Referee #1 comments: The manuscript by Benard et al. describes the results from a mesocosm experiment that was designed to investigate the responses of a natural phytoplankton community to warming and acidification. The authors observed a clear stimulation of phytoplankton growth by temperature whereas acidification had no or only a minor effect. Although many experimental studies have been conducted in recent years to investigate phytoplankton responses to warming or to acidification little is known about combined effects. The data provided by this study are thus potentially valuable and interesting. However, important information is lacking in the current version and need to be included and discussed to improve the value of this manuscript. The set-up of the experiment was designed to keep pH in the acidified mesocosms constant, yielding a decrease of pH after the bloom. This differs to the natural environment where a phytoplankton bloom can substantially modify (increase) pH. It also differs to the earlier mesocosms experiments that the authors reference in their discussion. I suggest that the authors discuss implications of the differences in the set-up of experiments.

Author's response to general comments: We thank the reviewer for the thorough evaluation of the manuscript and the insightful comments. The comment on our experimental approach is very appropriate and, as suggested, a new paragraph was added where its implications are discussed (see below). The experimental protocol where pH and pCO₂ are kept constant during the full experiment is indeed different to what would happen in nature where changes in photosynthesis and respiration during the bloom development would affect theses parameters. The main reason why these conditions were kept constant was to allow us to precisely measure and maximise inference of the effects of pH and pCO₂ on different processes (e.g. phytoplankton photosynthesis in this paper, and dimethylsulfide production in a companion paper to be submitted) taking place during all phases of the bloom.

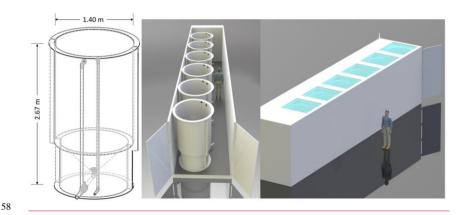
The following section has been added in the new version of the manuscript:

4.5: Implications and limitations

1 2

During our study, we chose to keep the pH constant during the whole experiment instead of allowing it to vary with changes in photosynthesis and respiration during the bloom phases. This approach differs from previous mesocosm experiments where generally no subsequent CO₂ manipulations are conducted after the initial targets are attained (Schulz et al. 2017 and therein). Keeping the pH and pCO₂ conditions stable during our study allowed us to precisely quantify the effect of the changing pH/pCO₂ on the processes taking place during the different phases of the bloom. Such control was not exercised in two of our mesocosms (i.e. the drifters). In these two mesocosms, the pH increased from 7.9 to 8.3 at 10°C, and from 7.9 to 8.7 at 15°C. Since the buffer capacity of acidified waters diminishes with increasing CO₂, the drift in pCO₂ and pH due to biological activity would have been even greater in the more acidified treatments (Delille et al., 2005; Riebesell et al., 2007). Hence, allowing the pH to drift in all mesocosms would have likely ended in an overlapping of the treatments where acidification effects would have been harder to detect. Thus, our experiment could be considered as an intermediate between strictly controlled small scale laboratory experiments and large scale pelagic mesocosm experiments in which only the initial conditions are set. By limiting pCO₂ decrease under high CO₂ drawdown due to photosynthesis during the bloom phase, we minimise confounding effects of

35	$\underline{pCO_2}\ potentially\ overlapping\ in\ association\ with\ high\ biological\ activity\ in\ the\ mesocosms.\ Hence,\ the\ experimental\ conditions$
36	could be considered as extreme examples of acidification conditions, due to the extent of pCO2 values studied. However, the
37	absence of OA effects on most biological parameters measured during our study, even under these extreme conditions,
38	strengthens the argument that the phytoplankton community in LSLE is resistant to OA.
39	
40	Referee comment: Nutrient concentration and irradiance are main factors controlling phytoplankton growth in seawater. The
41	authors should asses how these factors may have affected cell growth and primary production. This includes: 1. Add a drawing
42	of the set-up and placement of mesocosms and treatments within the container. Containers often bear the risk of self-shading,
43	which would need to be considered.
44	
45	Author's response: We thank the reviewer for pointing that out. Light is indeed an important driver of phytoplankton
46	photosynthesis and growth. As requested, the following information has been added in the text as well as a new figure:
47	
48	Modification (line 88)
49	Old sentence: Each enclosure is sealed with a Plexiglas cover allowing the transmission of 90 % of photosynthetically active
50	radiation (PAR; 400-700 nm), 50-85 % of solar UVB (280-315 nm) and 85-90 % of UVA (315-400 nm).
51	
52	New sentence: The mesocosms exhibit opaque walls and all lie on the same plane level as not to shade each other. Light
53	penetrates the mesocosms only through a sealed Plexiglas circular cover at their uppermost part. The cover allows the
54	transmission of 90 % of photosynthetically active radiation (PAR; 400-700 nm), 85-90 % of UVA (315-400 nm), and 50-
55	85 % of solar UVB (280–315 nm).
56	
57	New figure (A):
•	



In terms of the impact of nutrient concentrations on cell growth and primary production, we discuss this further: please refer to the specific comment about line 211 below.

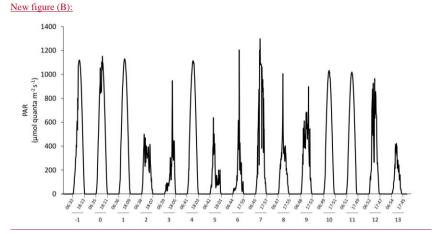
Referee comment: Please give absolute values for irradiance instead of % shading. What was the light:dark cycle? Since primary production measurements were carried out over 24h, some incubation hours will have been in the dark if natural sunlight was applied. It is important to inform about the potential role of dark respiration during the incubations. Since light intensity has been shown to co-affect phytoplankton responses to acidification and warming, it is absolutely necessary to show and discuss the light climate in mesocosms and incubations.

Author's response:

We acknowledge that light intensity has the potential to affect phytoplankton response to acidification and warming. Light intensity varied between days over the duration of the experiment (see new figure below). We cannot exclude the possible interaction of light intensity with acidification or warming, or the effect the varying intensity has on the dynamics of primary production. However, as all mesocosms were subjected to the same fluctuating natural irradiance, and that pCO₂ and temperature were the only factors changing between the treatments, we limited our interpretation to these two parameters.

New sentence (line 112): Incident light was variable during our experiment, with only few sunny days (Fig. B).





Referee comment: It is important to consider that net primary production was measured. Hence, responses to warming and acidification may not only be related to photosynthetic production but also to respiration processes. Please discuss.

Author's response:

We agree with the reviewer. The text was modified accordingly (Line 394).

Old sentence: The warming-induced decrease in carbon fixation measured during Phase II may thus result from an increase in respiration by the nitrogen-limited diatoms.

New sentence: The warming-induced increase in fixed carbon being release in the dissolved fraction likely stems from increased exudation by phytoplankton, or sloppy feeding / excretion following ingestion by grazers (Kim et al., 2011). The increase in fixed carbon released as dissolved organic carbon (DOC) measured during Phase II may also result from greater respiration by the nitrogen-limited diatoms during periods of darkness of the incubations, as dark phytoplankton respiration rates generally increase with temperature (Butrón et al., 2009; Robarts and Zohary, 1987). Moreover, the enclosures do not permit the sinking and export of particulates organic carbon (POC), allowing a further transformation into DOC by heterotrophic bacteria, a process that could be exacerbated under warming (Wohlers et al., 2009).

Referee comment: Given that the authors did not add nutrients to the natural seawater, the strong increase in biomass (from 10 to up to 30 µg/L Chl a in one day) after incubation is very surprising. What could have limited phytoplankton growth in situ? Please discuss. Author's response: The following phrases were added to address this phenomenon (line 302). New sentence: In situ nutrient conditions prior to the water collection were favourable for a bloom development. Based on previous studies, in situ phytoplankton growth was probably limited by light due to water turbidity and vertical mixing at the time of water collection (Levasseur et al. 1984). Grazing may also have played a role in keeping the in situ biomass of flagellates low prior to our sampling. However, a natural diatom fall bloom was observed in the days following the water collection in the adjacent region (Ferreyra, pers. comm.). The increased stability within the mesocosms, combined with the reduction of the grazing pressure (filtration on 250 µm) likely contributed to the fast accumulation of phytoplankton biomass. Referee comment: There was a strong drop in pH prior to the acidification treatment on day -3. What may have been the reason for this drop? Author's response (AR): The following modifications have been made to address the pH drop / pCO₂ rise at the onset of the experiment. Modification (line 200) Old sentence: The pH remained relatively stable throughout the experiment in the pH-controlled treatments, but decreased slightly during Phase II by an average of -0.14 ± 0.07 units relative to the target pH_T (Fig. 1a). New sentence: Following the filling of the mesocosms, the pH_T in all mesocosms decreased from an average of 7.84 to 7.53. Throughout the rest of the experiment after treatments were applied, the pH remained relatively stable in the pH-controlled treatments, but decreased slightly during Phase II by an average of -0.14 ± 0.07 units relative to the target pH_T (Fig. 1a). Addition (line 294) New sentence: The onset of the experiment was marked by an increase of pCO2 on the day following the filling of the mesocosms. This phenomenon often takes place at the beginning of such experiments when pumping tends to break phytoplankton cells and larger debris into smaller ones. We attribute the rapid fluctuations in pCO2 to the release of organic matter following the filling of the mesocosms with a stimulating effect on heterotrophic respiration, and hence CO2 production.

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Specific comments:

136	
137	Line 87: add diameter and height of mesocosms
138	AR: Line 87 has been modified as follow: (note that the dimensions of the mesocosms are now also presented in the new
139	figure A):
140	
141	Old sentence: The mesocosms are cylindrical with a cone-shaped bottom within which mixing is achieved using a propelled
142	fixed near the top.
143	
144	New sentence: The mesocosms are cylindrical (2.67 m \times 1.40 m) with a cone-shaped bottom within which mixing is achieved
145	using a propeller fixed near the top.
146	
147	Line 99ff: add total duration of the experiment to the description
148	AR: Line 100 has been modified as follow:
149	
150	Old sentence:
151	The water was collected at 5 m depth near Rimouski harbour (48° 28' 39.9" N, 68° 31' 03.0" W) on the 27th of September 2014
152	In situ conditions were: salinity = 26.52, temperature = $10 ^{\circ}\text{C}$, nitrate $(NO_3) = 12.8 \pm 0.6 \mu\text{mol L}^{-1}$, silicic acid
153	$(Si(OH)_4) = 16 \pm 2 \mu mol L^{-1}$, and soluble reactive phosphate (SRP) = $1.4 \pm 0.3 \mu mol L^{-1}$. The same day (indicated as day -5)
154	$\underline{\text{hereafter}}\text{, the water was filtered through a 250}\ \mum mesh while simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the simultaneously filling the 12 mesocosm tanks by gravity with the 12 mesocosm tanks by$
155	a custom made 'octopus' tubing system.
156	
157	New sentence:
158	The water was collected at 5 m depth near Rimouski harbour (48° 28' 39.9" N, 68° 31' 03.0" W) on the 27th of September 2014
159	(indicated as day -5 hereafter), and the experiment lasted until the 15th of October 2014 (day 13). In situ conditions were
160	$\underline{salinity} = 26.52, \ temperature = 10 \ ^{\circ}C, \ nitrate \ (NO_{3}^{-}) = 12.8 \pm 0.6 \ \mu mol \ L^{-1}, \ silicic \ acid \ (Si(OH)_{4}) = 16 \pm 2 \ \mu mol \ L^{-1}, \ and \ acid \ (Si(OH)_{4}) = 16 \pm 2 \ \mu mol \ L^{-1}, \ acid \ (Si$
161	soluble reactive phosphate (SRP) = $1.4 \pm 0.3 \mu\text{mol}L^{-1}$. On day -5, the water was filtered through a 250 μ m mesh while
162	simultaneously filling the 12 mesocosm tanks by gravity with a custom made 'octopus' tubing system.
163	
164	Line 104: give value for initial pCO2
165	AR: Line 104 has been modified as follow:
166	Old sentence: The initial in situ temperature of 10 °C was maintained in the twelve mesocosms for the first 24 h (day -4).
167	
168	New sentence: The initial pCO ₂ was 623 ± 7 µatm and the in situ temperature of 10 °C was maintained in the twelve
169	mesocosms for the first 24 h (day -4).

