

The reviewer is thanked for their time and detailed comments on the text.

On page 3, line 20: In these mesocosm experimental setup, HDPE containers were used. H₂O₂ is mainly produced through UV oxidation of dissolved organic material. How much UV radiation can be transmitted through the HDPE wall of these containers?

Given the thickness of any plastic ‘tanks’ UV should have been strongly attenuated with little transmission through the HDPE wall. This is now explicitly stated in the text. Line added: ‘In all cases, these HDPE containers likely strongly attenuated natural UV radiation, compared to ambient waters, which is expected to negatively affect photochemical formation of H₂O₂ (Cooper et al., 1988, 1994).’

On page 6, line 30: When the artificial light source is used in the experiment. How much UV light is provided by these artificial light sources?

We did not quantify this explicitly relative to ambient light, but based on the lamp’s spectral distribution, they are deficient in UV, especially compared to sunlight at low latitudes. We discuss this briefly later, but also now flag here as well, ‘Whilst the light condition for these experiments was selected to approximate the light intensity of ambient near-surface waters, the synthetic lighting is deficient in UV relative to ambient sunlight-especially at low latitudes.’

On page 7, line 20: I believe residue catalase on containers can affect H₂O₂ sampling and determinations, what cautious steps were taken to avoid that?

This is possible, but we did not see any evidence of analytical issues herein (which would have been evident in our standard additions of H₂O₂ during calibration). The plasticware used to prepare catalase was discarded, and our measurements (in catalase treated incubations) were made >12 h after any manipulation with catalase. Any added catalase was likely largely de-activated between addition and H₂O₂ measurements (catalase additions occurred at sunset, measurements of H₂O₂, as reported, were made the following day). An extra line of detail is added. ‘Plasticware used to handle catalase was discarded. No adverse effects of measuring H₂O₂ in catalase-manipulated solutions were found. As H₂O₂ measurements were made >12 h after catalase addition, this may reflect catalase de-activation under the incubated conditions.’

On page 8, line 11: Fe(II) can cause severe interferences in luminol based chemiluminescence determination of H₂O₂. For discrete samples, a short delay from sampling to determination may be enough time for Fe(II) to decay away, what measures were taken to avoid Fe(II) interferences in continuous sampling and determinations?

Yes this is certainly the case. In a companion text we describe Fe(II) concentrations for the same mesocosms discussed herein which included measurements of Fe(II) in ambient waters. For continuous measurements of H₂O₂ ‘in situ’, the apparatus had a loop to induce a delay of 1-2 minutes in the dark. Any short-lived radical species would thereby be reduced in concentration prior to measurement of H₂O₂. Fe(II) was below detection (<0.2 nM) in all the Mediterranean experiments discussed herein. In Patagonia/Svalbard/Gran Canaria, Fe(II) was measurable in situ and a residual Fe(II) signal was present in the dark-especially in Patagonia. Without any Fe(II) decay over 1-2 minutes, the maximum possible over-estimate of H₂O₂ in-situ would be <10%. We explicitly flag this uncertainty, it is challenging to be more accurate as Fe(II) and H₂O₂ are difficult to determine at the exact same time. ‘Fe(II) and photochemically generated radical species can interfere with the luminol based chemiluminescence used to determine H₂O₂. In batch measurements, the >15 minute time delay between sample collection analysis (during which time

the same in not exposed to light) is likely sufficient to minimize interference from these species which have much shorter half-lives and lower concentrations than H₂O₂. For in situ, continuous measurements, a sample loop was intentionally introduced such that seawater was displaced from ambient light for 2 minutes prior to analysis. Some residual Fe(II) may have therefore remained leading to over-estimation of H₂O₂ concentrations. As Fe(II) concentrations were quantified in all of the experiments and ambient waters described herein, we can determine that the maximum possible over-estimate of H₂O₂ concentrations for ambient waters during continuous analysis was <10%.'

On page 11, line 21: the artificial light source may not have provided enough UV radiation to match that of sunlight.

Ammended 'with natural lighting and especially higher UV light exposure'

On page 14, line 9: "diel" should be used in this situation instead of "diurnal".

Ammended.

On page 21, line 6: UV transparency of HDPE and polyurethane may explain the differences here.

Added: 'Whilst not explicitly measured, a likely factor to explain this difference is the reduced UV light exposure to water within sealed HDPE containers compared to open polyurethane bags.'

On page 21, line 8: whether the light source contains enough UV radiation is critical here.

Added: 'Lighting conditions, especially the UV component of light which acts as a major source of photochemical H₂O₂ formation in natural waters'

Experiment design and bacterial abundance control extracellular H₂O₂ concentrations during 4 series of mesocosm experiments.

Mark J. Hopwood¹, Nicolas Sanchez², Despo Polyviou³, Øystein Leiknes², Julian Gallego-Urrea⁴, Eric P. Achterberg¹, Murat V. Ardelan², Javier Aristegui⁵, Lennart Bach¹, Sengul Besiktepe⁶, Yohann Heriot¹, Ioanna Kalantzi⁷, Tuba Terbiyik Kurt⁸, Ioulia Santi⁷, Tatiana M. Tsagaraki⁹, David Turner⁴

Correspondence to: Mark J. Hopwood (mhopwood@geomar.de)

1 GEOMAR Helmholtz Centre for Ocean Research Kiel, Germany

2 Norwegian University of Science and Technology, Trondheim, Norway

3 Ocean and Earth Science, National Oceanography Centre Southampton, United Kingdom

10 4 Marine Sciences, University of Gothenburg, Sweden

5 Instituto de Oceanografía y Cambio Global, IOCAG, Universidad de Las Palmas de Gran Canaria, ULPGC, Las Palmas, Spain

6 The Institute of Marine Sciences and Technology, Dokuz Eylul University, Turkey

7 Institute of Oceanography, Hellenic Centre for Marine Research, Heraklion, Greece

15 8 Marine Biology, Çukurova University, Turkey

9 Department of Biological Sciences, University of Bergen, Norway

Abstract

The extracellular concentration of H₂O₂ in surface aquatic environments is controlled by a balance between photochemical production and the microbial synthesis of catalase and peroxidase enzymes to remove H₂O₂ from solution. In any kind of incubation experiment, the formation rates and equilibrium concentrations of ROS may be sensitive to both the experiment design, particularly to the regulation of incident light, and the abundance of different microbial groups, as both cellular H₂O₂ production and catalase/peroxidase enzyme production rates differ between species. Whilst there are extensive measurements of photochemical H₂O₂ formation rates and the distribution of H₂O₂ in the marine environment, it is poorly constrained how different microbial groups affect extracellular H₂O₂ concentrations, how comparable extracellular H₂O₂ concentrations within large scale incubation experiments are to those observed in the surface-mixed layer, and to what extent a miss-match with environmentally relevant concentrations of ROS in incubations could influence biological processes differently to what would be observed in nature. Here we show that both experiment design and bacterial abundance consistently exert control on extracellular H₂O₂ concentrations across a range of incubation experiments in diverse marine environments.

30

During 4 large scale (>1000 L) mesocosm experiments (in Gran Canaria, the Mediterranean, Patagonia and Svalbard) most experimental factors appeared to exert only minor, or no, direct effect on H₂O₂ concentrations. For example, in 3 of 4 experiments where pH was manipulated to 0.4-0.5 below ambient pH no significant change was evident in extracellular

H₂O₂ concentrations relative to controls. An influence was sometimes inferred from zooplankton density, but not consistently between different incubation experiments and no change in H₂O₂ was evident in controlled experiments using different densities of the copepod *Calanus finmarchicus* grazing on the diatom *Skeletonema costatum* (<1% change in [H₂O₂] comparing copepod densities from 1-10 L⁻¹). Instead, the changes in H₂O₂ concentration contrasting high/low zooplankton incubations appeared to arise from the resulting changes in bacterial activity. The correlation between bacterial abundance and extracellular H₂O₂ was stronger in some incubations than others (R² range 0.09 to 0.55), yet high bacterial densities were consistently associated with low H₂O₂. Nonetheless, the main control on H₂O₂ concentrations during incubation experiments relative to those in ambient, unenclosed waters was the regulation of incident light. In an open (lidless) mesocosm experiment in Gran Canaria, H₂O₂ was persistently elevated (2-6 fold) above ambient concentrations; whereas using closed high density polyethylene mesocosms in Crete, Svalbard and Patagonia H₂O₂ within incubations was always reduced (median 10-90%) relative to ambient waters.

45 1.0 Introduction

Reactive oxygen species (ROS), such as H₂O₂, are ubiquitous in surface aquatic environments due to photochemical formation (Van Baalen and Marler, 1966; Moore et al., 1993; Miller and Kester, 1994). Quantum yields for H₂O₂ formation increase with declining wavelength and so the ultraviolet (UV) portion of natural sunlight is a major source of H₂O₂ in surface aquatic environments (Cooper et al., 1988, 1994). Sunlight normalized H₂O₂ production rates therefore peak between wavelengths of 310-340 nm (Kieber et al., 2014). H₂O₂ is present at concentrations on the order of 10-100 nM in the ocean's surface mixed layer with its concentration generally declining sharply with depth (Price et al., 1998; Yuan and Shiller, 2001; Gerringa et al., 2004). Because its decay rate is slow (observed half-lives in seawater range from 10 to 120 h, Petasne and Zika 1997) compared to less stable ROS such as superoxide (O₂⁻) and the hydroxyl radical (OH[•]), extracellular H₂O₂ concentrations in surface waters show a pseudo-sinuuous diurnal cycle, with elevated H₂O₂ concentrations occurring during daylight hours (Price et al., 1998).

H₂O₂ features as a reactive intermediate in the natural biogeochemical cycling of many compound groups including halocarbons (Hughes and Sun, 2016), trace metals (Moffett and Zika, 1987; Voelker and Sulzberger, 1996; Hansel et al., 2015) and dissolved organic matter (DOM) (Cooper et al., 1988; Scully et al., 2003). Previous work has highlighted the susceptibility of a broad range of marine biota to elevated extracellular H₂O₂ concentrations (Bogosian et al., 2000; Morris et al., 2011) and argued that measurable negative effects on metabolism occur in some marine species at H₂O₂ concentrations within the range of ambient surface-mixed layer concentrations (Morris et al., 2011; Baltar et al., 2013). Peroxidase and catalase enzymes are widely produced by marine microbes to lower extracellular H₂O₂ concentrations and these enzymes are the dominant sink for H₂O₂ in the surface marine environment (Moffett and Zafiriou, 1990; Angel et al., 1999). The reliance of some species including strains of *Prochlorococcus*, which do not produce such enzymes, on other 'helper' organisms to

remove extracellular H₂O₂ underpins a theory of reductive evolution, ‘the Black Queen Hypothesis’ (BQH) (Morris et al., 2012). BQH infers that because the removal of extracellular H₂O₂ by any species is a communal benefit, there is an energetic benefit to be gained to an individual species by losing genes associated with extracellular H₂O₂ detoxification. Loss of these genes continues to be favourable to individual species until only a minority of community members poses the ability to
70 remove H₂O₂, and the benefit of further loss would be offset by the negative effects of increasing extracellular H₂O₂ concentrations (Morris et al., 2012).

It is already acknowledged that laboratory incubation studies using buffered growth media are often conducted at H₂O₂ concentrations 2-10× higher than those found in the surface ocean (Morris and Zinser, 2013). We have previously
75 hypothesized that the same may be generally true for meso-scale experiments (Hopwood et al., 2018b) because the relative stability of H₂O₂ means that the enclosure of water at the ocean’s surface within mesocosms can lead to elevated H₂O₂ concentrations. Yet there are presently few examples in the literature of incubation experiments where ROS concentrations are measured and therefore it is unknown how changes to other stressors, or changes to experimental design, affect extracellular ROS concentrations. In order to assess whether ROS could be a significant artefact in incubation experiments;
80 and to investigate how extracellular H₂O₂ concentrations respond to changes in DOC, pH and grazing pressure; here we collate data on H₂O₂ from a series of small to large scale (20-8000 L) incubation experiments with varying geographical location (Table 1).

2.0 Methods

Our rationale for the investigation of H₂O₂ trends during these 20-8000 L scale mesocosm and microcosm experiments is
85 that the experiment matrixes for each experiment permitted the changing of 1, 2 or 3 key variables (DOC, zooplankton, pH) whilst maintaining others (e.g. salinity, temperature, light) in a constant state across the mesocosm/microcosm experiment. The relationships between H₂O₂ and other chemical/biological parameters are therefore potentially easier to investigate than in the ambient water column where mixing and the vertical/lateral trends in H₂O₂ concentrations must also be considered. Additionally, two of the experiment designs described herein (see Table 1) were repeated in 3 geographic locations
90 facilitating direct comparisons between the experiment results with only limited mitigating factors concerning method changes.

2.1 Mesocosm set up and sampling

Eight incubation experiments (Table 1) were constructed using coastal seawater which was either collected through pumping from small boats deployed offshore, or from the end of a floating jetty. Three of these incubations were outdoor mesocosm
95 experiments (MesoPat, MesoArc and MesoMed) conducted using the same basic setup (based on that used in earlier experiments described by Larsen et al., 2015). For these three mesocosms, 10 identical cubic high density polyethylene

(HDPE) 1000-1500 L tanks were filled ~95% with seawater which was passed through nylon mesh (size as per Table 1) to remove mesozooplankton. The 10 closed mesocosm tanks were then held in position with a randomized treatment configuration and incubated at ambient seawater temperature. For MesoPat and MesoArc the mesocosms were tethered to a jetty. For MesoMed the mesocosms were held in a pool facility at the Hellenic Centre for Marine Research which was continuously flushed with seawater to maintain a constant temperature. An extra HDPE container (to which no additions were made) was also filled to provide an additional supply of un-manipulated seawater (without zooplankton, DOC, or nutrient additions) for calibration purposes and baseline measurements on day 0. During MesoMed, this surplus container was incubated alongside the mesocosms for the duration of the experiment without any further additions/manipulation. In all cases, these HDPE containers likely strongly attenuated natural UV radiation, compared to ambient waters, which is expected to negatively affect photochemical formation of H₂O₂ (Cooper et al., 1988, 1994).

