1	Tracing terrestrial versus marine sources of dissolved organic carbon in a coastal bay								
2	using stable carbon isotopes								
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21 Abstract

The sources of dissolved organic matter (DOM) in coastal waters are diverse, and they play 22 23 different roles in the biogeochemistry and ecosystems of the ocean. In this study, we measured dissolved organic carbon (DOC) and nitrogen (DON), δ^{13} C-DOC, and fluorescent dissolved 24 organic matter (FDOM) in coastal bay waters surrounded by large cities (Masan Bay, Korea) 25 to determine the different DOM sources in this region. The surface seawater samples were 26 collected in two sampling campaigns (Aug. 2011 and Aug. 2016). The salinities were in the 27 28 range of 10–21 in 2011 and 25–32 in 2016. In 2011, excess DOC was observed for high-salinity (16-21) waters, indicating that the excess source inputs were mainly from marine 29 autochthonous production according to the δ^{13} C-DOC values of -23.7% to -20.6%, the higher 30 31 concentrations of protein-like FDOM, and the lower DOC/DON (C/N) ratios (8-15). By contrast, excess DOC observed in high-salinity waters in 2016 was characterized by low 32 FDOM, more depleted δ^{13} C values of -28.8% to -21.1%, and high C/N ratios (13-45), 33 suggesting that the excess DOC source is from terrestrial C3 plants by direct land-seawater 34 interactions. Our results show that multiple DOM tracers such as δ^{13} C-DOC, FDOM, and C/N 35 36 ratios are powerful for discriminating the complicated sources of DOM occurring in coastal 37 waters.

38 **1. Introduction**

Dissolved organic matter (DOM) plays an important role in biogeochemical cycles 39 (e.g., de-oxygenation, acidification, photochemistry) and ecosystems of the ocean (Hansell and 40 Carlson, 2002). DOM composition depends on its parent organic matter and subsequent 41 biogeochemical processes. DOM in coastal waters originates from various sources including 42 43 (1) *in situ* production by primary production, exudation of aquatic plants, and their degradation (Markager et al., 2011; Carlson and Hansell, 2015), (2) terrestrial sources by the degradation 44 45 of soil and terrestrial plant matter (Opsahl and Benner, 1997; Bauer and Bianchi, 2011), and (3) anthropogenic sources such as industrial, agricultural, and domestic sewage (Griffith and 46 Raymond, 2011). 47

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Depending on the origin and composition of DOM, its behavior and cycling are 49 different: the labile fraction of DOM is decomposed rapidly through microbially or 50 photochemically mediated processes, whereas refractory DOM is resistant to degradation and 51 can persist in the ocean for millennia. In the coastal ocean, organic matter from terrestrial plant 52 53 litter or soils appears to be more refractory (Cauwet, 2002) and thus often behaves conservatively. In addition, refractory DOM is produced in the ocean by bacterial 54 transformation of labile DOM by reshaping its composition (Tremblay and Benner, 2006; Jiao 55 56 et al., 2010). However, it is still very difficult to determine the sources and characteristics of 57 DOM in coastal waters.

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59 There are many approaches to distinguish the source of DOM in coastal areas using 60 various tracers (Faganeli et al., 1988; Benner and Opsahl, 2001; Chen et al., 2004; Baker and 61 Spencer, 2004; Cawley et al., 2012; Lee and Kim, 2018). The stable carbon isotopic

composition of dissolved organic carbon (δ^{13} C-DOC) has been used to distinguish different 62 sources. In general, δ^{13} C values derived from C3 and C4 land plants are in the range of -23%63 64 -34% and -9% -17% (Deines, 1980), respectively, while those derived from marine phytoplankton range from -18 to -22‰ (Kelley et al., 1998; Coffin and Cifuentes, 1999). In 65 addition, the optically active fraction of DOM known as fluorescent DOM (FDOM) have been 66 successfully used for characterizing DOM (Coble et al., 1990; Coble, 1996). Fluorescence 67 excitation-emission matrices and parallel factor analysis (EEM-PARAFAC) technique has been 68 69 applied to trace the source of humic-like versus protein-like DOM in coastal waters and estuaries (Chen et al., 2004; Jaffé et al., 2004; Murphy et al., 2008). DOC/DON ratios are often 70 used to differentiate allochthonous versus autochthonous sources. The C/N ratios of terrestrial 71 72 organic carbon are usually higher than 12, while those of marine organic carbon from phytoplankton are almost constant ranging from 6 to 8 (Milliman et al., 1984; Lobbes et 73 al.,2000). However, the interpretation of isotopic ratio of bulk sample alone in complex coastal 74 environments is somewhat complicated by the overlap of the isotopic ranges. Thus, several 75 studies have used δ^{13} C-DOC combined with FDOM (Osburn and Stedmon, 2011; Osburn et 76 77 al., 2011; Ya et al., 2015; Lu et al., 2015) or carbon isotope ratios combined with C/N ratio (Thornton and McManus, 1994; Andrews et al., 1998; Wang et al., 2004; McCallister et al., 78 2006; Pradhan et al., 2014) to discriminate different sources of DOM in estuarine and coastal 79 80 waters. As far as we know, these three tracers together have not yet been used together to 81 determine DOM sources in coastal waters.

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83 Our study aimed at discriminating DOM sources in coastal waters, where various 84 sources are present, using δ^{13} C-DOC, FDOM, and DOC/DON ratios together. Masan bay is 85 surrounded by cities with thousands of industrial plants and a population of 1.1 million. In

86 association with large anthropogenic nutrient loading, this area has been recognized as a highly eutrophic embayment (Lee and Min, 1990; Yoo, 1991; Hong et al., 2010). Red tides and 87 hypoxic water mass in the bottom layer of the bay have occurred annually in spring and summer 88 (Lee et al., 2009). In addition, there are potential point sources of sewage treatment plants 89 (STPs) which manage domestic and industrial wastewater from Masan and Changwon cities. 90 91 Lee et al. (2011) revealed the origins of sewage and organic matter using dissolved sterols in 92 Masan Bay. They reported that the water samples from the creeks, inner bay, and nearby STP 93 were affected by sewage sources. Oh et al. (2017) showed that the excess DOC in bay water is produced by phytoplankton production. Therefore, Masan Bay is a suitable place to test the 94 applicability of these multiple tracers to differentiate complicated DOM sources in other areas 95 96 of the world's coastal regions.

