# Characterizing the origins of dissolved organic carbon in coastal seawater using stable carbon isotope and light absorption characteristics

Heejun Han<sup>1</sup>, Jeomshik Hwang<sup>1</sup>, Guebuem Kim<sup>1</sup>

5 <sup>1</sup>School of Earth and Environmental Sciences/Research Institute of Oceanography, Seoul National University, Seoul, 08826, South Korea

Correspondence to: Guebuem Kim (gkim@snu.ac.kr)

Abstract. In order to determine the origins of dissolved organic matter (DOM) occurring in the seawater of Sihwa Lake, we measured the stable carbon isotope ratios of dissolved organic carbon (DOC- $\delta^{13}$ C) and the optical properties (absorbance and fluorescence) of DOM in two different seasons (March 2017 and September 2018). Sihwa Lake is enclosed by a dyke along the western coast of South Korea, and the water is exchanged with the Yellow Sea twice a day through the sluice gates. The DOC concentrations were generally higher in lower salinity waters in both periods, and excess of DOC was also observed in 2017 in high-salinity waters. Here, the excess DOC represents any DOC concentrations higher than those in the incoming open-ocean seawater. The excess DOC occurring in the lower salinity waters originated mainly from marine sediments of tidal flats, based on the DOC- $\delta^{13}$ C values (-20.7±1.2%) and good correlations among the DOC, humic-like fluorescent DOM (FDOM<sub>H</sub>), and NH<sub>4</sub><sup>+</sup> concentrations. However, the origins of the excess DOC observed in 2017 appears to be from two different sources: one mainly from marine sources such as biological production based on the DOC-  $\delta^{13}$ C values (-19.1‰ to -20.5‰), and the other mainly from terrestrial sources by land-seawater interactions based on its depleted DOC- $\delta^{13}$ C values (-21.5% to -27.8%). This <u>terrestrial</u> DOM <u>source</u> observed in 2017 was likely associated with DOM on the reclaimed land, which experienced extended exposure to light and bacterial degradation as indicated by the higher spectral slope ratio  $(S_R)$  of light absorbance and no concurrent increases in the FDOM<sub>H</sub> and NH<sub>4</sub><sup>+</sup> concentrations. Our study demonstrates that the combination of these biogeochemical tools can be a powerful tracer of DOM sources and characteristics in coastal environments.

## 1 Introduction

Dissolved organic carbon (DOC), a major component of dissolved organic matter (DOM), is the largest reduced carbon pool in the ocean (Benner et al., 1992; Raymond and Spencer, 2014). DOC plays a significant role in coastal carbon dynamics and biogeochemical cycles, and thus it is important to understand the sources and characteristics of DOC (Vetter et al., 2007; Carson and Hansell, 2015). In the coastal ocean, DOM sources include (1) in-situ biological production (Carlson and Hansell, 2015), (2) terrestrial organic matter such as soils and plant matters (Opsahl and Benner, 1997; Bauer and Bianchi,

1

Heejun Han 5/2/21 1:36 PM

Deleted: coastal

Heeiun Han 5/2/21 1:36 PM

**Deleted:** a water body semi-enclosed by a dyke along the western coast of South Korea,

Heejun Han 5/2/21 1:37 PM

Deleted: of DOC

Heejun Han 5/2/21 1:38 PM

**Deleted:** higher concentrations of

Heeiun Han 5/2/21 1:39 PM

Deleted: land-derived

2011), and (3) anthropogenic sources such as industrial and agricultural wastewaters (Tedetti et al., 2010; Griffith and Raymond, 2011).

A part of DOM is known as colored dissolved organic matter (CDOM), which is the light-absorbing fraction of organic matter (Coble, 2007; Kim and Kim, 2016; Kim and Kim, 2018). The major fraction of CDOM, which emits fluorescence after absorbing light energy, is referred to as fluorescent DOM (FDOM) (Coble, 1996; Coble, 2007; Kim and Kim, 2016). Rivers are known as the major source of the humic-like FDOM (FDOM<sub>H</sub>) in the coastal ocean (Stedmon and Nelson, 2015; Kim and Kim, 2016), while aerobic microbial remineralization of sinking organic matter is the major source of FDOM<sub>H</sub> in the deep sea (Jørgensen et al., 2011; Catalá et al., 2015; Kim and Kim, 2016). On the other hand, Kim and Kim (2016) reported that anaerobic production in the bottom sediment of the deep East Sea (Japan Sea) accounted for approximately 10% of the total production of FDOM<sub>H</sub> in the deep-water column. Recently, the anaerobic process in the bottom sediment of coastal oceans was also suggested as an important source of FDOM<sub>H</sub> (Kim and Kim, 2018).

In order to decipher the sources of FDOM<sub>H</sub> and DOC in coastal waters, various approaches including optical properties and stable carbon isotope ratios ( $\delta^{13}$ C)\_have been developed. The humic-like component and protein-like component of FDOM were identified based on an excitation-emission matrix (EEM) spectroscopy method combined with a parallel factor analysis (PARAFAC) model (Bro, 1997; Coble, 2007; Kim and Kim, 2016; Kim and Kim 2018). The molecular weight and photochemical processes of DOM were characterized by the absorption spectral parameters of CDOM. For example, the absorption spectral slope ratio ( $S_R$ ) negatively correlates with the DOM molecular weight and increases upon irradiation (Helms et al., 2008; Hansen et al., 2016). The stable carbon isotopic composition of DOC (DOC– $\delta^{13}$ C) has been used to differentiate terrestrial versus marine DOC (Gearing, 1988; Wang et al., 2004; Lee and Kim, 2018; Lee et al., 2020). In general,  $\delta^{13}$ C values of terrestrial sources such as  $C_3$  and  $C_4$  plants range from –23‰ to –34‰ and –9‰ to –17‰, respectively, while those derived from marine phytoplankton range from –18‰ to –22‰ (Gearing, 1998).