Label (Project)	Location	Month / year	Duration / days	Manipulated drivers	Scale / L	Site	Design Fig. S1	H ₂ O ₂ data available
MesoPat (Ocean Certain)	Comau fjord, Patagonia	Nov 2014	11	DOC, grazing	1000	In-situ	I	Diurnal cycle. Limited time series
MultiPat (Ocean Certain)	Comau fjord, Patagonia	Nov 2014	8	DOC, grazing, pH	20	Temperature controlled room	II	Final [H ₂ O ₂]
MicroPat (Ocean Certain)	Comau fjord, Patagonia	Nov 2014	11	DOC, grazing	20	Temperature controlled room	III	Final [H ₂ O ₂]
MesoArc (Ocean Certain)	Kongsfjorden, Svalbard	July 2015	12	DOC, grazing	1250	In-situ	I	Diurnal cycle
MultiArc (Ocean Certain)	Kongsfjorden, Svalbard	July 2015	8	DOC, grazing, pH	20	Temperature controlled room	II	Limited time series
MesoMed (Ocean Certain)	Hellenic Centre for Marine Research, Crete	May 2016	12	DOC, grazing	1500	Outdoor temperature controlled pool	I	Diurnal cycle, H ₂ O ₂ time series, decay rates, H ₂ O ₂ spiked incubation

MultiMed (Ocean Certain)	Hellenic Centre for Marine Research, Crete	May 2016	9	DOC, grazing, pH	20	Temperature controlled room	II	Final [H ₂ O ₂]
Gran Canaria (The Future Ocean)	Taliarte Harbour, Gran Canaria	Mar 2016	28	pCO ₂	8000	In-situ	IV	Diurnal cycle, H ₂ O ₂ time series, H ₂ O ₂ spiked incubation

110 **Table 1** Details of experiments where H₂O₂ data were collected. Data from 8 separate experiments are presented, including 4 outdoor mesocosm experiments and 4 indoor microcosm/multistressor experiments. 'DOC' dissolved organic carbon.

Experiment	PAT (Patagonia)	ARC (Svalbard, Arctic)	MED (Crete, Mediteranean)	Gran Canaria
Mesocosm	MesoPat	MesoArc	MesoMed	Gran Canaria
Containers	HDPE 1000 L	HDPE 1250 L	HDPE 1500 L	Polyurethane 8000 L
Lighting	Ambient	Ambient	Ambient reduced ~50% with net	Ambient
Zooplankton treatment	+30 copepods L ⁻¹	+5 copepods L ⁻¹	+4 copepods L ⁻¹	NA
Macronutrient addition	N added as NO ₃	N added as NH ₄	N added as 50/50 NH ₄ /NO ₃	N added as NO ₃
Macronutrient addition timing	Daily	Daily	Daily	Day 18 only
Macronutrients added (per addition)	1.0 μM NO ₃ , 1.0 μM Si, 0.07 μM PO ₄	1.12 μM NH ₄ , 1.2 μM Si, 0.07 μM PO ₄ (11.4 μM Si added on day 1)	48 nM NO ₃ , 48 nM NH ₄ , 6 nM PO ₄	3.1 μM NO ₃ , 1.5 μM Si, 0.2 μM PO ₄
Screening of initial seawater	NA	200 μm	140 μm	3 mm
Multistressor	MultiPat	MultiArc	MultiMed	
Containers	HDPE collapsible 20 L	HDPE collapsible 20 L	HDPE collapsible 20 L	
Lighting	36 W lamps	36 W lamps	36 W lamps	
Light regime	15 h light / 9 h dark	24 h light	15 h light / 9 h dark	
Zooplankton treatment	+30 copepods L ⁻¹	+5 copepods L ⁻¹	+4 copepods L ⁻¹	
Macronutrient addition	Same as MesoPat	Same as MesoArc	Same as MesoMed	
Macronutrient addition timing	Daily	Daily	Daily	
Macronutrients added (per addition)	1.0 μM NO ₃ , 1.0 μM Si, 0.07 μM PO ₄	1.12 μM NH ₄ , 1.2 μM Si, 0.07 μM PO ₄	48 nM NO ₃ , 48 nM NH ₄ , 6 nM PO ₄	
pH post adjustment	7.54±0.09	7.76±0.03	7.64±0.02	
pH pre-adjustment	7.91±0.01	8.27±0.18	8.08±0.02	
Screening of initial seawater	200 μm	200 μm	140 μm	
Temperature / °C	13-18	4.0-7.0	19.9-21.5	
Microcosm	MicroPat			
Containers	HDPE collapsible 20 L			
Lighting	36 W lamps			
Light regime	15 h light / 9 h dark			
Containers	HDPE collapsible 20 L			
Grazing treatment	+30 copepods L ⁻¹			
Macronutrient addition timing	Daily			

Macronutrient addition	N was added as NO ₃
Macronutrients added (per addition)	1.0 μM NO ₃ , 1.0 μM Si, 0.07 μM PO ₄
Screening of initial seawater	200 μm
Temperature / °C	14-17

Table 2 Experiment details for each experiment. For a visual representation of experiment designs, the reader is referred to Supplementary Material. ‘HDPE’ high density polyethylene. ‘NA’ not applicable.

115 The 10-mesocosm experiment design matrix was the same for MesoPat, MesoArc and MesoMed (Fig. S1, design I). For these 3 mesocosm experiments, zooplankton were collected one day in advance of requirement using horizontal tows at ~30 m depth with a mesh net equipped with a non-filtering cod end. Collected zooplankton were then stored overnight in 100 L containers and non-viable individuals removed by siphoning prior to making zooplankton additions to the mesocosm containers. After filling the mesocosms, zooplankton (quantities as per Table 2) were then added to 5 of the containers to
 120 create contrasting high/low grazing conditions. Macronutrients (NO₃/NH₄, PO₄ and Si) were added to mesocosms daily (Table 2). Across both the 5-high and 5-low grazing tank treatments, a dissolved organic carbon (DOC) gradient was created by addition of glucose to provide carbon at 0, 0.5, 1, 2 and 3 times the Redfield Ratio (Redfield, 1934) with respect to added PO₄. Mesocosm water was sampled through silicon tubing (permanently fixed into each mesocosm lid) immediately after mixing of the containers using plastic paddles (also mounted within the mesocosms through the lids) with the first 2 L
 125 discarded in order to flush the sample tubing.

A 4th outdoor mesocosm experiment (Gran Canaria) used 8 cylindrical polyurethane bags with a depth of approximately 3 m, a starting volume of ~8000 L and no lid or screen on top (Hopwood et al., 2018b). After filling with coastal seawater the bags were allowed to stand for 4 days. A pH gradient across the 8 tanks was then induced (on day 0) by the addition of
 130 varying volumes of filtered, pCO₂ saturated seawater (resulting in pCO₂ concentrations from 400-1450 μatm, treatments outlined Fig. S1 IV) using a custom-made distribution device (Riebesell et al., 2013). A single macronutrient addition (3.1 μM nitrate, 1.5 μM silicic acid and 0.2 μM phosphate) was made on day 18 (Table 2).

2.2 Microcosm and multistressor set up and sampling

A 10-treatment microcosm (MicroPat) incubation mirroring the MesoPat 10 tank mesocosm (treatment design as per Fig. S1
 135 I, but with 6 × 20 L containers per treatment -one for each time point- rather than a single HDPE tank) and three 16-treatment multistressor experiments (MultiPat, MultiArc and MultiMed Fig. S1 II) were conducted using artificial lighting in temperature controlled rooms (Table 1, Fig. S1). For all 3 multistressor incubations (MultiPat, MultiArc and MultiMed) and the single microcosm incubation (MicroPat), coastal seawater (filtered through nylon mesh) was used to fill 20 L HDPE collapsible containers. The 20 L containers were arranged on custom made racks with light provided by a network of 36 W

140 lamps (Phillips, MASTER TL-D 90 De Luxe 36W/965 tubes). The number and orientation of lamps was adjusted to produce a light intensity of $80 \mu\text{mol quanta m}^{-2} \text{s}^{-1}$. A diurnal light regime representing spring/summer light conditions at each fieldsite was used and the tanks were agitated daily and after any additions (e.g. glucose, acid or macronutrient solutions) in order to ensure a homogeneous distribution of dissolved components. Whilst the light condition for these experiments was selected to approximate the light intensity of ambient near-surface waters, the synthetic lighting is deficient in UV relative to ambient sunlight-especially at low latitudes. In all 20 L scale experiments, macronutrients were added daily (as per Table 2). One 20 L container from each treatment set was 'harvested' for sample water each sampling day.

The experiment matrix used for the MicroPat incubation duplicated the MesoPat experiment design (Table 2) and thereby consisted of 10 treatments. The experiment matrix for the 3 multistressor experiments (MultiPat, MultiArc and MultiMed 150 outlined in Fig. S1 II) duplicated the corresponding mesocosm experiments at the same fieldsites (MesoPat, MesoArc and MesoMed), with one less C/glucose treatment and an additional pH manipulation (Table 2). The multistressor experiments thereby consisted of 16 treatments. pH manipulation was induced by adding a spike of HCl (trace metal grade) on day 0 only. For trace metal and H_2O_2 analysis, sample water from 20 L collapsible containers was extracted using a plastic syringe and silicon tubing which was mounted through the lid of each collapsible container.

155 Throughout, where changes in any incubation experiment are plotted against time, 'day 0' is defined as the day the experimental gradient (zooplankton, DOC, pCO_2) was imposed. Time prior to day 0 was intentionally introduced during some experiments to allow water to equilibrate with ambient physical conditions after container filling. H_2O_2 concentration varies on diurnal timescales and thus during each experiment where a time series of H_2O_2 concentration was measured, 160 sample collection and analysis occurred at the same time daily (± 0.5 h) and the order of sample collection was random.

2.3 Ancillary experiments

Four side experiments (1-4 below) were conducted to investigate potential links between bacterial/zooplankton abundance and extracellular H_2O_2 concentrations. Where specified, H_2O_2 concentrations were manipulated to form high, medium and low H_2O_2 conditions by adding aliquots of either a 1 mM H_2O_2 solution (prepared weekly from H_2O_2 stock) to increase H_2O_2 165 concentration, or bovine catalase (prepared immediately before use) to decrease H_2O_2 concentration. All treatments were triplicated. Catalase is photo-deactivated and biological activity to remove extracellular H_2O_2 follows the diurnal cycle (Angel et al., 1999; Morris et al., 2016), so catalase/ H_2O_2 additions were conducted at sunset in order to minimize the additions required. Bovine catalase was used as received (Sigma Aldrich) with stock solutions prepared from frozen enzyme (stored at -20°C). De-natured catalase was prepared by heating enzyme solution to $>90^\circ\text{C}$ for 10 min. Plasticware used to 170 handle catalase solution was discarded. No adverse effects of measuring H_2O_2 in catalase-manipulated solutions were found. As H_2O_2 measurements were made >12 h after catalase addition, this may reflect catalase de-activation under the incubated conditions.

(1) In Gran Canaria a 5 day experiment was conducted, using 5 L polypropylene bottles. After filling with offshore seawater, and the addition of macronutrients which matched the concentrations added to the Gran Canaria mesocosm (3.1 μM nitrate, 1.5 μM silicic acid and 0.2 μM phosphate), bottles were incubated under ambient light and temperature conditions within Taliarte Harbor. (2) In Crete, a similar 7 day incubation was conducted in the HCMR pool facility using 20 L HDPE containers. Seawater was extracted from the baseline MesoMed mesocosm (no DOC or zooplankton addition) on day 11 and then incubated without further additions except for H_2O_2 manipulation. After day 5 no further H_2O_2 manipulations were made. (3) As per (2), seawater was withdrawn from the baseline MesoMed mesocosm on day 11 and then incubated without further addition except for H_2O_2 manipulation in 500 mL trace metal clean LDPE bottles under the artificial lighting conditions used for the MultiMed incubation. (4) A short term (20 h) experiment was conducted in trace metal clean 4 L HDPE collapsible containers to investigate the immediate effect of grazing on H_2O_2 concentrations. Filtered (0.2 μm , Satorius) coastal seawater (S 32.8, pH 7.9) water was stored in the dark for 3 days before use. The diatom *Skeletonema costatum* (NIVA-BAC 36 strain culture (CAA) from the Norsk Institutt for vannforskning (NIVA)) was used as a model phytoplankton grown in standard f/2 medium (Guillard and Ryther, 1962). Each treatment consisted of a total volume of 2 L seawater and contained macronutrients, 7.5 ml of the original medium (resulting in an initial chlorophyll a concentration of 3 $\mu\text{g L}^{-1}$ in the incubations) and treated seawater containing the copepod *Calanus finmarchicus* corresponding to each desired density. The light regime was produced with fluorescent lighting with a mean luminous intensity of 80-90 $\mu\text{mol m}^{-2} \text{s}^{-1}$ and the temperature maintained at 10.5-10.9°C.

