Biogeochemical controls on ammonium accumulation in the surface layer of the Southern

Ocean

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Shantelle Smith^{1*}, Katye E. Altieri¹, Mhlangabezi Mdutyana^{1,2}, David R. Walker³, Ruan G. Parrott¹, Sedick Gallie³, Kurt A.M. Spence¹, Jessica M. Burger¹, Sarah E. Fawcett^{1,4}

Department of Oceanography, University of Cape Town, Private Bag X3, Rondebosch,
 Cape Town, South Africa

² Southern Ocean Carbon and Climate Observatory (SOCCO), CSIR, Rosebank, Cape Town, South Africa

³ Department of Conservation and Marine Sciences, Cape Peninsula University of Technology, Cape Town, South Africa

⁴ Marine and Antarctic Research centre for Innovation and Sustainability (MARIS), University of Cape Town, Cape Town, South Africa

The production and removal of ammonium (NH4+) are essential upper-ocean nitrogen cycle

* Corresponding author: smtsha023@myuct.ac.za

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1. Abstract

19 pathways, yet in the Southern Ocean where NH₄⁺ has been observed to accumulate in surface 20 waters, its mixed-layer cycling remains poorly understood. For surface seawater samples 21 collected between Cape Town and the marginal ice zone in winter 2017, we found that NH4+ 22 concentrations were five-fold higher than is typical for summer, and lower north than south of 23 the Subantarctic Front (0.01-0.26 µM versus 0.19-0.70 µM). Our observations confirm that 24 NH₄⁺ accumulates in the Southern Ocean's winter mixed layer, particularly in polar waters. NH₄⁺ 25 assimilation rates were highest near the Polar Front (12.9 ± 0.4 nM day⁻¹) and in the Subantarctic 26 Zone $(10.0 \pm 1.5 \text{ nM day}^{-1})$, decreasing towards the marginal jce zone $(3.0 \pm 0.8 \text{ nM day}^{-1})$ despite 27 the high ambient NH₄⁺ concentrations in these southernmost waters, likely due to the low 28 temperatures and limited light availability. By contrast, rates of NH₄⁺ oxidation were higher south than north of the Polar Front (16.0 \pm 0.8 versus 11.1 \pm 0.5 nM day⁻¹), perhaps due to the lower 29 light and higher iron conditions characteristic of polar waters. NH₄+ concentrations were also 30 31 measured on five transects of the Southern Ocean (Subtropical- to marginal ice zone) spanning 32 the 2018/2019 annual cycle. These measurements reveal that mixed-layer NH₄+ accumulation 33 south of the Subantarctic Front derives from sustained heterotrophic NH₄+ production in late 34 summer through winter that in net, outpaces NH₄+ removal by temperature-, light, and iron-35 limited microorganisms. Our observations thus imply that the Southern Ocean becomes a 36 biological source of CO₂ to the atmosphere for half the year not only because nitrate drawdown 37 is weak, but also because the ambient conditions favour net heterotrophy and NH4+ accumulation.

2. Introduction

The Southern Ocean impacts the Earth system through its role in global thermohaline circulation,

40 which drives the exchange of heat and nutrients among ocean basins (Frölicher et al., 2015;

\$41 Sarmiento et al., 2004). The Southern Ocean also plays an integral role in mediating climate, by

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transferring carbon to the deep ocean via its biological and solubility pumps (Sarmiento & Orr, 78 79 1991; Volk & Hoffert, 1985) and through the release of deep-ocean CO₂ to the atmosphere during 80 deep-water ventilation (i.e., CO₂ leak; Broecker & Peng, 1992; Lauderdale et al., 2013; Sarmiento 81 & Toggweiler, 1984). Upper Southern Ocean circulation is dominated by the eastward-flowing 82 Antarctic Circumpolar Current (ACC) that consists of a series of broad circumpolar bands 83 ("zones") separated by oceanic fronts. These fronts can drive water mass formation (Ito et al., 84 2010) and nutrient upwelling that supports elevated productivity (Sokolov & Rintoul, 2007).

85 Concentrations of the essential macronutrients, nitrate (NO₃-) and phosphate (PO₄³-), are 86 perennially high in Southern Ocean surface waters, in contrast to most of the global ocean. 87 Assimilation of these nutrients, and thus primary productivity in the Southern Ocean, is limited 88 by numerous overlapping factors, including temperature, light, micronutrient concentrations, and 89 grazing pressure (e.g., Boyd et al., 2001; Martin et al., 1990; Reay et al., 2001; Smith Jr & 90 Lancelot, 2004). The strength of these limitations varies with sector (i.e., longitude), zone (i.e., 91 latitude), and season, resulting in spatial and temporal variability in chlorophyll-a, primary 92 production, plankton community composition, and nutrient uptake regime (Mdutyana et al., 93 2020; Mengesha et al., 1998; Shadwick et al., 2015; Thomalla et al., 2011. In addition to the 94 seasonality of temperature and light, Southern Ocean ecosystems are influenced by seasonal 95 changes in nutrient availability. In winter, deep mixing replenishes the nutrients required for phytoplankton growth but the low temperatures and light levels impede biological activity 96 97 (Rintoul & Trull, 2001). Once the mixed layer shoals in spring and summer, phytoplankton 98 consume the available nutrients until some form of limitation (usually iron; Nelson et al., 2001; 99 Nicholson et al., 2019) sets in. This balance between wintertime nutrient recharge and 100 summertime nutrient drawdown is central to the Southern Ocean's role in setting atmospheric 101 CO₂ (Sarmiento & Toggweiler, 1984).

103 drives phytoplankton to increased reliance on recycled ammonium (NH₄⁺; Timmermans et al., 104 1998), the assimilation of which has a far lower iron requirement than that of NO₃ (Price et al., 105 1994). The extent to which phytoplankton rely on NO₃ versus NH₄ as their primary N source 106 has implications for Southern Ocean CO2 removal since phytoplankton growth fuelled by 107 subsurface NO₃ ("new production") must be balanced on an annual basis by the export of sinking 108 organic matter ("export production"; Dugdale & Goering, 1967), which drives CO2 sequestration 109 (i.e., the biological pump; Volk & Hoffert, 1985). By contrast, phytoplankton growth on NH₄+ or 110 other recycled N forms ("regenerated production") yields no net removal of CO2 to the deep 111 ocean (Dugdale & Goering, 1967). Considerable research has focused on NO₃ cycling in the 112 Southern Ocean mixed layer because of the importance of this nutrient for the biological pump 113 (e.g., Francois et al., 1992; Johnson et al., 2017; Mdutyana et al., 2020; Primeau et al., 2013;

The onset of iron limitation following the spring/early summer bloom in the Southern Ocean

114 Sarmiento & Toggweiler, 1984) and global ocean fertility (Fripiat et al., 2021; Sarmiento et al., 115 2004). By contrast, the cycling of regenerated N within the seasonally-varying mixed layer -

116 including the production of NH₄⁺ and its removal by phytoplankton and nitrifiers - remains

117 poorly understood.

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118 NH₄⁺ is produced in the euphotic zone as a by-product of heterotrophic metabolism (Herbert,

119 1999) and as a consequence of zooplankton_grazing (Lehette et al., 2012; Steinberg & Saba,

120 2008), and is removed by phytoplankton uptake (in euphotic waters) and nitrification (mainly in Deleted: Southern Ocean

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(Kopczyńska et al., 2007). Deleted: seasonal cycles

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166 aphotic waters). Heterotrophic bacteria can also consume NH₄+ (Kirchman, 1994) and have been 167 hypothesized to do so at significant rates in the Southern Ocean mixed layer in winter (Cochlan, 168 2008; Mdutyana et al., 2020). The assimilation of NH₄+ by phytoplankton requires relatively little 169 energy (Dortch, 1990) such that NH₄⁺ is usually consumed in the cuphotic zone as rapidly as it 170 is produced (Glibert, 1982; La Roche, 1983), resulting in very low surface NH₄⁺ concentrations 171 in the open ocean (<0.2 μM; Paulot et al., 2015). Additionally, NH₄⁺ is often the preferred N 172 source to small phytoplankton (Dortch 1990), which typically dominate when iron and/or light 173 are limiting (Deppeler & Davidson, 2017; Pearce et al., 2010; Tagliabue et al., 2014) since their 174 higher cell surface area-to-volume ratio renders them less vulnerable to diffusion- and/or light limitation (Finkel et al., 2004; Fujiki & Taguchi, 2002; Hudson & Morel, 1993; Mei et al., 2009). 175

176 In addition to the implications for size distribution, the dominant N source to phytoplankton is 177 indicative of their potential for CO₂ removal, as per the new production paradigm (Dugdale & Goering, 1967). The N isotopic composition (δ^{15} N, in ‰ vs. N₂ in air, = (15 N/ 14 N_{sample}/ 15 N/ 14 N_{air} 178 179 -1) x 1000) of particulate organic N (PON; a proxy for phytoplankton biomass) can be used to 180 infer the dominant N source to phytoplankton (Altabet, 1988; Fawcett et al., 2011; Lourey et al., 181 2003; Van Oostende et al., 2017) since the assimilation of subsurface NO₃ yields PON that is 182 higher in δ^{15} N than that fuelled by recycled NH₄+ uptake (Treibergs et al., 2014). As such, measurements of bulk δ^{15} N-PON can be used to infer the net N uptake regime (Altabet, 1988; 183 184 Fawcett et al., 2011; 2014; Lourey et al., 2003).

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Nitrification, the oxidation of NH₄⁺ to nitrite (NO₂⁻) and then NO₃⁻ by chemoautotrophic bacteria and archaea, was historically considered unimportant in euphotic zone waters due to the evidence for light inhibition of nitrifiers (Hooper & Terry, 1974; Horrigan & Springer, 1990; Olson, 1981) and the fact that they are outcompeted by phytoplankton for NH₄⁺ (Smith et al., 2014; Ward, 1985; 2005; Zakem et al., 2018). However, this view has been challenged in numerous ocean regions (Yool et al., 2007), including the Southern Ocean (Smart et al., 2015; Cavagna et al., 2015; Fripiat et al., 2015; Mdutyana et al., 2020). Wintertime upper-ocean NH₄⁺ dynamics thus have implications for annual estimates of carbon export potential, insofar as NO₃⁻ produced by nitrification in the winter mixed layer that is subsequently supplied to spring/summer phytoplankton communities constitutes a regenerated rather than a new N source on an annual basis (Mdutyana et al., 2020).

Surface concentrations of NH₄+ are typically near-zero in spring and early- to mid-summer in the open Southern Ocean (Daly et al., 2001; Henley et al., 2020; Sambrotto & Mace, 2000; Savoye et al., 2004) due to assimilation by phytoplankton. In late summer, a peak in NH₄+ concentration has been observed and attributed to enhanced bacterial and zooplankton activity following elevated phytoplankton growth (Becquevort et al., 2000; Dennett et al., 2001; Mengesha et al., 1998). The limited available observations suggest that wintertime surface NH₄+ concentrations are high (often >1 μM), particularly south of the Subantarctic Front (SAF) (Bianchi et al., 1997; Henley et al., 2020; Philibert et al., 2015; Mdutyana et al., 2020; Weir et al., 2020). It thus appears that NH₄+ is not depleted following the late summer peak in its concentration, which indicates enhanced NH₄+ regeneration, either coincident with (but in excess of) NH₄+ assimilation in winter and/or prior to this in late summer and/or autumn. Under these conditions, the Southern Ocean mixed layer may become net heterotrophic and thus a biological source of CO₂ to the atmosphere.

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285 Here, we focus mainly on NH4+ cycling in the Southern Ocean mixed layer in winter, a season 286 assumed to be largely biologically dormant (Arrigo et al., 2008; Schaafsma et al., 2018) and for 287 which NH₄⁺ cycle data are scarce. We confirm that NH₄⁺ accumulates throughout the winter 288 mixed layer south of the SAF, and examine the potential drivers thereof. Using NH4+ 289 concentration data collected over a full annual cycle, we propose that these drivers include a 290 contribution from the residual late-summer NH₄+ pool, sustained NH₄+ production in the autumn, 291 and winter, and limited wintertime NH₄⁺ uptake and oxidation that nonetheless exceed the rate of 292 in situ NH4+ production. Finally, from our temporally-resolved NH4+ concentration data, we 293 propose - for the first time - a measurement-based seasonal cycle for the mixed-layer NH₄+ pool 294 south of the SAF.

3. Methods

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3.1 Cruise tracks and sample collection

Samples were collected for a series of analyses on the southward (S) and northward (N) legs of a winter cruise between Cape Town, South Africa, and the marginal ice zone (MIZ) onboard the R/V SA Agulhas II (VOY25; 28 June to 13 July 2017) (Fig. 1). Samples were also collected for NH₄⁺ concentration analysis on three cruises onboard the R/V SA Agulhas II during 2018/19: early- and late summer samples were collected during the SANAE 58 Relief Voyage (6 December 2018 to 15 March 2019; VOY035); winter samples were collected during the SCALE 2019 (www.scale.org.za) winter cruise to the MIZ (18 July to 12 August 2019; VOY039); and spring samples were collected during the SCALE 2019 spring cruise to the MIZ (12 October to 20 November 2019; VOY040) (Fig. S1).