170 171 Line 113ff: Add total amount of volume sampled from the mesocosms each day 172 AR: The following was added to line 116: "Total amount of volume sampled every day was 24 L or less." 173 174 Line 166: Was the no replicate incubation? Was the error within treatment assessed? 175 AR: There was no replication of incubations. The number of bottles to handle was already quite extensive and the maximum 176 capacity of our incubators had been reached. We chose to adopt the strategy of an increased number of treatments 177 (mesocosms), which, even with reduced replication, allows greater power to characterise the functional relationships between 178 OA parameters and organism or ecosystem response (Riebesell et al. 2011 (Guide to Best Practices in Ocean Acidification)). 179 However, we conducted independent measures of particulate P_P, dissolved P_P, as well as total P_P every day in all the mesocosms 180 allowing us to verify that the fractions measured in particulate and dissolved P_p reliably added up to the total P_p. 181 182 Line 171: Give irradiance values 183 AR: The daily irradiances are now presented in Figure B. 184 185 **Line 202:** pCO2 was $1340 \pm 150 \mu$ atm on day -3; why was the value so high? 186 AR: The pCO2 doubled after the filling of the mesocosms most probably due to an increase in CO2 production following the 187 release of organic matter and the increase in heterotrophic respiration. We conclude that the filling of the mesocosms tends to 188 break phytoplankton cells and larger debris into smaller ones, with a stimulating effect on bacteria. Refer to additions on lines 189 200 and 294. 190 191 Line 211: The three nutrients displayed a similar temporal depletion pattern following the development of the phytoplanktonic 192 bloom.' I disagree the nutrients in the warm treatments were clearly reduced much faster. 193 AR: Right. We meant that the general pattern was similar between the three nutrients (nitrate, soluble reactive phosphate and 194 silicic acid) within each of the mesocosms, however we agree that clarity could be added here. We rephrased this part of the results section: 195 196 197 Old sentence: The three nutrients displayed a similar temporal depletion pattern following the development of the 198 phytoplanktonic bloom. 199 200 New sentences: Within individual mesocosms, concentrations of nitrate, silicic acid and soluble reactive phosphate displayed 201 similar temporal patterns following the development of the phytoplankton bloom. Overall, nutrient depletion was reached 202 within 5 days in all mesocosms at 10°C, exception made of the drifter which became nutrient-deplete by day 3. Nutrient depletion was reached slightly earlier within the 15°C mesocosms, all of them displaying exhaustion within 3 days of the 203

Referee comments #2: The manuscript of Bénard and collaborators reports on an experiment that has been conducted using indoor mesocosms (2.6 m3) to test for the effect of ocean acidification and warming on the development of a fall phytoplankton bloom in the Lower St. Lawrence Estuary. The experiment setup comprised 2 sets of 6 mesocosms installed in two temperature-controlled containers, that were filled with seawater sieved onto 250 microns. In one container, the water temperature was raised by 5°C compared to the mesocosms installed in the other container (10 vs. 15°C). A gradient approach (no replicates) has been considered for pCO2/pH covering a range of pH from 7.2 to 8.6. The experiment lasted 13 days and covered the development of a bloom and its decline. Major conclusions of this study are that pCO2 has no effect on all measured parameters and processes while increasing temperature led to a faster build-up of chlorophyll and higher particulate primary production rates. Overall, this is a very well written manuscript that deals with an important topic. The introduction is well documented and shows that while this topic is of great importance, a fair amount of studies has already been conducted, including studies using in situ mesocosms in various environments. Although I would like to ultimately recommend this manuscript for publication in BG, I am concerned by 3 major aspects of this work and would like the authors to answer these comments.

Author's response: We would like to thank the reviewer for the general evaluation of the manuscript and the insightful comments. We will further discuss the following comments of the reviewer.

Referee comment: Realism. The authors clearly mention that the surface mixed-layer pCO2 is strongly modulated by biological productivity, yet they decided to run an experiment during which a bloom is produced and where carbonate chemistry has been maintained as constant. This would be acceptable if well explained and discussed, but the problem is that "control" mesocosms were actually not controlled (consider changing their name. . .) and pH was left increasing while the bloom was forming to (what I consider to be) very high and potentially unrealistic pH (?) values of 8.6. In situ pH was apparently close to 7.8, these "control" mesocosms appear to me as "perturbed"! Besides this major concern, I have to admit I do not understand how carbonate chemistry was controlled. The authors mention that "acidification" was carried out over day -1. On that day, I actually also observe a sudden increase of pH for the "controls", pH8 and pH7.8. . . How did that happen? Naturally? Why was the increase in pH much higher in the controls than for the other mesocosms. Obviously, some information is missing here. Do you know the reason why pH decreased so fast between day -4 and day -3?

Author's response: First, following this comment, the "Controls" have been more appropriately renamed "Drifters" to clearly show that the pCO₂ was not controlled. We noted that Reviewer #1 also pointed out the shortcomings in the discussion of the different approaches to control pH during this type of experiment.

The following section has been added in the new version of the manuscript:

4.5: Implications and limitations

During our study, we chose to keep the pH constant during the whole experiment instead of allowing it to vary with changes in photosynthesis and respiration during the bloom phases. This approach differs from previous mesocosm experiments where generally no subsequent CO₂ manipulations are conducted after the initial targets are attained (Schulz et al. 2017 and therein). Keeping the pH and pCO₂ conditions stable during our study allowed us to precisely quantify the effect of the changing pH/pCO₂ on the processes taking place during the different phases of the bloom. Such control was not exercised in two of our mesocosms (i.e. the drifters). In these two mesocosms, the pH increased from 7.9 to 8.3 at 10°C, and from 7.9 to 8.7 at 15°C. Since the buffer capacity of acidified waters diminishes with increasing CO₂, the drift in pCO₂ and pH due to biological activity would have been even greater in the more acidified treatments (Delille et al., 2005; Riebesell et al., 2007). Hence, allowing the pH to drift in all mesocosms would have likely ended in an overlapping of the treatments where acidification effects would have been harder to detect. Thus, our experiment could be considered as an intermediate between strictly controlled small scale laboratory experiments and large scale pelagic mesocosm experiments in which only the initial conditions are set. By limiting pCO₂ decrease under high CO₂ drawdown due to photosynthesis during the bloom phase, we minimise confounding effects of pCO₂ potentially overlapping in association with high biological activity in the mesocosms. Hence, the experimental conditions could be considered as extreme examples of acidification conditions, due to the extent of pCO2 values studied. However, the absence of OA effects on the biological parameters measured during our study, even under these extreme conditions, strengthens the argument that the phytoplankton community in LSLE is resistant to OA.

 $\underline{\text{To further clarify how the acidification and pH treatments were applied, the following has been added.}$

Addition (line 112):

To attain initial targeted pH, CO₂-satured artificial seawater was added to the mesocosms that needed a pH lowering while mesocosms M2 (8.0), M4 (7.8), M6 (Drifter), M9 (8.0), M11 (Drifter) and M12 (7.8) were openly mixed to allow the degassing of the supersaturated CO₂. Once the mesocosms had reached their target pH, the automatic system controlled the sporadic addition of CO₂-saturated water to refrain the pH from rising. Only the "Drifters" were not controlled throughout the experiment.

Referee comment: Timing. The second concern I have is related to the division of the experiment in 2 phases. Phase 1 corresponds to the development of the diatom bloom extended up to the depletion of nitrate (day 0 to 4) and Phase 2 corresponds to the declining phase of the bloom in the absence of detectable nitrate. Except that this is not really true since temperature increased the speed at which chla built-up and nutrients were consumed (this is not really mentioned in 3.2). At 15°C, except for 1 mesocosm, nitrate was exhausted already on day 2 while at 10°C, NO3 in most mesocosms were actually exhausted on day 4. My point is that since T modified the timing of the bloom (and its decline), it does not seem correct to me to consider fixed periods. The build-up of chla and all related statistical analyses should be conducted at 15°C between day 0 and 2, and all tests related to the decline of the bloom between day 3 and 13. Would that change some of your results?

Author's response:

We agree with the reviewer on the need for a better representation of the different phases in relation to the treatments but would previously like to inform on our decision to initially choose a fixed period. Firstly, it is imperative to divide the experiment in two phases not to confound effects caused by different processes and conditions during the development or the decline of the bloom. We previously considered multiple division method for the experiment (day of nitrate exhaustion, maximum Chl a concentration, averaged day of nitrate exhaustion) and ultimately opted for the averaged day of nitrate exhaustion as to mark the end of the nutrient-rich development phase. By choosing a single day as the divider for the phases, we could accurately compare our results with numerous mesocosms experiment that also divide their experiment using fixed periods. However, in many of those cases, the distinction between the phases were sharply defined as timing was not an important factor.

Thus, we agree with the reviewer suggestion but would like to take it one step further. Assigning phases duration based on temperature treatments as the reviewer suggests (Phase I: days 0-4 at 10°C, and days 0-2 at 15°C) would exclude some data from mesocosms that are still in the growth phase from the analyses of that phase, as M3 and M5 maximum Chl a concentrations are attained on day 7, and M7 maximum Chl a concentration is on day 4. Therefore, we suggest modifying the phases for each mesocosm as follow: Phase I (day 0 to day of maximum Chl a concentration) and Phase II (day after maximum Chl a concentration to day 13). By doing so, all the analyses on the Phase I will be constrained to the Chl a accumulation phase for each mesocosm, while Phase II will be an accurate representation of the individual declining phase. This modification carries some changes in the significance of the results of the analyses, although it does not change the overall narrative or conclusion of the manuscript. Namely, the absence of acidification effects is still valid for all parameters measured, except for picocyanobacteria abundance at 15°C during Phase II which is shows a negative linear trend with increasing pCO₂. We already suggested that increases in grazing pressures could counteract the stimulating effect of increased CO2 availability on picocyanobacteria, and this is still valid. With regards to the temperature effects, the differences on the mean concentrations of Chl a is not significant in either phases. Although, the accumulation rate of Chl a is still higher at 15°C, reflecting the faster accumulation of Chl a. The temperature effects on particulate primary production during the both phases have also been discarded, yet our conclusion remains that the P_P is not affected over the full duration of the experiment. This will strengthen the conclusion that only the timing of the bloom development is affected by temperature, with negligible effects on the other parameters.

Moreover, we had already processed data in this manner and final figures can easily be produced to reflect the changes in the statistical analyses.

Referee comment: Grazing. I regret that potentially the most exciting result of this experiment suggesting that pCO2 "positive" effects on phytoplankton were actually masked by significant increases in micro-grazing is not more developed. I understand the politics behind the publication of papers from a joint experiment, it would just bring much more value to your

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333	Author's response:
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335	We agree with this comment. The impact of the different treatments on zooplankton abundance will be discussed in a
336	companion paper by colleagues.
337	
338	Minor comments
339	
340	<u>L217: concentrations were</u>
341	AR: "Concentrations where" changed to "concentrations were"
342	
343	L225: "suggesting a faster loss of pigments". Not really convinced by that Is the slope different?
344	AR: Following the changes made with regards to the statistical analyses, this section was adjusted.
345	
346	Old section (line 223-226):
347	During Phase II, we observed no significant effect of increasing pCO ₂ on the mean Chl a concentrations at the two temperatures
348	tested. Nevertheless, during that phase, the mean Chl a concentrations decreased from $18.2 \pm 0.9 \mu\text{g}\text{L}^{-1}$ at 10°C to
349	$12.4 \pm 0.7 \mu g L^{-1}$ at 15 °C, suggesting a faster loss of the pigments following the depletion of NO ₃ .
350	
351	New sentence:
352	During Phase II, we observed no significant effect of pCO ₂ , temperature, and the interaction of those factors on the mean
353	Chl a concentrations following the depletion of NO ₃ :
354	L230 : "The strong correlation" I do not understand this sentence. How a correlation can suggest anything about importance?
355	AR: The sentence has been modified.
356	
357	New sentence:
358	The correlation between the nanophytoplankton abundance and Chl a ($r^2 = 0.75$, $p < 0.001$, $df = 166$) suggests that this
359	phytoplankton size class was responsible for most of the biomass build-up throughout the experiment.
360	
361	
362	Figure 1a: label pHT in situ, why in situ?

paper if these results were incorporated and discussed. Top-down control is very often neglected in these OA-OW experiments.

<u>AR:</u> All pH_T are measured at 25°C and are computed to the temperatures of the mesocosms. The label " pH_T in situ" meant that the pH_T is calculated at the in situ temperature of each mesocosm. Therefore, for mesocosms M1–M6 the pH_T is computed at 10°C, while for mesocosms M7–M12 the pH_T is computed at 15°C. To avoid confusion, we changed the label to " pH_T ".

Experimental assessment of the sensitivity of an estuarine phytoplankton fall bloom to acidification and warming

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Correspondence: Robin Bénard (robin.benard.1@ulaval.ca)

Abstract. We investigated the combined effect of ocean acidification and warming on the dynamics of the phytoplankton fall bloom in the Lower St. Lawrence Estuary (LSLE), Canada. Twelve 2600 L mesocosms were set to initially cover a wide range of pH_T (pH on the total proton scale) from 8.0 to 7.2 corresponding to a range of pCO₂ from 440 to 2900 μatm, and two temperatures (in situ and +5 °C). The 13-day experiment captured the development and decline of a nanophytoplankton bloom dominated by the chain-forming diatom Skeletonema costatum. During the development phase of the bloom, increasing pCO2 influenced neither the magnitude nor the net growth rate of the nanophytoplankton bloom whereas increasing the temperature by 5 °C stimulated the chlorophyll a (Chl a) growth rate and maximal particulate primary production (P_P) by 7650 % and 63160 %, respectively. During the declining phase of the bloom, warming accelerated the loss of diatom cells, and paralleled by a gradual decrease in the abundance of photosynthetic picoeukaryotes and a bloom of picocyanobacteria. Increasing pCO₂ and warming did not influence the abundance of picoeukaryotes while picocyanobacteria abundance was reduced by the increase in pCO2 when combined with warming in the latter phase of the experiment. negatively affected Pr. Due to the countervailing responses of the plankton community to warming during the two phases of the experimentOver the full duration of the experiment, the time-integrated net primary production was not significantly affected by the pCO₂ treatments or warmingover the full duration of the study. The diatom bloom was paralleled by a gradual decrease in the abundance of photosynthetic picoeukaryotes and followed by a bloom of picocyanobacteria. Increasing pCO2 and warming did not influence the abundance of picoeukaryotes, but warming stimulated picoeyanobacteria proliferation. Overall, our results suggest that warming, rather than acidification, is more likely to alter phytoplankton autumnal bloom development in the LSLE in the decades to come. Future studies examining a broader gradient of temperatures should be conducted over a larger seasonal window in order to better constrain the potential effect of warming on the development of blooms in the LSLE and its impact on the fate of primary production.

