicolas Copernic,
10 29280 Plouzan e, France

11 2- now at Ocean Sciences Department, School of Environmental Sciences, University of Liverpool, Liverpool,
12 L69 3GP, United Kingdom

13 3- now at Ifremer, Centre de Brest, G eosciences Marines, Laboratoire des Cycles G eochimiques (LCG), 29280
14 Plouzan e, France

15 4- now at Department of Earth Sciences, Institute of Geochemistry and Petrology, ETH-Z urich, Z urich,
16 Switzerland

17 5- GEOMAR, Helmholtz Centre for Ocean Research Kiel, Wischhofstra e 1-3, 24148 Kiel, Germany

18 6- now at Department of Earth Sciences, Stellenbosch University, Stellenbosch, 7600, South Africa

19 7- now at Earth, Ocean and Atmospheric Science, Florida State University, Tallahassee, Florida, 32310, USA

20 8- Ifremer, Univ. Brest, CNRS, IRD, Laboratoire d'Oc eanographie Physique et Spatiale (LOPS), IUEM, F-
21 29280, Plouzan e, France

22
23 *Correspondence to: helene.planquette@univ-brest.fr*

24 25 **Abstract**

26 The aim of the GEOVIDE cruise (May-June 2014, R/V *Pourquoi Pas?*) was to provide a better understanding of
27 trace metal biogeochemical cycles in the North Atlantic Ocean. As marine particles play a key role in the global
28 biogeochemical cycle of trace elements in the ocean, we discuss the distribution of particulate iron (PFe), in
29 relation to the distribution of particulate aluminium (PAI), manganese (PMn) and phosphorus (PP). Overall, 32
30 full vertical profiles were collected for trace metal analyses, representing more than 500 samples. This resolution
31 provides a solid basis for assessing concentration distributions, elemental ratios, size-fractionation, and adsorptive
32 scavenging processes in key areas of the thermohaline overturning circulation. Total particulate iron
33 concentrations ranged from as low as 9 pmol L⁻¹ in surface waters of the Labrador Sea to 304 nmol L⁻¹ near the
34 Iberian margin, while median PFe concentrations of 1.15 nmol L⁻¹ were measured over the sub-euphotic ocean
35 interior.

36 Within the Iberian Abyssal Plain, the ratio of PFe to PAI was identical to the continental crust molar ratio (0.21
37 mol mol⁻¹), indicating the important influence of crustal particles in the water column. Overall, the lithogenic

38 component explained more than 87% of PFe variance along the section. Within the Irminger and Labrador basins,
39 the formation of biogenic particles led to an increase of the PFe/PAI ratio (up to 0.64 mol mol⁻¹) compared to the
40 continental crust ratio. Continental margins provide high quantities of particulate trace elements (up to 10 nmol
41 L⁻¹ of PFe) to the open ocean. For example, horizontal advection of PFe was visible more than 250 km away from
42 the Iberian margin. Additionally, several benthic nepheloid layers were observed more than 200 m above the
43 seafloor along the transect, especially in the Icelandic, Irminger and Labrador basins, suspending particles with
44 high PFe content of up to 89 nmol L⁻¹.

45

46 **1. Introduction**

47 Particles play a key role in the ocean where they drive the residence time of most elements (Jeandel and Oelkers,
48 2015), and strongly influence the global biogeochemistry of macro- and micro-nutrients including iron (Milne et
49 al., 2017). In the surface ocean, biological activity produces biogenic suspended matter through planktonic
50 organisms, while atmospheric deposition (Baker et al., 2013; Jickells et al., 2005), riverine discharge (Aguilar-
51 Islas et al., 2013; Berger et al., 2008; Ussher et al., 2004) or ice-melting (Hawkings et al., 2014; Lannuzel et al.,
52 2011, 2014) deliver mostly lithogenic derived particles to surface waters. These particulate inputs are highly
53 variable, both spatially and seasonally, in the world's oceans. At depth, benthic and shelf sediment resuspension
54 (e.g. Aguilar-Islas et al., 2013; Cullen et al., 2009; Elrod et al., 2004; Fitzwater et al., 2000; Hwang et al., 2010;
55 Lam et al., 2015; Lam and Bishop, 2008; McCave and Hall, 2002), and hydrothermal activity (Elderfield and
56 Schultz, 1996; Lam et al., 2012; Tagliabue et al., 2010, 2017; Trefry et al., 1985), provides important amounts of
57 particles to the water column. Moreover, authigenic particles can be produced *in-situ* by aggregation of colloids
58 (Bergquist et al., 2007) or oxidation processes (Bishop and Fleisher, 1987; Collier and Edmond, 1984). Thus,
59 oceanic particles result from a complex combination of these different sources and processes (Lam et al., 2015).
60 In the upper water column, the total iron pool is dominated by marine particles (Radic et al., 2011) which strongly
61 interact with the dissolved pool (e.g. Ellwood et al., 2014). Indeed, dissolved iron can be scavenged onto particles
62 (Gerringa et al., 2015; Rijkenberg et al., 2014), incorporated into biogenic particles (Berger et al., 2008) or
63 produced by remineralisation of particles (Dehairs et al., 2008; Sarthou et al., 2008). Interestingly, the concept of
64 “reversible scavenging” (i.e. release at depth of dissolved iron previously scavenged onto particles) has been
65 advocated recently (Dutay et al., 2015; Jeandel and Oelkers, 2015; Labatut et al., 2014), while other studies reveal
66 distinct dissolution processes of inorganic particulate iron (e.g. Oelkers et al., 2012; Cheize et al., 2018). Slow
67 dissolution of particulate iron at margins has also been evoked as a continuous fertilizer of primary production
68 and should be considered as a source of dissolved iron (e.g. Jeandel et al., 2011; Jeandel and Oelkers, 2015; Lam
69 and Bishop, 2008). Within or below the mixed layer, the rates of regeneration processes can also impact the
70 bioavailable pool of iron, among other trace metals (e.g. Ellwood et al., 2014; Nuester et al., 2014). However, the
71 rates of these processes are not yet fully constrained. The study of particulate iron is thus essential to better
72 constrain its marine biogeochemical cycle. Interest has grown in this subject over the last 10 years, in particular
73 (e.g. Bishop and Biscaye, 1982; Collier and Edmond, 1984; Frew et al., 2006; Lam et al., 2012; Milne et al., 2017;
74 Planquette et al., 2011, 2013; Sherrell et al., 1998) and, to our knowledge, only two studies have been performed
75 on an ocean-wide scale: the GA03 GEOTRACES North Atlantic Zonal Transect (Lam et al., 2015; Ohnemus and
76 Lam, 2015) and the GP16 GEOTRACES Pacific Transect (Lam et al., 2017; Lee et al., 2017).

77 Within this global context, this paper presents the particulate iron distribution of the North Atlantic Ocean, along
78 the GEOTRACES GA01 section (GEOVIDE), and discusses the various sources and processes affecting
79 particulate iron (PFe) distribution, using particulate aluminium (PAI), phosphorus (PP) or manganese (PMn)
80 distributions to support our conclusions.
81

81

82 **2. Methods**

83 2.1. Study area

84 Particulate samples were collected at 32 stations during the GEOVIDE (GEOTRACES GA01 section) cruise
85 between May and June 2014 aboard the R/V *Pourquoi Pas?* in the North Atlantic Ocean (Sarthou et al., 2018).
86 The sampling spanned several biogeochemical provinces (Figure 1), starting over the Iberian margin (IM, Stations
87 2, 4 and 1), and proceeding to the Iberian Abyssal Plain (IAP, Stations 11 to 17), the Western European Basin
88 (WEB, Station 19 to Station 29) and the Icelandic Basin (IcB, Stations 32 to 36). Then, samples were collected
89 above the Reykjanes Ridge (RR, Station 38), in the Irminger Basin (IrB, Stations 40 to 60), close to the Greenland
90 shelf (GS, Stations 53, 56 and 61), the Labrador Basin (LB, Stations 63 to 77) and finally close the Newfoundland
91 shelf (NS, Station 78) (Figure 1). The North Atlantic is characterized by a complex circulation (briefly described
92 in section 3.1 and in detail by Zunino et al. (2017) and García-Ibáñez et al. (2015)) and is one of the most
93 productive regions of the global ocean (Martin et al., 1993; Sanders et al., 2014).
94

94

95

96 2.2. Sampling

97 Samples were collected using the French GEOTRACES clean rosette, equipped with twenty-two 12 L GO-FLO
98 bottles (two bottles were leaking and were not deployed during the cruise). GO-FLO bottles (General Oceanic's)
99 were initially cleaned in the home laboratory (LEMAR) following the GEOTRACES procedures (Cutter and
100 Bruland, 2012). The rosette was deployed on a 14 mm Kevlar cable with a dedicated, custom-designed clean
101 winch. Immediately after recovery, the GO-FLO bottles were individually covered at each end with plastic bags
102 to minimize contamination. Bottles were then transferred into a clean container (class-100) for sampling. On each
103 cast, nutrient and/or salinity samples were taken to check potential leakage of the GO-FLO bottles.

104 Filters were cleaned following the GEOTRACES protocols (<http://www.geotraces.org/images/Cookbook.pdf>)
105 and kept in acid-cleaned 1 L LDPE bottles (Nalgene) filled with ultrapure water (Milli-Q, resistivity of 18.2 MΩ
106 cm) until use. All filters were 25 mm diameter in order to optimize the signal over the filter blank, except at the
107 surface depth where 47 mm diameter filters were used. The filters were mounted on acid-cleaned polysulfone
108 filter holders (Nalgene™). Prior to filtration, the GO-FLO bottles were shaken three times, as recommended in
109 the GEOTRACES cookbook to avoid settling of particles in the lower part of the bottle. GO-FLO bottles were
110 pressurized to < 8 psi with 0.2 μm filtered nitrogen gas (N₂, Air Liquide). Seawater was then filtered directly
111 through paired filters (Pall Gelman Supor™ 0.45 μm polyetersulfone, and Millipore mixed ester cellulose MF 5
112 μm) mounted in Swinnex polypropylene filter holders (Millipore), following Planquette and Sherrell (2012) inside
113 the clean container. Filtration was operated until the bottle was empty or until the filter clogged; the volume
114 filtered ranged from 2 L for surface samples to 11 L within the water column. After filtration, filter holders were
115 disconnected from the GO-FLO bottles and a gentle vacuum was applied using a syringe in order to remove any

116 residual water under a laminar flow hood. Filters were then removed from the filter holders with plastic tweezers
117 (which were rinsed with Milli-Q between samples). Most of the remaining seawater was removed via ‘sipping’
118 by capillary action, when placing the non-sampled side of the filter onto a clean 47 mm supor filter. Each filter
119 pair was then placed in an acid-cleaned polystyrene PetriSlide (Millipore), double bagged, and finally stored at -
120 20 °C until analysis at LEMAR. Between casts, filter holders were thoroughly rinsed with Milli-Q, placed in an
121 acid bath (5% Trace metal grade HCl) for 24 hours, then rinsed with Milli-Q.

122 At each station, process blanks were collected as follows: 2 L of a deep (1000 m) and a shallow (40 m) seawater
123 sample were first filtered through a 0.2 µm pore size capsule filter (Pall Gelman Acropak 200) mounted on to the
124 outlet of the GO-FLO bottle before passing through the particle sampling filter, which was attached directly to
125 the swinnex filter holder.

126

127 2.3. Analytical methods

128 In the home laboratory, sample handling was performed inside a clean room (Class 100). All solutions were
129 prepared using ultrapure water (Milli-Q) and all plasticware had been acid-cleaned before use. Frozen filters,
130 collected within the mixed layer or within nepheloid layers, were first cut in half using a ceramic blade: one filter
131 half was dedicated to total digestion (see below), while the other half was archived at -20 °C for SEM analyses or
132 acid leaching of “labile” metals following Berger et al. (2008) method (to be published separately).

133 Filters were digested following the method described in Planquette and Sherrell (2012). Filters were placed on
134 the inner wall of acid-clean 15 mL PFA vials (Savillex™), and 2 mL of a solution containing 2.9 mol L⁻¹
135 hydrofluoric acid (HF, suprapur grade, Merck) and 8 mol L⁻¹ nitric acid (HNO₃, Ultrapur grade, Merck) was added
136 to each vial. Vials were then closed and refluxed at 130 °C on a hot plate for 4 hours, after which the filters were
137 removed. After cooling, the digest solution was evaporated at 110 °C to near dryness. Then, 400 µL of
138 concentrated HNO₃ (Ultrapur grade, Merck) was added, and the solution was re-evaporated at 110 °C. Finally,
139 the obtained residue was dissolved with 3 mL of 0.8 mol L⁻¹ HNO₃ (Ultrapure grade, Merck). This archived
140 solution was transferred to an acid cleaned 15 mL polypropylene centrifuge tube (Corning®) and stored at 4 °C
141 until analyses.

142 All analyses were performed on a sector field inductively coupled plasma mass spectrometer (SF-ICP-MS
143 Element 2, Thermo-Fisher Scientific). Samples were diluted by a factor of 7 on the day of analysis in acid-washed
144 13 mm (outer diameter) rounded bottom, polypropylene centrifuge tubes (VWR) with 0.8 mol L⁻¹ HNO₃ (Ultrapur
145 grade, Merck) spiked with 1 µg L⁻¹ of indium (¹¹⁵In) solution in order to monitor the instrument drift. Samples
146 were introduced with a PFA-ST nebulizer connected to a quartz cyclonic spray chamber (Elemental Scientific
147 Incorporated, Omaha, NE) via a modified SC-Fast introduction system consisting of an SC-2 autosampler, a six-
148 port valve and a vacuum-rinsing pump. The autosampler was contained under a HEPA filtered unit (Elemental
149 Scientific). Two 6-point, matrix-matched multi-element standard curves with concentrations bracketing the range
150 of the samples were run at the beginning, the middle and the end of each analytical run. Analytical replicates were
151 made every 10 samples, while accuracy was determined by performing digestions of the certified reference
152 material BCR-414 (plankton, Community Bureau of Reference, Commission of the European Communities),
153 PACS-3 and MESS-4 (marine sediments, National Research Council Canada), following the same protocol used
154 for the samples. Recoveries were typically within 10% of the certified values (and within the error of the data,
155 taken from replicate measurements, Table 1). Once all data were normalized to an ¹¹⁵In internal standard and

156 quantified using an external standard curve, the dilution factor of the total digestion was accounted for. The
 157 elemental concentrations were obtained per filter (pmol/filter) and were then process blank-corrected. Finally,
 158 pmol/filter values were divided by the volume of water filtered through stacked filters.
 159 Total concentrations (sum of small size fraction (0.45-5 µm) and large (>5 µm) size fraction) of particulate trace
 160 elements are reported in Table S1.

161

162 2.4. Positive matrix factorisation

163 Positive Matrix Factorisation (PMF) was run to characterise the main factors influencing the particulate trace
 164 element variance along the GEOVIDE section. In addition to PFe, PAI, PMn, and PP, nine additional elements
 165 were included in the PMF: yttrium (Y), barium (Ba), lead (Pb), thorium (Th), titanium (Ti), vanadium (V), cobalt
 166 (Co), copper (Cu) and zinc (Zn). The PMF was conducted on samples where all elements were above their
 167 detection limits; after selection, 445 of the 549 existing data points were used. Analyses were performed using
 168 the PMF software, EPA PMF 5.0, developed by the USA Environmental Protection Agency (EPA). Three to six
 169 factor models were run on the data. The configuration that provided the lowest error estimation (i.e. was the most
 170 reliable) was the four factor model. To ensure stability, this model was run 100 times. After displacement, error
 171 estimation and bootstrap error estimation, the model was recognised as stable.

