



1 Rhizosphere to the atmosphere: contrasting methane pathways, fluxes
2 and geochemical drivers across the terrestrial-aquatic wetland
3 boundary
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18 **Key Words:**

19 Diffusion

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24 Iron reduction

25 Carbon cycle



26 **Abstract**

27 *Although wetlands represent the largest natural source of atmospheric CH₄, large*
 28 *uncertainties remain regarding the global CH₄ flux. Wetland hydrological oscillations*
 29 *contribute to this uncertainty, dramatically altering wetland area, water table height, soil*
 30 *redox potentials and CH₄ emissions. This study compares both terrestrial and aquatic CH₄*
 31 *fluxes over two distinct seasons in both permanent and seasonal remediated freshwater*
 32 *wetlands in subtropical Australia. We account for aquatic CH₄ diffusion and ebullition rates,*
 33 *and plant-mediated CH₄ fluxes from three distinct vegetation communities, thereby examining*
 34 *seasonal, diurnal and intra-habitat variability. CH₄ emission rates were related to underlying*
 35 *sediment geochemistry. For example, distinct negative relationships between Fe(III) and SO₄²⁻*
 36 *and CH₄ fluxes were observed, whereas distinct positive trends occurred between CH₄*
 37 *emissions and Fe(II) / AVS, where sediment Fe(III) and SO₄²⁻ were depleted. The highest CH₄*
 38 *emissions of the seasonal wetland were measured during flooded conditions and always during*
 39 *daylight hours, which is consistent with soil redox potential and temperature being important*
 40 *co-drivers of CH₄ flux. The highest CH₄ fluxes were consistently emitted from the permanent*
 41 *wetland (1.5 to 10.5 mmol m⁻² d⁻¹), followed by the Phragmites australis community within the*
 42 *seasonal wetland (0.8 to 2.3 mmol m⁻² d⁻¹), whilst the lowest CH₄ fluxes came from a region of*
 43 *forested Juncus sp. (-0.01 to 0.1 mmol m⁻² d⁻¹) which also corresponded with the highest*
 44 *sedimentary Fe(III) and SO₄²⁻. We suggest that wetland remediation strategies should consider*
 45 *geochemical profiles to help to mitigate excessive and unwanted methane emissions, especially*
 46 *during early system recovery periods.*



47 1.0 Introduction

48 Wetlands are considered one of the most valuable ecosystems on Earth (Costanza et al.,
 49 2014). They are biodiversity hotspots that provide ecosystem services such as water filtration,
 50 sediment trapping, floodwater retention and carbon (C) storage (Bianchi, 2007). Wetlands
 51 account for ~5.5% of terrestrial surfaces (Melton et al., 2013) and have been estimated to store
 52 from ~4% (Bridgham et al., 2014) to ~30% (Mitsch et al., 2013) of Earth's estimated 2500 Pg
 53 soil C pool (Lal, 2008). Pristine wetlands have long been considered net C sinks due to their
 54 high rates of productivity and low rates of decomposition (Mitsch et al., 2013); however due
 55 to their waterlogged nature and anaerobic soils, wetlands are ideal environments for the
 56 production of methane (CH₄), a potent greenhouse gas. As such, wetlands are recognised as
 57 Earth's largest natural source of CH₄ to the atmosphere ($185 \pm 21 \text{ Tg C yr}^{-1}$) (Saunois et al.,
 58 2016).

59 Resolving the drivers, pathways and effects of seasonal weather oscillations on wetland
 60 CH₄ sink or source behaviours is important to enable more accurate climate model projections
 61 and to reduce uncertainties in the global wetland CH₄ budget (Kirschke et al., 2013; Saunois et
 62 al., 2016). Mitsch et al. (2013) estimated that the average ratio of freshwater wetland CO₂
 63 sequestration to CH₄ emissions was 19.5:1. As CH₄ is 34 times more potent than carbon dioxide
 64 (CO₂) over a 100 year time scale (Stocker et al., 2013), this suggests that many freshwater
 65 wetlands may have a net positive radiative forcing effect on climate (Hernes et al., 2018).
 66 However, variability in geomorphology, wetland maturity, salinity and underlying
 67 geochemical composition all contribute to variable CH₄ dynamics (Bastviken et al., 2011;
 68 Poffenbarger et al., 2011; Whiting & Chanton, 2001). The lack of spatially-resolved wetland
 69 CH₄ emission data, as well as the limited number of studies constraining the multiple wetland
 70 CH₄ flux pathways (i.e. ebullition, diffusion and plant-mediated) coupled with ongoing
 71 anthropogenic conversion of wetland systems (Bartlett & Harriss, 1993; Neubauer &
 72 Megonigal, 2015; Saunois et al., 2016) further contribute to the uncertainties around CH₄
 73 regional to global scale budgets.

74 Extensive clearing and drainage of many coastal wetlands has occurred over the
 75 previous two centuries in order to accommodate agriculture, aquaculture and urban development
 76 (Armentano & Menges, 1986; Villa & Bernal, 2018; White et al., 1997). Drained wetlands can
 77 lead to rapid soil organic matter oxidation, and transform systems to net CO₂ sources (Deverel
 78 et al., 2016; Pereyra & Mitsch, 2018). Drainage systems can also reduce wetland inundation



79 periods and alter sediment redox-dependant geochemistry and microbially-mediated reactions
80 (Johnston et al., 2014), particularly those involving bioavailable iron (Fe(III)), sulphate (SO_4^{2-})
81) and nitrate (NO_3^-). Importantly, anaerobic carbon metabolism employing these terminal
82 electron acceptors (Fe(III), SO_4^{2-} , NO_3^-) competes thermodynamically with methanogenic
83 bacteria and archaea and thereby can inhibit CH_4 production (á Norði & Thamdrup, 2014;
84 Burdige, 2012; Karimian et al., 2018; Lal, 2008). With increasing value now placed on the
85 ecosystems services provided by wetlands, many degraded systems are now undergoing
86 remediation and re-flooding (Johnston et al., 2014). However, the ecosystem benefits, such as
87 enhanced biodiversity and water quality, may come at a price in the form of high initial CH_4
88 flux rates, and predicted net radiative forcing for several centuries post-remediation - thus
89 posing a ‘biogeochemical compromise’ (Hemes et al., 2018; Lal, 2008).

90 Within Australia, it has been estimated that more than 50% of natural wetlands have
91 been lost to land use change, drainage and degradation since European settlement (Finlayson
92 & Rea, 1999; ANCA, 1995). This equates to an estimated ~1.2 Pg C emitted to the atmosphere
93 through oxidation of soil organic carbon (Page & Dalal, 2011). Much of eastern Australia’s
94 freshwater coastal wetlands are underlain by Holocene derived sulphidic sediments (i.e pyrite
95 – Fe_2S , known as coastal acid sulphate soils; CASS) formed during periods of higher sea levels
96 (Walker, 1972; White et al., 1997). When CASS are drained, pyrite is oxidised, producing
97 sulphuric acid (H_2SO_4). This results in highly acidic soils with pH levels as low as 3 (Johnston
98 et al., 2014; Sammut et al., 1996). After rainfall events, groundwater transports H_2SO_4 from
99 the CASS landscapes into nearby creeks and estuaries (Sammut et al., 1996). The low pH
100 groundwater discharge also mobilises iron and aluminium, fuels aquatic deoxygenation, and
101 can lead to large fish kills and degradation of infrastructure (Jeffrey et al., 2016; Johnston et
102 al., 2003; White et al., 1997; Wong et al., 2010). Drained CASS wetlands typically contain
103 abundant reactive Fe(III) and exhibit complex sulphur and Fe cycling (Boman et al., 2008;
104 Burton et al., 2011; Burton et al., 2006). Wetland iron and sulfur cycling can profoundly
105 influence CH_4 production and consumption via a series of complex redox reactions coupled
106 with organic matter mineralisation (Holmkvist et al., 2011; Sivan et al., 2014). As such,
107 terminal electron acceptor availability is critical when considering wetland remediation and the
108 biogeochemical compromise paradigm.

109 Here we assess CH_4 emissions rates from a remediated freshwater CASS wetland in
110 subtropical eastern Australia, and compare fluxes from the permanent wetland and the adjacent
111 seasonal wetland ecotypes. We hypothesize that wetland CH_4 emissions will differ



significantly between the seasons and between the four wetland communities. We account for three atmospheric flux pathways for methane; ebullition, diffusion and plant-mediated fluxes, over diurnal cycles and within different seasons. CH₄ fluxes were also assessed in relation to the underlying soil properties, including sulphate, reactive iron III and iron II, acid volatile sulphur, chloride and organic carbon.

117

2.0 Methods

2.1 Study site

Cattai Wetland is located on the mid-coast of New South Wales, Australia. The reserve covers 500 hectares, featuring a shallow permanent wetland covering an area of approximately 16 hectares that is adjacent to a seasonal wetland and floodplain located to the south (Fig. 1). Both sites discharge into the nearby Cooperbrook Creek, a tributary of the larger Manning River estuary. The site was extensively cleared and low-lying areas drained during the early 1900's in order to aid agriculture and development in the region. As a result of this anthropogenic drainage, the oxidation of CASS produced sulphuric acid and episodic acidic discharge to adjacent creeks for many years (Tulau, 1999). To ameliorate acidic discharge, the natural hydrology of the site was restored in 2003 through the decommissioning of agricultural drains and removal of floodgates. Re-flooding of the CASS landscape has reduced the production of sulphuric acid, acid discharge and aluminium and iron mobilisation, hence improving the downstream water quality (GTCC, 2014).

The region receives a mean annual rainfall of 1180 mm with the majority falling during early autumn with an average maximal monthly rainfall in March (152 mm). The lowest rainfall generally occurs during the winter months with average minimal rainfall during September (60 mm). Average minimum and maximum summer temperatures range from 17.6 °C to 29 °C (January) and in winter range from 5.9 °C to 18.5 °C (July) (BOM, 2018). The dominant vegetation type within the permanent wetland is an introduced waterlily species (*Nymphaea capensis*), while the fringes of the wetland consist of wetland tree species; *Casuarina* sp. and *Melaleuca quinquenervia*. The seasonal wetland to the south is dominated by the sedge; *Juncus kraussii* (Veg A) and features scattered stands of *Phragmites australis* (Veg B) with areas of slightly higher elevation dominated by *Juncus kraussii* below *Casuarina* sp. (Veg C) (Fig. 1).

