

## Response to comments by anonymous referee #1

We thank the reviewer for her/his insightful comments and questions.

This work studied the effects of drying-wetting, soil aggregate size, litter addition and plant on N<sub>2</sub>O flux from floodplain soils. The authors used model soils and mesocosm experiments to conduct the research. As far as I can say, there are still much more space can be improved for this manuscript.

In general, it is interesting to know how soil N<sub>2</sub>O flux are controlled by different environmental factors. However, there are already many studies conducted in no matter drying-wetting and soil aggregation, or litter addition and vegetation effects. What the knowledge gaps do you want to fill? It should be clarified in the introduction part.

Reply: We concur with the reviewer that the specific objectives of this study were not sufficiently well stated. While the effects of microhabitats related to soil aggregates, the detritosphere and plant-soil interactions in the rhizosphere on N<sub>2</sub>O emissions from soils have been studied individually, little is known about their relative effects and interactions. In a mesocosm study, we investigated this aspect for the hot moments of N<sub>2</sub>O emissions from floodplain soils during the drying phase after flooding. In particular, aggregate size effects have not been investigated in this context (as stated on lines 80ff). A particular novel aspect of the study is the minimization of the potentially confounding factor “soil structure” by mixing a given aggregate size fraction with inert material replacing the removed smaller or larger fraction. As stated on line 71ff, previous studies employing isolated aggregate size fractions have provided partially inconsistent results.

The innovative aspects of objectives, will be further clarified in the introduction of the revised manuscript, with emphasis on the relevance of the research, and addressing also the potential with regards to filling knowledge gaps.

Here are technical questions:

1. Line 14-15, it is not accurate to write the buried organic matter and rhizosphere processes. Actually, the experiments were about litter addition and plant vegetation. It still takes several steps from litter to organic matter. And also, you didn't took the rhizosphere samples.

R1: We agree that the term “buried organic matter” is too unspecific. Although, *sensu stricto*, litter is “organic matter” as well, it indeed might be confused with further decomposed and transformed “soil organic matter”. We therefore have replaced “buried organic matter” with “buried litter”. We checked the entire manuscript and this is the only place where we used this rather unspecific term.

We also agree that at this point “rhizosphere processes” should be replaced by “plant-soil interactions”, even though in the later discussion mainly rhizosphere processes per se are invoked to explain the observed plant effects.

2. L148, for soil pH measurement, normally it is 10 g soil was mixed with 25 mL solution. The authors used 20 mL of solution, any references? The solution can be water or CaCl<sub>2</sub>, as far as I know, for alkaline soil, it is better to use water. In this study, the soil pH were ~ 8, any reasons to choose CaCl<sub>2</sub>?

R2: There are several soil-to-solution ratios recommended in the literature, among them also 1:2.5 (Blume et al. 2010. Scheffer/Schachtschabel – Lehrbuch der Bodenkunde, 16<sup>th</sup> ed., p. 151), or 1:1 (Thomas G.W. 1996. «Soil pH and Soil acidity» In: Sparks et al. (eds.) Methods of soil analysis – 3. Chemical methods. SSSA Book Series 5, pp. 475ff.). A soil-to-solution ratio of 1:2 for mineral soil samples – as has been used in our laboratories since more than 30 years – is also recommended by one of the newest method handbooks: Hendershot et al.

(2008) "Soil reaction and exchangeable acidity" In: Carter, M.R. (ed.) Soil sampling and methods of analysis. 2<sup>nd</sup> ed., Can. Soc. Soil Sci., chapter 16. Furthermore, this handbook, citing several individual studies, recommends CaCl<sub>2</sub> as suspending solution with several advantages over water, in particular also for agricultural soils whose pH is often comparatively high. There is no mentioning in this, or any of the other cited references, of a disadvantage in using CaCl<sub>2</sub> for carbonate containing soils.

More generally, soils are heavily buffered systems and the measured pH should be virtually independent of such small variation in ionic strength.

3. Have the authors ever considered the emission/uptake of N<sub>2</sub>O by the aboveground of plant? There are already many studies in this field, such as: Smart D R, Bloom A J. Wheat leaves emit nitrous oxide during nitrate assimilation[J]. Proceedings of the National Academy of Sciences, 2001, 98(14): 7875-7878. In this study, the authors measured N<sub>2</sub>O flux from the mesocosm have both soil and plant. This flux cannot be called soil flux, but may be soil/plant flux?

R3: In the introduction (line 96ff) we considered potential bypassing of the soil matrix by N<sub>2</sub>O fluxes via plant-internal aeration channels (aerenchyma). This phenomenon is well documented for *Poaceae* such as the Genus *Oryza* or *Phalaris arundinacea*. However, for willows (*Salix sp.*) such a process has, to our knowledge, not been documented yet. Although, considering that also adventitious roots of *Salix* species contain aerenchyma, we cannot exclude this process to occur in our case, our results do not indicate an enhanced N<sub>2</sub>O emission via the plant, since we observed the lowest flux rates as well as lowest total integrated emissions in the mesocosms with plants. Therefore we conclude that in our experiment, such a process, if present, was of minor importance in terms of modulating net N<sub>2</sub>O fluxes to the atmosphere. However, we agree that the possibility that part of the N<sub>2</sub>O fluxes from the planted soils occurred via plant-internal channels should be mentioned in discussion section 4.3. We also agree that emission fluxes should be termed "soil/plant flux" or "ecosystem flux" instead of "soil flux". Although nowhere in the manuscript we have used the term "soil flux", we agree that we need to clarify at respective prominent places in the manuscript that in the case of the treatments with willow emissions/fluxes relate to the whole soil/plant system and not to the soil alone.

4. L274, the author can show the data in support information.

R4: we will upload a file containing the supplementary information and adjust the text accordingly.

5. L313-315, the authors didn't check the statistics difference of soil chemical/physical properties between different treatments. Therefore, the hypothesis is not really correct before statistics analysis were done.

R5: The comparison of the initial physicochemical properties by t-tests with Welch's correction showed statistically significant differences for the C:N ratio and pH. However, C:N ratios of 12 and 16 can be considered ecologically similar in terms of soil organic matter degradability, in particular since both C<sub>org</sub> and total N do not differ that much. The higher pH in the macroaggregated model soil is probably due to a higher carbonate content, which also is not expected to strongly affect biogeochemical processes of the N cycle. These remarks will be added in the revised manuscript, and a new column will be added in table 1 with the results of the statistical analyses.

6. L346-347, Actually WFPS-SA value were not decreased to pre-flood even until the end of experiments (Fig. 2 a and b). The explanation might be low diffusion rate of N<sub>2</sub>O in SA treatments caused reduction of N<sub>2</sub>O to N<sub>2</sub>?

R6: Considering the high WFPS in the SAU treatment, the referee's remark represents a valid explanation for the observed low fluxes under the given circumstances. However, the relatively high redox potentials, which we invoke here, argue against sufficient anoxia for complete reduction of N<sub>2</sub>O to N<sub>2</sub>. Nevertheless, we will include this aspect in the discussion in section 4.1. of the revised manuscript.

7. L409, delete one dot

8. L457, delete DOI

9. Table 2, it would be better to explain the meanings of LAU, SAU....in the table caption.

10. L638-639, no dotted line in Fig. 3?

11. Fig. 2, it would be better to put WFPS in the right Y axis. And put WFPS-LA, WFPS-SA....in the figure legend.

12. Fig. 3e, the data are not completely shown.

13. Fig. 4, would be better to have the same unit (μM) for nitrate and nitrite/ammonium

R7-13: the authors consent with all these remarks and will make changes to the revised manuscript accordingly.

## Response to comments by Y. A. Teh (Referee)

We thank the reviewer for his supportive evaluation, insightful comments and questions. Addressing them will strongly improve the manuscript.

### GENERAL COMMENTS

This is a creative and interesting process-based experiment that uses different aggregate treatments (i.e. micro- versus macro-aggregate dominated) and plant-soil treatments (i.e. a gradient of “plant influence,” from rhizosphere to detritus-affected soil to plant-free soil) to determine how differences in soil structure and various levels of plant influence potentially influence N<sub>2</sub>O dynamics in soil. The factorial experimental design is powerful because it enables the investigators to assess not only main effects, but also evaluate the potential importance of synergistic effects among different treatments. Overall, it is my view that this paper was clearly written, with a well-justified experimental design, and a logical analysis of the data. The introduction to the paper clearly explains the basis and wider significance of this research, while the methods section explains the overall approach taken with clarity. The results section documents the main findings of the work succinctly, while the discussion takes a reasonable (and not overly speculative) approach to data interpretation, informed by the authors’ grasp of the current literature. The investigators’ comprehensive measurement of a range of environmental parameters is to be commended and enables them to make logical inferences about the role of different treatments and environmental factors in regulating N<sub>2</sub>O dynamics during different parts of the simulated water cycle. In particular, the investigators make good use of redox potential measurements to evaluate how changes in redox/O<sub>2</sub> availability could be driving N dynamics along the “plant influence” gradient that they have created in the laboratory.

However, while I am generally supportive of this research and believe it will make a valuable contribution to the wider body of knowledge on this topic, I do have a few general remarks that I believe need to be addressed before this paper can go forward to publication. First, I think the authors need to be open and transparent about the potential limitations of their research. For example, the soil structure treatments represent two extremes (large versus small aggregates), whereas in reality micro- and macro-aggregates would be mixed together. The authors need to explain how their experimental treatment could relate or correspond to real-world conditions, drawing if possible on pre-existing field or laboratory data (see points 1 and 5 below).

**R 1:** We agree that including a discussion of the implicit limitations of our experimental approach with respect to natural conditions will contribute to a better evaluation of the results of our study, and we thus will include this in a revised version of the manuscript.

By investigating two pedogenetically well-defined aggregate size fractions (4000 – 250  $\mu\text{m}$  and 250 – 0  $\mu\text{m}$ ; Tisdall and Oades, 1982) separately – but with soil structure

kept similar by replacing the removed fraction by inert material of the same size - , we aimed at evaluating the individual potential of these fractions to offer conditions for the soil microbial community to form N<sub>2</sub>O. Following the reviewer's suggestion, we propose to include a discussion of how these conditions relate to real-world conditions as follows. As detailed in our response R1 below, these two size fractions represent significant "components" both of our investigated original soil and of most other soils. However, we intentionally excluded interactions between the two soil aggregate size fractions to assess the individual potential of each fraction separately. Therefore we can neither assess any interactions between large and small aggregates, nor such with soil structures larger than 4mm, which all may also be important for N<sub>2</sub>O emissions under natural conditions. Since we have no data related to this, we prefer not to speculate about such effects in our paper.

Likewise, the authors need to be clearer about the limitations underlying their rhizosphere (Salix) treatment. It is difficult to generalise more widely about the effects of plant rhizospheres on N dynamics without examining a range of different plants (including single and multi-species mixtures), in order to tease-apart individual species effects from generic rhizosphere effects (see point 6 below); I think it is important, in the revised version of this text, that the authors acknowledge this limitation and spend a bit more time exploring what they believe could be more widely generalisable from their study, rather than what is species-specific.

**R II: For a reply the reader is kindly referred to R6**

Second, I do not believe that the authors have fully exploited their experimental design in the analysis of their data, and sincerely believe that more could be done to examine these data in greater depth. For example, as mentioned above, one of the strengths of a factorial experimental design is that the investigators can establish if there are synergistic interactions among different experimental treatments (e.g. aggregate X rhizosphere effects). However, the investigators do not appear to have examined if interactions among treatments occurred, or at least these findings are not reported if these tests were conducted. Moreover, I would suggest that the authors try more complex multivariate models to analyse their data; for instance, using approaches such as analysis of co-variance (ANCOVA), generalized linear models, or mixed effects models. The benefit of these more comprehensive multivariate models is that they enable the investigator to establish the relative importance of different treatments and continuous environmental variables in regulating flux.

