



1 **Carbon Dioxide and Methane Emissions from Red Sea Mangrove**
2 **Sediments**

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10

11 **Abstract**

12 Mangrove forests are highly productive tropical and subtropical coastal systems that provide a
13 variety of ecosystem services, including the sequestration of carbon. While mangroves are
14 reported to be the most intense carbon sinks among all forests, their sediments can also support
15 large emissions of greenhouse gases (GHG), such as carbon dioxide (CO₂) and methane (CH₄),
16 to the atmosphere. However, data derived from arid mangrove systems like the Red Sea are
17 lacking. Here, we report emission rates of CO₂ and CH₄ from mangrove sediments along the
18 Saudi Arabian coast of the Red Sea, and assess the relative role of these two gases in supporting
19 total GHG emissions. Diel CO₂ and CH₄ emission rates in Red Sea mangrove sediments ranged
20 from -3452 to 7500 μmol CO₂ m⁻² d⁻¹ and from 0.9 to 13.3 μmol CH₄ m⁻² d⁻¹, respectively. The
21 rates reported here fall within previously reported ranges for both CO₂ and CH₄, but maximum
22 CO₂ and CH₄ flux rates in the Red Sea are 10 to 100-fold below those previously reported for
23 mangroves elsewhere. Based on the isotopic composition of the CO₂ and CH₄ produced by
24 mangrove sediments, we identified the origin of the organic matter that supports GHG emissions.
25 In most of the mangrove stands, GHG emissions were supported by organic matter from mixed



26 sources while only in one mangrove stand the GHG emissions were supported by organic matter
27 derived from mangrove tissues. Moreover, the organic matter derived from mangrove tissues
28 reduced CO₂ fluxes and enhanced CH₄ production, pointing out the importance of the origin of
29 the organic matter in GHG emissions. Methane was the main source of CO₂-equivalents, despite
30 the comparatively low emission rates, in most of the sampled mangroves, and therefore deserves
31 careful monitoring in this region. Despite the mean net emission of CO₂ and CH₄ by Red Sea
32 mangroves reported here, these forests become net organic carbon sinks when taking into
33 account the existing carbon burial rates for the Red Sea mangroves. By further resolving GHG
34 fluxes in arid mangroves, we will better ascertain the role of these forests in global carbon
35 budgets.

36

37



38 **1 Introduction**

39 Mangrove forests, typically growing in the intertidal zones of tropical and subtropical coasts, are
40 highly productive components of coastal ecosystems and adapted to high salinity and anoxic
41 conditions associated with waterlogged sediments. Mangrove forests cover a global estimated
42 area of 137,760 km² (Giri et al., 2011) and are typically constrained by temperature, with
43 greatest biomass and species diversity in the equatorial zone (Alongi, 2012). Mangroves rank
44 amongst the most threatened ecosystems in the biosphere, with losses estimated at 50% of their
45 global extent over the past 50 years (Giri et al., 2011). These losses affect all mangrove regions,
46 except for the Red Sea, where mangrove coverage has increased by 13% over the past four
47 decades (Almahasheer et al., 2016).

48

49 Loss of mangrove forest represents a loss of valuable ecosystem services, including habitat and
50 nursery for marine species, coastal protection from erosion due to wave action, and the filtration
51 of harmful pollutants from terrestrial sources (Alongi, 2008), as well as loss of CO₂ sink capacity
52 and a source of emissions of greenhouse gases (GHG) from disturbed soil carbon stocks (Donato
53 et al., 2011; Alongi, 2014). Hence, mangrove conservation and restoration have been proposed as
54 important components of so-called Blue Carbon strategies to mitigate climate change (Duarte, et
55 al., 2013). Indeed, mangroves are reported to be the most intense carbon sinks among all forests,
56 supporting carbon sequestration rates and organic carbon stocks as much as five times higher
57 than those in terrestrial forests (Donato et al., 2011). While mangrove forests cover less than 1%
58 of total coastal ocean area, they contribute to almost 15% of total carbon sequestration in coastal
59 ecosystems (Alongi, 2012), making mangrove forests highly effective in terms of carbon
60 sequestration per unit area. The management of mangroves to maximize CO₂ removal and
61 subsequent storage is gaining momentum as a cost-effective strategy to mitigate climate change.

62

63 However, mangroves act as both carbon sinks and sources, as their sediments have been reported
64 to support large GHG emissions, in the forms of CO₂ and CH₄ (Allen et al., 2007; Kristensen et
65 al., 2008a; Chen et al., 2016). Whereas concerns are focused on GHG emissions following



66 mangrove disturbance, estimated at 0.02 – 0.12 Pg C yr⁻¹ globally (Donato et al., 2011),
67 undisturbed mangrove sediments also support GHG emissions (Purvaja and Ramesh, 2000;
68 Kristensen et al., 2008b; Chauhan et al., 2015). Previous studies on GHG emission rates from
69 mangrove sediments show highly variable fluxes, with mangroves reported to act from negligible
70 (Alongi, 2005) to considerable sources (Livesley and Andrusiak, 2012; Chen et al., 2016).
71 Comparisons of carbon sequestration rates between mangrove stands have revealed that climatic
72 conditions play an important role, with mangroves in the arid tropics, such as those in the Red
73 Sea, supporting the lowest carbon sequestration rates (Almahasheer et al. 2017). Likewise, GHG
74 emissions from mangrove forests may vary with climate, with most reported rates to-date derived
75 from the wet tropics (Alongi et al., 2005; Chauhan et al., 2015; Chen et al., 2016). Whereas Red
76 Sea mangroves are considered to play a minor role as CO₂ sinks, their role may be greater than
77 portrayed by low carbon burial rates if they also support very low GHG emissions, thereby
78 leading to a balance comparable to mangroves in the wet tropics.

79 Here we report emission rates of CO₂ and CH₄, along with their carbon isotopic composition,
80 from mangrove sediments along the Saudi coast of the Red Sea, and assess the relative role of
81 these two gases in supporting total GHG emissions as well as their fluctuations along the day-
82 night cycle.