Leg S of VOY25 in winter 2017 crossed the Atlantic sector and due to logistical constraints, involved only surface underway collections, while leg N bordered the Atlantic and Indian sectors (30°E; WOCE IO6 line) and included eight conductivity-temperature-depth (CTD) hydrocast stations. Frontal positions were determined using the ship's hull-mounted thermosalinograph. supported by temperature, salinity, and oxygen concentration data from CTD measurements made during leg N. The salinity and oxygen sensors were calibrated against seawater samples that were analyzed for salinity using a Portasal 8410A salinometer and for dissolved oxygen by

Winkler titration (Strickland & Parsons, 1972). Frontal positions were determined from sharp

314 gradients in potential temperature, salinity, potential density, and oxygen concentrations (Belkin

315 & Gordon, 1996; Lutjeharms & Valentine, 1984; Orsi et al., 1995). For leg N, the mixed layer

depth (MLD) was determined for each Niskin (up)cast as the depth between 10 m and 400 m at 316 317

which the Brunt Väisälä Frequency squared, N^2 , reached a maximum (Carvalho et al., 2017).

318 During leg S, samples were collected every four hours from the ship's underway system (~7 m

319 intake; "underway stations") while samples on leg N were collected from surface Niskin bottles

320 (~10 m, approximately 55% light depth) mounted on the CTD rosette ("CTD stations"). NH₄⁺

321 samples were also taken at 13 depths over the upper 500 m at the CTD stations. At all stations,

322 40 mL of unfiltered seawater was collected for the analysis of NH₄⁺ concentrations in duplicate

323 50 mL high density polyethylene (HDPE) bottles that had been stored ("aged") with

324 orthophthaldialdehyde (OPA) working reagent. Unfiltered seawater was collected in duplicate Deleted: , particularly

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349 50 mL polypropylene centrifuge tubes for the analysis of NO₂-, NO₂-, and PO₄³-, and in a single 350 tube for urea. Immediately following collection, NH₄⁺ and nutrient samples were frozen at -20°C.

Duplicate size-fractionated chlorophyll-a samples were collected by filtering seawater (500 mL) 351

352 through 25 mm-diameter glass fibre filters (0.3 µm and 2.7 µm; Sterlitech GF-75 and Grade D,

353 respectively). Acetone (90%) was added to foil-wrapped borosilicate tubes containing the filters

354 and incubated at -20 °C for 24 hours. Duplicate seawater samples (4 L) were also gently vacuum-355 filtered through combusted 47 mm-diameter, 0.3 µm, GF-75 filters for POC and PON

concentrations and δ^{15} N-PON. Filters were stored in combusted foil envelopes at -80°C.

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357 For microscopy, unfiltered seawater samples (250 mL) were collected during leg S in amber 358 glass bottles and immediately fixed by the addition of 2.5 mL of Lugol's iodine solution (2%

final concentration), then stored at low room temperature in the dark until analysis. For flow 359

360 cytometry, seawater samples were collected in triplicate 2 mL microcentrifuge tubes, fixed with

glutaraldehyde (1% final concentration), and stored at -80°C until analysis (Marie et al., 2005).

Ten incubation experiments were conducted during leg S to measure net primary production (NPP) In addition, four NPP experiments were conducted during leg N using seawater collected from Niskin bottles fired at ≈10 m. In all cases, pre-screened (200-µm mesh; to remove large grazers) seawater was collected in three 2-L polycarbonate bottles to which NaH13CO3 was added at ~5% of the estimated ambient DIC concentration, ¹³C enrichment was re-calculated post-cruise using measured DIC concentrations, and these enrichments were used in all NPP rate calculations. Bottles were incubated for 5 to 6.5 hours in custom-built deck-board incubators shaded with neutral-density screens to mimic the 55% light level, and supplied with running surface seawater. Following incubation, each sample was divided (1 L per size fraction) and gently vacuum filtered through combusted 0.3 µm, and 2.7 µm glass fibre filters that were stored in combusted foil at -80°C until analysis.

N uptake (as NO₃, NH₄⁺ and urea) and NH₄⁺ oxidation experiments were conducted at five stations during leg S, with NH₄⁺ oxidation measured at two additional stations at the ice edge (Fig. 1). On leg N, experiments were also conducted using seawater collected from ~10 m at the same four CTD stations as the NPP experiments. Duplicate 1 L polycarbonate bottles were amended with ¹⁵N-labeled NO₃, NH₄ or urea at ~10% of the ambient N concentration, estimated based on past wintertime measurements (Mdutyana et al., 2020) and, in the case of NH₄+, coincident shipboard analyses. ¹⁵N enrichment was re-calculated post-cruise using the measured nutrient concentrations, and these enrichments were used in all rate calculations. Incubations were carried out as for NPP. For NH₄+ oxidation, duplicate black 250 mL HDPE bottles were amended with 0.1 μ M 15 NH₄⁺ and 0.1 μ M 14 NO₂⁻ (the latter as a "trap" for the 15 NO₂⁻ produced by NH₄⁺ oxidation, Ward 2011). NH₄⁺ oxidation bottles were incubated for 24 hours under the same temperature conditions as the N uptake and NPP experiments. Subsamples (50 mL) were collected from each bottle immediately following tracer addition (T₀) and at the end of the experiments (T_f), and frozen at -20°C until analysis.

3.2 Sample processing

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3.2.1. Ammonium concentrations

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426 On all cruises, NH₄+concentrations were measured shipboard using the fluorometric method of 427 Holmes et al. (1999) and a Turner Designs Trilogy fluorometer 7500-000 equipped with a UV 428 module. The detection limit, calculated as twice the pooled standard deviation of all standards, 429 was 0.06 μM. To prevent possible in/efflux of ammonia (NH₃) due to the temperature difference 430 between surface waters and the shipboard laboratory, samples were frozen immediately upon 431 collection, for a maximum of 24 hours. OPA working reagent was added to the frozen samples 432 prior to defrosting them for analysis. Samples were slowly warmed to room temperature in a 433 water bath after OPA addition, incubated in the dark for four hours once defrosted, and then each 434 replicate was measured in triplicate. Standards and blanks were made daily using Type-1 Milli-435 Q water. Precision was $\pm 0.03 \mu M$ for replicate samples and standards.

436 During VOY040 (spring 2019), we investigated the possibility that the ship's underway system 437 alters the seawater NH₄⁺ concentrations (e.g., due to contamination or cell breakage). We 438 collected surface samples from the underway and Niskin bottles concurrently and measured an 439 average NH₄⁺ concentration difference of 0.07 ± 0.15 µM (n=17), with no noticeable trend of 440 one method consistently yielding higher/lower concentrations. We thus have no reason to doubt 441 NH₄⁺ concentrations measured for seawater samples collected from the ship's underway system.

3.2.2. Macronutrient concentrations

442 443 Following the winter 2017 cruise, duplicate seawater samples were analysed manually for NO2 444 and PO₄³ (Bendschneider & Robinson, 1952; Murphy & Riley, 1962) using a Thermo Scientific 445 Genesys 30 Visible spectrophotometer. Precision and detection limit was \pm 0.05 μ M, and 0.05 446 μM for NO₂ and $\pm 0.06 \mu M$ and $0.05 \mu M$ for PO₄. The concentrations of NO₃ + NO₂ and 447 "Si(OH), were measured using a Lachat QuickChem 8500 Series 2 flow injection autoanalyzer. 448 Aliquots of a certified reference material (JAMSTEC) were measured during each run to ensure 449 measurement accuracy (SD \leq 2%). The precision of the NO₃ + NO₂ and Si(OH)₄ measurements 450 was \pm 0.4 μ M and \pm 0.2 μ M, respectively, and the detection limit was 0.1 μ M and 0.2 μ M. NO₃ 451 concentrations were calculated by subtraction (i.e., NO₃ + NO₂ - NO₂), with error propagated 452 according to standard statistical practices. Urea-N (hereafter, urea) concentrations were 453 determined via the room-temperature, single-reagent colorimetric method (Revilla et al., 2005) 454 using a Thermo Scientific Genesys 30 Visible spectrophotometer; precision was \pm 0.04 μM and 455 the detection limit was 0.04 µM. 456 3.2.3. Chlorophyll-a concentrations

457 Chlorophyll-a concentrations ([chl-a]) were determined shipboard using the nonacidified 458 fluorometric method (Welschmeyer, 1994). The Turner Designs Trilogy fluorometer was 459 calibrated with an analytical standard (Anacystis nidulans, Sigma-Aldrich®) prior to and 460 following the cruise. The [chl-a] of the 0.3-2.7 µm size class (hereafter, "pico" size class) was 461 calculated by subtracting the measured [chl-a] of the >2.7 µm size class (hereafter, "nano+" size 462 class) from the >0.3 µm size class (hereafter, "bulk"). Given previous work showing that the 463 winter Southern Ocean phytoplankton community is composed primarily of small cells (i.e., typically <15 μm; e.g., Hewes et al., 1985; 1990; Weber & El-Sayed, 1987), we did not separate 464 micro- from nanophytoplankton. 465

3.2.4. Bulk POC, PON and δ^{15} N-PON

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505 The NPP and N uptake filters were fumed with hydrochloric acid in a desiccator for 24 hours to 506 remove inorganic carbon, then dried for 24 hours at 40°C and packaged into tin cups. Filters for

507 δ^{15} N-PON were dried in the same way, but not acidified. Samples were analysed using a Delta

508 V Plus isotope ratio mass spectrometer (IRMS) coupled to a Flash 260 elemental analyser, with

a detection limit of 0.17 μ mol C and 0.07 μ mol N and precision of ± 0.005 At% for C and

509 510 N. <u>Unused</u> pre-combusted filters (blanks) were <u>included in</u> each batch run, POC and PON content

511 was determined from daily standard curves of IRMS area versus known C and N masses. For the

512 isotope ratios, sample measurements were referenced to internal laboratory standards calibrated

513 against IAEA reference materials that were measured after every 5-7 samples.

514 3.2.5. Size-fractionated rates of NPP and N uptake

515 Carbon and N uptake rates (NPP, ρNH_4^+ , ρNO_3^- , $\rho Urea$) were calculated according to Dugdale

516 & Wilkerson (1986) as:

$$\rho M = \frac{[PM] x (At\%_{meas} - At\%_{amb})}{T x (At\%_{init} - At\%_{amb})}$$
(Eqn 1)

518 where,
$$At\%_{init} = \frac{(\lceil M \rceil \times At\%_{amb}) + (\lceil M_{tracer} \rceil \times At\%_{tracer})}{\lceil M \rceil + \lceil M_{tracer} \rceil}$$
 (Eqn 2)

519 Here, M is the species of interest (C, NH₄+, NO₃-, or urea); ρM is the uptake rate of that species

520 (nM hour⁻¹, i.e., nmol C or N L⁻¹ hour⁻¹); [PM] is the concentration of POC or PON (μM) on the

521 filters; [M] is the ambient concentration of DIC, NH₄₊, NO₃-, or urea at the time of sample

522 collection; [M_{tracer}] is the concentration of NaH¹³CO₃, ¹⁵NH₄+, ¹⁵NO₃-, or ¹⁵N-urea added to the

523 incubation bottles; and T is the incubation period (days). DIC concentrations were measured

524 shipboard using a VINDTA 3C instrument and ranged from 2017 to 2130 µM (Bakker et al.,

525 2016). The PM and ρM of the picoplankton size class was calculated by subtracting the

526 nanoplankton from the bulk measurements. Daily rates were computed by multiplying the hourly

rates by the number of daylight hours, the latter calculated using the sampling latitude and day

of the year (Forsythe et al., 1995).

The f-ratio (Eppley & Peterson, 1979), used to estimate the fraction of NPP potentially available

530 for export, was calculated as:

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$$f - ratio = \frac{\rho N O_3^-}{\sqrt{\rho N_{tot}}}$$
 (Eqn 3)

532 where $\rho N_{\text{tot}} = \rho N H_4^+ + \rho N O_3^- + \rho U rea$. Urea uptake was not measured at underway stations

50.7°S and 55.5°S (both in the Antarctic Zone); here, the f-ratio was calculated omitting oUrea. 533

534 For the two Antarctic Zone stations at which urea uptake was measured, including oUrea

decreased the f-ratio by 8-25% compared to that calculated using only ρNO₃- and ρNH₄+. 535

536 3.2.6. Ammonia oxidation rates

The azide method (McIlvin and Altabet 2005) was used to convert NO₂ produced by NH₄+

oxidation to N2O gas that was measured using a Delta V Plus IRMS with a custom-built purge-

539 and-trap front end (McIlvin & Casciotti, 2011). This configuration yields a detection limit of 0.2 Deleted: C

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Deleted: standardised to Merck Gel (δ^{15} N = 7.5‰, δ^{13} C = -20.1%; Merck), Valine (δ^{15} N = 12.1%, δ^{13} C = -26.8%; Sigma), Choc (δ^{15} N = 4.3%, δ^{13} C = -17.8%), and NH₄Cl (δ^{15} N = -0.6%),

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Deleted: >0.3 μm-filter (i.e.,

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Deleted: The specific carbon fixation rate (V_C) was calculated as ρC/POC and the specific uptake rate of total N (V_{Ntot}) was calculated as $\rho N_{tot}/PON$ (where $\rho N_{tot} = \rho N H_4^+$ $\rho NO_3^2 + \rho Urea$).

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nmol N with a δ^{15} N precision of \pm 0.1‰. The δ^{15} N of NO₂⁻ was derived from 45 N₂O/ 44 N₂O and the rate of NH₄⁺ oxidation (NH₄⁺ox· nM day⁻¹) was calculated following Peng et al. (2015) as:

$$NH_{4 \text{ ox}}^{+} = \frac{\Delta_{0}^{[15}NO_{2}^{-}]}{f_{NH_{2}^{+}}^{15} \times T}$$
 (Eqn 4)

Here, Δ[15NO₂-] is the change in the concentration of 15NO₂- (nM) between the start and end of the incubation, calculated as the difference in the measured δ15N of NO₂- between the T_f and T₀ samples, f¹⁵_{NH4} is the fraction of the NH₄+ substrate labelled with 15N at the start of the incubation, and T is the incubation length (days). All 15NO₂- produced during the incubations was assumed to derive from 15NH₄+ oxidation. The detection limit ranged from 0.02 to 0.11 nM day⁻¹, calculated according to Santoro et al. (2013).

