

Dear Reviewers,

Thank you for your patience and your comments on the manuscript. We are grateful for your time and believe the feedback has improved the manuscript. We hope you find it suitable for publication.

This version has incorporated all of your comments. As suggested, we developed and reformatted section 5.2 following the structure proposed by J. Farmer and we believe that it is written more clearly and will be easier for the reader to follow.

As suggested by M. Henehan, due to evidence of size-fraction effects on the  $\delta^{11}\text{B}_{\text{carbonate}}$  for *G. ruber*, we made a linear regression based only on smaller size fractions. The resulting calibration did not show a significant change in the sensitivity of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  compared to culture experiments (Henehan et al., 2013) but showed an offset of  $\sim -0.4\%$ .

The regression on smaller size fractions for *T. sacculifer* did not show a significant offset from the calibration from a culture study (Martinez-Boti et al., 2015b), which supports our discussion for *T. sacculifer* in the WEP.

The anonymous reviewer (AR) was concerned that although both *O. universa* and *T. sacculifer* have overlapping calcification depths at sites FC-02a and WP07-a, they record different  $\delta^{11}\text{B}_{\text{carbonate}}$  signatures. We attribute the difference in  $\delta^{11}\text{B}_{\text{carbonate}}$  to the different photosymbiosis characteristics of each species, and discuss this.

We also thank the AR for the note regarding gender assumptions during this process.

Best, Regards,

Report #1

Submitted on 02 Dec 2019

Referee #1: Jesse Farmer, jesse.farmer@princeton.edu

Revision review by Jesse Farmer

Guillermic et al. have made great progress in revising their manuscript. The introduction, background, materials and methods and results are publishable with very minor revisions. However, the manuscript loses focus and cohesiveness in the discussion, where the text becomes quite difficult to read and understand in its current iteration. I do note that, to my reading, this is an issue only with the writing presentation; the technical aspects of the discussion are sound and the figures are generally excellent and supportive of the text. Still, unfortunately I cannot yet recommend this for publication until the discussion text is improved. I encourage the coauthors to churn through a few rounds of revision to the discussion, with a particular focus on English grammar, and have provided detailed comments below that hopefully aid their work. I look forward to approving the revised manuscript for publication.

L84. Change to “samples from sites that are currently in quasi-equilibrium with the atmosphere” (less ambiguous)

**Changed**

L125-127. Suggest change to “Furthermore,  $\delta^{11}\text{B}$  differences between foraminifera species inhabiting waters of the same pH makes the acquisition of more coretop and culture data essential for applications of the  $\delta^{11}\text{B}$ -pH proxy.”

**Changed**

L149. State here which fractionation value you use in this study, e.g., “We use the fractionation of 27.2 ‰ from Klochko et al. (2006) in this study.”

**Changed**

L176-177. Be a little more specific here, e.g., “The extent to which these results apply to the planktonic foraminifera studied here are not known. Nonetheless, pH modulation of the calcifying fluid may influence the  $\delta^{11}\text{B}$  of planktonic foraminifera.”

**Changed**

Section 2.5. Please clarify- To what extent are these depth and habitat preferences global? Do foraminifera can show regional deviations in their depth & habitat preferences? Do we know? Many of the referenced studies of foraminifera habitats focus on the tropical/subtropical Atlantic. “Although the studies listed above showed evidence for species-specific living depth-habitat affinities, recent direct observations showed that environmental conditions (e.g. temperature, light) was locally responsible for the variability in the living depth of certain foraminifera species in the eastern North Atlantic (Rebotim et al., 2017). The same study showed evidence for a correspondence between living depth habitat and indirectly-derived calcification depth, supporting the approach utilized in this study. ”

L232. Change “drilled” to ”recovered” or “cored”. It is only appropriate to say “drilled” when a drilling system was used for core recovery (as is the case for ODP/IODP)

Changed for “cored”

L255. Good points on the response to my comment; please incorporate your response into a sentence in the manuscript here.

L256. “Hydrochloric acid was used to allow complete dissolution of the sample including Fe-Mn oxide and hydroxides if present. No matrix effect resulting from the mix HCl/HF was observed on the  $\delta^{11}\text{B}$ .”

L266. Typo, should read “3.5”

Fixed

L269. Change to “70 uL of carbonate sample dissolved in 1N HCl was loaded...”. This gets around the ambiguity of acid used in my above comment on L256-257.

Changed

L289. Remove “internal”

Removed

L291. Change “boron isotopes liquid standard” to “boric acid standard”

Changed

L303. Again on the 1N HCl dissolution from L256-257. If I follow your protocol, you have a foram aliquot dissolved in 1N HCl, to which you have added HNO<sub>3</sub> and HF for the ICP-MS measurement. Was HCl added to the standards to properly matrix-match? Or was the volume of HCl sufficiently small that you ignored it? Please specify.

The volume of HCl was sufficiently small (<1%) to be ignored.

Line 205 “any matrix effect (Misra et al., 2014b), the remaining HCl (<1%) was negligible.”

L339. Suggest “during their growth” in lieu of “along their ontogeny”

Removed instead following reviewer 2 comment. “We applied (based on uncertainties of our measurements) an uncertainty of  $\pm 10\text{m}$  for calcification depths  $> 70\text{ m}$  and an uncertainty of  $\pm 20\text{ m}$  when calcification depths  $< 70\text{ m}$ .”

L339-341. Are these uncertainties truly representative of foram depth migration during their lifespan, or are they more indicative of uncertainty in different measurements of foram depth habitats? Please specify.

“Direct observations of living depths of foraminifera remain limited. However, the depth uncertainties reported here are in line with the uncertainties calculated based on direct observations in the eastern North Atlantic which give a standard error on average living depths ranging from 6-22 m for the same species (Rebotim et al., 2017).”

Section 4.1. This is a fair response to my comment; please incorporate this response into the text by adding a paragraph in Section 3.8. It is certainly fine to use a best guess and reasoned approach to derive CDs, as long as the reader is aware of your approach. (Note- it might also

help to mock up a figure showing the data in Tables S6 and S7).

“Because both methods have their uncertainties (in one case, use of taxon-specific calibrations, and in the other, analytical limitations), both estimates of calcification depth were compared to published values for the basin, and where available, for the same site (Table S6). To select which calcification depth to use for further calculations, we first looked at CD<sub>1</sub>, CD<sub>2</sub> and CD<sub>3</sub>. If, two CDs were similar we selected that one, if CD<sub>1</sub> and CD<sub>2</sub> were different we chose literature values (CD<sub>3</sub>) when available. For some less studied species, like *G. tumida*, *G. menardii* or *P. obliquiloculata*, CD<sub>3</sub> was not always available but showed good correspondence with our CD<sub>2</sub>, moreover due to availability of Mg/Ca-temperature taxon-specific calibrations we preferentially use CD<sub>2</sub> for those species.”

#### Section 4.2

To what extent are your foraminifera samples and those from the literature in Fig. 5 from different size fractions? If so, given the noted  $\delta^{11}\text{B}$  differences between size fractions, is it worth regressing against literature data from different size fractions? How much of the “large uncertainty given variability in the data” (L399) might reflect size class differences? Please address in the text.

We have added more informations to the section 4.2.1. for *G. ruber*. Results can be in line with a size fraction effect for *G. ruber*, but not for *T. sacculifer*. However, no significant differences are observed yet between the different calibrations due to the limited datasets.

Starting line 400: “Samples were picked from the 250-300  $\mu\text{m}$  fraction, except for the WEP sites where they were picked from the 250-400  $\mu\text{m}$  fraction. Weight per shell averaged  $11 \pm 4 \mu\text{g}$  (n=4, SD) although the weight was not measured on the same sub-sample analyzed for  $\delta^{11}\text{B}$  and trace elements or at the WEP sites. In comparison to literature, the size fraction used for this study was smaller: Foster et al. (2008) used the 300-355 $\mu\text{m}$  fraction, Henehan et al. (2013) utilized multiple size fractions (250-300, 250-355, 300-355, 355-400 and 400-455  $\mu\text{m}$ ) and Raitzsch et al. (2018) used the 315-355  $\mu\text{m}$  fraction.

Our results for *G. ruber* (Fig. 5) are in close agreement with published data from other core-tops, sediment traps, tows, and culture experiments for  $\delta^{11}\text{B}_{\text{borate}} > 19 \text{‰}$  (Foster et al., 2008, Henehan et al., 2013, Raitzsch et al., 2018). However, the two datapoints from  $\delta^{11}\text{B}_{\text{borate}} < 19 \text{‰}$  are lower compared to previous studies. Elevated  $\delta^{11}\text{B}_{\text{carbonate}}$  values relative to  $\delta^{11}\text{B}_{\text{borate}}$  has been explained by the high photosynthetic activity (Hönisch et al., 2003; Zeebe et al., 2003). Three calibrations have been derived (Table 3). Linear regression on our data alone yields a slope of  $1.12 (\pm 1.67)$ . While this regression is not significantly different from a 1:1 line, the uncertainty term are significant given limited data in our study. Therefore, the sensitivity of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  of our linear regression is not statistically different from 1, the uncertainty on this regression is important due to our small dataset, thus not inconsistent with the low sensitivity trend of the culture experiments from Sanyal et al., (2001) or Henehan et al., (2013). The second calibration made compiling all data from literature shows a sensitivity similar (e.g.  $0.46 (\pm 0.34)$ ) to the one recently published by Raitzsch et al., (2018) (e.g.  $0.45 (\pm 0.16)$ , Table 3). The third linear regression made only on data from the 250-400  $\mu\text{m}$  fraction from our study and from the 250-300  $\mu\text{m}$  from Henehan et al. (2013) yields a slope of  $0.58 (\pm 0.91)$  similar to culture experiments from Henehan et al., (2013) (e.g.  $0.6 (\pm 0.16)$ , Table 3). This third calibration is

offset by  $\sim -0.4\%$  ( $p > 0.05$ ) compared to culture calibration from Henehan et al. (2013). The variability in our weight per shell based data from Henehan et al., (2013) can potentially imply a deviation down to  $1\%$  relative to its calibration line, which can be in line with the maximum deviation observed in our data ( $\sim 1.2\%$ ) and not inconsistent with a size effect explaining the offset in our calibration.

Line 431: “It is also noticeable that *T. sacculifer* (w/o sacc) samples from the WEP have a  $\delta^{11}\text{B}_{\text{carbonate}}$  close to expected  $\delta^{11}\text{B}_{\text{borate}}$  and are significantly lower compared to the combined *T. sacculifer* of other sites ( $p=0.01$ , unpaired t-test). When doing the regression using data from the 250-400  $\mu\text{m}$  fraction, our results are not significantly different from the regression through data that combine all size fractions (Fig. 5).

Also, suggest changing “fall (above/below) the 1:1 line” to “exhibit (higher/lower)  $\delta^{11}\text{B}$  compared to expected  $\delta^{11}\text{B}_{\text{borate}}$  at their collection location” throughout the text. It is less confusing and more precise. (see L396, 400)

Changed throughout the text

Section 4.2.1 *G. ruber*. This section is confusing as currently written. Some rephrasing ideas:

- L383: Rephrase to “However, our two datapoints from  $\delta^{11}\text{B}_{\text{borate}} < 19\%$  are lower compared to previous studies.” “Lighter  $\delta^{11}\text{B}$ ” is not correct; use either “lighter B isotopic composition” or “lower  $\delta^{11}\text{B}$ ”.

Changed

- L383-384: Regarding the comment from M. Henehan and response, are your samples from different size fractions than

I think I have given more informations in this section.

- L384-385: “Whilst...” Delete this unnecessary sentence. Deleted

- L385-386: “The positive offset from the 1:1 curve...” change to say elevated  $\delta^{11}\text{B}$  foram values relative to  $\delta^{11}\text{B}_{\text{borate}}$ . Changed

- If you want to discuss the properties of the calibration curves, you should state the curves in the text for the reader. You should say something like you do for *T. sacculifer* on L396-397: “Linear regression on our data alone yields a regression of \_\_\_\_\_. While this regression is not significantly different from a 1:1 line, the uncertainty terms are significant given limited data in our study. Therefore, our data are not inconsistent with the low sensitivity trend.... (append the rest of the paragraph as written)”.

Line 410: “Three calibrations have been derived (Table 3). Linear regression on our data alone yields a slope of  $1.12 (\pm 1.67)$ . While this regression is not significantly different from a 1:1 line, the uncertainty term are significant given limited data in our study. Therefore, the sensitivity of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  of our linear regression is not statistically different from 1, the uncertainty on this regression is important due to our small dataset, thus not inconsistent with the low sensitivity trend of the culture experiments from Sanyal et al., (2001) or Henehan et al., (2013). The second calibration made compiling all data from literature shows a sensitivity similar (e.g.  $0.46 (\pm 0.34)$ ) to the one recently published by Raitzsch et al., (2018) (e.g.  $0.45 (\pm 0.16)$ , Table 3). The third linear regression made only on data from the 250-400  $\mu\text{m}$  fraction from our study and from the 250-300  $\mu\text{m}$  from Henehan et al. (2013) yields a slope of  $0.58 (\pm 0.91)$  similar to culture experiments from Henehan et al., (2013) (e.g.  $0.6 (\pm 0.16)$ , Table 3).”

L400. Remove “below the 1:1 line” as no data are significantly below this line. “close to” is fine. Can you indicate the WEP samples on this plot with a star or other indication?  
Changed, I have highlighted the WEP samples in Fig. 5 for *T. sacculifer*.

L408-409. Rephrase to “, and is not significantly different from ( $p > 0.05$ ) the *O. universa* calibration previously reported by Hennehan et al. (2016) ( $0.95 \pm 0.17$ ).”  
Changed

L413. “For *O. universa* and all deep-dwelling species,”  
Changed

L417. Change “may remain” to “remains”  
Changed

L430. Typo *T. sacculifer* (sacc)  
Fixed

L432-433. What do you mean by interspecific B/Ca ratios? Please elaborate. I do not think it comes as any surprise that B/Ca ratios are different in different foraminifera species.  
This has been changed to: “This study supports species-specific B/Ca ratios as previously published (Yu et al., 2007; Tripathi et al., 2009, 2011; Allen and Hönisch, 2012; Hennehan et al., 2016).”

L441-444. Confusing. Please split into two sentences, with one about core site depth and one about calcification depth. Also please note that you see a weak decrease in B/Ca with increasing calcification depth, although it is significant ( $p < 0.05$ ).  
“When comparing data from all sites together, a weak decrease in B/Ca with increasing calcification depth is observed ( $R^2 = 0.11$ ,  $p < 0.05$ , Fig. S4). A correlation also exists between B/Ca and the water depths of the cores (not significant, Fig. S4).”

L454. Typo (w/o sacc)  
Fixed

L487-488. Instead of saying seasonality is not important, rephrase to “seasonality is of relatively minor impact on the carbonate system parameters at the sites we examined.”  
Changed

L499-503. Specify symbiont photosynthesis  
Added

L502-503. lower/lowest  
Changed: “Dinoflagellate-bearing foraminifera (*G. ruber*, *T. sacculifer* and *O. universa*) tend to have a higher symbiont density and photosynthesis activity while *P. obliquiloculata*, *G. menardii* and *N. dutertrei* have lower symbiont density and *P. obliquiloculata*, *N. dutertrei* lowest photosynthetic activity (Takagi et al., 2019).”

L508-549. This is difficult to understand as written and needs revising, otherwise these points will be completely lost by the reader.

L508-513. Present items in logical order: First what you observe (low d11B of deep-dwelling species relative to d11B borate), then context (lower symbiont density and photosynthetic activity in these forams), and combine this into an interpretation (lower symbiont activity leads to lower microenvironment pH and may explain the low d11B of these taxa).

**We reformatted this section according to your suggestions.**

L513. “and” instead of “et”

**Changed**

L514 and throughout. Do not use “they” or any pronouns, as it is not clear to what you are referring. Use the noun itself. Here, “they” = “symbionts”

**Changed**

L518. Be specific. “A deeper depth habitat will reduce the light intensity the symbionts receive, and as a consequence may lower symbiont photosynthetic activity, possibly reducing pH in the microenvironment surrounding the foraminifera”.

**Changed**

L521. What does “basically support” mean? Either the trend supports the fact or it does not.

**Fixed**

L524. Start a new paragraph here. “To test if the d11B signature was inferred to a light driven”- what does this mean?

**I changed this section according to your previous comment.**

L524-525. change to “we have independently calculated foraminifera (calcification?) depth based on various light insolation culture experiments and the microenvironment  $\Delta$ pH derived from our data”.

**Changed**

L535. Change to Microenvironment  $\Delta$ pH

**Changed**

L537-541. Please rephrase these sentences. I think this may be a key point, but I cannot follow it as currently written.

**We have edited this paragraph.**

L542. Change to “*G. menardii* and *G. tumida* are similar...”

**Changed**

L547-549. Need to add some commas here to make this understandable.

**We changed for: “We can also note that *P. obliquiloculata*, which has the lowest symbiont density and photosynthetic activity (Takagi et al., 2019), has the lowest microenvironment pH**

compared to other deeper-dweller species, supporting this respiration driven microenvironment”

L555-557. “more borate ion may be incorporated... more boric acid may be incorporated.” This is just a hypothesis.

Changed

L560-572. Again, this is difficult to read as currently written. Please revise and particularly correct English grammar.

Edited

L582. Change to “which supports previous paleo-reconstructions using existing calibrations of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$ ”.

Changed

L583. Which observations are these from Henehan et al. (2016)? Be specific.

Edited: “Our  $\delta^{11}\text{B}_{\text{carbonate}}$  date and their sensitivity to  $\delta^{11}\text{B}_{\text{borate}}$  for *O. universa* support previous data from Henehan et al., (2016).”

L590-592. Please rephrase and be much much more specific, I have no idea what you are trying to say here. What do you mean “add a weight/shell”? Is 11  $\mu\text{g}$  per shell small or large? Context needed!

I removed this part, as I have developed it in the other sections 4.2.1 and section 5.4.

L594. What correction? What coretop study?