142



143 **2.2 The aquatic CH₄ flux of the permanent wetland**

144 To quantify CH₄ ebullition rates, up to 12 ebullition domes were deployed during two
 145 distinct seasons (detailed below) at ~20 m intervals along a longitudinal transect, from the edge
 146 of the permanent wetland towards the centre. Each dome was carefully suspended below the
 147 water level by flotation rings, ensuring minimal disturbance of sediment and the water column.
 148 Gas samples were extracted from the headspace of each dome using a 300 mL gas tight syringe
 149 at periods of ~48 h. The volume was recorded and each sample then diluted using ambient air
 150 (1:729 ratio) and analysed in situ using a using a manufacturer calibrated cavity ring-down
 151 spectrometer (Picarro G2201-*i*) to determine CH₄ concentrations (ppm). Diffusive CH₄ fluxes
 152 from the permanent wetland were measured using a floating chamber with a portable
 153 greenhouse gas analyser (UGGA, Los Gatos Research). To account for spatial and temporal
 154 variability, measurements were conducted during both day-time and night-time, and sampling
 155 within vegetated areas featuring lilies (*Nymphaea capensis*), forested areas (*Melaleuca* sp.)
 156 and in areas where no aquatic vegetation was present (i.e. open water). A total of 39 CH₄
 157 floating chamber incubations averaging ~8 minutes in duration were recorded over the two
 158 campaigns. The average r^2 value of linear regressions of CH₄ concentrations versus time during
 159 chamber incubations was 0.97 ± 0.05 . One chamber measurement was disregarded as an outlier
 160 (as it was more than three times the standard deviation of the mean) and any chambers capturing
 161 ebullition bubbles (determined by a nonlinear increase in concentration) were also disregarded.
 162 The seasonal ebullition and diffusive CH₄ flux methods and measurements from the permanent
 163 wetland have previously been reported elsewhere (Jeffrey et al. submitted).

164 **2.3 Plant-mediated CH₄ fluxes**

165 Simultaneous time series chamber experiments were conducted over ~24 hours to
 166 measure CH₄ fluxes during each season from the three different wetland vegetation ecotypes.
 167 These ecotypes were *Juncus kraussii* (Veg A), *Phragmites australis* (Veg B) and *Juncus*
 168 *kraussii* amongst *Casuarina* sp. forest (Veg C) (Fig. 1). In each ecotype, 65 x 65 x 30 cm
 169 acrylic bases were installed four months before the first time series experiment, to minimise
 170 disturbance to the sediment profile and vegetative rhizosphere. Vegetative flux chambers were
 171 constructed of an aluminium frame with clear Perspex walls and roof that matched the areal
 172 footprint of the pre-inserted acrylic bases. The chambers were 100 cm, 150 cm and 50 cm high
 173 for at Veg A, B and C respectively. The custom sizes were tailored for the different vegetation



174 heights, whilst minimising chamber volume as much as possible. Each chamber was leak-tested
 175 under laboratory conditions prior to fieldwork.

176 Before each field incubation, chambers were flushed with atmospheric air then
 177 carefully lowered over the vegetation and onto the acrylic base ensuring an air tight seal. A
 178 small fan circulated internal air within each chamber. Air within the chamber was pumped
 179 through a closed loop from the top of the chamber using gas tubing (Bevaline), passing through
 180 a drying agent (Drierite desiccant) and then analysed in situ using a calibrated cavity ring-down
 181 spectrometers (Picarro G2201-*i* or LosGatos), recording the flux rate of CH₄ (ppm/sec). The
 182 gas flow was returned near the base inside each vegetation chamber closing the loop.
 183 Vegetation incubation times ranged from 6 to 15 minutes depending on the flux rate and were
 184 taken from triplicate sites to account for heterogeneity within each ecotype. The daytime
 185 measurements (after sunrise) were measured at ~10 minute intervals whilst night time
 186 measurements (after sunset) were taken at ~4 hourly intervals. CH₄ fluxes from the adjacent
 187 exposed sediments or shallow overlying water at each site were also measured at ~4 hourly
 188 intervals to determine the influence and role of plant-mediated CH₄ fluxes compared to non-
 189 vegetated CH₄ fluxes. Light and temperature loggers (Onset Hobo) measured the changes in
 190 diurnal air temperature (°C) and photosynthetically active radiation (PAR) at each site.

191

192 **2.4 Soil geochemistry and redox conditions**

193 A water logger (Minidiver) was deployed in the permanent wetland before the first
 194 campaign to monitor changes in water depth (cm) and temperature (°C). Field pH (pH_F) and
 195 the redox potential (E_{hF}; reported against standard hydrogen electrode) were determined in
 196 situ, by directly inserting the electrode into the soils (5 cm depth, 8 replicates) at each site. A
 197 composite sampling approach (3 cores) was used to collect sediment samples from each site,
 198 to determine organic C content, Fe(III)_{HCl}, Fe(II)_{HCl}, Cl, SO₄²⁻ and acid volatile sulphur (AVS).
 199 The cores were extracted in December 2016, by inserting a 4.0 cm diameter acrylic tube into
 200 the sediment to a depth of up to 50 cm. Cores were immediately sectioned into 2 cm increments
 201 to a depth of 20 cm, and 5 cm increments thereafter, ensuring higher vertical resolution in the
 202 organic rich near-surface sediments. Samples were immediately placed into air-tight bags, then
 203 frozen within 12 hr of collection at -16°C in a portable freezer and transferred to -80°C freezer
 204 in the laboratory. Frozen samples were thawed in an oxygen-free anaerobic chamber (1-5% H₂



205 in N₂), using an oxygen consuming palladium (Pd) catalyst. The defrosted samples were
206 homogenised using a plastic spatula.

207 AVS content was determined by adding 1-2 g of wet sediment with 6 M HCl:1 M L-
208 ascorbic acid. The liberated H₂S was captured in 5 ml of 3% Zn acetate in 2 M NaOH and then
209 quantified using iodometric titration. The reactive Fe fractions were determined using a
210 sequential extraction procedure optimised for acid sulphate soils based on Claff et al. (2010).
211 Poorly crystalline solid-phase Fe (II) and Fe (III) were determined by extracting 2 g wet sub-
212 samples with cold N₂-purged 1 M HCl for four hours. Aliquots of 0.45 µm-filtered extract were
213 analysed for Fe (II) [Fe(II)_{HCl}] and total Fe [Fe_{HCl}] using the 1,10-phenanthroline method with
214 the addition of hydroxylammonium chloride for total Fe (APHA, 2005). The Fe(III) [Fe(III)_{HCl}]
215 was determined by the difference of [Fe_{HCl}] – [Fe(II)_{HCl}]. Total organic carbon (TOC) and total
216 S (S_{Tot}) were determined via a LECO CNS-2000 carbon and sulfur analyser. Chloride and
217 sulfate concentrations were measured using filtered (0.45 µm) aliquot from a 1:5 water extract
218 of freshly defrosted wet soil, as per Rayment and Higginson (1992) via ion chromatography
219 using a Metrosep A Supp4-250 column, an RP2 guard column and eluent containing 2 mM
220 NaHCO₃, 2.4 mM Na₂CO₃ and 5% acetone, in conjunction with a Metrohm MSM module for
221 background suppression.

222

223 2.5 Calculations

224 Both the air-water and vegetative CH₄ fluxes were calculated for the chamber
225 deployments in the permanent wetland and seasonal wetland using the equation:

$$226 \quad F = (s(V/RT_{\text{air}}A))t \quad (1)$$

227 where s is the regression slope for each chamber incubation deployments (ppm sec⁻¹), V is the
228 chamber volume (m³), R is the universal gas constant, T_{air} is the air temperature inside the
229 chamber (K), A is the surface area of the chamber (m²) and t is the conversion factor from
230 seconds to day, and to mmol.

231 Ebullition rates (E_b) (mmol m⁻² d⁻¹) were calculated using the equation:

$$232 \quad E_b = ([\text{CH}_4] \cdot \text{CH}_4\text{Vol.}) / A \cdot V_m \cdot \text{Td} \quad (2)$$



where $[CH_4]$ is the CH_4 concentration in the collected gas (%), $CH_{4Vol.}$ is the gas volume sampled (L), A is the funnel area (m^2), V_m is the molar volume of CH_4 at in situ temperature (L) and T_d is deployment time (days).

3.0 Results

3.1 Hydrological Conditions

Prior to the first campaign in April 2017 (C1), an extreme hot/drying summer period occurred during early 2017 (Fig. 2). This resulted in an average wetland water column temperature of 23.3 ± 0.7 °C and a water depth in the permanent wetland as low as ~7.3 cm, with exposed sediments along the wetland perimeter during the preceding month. Total rainfall for the two weeks prior to C1 was 342 mm, with an additional 35 mm of rain occurring during C1 fieldwork (Fig. 2) thus raising the water column depth in the permanent wetland to 77.2 cm in less than four weeks. This C1 deployment was therefore categorized as the ‘post-dry/flooded’ period, where air temperatures ranged from 13.3 to 22.8 °C and the average water column temperature in the permanent wetland was 20.4 ± 0.5 °C. The second fieldwork campaign was conducted in September 2017 (C2) under cool/drying conditions, where air temperatures ranged from as low as 3.4 °C to 34.9 °C (Fig. 2), with cooler average water temperatures 12.6 ± 0.4 °C in the permanent wetland (Fig. 2). The depth of the permanent wetland at this time had dropped slightly to ~33 cm (Fig. 2).

3.2 Permanent and Seasonal Wetland CH_4 fluxes

The vegetation time series revealed diurnal variability of plant-mediated CH_4 emissions occurred at most ecotypes, with the highest CH_4 fluxes occurring during daytime around mid-day and the lowest CH_4 fluxes during the night time (Fig. 3, Table 1). The lowest CH_4 fluxes were found at Veg C with a net negative CH_4 flux observed during C2 time series. The CH_4 sediment fluxes measured amongst each vegetation time series were consistently much lower than the plant-mediated CH_4 fluxes indicating that the vegetation was indeed the main conduit for CH_4 to the atmosphere (Fig. 3, Table 1). The CH_4 fluxes were highly variable between the replicates at each site. Temperature and PAR followed similar diurnal trends to each other and had positive correlations to CH_4 emissions (Fig. 3).



CH₄ fluxes from the three vegetation types were higher in C1 than C2 (Fig. 4, Table 1). The highest CH₄ fluxes in each of the vegetation types always occurred during the daytime (Fig. 4, Table 1). *Phragmites sp.* (Veg B) consistently emitted the highest CH₄ fluxes (2.27 ± 1.42 mmol m⁻² d⁻² during C1 and 0.77 ± 0.46 mmol m⁻² d⁻¹ during C2). The Veg C ecotype within the seasonal wetland consistently produced the lowest CH₄ fluxes of all sites, with a net negative flux occurring during C2 (-0.01 ± 0.08 mmol m⁻² d⁻¹).