Light levels during all experiments (Table 1) were quantified using a planar Li-cor Q29891 sensor connected to a Li-cor Li-1400 data logger. Diurnal experiments measuring H_2O_2 concentrations in mesocosms or ambient surface (10 cm depth) seawater were conducted using flow injection apparatus with a continuous flow of seawater into the instrument through a PTFE line as described previously (Hopwood et al., 2018b). For extensive datasets, the diurnal range of H_2O_2 concentrations was determined as the difference between the means of the highest and lowest 10% of datapoints. Fe(II) and photochemically generated radical species can interfere with the luminol based chemiluminescence used to determine H_2O_2 . In batch measurements, the >15 minute time delay between sample collection analysis (during which time the same is not exposed to light) is likely sufficient to minimize interference from these species which have much shorter half-lives and lower concentrations than H_2O_2 (Yuan and Shiller, 1999). For in situ continuous measurements a sample loop was intentionally introduced such that seawater was displaced from ambient light for 2 minutes prior to analysis. Some residual Fe(II) may have therefore remained leading to over-estimation of H_2O_2 concentrations. As Fe(II) concentrations were quantified in all of the experiments and ambient waters described herein (Hopwood et al., 2018a), we can determine that the maximum possible over-estimate of H_2O_2 concentrations for ambient waters during continuous analysis (diurnal timeseries) was <10%.

2.4 Chemical analysis

H₂O₂

H₂O₂ samples were collected in opaque HDPE 125 mL bottles (Nalgene) which were pre-cleaned (1 day soak in detergent, 1 week soak in 1 M HCl, 3 rinses with de-ionized water) and dried under a laminar flow hood prior to use. Bottles were rinsed once with sample water, filled with no headspace and always analysed within 2 h of collection via flow injection analysis (FIA) using the Co(II) catalysed oxidation of luminol (Yuan and Shiller, 1999). FIA systems were assembled and operated exactly as per Hopwood et al., (2017) producing a detection limit of < 1 nM. Calibrations were run daily and with every new reagent batch using 6 standard additions of H₂O₂ (TraceSelect, Fluka) within the range 10-300 nM to aged (stored at room temperature in the dark for >48 h) seawater (unfiltered).

215 **Macronutrients**

Dissolved macronutrient concentrations (nitrate+nitrite, phosphate, silicic acid; filtered at 0.45 µm upon collection) were measured spectrophotometrically the same day as sample collection (Hansen and Koroleff, 2007). For experiments in Crete (MesoMed, MultiMed), phosphate concentrations were determined using the 'MAGIC' method (Rimmelin and Moutin, 2005). The detection limits for macronutrients thereby inevitably varied slightly between the different mesocosm/microcosm/multistressor experiments (Table 1), however this does not adversely affect the discussion of results herein.

Carbonate chemistry

pH_T (except where stated otherwise, 'pH' refers to the total pH scale reported at 25°C) was measured during the Gran Canaria mesocosm using the spectrophotometric technique of Clayton and Byrne (1993) with m-cresol purple in an automated Sensorlab SP101-SM system using a 25°C-thermostatted 1 cm flow-cell exactly as per González-Dávila et al., (2016). pH during the MesoPat/MicroPat/MultiPat experiments was measured similarly as per Gran Canaria using m-cresol. During MesoArc/MultiArc/MesoMed/MultiMed experiments pH was measured spectrophotometrically as per Reggiani et al., (2016).

Biological parameters

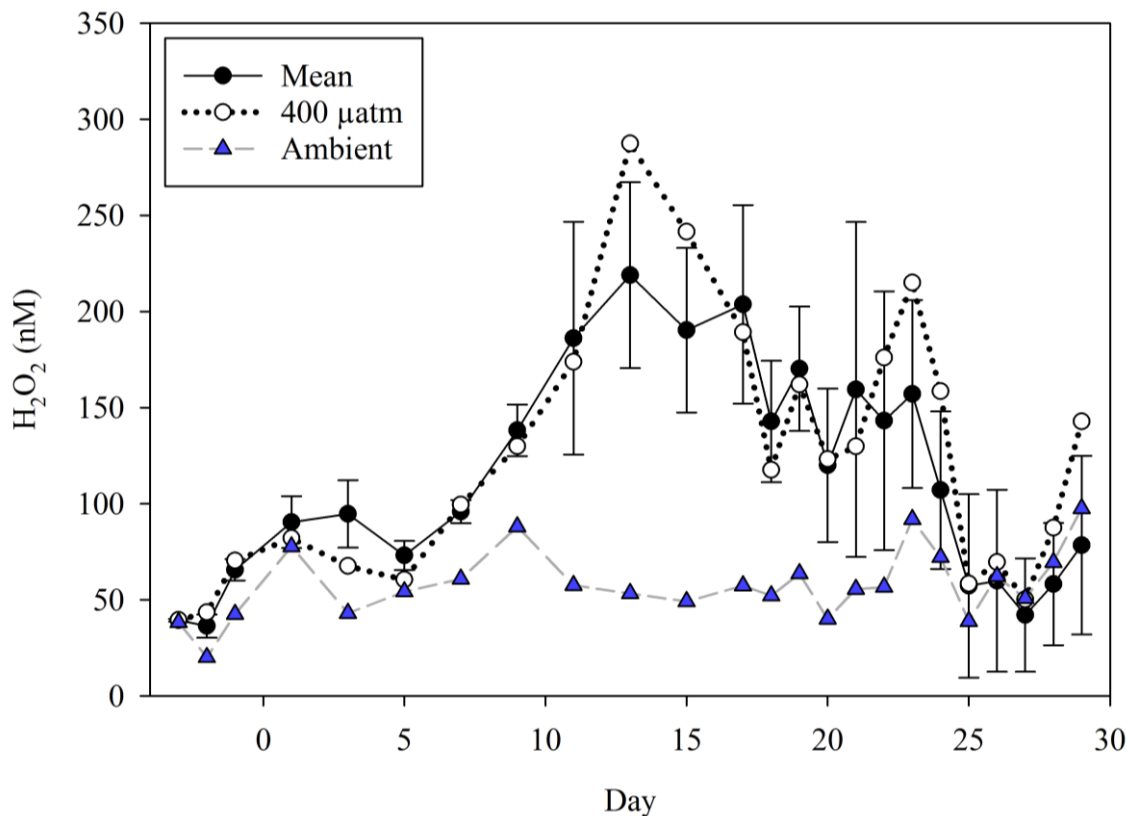
230 Chlorophyll a was measured by fluorometry as per Welschmeyer (1994). Bacterial production was determined by incorporation of tritium-labelled leucine (³H-Leu) using the centrifugation procedure of Smith and Azam (1992). Conversion of leucine to carbon (C) was done with the theoretical factor 3.1 kg C mol⁻¹ leucine. In Gran Canaria, flow cytometry was conducted on 2 mL water samples which were fixed with 1% paraformaldehyde (final concentration), flash frozen in liquid N₂ and stored at -80°C until analysis. Samples were analysed (FACSCalibur, Becton Dickinson) with a 15 mW laser set to excite at 488 nm (Gasol and del Giorgio, 2000). Subsamples (400 µL) for the determination of heterotrophic bacteria were stained with the fluorochrome SybrGreen-I (4 µL) at room temperature for 20 min and run at a flow rate of 16 µL min⁻¹. Cells were enumerated in a bivariate plot of 90° light scatter and green fluorescence. Molecular Probes latex beads (1 µm) were used as internal standards. In Crete (MesoMed/MultiMed), the flow cytometry was conducted similarly except for the

following minor changes: samples were fixed with 0.5% glutaraldehyde (final concentration), yellow-green microspheres (1 and 10 μm diameter, respectively) were used as internal references during the analysis of bacterial and nanoflagellate populations, and the flow rate was 79-82 $\mu\text{L min}^{-1}$. Subsamples (7-50 L) for zooplankton composition and abundance were preserved in 4% borax buffered formaldehyde solution and analysed microscopically.

3.0 Results

3.1 H_2O_2 time series during outdoor mesocosm incubations; MesoMed and Gran Canaria

In order to understand the controls on H_2O_2 concentrations in incubations, time series of H_2O_2 are first presented for those experiments with the highest resolution data. Also of interest are trends in bacterial productivity following the observation that H_2O_2 decay constants appear to correlate with bacterial abundance in a range of natural waters (Cooper et al., 1994). The concentration of H_2O_2 was followed in all treatments on all sampling days during the Gran Canaria and MesoMed mesocosms. In Gran Canaria, comparing mean ($\pm\text{SD}$) H_2O_2 in all mesocosms across a pCO_2 gradient (400-1450 μatm) with H_2O_2 in ambient seawater outside the mesocosms, H_2O_2 was generally elevated within the mesocosms compared to ambient seawater (ANOVA $p < 0.05$ for all treatments compared to ambient conditions). The mean and median ambient H_2O_2 concentration throughout the experiment was at least 40% lower than that in any mesocosm treatment (Fig. 1). This included the 400 μatm mesocosm which received no additions of any kind until the nutrient spike on day 18. The only exception was a short time period under post-bloom conditions when bacterial abundance peaked and daily integrated light intensity was relatively low (compared to the mean over the duration of the experiment) for 3 consecutive days (experiment days 25-27, Hopwood et al., 2018). No clear trend was observed with respect to the temporal trend in H_2O_2 and the pCO_2 gradient. H_2O_2 concentration in the baseline pCO_2 treatment was close to the mean (400-1450 μatm) for the duration of the 28 day experiment.



260 **Figure 1: A summary of H_2O_2 over the duration of a pCO_2 gradient mesocosm in Gran Canaria. Data from Hopwood et al., (2018). The mean (\pm SD) H_2O_2 from all pCO_2 treatments is contrasted with the concentration in ambient surface seawater immediately outside the mesocosms. In addition to its inclusion in the mean, the baseline $400 \mu\text{atm}$ pCO_2 treatment is shown separately to allow comparison with ambient surface seawater.**

265 During MesoMed (Fig. 2) an additional mesocosm tank was filled (Tank 11) and maintained without any additions (no macronutrients, no DOC, no zooplankton) alongside the 10 mesocosm containers. As per the Gran Canaria mesocosm, H_2O_2 concentrations were also followed in ambient seawater throughout the duration of the MesoMed experiment. MesoMed was however conducted in an outdoor pool facility, so the ambient concentration of H_2O_2 in coastal seawater refers to a site approximately 500 m away from the incubation pool. Ambient H_2O_2 was generally higher than that observed within the mesocosm with a median concentration of 120 nM around midday (Fig. 2(a)).

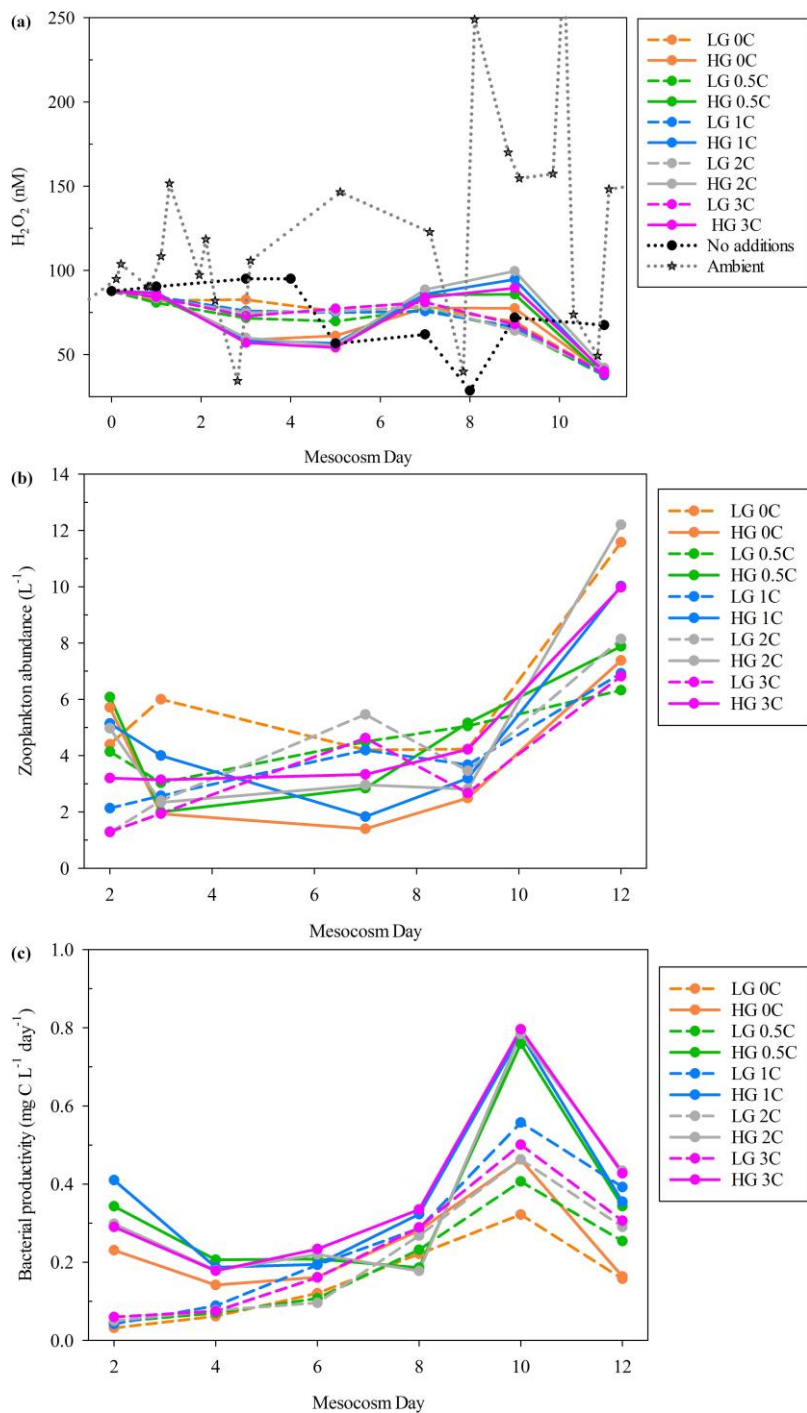


Figure 2: (a) H₂O₂ in all mesocosms during MesoMed in Gouves, Crete. A 10-treatment matrix (as per Fig. S1) was used (b) Zooplankton abundances showed a rapid convergence in the HG/LG (high grazing/low grazing) status of the mesocosms after day 2 (c) The trend in bacterial productivity showed broad similarity within the HG and LG treatment groups. 'C' refers to the ratio of added carbon (glucose) in each treatment.

