<u>Seasonality of nitrogen s</u>ources, and cycling and loading of nitrogen inin a New England river discerned from nitrate isotope ratios

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Abstract

- 15 from the Pawcatuck River. We conducted weekly monitoring of nutrients and nitrate (NO₃⁻) isotope ratios (¹⁵N/¹⁴N, ¹⁸O/¹⁶O and ¹⁷O/¹⁶O) at the mouth of the river and from the larger of two Waste Water Treatment Facilities (WWTFs) along the estuary, as well as seasonal along-river surveys. Our observations reveal a direct relationship between N loading and the magnitude of river discharge, and a consequent seasonality to N loading into the estuary rendering loading from the WWTFs and
- 20 from an industrial site upriver more important at lower river flows during warmer months, comprising ~23 % and ~18 % of N loading, respectively. Riverine nutrients derived predominantly from deeper groundwater and the industrial point source upriver during low base flow in summer, and from shallower groundwater and surface flow at higher river flows during colder months <u>- wherein-Loading of dissolved organic nitrogen appeared to increase with river discharge, ostensibly delivered</u>
- 25 by surface water. The NO₃⁻ associated with deeper groundwater had higher ¹⁵N/¹⁴N ratios than shallower groundwater, consistent with the expectation fractionation due to partial denitrification. Corresponding NO₃⁻ ¹⁸O/¹⁶O ratios were lower during the warm season, due to increased biological cycling in-river. Uncycled atmospheric NO₃⁻, detected from its unique mass-independent NO₃⁻¹⁷O/¹⁶O vs. ¹⁸O/¹⁶O fractionation, accounted for < 3 % of riverine NO₃⁻, even at elevated discharge. Along-river,
- 30 NO₃⁻¹⁵N/¹⁴N ratios showed a correspondence to regional land use, increasing from agricultural and forested catchments to the more urbanized watershed downriver, with the agricultural and urbanized portions of the watershed contributing disproportionately to total N loading. Corresponding NO₃⁻¹⁸O/¹⁶O ratios were lower during the warm season, a dynamic that we ascribe to increased biological cycling in-river. The evolution of The ¹⁸O/¹⁶O isotope ratios along-river were
- 35 <u>conformed to consistent with</u> the notion of nutrient spiraling, reflecting <u>the NO3</u>-input <u>of NO3</u> from the <u>watershed</u>-<u>catchment</u> and <u>from</u> in-river nitrification and its coincident removal by biological consumption. <u>These findings stress the importance of considering seasonality of riverine N sources</u>

and loading Uncycled atmospheric NO₃⁻, detected from its unique mass-independent NO₃⁻¹⁴⁷O/¹⁶O vs.
 ¹⁸O/¹⁶O fractionation, accounted for < 3 % of riverine NO₃⁻, even at elevated discharge. to We explore
 the implications of our findings for the mitigation eof eutrophication in Little Narragansett
 Bayreceiving estuaries. Our study further advances a conceptual framework that reconciles with the current theory of riverine nutrient cycling, from which to robustly interpret NO₃⁻ isotope ratios to constrain cycling and source partitioning in river systems.

1. Introduction

- 45 Human activities have resulted in a substantial increase in the delivery of nutrients from terrestrial to aquatic and marine systems (Gruber and Galloway, 2008). In marine systems, increased loading of reactive nitrogen (N) has resulted in coastal eutrophication, engendering the loss of valuable nearshore habitat such as seagrass beds and oyster reefs, depletion of dissolved oxygen (creating so-called "dead zones"), and increased frequency and severity of algal blooms – including
- 50 toxic brown and red tides causing fish kills (Heisler et al., 2008). In densely populated areas like the northeast United States, excess anthropogenic nitrogen loads originate from Waste Water Treatment Facilities (WWTFs), septic systems, industrial discharge, fertilizer applied to turf and agricultural lands, and atmospheric sources from industry and fossil fuel use (Valiela et al., 1997; McClelland et al., 2003, Latimer and Charpentier, 2010). The pervasive degradation of coastal marine ecosystems is alarming and of significant concern to coastal communities worldwide.

The transfer of nutrients from land to the coast is facilitated by rivers, which constitute an effective pipeline that collects nutrients from the watershed, ultimately discharging these to the coast. The mitigation of estuarine eutrophication thus relies on identifying primary sources of nutrients to riverine systems. Nutrients are fundamentally delivered to rivers from non-point sources:

- 60 from waters entering the river via surface runoff, sub-surface groundwater in the unsaturated zone, and groundwater within the water table. Nutrients also enter rivers from point sources, including WWTFs as well as industrial discharge, which can dominate N loading in urbanized watersheds (Howarth et al., 1996). The nutrient loads contained in surface and deeper groundwater entering rivers differ markedly depending on land use. In temperate pristine systems, soil and groundwater
- 65 concentrations are generally low, with reactive N originating from atmospheric deposition, biological N₂ fixation in soils, and from N in rocks and minerals (Hendry et al. 1984; Holloway et al. 1998; Morford et al., 2016). Higher concentrations of reactive N are found in waters draining agricultural and urbanized areas (Dubrovsky et al., 2010; Baron et al., 2013).

The N loaded to the watershed is partially attenuated through biological cycling in soils and
aquifers. Specifically, organic N is degraded to reduced N species that are oxidized (nitrified) to nitrate (NO₃⁻) in oxygenated zones of groundwater. NO₃⁻ is otherwise removed from anoxic groundwater by denitrification, reduced to inert N₂. Reactive N is further cycled and attenuated in-river: The hyporheic zone, where groundwater interchanges with stream and river water, creates a complex environment that can stimulate nitrification and denitrification, as oxic and anoxic pockets exist in close proximity (Sebilo et al., 2003; Harvey et al., 2013). Reactive nitrogen can be further attenuated by benthic denitrification within the river channel (Sebilo et al., 2003; Kennedy et al., 2008; Mulholland et al., 2008).

Identifying sources of N to rivers can be difficult due to the expanse and heterogeneity of the watershed, the long integration time of deeper groundwater, and the degree of biological N cycling
 in groundwater and in-river. While measurements of N concentrations along the river channel in relation to regional land use can offer insights in this regard, N sources can be further resolved using complementary measurements of the naturally occurring N and oxygen (O) isotope ratios of riverine NO₃⁻ (¹⁵N/¹⁴N and ¹⁸O/¹⁶O, respectively). Henceforth, we express the isotope ratios in delta notation:

$$\delta (\%_0) = \left(\frac{isotope\ ratio\ of\ sample}{isotope\ ratio\ of\ reference} - 1\right) \times 1000$$
(1)

- The reference for δ¹⁵N is N₂ in air, and for δ¹⁸O is Vienna Standard Mean Ocean Water (VSMOW). The N and O isotope ratios of NO₃⁻ provide constraints on N sources and cycling in part because respective N sources cover discrete ranges of δ¹⁵N and δ¹⁸O values (Kendall et al. 2007). Reactive N species from atmospheric deposition, biological N₂ fixation, and industrial N₂ fixation share overlapping ranges of δ¹⁵N values (≤ 0 ‰), which differ appreciably from those of livestock and human waste (8 25 ‰;
 Kendall, 1998; Böhlke, 2003; Xue et al., 2009). In contrast, the δ¹⁸O signatures of atmospheric NO₃⁻ (60 80 ‰) are distinct from those of industrial NO₃⁻ (~25 ‰) and from NO₃⁻ produced by nitrification, which aligns closely with that of ambient water ((≤ 1 ‰; Boshers et al. 2019 and references therein). Atmospheric NO₃⁻ is further distinguished by a mass-independent δ¹⁷O vs. δ¹⁸O fractionation that is not manifest in industrial and biological NO₃⁻ (Savarino and Thiemens, 1999).
- The isotope ratios of NO₃⁻ also provide constraints on N cycling because N and O isotopologues are differentially sensitive to respective biological N transformations (*reviewed by* Casciotti, 2016), implicating different mass balance considerations within the N cycle that permit differentiation of N sources from cycling. Briefly, in riverine systems where NO₃⁻ is the dominant N pool, $\delta^{15}N_{NO3}$

integrates across values of reactive N delivered from the watershed, minus NO₃⁻ removed by benthic denitrification (if associated with N isotopic fractionation; Sebilo et al., 2003). Values of $\delta^{15}N_{NO3}$ are additionally sensitive to isotopic fractionation due to internal cycling in-river – assimilation and remineralization to NO₃⁻ via nitrification – in systems where riverine N is otherwise partitioned comparably between oxidized and reduced pools (*i.e.*, NO₃⁻ vs. ammonium and particulate N; Sebilo et al., 2006). Riverine $\delta^{18}O_{NO3}$, in turn, integrates across values of exogenous NO₃⁻ delivered to the river from the watershed and from atmospheric deposition, those of NO₃⁻ produced in-river by nitrification, minus the NO₃⁻ lost concurrently to denitrification and assimilation (*see* Sigman et aland <u>Fripiat</u>,- 2019). Interpreted in tandem, NO₃⁻ N and O isotopologue ratios thus offer complementary constraints to identify important source terms and characterize inherent-cycling.

- Here we present a study of annual N loading from the Pawcatuck River to the Little Narragansett
 Bay in southern New England (U.S.A.), wherein we exploit measurements of the N and O isotope ratios of riverine NO₃⁻ to draw inferences on dominant N sources from the watershed and on riverine N cycling. The site is heavily impacted by nitrogen loading as evidenced by the history of the habitat: Vast seagrass beds of *Zostera marina* (eelgrass) historically established in Little Narragansett Bay were overtaken in the early 1990's by extensive mats of filamentous macroalgae dominated by the
- 115 Cladophoraceae clade, whose substantial biomass has been linked to frequent events of night-time hypoxia in the bay's shallow-water coves (Dodds and Gudder. 1992; Dillingham et al., 1993; D'avanzo and Kremer 1994; Tiner et al., 2003; Berezina and Golubkov, 2008; National Water Quality Monitoring Council, 2020). The evident eutrophication of the estuary has raised questions regarding the magnitude of N loading from the Pawcatuck River and from the local WWTFs, whose respective
- 120 contributions must be assessed in order to devise targets for mitigation. To this end, we conducted weekly monitoring of nutrients and NO₃⁻ isotopologue ratios at the mouth of the Pawcatuck River and of nutrients discharged from the larger of two WWTFs along the estuary, as well as parallel measurements of samples collected from seasonal along-river surveys. Utilizing NO₃⁻ isotopologue ratios to identify N sources has immediate local implications for management of the watershed,
- 125 allows for extrapolation to similar watersheds throughout the temperate zone, and most importantly, isolates the seasonal and flow-dependent nature of N cycling within a riverine system transitioning to an estuarine system. This last finding has direct relevance to water quality modeling efforts in temperate estuaries.