60 In this study, we used DOC-δ<sup>13</sup>C, FDOM, and S<sub>R</sub> values together to characterize different sources of DOM in Sihwa Lake, South Korea, one of the most dynamic coastal settings in terms of salinity changes, hypoxia, metal pollution, and eutrophication (Kim et al., 2009; Ra et al., 2011; Lee et al., 2014; Kim and Kim, 2014; Lee et al., 2017; Kim and Kim, 2018).

Heeiun Han 5/2/21 4:30 PM

Deleted: a

Heeiun Han 5/2/21 5:22 PM

Deleted:

Heejun Han 5/2/21 4:29 PM

Deleted: s

Heejun Han 5/2/21 4:45 PM

Deleted: were found to

## 2 Materials and methods

# 2.1 Study area and sampling

Sihwa Lake (126.6 °E; 37.3 °N) is located on the western coast of South Korea, with an area of 57 km² and an average depth of 3.2 m (maximum depth: 18 m). This lake was originally constructed as a land reclamation project planned by the government to provide agricultural land and water for the nearby metropolitan area during the 1980s and 1990s (Bae et al., 2010) (Fig. 1). Freshwater runs through six small streams into the lake, and four waterways connect the lake to the Banwol industrial complex (Fig. 1). Because the lake experienced serious deterioration of its water quality owing to wastewater discharges from industrial complexes under limited water circulation, sluice gates were constructed and opened twice a day for water exchanges between the lake and the Yellow Sea since 2012. The dyke is currently used as a tidal power plant (Lee et al., 2017) (Fig. 1). The total volume of Sihwa Lake water is ~3.3×10<sup>8</sup> m³. The water exchange rate is approximately  $3.4\times10^8$  m³ y⁻¹, enough to replace the entire reservoir in a year (Lee et al., 2003; Lee et al., 2014).

Water samples for the entire water column were collected on board a ship (~1 ton) in March 2017 and September 2018. In 2018, only surface water samples were collected at shallow stations (station numbers 1–6) as the water level of the reservoir was lower than in 2017. In order to investigate the effect of wastewater from the industrial complex, an additional sample was collected near the Banwol waterway (station B4) in 2018 (Fig. 1). The temperature and salinity were measured using a conductivity-temperature-depth (CTD) instrument (Ocean Seven 304, INDONAUT Srl).

The water samples were filtered through a pre-combusted (450 °C for 5h) GF/F filter (pore size:  $0.7 \mu m$ ; Whatman). The samples for the DOC and DOC- $\delta^{13}$ C analyses were acidified with 6M HCl (to a final sample pH of ~2) to avoid any bacterial activity and were stored in pre-combusted glass ampoules (Kim and Kim, 2010). Samples for the FDOM analysis were stored in pre-combusted amber vials in a refrigerator at 4°C. Samples for the dissolved inorganic nutrient analysis were stored frozen in polypropylene conical tubes.

# 2.2 Chemical analyses

80

Inorganic nutrient concentrations were measured using a nutrient auto-analyzer (New QuAAtro39, SEAL analytical). The analytical uncertainties were <5% for the reference materials for  $NO_X$  (KANSO Co., LTD). The dissolved oxygen (DO) concentration was determined using Winkler's method (Carpenter, 1965). The DOC concentration was measured using a high-temperature catalytic oxidation (HTCO) method using a total organic carbon (TOC) analyzer (TOC- $V_{CPH}$ , Shimadzu) (Kim and Kim, 2010). The accuracy of the DOC measurements was  $\pm 2 \, \mu \text{mol L}^{-1}$  for the reference material of deep seawater (DSR; ~43  $\, \mu \text{mol L}^{-1}$  DOC; University of Miami) (Hansell, 2005). The DOC– $\delta^{13}$ C values were measured using an isotope ratio mass spectrometer (IRMS; Isoprime, Elementar) connected to a TOC analyzer (Vario TOC cube, Elementar) (Panetta et al., 2008; Troyer et al., 2010). The DOC– $\delta^{13}$ C values were in very good agreement, within <5%, with the reported values of

Heejun Han 5/2/21 4:45 PM

Deleted: a certifie

the reference materials: IAEA-CH6 sucrose ( $\delta^{13}C = -10.45\pm0.03\%$ ), Suwannee River Fulvic Acid (SRFA;  $\delta^{13}C = -27.6\pm0.12\%$ ; International Humic Substances Society), and DSR (University of Miami) as reported by Lang et al. (2007) ( $\delta^{13}C = -21.7\pm0.3\%$ ) and Panetta et al. (2008) ( $\delta^{13}C = -21.4\pm0.3\%$ ).

# 105 2.3 Optical measurements

110

Fluorescence and absorbance spectra of the samples were measured using a spectrophotometer (Aqualog, Horiba). For the FDOM analysis, the emission and excitation wavelength ranges were set from 240 to 600 nm and from 250 to 500 nm, respectively, with 3 nm scanning intervals. The PARAFAC analysis for the EEM data was conducted using the Solo software (Eigenvector, 108 Inc.) (Han et al., 2020). The Raman and Rayleigh scattering signals, inner-filter effect, and blank subtraction were corrected with Milli-Q water (18.2  $M\Omega$ cm) using the Solo software (Bro, 1997; Zepp et al., 2004; Stedmon and Bro, 2008; Han et al., 2020). The PARAFAC results were validated by a split-half analysis and random initialization (Stedmon and Bro, 2008). The fluorescence intensities of FDOM were normalized with the Raman peak area of water and were presented in Raman Unit (RU) (Lawaetz and Stedmon, 2009).