Field Code Changed

1. Introduction

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Anthropogenic emissions have increased atmospheric carbon dioxide (CO₂) concentrations from their pre-industrial value of 280 to 412 ppm in 2017, and concentrations of 850-1370 ppm are expected by the end of the century under the business-asusual scenario RCP 8.5 (IPCC, 2013). The global ocean has already absorbed about 28 % of these anthropogenic CO2 emissions (Le Quéré et al., 2015), leading to a global pH decrease of 0.11 units (Gattuso et al., 2015), a phenomenon known as Ocean Acidification (OA). The surface ocean pH is expected to decrease by an additional 0.3-0.4 units under the RCP 8.5 scenario by 2100, and as much as 0.8 units by 2300 (Caldeira and Wickett, 2005; Doney et al., 2009; Feely et al., 2009). The accumulation of anthropogenic CO2 in the atmosphere also results in an increase in the Earth's heat content that is primarily absorbed by the ocean (Wijffels et al., 2016), leading to an expected rise of sea surface temperatures of 3 to 5 °C by 2100 (IPCC, 2013). Whereas the effect of increasing atmospheric CO₂ partial pressures (pCO₂) on ocean chemistry is relatively well documented, the potential impacts of OA on marine organisms and how their response to OA will be modulated by the concurrent warming of the ocean surface waters are still the subject of much debate (Boyd and Hutchins, 2012; Gattuso et al., 2013). Over the last decade, there has been increasing interest in the potential effects of OA on marine organisms (Kroeker et al., 2013). The first experiments were primarily conducted on single phytoplankton species (reviewed in Riebesell and Tortell, 2011), but subsequent mesocosm experiments highlighted the impact of OA on the structure and productivity of complex plankton assemblages (Riebesell et al., 2007, 2013). Due to their widely different initial and experimental conditions, these ecosystem-level experiments generated contrasting results (Schulz et al., 2017) but some general patterns nevertheless emerged. For example, diatoms generally benefit from higher pCO₂ through stimulated photosynthesis and growth rates since the increase in CO₂ concentrations compensates for the low affinity of RubisCO towards CO₂ (Giordano et al., 2005; Gao and Campbell, 2014). Although most phytoplankton species have developed carbon concentration mechanisms (CCM) to compensate for the low affinity of RubisCO towards CO2, CCM efficiencies differ between taxa, rendering predictions of the impact of a CO₂ rise on the downregulation of CCM rather difficult (Raven et al., 2014). For example, some studies unexpectedly reported no significant or very modest stimulation of primary production under elevated CO2 concentrations (Engel et al., 2005; Eberlein et al., 2017). OA can ultimately affect the structure of phytoplankton assemblages. Small cells such as photosynthetic picoeukaryotes can benefit directly from an increase in pCO2 as CO2 can passively diffuse through their boundary layer (Beardall et al., 2014), and the smallest organisms within this group could benefit most from the increase (Brussaard et al., 2013). Accordingly, OA experiments have typically favoured smaller phytoplankton cells (Yoshimura et al., 2010; Brussaard et al., 2013; Morán et al., 2015), although the proliferation of larger cells has also been reported (Tortell et al., 2002). Hence, generic predictions of phytoplankton community responses to OA are challenging. Few recent studies have investigated the combined effects of OA and warming on natural phytoplankton assemblages (Hare et al., 2007; Feng et al., 2009; Maugendre et al., 2015; Paul et al., 2015, 2016). Laboratory experiments have shown that OA

and warming could together increase photosynthetic rates, but at the expense of species richness, the reduction of diversity

predominantly imputable to warming (Tatters et al., 2013). Results of an experiment conducted with a natural planktonic community from the Mediterranean Sea showed no effect of a combined warming and decrease in pH on primary production, but higher picocyanobacteria abundances were observed in the warmer treatment (Maugendre et al., 2015). Shipboard microcosm incubations conducted in the northern South China Sea displayed higher phytoplankton biomass, daytime primary productivity and dark community respiration under warmer conditions, but these positive responses were cancelled at low pH (Gao et al., 2017). In contrast, a mesocosm experiment carried out with a fall planktonic community from the western Baltic Sea led to a decrease in phytoplankton biomass under warming, but combined warming and increased pCO₂ led to an increase in biomass (Sommer et al., 2015). Results from experiments where the impacts of pCO2 and temperature are investigated individually may be misleading as multiple stressors can interact antagonistically or synergistically, sometimes in a nonlinear, unpredictable fashion (Todgham and Stillman, 2013; Boyd et al., 2015; Riebesell and Gattuso, 2015; Gunderson et al., 2016). The Lower St. Lawrence Estuary (LSLE) is a large (9350 km²) segment of the greater St. Lawrence Estuary (d'Anglejan, 1990). From June to September, the LSLE is characterized by a dynamic succession in the phytoplankton community, mostly driven by changes in light and nutrient availability through variations in the intensity of vertical mixing (Levasseur et al., 1984). The spring and fall blooms are mostly comprised of diatoms, with simultaneous nitrate and silicic acid exhaustion ultimately limiting primary production (Levasseur et al., 1987; Roy et al., 1996). How OA and warming may affect these blooms and primary production has never been investigated in the LSLE. The OA problem is complex in estuarine and coastal waters where freshwater runoff, tidal mixing, and high biological activity contribute to variations in pCO2 and pH on different time scales (Duarte et al., 2013). The surface mixed-layer pCO₂ in the LSLE varies spatially from 139 to 548 µatm and is strongly modulated by biological productivity (Dinauer and Mucci, 2017). Surface pH_T has been shown to vary from 7.85 to 7.93 in a single tidal cycle in the LSLE, nearly as much as the world's oceans have experienced in response to anthropogenic CO₂ uptake over the last century (Caldeira and Wickett, 2005; Mucci et al., 2017). The main objective of this study was to experimentally assess the sensitivity of the LSLE phytoplankton fall assemblage to a

The main objective of this study was to experimentally assess the sensitivity of the LSLE phytoplankton fall assemblage to a large pCO₂ gradient at two temperatures (in situ and +5 °C). Whether lower trophic-level microorganisms thriving in a highly variable environment will show higher resistance or resilience to future anthropogenic forcings is still a matter of speculation.

2. Material and methods

2.1 Mesocosm setup

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The mesocosm system consists of two thermostated full-size ship containers each holding six 2600 L mesocosms (Aquabiotech Inc., Québec, Canada $^{\infty}$). The mesocosms are cylindrical (2.67 m × 1.40 m) with a cone-shaped bottom within which mixing is achieved using a propeller fixed near the top (Fig. 1). The mesocosms exhibit opaque walls and all lie on the same plane level as not to shade each other. Light penetrates the mesocosms only through a sealed Plexiglas circular cover at their uppermost part. The cover allows the transmission of 90 % of photosynthetically active radiation (PAR; 400–700 nm), 85–90 % of UVA (315–400 nm), and 50–85 % of solar UVB (280–315 nm). Each enclosure is sealed with a Plexiglas cover allowing the

transmission of 90 % of photosynthetically active radiation (PAR; 400–700 nm), 50–85 % of solar UVB (280–315 nm) and 85–90 % of UVA (315–400 nm). The mesocosms are equipped with individual, independent temperature probes (AQBT-Temperature sensor, accuracy ± 0.2 °C). Temperature in the mesocosms was measured every 15 minutes during the experiment, and the control system triggered either a resistance heater (Process Technology TTA1.8215) located near the middle of the mesocosm or a pump-activated glycol refrigeration system to maintain the set temperature. The pH in each mesocosm was monitored every 15 minutes using Hach® PD1P1 probes (± 0.02 pH units) connected to Hach® SC200 controllers, and positive deviations from the target values activated peristaltic pumps linked to a reservoir of artificial seawater equilibrated with pure CO₂ prior to the onset of the experiment. This system maintained the pH of the seawater in the mesocosms within ± 0.02 pH units of the targeted values by lowering the pH during autotrophic growth but could not increase the pH during bloom senescence when the pCO₂ rose and pH decreased.

2.2 Setting

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The water was collected at 5 m depth near Rimouski harbour (48° 28' 39.9" N, 68° 31' 03.0" W) on the 27th of September 2014 (indicated as day -5 hereafter), and the experiment lasted until the 15th of October 2014 (day 13). In situ conditions were: salinity = 26.52, temperature = 10 °C, nitrate (NO₃') = $12.8 \pm 0.6 \mu$ mol L⁻¹, silicic acid (Si(OH)₄) = $16 \pm 2 \mu$ mol L⁻¹, and soluble reactive phosphate (SRP) = $1.4 \pm 0.3 \,\mu \text{mol L}^{-1}$. On day -5, the water was filtered through a 250 μm mesh while simultaneously filling the 12 mesocosm tanks by gravity with a custom made 'octopus' tubing system. The water was collected at 5 m depth near Rimouski harbour (48° 28' 39.9" N, 68° 31' 03.0" W) on the 27th of September 2014. In situ conditions were: salinity = 26.52, temperature = 10° C, nitrate (NO₂) = 12.8 ± 0.6 umol L⁴, silicic acid (Si(OH)₄) = 16 ± 2 umol L⁴, and soluble reactive phosphate (SRP) = 1.4 ± 0.3 μmol L⁻¹. The same day (indicated as day 5 hereafter), the water was filtered through a 250 µm mesh while simultaneously filling the 12 mesocosm tanks by gravity with a custom made 'octopus' tubing system. The initial pCO₂ was 623 ± 7 µatm and the in situ temperature of 10 °C was maintained in the twelve mesocosms for the first 24 h (day -4). The initial in situ temperature of 10 °C was maintained in the twelve mesocosms for the first 24 h (day -4). After that period, the six mesocosms in one container were maintained at 10 °C while temperature was gradually increased to 15 °C during-over day -3 in the six mesocosms of the other container. To avoid subjecting the planktonic communities to excessive stress due to sudden changes in temperature and pH while setting the experiment, the mesocosms were left to acclimatize on day -2 before acidification was carried out over day -1. One mesocosm from each temperature-controlled container was not pH-controlled to assess the community response to the freely fluctuating pH. These two mesocosms were labelled "Controls Drifters" as the initial in situ pH was allowed to fluctuate over time with the development of the phytoplankton bloom. The other mesocosms were set to cover a range of pH_T of ca. 8.0 to ca. 27.2 corresponding to a pCO₂ gradient of zea. 440 to zea. 2900 µatm after acidification was carried out. To attain initial targeted pH, CO2-satured artificial seawater was added to the mesocosms that needed a pH lowering while mesocosms M2 (8.0), M4 (7.8), M6 (Drifter), M9 (8.0), M11 (Drifter) and M12 (7.8) were openly mixed to allow the degassing of the supersaturated CO₂. Once the mesocosms had reached their target pH, the automatic system controlled the sporadic addition of CO2-saturated water to refrain the pH from rising. Only the Drifters were not controlled throughout the experiment. Incident light was variable during our experiment, with only few sunny days (Fig. 2).

2.3 Seawater analysis

The mesocosms were sampled between 05:00 and 08:00 a.m. every day. Seawater for carbonate chemistry, nutrients, and primary production were collected directly from the mesocosms as close to sunrise as possible. Seawater was also collected in 20 L carboys for the determination of chlorophyll *a* (Chl *a*), taxonomy, and other variables. Total amount of volume sampled every day was 24 L or less. Samples for salinity were taken from the artificial seawater tanks and in the mesocosms on day -3, 3 and 13. The samples were collected in 250 mL plastic bottles and stored in the dark until analysis was performed using a Guildline Autosal 8400B Salinometer during the following months.

2.3.1 Carbonate chemistry

Carbonate chemistry parameters were determined using methods described in Mucci et al. (2017). Briefly, water samples for pH (every day) and total alkalinity (TA, every 3–4 days) measurements were, respectively, transferred from the mesocosms to 125 mL plastic bottles without headspace and 250 mL glass bottles. A few crystals of $HgCl_2$ were added to the glass bottles before sealing them with a ground-glass stopper and Apiezon® Type-M high-vacuum grease. The pH was determined within hours of collection, after thermal equilibration at 25.0 ± 0.1 °C, using a Hewlett-Packard UV-Visible diode array spectrophotometer (HP-8453A) and a 5 cm quartz cell with phenol red (PR; Robert-Baldo et al., 1985) and *m*-cresol purple (mCP; Clayton and Byrne, 1993) as indicators. Measurements were carried out at the wavelength of maximum absorbance of the protonated (HL) and deprotonated (L) indicators. Comparable measurements were carried out using a TRIS buffer prepared at a practical salinity of ea.-25 before and after each set of daily measurements (Millero, 1986).

The pH on the total proton concentration scale (pH_T) of the buffer solutions and samples at 25 °C was calculated according to

the equation of Byrne (1987), using the salinity of each sample and the HSO₄⁻ association constants given by Dickson (1990). The TA was determined on site within one day of sampling by open-cell automated potentiometric titration (Titrilab 865, Radiometer®) with a pH combination electrode (pHC2001, Red Rod®) and a dilute (0.025N) HCl titrant solution. The titrant was calibrated using Certified Reference Materials (CRM Batch#94, provided by A. G. Dickson, Scripps Institute of Oceanography, La Jolla, USA). The average relative error, based on the average relative standard deviation on replicate standard and sample analyses, was better than 0.15 %. The carbonate chemistry parameters at in situ temperature were then calculated using the computed pH_T at 25 °C in combination with the measured TA using CO₂SYS (Pierrot et al., 2006) and the carbonic acid dissociation constants of Cai and Wang (1998).