172

173 2.5. Derived and ancillary parameters

174 To investigate the proportion of lithogenic iron within the bulk particulate iron, we used the Upper Continental
 175 Crust (UCC) Fe/Al molar ratio (0.21) of Taylor and McLennan (1995) to calculate the lithogenic components of
 176 particles (%PFelitho) following Eq. (1):

177

$$178 \quad \%PFelitho = 100 * \left(\frac{PAI}{PFe} \right)_{sample} * \left(\frac{PFe}{PAI} \right)_{UCCratio} \quad (1)$$

179

180 The non-lithogenic PFe is obtained using Eq. (2):

181

$$182 \quad \%PFenon_litho = 100 - \%PFelitho \quad (2)$$

183

184 Note that while the %PFelitho and %PFenon-litho proxies are interesting tools to evaluate the importance of lithogenic
 185 and non-lithogenic (either biogenic or authigenic) fraction, they have to be used carefully, as the spatial and
 186 temporal variation of the lithogenic component ratios may involve uncertainties of the estimated fraction value.

187 In addition to PAI, PMn can be used as a tracer of inputs from shelf resuspension (Lam and Bishop, 2008), using
 188 a percentage of sedimentary inputs “%bulk sediment inputs” estimated according to the following equation:

$$189 \quad \%bulksedimentPMn = 100 * \left(\frac{PAI}{PMn} \right)_{sample} * \left(\frac{PMn}{PAI} \right)_{UCCratio} \quad (3)$$

190 with PAI/PMn being the ratio from the GEOVIDE samples and the PMn/PAI being the UCC value (0.0034; Taylor
 191 and McLennan, 1995).

192 This proxy can be a good indicator of sediment resuspension. We assume that particles newly resuspended in the
193 water column will have the same PMn/PAI ratio as the UCC ratio, leading to a “%bulk sediment Mn” of 100%.
194 This proxy assumes homogeneity of the sediment PMn/PAI ratio throughout the GEOVIDE section. However,
195 this may not be the case at every station. In consequence, this proxy should only be used to identify new benthic
196 resuspension at specific locations; inter-comparison between several locations may not be appropriate. When a
197 sample presents a “%bulk sediment Mn” greater than 100%, we have assigned a maximum value of 100%. As the
198 Mn cycle can also be influenced by biotic uptake (e.g. Peers and Price, 2004; Sunda and Huntsman, 1983), this
199 proxy is only used at depths where biologic activity was negligible (i.e. below 150m depth).

200 Potential temperature (θ°), salinity (S), and transmissometry data were retrieved from the CTD sensors (CTD
201 SBE911 equipped with a SBE43).

202

203

204 **3. Results**

205 *3.1. Hydrography setting*

206 Here, we briefly describe the hydrography encountered during the GEOVIDE section (Figure 2) as a thorough
207 description is available in García-Ibáñez et al. (2015). At the start of the section, the warm and salty Mediterranean
208 Water (MW, $S = 36.50$, $\theta^\circ = 11.7$ °C) was sampled between 600 and 1700 m in the Iberian Abyssal Plain (IAP).
209 MW resulted from the mixing between the Mediterranean Overflow Water (MOW) plume coming from the
210 Mediterranean Sea and local waters. Surface water above the Iberian Shelf was characterised by low salinity ($S =$
211 34.95) at station 2 and 4 compared to surrounding water masses. Close to the sea floor of the Iberian Abyssal
212 Basin, the North East Atlantic Deep Water (NEADW, $S = 34.89$, $\theta^\circ = 2.0$ °C) spread northward. The North
213 Atlantic Central Water (NACW, $S > 35.60$, $\theta^\circ > 12.3$ °C) was the warmest water mass of the transect and was
214 observed in the subsurface layer of the Western European Basin and Iberian Abyssal Plain. An old Labrador Sea
215 Water (LSW, $S = 34.87$, $\theta^\circ = 3.0$ °C) flowed inside the Western European and Icelandic Basins, between 1000
216 and 2500m depth.

217 In the Icelandic Basin, below the old LSW, the Iceland-Scotland Overflow Water (ISOW, $S = 34.98$, $\theta^\circ = 2.6$ °C)
218 spread along the Reykjanes Ridge slope. This cold water, originating from the Arctic, led to the formation of
219 NEADW after mixing with surrounding waters. North Atlantic hydrography was impacted by the northward
220 flowing of the North Atlantic Current (NAC), which carried warm and salty waters from the subtropical area. Due
221 to air-sea interactions and mixing with surrounding water, the NACW is cooled and freshened in the subpolar
222 gyre and is transformed in Subpolar Mode Water (SPMW). The formation of SPMW inside the Icelandic and
223 Irminger Basins leads to the formation of regional mode waters: the Iceland Subpolar Mode Water (IcSPMW, S
224 $= 35.2$, $\theta^\circ = 8.0$ °C) and the Irminger Subpolar Mode Water (IrSPMW, $S = 35.01$, $\theta^\circ = 5.0$ °C), respectively.
225 IcSPMW was a relatively warm water mass with potential temperature up to 7 °C (García-Ibáñez et al., 2015).
226 Another branch of the NAC mixed with Labrador Current waters to form the relatively fresh SubArctic
227 Intermediate Water (SAIW, $S = < 34.8$, 4.5 °C $< \theta^\circ < 6$ °C).

228 The Irminger Basin is a complex area with a multitude of water masses. In the middle of the basin, an old LSW,
229 formed one year before (Straneo et al., 2003), spread between 500 and 1200 m depth. Close to the bottom, the

230 Denmark Strait Overflow Water (DSOW, $S = 34.91$) flowed across the basin. Greenland coastal waters were
231 characterised by low salinity values, down to $S = 33$. The strong East Greenland Current (EGC) flowed southward
232 along the Greenland shelf in the Irminger Basin. At the southern tip of Greenland, this current enters the Labrador
233 Basin along the west coast of Greenland and followed the outline of the basin until the Newfoundland shelf. In
234 the Labrador Basin, the deep convection of SPMW at 2000 m was involved in the formation of the LSW ($S =$
235 34.9 , $\theta = 3.0^{\circ}\text{C}$) (García-Ibáñez et al., 2015; Yashayaev and Loder, 2009). Above the Newfoundland Shelf,
236 surface waters were affected by discharge from rivers and ice-melting and characterised by extremely low salinity
237 for open ocean waters, below 32 in the first 15 meters.

238 3.2. Section overview

239 Total particulate concentrations spanned a large range of concentrations from below detection (Table 1) to 304
240 nmol L^{-1} for PFe, 1544 nmol L^{-1} for PAI, 3.5 nmol L^{-1} for PMn and 402 nmol L^{-1} for PP. The ranges of
241 concentrations are comparable to other studies recently published (Table 2).

242 Along the section, PFe, PAI, and PMn were predominantly found ($> 90\%$) in particles larger than $5 \mu\text{m}$, except
243 in surface waters, reflecting a more heterogenous pattern, where $9 \pm 8.6\%$ of PFe, $10.9 \pm 15.4\%$ of PAI and 32.8
244 $\pm 16.6\%$ of PMn, $38.8 \pm 8.6\%$ of PP were hosted by smaller particles ($0.45\text{-}5 \mu\text{m}$). Data are shown in Figure 3.

245

246 3.3. Open Ocean stations: from the Iberian Abyssal Plain to the Labrador Basin

247 This concerns all stations from station 11 to 77, with the exception of stations 53, 56 and 61 which were sampled
248 close to the Greenland coast (Figure 1). Particulate iron concentration profiles showed identical patterns at all the
249 open ocean stations encountered along the section. Median PFe was low at 0.25 nmol L^{-1} within the first 100 m
250 and steadily increased with depth. However, at two stations, elevated concentrations were determined in the upper
251 100 m, up to 4.4 nmol L^{-1} at station 77 at 40 m depth and 7 nmol L^{-1} at station 63 between 70 and 100 m depth.
252 PFe concentrations gradually increased with depth, with a median PFe of 1.74 nmol L^{-1} below 1000m. Close to
253 the seafloor of some stations (26, 29, 32, 34, 49, 60, and 71), high concentrations of PFe were observed, up to 88
254 nmol L^{-1} (station 71 at 3736 m). These high PFe values were associated with low beam transmissometry values \leq
255 97% .

256 Particulate aluminium and manganese profiles were similar to PFe profiles, with low concentrations measured in
257 the first 100 m (1.88 nmol L^{-1} and 55 pmol L^{-1} , respectively) which increased towards the seafloor. Close to the
258 seafloor, high concentrations were determined at the same stations cited above for PFe, with a maximum of 264
259 nmol L^{-1} and 3.5 nmol L^{-1} for PAI and PMn respectively at station 71 (supplementary Table S1). Highest
260 particulate phosphorus concentrations were in the uppermost 50 m, with a median value of 66 nmol L^{-1} . Below
261 200 m depth PP concentrations decreased to values below 10 nmol L^{-1} . Inter-basins differences were observed
262 within surface samples, with median PP concentration being higher in the Irminger Basin (127 nmol L^{-1}) than in
263 the Iberian Abyssal Plain (28 nmol L^{-1}) (Figure 3).

264 Finally, above the Reykjanes Ridge, PP, PMn, PAI and PFe concentrations were in the same range as the
265 surrounding open ocean stations. However, close to the seafloor, high concentrations were measured, with PFe,

266 PAI, and PMn reaching 16.2 nmol L⁻¹, 28.8 nmol L⁻¹, and 0.51 nmol L⁻¹ at 1354 m depth, respectively (Figure 3
267 and Table S1).

268

269 3.4. Margins and Shelves: Iberian Margin (stations 1 to 4), Greenland coast (stations 53, 56
270 and 61) and Newfoundland Shelf (station 78)

271

272 The Iberian margin was characterised by low beam transmissometry values at station 2 (88 % at 140 m depth,
273 Figure 4a) suggesting high particle concentrations. Particulate iron concentrations varied from 0.02 nmol L⁻¹ to
274 304 nmol L⁻¹. Within the first 50 m, PFe concentrations decreased towards the shelf break where PFe dropped
275 from 2.53 nmol L⁻¹ (station 2) to 0.8 nmol L⁻¹ (Station 1). At all three stations, PFe concentrations increased with
276 depth and reached a maximum close to the seafloor. For example, 300 nmol L⁻¹ of PFe was determined at 138.5
277 m depth at station 2. Lithogenic tracers, such as PAI or PMn, presented similar profiles to PFe with concentrations
278 ranging from 0.11 and 1544 nmol L⁻¹, and from below detection limit to 2.51 nmol L⁻¹, respectively (Figure 3,
279 Table S1). Total particulate phosphorus concentrations were relatively low in surface waters ranging from values
280 below detection to 38 nmol L⁻¹; concentrations decreased with depth and were less than 0.7 nmol L⁻¹ below 1000
281 m depth.

282 In the vicinity of the Greenland shelf, PFe concentrations had a high median value of 10.8 nmol L⁻¹ and were
283 associated with high median PAI and PMn concentrations of 32.3 nmol L⁻¹ and 0.44 nmol L⁻¹, respectively.
284 Concentrations of PP were high at the surface with a value of 197 nmol L⁻¹ at 25 m depth of station 61. Then, PP
285 concentrations decreased strongly, to less than 30 nmol L⁻¹ below 100 meters depth. Furthermore, beam
286 transmissometry values in surface waters at these three stations, were the lowest of the entire section, with values
287 below 85 % (Figure 4a).

288 Close to the Newfoundland margin, surface waters displayed a small load of particulate trace metals as PFe, PAI,
289 and PMn concentrations were below 0.8 nmol L⁻¹, 2 nmol L⁻¹, and 0.15 nmol L⁻¹, respectively. Then, close to the
290 bottom of station 78, at 371 m depth, beam transmissometry values dropped to 94 % (Figure 4a) and were
291 associated with extremely high concentrations of PFe = 168 nmol L⁻¹, PAI = 559 nmol L⁻¹, and PMn = 2 nmol L⁻¹.
292 Total PP concentrations in the first 50 m ranged from 35 to 97 nmol L⁻¹. Below 50 m, PP remained relatively
293 high with values up to 16 nmol L⁻¹ throughout the water column. (Figure 3 and Table S1).

294

295 **4. Discussion**

296 Our goal was to investigate mechanisms that drive the distribution of PFe in the North Atlantic, in particular the
297 different routes of supply and removal. Possible sources of PFe include lateral advection offshore from margins,
298 atmospheric inputs, continental run-off, melting glaciers and icebergs, resuspended sediments, hydrothermal
299 inputs and biological uptake. Removal processes include remineralization, dissolution processes and sediment
300 burial.

301 In the following sections, we examine each of these sources and processes, explore the evidence for their relative
302 importance, and use compositional data to estimate the particle types and host phases for iron and associated
303 elements.

304 4.1. Analysis of the principal factors controlling variance: near-ubiquitous influence of crustal
305 particles in the water column

306 Positive matrix factorisation analysis (Figure 5) was undertaken on the entire dataset, in consequence, the factors
307 described below are highly influenced by the major variations of particulate element concentrations (usually at
308 the interfaces, i.e. margin, seafloor, surface layer). The first factor is characterised by lithogenic elements,
309 representing 86.8 % of the variance of PFe, 75.8 % of PAI and 90.5 % of PTi. The second factor is correlated with
310 both Mn and Pb and explains no less than 76.5 % and 77.0 % of their respective variances. Ohnemus and Lam
311 (2015) observed this co-relation between manganese and lead particles and explained it by the co-transport on
312 Mn-oxides (Boyle et al., 2005). The formation of barite explains the third factor and constrained 87.7 % of the Ba
313 variance in the studied regions. Biogenic barite accumulation within the mesopelagic layer is related to bacterial
314 activity and remineralisation of biogenic material (Lemaitre et al., 2018a). A biogenic component is the fourth
315 factor and explained most of PP variance, 83.7 %. The micronutrient trace metals, copper, cobalt and zinc, had
316 more than a quarter of their variances influenced by this factor. Note that the biogenic contribution to PFe and
317 other trace elements will be discussed in another paper (Planquette et al., in prep).

318 These results indicate that along the GA01 section, PFe distributions were predominantly controlled by lithogenic
319 material and to a smaller extent by remineralisation processes (PMF, factor 3 = 4.1 %). This does not rule out
320 some biogenic influences on PFe distribution, especially in the surface, but its contribution is most likely obscured
321 by the high lithogenic contribution.

