

**Tracing terrestrial versus marine sources of dissolved organic carbon in a coastal bay
using stable carbon isotopes**

Shin-Ah Lee^a, Tae-Hoon Kim^b, and Guebuem Kim^{a,*}

^aSchool of Earth and Environmental Sciences/Research Institute of Oceanography, Seoul
National University, Seoul 08826, Republic of Korea

^bDepartment of Earth and Marine Sciences, Jeju National University, Jeju, 63243, Republic of
Korea

*Corresponding author at: School of Earth and Environmental Sciences, Seoul National
University, Seoul 08826, Korea

E-mail address: gkim@snu.ac.kr (G.Kim)

Abstract

The sources of dissolved organic matter (DOM) in coastal waters are diverse, and they play different roles in the biogeochemistry and ecosystems of the ocean. In this study, we measured dissolved organic carbon (DOC) and nitrogen (DON), $\delta^{13}\text{C}$ -DOC, and fluorescent dissolved organic matter (FDOM) in coastal bay waters surrounded by heavily industrialized cities (Masan Bay, Korea) to determine the different DOM sources in this region. The surface seawater samples were collected in two sampling campaigns (Aug. 2011 and Aug. 2016). The salinities were in the range of 10–21 in 2011 and 25–32 in 2016. In 2011, excess DOC was observed for higher-salinity (16–21) waters, indicating that the excess source inputs were mainly from marine autochthonous production according to the $\delta^{13}\text{C}$ -DOC values of -23.7‰ to -20.6‰ , the higher concentrations of protein-like FDOM, and the lower DOC/DON (C/N) ratios (8–15). By contrast, the high DOC waters observed in high-salinity waters in 2016 were characterized by low FDOM, more depleted $\delta^{13}\text{C}$ values of -28.8‰ to -21.1‰ , and high C/N ratios (13–45), suggesting that the excess DOC source is from terrestrial C3 plants by direct land-seawater interactions. Our results show that multiple DOM tracers such as $\delta^{13}\text{C}$ -DOC, FDOM, and C/N ratios are powerful for discriminating the complicated sources of DOM occurring in coastal waters.

1. Introduction

Dissolved organic matter (DOM) plays an important role in biogeochemical cycles (e.g., de-oxygenation, acidification, photochemistry) and ecosystems of the ocean (Hansell and Carlson, 2002). DOM composition depends on its parent organic matter and subsequent biogeochemical processes. DOM in coastal waters originates from various sources including (1) *in situ* production by primary production, exudation of aquatic plants, and their degradation (Markager et al., 2011; Carlson and Hansell, 2015); (2) terrestrial sources by the degradation of soil and terrestrial plant matter (Opsahl and Benner, 1997; Bauer and Bianchi, 2011); and (3) anthropogenic sources (Griffith and Raymond, 2011).

Depending on the origin and composition of DOM, its behavior and cycling are different: a labile fraction of DOM is decomposed rapidly through microbially or photochemically mediated processes, whereas refractory DOM is resistant to degradation and can persist in the ocean for millennia. In the coastal ocean, organic matter from terrestrial plant litter or soils appears to be more refractory (Cauwet, 2002) and thus often behaves conservatively. In addition, refractory DOM is produced in the ocean by bacterial transformation of labile DOM by reshaping its composition (Tremblay and Benner, 2006; Jiao et al., 2010). However, it is still very difficult to determine the sources and characteristics of DOM in coastal waters.

The stable carbon isotopic composition of dissolved organic carbon ($\delta^{13}\text{C}$ -DOC) has been used to distinguish different sources. In general, $\delta^{13}\text{C}$ values derived from C3 and C4 land plants are in the range of -23‰ – -34‰ and -9‰ – -17‰ (Deines, 1980), respectively, while those derived from **marine phytoplankton** range from -18 to -22‰ (Kelley et al., 1998; Coffin

and Cifuentes, 1999). In addition, the optically active fraction of DOM known as fluorescent DOM (FDOM) have been successfully used for characterizing DOM (Coble et al., 1990; Coble, 1996). Fluorescence excitation-emission matrices and parallel factor analysis (EEM-PARAFAC) technique has been applied to trace the source of humic-like versus protein-like DOM in coastal waters and estuaries (Chen et al., 2004; Jaffé et al., 2004; Murphy et al., 2008; Huang and Chen, 2009). DOC/DON ratios are often used to differentiate allochthonous versus autochthonous sources (Thornton and McManus, 1994; Andrews et al., 1998; McCallister et al., 2006). The C/N ratios of terrestrial organic carbon are usually higher than 12, while those of marine organic carbon from phytoplankton are almost constant ranging from 6 to 8 (Milliman et al., 1984; Lobbes et al., 2000). Thus, multiple DOM tracers are more powerful for discriminating DOM sources in coastal waters where various sources are present (Faganeli et al., 1988; Wang et al., 2004; Osburn and Stedmon, 2011; Osburn et al., 2011; Cawley et al., 2012; Pradhan et al., 2014; Ya et al., 2015; Lee and Kim, 2018).

Masan bay is surrounded by cities with thousands of industrial plants and a population of 1.1 million. In association with large anthropogenic nutrient loading, this area has been recognized as a highly eutrophic embayment (Lee and Min, 1990; Yoo, 1991; Hong et al., 2010). The development of red tides and hypoxic water mass in the bottom layer of the bay has occurred annually in spring and summer (Lee et al., 2009). In addition, there are point sources from sewage treatment plants (STPs), which manage domestic and industrial wastewater from Masan and Changwon cities. Therefore, in this study, we attempted to use $\delta^{13}\text{C}$ -DOC, FDOM, and DOC/DON ratios to differentiate various sources and characteristics of DOM in Masan bay waters, which may include many different types of terrestrial/anthropogenic and marine sources of DOM.

2. Materials and methods

2.1 Study site

Masan Bay is located on the southeast coast of Korea with an area of approximately 80 km² (Fig. 1). The annual precipitation is approximately 1500 mm, and most of the precipitations occurs in the summer monsoon season. The amount of freshwater discharge into this bay is approximately $2.5 \times 10^8 \text{ m}^3 \text{ yr}^{-1}$ with significant seasonal variation. The tide is semi-diurnal, showing a maximum tidal amplitude of ~1.9 m (average = 1.3 m) during the sampling period. Due to topographic conditions, the current is very weak ($2\text{--}3 \text{ cm s}^{-1}$), and the residence times of water in the inner bay and in the entire bay are approximately 54 and 23 days, respectively (Lee et al., 2009). In the middle of the bay, an artificial island was constructed in 2015–2016 (Fig. 1) with an area of 0.64 km². **There is no marsh or wetland habitat.** The artificial island may have resulted in changes in water currents, residence times, and biogeochemical conditions.