“Results for *G. ruber* are the most scattered, potentially due to difference in test sizes (Henehan et al., 2013), or depth habitat, although we can not exclude undocumented diagenetic effects. Results reaffirm the importance of working with narrow size fractions (Henehan et al., 2013), the utilization of calibrations derived from the same size fraction or use of offsets to take into account this size fraction effect, and the importance of core-top studies before paleo-application.”

L597. Typo on species

“We also find that for two species, the boron proxy is a relatively straightforward recorder of ambient pH, with sensitivities close to unity for *O. universa* and *N. dutertrei*.”

References. Please double check these and their formatting. Some citations missing and formatting issues.

Checked

Figures.

Fig. 2. Change figure and caption to  $\text{B}(\text{OH})_3$  and  $\text{B}(\text{OH})_4^-$  to match main text. Note typo on Figure  $\alpha$ , should be 1.0272.

Changed



Report #2

Submitted on 02 Dec 2019

Anonymous Referee #2

This is my second review of this MS. As it stands, the edits are generally ok, but in some cases the authors chose to only reply to edits in the response and not always insert their response into the MS. Sometimes this was ok, but if a reviewer requests an explanation, the response should generally be incorporated into the manuscript. Reviewer 1 should also review the edits as some of Dr. Farmer's suggestions were also not implemented into the revised MS. I have specific comments below. The new section on symbiosis and the findings of Tagaki was very poorly written and should be edited prior to final publication.

**We have reviewed the edits and incorporated them, including the explanation for the depth habitat and the cleaning. We also have rewritten section 5.2.**

As an aside: please do not assume that anonymous reviewers are CIS gendered males. The opening response to my review was “We wish to thank this reviewer for HIS thorough review of our manuscript and HIS helpful comments”. Instead: We thank the reviewer for helpful comments and suggestions.

**We thank the reviewer for pointing this out.**

There is no evidence that foraminifers change their symbiont assemblages to ones more similar to what are found in deeper dwelling species. In addition, the paragraph beginning on line 195 where differences in the boron isotope data between *T. sacculifer*, *G. ruber*, and *O. universa* is explored states that *O. universa* exhibits lower  $\delta^{11}\text{B}$  because it lives deeper in the water column, but the depth habitats chosen for these species IN THIS STUDY overlap or are at the same depth according to table S7.

Line 176: Change: “We acknowledge this is speculative as it is based upon benthic foraminifer experiments” to “We acknowledge this process may not be the same for planktic species as these findings were based upon benthic foraminifer experiments”

**This has been changed from “The extent to which these results apply to the planktonic foraminifera studied here are not known. Nonetheless, pH modulation of the calcifying fluid may influence the  $\delta^{11}\text{B}$  of planktonic foraminifera.”**

Section 3.3: the reviewers responded to my comment but did not add any additional text to the MS. The reason the full reductive cleaning protocol was used should be included in this paragraph so that other readers are aware of the study by Misra 2014 and the purpose of the full cleaning protocol.

**This states: “Samples were then cleaned using full reductive and oxidative cleaning (Boyle and Keigwin, 1985; Barker et al., 2003). We utilized the reductive cleaning because some of the sites were not previously studied and previous comparisons have shown no effect on B/Ca (Misra et al., 2014b), nevertheless, Fe-Mn oxide and hydroxides can result in non-negligible content of Mg and B contamination. Overall, the samples did not present high Mn concentration. Reductive cleaning leads to a decrease in Mg/Ca which would result in deeper CDs, which is not the case when comparing with CD1 and CD3, we then no longer assume this decrease problematic for the purpose of this study.”**

Line 339: I suggest removing the clause “As foraminifera can migrate in the water column along their ontogeny” because it is misleading. Foraminifers can occupy a deeper depth habitat at the end of their ontogeny, but as written it still seems to imply that foraminifers migrate up and down, which they likely do not. Removing this clause bears no impact on the rest of the sentence.

Removed

Line 456: Again: I think this should be more carefully written. The use of the word “migrate” with foraminifer depth habitats usually implies that they move up and down in the water column, which they do not. This should be reworded to state that at the end of their life cycle they often transition to deeper waters prior to gametogenesis.

Changed for: “We note that calculation of absolute calcification depths can be challenging in some cases as many species often transition to deeper waters at the end of their life cycle prior to gametogenesis”

Line 495: *G. ruber*, *T. sacculifer* and *O. universa* do not have chrysophyte algal symbionts, only dinoflagellates. There is no mention of chrysophytes in the Anderson and Be paper nor in Spero 1987. *N. dutertrei* has pelagophyte symbionts not chrysophytes, confirmed using genetics, see Bird et al., 2018.

Thank you for those inputs. We have added the pelagophyte symbionts for *N. dutertrei*.

Paragraph beginning line 500 is poorly worded.

The substance does not change, but we have edited it.

Line 509: have microenvironments with lower “pH” than ambient seawater (insert pH)

Changed

Line 513: ‘et’?

Changed

Line 514: insert ‘a’ between also and function.

Added

Line 515: Should “for the purpose of this study” be a new sentence?

Split in two sentences

Line 537: Tagazaki should be Tagaki.

Changed

Line 540: specie is spelled incorrectly

Changed

Paragraph beginning on line 540: *O. universa* in this study occupy similar depths to the *G. ruber* and *G. sacculifer*. Thus, I do not agree with the discussion here.

That is actually where Tagaki’s study is relevant, as it shows that each species has their own characteristics in terms of photosymbiosis. What we found interesting is that potentially, *T. sacculifer*, which seems to have a higher potential for photosynthesis, might also be more sensitive to changes of insolation depending of its habitat in the water column. Also, its photosynthetic activity might be more effective at depth due to its higher symbiont density, reflected in higher  $\delta^{11}\text{B}$  carbonate than *O. universa* which has a lower potential for

photosynthesis **for the same depth.**

Line 560: “*T. sacculifer* has the potential to support more photosynthesis due to its higher symbiont density. Higher photosynthetic activity is observed compared to other species potentially supporting higher symbiont/host interactions (Tagaki et al., 2019). Those results could be in line with a greater sensitivity of *T. sacculifer*’s photosynthetic activity with changes in insolation/water depth.”

Line 577: “The low  $\delta^{11}\text{B}_{\text{carbonate}}$  of *O. universa* compared to *T. sacculifer* for the similar calcification depth at few sites (e.g. FC-02a, WP07-a) might reflect difference in photosynthetic potential between the two species, Tagaki et al. (2019) showed a lower photosynthetic potential for *O. universa* compared to *T. sacculifer*.”

Section 5.4 is very confusing. It seems haphazardly put together and some sentences are poorly structured. Paragraph 580 could be deleted.

**We have edited this part following suggestions from Jesse Farmer**

## Report #3 Henehan

The manuscript is greatly improved from its previous incarnation. Although it is for the previous reviewers to judge whether their comments are adequately dealt with, my sense is that they seem to be largely adequately discussed.

I am personally still a bit unconvinced in the point of plotting a calibration line through data that come from different size fractions for *G. ruber* and *T. sacculifer*, when there is clearly a known size fraction effect which is muddling the pH (/borate) signal and potentially influencing the slope. I guess particularly also in this study using such wide size fractions (perhaps by necessity due to sample limitations) means that there is always the possibility that the size distribution could have varied within this range and introduced inter-site differences that change the slope. The authors now provide some shell weight data to inform a little as to the possibility of there being inter-site differences in the sampled shell size fraction – I presume because there are no photos of the samples to measure the actual size distribution. This is at least something, but just telling us that the average shell weight varied is only of limited use- why not give us this data in the tables, so we can see if those boron samples that diverge most strongly from the existing calibration line of *ruber/sacculifer* were indeed on average the smallest test sizes? It would be helpful.

Unfortunately, no weight per shell determinations was done on the WEP samples. Also, the weight per shell we discuss in the manuscript are from other sub-samples picked for oxygen and carbon isotopes only.

We have derived the calibrations from the smaller size fractions, results for *G. ruber* can be in line with a size effect on the  $\delta^{11}\text{B}_{\text{carbonate}}$ , same sensitivity to  $\delta^{11}\text{B}_{\text{borate}}$  but offset of  $\sim 0.4\text{‰}$  from your culture calibration.

There are also still quite a few oddities in sentence structure, word choice, spelling/typos which I will outline below.

Otherwise, with these edits/additions, I have no objections the paper being published. Congratulations to the authors on some nice work.

Line 94: Anagnostou et al. (2016) also did this for core-top and Eocene samples.

**Added**

Line 144:  $\text{B}(\text{OH})_3$  and  $\text{B}(\text{OH})_4^-$ , as used here, are the more common notations for these aqueous species.. I would favour using these. But then for whatever reason in Fig. 2 and its caption (but I think nowhere else?) you have called them  $\text{H}_4\text{B}(\text{OH})_3$  etc. I think you should be consistent so the unfamiliar reader can relate what you are talking about better. Why not stick to what you have here to be more consistent with the literature?

**Changed**

Line 157: as far as I am aware Noireaux et al 2015 did not look at taxonomic differences in

forams- rather inorganic carbonate polymorphs. Either rephrase the sentence or delete the reference.

We removed it, this is a typo from previous version.

Line 196: within the parentheses, insert comma after the subscript borate, followed by 'hereafter' added

Line 204: Note that the culture slopes of *T. sacculifer* and *O. universa* are also within error of 1, when one propagates the error on each datapoint (which for *universa* in particular is rather large). "More core-top and culture calibrations are needed to refine those slopes and fully understand why different slopes if significant differences are observed, which is part of the motivation for this study"

Line 273: I may be being ignorant, but I have never heard of H or X sample cones, only skimmer cones. Is this correct here?

Thank you for picking on that, you are right, skimmer are X or H but sample are normal or jet. We utilized "normal" sample cones.

Line 282: delete 'the' before 11B.

Removed

Lines 285, 293, possibly elsewhere: Inconsistent nomenclature of JCp-1 coral standard. How I just wrote it is the official notation.

Changed

Line 300: Subscript needed for 3 on nitric acid.

Changed

Line 329: 'prerequisite' would be a better word than 'postulate'.

Changed

Line 331: apostrophe after the 's' of species.

Added

Line 426: typo in *obliquiloculata*.

Changed

Line 427: What do the authors mean by 'This study supports interspecific B/Ca ratios'.. ?

Rewrite.

"B/Ca data are species-specific and consistent with previous work (e.g., compiled in Henehan et al., 2016) with ratios higher for *G. ruber* > *T. sacculifer* (sacc) > *T. sacculifer* (w/o sacc) > *P. obliquiloculata* > *O. universa* >> *G. menardii* > *N. dutertrei* > *G. tumida* > *G. inflata* > *N. pachyderma* > *G. bulloides* (Fig. 7)."

Line 451: correct names are 'Kemle-von Mücke and Oberhänsli'

Changed

Line 486: Is this statement really true? See Shaked and de Vargas (2008), doi:10.3354/meps325059

I removed this statement and added the reference

Line 517-518: Sentence poorly written. Not exactly sure what it is the authors want to say. We have reedited this part

Line 532: missing s on species.  
Added

Line 537: ‘deep-dwelling species’ or ‘deeper-dwelling species’, not ‘deep dweller species’  
Changed thorough the text

Line 546: Joji Uchikawa, not Ushikawa.  
Changed thorough the text

Lines 550-551: Not sure it’s wise to pool all different size fractions like this- what does this truly tell you about pH?  
We made regressions based on the different size fractions.

Line 552: missing ‘and’ before ‘the observation’? Also ‘is’, not ‘are’.  
Changed

Line 567-568: “the main interest with utilizing boron-based proxies relates to the...” poorly written sentence. “The main aim of utilizing” or “The primary applications of boron-based proxies are”?  
Changed for: “The main aim of utilizing boron-based proxies relates to the reconstruction of past oceanic conditions”

Line 567-575: Odd paragraph separation, and first line of second paragraph in line 572 isn’t particularly self-contained. In line 574: ‘which is also supporting of’ should be ‘which supports’.  
Changed

Lines 576-578: I would suggest that Chalk et al. (2017)’s accurate downcore data would suggest there isn’t necessarily need for a calibration from the same location or setting? Is it fair to make such a bold statement, and advocate for every downcore reconstruction from now on to have a calibration point from the same setting, based on your sacculifer data alone (that could in theory be different in some places just because the size distribution isn’t exactly the same in all sample sites)?

Yes, only the WEP presents those low d11B in our data, I think it is still best explained by their calcification depths. At this stage, for reconstruction in the WEP, correction / calibration will have to be applied. Nevertheless, this is reassuring to observe that the sensitivity of the calibrations for *G. ruber* is the not significantly different between size fractions, which would make the use of an offset ok for the reconstruction.

Lines 579-580: What correction do you mean? And why? Correction relative to what, the culture calibration?

“Results for *G. ruber* are the most scattered, potentially due to difference in test sizes (Henehan et al., 2013), or depth habitat, although we can not exclude undocumented diagenetic effects. Results reaffirm the importance of working with narrow size fractions (Henehan et al., 2013), the utilization of calibrations derived from the same size fraction or use of offsets to take into account this size fraction effect, and the importance of core-top studies before paleo-application.”

Line 583: what is meant by ‘our sample add a weight/shell’?

**Edited**

Line 584: Suggest “Greater divergence” rather than “The higher divergence”

**Changed**

Line 588: insert ‘isotope-pH’ after boron.

**Added**

Line 592 (and elsewhere): in situ is not hyphenated.

**Addressed**

Lines 592-594: This sentence could be rewritten to explain a bit more clearly what you did.

**This section has been reedited**

Line 600: ‘Sparsest’ seems an odd choice of word- implies not only that there is more scatter, but there is also not as much data. ‘most scattered’ might be better?

**Changed**

Line 606: ‘can impact’ rather than ‘impacts’, as for example your deep dwellers are not very affected.

**Changed**

Line 614: insert ‘similarly’ before respiration

**Added**

Line 617: “those calibrations” – which calibrations are you referring to?

**“the calibrations published to date”**

Line 618: ‘taxon-specific’, not ‘taxa-specific’, I would think here?

**Changed**

Figure 5: Include the size fraction in each *ruber* and *sacculifer* dataset’s legend entry, not just one.. and why are the *ruber* culture data not plotted?

**Added and plotted the culture data**

Figure captions: In general the figure and table captions are a little spartan, and could be a bit

more in-depth and informative. Examples, for instance:

Fig. 4 caption, 'Pre-industrial data' is vague.. hydrographic data (also 'for the sites', rather than 'of the sites')? Are these actual measurements or interpolations? What are the dotted lines? Why are the temperatures seasonally defined but nothing else?

We utilized the GLODAP database that does not resolve seasonal variations. We have added "Dotted lines are the calculated uncertainties based errors on TA and DIC from the GLODAP database."

Fig.5 caption:  $\delta^{11}\text{B}$ borate was, not were. Are the measurements from core-tops, tows, cultures, etc? Are these just the MC-ICPMS data, if so state, so it's clear why there's no NTIMS measurements in here? How are the error bars defined, are they one sigma or two?

"Boron isotopic measurements of mixed-layer foraminifera plotted against the  $\delta^{11}\text{B}$ borate.  $\delta^{11}\text{B}$ borate was characterized by determination of the calcification depth of the foraminifera utilizing data presented in Fig. 4, A) *G. ruber*, B) *T. sacculifer*, C) *O. universa*. Mono-specific calibrations (Table 3) and error bars on  $\delta^{11}\text{B}$ borate were derived utilizing the wild bootstrap code from Henahan et al. (2016), errors on the  $\delta^{11}\text{B}$ carbonate for this study are reported as  $2\sigma$  of measured AE121 standards during the session of the sample. Calibrations were also derived on the 250-400 size fraction for *G. ruber* and *T. sacculifer* (black dashed lines). Data reported on those graphs have been measured with an MC-ICP-MS."

Fig. 6: 'plotted', not 'plot'.

Changed

Fig. 8: watch your superscripts. Does the light attenuation coefficient have a unit? State how the calcification depth marked by the grey band was derived.

So, no units were reported in Rink et al., 1998, but we believe it may be  $\text{m}^{-1}$

Fig. 10: say a little bit more about what the 'reconstructed' values come from and how they were calculated.

"**Figure 9:** Water depth pH profiles reconstructed at every site applying the mono-specific calibrations derived from our results (Table 3). Figure is showing measured  $\delta^{11}\text{B}_{\text{calcite}}$ ,  $\delta^{11}\text{B}_{\text{borate}}$  calculated according to different calibrations (see Table 3 and text), calculated pH based on  $\delta^{11}\text{B}$  ( $\text{pH}_{\delta^{11}\text{B}}$ ) and  $\text{pCO}_2$  calculated from  $\text{pH}_{\delta^{11}\text{B}}$  and alkalinity.

**Figure 10:** Evaluation of the reconstructed parameters,  $\delta^{11}\text{B}_{\text{borate}}$ , pH and  $\text{pCO}_2$  versus *in situ* parameter calculated in Fig. 9 (based on  $\delta^{11}\text{B}$  and alkalinity). The recalculated parameters are consistent with *in situ* data, except for *G. ruber*, this variability might be explained by the different test sizes within our size fractions."

Table 2: Is it necessary to have a column for the cleaning method when it's the same for every sample?

We have removed this column.

Quick question- Site A14 (to a lesser extent also FC-02a) has a really weird vertical pH gradient.. any idea what causes this?

We are not sure how to explain the A14 profile.





