Quantification of potential methane emissions associated with organic matter amendments following oxic soil inundation

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Abstract

Methane (CH_4) emissions are a potent contributor to global warming and wetlands can be a significant CH_4 source. In a microcosm study, we evaluated how the practice of amending soils with organic matter as part of wetland restoration projects may affect CH_4 production potential. Organic amendments including hay, manure, biosolids, composted yard waste, and compostwood mulch were evaluated at three different levels. Using 1-liter glass microcosms, we measured the production of biogenic gases over 60 days in two soils designated by texture:, a sandy loam (SL) and a sandy clay loam (SCL). Fresh organic amendments increased CH_4 production, leading to potentially higher global warming potential and wetland C loss, particularly in sandy soils. Organic amendments increasedand CH_4 production was more pronounced in the SL. We observed biogenic gas production in two sequential steady state phases: Phase 1 produced some CH_4 but was mostly carbon dioxide (CO_2) followed by Phase 2, two to six weeks later, with much higher total gas and nearly equal amounts of CH_4 and CO_2. If this is generally true in soils, it may be appropriate to report CH_4methane emissions in the context of inundation duration. The CH_4 from the SCL soil ranged from 0.003 – 0.8 cccm^3 /Kg^{-1} /day in Phase 1 to 0.75 – 28 cm^3 Kg^{-1} day in Phase 2, and from the SL range from 0.03 – 16 cc cm^3 Kg^{-1}

day ~~cm^3/Kg/day~~ in Phase 1 to 1.8 – 64 cm^3 Kg^{-1} day in Phase 2. ~~We had set out to identify an organic amendment that would promote iron (Fe) reduction without excess CH_4, but amendments were not needed to produce Fe and make soils hydric.~~ Adding fresh organic matter (e.g., hay) ~~resulted in both excess Fe^{2+} and CH_4~~increased ferrous iron (Fe^{2+}) ~~-~~concentrations whereas in some cases composted ~~amendments had little effect. The potential for excess methanogenesis should~~organic matter decreased both Fe^{2+} concentrations and CH_4 production. Methanogenesis normally increases following the depletion of reduce~~ai~~eable ~~iron~~Fe; however, we observed instances where this was not the case, suggesting other biogeochemical mechanisms ~~must be taken into account when considering organic matter amendments~~contribut~~ing~~ed to the shift in ~~mitigation wetlands.~~gas production.

Keywords Methane emissions, mitigation wetlands, organic amendments

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1 Introduction

The ecological benefits of wetlands are well documented, including their role as carbon (C) sinks to stabilize global climate (Mitsch et al., 2015). Driven in part by this ecological contribution, from 1970 to 2015 ~~new,~~ ~~(human-made~~~~constructed~~human-made ~~)~~ wetlands have increased 233% (Darrah et al., 2019). Between 2004 and 2009 the United States saw a net gain of 16,670 hectares of freshwater wetlands: 360,820 hectares of new wetlands to offset 344,140 hectares of existing (presumably ~~carbon~~C-sink) wetlands that were destroyed (Dahl, 2011). Although ~~human-made~~~~constructed~~created or restored wetlands may effectively sequester ~~carbon (~~C~~)~~, it may take hundreds of years to offset their radiative forcing due to methane (CH_4) emissions (Neubauer, 2014). With such a large number of new human-madeconstructedhuman-made wetlands, and their potential to increase global warming, it is vital to consider factors that may contribute to CH_4 emissions.