2.2 Sampling

Sampling was conducted in August 2011 and August 2016 in Masan Bay. Water samples were collected from the surface at 17 sites in 2011 and 10 sites in 2016, from the inner to the outer bay. The bay receives a large amount of freshwater discharge from the northernmost part of the region. **The averages of surface water temperature were $30.4 \pm 2.3^\circ\text{C}$ in 2011 and $26.5 \pm 0.7^\circ\text{C}$ in 2016.** All water samples were filtered through pre-combusted GF/F filters. Samples for FDOM analysis were stored at 4°C in pre-combusted amber vials. Samples for DOC, total dissolved nitrogen (TDN), and $\delta^{13}\text{C}$ -DOC analysis were stored in pre-combusted glass ampoules after acidifying to a pH ~2 with 6 M HCL. Samples analyzed for dissolved

inorganic **nitrogen** (DIN) were stored frozen in a HDPE bottle (Nalgene) prior to analysis.

2.3 Analytical methods

The concentrations of DOC and TDN were determined using a high-temperature catalytic oxidation (HTCO) analyzer (TOC-V_{CPH}, Shimadzu, Japan). The standardization for DOC analysis was performed using a calibration curve of acetanilide (C:N ratio = 8) in ultra-pure water. The acidified samples were purged with pure air carrier gas for two min to remove dissolved inorganic carbon. Samples were carried into a combustion tube heated to 720 °C where the DOC was converted **quantitatively** to CO₂. CO₂ gas was detected by a non-dispersive infrared detector (NDIR). Our DOC and TDN methods were verified using seawater reference samples for DOC of 44–46 μmol L⁻¹ and for TDN of 32–34 μmol L⁻¹, which were produced by the University of Miami (Hansell's lab) in the USA. Inorganic nutrients were measured using nutrient auto-analyzers (Alliance Instruments, FUTURA+ for 2011 samples and QuAAtro39, SEAL Analytical Ltd. for 2016 samples). Reference seawater materials (KANSO Technos, Japan) were used for the verification of analytical accuracy. DON concentrations were calculated based on the difference between the TDN and DIN concentrations.

The values of δ¹³C-DOC were determined using a TOC-IR-MS instrument (Isoprime, Elementar, Germany). The analytical method is the same as that used by Kim et al. (2015) and Lee and Kim (2018). Low carbon water (< 2 μM; University of Miami, Hansell's lab) was measured for blank corrections and used for preparing all standard samples. The blank correction procedure is the same as that reported previously (Panetta et al., 2008; De Troyer et al., 2010). Certified IAEA-CH6 sucrose (International Atomic Energy Agency, -10.45 ± 0.03‰) was used for standardization. The standard solution was measured for every ten

samples to monitor analyzer drift. Our measured values of $\delta^{13}\text{C}$ -DOC of the Deep-Sea Water Reference (University of Miami) samples were $\pm 0.3\text{‰}$ relative to the values provided by Panetta et al. (2008) and Lang et al. (2007).

FDOM was determined using a spectrofluorometer (FluoroMate FS-2, SCINCO) within two days from the sampling time. EEMs were collected for the emission (Em) wavelength range of 240–600 nm with 2 nm intervals and an excitation (Ex) wavelength range of 240–500 nm with 5 nm intervals. Each sample value was subtracted for the signal of Milli-Q water produced daily to remove Raman scattering peaks. All data were converted to ppb quinine sulfate equivalent (QSE) using a quinine sulfate standard solution dissolved in 0.1N sulfuric acid at Ex/Em of 350/450 nm. EEMs-PARAFAC was performed on MATLAB (R2013a) using a DOMFluor toolbox. We did not correct EEM data for inner filter effects before measurements, because the inner filter effects were found to be negligible for coastal water samples using this instrument (Lee and Kim, 2018).

3. Results and Discussion

3.1 Horizontal distributions of DOM

The salinity of surface seawater in August 2011 ranged from 10 to 21, while the salinity in August 2016 ranged from 25 to 32 (Table 1 and Fig. 2). The concentrations of DOC in both sampling periods ranged from 100 μM to 200 μM (Fig. 2), which fall within the DOC ranges commonly observed in coastal waters (Gao et al., 2010; Osburn and Stedmon, 2011; Kim et al., 2012). The highest concentration of DOC in 2011 (186 μM) was observed at station M4-1 in the middle of the bay, whereas the highest concentration of DOC in 2016 (191 μM) was observed at station M1, which is the innermost station in the bay. DOC concentrations were

lowest at the outermost stations in both sampling periods. Concentrations of DON were in the range of 7–24 μM in 2011 and 3–15 μM in 2016, with the highest value at M5-1 in 2011 and at M1 in 2016 (Fig. 2).

EEM-PARAFAC dataset analyses identified three components in the surface water samples. Based on the excitation-emission peak location (Coble, 2007) and the **comparison with data in the OpenFluor (Murphy et al., 2014)**, Component 1 (FDOM_M , $\text{Ex} = 290\text{--}320\text{ nm}$, $\text{Em} = 370\text{--}420\text{ nm}$) is associated with a marine humic-like component. Component 2 (FDOM_H , $\text{Ex} = 320\text{--}360\text{ nm}$, $\text{Em} = 420\text{--}460\text{ nm}$) is associated with a terrestrial humic-like component. Component 3 (FDOM_P , $\text{Ex} = 275\text{--}300\text{ nm}$, $\text{Em} = 340\text{--}360\text{ nm}$) is associated with a protein-like component, which is a product of microbial processes. We use Component 2 as a representative of terrestrial humic-like FDOM (FDOM_H) in this study because there was a good correlation ($r^2 = 0.95$) between Component 1 and Component 2.

FDOM_H is known to indicate humic substances from terrestrial, anthropogenic, or agricultural sources (Coble, 2007), whereas FDOM_P is likely related to autochthonous or anthropogenic sources (Coble, 1996; Hudson et al., 2007). The intensities of FDOM_H and FDOM_P in 2011 were in the range of 3.6–9.2 ppb QSE and 4–79 ppb QSE, respectively (Fig. 3). The intensities of FDOM_H and FDOM_P in 2016 were in the range of 2.7–0.6 ppb QSE and 4.8–2.1 ppb QSE, respectively (Fig. 3). An exceptionally higher concentration of FDOM_P was observed at station M4-1 (78 ppb QSE) relative to that of other stations (2–25 ppb QSE) in 2011 (Fig. 4d).

