

1 **Drivers of diffusive CH₄ emissions from shallow subarctic lakes on daily to multi-**
2 **year time scales**

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14 **Abstract**

15 Lakes and reservoirs contribute to regional carbon budgets via significant emissions of climate forcing trace
16 gases. Here, for improved modelling, we use 8 years of floating chamber measurements from three small,
17 shallow subarctic lakes (2010–2017, $n = 1306$) to separate the contribution of physical and biogeochemical
18 processes to the turbulence-driven, diffusion-limited flux of methane (CH_4) on daily to multi-year
19 timescales. Correlative data include 9 years of surface water concentration measurements (2009–2017, n
20 $= 606$) and in situ meteorological observations. We used the latter to compute near surface turbulence
21 based on similarity scaling and then applied the surface renewal model to compute gas transfer velocities.
22 Chamber fluxes averaged $6.9 \pm 0.3 \text{ mg m}^{-2} \text{ d}^{-1}$ and gas transfer velocities (k_{600}) averaged $4.0 \pm 0.1 \text{ cm h}^{-1}$.
23 Spectral analysis indicated that on timescales shorter than a month, emissions were driven by wind shear
24 whereas on longer timescales variations in water temperature governed the flux. Chamber derived gas
25 transfer velocities tracked the power-law wind speed relation of the model. Coefficients for the model and
26 dissipation rates depended on shear production of turbulence, atmospheric stability, and exposure to
27 wind. Fluxes increased with wind speed until daily average values exceeded 6.5 m s^{-1} , at which point
28 emissions were suppressed due to rapid water column degassing reducing the water–air concentration
29 gradient. Arrhenius-type temperature functions of the CH_4 flux ($E_a' = 0.90 \pm 0.14 \text{ eV}$) were robust ($R^2 \geq$
30 0.93 , $p < 0.01$) and also applied to the surface CH_4 concentration ($E_a' = 0.88 \pm 0.09 \text{ eV}$). These results
31 indicate that emissions were strongly coupled to production and supply to the water column. Our findings
32 show that accurate short- and long-term projections of lake CH_4 emissions can be based on distinct
33 weather- and climate controlled drivers.

34 1. Introduction

35 Inland waters are an important source of the radiatively active trace gas methane (CH₄) to the atmosphere
36 (Bastviken et al., 2011; Cole et al., 2007). On regional to global scales, an estimated 21–46% of ice-free
37 season CH₄ emissions from lakes, ponds and reservoirs occur via turbulence-driven diffusion-limited gas
38 exchange (Bastviken et al., 2011; DelSontro et al., 2018; Wik et al., 2016b) (hereafter abbreviated to
39 ‘diffusive fluxes’). Diffusive fluxes are often measured with floating chambers (Bastviken et al., 2004) but
40 gas transfer models are increasingly used, for example in regional emission budgets (Holgerson and
41 Raymond, 2016; Weyhenmeyer et al., 2015). Fluxes computed with modelled gas transfer velocities agree
42 to a certain extent with floating chambers and the eddy covariance technique in short-term
43 intercomparison campaigns (Bartosiewicz et al., 2015; Crill et al., 1988; Erkkilä et al., 2018). However, long-
44 term comparisons are needed to identify weather- and climate related controls on the flux that are
45 appropriate for seasonal assessments. Considering the increased use of process-based approaches in
46 regional emission estimates (Tan and Zhuang, 2015), understanding the mechanisms that drive the
47 components of the diffusive flux is imperative for improving emission estimates.

48

49 1.1 Drivers of diffusive CH₄ emissions

50 Diffusive fluxes at the air-water interface are estimated with a two-layer model (Liss and Slater, 1974):

$$F = k(C_{aq} - C_{air,eq}) \quad [1]$$

51 The flux F [mg m⁻² d⁻¹] depends on the concentration difference across a thin layer immediately below
52 the air-water interface ($\Delta[\text{CH}_4]$ in mg m⁻³), of which the upper boundary is in equilibrium with the
53 atmosphere ($C_{air,eq}$) and the base represents the bulk liquid (C_{aq}), and is limited by the gas transfer
54 velocity k [m d⁻¹]. k has been conceptualized as characterizing transfer across the diffusive boundary layer.
55 Other models envision exchange as driven by parcels of water intermittently in contact with the
56 atmosphere. In these surface renewal models, k depends on the frequency of the renewal events
57 (Csanady, 2001; Lamont and Scott, 1970). The resulting calculation for k is based on the Kolmogorov
58 velocity scale, $u_n = (\varepsilon\nu)^{1/4}$ where ε is dissipation rate of turbulent kinetic energy (TKE) and ν is kinematic
59 viscosity (Tennekes and Lumley, 1972). Progress has been made in understanding how to compute ε and
60 gas transfer rates as a function of wind speed and the heating and cooling at the lake’s surface (Tedford
61 et al., 2014). Comparisons between models and other flux estimation methods, such as eddy covariance,
62 illustrate the improved accuracy when computing gas transfer velocities using a turbulence-based as
63 opposed to wind based models (Czikowsky et al., 2018; Heiskanen et al., 2014; Mammarella et al., 2015).

64

65 A key control on emissions is the periodicity at which dissolved gases are brought to the air-water
66 interface. During stratification, the density gradient makes it difficult for wind driven mixing to bring gases
67 to the surface, and they may accumulate in the stratified regions. Conversely, thermal convection
68 associated with surface cooling can deepen the mixed layer and transfer stored gas to the surface (Crill et
69 al. 1988; Eugster et al. 2003). Nighttime emissions can be enhanced when the surface cools despite low
70 wind speeds (Podgrajsek et al., 2015; Poindexter et al., 2016). Temporal patterns of stratification and
71 mixing contribute to variability in diffusive CH₄ fluxes (López Bellido et al., 2009; Podgrajsek et al., 2016)
72 and concentrations (Loken et al., 2019; Natchimuthu et al., 2016). Periodic emissions from storage at depth

73 have been particularly difficult to resolve in lake emission budgets (Bastviken et al., 2004; Wik et al.,
74 2016b).

75
76 CH₄ emissions to the atmosphere also depend on the rates of methane metabolism regulated by substrate
77 availability and temperature-dependent shifts in enzyme activity and microbial community structure
78 (Borrel et al., 2011; McCalley et al., 2014; Tveit et al., 2015). Arrhenius-type relationships of CH₄ fluxes
79 have emerged from field studies (DelSontro et al., 2018; Natchimuthu et al., 2016; Wik et al., 2014) and
80 across latitudes and aquatic ecosystem types in synthesis reports (Rasilo et al., 2015; Yvon-Durocher et al.,
81 2014). However, the temperature sensitivity is modulated by biogeochemical factors that differ between
82 lake ecosystems, such as nutrient content (Davidson et al., 2018; Sepulveda-Jauregui et al., 2015),
83 methanotrophic activity (Duc et al., 2010; Lofton et al., 2014), predominant emission pathway (DelSontro
84 et al., 2016; Jansen et al., 2019) and warming history (Yvon-Durocher et al., 2017). In lakes, the air-water
85 concentration difference driving the flux (Eq. 1) is further affected by factors that dissociate production
86 from emission rates. These include biotic factors, such as aerobic and anaerobic methanotrophy, and
87 abiotic factors such as hydrologic inputs of terrestrially produced CH₄ (Miettinen et al., 2015; Paytan et al.,
88 2015) and storage-and-release cycles associated with transient stratification (Czikowsky et al., 2018;
89 Jammet et al., 2017; Vachon et al., 2019). Given these interacting functional dependencies, the magnitude
90 of fluxes has complex patterns of temporal variability.

91
92 Disentangling the physical and biogeochemical drivers of the diffusive CH₄ flux remains a challenge. The
93 component drivers respond differently to slow and fast changes in meteorological covariates (Baldocchi
94 et al., 2001; Koebisch et al., 2015) such that different mechanisms may explain the diel and seasonal
95 variability of the flux. For example, temperature affects emissions through convective mixing on short
96 timescales and through the rate of sediment methanogenesis on longer timescales; the diurnal cycle of
97 insolation may have a limited effect on production because the heat capacity of the water buffers the
98 temperature signal (Fang and Stefan, 1996). Similar phase lags and amplifications may lead to hysteretic
99 flux patterns, such as cold season emission peaks due to release of gases from the hypolimnion in dimictic
100 lakes (Encinas Fernández et al., 2014; López Bellido et al., 2009) or thermal inertia of lake sediments (Zimov
101 et al., 1997). Spectral analysis of the flux and its components can improve our understanding of the flux
102 variability by quantifying how much power is associated with key periodicities (Baldocchi et al., 2001).

103
104 Here we present a high-resolution, long-term dataset (2010–2017) of diffusive CH₄ fluxes from three
105 subarctic lakes estimated with floating chambers ($n = 1306$), and fluxes obtained by modelling using in situ
106 meteorological observations and surface water concentrations ($n = 535$). The surface renewal model is
107 used to compute gas transfer velocities. Arrhenius relationships of $\Delta[\text{CH}_4]$ and fluxes of CH₄ are also
108 calculated. Using spectral analysis of our time series data, we distinguish the temporal dependency of
109 abiotic and biotic controls on the flux. The effects of lake size and wind exposure are illustrated by
110 comparing results from the 3 different lakes.

111 **2. Materials and Methods**

112 **2.1 Field site**

113 CH₄ emissions were measured from three subarctic lakes of post-glacial origin (Kokfelt et al., 2010), located
114 around the Stordalen Mire in northern Sweden (68°21' N, 19°02' E, Fig. 1), a palsa mire complex underlain
115 by discontinuous permafrost (Malmer et al., 2005). The Mire (350 m a.s.l.) is part of a catchment that
116 connects Mt. Vuoskoåiveh (920 m a.s.l.) in the south to Lake Torneträsk (341 m a.s.l.) in the north (Lundin
117 et al., 2016; Olefeldt and Roulet, 2012). Villasjön is the largest and shallowest of the lakes (0.17 km², 1.3
118 m max. depth) and drains through fens into a stream feeding Mellersta Harrsjön and Inre Harrsjön, which
119 are 0.011 and 0.022 km² in size and have maximum depths of 6.7 m and 5.2 m, respectively (Wik et al.,
120 2011). The lakes are normally ice-free from the beginning of May through the end of October. Manual
121 observations were generally conducted between mid-June and the end of September. Diffusion accounts
122 for 17%, 52% and 34% of the ice-free CH₄ flux in Villasjön, Inre and Mellersta Harrsjön, respectively, with
123 the remainder emitted via ebullition (2010–2017; Jansen et al., 2019).

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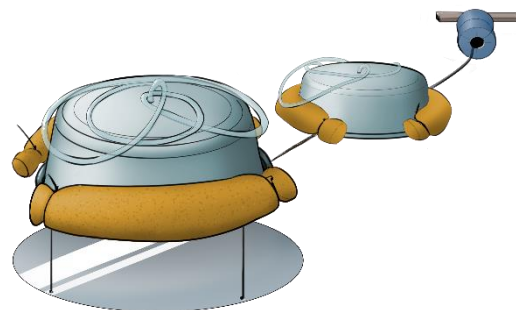
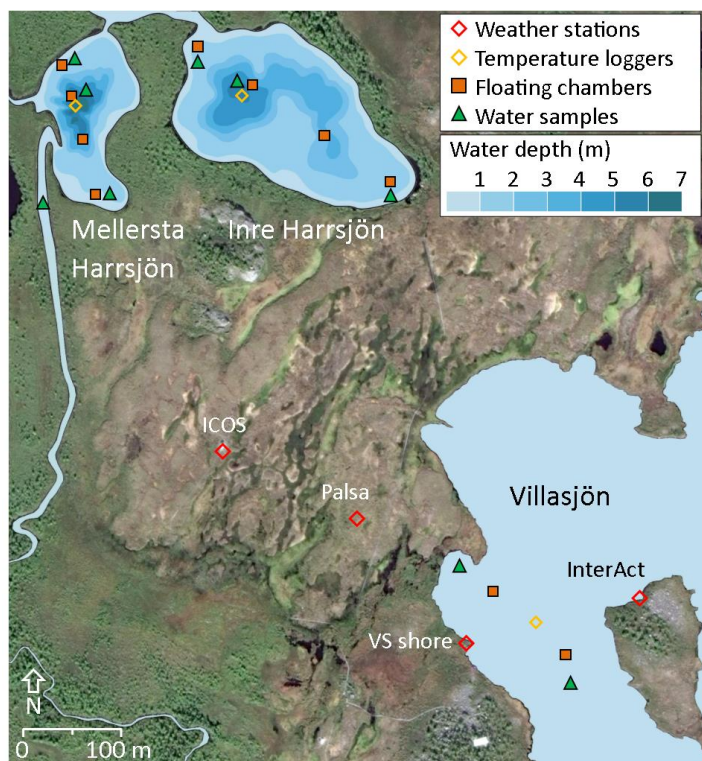


Figure 1 – Map of the Stordalen Mire field site (left). Chamber and sampling locations are shown as they were in 2015–2017. A schematic of the floating chamber pairs is shown to the right. Lake bathymetry from Wik et al. (2011). Satellite imagery: Google, DigitalGlobe, 2017.