275 H₂O₂ during the MesoMed experiment was relatively constant in terms of the range of concentrations measured over the 11
day duration of the experiment (Fig. 2), especially when compared to the Gran Canaria mesocosm (Fig. 1). A notable
clustering of the high ('HG') and low ('LG') zooplankton tanks was clearly observed between days 1 and 9 (Fig. 2) (addition
of zooplankton took place immediately after day 1 sampling). H₂O₂ concentration in the high zooplankton tanks initially
declined more strongly than the low zooplankton tanks, then re-bounded together after day 5 (Fig. 2). Dilution experiments
280 to estimate zooplankton grazing and zooplankton abundance (Fig. 2) both suggested that between days 3 and 7, the high/low
grazing status of the mesocosms converged i.e. grazing declined in the tanks to which zooplankton had initially been added
and increased in the tanks to which no zooplankton had been added such that initial 'high/low' grazing labels became
obsolete (Rundt, 2016). H₂O₂ concentration declined sharply in all treatments on day 11, except in the no-nutrient addition
mesocosm, coinciding with a pronounced increase in zooplankton abundance and occurring just after bacterial productivity
285 peaked in all treatments (Fig. 2).

H₂O₂ decay rate constants in the dark (measured using freshly collected seawater at the MesoMed fieldsite over 24 h and
assumed to be first order) were 0.049 h⁻¹ (unfiltered) and 0.036 h⁻¹ (filtered, Satorius 0.2 µm) corresponding to half-lives of
14 h and 19 h, respectively, which are within the range expected for coastal seawater (Petasne and Zika, 1997).

290 **3.2 H₂O₂ trends during 20 L scale indoor MultiPat, MultiMed and MicroPat incubations**

A sustained decline in H₂O₂ concentration was found whenever ambient seawater was moved into controlled temperature
rooms with artificial diel light cycles (e.g Fig. 3) which were used to incubate all 20 L scale multistressor and microcosm
experiments discussed herein (Table 1). Final H₂O₂ concentrations in these 20 L scale experiments were thereby generally
low compared to those measured in corresponding ambient surface waters and to the corresponding outdoor experiments in
295 the same locations with natural lighting (and higher UV light exposure).

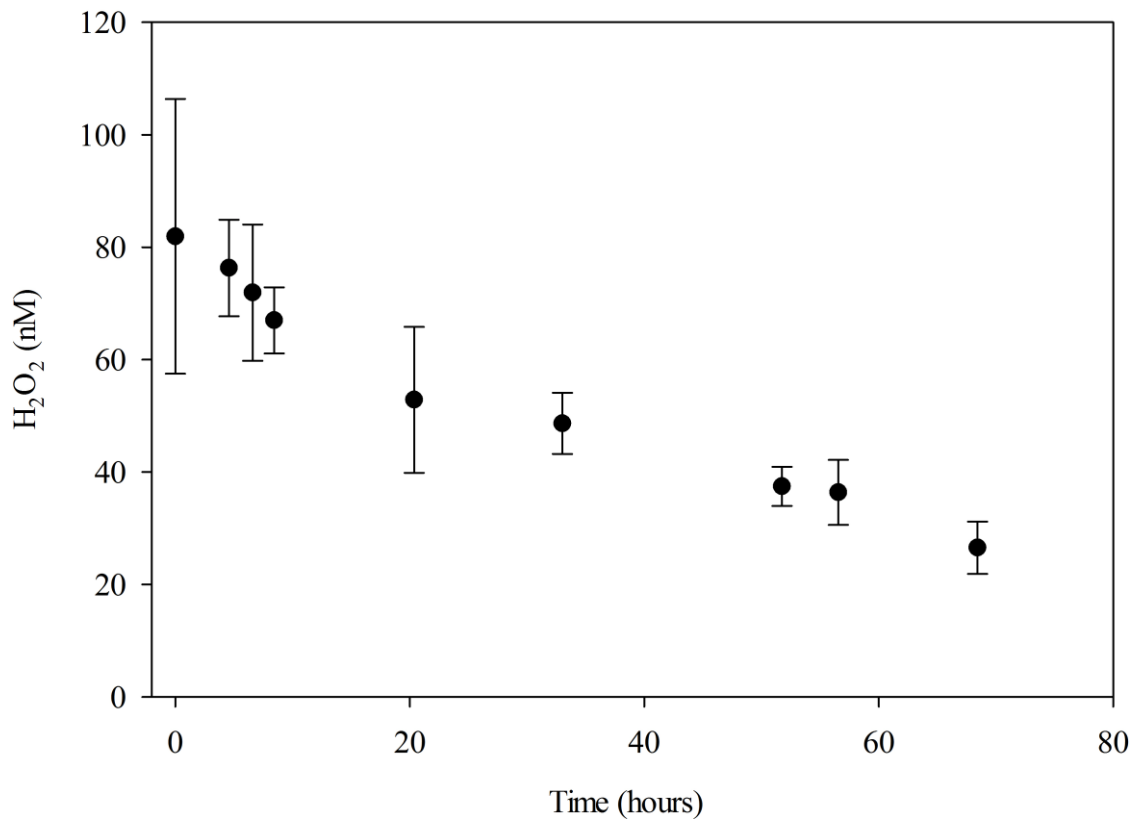
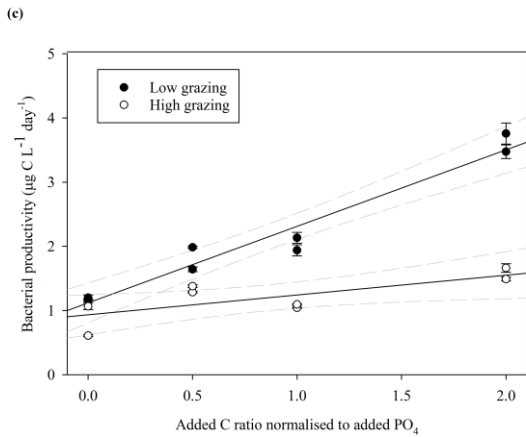
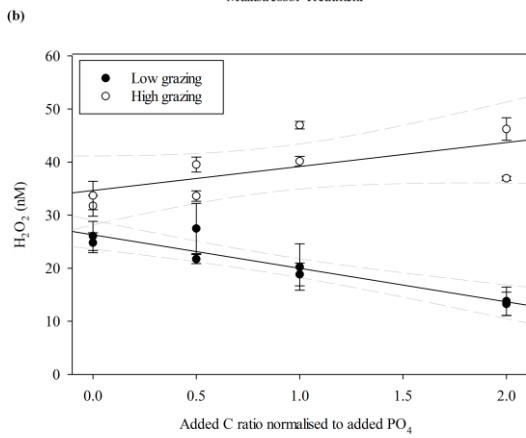
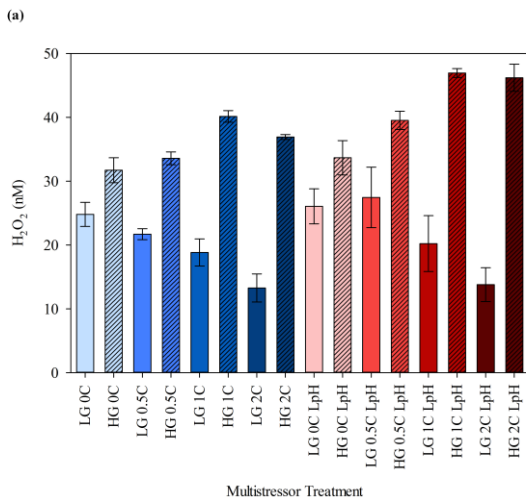


Figure 3: Seawater from MesoMed (without macronutrient, DOC or zooplankton amendment) was used to fill a 20 L HDPE container which was then incubated under the synthetic lighting used in the MultiMed experiment for 72 h with regular sub-sampling for analysis of H_2O_2 .



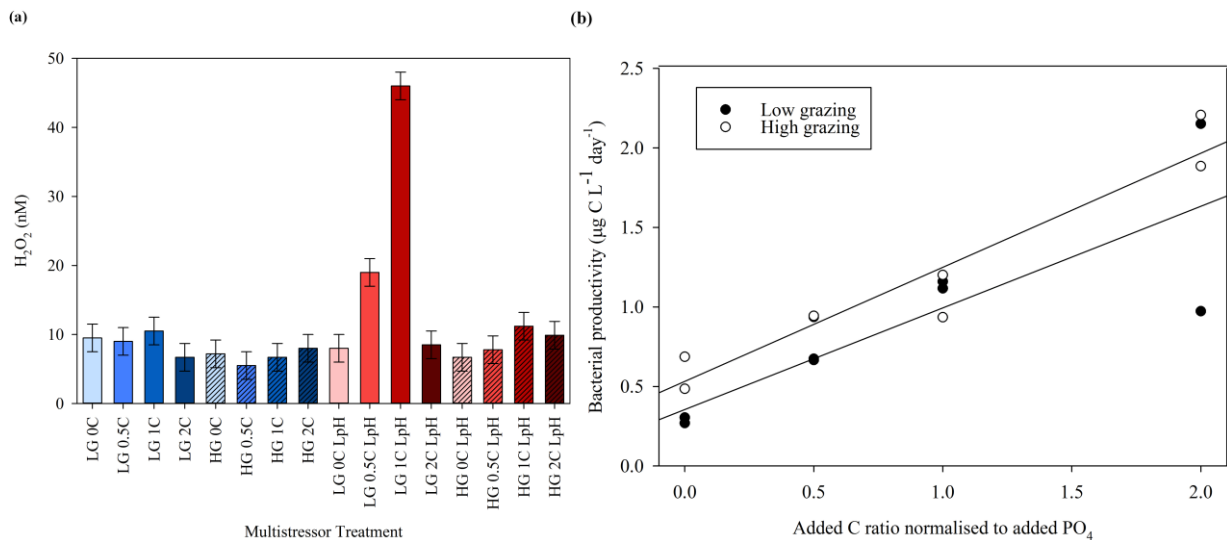
300

Figure 4: (a) H₂O₂ concentrations at the end of the MultiMed experiment (Day 9). Ambient pH (blue), low pH (red); high grazing (hatched); carbon (C, glucose) added at 0, 0.5, 1.0, 1.5 and 2.0 × Redfield carbon: phosphate ratio. (b) Plotting both ambient and low pH datapoints together, which exhibited no statistically significant difference in H₂O₂ concentrations, final H₂O₂ concentration showed contrasting trends between high and low grazing treatments over the added C gradient. 95% confidence intervals are shown. (c) Bacterial productivity, measured via leucine incorporation, during the same experiments.

305

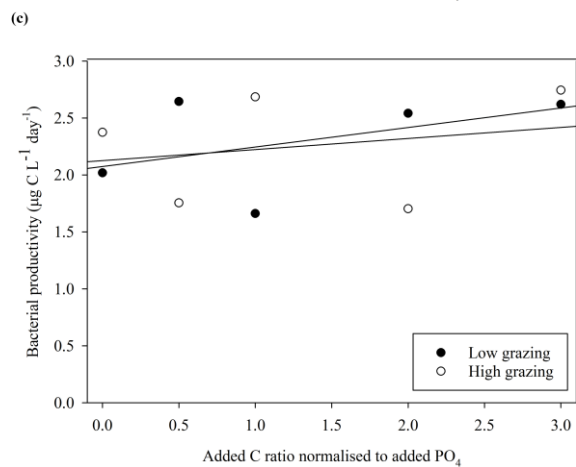
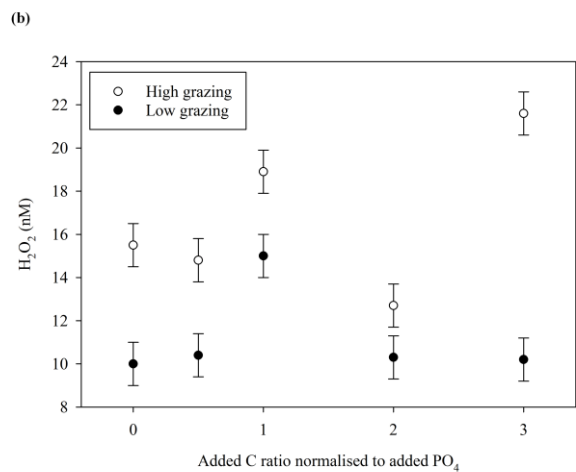
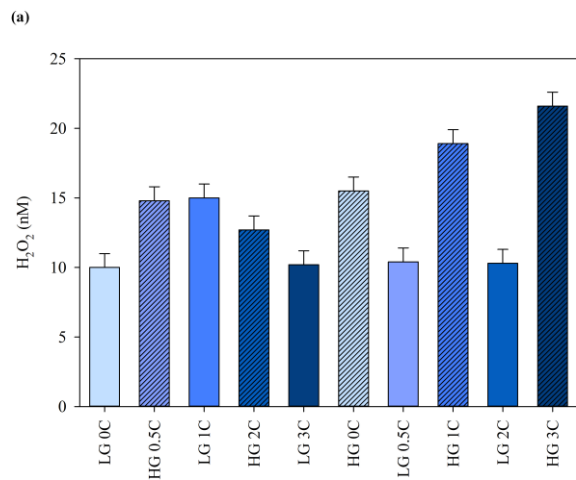
H₂O₂ concentrations by the end of the MultiMed experiments (day 9) were universally low compared to the range found in comparable ambient waters and the outdoor mesocosm incubation conducted at the same fieldsite (Fig. 2). As was the case in the MesoMed experiment, a clear difference was noted between H₂O₂ concentrations in the high and low zooplankton addition treatments (Fig. 4 (b)), with the high grazing always resulting in higher H₂O₂ concentrations (t test, p < 0.001). Any effect of pH was less obvious, with similar results obtained between ambient (initially 8.08 ± 0.02) and low (initially 7.64 ± 0.02) pH treatments (Fig. 4 (a)) and thus low and ambient pH treatments are not distinguished in Fig. 4 (b) and (c). An effect of the imposed C gradient on H₂O₂ concentrations was notable in both the high and low grazing treatments, yet the effect operated in the opposite direction (Fig. 4 (b)). In high grazing treatments, increasing C corresponded to increasing extracellular H₂O₂ concentrations (linear regression coefficient 4.5 ± 2.3); whereas in low grazing treatments, increasing C corresponded to decreasing extracellular H₂O₂ concentrations (linear regression coefficient -6.3 ± 0.97). Bacterial productivity increased with added C in both high (linear regression coefficient 0.31 ± 0.1) and low grazing treatments (linear regression coefficient 1.2 ± 0.1), but there was a more pronounced increase under low grazing conditions (Fig. 4 (c)).