3.2 Origin of excess DOM

The plot of DOC against salinity in 2011 showed two different mixing trends. The first slope showed a slight increase in DOC with decreasing salinity toward the innermost stations, including M1, M1-1, and M2 (Fig.4a, Group 1). The source of DOC in these lower salinity stations appears to have originated from land by natural and/or anthropogenic processes. The second trend showed a sharp rise in DOC (excess DOC in 2011) to the maximum at stations with salinities between 18 and 22 (Fig.4a, Group 2), indicating that there are other DOC sources at the high-salinity stations, besides the two end-member mixing. The excess DOC concentrations in higher-salinity waters could have been produced *in situ* by biological production and/or from land sources such as STP. The plot of DOC against salinity showed that DOC in 2016 was in a range similar to that of 2011, although there was much less influence from fresh water (Fig. 4a, Group 3). This plot shows that there was an addition of DOC (excess DOC) in 2016 for high-salinity water in the bay. The source of the excess DOC in this period could be from either biological production or land-seawater interactions. To determine the main sources of the excess DOC using $\delta^{13}\text{C}$ -DOC, FDOM, and DOC/DON ratios, the excess DOC stations are separated into three groups (Fig. 4a).

Group 1 includes low-salinity water stations (M1, M1-1, M2, M3, M5-1, M5-2, and M5-3) observed in 2011 (Fig. 1). $\delta^{13}\text{C}$ -DOC values for Group 1 ranged from -25.4‰ to -23.3‰ . We plotted a conservative mixing curve of $\delta^{13}\text{C}$ -DOC for two end-member mixing (Spiker, 1980; Raymond and Bauer, 2001). The assumed end-member values of DOC and $\delta^{13}\text{C}$ -DOC were $185\text{ }\mu\text{M}$ and -28‰ , respectively, for the terrestrial end-member ($S=0$) and $100\text{ }\mu\text{M}$ and -18‰ , respectively, for the marine end-member ($S=34$). The $\delta^{13}\text{C}$ -DOC values of Group 1 fit relatively well into this mixing curve ($\pm 1.5\text{‰}$), indicating that the conservative mixing between the terrestrial C3 land plant (-23‰ to -32‰) in freshwater and the open ocean

seawater results in DOC for Group 1. As such, the plot of $\delta^{13}\text{C}$ -DOC values versus C/N ratios also indicates that the excess DOC of Group 1 is from freshwater DOC (Fig. 5).

Group 2 includes high-salinity water stations (M4-1, M4-2, M6, M6-1, M7-1, M7-2, M8, M9, and M9-1) observed in 2011 (Fig. 1). The $\delta^{13}\text{C}$ -DOC values of Group 2 were in the range of -23.3‰ to -20.6‰ and were more enriched than the conservative mixing curve. These values are close to the marine $\delta^{13}\text{C}$ -DOC values (-22 to -18‰) (Fry et al., 1998), except for one station (M6), in this group (-23.3‰). The $\delta^{13}\text{C}$ -DOC values of Group 2 suggest that DOM was added *in situ* by biological production in seawater. As such, the plot of $\delta^{13}\text{C}$ -DOC values versus C/N ratios also indicates that the excess DOC of Group 2 is produced by marine phytoplankton (Fig. 5).

Group 3 includes high-salinity water stations (M1, M2, M3, M4, M5, M6, and M7) observed in 2016 (Fig. 1). Although all data were collected in the same wet season (August), the salinity ranges of both campaigns were different from those in 2011, with a narrow high salinity range in 2016. The $\delta^{13}\text{C}$ -DOC values for Group 3 also showed significantly different values relative to those sampled in 2011 (Group 1 and Group 2). The $\delta^{13}\text{C}$ -DOC values for Group 3 were depleted (-28.8‰ and -21.1‰) relative to the conservative mixing curve (Fig. 4b). The plot of $\delta^{13}\text{C}$ -DOC values versus C/N ratios indicates that the excess DOC of Group 3 is from C3 terrestrial plants through direct land-seawater interactions based on the fact that the excess DOC occurred in high-salinity (26–32) waters (Fig. 5).

FDOM_H showed a significant negative correlation with salinity ($r^2 = 0.89$). The concentrations were highest for Group 1 and lowest for Group 3. This result indicates that

humic DOM in this region was mainly from a terrestrial source and behaved conservatively in the freshwater and seawater mixing zone. This trend is commonly observed in coastal waters worldwide (Coble et al., 1998; Mayer et al., 1999). However, the concentration of FDOM_P showed no correlation with salinity. In general, FDOM_P shows non-conservative behavior in many estuaries owing to the extra source of DOC produced by *in situ* biological production (Benner and Opsahl, 2001). In the study region, a remarkably high FDOM_P concentration was observed at station M4-1 in 2011, where DOC concentration was highest. This trend also supports the argument based on the $\delta^{13}\text{C}$ -DOC results, where the main source of DOC at this station is from *in situ* biological production (Twardowski and Donaghay, 2001; Zhang et al., 2009).

Masan Bay has many potential land sources of DOM from different creeks. In addition, the treated sewage outflow from a STP is located near station M7-1 (Fig. 1). Many studies have been conducted to identify organic pollutants from STP (Kannan et al., 2010; Lee et al., 2011). In our study, however, station M7-1 did not show different DOM characteristics: (1) the concentrations of DOC, FDOM_H, and FDOM_P against salinity did not show anomalously higher or lower trends, relative to the other stations nearby. (2) The $\delta^{13}\text{C}$ -DOC values at M7-1 (-20.6‰) were close to the marine values, similar to those in other stations nearby, although they are known to be lighter in some US wastewater treatment plants (-26‰) (Griffith et al., 2009). (3) A fulvic-like peak was not observed, although a significantly higher fulvic-like peak (Ex/Em 320–340 nm/410–430 nm) was observed in treated sewage (Baker and Inverarity, 2004). (4) The increase of FDOM_P intensities at stations M7-1 and M7-2 was insignificant relative to those at stations M6-1 and M8, although FDOM_P is often used as a tracer of anthropogenic material including treated effluents (Hudson et al., 2007). Thus, we conclude

that the concentration of DOC at station M7-1 was not influenced by STP. This STP appears to reduce TOC concentrations to a level that cannot influence the DOC concentrations resulting from the mixing of other sources, as shown in several other estuaries (Abril et al., 2002). Combining tracers of $\delta^{13}\text{C}$ -DOC, C/N ratios, and FDOM_P , the excess DOC for Group 2 occurring in high salinity waters is likely from marine biological production, rather than from STP sources.