The PARAFAC model characterized one marine humic-like, one protein-like, and two terrestrial humic-like fluorescent components in Sihwa Lake, which are consistent with those of a previous study (Kim and Kim, 2018) (Fig. S1). The spectral shapes of the fluorescent components were compared with those in other results in the OpenFluor database (https://openfluor.lablicate.com) (Murphy et al., 2014). All components (C1 to C4) were matched with the major components from 36, 39, 62, and 19 studies, respectively, with similarity scores of 95%. The spectral characteristics of component 1 (FDOM<sub>C</sub>; Ex/Em = 342/427 nm) and component 3 (FDOM<sub>A</sub>; Ex/Em = 381/493 nm) are known to be associated with terrestrial humic-like DOM (Coble 2007). The spectral characteristics of component 2 (FDOM<sub>M</sub>; Ex/Em = 297/388 nm) is known to be associated with marine humic-like DOM originating from microbial remineralization (Coble, 2007; Jørgensen et al., 2011). Component 4 (FDOM<sub>P</sub>; Ex/Em = 282/322 nm) is characterized as a protein-like (tryptophan-like) DOM, which originates mainly from biological production (Coble, 2007). In this study, FDOM<sub>C</sub> was used to represent humic FDOM (FDOM<sub>H</sub>) since all humic-like components showed similar patterns.

UV-visible absorption spectra of the samples were measured with a scanning wavelength range of 240–700 nm. The absorption coefficient was calculated using the following equation:

$$a_{\lambda} = 2.303 A_{\lambda} / l \tag{1}$$

where  $\alpha$  is the absorption coefficient (m<sup>-1</sup>),  $A_{\lambda}$  is the absorbance, and l is the optical path length of the quartz cuvette (m). The  $S_R$  value was calculated as the ratio of the spectral slope of shorter wavelengths ( $S_{275-295}$ ) to that of longer wavelengths ( $S_{350-400}$ ) (Helms et al., 2008; Han et al., 2020). The spectral slope (S) was calculated using the following equation:

$$a_{\lambda} = a_{\lambda_{ref}} e^{-S(\lambda - \lambda_{ref})} \tag{2}$$

Heejun Han 5/2/21 4:47 PM

Deleted: C

Heejun Han 5/2/21 1:41 PM **Deleted:** as a representative of

where  $\alpha$  is the Napierian absorption coefficient (m<sup>-1</sup>),  $\lambda$  is the wavelength, and  $\lambda_{ref}$  is the reference wavelength (Twardowski et al., 2004; Helms et al., 2008).

#### 3 Results

150

In 2017, the vertical distribution of salinity indicated a well-mixed water column (Fig. 2). Similarly, the concentrations of DO and NH<sub>4</sub><sup>+</sup> were vertically uniform (Fig. 2). Horizontally, salinity increased from 28 at the innermost station to 32 at the outermost station (Fig. 2). The concentrations of DO and NH<sub>4</sub><sup>+</sup> were in the ranges of 7–13 mg L<sup>-1</sup> (average: 10.1±2.4 mg L<sup>-1</sup>) and 0.1–25 μmol L<sup>-1</sup> (average: 8.7±8.1 μmol L<sup>-1</sup>), respectively (Fig. 2). The DO concentration gradually increased with an increase in salinity along the transect, while the NH<sub>4</sub><sup>+</sup> concentration decreased with an increase in salinity (Fig. 2). The NH<sub>4</sub><sup>+</sup> concentration showed the lowest values (<1 μmol L<sup>-1</sup>) between station 10 and station 13 (Fig. 2).

In 2017, the DOC concentrations were in the range of 97–349 µmol L<sup>-1</sup> (average:  $184\pm76$  µmol L<sup>-1</sup>). The distribution of DOC concentrations was quite different from the salinity and DO distributions, with the highest concentrations in the surface water samples of stations 3, 4, 5, 6, 7, 8, and 9 and the bottom water samples of stations 3, 4, and 5 (Fig. 2). The DOC $-\delta^{13}$ C values ranged from -19.2% to -27.8% (average:  $-21.8\pm1.9\%$ ) (Fig. 2). The most depleted values were observed in the surface water samples of stations 5, 6, 7, 9, and 10 (-22.6% to -27.8%) (Fig. 2). The concentrations of FDOM<sub>C</sub> (terrestrial humic-like component 1), FDOM<sub>A</sub> (terrestrial humic-like component 2), FDOM<sub>M</sub> (marine humic-like component), and FDOM<sub>P</sub> (protein-like component) were in the ranges of 1.6-4.1 RU (average:  $2.3\pm0.8$  RU), 0.6-1.8 RU (average:  $1.1\pm0.3$  RU), 1.0-2.4 RU (average:  $1.5\pm0.5$  RU), and 1.6-6.1 RU (average:  $2.8\pm1.0$  RU), respectively (Fig. 2). The concentrations of all FDOM components were generally higher in the upstream stations and decreased with salinity (Fig. 2). The FDOM<sub>P</sub> concentration was slightly higher in the bottom water of station 10 (Fig. 2). The  $S_R$  values were in the range of 0.70-1.76 (average:  $1.21\pm0.20$ ), with higher values in the surface water samples of stations 6, 8, 9, and 10 (Fig. 2).

In 2018, salinity showed a wide range, between 18 and 30, compared with that in 2017 (Fig. 3). Low-salinity waters (salinity: 18–27) were observed from the innermost station to station 9 (Fig. 3). The concentrations of DO and NH<sub>4</sub><sup>+</sup> were in the ranges of 6–11 mg L<sup>-1</sup> (average: 8.2±1.6 mg L<sup>-1</sup>) and 0.4–25 μmol L<sup>-1</sup> (average: 13.1±7.9 μmol L<sup>-1</sup>), respectively (Fig. 3). The NH<sub>4</sub><sup>+</sup> concentrations in the outermost stations were lower than the detection limit (Fig. 3).