The permanent wetland showed an inverse trend with seven-fold higher diffusive fluxes during the cool/drying C2 (10.46 ± 15.81 mmol m⁻² d⁻¹) compared to the post-dry/flooded C1 (1.49 ± 2.75 mmol m⁻² d⁻¹), while the ebullition rates were similar during both seasons (Fig. 4, Table 1). Overall, the plant mediated CH₄ fluxes from the three seasonal wetland vegetation ecotypes (Veg A, B and C) were within the range of aquatic fluxes measured from the permanent wetland for the post-dry/flooded C1 time series, but not for the cool/drying C2 time series, when the permanent wetland CH₄ fluxes were much higher (Fig. 4).

3.3 Sediment core profiles and soil redox potentials

Average concentrations from soil cores (Table 1, Fig. 5) were based upon the top 20 cm of the profile, where the highest organic carbon concentrations were found. This upper rhizosphere depth zone is assumed to be an active area of carbon metabolism and CH₄ production and consumption (Nedwell & Watson, 1995). The Fe(III)_{HCl} concentrations were greater than Fe(II)_{HCl} at all three seasonal wetland sites, however the permanent wetland showed an opposite trend with low concentrations of both Fe(III) (5.6 ± 10.7 mmol kg⁻¹) and SO₄²⁻ (1.5 ± 1.0 mmol kg⁻¹) (Fig. 5, Table 1). The highest average concentrations of Fe(III)_{HCl} were found at the Veg C site (204.0 mmol kg⁻¹) and highest and similar concentrations of SO₄²⁻ were in Veg B and Veg C sediments (45.4 ± 41.0 mmol kg⁻¹ and 43.3 ± 16.7 mmol kg⁻¹) (Fig. 5, Table 1). Net positive redox potential was found at all four sites during C1 (under post-dry/flooded conditions) indicating a lag time between recent flooding and the onset of reducing conditions. In contrast, a negative redox potential was found within the permanent wetland and Veg B during C2, indicating reduced conditions under cool drying conditions (Table 1). The TOC concentrations (%) were highest in the upper profiles and similar across all sites (Fig. 5, Table 1) averaging $13.4 \pm 7.6\%$.



294 **3.4 Temperature and PAR**

295 Correlation plots for both temperature (°C) and sunlight (PAR) versus CH₄ emissions
 296 from the three vegetation ecotypes showed no distinct relationships with the exception of Veg
 297 B during C2 for PAR ($r^2=0.18$, $p<0.01$) and temperature (°C) ($r^2=0.35$, $p<0.001$). No clearer
 298 trends were observed by combining all site measurements, nor separating daytime fluxes and
 299 drivers from night time fluxes and drivers.

300

301 **4.0 Discussion**

302 **4.1 Geochemistry of the CASS landscape**

303 Sediment profiles provide insights to the historical geochemical changes that have
 304 occurred across the CASS landscapes of the four Cattai Wetland sites (Fig. 5). If we assume
 305 that relatively uniform deposition of late Holocene materials occurred, the differences between
 306 present day profiles are related to historical changes in hydrology and land use, topographic
 307 elevation, geochemical trajectories and vegetative carbon inputs. For example, the permanent
 308 wetland shows distinct differences to the adjacent seasonal wetland sites, with divergent
 309 geochemical signatures of both iron and sulphate that reflect the sustained inundation (Table
 310 1, Fig 5). The permanent wetland had significantly lower Fe(III) ($p<0.001$) and 11 to 30 fold
 311 lower SO₄²⁻ concentrations within the upper soil profile compared to the seasonal wetland. The
 312 ratio of Fe(III)_{HCl} to Fe(II)_{HCl} from the flooded soils of the permanent wetland was 0.03,
 313 indicating the sediments were almost completely depleted of Fe(III). Under reducing
 314 conditions where there is low SO₄²⁻ and little to no Fe(III) to competitively exclude
 315 methanogenesis, CH₄ production becomes more favourable. Indeed, CH₄ production was on
 316 average highest from the permanent wetland, especially when considering the dual CH₄
 317 pathways of ebullition and air-water diffusion (Table 1).

318 In addition to sulphate reduction, some depletion of the sulphur pool from the
 319 permanent wetland may have occurred due to drainage exports of sulphuric acid (H₂SO₄)
 320 discharging from the CASS landscape throughout the last century. Alternatively, reducing
 321 conditions induced by re-flooding freshwater wetlands is known to encourage the re-formation
 322 of AVS and pyrite (FeS₂) and produce alkalinity, thereby attenuating acid production and
 323 discharge (Burton et al., 2007; Johnston et al., 2014; Johnston et al., 2012) and reducing the
 324 total SO₄²⁻ pool of CASS landscapes. While the AVS concentrations found within the



325 permanent wetland (up to $18.5 \mu\text{mol g}^{-1}$) were a result of sulphate reduction induced by CASS
 326 wetland restoration, they nonetheless represent a relatively volatile form of sulphur, which is
 327 at risk of rapid oxidation during drought periods (Johnston et al., 2014; Karimian et al., 2017).
 328 The AVS concentrations of the permanent wetland sites were more than 20-fold higher than
 329 the three adjacent seasonal wetland sites, and represent a potentially volatile by-product and
 330 consequence of re-flooding CASS soil landscapes, in addition to leading to increases of CH_4
 331 emissions (Table 1).

332 The soil profile from the seasonal wetland Veg C habitat featured abundant $\text{Fe(III)}_{\text{HCl}}$
 333 ($\text{Fe(III)}_{\text{HCl}}$ to $\text{Fe(II)}_{\text{HCl}}$ ratio of 136) and also SO_4^{2-} . This was associated with the lowest fluxes
 334 of CH_4 for both seasonal sampling periods (Fig. 5, Table 1). Relatively low CH_4 fluxes from
 335 Veg C are likely due to the more oxidising conditions present at this site and the surfeit of
 336 thermodynamically favourable terminal electron acceptors (i.e. Fe(III) and SO_4^{2-}), which
 337 would competitively exclude organic matter degradation by methanogenic archaea (Postma &
 338 Jakobsen, 1996).

339 At the other seasonal wetland sites (Veg A and B), the average Fe(III) and SO_4^{2-}
 340 concentrations were intermediate, (i.e. lower than Veg C, but higher than the permanent
 341 wetland), although in the upper profile Veg B had more SO_4^{2-} while Veg A had more Fe(III)
 342 (Fig. 5, Table 1). CH_4 flux values from these sites were also intermediate (Table 1). Sediment
 343 profiles from both Veg A and Veg B indicated a degree of Fe reduction based on the ratio of
 344 $\text{Fe(III)}:\text{Fe(II)}$ which were 7.2 and 3.6 respectively. The redox potentials from Veg B during
 345 both C1 and C2 seasons (9.6 mV and -89.0 mV respectively) were consistently lower than Veg
 346 A during C1 and C2 seasons (46.5 mV and 12.0 mV respectively), which is consistent with the
 347 more reducing conditions encouraging CH_4 production in Veg B habitat. Further, as iron
 348 reduction yields more free energy than SO_4^{2-} reduction (which yields more free energy than
 349 methanogenesis) (Burdige, 2012), then Fe reduction at Veg A may outcompete CH_4 production
 350 ahead of SO_4^{2-} reduction at Veg B, which may help explain some of the differences in CH_4
 351 production between the two sites.

352 Regression analysis and Spearman rho coefficients summarise the spatial trends
 353 occurring between the average sediment parameters versus seasonal CH_4 fluxes from the
 354 different sites (Fig. 7). Positive significant trends occurred for Fe(II) , AVS and the $\text{Cl}:\text{SO}_4^{2-}$
 355 ratios with CH_4 flux rates ($r_s=0.88$, $p<0.01$) supporting our hypothesis that reducing conditions
 356 and a smaller pool of sediment Fe(III) and SO_4^{2-} facilitate higher CH_4 production rates.



357 Alternatively, the negative trends observed between soil redox potentials, SO_4^{2-} , Fe(III) and
358 CH_4 fluxes affirm that the abundance of thermodynamically favourable terminal electron
359 acceptors plays a role in attenuating CH_4 production at each site.

360

361 4.2 Plant-mediated CH_4 fluxes from the seasonal wetland

362 Plant-mediated CH_4 fluxes were highest during C1 under post-dry/flooded conditions
363 with 20–30 cm of standing waters in the seasonal wetland (Table 1). While waterlogged
364 conditions are an obvious driver of higher CH_4 production rates from saturated sediments in
365 addition to the geochemical differences (previously discussed), other drivers which may
366 explain these trends include differences in diurnal variability in temperature, PAR and plant
367 physiology, which may influence CH_4 gas transport pathways.

368 In vegetated seasonal wetlands, plant-mediated gas transport is recognised as a
369 dominant pathway for CH_4 emission to the atmosphere and accounts for up to 90% of total
370 wetland fluxes (Sorrell & Boon, 1994; Whiting & Chanton, 1992). For plant survival in near-
371 permanent inundation environments, oxygen transport occurs via the aerenchyma downwards
372 to the rhizome. This increases the plant performance by mitigating (i.e. oxidising) the
373 accumulation of phytotoxins such as sulphides and reducing metal ions around the roots
374 (Armstrong & Armstrong, 1990; Armstrong et al., 2006; Penhale & Wetzels, 1983). As oxygen
375 transfer to the rhizosphere occurs, an exchange of sedimentary CH_4 can be efficiently
376 transported from the rhizosphere to atmosphere, bypassing sedimentary oxidative processes
377 along the way (Fig. 8). This process in plants can be either convective (i.e. pressurised) or via
378 passive diffusive gas flow, both of which are adaptive traits of many wetland species
379 (Armstrong & Armstrong, 1991; Konnerup et al., 2011).

380 During both seasons the highest CH_4 fluxes from seasonal wetland vegetation were
381 emitted from *Phragmites australis* (Veg B) and always occurred during daylight (Table 1, Fig.
382 8). In *Phragmites australis* (Veg B), the presence of pressurised lacunar leaf culms drive a
383 mass flow of oxygen to the rhizome and back to the atmosphere via older (non-pressurised)
384 efflux culms (Henneberg et al., 2012; Sorrell & Boon, 1994). This process has been widely
385 studied in wetlands featuring this species, as it is one of the most productive and wide spread
386 flowering wetland species (Brix et al., 2001; Chanton et al., 2002; Clevering & Lissner, 1999;
387 Tucker, 1990). Kim et al. (1998) showed CH_4 emissions from *Phragmites australis* peaked
388 around midday and that daytime emissions were about 3-fold higher than night time emissions,



positively correlating with temperature and PAR. These were similar to our findings with highest CH₄ fluxes of each seasonal time series occurring near midday (10:50 am during C1; 4.88 mmol m⁻² d⁻² and 12:15 pm during C2; 2.06 mmol m⁻² d⁻²) (Fig. 3). We also found a positive significant relationship between CH₄ flux and both temperature and PAR during C2 ($r^2=0.35$, $p<0.001$ and $r^2=0.18$, $p<0.01$ respectively) (Fig. 6). The often high diurnal variability in CH₄ fluxes from *Phragmites australis* occurs as convective gas transport increases rhizospheric oxygen and CH₄ exchange via living culms during the daytime, whereas molecular diffusion during the night time facilitates a more passive and lower CH₄ flux pathway through dead culms (Armstrong & Armstrong, 1991; Chanton et al., 2002).

