

Experimental assessment of the sensitivity of an estuarine phytoplankton fall bloom to acidification and warming

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Abstract. We investigated the combined effect of ocean acidification and warming on the dynamics of the phytoplankton fall bloom in the Lower St. Lawrence Estuary (LSLE), Canada. Twelve 2600 L mesocosms were set to initially cover a wide range of pH_T (pH on the total proton scale) from 8.0 to 7.2 corresponding to a range of pCO₂ from 440 to 2900 µatm, and two temperatures (in situ and +5 °C). The 13-day experiment captured the development and decline of a nanophytoplankton bloom dominated by the chain-forming diatom *Skeletonema costatum*. During the development phase of the bloom, increasing pCO₂ influenced neither the magnitude nor the net growth rate of the nanophytoplankton bloom whereas increasing the temperature by 5 °C stimulated the chlorophyll *a* (Chl *a*) growth rate and maximal particulate primary production (P_P) by 76 % and 63 %, respectively. During the declining phase of the bloom, warming accelerated the loss of diatom cells, paralleled by a gradual decrease in the abundance of photosynthetic picoeukaryotes and a bloom of picocyanobacteria. Increasing pCO₂ and warming did not influence the abundance of picoeukaryotes while picocyanobacteria abundance was reduced by the increase in pCO₂ when combined with warming in the latter phase of the experiment. Over the full duration of the experiment, the time-integrated net primary production was not significantly affected by the pCO₂ treatments or warming. Overall, our results suggest that warming, rather than acidification, is more likely to alter phytoplankton autumnal bloom development in the LSLÉ in the decades to come. Future studies examining a broader gradient of temperatures should be conducted over a larger seasonal window in order to better constrain the potential effect of warming on the development of blooms in the LSLÉ and its impact on the fate of primary production.

1. Introduction

Anthropogenic emissions have increased atmospheric carbon dioxide (CO₂) concentrations from their pre-industrial value of 280 to 412 ppm in 2017, and concentrations of 850–1370 ppm are expected by the end of the century under the business-as-usual scenario RCP 8.5 (IPCC, 2013). The global ocean has already absorbed about 28 % of these anthropogenic CO₂ emissions (Le Quéré et al., 2015), leading to a global pH decrease of 0.11 units (Gattuso et al., 2015), a phenomenon known

32 as Ocean Acidification (OA). The surface ocean pH is expected to decrease by an additional 0.3–0.4 units under the RCP 8.5
33 scenario by 2100, and as much as 0.8 units by 2300 (Caldeira and Wickett, 2005; Doney et al., 2009; Feely et al., 2009). The
34 accumulation of anthropogenic CO₂ in the atmosphere also results in an increase in the Earth's heat content that is primarily
35 absorbed by the ocean (Wijffels et al., 2016), leading to an expected rise of sea surface temperatures of 3 to 5 °C by 2100
36 (IPCC, 2013). Whereas the effect of increasing atmospheric CO₂ partial pressures (pCO₂) on ocean chemistry is relatively well
37 documented, the potential impacts of OA on marine organisms and how their response to OA will be modulated by the
38 concurrent warming of the ocean surface waters are still the subject of much debate (Boyd and Hutchins, 2012; Gattuso et al.,
39 2013).

40 Over the last decade, there has been increasing interest in the potential effects of OA on marine organisms (Kroeker et al.,
41 2013). The first experiments were primarily conducted on single phytoplankton species (reviewed in Riebesell and Tortell,
42 2011), but subsequent mesocosm experiments highlighted the impact of OA on the structure and productivity of complex
43 plankton assemblages (Riebesell et al., 2007, 2013). Due to their widely different initial and experimental conditions, these
44 ecosystem-level experiments generated contrasting results (Schulz et al., 2017) but some general patterns nevertheless
45 emerged. For example, diatoms generally benefit from higher pCO₂ through stimulated photosynthesis and growth rates since
46 the increase in CO₂ concentrations compensates for the low affinity of RubisCO towards CO₂ (Giordano et al., 2005; Gao and
47 Campbell, 2014). Although most phytoplankton species have developed carbon concentration mechanisms (CCM) to
48 compensate for the low affinity of RubisCO towards CO₂, CCM efficiencies differ between taxa, rendering predictions of the
49 impact of a CO₂ rise on the downregulation of CCM rather difficult (Raven et al., 2014). For example, some studies
50 unexpectedly reported no significant or very modest stimulation of primary production under elevated CO₂ concentrations
51 (Engel et al., 2005; Eberlein et al., 2017). OA can ultimately affect the structure of phytoplankton assemblages. Small cells
52 such as photosynthetic picoeukaryotes can benefit directly from an increase in pCO₂ as CO₂ can passively diffuse through their
53 boundary layer (Beardall et al., 2014), and the smallest organisms within this group could benefit most from the increase
54 (Brussaard et al., 2013). Accordingly, OA experiments have typically favoured smaller phytoplankton cells (Yoshimura et al.,
55 2010; Brussaard et al., 2013; Morán et al., 2015), although the proliferation of larger cells has also been reported (Tortell et
56 al., 2002). Hence, generic predictions of phytoplankton community responses to OA are challenging.

57 Few recent studies have investigated the combined effects of OA and warming on natural phytoplankton assemblages (Hare
58 et al., 2007; Feng et al., 2009; Maudgandre et al., 2015; Paul et al., 2015, 2016). Laboratory experiments have shown that OA
59 and warming could together increase photosynthetic rates, but at the expense of species richness, the reduction of diversity
60 predominantly imputable to warming (Tatters et al., 2013). Results of an experiment conducted with a natural planktonic
61 community from the Mediterranean Sea showed no effect of a combined warming and decrease in pH on primary production,
62 but higher picocyanobacteria abundances were observed in the warmer treatment (Maudgandre et al., 2015). Shipboard
63 microcosm incubations conducted in the northern South China Sea displayed higher phytoplankton biomass, daytime primary
64 productivity and dark community respiration under warmer conditions, but these positive responses were cancelled at low pH
65 (Gao et al., 2017). In contrast, a mesocosm experiment carried out with a fall planktonic community from the western Baltic

66 Sea led to a decrease in phytoplankton biomass under warming, but combined warming and increased pCO₂ led to an increase
67 in biomass (Sommer et al., 2015). Results from experiments where the impacts of pCO₂ and temperature are investigated
68 individually may be misleading as multiple stressors can interact antagonistically or synergistically, sometimes in a nonlinear,
69 unpredictable fashion (Todgham and Stillman, 2013; Boyd et al., 2015; Riebesell and Gattuso, 2015; Gunderson et al., 2016).
70 The Lower St. Lawrence Estuary (LSLE) is a large (9350 km²) segment of the greater St. Lawrence Estuary (d'Anglejan,
71 1990). From June to September, the LSLE is characterized by a dynamic succession in the phytoplankton community, mostly
72 driven by changes in light and nutrient availability through variations in the intensity of vertical mixing (Levasseur et al.,
73 1984). The spring and fall blooms are mostly comprised of diatoms, with simultaneous nitrate and silicic acid exhaustion
74 ultimately limiting primary production (Levasseur et al., 1987; Roy et al., 1996). How OA and warming may affect these
75 blooms and primary production has never been investigated in the LSLE. The OA problem is complex in estuarine and coastal
76 waters where freshwater runoff, tidal mixing, and high biological activity contribute to variations in pCO₂ and pH on different
77 time scales (Duarte et al., 2013). The surface mixed-layer pCO₂ in the LSLE varies spatially from 139 to 548 µatm and is
78 strongly modulated by biological productivity (Dinauer and Mucci, 2017). Surface pH_T has been shown to vary from 7.85 to
79 7.93 in a single tidal cycle in the LSLE, nearly as much as the world's oceans have experienced in response to anthropogenic
80 CO₂ uptake over the last century (Caldeira and Wickett, 2005; Mucci et al., 2017).
81 The main objective of this study was to experimentally assess the sensitivity of the LSLE phytoplankton fall assemblage to a
82 large pCO₂ gradient at two temperatures (in situ and +5 °C). Whether lower trophic-level microorganisms thriving in a highly
83 variable environment will show higher resistance or resilience to future anthropogenic forcings is still a matter of speculation.

84 **2. Material and methods**

85 **2.1 Mesocosm setup**

86 The mesocosm system consists of two thermostated full-size ship containers each holding six 2600 L mesocosms (Aquabiotech
87 Inc., Québec, Canada). The mesocosms are cylindrical (2.67 m × 1.40 m) with a cone-shaped bottom within which mixing is
88 achieved using a propeller fixed near the top (Fig. 1). The mesocosms exhibit opaque walls and all lie on the same plane level
89 as not to shade each other. Light penetrates the mesocosms only through a sealed Plexiglas circular cover at their uppermost
90 part. The cover allows the transmission of 90 % of photosynthetically active radiation (PAR; 400–700 nm), 85–90 % of UVA
91 (315–400 nm), and 50–85 % of solar UVB (280–315 nm). The mesocosms are equipped with individual, independent
92 temperature probes (AQBT-Temperature sensor, accuracy ± 0.2 °C). Temperature in the mesocosms was measured every 15
93 minutes during the experiment, and the control system triggered either a resistance heater (Process Technology TTA1.8215)
94 located near the middle of the mesocosm or a pump-activated glycol refrigeration system to maintain the set temperature. The
95 pH in each mesocosm was monitored every 15 minutes using Hach® PD1P1 probes (± 0.02 pH units) connected to Hach®
96 SC200 controllers, and positive deviations from the target values activated peristaltic pumps linked to a reservoir of artificial
97 seawater equilibrated with pure CO₂ prior to the onset of the experiment. This system maintained the pH of the seawater in the

98 mesocosms within ± 0.02 pH units of the targeted values by lowering the pH during autotrophic growth but could not increase
99 the pH during bloom senescence when the $p\text{CO}_2$ rose and pH decreased.

100 **2.2 Setting**

101 The water was collected at 5 m depth near Rimouski harbour ($48^\circ 28' 39.9''$ N, $68^\circ 31' 03.0''$ W) on the 27th of September 2014
102 (indicated as day -5 hereafter), and the experiment lasted until the 15th of October 2014 (day 13). In situ conditions were:
103 salinity = 26.52, temperature = 10 °C, nitrate (NO_3^-) = $12.8 \pm 0.6 \mu\text{mol L}^{-1}$, silicic acid (Si(OH)_4) = $16 \pm 2 \mu\text{mol L}^{-1}$, and
104 soluble reactive phosphate (SRP) = $1.4 \pm 0.3 \mu\text{mol L}^{-1}$. On day -5, the water was filtered through a 250 μm mesh while
105 simultaneously filling the 12 mesocosm tanks by gravity with a custom made ‘octopus’ tubing system. The initial $p\text{CO}_2$ was
106 $623 \pm 7 \mu\text{atm}$ and the in situ temperature of 10 °C was maintained in the twelve mesocosms for the first 24 h (day -4). After
107 that period, the six mesocosms in one container were maintained at 10 °C while temperature was gradually increased to 15 °C
108 over day -3 in the six mesocosms of the other container. To avoid subjecting the planktonic communities to excessive stress
109 due to sudden changes in temperature and pH while setting the experiment, the mesocosms were left to acclimatize on day -2
110 before acidification was carried out over day -1. One mesocosm from each temperature-controlled container was not pH-
111 controlled to assess the community response to the freely fluctuating pH. These two mesocosms were labelled “Drifters” as
112 the initial in situ pH was allowed to fluctuate over time with the development of the phytoplankton bloom. The other
113 mesocosms were set to cover a range of pH_T of ~ 8.0 to ~ 7.2 corresponding to a $p\text{CO}_2$ gradient of ~ 440 to $\sim 2900 \mu\text{atm}$ after
114 acidification was carried out. To attain initial targeted pH, CO_2 -saturated artificial seawater was added to the mesocosms that
115 needed a pH lowering while mesocosms M2 (8.0), M4 (7.8), M6 (Drifter), M9 (8.0), M11 (Drifter) and M12 (7.8) were openly
116 mixed to allow the degassing of the supersaturated CO_2 . Once the mesocosms had reached their target pH, the automatic
117 system controlled the sporadic addition of CO_2 -saturated water to refrain the pH from rising. Only the Drifters were not
118 controlled throughout the experiment. Incident light was variable during our experiment, with only few sunny days (Fig. 2).

119 **2.3 Seawater analysis**

120 The mesocosms were sampled between 05:00 and 08:00 a.m. every day. Seawater for carbonate chemistry, nutrients, and
121 primary production were collected directly from the mesocosms as close to sunrise as possible. Seawater was also collected in
122 20 L carboys for the determination of chlorophyll *a* (Chl *a*), taxonomy, and other variables. Total amount of volume sampled
123 every day was 24 L or less. Samples for salinity were taken from the artificial seawater tanks and in the mesocosms on day -
124 3, 3 and 13. The samples were collected in 250 mL plastic bottles and stored in the dark until analysis was performed using a
125 Guildline Autosal 8400B Salinometer during the following months.

126 **2.3.1 Carbonate chemistry**

127 Carbonate chemistry parameters were determined using methods described in Mucci et al. (2017). Briefly, water samples for
128 pH (every day) and total alkalinity (TA, every 3–4 days) measurements were, respectively, transferred from the mesocosms to

129 125 mL plastic bottles without headspace and 250 mL glass bottles. A few crystals of HgCl_2 were added to the glass bottles
130 before sealing them with a ground-glass stopper and Apiezon[®] Type-M high-vacuum grease. The pH was determined within
131 hours of collection, after thermal equilibration at 25.0 ± 0.1 °C, using a Hewlett-Packard UV-Visible diode array
132 spectrophotometer (HP-8453A) and a 5 cm quartz cell with phenol red (PR; Robert-Baldo et al., 1985) and *m*-cresol purple
133 (mCP; Clayton and Byrne, 1993) as indicators. Measurements were carried out at the wavelength of maximum absorbance of
134 the protonated (HL) and deprotonated (L) indicators. Comparable measurements were carried out using a TRIS buffer prepared
135 at a practical salinity of 25 before and after each set of daily measurements (Millero, 1986).
136 The pH on the total proton concentration scale (pH_T) of the buffer solutions and samples at 25 °C was calculated according to
137 the equation of Byrne (1987), using the salinity of each sample and the HSO_4^- association constants given by Dickson (1990).
138 The TA was determined on site within one day of sampling by open-cell automated potentiometric titration (Titralab 865,
139 Radiometer[®]) with a pH combination electrode (pHC2001, Red Rod[®]) and a dilute (0.025N) HCl titrant solution. The titrant
140 was calibrated using Certified Reference Materials (CRM Batch#94, provided by A. G. Dickson, Scripps Institute of
141 Oceanography, La Jolla, USA). The average relative error, based on the average relative standard deviation on replicate
142 standard and sample analyses, was better than 0.15 %. The carbonate chemistry parameters at in situ temperature were then
143 calculated using the computed pH_T at 25 °C in combination with the measured TA using CO_2SYS (Pierrot et al., 2006) and
144 the carbonic acid dissociation constants of Cai and Wang (1998).

145 **2.3.3 Nutrients**

146 Samples for NO_3^- , $\text{Si}(\text{OH})_4$, and SRP analyses were collected directly from the mesocosms every day, filtered through
147 Whatman GF/F filters and stored at -20 °C in acid washed polyethylene tubes until analysis by a Bran and Luebbe Autoanalyzer
148 III using the colorimetric methods described by Hansen and Koroleff (2007). The analytical detection limit was $0.03 \mu\text{mol L}^{-1}$
149 ¹ for NO_3^- plus nitrite (NO_2^-), $0.02 \mu\text{mol L}^{-1}$ for NO_2^- , $0.1 \mu\text{mol L}^{-1}$ for $\text{Si}(\text{OH})_4$, and $0.05 \mu\text{mol L}^{-1}$ for SRP.

