



Evaluating the potential for Archaea to serve as ice nucleating particles

Jessie M. Creamean¹, Julio E. Ceniceros³, Lilyanna Newman², Allyson D. Pace², Thomas C.J. Hill¹, Paul J. DeMott1, and Matthew E. Rhodes²

¹Dept. of Atmospheric Science, Colorado State University, Fort Collins, CO, USA

²Dept. of Biology, College of Charleston, SC, USA

³Geological Sciences, University of Texas, El Paso, TX, USA

Correspondence to: Jessie M. Creamean (jessie.creamean@colostate.edu) and Matthew E. Rhodes (rhodesme@cofc.edu)

Abstract. Aerosols play a crucial role in cloud formation. Biologically-derived materials from microorganisms such as plant bacteria, fungi, pollen, and various vegetative detritus can serve as ice nucleating particles (INPs), some of which initiate glaciation in clouds at relatively warm freezing temperatures. However, determining the magnitude of the interactions between clouds and biologically-derived INPs remains a significant challenge due to the diversity and complexity of bioaerosols, and limited observations of such aerosols to facilitate cloud ice formation. Additionally, microorganisms from the domain Archaea have to date not been evaluated as INPs. Here, we present the first results reporting the ice nucleation activity of a subset of archaeal cells from Haloarchaea. Intact cells of *Halococcus morrhuae* and *Haloferax sulfurifontis* demonstrated the ability to induce freezing as warm as –18 °C, while lysed cells of *Haloquadratum walsbyi* and *Natronomonas pharaonis* were unable to serve as warm temperature INPs. Exposure to heat and peroxide digestion indicated that the INPs of intact cells were driven by organic (*H. morrhuae* and *H. sulfurifontis*) and possibly also heat-labile materials (*H. sulfurifontis* only). While halophiles are prominent in hypersaline environments such as the Great Salt Lake and the Dead Sea, other members of the Archaea, such as methanogens and thermophiles, are prevalent in anoxic systems in seawater, sea ice, marine sediments, glacial ice, permafrost, and other cold niches. Archaeal extremophiles are both diverse and highly abundant. Thus, assessing their ability to become airborne, and their abilities to impact cloud formation, is necessary to improve understanding of biological impacts on clouds.

25 1 Introduction

Through their impact on the atmospheric energy budget and hydrological cycle, clouds play a prominent role in shaping Earth's climate at both global and regional scales (Baker and Peter, 2008; Boucher et al., 2013). However, due to the complexity of the microphysical processes associated with cloud formation and dynamics, clouds remain some of the most poorly constrained atmospheric features in climate models. In turn, while atmospheric aerosols strongly impact cloud formation, albedo, lifetime, and precipitation formation processes, disentangling the relationships and feedbacks among aerosols, clouds, and precipitation remains a significant challenge (Sato and Suzuki, 2019; Stevens and Feingold, 2009). In particular, the role of aerosols called





ice nucleating particles (INPs), that induce glaciation in mixed-phase and ice clouds, is highly uncertain relative to other aerosol-cloud processes (Kanji et al., 2017). Improving understanding of such ice formation processes is crucial given most precipitation, globally, is initiated via the ice phase (Lohmann and Feichter, 2005).

In the Earth's troposphere, pure water remains in a supercooled liquid state until below -38 °C. At temperatures greater than -38 °C, the assistance of INPs such as mineral dust, volcanic ash, and select biologically-produced macromolecules (e.g., from pollen, fungi, bacteria, and other sources) is required to initiate the heterogeneous formation of primary ice embryos, that continue to grow into larger ice crystals (Hoose and Möhler, 2012; Kanji et al., 2017). Depending on their surface properties and structural makeup, mineral dust and volcanic ash can raise the freezing point of water from -38 °C to as high as approximately -12 to -15 °C (Murray et al., 2012). At temperatures above -15 °C, and specifically in the immersion freezing 40 regime (i.e., heterogeneous glaciation of a supercooled droplet), the majority of naturally-occurring INPs are biological in origin (Fröhlich-Nowoisky et al., 2016; Hoose and Möhler, 2012; Kanji et al., 2017; Morris et al., 2004). Intact and lysed cell wall fragments (Anderson and Ashworth, 1986; Du et al., 2017; O'Sullivan et al., 2015; Šantl-Temkiv et al., 2015), viable and nonviable cells (Anderson and Ashworth, 1986), and cellular byproduct materials such as exopolymeric substances, saccharides, and biosurfactants (Albers et al., 2017; Decho and Gutierrez, 2017; DeMott et al., 2018; Dreischmeier et al., 2017; Perkins et al., 2020; Zeppenfeld et al., 2019) have all been demonstrated to serve as biologically-derived INPs (Després et al., 2012; Fröhlich-Nowoisky et al., 2016). The most thoroughly-studied biologically-derived INPs of measurable abundance are strains of the bacterium *Pseudomonas syringae*, which is capable of forming ice at temperatures as high as -1 °C (Maki et al., 1974; Morris et al., 2008). The ability of *Pseudomonas syringae* to effectively facilitate ice formation is due to a specific ice nucleating protein (i.e., a protein that is ice nucleation active, designated Ina), that coordinates the crystallization of ice (Cochet and Widehem, 2000; Davies, 2014; Failor et al., 2017; Gurian-Sherman and Lindow, 1993; Hartmann et al., 2013; Šantl-Temkiv et al., 2015). The Ina protein is found in select lineages of bacteria all within the class Gammaproteobacteria (Warren, 1995). There is some evidence that slightly less effective, but highly abundant proteinaceous ice nucleating proteins can be found within fungi, but they remain to be fully categorized (Fröhlich-Nowoisky et al., 2015; Kunert et al., 2019). However, a recent study identifying the first Gram positive ice nucleating bacterium from cloud precipitation suggests that ice nucleating proteins are far more widespread within the bacterial domain than previously thought (Failor et al., 2017).

Even though many laboratory and field-based investigations have alluded to the importance of biologically-derived INPs, the relatively limited available observational data, and a very limited understanding and representation of secondary ice formation processes and their links to biologically-derived INPs in clouds, have caused models to produce equivocal results regarding the global significance of biological ice nucleation in cloud and precipitation formation (Burrows et al., 2013; Hoose et al., 2010b; Hummel et al., 2018; Phillips et al., 2009; Sesartic et al., 2012; Twohy et al., 2016; Vergara-Temprado et al., 2017). Climate models of all scales require information on INP sources to accurately represent ice nucleation and thus cloud microphysics, especially considering: (1) biologically-derived INPs form ice at cloud temperatures as high as -1 °C while





certain mineral dusts, aside some feldspars, glaciate modestly starting at -12 °C and (2) biologically-derived and mineral INP concentrations can vary by several orders of magnitude at any given temperature (Burrows et al., 2013; DeMott et al., 2016; Hoose et al., 2010a; Hoose et al., 2010c; Kanji et al., 2017; McCluskey et al., 2019; Petters and Wright, 2015; Vergara-Temprado et al., 2017).