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98 2. Materials and methods

99 2.1 Study site

Masan Bay is located on the southeast coast of Korea with an area of approximately 100 80 km² (Fig. 1). The annual precipitation is approximately 1500 mm, and most of the 101 precipitations occurs in the summer monsoon season. The amount of freshwater discharge into 102 this bay is approximately $2.5 \times 10^8 \text{ m}^3 \text{ yr}^{-1}$ with significant seasonal variation. The tide is semi-103 104 diurnal, showing a maximum tidal amplitude of ~ 1.9 m (average = 1.3 m) during the sampling period. Due to topographic conditions, the current is very weak $(2-3 \text{ cm s}^{-1})$, and the residence 105 times of water in the inner bay and in the entire bay are approximately 54 and 23 days, 106 107 respectively (Lee et al., 2009). In the middle of the bay, an artificial island was constructed in 2015-2016 (Fig. 1) with an area of 0.64 km². The artificial island may have resulted in changes 108 in water currents, residence times, and biogeochemical conditions. 109

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111 *2.2 Sampling*

Sampling was conducted in August 2011 and August 2016 in Masan Bay. Water 112 samples were collected from the surface at 17 sites in 2011 and 10 sites in 2016. The bay 113 receives a large amount of freshwater discharge from the northernmost part of the region. The 114 average surface water temperatures were 30.4 ± 2.3 °C in 2011 and 26.5 ± 0.7 °C in 2016. All 115 water samples were filtered through pre-combusted GF/F filters. Samples for FDOM analysis 116 117 were stored at 4°C in pre-combusted amber vials. Samples for DOC, total dissolved nitrogen (TDN), and δ^{13} C-DOC analysis were stored in pre-combusted glass ampoules after acidifying 118 to a pH ~2 with 6M HCL. Samples analyzed for dissolved inorganic nitrogen (DIN) were stored 119 120 frozen in a HDPE bottle (Nalgene) prior to analyses.

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122 *2.3 Analytical methods*

The concentrations of DOC and TDN were determined using a high-temperature 123 catalytic oxidation (HTCO) analyzer (TOC-V_{CPH}, Shimadzu, Japan). The standardization for 124 125 DOC analysis was performed using a calibration curve of acetanilide (C:N ratio = 8) in ultrapure water. The acidified samples were purged with pure air carrier gas for two min to remove 126 dissolved inorganic carbon. Samples were carried into a combustion tube heated to 720 °C 127 128 where the DOC was converted quantitatively to CO₂. CO₂ gas was detected by a non-dispersive infrared detector (NDIR). Our DOC and TDN methods were verified using the seawater 129 reference samples for DOC of 44–46 μ mol L⁻¹ and for TDN of 32–34 μ mol L⁻¹, which were 130 131 produced by the University of Miami (Hansell's lab, USA). Inorganic nutrients were measured using nutrient auto-analyzers (Alliance Instruments, FUTURA+ for 2011 samples and 132 QuAAtro39, SEAL Analytical Ltd. for 2016 samples). Reference seawater materials (KANSO 133

Technos, Japan) were used for the verification of analytical accuracy. DON concentrations
were calculated based on the difference between the TDN and DIN concentrations.

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The values of δ^{13} C-DOC were determined using a TOC-IR-MS instrument (Isoprime, 137 Elementar, Germany). The analytical method is the same as that used by Kim et al. (2015) and 138 Lee and Kim (2018). Low carbon water (< 2 μ M; University of Miami, Hansell's lab) was 139 measured for blank corrections and used for preparing all standard samples. The blank 140 141 correction procedure is the same as that reported previously (Panetta et al., 2008; De Troyer et al., 2010). Certified IAEA-CH6 sucrose (International Atomic Energy Agency, -10.45 ± 142 0.03‰) was used for standardization. The standard solution was measured for every ten 143 samples to monitor the drifting effect. Our measured values of δ^{13} C-DOC of the Deep-Sea 144 Water Reference (University of Miami) samples were $\pm 0.3\%$ relative to the values provided 145 by Panetta et al. (2008) and Lang et al. (2007). 146

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FDOM was determined using a spectrofluorometer (FluoroMate FS-2, SCINCO) 148 within two days from the sampling time. EEMs were collected for the emission (Em) 149 wavelength range of 240–600 nm with 2 nm intervals and an excitation (Ex) wavelength range 150 of 240-500 nm with 5 nm intervals. Each sample value was subtracted for the signal of Milli-151 152 Q water produced daily to remove Raman scattering peaks. All data were converted to ppb quinine sulfate equivalent (QSE) using a quinine sulfate standard solution dissolved in 0.1N 153 sulfuric acid at Ex/Em of 350/450 nm. We did not correct EEM data for inner filter effects 154 155 before measurements, because the inner filter effects were found to be negligible for coastal water samples using this instrument (Lee and Kim, 2018). EEMs-PARAFAC was performed 156

on MATLAB (R2013a) using a DOMFluor toolbox, and the three components (C1-C3) were
validated by split-half analysis (Figs. S1 and S2).