We note that isotope dilution (i.e., the dilution of ¹⁵NH₄+ by co-occurring ¹⁴NH₄+ regeneration) during the NH₄+ uptake and oxidation experiments could potentially lead to an underestimation of the rates (Glibert et al., 1982; Mdutyana, 2021). For the NH₄+ uptake experiments, their short duration (3 to 7.5 hours) would have rendered the effect of regeneration minor (Mdutyana et al., 2020). Moreover, the ¹⁵NH₄+ additions were high (100 nM) relative to both the ambient NH₄+ concentrations north of the SAF and the K_m values derived for NH₄+ uptake and oxidation in the winter Southern Ocean (150-405 nM and 28-137 nM, respectively; Mdutyana, 2021), making a significant dilution effect unlikely (Lipschultz, 2008). Finally, at the stations south of the SAF, the ambient NH₄+ concentrations were so high that even if the regeneration of ¹⁴NH₄+ occurred at an elevated rate (e.g., 50 nM day⁻¹; as has been measured in the late-summer Southern Ocean when remineralization is expected to be high; Goeyens et al., 1991), the ¹⁵N/¹⁴N of the NH₄+ pool would decrease by <1-2%. We thus consider the potential effect of isotope dilution to be minor.

A further consideration is possible stimulation of the NH₄⁺ uptake and oxidation rates by ¹⁵NH₄⁺ addition (Lipschultz, 2008). Given the K_m values listed above and the high ambient NH₄⁺ concentrations measured in the PFZ and AZ, a stimulation effect could only be significant at the stations north of the SAF where the NH₄⁺ concentrations were 10-100 nM, and even then, to a lesser extent for NH₄⁺ oxidation than NH₄⁺ uptake given that ammonia oxidizers in the winter Southern Ocean become saturated at NH₄⁺ concentrations of 100-200 nM (Mdutyana, 2021). The rates reported for the stations north of the SAF should therefore be considered "potential rates." However, since our focus is mainly on explaining the accumulation of NH₄⁺ south of the SAF, having "potential" rather than "true" rates for the STZ and SAZ does not affect our conclusions.

3.2.7 Plankton community composition

Microplankton groups (>1.5 μm) were identified and counted in a subsample (20 mL) from each, amber bottle using the Utermöhl technique (Utermöhl, 1958) and following the recommendations of Hasle (1978). Plankton groups and individual species were counted and identified using an inverted light microscope (Olympus CKX41) at 200x magnification. This level of magnification limited the cell sizes that could be reliably distinguished to >15 μm. For each sample, at least 100 cells were enumerated to ensure a statistically valid count.

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628 Pico- and nanoplankton cells (<15 μm) were enumerated using an LSR II flow cytometer (BD 629 Biosciences) equipped with blue, red, violet, and green lasers. Prior to analysis, 1 mL of sample 630 was incubated with 1% (v/v) SYBR Green-L(a DNA stain) at room temperature in the dark for 631 10 minutes (Marie et al., 1997). From light scatter and autofluorescence, the DNA-containing 632 particles were identified as nano- and picoeukaryotes, and Synechococcus. Additionally, small 633 heterotrophic prokaryotes (i.e., bacteria and possibly archaea; hereafter "bacteria") were 634 identified as DNA-containing particles with the lowest detectable autofluorescence (Marie et al., 635 1997; Gasol & del Giorgio, 2000) (see also Text S2). All particles lacking DNA were considered 636 detritus. The populations of interest were gated using FlowJo 10.3 software (TreeStar, Inc.; 637 www.flowjo.com). 638 In this study, we did not directly measure NH₄+ regeneration (i.e., heterotrophy). Instead, we use 639 the abundance of heterotrophic bacteria as a qualitative indicator of NH₄⁺ regeneration potential, 640 recognizing that cell abundance does not imply activity. Additionally, we estimate the rate of 641 NH₄+ production from our concentration and rate data (see section 3.3). The availability of 642 organic matter to heterotrophs is inferred from the abundance of detritus. 643 3.3 Mixed-layer NH₄⁺ residence time and NH₄⁺ production rate estimates 644 The residence time of the mixed-layer NH₄+ pool can be estimated using the measured ambient 645 NH₄⁺ concentrations and corresponding NH₄⁺ removal rates as $NH_{4}^{+}_{residence\ time} = \frac{[NH_{4}^{+}]}{NH_{4}^{+}_{removal\ rate}}$ 646 (Eqn 5) Here, NH₄⁺_{residence time} is the time period (days) over which a given NH₄⁺ concentration will be 647 648 depleted assuming a constant $NH_{4+removal rate}$. We set $NH_{4+removal rate} = \rho NH_{4+} + NH_{4+ox}$ in winter 649 and = ρNH₄⁺ in late summer given the evidence for negligible mixed-layer NH₄⁺ oxidation rates 650 in this latter season (Bianchi et al., 1997; Mdutyana et al., 2020). 651 652 To determine the contribution of late summer NH₄+ production to the wintertime NH₄+ pool (see 653 section 5.2), we define a rate of NH₄+concentration decline: 654 $_{NH_{4 \, rate \, of \, decline}}^{+} = NH_{4 \, production \, rate}^{+} - NH_{4 \, removal \, rate}^{+}$ (Eqn 6) 655 Here, NH₄⁺_{production rate_}is the NH₄⁺ flux required to compensate for NH₄⁺ removal over the late-656 summer-to-winter period, in order to yield the observed seasonal change in the ambient NH4+ 657 concentration. 658 The rate of NH₄+concentration decline can also be defined as: $NH_{4 \text{ rate of decline}}^{+} = \frac{[NH_{4}^{+}]_{\text{decline}}}{t}$ 659 Where [NH₄+]_{decline} is the difference between the late summer and winter NH₄+ concentrations 660 661 and t is the time period (days) over which the NH₄⁺ concentration declines. Setting Eqn 6 and 7

 $NH_{4 \text{ production rate}}^{+} = \frac{NH_{4}^{+}|_{decline}}{t} + NH_{4 \text{ removal rate}}^{+}$

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equal yields:

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Deleted: Pacific Blue (450/50). DNA-containing cells were isolated in each sample based on their detected

Deleted: on the FITC bandpass filter. Subsequently, based on their detected autofluorescence on the APC bandpass filter relative to the PE bandpass filter, the isolated DNA-containing cells were grouped into the following populations: Nano...

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Deleted: www.flowjo.com). Relative cellular sizes were determined using 60 μL of SPHERO™ Blank Calibration Particles, 1.8 – 2.2 μm in diameter, added to 1 mL of selected samples to yield a final concentration of ~6x10⁵ particles mL¹. Relative to the 1.8 – 2.2 μm calibration beads, nanoeukaryotes were larger than 2.2 μm, picoeukaryotes and heterotrophic cells were smaller than 1.8 μm, and Synechococcus exhibited a range of sizes around 2 μm, with two distinct subgroups; one of ~2 μm in size and another slightly larger than 2.2 μm (see Fig. S1). Synechococcus was isolated from the nanoeukaryotes by its pigment characteristics – both subgroups of Synechococcus had high PE relative to APC content (Barlow et al., 1985), whereas nanoeukaryotes had high APC and PE.

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Deleted:) were made in this study, potential heterotrophic activity is evaluated from

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(Eqn 8)

724 Where, $NH_4^+_{consumption \ rate} = \rho NH_4^+ + NH_4^+_{ox}$. Eqns 7 and 8 assume that the elevated wintertime Formatted: Justified, Tab stops: Not at 2.75 cm + 3 cm + 725 NH₄⁺ concentrations result from continuous NH₄⁺ production in excess of removal rather than 726 from sporadic events of removal and/or production occurring between late summer and winter. 727 3.4 Statistical analyses

728 The correlations among latitude, N concentrations, NPP, N assimilation rates, and NH₄⁺ oxidation

rates were investigated at the 5% significance level using the Pearson correlation coefficient and

730 the R packages, stats (R Core Team, 2020) and corrplot (Wei & Simko, 2017). Standard

731 deviations were propagated using standard statistical practices. Deleted: inorganic carbon and

Deleted: uptake

4. Results

4.1 Hydrography

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749

734 Sea surface temperature (SST) decreased by ~17 °C between Cape Town (~34°S) and the edge 735

of the MIZ (61.7°S), with similar gradients measured for legs S and N. During leg N, fairly deep

736 MLDs were observed (124-212 m), similar to June and July climatological MLDs compiled from

737 Argo float data for this region (Dong et al., 2008). While the focus of this study is the surface

738 (i.e., upper ~10 m), we <u>report</u> the <u>MLDs</u> here to <u>show</u> that sampling took place under typical

739 winter conditions, with the deep MLDs evincing ongoing winter mixing and associated nutrient

740 recharge. Where not specified, the trends discussed below refer to the surface data only.

741 Latitudinal variations in each parameter are assessed by comparing the various Southern Ocean

742 zones - the Subtropical Zone (STZ) north of the Subtropical Front (STF), the Subantarctic Zone

743 (SAZ) between the STF and the Subantarctic Front (SAF), the Polar Frontal Zone (PFZ) between

744 the SAF and the Polar Front (PF), and south of the PF, the Open and Polar Antarctic Zones (OAZ

745 and PAZ, which are divided by the Southern Antarctic Circumpolar Current Front (SACCF) and

746 collectively termed the Antarctic Zone (AZ); see Text S1 for detailed definitions of the fronts

747 and zones, and Fig. 1 and S1 for their positions at the time of sampling). For each parameter, the

748 $\underline{average \pm 1 \text{ standard deviation (SD)}}$ for each Southern Ocean zone is reported in Table 1.

4.2 Macronutrient concentrations

750 In winter 2017, the surface and mixed-layer concentrations of NH₄⁺ ranged from below detection

751 to 0.70 µM (Fig. 2a and b). Surface concentrations were higher in the PFZ, OAZ, and PAZ (0.42

752 \pm 0.01 μ M, 0.52 \pm 0.01 μ M, and 0.58 \pm 0.01 μ M, respectively) than in the STZ and SAZ (0.08 \pm

753 $0.03~\mu M$ and $0.06\pm0.01~\mu M$, respectively), with a sharp gradient observed at the SAF, South of

754 the SAF, high NH₄⁺ concentrations persisted near-homogeneously throughout the mixed layer,

755 with mixed layer averages ranging from $0.65 \pm 0.01 \,\mu\text{M}$ at station 58.5°S to $0.27 \pm 0.01 \,\mu\text{M}$ at

756 station 48.0°S and averaging $0.47 \pm 0.02 \mu M$, with concentrations that were below detection

757 north of the SAF (Fig. 2b). Below the mixed layer, NH₄+ concentrations decreased rapidly at all

758 stations to values below detection by 200 m.

759 The concentrations of $\underline{NO_3}$ and $\underline{PO_4}^3$ increased southwards from $\leq 10 \, \mu M$ and $\leq 1 \, \mu M$ in the STZ

760 to >20 μM and >1.5 μM in the PFZ, OAZ, and PAZ (Fig. 2c and S3a), with the sharpest gradients

761 occurring near the SAF. The concentrations of Si(OH)₄ increased rapidly across the PF, from an

762 average of $3.2\pm1.1~\mu\text{M}$ between 35.0°S and 48.0°S to $45.6\pm0.6~\mu\text{M}$ between 52.1°S and 58.9°S Deleted: from

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790 (Fig. <u>\$3b</u>). The NO_2 concentrations were consistently low across the transect (0.16 \pm 0.02 μ M;

791 Fig. S3c), as were the concentrations of urea $(0.20 \pm 0.04 \mu M_{\odot})$ Table 1), with slightly lower urea

792 concentrations observed in the SAZ than in the other zones.

793 4.3 Chlorophyll-a, POC and PON

794 The highest bulk, [chl-a] was observed near the South African continental shelf, decreasing across

795 the STF and remaining low thereafter (Fig. 3a). The proportion of chl-a in the nano+ size class.

796 varied across the region but was >50% at all stations, with higher (>80%) contributions near the

797 fronts and at many OAZ and PAZ stations (Fig. 3b). The nano+ contribution was ≤60% at only

798 five stations (three in the SAZ, two in the OAZ).

799 The concentrations of bulk POC and PON were highest north of the STF and slightly higher in

800 the OAZ than in the SAZ and PFZ (Fig. S4a and b). The contribution of the nano+ size fraction

801 to POC and PON across the transect was $77.1 \pm 22.6\%$ and $66.9 \pm 24.2\%$, respectively (Fig. S4c)

802 and d). The δ^{15} N-PON decreased southwards from the STZ and SAZ (1.7 \pm 1.0%) to the PFZ

803 and OAZ (0.5 \pm 0.5%; Fig. 4). Despite considerable differences among zones, the δ^{15} N-PON

804 was relatively homogenous within each zone.

805 4.4 Rates of net primary production, nitrogen uptake, and ammonium oxidation

806 Rates of bulk NPP were two- to six-fold higher in the SAZ and PFZ than has been reported

807 previously for the Atlantic sector in winter (Mdutyana et al., 2020; Froneman et al., 1999) (Fig.

808 5a). By contrast, NPP was low in the OAZ, consistent with previous measurements (Kottmeier

& Sullivan, 1987; Mdutyana et al., 2020). The relative contribution of the nano+ size class 809

generally decreased southwards, from \$5.4% at 37.0°S to 24.4% at 53.5°S, before increasing to 810

811 ≥80% near the SACCF.