In 2018, the concentrations of DOC were in the range of 101–195 μmol L<sup>-1</sup> (average: 130±32 μmol L<sup>-1</sup>). The DOC concentrations gradually decreased with an increase in the salinity (Fig. 3). The DOC–δ<sup>13</sup>C values ranged from –19.1‰ to – 21.5‰ (average: –20.0±0.6‰) (Fig. 3). The concentrations of FDOM<sub>C</sub>, FDOM<sub>A</sub>, FDOM<sub>M</sub>, and FDOM<sub>P</sub> were in the ranges of 1.4–5.1 RU (average: 1.9±0.9 RU), 1.3–4.1 RU (average: 1.8±0.7 RU), 1.4–4.9 RU (average: 2.1±0.9 RU), and 1.1–2.5 RU (average: 1.6±0.4 RU), respectively (Fig. 3). All FDOM<sub>H</sub> concentrations were higher in 2018 than those in 2017 (Fig. 3).

Heejun Han 5/2/21 4:47 PM

Deleted: concentrations

Heejun Han 5/2/21 1:41 PM

Deleted: of DO

Heejun Han 5/2/21 1:41 PM

Deleted: the

Heejun Han 5/2/21 1:41 PM

Deleted: the

Heejun Han 5/2/21 4:47 PM

Deleted: they were

The FDOM<sub>P</sub> concentrations were generally higher in the surface water samples, with a slightly higher value at station 12, where the salinity is slightly lower (Fig. 3). The  $S_R$  values were in the range of 0.72–1.08 (average: 0.87±0.10) (Fig. 3). The  $S_R$  values were relatively constant at all sampling stations (Fig. 3).

#### 4 Discussion

200

205

In both sampling periods, low-salinity waters showed higher DOC, lower DO, higher NH<sub>4</sub><sup>+</sup>, and higher FDOM<sub>H</sub> (FDOM<sub>C</sub>) concentrations (Figs. 2 and 3). As such, the DOC and FDOM<sub>H</sub> concentrations exhibited significant negative correlations against salinity, indicating low-salinity associated inputs of both DOM components (Figs. 4a and 4b). However, the patterns of the DOC distributions were very different from those of the FDOM<sub>H</sub> distributions, indicating a decoupling of major sources for both DOM components (Figs. 2 and 3). In this region, anthropogenic source inputs were found to be insignificant since there were no anomalous increases in DOC and FDOM<sub>H</sub> concentrations in 2018 at station B4, where the waterway connects to the Banwol industrial complex (Fig. 3).

In both periods, the concentrations of FDOM<sub>H</sub> showed significant correlations against salinities and  $NH_4^+$  concentrations, with two negative slopes against salinity and a single positive slope against  $NH_4^+$  (Figs. 4 and 5). This correlation trend suggests that the main source of FDOM<sub>H</sub> was dependent on the supply of  $NH_4^+$ , rather than freshwater inputs (Figs. 4a and 5a). In this lake, Kim and Kim (2018) hypothesized that FDOM<sub>H</sub> is produced by the anaerobic decomposition of organic matter in bottom sediments in the freshwater-seawater mixing zone. The hypothesis was, based on good correlations among the salinity,  $NH_4^+$ , and FDOM<sub>H</sub> concentrations as well as low Eh values in samples with higher FDOM<sub>H</sub> concentrations, although possible terrestrial inputs could not be excluded. Such correlations could also be observed if terrestrial fresh water, which had higher FDOM<sub>H</sub> and lower  $NH_4^+$  levels, passed through a salinity mixing zone where extremely high amounts of  $NH_4^+$  and low FDOM<sub>H</sub> are introduced from the bottom sediments. However, our  $DOC-8^{13}C$  values ( $-20.0\pm0.4\%$ ) observed in low-salinity (<28) water samples here exclude any possible significant contributions of terrestrial sources, indicating the effective degradation of terrestrial sources before they reached the mixing zone. In this tidal flat environment, the large inputs (i.e., seepage) of DOC and FDOM<sub>H</sub>, together with  $NH_4^+$ , might happen from marine sediments without significant decreases in the DO concentration (Kim et al., 2012), as the shore water runs back and forth over a wide sediment-surface area during the tidal cycle (Fig. 5). In this case, depending on the salinity ranges of the overlying waters, different slopes of FDOM<sub>H</sub> against salinities could be observed for different seasons as observed in this study (Fig. 4b).

In 2018, the distribution pattern of the DOC concentrations was similar to that of FDOM<sub>He</sub> showing a conservative mixing (Fig. 3). During this period, the freshwater contribution was greater compared with that in 2017, with the lowest salinity of 218. However, the DOC $-\delta^{13}$ C values in all samples in this period ranged from -19.1% to -21.5% (average:  $-20.0\pm0.6\%$ ), falling into the range of marine phytoplankton values (-18% to -22%) (Gearing, 1988) (Fig. 4c). The  $S_R$  values were

Heejun Han 5/2/21 4:48 PM

Deleted:

Heeiun Han 5/2/21 4:49 PM

Deleted: the

Heejun Han 5/2/21 4:50 PM

Deleted: the

Heejun Han 5/2/21 4:50 PM

Deleted:

Heejun Han 5/2/21 4:51 PM

Deleted: s

Heejun Han 5/2/21 4:51 PM

**Deleted:** es

Heejun Han 5/2/21 4:51 PM

Deleted: ay

Heejun Han 5/2/21 4:51 PM

Deleted: have

Heejun Han 5/2/21 4:51 PM

Deleted: ed

Heejun Han 5/2/21 4:51 PM

Deleted: s

Heejun Han 5/2/21 1:41 PM

Deleted:

Heejun Han 5/2/21 4:51 PM

**Deleted:** being approximately

Deleted:

relatively low and constant (average: 0.86±0.1) at all stations in 2018, indicating no significant changes in the input sources or the effects of photochemical and bacterial degradation (Moran et al., 2000; Helms et al., 2008) (Fig. 4d). Thus, we conclude that the main source of DOC in 2018 was from marine sediments, together with FDOM<sub>H</sub> and NH<sub>4</sub>, as external terrestrial sources of DOC were insignificant or effectively degraded on the tidal flat (Kim et al., 2012).