83

84 **2 Materials and Methods**

85 **2.1 Study area**

86 We sampled seven mangrove forests along the eastern coast of the Red Sea (Fig. 1). We
87 collected triplicate sediment cores (translucent PVC tubes, 30.5 cm in height and 9.5 cm in
88 diameter) at each station in order to measure CO₂ and CH₄ fluxes. Additionally, we analyzed
89 sediment chlorophyll *a* and nutrient (organic carbon and nitrogen) content. Mangrove sediments
90 were sampled five to ten meters from the forest edge, typically in the center of the mangrove
91 belt. We sampled two stations (S1 and S2) in January and February 2017 and the other five
92 mangrove stations (S3-S7) in March on board the R/V Thuwal as part of a scientific cruise. The
93 sediment cores collected from S1 and S2 were immediately transported to the laboratory and



94 placed in seawater baths and enclosed in environmental growth chambers (Percival Scientific
95 Inc., Perry, IA, USA) with 12h-light:12h-dark cycles at a constant temperature of 26°C. The
96 sediment cores collected during the scientific cruise were transported immediately on board and
97 placed in open aquarium tanks with running seawater in order to keep them close to *in situ*
98 temperature. Salinity and temperature were routinely recorded using a CTD.

99

100 **2.2 Sediment characteristics**

101 The chlorophyll *a* content of the sediment was measured by fluorometry. The surface layer of
102 each replicate core was collected and frozen until further analysis. Prior to chlorophyll *a*
103 extraction, the sediment samples were left at room temperature to thaw. The chlorophyll *a* was
104 extracted by adding 7 ml of 90% acetone to 2 ml of sediment sample. After a 24h incubation at
105 4° C in dark conditions, the samples were centrifuged and the chlorophyll *a* content in the
106 supernatant was measured on a Trilogy fluorometer. The nutrient (organic carbon and nitrogen)
107 content of the sediment was analyzed on an Organic Elemental Analyzer (Flash 2000) after
108 acidification of sediment samples.

109

110 **2.3 Measurement of greenhouse gas fluxes**

111 We measured CO₂ and CH₄ fluxes in mangrove sediments using two different techniques. The
112 CO₂ and CH₄ fluxes from S1 and S2 were measured using the closed water circuit technique and
113 the CO₂ and CH₄ fluxes from the rest of the stations sampled during the scientific cruise (S3-S7)
114 were measured using the headspace technique.

115

116 **2.3.1 Measurement of CO₂ and CH₄ fluxes in sediment core incubations using** 117 **closed water circuit technique**



118 We incubated mangrove sediment cores from stations S1 and S2 using a closed water circuit
119 technique in order to measure changes in CO₂ and CH₄ concentrations. Before starting the
120 incubation, the seawater above the sediment from each core was replaced by fresh seawater
121 collected from the same location, avoiding disturbance of the sediment. Then, the seawater from
122 the core was recirculated by a peristaltic pump in an enclosed water circuit through a membrane
123 equilibrator (Liqui-cel mini module, 3M, Minnesota, USA). This setup enables the equilibration
124 of gases in dissolution with an enclosed air circuit. The air from the enclosed air circuit was then
125 passed through a desiccant (calcium sulfate, WA Hammond Drierite Co., LTD, Ohio, USA)
126 column and flowed into a cavity ring-down spectrometer (CRDS; Picarro Inc., Santa Clara, CA,
127 USA) to continuously measure the CO₂ and CH₄ concentration. We ran the incubations for at
128 least 30 minutes under light (200 μmol photons m⁻² s⁻¹) and dark conditions.

129 The concentration of CO₂ in the water circuit (μmol ml⁻¹) was calculated by Eq. (1):

$$130 \quad [\text{CO}_2] = H_{cp} \times [\text{HP_CO}_2] \times (1 - p_{\text{H}_2\text{O}}), \quad (1)$$

131 where H_{cp} is the Henry constant (mol ml⁻¹ atm⁻¹) calculated using R marelac package (Soetaert
132 et al., 2016); $[\text{HP_CO}_2]$ is the given concentration of CO₂ (ppm), and $p_{\text{H}_2\text{O}}$ is the water vapor
133 pressure (atm).

134 The CO₂ fluxes were calculated from the change in CO₂ concentration over time during our
135 incubations, correcting by the seawater volume present in each core. Then, the fluxes were
136 transformed to an aerial basis (μmol m⁻² h⁻¹) by taking into account the core surface area. Finally,
137 the daily fluxes (μmol m⁻² d⁻¹) were calculated by multiplying the CO₂ flux obtained under light
138 conditions by the number of light hours plus the CO₂ flux obtained under dark conditions by the
139 number of dark hours.

140 The CH₄ fluxes were calculated in the same manner as for the CO₂ fluxes, with the exception
141 that the Henry constant was calculated using Eq. (2):

$$142 \quad \beta = H_{cp} \times (RT), \quad (2)$$



143 where H_{cp} is the Henry constant ($\text{mol ml}^{-1} \text{atm}^{-1}$), R is the ideal gas constant (82.057338 atm ml
144 $\text{mol}^{-1} \text{K}^{-1}$), T is standard temperature (273.15 K), and β is the Bunsen solubility coefficient of
145 CH_4 , extracted from Wiesenburg and Guinasso (1979).

