

1 **The importance of mineral determinations to PROFILE base**
2 **cation weathering release rates: A case study**

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17 **Abstract**

18 Accurate estimates of base cation weathering rates in forest soils are crucial for policy decisions on sustainable
19 biomass harvest levels and for calculations of critical loads of acidity. The PROFILE model is one of the most
20 frequently used methods to quantify weathering rates, where the quantitative mineralogical input has often been
21 calculated by the A2M (“Analysis to Mineralogy”) program based solely on geochemical data. The aim of this
22 study was to investigate how uncertainties in quantitative mineralogy, originating from modeled mineral
23 abundance and assumed stoichiometry, influence PROFILE weathering estimate, by using measured quantitative
24 mineralogy by X-ray powder diffraction (XRPD) as a reference. Weathering rates were determined for two sites,
25 one in Northern (Flakaliden) and one in Southern (Asa) Sweden. At each site, 3–4 soil profiles were analyzed at
26 10 cm depth intervals. Normative quantitative mineralogy was calculated from geochemical data and qualitative
27 mineral data with the A2M program using two sets of qualitative mineralogical data inputs to A2M: 1) A site-
28 specific mineralogy based on information about mineral identification and mineral chemical composition as
29 determined directly by XRPD and electron microprobe analyses (EMPA), and 2) regional mineralogy, representing
30 the assumed minerals present and assumed mineral chemical compositions for large geographical areas in Sweden,
31 as per previous published studies. Arithmetic means of the weathering rates determined from A2M inputs (W_{A2M})
32 were generally in relatively close agreement with those (W_{XRPD}) determined by inputs based on direct XRPD and
33 EMPA measurements. The hypothesis that using site-specific instead of regional mineralogy will improve the
34 confidence in mineral data input to PROFILE was supported for Flakaliden. However, at Asa, site-specific
35 mineralogies reduced the discrepancy for Na between W_{A2M} and W_{XRPD} but produced larger and significant
36 discrepancies for K, Ca and Mg. For Ca and Mg the differences between weathering rates based on different
37 mineralogies could be explained by differences in the content of some specific Ca- and Mg-bearing minerals, in
38 particular amphibole, apatite, pyroxene and illite. Improving the accuracy in the determination of these minerals
39 would reduce weathering uncertainties. High uncertainties in mineralogy, due for example to different A2M
40 assumptions, had surprisingly little effect on the predicted weathering of Na- and K-bearing minerals. This can be
41 explained by the fact that the weathering rate constants for the minerals involved, e.g. K-feldspar and micas, are
42 similar in PROFILE. Improving the description of the dissolution rate kinetics of the plagioclase mineral group as
43 well as major K-bearing minerals (K-feldspars and micas) should be a priority to help improve future weathering
44 estimates with the PROFILE model.

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47 **Definitions and abbreviations**

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49 Mineralogy = the identity (specific mineral or mineral group) and stoichiometry (specific mineral chemical
50 composition) of minerals that are present at a certain geographic unit, a particular site (*site-specific mineralogy*)
51 or a larger geographic province (*regional mineralogy*)

52 Quantitative mineralogy= the quantitative information (wt.%) on the abundance of specific minerals in
53 the soil.

54 **Abbreviations:**

55 M_{XRPD} = quantitative mineralogy based on XRPD (amount) and electron microprobe analysis (composition)

56 $M_{\text{A2M-reg}}$ = quantitative mineralogy calculated with the A2M model and using regional mineralogy input

57 $M_{\text{A2M-site}}$ = quantitative mineralogy calculated with the A2M model and using site-specific mineralogy input

58 W_{XRPD} = weathering rate based on quantitative mineralogy determined by direct XRPD and electron microprobe
59 analysis

60 W_{A2M} = weathering rate based on quantitative mineralogy determined by the A2M model (unspecific mineralogy
61 input)

62 $W_{\text{A2M-reg}}$ = weathering rate based on quantitative mineralogy determined by the A2M model, and assuming regional
63 mineralogy input.

64 $W_{\text{A2M-site}}$ = weathering rate based on quantitative mineralogy determined by the A2M model and assuming site-
65 specific mineralogy input.

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69 **1. Introduction**

70 The dissolution of minerals in soils and rocks during weathering represents, together with deposition, the most
71 important long-term supply of base cations for plant growth as well as acting as a buffer against soil and water
72 acidification. Quantifying weathering rates is therefore of key importance to guide modern forestry demands on
73 biomass removal by helping to identify threshold levels that are sustainable for base cation removal from soils and
74 waters. With the introduction of the harvest of forest biomass for energy production that includes whole tree
75 harvest and stump extraction, about 2–3 times more nutrients are exported from the forest compared to stem-only
76 harvest. As a result, issues of acidification and base cation supply are exacerbated and the sustainability of this
77 practice is questioned (Röser, 2008; de Jong et al. 2017). Regional nutrient balance calculations for Sweden have
78 indicated that net losses of base cations from forest soils can occur in stem-only harvest scenarios, and this trend
79 would be substantially exacerbated and more frequent in whole-tree harvesting scenarios, largely due to low
80 weathering rates (Sverdrup and Rosén, 1998; Akselsson et al., 2007a,b). Furthermore, the same effect occurs both
81 under current and projected future climate conditions (Akselsson et al., 2016).

82 The weathering rates included in these nutrient balance calculations are in most cases based on the PROFILE
83 model. This is a process-oriented model calculating steady-state weathering rates using transition state theory and
84 physical and geochemical properties of the soil such as temperature, soil moisture, soil mineralogy and
85 concentrations of base cation, hydrogen and organic acids. (Sverdrup, 1996). This model has been widely applied
86 in Europe, Canada and the US during the last several decades or more of weathering research (Olsson et al., 1993;
87 Langan et al., 1995; Kolka et al., 1996; Starr et al., 1998; Sverdrup and Rosén, 1998; Whitfield et al., 2006;
88 Akselsson et al., 2007a; Koseva et al., 2010; Stendahl et al., 2013). In some cases nutrient balance calculations
89 have also been based on the depletion method (Olsson et al., 1993).

90 Reliable weathering rate estimates are crucial for the accuracy of future nutrient budget calculations (Futter et al.,
91 2012). Regarding the accuracy of the PROFILE model, the importance of high accuracy in physical input
92 parameters for the modelled weathering rate outputs has been highlighted by Hodson et al. (1996) and Jönsson et
93 al. (1995). Among the various parameters Hodson et al. (1996) noted that the weathering response of the entire
94 soil profile depends critically on its mineralogy and as such any choice of the model user about mineralogical input
95 data may affect the model outcome significantly (Hodson et al., 1997). In most cases the mineralogical input to
96 the PROFILE model is also derived by modelling yet little attention has been given to the influence of modelled
97 versus directly measured mineralogical input data on calculated base cation release rates.

98 The most widely used method for direct quantitative mineralogical analysis of soil samples is X-ray powder
99 diffraction, and the accuracy that can be achieved has been demonstrated in round robin tests most notably the
100 Reynolds Cup (McCarty, 2002; Kleeberg, 2005; Omotoso et al., 2006, Raven and Self, 2017). Casetou-Gustafson
101 et al. (2018) made some independent assessment of the accuracy of their own XRPD data by geochemical cross
102 validation (i.e. the mineral budgeting approach of Andrist-Rangel et al., 2006). Nonetheless, we should stress that
103 like all analytical methods the determined weight fractions of minerals identified in a soil sample by XRPD will
104 have an associated uncertainty. Additionally, minerals present in minor amounts, nominally < 1% by weight, may
105 fall below the lower limit of detection of the XRPD method.

106 Due mainly to the relative ease of measurement and consequent ready availability of total element geochemical
107 data on soils, indirect methods of determining quantitative soil mineralogy, such as so called ‘normative’
108 geochemical calculations have been widely used to generate mineralogical data for use in the PROFILE model.
109 One such method is the normative “Analysis to Mineralogy” (A2M) program (Posch and Kurz, 2007) that has
110 commonly been used in PROFILE applications (Stendahl et al., 2013; Zanchi et al., 2014; Yu et al., 2016; 2018;
111 Kronnäs et al., 2019). Based on a quantitative geochemical analysis of a soil sample, typically expressed in weight
112 percent oxides, as well as on some assessment of the available minerals in the soil sample (minerals present) and
113 their stoichiometry (chemical compositions), A2M calculates all possible mineralogical compositions for the soil
114 sample. Thus the A2M output for a given soil sample input has multiple solutions and can be described as a
115 multidimensional mineralogical solution space. This necessitates a choice when using A2M output in applications
116 such as weathering rate studies, the convention for which has been to use the geometric mean mineralogical
117 compositions e.g. Stendahl et al. 2013. Casetou-Gustafson et al. (2018) compared the output of A2M with directly
118 determined XRPD mineralogies at two sites, applying A2M in two different ways. In the first case the information
119 on available minerals in the model input was obtained from direct XRPD mineral identifications and information
120 on mineral stoichiometry from direct microprobe analysis of the minerals at the specific site (hereafter denoted
121 "site-specific"). In the second case the mineral stoichiometry and mineral identity were both assumed based on an
122 expert assessment of the probable mineralogy at the regional scale as given by Warfvinge and Sverdrup (1995),
123 hereafter denoted "regional". Casetou-Gustafson et al. (2018) concluded that using A2M in combination with
124 regional input data yielded results with large deviations from directly (XRPD) measured quantitative mineralogy,
125 particularly for two of the major minerals, K-feldspar and dioctahedral mica. When site-specific mineralogical
126 input data was used, measured and modeled quantitative mineralogy showed a better correspondence for most
127 minerals. For a specific mineral and a specific site, however, the bias in determination of quantitative mineralogy
128 might be significant depending on the accuracy of input data to A2M, i.e. total geochemistry and/or mineral
129 stoichiometry (Casetou-Gustafson et al., 2018). Potential errors like these in mineralogical input data might be
130 assumed to affect the calculated weathering for different base cations significantly.

131 In the present study, we used the different mineralogical data from Casetou-Gustafson et al. (2018) to model
132 weathering rates of soils with the PROFILE model. Rates calculated based on measured mineral abundances using
133 quantitative XRPD in combination with measured mineral elemental compositions are taken as ‘reference’
134 weathering rates to which other rates are compared. Samples for this study were collected from podzolised till
135 soils from 8 soil profiles at two forest sites in northern and southern Sweden, respectively.