In 2017, the DOC sources were more complicated than in 2018, showing excess DOC concentrations regardless of salinity (Fig. 2). Here, this excess DOC represents any DOC concentrations higher than those in the incoming open-ocean seawater, from the Yellow Sea (the highest salinity and the lowest DOC concentrations observed in Sihwa Lake) (Fig. 4a). The samples with excess DOC levels observed in 2017 were nearly 69% higher than the DOC values of the background seawater (~110 μmol L<sup>-1</sup>) (Fig. 4a). The samples with excess DOC levels observed in 2017 were separated into two major groups (Group 1 and Group 2) based on their DOC concentrations, DOC-δ<sup>13</sup>C values, and salinities (Fig. 2). Group 1 (*n*=3) includes the samples observed in stations 12 and 13 (Fig. 2). Group 2 (*n*=11) includes the samples observed in the surface waters of stations 4, 5, 6, 7, 8, 9, and 10, and in the bottom waters of stations 3, 4, 5, and 6 (Fig. 2).

For the Group 1 samples, the DOC concentrations ranged from 144 to 223 μmol L<sup>-1</sup> (average: 175±42 μmol L<sup>-1</sup>) (Figs. 2 and 235 da). The DOC-δ<sup>13</sup>C values ranged from -19.1‰ to -22.1‰ (average: -20.6±1.5‰), which were close to the δ<sup>13</sup>C values of marine organisms (Figs. 2 and 4c). The S<sub>R</sub> values (average: 1.07±0.03) in this group were relatively low and constant compared with those of Group 2, indicating the presence of high-molecular-weight DOM (Figs. 2 and 4d). Thus, the higher DOC concentrations observed in this group appear to be from in-situ biological production.

240 For the Group 2 samples, the DOC concentrations ranged from 103 to 330 μmol L<sup>-1</sup> (average: 213±69 μmol L<sup>-1</sup>) (Figs. 2 and 4a). The DOC-δ<sup>13</sup>C values ranged from -21.5% to -27.8%, which include the signature of terrestrial C<sub>3</sub> plants (-23% to -32%) (Gearing, 1988) (Figs. 2 and 4c). For this group, increases in FDOM<sub>H</sub> concentrations were not observed, although the S<sub>R</sub> values (average: 1.26±0.2) were higher than those at other stations, likely due to the influence of highly degraded low-molecular-weight DOM (Helms et al., 2008) (Fig. 4d). Thus, this terrestrial DOM observed in 2017 seems to be associated with non-fluorescent and low-molecular-weight DOM on the reclaimed land by direct land-seawater interaction through the tidal inundation of seawater. This could happen if the terrestrial DOM experienced intense light exposure and/or bacterial degradation on land, a condition likely facilitated through the process of land reclamation.

Without the use of these multiple tracers, in this region, the higher DOC and FDOM<sub>H</sub> concentrations occurring in areas with
lower salinity waters could be attributed to inputs with terrestrial origins, while the higher DOC concentrations in high
salinity areas could be attributed to in-situ biological production. These multiple tracers could lead to the better identification
of different source inputs. Therefore, our results suggest that the combination of stable carbon isotope, FDOM, and S<sub>R</sub> values

Heejun Han 5/2/21 1:42 PM

Deleted: sources of

Heejun Han 5/2/21 4:52 PM

Deleted: large

Heejun Han 5/2/21 1:42 PM

Deleted: independent

Heejun Han 5/2/21 1:43 PM

Deleted: denote

Heejun Han 5/2/21 1:43 PM

Deleted: concentrations higher

Deleted: that of

Heejun Han 5/2/21 1:43 PM

Deleted: (background)

Heeiun Han 5/2/21 1:43 PM

**Deleted:** observed in 2018

Heejun Han 5/2/21 4:53 PM **Deleted:** higher

Heejun Han 5/2/21 4:54 PM

Deleted: are

Heejun Han 5/2/21 4:54 PM

Deleted: at

Heejun Han 5/2/21 4:55 PM

Deleted: However

Heejun Han 5/2/21 4:55 PM

Deleted: t

Heejun Han 5/2/21 4:55 PM

Deleted: an

can be a cruicial tool for deciphering the sources and characteristics of DOM in coastal waters where various DOM sources are present.

270 5 Conclusions

> Different sources and distributions of DOM were determined in different seasons using various tracers in Sihwa Lake, South Korea. Our results reveal that the higher DOC concentrations occurring in low-salinity water, which could be regarded as having terrestrial sources, were actually from marine sediment sources based on DOC- $\delta^{13}$ C values (-19.1% to -21.5%) together with significant correlations among the DOC, FDOM<sub>H</sub>, and NH<sub>4</sub><sup>+</sup> concentrations. On the other hand, the excess DOC concentrations occurring in high-salinity waters at some stations, which are generally believed to be from marine sources, were found to be from non-fluorescent, low-molecular-weight, terrestrial DOM sources from the reclaimed land area based on depleted DOC- $\delta^{13}$ C values (-21.5% to -27.8%) and high  $S_R$  values (1.26±0.2), without concurrent increases in FDOM<sub>H</sub> and NH<sub>4</sub><sup>+</sup> concentrations. Our study demonstrates that combinaing these multiple DOM tracers can be used successfully in other coastal settings where the sources and characteristics of DOM are complicated.