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160 **3. Results and Discussion**

161 *3.1 Horizontal distributions of DOM*

The salinity of surface seawater in August 2011 ranged from 10 to 21, while the salinity in 162 August 2016 ranged from 25 to 32 (Table 1 and Fig. 2). The concentrations of DOC in both 163 164 sampling periods ranged from 100 μ M to 200 μ M (Fig. 2), which fall within the DOC ranges commonly observed in coastal waters (Gao et al., 2010; Osburn and Stedmon, 2011; Kim et al., 165 2012). The highest concentration of DOC in 2011 (186 µM) was observed at station M4-1 in 166 167 the middle of the bay, whereas the highest concentration of DOC in 2016 (191 µM) was observed at station M1, which is the innermost station in the bay. DOC concentrations were 168 lowest at the outermost stations in both sampling periods. Concentrations of DON were in the 169 range of 7–24 μ M in 2011 and 3–15 μ M in 2016, with the highest value at M5-1 in 2011 and 170 at M1 in 2016 (Fig. 2). 171

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EEM-PARAFAC dataset analyses identified three components in the surface water 173 samples. EEMs contour plots and split-half validation results of three components are shown 174 175 in the supplementary (Figs. S1 and S2). Based on the comparison with the data in the OpenFluor (Murphy et al., 2014), Component 1 (FDOM_H, Ex/Em = 322/405 nm) is associated 176 with a terrestrial humic-like component (Liu et al., 2019; Dalmagro et al., 2019; Chen et al., 177 178 2016). Component 2 (FDOM_M, Ex/Em = 386/450 nm) is also associated with an allochthonous humic-like component (Wünsch et al., 2017). Component 3 (FDOM_P, Ex/Em = 280/330 nm) 179 is associated with a protein-like component, which is a product of microbial processes (Liu et 180

al., 2019; Murphy et al., 2011; Osburn et al., 2011). We use Component 1 as a representative of terrestrial humic-like FDOM (FDOM_H) in this study because there was a significant correlation ($r^2 = 0.95$) between Component 1 and Component 2.

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FDOM_H is known to indicate humic substances from terrestrial, anthropogenic, or 185 agricultural sources (Coble, 2007), whereas $FDOM_P$ is likely related to autochthonous or 186 anthropogenic sources (Coble, 1996; Hudson et al., 2007). The intensities of FDOM_H and 187 FDOM_P in 2011 were in the range of 3.6–9.2 ppb QSE and 4–79 ppb QSE, respectively (Fig. 188 3). The intensities of FDOM_H and FDOM_P in 2016 were in the range of 2.7-0.6 ppb QSE and 189 4.8–2.1 ppb QSE, respectively (Fig. 3). An exceptionally higher concentration of FDOM_P was 190 191 observed at station M4-1 (78 ppb QSE) relative to that of other stations (2–25 ppb QSE) in 2011 (Fig. 4d). 192

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194 *3.2 Origin of excess DOM*

The plot of DOC against salinity in 2011 showed two different mixing trends. The first 195 196 slope showed a slight increase in DOC with decreasing salinity toward the innermost stations, including M1, M1-1, and M2 (Fig.4a, Group 1). The second trend showed a sharp rise in DOC 197 (excess DOC in 2011) to the maximum at stations with salinities between 18 and 22 (Fig.4a, 198 199 Group 2), indicating that there are other DOC sources at the high-salinity stations, besides the two end-member mixing. The plot of DOC against salinity showed that DOC in 2016 was in a 200 range similar to that of 2011, although there was much less influence from fresh water (Fig. 4a, 201 202 Group 3). This plot shows that there was an addition of DOC (excess DOC) in 2016 for highsalinity water in the bay. The potential sources of excess DOC occurring in this bay water may 203 include terrestrial freshwater in creeks, STP water, direct land-seawater interaction, and in situ 204

biological production. The creek water may also include various anthropogenic sources (i.e., industrial, agricultural, and domestic sewage) as well as natural land sources. There are no saltmarsh or wetland habitats in Masan Bay. To determine the main sources of the excess DOC using δ^{13} C-DOC, FDOM, and DOC/DON ratios, the excess DOC stations are separated into three groups (Group 1, Group 2 in 2011, and Group 3 in 2016) (Fig. 4a).

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Group 1 includes low-salinity water stations (M1, M1-1, M2, M3, M5-1, M5-2, and 211 M5-3) observed in 2011 (Fig. 1). δ^{13} C-DOC values for Group 1 ranged from -25.4‰ to 212 -23.3‰. We plotted a conservative mixing curve of δ^{13} C-DOC for two end-member mixing 213 (Spiker, 1980; Raymond and Bauer, 2001). The assumed end-member values of DOC and δ^{13} C-214 215 DOC were 185 µM and -28‰ (Raymond and Bauer, 2001), respectively, for the terrestrial end-member (S=0) and 100 µM and -18‰ (Kelley et al., 1998), respectively, for the marine 216 end-member (S=34). The δ^{13} C values of Group 1 fall into the mixing line or are slightly heavier 217 than the mixing line within 1.5 ‰, indicating the conservative mixing between the terrestrial 218 C3 land plant (-23‰ to -32‰; Deines, 1980) in freshwater and the open ocean seawater. The 219 slightly heavier values could be produced by *in situ* biological production during the mixing 220 processes. As such, the plot of δ^{13} C-DOC values versus C/N ratios also indicates that the excess 221 DOC of Group 1 is from freshwater DOC (Fig. 5a). 222

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Group 2 includes high-salinity water stations (M4-1, M4-2, M6, M6-1, M7-1, M7-2, M8, M9, and M9-1) observed in 2011 (Fig. 1). The δ^{13} C-DOC values of Group 2 were in the range of -23.3‰ to -20.6‰ and were more enriched than the conservative mixing curve. These values are close to the marine δ^{13} C-DOC values (-22 to -18‰) (Fry et al., 1998), except for one station (M6), in this group (-23.3‰). The δ^{13} C-DOC values of Group 2 suggest that excess DOM was added *in situ* by biological production in seawater. As such, the plot of δ^{13} C-DOC values versus C/N ratios also indicates that the excess DOC of Group 2 is produced by marine phytoplankton (Fig. 5a).