524 **2.3.3 Nutrients**

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- 525 Samples for NO₅, Si(OH)₄, and SRP analyses were collected directly from the mesocosms every day, filtered through
- 526 Whatman GF/F filters and stored at -20 °C in acid washed polyethylene tubes until analysis by a Bran and Luebbe Autoanalyzer
- 527 III using the colorimetric methods described by Hansen and Koroleff (2007). The analytical detection limit was 0.03 µmol L
- 528 1 for NO₃ plus nitrite (NO₂), 0.02 -μmol L-1 for NO₂, 0.1 -μmol L-1 for Si(OH)₄, and 0.05- μmol L-1 for SRP.

2.3.4 Plankton biomass, composition and enumeration

- Duplicate subsamples (100 mL) for Chl a determination were filtered onto Whatman GF/F filters. Chl a concentrations were
 - measured using a 10-AU Turner Designs fluorometer, following a 24 h extraction in 90 % acetone at 4 °C in the dark without
- 532 grinding (acidification method: Parsons et al., 1984). The analytical detection limit for Chl-a was 0.05 μg L⁻¹.
- Pico- (0.2-2 µm) and nanophytoplankton (2-20 µm) cell abundances were determined daily by flow cytometry. Sterile
 - cryogenic polypropylene vials were filled with 4.95 mL of seawater to which 50 µL of glutaraldehyde Grade I (final
- concentration = 0.1 %, Sigma Aldrich; Marie et al., 2005) were added. Duplicate samples were flash frozen in liquid nitrogen
 - after standing 15 minutes at room temperature in the dark. These samples were then stored at -80 °C until analysis. After
 - thawing to ambient temperature, samples were analyzed using a FACS Calibur flow cytometer (Becton Dickinson) equipped
 - with a 488 nm argon laser. The abundances of nanophytoplankton and picophytoplankton, which includes photosynthetic
 - picoeukaryotes and picocyanobacteria, were determined by their autofluorescence characteristics and size (Marie et al., 2005).
- 540 The biomass accumulation and nanophytoplankton growth rates were calculated by the following equation:
- 541 $\mu = \ln (N_2/N_1)/(t_2-t_1),$ ______(1)
- where N_1 and N_2 are the biomass or cell concentrations at given times t_1 and t_2 , respectively.
- 543 Microscopic identification and enumeration for eukaryotic cells larger than 2 µm was conducted on samples taken from each
 - mesocosm on three days: day -4, the day when maximum Chl a was attained in each mesocosm, and day 13. Samples of
 - 250 mL were collected and preserved with acidic Lugol solution (Parsons et al., 1984), then stored in the dark until analysis.
- 546 Cell identification was carried out at the lowest possible taxonomic rank using an inverted microscope (Zeiss Axiovert 10) in
 - accordance with Lund et al. (1958). The main taxonomic references used to identify the phytoplankton were Tomas (1997)
 - and Bérard-Therriault et al. (1999).

2.3.5 Primary production

- 550 Primary production was determined daily using the ¹⁴C-fixation incubation method (Knap et al., 1996; Ferland et al., 2011).
- 551 One clear and one dark 250 mL polycarbonate bottle were filled from each mesocosm at dawn and spiked with 250 μL of
- NaH¹⁴CO₃ (80 μCi mL⁻¹). One hundred μL of 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) (0.02 mol L⁻¹) was added to
- the dark bottles to prevent active fixation of ¹⁴C by phytoplankton (Legendre et al., 1983). The total amount of radioisotope in
- 554 each bottle was determined by immediately pipetting 50 µL subsamples into a 20 mL scintillation vial containing 10 mL of

scintillation cocktail (EcolumeTM) and 50 μ L of ethanolamine (Sigma). Bottles were placed in separate incubators, at either 10 °C or 15 °C, under reduced (30 %) natural light for 24 h, which corresponds to the light transmittance at mid-mesocosm depth.

At the end of the incubation periods, 3 mL were transferred to a scintillation vial for determination of the total primary production (P_T), 3 mL were filtered through a syringe filter (GD/X 0.7 μ m) to estimate daily photosynthetic carbon fixation released in the dissolved organic carbon pool (P_D). The remaining volume was filtered onto a Whatman GF/F filter to measure the particulate primary production (P_P). Vials containing the P_T and P_D samples were acidified with 500 μ L of HCl 6 N, allowed to sit for 3 h under a fume hood, then neutralized with 500 μ L of NaOH 6 N. The vials containing the filters were acidified with 100 μ L of 0.05 N HCl and left to fume for 12 h. Fifteen mL of scintillation cocktail were added to the vials and they were stored pending analysis using a Tri-Carb 4910TR liquid scintillation counter (PerkinElmer). Rates of carbon fixation into particulate and dissolved organic matter were calculated according to Knap et al. (1996) using the dissolved inorganic carbon concentration computed for each mesocosm at the beginning of the daily incubations and multiplied by a factor of 1.05 to correct for the lower uptake of 14 C compared to 12 C.

2.4 Statistical analysis

All statistical analyses were performed using R (nlme package). A general least squares (gls) model approach was used to test the linear effects of the two treatments (temperature, pCO₂), and of their interactions on the measured variables (Paul et al., 2016; Hussherr et al., 2017). The analysis was conducted independently on two different time periods: Phase I (day 0 to day of maximum Chl a concentration) was calculated individually for each mesocosm, whereas Phase II (day after maximum Chl a concentrations) corresponded to the declining phase of the bloom (Table 1). Phase I (day 0 to day 4) corresponded to the development of the diatom bloom and extended up to the depletion of nitrate, whereas Phase II (day 5 to day 13) corresponded to the declining phase of the bloom in the absence of detectable nitrate. Averages (or time-integration in the case of primary production) of the response variables were calculated separately over the two phases and were plotted against pCO₂. Separate regressions were performed with pCO₂ as the continuous factor for each temperature when a temperature effect or interaction with pCO₂ was detected in the gls model. Otherwise, the model included data from both temperatures and the interaction with pCO₂. Normality of the residuals was determined using a Shapiro-Wilk test (p > 0.05) and data were transformed (natural logarithm or square root) if required. As explained by Havenhand et al. (2010), the gradient approach, instead of treatment replication, is particularly suitable when few experimental units are available such as in large volume mesocosm experiments. In addition, squared Pearson's correlation coefficients (r²) with a significance level of 0.05 were used to evaluate correlations between key variables.

3. Results

3.1 Seawater chemistry

Water salinity was 26.52 ± 0.03 on day -4 in all mesocosms and remained constant throughout the experiment, averaging 26.54 ± 0.02 on day 13. The TA was practically invariant in the mesocosms, averaging $2057 \pm 2 \,\mu$ mol kg_{sw}^{-1} on day -4 and $2058 \pm 2 \,\mu$ mol kg_{sw}^{-1} on day 13. Following the filling of the mesocosms, the pH_T in all mesocosms decreased from an average of 7.84 to 7.53. Throughout the rest of the experiment after treatments were applied, the pH remained relatively stable in the pH-controlled treatments, but decreased slightly during Phase II by an average of -0.14 ± 0.07 units relative to the target pH_T (Fig. 3a). The pH remained relatively stable throughout the experiment in the pH-controlled treatments, but decreased slightly during Phase II by an average of -0.14 ± 0.07 units relative to the target pH_T (Fig. 1a). Given a constant TA, pH variations were accompanied by variations in pCO₂, from an average of $1340 \pm 150 \,\mu$ atm on day -3, and ranging from 564 to 2902 μ atm at 10 °C, and from 363 to 2884 μ atm at 15 °C on day 0 following the acidification (Fig. 34b; Table 1). The pH_T in the DrifterControls (M6 and M11) increased from 7.896 and 7.862 on day 0 at 10 °C and 15 °C, respectively, to 8.307 and 8.554 on day 13, reflecting the balance between CO₂ uptake and metabolic CO₂ production over the duration of the experiment. On the last day, pCO₂ in all mesocosms ranged from 186 to 3695 μ atm at 10 °C, and from 90 to 3480 μ atm at 15 °C. The temperature of the mesocosms in each container remained within \pm 0.1 °C of the target temperature throughout the experiment and averaged 10.04 ± 0.02 °C for mesocosms M1 through M6, and 15.0 ± 0.1 °C for mesocosms M7 through M12 (Fig. 34c; Table 1).

3.2 Dissolved inorganic nutrient concentrations

Nutrient concentrations averaged $9.1 \pm 0.5 \,\mu\text{mol} \,L^{-1}$ for NO_3 , $13.4 \pm 0.3 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$, and $0.91 \pm 0.03 \,\mu\text{mol} \,L^{-1}$ for $Si(OH)_4$ fell below the development of the phytoplanktonic bloom. Overall, $Si(OH)_4$ depletion and $Si(OH)_4$ fell below the detection limit between day 1 and 5 in all mesocosms except for those whose $Si(OH)_4$ fell below the detection limit between day 1 and 5 in all mesocosms except for those whose $Si(OH)_4$ fell below the detection limit between day 1 and 5 in all mesocosms except for those whose $Si(OH)_4$ depletion occurred on day 9. Variations in $Si(OH)_4$ concentrations followed closely those of $Si(OH)_4$ in all mesocosms except again for those set at $Si(OH)_4$ and $Si(OH)_4$ in which undetectable values were reached on day 9.

3.3 Phytoplankton biomass

Chl a concentrations where below 1 μ g L⁻¹ just after the filling of the mesocosms, and averaged 5.9 ± 0.6 μ g L⁻¹ on day 0 (Fig. $\underline{42}a$). They then quickly increased to reach maximum concentrations around 27 ± –2 μ g L⁻¹ on day 3 ± 2, and decreased progressively until the end of the experiment, reaching 11 ± 1 μ g L⁻¹ and 2.4 ± 0.2 μ g L⁻¹ at 10 °C and 15 °C on day 13. During Phase I, results from the gls model show no significant relationships between the mean Chl a concentrations and pCO₂, temperature, and the interaction of the two factors at the two temperatures tested but significantly higher Chl a values at 15 °C than at 10 °C (Fig. $\underline{42}b$; Table $\underline{24}$). During this phase, the accumulation rate of Chl a was positively affected by temperature, increasing by ~76 %, but was not affected bynot-by the pCO₂ gradient at either temperature (Fig. $\underline{53}a$; Table $\underline{32}$). The maximum Chl a concentrations reached during the bloom were not affected by the two treatments (Fig. $\underline{53}b$; Table $\underline{32}$). During Phase II, we observed no significant effect of increasing pCO₂, temperature, and the interaction of those factors on the mean Chl a concentrations following the depletion of NO₃. (Fig. 4c; Table 4) at the two temperatures tested. Nevertheless, during that phase, the mean Chl a concentrations decreased from 18.2 ± 0.9 μ g L⁺ at 10 °C to 12.4 ± 0.7 μ g L⁺ at 15 °C, suggesting a faster loss of the pigments following the depletion of NO₃.

3.4 Phytoplankton size-class

- Nanophytoplankton abundance varied from $8 \pm 1 \times 10^6$ cells L⁻¹ on day 0 to an average maximum of $36 \pm 10 \times 10^6$ cells L⁻¹ at the peak of the bloom (Fig. 42d). At both temperatures, nanophytoplankton abundance increased until at least days 2 or 4 and decreased or remained stable thereafter. The strong correlation between the nanophytoplankton abundance and Chl a ($r^2 = 0.7582$, p < 0.001, df = 166) suggests that this phytoplankton size class was responsible for most of the biomass build-up in the mesocosmsthroughout the experiment. As observed for the mean Chl a concentration, the mean abundance of nanophytoplankton was not significantly affected by the pCO₂ gradient at the two temperatures investigated during Phase I, but showed higher values at 15 °C ($2631 \pm 3.2 \times 10^6$ cells L⁻¹) than at 10 °C ($13.14 \pm 2.1 \times 10^6$ cells L⁻¹) (Fig. 42e; Table 21). Likewise, the growth rate of nanophytoplankton during Phase I was not influenced by the pCO₂ gradient at the two temperatures but was significantly higher in the warm treatment (Fig. 53c; Table 32). During Phase II, no relationship was found between the mean nanophytoplankton abundance and the pCO₂ gradient, at the twothe temperatures, and no the pCO₂ × temperature effect interaction was observed (Fig. 42f; Table 43).
- Initial abundance of photosynthetic picoeukaryotes was $10 \pm 2 \times 10^6$ cells L⁻¹, accounting for more than 80 % of total plankton cells in the $0.2–20~\mu m$ size fraction. The abundance of this plankton size fraction decreased slightly through Phase I and their number remained relatively stable at $43.3 \pm 30.2 \times 10^6$ cells L⁻¹ throughout Phase II (Fig. 42g). We found no relationship between the abundance of picoeukaryotes and the pCO₂ gradient at the two temperatures investigated during both Phases I and II, and no temperature effect was observed either (Fig. 42h, i; Tables 21 and 43).
- Picocyanobacteria exhibited a different pattern than the nanophytoplankton and picoeukaryotes (Fig. $\underline{42}$ j). Their abundance was initially low $(1.7 \pm 0.3 \times 10^6 \text{ cells L}^{-1} \text{ on day 0})$, remained relatively stable during Phase I, and increased rapidly during

Phase II, accounting for \sim ea.-50 % of the total picophytoplankton cell counts toward the end of the experiment. During Phase II, the mean picocyanobacteria abundance was not influenced by the pCO₂ gradient but was higher at 15 °C (1.4 ± 0.2 × 10⁶ cells L⁻¹) than at 10 °C (0.95 ± 0.05 × 10⁶ cells L⁻¹) or temperature (Fig. 42k; Table 21). During Phase II, their mean picocyanobacteria abundance was not significantly affected by pCO₂ at in situ temperature, remained higher aHowever, mean picocyanobacteria were higher at 15 °C, with the pCO₂ gradient responsible for a ~33% reduction of picocyanobacteria abundance from the Drifter to the more acidified treatment (4.5 4 ± 0.3 - 2 × 10⁶ cells L⁻¹ vs. 3.0 ± 0.3 × 10⁶ cells L⁻¹) than at 10 °C (2.6 ± 0.1 × 10⁶ cells L⁻¹), and again no significant effect of pCO₂ was detected (Fig 42l; Table 43).

