

1 Major role of ammonia-oxidizing bacteria in N₂O production in the Pearl River Estuary

2 Li Ma^{1,2}, Hua Lin^{1,2,3}, Xiabing Xie¹, Minhan Dai^{1,2}, Yao Zhang^{1,2}

3 ¹State Key Laboratory of Marine Environmental Science, Xiamen University, Xiamen 361101, China

4 ²College of Ocean and Earth Sciences, Xiamen University, Xiamen 361101, China

5 ³Key Laboratory of Marine Ecosystem and Biogeochemistry, State Oceanic Administration, Second
6 Institute of Oceanography, Ministry of Natural Resources, Hangzhou 310012, China

7 Correspondence to: Yao Zhang (yaozhang@xmu.edu.cn)

8 **Abstract.** Nitrous oxide (N₂O) has significant global warming potential as a greenhouse gas. Estuarine
9 and coastal regimes are the major zones of N₂O production in the marine system. However, biological
10 sources of N₂O in estuarine ecosystems remain controversial, but are of great importance for
11 understanding global N₂O emission patterns. Here, we measured concentrations and isotopic
12 compositions of N₂O as well as distributions of ammonia-oxidizing bacterial and archaeal *amoA* and
13 denitrifier *nirS* genes by quantitative polymerase chain reaction along a salinity gradient in the Pearl
14 River Estuary, and performed in situ incubation experiments to estimate N₂O yields. Our results
15 indicated that nitrification predominantly occurred, with significant N₂O production during ammonia
16 oxidation, in the hypoxic waters of the upper estuary where the maximum N₂O and $\Delta\text{N}_2\text{O}_{\text{excess}}$
17 concentrations were observed, although minor denitrification might be concurrent at the site with the
18 lowest dissolved oxygen. Ammonia-oxidizing β -proteobacteria (AOB) were significantly positively
19 correlated with all N₂O-related parameters, although their *amoA* gene abundances were distinctly lower
20 than ammonia-oxidizing Archaea (AOA) throughout the estuary. Furthermore, the N₂O production rate
21 and the N₂O yield normalized to *amoA* gene copies or transcripts estimated a higher relative
22 contribution of AOB to the N₂O production in the upper estuary. Taken together, the in situ incubation
23 experiments, N₂O isotopic composition and concentrations, and gene datasets suggested that the high
24 concentration of N₂O (oversaturated) is mainly produced from strong nitrification by the relatively high
25 abundance of AOB in the upper reaches and is the major source of N₂O emitted to the atmosphere in the
26 Pearl River Estuary.

1 **1 Introduction**

2 Nitrous oxide (N_2O) is a potent greenhouse gas with global warming potential 298 times that of carbon
3 dioxide (CO_2) on a 100 yr timescale, and contributes to stratospheric ozone depletion as a major
4 precursor of free radicals (Ravishankara et al., 2009). N_2O emissions from soils and marine systems are
5 estimated to account for 56%–70% (6–7 Tg $\text{N}_2\text{O-N yr}^{-1}$) (Syakila and Kroeze, 2011; Butterbach-Bahl et
6 al., 2013; Hink et al., 2017) and 30% (4 Tg $\text{N}_2\text{O-N yr}^{-1}$) (Nevison et al., 2004; Naqvi et al., 2010; Voss
7 et al., 2013) of the total global N_2O emissions, respectively. The main processes responsible for N_2O
8 emissions are microbial transformation of ammonia, nitrite, and nitrate through nitrification and
9 denitrification (Butterbach-Bahl et al., 2013). It has been estimated that oceanic N_2O production is
10 dominated by nitrification, whereas only 7% is contributed by denitrification (Freing et al., 2012).

11 N_2O is released as a byproduct during nitrification via incomplete oxidation of hydroxylamine
12 (NH_2OH) to nitrite (NO_2^-) by ammonia-oxidizing bacteria (AOB) (Stein, 2011). This process may be
13 enhanced under suboxic conditions (Naqvi et al., 2010). While no equivalent of the hydroxylamine-
14 oxidoreductase that catalyzes N_2O formation through NH_2OH oxidation has been found in ammonia-
15 oxidizing archaea (AOA) (Hatzenpichler, 2012), recent studies indicated that AOA possibly produces
16 hybrid N_2O via a combination of an ammonia oxidation intermediate (NH_2OH , HNO , or NO) and NO_2^-
17 (Stieglmeier et al., 2014; Frame et al., 2017). In addition, AOB have been shown to produce N_2O from
18 NO_2^- during nitrifier denitrification (Shaw et al., 2006). This process is also promoted under micro-oxic
19 and anoxic conditions (Yu et al., 2010). Denitrification by heterotrophic denitrifiers is another major
20 pathway of N_2O production in marine environments, occurring under anoxic conditions or at the
21 suboxic–anoxic interface (Naqvi et al., 2010; Yamagishi et al., 2007; Ji et al., 2018). NO_2^- is reduced by
22 a copper-containing (NirK) or cytochrome cd1-containing nitrite reductase (NirS) to nitric oxide (NO),
23 and then by a heme-copper NO reductase (NOR) to N_2O (Coyne et al., 1989; Treusch et al., 2005;
24 Abell et al., 2010; Bartossek et al., 2010; Lund et al., 2012; Graf et al., 2014). As an intermediary
25 product during denitrification, production and further reduction of N_2O are sensitive to different O_2
26 conditions (Babbin et al., 2015; Ji et al., 2015).

27 Biological nitrogen transformations are catalyzed by various microbial enzymes, of which
28 ammonium monooxygenase (AMO) and nitrite reductases (NIRs) are key enzymes responsible for

1 nitrification and denitrification, respectively (Canfield et al., 2010). The genes encoding for AMO
2 subunit A (*amoA*) and NIRs (*nirS* and *nirK*) have been widely applied as functional marker genes to
3 identify the distribution of ammonia oxidizers and denitrifiers. Previous studies have shown significant
4 correlations of *amoA* with spatial variations of N₂O emissions or N₂O production rates in soils and
5 oceans (Avrahami and Bohanann, 2009; Santoro et al., 2011; Löscher et al., 2012). In addition,
6 significant relationships between *nirK* or *nirS* abundances and N₂O emissions were observed in
7 grasslands (Čuhel et al., 2010), arable soils (Clark et al., 2012; Jones et al., 2014), and the ocean
8 (Arévalo-Martínez et al., 2015).

9 Estuaries are highly impacted by coastal nutrient pollution and eutrophication because of
10 anthropogenic activity; they play a significant role in nitrogen cycling at the land–sea interface (Bricker
11 et al., 2008; Damashek et al., 2016; Damashek and Francis 2018). Estuarine and coastal regimes have
12 long been recognized as major zones of N₂O production in the marine system (Seitzinger and Kroeze,
13 1998; Mortazavi et al., 2000; Usui et al., 2001; Kroeze et al., 2010; Allen et al., 2011). In particular,
14 eutrophic estuaries with extensive oxygen-deficient zones have been considered hotspot regions for
15 N₂O production (Abril et al., 2000; De Wilde and De Bie, 2000; Garnier et al., 2006; Lin et al., 2016),
16 with oversaturated N₂O and high N₂O concentrations and flux (De Wilde and De Bie 2000; De Bie et al.
17 2002; Garnier et al., 2006; Rajkumar et al., 2008; Barnes and Upstill-Goddard 2011; Lin et al., 2016).
18 The dynamics of N₂O emissions in these ecosystems are regulated by complex physical and
19 biogeochemical processes; for example, mixing between freshwater and oceanic waters influences the
20 biogeochemistry of estuarine waters as well as microbial activity (Huertas et al., 2018; Laperriere et al.,
21 2019).

22 Nitrification is often credited as the dominant N₂O production pathway in estuaries (De Bie et al.
23 2002; Barnes and Upstill-Goddard 2011; Kim et al. 2013; Lin et al. 2016; Huertas et al., 2018;
24 Laperriere et al., 2019). Although AOA frequently outnumber AOB and dominate microbial
25 communities, their contribution to nitrification remains controversial in estuarine and coastal waters
26 (Bernhard et al., 2010; Zhang et al., 2014; Hou et al., 2018). Furthermore, the relative contributions of
27 AOB and AOA to N₂O production is inconclusive (Monteiro et al., 2014) and there is a potential niche
28 overlap between nitrifiers and denitrifiers in low oxygen conditions (Frame and Casciotti, 2010; Zhang

1 et al., 2014; Penn et al., 2016). AOB are reported to thrive in hypoxic environments and denitrification
2 in the oxic ocean is suggested to occur within anoxic particle interiors (Frame and Casciotti, 2010; Ni et
3 al., 2014). It is therefore of great importance to elucidate the biological sources of N₂O production in
4 estuarine ecosystems to better understanding global N₂O emission patterns.

5 The Pearl River Estuary (PRE) is one of the world's most complex estuarine systems with a total
6 discharge of $285.2 \times 10^9 \text{ m}^3 \text{ yr}^{-1}$ (Dai et al., 2014). The PRE is surrounded by complex regions with a
7 rich nitrogen supply that produces eutrophic waters (Dai et al., 2008). Moreover, increased oxygen
8 consumption by organic matter degradation leads to the formation of hypoxic zones in the upper reaches
9 of the PRE (Dai et al., 2006; He et al., 2014), which may support strong nitrification, denitrification,
10 and N₂O production (Lin et al., 2016). In this study, N₂O-related biogeochemical parameters were
11 measured, and distributions of AOB and AOA *amoA* and denitrifier *nirS* genes were quantified by
12 quantitative polymerase chain reaction (qPCR) to investigate the relationship between N₂O production
13 and spatial distribution of nitrifiers and denitrifiers along a salinity gradient in the PRE (Fig. 1).
14 Moreover, in situ incubation experiments were performed in the hypoxic upper estuary to estimate (1)
15 nitrification and N₂O production rates, (2) whether denitrification occurred during nitrification, and (3)
16 N₂O yield (mol N₂O-N produced per mol ammonia oxidized). By combining the genetic datasets and
17 incubation estimates, this study thus identified the relative contributions of AOB and AOA in producing
18 N₂O in the PRE.

19 **2 Materials and methods**

20 **2.1 Study area and sampling**

21 A total of 22 sites along the salinity gradient of the PRE were sampled during a research cruise in July
22 2015, including 11 sites in the upper reaches (upstream of the Humen outlet) and 11 sites in the lower
23 reaches (Lingdingyang) (Fig. 1). Water samples were taken from the surface (2 m) and bottom (4–15 m)
24 of each site by using a conductivity, temperature, and depth (CTD) rosette sampling system (SBE 25;
25 Sea-Bird Scientific, USA) fitted with 12 L Niskin bottles (General Oceanics). A total of 16 samples
26 (from two depths at eight sites) were subjected to gene analysis (Fig. 1). A total of 1 L of water for gene

1 analysis was serially filtered through 0.8 μm and then 0.22 μm pore size polycarbonate membrane filters
2 (47 mm diameter, Millipore) within 30 min at a pressure <0.03 MPa to retain the particle-associated
3 (PA) communities (>0.8 μm) and free-living (FL) communities (0.22–0.8 μm). For the upper estuary
4 samples, more membrane filters were used to avoid the filters clogging. RNAlater solution (Ambion,
5 Austin, Texas, USA) was quickly added to the samples to prevent RNA degradation. All of the filters
6 were immediately flash frozen in liquid nitrogen and then stored at -80 $^{\circ}\text{C}$ until further analysis. Water
7 samples for nutrient determination were filtered through 0.45 μm pore size cellulose acetate membranes
8 and then immediately frozen at -20 $^{\circ}\text{C}$ until further analysis. Water samples for dissolved N_2O were
9 collected into 125 mL headspace glass bottles to which 100 μL of saturated HgCl_2 was added; the
10 bottles were immediately closed with rubber stoppers and aluminum crimp-caps and stored in the dark
11 at 4 $^{\circ}\text{C}$ until analysis in the laboratory. All N_2O samples were collected during the July 2015 cruise
12 except for samples from sites P03, P05, A01, A06, and A10 intended for N_2O isotopic composition
13 analyses, which were sampled during a cruise in March 2010. Total suspended material (TSM) was
14 collected by filtering 1–4 L of water onto pre-combusted and pre-weighed glass fiber filters (GF/Fs)
15 (Whatman), and then stored at -20 $^{\circ}\text{C}$ until weighing in the laboratory.

16 **2.2 Biogeochemical parameters, N_2O emissions and isotopic analysis of environmental samples**

17 Temperature and salinity were measured with the SBE 25 CTD system. Dissolved oxygen (DO)
18 concentrations were measured using the Winkler method (Dai et al., 2006). Ammonia was measured
19 using the indophenol blue spectrophotometric method (Pai et al., 2001) on board; nitrate, nitrite, and
20 silicate were analyzed using routine spectrophotometric methods with a Technicon AA3 Auto-Analyzer
21 (Bran-Lube, GmbH) (Han et al., 2012). N_2O concentrations were analyzed by gas chromatography (GC,
22 Agilent 6890 μECD) coupled with a purge-trap system (Tekmar Velocity XPT) at 25 $^{\circ}\text{C}$ (Lin et al.,
23 2016). N_2O standard gases of 1.02 and 2.94 ppmv $\text{N}_2\text{O}/\text{N}_2$ (National Center of Reference Material,
24 China, Beijing) were used. The relative standard deviation of the slope of the standard working curve
25 was 1.77% ($n=8$). The detection limit was calculated to be ~ 0.1 nmol L^{-1} and the precision was better
26 than $\pm 5\%$. When water samples were analyzed, every 5–10 samples were spiked with N_2O standards to
27 calibrate the GC.