While there have been ongoing efforts to identify and characterize INPs in the eukaryotic and bacterial domains (Dreischmeier et al., 2017; Failor et al., 2017; Fröhlich-Nowoisky et al., 2016; Hill et al., 2014; Kanji et al., 2017; Kunert et al., 2019; Ling et al., 2018; Morris et al., 2008; Pummer et al., 2012; Qiu et al., 2019), to date, there has been no study that attempts to identify INPs within the domain Archaea. This is in no small part due the fact that, until relatively recently, the Archaea were believed to be largely relegated to marginal existences in extreme environments on our planet. Though recent studies have begun to reveal the true ubiquity and abundance of Archaea in Earth's ecosystems—including seawater, ocean sediment, plankton, soil, marine and terrestrial biofilms, and sea ice, where they can comprise up to 40% of the microbial ecosystem (Cavicchioli, 2011; Flemming and Wuertz, 2019; Hoshino and Inagaki, 2019; Junge et al., 2004; Mondav et al., 2014; Munson et al., 1997; Ochsenreiter et al., 2002; Santoro et al., 2019)—many Archaea remain uncultured and unculturable in laboratory settings, further complicating investigations into their possible propensities to serve as INPs. At the same time, however, the archaeal domain contains both unique cell wall compositions and cell surface structures not present in the other domains (Albers and Meyer, 2011). While appearing highly similar to the bacterial domain in both size and appearance, the archaeal cell envelope is distinct in several ways. In contrast to the Bacteria, most cultured Archaea maintain a proteinaceous surface layer, or Slayer, that provides the cell with structural stability. In many Archaea, the S-layers provide the entirety of the cell envelope. Comparatively few Archaea contain additional cell envelope polymers, and the ones that do, do not produce the near ubiquitous bacterial polymer peptidoglycan. Some Archaea do produce a structurally similar polymer, pseudomurein. The archaeal Slayers are often composed of a single protein or glycoprotein arranged in symmetrical patterns, with hexagonal symmetry being the most common (Albers and Meyer, 2011). As with the Bacteria, many surface exposed proteins are modified in a variety of ways. In addition to the S-layer itself, the archaeal domain contains its own regimen of surface proteins and structures that interact with the external environment. Thus, an entire domain worth of cell surface structures remains to be assessed for potential INP activity.

Here we present a first attempt to assess the potential for members of the domain Archaea to serve as INPs. We initiated our investigation with four members of the obligate halophilic lineage, the Haloarchaea: *Halococcus morrhuae, Haloferax sulfurifontis, Haloquadratum walsbyi*, and *Natronomonas pharaonis*. Together, the four Haloarchaea represented a variety of cell surface designs. Halococcus is one of a limited number of genera belonging to the Archaea that does not possess an Slayer. Instead they possesses a cell envelope that includes highly sulphated heteropolysaccharides (Schleifer et al., 1982). They can be several microns in size, exhibit a cocci (i.e., spheroidal) morphology, and do not lyse in fresh water (Legat et al., 2010). *Haloferax* possesses exclusively a sulphur-rich S-layer and exists as rod- or irregular-shaped cells several microns in size

https://doi.org/10.5194/bg-2020-367 Preprint. Discussion started: 4 November 2020

© Author(s) 2020. CC BY 4.0 License.



100

105



(Elshahed et al., 2004). *Haloquadratum* can produce halomucin in addition to its S-layer and grows in it is trademark square morphology, a couple microns in size (Burns et al., 2007). *Natronomonas* exist as rods several microns in length and prefer alkaline hypersaline environments (Falb et al., 2005).

We opted to initiate our investigation with members of the Haloarchaea for a variety of reasons, including: (1) the diversity of cell surface compositions, (2) the comparative ease of culture, and (3) the regional dominance of the Haloarchaea in relatively large hypersaline environments. While few largescale geographic areas are dominated by members of the domain Archaea, one notable exception is hypersaline bodies of water such as the north basin of the Great Salt Lake and the Dead Sea. These waterbodies extend over hundreds of square kilometres, often contain upwards of 90% Archaea, and can impact the local climate and weather patterns (Carpenter, 1993). Thus, when investigating the impact of potential archaeal INPs in an environment, halophiles and hypersaline offer attractive starting points.

To fully understand the interaction of Earth's climate with the microbial world, it is imperative to include the impact of the archaeal domain, since even when less prevalent, the possession of ice nucleation activity will enable a species to exert an outsized effect on its environs. And to fully understand the potential impact of the archaeal domain on Earth's climate, it is important to assess the potential for members of the Archaea to serve as INPs and contribute to cloud formation.

110 2 Materials and methods

2.1 Cell cultures

Saline media for all four haloarchaeal species tested, *H. morrhuae* (medium 97), *H. sulfurifontis* (medium 1018), *H. walsbyi* (medium 1091), and *N. pharaonis* (medium 371) were prepared according to the recipes provided by the DSMZ-German Collection of Microorganisms and Cell Cultures (https://www.dsmz.de/) with the following alteration: one gram per litre of glycerol was supplemented to medium 1091 for growth of *H. walsbyi*. All cultures were grown at 37 °C and 100 rpm until midlog phase (i.e., midway through the period of exponential cell growth) at the College of Charleston and the purity and cell density were monitored optically at 1000× magnification throughout. Table 1 provides the cell concentrations and salinities of all four prepared cultures. Upon achieving log-phase growth (i.e., the period characterized by cell doubling), cultures were shipped overnight to the Colorado State University (CSU) on ice.

120 **2.2** Preparation of samples for ice nucleation measurements

Due to the high salinities of the media—which would have caused significant freezing point depression (for reference, seawater is typically 30 – 35 ppt (Bodnar, 1993)), samples were diluted to reduced salinities shown in Table 2. This reduction in salinity also inherently caused some of the haloarchaeal cells to lyse. Testing ice nucleation responses from cell lysis is relevant given: (1) cell fragments from fungi and bacteria have been previously observed to serve as INPs (Anderson and Ashworth, 1986;



135

140

145



Du et al., 2017; O'Sullivan et al., 2015; Šantl-Temkiv et al., 2015) and (2) a less saline environment is more atmospherically relevant for how archaea might behave once incorporated in a relatively dilute cloud drop prior to immersion freezing. Cell lysis thresholds were determined by serially diluting cultures with deionized water as shown in Table 2 and microscopically assessing cellular survival. Samples of active, log-phase cultures were diluted 1:15 with deionized water. Additional tests were conducted on *H. morrhuae* cultures diluted to 1:6 and 1:30 with cells remaining intact at all dilutions tested. Cultures were checked microscopically after dilution for cell density and to ascertain if cell lysis had occurred.

2.3 Immersion freezing ice nucleation experiments

INP concentrations were measured using the CSU Ice Spectrometer (IS) (Hiranuma et al., 2015; McCluskey et al., 2018; Suski et al., 2018), which is an immersion freezing measurement device suited to testing aliquots of liquid culture. The IS is constructed using two 96-well aluminium incubation blocks, designed for incubating polymerase chain reaction (PCR) plates, placed end-to-end and encased on their sides and base by cold plates. Immersion freezing temperature spectra were obtained by dispensing 50-μL aliquots of cell suspensions into sterile, 96-well PCR trays in a laminar flow cabinet. Each sample consisted of 24 aliquots. PCR plates were then placed into the blocks of the IS, the device cooled at 0.33°C min⁻¹ to approximately –29 °C using a recirculating low temperature bath, and the freezing of wells was recorded every 0.5 °C. Frozen aliquots were detected with a CCD camera system controlled with LabVIEW (LabVIEW; NI, Inc.). From the number of wells frozen at each temperature step, the fraction frozen is calculated. Standard error was calculated for each spectrum. Deionized water and uninoculated diluted media were run as controls for all species during each experiment.