136 The primary objective of this study was to describe and quantify the effect of differences in mineralogy input on
137 PROFILE weathering rates, leaving all other input parameters of the PROFILE model constant to isolate the effects
138 of variation in input of mineral stoichiometry and abundance. A first specific aim was to determine the
139 uncertainties in weathering rates caused by uncertainties in normative quantitative mineralogy. This was
140 approached by comparing PROFILE runs using modeled mineralogies based on the presence of minerals of a
141 specific site or a larger geographic region (i.e. site-specific and regional mineralogy) with PROFILE runs using
142 the directly measured mineralogy. The latter was assumed to represent the ‘true’ mineralogy at each site. The
143 comparison of PROFILE weathering rates, based on XRPD versus A2M mineralogy, was done using 1000 random
144 solutions per sample from the entire multidimensional A2M mineralogical solution space. In the following,

145 weathering rates calculated by PROFILE based on XRPD and A2M mineralogies are denoted W_{XRPD} and W_{A2M} ,
146 respectively.

147 A second specific aim was to investigate how the over- or underestimation of W_{A2M} in relation to W_{XRPD} mirrors
148 the over- or underestimation of mineral contents estimated with A2M.

149 The following hypotheses were made:

150 (1) PROFILE weathering rates obtained with normative quantitative mineralogy calculated based on site-specific
151 mineralogical information about mineral identity and mineral stoichiometry, are more similar to the reference
152 weathering rates than PROFILE runs obtained with normative quantitative mineralogy calculated based on
153 regional information only.

154 (2) Over- and underestimations of weathering rates of different base cations by the PROFILE model can be
155 explained by over- or underestimations of mineral contents of a few specific minerals.

156

157 **2. Materials and methods**

158 **2.1 Study sites**

159 Two experimental forest sites, Asa in southern, and Flakaliden in northern Sweden, were used for the study (Table
160 1). Both sites have Norway spruce (*Picea abies* (L.) Karst) stands of uniform age, but differ in climate. Flakaliden
161 is located in the boreal zone with long cold winters, whereas Asa is located in the hemiboreal zone. The soils have
162 similar texture (Sandy loamy till), soil types (Spodosols) and moisture conditions. According to the geographical
163 distribution of mineralogy types in Sweden the sites belong to different regions (Warfvinge and Sverdrup, 1995).
164 The experiments, which started in 1986, aimed at investigating the effects of optimized water and nutrient supply
165 on tree growth and carbon cycling in Norway spruce forests (Linder 1995, Albaugh et al. 2009). The sites are
166 incorporated in the Swedish Infrastructure for Ecosystem Science (SITES).

167 **2.2 Soil sampling and stoniness determination**

168 Soil sampling was performed in October 2013 and March 2014 in the border zone of four plots each of the sites.
169 Plots selected for sampling were untreated control plots (K1 and K4 at Asa, 10B and 14B at Flakaliden) and
170 fertilized 'F' plots (F3, F4 at Asa, 15A, 11B at Flakaliden). A rotary drill was used in order to extract one intact
171 soil core per plot (17 cm inner diameter) except for plot K4, F3 and F4 at the Asa site. A 1 x 1 m soil pit was
172 excavated at each of the three latter plots due to inaccessible terrain for forest machinery. The maximum mineral
173 soil depth varied between 70–90 cm in Flakaliden and 90–100 cm in Asa.

174

175 The volume of stones and boulders was determined with the penetration method by Viro (1952), and by applying
176 penetration data to the functions by Stendahl et al. (2009). A metal rod was penetrated at 16 points per plot into
177 the soil until the underground was not possible to penetrate any further, or to the depth 30 cm. There was a higher
178 average stoniness at Flakaliden than Asa (39 vol-% compared to 29 vol-% in Asa) that could partially explain the
179 lower maximum sampling depth at Flakaliden.

180 **2.3 Sample preparation**

181 Soils samples for chemical analyses were taken at 10 cm depth intervals in the mineral soil. Prior to analysis all
182 soil samples were dried at 30–40 °C and sieved at 2 mm mesh. Soil chemical analyses were performed on the fine
183 earth fraction (< 2mm).

184 **2.4 Analysis of geochemistry, total carbon and soil texture**

185 Total carbon was determined using a LECO elemental analyzer according to ISO 10694. Analysis of total
186 geochemical composition, conducted by ALS Scandinavia AB, was made by inductively coupled plasma
187 spectrometry (ICP-MS). Prior to analyses, the samples were ignited at 1000° C to oxidize organic matter and
188 grinded with an agate mortar. Particle size distribution was analyzed by wet sieving and sedimentation (Pipette
189 method) in accordance with ISO 11277. More details about the analytical procedure was given by Casetou-
190 Gustafson et al. (2018).

191 **2.5 Determination of quantitative mineralogy**

192 A detailed description of methods used to quantify mineralogy of the samples was given by Casetou et al. (2018)
193 and these are described in brief below.

194 **2.5.1 Measured mineralogy**

195 Quantitative soil mineralogy was determined with the X-ray powder diffraction technique, XRPD (M_{XRPD}) (Hillier
196 1999, 2003) (Table S1a-b). Preparation of samples for determination of XRPD patterns was made from spray
197 drying slurries of micronized soil samples (<2 mm) in ethanol. Quantitative mineralogical analysis of the
198 diffraction data was performed using a full pattern fitting approach (Omotoso et al., 2006). In the fitting process,
199 the measured diffraction pattern is modelled as a weighted sum of previously recorded and carefully verified
200 standard reference patterns of the prior identified mineral components. The chemical composition of the various
201 minerals present in the soils was determined by electron microprobe analysis (EMPA) (Table S6).

202 **2.5.2 Calculated mineralogy**

203 The A2M program (Posch and Kurz, 2007) was used to calculate quantitative mineralogical composition (M_{A2M})
204 from geochemical data. Based on a set of pre-determined data on mineral identity and stoichiometry, the model
205 outcome is a range of equally possible mineralogical compositions. The multidimensional structure of this
206 normative mineralogy model is a consequence of the number of minerals being larger than the number of analysed
207 elements, where a specific element can often be contained in several different minerals. A system of linear
208 equations is used to construct an M-N dimensional solution space (Dimension M= Number of minerals, Dimension
209 N=number of oxides). In this study we used one thousand solutions to cover the range of possible quantitative
210 mineralogies that may occur at a specific site.

211
212 A2M was used to calculate 1000 quantitative mineralogies each for two different sets of mineral identity and
213 element stoichiometry, $M_{A2M-reg}$ (regional) and $M_{A2M-site}$ (site-specific). Regional mineralogy refers to the mineral
214 identity and stoichiometry for the four major mineralogical provinces in Sweden as suggested by Warfvinge and
215 Sverdrup (1995), of which Asa and Flakaliden belong to different regions (Table S5). Site-specific mineralogy

216 refers to the measured mineral identity and stoichiometry determined by the XRPD and electron microprobe
217 analyses of the two sites (Table S6) (Casetou-Gustafson et al., 2018).

218 **2.6. Estimation of weathering rates with PROFILE**

219 **2.6.1 PROFILE model description**

220 The biogeochemical PROFILE model can be used to study the steady state weathering (i.e. stoichiometric mineral
221 dissolution) of soil profiles, as weathering is known to be primarily determined by the physical soil properties at
222 the interface of wetted mineral surfaces and the soil solution. PROFILE is a multilayer model, thus, for each soil
223 layer, parameters are specified based on field measurements and estimation methods (Warfvinge and Sverdrup,
224 1995). Furthermore, isotropic, well mixed soil solution conditions are assumed to prevail in each layer as well as
225 surface limited dissolution in line with early views by Aagard and Helgeson (1982) and Cou and Wollast (1985)
226 (Sverdrup, 1996). Based on these major assumptions, PROFILE calculates chemical weathering rates from a series
227 of kinetic reactions that are described by laboratory determined dissolution rate coefficients and soil solution
228 equilibria (i.e. transition state theory) (Sverdrup and Warfvinge, 1993). The PROFILE version (September 2018)
229 that was used in this study is coded to produce information on the weathering contribution of specific minerals,
230 which allowed us to test our second hypothesis. This version is based on the weathering rates of 15 minerals. Of
231 these, apatite, pyroxene, illite, dolomite and calcite were not found at the two study sites according to XRPD data
232 (Table S1).

233 **2.6.2 PROFILE parameter estimation**

234 The only parameter that was changed between different PROFILE runs was the quantitative mineralogy for each
235 soil layer, as described above. Hence, PROFILE estimated weathering rates (W) based on measured mineralogy
236 (W_{XRPD}), and the two versions of A2M calculated mineralogy, regional ($W_{\text{A2M-reg}}$), and site-specific ($W_{\text{A2M-site}}$). In
237 the regional mineralogy, plagioclase is assumed to occur as pure anorthite and pure albite for simplification as has
238 been used in previous studies (Stendahl et al., 2013; Zanchi et al., 2014). This simplification was done in order to
239 avoid having a number of minerals containing different amount of Ca and Na, as a result of plagioclase forming a
240 continuous solid solution series, since it would not affect the weathering rates.

241
242 The physical soil layer specific parameters, that were kept constant between different profile runs, were exposed
243 mineral surface area, stoniness, soil bulk density and soil moisture (Table 2). Exposed mineral surface area was
244 estimated from soil bulk density and texture analyses in combination with an algorithm specified in Warfvinge
245 and Sverdrup (1995) and critically discussed in Hodson et al. (1998). The volumetric field soil water content in
246 Flakaliden and Asa was estimated to be $0.25 \text{ m}^3 \text{ m}^{-3}$ according to the moisture classification scheme described in
247 Warfvinge and Sverdrup (1995). It was used to describe the volumetric water content for each soil pit.