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Group 3 includes high-salinity water stations (M1, M2, M3, M4, M5, M6, and M7) 233 observed in 2016 (Fig. 1). Although all data were collected in the same wet season (August), 234 the salinity ranges of both campaigns were different from those in 2011, with a narrow high 235 salinity range in 2016. The δ^{13} C-DOC values for Group 3 also showed significantly different 236 values relative to those sampled in 2011 (Group 1 and Group 2). The δ^{13} C-DOC values for 237 Group 3 were depleted (-28.8‰ and -21.1‰) relative to the conservative mixing curve (Fig. 238 4b). The plot of δ^{13} C-DOC values versus C/N ratios indicates that the excess DOC of Group 3 239 is from C3 terrestrial plants through direct land (including the possible sources from a newly-240 built artificial island)-seawater interactions, based on the fact that the excess DOC occurred in 241 high-salinity (26–32) waters (Fig. 5a). 242

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FDOM_H showed a significant negative correlation with salinity ($r^2 = 0.89$). The 244 concentrations were highest for Group 1 and lowest for Group 3. This result indicates that 245 humic DOM in this region was mainly from a terrestrial source and behaved conservatively in 246 247 the freshwater and seawater mixing zone. This trend is commonly observed in coastal waters worldwide (Coble et al., 1998; Mayer et al., 1999). However, the concentration of FDOM_P 248 showed no correlation with salinity. In general, FDOM_P shows non-conservative behavior in 249 250 many estuaries owing to the extra source of DOC produced by *in situ* biological production (Benner and Opsahl, 2001). In the study region, a remarkably high FDOM_P concentration was 251 observed at station M4-1 in 2011, where DOC concentration was highest. This trend also 252

supports the argument, based on the δ^{13} C-DOC results, that the main source of DOC at this station is from *in situ* biological production. We observed the decoupling between DOC and FDOM_H because FDOM_H is not the major portion of DOC in this bay, except M4-1 station.

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Masan Bay has many potential land sources of DOM from different creeks. In addition, 257 the treated sewage outflow from a STP is located near station M7-1 (Fig. 1). Many studies have 258 been conducted to identify organic pollutants from STP (Kannan et al., 2010; Lee et al., 2011). 259 260 In our study, however, station M7-1 did not show different DOM characteristics: (1) the concentrations of DOC, FDOM_H, and FDOM_P against salinity did not show anomalously 261 higher or lower trends, relative to the other stations nearby. (2) The δ^{13} C-DOC values at M7-1 262 263 (-20.6‰) were close to the marine values (Fry et al., 1998), similar to those in other stations nearby, although they are known to be lighter in some US wastewater treatment plants (-26%)264 (Griffith et al., 2009). (3) A fulvic-like peak was not observed, although a significantly higher 265 fulvic-like peak (Ex/Em 320-340 nm/410-430 nm) was observed in treated sewage (Baker and 266 Inverarity, 2004). (4) The increase of FDOM_P intensities at stations M7-1 and M7-2 was 267 268 insignificant relative to those at stations M6-1 and M8, although FDOM_P is often used as a tracer of anthropogenic material including treated effluents (Hudson et al., 2007). Thus, we 269 conclude that the concentration of DOC at station M7-1 was not influenced by STP. This STP 270 271 appears to reduce TOC concentrations to a level that cannot influence the DOC concentrations resulting from the other mixing sources, as shown in several other estuaries (Abril et al., 2002). 272 In general, anomalously high FDOM_P was observed for anthropogenic sources (Coble, 273 274 1996; Baker et al., 2003). The δ^{13} C values of sewage effluents generally ranged from -22‰ to

276 24‰ to -28% (Griffith et al., 2009). The δ^{13} C vs FDOM_P plot (Fig. 5b) shows that there was

-28.5‰ (Andrews et al., 1998; Barros et al., 2010), and those of STP effluents ranged from -

277 no increase in FDOM_P concentrations for samples which had depleted δ^{13} C values. Thus, we 278 conclude that there was no significant DOC input from untreated sewage or STP sources in this 279 bay.

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281 **4. Conclusions**

We determined the sources of DOM in 2011 and 2016 using the δ^{13} C-DOC, FDOM, 282 and DOC/DON ratios. The main sources were separated into three groups based on DOC 283 284 concentrations versus salinity plots. The DOM concentrations in the first group in 2011, which included the lowest salinity waters, were found to be mixtures of terrestrial DOM and open-285 ocean DOM sources based on the δ^{13} C values of -25.4‰ to -23.3‰ and a good correlation 286 287 between FDOM_H and salinity. The excess DOC concentrations in the second group in higher salinity waters in 2011 were found to be produced *in situ* by biological production based on 288 more enriched δ^{13} C-DOC values (-22.0% to -20.6%), high FDOM_P concentrations, and low 289 C/N ratios. The excess DOC concentrations in the third group in high salinity waters in 2016 290 seemed to be produced by direct interaction between land and seawater based on more depleted 291 δ^{13} C-DOC values (-28.8‰ and -21.1‰), low FDOM concentrations, and high C/N ratios. Our 292 results show that the combination of multiple DOM tracers, including δ^{13} C-DOC, FDOM, and 293 DOC/DON ratios, is powerful for discriminating the complicated sources of DOM occurring 294 295 in coastal waters.

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297 Data availability

All data used in this paper can be accessed by email to the corresponding author upon request.