125

126 **2.2 Floating chambers**

127 We used floating chambers to directly measure the turbulence-driven diffusive CH₄ flux across the air-
128 water interface (Fig. 1). They consisted of plastic tubs covered with aluminium tape to reflect incoming
129 radiation and were equipped with polyurethane floats and flexible sampling tubes capped at one end with
130 3-way stopcocks (Bastviken et al., 2004). Depending on flotation depth, each chamber covered an area
131 between 610 and 660 cm² and contained a headspace of 4 to 5 litres. Chambers were deployed in pairs
132 with a plastic shield mounted 30 cm below one chamber of each pair to deflect methane bubbles rising
133 from the sediment. Every 1–2 weeks during the ice-free seasons of 2010 to 2017, 2–4 chamber pairs were
134 deployed in Villasjön and 4–7 chamber pairs in Inre and Mellersta Harrsjön in different depth zones (Fig.
135 1). The number of chambers and deployment intervals exceeded the minimum needed to resolve the
136 spatiotemporal variability of the flux (Wik et al., 2016a). Over a 24 hour period, 2–4 60 mL headspace
137 samples were collected from each chamber using polypropylene syringes and the flotation depth and air
138 temperature were noted in order to calculate the headspace volume. The 24-hour deployment period
139 integrates diel variations in the gas transfer velocity (Bastviken et al., 2004).

140
141 The fluxes reported here are from the shielded chambers only. To check that the shields were not reducing
142 fluxes from turbulent processes such as convection, we compared fluxes from shielded and unshielded
143 chambers on days when the lake mean bubble flux was <1% of the lake mean diffusive flux (bubble traps,
144 2009–2017; Jansen et al., 2019; Wik et al., 2013). Averaged over the three lakes, the difference was
145 statistically significant ($0.20 \pm 0.16 \text{ mg m}^{-2} \text{ d}^{-1}$ ($n = 58$) (mean \pm 95% CI)), but small in relative terms (6% of
146 the mean flux). Conversely, some types of floating chambers can enhance gas transfer by creating artificial
147 turbulence when dragging through the water (Matthews et al., 2003; Vachon et al., 2010; Wang et al.,
148 2015). Ribas-Ribas et al. (2018) compared acoustic Doppler velocimeter measurements inside and outside
149 the perimeter of a chamber of similar design, size and flotation depth as those used in this study, and,
150 based on a comparison of measured TKE dissipation rates and computed gas transfer velocities, concluded
151 that the chambers did not cause artificial turbulence.

152 153 **2.3 Water samples**

154 Surface water samples were collected 0.2–0.4 m below the surface at 2–3 different locations in each lake,
155 at one to two-week intervals from June to October (Fig. 1). Samples were collected from shore with a 4 m
156 Tygon tube attached to a float to avoid disturbing the sediments (2009–2014), and from a rowboat over
157 the deepest points of Inre and Mellersta Harrsjön (2010–2017) and at shallows (<1 m water depth) on
158 either end of the lakes (2015–2017) using a 1.2 m L x 3.2 mm ID Tygon tube. In addition, water samples
159 were collected at the deepest point of Inre and Mellersta Harrsjön at 1 m intervals down to 0.1 m from
160 the sediment surface with a 7.5 m L x 6.4 mm ID fluorinated ethylene propylene (FEP) tube. Subsequently,
161 60 mL polypropylene syringes were rinsed thrice with sample water before duplicate bubble-free samples
162 were collected, and were capped with airtight 3-way stopcocks. 30 mL samples were equilibrated with 30
163 mL headspace and shaken vigorously by hand for 2 minutes (2009–2014) or on a mechanical shaker at 300
164 rpm for 10 minutes (2015–2017). Prior to 2015, outside air – with a measured CH₄ content – was used as
165 headspace. From 2015 on we used an N₂ 5.0 headspace (Air Liquide). Water sample conductivity was
166 measured over the ice-free season of 2017 ($n = 323$) (S230, Mettler-Toledo), and converted to specific
167 conductance using a temperature-based approach.

168 **2.4 Concentration measurements**

169 Gas samples were analysed within 24 hours after collection at the Abisko Scientific Research Station, 10
170 km from the Stordalen Mire. Sample CH₄ contents were measured on a Shimadzu GC-2014 gas
171 chromatograph which was equipped with a flame ionization detector (GC-FID) and a 2.0 m long, 3 mm ID
172 stainless steel column packed with 80/100 mesh HayeSep Q and used N₂ >5.0 as a carrier gas (Air Liquide).
173 For calibration we used standards of 2.059 ppm CH₄ in N₂ (Air Liquide). 10 standard measurements were
174 made before and after each run. After removing the highest and lowest values, relative standard
175 deviations of the standard runs were generally less than 0.25%.

176

177 **2.5 Water temperature, pressure, density and mixed layer depth**

178 Water temperature was measured every 15 minutes from 2009 to 2018 with temperature loggers (HOBO
179 Water Temp Pro v2, Onset Computer) in Villasjön and at the deepest locations within Inre and Mellersta
180 Harrsjön. Sensors were deployed at 0.1, 0.3, 0.5, 1.0 m depth in all lakes, with additional sensors at 3.0,
181 5.0 m (IH and MH) and at 6.7 m (MH). Sensors were intercalibrated prior to deployment in a well-mixed
182 water tank, and by comparing readouts just before and during ice-on when the water column was
183 isothermal. In this way a precision of <0.05 °C was achieved. The bottom sensors were buried in the surface
184 sediment and were excluded from in situ intercalibration. Water pressure was measured in Mellersta
185 Harrsjön (5.5 m) with a HOBO U20 Water Level logger (Onset Computer). Water density was computed
186 from temperature and salinity (Chen and Millero, 1977), using lake-averaged specific conductivity and a
187 salinity factor [mS cm⁻¹ / g kg⁻¹] of 0.57. The salinity factor was based on a linear regression of simultaneous
188 measurements of conductivity and dissolved solids (R² = 0.99, n = 7) in five lakes in the Torneträsk
189 catchment (Miljödata-MVM, 2017). We defined the depth of the surface mixing layer (z_{mix}) at a density
190 gradient threshold (dp/dz) of 0.03 kg m⁻³ m⁻¹ (Rueda et al., 2007).

191

192 **2.6 Meteorology**

193 Meteorological data was collected from four different masts on the Mire, and collectively covered a period
194 from June 2009 to October 2017 with half-hourly measurements of wind speed, air temperature, relative
195 humidity, air pressure and irradiance (Fig. 1, Table 1). Wind speed was measured with 3D sonic
196 anemometers at the Palsa tower (z = 2.0 m), the Villasjön shore tower (z = 2.9 m), at the InterAct Lake
197 tower (z = 2.0 m) and at the Integrated Carbon Observation System (ICOS) site (z = 4.0 m). Air temperature
198 and relative humidity were measured at the Palsa tower, at the Villasjön shore tower (Rotronic MP100a
199 (2012–2015) / Vaisala HMP155 (2015–2017)) and at the InterAct lake tower. Incoming and outgoing
200 shortwave and long wave radiation were monitored with net radiometers at the Palsa tower (Kipp & Zonen
201 CNR1) and at the InterAct lake tower (Kipp & Zonen CNR4). Precipitation data was collected with a
202 WeatherHawk 500 at the ICOS site. Overlapping measurements were cross-validated and averaged to form
203 a single timeseries.

204

205 **Table 1** – Location and instrumentation of meteorological observations on the Stordalen Mire, 2009–2018.

Identifier	Period	Location	Wind	Air temp. and humidity	Radiation	Ref.
Palsa tower	2009-11	68°21'19.68"N 19° 2'52.44"E	C-SAT 3 <i>Campbell Scientific</i>	HMP-45C <i>Campbell Scientific</i>	CNR-1 <i>Kipp & Zonen</i>	Olefelt et al., 2012
Villasjön shore tower	2012-18	68°21'14.58"N 19° 3'1.07"E	R3-50 <i>Gill</i>	MP100a, <i>Rotronic</i> HMP155, <i>Vaisala</i>	REBS Q7.1 <i>Campbell Sci.</i>	Jammet et al., 2015
InterAct Lake tower	2012-18	68°21'16.22"N 19° 3'14.98"E	uSonic 3 Scientific <i>Metek</i>	CS215 <i>Campbell Scientific</i>	CNR-4 <i>Kipp & Zonen</i>	n/a
ICOS site	2013-18	68°21'20.59"N 19° 2'42.08"E	Weatherhawk 500 <i>Campbell Scientific</i>			n/a

206

207 **2.7 Computation of CH₄ storage and residence time**

208 The amount of CH₄ stored in the water column [g CH₄ m⁻²] was computed by weighting and then adding
 209 each concentration measurement by the volume of the 1 m depth interval within which it was collected.
 210 For the upper 2 m of the two deeper lakes we separately computed storage in the vegetated littoral zone
 211 from near-shore concentration measurements, as these values could be different from those further from
 212 shore due to outgassing and oxidation during horizontal transport (DelSontro et al., 2017). We computed
 213 the average residence time of CH₄ in the lake by dividing the amount stored by the lake mean surface flux.
 214 Residence times computed with this approach should be considered upper limits, because in this
 215 calculation we assumed that removal processes other than surface emissions, such as microbial oxidation,
 216 were negligible or took place at the sediment-water interface with minimal effect on water column CH₄.

217

218 **2.8 Flux calculations**

219 In order to calculate the chamber flux with Eq. 1, we estimated the gas transfer velocity, k_{ch} [cm h⁻¹] from
 220 the time-dependent equilibrium chamber headspace concentration $C_{h,eq}(t)$ [mg m⁻³] (Bastviken et al.,
 221 2004):

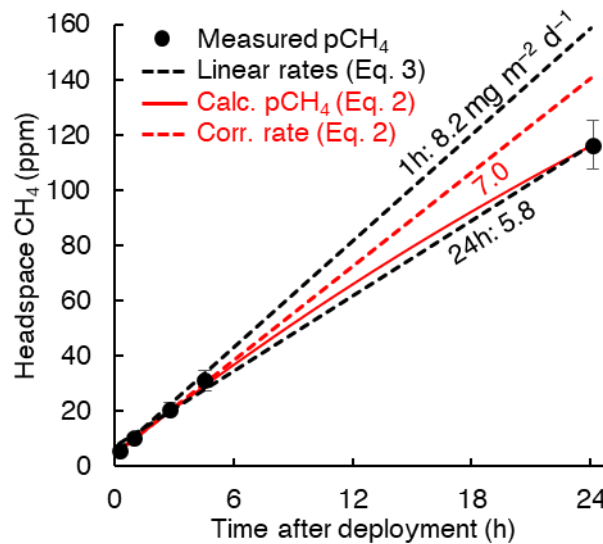
$$(C_{aq} - C_{h,eq}(t)) = (C_{aq} - C_{h,eq}(t_0))e^{-\frac{K_H R T_{water} A}{V} k_{ch} t} \quad [2]$$

222 where K_H is Henry's law constant for CH₄ [mg m⁻³ Pa⁻¹] (Wiesenburg and Guinasso, 1979), R is the universal
 223 gas constant [m³ Pa mg⁻¹ K⁻¹], T_{water} is the surface water temperature [K] and V and A are the chamber
 224 volume [m³] and area [m²], respectively. This method accounts for gas accumulation in the chamber
 225 headspace, which reduces the concentration gradient and limits the flux (Eq. 1) (Fig. 2). For a subset of
 226 chamber measurements where simultaneous water concentration measurements were unavailable ($n =$
 227 949) we computed the flux from the headspace concentrations alone:

$$F = c_1 M \frac{\partial x_h}{\partial t} \frac{PV}{RT_{air} A} \quad [3]$$

228 $\partial x_h / \partial t$ is the headspace CH₄ mole fraction change [mol mol⁻¹ d⁻¹] computed with ordinary least squares
 229 (OLS) linear regression (Fig. 2), M is the molar mass of CH₄ (0.016 mg mol⁻¹), P is the air pressure [Pa], T_{air}
 230 is the air temperature [K]. Scalar c_1 corrects for the accumulation of CH₄ gas in the chamber headspace
 231 and increases over the deployment time. Comparing both chamber flux calculation methods we find $c_1 =$
 232 1.21 for 24 hour deployments (OLS, $R^2 = 0.85$, $n = 357$). Chambers were sampled up to 4 times during their
 233 24 hour deployment (at 10 minutes, 1–5 hours and 24 hours) which allowed us to compute fluxes at time
 234 intervals of 1 hour and 24 hours. P and T_{air} were averaged over the relevant time interval.