150 **2.3.4 Plankton biomass, composition and enumeration**

151 Duplicate subsamples (100 mL) for Chl *a* determination were filtered onto Whatman GF/F filters. Chl *a* concentrations were
152 measured using a 10-AU Turner Designs fluorometer, following a 24 h extraction in 90 % acetone at 4 °C in the dark without
153 grinding (acidification method: Parsons et al., 1984). The analytical detection limit for Chl *a* was $0.05 \mu\text{g L}^{-1}$.
154 Pico- (0.2–2 μm) and nanophytoplankton (2–20 μm) cell abundances were determined daily by flow cytometry. Sterile
155 cryogenic polypropylene vials were filled with 4.95 mL of seawater to which 50 μL of glutaraldehyde Grade I (final
156 concentration = 0.1 %, Sigma Aldrich; Marie et al., 2005) were added. Duplicate samples were flash frozen in liquid nitrogen
157 after standing 15 minutes at room temperature in the dark. These samples were then stored at -80 °C until analysis. After
158 thawing to ambient temperature, samples were analyzed using a FACS Calibur flow cytometer (Becton Dickinson) equipped
159 with a 488 nm argon laser. The abundances of nanophytoplankton and picophytoplankton, which includes photosynthetic

160 picoeukaryotes and picocyanobacteria, were determined by their autofluorescence characteristics and size (Marie et al., 2005).

161 The biomass accumulation and nanophytoplankton growth rates were calculated by the following equation:

$$162 \mu = \ln(N_2/N_1) / (t_2 - t_1), \quad (1)$$

163 where N_1 and N_2 are the biomass or cell concentrations at given times t_1 and t_2 , respectively.

164 Microscopic identification and enumeration for eukaryotic cells larger than 2 μm was conducted on samples taken from each
165 mesocosm on three days: day -4, the day when maximum Chl *a* was attained in each mesocosm, and day 13. Samples of
166 250 mL were collected and preserved with acidic Lugol solution (Parsons et al., 1984), then stored in the dark until analysis.

167 Cell identification was carried out at the lowest possible taxonomic rank using an inverted microscope (Zeiss Axiovert 10) in
168 accordance with Lund et al. (1958). The main taxonomic references used to identify the phytoplankton were Tomas (1997)
169 and Bérard-Therriault et al. (1999).

170 **2.3.5 Primary production**

171 Primary production was determined daily using the ^{14}C -fixation incubation method (Knap et al., 1996; Ferland et al., 2011).
172 One clear and one dark 250 mL polycarbonate bottle were filled from each mesocosm at dawn and spiked with 250 μL of
173 $\text{NaH}^{14}\text{CO}_3$ (80 $\mu\text{Ci mL}^{-1}$). One hundred μL of 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) (0.02 mol L^{-1}) was added to
174 the dark bottles to prevent active fixation of ^{14}C by phytoplankton (Legendre et al., 1983). The total amount of radioisotope in
175 each bottle was determined by immediately pipetting 50 μL subsamples into a 20 mL scintillation vial containing 10 mL of
176 scintillation cocktail (EcolumeTM) and 50 μL of ethanolamine (Sigma). Bottles were placed in separate incubators, at either
177 10 $^\circ\text{C}$ or 15 $^\circ\text{C}$, under reduced (30 %) natural light for 24 h, which corresponds to the light transmittance at mid-mesocosm
178 depth.

179 At the end of the incubation periods, 3 mL were transferred to a scintillation vial for determination of the total primary
180 production (P_T), 3 mL were filtered through a syringe filter (GD/X 0.7 μm) to estimate daily photosynthetic carbon fixation
181 released in the dissolved organic carbon pool (P_D). The remaining volume was filtered onto a Whatman GF/F filter to measure
182 the particulate primary production (P_P). Vials containing the P_T and P_D samples were acidified with 500 μL of HCl 6 N, allowed
183 to sit for 3 h under a fume hood, then neutralized with 500 μL of NaOH 6 N. The vials containing the filters were acidified
184 with 100 μL of 0.05 N HCl and left to fume for 12 h. Fifteen mL of scintillation cocktail were added to the vials and they were
185 stored pending analysis using a Tri-Carb 4910TR liquid scintillation counter (PerkinElmer). Rates of carbon fixation into
186 particulate and dissolved organic matter were calculated according to Knap et al. (1996) using the dissolved inorganic carbon
187 concentration computed for each mesocosm at the beginning of the daily incubations and multiplied by a factor of 1.05 to
188 correct for the lower uptake of ^{14}C compared to ^{12}C .

189 2.4 Statistical analysis

190 All statistical analyses were performed using R (nlme package). A general least squares (gls) model approach was used to test
191 the linear effects of the two treatments (temperature, pCO₂), and of their interactions on the measured variables (Paul et al.,
192 2016; Hussherr et al., 2017). The analysis was conducted independently on two different time periods: Phase I (day 0 to day
193 of maximum Chl *a* concentration) was calculated individually for each mesocosm, whereas Phase II (day after maximum Chl *a*
194 concentrations) corresponded to the declining phase of the bloom (Table 1). Averages (or time-integration in the case of
195 primary production) of the response variables were calculated separately over the two phases and were plotted against pCO₂.
196 Separate regressions were performed with pCO₂ as the continuous factor for each temperature when a temperature effect or
197 interaction with pCO₂ was detected in the gls model. Otherwise, the model included data from both temperatures and the
198 interaction with pCO₂. Normality of the residuals was determined using a Shapiro-Wilk test ($p > 0.05$) and data were
199 transformed (natural logarithm or square root) if required. As explained by Havenhand et al. (2010), the gradient approach,
200 instead of treatment replication, is particularly suitable when few experimental units are available such as in large volume
201 mesocosm experiments. In addition, squared Pearson's correlation coefficients (r^2) with a significance level of 0.05 were used
202 to evaluate correlations between key variables.

203 3. Results

204 3.1 Seawater chemistry

205 Water salinity was 26.52 ± 0.03 on day -4 in all mesocosms and remained constant throughout the experiment, averaging
206 26.54 ± 0.02 on day 13. The TA was practically invariant in the mesocosms, averaging $2057 \pm 2 \mu\text{mol kg}_{\text{sw}}^{-1}$ on day -4 and
207 $2058 \pm 2 \mu\text{mol kg}_{\text{sw}}^{-1}$ on day 13. Following the filling of the mesocosms, the pH_T in all mesocosms decreased from an average
208 of 7.84 to 7.53. Throughout the rest of the experiment after treatments were applied, the pH remained relatively stable in the
209 pH-controlled treatments, but decreased slightly during Phase II by an average of -0.14 ± 0.07 units relative to the target pH_T
210 (Fig. 3a). Given a constant TA, pH variations were accompanied by variations in pCO₂, from an average of $1340 \pm 150 \mu\text{atm}$
211 on day -3, and ranging from 564 to 2902 μatm at 10 °C, and from 363 to 2884 μatm at 15 °C on day 0 following the
212 acidification (Fig. 3b; Table 1). The pH_T in the Drifters (M6 and M11) increased from 7.896 and 7.862 on day 0 at 10 °C and
213 15 °C, respectively, to 8.307 and 8.554 on day 13, reflecting the balance between CO₂ uptake and metabolic CO₂ production
214 over the duration of the experiment. On the last day, pCO₂ in all mesocosms ranged from 186 to 3695 μatm at 10 °C, and from
215 90 to 3480 μatm at 15 °C. The temperature of the mesocosms in each container remained within ± 0.1 °C of the target
216 temperature throughout the experiment and averaged 10.04 ± 0.02 °C for mesocosms M1 through M6, and 15.0 ± 0.1 °C for
217 mesocosms M7 through M12 (Fig. 3c; Table 1).

218 **3.2 Dissolved inorganic nutrient concentrations**

219 Nutrient concentrations averaged $9.1 \pm 0.5 \mu\text{mol L}^{-1}$ for NO_3^- , $13.4 \pm 0.3 \mu\text{mol L}^{-1}$ for $\text{Si}(\text{OH})_4$, and $0.91 \pm 0.03 \mu\text{mol L}^{-1}$ for
220 SRP on day 0 (Fig. 3d, e, f). Within individual mesocosms, concentrations of nitrate, silicic acid and soluble reactive phosphate
221 displayed similar temporal patterns following the development of the phytoplankton bloom. Overall, NO_3^- depletion was
222 reached within 5 days in all mesocosms at 10°C , exception made of the Drifter which became nutrient-deplete by day 3.
223 Nutrient depletion was reached slightly earlier within the 15°C mesocosms, all of them displaying exhaustion within 3 days
224 of the experiment. Accordingly, bloom development and primary production within each mesocosm were eventually limited
225 by the supply in nutrients, irrespective of the temperature or pH treatment. Likewise, $\text{Si}(\text{OH})_4$ fell below the detection limit
226 between day 1 and 5 in all mesocosms except for those whose pH_T was set at 7.2 and 7.6 at 10°C (M5 and M3) and in which
227 $\text{Si}(\text{OH})_4$ depletion occurred on day 9. Variations in SRP concentrations followed closely those of NO_3^- in all mesocosms except
228 again for those set at pH 7.2 and 7.6 in which undetectable values were reached on day 9.

229 **3.3 Phytoplankton biomass**

230 Chl *a* concentrations were below $1 \mu\text{g L}^{-1}$ just after the filling of the mesocosms, and averaged $5.9 \pm 0.6 \mu\text{g L}^{-1}$ on day 0 (Fig.
231 4a). They then quickly increased to reach maximum concentrations around $27 \pm 2 \mu\text{g L}^{-1}$ on day 3 ± 2 , and decreased
232 progressively until the end of the experiment, reaching $11 \pm 1 \mu\text{g L}^{-1}$ and $2.4 \pm 0.2 \mu\text{g L}^{-1}$ at 10°C and 15°C on day 13. During
233 Phase I, results from the gls model show no significant relationships between the mean Chl *a* concentrations and pCO_2 ,
234 temperature, and the interaction of the two factors (Fig. 4b; Table 2). During this phase, the accumulation rate of Chl *a* was
235 positively affected by temperature, increasing by $\sim 76\%$, but was not affected by the pCO_2 gradient at either temperature (Fig.
236 5a; Table 3). The maximum Chl *a* concentrations reached during the bloom were not affected by the two treatments (Fig. 5b;
237 Table 3). During Phase II, we observed no significant effect of pCO_2 , temperature, and the interaction of those factors on the
238 mean Chl *a* concentrations following the depletion of NO_3^- (Fig. 4c; Table 4).

239 **3.4 Phytoplankton size-class**

240 Nanophytoplankton abundance varied from $8 \pm 1 \times 10^6 \text{ cells L}^{-1}$ on day 0 to an average maximum of $36 \pm 10 \times 10^6 \text{ cells L}^{-1}$ at
241 the peak of the bloom (Fig. 4d). At both temperatures, nanophytoplankton abundance increased until at least days 2 or 4 and
242 decreased or remained stable thereafter. The correlation between the nanophytoplankton abundance and Chl *a* ($r^2 = 0.75$,
243 $p < 0.001$, $\text{df} = 166$) suggests that this phytoplankton size class was responsible for most of the biomass build-up throughout
244 the experiment. As observed for the mean Chl *a* concentration, the mean abundance of nanophytoplankton was not
245 significantly affected by the pCO_2 gradient at the two temperatures investigated during Phase I, but showed higher values at
246 15°C ($26 \pm 2 \times 10^6 \text{ cells L}^{-1}$) than at 10°C ($14 \pm 1 \times 10^6 \text{ cells L}^{-1}$) (Fig. 4e; Table 2). Likewise, the growth rate of
247 nanophytoplankton during Phase I was not influenced by the pCO_2 gradient at the two temperatures but was significantly
248 higher in the warm treatment (Fig. 5c; Table 3). During Phase II, no relationship was found between the mean

249 nanophytoplankton abundance and the pCO₂ gradient, the temperature, and the pCO₂ × temperature interaction (Fig. 4f; Table
250 4).

251 Initial abundance of photosynthetic picoeukaryotes was $10 \pm 2 \times 10^6$ cells L⁻¹, accounting for more than 80 % of total plankton
252 cells in the 0.2–20 μm size fraction. The abundance of this plankton size fraction decreased slightly through Phase I and their
253 number remained relatively stable at $4 \pm 3 \times 10^6$ cells L⁻¹ throughout Phase II (Fig. 4g). We found no relationship between the
254 abundance of picoeukaryotes and the pCO₂ gradient at the two temperatures investigated during both Phases I and II, and no
255 temperature effect was observed either (Fig. 4h, i; Tables 2 and 4).

256 Picocyanobacteria exhibited a different pattern than the nanophytoplankton and picoeukaryotes (Fig. 4j). Their abundance was
257 initially low ($1.7 \pm 0.3 \times 10^6$ cells L⁻¹ on day 0), remained relatively stable during Phase I, and increased rapidly during Phase
258 II, accounting for ~50 % of the total picophytoplankton cell counts toward the end of the experiment. During Phase I, the mean
259 picocyanobacteria abundance was not influenced by the pCO₂ gradient or temperature (Fig. 4k; Table 2). During Phase II, the
260 mean picocyanobacteria abundance was not significantly affected by pCO₂ at in situ temperature. However, mean
261 picocyanobacteria were higher at 15 °C, with the pCO₂ gradient responsible for a ~33% reduction of picocyanobacteria
262 abundance from the Drifter to the more acidified treatment ($4.4 \pm 0.2 \times 10^6$ cells L⁻¹ vs. $3.0 \pm 0.3 \times 10^6$ cells L⁻¹) (Fig 4l; Table
263 4).

264 3.5 Phytoplankton taxonomy

265 The taxonomic composition of the planktonic assemblage larger than 2 μm was identical in all treatments at the beginning of
266 the experiment, and was mainly composed of the cosmopolitan chain-forming centric diatom *Skeletonema costatum* (*S.*
267 *costatum*) and the cryptophyte *Plagioselmis prolonga* var. *nordica* (Fig. 6). At the peak of the blooms (maximum Chl *a*
268 concentrations), the species composition did not vary between the pCO₂ treatments and between the two temperatures tested.
269 *S. costatum* was the dominant species in all mesocosms (70–90 % of the total number of eukaryotic cells), except for one
270 mesocosm (M3, pH 7.6 at 10 °C) where a mixed dominance of *Chrysochromulina* spp. (a prymnesiophyte of 2–5 μm) and *S.*
271 *costatum* was observed (Fig. 6a). *S. costatum* accounted for 80–90 % of the total eukaryotic cell counts in all mesocosms at
272 the end of the experiment carried out at 10 °C. At 15 °C, the composition of the assemblage had shifted toward a dominance
273 of unidentified flagellates and choanoflagellates (2–20 μm) in all mesocosms with these two groups accounting for 55–80 %
274 of the total cell counts while diatoms showed signs of loss of viability as indicated by the presence of empty frustules (Fig.
275 6b).

276 3.6 Primary production

277 P_P increased in all mesocosms during Phase I of the experiment, in parallel with the increase in Chl *a* (Fig. 7a). P_P maxima
278 were attained on days 3–4, except for the 15 °C Drifter (M11) where P_P peaked on day 1. We found no significant effect of
279 the pCO₂ gradient, temperature and the pCO₂ × temperature interaction on the time-integrated P_P during both Phases I and II
280 (Fig. 7b, c; Tables 2 and 4). Similarly, the absence of significant treatment effects remained when normalizing P_P per unit of

281 Chl *a* (Fig. 7g, h, i). Initial Chl *a*-normalized P_P values were $3.3 \pm 0.5 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ and reached maxima between
282 $3.7 \pm 0.3 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ and $5.7 \pm 0.6 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ at 10°C and 15°C , respectively. These values then
283 decreased to $2.2 \pm 0.6 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ and $0.9 \pm 0.2 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ on the last day of the experiment. During
284 Phase I, the mean Chl *a*-normalized P_P was not significantly affected by the $p\text{CO}_2$ gradient or warming, as observed for the
285 mean Chl *a* concentrations and time-integrated P_P over that phase (Fig. 7h; Table 2). During Phase II, the log of the mean
286 Chl *a*-normalized P_P was not significantly affected by the $p\text{CO}_2$ gradient, the temperature, or the interaction of these factors
287 (Fig. 7i; Table 4).