322 To further investigate the influence of crustal material on the distribution of PFe, it is instructive to examine the
323 distribution of the PFe to PAI molar ratio, and the resulting %PFe_{litho} (see section 2.6 for definition of this
324 parameter) along the section (Figure 6). Overall, the estimated lithogenic contribution to PFe varies from 25 %
325 (west of the Irminger Basin, station 60, 950 m depth) to 100 % at stations located within the Western European
326 Basin. Note that 100% of estimated lithogenic PFe does not necessary mean that biogenic particles are absent;
327 they may just be masked by the dominance of lithogenic particles. Important inter-basin variations are observed
328 along the section (Figure 6). The IAP and WEB displayed high median values of the proxy %PFe_{litho}, 90 % (Figure
329 6b); this could be linked to the lateral advection of iron rich lithogenic particles sourced from the Iberian margin
330 and to atmospheric inputs (Shelley et al., 2017). Then, between stations 26 and 29, the %PFe_{litho} proxy values
331 dramatically decreased, and reached values less than 55 % in the Iceland, Irminger and the Labrador basins (Figure
332 6b). This feature is likely associated with the presence of the Sub-Arctic Front located between 49.5 and 51 °N
333 latitude and 23.5 and 22 °W longitude (Zunino et al., 2017). Indeed, this front which separates cold and fresh
334 water of subpolar origin from warm and salty water of subtropical origin was clearly identifiable by the steep
335 gradient of the isohalines between station 26 and 29; salinity dropping from 35.34 to 35.01 (Figure 2). Lower
336 %PFe_{litho} proxy values, could be associated with higher proportions of PFe from biogenic origin, especially in the
337 case of the LSW.

338

339 4.2. Tracking the different inputs of particulate iron

340 4.2.1. Inputs at margins: Iberian, Greenland and Newfoundland

341 Inputs of iron from continental margin sediments supporting the high productivity found in shallow coastal regions
342 have been demonstrated in the past (e.g. Cullen et al. 2009, Elrod et al. 2004, Jeandel et al. 2011, Ussher et al.
343 2007) and sometimes, were shown to be advected at great distances from the coast (e.g. Lam and Bishop, 2008).
344 In the following section, we will investigate these possible sources in proximity to the different margins
345 encountered.

346

347 *The Iberian margin*

348 The Iberian margin was an important source of lithogenic-derived iron-rich particles to the Atlantic Ocean during
349 GEOVIDE; shelf resuspension impacts were perceptible up to 280 km from the margin (Station 11) in the Iberian
350 Abyssal Plain (Figure 8).

351 On the shelf, at station 2, high sediment resuspension resulted in the low beam transmissometry value (87.6 %) at
352 the immediate vicinity of the seafloor (153 m depth). This sediment resuspension led to an extensive input of
353 lithogenic particles within the water column associated with high concentrations of PFe (304 nmol L⁻¹), PAI (1500
354 nmol L⁻¹), and PMn (2.5 nmol L⁻¹) (Figure 3, Table S1). Moreover, 100 % of PFe was estimated to have a
355 lithogenic origin (Figure 8b) while 100 % of the PMn was estimated to be the result of a recent sediment
356 resuspension according to the %PFe_{litho} and “%bulk sediment Mn” proxies (Figure 8b and c), confirming the
357 resuspended particle input. In addition, ADCP data acquired during GEOVIDE (Zunino et al., 2017) and several
358 other studies have reported an intense current spreading northward coming from the Straits of Gibraltar and
359 Mediterranean Sea, leading to the strong resuspension of benthic sediments above the Iberian Shelf (e.g. Biscaye
360 and Eittrheim, 1977, Eittrheim et al., 1976, McCave and Hall, 2002, Spinrad et al., 1983).

361 At distance from the shelf, within the Iberian Abyssal Plain, an important lateral advection of PFe from the margin
362 was observable (Figure 8a). These lateral inputs occurred at two depth ranges: between 400 and 1000 m as seen
363 at stations 4 and 1, with PFe concentrations reaching 4 nmol L⁻¹, and between 2500 m and the bottom (3575 m)
364 of station 1, with PFe concentrations reaching 3.5 nmol L⁻¹. While 100 % of PFe had a lithogenic signature, the
365 sedimentary source input estimation decreased, between 40 % and 90 % of the PMn (Figure 8b). Transport of
366 lithogenic particles was observable until station 11 (12.2°W) at 2500 m where PFe concentration was 7.74 nmol
367 L⁻¹ and 60 % of PMn had a sedimentary origin (Figure 4). It is noteworthy that no increase in PFe, PMn or PAI
368 was observed between 500 and 2000 m depth, where the MOW spreads (García-Ibáñez et al., 2015). This is
369 consistent with the observed dissolved iron (DFe) concentrations (Tonnard et al., 2018, this issue), yet in contrast
370 to dissolved aluminium (DAI) concentrations (Menzel-Barraqueta et al., 2018, this issue) which were high in the
371 MOW, and with the study of Ohnemus and Lam (2015) that reported a maximum PFe concentration at 695 m
372 depth associated with the particle-rich Mediterranean Overflow Water (Eittrheim et al., 1976) in the IAP. However,
373 their station was located further south of our station 1. The shallower inputs observed at stations 1 and 4 could
374 therefore be attributed to sediment resuspension from the Iberian margin and nepheloid layer at depth for station
375 1.

376 Surface coastal waters of the Iberian Shelf are impacted by the runoff for the Tagus River, which is characterised
377 by high suspended matter discharges, ranging between 0.4 to 1 × 10⁶ tons yr⁻¹, and with a high anthropogenic
378 trace element signature (Jouanneau et al., 1998). During the GEOVIDE section, the freshwater input was
379 observable at stations 1, 2 and 4 in the first 20 m; salinity was below 35.2 while surrounding water masses had

380 salinity up to 35.7. Within the freshwater plume, particulate concentrations were high at station 2 with PFe of 1.83
381 nmol L^{-1} . Further away from the coast, the particulate concentrations remained low at 20m depth, with PFe, PAI,
382 and PMn concentrations of 0.77 nmol L^{-1} , 3.5 nmol L^{-1} , and 0.04 nmol L^{-1} , respectively at station 1. The low
383 expansion of the Tagus plume is likely due to the rapid settling of suspended matter. Indeed, our coastal station 2
384 was located approximately 50 km from the Iberian coast, whereas the surface particle load can only be observed
385 at a maximum distance of 30 km from the Tagus estuary (Jouanneau et al., 1998), Overall, the Iberian margin
386 appears to be an important source of lithogenic-derived iron rich particles to the Atlantic Ocean.

387

388 *South Greenland*

389 During GEOVIDE, the Greenland shelves were a source of particulate-rich meteoric water leading to a transfer
390 of DFe to PFe by enhanced biological activity. Indeed, both East (station 53) and West (station 61) Greenland
391 shelves had high concentration of particles (beam transmissometry of 83 %, Figure 4a) and particulate trace
392 elements, reaching 22.1 nmol L^{-1} (at 100 m depth) and 18.7 nmol L^{-1} (at 136 m depth) of PFe, respectively. Several
393 studies have already demonstrated the importance of icebergs and sea ice melting as sources of dissolved and
394 particulate iron (e.g. van der Merwe et al., 2011a, 2011b; Planquette et al., 2011; Raiswell et al., 2008). The
395 Greenland shelf is highly influenced by external fresh water inputs such as sea-ice-melting or riverine runoff
396 (Fragoso et al., 2016), which are important sources of iron to the Greenland Shelf (Bhatia et al., 2013; Hawkings
397 et al., 2014; Statham et al., 2008). During the cruise, the relative freshwater observed ($S < 33 \text{ psu}$) within the first
398 25 m of stations 53 and 61 was associated with high PFe (19 nmol L^{-1}), PAI (61 nmol L^{-1}), PMn (0.6 nmol L^{-1})
399 and low beam transmissometry ($\leq 85\%$) (Figure 4a and Table S1). The associated particles were enriched in iron
400 compared to aluminium, as PFe/PAI ratio was 0.3 within the meteoric water plume. The high PP concentrations
401 (reaching 197 nmol L^{-1}) resulting from high biological production ($\text{Chl-a} = 6.21 \text{ mg m}^{-3}$ at 24 m at station 61),
402 induced by the supply of bioavailable dissolved iron (surface DFe of 0.79 nM at station 61) from meteoric water
403 (Raiswell et al., 2008; Statham et al., 2008; Tonnard et al. 2018), led to a transfer of DFe to the particulate phase.
404 This is in line with the finding that around 30 % PFe had a non-lithogenic origin. In addition, only 40% PMn
405 originated from resuspended sediments. Interestingly, these two proxies remained constant from the seafloor to
406 the surface (Station 49, Figure 8), with around 25 % of the PMn of sedimentary origin, which could be due to
407 important mixing occurring on the shelf. The lithogenic PFe could result from the release of PFe from Greenland
408 bedrock captured during the ice sheet formation on land.

409 The spatial extent of the off-shelf lateral transport of particles was not important on the east Greenland coast.
410 Indeed, no visible increase of particulate trace metal concentrations was visible at the first station off-shelf, station
411 60 (Figure 8), except at 1000 m depth, where a strong increase (up to 90 %) of sedimentary PMn was seen. This
412 is probably due to the East Greenland Coastal Current (EGCC) that was located at station 53 which constrained
413 these inputs while stations 56 and 60 were under the influence of another strong current, the East Greenland-
414 Irminger current (EGIC) (Zunino et al., 2017). To the west of the Greenland margin, lateral transport of particles
415 was slightly more important. Noticeable concentrations of particulate lithogenic elements were observable until
416 station 64 located 125 km away from shoreline. These particles had a decreased PFe lithogenic contribution (50
417 %) with a similar (25 %) sedimentary PMn content than closer to the margin (Figure 8b and c). The increasing
418 nature of non-lithogenic PFe is linked to the bloom in surface waters (PFe/PAI ratio of $0.30 \text{ mol mol}^{-1}$, PP of 197
419 nmol L^{-1} and Chl-a concentration of 6.21 mg m^{-3} at station 61), with the gravitational settling of biogenic PFe.

420 Therefore, particles newly resuspended from Greenland sediments are an important source, representing around
421 one third of the pMn pool, combined with surface inputs such as riverine runoff and/or ice-melting that are
422 delivering particles on the shelf, and also biological production. Unlike the Iberian shelf, the Greenland margin
423 was not an important provider of particulate metals inside the Irminger and Labrador Basins, due to the circulation
424 that constrained the extent of the margin plume.

425

426

427 *The Newfoundland Shelf*

428 Previous studies have already described the influence of fresh water on the Newfoundland shelf from the Hudson
429 Strait and/or Canadian Arctic Archipelago (Fragoso et al., 2016; Yashayaev, 2007). Yashayaev (2007) also
430 monitored strong resuspension of sediments associated with the spreading of the Labrador Current along the West
431 Labrador margin.

432 Close to the Newfoundland coastline, at station 78, high fresh water discharge (≤ 32 psu) was observed in surface
433 waters (Benetti et al., 2017). Interestingly, these freshwater signatures were not associated with elevated
434 particulate trace metal concentrations. Distance of meteoric water sources implied a long travel time for the water
435 to spread through the Labrador Basin to our sampling stations. Along the journey, particles present originally may
436 have been removed from the water column by gravitational settling.

437 The proportion of lithogenic PFe was relatively high and constant throughout the water column, with a median
438 value of 70 %. At station 78, 95 % of the PMn had a sedimentary origin close to the seafloor (371 m). The
439 spreading of the recent sediment resuspension was observable until 140 m depth where the contribution of
440 sedimentary Mn was still 51% (Figure 8c, Table S2). This could correspond to an intense nepheloid layer as
441 previously reported by Biscaye and Eittrheim (1977) (see also section 3.3.2). The high PFe concentration (184
442 nmol L⁻¹, station 78, 371 m depth, Figure 8b) associated with a high percentage of sedimentary PMn (95%)
443 observed at the bottom of this station, was therefore the result of an important resuspension of shelf sediments.
444 This was confirmed with low transmissometry values of 95 % (Figure 4a).

445

446 Along the GEOVIDE section, continental shelves provided an important load of particles to the surrounding water
447 column. The three margins sampled during GEOVIDE behaved very differently; the Iberian margin discharged
448 high quantities of lithogenic particles far from the coast while the Greenland and Newfoundland margins did not
449 reveal important PFe concentrations. Spreading of particles is tightly linked to hydrodynamic conditions, which
450 in the case of the Greenland margin, prevented long distance seeding of PFe. Moreover, each margin showed a
451 specific PFe/PAI ratio (Figure 9) indicating different composition of the resuspended particles. Resuspended
452 particles represent the composition of sediment at the margin if redox transformation of iron and aluminium are
453 considered negligible under these circumstances. Differences between margins were due to the presence of non-
454 crustal particles, either biogenic or authigenic. Biological production in surface waters and authigenic formation
455 of iron hydroxide produced particles with a higher PFe/PAI content and their export through the water column to
456 the sediment increased the PFe/PAI ratio at depth. Regions where biological production is intense such as in the
457 vicinity of Newfoundland presented higher PFe/PAI ratios of resuspended benthic particles.

458

459

4.2.2 Benthic resuspended sediments

460 Along the GEOVIDE section, Benthic nepheloid layers (BNLs) provided high concentrations of particulate trace
461 elements to the deep open ocean, contributing significantly to the total budget of iron. BNLs were observable in
462 each province, although intensities varied (Figures 3 and 10).

463 In BNLs located within the WEB, PFe concentrations reached up to 10 nmol L^{-1} (stations 26 and 29, Figure 10a;
464 Table S1). These concentrations were lower than PFe concentrations in BNLs from the Icelandic (stations 32 and
465 34), Irminger (stations 42 and 44) and Labrador Basins (stations 68, 69 and 71), where benthic resuspension led
466 to PFe concentrations higher than 40 nmol L^{-1} , even reaching 89 nmol L^{-1} at the bottom of station 71 (3736 m
467 depth). Moreover, in the Irminger and Labrador Basins, PFe/PAI molar ratios within BNLs were higher than the
468 ones measured within the WEB at station 26 and 29. In the Irminger Basin, PFe/PAI reached 0.4 mol mol^{-1} (Figure
469 10b), which could reveal a mixture of lithogenic and biogenic matter that had been previously exported. This
470 feature was also observed in the Labrador Basin, with PFe/PAI ratio ranging between 0.34 and $0.44 \text{ mol mol}^{-1}$. In
471 contrast, BNLs sampled in the WEB clearly have a lithogenic imprint, with PFe/PAI molar ratios close to the
472 crustal one. Resuspended sediments with a non-crustal contribution seem to have higher PFe contents than
473 sediments with lithogenic characteristics. Nevertheless, interestingly all BNLs present during GEOVIDE were
474 spreading identically, with impacts observable up to 200 m above the oceanic seafloor (Figure 10), as reflected in
475 beam transmissometry values, and PFe concentrations, which returned to background levels at 200 m above the
476 seafloor. The presence of these BNLs has also been reported by Le Roy et al. (2018) using radium-226 activity.
477 Important differences of PFe intensities could also be due to different hydrographic components and topographic
478 characteristics. BNLs occur due to strong hydrographic stresses (i.e. boundary currents, benthic storms and deep
479 eddies) interacting with the ocean floor (Biscaye and Eittrheim, 1977; Eittrheim et al., 1976; Gardner et al., 2017,
480 2018). They are, by definition, highly variable geographically and temporally, but we have no physical data which
481 would allow us to investigate this hypothesis further.

482

483 4.2.3. Reykjanes Ridge inputs

484 Above the Reykjanes Ridge (Station 38), high PFe concentrations were determined, reaching 16 nmol L^{-1} just
485 above the seafloor, while increased DFe concentrations were reported to the east of the ridge (Tonnard et al.,
486 2018, this issue). The exact sources of iron-rich particles cannot be well constrained, as they could come from
487 active hydrothermal vents or resuspension of particulate matter from new crustal matter produced at the ridge.
488 According to the oceanic circulation (Zunino et al., 2017; Garcia-Ibanez et al., 2017), hydrothermal particles could
489 have been seen in the ISOW within the Icelandic Basin. Nevertheless, at the vicinity of the ridge, scanning electron
490 microscope (SEM) analyses of our samples did reveal several biological debris and clays but not the presence of
491 iron (oxy-)hydroxide particles (supplementary figure S1), which are known to be produced close to hydrothermal
492 vents (Elderfield and Schultz, 1996). Their absence could thus indicate an absence of vents. However, data from
493 other proxies, such as helium-3, would be necessary to confirm the presence or absence of an hydrothermal source
494 close to station 38.