At the end of the MultiPat experiment (day 8), H₂O₂ concentrations were similarly low compared to ambient surface waters at the Patagonia fieldsite (Fig. 5 (a)), although there was a greater range of results. In the low pH treatment (initially 7.54 ± 0.09), H₂O₂ concentrations were significantly higher (Mann-Whitney Rank Sum test p=0.02) compared to the unmodified pH treatment (initially 8.01 ± 0.02). However, two of the low pH treatments with particularly high H₂O₂ were outliers (defined as 1.5 IQR) when considering the data as consisting of two pH groups. Without these two datapoints, there would be no significant difference between H₂O₂ in high and low treatments (p=0.39). Contrary to the results from the MultiMed experiment (Fig. 4), there was no significant difference between high/low grazing treatments (Mann-Whitney Rank Sum test p=0.65). Bacterial productivity also showed similar results between the high and low grazing treatments (Fig. 5 (b)). Data from day 5 (the last day bacterial productivity was measured) showed a similar gradient in increased bacterial productivity with added C for both high/low grazing treatment groups (linear regressions HG 0.64, R² 0.70 and LG 0.72, R² 0.92).



330 **Figure 5: (a) H₂O₂ concentrations at the end of the MultiPat experiment. Normal pH (blue), low pH (red); high grazing (hashed); DOC added at 0, 0.5, 1.0, and 2.0 × Redfield carbon (C):phosphate ratio indicated by increasing colour density. (b) Plotting both high and low pH datapoints together (which exhibited no statistically significant difference in H₂O₂ concentrations), bacterial productivity showed similar trends between the HG and LG treatments.**

The MicroPat experiment, also conducted using 20 L HDPE containers and artificial lighting, yielded no clear trend with respect to H₂O₂ concentrations over the imposed C gradient (Fig. 6, day 11), but the high grazing treatments were associated with higher H₂O₂ concentrations (t-test, p=0.017). Bacterial productivity was not systematically different across the high/low grazing treatment groups, nor was there as clear a trend in bacterial productivity with respect to the added C gradient (Fig. 6 (c)) compared to the MultiPat (Fig. 5 (b)) or MultiMed (Fig. 4 (c)) experiments.



340 **Figure 6:** (a) H₂O₂ concentrations at the end of the MicroPat experiment. High grazing treatments are hashed; DOC added at 0, 0.5, 1.0, 2.0 and 3.0 × Redfield carbon (C):phosphate ratio indicated by increasing colour density. (b) No clear trend was evident across the DOC gradient, but high grazing was consistently associated with higher H₂O₂ concentration. (c) Bacterial productivity in the same experiment.

3.2 Diurnal cycling of H₂O₂; results from the Mediterranean

345 In addition to the trends observed over the duration of multi-day incubation experiments, a diurnal variability in H₂O₂ concentrations is expected. The diurnal cycle of H₂O₂ concentrations during MesoMed was followed in the no-addition tank (number 11) over 2 days with markedly different H₂O₂ concentrations (Fig. 4). An additional cycle was monitored at a nearby coastal pier (Gouves) for comparative purposes. The mean difference between mid-afternoon and early-morning H₂O₂ could also be deduced from discrete time points collected over the experimental duration in seawater close to the pool facility. All time series are plotted against local time (UTC+1). Sunrise/sunset was as follows: (May 15) 06:15, 20:17; (May 19) 06:12, 20:20. All three time series showed the expected peak in H₂O₂ concentrations during daylight hours, but the timing of peak H₂O₂ concentration and the range of concentrations observed differed between mesocosms and coastal seawater. The intraday range in H₂O₂ concentrations in Gouves, and the afternoon peak in H₂O₂, (Fig. 7) was similar to that observed previously in Gran Canaria (Hopwood et al., 2018b). Yet both the mesocosm diurnal time series exhibited notably limited diurnal ranges and peak H₂O₂ concentration occurred earlier, around midday (Fig. 7), than in coastal waters.

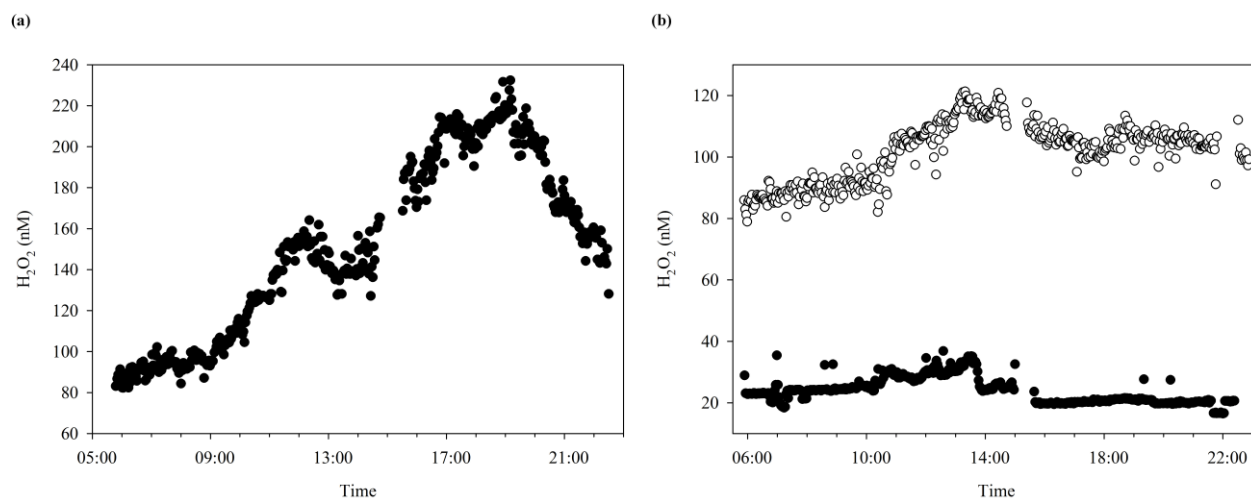


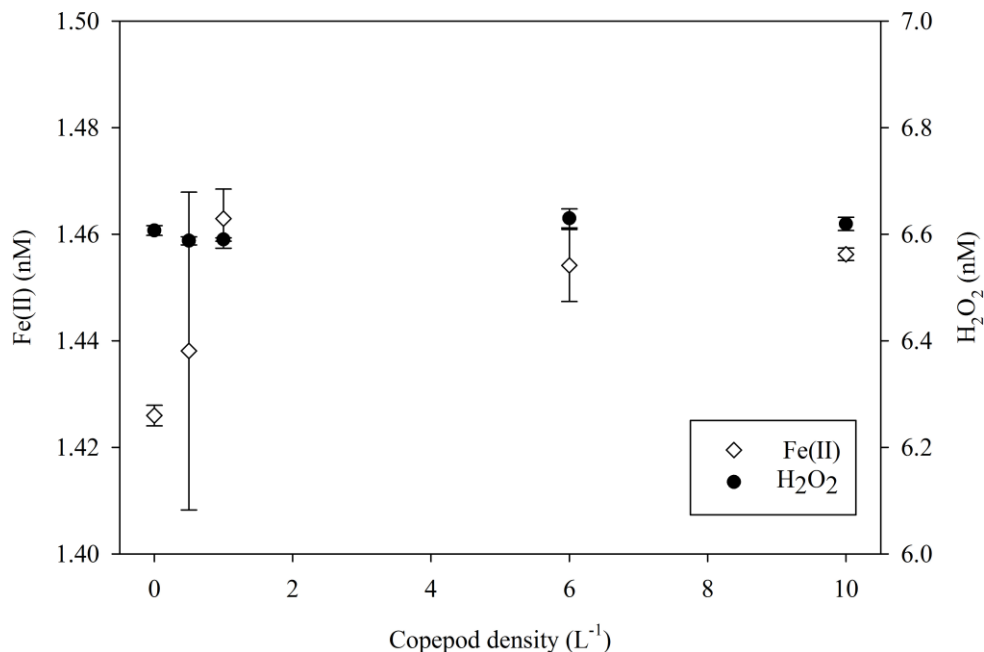
Figure 7: (a) Diurnal cycling of H₂O₂ in coastal seawater (Gouves, Crete 17 May) and (b) in the no addition tank (number 11) during the MesoMed mesocosm on May 15 (open circles) and May 19 (closed circles) 2016 (experiment days 4 and 8, respectively).

3.3 Ancillary experiments to investigate links between microbial groups (bacterial, zooplankton) and extracellular

H₂O₂

360 In addition to comparing H₂O₂ concentrations in different incubation experiments to assess the effect of experiment setup on extracellular H₂O₂ concentrations, potential links between microbial groups and H₂O₂ were explored. The MesoPat/Arc/Med, MicroPat and MultiPat/Arc/Med experiments all included a high/low zooplankton addition treatment (Table 2). Over a 20 h incubation (4 h darkness, 16 h light) in an experiment with varying concentrations of copepods (0-25 L⁻¹) grazing on an intermediate density of a diatom (initially 3 µg L⁻¹ chlorophyll a), H₂O₂ concentrations showed no inter-treatment differences (Fig. 8). A diatom was selected as phytoplankton stock because cell normalized H₂O₂ production rates for diatoms appear to be generally at the low end of the observed range for phytoplankton groups (Schneider et al., 2016). Fe(II) concentration

(measured at the same time as per Hopwood et al., 2018a) also appeared to be unaffected by the copepod density as the difference between treatments was almost negligible (<0.04 nM).



370

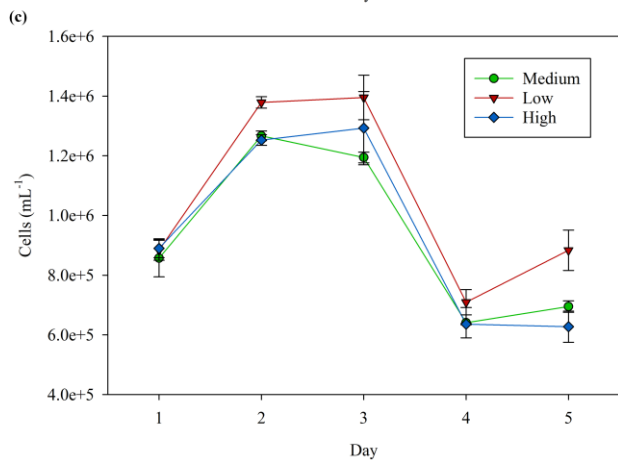
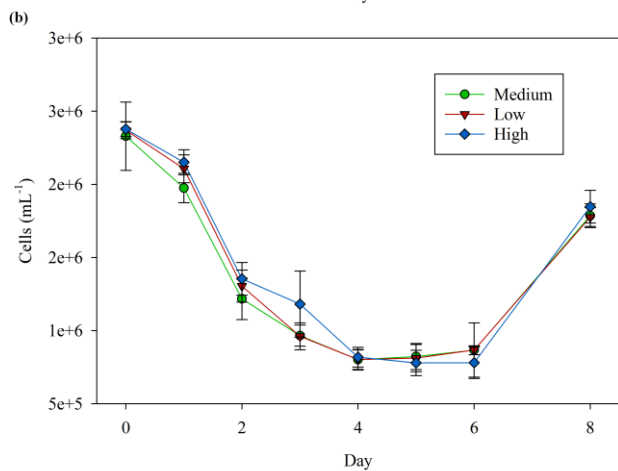
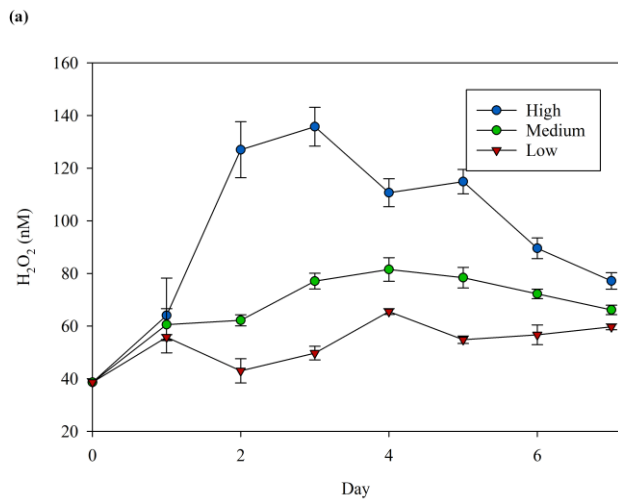
Figure 8: H₂O₂ and Fe(II) concentrations in a culture of diatoms growing in coastal seawater after 20 h of incubation with a zooplankton gradient imposed by addition of copepods.

