Biogeosciences Discussions



- 1 Reconstructing past variations in environmental conditions and paleoproductivity
- 2 over the last ~8000 years off Central Chile (30° S)
- 3
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# 32 Abstract

33	This study aims at establishing past variations of the main oceanographic and climatic
34	features in the Central Chilean coast, using recent sedimentary records of a transitional
35	semi-arid ecosystem susceptible to environmental forcing conditions. Coquimbo (30°S)
36	region is characterized by dry summers and short rainfall periods during winter. The
37	relatively wet-winter climate results from the interactions between the southern westerly
38	winds and the South Pacific Anticyclone (SPA); in summer, the SPA moves southwards
39	while in winter it returns to the north, allowing the passage of storm fronts. This semi-
40	arid zone is strongly affected by variations associated with El Niño-Southern Oscillation
41	(ENSO), caused by seasonal latitudinal changes in the SPA that produce high variability
42	and precipitation in Chilean mid-latitudes. Sediment cores were retrieved in two bays,
43	Guanaqueros and Tongoy, for geochemical analyses including: sensitive redox trace
44	elements, biogenic opal, total organic carbon (TOC), diatoms, stable isotopes of organic
45	carbon and nitrogen. The results suggest a main dry phase of high primary productivity
46	concomitant with high fluxes of organic compounds to the bottom and suboxic-anoxic
47	conditions in the sediments. This period reached a maximum at cal BC ~4500, followed
48	by a continuous increase in wet conditions, low primary productivity and a more
49	oxygenated environment towards the present, being remarkably stronger in the last 2000
50	years. We suggest that this might be associated with greater El Niño frequencies or
51	similar conditions that increase precipitation, concomitantly with the introduction of
52	oxygenated waters to coastal zones by the propagation of equatorial origin waves.
53	
54	Keywords: paleoproductivity, paleoredox, trace metals, diatoms, opal, organic carbon,
55	ENSO, Coquimbo, SE-Pacific

56





# 57 1. Introduction

58	The northern-central Chilean continental margin (18–30°S) has distinct characteristics
59	in terms of topography, climate and oceanographic conditions, which modulate primary
60	productivity and the chemical composition of the water column. This zone is
61	characterized by several foci of high primary production (0.5-9.3 g C m <sup>-2</sup> d <sup>-1</sup> ; González
62	et al., 1998; Daneri et al., 2000, Thomas et al., 2001) off Iquique (21°S), Antofagasta
63	$(23^{\circ}S)$ and Coquimbo $(30^{\circ}S)$ , resulting from the influx of nutrient-rich waters due to a
64	semi-permanent upwelling forced by local winds. This productivity takes place close to
65	the coast above the narrow continental shelf, allowing for the development of relevant
66	fisheries and accounting for up to 40% of total annual catches in the Chilean margin
67	(Escribano et al., 2004 and references therein).
68	Upwelling areas are recognized as special zones where the considerable amount of
69	organic material produced in the photic zone causes large particle depositions on the
70	bottom. The remineralization of this organic carbon promotes anoxic/suboxic
71	environments where chemosynthetic and authigenic reactions occur. The distribution of
72	chemical compounds present in the sediments and water column are driven by
73	diagenetic reactions, involving metals and nutrients, related to the destruction of organic
74	matter and thus are linked to the diagenetic sequence (oxic or anoxic). In the case of
75	trace metals, these reactions result in soluble or insoluble phases, which are particular
76	for each metal, depending on the oxidative state of the sediments and bottom waters
77	(Chester, 1990; Guieu et al., 1998; Wells et al., 1998).
78	The high productivity observed in these boundary current ecosystems develops and
79	maintains zones of low dissolved oxygen content known as oxygen minimum zones
80	(OMZs). These naturally occurring regions are found in the upwelling areas of the
81	Eastern Boundary Currents, developed along the North and South Pacific Ocean where
82	their intensity, thickness, and temporal stability vary as a function of latitude (Helly and
83	Levin, 2004, Ulloa et al., 2012). To the north (e.g. at 21°S) and off Peru, the OMZ is
84	permanently present, can extend into the euphotic zone and, in the case of northern
85	Chile and southern Peru, it shows no significant interface with the benthic environment
86	due to the presence of a narrow continental shelf (Helly and Levin, 2004).
87	Although ecosystems in Eastern Boundary Currents represent less than 5% of the
88	ocean's surface, they support regions of unique microbial processes that influence
89	global element cycles. The nature of the chemical processes that are taking place in

90 surface sediments and bottom waters establishes the enrichment or depletion of trace





- 91 elements within sediments. This could respond to different causes and is not always
- 92 directly related to the amount of organic carbon that has settled on the bottom.
- 93 Diagenetic reactions could modify the original extension of metal enrichment that
- 94 occurs in the upper sediment layer, reflecting the redox conditions during early
- 95 diagenesis in the sediments (Nameroff et al., 2002) and the environmental conditions of
- 96 the overlying water, i.e. oxygen content (Zheng et al., 2000; McManus et al., 2002;
- 97 Siebert et al., 2003), both intimately related to the biological pump.
- 98 Climatic trends in the last century point to the beginning of a global warming period,
- mainly observed by a  $CO_2$  increase in the atmosphere (Bolin, 1992; Sarmiento and
- 100 Gruber, 2002), and corroborated by studies on ice and coral growth records (Mosley-
- 101 Thompson et al., 2006; Schneider and Steig, 2008; Nurhati et al., 2009). Higher mean
- 102 temperatures produce the stratification of the oceans suppressing vertical nutrient
- transport that sustains primary production, as observed after 1999 (Behrenfeld et al.,
- 104 2006). The El Nino/Southern Oscillation cycles (ENSO) are the most important
- 105 phenomena with a major impact on sea life from phytoplankton to marine mammals-
- and are closely related to changes in atmospheric variables. Their frequency seems to
- 107 have intensified since cal BP 3200 (Marchant et al., 1999), while extreme variations
- 108 have been observed in the last 200 yrs. Heavy rainfall episodes normally occur during
- 109 strong El Niño conditions (Montecinos and Aceituno, 2003), which can also be
- observed in sedimentary records in northern Chile (23°S; Garreaud and Rutllant, 1996;
- 111 Vargas et al., 2007). In addition, the deepening of nutrient-rich Equatorial Subsurface
- 112 Water (ESW) reduces primary productivity in this zone, considered the most productive
- 113 area in the global ocean promoting biogeochemical changes validated through organic
- and inorganic sedimentary records (Muñoz et al., 2012 and references therein).
- 115 Our work focusses on past variations of environmental conditions and marine
- 116 productivity in recent sedimentary records from a transitional semi-arid ecosystem of
- 117 Central Chilean coast, susceptible to oceanographic and climatic forcing. The study area
- (Fig. 1) provides an adequate platform to observe environmental variability within
- 119 different time scales. We suggest that the low continental inputs allow us to clearly
- 120 identify wet/dry as well as high/low export primary production periods because
- sediments in this area receive the organic flux mainly from the upwelling foci in the
- area. Moreover, the organic flux is restricted to the narrow shelf where low oxygen
- 123 content has been detected promoting the preservation of inorganic (trace metals) and
- 124 organic proxies.





# 125 2. Study area

126	The Coquimbo region constitutes a border area between the most arid zones of northern
127	Chile (Atacama Desert) and the more mesic Mediterranean climate of central Chile
128	(Montecinos et al., 2015). Here, the shelf is narrow and several small bays trace the
129	coast line. Favorable winds throughout the year promote an important upwelling center
130	in the southern part of the region (Lengua de Vaca Point), developing high biomass
131	along a narrow coastal area (Moraga-Opazo et al., 2011), and reaching maximum
132	concentrations of ~20 mg m <sup>-3</sup> (Torres et al., 2009). High variability in pigment
133	concentrations have been attributed to high variability in physical processes, which
134	modulate Chl-a distribution along the coast. The bays of Tongoy and Coquimbo
135	register a bipolar circulation consisting of counter rotating gyres: clockwise to the south
136	of the bays and counterclockwise to the north (Moraga-Opazo et al., 2006; 2011).
137	Additionally, large squirts of photosynthetic pigments are observed from the coast to the
138	open ocean with large eddies acting as retention zones counteracting the transport of
139	Chl-a offshore (Rutllant and Montecino, 2002; Marín et al., 2003).
140	The Tongoy and Guanaqueros bays are located at the southern edge of a broad
141	embayment between small islands in the north (29°S; Choros, Damas and Chañaral) and
142	Lengua de Vaca Point in the south (30°S) (Fig. 1), protected from predominant
143	southerly winds. Tongoy Bay is a narrow marine basin (10 km at its maximum width)
144	with a maximum depth of ~100 m. To the northeast lies Guanaqueros Bay, a smaller
145	and shallower basin. Recent oceanographic studies in this zone suggest the relevance of
146	low dissolved oxygen water intrusions from the shelf to shallow waters (Fig. 2), which
147	seem related to sea level decreases resulting from local wind annual cycles at a regional
148	meso-scale (Gallardo et al., 2017)). Furthermore, in the shallow waters of Tongoy Bay,
149	the high primary productivity results in high TOC in the water column that allows for
150	the deposition of fine material on the bottom; TOC increases concurrently with the
151	periods of low oxygen conditions (Fig. 3; Muñoz et al., unpublished data). The spatial
152	and temporal variability of this process is still under study.
153	Sedimentological studies are scarce in the northern-central Chilean shelf. A few
154	technical reports indicate that sediments between 27°S and 30°S are composed of very
155	fine sand and silt with relatively low organic carbon content (<3 and ~5%), except at
156	very limited coastal areas where organic material accounts for around ~16% (Muñoz,
157	unpublished data; FIP2005-61 Report, www.fip.cl). Coastal weathering is the main
158	source of continental input due to scarce river flows and little rainfall in the zone (0.5 to





- 159 ~20 mm yr<sup>-1</sup>; https://es.climate-data.org/location/940/, Fig.1). Freshwater discharges are
- represented by creeks, which receive the drainage of the coastal range forming wetland
- areas in the coast and even small estuaries, as observed in Tongoy and Pachingo. These
- basins cover  $\sim$  300 and 487 km<sup>2</sup>, respectively. The water volume in the estuaries is
- 163 maintained by the influx of seawater mixed with groundwater supply. No surface flux to
- the sea is observed. Therefore, freshwater discharge occurs only during high rainfall
- 165 periods in the coastal zone (DGA, 2011), which normally takes place during El Niño
- 166 years when higher runoff has been recorded in the area (Valle-Levinson et al., 2000)
- 167 normally during the June-July-August quarter (Garreaud et al., 2009). In this scenario,
- 168 marine sediments are often highly influenced by primary production in the water
- 169 column. This organic input is relevant for the sediment composition and, thus,
- 170 sedimentary records could clearly reveal the variability in primary production and
- 171 hence, the oceanographic conditions over the shelf, which ultimately respond to major
- 172 atmospheric patterns.
- 173