288 P_D was low at the beginning of the experiment, averaging $1.5 \pm 0.4 \mu\text{mol C L}^{-1} \text{d}^{-1}$, increased progressively during Phase I to
289 reach maximum values of $6\text{--}48 \mu\text{mol C L}^{-1} \text{d}^{-1}$ between days 4 and 8, and decreased thereafter (Fig. 7d). Time-integrated P_D
290 was not significantly affected by the $p\text{CO}_2$ gradient, the temperature, and the $p\text{CO}_2 \times$ temperature interaction during the two
291 phases (Fig. 7e, f; Tables 2 and 4). Chl *a*-normalized P_D was low on day 0, averaging $0.3 \pm 0.1 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$, reached
292 maximum values of $1.0 \pm 0.2 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ and $1.6 \pm 0.2 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ at 10°C and 15°C , then
293 respectively decreased to $0.17 \pm 0.05 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ and $0.6 \pm 0.2 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$ by the end of the
294 experiment (Fig. 7j). During Phase I, the mean Chl *a*-normalized P_D was affected neither by the $p\text{CO}_2$ gradient, the temperature,
295 nor by the interaction between those factors (Fig. 7k; Table 2). During Phase II, the log of the mean Chl *a*-normalized P_D was
296 not affected by $p\text{CO}_2$ at either temperature tested, but significantly increased with warming (Fig. 7l; Table 4).

297 Figure 6 shows the influence of the treatments on maximum P_P and P_D as well as on the time-integrated P_P and P_D over the full
298 length of the experiment. We found no effect of the $p\text{CO}_2$ gradient on the maximum P_P values at the two temperatures tested,
299 but warming increased the maximum P_P values from $66 \pm 13 \mu\text{mol C L}^{-1} \text{d}^{-1}$ to $126 \pm 8 \mu\text{mol C L}^{-1} \text{d}^{-1}$ (Fig. 8a; Table 5). The
300 time-integrated P_P over the full duration of the experiment was not affected by the $p\text{CO}_2$ gradient or the increase in temperature
301 (Fig. 8b; Table 5). The maximum P_D values were significantly affected by the treatments (Fig 8c; Table 5). Maximum P_D
302 decreased with increasing $p\text{CO}_2$ at in situ temperature but warming cancelled this effect (antagonistic effect). Nevertheless,
303 the time-integrated P_D over the whole experiment did not vary significantly between treatments, although a decreasing
304 tendency with increasing $p\text{CO}_2$ at 10°C and an increasing tendency with warming can be seen in Fig. 8d (Table 5).

305 4. Discussion

306 4.1 General characteristics of the bloom

307 The onset of the experiment was marked by an increase of $p\text{CO}_2$ on the day following the filling of the mesocosms. This
308 phenomenon often takes place at the beginning of such experiments when pumping tends to break phytoplankton cells and
309 larger debris into smaller ones. We attribute the rapid fluctuations in $p\text{CO}_2$ to the release of organic matter following the filling
310 of the mesocosms with a stimulating effect on heterotrophic respiration, and hence CO_2 production. Then, a phytoplankton
311 bloom, numerically dominated by the centric diatom *S. costatum*, took place in all mesocosms, regardless of treatments (Fig.
312 6). *S. costatum* is a common phytoplankton species in the St. Lawrence Estuary and in coastal waters (Kim et al., 2004; Starr

313 et al., 2004; Annane et al., 2015). The length of the experiment (13 days) allowed us to capture both the development and
314 declining phases of the bloom. The exponential growth phases lasted 1–4 days depending on the treatments, but maximal Chl *a*
315 concentrations were reached only after 7 days in two of the twelve mesocosms (Fig. 4a; Table 1). The suite of measurements
316 and statistical tests conducted did not provide any clues as to the underlying causes for the lower rates of biomass accumulation
317 measured in these two mesocosms. Since statistical analyses conducted with or without these two apparent outliers gave similar
318 results, they were not excluded from the analyses.

319 In situ nutrient conditions prior to the water collection were favourable for a bloom development. Based on previous studies,
320 in situ phytoplankton growth was probably limited by light due to water turbidity and vertical mixing at the time of water
321 collection (Levasseur et al. 1984). Grazing may also have played a role in keeping the in situ biomass of flagellates low prior
322 to our sampling. However, a natural diatom fall bloom was observed in the days following the water collection in the adjacent
323 region (Ferreira, pers. comm.). The increased stability within the mesocosms, combined with the reduction of the grazing
324 pressure (filtration on 250 μm) likely contributed to the fast accumulation of phytoplankton biomass. During the development
325 phase of the bloom, the concentration of all three monitored nutrients decreased, with NO_3^- and $\text{Si}(\text{OH})_4$ reaching undetectable
326 values. This nutrient co-depletion is consistent with results from previous studies suggesting a co-limitation of diatom blooms
327 by these two nutrients in the St. Lawrence Estuary (Levasseur et al., 1987, 1990). Variations in P_P roughly followed changes
328 in Chl *a*, and, as expected, the maximum Chl *a*-normalized P_P ($5 \pm 2 \mu\text{mol C} (\mu\text{g Chl } a)^{-1} \text{d}^{-1}$) was reached during the
329 exponential growth phase in all mesocosms. Decreases in total phytoplankton abundances and P_P followed the bloom peaks
330 and the timing of the NO_3^- and $\text{Si}(\text{OH})_4$ depletions. A clear succession in phytoplankton size classes characterized the
331 experiment. Nanophytoplankton cells were initially present in low abundance and became more numerous as the *S. costatum*
332 diatom bloom developed. The correlation ($r^2 = 0.83$, $p < 0.001$, $df=34$) between the abundance of nanophytoplankton and *S.*
333 *costatum* enumeration suggests that this cell size class can be used as a proxy of *S. costatum* counts in all mesocosms
334 throughout the experiment. Nanophytoplankton cells accounted for $79 \pm 7\%$ of total counts of cells $< 20 \mu\text{m}$ on the day of the
335 maximum Chl *a* concentration. Accordingly, nanophytoplankton exhibited the same temporal trend as Chl *a* concentrations.
336 During Phase II, nanophytoplankton abundances remained roughly stable at in situ temperature but decreased at 15°C towards
337 the end of the experiment. Photosynthetic picoeukaryotes were originally abundant and decreased throughout the experiment
338 whereas picocyanobacteria abundances increased during Phase II. This is a typical phytoplankton succession pattern for
339 temperate systems where an initial diatom bloom growing essentially on allochthonous nitrate gives way to smaller species
340 growing on regenerated forms of nitrogen (Taylor et al., 1993).

341 **4.2 Phase I (Diatom bloom development)**

342 Our results show no significant effect of increasing pCO_2 /decreasing pH on the mean abundance and net accumulation rate of
343 the diatom-dominated nanophytoplankton assemblage during the development of the bloom (Figs. 4e and 5c). These results
344 suggest that *S. costatum*, the species accounting for most of the biomass accumulation during the bloom, neither benefited
345 from the higher pCO_2 nor was negatively impacted by the lowering of pH. Assuming that *S. costatum* was also responsible for

346 most of the carbon fixation during the bloom development phase, the absence of effect on P_P and Chl *a*-normalized P_P following
347 increases in pCO_2 brings additional support to our conclusion. *S. costatum* operates a highly efficient CCM, minimizing the
348 potential benefits of thriving in high CO_2 waters (Trimborn et al., 2009). This may explain why the strain present in the LSLE
349 did not benefit from the higher pCO_2 conditions. Likewise, a mesocosm experiment conducted in the coastal North Sea showed
350 no significant effect of increasing pCO_2 on carbon fixation during the development of the spring diatom bloom (Eberlein et
351 al., 2017).

352 In addition to the aforementioned insensitivity to increasing pCO_2 , our results point towards a strong resistance of *S. costatum*
353 to severe pH decline. During our study, surprisingly constant rates of Chl *a* accumulation and nanophytoplankton growth (Fig.
354 5a, c), as well as maximum P_P (Fig. 8a), were measured during the development phase of the bloom over a range of pH_T
355 extending from 8.6 to 7.2 (Fig. 3a). In a recent effort to estimate the causes and amplitudes of short-term variations in pH_T in
356 the LSLE, Mucci et al. (2017) showed that pH_T in surface waters was constrained within a range of 7.85 to 7.93 during a 50-
357 h survey over two tidal cycles at the head of the Laurentian Channel. It is notable that even the upwelling of water from 100 m
358 depth or of low-oxygen LSLE bottom water would not decrease pH_T beyond ~ 7.75 and ~ 7.62 , respectively (Mucci et al., 2017
359 and references therein). Our results show that the phytoplankton assemblage responsible for the fall bloom may tolerate even
360 greater pH_T excursions. In the LSLE, such conditions may arise when the contribution of the low pH_T (7.12) freshwaters of
361 the Saguenay River to the LSLE surface waters is amplified during the spring freshet. However, considering that comparable
362 studies conducted in different environments have reported negative effects of decreasing pH on diatom biomass accumulation
363 (Hare et al., 2007; Hopkins et al., 2010; Schulz et al., 2013), it cannot be concluded that all diatom species thriving in the
364 LSLE are insensitive to acidification.

365 In contrast to the pCO_2 treatment, warming affected the development of the bloom in several ways. Increasing temperature by
366 5 °C significantly increased the accumulation rate of Chl *a*, and the nanophytoplankton growth rate during Phase I of the
367 bloom. The positive effects of warming on maximum P_P during the development phase of the bloom most likely reflect the
368 sensitivity of photosynthesis to temperature (Sommer and Lengfellner, 2008; Kim et al., 2013). It could also be related to
369 optimal growth temperatures, which are often higher than in situ temperatures in marine phytoplankton (Thomas et al., 2012;
370 Boyd et al., 2013). In support of this hypothesis, previous studies have reported optimal growth temperatures of 20–25 °C for
371 *S. costatum*, which is 5–10 °C higher than the warmer treatment investigated in our study (Suzuki and Takahashi, 1995;
372 Montagnes and Franklin, 2001). Extrapolating results from a mesocosm experiment to the field is not straightforward, as little
373 is known of the projected warming of the upper waters of the LSLE in the next decades. In the Gulf of St. Lawrence, positive
374 temperature anomalies in surface waters have varied from 0.25 to 0.75 °C per decade between 1985 and 2013 (Larouche and
375 Galbraith, 2016). In the LSLE, warming of surface waters will likely result from a complex interplay between heat transfer at
376 the air-water interface and variations in vertical mixing and upwelling of the cold intermediate layer at the head of the Estuary
377 (Galbraith et al., 2014). Considering current uncertainties regarding future warming of the LSLE, studies should be conducted

378 over a wider range of temperatures in order to better constrain the potential effect of warming on the development of the
379 blooms in the LSLE.

380 Picoeukaryotes showed a more or less gradual decrease in abundance during Phase I, and our results show that this decline
381 was not influenced by the increases in pCO₂ (Fig. 4g, h; Table 2). Picoeukaryotes are expected to benefit from high pCO₂
382 conditions even more so than diatoms as CO₂ can passively diffuse through their relatively thin boundary layer precluding the
383 necessity of a costly uptake mechanism such as a CCM (Schulz et al., 2013). This hypothesis has been supported by several
384 studies showing a stimulating effect of pCO₂ on picoeukaryote growth (Bach et al., 2016; Hama et al., 2016; Schulz et al.,
385 2017 and references therein). On the other hand, in nature, the abundance of picoeukaryotes generally results from a delicate
386 balance between cell division rates and cell losses through microzooplankton grazing and viral attacks. The few experiments,
387 including the current study, reporting the absence or a modest effect of increasing pCO₂ on the abundance of eukaryotic
388 picoplankton attribute their observations to an increase in nano- and microzooplankton grazing (Rose et al., 2009; Neale et al.,
389 2014). During our experiment, the biomass of microzooplankton increased with increasing pCO₂ by ca. 200–300 % at the two
390 temperatures tested (Ferreira and Lemli, unpubl. data). Thus, it is possible that a positive effect of increasing pCO₂ and
391 warming on picoeukaryote abundances might have been masked by higher picoeukaryote losses due to increased
392 microzooplankton grazing.

393 **4.3 Phase II (declining phase of the bloom)**

394 The gradual decrease in nanophytoplankton abundances coincided with an increase in the abundance of picocyanobacteria
395 (Fig. 4j). At in situ temperature, the picocyanobacteria abundance during Phase II was unaffected by the increase in pCO₂ over
396 the full range investigated (Fig. 4l; Table 4). The lack of positive response of picocyanobacteria to elevated pCO₂ was
397 somewhat surprising considering that they have less efficient CCMs than diatoms (Schulz et al., 2013). Accordingly, several
398 studies have reported a stimulation of the net growth rate of picocyanobacteria under elevated pCO₂ in different environments
399 (coastal Japan, Mediterranean Sea, and Raunefjorden in Norway) and under different nutrient regimes, i.e. bloom and post-
400 bloom conditions (Hama et al., 2016; Sala et al., 2016; Schulz et al., 2017). However, studies have also shown no direct effect
401 of elevated pCO₂ on the net growth of picocyanobacteria during studies conducted in the Subtropical North Atlantic and the
402 South Pacific (Law et al., 2012; Lomas et al., 2012). In our study, picocyanobacteria abundance was even reduced when high
403 CO₂ was combined with warming. Similar negative effects of CO₂ on picocyanobacteria (particularly *Synechococcus*) have
404 also been observed under later stages of bloom development, i.e. nutrient depletion, either caused by competition or grazing
405 (Paulino et al., 2008; Hopkins et al., 2010). A potential increase in grazing pressure, following the rise in heterotrophic
406 nanoflagellates abundance (e.g. choanoflagellates; Fig. 6b) measured under high pCO₂ and warmer conditions, could explain
407 the ostensible negative effect of increasing pCO₂ on picocyanobacteria abundance in our experiment. Despite the absence of
408 grazing measurements during our study, our results support the hypothesis that the potential for increased picocyanobacteria
409 population growth under elevated pCO₂ and temperature is partially dependent on different grazing pressures (Fu et al., 2007).

410 Neither warming nor acidification affected the net particulate carbon fixation during the declining phase of the bloom. In our
411 study, the time-integrated P_P and Chl *a*-normalized P_P were not significantly affected by the increase in pCO_2 during Phase II
412 at the two temperatures tested (Fig. 7; Table 4). This result is surprising since nitrogen-limited cells have been shown to be
413 more sensitive to acidification, resulting in a reduction in carbon fixation rates due to higher respiration (Wu et al., 2010; Gao
414 and Campbell, 2014; Raven et al., 2014). Although our measurements do not allow to discriminate between the contributions
415 of the different phytoplankton size classes to carbon fixation, we can speculate that diatoms, which were still abundant during
416 Phase II, contributed to a significant fraction of the primary production. If so, these results suggest that *S. costatum* remained
417 insensitive to OA even under nutrient stress. However, in contrast to Phase I, increasing the temperature by 5 °C during Phase
418 II significantly increased the Chl *a*-normalized P_D . The warming-induced increase in fixed carbon being release in the dissolved
419 fraction likely stems from increased exudation by phytoplankton, or sloppy feeding / excretion following ingestion by grazers
420 (Kim et al., 2011). The increase in fixed carbon released as dissolved organic carbon (DOC) measured during Phase II may
421 also result from greater respiration by the nitrogen-limited diatoms during periods of darkness of the incubations, as dark
422 phytoplankton respiration rates generally increase with temperature (Butrón et al., 2009; Robarts and Zohary, 1987). Moreover,
423 the enclosures do not permit the sinking and export of particulates organic carbon (POC), allowing a further transformation
424 into DOC by heterotrophic bacteria, a process that could be exacerbated under warming (Wohlers et al., 2009).

425 **4.4 Effect of the treatments on primary production over the full experiment**

426 As mentioned above, increasing pCO_2 had no effect on time-integrated P_P during the two phases of the bloom, and warming
427 only affected the maximum P_P . As a result, primary production rates integrated over the whole duration of the experiment were
428 not significantly different between the two temperatures tested. Although not statistically significant, the time-integrated P_D
429 over the full experiment displays a slight decrease with increasing pCO_2 at 10 °C and overall higher values in the warmer
430 treatment (Fig. 8d; Table 5). Previous studies have reported increases of DOC exudation (Engel et al., 2013), but also
431 decreasing DOC concentrations at elevated pCO_2 under nitrate limitation (Yoshimura et al., 2014). The increase in DOC
432 exudation is attributed to a stimulation of photosynthesis resulting from its sensitivity to higher pCO_2 (Engel et al. 2013), but
433 the causes for a decrease in DOC concentrations at high pCO_2 are less clear and potentially attributable to an increase in
434 transparent exopolymer particle (TEP) production (Yoshimura et al, 2014). Elevated TEP production under high pCO_2
435 conditions has been measured both at the peak of a bloom in a mesocosm study (Engel et al., 2014), and in post-bloom nutrient
436 depleted conditions (MacGilchrist et al., 2014). However, during our study, TEP production decreased under high pCO_2
437 (Gaaloul, 2017). Thus, the apparent decrease in P_D cannot be attributed to a greater conversion of exuded dissolved
438 carbohydrate into TEP. The apparent rise in P_D under warming is consistent with previous studies reporting similar increases
439 in phytoplankton dissolved carbon release with temperature (Morán et al., 2006; Engel et al., 2011). Although these apparent
440 changes in P_D with increasing pCO_2 and warming require further investigations, they suggest that a larger proportion (~15 %
441 of P_T at 15 °C compared to 10 % at 10 °C) of the newly fixed carbon could be exuded and become available for heterotrophic
442 organisms under warmer conditions.