Some researchers have suggested that Organic amendments such as straw, wood mulch, manure, and biosolids, mixed into the soil, are thought to accelerate C storage by enhancing the conversion of plant-derived compounds to microbial residues (Richardson et al., 2016). Microbial residues, largely aliphatic-C from cell membrane lipids, can accumulate in soil and are not directly accessible by methanogens (Chen et al., 2018). Plants contribute both above and belowground organic matter (OM). MMicrobial residues, largely aliphatic-C from cell membrane lipids, can accumulate under anoxic conditions and are not directly accessible by methanogens (Chen et al., 2018). Belowground plant materials are preferentially converted to soil organic carbon (SOC) (Mazzilli et al., 2015). In saturated soils root residues of wetland plants contain suberin and cutin (Watanabe et al., 2013), which persist, reducing biogenic gas production (Mikutta et al., 2006). Before contributing to SOC, standing litter in natural wetlands is partially decomposed by fungi (Kuehn et al., 2011), and further decomposed by aerobic bacteria (Yarwood, 2018). Allochthonous organic amendments are derived from above-ground material, but they have not been subjected to wetland biogeochemical processes. Studies suggest these materials are less amenable to soil C stabilization compared to natural plant inputs and may increase CH_4 production (Scott et al., 2020). In addition to increasing CH_4 production directly, organic amendments may cause SOC priming that produces additional CH_4 (Nottingham et al., 2009), and can lead to an increase in iron (Fe) reduction and ~~iron~~ toxicity (Saaltink et al., 2017).

Iron ~~(Fe)~~ oxides play multiple roles in anoxic soils, being both an electron acceptor for organic C metabolism (Straub et al., 2001), and a stabilizing agent for SOC on mineral surfaces (Lehmann and Kleber, 2015). As a metabolite, Fe reduction competes with CH_4 production (Huang et al., 2009) and can facilitate sulfur recycling (which also competes with CH_4 production) in freshwater sediments (Hansel et al., 2015). However, ~~some~~ recent literature suggests the relationship of ~~iron~~Fe reduction and methanogenesis is more complex. Some methanogens appear capable of switching between methanogenesis and ~~iron~~Fe reduction (Sivan et al., 2016). In cultures with *Methanosarcina acetivorans*, adding ~~iron~~Fe oxides increased methane production (Ferry, 2020), presumably by the utilization of a metabolic pathway where electron flow is bifurcated with some electrons going toward ~~iron~~Fe reduction to increase energy yield (Zhuang et al., 2015; Prakash et al., 2019). In systems that are near pH neutral, Fe reduction does not necessarily have an energetic competitive advantage over CH_4 production (Bethke et al., 2011). In addition to influencing metabolic pathways, metal-oxide surfaces can stabilize organic matter, making it less bioavailable, which can affect both Fe reduction (Poggenburg et al., 2018), ~~and~~ C mineralization (Amendola et al., 2018; Lalonde et al., 2012) and production of CH_4. ~~, one of the primary methods for determining if soils are hydric National Technical Committee for Hydric Soils (NTCHS 2015), a key indicator of wetland success under mitigation guidelines.~~

We carried out a lab experiment using organic amendments commonly used in wetland restoration (biosolids (Bloom~~®~~®) - B, manure - M, composted yard waste (LeafGro®) - L, wood chips - W, and hay - H) and measured how they affected CH_4 production and Fe ~~-~~reduction. ~~A series of 1-liter~~One-~~-~~liter (1-L) glass ~~-~~jar microcosms were incubated with two different soils~~, a sandy clay loam (SCL), and a sandy loam (SL), both~~ ~~from~~ collected from sites where~~recently created~~ freshwater wetlands were recently created. The microcosms were kept under anaerobic conditions to compare the ability of these substrates to support anaerobic metabolism. We hypothesized that organic amendments would stimulate dissimilatory Fe-reduction in soils (measured as soluble ferrous iron, Fe^{2+}). Further, we hypothesized that amendments promoting Fe reduction would limit methanogenesis. We also tested differences between cured (i.e., aged/composted) and uncured (fresh) organic amendments and hypothesized that uncured amendments would increase Fe reduction due to the presence of more labile, soluble, compounds. In the United States organic amendments are often required in ~~-~~mitigation wetlands, that is, wetlands created or restored to offset wetland losses;~~.~~ h~~. H~~owever, there has not been a systematic evaluation of whether or not amendments promote hydric soil conditions (Fe reduction), ~~or may~~ lead to Fe toxicity (~~excess~~from Fe reduction), or ~~my cause excess~~~~may~~ increase CH_4 production.