**R III: We fully agree that an experiment has to be analyzed according to its experimental design. In our case, this includes the interaction of aggregate size and soil treatment (unamended, litter addition, plant presence). We in fact have included this term in all ANOVA models, but failed to report the results when the term was not statistically significant or only weakly significant. We will fix this in the revised version. The structure of our experimental treatments is not hierarchical so that no mixed model is required. Such a model would only be necessary if one would analyse the**

time series data, i.e. if one had several values per microcosm. We have considered this but decided not to do so, for the following reasons:

(1) our focus was on the *average response* during distinct phases that we have identified in our time series, in particular during “hot moments” after wetting; working with average time-series data provides an answer to hypotheses about whether total emissions during this period, for example, differ between treatments; in other words, our hypotheses were about cumulated fluxes during a period, and we therefore carried out these analyses at this level.

(2) the processes we observed are extremely dynamic; fitting a full time series model would almost certainly have resulted in significant time x treatment interactions – such effects would primarily be driven by the peak values of e.g. N<sub>2</sub>O emissions after wetting; whether treatment differences for these single measurements reflect true differences in time and extent of peak fluxes is uncertain... it in fact is very likely that the true peak occurred a short time before or after these measurements, and this may be treatment specific. Again, we were not interested in whether the maximum flux occurred a bit earlier or later in time (this may not be reproducible anyways), but whether total emissions during the hot moment changed. Working with such aggregated data solves the problem of subtle shifts in emission timing, and gives extreme values much less weight.

(3) the proper modelling of the time series is very complicated: this involved heterogeneous variances (because large values scatter more) and the modelling of serial correlations (because subsequent values are not independent). On the time-aggregated scale, these problems do not occur. We also could log-transform the data to compare the treatments, which was not possible on the raw data because (a) negative values occurred due to measurement error, and (b) we were asking questions about total fluxes (e.g. grams of N<sub>2</sub>O emitted) and not relative effects.

In summary, we agree that more complex analyses can potentially be done. However, we have deliberately focused on (1) the aggregation level that matched the questions we were asking, and (2) the aggregation level at which statistical procedures were robust. We agree that we did not document this very well and propose to address this in the revision.

Third, I agree with the first referee that the authors need to spend a bit more time clearly highlighting what knowledge gaps this paper fills. As the first referee indicates, there are already existing studies that have examined the individual effects of all the variables discussed here. In order to make this paper more impactful, the authors need to articulate how this specific study is unique or advances our current state-of-knowledge (e.g. does the factorial design add knowledge or insight?). Specific comments are provided in the section below.

R IV: We concur with both reviewers that the specific objectives of this study were not sufficiently well stated. As mentioned in our response to Reviewer 1, this aspect will be addressed. We will clarify that, while the effects of microhabitats related to soil aggregates, the detritosphere and plant-soil interactions in the rhizosphere on N<sub>2</sub>O emissions from soils have been studied individually, little is known about their relative

effects and interactions. In our mesocosm study, we investigated this aspect for the hot moments of N<sub>2</sub>O emissions from floodplain soils during the drying phase after flooding. In particular, aggregate size effects have not been investigated in this context (as stated on lines 79f). A particular novel aspect of the study is the minimization of the potentially confounding factor “soil structure” by mixing a given aggregate size fraction with inert material replacing the removed smaller or larger fraction. As stated on line 71ff, previous studies employing isolated aggregate size fractions have provided partially inconsistent results possibly linked to some extent to the changes in soil structure by aggregate separation.

The better specified objectives and novel aspects will be included in the introduction of the revised manuscript.

### SPECIFIC COMMENTS

1. Lines 136-137: For experimental purposes, the investigators have created quasi-artificial system conditions, with treatments either containing macro- or microaggregates. While I fully understand why this was done, it would be useful to understand (even qualitatively) how close or far from reality these treatments are. For example, what was the proportion of macro- and micro-aggregates under natural conditions?

R1: The original floodplain soil consisted of  $18.5 \pm 4.6$  % aggregates smaller than 250  $\mu\text{m}$  and  $81.5 \pm 4.6$  % macroaggregates (mean  $\pm$  sd; n = 10). We composed our model soils of a 1:1 mixture of isolated aggregates and inert matrix material. This is different from the original soil composition, but well within the range of published top soil aggregate size distributions (e.g. Cantón et al., 2009; Gajić et al., 2010; Six et al., 2000). 50% microaggregates may be more than what is found in most natural or agricultural soils. Nevertheless, we chose to use equal amounts of small and large aggregates to be able to separate effects of aggregate size from effects of aggregate amount (soil mass). To reflect these reasonings, we propose to discuss the distribution of small and large aggregates in the original soil (material and method section of the revised manuscript). The discussion of relevance would be added to the discussion in section 4.1 and in the conclusions. For additional considerations on the effect of flood disturbance on small-scale heterogeneity and dynamics of aggregate size distribution see R5 below.

2. Line 173: Clarity of expression; consider revising this section to read “The mesocosm experiment had a factorial experimental design consisting of two factors (model soil and plant-soil treatment), with the first factor containing two levels (macroaggregates, microaggregates) and the second factor containing three levels (unamended, litter added, plant present). This experimental design resulted in six treatments, each replicated six times.”

R2: The authors concur with this remark and will adjust this part accordingly

3. Line 179-180: What was the rationale for autoclaving the leaves? Under natural conditions, these leaves would contain their own microbial community which could contribute to N<sub>2</sub>O dynamics, and autoclaving means that the results will be biased towards the activity of the soil community (or, spore-forming phyllosphere microbes able to resist the effects of autoclaving).

R3: Since we specifically wanted to test the effect of additional labile C available to the N<sub>2</sub>O producing or consuming soil microbial community, we decided to eliminate, or at least reduce the effect of and interaction with the phyllosphere of the collected leaves by sterilization. We are aware that this introduces a certain bias. However, so far there are no direct effects of the phyllosphere community on N<sub>2</sub>O production described in the literature. The only role of these organisms in plant-atmosphere interactions reported in the literature is in capturing/consuming methane and/or volatile organic carbon compounds (Bringel and Couée, 2015). On the other hand, we cannot say anything about potential effects of interactions between the phyllosphere and soil communities on N<sub>2</sub>O production/consumption. These remarks will be added to the discussion section of the litter effects, 4.2., in the revised version of the manuscript.

4. Lines 221-232: Further detail on the statistical analyses are required here. For example, what were the independent variables used in the ANOVA? Did the model include interaction terms? Given that sampling was conducted over different periods of time, did the authors use a repeated measures ANOVA, to account for the effects of time?

R4: The independent variables for the two way ANOVA were SOIL TREATMENT (unamended, litter addition, plant presence) and AGGREGATE SIZE. The ANOVA model also included interactions, which were indeed significant for some of the parameter. However, we did not report the cases where the interaction was not or only weakly statistically significant. We will address this in the revision.

Our hypotheses were related to total fluxes during hot moments, which is why we did not analyze the time series but aggregated data. The rationale for this was already explained in detail above (R III).

5. Lines 300-353: This is an interesting and well-written part of the discussion. However, I do think that this part of the discussion could be improved by trying to link back the findings from the experiment to natural conditions (see point 1). For example, under natural conditions, what is the relative distribution of macro- or micro-aggregates? Based on your understanding/knowledge of the natural aggregate distributions, what patterns or processes do you think will dominate in a natural setting? While I realise this might be somewhat speculative (unless other data, such as field measurements, are available), I think it's an important talking point, as it will enable the reader to relate these findings (derived under somewhat artificial conditions) to the real world.



R5: For our assessment and evaluation of the relative distribution of macro- and micro-aggregates in our experimental soil and other soils reported in the literature see R1.

Furthermore, the frequent hydrological disturbance in floodplains creates a highly dynamic and small-scaled spatial mosaic of different aggregate size distributions. Therefore, the results on the individual potentials of differently sized aggregates to emit N<sub>2</sub>O and their respective interactions with plant roots and litter accumulation could help to better understand the seemingly erratic spatial and temporal distribution of enhanced N<sub>2</sub>O emissions from floodplain areas. Considering our results, one could speculate that zones with a relatively high percentage of macroaggregates would be prone to particularly high emissions during hot moments. In a revised manuscript, these considerations would be added also to the discussion in section 4.1.

6. Lines 380-406: The discussion of potential direct and indirect effects facilitated by the presence of an active root system is interesting and well-reasoned. However, I was left wondering as to how generalizable these findings are, given the wide range of traits displayed by different plants? I.e. to what extent are the trends identified here unique to *Salix*, and to what extent are these patterns more widely generalizable? I think it is important that the authors develop this section a bit further, in particular acknowledging this limitation more frankly.

R6: Different plant species may indeed exert different rhizosphere effects (for an overview of potential rhizosphere effects see the current manuscript lines 81 to 101). Thus, strictly speaking, this study is directly relevant only for *salix* sp.. However, this is an important plant genus adapted to temporary flooding and thus often found in river floodplains. While oxygen depletion by root exudation stimulated microbial respiration, discussed as one process potentially reducing N<sub>2</sub>O emissions in our study, likely occurs in the rhizosphere of any plant, rhizosphere aeration as alternative process is restricted to plants possessing aerenchyma. However, the latter is a trait of many plants adapted to temporary flooding. It has been described also for the grass family of poaceae, or for ash, and It would not be surprising to find this trait in other Salicaceae like poplar sp. and other species of softwood floodplain forests.

#### References:

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- Cantón, Y., Solé-Benet, A., Asensio, C., Chamizo, S. and Puigdefábregas, J.: Aggregate stability in range sandy loam soils Relationships with runoff and erosion, *CATENA*, 77(3), 192–199, doi:10.1016/j.catena.2008.12.011, 2009.
- Gajić, B., Đurović, N. and Dugalić, G.: Composition and stability of soil aggregates in Fluvisols under forest, meadows, and 100 years of conventional tillage, *J. Plant Nutr. Soil Sci.*, 173(4), 502–509, doi:10.1002/jpln.200700368, 2010.
- Six, J., Paustian, K., Elliott, E. T. and Combrink, C.: Soil Structure and Organic Matter, *Soil Sci. Soc. Am. J.*, 64(2), 681, doi:10.2136/sssaj2000.642681x, 2000.