235 Figure 2 shows that the headspace correction is necessary to avoid underestimating fluxes. The headspace-
 236 corrected flux (dashed red line) equals the initial slope of Eq. 2 (solid red line) and is about 21% higher
 237 than the non-corrected flux (lower dashed black line in Fig. 2). However, both Eq. 2 (solid red line) and Eq.
 238 3 with $c_1 = 1$ (dashed black lines) fit the concentration data ($R^2 \geq 0.98$ for 94% of 24-hour flux
 239 measurements). This similarity results partly because the fluxes were low enough to keep headspace
 240 concentrations well below equilibrium with the water column. Short-term measurements (upper dashed
 241 black line) may omit the need for headspace correction (Bastviken et al., 2004). Because concentration
 242 measurements were not available for all chamber observations, we used multi-year mean values of $\Delta[\text{CH}_4]$
 243 and k_{ch} to compute c_1 as a function of chamber deployment time. For 24 hour chamber deployments, $c_1 =$
 244 1.21.



245 **Figure 2** – Example of chamber headspace CH_4 concentrations versus deployment time. Measured
 246 concentrations (dots) are averages from 2015–2017 (0.1h) and 2011 (1h–24h); error bars represent the
 247 95% confidence intervals. Linear regressions (dashed black lines) show the rate increase over 1 hour (two
 248 measurements) and over 24 hours (five measurements). The solid red line represents chamber
 249 concentrations computed with Eq. 2. The rate increase associated with the mean 24h flux corrected for
 250 headspace accumulation is shown as a dashed red line (Eq. 1 with k_{ch} from Eq. 2, or Eq. 3 with $c_1 = 1.21$).
 251 Labels denote fluxes calculated from the linear regression slopes (Eq. 3, black) and from Eq. 2 (red).

2.9 Computing gas transfer velocities with the surface renewal model

We used the surface renewal model (Lamont and Scott, 1970) formulated for small eddies at Reynolds numbers >500 (MacIntyre et al., 1995; Theofanous et al., 1976) to estimate k :

$$k_{mod} = \alpha(\varepsilon\nu)^{\frac{1}{4}} Sc^{-\frac{1}{2}} \quad [4]$$

where the hydrodynamic and thermodynamic forces driving gas transfer are expressed, respectively, as the TKE dissipation rate ε [m^2s^{-3}], and the dimensionless Schmidt number Sc , defined as the ratio of the kinematic viscosity ν [m^2s^{-1}] to the free solution diffusion coefficient D_0 [m^2s^{-1}] (Jähne et al., 1987; Wanninkhof, 2014). The scaling parameter α has a theoretical value of 0.37 (Katul et al., 2018), but is often estimated empirically (α') to calibrate the model (e.g. Wang et al., 2015). To allow for a qualitative comparison between model and chamber fluxes, we took ratios of k_{ch} (floating chambers) and $(\varepsilon\nu)^{\frac{1}{4}} Sc^{-\frac{1}{2}}$ (surface renewal model, half-hourly values of k_{mod} averaged over each chamber deployment period), and determined $\alpha' = 0.23 \pm 0.02$ for all lakes (mean \pm 95% CI, $n = 334$) (Fig. 3), and $\alpha' = 0.31 \pm 0.06$ ($n = 67$) for Villasjön, $\alpha' = 0.25 \pm 0.03$ ($n = 136$) for Inre Harsjön and $\alpha' = 0.17 \pm 0.02$ ($n = 131$) for Mellersta Harsjön (Supplementary Fig. 1). Calibrating the model in this way allowed us to assess whether chamber flux relationships with wind speed and temperature were reproduced by the model. For similar comparative purposes, k -values were normalized to a Schmidt number of 600 (CO_2 at 20°C) (Wanninkhof, 1992): $k_{600} = (600/Sc)^{-0.5}k$. The wind speed at 10 m (U_{10}) was computed from measured wind speed following Smith (1988), assuming a neutral atmosphere.

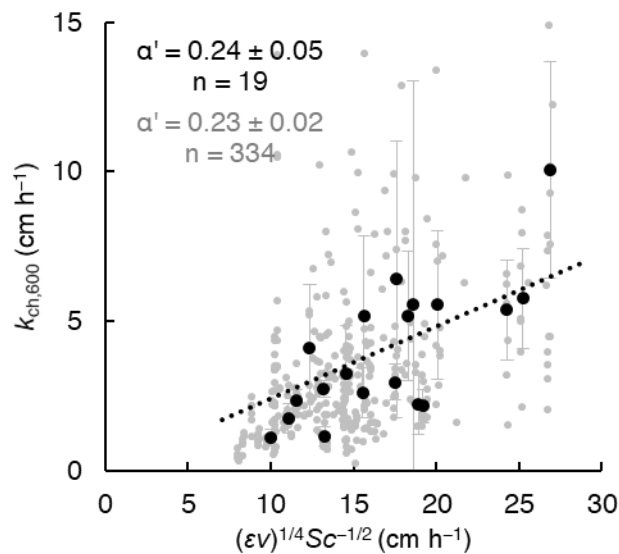


Figure 3 – Determination of the model scaling parameter α' via comparison between gas transfer velocities from floating chambers (Eq. 2) and the surface renewal model (Eq. 4 with $\alpha' = 1$ and $Sc = 600$, half-hourly values averaged over each chamber's 24 hour deployment period) for all three lakes. Dots represent individual chamber deployments (grey) and multi-chamber means for each weekly deployment in 2016 and 2017, when concentration measurements were taken simultaneously with, and in close proximity to the chamber measurements (black). Mean ratios, and therefore α' , are represented by the slopes of the dotted lines. Error bars represent 95% confidence intervals of the means.

292 We used a parametrization by Tedford et al. (2014) based on Monin-Obukhov similarity theory to estimate
 293 the TKE dissipation rate at half-hourly time intervals:

$$\varepsilon = \begin{cases} 0.56 u_{*w}^3 / \kappa z + 0.77\beta & \text{if } \beta > 0 \text{ (cooling)} \\ 0.6 u_{*w}^3 / \kappa z & \text{if } \beta \leq 0 \text{ (heating)} \end{cases} \quad [5]$$

294 where u_{*w} is the water friction velocity [m s^{-1}], κ is the von Kármán constant, z is depth below the water
 295 surface (0.15 m, the depth for which Eq. 5 was calibrated). We determined u_{*w} from the air friction velocity
 296 u_{*a} assuming equal shear stresses (τ) on both sides of the air-water interface; $\tau = \rho_a u_{*a}^2 = \rho_w u_{*w}^2$, and
 297 taking into account atmospheric stability (MacIntyre et al., 2014; Tedford et al., 2014). β is the buoyancy
 298 flux [$\text{m}^2 \text{s}^{-3}$], which accounts for turbulence generated by convection (Imberger, 1985):

$$\beta = \alpha_T g Q_{eff} / c_{pw} \rho_w \quad [6]$$

299 Here, α_T is the thermal expansion coefficient [$\text{m}^3 \text{K}^{-1}$] (Kell, 1975), g is the standard gravity [m s^{-2}], c_{pw} [J
 300 $\text{kg}^{-1} \text{K}^{-1}$] is the water specific heat and ρ_w [kg m^{-3}] is the water density. Q_{eff} [W m^{-2}] represents the net
 301 heat flux into the mixing layer and is the sum of net shortwave and long-wave radiation and sensible and
 302 latent heat fluxes. Penetration of radiation into the water column was evaluated across seven wavelength
 303 bands via Beer's Law (Jellison and Melack, 1993). An attenuation coefficient of 0.74 was computed for the
 304 visible portion of the spectrum from Secchi depth (2.3 m: Karlsson et al., 2010) following Idso and Gilbert
 305 (1974). Net longwave radiation ($LW_{net} = LW_{out} - LW_{in}$) was computed via measurements of LW_{in} (Table 1)
 306 and $LW_{out} = \sigma T^4$, where σ is the Stefan-Boltzmann constant ($5.67 \times 10^{-8} \text{ W m}^{-2} \text{K}^{-4}$) and T is the surface
 307 water temperature in K. LW_{net} timeseries were gap-filled with ice-free mean values for each lake. Sensible
 308 and latent heat fluxes were computed with bulk aerodynamic formula (MacIntyre et al., 2002). Both Q_{eff}
 309 and β are here defined as positive when the heat flux is directed out of the water, for example when the
 310 surface water cools.

311
 312 Direct measurements of ε in an Arctic pond (1 m depth, 0.005 km^2 surface area) demonstrate that Equation
 313 5 can characterize near-surface turbulence in small, sheltered water bodies similar to the lakes studied
 314 here (MacIntyre et al., 2018). When the near surface was strongly stratified at instrument depth (buoyancy
 315 frequencies ($N = \sqrt{g/\rho_w \times d\rho_w/dz}$) > 25 cycles per hour (cph)), the required assumption of
 316 homogeneous isotropic turbulence was not met and Equation 5 could not be evaluated. We observed
 317 cases with $N > 25$ cph $< 3\%$ of the time.

318

319 **2.10 Calculation of binned means**

320 We binned data to assess correlations between the flux and environmental covariates. Half-hourly values
 321 of water temperature and wind speed were averaged over the deployment period of each chamber
 322 (fluxes), and over 24 hours prior to the collection of each water sample (concentrations), reflecting the
 323 mean residence time of CH_4 in the water column. Fluxes, concentrations and k -values were then binned in
 324 10 day, 1 $^\circ\text{C}$ and 0.5 m s^{-1} bins to obtain relationships with time, water temperature and wind speed,
 325 respectively. The 10 day bins typically contained at least one sampling day for each overlapping year, and
 326 enabled representative averaging across years. Lake-dependent variables (e.g. flux) were normalized by
 327 lake to obtain a single timeseries (divided by the lake mean, multiplied by the overall mean).

328

329 **2.11 Calculation of the empirical activation energy**

330 Chamber and modelled fluxes as well as concentrations were fitted to an Arrhenius-type temperature
331 function (e.g. Wik et al., 2014; Yvon-Durocher et al., 2014):

$$F = e^{-E_a'/k_B T + b} \quad [7]$$

332 where k_B is the Boltzmann constant (8.62×10^{-5} eV K⁻¹) and T is the water temperature in K. The empirical
333 activation energy (E_a' , in electron volts (eV), 1 eV = 96 kJ mol⁻¹) was computed with a linear regression of
334 natural logarithm of the fluxes and concentrations onto the inverse temperature (1/K), of which b is the
335 intercept.

336

337 **2.12 Timescale analysis: power spectra and climacogram**

338 We computed power spectra for near-continuous timeseries of the surface sediment, water- and air
339 temperature and the wind speed according to Welch's method (pwelch in MATLAB 2018a), which splits
340 the signal into overlapping sections and applies a cosine tapering window to each section (Hamming,
341 1989). Data gaps were filled by linear interpolation. We removed the linear trend from original timeseries
342 to reduce red noise, and block-averaged spectra (8 segments with 50% overlap) to suppress aliasing at
343 higher frequencies. We normalized the spectral densities by multiplying by the frequency and dividing by
344 the variance of the original timeseries (Baldocchi et al., 2001).

345

346 We evaluated our discontinuous (fluxes, concentrations) and continuous (meteorology) timeseries with a
347 climacogram, an intuitive way to visualize a continuum of variability (Dimitriadis and Koutsoyiannis, 2015).
348 It displays the change of the standard deviation (σ) with averaging timescale (t_{avg}). Variables were
349 normalized by lake to create a single timeseries at half-hourly resolution (e.g. 48 entries for each 24-hour
350 chamber flux). To compute each standard deviation ($\sigma(t_{avg})$) data were binned according to averaging
351 timescale, which ranged from 30 minutes to 1 year. Because of the discontinuous nature of the datasets,
352 n bins were distributed randomly across the time series. We chose $n = 100000$ to ensure that the 95%
353 confidence interval of the standard deviation at the smallest bin size was less than 1% of the value of σ
354 (Sheskin, 2007). To allow for comparison between variables we normalized each σ -series by its initial,
355 smallest-bin value: $\sigma_{norm} = \sigma/\sigma_{init}$. For timescales < 1 week we used 1-hour chamber observations, noting
356 that sparse, daytime-only observations of concentrations and 1-hour fluxes may underestimate short-term
357 variability (σ_{init}). We use the climacogram to test whether the variability of the diffusive CH₄ flux is
358 contained within meteorological variability, as for terrestrial ecosystem processes (Pappas et al., 2017).