### 174 **3. Materials and methods**

# 175 **3.1. Sampling**

- 176 Sediment cores were retrieved from two bays in the Coquimbo region: Bahía
- 177 Guanaqueros (core BGGC5; 30°09' S, 71°26' W; 89 m water depth) and Bahía Tongoy
- 178 (core BTGC8; 30°14' S, 71°36' W; 85 m water depth) (Fig. 1.), using a gravity corer
- 179 (KC-Denmark) in May 2015, on board the L/C Stella Maris II owned by the
- 180 Universidad Católica del Norte. The length of the cores was 126 cm for BGGC5 and 98
- 181 cm for BTGC8. Both cores were cut along the main axis and a general visual
- characterization was done. Different textures and color layers were identified using theMunsell color chart.
- 184 Subsequently, the cores were sliced into 1-cm sections and subsamples were separated
- 185 for grain size measurements, magnetic susceptibility, trace elements, biogenic opal, C
- and N stable isotope signatures ( $\delta^{13}$ C,  $\delta^{15}$ N), and TOC analyses. The samples were first
- 187 kept frozen ( $-20^{\circ}$  C) and then freeze-dried before laboratory analyses.
- 188 The magnetic signal indicates the concentrations and compositions of magnetic
- 189 minerals and is usually used combined with others detrital proxies -as grain size- to
- 190 establish changes in sedimentary processes closely controlled by climatic conditions.
- 191 We considered redox trace elements measurements that respond to local hypoxia (U,
- 192 Mo and Re) as well as nutrient-type elements, which follow the organic fluxes to the





193	sediments (Ba, Ni Cu, P). Additionally, we measured Fe and Mn relevant in adsorption-
194	desorption and scavenging processes of dissolved elements in the bottom water and
195	sediments. We also measured Ca, K and Pb that have relevant continental sources by
196	coastal erosion, weathering and eolian transport, which is also true for Fe and Mn.
197	Besides, Ca accumulation within the sediments depends on the carbonate productivity
198	and dissolution, being used as paleoproductivity proxy (Paytan, 2008; Govin et al.,
199	2012). We use Al as a normalizing parameter for enrichment/depletion of elements due
200	to its conservative behavior. The crustal contribution and the elements are presented as
201	metal/Al ratios. The authigenic enrichment factor of elements was estimated according
202	to: $EF = (Me/Al)_{sample} / (Me/Al)_{detrital}$ ; where $(Me/Al)_{sample}$ is the bulk sample metal (Me)
203	concentration normalized to Al content and the denomination "detrital" indicates a
204	lithogenic background (Böhning et al., 2009). Detrital concentrations ([Me] <sub>detrital</sub> and
205	$[Al]_{detrital}$ ) were established considering the local TM abundance, which is more accurate
206	than using mean Earth crust values (Van der Weijden, 2002). We used the average of or
207	element concentrations at the surface sediments (0-3 cm) of Pachingo wetland (Table
208	1).
209	Diatoms and siliceous microfossils were identified and counted; jointly with biogenic
210	opal content, they constitute our proxies of siliceous export production. Pollen grains
211	were also identified and counted, and used to identify wet and dry environmental
212	conditions. TOC and stable isotopes of organic matter were used to identify the
213	variability of organic fluxes to the bottom and establish biogeocheemical changes in the
214	organic matter remineralization.
215	<sup>14</sup> C and <sup>210</sup> Pb were analyzed at selected sediment levels and foraminifera species were
216	selected for radiocarbon analyses (Table 2).
217	
218	3.2, Geochronology ( <sup>210</sup> Pb and <sup>14</sup> C)
219	<sup>210</sup> Pb activities were quantified through alpha spectrometry of its daughter <sup>210</sup> Po in
220	secular equilibrium with <sup>210</sup> Pb, using <sup>209</sup> Po as a yield tracer (Flynn, 1968). The chemical
221	procedure considered a total digestion of the sediment samples and then autoplated onto
222	silver disks at ~75°C for 3 three hours in the presence of ascorbic acid. The $^{210}$ Po
223	activity was counted in a CANBERRA QUAD alpha spectrometer, model 7404, until
224	the desired counting statistics was achieved (4–10% 1 $\sigma$ errors) in the Chemical
225	Oceanography Laboratory of Universidad de Concepción. <sup>210</sup> Po activity –assumed to be

226 in secular equilibrium with <sup>210</sup>Pb– was calculated using the ratio between natural





radionuclide and the tracer, which is multiplied by the activity of the tracer at the time 227 of plating. The period elapsed between plating and counting produces <sup>210</sup>Po decay (half-228 life: 138 days) and between sampling and plating <sup>210</sup>Pb decay (half-life: 22.3 yr); 229 230 counting was corrected to these elapsed times even when there was a short time period 231 between the collection date and the time of sample analysis (less than one year). Ages were estimated using the inventories of the activities in excess (<sup>210</sup>Pb<sub>xs</sub>, unsupported), 232 based on the Constant Rate of Supply Model (CRS, Appleby and Oldfield, 1978). 233 Unsupported activities were determined as the difference between <sup>210</sup>Pb and <sup>226</sup>Ra 234 activities measured in some sediment column intervals. <sup>226</sup>Ra was measured with a 235 gamma spectrometry at the Laboratoire Géosciences of the Université de Montpellier 236 (France). Standard deviations (SD) of the <sup>210</sup>Pb inventories were estimated propagating 237 counting uncertainties (Bevington and Robinson, 1992) (Table S1, supplementary data). 238 Radiocarbon measurements were performed on a mix of planktonic foraminifera species 239 in core BGGC5 whereas the benthic foraminifera species Bolivina plicata was selected 240 for core BTGC8 (Table 2). Freeze-dried sediment was washed over a 63 µm mesh-size 241 242 sieve and dried after washing at 50°C. At least 2 mg of mixed planktonic foraminifera 243 were picked from the 125-250 µm fraction. The samples were submitted to the National 244 Ocean Sciences AMS Facility (NOSAMS) of the Woods Hole Oceanographic Institution (WHOI). The Fraction Modern (Fm) was corrected by the  $\delta^{13}$ C value, and 245 246 ages were calculated using 5568 (yrs) as the half-life of radiocarbon. The ages were 247 converted to calendar years before present using the calibration curve Calpal2007 HULU (Jöris and Weninger, 1998), and considering an age reservoir effect 248 that compares the <sup>14</sup>C value in the foraminifera core with ages estimated from the CRS 249 model. The deviation from the global mean reservoir age (DR) was estimated 250 subtracting the value corresponding to the age obtained with the CRS model in the 251 marine 13.14c curve (Reimer et al., 2013) from the apparent <sup>14</sup>C age measured at depths 252 of 5 and 10 cm for cores BTGC8 and BGGC5, respectively (Sabatier et al., 2010; Table 253 3). The time scale was obtained according to the best fit of  ${}^{210}$ Pb<sub>xs</sub> curves and  ${}^{14}$ C points, 254 using the CLAM 2.2 software and Marine curve 13C (Reimer et al, 2013), and 255 considering a global reservoir of 441 years (Fig. 4). 256 257

258 3.3. Geophysical characterization





- 259 Magnetic susceptibility  $(SIx10^{-8})$  was measured with a Bartington Susceptibility Meter
- 260 MS2E in the Sedimentology Laboratory at Centro Eula, Universidad de Concepción.
- 261 Mean values from three measurements were calculated for each sample.
- 262 Grain size was determined using a Mastersizer 2000 laser particle analyzer, coupled to a
- 263 Hydro 2000–G Malvern in the Sedimentology Laboratory of Universidad de Chile.
- 264 Skewness, sorting and kurtosis were evaluated using the GRADISTAT statistical
- software (Blott and Pye, 2001), which includes all particle size spectra.
- 266

## 267 **3.4. Trace elements analysis**

268 Trace element analyses were performed by ICP-MS (Inductively Coupled Plasma-Mass 269 Spectrometry) and carried out at Université de Montpellier 2, France (OSU 270 OREME/AETE regional facilities), using an Agilent 7700x. About 50 mg of samples and geochemical reference materials (UBN, BEN and MAG1) were dissolved twice 271 272 through the conventional digestion method using a concentrated HF-HNO<sub>3</sub>-HClO<sub>4</sub> mix (1:1:0.1) in Savillex screw-top Teflon beakers at 120°C, on a hot plate during 48h. 273 274 Following digestion, the samples were subjected to three evaporation steps in order to 275 remove fluorine. Shortly before analysis, samples were dissolved in 2 ml of concentrated HNO<sub>3</sub> and transferred to 20 ml polypropylene bottles. Final sample 276 277 preparation was undertaken by dilution with ultrapure water to a sample-solution weight 278 ratio of 1: 4000-5000 and the addition of a known weight of internal standard solution 279 consisting of 1 ppb of In and Bi. Internal standardization used ultra-pure solution enriched in In and Bi, both elements whose natural abundances in geological samples 280 do not contribute significantly to the added internal standard. This is used to deconvolve 281 mass-dependent sensitivity variations of both matrix and instrumental origin, occurring 282 283 during the course of an analytical session.

Sample introduction uses a peristaltic pump, a micro-nebulizer and a cooled double-284 285 pass Scott type spray chamber. The uptake time (typically 45 s) is set to facilitate stable 286 analyte signals prior to a 120 seconds analysis for each sample. Elements with an atomic mass lower than 80 were analyzed in collision mode using He; heavier elements 287 288 were analyzed in no-gas mode. A wash out procedure consisting of 60 seconds with 289 HNO<sub>3</sub> 10% and 120 seconds with 2% HNO<sub>3</sub> has been found appropriate to achieve 290 instrument blank level. The total time for analysis of a single sample solution is c. 3 291 minutes. Mean concentrations for the analyzed samples were determined by external 292 calibrations prepared daily from multi- and mono-elemental solutions, with





293 concentrations in the range of 0.05-10 ppb for trace elements and of 1-10 ppm for 294 major elements (Ca, K). Polyatomic interferences were controlled by running the 295 machine at an oxide production level <1%. Typical analytical precisions attained by this 296 technique are generally between 1% and 3%, relative standard deviation. Accuracy has 297 been assessed with an analysis of international reference materials and results show 298 agreement generally better than  $\pm 5\%$  with reference values.