# Résultats de la comparaison

Ancien fichier :

**Maxence Guillermic coretop manuscript - 03092020.pdf**

**31 pages (736 KB)**  
3/9/2020 9:14:32 AM

par  
rapport  
à

Nouveau fichier :

**Maxence Guillermic coretop manuscript - 11122019.pdf**

**30 pages (742 KB)**  
2/3/2020 4:10:37 PM

**Total des modifications Contenu**

**1187** **795** remplacements  
**187** insertions  
**205** suppressions

Comparaison du texte  
seulement

**Styles et  
annotations**

**0** style  
**0** annotations

[Consulter la 1re modification \(page 1\)](#)

1 Seawater pH reconstruction using boron isotopes in multiple planktonic foraminifera species with  
2 different depth habitats and their potential to constrain pH and pCO<sub>2</sub> gradients  
3

4  
5 Maxence Guillermic<sup>1,2</sup>, Sambuddha Misra<sup>3,4</sup>, Robert Eagle<sup>1,2</sup>, Alexandra Villa<sup>2,5</sup>, Fengming Chang<sup>6</sup>,  
6 Aradhna Tripathi<sup>1,2</sup>  
7  
8  
9

10  
11 <sup>1</sup> Department of Earth, Planetary, and Space Sciences, Department of Atmospheric and Oceanic  
12 Sciences, Institute of the Environment and Sustainability, UCLA, University of California – Los  
13 Angeles, Los Angeles, CA 90095 USA

14 <sup>2</sup> Laboratoire Géosciences Océan UMR6538, UBO, Institut Universitaire Européen de la Mer, Rue  
15 Dumont d'Urville, 29280, Plouzané, France

16 <sup>3</sup> Indian Institute of Science, Centre for Earth Sciences, Bengaluru, Karnataka 560012, India

17 <sup>4</sup> The Godwin Laboratory for Palaeoclimate Research, Department of Earth Sciences, University of  
18 Cambridge, UK

19 <sup>5</sup> Department of Geology, University of Wisconsin-Madison, Madison, WI 53706 USA

20 <sup>6</sup> Key Laboratory of Marine Geology and Environment, Institute of Oceanology, Chinese Academy of  
21 Sciences, Qingdao 266071, China  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36

37 Submitted to Biogeosciences  
38  
39  
40

41 \*Corresponding author:

42 E-mail address: [maxence.guillermic@gmail.com](mailto:maxence.guillermic@gmail.com)  
43

44 **ABSTRACT**

45

46 Boron isotope systematics of planktonic foraminifera from core-top sediments and culture experiments have been  
47 studied to investigate the sensitivity of  $\delta^{11}\text{B}$  of their calcite tests to seawater pH. However, our knowledge of the  
48 relationship between  $\delta^{11}\text{B}$  and pH remains incomplete for many taxa. Thus, to expand the potential scope of  
49 application of this proxy, we report  $\delta^{11}\text{B}$  data for 7 different species of planktonic foraminifera from sediment  
50 core-tops. We utilize a method for the measurement of small samples of foraminifera and calculate the  $\delta^{11}\text{B}$ -calcite  
51 sensitivity to pH for *Globigerinoides ruber*, *Trilobus sacculifer* (sacc or w/o sacc), *Orbulina universa*, *Pulleniatina*  
52 *obliquiloculata*, *Neogloboquadrina dutertrei*, *Globorotalia menardii* and *Globorotalia tumida*, including for  
53 unstudied core-tops and species. The sensitivity of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  (eg.  $\Delta\delta^{11}\text{B}_{\text{carbonate}}/\Delta\delta^{11}\text{B}_{\text{borate}}$ ) in core-  
54 tops is consistent with previous studies for *T. sacculifer* and *G. ruber* and close to unity for *N. dutertrei*, *O. universa*  
55 and combined deep-dwelling species. Deep-dwelling species closely follow the core-top calibration for *O.*  
56 *universa*, which is attributed to respiration-driven microenvironments, likely caused by light limitation and/or  
57 symbiont/host interactions. These taxa have diverse ecological preferences and are from sites that span a range of  
58 oceanographic regimes, including some that are in regions of air-sea equilibrium and others that are out of  
59 equilibrium with the atmosphere. Our data support the premise that utilizing boron isotope measurements of  
60 multiple species within a sediment core can be utilized to constrain vertical profiles of pH and  $\text{pCO}_2$  at sites  
61 spanning different oceanic regimes, thereby constraining changes in vertical pH gradients and yielding insights  
62 into the past behavior of the oceanic carbon pumps.

## 63 1. Introduction

64 The oceans are absorbing a substantial fraction of anthropogenic carbon emissions resulting in declining  
65 surface ocean pH (Fig. 1; IPCC, 2014). Yet there is a considerable uncertainty over the magnitude of future pH  
66 change in different parts of the ocean and the response of marine biogeochemical cycles to physio-chemical  
67 parameters (T, pH) caused by climate change (Bijma et al., 2002; Ries et al., 2009). Therefore, there is an increased  
68 interest in reconstructing past seawater pH (Hönisch and Hemming, 2005; Liu et al., 2009; Wei et al., 2009;  
69 Douville et al., 2010), in understanding spatial variability in aqueous pH and carbon dioxide ( $p\text{CO}_2$ ) (Foster et al.,  
70 2008; Martinez-Boti et al., 2015b; Raitzsch et al., 2018), and in studying the response of the biological carbon  
71 pump utilizing geochemical proxies (Yu et al., 2007, 2010, 2016).

72 Although proxies for carbon cycle reconstruction are complex in nature (Pagani et al., 2005; Tripathi et al.,  
73 2009, 2011; Allen and Hönisch, 2012), the boron isotope composition of foraminiferal tests is emerging as one of  
74 the more robust candidates (Hönisch et al., 2005, 2009; Ni et al., 2007; Foster et al., 2008, 2012; Bartoli et al.,  
75 2011; Henehan et al., 2013; Martinez-Boti et al., 2015b; Chalk et al., 2017). The study of laboratory cultured  
76 foraminifera has demonstrated a systematic dependence of the boron isotope composition of tests on ambient pH  
77 (Sanyal et al., 1996, 2001; Henehan et al., 2013, 2016). Core-top measurements on globally distributed samples  
78 also show a  $\delta^{11}\text{B}$  sensitivity to pH with taxa-specific offsets from the theoretical fractionation line of borate ion  
79 (Rae et al., 2011; Henehan et al., 2016; Raitzsch et al., 2018).

80 Knowledge of seawater pH, in conjunction with constraints on one other carbonate system parameter  
81 (Total Alkalinity (TA), DIC (dissolved inorganic carbon),  $[\text{HCO}_3^-]$ ,  $[\text{CO}_3^{2-}]$ ), can be utilized to constrain aqueous  
82  $p\text{CO}_2$ . Application of empirical calibrations for boron isotopes, determined for select species of foraminifera from  
83 core-tops and laboratory cultures, has resulted in accurate reconstructions of  $p\text{CO}_2$  utilizing downcore samples  
84 from sites that are in quasi-equilibrium with the atmosphere at present.  $\delta^{11}\text{B}_{\text{carbonate}}$  based reconstructed values of  
85  $p\text{CO}_2$  are analytically indistinguishable from ice core  $\text{CO}_2$  records (Hönisch et al., 2005, 2009; Foster et al., 2008;  
86 Henehan et al., 2013; Chalk et al., 2017).

87 Therefore, the last decade has produced several studies aiming at reconstructing past seawater pH using  
88 boron isotopes to constrain atmospheric  $p\text{CO}_2$  in order understand the changes in the global carbon cycle (Hönisch  
89 et al., 2005, 2009; Foster et al., 2008, 2012, 2014; Seki et al., 2010; Bartoli et al., 2011; Henehan et al., 2013;  
90 Martinez-Boti et al., 2015a, 2015b; Chalk et al., 2017). In addition to reconstructing atmospheric  $p\text{CO}_2$ , in a few  
91 studies, the  $\delta^{11}\text{B}$  proxy has been applied to mixed-layer planktonic foraminifera at sites out of equilibrium with  
92 the atmosphere to constrain past air-sea fluxes (Foster et al., 2014; Martinez-Boti et al., 2015b). A small body of  
93 work has examined whether data for multiple species in core-top (Foster et al., 2008) and down-core samples could  
94 be used to constrain vertical profiles of pH through time (Palmer et al., 1998; Pearson and Palmer, 1999).

95 In this study, we make critical additions to the emerging pool of boron isotope data of core-top planktonic  
96 foraminifera from different oceanographic regimes, including data for species that have not previously been  
97 examined. We utilize a low-blank (15 pg B to 65 pg B), high precision (2sd on the international standard JCP-1 is  
98 0.20 ‰, n=6)  $\delta^{11}\text{B}_{\text{carbonate}}$  analysis method (down to ~250  $\mu\text{g CaCO}_3$ ), modified after Misra et al. (2014), to study  
99 multiple species of planktonic foraminifera from sediment core-tops that span a range of oceanographic regimes,  
100 including open-ocean oligotrophic settings and marginal seas. We constrain calibrations for different species, and  
101 compare results to published work (Foster et al., 2008; Henehan et al., 2013; Henehan et al., 2016; Martinez-Boti  
102 et al., 2015b; Raitzsch et al., 2018). We also test whether these data support the application of boron isotope

103 measurements of multiple species within a sediment core as a proxy for constraining vertical profiles of pH and  
104 pCO<sub>2</sub>.

105

## 106 2. Background

### 107 2.1 Planktonic foraminifera as archives of seawater pH

108 Planktonic foraminifera are used as archives of past environmental conditions within the mixed layer and  
109 thermocline, as their chemical composition is correlated with the physio-chemical parameters of their calcification  
110 environment (Ravelo and Fairbanks, 1992; Elderfield and Ganssen, 2000; Dekens et al., 2002; Anand et al., 2003;  
111 Sanyal et al., 2001; Ni et al., 2007; Henehan et al., 2013, 2015, 2016; Howes et al., 2017; Raitzch et al., 2018).  
112 The utilization of geochemical data for multiple planktonic foraminifera species with different ecological  
113 preferences to constrain vertical gradients has been explored in several studies. The framework for such an  
114 approach was first developed using modern samples of planktonic foraminifera for oxygen isotopes, where it was  
115 proposed as a tool to constrain vertical temperature gradients and study physical oceanographic conditions during  
116 periods of calcification (Ravelo and Fairbanks, 1992).

117 Because planktonic foraminifera species complete their lifecycle in a particular depth habitat due to their  
118 ecological preference (Ravelo and Fairbanks, 1992; Farmer et al., 2007), it is theoretically possible to reconstruct  
119 water column profiles of pH using data from multiple taxa (Palmer and Pearson, 1998; Anagnostou et al., 2016).  
120 The potential use of an analogous approach to reconstruct past profiles of seawater pH was first highlighted by  
121 Palmer and Pearson (1998) on Eocene samples to constrain pH-depth gradients. However, in these boron isotope-  
122 based studies, it was assumed that boron isotope offset from seawater and foraminiferal carbonate were constant,  
123 which is an assumption not supported by subsequent studies (e.g., Hönisch et al., 2003; Foster et al., 2008; Henehan  
124 et al., 2013, 2016; Raitzsch et al., 2018; Rae, 2018). Furthermore,  $\delta^{11}\text{B}$  differences between foraminifera species  
125 that inhabit waters that are the same pH makes the acquisition of more core-top and culture data essential for  
126 applications of the proxy.

127

### 128 2.2 Boron systematics in seawater

129 Boron is a conservative element in seawater with a long residence time ( $\tau_B \sim 14$  Myr) (Lemarchand et al.,  
130 2002a). In seawater, boron exists as trigonal boric acid  $\text{B}(\text{OH})_3$  and tetrahedral borate ion  $\text{B}(\text{OH})_4^-$  (borate). The  
131 relative abundance of boric acid and borate ion is a function of the ambient seawater pH. At standard open ocean  
132 conditions ( $T = 25$  °C and  $S = 35$ ), the dissociation constant of boric acid is 8.60 (Dickson, 1990), implying that  
133 boron mainly exists in the form of boric acid in seawater. Since the  $\text{pK}_B$  and seawater pH (e.g.,  $\sim 8.1$ , NBS) values  
134 are similar, it implies that small changes in seawater pH will induce strong variations in the abundance of the two  
135 boron species (Fig. 2).

136 Boron has two stable isotopes,  $^{10}\text{B}$  and  $^{11}\text{B}$ , with average relative abundances of 19.9 and 80.1 %,  
137 respectively. Variations in B isotope ratio are expressed in conventional delta ( $\delta$ ) notation:

138

$$139 \quad \delta^{11}\text{B} (\text{‰}) = 1000 \times \left( \frac{{}^{11}\text{B}/{}^{10}\text{B}_{\text{Sample}}}{{}^{11}\text{B}/{}^{10}\text{B}_{\text{NIST 951-a}}} - 1 \right) \quad (1)$$

140

141 where positive values represent enrichment in the heavy isotope  $^{11}\text{B}$ , and negative values enrichment in the light  
142 isotope  $^{10}\text{B}$ , relative to the standard reference material. Boron isotope values are reported versus the NIST SRM  
143 951 (Cantazaro et al., 1970).

144  $\text{B}(\text{OH})_3$  is enriched in  $^{11}\text{B}$  compared to  $\text{B}(\text{OH})_4^-$  with a constant offset between the two chemical  
145 species, within the range of physio-chemical variation observed in seawater, given by the fraction factor ( $\alpha$ ). The  
146 fractionation ( $\epsilon$ ) between  $\text{B}(\text{OH})_3$  and  $\text{B}(\text{OH})_4^-$  of  $27.2 \pm 0.6 \%$  has been empirically determined by Klochko et  
147 al., (2006) in seawater. Note, Nir et al., (2015) calculate this fractionation, using an independent method, to be  $26$   
148  $\pm 1 \%$ , which is within the analytical uncertainty of the Klochko et al., (2006) value.

149

### 150 **2.3 Boron isotopes in planktonic foraminifera calcite**

151 Many biogenic carbonate-based geochemical proxies are affected by “vital effects” or biological  
152 fractionations (Urey et al., 1951). The  $\delta^{11}\text{B}_{\text{carbonate}}$  in foraminifera exhibits species-specific offsets (see Rae et al.,  
153 2018 for review) compared to theoretical predictions for the boron isotopic composition of  $\text{B}(\text{OH})_4^-$  ( $\alpha=1.0272$ ,  
154 Klochko et al., 2006). As the analytical and technical aspects of boron isotope measurements have improved  
155 (Foster et al., 2008; Rae et al., 2011; Misra et al., 2014; Lloyd et al., 2018), evidence for taxonomic differences  
156 have not been eliminated, but have become increasingly apparent (Foster et al., 2008, 2018; Henehan et al 2013,  
157 2016; Noireaux et al., 2015; Foster et al., 2016; Rae et al., 2018; Raitzsch et al., 2018).

158 At present, culture and core-top calibrations have been published for several planktonic species including  
159 *Trilobatus sacculifer*, *Globigerinoides ruber*, *Globigerina bulloides*, *Neogloboquadrina pachyderma*, *Orbulina*  
160 *universa* (Foster et al., 2008; Henehan et al., 2013; Henehan et al., 2015; Sanyal et al., 1996; Sanyal et al., 2001).  
161 Although the boron isotopic composition of several species of foraminifera are now commonly used tools for  
162 reconstructing surface seawater pH, for other species, there is a lack of data constraining boron isotope sensitivity  
163 between foraminiferal carbonate and borate ion in seawater.

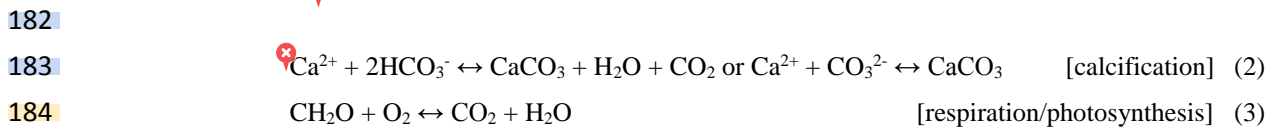
164

### 165 **2.4 Origin of biological fractionations in foraminifera**

166 Perforate foraminifera are calcifying organisms that maintain a large degree of biological control over  
167 their calcification space, and thus, mechanisms of biomineralization may be of significant importance in  
168 controlling the  $\delta^{11}\text{B}$  of the biogenic calcite. The biomineralization of foraminifera is based on seawater  
169 vacuolization (Erez, 2003; de Nooijer et al., 2014) with parcels of seawater being isolated by an organic matrix  
170 thereby creating a vacuole filled with seawater. Recent work has also demonstrated that even if the chemical  
171 composition of the reservoirs is modified by the organism, seawater is directly involved in the calcification process  
172 with vacuoles formed at the periphery of the shell (de Nooijer et al., 2014). Culture experiments by Rollion-Bard  
173 and Erez., (2010) have proposed that the pH at the site of biomineralization is elevated to an upper pH limit of  $\sim 9$   
174 for the shallow-water, symbiont-bearing benthic foraminifera *Amphistegina lobifera*, which would support a pH  
175 modulation of a calcifying fluid in foraminifera. We acknowledge this is speculative as it is based upon benthic  
176 foraminifer experiments.

177 For taxa with symbionts, the microenvironment surrounding the foraminifera is chemically different from  
178 seawater due to photosynthetic activity (Jorgensen et al., 1985; Rink et al., 1998; Köhler-Rink and Köhl, 2000).  
179 Photosynthesis by the symbionts elevates the pH of the microenvironment (Jorgensen et al., 1985; Rink et al.,

180 1998; Wolf-Gladrow et al., 1999; Köhler-Rink and Kühl, 2000), while calcification and respiration decrease it  
181 (Equation 2 and 3).



186  $\delta^{11}\text{B}$  in foraminifera is primary controlled by seawater pH, but is also dependent of the pH alteration of  
187 microenvironments due to calcification, respiration and symbiont photosynthesis.  $\delta^{11}\text{B}$  should therefore reflect the  
188 relative dominance of these processes and may account for species-specific  $\delta^{11}\text{B}$  offsets. Theoretical predictions  
189 from Zeebe et al. (2003) and foraminiferal data from Hönisch et al., (2003) highlighted the dominance of  
190 microenvironment pH in  $\delta^{11}\text{B}$  signature of foraminifera. Their work also suggested that for a given species, there  
191 should be a constant offset observed between the boron isotope composition of foraminifera and borate ion over a  
192 large range of pH, imparting confidence in utilizing species-specific boron isotope data as a proxy for seawater  
193 pH.