495

496 4.2.4. Atmospheric inputs

497 Atmospheric deposition is an important source of trace elements in surface of the open ocean (e.g. Jickells et al.,
498 2005). Atmospheric inputs, both wet and dry, were reported to be low during the GEOVIDE cruise (Menzel

499 Barraqueta et al., 2018b, this issue; Shelley et al., 2017; 2018). In fact, oceanic particle measurements in surface
500 waters along the section did not reveal high PFe or PAI concentrations. One pattern is interesting to note: the
501 surface waters of the Iberian Abyssal Plain and Western European Basin, between stations 11 and 23 presented a
502 characteristic feature with really low PFe/PAI elemental ratios, of 0.11, smaller than the UCC ratio of 0.21 (Figure
503 6a). Such low ratios have been reported in the same region by Barrett et al. (2012). One possible explanation is
504 given by Buck et al. (2010) who described Fe-depleted aerosols in this area of the North Atlantic with PFe/PAI
505 ratio below UCC ratio. However, Shelley et al. (2017) found a higher PFe/PAI ratio around 0.25 in this area (their
506 samples geoa5-6). This result, highlights some of the difficulties in linking atmospheric inputs to water column
507 data (Baker et al., 2016), and implies a probable fractionation after aerosol deposition. In addition, there is high
508 spatial and temporal variability of atmospheric deposition (Mahowald et al., 2005) and a certain degree of
509 uncertainty about the dissolution processes of atmospherically-transported particles (Bonnet and Guieu, 2004).

510
511

512 **5. Conclusions**

513

514 The investigation of the PFe composition of suspended particulate matter along the GEOVIDE section in the
515 North Atlantic reflects the pervasive influence of crustal particles, augmented by sedimentary inputs at margins,
516 and within benthic nepheloid layers at depths. In consequence, variance of particulate iron along the section is
517 mainly explained by lithogenic factors.

518 Resuspension of sedimentary particles from continental shelves are responsible of high particulate iron
519 concentrations within the surrounding water column and could be observed at long distances from the shelf, in
520 the case of the Iberian margin. Our results also demonstrate the impact of Arctic meteoric water on the Greenland
521 shelf, while in surface waters, the enhancement of productivity by new bioavailable iron is leading to a transfer
522 of dissolved iron to the particulate phase. Benthic nepheloid layer are providing important concentration of
523 particles to the water column; they were observed in most of the oceanic basin encountered along the GEOVIDE
524 section.

525 Overall, PFe distributions in the North Atlantic are strongly influenced by sources at its boundaries (i.e.
526 continental margins and seafloor). When combined with other datasets from the GEOTRACES program in a
527 modelling study, for example, use of this data will facilitate a greater understanding of particulate iron cycling in
528 the North Atlantic.

529

530

531

532 **Acknowledgments**

533 We are greatly indebted to the captain and crew of the N/O Pourquoi Pas? for their help during the GEOVIDE
534 mission and clean rosette deployment. We would like to give special thanks to Fabien Pérault and Emmanuel de
535 Saint Léger for their technical expertise, to Catherine Schmechtig for the GEOVIDE database management and
536 Greg Cutter for his guidance in setting up the new French clean sampling system. We would like to thank both
537 reviewers for constructive comments that greatly improved this manuscript.

538 We also would like to thank Reiner Schlitzer for the Ocean Data View software (ODV).

539 This work was supported by the French National Research Agency (ANR-13-BS06-0014, ANR-12-PDOC-0025-
540 01), the French National Centre for Scientific Research (CNRS-LEFE-CYBER), the LabexMER (ANR-10-
541 LABX-19), and Ifremer. It was supported for the logistic by DT-INSU and GENAVIR.

542

543 **References**

544 Aguilar-Islas, A. M., Rember, R., Nishino, S., Kikuchi, T. and Itoh, M.: Partitioning and lateral transport of iron
545 to the Canada Basin, *Polar Sci.*, 7(2), 82–99, doi:10.1016/j.polar.2012.11.001, 2013.

546 Baker, A. R., Adams, C., Bell, T. G., Jickells, T. D. and Ganzeveld, L.: Estimation of atmospheric nutrient inputs
547 to the Atlantic Ocean from 50°N to 50°S based on large-scale field sampling: Iron and other dust-associated
548 elements, *Global Biogeochem. Cycles*, 27(3), 755–767, doi:10.1002/gbc.20062, 2013.

549 Baker, A. R., Landing, W. M., Bucciarelli, E., Cheize, M., Fietz, S., Hayes, C. T., Kadko, D., Morton, P. L.,
550 Rogan, N., Sarthou, G., Shelley, R. U., Shi, Z., Shiller, A. and van Hulten, M. M. P.: Trace element and isotope
551 deposition across the air–sea interface: progress and research needs, *Philos. Trans. R. Soc. A Math. Phys. Eng.*
552 *Sci.*, 374(2081), 20160190, doi:10.1098/rsta.2016.0190, 2016.

553 Barrett, P. M., Resing, J. A., Buck, N. J., Buck, C. S., Landing, W. M. and Measures, C. I.: The trace element
554 composition of suspended particulate matter in the upper 1000m of the eastern North Atlantic Ocean: A16N, *Mar.*
555 *Chem.*, 142–144, 41–53, doi:10.1016/j.marchem.2012.07.006, 2012.

556 Benetti, M., Reverdin, G., Lique, C., Yashayaev, I., Holliday, N. P., Tynan, E., Torres-Valdes, S., Lherminier, P.,
557 Tréguer, P., and Sarthou, G.: Composition of freshwater in the spring of 2014 on the southern Labrador shelf and
558 slope, *Journal of Geophysical Research: Oceans*, 122, 1102–1121, 10.1002/2016jc012244, 2017.

559 Berger, C. J. M., Lippiatt, S. M., Lawrence, M. G. and Bruland, K. W.: Application of a chemical leach technique
560 for estimating labile particulate aluminum, iron, and manganese in the Columbia River plume and coastal waters
561 off Oregon and Washington, *J. Geophys. Res.*, 113, C00B01, doi:10.1029/2007JC004703, 2008.

562 Bergquist, B. A., Wu, J. and Boyle, E. A.: Variability in oceanic dissolved iron is dominated by the colloidal
563 fraction, *Geochim. Cosmochim. Acta*, 71(12), 2960–2974, doi:10.1016/j.gca.2007.03.013, 2007.

564 Bhatia, M. P., Kujawinski, E. B., Das, S. B., Breier, C. F., Henderson, P. B. and Charette, M. A.: Greenland
565 meltwater as a significant and potentially bioavailable source of iron to the ocean, *Nat. Geosci.*, 6(4), 274–278,
566 doi:10.1038/ngeo1746, 2013.

567 Biscaye, P. E. and Eitrem, S. L.: Suspended Particulate Loads and Transports in the Nepheloid Layer of the
568 Abyssal Atlantic Ocean, *Dev. Sedimentol.*, 23(C), 155–172, doi:10.1016/S0070-4571(08)70556-9, 1977.

569 Bishop, J. K. B. and Biscaye, P. E.: Chemical characterization of individual particles from the nepheloid layer in
570 the Atlantic Ocean, *Earth Planet. Sci. Lett.*, 58(2), 265–275, doi:10.1016/0012-821X(82)90199-6, 1982.

571 Bishop, J. K. B. and Fleisher, M. Q.: Particulate manganese dynamics in Gulf Stream warm-core rings and
572 surrounding waters of the N.W. Atlantic, *Geochim. Cosmochim. Acta*, 51(10), 2807–2825, doi:10.1016/0016-
573 7037(87)90160-8, 1987.

574 Bonnet, S. and Guieu C.: Dissolution of atmospheric iron in seawater, *Geophys. Res. Lett.*, 31(3), L03303,
575 doi:10.1029/2003GL018423, 2004.

576 Boyle, E. A., Bergquist, B. A., Kayser, R. A. and Mahowald, N.: Iron, manganese, and lead at Hawaii Ocean
577 Time-series station ALOHA: Temporal variability and an intermediate water hydrothermal plume, *Geochim.*
578 *Cosmochim. Acta*, 69(4), 933–952, doi:10.1016/j.gca.2004.07.034, 2005.

579 Buck, C. S., Landing, W. M., Resing, J. A. and Measures, C. I.: The solubility and deposition of aerosol Fe and
580 other trace elements in the North Atlantic Ocean: Observations from the A16N CLIVAR/CO2repeat hydrography
581 section, *Mar. Chem.*, 120(1–4), 57–70, doi:10.1016/j.marchem.2008.08.003, 2010.

582 Cheize, M., Planquette, H. F., Fitzsimmons, J. N., Pelleter, E., Sherrell, R. M., Lambert, C., Bucciarelli, E.,
583 Sarthou, G., Le Goff, M., Liorzou, C., Chéron, S., Viollier, E., and Gayet, N.: Contribution of resuspended
584 sedimentary particles to dissolved iron and manganese in the ocean: An experimental study, *Chemical Geology*.
585 doi: 10.1016/j.chemgeo.2018.10.003, 2018.

586 Collier, R. and Edmond, J.: The trace element geochemistry of marine biogenic particulate matter, *Prog.*
587 *Oceanogr.*, 13(2), 113–199, doi:10.1016/0079-6611(84)90008-9, 1984.

588 Cullen, J. T., Chong, M. and Ianson, D.: British columbia continental shelf as a source of dissolved iron to the
589 subarctic northeast Pacific Ocean, *Global Biogeochem. Cycles*, 23(4), 1–12, doi:10.1029/2008GB003326, 2009.

590 Cutter, G. A. and Bruland, K. W.: Rapid and noncontaminating sampling system for trace elements in global
591 ocean surveys, *Limnol. Oceanogr. Methods*, 10(JUNE), 425–436, doi:10.4319/lom.2012.10.425, 2012.

592 Dammshäuser, A., Wagener, T., Garbe-Schönberg, D. and Croot, P.: Particulate and dissolved aluminum and
593 titanium in the upper water column of the Atlantic Ocean, *Deep. Res. Part I Oceanogr. Res. Pap.*, 73, 127–139,
594 doi:10.1016/j.dsr.2012.12.002, 2013.

595 Dehairs, F., Jacquet, S., Savoye, N., Van Mooy, B. A. S., Buesseler, K. O., Bishop, J. K. B., Lamborg, C. H.,
596 Elskens, M., Baeyens, W., Boyd, P. W., Casciotti, K. L. and Monnin, C.: Barium in twilight zone suspended
597 matter as a potential proxy for particulate organic carbon remineralization: Results for the North Pacific, *Deep.*
598 *Res. Part II Top. Stud. Oceanogr.*, 55(14–15), 1673–1683, doi:10.1016/j.dsr2.2008.04.020, 2008.

599 Dutay, J. C., Tagliabue, A., Kriest, I. and van Hulten, M. M. P.: Modelling the role of marine particle on large
600 scale ^{231}Pa , ^{230}Th , Iron and Aluminium distributions, *Prog. Oceanogr.*, 133, 66–72,
601 doi:10.1016/j.pocean.2015.01.010, 2015.

602 Eitrem, S., Thorndike, E. M. and Sullivan, L.: Turbidity distribution in the Atlantic Ocean, *Deep. Res. Oceanogr.*
603 *Abstr.*, 23(12), 1115–1127, doi:10.1016/0011-7471(76)90888-3, 1976.

604 Elderfield, H. and Schultz, A.: Mid-Ocean Ridge Hydrothermal Fluxes and the Chemical Composition of the
605 Ocean, *Annu. Rev. Earth Planet. Sci.*, 24(1), 191–224, doi:10.1146/annurev.earth.24.1.191, 1996.

606 Ellwood, M. J., Nodder, S. D., King, A. L., Hutchins, D. A., Wilhelm, S. W. and Boyd, P. W.: Pelagic iron cycling
607 during the subtropical spring bloom, east of New Zealand, *Mar. Chem.*, 160, 18–33,
608 doi:10.1016/j.marchem.2014.01.004, 2014.

609 Elrod, V. A., Berelson, W. M., Coale, K. H. and Johnson, K. S.: The flux of iron from continental shelf sediments:
610 A missing source for global budgets, *Geophys. Res. Lett.*, 31(12), 2–5, doi:10.1029/2004GL020216, 2004.

611 Fitzwater, S. E., Johnson, K. S., Gordon, R. M., Coale, K. H. and Smith, W. O.: Trace metal concentrations in
612 the Ross Sea and their relationship with nutrients and phytoplankton growth, *Deep. Res. Part II Top. Stud.*
613 *Oceanogr.*, 47(15–16), 3159–3179, doi:10.1016/S0967-0645(00)00063-1, 2000.

614 Fragoso, G. M., Poulton, A. J., Yashayaev, I. M., Head, E. J. H., Stinchcombe, M. C. and Purdie, D. A.:
615 Biogeographical patterns and environmental controls of phytoplankton communities from contrasting
616 hydrographical zones of the Labrador Sea, *Prog. Oceanogr.*, 141, 212–226, doi:10.1016/j.pocean.2015.12.007,
617 2016.

618 Frew, R. D., Hutchins, D. A., Nodder, S., Sanudo-Wilhelmy, S., Tovar-Sanchez, A., Leblanc, K., Hare, C. E. and
619 Boyd, P. W.: Particulate iron dynamics during FeCycle in subantarctic waters southeast of New Zealand, *Global*
620 *Biogeochem. Cycles*, 20(1), 1–15, doi:10.1029/2005GB002558, 2006.

621 García-Ibáñez, M. I., Pardo, P. C., Carracedo, L. I., Mercier, H., Lherminier, P., Ríos, A. F. and Pérez, F. F.:
622 Structure, transports and transformations of the water masses in the Atlantic Subpolar Gyre, *Prog. Oceanogr.*, 135,
623 18–36, doi:10.1016/j.pocean.2015.03.009, 2015.

624 Gardner, W. D., Tucholke, B. E., Richardson, M. J. and Biscaye, P. E.: Benthic storms, nepheloid layers, and
625 linkage with upper ocean dynamics in the western North Atlantic, *Mar. Geol.*, 385, 304–327,
626 doi:10.1016/j.margeo.2016.12.012, 2017.

627 Gardner, W. D., Richardson, M. J. and Mishonov, A. V.: Global assessment of benthic nepheloid layers and
628 linkage with upper ocean dynamics, *Earth Planet. Sci. Lett.*, 482, 126–134, doi:10.1016/j.epsl.2017.11.008, 2018.

629 Gerringa, L. J. A., Rijkenberg, M. J. A., Schoemann, V., Laan, P. and de Baar, H. J. W.: Organic complexation
630 of iron in the West Atlantic Ocean, *Mar. Chem.*, 177, 434–446, doi:10.1016/j.marchem.2015.04.007, 2015.