300 Author contribution

301	GK conceptualized the study. SL and TK collected the samples. SL performed the analyses. SL
302	and GK wrote the manuscript and all authors contributed to the interpretation and discussion
303	of the results.
304	
305	Competing interests
306	The authors declare that they have no conflict of interest.
307	
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317	References
318	Abril, G., Nogueira, M., Etcheber, H., Cabeçadas, G., Lemaire, E., and Brogueira, M.:
319	Behaviour of organic carbon in nine contrasting European estuaries, Estuarine Coastal Shelf
320	Sci., 54, 241-262, 2002.
321	Andrews, J., Greenaway, A., and Dennis, P.: Combined carbon isotope and C/N ratios as
322	indicators of source and fate of organic matter in a poorly flushed, tropical estuary: Hunts Bay,

323 Kingston Harbour, Jamaica, Estuarine Coastal Shelf Sci., 46, 743-756, 1998.

- Baker, A., and Inverarity, R.: Protein-like fluorescence intensity as a possible tool for
 determining river water quality, Hydrol. Process., 18, 2927-2945, 2004.
- Baker, A., and Spencer, R. G.: Characterization of dissolved organic matter from source to sea
 using fluorescence and absorbance spectroscopy, Sci. Total Environ., 333, 217-232, 2004.
- Bauer, J. E. and Bianchi, T. S.: Dissolved organic carbon cycling and transformation, in:
 Treatise on estuarine and coastal science, edited by: Wolanski, E. and Mcluski, D. S., 5, 7-67,
 Academic Press, Waltham, 2011.
- 331 Beaupré, S. R.: Chapter 6 The Carbon Isotopic Composition of Marine DOC, in:
- Biogeochemistry of Marine Dissolved Organic Matter (Second Edition), edited by: Hansell, D.
- A., and Carlson, C. A., Academic Press, Boston, 335-368, 2015
- Benner, R., and Opsahl, S.: Molecular indicators of the sources and transformations of dissolved organic matter in the Mississippi river plume, Org. Geochem., 32, 597-611, 2001.
- Carlson, C. A., and Hansell, D. A.: Chapter 3 DOM Sources, Sinks, Reactivity, and Budgets,
- in: Biogeochemistry of Marine Dissolved Organic Matter (Second Edition), edited by: Hansell,
- 338 D. A., and Carlson, C. A., Academic Press, Boston, 65-126, 2015
- Cauwet, G.: Chapter 12 DOM in the Coastal Zone, in: Biogeochemistry of Marine Dissolved
 Organic Matter, edited by: Hansell, D. A., and Carlson, C. A., Academic Press, San Diego,
 579-609, 2002.
- 342 Cawley, K. M., Ding, Y., Fourqurean, J., and Jaffé, R.: Characterising the sources and fate of dissolved organic matter in Shark Bay, Australia: a preliminary study using optical properties 343 344 and stable carbon isotopes, Mar. Freshwater Res., 63, 1098-1107, https://doi.org/10.1071/MF12028, 2012. 345

- Chen, R. F., Bissett, P., Coble, P., Conmy, R., Gardner, G. B., Moran, M. A., Wang, X., Wells,
- M. L., Whelan, P., and Zepp, R. G.: Chromophoric dissolved organic matter (CDOM) source
 characterization in the Louisiana Bight, Mar. Chem., 89, 257-272, 2004.
- 349 Chen, M., Kim, J.-H., Nam, S.-I., Niessen, F., Hong, W.-L., Kang, M.-H., and Hur, J. J.:
- Production of fluorescent dissolved organic matter in Arctic Ocean sediments, Sci. Rep. 6,
 39213, 2016.
- Coble, P. G., Green, S. A., Blough, N. V., and Gagosian, R. B.: Characterization of dissolved
 organic matter in the Black Sea by fluorescence spectroscopy, Nature, 348, 432, 1990b.
- 354 Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation-
- emission matrix spectroscopy, Mar. Chem., 51, 325-346, 1996.
- Coble, P. G., Del Castillo, C. E., and Avril, B.: Distribution and optical properties of CDOM in
 the Arabian Sea during the 1995 Southwest Monsoon, Deep Sea Research Part II: Topical
 Studies in Oceanography, 45, 2195-2223, 1998.
- Coble, P. G.: Marine optical biogeochemistry: the chemistry of ocean color, Chemical reviews,
 107, 402-418, 2007.
- Coffin, R. B., and Cifuentes, L. A.: Stable isotope analysis of carbon cycling in the Perdido
 Estuary, Florida, Estuaries, 22, 917-926, 1999.
- 363 Dalmagro, H. J., Lathuillière, M. J., Sallo, F. d. S., Guerreiro, M. F., Pinto, O. B., de Arruda, P.
- 364 H., Couto, E. G., and Johnson, M. S.: Streams with Riparian Forest Buffers versus
- 365 Impoundments Differ in Discharge and DOM Characteristics for Pasture Catchments in
- 366 Southern Amazonia, Water, 11, 390, 2019.

De Troyer, I., Bouillon, S., Barker, S., Perry, C., Coorevits, K., and Merckx, R.: Stable isotope
analysis of dissolved organic carbon in soil solutions using a catalytic combustion total organic
carbon analyzer-isotope ratio mass spectrometer with a cryofocusing interface, Rapid Commun.
Mass Spectrom., 24, 365-374, 2010.

- Deines, P.: Chapter 9 The isotopic composition of reduced organic carbon, in Handbook of
 environmental isotope geochemistry, edited by Fritz, P., and Fontes, J. Ch., Elsevier Science,
 373 329-406, 1980.
- Faganeli, J., Malej, A., Pezdic, J., and Malacic, V.: C: N: P ratios and stable c-isotopic ratios as
 indicators of sources of organic-matter in the gulf of trieste (northern adriatic), Oceanolog.
 Acta, 11, 377-382, 1988.
- Fry, B., Hopkinson, C. S., Nolin, A., and Wainright, S. C.: ¹³C/¹²C composition of marine
 dissolved organic carbon, Chem. Geol., 152, 113-118, 1998.
- Gao, L., Fan, D., Li, D., and Cai, J.: Fluorescence characteristics of chromophoric dissolved
 organic matter in shallow water along the Zhejiang coasts, southeast China, Mar. Environ. Res.,
 69, 187-197, 2010.
- 382 Griffith, D. R., Barnes, R. T., and Raymond, P. A.: Inputs of fossil carbon from wastewater 383 treatment plants to US rivers and oceans, Environ. Sci. Technol., 43, 5647-5651, 2009.
- Griffith, D. R. and Raymond, P. A.: Multiple-source heterotrophy fueled by aged organic
 carbon in an urbanized estuary, Mar. Chem., 124, 14-22, 2011.
- Hansell, D. A. and C. A. Carlson: Biogeochemistry of Marine Dissolved Organic Matter.
- 387 Academic Press, San Diego, 774 pp. 2002.