194 Comparison of boron isotope data for multiple planktonic foraminiferal species indicate that taxa with  
195 high levels of symbiont activity such as *T. sacculifer* and *G. ruber* show higher  $\delta^{11}\text{B}$  values than the  $\delta^{11}\text{B}$  of ambient  
196 borate (Foster et al., 2008, Henehan et al., 2013, Raitzsch et al., 2018). The sensitivities ( $\Delta\delta^{11}\text{B}_{\text{carbonate}}/\Delta\delta^{11}\text{B}_{\text{borate}}$   
197 referred to as the slope) of existing calibrations suggest a different species-specific sensitivity for these species  
198 compared to other taxa (Sanyal et al., 2001; Henehan et al., 2013; Henehan et al., 2015; Raitzsch et al., 2018). For  
199 example, *Orbulina universa* exhibits a lower  $\delta^{11}\text{B}$  than *in situ*  $\delta^{11}\text{B}$  values of borate ion (Henehan et al., 2016),  
200 consistent with the species living deeper in the water column characterized by reduced photosynthetic activity.

201 It is possible that photosynthetic activity by symbionts might not be able to compensate for changes in  
202 calcification and/or respiration, leading to an acidification of the microenvironment. It is interesting to note that  
203 for *O. universa* the slope determined for the field-collected samples is not statistically different from unity ( $0.95 \pm$   
204  $0.17$ ) (Henehan et al. 2016), while culture experiments report slopes of  $\leq 1$  for multiple species including *G. ruber*  
205 (Henehan et al., 2013), *T. sacculifer* (Sanyal et al., 2001), and *O. universa* (Sanyal et al., 1999). More core-top and  
206 culture calibrations are needed to fully understand why different slopes are observed, which is part of the  
207 motivation for this study.

208

## 209 2.5 Planktic foraminifera depth and habitat preferences

210 The preferred depth habitat of different species of planktonic foraminifera depends on their ecology,  
211 which in turn relies on the hydrographic conditions. For example, *G. ruber* is commonly found in the mixed layer  
212 (Fairbanks and Wiebe, 1980; Dekens et al., 2002; Farmer et al., 2007) during the summer (Deuser et al., 1981)  
213 whereas *T. sacculifer* is present in the mixed layer until mid-thermocline depths (Farmer et al., 2007) during spring  
214 and summer (Deuser et al., 1981, 1989). Specimens of *P. obliquiloculata* and *N. dutertrei* are abundant during  
215 winter months (Deuser et al., 1989), with an acme in the mixed layer (~60m) for *P. obliquiloculata*, and at mid-  
216 thermocline depths for *N. dutertrei* (Farmer et al., 2007). In contrast, *O. universa* tends to record annual average  
217 conditions within the mixed layer. Specimens of *G. menardii* calcify within the seasonal thermocline (Fairbanks  
218 et al., 1982, Farmer et al., 2007, Regenberg et al., 2009), and in some regions in the upper thermocline (Farmer et



219 al., 2007), and records annual temperatures. *G. tumida* is found at the lower thermocline or below the thermocline  
220 and records annual average conditions (Fairbanks and Wiebe, 1980; Farmer et al., 2007, Birch et al., 2013).

221

### 222 3. Materials and Methods

223

#### 224 3.1 Localities studied

225 Core-top locations were selected to span a broad range of seawater pH, carbonate system parameters, and  
226 oceanic regimes. Samples from Atlantic Ocean (CD107-A), Indian Ocean (FC-01a and FC-02a), Arabian Sea  
227 (FC-13a and FC-12b) and Pacific Ocean (WP07-01, A14, and Ocean Drilling Program 806A and 807A) were  
228 analyzed; characteristics of the sites are summarized in Table 1 and S7, Fig. 3, and Fig. 4.

229 Atlantic site CD107-a (CD107 site A) was drilled in 1997 by the Benthic Boundary Layer program  
230 (BENBO) (K.S. Black et al., 1997 - cruise report RRS Charles Darwin Cruise 107). Arabian Sea sites FC-12b  
231 (CD145 A150) and FC-13a (CD145 A3200) were retrieved by the *Charles Darwin* in the Pakistan Margin in 2004  
232 (B.J. Bett et al., 2003 - cruise report n°50 RRS Charles Darwin Cruise 145). A14 was recovered by box corer in  
233 the southern area of the South China Sea in 2012. Core WP07-01 was obtained from the Ontong Java Plateau using  
234 a giant piston corer during the Warm Pool Subject Cruise in 1993. Holes 806A and 807A were retrieved on Leg  
235 130 by the Ocean Drilling Program (ODP). The top 10 cm of sediment from CD107-A have been radiocarbon  
236 dated to be Holocene <3 ky (Thomson et al., 2000). Samples from multiple box cores from Indian Ocean sites  
237 were radiocarbon dated as Holocene <7.3 ky (Wilson et al., 2012). Samples from western equatorial Pacific Site  
238 806B, close to site WP07-01, are dated to between 7.3-8.6 ky (Lea et al., 2000). Arabian Sea and Pacific core-top  
239 samples were not radiocarbon dated but are assumed to be Holocene.

240

#### 241 3.2 Species

242 Around 50-100 foraminifera shells were picked from the 400-500 µm fraction size for *Globorotalia*  
243 *menardii* and *Globorotalia tumida*, >500 µm for *Orbulina universa*, from the 250-400 µm fraction size for  
244 *Trilobatus sacculifer* (w/o sacc, without sacc-like final chamber), *Trilobatus sacculifer* (sacc, sacc-like final  
245 chamber), *Globigerinoides ruber* (white, sensu stricto), *Neogloboquadrina dutertrei*, *Pulleniatina obliquiloculata*.  
246 The samples picked for analyses were visually well preserved.

247

#### 248 3.3 Sample cleaning

249 Briefly, picked foraminifera were gently cracked open, clay removed and checked for coarse-grained  
250 silicates. The next stages of sample processing and chemical separation were performed in a class 1000 clean lab  
251 equipped with boron-free HEPA filters. Samples were then cleaned using full reductive and oxidative cleaning  
252 (Boyle and Keigwin, 1985; Barker et al., 2003). final leaching step with 0.001N HCl was done before dissolution  
253 in 1N HCl. Each sample was divided into two aliquots: an aliquot for boron purification and one aliquot for trace  
254 element analysis.

255

#### 256 3.4 Reagents

257 Double-distilled HNO<sub>3</sub> and HCl acids (from Merck® grade) and a commercial bottle of HF Ultrapure  
258 grade were used at Brest. Double-distilled acids were used at Cambridge. All acids and further dilutions were  
259 prepared using double-distilled 18.2 MΩ.cm<sup>-1</sup> MQ water. Working standards for isotope ratio and trace element

260 measurements were freshly diluted on a daily basis with the same acids used for sample preparation to avoid any  
261 matrix effect.

262

### 263 **32.5 Boron isotopes**

264 Boron purification for isotopic measurement was done utilizing microdistillation method developed by  
265 Gaillardet et al., (2001), for Ca-rich matrices by Wang et al., (2010) and adapted at Cambridge by Misra et al.,  
266 (2014a). 70  $\mu\text{L}$  of dissolved carbonate sample was loaded on a cap of a clean fin legged 5 mL conical beaker  
267 upside down. The tightly closed beaker was put on a hotplate at  $95^\circ\text{C}$  for 15 hours. The beakers were taken off the  
268 hotplate and were allowed to cool for 15 min. The cap where the residue formed was replaced by a clean one.  
269 Then, 100  $\mu\text{L}$  of 0.5% HF were added to the distillate.

270 Boron isotopic measurements were carried out on a Thermo Scientific @Neptune+ MC-ICP-MS at the  
271 University of Cambridge. Neptune+ was equipped with Jet interface and two  $10^{13} \Omega$  resistors. The instrumental  
272 setup included Savillex® 50 $\mu\text{l}/\text{min}$  C-flow self-aspirating nebulizer, single pass Teflon® Scott-type spray chamber  
273 constructed utilizing Savillex® column components, 2.0 mm Pt injector from ESI®, Thermo® Ni 'H' type sample  
274 cone and 'X' type skimmer cones. Both isotopes of boron were determined utilizing  $10^{13} \Omega$  resistors (Misra et al.,  
275 2014a; Lloyd et al., 2018).

276 The sample size for boron isotope analyses typically ranged from 10 ppb B (~5 ng B) to 20 ppb B samples  
277 (~10 ng B). Instrumental sensitivity for  $^{11}\text{B}$  was 17 mV/ppb B (eg. 170 mV for 10ppb B) in wet plasma at 50 $\mu\text{l}/\text{min}$   
278 sample aspiration rate. Intensity of  $^{11}\text{B}$  for a sample at 10ppb B was typically  $165\text{mV} \pm 5\text{mV}$  closely matched the  
279  $170\text{mV} \pm 5\text{mV}$  of the standard. Due to the low boron content of the samples extreme care was taken to avoid boron  
280 contamination during sample preparation and reduce memory effect during analysis. Procedural boron blanks  
281 ranged from 15pg B to 65 pg B (contributed to less than <1% of the sample signal). The acid blank during analyses  
282 was measured at  $\leq 1\text{mV}$  on the  $^{11}\text{B}$ , meaning a contribution < 1% of the sample intensity, no memory effect was  
283 observed within and across sessions.

284 Analyses of external standards were done to ensure data quality. For  $\delta^{11}\text{B}$  measurements two carbonate  
285 standards were utilized: the JCP-1 (Geological Survey of Japan, Tsukuba, Japan) international standard (Gutjahr  
286 et al., 2014) and the NEP internal coral (Porites sp.,  $\delta^{11}\text{B} = 26.12 \pm 0.92 \text{‰}$ , 2SD, n=33 Holcomb et al., 2015 and  
287 Sutton et al., 2018, Table S2) from University of Western Australia/Australian National University. Certified boron  
288 isotopes liquid standard, the ERM® AE121 ( $\delta^{11}\text{B} = 19.9 \pm 0.6 \text{‰}$ , SD, certified) was used to monitor reproducibility  
289 and drift during each session (Vogl and Rosner, 2011; Foster et al., 2013; Misra et al., 2014). Results for the  
290 isotopic composition of the NEP standard are shown in Table S2, average values are  $\delta^{11}\text{B}_{\text{NEP}} = 25.70 \pm 0.93 \text{‰}$   
291 (2SD, n=22) over different 7 analytical sessions with each number representing an ab-initio processed sample -  
292 this study). Our results are within error of published values of  $26.20 \pm 0.88 \text{‰}$  (2SD, n = 27) and  $25.80 \pm 0.89 \text{‰}$   
293 (2SD, n = 6) by Holcomb et al. (2015) and Sutton et al. (2018) respectively. Chemically cleaned JCP<sub>1</sub> samples  
294 were measured at  $24.06 \pm 0.20$  (2SD, n=6) and is within error of published values of  $24.37 \pm 0.32 \text{‰}$  and  $24.42 \pm$   
295  $0.28 \text{‰}$  by Holcomb et al. (2015) and Sutton et al. (2018) respectively.

296

### 297 **3.6 Trace elements**

298 The calcium concentration of each sample was measured on an ICP-AES ® Ultima 2 HORIBA at the  
299 Pôle spectrometrie Océan (PSO), UMR6538 (Plouzané, France). Samples were then diluted to fixed calcium

300 concentrations (typically 10 ppm or 30 ppm Ca) using 0.1 M HNO<sub>3</sub> & 0.3 M HF matching multi-element standards  
301 Ca concentration to avoid any matrix effect (Misra et al., 2014b). Trace elements (e.g. X/Ca ratios) were analyzed  
302 on a Thermo Scientific ® Element XR HR-ICP-MS at the PSO, Ifremer (Plouzané, France).

303 Trace element analyses were done at a Ca concentration of 10 or 30 ppm. The typical blanks for a 30 ppm  
304 Ca session were: <sup>7</sup>Li < 2%, <sup>11</sup>B < 7%, <sup>25</sup>Mg < 0.2% and <sup>43</sup>Ca < 0.02%. Additionally, blanks for a 10 ppm Ca session  
305 were: <sup>7</sup>Li < 2.5%, <sup>11</sup>B < 10%, <sup>25</sup>Mg < 0.4% and <sup>43</sup>Ca < 0.05%. Due to strong memory effect for boron and  
306 instrumental drift on the Element XR, long sessions of conditioning were done prior analyses. Boron blanks were  
307 driven below 5% of signal intensity usually after 4 to 5 days of continuous analyses of carbonate samples. External  
308 reproducibility was determined on the consistency standard Cam-Wuellestorf (courtesy of the University of  
309 Cambridge) (Misra et al., 2014b), Table S3. Our X/Ca ratio measurements on the external standard Cam-  
310 Wuellestorf were all the time within error of the published value (Table S3) validating the robustness of our trace  
311 elements data. Analytical uncertainty of a single measurement was calculated from the reproducibility of the Cam-  
312 Wuellestorf, measured during a particular mass spectrometry session. The analytical uncertainties on the X/Ca  
313 ratios are: 0.4 µmol/mol for Li/Ca, 7 µmol/mol for B/Ca and 0.01 mmol/mol for Mg/Ca (2SD, n=31) respectively.  
314

### 315 3.7 Oxygen isotopes

316 Carbonate δ<sup>13</sup>C and δ<sup>18</sup>O were measured on a Gas Bench II coupled to a Delta V mass spectrometer at the  
317 stable isotope facility of Pôle spectrometrie Océan (PSO), Plouzané. Around 20 shells were weighed, crushed and  
318 clay removed. The recovered foraminifera were weighed in tubes and flushed with He gas. Samples were then  
319 digested in phosphoric acid and analyzed. Results were calibrated to the VPDB scale by international standard  
320 NBS19 and analytical precision on the in-house standard Ca21 was better than 0.11‰ for δ<sup>18</sup>O (1SD, n=5) and  
321 0.03‰ for δ<sup>13</sup>C (1SD, n=5).  
322

### 323 3.8 Calcification depth determination

324 We utilized two different chemo-stratigraphic methods to estimate the calcification depth in this study  
325 (Table S6 and S7). The first method, commonly used in paleoceanography, utilizes δ<sup>18</sup>O measurements of the  
326 carbonate (δ<sup>18</sup>O<sub>c</sub>) to estimate calcification depths (referred to as δ<sup>18</sup>O-based calcification depths) (Schmidt et al.,  
327 2002; Mortyn et al., 2003; Sime et al., 2005; Farmer et al., 2007; Birsh et al., 2013). The second method utilizes  
328 Mg/Ca-based temperature estimates (T<sub>Mg/Ca</sub>) to constrain calcification depths (Quintana Krupinski et al., 2017). In  
329 both cases, the postulate was that vertical profiles of seawater temperature are available for different seasons in  
330 ocean atlases and cruise reports, and that hydrographic data and geochemical proxy signatures can be compared  
331 to assess the depth in the water column that represents the species maximum abundance.

332 The two different methods to estimate calcification depth were then compared to published depth  
333 estimates for the basin, and where available, for the same site (Table S6). We chose literature values for  
334 calcification depths when available, or depths that were closest to what is known for the region or basin. As  
335 foraminifera can migrate in the water column along their ontogeny, we applied (based on uncertainties of our  
336 measurements) an uncertainty of ±10m for calcification depths > 70 m and an uncertainty of ±20 m when  
337 calcification depths < 70 m. The depth habitats utilized to derive *in situ* parameters are summarized in Table S7.  
338

### 339 3.9 δ<sup>11</sup>B<sub>borate</sub>

340 Two carbonate system parameters are needed to fully constrain the carbonate system. Following the  
341 approach of Foster et al., (2008) we used the GLODAP database (Key et al., 2004) corrected for anthropogenic  
342 inputs in order to estimate pre-industrial carbonate system parameters at each site. Temperature, salinity and  
343 pressure for each site are from the World ocean database 2013 (Boyer et al., 2013). We utilized the R<sup>®</sup> code in  
344 Henehan et al, (2016) (courtesy of Michael Henehan) to calculate the  $\delta^{11}\text{B}_{\text{borate}}$  and derive our calibrations.  
345 Uncertainty for  $\delta^{11}\text{B}_{\text{borate}}$  utilizing the code was similar to the one calculated by applying 2 standard deviations of  
346 the calculated  $\delta^{11}\text{B}_{\text{borate}}$  within the limits imposed by the calcification depth.

347 The Matlab<sup>®</sup> template provided by Zeebe and Wolf-Gladow, (2001) was used to calculate pCO<sub>2</sub> from  
348 TA; temperature, salinity and pressure were included into the calculations. Total boron was calculated from Lee  
349 et al., (2010), K<sub>1</sub> and K<sub>2</sub> were calculated from Mehrbach et al. (1973) refitted by Dickson and Millero (1987).

350 Statistical tests were made utilizing GraphPad<sup>®</sup> software, linear regressions for calibration where derived  
351 utilizing R<sup>®</sup> code in Henehan et al, (2016) (courtesy of Michael Henehan) with a k=500.

352

## 353 4. Results

354

### 355 4.1 Depth habitat

356 The calcification depths utilized in this paper are summarized in Tables S6 and S7, including a comparison  
357 of calcification depth determination methods. The calculated calcification depths are consistent with the ecology  
358 of each species and the hydrography of the sites. Specimens of *G. ruber* and *T. sacculifer* appear to be living in  
359 the shallow mixed layer (0-100 m), with *T. sacculifer* living or migrating deeper than *G. ruber* (down to 125 m).  
360 Specimens of *O. universa* and *P. obliquiloculata* are living in the upper thermocline; *G. menardii* is found in the  
361 upper thermocline until the thermocline depth specific to the location; *N. dutertrei* is living around the thermocline  
362 depth and specimens of *G. tumida* are found in the lower thermocline.

363 Data from both approaches implies that some species inhabit deeper environments in the Western  
364 Equatorial Pacific (WEP) relative to the Arabian Sea, which in turn are deeper dwelling than in the Indian Ocean.  
365 In some cases, we find evidence for differences in habitat depth of up to ~100m between the WEP and the Arabian  
366 Sea. This trend is observed for *G. ruber* and *T. sacculifer*, but not for *O. universa*.