Thermal and chemical treatments were conducted to isolate heat-labile (e.g., proteinaceous) and organic INPs in the diluted samples. The stability (or lack thereof) of INPs to these treatments provides an indication of composition. To assess the contribution of heat-labile entities, a 1.5-mL aliquot of suspension was tested after heating to 95 °C for 20 min. To remove all organic INPs, 0.75 mL of 30% H₂O₂ was added to a 1.5-mL aliquot of suspension and the mixture heated to 95 °C for 20 min while illuminated with UVB fluorescent bulbs to generate hydroxyl radicals (residual H₂O₂ is removed using catalase (McCluskey et al., 2018)), and the sample retested in the IS. Remaining INPs were possibly aggregates of cellular material that were not fully digested, inorganic INPs in the media, or other biological materials resistant to heat and peroxide digestion.

3 Results and discussion

${\bf 3.1\ Interspecies\ comparison\ of\ haloarchaeal\ ice\ nucleation\ abilities}$

Figure 1 shows the results of the 1:15 dilutions of each of the haloarchaeal species. The media for all four species contributed modestly to background freezing as compared to deionized water controls. *H. morrhuae* performed best as an INP, initiating the freezing of water at temperatures as high as -17.6 °C. *H. sulfurifontis* also demonstrated enhanced capability to serve as an INP, freezing water at temperatures above that of deionized water and sterile media, with a freezing onset of -19.2 °C. These



170

175

180

185



haloarchaea are not as proficient ice nucleators as other more commonly-studied biologically-derived entities such as certain bacteria (up to -1.3 °C) (Kim et al., 1987; Lindow et al., 1989; Maki et al., 1974; Vali et al., 1976), fungi (up to -1 °C) (Kunert et al., 2019; Richard et al., 1996), lichen (up to -2 °C) (Kieft, 1988; Kieft and Ruscetti, 1990), and pollen (up to -8 to -12 °C) (Hader et al., 2014; Pummer et al., 2011). The haloarchaeal species fall closer in line with less effective biologically-derived INPs such as fungal spores (onset freezing temperatures reported up to -10 °C, but typically initiate freezing < -25 °C) (Iannone et al., 2011; Jayaweera and Flanagan, 1982) and phytoplankton (observed up to -3 °C but typically < -20 °C) (Knopf et al., 2011). *H. morrhuae* cells were also more effective at nucleating ice even though total cell concentrations were three times higher for *H. sulfurifontis* (Table 2). One possible explanation is that *H. morrhuae* is unusual among the Archaea in that it has a cell envelop composed of polysaccharides, which have been shown to serve as markers for ice nucleating activity (Zeppenfeld et al., 2019). In general, although not "warm temperature" INPs (i.e., that glaciate > -15 °C), these haloarchaea are comparable in ice nucleation activity to fungal spores and phytoplankton and can thus contribute to INP populations at very relevant atmospheric freezing temperatures.

As expected ice nucleation activity did not occur in all tested haloarchaeal species, just as with bacteria (Karimi et al., 2020; Szyrmer and Zawadzki, 1997). Both *H. walsbyi* and *N. pharaonis* demonstrated no ability to serve as INPs (i.e., the fractions frozen were not higher than the media blanks). Interestingly, the dilution of the log-phase cultures 1:15 with deionized water resulted in the complete lysis of *H. walsbyi* and the partial lysis of *N. pharaonis* as opposed to *H. morrhuae* and *H. sulfurifontis*, which both remained intact. These results indicate that of the species studied, lysed haloarchaeal cells (i.e., cell fragments) do not enhance ice nucleation abilities, and possibly even suppress it.

3.2 Response of haloarchaeal species to heat and peroxide treatments

Each sample (i.e., all dilutions listed in Table 2) was subject to thermal and peroxide treatments to obtain the heat-labile (proteinaceous) and organic frozen fractions in addition to the total (unamended) frozen fractions. Figure 2 shows these spectra for the 1:6, 1:15, and 1:30 dilutions of *H. morrhuae*. The ice nucleating activity of all three dilutions was reduced by heat to the same degree (~0.5 °C) and especially reduced by peroxide, indicating the samples contained some heat-labile, likely proteinaceous, INPs but a more substantial amount of some other biogenic organic INP (Conen et al., 2011; Hill et al., 2016; McCluskey et al., 2018). Interestingly, increasing the dilution, such that the sample contained less culture to deionized water, led to a larger decrease in fraction frozen from the peroxide treatments. For the 1:6 dilution, frozen fractions dropped up to 1.3 °C and 1.9 °C for the heat and peroxide treatments, respectively. For the largest decrease, the 1:30 dilution dropped by up to 3.1 °C and 4.2 °C for heat and peroxide treatments, respectively. One conceivable explanation for the increased efficacy with increasing dilution is that the peroxide (the same volume was used in each sample) remained in higher concentration to digest a lower concentration of cells; thus, less residual organic material was available to serve as INPs. It is possible that higher volumes of peroxide would be needed for higher cell concentrations to eliminate all organic material, depending on the location of the organic material (i.e., if it is extra- or inter-cellular).



190

210

215



For the other three haloarchaeal species, only *H. sulfurifontis* exhibited a decrease in frozen fractions—fraction frozen dropped by up to 3.4 °C and 4.1 °C for heat and peroxide, respectively. This decrease was mostly observed when the fraction frozen was approximately 0.2 to 0.8, as opposed to *H. morrhuae*, where the difference in temperature from the unamended to the treated frozen fractions were roughly equivalent throughout the spectra. Interestingly, *H. sulfurifontis* INP spectra were more responsive to the heat treatment as opposed to peroxide, indicating this species heat-labile, probably proteinaceous, INPs, the opposite of *H. morrhuae*. *H. walsbyi* (lysed) and *N. pharaonis* (partially lysed) showed no response to the treatments (not shown), because there were essentially no INPs in the original dilutions above the background INPs in the media.

4 Conclusions

Here, we present the first reported results on the ice nucleation activity of the domain Archaea. Specifically, we focus on four species within the Haloarchaea due to their diversity of cell surface compositions, ease of culturability, and regional presence within relatively large hypersaline environments. All species were introduced to hyposaline conditions to: (1) negate freezing point depression due to salinity and (2) more accurately reflect the expected environmental conditions of ice nucleation in clouds, as, in theory, aerosolized haloarchaeal cells followed by immersion in cloud drops would yield similar, low salinities. 200 While intact cells of *H. morrhuae* and *H. sulfurifontis* demonstrated ability as INPs, inducing freezing up to -18 °C, lysed cells of H. walsbyi and N. pharaonis did not exhibit ice nucleation activity over the temperature range accessible in the tests. The intact cells that demonstrated ice nucleation activity contained both heat-stable organic ice nucleating entities (H. morrhuae) and heat-labile material (H. sulfurifontis). This may be negated at cloud level salinities where H. sulfurifontis would be expected to lyse as well. H. morrhuae on the other hand remained intact at all salinities. It is also rare among both the 205 Haloarchaea, in particular, and the Archaea, as whole, in that its surface is completely devoid of a proteinaceous S-layer. Other members of both the domain Archaea and the class Haloarchaea have genetic motifs indicative of ice-binding surface proteins. Therefore, further work is needed to directly evaluate the surface properties to disentangle which constituents are responsible for ice formation in the Halococcus, and others within Haloarchaea and Archaea, in general.