359

360 **2.13 Statistics**

361 We used Analysis of Variance (ANOVA) and the t-test to compare means of different groups. The use of
362 means, rather than medians was necessary because annual emissions can be determined by rare, high-
363 magnitude emission events. Parametric tests were justified because of the large number of samples in
364 each analysis, in accordance with the central limit theorem. Linear regressions were performed with the
365 ordinary least squares method (OLS): reported p -values refer to the significance of the regression slope.
366 Non-linear regressions were optimized with the Levenberg-Marquardt algorithm for non-linear least
367 squares with confidence intervals based on bootstrap replicates ($n = 1999$). Computations were done in
368 MATLAB 2018a and in PAST v3.25 (Paleontological Statistics software package) (Hammer et al., 2001).

369 **3. Results**

370 **3.1 Measurements and models**

371 Chamber fluxes averaged $6.9 \text{ mg m}^{-2} \text{ d}^{-1}$ (range 0.2–32.2, $n = 1306$) and closely tracked the temporal
372 evolution of the surface water concentrations (mean 11.9 mg m^{-3} , range 0.3–120.8, $n = 606$), with the
373 higher values in each lake measured in the warmest months (July and August, Fig 4a,e). Diffusive fluxes
374 increased with wind speed and water temperature (Fig 4b,c). Reduced emissions were measured in the
375 shoulder months (June and September) and were associated with lower water temperatures. We also
376 observed abrupt reductions of the flux at wind speeds lower than 2 m s^{-1} and higher than 6.5 m s^{-1} . Surface
377 water concentrations generally increased with temperature and peaked in the summer months, but unlike
378 the chamber fluxes they decreased with increasing wind speed (Fig. 4f,g). Relationships with wind speed
379 were approximately linear, while relationships with temperature fitted an Arrhenius-type exponential
380 function (Eq. 7). Activation energies were not significantly different when using either surface water or
381 sediment temperature ($E_a' = 0.90 \pm 0.14 \text{ eV}$, $R^2 = 0.93$, $E_a' = 1.00 \pm 0.17$, $R^2 = 0.93$, respectively, mean \pm 95%
382 CI). The fluxes, concentrations and the wind speed were non-normally distributed (Fig. 4d,h,o). Surface
383 water temperatures (0.1–0.5 m) were normally distributed around the mean of each individual month of
384 the ice-free season (Fig. 4n), but the composite distribution was bimodal.

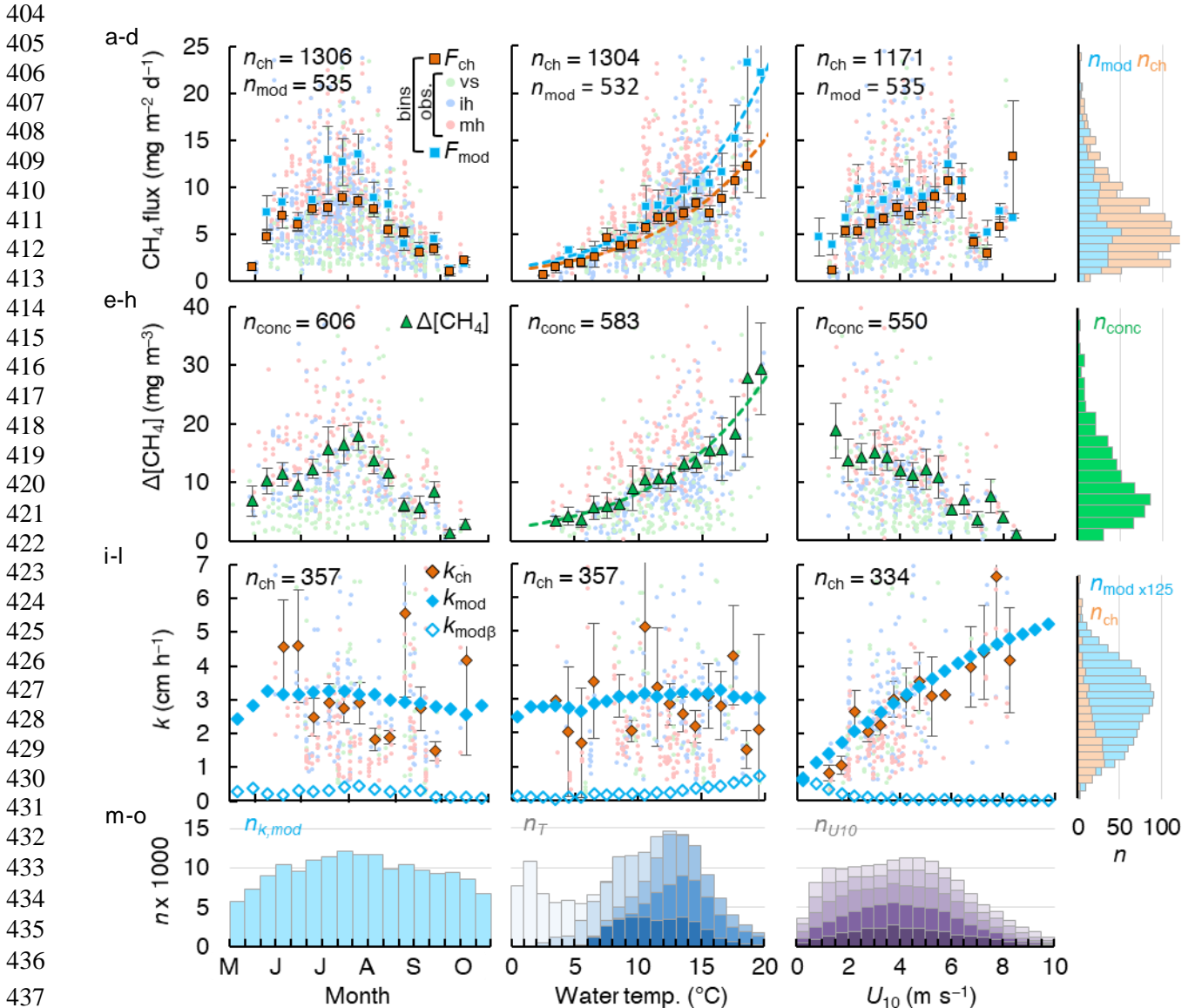
385
386 Fluxes computed with the surface renewal model (Eq. 1 using k_{mod}) closely resembled the chamber fluxes
387 (Eq. 3) in terms of temporal evolution (Fig. 4a) and correlation with environmental drivers (Fig. 4b,c). Mean
388 model fluxes were slightly higher than the chamber fluxes in Villasjön and Inre Harrsjön, and slightly lower
389 in Mellersta Harrsjön (Table 2). Model fluxes were significantly different between littoral and pelagic zones
390 in Inre and Mellersta Harrsjön (paired t-tests, $p \leq 0.02$), reflecting spatial differences in the surface water
391 concentration (Table 2). Similar to the chamber fluxes, the air-water concentration difference ($\Delta[\text{CH}_4]$)
392 explained most of the temporal variability of the modelled emissions; both k_{mod} (Eq. 4) and k_{ch} (Eq. 2) were
393 functions of U_{10} (Fig. 4k) and did not display a distinctive seasonal pattern (Fig. 4i). Modelled fluxes
394 decreased at higher wind speeds when surface concentrations decreased, and displayed a cut-off at daily
395 mean $U_{10} \geq 6.5 \text{ m s}^{-1}$, similar to the chamber fluxes, but not at $U_{10} < 2.0 \text{ m s}^{-1}$. The temperature sensitivity
396 of the modelled fluxes ($E_a' = 0.97 \pm 0.12 \text{ eV}$, mean \pm 95% CI, $R^2 = 0.94$) did not differ significantly from that
397 of the chamber fluxes.

398

399 **Table 2** – CH₄ fluxes from floating chambers and the surface renewal model, and surface CH₄ concentrations. Data
 400 from 2014 was excluded from the model flux means because of a substantial bias in the timing of sample collection.
 401 Model fluxes for each lake were computed with distinct scaling parameter values (Supplementary Fig. 1).

Location	Chamber flux (mg m ⁻² d ⁻¹)		Modelled flux (mg m ⁻² d ⁻¹)		Surface concentration (mg m ⁻³)	
	mean ± 95% CI	<i>n</i>	mean ± 95% CI	<i>n</i>	mean ± 95% CI	<i>n</i>
Overall	6.9 ± 0.3	1306	7.6 ± 0.5	501	11.9 ± 0.9	606
Villasjön	5.2 ± 0.5	249	7.0 ± 0.9	149	8.3 ± 1.1	183
Inre Harrsjön	6.6 ± 0.4	532	7.6 ± 0.7	176	10.2 ± 1.0	211
Shallow (<2 m)	6.0 ± 0.6	219	8.4 ± 0.9	113	11.1 ± 1.3	133
Intermediate (2-4 m)	7.1 ± 0.6	212				
Deep (>4 m)	6.6 ± 0.8	101	7.0 ± 0.9	63	8.6 ± 1.4	78
Mellersta Harrsjön	8.0 ± 0.4	525	7.7 ± 0.7	176	16.7 ± 2.0	212
Shallow (<2 m)	8.1 ± 0.6	272	8.3 ± 0.9	113	18.2 ± 2.7	134
Intermediate (2-4 m)	7.8 ± 0.7	154				
Deep (>4 m)	8.0 ± 1.0	99	6.8 ± 0.9	63	14.1 ± 2.7	78

402
 403



438 **Figure 4** – Scatterplots of the CH₄ flux (a-c), CH₄ air-water concentration difference (e-g) and gas transfer
 439 velocity (i-k) versus time, surface water temperature and wind speed, as well as the histograms of the
 440 aforementioned variables (d,h,i, m-o). In each scatter plot binned means of the flux (squares, a-c),
 441 concentrations (triangles, e-g) and gas transfer velocities (rhombuses, i-k) are represented by large
 442 symbols with 95% confidence intervals (error bars). Orange and light blue symbols reflect chamber-derived
 443 and model-derived binned values, respectively. Model k was computed with $\alpha' = 0.23$. Bin sizes were 10
 444 days, 1 °C and 0.5 m s⁻¹ for time, surface water temperature and U_{10} , respectively. Small green, blue and
 445 red dots represent individual measurements in Villasjön, Inre Harrsjön and Mellersta Harrsjön,
 446 respectively. Open rhombus symbols in panels i-k represent the buoyancy component of the gas transfer
 447 velocity, closed rhombus symbols include both the wind-driven and buoyancy-driven components. Dashed
 448 lines in panels b and f represent fitted Arrhenius functions (Eq. 7). Histograms of modelled (light blue) and
 449 measured (light orange) quantities (d,h,i) overlap. Histograms of the surface water temperature (m) and
 450 U_{10} (o) are stacked by month, from June (darkest shade) to October (lightest shade).