4. Conclusions

We determined the sources of DOM in 2011 and 2016 using the $\delta^{13}\text{C}$ -DOC, FDOM , and DOC/DON ratios. The main sources were separated into three groups based on DOC concentrations versus salinity plots. The DOM concentrations in the first group in 2011, which included the lowest salinity waters, were found to be mixtures of terrestrial DOM and open-ocean DOM sources based on the $\delta^{13}\text{C}$ values of -25.4‰ to -23.3‰ and a good correlation between FDOM_H and salinity. The excess DOC concentrations in the second group in higher salinity waters in 2011 were found to be produced *in situ* by biological production based on more enriched $\delta^{13}\text{C}$ -DOC values (-22.0‰ to -20.6‰), high FDOM_P concentrations, and low C/N ratios. The excess DOC concentrations in the third group in high salinity waters in 2016 seemed to be produced by direct interaction between land and seawater based on more depleted $\delta^{13}\text{C}$ -DOC values (-28.8‰ and -21.1‰), low FDOM concentrations, and high C/N ratios. Our results show that the combination of multiple DOM tracers, including $\delta^{13}\text{C}$ -DOC, FDOM , and DOC/DON ratios, is powerful for discriminating the complicated sources of DOM occurring in coastal waters.

Data availability

All data used in this paper can be accessed by email to the corresponding author upon request.

Author contribution

GK conceptualized the study. SL and TK collected the samples. SL performed the analyses. SL and GK wrote the manuscript and all authors contributed to the interpretation and discussion of the results.

Competing interests

The authors declare that they have no conflict of interest.

Acknowledgements

We thank members of the Environmental and Marine Biogeochemistry Laboratory (EMBL) for their assistance with sampling and laboratory analyses. We would like to thank two anonymous reviewers for constructive discussion and valuable comments.

Financial support

This work was supported by the National Research Foundation of Korea (NRF) grant funded by the Korean government (MEST) (NRF-2018R1A2B3001147).

References

Abril, G., Nogueira, M., Etcheber, H., Cabeçadas, G., Lemaire, E., and Brogueira, M.: Behaviour of organic carbon in nine contrasting European estuaries, Estuarine Coastal Shelf Sci., 54, 241-262, 2002.

301 Andrews, J., Greenaway, A., and Dennis, P.: Combined carbon isotope and C/N ratios as
 302 indicators of source and fate of organic matter in a poorly flushed, tropical estuary: Hunts Bay,
 303 Kingston Harbour, Jamaica, *Estuarine Coastal Shelf Sci.*, 46, 743-756, 1998.

304 Baker, A., and Inverarity, R.: Protein-like fluorescence intensity as a possible tool for
 305 determining river water quality, *Hydrological Processes*, 18, 2927-2945, 2004.

306 Bauer, J. E. and Bianchi, T. S.: Dissolved organic carbon cycling and transformation, in:
 307 *Treatise on estuarine and coastal science*, edited by: Wolanski, E. and Meluski, D. S., 5, 7-67,
 308 Academic Press, Waltham, 2011.

309 Beaupré, S. R.: Chapter 6 - The Carbon Isotopic Composition of Marine DOC, in:
 310 *Biogeochemistry of Marine Dissolved Organic Matter (Second Edition)*, edited by: Hansell, D.
 311 A., and Carlson, C. A., Academic Press, Boston, 335-368, 2015

312 Benner, R., and Opsahl, S.: Molecular indicators of the sources and transformations of
 313 dissolved organic matter in the Mississippi river plume, *Org. Geochem.*, 32, 597-611, 2001.

314 Carlson, C. A., and Hansell, D. A.: Chapter 3 - DOM Sources, Sinks, Reactivity, and Budgets,
 315 in: *Biogeochemistry of Marine Dissolved Organic Matter (Second Edition)*, edited by: Hansell,
 316 D. A., and Carlson, C. A., Academic Press, Boston, 65-126, 2015

317 Cauwet, G.: Chapter 12 - DOM in the Coastal Zone, in: *Biogeochemistry of Marine Dissolved*
 318 *Organic Matter*, edited by: Hansell, D. A., and Carlson, C. A., Academic Press, San Diego,
 319 579-609, 2002.

320 Cawley, K. M., Ding, Y., Fourqurean, J., and Jaffé, R.: Characterising the sources and fate of
 321 dissolved organic matter in Shark Bay, Australia: a preliminary study using optical properties

322 and stable carbon isotopes, *Mar. Freshwater Res.*, 63, 1098-1107,
 323 <https://doi.org/10.1071/MF12028>, 2012.

324 Chen, R. F., Bissett, P., Coble, P., Conmy, R., Gardner, G. B., Moran, M. A., Wang, X., Wells,
 325 M. L., Whelan, P., and Zepp, R. G.: Chromophoric dissolved organic matter (CDOM) source
 326 characterization in the Louisiana Bight, *Mar. Chem.*, 89, 257-272, 2004.

327 Coble, P. G., Green, S. A., Blough, N. V., and Gagosian, R. B.: Characterization of dissolved
 328 organic matter in the Black Sea by fluorescence spectroscopy, *Nature*, 348, 432, 1990b.

329 Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation-
 330 emission matrix spectroscopy, *Mar. Chem.*, 51, 325-346, 1996.

331 Coble, P. G., Del Castillo, C. E., and Avril, B.: Distribution and optical properties of CDOM in
 332 the Arabian Sea during the 1995 Southwest Monsoon, *Deep Sea Research Part II: Topical*
 333 *Studies in Oceanography*, 45, 2195-2223, 1998.

334 Coble, P. G.: Marine optical biogeochemistry: the chemistry of ocean color, *Chemical reviews*,
 335 107, 402-418, 2007.

336 Coffin, R. B., and Cifuentes, L. A.: Stable isotope analysis of carbon cycling in the Perdido
 337 Estuary, Florida, *Estuaries*, 22, 917-926, 1999.

338 De Troyer, I., Bouillon, S., Barker, S., Perry, C., Coorevits, K., and Merckx, R.: Stable isotope
 339 analysis of dissolved organic carbon in soil solutions using a catalytic combustion total organic
 340 carbon analyzer-isotope ratio mass spectrometer with a cryofocusing interface, *Rapid Commun.*
 341 *Mass Spectrom.*, 24, 365-374, 2010.

342 Deines, P.: The isotopic composition of reduced organic carbon, Handbook of environmental
 343 isotope geochemistry, 329-406, 1980.

344 Faganeli, J., Malej, A., Pezdic, J., and Malacic, V.: C: N: P ratios and stable c-isotopic ratios as
 345 indicators of sources of organic-matter in the gulf of trieste (northern adriatic), Oceanolog.
 346 Acta, 11, 377-382, 1988.

347 Fry, B., Hopkinson, C. S., Nolin, A., and Wainright, S. C.: $^{13}\text{C}/^{12}\text{C}$ composition of marine
 348 dissolved organic carbon, Chem. Geol., 152, 113-118, 1998.

349 Gao, L., Fan, D., Li, D., and Cai, J.: Fluorescence characteristics of chromophoric dissolved
 350 organic matter in shallow water along the Zhejiang coasts, southeast China, Marine
 351 environmental research, 69, 187-197, 2010.

