



# The carbon footprint of a Malaysian tropical reservoir: measured versus modeled estimates highlight the underestimated key role of downstream processes

Cynthia Soued<sup>1</sup>, Yves T. Prairie<sup>1</sup>

5 <sup>1</sup>Groupe de Recherche Interuniversitaire en Limnologie et en Environnement Aquatique (GRIL), Département des Sciences Biologiques, Université du Québec à Montréal, Montréal, H2X 3X8, Canada.

*Correspondence to:* Cynthia Soued (cynthia.soued@gmail.com)

**Abstract.** Reservoirs are important sources of greenhouse gases (GHG) to the atmosphere and their number is rapidly increasing, especially in tropical regions. Accurately predicting their current and future emissions is essential but hindered by fragmented data on the subject, which often fail to include all emission pathways (diffusion, ebullition, degassing, and downstream emissions) and the high spatial and temporal flux variability. Here we conducted a comprehensive sampling of Batang Ai reservoir (Malaysia), and compared field-based versus modeled estimates of its annual carbon footprint for each emission pathway. Carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>) diffusive fluxes were higher in upstream reaches. Reducing spatial and temporal sampling resolution resulted in up to 64 and 28 % change in flux estimate respectively. Most GHGs present in discharged water were degassed at the turbines, and the remainder were gradually emitted along the outflow river, leaving time for CH<sub>4</sub> to be partly oxidized to CO<sub>2</sub>. Overall, the reservoir emitted 2 639 gCO<sub>2</sub>eq m<sup>-2</sup> yr<sup>-1</sup>, with 90 % occurring downstream of the dam, mostly in the form of CH<sub>4</sub>. These emissions, largely underestimated by predictions, are mitigated by CH<sub>4</sub> oxidation upstream and downstream of the dam, but could have been drastically reduced by slightly raising the water intake elevation depth. CO<sub>2</sub> diffusion and CH<sub>4</sub> ebullition were lower than predicted, whereas modeled CH<sub>4</sub> diffusion was accurate. Investigating latter discrepancies, we conclude that exploring morphometry, soil type, and stratification patterns as predictors can improve modeling of reservoir GHG emissions at local and global scales.

## 1 Introduction

Reservoirs provide a variety of services to humans (water supply, navigation, flood control, hydropower) and cover an estimated area exceeding 0.3 million km<sup>2</sup> globally (Lehner et al., 2011). This area is increasing, with an expected rapid growth of the hydroelectric sector in the next two decades (International Hydropower Association (IHA), 2015), mainly in tropical and subtropical regions (Zarfl et al., 2015). The flooding of terrestrial landscapes can change its carbon (C) balance into significant greenhouse gas (GHG) sources to the atmosphere (Prairie et al., 2018; Rudd et al., 1993; Teodoru et al., 2012). While part of reservoir GHG emissions would occur naturally (not legitimately attributable to damming), the remainder results from newly created environments favoring C mineralization, particularly methane (CH<sub>4</sub>) production (flooded organic-rich anoxic soils) (Prairie et al., 2018). Field studies have revealed a wide range in measured fluxes, with spatial and temporal variability sometime spanning several orders of magnitude within a single reservoir (Paranaíba et al.,



2018; Sherman and Ford, 2011). Moreover, reservoirs can emit GHG through several pathways: surface diffusion at the air-water interface, ebullition from the sediments, and for some reservoirs (mostly hydroelectric) through degassing upon water discharge and downstream emissions of excess gas in the outflow river. The relative contribution of these flux pathways to total emissions is extremely variable. While diffusion is the most frequently sampled, it is often not the main emission pathway (Demarty and Bastien, 2011). Indeed, measured ebullition, degassing, and downstream emissions range from negligible to several order of magnitude higher than diffusion in different reservoirs (Bastien and Demarty, 2013; DelSontro et al., 2010; Galy-Lacaux et al., 1997; Keller and Stallard, 1994; Kemenes et al., 2007; Teodoru et al., 2012; Venkiteswaran et al., 2013), making it a challenge to model total reservoirs GHG emissions.

Literature syntheses over the past 20 years have yielded highly variable global estimates of reservoirs GHG footprint, ranging from 0.5 to 2.3 PgCO<sub>2</sub>eq.yr<sup>-1</sup> (Barros et al., 2011; Bastviken et al., 2011; Deemer et al., 2016; St. Louis et al., 2000). These estimates are based on global extrapolations of averages of sampled systems, representing an uneven spatial distribution biased toward North America and Europe, and an uneven mixture of emission pathways. Recent studies have highlighted the lack of spatial and temporal resolution as well as the frequent absence of some flux pathways (especially degassing, downstream, and N<sub>2</sub>O emissions) in most reservoir GHG assessments (Beaulieu et al., 2016; Deemer et al., 2016). More recently, studies have focused on identifying drivers of reservoir GHG flux variability. Using global empirical data, Barros et al. (2011) proposed the first quantitative models for reservoir carbon dioxide (CO<sub>2</sub>) and CH<sub>4</sub> surface diffusion as a negative function of reservoir age, latitude, and mean depth (for CO<sub>2</sub> only), and a positive function of dissolved organic carbon (DOC) inputs (Barros et al., 2011). An online tool (G-res) for predicting reservoir C emissions was later developed on the basis of a similar empirical modeling approach of measured reservoir fluxes with globally available environmental data (UNESCO/IHA, 2017). Modeling frameworks to predict GHG emissions from existing and future reservoirs are essential tools for reservoir management. However, their accuracy is directly related to available information and inherently affected by gaps and biases of the published literature. For example, while the G-res model predicts reservoir CO<sub>2</sub> and CH<sub>4</sub> diffusion as well as CH<sub>4</sub> ebullition and degassing on the basis of temperature, age, % littoral zone and soil organic C, it does not consider N<sub>2</sub>O emissions, CO<sub>2</sub> degassing, and emissions in the downstream river due to scarcity of data. Overall, the paucity of comprehensive empirical studies limits our knowledge of reservoir GHG dynamics at a local scale, introducing uncertainties in large scale estimates and hindering model development.

The research reported here focusses on building a comprehensive assessment of GHG fluxes of Batang Ai, a tropical reservoir in south-east Asia (Malaysia), over four sampling campaigns spanning two years with an extensive spatial coverage. The main objective of this study is to provide a comprehensive account of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes from diffusion, ebullition, degassing, and downstream river (accounting for riverine CH<sub>4</sub> oxidation) to better understand what shapes their relative contributions and their potential mitigation. The second objective is to compare our measured values with modeled estimates from each pathway and gas species to locate where the largest discrepancies are, and thereby identify research avenues for improving the current modeling framework.



## 2 Materials and methods

### 2.1 Study site and sampling campaigns

Batang Ai is a hydroelectric reservoir located on the Borneo Island in the Sarawak province of Malaysia (latitude 1.16° and longitude 111.9°). The regional climate is tropical equatorial with a relatively constant temperature throughout the year, on average 23°C in the morning to 32°C during the day. Annual rainfall varies between 3300 and 4600 mm with two monsoon seasons: November to February (northeast monsoon), and June to October (southwest) (Sarawak Government, 2019). Batang Ai reservoir was impounded in 1985 with no prior clearing of the vegetation, and has a dam wall of 85 m in height, a mean depth of 34 m, and a total area of 68.4 km<sup>2</sup>. The reservoir catchment consists of 1149 km<sup>2</sup> of mostly forested land where human activities are limited to a few traditional habitations and associated croplands, and localized aquaculture sites within the reservoir main basin. The reservoir has two major inflows: the Batang Ai and Engkari rivers, which flow into two reservoir branches merging upstream of the main reservoir basin (Figure 1). Four sampling campaigns were conducted: 1) November 14<sup>th</sup> to December 5<sup>th</sup> 2016 (Nov-Dec 2016), 2) April 19<sup>th</sup> to May 3<sup>th</sup> 2017 (Apr-May 2017), 3) February 28<sup>th</sup> to March 13<sup>th</sup> 2018 (Feb-Mar 2018), and 4) August 12<sup>th</sup> to 29<sup>th</sup> 2018 (Aug 2018).

### 2.2 Water chemistry

Surface concentrations of DOC, total phosphorus (TP), total nitrogen (TN), and chlorophyll a (Chla) were measured at all surface diffusion sampling sites (Figure 1) and during each campaign. For TP and TN, we collected non-filtered water in acid-washed glass vials stored at 4°C until analysis. TP was measured by spectrophotometry using the standard molybdenum blue method after persulphate digestion. TN analyses were performed by alkaline persulphate digestion to NO<sub>3</sub>, subsequently measured on an Alpkem Flow Solution IV autoanalyser. Water filtered at 0.45 μm was used for DOC analysis with a Total Organic Carbon analyser 1010-OI following sodium persulphate digestion. Chla was analysed through spectrophotometry following filtration on Whatman (GF/F) filters and extraction by hot 90 % ethanol solution.