1 The excess N₂O ($\Delta N_{2}O_{\text{excess}}$) and N₂O saturation (S%) were calculated with Eq. (1) and (2):

$$2 \Delta N_{2}O_{\text{excess}} = N_{2}O_{\text{observed}} - N_{2}O_{\text{equilibrium}} \quad (1)$$

$$3 S\% = N_{2}O_{\text{observed}} / N_{2}O_{\text{equilibrium}} \times 100\% \quad (2)$$

4 where $N_{2}O_{\text{observed}}$ represents the measured concentrations of N₂O in the water, and the equilibrium
5 values of N₂O ($N_{2}O_{\text{equilibrium}}$) were calculated by Eq. (3) and (4) (Weiss and Price, 1980):

$$6 N_{2}O_{\text{equilibrium}} = xF \quad (3)$$

$$7 \ln F = A_1 + A_2(100/T) + A_3 \ln(T/100) + A_4(T/100)^2 + S[B_1+B_2(T/100) + B_3(T/100)^2] \quad (4)$$

8 where x is the mole fraction of N₂O in the atmosphere and T is the absolute temperature. In this study,
9 we used the global mean atmospheric N₂O (327 ppb) from 2015 (<http://www.esrl.noaa.gov/gmd>). The
10 fitted function F and constants A_1 , A_2 , A_3 , A_4 , B_1 , B_2 and B_3 were proposed by Weiss and Price
11 (1980).

12 The N₂O flux (F_{N_2O} , $\mu\text{mol m}^{-2} \text{d}^{-1}$) through the air–sea interface was estimated based on Eq. (5):

$$13 F_{N_2O} = k_{N_2O} \times \rho \times K_H^{N_2O} \times \Delta pN_2O = k_{N_2O} \times 24 \times 10^{-2} \times (N_{2}O_{\text{observed}} - N_{2}O_{\text{equilibrium}}) \quad (5)$$

14 where k_{N_2O} (cm h^{-1}) is the N₂O gas transfer velocity depending on wind and water temperature, $K_H^{N_2O}$ is
15 the solubility of N₂O, and ΔpN_2O is the average sea–gas N₂O partial pressure difference. k_{N_2O} was
16 estimated using Eq. (6) according to Wanninkhof (1992):

$$17 k_{N_2O} = 0.31 \times u_{\text{av}}^2 \times (Sc_{N_2O}/600)^{-0.5} \quad (6)$$

18 where u_{av} is the average wind speed 10 m above the water surface. In this study, a CO₂ Schmidt number
19 (Sc) of 600 at 20 °C in fresh water (Wanninkhof, 1992) was used for estuarine systems (Raymond and
20 Cole, 2001). The Sc is defined as the kinematic viscosity of water divided by the diffusion coefficient of
21 the gas and calculated from temperature (Wanninkhof, 1992). For N₂O in waters with salinities <35 and
22 temperatures ranging from 0–30 °C, Sc_{N_2O} was estimated using Eq. (7) according to Wanninkhof (1992):

$$23 Sc_{N_2O} = 2055.6 - 137.11 t + 4.3173 t^2 - 0.05435 t^3 \quad (7)$$

24 where t is the in situ temperature of the sampling site.

25 To determine the isotopic composition of N₂O, the gas samples were introduced into a trace gas
26 cryogenic pre-concentration device (PreCon, Thermo Finnigan), as described in Cao et al. (2008) and
27 Zhu et al. (2008), and then $\delta^{15}\text{N-N}_2\text{O}$ was analyzed using an isotope ratio mass spectrometer (IRMS,

1 Thermo Finnigan MAT-253, Bremen, Germany). The molecular ions of N₂O (N₂O⁺, m/z 44, 45 and 46)
2 were quantified by IRMS to calculate isotope ratios for the entire molecule (¹⁵N/¹⁴N and ¹⁸O/¹⁶O). The
3 δ¹⁵N values of N₂O in samples were calculated using the ¹⁵N/¹⁴N of the pure N₂O reference gas and
4 samples (Frame and Casciotti, 2010; Mohn et al., 2014). The reference gas was previously calibrated
5 against N₂O isotopic standard gas (δ¹⁵N (vs Air-N₂) = -0.320‰) produced by Shoko Co. Ltd. (Tokyo,
6 Japan) and the δ¹⁵N value (vs Air-N₂) of the N₂O reference gas is 6.579±0.030‰. The precision of the
7 method for δ¹⁵N-N₂O was estimated as 0.3‰.

8 **2.3 Nucleic acid extraction and qPCR**

9 DNA was extracted using the FastDNA™ SPIN Kit for Soil (MP, USA) according to the
10 manufacturers' protocol with minor modifications. RNA was extracted using TRIzol reagent (Ambion,
11 Austin, Texas, USA), and then eluted with 50 µL of RNase-free water. The extracted RNA was treated
12 with DNase I (Invitrogen, Carlsbad, CA) to remove any residual DNA. DNA contamination was
13 checked by amplifying the bacterial 16S rRNA genes before reverse transcription. Total RNA without
14 DNA contamination was reverse transcribed to synthesize single-strand complementary DNA (cDNA)
15 using the First-Strand cDNA Synthesis Kit (Invitrogen, Austin, Texas, USA).

16 The transcript and copy abundances of bacterial and archaeal *amoA* genes and bacterial *nirS* genes
17 were examined using qPCR and a CFX96 Real Time PCR system (BIO-RAD, Singapore). The β-
18 proteobacterial and archaeal *amoA* were amplified using primer sets amoA-1F and amoA-2R (Kim et
19 al., 2008) and Arch-amoAF and Arch-amoAR (Francis et al., 2005), respectively; *nirS* was amplified
20 using primers nirS-1F and nirS-3R (Braker et al., 1998; Huang et al., 2011). Quantitative PCR
21 amplification for the β-proteobacterial and archaeal *amoA* was carried out as described previously
22 (Mincer et al., 2007; Hu et al., 2011). For the amplification of *nirS*, the qPCR reaction mixture was
23 prepared in accordance with Zhang et al. (2014) and thermal cycling conditions were as described in
24 Huang et al. (2011). Standards for the qPCR reactions consisted of serial 10-fold dilutions (10⁷ to 10⁰
25 copies per µL) of plasmid DNA containing amplified fragments of the targeted genes (accession
26 numbers MH458281 for β-proteobacterial *amoA*, KY387998 for archaeal *amoA* and KF363351 for
27 *nirS*). The amplification efficiencies of qPCR were always between 85%–95% with R² >0.99. The

1 specificity of the qPCR reactions was confirmed by melting curve analysis, agarose gel electrophoresis
2 and sequencing analysis. Inhibition tests were performed by 2-fold and 5-fold dilutions of all samples
3 and indicated that our samples were not inhibited.

4 **2.4 Incubation experiments**

5 Incubation experiments were performed in the surface and bottom waters at sites P01 (2 and 5 m water
6 depth) and P05 (2 and 12 m) upstream of the Humen outlet (Fig. 1). Water samples were collected from
7 Niskin bottles through a clean polytetrafluoroethylene (Teflon) silicone hose, and carefully filled into
8 125 mL clean headspace glass bottles without gas bubbles. The bottles were immediately closed with an
9 air-tight butyl rubber stopper and aluminum crimp-cap. A total of 43 bottles were set up for surface and
10 bottom at sites P01 and 34 bottles at P05. Samples from four parallel bottles were taken to determine the
11 initial (t_0) dissolved N_2O concentration, and triplicate samples were taken to measure the initial
12 dissolved inorganic nitrogen (DIN) concentration, which included ammonium, nitrite, and nitrate. The
13 remaining 36 (P01) and 27 (P05) bottles were incubated in the dark at in situ temperatures (± 1 °C). At
14 site P01, samples from six parallel bottles were taken at 3, 6, 18, and 24 h during the incubation
15 experiment for N_2O determination after injecting saturated mercuric chloride ($HgCl_2$, 1:100 v:v) into the
16 bottles; triplicate samples were also taken at the same time for DIN measurements by filtering through
17 0.7 μm pore size GF/Fs under pressure <0.03 MPa. Concentrations of N_2O , ammonium, nitrite, and
18 nitrate were measured as described in Sect. 2.2. At site P05, samples were taken after 3, 6, and 12 h
19 incubation and the other procedures were the same as described for site P01.

20 The effect of DIN assimilation is negligible during incubation in the dark (Ward, 2008). Therefore,
21 the potential processes of nitrogen transformation and N_2O production can be determined according to
22 “mass balance” in a closed incubation system. The main processes were analyzed based on the dynamic
23 variations of DIN (ΔDIN), ammonia ($\Delta NH_3 + NH_4^+$), nitrite (ΔNO_2^-), nitrate (ΔNO_3^-), and N_2O (ΔN_2O)
24 concentrations during incubation. The average rates of nitrification and N_2O production were estimated
25 using the slopes of the linear regression between concentrations versus incubation time when DIN was
26 in balance (i.e. no denitrification). All of the concentration-based rates described from the incubations
27 represent net rates. The N_2O yield during nitrification was calculated with Eq. (8):

$$1 \quad \text{N}_2\text{O}_{\text{yield}} (\%) = \Delta\text{N}_2\text{O-N} / \Delta(\text{NH}_3 + \text{NH}_4^+)\text{-N} \quad (8)$$

2 **2.5 Statistical analyses**

3 Since a normal distribution of the individual data sets was not always met, we used the non-parametric
4 Wilcoxon rank-sum tests for comparing two variables. The bivariate correlations between
5 environmental factors and functional genes were described by Spearman correlation coefficients (ρ
6 value). False discovery rate (FDR)-based multiple comparison procedures were applied to evaluate the
7 significance of multiple hypotheses and identify truly significant comparisons (FDR-adjusted P value)
8 (Pike, 2011). The maximum gradient length of detrended correspondence analysis was shorter than 3.0,
9 thus redundancy analysis (RDA) based on the qPCR data was used to analyze variations in the AOA
10 and AOB distributions under environmental constraints in the software R (version 3.4.4) Vegan 2.5–3
11 package. The qPCR-based relative abundances and environmental factors were normalized via Z
12 transformation (Magalhães et al., 2008). The null hypothesis, that the community was independent of
13 environmental parameters, was tested using constrained ordination with a Monte Carlo permutation test
14 (999 permutations). Significant environmental parameters ($P < 0.05$) without multicollinearity (variance
15 inflation factor < 20) (Ter Braak, 1986) were obtained. Standard and partial Mantel tests were run in R
16 (version 3.4.4, Vegan 2.5–3 package) to determine the correlations between environmental factors and
17 the AOA and AOB distributions. Dissimilarity matrices of communities and environmental factors were
18 based on Bray-Curtis and Euclidean distances between samples, respectively. Based on Spearman
19 correlation, the significance of the Mantel statistics was obtained after 999 permutations. Statistical tests
20 were assumed to be significant at a P value of < 0.05 .

21 **3 Results**

22 **3.1 Distribution of nutrients, DO, and N₂O along a salinity transect of the PRE**

23 The studied transect was divided into a northern region upstream of the Humen outlet and southern area
24 (Lingdingyang) (Fig. 1); these regions have distinct biogeochemical characteristics. Salinity exhibited
25 low values (0.1 to 4.4) upstream of the Humen outlet, and sharply increased from 0.7 to 34.2
26 downstream in Lingdingyang (Fig. 2a). The ammonium/ammonia concentrations decreased from 167.2

1 $\mu\text{mol L}^{-1}$ (site P01 surface water) to $20.9 \mu\text{mol L}^{-1}$ (site P07 bottom water) upstream of the Humen
2 outlet and consistently decreased downstream in Lingdingyang ($5.7 \mu\text{mol L}^{-1}$ to below detection limit)
3 (Fig. 2b). Correspondingly, the sum of nitrate and nitrite concentrations increased from $93.6 \mu\text{mol L}^{-1}$
4 (site P01 bottom water) to $172.3 \mu\text{mol L}^{-1}$ (site P03 surface water) upstream, but it sharply decreased
5 seaward to Lingdingyang (Fig. 2c). The DO concentrations were distinctly lower upstream of the
6 Humen outlet with nearly one-half of the samples below the hypoxic threshold ($63.0 \mu\text{mol L}^{-1}$; Rabalais
7 et al., 2010). Generally, the DO concentrations increased seaward from 155.7 to $238.0 \mu\text{mol L}^{-1}$ in the
8 surface waters of the Lingdingyang area, whereas they varied from 74.0 to $183.3 \mu\text{mol L}^{-1}$ in the
9 bottom waters (Fig. 2d).

10 In contrast to the DO concentrations, the N_2O concentrations were distinctly higher upstream of the
11 Humen outlet (48.9 – $148.2 \text{ nmol L}^{-1}$) than in Lingdingyang, where they decreased seaward from 24.6 to
12 5.4 nmol L^{-1} (Fig. 2e). Similarly, higher $\Delta\text{N}_2\text{O}_{\text{excess}}$ (42.0 – $141.3 \text{ nmol L}^{-1}$) with saturations from
13 701.1% to 2175.1% was observed upstream; lower $\Delta\text{N}_2\text{O}_{\text{excess}}$ (-1.4 – 17.8 nmol L^{-1}) was present in the
14 Lingdingyang area with the saturations ranging from 86% to 363% (Fig. 2f). The estimated water–air
15 N_2O fluxes were 100.4 to $344.0 \mu\text{mol m}^{-2} \text{ d}^{-1}$ upstream and decreased in Lingdingyang (42.4 to -2.6
16 $\mu\text{mol m}^{-2} \text{ d}^{-1}$) (Fig. 2g). Together, the PRE acts as a N_2O source that releases to the atmosphere and
17 notably, a significant negative relationship was observed between $\Delta\text{N}_2\text{O}_{\text{excess}}$ or N_2O flux and DO ($P <$
18 0.01 for each) in the upper estuary (Fig. 2i and j). The isotopic compositions of N_2O ($\delta^{15}\text{N}\text{-N}_2\text{O}$)
19 showed an enrichment of $^{15}\text{N}_2\text{O}$ seaward, varying from -27.9 to 7.1‰ (Fig. 2h). Overall, upstream of
20 the Humen outlet was characterized by hypoxic waters rich in nitrogen-based nutrients, where
21 ammonium concentrations decreased and the sum of nitrite and nitrate concentrations increased seaward,
22 corresponding to distinctly higher N_2O fluxes released to the atmosphere.