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812 The bulk NH₄+ uptake rates (ρNH₄+) generally increased southwards from the STZ to the SAZ

and PFZ, and then decreased across the OAZ to reach a minimum at the southernmost station

814 (Fig. 5b). In the nano+ size fraction, ρNH₄+ changed little latitudinally, although it was slightly

815 lower in the PFZ than in the other zones. The contribution of nanoplankton to ρNH₄⁺ ranged from

816 32.8% in the PFZ to 71.9% in the STZ. The bulk NO_3 uptake rates (ρNO_3) were also low in the

817 STZ, while the highest pNO₃ was measured in the SAZ, with the rate then decreasing

818 southwards. ρNO_3 in the nano+ size class followed the same trend as total community ρNO_3 ,

819 with the nanoplankton accounting for $71.5 \pm 0.3\%$ of bulk ρNO_3 on average. The rates of bulk

820 urea uptake (ρUrea) were highest in the STZ, with the SAZ and the PFZ hosting similar rates,

821 and the lowest rates were measured in the OAZ. ρUrea for the nano+ size class followed a similar

822 trend to bulk ρUrea, and nanoplankton accounted for 51.8% of ρUrea in the SAZ, increasing to

823 100% in the PAZ. The uptake rates of the different N forms were not significantly correlated

824 with one another or with the ambient N concentrations (Table S1).

Ammonium oxidation rates (NH₄+_{ox}) increased southwards, with higher NH₄+_{ox} in the OAZ and

826 PAZ than in the STZ, SAZ, and PFZ (Fig. 5c). NH₄ ox was generally comparable to previous

wintertime measurements from the surface of the open Southern Ocean (Mdutyana et al., 2020).

828 NH₄⁺_{ox} was not correlated with the ambient NH₄⁺ concentration (<u>Table S1</u>). Deleted: S2c

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Deleted:), consistent with previous autumn and winter studies (Froneman et al., 1999; Philibert et al., 2015; Scharek et al., 1994).

Deleted: >2.7 µm

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Deleted: The ratio of bulk POC:chl-a (weight:weight) was on average low in the STZ, SAZ, and PFZ, and reached a maximum in the OAZ (Fig. 4a). Contrastingly, the ratio of POC:PON (mol:mol) appeared to decrease southwards, although there was no significant difference among zones (pvalue > 0.05) (Fig. 4b).

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871	4.5 Plankton community composition	ļ
872	Microplankton abundance was Jow, with the highest cell counts recorded at stations 37.2°S and	B
873	41.3°S in the STZ and no cells counted at 38.1°S (STZ) and 55.5°S (OAZ) (Fig. 6a). On average,	
874	microplankton abundance was higher in the STZ than in the SAZ, PFZ, and OAZ. The greatest	_
875	diversity of microplankton groups was observed at 41.3°S in the STZ and at 50.0°S near the PF.	
876	Centric diatoms (including Planktoniella, Coscinodiscus, and Thalassiosira species) were	1
877	detected only at the southernmost station, 58.9°S (3 cells mL ⁻¹). Pennate diatoms (including	K
878	Pseudo-nitzschia, Pleurosigma, and Navicula species) were more abundant in the STZ, PFZ, and	,
879	OAZ, with negligible abundances in the SAZ. Higher pennate diatom abundances occurred near	Ζ,
880	the PF (7 cells mL ⁻¹), as has been observed in summer (e.g., Bracher et al., 1999). Dinoflagellates	
881	were identified at every station except 38.1°S and were most abundant in the STZ and PFZ. At	1
882	all but three stations, small (~15 μm) dinoflagellates were the most abundant group, although the	//
883	larger Protoperidinium dinoflagellate species (mainly heterotrophic; Jeong & Latz, 1994) were	//,
884	almost as abundant in the PFZ and at 54.0°S. Microzooplankton (i.e., ciliates, 20-200 μm) were	K
885	most abundant in the STZ, and were also present in the PFZ at 46.1°S (3 cells mL ⁻¹) and 48.9°S	
886	(3 cells mL^{-1}) and in the OAZ at 50.0°S (1 cells mL^{-1}) and 54.0°S (4 cells mL^{-1}). All other	
887	stations were characterized by negligible (<1 cells mL ⁻¹) microzooplankton abundances.	
888	Nano- and picoeukaryotes, Synechococcus, and heterotrophic bacteria (collectively, "small	/,
889	cells"), were roughly 10 ³ -times more abundant than the microplankton (Fig. 6b). Notwithstanding	/
890	a lack of data from the STZ, the highest small cell abundances occurred in the SAZ near the SAF.	
891	Across the transect, picoeukaryotes were generally more abundant than all other phytoplankton	1
892	groups (average picoeukaryote contribution to total small cells of 12-54%; nanoeukaryotes of 7-	$/\!\!/$
893	39%; Synechococcus of 15-42%). A similar trend has been observed for the Southern Ocean in	//
894	spring (Detmer & Bathmann, 1997) and late summer (Fiala et al., 1998), in contrast to mid-	$/\!/$
895	summer observations showing nanoplankton dominance (e.g., Ishikawa et al., 2002; Weber &	/ ,
896	El-Sayed, 1987). Additionally, picoeukaryotes were two- to three orders of magnitude more	/
897	abundant in the SAZ and PFZ than in the OAZ. Nanoeukaryotes dominated near the PF at 50.0°S	//
898	(39%) and in the southern OAZ at 55.5°S (36%), while Synechococcus dominated at 42.7°S and	
899	54.0°S (42% and 33%, respectively). <u>In general, nanoeukaryote</u> abundance was higher in the	M_{I}
900	SAZ than in the PFZ and OAZ, as was that of Synechococcus.	///
901	The contribution of heterotrophic bacteria to total small cells varied considerably (10-62%),	
902	reaching a maximum south of the PF at 53.0°S and 57.8°S (62% and 50%), and with higher	I/
903	abundances in the SAZ than in the PFZ and OAZ (Fig. 7). Additionally, heterotrophic bacterial	Ζ,
904	abundances were ten-fold lower to two-fold higher than the total pico- and nanophytoplankton	
905	cell counts. Detrital particles were most abundant near the southern edge of the SAF, and were	K
906	generally more abundant in the PFZ than in the SAZ and OAZ (Fig. <u>\$5</u>).	<
907	4.6 2018/19 cruises: ammonium concentrations	
908	In early summer, surface NH ₄ + concentrations were uniformly low across the transect (average	
909	of 0.11 + 0.09 µM. Fig. 8a). South of the SAF, NH ₄ ⁺ increased to an average concentration of	1

 $0.81 \pm 0.92~\mu\text{M}$ by late summer (Fig. 8b). By winter 2019, the NH₄⁺ concentrations south of the SAF were ~40% lower than they had been in late summer (Fig. 8c), and were similar to those

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957 observed in winter 2017 (0.50 \pm 0.30 μ M and 0.52 \pm 0.11 μ M, respectively), confirming that our 958 2017 observations are generally representative of the wintertime Southern Ocean. By early 959 spring, the NH₄⁺ concentrations south of the SAF had declined to near or below detection (0.09 960 ± 0.08 μM; Fig. 8d) before rising again by late spring to an average value only slightly lower 961 than that measured in winter (0.40 \pm 0.74 μ M; Fig. 8e). However, the late-spring NH₄⁺ 962 concentrations were only elevated in the PFZ (range of 0.11 ± 0.01 to 4.39 ± 0.03 µM, average 963 of $0.77 \pm 1.11 \mu M$), as has been observed previously (Bathmann et al., 1997). Excluding the PFZ 964 data yields a far lower late-spring average of $0.17 \pm 0.11 \,\mu\text{M}$ south of the SAF, which we take 965 as more broadly representative of this season.

4.7 Mixed-layer NH₄⁺ residence time and NH₄⁺ production rate estimates

The NH₄⁺_{residence time} in winter 2017 south of the SAF, computed using Eqn 5, ranged from 10 to 967 968 38 days (median of 21 days). These values were estimated using wintertime measurements only and as such, may not be representative of the transition from summer to winter. To refine our 969 970 estimates, we use the average ρNH_4^+ and NH_4^+ concentration measured south of the SAF in late 971 summer (50.6 ± 24.0 nM day⁻¹ and 0.81 ± 0.92 μ M, respectively; Deary, 2020), which yields an

972 NH₄⁺residence time of 2 to 27 days (median of 5 days).

> The NH₄+production rate, calculated using Eqn 8 and an [NH₄+]_{decline} of 330 nM (i.e., 810 nM – 480 nM), t of 141 days, and NH₄⁺_{removal rate} of 50.6 ± 24.0 nM day⁻¹ (here, the average late-summer ρNH_4^+ south of the SAF is used to approximate NH_4^+ removal rate), was 52.9 ± 25.0 nM day-1. If we instead use the average NH₄⁺_{removal rate} and NH₄⁺ concentration measured in winter 2017 (21.4 ± 0.6 nM day^{-1} and $520 \pm 110 \text{ nM}$), the NH₄+production rate was $23.4 \pm 6.6 \text{ nM day}^{-1}$. Using the range of NH₄+_{removal rate} values and the average ambient NH₄+ concentration measured south of the SAF in winter 2017 (16.7 to 31.2 nM day-1 and 520 nM) and late summer 2019 (22.6 to 98.6 nM day-¹ and 810 nM), we calculate that over the late-summer-to-winter transition, the NH₄+_{production rate} ranged from 18.8 to 100.9 nM day-1.

Discussion

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5.1 Drivers of NH₄+ cycling in the surface layer of the Southern Ocean

Previous work has suggested that NH₄⁺ accumulates in the Southern Ocean mixed layer following the late summer increase in heterotrophy, then decreases into autumn as heterotrophic activity subsides, to be depleted by winter due to advective processes and biological removal (Koike et al., 1986; Serebrennikova & Fanning, 2004). However, our data show that NH₄+ concentrations are elevated in the mixed layer in winter, particularly south of the SAF (Fig. 2). Similarly elevated winter surface-layer NH₄⁺ has been observed previously in both the Atlantic and Indian sectors, with concentrations typically increasing towards the south (Philibert et al., 2015; Mdutyana et al., 2020; Bianchi et al., 1997). Numerous overlapping processes are likely involved in setting the ambient NH₄⁺ concentrations, as summarized in Fig. 2. In this study, we directly measured the rates of NH₄ uptake and oxidation, and estimated the rates of NH₄ production, along with qualitatively evaluating heterotrophy from the relative abundance of heterotrophic bacteria, phytoplankton, and detritus. For the NH₄⁺ cycle processes shown in Fig. 2 that are not quantified or inferred from our dataset, we consider their potential role in Southern Ocean NH₄+ cycling based on findings reported in the literature.

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<#>5.1 Drivers of NH₄+ cycling in the surface layer of the Southern Ocean

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Deleted: <#>by different size fractions of the winter plankton community, as well as

Deleted: <#>oxidation. We infer the contribution of heterotrophic bacteria and microzooplankton to NH4+

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1026 The high NH₄+ concentrations observed south of the SAF in winter, may result from net NH₄+ 1027 accumulation during late summer, autumn, and/or winter. The persistence of elevated NH4+ 1028 concentrations that are near-homogeneously distributed throughout the mixed layer is consistent 1029 with a residence time for the winter NH₄⁺ reservoir in excess of the time-scale for upper-ocean 1030 mixing. Indeed, we calculate a median residence time of 21 days south of the SAF, compared to 1031 2 days north of the SAF. One implication of the long residence time computed for the polar zones 1032 is that the wintertime NH4+ pool likely reflects both ongoing processes and those that occurred 1033 earlier in the year. We posit that the elevated NH₄+ concentrations south of the SAF may result 1034 from higher wintertime rates of NH₄⁺ production than removal and/or from the gradual but 1035 incomplete depletion in winter of NH₄⁺ produced mainly in late summer and autumn. We evaluate 1036 both possibilities throughout the discussion below.

5.1.1 Ammonium removal

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Ammonium assimilation - Microbial growth is limited in the winter Southern Ocean (Arrigo et al., 2008; Smith Jr et al., 2000, Takao et al., 2012), resulting in low cell abundances and nutrient uptake rates (Church et al., 2003; Iida & Odate, 2014; Mdutyana et al., 2020). However, while the concentrations of chl-a and rates of NPP were low across our transect, they were not negligible (Fig. 3a and 5a), consistent with previous reports for this season (Mordy et al., 1995; Pomeroy & Wiebe, 2001). Southern Ocean phytoplankton are adapted to survive suboptimal conditions; for example, numerous species achieve their maximum growth rates at temperatures that are considerably lower than the optimal growth temperatures of temperate and tropical species (2-9 °C versus 10-30 °C and 15-35 °C, respectively), with sharp declines in growth rates observed at temperatures outside this range (Boyd et al., 2013; Coello-Camba & Agusti, 2017; Fiala & Oriol, 1990). In addition, ice-free Southern Ocean waters typically extend to <60°S in the eastern Atlantic and western Indian sectors in winter, so that even though irradiance levels may not be optimal for phytoplankton growth, there is always some light available for photosynthesis. The hostile wintertime conditions of the open Southern Ocean do not, therefore, prevent ecosystem functioning, although the microbial dynamics and associated biogeochemical processes differ from those occurring in summer (Smart et al., 2015; Mdutyana et al., 2020).

previous wintertime observations (ranging from 32-66 nM day⁻¹; Cota et al., 1992; Mdutyana et al., 2020; Philibert et al., 2015). Such low rates, if generally representative of winter, would limit mixed-layer NH₄⁺ drawdown, especially south of the PF where ρNH₄⁺ was particularly low. Recycled N (NH₄⁺ + urea) nonetheless accounted for most of the N <u>assimilated during winter</u>,

We measured fairly low surface NH₄+ uptake rates (3.0-13.2 nM day⁻¹; Fig. 5b) compared to

including in the AZ (Fig. 5b).