443 **4.5 Implications and limitations**

444 During our study, we chose to keep the pH constant during the whole experiment instead of allowing it to vary with changes
445 in photosynthesis and respiration during the bloom phases. This approach differs from previous mesocosm experiments where
446 generally no subsequent CO₂ manipulations are conducted after the initial targets are attained (Schulz et al. 2017 and therein).
447 Keeping the pH and pCO₂ conditions stable during our study allowed us to precisely quantify the effect of the changing
448 pH/pCO₂ on the processes taking place during the different phases of the bloom. Such control was not exercised in two of our
449 mesocosms (i.e. the Drifters). In these two mesocosms, the pH_T increased from 7.9 to 8.3 at 10 °C, and from 7.9 to 8.7 at 15 °C.
450 Since the buffer capacity of acidified waters diminishes with increasing CO₂, the drift in pCO₂ and pH due to biological activity
451 would have been even greater in the more acidified treatments (Delille et al., 2005; Riebesell et al., 2007). Hence, allowing
452 the pH to drift in all mesocosms would have likely ended in an overlapping of the treatments where acidification effects would
453 have been harder to detect. Thus, our experiment could be considered as an intermediate between strictly controlled small scale
454 laboratory experiments and large scale pelagic mesocosm experiments in which only the initial conditions are set. By limiting
455 pCO₂ decrease under high CO₂ drawdown due to photosynthesis during the development of the bloom phase, we minimise
456 confounding effects of pCO₂ potentially overlapping in association with high biological activity in the mesocosms. Hence, the
457 experimental conditions could be considered as extreme examples of acidification conditions, due to the extent of pCO₂ values
458 studied. However, the absence of OA effects on most biological parameters measured during our study, even under these
459 extreme conditions, strengthens the argument that the phytoplankton community in LSLE is resistant to OA.

460 **5. Conclusion**

461 Our results reveal a remarkable resistance of the different phytoplankton size classes to the large range of pCO₂/pH investigated
462 during our study. It is noteworthy that the plankton assemblage was submitted to decreases in pH far exceeding those that they
463 are regularly exposed to in the LSLE. The resistance of *S. costatum* to the pCO₂ treatments suggests that the acidification of
464 surface waters of the LSLE will not affect the development rate and the amplitude of fall blooms dominated by this species.
465 Photosynthetic picoeukaryotes and picocyanobacteria thriving alongside the blooming diatoms were also insensitive to
466 acidification. In contrast to the pCO₂ treatments, warming the water by 5 °C had multiple impacts on the development and
467 decline of the bloom. The 5 °C warming hastened the development of the diatom bloom (albeit with no increase in total cells
468 number) and increased the abundance of picocyanobacteria during Phase II despite a reduction under high pCO₂. These
469 temperature-induced variations in the phytoplankton assemblage were accompanied by an increase in maximal P_P and suggest
470 a potential increase in P_D under warming, although no significant changes in time-integrated P_P and P_D were observed over the
471 phases or the full temporal scale of the experiment. Overall, our results indicate that warming could have more important
472 impacts than acidification on phytoplankton bloom development in the LSLE in the next decades. Future studies should be
473 conducted and specifically designed to better constrain the potential effects of warming on phytoplankton succession and
474 primary production in the LSLE.

475 *Data availability.* The data are freely accessible via <https://doi.org/10.1594/PANGAEA.886887>, or can be obtained by
476 contacting the author (robin.benard.1@ulaval.ca).

477 *Author contributions.* R. B enard was responsible for the experimental design elaboration, data sampling and processing, and
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489 **References**

490 Annane, S., St-Amand, L., Starr, M., Pelletier, E., and Ferreyra, G. A.: Contribution of transparent exopolymeric particles
491 (TEP) to estuarine particulate organic carbon pool, *Mar. Ecol. Prog. Ser.*, 529, 17–34, doi:10.3354/meps11294, 2015.

492 Bach, L. T., Taucher, J., Boxhammer, T., Ludwig, A., Aberle-Malzahn, N., Abrahamsson, K., Alm en, A. K., Asplund, M. E.,
493 Audritz, S., Boersma, M., Breitbarth, E., Bridges, C., Brussaard, C., Brutemark, A., Clemmesen, C., Collins, S., Crawford, K.,
494 Dahlke, F., Deckelnick, M., Dittmar, T., Doose, R., Dupont, S., Eberlein, T., Endres, S., Engel, A., Engstr om- st, J., Febiri,
495 S., Fleischer, D., Fritsche, P., Gledhill, M., G ttler, G., Granberg, M., Grossart, H. P., Grifos, A., Hoffmann, L., Karlsson, A.,
496 Klages, M., John, U., Jutfelt, F., K oster, I., Lange, J., Leo, E., Lischka, S., Lohbeck, K., Lundve, B., Mark, F. C., Meyerh ofer,
497 M., Nicolai, M., Pansch, C., Petersson, B., Reusch, T., De Moraes, K. R., Schartau, M., Scheinin, M., Schulz, K. G., Schwarz,
498 U., Stenegren, M., Stiasny, M., Storch, D., Stuhr, A., Sswat, L., Svensson, M., Thor, P., Voss, M., Van De Waal, D., Wannicke,
499 N., Wohlrab, S., Wulff, A., Achterberg, E. P., Alguer -Mu niz, M., Anderson, L. G., Bellworthy, J., B udenbender, J., Czerny,
500 J., Ericson, Y., Esposito, M., Fischer, M., Haunost, M., Helleman, D., Horn, H. G., Hornick, T., Meyer, J., Sswat, M., Zark,
501 M., and Riebesell, U.: Influence of ocean acidification on a natural winter-to-summer plankton succession: First insights from
502 a long-term mesocosm study draw attention to periods of low nutrient concentrations, *PLoS ONE*, 11(8), 1–33,
503 doi:10.1371/journal.pone.0159068, 2016.

504 Beardall, J., Stojkovic, S., and Gao, K.: Interactive effects of nutrient supply and other environmental factors on the sensitivity
505 of marine primary producers to ultraviolet radiation: Implications for the impacts of global change, *Aquat. Biol.*, 22, 5–23,
506 doi:10.3354/ab00582, 2014.

507 Bérard-Therriault, L., Poulin, M., and Bossé, L.: Guide d'identification du phytoplancton marin de l'estuaire et du golfe du
508 Saint-Laurent incluant également certains protozoaires., Canadian Special Publication of Fisheries and Aquatic Sciences, 128,
509 1–387, doi :10.1139/9780660960579, 1999.

510 Boyd, P. W., and Hutchins, D. A.: Understanding the responses of ocean biota to a complex matrix of cumulative
511 anthropogenic change, *Mar. Ecol. Prog. Ser.*, 470, 125–135, doi:10.3354/meps10121, 2012.

512 Boyd, P. W., Rynearson, T. A., Armstrong, E. A., Fu, F., Hayashi, K., Hu, Z., Hutchins, D. A., Kudela, R. M., Litchman, E.,
513 Mulholland, M. R., Passow, U., Strzepek, R. F., Whittaker, K. A., Yu, E., and Thomas, M. K.: Marine Phytoplankton
514 temperature versus growth responses from polar to tropical waters - outcome of a scientific community-wide study, *PLoS*
515 *ONE*, 8(5), doi:10.1371/journal.pone.0063091, 2013.

516 Boyd, P. W., Lennartz, S. T., Glover, D. M., and Doney, S. C.: Biological ramifications of climate-change-mediated oceanic
517 multi-stressors, *Nat. Clim. Chang.*, 5(1), 71–79, doi:10.1038/nclimate2441, 2015.

518 Brussaard, C. P. D., Noordeloos, A. A. M., Witte, H., Collenteur, M. C. J., Schulz, K., Ludwig, A., and Riebesell, U.: Arctic
519 microbial community dynamics influenced by elevated CO₂ levels, *Biogeosciences*, 10(2), 719–731, doi:10.5194/bg-10-719-
520 2013, 2013.

521 Butrón, A., Iriarte, A., and Madariaga, I.: Size-fractionated phytoplankton biomass, primary production and respiration in the
522 Nervión-Ibaizabal estuary: A comparison with other nearshore coastal and estuarine ecosystems from the Bay of Biscay, *Cont.*
523 *Shelf Res.*, 29(8), 1088–1102, doi:10.1016/j.csr.2008.11.013, 2009.

524 Byrne, R. H.: Standardization of Standard Buffers by Visible Spectrometry, *Anal. Chem*, 59, 1479–1481,
525 doi:10.1021/ac00137a025, 1987.

526 Cai, W. J., and Wang, Y.: The chemistry, fluxes, and sources of carbon dioxide in the estuarine waters of the Satilla and
527 Altamaha Rivers, Georgia, *Limnol. Oceanogr.*, 43(4), 657–668, doi:10.4319/lo.1998.43.4.0657, 1998.

528 Caldeira, K., and Wickett, M. E.: Ocean model predictions of chemistry changes from carbon dioxide emissions to the
529 atmosphere and ocean, *J. Geophys. Res.*, 110(C9), 1–12, doi:10.1029/2004JC002671, 2005.

530 Clayton, T. D., and Byrne, R. H.: Spectrophotometric seawater pH measurements: total hydrogen ion concentration scale
531 calibration of m-cresol purple and at-sea results, *Deep. Res. Part I*, 40(10), 2115–2129, doi:10.1016/0967-0637(93)90048-8,
532 1993.

533 d'Anglejan, B.: Recent sediments and sediment transport processes in the St. Lawrence Estuary, in: *Oceanography of a large-*
534 *scale estuarine system*, Eds: El-Sabh, M. I., and Silverberg, N., Springer-Verlag, New York, USA, 109–129, doi:
535 10.1002/9781118663783.ch6, 1990.

536 Delille, B., Harlay, J., Zondervan, I., Jacquet, S., Chou, L., Wollast, R., Bellerby, R. G. J., Frankignoulle, M., Borges, A. V.,
537 Riebesell, U. and Gattuso, J. P.: Response of primary production and calcification to changes of pCO₂during experimental

538 blooms of the coccolithophorid *Emiliana huxleyi*, *Global Biogeochem. Cycles*, 19(2), 1–14, doi:10.1029/2004GB002318,
539 2005.

540 Dickson, A. G.: Standard potential of the reaction: $\text{AgCl(s)} + 1/2\text{H}_2\text{(g)} = \text{Ag(s)} + \text{HCl(aq)}$ and the standard acidity constant of
541 the ion HSO_4^- in synthetic sea water from 273.15 to 318.15 K, *J. Chem. Thermodyn.*, 22(2), 113–127, doi:10.1016/0021-
542 9614(90)90074-Z, 1990.

543 Dinauer, A., and Mucci, A.: Spatial variability in surface-water pCO_2 and gas exchange in the world's largest semi-enclosed
544 estuarine system: St. Lawrence Estuary (Canada), *Biogeosciences*, 14(13), 3221–3237, doi:10.5194/bg-14-3221-2017, 2017.

545 Doney, S. C., Fabry, V. J., Feely, R. A., and Kleypas, J. A.: Ocean acidification: The other CO_2 problem, *Ann. Rev. Mar. Sci.*,
546 1(1), 169–192, doi:10.1146/annurev.marine.010908.163834, 2009.

547 Duarte, C. M., Hendriks, I. E., Moore, T. S., Olsen, Y. S., Steckbauer, A., Ramajo, L., Carstensen, J., Trotter, J. A., and
548 McCulloch, M.: Is ocean acidification an open-ocean syndrome? Understanding anthropogenic impacts on seawater pH,
549 *Estuaries Coasts*, 36(2), 221–236, doi:10.1007/s12237-013-9594-3, 2013.

550 Eberlein, T., Wohlrab, S., Rost, B., John, U., Bach, L. T., Riebesell, U., and Van De Waal, D. B.: Effects of ocean acidification
551 on primary production in a coastal North Sea phytoplankton community, *PLoS ONE*, 12(3), e0172594,
552 doi:10.1371/journal.pone.0172594, 2017.

553 Engel, A., Zondervan, I., Aerts, K., Beaufort, L., Benthien, A., Chou, L., Delille, B., Gattuso, J.-P., Harlay, J., Heeman, C.,
554 Hoffmann, L., Jacquet, S., Nejstgaard, J., Pizay, M.-D., Rochelle-Newall, E., Schneider, U., Terbrueggen A., and Riebesell,
555 U.: Testing the direct effect of CO_2 concentration on a bloom of the coccolithophorid *Emiliana huxleyi* in mesocosm
556 experiments, *Limnol. Oceanogr.*, 50(2), 493–507, doi:10.4319/lo.2005.50.2.0493, 2005.

557 Engel, A., Händel, N., Wohlers, J., Lunau, M., Grossart, H.-P., Sommer, U., and Riebesell, U.: Effects of sea surface warming
558 on the production and composition of dissolved organic matter during phytoplankton blooms: Results from a mesocosm study,
559 *J. Plankton Res.*, 33(3), 357–372, doi:10.1093/plankt/fbq122, 2011.

560 Engel, A., Borchard, C., Piontek, J., Schulz, K. G., Riebesell, U., and Bellerby, R.: CO_2 increases ^{14}C primary production in
561 an Arctic plankton community, *Biogeosciences*, 10(3), 1291–1308, doi:10.5194/bg-10-1291-2013, 2013.

562 Engel, A., Piontek, J., Grossart, H.-P., Riebesell, U., Schulz, K. G., and Sperling, M.: Impact of CO_2 enrichment on organic
563 matter dynamics during nutrient induced coastal phytoplankton blooms, *J. Plankton Res.*, 36(3), 641–657,
564 doi:10.1093/plankt/fbt125, 2014.

565 Feely, R. A., Doney, S. C., and Cooley, S. R.: Ocean acidification: present conditions and future changes in a high- CO_2 world,
566 *Oceanography*, 22(4), 36–47, doi:10.5670/oceanog.2009.95, 2009.

567 Feng, Y., Hare, C. E., Leblanc, K., Rose, J. M., Zhang, Y., DiTullio, G. R., Lee, P. A., Wilhelm, S. W., Rowe, J. M., Sun, J.,
568 Nemcek, N., Gueguen, C., Passow, U., Benner, I., Brown, C., and Hutchins, D. A.: Effects of increased pCO_2 and temperature
569 on the North Atlantic spring bloom. I. The phytoplankton community and biogeochemical response, *Mar. Ecol. Prog. Ser.*,
570 388, 13–25, doi:10.3354/meps08133, 2009.

571 Ferland, J., Gosselin, M., and Starr, M.: Environmental control of summer primary production in the Hudson Bay system: The
572 role of stratification, *J. Mar. Syst.*, 88(3), 385–400, doi:10.1016/j.jmarsys.2011.03.015, 2011.

573 Gaaloul, H.: Effets du changement global sur les particules exopolymériques transparentes au sein de l'estuaire maritime du
574 Saint-Laurent, M.Sc. thesis, Université du Québec à Rimouski, Canada, 133 pp., 2017.

575 Galbraith, P. S., Chassé, J., Gilbert, D., Larouche, P., Caverhill, C., Lefaiivre, D., Brickman, D., Pettigrew, B., Devine, L., and
576 Lafleur, C.: Physical Oceanographic Conditions in the Gulf of St. Lawrence in 2013, *DFO Can. Sci. Advis. Sec. Res. Doc.*,
577 2014/062(November), vi + 84 pp, 2014.

578 Gao, K., and Campbell, D. A.: Photophysiological responses of marine diatoms to elevated CO₂ and decreased pH: A review,
579 *Funct. Plant Biol.*, 41(5), 449–459, doi:10.1071/FP13247, 2014.