23 **3.2 Distributions of *amoA* and *nirS* genes along the salinity transect**

24 The total abundance of AOA *amoA* (sum of FL and PA communities) varied from 3.10×10^3 to 6.87×10^5
25 copies L^{-1} in the surface waters (Fig. 3a) and 6.40×10^4 to 4.21×10^7 copies L^{-1} in the bottom waters; an
26 increase along the salinity transect was observed in the bottom (Fig. 3b). In contrast, the total abundance

1 of AOB *amoA* generally decreased seaward along the salinity transect for the surface (4.23×10^2 to
2 2.13×10^4 copies L^{-1}) and bottom waters (4.49×10^3 to 8.79×10^4 copies L^{-1}) (Fig. 3c and d). Overall, the
3 abundance of AOA *amoA* was significantly higher than AOB ($P < 0.01$). The total abundance of *nirS*
4 varied from 9.12×10^4 to 2.00×10^7 copies L^{-1} and was higher than both AOA ($P < 0.05$) and AOB *amoA*
5 ($P < 0.01$) in the surface waters and AOB *amoA* in the bottom water ($P < 0.01$) (Fig. 3e and f). Notably,
6 these three genes were predominantly distributed in the PA communities compared to the FL
7 communities in the PRE transect (Fig. 3). The transcripts of the three genes were analyzed in the PA
8 communities of the two incubation sites upstream of the Humen outlet. The transcript abundances of
9 AOA *amoA* (7.44×10^3 to 4.62×10^5 transcripts L^{-1}) were one to three orders of magnitude higher than
10 AOB *amoA* (3.62×10^2 to 5.00×10^2 transcripts L^{-1}) at P01 (Fig. 3a–d), whereas the transcript
11 abundances of AOB *amoA* were relatively higher at P05 (AOB = 8.96×10^4 to 3.83×10^5 transcripts L^{-1} ;
12 AOA = 1.26×10^4 to 1.39×10^5 transcripts L^{-1}). The *nirS* gene showed a similar transcript level with
13 AOA *amoA* at P01 (2.20×10^4 to 6.69×10^4 transcripts L^{-1}), but one order of magnitude lower transcript
14 level than both AOA and AOB *amoA* at P05 (8.59×10^3 to 1.12×10^4 transcripts L^{-1}) (Fig. 3e and f).

15 **3.3 Correlations between genes abundances and biogeochemical parameters**

16 We analyzed the correlations between the genes abundances of AOA, AOB, or denitrifiers and
17 biogeochemical parameters. The results indicate that AOA *amoA* abundance was significantly
18 correlated ($P < 0.05$ – 0.01) to the hydrographic parameters temperature (negative) and salinity (positive),
19 as well as silicate concentration (negative) (Table 1). However, AOB *amoA* abundance was
20 significantly correlated ($P < 0.05$ – 0.01) to TSM concentration (positive), pH (negative), and DO
21 (negative). Notably, there were positive correlations between AOB *amoA* abundances and all N_2O
22 parameters as well as ammonia concentrations (Table 1; $P < 0.05$ – 0.01) except for the extremely low-
23 abundance of FL AOB. No significant Spearman correlations were found between bacterial nitrite
24 reductase *nirS* abundance and the measured biogeochemical parameters.

25 The RDA was used to further analyze variations in the AOA and AOB distributions under
26 environmental constraints. The results confirmed that the relatively high AOB abundances in the upper
27 estuary were constrained by low salinity water, high nitrite and TSM concentrations, low DO conditions,

1 and high N₂O concentrations whereas high salinity water and opposite environmental conditions
2 constrained the relatively high AOA abundances in the Lingdingyang area (Fig. 4). These constraints
3 explained 89.3% of the variation in the ammonia oxidizer distribution along the PRE. Apparently, the
4 communities with relatively high AOB abundances in the upper estuary positively influenced the
5 concentration of N₂O in the water.

6 **3.4 Nitrogen transformation and N₂O production in the incubation experiments**

7 The in situ biogeochemical conditions of the incubation experiments are shown in Fig. 2 and listed in
8 Table S1. Site P01 exhibited the lowest in situ DO concentrations (30.0 μmol L⁻¹ in the bottom water
9 and 30.9 μmol L⁻¹ in the surface water). The concentration of DIN was generally unchanged in the
10 early-to-middle (0–18 h) phase for the P01 surface water and early (0–6 h) phase for the P01 bottom
11 water, but showed a distinct decrease in the end phase (Fig. 5a). The ammonia and nitrite concentrations
12 consistently decreased and increased, respectively, during the incubation experiments; the nitrate
13 concentrations decreased in the end phase after a slight increase (Fig. 5b). These results clearly indicate
14 that nitrification occurred during the entire P01 incubations, and suggest that denitrification may be
15 present in the end phase (Fig. 5g). The rates of ammonia oxidation during the entire incubations and
16 nitrite oxidation during the early or early-to-middle phases were estimated by linear regressions of
17 ammonia and nitrate concentrations, respectively (Fig. 5a and b; Table 2). Correspondingly, the
18 estimated average N₂O production rate (24 h) was 0.62 nmol L⁻¹ h⁻¹ in P01 surface water and 0.70 nmol
19 L⁻¹ h⁻¹ in P01 bottom water; the estimated N₂O production rates from nitrification were 0.60 nmol L⁻¹
20 h⁻¹ in the surface water (18 h) and 1.61 nmol L⁻¹ h⁻¹ in the bottom water (6 h; Fig. 5c). Thus, the
21 estimated N₂O yield in the surface and bottom waters based on nitrification was 0.26% and 0.30%
22 (Table 2).

23 In the incubation experiments at site P05, the DIN concentrations remained unchanged (Fig. 5d)
24 and the ammonia concentrations consistently decreased and the nitrite and nitrate concentrations
25 increased (Fig. 5e). The rates of ammonia and nitrite oxidation were also estimated by linear regressions
26 of ammonia and nitrate concentrations, respectively (Fig. 5d and e; Table 2). The ammonia oxidation
27 rates were approximately equal to the sum of the increased nitrite and nitrate concentration rates. Thus,

1 nitrification occurred during the incubation experiments without denitrification. The estimated N₂O
2 production rates from nitrification were 1.15 nmol L⁻¹ h⁻¹ in the P05 surface water and 1.41 nmol L⁻¹
3 h⁻¹ in the P05 bottom water (Fig. 5f); the estimated N₂O yields based on nitrification were 0.21%
4 (surface) and 0.32% (bottom) (Table 2).

5 The N₂O production rates and yields normalized to total AOA and AOB *amoA* gene copies (sum
6 of PA and FL fractions or only PA fractions) or transcripts (only PA fraction) were calculated (Table
7 S3). The highest average *amoA* gene copy-specific N₂O production rates and yields were in the surface
8 water of site P05, where the highest nitrification rate was observed (Table 2). The highest average
9 *amoA* gene transcript-specific N₂O production rates and yields were in the bottom water of site P01,
10 where the highest N₂O production rate was observed (Table 2).

11 **4 Discussion**

12 **4.1 Contribution of nitrification versus denitrification to N₂O production in the hypoxic upper** 13 **estuary**

14 The spatial variations of N₂O concentration, its saturation, and water–air N₂O flux along the PRE are
15 consistent with our previous study (Lin et al., 2016), indicating that higher N₂O in the upper estuary
16 ensures the PRE acts as a source of atmospheric N₂O. The in situ incubation experiments clearly
17 indicated that nitrification predominantly occurred in the hypoxic waters of the upper estuary along with
18 significant N₂O production, and suggested that denitrification could be concurrent at the lowest DO site
19 (P01) where the maximum N₂O and $\Delta N_2O_{\text{excess}}$ concentrations were observed. These results confirm
20 previous speculation that extreme enrichment of ammonia in the water column due to high loads of
21 anthropogenic-sourced nutrients and organic matter in an upper estuary (Dai et al., 2008; He et al., 2014)
22 could result in strong nitrification under low O₂ solubility conditions (Dai et al., 2008); thus, N₂O is
23 produced as a byproduct through nitrification and is oversaturated in the PRE (Lin et al., 2016). The
24 PRE sediments also act as a source of N₂O, which is released into the overlying waters through
25 denitrification (Tan et al., 2019); however, in estuarine waters, nitrification apparently is the main
26 source of N₂O production. Previous studies also proposed that nitrification may be the major source of
27 N₂O production in the water column in estuarine systems, such as the Guadalquivir (Huertas et al.,

1 2018), Schelde (De Wilde and De Bie, 2000), and Chesapeake Bay (Laperriere et al., 2019). However,
2 in the estuarine sediments, N₂O production was attributed to both nitrification and denitrification, such
3 as in the Tama (Japan) (Usui et al., 2001) and Yangtze (China) estuaries (Liu et al., 2019; Wang et al.,
4 2019), where denitrification is the major nitrogen removal pathway with N₂O production and
5 consumption.

6 The isotopic composition of N₂O ($\delta^{15}\text{N-N}_2\text{O}$) was consistent with the above interpretation.
7 According to previous studies (Table S2), the $\delta^{15}\text{N}$ of N₂O produced during ammonia oxidation by
8 AOB strains ranged from -68‰ to -6.7‰ (Yoshida, 1988; Sutka et al., 2006; Mandernack et al., 2009;
9 Frame and Casciotti, 2010; Jung et al., 2014; Toyoda et al., 2017) and $6.3\text{--}10.2\text{‰}$ in a marine AOA
10 strain (Santoro et al., 2011). The $\delta^{15}\text{N}$ of N₂O produced during denitrification ranged from -37.2‰ to
11 -7.9‰ (Toyoda et al., 2005); during nitrifier-denitrification by AOB strains it ranged from $-57.6 \pm$
12 4.1‰ to -21.5‰ (Sutka et al., 2003; Sutka et al., 2006; Frame and Casciotti, 2010). Therefore, the
13 much lower $\delta^{15}\text{N-N}_2\text{O}$ (-27.9‰ to -12.6‰) upstream of the Humen outlet is consistent with AOB
14 nitrification or denitrification processes, whereas enriched $^{15}\text{N-N}_2\text{O}$ ($5.2\text{--}7.1\text{‰}$) in the lower reaches
15 approaches AOA nitrification and air $^{15}\text{N-N}_2\text{O}$ (Santoro et al., 2011). Taken together, the isotopic
16 compositions of N₂O (Fig. 2h) and N₂O concentration distribution (Fig. 2e–g) suggest that the high
17 concentrations of N₂O (oversaturation) were produced from strong nitrification by AOB and probably
18 concurrent minor denitrification in the upper estuary, however in the lower reaches, low concentrations
19 of N₂O could be explained by AOA nitrification or water atmospheric exchange of N₂O.

20 **4.2 Correlations of AOB versus AOA with N₂O-related biogeochemical parameters along the PRE**

21 The more abundant AOA *amoA* genes, relative to AOB, and the more abundant genes in the PA
22 communities than FL communities are consistent with our previous study in the PRE (Hou et al., 2018),
23 which also reported significant positive correlations between the AOB *amoA* gene abundance and the
24 oxidation rate of ammonia to nitrate. This suggests that AOB might be active in the ammonium and
25 particle-enriched PRE despite their low abundance (Füssel, 2014; Hou et al., 2018). Lower oxygen
26 availability in particle micro-niches has been reported to be favorable for both nitrification and
27 denitrification potential in oxygenated water (Kester et al., 1997). The Spearman correlations and RDA

1 analyses in this study indicate that high nutrient and TSM concentrations and low DO and pH
2 conditions were favourable for relatively high abundance of AOB in the upper estuary, which is also
3 consistent with our previous PRE study that found high TSM concentrations and low DO and pH
4 influenced substrate availability and thus AOB distribution (Hou et al., 2018). Moreover, AOB *amoA*
5 abundances positively correlated to all N₂O-related parameters as revealed by the Spearman correlations
6 and RDA analyses, suggesting a significant influence of AOB (mainly the PA fraction) on N₂O
7 production/emission in the upper estuary. However, compared to AOB, AOA *amoA* distribution along
8 the PRE transect appears to be regulated more by water mixing since AOA was significantly correlated
9 to the hydrographic parameters and silicate concentration.

10 To further eliminate the co-varying effects of water mixing, substrate availability, and N₂O-related
11 parameters along the salinity transect, and to identify the intrinsic/direct relationship between ammonia
12 oxidizers and N₂O production, we performed standard and partial Mantel tests. We defined four types of
13 environmental constraints: water mixing parameters (temperature, salinity, and silicate), substrate
14 parameters (ammonia/ammonium, nitrite, and nitrate), parameters influencing substrate availability (DO,
15 TSM, and pH), and N₂O-related parameters (N₂O and $\Delta N_2O_{\text{excess}}$). For the water mixing parameters, we
16 analyzed the relationships between potential temperature (θ), salinity, and silicate concentration with a
17 three-dimensional scatter plot (Fig. S1) that indicates low salinity and high silicate contents were the
18 best indicators for river input in the ocean (Moore, 1986). Thus, we chose temperature, salinity, and
19 silicate as proxies to trace estuarine water masses and mixing. Water mixing parameters (standard and
20 partial Mantel tests, $P < 0.01$) and those influencing substrate availability (standard and partial Mantel
21 tests, $P < 0.05$) significantly controlled variations in the distribution of AOA and AOB along the PRE
22 transect (Fig. 6a and c), supporting the Spearman and RDA conclusions. Notably, variations in the
23 distribution of AOA and AOB were significantly correlated with N₂O production (standard and partial
24 Mantel test, $P < 0.01$) after eliminating the co-varying effects of other parameters (Fig. 6d),
25 demonstrating the significant contribution of ammonia oxidizers to N₂O production.