### 2.3 Surface diffusive flux

Surface diffusion of CO<sub>2</sub> and CH<sub>4</sub> to the atmosphere were measured at 36 sites in the reservoir and 3 sites in the inflow rivers (Figure 1), and sampling of the same sites was repeated each campaign (with a few exceptions). Fluxes were measured using a static air tight floating chamber connected in a closed loop to an Ultraportable gas Analyser (UGGA from LGR). Diffusive flux rates ( $F_{\text{gas}}$ ) were derived from the linear change in CO<sub>2</sub> and CH<sub>4</sub> partial pressures (continuously monitored at 1 Hz for a minimum of 5 min) through time inside the chamber using the following Eq. (1):

$$F_{\text{gas}} = \frac{sV}{mV s}, \quad (1)$$



95 Where  $s$  is the gas accumulation rate in the chamber,  $V = 25$  L the chamber volume,  $S = 0.184$  m<sup>2</sup> the chamber surface area, and  $mV$  the gas molar volume at current atmospheric pressure.

N<sub>2</sub>O diffusive flux was estimated at 7 of the sampled sites (Figure 1) using the following Eq. (2) (Lide, 2005):

$$F_{\text{gas}} = k_{\text{gas}} (C_{\text{gas}} - C_{\text{eq}}), \quad (2)$$

100 where  $k_{\text{gas}}$  is the gas exchange coefficient (m d<sup>-1</sup>),  $C_{\text{gas}}$  is the gas concentration in the water and  $C_{\text{eq}}$  is the theoretical gas concentration at equilibrium given measured water temperature, atmospheric pressure and ambient gas concentration.  $C_{\text{N}_2\text{O}}$  was measured using the headspace technique, with a 1.12 L sealed glass serum bottle containing surface water and a 0.12 L headspace of ambient air. After shaking the bottle for two minutes to achieve air-water equilibrium, the headspace gas was extracted from the bottle with an airtight syringe and injected in previously evacuated 9 mL glass vial capped with an air tight butyl stopper and aluminium seal. Three analytical replicates and a local sample of ambient air were taken at each site  
105 and analysed by gas chromatography using a Shimadzu GC-2040, with a Poropak Q column to separate gases and an ECD detector.  $k_{\text{N}_2\text{O}}$  was derived from the measured  $k_{\text{CH}_4}$  using the following Eq. (3) (Cole and Caraco, 1998; Ledwell, 1984):

$$k_{\text{N}_2\text{O}} = \left( \frac{Sc_{\text{N}_2\text{O}}}{Sc_{\text{CH}_4}} \right)^{-0.67} k_{\text{CH}_4}, \quad (3)$$

110 where  $Sc$  is the gas Schmidt number (Wanninkhof, 1992),  $k_{\text{CH}_4}$  was calculated by rearranging Eq. (2) for CH<sub>4</sub>, with known values of  $F_{\text{gas}}$ ,  $C_{\text{gas}}$ , and  $C_{\text{eq}}$ . CH<sub>4</sub> and CO<sub>2</sub> concentrations in the water were measured using the headspace technique. Surface water was collected in a 60 mL gas-tight plastic syringe in which a 30 mL headspace was created (using either ambient air or carbon free air). The syringe was shaken for 2 min to achieve air-water gas equilibrium. The gas phase was then injected in a 12 mL air-tight pre-evacuated vial and subsequently analysed on a Shimadzu GC-8A Gas chromatograph with flame ionization detector. The samples were also analyzed for isotopic  $\delta^{13}\text{CO}_2$  and  $\delta^{13}\text{CH}_4$  signatures using a Cavity Ring Down Spectrometer (CRDS) equipped with a Small Sample Isotopic Module (SSIM, Picarro G2201 -i, Picarro Inc).

#### 115 2.4 Ebullition flux

Sediment gas ebullition was measured at four sites in the reservoir and two sites in the inflows (Figure1) by deploying 0.785 m<sup>2</sup> underwater inverted funnel traps at 2 to 3 m deep for approximately 20 days in the reservoir and 1h in the inflows. The top part of a closed plastic syringe was fixed to the narrow end of the funnel trap where the emerging bubbles accumulated. Upon recovery, bubble gas volume was measured, collected from the syringe, and injected in 12 mL pre-evacuated air tight  
120 vials for CO<sub>2</sub> and CH<sub>4</sub> concentration analyses (using the aforementioned method). Ebullition rate was calculated assuming the original bubble composition was similar to bubbles collected almost right after ascent in the inflows sites, which was 100 % CH<sub>4</sub>. Hence we considered CO<sub>2</sub> and N<sub>2</sub>O ebullition to be null.

In order to estimate the potential for sediment accumulation fueling ebullition in the littoral zone, we calculated the mud energy boundary depth (EBD, below which fine grained sediments accumulation occurs) using the reservoir surface area (E  
125 in km<sup>2</sup>) as the exposure parameter in the following Eq. (4) (Rowan et al., 1992):



$$EBD = 2.685 E^{0.305}, \quad (4)$$

## 2.5 Downstream emissions and CH<sub>4</sub> oxidation

Downstream emissions of CO<sub>2</sub> and CH<sub>4</sub> were calculated using the following Eq. (5):

$$F_d = Q (C_{up} - C_{base}), \quad (5)$$

130 Where  $F_d$  is the flux to the atmosphere,  $Q$  is the water discharge,  $C_{up}$  is the gas concentration upstream of the dam at the  
water withdrawal depth, and  $C_{base}$  is the gas concentration considered as a natural baseline in the outflow river (previous to  
reservoir construction). Since no measurements of pre-impoundment gas concentration are available, we considered two  
values as upper and lower bonds for  $C_{base}$ : 1) the gas concentration at the surface of the reservoir right upstream of the dam,  
and 2) the gas concentration in the outflow 19 km downstream of the dam. Degassing emissions (gas emitted right after  
135 water discharge due to pressure release) were calculated with Eq. (5) replacing  $C_{base}$  by the gas concentration measured at the  
powerhouse right after water discharge. To define  $C_{up}$ , we measured in each campaign, CO<sub>2</sub> and CH<sub>4</sub> concentrations in a  
vertical profile right upstream of the dam at each one to three meters from 0 up to 32 m using a multi-parameter probe  
equipped with depth, oxygen, and temperature sensors (Yellow Spring Instruments, YSI model 600XLM-M) attached to a 12  
Volt submersible Tornado pump (Proactive Environmental Products) for water collection. Gas concentration and  $\delta^{13}C$  were  
140 measured as previously described. The water withdrawal depth ranged from 20 to 23 m and was estimated based on known  
values of water intake and water level elevations compared to sea level.  $C_{up}$  was defined as the average measured gas  
concentrations in the range of  $\pm 1$  m of the withdrawal depth. Gas concentration was measured at four locations in the  
outflow, at 0 (power house), 0.6, 2.7, and 19 km downstream of the dam.

Estimates of downstream CH<sub>4</sub> oxidation were obtained, for each sampling campaign, by calculating the fraction of CH<sub>4</sub>  
145 oxidized ( $F_{ox}$ ) using the following Eq. (6):

$$F_{ox} = \frac{-(\ln(\delta^{13}CH_{4resid}+1000)-\ln(\delta^{13}CH_{4source}+1000))\left(1-\frac{[CH_4]_{resid}}{[CH_4]_{source}}\right)}{\left(1-\frac{1}{\alpha}\right)\ln\left(\frac{[CH_4]_{resid}}{[CH_4]_{source}}\right)}, \quad (6)$$

Eq. (6) is based on a non-steady state isotopic model developed considering evasion (emission to the atmosphere) and  
oxidation as the two loss processes for CH<sub>4</sub> in the outflow river, assuming negligible isotopic fractionation for evasion  
(Knox et al., 1992) and a fractionation of  $\alpha = 1.02$  for oxidation (Coleman et al., 1981) (see derivation in Supplementary  
150 Information).  $[CH_4]_{source}$ ,  $[CH_4]_{resid}$ ,  $\delta^{13}CH_{4source}$ , and  $\delta^{13}CH_{4resid}$  are the concentrations of CH<sub>4</sub> and their corresponding  
isotopic signatures at the beginning of the outflow (km 0) and 19 km downstream, representing the source and residual pools  
of CH<sub>4</sub> respectively. The amount of CH<sub>4</sub> oxidized to CO<sub>2</sub> along the 19 km of river stretch for each sampling campaign was  
calculated as the product of  $F_{ox}$  and  $[CH_4]_{source}$ . Note that downstream N<sub>2</sub>O emissions were considered null since N<sub>2</sub>O  
145 concentrations measured in the deep reservoir layer were lower than concentrations in the outflow.