While halophilic archaea are prominent in hypersaline environments throughout the globe, such as the Great Salt Lake and the Dead Sea, other members of the domain Archaea such as methanogens and thermophiles are prevalent in anoxic systems in seawater, sea ice, marine sediments, glacial ice, permafrost, hot springs, submarine hydrothermal vents, and hot, dry deserts (Amend and Shock, 2001; Collins et al., 2010; Oremland and Taylor, 1978; Price, 2007; Thauer et al., 2008; Thummes et al., 2007; van der Maarel et al., 1999). However, some studies allude to the fact that these extremophiles are not confined to extreme living conditions, which qualifies them as one of the most abundant prokaryotes on Earth (DeLong, 1998). Thus, these microorganisms are more ubiquitous than one might think, are present in the atmosphere (Fröhlich-Nowoisky et al., 2014), and may affect cloud formation (Amato et al., 2017). Indeed, the order Halobacteriales, which contains *H. morrhuae*, has been found to be present in continental air and was relatively abundant among the Archaea found in marine air (Fröhlich-Nowoisky



220



et al., 2014). Further, Archaea accounted for several percent of all sequences in boundary layer air sampled from 45 – 50 °S over the Southern Ocean (Uetake et al., 2020). Future work should focus on characterizing a wide range of environmentally-relevant Archaea, such as methanogens (see Creamean et al., 2020) and ammonia oxidizers (see Fröhlich-Nowoisky et al., 2014), for their ice nucleating properties and address how they may be important for regional cloud formation, and hence weather and climate, as the impacts of the Archaea on climate processes is currently unknown.

Acknowledgments: The authors would like to acknowledge Prof. Stephen Schmidt and Dr. Pacifica Sommers of the University of Colorado, Boulder for assisting with preservation and culturing of the haloarchaea after arrival in Colorado. J. Ceniceros was supported by the U.S. Department of Commerce, National Oceanic and Atmospheric Administration, Educational Partnership Program under Agreement No. NA16SEC4810006. This work was funded by a NASA EPSCoR Research Infrastructure Development for the state of South Carolina.





Table 1. Medium recipes, initial cell concentrations, and salinities for the four haloarchaeal species after culturing.

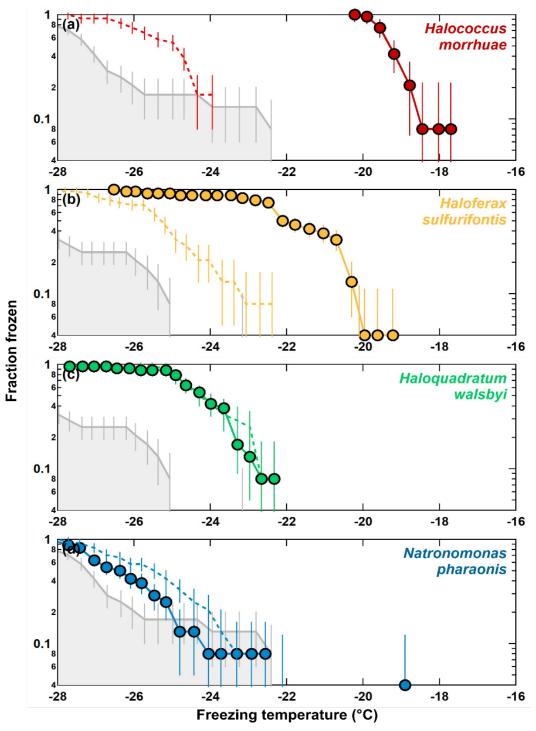
Haloarchaeal species	DSMZ medium recipe	Initial cell concentration (mL ⁻¹)	Salinity in ppt (%)
Halococcus morrhuae	97	$2.7 \times 10^9 *$	160 (16.0%)
Haloferax sulfurifontis	1018	$3.5 \times 10^9 *$	177 (17.7%)
Haloquadratum walsbyi	1091	2.4×10^{8}	210 (21.0%)
Natronomonas pharaonis	371	$1.1 \times 10^9 *$	207 (20.7%)

^{*} Initial cell concentrations were too many to count (TMTC) optically, thus were estimated based on diluted samples, which were 1:15 in deionized water dilution for *H. morrhuae* and 1:6 for *H. sulfurifontis* and *N. pharaonic*.

Table 2. Samples produced for all four haloarchaeal species. Diluted cell concentrations and salinities were calculated based on the volume of culture mixed with media and dilution factor. For the cell concentrations, "n/a" = not applicable due to lysing.

Haloarchaeal species	Cell state	Dilution factor	Diluted cell conc (mL ⁻¹)	Diluted salinity in ppt (%)
Halococcus morrhuae	intact	1:6	4.5×10 ⁸	26.7 (2.7%)
	intact	1:15	1.8×10^{8}	10.7 (1.1%)
	intact	1:30	9.0×10^{7}	5.3 (0.5%)
Haloferax sulfurifontis	intact	1:15	5.8×10^{8}	11.8 (1.2%)
Haloquadratum walsbyi	lysed	1:15	n/a	14.0 (1.4%)
Natronomonas pharaonis	lysed	1:15	n/a	13.8 (1.4%)





240 Figure 1. Freezing spectra for each of the haloarchaeal species diluted 1:15 in deionized water: (a) *H. morrhuae*, (b) *H. sulfurifontis*, (c) *H. walsbyi*, and (d) *N. pharaonis*. *H. morrhuae* and *H. sulfurifontis* did not lyse, *H. walsbyi* lysed, and *N. pharaonis* partially lysed. Dashed lines indicate the diluted sterile media controls and grey shading represents the deionized water blanks processed for each species. Error bars indicate standard error.





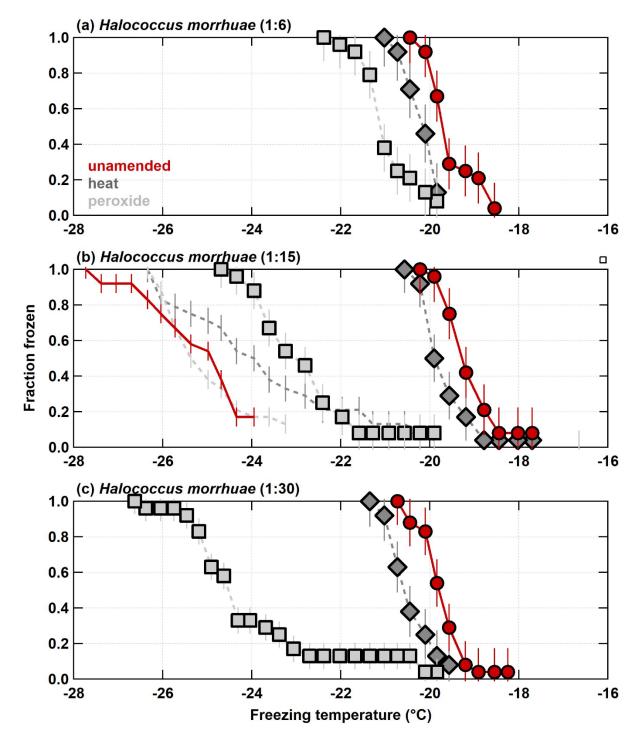


Figure 2. Unamended, heat-labile (heat), and organic (H_2O_2) frozen fractions for each of the three H. morrhuae dilutions from the processing treatments. Plain lines in panel (b) represent spectra for the sterile media controls that also underwent the thermal and peroxide treatments.