# List of all relevant changes

## Adjustments according to our responses to comments by anonymous referee #1

- Line 14: Terminology was specified in the abstract (Ref. 1, R1)
- Line 120 – 127: An improved objectives section has been added to the introduction section (Ref. 1, general reply; Ref. 2, RIV)
- Line 157: An appropriate reference was added (Ref. 1, R2)
- Line 295: Added reference to supplementary information (Ref. 1, R4)
- Line 335 – 337: remarks concerning statistical analysis of the initial physicochemical soil properties and adjustments to Table 1 added in to the manuscript (Ref. 1, R5)
- Line 372 – 374: In response to the referees remark concerning the effect of diffusion limitation on N<sub>2</sub>O reduction, according text was added to the discussion in section 4.1. (Ref. 1, R6)
- Line 431 – 437: Additional paragraph concerning potential bypassing of the soil matrix by N<sub>2</sub>O fluxes via plant-internal aeration channels added in discussion section 4.3. (Ref. 1, R3)

References/Figures/Tables: several minor adjustments on graphs and figure captions according the referee's remarks (Ref. 1, R7-13)

## Adjustments according to our responses to comments by anonymous referee #2

- Line 138 – 139: Additional information on pre-experimental conditions provided (Ref. 2, R1)
- Line 148 – 153: Additional information on experimental conditions and limitations provided (Ref. 2, RI, R1)  
331 – 335
- Line 182 – 185: Adjustment of phrasing according the referee's remarks (Ref. 2, R2)
- Line 192 – 195: Rationale for sterilization of leaf litter added (Ref. 2, R3)
- Line 234 – 237: Statistics section specified (Ref. 2, RIII, R4)  
241 – 245
- Line 380 – 384: Discussion paragraph added about extrapolating our findings to natural conditions (Ref. 2, R5)
- Line 442 – 447: Discussion paragraph added about the generalizability of our findings in the plant treatments (Ref. 2; R6)

# 1 **Alteration of nitrous oxide emissions from floodplain soils by** 2 **aggregate size, litter accumulation and plant–soil interactions**

3 Martin Ley<sup>1,2</sup>, Moritz F. Lehmann<sup>2</sup>, Pascal A. Niklaus<sup>3</sup>, and Jörg Luster<sup>1</sup>

4 <sup>1</sup>Swiss Federal Institute for Forest, Snow and Landscape Research WSL, Zürcherstrasse 111, 8903 Birmensdorf,  
5 Switzerland

6 <sup>2</sup>Department of Environmental Sciences, University of Basel, Bernoullistrasse 30, 4056 Basel, Switzerland

7 <sup>3</sup>Department of Evolutionary Biology and Environmental Studies, University of Zürich, Winterthurerstrasse 190,  
8 8057 Zurich, Switzerland

9 *Correspondence to:* Martin Ley (martin.ley@wsl.ch)

10 **Abstract.** Semi–terrestrial soils such as floodplain soils are considered potential hotspots of nitrous oxide (N<sub>2</sub>O)  
11 emissions. Microhabitats in the soil, such as within and outside of aggregates, in the detritusphere, and/or in the  
12 rhizosphere, are considered to promote and preserve specific redox conditions. Yet, our understanding of the  
13 relative effects of such microhabitats and their interactions on N<sub>2</sub>O production and consumption in soils is still  
14 incomplete. Therefore, we assessed the effect of aggregate size, **buried leaf litter, and plant–soil interactions** on  
15 the occurrence of enhanced N<sub>2</sub>O emissions under simulated flooding/drying conditions in a mesocosm  
16 experiment. We used two model soils with equivalent structure and texture, comprising macroaggregates (400–  
17 250 µm) or microaggregates (< 250 µm) from a N-rich floodplain soil. These model soils **were** planted **either**  
18 with basket willow (*Salix viminalis* L.), mixed with leaf litter, or left unamended. After 48 hours of flooding, a  
19 period of enhanced N<sub>2</sub>O emissions occurred in all treatments. The unamended model soils with macroaggregates  
20 emitted significantly more N<sub>2</sub>O during this period than those with microaggregates. Litter addition modulated the  
21 temporal pattern of the N<sub>2</sub>O emission, leading to short-term peaks of high N<sub>2</sub>O fluxes at the beginning of the  
22 period of enhanced N<sub>2</sub>O emissions. The presence of *S. viminalis* strongly suppressed the N<sub>2</sub>O emission from the  
23 macroaggregated model soil, masking any aggregate-size effect. Integration of the flux data with data on soil  
24 bulk density, moisture, redox potential and soil solution composition suggest that macroaggregates provided  
25 more favorable conditions for spatially coupled nitrification–denitrification, which are particularly conducive to  
26 net N<sub>2</sub>O production, than microaggregates. The local increase in organic carbon in the detritusphere appears to  
27 first stimulate N<sub>2</sub>O emissions, but ultimately, respiration of the surplus organic matter shifts the system towards  
28 redox conditions where N<sub>2</sub>O reduction to N<sub>2</sub> dominates. Similarly, the low emission rates in the planted soils can  
29 be best explained by root exudation of low-molecular weight organic substances supporting complete  
30 denitrification in the anoxic zones, but also by the inhibition of denitrification in the zone, where rhizosphere  
31 aeration takes place. Together, our experiments highlight the importance of microhabitat formation in regulating  
32 oxygen (O<sub>2</sub>) content and the completeness of denitrification in soils during drying after saturation. Moreover,  
33 they will help to better predict the conditions under which hotspots, and moments, of enhanced N<sub>2</sub>O emissions  
34 are most likely to occur in hydrologically dynamic soil systems like floodplain soils.

Kommentar [MaL1]: Ref. 1, R1

35

## 36 1. Introduction

37 Nitrous oxide (N<sub>2</sub>O) is a potent greenhouse gas with a global warming potential over a 100 year time horizon  
38 298 times higher than the one of carbon dioxide (Forster et al., 2007). Given its role as climate-relevant gas and  
39 in the depletion of stratospheric ozone (Ravishankara et al., 2009), the steady increase of its average atmospheric  
40 concentration of 0.75ppb yr<sup>-1</sup> (Hartmann et al., 2013) asks for a quantitative understanding of its sources and the  
41 factors that control its production. On a global scale, vegetated soils are the main natural terrestrial sources of  
42 N<sub>2</sub>O. Agriculture is the main anthropogenic source and the main driver of increasing atmosphere N<sub>2</sub>O  
43 concentrations (Ciais et al., 2013).

44 In soils, several biological nitrogen (N) transformation processes produce N<sub>2</sub>O either as a mandatory  
45 intermediate or as a by-product (Spott et al., 2011). Under oxic conditions, the most important process is obligate  
46 aerobic nitrification that yields N<sub>2</sub>O as by-product when hydroxylamine decomposes (Zhu et al., 2013). Under  
47 low oxygen (O<sub>2</sub>) availability, nitrifier denitrification and heterotrophic denitrification with N<sub>2</sub>O as intermediate  
48 become more relevant (Philippot et al., 2009). At stably anoxic conditions and low concentrations of nitrate  
49 (NO<sub>3</sub><sup>-</sup>), complete denitrification consumes substantial amounts of previously produced N<sub>2</sub>O by further reduction  
50 to N<sub>2</sub> (Baggs, 2008; Vieten et al., 2009). In environments that do not sustain stable anoxia, but undergo sporadic  
51 transitions between oxic and anoxic conditions, the activity of certain N<sub>2</sub>O reductases can be suppressed by  
52 transiently elevated O<sub>2</sub> concentration and thus can lead to the accumulation of N<sub>2</sub>O (Morley et al., 2008).

53 Nitrous oxide emissions from soils depend on the availability of carbon (C) and N substrates that fuel the  
54 involved microbial processes. On the other hand, given its dependency on O<sub>2</sub>, N<sub>2</sub>O production is also governed  
55 by the diffusive supply of O<sub>2</sub> through soils. Similarly, soil N<sub>2</sub>O emissions are modulated by diffusive N<sub>2</sub>O  
56 transport from the site of production to the soil surface (e.g. Böttcher et al., 2011; Heincke and Kaupenjohann,  
57 1999). Substrate availability, gas diffusivity, and the distribution of soil organisms are highly heterogeneous in  
58 soils at a small scale, with micro-niches in particular within soil aggregates, within the detritosphere, and within  
59 the rhizosphere. These can result in “hot spots” with high denitrification activity (Kuzyakov and Blagodatskaya,  
60 2015).

61 Soil aggregate formation is a key process in building soil structure and pore space. Soil aggregates undergo  
62 different stages in their development, depending on the degradability of the main binding agent (Tisdall and  
63 Oades, 1982). Initially, highly persistent primary organo–mineral clusters (20–250 µm) are held together by root  
64 hairs and hyphae, thus forming macroaggregates (> 250 µm). Upon decomposition of these temporary binding  
65 agents and the subsequent disruption of the macroaggregates, microaggregates (< 250 µm) are released (Elliott  
66 and Coleman, 1988; Oades, 1984; Six et al., 2004). These consist of clay-encrusted fragments of organic debris  
67 coated with polysaccharides and proteins. This multi-stage development leads to a complex relationship between  
68 aggregate size, intra-aggregate structure and soil structure (Ball, 2013; Totsche et al., 2017), which influences  
69 soil aeration, substrate distribution and pore water dynamics (Six et al., 2004). Often, micro-site heterogeneity  
70 increases with aggregate size, thus fostering the simultaneous activity of different N<sub>2</sub>O producing microbial  
71 communities with distinct functional traits (Bateman and Baggs, 2005). Aggregate size effects on N<sub>2</sub>O  
72 production and consumption have generally been studied in static batch incubation experiments with a  
73 comparatively small number of isolated aggregates of uniform size, at constant levels of water saturation (Diba  
74 et al., 2011; Drury et al., 2004; Jahangir et al., 2011; Khalil et al., 2005; Sey et al., 2008), and through modelling  
75 approaches (Renault and Stengel, 1994; Stolk et al., 2011). Previous work provided partially inconsistent results,  
76 which led to an ongoing discourse about the interplay of physicochemical properties and different aggregate

77 sizes in controlling N<sub>2</sub>O emission. Such inconsistencies may in parts be attributed to the use of different  
78 aggregate size classes, changes in soil structure by aggregate separation, other methodological constraints (water  
79 saturation, redox potential), and differences in microbial communities. The effects of specific aggregate sizes  
80 within a simulated soil structure, in combination with fluctuating water saturation, on soil N<sub>2</sub>O emissions have,  
81 to our knowledge, not been addressed specifically.

82 Similar to soil aggregates, the detritosphere and the rhizosphere (the zone of the soil that is affected by root  
83 activity) (Baggs, 2011; Luster et al., 2009), can be considered biogeochemical hot spots (Kuzyakov and  
84 Blagodatskaya, 2015; Myrold et al., 2011). Here, carbon availability is much higher than in the bulk soil and  
85 thus rarely limiting microbial process rates. The detritosphere consists of dead organic material, which spans a  
86 wide range of recalcitrance to microbial decomposition. Spatially confined accumulations of variably labile soil  
87 litter form microhabitats that are often colonized by highly active microbial communities (Parkin, 1987).  
88 Aggregation of litter particles has been shown to affect N<sub>2</sub>O emissions (Loecke and Robertson, 2009). Hill  
89 (2011) identified buried organic-rich litter horizons in a stream riparian zone as hot spots of N cycling. Similarly,  
90 in the rhizosphere, root exudates and exfoliated root cells provide ample degradable organic substrate for soil  
91 microbes (Robertson and Groffman, 2015). Yet, plant growth may also affect soil microbial communities  
92 through competition for water and nutrients (e.g., fixed N) (Bender et al., 2014; Myrold et al., 2011). The  
93 combined effects of these plant–soil interactions on N<sub>2</sub>O production have been reviewed by Philippot et al.  
94 (2009). Root-derived bioavailable organic compounds can stimulate heterotrophic microbial activity, specifically  
95 N mineralization and denitrification. Nitrification in turn can be enhanced by the elevated N turnover and  
96 mineralization rates, but may also be negatively affected by specific inhibitors released from the root or through  
97 plant-driven ammonium depletion. The ability of some plants adapted to water-saturated conditions to  
98 „pump“ air into the rhizosphere via aerenchyma (gas conductive channels in the root) leads to an improved  
99 oxygenation of the rhizosphere and a stimulation of nitrification (Philippot et al., 2009). Surrounded by  
100 otherwise anoxic sediments, such aerated micro-environments may create optimal conditions for coupled  
101 nitrification–denitrification (Baldwin and Mitchell, 2000; Koschorreck and Darwich, 1998). On the other hand,  
102 transport of N<sub>2</sub>O produced in the soil to the atmosphere is may be facilitated via these plant-internal channels,  
103 bypassing diffusive transport barriers and enhancing soil–atmosphere gas fluxes (Jørgensen et al., 2012).