580 Gao, G., Jin, P., Liu, N., Li, F., Tong, S., Hutchins, D. A., and Gao, K.: The acclimation process of phytoplankton biomass,
581 carbon fixation and respiration to the combined effects of elevated temperature and pCO₂ in the northern South China Sea,
582 *Mar. Pollut. Bull.*, 118(1–2), 213–220, doi:10.1016/j.marpolbul.2017.02.063, 2017.

583 Gattuso, J. P., Mach, K. J., and Morgan, G.: Ocean acidification and its impacts: An expert survey, *Clim. Change*, 117(4),
584 725–738, doi:10.1007/s10584-012-0591-5, 2013.

585 Gattuso, J.-P., Magnan, A., Bille, R., Cheung, W. W. L., Howes, E. L., Joos, F., Allemand, D., Bopp, L., Cooley, S. R., Eakin,
586 C. M., Hoegh-Guldberg, O., Kelly, R. P., Portner, H.-O., Rogers, a. D., Baxter, J. M., Laffoley, D., Osborn, D., Rankovic, A.,
587 Rochette, J., Sumaila, U. R., Treyer, S., and Turley, C.: Contrasting futures for ocean and society from different anthropogenic
588 CO₂ emissions scenarios, *Science*, 349(6243), doi:10.1126/science.aac4722, 2015.

589 Giordano, M., Beardall, J., and Raven, J. A.: CO₂ concentrating mechanisms in algae: Mechanisms, environmental
590 modulation., and evolution, *Annu. Rev. Plant Biol.*, 56(1), 99–131, doi:10.1146/annurev.arplant.56.032604.144052, 2005.

591 Gunderson, A. R., Armstrong, E. J., and Stillman, J. H.: Multiple stressors in a changing World: The need for an improved
592 perspective on physiological responses to the dynamic marine environment, *Ann. Rev. Mar. Sci.*, 8(1), 357–378,
593 doi:10.1146/annurev-marine-122414-033953, 2016.

594 Hama, T., Inoue, T., Suzuki, R., Kashiwazaki, H., Wada, S., Sasano, D., Kosugi, N., and Ishii, M.: Response of a phytoplankton
595 community to nutrient addition under different CO₂ and pH conditions, *J. Oceanogr.*, 72(2), 207–223, doi:10.1007/s10872-
596 015-0322-4, 2016.

597 Hansen, H. P., and Koroleff, F.: Determination of nutrients, in: *Methods of Seawater Analysis*, 3, Eds: Grasshoff K., Kremling,
598 K., and Ehrhardt, M., Wiley-VCH Verlag GmbH, Weinheim, Germany, 159–228, doi:10.1002/9783527613984.ch10, 2007.

599 Hare, C. E., Leblanc, K., DiTullio, G. R., Kudela, R. M., Zhang, Y., Lee, P. A., Riseman, S., and Hutchins, D. A.: Consequences
600 of increased temperature and CO₂ for phytoplankton community structure in the Bering Sea, *Mar. Ecol. Prog. Ser.*, 352, 9–16,
601 doi:10.3354/meps07182, 2007.

602 Havenhand, J., Dupont, S., and Quinn, G. P.: Designing ocean acidification experiments to maximise inference, in *Guide to*
603 *best practices for ocean acidification research and data reporting*, Eds: Riebesell, U., Fabry, V. J., and Gattuso, J.-P.,
604 Publications Office of the European Union, Luxembourg, 67–80, 2010.

605 Hopkins, F. E., Turner, S. M., Nightingale, P. D., Steinke, M., Bakker, D., and Liss, P. S.: Ocean acidification and marine
606 trace gas emissions, *Proc. Natl. Acad. Sci. U.S.A.*, 107(2), 760–765, doi:10.1073/pnas.0907163107, 2010.

607 Husserr, R., Levasseur, M., Lizotte, M., Tremblay, J. É., Mol, J., Thomas, H., Gosselin, M., Starr, M., Miller, L. A., Jarniková,
608 T., Schuback, N., and Mucci, A.: Impact of ocean acidification on Arctic phytoplankton blooms and dimethyl sulfide
609 concentration under simulated ice-free and under-ice conditions, *Biogeosciences*, 14(9), 2407–2427, doi:10.5194/bg-14-2407-
610 2017, 2017.

611 IPCC: Working Group I Contribution to the Fifth Assessment Report Climate Change 2013: The Physical Science Basis,
612 Intergov. Panel Clim. Chang., 1535, doi:10.1017/CBO9781107415324., 2013.

613 Kim, K. Y., Garbary, D. J., and McLachlan, J. L.: Phytoplankton dynamics in Pomquet Harbour, Nova Scotia: a lagoon in the
614 southern Gulf of St Lawrence, *Phycologica*, 43(3), 311–328, 2004.

615 Kim, J. M., Lee, K., Shin, K., Yang, E. J., Engel, A., Karl, D. M. and Kim, H. C.: Shifts in biogenic carbon flow from particulate
616 to dissolved forms under high carbon dioxide and warm ocean conditions, *Geophys. Res. Lett.*, 38(8),
617 doi:10.1029/2011GL047346, 2011.

618 Kim, J. H., Kim, K. Y., Kang, E. J., Lee, K., Kim, J. M., Park, K. T., Shin, K., Hyun, B., and Jeong, H. J.: Enhancement of
619 photosynthetic carbon assimilation efficiency by phytoplankton in the future coastal ocean, *Biogeosciences*, 10(11), 7525–
620 7535, doi:10.5194/bg-10-7525-2013, 2013.

621 Knap, A., Michaels, A., Close, A. R., Ducklow, H., and Dickson, A. G.: Protocols for the Joint Global Ocean Flux Study
622 (JGOFS) core measurements, JGOFS Rep No. 19, Reprint of the IOC Manuals and Guides No. 29, UNESCO, Bergen, Norway,
623 doi:10013/epic.27912, 1996.

624 Kroeker, K. J., Kordas, R. L., Crim, R., Hendriks, I. E., Ramajo, L., Singh, G. S., Duarte, C. M., and Gattuso, J. P.: Impacts of
625 ocean acidification on marine organisms: Quantifying sensitivities and interaction with warming, *Glob. Chang. Biol.*, 19(6),
626 1884–1896, doi:10.1111/gcb.12179, 2013.

627 Larouche, P., and Galbraith, P. S.: Canadian coastal seas and Great Lakes sea surface temperature climatology and recent
628 trends, *Can. J. Remote Sens.*, 42(3), 243–258, doi:10.1080/07038992.2016.1166041, 2016.

629 Law, C. S., Breitbarth, E., Hoffmann, L. J., McGraw, C. M., Langlois, R. J., Laroche, J., Marriner, A., and Safi, K. A.: No
630 stimulation of nitrogen fixation by non-filamentous diazotrophs under elevated CO₂ in the South Pacific, *Glob. Chang. Biol.*,
631 18, 3004–3014, 2012.

632 Le Quéré, C., Moriarty, R., Andrew, R. M., Canadell, J. G., Sitch, S., Korsbakken, J. I., Friedlingstein, P., Peters, G. P., andres,
633 R. J., Boden, T. A., Houghton, R. A., House, J. I., Keeling, R. F., Tans, P., Arneeth, A., Bakker, D. C. E., Barbero, L., Bopp,
634 L., Chang, J., Chevallier, F., Chini, L. P., Ciais, P., Fader, M., Feely, R. A., Gkritzalis, T., Harris, I., Hauck, J., Ilyina, T., Jain,
635 A. K., Kato, E., Kitidis, V., Klein Goldewijk, K., Koven, C., Landschützer, P., Lauvset, S. K., Lefèvre, N., Lenton, A., Lima,
636 I. D., Metzl, N., Millero, F., Munro, D. R., Murata, A., S. Nabel, J. E. M., Nakaoka, S., Nojiri, Y., O’Brien, K., Olsen, A.,
637 Ono, T., Pérez, F. F., Pfeil, B., Pierrot, D., Poulter, B., Rehder, G., Rödenbeck, C., Saito, S., Schuster, U., Schwinger, J.,
638 Séférian, R., Steinhoff, T., Stocker, B. D., Sutton, A. J., Takahashi, T., Tilbrook, B., Van Der Laan-Luijkx, I. T., Van Der

639 Werf, G. R., Van Heuven, S., Vandemark, D., Viovy, N., Wiltshire, A., Zaehle, S., and Zeng, N.: Global Carbon Budget 2015,
640 Earth Syst. Sci. Data, 7(2), 349–396, doi:10.5194/essd-7-349-2015, 2015.

641 Legendre, L., Demers, S., Yentsch, C. M., and Yentsch, C. S.: The ¹⁴C method: Patterns of dark CO₂ fixation and DCMU
642 correction to replace the dark bottle, Limnol. Oceanogr., 28(5), 996–1003, doi:10.4319/lo.1983.28.5.0996, 1983.

643 Levasseur, M., Therriault, J.-C., and Legendre, L.: Hierarchical control of phytoplankton succession by physical factors, Mar.
644 Ecol. Prog. Ser., 19, 211–222, doi:10.3354/meps019211, 1984.

645 Levasseur, M. E., and Therriault, J.-C.: Phytoplankton biomass and nutrient dynamics in a tidally induced upwelling: the role
646 of the NO₃:SiO₄ ratio, Mar. Ecol. Prog. Ser., 39, 87–97, 1987.

647 Levasseur, M. E., Harrison, P. J., Heimdal, B. R., and Therriault, J.-C.: Simultaneous nitrogen and silicate deficiency of a
648 phytoplankton community in a coastal jet-front, Mar. Biol., 104(2), 329–338, doi:10.1007/BF01313275, 1990.

649 Lomas, M. W., Hopkinson, B. M., Losh, J. L., Ryan, D. E., Shi, D. L., Xu, Y., and Morel, F. M. M.: Effect of ocean acidification
650 on cyanobacteria in the subtropical North Atlantic, Aquat. Microb. Ecol., 66(3), 211–222, doi:10.3354/ame01576, 2012.

651 Lund, J. W. G., Kipling, C., and Le Cren, E. D.: The inverted microscope method of estimating algal numbers and the statistical
652 basis of estimates by counting, Hydrobiologia, 11, 143–170, 1958.

653 MacGilchrist, G. A., Shi, T., Tyrrell, T., Richier, S., Moore, C. M., Dumousseaud, C., and Achterberg, E. P.: Effect of enhanced
654 pCO₂ levels on the production of dissolved organic carbon and transparent exopolymer particles in short-term bioassay
655 experiments, Biogeosciences, 11(13), 3695–3706, doi:10.5194/bg-11-3695-2014, 2014.

656 Marie, D., Simon, N., and Vaultot, D.: Phytoplankton cell counting by flow cytometry, Algal Cult. Tech., 253–267,
657 doi:10.1016/B978-012088426-1/50018-4, 2005.

658 Maugendre, L., Gattuso, J. P., Louis, J., De Kluijver, A., Marro, S., Soetaert, K., and Gazeau, F.: Effect of ocean warming and
659 acidification on a plankton community in the NW Mediterranean Sea, ICES J. Mar. Sci., 72(6), 1744–1755,
660 doi:10.1093/icesjms/fsu161, 2015.

661 Millero, F. J.: The pH of estuarine waters, Limnol. Oceanogr., 31(4), 839–847, doi:10.4319/lo.1986.31.4.0839, 1986.

662 Montagnes, D. J. S., and Franklin, M.: Effect of temperature on diatom volume, growth rate, and carbon and nitrogen content:
663 Reconsidering some paradigms, Limnol. Oceanogr., 46(8), 2008–2018, doi:10.4319/lo.2001.46.8.2008, 2001.

664 Morán, X. A. G., Sebastián, M., Pedrós-Alió, C., and Estrada, M.: Response of Southern Ocean phytoplankton and
665 bacterioplankton production to short-term experimental warming, Limnol. Oceanogr., 51(4), 1791–1800,
666 doi:10.4319/lo.2006.51.4.1791, 2006.

667 Morán, A. G., Alonso-sa, L., Nogueira, E., Ducklow, H. W., Gonza, N., Calvo-dí, A., Arandia-gorostidi, N., Dí, L., Huete-
668 stauffer, T. M., Rey, U., and Carlos, J.: More, smaller bacteria in response to ocean's warming?, Proc. R. Soc., 282(1810), 1–
669 9, doi:http://dx.doi.org/10.1098/rspb.2015.0371, 2015.

670 Mucci, A., Levasseur, M., Gratton, Y., Martias, C., Scarratt, M., Gilbert, D., Tremblay, J.-É., Ferreyra, G., and Lansard, B.:
671 Tidally-induced variations of pH at the head of the Laurentian Channel, Can. J. Fish. Aquat. Sci., doi:10.1139/cjfas-2017-
672 0007, 2017.

673 Neale, P. J., Sobrino, C., Segovia, M., Mercado, J. M., Leon, P., Cortés, M. D., Tuite, P., Picazo, A., Salles, S., Cabrerizo, M.
674 J., Prasil, O., Montecino, V., and Reul, A.: Effect of CO₂, nutrients and light on coastal plankton. I. Abiotic conditions and
675 biological responses, *Aquat. Biol.*, 22, 25–41, doi:10.3354/ab00587, 2014.

676 Parsons, T. R., Maita, Y., and Lalli, C. M.: A manual of chemical and biological methods for seawater analysis, Permagon
677 Press, New York, 1984.

678 Paul, C., Matthiessen, B., and Sommer, U.: Warming, but not enhanced CO₂ concentration, quantitatively and qualitatively
679 affects phytoplankton biomass, *Mar. Ecol. Prog. Ser.*, 528, 39–51, doi:10.3354/meps11264, 2015.

680 Paul, C., Sommer, U., Garzke, J., Moustaka-Gouni, M., Paul, A., and Matthiessen, B.: Effects of increased CO₂ concentration
681 on nutrient limited coastal summer plankton depend on temperature, *Limnol. Oceanogr.*, 61(3), 853–868,
682 doi:10.1002/lno.10256, 2016.

683 Paulino, A. I., Egge, J. K. and Larsen, A.: Effects of increased atmospheric CO₂ on small and intermediate sized osmotrophs
684 during a nutrient induced phytoplankton bloom, *Biogeosciences*, 5(3), 739–748, doi:10.5194/bg-5-739-2008, 2008. Pierrot, D.,
685 Lewis, E., and Wallace, D. W. R.: MS Excel program developed for CO₂ system calculations, Carbon Dioxide Information
686 Analysis Center, ORNL/CDIAC-105a, Oak Ridge National Laboratory, US Department of Energy, Oak Ridge, 592 Tennessee,
687 2006.

688 Raven, J. A., Beardall, J., and Giordano, M.: Energy costs of carbon dioxide concentrating mechanisms in aquatic organisms,
689 *Photosynth. Res.*, 121, 111–124, 2014.

690 Riebesell, U., and Gattuso, J.-P.: Lessons learned from ocean acidification research, *Nat. Clim. Chang.*, 5(1), 12–14,
691 doi:10.1038/nclimate2456, 2015.

692 Riebesell, U., and Tortell, P. D.: Effects of ocean acidification on pelagic organism and ecosystems, in *Ocean Acidification*,
693 Eds: Gattuso J.-P., and Hansson L., Oxford University Press, New York, 99–121, 2011.

694 Riebesell, U., Schulz, K. G., Bellerby, R. G. J., Botros, M., Fritsche, P., Meyerhöfer, M., Neill, C., Nondal, G., Oschlies, a,
695 Wohlers, J., and Zöllner, E.: Enhanced biological carbon consumption in a high CO₂ ocean., *Nature*, 450(7169), 545–548,
696 doi:10.1038/nature06267, 2007.

697 Riebesell, U., Czerny, J., Von Bröckel, K., Boxhammer, T., Büdenbender, J., Deckelnick, M., Fischer, M., Hoffmann, D.,
698 Krug, S. A., Lentz, U., Ludwig, A., Mucche, R., and Schulz, K. G.: Technical Note: A mobile sea-going mesocosm system -
699 New opportunities for ocean change research, *Biogeosciences*, 10(3), 1835–1847, doi:10.5194/bg-10-1835-2013, 2013.