280

285

295

275

Data availability. All data are available upon request to the corresponding author.

Author contributions. GK contributed to the conceptualization of the manuscript. HH and GK were involved in planning the research. HH collected samples and performed the analyses. All authors were involved in analyzing the results and writing the paper.

Competing interests. The authors declare that they have no conflict of interest.

Acknowledgements. This research was supported by the National Research Foundation (NRF) of Korea (NRF-2018R1A2B3001147) funded by the South Korean government. We would like to thank all lab members for their assistance.

## References

Bae, Y. H., Kim, K. O., and Choi, B. H.: Lake Sihwa tidal power plant project, Ocean. Eng., 37, 454-463, https://doi.org/10.1016/j.oceaneng.2010.01.015, 2010.

Bauer, J. E. and Bianchi, T. S.: Dissolved organic carbon cycling and transformation, in: Treatise on estuarine and coastal science, edited by: Wolanski, E. and Mcluski, D. S., 5, 7-67, Academic Press, Waltham, 2011.

Benner, R., Pakulski, J. D., McCarthy, M., Hedges, J. I., and Hatcher, P. G.: Bulk chemical characteristics of dissolved organic matter in the ocean, Science, 255, 1561-1564, https://doi.org/10.1126/science.255.5051.1561, 1992.

Deleted: ritical

leeiun Han 5/2/21

Deleted: the

Heejun Han

Deleted: the

Deleted: tion of

Heejun Han 5/2

Deleted: waters

Bro, R.: PARAFAC. Tutorial and applications, Chemometr. Intell. Lab., 38, 149-171, 1997.

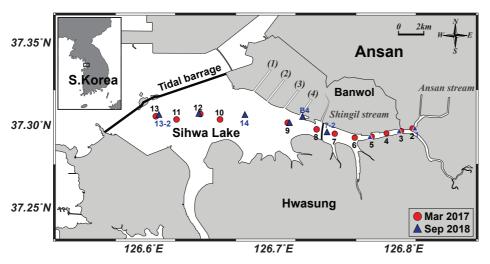
320

- Carlson, C. A. and Hansell, D. A.: Chapter 3 DOM sources, sinks, reactivity, and budgets, in: Biogeochemistry of marine dissolved organic matter (Second Edition), edited by: Hansell, D. A. and Carlson, C. A., 65–126, Academic Press, Boston. 2015.
  - Carpenter, J. H.: The Chesapeake Bay institute technique for the Winkler dissolved oxygen method, Limnol. Oceanogr., 10, 141–143, 1965.
- Catalá, T. S., Reche, I., Álvarez, M., Khatiwala, S., Guallart, E. F., Benítez-Barrios, V. M., Fuentes-Lema, A., RomeraCastillo, C., Nieto-Cid, M., Pelejero, C., Fraile-Nuez, E., Ortega-Retuerta, E., Marrasé, C., and Álvarez-Salgado, X. A.:
  Water mass age and aging driving chromophoric dissolved organic matter in the dark global ocean, Global
  Biogeochem. Cy., 29, 917–934, https://doi.org/10.1002/2014GB005048, 2015.
  - Coble, P. G.: Characterization of marine and terrestrial DOM in seawater using excitation-emission matrix spectroscopy, Mar. Chem., 51, 325–346, https://doi.org/10.1016/0304-4203(95)00062-3, 1996.
- 315 Coble, P. G.: Marine optical biogeochemistry: the chemistry of ocean color, Chem. Rev., 107, 402–418, https://doi.org/10.1021/cr050350+, 2007.
  - Gearing, J. N.: The use of stable isotope ratios of tracing the nearshore-offshore exchange of organic matter, in: Coastal-offshore ecosystem interactions, edited by: Jansson, B.-O., 69–101, Springer-Verlag, Berlin, 1988.
  - Griffith, D. R. and Raymond, P. A.: Multiple-source heterotrophy fuelled by aged organic carbon in an urbanized estuary, Mar. Chem., 124, 14–22, https://doi.org/10.1016/j.marchem.2010.11.003, 2011.
  - Han, H., Kim, G., Seo, H., Shin, K.-H., and Lee, D.-H.: Significant seasonal changes in optical properties of brown carbon in the midlatitude atmosphere, Atmos. Chem. Phys., 20, 2709–2718, https://doi.org/10.5194/acp-20-2709-2020, 2020.
  - Hansell, D. A.: Dissolved organic carbon reference material program, Eos Trans. AGU., 86, 318, https://doi.org/10.1029/2005E0350003, 2005.
- 325 Hansen, A. M., Kraus, T. E. C., Pellerin, B. A., Fleck, J. A., Downing, B. D., and Bergamaschi, B. A.: Optical properties of dissolved organic matter (DOM): Effects of biological and photolytic degradation, Limnol. Oceanogr., 61, 1015–1032, https://doi.org/10.1002/lno.10270, 2016.
  - Helms, J. R., Stubbins, A., Ritchie, J. D., and Minor, E. C.: Absorption spectral slopes and slope ratios as indicators of molecular weight, source, and photobleaching of chromophoric dissolved organic matter, Limnol. Oceanogr., 53, 955–969, https://doi.org/10.4319/lo.2008.53.3.0955, 2008.
  - Jørgensen, L., Stedmon, C. A., Kragh, T., Markager, S., Middelboe, M., Søndergaard, M.: Global trends in the fluorescence characteristics and distribution of marine dissolved organic matter, Mar. Chem., 126, 139–148, https://doi.org/10.1016/j.marchem.2011.05.002, 2011.
- Kim, J. and Kim, G.: Significant anaerobic production of fluorescent dissolved organic matter in the deep East Sea (Sea of Japan), Geophys. Res. Lett., 43, 7609–7616, https://doi.org/10.1002/2016GL069335, 2016.