299

#### **300 3.5. TOC and stable isotopes**

TOC and stable isotope ( $\delta^{15}$ N and  $\delta^{13}$ C) analyses were performed at the Institut für 301 302 Geographie, Friedrich Alexander Universität (FAU) Erlangen-Nürnberg, Germany. Dry 303 material was placed into tin and silver capsules for N and C analyses respectively, and 304 combusted at 1060° C in a continuous helium flow in an elemental analyzer (NC2500, Carlo Erba), in the presence of chromium oxide and silvered cobalt oxide. The resulting 305 gases, were passed over copper wires at 650° C to reduce nitrogen and excess oxygen. 306 Thereafter, water vapor was trapped with Mg(ClO<sub>4</sub>)<sub>2</sub> and the remaining gases (N<sub>2</sub> and 307 CO<sub>2</sub>) were separated in a gas chromatography column at 45° C. N<sub>2</sub> and CO<sub>2</sub> were 308 passed successively via a ConFloII interface into the isotope-ratio-mass spectrometer 309 310 (Delta Plus, Thermo-Finnigan) and isotopically analyzed. Carbon and nitrogen contents 311 were determined from the peak-area-versus-sample-weight ratio of each individual 312 sample and calibrated with the elemental standards cyclohexanone-2,4-313 dinitrophenylhydrazone (C12H14N4O4) and atropine (C17H23NO3) (Thermo Quest). A laboratory-internal organic standard (Peptone) with known isotopic composition was 314 used for final isotopic calibrations. Stable isotope ratios are reported in the  $\delta$  notation as 315 the deviation relative to international standards (Vienna Pee Dee Belemnite for  $\delta^{13}$ C and 316 atmospheric N<sub>2</sub> for  $\delta^{15}$ N), so  $\delta^{13}$ C or  $\delta^{15}$ N = [(R sample/R standard) - 1] x 10<sup>3</sup>, where R 317 is  ${}^{13}C/{}^{12}C$  or  ${}^{15}N/{}^{14}N$ , respectively. Typical precision of the analyses was  $\pm 0.1\%$  for 318  $\delta^{15}$ N and  $\delta^{13}$ C. 319

320

### 321 **3.6. Biogenic opal**

Biogenic opal was estimated following the procedure described by Mortlock and Froelich (1989) with a slight modification, which consists in extracting 50 mg of sediment with 1 M NaOH (instead of 2 M Na<sub>2</sub>CO<sub>3</sub>) at 85°C for 5 hours. Extraction and analysis by molybdate-blue spectrophotometry (Hansen and Koroleff, 1999) were





conducted at the laboratories of Marine Organic Geochemistry and Paleoceanography,
University of Concepción, Chile. Values are expressed as biogenic opal by multiplying
the Si (%) by 2.4 (Mortlock and Froelich, 1989). Analytical precision was ± 0.5%.
Accumulation rates were determined based on sediment mass accumulation rates and
amount of opal at each core section in %.

331

#### 332 **3.7. Diatoms and siliceous microfossils**

Smear slides for qualitative abundances of siliceous microfossils were carried out every 333 centimeter following the Ocean Drilling Program (ODP) protocol described by 334 335 Mazzullo et al. (1988.) To determine the quantitative abundance of siliceous 336 microfossils (diatoms, silicoflagellates, sponge spicules, crysophyts and phytoliths), ~ 337 0.5 g of freeze-dried sediment was treated according to Schrader and Gersonde (1978). Samples were chosen every ~4, 8 and 12 cm for BGGC5 and at an average of 6 cm for 338 339 BTGC8. Permanent slides were prepared by placing a defined sample volume (0.2 ml) onto microscope slides that were then air-dried and mounted with Naphrax mounting 340 341 medium (refraction index =1.3). Siliceous microfossils were identified and counted under an Olympus CX31 microscope with phase contrast. 1/5 of the slides were counted 342 at 400X for siliceous microfossils and one transect at 1000x was counted for 343 Chaetoceros resting spores. Two slides per sample were counted; the estimated 344 counting error was 15%. Total diatom abundances are given in valves g<sup>-1</sup> of dry 345 346 sediments.

347

#### 348 **3.8. Pollen**

Sample preparation for pollen analysis was conducted following the standard 349 350 methodology for sediment samples (Faegri and Iversen, 1989), which includes 351 deflocculating with 10% KOH, carbonate dissolution with a 5% HCl treatment, silica 352 dissolution with 30% HF, and cellulose removal via acetolysis reactions. Samples were 353 mounted with liquid glycerol and sealed permanently with paraffin wax. Pollen identification was conducted under a stereomicroscope 400 fold magnification with the 354 355 assistance of the Heusser (1971) pollen catalogue. A total of 100-250 terrestrial pollen 356 grains were counted on each sample depending on their abundance. Pollen percentage 357 for each taxon was calculated from the total sum of terrestrial pollen. The percentage of 358 aquatic pollen and fern spores was calculated based on the total terrestrial sum plus the 359 respective group. Pollen percentage diagrams were generated using the Tilia software





- 360 (E. Grimm, Illinois State Museum, Springfield, IL. USA). The diagram was divided into
- 361 "zones" based on the identification of the most important changes in pollen percentage
- and assisted by a cluster ordination (CONISS) performed by the same software.
- We further summarize pollen-based precipitation trends by calculating a Pollen Moisture Index (PMI), which is defined as the normalized ratio between Euphorbiaceae (wet coastal scrubland) and Chenopodiaceae (arid scrubland). Thus, positive (negative) values of this index indicate the relative expansion (reduction) of coastal scrubland under relatively wetter (drier) conditions.
- 368

### 369 **4. Results**

### 370 4.1. Geochronology

- <sup>210</sup>Pb<sub>xs</sub> (unsupported activity) was obtained from the surface down to 8 cm depth in the 371 two cores, with an age of ~cal AD 1860 at 7 cm in both of them (Table S1). Greater 372 surface activities were obtained for core BGGC5 (13.48  $\pm$  0.41 dpm g<sup>-1</sup>) compared to 373 core BTGC8 (5.80  $\pm$  0.19 dpm g<sup>-1</sup>), showing an exponential decay with depth (Fig. 4). 374 A recent sedimentation rate of  $0.11\pm0.01$  cm yr<sup>-1</sup> was estimated. 375 376 The age reservoir estimation was similar for both cores, 441 and 442 years for BGGC5 and BTGC8, respectively (Table 3). The age model provided a maximum age of cal BC 377 6300 for core BGGC5, and cal BC 5760 for core BTGC8 (Fig. 4). A mean 378
- sedimentation rate of 0.02 cm  $yr^{-1}$  was estimated for core BGGC5, with a period of
- relative low values  $(0.01 \text{ cm yr}^{-1})$  between cal BC 4000 and 2000. For BTGC8,

sedimentation rates were less variable and around  $0.013 \text{ cm yr}^{-1}$  in the entire core.

382

### 383 4.2. Geophysical characterization

384 The sediments retrieved from the bays showed fine grains in the range of very fine sand and silt in the southern areas. There, the grain size distribution was mainly unimodal, 385 386 very leptokurtic, better sorted and skewed to fine grain when compared to sediments 387 from the northern areas. Sediment cores obtained from the northern areas were sandy (coarse sand and gravel), with abundant calcareous debris. Longer cores of soft 388 sediment were retrieved at the southern areas (BGGC5 and BTGC8), where the silty 389 component varied between 40 % and 60 % (Fig. 1 and 5a,b). The clay component was 390 very low at both cores (<2%). The sediment's color ranged from very dark gravish 391 brown to dark olive brown (2.5Y 3/3-3/2) at Guanaqueros Bay (BGGC5) and from dark 392 393 olive gray to olive gray (5Y 3/2-4/2) at Tongoy Bay (BTGC8). Visible macro-remains





394 (snails and fish vertebrae) were found and weak laminations were identified at both cores. The magnetic susceptibility showed higher values close to the surface, up to 127 395 SI  $x10^{-8}$  at BGGC5 and relative lower values (85 SI  $x10^{-8}$ ) at BTGC8. At greater depths, 396 however, the values were very constant, around  $5-8 \times 10^{-8}$  SI at BGGC5 core and 397 around 12-20 x10<sup>-8</sup> SI at BTGC8 core. In both cores, susceptibility increases 398 399 substantially after cal AD 1800 (Figs. 5a, 5b). Lower bulk densities were estimated at the core BGGC5 (0.7–0.9 g cm<sup>-3</sup>) compared to the core BTGC8 (>1 g cm<sup>-3</sup>) (Fig. 5a, 400 5b). In accordance with this, the mean grain size was  $60-80 \ \mu\text{m}$  at Guanaqueros Bay 401 402 (BTGC8), compared to 50-60 µm at Tongoy Bay (BGGC5). Both cores were negatively skewed, with values of -1 to -1.2 at BGGC5, and -1 to -2.5 at BTGC8. Minor 403 404 increases towards coarser grain size were observed in the last 2000 years, especially at Tongoy Bay (BTGC8). In both cases, grain size distributions were strongly leptokurtic. 405 Ca/Fe ratio also diminished in time, except at core BTGC8 where it was only observed 406 407 during the last ~3000 years. The diminishing of the Ca/Fe ratio is due to a decrease in Ca content mainly but also because of a slight increase in Fe within the sediments (Figs. 408 409 6a, 6b).

410

#### 411 **4.3. Biogenic components**

#### 412 4.3.1. Siliceous microfossils and biogenic opal

Total diatom abundance fluctuated between  $5.52 \times 10^5$  and  $4.48 \times 10^7$  valves g<sup>-1</sup> in core BGGC5. Total diatom abundance showed a good correlation with biogenic opal content at BGGC5 (R<sup>2</sup> =0.52, P<0.5), with the highest values from 72–74 cm to the bottom of the core, corresponding to cal BC 3390–3790. In contrast, diatom abundance and biogenic opal were much lower in core BTGC8 (< 2 ×10<sup>5</sup> valves g<sup>-1</sup> and <3%, respectively). Here, the siliceous assemblage was almost completely conformed by *Chaetoceros* resting spores (RS) (Fig. 7).