One possible reason CH₄ fluxes were lower from Veg A than Veg B despite their close geographical location, may be due to the passive gas diffusion mechanism utilised by *Juncus sp.* (Henneberg et al., 2012). Unlike the pressurised conductive gas flow mechanisms of Veg B, many wetland rush species (such as Veg A) employ passive diffusive gas flow to survive within water logging environments (Brix et al., 1992; Konnerup et al., 2011). Despite diffusion being a less efficient gas transport mechanism (Konnerup et al., 2011), plant-mediated CH₄ diffusion is recognised as the dominant pathway for CH₄ emissions from many seasonal wetland species. During C1 and C2, day time fluxes (diffusive) from Veg A were only 19% and 33% higher than night time fluxes (diffusive). In comparison, at Veg B these day:night ratios were almost triple this (67% and 94% higher) during the same periods. This may potentially be due to the more efficient daytime conductive gas transfer pathway of CH₄ through Veg B (*Phragmites australis*) compared to the more passive diffusive CH₄ gas transfer pathway of Veg A (*Juncus kraussii*). This suggests that non-pressurized pathways may result in lower net rhizosphere-atmosphere gas exchange of CH₄ from seasonal wetland vegetation.

The *Juncus kraussii* below *Casuarina sp.* trees (Veg C) emitted nominal fluxes of CH₄ during both time series campaigns and was a net sink for CH₄ during C2 (Table 1, Fig. 8). Although wetland trees have recently been shown to contribute significantly to CH₄ fluxes from flooded environments (Pangala et al., 2017), we could not quantify or constrain the role of trees as a conduit of methane to the atmosphere at this site. Regardless, there were clearly lower CH₄ fluxes through the Veg C (*Juncus kraussii*) compared to the Veg A (*Juncus kraussii*). As the species at ground level were identical, these differences are not related to vegetative gas transport mechanisms, nor organic carbon content (Table 1). Shading by the overhanging trees may inhibit the daytime diffusive CH₄ gas transport through Veg C assumable to lower rates of photosynthesis, however PAR was only lower during C2 (Fig. 7)



and so does not appear to explain the CH₄ flux differences observed during C1. The differences are therefore likely explained by the higher positive redox potentials (Table 1) and more abundant thermodynamically favourable terminal electron acceptors (i.e. Fe(III) and SO₄²⁻) (Fig. 5) all of which can inhibit methane production within the sediments (Burdige, 2012).

4.3 Permanent Wetland CH₄ fluxes

Diffusive CH₄ fluxes from the permanent wetland varied considerably between seasons; however, ebullition fluxes were similar (Table 1, Fig. 8). The highest seasonal CH₄ fluxes for both ebullition and diffusion (2.1 mmol m⁻² d⁻¹ and 10.5 mmol m⁻² d⁻¹ respectively) occurred during C2 despite cooler conditions (Fig. 2, Fig. 8). This however was the opposite trend to the seasonal wetland CH₄ fluxes (Table 1, Fig. 8). One reason may be due to the antecedent hydrological conditions before C1 (Fig. 2). Jeffrey et al (submitted) reported that a water level drawdown of the permanent wetland after a hot and drying summer period exposed some of the permanent wetland sediments to oxidative conditions. This may have oxidised a portion of the labile sedimentary carbon pool prior to C1 sampling of the permanent wetland, therefore reducing the total CH₄ pool observed during C1 sampling. A lag time (ranging from weeks to months) for recovery of the CH₄ pool post-drought has been observed in other systems (Boon et al., 1997) and also during lab-based experiments (Freeman et al., 1992; Knorr et al., 2008). This may explain the higher CH₄ fluxes during C2 when the system had had sufficient time to recover, despite lower water column temperatures that would normally reduce microbial metabolism rates. This hypothesis is also supported by the shift of net positive redox potential of the permanent wetland during C1 (71.7 ± 65 mV), to a strong negative redox potential during C2 (-216 ± 42 mV) indicating that there was a time lag for reducing conditions to recover within the permanent wetland for C2. This highlights the critical role of antecedent hydrological conditions and how dynamic weather oscillations of drought and floods (a common occurrence of many Australian wetland systems), strongly influence the redox potentials, soil geochemistry and ultimately CH₄ fluxes.

4.4 Implications and conclusions

Permanent wetland emissions account for the majority of the global wetland CH₄ budget however both subtropical systems and southern hemisphere systems are poorly



represented (Bartlett & Harriss, 1993; Bastviken et al., 2011) (Fig. 9). Further, the fluxes from seasonal wetlands are poorly constrained (Pfeifer-Meister et al., 2018) due to their intermittent nature and variability of intra-seasonal areal extent, which may compound why natural wetlands have the largest uncertainty of the global methane budget (Kirschke et al., 2013; Saunio et al., 2016). Although the temporal resolution of our study cannot be up scaled to realistic annual estimates, our high resolution sampling strategy provided insights to daily CH_4 flux rates revealing distinct differences between different vegetation types across the terrestrial aquatic wetland boundary. Our seasonal emissions rates were at the low end of the scale of measurements made in southern hemisphere subtropical systems but within range of northern hemisphere subtropical systems of similar latitudes (Fig. 9).

Although remediating degraded wetlands through re-flooding is a common technique to improve biodiversity, increase C sequestration and improve downstream water quality issues (Johnston et al., 2014; Johnston et al., 2004), our results propose a nuanced dilemma for land use managers, as wetland restoration can have net positive radiative forcing effects on the Earth's climate due to high rates of CH_4 production (Mitsch et al., 2013). This has also been shown to be particularly high during early remediation periods (Hemes et al., 2018). Our results suggest that seasonal wetlands emit less CH_4 on an areal basis than permanent wetlands, yet carbon accumulation in these soils may be lower (Brown et al. (in publication)). Longer-term studies over annual cycles encompassing seasonal drivers and CH_4 fluxes would further test this hypothesis of the different drivers between seasonal and permanent wetland systems.

Our results also suggest that selective hydrological restoration of wetlands featuring sediments with abundant thermodynamically favourable terminal electron acceptors (i.e. Fe(III) or SO_4^{2-}) may be a (partial) biogeochemical solution (also suggested by Hemes et al. (2018)) to both remediate degraded sites whilst simultaneously mitigating some CH_4 emissions. When Fe(III) and SO_4^{2-} are abundant in anaerobic environments they provide preferential terminal electron acceptors for microbial metabolism and thus limit methanogenesis via competitive exclusion (Acht nich et al., 1995). However, high rates of sulphate reduction coupled with Fe reduction can also lead to the accumulation of metal sulphide minerals e.g. pyrite and AVS (Johnston et al., 2014). Under permanently saturated and low oxygen conditions, metal sulphides will steadily accumulate and remain relatively benign. However, if the saturated state of remediated sites cannot be maintained, AVS may react with oxygen resulting in undesirable production of acidity and low pH conditions. Therefore the remediation of wetlands for carbon storage should involve careful site selection



486 to both limit CH₄ production and to avoid redox related geochemical by-products with
487 detrimental environmental effects.

488 This study has highlighted how sediment geochemistry is intimately related to CH₄
489 production and consumption. While high sulphate and Fe(III) favour lower CH₄ production,
490 sites featuring more reducing conditions and depleted sulphate and Fe(III) favour the highest
491 CH₄ fluxes. Results reveal distinct differences between the areal CH₄ fluxes of four different
492 eco-types located within a remediated subtropical Australian wetland and indicate high
493 seasonal variability. By combining novel and well established techniques we delineated several
494 CH₄ pathways of both seasonal and permanent wetland sources (ebullition, diffusion and plant-
495 mediated pathways) and linked these to seasonal drivers. This provided evidence that soil
496 geochemistry is an important factor to consider for wetland remediation in the context of CH₄
497 production and mitigation strategies. The CH₄ emissions results were comparable to other
498 wetlands of similar latitudes and contribute important data for both the understudied southern
499 hemisphere wetlands and seasonal subtropical wetland ecotypes.

500

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508 (www.ian.umces.edu/symbols).

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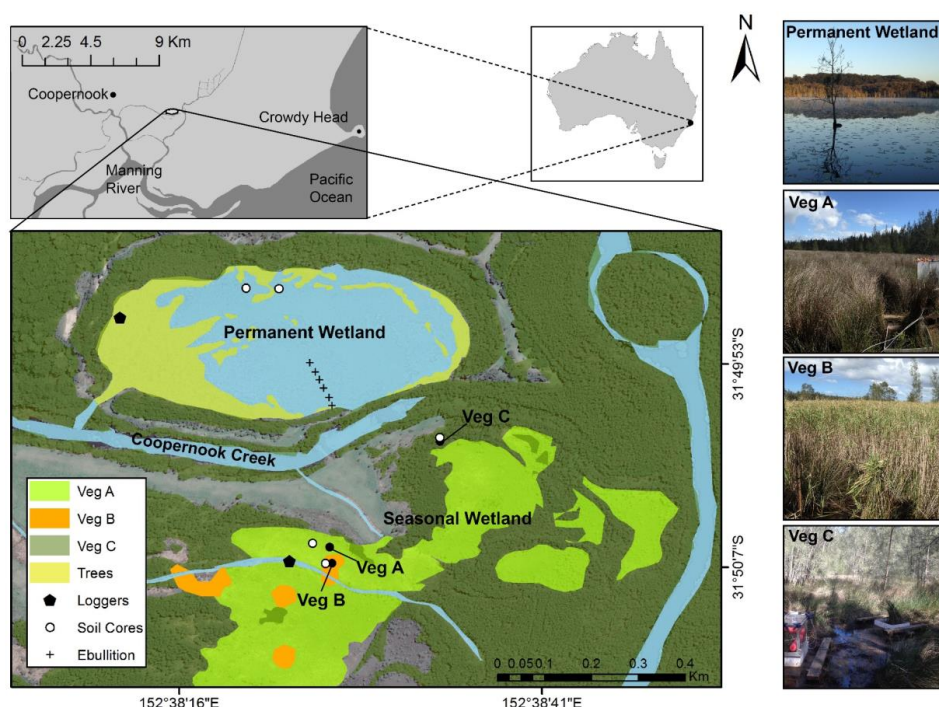
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Figures