631 Hawkins, J. R., Wadham, J. L., Tranter, M., Raiswell, R., Benning, L. G., Statham, P. J., Tedstone, A., Nienow,
632 P., Lee, K. and Telling, J.: Ice sheets as a significant source of highly reactive nanoparticulate iron to the oceans,
633 *Nat. Commun.*, 5(May), 1–8, doi:10.1038/ncomms4929, 2014.

634 Hwang, J., Druffel, E. R. M. and Eglinton, T. I.: Widespread influence of resuspended sediments on oceanic
635 particulate organic carbon: Insights from radiocarbon and aluminum contents in sinking particles, *Global*
636 *Biogeochem. Cycles*, 24(4), 1–10, doi:10.1029/2010GB003802, 2010.

637 Jeandel, C. and Oelkers, E. H.: The influence of terrigenous particulate material dissolution on ocean chemistry
638 and global element cycles, *Chem. Geol.*, 395, 50–66, doi:10.1016/j.chemgeo.2014.12.001, 2015.

639 Jeandel, C., Peucker-Ehrenbrink, B., Jones, M. T., Pearce, C. R., Oelkers, E. H., Godderis, Y., Lacan, F., Aumont,
640 O. and Arsouze, T.: Ocean margins: The missing term in oceanic element budgets?, *Eos, Transactions American*
641 *Geophysical Union*, 92(26), 217–224, doi: 10.1029/2011EO260001, 2011.

642 Jickells, T. D., An, Z. S., Andersen, K. K., Baker, A. R., Bergametti, C., Brooks, N., Cao, J. J., Boyd, P. W., Duce,
643 R. A., Hunter, K. A., Kawahata, H., Kubilay, N., LaRoche, J., Liss, P. S., Mahowald, N., Prospero, J. M.,
644 Ridgwell, A. J., Tegen, I. and Torres, R.: Global iron connections between desert dust, ocean biogeochemistry,
645 and climate, *Science* (80-.), 308(5718), 67–71, doi:10.1126/science.1105959, 2005.

646 Jouanneau, J. M., Garcia, C., Oliveira, A., Rodrigues, A., Dias, J. A. and Weber, O.: Dispersal and deposition of
647 suspended sediment on the shelf off the Tagus and Sado estuaries, S.W. Portugal, *Prog. Oceanogr.*, 42(1–4), 233–
648 257, doi:10.1016/S0079-6611(98)00036-6, 1998.

649 Labatut, M., Lacan, F., Pradoux, C., Chmeleff, J., Radic, A., Murray, J. W., Poitrasson, F., Johansen, A. M., Thil,
650 F., Lacan, F., Pradoux, C., Chmeleff, J., Radic, A., Murray, J. W., Poitrasson, F., Johansen, A. M. and Thil, F.:
651 Iron sources and dissolved-particulate interactions in the seawater of the Western Equatorial Pacific, iron isotope
652 perspectives., *Global Biogeochem Cycles*, 1044–1065, doi:10.1002/2014GB004928, 2014.

653 Lam, P. J. and Bishop, J. K. B.: The continental margin is a key source of iron to the HNLC North Pacific Ocean,
654 *Geophys. Res. Lett.*, 35(7), 1–5, doi:10.1029/2008GL033294, 2008.

655 Lam, P. J., Ohnemus, D. C. and Marcus, M. A.: The speciation of marine particulate iron adjacent to active and
656 passive continental margins, *Geochim. Cosmochim. Acta*, 80, 108–124, doi:10.1016/j.gca.2011.11.044, 2012.

657 Lam, P. J., Ohnemus, D. C. and Auro, M. E.: Size-fractionated major particle composition and concentrations
658 from the US GEOTRACES North Atlantic Zonal Transect, *Deep. Res. Part II Top. Stud. Oceanogr.*, 116, 303–
659 320, doi:10.1016/j.dsr2.2014.11.020, 2015.

660 Lam, P. J., Lee, J. M., Heller, M. I., Mehic, S., Xiang, Y. and Bates, N. R.: Size-fractionated distributions of
661 suspended particle concentration and major phase composition from the U.S. GEOTRACES Eastern Pacific Zonal
662 Transect (GP16), *Mar. Chem.*, (April), 0–1, doi:10.1016/j.marchem.2017.08.013, 2017.

663 Lannuzel, D., Bowie, A. R., van der Merwe, P. C., Townsend, A. T. and Schoemann, V.: Distribution of dissolved
664 and particulate metals in Antarctic sea ice, *Mar. Chem.*, 124(1–4), 134–146, doi:10.1016/j.marchem.2011.01.004,
665 2011.

666 Lannuzel, D., Van der Merwe, P. C., Townsend, A. T. and Bowie, A. R.: Size fractionation of iron, manganese
667 and aluminium in Antarctic fast ice reveals a lithogenic origin and low iron solubility, *Mar. Chem.*, 161, 47–56,
668 doi:10.1016/j.marchem.2014.02.006, 2014.

669 Lee, J. M., Heller, M. I. and Lam, P. J.: Size distribution of particulate trace elements in the U.S. GEOTRACES
670 Eastern Pacific Zonal Transect (GP16), *Mar. Chem.*, 201(September 2017), 108–123,
671 doi:10.1016/j.marchem.2017.09.006, 2017.

672 Lemaître, N., planquette, H., Planchon, F., Sarthou, G., Jacquet, S., Garcia-Ibanez, M. I., Gourain, A., Cheize,
673 M., Monin, L., Andre, L., Laha, P., Terryn, H., and Dehairs, F.: Particulate barium tracing significant mesopelagic
674 carbon remineralisation in the North Atlantic, *Biogeosciences*, doi:10.5194/bg-15-2289-2018, 2018a.

675 Lemaitre, N., Planchon, F., Planquette, H., Dehairs, F., Fonseca-Batista, D., Roukaerts, A., Deman, F., Tang, Y.,
676 Mariez, C., and Sarthou G.: High variability of export fluxes along the North Atlantic GEOTRACES section
677 GA01: Particulate organic carbon export deduced from the ²³⁴Th method, *Biogeosciences*, doi:10.5194/bg-2018-
678 190, 2018b.

679 Le Roy, E., Sanial, V., Charette, M.A., Van Beek, P., Lacan, F., Jacquet, S.H., Henderson, P.B., Souhaut, M.,
680 García-Ibáñez, M.I., Jeandel, C. and Pérez, F.: The ²²⁶Ra-Ba relationship in the North Atlantic during
681 GEOTRACES-GA01, *Biogeosciences*, doi:10.5194/bg-2017-478, 2018.

682 Marsay, C. M., Lam, P. J., Heller, M. I., Lee, J. M. and John, S. G.: Distribution and isotopic signature of ligand-
683 leachable particulate iron along the GEOTRACES GP16 East Pacific Zonal Transect, *Mar. Chem.*, (November
684 2016), 1–14, doi:10.1016/j.marchem.2017.07.003, 2017.

685 Martin, J. H., Fitzwater, S. E., Michael Gordon, R., Hunter, C. N. and Tanner, S. J.: Iron, primary production and
686 carbon-nitrogen flux studies during the JGOFS North Atlantic bloom experiment, *Deep. Res. Part II*, 40(1–2),
687 115–134, doi:10.1016/0967-0645(93)90009-C, 1993.

688 McCave, I. N. and Hall, I. R.: Turbidity of waters over the Northwest Iberian continental margin, *Prog. Oceanogr.*,
689 52(2–4), 299–313, doi:10.1016/S0079-6611(02)00012-5, 2002.

690 Menzel Barraqueta, J.L., Schlosser, C., Planquette, H., Gourain, A., Cheize, M., Boutorh, J., Shelley, R., Pereira
691 Contreira, L., Gledhill, M., Hopwood, M.J. and Lherminier, P.: Aluminium in the North Atlantic Ocean and the
692 Labrador Sea (GEOTRACES GA01 section): roles of continental inputs and biogenic particle removal.
693 *Biogeosciences*, 1-28, doi: 10.5194/bg-2018-39, 2018.

694 Menzel Barraqueta, J.-L., Klar, J. K., Gledhill, M., Schlosser, C., Shelley, R., Planquette, H., Wenzel, B.,
695 Sarthou, G., and Achterberg, E. P.: Atmospheric aerosol deposition fluxes over the Atlantic Ocean: A
696 GEOTRACES case study, *Biogeosciences Discuss.*, <https://doi.org/10.5194/bg-2018-209>, in review, 2018b.

697
698 Milne, A., Schlosser, C., Wake, B. D., Achterberg, E. P., Chance, R., Baker, A. R., Forryan, A. and Lohan, M.
699 C.: Particulate phases are key in controlling dissolved iron concentrations in the (sub)tropical North Atlantic,
700 *Geophys. Res. Lett.*, 44(5), 2377–2387, doi:10.1002/2016GL072314, 2017.

701 Nuester, J., Shema, S., Vermont, A., Fields, D. M. and Twining, B. S.: The regeneration of highly bioavailable
702 iron by meso- and microzooplankton, *Limnol Oceanogr*, 59(4), 1399–1409, doi:10.4319/lo.2014.59.4.1399, 2014.

703 Oelkers, E. H., Jones, M. T., Pearce, C. R., Jeandel, C., Eiriksdottir, E. S. and Gislason, S. R.: Riverine particulate
704 material dissolution in seawater and its implications for the global cycles of the elements, *Geosci.*, 344(11–12),
705 646–651, doi:10.1016/j.crte.2012.08.005, 2012.

706 Ohnemus, D. C. and Lam, P. J.: Cycling of lithogenic marine particles in the US GEOTRACES North Atlantic
707 transect, *Deep. Res. Part II Top. Stud. Oceanogr.*, 116, 283–302, doi:10.1016/j.dsr2.2014.11.019, 2015.

708 Peers, G. and Price, N. M.: A role for manganese in superoxide dismutases and growth of iron-deficient diatoms,
709 *Limnol. Oceanogr.*, 49(5), 1774–1783, doi:10.4319/lo.2004.49.5.1774, 2004.

710 Planquette, H. and Sherrell, R. M.: Sampling for particulate trace element determination using water sampling
711 bottles: Methodology and comparison to in situ pumps, *Limnol. Oceanogr. Methods*, 10(5), 367–388,
712 doi:10.4319/lom.2012.10.367, 2012.

713 Planquette, H., Fones, G. R., Statham, P. J. and Morris, P. J.: Origin of iron and aluminium in large particles (>
714 53 µm) in the Crozet region, Southern Ocean, *Mar. Chem.*, 115(1–2), 31–42, doi:10.1016/j.marchem.2009.06.002,
715 2009.

716 Planquette, H., Sanders, R. R., Statham, P. J., Morris, P. J. and Fones, G. R.: Fluxes of particulate iron from the
717 upper ocean around the Crozet Islands: A naturally iron-fertilized environment in the Southern Ocean, *Global*
718 *Biogeochem. Cycles*, 25(2), doi:10.1029/2010GB003789, 2011.

719 Planquette, H., Sherrell, R. M., Stammerjohn, S. and Field, M. P.: Particulate iron delivery to the water column
720 of the Amundsen Sea, Antarctica, *Mar. Chem.*, 153, 15–30, doi:10.1016/j.marchem.2013.04.006, 2013.

721 Radic, A., Lacan, F. and Murray, J. W.: Iron isotopes in the seawater of the equatorial Pacific Ocean: New
722 constraints for the oceanic iron cycle, *Earth Planet. Sci. Lett.*, 306(1–2), 1–10, doi:10.1016/j.epsl.2011.03.015,
723 2011.

724 Raiswell, R., Benning, L. G., Tranter, M. and Tulaczyk, S.: Bioavailable iron in the Southern Ocean: The
725 significance of the iceberg conveyor belt, *Geochem. Trans.*, 9(1), 7, doi:10.1186/1467-4866-9-7, 2008.

726 Rijkenberg, M. J. A., Middag, R., Laan, P., Gerringa, L. J. A., Van Aken, H. M., Schoemann, V., De Jong, J. T.
727 M. and De Baar, H. J. W.: The distribution of dissolved iron in the West Atlantic Ocean, *PLoS One*, 9(6), 1–14,
728 doi:10.1371/journal.pone.0101323, 2014.

729 Sanders, R., Henson, S. A., Koski, M., De La Rocha, C. L., Painter, S. C., Poulton, A. J., Riley, J., Salihoglu, B.,
730 Visser, A., Yool, A., Bellerby, R. and Martin, A. P.: The Biological Carbon Pump in the North Atlantic, *Prog.*
731 *Oceanogr.*, 129(PB), 200–218, doi:10.1016/j.pocean.2014.05.005, 2014.

732 Sarthou, G., Lherminier, and the GEOVIDE team: Introduction to the French GEOTRACES North Atlantic
733 Transect (GA01): GEOVIDE cruise, *Biogeosciences*, 15, 7097–7109, <https://doi.org/10.5194/bg-15-7097-2018>,
734 2018.

735 Sarthou, G., Vincent, D., Christaki, U., Obernosterer, I., Timmermans, K. R. and Brussaard, C. P. D.: The fate of
736 biogenic iron during a phytoplankton bloom induced by natural fertilisation: Impact of copepod grazing, *Deep.*
737 *Res. Part II Top. Stud. Oceanogr.*, 55(5–7), 734–751, doi:10.1016/j.dsr2.2007.12.033, 2008.

738 Schlosser, C., Schmidt, K., Aquilina, A., Homoky, W. B., Castrillejo, M., Mills, R. A., Patey, M. D., Fielding, S.,
739 Atkinson, A. and Achterberg, E. P.: Mechanisms of dissolved and labile particulate iron supply to shelf waters
740 and phytoplankton blooms off South Georgia, Southern Ocean, *Biogeosciences*, 15, 4973–4993, doi: 10.5194/bg-
741 15-4973-2018, 2018.

742 Shelley, R. U., Landing, W. M., Ussher, S. J., Planquett, H. and Sarthou, G.: Characterisation of aerosol
743 provenance from the fractional solubility of Fe (Al, Ti, Mn, Co, Ni, Cu, Zn, Cd and Pb) in North Atlantic aerosols
744 (GEOTRACES GA01 and GA03), *Biogeosciences*, 15, 2271–2288, doi: 10.5194/bg-15-2271-2018, 2018

745 Shelley, R. U., Landing, W. M., Ussher, S. J., Planquette, H. and Sarthou, G.: Regional trends in the fractional
746 solubility of Fe and other metals from North Atlantic aerosols (GEOTRACES cruises GA01 and GA03) following
747 a two-stage leach, *Biogeosciences*, 15(1), 2271–2288, doi:10.5194/bg-15-2271-2018, 2018.

748 Sherrell, R. M., Field, P. M. and Gao, Y.: Temporal variability of suspended mass and composition in the
749 Northeast Pacific water column: Relationships to sinking flux and lateral advection, *Deep. Res. Part II Top. Stud.*
750 *Oceanogr.*, 45(4–5), 733–761, doi:10.1016/S0967-0645(97)00100-8, 1998.

751 Spinrad, R. W., Zaneveld, J. R. and Kitchen, J. C.: A Study of the Optical Characteristics of the Suspended Particles
752 Benthic Nepheloid Layer of the Scotian Rise, *J. Geophys. Res.*, 88, 7641–7645, doi:10.1029/83J003C, 1983.

753 Statham, P. J., Skidmore, M. and Tranter, M.: Inputs of glacially derived dissolved and colloidal iron to the coastal
754 ocean and implications for primary productivity, *Global Biogeochem. Cycles*, 22(3), 1–11,
755 doi:10.1029/2007GB003106, 2008.