2 Materials and methods

2.1 Microcosm setup

Saturated incubations were established using soil from two recent mitigation wetlands located in Maryland, USA. The first site (76°50'40.35"W, 38°47'5.41"N) was most recently a horse pasture and will be referred to as SCL denoting the texture (sandy clay loam~~), a mesic anthrudult.~~). The second site (75°47'40.20"W, 39°1'52.42"N) was most recently a corn/soy farm with tile drains and was likely a wetland prior to conversion to farmland. The second site will be referred to as SL (sandy loam). Both sites had been recently graded to establish wetland topography, so the upper portion of the soils, where soil samples were collected, were mixed endo- and umbr-aquic horizons~~. All soils were air-dried~~ but with no ped structure. Soil was collected from these recently constructed surface horizons to a depth of 15 cm, a typical depth for mixing-in organic amendments, sieved (2mm~~),~~) and homogenized prior to use. Additional soil information is shown on Supplemental Table ~~X~~S1.

Microcosm experiments were conducted in ~~1000-mL~~1-L glass straight-sided wide-mouth food canning jars. Each microcosm had a total of 600cc of solid material and was filled with water for a total volume of 660cc. The volumes needed to be precise in order to facilitate headspace and liquid sampling and to allow space for soil expansion. When amendments were added, an equal volume of soil needed to be removed so the total volume of solid material was a constant 600cc. At the start of the experiment, the headspace was purged with nitrogen gas. The incubation temperature was 20°C. Jar lids had precision drilled holes fitted with grey butyl rubber stoppers, making it possible to non-destructively remove the overlying liquid (for Fe and pH analyses) using a 7.5 cm needle. Since the head-space pressure increased due to biogenic gas production, atmospheric pressure was re-established during gas sampling events by piercing the septa with a 24-gauge needle connected to a 50mL gas-tight syringe. This procedure allowed us to record the total volume of gas produced and collect gas samples (0.01 - 1000 μL) under atmospheric pressure (Supplemental Figure S1). A small coating of silicone applied to stoppers after piercing prevented leaks. All microcosm trials were run with three replicates except where noted.

2.2 Microcosm Experiments

2.2.1 Experiment 1

We measured CH_4 and Fe^{2+} production with various organic amendments, including composted yard waste (L), composted wood chips (W), class 1 biosolids ~~(Bloom®) -~~ - (B~~,~~), manure ~~-~~(M~~,~~), and hay ~~-~~(H) at three treatment levels: 8.8% (v/v), 26%, and 53%, in two soils, a SL and a SCL. We used horse M for the SCL incubations and cow M for the SL incubations. This matched the wetland mitigation conditions at each field location. The treatment levels reflect the Maryland Department of Environment (MDE) recommendation for wetland restoration (60 cubic yards per acre assuming a 6” mixing depth) = 1x, 3x, and 6x the MDE recommended level. All amendments were sieved to 5mm. Hay was chopped with a Wiley mill, blended, or cut with scissors until it could easily pass a 5mm sieve.

2.2.2 Experiment 2

We measured CH_4 and Fe^{2+} production using cured (aged) and uncured (fresh) organic materials. We used two amendments, B and M. The two cured materials were from the same two sources as the fresh material but ~~were~~had been cured for a minimum of 3 months. We added the same amount of amendment to each microcosm based on ~~organic matter (~~OM content. Each amendment was evaluated for OM by loss-on-ignition (LOI) (550°C for 2h). Based on the percent OM we adjusted the amount of amendment so the final ~~dose~~loading rate was 20g OM/ ~~600cc~~600 cm^3 soil. The microcosm setup was the same as Experiment 1 except we used the same volume of soil (~~600cc~~600 cm^3) in all microcosms. These microcosms were incubated for 13 days and sampled periodically for Fe^{2+} and biogenic gases.

2.2.3 Experiment 3

We measured a) CH_4 and b) Fe^{2+} production as a function of pH. We used H leachate as a substrate (McMahon et al., 2005). We leached 5.63 g H with 125 cm^3 cold de-ionized water, shaking horizontally at 5°C for 24 hours. The leachate was filtered to 20 µm and immediately placed into jars with 600 ~~cc~~cm^3 SL soil and incubated for 22 days. The pH was adjusted to target levels of 5.6, 6.1, and 6.6 using a non-substrate buffer: 2-(N-morpholino) ethanesulfonic acid (MES). To determine the necessary concentration of MES, we titrated SL (pH 5.8) to our maximum desired pH (6.6). We determined that the buffering capacity of the soils corresponded to ~ 2 mN in the 125 ~~cc~~cm^3 of liquid (leachate volume), so we prepared microcosms using 125 ~~cc~~cm^3 of 20 mN MES buffer.