3.5 Phytoplankton taxonomy

The taxonomic composition of the planktonic assemblage larger than 2 μm was identical in all treatments at the beginning of the experiment, and was mainly composed of the cosmopolitan chain-forming centric diatom *Skeletonema costatum* (*S. costatum*) and the cryptophyte *Plagioselmis prolonga* var. *nordica* (Fig. 64). At the peak of the blooms (maximum Chl *a* concentrations), the species composition did not vary between the pCO₂ treatments and between the two temperatures tested. *S. costatum* was the dominant species in all mesocosms (70–90 % of the total number of eukaryotic cells), except for one mesocosm (M3, pH 7.6 at 10 °C) where a mixed dominance of *Chrysochromulina* spp. (a prymnesiophyte of 2–5 μm) and *S. costatum* was observed (Fig. 64a). *S. costatum* accounted for 80–90 % of the total eukaryotic cell counts in all mesocosms at the end of the experiment carried out at 10 °C. At 15 °C, the composition of the assemblage had shifted toward a dominance of unidentified flagellates and choanoflagellates (2–20 μm) in all mesocosms with these two groups accounting for 55–80 % of the total cell counts while diatoms showed signs of loss of viability as indicated by the presence of empty frustules (Fig. 64b).

3.6 Primary production

 P_P increased in all mesocosms during Phase I of the experiment, in parallel with the increase in Chl a (Fig. 75a). P_P maxima were attained on days 3–4, except for the 15 °C <u>DrifterControl</u> (M11) where P_P peaked on day 1. We found no significant effect of the pCO₂ gradient, temperature and the pCO₂ × temperature interaction—on the time-integrated P_P at the two temperatures during both Phases I and II (Fig. 75b, c; Tables 24 and 43), but time integrated P_P was higher at 15 °C than at 10 °C during Phase II (Tables 1 and 3). Similarly, opposite responses to warming were observed the absence of significant treatment effects remained when normalizing P_P per unit of Chl a (Fig. 75g, h, i). Initial Chl a-normalized P_P values were $3.3 \pm 0.5 \,\mu$ mol C (μ g Chl a)-1 d-1 and reached maxima between $3.7 \pm 0.3 \,\mu$ mol C (μ g Chl a)-1 d-1 and $5.76.0 \pm 0.67 \,\mu$ mol C (μ g Chl a)-1 d-1 or the last day of the experiment (Fig. 5g). During Phase I, the mean Chl a-normalized P_P was not significantly affected by the pCO₂ gradient or higher under warming, but as observed for the mean Chl a concentrations and time-integrated P_P over that phase, was not significantly affected by the pCO₂ gradient (Fig. 75h; Table 24). During Phase II, the log of the mean Chl a-normalized P_P was not significantly affected by the pCO₂ gradient (Fig. 75h; Table 24). During Phase II, the log of the mean Chl a-normalized P_P was not significantly affected by the pCO₂ gradient

gradient, the temperature, or the interaction of these factors decreased with increasing temperature, with values of $2.2 \pm 0.2 \,\mu\text{mol} \, \text{C} \, (\mu\text{g} \, \text{Chl} \, a)^{+} \, \text{d}^{+} \, \text{and} \, 1.6 \pm 0.1 \, \mu\text{mol} \, \text{C} \, (\mu\text{g} \, \text{Chl} \, a)^{+} \, \text{d}^{+} \, \text{at} \, 10 \, ^{\circ} \text{C} \, \text{and} \, 15 \, ^{\circ} \text{C}, \text{ respectively} (Fig. 75i; Table 43).$ No significant effect of pCO₂-was detected.

 P_D was low at the beginning of the experiment, averaging $1.5 \pm 0.4~\mu$ mol C $L^{-1}~d^{-1}$, increased progressively during Phase I to reach maximum values of up-to-6_48 μ mol C $L^{-1}~d^{-1}~on$ -between days 4 and 58_at the beginning of Phase II, and decreased thereafter (Fig. 75d). Time-integrated P_D was not significantly affected neither-by the pCO₂ gradient, the temperature, and the pCO₂ × temperature interaction at the two temperatures tested nor by temperature-during the two pPhases (Fig. 75e, f; Tables 24 and 43). Chl a-normalized P_D was low on day 0, averaging $0.3 \pm 0.1~\mu$ mol C (μ g Chl a)⁻¹ d⁻¹, reached maximum values of $1.04 \pm 0.5 - 2~\mu$ mol C (μ g Chl a)⁻¹ d⁻¹ at the beginning of Phase II and $1.6 \pm 0.2~\mu$ mol C (μ g Chl a)⁻¹ d⁻¹ at $10~^{\circ}$ C and $15~^{\circ}$ C, then and-respectively decreased to $0.173 \pm 0.405~\mu$ mol C (μ g Chl a)⁻¹ d⁻¹ and $0.6 \pm 0.2~\mu$ mol C (μ g Chl a)⁻¹ d⁻¹ by the end of the experiment (Fig. 75j). During Phases I₂-and II, the mean Chl a-normalized P_D wasere affected neither by the pCO₂ gradient, the temperature and pCO₂ gradient, nor by the interaction between those factors (Fig. 75k,-1; Tables 24 and 3). During Phase II, the log of the mean Chl a-normalized P_D was not affected by pCO₂ at either temperature tested, but significantly increased with warming (Fig. 71; Table 4).

Figure 6 shows the influence of the treatments on maximum P_P and P_D as well as on the time-integrated P_P and P_D over the full length of the experiment. We found no effect of the pCO₂ gradient on the maximum P_P values at the two temperatures tested, but warming increased the maximum P_P values from $66 \pm 13 \mu mol C L^{-1} d^{-1}$ to $126 \pm 8 \mu mol C L^{-1} d^{-1}$ (Fig. 86a; Table 54). The time-integrated P_P over the full duration of the experiment was not affected by the pCO₂ gradient or the increase in temperature (Fig. 86b; Table 54). The absence of temperature effect results from the countervailing responses in time-integrated P_P between Phases I and II. The maximum P_D values were significantly affected by the treatments (Fig 86c; Table 54). Maximum P_D decreased with increasing pCO₂ at in situ temperature but warming cancelled this effect (antagonistic effect). Nevertheless, the time-integrated P_D over the whole experiment did not vary significantly between treatments, although a decreasing tendency with increasing pCO₂ at 10 °C and an increasing tendency with warming can be seen in Fig. 86d (Table 54).

4. Discussion

4.1 General characteristics of the bloom

The onset of the experiment was marked by an increase of pCO₂ on the day following the filling of the mesocosms. This phenomenon often takes place at the beginning of such experiments when pumping tends to break phytoplankton cells and larger debris into smaller ones. We attribute the rapid fluctuations in pCO₂ to the release of organic matter following the filling of the mesocosms with a stimulating effect on heterotrophic respiration, and hence CO₂ production. Then, Aa phytoplankton bloom, numerically dominated by the centric diatom *S. costatum*, took place in all mesocosms, regardless of treatments (Fig.

et al., 2004; Annane et al., 2015). The length of the experiment (13 days) allowed us to capture both the development and declining phases of the bloom. The exponential growth phases lasted 1-4 days depending on the treatments, but maximal Chl a concentrations were reached only after 7 days in two of the twelve mesocosms (Fig. 42a; Table 1). The suite of measurements and statistical tests conducted did not provide any clues as to the underlying causes for the lower rates of biomass accumulation measured in these two mesocosms. Since statistical analyses conducted with or without these two apparent outliers gave similar results, they were not excluded from the analyses. In situ nutrient conditions prior to the water collection were favourable for a bloom development. Based on previous studies, in situ phytoplankton growth was probably limited by light due to water turbidity and vertical mixing at the time of water collection (Levasseur et al. 1984). Grazing may also have played a role in keeping the in situ biomass of flagellates low prior to our sampling. However, a natural diatom fall bloom was observed in the days following the water collection in the adjacent region (Ferreyra, pers. comm.). The increased stability within the mesocosms, combined with the reduction of the grazing pressure (filtration on 250 µm) likely contributed to the fast accumulation of phytoplankton biomass. During the development phase of the bloom, the concentration of all three monitored nutrients decreased, with NO₃ and Si(OH)₄ reaching undetectable values. This nutrient co-depletion is consistent with results from previous studies suggesting a co-limitation of diatom blooms by these two nutrients in the St. Lawrence Estuary (Levasseur et al., 1987, 1990). Variations in PP roughly followed changes in Chl a, and, as expected, the maximum Chl a-normalized P_P (5 ± 2 µmol C (µg Chl a)-1 d-1) was reached during the exponential growth phase in all mesocosms. Decreases in total phytoplankton abundances and P_P followed the bloom peaks and the timing of the NO₃ and Si(OH)₄ depletions. A clear succession in phytoplankton size classes characterized the experiment. Nanophytoplankton cells were initially present in low abundance and became more numerous as the S. costatum diatom bloom developed. The strong-correlation ($r^2 = 0.83$, p < 0.001, df=34) between the abundance of nanophytoplankton and S. costatum enumeration suggests that this cell size class can be used as a proxy of S. costatum counts in all mesocosms throughout the experiment. Nanophytoplankton cells accounted for 79 ± 7 % of total counts of cells < 20 μ m on the day of the maximum Chl a concentration. Accordingly, nanophytoplankton exhibited the same temporal trend as Chl a concentrations. During Phase II, nanophytoplankton abundances remained roughly stable at in situ temperature but decreased at 15 °C towards the end of the experiment. Photosynthetic picoeukaryotes were originally abundant and decreased throughout the experiment whereas picocyanobacteria abundances increased during Phase II. This is a typical phytoplankton succession pattern for temperate systems where an initial diatom bloom growing essentially on allochthonous nitrate gives way to smaller species

64). S. costatum is a common phytoplankton species in the St. Lawrence Estuary and in coastal waters (Kim et al., 2004; Starr

4.2 Phase I (Diatom bloom development)

growing on regenerated forms of nitrogen (Taylor et al., 1993).

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Our results show no significant effect of increasing pCO₂/decreasing pH on the mean abundance and net accumulation rate of the diatom-dominated nanophytoplankton assemblage during the development of the bloom (Figs. 42e and 53c). These results suggest that *S. costatum*, the species accounting for most of the biomass accumulation during the bloom, neither benefited

from the higher pCO2 nor was negatively impacted by the lowering of pH. Assuming that S. costatum was also responsible for most of the carbon fixation during the bloom development phase, the absence of effect on P_P and Chl a-normalized P_n following increases in pCO₂ brings additional support to our conclusion. S. costatum operates a highly efficient CCM, minimizing the potential benefits of thriving in high CO₂ waters (Trimborn et al., 2009). This may explain why the strain present in the LSLE did not benefit from the higher pCO₂ conditions. Likewise, a mesocosm experiment conducted in the coastal North Sea showed no significant effect of increasing pCO₂ on carbon fixation during the development of the spring diatom bloom (Bach et al., 2017; Eberlein et al., 2017). In addition to the aforementioned insensitivity to increasing pCO₂, our results point towards a strong resistance of S. costatum to severe pH decline. During our study, surprisingly constant rates of Chl a accumulation and nanophytoplankton growth (Fig. 53a, c), as well as maximum P_P (Fig. 86a), were measured during the development phase of the bloom over a range of pH_T extending from 8.6 to 7.2 (Fig. 34a). In a recent effort to estimate the causes and amplitudes of short-term variations in pH_T in the LSLE, Mucci et al. (2017) showed that pH_T in surface waters was constrained within a range of 7.85 to 7.93 during a 50h survey over two tidal cycles at the head of the Laurentian Channel. It is notable that even the upwelling of water from 100 m depth or of low-oxygen LSLE bottom water would not decrease pH_T beyond ~ea. 7.75 and ~7.62, respectively (Mucci et al., 2017 and references therein). Our results show that the phytoplankton assemblage responsible for the fall bloom may tolerate even greater pH_T excursions. In the LSLE, such conditions may arise when the contribution of the low pH_T (7.12) freshwaters of the Saguenay River to the LSLE surface waters is amplified during the spring freshet. However, considering that comparable studies conducted in different environments have reported negative effects of decreasing pH on diatom biomass accumulation (Hare et al., 2007; Hopkins et al., 2010; Schulz et al., 2013), it cannot be concluded that all diatom species thriving in the LSLE are insensitive to acidification. In contrast to the pCO₂ treatment, warming affected the development of the bloom in several ways. Increasing temperature by 5 °C significantly increased the accumulation rate of Chl a, and the nanophytoplankton growth rate, as well as the timeintegrated P_P and Chl a normalized P_P during Phase I of the bloom. The positive effects of warming on the Chl a normalized maximum P_P during the development phase of the bloom most likely reflect the sensitivity of photosynthesis to temperature (Sommer and Lengfellner, 2008; Kim et al., 2013). It could also be related to optimal growth temperatures, which are often higher than in situ temperatures in marine phytoplankton (Thomas et al., 2012; Boyd et al., 2013). In support of this hypothesis, previous studies have reported optimal growth temperatures of 20-25 °C for S. costatum, which is 5-10 °C higher than the warmer treatment investigated in our study (Suzuki and Takahashi, 1995; Montagnes and Franklin, 2001). Extrapolating results from a mesocosm experiment to the field is not straightforward, as little is known of the projected warming of the upper waters of the LSLE in the next decades. In the Gulf of St. Lawrence, positive temperature anomalies in surface waters have varied from 0.25 to 0.75 °C per decade between 1985 and 2013 (Larouche and Galbraith, 2016). In the LSLE, warming of surface

waters will likely result from a complex interplay between heat transfer at the air-water interface and variations in vertical

mixing and upwelling of the cold intermediate layer at the head of the Estuary (Galbraith et al., 2014). Considering current

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uncertainties regarding future warming of the LSLE, studies should be conducted over a wider range of temperatures in order to better constrain the potential effect of warming on the development of the blooms in the LSLE.