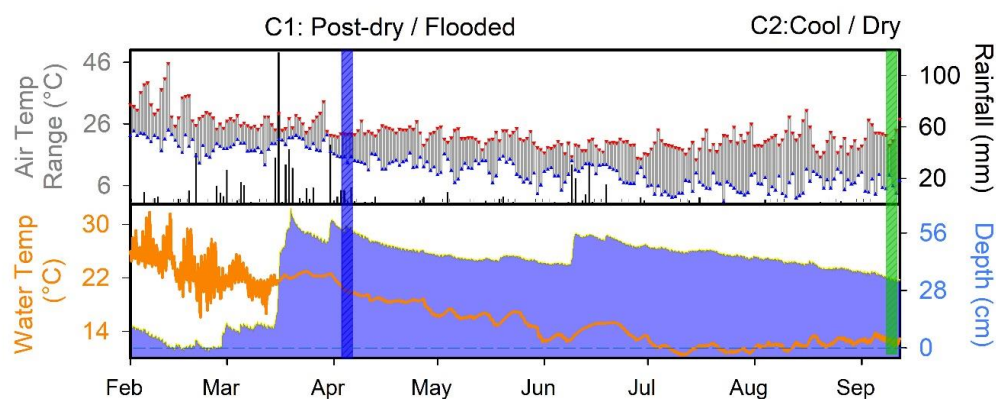
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514 **Fig. 1** The seasonal wetland study sites consisting of Veg A (*Juncus kraussii*), Veg B
 515 (*Phragmites australis*), Veg C (*Juncus kraussii* below *Casuarina* sp.) and the permanent
 516 wetland indicating sediment coring sites, ebullition replicate transect, 24 h vegetation time
 517 series sites and imagery of vegetation ecotypes.

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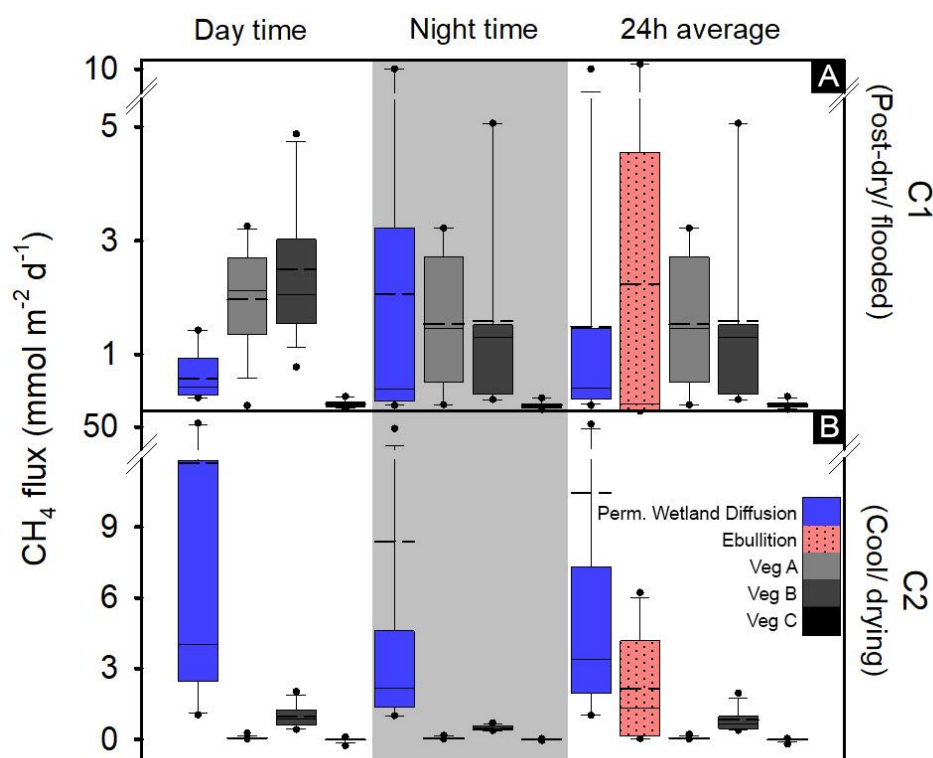
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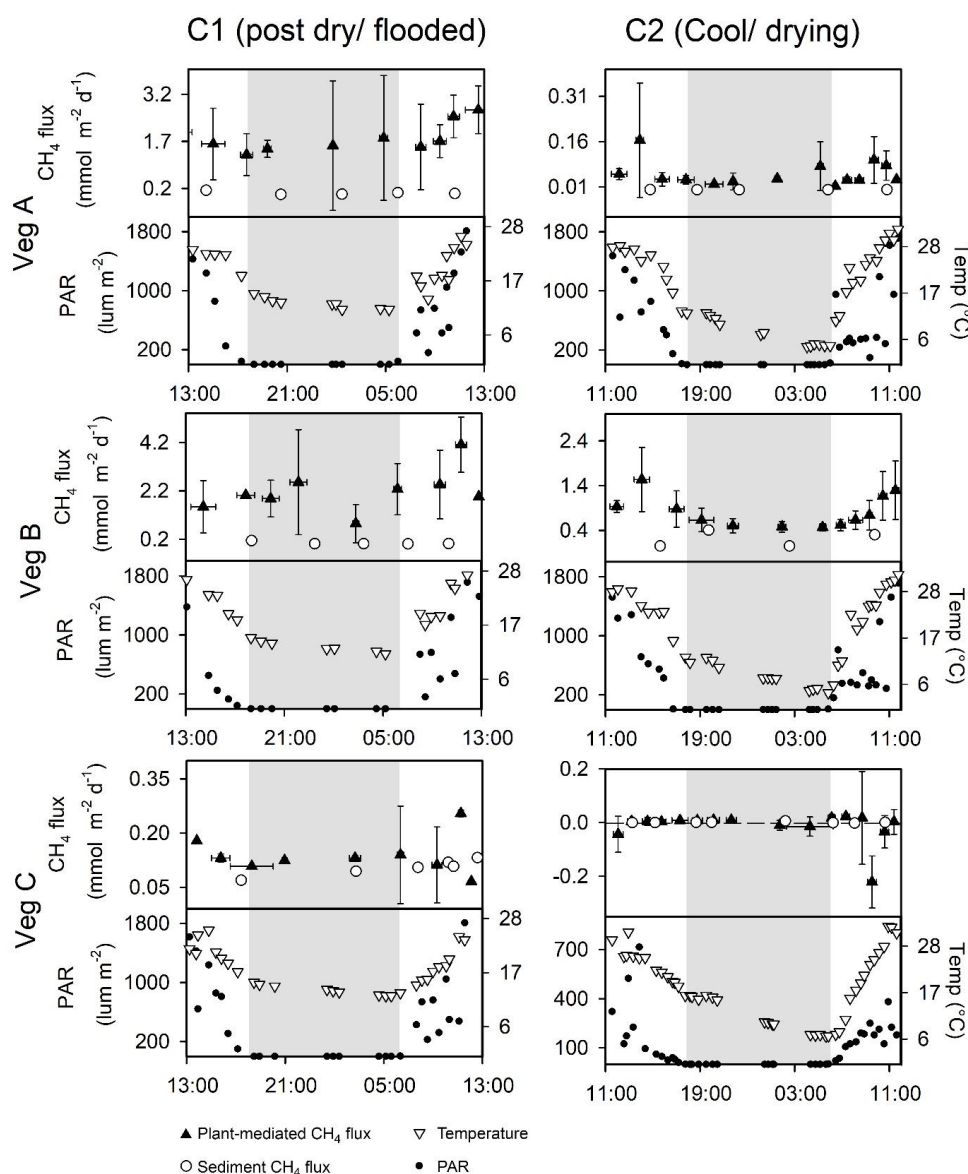
521 **Figure 2.** Hydrograph for the seven months of 2017 indicating daily rainfall, maximum/
 522 minimum air temperature, water temperature and antecedent hydrology. Vertical coloured
 523 bands represent the two fieldwork campaigns.

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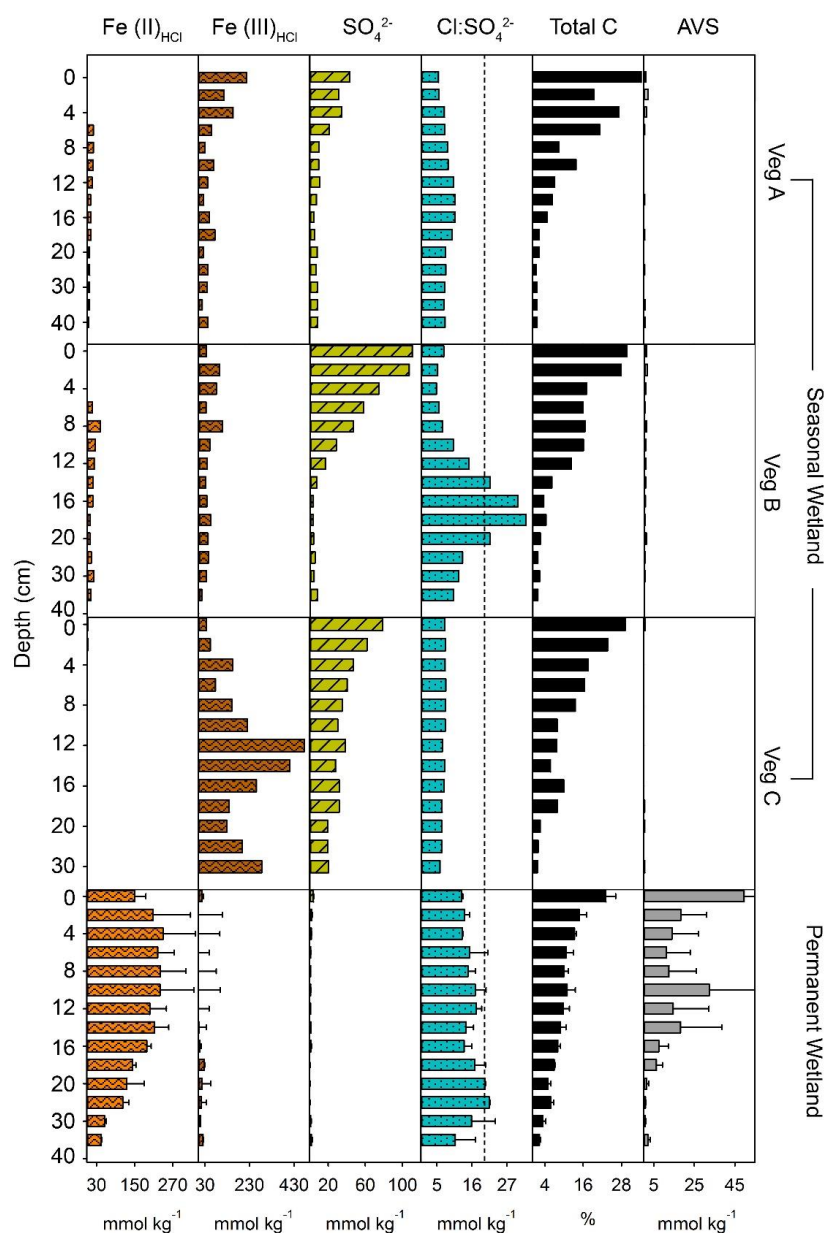
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526 **Figure 3.** Simultaneous 24 h time series of vegetative CH_4 fluxes from the seasonal wetland
 527 ecotypes at Cattai Wetland during C1: post-dry/flooded (Apr 2017) and C2: cool/drying
 528 conditions (Sep 2017). The vertical error bars of the plant-mediated CH_4 flux ($\text{mmol m}^{-2} \text{d}^{-1}$)
 529 represent standard deviation of the triplicate time series measurements taken from each site and
 530 horizontal bars represent the total aggregated time period represented by replicate chambers.
 531 The grey shading indicates night-time. Note: Different y-axis scales for CH_4 to highlight
 532 diurnal trends.