248
249 Another group of parameters kept constant was chemical soil layer specific parameters. Aluminum solubility
250 coefficient needed for solution equilibrium reactions, defined as $\log\{\text{Al}^{3+}\}+3\text{pH}$, was estimated applying a
251 function developed from previously published data (Simonsson and Berggren, 1998) on our own total carbon and
252 oxalate extractable aluminum measurements. The function is based on the finding that the Al solubility in the
253 upper B-horizon of Podzols is closely related to the molar ratio of aluminum to carbon in pyrophosphate extracts,

254 and that below the threshold value of 0.1, Al solubility increases with the Al_p/C_p ratio (Simonsson and Berggren,
255 1998). Thus, a function was developed for application to our own measurements of Al_{ox} and C_{tot} based on the
256 assumption that it is possible to use the Al_{ox}/C_{tot} ratio instead of the Al_p/C_p ratio. Data on soil solution DOC were
257 available from lysimeters installed at 50 cm depth for plot K4 and K1 in Asa and 10B and 14B in Flakaliden, and
258 these values were also applied to soil depths below 50 cm (H. Grip, unpublished data). The E-horizon (0–10 cm
259 at Flakaliden) and A-horizon (0–10 cm at Asa) were characterized by higher DOC values based on previous
260 findings (Fröberg et al., 2013) and the classification scheme of DOC in Warfvinge and Sverdrup (1995). Partial
261 CO_2 pressure values in the soil were taken from the default estimate of Warfvinge and Sverdrup (1995).

262

263 Other site-specific parameters that were kept constant between PROFILE runs were evapotranspiration,
264 temperature, atmospheric deposition, precipitation, runoff and nutrient uptake. Temperature is one of the important
265 factors that regulate the weathering rate, and for steady-state calculations in PROFILE the mean annual
266 temperature is used. Kronnäs et al. (2019) demonstrates how weathering rates varies between seasons, due to e.g.
267 variations in temperature. Precipitation is used in PROFILE to calculate vertical water flow through the soil profile.
268 The main effect of precipitation on weathering rates is its impact on soil moisture, but in PROFILE soil moisture
269 is not internally modeled, but given as input. An estimate of the average evaporation per site was derived from
270 annual averages of precipitation and runoff data using a general water balance equation. Deposition data from two
271 sites of the Swedish ICP Integrated Monitoring catchments, Aneboda (for Asa) and Gammtratten (for Flakaliden)
272 (Löfgren et al., 2011) were used to calculate the total deposition. The canopy budget method of Staelens et al.
273 (2008) was applied as in Zetterberg et al. (2014) for Ca^{2+} , Mg^{2+} , K^+ , Na^+ . The canopy budget model is commonly
274 used for elements that are prone to canopy leaching (Ca^{2+} , Mg^{2+} , K^+ , Na^+ , SO_4^{2-}) or canopy uptake (NH_4^+ , NO_3^-)
275 and calculates the total deposition (TD) as the sum of dry deposition (DD) and wet deposition (WD). Wet
276 deposition was estimated based on the contribution of dry deposition to bulk deposition, both for base cations and
277 anions, using dry deposition factors from Karlsson et al. (2012, 2013). Base cation and nitrogen accumulation rate
278 in above-ground tree biomass (i.e. bark, stemwood, living and dead branches, needles) was estimated as the
279 average accumulation rate over a 100 years rotation length in Flakaliden compared to a 73 years rotation length in
280 Asa. These calculations were based on Heureka simulations using the StandWise application (Wikström et al.,
281 2011) for biomass estimates in combination with measured nutrient concentrations in above-ground biomass (S.
282 Linder unpubl. data).

283 **2.7 A definition of significant discrepancies between W_{A2M} and W_{XRPD}**

284 A consequence of the mathematical structure of the A2M program is that the final solution space of possible
285 quantitative mineralogies produces an uncertainty range of weathering estimates, but in a different sense than the
286 uncertainty caused by e.g. uncertainties in chemical analyses, because all mineralogies produced within this range
287 are equally likely. Thus, here we define a significant discrepancy between W_{XRPD} and W_{A2M} to occur when the
288 former is outside the range of the latter, as illustrated in Fig. 1a. The opposite case is a non-significant discrepancy,
289 when the weathering rates based on XRPD are contained in the weathering range based on A2M (Figure 1b).

290

291 The uncertainty range of W_{A2M} can potentially be reduced by reducing uncertainties in analyses of soil
292 geochemistry but most particularly by definitions of available minerals and their stoichiometry. Furthermore, some

293 discrepancies between W_{XRPD} and W_{A2M} might also arise due to limitations of the XRPD method, particularly
294 when minerals occur near or below the detection limit.

295 **2.8 Statistical analyses**

296 In order to quantify the effect of mineralogy on PROFILE weathering rates two statistical measures were used to
297 describe the discrepancies between W_{XRPD} and W_{A2M} . Firstly, root mean square errors (RMSE) of the differences
298 between W_{XRPD} and the arithmetic mean of weathering rates based on regional and site-specific mineralogy, i.e.,
299 $W_{A2M-reg}$ and $W_{A2M-site}$, were calculated:

$$301 \quad RMSE = \sqrt{\frac{1}{n} \sum_{i=1}^n (W_{XRPD_i} - W_{A2M_i})^2} \quad \text{Eq. (1)}$$

302
303 RMSE's were calculated individually for each element, soil layer and soil profile for two data sets. An RMSE
304 expressing the error of the aggregated, total weathering rates in the 0–50 cm soil horizon was calculated to test our
305 first hypothesis (RMSE of total weathering). In addition, an RMSE expressing the errors originating from
306 discrepancies between W_{XRPD} and W_{A2M} for individual minerals was also calculated (RMSE of weathering by
307 mineral). In the latter case, sums of RMSE's by mineral were calculated for each element and soil profile by
308 analogy with the summing up of weathering rates for the whole 0–50 cm soil profile.

309
310 Secondly, relative discrepancies (i.e. average percentage of over- or underestimation of W_{A2M} compared to W_{XRPD})
311 were calculated as the absolute discrepancy divided by the measured value.

$$313 \quad \text{Relative error} = \left(\frac{W_{A2M_i} - W_{XRPD_i}}{W_{XRPD_i}} \right) 100 \quad \text{Eq. (2)}$$

314
315 Relative errors were calculated for each site by comparing the average sum of W_{A2M} in the upper mineral soil (0–
316 50 cm) with the sum of W_{XRPD} in the upper mineral soil.

317 Statistical plotting of results was performed using R (version 3.3.0) (R Core Team, 2016) and Excel 2016.

318 **3. Results**

319 **3.1 Weathering rates based on XRPD mineralogy**

320 Weathering estimates with PROFILE are hereafter presented as the sum of weathering rates in the 0–50 cm soil
321 horizon, since this soil depth is commonly used in weathering rate studies. Information on individual, and deeper
322 soil layers (50-100 cm) is given in Table S2.

323
324 Weathering rates of the base cations based on quantitative XRPD mineralogy (W_{XRPD}), i.e. the reference
325 weathering rates, were ranked in the same order at both sites, with $Na > Ca > K > Mg$ (Table S2). On average,
326 weathering rates of Na, Ca, K and Mg at Asa were 17.7, 8.4, 5.6 and 3.6 $\text{mmol}_c \text{ m}^{-2} \text{ yr}^{-1}$, respectively.
327 Corresponding figures for Flakaliden were of similar magnitude, i.e., 14.8, 9.8, 5.7 and 5.6 $\text{mmol}_c \text{ m}^{-2} \text{ yr}^{-1}$. The
328 variation in weathering rates between soil profiles was smaller at Asa than at Flakaliden, as the standard deviation

329 in relation to the means for different elements ranged between 0.2-2.3 at Asa, and 2.0 –5.7 at Flakaliden (Table
330 S2).

331

332 3.2 Comparison between weathering rates based on XRPD and A2M mineralogy

333 At Flakaliden, $W_{A2M-site}$ was generally in closer agreement with W_{XRPD} than $W_{A2M-reg}$ (Fig. 2b), in line with the first
334 hypothesis. The discrepancies between W_{XRPD} and W_{A2M} were small and non-significant for Mg regardless of the
335 mineralogy input used in A2M, although the estimated discrepancies were reduced when site-specific mineralogy
336 was used. The use of regional mineralogy in A2M underestimated K release rates compared to W_{XRPD} , and the
337 discrepancy was significant. Using site-specific mineralogy resulted in smaller and non-significant discrepancy
338 for K release rates. A similar response to different mineralogies was revealed for Ca, although the result varied
339 more among soil profiles. In contrast to K and Ca, the release of Na was overestimated by both $W_{A2M-site}$ and W_{A2M-}
340 reg compared to W_{XRPD} . The discrepancies were significant regardless of the mineralogy input used in A2M,
341 although using site-specific mineralogy slightly reduced the discrepancy. The generally closer agreement between
342 $W_{A2M-site}$ and W_{XRPD} than $W_{A2M-reg}$ at Flakaliden was also indicated by the lower RMSEs of total weathering for all
343 base cations when site-specific mineralogy was used (Fig. 3a). Relative RMSE were below 20 % for $W_{A2M-reg}$, but
344 below 10 % for $W_{A2M-site}$. However, RMSE for Na was only slightly smaller for $W_{A2M-site}$ than $W_{A2M-reg}$ (16 % for
345 $W_{A2M-site}$).

346

347 PROFILE weathering rates for Asa revealed a different pattern compared to Flakaliden, and the results for Ca, Mg
348 and K were contradictory to hypothesis one. $W_{A2M-reg}$ was in close agreement with W_{XRPD} for K, Ca and Mg, and
349 the small discrepancies were non-significant (Fig. 2a). However, $W_{A2M-reg}$ for Na was consistently overestimated
350 compared to W_{XRPD} and the discrepancies were significant. Using site-specific mineralogy improved the fit
351 between W_{XRPD} and W_{A2M} for Na but had rather the opposite effect on the other base cations at this site. For K, Ca
352 and Mg, $W_{A2M-site}$ overestimated weathering rates, and resulted in significant discrepancies, and larger RMSE,
353 whereas the discrepancies for Na were reduced and non-significant (Fig. 3a). At Asa, the highest relative RMSEs
354 of total weathering occurred for Ca and Mg with $W_{A2M-site}$ (> 30 %) (Fig. 3a). Large standard deviations were due
355 to a single soil profile, F4. The better consistency with $W_{A2M-reg}$ was indicated by RMSE below 10 % for Ca and
356 Mg, and that RMSE for Mg was half of the error with $W_{A2M-site}$. Only for Na, RMSE was lower for $W_{A2M-site}$ than
357 with $W_{A2M-reg}$.