- Hong, S. H., Kannan, N., Jin, Y., Won, J. H., Han, G. M., and Shim, W. J.: Temporal trend,
 spatial distribution, and terrestrial sources of PBDEs and PCBs in Masan Bay, Korea, Mar.
 Pollut. Bull., 60, 1836-1841, 2010.
- Hudson, N., Baker, A., and Reynolds, D.: Fluorescence analysis of dissolved organic matter in
- natural, waste and polluted waters—a review, River Res. Appl., 23, 631-649, 2007.
- Jaffé, R., Boyer, J., Lu, X., Maie, N., Yang, C., Scully, N., and Mock, S.: Source characterization of dissolved organic matter in a subtropical mangrove-dominated estuary by fluorescence analysis, Mar. Chem., 84, 195-210, 2004.
- Jiao, N., Herndl, G. J., Hansell, D. A., Benner, R., Kattner, G., Wilhelm, S. W., Kirchman, D.
- L., Weinbauer, M. G., Luo, T., and Chen, F.: Microbial production of recalcitrant dissolved
 organic matter: long-term carbon storage in the global ocean, Nat. Rev. Microbiol., 8, 593-599,
 2010.
- Kannan, N., Hong, S. H., Yim, U. H., Kim, N. S., Ha, S. Y., Li, D., and Shim, W. J.: Dispersion
 of organic contaminants from wastewater treatment outfall in Masan Bay, Korea, J. Toxicol.
- 402 Environ. Health Sci., 2, 200-206, 2010.
- Kelley, C. A., Coffin, R. B., and Cifuentes, L. A.: Stable isotope evidence for alternative
 bacterial carbon sources in the Gulf of Mexico, Limnol. Oceanogr., 43, 1962-1969, 1998.
- Kim, T. -H., Waska, H., Kwon, E., Suryaputra, I. G. N., and Kim, G.: Production, degradation,
- and flux of dissolved organic matter in the subterranean estuary of a large tidal flat, Mar. Chem.,
 142, 1-10, 2012.
- 408 Kim, T. -H., Kim, G., Lee, S. A., and Dittmar, T.: Extraordinary slow degradation of Dissolved
- 409 Organic Carbon (DOC) in a cold marginal sea, Sci. Rep., 5, 13808, 2015.

- Lamb, A. L., Wilson, G. P., and Leng, M. J.: A review of coastal palaeoclimate and relative
 sea-level reconstructions using δ¹³C and C/N ratios in organic material. Earth-Sci. Rev, 75(14), 29-57, 2006.
- Lang, S. Q., Lilley, M. D., and Hedges, J. I.: A method to measure the isotopic (¹³C)
 composition of dissolved organic carbon using a high temperature combustion instrument, Mar.
 Chem., 103, 318-326, 2007.
- Lee, C.-W., and Min, B.-Y.: Pollution in Masan Bay, a matter of concern in South Korea, Mar.
 Pollut. Bull., 21, 226-229, 1990.
- 418 Lee, H. J., Hong, S. H., Kim, M., Ha, S. Y., An, S. M., and Shim, W. J.: Tracing origins of
- sewage and organic matter using dissolved sterols in Masan and Haengam Bay, Korea, Ocean
 Sci. J., 46, 95-103, 2011.
- Lee, S.-A., and Kim, G.: Sources, fluxes, and behaviors of fluorescent dissolved organic matter
 (FDOM) in the Nakdong River Estuary, Korea, Biogeosciences, 15, 1115-1122, 2018.
- Lee, Y.-W., Hwang, D.-W., Kim, G., Lee, W.-C., and Oh, H.-T.: Nutrient inputs from submarine groundwater discharge (SGD) in Masan Bay, an embayment surrounded by heavily industrialized cities, Korea, Sci. Total Environ., 407, 3181-3188, 2009.
- Liu, C., Du, Y., Yin, H., Fan, C., Chen, K., Zhong, J., and Gu, X.: Exchanges of nitrogen and phosphorus across the sediment-water interface influenced by the external suspended particulate matter and the residual matter after dredging, Environ. Pollut., 246, 207-216, 2019.
- 429 Lobbes, J. M., Fitznar, H. P., and Kattner, G.: Biogeochemical characteristics of dissolved and
- 430 particulate organic matter in Russian rivers entering the Arctic Ocean, Geochim. Cosmochim.
- 431 Acta, 64, 2973-2983, 2000.

- Lu, Y., Edmonds, J. W., Yamashita, Y., Zhou, B., Jaegge, A., and Baxley, M.: Spatial variation
 in the origin and reactivity of dissolved organic matter in Oregon-Washington coastal waters,
 Ocean Dyn., 65, 17-32, 2015.
- 435 Markager, S., Stedmon, C. A., and Søndergaard, M.: Seasonal dynamics and conservative
- 436 mixing of dissolved organic matter in the temperate eutrophic estuary Horsens Fjord, Estuarine
- 437 Coastal Shelf Sci., 92, 376-388, 2011.
- Mayer, L. M., Schick, L. L., and Loder, T. C.: Dissolved protein fluorescence in two Maine
 estuaries, Mar. Chem., 64, 171-179, 1999.
- 440 McCallister, S. L., Bauer, J. E., Ducklow, H. W., and Canuel, E. A.: Sources of estuarine
- dissolved and particulate organic matter: a multi-tracer approach, Org. Geochem., 37, 454-468,
 2006.
- Milliman, J. D., Qinchun, X., and Zuosheng, Y.: Transfer of particulate organic carbon and
 nitrogen from the Yangtze River to the ocean, Am. J. Sci., 284, 824-834, 1984.
- Murphy, K. R., Stedmon, C. A., Waite, T. D., and Ruiz, G. M.: Distinguishing between
 terrestrial and autochthonous organic matter sources in marine environments using
 fluorescence spectroscopy, Mar. Chem., 108, 40-58, 2008.
- Murphy, K. R., Hambly, A., Singh, S., Henderson, R. K., Baker, A., Stuetz, R., and Khan, S. J.:
 Organic matter fluorescence in municipal water recycling schemes: toward a unified
 PARAFAC model, Environ. Sci. Technol., 45, 2909-2916, 2011.
- Murphy, K. R., Stedmon, C. A., Wenig, P., and Bro, R.: OpenFluor–an online spectral library
 of auto-fluorescence by organic compounds in the environment, Anal. Methods, 6, 658-661,
 2014.