700 Robarts, R. D., and Zohary, T.: Temperature effects on photosynthetic capacity, respiration., and growth rates of bloom-
701 forming cyanobacteria, *New Zeal. J. Mar. Freshw. Res.*, 21(3), 391–399, doi:10.1080/00288330.1987.9516235, 1987.

702 Robert-Baldo, G., Morris, M., and Byrne, R.: Spectrophotometric determination of seawater pH using phenol red, *Anal. Chem.*,
703 3(57), 2564–2567, doi:10.1021/ac00290a030, 1985.

704 Rose, J. M., Feng, Y., Gobler, C. J., Gutierrez, R., Harel, C. E., Leblanc, K., and Hutchins, D. A.: Effects of increased pCO₂
705 and temperature on the North Atlantic spring bloom. II. Microzooplankton abundance and grazing, *Mar. Ecol. Prog. Ser.*, 388,
706 27–40, doi:10.3354/meps08134, 2009.

707 Roy, S., Chanut, J.-P., Gosselin, M., and Sime-Ngando, T.: Characterization of phytoplankton communities in the Lower St.
708 Lawrence Estuary using HPLC-detected pigments and cell microscopy, *Mar. Ecol. Prog. Ser.*, 142, 55–73,
709 doi:10.3354/meps142055, 1996.

710 Sala, M. M., Aparicio, F. L., Balagué, V., Boras, J. A., Borrull, E., Cardelús, C., Cros, L., Gomes, A., López-Sanz, A., Malits,
711 A., Martínez, R. A., Mestre, M., Movilla, J., Sarmiento, H., Vázquez-Domínguez, E., Vaqué, D., Pinhassi, J., Calbet, A., Calvo,
712 E., Gasol, J. M., Pelejero, C., and Marrasé, C.: Contrasting effects of ocean acidification on the microbial food web under
713 different trophic conditions, *ICES J. Mar. Sci.*, 73(3), 670–679, doi:10.1093/icesjms/fsv130, 2016.

714 Schulz, K. G., Bellerby, R. G. J., Brussaard, C. P. D., Büdenbender, J., Czerny, J., Engel, A., Fischer, M., Koch-Klavsen, S.,
715 Krug, S. A., Lischka, S., Ludwig, A., Meyerhöfer, M., Nondal, G., Silyakova, A., Stuhr, A., and Riebesell, U.: Temporal
716 biomass dynamics of an Arctic plankton bloom in response to increasing levels of atmospheric carbon dioxide, *Biogeosciences*,
717 10(1), 161–180, doi:10.5194/bg-10-161-2013, 2013.

718 Schulz, K. G., Bach, L. T., Bellerby, R. G. J., Bermudez, R., Budenbender, J., Boxhammer, T., Czerny, J., Engel, A., Ludwig,
719 A., Meyerhofer, M., Larsen, A., Paul, A., Sswat, M., and Riebesell, U.: Phytoplankton blooms at increasing levels of
720 atmospheric carbon dioxide: experimental evidence for negative effects on prymnesiophytes and positive on small
721 picoeukaryotes, *Front. Mar. Sci.*, 4, 64, doi:10.3389/fmars.2017.00064, 2017.

722 Sommer, U., and Lengfellner, K.: Climate change and the timing, magnitude, and composition of the phytoplankton spring
723 bloom, *Glob. Chang. Biol.*, 14(6), 1199–1208, doi:10.1111/j.1365-2486.2008.01571.x, 2008.

724 Sommer, U., Paul, C., and Moustaka-Gouni, M.: Warming and ocean acidification effects on phytoplankton - From species
725 shifts to size shifts within species in a mesocosm experiment, *PLoS ONE*, 10(5), 17, doi:10.1371/journal.pone.0125239, 2015.

726 Starr, M., St-Amand, L., Devine, L., Bérard-Therriault, L., and Galbraith, P. S.: State of phytoplankton in the Estuary and Gulf
727 of St. Lawrence during 2003, *CSAS Res. Doc.*, 2004/123, 35, 2004.

728 Suzuki, Y., and Takahashi, M.: Growth responses of several diatom species isolated from various environments to temperature,
729 *J. Phycol.*, 31(6), 880–888, doi:10.1111/j.0022-3646.1995.00880.x, 1995.

730 Tatters, A. O., Roleda, M. Y., Schnetzer, A., Fu, F., Hurd, C. L., Boyd, P. W., Caron, D. A., Lie, A. A. Y., Hoffmann, L. J.,
731 and Hutchins, D. A.: Short- and long-term conditioning of a temperate marine diatom community to acidification and
732 warming., *Philos. Trans. R. Soc. Lond. B. Biol. Sci.*, 368(1627), 20120437, doi:10.1098/rstb.2012.0437, 2013.

733 Taylor, A. H., Harbour, D. S., Harris, R. P., Burkill, P. H., and Edwards, E. S.: Seasonal succession in the pelagic ecosystem
734 of the North Atlantic and the utilization of nitrogen, *J. Plankton Res.*, 15(8), 875–891, doi:10.1093/plankt/15.8.875, 1993.

735 Thomas, M. K., Kremer, C. T., Klausmeier, C. a and Litchman, E.: A global pattern of thermal adaptation in marine
736 phytoplankton., *Science*, 338(6110), 1085–1088, doi:10.1126/science.1224836, 2012.

737 Todgham, A. E., and Stillman, J. H.: Physiological responses to shifts in multiple environmental stressors: Relevance in a
738 changing world, *Integr. Comp. Biol.*, 53(4), 539–544, doi:10.1093/icb/ict086, 2013.

739 Tomas, C. R. (ed): *Identifying Marine Phytoplankton*, Academic Press: San Diego, 858 pp., 1997.

740 Tortell, P. D., DiTullio, G. R., Sigman, D. M., and Morel, F. M. M.: CO₂ effects on taxonomic composition and nutrient
741 utilization in an Equatorial Pacific phytoplankton assemblage, *Mar. Ecol. Prog. Ser.*, 236, 37–43, doi:10.3354/meps236037,
742 2002.

743 Trimborn, S., Wolf-Gladrow, D., Richter, K. U., and Rost, B.: The effect of pCO₂ on carbon acquisition and intracellular
744 assimilation in four marine diatoms, *J. Exp. Mar. Bio. Ecol.*, 376(1), 26–36, doi:10.1016/j.jembe.2009.05.017, 2009.

745 Wijffels, S., Roemmich, D., Monselesan, D., Church, J., and Gilson, J.: Ocean temperatures chronicle the ongoing warming
746 of Earth, *Nat. Clim. Chang.*, 6(2), 116–118, doi:10.1038/nclimate2924, 2016.

747 Wohlers, J., Engel, A., Zollner, E., Breithaupt, P., Jurgens, K., Hoppe, H.-G., Sommer, U. and Riebesell, U.: Changes in
748 biogenic carbon flow in response to sea surface warming, *Proc. Natl. Acad. Sci.*, 106(17), 7067–7072,
749 doi:10.1073/pnas.0812743106, 2009.

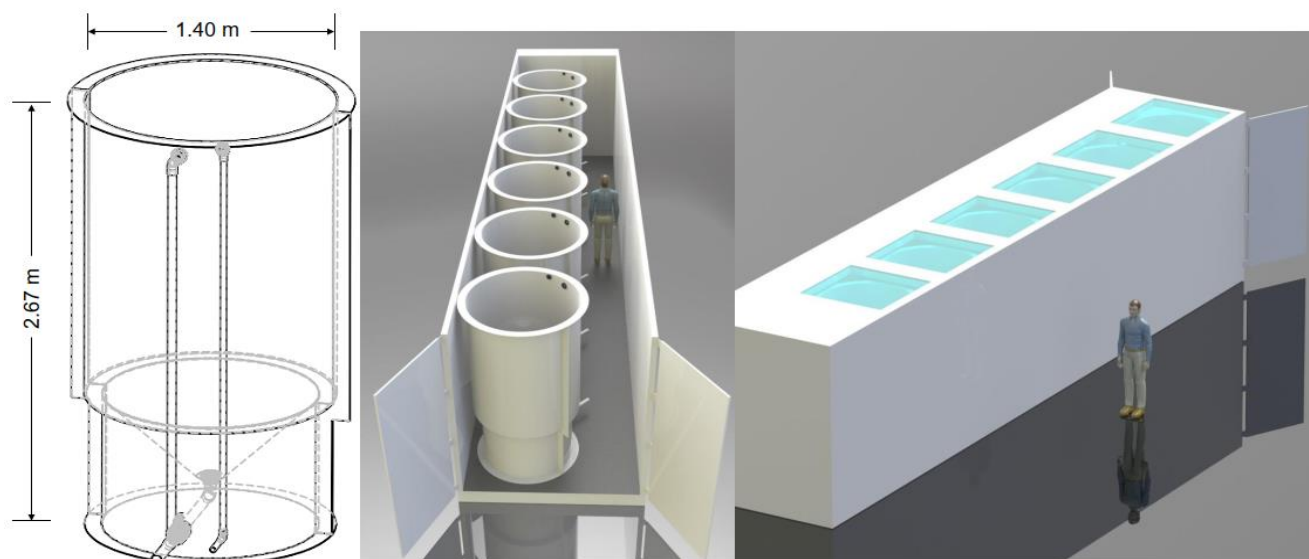
750 Wu, Y., Gao, K., and Riebesell, U.: CO₂-induced seawater acidification affects physiological performance of the marine diatom
751 *Phaeodactylum tricornutum*, *Biogeosciences*, 7(9), 2915–2923, doi:10.5194/bg-7-2915-2010, 2010.

752 Yoshimura, T., Nishioka, J., Suzuki, K., Hattori, H., Kiyosawa, H., and Watanabe, Y. W.: Impacts of elevated CO₂ on organic
753 carbon dynamics in nutrient depleted Okhotsk Sea surface waters, *J. Exp. Mar. Bio. Ecol.*, 395(1–2), 191–198,
754 doi:10.1016/j.jembe.2010.09.001, 2010.

755 Yoshimura, T., Sugie, K., Endo, H., Suzuki, K., Nishioka, J., and Ono, T.: Organic matter production response to CO₂ increase
756 in open subarctic plankton communities: Comparison of six microcosm experiments under iron-limited and -enriched bloom
757 conditions, *Deep Res. Part I Oceanogr. Res. Pap.*, 94, 1–14, doi:10.1016/j.dsr.2014.08.004, 2014.

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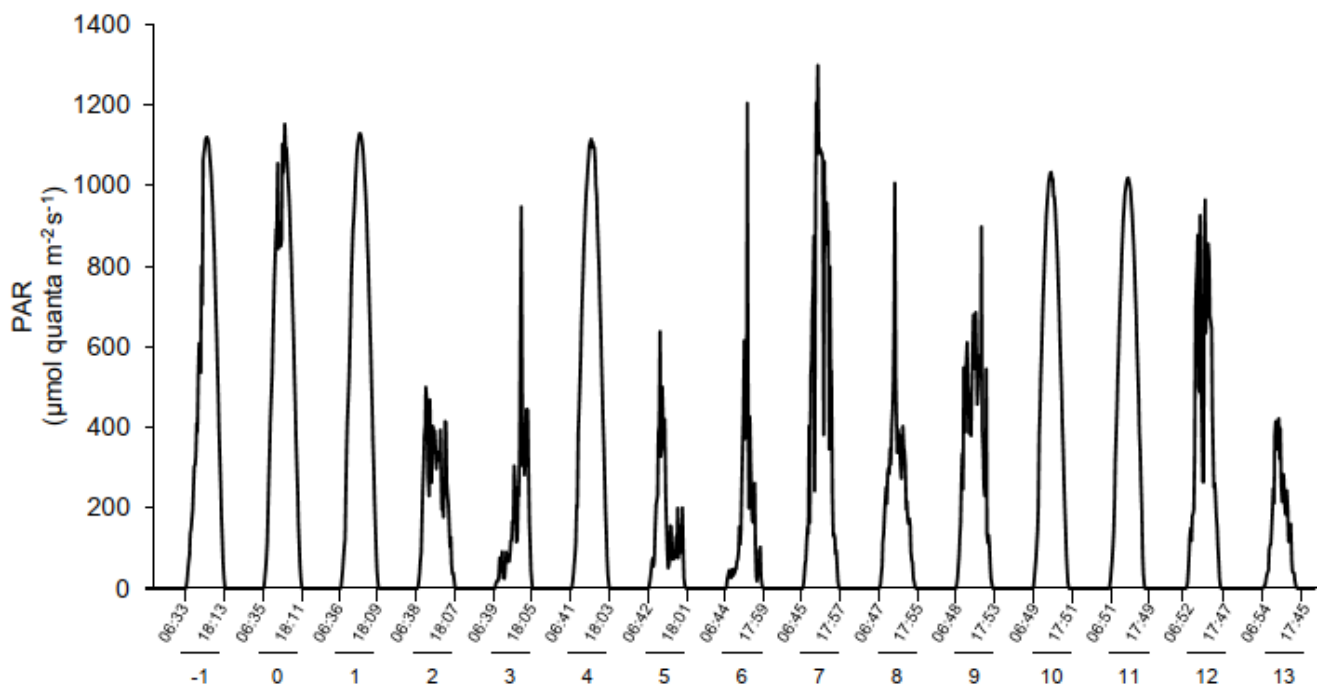
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Figure 1. Schematic drawing including mesocosm dimensions and placement within the containers (Aquabiotech Inc, Québec, Canada). The whole setup includes a second container holding 6 more mesocosms not depicted here.

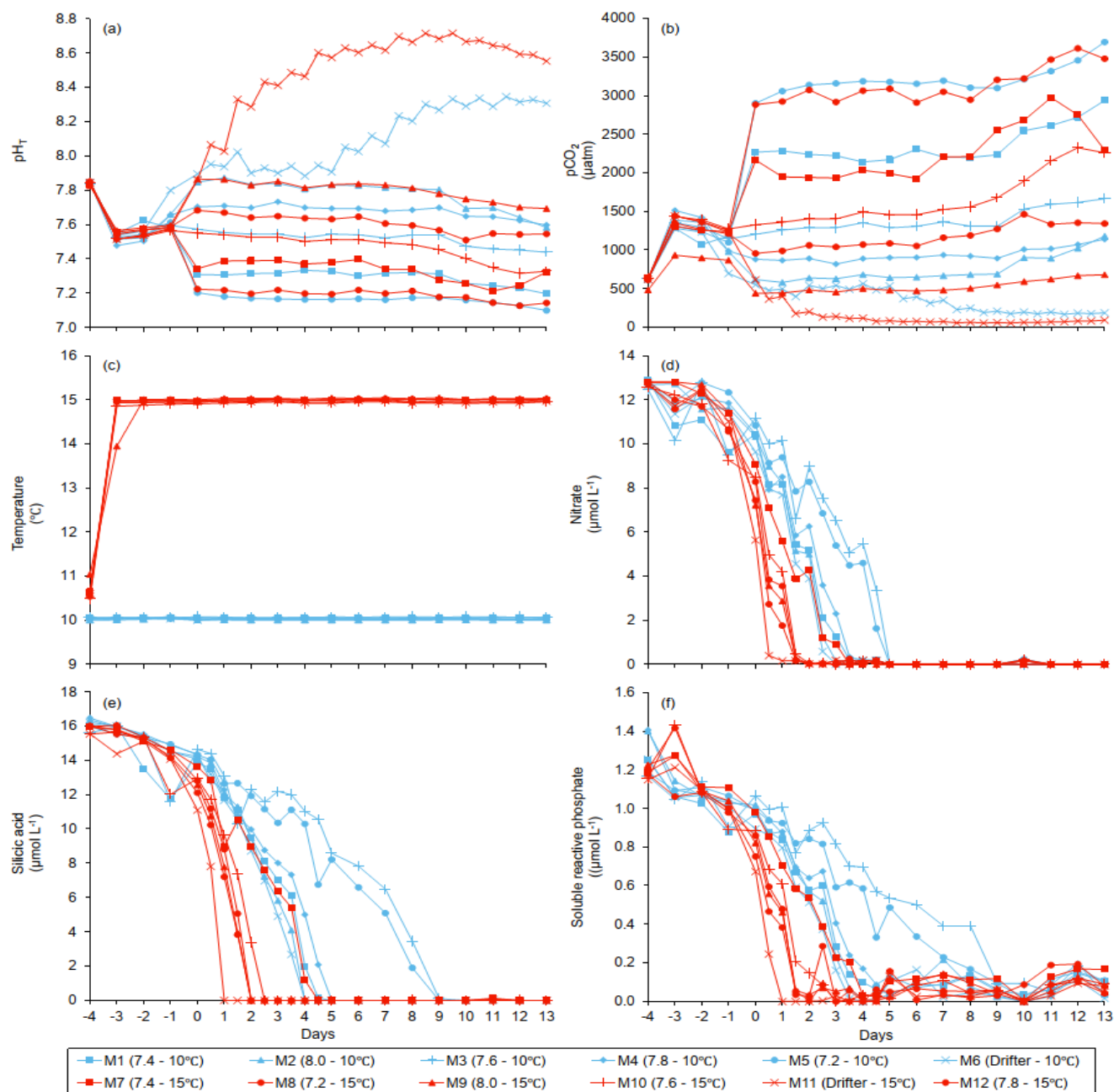
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766 **Figure 2. Changes in incident photosynthetic active radiation (PAR) at the top of the mesocosms level during the experiment as**
 767 **measurement by a Satlantic HyperOCR hyperspectral radiometer and integrated in the 400-700 nm range. Local sunrise and sunset**
 768 **times (EDT) are indicated with the corresponding days of the experiment.**