2. Methods

130 2.1 Site Description

The Pawcatuck River watershed (\sim 760 km²) is located predominantly in the state of Rhode Island (RI) with as small portion in eastern Connecticut (Figure 1). The river originates at Worden Pond in



Figure 1. <u>Map of (a) Map of the Pawcatuck River watershed (URIEDC_RIGIS, 2019) and associated land use (<u>URIEDC_RIGIS, 2019</u>; U.S. Geological Survey, 2011)._-and (b) <u>Map of sampling locations, landmarks, including the</u> <u>Westerly and Pawcatuck waste water treatment facilities (WWTF)</u> and <u>river</u> discharge gauges along the Pawcatuck River (U.S. Geological Survey, 2005; U.S. Census Bureau, 2017).</u> Wakefield, RI, and extends 47 km southwest to Westerly, RI. It is joined by the Wood River, which originates in northern RI and runs 29 km south to Wood River Junction. The drainage basin is mostly

- 135 flat, hosting terrain with forests and wetlands (73 %) and relatively low human population (~56,400; based on a dasymetric analysis of the 2010 U.S. Census Bureau population data in the watershed; Vaudrey et al., 2017) owing in part to state land trust holdings that protect ~22 % of the watershed in RI from development (Dillingham et al. 1992; U.S. Geological Survey, 2011). Agricultural areas comprise 8 % of land use (U.S. Geological Survey, 2011) and are mostly located in the upper
- 140 watershed, which hosts a number of turf farms: In 2005, Washington county where the Pawcatuck River originates – was noted as having the highest density of turf farms in the United States (U.S. Environmental Protection Agency, 2005). Urbanized and developed land usage comprises 13 % of the total watershed with the majority of the urban areas concentrated on the lower 19 km portion of the river, between Bradford and Westerly (U.S. Geological Survey, 2011).
- 145 Three nutrient discharge permits are allotted along the river by the RI Department of Environmental Management (RI DEM; Figure 1): Kenyon Industries, a fabric processing plant, is located approximately 7 km downstream from Worden Pond. Two WWTFs discharge into the estuary and are located 1 km downstream of the Westerly Bridge, approximately 47 km downstream of Worden Pond.
- 150 2.2 Sample collection

We conducted four distinct sampling regimens: (a) weekly river samplings at the mouth of the river, (b) weekly WWTF effluent samplings, (c) seasonal along-river surveys, and (d) rainwater samplings. We collected weekly river samples (a) from January 10, 2018 through to January 12, 2019 at two sites: The Stillman Bridge near the mouth of the freshwater portion of the river, and ~1 km downstream at the Westerly Bridge, which marks the limit of seawater intrusion (Figure 1, S1). (b) We obtained samples of wastewater treatment effluent collected weekly at the Westerly Waste Water Treatment Facility (W-WWTF) from June 6, 2018 to May 22, 2019. (c) We conducted three seasonal along-river surveys on May 21, 2018, November 9, 2018, and March 12, 2019 at 15 discrete sampling stations between Worden Pond and the Westerly Bridge. Additionally, we performed a highly resolved sampling (approximately every 0.75 km) of the lower river from Potter Hill Dam (Station 11) to Westerly (Station 15) aboard kayaks in May 2017 (Figure 1). No samples were collected directly from the retention ponds or their outflow at Kenyon Industries (Figure 1). (d) We collected

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rainwater samples following rain events from a rooftop collector at the Avery Point campus in Groton, CT (approximately 18 km west of the Pawcatuck River), from September 6, 2018 to December 2, 2018, in order to define regional NO₃⁻ isotopic endmembers.

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Weekly samplings at the Stillman and Westerly bridges occurred around sunrise, before the onset of photosynthetic activity, whereas along-river samples were collected sequentially from sunrise to mid-day. During each sample collection, river temperature and dissolved oxygen concentrations were measured *in situ* with a Thermo Orion Star A123 portable dissolved oxygen meter. At each site, river

- 170 water was collected at ~0.5 m depth with a Van Dorn bottle and transferred into a 5 L carboy for transport, on ice, back to the laboratory for processing. In the laboratory, the conductivity of each sample was measured with an Oakton CON 450 conductivity meter. Sub-samples for analyses of dissolved nutrient and NO₃⁻ isotope ratios were filtered through pre-combusted 25 mm GF/F glass fiber filters and collected in acid washed polypropylene bottles, then stored at -20°C pending analysis.
- 175 The filters were placed in pre-combusted aluminum foil and frozen at -20°C in preparation for particulate nitrogen isotope ratio analyses. Samples for chlorophyll-a analysis were similarly collected onto 25 mm GF/F filters.

The weekly effluent samples at the Westerly WWTF were collected by facility personnel into 0.5 L acid-washed polypropylene bottles and frozen pending monthly pick-ups by our team. Two types of samples were collected on a weekly basis: grab and composite samples. Grab samples correspond to treated effluent collected prior to its release to the river, while composite samples are effluent collected continually over a 24-hour period, thus providing a concentration-weighted daily average. In the laboratory, samples for nutrient analysis were thawed and filtered through a 25 mm GF/F filter and frozen at -20 °C pending analysis. Samples for particulate N analysis were not collected from the WWTF.

Rainwater samples were collected into trace-metal-clean 1-L Teflon bottles outfitted with a glass funnel to create a vapor lock preventing evaporation. These samples were stored unfiltered at -20°C pending nutrient and NO_3^- isotope ratio analyses.

2.3 Nutrient analyses

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The NO₃⁻ concentration, [NO₃⁻], in river and WWTF samples was measured by conversion to nitric oxide in a hot Vanadium III solution followed by detection on a chemiluminescent NOx analyzer (Teledyne[™]; Braman and Hendrix, 1989). Incident nitrite in the samples was first reacted with Griess 195

reagents (Strickland and Parsons, 1972) before injection into the hot Vanadium (III) solution in order to detect NO₃⁻ only. The concentration of nitrite, [NO₂⁻], in river samples was measured by conversion to nitric oxide in hot iodine solution, followed by detection on the chemiluminescent NO_x analyzer (Garside, 1982). For the rainwater samples, $[NO_3^-]$ and $[NO_2^-]$ were measured on a SmartChem discrete nutrient autoanalyzer (Unity Scientific[™]) using standard protocols adapted for the SmartChem (Strickland and Parsons, 1972; U. S. Environmental Protection Agency, 1993b; 4500-NO₂, 2018; 4500-NO₃⁻, 2018). Concentrations of ammonium, $[NH_4^+]$, and phosphate, $[PO_4^{3-}]$, in river and 200 WWTF samples were measured on a SmartChem autoanalyzer using standard protocols (Murphy and Riley, 1962; Strickland and Parsons, 1972; U.S. Environmental Protection Agency, 1978, 1993; 4500-NH3, 2018; 4500-P, 2018).

The concentration of total dissolved nitrogen, [TDN], in filtered river and WWTF samples was measured by persulfate oxidation to NO₃, then measured via chemiluminescent NOx analyzer as 205 described above (Sólorzano and Sharp, 1980; Knapp et al, 2005). The persulfate reagent was first recrystallized following protocol by Grasshoff et al. (1999). A ratio of sample to reagent of 5 to 10 was used in the oxidations. Reagent blanks accounted for \leq 0.3 % of the TDN signal. The incident concentration of dissolved organic nitrogen, [DON] was calculated as the difference between [TDN] and dissolved inorganic nitrogen, [DIN], where $[DIN] = [NO_3^-] + [NO_2^-] + [NH_4^+]$.

210 2.4 Chlorophyll-a analyses

> Chlorophyll-a was extracted from duplicate 25 mm GF/F filter samples in 5 mL of 90 % acetone, incubated overnight at -20°C and quantified by florescence detection on a Turner Trilogy Laboratory Fluorometer (Arar and Collins, 1997).

2.5 NO₃⁻ Isotope ratio analyses

- 215 The nitrogen and oxygen isotope ratios of NO₃⁻, ¹⁵N/¹⁴N, ¹⁸O/¹⁶O, and ¹⁷O/¹⁶O were analyzed using the denitrifier method in samples where $[NO_3] \ge 1.5 \mu M$ (Sigman et al. 2001; Casciotti et al, 2002; Kaiser et al. 2007). Briefly, NO_3^- was converted quantitatively to a nitrous oxide (N_2O) analyte by denitrifying bacteria that lack a terminal reductase (*Pseudomonas chlororaphis* f. sp. aureofaciens; ATCC[®] 13985[™]), followed by analysis of the N₂O product at the University of Connecticut on a Thermo
- 220 Delta V GC-IRMS prefaced with a custom-modified Gas Bench II device with two cold traps and a PAL autosampler (Casciotti et al., 2002). The NO3⁻¹⁷O/¹⁶O in rainwater (as well as ¹⁸O/¹⁶O) was similarly

analyzed by bacterial conversion to N_2O , followed by pyrolysis in a gold tube to N_2 and O_2 and analysis on a Thermo Delta V GC-IRMS at Brown University (Kaiser et al. 2007).

Coupled $\delta^{15}N_{\text{NO3}}$ and $\delta^{18}O_{\text{NO3}}$ analyses at UConn and Brown University were calibrated from parallel analyses of NO₃⁻ reference materials USGS-34 (δ¹⁵N: -1.8 ‰ vs. air; δ¹⁸O: -27.9 ‰ vs. VSMOW) 225 and IAEA-N3 (δ^{15} N: +4.7 ‰ vs. air; δ^{18} O: +25.6 ‰ vs. VSMOW). Samples were analyzed in triplicate among two or more batch analyses. Reproducibility averaged 0.2‰ for $\delta^{15}N_{NO3}$ and 0.3‰ for $\delta^{18}O_{NO3}$. Coupled analyses of $\delta^{18}O_{NO3}$ and $\delta^{17}O_{NO3}$ of rainwater NO₃⁻ and some of the river samples were calibrated with USGS-34 (Δ^{17} O: -0.1 ‰ vs. VSMOW) and USGS-35 (δ^{18} O +57.5 ‰ vs. VSMOW; Δ^{17} O: +21.6 ‰ vs. VSMOW). The mass independent fractionation of NO₃^{- 17}O vs. ¹⁸O- (Δ^{17} O vs. VSMOW) is 230 calculated from Thiemens (1999):

$$\Delta^{17} O = \delta^{17} O - 0.52 \times \delta^{18} O \tag{2}$$

The analytical reproducibility for $\Delta^{17}O_{NO3}$ averaged 0.3% based upon the pooled standard deviation of repeated measures of reference materials. The fraction (%) of atmospheric NO₃⁻ in river water was 235 derived from a two-end-member mixing equation of river water NO₃⁻ ($\Delta^{17}O$ = 0) with the corresponding atmospheric NO₃⁻ Δ^{17} O value (19.7 to 27.2 ‰; Section S1), with an associated uncertainty of ~1 % based on the pooled standard deviations of Monte Carlo error propagations.

2.6 Particulate nitrogen analyses

Particulate nitrogen (PN) in river samples was collected on pre-combusted 25 mm GF/F glass fiber 240 filters that were freeze dried then compacted into tin capsules for analysis on a Costech Elemental Analyzer connected to a Thermo Delta V isotope ratio mass spectrometer via a Conflow IV interface. Samples were calibrated with aliquots of recognized reference materials USGS 40 and 41 ($\delta^{15}N$ = -4.52 and +47.57 ‰ vs. air, respectively), achieving an analytical precision of ~0.3 ‰.

2.7 Nutrient flux estimates

245 Instantaneous nutrient fluxes were estimated from the product of the nutrient concentration and the corresponding mean daily river discharge recorded by USGS gauges, or discharge reported by the Westerly WWTF. Given the large number of available river flow data from the USGS gauge at the Stillman Bridge and effluent discharge from the WWTF in relation to comparatively fewer concentration data, annual fluxes (\underline{L}) of respective constituents were calculated using Beale's ratio estimator (Beale 1962; Quiblé et al. 2006), which includes a bias correction factor (the term in 250 parentheses) to accountaccounts for the covariance between load and river flow values (Eq. 3):

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$$L = CQ \frac{\mu_q}{Q} n \left(\frac{1 + \frac{1}{n_d} \frac{SQ}{QQQ}}{1 + \frac{1}{n_d} \frac{QQ}{QQ}} \right)$$

$$S_{CQ} = \frac{1}{n_{s-1}} \left(\sum_{i=1}^{n_d} C_i Q_i - n_d QCQ \right)$$
(3a)

$$S_{Q^2} = \frac{1}{n_d - 1} \left(\sum_{i=1}^{n_d} Q_i - n_d Q^2 \right)$$
(3c)

255 The term μ_q is the mean of all river discharge measurements, C_i is the concentration on day i, Q_i the average river discharge on day i, n the total number of days for the period of load estimation and n_d is the number of observations of C_i. Overbars denote sample arithmetic means, and L is the resulting load. We further conducted bootstrap analyses to provide estimates of the uncertainty of the respective fluxes.