451 3.2 Meteorology and mixing regime

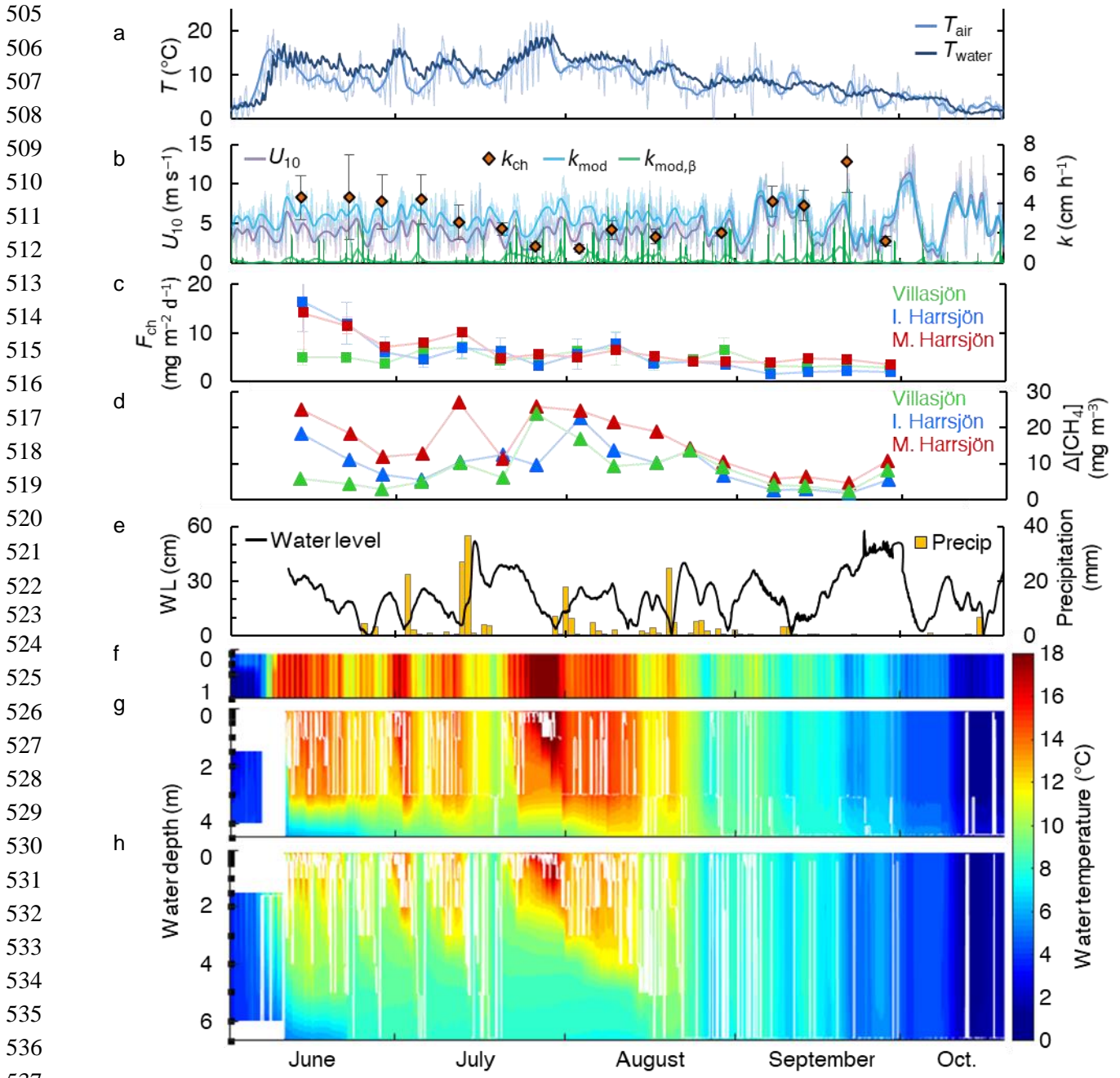
452 Throughout the ice-free season the lakes were weakly stratified (Table 3). Figure 5 shows a timeseries of
453 the mixed layer depth and water temperature in the deeper lakes, along with wind speed, air temperature
454 and precipitation for the ice-free period of 2017. The ice-free period consisted of two phases. In the first,
455 air and surface water temperatures were higher and the two deeper lakes stratified. Wind speeds
456 increased to mean values approaching 5 m s^{-1} for a few days at a time and then decreased for a day or
457 two. Deep mixing events followed surface cooling and heavy rainfall. Water level maxima and surface
458 temperature minima were observed 2-3 days after rainfall events, for example between 15 and 18 July
459 2017 (Fig. 5e). In the second phase, wind speeds were persistently higher ($U_{10} > 5 \text{ m s}^{-1}$), air and surface
460 water temperatures declined and all lakes mixed to the bottom. Strong nocturnal cooling on 16 August
461 2017 broke up stratification and the lakes remained well-mixed until ice-on (20 October). Throughout the
462 ice-free seasons from 2009–2018, stratified periods ($z_{\text{mix}} \leq 1 \text{ m}$) lasted for 7 hours on average and were
463 common (31% and 45% of the time in Inre and Mellersta Harrsjön, respectively), but were frequently
464 disrupted by deeper mixing events. Shallow mixing ($z_{\text{mix}} \leq z_{\text{mean}}$) occurred on diel timescales. Deeper mixing
465 occurred at longer intervals (days-weeks), and more frequently toward the end of the ice-free season (Fig.
466 5g,h) in association with higher wind speeds.

467 Fluxes and near surface concentrations also varied within these periods, with concentrations and fluxes
468 higher in the warmer, stratified period and lower in the colder, mixed periods. In 2017, the highest
469 concentrations and fluxes occurred earlier in the season, with the initial high values in the two deeper
470 lakes indicative of residual CH_4 that had not evaded immediately after ice-off, around 1 June 2017 (Fig.
471 5c,d). As residual CH_4 was emitted, near surface concentrations declined, and then in the first half of the
472 stratified period (July 2017, Fig. 5d), particularly in Mellersta Harrsjön, increased with increased rainfall
473 and with temperature. During this period, k_{ch} and k_{mod} were similar. Decreases in k_{ch} occurred when air
474 temperatures increased above surface water temperatures in the day leading to a stable atmosphere and
475 when near surface temperatures were warmer, and depending upon the lake, stratified to the surface.
476 Thus, lower fluxes occurred during the second part of the stratified period (August 2017, Fig. 5c) when
477 surface concentrations increased during warming periods when winds were light, the atmosphere was
478 stable during the day, and the upper water column was strongly stratified. Fluxes and concentrations were
479 lower in the autumn mixed periods, by which time the lakes had degassed, and with the colder surface
480 sediment temperatures, rates of production had decreased.

481 The modelled gas transfer velocity generally followed the temporal pattern of the wind speed (Fig. 4b).
482 Due to model calibration, the modelled gas transfer velocities (Fig. 4b, blue line) tracked those derived
483 from chamber observations (Fig. 4b, orange rhombuses). Discrepancies pointed to a mismatch between
484 24-hour integrated chamber fluxes and surface concentrations measured at a single point in time. For
485 example, measuring a low surface concentration in the de-gassed water column after a windy period
486 during which the surface flux was high led to an overestimated k_{ch} on 21 September 2017. Contrastingly,
487 k_{ch} was lower than k_{mod} on 3 August 2017 due to elevated surface concentrations and a low chamber flux
488 associated with a warm and stratified period preceding water sampling.

489
490 The temperature of the surface mixed layer exceeded the air temperature by $1.6 \text{ }^\circ\text{C}$ on average (Fig. 5a),
491 such that the atmospheric boundary layer over the lakes was often unstable, particularly at night during
492 warm periods as well as during the many cold fronts. We computed an unstable atmosphere over the lakes

493 ($z/L_{MO,a} < 0$, where z is the measurement height and $L_{MO,a}$ is the air-side Monin-Obukhov length; Foken
494 2006) $\sim 76\%$ of the time during ice-free seasons. Atmospheric instability increases sensible and latent heat
495 fluxes (Brutsaert, 1982), enhancing the cooling rate. Thus, buoyancy fluxes were positive at night and
496 during cold fronts throughout the ice-free season (Fig 5b, Fig. 4i-k). The magnitude of buoyancy flux during
497 cooling periods tended to range from 10^{-8} to 10^{-7} $\text{m}^2 \text{s}^{-3}$ in the stratified period and decreased as water
498 temperatures cooled in autumn (Fig. 4i,j). TKE dissipation rates at 0.15 m were high, with values often
499 between 10^{-6} and 10^{-5} $\text{m}^2 \text{s}^{-3}$, although values did fall as low as 10^{-8} $\text{m}^2 \text{s}^{-3}$ when winds were light.
500 Comparison of these two terms indicated that buoyancy flux during cooling was typically two orders of
501 magnitude less than ϵ and was only equal to it during the lightest winds (Fig. 4k). Consequently, its
502 contribution to the gas transfer coefficient was minor (Fig. 7). Averaged over all ice-free seasons (2009–
503 2017) the buoyancy flux contributed only 8% to the TKE dissipation rate, but up to 90% during rare, very
504 calm periods ($U_{10} \leq 0.5 \text{ m s}^{-1}$, Fig. 4k) and up to 25% on during the warmest periods ($T_{\text{surf}} \geq 18 \text{ }^\circ\text{C}$, Fig. 4j).

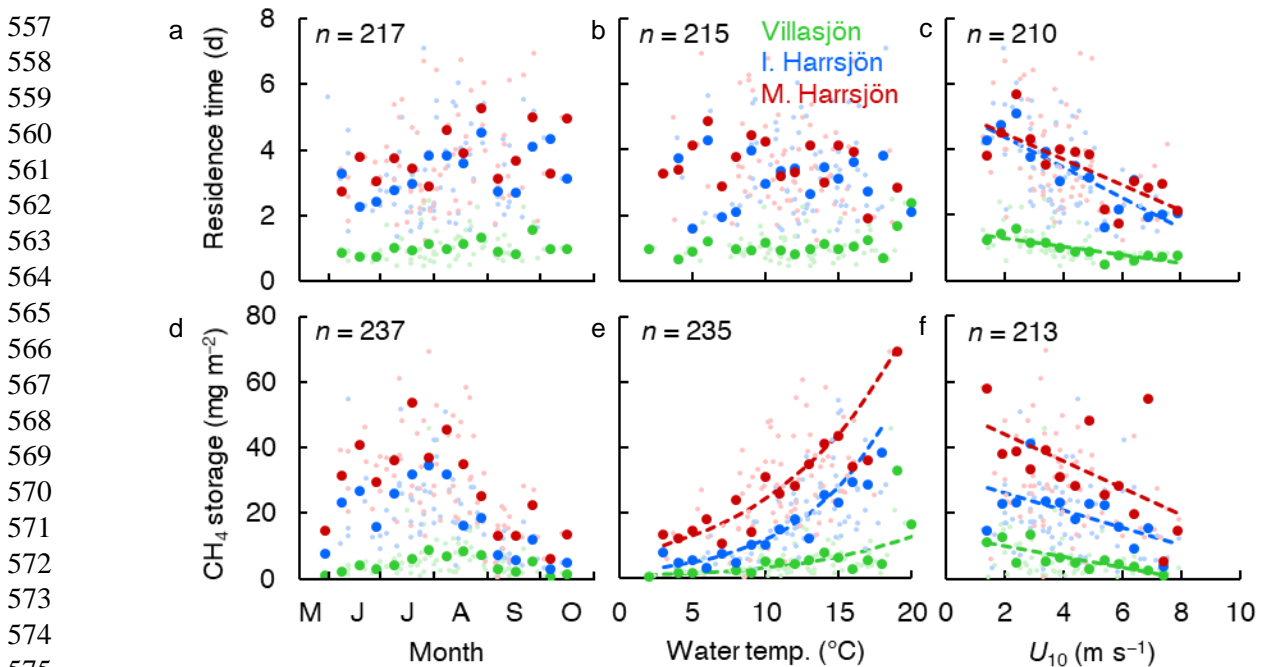


538 **Figure 5** – Timeseries of air and surface mixed-layer temperature (three-lake mean) (a), wind speed, gas
 539 transfer velocity from the surface renewal model (k_{mod} and its buoyancy component, $k_{\text{mod},\beta}$) and from
 540 chamber observations (k_{ch}) (three-lake mean values, error bars represent 95% confidence intervals) (b),
 541 chamber CH_4 flux (c), air-water CH_4 concentration difference (d), precipitation and changes in water level
 542 in Mellersta Harrsjön (e) and the water temperature in Villasjön (f), Inre Harrsjön (g) and Mellersta
 543 Harrsjön (h) during the ice-free season of 2017 (1 June to 20 October). The white lines in panels f-h
 544 represent the depth of the surface mixed layer. Thin and thick lines in panels a and b represent half-hourly
 545 and daily means, respectively. In panel a only the half-hourly timeseries of T_{water} was plotted.

546 **Table 3** – Lake morphometry, temperature of the surface mixing layer, buoyancy frequency and CH₄ residence time.
 547 Mean values were calculated over the ice-free seasons of 2009–2017.

Lake	Area (ha)	Depth (m)		Mixing layer temp. (°C)		N (cycles h ⁻¹)		CH ₄ residence time (days)	
		mean	max	mean ± SD	n	mean ± SD	n	mean ± SD	n
Villasjön	17.0	0.7	1.3	9.9 ± 5.5	148976	5.7 ± 8.0	59552	1.0 ± 0.4	72
Inre Harrsjön	2.3	2.0	5.2	10.1 ± 5.2	278752	5.2 ± 6.9	66757	3.4 ± 1.9	73
Mellersta Harrsjön	1.1	1.9	6.7	9.2 ± 4.9	278014	5.3 ± 9.0	61268	3.7 ± 1.7	72

548
 549 **3.3 CH₄ storage and residence times**
 550 Residence times of stored CH₄ varied between 12 hours and 7 days and were inversely correlated with
 551 wind speed in all three lakes (OLS: R² ≥ 0.57, Fig. 6). The mean residence time was shortest in the shallowest
 552 lake, and was not significantly different between the two deeper lakes (paired t-test, *p* < 0.01, Table 3).
 553 We did not find a statistically significant linear correlation between the residence time and day of year or
 554 the water temperature. CH₄ storage was greatest in the deeper lakes and displayed patterns similar to the
 555 surface concentrations, increasing in the warmest months with water temperature and decreasing with
 556 wind speed.