1060 The <u>available</u> δ^{15} N-PON data suggest that <u>the preferential</u> reliance <u>of phytoplankton</u> on recycled 1061 N may have persisted from the late summer. In theory, PON generated in early- through mid-1062 summer from the assimilation of upwelled NO₃⁻ (δ^{15} N-NO₃⁻ of 5.2% in the AZ and 6.2% in the 1063 SAZ; Smart et al., 2015; Fripiat et al., 2019; 2021) will have a δ^{15} N of ~0% in the AZ and 1-2% 1064 in the SAZ given the isotope effect of NO₃- assimilation and the degree of seasonal NO₃-1065 drawdown (Sigman et al., 1999; Granger et al., 2004; 2010). Such δ^{15} N-PON values have indeed 1066 been measured in the early- and mid-summer Southern Ocean (Lourey et al., 2003; Smart et al., 1067 2020; Soares et al., 2015). By late summer, δ^{15} N-PON has been observed to decline to between

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1099 -5 and -1%, with the lowest values occurring in the AZ (Lourey et al., 2003; Smart et al., 2020; Deleted: to 1100 Trull et al., 2008). Since the δ^{15} N of recycled N is expected to be low (<0%; Checkley & Miller, Deleted: 1101 1989, Macko et al., 1986), the early-to-late summer decline in δ^{15} N-PON implicates a switch Deleted: 1102 from dominantly NO₃-- to dominantly recycled N-supported phytoplankton growth (Lourey et 1103 al., 2003). For the SAZ, the subsequent late summer-to-winter rise in δ^{15} N-PON (i.e., from \sim -1104 1‰ to 1-2.5‰; Fig. 4) has previously been attributed to PON decomposition by heterotrophic Deleted: 4c 1105 bacteria (Smart et al., 2020), during which ¹⁴N-NH₄⁺ is preferentially remineralized, leaving the 1106 remaining PON enriched in ¹⁵N (Möbius, 2013). That NH₄+ concentrations are not elevated in 1107 the SAZ mixed layer in winter (Fig. 2b) indicates that the remineralized NH₄⁺ is rapidly re-Deleted: .) 1108 assimilated by phytoplankton and/or oxidized to NO₂ in this zone. In the AZ, the much lower 1109 $\delta^{15}\text{N-PON}$ of -3 to -1‰ that we observe in winter surface waters requires the sustained 1110 <u>assimilation</u> of low- δ^{15} N N (i.e., recycled N) to offset a remineralization-driven δ^{15} N rise <u>akin</u> to Deleted: consumption 1111 that of the SAZ. We conclude that Southern Ocean phytoplankton preferentially consume Deleted: NH4+ and urea 1112 regenerated N from late summer until at least July (albeit at low rates in winter), particularly Deleted: similar 1113 Deleted: dominantly south of the PF. 1114 The fact that NH4+ accumulated in the winter mixed layer despite, being the preferred Deleted: the 1115 phytoplankton N source in late summer through winter implies that low rates of NH₄⁺ uptake Deleted: concentration was high 1116 contributed to its accumulation. Multiple factors may cause low rates of photoautotrophic NH₄⁺ Deleted: NH4+ 1117 assimilation, including deplete NH4+ and micronutrient concentrations, light limitation, and low Deleted: the 1118 temperatures. North of the SAF, NH₄+ concentrations below detection likely limited ρNH₄+, as Deleted: of this N form. 1119 evidenced by the fact that in a series of experiments conducted on the same cruise, ρNH₄⁺ Deleted: uptake 1120 increased with the addition of NH₄⁺ at these stations (Mdutyana, 2021). By contrast, south of the 1121 SAF, NH₄⁺ concentrations were similar to or higher than the half-saturation constant (K_m) derived 1122 for NH₄⁺ uptake in the winter Southern Ocean (0.2 to 0.4 μM; Mdutyana, 2021), suggesting that 1123 something other than NH₄⁺ availability was limiting to phytoplankton at these latitudes. 1124 Iron is not directly involved in NH4+ assimilation but is required for electron transport during 1125 photosynthesis and respiration (Raven, 1988). While iron limitation is widespread across the 1126 Southern Ocean (Janssen et al., 2020; Pausch et al., 2019; Viljoen et al., 2019), iron availability 1127 appears to be higher in winter than during other seasons (Mtshali et al., 2019; Tagliabue et al., 1128 2014) due to enhanced mixing, storms, and increased aeolian deposition (Coale et al., 2005; 1129 Honjo et al., 2000; Sedwick et al., 2008). The fact that ρNO₃⁻ and ρNH₄⁺ were generally similar 1130 across the transect (Fig. 5b) argues against a dominant role for iron in controlling ρNH₄+ since 1131 NO₃⁻ assimilation has a far higher iron requirement than NH₄⁺ assimilation (Morel et al., 1991). Deleted: consumption 1132 In contrast to NH4+ and iron availability, light limitation is exacerbated in winter due to low 1133 insolation, increased cloud-cover, and mixed layers that can be hundreds of meters deeper than 1134 the euphotic zone (Buongiorno Nardelli et al., 2017; Sallée et al., 2010). Light is thus often Deleted: Brightman & Smith Jr., 1989; 1135 considered the dominant constraint on Southern Ocean primary productivity in this season 1136 (Thomalla et al., 2011; Llort et al., 2019; Wadley et al., 2014). However, since NH₄+assimilation Deleted: consumption 1137 by phytoplankton is fairly energetically inexpensive (Dortch, 1990), it should occur even under 1138 low light conditions (recognizing that light remains critical for coincident CO2 fixation). 1139 Heterotrophic bacteria can also consume NH₄⁺ (Kirchman, 1994), including in the dark, as they Deleted: since 1140 derive energy from organic carbon oxidation rather than light. At an ecosystem level, therefore,

1 60 NH₄* <u>assimilation</u> may not be primarily limited by light, although this parameter clearly strongly controls the rate <u>and distribution</u> of NPP (Fig. 5a).

1162 Previous observations suggest that temperature can influence NH₄ uptake, especially in winter 1163 (Glibert, 1982; Reay et al., 2001). The negative effect of temperature appears to be enhanced 1164 under high-nutrient and low-light conditions, at least in the case of phytoplankton growth rates 1165 (Baird et al., 2001). Experiments conducted coincident with our sampling showed that the 1166 maximum rate of NH4+ uptake (Vmax) achievable by the in situ community was strongly 1167 negatively correlated with temperature and latitude (Mdutyana, 2021), with the latter parameter 1168 representing the combined role of light, temperature, and possibly iron, the average concentration 1169 of which appears to increase from the SAZ to the AZ (Tagliabue et al., 2012). We conclude that 1170 these three drivers, along with NH₄+ availability north of the SAF, may all play a role in 1171 controlling photoautotrophic NH4+ assimilation in the winter Southern Ocean, with complex 1172 interactions among them that are difficult to disentangle.

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In addition to physical and chemical limitations, microbial preference for other N species may impact NH_4^+ depletion. For example, the preferential uptake of urea and/or other dissolved organic N (DON) species by some organisms (e.g., picoeukaryotes, cyano- or heterotrophic bacteria) could cause a net decrease in the total NH_4^+ uptake rates. While urea has been shown to constitute a large fraction of the total N assimilated by Southern Ocean phytoplankton in summer and autumn (albeit mainly in the SAZ; Joubert et al., 2011; Thomalla et al., 2011), we measured fairly low ρ Urea (Fig. 5b), which is perhaps unsurprising given the low ambient urea concentrations (Table 1). The exceptions were stations 37°S and 43.0°S where ρ Urea was higher than ρ NH₄+, coincident with very low ambient NH₄+ (0.10 μ M and below detection) and relatively high urea concentrations (0.36 μ M and 0.15 μ M).

Community composition can also alter the N uptake regime. Small phytoplankton, such as the numerically-dominant nano- and picoeukaryotes, are more likely to consume NH₄+ and urea than NO₃ (Koike et al., 1986; Lee et al., 2012; 2013), especially under conditions of severe iron and light <u>Jimitation</u> (Sunda & Huntsman, 1997). Across our transect, <u>reduced N (i.e., NH₄+ transparent</u> urea) uptake exceeded NO3 uptake for both the total phytoplankton community (transect average of 12.0 ± 0.9 nM day⁻¹ for reduced N versus 5.8 ± 1.0 nM day⁻¹ for NO₃⁻; f-ratio of 0.36) and the <u>pico</u> size fraction ($5.0 \pm 1.2 \text{ nM day}^{-1}$ versus $1.9 \pm 1.2 \text{ nM day}^{-1}$; f-ratio of 0.27; Fig. 5b). That said, the NO₃⁻ uptake rates were not negligible, including in the <u>pico</u> size fraction. In the PFZ and AZ, NO₃ uptake by the picoplankton was far more strongly correlated with the abundance of picoeukaryotes than Synechococcus (r = 0.75 and 0.03, respectively), consistent with observations of dominant reliance on NO₃- by picoeukaryotes and NH₄+ by Synechococcus in other ocean regions (Fawcett et al., 2011; 2014; Painter et al., 2014). Additionally, Synechococcus abundance was strongly correlated with NH₄⁺ concentration south of the SAF (r = 0.65). In the nano+ size class, NO₃- uptake was likely driven in the SAZ by dinoflagellates and nanoeukaryotes, and in the PFZ and AZ by diatoms, which remain active in these zones in winter (Weir et al., 2020). By contrast, nanoeukaryotes, which have a higher per-cell nutrient requirement than the equally-abundant picoeukaryotes, may have dominated NH₄+ uptake in the PFZ and AZ given that higher nanoeukaryote abundances corresponded with lower NH₄+ concentrations at a number of stations (e.g., stations 50.0°S, 51.1°S, and 55.5°S; Fig. 6b).

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Deleted: Casey et al., 2009; Fawcett et al., 2011, 2014; Treibergs et al., 2014; Painter et al., 2014). Nonetheless, *Synechococcus* can consume all N forms (Capone et al., 2008 and references therein) and has evolved strategies to conserve iron by using other trace metals in some enzymes (Palenik et al., 2003). Thus, *Synechococcus* may be adapted to consume NO₃⁻ in the Southern Ocean when reduced N concentrations are near depletion (e.g., north of the SAF in winter), but are likely to consume NH₄⁺ as long as it is available, as implied by their strong correlation

1244 The low abundances of diatoms and dinoflagellates and absence of coccolithophores across our 1245 transect (Fig. 6a) is expected given the limitations imposed on nutrient uptake and CO₂ fixation

1246 by winter Southern Ocean conditions. The lower surface area-to-volume ratio of Jarge cells 1247 means that they rapidly experience diffusion-limitation of NH₄⁺ and micronutrient uptake and are

1248 more susceptible to light limitation (Finkel et al., 2004), resulting in their being outcompeted by 1249 smaller species for essential resources (Franck et al., 2005; Cavender-Bares et al., 1999). The 1250 near-absence of centric diatoms is also best explained thus, particularly given their low surface

1251 area-to-volume ratio compared to the more-abundant pennate species (Kobayashi & Takahashi, 1252

2002) that are more likely to consume NH₄+ (Semeneh et al., 1998). Diatom success in winter

1253 may also be limited by enhanced mixing, as this group generally prefers stratified waters

1254 (Kopczynska et al., 2007).

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1255 In sum, NH4+ uptake rates were low across our transect but not negligible, indicating that 1256 phytoplankton activity in winter, which is dominated by smaller species, is a sink for NH₄⁺. The 1257 hostile conditions of the winter Southern Ocean imposed limitations on NH4+ uptake that varied 1258 with latitude, with NH₄+ concentrations controlling ρNH₄+ north of the SAF, while light and 1259 temperature were important south of the SAF, Additionally, Synechococcus, nanoeukaryotes, and 1260 pennate diatoms likely dominated NH₄⁺ assimilation, consistent with previous observations from

the Southern Ocean and elsewhere (Klawonn et al., 2019; Semeneh et al., 1998).

Ammonium oxidation - Nitrification removes more mixed-layer NH4+ in winter than phytoplankton assimilation south of the PF, with NH₄+ oxidation rates that were two- to fivetimes the co-occurring NH₄+ uptake rates (Fig. 5c). The comparative success of ammonia oxidisers may be due to decreased competition with phytoplankton for NH4+ augmented by decreased photoinhibition (Wan et al., 2018; Lu et al., 2020), elevated NH₄⁺ availability (Baer et al., 2014; Mdutyana et al., 2020; Mdutyana, 2021), and the minimal effect of temperature on NH₄⁺ oxidation (Bianchi et al., 1997; Baer et al., 2014; Horak et al., 2013; Mdutyana 2021). One implication of the dominance of NH₄⁺ oxidation in winter is that in addition to the limitations on photoautotrophic NH₄⁺ assimilation discussed above, low phytoplankton success in the AZ may result from nitrifiers outcompeting phytoplankton for scarce resources (e.g., trace elements required for enzyme functioning, such as iron and copper; Amin et al., 2013; Maldonado et al.,

2006; Shafiee et al., 2019) under conditions of low incident light and enhanced mixing.

The K_m derived for NH₄⁺ oxidation in the winter Southern Ocean has recently been reported to be low (0.03 to 0.14 μ M), with ammonia oxidizers observed to become saturated at ambient NH₄⁺ concentrations of ~0.1-0.2 µM (Mdutyana, 2021). This means that south of the SAF in winter 2017, ammonia oxidizers were not substrate limited (as implied by the lack of correlation between NH₄+ox and NH₄+ concentration; <u>Table S1</u>), which raises the question of why NH₄+ oxidation did not occur at higher rates. The answer may indirectly involve temperature, in that psychrophilic organisms can be less responsive to high substrate concentrations at low temperatures (Baer et al., 2014). Another possibility is that NH₄⁺ oxidation was iron-limited (Shiozaki et al., 2016; Shafiee et al., 2019; Mdutyana, 2021). In any case, ammonia oxidisers were moderately successful across the surface Southern Ocean in winter, with low light, reduced competition with phytoplankton, and substrate repletion likely explaining the elevated NH4+ oxidation rates south of the PF compared to the stations to the north.