2.2.4 Experiment 4

We measured Fe^{2+} production using leached H as a substrate (as in Experiment 3) but compared these finding to those with unleached H, and the H residuals.

2.3 Soil, Liquid, and Gas Analyses

Prior to the start of the experiments, we analyzed the SL and SCL for soil texture, percent soil C, and extractable ~~iron~~ Fe (Supplemental Table S1). Soil texture was determined by ~~hygrometer method:~~adding 50 g soil ~~was added~~ to a 1000 ml cylinder with 0.5% hexametaphosphate. Sand settled after 1 minute and silt after 24 hours. Soil moisture content was determined as weight loss of approximately 5 g of soil dried at 105°C for 48 hours. We determined percent soil C using thermal combustion ~~analysis~~ at 950°C on a LECO CHN-2000 analyzer (LECO Corp., St. Joseph, MI). Iron extractions were performed sequentially with 1 M hydroxylamine hydrochloride (HHCL) in 25% v/v acetic acid; 50 g / 1 sodium dithionite in solution 0.35 M ace-tic acid / 0.2 M sodium citrate buffered to pH 4.8; 0.2 M ammonium oxalate / 0.17 M oxalic acid (pH 3.2) (Poulton and Canfield, 2005). The HHCL extraction targets bioavailable iron, primarily ferrihydrite and lepidocrocite. Dithionite also includes more crystalline iron oxide forms, hematite and goethite. Oxalate includes the bioavailable iron oxides and magnetite.

Throughout the experiments we measured Fe^{2+}, pH, and biogenic gases in the headspace. In some cases, Fe^{2+} and pH were measured only at the end of the incubation. Using a 3" needle, we extracted 0.3 - 1 ~~cc~~cm^3 (for Fe^{2+}) and 1 ~~cc~~cm^3 (for pH) of the supernatant liquid to avoid disturbing soil in the jars. ~~Liquid samples~~Samples of liquid supernatant were removed during gas sampling, when atmospheric pressure was maintained, to avoid loss of biogenic gases and atmospheric contamination. For the final sample point the jar contents were thoroughly mixed prior to sampling to include pore water and gases. Ferrous iron in supernatant liquid was measured with a HACH DR4000 spectrophotometer. The spectrophotometer was also used to measure Fe in the Fe-oxide extractions. Prior to analysis, extracted Fe-oxides were reduced by adding thioglycolic acid. To confirm the spectrophotometer accuracy, a subset of samples was also analyzed on a PerkinElmer PinAAcle 900T atomic absorption spectrometer. An Orion 9142BN electrode was used to determine pH.

Gas samples were collected in 12 ~~cc~~cm^3 N~~-~~-purged exetainer vials and analyzed by injecting 5 ~~cc~~cm^3 into a Varian Model 450-GC gas chromatograph. Since sample volume was typically 1 ~~cc~~cm^3 or less, 5 ~~cc~~cm^3 nitrogen gas was added to the vials immediately prior to analysis for CO_2 and CH_4, and measured concentrations were corrected for dilution and prior headspace gas concentrations. We also performed ~~For~~ fluorescent spectral scans on ~~-~~dissolved organic matter that was extracted from organic materials with 1:10 solid (weight) / deionized water (volume) for 24 hours and filtered to 0.45 µm (Fischer et al. 2020). After diluting samples, emission spectra were recorded using an Aqualog fluorometer (Horiba Scientific; Edison, NJ).