1 **4.3 Contribution of AOB versus AOA to N₂O production**

2 We attempted to accurately assess the relative contributions of AOA and AOB to N₂O production in the
3 PRE by plotting the N₂O production rates (Fig. 7a) and yields (Fig. 7b) normalized to total (sum of
4 AOA and AOB) *amoA* gene copies or transcripts at sites P01 and P05 along the x-y axes that represent
5 the relative contributions of AOA and AOB to the total *amoA* gene or transcript pools. Notably,
6 compared to AOA, higher AOB abundance in the *amoA* gene-based DNA or cDNA pool resulted in
7 distinctly higher (disproportionately higher relative to enhanced abundance) average *amoA* gene copy
8 or transcript-specific N₂O production rates (Fig. 7a) and yields (Fig. 7b), suggesting that AOB may
9 have higher cell-specific activities in the upper estuary and thus be more active in producing N₂O than
10 AOA. Previous studies based on pure cultures of AOB and AOA strains provided evidence that AOB
11 have higher N₂O yields (0.09 to 26%) (Yoshida and Alexander, 1970; Goreau et al., 1980) than AOA
12 (0.002 to 0.09%) during ammonia oxidation (Löscher et al., 2012; Stieglmeier et al., 2014). The higher
13 N₂O yield from AOB has also been observed in soils despite a lower abundance of AOB (Hink et al.,
14 2017; Hink et al., 2018). Based on results indicated by Fig. 7, we conclude that AOB may have higher
15 relative contributions to the high N₂O production in the upper estuary where low DO, high
16 concentrations of N₂O and Δ N₂O, and high N₂O flux were observed.

17 Ammonia oxidizers are sensitive to oxygen during N₂O production (Santoro et al., 2011; Löscher
18 et al., 2012; Stieglmeier et al., 2014). Studies based on pure cultures of AOB strains *Nitrosomonas*
19 *marina* NM22 and *Nitrosococcus oceani* NC10, and AOA strain *Nitrosopumilus maritimus* showed
20 higher N₂O yields and production during nitrification by both AOA and AOB when O₂ concentrations
21 varied from aerobic to hypoxic conditions (Löscher et al., 2012). However, when O₂ concentrations
22 varied from hypoxic to anaerobic conditions (i.e. in a lower O₂ concentration range), the AOB strain
23 *Nitrospira multiformis* and AOA strains *Nitrososphaera viennensis* and *Nitrosopumilus maritimus*
24 showed that AOB had distinctly higher N₂O yields at lower oxygen conditions and, in contrast, AOA
25 had lower N₂O yields at lower oxygen concentrations (Stieglmeier et al., 2014). In addition, results from
26 the cultured AOB strain *Nitrosomonas marina* C-113a indicated increasing N₂O yields with higher cell
27 concentrations (Frame and Casciotti, 2010). This evidence supports our conclusions that the high

1 concentration of N₂O (oversaturated) may be mainly produced from strong nitrification by the high
2 abundance of AOB in the low DO conditions in the upper estuary.

3 In addition, it is possible that comammox (COMplete AMMONia OXidiser) species, newly
4 discovered in terrestrial systems (Daims et al., 2015; Santoro, 2016; Kits et al., 2017), are also involved
5 in N₂O production (Hu and He, 2017) given the similar ammonia oxidation pathway to AOB. It has
6 been further reported that the comammox *Nitrospira inopinata* has a lower N₂O yield than AOB due to
7 a lack of NO reductases and the formation of N₂O from the abiotic conversion of hydroxylamine (Kits
8 et al., 2019). However, comammox has not been widely observed in estuarine waters. Also, *nirK*-type
9 denitrifiers may contribute to N₂O production despite being much less abundant than *nirS*-
10 type denitrifiers (Huang et al., 2011; Maeda et al., 2017). Furthermore, *nirS*-type denitrifiers are more
11 likely to be capable of complete denitrification because of a higher co-occurrence of the N₂O reductase
12 gene (*nosZ*) with *nirS* than *nirK* (Graf et al., 2014). However, there is currently no direct evidence that
13 denitrification or nitrifier-denitrification is responsible for N₂O production in the PRE water column. A
14 release of N₂O into the overlying waters through denitrification was reported for PRE sediments (Tan et
15 al., 2019). Further study is needed to clarify the potential of both *nirK* and *nirS*-type denitrifiers in N₂O
16 production from the interface between sediment and water in the PRE.

17 **5 Conclusions**

18 This study explored the relative contributions of AOB and AOA in producing N₂O in the PRE by
19 combining isotopic compositions and concentrations of N₂O, distributions and transcript levels of AOB
20 and AOA *amoA* and denitrifier *nirS* genes, and incubation estimates of nitrification and N₂O production
21 rates. Our findings indicate that the high concentrations of N₂O and $\Delta N_2O_{\text{excess}}$ and the much lower
22 $\delta^{15}\text{N-N}_2\text{O}$ are primarily attributed to strong nitrification by AOB. There is also probably concurrent
23 minor denitrification in the upper estuary where AOB abundances are higher before decreasing seaward
24 along the salinity transect. Low concentrations of N₂O and $\Delta N_2O_{\text{excess}}$ and enriched ¹⁵N₂O could be
25 explained by AOA nitrification in the lower reaches of the estuary. Collectively, AOB contributed the
26 major part of N₂O production in the upper estuary, which is the major source of N₂O emitted to the
27 atmosphere in the PRE.

1 **Data availability**

2 All data can be accessed in the form of Excel spreadsheets via the corresponding author.

3 **The Supplement related to this article is available online.**

4 **Author contribution**

5 M.D. and Y.Z. conceived and designed the experiments. L.M., H.L., and X.X. performed the
6 experiments. L.M., Y.Z., H.L., and X.X. analyzed the data. L.M. and Y.Z. wrote the paper. All authors
7 contributed to the interpretation of results and critical revision.

8 **Competing interests**

9 The authors declare no conflicts of interest.

10 **Acknowledgments**

11 We thank Qing Li for measuring ammonia concentrations on board, Jian-Zhong Su and Liguu Guo for
12 measuring dissolved oxygen concentrations on board, and Tao Huang and Lifang Wang for measuring
13 nitrate and nitrite concentrations. We also thank Lei Hou for her assistance in qPCR measurements and
14 data analysis, and Mingming Chen and Huade Zhao for their assistance with the software. This research
15 was supported by the National Key Research and Development Programs (2016YFA0601400) and
16 NSFC projects (41676125, 41721005, and 41706086). We thank Kara Bogus, PhD, from Liwen Bianji,
17 Edanz Editing China (www.liwenbianji.cn/ac), for editing the English text of a draft of this manuscript.

References

- Abell, G. C. J., Revill, A. T., Smith, C., Bissett, A. P., Volkman, J. K., and Robert, S. S.: Archaeal ammonia oxidizers and *nirS*-type denitrifiers dominate sediment nitrifying and denitrifying populations in a subtropical macrotidal estuary, *ISME J.*, 4, 286–300, 2010.
- 5 Abril, G., Riou, S. A., Etcheber, H., Frankignoulle, M., de Wit, R., Middelburg, J. J.: Transient tidal time-scale, nitrogen transformations in an estuarine turbidity maximum fluid mud system (the Gironde, south-west France), *Estuar. Coast. Shelf S.* 50, 703–715, 2000.
- Allen, D., Dalal, R. C., Rennenberg, H., and Schmidt, S.: Seasonal variation in nitrous oxide and methane emissions from subtropical estuary and coastal mangrove sediments, Australia, *Plant Biol.*, 13, 126–133, 2011.
- 10 Arévalo-Martínez, D. L., Kock, A., Löscher, C. R., Schmitz, R. A., and Bange, H. W.: Massive nitrous oxide emissions from the tropical South Pacific Ocean, *Nat. Geosci.*, 8, 530–533, 2015.
- Avrahami, S., and Bohannan, J. M.: N₂O emission rates in a California meadow soil are influenced by fertilizer level, soil moisture and the community structure of ammonia-oxidizing bacteria, *Glob. Change Biol.*, 15, 643–655, 2009.
- 15 Babbin, A. R., Bianchi, D., Jayakumar, A., and Ward, B. B.: Rapid nitrous oxide cycling in the suboxic ocean, *Science*, 348, 1127–1129, 2015.
- Barnes, J., and Upstill-Goddard, R. C.: N₂O seasonal distributions and air–sea exchange in UK estuaries: Implications for the tropospheric N₂O source from European coastal waters, *J. Geophys. Res.*, 20 116, G01006, 2011. <http://dx.doi.org/10.1029/2009JG001156>.
- Bartossek, R., Nicol, G.W., Lanzen, A., Klenk, H. P., and Schleper, C.: Homologues of nitrite reductases in ammonia-oxidizing archaea: diversity and genomic context, *Environ. Microbiol.*, 12, 1075–1088, 2010.
- Bernhard, A. E., Landry, Z. C., Blevins, A., de la Torre, J. R., Giblin, A. E., and Stahl, D. A.: 25 Abundance of ammonia-oxidizing archaea and bacteria along an estuarine salinity gradient in relation to potential nitrification rates, *Appl. Environ. Microbiol.*, 76, 1285–1289, 2010.

- Braker, G., Fesefeldt, A., and Witzel, K. P.: Development of PCR primer systems for amplification of nitrite reductase genes (*nirK* and *nirS*) to detect denitrifying bacteria in environmental samples, *Appl. Environ. Microbiol.*, 64, 3769–3775, 1998.
- Bricker, S. B., Longstaff, B., Dennison, W., Jones, A., Boicourt, K., Wicks, C., and Woerner, J.: Effects of nutrient enrichment in the nation's estuaries: a decade of change, *Harmful Algae*, 8, 21–32, 2008.
- Butterbach-Bahl, K., Baggs, E. M., Dannenmann, M., Kiese, R., and Zechmeister-Boltenstern, S.: Nitrous oxide emissions from soils: how well do we understand the processes and their controls? *Philos. Trans. R. Soc. B-Biol. Sci.*, 368, <https://doi.org/10.1098/rstb.2013.0122>, 2013.
- Canfield, D. E., Glazer, A. N., and Falkowski, P. G.: The Evolution and Future of Earth's Nitrogen Cycle, *Science*, 330, 192–196, 2010.
- Cao, Y., Sun, G., Han, Y., Sun, D., and Wang, X.: Determination of nitrogen, carbon and oxygen stable isotope ratios in N₂O, CH₄, and CO₂ at natural abundance levels by mass spectrometer, *Acta Pedologica Sinica.*, 45, 249–258, 2008 (Chinese).
- Clark, I. M., Buchkina, N., Jhureea, D., Goulding, K. W. T., and Hirsch, P. R.: Impacts of nitrogen application rates on the activity and diversity of denitrifying bacteria in the Broadbalk Wheat Experiment, *Philos. Trans. R. Soc. B-Biol. Sci.*, 367, <https://doi.org/10.1098/rstb.2011.0314>, 2012.
- Coyne, M. S., Arunakumari, A., Averill, B. A., and Tiedje, J. M.: Immunological identification and distribution of dissimilatory heme cd1 and non-heme copper nitrite reductases in denitrifying bacteria, *Appl. Environ. Microbiol.*, 55, 2924–2931, 1989.
- Čuhel, J., Šimek, M., Laughlin, R. J., Bru, D., Chèneby, D., Watson, C. J., and Philippot, L.: Insights into the Effect of Soil pH on N₂O and N₂ Emissions and Denitrifier Community Size and Activity, *Appl. Environ. Microbiol.*, 76, 6, 1870–1878, 2010.
- Dai, M., Guo, X., Zhai, W., Yuan, L., Wang, B., Wang, L., Cai, P., Tang, T., and Cai, W. J.: Oxygen depletion in the upper reach of the Pearl River estuary during a winter drought, *Mar. Chem.*, 102, 159–169, 2006.

- Dai, M., Wang, L., Guo, X., Zhai, W., Li, Q., He, B., and Kao, S. J.: Nitrification and inorganic nitrogen distribution in a large perturbed river/estuarine system: the Pearl River Estuary, China, *Biogeosciences*, 5, 1545–1585, 2008.
- Dai, M., Gan, J., Han, A., Kung, H. S., and Yin, Z.: Physical dynamics and biogeochemistry of the Pearl River plume, In *Biogeochemical dynamics at major river-coastal interfaces, Linkages with Global Change*. Bianchi, T. S., Allison, M. A., and Cai, W. J. (eds). Cambridge University Press, pp. 321–351, 2014.
- Daims, H., Lebedeva, E. V., Pjevac, P., Han, P., Herbold, C., Albertsen, M., Jehmlich, N., Palatinszky, M., Vierheilig, J., and Bulaev, A.: Complete nitrification by *Nitrospira* bacteria, *Nature*, 528, 504–509, 2015.
- Damashek, J., Casciotti, K. L., and Francis, C. A.: Variable Nitrification Rates Across Environmental Gradients in Turbid, Nutrient-Rich Estuary Waters of San Francisco Bay, *Estuaries and Coasts*, 39, 1050–1071, 2016.
- Damashek, J. and Francis, C. A.: Microbial Nitrogen Cycling in Estuaries: From Genes to Ecosystem Processes, *Estuaries and Coasts*, 41, 626–660, 2018.
- De Wilde, H. P. J., and De Bie, M. J. M.: Nitrous oxide in the Schelde Estuary: production by nitrification and emission to the atmosphere, *Mar. Chem.*, 69, 203–216, 2000.
- De Bie, M. J. M., Middelburg, J. J., Starink, M., and Laanbroek, H. J.: Factors controlling nitrous oxide at the microbial community and estuarine scale, *Mar. Ecol- Prog. Ser.*, 240, 1–9, 2002.
- Frame, C. H., and Casciotti, K. L.: Biogeochemical controls and isotopic signatures of nitrous oxide production by a marine ammonia-oxidizing bacterium, *Biogeosciences*, 7, 2695–2709, 2010.
- Frame, C. H., Lau, E., Joseph Nolan IV, E., Goepfert, T. J., and Lehmann, M. F.: Acidification Enhances Hybrid N₂O Production Associated with Aquatic Ammonia-Oxidizing Microorganisms, *Front. Microbiol.*, 7, 1–23, 2017.
- Francis, C. A., Roberts, K. J., Beman, J. M., Santoro, A. E., and Oakley, B. B.: Ubiquity and diversity of ammonia-oxidizing archaea in water columns and sediments of the ocean, *P. Natl. Acad. Sci. USA*, 102, 14683–14688, 2005.