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Deleted:) and were more abundant. That said, we did not observe a clear relationship between pennate diatom abundance and NH₄+ concentration, except proximate to the PF (stations 47.9°S, 48.9°S, and 50.0°S) where higher pennate abundance was associated with lower NH4+

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Although NH4+ oxidisers appear to be truly psychrophilic given the southward increase in NH₄+ oxidation rates, the effect of temperature is difficult to disentangle in an environment with multiple overlapping drivers. While several studies have reported a minimal effect of temperature on NH₄+ oxidation rates

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Deleted: 2013; Mdutyana et al., in review), nitrifiers in the winter Southern Ocean may yet be living at suboptimal temperatures (Jones et al., 1988). Indeed, a relative inefficiency of NH₄+ oxidation at low temperatures could be inferred from the general southward increase in the ratio of NH₄+ to NO₂- concentration (NH₄+:NO₂-; Fig.

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NH₄+ production must have been sustained during the winter to maintain a mixed-layer NH₄+

pool south of the SAF that was high in concentration relative to the early summer. Indeed, the residence time estimated for NH₄ in winter (10 to 38 days) is considerably shorter than the

1352 1353 transition from late summer to winter (approximately three months), indicating that heterotrophic

1354 NH₄⁺ production, which would have occurred coincident with NH₄⁺ consumption, must, have

1355 been ongoing in winter. We estimate the rate of this wintertime NH_4^+ production to be 23.4 ± 6.6

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Heterotrophic activity by bacteria - Heterotrophic bacteria contribute significantly to NH4+ production in the Southern Ocean (Hewes et al., 1985; Koike et al., 1986; Tréguer & Jacques, 1992), including in winter (Rembauville et al., 2017). In our dataset, lower ratios of photosynthetic-to-heterotrophic cells were observed at stations with higher NH₄+ concentrations (e.g., stations 48.9°S, 53.0°S, 54.0°S, and 57.8°S; Fig.7a), consistent with a role for the heterotrophic bacteria present at the time of sampling in generating the ambient NH₄+ pool. The potential for ongoing heterotrophic activity can also be inferred from the high detrital particle counts along the transect (Fig. 7b). However, since heterotrophic bacteria are likely more active in late summer and autumn when the temperature and the supply of labile PON are higher (Becquevort et al., 2000; Dennett et al., 2001; Pomeroy & Wiebe, 2001; Smart et al., 2020), we expect that the winter NH₄+ pool includes NH₄+ produced in late summer and autumn. A further consideration is assimilation of NH₄⁺ by heterotrophic bacteria, reported to occur at elevated rates in the Southern Ocean mixed layer in winter (Mdutyana et al. 2020; Text S3). If this process is a persistent feature of the winter Southern Ocean, it will decrease the net contribution of heterotrophic bacteria to NH4+ accumulation. We conclude that it is unlikely that the surface NH4+ pool measured in winter derived solely from wintertime bacterial NH₄+ production given that yet

Heterotrophic activity by zooplankton - While the microzooplankton enumerated in this study occurred at very low abundances, those that were present likely contributed to the NH₄⁺ flux. For example, at stations 48.9°S and 54.0°S in the PFZ and AZ respectively, both the ratios of photosynthetic-to-heterotrophic cells and the absolute abundances of heterotrophic bacteria were low, while the microzooplankton abundances and NH4+ concentrations were elevated compared to nearby stations. The implication of these observations is that elevated microzooplankton abundances may help to explain high NH4+ concentrations in waters with low numbers of heterotrophic bacteria, although we note that this scenario only occurred at two stations. On balance, we posit that microzooplankton are Jess important for wintertime NH₄+ production than heterotrophic bacteria given their low abundances in the surface layer (Fig. 6a; Atkinson et al.,

higher NH₄⁺ concentrations have been observed in late summer_and autumn (Becquevort et al.,

2000; Dennett et al., 2001), including in the present study (see section 5.2 below).

1386 Above, we have assumed that NH₄⁺ production is the direct result of heterotrophy. However, 1387 there are other possible mechanisms of NH₄+ supply that should be considered. We briefly 1388 address some of these processes below, noting that for most, there are very few to no observations

1389 available from the Southern Ocean. Deleted: inputs

Deleted: NH4+ production, although

Deleted: Although not measured directly in this study,

Deleted: be...ave been sustained during the winter to retain 81

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Deleted: is...as high in concentration relative to the early summer. With low or no NH4+ production in the autumn and

Deleted: pool south of the SAF would be depleted ...n winter (10 to 38 days (median of 21 days) given... is considerably shorter than the consumption rate (ρNH4++...ransition from late summer to winter (approximately three months), indicating that heterotrophic NH₄+_{ox}) and NH₄+ concentration measured at each station (Text S2). Heterotrophic NH4+...production, which would have occurred coincident with NH₄⁺ consumption, must, therefore, be [30]

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1628 DON cycling - NH₄+ can be released by heterotrophic bacteria that directly consume DON (e.g.,

1629 urea: Billen, 1983; Tupas & Koike, 1990), and possibly also by ammonia oxidisers that convert

1630 DON to NH_4^+ intracellularly, through the equilibration <u>of the</u> intra- and extracellular NH_4^+ pools

1631 (Kitzinger et al., 2019). DON can also be converted to NH₄⁺ through photodegradation by UV

1632 radiation (e.g., Aarnos et al., 2012). Bacterial decomposition of DON (rather than PON) to NH₄⁺

1633 is implicit in most estimates of ammonification, however, and cellular NH4+ efflux by ammonia

1634 oxidisers is likely extremely low given that they require NH₄⁺ to fix CO₂. Additionally, the low

1635 light <u>flux to</u> the <u>surface</u> Southern Ocean in <u>winter means</u> that photodegradation <u>will not</u> yield a

1636

significant supply of NH₄⁺. Thus, DON conversion to NH₄⁺, through any mechanism, is probably

1637 negligible.

1638 External inputs of ammonium - High surface ocean NH₄⁺ concentrations may theoretically derive

1639 from external inputs of NH₄+, such as N₂ fixation, NH₄+ aerosol deposition, or sea-ice melt. N₂

1640 fixation should be below detection in the winter Southern Ocean due to the cold temperatures,

1641 low light and iron conditions, and high NO₃⁻ concentrations (Jiang et al., 2018; Knapp et al.,

1642 2012; Kustka et al., 2003). NH₄+ aerosols are unlikely to be abundant over regions of the Southern

1643

Ocean remote from islands and coastal Antarctica, particularly in winter when NH4+ aerosol

1644 concentrations have been shown to reach a minimum (Legrand et al., 1998; Xu et al., 2019).

1645 Moreover, the aerosols that are present over the open Southern Ocean will derive mainly from 1646

surface_ocean NH3 efflux; once re-deposited, this NH4+ does not constitute a new input to surface

1647 waters (Altieri et al., 2021), Finally, since our sampling took place before the sea-ice reached its

1648 northernmost extent (Cavalieri & Parkinson, 2008), the dominant process would have been sea-

1649 ice formation rather than sea-ice melt, the latter an occasional source of NH₄+ (Kattner et al.,

1650 2004; Zhou et al., 2014). In any case, we observed elevated NH₄+ concentrations as far north as

1651 46°S, <u>~1700</u> km beyond the <u>influence</u> of sea-ice melt.

1652 5.2 Seasonal cycling of NH₄+ in the Southern Ocean mixed layer south of the SAF

1653 The NH₄⁺ concentration data collected over the 2018/19 annual cycle provide context for 1654 interpreting our winter 2017 dataset, allowing us to address our hypothesis that NH₄⁺ production

1655 in late summer and autumn contributes to the elevated NH₄⁺ concentrations measured in winter.

1656 The very low NH₄⁺ concentrations observed in early summer (Fig. 8a) are consistent with high 1657 rates of phytoplankton NH₄⁺ assimilation during the spring and early-summer growing period

1658 (Mdutyana et al., 2020; Savoye et al., 2004; Daly et al., 2001). By late summer, the NH₄⁺

1659 concentrations, increased (Fig. 8b) due to elevated heterotrophic activity (i.e., bacterial

1660 decomposition and zooplankton grazing) following the accumulation of algal biomass

1661 (Mengesha et al., 1998; Le Moigne et al., 2013), coupled with iron- and/or silicate-limitation of

1662 phytoplankton (Hiscock et al., 2003; Sosik & Olson, 2002) and enhanced grazing pressure

1663 (Becquevort et al., 2000). Mixed-layer NH₄+ remained high between late summer and winter,

1664 (Fig. 8b-c), likely due to sustained heterotrophic NH₄⁺ production in excess of NH₄⁺ removal.

1665 <u>This notion</u> is supported by <u>estimates</u> of the residence time of NH_{θ^+} . We calculate that <u>in summer</u>,

1666 the in situ NH₄+ pool would be depleted in 2 to 27 days (median of 5 days) without coincident

1667 NH_4^+ production. In addition, the net decline in NH_4^+ concentration of $0.31 \pm 0.97 \mu M$ between

1668 late summer and winter requires an average NH₄⁺ production rate of 52.8 ± 25.0 nM/day given

1669 the observed NH₄⁺ assimilation rates. This estimate is remarkably similar to the only Deleted: produced through the release of NH₄+ by ...eleased by heterotrophic bacteria that directly consume DON (e.g. urea) (... Billen, 1983; Tupas & Koike, 1990), and possibly released...lso by ammonia oxidisers,...that convert DON to NH4+ intracellularly, through the equilibration between...f the intra- and extracellular NH₄₊ pools (Kitzinger et al., 2019). DON can also be converted to NH₄+ through photodegradation by UV radiation (e.g., Aarnos et al., 2012).

However, bacterial ...acterial decomposition of DON (rather than PON) to NH₄+ is implicit in most estimates qualitative...of ammonification, however, and quantitative, of

heterotrophic bacterial remineralization....ellular NH4+ efflux by ammonia oxidisers is likely extremely low given that they require NH₄⁺ to fix CO₂. Additionally, the low light levels of...lux to the wintertime...urface Southern Ocean mean...n winter means that photodegradation is unlikely to ...ill not yield a significant supply of NH4+ flux. We thus conclude that

Deleted: from nitrogen...2 fixation, NH₄+ aerosol deposition, or sea-ice melt. Nitrogen

Deleted: negligible

Deleted: extremely ...old temperatures, low light and iron availability...onditions, and high NO₃⁻ concentrations (Jiang al., 2018; Knapp et al., 2012; Kustka et al., 2003). Similar [46]

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Moved up [6]: 1998; Xu et al., 2019).

Deleted: NH4+ deposition to the surface Southern Ocean is thus likely minimal....Finally, since our sampling took place before the sea-ice reached its northernmost extent (Cavalieri & Parkinson, 2008), the dominant process would have been sea-ice formation rather than sea-ice melt, the latter a...n occasional source of NH4+ at times ...Kattner et al., 2004; Zhou et al., 2014), although probably not during our study.481

Deleted: To contextualize our wintertime observations, we need to explore the seasonality of the NH4+ pool in the surface Southern Ocean, especially given ...he NH4+ concentration497

Deleted: every two hours between Cape Town and Antarctica (early- and late summer) or the MIZ (winter and spring), and analysed as described in section 3.2.1 for winter 2017.

Deleted: significantly as the growing season progressed, reaching an average concentration of $0.81 \pm 0.92~\mu M$ by late summer ...Fig. 9b). This NH4+ increase can be explained [51]

Deleted: Using the NH₄+ concentrations and ρNH₄+ measured in late summer 2019 (Deary, 2020), w

Deleted: Indeed, given the average ρNH₄+ south of the SAF in late summer ($50.6 \pm 24.0 \text{ nM/day}$),

Deleted: We further calculate that

Deleted: $33...1 \pm 0.97 \mu M$ between late summer and winter (a roughly four-month period) ... equires an average NH4+ production rate of 52.9... \pm 25.0 nM/day. ...

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Deleted: comparable to NH₄+ remineralisation rates measured in the AZ near the Antarctic Peninsula in summer (

1875 measurements of NH₄⁺ regeneration available for the Southern Ocean, measured near the 1876 Antarctic Peninsula in summer (average of 55 nM day⁻¹; Goeyens et al., 1991).

By, early spring, the NH₄⁺ concentrations had declined (Fig. 8d), implicating increased photosynthetic activity, and thus nutrient assimilation, following the alleviation of light-limitation. We suggest that any NH₄⁺ remaining in late winter, would have been consumed in early spring prior to significant NO₃⁻ drawdown because far less energy (i.e., light) is required for its assimilation (Dortch, 1990). The high NH₄⁺ concentrations subsequently observed in late spring (mainly in the PFZ; Fig. 8e) can be explained by elevated heterotrophic activity in response to high levels of regional phytoplankton growth driven by frontal upwelling of limiting nutrients (Becquevort et al., 2000; Mayzaud et al., 2002).