104 The dynamics of N<sub>2</sub>O emissions are strongly coupled to the dynamics of pore water. Re-wetting of previously  
105 dried soil can lead to strong N<sub>2</sub>O emissions (Goldberg et al., 2010; Ruser et al., 2006), likely fostered by a  
106 wetting-induced flush in N mineralization (Baldwin and Mitchell, 2000). On the other hand, the drying-phase  
107 after water saturation of sediments and soils can lead to a period of enhanced N<sub>2</sub>O emissions (e.g. Baldwin and  
108 Mitchell, 2000; Groffman and Tiedje, 1988; Rabot et al., 2014; Shrestha et al., 2012) when water-filled pore  
109 space (WFPS) exceeds 60% (Beare et al., 2009; Rabot et al., 2014). The increased N<sub>2</sub>O production has been  
110 attributed to enhanced coupled nitrification–denitrification (Baldwin and Mitchell, 2000). Depending on the  
111 spatial distribution of water films around soil particles and tortuosity (which is a function of aggregate size and  
112 soil structure), the uneven drying of the soil after full saturation may generate conditions that are conducive to  
113 the formation of anaerobic zones in otherwise oxic environments (Young and Ritz, 2000). Pore water thereby  
114 acts as a diffusion barrier for gas exchange, limiting the O<sub>2</sub> availability in the soil pore space (Butterbach-Bahl et  
115 al., 2013). Moreover, pore water serves as a medium for the diffusive dispersal of dissolved C and N substrates,  
116 e.g. from the site of litter decomposition to spatially separated N<sub>2</sub>O producing microbial communities (Hu et al.,  
117 2015). Therefore, fluctuations in water saturation efficiently promote the development of hot spots and hot

118 moments of N<sub>2</sub>O emissions in floodplain soils and other semi-terrestrial soils (Hefting et al., 2004; Shrestha et al.,  
119 2012).

120 The main objective of the present experimental study was to assess both the relative and combined effects of soil  
121 microhabitats associated with soil aggregates, the detritosphere and plant–soil interactions on N<sub>2</sub>O emissions  
122 from floodplain soils under changing pore-space saturation. We simulated a flooding event in mesocosm  
123 experiments with main focus on the dynamics of N<sub>2</sub>O emissions during hot moments in the drying phase after  
124 flooding. To isolate the effect of aggregate-size and to minimize confounding effects of differences in soil  
125 structure, we prepared model soils by mixing aggregate size fractions of a floodplain soil with suitable inert  
126 material. The combined effects of soil aggregate size and plant detritus or plant-soil interactions were addressed  
127 by mixing the model soils with leaf litter or by planting them with willow cuttings (*Salix viminalis* L.).

Kommentar [MaL2]: Ref 1, Reply;  
Ref 2, R IV

128 We demonstrate that the level of soil aggregation significantly affects N<sub>2</sub>O emission rates from floodplain soils  
129 through its modulating control on the model soil's physicochemical properties. We further show that these  
130 effects can be modified by the presence of a detritosphere and by root–soil interactions, changing carbon and N  
131 substrate availability and redox conditions.

## 132 2. Material and methods

### 133 2.1 Model soils

134 In February 2014, material from the uppermost 20 cm of a N-rich gleyic Fluvisol (calcaric, humic siltic) with  
135 20% sand and 18% clay (Samaritani et al., 2011) was collected in the restored Thur River floodplain near  
136 Niederneunforn (NE Switzerland 47°35' N, 8°46' E, 453 m.a.s.l.; MAT 9.1 °C; MAP 1015 mm). After removing  
137 plant residues such as roots, twigs and leaves, the soil was mixed and air-dried to a gravimetric water content of  
138 24.7 ± 0.4 %. In the next step, the original floodplain soil material, consisting of 18.5 ± 4.6 % aggregates smaller  
139 than 250 µm and 81.5 ± 4.6 % macroaggregates (mean ± SD; n = 10), was separated into a macroaggregate  
140 fraction (250–4000 µm) and a microaggregate fraction (< 250 µm) by dry sieving. The threshold of 250 µm  
141 between macroaggregates and microaggregates was chosen based on Tisdall and Oades (1982). Soil aggregate  
142 fractions were then used to re-compose model soils. In order to preserve soil structure, the remaining aggregate  
143 size fractions were complemented with an inert matrix replacing the removed aggregate size fraction of the  
144 original soil. Model Soil 1 (LA) was composed of soil macroaggregates mixed in a 1:1 (w/w) ratio with glass  
145 beads of 150–250 µm size serving as inert matrix material replacing the microaggregates of the original soil.  
146 Similarly, Model Soil 2 (SA) was composed of soil microaggregates mixed at the same ratio with fine quartz  
147 gravel of 2000–3200 µm size. To generate an even mixture of original soil aggregates and the respective inert  
148 matrix a Turbula mixer (Willy A. Bachofen AG, Muttenz, Switzerland) was used. The proportions of the  
149 aggregate size fractions in the model soils were different from the original soil, and 50% microaggregates may  
150 be more than what is found in most natural or agricultural soils (often less than 10 %). Nevertheless, we chose to  
151 use equal amounts of micro- and macroaggregates, in order to be able to separate the effects of aggregate size  
152 from effects of aggregate amount (soil mass). These proportions were still well in the range of common top soils  
153 (e.g. Cantón et al., 2009; Gajić et al., 2010; Six et al., 2000). The physicochemical properties of the two soils  
154 were determined by analysing three random samples of each model soil. Texture of the complete model soils  
155 was determined using the pipette method (Gee and Bauder, 1986) and pH was measured potentiometrically in a  
156 stirred slurry of 10 g soil in 20 ml of 0.01 M CaCl<sub>2</sub>, as recommended in Hendershot et al. (2007). Additionally

Kommentar [MaL3]: Ref 2, R1

Kommentar [MaL4]: Ref 2; RI, R1

Kommentar [MaL5]: Ref 1, R2

157 organic carbon ( $C_{org}$ ) and total nitrogen (TN) were analysed in both aggregate size fractions without the inert  
158 material, using the method described by Walthert et al. (2010). The two model soils displayed very similar  
159 physicochemical properties (Table 1), except for the C:N ratio that was lower in macroaggregates than in  
160 microaggregates. The latter was due to the slightly lower organic carbon content in concert with slightly higher  
161 TN values in the macroaggregates. The high calcium carbonate ( $CaCO_3$ ) content of the source material of our  
162 model soils ( $390 \pm 3 \text{ g } CaCO_3 \text{ kg}^{-1}$ ; Samaritani et al., 2011) buffered the systems at an alkaline pH of  $8.00 \pm 0.02$   
163 for LA and  $7.56 \pm 0.01$  for SA respectively (Table 1), ensuring that the activity of key N-transforming enzymes  
164 was not hampered by too low pH, and that the potential for simultaneous production and consumption of  $N_2O$  in  
165 our experiment was fully intact (Blum et al., 2018; Frame et al., 2017).

## 166 2.2 Mesocosms

167 For the mesocosm experiments, transparent polyvinyl chloride (PVC) cylinders with polymethyl methacrylate  
168 (PMMA) couplings were used. A mesocosm comprised a bottom column section, containing the soil material  
169 and a drainage layer as described below, and the upper headspace section with a detachable headspace chamber  
170 (Fig. 1). Each column section was equipped with two suction cups (Rhizon MOM Soil Moisture Samplers,  
171 Rhizosphere Research Products, Netherlands; pore size  $0.15 \mu\text{m}$ ) for soil solution sampling. The suction cups  
172 were horizontally inserted at 5 cm and 20 cm below soil surface. For redox potential measurements, two custom-  
173 made Pt electrodes (tip with diameter of 1 mm and contact length of 5 mm) were placed horizontally at a  $90^\circ$   
174 angle to the suction cups at the same depths, with the sensor tip being located 5 cm from the column wall. A  
175 Ag/AgCl reference electrode (B 2820, SI Analytics, Germany) was installed as shown in Fig. 1. A volumetric  
176 water content (VWC) sensor (EC-5, Decagon, USA) was installed 15 cm below the soil surface. To avoid  
177 undesired waterlogging, each column section contained a 5 cm thick drainage layer composed of quartz sand  
178 with the grain size decreasing with depth from 1 mm to 5.6 mm (Fig. 1). The upper cylinder section was  
179 equipped with three way valves for gas sampling, and an additional vent for pressure compensation.

## 180 2.3. Experimental setup

181 The mesocosm experiment had a factorial experimental design consisting of two factors (MODEL SOIL and  
182 TREATMENT), with the first factor containing two levels (macroaggregates, microaggregates) and the second  
183 factor containing three levels (unamended, litter added, plant presence). This experimental design resulted in six  
184 treatments, each replicated six times (Table 2). As basic material, each mesocosm contained 8.5 kg of either of  
185 the two model soils. Unamended model soils were used to investigate exclusively the effect of aggregate size,  
186 abbreviated as LAU (large aggregates, unamended) and SAU (small aggregates, unamended), respectively. In  
187 order to specifically assess the effect of enhanced availability of labile C in the detritusphere for the  $N_2O$   
188 producing or consuming soil microbial community, two sets of mesocosms were amended with freshly collected  
189 leaves of Basket Willow (*Salix viminalis* L.). Those leaves were cut into small pieces, autoclaved, and then  
190 added to the model soil components ( $8 \text{ g kg}^{-1}$  model soil) during the mixing procedure to create treatments LAL  
191 (large aggregates, litter) and SAL (small aggregates, litter), respectively. The sterilization step was included to  
192 create equal starting conditions in all litter treatments by reducing any potential effect of, and interaction with,  
193 the phyllosphere microbial community even though a direct involvement of the phyllosphere community in  $N_2O$   
194 production was unlikely according to the literature (Bringel and Couée, 2015). A third set of mesocosms was  
195 planted with cuttings collected from the same *Salix viminalis* creating treatments LAP (large aggregates, plant)

Kommentar [MaL6]: Ref. 2, R2

Kommentar [MaL7]: Ref. 2, R3

196 and SAP (small aggregates, plant), respectively to evaluate the effects of root–soil interactions in the respective  
197 model soils. For each mesocosm one cutting was inserted 10 cm into the soil, protruding from the surface about  
198 3 cm.

199 The addition of leaf litter to the model soils led to an increase of  $C_{org}$  and TN in LAL relative to LAU by 41 %  
200 and 35 %, respectively, and in SAL relative to SAU by 58 % and 44 % respectively. The bulk density of the  
201 unamended model soil SAU ( $1.27 \pm 0.01 \text{ g cm}^{-3}$ ) was slightly higher than the one of LAU ( $1.22 \pm 0.01 \text{ g cm}^{-3}$ ;  
202 adj.  $P$ :  $< 0.0001$ ). Regarding the litter addition treatments, the bulk density of LAL ( $1.13 \pm 0.01 \text{ g cm}^{-3}$ ) was  
203 significantly smaller than the one of LAU (adj.  $P$ :  $< 0.0001$ ), whereas the bulk density of SAL ( $1.27 \pm 0.02 \text{ g cm}^{-3}$ )  
204 did not differ significantly from the one of SAU. The soils in the treatments with plants exhibited a similar  
205 bulk density (LAP:  $1.23 \pm 0.02 \text{ g cm}^{-3}$ ; SAP:  $1.24 \pm 0.01 \text{ g cm}^{-3}$ ) as in the respective unamended treatments.