- Freing, A., Wallace, D. W. R., and Bange, H. W.: Global oceanic production of nitrous oxide, *Philos. Trans. R. Soc. B-Biol. Sci.*, 367, <https://doi.org/10.1098/rstb.2011.0360>, 2012.
- Füssel, J.: Impacts and importance of ammonia and nitrite oxidation in the marine nitrogen cycle, PhD thesis, Max Planck Institute for Microbial Ecology, Bremen, Germany, 166 pp., 2014.
- 5 Garnier, J., C̄bron, A., Tallec, G., Billen, G., Sebilo, M., and Martinez, A.: Nitrogen behaviour and nitrous oxide emission in the tidal Seine River estuary (France) as influenced by human activities in the upstream watershed. *Biogeochemistry*, 77, 305–326, 2006.
- Goreau, T. J., Kaplan, W. A., Wofsy, S. C., McElroy, M. B., Valois, F. W., and Watson, S. W.: Production of NO_2^- and N_2O by Nitrifying Bacteria at Reduced Concentrations of Oxygen, *Appl. Environ. Microbiol.*, 40, 526–532, 1980.
- 10 Graf, D. R. H., Jones, C. M., and Hallin, S.: Intergenomic comparisons highlight modularity of the denitrification pathway and underpin the importance of community structure for N_2O emissions. *PloS One* 9: e114118. <https://doi.org/10.1371/journal.pone.0114118.s008>. 2014.
- Han, A., Dai, M., Kao, S.-J., Gan, J., Li, Q., Wang, L., Zhai, W., and Wang, L.: Nutrient dynamics and biological consumption in a large continental shelf system under the influence of both a river plume and coastal upwelling, *Limnol. Oceanogr.*, 57, 486–502, 2012.
- 15 Hatzenpichler, R.: Diversity, Physiology, and Niche Differentiation of Ammonia Oxidizing Archaea, *Appl. Environ. Microbiol.*, 78, 7501–7510, 2012.
- He, B., Dai, M., Zhai, W., Guo, X., and Wang, L.: Hypoxia in the upper reaches of the Pearl River Estuary and its maintenance mechanisms: A synthesis based on multiple year observations during 2000-2008, *Mar. Chem.*, 167, 13–24, 2014.
- 20 Hink, L., Nicol, G. W., and Prosser, J. I.: Archaea produce lower yields of N_2O than bacteria during aerobic ammonia oxidation in soil, *Environ. Microbiol.*, 19, 4829–4837, 2017.
- Hink, L., Gubry-Rangin, C., Nicol, G. W., and Prosser, J. I.: The consequences of niche and physiological differentiation of archaeal and bacterial ammonia oxidisers for nitrous oxide emissions, *ISME J.*, 12, 1084–1093, 2018.
- 25

- Hou, L., Xie, X., Wan, X., Kao, S. J., Jiao, N., and Zhang, Y.: Niche differentiation of ammonia and nitrite oxidizers along a salinity gradient from the Pearl River estuary to the South China Sea, *Biogeosciences*, 15, 5169–5187, 2018.
- Huang, S., Chen, C., Yang, X., Wu, Q., and Zhang, R.: Distribution of typical denitrifying functional genes and diversity of the *nirS*-encoding bacterial community related to environmental characteristics of river sediments, *Biogeosciences*, 8, 3041–3051, 2011.
- Hu, A., Jiao, N., and Zhang, C. L.: Community structure and function of planktonic *Crenarchaeota*: changes with depth in the South China Sea, *Microb. Ecol.*, 62, 549–563, 2011.
- Hu, H. W., and He, J. Z.: Comammox—a newly discovered nitrification process in the terrestrial nitrogen cycle, *J. Soils Sediments*, 17, 2709–2717, 2017.
- Huertas, I. E., Flecha, S., Navarro, G., and Perez, F. F., de la Paz, M.: Spatio-temporal variability and controls on methane and nitrous oxide in the Guadalquivir Estuary, Southwestern Europe, *Aquat Sci.*, 80, 29, 2018.
- Ji, Q., Babbin, A. R., Jayakumar, A., Oleynik, S., and Ward, B. B.: Nitrous oxide production by nitrification and denitrification in the Eastern Tropical South Pacific oxygen minimum zone, *Geophys. Res. Lett.*, 42, 10, 755–10, 764, 2015.
- Ji, Q., Buitenhuis, E., Suntharalingam, P., Sarmiento, J. L., and Ward, B. B.: Global Nitrous Oxide Production Determined by Oxygen Sensitivity of Nitrification and Denitrification, *Global Biogeochem. Cy.*, 32, 1790–1802, 2018.
- Jones, C. M., Spor, A., Brennan, F. P., Breuil, M. C., Bru, D., Lemanceau, P., Griffiths, B., Hallin, S., and Philippot, L.: Recently identified microbial guild mediates soil N₂O sink capacity, *Nat. Clim. Chang.*, 4, 801–805, 2014.
- Jung, M. Y., Well, R., Min, D., Gieseemann, A., Park, S. J., Kim, J. G., and Rhee, S. K.: Isotopic signatures of N₂O produced by ammonia-oxidizing archaea from soils, *ISME J.*, 8, 1115–1125, 2014.
- Kester, R. A., de Boer, W., and Laanbroek, H. J.: Production of NO and N₂O by pure cultures of nitrifying and denitrifying bacteria during changes in aeration, *Appl. Environ. Microbiol.*, 63, 3872–3877, 1997.

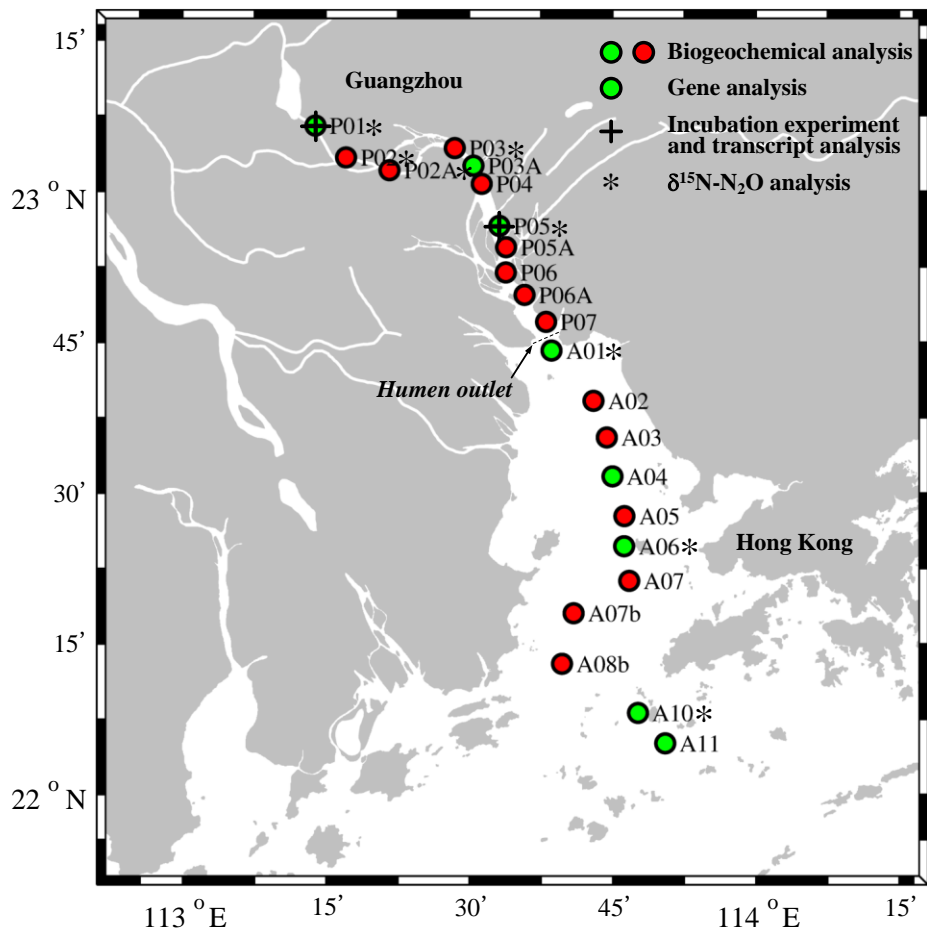
- Kim, O. S., Junier, P., Imhoff, J. F., and Witzel, K. P.: Comparative analysis of ammonia monooxygenase (*amoA*) genes in the water column and sediment–water interface of two lakes and the Baltic Sea, *FEMS Microbiol. Ecol.*, 66, 367–378, 2008.
- Kim, I. N., Lee, K., Bange, H.W., Macdonald, A. M.: Interannual variation in summer N₂O concentration in the hypoxic region of the northern gulf of Mexico, 1985–2007, *Biogeosciences*, 10, 6783–6792, 2013.
- Kits, K. D., Sedlacek, C. J., Lebedeva, E. V., Han, P., Bulaev, A., Pjevac, P., Daebeler, A., Romano, S., Albertsen, M., Stein, L. Y., Daims, H., and Wagner, M.: Kinetic analysis of a complete nitrifier reveals an oligotrophic lifestyle, *Nature*, 549, 269–272, 2017.
- 10 Kits, K. D., Jung, M.Y., Vierheilig, J., Pjevac, P., Sedlacek, C. J., Liu, S., Herbold, C., Stein, L. Y., Richter, A., Wisse, H., Brüggemann, N., Wagner, M., and Daims, H.: Low yield and abiotic origin of N₂O formed by the complete nitrifier *Nitrospira inopinata*, *Nat. Commun.*, 2019. <https://doi.org/10.1038/s41467-019-09790-x>
- Kroeze, C., Dumont, E., and Seitzinger, S.: Future trends in emissions of N₂O from rivers and estuaries, 15 *J. Integr. Environ. Sci.*, 7, 71–78, 2010.
- Laperriere, S. M., Nidzieko, N. J., Fox, R. J., Fisher, A. W., Santoro, A. E.: Observations of variable ammonia oxidation and nitrous oxide flux in a eutrophic estuary, *Estuar Coast*, 42, 33–44, 2019.
- Lin, H., Dai, M., Kao, S. J., Wang, L., Roberts, E., Yang, J., Huang, T., and He, B.: Spatiotemporal variability of nitrous oxide in a large eutrophic estuarine system: The Pearl River Estuary, China, 20 *Mar. Chem.*, 182, 14–24, 2016.
- Liu, C., Hou, L., Liu, M., Zheng, Y., Yin, G., Han, P., Dong, H., Gao, J., Gao, D., Chang, Y., Zhang, Z.: Coupling of denitrification and anaerobic ammonium oxidation with nitrification in sediments of the Yangtze Estuary: Importance and controlling factors, *Estuar. Coast. Shelf. S.*, 220, 64–72, 2019.
- 25 Löscher, C. R., Kock, A., Könneke, M., LaRoche, J., Bange, H. W., and Schmitz, R. A.: Production of oceanic nitrous oxide by ammonia-oxidizing archaea, *Biogeosciences*, 9, 2419–2429, 2012.
- Lund, M. B., Smith, J. M., and Francis, C. A.: Diversity, abundance and expression of nitrite reductase (*nirK*)-like genes in marine thaumarchaea, *ISME J.*, 6, 1966–1977, 2012.

- Maeda, K., Toyoda, S., Philippot, L., Hattori, S., Nakajima, K., Ito, Y., and Yoshida, N.: Relative Contribution of *nirK*- and *nirS*- Bacterial Denitrifiers as Well as Fungal Denitrifiers to Nitrous Oxide Production from Dairy Manure Compost, *Environ. Sci. Technol.*, 51, 14083–14091, 2017.
- Magalhães, C., Bano, N., Wiebe, W. J., Bordalo, A. A., and Hollibaugh, J. T.: Dynamics of nitrous oxide reductase genes (*nosZ*) in intertidal rocky biofilms and sediments of the Douro River Estuary (Portugal), and their relation to N-biogeochemistry, *Microb. Ecol.*, 55, 259–269, 2008.
- Mandernack, K. W., Mills, C. T., Johnson, C. A., Rahn, T., and Kinney, C.: The $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ values of N_2O produced during the co-oxidation of ammonia by methanotrophic bacteria, *Chem. Geol.*, 267, 96–107, 2009.
- 10 Mincer, T. J., Church, M. J., Taylor, L. T., Preston, C., Karl, D. M., and DeLong, E. F.: Quantitative distribution of presumptive archaeal and bacterial nitrifiers in Monterey Bay and the North Pacific Subtropical Gyre, *Environ. Microbiol.*, 9, 1162–1175, 2007.
- Mohn, J., Wolf, B., Toyoda, S., Lin, C. T., Liang, M. C., Brüggemann, N., Wissel, H., Dyckmans, A. E. S, J., Szwec, L., Ostrom, N. E., Casciotti, K. L., Forbes, M., Giesemann, A., R., Doucett, R. R.,
15 Well, Yarnes, C. T., Ridley, A. R., Kaiser, J., and Yoshida, N.: Interlaboratory assessment of nitrous oxide isotopomer analysis by isotope ratio mass spectrometry and laser spectroscopy: current status and perspectives, *Rapid Commun Mass Spectrom.* 28, 1995–2007, 2014.
- Monteiro, M., S éneca, J., and Magalhães, C.: The History of Aerobic Ammonia Oxidizers: from the First Discoveries to Today, *J. Microbiol.*, 52, 537–547, 2014.
- 20 Moore, W. S., Sarmiento, J. L., and Key, R. M.: Tracing the Amazon component of surface Atlantic water using ^{228}Ra , salinity and silica, *J. Geophys. Res.*, 91, 2574–2580, 1986.
- Mortazavi, B., Iverson, R. L., Huang, W., Graham Lewis, F., Caffrey, J. M.: Nitrogen budget of Apalachicola Bay, a bar-built estuary in the northeastern Gulf of Mexico, *Mar. Ecol.-Prog. Ser.*, 195, 1–14, 2000.
- 25 Naqvi, S. W. A., Bange, H. W., Far ás, L., Monteiro, P. M. S., Scranton, M. I., and Zhang, J.: Marine hypoxia/anoxia as a source of CH_4 and N_2O , *Biogeosciences*, 7, 2159–2190, 2010.
- Nevison, C. D., Lueker, T. J., and Weiss, R. F.: Quantifying the nitrous oxide source from coastal upwelling, *Glob. Biogeochem. Cycle*, 18, 1–17, 2004.