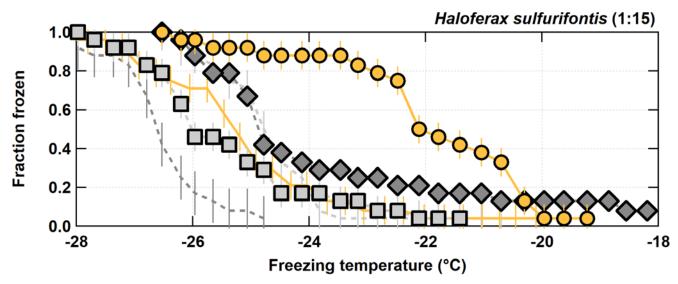


Figure 3. Unamended, heat-labile (heat), and organic (H₂O₂) frozen fractions for each of the 1:15-diluted *H. sulfurifontis* from the processing treatments. Plain lines represent spectra for the sterile media controls that also underwent the thermal and peroxide treatments. *H. walsbyi* and *N. pharaonis* are not shown since they exhibited no INP activity in their unamended spectra.





255 References

265

280

- Albers, S. V., and Meyer, B. H.: The archaeal cell envelope. 2011.
- Albers, S. V., Eichler, J., and Aebi, M.: Chapter 22 Archaea., in: Essentials of Glycobiology (3rd edition), edited by: Varki, A., Cummings, R. D., and Esko, J. D., Cold Spring Harbor Laboratory Press, Cold Spring Harbor, New York, 2017.
- Amato, P., Joly, M., Besaury, L., Oudart, A., Taib, N., Moné, A. I., Deguillaume, L., Delort, A.-M., and Debroas, D.: Active microorganisms thrive among extremely diverse communities in cloud water, PLOS ONE, 12, e0182869, 10.1371/journal.pone.0182869, 2017.
 - Amend, J. P., and Shock, E. L.: Energetics of overall metabolic reactions of thermophilic and hyperthermophilic Archaea and Bacteria, FEMS Microbiology Reviews, 25, 175-243, 10.1111/j.1574-6976.2001.tb00576.x, 2001.
 - Anderson, J. A., and Ashworth, E. N.: The Effects of Streptomycin, Desiccation, and UV Radiation on Ice Nucleation by Pseudomonas viridiflava, 80, 956-960, 10.1104/pp.80.4.956 %J Plant Physiology, 1986.
 - Baker, M. B., and Peter, T.: Small-scale cloud processes and climate, Nature, 451, 299-300, 10.1038/nature06594, 2008.
 - Bodnar, R. J.: Revised equation and table for determining the freezing point depression of H2O-Nacl solutions, Geochimica et Cosmochimica Acta, 57, 683-684, https://doi.org/10.1016/0016-7037(93)90378-A, 1993.
- Boucher, O., Randall, D., Artaxo, P., Bretherton, C., Feingold, G., Forster, P., Kerminen, V.-M., Kondo, Y., Liao, H.,
 Lohmann, U., Rasch, P., Satheesh, S. K., Sherwood, S., Stevens, B., and Zhang, X. Y.: Clouds and Aerosols, in:
 Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report
 of the Intergovernmental Panel on Climate Change, edited by: Stocker, T. F., Qin, D., Plattner, G.-K., Tignor, M.,
 Allen, S. K., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P. M., Cambridge University Press, Cambridge,
 United Kingdom and New York, NY, USA, 571–658, 2013.
- Burns, D. G., Janssen, P. H., Itoh, T., Kamekura, M., Li, Z., Jensen, G., Rodríguez-Valera, F., Bolhuis, H., and Dyall-Smith, M. L.: Haloquadratum walsbyi gen. nov., sp. nov., the square haloarchaeon of Walsby, isolated from saltern crystallizers in Australia and Spain, International journal of systematic and evolutionary microbiology, 57, 387-392, 10.1099/ijs.0.64690-0, 2007.
 - Burrows, S. M., Hoose, C., Pöschl, U., and Lawrence, M. G.: Ice nuclei in marine air: biogenic particles or dust?, Atmos. Chem. Phys., 13, 245-267, 10.5194/acp-13-245-2013, 2013.
 - Carpenter, D. M.: The Lake Effect of the Great Salt Lake: Overview and Forecast Problems, Weather and Forecasting, 8, 181-193, 10.1175/1520-0434(1993)008<0181:TLEOTG>2.0.CO;2, 1993.
 - Cavicchioli, R.: Archaea Timeline of the third domain. 2011.
 - Cochet, N., and Widehem, P.: Ice crystallization by Pseudomonas syringae, Applied Microbiology and Biotechnology, 54, 153-161, 10.1007/s002530000377, 2000.
 - Collins, R. E., Rocap, G., and Deming, J. W.: Persistence of bacterial and archaeal communities in sea ice through an Arctic winter, Environmental Microbiology, 12, 1828-1841, 10.1111/j.1462-2920.2010.02179.x, 2010.
 - Conen, F., Morris, C. E., Leifeld, J., Yakutin, M. V., and Alewell, C.: Biological residues define the ice nucleation properties of soil dust, Atmos. Chem. Phys., 11, 9643-9648, 10.5194/acp-11-9643-2011, 2011.
- 290 Creamean, J. M., Hill, T. C. J., DeMott, P. J., Uetake, J., Kreidenweis, S., and Douglas, T. A.: Thawing permafrost: an overlooked source of seeds for Arctic cloud formation, Environmental Research Letters, 15, 084022, 10.1088/1748-9326/ab87d3, 2020.
 - Davies, P. L.: Ice-binding proteins: a remarkable diversity of structures for stopping and starting ice growth, Trends in Biochemical Sciences, 39, 548-555, 10.1016/j.tibs.2014.09.005, 2014.
- 295 Decho, A. W., and Gutierrez, T.: Microbial Extracellular Polymeric Substances (EPSs) in Ocean Systems, Frontiers in Microbiology, 8, 922, 2017.
 - DeLong, E.: Archaeal Means and Extremes, 280, 542-543, 10.1126/science.280.5363.542 %J Science, 1998.
- DeMott, P. J., Hill, T. C. J., McCluskey, C. S., Prather, K. A., Collins, D. B., Sullivan, R. C., Ruppel, M. J., Mason, R. H., Irish, V. E., Lee, T., Hwang, C. Y., Rhee, T. S., Snider, J. R., McMeeking, G. R., Dhaniyala, S., Lewis, E. R., Wentzell, J. J. B., Abbatt, J., Lee, C., Sultana, C. M., Ault, A. P., Axson, J. L., Diaz Martinez, M., Venero, I., Santos-Figueroa, G., Stokes, M. D., Deane, G. B., Mayol-Bracero, O. L., Grassian, V. H., Bertram, T. H., Bertram, A. K., Moffett, B. F., and Franc, G. D.: Sea spray aerosol as a unique source of ice nucleating particles, 113, 5797-5803, 10.1073/pnas.1514034112 %J Proceedings of the National Academy of Sciences, 2016.