206 The experiments were conducted inside a climate chamber set to constant temperature ( $20 \pm 1 \text{ }^\circ\text{C}$ ) and relative  
207 air humidity ( $60 \pm 10\%$ ), with a light/dark cycle of 14/10 h (PAR  $116.2 \pm 13.7 \text{ } \mu\text{mol m}^{-2} \text{ s}^{-1}$ ). The experimental  
208 period was divided into four consecutive phases: The conditioning phase (Phase 1) lasted for 15 weeks and  
209 allowed the model soils to equilibrate and the plants to develop a root system. This was followed by the first  
210 experimental phase of nine days (Phase 2), serving as a reference period under steady-state conditions. During  
211 Phases 1 and 2, the soils were continuously irrigated with artificial river water ( $\text{Na}^+$ :  $0.43 \text{ } \mu\text{M}$ ;  $\text{K}^+$ :  $0.06 \text{ } \mu\text{M}$ ;  
212  $\text{Ca}^{2+}$ :  $1.72 \text{ } \mu\text{M}$ ;  $\text{Mg}^{2+}$ :  $0.49 \text{ } \mu\text{M}$ ;  $\text{Cl}^-$ :  $4.04 \text{ } \mu\text{M}$ ;  $\text{NO}_3^-$ :  $0.16 \text{ } \mu\text{M}$ ;  $\text{HCO}_3^-$ :  $0.5 \text{ } \mu\text{M}$ ;  $\text{SO}_4^{2-}$ :  $0.11 \text{ } \mu\text{M}$ ; pH: 7.92) via  
213 suction cups, to maintain a volumetric water content of  $35 \pm 5 \%$ . In Phase 3, the mesocosms were flooded by  
214 pumping artificial river water through the drainage vent at the bottom into the cylinder ( $10 \text{ mL min}^{-1}$ , using a  
215 peristaltic pump; IPC-N-24, Ismatec, Germany) until the water level was 1 cm above the soil surface. After 48 h  
216 of flooding, the water was allowed to drain and the soil to dry for 18 days without further irrigation (Phase 4).

## 217 2.4 Sampling and analyses

218 During the entire experiment, water content and redox potential were automatically logged every 5 minutes  
219 (EM5b, Decagon, USA and CR1000, Campbell scientific, USA, respectively).

220 At selected time points during the experiment, soil-emitted gas and soil solution were sampled. For  $\text{N}_2\text{O}$  flux  
221 measurements, 20, 40 and 60 minutes after closing the mesocosms, headspace gas samples (20 mL) were  
222 collected using a syringe and transferred to pre-evacuated exetainers. The samples were analyzed for their  $\text{N}_2\text{O}$   
223 concentration using a gas chromatograph (Agilent 6890, Santa Clara, USA; Porapak Q column, Ar/ $\text{CH}_4$  carrier  
224 gas, micro-ECD detector). Measured headspace  $\text{N}_2\text{O}$  concentrations were converted to moles using the ideal gas  
225 law and headspace volume. The  $\text{N}_2\text{O}$  efflux rates were calculated as the slope of the linear regression of the  $\text{N}_2\text{O}$   
226 amounts at the three sampling times, relative to the exposed soil surface area (Fig. 1, Shrestha et al., 2012).

227 For soil water sampling, 20 mL of soil solution were collected using the suction cups. Water samples were  
228 analyzed for dissolved organic carbon (DOC) and TN concentrations with an elemental analyzer (Formacs<sup>HT/TN</sup>,  
229 Skalar, The Netherlands). Nitrate and ammonium concentrations were measured by ion chromatography (IC 940,  
230 Metrohm, Switzerland), and nitrite ( $\text{NO}_2^-$ ) concentrations were determined photometrically (DR 3900, Hach  
231 Lange, Germany).

## 232 2.5 Data analyses

233 We were interested in effects on cumulated  $\text{N}_2\text{O}$  emissions during hot moments following flooding. We  
234 therefore analyzed data aggregated over this period rather than the raw full time series data. This procedure also



235 avoided potential issues with small shifts in the timing of emissions that might have been significant but which  
236 were irrelevant for the total fluxes we focused on. The total amount of N<sub>2</sub>O emitted during the period of  
237 enhanced N<sub>2</sub>O fluxes in Phase 4, Q<sub>tot</sub>, was calculated by integrating the N<sub>2</sub>O fluxes between day 11 and 25 of the  
238 experiment as follows:

$$Q_{tot} = \frac{1}{2} \sum_{n=1}^{n_{max}} [\Delta_n \times (q_n + q_{n+1})] \quad (1)$$

239 where  $\Delta_n$  is the time period between the n<sup>th</sup> and the n+1<sup>th</sup> measurement, and  $q_n$  and  $q_{n+1}$  the mean flux on the n<sup>th</sup>  
240 and n+1<sup>th</sup> measurement day, respectively. “n=1” refers to day 11, and  $n_{max}$  to day 25 of Phase 4. The integrated  
241 N<sub>2</sub>O fluxes, as well as the average DOC and N-species concentrations in the soil solution during this period were  
242 analyzed by performing two-way ANOVAs with the fixed terms TREATMENT and MODEL SOIL including their  
243 interaction. In case of significant MODEL SOIL, TREATMENT or MODEL SOIL × TREATMENT effects, their causes  
244 were inspected with the Tukey’s honestly significant difference (HSD) post hoc test. For all data, the residuals of  
245 the ANOVA models were inspected, and the Shapiro–Wilk normality test was applied to ensure that the values  
246 follow a Gaussian distribution. In case that this requirement for ANOVA was not met, the respective data set  
247 was log-transformed. Significance and confidence levels were set at  $\alpha < 0.05$ . The results of the performed  
248 ANOVAs are summarized in Table 3. For the statistical analyses we used GraphPad Prism (GraphPad Software  
249 Inc., 2017) and R (R Core Team, 2018).

Kommentar [MaL8]: Ref 2, R III

Kommentar [MaL9]: Ref 2, R III, R4

## 250 3. Results

### 251 3.1 Soil moisture and redox potential

252 During Phase 1 and 2, saturation levels stabilized at  $53.0 \pm 2.1\%$  WFPS (water filled pore space) in the  
253 treatments with LA soils, and were slightly higher in SA treatments ( $57.8 \pm 2.0\%$ ) (Fig. 2). The flooding of the  
254 mesocosms for 48 h with artificial river water raised the WFPS for all LA soils to  $87.8 \pm 0.1\%$ , significantly  
255 exceeding the increase of WFPS in SA soils ( $80.6 \pm 0.1\%$ ). The water release from the system after the  
256 simulated flood resulted in an immediate drop of the WFPS, except for the LAU treatment (Fig. 2). This was  
257 followed by slow drying for 1 week, and a more marked decrease in WFPS during the second week after the  
258 flood. During the latter period, the plant treatments dried faster than the other treatments. As a result, at the end  
259 of the experiment, WFPS was still above pre-flood values in unamended and litter treatments, while WFPS  
260 levels in the treatments with plants were lower than before the flooding.

261 The time course of the redox potential measured in 5 cm and 20 cm depth exhibited distinct patterns depending  
262 on the respective model soil (Fig. 3). In all treatments, flooding induced a rapid decrease of the redox potential to  
263 values below 250 mV within 36 hours. Upon water release, the redox potential returned rapidly to pre-flood  
264 values at both measurement depths only in SA soils. In the LA treatments (most pronounced in LAL), soils at 20  
265 cm depth underwent a prolonged phase of continued reduced redox condition, returning to the initial redox levels  
266 only towards the end of the experiment.

### 267 3.2 Hydrochemistry of soil solutions

268 Considering individual treatments, DOC concentrations varied only little with time. Yet, the DOC concentrations  
269 were generally much higher in treatments with LA than with SA soils. This main effect of MODEL SOIL was

270 highly significant, as was the interaction with TREATMENTS due to a smaller difference in the litter addition  
271 treatments than in the unamended and plant treatments (Table 3). Nitrate was the most abundant dissolved  
272 reactive N species in the soil solution, with pre-flood concentrations of 1 to 5 mM (Fig. 4d–f). In the unamended  
273 and plant treatments,  $\text{NO}_3^-$  concentrations were markedly higher in SA than in LA soils, whereas they were  
274 similar in both litter addition treatments. Two distinct temporal patterns in the evolution of  $\text{NO}_3^-$  concentration  
275 could be discerned. In the unamended and litter-addition treatments,  $\text{NO}_3^-$  concentrations decreased after the  
276 flooding, consistently reaching a minimum on day 19, in the case of the litter treatments below the detection  
277 limit of 0.2  $\mu\text{M}$ , before increasing again during the latter drying phase (Fig. 4d,e). In contrast, in the treatments  
278 with plants,  $\text{NO}_3^-$  concentrations steadily declined from concentrations of 1–2 mM to around 0.5 mM at the end  
279 of the experiment (Fig. 4f). Nitrite was found at significant concentrations only in LA soils, with highest  
280 concentrations in the LAU treatment right after the flooding (33.6  $\mu\text{M}$ ) and decreasing concentrations throughout  
281 the remainder of the experiment (Fig. 4g–i). In SA soils  $\text{NO}_2^-$  concentration was always  $< 5 \mu\text{M}$ , without much  
282 variation. Similarly, in most treatments except SAL, ammonium ( $\text{NH}_4^+$ ) concentrations were  $< 10 \mu\text{M}$ , and  
283 particularly towards the end of the experiment very close to the detection limit (Fig. 4j, 4l). In the SAL treatment,  
284  $\text{NH}_4^+$  concentrations peaked 5 days after the flood with concentrations of around 70  $\mu\text{M}$  (Fig. 4k). This deviation  
285 from the other temporal patterns prompted a significant interaction effect between MODEL SOIL and TREATMENTS.

### 286 3.3 Nitrous oxide emissions

287 During Phase 2 (i.e., before the flooding),  $\text{N}_2\text{O}$  fluxes were generally low ( $< 1 \mu\text{mol m}^{-2} \text{h}^{-1}$ ; Fig. 2), however,  
288 fluxes in the LAL treatment were significantly higher than in the other treatments (adj.  $P = 0.002$ – $0.039$ ; Fig. 2).  
289 The flooding triggered the onset of a “hot moment”, defined here as period with strongly increased  $\text{N}_2\text{O}$   
290 emissions, which lasted for about one week independent of the treatment (Fig. 2). The maximum efflux was  
291 observed immediately after the flood. The subsequent decline in  $\text{N}_2\text{O}$  emission rates followed different patterns  
292 among the various treatments. Normalizing the  $\text{N}_2\text{O}$  flux to the maximum measured efflux for each replicated  
293 treatment revealed a slower decrease with time for the unamended soils than for the litter and plant treatments  
294 (Fig. S1). The strongest peak emissions were observed in the LAL treatment ( $91.6 \pm 14.0 \mu\text{mol m}^{-2} \text{h}^{-1}$ ; mean  $\pm$   
295 SD). Throughout most of the drying phase, the LAU and LAL treatments exhibited higher  $\text{N}_2\text{O}$  emissions than  
296 the corresponding SAU and SAL experiments. In contrast, there was no such difference in the treatments with  
297 plant cuttings, and peak  $\text{N}_2\text{O}$  emissions were overall lower than in the other treatments. The integrated  $\text{N}_2\text{O}$   
298 fluxes during the hot moment (days 11 to 25 of the experiment) were significantly higher for the LAU and LAL  
299 than for all other treatments (Fig. 5), and the aggregate size effect was also significant within the unamended (adj.  
300  $P = 0.045$ ) and litter-addition treatments (adj.  $P = 0.008$ ). The integrated  $\text{N}_2\text{O}$  emissions in the two plant  
301 treatments did not differ significantly from each other, but were significantly smaller than in the LAU (adj.  $P =$   
302  $0.001$ ), and the LAL (adj.  $P = 0.005$ ) treatments. Overall, the effects of MODEL SOIL and TREATMENTS were  
303 significant, as was the interaction between the two factors due to the different aggregate size effect in the plant  
304 compared to the unamended and litter addition treatments (Table 3).