## 155 2.6 Ecosystem scale C footprint

Batang Ai annual C footprint was calculated as the sum of surface diffusion, ebullition, and downstream emissions of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O considering a greenhouse warming potential of 1, 34, and 298 respectively over a 100 years lifetime period (Myhre et al., 2013). For each flux pathway, annual flux was estimated as the average of the sampling campaigns. Ecosystem scale estimate of surface diffusion was calculated for each campaign as the average of measured flux rates applied to the reservoir area for N<sub>2</sub>O, and for CO<sub>2</sub> and CH<sub>4</sub> it was obtained by spatial interpolation of measured fluxes over the reservoir area based on inverse distance weighting with a power of two (a power of one yields similar averages, CV < 11 %) using package gstat version 1.1-6 in the R version 3.4.1 software (Pebesma, 2004; R Core Team, 2017). Ebullition at the reservoir scale was calculated as the average of measured reservoir ebullition rates applied to the littoral area (< 3 m deep). The estimated C emissions of Batang Ai based on measured data was compared to values derived from the G-res model (UNESCO/IHA, 2017) and the model presented in Barros et al. (2011). Details on models equations and input variables are presented in the Supplementary Information (Table S2 and S3).

## 3 Results and discussion

### 3.1 Water chemistry

The reservoir is stratified throughout the year with a thermocline around 13 m and mostly anoxic conditions in the hypolimnion of the main basin (Figure S1). The system is oligotrophic, with very low concentrations of DOC, TP, TN, and Chla averaging 0.9 mg L<sup>-1</sup>, 5.9, 0.11, and 1.3 µg L<sup>-1</sup> respectively (Table S1), and high water transparency (Secchi depth > 5 m). In the reservoir inflows, concentrations of measured chemical species are slightly higher but still in the oligotrophic range (Table S1), however the transparency is much lower due to turbidity (Secchi < 0.5).

### 3.2 Surface diffusion

Measured CO<sub>2</sub> diffusion in the reservoir averaged 7.7 mmol m<sup>-2</sup> d<sup>-1</sup> (Table S1), which is on the low end compared to other reservoirs (Deemer et al., 2016) and even to natural lakes (Sobek et al., 2005), but similar to CO<sub>2</sub> fluxes measured in two reservoirs in Laos (Chanudet et al., 2011). CO<sub>2</sub> diffusion across all sites ranged from substantial uptake to high emissions (from -30.8 to 593.9 mmol m<sup>-2</sup> d<sup>-1</sup>, Table S1) reflecting a large spatial and temporal variability. Spatially, CO<sub>2</sub> fluxes measured in the main basin and branches had similar averages of 7.9 and 7.3 mmol m<sup>-2</sup> d<sup>-1</sup> respectively, contrasting with higher and more variable values in the inflows (mean of 137.3 mmol.m<sup>-2</sup>.d<sup>-1</sup>, Figures 2). Within the reservoir, CO<sub>2</sub> fluxes were variable (SD ± 18.2 mmol m<sup>-2</sup> d<sup>-1</sup>) but did not follow a consistent pattern, and might reflect pre-flooding landscape heterogeneity (Teodoru et al., 2011). Temporally, highest average reservoir CO<sub>2</sub> fluxes were measured in Apr-May 2017, when no CO<sub>2</sub> uptake was observed, contrary to other campaigns, especially Feb-Mar and Aug 2018, when CO<sub>2</sub> uptake was



185 common (Figure S2) and average Chla concentrations were the highest. This reflects the important role of metabolism (namely CO<sub>2</sub> consumption by primary production) in modulating surface CO<sub>2</sub> fluxes in Batang Ai.

All measured CH<sub>4</sub> diffusive fluxes were positive and ranged from 0.03 to 113.4 mmol m<sup>-2</sup> d<sup>-1</sup> (Table S1). Spatially, CH<sub>4</sub> fluxes were progressively higher moving further upstream (Figure 2 and S3) with decreasing water depth and increasing connection to the littoral. This gradient in morphometry induces an increasingly greater contact of the water with bottom and littoral sediments, where CH<sub>4</sub> is produced, explaining the spatial pattern of CH<sub>4</sub> fluxes. CH<sub>4</sub> diffusive fluxes also varied 190 temporally, but to a lesser extent than CO<sub>2</sub>, being on average highest in Aug 2018 in the reservoir and in Nov-Dec 2016 in the inflows.

Reservoir N<sub>2</sub>O diffusive fluxes (measured with a limited spatial resolution) averaged -0.2 (SD ± 2.1) nmol m<sup>-2</sup> d<sup>-1</sup> (Table S1). The negative value indicates that the system acts as a slight net sink of N<sub>2</sub>O, absorbing an estimated 2.1 gCO<sub>2</sub>eq m<sup>-2</sup> yr<sup>-1</sup> (Table 2). Atmospheric N<sub>2</sub>O uptake have previously been reported in aquatic systems and linked to low oxygen and nitrogen 195 content conducive to complete denitrification which consumes N<sub>2</sub>O (Soued et al., 2016; Webb et al., 2019). These environmental conditions match observations in Batang Ai, with a low average TN concentration of 0.11 mg L<sup>-1</sup> (Table S1) and anoxic deep waters (Figure S1).

### 3.3 Ebullition

We calculated that CH<sub>4</sub> ebullition rates in Batang Ai's littoral area ranged from 0.02 to 0.84 mmol m<sup>-2</sup> d<sup>-1</sup>, which contrasts 200 with rates measured in its inflows that are several orders of magnitude higher (52 and 103 mmol m<sup>-2</sup> d<sup>-1</sup>). Similar patterns were observed in other reservoirs, where inflow arms where bubbling hot spots due to a higher organic C supply driven by terrestrial matter deposition (DeSontro et al., 2011; Grinham et al., 2018). Since ebullition rates are notoriously heterogeneous and were measured at only four sites in the reservoir, they may not reflect ecosystem-scale rates. However, our attempt to manually provoke ebullition at several other sites (by physically disturbing the sediments) did not result in any 205 bubble release, confirming the low potential for ebullition in the reservoir littoral zone. Moreover, we calculated that fine grained sediment accumulation is unlikely at depths shallower than 9.7 m (estimated EBD) in Batang Ai. This, combined with the reservoir steep slope, prevents the sustained accumulation of organic material in littoral zones (Blais and Kalff, 1995), hence decreasing the potential for CH<sub>4</sub> production and bubbling there. Also, apparent littoral sediment composition in the reservoir; dense clay with low porosity, may further hinder bubble formation and emission (de Mello et al., 2018).

### 210 3.4 Degassing and downstream emissions

Emissions downstream of the dam, expressed on a reservoir-wide areal basis, ranged from 19.3 to 38.6 mmol m<sup>-2</sup> d<sup>-1</sup> for CO<sub>2</sub> and from 5.9 to 13.9 mmol m<sup>-2</sup> d<sup>-1</sup> for CH<sub>4</sub> (Table 1). The amount of CO<sub>2</sub> exiting the reservoir varied little between sampling campaigns (CV = 3 %) contrary to CH<sub>4</sub> (CV = 28 %, Table 1 and Figure 3). Higher temporal variability of CH<sub>4</sub> concentration in discharged water is likely modulated by microbial CH<sub>4</sub> oxidation in the reservoir water column upstream of 215 the dam. Evidence of high CH<sub>4</sub> oxidation are apparent in reservoir water column profiles, showing a sharp decline of CH<sub>4</sub>



concentration and increase of  $\delta^{13}\text{CH}_4$  right around the water withdrawal depth (Figure S1). This vertical pattern results from higher oxygen availability when moving up in the hypolimnion (Figure S1), promoting  $\text{CH}_4$  oxidation at shallower depths. Once GHGs have exited the reservoir, a large fraction (40 and 65 % for  $\text{CO}_2$  and  $\text{CH}_4$  respectively) is immediately lost to the atmosphere as degassing emissions (Table 1), which is in line with previous literature reports (Kemenes et al., 2016). Along the outflow river,  $\text{CO}_2$  and  $\text{CH}_4$  concentrations gradually decreased,  $\delta^{13}\text{CO}_2$  remained stable, whereas  $\delta^{13}\text{CH}_4$  steadily increased indicating riverine  $\text{CH}_4$  oxidation (Figure 3). We estimated that riverine  $\text{CH}_4$  oxidation ranged from 0.38 to 1.80  $\text{mmol m}^{-2} \text{d}^{-1}$  (expressed per  $\text{m}^2$  of reservoir area for comparison), transforming 18 to 32 % (depending on the sampling campaign) of the  $\text{CH}_4$  to  $\text{CO}_2$  within the first 19 km of the outflow. Riverine oxidation rates did not co-vary temporally with water temperature, oxygen availability, or  $\text{CH}_4$  concentrations (known as typical drivers (Thottathil et al., 2019)), hence they might be regulated by other factors like light and microbial assemblages (Murase and Sugimoto, 2005; Oswald et al., 2015). Overall, riverine oxidation of  $\text{CH}_4$  to  $\text{CO}_2$  (which has a 34 times lower warming potential) reduced radiative forcing of downstream emissions (excluding degassing) by, on average, 18 %, and the total annual reservoir C footprint by 6 %. Despite having a measurable impact on reservoir C emissions,  $\text{CH}_4$  oxidation downstream of dams was only estimated in two other reservoirs to our knowledge (Guérin and Abril, 2007; Kemenes et al., 2007). Accounting for this process is particularly important in systems where downstream emissions are large, a common situation in tropical reservoirs (Demarty and Bastien, 2011). While additional data on the subject is needed, our results provide one of the first basis for understanding  $\text{CH}_4$  oxidation downstream of dams, and eventually integrating this component to global models (from which it is currently absent).