146 **2.3.2 Measurement of CO_2 and CH_4 fluxes in sediment core incubations using the** 147 **headspace technique**

148 Mangrove sediment cores from stations S3 to S7 were incubated using a headspace technique in
149 order to measure changes in CO_2 and CH_4 concentrations. Before starting the incubation, the
150 seawater above the sediment from each core was replaced by fresh seawater from the running
151 seawater system, leaving a headspace of 200 ml. Each core was sealed with a stopper equipped
152 with a gas-tight valve serving as a headspace sampling port. The sealed core was left for 1 hour
153 before the first headspace sampling to allow equilibration between seawater and air phases. Each
154 core was sampled with a syringe, withdrawing 15 ml of air from the equilibrated headspace.
155 Headspace samples were periodically drawn from each sediment incubation over a 24-hour
156 incubation period. The CO_2 and CH_4 concentrations in the headspace samples along with their
157 isotopic composition ($\delta^{13}\text{C}\text{-CO}_2$ and $\delta^{13}\text{C}\text{-CH}_4$) were measured with a cavity ring-down
158 spectrometer (CRDS; Picarro Inc., Santa Clara, CA, USA) connected to a small sample isotopic
159 module extension (SSIM A0314, Picarro Inc., Santa Clara, CA, USA). We ran standards (730
160 ppm CO_2 , 1.9 ppm CH_4) before and after every three samples.

161 The concentration of dissolved CO_2 in the seawater after equilibrium was calculated from the
162 concentration in the equilibrated headspace (ppm) as described previously by Wilson et al.
163 (2012) for other gases:

$$164 \quad [\text{CO}_2]_w = 10^{-6} \beta m_a p_{dry}, \quad (3)$$

165 where β is the Bunsen solubility coefficient of CO_2 ($\text{mol ml}^{-1} \text{atm}^{-1}$), m_a is the given
166 concentration of CO_2 in the equilibrated headspace (ppm), and p_{dry} is atmospheric pressure (atm)
167 of dry air. The Bunsen solubility coefficient of CO_2 was calculated using Eq. (4):

$$168 \quad \beta = H_{cp} \times (RT) \quad (4)$$



169 where H_{cp} is the Henry constant ($\text{mol ml}^{-1} \text{atm}^{-1}$) calculated using R marelac package (Soetaert
170 et al., 2016), R is the ideal gas constant ($82.057338 \text{ atm ml mol}^{-1} \text{K}^{-1}$) and T is standard
171 temperature (273.15 K). The atmospheric pressure of dry air (p_{dry}) was calculated using Eq. (5):

$$172 \quad p_{dry} = p_{wet} (1 - \%H_2O) \quad (5)$$

173 where p_{wet} is the atmospheric pressure of wet air corrected by the effect of multiple syringe
174 draws from the same core, applying Boyle's law.

175 The initial concentration of dissolved CO_2 in seawater before equilibrium was then calculated as:

$$176 \quad [\text{CO}_2]_{aq} = ([\text{CO}_2]_w V_w + 10^{-6} m_a V_a) / V_w \quad (6)$$

177 where $[\text{CO}_2]_w$ is the concentration of dissolved CO_2 in the seawater after equilibrium, V_w is the
178 volume of seawater (ml) and V_a is the headspace volume (ml) in the core. Finally, treating the
179 gas as ideal, the units were converted to nM using Eq. (7):

$$180 \quad [\text{CO}_2]_{aq} = 10^9 * p_{dry}[\text{CO}_2]_{aq} / (RT) \quad (7)$$

181 where R is the ideal gas constant ($0.08206 \text{ atm l mol}^{-1} \text{K}^{-1}$) and T is temperature (K).

182 The CO_2 fluxes were calculated from the change in CO_2 concentration over time during our
183 incubations, correcting by the seawater volume present in each core. Then, the fluxes were
184 transformed to an aerial basis ($\mu\text{mol m}^{-2} \text{d}^{-1}$) by taking into account the core surface area. Finally,
185 the day and night fluxes ($\mu\text{mol m}^{-2} \text{h}^{-1}$) were calculated from the change in CO_2 concentration
186 between consecutive samplings during day and night time, respectively.

187 The CH_4 fluxes were calculated in the same manner as for the CO_2 fluxes, with the exception
188 that the Bunsen solubility coefficient of CH_4 was calculated according to Wiesenburg and
189 Guinasso (1979).

190

191 **2.4 Isotopic composition of CO_2 ($\delta^{13}\text{C}$ - CO_2) and CH_4 ($\delta^{13}\text{C}$ - CH_4)**



192 The isotopic signature of the CO₂ and CH₄ produced in the sediment incubations was estimated
193 by conducting keeling plots (Pataki et al. 2003; Thom et la. 2003; Garcias-Bonet and Duarte
194 2017). Briefly, the δ¹³C of the CO₂ and CH₄ produced was extracted from the intercept of the
195 linear regression between the inverse of the gas partial pressure and the isotopic signature.

196

197 **3 Results**

198 The mean (± SE) diel CO₂ and CH₄ emission rates for the seven sites were 372 ± 1309 μmol CO₂
199 m⁻² d⁻¹ and 5.6 ± 1.6 μmol CH₄ m⁻² d⁻¹, respectively. We observed high variability among the
200 seven mangrove forest sites studied, with net CO₂ and CH₄ diel emission rates ranging from -
201 3452 to 7500 μmol CO₂ m⁻² d⁻¹ and from 0.9 to 13.3 μmol CH₄ m⁻² d⁻¹, respectively (Table 1).

202 Mangrove sediments absorbed CO₂ during daytime and emitted CO₂ during night time, with
203 means (± SE) of -54.6 ± 37 μmol CO₂ m⁻² d⁻¹ and 86 ± 120 μmol CO₂ m⁻² d⁻¹ respectively (Table
204 1, Fig. 2). However, in three out of seven sites, heterotrophic activities outbalanced
205 photosynthesis on a 24h basis. At two sites, S3 and S6, we found an increase of the CO₂
206 emissions between day and night, contradictory to the classical daytime primary production–
207 night-time respiration pattern, possibly indicative of a light mediated increase of heterotrophic
208 processes.