533

534 **Figure 4.** Seasonal fluxes of CH₄ from diurnal sampling and ebullition from the permanent
 535 wetland and adjacent 24 h time series of the seasonal wetland vegetation types A, B and C.
 536 Note: Dashed line represents the average, solid line represents the median and dots represent
 537 5th and 95th percentiles.

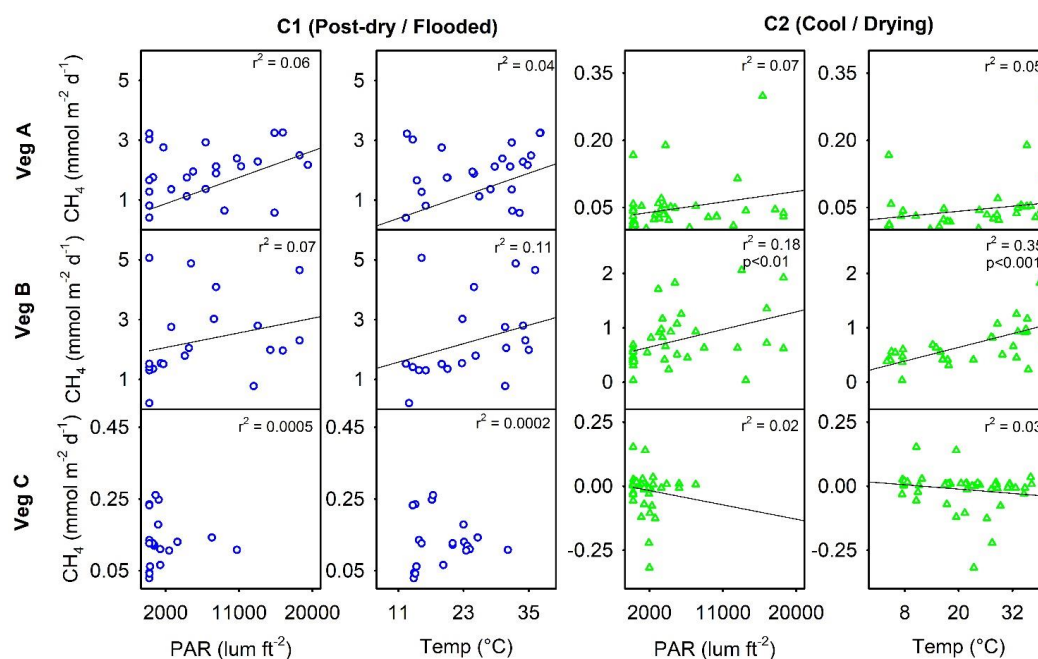


538

539 **Figure 5.** Soil profiles of the permanent and seasonal wetland sites indicating Fe(II)_{HCl},
 540 Fe(III)_{HCl}, SO₄²⁻, Cl:SO₄²⁻ (a proxy for depletion of marine-derived sulphate, where >20 is
 541 broadly indicative of SO₄²⁻ reduction and <8 CASS pyrite oxidation (Mulvey, 1993)), total C
 542 and acid volatile sulphur (AVS). Note: The permanent wetland profiles are averages from two
 543 adjacent sites with error bars representing the standard deviation.



544



545

546 **Figure 6.** Correlations of CH₄ with temperature (°C) and photo-synthetically active radiation
 547 (PAR) (lum ft⁻²) for the three seasonal wetland vegetation sites of Cattai Wetland during two
 548 seasonal campaigns.

549

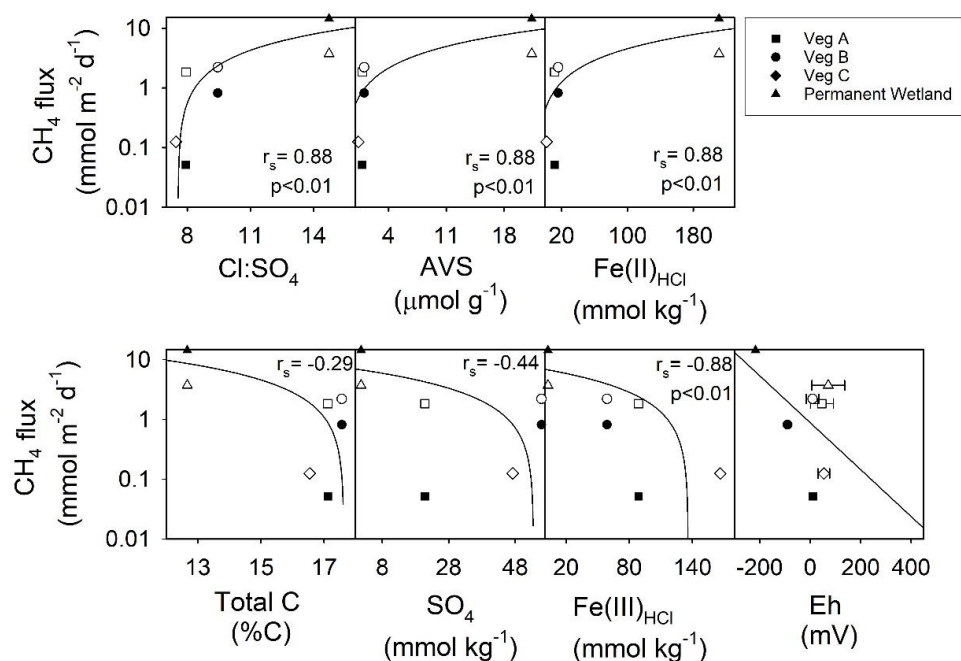
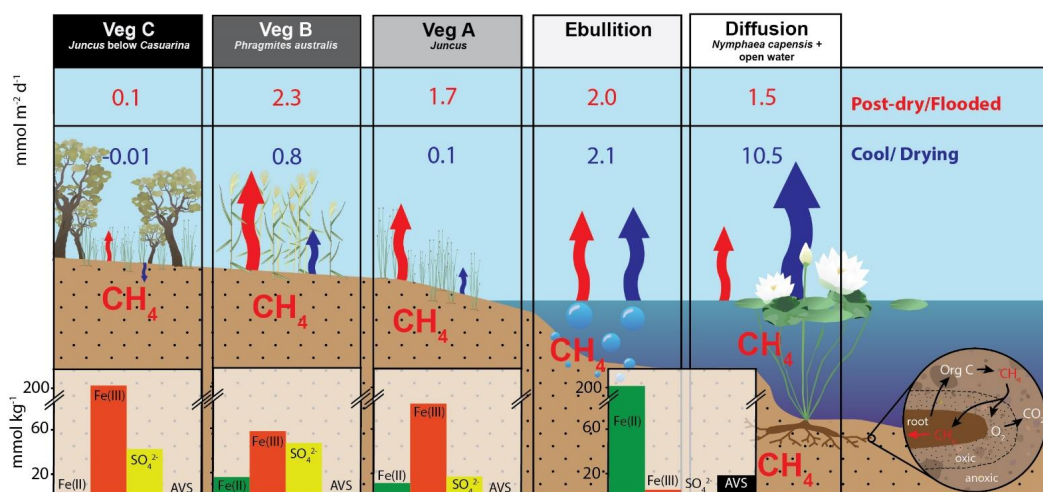


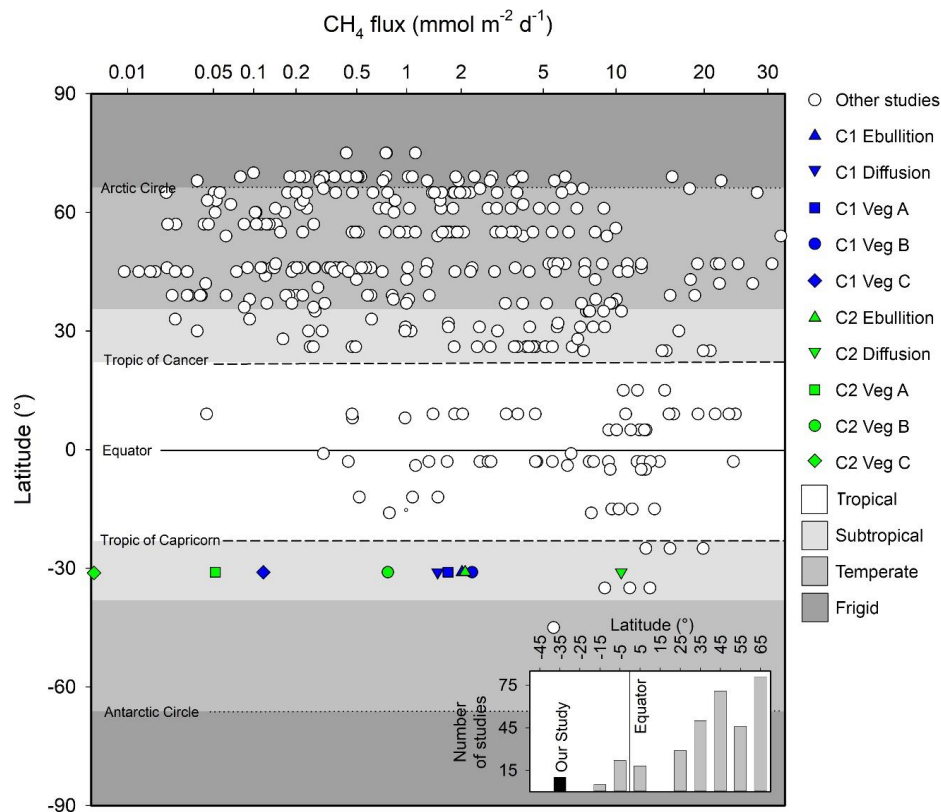
Figure 7. Regression analysis of average daily CH_4 fluxes ($\text{mmol m}^{-2} \text{d}^{-1}$) vs subsoil parameters of 0-20 cm core depth (i.e. CH_4 'active' zone). Note: Log scale y-axis of CH_4 fluxes from the four wetland ecotypes over two seasons. Note: The r_s values calculated using Spearman rho are for C1 (black shapes) and C2 (white shapes).