420 A total of 135 and 8 diatom taxa were identified in cores BGGC5 and BTGC8 421 respectively, where the core BTGC8 registered very low abundances of diatoms. In general, diatoms were the most important assemblage of siliceous microfossils (96%), 422 423 followed by sponge spicules (3%). The contribution of phytoliths and chrysophyte cysts was less than 2% in core BGGC5. Chaetoceros (RS) dominated the diatom assemblage 424 (~90%; Fig. 7), and included the species C. radicans, C. cinctus, C. constrictus, C. 425 vanheurckii, C. coronatus, C. diadema, and C. debilis. Other species recorded of 426 427 upwelling group (mainly in core BGGC5) were: Skeletonema japonicum, and





428 Thalassionema nitzschioides var. nitzschioides (Table S2). Freshwater diatoms 429 (Diploneis papula, Cymbella tumida, Fragilaria capucina, Diatoma elongatum) and 430 non-planktonic diatoms (Cocconeis scutellum, C. costata and Gramatophora angulosa) 431 accounted for ~0.1-5%; while the group of coastal planktonic diatoms accounted for 432  $\sim 0.3-6\%$  of the total assemblage. The main planktonic diatoms were (*Rhizosolenia* imbricata, and Thalassiosira eccentrica). Oceanic-warm diatoms (Roperia tesselata, 433 Th. nitzschioides var inflatula) and the tycoplanktonic diatom group were rare with less 434 435 than 1%.

436

### 437 **4.3.2. TOC and stable isotopes distribution**

438 Consistent with opal and diatoms, core BGGC5 showed higher values of TOC 439 (between 2 % and 5 %) compared with less than ~1.5 % in core BTGC8 (Fig. 5a,b). 440 Furthermore,  $\delta^{13}$ C was slightly higher at core BTGC8 (-20 ‰ to -21 ‰) compared with core BGGC5 (-21 ‰ to -22 ‰), the former also showing slightly increased 441 values of  $\delta^{15}$ N from the deeper sections to the surface of the core (<7 ‰ to >10 ‰). 442 This increase was less evident at core BGGC5, with values of ~9 ‰ at depth to >10 ‰ 443 on the surface (Fig. 5a,b). Diminishing TOC contents was related to slightly higher 444  $\delta^{13}$ C values (~ -20 ‰) in both cores. 445

446

#### 447 **4.3.3. Pollen record**

Initial surveys on core BTGC8 (Tongoy Bay) revealed extremely low pollen abundances which impeded further palynology work. A complete pollen analysis was only conducted for core BGGC5 (Guanaqueros Bay). The pollen record of core BGGC5 consisted of 29 samples shown in Figure 8. The record was divided in five general zones following visual observation of changes in the main pollen types and also assisted by the cluster analysis CONISS.

Zone BG-1 (cal BC 6250 – cal BC 5650): This zone is dominated by the herbaceous
taxa Chenopodiaceae, Ast. *Leucheria-type*, Ast. Asteroideae, Apiaceae with relatively

456 high values of the wetland genus *Typha spp*.

Zone BG-2 (cal BC 5650 – cal BC 4600): This zone is also dominated by
Chenopodiaceae, Ast. *Leucheria-type* and Ast. Asteroideae. In addition, other nonarboreal elements such as Ast. *Ambrosia-type*, Poaceae, Brassicaceae and *Chorizanthe spp.* expand notably.





- 461 Zone BG-3 (cal BC 4600 cal BC 1450): This zone is marked by a steady decline in
- 462 Chenopodiaceae and Ast. Leucheria-type, and by the expansion of several other
- 463 herbaceous elements, such as Euphorbiaceae, Ast. *Baccharis-type* and Brassicaceae.
- 464 Zone BG-4 (cal BC 1450 cal AD 1800): This zone is mostly dominated by Ast.
- 465 Asteroideae, and marked by the decline of Chenopodiaceae and Ast. Leucheria-type.
- 466 Other coastal taxa such as Euphorbiaceae, Ast. Baccharis-type, Ast. Chichorioideae,
- 467 *Quillaja saponaria*, Brassicaceae and *Salix spp.* also expand in this zone.
- Zone BG-5 (cal AD 1800 cal AD 2014): The upper portion of the record is
  dominated by Ast. Asteroideae and Poaceae, and marked by increments of
  Geraniaceae, Ast. Mutisieae, Myrtaceae and *Q. saponaria*. Additionally, this zone
  includes introduced pollen types such as *Rumex* and *Pinus*. The latter is not shown in
  the diagram of Figure 8 because its abundance was minimal.
- The most distinctive change revealed in core BGGC-5 is a long-term decrease in Chenopodiaceae and increments in Euphorbiaceae and Ast. Asteroideae. Along with this trend, a later expansion of several other representatives of the coastal shrub land starts at around cal BC 4600.
- 477

#### 478 **4.4. Trace element distributions**

479 Trace element distributions are shown in figures 6a and 6b for Guanaqueros (BGGC5) and Tongoy Bays (BTGC8), respectively. Trace metals sensitive to the presence of 480 481 oxygen (U, Re, Mo) showed increasing metal/Al ratios from the base of the core (cal BC ~6300) until cal BC 4500 in core BGGC5. After this maximum, ratios presented a 482 slight increase towards the beginning of the recent era (cal AD 1) followed by a sharp 483 decrease until present. Similarly, the metal ratios in the core BTGC8 increase over 484 485 time, yet the maximum was observed at cal AD 1000. The exception of this trend was 486 Mo which exhibited maximum values until cal BC 4000 and then a steady decrease 487 towards the present. Additionally, metal/Al values were higher at core BGGC5. Iron 488 displayed a clear increase around cal BC 2000 at core BGGC5, and at core BTGC8 was observed a slight increase after cal BC ~3000. Manganese did not show a clear 489 trend. 490

A second element group (metal/Al ratios), including Cd, Ni and P (related to primary
productivity and organic fluxes), showed a similar pattern than Mo/Al towards the
bottom of core BGGC5, i.e. the highest values around cal BC 4500 and a constant
reduction towards the present. A third group, consisting of Ba, P and Ca, exhibited a





less clear pattern. The Cd/Al and Ni/Al ratios in core BTGC8 showed only slightly
decreasing values, and the maximum values were very low compared to the BGGC5
core. The same pattern is observed for other elements. Metal/Al ratios for Ba, Ca and
P were lower and presented a long-term decreasing pattern towards the present.
An exception to the previously described patterns was Cu/Al, which peaked at cal BC
2000 and showed a conspicuous increase in the past ~150 years. This was also

observed at core BTGC8, but with lower concentrations than at core BGGC5.

502

### 503 5. Discussion

### 504 5.1. Sedimentary composition of the cores: terrestrial *versus* biogenic inputs

505 The sediments in the southern zones of the bays constitute a sink of fine particles 506 transported from northern areas and the shelf (Fig. 5a, 5b), responding to the water circulation in the Guanaqueros and Coquimbo Bays described as bipolar, i.e. two 507 508 counter-rotating gyres which are counterclockwise to the north and clockwise to the 509 south (Valle-Levinson and Moraga, 2006). This is the result of the wind and a 510 coastline shape delimited by two prominent points to the north and south. In the case of Tongoy Bay (the southernmost bay of the system), circulation shows a different 511 pattern due to its northern direction compared to Guanaqueros Bay, which opens to the 512 west. The cyclonic recirculation in Tongoy Bay seems to be part of a gyre larger than 513 the Bay's circulation (Moraga et al., 2011). This could explain differences in sediment 514 515 particle distribution and composition between the bays. At Tongoy Bay, there are less organic carbon accumulation (< 2 %), siliceous microfossils and pollen (Figs. 5, 7 and 516 8). Similarly, in Guanaqueros Bay TOC contents are only slightly higher (> 2 %), 517 especially before cal BC 1000 (~ 4 %). However, sediments there contain enough 518 519 microfossils to establish differences in primary productivity periods and also provide a 520 pollen record evidencing the prevailing environmental conditions.

The stable isotopes measured in the study area were in the range of marine 521 sedimentary particles for southern oceans at low and mid latitudes ( $\delta^{13}$ C; -20 ‰ - -24 522 523 ‰; Williams 1970; Rau et al., 1989; Ogrinc et. al. 2005), and slightly lower than the TOC composition at the water column (-18 ‰, Fig. 3). This suggests that the organic 524 525 particles that settle on the bottom are a more refractory material (C/N: 9-11), remineralized during particle transportation and sedimentation. This results in lighter 526 isotopic compositions, especially at core BTGC8. Besides, the  $\delta^{15}$ N and  $\delta^{13}$ C of settled 527 particles are more negative at surface sediments due to a preferential degradation of 528





molecules rich in <sup>13</sup>C and <sup>15</sup>N, resulting in more negative values and higher C/N ratios 529 at sediments than in suspended particles (Fig. 3, 5a, 5b). However, this is also due to 530 531 the stronger diagenetic reactions observed near the bottom layer (Nakanishi and 532 Minagawa, 2003), indicating that the particles' diagenesis and time of transportation 533 over the shelf should be a relevant factor in this differentiation. Thus, these sediments are composed by winnowed particles transported by water circulating over the shelf, 534 and the isotopic variations should not establish clearly the contribution of terrestrial 535 536 inputs.

537 Magnetic susceptibility (MS) measurements revealed lower values throughout both cores (BGGC5:  $5 - 8 \times 10^{-8}$  SI; BTGC8:  $12 - 20 \times 10^{-8}$  SI), except at dates after ~ cal 538 AD 1800, when the susceptibility increases substantially to values similar to those 539 observed in the Pachingo wetland  $(40 - \sim 200 \text{ x}10^{-8} \text{ SI}; \text{ unpublished data})$  on the 540 southern side of Tongoy Bay. Magnetite has strong response to magnetic fields and its 541 542 concentration is considered proportional to magnetic susceptibility (Dearing, 1999). 543 Besides, mineral post-depositional transformations (alteration of magnetite minerals) 544 and dilution by biogenic components (carbonates, silicates) should also be relevant in the MS intensity in zones with high organic accumulation rates (Hatfield and Stoner, 545 2013). However, this is not expected to be the case for our cores and the MS should be 546 mainly accounting for the source of the particles. High MS measurements in the bay 547 sediments would be associated with relevant terrestrial input. The area is surrounded 548 549 by several creeks that are only active during major flooding events, with greater impacts in Tongoy Bay compared to Guanaqueros Bay. An important increment in the 550 contribution of terrestrial material has occurred in Tongoy Bay in recent times (Ortega 551 et al., in review), which is diluting organic proxy records and increasing the grain size. 552 553 Our records indicate a slight increase in mean grain size at both bays, supported also 554 by a slight decrease in Ca/Fe ratio indicating more Fe input from continental erosion 555 (Fig. 5a, 5b).