358

359 A complementary illustration of the relationships between weathering rates based on XRPD and A2M is shown in
360 Fig. 4 and provided as Tables S3 and S4, which includes all data from individual soil layers 0–50 cm. A general
361 picture is that $W_{A2M-site}$ was less dispersed along the 1:1-line than $W_{A2M-reg}$, in particular for Flakaliden. On the
362 other hand, for weathering rates in the lower range (< 5 mmol_c m⁻² yr⁻¹) site-specific mineralogy tended to generate
363 both over- and underestimated weathering rates. In most soil profiles, deviations from the 1:1-line were frequent
364 in soil layers below 20 cm. For Na under- and overestimations occurred in the whole range of weathering
365 estimates,

366 3.3 Mineral-specific contribution to weathering rates

367 In spite of its intermediate dissolution rate plagioclase was, due to its abundance, the most important Na-bearing
368 mineral determined in this study (Table 3 and Fig. 5). Plagioclase is a variable group of minerals with different
369 stoichiometric proportions of Ca and Na, from the purely sodic albite on the one hand to the purely calcic anorthite
370 on the other hand (Table S5) as well as with intermediate compositions (Table S6). For simplicity, they will be
371 referred to in this study as sodic and calcic plagioclase. Based on the same quantitative mineralogy (i.e. same
372 elemental compositions and identity of minerals), W_{XRPD} and $W_{\text{A2M-site}}$ gave strong weight to both calcic and sodic
373 plagioclase in estimating Na release rates, but $W_{\text{A2M-site}}$ gave stronger weight to calcic versus sodic plagioclase at
374 Asa, and vice-versa at Flakaliden (Fig. 5). In spite of these differences, the resultant release rates of Na according
375 to $W_{\text{A2M-site}}$ and W_{XRPD} were rather similar (Fig. 5).

376
377 Total Na release rates of $W_{\text{A2M-reg}}$ compared to W_{XRPD} were moderately overestimated. The relative RMSE of
378 weathering by specific Na-containing minerals were of more similar magnitude for Na at Flakaliden compared to
379 Asa (Fig. 3b). However, the standard deviations of RMSE were relatively large at Flakaliden, due to large RMSE
380 for albite in one specific soil profile (11B) (Table S7). Contrary to relative RMSE of total weathering, the relative
381 RMSE of weathering by specific minerals was lower for Na at Asa with regional than site-specific mineralogy.

382
383 According to W_{XRPD} , calcic plagioclase weathering was the most important source to Ca release at Flakaliden, and
384 the second most important source at Asa after epidote (Fig. 5). As for Na, $W_{\text{A2M-site}}$ gave stronger weight to calcic
385 plagioclase than W_{XRPD} at Asa. It was the other way around for $W_{\text{A2M-site}}$ at Flakaliden and the regional mineralogy
386 (i.e. W_{XRPD} gave stronger weight to calcic plagioclase than $W_{\text{A2M-site}}$). Another important Ca source in weathering
387 estimates based on A2M was apatite. This mineral was not detected in the XRPD analyses but was included in
388 both A2M mineralogies as a necessary means to allocate measured total phosphorus content to a specific mineral
389 (Casetou-Gustafson et al. 2018).

390
391 Similar to Na, relative RMSE of weathering by Ca-containing minerals were several magnitudes larger than RMSE
392 of the total weathering of Ca. In other words, although an overall similar weathering rates might be generated by
393 the PROFILE model based on different quantitative mineralogies, the underlying modelled contributions from
394 different minerals can be markedly different. At Flakaliden, the mean relative RMSE by specific minerals were
395 larger for regional than site-specific mineralogy at Flakaliden (Fig. 3b). However, the difference was not
396 significant since the standard deviations were high, probably due to larger RMSE for Ca-bearing minerals in soil
397 profile 11B (Table S7).

398
399 A general picture of the mineral contribution to Mg release is that W_{XRPD} placed most weight to amphibole whereas
400 in W_{A2M} Mg release was more equally distributed among other minerals, notably hydrobiotite, trioctahedral mica
401 and vermiculite. At Asa, and to an even larger extent at Flakaliden, Mg release by A2M mineralogies was
402 determined by a higher contribution of minerals with high dissolution rates (Fig. 5 and Table 3) (i.e. In $W_{\text{A2M-site}}$,
403 hydrobiotite and trioctahedral mica; In $W_{\text{A2M-reg}}$, muscovite and vermiculite at Asa and biotite and illite at
404 Flakaliden). At Asa, less weight was given to amphibole by $W_{\text{A2M-site}}$ compared to W_{XRPD} . At Flakaliden, the $W_{\text{A2M-}}$
405 $_{\text{site}}$ was close W_{XRPD} in spite of the very different allocations of weathering rates to different minerals. The

406 underestimation of Mg release by $W_{A2M-reg}$ was largely explained by the lower weight given to amphibole in both
407 A2M scenarios (Fig. 5). However, A2M gave larger weight to other minerals. The sums of RMSEs of weathering
408 from specific Mg-bearing minerals were much larger for regional than site-specific mineralogy at Flakaliden and
409 reached a maximum value of 156 %. A contributing factor were generally larger RMSE for the mineral
410 contribution of amphibole to Mg weathering and the fact that pyroxene contributed to the RMSEs of the total
411 weathering of Mg. Furthermore, a large standard deviation for the sum of RMSE of specific minerals (Fig. 3b)
412 was caused by soil profile 11B where more weight was placed on amphibole and biotite in contributing to Mg
413 weathering (Table S7). The two A2M mineralogies resulted in the same RMSEs for Mg-bearing minerals at Asa
414 (Fig. 3b).

415

416 Potassium release rates were largely dominated by K-Feldspar weathering in both W_{XRPD} and $W_{A2M-site}$. However,
417 K release by $W_{A2M-reg}$ (Fig. 5) were largely determined by micas at both sites. Together with Mg, these elements
418 had also the lowest weathering rates, indicating that differences between $W_{A2M-reg}$ and W_{XRPD} in relative terms were
419 not correlated with the magnitude of weathering. Unlike the other base cations, relative RMSE of K-bearing
420 minerals were lower at both sites when site-specific mineralogy was used instead of regional (Fig. 3b), and the
421 mineral specific RMSEs were also of similar magnitude as the RMSE of the total weathering (Fig.3a). $W_{A2M-site}$ of
422 K (Fig. 3b), were not several magnitudes larger than RMSE of the total weathering (Fig. 3a). The largest relative
423 RMSEs of K-containing minerals were reached by $W_{A2M-reg}$ at Flakaliden in soil profile 11B, indicated by the high
424 standard deviation.

425 4. Discussion

426 4.1 General range of weathering rates in relation to expectations from other sensitivity studies, and the 427 range of discrepancies between W_{XRPD} and W_{A2M}

428 To our knowledge, the present study is the first to have examined the sensitivity of the PROFILE model on real
429 case study differences of directly measured mineralogy versus indirectly determined normative mineralogy.
430 However, a few systematic studies have been made previously to test the influence of mineralogy inputs, amongst
431 other input parameters, to PROFILE weathering rates. Jönsson et al. (1995) concluded that uncertainty in
432 quantitative mineralogy could account for a variation from the best weathering estimate of about 20 %, and that
433 variations in soil physical and chemical parameters could be more important. The sensitivity analysis of Jönsson
434 et al. (1995) was made by a Monte Carlo simulation where mineralogical inputs were varied by ± 20 % of abundant
435 minerals, and up to ± 100 % of minor minerals. Shortly after, Hodson et al. (1996) examined the sensitivity of the
436 PROFILE model with respect to the sensitivity of weathering of specific minerals and concluded that large
437 uncertainties especially in soil mineralogy, moisture, bulk density, temperature and surface area determinations
438 will have a larger effect on weathering rates than was reported by Jönsson et al. (1995).

439 Compared with the sensitivity analyses by Jönsson et al. (1995), the range of uncertainty in dominating mineral
440 inputs used in the present study was of similar order of magnitude. For this study we used the XRPD measured
441 (M_{XRPD}) and A2M estimated mineralogies (M_{A2M}) determined by Casetou-Gustafson et al. (2018). For example,
442 they concluded that $M_{A2M-reg}$ produced a low relative RMSE of total plagioclase (7 – 11 %) but higher relative
443 RMSE for less abundant minerals, such as dioctahedral mica (90 – 106 %). They also showed that when regional

444 mineral identity and assumed stoichiometry was replaced by site-specific mineralogy ($M_{A2M-site}$), the bias in
445 quantitative mineralogy was reduced.

446 Thus, given this bias in quantitative mineralogy input to PROFILE, discrepancies of W_{A2M} from W_{XRPD} at our
447 study sites should have been on the order of 20 % or less, and site-specific mineralogy inputs should produce
448 weathering rates with lower discrepancies than regional mineralogy. The result of this study was in agreement
449 with this expectation for all elements at Flakaliden but only for Na at Asa. The different quantitative mineralogies
450 resulted in discrepancies between W_{A2M} and W_{XRPD} that differed with site (Fig. 3a, 5).

451 **4.2 Is $W_{A2M-site}$ more consistent than $W_{A2M-reg}$?**

452 Our first hypothesis, that using site-specific mineralogy in the PROFILE model compared to regional mineralogy,
453 should result in weathering rates closer to XRPD-based mineralogy, and thus be more consistent, was generally
454 supported for Flakaliden, but only for Na at Asa. This result was revealed from both the occurrence of significant
455 discrepancies as well as the RMSE of the total weathering rates. Thus, the results did not support our first
456 hypothesis in a consistent way. The possible reasons for this outcome are discussed below, based on the analysis
457 of how different minerals contributed to the overall weathering rates.

458 **4.3 How are discrepancies between W_{A2M} and W_{XRPD} correlated to bias in determinations of quantitative** 459 **mineralogy**

460 The version of the PROFILE model used in this study allowed a close examination of the per element weathering
461 rate contributions obtained from different minerals that provide some insight into the causes to the total W_{A2M}
462 discrepancies.