Picoeukaryotes showed a more or less gradual decrease in abundance during Phase I, and our results show that this decline was not influenced by the increases in pCO₂ (Fig. 42g, h; Table 24). Picoeukaryotes are expected to benefit from high pCO₂ conditions even more so than diatoms as CO₂ can passively diffuse through their relatively thin boundary layer precluding the necessity of a costly uptake mechanism such as a CCM (Schulz et al., 2013). This hypothesis has been supported by several studies showing a stimulating effect of pCO₂ on picoeukaryote growth (Bach et al., 2016; Hama et al., 2016; Schulz et al., 2017 and references therein). On the other hand, in nature, the abundance of picoeukaryotes generally results from a delicate balance between cell division rates and cell losses through microzooplankton grazing and viral attacks. The few experiments, including the current study, reporting the absence or a modest effect of increasing pCO₂ on the abundance of eukaryotic picoplankton attribute their observations to an increase in nano- and microzooplankton grazing (Rose et al., 2009; Neale et al., 2014). During our experiment, the biomass of microzooplankton increased with increasing pCO₂ by ca. 200–300 % at the two temperatures tested (Ferreyra and Lemli, unpubl. data). Thus, it is possible that a positive effect of increasing pCO₂ and warming on picoeukaryote abundances might have been masked by higher picoeukaryote losses due to increased microzooplankton grazing.

4.3 Phase II (declining phase of the bloom)

The gradual decrease in nanophytoplankton abundances coincided with an increase in the abundance of picocyanobacteria (Fig. 42j). Regardless of At in situ temperature, the picocyanobacteria abundance during Phase II was unaffected by the increase in pCO2 over the full range investigated (Fig. 421; Table 43). The lack of positive response of picocyanobacteria to elevated pCO₂ was somewhat surprising considering that they have less efficient CCMs than diatoms (Schultz et al., 2013). Accordingly, several studies have reported a stimulation of the net growth rate of picocyanobacteria under elevated pCO2 in different environments (coastal Japan, Mediterranean Sea, and Raunejforden in Norway) and under different nutrient regimes, i.e. bloom and post-bloom conditions (Hama et al., 2016; Sala et al., 2016; Schulz et al., 2017). Consistent with our observations, hHowever, Law et al. (2012) and Lomas et al. (2012) studies have also shown observed no direct effect of elevated pCO₂ on the net growth of picocyanobacteria during studies conducted in the Subtropical North Atlantic and the South Pacific (Law et al., 2012; Lomas et al., 2012). In our study, picocyanobacteria abundance was even reduced when high CO₂a was combined with warming. Similar negative effects of CO₂ on picocyanobacteria (particularly Synechococcus) have also been observed under later stages of bloom development, i.e. nutrient depletion, either caused by competition or grazing (Paulino et al., 2008; Hopkins et al., 2010). A potential increase in grazing pressure, following the rise in heterotrophic nanoflagellates abundance (e.g. choanoflagellates; Fig. 64b) measured under high pCO2 and warmer conditions, could explain the ostensible absence of stimulationnegative effect of increasing pCO₂ on picocyanobacteria by increasing pCO₂ abundance in our experiment. Despite the absence of grazing measurements during our study, our results support the hypothesis that the potential

for increased picocyanobacteria population growth under elevated pCO_2 and temperature is partially dependent on different grazing pressures (Fu et al., 2007).

WNeither warming, not nor acidification, affected the net particulate carbon fixation during the declining phase of the bloom. In our study, the time-integrated P_pprimary production and Chl a-normalized primary productionP_p were not significantly affected by the increase in pCO₂ during Phase II at the two temperatures tested (Fig. 75.; Table 43). This result is surprising since nitrogen-limited cells have been shown to be more sensitive to acidification, resulting in a reduction in carbon fixation rates due to higher respiration (Wu et al., 2010; Gao and Campbell, 2014; Raven et al., 2014). Although our measurements do not allow to discriminate between the contributions of the different phytoplankton size classes to carbon fixation, we can speculate that diatoms, which were still abundant during Phase II, contributed to a significant fraction of the primary production. If so, these results suggest that S. costatum remained insensitive to OA even under nutrient stress. However, in contrast to Phase I, increasing the temperature by 5 °C during Phase II significantly reduced Pp and their the Chl anormalized P_PP_D. Dark phytoplankton respiration rates generally increase with temperature (Butrón et al., 2009; Robarts and Zohary, 1987). The warming-induced decrease in carbon fixation increase in fixed carbon being release in the dissolved fraction likely stems from increased exudation by phytoplankton, or sloppy feeding / excretion following ingestion by grazers (Kim et al., 2011). The increase in fixed carbon released as dissolved organic carbon (DOC) measured during Phase II may also result from greater respiration by the nitrogen-limited diatoms during periods of darkness of the incubations, as dark phytoplankton respiration rates generally increase with temperature (Butrón et al., 2009; Robarts and Zohary, 1987). Moreover, the enclosures do not permit the sinking and export of particulates organic carbon (POC), allowing a further transformation into DOC by heterotrophic bacteria, a process that could be exacerbated under warming (Wohlers et al., 2009). measured during Phase II may thus result from an increase in respiration by the nitrogen-limited diatoms.

4.4 Effect of the treatments on primary production over the full experiment

As mentioned above, increasing pCO₂ had no effect on time-integrated P_P during the two phases of the bloom, andbut warming resulted respectively in a positive and negative effect during Phases I and Honly affected the maximum P_P.—As a result, primary production rates integrated over the whole duration of the experiment were not significantly different between the two temperatures tested. Although not statistically significant, the time-integrated P_Ds over the full experiment displays a slight decrease with increasing pCO₂ at 10 °C and overall higher values in the warmer treatment (Fig. 86d; Table 54). Previous studies have reported increases of dissolved organic carbon (DOCDOC) exudation (Engel et al., 2013), but also decreasing DOC concentrations at elevated pCO₂ under nitrate limitation (Yoshimura et al., 2014). The increase in DOC exudation is attributed to a stimulation of photosynthesis resulting from its sensitivity to higher pCO₂ (Engel et al. 2013), but the causes for a decrease in DOC concentrations at high pCO₂ are less clear and potentially attributable to an increase in transparent exopolymer particle (TEP) production (Yoshimura et al., 2014). Elevated TEP production under high pCO₂ conditions has been measured both at the peak of a bloom in a mesocosm study (Engel et al., 2014), and in post-bloom nutrient depleted conditions (MacGilchrist et al., 2014). However, during our study, TEP production decreased under high pCO₂ (Gaaloul, 2017). Thus,

the apparent decrease in P_D cannot be attributed to a greater conversion of exuded dissolved carbohydrate into TEP. The apparent rise in P_D under warming is consistent with previous studies reporting similar increases in phytoplankton dissolved carbon release with temperature (Morán et al., 2006; Engel et al., 2011). Although these apparent changes in P_D with increasing pCO₂ and warming require further investigations, they suggest that a larger proportion ($\underline{\sim}$ ea.-15 % of P_T at 15 °C compared to 10 % at 10 °C) of the newly fixed carbon could be exuded and become available for heterotrophic organisms under warmer conditions.

4.5 Implications and limitations

 During our study, we chose to keep the pH constant during the whole experiment instead of allowing it to vary with changes in photosynthesis and respiration during the bloom phases. This approach differs from previous mesocosm experiments where generally no subsequent CO2 manipulations are conducted after the initial targets are attained (Schulz et al. 2017 and therein). Keeping the pH and pCO₂ conditions stable during our study allowed us to precisely quantify the effect of the changing pH/pCO₂ on the processes taking place during the different phases of the bloom. Such control was not exercised in two of our mesocosms (i.e. the Drifters). In these two mesocosms, the pH_T increased from 7.9 to 8.3 at 10 °C, and from 7.9 to 8.7 at 15 °C. Since the buffer capacity of acidified waters diminishes with increasing CO2, the drift in pCO2 and pH due to biological activity would have been even greater in the more acidified treatments (Delille et al., 2005; Riebesell et al., 2007). Hence, allowing the pH to drift in all mesocosms would have likely ended in an overlapping of the treatments where acidification effects would have been harder to detect. Thus, our experiment could be considered as an intermediate between strictly controlled small scale laboratory experiments and large scale pelagic mesocosm experiments in which only the initial conditions are set. By limiting pCO₂ decrease under high CO₂ drawdown due to photosynthesis during the development of the bloom phase, we minimise confounding effects of pCO₂ potentially overlapping in association with high biological activity in the mesocosms. Hence, the experimental conditions could be considered as extreme examples of acidification conditions, due to the extent of pCO2 values studied. However, the absence of OA effects on most biological parameters measured during our study, even under these extreme conditions, strengthens the argument that the phytoplankton community in LSLE is resistant to OA.

5. Conclusion

Our results reveal a remarkable resistance of the different phytoplankton size classes to the large range of pCO₂/pH investigated during our study. It is noteworthy that the plankton assemblage was submitted to decreases in pH far exceeding those that they are regularly exposed to in the LSLE. The resistance of *S. costatum* to the pCO₂ treatments suggests that the acidification of surface waters of the LSLE will not affect the development rate and the amplitude of fall blooms dominated by this species. Photosynthetic picoeukaryotes and picocyanobacteria thriving alongside the blooming diatoms were also insensitive to acidification. In contrast to the pCO₂ treatments, warming the water by 5 °C had multiple impacts on the development and decline of the bloom. The 5 °C warming hastened the development of the diatom bloom (albeit with no increase in total cells

872 number) and increased the abundance of picocyanobacteria during Phase II despite a reduction under high pCO2. These 873 temperature-induced variations in the phytoplankton assemblage were accompanied by an increase in maximal P_P and suggest 874 a potential increase in P_D under warming, although no significant changes in time-integrated P_P and P_D were observed over the 875 phases or the full temporal scale of the experiment, respectively, by higher then lower Pp during the development and declining 876 phases of the diatom bloom. Due to these contrasting responses, warming had no net effect on P_P over the full temporal scale 877 of the experiment. Overall, our results indicate that warming could have more important impacts than acidification on 878 phytoplankton bloom development in the LSLE in the next decades. Future studies should be conducted and specifically 879 designed to better constrain the potential effects of warming on phytoplankton succession and primary production in the LSLE.

Data availability. The data have been submitted to be freely accessible via https://issues.pangaea.de/browse/PDI-16607, or can be obtained by contacting the author (robin.benard.1@ulaval.ca).

Author contributions. R. Bénard was responsible for the experimental design elaboration, data sampling and processing, and the redaction of this article. Several co-authors supplied specific data included in this article, and all co-authors contributed to this final version of the article.

Competing interests. The authors declare that they have no conflict of interest.

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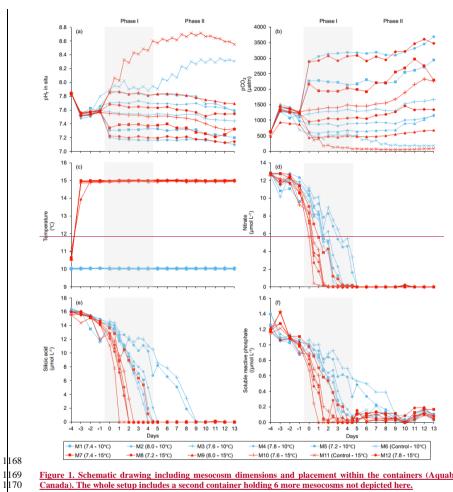


Figure 1. Schematic drawing including mesocosm dimensions and placement within the containers (Aquabiotech Inc, Québec, Canada). The whole setup includes a second container holding 6 more mesocosms not depicted here.