Recent information indicates that during the intensification of southern winds the upwelling develops a nutrient-rich and low-oxygen flow within the bay's southern areas (Gallardo et al., 2017), which promotes phytoplankton blooms and low oxygen events. Decreasing concentrations of Ca from the deepest part of both cores to the surface was interpreted as decreasing primary productivity (Keshav and Achyuthan, 2015; Sun et al., 2016), but higher concentrations were measured in core BGGC5 compared with core BTGC8, where more terrestrial influence is being suggested. The





563 slight increase of K/Ca ratio in time, from bottom to the surface, should also be 564 interpreted as a slight increase in continental input, since K is related to siliciclastic 565 material from coastal erosion, fluvial and groundwater inputs. However, the variation 566 of Ca was larger (Fig.6a, 6b), resulting in higher K/Ca ratios at the surface. This 567 indicating that the continental input has not changed much in time but rather the 568 primary productivity has decreased (Fig. 5a, 5b).

569

#### 570 **5.2. Temporal variability of proxies for primary productivity**

571 Several elements participating in phytoplankton growth are useful to interpret 572 variations in primary productivity in time, as they are preserved within the sediments 573 under suboxic-anoxic conditions. This produces enrichment over crustal abundance, 574 which distinguishes them from continental inputs. The presence of free dissolved sulfides produced by sulfate reduction reactions in the diagenesis of organic matter is 575 576 relevant for metal precipitation in pore water (Calvert and Pedersen, 1993; Morse and 577 Luther, 1999). At the same time, organic matter remineralization releases ions to the 578 pore water where they could form organic complexes and insoluble metal sulfides. 579 Conversely, they could be incorporated into pyrite as Cd, Ni and Cu, showing different degrees of trace metal pyritization (Huerta-Diaz and Morse, 1992). Ca, Sr, 580 Cd and Ni profiles suggest a lower proportion of organic deposition in time (Fig. 6a, 581 6b), consistent with the slight reduction of TOC content in time observed in the 582 583 sediments (Figs. 5a, 5b), and concomitantly with other elements related to organic fluxes to the bottom and primary productivity. In the case of Ba, it is actively 584 incorporated into phytoplankton biomass or adsorbed onto Fe oxyhydroxides, 585 increasing the Ba flux towards the sediments, where it is also released during organic 586 587 matter diagenesis. Ba is precipitating in microenvironments where Ba-sulfate reaches 588 supersaturation (Tribovillard et al., 2006 and references therein), but it is dissolved in 589 suboxic-anoxic environments or where sulfate is significantly depleted (Torres et al., 590 1996; Dymond et al., 1992). Therefore, it is better preserved in less anoxic environments with moderate productivity, expected to be the case of our study site 591 (Gross Primary Productivity =0.35 to 2.9 g C m<sup>-1</sup>d<sup>-1</sup>; Daneri et al., 2000). Hence, the 592 slight increase of Ba from cal BC 4000 (Fig. 6a) to the present should rather be the 593 594 response to a less anoxic environment than to an increase in primary productivity. This is consistent with the reduction in TOC and other nutrient-type elements (Ni, Sr, Ca, 595 596 Cd), and results in a low negative correlation with TOC (-0.59; Table 4) due to the Ba





597 remobilization in anoxic conditions after high organic material deposition during cal BC 4000-4500. On the other hand, P distribution showed a trend similar to that of 598 599 TOC and other elements related to organic fluxes to the bottom, although with a lower correlation. The accumulation of P depends on the deposition rate of organic P (dead 600 plankton, bones and fish scales) to the bottom, and is actively remineralized during 601 aerobic or anaerobic bacterial activity. Dissolved P diffuses towards the water column 602 603 where part of it could be adsorbed onto Fe oxides that maintain this element within the 604 sediments. P is buried during a continued sedimentation process and could be released to the pore water under anoxic conditions, when oxides are reduced, creating the 605 environmental conditions for phosphorite and carbonate-fluorapatite precipitation. 606 Normally, this takes place in sites with high sedimentation rates and high organic 607 608 matter fluxes to the bottom (Filippelli, 1997; Cha et al., 2005), which was not the case for our study area (<0.02 cm yr<sup>-1</sup>). In spite of this difference, P and TOC showed a 609 610 decreasing trend towards the present, suggesting reducing flux of organic matter over time, which was also observed for Ni and Cd distributions. Alternatively, it could be 611 612 explained by the increased remineralization of the organic material settled on the 613 bottom (Figs. 6a, 6b).

614 Productivity reconstructions were based on diatom relative abundances and biogenic 615 opal content only in core BGGC5, since core BTGC8 registered valve counts that were too low (<1% in relative diatom abundance). However, at both cores diatom 616 617 assemblages were represented mainly by Chaetoceros resting spores, which are used 618 as upwelling indicators, showing increased concentrations during periods of high productivity and upwelling (Abrantes 1988, Vargas et al., 2004). In addition, 619 620 Chaetoceros resting spores are highly silicified and well preserved in coastal 621 sediments (Blasco et al., 1981). The downcore siliceous productivity based on opal 622 distribution (Fig. 7) distinguished three main periods of increased productivity: (1) recent time cal AD 2014 – cal AD ~1800, (2) cal AD 1 – cal BC 2300 and (3) > cal 623 BC 4300. The mean opal accumulation rate in the second period was  $11.8 \pm 4.8$  g m<sup>2</sup> 624 yr<sup>-1</sup>, when spicules and minerals (quarz, framboid pyrite) where abundant in smear 625 slides. During the third period, accumulation increased highly to  $\sim 30.1 \pm 14.5$  g m<sup>2</sup> yr<sup>-</sup> 626 627 <sup>1</sup>, when the *Chaetoceros* spores were predominant, indicating upwelling intensification 628 and low spicules and minerals were observed in the slides. This is partially consistent with the nutrient-type element distributions. Although the first period was too short, 629 high opal accumulation and high Cd/U ratios could also be observed, which increased 630





toward the present (mean opal value of  $32.3 \pm 22.4$  g m<sup>2</sup> yr<sup>-1</sup>). Similarly, Cu and Fe 631 also increased in recent times (Fig. 6a), contributing to fertilize the environment and 632 633 promoting primary productivity. The second period was not clearly identified in terms of metals, except for Fe which shows a conspicuous increment in this period (Fig. 6a). 634 635 During the third period, all metal proxies showed primary productivity increases before cal BC 4500, as indicated by opal accumulation within the sediments. In 636 anoxic-suboxic environments Cd/U ratios could vary between 0.2 and 2 (Nameroff et 637 638 al., 2002), the high concentrations of both elements reflect anoxic conditions but their 639 different behavior could result in variable Cd/U ratios in suboxic environments. Here, the Cd and U accumulation on sediments resulted in high Cd/U ratios (>2; Fig. 7) 640 641 during periods with high opal accumulation in the cores, especially in the third period, 642 and even in core BTGC8; and lower ratios (< 1; Fig. 7) when opal was low, indicating 643 higher variations in the primary productivity in time with moderated changes in 644 oxygen conditions at the bottoms. Opal showed good correlations with Ni and Cd 645 (~0.70; Table 4; Fig. 6a), all suggesting the relevance of bottom organic fluxes for 646 element accumulation within the sediments, and establishing a clear period of higher primary productivity around cal BC 4500, when lowest oxygen conditions prevailed 647 648 (Fig. 7).

649

#### **5.3.** Temporal variability of proxies for bottom water oxygenation

651 U, Re and Mo distributions in core BGGC5 indicate that anoxic or suboxic conditions were developed from cal BC 6000 to ~ cal AD 1-1000. After this period and towards 652 the present, however, a remarkable reduction in their concentration suggests a more 653 654 oxygenated bottom environment, concurrent with lower organic fluxes to the 655 sediments. The Re profile shows the influence of suboxic waters not necessarily 656 associated with increased organic matter fluxes to the bottom. Since this element is not 657 scavenged by organic particles, its variability is directly related to oxygen changes 658 (Calvert and Pedersen, 2007, and references therein). Additionally, it is strongly enriched above crustal abundance in suboxic conditions (Colodner et al., 1993; 659 660 Crusius et al 1996), being >10 times in core BGGC5 (Fig. 9) before cal AD 1. In the 661 same way, U exhibits a similar pattern, and although organic deposition has an impact 662 on its distribution (Zheng et al., 2002), it also relates to changes in bottom oxygen 663 conditions. This is because its shift from a soluble conservative behavior to non-664 conservative and insoluble solely depends on the redox potential change that occurs





665 near the Fe(III) reduction zone (Klinkhammer and Palmer, 1991). Molybdenum, which showed high increases at cal BC 4500, also indicates the presence of sulfidic 666 667 conditions, as shown by the Re distribution highly enriched in anoxic environments (Colodner et al., 1993), and by the reduction of Re(VII) to Re(IV) forming  $ReO_2$  or 668 669 ReS (Calvert and Pedersen, 2007). Rhenium, U and Mo enrichment is relevant to decipher the redox condition within the sediments, even in places with high lithogenic 670 input that could obscure the authigenic enrichment of other elements under similar 671 672 conditions (Crusius et al., 1996). In both places, the concentrations of these elements 673 showed values above the crustal abundance, especially in core BGGC5 (Fig. 9), with 674 Re and Mo becoming more enriched than U. This suggests that the presence of anoxic 675 conditions were stronger around cal BC 4500-5000. The most relevant enrichment 676 was observed for Cd (> 30, Fig 9), which could similarly indicate the sulfidic condition within the sediments that allows Cd precipitation. It is also supported by Mo 677 678 enrichment, since its accumulation within the sediments is highly controlled by sulfide 679 concentrations (Chaillou et al., 2002; Nameroff et al., 2002; Sundby et al., 2004).