### 3.5 Importance of sampling resolution

High spatial and temporal sampling resolution have been recently highlighted as an important but often lacking aspect of reservoir C footprint assessments (Deemer et al., 2016; Paranaíba et al., 2018). Reservoir scale fluxes are usually derived from applying an average of limited flux measurements to the entire reservoir area. For Batang Ai, this method overestimates by 14 % ( $130 \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ ) and 64 % ( $251 \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ )  $\text{CO}_2$  and  $\text{CH}_4$  diffusive fluxes respectively compared to spatial interpolation. This is due to the effect of extreme values that are very constrained in space but have a disproportionate effect on the overall flux average. Also, reducing temporal sampling resolution to one campaign instead of four changes the reservoir C footprint estimate by up to 28 %. An additional source of uncertainty in reservoir flux estimates is the definition of a baseline value representing natural river emissions in order to calculate downstream emissions of excess gas in the outflow attributable to damming. In Batang Ai, downstream emission was estimated on the basis of a potential range in natural baseline emissions, however, measured values of GHG concentrations in the pre-impounded river would have substantially reduced the estimate uncertainty. Results from Batang Ai reinforce the importance of pre and post-impoundment sampling resolution and upscaling methods in annual reservoir-scale GHG flux estimates.



### 3.6 Reservoir C footprint and potential mitigation

Most of Batang Ai emissions occur downstream of the dam through degassing (60.2 %) and from the outflow river surface (29.7 %), while surface diffusive fluxes contributed only 10.0 %, and ebullition 0.13 % (Table 2). In all pathways, CH<sub>4</sub> fluxes were higher than CO<sub>2</sub> and N<sub>2</sub>O (especially for degassing), accounting for 76.7 % of Batang Ai CO<sub>2</sub>eq emissions. This distribution of the flux can be attributed mostly to the accumulation of large quantities of CH<sub>4</sub> in the hypolimnion, combined with the fact that the withdrawal depth is located within this layer, allowing the accumulated gas to escape to the atmosphere. Previous studies on reservoirs with similar characteristics to Batang Ai (tropical climate with a permanent thermal stratification and deep water withdrawal) have also found degassing and downstream emissions to be the major emission pathways, especially for CH<sub>4</sub> (Galy-Lacaux et al., 1997; Kemenes et al., 2007).

Overall, we estimated that the reservoir emits on average  $2\,639 (\pm 651) \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$  which corresponds to  $0.181 \text{ TgCO}_2\text{eq yr}^{-1}$  over the whole system. In comparison, the annual areal emission rate (diffusion and ebullition) of the inflows, based on a more limited sampling resolution, is estimated to range from 10.8 to 52.5 kgCO<sub>2</sub>eq m<sup>-2</sup> yr<sup>-1</sup>, mainly due to extremely high ebullition. When applied to the approximated surface area of the river before impoundment (1.52 km<sup>2</sup>), this rate translates to 0.016 – 0.080 TgCO<sub>2</sub>eq (Table 2), assuming a similar flux rates in the current inflows and pre-impoundment river. While the emission rate of the river per unit of area is an order of magnitude higher than for the reservoir, its estimated total flux remains 2.3 to 11.3 times lower due to a much smaller surface. Higher riverine emissions rates are probably due to a shallower depth and higher inputs of terrestrial organic matter, both conducive to CO<sub>2</sub> and CH<sub>4</sub> production and ebullition. Changing the landscape hydrology to a reservoir drastically reduced areal flux rates, especially ebullition, however, it widely expanded the volume of anoxic environments (sediments and hypolimnion), creating a vast new space for CH<sub>4</sub> production. The new hydrological regime also created an opportunity for the large quantities of gas produced in deep layers to easily escape to the atmosphere through the outflow and downstream river.

One way to reduce reservoir emissions is to ensure low GHG concentrations at the turbine water withdrawal depth. In Batang Ai, maximum CO<sub>2</sub> and CH<sub>4</sub> concentrations are found in the reservoir deep layers, and rapidly decrease from 20 to 10 m for CO<sub>2</sub> and from 25 to 15 m for CH<sub>4</sub> (Figure S1). This pattern is commonly found in lakes and reservoirs and results from thermal stratification and biological processes (aerobic respiration and CH<sub>4</sub> oxidation). Knowing this concentration profile, degassing and downstream emissions could have been reduced in Batang Ai by elevating the water withdrawal depth to avoid hypolimnetic gas release. We calculated that elevating the water withdrawal depth by 1, 3, and 5 m would result in a reduction of degassing and downstream emissions by 1, 13, and 22 % for CO<sub>2</sub> and by 27, 89, and 98 % for CH<sub>4</sub>, respectively (Figure S4). Consequently, a minor change in the dam design could have drastically reduced Batang Ai's C footprint. This should be taken in consideration in future reservoir construction, especially in tropical regions.



### 3.7 Measured versus modeled fluxes

Based on measurements, Batang Ai emits on average  $113 (\pm 15) \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$  via surface  $\text{CO}_2$  diffusion. This value is 41 times lower than predicted by Barros et al. model ( $4671 \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ , Table 2) based on reservoir age, DOC inputs  
280 (derived from DOC water concentration), and latitude. The high predicted value for Batang Ai, being a relatively old reservoir with very low DOC concentration, is mainly driven by its low latitude. While reservoirs in low latitudes globally have higher average  $\text{CO}_2$  fluxes due to higher temperature and often dense flooded biomass (Barros et al., 2011; St. Louis et al., 2000), our results provide a clear example that not all tropical reservoirs have high  $\text{CO}_2$  emissions by simple virtue of their geographical location. Despite high temperature, Batang Ai's very low water organic matter content (Table S1) offers  
285 little substrate for net heterotrophy, and its strong permanent stratification creates a physical barrier potentially retaining  $\text{CO}_2$  derived from flooded biomass in the hypolimnion.

In comparison, the G-res model predicts a diffusive  $\text{CO}_2$  flux of  $577 (509\text{-}655) \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ , which includes the flux naturally sustained by catchment carbon inputs ( $397 \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ , predicted flux 100 years after flooding) and the flux derived from organic matter flooding ( $180 \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ ). While the predicted G-res value is much closer than that  
290 predicted from the Barros et al. model, it still overestimates measured flux, mostly the natural baseline (catchment derived) part of it. The G-res predicts baseline  $\text{CO}_2$  effluxes as a function of soil C content, a proxy for C input to the reservoir. While Batang Ai soil is rich in organic C ( $\sim 50 \text{ g kg}^{-1}$ ), it also has a high clay content ( $> 40 \%$ ) (ISRIC - World Soil Information, 2019; Wasli et al., 2011) which is known to bind with organic matter and reduce its leaching to the aquatic environment (Oades, 1988). This may explain the unusually low DOC concentration in the reservoir and its inflows ( $0.3$  to  $1.8 \text{ mg L}^{-1}$ ,  
295 Table S1) that are among the lowest reported in freshwaters globally (Sobek et al., 2007). Had G-res been able to capture the baseline emissions more correctly in Batang Ai (close to zero given the very low DOC inputs), predictions would have nearly matched observations. Finally, the G-res model is not suitable to predict  $\text{CO}_2$  uptake, which was observed in 32 % of flux measurements in Batang Ai due to an occasionally net autotrophic surface metabolism favored under low C inputs (Bogard and del Giorgio, 2016). The modeling framework of reservoirs diffusive  $\text{CO}_2$  flux would benefit from a better  
300 predictive capacity of internal metabolism and catchment C inputs in a diverse range of landscape properties.

Our field-based estimate of Batang Ai  $\text{CH}_4$  surface diffusion is  $153 (\pm 43) \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$ , which differs by only 5 % and 15 % from the G-res and Barros et al. modeled predictions of 161 (132-197) and  $176 \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$  respectively (Table 2). Both models use as predictors age, a proxy for water temperature (air temperature or latitude), and an indicator of reservoir morphometry (% littoral area or mean depth), and Barros et al. also uses DOC input (Table S3). Similar predictors were  
305 identified in a recent global literature analysis, which also pointed out the role of trophic state in  $\text{CH}_4$  diffusion, with Batang Ai falling well in the range of flux reported in other oligotrophic reservoirs (Deemer et al., 2016). Overall, our results show that global modelling frameworks for  $\text{CH}_4$  diffusive emissions capture reasonably well the reality of Batang Ai.