## 305 4. Discussion

306 In our experiment, we could confirm the occurrence of periods of enhanced  $\text{N}_2\text{O}$  emissions in the drying phase  
307 shortly after flooding, as expected based on previous research (Baldwin and Mitchell, 2000; Groffman and

Kommentar [MaL10]: Ref 1; R4

308 Tiedje, 1988; Rabot et al., 2014; Shrestha et al., 2012). We observed that the six treatments had a substantial  
309 effect on the magnitude and temporal pattern of N<sub>2</sub>O emissions that could only be captured by observations at  
310 relatively high temporal resolution. The fast occurrence of strong N<sub>2</sub>O fluxes over a comparatively short period  
311 in the litter-amended treatment on the one side, and the relatively weak response to the flooding in the plant  
312 treatment on the other, suggests complex interactive mechanisms related to distinct microhabitat effects leading  
313 to characteristic periods of enhanced N<sub>2</sub>O emission. Rabot et al. (2014) explained N<sub>2</sub>O emission peaks during the  
314 desaturation phase with the release of previously produced and entrapped N<sub>2</sub>O. Such a mechanism may partly  
315 contribute to high N<sub>2</sub>O emissions in our experiment initially, but the continuing depletion of NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup>  
316 during the phase of high N<sub>2</sub>O emissions indicates that the flooding and drying has strong effects on N  
317 transformations mediated by microorganisms in the soil (e.g., the balance and overall rates of nitrification,  
318 nitrifier–denitrification, and denitrification). Hence, physical controls alone clearly do not explain the observed  
319 timing and extent of hot moments with regard to N<sub>2</sub>O emission. In the following sections we will discuss how  
320 the effect of flooding on microbial N<sub>2</sub>O production is modulated by differential microhabitat formation (and  
321 hence redox conditions) in the various treatments.

#### 322 4.1 Effect of aggregate size on N<sub>2</sub>O emissions

323 LA model soils exhibited both higher peak and total N<sub>2</sub>O emissions during the hot moment in the drying phase  
324 than SA model soils (Figs. 2 and 5). By contrast, in the presence of a growing willow, there was no detectable  
325 effect of aggregate size on the overall N<sub>2</sub>O emission (further discussion below). The aggregate size effects  
326 observed in the unamended and litter treatments can be explained by factors controlling (i) gas diffusion (e.g.  
327 water film distribution, tortuosity of the intra-aggregate pore space) and (ii) decomposition of encapsulated soil  
328 organic matter (SOM) regulating the extent of N<sub>2</sub>O formation (Neira et al., 2015). In order to isolate the effect of  
329 aggregate size (i.e., to minimize the effect of other factors that are likely to influence gas diffusion), we created  
330 model soils of similar soil structure and texture (see Materials and Methods). We thereby implicitly accepted that  
331 potential interactions of the two size fractions with each other, or with soil structures larger than 4 mm could not  
332 be assessed in this experiment. Although this approach thus represents only an approximation of real-world  
333 conditions it was still an improvement compared to experiments where no attempts were made to conserve soil  
334 structure. Similarly, the bulk soil chemical properties of the two aggregate size fractions, such as C<sub>org</sub> content  
335 and TN, are essentially the same. Differences in the initial C:N ratio and pH, although statistically significant,  
336 can be considered equivalent in the ecological context, e.g., in terms of organic matter degradability. Therefore,  
337 we assume in the following that the differences in N<sub>2</sub>O emissions among the treatments can mainly be attributed  
338 to size-related aggregate properties and their interactions with litter addition or rhizosphere effects.

339 During Phase 3 with near-saturated conditions, no aggregate size effect was observed. High WFPS seem to have  
340 limited the gas diffusion (O<sub>2</sub> and N<sub>2</sub>O) independent of the aggregate size, limiting soil–atmosphere gas exchange  
341 in both model soils equally (Neira et al., 2015; Thorbjørn et al., 2008). As a consequence of inhibited gas  
342 exchange/soil aeration, a sharp drop in the redox potential was observed in all treatments, indicating a rapid  
343 decline in O<sub>2</sub> availability to suboxic/anoxic conditions. Together with an incipient decrease in soil solution NO<sub>3</sub><sup>-</sup>,  
344 this indicates that N<sub>2</sub>O production is primarily driven by denitrification in this phase.

345 The aggregate size effects on the formation of moments of enhanced N<sub>2</sub>O emission became evident during the  
346 subsequent drying period. During the initial drying phase, when a heterogeneous distribution of water films  
347 around soil particles/aggregates develops (Young and Ritz, 2000), the macroaggregates in the LA model soils

Kommentar [MaL11]: Ref 2; RI

Kommentar [MaL12]: Ref 1, R5

348 appear to foster micro-environmental conditions that are more beneficial to N<sub>2</sub>O production. This could be  
349 related to the longer diffusive distances for re-entering O<sub>2</sub> caused by the higher tortuosity of the intra-aggregate  
350 pore space of macroaggregates, as reported by Ebrahimi and Or (2016). This may have helped to maintain, or  
351 even extend, reducing conditions due to microbial activity inside the core of macroaggregates during drying.  
352 Thus, on the one hand, large aggregates favor the emergence of anoxic microhabitats expanding the zones where  
353 denitrification occurs. On the other hand, the overall higher porosity of the LA soils supports a better aeration in  
354 drained parts of the soil (Sey et al., 2008), and aerobic processes (e.g., nitrification) are supported. As a result,  
355 ideal conditions for spatially coupled nitrification–denitrification are created (Baldwin and Mitchell, 2000;  
356 Koschorreck and Darwich, 1998). Indeed, the emergence of heterogeneously distributed, spatially confined  
357 oxygen minimum zones during soil drying may be reflected by the high variability of the redox conditions  
358 observed in replicate mesocosms and, on average, the tendency towards lower redox potentials for a prolonged  
359 period of time in the subsoils of the LA model soils (Fig. 3 d–f). In this context, the relevance of water films for  
360 the emergence of periods of enhanced N<sub>2</sub>O emissions is further highlighted by the fact that elevated flux rates  
361 were only observed as long as the WFPS was above 65%. This is consistent with work by Rabot et al. (2014)  
362 and Balaine et al. (2013), who found similar soil water saturation thresholds for elevated N<sub>2</sub>O emissions from  
363 soils, attributing this phenomenon to suboptimal environmental conditions for both nitrification and  
364 denitrification at lower saturation levels.

365 Given the arguments above, we assume that N<sub>2</sub>O emissions during the drying phase originate to a large degree  
366 from heterotrophic denitrification, and that they are governed mainly by the aggregate-size dependent redox  
367 conditions within the semi-saturated soils. This conclusion stands in good agreement with findings from Drury et  
368 al. (2004), who found higher production of N<sub>2</sub>O due to enhanced denitrification with increasing size of intact  
369 arable soil aggregates in a laboratory incubation study. In contrast, the much lower emissions from the SA  
370 treatments can best be explained by a rapid return to pre-flood, i.e. oxic redox conditions in most of the pore  
371 space, under which N<sub>2</sub>O production driven by denitrification is inhibited. **Enhanced reduction of N<sub>2</sub>O to N<sub>2</sub> in  
372 the SA versus LA treatments seems less likely as an explanation for lowered net N<sub>2</sub>O emission rates, since the  
373 relatively high redox potential represents an impediment to complete denitrification to N<sub>2</sub>.** Furthermore,  
374 according to Manucharova et al. (2001) and Renault and Stengel (1994), aggregates smaller than 200 µm are  
375 simply not large (and reactive) enough (i.e., molecular diffusive distances for oxygen are too short) to develop  
376 suboxic or anoxic conditions in the center, let alone denitrifying zones. Hence, only a relatively small fraction of  
377 the total number of microaggregates in the SA soils would have been large enough (between 200 and 250 µm) to  
378 host denitrification and act as site of anaerobic N<sub>2</sub>O production.

379 **Under natural conditions, frequent hydrological disturbance in floodplains creates a highly dynamic and small-  
380 scaled mosaic of different aggregate size distributions. In this regard, our results, demonstrating the effect  
381 aggregate size has on N<sub>2</sub>O emissions, may help to understand the seemingly erratic spatial and temporal  
382 distribution of enhanced N<sub>2</sub>O emissions from floodplain areas. Moreover they imply that zones with a relatively  
383 high percentage of macroaggregates would be particularly prone to high emissions of N<sub>2</sub>O after a flood event.**

Kommentar [MaL13]: Ref 1, R6

Kommentar [MaL14]: Ref 2, R 5

#### 384 4.2 Litter effect on N<sub>2</sub>O emissions

385 We expected that litter addition would increase N<sub>2</sub>O emissions from model soils with both small and large  
386 aggregates, as was found earlier (e.g. Loecke and Robertson, 2009; Parkin, 1987). The addition of litter to the  
387 model soils changed the temporal dynamics of the N<sub>2</sub>O emission substantially, but its effect on the net integrated

388 N<sub>2</sub>O emission was rather minor (Fig. 5). More precisely, highest peak emission rates of all treatments were  
389 observed in the LAL treatment, but peak emission rates were followed by a faster return to low pre-flood  
390 emission rates in the LAL and the SAL treatments relative to the unamended treatments (Fig 2). This confirms  
391 that surplus organic carbon can, on short-term, boost N<sub>2</sub>O emissions, particularly in the large-aggregate  
392 treatment. The fast mid-term return to low N<sub>2</sub>O emission suggests that N<sub>2</sub>O production by heterotrophic  
393 denitrification either becomes limited by substrates other than carbon, and/or that the carbon added to the soils  
394 affects the redox-biogeochemistry in a way that shifts the balance between N<sub>2</sub>O production and consumption in  
395 favor of consumption. Loecke and Robertson (2009) reported similar temporal N<sub>2</sub>O emission patterns in field  
396 experiments with litter-amended soil, and attributed the observed dynamic of a rapid decline after peak emission  
397 to an increased demand for terminal electron acceptors during denitrification shortly after the carbon addition.  
398 Nitrate/nitrite limitation leads, under stable anoxic conditions, ultimately to the complete reduction of produced  
399 N<sub>2</sub>O to N<sub>2</sub> decreasing net N<sub>2</sub>O emission. Indeed, the rapid decrease in N<sub>2</sub>O emissions after the emission rate peak  
400 in the litter addition treatments was accompanied by the complete depletion of NO<sub>3</sub><sup>-</sup> in the soil solution at low  
401 redox potential, suggesting nitrate limitation. The increased demand for electron acceptors can be attributed to  
402 the increased availability of labile C compounds and nutrients provided by the mineralization of litter, and the  
403 concomitant stimulation of aggregate-associated microbial communities during the flooding (Li et al., 2016). At  
404 the same time, the litter-stimulated soil respiration increases the soil's oxygen demand, maintaining stable low  
405 redox conditions for a longer period of time during the drying phase. Since high activity of N<sub>2</sub>O reductase  
406 requires very low O<sub>2</sub> concentrations (Morley et al., 2008), such conditions may be particularly favorable for  
407 complete denitrification to N<sub>2</sub>, an additional, or alternative, explanation for the low N<sub>2</sub>O emission rates shortly  
408 after the N<sub>2</sub>O emission peak.