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260 3. Results

3.1 Weekly river samplings

The concentration of NO3⁻ measured in samples collected weekly at the Stillman Bridge was lowest in winter and highest in the summer months, ranging from to 9.7 μ M to as high as 73.5 μ M, with a median value of 30.4 μM (Figure 2a). Comparable concentrations were detected at the 265 Westerly Bridge at each sampling, except for instances where the site experienced saltwater intrusions, evidenced by elevated conductivities (data not shown) - at which times [NO3-] at the Westerly Bridge was lower due to lower concentrations in the seawater endmember. The concentration of NO2⁻ was negligible in all samples. At both bridge sites, [NO3⁻] decreased with increasing river discharge (Figure 2b; Table 1). The [NO₃] at the Stillman Bridge, upstream of potential 270 seawater intrusion, also correlated directly with conductivity (Figure 3a). Values of $\delta^{15}N_{NO3}$ were lowest in winter and increased in summer, ranging from 5.3 ‰ to 9.4 ‰ - thus decreasing with increasing river discharge (Figure 2c-d; Table 1). Values of $\delta^{18}O_{NO3}$ followed a contrasting trend, being lower during the summer months and increasing in winter months, with values ranging from 1.6 ‰ to 6.8 ‰, barring notwithstanding a single an outlying value of 8.1 ‰ (Figure 2e). Values of $\delta^{18}O_{NO3}$ 275 at the bridges increased directly with discharge (Figure 2f; Table 1). Measurements of $\Delta^{17}O_{NO3}$ at the Stillman Bridge ranged from -0.5 to 1.9 %. The fraction of atmospheric NO₃⁻ estimated based on the $\Delta^{17}O_{NO3}$ values of precipitation recorded at Avery Point (Section S1; Fig. S2) indicated that uUncycled atmospheric NO3⁻ was not detected in the majority of the river samples analyzed, with only 10 of 41 samples showing values above our lower limit of detection of ~ 1 % atmospheric NO₃⁻. The fraction of

280 atmospheric NO₃⁻ was otherwise < 3%, notwithstanding a single sample in which atmospheric NO₃⁻ accounting for ~7 % of total riverine NO₃⁻ (Figure 2g, S2; Section S1). Values of $\Delta^{17}O_{NO3}$ nevertheless correlated with river discharge (Figure 2h; Table 1).



Figure 2. Weekly measurements of solute concentrations and NO₃⁻ isotopologue ratios at the Stillman and Westerly bridges vs. the sampling date (superposed onto river discharge), and vs. the mean daily river discharge recorded at the Stillman Bridge. The secondary axis on left-hand panels is the river discharge (x 10⁶ m³ d⁻¹). [NO₃⁻¹] vs. (a) sampling date and (b) discharge; $\delta^{15}N_{NO3}$ vs. (c) sampling date and (d) discharge; $\delta^{18}O_{NO3}$ vs. (e) sampling date and (f) discharge; $\Delta^{17}O_{NO3}$ vs. (g) sampling date and (h) discharge; [NH₄⁺¹] vs. (i) sampling date and (j) [NH₄⁺¹] vs. discharge; [PO₄³⁻¹] vs. (k) sampling date and (l) discharge; [DON] vs. (m) sampling date and (n) [DON] vs. discharge. Statistical fits of least-squares linear regressions are reported in Table 1.

The concentration of NH₄⁺ recorded weekly at the bridges was consistently lower than corresponding 285 $[NO_3^{-1}]$. In contrast to $[NO_3^{-1}]$, $[NH_4^{+1}]$ at the bridges was lowest in summer and higher in winter, ranging from below detection to 7.8 μ M, and correlated directly with discharge (Figure 2i-j). The [PO₄³⁻] ranged from 0.1 μ M to 2.7 μ M with one sample as high as 5.9 μ M during a single sampling event, exhibiting higher concentrations occurring in summer months, thus mirroring $[NO_3^-]$ (Figure 2k). Concentrations of PO₄³⁻ appeared to correlate inversely with discharge, yet only at the Westerly Bridge but not the Stillman Bridge (Figure 2I; Table 1).

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The concentration of DON at the bridge sites ranged from 9 to 56 μ M, appeared similar among seasons, and did not show a statistically significant relationship to river discharge (Figure 2m-n; Table 1). Nevertheless, [DON] and coincident [DIN] were inversely correlated, albeit weakly so, and significantly so only at the Stillman Bridge (Figure 3b; Table 1). In turn, [PN] exhibited median values of 2.6 µM from May through Oct, and 2.9 µM during the colder season, showing no values greater than 7 µM; no correlation of [PN] with river discharge was evident (Figure S3a-b; Table 1). Concentrations of chlorophyll-a, which we measured only from June through December, ranged from 0.5 µg L⁻¹ to 12.1 µg L⁻¹, with higher values occurring in late summer to early fall. Chlorophyll-a showed no correlation with discharge (Figure S3e-f; Table 1).



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The daily riverine flux of dissolved inorganic nitrogen (DIN) delivered to the estuary from the Pawcatuck River, computed from the product of river discharge and the sum of $[NO_3^-]$ and $[NH_4^+]$



Figure 3. (a) [NO₃⁻] in weekly samples at the Stillman Bridge vs. conductivity (log scale). (b) Weekly [DON] at the Stillman and Westerly bridges vs. [DIN].

recorded at the bridges, varied ~10-fold over the annual sampling period, ranging from 0.1 to 1.1 (x 10⁵) moles of N_{DIN} per day – omitting a single outlier of 1.8 x 10⁵ moles of N_{DIN} per day (Figure 4a). The riverine DIN flux increased directly with river discharge, such that it was lowest in summer, averaging 0.2 ± 0.1 (x 10⁵) moles of N_{DIN} per day from May through October (Figure 4b; Table 1). The riverine DON flux, in turn, ranged from < 0.1 to 2.0 (x 10⁵) moles of N_{DON} per day, and also increased directly with discharge (Figure 4c-d; Table 1). The total riverine N flux (TN flux), which is the sum of respective DIN, DON and PN fluxes, ranged from 0.2 to 3.0 (x 10⁵) moles of N_{TN} per day and correlated directly with discharge (Figure 4e-f; Table 1).

310 3.2 WWTF samples

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Nutrient concentrations measured in samples collected weekly at the Westerly WWTF, consisting of both grab and composite samples, ranged from 30 to 527 μ M for [NO₃⁻], 1.3 to 1070 μ M for [NH₄⁺], 11.7 to 1168 μ M for [DON], and 2.7 to 26.5 μ M for [PO₄³⁻] (Figure 5a, c, e, g). Concentrations of NO₃⁻ and NH₄⁺ were highly correlated similar in grab vs. composite samples

- 315 (Figure S4a-b). The [NO₃⁻] and [NH₄⁺] measured at UConn were similar to those reported by the Westerly WWTF (Figure S4c-d). The [DON] measured at UConn showed poor correspondence to the facility-reported [TON] (total organic nitrogen) for the few corresponding sampling dates, although these sample types may arguably not be comparable as the UConn analyses did not include [PN] (Figure S4e).
- Both [NO₃⁻] and [PO₄³⁻] were higher in summer months when facility discharge was lower, at which time [NH₄⁺] was lower. The concentration of NO₃⁻ correlated inversely with the facility-reported discharge, whereas [NH₄⁺] correlated directly with discharge (Table 1). There was an apparent increase in [DON] with discharge, albeit, with high variability during high flow in winter months, whereas facility-reported [TON] did not correlate with discharge (Figure 5f; Table 1). Our limited
 [PO₄³⁻] measurements were not significantly correlated with facility-reported discharge (Figure 5h;
- Table 1).

In contrast to the riverine N fluxes, which increased with river discharge, the DIN and TON fluxes from the Westerly WWTF were remarkably constant, and were substantially lower than corresponding riverine fluxes, averaging 3.2×10^3 moles of N_{DIN} per day, 1.0×10^3 moles of N_{TON} per day, and 4.1×10^3 moles of N_{TN} per day in 2018 (Figure 4 a-f; Table S1). The daily TN loading at the

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Westerly WWTF was notably lower than the permitted allowable daily discharge from May through November of 13.5×10^4 moles of N_{TN} per day.



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Х	Y	Location	Slope	Intercept	r ²	p-value
	[NO -] (NA)	S	-9.1	50.6	0.50	**
	[ΝΟ ₃] (μινι)	W	-7.9	47.8	0.44	**
	\$15N1 (0()	S	-0.3	7.9	0.16	**
	0 ⁻³ N _{NO3} (‰)	W	-0.3	7.9	0.15	**
	S180 (0()	S	0.5	2.7	0.34	**
	0 ⁻⁵ U _{NO3} (‰)	W	0.4	2.9	0.31	**
	Δ ¹⁷ O _{NO3} (‰)	S	0.1	-0.1	0.15	*
	[S	0.6	0.5	0.27	**
	[NH4 ⁺] (μΝΙ)	W	0.7	0.6	0.29	**
	(DO 3-1 (DA)	S	0.0	0.8	0.00	0.78
River	[PO₄³] (μM)	W	-0.1	1.0	0.09	*
Discharge (x 10 ⁶ m ³ d ⁻¹)		S	-0.2	34.2	0.15	<u>0.12**</u>
	[DON] (μΜ)	W	-0.3	33.7	0.08	0.10
		S	0.4	2.4	0.09	0.07
	[PN] (µM)	W	0.3	2.8	0.06	0.18
	r	S	-0.3	3.7	0.01	0.64
	[chl-a] (µg L ⁻⁺)	W	-0.5	4.7	0.03	0.45
		S	2.6 x 10 ⁴	Forced zero	0.85	**
	DIN flux (mol d ⁻¹)	W	2.7 x 10 ⁴	Forced zero	0.86	**
		S	3.0 x 10 ⁴	Forced zero	0.91	**
	DON flux (mol d ⁻¹)	W	2.7 x 10 ⁴	Forced zero	0.94	**
		S	6.0 x 10 ⁴	Forced zero	0.93	**
	IN flux (mol d ⁻⁺)	W	6.0 x 10 ⁴	Forced zero	0.95	**
		S	-0.2	34.0	0.17	**
[DIN] (µM)	[DON] (µM)	W	-0.3	33.7	0.08	0.10
		WWTF comp	-268	514	0.43	**
	[NO₃ ⁻] (µM)	WWTF comp-r	-240	439	0.31	**
		WWTF grab	-268	544	0.42	**
		WWTF comp	232	-83	0.09	0.10
W-WWTF	[NH₄⁺] (μM)	WWTF comp-r	449	-256	0.26	**
Discharge		WWTF grab	141	-75	0.15	*
(x 10 ⁴ m³ d⁻¹)	DON (uM)	WWTF comp	262	-8	0.10	*
	υ σττ (μ)	WWTF grab	102	31	0.07	0.18
	TON (μM)	WWTF comp-r	-11	115	0.00	0.75
	[PO₄³-] (µM)	WWTF comp	5.0 x 10 ⁻²	8.0	0.05	0.30
	1 J (F- 7	WWTF grab	4.0 x 10 ⁻²	3.7	0.08	0.12
	δ ¹⁵ N _{NO3} (‰)	S	2.6 x 10 ⁴	6.6	0.43	**
1/(NO ₃ ⁻ flux)		W	1.9 x 10 ⁴	6.8	0.39	**
(d mol ⁻¹)	δ ¹⁸ Ο _{NO3} (‰)	S	-2.1 x 10 ⁴	4.2	0.25	**
	- 10	W	-1.0×10^4	4.7	0.26	**
	δ¹ ⁸ O _{NO3} -corr (‰)	S	-1.6 x 10 ⁴	3.8	0.26	**

Table 1. Correlation coefficients, corresponding intercepts, coefficients of determination (r^2) and statistical probability of least-squared regression analyses from property-property plots of riverine solutes and fluxes. Statistically significant relationships are signaled by an asterisks (p-value $\leq 0.05^*$; $\leq 0.01^{**}$).