557

558 **Figure 8.** Conceptual model summarising the terrestrial and aquatic CH₄ fluxes (mmol m⁻² d⁻¹) and sediment core profile parameters (mmol kg⁻¹) of the permanent and seasonal wetlands
 559 during C1 (post-dry/flooded conditions) and C2 (cool/drying conditions) of Cattai Wetland.
 560 Conceptual diagram expanded from Jeffrey et al. (in publication) and rhizome process insert
 561 adapted from (Conrad, 1993).
 562

563



564

565 **Figure 9.** Summary of major CH₄ wetland reviews by Bartlett and Harriss (1993), Bastviken
566 et al. (2011) and modelled fluxes by Cao et al. (1998) adapted from Jeffrey et al., (in
567 publication) highlighting latitudinal trends and bias from a variety of wetland systems. Inset
568 figure highlights number of studies in these reviews by latitudinal increments of 10° poleward
569 of the equator. Note: x axis scaled to highlight subtle differences between studies.

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List of Tables

573 **Table 1.** Summary of plant-mediated CH₄ fluxes from the seasonal wetland time series and
 574 diurnal CH₄ fluxes and ebullition from the permanent wetland during C1 (post-dry/ flooded)
 575 and C2 (cool/ drying). The corresponding sediment core data are average concentrations from
 576 0 to 20 cm below ground level.

CH ₄ flux (mmol m ⁻² d ⁻¹)	Permanent Wetland		Seasonal Wetland Sites		
	Ebullition	Diffusion	Veg A	Veg B	Veg C
C1 - Sediment flux			0.06	0.04	0.10
C1 - Day time		0.57	1.79	2.64	0.13
C1 - Night time		2.07	1.50	1.59	0.10
C1 - Daily average	2.02	1.49	1.70	2.27	0.12
C2 - Sediment flux			0.0004	0.20	0.0003
C2 - Day time		11.72	0.06	0.94	0.13
C2 - Night time		8.39	0.04	0.48	0.10
C2 - Daily average	2.10	10.46	0.05	0.77	-0.01
Sediment core average (0-20cm)					
Fe _{HCl} (II) (mmol kg ⁻¹)		202.3	11.6	15.4	1.5
Fe _{HCl} (III) (mmol kg ⁻¹)		5.6	83.3	56.1	204.0
SO ₄ ²⁻ (mmol kg ⁻¹)		1.5	17.6	45.4	43.3
Cl:SO ₄ ²⁻		14.8	8.4	13.9	7.4
AVS (μmol g ⁻¹)		18.5	0.7	0.9	0.3
TOC (% C)		11.6	14.3	14.8	14.6
C1 - Redox Eh (mV)		71.7	46.5	9.6	54.4
C2 - Redox Eh (mV)		-216	12	-89	424

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References

- á Norði, K., and Thamdrup, B.: Nitrate-dependent anaerobic methane oxidation in a
freshwater sediment, *Geochimica et Cosmochimica Acta*, 132, 141-150, 2014.
- Achtnich, C., Bak, F., and Conrad, R.: Competition for electron donors among nitrate
reducers, ferric iron reducers, sulfate reducers, and methanogens in anoxic paddy soil,
Biology and Fertility of Soils, 19, 65-72, 1995.
- ANCA: Issues paper for the Wise use Workshop, 4–6 December. Wetlands and migratory
Wildlife Unit of the Australian Nature Conservation Agency, Canberra. 42 pp, 1995.
- Armentano, T., and Menges, E.: Patterns of change in the carbon balance of organic soil-
wetlands of the temperate zone, *The Journal of Ecology*, 755-774, 1986.
- Armstrong, J., and Armstrong, W.: Light-enhanced convective throughflow increases
oxygenation in rhizomes and rhizosphere of *Phragmites australis* (Cav.) Trin. ex Steud,
New Phytologist, 114, 121-128, 1990.
- Armstrong, J., and Armstrong, W.: A convective through-flow of gases in *Phragmites*
australis (Cav.) Trin. ex Steud, *Aquatic Botany*, 39, 75-88, 1991.
- Armstrong, J., Jones, R., and Armstrong, W.: Rhizome phyllosphere oxygenation in
Phragmites and other species in relation to redox potential, convective gas flow,
submergence and aeration pathways, *New Phytologist*, 172, 719-731, 2006.
- Association, A. P. H., Association, A. W. W., Federation, W. P. C., and Federation, W. E.:
Standard methods for the examination of water and wastewater, American Public
Health Association., 1915.
- Bartlett, K. B., and Harriss, R. C.: Review and assessment of methane emissions from
wetlands, *Chemosphere*, 26, 261-320, 1993.
- Bastviken, D., Tranvik, L. J., Downing, J. A., Crill, P. M., and Enrich-Prast, A.: Freshwater
methane emissions offset the continental carbon sink, *Science*, 331, 50, 2011.
- Bianchi, T. S.: *Biogeochemistry of estuaries*, Oxford University Press New York, 2007.
- Boman, A., Åström, M., and Fröjdö, S.: Sulfur dynamics in boreal acid sulfate soils rich in
metastable iron sulfide—the role of artificial drainage, *Chemical Geology*, 255, 68-77,
2008.
- Boon, P. I., Mitchell, A., and Lee, K.: Effects of wetting and drying on methane emissions
from ephemeral floodplain wetlands in south-eastern Australia, *Hydrobiologia*, 357, 73-
87, 1997.



- 612 Bridgham, S. D., Moore, T. R., Richardson, C. J., and Roulet, N. T.: Errors in greenhouse
613 forcing and soil carbon sequestration estimates in freshwater wetlands: a comment on
614 Mitsch et al. (2013), *Landscape Ecology*, 29, 1481-1485, 2014.
- 615 Brix, H., Sorrell, B. K., and Orr, P. T.: Internal pressurization and convective gas flow in
616 some emergent freshwater macrophytes, *Limnology and Oceanography*, 37, 1420-1433,
617 1992.
- 618 Brix, H., Sorrell, B. K., and Lorenzen, B.: Are Phragmites-dominated wetlands a net source
619 or net sink of greenhouse gases?, *Aquatic Botany*, 69, 313-324, 2001.
- 620 Burdige, D.: Estuarine and coastal sediments–coupled biogeochemical cycling, *Treatise on*
621 *Estuarine and Coastal Science*, 5, 279-316, 2012.
- 622 Burton, E., Bush, R. T., Johnston, S. G., Sullivan, L. A., and Keene, A. F.: Sulfur
623 biogeochemical cycling and novel Fe–S mineralization pathways in a tidally re-flooded
624 wetland, *Geochimica et Cosmochimica Acta*, 75, 3434-3451, 2011.
- 625 Burton, E. D., Bush, R. T., and Sullivan, L. A.: Elemental sulfur in drain sediments
626 associated with acid sulfate soils, *Applied Geochemistry*, 21, 1240-1247, 2006.
- 627 Burton, E. D., Bush, R. T., Sullivan, L. A., and Mitchell, D. R. G.: Reductive transformation
628 of iron and sulfur in schwertmannite-rich accumulations associated with acidified
629 coastal lowlands, *Geochimica et Cosmochimica Acta*, 71, 4456-4473,
630 10.1016/j.gca.2007.07.007, 2007.
- 631 Cao, M., Gregson, K., and Marshall, S.: Global methane emission from wetlands and its
632 sensitivity to climate change, *Atmospheric environment*, 32, 3293-3299, 1998.
- 633 Chanton, J. P., Arkebauer, T. J., Harden, H. S., and Verma, S. B.: Diel variation in lacunal
634 CH₄ and CO₂ concentration and $\delta^{13}\text{C}$ in *Phragmites australis*, *Biogeochemistry*, 59,
635 287-301, 2002.
- 636 Claff, S. R., Sullivan, L. A., Burton, E. D., and Bush, R. T.: A sequential extraction
637 procedure for acid sulfate soils: partitioning of iron, *Geoderma*, 155, 224-230, 2010.
- 638 Clevering, O. A., and Lissner, J.: Taxonomy, chromosome numbers, clonal diversity and
639 population dynamics of *Phragmites australis*, *Aquatic Botany*, 64, 185-208, 1999.
- 640 Conrad, R.: Mechanisms controlling methane emission from wetland rice fields, in:
641 *Biogeochemistry of Global Change*, Springer, 317-335, 1993.
- 642 Costanza, R., de Groot, R., Sutton, P., van der Ploeg, S., Anderson, S. J., Kubiszewski, I.,
643 Farber, S., and Turner, R. K.: Changes in the global value of ecosystem services,
644 *Global Environmental Change*, 26, 152-158, 2014.