367 Some differences in calcification depth are observed between the two calcification depth determination  
368 methods. These differences might be due to the choice of calibrations. Alternatively, our uncertainties for  $\delta^{18}\text{O}$   
369 implies larger uncertainties on the calcification depth determination using this approach, compared to Mg/Ca  
370 measurements.

371

### 372 4.2 Empirical calibrations of foraminiferal $\delta^{11}\text{B}_{\text{carbonate}}$ to $\delta^{11}\text{B}_{\text{borate}}$

373 Results for the different species analyzed in this study are presented in Fig. 5, Fig. 6 and summarized in  
374 Table 2; additionally, published calibrations for comparison are summarized in Table 3.

375

#### 376 4.2.1 *G. ruber*

377 Our results for *G. ruber* (Fig. 5) are in good agreement with published data from other core-tops, sediment  
378 traps, tows, and culture experiments for  $\delta^{11}\text{B}_{\text{borate}} > 19\text{‰}$  (Foster et al., 2008, Henehan et al., 2013, Raitzsch et al.,  
379 2018). However, for  $\delta^{11}\text{B}_{\text{borate}} < 19\text{‰}$  our results show lighter  $\delta^{11}\text{B}_{\text{carbonate}}$  compared to published values. Whilst this

380 species has been widely studied previously, the sites selected in this study allow us to extend the calibration. The  
381 positive offset from the 1:1 curve has been explained by the high photosynthetic activity (Hönisch et al., 2003;  
382 Zeebe et al., 2003). Two calibrations have been derived. Utilizing only our data, the sensitivity of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  
383  $\delta^{11}\text{B}_{\text{borate}}$  of our linear regression is not statistically different from 1, the uncertainty on this regression is important  
384 due to our small dataset and not inconsistent with the low sensitivity trend of the culture experiments from Sanyal  
385 et al., (2001) or Henehan et al., (2013). The second calibration made compiling all data from literature shows a  
386 sensitivity similar (e.g.  $0.46 (\pm 0.34)$ ) to the one recently published by Raitzsch et al., (2018) (e.g.  $0.45 (\pm 0.16)$ ,  
387 Table 3).

388

#### 389 4.2.2 *T. sacculifer*

390  $\delta^{11}\text{B}_{\text{carbonate}}$  results for *T. sacculifer* (sacc and w/o sacc) (Fig. 5) are compared to published data (Foster et  
391 al., 2008; Martinez-Boti et al., 2015b, Raitzsch et al., 2018). Results for *T. sacculifer* are in good agreement with  
392 the literature and fall above the 1:1 line. Linear regression on our data yields a slope of  $1.3 \pm 0.2$  but is not  
393 statistically different to the results from Martinez-Boti et al., (2015b) (Table 3), ( $p > 0.05$ ). However, when  
394 compiled with published data using the bootstrap method a slope of  $0.83 \pm 0.48$  is calculated, with a large  
395 uncertainty given the variability in the data. It is also noticeable that *T. sacculifer* (w/o sacc) samples from the  
396 WEP have a  $\delta^{11}\text{B}_{\text{carbonate}}$  close or below the 1:1 line and are significantly lower compared to the combined *T.*  
397 *sacculifer* of other sites ( $p = 0.01$ , unpaired t-test).

398

#### 399 4.2.3 *O. universa* and deeper dwelling species: *N. dutertrei*, *P. obliquiloculata*, *G. menardii* and *G. tumida*

400 Our results for *O. universa* (Fig. 5), *N. dutertrei*, *P. obliquiloculata*, *G. menardii* and *G. tumida* (Fig. 6)  
401 fall below the 1:1 line. These data for *O. universa* are not statistically different from the Henehan et al. (2016)  
402 calibration ( $p > 0.05$ ). Our results for *N. dutertrei* expand upon the initial measurements presented in Foster et al.,  
403 (2008). The different environments experienced by *N. dutertrei* in our study permit us to extend the range and  
404 derive a calibration for this species; the slope is close to unity ( $0.93 \pm 0.55$ ), and is similar to the ( $0.95 \pm 0.17$ )  
405 previously reported by Henehan et al., (2016) for *O. universa* and not statistically different ( $p > 0.05$ ). The data for  
406 *P. obliquiloculata* exhibits the largest offset from the theoretical line. The range of  $\delta^{11}\text{B}_{\text{borate}}$  from the samples we  
407 have of *G. menardii* and *G. tumida* is not sufficient to derive calibrations, but the points are in good agreement  
408 with the *N. dutertrei* calibration and Henehan et al. (2016) calibration for *O. universa*.

409 For all species, the slopes are not statistically different from Henehan et al. (2016) ( $p > 0.05$ ) and are close  
410 to unity. If data for deep-dwelling foraminiferal species are pooled together with each other and with data from  
411 Henehan et al., (2016) and Raitzsch et al., (2018), we calculate a slope of  $0.95 (\pm 0.13)$  ( $R^2 = 0.7987$ ,  $p < 0.0001$ ); if  
412 only our data are used, we calculate a slope that is not significantly different ( $0.82 \pm 0.27$ ;  $p < 0.05$ ). However, it  
413 may remain premature to assume that a unique calibration with a slope of  $\sim 0.9$  can be used for all deeper-dwelling  
414 species; more data is needed for *P. obliquiloculata*, *G. menardii* and *G. tumida* to robustly test this assertion.

415

#### 416 4.2.4 Comparison of core-top and culture data

417 The data for *G. ruber* and *T. sacculifer* from the core-tops we measured are broadly consistent with  
418 previous published results. The calibrations between these core-top derived estimates and culture experiments are  
419 not statistically different due to small datasets and uncertainties on the linear regressions (Henehan et al., 2013;

420 Marinez-Boti et al., 2015; Raitzsch et al., 2018; Table 3). The sensitivities of the species analyzed are not  
421 statistically different and are close to unity.

422

### 423 4.3 B/Ca ratios

424 B/Ca ratios are presented in Table 2 and Fig. 7. Values are species-specific consistent with previous work  
425 (e.g., compiled in Henehan et al., 2016) with ratios higher for *G. ruber* > *T. sacculifer* > *T. sacculifer* (w/o sacc) >  
426 *P. obliquiloculata* > *O. universa* >> *G. menardii* > *N. dutertrei* > *G. tumida* > *G. inflata* > *N. pachyderma* > *G.*  
427 *bulloides* (Fig. 7). This study supports interspecific B/Ca ratios (Yu et al., 2007; Tripathi et al., 2009, 2011; Allen  
428 and Hönisch, 2012; Henehan et al., 2016). Differences between surface- and deep-dwelling foraminifera are  
429 observed, with lower values and a smaller range for the deeper dwelling taxa (58-126  $\mu\text{mol/mol}$  vs 83-190  
430  $\mu\text{mol/mol}$  for shallow dwellers), however, the trend for the surface-dwellers can also be driven by interspecies  
431 B/Ca variability. The B/Ca data for deep-dwelling taxa exhibits a significant correlation with  $[\text{B}(\text{OH})_4^-]/[\text{HCO}_3^-]$   
432 ( $p < 0.05$ ), but no correlation with  $\delta^{11}\text{B}_{\text{carbonate}}$  and temperature (Fig. S3). Surface-dwelling species have B/Ca ratios  
433 that exhibit significant correlations with  $[\text{B}(\text{OH})_4^-]/[\text{HCO}_3^-]$ ,  $\delta^{11}\text{B}_{\text{carbonate}}$  and temperature. The sensitivity of B/Ca  
434 to  $[\text{B}(\text{OH})_4^-]/[\text{HCO}_3^-]$  is lower for deep-dwelling species compared to surface dwelling species. When all the B/Ca  
435 data are compiled, significant trends are observed with  $[\text{B}(\text{OH})_4^-]/[\text{HCO}_3^-]$ ,  $\delta^{11}\text{B}_{\text{carbonate}}$  and temperature (Fig. S3).  
436 We also observe that if we compare data from all sites together, correlations exist between B/Ca and the water  
437 depths of the cores (not significant, Fig. S4) but these correlations may also be related to the different the depth  
438 habitats of different taxa in each region, a significant trend is observed when all the data are plotted ( $R^2=0.11$ ,  
439  $p < 0.05$ , Fig. S4).

440

## 441 5. Discussion

442

### 443 5.1 Sources of uncertainty relating to depth habitat and seasonality at studied sites

444

#### 445 5.1.1 Depth habitats and $\delta^{11}\text{B}_{\text{borate}}$

446 Because foraminifera will record ambient environmental conditions during calcification, the accurate  
447 characterization of *in-situ* data is needed not only for calibrations, but also to understand the reconstructed record  
448 of pH or  $p\text{CO}_2$ . The species we examined are ordered here from shallower to deeper depth habitats: *G. ruber* > *T.*  
449 *sacculifer* (sacc) > *T. sacculifer* (w/o sacc) > *O. universa* > *P. obliquiloculata* > *G. menardii* > *N. dutertrei* > *G.*  
450 *tumida* (this study; Birch et al., 2013; Farmer et al., 2007), although the specific water depth will vary depending  
451 on the hydrology of the site (Kemle-von and Oberhänhsl, 1999). We note that calculation of absolute calcification  
452 depths can be challenging in some cases as many species migrate during their ontogeny (Steinhardt et al., 2015).

453 We find that assumptions about the specific depth habitat a species of foraminifera is calcifying over, in  
454 a given region, can lead to differences of a few per mil in calculated isotopic compositions of borate (Fig. 4).  
455 Hence this can cause a bias in calibrations if calcification depths are assumed instead of being calculated (i.e., with  
456  $\delta^{18}\text{O}$  and/or Mg/Ca). Factors including variations in thermocline depth can impact depth habitats for some taxa.  
457 At the sites we examined, most of the sampled species live in deeper depth habitats in the WEP relative to the  
458 Indian Ocean, which in turn is characterized by deeper depth habitats than in the Arabian Sea. In the tropical  
459 Pacific, *T. sacculifer* is usually found deeper than *G. ruber* except at sites characterized by a shallow thermocline,

460 in which case they tend to overlap their habitat (e.g., ODP Site 806 in the WEP which has a deeper thermocline  
461 than at ODP Site 847 in the Eastern Equatorial Pacific; EEP) (Rickaby et al., 2005). The difference in depth habitats  
462 for *T. sacculifer* and *N. dutertrei* between the WEP and EEP can be as much as almost 100 m (Rickaby et al.,  
463 2005).

#### 465 5.1.2 Seasonality and *in-situ* $\delta^{11}\text{B}_{\text{borate}}$

466 As discussed by Raitzsch et al., (2018), depending of the study area, foraminiferal fluxes can change  
467 throughout the year, so seasonality can have a major impact on hydrographic carbonate parameters calculations  
468 for any given water depth. We therefore recalculated the theoretical  $\delta^{11}\text{B}_{\text{borate}}$  using seasonal data for temperature  
469 and salinity and annual values for TA and DIC for each depth at each site. The GLODAP (2013) database does  
470 not provide seasonal TA or DIC values.

471 The low sensitivity of  $\delta^{11}\text{B}_{\text{borate}}$  to temperature and salinity means that calculated  $\delta^{11}\text{B}_{\text{borate}}$  for each water  
472 depth at our sites were not strongly impacted (Fig. S1). Thus, these findings support Raitzsch et al. (2018), who  
473 concluded that calculated  $\delta^{11}\text{B}_{\text{borate}}$  values corrected for seasonality was within error of non-corrected values for  
474 each water depth. As Raitzsch et al, (2018) highlight, seasonality might be more important at high latitude sites  
475 where seasonality is more marked, however, the seasonality of primary production will also be more tightly  
476 constrained due to the seasonal progression of winter light limitation and intense vertical mixing and summer  
477 nutrient limitation.

478 Data for our sites suggests that most  $\delta^{11}\text{B}_{\text{borate}}$  variability we observe does not come from seasonality but  
479 from the assumed water depths for calcification. With the exception of a few specific areas such as the Red Sea  
480 (Henehan et al., 2016, Raitzsch et al., 2018), at most sites examined, seasonal  $\delta^{11}\text{B}_{\text{borate}}$  at a fixed depth does not  
481 vary by more than ~0.2%. We conclude that seasonality is not an important factor impacting carbonate system  
482 parameters at the sites we examined.

#### 484 5.2 $\delta^{11}\text{B}$ , microenvironment pH and depth habitats

485 In planktonic foraminifera, algal symbiosis is the more common symbiotic relationship. For most of  
486 planktonic foraminifera, the host presents only one species of symbionts (Gast and Caron, 2001). The family  
487 Globigerinidae, including *G. ruber*, *T. sacculifer* and *O. universa*, commonly have dinoflagellates or chrysophyte  
488 algal symbionts (Anderson and Be, 1976; Spero, 1987). The families Pulleniatinidae, Globorotaliidae, including  
489 *N. dutertrei*, *P. obliquiloculata*, *G. menardii* and *G. tumida*, have chrysophyte algal symbionts (Gastrich, 1988).

490 The relationship between the symbionts and the host is complex by nature. Nevertheless, this symbiotic  
491 relationship provides energy (Hallock, 1981b) and promotes calcification of the foraminifera (Duguay, 1983; Erez  
492 et al., 1983) by providing the inorganic carbon to the host (Jorgensen et al., 1985). Also, for *T. sacculifer* and *O.*  
493 *universa* photosynthesis increases with higher insolation (Jorgensen et al., 1985; Rink et al., 1998).

494 Dinoflagellate-bearing foraminifera (*G. ruber*, *T. sacculifer* and *O. universa*) tend to have a higher  
495 symbiont density and photosynthesis activity while *P. obliquiloculata*, *G. menardii* and *N. dutertrei* have lowered  
496 symbiont density and *P. obliquiloculata*, *N. dutertrei* lower photosynthetic activity (Takagi et al., 2019). *P.*  
497 *obliquiloculata* showed the minimum symbiont density and photosynthetic activity (Takagi et al., 2019).

498 It is now accepted that the foraminifera  $\delta^{11}\text{B}$  signature comes from the microenvironment pH (Jorgensen  
499 et al., 1985; Rink et al., 1998; Köhler-Rink and Köhl, 2000, Hönisch et al., 2003; Zeebe et al., 2003). Foraminifera

500 with high photosynthetic activity and symbiont density like *G. ruber* and *T. sacculifer* present a pH of  
501 microenvironment higher than ambient seawater,  $\delta^{11}\text{B}$  higher than 1:1 line (Foster et al., 2008, Henehan et al.,  
502 2013, Raitzsch et al., 2018). The opposite can also be true, from our study, species with lower photosynthetic  
503 activity and lower symbiont density present microenvironments lower than ambient seawater,  $\delta^{11}\text{B}$  lower than 1:1  
504 line (Martinez-Boti et al., 2015b; Henehan et al., 2016), this is the case in our data for *N. dutertrei*, *G. menardii*  
505 and *P. obliquiloculata* and likely *G. tumida*. Nevertheless, the low  $\delta^{11}\text{B}$  of *O. universa* and *T. sacculifer* (w/o sacc)  
506 from the WEP are difficult to reconcile with a high photosynthetic activity compared to *T. sacculifer* et *G. ruber*.

507 The photosynthetic activity is also function of the light level they received which is, in the natural system,  
508 dependent of their depth in the water column, for the purpose of this study we will not consider turbidity which  
509 also influences the light penetration in the water column. In this case, the photosynthetically active foraminifera  
510 living close to the surface should see their microenvironment pH (thus  $\delta^{11}\text{B}$ ) more sensitive to water depth changes.  
511 A deeper depth habitat will change the light intensity they received and as a consequence may lower their  
512 photosynthetic activity reducing their microenvironment pH. This thought is supported by the significant trend  
513 observed between our  $\Delta^{11}\text{B}$  and the calcification depth for *G. ruber* and *T. sacculifer* of our sites (Fig. S2). This  
514 trend basically supports the fact that the microenvironment pH decrease with calcification depth. We observe a  
515 decrease of  $\delta^{11}\text{B}$  in the WEP for *T. sacculifer* (w/o sacc), significantly different from the other sites ( $p < 0.05$ ). The  
516  $\Delta^{11}\text{B}$  of *G. ruber*, *T. sacculifer* (w/o sacc and sacc) is also significantly lower in the WEP compared to the other  
517 sites ( $p < 0.05$ ). To test if the  $\delta^{11}\text{B}$  signature was inferred to a light driven, we have been able to independently  
518 calculate the depth of the foraminifera based on various light insolation culture experiments (Jorgensen et al.,  
519 1985) and the  $\Delta$ microenvironment pH derived from our data (Fig. 8A and B). This exercise verified that this low  
520  $\delta^{11}\text{B}$  can be explained by the reduced light environment due to a deeper depth habitat in the WEP (Fig. 8B). Also,  
521 *T. sacculifer* has the potential to support more photosynthesis due to its higher symbiont density. Higher  
522 photosynthetic activity is observed compared to other species potentially supporting higher symbiont/host  
523 interactions. Those results could be in line with a greater sensitivity of *T. sacculifer* photosynthetic activity with  
524 changes in insolation/water depth. It can also be noted that this species presents the largest variations in symbiont  
525 density versus its test size. When applied to the other species *O. universa* data suggest a microenvironment pH  
526 0.10 to 0.20 lower than ambient seawater pH which would be in line with species living deeper than 50m (light  
527 compensation point (Ec), Rink et al., 1998) also consistent with our calcification depth reconstructions.  
528  $\Delta$ microenvironment pH is higher in *T. sacculifer* > *G. ruber* > *T. sacculifer* (w/o sacc - WEP) > *O. universa*, *N.*  
529 *dutertrei*, *G. menardii*, *G. tumida* > *P. obliquiloculata* in line with photosymbiosis findings from Tagazaki et al.,  
530 (2019). Also, the higher  $\delta^{11}\text{B}$  data from the African upwelling published by Raitzsch et al., (2018) for *G. ruber*  
531 and *O. universa* might reflect the higher microenvironment pH due to a shallower depth habitat. This could  
532 highlight a potential issue with calibration when applied to sites with different oceanic regimes as the  $\delta^{11}\text{B}$  species-  
533 specific calibrations could be also location-specific for the mixed dweller species.