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Figure 5. Nutrient concentrations discharged from the Westerly WWTF: $[NO_3^-]$ vs. (a) date and (b) facility discharge; $[NH_4^+]$ vs. (c) date and (d) facility discharge; [DON] and facility-reported [TON] vs. (e) date and (f) facility discharge; $[PO_4^{3^-}]$ vs. (g) date and (h) facility discharge. Grey line corresponds to the WWTF mean daily water discharge with reference to secondary axis on left-hand panels (x 10^4 m³ d⁻¹). Statistical fits of least-squares linear regressions are reported in Table 1.

3.3 Along-river samplings

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Samples collected at stations along the length of the river showed both spatial and seasonal patterns in nutrients and NO_3^- isotope ratios (Figure 6, S5). On average, $[NO_3^-]$ varied differed among sampling dates ($F_{2,12}$ = 122.4, p < 0.0001) and was lower during the November 2018 sampling than

during the May 2018 and March 2019 samplings at all river sites (Tukey HSD, both p < 0.05; Figure 6b; Table S2). $[NO_3^-]$ tended to increase along river sections ($F_{3,9} = 32.1$, p < 0.0001), but the specific patterns varied among sampling dates ($F_{6,12}$ = 107.2, p < 0.0001). In the source basin at Worden Pond,

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 $[NO_3]$ ranged from 0.4 to 6 μ M among sampling events and increased to date-specific maxima of 10 - 65 µM between Stations 2 and 4 (Biscuit City Road to Wood River Junction). Concentrations decreased downstream of the Wood River inflow (between river sections 2 and 3) to values as low as 7 μ M in November and as high as 29 μ M in March (Tukey HSD, both p < 0.01), although these two sections of the river were similar in May 2018 (Tukey HSD, both p > 0.9). [NO₃⁻] increased between Potter Hill Dam (Station 11) and the Stillman and Westerly bridges (between sections 3 and 4) during 350 all sampling campaigns (Tukey HSD, all p < 0.05), with final concentrations of 10 μ M in November and 32 µM in March (Figure 6b).

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Values of $\delta^{15}N_{NO3}$ differed among samplings dates (F_{2.15} = 16.6, p < 0.001) and along the river (F_{3.9} = 26.2, p < 0.001; Figure 6c). On average, values were lowerst in March 2019, at which time [NO₃-] 355 was relatively elevated, than in May and November 2018 (Tukey HSD: both p < 0.001), although a sample in the uppermost river section (Station 2) had higher $\delta^{15}N_{NO3}$ values in March than in May (see Discussion). Values during the March 2019 sampling ranged from 3.4 ‰ at Station 3 to 6.7 ‰ at the bridges (river section 4). Values in November 2018, which were similar to those in May 2018, ranged from 5.8 % at Station 3 to 8.5 % at the bridges. NO₃⁻ delivered by the Wood River (Station 6) had 360 $\delta^{15}N_{NO3}$ values similar to or greater than those of NO_3^- originated upstream in the Pawcatuck River.

In contrast to $\delta^{15}N_{NO3}$, $\delta^{18}O_{NO3}$ values tended to decrease downriver (F_{3,9} = 8.6, p < 0.01; Figure 6d), despite relatively large variability. Relative maxima between 3.2 and 5.0 ‰ were apparent at Stations 3 and 4 (river section 2), decreasing to values to values oscillating between 2.7 and 4.8 %toward the bridges ($F_{3,9}$ = 8.6, p < 0.01; Figure 6d). The $\delta^{18}O_{NO3}$ values upriver were generally higher 365 in November (in contrast to $\delta^{15}N_{NO3}$) but otherwise occupied comparable ranges among sampling dates. Values contributed by the Wood River were similar to or marginally greater than those upstream in the Pawcatuck River on corresponding dates.

The concentration of NH₄⁺ did not vary systematically across river sections ($F_{3,9} = 3.2$, p = 0.08), ranging from 0.4 to 6.8 μ M (Figures S5). [NH₄⁺] was greater during the May 2018 sampling than during March or November, and this effect did not vary significantly across the river ($F_{6,12} = 2.1$, p = 0.12).

The concentration of PO_4^{3-} varied over both space ($F_{3,9}$ = 45.2, p < 0.0001) and time ($F_{2,12}$ = 72.0, p < 0.0001), and these effects were non-additive (F_{6,12} = 32.2, p < 0.0001). [PO₄³⁻] was relatively homogeneous across the river sections in May 2018 (Tukey HSD, all pairwise p > 0.05) but increased in river section 2 in both March and November compared to neighboring sections up and downstream (Tukey HSD, all p < 0.01; Figure 6e). Across all sampling dates, $[PO_4^{3-}]$ ranged from 0.2 to 0.5 μ M in and near Worden Pond (river section 1) and peaked at values between 0.7 and 2.9 μ M, in river section 2. Further downstream, $[PO_4^{3-}]$ ranged from 0.5 and 1.2 μ M. $[PO_4^{3-}]$ in the Wood River (Station 6) was relatively low and similar to that at Worden.

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Figure 6. Solute concentrations and nitrate isotopologue ratios observed at discrete locations along the Pawcatuck River from its origin at Worden Pond to the Westerly Bridge during along-river sampling campaigns. (a) Mean daily river discharge recorded at three flow gauges along the Pawcatuck River during the sampling campaigns; (b) [NO₃⁻], (c) δ^{15} N_{NO3}, (d) δ^{18} O_{NO3}, and (e) [PO₄³⁻] measured at stations along-river. Open circles designate the lower Wood River that inflows to the Pawcatuck.

4. Discussion

380 4.1 Nutrient source attribution

At the Stillman and Westerly bridges, concentrations of NO₃⁻ – the principal component of DIN – scaled inversely with discharge, wherein higher concentrations occurred during summer at low base flow. This relationship suggests the bulk of riverine DIN during low base flow originated from groundwater and point sources along the river catchment. Given that there is only one documented point source upstream of Stillman and Westerly bridges, we surmise that DIN at low base flow originated predominantly from groundwater and partially from discharge at Kenyon Industries. That [NO3-] at the Stillman Bridge increased in proportion to conductivity also suggests a groundwater source for bulk riverine nutrients at low base flow, although an analogous trend would could admittedly arise from loading by a point sources.

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During wetter months in winter, increased input of shallow groundwater and surface runoff (henceforth collectively referred to as "shallow flow") diluted the low base flow $[NO_3^-]$, thus lowering riverine concentrations, a dynamic documented in other temperate rivers (Mulholland and Hill 1997; Dubrovski et al. 2010). Nevertheless, the daily DIN flux increased with discharge, indicating that DIN is also imported to the river by shallow flow, albeit, at a lower concentration than low base flow DIN. 395 From the slope of the DIN flux to discharge relationship, the daily DIN flux increased by 2.6 ± 0.3 -(x 10^4)-moles of N per additional 10^6 m³ of discharge, suggesting the DIN concentration of shallow flow from the catchment averaged $26 \pm 3 \,\mu$ M (Table 1). The relationship between [DIN] and river discharge, which we initially presumed linear, is then better described by a two end-member mixing curve comprised of low base flow [DIN] mixing with shallow flow [DIN] (Figure 7), assuming a 60 µM end-

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 $[DIN]_i = [(Q_i - 2.20 \times 10^8) * 26 \times 10^{-6} + 2.20 \times 10^8 * 6460 \times 10^{-6}]/Q_i$

member approximated from the median [DIN] at low base flow:

The term Q_i is mean river discharge on day *i* in units of L d⁻¹, from which the subtracted value of 2.2 x 10⁸ L d⁻¹ is the asymptote of low base flow (i.e., the lowest river discharge observed in 2018), and [DIN]_i is the corresponding concentration in units of moles L⁻¹. The low base flow [DIN] end-member of $64 \pm 9 \,\mu$ M derives from the best fit of the equation to the data. Implicit in Eq. 4 is the assumption of negligible in-river N consumption, a notion supported by the low incident [PN]; we return to this dynamic further below. The mixing relationship can serve to approximate the DIN flux from the Pawcatuck River into Little Narragansett Bay based on the river discharge recorded continually by the Formatted: Not Highlight

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(4)

USGS at the Stillman Bridge.-<u>—The annual DIN load returned by this function based on the</u>
 continuous (15-min interval) discharge records at the Stillman Bridge for 2018 is 2<u>1.4 x 10⁶ mol N yr</u>.
 <u>1</u>, slightly higher than our direct estimate of 20.2 (± 2.0) x 10⁶ mol N yr.
 <u>1</u>, vet reassuringly within the uncertainty (Table 2).

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The inverse correlation of [DON] with [DIN], in turn, suggests that [DON] is transported into the river by shallow ground water and surface flow from the catchment. Shallow flow, which increases with increased precipitation, is apt to transport organic material from soils and surface plant materials (Elwood and Turner, 1989; Mulholland et al. 1990; Pabich et al. 2001). The import of DON by shallow flow is consistent with the visibly elevated concentrations of riverine tannins. In this regard, the lack of direct correlation of [DON] to discharge is surprising, but may be masked by the relatively high variability of the [DON] measurements, even between replicate water samples.



Figure 7: Mixing curve of low base flow $[NO_3]$ with shallow flow $[NO_3]$ superposed onto weekly measurements of $[NO_3^-]$ at the vs. the corresponding mean daily discharge at the Stillman Bridge (Eq. 4).

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- Nutrient loading from the Pawcatuck River into Little Narragansett Bay was investigated previously by Fulweiler and Nixon (2005). As discerned herein, they observed an inverse relationship of [DIN] to discharge from biweekly measurements at the Stillman Bridge over an annual cycle. Contrary to our interpretations, however, they argued that the decline in [DIN] with discharge was due to seasonal uptake by vegetation within the catchment, specifically during spring. They observed the lowest [DIN] in spring, corresponding to the highest discharge during their annual study period.
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Here, we otherwise argue that increased water discharge dilutes the low base-flow nutrients derived

from groundwater and point source discharge, such that concentrations are most elevated at low base flow. While the concentration is lower during periods of high river flow, the riverine DIN flux nevertheless increases with discharge, carrying nutrients imported by shallow flow.

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Fulweiler and Nixon (2005) also observed that [DON] and [DIN] were inversely correlated, as in the current study, and further detected a positive correlation between [DON] and discharge, corroborating our earlier inference that such a relationship should be manifest. They reasoned that the greater remineralization of bioavailable DON in summer, at low discharge, could explain this trend, given the greater in-river residence time of DON at low base flow. While the mineralization of DON may be significant during the warm season (e.g., Brookshire et al. 2005), we otherwise contend that the increased [DON] with discharge may reflect import from the catchment via shallow flow.