- 645 Deverel, S. J., Ingram, T., and Leighton, D.: Present-day oxidative subsidence of organic
646 soils and mitigation in the Sacramento-San Joaquin Delta, California, USA,
647 Hydrogeology journal, 24, 569-586, 2016.
- 648 Finlayson, C. M., and Rea, N.: Reasons for the loss and degradation of Australian wetlands,
649 Wetlands Ecology and Management, 7, 1-11, 1999.
- 650 Freeman, C., Lock, M., and Reynolds, B.: Fluxes of CO₂, CH₄ and N₂O from a Welsh
651 peatland following simulation of water table draw-down: Potential feedback to climatic
652 change, Biogeochemistry, 19, 51-60, 1992.
- 653 Hemes, K. S., Chamberlain, S. D., Eichelmann, E., Knox, S. H., and Baldocchi, D. D.: A
654 biogeochemical compromise: The high methane cost of sequestering carbon in restored
655 wetlands, Geophysical Research Letters, 45, 6081-6091, 2018.
- 656 Henneberg, A., Sorrell, B. K., and Brix, H.: Internal methane transport through *Juncus*
657 effusus: experimental manipulation of morphological barriers to test above-and below-
658 ground diffusion limitation, New Phytologist, 196, 799-806, 2012.
- 659 Holmkvist, L., Ferdelman, T. G., and Jørgensen, B. B.: A cryptic sulfur cycle driven by iron
660 in the methane zone of marine sediment (Aarhus Bay, Denmark), Geochimica et
661 Cosmochimica Acta, 75, 3581-3599, 2011.
- 662 Jeffrey, L. C., Maher, D. T., Santos, I. R., McMahon, A., and Tait, D. R.: Groundwater, Acid
663 and Carbon Dioxide Dynamics Along a Coastal Wetland, Lake and Estuary Continuum,
664 Estuaries and Coasts, 39, 1325-1344, 2016.
- 665 Johnston, S. G., Slavich, P. G., Sullivan, L. A., and Hirst, P.: Artificial drainage of
666 floodwaters from sulfidic backswamps: effects on deoxygenation in an Australian
667 estuary, Marine and Freshwater Research, 54, 781-795, 2003.
- 668 Johnston, S. G., Slavich, P. G., and Hirst, P.: The effects of a weir on reducing acid flux from
669 a drained coastal acid sulphate soil backswamp, Agricultural Water Management, 69,
670 43-67, 2004.
- 671 Johnston, S. G., Keene, A. F., Burton, E. D., Bush, R. T., and Sullivan, L. A.: Quantifying
672 alkalinity generating processes in a tidally remediating acidic wetland, Chemical
673 Geology, 304, 106-116, 2012.
- 674 Johnston, S. G., Burton, E. D., Aaso, T., and Tuckerman, G.: Sulfur, iron and carbon cycling
675 following hydrological restoration of acidic freshwater wetlands, Chemical Geology,
676 371, 9-26, 2014.



- 677 Karimian, N., Johnston, S. G., and Burton, E. D.: Acidity generation accompanying iron and
678 sulfur transformations during drought simulation of freshwater re-flooded acid sulfate
679 soils, *Geoderma*, 285, 117-131, 2017.
- 680 Karimian, N., Johnston, S. G., and Burton, E. D.: Iron and sulfur cycling in acid sulfate soil
681 wetlands under dynamic redox conditions: A review, *Chemosphere*, 197, 803-816,
682 10.1016/j.chemosphere.2018.01.096, 2018.
- 683 Kim, J., Verma, S., Billesbach, D., and Clement, R.: Diel variation in methane emission from
684 a midlatitude prairie wetland: significance of convective throughflow in *Phragmites*
685 *australis*, *Journal of Geophysical Research: Atmospheres*, 103, 28029-28039, 1998.
- 686 Kirschke, S., Bousquet, P., Ciais, P., Saunio, M., Canadell, J. G., Dlugokencky, E. J.,
687 Bergamaschi, P., Bergmann, D., Blake, D. R., Bruhwiler, L., Cameron-Smith, P.,
688 Castaldi, S., Chevallier, F., Feng, L., Fraser, A., Heimann, M., Hodson, E. L.,
689 Houweling, S., Josse, B., Fraser, P. J., Krummel, P. B., Lamarque, J.-F., Langenfelds,
690 R. L., Le Quéré, C., Naik, V., O'Doherty, S., Palmer, P. I., Pison, I., Plummer, D.,
691 Poulter, B., Prinn, R. G., Rigby, M., Ringeval, B., Santini, M., Schmidt, M., Shindell,
692 D. T., Simpson, I. J., Spahni, R., Steele, L. P., Strode, S. A., Sudo, K., Szopa, S., van
693 der Werf, G. R., Voulgarakis, A., van Weele, M., Weiss, R. F., Williams, J. E., and
694 Zeng, G.: Three decades of global methane sources and sinks, *Nature Geoscience*, 6,
695 813-823, 10.1038/ngeo1955, 2013.
- 696 Knorr, K.-H., Glaser, B., and Blodau, C.: Fluxes and ^{13}C isotopic composition of dissolved
697 carbon and pathways of methanogenesis in a fen soil exposed to experimental drought,
698 *Biogeosciences Discussions*, 5, 1319-1360, 2008.
- 699 Konnerup, D., Sorrell, B. K., and Brix, H.: Do tropical wetland plants possess convective gas
700 flow mechanisms?, *New Phytol*, 190, 379-386, 10.1111/j.1469-8137.2010.03585.x,
701 2011.
- 702 Lal, R.: Carbon sequestration, *Philosophical Transactions of the Royal Society of London B:*
703 *Biological Sciences*, 363, 815-830, 2008.
- 704 Melton, J., Wania, R., Hodson, E., Poulter, B., Ringeval, B., Spahni, R., Bohn, T., Avis, C.,
705 Beerling, D., and Chen, G.: Present state of global wetland extent and wetland methane
706 modelling: conclusions from a model intercomparison project (WETCHIMP),
707 *Biogeosciences*, 10, 753-788, 2013.
- 708 Mitsch, W. J., Bernal, B., Nahlik, A. M., Mander, Ü., Zhang, L., Anderson, C. J., Jørgensen,
709 S. E., and Brix, H.: Wetlands, carbon, and climate change, *Landscape Ecology*, 28,
710 583-597, 2013.



- 711 Mulvey, P.: Pollution, prevention and management of sulphidic clays and sands. Proceedings
712 National Conference on Acid Sulphate Soils. (Ed. R Bush) pp, 116-129, 1993.
- 713 Nedwell, D. B., and Watson, A.: CH₄ production, oxidation and emission in a UK
714 ombrotrophic peat bog: influence of SO₄²⁻ from acid rain, Soil Biology and
715 Biochemistry, 27, 893-903, 1995.
- 716 Neubauer, S. C., and Megonigal, J. P.: Moving beyond global warming potentials to quantify
717 the climatic role of ecosystems, Ecosystems, 18, 1000-1013, 2015.
- 718 Page, K., and Dalal, R.: Contribution of natural and drained wetland systems to carbon
719 stocks, CO₂, N₂O, and CH₄ fluxes: an Australian perspective, Soil Research, 49, 377-
720 388, 2011.
- 721 Pangala, S. R., Enrich-Prast, A., Basso, L. S., Peixoto, R. B., Bastviken, D., Hornibrook, E.
722 R., Gatti, L. V., Marotta, H., Calazans, L. S. B., and Sakuragui, C. M.: Large emissions
723 from floodplain trees close the Amazon methane budget, Nature, 552, 230, 2017.
- 724 Penhale, P. A., and Wetzel, R. G.: Structural and functional adaptations of eelgrass (*Zostera*
725 *marina* L.) to the anaerobic sediment environment, Canadian Journal of Botany, 61,
726 1421-1428, 1983.
- 727 Pereyra, A. S., and Mitsch, W. J.: Methane emissions from freshwater cypress (*Taxodium*
728 *distichum*) swamp soils with natural and impacted hydroperiods in Southwest Florida,
729 Ecological Engineering, 114, 46-56, 2018.
- 730 Pfeifer-Meister, L., Gayton, L. G., Roy, B. A., Johnson, B. R., and Bridgham, S. D.:
731 Greenhouse gas emissions limited by low nitrogen and carbon availability in natural,
732 restored, and agricultural Oregon seasonal wetlands, PeerJ, 6, 10.7717/peerj.5465,
733 2018.
- 734 Poffenbarger, H. J., Needelman, B. A., and Megonigal, J. P.: Salinity influence on methane
735 emissions from tidal marshes, Wetlands, 31, 831-842, 2011.
- 736 Postma, D., and Jakobsen, R.: Redox zonation: equilibrium constraints on the Fe (III)/SO₄²⁻
737 reduction interface, Geochimica et Cosmochimica Acta, 60, 3169-3175, 1996.
- 738 Rayment, G., and Higginson, F. R.: Australian laboratory handbook of soil and water
739 chemical methods, Inkata Press Pty Ltd, 1992.
- 740 Sammut, J., White, I., and Melville, M.: Acidification of an estuarine tributary in eastern
741 Australia due to drainage of acid sulfate soils, Marine and Freshwater Research, 47,
742 669-684, 1996.



- 743 Saunois, M., Bousquet, P., Poulter, B., Peregon, A., Ciais, P., Canadell, J. G., Dlugokencky,
744 E. J., Etiope, G., Bastviken, D., and Houweling, S.: The global methane budget 2000–
745 2012, *Earth System Science Data* (Online), 8, 2016.
- 746 Sivan, O., Antler, G., Turchyn, A. V., Marlow, J. J., and Orphan, V. J.: Iron oxides stimulate
747 sulfate-driven anaerobic methane oxidation in seeps, *Proceedings of the National*
748 *Academy of Sciences*, 111, E4139-E4147, 2014.
- 749 Sorrell, B. K., and Boon, P. I.: Convective gas flow in *Eleocharis sphacelata* R. Br.: methane
750 transport and release from wetlands, *Aquatic Botany*, 47, 197-212, 1994.
- 751 Tucker, G. C.: The genera of Arundinoideae (Gramineae) in the southeastern United States,
752 *Journal of the Arnold Arboretum*, 71, 145-177, 1990.
- 753 Tulau, M.: Acid Sulphate Soil Priority Management Areas on the Lower Hastings Camden
754 Haven Floodplains, Department of Land and Water Conservation, 1999.
- 755 Villa, J. A., and Bernal, B.: Carbon sequestration in wetlands, from science to practice: An
756 overview of the biogeochemical process, measurement methods, and policy framework,
757 *Ecological Engineering*, 114, 115-128, 10.1016/j.ecoleng.2017.06.037, 2018.
- 758 Walker, P. H.: Seasonal and stratigraphic controls in coastal floodplain soils, *Soil Research*,
759 10, 127-142, 1972.
- 760 White, I., Melville, M., Wilson, B., and Sammut, J.: Reducing acidic discharges from coastal
761 wetlands in eastern Australia, *Wetlands Ecology and Management*, 5, 55-72, 1997.
- 762 Whiting, G. J., and Chanton, J. P.: Plant-dependent CH₄ emission in a subarctic Canadian
763 fen, *Global biogeochemical cycles*, 6, 225-231, 1992.
- 764 Whiting, G. J., and Chanton, J. P.: Greenhouse carbon balance of wetlands: methane emission
765 versus carbon sequestration, *Tellus B: Chemical and Physical Meteorology*, 53, 521-
766 528, 2001.
- 767 Wong, V. N., Johnston, S. G., Bush, R. T., Sullivan, L. A., Clay, C., Burton, E. D., and
768 Slavich, P. G.: Spatial and temporal changes in estuarine water quality during a post-
769 flood hypoxic event, *Estuarine, Coastal and Shelf Science*, 87, 73-82, 2010.

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