534 Microenvironment pH results for *N. dutertrei*, *G. menardii*, *G. tumida*, are similar to *O. universa* and  
535 suggest a threshold for respiration driven  $\delta^{11}\text{B}$  signature. This threshold can be driven by a change of  
536 photosynthetic activity due to lower light intensity at deeper depth and/or a change in the symbiont assemblage  
537 with non-dinoflagellate symbionts at deeper depth. We can explain this threshold because deep dweller species do  
538 not experience important changes of insolation at those depths so their microenvironments should be respiration  
539 driven and relatively stable. We can also note that *P. obliquiloculata* which has the lowest symbiont density and



540 photosynthetic activity has the lowest microenvironment pH compared to other deeper dweller species supporting  
541 this respiration driven microenvironment.

542

### 543 5.3 $\delta^{11}\text{B}$ sensitivity to $\delta^{11}\text{B}_{\text{borate}}$ and relationship with B/Ca signatures

544  $\delta^{11}\text{B}_{\text{carbonate}}$  and B/Ca data have shown to be sensitive to precipitation rate with at higher precipitation rate  
545 increasing  $\delta^{11}\text{B}_{\text{carbonate}}$  (Farmer et al., 2019) and B/Ca (Farmer et al., 2019; Gabitov et al., 2014; Kaczmarek et al.,  
546 2016; Mavromatis et al., 2015; Ushikawa et al., 2015). A recent study from Farmer et al. (2019) has proposed that  
547 in foraminifera at higher precipitation rates, more borate ion is incorporated into the carbonate mineral, while at  
548 lower precipitation rates, more boric acid is incorporated. They also suggest this may explain low sensitivities of  
549 culture experiments.

550 When combining all literature data, *T. sacculifer* and *G. ruber* have sensitivities of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$   
551 of  $0.83 \pm 0.48$  and  $0.46 \pm 0.34$  respectively in line with previous literature and paleo- $\text{CO}_2$  reconstructions. Also, if  
552 we only take into account our data, the observation that the sensitivity of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  are not statistically  
553 different from unity for most of the species investigated we can speculate that for these taxa, changes in  
554 precipitation rate and contributions of boric acid are not likely to be important. If considering only the data from  
555 this study, *G. ruber* ( $1.12 \pm 1.67$ ) and *T. sacculifer* ( $1.38 \pm 1.35$ ) present higher sensitivities of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  
556  $\delta^{11}\text{B}_{\text{borate}}$ . We can then again speculate that the observed high values for  $\delta^{11}\text{B}_{\text{carbonate}}$  at high seawater pH can be due  
557 to higher precipitation rates. We note this could also be consistent with the higher sensitivity of B/Ca signatures  
558 in these two surface dwelling species to ambient  $[\text{B}(\text{OH})_4^-]/[\text{HCO}_3^-]$  relative to deeper dwelling species. Those  
559 interspecific differences still remain to be explained, however, part of this variability is likely due to changes in  
560 the carbonate chemistry of the microenvironment resulting in changing competition between borate and  
561 bicarbonate. A caveat is that we can not exclude specific biological processes, and that in taxa with a non  
562 respiration-driven microenvironment, changes in day/night calcification ratios also impacting observed values. As  
563 indicated by Farmer et al., (2019), studies of calcite precipitation rates in foraminifera may help to improve our  
564 understanding of the fundamental basis of boron-based proxies.

565

### 566 5.4 Evaluation of species for pH reconstructions and water depth pH reconstructions

567 This data set allows us to reassess the utility of boron-based proxies for the carbonate system. The main  
568 interest with utilizing boron-based proxies relates to the reconstruction of past oceanic conditions - specifically pH  
569 and  $\text{pCO}_2$ . Mixed-layer species (eg. *G. ruber* and *T. sacculifer*) are potential archives for atmospheric  $\text{CO}_2$   
570 reconstructions. Other species can shed light on other aspects of the carbon cycle including the physical and  
571 biological carbon pumps.

572 There are a few main inferences we can make. When compiled with data from the literature, sensitivities  
573 of  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  for *G. ruber* and *T. sacculifer* are similar to previous studies (Martinez-Boti et al., 2015b;  
574 Raitzsch et al., 2018) which is also supporting of previous paleo-reconstructions. Our data also support the  
575 observations of Henehan et al., (2016) for *O. universa*.

576 In order to derive accurate reconstructions of past ambient pH and  $\text{pCO}_2$ , accurate species-specific  
577 calibrations need to be used that are constrained by core-tops or samples from similar types of settings (Fig. 9, 10,  
578 S6). Lighter  $\delta^{11}\text{B}$  signatures in *T. sacculifer* (w/o sacc) are observed in the WEP, which may be explained by the  
579 deeper depth habitat for these taxa, where lower light levels might reduce symbiont photosynthetic activity. Also,

580 correction will be needed for *T. sacculifer* (w/o sacc) in the WEP. When applying the calibrations n°2 and 4 to *T.*  
581 *sacculifer* and *G. ruber* (compilation of all data, Table 3) our data show more variability, especially for *G. ruber*  
582 which lead to the larger mismatch compared to *in-situ* parameters. Henehan et al., (2013) reported a lighter  $\delta^{11}\text{B}$   
583 with smaller test size, our sample add a weight/shell of  $11 \pm 4 \mu\text{g}$  (n=4, SD) which, despite a narrow range, could  
584 still explain this variability. The higher divergence of reconstructed values from *in-situ* measurements are observed  
585 at site WPO7-01 for both *T. sacculifer* (w/o sacc) and *G. ruber*. More data would be needed to determine a proper  
586 correction for both species, coretop study will be determinant for future downcore reconstructions, especially in  
587 the WEP.

588 We also find that for two species, the boron proxy is a relatively straightforward recorder of ambient pH,  
589 with sensitivities close to unity for *O. universa*, and *N. dutertrei*. There is also promise in using multiple  
590 species in a sample from different hydrographic regimes to reconstruct vertical profiles of pH and pCO<sub>2</sub>. We are  
591 able to reproduce pH and pCO<sub>2</sub> profiles from multiple sites with different water column structures (Fig. 9) with  
592 those reconstructions within error of the *in-situ* values, for most sites. In order to avoid circularity, to validate these  
593 calibrations, we recalculated ambient pH and pCO<sub>2</sub> by first excluding site-specific data and then recalculating  
594 species-specific calibrations, followed by application to each specific site. The comparison of the two methods  
595 does not show significant differences and validates the robustness of the calibrations (Fig. S5). We utilized the  
596 calibrations derived from our data for *G. ruber* (calibration n°1 and 2, Table 3), *T. sacculifer* (calibration n°3 and  
597 4, Table 3), *O. universa* (calibration n°8, Table 3), for *P. obliquiloculata* (calibration n°11, Table 3), and for *N.*  
598 *dutertrei*, *G. tumida* and *G. menardii* the calibration made on the compilation of the deep-dweller (calibration  
599 n°13, Table 3). Results are shown in Fig. 9 and evaluated in Fig. 10. For *G. menardii*, more data would be helpful  
600 to provide additional constraints. Results for *G. ruber* are the sparsest, potentially due to difference in test sizes  
601 (Henehan et al., 2013) or undocumented diagenetic effects. Results reaffirm the importance of working with  
602 narrow size fractions (Henehan et al., 2013) and the importance of core-top study to determine corrections.

603

## 604 6. Conclusions and future implications

605 Our study has extended the boron isotope proxy with data for new species and sites. The work supports  
606 previous work showing that depth habitats of foraminifera vary depending on the oceanic regime, and this impacts  
607 boron isotope signatures. Low  $\delta^{11}\text{B}$  values in the WEP compared to other regions for *T. sacculifer* (w/o sacc) may  
608 be explained by a reduction in microenvironment pH due to a deeper depth habitat associated with reduced  
609 irradiance and thus photosynthetic activity. Those results might also highlight a potential need for studying core-  
610 tops in order to establish what factors are important to accurately develop reconstructions in different areas.

611 The sensitivity of  $\delta^{11}\text{B}$  to pH is in line with previously published data for *T. sacculifer*, *G. ruber*. The  
612 sensitivity of  $\delta^{11}\text{B}$  to pH of *O. universa* (mixed-dweller), *N. dutertrei*, *G. menardii* and *G. tumida* (deep-dwellers)  
613 are similar but more data are needed to fully determine those sensitivities. The similarity of boron isotope  
614 calibrations for deep-dwelling taxa might be related to respiration-driven microenvironments.

615 Reconstruction of seawater pH and carbonate system parameters is achievable using foraminiferal  $\delta^{11}\text{B}$   
616 but additional core-top and down-core studies reconstructing depth profiles will be needed in order to further verify  
617 those calibrations. Past pH and pCO<sub>2</sub> water depth profiles can potentially be created by utilizing multiple  
618 foraminiferal species in concert with taxa-specific calibrations for similar settings. This approach has much

619 potential for enhancing our understanding of the past workings of the oceanic carbon cycle, and the biological  
620 pump.  
621

622 **Author contribution**

623 R.E and A.T. wrote the proposals that funded the work. A.T. and F.C. provided the samples. M.G., S.M. and A.T.  
624 contributed to the experimental design. A.V. helped for sample preparation. M.G. and S.M contributed to  
625 developing the method of boron isotope analysis. M.G. performed the measurements with assistance from S.M.  
626 M.G conducted the data analysis. M.G. drafted the paper, which was edited by all authors. Interpretation was led  
627 by M.G., A.T., S.M. with input from R.E., A.V. and F.C.

628

629 **Competing interests**

630 The authors declare that they have no conflict of interest.

631

632 **Acknowledgments:**

633 The authors wish to thank Jesse Farmer for his valuable and detailed comments on the actual and a previous version  
634 of the manuscript. We wish to thank Michael Henehan for helpful discussion, comments on the manuscript and  
635 help with the code. We also want to thank the anonymous reviewer for helpful comments. Lea Bonnin for  
636 assistance with picking samples, the IODP repository for provision of samples, the Tripati Laboratory (UCLA) for  
637 their technical support, Mervyn Greaves, Madeleine Bohlin (University of Cambridge) for technical support and  
638 use of laboratory space, Yoan Germain, Emmanuel Ponzevera and Oanez Lebeau for technical support and use of  
639 laboratory space in Brest, Jill Sutton for helpful conversation on the manuscript. Research is supported by DOE  
640 BES grant DE-FG02-13ER16402, by the International Research Chair Program that is funded by the French  
641 government (LabexMer ANR-10-LABX-19-01), and IAGC student research grant 2017.

642

643 **References**

- 644 Allen, K. A. and Hönisch, B.: The planktic foraminiferal B/Ca proxy for seawater carbonate chemistry: A critical  
645 evaluation, *Earth Planet. Sci. Lett.*, 345–348, 203–211, 2012.
- 646 Anand, P., Elderfield, H. and Conte, M. H.: Calibration of Mg/Ca thermometry in planktonic foraminifera from a  
647 sediment trap time series. *Paleoceanography* 18, 2003.
- 648 Anderson, O. R. and Bé, A.W. H.: The ultrastructure of a planktonic foraminifer, *Globigerinoides sacculifer*  
649 (Brady), and its symbiotic dinoflagellates, *J. Foramin. Res.*, 6, 1–21, 1976.
- 650 Arbuszewski, J., DeMenocal, P., Kaplan, A. and Farmer, E. C.: On the fidelity of shell-derived  $\delta^{18}\text{O}$  seawater  
651 estimates, *Earth Planet. Sci. Lett.*, 300, 185–196, 2010.
- 652 Axelsson, M. D., Rodushkin, I., Ingri, J. and Öhlander, B.: Multielemental analysis of Mn–Fe nodules by ICP-  
653 MS: optimisation of analytical method, *Analyst*, 127, 76–82, 2002.
- 654 Babila, T.L., Rosenthal, Y., Conte, M.H.: Evaluation of the biogeochemical controls on B/Ca of *Globigerinoides*  
655 *ruber* white from the Oceanic Flux Pro-gram, Bermuda. *Earth Planet. Sci. Lett.* 404, 67–76, 2014.
- 656 Barker S., Greaves M. and Elderfield H.: A study of cleaning procedures used for foraminiferal Mg/Ca  
657 paleothermometry. *Geochemistry, Geophys. Geosystems* 4, 1–20, 2003.
- 658 Bartoli, G., Hönisch, B. and Zeebe, R. E.: Atmospheric CO<sub>2</sub> decline during the Pliocene intensification of  
659 Northern Hemisphere glaciations. *Paleoceanography* 26, 1–14, 2011.
- 660 Bemis, B. E., Spero, H. J., Bijma, J. and Lea, D. W.: Reevaluation of the oxygen isotopic composition of  
661 planktonic foraminifera: Experimental results and revised paleotemperature equations. *Paleoceanography*  
662 13, 150–160, 1998.
- 663 Bemis, B. E., Spero, H. J. and Thunell, R. C.: Using species-specific paleotemperature equations with  
664 foraminifera: a case study in the Southern California Bight, *Mar. Micropaleontol.*, 46, 405–430, 2002.
- 665 Bijma, J., Faber Jr., W.W., Hemleben, C.: Temperature and salinity limits for growth and survival of some  
666 planktonic foraminifera in laboratory cultures, *J. Foraminiferal Res.* 20 (2), 95–116, 1990.
- 667 Bijma, J., Hönisch, B. and Zeebe, R. E.: Impact of the ocean carbonate chemistry on living foraminiferal shell  
668 weight: Comment on “Carbonate ion concentration in glacial-age deep waters of the Caribbean Sea” by W.  
669 S. Broecker and E. Clark, *Geochemistry, Geophys. Geosystems*, 3, 1–7, 2002.
- 670 Birch, H., Coxall, H. K., Pearson, P. N., Kroon, D. and O’Regan, M.: Planktonic foraminifera stable isotopes and  
671 water column structure: Disentangling ecological signals, *Mar. Micropaleontol.*, 101, 127–145, 2013.
- 672 Boyer, T.P., Antonov, J. I., Baranova, O. K., Coleman, C., Garcia, H. E., Grodsky, A., Johnson, D. R., Locarnini,  
673 R. A., Mishonov, A. V., O’Brien, T.D., Paver, C.R., Reagan, J.R., Seidov, D., Smolyar, I. V., and Zweng,  
674 M. M.: World Ocean Database, NOAA Atlas NESDIS 72, S. Levitus, Ed., A. Mishonov, Technical Ed.,  
675 Silver Spring, MD, 209, 2013.
- 676 Boyle, E. A.: Manganese carbonate overgrowths on foraminifera tests, *Geochim. Cosmochim. Acta.*, 47, 1815–  
677 1819, 1983.
- 678 Branson, O., Kaczmarek, K., Redfern, S. A. T., Misra, S., Langer, G., Tyliszczak, T., Bijma, J. and Elderfield,  
679 H.: The coordination and distribution of B in foraminiferal calcite, *Earth Planet. Sci. Lett.*, 416, 67–72,  
680 2015.

- 681 Catanzaro, E.J., Champion, C.E., Garner, A.L., Marinenko, G., Sappenfield, K.M. and Shields, W.R.: Boric  
682 Acid; Isotopic and Assay Standard Reference Materials. U.S. Natl. Bur. Stand. Spec., Publ. 260-17, 70p,  
683 1970.
- 684 Chalk, T. B., Hain, M. P., Foster, G. L., Rohling, E. J., Sexton, P. F., Badger, M. P. S., Cherry, S. G., Hasenfratz,  
685 A. P., Haug, G. H., Jaccard, S. L., Martínez-García, A., Pälike, H., Pancost, R. D. and Wilson, P. A.:  
686 Causes of ice age intensification across the Mid-Pleistocene Transition, Proc. Natl. Acad. Sci., 114,  
687 13114–13119, 2017.
- 688 Coadic, R., Bassinot, F., Dissard, D., Douville, E., Greaves, M. and Michel, E.: A core-top study of dissolution  
689 effect on B/Ca in Globigerinoides sacculifer from the tropical Atlantic: Potential bias for paleo-  
690 reconstruction of seawater carbonate chemistry, Geochemistry, Geophys. Geosystems 14, 1053–1068,  
691 2013.
- 692 de Nooijer, L. J., Spero, H. J., Erez, J., Bijma, J. and Reichart, G. J.: Biomineralization in perforate foraminifera.  
693 Earth-Science Rev., 135, 48–58, 2014.
- 694 Dekens, P. S., Lea, D. W., Pak, D. K. and Spero, H. J.: Core top calibration of Mg/Ca in tropical foraminifera:  
695 Refining paleotemperature estimation, Geochemistry, Geophys. Geosystems 3, 1–29, 2002.
- 696 Deuser, W.G., Ross, E.H., Hemleben, Ch., Spindler, M.: Seasonal changes in species composition, numbers,  
697 mass, size, and isotopic composition of planktonic foraminifera settling into the deep Sargasso Sea,  
698 Palaeogeogr., Palaeoclimat., Palaeoecol., 33:103-127, 1981.
- 699 Deuser, W. G. and Ross, E. H.: Seasonally abundant planktonic foraminifera of the Sargasso Sea; succession,  
700 deep-water fluxes, isotopic compositions, and paleoceanographic implications, J. Foraminifer. Res. 19,  
701 268–293, 1989.
- 702 Dickson, A. G.: Thermodynamics of the dissociation of boric acid in synthetic seawater from 273.15 to 318.15  
703 K., Deep Sea Res., Part A, Oceanogr. Res. Pap. 37, 755–766, 1990.
- 704 Dickson, A.G., Millero, F.J.: A comparison of the equilibrium constants for the dissociation of carbonic acid in  
705 seawater media, Deep-Sea Res., 34, 1733–1743, 1987.
- 706 Douville, E., Paterne, M., Cabioch, G., Louvat, P., Gaillardet, J., Juillet-Leclerc, A. and Ayliffe, L.: Abrupt sea  
707 surface pH change at the end of the Younger Dryas in the central sub-equatorial Pacific inferred from  
708 boron isotope abundance in corals (Porites), Biogeosciences 7, 2445–2459, 2010.
- 709 Duguay, L.E.: Comparative laboratory and field studies on calcification and carbon fixation in foraminiferal-  
710 algal associations, Journal of Foraminiferal Research 13, 252-261, 1983.
- 711 Duplessy, J., Labeyrie, L., Juilletleclerc, A., Maitre, F., Duprat, J. and Sarthein, M.: Surface salinity  
712 reconstruction of the north-atlantic ocean during the last glacial maximum, Oceanol. Acta, 14, 311–324,  
713 1991.
- 714 Elderfield, H., Yu, J., Anand, P., Kiefer, T. and Nyland, B.: Calibrations for benthic foraminiferal Mg/Ca  
715 paleothermometry and the carbonate ion hypothesis, Earth Planet. Sci. Lett., 250, 633–649., 2006.
- 716 Elderfield, H. and Granssen, G.: Past temperatures and O18 of surface ocean waters inferred from foraminiferal  
717 Mg/Ca ratios, Nature 405, 442–445, 2000.
- 718 Erez J.: Calcification Rates, Photosynthesis and Light in Planktonic Foraminifera. In: Westbroek P., de Jong  
719 E.W. (eds) Biomineralization and Biological Metal Accumulation. Springer, Dordrecht, 1983.
- 720 Erez, J.: The Source of Ions for Biomineralization in Foraminifera and Their Implications for Paleocceanographic  
721 Proxies, Rev. Mineral. Geochemistry, 54, 115–149, 2003.