2.4 Data analysis

Unless otherwise noted, statistical determinations were done using ANOVA in R or SAS. The Fe^{2+} concentrations were evaluated using contrasts for each of the amendments compared to the control using the R multcomp package. The gas curves were modelled as piecewise, bimodal linear functions using the R “Segmented” package (Muggeo, 2008). Breakpoints were determined using the total gas curves but, in some cases, Segmented could not identify a breakpoint in the total gas curve, so CH_4 curves were used as noted in Supplemental Figures S2 & S3. Gas curves from H amendments did not fit a piecewise model and were modelled as sigmoidal functions using the SSgompertz function in R. However, SSgompertz is sensitive to data scatter, particularly at the beginning and end of the curve, so ~~in two cases, the total~~ the gas ~~and CO_2~~ curves for H6x in the SL ~~,~~ were fit ~~the data~~ with a power function in Excel.

3 Results

3.1 Experiment 1a: Effect of organic amendments and soil type on CH_4 gas production

Results for CH_4, CO_2, and tTotal gas production rates are shown on The addition of organic amendments increased CH_4 production (Table 1). The amount of the increase depended on the soil type, when CH_4 samples were collected, amendment type, and dose. In general, the SL soil produced 2.4 times as much CH_4 as the SCL (Table 1 and Supplemental Figure S4a). GMethane gas production occurred in two distinct steady-state gas production periods, which we identified as Phase 1, and then after a breakpoint, Phase 2. (Figure 1) with (iIndividual gas curves are shown in Supplemental Figures S22 (SCL) and S33 (SL).) Therefore, we reported Phase 1 & 2 gas production rates, as well as the breakpoint (Table 1). A typical piecewise gas production curve is shown in Supplemental Figure S5 with individual[AB1]Some CH_4 was produced almost immediately upon inundation (Phase 1Table 1a), but after the breakpoint (40 days in both the SL and SCL soils), there is a large increase in CH_4 as well as an average 4.7x ± 1.9 increase in total gas production (Supplemental Table 1bS2). One of our amendments, H, did not fit the linear bimodal pattern, so we reported rates separately on Table 1c.

Gas production varied by The amount of the increase depended on the soil texture, the incubation time point when CH_4 samples were collected, amendment type, and dose. gas curves shown in Supplemental Figures S2 (SCL) and S3 (SL). Supplemental Table S2 shows CH_4 and includes total gas and carbon dioxide (CO_2). In general, the SL soil produced 2.6 times as much total gas (Figure 2a) and 2.4 times as much CH_4 as the SCL (Figure 2b). In the SCL soil, CH_4 production in Phase 1 was 0.003

~~cc~~cm^3 CH_4^{-1} ~~/~~Kg $soil^{-1}$ ~~/~~day and with amendments increased to as much as 0.8 ~~cc~~ cm^3 CH_4^{-1} Kg $soil^{-1}$ day ~~cm^3/Kg/day~~ (Table ~~1~~1a). In Phase 2 ~~CH_4 was~~ 1.9 ~~cc~~ cm^3 CH_4^{-1} Kg $soil^{-1}$ day was ~~produced in~~produced in control soils ~~cm^3/Kg/day~~ and with amendments increased to as much as 28 ~~cc~~ cm^3 CH_4^{-1} Kg $soil^{-1}$ day ~~cm^3/Kg/day~~ (Table ~~1~~1b). In the SL soil, amendments increased the rate ~~CH_4~~ from 0.04 to 16 ~~cc~~ cm^3 CH_4^{-1} Kg $soil^{-1}$ day ~~cm^3/Kg/day in~~ Phase 1 and from 1.8 to 64 cm^3 CH_4^{-1} Kg $soil^{-1}$ day in Phase 2.

Gas production rates generally increase~~dd~~d with amendment ~~dose~~loading rate (Table 1a & b), as expected. ~~.~~With the exception of L in the SL, all amendments reduced the ~~breakpoint.~~time required to transition from Phase 1 to Phase 2 (i.e. the breakpoint). Biosolids caused the largest shift, decreasing the breakpoint to as little as 5 days. While amendments generally increased CH_4 production there were exceptions. Low ~~dose~~loading rates of cured amendments (L and W) had lower CH_4 production rates than unamended soil: L1 in Phase 1 in both soils; L3 in the SL; L3 in the SCL (Phase 2 only); and W1 in the SCL (Phase 2). Biosolids (B1) also lowered CH_4 production rates in the SL ~~both~~ soil~~s~~ (Phase 1) (Table ~~1~~1a). We examined the normalized CH_4 production rates (per g C in soil), but in most cases results were not statistically different at $p < 0.05$ (Supplemental Figure S4). The general trends indicate uncured amendments (e.g. B and M) produce more methane per unit carbon than cured amendments (L).