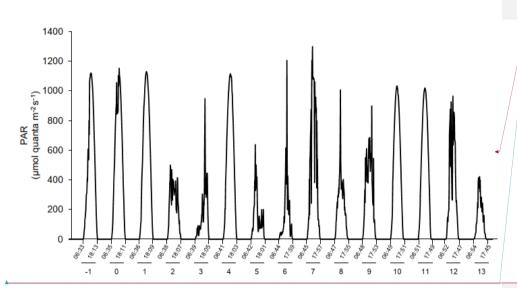
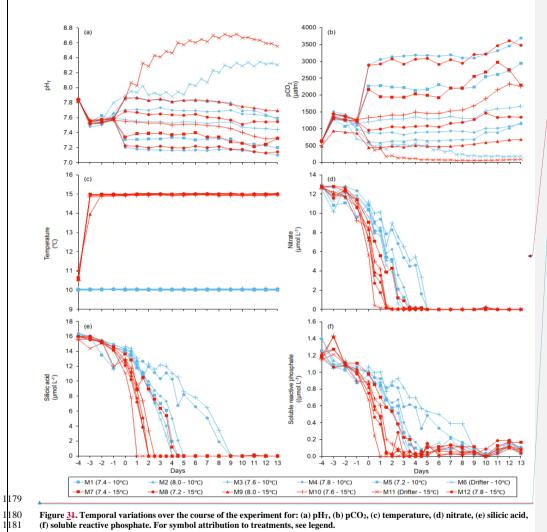


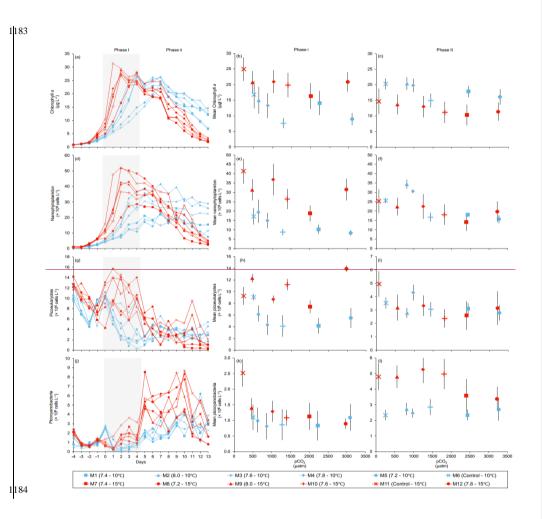
Figure 2. Changes in incident photosynthetic active radiation (PAR) at the top of the mesocosms level during the experiment as measurement by a Satlantic HyperOCR hyperspectral radiometer and integrated in the 400-700 nm range. Local sunrise and sunset times (EDT) are indicated with the corresponding days of the experiment.

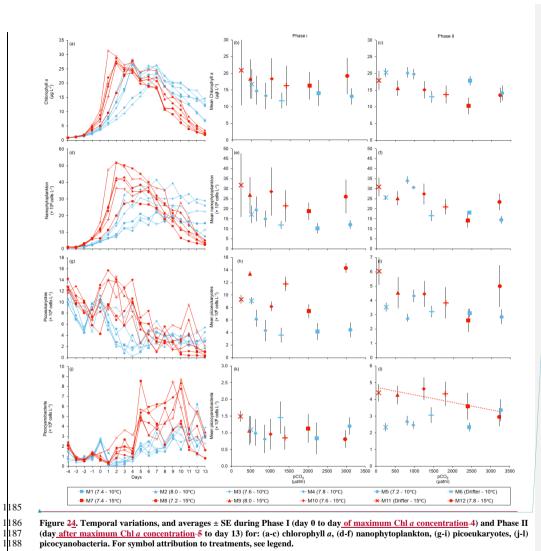




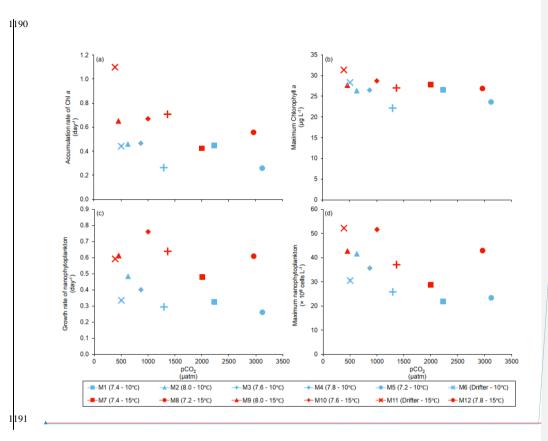
(f) soluble reactive phosphate. For symbol attribution to treatments, see legend.

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 $\textbf{Figure 24.} \ \textbf{Temporal variations, and averages} \pm \textbf{SE} \ \textbf{during Phase I (day 0 to day} \ \underline{\textbf{of maximum Chl}} \ \underline{\textbf{a concentration}} \ \underline{\textbf{4}}) \ \textbf{and Phase II}$ (day after maximum Chl a concentration 5 to day 13) for: (a-c) chlorophyll a, (d-f) nanophytoplankton, (g-i) picocukaryotes, (j-l) picocyanobacteria. For symbol attribution to treatments, see legend.



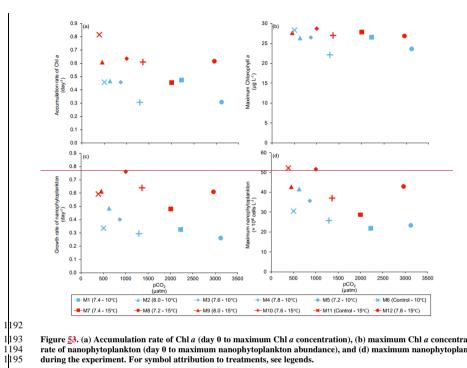
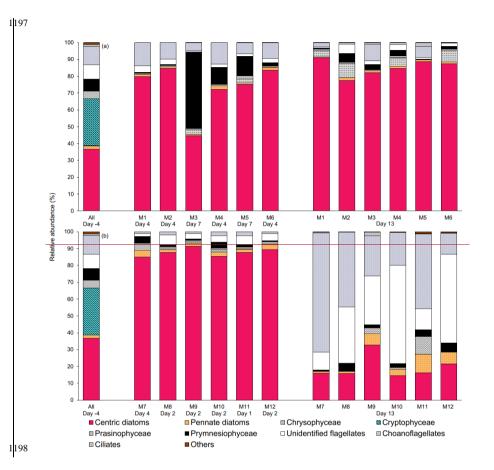


Figure 53. (a) Accumulation rate of Chl a (day 0 to maximum Chl a concentration), (b) maximum Chl a concentrations, (c) growth rate of nanophytoplankton (day 0 to maximum nanophytoplankton abundance), and (d) maximum nanophytoplankton abundance during the experiment. For symbol attribution to treatments, see legends.





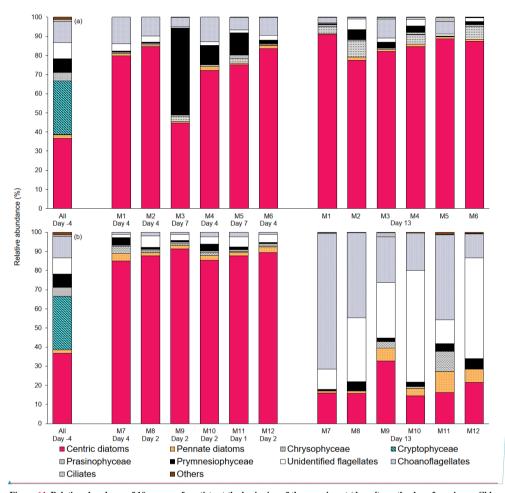
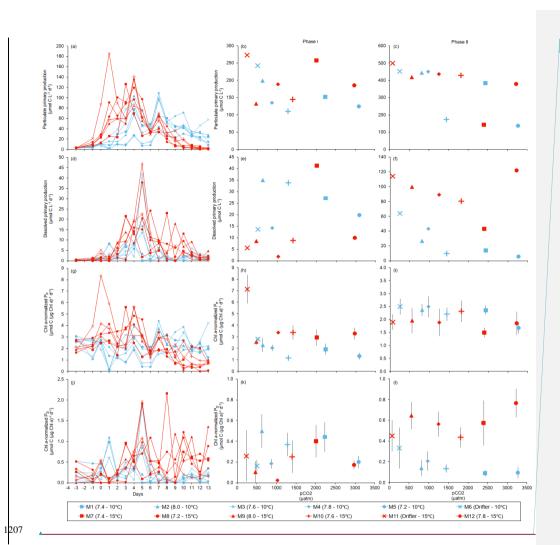


Figure 64. Relative abundance of 10 groups of protists at the beginning of the experiment (day -4), on the day of maximum Chl a concentrations in each mesocosm, and at the end of the experiment (day 13) for (a) 10 °C and (b) 15 °C mesocosms. The group « others » include dinoflagellates, Chlorophyceae, Dictyochophyceae, Euglenophyceae, heterotrophic groups, and unidentified cells. Each bar plot represents a mesocosm at a given time. The bar plot on day -4 represents the initial community assemblage before temperature manipulation and acidification, and is therefore the same for each temperature treatment. For symbol attribution to treatments, see legend.



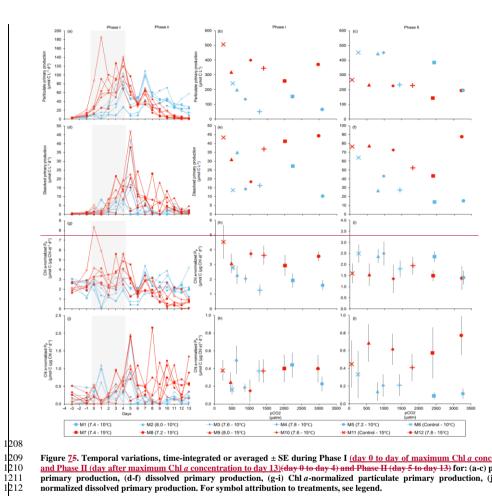
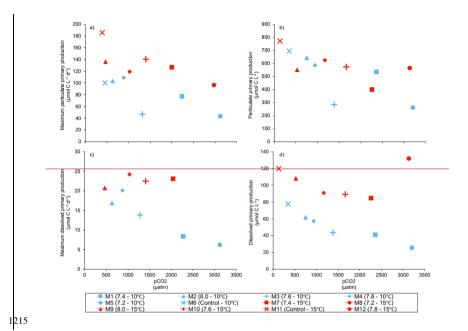


Figure 75. Temporal variations, time-integrated or averaged \pm SE during Phase I (day 0 to day of maximum Chl a concentration and Phase II (day after maximum Chl a concentration to day 13)(day 0 to day 4) and Phase II (day 5 to day 13) for: (a-c) particulate primary production, (d-f) dissolved primary production, (g-i) Chl a-normalized particulate primary production, (j-i) Chl anormalized dissolved primary production. For symbol attribution to treatments, see legend.





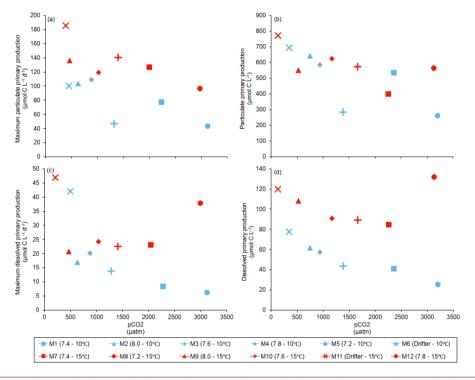


Figure 86. (a) Maximum particulate primary production, (b) time-integrated particulate primary production (c) maximum dissolved primary production, and (d) time-integrated dissolved primary production over the full course of the experiment (day 0 to day 13). For symbol attribution to treatments, see legend.

<u> </u>		Phas	se I	Pha	se II	Day 0-13	
Mesocosm	Day of	pH _T	pCO ₂ (µatm)	pH _T	pCO ₂	Temperature (°C)	-
<u>M1</u> (7.4 – 10 °C)	4	7.32 ± 0.01	2231 <u>.</u> ± 25	7.28 ± 0.02	2437 <u>.</u> ±92	10.06 ± 0.01	4
<u>M2</u> (8.0 – 10 °C)	<u>4</u>	7.84 <u>± 0.01</u>	628 ± 16	7.74 ± 0.03	<u>814 ± 65</u>	10.00 ± 0.01	4
<u>M3</u> (7.6 − 10 °C)	7	7.54 <u>± 0.01</u>	1294 <u>± 18</u>	7.48 ± 0.02	1503 <u>± 64</u>	10.07 ± 0.01	4
<u>M4</u> (7.8 – 10 °C)	<u>4</u>	7.71 <u>± 0.01</u>	868 ± 13	7.66 ± 0.01	976 <u>± 29</u>	10.04 ± 0.01	4
<u>M5</u> (7.2 − 10 °C)	7	7.17 <u>.±</u> 0.01	3122 ± 35	7.15 ± 0.01	3315 ± 94	10.03 ± 0.01	4
<u>M6</u> (Drifter – 10 °C)	<u>4</u>	7.93 <u>± 0.01</u>	503 ± 15	8.22 ± 0.03	251,±25	10.02 ± 0.01	4
<u>M7</u> (7.4 – 15 °C)	<u>4</u>	7.38 ± 0.01	2004 <u>± 44</u>	7.31 ± 0.02	2399 <u>,</u> ± 120	15.00 ± 0.01	-
<u>M8</u> (7.2 – 15 °C)	2	7.21 <u>± 0.01</u>	2961 <u>± 58</u>	7.18 ± 0.01	3179 <u>± 74</u>	15.01 ± 0.01	
<u>M9</u> (8.0 – 15 °C)	2	7.85 ± 0.01	454 ± 13	7.79 ± 0.02	545 ± 25	15.03 ± 0.01	
<u>M10</u> (7.6 – 15 °C)	2	7.54 ± 0.01	1364 ± 22	7.44 ± 0.02	1746 ± 106	14.94 ± 0.01	
<u>M11</u> (Drifter – 15 °C)	1	8.07 <u>± 0.01</u>	388 ± 90	8.59 ± 0.02	<u>84 ± 7</u>	14.96 ± 0.02	
<u>M12</u> (7.8 – 15 °C)	2	7.67 <u>, ± 0.01</u>	1001 <u>±31</u>	7.59 ± 0.01	<u>1215,44±</u>	14.98 ± 0.02	•