- 722 Fairbanks, R. G. and Wiebe, P. H.: Foraminifera and Chlorophyll Maximum: Vertical Distribution, Seasonal  
723 Succession, and Paleoceanographic Significance, *Science*, 209, 1524–1526, 1980.
- 724 Fairbanks, R. G., Sverdløve, M., Free, R., Wiebe, P. H. and Bé, A. W. H.: Vertical distribution and isotopic  
725 fractionation of living planktonic foraminifera from the Panama Basin, *Nature*, 298, 841–844, 1982.
- 726 Farmer, E. C., Kaplan, A., de Menocal, P. B. and Lynch-Stieglitz, J.: Corroborating ecological depth preferences  
727 of planktonic foraminifera in the tropical Atlantic with the stable oxygen isotope ratios of core top  
728 specimens, *Paleoceanography*, 22, 1–14, 2007.
- 729 Feely, R.: Impact of Anthropogenic CO<sub>2</sub> on the CaCO<sub>3</sub> System in the Oceans, *Science*, 305, 362–366, 2004.
- 730 Ferguson, J. E., Henderson, G. M., Kucera, M. and Rickaby, R. E. M.: Systematic change of foraminiferal  
731 Mg/Ca ratios across a strong salinity gradient, *Earth Planet. Sci. Lett.*, 265, 153–166, 2008.
- 732 Foster, G. L.: Seawater pH, pCO<sub>2</sub> and [CO<sub>3</sub><sup>2-</sup>] variations in the Caribbean Sea over the last 130 kyr: A boron  
733 isotope and B/Ca study of planktic foraminifera, *Earth Planet. Sci. Lett.*, 271, 254–266, 2008.
- 734 Foster, G. L. and Sexton, P. F.: Enhanced carbon dioxide outgassing from the eastern equatorial Atlantic during  
735 the last glacial, *Geology*, 42, 1003–1006, 2014.
- 736 Foster, G. L., Lear, C. H. and Rae, J. W. B.: The evolution of pCO<sub>2</sub>, ice volume and climate during the middle  
737 Miocene, *Earth Planet. Sci. Lett.*, 341–344, 243–254, 2012.
- 738 Foster, G. L. and Rae, J. W. B.: Reconstructing Ocean pH with Boron Isotopes in Foraminifera, *Annu. Rev.*  
739 *Earth Planet. Sci.*, 44, 207–237, 2016.
- 740 Gabitov, R. I., Rollion-bard, C., Tripathi, A. and Sadekov, A.: In situ study of boron partitioning between calcite  
741 and fluid at different crystal growth rates, *Geochim. Cosmochim. Acta*, 137, 81–92, 2014.
- 742 Gaillardet, J., Lemarchand, D., Göpel, C. and Manhès, G.: Evaporation and Sublimation of Boric Acid :  
743 Application for Boron Purification from Organic Rich Solutions, *Geostand. Newsl.*, 25, 67–75, 2001.
- 744 Gast, R. J. and Caron D. A.: Photosymbiotic associations in planktonic foraminifera and radiolaria, 1–7, 2001.
- 745 Gastrich, M.D.: Ultrastructure of a new intracellular symbiotic alga found within planktonic foraminifera,  
746 *Journal of Phycology* 23, 623-632, 1988.
- 747 Gattuso, J.P. and Hansson, L.: *Ocean acidification*, Oxford University Press, 2011.
- 748 Gutjahr, M., Bordier, L., Douville, E., Farmer, J., Foster, G. L., Hathorne, E., Hönisch, B., Lemarchand, D., Louvat,  
749 P., McCulloch, M., Noireaux, J., Pallavicini, N., Rodushkin, I., Roux, P., Stewart, J., Thil, F. You, C.F.:  
750 Boron Isotope Intercomparison Project (BIIP): Development of a new carbonate standard for stable isotopic  
751 analyses. In EGU general assembly conference abstracts, Vol. 16, (2014).
- 752 Hallock P.: *Algal Symbiosis : A Mathematical Analysis Marine Biology* 62, 249-255, 1981b.
- 753 Hemming, N. G. and Hanson, G. N. Boron isotopic composition and concentration in modern marine carbonates,  
754 *Geochim. Cosmochim. Acta*, 56, 537–543, 1992.
- 755 Hendry, K.R., Rickaby, R.E.M., Meredith, M.P., Elderfield, H.: Controls on stable isotope and trace metal  
756 uptake in *Neogloboquadrina pachyderma* (sinistral) from an Antarctic sea-ice environment. *Earth Planet.*  
757 *Sci. Lett.* 278, 67–77, 2009.

- 758 Henehan, M. J., Foster, G. L., Bostock, H. C., Greenop, R., Marshall, B. J. and Wilson, P. A.: A new boron  
759 isotope-pH calibration for *Orbulina universa*, with implications for understanding and accounting for ‘vital  
760 effects’, *Earth Planet. Sci. Lett.*, 454, 282–292, 2016.
- 761 Henehan, M. J., Foster, G. L., Rae, J. W. B., Prentice, K. C., Erez, J., Bostock, H. C., Marshall, B. J. and Wilson,  
762 P. A.: Evaluating the utility of B/Ca ratios in planktic foraminifera as a proxy for the carbonate system: A  
763 case study of *Globigerinoides ruber*, *Geochemistry, Geophys. Geosystems* 16, 1052–1069, 2015.
- 764 Henehan, M. J., Rae, J. W. B., Foster, G. L., Erez, J., Prentice, K. C., Kucera, M., Bostock, H. C., Martínez-Botí,  
765 M. A., Milton, J. A., Wilson, P. A., Marshall, B. J. and Elliott, T.: Calibration of the boron isotope proxy in  
766 the planktonic foraminifera *Globigerinoides ruber* for use in palaeo-CO<sub>2</sub> reconstruction, *Earth Planet. Sci.*  
767 *Lett.* 364, 111–122, 2013.
- 768 Holcomb, M., Decarlo, T. M., Schoepf, V., Dissard, D., Tanaka, K. and McCulloch, M.: Cleaning and pre-  
769 treatment procedures for biogenic and synthetic calcium carbonate powders for determination of elemental  
770 and boron isotopic compositions, *Chem. Geol.*, 398, 11–21, 2015.
- 771 Hönisch, B., Hemming, N. G., Archer, D., Siddall, M. and McManus, J. E.: Atmospheric Carbon Dioxide  
772 Concentration Across the Mid-Pleistocene Transition, *Science*, 324, 1551–1554, 2009.
- 773 Hönisch, B., Bijma, J., Russell, A. D., Spero, H. J., Palmer, M. R., Zeebe, R. E. and Eisenhauer, A.: The  
774 influence of symbiont photosynthesis on the boron isotopic composition of foraminifera shells, *Mar.*  
775 *Micropaleontol.*, 49, 87–96, 2003.
- 776 Hönisch, B. and Hemming, N. G.: Ground-truthing the boron isotope-paleo-pH proxy in planktonic foraminifera  
777 shells: Partial dissolution and shell size effects, *Paleoceanography* 19, 1–13, 2004.
- 778 Hönisch, B., Bickert, T. and Hemming, N. G.: Modern and Pleistocene boron isotope composition of the benthic  
779 foraminifer *Cibicides wuellerstorfi*, *Earth Planet. Sci. Lett.*, 272, 309–318, 2008.
- 780 Howes, E. L., Kaczmarek, K., Raitzsch, M., Mewes, A., Bijma, N., Horn, I., Misra, S., Gattuso, J. P. and Bijma,  
781 J.: Decoupled carbonate chemistry controls on the incorporation of boron into *Orbulina universa*,  
782 *Biogeosciences*, 14, 415–430, 2017.
- 783 IPCC: Climate Change 2014 - The Physical Science Basis, edited by Intergovernmental Panel on Climate  
784 Change, Cambridge University Press, Cambridge., 2014.
- 785 Jørgensen, B. B., Erez, J., Revsbech, P. and Cohen, Y.: Symbiotic photosynthesis in a planktonic foraminifera,  
786 *Globigerinoides sacculifer* (Brady), studied with microelectrodes, *Limnol. Oceanogr.*, 30, 1253–1267  
787 1985.
- 788 Kaczmarek, K., Nehrke, G., Misra, S., Bijma, J. and Elderfield, H.: Investigating the effects of growth rate and  
789 temperature on the B/Ca ratio and  $\delta^{11}\text{B}$  during inorganic calcite formation, *Chem. Geol.*, 421, 81–92,  
790 2016.
- 791 Kemle-von Mücke S. and Oberhänsli H.: The Distribution of Living Planktic Foraminifera in Relation to  
792 Southeast Atlantic Oceanography, *Use Proxies Paleocyanogr.*, 91–115, 1999.
- 793 Key, R.M.: A global ocean carbon climatology: Results from Global Data Analysis Project (GLODAP), *Global*  
794 *Biogeochem. Cycles*, 18, GB4031, 2004.
- 795 Kim, S.-T. and O’Neil, J. R.: Equilibrium and nonequilibrium oxygen isotope effects in synthetic carbonates,  
796 *Geochim. Cosmochim. Acta*, 61, 3461–3475, 1997.
- 797 Klochko, K., Cody, G. D., Tossell, J. A., Dera, P. and Kaufman, A. J.: Re-evaluating boron speciation in  
798 biogenic calcite and aragonite using  $^{11}\text{B}$  MAS NMR, *Geochim. Cosmochim. Acta*, 73, 1890–1900, 2009.



- 799 Klochko, K., Kaufman, A. J., Yao, W., Byrne, R. H. and Tossell, J. A.: Experimental measurement of boron  
800 isotope fractionation in seawater, *Earth Planet. Sci. Lett.*, 248, 276–285, 2006.
- 801 Köhler-Rink, S. and Kühl, M.: Microsensor studies of photosynthesis and respiration in larger symbiotic  
802 foraminifera. I. The physico-chemical microenvironment of *Marginopora vertebralis*, *Amphistegina*  
803 *lobifera* and *Amphisorus hemrichii*, *Mar. Biol.*, 137, 473–486, 2000.
- 804 Köhler-Rink, S. and Kühl, M., Microsensor studies of photosynthesis and respiration in the larger symbiont  
805 bearing foraminifera *Amphistegina lobifera*, and *Amphisorus hemprichii*, *Ophelia*, 55, 111–122, 2001.
- 806 Lea, D. W., Pak, D. K. and Spero, H. J.: Climate impact of late quaternary equatorial Pacific sea surface  
807 temperature variations, *Science*, 289, 1719–1724, 2000.
- 808 Lemarchand, D., Gaillardet, J., Lewin, A. and Allègre, C. J.: Boron isotope systematics in large rivers:  
809 Implications for the marine boron budget and paleo-pH reconstruction over the Cenozoic, *Chem. Geol.*,  
810 190, 123–14, 2002.
- 811 Liu, Y., Liu, W., Peng, Z., Xiao, Y., Wei, G., Sun, W., He, J., Liu, G. and Chou, C.-L.: Instability of seawater  
812 pH in the South China Sea during the mid-late Holocene: Evidence from boron isotopic composition of  
813 corals, *Geochim. Cosmochim. Acta*, 73, 1264–1272, 2009.
- 814 Lloyd, N. S., Sadekov, A. Y. and Misra, S.: Application of 1013ohm Faraday cup current amplifiers for boron  
815 isotopic analyses by solution mode and laser ablation multicollector inductively coupled plasma mass  
816 spectrometry, *Rapid Commun. Mass Spectrom.*, 32, 9–18, 2018.
- 817 Martínez-Botí, M. A., Foster, G. L., Chalk, T. B., Rohling, E. J., Sexton, P. F., Lunt, D. J., Pancost, R. D.,  
818 Badger, M. P. S. and Schmidt, D. N.: Plio-Pleistocene climate sensitivity evaluated using high-resolution  
819 CO<sub>2</sub> records, *Nature*, 518, 49–54, 2015a.
- 820 Martínez-Botí M. A., Marino G., Foster G. L., Ziveri P., Henehan M. J., Rae J. W. B., Mortyn P. G. and Vance  
821 D.: Boron isotope evidence for oceanic carbon dioxide leakage during the last deglaciation. *Nature*, 518,  
822 219–222, 2015b.
- 823 Martínez-Botí, M. A., Mortyn, P. G., Schmidt, D. N., Vance, D. and Field, D. B.: Mg/Ca in foraminifera from  
824 plankton tows: Evaluation of proxy controls and comparison with core tops, *Earth Planet. Sci. Lett.*, 307,  
825 113–125, 2011.
- 826 Mavromatis, V., Montouillout, V., Noireaux, J., Gaillardet, J. and Schott, J.: Characterization of boron  
827 incorporation and speciation in calcite and aragonite from co-precipitation experiments under controlled  
828 pH, temperature and precipitation rate, *Geochim. Cosmochim. Acta*, 150, 299–313, 2015.
- 829 McCulloch, M. T., D’Olivo, J. P., Falter, J. L., Georgiou, L., Holcomb, M., Montagna, P. and Trotter, J. A.:  
830 Boron Isotopic Systematics in Scleractinian Corals and the Role of pH Up-regulation, *Boron Isot. Adv.*  
831 *Isot. Geochemistry*, 2018.
- 832 Millero, F.: Speciation of metals in natural waters, *Geochem. Trans.*, 2, 57, 2001.
- 833 Millero, F., Woosley, R., DiTrollo, B. and Waters, J.: Effect of Ocean Acidification on the Speciation of Metals  
834 in Seawater, *Oceanography* 22, 72–85, 2009.
- 835 Misra, S., Greaves, M., Owen, R., Kerr, J., Elmore, A. C. and Elderfield, H.: Determination of B/Ca of natural  
836 carbonates by HR-ICP-MS, *Geochemistry, Geophys. Geosystems*, 15, 1617–1628, 2014a.
- 837 Misra, S., Owen, R., Kerr, J., Greaves, M. and Elderfield, H.: Determination of  $\delta^{11}\text{B}$  by HR-ICP-MS from mass  
838 limited samples: Application to natural carbonates and water samples, *Geochim. Cosmochim. Acta*, 140,  
839 531–552, 2014b.

- 840 Mortyn, P. G. and Charles, C. D.: Planktonic foraminiferal depth habitat and  $\delta^{18}\text{O}$  calibrations: Plankton tow  
841 results from the Atlantic sector of the Southern Ocean, *Paleoceanography*, 18, 2003.
- 842 Mulitza, S., Boltovskoy, D., Donner, B., Meggers, H., Paul, A. and Wefer, G.: Temperature: $\delta^{18}\text{O}$  relationships  
843 of planktonic foraminifera collected from surface waters, *Palaeogeogr. Palaeoclimatol. Palaeoecol.*, 202,  
844 143–152, 2003.
- 845 Ni, Y., Foster, G. L., Bailey, T., Elliott, T., Schmidt, D. N., Pearson, P., Haley, B. and Coath, C.: A core top  
846 assessment of proxies for the ocean carbonate system in surface-dwelling foraminifers, *Paleoceanography*  
847 22, 2007.
- 848 Nir, O., Vengosh, A., Harkness, J. S., Dwyer, G. S. and Lahav, O.: Direct measurement of the boron isotope  
849 fractionation factor: Reducing the uncertainty in reconstructing ocean paleo-pH, *Earth Planet. Sci. Lett.*,  
850 414, 1–5, 2015.
- 851 Noireaux, J., Mavromatis, V., Gaillardet, J., Schott, J., Montouillout, V., Louvat, P., Rollion-Bard, C. and  
852 Neuville, D. R.: Crystallographic control on the boron isotope paleo-pH proxy, *Earth Planet. Sci. Lett.*,  
853 430, 398–407, 2015.
- 854 Orr, J. C., Fabry, V. J., Aumont, O., Bopp, L., Doney, S. C., Feely, R. A., Gnanadesikan, A., Gruber, N., Ishida,  
855 A., Joos, F., Key, R. M., Lindsay, K., Maier-Reimer, E., Matear, R., Monfray, P., Mouchet, A., Najjar, R.  
856 G., Plattner, G. K., Rodgers, K. B., Sabine, C. L., Sarmiento, J. L., Schlitzer, R., Slater, R. D., Totterdell, I.  
857 J., Weirig, M. F., Yamanaka, Y. and Yool, A.: Anthropogenic ocean acidification over the twenty-first  
858 century and its impact on calcifying organisms, *Nature*, 437, 681–686, 2005.
- 859 Pagani, M.: Marked Decline in Atmospheric Carbon Dioxide Concentrations During the Paleogene, *Science*,  
860 309, 600–603, 2005.
- 861 Palmer, M. R., Pearson, P. N. and Cobb, S. J., Reconstructing Past Ocean pH-Depth Profiles, *Science*, 282,  
862 1468–1471, 1998.
- 863 Pearson, P. N. and Palmer, M. R.: Middle Eocene seawater pH and atmospheric carbon dioxide concentrations,  
864 *Science*, 284, 1824–1826, 1999.
- 865 Peeters, F. J. C. and Brummer, G.-J. a.: The seasonal and vertical distribution of living planktic foraminifera in  
866 the NW Arabian Sea, *Geol. Soc. London, Spec. Publ.*, 195, 463–497, 2002.
- 867 Quintana Krupinski, N. B., Russell, A. D., Pak, D. K. and Paytan, A.: Core-top calibration of B/Ca in Pacific  
868 Ocean *Neoglobobulimina incompacta* and *Globigerina bulloides* as a surface water carbonate system proxy,  
869 *Earth Planet. Sci. Lett.*, 466, 139–151, 2017.
- 870 Rae, J.W.B.: Boron Isotopes in Foraminifera: Systematics, Biomineralisation, and CO<sub>2</sub> Reconstruction. In:  
871 Marschall, H., Foster, G. (eds), *Boron Isotopes. Advances in Isotope Geochemistry*. Springer, Cham, 2018.
- 872 Rae, J. W. B., Foster, G. L., Schmidt, D. N. and Elliott, T.: Boron isotopes and B/Ca in benthic foraminifera:  
873 Proxies for the deep ocean carbonate system, *Earth Planet. Sci. Lett.*, 302, 403–413, 2011.
- 874 Raitzsch, M., Bijma, J., Benthien, A., Richter, K.-U., Steinhöfel, G. and Kučera, M.: Boron isotope-based  
875 seasonal paleo-pH reconstruction for the Southeast Atlantic – A multispecies approach using habitat  
876 preference of planktonic foraminifera, *Earth Planet. Sci. Lett.*, 487, 138–150, 2018.
- 877 Ravelo, A. C. and Fairbanks, R. G.: Oxygen isotopic composition of multiple species of planktonic foraminifera:  
878 recorder of the modern photic zone temperature gradient, *Palaeogeogr. Palaeoclimatol. Palaeoecol.*, 7,  
879 815–831, 1992.