463 **4.3.1 Sodium release rates**

464 A biased determination of mineralogy may not necessarily result in a corresponding bias of PROFILE weathering
465 estimates if the discrepancies are cancelling each other out, and if dissolution rates of the different minerals are
466 rather similar. This was probably the case for Na. At both study sites and for both W_{XRPD} and W_{A2M} , Na release
467 rates were largest for plagioclase minerals. The Na release from $W_{A2M-site}$ and $W_{A2M-reg}$ were close to W_{XRPD} at both
468 study sites (i.e. all weathering rates were in the range of 17-19 $\text{mmol}_c \text{ m}^{-2} \text{ yr}^{-1}$) nonetheless $W_{A2M-site}$ placed more
469 weight to calcic plagioclase and $W_{A2M-reg}$ more weight to albitic plagioclase (Fig.5). Contrary to our second
470 hypothesis, the relatively high precision in total release rates (i.e.<10%; Fig. 3a) of Na was not correlated to the
471 actual low precision in mineral contribution to the total Na release rates (i.e. >30 %; Fig. 3b). The latter can be
472 explained by the fact that in PROFILE all types of plagioclase have the same dissolution rate coefficients (Table
473 3). Due to this, and in combination with the fact that plagioclase type minerals are a major source for Na, the
474 mineralogical uncertainty in estimating Na release rates with PROFILE was relatively low in this study (i.e. <20
475 %). In context, however, we note that it is generally accepted that under natural conditions different plagioclase
476 minerals weather at different rates, (Allen and Hajek, 1989, Blum and Stillings, 1995).

477 **4.3.2 Calcium release rates**

478 According to W_{XRPD} and W_{A2M} , a key mineral for Ca release rates was calcic plagioclase at Flakaliden and epidote
479 at Asa. In line with our second hypothesis, the overestimation of calcic plagioclase in $M_{A2M-site}$ at Asa at the expense

480 of epidote and amphibole (Casetou-Gustafson et al., 2018) was directly reflected in the significant discrepancy
481 and overestimated weathering rates of Ca by $W_{A2M-site}$ compared to W_{XRPD} (Fig. 5, and 1a). This discrepancy was
482 due to differences between $W_{A2M-site}$ and W_{XRPD} in the mineral stoichiometry of calcic plagioclases, and not in
483 geochemistry, as the same geochemical analyses were also used for $W_{A2M-reg}$.

484

485 At Flakaliden, A2M based on site-specific mineralogy overestimated epidote at the expense of amphibole
486 (Casetou-Gustafson et al., 2018), leading to an underestimation of Ca weathering rates from amphibole compared
487 to epidote (Fig. 5). On the other hand, at Asa, it was the regional mineralogy input to A2M that resulted in
488 overestimated amounts of epidote at the expense of dioctahedral vermiculite and amphibole, and this bias was
489 directly reflected in the underestimated release of Ca from amphibole in $W_{A2M-reg}$. Conversely, the relatively small
490 and non-significant discrepancies of Ca release by $W_{A2M-site}$ at Flakaliden and by $W_{A2M-reg}$ at Asa did not depend
491 on a high precision in estimating the contribution from different minerals, since the precision was actually low. In
492 these cases, the good fits seem to be simply coincidental. Owing to differences in dissolution rates, Ca-bearing
493 minerals tend to compensate each other in terms of the total weathering rate that is calculated. This compensatory
494 effect is perhaps the reason why by coincidence, both $W_{A2M-reg}$ and $W_{A2M-site}$ discrepancies for Ca diverge in
495 different directions at Asa compared to Flakaliden.

496

497 Another source of uncertainty associated with the release of Ca is the role of minerals with high dissolution rates
498 that occur in low abundance, for example apatite, pyroxene and calcite. Apatite was included in M_{A2M} , but if present
499 in the soils studied was below the detection limit of 1 wt.% in the XRPD analyses as were pyroxene and calcite
500 (Casetou-Gustafson et al., 2018). Additionally, the assumption made in the A2M calculations that all P determined
501 in the geochemical analyses is allocated to apatite will likely overestimate the abundance of this mineral since soil
502 P can also occur bound to Fe and Al oxides and soil organic matter in acidic mineral soils (Weil and Brady, 2016).
503 The relatively high abundance of paracrystalline Fe-oxyhydroxide and Al-containing allophane and imogolite at
504 Flakaliden indicates that this could be the case, at least at Flakaliden.

505

506 Regarding pyroxene, XRPD might also have failed to detect and quantify pyroxene due to low abundance at
507 Flakaliden (Casetou-Gustafson et al., 2018). Analytical limitations of XRPD would thus imply that W_{XRPD} of Ca
508 might be underestimated at Flakaliden and Asa. However, in the absence of XRPD detection it is also possible that
509 $M_{A2M-reg}$ overestimated the pyroxene contents at Flakaliden. Thus, apatite and pyroxene added relatively large
510 uncertainties to the weathering estimates of Ca at Flakaliden due to the fact that they have a low abundance in
511 combination with very high dissolution rates. In terms of other reactive trace mineral phases, White et al. (1996,
512 2017) has highlighted the importance of small amounts of calcite in intact granitoid rocks and its significance for
513 Ca found in watershed studies. They also noted that in laboratory leaching experiments on the rocks they studied,
514 reactive calcite became exhausted after just 1.5 years. Given the trace concentrations involved and the high
515 solubility of calcite it is doubtful that calcite is or has been of any long-lived significance in the soil profiles
516 studied, even though they are derived largely from rocks of granitic composition. Although, the results of White
517 et al. (1996, 2017) do suggest that calcite present in the in-situ granitoid rocks underlying the soils may well
518 contribute to Ca export from the catchment. Additionally, the overestimation of the slowly weatherable mineral
519 illite by $M_{A2M-reg}$ (Casetou-Gustafson et al., 2018) resulted in an underestimation of Ca release by $W_{A2M-reg}$ at

520 Flakaliden, since less Ca was allocated to the more weatherable minerals. Although, it should also be noted
521 parenthetically that Ca can only occur as an exchangeable cation in illite, it is not an element that occurs as part of
522 the illite crystal structure, such that the 'illite' composition used in PROFILE is in need of some revision.

523 4.3.3 Magnesium release rates

524 At both study sites, a large number of Mg-containing minerals contributed to the release of Mg, but amphibole
525 was the predominant mineral according to W_{XRPD} and $W_{\text{A2M-site}}$. The only significant discrepancy in Mg release
526 rates was revealed for $W_{\text{A2M-site}}$ at Asa, which resulted in an overestimation by 41 %. This overestimation was an
527 effect of underestimated contribution from amphibole in combination with overestimated contributions from
528 hydrobiotite and trioctahedral mica. This result for Asa supported our second hypothesis. At Flakaliden, $W_{\text{A2M-site}}$
529 produced the same shift in the contribution of Mg by minerals, but the net effect was a very small and non-
530 significant discrepancy to W_{XRPD} . As was noted for Ca, the different outcomes of using site-specific mineralogies
531 at Asa and Flakaliden has no systematic underlying pattern.

532 Using PROFILE based on regional mineralogy resulted in surprisingly low and non-significant discrepancies in
533 Mg release rate, despite both the qualitative and quantitative mineralogies being very different from XRPD,
534 particularly at Flakaliden. For example, both pyroxene and illite were included in $M_{\text{A2M-reg}}$, but not in M_{XRPD} . Thus,
535 at Flakaliden, the overestimation of illite in $M_{\text{A2M-reg}}$ caused an underestimation of Mg release rates comparable to
536 the underestimation of Ca release rates.

537 4.3.4 Potassium release rates

538 Weathering of K-feldspar was the most important source of K release by PROFILE regardless of the different
539 types of mineralogy input. Casetou-Gustafson et al. (2018) showed a strong negative correlation between $M_{\text{A2M-reg}}$
540 and M_{XRPD} for two of the major K-bearing minerals observed at both study sites, i.e., illite (or dioctahedral mica,
541 muscovite) and K-feldspar. Contrary to our second hypothesis, the results of the present study demonstrate that
542 over- or underestimation of $W_{\text{A2M-reg}}$ compared to W_{XRPD} cannot be explained by significant negative correlation of
543 illite and K-feldspar in $M_{\text{A2M-reg}}$. However, this is likely related to the fact that illite and K-feldspar have the lowest
544 and also quite similar dissolution rates among minerals included in PROFILE (i.e. the highest dissolution
545 coefficients, Table 3). Although very different inputs in relation to K bearing minerals produced very similar
546 outputs, we note that this appears contradictory to differences in the behaviour of K-feldspars and K-micas as
547 sources of K via weathering to plants as reviewed for example by Thompson and Ukrainczyk (2002). Additionally
548 we note that Hodson et al. (1997) compared reaction rate constants for different minerals from Sverdrup et al.
549 (1990) with their own calculations and the discrepancies were relatively large for some minerals, e.g. muscovite.

550 5. Concluding remarks

551 • Based on comparing the full solution span of normative mineralogy from the A2M program to measured
552 reference mineralogy from XRPD overall similar weathering rates were generated by the different
553 mineralogical inputs to the PROFILE model. However, the underlying contributions from different
554 minerals to the overall rates differed markedly. Although the similarity of overall rates lends some support
555 to the use of normative mineralogy as input to weathering models, the details of the comparison reveal
556 potential short-comings and room for improvements in the use of normative mineralogies.

- 557 • Compared with regional mineralogy, weathering rates based on site-specific mineralogy were more
558 comparable to the reference rates generated from XRPD mineralogy, in line with hypothesis 1, at one of
559 the study sites (Flakaliden), but not at the other (Asa). Thus, although intuitively the more detailed site
560 specific quantitative mineralogy data might be expected to give more comparable results, this is not
561 supported by this study.
- 562 • For Ca and Mg the differences between weathering rates based on different mineralogies could be
563 explained by differences in the content (modelled or actual) of some specific Ca- and Mg-bearing
564 minerals, e.g. amphibole, apatite, pyroxene, calcite and illite. Improving certainty in relation to presence
565 versus absence of some of these minerals and if present accurate quantification at low levels would reduce
566 weathering rate calculation uncertainties.
- 567 • High uncertainties in mineralogy, due for example to different A2M assumptions, had surprisingly low
568 effect on the weathering from Na- and K-bearing minerals. This can be explained by the fact that the
569 weathering rate constants for the minerals involved, e.g. the plagioclase feldspars and K-feldspar and
570 dioctahedral micas, are similar in PROFILE such that they compensate each other in the overall
571 weathering rate outputs for these elements, a situation that is unlikely to reflect reality.
- 572 • For more in-depth analysis of the uncertainties in weathering rates caused by mineralogy, the rate
573 coefficients of minerals should be revisited and their uncertainties assessed. A revision of rate constants
574 could lead to results more in line with hypothesis 1.