#### 409 **4.3 Effects of *Salix viminalis***

410 Planted willow cuttings resulted in relatively low maximum N<sub>2</sub>O emission rates (LAP: 19.75 ± 9.31 μmol m<sup>-2</sup> h<sup>-1</sup>;  
411 SAP: 15.07 ± 12.07 μmol m<sup>-2</sup> h<sup>-1</sup>; mean ± SD), independent of aggregate size. The high values for WFPS  
412 throughout the hot moment, and a low redox potential in the subsoil, imply optimal conditions for denitrification  
413 or nitrifier denitrification, but compared to unamended and litter-addition treatments, only little N<sub>2</sub>O was emitted  
414 (both during peak N<sub>2</sub>O emission rates and with regards to the integrated N<sub>2</sub>O flux). *S. viminalis* suppressed peak  
415 N<sub>2</sub>O emissions, overriding the positive effect of large aggregates on N<sub>2</sub>O emissions observed otherwise. The  
416 specific mechanisms involved are uncertain. Fender et al. (2013) found in laboratory experiments with soil from  
417 a temperate broad-leaved forest planted with ash saplings (*Fraxinus excelsior* L.) N<sub>2</sub>O fluxes and plant effects  
418 very similar to the ones observed in our study. They attributed reduced N<sub>2</sub>O emissions in presence of ash partly  
419 to plant uptake of nutrients that reduced NO<sub>3</sub><sup>-</sup> availability to denitrifiers. Fast-growing plant species like *Salix* are  
420 particularly effective in removing soil inorganic N (Kowalik and Randerson, 1994). Such a causal link between  
421 reduced N<sub>2</sub>O emissions and plant growth is, however, not supported by our data. More precisely, the NO<sub>3</sub><sup>-</sup>  
422 concentrations during the hot moment of N<sub>2</sub>O emissions were always relatively high (> 0.5 mM) and above the  
423 levels observed in the litter treatments.

424 An alternative explanation for the reduced N<sub>2</sub>O emissions in the plant treatments could be rhizosphere aeration  
425 by aerenchyma, a physiological trait of *Salix viminalis* roots, which prevents the formation of anoxia in their  
426 close vicinity (Blom et al., 1990; Randerson et al., 2011), and thus inhibits anaerobic N<sub>2</sub>O production. Indeed,  
427 redox potentials in the topsoil were higher in SAP and LAP compared to the other treatments. By contrast, the

428 redox potential in the saturated subsoil below was even lower than observed for the unamended soils. This  
429 indicates that the aeration effect by aerenchyma is constrained to the upper soil, or is, in the deeper soil portions,  
430 compensated by respiratory rhizosphere processes. On the other hand, aerenchyma can also aid in the gas  
431 exchange between the soil and the atmosphere, leading to an accelerated transport of N<sub>2</sub>O by bypassing the soil  
432 matrix. This phenomenon is well documented for various grasses such as *Oryza* (Baruah et al., 2010), *Triticum*  
433 (Smart and Bloom, 2001) or *Phalaris arundinacea* (Jørgensen et al., 2012). However, we are not aware of any  
434 reports on enhanced N<sub>2</sub>O emissions via aerenchyma by willows (*Salix sp.*), and indeed, our results do not  
435 indicate any increased N<sub>2</sub>O emission via plants. In fact, we observed the lowest ecosystem flux rates and lowest  
436 total integrated N<sub>2</sub>O emissions in the mesocosms with *S. viminalis*.

Kommentar [MaL15]: Ref 1; R3

437 According to Fender et al. (2013), in vegetated soils, microbial respiration is stimulated by deposition of root  
438 exudates, which in concert with root respiration in a highly saturated pore space, leads to severe and ongoing  
439 oxygen depletion. Under such stable anoxic conditions complete denitrification would take place generating N<sub>2</sub>  
440 and not N<sub>2</sub>O as the dominant final product and therefore N<sub>2</sub>O emissions would be low.

441 While oxygen depletion by root-exudation-stimulated microbial respiration likely occurs in the rhizosphere of  
442 any plant, rhizosphere aeration is restricted to plants possessing aerenchyma. However, the latter is a  
443 characteristic of many plants adapted to temporary flooding, and has been described also for *Poaceae*, or for ash.  
444 Furthermore, it is reasonable to expect this trait to be found in other *Salicaceae* like *Populus sp.* and other  
445 species of softwood floodplain forests. In areas with monospecific stands of, for example *Salix sp.*, which are  
446 often found on restored river banks, this N<sub>2</sub>O-emission reducing trait can be a welcome side effect.

Kommentar [MaL16]: Ref 2; R6

## 447 5. Conclusions

448 In this study, we investigated the distinct effects of aggregate size, surplus organic carbon from litter and  
449 vegetation on N<sub>2</sub>O emission from model soils after flooding. Flooding and drying were always associated with  
450 hot moments of N<sub>2</sub>O production, most likely due to heterotrophic denitrification as result of suboxic O<sub>2</sub> levels at  
451 high WFPS. Our results demonstrate that aggregate size is a very important factor in modulating N<sub>2</sub>O emission  
452 from soils under changing pore space water saturation. Aggregates of a diameter > 250 µm appear to foster  
453 suboxic microhabitats that favor denitrification and associated N<sub>2</sub>O emission. This soil aggregate size effect may  
454 be amplified in the presence of excess carbon substrate, as long as heterotrophic denitrification, as the main N<sub>2</sub>O  
455 producing process, is not electron-acceptor limited, and extremely reducing conditions in organic rich soils do  
456 not promote complete denitrification leading to further reduction of N<sub>2</sub>O to N<sub>2</sub>. On the other hand, the higher  
457 porosity of the soils with macroaggregates may aid in the formation of microsites at the surface of aggregates  
458 where nitrification is re-initialized during drying, supporting favorable conditions for spatially coupled  
459 nitrification–denitrification. The mechanisms by which processes in the rhizosphere of *Salix viminalis* effectively  
460 suppress N<sub>2</sub>O emissions, and thus mask any aggregate size effect, remain ambiguous. Distinct physiological  
461 features of *Salix viminalis*, its root metabolism, in combination with microbial respiration can lead to the  
462 simultaneous aeration of some parts of the rhizosphere, and the formation of strongly reducing zones in others.  
463 In both cases, redox conditions seem to be impedimental for extensive net N<sub>2</sub>O production.

464 Our results demonstrate the importance and complexity of the interplay between soil aggregate size, labile  
465 organic C availability, respiratory processes in the rhizosphere, and plant-induced aeration of soils under  
466 changing soil water content. Those interactions emerged as modulators of N<sub>2</sub>O emissions by controlling the O<sub>2</sub>

467 distribution in the soil matrix. Indeed, O<sub>2</sub> appears as the unifying master variable that ultimately sets the  
468 boundary conditions for N<sub>2</sub>O production and/or consumption.

469 The main scope of this work was to expand our knowledge on the controls on net N<sub>2</sub>O emissions from floodplain  
470 soils. The systematic relationships observed in this study are likely to help anticipating where and when hotspots  
471 and hot moments of N<sub>2</sub>O emissions are most likely to occur in hydrologically dynamic soil systems like  
472 floodplain soils. Further understanding of the complex interaction between plants and soil microorganisms, the  
473 detritusphere, and soil aggregation, as well as their influence on N turnover and N<sub>2</sub>O accumulation in soils,  
474 should focus on how the parameters tested affect the actual activity of the nitrifying and denitrifying  
475 communities, with an in-depth investigation into the biogeochemical pathways involved.

476 *Data availability.* Data will be openly available at <https://datadryad.org/>

477 *Competing interests.* The authors declare that they have no conflict of interest.

478 *Authors contributions.* The initial concept of the experiment was developed by JL, MFL and PAN. ML planned  
479 the experiment in detail, set it up and performed it. PAN supervised the measurement of N<sub>2</sub>O gas concentrations,  
480 whereas ML conducted all other measurements and data analyses. ML wrote the manuscript with major  
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678

679 **Table 1: Physicochemical properties of the two aggregate size fractions (macroaggregates and microaggregates) and**  
 680 **added leaf litter. C<sub>org</sub> and TN of the aggregates were measured in triplicates. The leaf litter was analyzed in**  
 681 **quadruplicates. Final pH and texture of model soil 1 and 2 were measured in duplicates (means ± SD). Significant**  
 682 **differences in the t-tests (P < 0.05) are highlighted in bold.**

		Macroaggregates	Microaggregates	Macroaggregates vs. Microaggregates	Litter ( <i>Salix v. L.</i> )
C <sub>org</sub>	g kg <sup>-1</sup>	19.22 ± 0.55	21.56 ± 2.39	P = 0.229	459.9 ± 2.55
Total N	g kg <sup>-1</sup>	1.58 ± 0.02	1.35 ± 0.14	P = 0.106	27.39 ± 0.15
C:N ratio		12.16 ± 0.22	15.99 ± 0.71	<b>P = 0.007</b>	16.79 ± 0.06
		Model soil 1	Model soil 2	Model soil 1 vs. Model soil 2	
pH (CaCl <sub>2</sub> )		8 ± 0.02	7.56 ± 0.01	<b>P = 0.009</b>	
sand	%	71.25 ± 0.05	70.7 ± 0.50	P = 0.469	
silt	%	20 ± 0.30	21.1 ± 0.60	P = 0.285	
clay	%	8.75 ± 0.25	8.2 ± 0.10	P = 0.240	

683

684 **Table 2: Overview of treatments in the flooding–drying experiment. Model Soil 1, containing soil macroaggregates is**  
 685 **abbreviated LA, whereas Model Soil 2 contains soil microaggregates and is abbreviated SA. The last character of each**  
 686 **abbreviation stands for unamended (U), litter addition (L) and plant presence (P). Each treatment was replicated six**  
 687 **times.**

	LAU	SAU	LAL	SAL	LAP	SAP
Model Soil 1 (LA)	+	-	+	-	+	-
Model Soil 2 (SA)	-	+	-	+	-	+
Leaf litter ( <i>Salix v.</i> )	-	-	+	+	-	-
<i>Salix v.</i>	-	-	-	-	+	+

688

689 **Table 3: Results of the two-way analysis of variance (ANOVA) of the integrated fluxes (Q<sub>tot</sub>) and the mean**  
 690 **concentrations of chemical properties in soil solution (n=6) during the period of enhanced N<sub>2</sub>O emissions (from day 11**  
 691 **to day 25). Shown are P values with significant differences (P < 0.05) highlighted in bold characters.**

	Q <sub>tot</sub>	DOC	NO <sub>3</sub> <sup>-</sup>	NO <sub>2</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>
TREATMENT	<b>0.0003</b>	<b>0.0133</b>	0.0988	<b>&lt; 0.0001</b>	<b>0.0007</b>
MODEL SOIL	<b>0.0002</b>	<b>&lt; 0.0001</b>	0.2181	<b>&lt; 0.0001</b>	<b>0.0004</b>
TREATMENT × MODEL SOIL	<b>0.0145</b>	<b>&lt; 0.0001</b>	0.0668	0.1174	<b>&lt; 0.0001</b>

692

693

694 **Figure Captions**

695 **Figure 1:** Schematic of a mesocosm with gas sampling valves (1), Ag/AgCl reference electrode (2), Pt redox electrodes  
696 (3), suction cups (4), volumetric water content sensors (5), vent (6), and water inlet/outlet (7). The top part is only  
697 attached during gas sampling.