352 Griffith, D. R., Barnes, R. T., and Raymond, P. A.: Inputs of fossil carbon from wastewater
 353 treatment plants to US rivers and oceans, Environ. Sci. Technol., 43, 5647-5651, 2009.

354 Griffith, D. R. and Raymond, P. A.: Multiple-source heterotrophy fueled by aged organic
 355 carbon in an urbanized estuary, Mar. Chem., 124, 14-22, 2011.

356 Hansell, D. A. and C. A. Carlson: Biogeochemistry of Marine Dissolved Organic Matter.
 357 Academic Press, San Diego, 774 pp. 2002.

358 Hong, S. H., Kannan, N., Jin, Y., Won, J. H., Han, G. M., and Shim, W. J.: Temporal trend,
 359 spatial distribution, and terrestrial sources of PBDEs and PCBs in Masan Bay, Korea, Mar.
 360 Pollut. Bull., 60, 1836-1841, 2010.

361 Huang, W., and Chen, R. F.: Sources and transformations of chromophoric dissolved organic
 362 matter in the Neponset River Watershed, *Journal of Geophysical Research: Biogeosciences*,
 363 114, 2009.

364 Hudson, N., Baker, A., and Reynolds, D.: Fluorescence analysis of dissolved organic matter in
 365 natural, waste and polluted waters—a review, *River Research and Applications*, 23, 631-649,
 366 2007.

367 Jaffé, R., Boyer, J., Lu, X., Maie, N., Yang, C., Scully, N., and Mock, S.: Source
 368 characterization of dissolved organic matter in a subtropical mangrove-dominated estuary by
 369 fluorescence analysis, *Mar. Chem.*, 84, 195-210, 2004.

370 Jiao, N., Herndl, G. J., Hansell, D. A., Benner, R., Kattner, G., Wilhelm, S. W., Kirchman, D.
 371 L., Weinbauer, M. G., Luo, T., and Chen, F.: Microbial production of recalcitrant dissolved
 372 organic matter: long-term carbon storage in the global ocean, *Nat. Rev. Microbiol.*, 8, 593-599,
 373 2010.

374 Kannan, N., Hong, S. H., Yim, U. H., Kim, N. S., Ha, S. Y., Li, D., and Shim, W. J.: Dispersion
 375 of organic contaminants from wastewater treatment outfall in Masan Bay, Korea, *Toxicology*
 376 *and Environmental Health Sciences*, 2, 200-206, 2010.

377 Kelley, C. A., Coffin, R. B., and Cifuentes, L. A.: Stable isotope evidence for alternative
 378 bacterial carbon sources in the Gulf of Mexico, *Limnol. Oceanogr.*, 43, 1962-1969, 1998.

379 Kim, T. -H., Waska, H., Kwon, E., Suryaputra, I. G. N., and Kim, G.: Production, degradation,
 380 and flux of dissolved organic matter in the subterranean estuary of a large tidal flat, *Mar. Chem.*,
 381 142, 1-10, 2012.

382 Kim, T. -H., Kim, G., Lee, S. A., and Dittmar, T.: Extraordinary slow degradation of Dissolved
 383 Organic Carbon (DOC) in a cold marginal sea, *Sci. Rep.*, 5, 13808, doi:10.1038/srep13808,
 384 2015.

385 Lamb, A. L., Wilson, G. P., and Leng, M. J.: A review of coastal palaeoclimate and relative
 386 sea-level reconstructions using $\delta^{13}\text{C}$ and C/N ratios in organic material. *Earth-Sci. Rev.*, 75(1-
 387 4), 29-57, 2006.

388 Lang, S. Q., Lilley, M. D., and Hedges, J. I.: A method to measure the isotopic (^{13}C)
 389 composition of dissolved organic carbon using a high temperature combustion instrument, *Mar.*
 390 *Chem.*, 103, 318-326, 2007.

391 Lee, C.-W., and Min, B.-Y.: Pollution in Masan Bay, a matter of concern in South Korea, *Mar.*
 392 *Pollut. Bull.*, 21, 226-229, 1990.

393 Lee, H. J., Hong, S. H., Kim, M., Ha, S. Y., An, S. M., and Shim, W. J.: Tracing origins of
 394 sewage and organic matter using dissolved sterols in Masan and Haengam Bay, Korea, *Ocean*
 395 *Sci. J.*, 46, 95-103, 2011.

396 Lee, S.-A., and Kim, G.: Sources, fluxes, and behaviors of fluorescent dissolved organic matter
 397 (FDOM) in the Nakdong River Estuary, Korea, *Biogeosciences*, 15, 1115-1122, 2018.

398 Lee, Y.-W., Hwang, D.-W., Kim, G., Lee, W.-C., and Oh, H.-T.: Nutrient inputs from submarine
 399 groundwater discharge (SGD) in Masan Bay, an embayment surrounded by heavily
 400 industrialized cities, Korea, *Sci. Total Environ.*, 407, 3181-3188, 2009.

401 Lobbes, J. M., Fitznar, H. P., and Kattner, G.: Biogeochemical characteristics of dissolved and
 402 particulate organic matter in Russian rivers entering the Arctic Ocean, *Geochim. Cosmochim.*
 403 *Acta*, 64, 2973-2983, 2000.

404 Markager, S., Stedmon, C. A., and S ndergaard, M.: Seasonal dynamics and conservative
 405 mixing of dissolved organic matter in the temperate eutrophic estuary Horsens Fjord, Estuarine
 406 Coastal Shelf Sci., 92, 376-388, 2011.

407 Mayer, L. M., Schick, L. L., and Loder, T. C.: Dissolved protein fluorescence in two Maine
 408 estuaries, Mar. Chem., 64, 171-179, 1999.

409 McCallister, S. L., Bauer, J. E., Ducklow, H. W., and Canuel, E. A.: Sources of estuarine
 410 dissolved and particulate organic matter: a multi-tracer approach, Org. Geochem., 37, 454-468,
 411 2006.

412 Milliman, J. D., Qinchun, X., and Zuosheng, Y.: Transfer of particulate organic carbon and
 413 nitrogen from the Yangtze River to the ocean, Am. J. Sci., 284, 824-834, 1984.

414 Murphy, K. R., Stedmon, C. A., Waite, T. D., and Ruiz, G. M.: Distinguishing between
 415 terrestrial and autochthonous organic matter sources in marine environments using
 416 fluorescence spectroscopy, Mar. Chem., 108, 40-58, 2008.

417 Murphy, K. R., Stedmon, C. A., Wenig, P., and Bro, R. J. A. M.: OpenFluor--an online spectral
 418 library of auto-fluorescence by organic compounds in the environment, 6, 658-661, 2014.

419 Osburn, C. L., and Stedmon, C. A.: Linking the chemical and optical properties of dissolved
 420 organic matter in the Baltic-North Sea transition zone to differentiate three allochthonous
 421 inputs, Mar. Chem., 126, 281-294, 10.1016/j.marchem.2011.06.007, 2011.