575

576 **6. Authors contribution**

577 Authors contributed to the study as in the following: S. Casetou-Gustafson: study design, data treatment, PROFILE
578 model analyses, interpretation and writing. C. Akselsson: study design, PROFILE model development,
579 interpretation and writing. B.A. Olsson: study design, interpretation and writing. S. Hillier: interpretation and
580 writing.

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721 **Table 1.** Characteristics of the study sites.

| Site | Asa | Flakaliden |
|---|-----------------------|--|
| Coordinates ^a | 57° 08' N; 14° 45' E | 64° 07' N; 19° 27' E |
| Elevation (m a.s.l.) ^a | 225-250 | 310-320 |
| Mean annual precipitation (mm) ^b | 688 | 523 |
| Mean annual air temperature (°C) ^b | 5.5 | 1.2 |
| Bedrock ^c | Acidic intrusive rock | Quartz-feldspar-rich sedimentary rock |
| Soil texture ^d | Sandy loam | Sandy loam |
| Type of quaternary deposit ^d | Sandy loamy till | Sandy loamy till |
| Soil moisture regime (Soil taxonomy) ^e | Udic | Udic |
| Soil type (USDA soil taxonomy) ^e | Spodosols | Spodosols |
| Region/province ^f | 3 | 1 |

^aBergh et al. 2005

^bLong-term averages of annual precipitation and temperature data (1961-1990) from nearest SMHI meteorological stations (Asa: Berg; Flakaliden: Kulbäcksliden)

^cSGU bedrock map (1:50000)

^dSoil texture based on own particle size distribution analysis by wet sieving according to ISO 11277

^eUSDA Soil Conservation service, 2014

^f Warfvinge and Sverdrup (1995)

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723 **Table 2.** PROFILE parameter description.

| Parameter | Description | Unit | Source |
|--|-------------|--|--|
| Temperature | Site | °C | Measurements from nearby SMHI stations |
| Precipitation | Site | m yr | Measurements from nearby SMHI stations |
| Total deposition | Site | mmol _c m ⁻² yr ⁻¹ | Measurements of open field and throughfall deposition available from nearby Swedish ICP Integrated Monitoring Sites |
| BC net uptake | Site | mmol _c m ⁻² yr ⁻¹ | Previously measured data from Asa and Flakaliden: Element concentration in biomass from Linder (unpublished data). Biomass data from Heureka simulations. |
| N net uptake | Site | mmol _c m ⁻² yr ⁻¹ | Previously measured data from Asa and Flakaliden: Element concentration in biomass from Linder (unpublished data). Biomass data from Heureka simulations. |
| BC in litterfall | Site | mmol _c m ⁻² yr ⁻¹ | Literature data for element concentrations from Hellsten et al. 2013 |
| N in litterfall | Site | mmol _c m ⁻² yr ⁻¹ | Literature data for element concentrations from Hellsten et al. 2013 |
| Evapofraction | Site | Fraction | Own measurements and measurements from nearby Swedish Integrated Monitoring Sites |
| Mineral surface area | Soil | m ² m ⁻³ | Own measurements used together with Eq. 5.13 in Warfvinge and Sverdrup (1995) |
| Soil bulk density | Soil | kg m ⁻³ | Own measurements |
| Soil moisture | Soil | m ³ m ⁻³ | Based on paragraph 5.9.5 in Warfvinge and Sverdrup (1995) |
| Mineral composition | Soil | Weight fraction | Own measurements |
| Dissolved organic carbon | Soil | mg l ⁻¹ | Previously measured data from Asa and Flakaliden: Measurements for B-horizon from Harald Grip and previously measured data from Fröberg et al. 2013 |
| Aluminium solubility coefficient | Soil | kmol m ⁻³ | Own measurements for total organic carbon and oxalate extractable aluminium together with function developed from previously published data (Simonsson and Berggren, 1998) |
| Soil solution CO ₂ partial pressure | Soil | atm. | Base on paragraph 5.10.2 in Warfvinge and Sverdrup (1995) |

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726 **Table 3** Mineral dissolution rate coefficients ($\text{kmol}_c \text{ m}^{-2} \text{ s}^{-1}$) used in PROFILE for the reactions with H^+ , H_2O ,
727 CO_2 and organic ligands (R^-) (Warfvinge and Sverdrup, 1995).

| Mineral | pKH | pKH ₂ O | pKCO ₂ | pKR |
|----------------------|------|--------------------|-------------------|------|
| Pyroxene | 12.3 | 17.5 | 15.8 | 14.4 |
| Apatite | 12.8 | 15.8 | 15.8 | 19.5 |
| Hornblende | 13.3 | 16.3 | 15.9 | 14.4 |
| Epidote | 14 | 17.2 | 16.2 | 14.4 |
| Plagioclase | 14.6 | 16.8 | 15.9 | 14.7 |
| K-Feldspar | 14.7 | 17.2 | 16.8 | 15 |
| Biotite | 14.8 | 16.7 | 15.8 | 14.8 |
| Chlorite | 14.8 | 17 | 16.2 | 15 |
| Vermiculite | 14.8 | 17.2 | 16.2 | 15.2 |
| Muscovite and Illite | 15.2 | 17.5 | 16.5 | 15.3 |

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745 **Figure captions**

746 **Figure 1.** The first scenario for describing the effect of mineralogy on weathering rates in the upper mineral soil
747 for a specific soil profile (a) happens when the PROFILE weathering rate based on XRPD (reference weathering
748 rates) is not contained in the weathering range produced using PROFILE in combination with the full A2M
749 solution space. There are two possible explanations of why a significant discrepancy introduces an uncertainty
750 range, i.e. (1) due to uncertainties related to the mineralogical A2M input and (2) due to uncertainties related to
751 the limitation of the XRPD method itself (i.e. detection limit). The second scenario (b) occurs when the reference
752 weathering rate is contained in the full A2M weathering span. In this case we speak of ‘non-significant
753 discrepancies’.

754 **Figure 2.** Comparison of PROFILE weathering rates of base cations ($\text{mmol}_c \text{ m}^{-2} \text{ yr}^{-1}$) at Asa (a) and Flakaliden
755 (b) sites in the 0–50 cm horizon based on XRPD mineralogy (vertical dashed lines) with PROFILE weathering
756 rates based on one thousand random regional A2M mineralogies versus one thousand random site-specific A2M
757 mineralogies. Data presented are from four different soil profiles per site. Regional graph for soil profile 10B at
758 Flakaliden is missing since A2M did not calculate 1000 solutions for soil layer 20-30, due to “Non-positive
759 solution”.

760 **Figure 3.** Root-mean square error (RMSE) of average PROFILE weathering rates ($\text{mmol}_c \text{ m}^{-2} \text{ yr}^{-1}$) of one
761 thousand A2M mineralogies per soil layer, compared to weathering rates based on XRPD mineralogy per soil
762 layer. Comparisons are based on the total weathering per element (A) and on the sum of mineral contributions to
763 total weathering per element (B). RMSE describes the prediction accuracy for a single soil layer.

764 **Figure 4.** Comparison of PROFILE weathering rates based on XRPD mineralogy ($\text{mmol}_c \text{ m}^{-2} \text{ yr}^{-1}$) with
765 PROFILE weathering rates based on regional A2M mineralogy (upper figures) versus site-specific mineralogy
766 (lower figures). Each data point represents a mean of one thousand PROFILE weathering rates for a specific soil
767 depth of one of 4 soil profiles per site.

768 **Figure 5.** Comparison of sums of PROFILE base cation weathering rates for different minerals in the upper
769 mineral soil (0-50 cm) based on XRPD mineralogy and the average PROFILE base cation weathering rate (i.e.
770 based on one thousands input A2M mineralogies per mineral) according to the two normative mineralogical
771 methods and for each study site (i.e. Asa site-sepcific, Flakaliden site-specific, Asa regional, Flakaliden
772 regional). For W_{A2M} , relative error (% of W_{XRPD} estimate) are given at the end of each bar to illustrate the
773 average deviation of W_{A2M} and W_{XRPD} in the upper mineral soil. *=significant discrepancy as defined in section
774 2.7. Vrm1=Trioctahedral vermiculite; Vrm2=Diocahdreal vermiculite. Information on chemical compositions of
775 minerals are given in Table S5 and S6.

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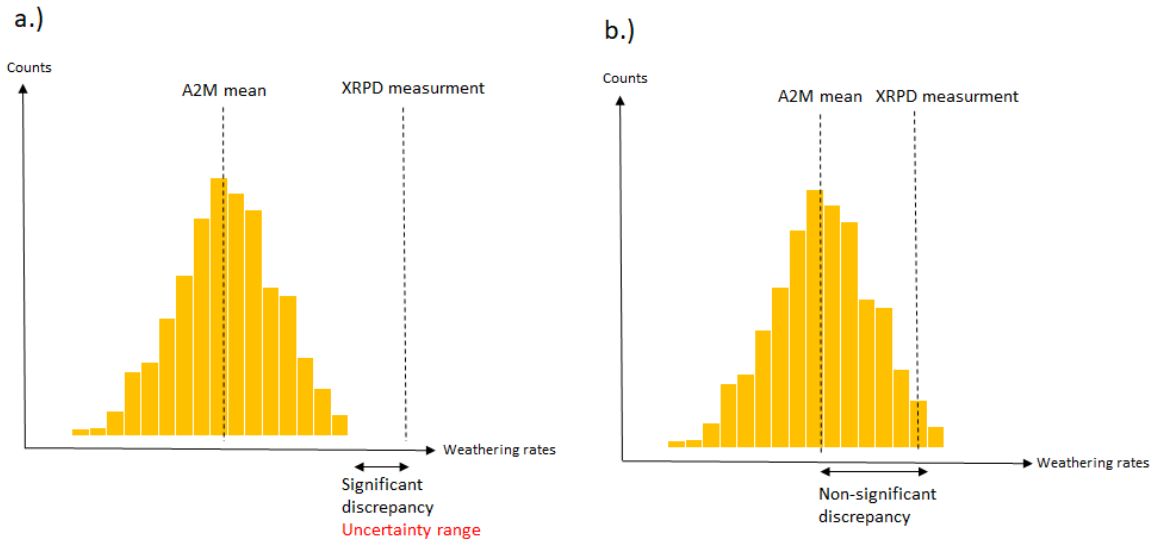
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787 Figure 1a,b