209 Methane emissions did not show circadian patterns with linear increases in CH₄ concentration in
210 our incubations (Fig. 2) and with similar light and dark rates (0.26 ± 0.08 and 0.21 ± 0.07 μmol
211 CH₄ m⁻² h⁻¹ (mean ± SE), respectively (Table 1)). In terms of total GHG contribution, the mean
212 CO₂-equivalents (CO₂e) emission was 564 ± 1284 μmol CO₂e m⁻² d⁻¹ (mean ± SE); mangrove
213 sediments were net emitters of CO₂e in three out of seven sites (Table 1), and in five out of seven
214 mangrove stands sampled, CH₄ was the main source of CO₂e to the atmosphere.

215 While no overall trend was revealed through the relationship between day and night fluxes for
216 CO₂ and CH₄ (Fig. 3), consistencies are evident at specific mangrove stations. For example,
217 night CO₂ emissions are clearly visible at S2, while S3 appears to emit CO₂ during daylight
218 hours. No relationship was apparent between GHG fluxes and the densities of organic carbon or



219 nitrogen in the sediment. There was no discernible trend between gas fluxes and chlorophyll *a*
220 content in surface sediments.

221 The isotopic signatures of the produced CO₂ ($\delta^{13}\text{C-CO}_2$) ranged from -11.21 to -25.72 ‰ as
222 derived from keeling plots (Fig. 4, Table 1). The $\delta^{13}\text{C-CO}_2$ was similar for almost all stations,
223 with the exception of S3 that had a $\delta^{13}\text{C-CO}_2$ of -25.72 ‰. The isotopic composition of the
224 produced CH₄ ($\delta^{13}\text{C-CH}_4$) ranged from -71.28 to -87.08 ‰, with a mean $\delta^{13}\text{C}$ signature of -80.61
225 ‰ (Fig. 4, Table 1). The data set is available from Sea et al. (2018).

226

227 **4 Discussion**

228 The CO₂ and CH₄ emissions reported in this study show that Red Sea mangrove sediments can
229 act as a source of GHG to the atmosphere. Values reported from this study fall within previously
230 reported ranges for both CH₄ and CO₂, but maximum CH₄ and CO₂ flux rates in the Red Sea are
231 over 10 to 100 fold below those reported elsewhere (Table 2). The variability in GHG emission
232 rates reported here could be attributed to spatial differences, as cores were taken from different
233 parts of each forest. Previous studies report significant discrepancies in emission rates in fringe
234 versus forest positions (Allen et al., 2007), although this is likely to be a minor source of
235 variability provided the narrow belt Red Sea mangroves typically form, about 30 m, due to the
236 restricted tidal range of about 0.6 m in this region.

237 The uniformity of day and night emission rates for CH₄ we observed in Red Sea mangrove
238 stands is consistent with previous work reporting that emission rates for CH₄ do not vary
239 significantly during light and dark hours in mangrove forests (Allen et al., 2007). It has been
240 suggested instead that variables such as sediment temperature are more significant in their
241 contributions to emission rates (Allen et al., 2007; Allen et al., 2011). Indeed, seasonal studies of
242 longer duration have reported increased emission rates during warmer seasons (Chen et al., 2016;
243 Livesley and Andrusiak, 2012). Methane concentrations typically remain low due to anaerobic
244 methane oxidation processes that take place near sediment surfaces (Kristensen et al., 2008a),
245 consistent with the low CH₄ emission rates from Red Sea mangrove sediments observed here.
246 Additionally, environments of high salinity like the Red Sea have been associated with decreased



247 CH₄ emissions, as sulfate-reducing bacteria are thought to outcompete methanogens
248 (Poffenbarger et al., 2011).

249 There were no relationships between GHG fluxes and sediment properties, such as chlorophyll *a*,
250 nitrogen density, and organic carbon density, suggesting that other factors have greater influence
251 on GHG flux rates in this region. Since mangroves can receive large contributions of organic
252 carbon from other sources (Newell et al., 1995), such as algal mats, seagrass and seaweed, the
253 examination of the isotopic composition of emitted carbon provides insights into the origin of the
254 organic carbon supporting GHG fluxes in mangrove sediments. The isotopic signature of the
255 CO₂ ($\delta^{13}\text{C-CO}_2$) produced by mangrove sediments in four out of the five mangrove stands with
256 available isotopic data was heavier (from -11.2 ± 0.6 to -15.9 ± 1.1 ‰ (Table 1)) than the
257 isotopic signature of mangrove tissues, suggesting the decomposition of organic matter from
258 mixed sources (Kennedy et al. 2010). Specifically, the isotopic signature of the mangroves found
259 in the central Red Sea has been recently reported as $\delta^{13}\text{C}_{\text{leaves}} = -26.98 \pm 0.15$ ‰, $\delta^{13}\text{C}_{\text{stems}} = -$
260 25.75 ± 0.16 ‰ and $\delta^{13}\text{C}_{\text{roots}} = -24.90 \pm 0.17$ ‰ for mangrove leaves, stems and roots while the
261 mean isotopic signature of other primary producers in the central the Red Sea has been reported
262 as $\delta^{13}\text{C}_{\text{seaweed}} = -12.8 \pm 0.5$ ‰ and $\delta^{13}\text{C}_{\text{seagrass}} = -8.2 \pm 0.2$ ‰ for seaweed and seagrass tissues,
263 respectively (Almahasheer et al. 20170). However, in one mangrove stand (S3) the $\delta^{13}\text{C-CO}_2$
264 was much lighter (-25.72 ± 0.21 ‰), indicative of mangrove tissues. Thus, according to the
265 isotopic signature, the CO₂ produced in mangrove sediments would be supported by mangrove
266 biomass in only one mangrove stand out of the five sampled sites with available isotopic data.
267 Moreover, the mean isotopic signature of the CH₄ produced in mangrove sediments ($\delta^{13}\text{C-CH}_4 =$
268 -80.6 ‰) confirms its biogenic origin, which normally ranges from -40 to -80 ‰, depending on
269 the isotopic signature of the organic compounds being biologically decomposed (Reeburgh,
270 2014). The lowest $\delta^{13}\text{C-CH}_4$ was detected in S3, coinciding with the lowest $\delta^{13}\text{C-CO}_2$ value,
271 suggesting that the organic matter being decomposed by methanogens came from mangrove
272 tissues as well.