680 Something similar occurs in Tongoy Bay (core BTGC8), but trace metal concentrations are lower for all elements and also for TOC, suggesting that it has 681 limited influence on metal accumulation within the sediments. Thus, these elements 682 suggest anoxic conditions within the sediments in both places around cal BC 683 4500–5000 (Fig. 6a, 6b). After this period, a second maximum but less intense anoxia 684 685 is observed at the beginning of the recent era (cal AD 1-1000), continuing with a conspicuous oxygenation until present times. This interpretation based on the 686 distribution of U, Re and Mo complements the observations of nutrient-type elements 687 pointing both to oxygenation changes in time and to changes in organic fluxes to the 688 689 sediments. A less prominent accumulation of nutrient-type elements (Ni, Cd, Ba, Ca 690 and P) would indicate lower organic material deposition to the sediments but 691 promoting anoxic conditions within the sediments and lower sulfide content with time, 692 which are nevertheless high enough to sustain Mo accumulation until cal AD 1. After that, a notorious decrease in Re, U and Mo accumulation was observed, suggesting 693 694 that the oxygenation of the bottom becomes relevant. This could also explain the 695 conspicuous increase of Cu/Al and Fe/Al in recent times due to the presence of oxides 696 (Fig. 6a, 6b). Apparently, a low level of dissolved Cu is maintained by the complexation with organic compounds produced by phytoplankton and Cu adsorption 697 698 on Fe oxides (Peacock and Sherman, 2004; Vance et al., 2008; Little et al., 2014), with





699 both processes increasing Cu in the particulate phase over surface sediments. At our study sites, Fe and Cu concentrations were higher in surface sediments, probably 700 701 related to an increase in Fe and Cu availability in the environment (Fig 6a, 6b). This 702 could be in turn associated with mining activities carried out in the area since the 703 beginning of cal AD 1900's. At present, the suboxic conditions within the bays result 704 from the influence of adjacent water masses with low oxygen contents, related to the 705 oxygen minimum zone (OMZ) (Fig. 2) centered at ~250 m. Upwelling promotes the intrusion of these waters towards the bays, with strong seasonality. Transition times 706 707 develop in short periods by changes in wind directions and intensities along the coast. 708 Additionally, oceanic variability along the western coast of South America is influenced by 709 equatorial Kelvin waves on a variety of timescales, from intraseasonal (Shaffer et al., 1997) 710 and seasonal (Pizarro et al., 2002; Ramos et al., 2006) to interannual (Pizarro et al., 2002; 711 Ramos et al., 2008). Coastal-trapped Kelvin waves originating from the equator can propagate 712 along the coast, modify the stability of the regional current system and the pycnocline, and 713 trigger extratropical Rossby waves (Pizarro et al., 2002; Ramos et al., 2006; 2008). This 714 oceanographic feature will generate changes in oxygen content within the bays with major impacts on redox sensitive elements in surface sediments; thus, the increased 715 716 frequency and intensity of this variability would result in a mean effect which is 717 observed as a gradual change in metal contents in time.

718

#### 719 **5.6.** Climatic interpretations

720 Past environmental changes are analogue with the present seasonal meridional 721 displacement of the ITCZ (Intertropical Convergence Zone) and SPSH (South Pacific Subtropical High) expansion/contraction, establishing cold-dry/warm-humid climate 722 723 conditions ('winter like/summer like') (Hebbeln et al., 2002; Lamy et al., 2001, 2002, 724 2010). In this regard, the Holocene climate of the semi-arid zone of Chile has alternated 725 between wet and dry phases, associated with the intensity and latitudinal position of the 726 Southern Westerly Wind belt (SWW). Studies based on pollen records in southern 727 coastal areas of Coquimbo (30°S) indicate that wet conditions were predominant before cal BC 6750, which brought the expansion of swamp forests areas along the coast. This 728 wet period was followed by a long-lasting arid phase that culminated in peak dry 729 730 conditions at cal BC 5550-3750 (Maldonado and Villagrán, 2006; Maldonado and 731 Rozas, 2008). This timing matches the relative dry conditions detected in the first 732 portion of our pollen reconstruction, indicated by relative low values of the Pollen





733 Moisture Index (Fig. 10). Our results further suggest that these dry conditions were followed by trends towards increasing precipitation from the mid-Holocene onwards 734 (from cal BC ~4500 to ~ cal AD 1800), as indicated by the increasing values in the 735 736 Pollen Moisture Index and continental runoff (Fig. 10). Al and Pb are considered 737 indicators of continental particle inputs to marine waters transported mainly by river runoff and dust (Saito et al 1992; Calvert and Pedersen, 2007; Govin et al., 2012; Xu et 738 al. 2015; Ohnemus and Lam, 2015). In our cores, these elements showed a trend similar 739 740 to pollen, i.e., a gradual rise in time, suggesting increased humid conditions during 741 recent periods. Therefore, this pattern of humidity and continental influence variability 742 in Guanaqueros Bay sediments suggests an increase in winter rainfall or an increased 743 frequency of episodic rainfall events since ~250 Cal BC (Ortega et al., submitted).

744 South of 35°S, an overall strengthening of SWW and a poleward shift driven by orbital forcings seem to be operating from the mid-Holocene (7 kyr BP) to pre-industrial 745 746 modern times (250 yr BP) (Varma et al., 2012), promoting more humid conditions in 747 this period. Our records indicates increases in grain size, K/Ca ratios, and Fe and Al 748 contents over time, all of which point to higher continental inputs, most probably by increasing rainfall events, which should be an important source of Fe in northern 749 Chilean margin, related to primary productivity increases in sedimentary records 750 (Dezileau et al., 2004). This study shows slight increases in Fe concentrations at cal BC 751 752  $\sim$ 1500–2000 and at cal BC  $\sim$ 4000; both periods showing clear increases in primary 753 productivity. However, maximum productivity observed at cal BC ~4000 -interpreted 754 from the accumulation of nutrient type elements, and the distribution of opal and diatoms- was consistent with overall dry environmental conditions (Fig. 6, 8), also 755 connected to an anoxic sulfidic environment. This could be related to an increased 756 757 intensity of the anticyclone and a consequently higher upwelling during this period (Frugone-Álvarez et al., 2017). 758

759 Peak drying with higher productivity and lower oxygen conditions was followed by a 760 gradually decreasing primary productivity and organic fluxes. The maximum productivity occurred during the driest period studied (mid-Holocene), suggesting 761 762 stronger upwelling that was weakening in time concomitantly with the increase of 763 humid conditions. In this case, weakened upwelling could be interpreted as an intrusion 764 of less nutrient-enriched upwelled waters over the shelf, influenced by remote equatorial 765 waves as observed today. Normally, the wet/dry phases associated with ENSO-like 766 conditions also influence bottom oxygenation which has been observed from the central





767 Peruvian coast to the southern Chilean regions (12 °S - 36 °S) (Escribano et al., 2004; Gutiérrez et al., 2008); thus, OMZ is expected to be less/more intense during warm/cold 768 769 phases. Besides, the Walker circulation strength and the expansion/contraction of the 770 SPSH also control marine productivity (Salvatteci et al., 2014), a key variable closely 771 related to the supply of nutrient rich and poor oxygen waters belonging to the OMZ. 772 Thus, warm periods tend to be associated with low productivity and weak OMZ 773 (Salvatteci et al., 2014). In this sense, an increase in the frequency of ENSO-like warm periods influences bottom oxygenation, resulting in a less reduced environment 774 775 concomitantly with less organic fluxes from primary productivity and less oxygen 776 consumption during organic matter diagenesis.

777

#### 778 6. Conclusions

The circulation seems to affect both places differently, leaving more variable grain compositions and higher TOC contents in the Guanaqueros Bay (core BGGC5) than in the Tongoy Bay (core BTGC8). This difference should be interpreted as an increase in the time of particle transportation resulting in grain size selection (more leptokurtic at core BTGC8), especially after cal AD 1. Furthermore, in both bays, constantly decreasing TOC contents were observed after cal BC ~2000 to the present, probably due to higher oxygenation of the bay bottom in time.

In the investigated sediments, differences in redox conditions could be reconstructed 786 787 showing a clear decreasing trend in oxygen bottoms before the beginning of recent 788 time (cal AD 1), followed by a rapid change to a more oxygenated environment in the last 2000 years. The environmental conditions at bottom waters were relevant in the 789 metal enrichment factor above crustal abundance within the sediments (highest EFs), 790 791 since low organic carbon accumulation and low sedimentation rates have been 792 estimated, indicating that the accumulation of these elements (U, Mo and Re) depends 793 mainly on oxygen content instead of on organic carbon burial rates. Apparently, a 794 maximum suboxia-anoxia occurred at cal BC ~4500 year BC, when peak U and Re 795 where recorded, probably due to the presence of a sulfidic environment.

796 The nutrient-type elements follow a similar trend, reduced at present and showing

<sup>797</sup> higher accumulation rates around cal BC 4500 (Ca, Ni, P and Cd). Their distribution is

ronsistent with the diatom and opal distributions, showing their dependence on primary

799 productivity and organic carbon burial rates. If the kinetics reaction is working at low





- 800 rates for these elements, they should be highly influenced during oxygenation periods,
- something that seems to have been operating with higher frequencies.
- 802 The record of continental proxies establishes a continuous increment in wet
- 803 conditions, consistent with previous reconstructions in central Chile. The most
- distinctive changes were observed after cal BC 4500, when an overall expansion of the
- 805 coastal vegetation occurred as a result of a progressive increase in precipitation and
- river runoffs, expanding the grain size of the sediments and the increase
- 807 concentrations of elements that has relevant continental source (Al, Fe, K and Pb).
- 808 Increased regional precipitations amounts have been commonly interpreted by a
- 809 northward movement of the Southern Westerly Winds belts, but the increased
- 810 frequency of El Niño events have also introduced a high variability of humidity in the
- 811 late Holocene. Thus, the apparent increase of oxygen conditions at bottoms would be
- the result of this oceanographic feature, which introduced a more oxygenated water
- 813 mass to the shelf and bays, temporarily changing the redox conditions in surface
- sediments and affecting the sensitive elements to redox potential change in the
- 815 environment. Additionally, this also affected the accumulation of organic matter due
- to an intensification of its remineralization, showing a decreasing trend in nutrient type
- 817 element accumulation and organic carbon burial rates towards the present.
- 818

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# Tables

Table 1. Concentration of elements in Pachingo wetland sediments, considered as lithogenic background for the study area. The values correspond to mean concentrations in surface sediments (0-3 cm).

Element	Metal/Al x 10 <sup>3</sup>	S
Ca	686.5	139.3
Fe	591.3	84.5
Р	8.6	0.7
Sr	5.7	0.6
Ba	5.6	0.1
Cu	0.258	0.019
Ni	0.174	0.005
U	0.020	0.003
Mo	0.020	0.003
Cd	0.0021	0.0003
Re	0.00004	0.00001





Table 2. Radiocarbon dates for BGGC5 and BTGC8 sediment cores collected from mixed planktonic foraminifera and monospecific benthic foraminifera (*Bolivina plicata*), respectively. The <sup>14</sup>C-AMS was performed at NOSAM-WHOI. The lab code and conventional ages collected from each core section is indicated. For error calculations see http://www.whoi.edu/nosams/radiocarbon-data-calculations.