At the end of the MesoMed experiment, seawater (extracted from the baseline treatment from the mesocosm on day 11) was used in two side experiments. During both the extracellular H₂O₂ concentration was manipulated, with each treatment
375 triplicated. In all cases the mean (\pm SD) of three replicate treatments is reported. The high-medium-low H₂O₂ concentration gradient used in each experiment was determined by considering the ambient concentration of H₂O₂ in the mesocosms (e.g. Fig. 2) and in ambient seawater close to the mesocosm facility. After the first daily H₂O₂ measurements were made, the required spikes to maintain the desired H₂O₂ gradient were calculated based on measured rates of H₂O₂ decay. H₂O₂ and catalase spikes were then added at sunset followed by gentle mixing.

380

A test specifically to investigate the effect of the multistressor/microcosm experimental set up on bacterial activity was conducted in 500 mL trace metal clean LDPE bottles under the artificial lighting conditions (~ 80 $\mu\text{mol quanta m}^{-2} \text{ s}^{-2}$) used for the MultiMed experiment. H₂O₂ concentrations again verified that manipulation with H₂O₂ spikes successfully created a low, medium and high H₂O₂ treatment (mean for triplicate low/medium/high treatments: 40 ± 2 , 120 ± 6 , 230 ± 7 nM H₂O₂).
385 Bacterial production showed no statistically significant (ANOVA, $p=0.562$) difference between triplicate low (1.69 ± 0.28 $\mu\text{g C L}^{-1} \text{ day}^{-1}$), medium (1.30 ± 0.60 $\mu\text{g C L}^{-1} \text{ day}^{-1}$) and high (1.29 ± 0.56 $\mu\text{g C L}^{-1} \text{ day}^{-1}$) H₂O₂ treatments.

For a concurrent manipulation in the Mediterranean using 20 L HDPE containers incubated outdoors, a gradient in H₂O₂ concentrations was similarly imposed. These manipulations successfully produced a clear gradient of H₂O₂ conditions with relatively consistent H₂O₂ concentrations within each triplicated set (Fig. 9 (a)). After day 5 no further manipulations were conducted and H₂O₂ accordingly began to converge towards the medium (no H₂O₂ spike, no active catalase spike) treatment. Flow cytometry, conducted on low/medium/high samples at 8 × 24 h intervals over the experiment duration, measured no significant (ANOVA, $p > 0.05$) difference between the 3 treatments for cell counts of any group (bacteria are shown as an example, Fig. 9 (b)).



395

Figure 9: (a) H_2O_2 gradient during the 20 L scale Mediterranean side experiment where a H_2O_2 gradient was created with H_2O_2 spikes and catalase (b) bacteria abundance during the same Mediterranean experiment (c) bacteria abundance for a similar incubation in Gran Canaria. Mean and standard deviations of triplicate treatments are plotted in all cases.

400 A similar side experiment was conducted in Gran Canaria, but one critical difference was the addition of macronutrients at the start of the experiment, as per the mesocosm at the same location (Table 2). Measurement of H₂O₂ concentrations, which were initially 43 ± 1 nM (mean of all 3 × 3 replicates at day 0), confirmed that a gradient was maintained over the 5-day duration of the experiment (mean 210 ± 113, 62 ± 14 and 47 ± 8 nM in the high, medium and low H₂O₂ treatments, respectively). Some modest shifts in phytoplankton group abundance were observed over the duration of this experiment.

405 Slightly higher cell counts of bacteria were consistently observed in the low H₂O₂ treatment relative to the medium and high H₂O₂ treatment (Fig. 9 (c)). Only the difference between the low and medium/high treatments was significant (ANOVA, p=0.028)- no significant difference was found between the medium and high H₂O₂ treatments (ANOVA, p=0.81).

4 Discussion

4.1 Bacteria, zooplankton and extracellular H₂O₂ trends

410 During all meso/multi/micro experiments and the Gran Canaria mesocosm (Table 1), data was available on the abundance of bacteria and zooplankton throughout the experiment. We focus on zooplankton because of the top-down control they may exert on primary production and the potential for grazing to release trace species into solution which may affect H₂O₂ biogeochemistry. Bacteria were a key focus because of the hypothesis that bacteria are, via the production of peroxidase/catalase enzymes, the main sink for H₂O₂ in surface aquatic environments (Cooper et al., 1994).

415

Throughout, no clear effect was evident of changing pH on H₂O₂ concentrations. The 440-1450 μatm pCO₂ gradient applied in Gran Canaria, which corresponded to a pH range of approximately 7.5-8.1, and the contrasting ambient/low pH (a reduction in pH of 0.4-0.5 from ambient waters was imposed) applied during 3 multistressor incubations (Table 2) exhibited no obvious change in equilibrium extracellular H₂O₂ concentration. Similarly no change was evident in Gran Canaria when

420 contrasting the diurnal cycling of H₂O₂ in the 400 and 1450 μatm pCO₂ treatments (Hopwood et al., 2018b). In the incubation experiments, whenever there was a sustained difference in extracellular H₂O₂ concentrations between treatment groups (MesoMed Fig. 2 and MultiMed Fig. 4), the main difference arose between ‘high’ and ‘low’ zooplankton addition treatments. However, determining the underlying reason for this was complicated by the shifts in zooplankton abundance during the experiments (e.g. Fig. 2 (b)).

425

The MultiPat (Fig. 5) and MicroPat (Fig. 6) incubations showed no significant effect of increased zooplankton abundance on extracellular H₂O₂. Two reasons for this can be considered. First, in Patagonia the initial ratio of zooplankton between the high and low treatments was the smallest of the experiments herein (17:14) and thus a large difference might not have been anticipated compared to the experiments where this initial ratio was always considerably higher. However, the mean ratio of

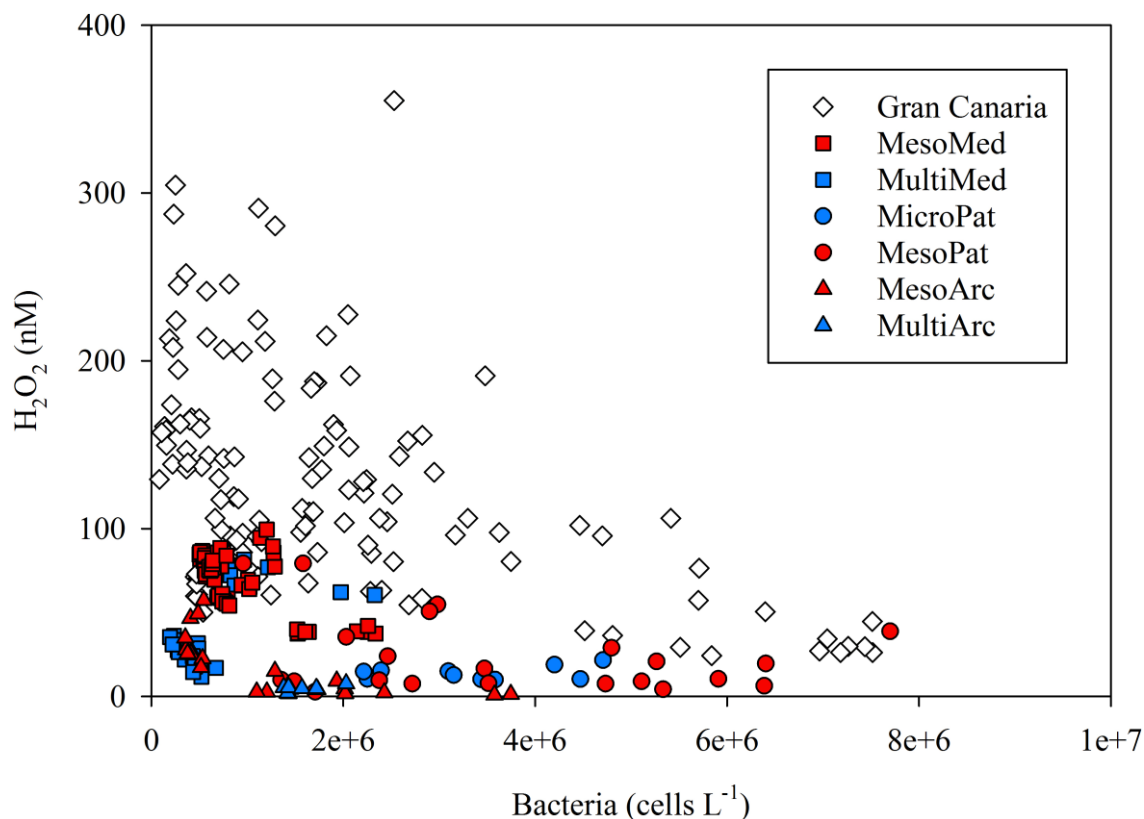
430 HG:LG zooplankton by the end of MultiPat had increased to 9:5. By comparison, during MesoMed (when the HG:LG

zooplankton abundance converged during the experiment, Fig. 2(b)) the HG:LG ratio after day 1 varied within the range 0.32-1.6 and thus the final ratio of 1.8 in MultiPat was not particularly low. A more distinct difference however arose in bacterial productivity (Fig. 5 (b)). Unlike MesoMed, MultiPat and MicroPat showed little difference in bacterial productivity between the high and low grazing treatments. Thus the effects of zooplankton with respect to shifts in the abundance of other microbial groups (rather than grazing itself) may be the underlying reason why extracellular H₂O₂ concentrations sometimes, but not consistently, changed between high and low grazing treatments. Second, in any case H₂O₂ concentrations at the end of the Patagonian experiments (MesoPat, MicroPat and MultiPat) were also very low (almost universally <20 nM) and thus the signal:noise ratio unfavourable for detecting differences between treatments.

Furthermore, the effect of higher zooplankton populations was not a consistent positive/negative change in extracellular H₂O₂. During the post-nutrient addition phase in Gran Canaria, the single treatment with slower nutrient drawdown (mesocosm 7) due to high grazing pressure exhibited relatively high H₂O₂ (Hopwood et al., 2018b). During MesoMed, increases in zooplankton abundance coincided with decreases in H₂O₂ concentration (Fig. 2). Similarly, during MultiMed (Fig. 4), the effect of adding zooplankton was the same; high zooplankton treatments exhibited low H₂O₂ concentration. As high zooplankton are correlated during some experiments, and anti-correlated in others, with H₂O₂, the underlying cause did not appear to be that H₂O₂ is generally produced by the process of grazing (i.e. as a by-product of feeding). Further support for this argument was found in the results of a simple side experiment adding copepods (*Calanus finmarchicus*) to a diatom culture (*Skeletonema costatum*) (Fig. 8). No measurable change in extracellular H₂O₂ concentration was found at higher densities of copepods either during a 16 h light incubation, or after 4 h of incubation in the dark (Fig. 8). There are two obvious limitations in this experiment; a different result may have been obtained with a different combination of copepod and phytoplankton, and standard f/2 medium contains the ligand ethylenediaminetetraacetic acid (EDTA) which may affect H₂O₂ formation rates by complexing trace species involved in H₂O₂ cycling (e.g. dissolved Fe and Cu). Nonetheless, it is known that cellular ROS production rates vary at the species level (Schneider et al., 2016; Cho et al., 2017), so shifts in species composition as a result of zooplankton addition are a plausible underlying cause of changes in extracellular H₂O₂ concentration. We summarise that any correlation between H₂O₂ and zooplankton thereby appears to have arisen from the resulting change in the abundance of microbial species, and thus the net contribution of biota to extracellular H₂O₂ concentration, rather than from the act of grazing itself.

Bacteria are expected to be a dominant H₂O₂ sink in most aquatic environments (Cooper et al., 1994). Here the correlation between extracellular H₂O₂ and bacteria cell counts was much stronger in some experiments than others (R² from 0.09-0.55). A key reason for this may simply be the generally low H₂O₂ concentrations measured in most of our experiments. At the low H₂O₂ concentrations of <50 nM observed during most experiments, the influence of any parameter on H₂O₂ removal would be more challenging to determine from an analytical perspective due to reduced signal:noise ratio. However, the H₂O₂-defence mechanism of organisms may also be sensitive to ambient H₂O₂ concentrations. Morris et al., (2016) suggest that

465 microbial communities exposed to high H_2O_2 have elevated H_2O_2 defences. If the microbial communities here exhibited a dynamic response to H_2O_2 concentrations in terms of their extracellular H_2O_2 removal rates, this would dampen the correlation between bacterial abundance and H_2O_2 concentrations. Combining all available H_2O_2 concentrations for which the corresponding total bacterial cell counts are available (Fig. 10) from all experiments (except the side experiments where H_2O_2 was manipulated using catalase or H_2O_2 spikes), provides some limited evidence for the dominance of bacteria as a
470 H_2O_2 sink. There was a notable absence of high- H_2O_2 , high-bacteria datapoints in any experiment (Fig. 10). The observed distribution is therefore consistent with a scenario where bacteria dominate H_2O_2 removal, but other factors (possibly including experiment design, see s4.2) can also lead to low H_2O_2 conditions independently of bacterial abundance.