The mean [DIN] imported by shallow flow inferred herein is relatively low (~26 μ M), in the range of 15 to 70 μM generally observed in surface and shallow groundwater of undeveloped catchments across the US, and substantially lower than the range of 100 to 700 μM observed in shallow streams 440 draining agriculture catchments (Dubrovsky et al., 2010). However, it is greater than the [DIN] of \leq 5 μM recorded in shallow streams draining pristine forested catchments in the Northeast U.S.A., which are otherwise dominated by DON (Dickerman et al. 1989; Hedin et al. 1998). The [DIN] of \sim 6465 μ M recorded here at low base flow, which likely reflects that of deeper groundwater (barring a substantial point source input) is also within the range reported for groundwater NO₃ in undeveloped

- 445 catchments, albeit, at the higher end of this range (of 7 to 75 μ M; Mueller et al., 1995), and falling within the range reported for groundwater NO₃⁻ in southern RI (0 - 91 μ M; Moran et al., 2014). The mean low base flow [DIN] observed here is substantially lower than concentrations typical of groundwater in agricultural catchments, but higher than the [DIN] that was observed in the groundwater reservoir of the Upper Wood River in the 1980's (median \leq 11 μ M μ M; Dickerman et al. 1989; Dickerman and Bell, 1993) - suggesting that anthropogenic input to the deeper groundwater
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4.2 Corroborating insights from NO₃⁻ isotope ratios

N reservoir of the Pawcatuck watershed has increased over time.

We turn to the N and O isotope composition of NO_3^- to further investigate relationships of nutrients with river discharge and to characterize N sources and cycling in the river. Like [NO₃⁻], the

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isotope ratios of NO₃⁻ co-varied with discharge. Values of $\delta^{15}N_{NO3}$ decreased with discharge, suggesting that (a) NO₃⁻ added by shallow flow had lower $\delta^{15}N_{NO3}$ values than low base flow NO₃⁻, and/or (b) $\delta^{15}N_{NO3}$ values at low base flow increased during warmer months compared to their groundwater end-member due to biological cycling in-river. Concurrently, $\delta^{18}O_{NO3}$ values increased with discharge, suggesting that (c) NO₃⁻ added by shallow flow had higher $\delta^{18}O_{NO3}$ values than low

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base flow NO₃⁻, and/or (d) $\delta^{18}O_{NO3}$ values decreased in summer due to biological cycling. We consider these hypotheses in turn.

4.2.1 Sources of DIN in shallow flow evidenced from $\delta^{\rm 15}N_{\rm NO3}$ values



Figure 8. Modified Keeling Plot of NO₃ isotopologue ratios at the Westerly and Stillman bridges vs. the inverse of the daily NO₃⁻ flux: (a) $\delta^{15}N_{NO3}$ vs. the inverse of the NO₃⁻ flux, (b) $\delta^{18}O_{NO3}$ and $\delta^{18}O_{NO3}$ corrected for atmospheric NO₃⁻ vs. the inverse of the NO₃⁻ flux.

In order to evaluate whether the lower $\delta^{15}N$ DIN values observed at higher discharge can be explained by the addition of relatively low $\delta^{15}N$ DIN by shallow flow, we plotted the $\delta^{15}N_{NO3}$ values recorded at the Stillman Bridge vs. the inverse of the corresponding NO₃⁻ flux (i.e., an adapted Keeling Plot; Keeling, 1958, 1961; Figure 8a). Because we lack measurements of the $\delta^{15}N$ values of the incident NH₄⁺ pool (which we could not assess due to an analytical interference from dissolved

organic material; see Zhang et al. 2007), we assume that the N isotope composition of NO₃⁻ captures 470 that of bulk DIN, on the basis that NH₄⁺ imported from the catchment was largely nitrified in-river, wherein NH_4^+ accounted for only a small fraction of the DIN reservoir. The riverine $\delta^{15}N_{NO3}$ data conform to a linear relationship expected for the addition of DIN with a lower mean δ^{15} N to the low base flow reservoir (Table 1). The intercept of the resulting linear regression suggests that the NO₃associated with increased discharge had a mean δ^{15} N value of ~6.7 \pm 0.2 ‰ (Table 1), compared to a

- low base flow value of 2^{-8} -% observed at the bridges. The average $\delta^{15}N_{NO3}$ value of atmospheric 475 NO_3^- in rainwater was -2.5 ± 2.1 ‰ (Section S1; Figure S2), indicating that NO_3^- added by shallow flow did not originate predominantly from direct atmospheric deposition as uncycled atmospheric NO₃. While the $\delta^{15}N_{NO3}$ of atmospheric NO₃⁻ could conceivably be fractionated by biological cycling in-river following its import by shallow flow, increased discharge occurred largely during the cold season, at 480 which time biological cycling in-river was presumably curtailed. Thus, we surmise that the DIN added by shallow flow did not originate from direct atmospheric deposition as uncycled atmospheric NO₃, but rather derived from catchment soils and shallow groundwater. The $\delta^{15}N_{NO3}$ end-member value of ~6.7 % is in the upper range observed for soil NO₃⁻ in temperate forested catchments (Mayer et al. 2002; Barnes and Raymond, 2009). While the net sources of reactive N to forested soils are 485 atmospheric deposition and biological N₂ fixation – which have relatively low δ^{15} N values (≤ 0 %) –
- partial denitrification in soils and shallow groundwater increases the δ^{15} N of the soil N reservoir to values of ~5 ‰ (Amundson et al., 2003; Houlton et al., 2006; McMahon and Böhlke 2006; Houlton and Bai, 2009). The NO₃- imported by shallow flow draining urbanized systems has comparatively higher $\delta^{15}N_{NO3}$ values (≥ 10 %; e.g., Divers et al., 2014), while NO₃⁻ in soils and shallow groundwater
- 490 in agricultural systems generally falls within a lower range of values between 2 - 4 ‰ (Green et al. 2008; Böhlke et al. 2009; Lin et al., 2019). The watershed of the Pawcatuck River is largely forested, yet hosts agricultural and urbanized sections that ostensibly contributed to the mean $\delta^{15}N_{NO3}$ endmember imported by shallow flow. Thus, while DIN added to the river by shallow flow at high discharge had a mean δ^{15} N value consistent with expectations for a largely forested catchment, inputs from agricultural and urbanized catchments may be rendered undiscernible due to their opposing
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4.2.2 Negligible fraction of uncycled atmospheric NO₃⁻ confirmed by O isotope ratios

contributions to the mean $\delta^{\rm 15} N_{\rm NO3}$ value in shallow flow.

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The inference that uncycled atmospheric NO₃⁻ did not contribute substantially to the increased NO_3^- flux at higher discharge is corroborated by the $\Delta^{17}O_{NO3}$ measurements at the Stillman Bridge. 500 The low values of NO3- in rainwater observed evidenced only a slight contribution of < 3% uncycled atmospheric NO₃ to total riverine NO₃ in a few samples, suggesting efficient processing of atmospheric NO₃- in soils shallow groundwater (Mengis et al, 2001; Barnes et al. 2008). This observation is further echoed in a recent meta-analysis of North American rivers, wherein the contribution of uncycled atmospheric NO3⁻ to base flow was inferred to be generally modest 505 (Sebestyen et al. 2019). The NO3⁻ delivered to the Pawcatuck River by shallow flow evidently originated from a reservoir that was biologically cycled within catchment soils - and potentially inriver – thus losing its atmospheric Δ^{17} O signature.

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A Keeling plot of $\delta^{18}O_{NO3}$ values vs. the inverse of the NO₃⁻ flux at the bridges suggests that NO₃⁻ added by surface flow had a mean $\delta^{18}O_{NO3}$ value of $\sim 4.5 \pm 0.4$ % (Figure 8b; Table 1), compared to a mean low base flow value of 2.8 \pm 0.2 %. Although the contribution of uncycled atmospheric NO₃⁻ to the riverine reservoir was modest, we nevertheless consider that the increase in $\delta^{18}O_{NO3}$ values with discharge may derive in part from uncycled atmospheric NO₃⁻, given the direct relationship of $\Delta^{17}O_{NO3}$ to discharge, and considering the characteristically elevated $\delta^{18}O_{NO3}$ values of 60 - 80 ‰ observed in the local rainwater NO₃⁻. Indeed, when the weighted contribution of atmospheric NO₃⁻ is subtracted 515 from individual $\delta^{18}O_{NO3}$ values (attributed from corresponding $\Delta^{17}O$ measurements, accounting for precipitation-dependent differences in the mean Δ^{17} O and $\delta^{18}O_{NO3}$ values of rainwater), the intercept of the Keeling plot decreases slightly to \sim 3.8 ± 0.2 ‰, nevertheless remaining greater than the $\delta^{18}O_{NO3}$ of low base flow NO₃⁻ (Table 1). The higher increase in $\delta^{18}O_{NO3}$ with increasing higher discharge is thus partially explained by the small component of uncycled atmospheric [NO₃⁻] with elevated $\delta^{18}O_{NO3}$

520 values.

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The $\delta^{18}O_{NO3}$ signature of 3.8 ‰ for NO₃⁻ added with increasing discharge (minus the uncycled atmospheric NO₃⁻) is in the range generically observed for soil NO₃⁻ (Kendall et al., 2007; Michener and Lajtha, 2007). It has traditionally been ascribed to that expected for newly nitrified NO₃, based on an empirical metric stipulating that the $\delta^{18}O_{NO3}$ values produced by nitrification derive from the fractional contribution of the reactants, namely 1/3 δ^{18} O of O₂ + 2/3 δ^{18} O of H₂O (Anderson and Hooper, 1983; Hollocher 1984; Kendall et al., 2007). Considering that the $\delta^{18}O_{H20}$ of Pawcatuck river water is -7 ‰ and the $\delta^{18}O_{02}$ of atmospheric oxygen is ~23.5 ‰ (Kroopnick and Craig, 1972), the nitrification $\delta^{18}O_{NO3}$ value thus expected is on the fortuitous order of 3.2 ‰. This empirical metric,

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however, demonstrably overlooks substantive isotope effects associated with O-atom incorporation 530 into the NO₃⁻ molecule during nitrification and isotopic exchange of the nitrite intermediate with water, which otherwise give way to nitrified NO_3^- whose $\delta^{18}O_{NO3}$ value is close to that of ambient water (Sigman et al., 2009; Casciotti et al., 2008; Buchwald and Casciotti, 2010; Snider et al., 2010; Boshers et al., 2019). This consideration explains frequent observations of relatively low $\delta^{18}O_{NO3}$ in some soils and saturated systems, which are not explained by simple fractional contribution of 535 reactants (Hinkle et al. 2008; Xue et al., 2009; Fang et al. 2012; Veale et al., 2019). Thus, we posit that the O isotope composition of the NO_3^- imported into the river with increased discharge, which is typical of that in soils and shallow groundwater, does not strictly indicate that shallow flow NO3originated from proximate nitrification therein, as generally presumed - but-lt also signals that NO3underwent partial denitrification in soils and shallow groundwater, resulting in a coupled increase in 540 its δ^{15} N and δ^{18} O relative to source values (Houlton et al. 2006; Granger and Wankel 2016; Boshers et al. 2019). Although increased discharge occurred largely in winter, some in-river biological cycling during colder months could additionally influence the shallow flow $\delta^{18}O_{NO3}$ end-member, specifically reducing it from its soil value due to the nitrification of incident NH₄⁺. Thus, $\delta^{18}O_{NO3}$ values imported by shallow flow, once adjusted for modest contributions of uncycled atmospheric NO₃⁻, fall within the 545 range typically observed in soils, potentially modified by nitrification in-river.