273 Interestingly, the mangrove with the lightest $\delta^{13}\text{C-CO}_2$ and $\delta^{13}\text{C-CH}_4$ (S3), showed the lowest
274 daily CO₂ flux (-1524 ± 686 $\mu\text{mol CO}_2 \text{ m}^{-2} \text{ d}^{-1}$) but the highest CH₄ emission rate (13.3 ± 9.5
275 $\mu\text{mol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$), compared to the fluxes detected in the rest of mangrove stands with available
276 isotopic data. Part of the variability in the CO₂ ($R^2=0.42$) and CH₄ ($R^2=0.40$) emission rate seems



277 to be explained by the origin of the organic matter being decomposed, estimated here as the
278 $\delta^{13}\text{C-CO}_2$ and $\delta^{13}\text{C-CH}_4$. Organic matter with lighter isotopic composition would enhance CO_2
279 emissions; whereas organic matter with heavier isotopic composition would enhance CH_4
280 emissions (Fig. 5), suggesting a different preferential use of organic matter by different microbial
281 groups in mangrove sediments. This corroborates the importance of the origin of the organic
282 carbon stored in mangrove sediments on their GHG emissions.

283 This study is first in reporting CO_2 and CH_4 fluxes from Red Sea mangrove sediments,
284 contributing to the scant data on arid mangrove systems (Atwood et al. 2017, Almahasheer et al.
285 2017) essential to establish a solid baseline on GHG emissions for future studies. Seasonal
286 variation should be considered in future studies on GHG emissions by Red Sea mangroves to
287 better determine annual emission rates from this system, which reaches some of the warmest
288 temperatures experienced by mangrove forests worldwide. Similarly, a wider spatial coverage
289 within the mangrove forest should be considered to confidently determine net GHG fluxes that
290 can be upscaled to the entire stock of Red Sea mangroves. While many studies use the static
291 chamber technique to determine concentrations of methane and carbon dioxide, the highly
292 sensitive CRDS was able to detect trace gases at very low concentrations, making this device
293 useful in future flux studies. However, it is recognized that inherent differences in reported
294 emissions exist as a result of two different sampling techniques used in this study. Establishing a
295 unified GHG sampling technique for mangrove sediments will aid future researchers in
296 establishing total carbon budgets and accurately informing policymakers of their findings.

297 Methane emission rates from Red Sea mangrove sediments, although quite low, become more
298 substantial when considered in terms of global warming potential. In this study, CH_4 was, despite
299 the comparatively low emission rates, the main source of CO_2e in the majority of sampled
300 mangroves, and therefore deserves careful monitoring in this region. Reported organic carbon
301 burial rates of Red Sea mangroves of $3.42 \text{ mmol C m}^{-2} \text{ d}^{-1}$ (Almahasheer et al. 2017), are 10
302 times larger than the combined average CO_2 and CH_4 emission rates reported here (0.37 mmol C
303 $\text{m}^{-2} \text{ d}^{-1}$), identifying mangrove sediments as net carbon sinks. The balance is lower when
304 computed in terms of CO_2e , the relevant metric in terms of radiative forcing, but still resulted in
305 a net CO_2e removal rate of about $2.9 \text{ mmol C m}^{-2} \text{ d}^{-1}$ by Red Sea mangrove sediments. Whereas
306 our results show that, despite sizeable CH_4 and CO_2 emissions, Red Sea mangrove sediments are



307 net sinks for organic carbon; their contribution to climate change, by either promoting it or
308 mitigating it is not reflected on this balance. The role of Red Sea mangrove sediments in climate
309 change depends on the changes experienced. In contrast to mangrove forests elsewhere around
310 the world, Red Sea mangroves are not in decline and, rather, have expanded 12% from the 120
311 Km² occupied in 1972 to 132 Km² in 2013 (Almhasheer et al. 2016). Hence, Red Sea mangroves
312 would have had a modest, but measurable contribution to climate change mitigation. Moreover,
313 protection measures and further reforestation efforts are being deployed along the Red Sea that
314 will further expand the area of mangroves (Almhasheer et al. 2016). The rationale for conserving
315 mangroves, in the climate change context, is not adequately represented by their net carbon sink
316 capacity when undisturbed, but rather by the emissions resulting from their disturbance. Indeed,
317 previous studies analyzing anthropogenic impacts on methane emission rates from mangrove
318 sediments have shown that disturbance significantly increases methane emissions (Purvaja and
319 Ramesh, 2001; Chen et al., 2011). This provides an additional rationale to conserve, and
320 continue to expand, Red Sea mangroves.
321 *Data availability.* All data will be accessible in the repository Pangea pending manuscript
322 acceptance.

323 *Competing interests.* The authors declare that they have no conflict of interest.

324

325 **Author contribution**

326 MAS, NG-B, VS and CMD designed the study. MAS and NG-B performed the measurements
327 and calculations. MAS, NG-B, VS and CMD interpreted the results. All authors contributed
328 substantially to the final manuscript.

329

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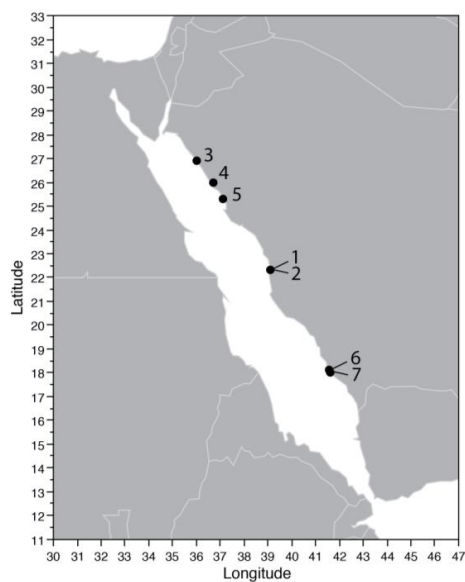
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429

430 **Fig. 1.** Mangrove stands sampled along the Saudi coast of the Red Sea. Numbers indicate
431 positions of sampling sites from this study. S1 and S2: King Abdullah University of Science and
432 Technology; S3: Duba; S4 and S5: Al Wahj; S6 and S7: Farasan Banks.