756 Straneo, F., Pickart, R. S. and Lavender, K.: Spreading of Labrador sea water: An advective-diffusive study based
757 on Lagrangian data, *Deep. Res. Part I Oceanogr. Res. Pap.*, 50(6), 701–719, doi:10.1016/S0967-0637(03)00057-
758 8, 2003.

759 Sunda, W. G. and Huntsman, S. A.: Effect of Competitive Interactions Between Manganese and Copper on
760 Cellular Manganese and Growth in Estuarine and Oceanic Species of the Diatom *Thalassiosira*, *Limnol.*
761 *Oceanogr.*, 28(5), 924–934, doi:10.4319/lo.1983.28.5.0924, 1983.

762 Tagliabue, A., Bopp, L., Dutay, J. C., Bowie, A. R., Chever, F., Jean-Baptiste, P., Bucciarelli, E., Lannuzel, D.,
763 Remenyi, T., Sarthou, G., Aumont, O., Gehlen, M. and Jeandel, C.: Hydrothermal contribution to the oceanic
764 dissolved iron inventory, *Nat. Geosci.*, 3(4), 252–256, doi:10.1038/ngeo818, 2010.

765 Tagliabue, A., Bowie, A. R., Boyd, P. W., Buck, K. N., Johnson, K. S. and Saito, M. A.: The integral role of iron
766 in ocean biogeochemistry, *Nature*, 543(7643), 51–59, doi:10.1038/nature21058, 2017.

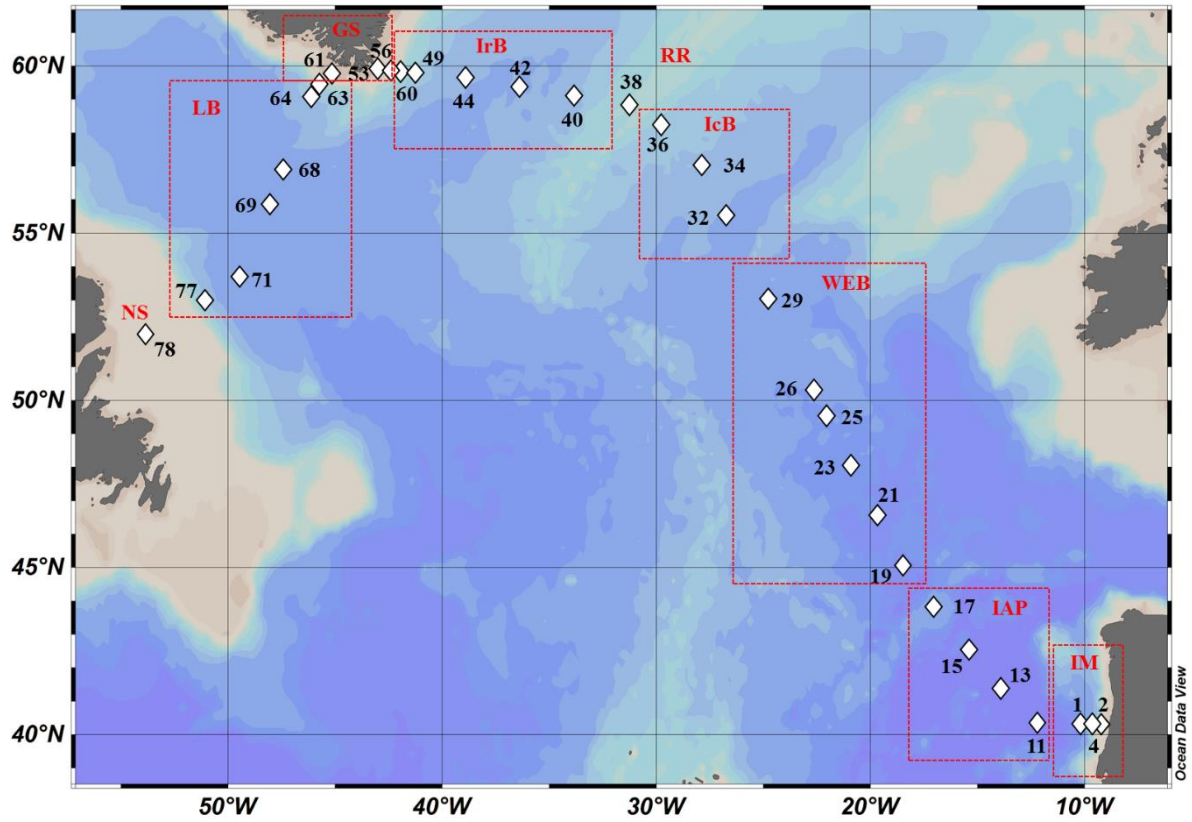
- 767 Taylor, S. and McLennan, S.: The geochemical evolution of the continental crust, *Rev. Geophys.*, 33(2), 241–
768 265, doi:10.1029/95RG00262, 1995.
- 769 Tebo, B. M. and Emerson, S. R.: Effect of Oxygen Tension Manganese (II) Concentration and Temperature on
770 the Microbially Catalyzed Manganese-Ii Oxidation Rate in a Marine Fjord, *Appl. Environ. Microbiol.*, 50(5),
771 1268–1273, 1985.
- 772 Tebo, B. M., Neelson, K. H., Emerson, S. and Jacobs, L.: Microbial mediation of Mn(II) and Co(II) precipitation
773 at the o₂/H₂S interfaces in two anoxic fjords, 29(6), 1247–1258, 1984.
- 774 Tonnard, M., Planquette, H., Bowie, A. R., van der Merwe, P., Gallinari, M., Desprez de Gésincourt, F., Germain,
775 Y., Gourain, A., Benetti, M., Reverdin, G., Tréguer, P., Boutorh, J., Cheize, M., Menzel Barraqueta, J.L., Pereira-
776 Contreira, L., Shelley, R., Lherminier, P., and Sarthou, G.: Dissolved iron in the North Atlantic Ocean and
777 Labrador Sea along the GEOVIDE section (GEOTRACES section GA01), *Biogeosciences Discuss.*,
778 <https://doi.org/10.5194/bg-2018-147>, 2018
- 779 Trefry, J. H., Trocine, R. P., Klinkhammer, G. P. and Rona, P. A.: Iron and copper enrichment of suspended
780 particles in dispersed hydrothermal plumes along the mid-Atlantic Ridge, *Geophys. Res. Lett.*, 12(8), 506–509,
781 doi:10.1029/GL012i008p00506, 1985.
- 782 Ussher, S. J., Achterberg, E. P. and Worsfold, P. J.: Marine biogeochemistry of iron, *Environ. Chem.*, 1(2), 67–
783 80, doi:10.1071/EN04053, 2004.
- 784 Ussher, S. J., Worsfold, P. J., Achterberg, E. P., LaëS, A., Blain, S., Laan, P., de Baar, H. J. W.: Distribution and
785 redox speciation of dissolved iron on the European continental margin, *Limnol. Oceanogr.*, 52(6), 2530–2539,
786 doi:10.4319/lo.2007.52.6.2530, 2007.
- 787 Van der Merwe, P., Lannuzel, D., Bowie, A. R., Mancuso Nichols, C. A. and Meiners, K. M.: Iron fractionation
788 in pack and fast ice in East Antarctica: Temporal decoupling between the release of dissolved and particulate iron
789 during spring melt, *Deep. Res. Part II Top. Stud. Oceanogr.*, 58(9–10), 1222–1236,
790 doi:10.1016/j.dsr2.2010.10.036, 2011a.
- 791 Van Der Merwe, P., Lannuzel, D., Bowie, A. R. and Meiners, K. M.: High temporal resolution observations of
792 spring fast ice melt and seawater iron enrichment in East Antarctica, *J. Geophys. Res. Biogeosciences*, 116(3), 1–
793 18, doi:10.1029/2010JG001628, 2011b.
- 794 Weinstein, S. E. and Moran, S. B.: Distribution of size-fractionated particulate trace metals collected by bottles
795 and in-situ pumps in the Gulf of Maine-Scotian Shelf and Labrador Sea, *Mar. Chem.*, 87(3–4), 121–135,
796 doi:10.1016/j.marchem.2004.02.004, 2004.
- 797 Yashayaev, I.: Hydrographic changes in the Labrador Sea, 1960-2005, *Prog. Oceanogr.*, 73(3–4), 242–276,
798 doi:10.1016/j.pocean.2007.04.015, 2007.

799 Yashayaev, I. and Loder, J. W.: Enhanced production of Labrador Sea Water in 2008, *Geophys. Res. Lett.*, 36(1),
800 doi:10.1029/2008GL036162, 2009.

801 Zunino, P., Lherminier, P., Mercier, H., Danialt, N., García-Ibáñez, M. I., and Pérez, F. F.: The GEOVIDE
802 cruise in May–June 2014 reveals an intense Meridional Overturning Circulation over a cold and fresh subpolar
803 North Atlantic. *Biogeosciences*, 14(23), 5323, 2017.

804
805
806
807
808
809
810
811
812
813
814
815
816
817
818
819
820
821
822
823
824
825
826
827
828
829
830
831
832
833
834
835
836
837

838 Figure 1: Map of stations where suspended particle samples were collected with GO-FLO bottles during the GEOVIDE
 839 cruise (GA01) in the North Atlantic Ocean. Biogeochemical provinces are indicated by red squares, IM: Iberian
 840 Margin, IAP: Iberian Abyssal Plain, WEB: Western European Basin, IcB: Iceland Basin, RR: Reykjanes Ridge, IrB:
 841 Irminger Basin, GS: Greenland Shelf, LB: Labrador Basin, NS: Newfoundland Shelf. This figure was generated using
 842 Ocean Data View (Schlitzer, R., Ocean Data View, odv.awi.de, 2017).

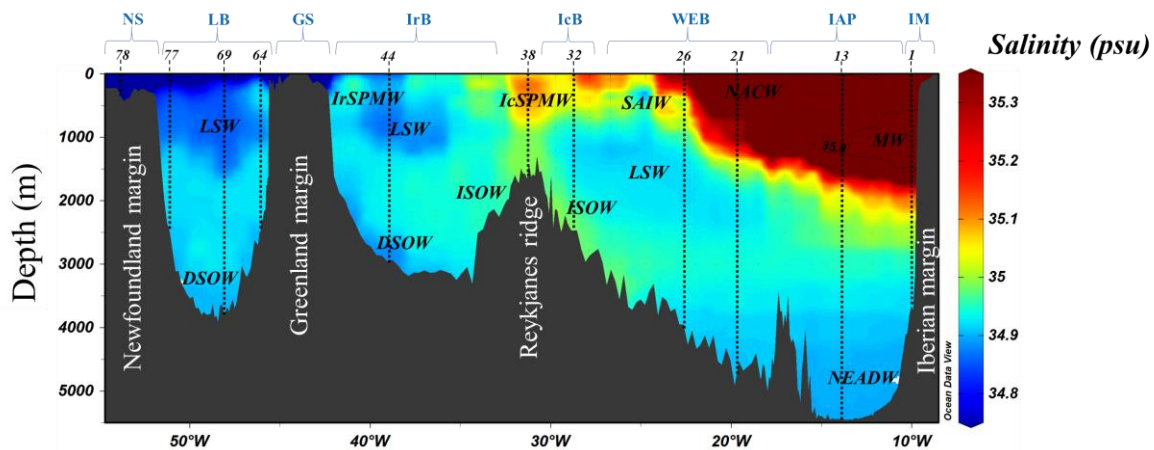


843
 844
 845
 846
 847
 848
 849
 850
 851
 852
 853
 854
 855
 856
 857
 858

859 Figure 2: Salinity section during the GEOVIDE cruise with water masses indicated in black italic font. A salinity
 860 contour of 35.8 psu have been applied to identify the Mediterranean Water (MW) to the east. From right to left; North
 861 East Atlantic Deep Water (NEADW); North Atlantic Central Water (NACW); Labrador Sea Water (LSW); Sub-
 862 Arctic Intermediate Water (SAIW); Iceland-Scotland Overflow Water (ISOW); Iceland Sub-Polar Mode Water
 863 (IcSPMW); Denmark Strait Overflow Water (DSOW); Irminger Sub-Polar Mode Water (IrSPMW). Station locations
 864 are indicated by the numbers above the section and biogeochemical provinces are indicated in blue font above station
 865 numbers. This figure was generated using Ocean Data View (Schlitzer, R., Ocean Data View, odv.awi.de, 2017).