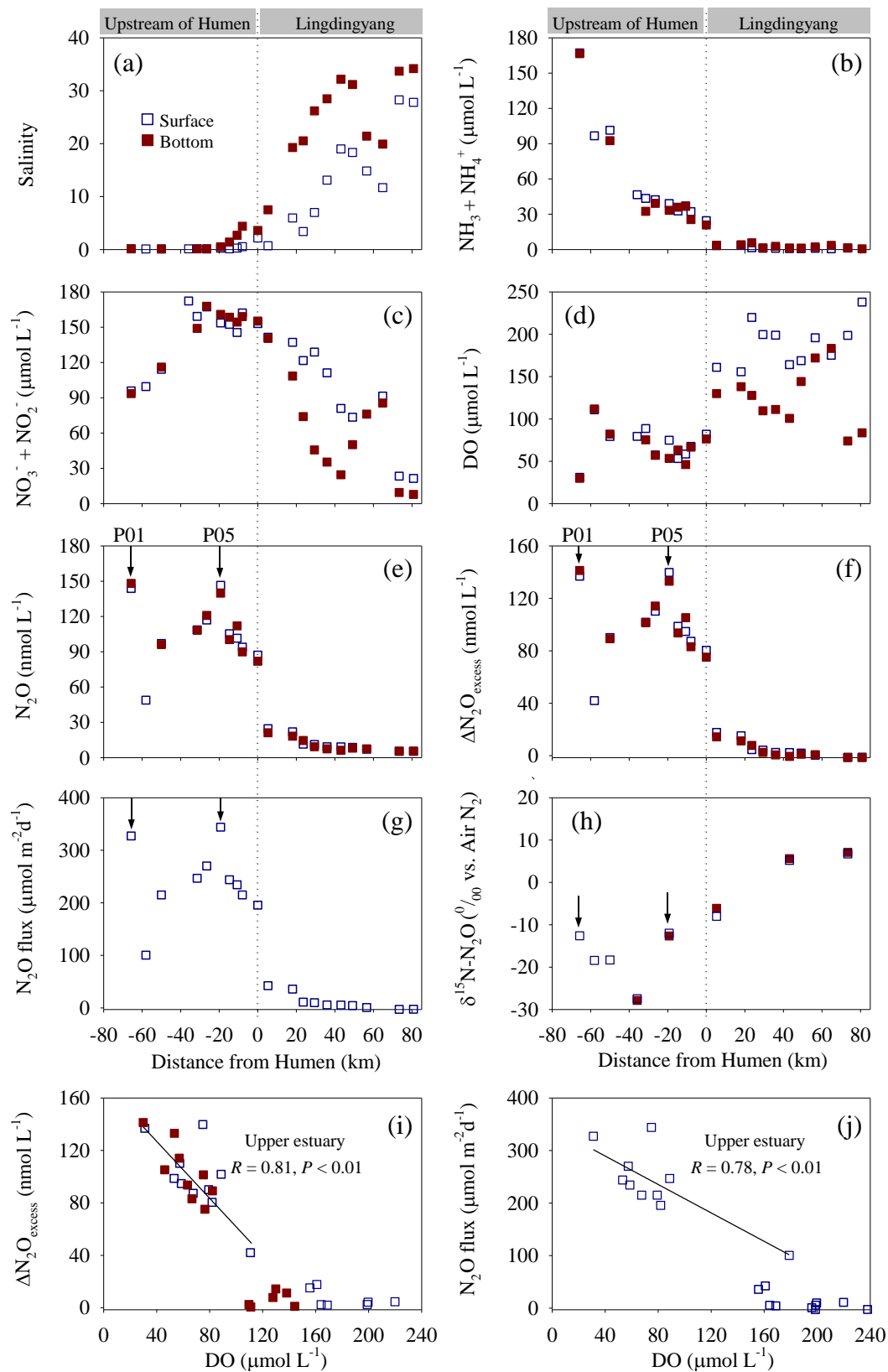
- Ni, B. J., Peng, L., Law, Y., Guo, J., and Yuan, Z.: Modeling of Nitrous Oxide Production by Autotrophic Ammonia-Oxidizing Bacteria with Multiple Production Pathways, *Environ. Sci. Technol.*, 48, 3916–3924, 2014.
- Pai, S. C., Tsau, Y. J., and Yang, T. I.: pH and buffering capacity problems involved in the determination of ammonia in saline water using the indophenol blue spectrophotometric method, *Analytica Chimica Acta*, 434, 209–216, 2001.
- Penn, J., Weber, T., and Deutsch, C.: Microbial functional diversity alters the structure and sensitivity of oxygen deficient zones, *J. Geophys. Res. Lett.*, 43, 9773–9780, 2016.
- Pike, N.: Using false discovery rates for multiple comparisons in ecology and evolution, *Methods in Ecology and Evolution*, 2, 278–282, 2011.
- Rabalais, N. N., D'iaz, R. J., Levin, L. A., Turner, R. E., Gilbert, D., and Zhang, J.: Dynamics and distribution of natural and human-caused hypoxia, *Biogeosciences*, 7, 585–619, 2010.
- Rajkumar, A. N., Barnes, J., Ramesh, R., Purvaja, R., and Upstill-Goddard, R. C.: Methane and nitrous oxide fluxes in the polluted Adyar River and estuary, SE India., *Mar. Pollut. Bull.*, 56, 2043–2051, 2008.
- Ravishankara, A. R., Daniel, J. S., and Portmann, R. W.: Nitrous Oxide (N₂O): The Dominant Ozone-Depleting Substance Emitted in the 21st Century, *Science*, 326, 123–125, 2009.
- Raymond, P. A., and Cole, J. J.: Gas exchange in rivers and estuary: choosing a gas transfer velocity, *Estuaries Coasts*, 24, 312–317, 2001.
- Santoro, A. E., Buchwald, C., McIlvin, M. R., and Casciotti K. L.: Isotopic Signature of N₂O Produced by Marine Ammonia-Oxidizing Archaea, *Science*, 333, 1282–1285, 2011.
- Santoro, A. E.: The do-it-all nitrifier, *Science*, 351, 342–343, 2016.
- Seitzinger, S. P., and Kroeze, C.: Global distribution of nitrous oxide production and N inputs in freshwater and coastal marine ecosystems, *Glob. Biogeochem. Cycle*, 12, 93–113, 1998.
- Shaw, L. J., Nicol, G. W., Smith, Z., Fear, J., Prosser, J. I., Baggs, E. M.: *Nitrosospira* spp. can produce nitrous oxide via a nitrifier denitrification pathway, *Environ. Microbiol.*, 8, 214–222, 2006.
- Stein, L. Y.: Surveying N₂O-Producing Pathways in Bacteria, *Methods Enzymol.*, 486, 131–152, 2011.

- Stieglmeier, M., Mooshammer, M., Kitzler, B., Wanek, W., Zechmeister-Boltenstern, S., Richter, A., and Schleper, C.: Aerobic nitrous oxide production through N-nitrosating hybrid formation in ammonia-oxidizing archaea, *ISME J.*, 8, 1135–1146, 2014.
- 5 Sutka, R. L., Ostrom, N. E., Ostrom, P. H., Gandhi, H., and Breznak, J. A.: Nitrogen isotopomer site preference of N₂O produced by *Nitrosomonas europaea* and *Methylococcus capsulatus* Bath, *Rapid Commun. Mass Spectrom.*, 17, 738–745, 2003.
- Sutka, R. L., Ostrom, N. E., Ostrom, P. H., Breznak, J. A., Gandhi, H., Pitt, A. J., and Li, F.: Distinguishing Nitrous Oxide Production from Nitrification and Denitrification on the Basis of Isotopomer Abundances, *Appl. Environ. Microbiol.*, 72, 638–644, 2006.
- 10 Syakila, A., and Kroeze, C.: The global nitrogen budget revisited, *J. Greenhouse Gas Meas. Manag.*, 1, 17–26, 2011.
- Tan, E., Zou, W., Jiang, X., Wan, X., Hsu, T. C., Zheng, Z., Chen, L., Xu, M., Dai, M., Kao, S.: Organic matter decomposition sustains sedimentary nitrogen loss in the Pearl River Estuary, China, *Sci. Total. Environ.*, 648, 508–517, 2019
- 15 Ter Braak, C. J. F. Canonical correspondence analysis: a new eigenvector technique for multivariate direct gradient analysis, *Ecology*, 67, 1167–1179, 1986.
- Toyoda, S., Mutoke, H., Yamagishi, H., Yoshida, N., Tanji, Y.: Fractionation of N₂O isotopomers during production by denitrifier, *Soil. Biol. Biochem.*, 37, 1535–1545, 2005.
- Toyoda, S., Yoshida, N., and Koba, K.: Isotopocule analysis of biologically produced nitrous oxide in various environments, *Mass Spectrom. Rev.*, 36, 135–160, 2017.
- 20 Treusch, A. H., Leininger, S., Kletzin, A., Schuster, S. C., Klenk, H., and Schleper, C.: Novel genes for nitrite reductase and Amo-related proteins indicate a role of uncultivated mesophilic crenarchaeota in nitrogen cycling, *Environ. Microbiol.*, 7, 1985–1995, 2005.
- Usui, T., Koike, I., and Ogura, N.: N₂O production, nitrification and denitrification in an estuarine sediment, *Estuar. Coast. Shelf Sci.*, 52, 769–781, 2001.
- 25 Voss, M., Bange, H. W., Dippner, J. W., Middelburg, J. J., Montoya, J. P., and Ward, B.: The marine nitrogen cycle: recent discoveries, uncertainties and the potential relevance of climate change, *Philos. Trans. R. Soc. B-Biol. Sci.*, 368, 20130121, 2013.