320



- DeMott, P. J., Mason, R. H., McCluskey, C. S., Hill, T. C. J., Perkins, R. J., Desyaterik, Y., Bertram, A. K., Trueblood, Jonathan V., Grassian, V. H., Qiu, Y., Molinero, V., Tobo, Y., Sultana, C. M., Lee, C., and Prather, K. A.: Ice nucleation by particles containing long-chain fatty acids of relevance to freezing by sea spray aerosols, Environmental Science: Processes & Impacts, 20, 1559-1569, 10.1039/C8EM00386F, 2018.
 - Després, V., Huffman, J. A., Burrows, S. M., Hoose, C., Safatov, A., Buryak, G., Fröhlich-Nowoisky, J., Elbert, W., Andreae, M., Pöschl, U., and Jaenicke, R.: Primary biological aerosol particles in the atmosphere: a review, Tellus B: Chemical and Physical Meteorology, 64, 15598-15598, 10.3402/tellusb.v64i0.15598, 2012.
 - Dreischmeier, K., Budke, C., Wiehemeier, L., Kottke, T., and Koop, T.: Boreal pollen contain ice-nucleating as well as ice-binding 'antifreeze' polysaccharides, Scientific Reports, 7, 41890, 10.1038/srep41890, 2017.
 - Du, R., Du, P., Lu, Z., Ren, W., Liang, Z., Qin, S., Li, Z., Wang, Y., and Fu, P.: Evidence for a missing source of efficient ice nuclei, Scientific Reports, 7, 39673, 10.1038/srep39673, 2017.
- Elshahed, M. S., Savage, K. N., Oren, A., Gutierrez, M. C., Ventosa, A., and Krumholz, L. R.: Haloferax sulfurifontis sp. nov., a halophilic archaeon isolated from a sulfide- and sulfur-rich spring, 54, 2275-2279, https://doi.org/10.1099/ijs.0.63211-0, 2004.
 - Failor, K. C., Schmale, D. G., Vinatzer, B. A., and Monteil, C. L.: Ice nucleation active bacteria in precipitation are genetically diverse and nucleate ice by employing different mechanisms, ISME Journal, 11, 2740-2753, 10.1038/ismej.2017.124, 2017.
 - Falb, M., Pfeiffer, F., Palm, P., Rodewald, K., Hickmann, V., Tittor, J., and Oesterhelt, D.: Living with two extremes: conclusions from the genome sequence of Natronomonas pharaonis, Genome research, 15, 1336-1343, 10.1101/gr.3952905, 2005.
- Flemming, H.-C., and Wuertz, S.: Bacteria and archaea on Earth and their abundance in biofilms, Nature Reviews Microbiology, 17, 247-260, 10.1038/s41579-019-0158-9, 2019.
 - Fröhlich-Nowoisky, J., Ruzene Nespoli, C., Pickersgill, D. A., Galand, P. E., Müller-Germann, I., Nunes, T., Gomes Cardoso, J., Almeida, S. M., Pio, C., Andreae, M. O., Conrad, R., Pöschl, U., and Després, V. R.: Diversity and seasonal dynamics of airborne archaea, Biogeosciences, 11, 6067-6079, 10.5194/bg-11-6067-2014, 2014.
- Fröhlich-Nowoisky, J., Hill, T. C. J., Pummer, B. G., Yordanova, P., Franc, G. D., and Pöschl, U.: Ice nucleation activity in the widespread soil fungus <i>Mortierella alpina</i>, Biogeosciences, 12, 1057-1071, 10.5194/bg-12-1057-2015, 2015.
 - Fröhlich-Nowoisky, J., Kampf, C. J., Weber, B., Huffman, J. A., Pöhlker, C., Andreae, M. O., Lang-Yona, N., Burrows, S. M., Gunthe, S. S., Elbert, W., Su, H., Hoor, P., Thines, E., Hoffmann, T., Després, V. R., and Pöschl, U.: Bioaerosols in the Earth system: Climate, health, and ecosystem interactions, Atmospheric Research, 182, 346-376, https://doi.org/10.1016/j.atmosres.2016.07.018, 2016.
 - Gurian-Sherman, D., and Lindow, S. E.: Bacterial ice nucleation: Significance and molecular basis. 1993.
 - Hader, J. D., Wright, T. P., and Petters, M. D.: Contribution of pollen to atmospheric ice nuclei concentrations, Atmospheric Chemistry and Physics, 14, 5433-5449, 10.5194/acp-14-5433-2014, 2014.
- Hartmann, S., Augustin, S., Clauss, T., Wex, H., Šantl-Temkiv, T., Voigtländer, J., Niedermeier, D., and Stratmann, F.: Immersion freezing of ice nucleation active protein complexes, Atmos. Chem. Phys., 13, 5751-5766, 10.5194/acp-13-5751-2013, 2013.
 - Hill, T. C. J., Moffett, B. F., Demott, P. J., Georgakopoulos, D. G., Stump, W. L., and Franc, G. D.: Measurement of ice nucleation-active bacteria on plants and in precipitation by quantitative PCR, Applied and environmental microbiology, 80, 1256-1267, 10.1128/AEM.02967-13, 2014.
- 345 Hill, T. C. J., DeMott, P. J., Tobo, Y., Froehlich-Nowoisky, J., Moffett, B. F., Franc, G. D., and Kreidenweis, S. M.: Sources of organic ice nucleating particles in soils, Atmospheric Chemistry and Physics, 16, 7195-7211, 10.5194/acp-16-7195-2016, 2016.
- Hiranuma, N., Augustin-Bauditz, S., Bingemer, H., Budke, C., Curtius, J., Danielczok, A., Diehl, K., Dreischmeier, K., Ebert, M., Frank, F., Hoffmann, N., Kandler, K., Kiselev, A., Koop, T., Leisner, T., Mohler, O., Nillius, B., Peckhaus, A.,
 Rose, D., Weinbruch, S., Wex, H., Boose, Y., DeMott, P. J., Hader, J. D., Hill, T. C. J., Kanji, Z. A., Kulkarni, G., Levin, E. J. T., McCluskey, C. S., Murakami, M., Murray, B. J., Niedermeier, D., Petters, M. D., O'Sullivan, D., Saito, A., Schill, G. P., Tajiri, T., Tolbert, M. A., Welti, A., Whale, T. F., Wright, T. P., and Yamashita, K.: A comprehensive