Core		mass	Lab. Code	modern	fraction	Conventiona	al Age
identification	material	(mg)	NOSAM	pMC	error	BP	error
BGGC5				•			
10-11	Planktonic foraminifera (mix)	1.8	OS-122160	0.8895	0.0027	940	25
18-19	Planktonic foraminifera (mix)	1.1	OS-122141	0.7217	0.0024	2,620	25
31-32	Planktonic foraminifera (mix)	2.7	OS-122161	0.6590	0.0021	3,350	25
45-46	Planktonic foraminifera (mix)	2	OS-122162	0.6102	0.0017	3,970	25
55-56	Planktonic foraminifera (mix)	1.6	OS-122138	0.5864	0.0025	4,290	35
66-67	Planktonic foraminifera (mix)	2.8	OS-122304	0.5597	0.0018	4,660	25
76-77	Planktonic foraminifera (mix)	2.6	OS-122163	0.4520	0.0016	6,380	30
96-97	Planktonic foraminifera (mix)	1.1	OS-122139	0.4333	0.0033	6,720	60
115-116	Planktonic foraminifera (mix)	4.7	OS-122164	0.3843	0.0016	7,680	35
125-126	Planktonic foraminifera (mix)	1.4	OS-122140	0.3964	0.0025	7,430	50
BTGC8							
5-6	Benthic foraminifera (Bolivina plicata)	4.2	OS-130657	0.8953	0.0017	890	15
20-21	Benthic foraminifera (Bolivina plicata)	7.7	OS-123670	0.7337	0.0021	2,490	25
30-31	Benthic foraminifera (Bolivina plicata)	13	OS-123671	0.6771	0.0016	3,130	20
40-41	Benthic foraminifera (Bolivina plicata)	11	OS-123672	0.6507	0.0019	3,450	25
50-51	Benthic foraminifera (Bolivina plicata)	8.7	OS-123673	0.5877	0.0014	4,270	20
60-61	Benthic foraminifera (Bolivina plicata)	13	OS-123674	0.5560	0.0018	4,720	25
71-72	Benthic foraminifera (Bolivina plicata)	10	OS-123675	0.4930	0.0013	5,680	20
80-81	Benthic foraminifera (Bolivina plicata)	7.3	OS-123676	0.4542	0.0012	6,340	20
90-91	Benthic foraminifera (Bolivina plicata)	6.8	OS-123677	0.4259	0.0015	6,860	30
96-97	Benthic foraminifera (Bolivina plicata)	6.8	OS-123678	0.3903	0.0013	7,560	25





Table 3. Reservoir age (DR) estimation considering the  $^{210}$ Pb age determined with the CRS model (McCaffrey and Thomson, 1980) at a selected depth sections of the core, compared with  $^{14}$ C ages (yr BP) from marine13.14 curve (Reimer et al., 2013), according to Sabatier et al. (2010).

		Age from	Age	140 : 12	<sup>14</sup> C DD		
		CRS	years	C marine 13	C age BP		
Core	cm	model	$BP^{a}$	curve	from foram.	DR	S
BGGC5	10.5	1828	122	499	940	441	15
BTCG8	5.5	1908	42	448	890	442	17

a. Before present=1950





BGGC	5															
	Al	Р	К	Ca	Mn	Fe	Ni	Cu	Mo	Cd	Re	Sr	U	Ba	Opal	TOC
Al	1.00	-0.62	0.49	-0.48	0.64	0.60	-0.75	0.56	-0.10	-0.73	-0.08	-0.33	0.08	0.49	-0.52	-0.44
Р		1.00	-0.31	0.37	-0.45	-0.56	0.56	-0.57	0.01	0.61	-0.11	0.39	-0.12	-0.20	0.49	0.24
K			1.00	-0.24	0.90	0.83	-0.29	0.47	0.28	-0.42	0.33	-0.12	0.50	0.26	-0.25	-0.19
Ca				1.00	-0.47	-0.50	0.44	-0.64	0.23	0.59	0.39	0.92	0.30	-0.60	0.18	0.32
Mn					1.00	0.94	-0.51	0.68	-0.01	-0.68	0.07	-0.32	0.24	0.43	-0.39	-0.31
Fe						1.00	-0.49	0.81	0.03	-0.70	0.11	-0.40	0.23	0.36	-0.37	-0.21
Ni							1.00	-0.51	0.49	0.91	0.35	0.25	0.26	-0.70	0.72	0.64
Cu								1.00	-0.12	-0.71	-0.06	-0.61	0.00	0.31	-0.39	-0.07
Мо									1.00	0.50	0.88	0.10	0.91	-0.48	0.33	0.36
Cd										1.00	0.36	0.42	0.27	-0.67	0.70	0.54
Re											1.00	0.27	0.92	-0.50	0.16	0.38
Sr												1.00	0.24	-0.36	0.05	0.17
U													1.00	-0.39	0.10	0.29
Ba														1.00	-0.30	-0.59
Opal															1.00	0.35
TOC																1.00
BTGC	8															
BTGC	8 Al	Р	K	Ca	Mn	Fe	Ni	Cu	Мо	Cd	Re	Sr	U	Ba	Opal	тос
BTGC Al	8 Al 1.00	<b>P</b> -0.19	<b>K</b> -0.17	<b>Ca</b> -0.37	<b>Mn</b> -0.02	<b>Fe</b> -0.03	Ni -0.39	<b>Cu</b> -0.04	<b>Mo</b> -0.39	<b>Cd</b> 0.02	<b>Re</b> -0.13	<b>Sr</b> -0.58	U -0.19	<b>Ba</b> 0.07	<b>Opal</b> -0.41	<b>TOC</b> -0.29
BTGC Al P	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23	<b>Ca</b> -0.37 0.00	<b>Mn</b> -0.02 0.43	<b>Fe</b> -0.03 0.28	Ni -0.39 0.58	<b>Cu</b> -0.04 0.23	<b>Mo</b> -0.39 0.37	Cd 0.02 0.13	<b>Re</b> -0.13 -0.04	<b>Sr</b> -0.58 0.30	U -0.19 0.14	<b>Ba</b> 0.07 -0.14	<b>Opal</b> -0.41 0.56	<b>TOC</b> -0.29 0.13
BTGC Al P K	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	<b>Ca</b> -0.37 0.00 -0.02	Mn -0.02 0.43 0.54	Fe -0.03 0.28 0.41	Ni -0.39 0.58 0.43	Cu -0.04 0.23 0.22	Mo -0.39 0.37 -0.11	Cd 0.02 0.13 0.05	<b>Re</b> -0.13 -0.04 -0.04	Sr -0.58 0.30 0.19	U -0.19 0.14 -0.28	<b>Ba</b> 0.07 -0.14 0.28	<b>Opal</b> -0.41 0.56 0.26	<b>TOC</b> -0.29 0.13 0.20
BTGC Al P K Ca	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33	Fe -0.03 0.28 0.41 -0.27	Ni -0.39 0.58 0.43 0.00	Cu -0.04 0.23 0.22 -0.23	<b>Mo</b> -0.39 0.37 -0.11 0.39	Cd 0.02 0.13 0.05 0.01	<b>Re</b> -0.13 -0.04 -0.04 0.33	<b>Sr</b> -0.58 0.30 0.19 0.50	U -0.19 0.14 -0.28 0.47	<b>Ba</b> 0.07 -0.14 0.28 -0.34	<b>Opal</b> -0.41 0.56 0.26 0.20	<b>TOC</b> -0.29 0.13 0.20 0.34
Al P K Ca Mn	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21	Ni -0.39 0.58 0.43 0.00 0.64	Cu -0.04 0.23 0.22 -0.23 0.01	<b>Mo</b> -0.39 0.37 -0.11 0.39 0.05	Cd 0.02 0.13 0.05 0.01 0.33	<b>Re</b> -0.13 -0.04 -0.04 0.33 0.15	Sr -0.58 0.30 0.19 0.50 0.32	U -0.19 0.14 -0.28 0.47 -0.02	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24	Opal -0.41 0.56 0.26 0.20 0.32	<b>TOC</b> -0.29 0.13 0.20 0.34 0.00
BTGC Al P K Ca Mn Fe	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13	Cu -0.04 0.23 0.22 -0.23 0.01 0.71	Mo -0.39 0.37 -0.11 0.39 0.05 -0.40	Cd 0.02 0.13 0.05 0.01 0.33 -0.48	<b>Re</b> -0.13 -0.04 -0.04 0.33 0.15 -0.67	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37	U -0.19 0.14 -0.28 0.47 -0.02 -0.62	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24 0.13	Opal -0.41 0.56 0.26 0.20 0.32 0.14	<b>TOC</b> -0.29 0.13 0.20 0.34 0.00 0.10
BTGC Al P K Ca Mn Fe Ni	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24	Mo -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20	<b>Re</b> -0.13 -0.04 -0.04 0.33 0.15 -0.67 0.25	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37 0.64	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24 0.13 -0.16	Opal -0.41 0.56 0.26 0.20 0.32 0.14 0.80	TOC           -0.29           0.13           0.20           0.34           0.00           0.10           0.45
BTGC Al P K Ca Mn Fe Ni Cu	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	Mo -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20 -0.68	<b>Re</b> -0.13 -0.04 -0.04 0.33 0.15 -0.67 0.25 -0.56	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37 0.64 -0.22	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24 0.13 -0.16 -0.10	<b>Opal</b> -0.41 0.56 0.26 0.20 0.32 0.14 <b>0.80</b> 0.21	TOC           -0.29           0.13           0.20           0.34           0.00           0.10           0.45           0.37
BTGC Al P K Ca Mn Fe Ni Cu Mo	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	Mo           -0.39           0.37           -0.11           0.39           0.05           -0.40           0.56           -0.25           1.00	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20 -0.68 0.45	Re           -0.13           -0.04           0.33           0.15           -0.67           0.25           -0.56           0.59	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37 0.64 -0.22 0.66	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24 0.13 -0.16 -0.10 -0.41	<b>Opal</b> -0.41 0.56 0.20 0.32 0.14 <b>0.80</b> 0.21 0.58	TOC           -0.29           0.13           0.20           0.34           0.00           0.10           0.45           0.37           0.30
BTGC Al P K Ca Mn Fe Ni Cu Mo Cd	8 <u>Al</u> 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	<b>Mo</b> -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25 1.00	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20 -0.68 0.45 1.00	<b>Re</b> -0.13 -0.04 -0.04 0.33 0.15 -0.67 0.25 -0.56 0.59 0.56	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37 0.64 -0.22 0.66 0.39	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69 0.52	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24 0.13 -0.16 -0.10 -0.41 0.11	Opal           -0.41           0.56           0.20           0.32           0.14           0.80           0.21           0.58           0.10	TOC           -0.29           0.13           0.20           0.34           0.00           0.10           0.45           0.37           0.30           -0.12
Al P K Ca Mn Fe Ni Cu Mo Cd Re	8 <u>Al</u> 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	Mo -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25 1.00	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20 -0.68 0.45 1.00	Re           -0.13           -0.04           -0.33           0.15           -0.67           0.25           -0.56           0.59           0.56           1.00	Sr           -0.58           0.30           0.19           0.50           0.32           -0.37           0.64           -0.22           0.66           0.39           0.53	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69 0.52 0.83	<b>Ba</b> 0.07 -0.14 0.28 -0.34 0.24 0.13 -0.16 -0.10 -0.41 0.11 -0.16	Opal           -0.41           0.56           0.20           0.32           0.14           0.80           0.21           0.58           0.10           0.13	TOC           -0.29           0.13           0.20           0.34           0.00           0.10           0.45           0.37           0.30           -0.12           0.17
Al P K Ca Mn Fe Ni Cu Mo Cd Re Sr	8 <u>Al</u> 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	<b>Mo</b> -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25 1.00	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20 -0.68 0.45 1.00	Re           -0.13           -0.04           -0.33           0.15           -0.67           0.25           -0.56           0.59           0.56           1.00	Sr           -0.58           0.30           0.19           0.50           0.32           -0.37           0.64           -0.22           0.66           0.39           0.53           1.00	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69 0.52 0.83 0.58	Ba           0.07           -0.14           0.28           -0.34           0.23           -0.16           -0.10           -0.41           0.11           -0.16	Opal -0.41 0.56 0.20 0.32 0.14 0.80 0.21 0.58 0.10 0.13 0.52	TOC           -0.29           0.13           0.20           0.34           0.00           0.10           0.45           0.37           0.30           -0.12           0.17           0.23
BTGC Al P K Ca Mn Fe Ni Cu Mo Cd Re Sr U	8 Al 1.00	<b>P</b> -0.19 1.00	<b>K</b> -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	<b>Mo</b> -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25 1.00	Cd 0.02 0.13 0.05 0.01 0.33 -0.48 0.20 -0.68 0.45 1.00	Re           -0.13           -0.04           -0.33           0.15           -0.67           0.25           -0.56           0.59           0.56           1.00	Sr           -0.58           0.30           0.19           0.50           0.32           -0.37           0.64           -0.22           0.66           0.39           0.53           1.00	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69 0.52 0.83 0.58 1.00	Ba           0.07           -0.14           0.28           -0.34           0.24           0.13           -0.16           -0.10           -0.41           0.11           -0.16	Opal           -0.41           0.56           0.26           0.32           0.14           0.80           0.21           0.58           0.10           0.13           0.52	TOC           -0.29           0.13           0.20           0.34           0.00           0.45           0.37           0.30           -0.12           0.17           0.23
BTGC Al P K Ca Mn Fe Ni Cu Mo Cd Re Sr U Ba	8 Al 1.00	<b>P</b> -0.19 1.00	К -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	Mo -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25 1.00	Cd           0.02           0.13           0.05           0.01           0.33           -0.48           0.20           -0.68           0.45           1.00	Re           -0.13           -0.04           -0.33           0.15           -0.67           0.25           -0.56           0.59           0.56           1.00	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37 0.64 -0.22 0.66 0.39 0.53 1.00	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69 0.52 0.83 0.58 1.00	Ba           0.07           -0.14           0.28           -0.34           0.24           0.13           -0.16           -0.10           -0.41           0.11           -0.16           -0.13           -0.19           1.00	Opal           -0.41           0.56           0.20           0.32           0.14           0.80           0.21           0.58           0.10           0.13           0.52           0.21	TOC           -0.29           0.13           0.20           0.34           0.00           0.45           0.37           0.30           -0.12           0.17           0.23           0.00
BTGC Al P K Ca Mn Fe Ni Cu Mo Cd Re Sr U Ba Opal	8 Al 1.00	<b>P</b> -0.19 1.00	К -0.17 0.23 1.00	Ca -0.37 0.00 -0.02 1.00	Mn -0.02 0.43 0.54 -0.33 1.00	Fe -0.03 0.28 0.41 -0.27 0.21 1.00	Ni -0.39 0.58 0.43 0.00 0.64 0.13 1.00	Cu -0.04 0.23 0.22 -0.23 0.01 0.71 0.24 1.00	Mo -0.39 0.37 -0.11 0.39 0.05 -0.40 0.56 -0.25 1.00	Cd           0.02           0.13           0.05           0.01           0.33           -0.48           0.20           -0.68           0.45           1.00	Re           -0.13           -0.04           -0.33           0.15           -0.67           0.25           -0.56           0.59           0.56           1.00	<b>Sr</b> -0.58 0.30 0.19 0.50 0.32 -0.37 0.64 -0.22 0.66 0.39 0.53 1.00	U -0.19 0.14 -0.28 0.47 -0.02 -0.62 0.19 -0.61 0.69 0.52 0.83 0.58 1.00	Ba           0.07           -0.14           0.28           -0.34           0.24           0.13           -0.16           -0.10           -0.41           0.11           -0.16           -0.13           -0.19           1.00	Opal           -0.41           0.56           0.20           0.32           0.14           0.80           0.21           0.58           0.10           0.13           0.52           0.21	<b>TOC</b> -0.29 0.13 0.20 0.34 0.00 0.10 0.45 0.37 0.30 -0.12 0.17 0.23 0.00 -0.42 0.39