Measured estimate of reservoir-scale  $\text{CH}_4$  ebullition averaged  $3.4 (\pm 1.9) \text{ gCO}_2\text{eq m}^{-2} \text{ yr}^{-1}$  (Table 2), which is one of the lowest reported globally in reservoirs (Deemer et al., 2016), and is an order of magnitude lower than the 52 (32 - 83)  $\text{gCO}_2\text{eq}$



310  $\text{m}^{-2} \text{yr}^{-1}$  predicted by the G-res model (the only available model for reservoir ebullition). This contrasts with the perception  
that tropical reservoirs consistently have high ebullitive emissions, and support the idea that the supply of sediment organic  
matter, rather than temperature, is the primary driver of ebullition (Grinham et al., 2018). Batang Ai sediment properties and  
focusing patterns mentioned earlier could explain the model overestimation of  $\text{CH}_4$  ebullition. The G-res model considers the  
fraction of littoral area and horizontal radiance (a proxy for heat input) as predictors of ebullition rate, but does not integrate  
315 other catchment properties. Building a stronger mechanistic understanding of the effect of sediment composition and  
accumulation patterns on  $\text{CH}_4$  bubbling may improve our ability to more accurately predict reservoir ebullition flux.  
Our empirical estimate shows that  $503 (\pm 66)$  and  $1869 (\pm 518)$   $\text{gCO}_2\text{eq m}^{-2} \text{yr}^{-1}$  are emitted as  $\text{CO}_2$  and  $\text{CH}_4$  respectively  
downstream of the dam (including degassing), accounting for 90 % of Batang Ai C emission (Table 2). Currently there are  
no available model predicting downstream C emissions from reservoirs, except the G-res model which is able to predict only  
320 the  $\text{CH}_4$  degassing part of this flux. Modeled  $\text{CH}_4$  degassing in Batang Ai is  $468 (266-832)$   $\text{gCO}_2\text{eq m}^{-2} \text{yr}^{-1}$  compared to an  
estimated  $1342 (\pm 372)$   $\text{gCO}_2\text{eq m}^{-2} \text{yr}^{-1}$  based on our measurements. Predictive variables used to model  $\text{CH}_4$  degassing are  
modeled  $\text{CH}_4$  diffusive fluxes (based on % littoral area and temperature) and water retention time (Table S3). In Batang Ai,  
the strong and permanent stratification favors oxygen depletion in the hypolimnion which promotes deep  $\text{CH}_4$  accumulation  
combined with a decoupling between surface and deep water layers. The model relies strongly on surface  $\text{CH}_4$  patterns to  
325 predict excess  $\text{CH}_4$  in the deep layer, which could explain why it underestimates  $\text{CH}_4$  degassing in Batang Ai. Improving this  
aspect of the model requires a better capacity to predict  $\text{CH}_4$  accumulation in deep reservoir layers based on globally  
available data. In this regard, our results suggest that the role of stratification is a promising key element to investigate.

### 3.8 Conclusions

The comprehensive GHG portrait of Batang Ai highlights the importance of spatial and temporal sampling resolution and the  
330 inclusion of all flux components in reservoir GHG assessments. Gas dynamics downstream of the dam (degassing, outflow  
emissions and  $\text{CH}_4$  oxidation), commonly not assessed in reservoir GHG studies, are major elements in Batang Ai. We  
suggest that these emissions could have been greatly diminished with a minor change to the dam design (withdrawal depth  
decrease). Mitigating GHG emissions from future reservoirs depends on the capacity to predict GHG fluxes from all  
pathways. In this regard, the comparison between Batang Ai measured and modeled GHG flux estimates allowed us to  
335 identify knowledge gaps based on which we propose the four following research avenues. 1) Refine the modeling of  
reservoir  $\text{CO}_2$  diffusion by studying its link with metabolism and organic matter leaching from different soil types. 2)  
Examine the potential for  $\text{CH}_4$  ebullition in littoral zones in relation to patterns of organic matter sedimentation linked to  
morphometry. 3) Improve the modeling of  $\text{CH}_4$  degassing by better defining drivers of hypolimnetic  $\text{CH}_4$  accumulation,  
namely thermal stratification. 4) Gather additional field data on GHG dynamics downstream of dams (degassing, river  
340 emissions, and river  $\text{CH}_4$  oxidation) in order to incorporate all components of the flux to the modeling of reservoirs C  
footprint.



### Author contribution

CS contributed to conceptualization, methodology, validation, formal analysis, investigation, data curation, writing - original draft, writing – review and editing, and project administration. YTP contributed to Methodology, validation, investigation, resources, writing – review and editing, supervision, and funding acquisition.

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### Competing interests

The authors declare that they have no conflict of interest.

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## References

- Barros, N., Cole, J. J., Tranvik, L. J., Prairie, Y. T., Bastviken, D., Huszar, V. L. M., del Giorgio, P. and Roland, F.: Carbon emission from hydroelectric reservoirs linked to reservoir age and latitude, *Nat. Geosci.*, 4(9), 593–596, doi:10.1038/ngeo1211, 2011.
- 360 Bastien, J. and Demarty, M.: Spatio-temporal variation of gross CO<sub>2</sub> and CH<sub>4</sub> diffusive emissions from Australian reservoirs and natural aquatic ecosystems, and estimation of net reservoir emissions, *Lakes Reserv. Res. Manag.*, 18(2), 115–127, doi:10.1111/lre.12028, 2013.
- Bastviken, D., Tranvik, L. J., Downing, J. A., Crill, P. M. and Enrich-Prast, A.: Freshwater Methane Emissions Offset the Continental Carbon Sink, *Science* (80-. ), 331(6013), 50–50, doi:10.1126/science.1196808, 2011.
- 365 Beaulieu, J. J., McManus, M. G. and Nietch, C. T.: Estimates of reservoir methane emissions based on a spatially balanced probabilistic-survey, *Limnol. Oceanogr.*, 61(S1), S27–S40, doi:10.1002/lno.10284, 2016.
- Blais, J. M. and Kalff, J.: The influence of lake morphometry on sediment focusing, *Limnol. Oceanogr.*, 40(3), 582–588, doi:10.4319/lo.1995.40.3.0582, 1995.
- Bogard, M. J. and del Giorgio, P. A.: The role of metabolism in modulating CO<sub>2</sub> fluxes in boreal lakes, *Global Biogeochem. Cycles*, 30(10), 1509–1525, doi:10.1002/2016GB005463, 2016.
- 370 Chanudet, V., Descloux, S., Harby, A., Sundt, H., Hansen, B. H., Brakstad, O., Serça, D. and Guerin, F.: Gross CO<sub>2</sub> and CH<sub>4</sub> emissions from the Nam Ngum and Nam Leuk sub-tropical reservoirs in Lao PDR, *Sci. Total Environ.*, 409(24), 5382–5391, doi:10.1016/j.scitotenv.2011.09.018, 2011.
- Cole, J. J. and Caraco, N. F.: Atmospheric exchange of carbon dioxide in a low-wind oligotrophic lake measured by the addition of SF<sub>6</sub>, *Limnol. Oceanogr.*, 43(4), 647–656, doi:10.4319/lo.1998.43.4.0647, 1998.
- 375 Coleman, D. D., Risatti, J. B. and Schoell, M.: Fractionation of carbon and hydrogen isotopes by methane-oxidizing bacteria, *Geochim. Cosmochim. Acta*, 45(7), 1033–1037, doi:10.1016/0016-7037(81)90129-0, 1981.
- Deemer, B. R., Harrison, J. A., Li, S., Beaulieu, J. J., DelSontro, T., Barros, N., Bezerra-Neto, J. F., Powers, S. M., dos Santos, M. A. and Vonk, J. A.: Greenhouse Gas Emissions from Reservoir Water Surfaces: A New Global Synthesis, *Bioscience*, 66(11), 949–964, doi:10.1093/biosci/biw117, 2016.
- 380 DelSontro, T., McGinnis, D. F., Sobek, S., Ostrovsky, I. and Wehrli, B.: Extreme Methane Emissions from a Swiss Hydropower Reservoir: Contribution from Bubbling Sediments, *Environ. Sci. Technol.*, 44(7), 2419–2425,



doi:10.1021/es9031369, 2010.

385 DelSontro, T., Kunz, M. J., Kempter, T., Wüest, A., Wehrli, B. and Senn, D. B.: Spatial Heterogeneity of Methane Ebullition in a Large Tropical Reservoir, *Environ. Sci. Technol.*, 45(23), 9866–9873, doi:10.1021/es2005545, 2011.