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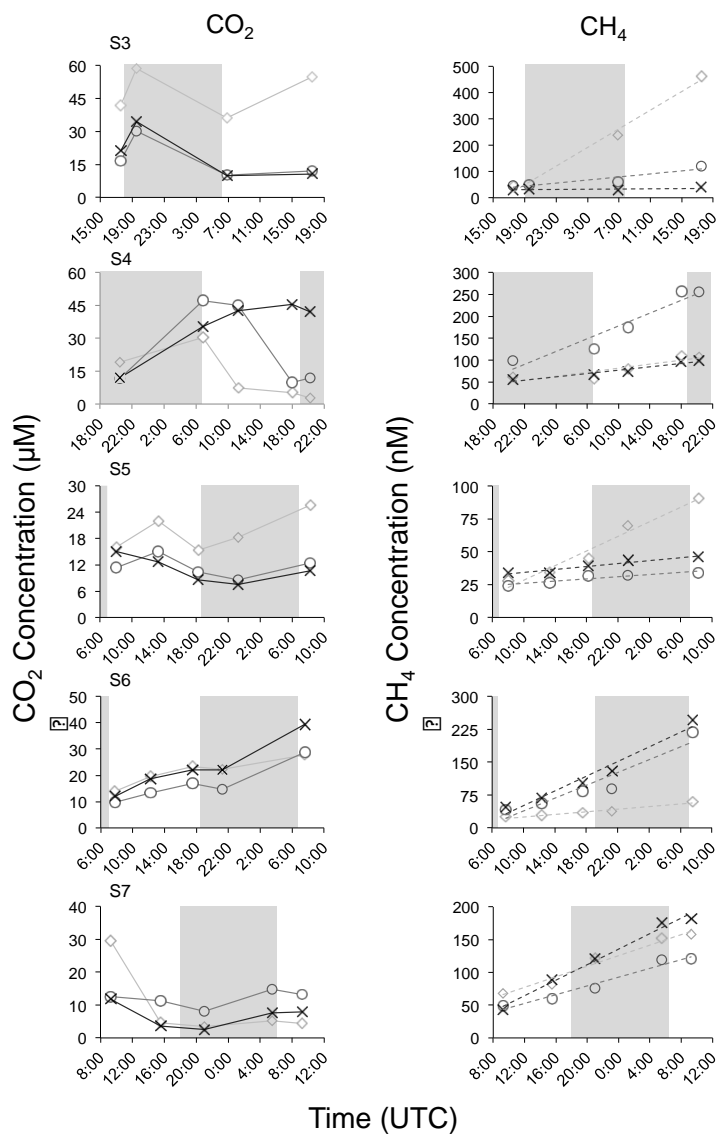
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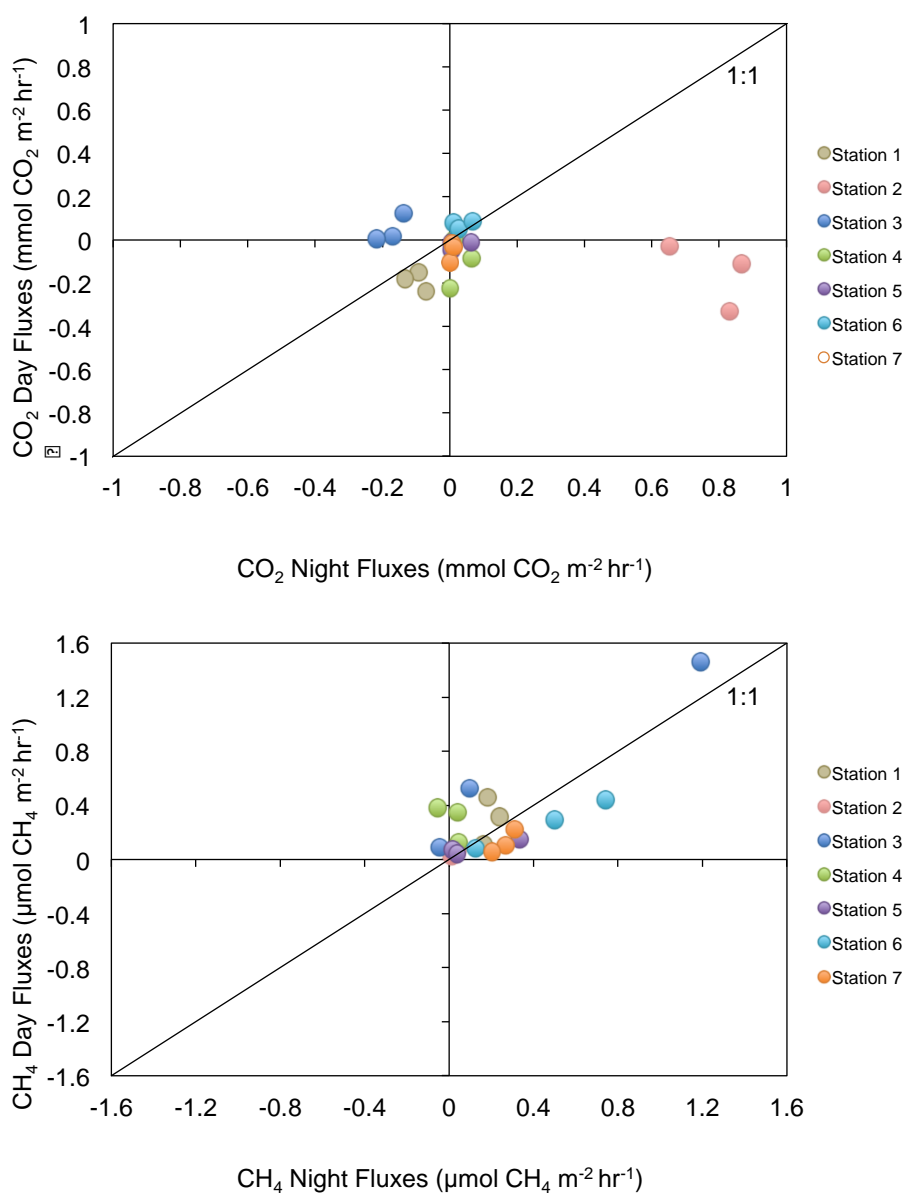
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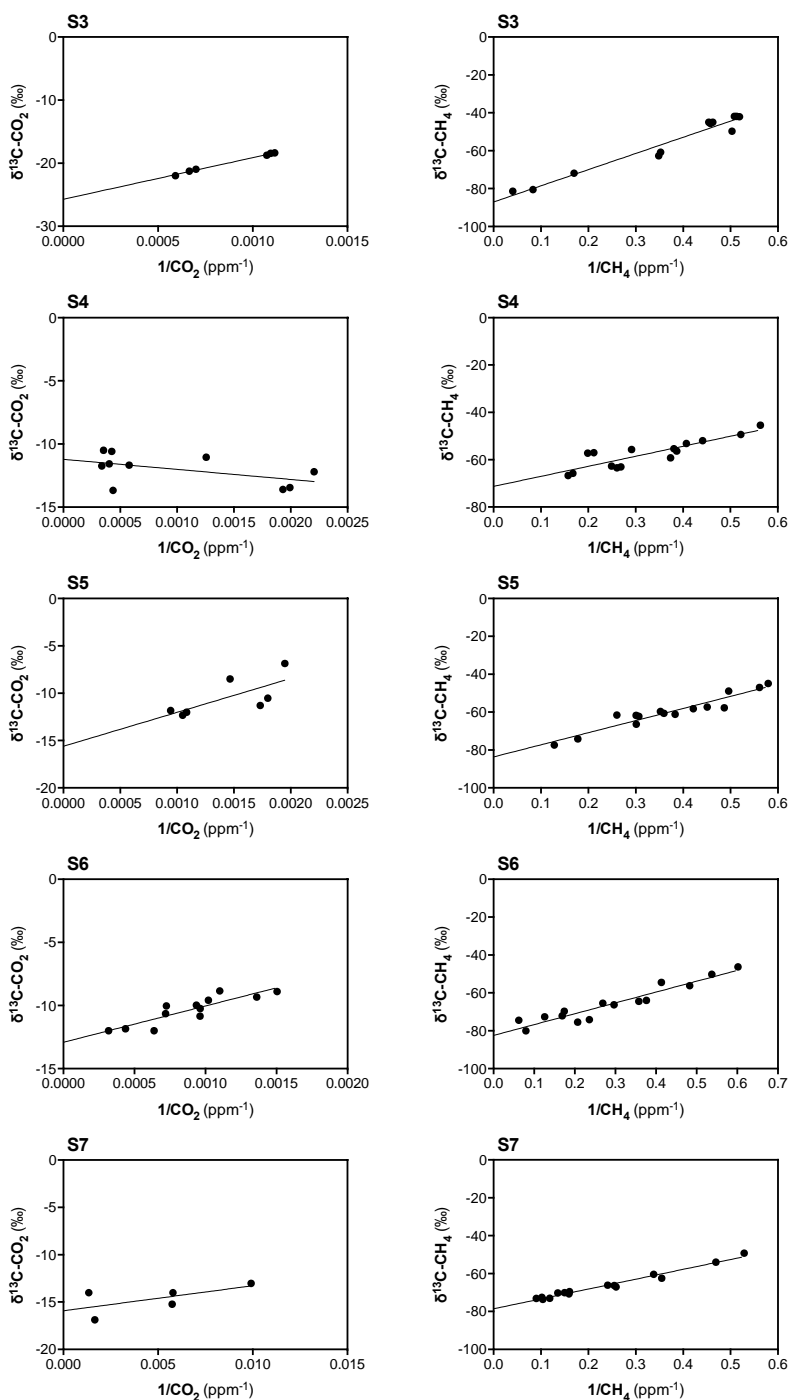
440 **Fig. 2.** Change in CO₂ (left panels) and CH₄ (right panels) concentrations over time in triplicated
 441 mangrove sediment cores from mangrove stations S3-S7. Shaded areas represent night time and
 442 each replicate is coded by different symbols.