4.2.3 Values of $\delta^{15}N_{NO3}$ at low base flow reflect groundwater DIN

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The higher $\delta^{15}N_{NO3}$ values at low base flow compared to shallow flow may derive directly from those of the ground-water end-members and point source(s); The $\delta^{15}N_{NO3}$ values in deeper groundwater are generally higher than in shallower groundwater above, <u>more</u> fractionated by denitrification in the saturated zone (e.g., Böhlke et al. 2006). Alternatively, the higher NO₃⁻ isotope ratios at low base flow may result from increased biological cycling in summer – modifying the isotope composition of low base flow NO₃⁻ relative to its groundwater and/or point source values. The expectation of increased biological activity in summer months is consistent with the incident decrease in [NH₄⁺] with lower discharge, which can be explained by a seasonal increase in algal assimilation and nitrification. Fulweiler and Nixon (2005) similarly observed lower [NH₄⁺] in the summer, but saw no correlation to river discharge, further supporting our contention that increased seasonal biological cycling underlies the [NH₄⁺] dynamics, rather than river discharge.

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The extent to which the coincident NO₃⁻ pool is also assimilated during summer months – and isotopically fractionated – is unclear. The fraction of the NO_3^- pool assimilated by algae may be 560 modest, even in summer, on the basis that the phytoplankton biomass was relatively small due to the high tannin content of the river water, which limiteds light penetration. Median chlorophyll-a concentrations in summer were ~1.3 μ g L⁻¹ at the Stillman and Westerly bridges – save for late summer where higher concentrations were detected – while the median [PN] was ~2.5 μ M, and no greater than 7 μ M. There are, however, populations of emergent plants along some shallow reaches 565 of the river, which may assimilate NO₃ as well as reduced N substrates. Nevertheless, the inference that the riverine NO₃⁻ pool is minimally assimilated, even in summer, appears consistent with alongriver distribution of NO_3^- isotope ratios. If a sizeable fraction of the incident NO_3^- pool were assimilated into biomass during summer months, both the $\delta^{15}N_{NO3}$ and $\delta^{18}O_{NO3}$ values of low base

570 2004; Johannsen et al., 2008). However, the $\delta^{15}N_{NO3}$ increase along-river observed during the seasonal surveys, which could be construed as signaling partial assimilation of riverine NO₃-, was not matched by coincident along-river increases in $\delta^{18}O_{NO3}$ values. Similarly, [PN] and chlorophyll-a did not increase along-river, as would otherwise be expected for the progressive and sizeable conversion of the NO₃⁻ pool into the particulate pools (Figure S6c-d). Thus, we rule out a dominant influence of 575 algal assimilation in fractionating the riverine NO₃⁻ isotope ratios.

flow NO_3^- would expectedly increase in proportion to the fraction of NO_3^- assimilated (Granger et al.,

A more nuanced framework from which to interpret the NO₃ isotope ratios is afforded by the concept of riverine nutrient spiraling, namely, the continual assimilation of nutrients in the water column, the remineralization of organic material in sediments, and the return of remineralized nutrients to the water column where they can undergo assimilation into new biomass (reviewed by Ensign and Doyle, 2006; Harvey et al., 2013). A small fraction of the NO₃⁻ pool is likely assimilated during the growing season, resulting in the production of PN with a lower δ^{15} N than coincident NO₃due to N isotope fractionation during assimilation (Needoba et al. 2003; Figure 9). Considering the small summertime pools of PN and NH₄⁺ relative to the NO₃⁻ pool, $\delta^{15}N_{NO3}$ values will be minimally fractionated by assimilation. Moreover, the concomitant recycling of PN and its subsequent nitrification will ostensibly regenerate NO₃⁻ with a $\delta^{15}N_{NO3}$ value roughly equivalent to that assimilated into organic material then ammonified – given an approximate steady state between NO3⁻

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assimilation and nitrification – such that $\delta^{15}N_{NO3}$ values will not incur a progressive increase from

continual assimilation along-river. These dynamics will result in little net change in riverine $\delta^{15}N_{NO3}$ values relative to the mean catchment end-member.

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The NO₃⁻ isotope ratios could, however, be influenced by denitrification in-river (Kellman and Hillaire-Marcel 1998; Figure 9). While direct benthic denitrification does not communicate an isotope enrichment to NO₃ in the overlying water column due to a reservoir effect (Brandes and Devol, 1997; Sebilo et al., 2003; Lehmann et al., 2005), δ^{15} N- and δ^{18} O-enriched NO $_3^-$ from the sediment depth of denitrification can be entrained into the water column by hyporheic flows in the riparian zone (Sebilo 595 et al., 2003). Moreover, coupled nitrification-denitrification can fractionate the N isotopologues of NH_4^+ in surface sediments in proportion to the corresponding fraction of nitrified NO_3^- lost concurrently to denitrification, thus contributing to an increase in δ^{15} N of the water column reactive N reservoir (Brandes and Devol 1997; Granger et al, 2011).

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The along-river increase in $\delta^{15}N_{NO3}$ values could then result from isotopic fractionation by sedimentary denitrification. Yet a downstream increase in $\delta^{15}N_{NO3}$ was notably apparent in all seasons, not only in summer. On the presumption that water-column and benthic N cycling were substantially reduced during the March 2019 sampling when river waters were colder (average temperature of 5.9 °C), we surmise that the increase in $\delta^{15}N_{NO3}$ values along-river arises principally from differences in the δ^{15} N of DIN input from respective reaches of the catchment – although some influence of benthic denitrification on riverine $\delta^{15}N_{NO3}$ values cannot be ruled out. We thus interpret the riverine $\delta^{15}N_{NO3}$ values to predominantly reflect the N isotope composition of DIN input from the catchment. We return to this insight in a subsequent section, to identify N sources along the catchment.

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The $\delta^{18}O_{NO3}$ values along-river can also be interpreted within the framework of nutrient spiraling. As with $\delta^{15}N_{NO3}$, the riverine $\delta^{18}O_{NO3}$ values integrate the contribution of NO_3^- imported from the catchment (including uncycled atmospheric NO $_3^-$), the NO $_3^-$ produced by nitrification in-river, and the NO₃⁻ consumed by assimilation and by denitrification (Figure 9). Without continual exogenous input from the catchment, $\delta^{18}O_{NO3}$ values of an initial NO₃⁻ reservoir would theoretically converge downriver onto a steady-state value dictated by the $\delta^{18}O_{NO3}$ of newly nitrified NO₃⁻ and the effective isotope effect for NO3⁻ consumption, by assimilation and denitrification: For instance, assuming a 615 $\delta^{18}O_{NO3}$ value of -6 ‰ for newly nitrified NO₃⁻ ($\delta^{18}O_{H2O}$ + 1 ‰; Casciotti et al., 2008; Sigman et al., 2009; Buchwald & Casciotti, 2010, Granger et al., 2013; Boshers et al., 2019), a canonical NO₃-

4.2.4 Influence of in-river biological cycling on $\delta^{18}O_{NO3}$ values at low base flow

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assimilation isotope effect of 5 ‰ (Needoba et al., 2003), and no influence of sedimentary denitrification on water column $\delta^{18}O_{NO3}$, values downriver would asymptote to -1 ‰. The $\delta^{18}O_{NO3}$ 620 introduced continuously along the catchment had $\delta^{18}O_{\text{NO3}}$ values greater than -1 ‰ – assuming roughly equivalent in-river assimilation and nitrification fluxes. These greater $\delta^{18}O_{NO3}$ values may also signal some influence of sedimentary denitrification in fractionating the water-column $\delta^{18}O_{NO3}$. Observations of decreasing along-river values are then consistent with the notion of higher 625 catchment $\delta^{18}O_{NO3}$ end-member values converging onto lower values determined by the ratio of nitrification to consumption in-river - and associated isotopic fractionation. Within this framework, $\delta^{18}O_{NO3}$ values in winter, when biological cycling is dampened, would expectedly increase to values closer to the catchment sources, a prediction that appears to be borne out in our observations. Barnes et al. (2008) similarly observed higher $\delta^{18}O_{NO3}$ values during the cold season in steams draining forested watersheds in the Northeastern U.S.A. The riverine $\delta^{18}O_{NO3}$ values thus afford insights into

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N sources and cycling that are consistent with expectations for nutrient spiraling.



Figure 9. Conceptual illustration of the influence of nutrient spiraling on the N and O isotope ratios of riverine NO₃⁻. Nutrient spiraling describes the cycling of nutrients as they are assimilated from the water column into biomass that is temporarily retained on the benthos, then mineralized and released back into the water column or denitrified. (a) The δ^{15} N of the riverine NO₃⁻ reservoir integrates the NO₃⁻ and NH₄⁺ delivered continually from groundwater (δ^{15} N_{NO3-GW} and δ^{15} N_{NH4-GW}), minus the NO₃⁻ removed concurrently by sedimentary denitrification – the δ^{15} N of which depends on the sedimentary isotope fractionation communicated to the water column reservoir, ¹⁵c_D. Given the small size of the respective PN and NH₄⁺ pools relative to NO₃⁻, ammonification and subsequent nitrification produce NO₃⁻ with a δ^{15} N_{NO3-N} value approximating that lost concurrently to assimilation (δ^{15} N_{NO3} - ¹⁵ c_A), notwithstanding the NH₄⁺ input from groundwater. The input of groundwater NH₄⁺ (δ^{15} N_{NH4-GW}) implicitly subsumes the input of reactive allochthonous PN and DON. (b) The riverine δ^{18} O_{NO3-N}, minus NO₃⁻ lost to algal assimilation and sedimentary denitrification – whose respective values depend on the net isotope effects associated with assimilation and denitrification, ¹⁸c_A and ¹⁸c_D.

4.2.5 Regional N sources to the Pawcatuck River

Observations from the along-river surveys provide insights into the contribution of different reaches of the catchment to the riverine N reservoir. Areas of disproportionate loading can be identified from distinct concentration increases, and areas of lesser loading and/or net attenuation from concentration decreases. Reaches of the river that exhibit disproportionate loading present potential targets for mitigation. As detailed above, we interpret changes in $\delta^{15}N_{NO3}$ values along-river to primarily reflect differences in the $\delta^{15}N$ of DIN inputs from respective reaches of the catchment,

640 thus serving to identify dominant regional N sources.

Surface water in Woarden Pond had relatively low nutrient concentrations, which remained similarly low at Biscuit City Road (Station 2) in two of three samplings. The otherwise extremely elevated [NO₃⁻] at Station 2 in March 2018 decreased downstream at Station 3 by > 40 %, more than can be explained by either dilution from additional inflow or by denitrification. This elevated concentration may then reflect the inadvertent sampling of a groundwater plume or a localized reach of slow-flowing water, rather than the mean river composition. Monitoring at Biscuit City Road from 2007-2016 by the Wood-Pawcatuck Watershed Association similarly reveals relatively low median [NO₃⁻] values of ~1 μM during fall samplings, punctuated by stochastic instances of elevated concentrations, as high as 41 μM (Figure S6; Wood-Pawcatuck Watershed Association, 2020). The δ¹⁵N_{NO3} value recorded at this station during the March 2019 sampling was 6.2 ‰ and the δ¹⁸O_{NO3} was 4.8 ‰, values consistent with either a groundwater plume or a slow-flowing reach of the river.