Demarty, M. and Bastien, J.: GHG emissions from hydroelectric reservoirs in tropical and equatorial regions: Review of 20 years of CH<sub>4</sub> emission measurements, *Energy Policy*, 39(7), 4197–4206, doi:10.1016/j.enpol.2011.04.033, 2011.

390 Galy-Lacaux, C., Delmas, R., Jambert, C., Dumestre, J.-F., Labroue, L., Richard, S. and Gosse, P.: Gaseous emissions and oxygen consumption in hydroelectric dams: A case study in French Guyana, *Global Biogeochem. Cycles*, 11(4), 471–483, doi:10.1029/97GB01625, 1997.

Grinham, A., Dunbabin, M. and Albert, S.: Importance of sediment organic matter to methane ebullition in a sub-tropical freshwater reservoir, *Sci. Total Environ.*, 621, 1199–1207, doi:10.1016/j.scitotenv.2017.10.108, 2018.

Guérin, F. and Abril, G.: Significance of pelagic aerobic methane oxidation in the methane and carbon budget of a tropical reservoir, *J. Geophys. Res. Biogeosciences*, 112(G3), doi:10.1029/2006JG000393, 2007.

395 International Hydropower Association (IHA): A brief history of hydropower, [online] Available from: <https://www.hydropower.org/a-brief-history-of-hydropower> (Accessed 11 July 2019), 2015.

ISRIC - World Soil Information: SoilGrids v0.5.3., [online] Available from: [https://soilgrids.org/#/?layer=ORCDRC\\_M\\_sl4\\_250m&vector=1](https://soilgrids.org/#/?layer=ORCDRC_M_sl4_250m&vector=1) (Accessed 1 May 2019), 2019.

400 Keller, M. and Stallard, R. F.: Methane emission by bubbling from Gatun Lake, Panama, *J. Geophys. Res.*, 99(D4), 8307, doi:10.1029/92JD02170, 1994.

Kemenes, A., Forsberg, B. R. and Melack, J. M.: Methane release below a tropical hydroelectric dam, *Geophys. Res. Lett.*, 34(12), 1–5, doi:10.1029/2007GL029479, 2007.

Kemenes, A., Forsberg, B. R. and Melack, J. M.: Downstream emissions of CH<sub>4</sub> and CO<sub>2</sub> from hydroelectric reservoirs (Tucuruí, Samuel, and Curuá-Una) in the Amazon basin, *Inl. Waters*, 6(3), 295–302, doi:10.1080/IW-6.3.980, 2016.

405 Knox, M., Quay, P. D. and Wilbur, D.: Kinetic isotopic fractionation during air-water gas transfer of O<sub>2</sub>, N<sub>2</sub>, CH<sub>4</sub>, and H<sub>2</sub>, *J. Geophys. Res.*, 97(C12), 20335, doi:10.1029/92JC00949, 1992.

Ledwell, J. J.: The Variation of the Gas Transfer Coefficient with Molecular Diffusivity, in *Gas Transfer at Water Surfaces*,



- pp. 293–302, Springer Netherlands, Dordrecht., 1984.
- 410 Lehner, B., Liermann, C. R., Revenga, C., Vörösmarty, C., Fekete, B., Crouzet, P., Döll, P., Endejan, M., Frenken, K., Magome, J., Nilsson, C., Robertson, J. C., Rödel, R., Sindorf, N. and Wisser, D.: High-resolution mapping of the world's reservoirs and dams for sustainable river-flow management, *Front. Ecol. Environ.*, 9(9), 494–502, doi:10.1890/100125, 2011.
- Lide, D.: *CRC Handbook of Chemistry and Physics*, edited by CRC Press, Boca Raton, FL. [online] Available from: <http://www.hbcpnetbase.com>, 2005.
- 415 St. Louis, V. L., Kelly, C. a., Duchemin, É., Rudd, J. W. M. and Rosenberg, D. M.: Reservoir Surfaces as Sources of Greenhouse Gases to the Atmosphere: A Global Estimate, *Bioscience*, 50(9), 766, doi:10.1641/0006-3568, 2000.
- de Mello, N. A. S. T., Brighenti, L. S., Barbosa, F. A. R., Staehr, P. A. and Bezerra Neto, J. F.: Spatial variability of methane (CH<sub>4</sub>) ebullition in a tropical hypereutrophic reservoir: silted areas as a bubble hot spot, *Lake Reserv. Manag.*, 34(2), 105–114, doi:10.1080/10402381.2017.1390018, 2018.
- 420 Murase, J. and Sugimoto, A.: Inhibitory effect of light on methane oxidation in the pelagic water column of a mesotrophic lake (Lake Biwa, Japan), *Limnol. Oceanogr.*, 50(4), 1339–1343, doi:10.4319/lo.2005.50.4.1339, 2005.
- Myhre, G., Shindell, D., Bréon, F.-M., Collins, W., Fuglestedt, J., Huang, J., Koch, D., Lamarque, J.-F., Lee, D., Mendoza, B., Nakajima, T., Robock, A., Stephens, G., Takemura, T. and Zhang, H.: Anthropogenic and Natural Radiative Forcing, in *Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change*, pp. 659–740, Cambridge, United Kindom. [online] Available from: [http://www.climatechange2013.org/images/report/WG1AR5\\_Chapter08\\_FINAL.pdf](http://www.climatechange2013.org/images/report/WG1AR5_Chapter08_FINAL.pdf), 2013.
- Oades, J. M.: The retention of organic matter in soils, *Biogeochemistry*, 5(1), 35–70, doi:10.1007/BF02180317, 1988.
- Oswald, K., Milucka, J., Brand, A., Littmann, S., Wehrli, B., Kuypers, M. M. M. and Schubert, C. J.: Light-Dependent Aerobic Methane Oxidation Reduces Methane Emissions from Seasonally Stratified Lakes, edited by C. Lovejoy, *PLoS One*, 10(7), e0132574, doi:10.1371/journal.pone.0132574, 2015.
- 430 Paranaíba, J. R., Barros, N., Mendonça, R., Linkhorst, A., Isidorova, A., Roland, F., Almeida, R. M. and Sobek, S.: Spatially Resolved Measurements of CO<sub>2</sub> and CH<sub>4</sub> Concentration and Gas-Exchange Velocity Highly Influence Carbon-Emission Estimates of Reservoirs, *Environ. Sci. Technol.*, 52(2), 607–615, doi:10.1021/acs.est.7b05138, 2018.
- Pebesma, E. J.: Multivariable geostatistics in S: the gstat package, *Comput. Geosci.*, 30(7), 683–691, doi:10.1016/j.cageo.2004.03.012, 2004.



- Prairie, Y. T., Alm, J., Beaulieu, J., Barros, N., Battin, T., Cole, J., del Giorgio, P., DelSontro, T., Guérin, F., Harby, A., Harrison, J., Mercier-Blais, S., Serça, D., Sobek, S. and Vachon, D.: Greenhouse Gas Emissions from Freshwater Reservoirs: What Does the Atmosphere See?, *Ecosystems*, 21(5), 1058–1071, doi:10.1007/s10021-017-0198-9, 2018.
- 440 R Core Team: R: A language and environment for statistical computing, [online] Available from: <https://www.r-project.org/>, 2017.
- Rowan, D. J., Kalf, J. and Rasmussen, J. B.: Estimating the Mud Deposition Boundary Depth in Lakes from Wave Theory, *Can. J. Fish. Aquat. Sci.*, 49(12), 2490–2497, doi:10.1139/f92-275, 1992.
- Rudd, J., Harris, R. and Kelly, C. A.: Are Hydroelectric Reservoirs Significant Sources of Greenhouse Gases?, *Ambio*, 22(4), 246–248 [online] Available from: [http://www.jstor.org/stable/4314078?seq=1#page\\_scan\\_tab\\_contents](http://www.jstor.org/stable/4314078?seq=1#page_scan_tab_contents), 1993.
- 445 Sarawak Government: The Geography of Sarawak, [online] Available from: [https://www.sarawak.gov.my/web/home/article\\_view/159/176/](https://www.sarawak.gov.my/web/home/article_view/159/176/) (Accessed 3 May 2019), 2019.
- Sherman, B. and Ford, P.: Methane Emissions from Two reservoirs in a steep, sub- Tropical rainforest catchment, in *Science Forum and Stakeholder Engagement: Building Linkages, Collaboration and Science Quality*, edited by D. K. Begbie and S. L. Wakem, pp. 1–9, Urban Water Security Research Alliance, Brisbane, Queensland., 2011.
- 450 Sobek, S., Tranvik, L. J. and Cole, J. J.: Temperature independence of carbon dioxide supersaturation in global lakes, *Global Biogeochem. Cycles*, 19(2), n/a–n/a, doi:10.1029/2004GB002264, 2005.
- Sobek, S., Tranvik, L. J., Prairie, Y. T., Kortelainen, P. and Cole, J. J.: Patterns and regulation of dissolved organic carbon: An analysis of 7,500 widely distributed lakes, *Limnol. Oceanogr.*, 52(3), 1208–1219, doi:10.4319/lo.2007.52.3.1208, 2007.
- 455 Soued, C., del Giorgio, P. A. and Maranger, R.: Nitrous oxide sinks and emissions in boreal aquatic networks in Québec, *Nat. Geosci.*, 9(2), 116–120, doi:10.1038/ngeo2611, 2016.
- Teodoru, C. R., Prairie, Y. T. and del Giorgio, P. A.: Spatial Heterogeneity of Surface CO<sub>2</sub> Fluxes in a Newly Created Eastmain-1 Reservoir in Northern Quebec, Canada, *Ecosystems*, 14(1), 28–46, doi:10.1007/s10021-010-9393-7, 2011.
- 460 Teodoru, C. R., Bastien, J., Bonneville, M.-C., del Giorgio, P. A., Demarty, M., Garneau, M., Hélie, J.-F., Pelletier, L., Prairie, Y. T., Roulet, N. T., Strachan, I. B. and Tremblay, A.: The net carbon footprint of a newly created boreal hydroelectric reservoir, *Global Biogeochem. Cycles*, 26(2), doi:10.1029/2011GB004187, 2012.
- Thottathil, S. D., Reis, P. C. J. and Prairie, Y. T.: Methane oxidation kinetics in northern freshwater lakes, *Biogeochemistry*, 143(1), 105–116, doi:10.1007/s10533-019-00552-x, 2019.