475 **Figure 10: Bacterial cell counts and H_2O_2 for all available data from all incubation experiment time-points where both measurements were made within 24 h of each other.**

4.2 Changes in extracellular H_2O_2 due to experiment design

When all available H_2O_2 datapoints were normalized to ambient H_2O_2 at the respective fieldsite, which varied between our locations (Table 3), some qualitative inter-experiment trends were evident. Experiments incubated with artificial lighting (MultiPat/Arc/Med and MicroPat) generally exhibited the lowest concentrations, while higher normalized H_2O_2
480 concentrations were observed in the closed HDPE mesocosms (MesoMed, MesoPat, MesoArc) and then the open Gran

Canaria mesocosm experiment (Fig. 11 (b) and (c)). This is not surprising considering the light arrangements for these experiments (Table 2). The Gran Canaria experiment was practically unshaded with surface seawater exposed to natural sunlight. The closed HDPE mesocosms (MesoMed, MesoPat, MesoArc) experienced natural sunlight but after attenuation through 1-2 cm of HDPE plastic. Whilst the transmission of different light wavelengths through these HDPE containers was not tested during our experiments, 1-2 cm of polyethylene should strongly attenuate the UV component of sunlight. The 20 L scale experiments (MultiMed, MultiPat, MultiArc and MicroPat) were conducted using identical synthetic lighting with lamps selected to as closely as possible replicate the wavelength distribution of natural sunlight. However, the fluorescent light distribution is still deficient, relative to sunlight, in wavelengths <400 nm, which is the main fraction of light that drives H₂O₂ formation in surface seawater (Kieber et al., 2014), and these containers still mitigated the limited UV exposure with a 1 mm HDPE layer which would further reduce anythe UV component of incoming light.

Location	Season	Latitude	Salinity	Temperature / °C	H ₂ O ₂ / nM
Taliarte, Gran Canaria	March 2016	30.0° N	36.6-36.8	18-19	10-50
Gouves, Crete	May 2016	35.3° N	NA	19-20 ^a	34-410 ^b
Comau fjord, Patagonia	November 2014	42.4° S	3.9-12.8	9.7-13	120-680
Kongsfjorden, Svalbard	July 2015	78.9° N	9.0-35.2	5.0-9.0	10-100

Table 3. Range of water properties in freshly collected coastal seawater at each site where the mesocosms were conducted. ‘NA’ not applicable. ^a Temperature of pool facility at HCMR, ^b Coastal seawater approximately 500 m from HCMR facility.

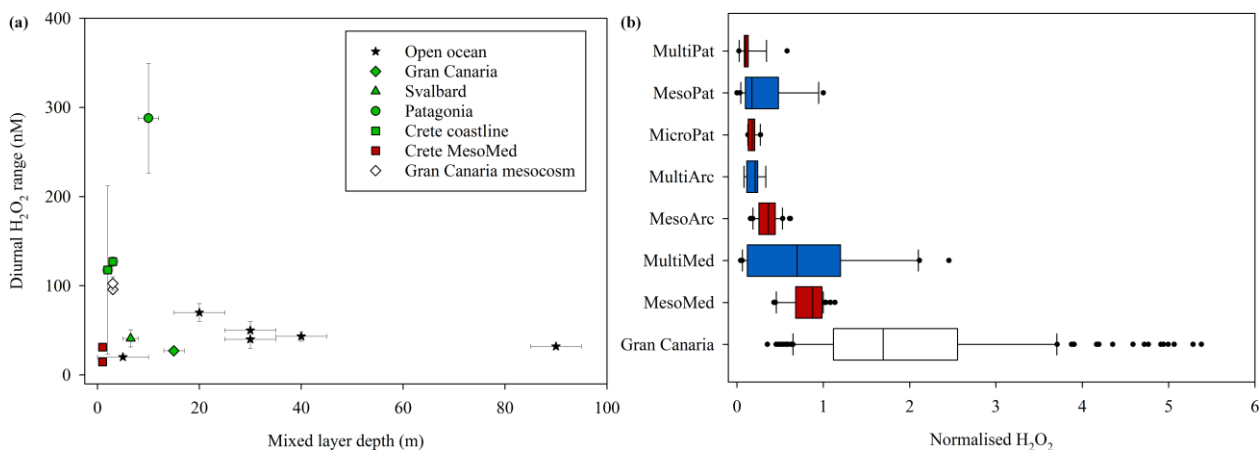


Figure 11: (a) Observed diurnal ranges in H₂O₂ concentrations. Black stars show literature surface marine values and green shapes in-situ experiments corresponding to experiment field site locations (b) H₂O₂ across all experiments as a fraction of ambient H₂O₂. For the Meso/Multi fieldsites (Mediterranean, Arctic and Patagonia) red bars are outdoor mesocosms and blue shapes indoor incubations. Median, 10th/25th/75th/90th percentiles and all outliers are shown.

During all periods when high resolution H₂O₂ time series were obtained, a clear diurnal trend was observed with a peak in H₂O₂ concentration occurring around midday (Fig. 7). Yet the range of concentrations within the two MesoMed diurnal

experiments (31.2 ± 2.3 nM and 14.5 ± 2.7 nM) was limited compared to those observed previously within a Gran Canaria mesocosm (96 ± 4 and 103 ± 8 nM, Hopwood et al., 2018). For comparison, the diurnal ranges reported in further offshore surface waters of the Atlantic, Gulf of Mexico and sub-tropical equatorial Pacific along the Peruvian shelf are 20-30 nM (Yuan and Shiller, 2001), 40-70 nM (Zika et al., 1985) and 40 nM,¹ respectively with no clear systematic trend associated with changes in mixed layer depth (Fig. 11 (a)). Within mesocosms and the coastal mesocosm fieldsites, the range was more variable. Notably, the MesoMed diurnal ranges (15 and 31 nM) were considerably lower than that observed at two corresponding coastal sites (one monitored over a single diurnal cycle, 127 ± 5 nM; one at regular intervals over the duration of the experiment, 118 ± 94 nM). Whereas, conversely, for the Gran Canaria mesocosm the ~100 nM diurnal range was much greater than that observed (27.0 ± 3.1 nM) in ambient surface waters (Fig. 11 (a)).

510

There are inevitably limits to what can be determined from contrasting available data on H₂O₂ concentration from multiple incubation experiments due to the different experiment designs (see Table 2). Yet the experiment setup with respect to moderating light during an experiment appears to be critical to establishing the equilibrium H₂O₂ concentration and can either enhance or retard the extracellular concentration of H₂O₂ during the experiment. The diurnal range plotted for all mesocosm experiments reflected increased H₂O₂ concentrations during daylight hours. This concentration range was suppressed in the closed HDPE containers (e.g. MesoMed), yet enhanced in open polyurethane bags (Gran Canaria). Whilst not explicitly measured, a likely factor to explain this difference is the reduced UV light exposure to water within sealed HDPE containers compared to open polyurethane bags. During the multistressor and microcosm experiments, incubated indoors in 20 L HDPE containers, the diurnal range in H₂O₂ concentrations was suppressed sufficiently that no increase in H₂O₂ was apparent during simulated daylight hours. Lighting conditions, especially the UV component of light which acts as a major source of photochemical H₂O₂ formation in natural waters, for the experiments therefore could explain both the contrasting change in the diurnal range of H₂O₂ (Fig. 11a), and the shift in the gradient between bacteria and H₂O₂ under different experiment conditions (Fig. 10).

520

4.3 ROS, bacteria and the Black Queen Hypothesis

Results from experiments where H₂O₂ concentrations were manipulated were mixed. In a side experiment after MesoMed, there was no evidence of strong positive or negative effects of H₂O₂ concentrations on any specific microbial group (Fig. 9). In Gran Canaria, under different experimental conditions (macronutrients were added, whereas for the MesoMed side experiment no macronutrient spike was added), a small increase in bacterial abundance was found at low H₂O₂ concentrations (+27%, Fig. 9 (c)). This result alone should be interpreted with caution, as the addition of catalase can have other effects in addition to lowering H₂O₂ concentration (Morris, 2011), yet it is intriguing to consider the role of H₂O₂ as an intermediate in the cycling of DOM alongside the role of bacteria as the dominant H₂O₂ sink.

530

¹ Unpublished data kindly provided by Insa Rapp (GEOMAR).

Photochemistry both enhances the lability of DOM (Bertilsson and Tranvik, 1998; Keiber et al., 1990) (thus making it more bioavailable as a substrate for bacteria) and causes the direct photochemical oxidation of DOM into dissolved inorganic carbon (Miller and Zepp, 1995; Granéli et al., 1996) (thus rendering it unavailable as a substrate for bacteria). ROS may enhance both of these processes, but few attempts have been made to determine the effect of manipulating ROS concentrations on photochemical DOM degradation rates, especially in the marine environment and at nanomolar concentrations (Pullin et al., 2004). Yet in experiments using furfuryl alcohol to suppress ROS in lake water, the rate of dissolved inorganic carbon formation when exposed to light decreased 20% and bacterial populations when later incubated in this ROS-quenched water were 4-fold higher than water with 'normal' ROS activity (Scully et al., 2003) implying that ROS removal was beneficial for bacteria. The results of experiments conducted in freshwater environments are not directly applicable to the marine environment, due to the different conditions in the ambient water column, but it is plausible that a similar mechanism underpinned the increase in bacteria abundance observed in Gran Canaria following the artificial lowering of H₂O₂ concentrations (Fig. 9). A large difference in bacterial populations between the presence and absence of some ROS species (Scully et al., 2003) raises interest in how important an influence changes in ROS concentration could be on the availability of DOM for bacterial productivity in the surface marine environment when more subtle changes are made to ambient H₂O₂ concentrations. If heterotrophic bacteria are the dominant H₂O₂ sink (Cooper et al., 1994), which the observed trend between bacterial abundance and extracellular H₂O₂ across a broad range of incubation experiments is consistent with (Fig. 10), this is also interesting in light of the Black Queen Hypothesis. BQH (Morris et al., 2012) assumes that the sole major benefit of producing enzymes that remove extracellular H₂O₂ is protection against the oxidative stress associated with high H₂O₂ concentrations- which is a communal benefit (Zinser, 2018). Yet, if increasing extracellular H₂O₂ concentrations accelerate the degradation of labile DOM to dissolved inorganic carbon, a second benefit of H₂O₂ removal in productive waters is the enhanced availability of this DOM to heterotrophs. Thus, under some circumstances, it could possibly be more favourable for heterotrophic species to maintain genes associated with the removal of H₂O₂ than autotrophic species because, in addition to the shared communal benefit of lowering oxidative stress, heterotrophs would suppositionally benefit more directly than autotrophs from the enhanced stability of labile DOM under low H₂O₂ conditions. However, whilst H₂O₂ is a reactive species, at the concentrations present in the marine environment the direct effects of changing H₂O₂ concentration on the abundances of different microbial groups (e.g. Fig. 9) are clearly minor. A specific challenge with determining the effect(s) of H₂O₂ concentration on any biogeochemical processes, and vice-versa, is that the diurnal variability in H₂O₂ concentration is always large compared to inter-treatment differences in H₂O₂ concentration within individual experiments (e.g. Fig. 11). High resolution data is therefore clearly required to properly interpret H₂O₂-microbial interactions and to better quantify the subtle links between H₂O₂ cycling and microbial functioning.

5 Conclusions

565 Extracellular H₂O₂ concentrations and bacterial abundances over a broad range of incubation experiments conducted in the marine environment support the hypothesis that bacterially produced enzymes are the dominant H₂O₂ sink. If heterotrophic bacteria are generally the main sink for H₂O₂ in surface marine environments, it is of interest to determine whether changes in extracellular H₂O₂ concentration measurably affect the photochemical transformation of DOM transformation to dissolved inorganic carbon. If increasing equilibrium ROS concentrations decreases the availability of labile DOM as a substrate for heterotrophs, this may affect which group/species produce catalase/peroxidase enzymes in productive waters.

570

It was apparent from comparing multiple experiments that incubation experiment design is also a strong influence on H₂O₂ concentrations. Closed HDPE mesocosms exhibited concentrations 10-90% lower than those expected in the corresponding ambient seawater, whereas an open (lidless) mesocosm exhibited concentrations 2-6 fold higher than ambient seawater. The diurnal range in H₂O₂ within incubations was also correspondingly increased in experiments where H₂O₂ concentration was artificially high, and vice-versa where H₂O₂ concentration was artificially low, suggesting enhanced, or reduced, photochemical stress over the diurnal cycle. Incubated experiments thus poorly mimic the biogeochemistry of reactive photo-chemically formed trace species.

575

4 Author Contributions

580 MH, DP, JG, EA, DT and MA designed the study. MH, NS, DP, ØL, JG, MA, JA, SB, YH, IK, TK and TT undertook work at one or more of the mesocosm/microcosm/multistressor experiments. MH, NS, DP, ØL, JG, JA, LB, SB, YH, TK, IS and TT conducted analytical work. MH, NS, DP, SB and TT interpreted the data. MH coordinated the writing of the manuscript with input from other authors.

5 Acknowledgements

585 The Ocean Certain and KOSMOS/PLOCAN teams assisting with all aspects of experiment logistics and organisation are thanked sincerely for their efforts. Labview software for operating the H₂O₂ FIA system was designed by P Croot, M Heller, C Neill and W King. Financial aid from the European Commission (OCEAN-CERTAIN, FP7- ENV-2013-6.1-1; no: 603773) is gratefully acknowledged. JA was supported by a Helmholtz International Fellow Award, 2015 (Helmholtz Association, Germany).

6 References

590 Angel, L. D., Fiedler, U., Eden, N., Kress, N., Adelong, D. and Herut, B.: Catalase activity in macro- and microorganisms as an indicator of biotic stress in coastal waters of the eastern Mediterranean Sea, *Helgol. Mar. Res.*, 53(3), 209–218, doi:10.1007/s101520050025, 1999.