576 **Figure 6** – Scatterplots of the CH₄ residence time (a-c) and storage (d-f) versus time, surface water
 577 temperature and wind speed. Symbol colours represent the different lakes. Large symbols represent
 578 binned means, small symbols represent individual estimates. Bin sizes were 10 days, 1 °C and 0.5 m s⁻¹ for
 579 time, water temperature and U₁₀, respectively. Each storage observation was paired with T and U₁₀
 580 averaged over the 24h (Villasjön) and 72h (Inre and Mellersta Harrsjön) prior to water sampling, reflecting
 581 average conditions during CH₄ residence times. The linear regressions of the residence time onto time (a)
 582 and temperature (b) were not statistically significant (*p* = 0.07–0.10). Linear relations of binned quantities
 583 and U₁₀ were statistically significant (c: *p* ≤ 0.002; f: *p* ≤ 0.04). Arrhenius-type functions (Eq. 7) adequately
 584 described the storage-temperature relation in each lake (e: R² ≥ 0.70, *p* < 0.001).

585 3.4 Variability

586 Chamber fluxes and surface water concentrations differed significantly between lakes (ANOVA, $p < 0.001$,
587 $n = 287$, $n = 365$) (Table 2). Both quantities were inversely correlated with lake surface area. CH₄
588 concentrations in the stream feeding the Mire ($22.2 \pm 5.1 \text{ mg m}^{-3}$, $n = 29$, mean \pm 95% CI), were significantly
589 higher than those in the lakes (Table 2). Surface water concentrations over the deep parts of the deeper
590 lakes ($\geq 2 \text{ m}$ water depth) were lower than those in the shallows ($< 2 \text{ m}$) by 21 to 26% for Inre and Mellersta
591 Harrsjön, respectively. However, the diffusive CH₄ flux did not differ significantly between depth zones in
592 either Inre Harrsjön (ANOVA, $p = 0.27$, $n = 290$) or Mellersta Harrsjön (ANOVA, $p = 0.90$, $n = 293$), or
593 between zones of high and low CH₄ ebullition in Villasjön (paired t-test, $p = 0.27$, $n = 89$). The similar fluxes
594 inshore and offshore present a contrast with ebullition, for which the highest fluxes were consistently
595 observed in the shallow lake and littoral areas of the deeper lakes (Jansen et al., 2019; Wik et al., 2013).

596
597 Relations between the flux and its drivers — temperature, wind speed and the surface concentration —
598 manifested on different timescales (Fig. 7). Over the ice-free season both the CH₄ fluxes and surface water
599 concentrations tracked changes in the water temperature. The wind speed (U_{10}) showed less variability
600 over seasonal (CV = 7%, $n = 17$) than over diel timescales (CV = 12%, $n = 24$) and displayed a clear diurnal
601 maximum. The surface water/sediment temperature varied primarily on a seasonal timescale (CV =
602 52%/45%, $n = 17$), and less on diel timescales (CV = 3%/2%, $n = 24$). Similar to the wind speed the gas
603 transfer velocity varied primarily on diel timescales (Fig. 7), albeit with a lower amplitude. This was in part
604 because $k_{mod} \propto u^{3/4}$ (Eq. 4), and because the drag coefficient, used to compute the water-side friction
605 velocity in Equation 5, increases at lower wind speeds and under an unstable atmosphere, which was
606 typically the case. The surface concentration correlated with wind speed and temperature (Fig. 4f,g), and
607 showed both seasonal and diel variability. On diel timescales $\Delta[\text{CH}_4]$ and k_{mod} were out of phase; $\Delta[\text{CH}_4]$
608 peaked just before noon, when the gas transfer velocity reached its maximum value (Fig. 7b,d). However,
609 binned means of the 1-hour chamber fluxes ($F_{ch}(1h)$) were not significantly different at the 95% confidence
610 level (error bars) and did not show a clear diel pattern (Fig. 7b). Temporal patterns of fluxes and
611 concentrations were very similar between the lakes (Supplementary Fig. 2 and 3).

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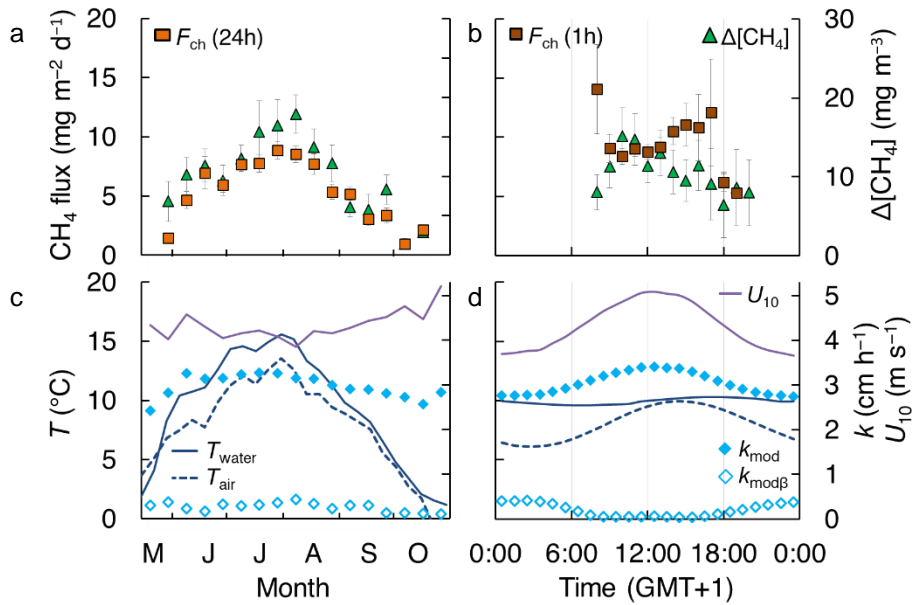
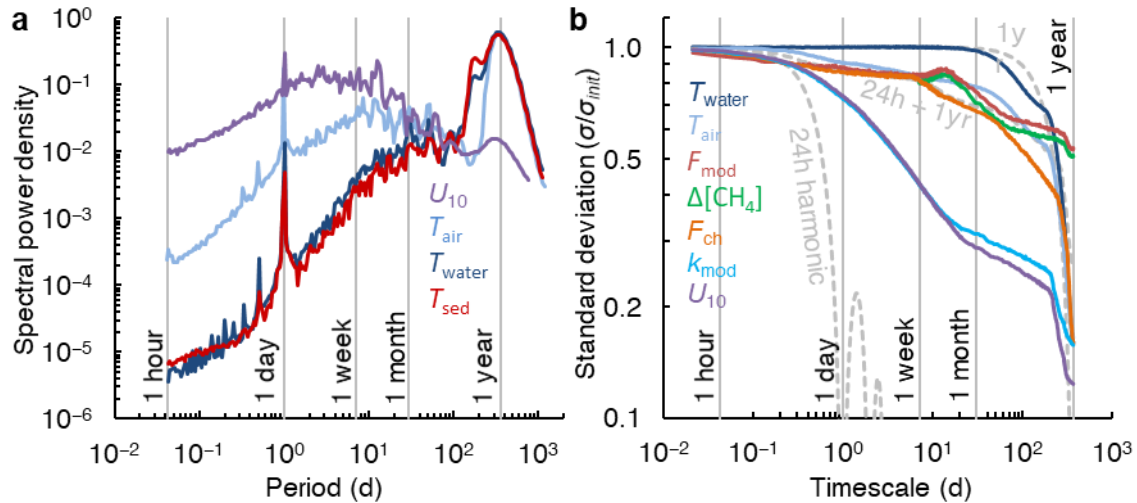


Figure 7 – Temporal patterns of CH₄ chamber fluxes, concentrations (a,b), gas transfer velocity, air and surface water temperature and wind speed (c,d). Bin sizes are 10 days (a,c) and 1 hour (b,d). Error bars represent 95% confidence intervals of the binned means. Temporal patterns in each individual lake are shown in Supplementary Figures 2 and 3.

632 **3.5 Timescale analysis**

633 The spectral density plot (Fig. 8a) disentangles dominant timescales of variability of the drivers of the flux.
634 The power spectra of wind speed and temperature peaked at periods of 1 day and 1 year, following well-
635 known diel and annual cycles of insolation and seasonal variations in climate (Baldocchi et al., 2001). The
636 diel spectral peak was subdued for the surface sediment temperature. For U_{10} , the overall spectral density
637 maximum between 1 day and 1 week, and somewhat longer in spectra for the ice-free period only
638 (Supplementary Fig. 4), corresponds to synoptic-scale weather variability, such as the passage of fronts
639 (MacIntyre et al., 2009). U_{10} and T_{air} also exhibit spectral density peaks at 1–3 weeks, which could be
640 associated with persistent atmospheric blocking typical of the Scandinavian region (Tyrllis and Hoskins,
641 2008). While the temperature variability was concentrated at annual timescales, the wind speed varied
642 primarily on timescales shorter than about a month and often shorter than a week.

643
644 The climacogram (Fig. 8b) reveals that the variability of the chamber flux and the gas transfer velocity was
645 enveloped by that of the water temperature and the wind speed, as was the surface concentration
646 difference for timescales < 5 months. The distribution of variability over the different timescales is similar
647 to that shown in the spectral density plot (Fig. 8a). The standard deviation of the water temperature did
648 not change from its initial value ($\sigma/\sigma_{\text{init}} = 1$) until timescales of about 1 month, following the 1 year
649 harmonic. In contrast, most of the variability of the wind speed was concentrated at time scales shorter
650 than 1 month. The variability of the chamber and modelled fluxes first tracked that of the wind speed, but
651 for timescales longer than about 1 month the decrease in variability resembled that of water temperature.
652 The variability of the modelled fluxes followed that of the surface concentration difference rather than the
653 gas transfer velocity. However, the coarse sampling resolution of the fluxes and concentrations may have
654 led to an underestimation of both the variability at <1-week timescales (Fig. 7b) and the value of σ_{init} .
655 Finally, the climacogram shows that k_{mod} retains about 72% of its variability at 24-hour timescales, which
656 justifies our averaging over chamber deployment periods for comparison with k_{ch} and the computation of
657 the model scaling parameter α' (Fig. 3).



658
 659 **Figure 8** – Timescale analysis of the diffusive CH_4 flux and its drivers. **a**: Normalized spectral density of
 660 whole-year near-continuous timeseries of the air temperature (T_{air}), temperature of the surface water and
 661 ice (0.1–0.5 m, T_{water}), temperature of the surface sediment in Mellersta Harrsjön (T_{sed}) and the wind speed
 662 (U_{10}). **b**: Climacogram of the measured and modelled CH_4 flux (F_{ch} , F_{mod}), the air and surface water
 663 temperature (T_{air} , T_{water}), water-air concentration difference ($\Delta[CH_4]$), modelled gas transfer velocity (k_{mod})
 664 and the wind speed (U_{10}) during the ice-free seasons of 2009–2017. Dashed, light-grey curves represent
 665 (combinations of) trigonometric functions of mean 0 and amplitude 1 with a specified period. 24h and 1yr
 666 harmonic functions were continuous over the dataset period while the 24h + 1yr harmonic was limited to
 667 periods when chamber flux data were available. Panel a is based on continuous timeseries that include
 668 the ice-cover seasons: Supplementary Figure 4 shows spectral density plots for individual ice-free seasons.

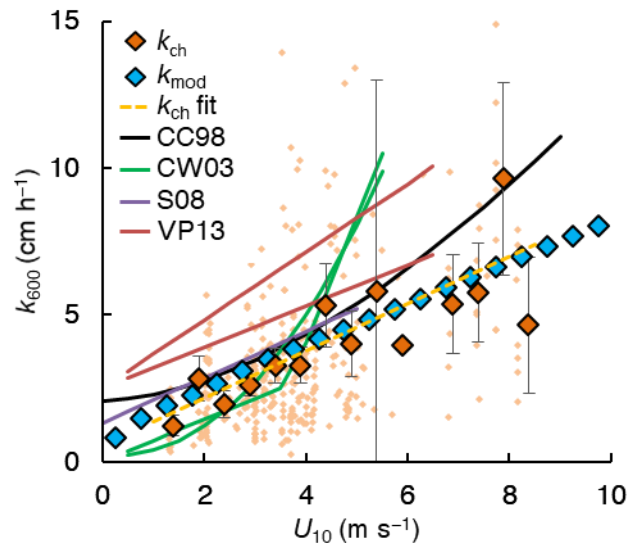
669 **4. Discussion**

670 **4.1 Magnitudes of fluxes and gas transfer velocities**

671 Overall, diffusive CH₄ emissions from the Stordalen Mire lakes ($6.9 \pm 0.3 \text{ mg m}^{-2} \text{ d}^{-1}$, mean \pm 95% CI) were
672 lower than the average of postglacial lakes north of 50°N, but within the interquartile range (mean 12.5,
673 IQR 3.0–17.9 $\text{mg m}^{-2} \text{ d}^{-1}$, Wik et al., 2016b). Emissions are also at the lower end of the range for northern
674 lakes of similar size (0.01–0.2 km²) (1–100 $\text{mg m}^{-2} \text{ d}^{-1}$, Wik et al., 2016b). As emissions of the Stordalen
675 Mire lakes do not appear to be limited by substrate quality or quantity (Wik et al., 2018), but strongly
676 depend on temperature (Fig. 4b), the difference is likely because a majority of flux measurements from
677 other postglacial lakes were conducted in the warmer, subarctic boreal zone. Boreal lake CH₄ emissions
678 are generally higher for lakes of similar size: 20–40 $\text{mg m}^{-2} \text{ d}^{-1}$ (binned means), $n = 91$ (Rasilo et al., 2015);
679 $\sim 12 \text{ mg m}^{-2} \text{ d}^{-1}$, $n = 72$ (Juutinen et al., 2009).