Table 4. Spearman rank order correlations for geochemical data. Significant values >0.8 are indicated in bold.





# Figures



Figure 1. Study area showing the position of sampling stations. Sediment cores were retrieved from Guanaqueros Bay (BGGC5) and from Tongoy Bay (BTGC8) at water depths of 89 and 85 m, respectively. Information of dissolved oxygen (DO) in the water column and of suspended organic particles collected at ST1, ST14 and ST16 sampling sites was gathered in a previous project (INNOVA 07CN13 IXM-150). Climograph of the region is showing the average precipitation in mm (bars) and temperatures in °C (min, max and average) over 12-month period.







Figure 2. Dissolved Oxygen (DO) time series in the water column measured between October 2010 and January 2011, at stations St1 and St16 off Tongoy Bay, Coquimbo (30°S).







Figure 3. Suspended particulate matter composition (TOC % and  $\delta^{13}$ Corg) measured in the water column between October 2010 and October 2011, at station St14, Tongoy Bay, Coquimbo (30°S).







Figure 4. Age model based on <sup>14</sup>CAMS and <sup>210</sup>Pb measurements. The time scale was obtained according to the best fit of curves of <sup>210</sup>Pb<sub>xs</sub> and <sup>14</sup>C points using CLAM 2.2 software and Marine curve <sup>13</sup>C (Reimer et al., 2013).

a)

-4000 -5000

-6000 Age (BC/AD)

5Y3/2 5Y4/2 0 50 100

Susceptibility

SI 10<sup>-8</sup> — mean grain size sand silt clay





Dry bulk density Mean grain size (2 cm<sup>3</sup>) (um) Granulometry (%) Skewness TOC (%) δ<sup>13</sup>C (‰) Ca/Fe (g cm<sup>3</sup>) (μm) Granulometry (%) Skewness 0.3 0.6 0.9 1.2 20 40 60 80100 0 25 50 75100 -1.6 -1.2 -0.8 -0.4 Color 0123456 -24 -22 -20 -18 0 1 2 3 2000 1000 0 -1000 -2000 -3000 4000 -5000 -ANN -6000 -7000 0 50 100 150 3 4 5 6 7 0 5 10 15 20 4 6 8 10 12 0.2 0.3 0.4 0.5 Age (BC/AD) sand silt Susceptibility SI 10<sup>-8</sup> Kurtosis C/N δ<sup>15</sup>N (‰) K/Ca 2.5Y3/3 clay Mean grain size b) Dry Bulk Density Mean grain size Granulometry δ<sup>13</sup>C (‰) TOC (%) Ca/Fe Skewness (g cm<sup>3</sup>) (µm) (%) Color 0.5 1.0 1.5 20 40 60 80 0 25 50 75100 -3 -2 0 -22-21-20-19-18 0 1 2 3 -1 2 2000 M~~M 1000 0 -1000 -2000 -3000

Figure 5. Sediment characterization of sediment cores retrieved from (a) Guanaqueros Bay (BGGC5) and (b) Tongoy Bay (BTGC8). Distribution in depth core of color, dry bulk density, statistical parameters (skewness, mean grain size, kurtosis), organic components (TOC, stable isotopes) and chemical composition (K/Ca, Ca/Fe).

3

Kurtosis

0

C/N

5 10 15 20 2 4 6 8 1012 0.2

δ<sup>15</sup>N (‰)

0.4 0.6

K/Ca

6 8 10 12

46







b)



Figure 6. Trace element distribution in sediment cores retrieved from (a) Guanaqueros Bay (BGGC5) and (b) Tongoy Bay (BTGC8), off Coquimbo (30°S).







Figure 7. Diatom abundance, opal accumulation and temporal variations in the relative abundance of *Chaetoceros* resting spores in BGGC5 and BTGC8 cores (Guanaqueros and Tongoy Bay, respectively). Cd/U distribution was included as a proxy for redox condition.







Figure 8. Pollen record in BGGC5 core.







Figure 9. Authigenic enrichment factor (EF) of trace elements in BGGC5 core. Lithogenic background as estimated from surface sediments of Pachingo wetland cores (see text).







Figure 10. Pollen Moisture Index defined as the normalized ratio between Euphorbiaceae (wet coastal shrub land) and Chenopodiaceae (arid scrubland). Positive (negative) values for this index indicate the relative expansion (reduction) of coastal vegetation under wetter (drier) conditions. Pb and Al distribution at BGGC5 core, representatives of terrigenous input to the bay.