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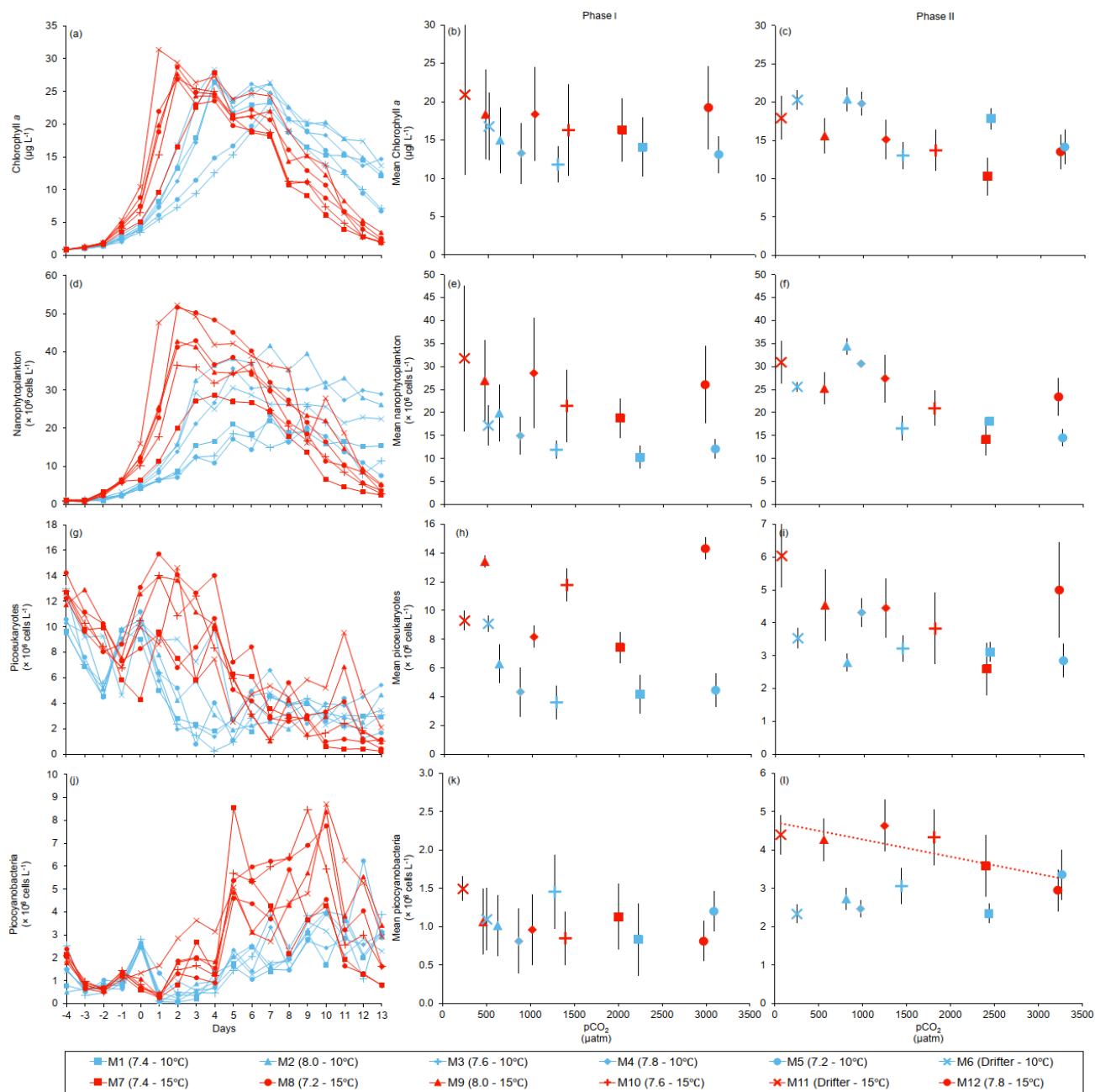
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Figure 3. Temporal variations over the course of the experiment for: (a) pH_r, (b) pCO₂, (c) temperature, (d) nitrate, (e) silicic acid, (f) soluble reactive phosphate. For symbol attribution to treatments, see legend.

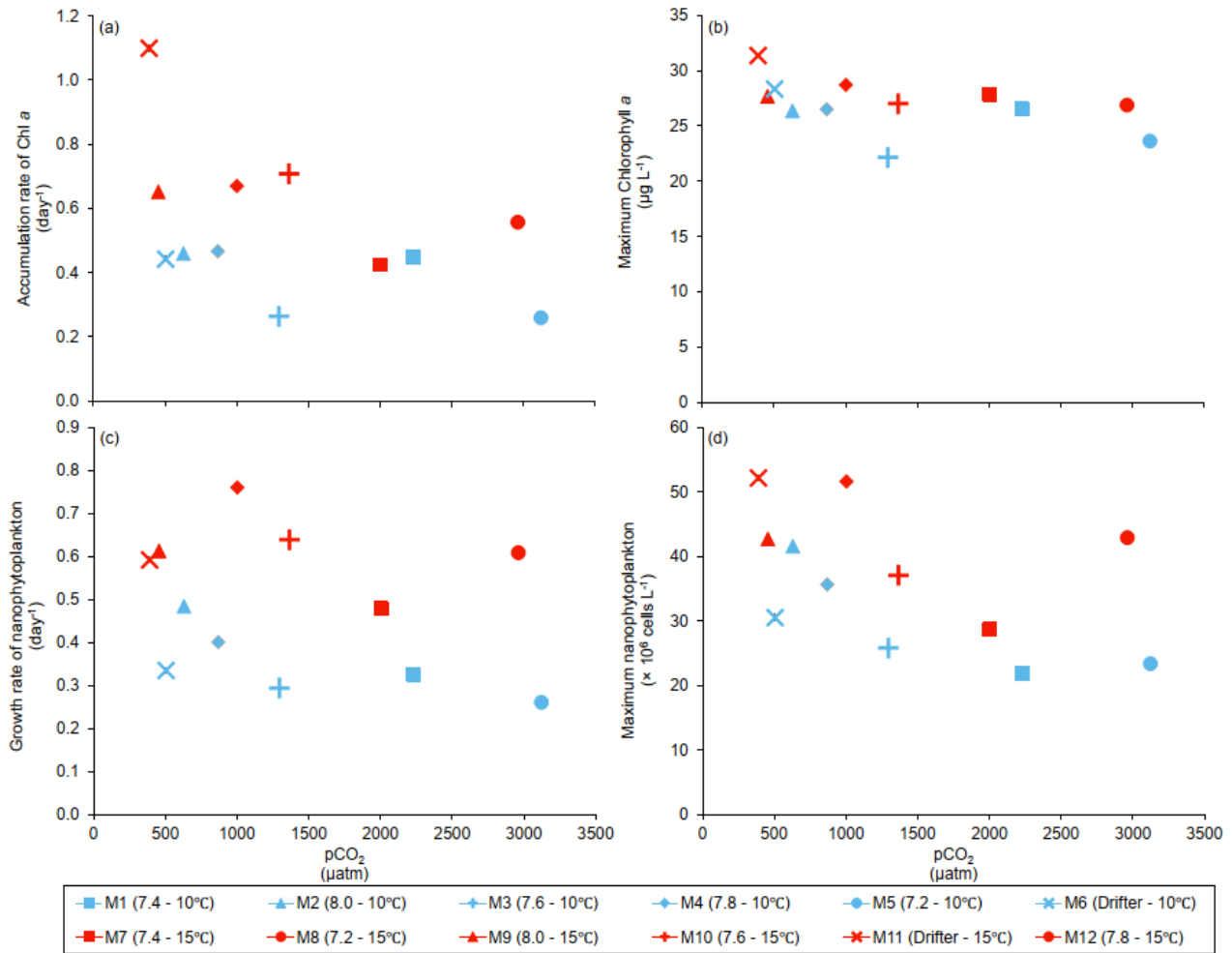
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776 **Figure 4. Temporal variations, and averages \pm SE during Phase I (day 0 to day of maximum Chl *a* concentration) and Phase II (day**
 777 **after maximum Chl *a* concentration to day 13) for: (a-c) chlorophyll *a*, (d-f) nanophytoplankton, (g-i) picoeukaryotes, (j-l)**
 778 **picocyanobacteria. For symbol attribution to treatments, see legend.**

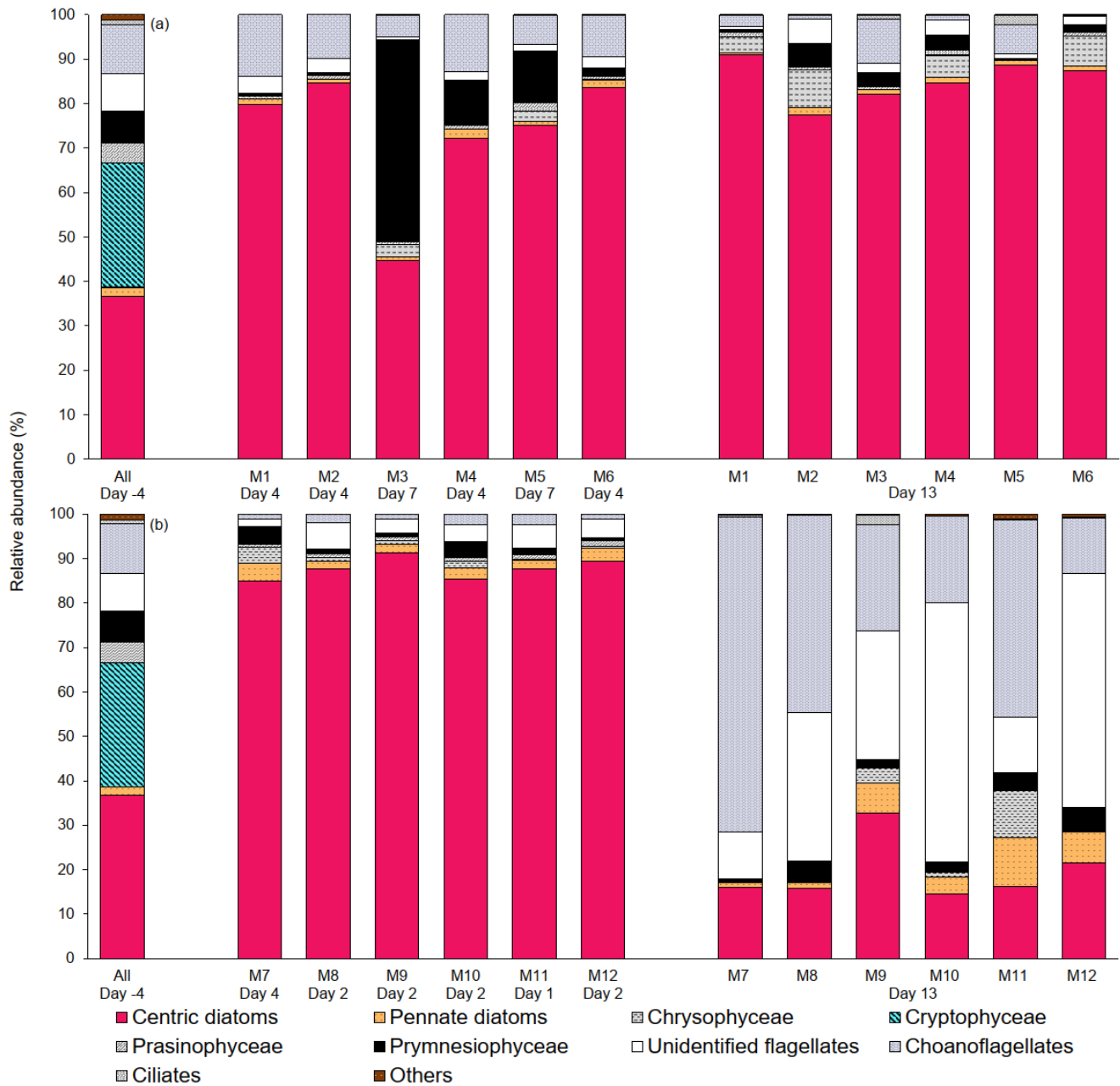
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781 **Figure 5. (a) Accumulation rate of Chl *a* (day 0 to maximum Chl *a* concentration), (b) maximum Chl *a* concentrations, (c) growth**
 782 **rate of nanophytoplankton (day 0 to maximum nanophytoplankton abundance), and (d) maximum nanophytoplankton abundance**
 783 **during the experiment. For symbol attribution to treatments, see legends.**

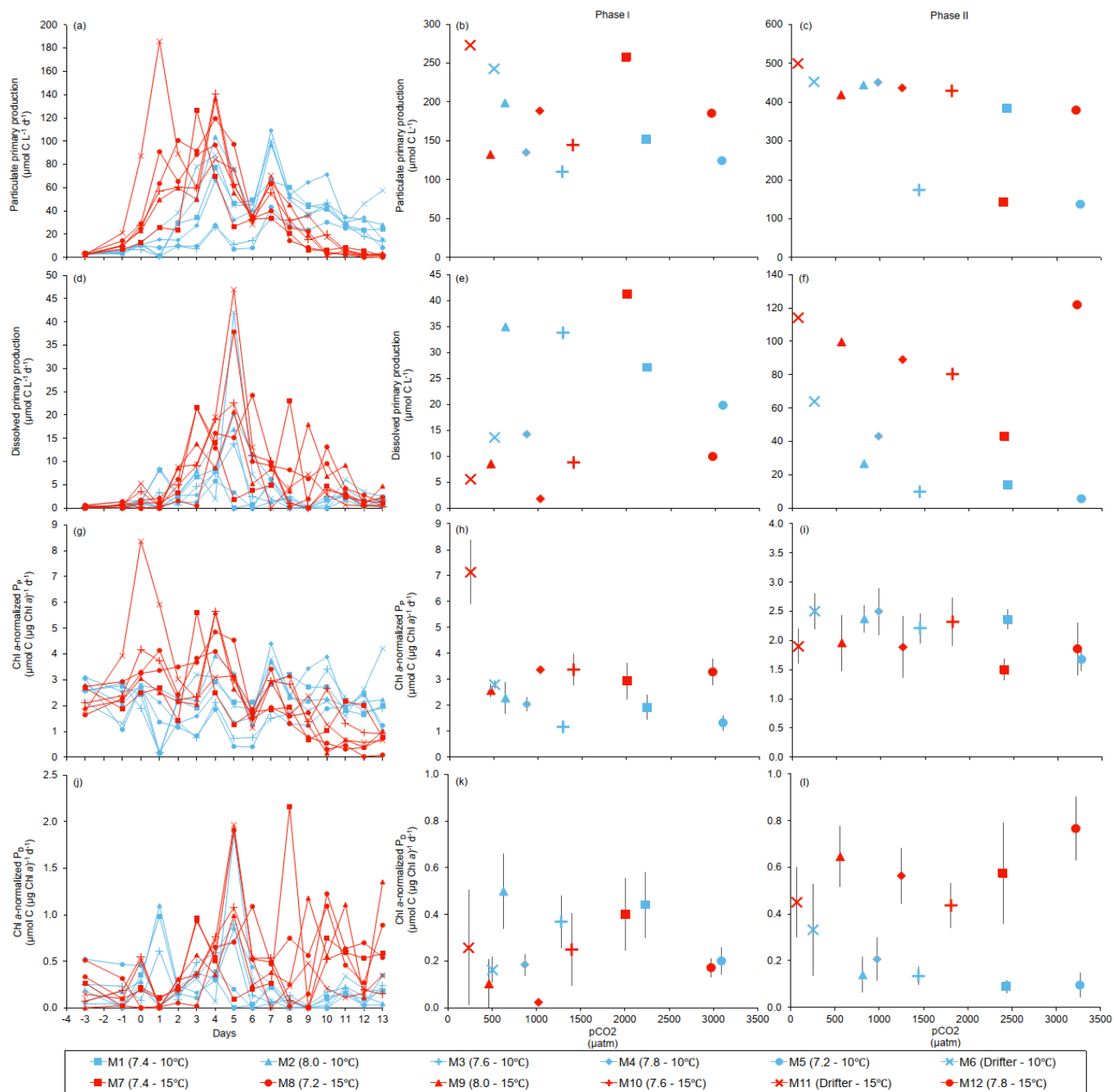
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786 **Figure 6. Relative abundance of 10 groups of protists at the beginning of the experiment (day -4), on the day of maximum Chl *a***
 787 **concentrations in each mesocosm, and at the end of the experiment (day 13) for (a) 10 °C and (b) 15 °C mesocosms. The group**
 788 **« others » include dinoflagellates, Chlorophyceae, Dictyochophyceae, Euglenophyceae, heterotrophic groups, and unidentified cells.**
 789 **Each bar plot represents a mesocosm at a given time. The bar plot on day -4 represents the initial community assemblage before**
 790 **temperature manipulation and acidification, and is therefore the same for each temperature treatment. For symbol attribution to**
 791 **treatments, see legend.**