422 Osburn, C. L., Wigdahl, C. R., Fritz, S. C., and Saros, J. E.: Dissolved organic matter
 423 composition and photoreactivity in prairie lakes of the US Great Plains, Limnol. Oceanogr., 56,
 424 2371-2390, 2011.

425 Panetta, R. J., Ibrahim, M., and G  linas, Y.: Coupling a High-Temperature Catalytic Oxidation
 426 Total Organic Carbon Analyzer to an Isotope Ratio Mass Spectrometer To Measure Natural-
 427 Abundance $\delta^{13}\text{C}$ -Dissolved Organic Carbon in Marine and Freshwater Samples, *Anal. Chem.*,
 428 80, 5232-5239, 2008.

429 Pradhan, U., Wu, Y., Shirodkar, P., Zhang, J., and Zhang, G.: Sources and distribution of
 430 organic matter in thirty five tropical estuaries along the west coast of India-a preliminary
 431 assessment, *Estuarine Coastal Shelf Sci.*, 151, 21-33, 2014.

432 Raymond, P. A., and Bauer, J. E.: DOC cycling in a temperate estuary: a mass balance approach
 433 using natural ^{14}C and ^{13}C isotopes, *Limnol. Oceanogr.*, 46, 655-667, 2001.

434 Spiker, E.: The Behavior of C-14 and C-13 in Estuarine Water-Effects of Insitu CO_2 Production
 435 and Atmospheric Exchange, *Radiocarbon*, 22, 647-654, 1980.

436 Thornton, S., and McManus, J.: Application of organic carbon and nitrogen stable isotope and
 437 C/N ratios as source indicators of organic matter provenance in estuarine systems: evidence
 438 from the Tay Estuary, Scotland, *Estuarine Coastal Shelf Sci.*, 38, 219-233, 1994.

439 Tremblay, L., and Benner, R.: Microbial contributions to N-immobilization and organic matter
 440 preservation in decaying plant detritus, *Geochim. Cosmochim. Acta*, 70, 133-146, 2006.

441 Twardowski, M. S., and Donaghay, P. L.: Separating in situ and terrigenous sources of
 442 absorption by dissolved materials in coastal waters, *Journal of Geophysical Research: Oceans*
 443 (1978–2012), 106, 2545-2560, 2001.

444 Wang, X.-C., Chen, R. F., and Gardner, G. B.: Sources and transport of dissolved and
 445 particulate organic carbon in the Mississippi River estuary and adjacent coastal waters of the
 446 northern Gulf of Mexico, *Mar. Chem.*, 89, 241-256, 2004.

447 Ya, C., Anderson, W., and Jaffé, R.: Assessing dissolved organic matter dynamics and source
448 strengths in a subtropical estuary: Application of stable carbon isotopes and optical properties,
449 Cont. Shelf Res., 92, 98-107, 2015.

450 Yoo, K.: Population dynamics of dinoflagellate community in Masan Bay with a note on the
451 impact of environmental parameters, Mar. Pollut. Bull., 23, 185-188, 1991.

452 Zhang, Y., van Dijk, M. A., Liu, M., Zhu, G., and Qin, B.: The contribution of phytoplankton
453 degradation to chromophoric dissolved organic matter (CDOM) in eutrophic shallow lakes:
454 field and experimental evidence, water research, 43, 4685-4697, 2009.

Table 1. Salinity, DOC, FDOM_H, FDOM_P, and $\delta^{13}\text{C}$ -DOC in surface water of Masan Bay in August 2011 and August 2016.

sampling campaign	station	salinity	DOC μM	FDOM _H ppbQSE	FDOM _T ppbQSE	$\delta^{13}\text{C}$ - DOC ‰	DON μM	DOC/DON
Aug. 2011	M1	14.0	148	6.7	13.6	-25.4	12	12
	M1-1	12.8	151	9.2	14.3	-24.3	7	21
	M2	10.2	157	9.0	5.4	-24.6	11	14
	M3	16.3	147	8.2	14.7	n/a	16	9
	M4-1	19.0	186	7.1	78.7	-21.9	13	15
	M4-2	18.6	155	6.9	8.3	-21.6	10	15
	M5-1	17.7	138	4.5	4.5	-23.3	24	6
	M5-2	18.4	133	5.8	20.9	-24.5	11	12
	M5-3	18.9	135	8.0	11.3	-23.7	13	11
	M6	18.4	146	6.6	24.8	-23.3	19	8
	M6-1	19.2	142	5.5	7.4	n/a	9	15
	M7-1	19.5	157	5.8	10.5	-20.6	11	15
	M7-2	18.9	148	5.6	9.6	-21.5	12	12
	M8	19.5	152	5.6	7.6	-21.5	15	10
	M9	18.8	149	5.6	14.5	-21.9	10	15
	M9-1	19.1	154	5.1	10.2	-21.0	12	13
	M9-2	20.8	106	3.6	13.1	-22.0	8	13
Aug. 2016	M1	29.2	191	2.7	4.8	-22.8	15	13
	M2	29.9	164	2.0	3.4	-21.1	7	22
	M3	26.0	155	2.5	3.8	-28.8	8	19
	M4	27.4	149	1.9	3.5	-22.6	9	17
	M5	25.5	165	1.8	3.3	-23.5	10	16
	M6	30.5	147	1.1	3.0	-23.7	6	26
	M7	31.4	166	1.3	4.4	-26.2	4	45
	M8	32.0	123	0.8	2.3	-23.7	5	26
	M9	32.0	146	0.6	2.1	-24.4	5	30
	M10	31.9	130	0.7	2.7	-25.0	3	39

n/a = not available.