Both $[NO_3^-]$ and $[PO_4^-]$ increased thereon at Kenyon and Wood River Junction (Stations 3 and 4, respectively) in all sampling campaigns. Associated $\delta^{15}N_{\text{NO3}}$ values were relatively low during the March 2019 sampling ($\leq 4 \%$) – coincident with more elevated [NO₃⁻] – potentially signaling the input 655 of DIN by shallow flow from proximate turf farms (Kreitler et al., 1978; Katz et al., 1999; Townsend et al., 2002; Deutsch et al, 2005). Input of uncycled atmospheric NO₃ by surface flow due to regional snow melt, which could also explain lower $\delta^{15}N_{\text{NO3}}$ values, is not supported by the corresponding $\delta^{18}O_{NO3}$ values, which would otherwise be disproportionately high. Moreover, there was little to no snow accumulated snow in March 2019. The increased nutrient concentrations observed at Stations 3 and 4 in all sampling campaigns also likely derived in part from the retention ponds at Kenyon 660 Industries, in light of a permitted discharge of 7,500 moles N and 950 moles P per day (U.S. Environmental Protection Agency, 2010). Corresponding $\delta^{15}N_{NO3}$ values at Stations 3 and 4 during the May and November 2018 samplings were ~6 ‰, which could indicate input from deeper agricultural groundwater, or could reflect discharge by Kenyon Industries, for which we do not have end-member 665 values.

Inflow from the less impacted Wood River evidently diluted nutrient concentrations in the Pawcatuck River (Station 7). The Wood River contributes significantly to the total discharge of the Pawcatuck <u>R</u>river ($\geq 14 \pm 5$ % of total – based on discharge at Hope Valley USGS gauge), draining a more forested watershed that harbors fewer agricultural areas than the lower Pawcatuck River. The [NO₃-] in all sampling campaigns remained relatively invariant downstream of the Wood River inflow

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through the largely forested catchment to Potter Hill Dam (Station 11), while $\delta^{15}N_{NO3}$ values increased

marginally. The increases in $[NO_3^-]$ and $\delta^{15}N_{NO3}$ thereon to the Stillman and Westerly bridges indicate DIN input from groundwater in the more populated portion of the watershed. The population density and associated septic systems increase considerably in the vicinity of the town of Westerly (Wood-

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Pawcatuck Watershed Association, 2016). Septic leachate and urban runoff are typically associated with relatively higher δ^{15} N values, on the order of 8 to 15 ‰ (Kendall et al. 2007; Böhlke et al., 2009; Kasper et al, 2015). Thus, changes in land use along the catchment best explain the $\delta^{15}N_{NO3}$ increase in the lower portion of the river.

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In all, the substantial difference in [DIN] between Stations 2 and 5 signals disproportionate input from this section of the watershed, likely owing to the proximity of turf farms and discharge from Kenyon Industries. Indeed, the riverine DIN flux at Wood River Junction amounted to 28 ± 11 % of the DIN flux recorded at the Stillman bridge among the 3 sampling dates, while accounting for only $11 \pm$ 2 % of the riverine discharge. A fraction of the N loaded in this portion of the river may arguably be partially attenuated by denitrification along-river; nevertheless, this regional input remains 685 substantial even assuming some biological attenuation. This portion of the river also contributed disproportionately to the riverine PO43- burden, although we do not explicitly consider this contribution in relation to the total discharge into the estuary, given the complex geochemistry of PO_4^{3-} that involves adsorption and release from authigenic particles in sediments (Froelich, 1988).

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The increase in [DIN] and $\delta^{15}N_{NO3}$ values in the lower portion of the river, in light of the large coincident river discharge, also signals a disproportionate contribution from the urbanized portion of the catchment. However, lacking estimates of river discharge at Potter Hill Dam (Station 11), we cannot deduce the fractional contribution from this portion of the watershed confidently. Nevertheless, assuming a $\delta^{15}N$ input from the urbanized catchment of 10 ‰ and a mean $\delta^{15}N_{NO3}$ of 6_‰ at Potter Hill Dam, compared to 8 ‰ at the Westerly Bridge, the DIN added to the river within 695 this reach would amount to ~50 % of the total riverine N load. Otherwise, assuming a δ^{15} N input of 15 ‰, the DIN contributed from the urbanized reach would otherwise amount to \sim 20 % of total.

4.3 N loading into Little Narragansett Bay

4.3.1 Riverine contributions

- Too Estimates of the annual N loading from the Pawcatuck River into Little Narragansett Bay for 2018, compiled from our weekly measurements at the Stillman bridge, were 20.2 x 10⁶ moles yr⁻¹ for DIN and 40.3 x 10⁶ moles yr⁻¹ for TN, albeit, with uncertainty associated with the TN loading estimate given the variability of our DON measurements (Table 2). These values are considerably larger than those estimated from biweekly measurements at the Stillman Bridge for 2002 by Fulweiler and Nixon
- 705 (2005), which were 7.2 x 10⁶ moles yr⁻¹ for DIN and 16.0 x 10⁶ moles yr⁻¹ for TN. The greater N loading in 2018 could arise from (a) an increase in groundwater concentrations and/or point source discharge, evident at low base flow, and/or (b) increased N loading by shallow flow. The latter could result from increased atmospheric N deposition, greater annual precipitation, and/or higher surface N concentrations imported by shallow flow. We examine these hypotheses in turn.
- In 2002, the [DIN] at low base flow, which reflects that associated with deeper groundwater and point source discharge, was ~50 μM (Fulweiler and Nixon 2005) thus lower than the value of <u>~64~60</u> μM observed by us suggesting lower DIN inputs from deeper groundwater and/or point sources in 2002. This inference is supported by monitoring data of the Wood-Pawcatuck Watershed Association from 1989 to 2017, which documented a discernible increase in riverine [NO₃⁻] of approximately 1
- 715 μM per year at Bradford (Station 8) in the month of October, and a slighter rate of increase 0.3 μM per year in May (Figure S7a). These trends are not explained by a secular change in monthly precipitation (Figure S7b). Given that mean river discharge is generally higher in May than in October, the greater rate of increase in October suggests an increase in [DIN] of deeper groundwater entering the river and/or an increase in point source discharge up-river. Assuming a 10 μM difference in
- [DIN] during low base flow at the Stillman Bridge in 2018 compared to 2002, and a year-round discharge of deeper groundwater of 0.25 x 10⁶ m³ d⁻¹ (based on the mean low base flow discharge), 0.9 x 10⁶ moles of additional DIN were potentially delivered in 2018 from the increased groundwater or point source [DIN]. This greater DIN input from deeper groundwater and/or point sources, however, only explains a small fraction of the additional loading of 12.9 x 10⁶ moles DIN estimated for 2018 compared to 2002.
 - Regional atmospheric deposition of DIN has decreased ~67 % since 2000, from ~95 μM to 36 μM◀ DIN in 2018 (NOAA National Atmospheric Deposition Program, 2019), which should have resulted in

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a lower riverine N flux given similar annual precipitation. However, 2002 was a drought year, whereas 2018 was the third wettest year on record in Washington County, RI, with total precipitation at 152

- 730 cm compared to an 80-year mean of 114 cm (NOAA National Centers for Environmental information, 2019). River discharge was thus a substantially lower in 2002, at 303 x 10⁶ m³ yr⁻¹, compared to 702 x 10⁶ m³ yr⁻¹ in 2018 (Table 2). The larger riverine N loading in 2018 is thus explained by greater precipitation and consequent discharge above low base flow, importing additional DIN (and DON) into the river via shallow flow. Assuming a comparable [DIN] delivered by shallow flow between then
- and now (26 ± 3 μmol L⁻¹), the greater discharge in 2018 entails an additional DIN influx of 10.1 (± 1.2) (x 10⁶) moles yr⁻¹, accounting for most of the estimated difference of 12.9 x 10⁶ moles DIN yr⁻¹ between 2018 and 2002. The greater discharge in 2018 ostensibly <u>resulted in in</u> increased the-DON influx into the river-concurrently, although the variability of our DON measurements precludes a robust estimate of this additional flux.
- In all, the greater DIN loading in 2018 compared to 2002 is explained in small part by an apparent* increase of [DIN] at low base flow deriving from a parallel increase in groundwater [DIN] or a potential increase in point source discharge by Kenyon Industries and in greater part by a substantial difference in annual river discharge and associated import of DIN (and DON). <u>Parenthetically, Eq. 4</u>
 returns a load of 10.1 x 10⁶ mol DIN yr¹ for 2002 when adjusting low base flow to 50 µM, compared
 to the observed flux of 7.2 x 10⁶ mol DIN yr¹ (Fulweiler and Nixon, 2005), suggesting that the import

of DIN from shallow groundwater may also have increased in the last two decades.

Using our mixing curve algorithm (Equation 4), we derive an estimate of the DIN loading by the Pawcatuck River in 2017 – a year posting a median river discharge – of 15.2 × 10⁶ moles N yr⁻¹ (Table 2). This estimate is 25% lower than that for 2018, yet more than double estimate for 2002, highlighting the impact of inter annual variability in river discharge on the annual N-load.

Extrapolating DIN discharge for other years with various river flows allows for a comparison to independent estimates of nitrogen loads from the watershed. Vaudrey et al. (2017) utilized a landuse model to estimate the TN load from the Pawcatuck River watershed at <u>14.6 x 10⁶ 16.8 x 10⁶</u>-moles TN yr⁻¹, <u>based on</u>, <u>This estimate was based on 2011 land cover data</u>, 2010 census data, precipitation from 2013 to 2015, <u>and included WWTF-reported data from 2011 to 2014</u>. The land-use model thus relies on "average" flow states and does not capture the variability in TN load associated with interannual variability in river discharge. Removing the WWTF-inputs occurring downstream of the Stillman Bridge allows for a comparison to this study, resulting in an estimate of TN loading of 14.6 x Formatted: Tab stops: Not at 0.25"

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10⁶ moles TN yr¹. In comparison, uUsing the mixing curve algorithm (Equation 4) and river flow during 760 for the 2013-2015 period, and further assuming that DIN accounted for roughly half of TN as in our measurements, DIN loading is otherwise estimated as 13.73 x 106 moles DIN yr-1 and TN loading at 2726.37 x 10⁶ moles TN yr⁻¹ for this period (Table 2). Theis TN load thusing estimated is thus 122.74 x 10⁶ moles TN yr⁻¹ higher than the land-use model estimate, leaving nearly 45 % of TN apparently unaccounted for. However, our TN measurements include both labile and non-labile N, while the 765 land-use model represents reactive TN and does not account for non-labile species. On the basis that refractory humified allochthonous organic material dominates the DON pool in the Pawcatuck River - an N pool that is largely unavailable to micro-organisms (Stevenson 1994) - a value of 45 % of TN being non-labile could be consistent with this system, if only ~10 % of riverine DON were labile on pertinent time scales._-Seitzinger et al. (1997) otherwise estimated that 40 – 70 % of DON from the 770 Delaware River was reactive on pertinent time scales. The data at hand do not permit us to resolve this guandary, although characterizing the reactivity of DON from the Pawcatuck River is evidently crucial to mitigating eutrophication in the bay.