866

867



868

869

870

871

872

873

874

875

876

877

878

879

880

881

882

883

884

885

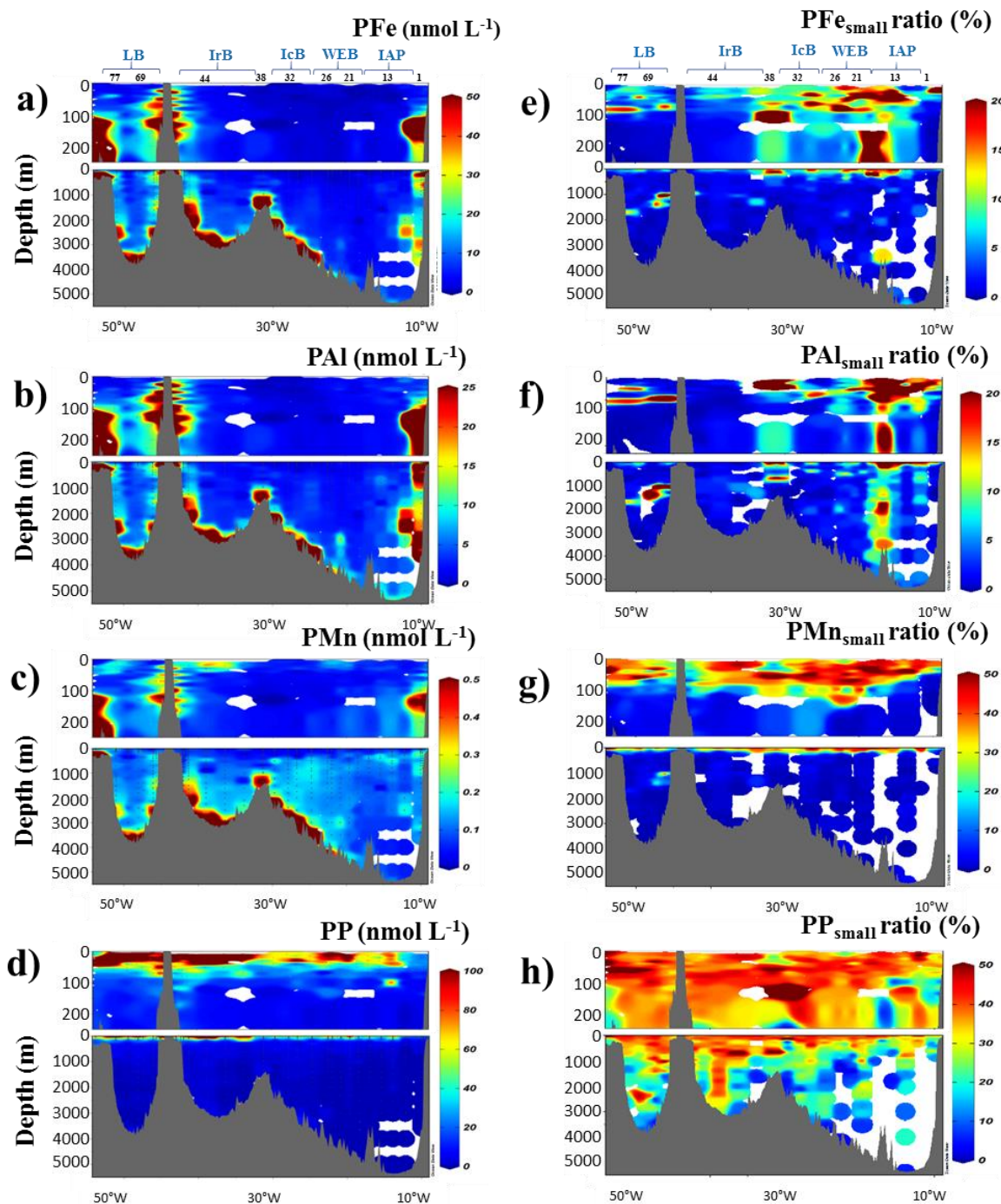
886

887

888

889
890
891
892
893
894

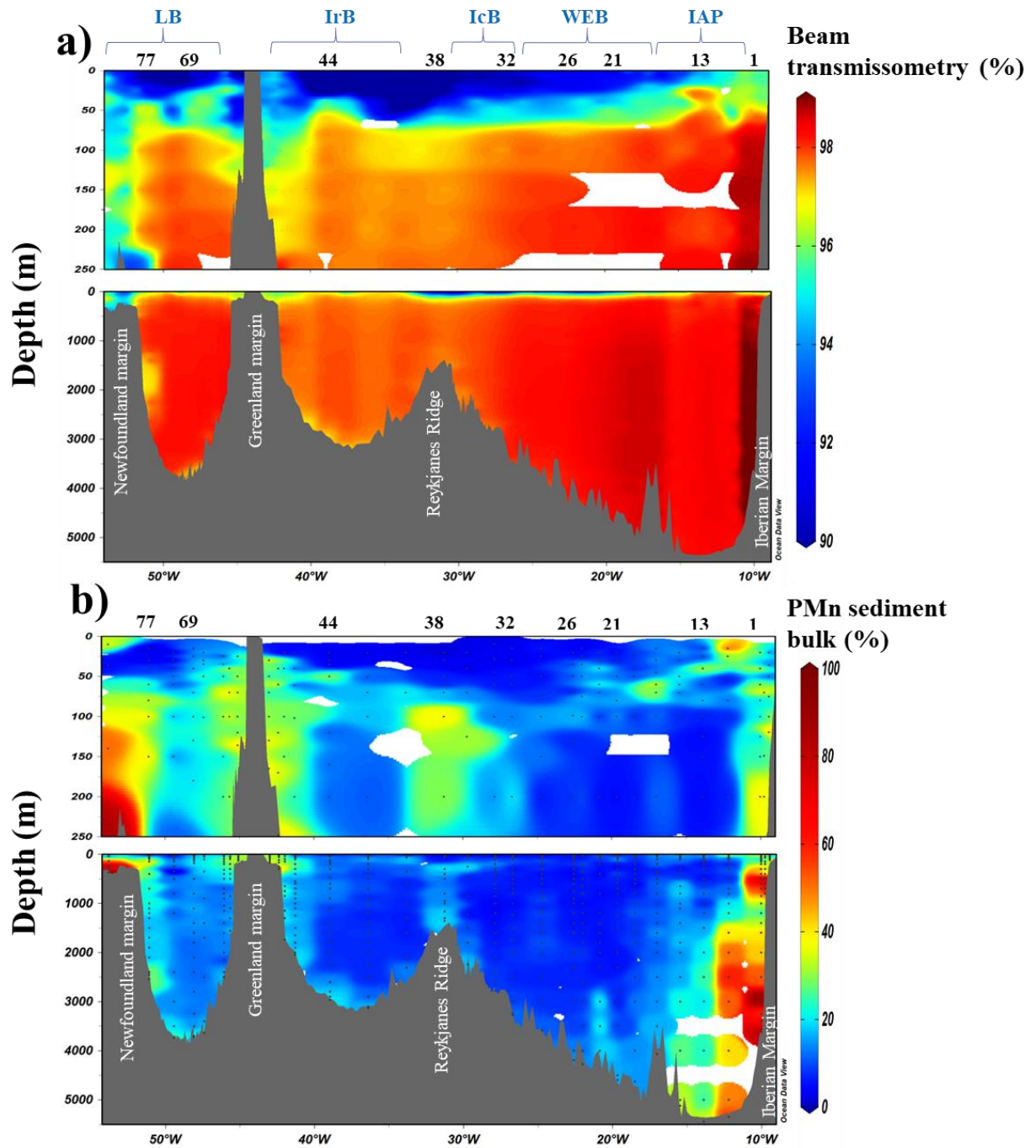
Figure 3: Left panel: Distribution of total particulate (a) iron (PFe), (b) aluminium (PAI), (c) manganese (PMn) and (d) phosphorus (PP) concentrations (nmol L^{-1}) in the first 250 m and the entire water column along the GEOVIDE section in the North Atlantic Ocean. Right panel: Contribution of the small size fraction ($0.45\text{-}5\ \mu\text{m}$) expressed as a percentage (%) of the total concentration of (e) PFe, (f) PAI, (g) PMn and (h) PP. Station IDs and biogeochemical regions are indicated on top of sections a and e. This figure was generated using Ocean Data View (Schlitzer, R., Ocean Data View, odv.awi.de, 2017).



895
896
897
898

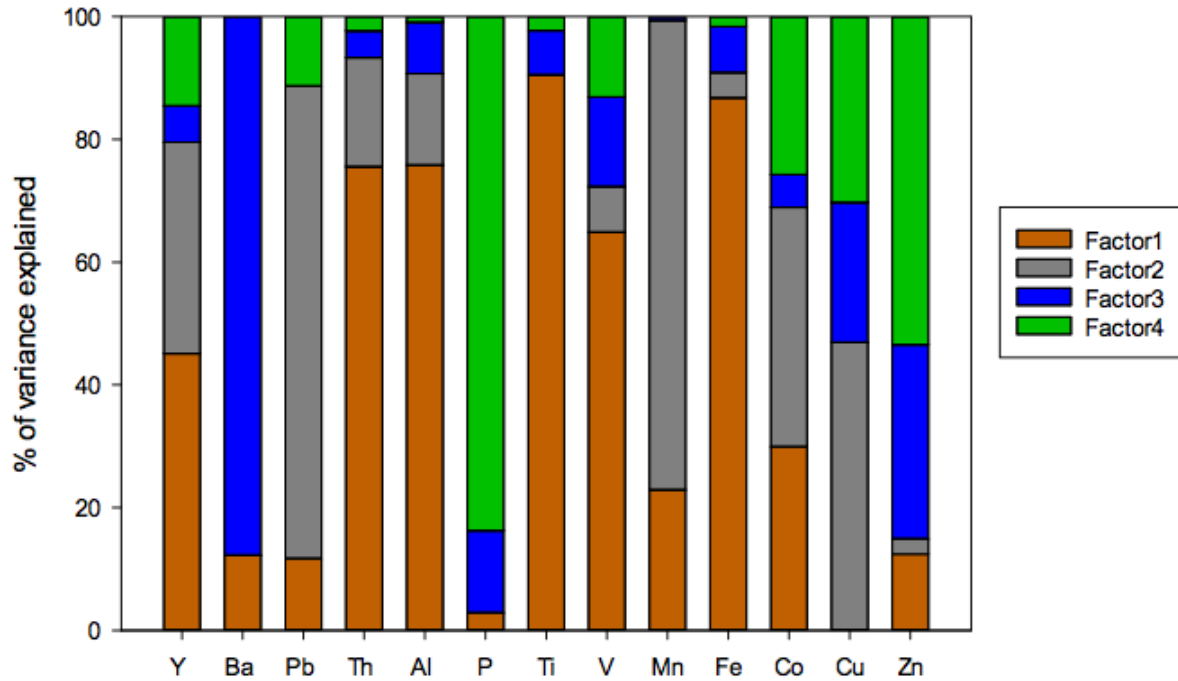
899
900
901
902

Figure 4: Section of derived contributions of sedimentary inputs along the GA01 section with (a) beam transmissometry (%) and (b) manganese bulk sediment proxy (%) based on Eq (3). Station IDs and biogeochemical region are indicated above section (a) in black numbers and blue letters, respectively. This figure was generated using Ocean Data View (Schlitzer, R., Ocean Data View, odv.awi.de, 2017).



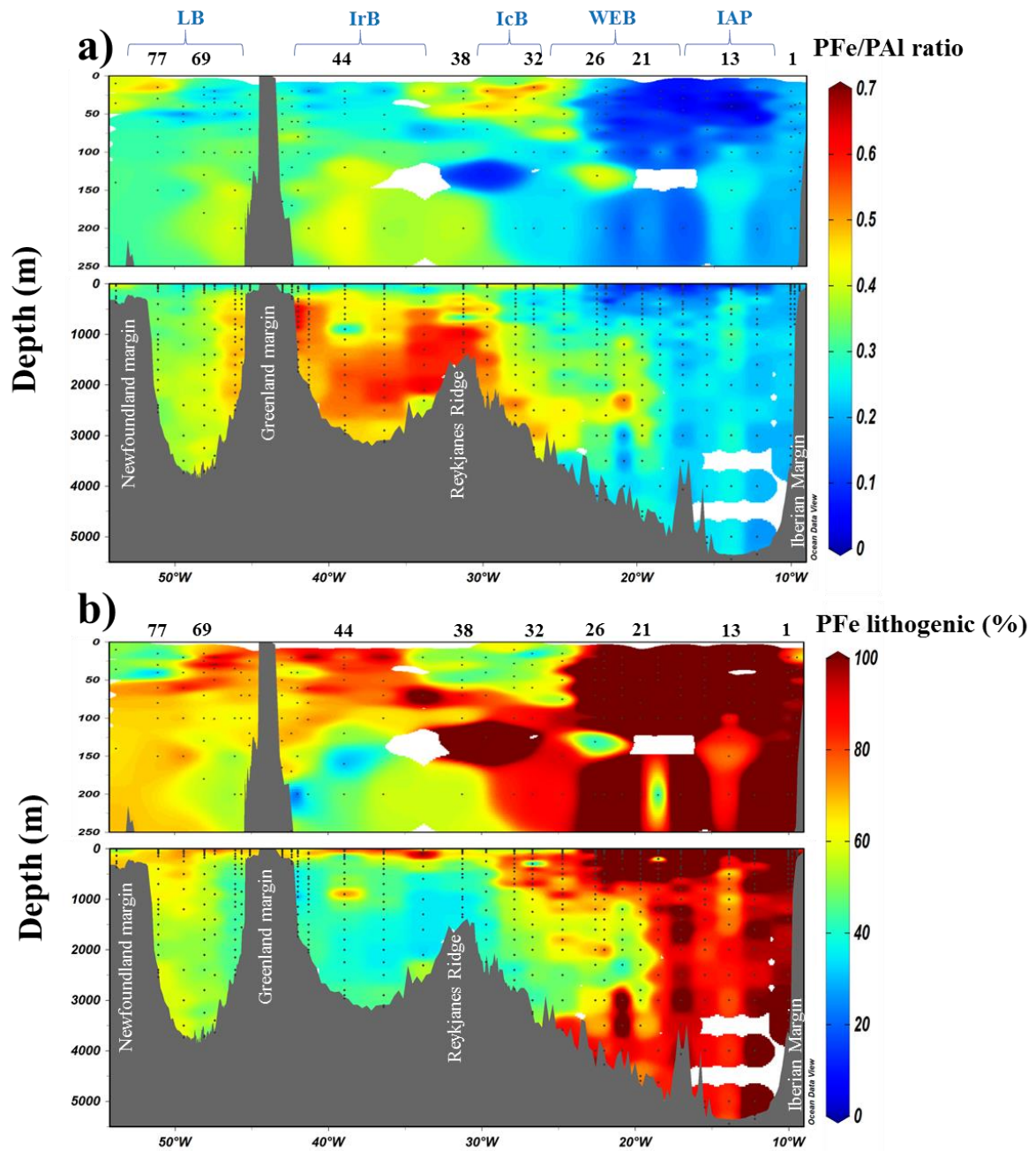
903
904
905
906
907
908

909 **Figure 5: Factor fingerprint of the positive matrix factorisation conducted on 445 particles samples collected along the**
 910 **GA01 section. The four main factors influencing the particulate trace element variance are represented in a stacked**
 911 **bar chart of the percentage of variance explained per element. Factor 1 is dominated by the lithogenic elements, e.g.**
 912 **Th, Al, Ti and Fe. Factor 2 is associated with Pb and Mn variances. Biogenic barite formation mainly influences factor**
 913 **3. Factor 4 is dominated by biogenic elements, e.g. P, Co, Cu and Zn.**



914
 915
 916
 917
 918
 919
 920
 921
 922
 923
 924
 925
 926
 927
 928
 929
 930

931 Figure 6: a) Section of the PFe to PAI molar ratio (mol mol^{-1}) during the GEOVIDE cruise (GA01) and (b) contribution
 932 (%) of lithogenic particulate iron ($\text{PFe}_{\text{Litho}}$) based on Eq. (1). Station IDs and biogeochemical provinces are indicated
 933 above each section in black numbers and blue letters, respectively. This figure was generated using Ocean Data View
 934 (Schlitzer, R., Ocean Data View, odv.awi.de, 2017).