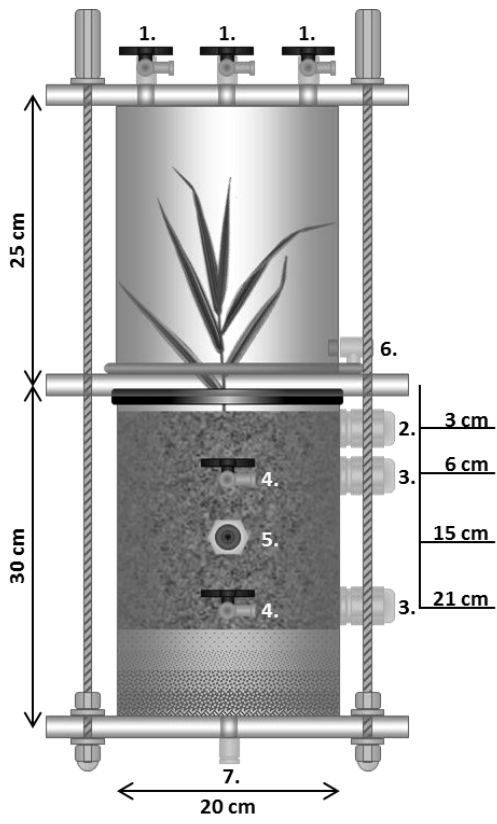
698 **Figure 2:** Mean N<sub>2</sub>O emission during the flooding–drying experiment from large-aggregate model soil (LA; filled  
699 circles) and small-aggregate model soil (SA, open circles). The corresponding water-filled pore space (WFPS) in LA  
700 (filled triangles) and SA (open triangles) are depicted on the right Y-axis. Unamended soils (A), litter addition (B) and  
701 plant treatment (C). Flooding phase indicated by the grey area. Symbols indicate means; error bars are SE; n= 6.

702 **Figure 3:** Redox potential relative to standard hydrogen electrode during the flooding–drying experiment in 5 cm and  
703 20 cm depth (mean ± SE; n=6). Unamended soils (a and d, respectively), litter addition (b and e, respectively), plant  
704 treatment (c and f, respectively). LA (filled circles) and SA (open circles); the dotted line at 250 mV marks the  
705 threshold, below which denitrification is expected to occur.

706 **Figure 4:** DOC (circles), nitrate (squares), nitrite (diamonds) and ammonium (triangles) concentrations in pore water  
707 during the flooding–drying experiment. LA (filled symbols) and SA (empty symbols). Unamended soils (a, d, g and j,  
708 respectively), litter addition (b, e, h and k, respectively) and plant treatment (c, f, j and l, respectively).; (mean ± SE;  
709 n=6).

710 **Figure 5:** Integrated N<sub>2</sub>O fluxes over the 14 days period of elevated N<sub>2</sub>O emissions in the drying phase of the flooding–  
711 drying experiment (mean ± SE; n= 6). Black bars represent Model Soil 1 (macroaggregates 250-4000µm) whereas  
712 Model Soil 2 (microaggregates < 250µm) is depicted as white bars. Significant differences among the six treatments  
713 are denoted by different lower case letters at adj. P < 0.05.

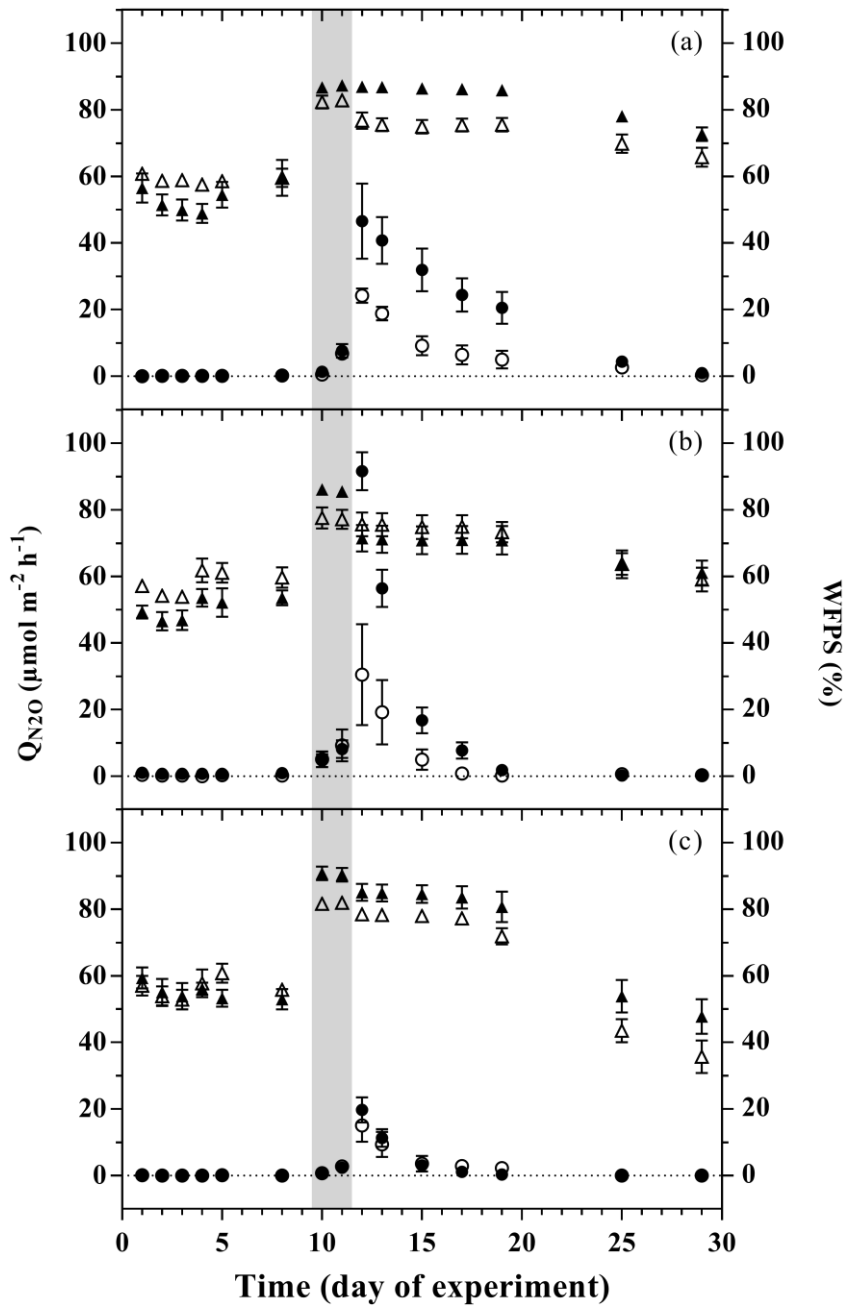
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716 Figure 1

717

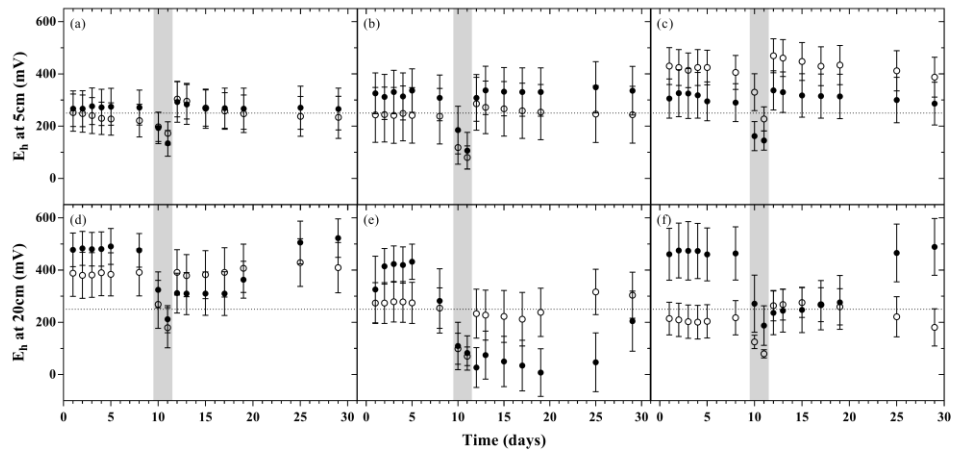


718

719 **Figure 2**

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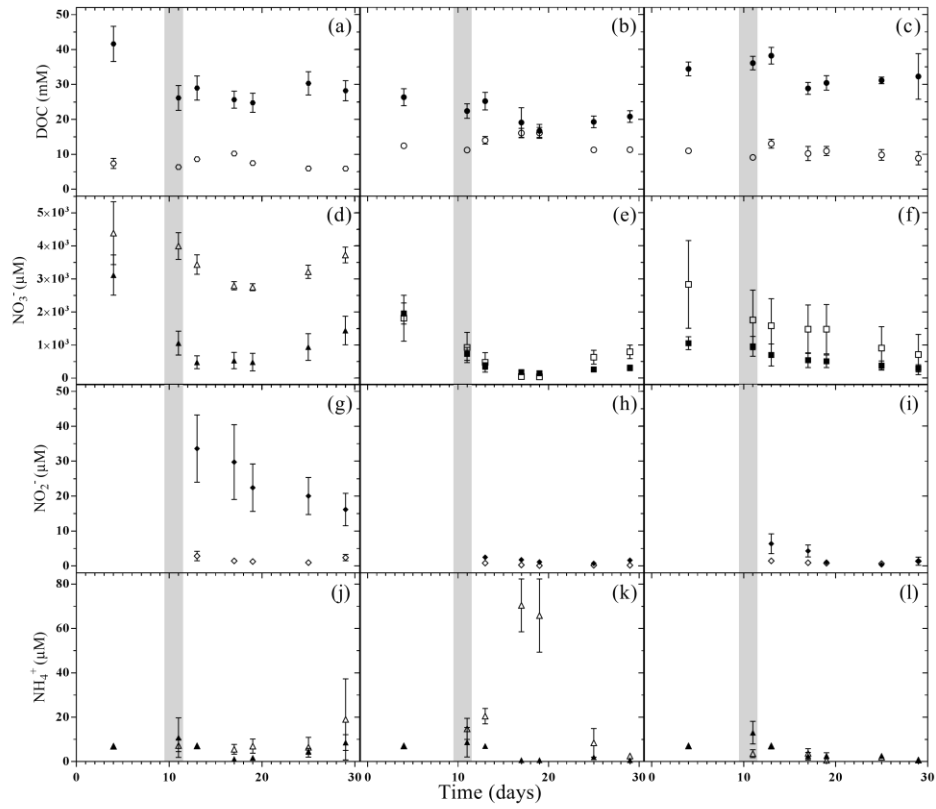




721

722 **Figure 3**

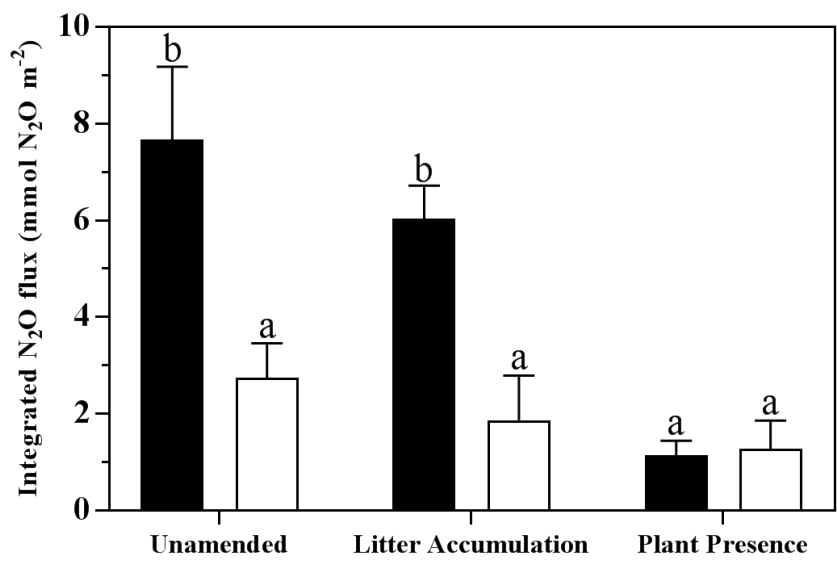
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725 **Figure 4**

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727

728 Figure 5