- Oh, Y. H., Lee, Y.-W., Park, S. R., and Kim, T.-H.: Importance of dissolved organic carbon flux
 through submarine groundwater discharge to the coastal ocean: Results from Masan Bay, the
 southern coast of Korea, J. Mar. Syst., 173, 43-48, 2017.
- Osburn, C. L., and Stedmon, C. A.: Linking the chemical and optical properties of dissolved
 organic matter in the Baltic-North Sea transition zone to differentiate three allochthonous
 inputs, Mar. Chem., 126, 281-294, 2011.
- Osburn, C. L., Wigdahl, C. R., Fritz, S. C., and Saros, J. E.: Dissolved organic matter
 composition and photoreactivity in prairie lakes of the US Great Plains, Limnol. Oceanogr., 56,
 2371-2390, 2011.
- Panetta, R. J., Ibrahim, M., and Gélinas, Y.: Coupling a High-Temperature Catalytic Oxidation
 Total Organic Carbon Analyzer to an Isotope Ratio Mass Spectrometer To Measure NaturalAbundance δ13C-Dissolved Organic Carbon in Marine and Freshwater Samples, Anal. Chem.,
 80, 5232-5239, 2008.
- Pradhan, U., Wu, Y., Shirodkar, P., Zhang, J., and Zhang, G.: Sources and distribution of
 organic matter in thirty five tropical estuaries along the west coast of India-a preliminary
 assessment, Estuarine Coastal Shelf Sci., 151, 21-33, 2014.
- 470 Raymond, P. A., and Bauer, J. E.: DOC cycling in a temperate estuary: a mass balance approach
 471 using natural 14C and 13C isotopes, Limnol. Oceanogr., 46, 655-667, 2001.
- 472 Spiker, E.: The Behavior of C-14 and C-13 in Estuarine Water-Effects of Insitu Co2 Production
- and Atmospheric Exchange, Radiocarbon, 22, 647-654, 1980.

Thornton, S., and McManus, J.: Application of organic carbon and nitrogen stable isotope and
C/N ratios as source indicators of organic matter provenance in estuarine systems: evidence
from the Tay Estuary, Scotland, Estuarine Coastal Shelf Sci., 38, 219-233, 1994.

- 477 Tremblay, L., and Benner, R.: Microbial contributions to N-immobilization and organic matter
- 478 preservation in decaying plant detritus, Geochim. Cosmochim. Acta, 70, 133-146, 2006.
- 479 Twardowski, M. S., and Donaghay, P. L.: Separating in situ and terrigenous sources of
 480 absorption by dissolved materials in coastal waters, J. Geophys. Res, 106, 2545-2560, 2001.
- Wang, X.-C., Chen, R. F., and Gardner, G. B.: Sources and transport of dissolved and
 particulate organic carbon in the Mississippi River estuary and adjacent coastal waters of the
 northern Gulf of Mexico, Mar. Chem., 89, 241-256, 2004.
- Wünsch, U. J., Murphy, K. R., and Stedmon, C. A.: The one-sample PARAFAC approach
 reveals molecular size distributions of fluorescent components in dissolved organic matter,
 Environ. Sci. Technol., 51, 11900-11908, 2017.
- Ya, C., Anderson, W., and Jaffé, R.: Assessing dissolved organic matter dynamics and source
 strengths in a subtropical estuary: Application of stable carbon isotopes and optical properties,
 Cont. Shelf Res., 92, 98-107, 2015.
- Yoo, K.: Population dynamics of dinoflagellate community in Masan Bay with a note on the
 impact of environmental parameters, Mar. Pollut. Bull., 23, 185-188, 1991.
- Zhang, Y., van Dijk, M. A., Liu, M., Zhu, G., and Qin, B.: The contribution of phytoplankton
 degradation to chromophoric dissolved organic matter (CDOM) in eutrophic shallow lakes:
 field and experimental evidence, Water Res., 43, 4685-4697, 2009.

Table 1. Salinity, DOC, FDOM_H, FDOM_P, and δ^{13} C-DOC in surface water of Masan Bay in 496 August 2011 and August 2016.

Sampling	station	salinity	DOC	FDOM _H	FDOM _T	δ ¹³ C- DOC	DON	DOC/DON
			(µM)	(ppbQSE)	(ppbQSE)	(‰)	(µM)	
Aug. 2011	M1	14.0	148	6.7	13.6	-25.4	12	12
	M1-1	12.8	151	9.2	14.3	-24.3	7	21
	M2	10.2	157	9.0	5.4	-24.6	11	14
	M3	16.3	147	8.2	14.7	n/a	16	9
	M4-1	19.0	186	7.1	78.7	-21.9	13	15
	M4-2	18.6	155	6.9	8.3	-21.6	10	15
	M5-1	17.7	138	4.5	4.5	-23.3	24	6
	M5-2	18.4	133	5.8	20.9	-24.5	11	12
	M5-3	18.9	135	8.0	11.3	-23.7	13	11
	M6	18.4	146	6.6	24.8	-23.3	19	8
	M6-1	19.2	142	5.5	7.4	n/a	9	15
	M7-1	19.5	157	5.8	10.5	-20.6	11	15
	M7-2	18.9	148	5.6	9.6	-21.5	12	12
	M8	19.5	152	5.6	7.6	-21.5	15	10
	M9	18.8	149	5.6	14.5	-21.9	10	15
	M9-1	19.1	154	5.1	10.2	-21.0	12	13
	M9-2	20.8	106	3.6	13.1	-22.0	8	13
Aug. 2016	M1	29.2	191	2.7	4.8	-22.8	15	13
	M2	29.9	164	2.0	3.4	-21.1	7	22
	M3	26.0	155	2.5	3.8	-28.8	8	19
	M4	27.4	149	1.9	3.5	-22.6	9	17
	M5	25.5	165	1.8	3.3	-23.5	10	16
	M6	30.5	147	1.1	3.0	-23.7	6	26
	M7	31.4	166	1.3	4.4	-26.2	4	45
	M8	32.0	123	0.8	2.3	-23.7	5	26
	M9	32.0	146	0.6	2.1	-24.4	5	30
	M10	31.9	130	0.7	2.7	-25.0	3	39

n/a = not available.