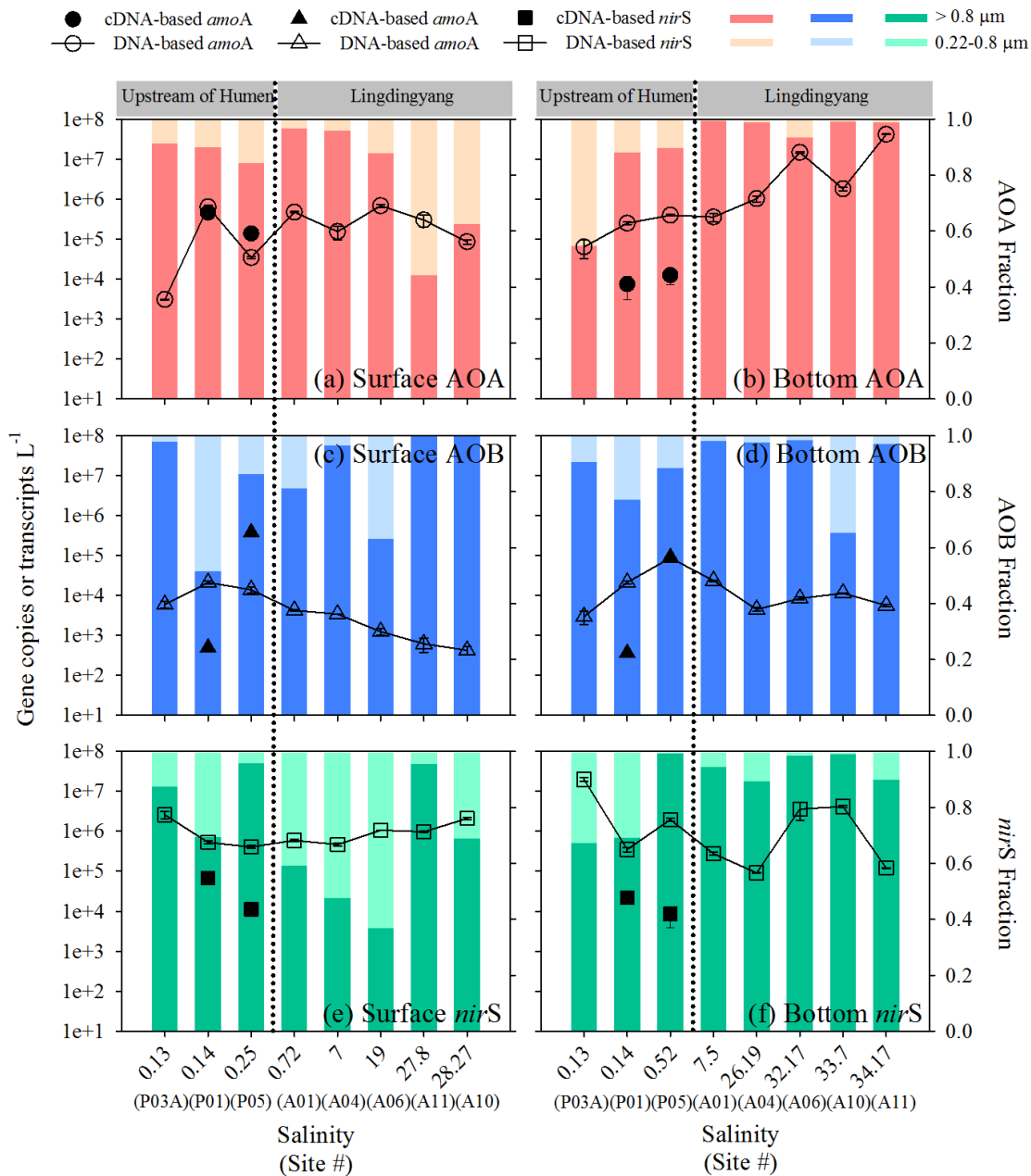
- Wang, J., Kan, J., Qian, G., Chen, J., Xia, Z., Zhang, X., Liu, H., and Sun, J.: Denitrification and anammox: Understanding nitrogen loss from Yangtze Estuary to the east China sea (ECS), *Environ. Pollut.*, 2019. <https://doi.org/10.1016/j.envpol.2019.06.025>.
- Wanninkhof, R. I. K.: Relationship between wind speed and gas exchange over the ocean, *J. Geophys. Res.-Oceans*, 97, 7373–7382, 1992.
- 5 Ward, B. B. Nitrification in marine systems, In: Capone, D. G., Bronk, D. A., Mulholl, M. R., and Carpenter, E. J. (ed), *Nitrogen in the Marine Environment (2nd Edition)* [M]. Burlington: Academic Press, 199–261, 2008.
- Weiss, R. F., and Price, B. A.: Nitrous oxide solubility in water and seawater, *Mar. Chem.*, 8, 347–359, 10 1980.
- Yamagishi, H., Westley, M. B., Popp, B. N., Toyoda, S., Yoshida, N., Watanabe, S., Koba, K., and Yamanak, Y.: Role of nitrification and denitrification on the nitrous oxide cycle in the eastern tropical North Pacific and Gulf of California, *J. Geophys. Res.*, 112, G02015, doi:10.1029/2006JG000227, 2007.
- 15 Yoshida, N.: ¹⁵N-depleted N₂O as a product of nitrification, *Nature*, 335, 528–529, 1988.
- Yoshida, T., and Alexander, M.: Nitrous Oxide Formation by *Nitrosomonas europaea* and Heterotrophic Microorganisms, *Soil Sci. Soc. Amer. Proc.*, 34, 880–882, 1970.
- Yu, R., Kampschreur, M. J., Mark van Loosdrecht, C. M., and Chandran, K.: Mechanisms and Specific Directionality of Autotrophic Nitrous Oxide and Nitric Oxide Generation during 20 Transient Anoxia, *Environ. Sci. Technol.*, 44, 1313–1319, 2010.
- Zhang, Y., Xie, X., Jiao, N., Hsiao, S. S. Y., and Kao, S. J.: Diversity and distribution of *amoA*-type nitrifying and *nirS*-type denitrifying microbial communities in the Yangtze River estuary, *Biogeosciences*, 11, 2131–2145, 2014.
- Zhu, R., Liu, Y., Li, X., Sun, J., Xu, H. and Sun, L.: Stable isotope natural abundance of nitrous oxide emitted from Antarctic tundra soils: effects of sea animal excrement depositions, *Rapid Commun. Mass Spectrom.*, 22, 3570–3578, 2008.
- 25



1
 2 **Figure 1:** Map of the PRE showing the sampling sites. Biogeochemical analyses were performed on
 3 samples from all sites (green and red circles). The green circles indicate sites where genes were
 4 analyzed. The black crosses indicate in situ incubation experiment sites (P01 and P05). The black
 5 asterisks indicate sites where the isotopic composition of N_2O was analyzed.

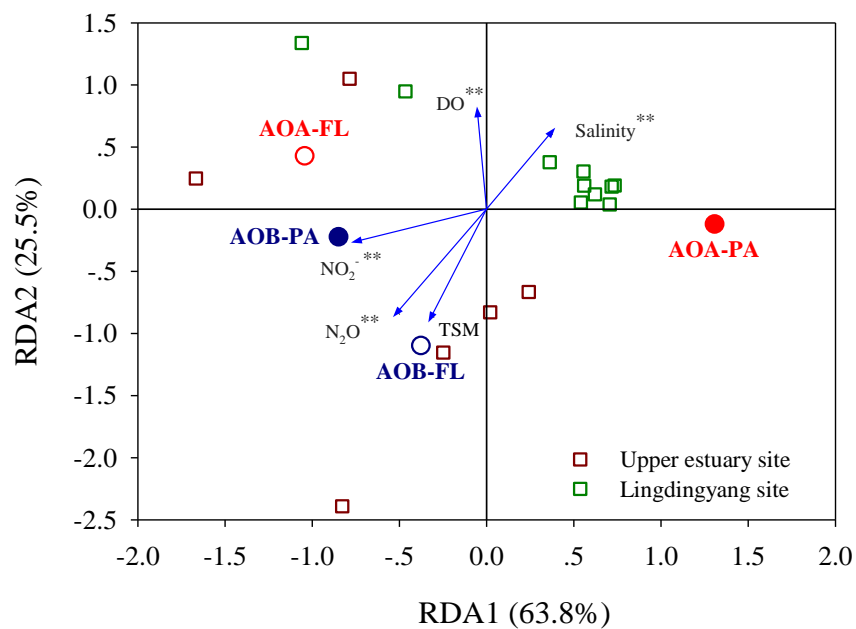


1 **Figure 2:** Distribution of biogeochemical factors along the PRE transect. (a) Salinity, (b) $\text{NH}_3+\text{NH}_4^+$, (c)
2 $\text{NO}_2^- + \text{NO}_3^-$, (d) DO, (e) N_2O , and (f) $\Delta\text{N}_2\text{O}_{\text{excess}}$ concentrations, (g) N_2O flux, (h) $\delta^{15}\text{N}-\text{N}_2\text{O}$, (i)
3 $\Delta\text{N}_2\text{O}_{\text{excess}}$ vs. DO, and (j) N_2O flux vs. DO. The dashed lines show the division of the transect into the
4 northern (upstream of the Humen outlet) and southern (Lingdingyang) areas. The arrows indicate the
5 sites where in situ incubation experiments were performed.



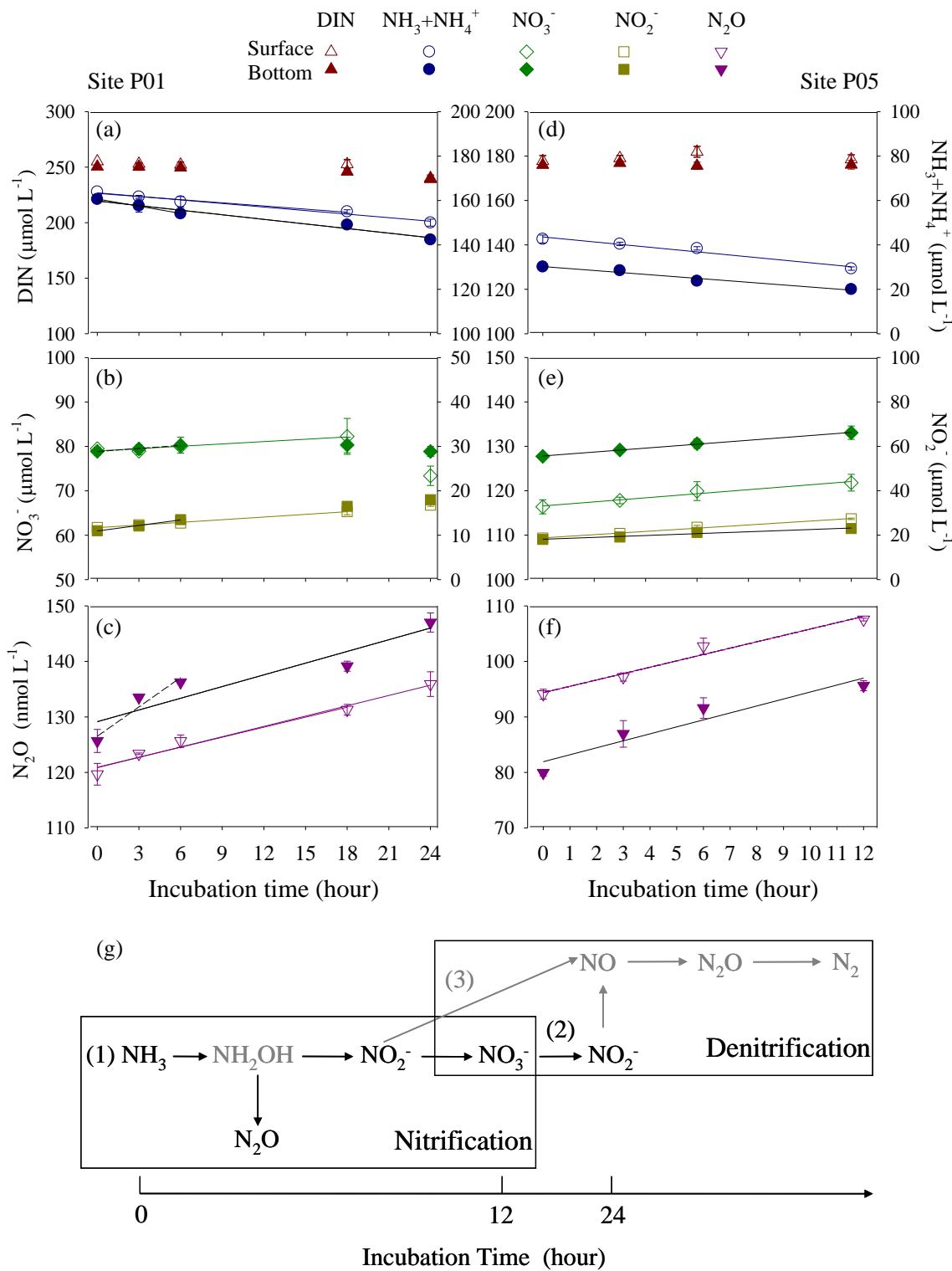
1
 2 **Figure 3:** Abundance distribution of AOA and AOB *amoA* and bacterial *nirS* along the salinity
 3 gradient in the PRE. Abundances of AOA *amoA* genes (open circles) and PA transcripts (closed circles)
 4 and the relative abundances of PA and FL AOA *amoA* genes in (a) surface and (b) bottom waters.
 5 Abundances of AOB *amoA* genes (open triangles) and PA transcripts (closed triangles) and the relative
 6 abundances of PA and FL AOB *amoA* genes in (c) surface and (d) bottom waters. Abundances of

- 1 bacterial *nirS* genes (open squares) and PA transcripts (closed squares) and the relative abundances of
- 2 PA and FL *nirS* genes in (e) surface and (f) bottom waters. The dashed lines indicate the division into
- 3 the northern (upstream of the Humen outlet) and southern (Lingdingyang) areas.

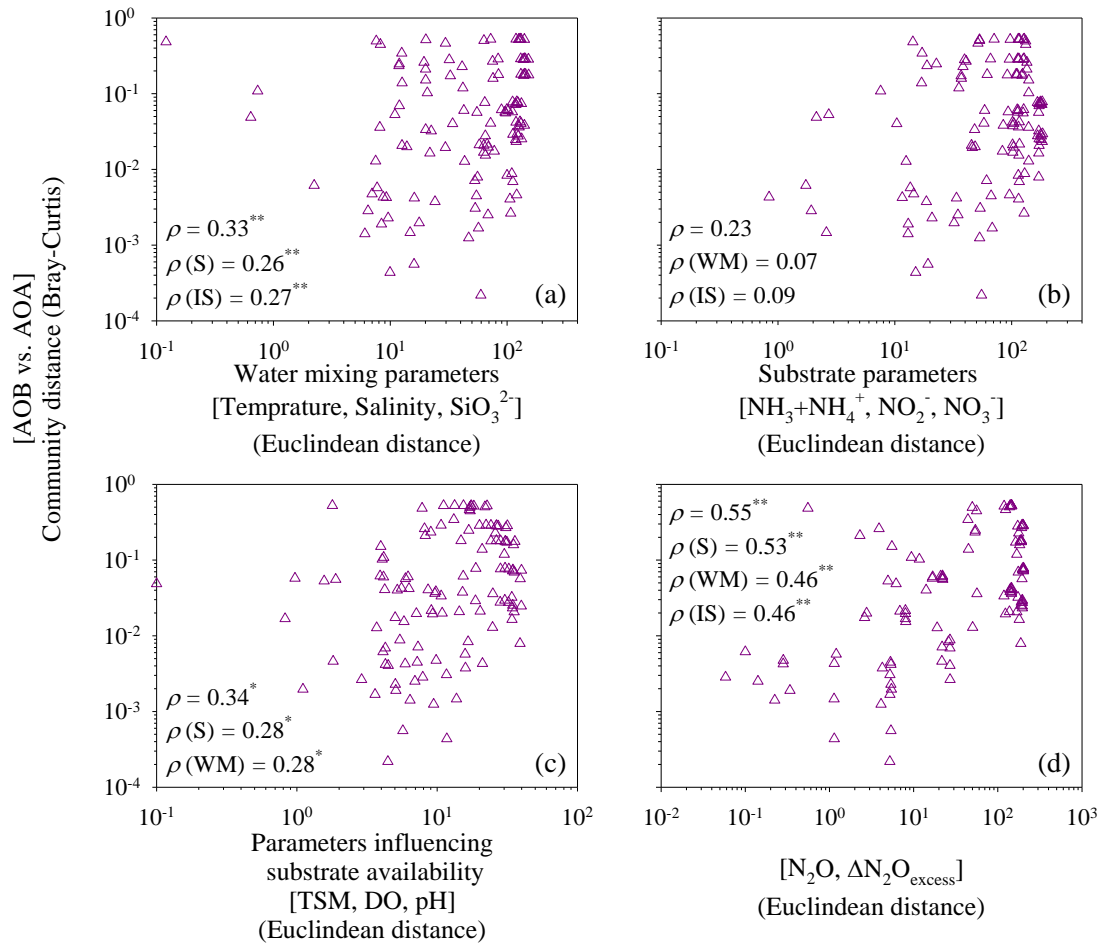


1

2 **Figure 4:** RDA of the relative abundance of AOA *amoA* and AOB *amoA* under biogeochemical
 3 constraints. Each square represents an individual sample. Vectors represent environmental variables. **P*
 4 < 0.05, ***P* < 0.01 (Monte Carlo permutation test).