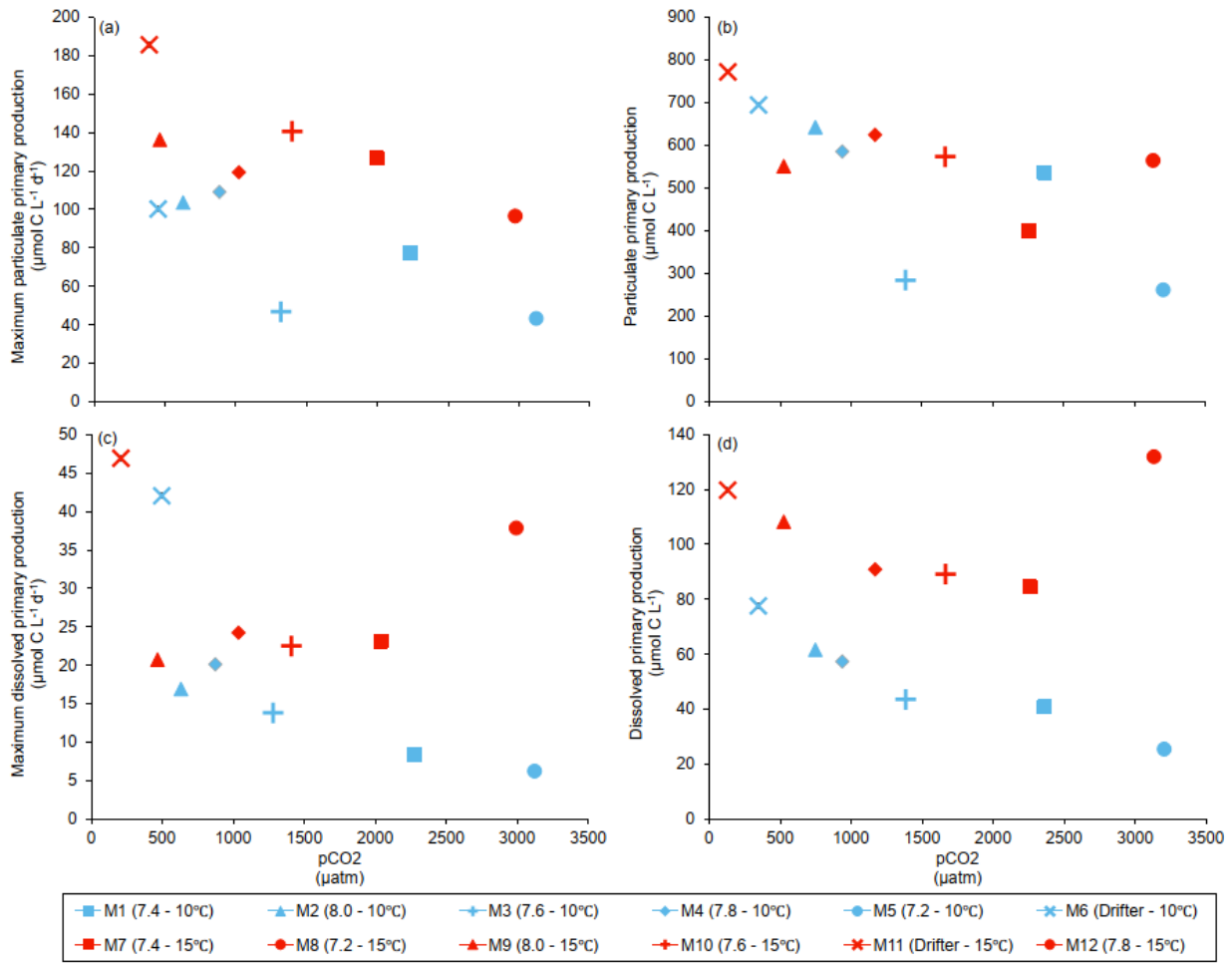
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794 **Figure 7. Temporal variations, time-integrated or averaged \pm SE during Phase I (day 0 to day of maximum Chl *a* concentration)**
 795 **and Phase II (day after maximum Chl *a* concentration to day 13) for: (a-c) particulate primary production, (d-f) dissolved primary**
 796 **production, (g-i) Chl *a*-normalized particulate primary production, (j-l) Chl *a*-normalized dissolved primary production. For**
 797 **symbol attribution to treatments, see legend.**

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800 **Figure 8. (a) Maximum particulate primary production, (b) time-integrated particulate primary production (c) maximum dissolved**
 801 **primary production, and (d) time-integrated dissolved primary production over the full course of the experiment (day 0 to day 13).**
 802 **For symbol attribution to treatments, see legend.**

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Table 1. Day of maximum Chl *a* concentration, the associated average pH_T (total hydrogen ion scale), and average pCO₂ over each individually defined phase. Phase I is defined from day 0 until day of maximum Chl *a* for each mesocosm, while Phase II is defined from the day after maximum Chl *a* until day 13. Average temperature over day 0 to day 13 is also presented for each mesocosm. Average values are presented with ± standard errors.

Mesocosm	Day of max Chl <i>a</i>	Phase I		Phase II		Day 0–13
		pH _T	pCO ₂ (μatm)	pH _T	pCO ₂ (μatm)	Temperature (°C)
M1 (7.4 – 10 °C)	4	7.32 ± 0.01	2231 ± 25	7.28 ± 0.02	2437 ± 92	10.06 ± 0.01
M2 (8.0 – 10 °C)	4	7.84 ± 0.01	628 ± 16	7.74 ± 0.03	814 ± 65	10.00 ± 0.01
M3 (7.6 – 10 °C)	7	7.54 ± 0.01	1294 ± 18	7.48 ± 0.02	1503 ± 64	10.07 ± 0.01
M4 (7.8 – 10 °C)	4	7.71 ± 0.01	868 ± 13	7.66 ± 0.01	976 ± 29	10.04 ± 0.01
M5 (7.2 – 10 °C)	7	7.17 ± 0.01	3122 ± 35	7.15 ± 0.01	3315 ± 94	10.03 ± 0.01
M6 (Drifter – 10 °C)	4	7.93 ± 0.01	503 ± 15	8.22 ± 0.03	251 ± 25	10.02 ± 0.01
M7 (7.4 – 15 °C)	4	7.38 ± 0.01	2004 ± 44	7.31 ± 0.02	2399 ± 120	15.00 ± 0.01
M8 (7.2 – 15 °C)	2	7.21 ± 0.01	2961 ± 58	7.18 ± 0.01	3179 ± 74	15.01 ± 0.01
M9 (8.0 – 15 °C)	2	7.85 ± 0.01	454 ± 13	7.79 ± 0.02	545 ± 25	15.03 ± 0.01
M10 (7.6 – 15 °C)	2	7.54 ± 0.01	1364 ± 22	7.44 ± 0.02	1746 ± 106	14.94 ± 0.01
M11 (Drifter – 15 °C)	1	8.07 ± 0.01	388 ± 90	8.59 ± 0.02	84 ± 7	14.96 ± 0.02
M12 (7.8 – 15 °C)	2	7.67 ± 0.01	1001 ± 31	7.59 ± 0.01	1215 44±	14.98 ± 0.02

808

809 **Table 2. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂, and their interaction during**
 810 **Phase I (day 0 to day of maximum Chl *a* concentration). Separate analysis with pCO₂ as a continuous factor were performed when**
 811 **temperature had a significant effect. Chl *a* concentration, nanophytoplankton abundance, picoeukaryote abundance,**
 812 **picrocyanobacteria abundance, particulate and dissolved primary production, and Chl *a*-normalized particulate and dissolved**
 813 **primary production. Significant results are in bold. *p < 0.05.**

Response Variable	Factor	df	t-value	p-value
Mean Chl <i>a</i> concentration (µg L ⁻¹)	Temperature	8	2.004	0.080
	pCO ₂	8	-0.464	0.655
	pCO ₂ x Temperature	8	0.244	0.813
Mean nanophytoplankton abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	2.725	0.026*
	pCO ₂ (10°C)	4	-2.285	0.084
	pCO ₂ (15°C)	4	-1.191	0.299
Mean picoeukaryote abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	1.056	0.322
	pCO ₂	8	-1.159	0.280
	pCO ₂ x Temperature	8	1.125	0.293
Mean picocyanobacteria abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	0.891	0.399
	pCO ₂	8	0.991	0.351
	pCO ₂ x Temperature	8	-1.166	0.277
Particulate primary production (µmol C L ⁻¹)	Temperature	8	-0.124	0.905
	pCO ₂	8	-1.011	0.342
	pCO ₂ x Temperature	8	0.867	0.411
Dissolved primary production (µmol C L ⁻¹)	Temperature	8	-1.429	0.191
	pCO ₂	8	-0.569	0.585
	pCO ₂ x Temperature	8	0.723	0.490
Chl <i>a</i> -normalized particulate primary production (µmol C (µg Chl <i>a</i>) ⁻¹ d ⁻¹)	Temperature	8	1.689	0.130
	pCO ₂	8	0.107	0.918
	pCO ₂ x Temperature	8	-0.381	0.713
Chl <i>a</i> -normalized dissolved primary production (µmol C (µg Chl <i>a</i>) ⁻¹ d ⁻¹)	Temperature	8	-1.046	0.326
	pCO ₂	8	-0.381	0.713
	pCO ₂ x Temperature	8	0.449	0.665

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815 **Table 3. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂ and their interaction. Separate**
 816 **analysis with pCO₂ as a continuous factor were performed when temperature had a significant effect. Accumulation rate of Chl *a***
 817 **(day 0 to maximum Chl *a* concentration), maximum Chl *a* concentration, growth rate of nanophytoplankton (day 0 to maximum**
 818 **nanophytoplankton abundance), and maximum nanophytoplankton abundance. Significant results are in bold. *p < 0.05.**

Response Variable	Factor	df	t-value	p-value
Accumulation rate of Chl <i>a</i> (day ⁻¹)	Temperature	8	2.679	0.028*
	pCO ₂ (10 °C)	4	-1.476	0.214
	pCO ₂ (15 °C)	4	-1.759	0.154
Maximum Chl <i>a</i> concentration (µg L ⁻¹)	Temperature	8	1.305	0.228
	pCO ₂	8	-0.387	0.709
	pCO ₂ × Temperature	8	0.022	0.983
Growth rate of nanophytoplankton (day ⁻¹)	Temperature	8	2.534	0.035*
	pCO ₂ (10 °C)	4	-0.882	0.403
	pCO ₂ (15 °C)	4	0.601	0.564
Maximum nanophytoplankton abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	1.380	0.205
	pCO ₂	8	-0.735	0.484
	pCO ₂ × Temperature	8	0.302	0.770

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820 **Table 4. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂, and their interaction during**
821 **Phase II (day after maximum Chl *a* to day 13). Separate analysis with pCO₂ as a continuous factor were performed when**
822 **temperature had a significant effect. Chl *a* concentration, nanophytoplankton abundance, picoeukaryote abundance,**
823 **picrocyanobacteria abundance, particulate and dissolved primary production, and Chl *a*-normalized particulate and dissolved**
824 **primary production. Significant results are in bold. *p < 0.05, **p < 0.01, ***p < 0.001.**

Response Variable	Factor	df	t-value	p-value
Mean Chl <i>a</i> concentration (µg L ⁻¹)	Temperature	8	-1.539	0.162
	pCO ₂	8	0.733	0.484
	pCO ₂ x Temperature	8	0.156	0.880
Mean nanophytoplankton abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	-0.528	0.612
	pCO ₂	8	1.264	0.242
	pCO ₂ x Temperature	8	0.699	0.505
Mean picoeukaryotes abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	1.628	0.142
	pCO ₂	8	0.226	0.827
	pCO ₂ x Temperature	8	-0.521	0.617
Mean picocyanobacteria abundance (× 10 ⁶ cells L ⁻¹)	Temperature	8	5.983	<0,001***
	pCO ₂ (10°C)	4	1.480	0.213
	pCO ₂ (15°C)	4	-3.051	0.038*
Particulate primary production (µmol C L ⁻¹)	Temperature	8	-0.015	0.988
	pCO ₂	8	-0.940	0.375
	pCO ₂ x Temperature	8	0.460	0.658
Dissolved primary production (µmol C L ⁻¹)	Temperature	8	1.894	0.095
	pCO ₂	8	-1.145	0.285
	pCO ₂ x Temperature	8	0.847	0.422
(Log) Chl <i>a</i> -normalized particulate primary production (µmol C (µg Chl <i>a</i>) ⁻¹ d ⁻¹)	Temperature	8	-2.288	0.052
	pCO ₂	8	-1.491	0.174
	pCO ₂ x Temperature	8	1.105	0.301
(Log) Chl <i>a</i> -normalized dissolved primary production (µmol C (µg Chl <i>a</i>) ⁻¹ d ⁻¹)	Temperature	8	2.357	0.046*
	pCO ₂ (10°C)	4	-2.573	0.062
	pCO ₂ (15°C)	4	1.345	0.250

825

826 **Table 5. Results of the generalized least squares models (gls) tests for the effects of temperature, pCO₂ and their interaction. Separate**
 827 **analysis with pCO₂ as a continuous factor were performed when temperature had a significant effect. Maximum particulate and**
 828 **dissolved primary production, and time-integration over the full duration of the experiment (day 0 to day 13). Natural logarithm**
 829 **transformation is indicated in parentheses when necessary, significant results are in bold. *p < 0.05, **p < 0.01.**

Response Variable	Factor	df	t-value	p-value
Maximum particulate primary production ($\mu\text{mol C L}^{-1} \text{d}^{-1}$)	Temperature	8	2.466	0.039*
	pCO ₂ (10 °C)	4	-2.328	0.080
	pCO ₂ (15 °C)	4	-2.394	0.075
Time-integrated particulate primary production ($\mu\text{mol C L}^{-1} \text{d}^{-1}$)	Temperature	8	-0.055	0.958
	pCO ₂ (10 °C)	4	-1.300	0.230
	pCO ₂ (15 °C)	4	0.801	0.446
(Log) Maximum dissolved primary production ($\mu\text{mol C L}^{-1}$)	Temperature	8	-0.659	0.528
	pCO ₂	8	-3.342	0.010**
	pCO ₂ × Temperature	8	2.858	0.021*
Time-integrated dissolved primary production ($\mu\text{mol C L}^{-1}$)	Temperature	8	1.687	0.130
	pCO ₂	8	-2.153	0.063
	pCO ₂ × Temperature	8	1.880	0.097

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