- Van Baalen, C. and Marler, J. E.: Occurrence of hydrogen peroxide in sea water, *Nature*, 211(5052), 951, doi:10.1038/211951a0, 1966.
- 595 Baltar, F., Reinthaler, T., Herndl, G. J. and Pinhassi, J.: Major Effect of Hydrogen Peroxide on Bacterioplankton Metabolism in the Northeast Atlantic, *PLoS One*, 8(4), e61051, doi:10.1371/journal.pone.0061051, 2013.
- Bertilsson, S. and Tranvik, L. J.: Photochemically produced carboxylic acids as substrates for freshwater bacterioplankton, *Limnol. Oceanogr.*, 43(5), 885–895, doi:10.4319/lo.1998.43.5.0885, 1998.
- Bogosian, G., Aardema, N. D., Bourneuf, E. V., Morris, P. J. L. and O’Neil, J. P.: Recovery of hydrogen peroxide-sensitive
600 culturable cells of *Vibrio vulnificus* gives the appearance of resuscitation from a viable but nonculturable state, *J. Bacteriol.*, 182(18), 5070–5075, doi:10.1128/JB.182.18.5070-5075.2000, 2000.
- Cho, K., Kasaoka, T., Ueno, M., Basti, L., Yamasaki, Y., Kim, D. and Oda, T.: Haemolytic activity and reactive oxygen species production of four harmful algal bloom species, *Eur. J. Phycol.*, 52(3), 311–319, doi:10.1080/09670262.2017.1286525, 2017.
- 605 Clayton, T. D. and Byrne, R. H.: Spectrophotometric seawater pH measurements: total hydrogen ion concentration scale calibration of m-cresol purple and at-sea results, *Deep Sea Res. Part I Oceanogr. Res. Pap.*, 40(10), 2115–2129, doi:10.1016/0967-0637(93)90048-8, 1993.
- Cooper, W. J., Zika, R. G., Petasne, R. G. and Plane, J. M. C.: Photochemical formation of hydrogen peroxide in natural waters exposed to sunlight, *Environ. Sci. Technol.*, 22(10), 1156–1160, doi:10.1021/es00175a004, 1988.
- 610 Cooper, W. J., Shao, C. W., Lean, D., Gordon, A. and Scully, F. E.: Factors affecting the distribution of H₂O₂ in surface waters, in *Environmental Chemistry of Lakes and Reservoirs*, vol. 237, pp. 391–422., 1994.
- Gasol, J. M. and del Giorgio, P. A.: Using flow cytometry for counting natural planktonic bacteria and understanding the structure of planktonic bacterial communities., *Sci. Mar.*, 64, 197 – 224, doi:10.3989/scimar.2000.64n2197, 2000.
- Gerringa, L. J. A., Rijkenberg, M. J. A., Timmermans, R. and Buma, A. G. J.: The influence of solar ultraviolet radiation on
615 the photochemical production of H₂O₂ in the equatorial Atlantic Ocean, *J. Sea Res.*, 51(1), 3–10, doi:10.1016/j.seares.2003.03.002, 2004.
- González-Dávila, M., Santana-Casiano, J. M., Petihakis, G., Ntoumas, M., Suárez de Tangil, M. and Krasakopoulou, E.: Seasonal pH variability in the Saronikos Gulf: A year-study using a new photometric pH sensor, *J. Mar. Syst.*, 162, 37–46, doi:10.1016/j.jmarsys.2016.03.007, 2016.
- 620 Granéli, W., Lindell, M. and Tranvik, L. J.: Photo-oxidative production of dissolved inorganic carbon in lakes of different humic content, *Limnol. Oceanogr.*, 41(4), 698–706, doi:10.4319/lo.1996.41.4.0698, 1996.
- Guillard, R. R. L. and Ryther, J. H.: Studies of marine planktonic diatoms. I. *Cyclotella nana* Hustedt, and *Detonula confervacea* (Cleve) Gran., *Can. J. Microbiol.*, 8(2), 229–239, doi:10.1139/m62-029, 1962.
- Hansel, C. M., Ferdelman, T. G. and Tebo, B. M.: Cryptic cross-linkages among biogeochemical cycles: Novel insights from
625 reactive intermediates, *Elements*, 11(6), 409–414, doi:10.2113/gselements.11.6.409, 2015.
- Hansen, H. P. and Koroleff, F.: Determination of nutrients, in *Methods of Seawater Analysis*, pp. 159–228., 2007.

- Hopwood, M. J., Rapp, I., Schlosser, C. and Achterberg, E. P.: Hydrogen peroxide in deep waters from the Mediterranean Sea, South Atlantic and South Pacific Oceans, *Sci. Rep.*, 7, 43436, doi:10.1038/srep43436, 2017.
- Hopwood, M. J., Santana-González, C., Gallego-Urrea, J., Sanchez, N., Achterberg, E. P., Ardelan, M. V., Gledhill, M.,
630 González-Dávila, M., Hoffmann, L., Leiknes, Ø., Santana-Casiano, J. M., Tsagaraki, T. M. and Turner, D.: Fe(II) stability in coastal seawater during experiments in Patagonia, Svalbard and Gran Canaria, *Biogeosciences Discuss.*, 2018, 1–29, doi:10.5194/bg-2018-439, 2018a.
- Hopwood, M. J., Riebesell, U., Arístegui, J., Ludwig, A., Achterberg, E. P. and Hernández, N.: Photochemical vs. Bacterial Control of H₂O₂ Concentration Across a pCO₂ Gradient Mesocosm Experiment in the Subtropical North Atlantic, *Front. Mar. Sci.*, 5, 105, doi:10.3389/fmars.2018.00105, 2018b.
- Hughes, C. and Sun, S.: Light and brominating activity in two species of marine diatom, *Mar. Chem.*, 181, 1–9, doi:10.1016/j.marchem.2016.02.003, 2016.
- Keiber, R. J., Zhou, X. and Mopper, K.: Formation of carbonyl compounds from UV-induced photodegradation of humic substances in natural waters: Fate of riverine carbon in the sea, *Limnol. Oceanogr.*, 35(7), 1503–1515,
640 doi:10.4319/lo.1990.35.7.1503, 1990.
- Kieber, D. J., Miller, G. W., Neale, P. J. and Mopper, K.: Wavelength and temperature-dependent apparent quantum yields for photochemical formation of hydrogen peroxide in seawater., *Environ. Sci. Process. Impacts*, 16(4), 777–91, doi:10.1039/c4em00036f, 2014.
- Larsen, A., Egge, J. K., Nejstgaard, J. C., Di Capua, I., Thyraug, R., Bratbak, G. and Thingstad, T. F.: Contrasting response
645 to nutrient manipulation in Arctic mesocosms are reproduced by a minimum microbial food web model., *Limnol. Oceanogr.*, 60(2), 360–374, doi:10.1002/lno.10025, 2015.
- Miller, W. L. and Kester, D. R.: Peroxide variations in the Sargasso Sea, *Mar. Chem.*, 48(1), 17–29, doi:10.1016/0304-4203(94)90059-0, 1994.
- Miller, W. L. and Zepp, R. G.: Photochemical production of dissolved inorganic carbon from terrestrial organic matter:
650 Significance to the oceanic organic carbon cycle, *Geophys. Res. Lett.*, 22(4), 417, doi:10.1029/94GL03344, 1995.
- Moffett, J. W. and Zafiriou, O. C.: An investigation of hydrogen peroxide chemistry in surface waters of Vineyard Sound with H₂18O₂ and 18O₂, *Limnol. Oceanogr.*, 35(6), 1221–1229, doi:10.4319/lo.1990.35.6.1221, 1990.
- Moffett, J. W. and Zika, R. G.: Reaction kinetics of hydrogen peroxide with copper and iron in seawater, *Environ. Sci. Technol.*, 21(8), 804–810, doi:10.1021/es00162a012, 1987.
- 655 Moore, C. a., Farmer, C. T. and Zika, R. G.: Influence of the Orinoco River on hydrogen peroxide distribution and production in the eastern Caribbean, *J. Geophys. Res.*, 98(C2), 2289, doi:10.1029/92JC02767, 1993.
- Morris, J. and Zinser, E. R.: Continuous hydrogen peroxide production by organic buffers in phytoplankton culture media, *J. Phycol.*, 49(6), 1223–1228, doi:10.1111/jpy.12123, 2013.
- Morris, J. J.: The ‘Helper’ Phenotype: A Symbiotic Interaction Between *Prochlorococcus* and Hydrogen Peroxide
660 Scavenging Microorganisms, University of Tennessee., 2011.

- Morris, J. J., Johnson, Z. I., Szul, M. J., Keller, M. and Zinser, E. R.: Dependence of the cyanobacterium *Prochlorococcus* on hydrogen peroxide scavenging microbes for growth at the ocean's surface, *PLoS One*, 6(2), e16805, doi:10.1371/journal.pone.0016805, 2011.
- 665 Morris, J. J., Lenski, R. E. and Zinser, E. R.: The black queen hypothesis: Evolution of dependencies through adaptive gene loss, *MBio*, 3(2), doi:10.1128/mBio.00036-12, 2012.
- Morris, J. J., Johnson, Z. I., Wilhelm, S. W. and Zinser, E. R.: Diel regulation of hydrogen peroxide defenses by open ocean microbial communities, *J. Plankton Res.*, 38(4), 1103–1114, doi:10.1093/plankt/fbw016, 2016.
- Petasne, R. G. and Zika, R. G.: Hydrogen peroxide lifetimes in South Florida coastal and offshore waters, *Mar. Chem.*, 56(3–4), 215–225, doi:10.1016/S0304-4203(96)00072-2, 1997.
- 670 Price, D., Mantoura, R. F. C. and Worsfold, P. J.: Shipboard determination of hydrogen peroxide in the western Mediterranean sea using flow injection with chemiluminescence detection, *Anal. Chim. Acta*, 377(2–3), 145–155, doi:10.1016/S0003-2670(98)00621-7, 1998.
- Pullin, M. J., Bertilsson, S., Goldstone, J. V. and Voelker, B. M.: Effects of sunlight and hydroxyl radical on dissolved organic matter: Bacterial growth efficiency and production of carboxylic acids and other substrates, *Limnol. Oceanogr.*, 675 49(6), 2011–2022, doi:10.4319/lo.2004.49.6.2011, 2004.
- Redfield, A. C.: On the proportions of organic derivations in sea water and their relation to the composition of plankton, in James Johnstone Memorial Volume, edited by R. J. Daniel, pp. 177–192, University Press of Liverpool, Liverpool., 1934.
- Reggiani, E. R., King, A. L., Norli, M., Jaccard, P., Sørensen, K. and Bellerby, R. G. J.: FerryBox-assisted monitoring of mixed layer pH in the Norwegian Coastal Current, *J. Mar. Syst.*, 162, 29–36, doi:10.1016/j.jmarsys.2016.03.017, 2016.
- 680 Riebesell, U., Czerny, J., Von Bröckel, K., Boxhammer, T., Büdenbender, J., Deckelnick, M., Fischer, M., Hoffmann, D., Krug, S. A., Lentz, U., Ludwig, A., Mucche, R. and Schulz, K. G.: Technical Note: A mobile sea-going mesocosm system - New opportunities for ocean change research, *Biogeosciences*, 10(3), 1835–1847, doi:10.5194/bg-10-1835-2013, 2013.
- Rimmelin, P. and Moutin, T.: Re-examination of the MAGIC method to determine low orthophosphate concentration in seawater, *Anal. Chim. Acta*, 548(1), 174–182, doi:10.1016/j.aca.2005.05.071, 2005.
- 685 Rundt, C.: Organic Carbon Enrichment of Mediterranean Waters: Effects on the Pelagic Microbial Food Web with Emphasis on Microzooplankton Grazing., University of Bremen., 2016.
- Schneider, R. J., Roe, K. L., Hansel, C. M. and Voelker, B. M.: Species-Level Variability in Extracellular Production Rates of Reactive Oxygen Species by Diatoms, *Front. Chem.*, 4, 5, doi:10.3389/fchem.2016.00005, 2016.
- Scully, N. M., Cooper, W. J. and Tranvik, L. J.: Photochemical effects on microbial activity in natural waters: The 690 interaction of reactive oxygen species and dissolved organic matter, in *FEMS Microbiology Ecology*, vol. 46, pp. 353–357., 2003.
- Smith, D. C. and Azam, F.: A simple, economical method for measuring bacterial protein synthesis rates in seawater using 3H-leucine, *Mar. Microb. Food Webs*, 6(2), 107–114, 1992.
- Voelker, B. M. and Sulzberger, B.: Effects of Fulvic Acid on Fe(II) Oxidation by Hydrogen Peroxide, *Environ. Sci.*

- 695 Technol., 30(4), 1106–1114, doi:10.1021/es9502132, 1996.
- Welschmeyer, N. A.: Fluorometric analysis of chlorophyll a in the presence of chlorophyll b and pheopigments, *Limnol. Oceanogr.*, 39(8), 1985–1992, doi:10.4319/lo.1994.39.8.1985, 1994.
- Yuan, J. and Shiller, A.: The distribution of hydrogen peroxide in the southern and central Atlantic ocean, *Deep Sea Res. Part II Top. Stud. Oceanogr.*, 48(13), 2947–2970, doi:10.1016/S0967-0645(01)00026-1, 2001.
- 700 Yuan, J. C. and Shiller, A. M.: Determination of subnanomolar levels of hydrogen peroxide in seawater by reagent-injection chemiluminescence detection, *Anal. Chem.*, 71(10), 1975–1980, doi:10.1021/ac981357c, 1999.
- Zika, R. G., Moffett, J. W., Petasne, R. G., Cooper, W. J. and Saltzman, E. S.: Spatial and temporal variations of hydrogen peroxide in Gulf of Mexico waters, *Geochim. Cosmochim. Acta*, 49(5), 1173–1184, doi:10.1016/0016-7037(85)90008-0, 1985.
- 705 Zinser, E. R.: Cross-protection from hydrogen peroxide by helper microbes: The impacts on the cyanobacterium *Prochlorococcus* and other beneficiaries in marine communities, *Environ. Microbiol. Rep.*, doi:10.1111/1758-2229.12625, 2018.