Rhode Island met an ambitious goal of a 50 % reduction in N loading to Narragansett Bay in 2012 relative to 1995-1996 loads, but the Pawcatuck River was not included in these reduction priorities. 775 This oversight is evident in the loads we currently see to the Pawcatuck River relative to loads in rivers draining to Narragansett Bay, located just east of the Pawcatuck River watershed. In the early 1980s through the early 2000s, the TN load normalized to watershed area for these rivers Woonasquatucket, Moshassuck, Blackstone, Taunton, and Pawtuxet rivers ranged from 9.3 to 14.9 kg ha⁻¹ yr⁻¹ , with an average of 11.8 kg ha⁻¹ y⁻¹ (as reviewed in Narragansett Bay Estuary Program, 2017; Nixon et al., 1995; 780 Nixon et al., 2008). Compared to this time period, the Pawcatuck River's current load of 7.4 kg N ha⁻¹ yr⁻¹ is relatively low. However, by the late 2000s, these riverine loads were substantially reduced to an average of 6.6 kg ha⁻¹ yr⁻¹, and have continued to decline, achieving an average of 4.8 kg ha⁻¹ yr⁻¹ in recent N budgets developed for the 2013-2015 time period (as reviewed in Narragansett Bay Estuary Program, 2017; Krumholz, 2012), with one - One exception to the success achieved in riverine 785 loads to Narragansett Bay isof-(i.e., the Ten Mile River, which drains an urbanized watershed in East Providence, RI, with a 2013-2015 load estimate of 10.2 kg ha⁻¹-yr⁻¹. However, the Ten Mile River watershed is relatively small (~20 % of the Pawcatuck River watershed), accounting for only 7% of the riverine load to Narragansett Bay <u>; (</u>Narragansett Bay Estuary Program, 2017<u>; Krumholz, 2012</u>). Export from pristine temperate zones prior to human disturbances is estimated to have been on the

790 order 1.3 kg N ha⁻¹ yr⁻¹ (Howarth et al. 1995; 1996). Most Narragansett Bay rivers are moving toward this pristine condition, whereas the Pawcatuck River has shown an increase in N load over time.

4.3.2 Point source loading from the WWTFs and Kenyon Industries

The Westerly and Pawcatuck WWTFs downstream of the Stillman and Westerly bridges 795 accounted for a relatively modest fraction of the total annual nitrogen loading into the estuary, approximately 7 % (Table 2). This estimate does not consider loading from the catchment downstream of the Stillman and Westerly bridges, which would modestly lower the relative contributions of the WWTFs. Fulweiler and Nixon (2005) otherwise estimated that the WWTFs accounted for 18 % of annual N loading into the estuary, albeit, relying on a WWTF loading estimate 800 of 6 x 10⁶ moles TN yr⁻¹, a flux notably higher than that of 1.45 x 10⁶ moles TN yr⁻¹ reported by Rhode Island Department of Environmental Management and the Connecticut Department of Energy and Environmental Protection for 2002 (Vaudrey et al., 2017). Replacing the higher WWTF load in the Fulweiler and Nixon's (2005) estimate with the lower load reported by the States yields a match to the current study, indicating that the WWTFs accounted for about 5 % of the total annual load. Independent estimates by Vaudrey et al. (2017) derived from a land-use model suggest that WWTF 805 effluents contribute \sim 13 % of the riverine-plus-WWTFs TN discharged to the estuary on an annual basis, but this load included only reactive nitrogen and did not estimate the non-labile fraction measured in this and Fulweiler and Nixon's (2005) studies; including an estimate of the non-labile

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The annual N loading into the Pawcatuck River from Kenyon Industries, as monitored by the Rhode Island Department of Environmental Monitoring from 2011-2013, was 2.7 x 10⁶ moles TN per year, thus accounting for 6 % of the annual riverine-plus-WWTFs loading to the estuary, an input comparable to that of the WWTFs (Table 2). Loading by Kenyon Industries is notable in that it is approximately equivalent to the amount of fertilizer applied to agricultural, hay, and pasture lands throughout the whole watershed (Vaudrey et al., 2017).

fraction brings the annual contribution from WWTFs down to 8 % of the TN.

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The overgrowth of nuisance macroalgae in Little Narragansett Bay is presumably fueled predominantly by nutrients delivered during warmer months, at which time riverine N loading is at a relative minimum (Table 2). While the fraction of TN loading to the estuary by the WWTFs was negligible during colder months (< 5 %), this proportion increased to 21 % during the warmer months in 2018, from May 1st to into October 31st. The estimated contribution from Kenyon Industries similarly increased to 16 % of total N loading during the warmer months. The influence of these point sources on algal growth during the warm season is likely to be even greater, considering that an important fraction of the total N flux from the Pawcatuck River derived from DON (38 % from May to

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November), of which only a fraction may be bioavailable on pertinent time scales. Assuming a median reactivity of riverine DON of 50 % (Seitzinger et al., 1997), the WWTFs and Kenyon industries could account for as much as 25 and 19 % of labile N loading to the estuary during the warm season, respectively, given a riverine DIN loading of 2.6 x 10⁶ mol N_{DIN} from May through October. Thus, we estimate that the WWTFs contributed between 21 - 25 % of N loading to the estuary during the warm season, and Kenyon Industries contributed 16 - 19 %.

830 4.4 *implications for the mitigation of eutrophication*

Our analysis suggests that the Pawcatuck River is strongly impacted by anthropogenic N input. Compounding the problem, the drainage basin of the river is large relative to the receiving estuary, explaining the severe eutrophication therein. The DIN concentrations and NO₃⁻ isotope ratios indicate substantive inputs of reactive N to the river from agricultural and/or point sources along the upper river catchment, and from urbanized sources along the lower reach of the river. The reactive N loaded annually into Little Narragansett Bay from the Pawcatuck River is highly influenced by the amplitude of river discharge, increasing with discharge due to the additional import of reactive N by shallow flow. Loading during the warmer months in 2018 was thus substantially lower than in colder months due to lower summertime precipitation, rendering point source discharges from Kenyon Industries and WWTFs more important to the total N loading to the estuary during the major growing season.

Reductions in summertime discharge by Kenyon Industries and the Westerly WWTF offer the most expeditious targets to decrease N loading into the estuary, albeit, at considerable cost. The disproportionate loading from the catchment of the upper river also begs more tempered applications of agricultural fertilizers at adjacent turf farms and expansion of riparian buffers, in order

to effect reductions in shallow and deeper groundwater N concentrations. In the more populated portion of the watershed, N reductions could be achieved by augmenting linkage of households to the sewer line, transitioning traditional septic systems to advanced, N-removing septic systems, and encouraging the dismantling of outdated, legacy cesspools (Amador et al. 2017; Narragansett Bay Estuary Program, 2017). Within the watershed draining directly to the estuarine portion of the
Pawcatuck River south of the Westerly Bridge, 90 % of the households are connected to sewer (Vaudrey et al., 2017). In the remainder of the watershed, where groundwater drains to a freshwater body (wetland, pond, river) prior to entering the estuary, only 21 % of people are on sewer. This

distribution reflects the urban nature of the watershed near the coast and the more rural character

of the watershed further inland. Finally, restricting the use of lawn fertilizers and lessening the extent 855 of impervious surfaces in and around Westerly would further aid in reducing loading from storm water.

While reductions in N loading are necessary to mitigate eutrophication in Little Narragansett Bay, target N loads have yet to be adopted by Rhode Island or Connecticut. A TN load of 50 kg ha⁻¹estuary yr ¹ (3.6 x 10^3 mol ha⁻¹_{estuary} yr⁻¹), which is generally supportive of eelgrass, has been proposed by the 860 scientific community (Hauxwell et al., 2003; Latimer and Rego, 2010). In 2018, the DIN and TN loads to Little Narragansett Bay were 37 x 10³ and 74 x 10³ moles N ha⁻¹ yr⁻¹, respectively (given a 583 ha area of estuary downstream of the Westerly Bridge), suggesting that an astounding 10 to 20-fold reduction in N loading may be required to recover eelgrass beds. We consider that a fraction of this N load may escape the estuary directly and not be retained therein, reducing the effective annual 865 estuarine N load. Moreover, seasonality of nitrogen delivery coupled with the warm summer growing season may point the way towards targeted summer reductions that could have greater impact on the eutrophic status of the system. Regardless, immediate mitigation efforts are necessary at this junction, not purely to realize reductions in N loading, but, more soberly, to prevent further increases in N loading to the Pawcatuck River and continuing degradation of the river and estuary.

870 5. Conclusions

> Our findings illustrate the utility of NO3⁻ isotopologue ratios in differentiating among N sources, with implications for the management of N loading from of the watershed. In particular, the seasonal and flow-dependent nature of N loading and cycling uncovered herein presents important considerations for mitigation efforts.

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Our interpretations of NO₃⁻ isotopologues dynamics also move beyond the traditional source attribution framework in an effort to reconcile with current theory of riverine N biogeochemistry. Nutrient spiraling theory offers a powerful conceptual basis to differentiate the influences of N sources vs. cycling on NO3- isotopologue distributions. Continued inquiry in the context of this framework is bound to yield novel and unexpected insights on N isotopologue cycling and, more 880 fundamentally, on river biogeochemistry.

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	River Discharge 10 ⁶ m ³ yr ⁻¹	DIN Flux 10 ⁶ mol N yr ⁻¹	TON Flux 10 ⁶ mol N yr ⁻¹	TN Flux 10 ⁶ mol N yr ⁻¹	TN Flux 10 ⁶ mol N yr⁻¹ (May-Nov)	TN loading kg N ha ⁻¹ yr ⁻¹	% of TN flux annual	% of TN flux (May-Nov)
Stillman bridge (2018)	702	20.1 ± 2.0	20.2 ± 2.8	40.3 ± 6.8	4.8 ± 0.6	7.4 [§]	-	-
Kenyon Industries	-	-	-	2.7*	1.0€	-	6	16
Westerly WWTF ¹	-	1.4 ± 0.1	1.1 ± 0.9	2.5 ± 0.1	1.2 ± 0.3	_	6	20
Pawcatuck WWTF [†]	-	-	-	0.3	<0.1	-	<1	2
Stillman bridge (2002) [‡]	303	7.2	6.2 ^f	16.0	-	3.1 [§]	-	-
Land use model ^c	-	-	-	17.8	-	3.6¥	-	-
Mixing model (2013-15) [£]	408	13. <mark>71</mark>	-	27 6 .3 2 ^π	-	_	-	-

Table 2. Estimates of annual N loading into Little Narragansett Bay from the Pawcatuck River at Stillman Bridge (inclusive of Kenyon Industries) and from the Westerly and Pawcatuck Waste Water Treatment Facilities downstream.

^{*}Based on a watershed area of 760 km²; ^{*}DEM-monitored loading in 2012. [¢]Permitted seasonal loading; [†]Measured; [†]Reported; [†]Fulweiler & Nixon (2005);^fas DON only; [¢]Vaudrey (2016); [¢]Eq. 4; ^{*}Based on a watershed area of 660 km² established from ArcHydro. ^{*}Assuming that DON loading is equivalent to coincident DIN loading.

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Data availability. All data were submitted to Rhode Island Department of Environmental
 Management and the Connecticut Department of Energy and Environmental Protection, and
 will also be archived online in the PANGAEA data repository.

Author contributions. VRR and JG conceive the research question, designed the study approach, led the field survey, ensured data curation and conducted formal analysis. SC, MLB, CPK, LAT, and HCW assisted with data collection and analysis. CMM assisted with statistical analyses. CRT and HH provided use of specialized facilities. JG and JMPV secured funding for the investigation. VRR

895 HH provided use of specialized facilities. JG and JMPV secured funding for the investigation. VRR and JG wrote the first draft of the paper, and all co-authors contributed to writing review and editing.

Competing interests. The authors declare that they have no conflicts of interest.

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905 Harbors (CUSH) inspired our effort to determine sources of nutrients to the estuary. <u>Comments</u> by two anonymous reviewers helped improve the manuscript.

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