From our six transects of surface NH₄+ concentrations across the Southern Ocean, we propose a seasonal cycle for mixed-layer NH₄+ south of the SAF (Fig. 8f). Our proposal is consistent with previous characterizations of the early summer-to-autumn evolution of Southern Ocean NH₄+ concentrations (i.e., from below detection due to phytoplankton assimilation to elevated due to net heterotrophy). However, it contradicts the hypothesis that NH₄+ will subsequently decline due to persistent but low rates of photosynthesis that yield insufficient biomass to support elevated heterotrophy in autumn, thus driving a coincident decrease in photosynthetic and heterotrophic activity (Koike et al., 1986; Serebrennikova & Fanning, 2004). Instead, our data evince a gradual decline in mixed-layer NH₄⁺ concentrations from late summer through winter. This decline can be explained by heterotrophic NH₄⁺ production outpacing NH₄⁺ removal in late summer/autumn, with NH₄⁺ regeneration then decreasing during winter to lower rates than the combined rate of NH₄+ assimilation and exidation. By late spring, NH₄+ reaches concentrations similar to those observed in early summer as the improved growing conditions (i.e., elevated light and iron availability; Ellwood et al., 2008; Mtshali et al., 2019) allow phytoplankton to rapidly consume any NH₄⁺ remaining at the end of winter and subsequently produced in spring. An exception to this scenario is elevated, localized NH₄+ production near fronts, such as we observed in late spring 2019, which likely resulted from biological activity supported by frontal upwelling of silicateand iron-bearing Upper Circumpolar Deep Water (Prézelin et al., 2000).

6. Summary and implications

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Our study of the upper Southern Ocean, focused on the infrequently-sampled winter season, provides new insights into the internal cycling of N in the mixed layer of a globally-important region. We attribute the elevated NH₄⁺ concentrations that persist in the winter mixed layer south of the SAF to sustained heterotrophic NH₄⁺ production in excess of NH₄⁺ removal, driven by temperature-, light-, and possibly iron-limitation of phytoplankton and nitrifiers. We further suggest that heterotrophic bacteria are the main NH₄⁺ producers in winter and that the contribution of external sources to the Southern Ocean's mixed-layer NH₄⁺ pool is negligible. From observations of surface NH₄⁺ concentrations made between December 2018 and November 2019, we deduce that the elevated mixed-layer NH₄⁺ concentrations measured in winter cannot be due solely to wintertime NH₄⁺ production. Instead, we propose that NH₄⁺ accumulates to its highest concentrations in late summer following the peak phytoplankton growing season, after which sustained heterotrophy throughout the autumn and winter prevents this NH₄⁺ from being fully depleted until the early spring, even though the rate of NH₄⁺ removal must exceed that of

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Deleted: the...early spring, the NH₄* concentrations south of the SAF ...ad declined to near or below the methodological detection limit $(0.09 \pm 0.08 \,\mu\text{M}; \text{Fig. 9d})$ [53]

Deleted: -...and thus nutrient assimilation, -

Deleted: that results in the consumption of nutrients introduced into surface waters in winter. We postulate that the residual NH₄⁺

Deleted: /early spring

Deleted: NH₄* concentrations south of the SAF rose again by the late spring to an average value only slightly lower than that measured in winter (0.37 ± 0.69 μM; Fig. 9e). However, late-spring NH₄* concentrations were only elevated...he high NH₄* concentrations subsequently observed in late spring (mainly in the PFZ (range of 0.11 ± 0.01 to 4.39 ± 0.03 μM, average of 0.71 ± 1.04 μM), as has been observed previously (Bathmann et al., 1997), which we attribute to increased heterotrophic activity ... Fig. 8e) can be explained by elevated heterotrophic activity in response to elevated ...igh levels of regional springtime ...hytoplankton growth driven by frontal upwelling of limiting nutrients (Becquevort et al., 2000; Mayzaud et al., 2002). Excluding the PFZ data yields a far lower late-spring average NH₄* concentration of 0.18 ± 0.14 μM, which we take as broadly representative of this season...

Deleted: Using...rom our high-resolution NH₄⁺ concentration measurements...ix transects of surface NH₄⁺ concentrations across the Southern Ocean, we propose a seasonal cycle for mixed-layer NH₄⁺ south of the SAF (Fig. 9f...f). Our proposal is consistent with previous characterizations of the early summer-to-autumn evolution of Southern Ocean NH₄⁺ concentrations (i.e., from below detection due to phytoplankton uptake...ssimilation to elevated due to net heterotrophic activity), but...eterotrophy). However, it contradicts the hypothesis that NH₄⁺ will subsequently decline due to persistent but low rates of photosynthesis that yield insufficient biomass to support late-summer...levated heterotrophy in autumn, thus resulting in...riving a coincident decrease in photosynthetic and heterotrophic activity (Koikso)

Deleted: 5.3 Implications

Potential for ammonium inhibition of nitrate uptake – The low rates of NO₃ uptake characteristic of winter Southern Ocean surface waters have been attributed to light, 157

Moved down [7]: Goeyens et al., 1995; Philibert et al., 2015; Reay et al.,

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Deleted: 2001), as has been observed in other regions (Dortch, 1990; Flynn et al., 2018). Previous Southern Ocean studies have identified an inhibitory effect of NH₄* on NO₃* uptake at NH₄* concentrations >1 μM (and occasionally... [58]

Moved down [8]: 2015). Surface ocean NH₄⁺ concentrations play a central role in determining the sign and magnitude of the air-sea NH₃ flux, along with wind speed, surface ocean temperature, and pH.

Deleted: Therefore, the biogeochemical pathways that drive seasonality in surface ocean NH₄⁺ concentrations are an important control on the remote Southern Ocean air-sea NH₄)

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Deleted:, conducted in...of the upper Southern Ocean during... focused on the infrequently-sampled winter season, provides new insights into the internal cycling of N in the mixed layer of a globally-important region. We used ____[60]

2311 NH₄⁺ production over this period. Measurements of heterotrophic NH₄⁺ production rates are 2312 required to confirm the hypothesized seasonal cycle of NH₄⁺ in the Southern Ocean mixed layer, 2313 and higher spatial resolution sampling of plankton community composition and N removal rates 2314 may help to explain local variability in NH₄⁺ concentrations, particularly near the fronts, Formatted: English (US) 2315 The persistence of elevated NH₄⁺ concentrations across the polar Southern Ocean between late 2316 summer and winter implies that the mixed layer is a biological source of CO2 to the atmosphere 2317 for half the year, not only because NO₃ drawdown is weak at this time (e.g., Gibson & Trull, 2318 1999; Gray et al., 2018; Hauck et al., 2015; Mongwe et al., 2018; Shadwick et al., 2015), but also 2319 because the ambient conditions allow for NH₄⁺ accumulation. There are additional implications 2320 of our observations. For example, NH₄+ concentrations >1 μM (and at times >0.5 μM) have been 2321 reported to inhibit NO₃⁻ assimilation, including in the Southern Ocean (Cochlan, 1986; Goevens Deleted: uptake 2322 et al., 1995; Philibert et al., 2015; Reay et al., 2001). Inhibition of NO3⁻ assimilation due to the Moved (insertion) [7] 2323 seasonal accumulation of NH₄⁺ would constitute an inefficiency in the biological pump. Formatted: Font colour: Auto, English (UK) 2324 However, we observed little evidence of this effect in winter 2017 - the southward decrease in Deleted: Figure Captions 2325 ρNO_3^- was not stronger than that of ρNH_4^+ despite the latitudinal increase in NH_4^+ concentration, Deleted: amount to 2326 Deleted: and we observed no relationship between NH₄+ concentration and the proportion of NO₃- to 2327 Deleted: for example, NO₃-+NH₄+ uptake (i.e., the f-ratio; Table S1). Deleted: X 2328 The implications of NH₄+ cycling extend beyond the upper ocean to the atmosphere, since Formatted: Highlight 2329 ammonium aerosols that influence Earth's albedo (Tevlin & Murphy, 2019) are formed in the 2330 marine boundary layer from reactions of NH₃ gas with acidic species. In the remote Southern 2331 Deleted: the Ocean, marine NH₃ emissions, which are the largest natural contributors to NH₃ globally, are 2332 likely the dominant local source of NH3 to the atmosphere (Paulot et al., 2015). Surface ocean Deleted: is 2333 NH₄⁺ concentrations play a central role in determining the sign and magnitude of the air-sea NH₃ Deleted: to be 2334 flux, along with wind speed, surface ocean temperature, and pH. Therefore, the biogeochemical Moved (insertion) [8] 2335 pathways that underpin seasonal changes in surface ocean NH₄⁺ concentrations represent an Deleted: are

important control on the remote Southern Ocean air-sea NH3 flux, with consequences for aerosol

composition, cloud formation, and climate (Altieri et al., 2021).

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Acknowledgements

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2350 We are grateful to Captain Knowledge Bengu and the crew of the R/V SA Agulhas II, and Chief 2351 Scientists Hermann Luyt, Marcello Vichi, and Thomas Ryan-Keogh. We thank Tahlia Henry for 2352 CTD operations and CTD and SDS data processing. We are grateful to the students from the 2353 Cape Peninsula University of Technology for help with sample collection and analysis of chl-a, 2354 and thank Raquel Flynn, Mishka Rawatlal, and Raymond Roman for assistance with nutrient 2355 analyses. We acknowledge the Flow Cytometry Core Facility at the University of Cape Town 2356 (UCT) and the efforts of Ian Newton at the Stable Light Isotope Laboratory (UCT). This work 2357 was supported by the South African Departments of Forestry, Fisheries, and Environment 2358 (formerly Environmental Affairs) and Science and Innovation (DSI), and the National Research 2359 Foundation (NRF) through the South African National Antarctic Program (SANAP; 110732 to 2360 K.E.A and 105539, 110735, and 129232 to S.E.F.), Equipment-related Travel and Training Grant 2361 (118615 to K.E.A.), Competitive Support for Rated Researchers Grant (111716 to K.E.A.), and 2362 Incentive Fund (115335 to S.E.F.). S.S., M.M., K.A.M.S., and J.M.B. acknowledge funding from 2363 the NRF through postgraduate scholarships (120105, 112380, 113193, and 108757). S.S. was 2364 partially supported by a UCT Vice-Chancellor Research Scholarship and M.M. by the UCT 2365 Harry Crossley Foundation Research Fellowship. S.E.F. and K.E.A. acknowledge the support of 2366 the UCT Vice-Chancellor Future Leaders 2030 programme. S.E.F. acknowledges an African 2367 Academy of Sciences/Royal Society FLAIR fellowship and K.E.A. acknowledges support from 2368 UCT through a University Research Council Launching Grant and a University Equipment 2369 Committee Grant. We further acknowledge the support of the DSI Biogeochemistry Research 2370 Infrastructure Platform (BIOGRIP).

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Moved down [9]: Figure 1: Winter 2017 cruise track overlaid on sea surface temperature (SST) measured by the hull-mounted thermosalinograph. The underway (Leg S) and CTD (Leg N) stations are indicated by white circles. Stations at which net primary production (NPP), nitrogen uptake, and ammonium oxidation experiments were conducted are denoted by red squares. The pink triangles indicate stations where only NPP experiments were conducted while the green circles show stations where only ammonium oxidation was

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Figure and Table Captions

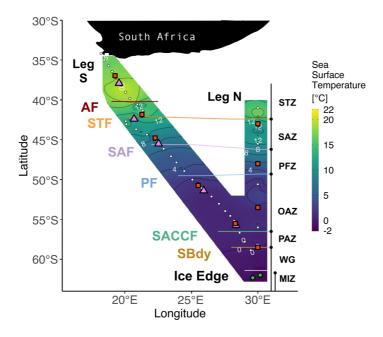


Figure 1: Winter 2017 cruise track overlaid on sea surface temperature (SST) measured by the hull-mounted thermosalinograph. The underway (Leg S) and CTD (Leg N) stations are indicated by white circles. Stations at which net primary production (NPP), nitrogen uptake, and ammonium oxidation experiments were conducted are denoted by red squares. The pink triangles indicate stations where only NPP experiments were conducted while the green circles show stations where only ammonium oxidation was measured. Solid lines indicate the positions of the fronts, identified from measurements of temperature and salinity, Abbreviations for fronts: AF – Agulhas Front (~40.2°S); STF – Subtropical Front (~42.1°S); SAF – Subantarctic Front (~45.6°S); PF – Polar Front (~49.5°S); SACCF – Southern Antarctic Circumpolar Current Front (~56.5°S); SBDY – Southern Boundary (~58.5°S). Abbreviations for zones: STZ – Subtropical Zone; SAZ – Subantarctic Zone; PFZ – Polar Frontal Zone; OAZ – Open Antarctic Zone; PAZ – Polar Antarctic Zone; WG – Weddell Gyre; MIZ – Marginal Ice Zone. Together, the OAZ and PAZ constitute the Antarctic Zone (AZ). See Text S1 for detailed definitions of the fronts and zones. Figure produced using the package ggplot2 (Wickham, 2016).

Table 1: Mean (\pm 1 SD) of surface ocean POC, PON, chl-a, and nutrient concentrations, cell abundances, and nutrient uptake rates measured in each zone of the Southern Ocean in winter 2017. Where no SD is given, only one sample was measured. The >0.3 μm and >2.7 μm size fractions are referred to as "bulk" and "nano+", respectively, "% of nano+" refers to the average relative contribution of the nano+, size fraction to total chl-a, POC, or PON, calculated for each station within a zone. The f-ratio including pUrea

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is only shown for zones where ρ Urea was measured at all stations. "ND" indicates no data available. Abbreviations as in Figure 1.