- 880 Regenberg, M., Nürnberg, D., Steph, S., Groeneveld, J., Garbe-Schönberg, D., Tiedemann, R. and Dullo, W.-C.:  
 881 Assessing the effect of dissolution on planktonic foraminiferal Mg/Ca ratios: Evidence from Caribbean  
 882 core tops, *Geochemistry, Geophys. Geosystems*, 7, 2006.
- 883 Regenberg, M., Steph, S., Nürnberg, D., Tiedemann, R. and Garbe-Schönberg, D.: Calibrating Mg/Ca ratios of  
 884 multiple planktonic foraminiferal species with  $\delta^{18}\text{O}$ -calcification temperatures: Paleothermometry for the  
 885 upper water column, *Earth Planet. Sci. Lett.*, 278, 324–336, 2009.
- 886 Rickaby, R. E. M. and Halloran, P.: Cool La Nina During the Warmth of the Pliocene?, *Science*, 307, 1948–  
 887 1952, 2005.
- 888 Ries, J. B., Cohen, A. L. and McCorkle, D. C.: Marine calcifiers exhibit mixed responses to CO<sub>2</sub>-induced ocean  
 889 acidification, *Geology*, 37, 1131–1134, 2009.
- 890 Rink, S., Kühl, M., Bijma, J. and Spero, H. J.: Microsensor studies of photosynthesis and respiration in the  
 891 symbiotic foraminifer *Orbulina universa*, *Mar. Biol.*, 131, 583–595, 1998.
- 892 Rollion-Bard, C. and Erez, J.: Intra-shell boron isotope ratios in the symbiont-bearing benthic foraminiferan  
 893 *Amphistegina lobifera*: Implications for  $\delta^{11}\text{B}$  vital effects and paleo-pH reconstructions, *Geochim.*  
 894 *Cosmochim. Acta*, 74, 1530–1536, 2010.
- 895 Rostek, F., Ruhland, G., Bassinot, F. C., Muller, P. J., Labeyrie, L. D., Lancelot, Y. and Bard, E.: Reconstructing  
 896 Sea-Surface Temperature and Salinity Using  $\delta^{18}\text{O}$  and Alkenone Records, *Nature*, 364, 319–321, 1993.
- 897 Russell, A. D., Hönisch, B., Spero, H. J. and Lea, D. W.: Effects of seawater carbonate ion concentration and  
 898 temperature on shell U, Mg, and Sr in cultured planktonic foraminifera, *Geochim. Cosmochim. Acta*, 68,  
 899 4347–4361, 2004.
- 900 Sanyal, A., Bijma, J., Spero, H. J. and Lea, D. W.: Empirical relationship between pH and the boron isotopic  
 901 composition of *Globigerinoides sacculifer*: Implications for the boron isotopes paleo-pH proxy,  
 902 *Paleoceanography*, 16, 515–519, 2001.
- 903 Sanyal, A., Hemming, N. G., Broecker, W. S., Lea, D. W., Spero, H. J., & Hanson, G. N. Oceanic pH control on  
 904 the boron isotopic composition of foraminifera: evidence from culture experiments, *Paleoceanography*,  
 905 11(5), 513-517, 1996.
- 906 Schmidt, G. A. and Mulitza, S.: Global calibration of ecological models for planktic foraminifera from core-top  
 907 carbonate oxygen-18, *Mar. Micropaleontol.*, 44, 125–140, 2002.
- 908 Seki, O., Foster, G. L., Schmidt, D. N., Mackensen, A., Kawamura, K. and Pancost, R. D.: Alkenone and boron-  
 909 based Pliocene pCO<sub>2</sub> records, *Earth Planet. Sci. Lett.*, 292, 201–211, 2010.
- 910 Shirayama, Y.: Effect of increased atmospheric CO<sub>2</sub> on shallow water marine benthos, *J. Geophys. Res.*, 110,  
 911 C09S08, 2005.
- 912 Sime, N. G., De La Rocha, C. L. and Galy, A.: Negligible temperature dependence of calcium isotope  
 913 fractionation in 12 species of planktonic foraminifera, *Earth Planet. Sci. Lett.*, 232, 51–66, 2005.
- 914 Spero H. J.: Symbiosis in the planktonic foraminifer, *Orbulina universa*, and the isolation of its symbiotic  
 915 dinoflagellate, *gymnodinium beii* sp.nov, *J. Phycol* 23, 307-317, 1987.
- 916 Sutton, J. N., Liu, Y. W., Ries, J. B., Guillermic, M., Ponzevera, E. and Eagle, R. A.:  $\delta^{11}\text{B}$  as monitor of  
 917 calcification site pH in divergent marine calcifying organisms, *Biogeosciences*, 15, 1447–1467, 2018.
- 918 Takagi H., Kimoto K., Fujiki T., Saito H., Schmidt C. and Kucera M.: Characterizing photosymbiosis in modern  
 919 planktonic foraminifera, *biogeosciences*, 3377–3396, 2019.

- 920 Thomson, J., Brown, L., Nixon, S., Cook, G. T. and MacKenzie, A. B.: Bioturbation and Holocene sediment  
921 accumulation fluxes in the north-east Atlantic Ocean (Benthic Boundary Layer experiment sites), *Mar.*  
922 *Geol.*, 169, 21–39, 2000.
- 923 Tripathi, A.: Deep-Sea Temperature and Circulation Changes at the Paleocene-Eocene Thermal Maximum.  
924 *Science*, 308, 1894–1898, 2005.
- 925 Tripathi, A. K., Roberts, C. D. and Eagle, R. A.: Coupling of CO<sub>2</sub> and Ice Sheet Stability Over Major Climate  
926 Transitions of the Last 20 Million Years, *Science*, 326, 1394–1397, 2009.
- 927 Tripathi, A. K., Roberts, C. D., Eagle, R. A. and Li, G.: A 20 million year record of planktic foraminiferal B/Ca  
928 ratios: Systematics and uncertainties in pCO<sub>2</sub> reconstructions, *Geochim. Cosmochim. Acta*, 75, 2582–  
929 2610, 2011.
- 930 Uchikawa, J., Penman, D. E., Zachos, J. C. and Zeebe, R. E.: Experimental evidence for kinetic effects on B/Ca  
931 in synthetic calcite: Implications for potential B(OH)<sub>4</sub><sup>-</sup> and B(OH)<sub>3</sub> incorporation, *Geochim. Cosmochim.*  
932 *Acta*, 150, 171–191, 2015.
- 933 Urey, H.C., Lowenstam, H.A., Epstein, S. & McKinney, C.R.: Measurement of paleo-temperature and  
934 temperatures of the upper cretaceous of England, Denmark, and the southeastern United-States. *Geol. Soc.*  
935 *Am. Bull.*, 62, 399-416, 1951.
- 936 Wang, B.-S., You, C.-F., Huang, K.-F., Wu, S.-F., Aggarwal, S. K., Chung, C.-H. and Lin, P.-Y.: Direct  
937 separation of boron from Na- and Ca-rich matrices by sublimation for stable isotope measurement by MC-  
938 ICP-MS, *Talanta*, 82, 1378–1384, 2010.
- 939 Wang, G., Cao, W., Yang, D. and Xu, D.: Variation in downwelling diffuse attenuation coefficient in the  
940 northern South China Sea, *Chinese J. Oceanol. Limnol.*, 26, 323–333, 2008.
- 941 Weare, B. C., Strub, P. T. and Samuel, M. D.: Annual Mean Surface Heat Fluxes in the Tropical Pacific Ocean, *J.*  
942 *Phys. Oceanogr.*, 11, 705–717, 1981.
- 943 Wei, G., McCulloch, M. T., Mortimer, G., Deng, W. and Xie, L.: Evidence for ocean acidification in the Great  
944 Barrier Reef of Australia, *Geochim. Cosmochim. Acta*, 73, 2332–2346, 2009.
- 945 Wilson, D. J., Piotrowski, A. M., Galy, A. and McCave, I. N.: A boundary exchange influence on deglacial  
946 neodymium isotope records from the deep western Indian Ocean, *Earth Planet. Sci. Lett.*, 341–344, 35–47,  
947 2012.
- 948 Wolf-Gladrow, D. A., Riebesell, U., Burkhardt, S. and Buma, J.: Direct effects of CO<sub>2</sub> concentration on growth  
949 and isotopic composition of marine plankton, *Tellus B Chem. Phys. Meteorol.*, 51, 461–476, 1999.
- 950 Yu, J., Menviel, L., Jin, Z. D., Thornalley, D. J. R., Barker, S., Marino, G., Rohling, E. J., Cai, Y., Zhang, F.,  
951 Wang, X., Dai, Y., Chen, P. and Broecker, W. S.: Sequestration of carbon in the deep Atlantic during the  
952 last glaciation, *Nat. Geosci.*, 9, 319–324, 2016.
- 953 Yu, J., Thornalley, D. J. R., Rae, J. W. B. and McCave, N. I.: Calibration and application of B/Ca, Cd/Ca, and δ  
954 <sup>11</sup>B in *Neogloboquadrina pachyderma* (sinistral) to constrain CO<sub>2</sub> uptake in the subpolar North Atlantic  
955 during the last deglaciation, *Paleoceanography*, 28, 237–252, 2013.
- 956 Yu, J., Foster, G. L., Elderfield, H., Broecker, W. S. and Clark, E.: An evaluation of benthic foraminiferal B/Ca  
957 and δ<sup>11</sup>B for deep ocean carbonate ion and pH reconstructions, *Earth Planet. Sci. Lett.*, 293, 114–120, 20,  
958 2010.
- 959 Yu, J., Elderfield, H., Hönisch, B.: B/Ca in planktonic foraminifera as a proxy for surface seawater pH.  
960 *Paleoceanography*22, PA2202, 2007.

- 961 Yu, J., Day, J., Greaves, M. and Elderfield, H., Determination of multiple element/calcium ratios in foraminiferal  
962 calcite by quadrupole ICP-MS, *Geochemistry, Geophys. Geosystems* 6, 2005.
- 963 Zeebe, R. E. and Wolf-Gladrow, D., *CO<sub>2</sub> in Seawater: Equilibrium, Kinetics, Isotopes* Elsevier Oceanography  
964 Series 65, Amsterdam, 2001.
- 965 Zeebe, R. E., Wolf-Gladrow, D. A., Bijma, J. and Hönisch, B., Vital effects in foraminifera do not compromise  
966 the use of  $\delta^{11}\text{B}$  as a paleo- pH indicator: Evidence from modeling, *Paleoceanography*, 18, 2003.

967 **Figure caption**

968

969 **Figure 1:** Reactions governing dissolved inorganic carbon equilibria.

970

971 **Figure 2:** (A) Speciation of  $H_3BO_3$  and  $H_4BO_4^-$  as function of seawater pH (total scale), (B)  $\delta^{11}B$  of dissolved  
972 inorganic boron species as a function of seawater pH, (C) sensitivity of  $\delta^{11}B$  of  $H_4BO_4^-$  for a pH ranging from 7.6  
973 to 8.4.  $T=25^\circ C$ ,  $S=35$ ,  $\delta^{11}B=39.61 \text{ ‰}$  (Foster et al., 2010), dissociation constant  $\alpha = 1.0272$  (Klochko et al., 2006).

974

975 **Figure 3:** Map showing locations of the core-tops used in this study (white diamonds). Red open circles represent  
976 the sites used for *in-situ* carbonate parameters from GLODAP database (Key et al., 2004).

977

978 **Figure 4:** Pre-industrial data versus depth of the sites used in this study. The figure shows seasonal temperatures  
979 (extracted from World Ocean Database 2013), density anomaly ( $kg/m^3$ ), pre-industrial pH and pre-industrial  $\delta^{11}B$   
980 of  $H_4BO_4^-$  (calculated from the GLODAP database and corrected for anthropogenic inputs).

981

982 **Figure 5:** Boron isotopic measurements of mixed-layer foraminifera plotted against the  $\delta^{11}B_{borate}$ .  $\delta^{11}B_{borate}$  were  
983 characterized by determination of the calcification depth of the foraminifera, A) *G. ruber*, B) *T. sacculifer*, C) *O.*  
984 *universa*. Mono-specific calibrations are summarized in Table 3.

985

986 **Figure 6:** Boron isotopic measurements of deep-dwelling foraminifera ( $\delta^{11}B_{carbonate}$ ) plot against  $\delta^{11}B_{borate}$ .  $\delta^{11}B_{borate}$   
987 were characterized by determining the calcification depth of foraminifera, A) *P. obliquiloculata*, B) *G. menardii*,  
988 C) *N. dutertrei*, D) *G. tumida* and E) Compilation of deep dweller species. Mono-specific calibrations are  
989 summarized in Table 3.

990

991 **Figure 7:** Boxplots of B/Ca ratios for multiple species, *T. sacculifer* (this study; Foster et al., 2008; Ni et al; 2007;  
992 Seki et al., 2010), *G. ruber* (this study; Babila et al., 2014; Foster et al., 2008; Ni et al., 2007), *G. inflata*, *G.*  
993 *bulloides* (Yu et al., 2007), *N. pachyderma* (Hendry et al., 2009; Yu et al., 2013), *N. dutertrei* (this study; Foster  
994 et al., 2008), *O. universa*, *P.obliquiloculata*, *G. menardii*, *G. tumida* (this study).

995

996 **Figure 8:** A) Boxplot showing the calculated microenvironment pH difference ( $\Delta$ microenvironment pH) between  
997 microenvironment and external pH based on the  $\delta^{11}B$  data. B) This figure shows that a decrease in insolation can  
998 explain the low  $\delta^{11}B$  from the WEP. Light penetration profile in the Western Pacific, with  $E_0$  in the WEP of 220  
999  $J.s^{-1}m^{-2}$  (Weare et al., 1981) and a light attenuation coefficient of 0.028 (Wang et al., 2008). Theoretical depths  
1000 were calculated for a decrease in microenvironment pH of  $\Delta pH_1 = -0.02$  (e.g. WP07-a);  $\Delta pH_1 = -0.04$  (e.g. A14),  
1001  $\Delta pH_2 = -0.06$  (e.g. 806A). Light penetration corresponding to  $E_c$  is  $\sim 12\%$ ,  $\Delta pH_0 \sim 7\%$ ,  $\Delta pH_1 \sim 5\%$ ,  $\Delta pH_2 \sim 1\%$   
1002 respective depth are 75m, 90m, 110m and 150m. Grey band is the calcification depth of *T. sacculifer* (w/o sacc)  
1003 utilized in this study.

1004

1005 **Figure 9:** Water depth pH profiles reconstructed at every site applying the mono-specific calibrations derived from  
1006 our results (Table 3). Figure is showing measured  $\delta^{11}\text{B}_{\text{calcite}}$ ,  $\delta^{11}\text{B}_{\text{borate}}$  calculated according to different calibrations  
1007 (see Table 3 and text), calculated pH based on  $\delta^{11}\text{B}$  ( $\text{pH}_{\delta^{11}\text{B}}$ ) and  $\text{pCO}_2$  calculated from  $\text{pH}_{\delta^{11}\text{B}}$  and alkalinity.  
1008  
1009 **Figure 10:** Evaluation of the reconstructed parameters,  $\delta^{11}\text{B}_{\text{borate}}$ , pH and  $\text{pCO}_2$  versus *in-situ* parameter. The  
1010 recalculated parameters are consistent with *in-situ* data, except for *G. ruber*. This variability might be explained  
1011 by the different test sizes.

1012 **Table caption**

1013

1014 **Table 1:** Box-core information

1015

1016 **Table 2:** Analytical results of  $\delta^{13}\text{C}$ ,  $\delta^{18}\text{O}$ ,  $\delta^{11}\text{B}$  and elemental ratios Li/Ca, B/Ca and Mg/Ca

1017

1018 **Table 3:** Species-specific  $\delta^{11}\text{B}_{\text{carbonate}}$  to  $\delta^{11}\text{B}_{\text{borate}}$  calibrations from literature and from our data