935

936

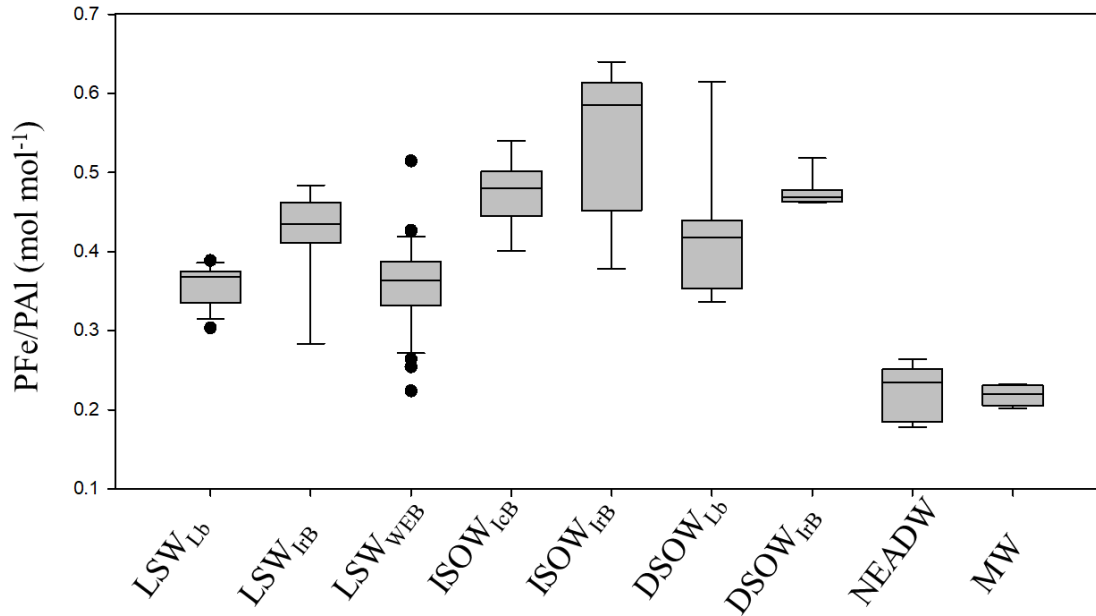
937

938

939

940

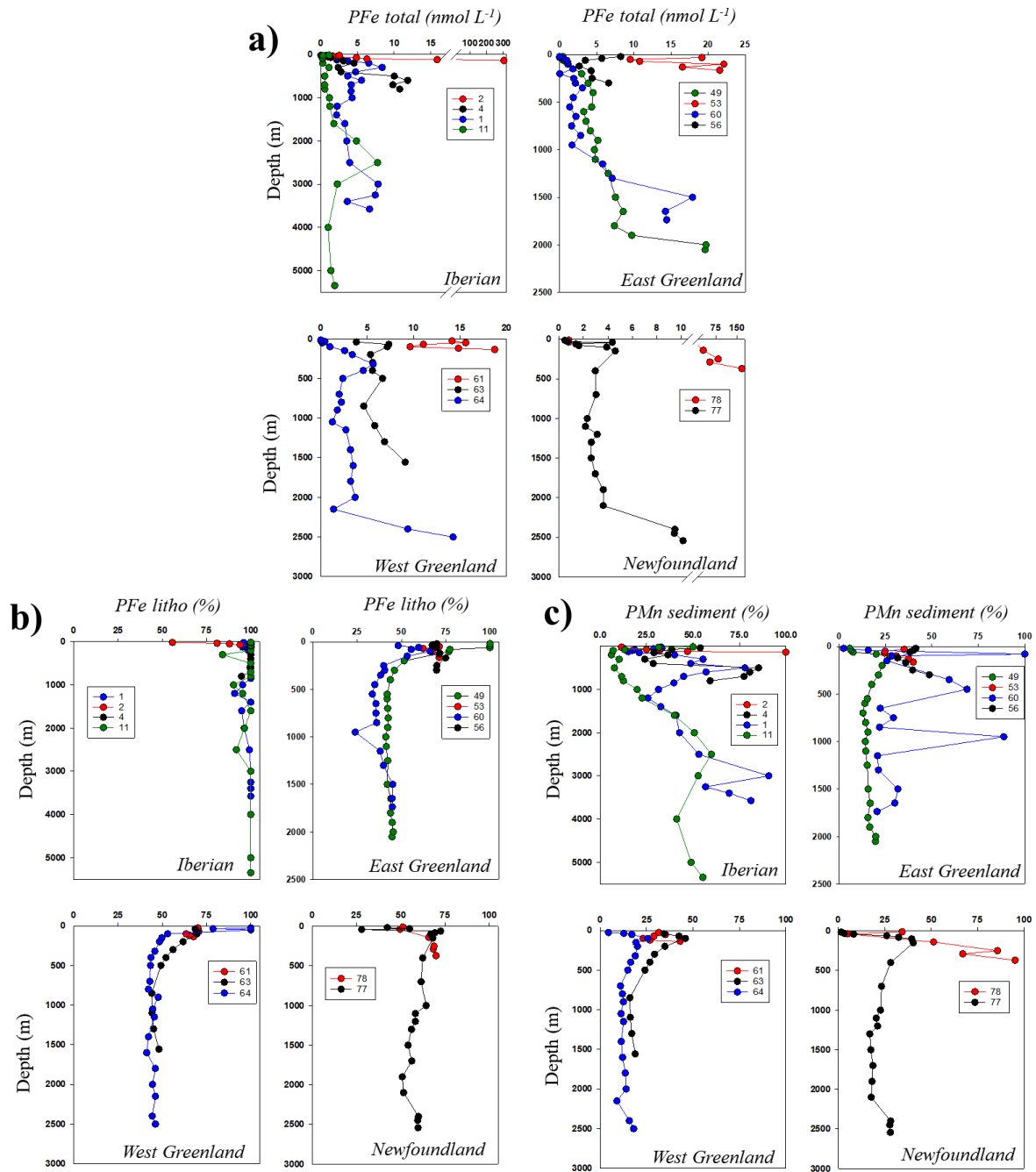
941 **Figure 7: Box and whisker diagram of PFe/PAI molar ratio in nine water masses sampled along the GA01 section in**
 942 **the North Atlantic Ocean. Water masses are defined in section 3.1 and in Figure 2. The PFe/PAI median values for each**
 943 **water masses with the biogeochemical provinces in subscript were as follows: LSW_{LB} = 0.37; LSW_{IFB} = 0.44; LSW_{WEB}**
 944 **= 0.36; ISOW_{ICB} = 0.48; ISOW_{IFB} = 0.58; DSOW_{LB} = 0.42; DSOW_{IFB} = 0.47; NEADW_{IAP} = 0.23; MW = 0.22 mol mol⁻¹.**
 945 **The difference in PFe/PAI between water masses is statistically significant (Kruskal-Wallis test; p = < 0.001 excluding**
 946 **water masses for which we had less than 5 data points for PFe/PAI). Noted that the UCC PFe/PAI ratio reported from**
 947 **Taylor and McLennan, (1995) is 0.21 mol mol⁻¹.**
 948



949
 950
 951
 952
 953
 954
 955
 956
 957
 958
 959
 960
 961
 962
 963
 964
 965

966
967
968

Figure 8: Vertical profiles of (a) PFe (nmol L⁻¹), (b) lithogenic proportion of particulate iron (PFe_{litho}, %) and (c) sedimentary proportion of particulate manganese (PMn sediment, %) at the Iberian, East-West Greenland and Newfoundland margins.



969

970

971

972

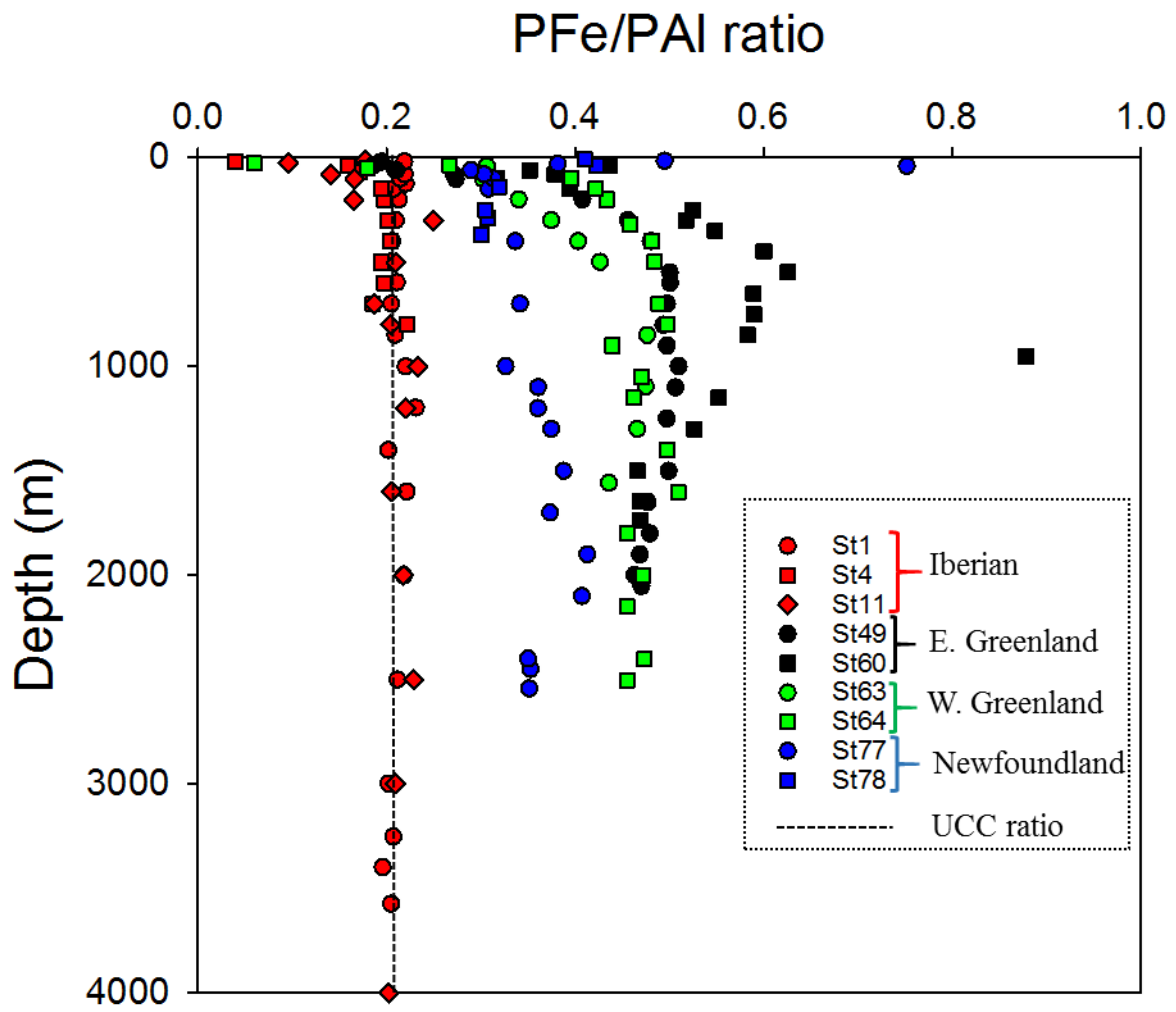
973

974

975

976
977
978

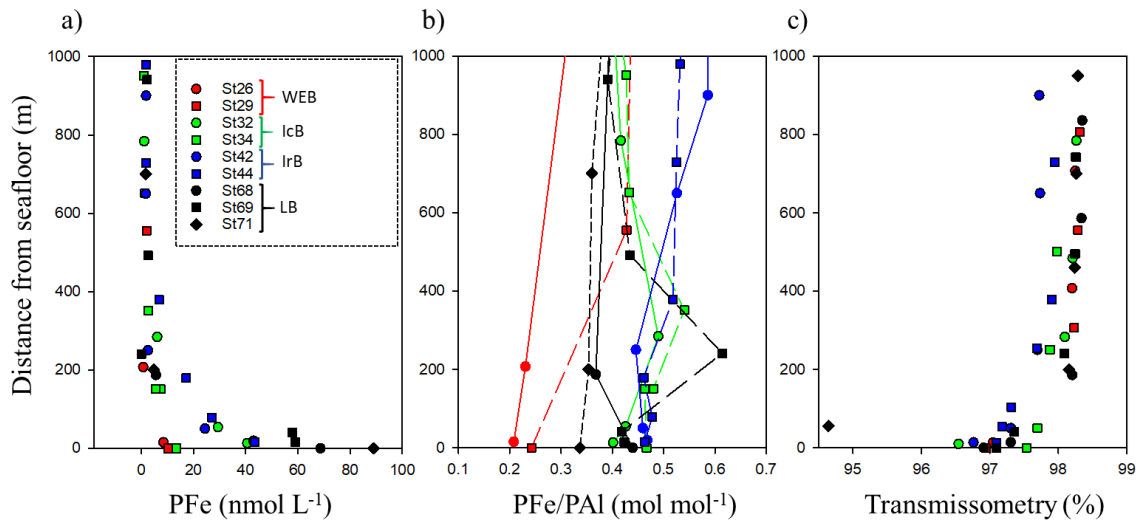
Figure 9: Scatter of the PFe/PAI ratio at the Iberian (red dots), East Greenland (black dots), West Greenland (green dots) and Newfoundland margins (blue dots). Dashed line indicate the UCC PFe/PAI ratio (Taylor and McLennan, 1995).



979
980
981
982
983
984
985
986
987
988
989

990 **Figure 10: Benthic Nepheloid Layers (BNLs) encountered along the GA01 section and observed through a) PFe total;**
 991 **b) PFe/PAI ratio and c) beam transmissometry (%) as a function of depth above the seafloor (m) at selected stations**
 992 **where a decrease in transmissometry was recorded in the West European (red dots), the Iceland (green dots), the**
 993 **Irminger (blue dots) and the Labrador Basins. Noted that the UCC PFe/PAI ratio reported from Taylor and McLennan,**
 994 **(1995) is 0.21 mol mol⁻¹.**

995



996

997

998

999

1000

1001

1002

1003

1004

1005

1006

1007

1008

1009

1010

1011

		Fe	Al	P	Mn
Blank (nmol L ⁻¹)	5µm filter	0.072	0.100	0.511	0.003
	0.45µm filter	0.132	0.164	1.454	0.005
Limit of detection (nmol L ⁻¹)	5µm filter	0.011	0.030	0.365	0.001
	0.45µm filter	0.026	0.046	1.190	0.001
Recovery CRM (%)	BCR-414 (n=10)	88 ± 7			94 ± 7
	MESS-4 (n=5)	98 ± 14	97 ± 14	80 ± 30	110 ± 18
	PACS-3 (n=8)	101 ± 9	99 ± 14	91 ± 34	112 ± 11

1012

1013 **Table 1: Blank and limit of detection (nmol L⁻¹) of the two filters and certified reference material (CRM)**
1014 **recoveries during GEOVIDE suspended particle digestions.**

1015

1016

1017

1018

1019

1020

Location	Depth range	PFe	PAI	PMn	PP	Fraction	Author	Year
N. Atlantic (>40°N)	All	bdl-304	bdl-1544	bdl-3.5	bdl-402	>0.45µm	This study	
Labrador Sea	0-250	0.1-1.2	0.1-1.5			>53 µm	Weinstein et al.	2004
Labrador Sea	0-250	2.5	3.6	0.05		0.4– 10µm	Weinstein et al.	2004
N. Atlantic (25-60°N)	Upper 1000m	0.29-1.71	0.2-19.7			0.4µm	Barrett et al.	2012
N. Atlantic	All	0-938	0-3600			0.8–51 µm	Ohnemus et al.	2015
Gulf of Maine	0-300	34.8	109			>0.4 µm	Weinstein et al.	2004
Eastern tropical N.A.	0-200		0.59-17.7			>0.2 µm	Dammshausser et al.	2013
Eastern tropical N.A.	0-600	ND-12				1–51 µm	Lam et al.	2012
Sub-tropical N.A.	All	ND-140	ND-800			>0.45µm	Milne et al.	2017
Meridional Atlantic	0-200		0.35-16.1			>0.2 µm	Dammshausser et al.	2013
Northeast Pacific	0-3557		0.0-54.2			1-53µm	Sherrell et al.	1998
Eastern tropical S.Pacific	All	bdl-159	bdl-162	bdl-8.7	bdl-983	>0.8 µm	Lee et al.	2017
South Georgia Shelf	All	0.87-267	0.6-195	0.01-3.85		>1 µm	Schlosser et al.	2017
Southern Ocean	30-340	0.15–13.2	0.11–25.5			>53 µm	Planquette et al.	2009
East Antarctic	Surface		0.02-10.67	0.01-0.14		>0.2 µm	Lannuzel et al.	2011
East Antarctic	Fast ice	43-10385	121-31372	1-307		>0.2 µm	Lannuzel et al.	2014
Ross Sea	All	0.68-57.3	ND-185	ND-1.4	5.4-404	>0.4 µm	Marsay et al.	2017

1021

1022 **Table 2: Concentration (in nmol L⁻¹) of particulate trace elements (PFe, PAI, PMn and PP) in suspended**
1023 **particles collected in diverse regions of the world's ocean. Bdl: below detection limit, ND: non-determined.**

1024

1025