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	STZ	SAZ	PFZ	OAZ	PAZ
NH ₄ ⁺ (μM)	0.08±0.03	0.06±0.01	0.42±0.01	0.52±0.01	0.58±0.01
PO ₄ ³⁻ (μM)	0.44±0.07	0.90±0.06	1.59±0.1	2.00±0.13	1.99±0.09
NO ₃ - (μM)	3.6±0.2	10.5±0.5	21.5±0.2	26.7±0.4	27.5±0.4
Si(OH) ₄ (µM)	2.6±0.1	2.5±1.8	6.6±0.1	40.3±0.5	45.0±0.8
NO ₂ - (μM)	0.15±0.02	0.13±0.02	0.17±0.02	0.19±0.01	0.21±0.02
Urea (µM)	0.23±0.04	0.11±0.04	0.26±0.08	0.24	0.21±0.03
chl-a (bulk) (µg L ⁻¹)	0.65±0.08	0.43±0.05	0.35±0.03	0.25±0.02	0.21±0.00
chl-a (nano+) (µg L ⁻¹)	0.50±0.05	0.30±0.04	0.24±0.02	0.18±0.02	0.17±0.02
chl-a (pico) (µg L ⁻¹)	0.15±0.1	0.13±0.07	0.11±0.04	0.06±0.03	0.04±0.02
chl-a (% of nano+)	77.5±13.9	73.1±10.9	69.8±8.7	76.7±11.3	80.1±8.5
POC (bulk) (μM)	4.4±6.7	3.4±0.4	3.2±0.3	3.4±0.5	3.5+0.2
POC (nano+) (µM)	2.6±0.5	2.6±0.4	1.9±1.2	1.9±0.4	4.6
PON (bulk) (μM)	0.6±0.2	0.5±0.1	0.4±0.1	0.5±0.1	0.5±0.1
PON (nano+) (μM)	0.3±0.1	0.3±0.1	0.2±0.3	0.2±0.1	0.4±0.0
POC (% of nano+)	79.7±24.6	79.6±19.0	50.9±33.2	77.2±21.8	ND
PON (% of nano+)	69.0±31.9	67.1±17.2	53.8±24.1	67.0±21.9	51.1±24.7
POC:chl-a (g g ⁻¹)	103.0±22.1	102.5±14.4	122.5±11	234.1±29.2	219.3±1.0
POC:PON (M/M)	7.81±6.49	6.90±1.25	7.13±0.71	6.72±1.62	5.80±3.75
δ^{15} N-PON	1.4±0.9	1.2±1.0	0.3±0.5	-1.3±0.5	-1.3±0.4
NPP (bulk) (nM day ⁻¹)	497.1±42.4	277.5±21.3	289.7±19.2	85.3±26.1	27.7±0.2
NPP (nano+) (nM day ⁻¹)	384.7±29.7	178.2±23.4	193.5	49.6±5.0	ND
ρNH ₄ ⁺ (bulk) (nM day ⁻¹)	5.7±0.8	8.9±1.1	12.9±0.4	4.8±0.1	3.0±0.8
ρNH ₄ ⁺ (nano+) (nM day ⁻¹)	4.0±1.1	4.1±1.2	4.2±4.7	3.1±0.4	ND
ρNO ₃ - (bulk) (nM day-1)	4.1±0.4	11.5±1.4	5.9±1	3.6±0.4	3.7±1.8
ρNO ₃ - (nano+) (nM day-1)	3.4±0.3	6.6±0.4	4.3±0.4	2.6±0.8	2.7±1.2
ρUrea (bulk) (nM day ⁻¹)	7.5±0.6	6.9±0.3	6.5±1.0	2.1±0.3	0.6±0.01
ρUrea (nano+) (nM day ⁻¹)	4.9±0.3	3.8±0.2	4.0±0.6	1.3±0.2	0.7±0.4
f-ratio (bulk) (including ρUrea)	0.21±0.31	0.43±0.11	0.23±0.18	ND	0.51±0.53
f-ratio (bulk) (excluding ρUrea)	0.43±0.32	0.57±0.12	0.31±0.18	0.43±0.16	0.55±0.54
NH ₄ +ox (nM day-1)	9.3±0.5	12.9±0.6	11.1	17.7±0.6	14.3±1.0
Total microplankton (cells mL ⁻¹)	13±11	5±3	9±3	6±6	4±2
Centric diatoms (cells mL ⁻¹)	<1	<1	<1	<1	1±2
Pennate diatoms (cells mL ⁻¹)	2±4	<1	2±1	2±3	<1
Dinoflagellates (cells mL ⁻¹)	7±6	4±0	6±2	3±2	2±0
Micro-zooplankton (cells mL ⁻¹)	4±3	<1	2±2	1±2	<1
Nanoeukaryotes (cells mL ⁻¹)	ND	2.2±1.4 E+03	1.5±0.7 E+03	1.6±0.7 E+03	1.4E+03
Picoeukaryotes (cells mL ⁻¹)	ND	4.5±2.9 E+03	4.9±3.7 E+03	1.5±0.5 E+03	8E+02
Synechococcus (cells mL ⁻¹)	ND	3.8±1.8 E+03	2.3±1.1 E+03	1.4±0.2 E+03	1E+03

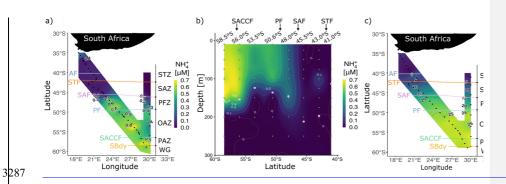


Figure 2: Concentrations of dissolved ammonium (NH₄⁺) a) at the surface for Legs S and N and b) with depth (0-300 m) for Leg N, and c) concentrations of nitrate (NO₃⁻) at the surface for Legs S and N. Pink circles in panel b show the mixed layer depth at the CTD stations. Abbreviations are as in Figure 1. Figure produced using the package ggplot2 (Wickham, 2016).

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a) 30°S 30°S South Africa South Africa 35°S 35°S Bulk Nano+ Chl-a Chl-a 40°S [µg L⁻¹] 40°S [%] 1.0 100 Patitnde 50°S: Latitude 100 80 8.0 SAZ SAZ 45°S 60 0.6 PFZ PFZ 40 0.4 50°S 0.2 20 OAZ OAZ 0.0 55°S 55°S PAZ PAZ SBdy WG WG 60°S 18°E 21°E 24°E 27°E 30°E 33°E 18°E 21°E 24°E 27°E 30°E 33°E Longitude Longitude

Figure 3: a) Bulk chlorophyll-a (chl-a) concentrations and b) the proportion of chlorophyll-a in the nano+size fraction at the surface for Legs S and N. Abbreviations are as in Figure 1. Figure produced using the package ggplot2 (Wickham, 2016),

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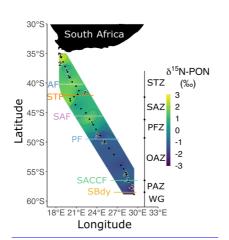


Figure 4: Bulk δ^{15} N-PON at the surface for Leg S in winter 2017. Two stations nearest South Africa at which biomass concentrations were extremely high have been excluded. Abbreviations are as in Figure 1. Figure produced using the package ggplot2 (Wickham, 2016).

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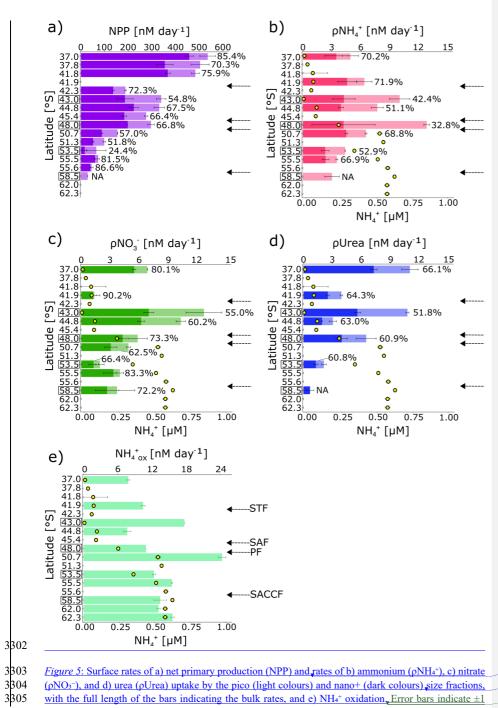
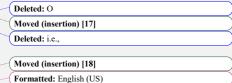


Figure 5: Surface rates of a) net primary production (NPP) and rates of b) ammonium (pNH₄+), c) nitrate (ρNO₃-), and d) urea (ρUrea) uptake by the pico (light colours) and nano+ (dark colours) size fractions, with the full length of the bars indicating the bulk rates, and e) NH_4^+ oxidation, Error bars indicate ± 1

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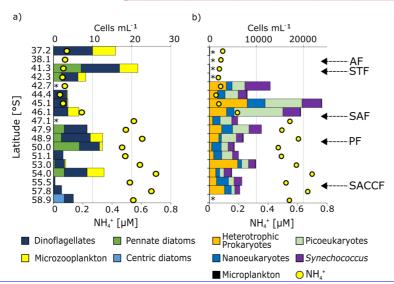


Figure 6: Surface community composition for a) plankton \geq 15 µm (enumerated by microscopy) and b) the total community <15 µm (enumerated by flow cytometry). For context, the surface NH₄+ concentration at each station is shown by the yellow circles, * indicates stations at which no measurements were made while the absence of a bar with no * indicates that no cells were detected. Note that the abundances shown on panel b (top x-axis) are >2 orders of magnitude greater than those shown in panel a. The "microplankton" shown in panel a are included on panel b (slim black bars) to illustrate the difference in abundance between the micro- and pico+nano populations. The frontal positions are indicated on panel b, with abbreviations as in Figure 1.

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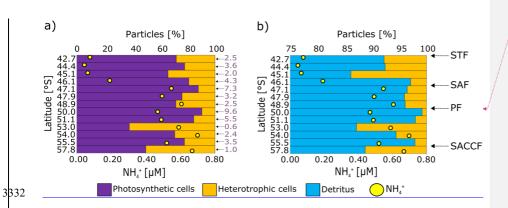


Figure 7: Relative abundances of a) total photosynthetic versus heterotrophic bacteria and b) detritus versus heterotrophic bacteria at the surface for Leg S. The surface NH₄+ concentration at each station is indicated by the yellow dots. The values in maroon text on the right side of panel a are the photosynthetic-to-heterotrophic cell ratios. The upper x-axis in panel b begins at 75% in order to highlight the (much smaller) heterotrophic bacterial contribution to the summed detrital + heterotrophic particles. Frontal abbreviations are as in Figure 1.

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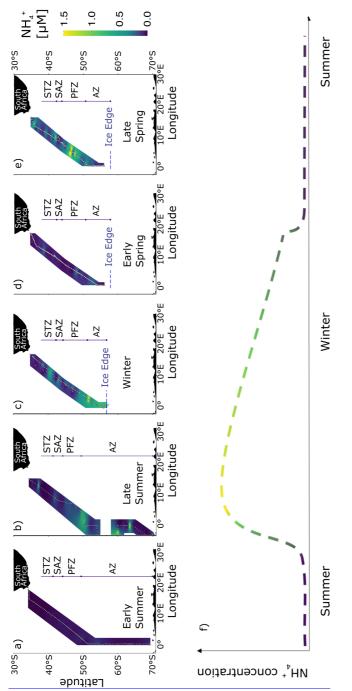
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early spring 2019, and e) late spring 2019. f) The proposed seasonal cycle of NH4* concentrations in the mixed layer south of the Subantarctic Front. The colour gradient in panel f shows the transition between late summer and late winter. Panels a and b cover a latitudinal extent of 30-70°S, while panels e-e cover 30-60°S due to the presence of sea-ice. Abbreviations are as in Figure 1, with AZ referring to the combined OAZ and PAZ. Figure produced using the package Figure 8: Surface concentrations of NH4* across the eastern Atlantic sector of the Southern Ocean measured between December 2018 and November 2019. Five unique transects (additional to the winter 2017 dataset presented in Fig. 2a) are shown: a) early summer 2018, b) late summer 2019, c) winter 2019, d) ggplot2 (Wickham, 2016).

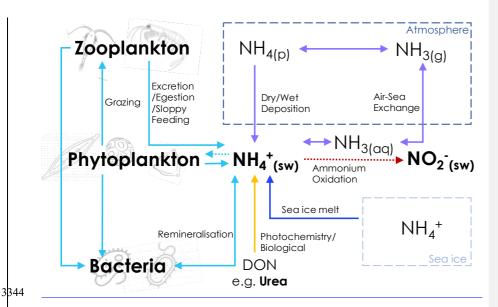


Figure 9: Schematic of the possible mixed-layer NH₄⁺ assimilation and production pathways. Bold text-indicates components of the NH₄⁺ cycle that were directly measured in this study (seawater concentrations of NH₄⁺, NO₂⁻, and urea; phytoplankton, bacterial, and microzooplankton cell abundances), and dotted lines indicate processes for which we have direct rate measurements (phytoplankton uptake of NH₄⁺; oxidation of NH₄⁺ to NO₂). Dashed-line boxes represent the atmosphere and sea-ice, with all other processes occurring in the ocean. DON – dissolved organic nitrogen; NH_{3(a0)} – aqueous (seawater) ammonia; NH_{4(p)} – ammonium aerosols (including ammonium sulphate, ammonium bisulphate, and ammonium nitrate); NH_{3(a)} – ammonia gas.

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Figure 8: Surface concentrations of NH₄* across the eastern Atlantic sector of the Southern Ocean measured between December 2018 and November 2019. Five unique transects (additional to the winter 2017 dataset presented in Fig. 2a) are shown: a) early summer 2018, b) late summer 2019, c) winter 2019, d) early spring 2019, and e) late spring 2019. f) The Proposed proposed seasonal cycle of NH₄* concentrations in the mixed layer south of the Subantarctic Front. The colour gradient in panel f indicates shows the transition period between late summer and late winter. Panels a and b cover a latitudinal extent of 30-70°S, while panels c-e cover 30-60°S due to the presence of sea-ice. Abbreviations are as in Figure 1, with AZ referring to the combined OAZ and PAZ. Figure produced using the package ggplot2 (Wickham, 2016).¶

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