443



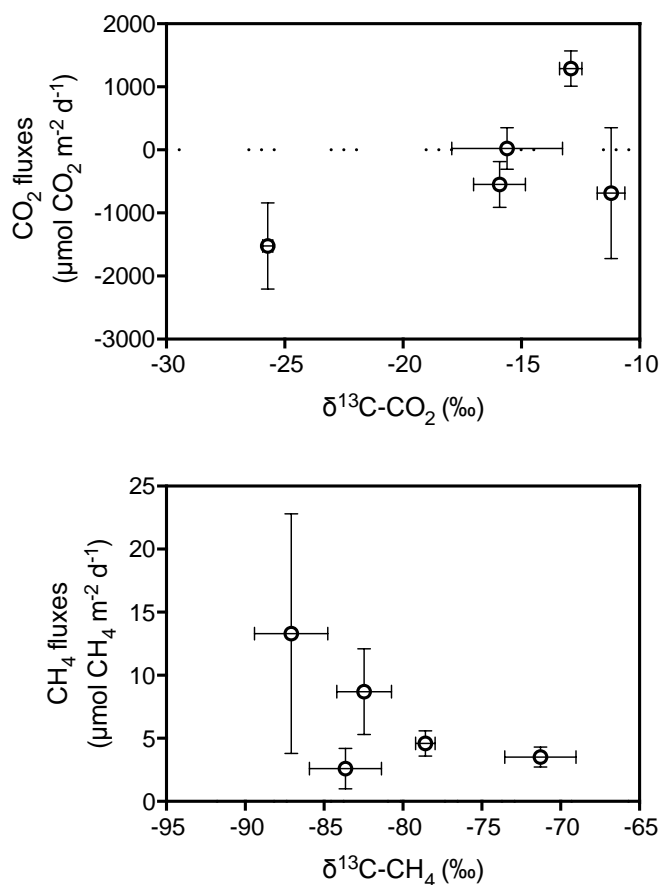
444

445 **Fig. 3.** Relationship between day and night fluxes for CO₂ (top panel) and CH₄ (bottom panel) at
446 all mangrove stations.