UNESCO/IHA: The GHG Reservoir Tool (G- res), [online] Available from: <https://g-res.hydropower.org/> (Accessed 1 May 2019), 2017.

465 Venkiteswaran, J. J., Schiff, S. L., St. Louis, V. L., Matthews, C. J. D., Boudreau, N. M., Joyce, E. M., Beaty, K. G. and Bodaly, R. A.: Processes affecting greenhouse gas production in experimental boreal reservoirs, *Global Biogeochem. Cycles*, 27(2), 567–577, doi:10.1002/gbc.20046, 2013.

Wanninkhof, R.: Relationship between wind speed and gas exchange over the ocean, *J. Geophys. Res.*, 97(C5), 7373, doi:10.1029/92JC00188, 1992.

470 Wasli, M. E., Tanaka, S., Kendawang, J. J., Abdu, A., Lat, J., Morooka, Y., Long, S. M. and Sakurai, K.: Soils and Vegetation Condition of Natural Forests and Secondary Fallow Forests within Batang Ai National Park Boundary , Sarawak , Malaysia, *Kuroshio Sci.*, 5(1), 67–76, 2011.

Webb, J. R., Hayes, N. M., Simpson, G. L., Leavitt, P. R., Baulch, H. M. and Finlay, K.: Widespread nitrous oxide undersaturation in farm waterbodies creates an unexpected greenhouse gas sink, *Proc. Natl. Acad. Sci.*, 201820389, doi:10.1073/pnas.1820389116, 2019.

Zarfl, C., Lumsdon, A. E., Berlekamp, J., Tydecks, L. and Tockner, K.: A global boom in hydropower dam construction, *Aquat. Sci.*, 77(1), 161–170, doi:10.1007/s00027-014-0377-0, 2015.



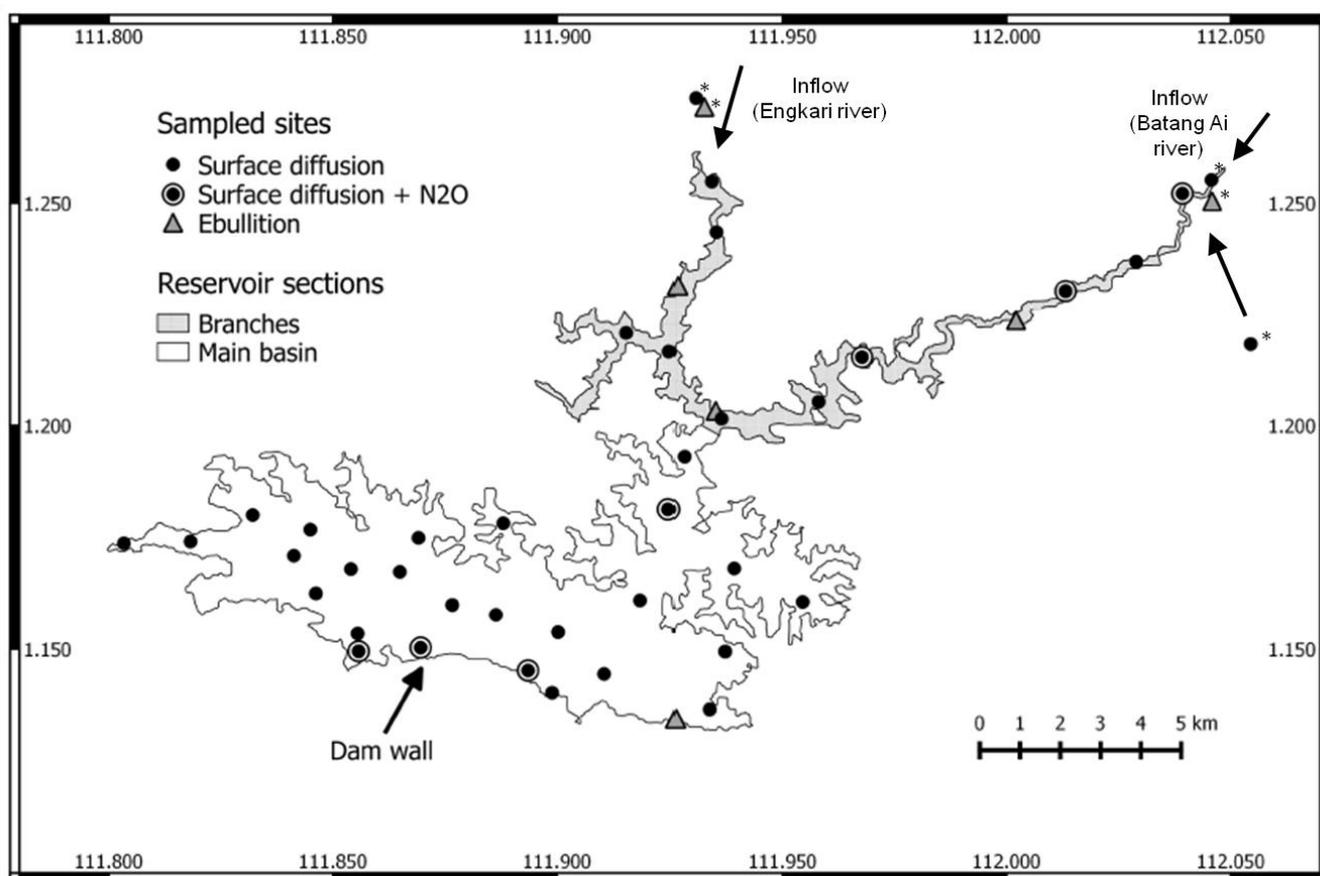
480 **Table 1: CO<sub>2</sub> and CH<sub>4</sub> dynamics downstream of the dam: gas export rate from upstream to downstream of the dam, degassing, oxidation (production for CO<sub>2</sub> and consumption for CH<sub>4</sub>), and resulting total emissions to the atmosphere (expressed as a range of values based on the two different baseline concentrations). Units are in mmol m<sup>-2</sup> d<sup>-1</sup> of reservoir surface area.**

	GHG downstream of the dam (mmol m <sup>-2</sup> d <sup>-1</sup> )			
	Exported	Degassed	CO <sub>2</sub> prod. / CH <sub>4</sub> cons. by oxidation	Total emitted
<b>CO<sub>2</sub></b>				
Nov-Dec 2016	40.62	15.26	0.90	27.9 - 38.6
Apr-May 2017	37.80	14.91	0.59	24.7 - 36.1
Feb-March 2018	37.98	9.58	1.80	19.3 - 36.9
Aug 2018	38.07	21.67	0.38	30 - 36.9
<b>CH<sub>4</sub></b>				
Nov-Dec 2016	14.84	11.56	0.90	13.76 - 13.93
Apr-May 2017	7.32	4.00	0.59	5.9 - 6.73
Feb-March 2018	12.47	4.92	1.80	8.91 - 10.67
Aug 2018	10.71	9.54	0.38	10.05 - 10.32