680
681 The gas transfer velocity in the Stordalen Mire lakes was similar to that predicted from wind-based models
682 of Cole and Caraco (1998) and Crusius and Wanninkhof (2003) at low wind speeds (Fig. 9). Both were based
683 on tracer experiments with sampling over several days, and thus, like our approach, are integrative
684 measures. The slope of the linear wind- k_{ch} relation (OLS: 0.81 ± 0.21 , slope \pm 95% CI, $R^2 = 0.20$ and $p < 0.01$
685 for the individual k_{ch} estimates (small orange rhombuses in Fig. 9)) was similar to that reported by Soumis
686 et al. (2008) (0.78 for a 0.06 km² lake), who also used a mass balance approach, and Vachon and Prairie
687 (2013) (0.70–1.16 for lakes 0.01–0.15 km²). Part of the difference with the models of Vachon and Prairie
688 (2013), Cole and Caraco (1998) and Soumis et al. (2008) was caused by the offset at 0 wind speed, which
689 may stem from a larger contribution of the buoyancy flux in their lakes than we computed for our lakes
690 with the surface renewal model (Crill et al., 1988; Read et al., 2012) or from remnant wind shear turbulence
691 (MacIntyre et al., 2018). While fetch limitation can reduce gas transfer at high wind speeds in small lakes
692 (Vachon and Prairie, 2013; Wanninkhof, 1992), and the lakes studied here are at the low end of the size
693 spectrum of water bodies in which the gas transfer models in Fig. 9 were developed (Table S1), there are
694 a number of other explanations for the low values we obtained. We further discuss these in section 4.5
695 after evaluating drivers of fluxes.

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707 **Figure 9** – Normalized gas transfer velocities (k_{600}) versus the wind speed at 10 m (U_{10}). Binned values
708 (large rhombuses, k_{ch} and k_{mod} , bin size = 0.5 m s⁻¹) and individual observations (small rhombuses, k_{ch}) from
709 floating chambers (k_{ch}) and the surface renewal model (k_{mod} with $\alpha' = 0.23$). Error bars represent 95%
710 confidence intervals of the binned means. Solid lines represent models from the literature: Cole and
711 Caraco (1998) (CC98), Crusius and Wanninkhof (2003) (bilinear and power law models) (CW03), Soumis et
712 al. (2008) (S08) and Vachon and Prairie (2013) (VP13) for lake surface areas of 0.01 and 0.15 km².
713 Supplementary Table 1 lists the model equations and calibration ranges. A power-law regression model is
714 shown for the individual k_{ch} datapoints ($n = 334$): $k_{600} = 0.77 \times U_{10}^{1.02} + 0.62$ (dashed yellow line).

715 4.2 Drivers of flux

716 Methane emitted from lakes in wetland environments can be produced in situ, or be transported in from
717 the surrounding landscape (Paytan et al., 2015). The distinction is important because some controls on
718 terrestrial methane production, such as water table depth (Brown et al., 2014), are irrelevant in lakes. In
719 the Stordalen Mire lakes, the Arrhenius-type relation of CH₄ fluxes and concentrations (Fig. 4b,f) together
720 with short CH₄ residence times (Fig. 6) suggest that efficient redistribution of dissolved CH₄ strongly
721 coupled emissions to sediment methane production. High CH₄ concentrations in the stream (section 3.4)
722 further suggest that external inputs of CH₄ — produced in the fens and transported into the stream with
723 surface runoff, or produced in stream sediments — may have elevated emissions in Mellersta Harrsjön
724 (Lundin et al., 2013). However, although the Mire exports substantial quantities of DOC and presumably
725 CH₄ from the water-logged fens to the lakes (Olefeldt and Roulet, 2012), after rainy periods we observed
726 either a decrease in $\Delta[\text{CH}_4]$ (13–19 July 2017, Fig. 5) or no significant change (3–6 July and 21–27 August
727 2017, Fig. 5). It remains unclear whether such reduced storage resulted from lower methanogenesis rates
728 associated with the temperature drop after rainfall, convection-induced degassing, or lake water
729 displacement or dilution by surface runoff.

730
731 Turbulent transfer was dominated by wind shear in the Stordalen Mire lakes, and we computed a minor
732 contribution (~8%) of the buoyancy-controlled fraction of k . Our result differs from that in Read et al.
733 (2012) who found that buoyancy flux dominated turbulence production in temperate lakes 0.1 km² in size
734 and smaller. For the Stordalen lakes we computed higher ice-free season mean values of u_{*w} , as well as
735 lower values of the water-side vertical friction velocity, $w_{*w} = (\beta z_{mix})^{1/3}$, (1.2–1.8 mm s⁻¹) than they
736 report (2.0–7.5 mm s⁻¹, $n = 40$ lakes). The difference here results from high wind speeds and often colder
737 surface waters compared to many temperate lakes. Therefore, values of sensible and latent heat fluxes
738 are lower in our lakes than in lakes in warmer regions. Many small lakes have low wind speeds particularly
739 at night. Consequently, the temperate lakes surveyed in Read et al. (2012), will have a larger contribution
740 of buoyancy flux to the gas transfer coefficient at night (MacIntyre and Melack, 2009). The contribution of
741 convection also depends on the wind-sheltering properties of the landscape surrounding the lake
742 (Kankaala et al., 2013; Markfort et al., 2010). Depending on the turbulence environment, the buoyancy
743 flux is thus weighed differently in parameterizations of ϵ (Heiskanen et al., 2014; Tedford et al., 2014) and
744 in wind-based models (offsets at $U_{10} = 0$ in Fig. 9), contributing to significant differences between model
745 realizations of k (Dugan et al., 2016; Erkkilä et al., 2018; Schilder et al., 2016).

746
747 The distinct spectral peaks of temperature and U_{10} (Fig. 8) indicate that flux dependencies on these
748 parameters (Fig. 4b,c) acted on different timescales. This difference has implications for the choice of
749 models or proxies of the flux in predictive analyses. For lakes that mix frequently and a climatology similar
750 to that of the Stordalen Mire (Malmer et al., 2005), temperature-based proxies (e.g. Thornton et al., 2015)
751 would resolve most of the variability of the ice-free diffusive CH₄ flux at timescales longer than a month.
752 Advanced gas transfer models that account for atmospheric stability and rapid variations in wind shear,
753 such as we have used here, allowed us to resolve variability in flux at timescales shorter than about a
754 month. Our results are representative of small, wind-exposed lakes in cold environments, where, as a
755 result of considerable wind driven mixing, fluxes are lower than would be predicted in lakes where
756 buoyancy fluxes during heating and cooling are higher.

757 **4.3 Storage and stability**

758 The robust temperature-sensitivity of lake methane emissions (Fig. 4b,f) (Wik et al., 2014; Yvon-Durocher
759 et al., 2014) is driven by biotic and abiotic mechanisms. Lake mixing can modulate temperature relations
760 by periodically decoupling production from emission rates (Engle and Melack, 2000). Here, enhanced CH₄
761 accumulation during periods of stratification may have contributed to concentration and storage maxima
762 in July and August (Fig. 4e, 6d). However, as the CH₄ residence time was invariant over the season and with
763 temperature (Fig. 6a,b), the storage-temperature relation (Fig. 6e) likely reflects rate changes in sediment
764 methanogenesis rather than inhibited mixing. For example, the highest CH₄ concentrations in our dataset
765 ($59.1 \pm 26.4 \text{ mg m}^{-3}$, $n = 37$) were measured during a period with exceptionally high surface water
766 temperatures ($T_{\text{water}} = 18.5 \pm 3.6 \text{ }^\circ\text{C}$) that lasted from 23 June to 30 July 2014. Emissions during this period
767 comprised 29%–56% (depending on lake) of the 2014 ice-free diffusive flux, while the peak quantity of
768 accumulated CH₄ comprised <5%. Two mechanisms may explain the lack of CH₄ accumulation. First,
769 stratification was frequently disrupted by vertical mixing (Fig. 5g-h) and concurrent hypolimnetic CH₄
770 concentrations were not significantly different from (Inre Harsjön, 2010–2017, paired t-test, $p = 0.12$, $n =$
771 32) or lower than (Mellersta Harsjön, 2010–2017, paired t-test, $p < 0.01$, $n = 35$) those in the surface mixed
772 layer. Second, stratification often was not strong enough to affect gas transfer velocities. Even when
773 assuming ϵ was suppressed by an order of magnitude for $N > 25$ and by two orders of magnitude for $N > 40$
774 (MacIntyre et al., 2018), k_{mod} was only slightly lower (2.8 cm h^{-1}) than the multi-year mean (3.0 cm h^{-1}).
775 Thus, in weakly stratified lakes with strong wind mixing, the temperature sensitivity of diffusive CH₄
776 emissions may be observed without significant modulation by stratification.

777
778 Degassing (Fig. 4c,g) prevented an unlimited increase of the emission rate with the gas transfer velocity.
779 In this way, $\Delta[\text{CH}_4]$ acted as a negative feedback that maintained a quasi-steady state between CH₄
780 production and removal processes throughout the ice-free season. In all three lakes CH₄ residence times
781 were inversely proportional to the wind speed (Fig. 6c), indicating an imbalance between production and
782 removal processes. We hypothesize that the imbalance exists because the variability of wind speed peaked
783 on shorter timescales than that of the water temperature (Fig. 8a). Changes in wind shear periodically
784 pushed the system out of production-emission equilibrium, allowing for transient degassing and
785 accumulation of dissolved CH₄. The temporal variability of dissolved gas concentrations is likely higher in
786 shallow wind-exposed systems with limited buffer capacity (Natchimuthu et al., 2016, 2017), and should
787 be taken into account when applying gas transfer models to small lakes and ponds.

788
789 Rapid degassing occurred at $U_{10} \geq 6.5 \text{ m s}^{-1}$ (Fig. 4c). Gas fluxes at high wind speeds may have been
790 enhanced by the kinetic action of breaking waves (Terray et al., 1996) or through microbubble-mediated
791 transfer. Wave breaking was observed on the Stordalen lakes at wind speeds $\geq 7 \text{ m s}^{-1}$. Microbubbles of
792 atmospheric gas (diameter $< 1 \text{ mm}$) can form due to photosynthesis, rain or wave breaking (Woolf and
793 Thorpe, 1991) and remain entrained for several days (Turner, 1961). Due to their relatively large surface
794 area they quickly equilibrate with sparingly soluble gases in the water column, providing an efficient
795 emission pathway to the atmosphere when the bubbles rise to the surface (Merlivat and Memery, 1983).
796 In inland waters microbubble emissions of CH₄ have only been indirectly inferred from differences in CO₂
797 and CH₄ gas transfer velocities (McGinnis et al., 2015; Prairie and del Giorgio, 2013), and more work is
798 needed to evaluate their significance in relatively sheltered systems.

799 **4.4 Timescales of variability**

800 Overall, the short-term variability of the flux due to wind speed ($1.1\text{--}13.2 \text{ mg m}^{-2} \text{ d}^{-1}$) was similar to the
801 long-term variability due to temperature ($0.7\text{--}12.2 \text{ mg m}^{-2} \text{ d}^{-1}$) (ranges of the binned means, Fig. 4b-c).
802 The diel patterns in the mixed layer depth (Fig. 5) and the gas transfer velocity (Fig. 7d) and daytime
803 variation of the surface concentration (Fig. 7b) were indicative of daily storage-and-release cycles,
804 resulting in a model flux difference of about $5 \text{ mg m}^{-2} \text{ d}^{-1}$ between morning and afternoon; about half the
805 mean seasonal range (Fig. 7a). Diel variability of lake methane fluxes has been observed at Villasjön (eddy
806 covariance, Jammet et al., 2017) and elsewhere (Bastviken et al., 2004, 2010; Crill et al., 1988; Erkkilä et
807 al., 2018; Eugster et al., 2011; Hamilton et al., 1994; Podgrajsek et al., 2014). Similarly, diel patterns in the
808 gas transfer velocity have been observed with the eddy covariance technique (Podgrajsek et al., 2015) and
809 in model studies (Erkkilä et al., 2018). Apparent offsets between the diurnal peaks of the flux, surface
810 concentrations and drivers (Fig 7b,d) have been noted previously (Koebsch et al., 2015), but have yet to
811 be explained. Continuous eddy covariance measurements in lakes where the dominant emission pathway
812 is turbulence-driven diffusion could help characterize flux variability on short timescales (e.g. Bartosiewicz
813 et al., 2015).