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Response Variable	Factor	df	t-value	p-value
	<u>Temperature</u> Temperat	00	2.004 2.3	0.0800.0
	ure	<u>8</u> 8	07	50*
Mean Chl a concentration	<u>pCO₂pCO₂-(10 °C)</u>	<u>8</u> 4	<u>-0.464</u> - <u>1.362</u>	<u>0.655</u> 0.2 45
$(\mu g L^{-1})$	<u>pCO</u> 2 x		0.244-	0.813 0.2
	TemperaturepCO ₂ (15 °C)	<u>8</u> 4	1.263	5.615 6.2
	Temperature Temperat	<u>8</u> 8	2.725 <mark>2.9</mark> 80	0.026*0.
Mean nanophytoplankton abundance (× 106 cells L-1)	pCO ₂ (10°C)pCO ₂	<u>4</u> 4	-2.285- 2.729	0.0840.0 53
(4.16 cells 2)	pCO ₂ (15°C)pCO ₂ (15°C)	<u>4</u> 4	-1.191- 1.231	0.2990.2 86
	<u>Temperature</u> Temperat	<u>8</u> 8		0.322 0.2 81
Mean picoeukaryote abundance	pCO ₂ pCO ₂	<u>8</u> 8	-1.159 -1.070	0.2800.3 16
$(\times 10^6 \text{ cells L}^{-1})$	$\frac{\text{pCO}_{\underline{2}}}{\text{TemperaturepCO}_{\underline{2}} \longrightarrow \times}$	<u>8</u> 8	1.125 _{1.0} 85	0.2930.3 09
	Temperature			
	<u>Temperature</u> Temperat	<u>8</u> 8		0.3990.0
Mean picocyanobacteria abundance	ure		66	15*
$(\times 10^6 \text{ cells L}^{-1})$	$\underline{\text{pCO}}_{\underline{2}}\underline{\text{pCO}}_{\underline{2}}\underline{\text{(10 °C)}}$	<u>8</u> 4	0.991 _{0.1}	0.351 _{0.9}

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	pCO ₂ x TemperaturepCO ₂ (15 °C)	<u>8</u> 4	-1.166- 2.268	0.277 _{0.0} 86
	Temperature Temperat	<u>8</u> 8	<u>-</u> 0.124 <mark>2.6</mark> 90	0.905 0.0 28*
Particulate primary production (μmol C L ⁻¹)	<u>pCO₂pCO₂ (10 °C)</u>	<u>8</u> 4	<u>-1.011</u> - <u>1.617</u>	<u>0.342</u> 0.1 81
	$\frac{\text{pCO}_{\underline{2}}}{\text{TemperaturepCO}_{\underline{2}}} \times \frac{\text{X}}{\text{(15 °C)}}$	<u>8</u> 4	0.867- 0.992	<u>0.411</u> 0.3 78
	Temperature Temperat	<u>8</u> 8	<u>1.429</u> 0.7 56	<u>0.191</u> 0.4 72
Dissolved primary production (μmolC-L-¹)	pCO ₂ pCO ₂	<u>8</u> 8	<u>-0.569</u> - 0.901	<u>0.585</u> 0.3 94
	$\frac{\text{pCO}_2}{\text{Temperature} \text{pCO}_2} - \times \\ \text{Temperature}$	<u>8</u> 8	<u>0.723</u> 0.9 56	<u>0.490</u> 0.3 67
	Temperature Temperat	<u>8</u> 8	1.689 2.5 92	0.130 0.0 32*
Chl a-normalized particulate primary production	<u>pCO₂ pCO₂ (10 °C)</u>	<u>8</u> 4	<u>0.107</u> - 1.467	<u>0.918</u> 0.2 16
(μmol C (μg Chl a) ⁻¹ d ⁻¹)	$\frac{\text{pCO}_{\underline{2}}}{\text{Temperature}_{\overline{P}}\text{CO}_{\underline{2}}} \times \frac{\text{X}}{\text{(15 °C)}}$	<u>8</u> 4	<u>-0.381</u> - <u>0.840</u>	<u>0.713</u> 0.4 48
Chl a-normalized dissolved primary production	Temperature Temperat	<u>8</u> 8	<u>-1.046</u> - <u>0.350</u>	<u>0.326</u> 0.7 35
(μ mol C (μ g Chl a) ⁻¹ d ⁻¹)	pCO ₂ pCO ₂	<u>8</u> 8	<u>-0.381</u> - 0.397	<u>0.713</u> 0.7 02

	$\frac{pCO_2x}{TemperaturepCO_2 \times}$ $\frac{Temperature}{Temperature}$	<u>8</u> 8	0.4490.5 22	0.6650.6 16
1232				

Table 32. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂ and their interaction. Separate analysis with pCO₂ as a continuous factor were performed when temperature had a significant effect. Accumulation rate of Chl a (day 0 to maximum Chl a concentration), maximum Chl a concentration, growth rate of nanophytoplankton (day 0 to maximum nanophytoplankton abundance), and maximum nanophytoplankton abundance. Significant results are in bold. *p < 0.05.

Response Variable	Factor	df	t-value	p-value
Accumulation rate of Chl a	Temperature	8	2.679	0.028*
(day-1)	pCO ₂ (10 °C)	4	-1.476	0.214
(day)	pCO ₂ (15 °C) 4 -1.759 0.154 Temperature 8 1.305 0.228 pCO ₂ 8 -0.387 0.709	0.154		
M · Cli	Temperature	8	1.305	0.228
Maximum Chl a concentration	pCO_2	8	-0.387	0.709
(μg L ⁻¹)	$pCO_2 \times Temperature \\$	4 -1.759 8 1.305	0.983	
	Temperature	8	2.534	0.035*
Growth rate of nanophytoplankton	pCO ₂ (10 °C)	4	-0.882	0.403
(day ⁻¹)	pCO ₂ (15 °C)	4	0.601	0.564
W	Temperature	8	1.380	0.205
Maximum nanophytoplankton abundance	pCO_2	8	-0.735	0.484
$(\times 10^6 \text{ cells L}^{-1})$	$pCO_2 \times Temperature \\$	8	0.302	0.770



Table 43. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂, and their interaction during Phase II (day after maximum Chl a^5 to day 13). Separate analysis with pCO₂ as a continuous factor were performed when temperature had a significant effect. Chl a concentration, nanophytoplankton abundance, picoeukaryote abundance, picocyanobacteria abundance, particulate and dissolved primary production, and Chl a-normalized particulate and dissolved primary production. Significant results are in bold. *p < 0.05, **p < 0.01, ***p < 0.001.

Response Variable	Factor	df	t-value	p-value	
	<u>Temperature</u> Temperat	<u>88</u>	<u>-1.539</u> -	<u>0.162</u> 0.007	
	ure	<u>0</u> 0	3.600	**	
Mean Chl a concentration	<u>pCO₂ pCO₂ (10 °C)</u>	<u>8</u> 4	<u>0.733</u> - 2.72 4	<u>0.484</u> 0.073	
(μg L ⁻¹)	<u>рСО₂ х</u>		0.456		
	TemperaturepCO ₂	<u>8</u> 4	0.156- 1.263	<u>0.880</u> 0.275	
	(15 °C)		1.203		
	<u>Temperature</u> Temperat	88	<u>-0.528</u> -	0.612 0.181	
	ure	<u>0</u> 0	1.465	<u>0.612</u> 0.161	
Mean nanophytoplankton abundance	pCO ₂ pCO ₂	<u>8</u> 8	1.264- 1.539	<u>0.242</u> 0.162	
$(\times 10^6 \text{ cells L}^{-1})$	pCO ₂ x		0.5001		
	TemperaturepCO₂ ×	<u>8</u> 8	0.6991.	<u>0.505</u> 0.345	
	Temperature		003		
	<u>Temperature</u> Temperat	88	<u>1.628</u> 0.	0.142 0.577	
	ure		581	<u>0.142</u> 0.577	
Mean picoeukaryotes abundance	pCO ₂ pCO ₂	<u>8</u> 8	<u>0.226</u> 0. 294	<u>0.827</u> 0.776	
$(\times 10^6 \text{ cells L}^{-1})$	<u>pCO₂ x</u>		0.534		
	TemperaturepCO ₂	<u>8</u> 8	<u>-0.521</u> - 0.698	<u>0.617</u> 0.505	
	×Temperature		0.050		
	<u>Temperature</u> Temperat	88	<u>5.983</u> 6.	<u><0,001***</u> <	
	ure	<u>0</u> 0	107	0.001***	
Mean picocyanobacteria abundance	pCO ₂ (10°C)pCO ₂	<u>4</u> 4	<u>1.480</u> 0.	<u>0.2130.709</u>	
$(\times 10^6 \text{ cells L}^{-1})$	(10 °C)	<u> </u>	401	<u>0.213</u> 0.709	
	pCO ₂ (15°C) _p CO ₂	44	<u>-3.051</u> -	<u>0.038*</u> 0.07	
	(15 °C)	44	2.347	9	

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	<u>Temperature</u> Temperat	88	<u>-0.015</u> -	<u>0.988</u> 0.012
Particulate primary production $(\mu mol \; C \; L^{\text{-}1})$	ure	<u></u> 0	2.248	坐
	<u>pCO₂ pCO₂ (10 °C)</u>	84	<u>-0.940</u> -	0.375 0.094
	<u></u>	_	2.186	
	<u>pCO₂ x</u>		0.460	460- 0.6580.075 390
	TemperaturepCO ₂	<u>8</u> 4		
	(15 °C)		2.390	
	<u>Temperature</u> Temperat		<u>1.894</u> 1.	
	ure	<u>8</u> 8	154	<u>0.095</u> 0.282
			-1.145-	
Dissolved primary production	pCO ₂ pCO ₂	<u>8</u> 8	1.701	<u>0.285</u> 0.127
$(\mu mol\ C\ L^{\text{-}1})$	pCO ₂ x			
	-	88	<u>0.847</u> 1.	0.422 0.208
	<u>Temperature</u> p CO₂ ×	<u>0</u> 0	369	<u>0.422</u> 0.208
	Temperature		2 200	0.0500.010
(Log) Chl a-normalized particulate	<u>Temperature</u> Temperat	<u>8</u> 8	-2.288-	0.0520.010
primary production	ure		3.387	<u>**</u>
(μmol C (μg Chl a) d-l)Chl a-	<u>pCO₂ pCO₂ (10 °C)</u>	<u>8</u> 4	<u>-1.491</u> -	<u>0.174</u> 0.090
normalized particulate primary			2.226	
production	pCO ₂ x		<u>1.105</u> -	
· (μmol C (μg Chl a)- ¹ d- ¹)	Temperature _P CO ₂	<u>8</u> 4	0.366	<u>0.301</u> 0.733
	(15 °C)			
(Log) Chl a-normalized dissolved	<u>Temperature</u> Temperat	-88	2.357 ₁ .	<u>0.046*</u> 0.07
primary production	ure	_	973	3
(μmol C (μg Chl a) d-1)Chl a-	pCO ₂ (10°C) pCO 2	48	<u>-2.573</u> -	0.062 0.103
normalized dissolved primary	<u></u>	_	1.838	
production	$pCO_2 (15^{\circ}C)pCO_2 \times$	48	<u>1.345</u> 1.	0.250 0.100
$(\mu \text{mol C} (\mu \text{g Chl } a)^{-1} \text{d}^{-1})$	Temperature	40	860	<u>0.230</u> 0.100

Table $\underline{54}$. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂ and their interaction. Separate analysis with pCO₂ as a continuous factor were performed when temperature had a significant effect. Maximum particulate and dissolved primary production, and time-integration over the full duration of the experiment (day 0 to day 13). Natural logarithm transformation is indicated in parentheses when necessary, significant results are in bold. *p < 0.05, **p < 0.01.

Response Variable	Factor	<u>d</u> Ðf	t-value	p-value
	Temperature	8	2.466	0.039*
Maximum particulate primary production (μmol C L ⁻¹ d ⁻¹)	pCO ₂ (10 °C)	4	-2.328	0.080
(pinor e E · C ·)	pCO ₂ (15 °C)	4	-2.394	0.075
	Temperature	8	-0.055	0.958
Time-integrated particulate primary production $(\mu \text{mol } C \text{ L}^{-1} \text{ d}^{-1})$	pCO ₂ (10 °C)	4	-1.300	0.230
(pinor e E · C ·)	pCO ₂ (15 °C)	4	0.801	0.446
	Temperature	8	-0.659	0.528
(Log) Maximum dissolved primary production (µmol C L ⁻¹)	pCO_2	8	-3.342	0.010**
(pinor C L)	$pCO_2 \times Temperature \\$	8	2.858	0.021*
	Temperature	8	1.687	0.130
Time-integrated dissolved primary production (µmol C L-1)	pCO_2	8	-2.153	0.063
(Amore 2)	$pCO_2 \times Temperature \\$	8	1.880	0.097