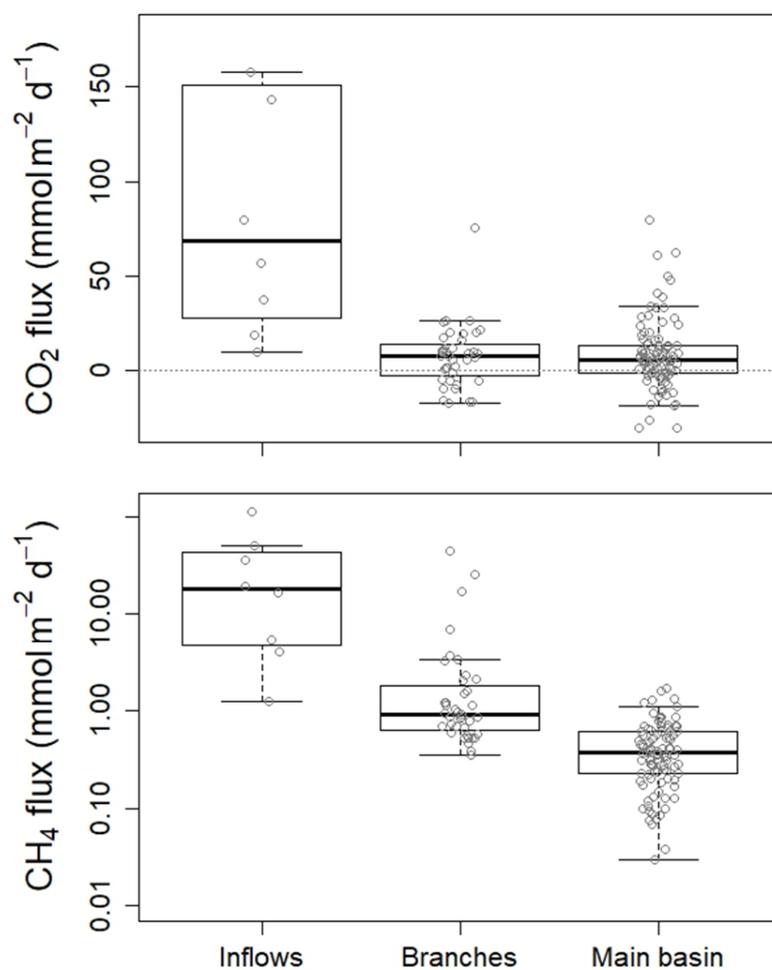


485 **Table 2: Estimated reservoir and inflow areal and total GHG fluxes to the atmosphere ( $\pm$  standard error or 95 % confidence interval based on model standard error) from different pathways based on measured and modeled approaches.**

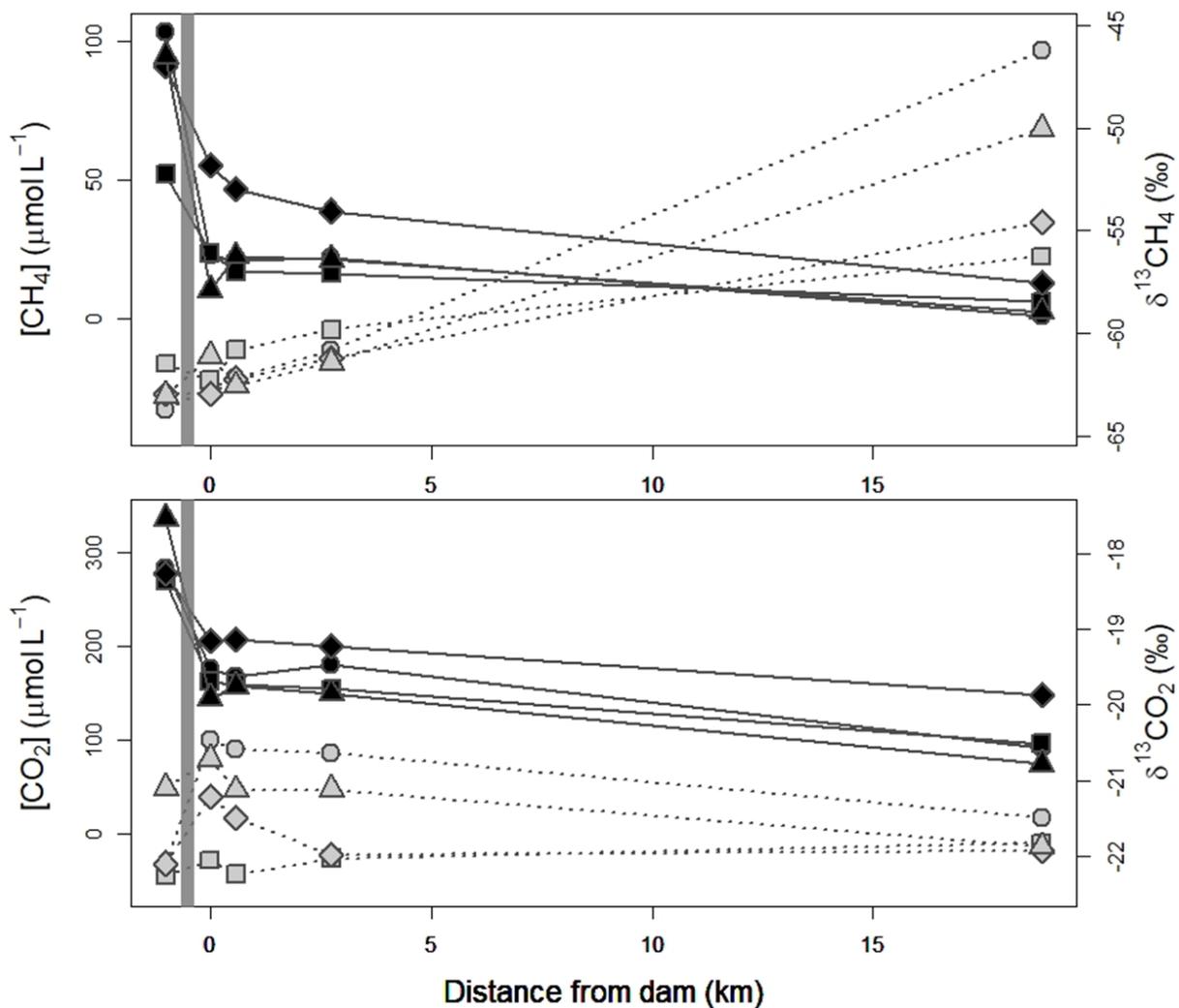
	Diffusion		N <sub>2</sub> O	Ebullition		Degassing		Downstream river emissions		Total
	CO <sub>2</sub>	CH <sub>4</sub>		CH <sub>4</sub>	CO <sub>2</sub>	CH <sub>4</sub>	CO <sub>2</sub>	CH <sub>4</sub>		
<i>Flux rate (gCO<sub>2</sub>eq m<sup>-2</sup> yr<sup>-1</sup>)</i>										
Reservoir (measured)	113 ( $\pm$ 15)	153 ( $\pm$ 43)	-2.1 ( $\pm$ 7.1)	3.4 ( $\pm$ 1.9)	247 ( $\pm$ 32)	1342 ( $\pm$ 372)	256 ( $\pm$ 34)	527 ( $\pm$ 146)	2639 ( $\pm$ 651)	
Reservoir (G-res model)	577 (509 - 655)	161 (132 - 197)		52 (32 - 83)		468 (266 - 832)			1258 (1041 - 1636)	
Reservoir (Barros et al. model)	4671	176							4847	
Inflows (measured)	156 - 9538	248 - 22510	NA	10377 - 20498	0	0	0	0	10781 - 52546	
<i>Total flux (TgCO<sub>2</sub>eq yr<sup>-1</sup>)</i>										
Reservoir (measured)	0.008	0.010	-0.0001	0.0002	0.017	0.092	0.018	0.036	0.181	
River*	0 - 0.014	0 - 0.034	NA	0.016 - 0.031	0.000	0.000	0.000	0.000	0.016 - 0.08	



490 Figure 1: Map of Batang Ai showing the location of sampled sites and reservoir sections. \* Represents the reservoir inflow sites.



495 **Figure 2:** Boxplots of measured CH<sub>4</sub> (on a log axis) and CO<sub>2</sub> fluxes grouped according to spatial position. Boxes are bounded by the 25<sup>th</sup> and 75<sup>th</sup> percentile and show medians (solid lines), and whiskers show 10<sup>th</sup> and 90<sup>th</sup> percentiles. Gray circles show single data points.



500 Figure 3: Concentrations (black symbols and solid line) and  $\delta^{13}\text{C}$  (gray symbols and dotted lines) of  $\text{CO}_2$  and  $\text{CH}_4$  from right upstream of the dam (gray band) to 19 km downstream in the outflow river. Circles, squares, diamonds, and triangles represent values from Nov-Dec 2016, Apr-May 2017, Feb-Mar 2018, and Aug 2018 respectively.