375



- laboratory study on the immersion freezing behavior of illite NX particles: a comparison of 17 ice nucleation measurement techniques, Atmospheric Chemistry and Physics, 15, 2489-2518, 10.5194/acp-15-2489-2015, 2015.
- Hoose, C., Kristjansson, J. E., and Burrows, S. M.: How important is biological ice nucleation in clouds on a global scale?, Environmental Research Letters, 5, 10.1088/1748-9326/5/2/024009, 2010a.
 - Hoose, C., Kristjánsson, J. E., and Burrows, S. M.: How important is biological ice nucleation in clouds on a global scale?, Environmental Research Letters, 5, 024009, 10.1088/1748-9326/5/2/024009, 2010b.
- Hoose, C., Kristjansson, J. E., Chen, J. P., and Hazra, A.: A Classical-Theory-Based Parameterization of Heterogeneous Ice Nucleation by Mineral Dust, Soot, and Biological Particles in a Global Climate Model, Journal of the Atmospheric Sciences, 67, 2483-2503, 10.1175/2010jas3425.1, 2010c.
 - Hoose, C., and Möhler, O.: Heterogeneous ice nucleation on atmospheric aerosols: a review of results from laboratory experiments, Atmos. Chem. Phys., 12, 9817-9854, 10.5194/acp-12-9817-2012, 2012.
 - Hoshino, T., and Inagaki, F.: Abundance and distribution of Archaea in the subseafloor sedimentary biosphere, The ISME Journal, 13, 227-231, 10.1038/s41396-018-0253-3, 2019.
 - Hummel, M., Hoose, C., Pummer, B., Schaupp, C., Fröhlich-Nowoisky, J., and Möhler, O.: Simulating the influence of primary biological aerosol particles on clouds by heterogeneous ice nucleation, Atmos. Chem. Phys., 18, 15437-15450, 10.5194/acp-18-15437-2018, 2018.
- Iannone, R., Chernoff, D. I., Pringle, A., Martin, S. T., and Bertram, A. K.: The ice nucleation ability of one of the most abundant types of fungal spores found in the atmosphere, Atmos. Chem. Phys., 11, 1191-1201, 10.5194/acp-11-1191-2011, 2011.
 - Jayaweera, K., and Flanagan, P.: Investigations on biogenic ice nuclei in the Arctic atmosphere, Geophys Res Lett, 9, 94-97, 10.1029/GL009i001p00094, 1982.
 - Junge, K., Eicken, H., and Deming, J. W.: Bacterial Activity at -2 to -20°C in Arctic Wintertime Sea Ice, 70, 550-557, 10.1128/AEM.70.1.550-557.2004 %J Applied and Environmental Microbiology, 2004.
 - Kanji, Z. A., Ladino, L. A., Wex, H., Boose, Y., Burkert-Kohn, M., Cziczo, D. J., and Krämer, M.: Overview of ice nucleating particles, Meteorological Monographs, 58, 1.1-1.33, 10.1175/AMSMONOGRAPHS-D-16-0006.1, 2017.
- Karimi, B., Nosrati, R., Fazly Bazzaz, B. S., Mirpour, M., Malboobi, M., and Owlia, P.: A comparative evaluation of freezing criteria and molecular characterization of epiphytic ice-nucleating (Ice+) and non-ice-nucleating (Ice-) Pseudomonas syringae and Pseudomonas fluorescens, Journal of Plant Pathology, 102, 169-178, 10.1007/s42161-019-00402-7, 2020.
 - Kieft, T. L.: Ice Nucleation Activity in Lichens, 54, 1678-1681, 1988.
 - Kieft, T. L., and Ruscetti, T.: Characterization of biological ice nuclei from a lichen, 172, 3519-3523, 10.1128/jb.172.6.3519-3523.1990 %J Journal of Bacteriology, 1990.
- Kim, H. K., Orser, C., Lindow, S. E., and Sands, D. C.: Xanthomonas campestris pv. translucens strains active in ice nucleation, Plant Dis, 71, 994-997, 10.1094/pd-71-0994, 1987.
 - Knopf, D. A., Alpert, P. A., Wang, B., and Aller, J. Y.: Stimulation of ice nucleation by marine diatoms, Nature Geoscience, 4, 88-90, 10.1038/ngeo1037, 2011.
- Kunert, A. T., Pöhlker, M. L., Tang, K., Krevert, C. S., Wieder, C., Speth, K. R., Hanson, L. E., Morris, C. E., Schmale Iii, D. G., Pöschl, U., and Fröhlich-Nowoisky, J.: Macromolecular fungal ice nuclei in Fusarium: effects of physical and chemical processing, Biogeosciences, 16, 4647-4659, 10.5194/bg-16-4647-2019, 2019.
 - Legat, A., Gruber, C., Zangger, K., Wanner, G., and Stan-Lotter, H.: Identification of polyhydroxyalkanoates in Halococcus and other haloarchaeal species, Applied Microbiology and Biotechnology, 87, 1119-1127, 10.1007/s00253-010-2611-6, 2010.
- Lindow, S. E., Lahue, E., Govindarajan, A. G., Panopoulos, N. J., and Gies, D.: Localization of ice nucleation activity and the iceC gene product in Pseudomonas syringae and Escherichia coli, Molecular plant-microbe interactions: MPMI, 2, 262-272, 10.1094/mpmi-2-262, 1989.
 - Ling, M. L., Wex, H., Grawe, S., Jakobsson, J., Löndahl, J., Hartmann, S., Finster, K., Boesen, T., and Šantl-Temkiv, T.: Effects of Ice Nucleation Protein Repeat Number and Oligomerization Level on Ice Nucleation Activity, Journal of Geophysical Research: Atmospheres, 123, 1802-1810, 10.1002/2017JD027307, 2018.
 - Lohmann, U., and Feichter, J.: Global indirect aerosol effects: a review, Atmos. Chem. Phys., 5, 715-737, 10.5194/acp-5-715-2005, 2005.



430



- Maki, L. R., Galyan, E. L., Chang-Chien, M. M., and Caldwell, D. R.: Ice nucleation induced by pseudomonas syringae, Applied microbiology, 28, 456-459, 1974.
- McCluskey, C. S., Hill, T. C. J., Sultana, C. M., Laskina, O., Trueblood, J., Santander, M. V., Beall, C. M., Michaud, J. M., Kreidenweis, S. M., Prather, K. A., Grassian, V., and DeMott, P. J.: A Mesocosm Double Feature: Insights into the Chemical Makeup of Marine Ice Nucleating Particles, Journal of the Atmospheric Sciences, 75, 2405-2423, 10.1175/Jas-D-17-0155.1, 2018.
- McCluskey, C. S., DeMott, P. J., Ma, P. L., and Burrows, S. M.: Numerical Representations of Marine Ice-Nucleating Particles in Remote Marine Environments Evaluated Against Observations, Geophys Res Lett, 46, 7838-7847, 10.1029/2018gl081861, 2019.
 - Mondav, R., Woodcroft, B. J., Kim, E.-H., McCalley, C. K., Hodgkins, S. B., Crill, P. M., Chanton, J., Hurst, G. B., VerBerkmoes, N. C., Saleska, S. R., Hugenholtz, P., Rich, V. I., and Tyson, G. W.: Discovery of a novel methanogen prevalent in thawing permafrost, Nature Communications, 5, 3212, 10.1038/ncomms4212, 2014.
- Morris, C. E., Georgakopoulos, D. G., and Sands, D. C.: Ice nucleation active bacteria and their potential role in precipitation, J. Phys. IV France, 121, 87-103, 10.1051/jp4:2004121004, 2004.
 - Morris, C. E., Sands, D. C., Vinatzer, B. A., Glaux, C., Guilbaud, C., Buffière, A., Yan, S., Dominguez, H., and Thompson, B. M.: The life history of the plant pathogen Pseudomonas syringae is linked to the water cycle, ISME J, 2, 321-334, 10.1038/ismej.2007.113, 2008.
- 420 Munson, M. A., Nedwell, D. B., and Embley, T. M.: Phylogenetic diversity of Archaea in sediment samples from a coastal salt marsh, 63, 4729-4733, 1997.
 - Murray, B. J., O'Sullivan, D., Atkinson, J. D., and Webb, M. E.: Ice nucleation by particles immersed in supercooled cloud droplets. 2012.
- O'Sullivan, D., Murray, B. J., Ross, J. F., Whale, T. F., Price, H. C., Atkinson, J. D., Umo, N. S., and Webb, M. E.: The relevance of nanoscale biological fragments for ice nucleation in clouds, Scientific Reports, 5, 8082, 10.1038/srep08082, 2015.
 - Ochsenreiter, T., Pfeifer, F., and Schleper, C.: Diversity of Archaea in hypersaline environments characterized by molecular-phylogenetic and cultivation studies, Extremophiles, 6, 267-274, 10.1007/s00792-001-0253-4, 2002.
 - Oremland, R. S., and Taylor, B. F.: Sulfate reduction and methanogenesis in marine sediments, Geochimica et Cosmochimica Acta, 42, 209-214, https://doi.org/10.1016/0016-7037(78)90133-3, 1978.
 - Perkins, R. J., Vazquez de Vasquez, M. G., Beasley, E. E., Hill, T. C. J., Stone, E. A., Allen, H. C., and DeMott, P. J.: Relating Structure and Ice Nucleation of Mixed Surfactant Systems Relevant to Sea Spray Aerosol, The Journal of Physical Chemistry A, 10.1021/acs.jpca.0c05849, 2020.
- Petters, M. D., and Wright, T. P.: Revisiting ice nucleation from precipitation samples, Geophys Res Lett, 42, 8758-8766, 10.1002/2015gl065733, 2015.
 - Phillips, V. T. J., Andronache, C., Christner, B., Morris, C. E., Sands, D. C., Bansemer, A., Lauer, A., McNaughton, C., and Seman, C.: Potential impacts from biological aerosols on ensembles of continental clouds simulated numerically, Biogeosciences, 6, 987-1014, 10.5194/bg-6-987-2009, 2009.
- Price, P. B.: Microbial life in glacial ice and implications for a cold origin of life, FEMS Microbiology Ecology, 59, 217-231, 10.1111/j.1574-6941.2006.00234.x, 2007.
 - Pummer, B. G., Bauer, H., Bernardi, J., Bleicher, S., and Grothe, H.: Birch and conifer pollen are efficient atmospheric ice nuclei, 11, 27219, 10.5194/acpd-11-27219-2011, 2011.
 - Pummer, B. G., Bauer, H., Bernardi, J., Bleicher, S., and Grothe, H.: Suspendable macromolecules are responsible for ice nucleation activity of birch and conifer pollen, Atmospheric Chemistry and Physics, 12, 2541-2550, 10.5194/acp-12-2541-2012, 2012.
 - Qiu, Y., Hudait, A., and Molinero, V.: How Size and Aggregation of Ice-Binding Proteins Control Their Ice Nucleation Efficiency, Journal of the American Chemical Society, 141, 7439-7452, 10.1021/jacs.9b01854, 2019.
 - Richard, C., Martin, J. G., and Pouleur, S.: Ice nucleation activity identified in some phytopathogenic Fusarium species, Phytoprotection, 77, 83-92, https://doi.org/10.7202/706104ar, 1996.
- 450 Šantl-Temkiv, T., Sahyoun, M., Finster, K., Hartmann, S., Augustin-Bauditz, S., Stratmann, F., Wex, H., Clauss, T., Nielsen, N. W., Sørensen, J. H., Korsholm, U. S., Wick, L. Y., and Karlson, U. G.: Characterization of airborne ice-nucleation-