- Kim, J. and Kim, T.-H.: Distribution of humic fluorescent dissolved organic matter in lake Shihwa: the role of the redox condition, Estuar. Coast., https://doi.org/10.1007/s12237-018-00491-0, 2018.
- Kim, K. T., Kim, E. S., Cho, S. R., and Park, J. K.: Distribution and temporal change of heavy metals in the surface sediments of lake Shihwa and adjacent sea, Korea, J. Coast. Res., 56, 817–821, 2009.
- 340 Kim, T.-H. and Kim, G.: Distribution of dissolved organic carbon (DOC) in the southwestern East Sea in summer, Ocean Polar Res., 32, 291–297, 2010.
  - Kim, T.-H. and Kim, G.: Estimating benthic fluxes of trace elements to hypoxic coastal waters using <sup>210</sup>Po, Estuar. Coast. Shelf. Sci., 151, 324–330, https://doi.org/10.1016/j.ecss.2014.05.008, 2014.
- Kim, T.-H., Waska, H., Kwon, E., Suryaputra, G. N., and Kim, G.: Production, degradation, and flux of dissolved organic matter in the subterranean estuary of a large tidal flat, Mar. Chem., 142–144, 1–10, http://dx.doi.org/10.1016/j.marchem.2012.08.002, 2012.
  - Koepfler, E. T., Benner, R., and Montagna, P. A.: Variability of dissolved organic carbon in sediments of a Seagrass bed and an unvegetated area within an estuary in Southern Texas, Estuaries, 16, 391–404, 1993.
  - Lang, S. Q., Lilley, M. D., and Hedge, J. I.: A method to measure the isotopic (<sup>13</sup>C) composition of dissolved organic carbon using a high temperature combustion instrument, Mar. Chem., 103, 318–326, 2007.
  - Lawaetz, A. J. and Stedmon, C. A.: Fluorescence intensity calibration using the raman scatter peak of water, Appl. Spectrosc., 62, 936–940, 2009.
  - Lee, C.-H., Lee, B.-Y., Chang, W. K., Hong, S., Song, S. J., Park, J., Kwon, B., Khim, J. S.: Environmental and ecological effects of Lake Shihwa reclamation project in South Korea: A review, Ocean. Coast. Manage., 102, 545–558, https://doi.org/10.1016/j.ocecoaman.2013.12.018, 2014.

- Lee, J., Hong, S., Yoon, S. J., Kwon, B., Ryu, J., Giesy, J. P., Allam, A. A., Al-khedhairy, A. A., and Khim, J. S.: Long-term changes in distributions of dioxin-like and estrogenic compounds in sediments of Lake Sihwa, Korea: Revisited mass balance, Chemosphere, 181, 767–777, https://doi.org/10.1016/j.chemosphere.2017.04.074, 2017.
- Lee, S.-A., Kim, T.-H., and Kim, G.: Tracing terrestrial versus marine sources of dissolved organic carbon in a coastal bay using stable carbon isotopes, Biogeosciences, 17, 135–144, https://doi.org/10.5194/bg-17-135-2020, 2020.
  - Lee, S.-I., Kim, B.-C., and Oh, H.-J.: Evaluation of lake modification alternatives for Lake Sihwa, Korea, Environ. Manage., 29, 57–66, https://doi.org/10.1007/s00267-001-0040-0, 2002.
  - Moran, M. A., Sheldon, W. M., and Zepp, R. G.: Carbon loss and optical property changes during long-term photochemical and biological degradation of estuarine dissolved organic matter, Limnol. Oceanogr., 45, 1254–1264, 2000.
- Murphy, K. R., Stedmon, C. A., Wenig, P., and Bro, R.: OpenFluor- an online spectral library of auto-fluorescence by organic compounds in the environment, Anal. Methods., 6, 658–661, https://doi.org/10.1039/C3AY41935E, 2014.
  - Opsahl, S. and Benner, R.: Distribution and cycling of terrigenous dissolved organic matter in the ocean, Nature, 386, 480–482, https://doi.org/10.1038/386480a0, 1997.