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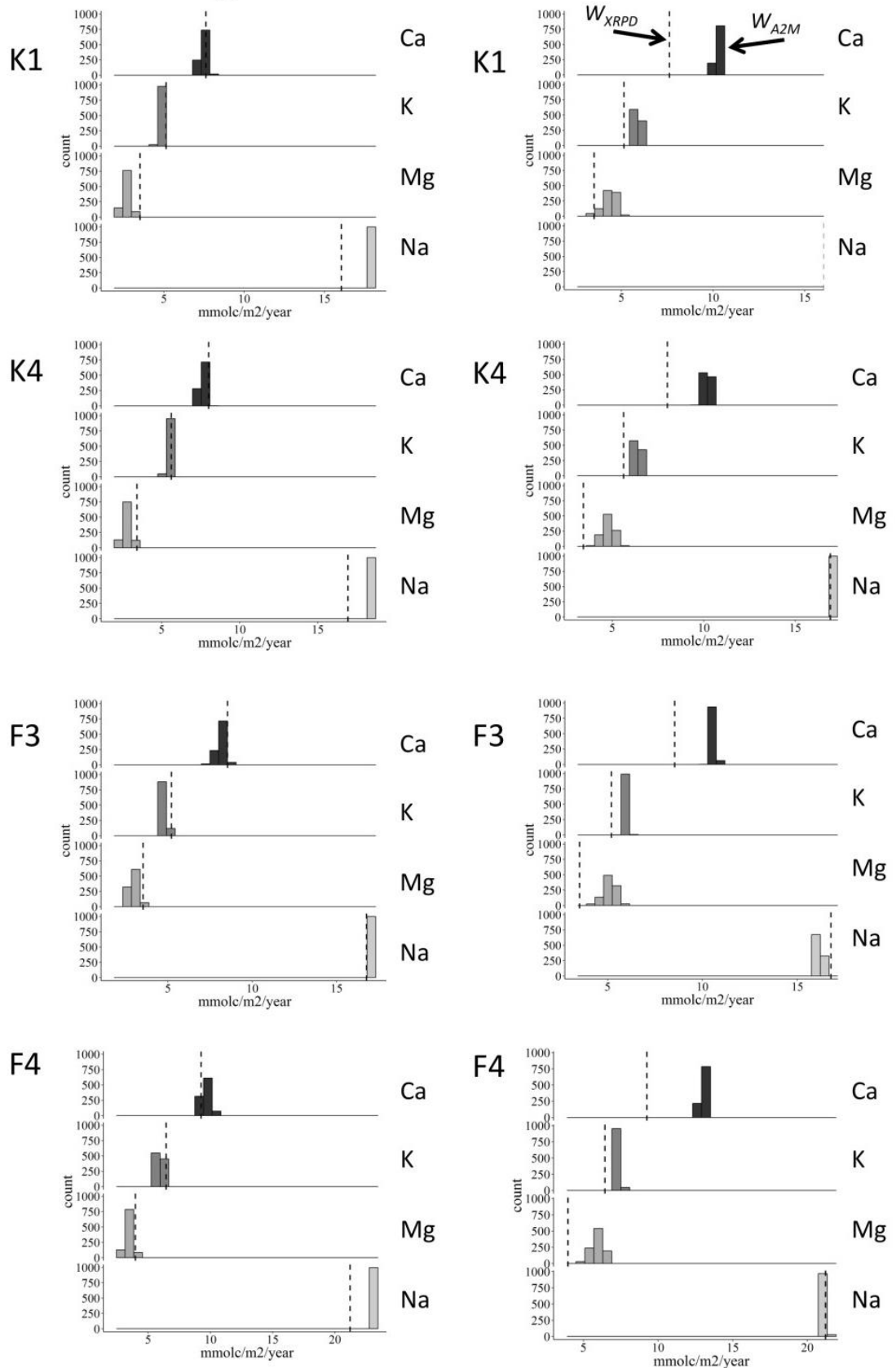
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a.

Asa

Regional

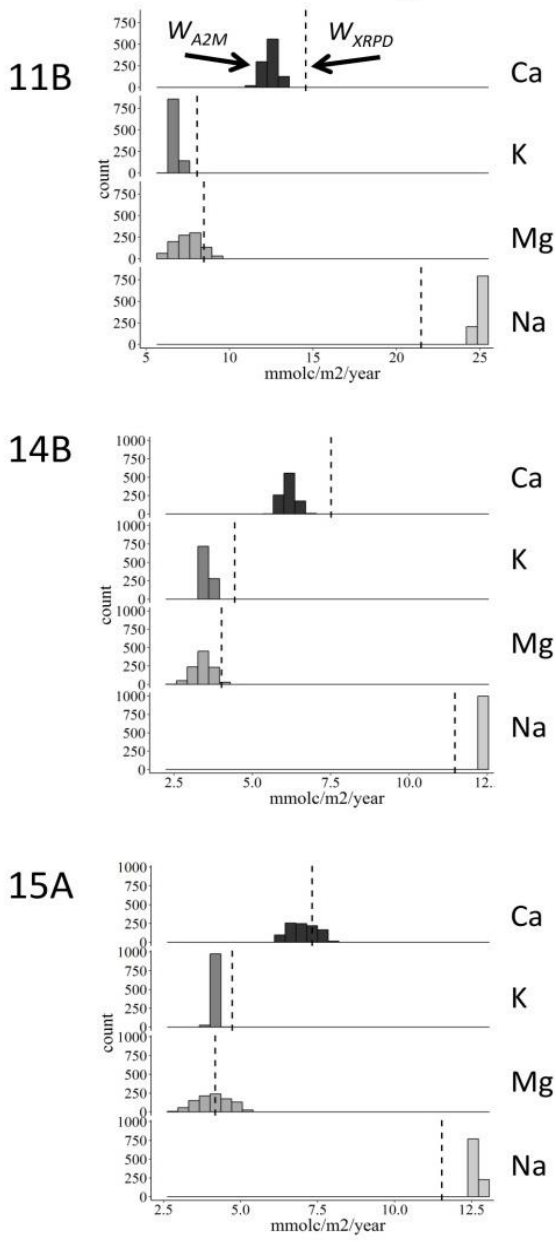
Site-specific



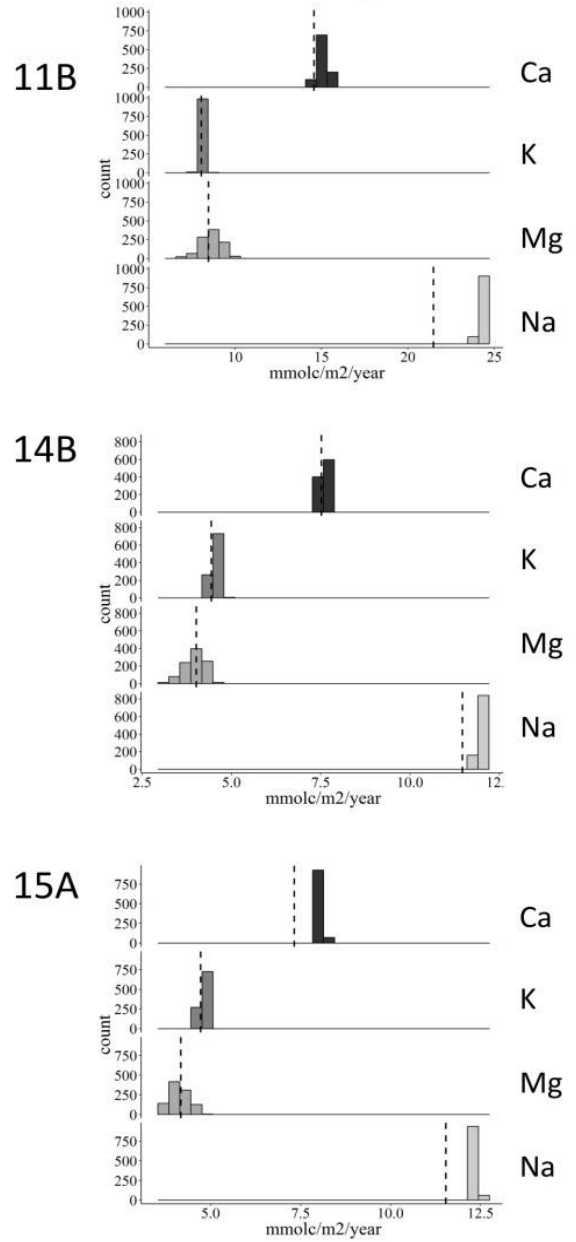
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807 Figure 2a