470

480



- active bacteria and bacterial fragments, Atmospheric Environment, 109, 105-117, https://doi.org/10.1016/j.atmosenv.2015.02.060, 2015.
- Santoro, A. E., Richter, R. A., and Dupont, C. L.: Planktonic Marine Archaea, Annual Review of Marine Science, 11, 131-158, 10.1146/annurev-marine-121916-063141, 2019.
 - Sato, Y., and Suzuki, K.: How do aerosols affect cloudiness?, Science, 363, 580, 10.1126/science.aaw3720, 2019.
 - Schleifer, K. H., Steber, J., and Mayer, H.: Chemical Composition and Structure of the Cell Wall of Halococcus morrhuae, Zentralblatt für Bakteriologie Mikrobiologie und Hygiene: I. Abt. Originale C: Allgemeine, angewandte und ökologische Mikrobiologie, 3, 171-178, https://doi.org/10.1016/S0721-9571(82)80030-6, 1982.
- 460 Sesartic, A., Lohmann, U., and Storelvmo, T.: Bacteria in the ECHAM5-HAM global climate model, Atmospheric Chemistry and Physics, 12, 8645-8661, 10.5194/acp-12-8645-2012, 2012.
 - Stevens, B., and Feingold, G.: Untangling aerosol effects on clouds and precipitation in a buffered system, Nature, 461, 607-613, 10.1038/nature08281, 2009.
- Suski, K. J., Hill, T. C. J., Levin, E. J. T., Miller, A., DeMott, P. J., and Kreidenweis, S. M.: Agricultural harvesting emissions of ice-nucleating particles, Atmospheric Chemistry and Physics, 18, 13755-13771, 10.5194/acp-18-13755-2018, 2018.
 - Szyrmer, W., and Zawadzki, I.: Biogenic and Anthropogenic Sources of Ice-Forming Nuclei: A Review, Bulletin of the American Meteorological Society, 78, 209-228, 10.1175/1520-0477(1997)078<0209:BAASOI>2.0.CO;2, 1997.
 - Thauer, R. K., Kaster, A.-K., Seedorf, H., Buckel, W., and Hedderich, R.: Methanogenic archaea: ecologically relevant differences in energy conservation, Nature Reviews Microbiology, 6, 579-591, 10.1038/nrmicro1931, 2008.
 - Thummes, K., Schäfer, J., Kämpfer, P., and Jäckel, U.: Thermophilic methanogenic Archaea in compost material: Occurrence, persistence and possible mechanisms for their distribution to other environments, Systematic and Applied Microbiology, 30, 634-643, https://doi.org/10.1016/j.syapm.2007.08.001, 2007.
- Twohy, C. H., McMeeking, G. R., DeMott, P. J., McCluskey, C. S., Hill, T. C. J., Burrows, S. M., Kulkarni, G. R., Tanarhte, M., Kafle, D. N., and Toohey, D. W.: Abundance of fluorescent biological aerosol particles at temperatures conducive to the formation of mixed-phase and cirrus clouds, Atmos. Chem. Phys., 16, 8205-8225, 10.5194/acp-16-8205-2016, 2016.
 - Uetake, J., Hill, T. C. J., Moore, K. A., DeMott, P. J., Protat, A., and Kreidenweis, S. M.: Airborne bacteria confirm the pristine nature of the Southern Ocean boundary layer, Proceedings of the National Academy of Sciences, 117, 13275, 10.1073/pnas.2000134117, 2020.
 - Vali, G., Christensen, M., Fresh, R. W., Galyan, E. L., Maki, L. R., and Schnell, R. C.: Biogenic Ice Nuclei. Part II: Bacterial Sources, Journal of the Atmospheric Sciences, 33, 1565-1570, 10.1175/1520-0469(1976)033<1565:BINPIB>2.0.CO;2, 1976.
- van der Maarel, M. J. E. C., Sprenger, W., Haanstra, R., and Forney, L. J.: Detection of methanogenic archaea in seawater particles and the digestive tract of a marine fish species, FEMS Microbiology Letters, 173, 189-194, 10.1111/j.1574-6968.1999.tb13501.x, 1999.
 - Vergara-Temprado, J., Murray, B. J., Wilson, T. W., O'Sullivan, D., Browse, J., Pringle, K. J., Ardon-Dryer, K., Bertram, A. K., Burrows, S. M., Ceburnis, D., DeMott, P. J., Mason, R. H., O'Dowd, C. D., Rinaldi, M., and Carslaw, K. S.: Contribution of feldspar and marine organic aerosols to global ice nucleating particle concentrations, Atmos. Chem. Phys., 17, 3637-3658, https://doi.org/10.5194/acp-17-3637-2017, 2017.
 - Warren, G. J.: Biological Ice Nucleation and Its Applications, edited by: Lee, R. E., Warren, G. J., and Gusta, L. V., American Phyotpathological Society Press, St. Paul, MN, 1995.
- Zeppenfeld, S., van Pinxteren, M., Hartmann, M., Bracher, A., Stratmann, F., and Herrmann, H.: Glucose as a Potential Chemical Marker for Ice Nucleating Activity in Arctic Seawater and Melt Pond Samples, Environmental Science & Technology, 53, 8747-8756, 10.1021/acs.est.9b01469, 2019.