- Panetta, R. J., Ibrahim, M., and Gélinas, Y.: Coupling a high-temperature catalytic oxidation total organic carbon analyzer to
   an isotope ratio mass spectrometer to measure natural-abundance δ<sup>13</sup>C-dissolved organic carbon in marine and freshwater samples, Anal. Chem., 80, 5232–5239, https://doi.org/10.1021/ac702641z, 2008.
  - Ra, K., Bang, J.-H., Lee, J.-M., Kim, K.-T., and Kim, E.-S.: The extent and historical trend of metal pollution recorded in core sediments from the artificial Lake Shihwa, Korea, Mar. Pollut. Bull., 62, 1814–1821, https://doi.org/10.1016/j.marpolbul.2011.05.010, 2011.
- Raymond, P. A. and Spencer, R. G. M.: Chapter 11 Riverine DOM, in: Biogeochemistry of marine dissolved organic matter (Second Edition), edited by: Hansell, D. A. and Carlson, C. A., 509–535, Academic Press, Boston, 2015.
  - Stedmon, C. A. and Bro, R.: Characterizing dissolved organic matter fluorescence with parallel factor analysis: a tutorial, Limnol. Oceanogr-Meth., 6, 572–579, 2008.
- Stedmon, C. A. and Nelson, N. B.: Chapter 10 The optical properties of DOM in the ocean, in: Biogeochemistry of marine dissolved organic matter (Second Edition), edited by: Hansell, D. A. and Carlson, C. A., 481–508, Academic Press, Boston, 2015.
  - Tedetti, M., Guigue, C., and Goutx, M.: Utilization of a submersible UV fluorometer for monitoring anthropogenic inputs in the Mediterranean coastal waters, Mar. Pollut. Bull., 60, 350–362, https://doi.org/10.1016/j.marpolbul.2009.10.018, 2010
- 385 Troyer, I. D., Bouillon, S., Barker, S., Perry, C., Coorevits, K., and Merckx, R.: Stable isotope analysis of dissolved organic carbon in soil solutions using a catalytic combustion total organic carbon analyzer-isotope ratio mass spectrometer with a cryofocusing interface, Rapid Commun. Mass Spectrom., 24, 365–374, https://doi.org/10.1002/rcm.4403, 2010.
  - Twardowski, M. S., Boss, E., Sullivan, J. M., and Donaghay, P. L.: Modeling the spectral shape of absorbing chromophoric dissolved organic matter, Mar. Chem., 89, 69–88, https://doi.org/10.1016/j.marchem.2004.02.008, 2004.
- 390 Vetter, T. A., Perdue, E. M., Ingall, E., Koprivnjak, J.-F., and Pfromm, P. H.: Combining reverse osmosis and electrodialysis for more complete recovery of dissolved organic matter from seawater, Sep. Purif. Technol., 56, 383–387, https://doi.org/10.1016/j.seppur.2007.04.012, 2007.
  - Wang, X.-C., Chen, R. F., and Gardner, G. B.: Sources and transport of dissolved and particulate organic carbon in the Mississippi River estuary and adjacent coastal waters of the northern Gulf of Mexico, Mar. Chem., 89, 241–256, https://doi.org/10.1016/j.marchem.2004.02.014, 2004.
  - Zepp, R. G., Sheldon, W. M., and Moran, M. A.: Dissolved organic fluorophores in southeastern US coastal waters: correction method for eliminating Rayleigh and Raman scattering peaks in excitation-emission matrices, Mar. Chem., 89, 15–36, https://doi.org/10.1016/j.marchem.2004.02.006, 2004.



410 Figure 1: Map of sampling stations in Sihwa Lake, South Korea.

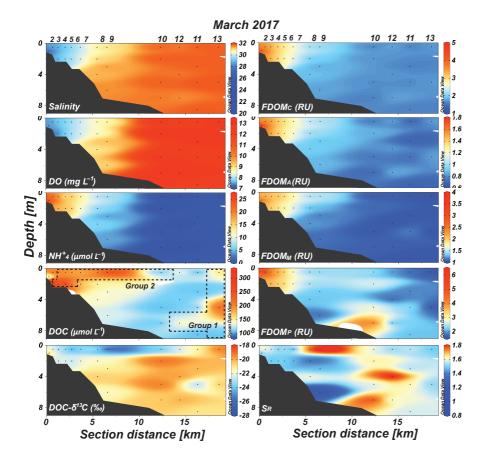


Figure 2: Vertical distributions of salinity, DO, NH<sub>4</sub><sup>+</sup>, DOC, DOC-δ<sup>13</sup>C, FDOM<sub>C</sub>, FDOM<sub>A</sub>, FDOM<sub>B</sub>, FDOM<sub>B</sub>, and S<sub>R</sub> values in Sihwa Lake in March 2017. The dashed blocks represent stations belonging to the Group 1 and the Group 2, which were separated based on the distributions of DOC, DOC-δ<sup>13</sup>C, and salinity.

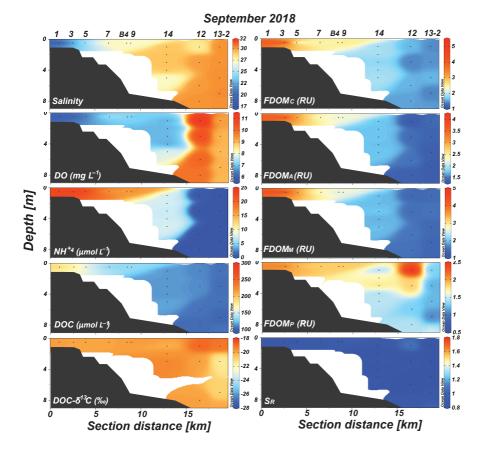


Figure 3: Vertical distributions of salinity, DO,  $NH_4^+$ , DOC, DOC $-\delta^{13}$ C, FDOM<sub>C</sub>, FDOM<sub>A</sub>, FDOM<sub>M</sub>, FDOM<sub>P</sub>, and  $S_R$  values in Sihwa Lake in September 2018.

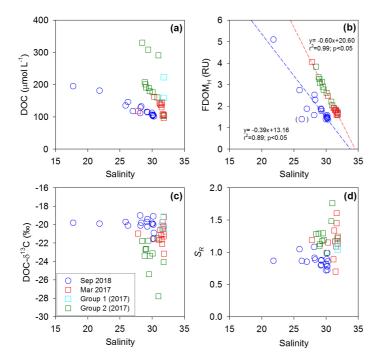


Figure 4: Plots of salinity versus (a) DOC concentrations, (b) FDOM<sub>H</sub> concentrations, (c) DOC $-\delta^{13}$ C values, and (d)  $S_R$  values in Sihwa Lake in March 2017 (red square) and September 2018 (blue circle). The dashed lines represent the regression lines.

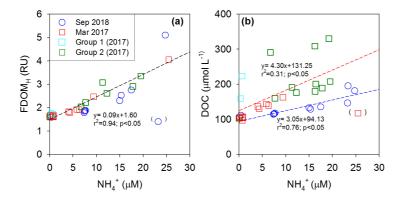


Figure 5: Plots of  $\mathrm{NH_4}^+$  concentrations versus (a) FDOM<sub>H</sub> concentrations and (b) DOC concentrations in Sihwa Lake in March 2017 (red square) and September 2018 (blue circle). The dashed lines represent the regression lines.