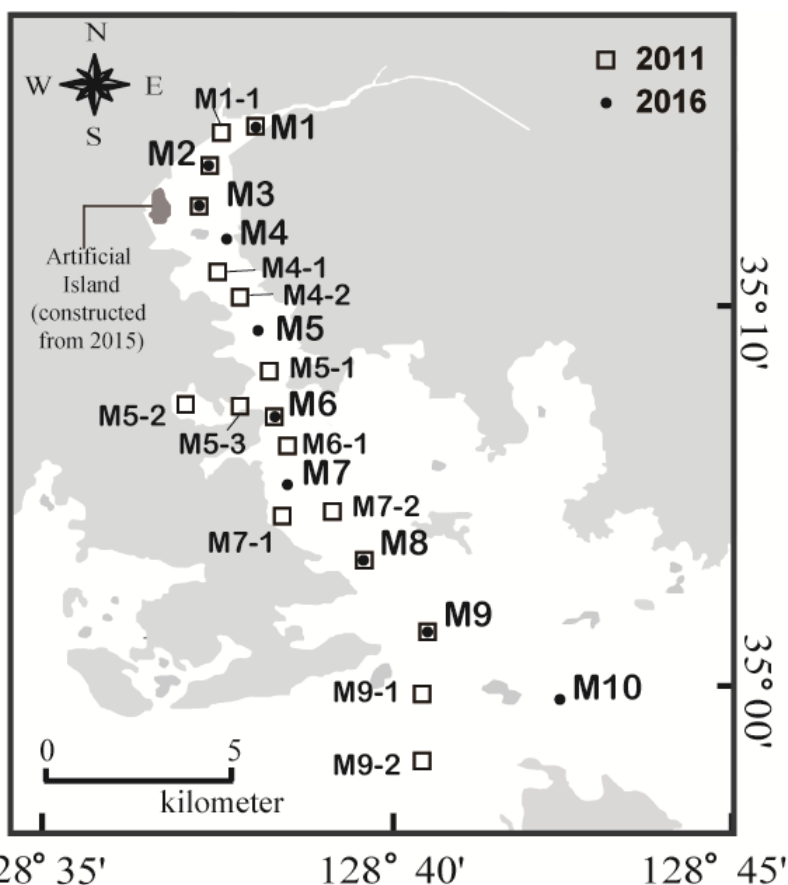


Figure 1. A map showing the sampling stations for DOC, $\delta^{13}\text{C}$ -DOC, FDOM, and DOC/DON ratio in Masan Bay, Korea, in 2011 and 2016.

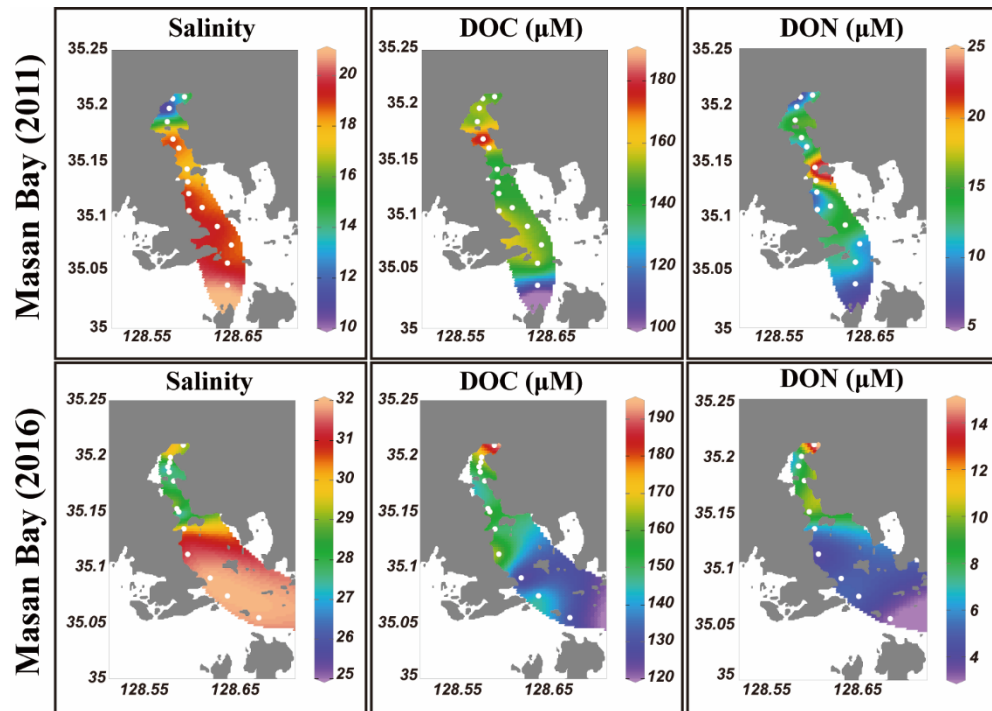


Figure 2. Surface distributions of salinity, DOC, and DON in Masan Bay, Korea, in 2011 and 2016.

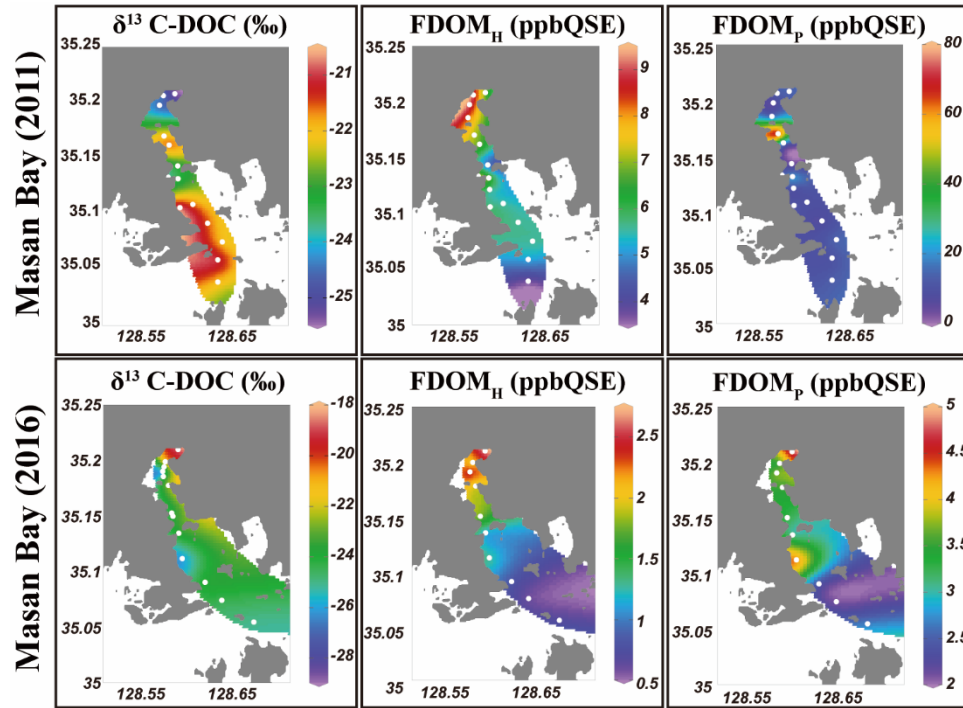


Figure 3. Surface distributions of $\delta^{13}\text{C-DOC}$, FDOM_H , and FDOM_P in Masan Bay, Korea, in 2011 and 2016.