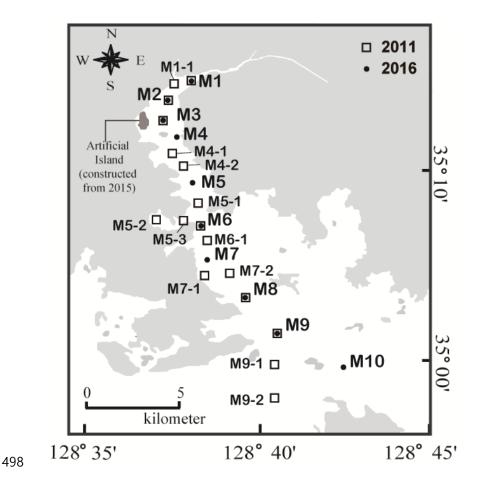
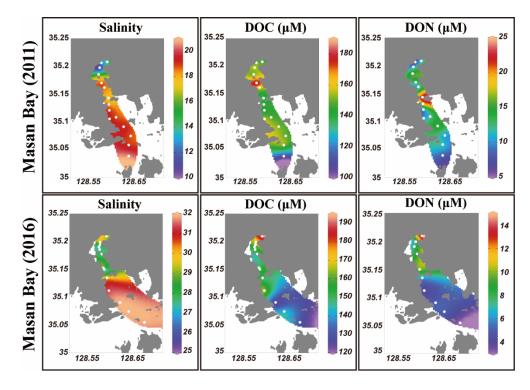


Figure 1. A map showing the sampling stations for DOC, δ^{13} C-DOC, FDOM, and DOC/DON

ratio in Masan Bay, Korea, in 2011 and 2016.



502 Figure 2. Surface distributions of salinity, DOC, and DON in Masan Bay, Korea, in 2011 and

503 2016.

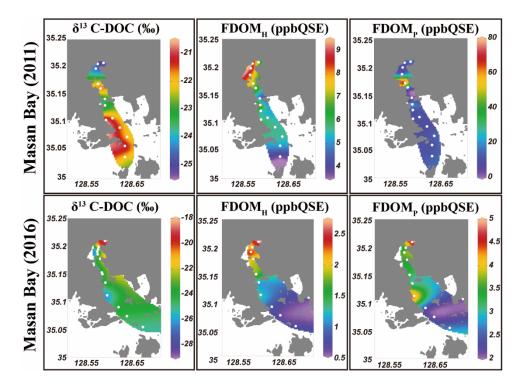


Figure 3. Surface distributions of δ^{13} C-DOC, FDOM_H, and FDOM_P in Masan Bay, Korea, in

506 2011 and 2016.

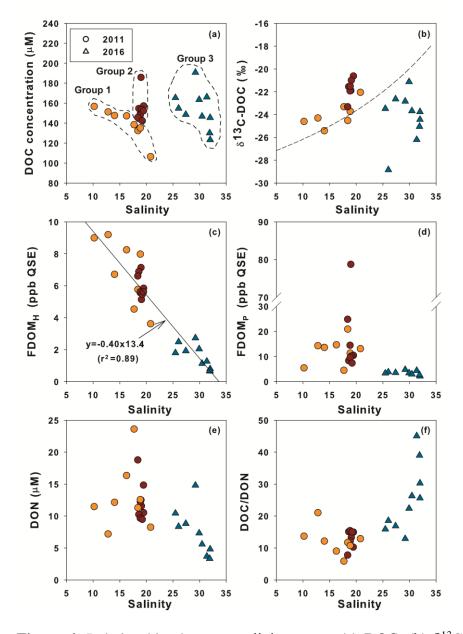
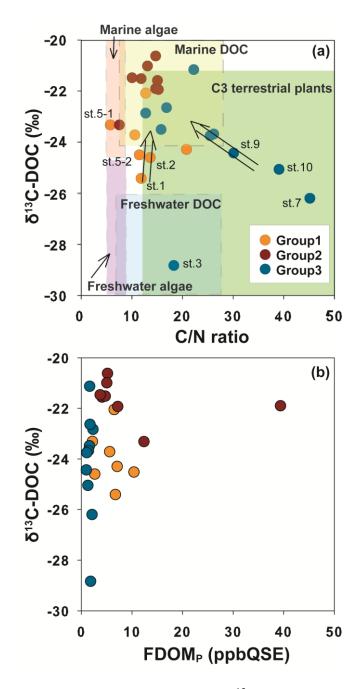


Figure 4. Relationships between salinity versus (a) DOC, (b) δ^{13} C-DOC, (c) FDOM_H, (d) FDOM_T, (e) DON, and (f) DOC/DON values. The DOC concentrations are divided into three groups based on probable different sources (in the dashed circles). The dashed line (b) represents the binary conservative mixing line for δ^{13} C-DOC between the terrestrial endmember and the marine end-member. The solid line (c) represents a linear regression fit of the data.

514



516 **Figure 5.** Relationships between δ^{13} C-DOC values versus (a) DOC/DON ratio and (b) FDOM_P 517 in Masan Bay, Korea. The ranges of DOC/DON ratio and δ^{13} C-DOC values for each group are 518 based on the values reported by Lamb et al. (2006) and Beaupré (2015).