448 **Fig. 4.** Keeling plots for mangrove stations S3-S7, showing the linear regression of the inverse of
449 CO₂ concentration (left panels) and CH₄ concentration (right panels) versus $\delta^{13}\text{C}\text{-CO}_2$ and $\delta^{13}\text{C}\text{-}$
450 CH₄. Y-intercepts were used to estimate the isotopic signatures of produced gases.



451

452 **Fig. 5.** Relation between the carbon isotopic signature of the produced CO₂ ($\delta^{13}\text{C}\text{-CO}_2$) and CO₂
453 fluxes (top panel) and carbon isotopic signature of the produced CH₄ ($\delta^{13}\text{C}\text{-CH}_4$) and the CH₄
454 fluxes (bottom panel) in Red Sea mangroves. Error bars indicate standard error of the mean.

455

456



Table 1. Summary of greenhouse gas fluxes and sediment characteristics from studied mangrove forests. CH₄ fluxes in brackets represent CO₂ equivalents in terms of global warming potential for a time horizon of 100 years (GWP₁₀₀), taking into account climate-carbon feedback as suggested by the AR5 of IPCC (Myhre et al., 2013). Data represent the mean ± SEM and nd means no data available.

Station	CO ₂ Day Flux ($\mu\text{mol m}^{-2} \text{hr}^{-1}$)	CH ₄ Day Flux ($\mu\text{mol m}^{-2} \text{hr}^{-1}$)	CO ₂ Night Flux ($\mu\text{mol m}^{-2} \text{hr}^{-1}$)	CH ₄ Night Flux ($\mu\text{mol CH}_4 \text{ m}^{-2} \text{hr}^{-1}$)	Daily CO ₂ Flux ($\mu\text{mol m}^{-2} \text{d}^{-1}$)	Daily CH ₄ Flux ($\mu\text{mol m}^{-2} \text{d}^{-1}$)	$\delta^{13}\text{C-CO}_2$ (‰)	$\delta^{13}\text{C-CH}_4$ (‰)	Nitrogen Density (mg cm^{-2})	C _{org} Density (mg cm^{-2})	Chl <i>a</i> ($\mu\text{g Chl } a \text{ g}^{-1}$ sediment)
1	-188 ± 25	0.30 ± 0.17 [10.2]	-99 ± 18	0.19 ± 0.04 [6.46]	-3452 ± 271	5.9 ± 1.3 [201]	nd	nd	nd	nd	nd
2	-157 ± 89	0.05 ± 0.02 [1.7]	782 ± 66	0.03 ± 0.01 [1.02]	7500 ± 894	0.9 ± 0.25 [31]	nd	nd	nd	nd	nd
3	49 ± 37	0.69 ± 0.4 [23.46]	-176 ± 23	0.42 ± 0.39 [14.28]	-1524 ± 686	13.3 ± 9.5 [452]	-25.7 ± 0.2	-87.1 ± 2.3	1.03 ± 0.05	13.33 ± 1.01	nd
4	-86 ± 79	0.28 ± 0.1 [9.52]	29 ± 19	0.01 ± 0.03 [0.34]	-684 ± 1038	3.5 ± 0.8 [119]	-11.1 ± 0.6	-71.3 ± 2.3	0.80 ± 0.03	8.98 ± 0.86	1.02 ± 0.05
5	-22 ± 11	0.09 ± 0.03 [3.06]	24 ± 20	0.13 ± 0.10 [4.42]	23 ± 331	2.6 ± 1.6 [88]	-15.6 ± 2.3	-83.6 ± 2.3	1.12 ± 0.05	13.34 ± 0.98	1.03 ± 0.04
6	73 ± 10	0.27 ± 0.10 [9.18]	35 ± 17	0.45 ± 0.18 [15.30]	1289 ± 280	8.7 ± 3.4 [296]	-12.9 ± 0.5	-82.5 ± 1.7	1.51 ± 0.14	10.58 ± 0.82	0.43 ± 0.14
7	-51 ± 28	0.13 ± 0.05 [4.42]	5 ± 3	0.26 ± 0.03 [8.84]	-547 ± 363	4.6 ± 1.0 [156]	-15.9 ± 1.1	-78.6 ± 0.6	3.30 ± 0.55	33.43 ± 6.69	1.86 ± 0.12



Table 2. Comparison of GHG fluxes from global mangrove forests and Red Sea mangroves. Literature values converted from reported form for comparison purposes.

		CO ₂ (mmol m ⁻² d ⁻¹)		CH ₄ (μmol m ⁻² d ⁻¹)	
		Minimum	Maximum	Minimum	Maximum
Kristensen, 2008a	Global	2	373	0	5000
Kristensen, 2008b	Tanzania	28	115	0	87.6
Livesley & Andrusiak, 2012	Australia	50	150	50	749
Alongi, 2005	China	17	121	5	66
Alongi, 2014	Global	49	69	0	5100
Chen, 2016	China	-16.9	279.2	-2.1	8015.1
Allen, 2007	Australia	-	-	4.5	25974
Allen, 2011	Australia	-	-	70.3	2348
Chuang, 2015	Mexico	-	-	12	11000
This Study	Red Sea	-3.5	7.5	0.9	13.3