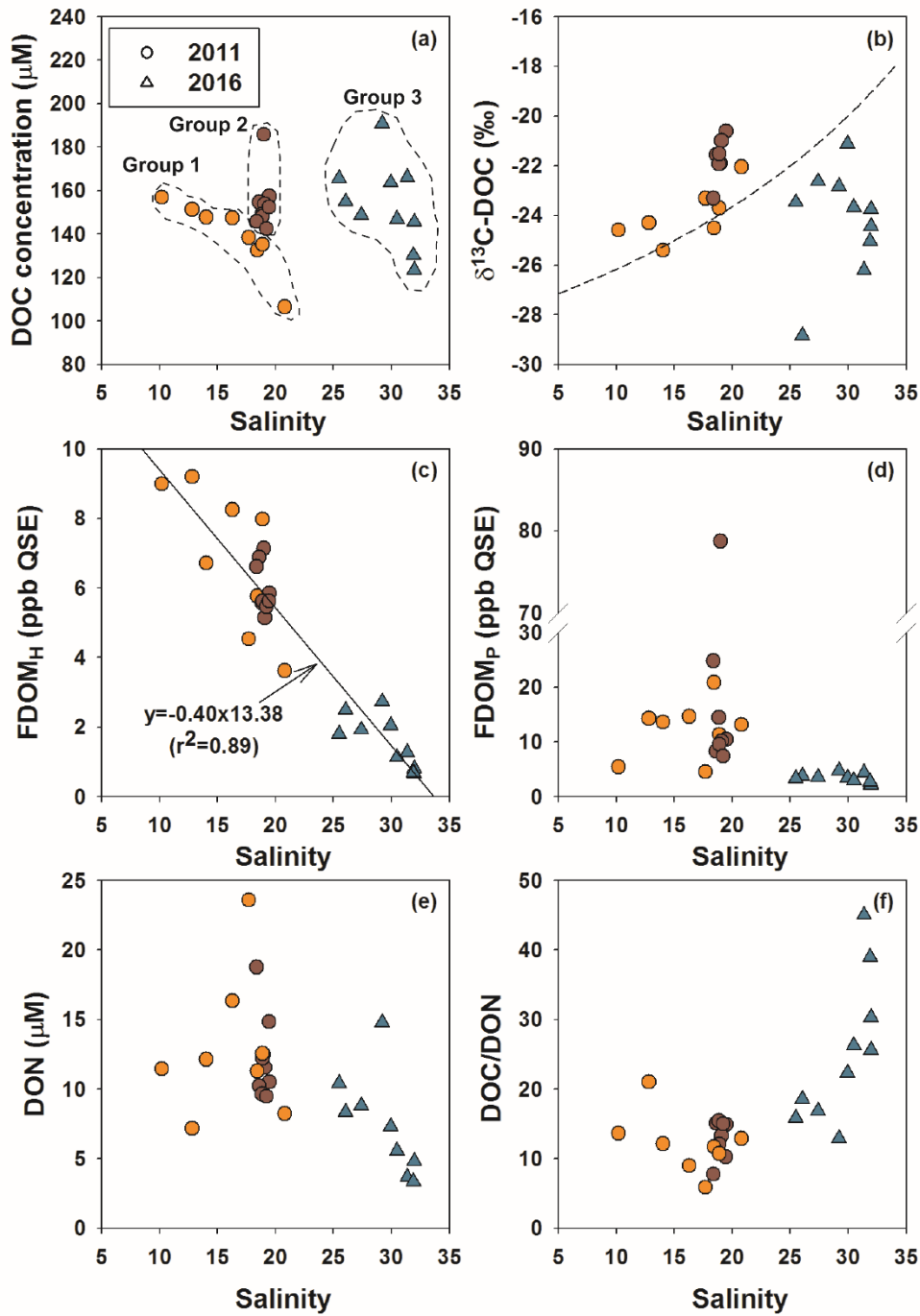


Figure 4. Relationships between salinity versus (a) DOC, (b) $\delta^{13}\text{C-DOC}$, (c) FDOM_H , (d) FDOM_T , (e) DON, and (f) DOC/DON values. The DOC groups are included in the dashed circles. The dashed line (b) represents the binary conservative mixing line for $\delta^{13}\text{C-DOC}$ between the terrestrial end-member and the marine end-member.

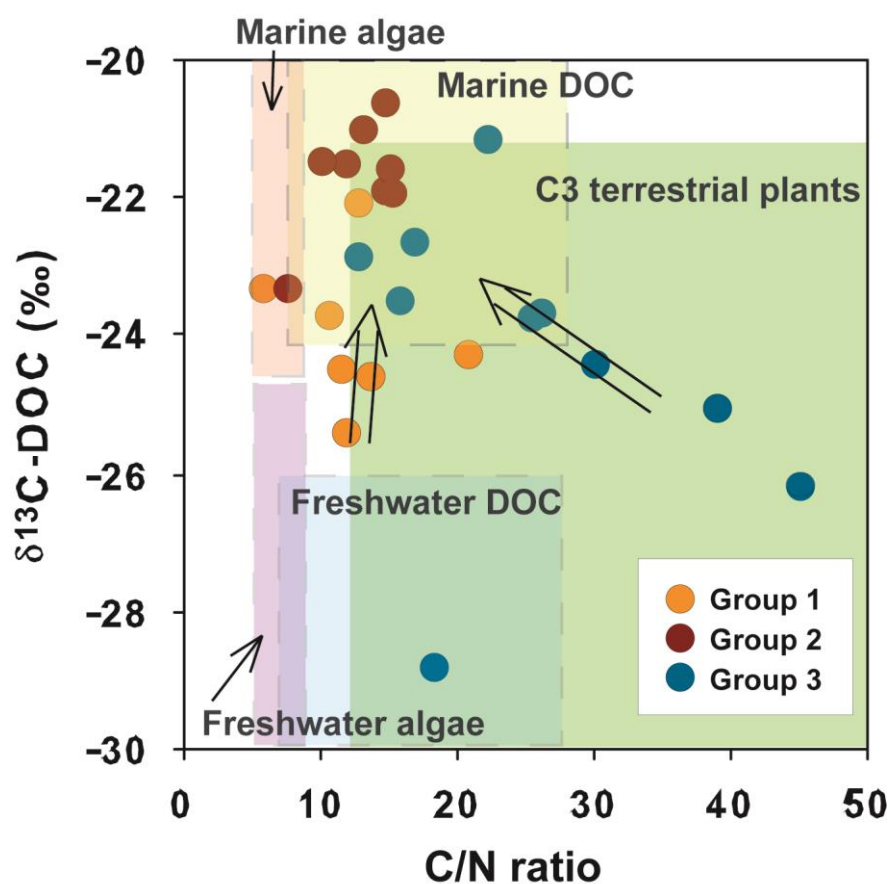


Figure 5. The relationship between DOC/DON ratios and $\delta^{13}\text{C-DOC}$ values in Masan Bay. The ranges of DOC/DON ratios and $\delta^{13}\text{C-DOC}$ values are based on the values reported by Lamb et al. (2006) and Beaupré (2015).