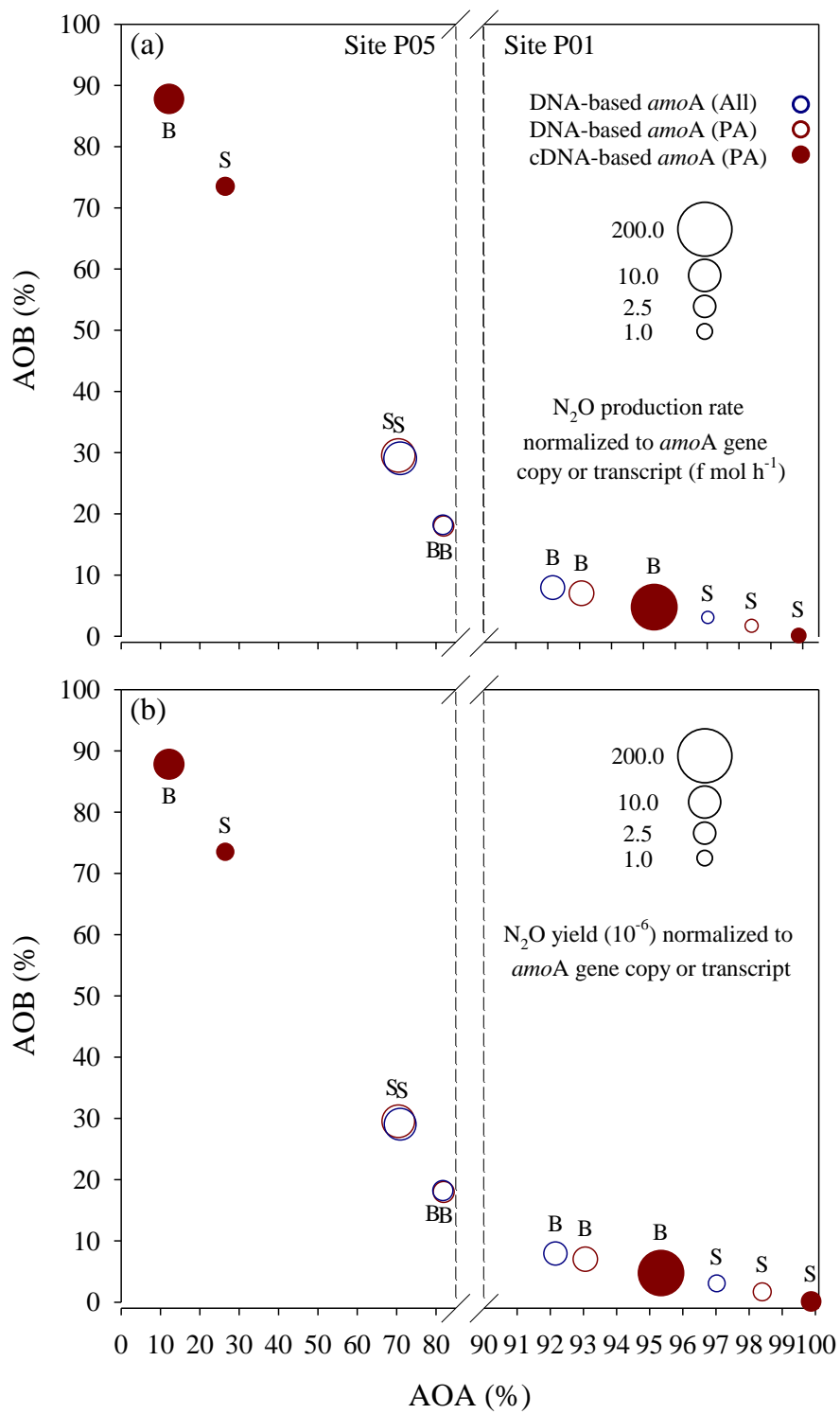
1 **Figure 5:** Variations in nitrogen compounds and N₂O concentrations at sites P01 and P05 during the
2 incubation experiments in surface (open symbols) and bottom (closed symbols) waters. (a, d) Total DIN
3 (brown triangles) and NH₃+NH₄⁺ (blue circles), (b, e) NO₃⁻ (green diamonds) and NO₂⁻ (dark yellow
4 squares), (c, f) N₂O (purple inverted triangles). Linear regressions depend on whether variations in DIN
5 concentration against time retain “mass balance” in a closed incubation system. The linear regressions
6 of ammonia were used to estimate ammonia oxidation rates in (a) P01 over 18 and 24 h (surface water,
7 blue lines) and 6 and 24 h (bottom water, black lines), and (d) P05 over 12 h (surface, blue line; bottom,
8 black line). The linear regressions of nitrate estimated nitrite oxidation rates in (b) P01 over 18 h
9 (surface water, green line) and 6 h (bottom water, black line), and (e) P05 after 12 h (surface, green line;
10 bottom, black line). The nitrite linear regressions after 18 h (surface water, dark yellow line) and 6 h
11 (bottom water, black line) in P01 and 12 h (surface, dark yellow line; bottom, black line) in P05 are also
12 shown, but do not indicate oxidation rates. The N₂O linear regressions were used to estimate N₂O
13 production rates in (c) P01 after 18 and 24 h (surface water, purple lines) and 6 and 24 h (bottom water,
14 black lines; dashed line, no statistical significance test), and (f) P05 after 12 h (surface, purple line;
15 bottom, black line). All regression equations, R^2 , and P values are shown in Table 2. (g) A diagram
16 showing transformations of nitrogen compounds and N₂O production during incubation experiments.
17 Nitrification (1) occurred during the entire P01 and P05 incubations and denitrification (2 and/or 3) may
18 be present in the end phase of the P01 incubation. The gray arrows indicate the pathways of nitrogen
19 loss unanalyzed here, and the gray compounds indicate the unmeasured nitrogen compound.



1

2 **Figure 6:** Correlations between the relative abundance of AOB versus AOA and (a) water mixing
 3 parameters (temperature, salinity, and silicate), (b) substrate parameters (ammonia/ammonium, nitrite,
 4 and nitrate), (c) parameters influencing substrate availability (TSM, DO, and pH), or (d) N_2O
 5 parameters (N_2O and $\Delta\text{N}_2\text{O}$). The ammonia oxidizers matrix was calculated according to the relative
 6 AOA and AOB abundances. Dissimilarity matrices of the relative abundance of AOB *amoA* and AOA
 7 *amoA* were based on Bray-Curtis distances and environmental factors were based on Euclidean
 8 distances between samples. Standard and partial Mantel tests were run to measure the correlation
 9 between two matrices. Spearman correlation coefficient (ρ) values are shown for standard (first value)

1 and partial Mantel (second, third, and fourth) tests. The P values were calculated using the distribution
2 of the Mantel test statistics estimated from 999 permutations. * $P < 0.05$; ** $P < 0.01$.



1 **Figure 7:** N₂O (a) production rates and (b) yields normalized to total *amoA* gene copy or transcript
2 numbers of AOA and AOB in a given sample. They are presented along the x-y axes that represent the
3 relative contributions of AOA and AOB to the total *amoA* gene or transcript pools. S, surface; B,
4 bottom.

1 **Table 1** Rho (ρ) values for the relationships between nitrifier and denitrifier gene abundances and biogeochemical parameters in the PRE.

Biogeochemical parameters	PA + FL			PA ($> 0.8 \mu\text{m}$)			FL ($0.22\text{--}0.8 \mu\text{m}$)		
	AOA- <i>amoA</i> (n = 16)	AOB- <i>amoA</i> (n = 16)	<i>nirS</i> (n = 16)	AOA- <i>amoA</i> (n = 16)	AOB- <i>amoA</i> (n = 14)	<i>nirS</i> (n = 16)	AOA- <i>amoA</i> (n = 16)	AOB- <i>amoA</i> (n = 16)	<i>nirS</i> (n = 16)
Temperature	-0.694*	0.359	0.085	-0.676*	0.303	0.165	-0.438	0.358	0.229
Salinity	0.644*	-0.339	-0.018	0.604*	-0.270	-0.047	0.403	-0.351	-0.356
SiO ₃ ⁻	-0.541*	0.559*	0.206	-0.497	0.503*	0.282	-0.350	0.481	0.238
TSM	-0.109	0.668*	0.047	-0.097	0.612*	0.194	0.191	0.565*	-0.071
pH	0.381	-0.656*	0.157	0.316	-0.615*	0.088	0.377	-0.605*	-0.059
DO	-0.074	-0.771**	-0.026	-0.121	-0.729**	-0.144	0.009	-0.697*	0.218
NH ₃ /NH ₄ ⁺	-0.482	0.646*	0.068	-0.482	0.571*	0.196	-0.325	0.587*	0.000
NO ₃ ⁻	-0.485	0.359	-0.138	-0.444	0.353	-0.112	-0.588*	0.213	0.115
NO ₂ ⁻	-0.588*	0.447	0.126	-0.556*	0.356	0.212	-0.421	0.288	0.265
N ₂ O	-0.421	0.641*	-0.194	-0.356	0.606*	-0.121	-0.385	0.490	0.047
$\Delta\text{N}_2\text{O}_{\text{excess}}$	-0.527*	0.559*	-0.160	-0.480	0.517*	-0.081	-0.369	0.504	0.096
N ₂ O flux ^a	-0.190 (n = 8)	1.000** (n = 8)	-0.524 (n = 8)	-0.143 (n = 8)	1.000** (n = 8)	-0.310 (n = 8)	-0.571 (n = 8)	0.657 (n = 6)	-0.524 (n = 8)

2 ^aSurface data; * False discovery rate (FDR)-adjusted $P < 0.05$; ** FDR-adjusted $P < 0.01$.

1 **Table 2** Linear regressions of ammonia, nitrite, nitrate, and N₂O concentrations against time and N₂O yields during incubation experiments.

Site_Layer	Time (hour)	$\Delta(\text{NH}_3+\text{NH}_4^+)$ ($\mu\text{mol L}^{-1} \text{h}^{-1}$)			ΔNO_2^- ($\mu\text{mol L}^{-1} \text{h}^{-1}$)			ΔNO_3^- ($\mu\text{mol L}^{-1} \text{h}^{-1}$)			$\Delta\text{N}_2\text{O}$ ($\text{nmol L}^{-1} \text{h}^{-1}$)			N ₂ O yield (%)
		Equation	R^2	Rate ^a	Equation	R^2	Rate ^a	Equation	R^2	Rate ^a	Equation	R^2	Rate ^a	
P01_S	18	$y = -0.47x + 163.20$	0.96*	0.47	$y = 0.20x + 11.69$	1.00**	0.20	$y = 0.18x + 78.98$	0.90*	0.18	$y = 0.60x + 120.93$	0.96*	0.60 ^b	0.26 ^b
	24	$y = -0.53x + 163.44$	0.98**	0.53	–	–	–	–	–	–	$y = 0.62x + 120.85$	0.98**	0.62	– ^c
P01_B	6	$y = -1.08x + 160.65$	1.00*	1.08	$y = 0.42x + 10.95$	1.00*	0.42	$y = 0.23x + 78.84$	0.98	0.23	$y = 1.61x + 127.04$	0.98	1.61 ^b	0.30 ^b
	24	$y = -0.69x + 159.76$	0.96**	0.69	–	–	–	–	–	–	$y = 0.70x + 129.14$	0.86*	0.70	– ^c
P05_S	12	$y = -1.12x + 43.58$	0.96*	1.12	$y = 0.73x + 18.78$	1.00**	0.73	$y = 0.46x + 116.58$	0.98**	0.46	$y = 1.15x + 79.79$	0.98**	1.15 ^b	0.21 ^b
P05_B	12	$y = -0.89x + 30.25$	0.96*	0.89	$y = 0.42x + 18.17$	0.96*	0.42	$y = 0.44x + 127.83$	1.00**	0.44	$y = 1.41x + 81.57$	0.96*	1.41 ^b	0.32 ^b

2 ^aThese rates are net rates since $\Delta(\text{NH}_3+\text{NH}_4^+)$ is the net consumption and ΔNO_2^- , ΔNO_3^- , and $\Delta\text{N}_2\text{O}$ is the net production during incubation.

3 ^bThese rates and yields (when only nitrification occurred) were used to calculate the average *amoA* gene copy-specific N₂O production rates and
4 N₂O yields in Figure 7.

5 ^cNo estimation of N₂O yield was made due to nitrification and denitrification may occur concurrently and DIN was not in balance.

6 * $P < 0.05$; ** $P < 0.01$.

7 –No regression analysis or no estimation made due to DIN was not in balance.