b. Flakaliden Regional



Site-specific



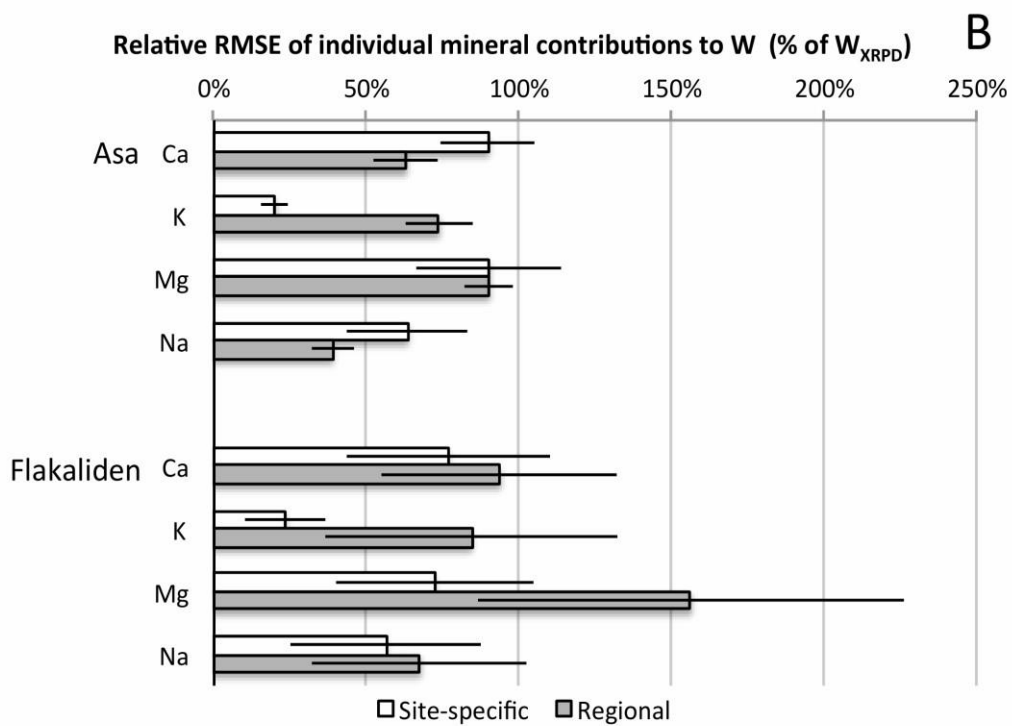
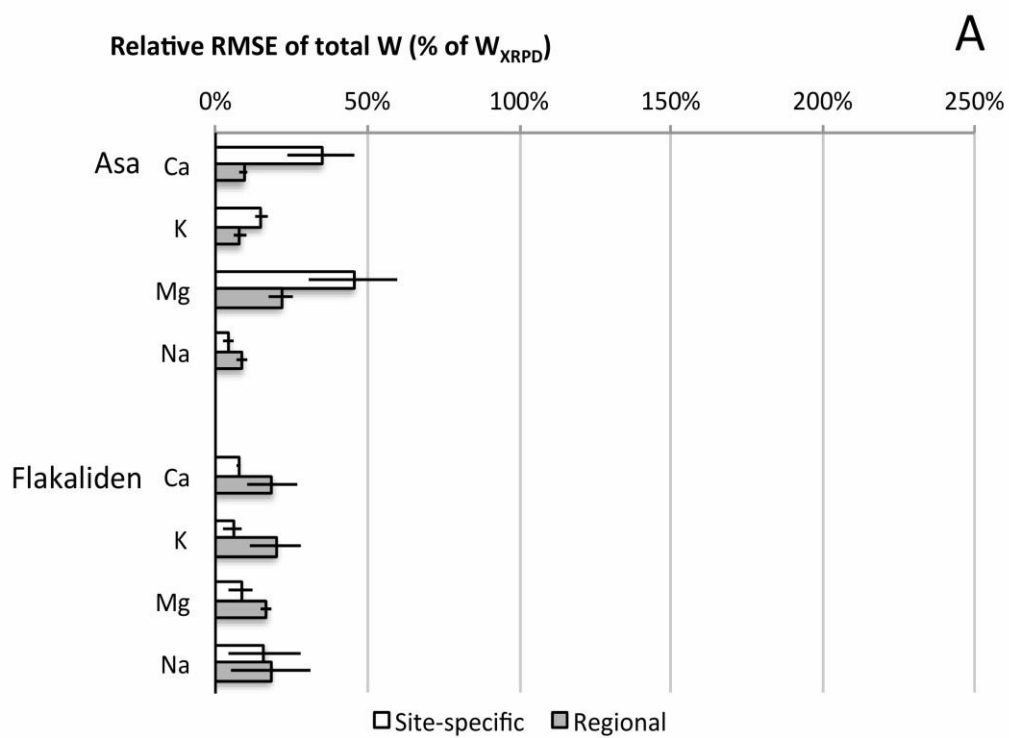
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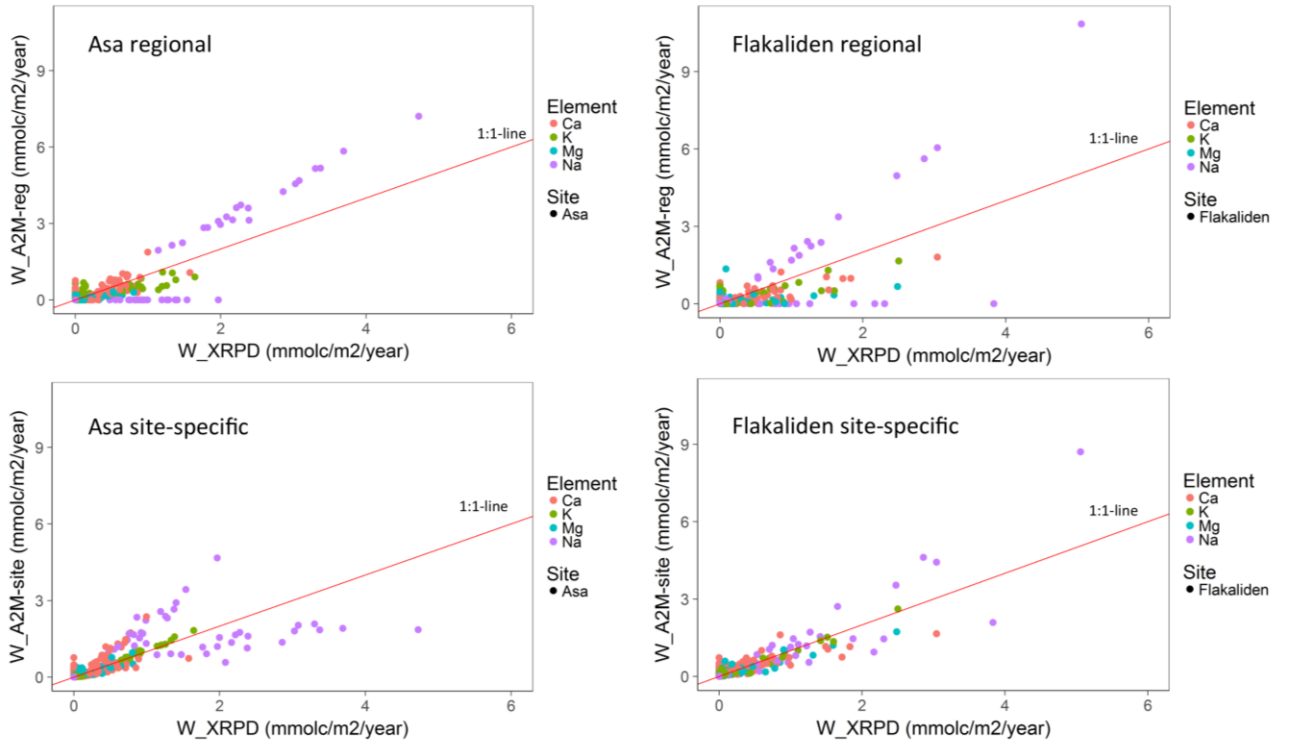


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814 Figure 3, A, B

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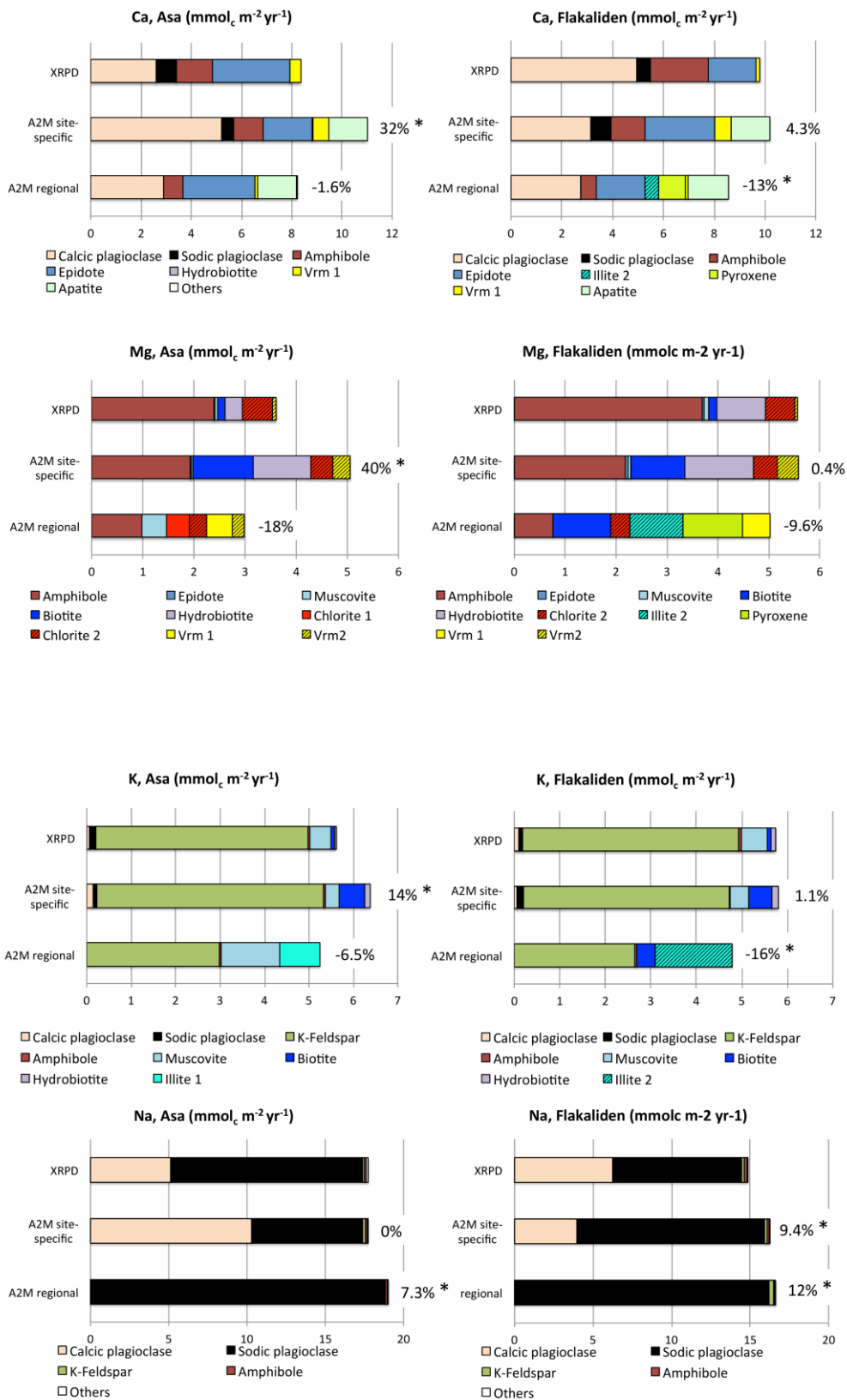


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818 Figure 4

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Figure 5