814
815 The CH_4 residence times (1–3 days) were not much longer than the diel timescale of vertical mixing (Fig.
816 5g,h). As a result, horizontal concentration gradients developed in the deeper lakes (Table 2). The $23 \pm$
817 11% concentration difference between depth zones in the deeper lakes (mean \pm 95%) fits transport model
818 predictions of DelSontro et al. (2017) for small lakes ($< 1 \text{ km}^2$) that highlight the role of outgassing and
819 oxidation during transport from production zones in the shallow littoral zones or the deeper sediments
820 (Hofmann, 2013). Concentration gradients may also have been caused by physical processes, such as
821 upwelling due to thermocline tilting (Heiskanen et al., 2014). Higher resolution measurements, for
822 example with automated equilibration systems (Erkkilä et al., 2018; Natchimuthu et al., 2016), are needed
823 to assess how much of the spatial and diel patterns of the CH_4 concentration can be explained by physical
824 drivers such as gas transfer and mixed layer deepening (Eugster et al., 2003; Vachon et al., 2019), or by
825 biological processes such as methanogenesis and microbial oxidation (Ford et al., 2002).

826
827 Gas transfer models can only deliver accurate fluxes if they are combined with measurements that capture
828 the full spatiotemporal variability of the surface concentration (Erkkilä et al., 2018; Hofmann, 2013;
829 Natchimuthu et al., 2016; Schilder et al., 2016). The short CH_4 residence times and diel pattern of $\Delta[\text{CH}_4]$
830 suggest that weekly sampling did not capture the full temporal variability of the surface concentrations.
831 Especially after episodes of high wind speeds and lake degassing (Fig. 4c,g), concentrations may not have
832 been representative of the 24-hour chamber deployment period.

833 4.5 Model-chamber comparison

834 It is fundamental to our understanding of controls on fluxes to determine why empirically derived values
835 of the model scaling parameter α' are relatively low in this study (0.17–0.31) compared to the theoretical
836 value of $\sqrt{2/15} \cong 0.37$ (Katul et al., 2018), and why they were different in the three lakes. Differences
837 in α' resulted from k_{ch} , with mean (\pm 95% CI) values estimated at 3.5 ± 0.7 ($n = 74$), 3.1 ± 0.4 ($n = 131$) and
838 2.5 ± 0.6 ($n = 142$) cm h^{-1} in Villasjön, Inre Harrsjön and Mellersta Harrsjön, respectively, while k_{mod} did not
839 differ significantly between lakes (ANOVA, $p < 0.001$). Synthesis studies show that scaling parameter values
840 can vary between 0.1 and 0.7 over the range of moderate to high dissipation rates computed for the
841 Stordalen Mire lakes (Eq. 5: $\varepsilon = 10^{-7}$ – 10^{-5} $\text{m}^2 \text{s}^{-3}$) (Esters et al., 2017; Wang et al., 2015 and references
842 therein). In such cases ε has been measured directly with acoustic Doppler- or particle image velocimetry
843 and compared with independent estimates of k using chambers (Gålfalk et al., 2013; Tokoro et al., 2008;
844 Vachon et al., 2010; Wang et al., 2015), eddy covariance observations (Heiskanen et al., 2014) or the
845 gradient flux technique (Zappa et al., 2007) and a sparingly soluble tracer, such as CO_2 or SF_6 . Measured
846 and modelled lake CO_2 fluxes agree reasonably well if Eq. 4 and Eq. 5 are used with a multi-study mean α'
847 of 0.5 (Bartosiewicz et al., 2015; Czikowsky et al., 2018; Erkkilä et al., 2018; Mammarella et al., 2015), but
848 the agreement is less clear for CH_4 fluxes (Bartosiewicz et al., 2015). The observed variability in α' could be
849 explained by chemical or biological factors that limit surface exchange, or by the variable contributions of
850 wind sheltering, atmospheric stability, and within lake stratification and mixing. Here, the low α' value may
851 imply an underestimation of k derived from chamber observations or an overestimation of dissipation
852 rates used in the modelling of gas transfer velocities.

853
854 An underestimation of chamber-derived gas transfer velocities may have resulted from an overestimation
855 of C_{aq} in Equation 1. In most freshwater systems a significant fraction of CH_4 is removed through microbial
856 oxidation (Bastviken et al., 2002). This additional removal process invalidates the implicit assumption in
857 Eq. 1 and 2 that all dissolved CH_4 that we measure in the surface water is emitted to the atmosphere.
858 Omitting oxidation would bias $\Delta[\text{CH}_4]$ high, and k_{ch} low. The Stordalen Mire lakes remained oxygenated
859 throughout the ice-free season and CH_4 stable isotopes indicate that between 24% (Villasjön) and 60%
860 (Inre and Mellersta Harrsjön) of CH_4 in the water column was continually oxidized (Jansen et al., 2019).
861 This may explain not only the low scaling parameter value compared to those found with other tracers,
862 but also why α' was higher in Villasjön (0.31, $n = 67$) than in the deeper lakes (0.17–0.25, $n = 267$)
863 (Supplementary Fig. 1). However, more work is needed to establish how the oxidation effect partitioned
864 between CH_4 reservoirs in the water column, where it would affect surface emissions, and the sediment.
865 An increase in surface concentrations which typically occurs at night would not have been manifest (Crill
866 et al., 1988; Czikowsky et al., 2018) because there was, apart from the period just after ice-off in 2017, no
867 significant CH_4 accumulation below the mixing layer throughout the ice-free seasons. Indeed, CH_4
868 concentrations within the 0.1–1 m surface layer of the deeper lakes (Table 2) were not significantly
869 different from those at greater depth (Inre Harrsjön: 12.2 ± 2.7 mg m^{-3} , $n = 292$; Mellersta Harrsjön: 17.7
870 ± 4.9 mg m^{-3} , $n = 405$; means \pm 95% CI).

871
872 An overestimation of gas transfer velocities computed with the surface renewal model may result if actual
873 dissipation rates are lower than we compute. Such occurs under high wind shear when more of the
874 introduced turbulent kinetic energy is used for mixing the water column and deepening the mixing layer,

875 and less is dissipated (Ivey and Imberger, 1991; Jonas et al., 2003). When this occurs, the coefficient on
876 u_*w^3 in Eq. 5 may have a lower value (Tedford et al., 2014), which translates to a reduced estimate of ϵ and
877 increased α' values. A similar decrease of ϵ can be assumed during heating, when strong stratification ($N >$
878 25 cph) dampens turbulent dissipation (MacIntyre et al., 2010, 2018), however, such stratification was
879 intermittent in our study (Fig. 5f-h).

880
881 Reduced gas transfer velocities and between-lake differences in k_{ch} could also be due to differences in
882 atmospheric forcing. First, the wind speed may have been lower over the lakes than on the Mire due to
883 the slight elevation (<1 m) of the surrounding peatland hummocks (Markfort et al., 2010). The wind-
884 sheltering effect of tall shrubs (*Betula nana* L, Malmer et al., 2005) on the shores of the deeper lakes (Fig.
885 1) was readily noticed during sample collection, particularly in Mellersta Harrsjön. Second, atmospheric
886 stability was different over the three lakes. The atmosphere was stable ($z/L_{MO,a} > 0$) over Mellersta
887 Harrsjön, Inre Harrsjön, and Villasjön during 29%, 21% and 22% of the ice free periods (2009–2017),
888 respectively, with drag coefficients $\sim 16\%$ lower than their neutral value during these times. The effect was
889 more pronounced when winds were light during daytime heating, with somewhat higher frequency during
890 autumn. Colder incoming stream water flowing into Mellersta Harrsjön may have contributed to lower
891 surface water temperatures in this lake (Table 3), with the discrepancy more noticeable as lake level rose
892 (Fig. 5e-h). More frequent periods with a stable atmosphere above Mellersta Harrsjön reduced sensible
893 and latent heat fluxes and are a likely cause of the increased stratification of the surface layer: water at
894 0.1 m was sometimes 0.5 °C to 2 °C warmer than at 0.3 m in Mellersta Harrsjön (5% of the time during ice-
895 free seasons) when temperatures were isothermal in the upper 0.5 m in Villasjön and Inre Harrsjön.
896 Greater near-surface stratification coupled with lower winds than measured on the Mire would have led
897 to the lower values of k and α' obtained in this lake. While this analysis points to the challenges in modelling
898 fluxes when meteorological instrumentation is not situated on the lakes, it also suggests that a solution is
899 to use lower values of α' when modelling k for sheltered water bodies.

900
901 In summary, the model scaling parameter α' computed in this study are lower than the theoretical value
902 of 0.37 and the 0.5 recently obtained in eddy covariance studies in which fluxes were measured with CO₂
903 and modelled. The discrepancy may be explained by surface CH₄ concentrations decreasing due to
904 microbial oxidation over the same timescale as our chamber measurements. Alternate explanations take
905 into account the magnitude of wind shear and degree of sheltering. Differences in α' between lakes
906 indicate the care required in modelling emissions from sheltered lakes; the overall cooler surface water
907 temperatures in the lake with greater stream inflows points to a new control on emissions. That is, when
908 stream inflows lead to surface water temperatures cooler than air temperature in sheltered lakes, a stable
909 atmosphere results which leads to a reduced momentum flux, lower emissions, and a longer time over
910 which methane oxidation can occur. The cooling effect may be especially pronounced in northern
911 landscapes underlain by permafrost, where the temperature of meltwater streams and subsurface flow in
912 the active layer remains low throughout the year. Thus, these comparisons of modelled and measured
913 fluxes point to new areas of research.

914 **5. Summary and conclusions**

915 In this study we combined a unique, multi-year dataset with a modelling approach to better understand
916 environmental controls on turbulence-driven diffusion-limited CH₄ emissions from small, shallow lakes.
917 Floating chambers estimated the seasonal mean flux at 6.9 mg m⁻² d⁻¹ and illustrated how the flux
918 depended on temperature and wind speed. Wind shear controlled the gas transfer velocity while thermal
919 convection and release from storage were minor drivers of the flux. CH₄ fluxes and surface concentrations
920 fitted an Arrhenius-type temperature function ($E_a' = 0.88\text{--}0.97$ eV), suggesting that emissions were
921 strongly coupled to rates of methanogenesis in the sediment. However, temperature was only an accurate
922 proxy of the flux on averaging timescales longer than a month. On shorter timescales wind-induced
923 variability in the gas transfer velocity, mixing layer depth, and storage decoupled production from emission
924 rates. Transient changes in the lake mixing regime allowed for periodic CH₄ accumulation and resulted in
925 an inverse relationship between wind speed and surface concentrations. In this way, the air-water
926 concentration difference acted as a negative feedback to emissions and prevented complete degassing of
927 the lakes, except at high wind speeds ($U_{10} \geq 6.5$ m s⁻¹).

928
929 Freshwater flux studies are increasingly focused on understanding mechanisms and developing proxies for
930 use in upscaling efforts and process-based models. Simple temperature- or wind-based proxies can yield
931 accurate flux estimates, but model parameters, such as E_a' and α' , must be calibrated to local conditions
932 to reflect relevant biotic and abiotic processes at appropriate timescales. Our study highlights the
933 importance of non-linear feedbacks, such as shallow lake degassing at high wind speeds, as well as
934 microbial removal processes and the need to consider the timescale over which fluxes occur relative to
935 the timescale over which CH₄ can be oxidized. Such biological removal processes may violate the
936 fundamental assumption of gas transfer models that all gas measured in the surface mixing layer is emitted
937 to the atmosphere. Advanced gas transfer models can only improve the accuracy of flux estimates if they
938 are paired with observations that capture the meteorological conditions over the lake and the
939 spatiotemporal variability of dissolved gas concentrations. Therefore, field measurements remain
940 necessary to inform, calibrate and validate models. Our results indicate that the timescale of driver
941 variability can inform the frequency of field measurements necessary to yield representative datasets for
942 novel proxy development.

943 **6. Data availability**

944 Data are available at www.bolin.su.se/data/. Surface renewal model code is available by contacting SM.

945

946 **7. Author contributions**

947 JJ, MW and PC designed the study. Fieldwork and laboratory measurements were conducted by JJ, JS and
948 MW. SM developed the surface renewal model code, with contributions from AC. JJ performed the
949 analyses and prepared the manuscript with contributions from BT, PC and SM.

950

951 **8. Competing interests**

952 The authors declare that they have no conflict of interest.

953

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967

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