Using fresh H, biogenic gas production followed a sinusoidal pattern and we reported maximum CH_4 production rate at the inflection point (Table ~~1~~1c). Hay was prone to floating ~~and~~ at higher ~~dose~~loading rates and was ~~mostly~~ present in the water column above the ~~soil~~ surface (not in contact with soil). In the ~~in-stances~~instances where this occurred (H3 and H6 in the SCL), there was a decrease in overall gas production rate and very low CH_4 – much lower than unamended soils (Table 1c and Supplemental Figure~~s~~ S2~~2~~z~~ & S3z~~). Floating also occurred in one replicated for H6 in SL – the pattern is shown on Supplemental Figures S2&3z, but not used in the average reported value (Table 1c).

3.2 Experiment 1b: Effect of organic amendments and soil type on Fe^{2+}

The type and ~~dose~~loading rate of organic amendments affected total soluble Fe^{2+} production, compared to the unamended control, in ~~only~~ a limited number of cases (Figure 3~~1~~, Supplemental Table S2). In the SL soil, L caused a decrease ($p < 0.05$) in supernatant Fe^{2+} concentrations whereas H increased supernatant Fe^{2+} in both soils ($p < 0.05$). In a separate set of experiments, we documented the relationship between supernatant Fe and pore water Fe (Supplemental Figure S5). Soil type affected the amount of soluble Fe^{2+} produced ($p < 0.05$). We did not see a difference in Fe^{2+} in the unamended microcosms even though the SCL had 2.2x the amount of hydrochloramine hydrochloride extractable Fe (FeHHCl) compared to the SL and had 7.6x more dithionite extractable Fe (Supplemental Table S1). Of the FeHHCl in soil, 19% or less in the SCL and 61% or less in the SL was reduced to Fe^{2+}. Hay was an exception, where up to 155 % of the FeHHCl in the SCL and 236 % in the SL was reduced to Fe^{2+} (Supplemental Table S2~~3~~). During the SL soil incubations, aqueous Fe^{2+} was measured simultaneous to CH_4 production. In the H and M treatments, there was a marked increase in CH_4 production when Fe^{2+} became asymptotic. However, with the other amendments, Fe^{2+} production continued or even increased during periods of high CH_4 production. Figure 4 shows two examples that highlight this pattern~~, is a subset of the~~ and ~~for~~ the complete set of curves is in ~~(~~Supplemental Figure S6~~6)~~.

3.3 Experiment 2a: Effect of cured versus fresh organic amendments on CH_4 gas production

In Experiment 1a, it appeared that curing may have had an effect on CH_4 production. Fresh H produced the most CH_4. The H1 trials had maximum production rates of 18.2 and 27.8 ~~mg~~ cm^3 CH_4^{-1} Kg soil^{-1} day ~~cm^3/Kg/day~~ in the SCL and SL soils, respectively (Table ~~1~~1c). The H3 and H6 ~~dose~~loading rates would likely have been higher had some portion of the H not floated. The M6 trials produced the most CH_4 at 27.7 and 64.0 ~~mg~~ cm^3 CH_4^{-1} Kg soil^{-1} day ~~cm^3/Kg/day~~ in the SCL and SL soils, respectively. Of the amendments used, M was cured the least (after fresh H, which was uncured). LeafGro, a commercial composted yard waste, ~~was cured the most~~ was cured the most and produced very little CH_4, in some cases less than the controls. Since we could not specify precisely how long the organic material had been cured, we conducted a separate experiment with organic materials ~~that had been cured~~ of known curing periods (at least 90 days~~,~~), using B and M. Rather than use the same volumetric quantities, we used the same ~~dose~~loading rate based on OM content. The results confirmed that curing has a strong influence on CH_4 production. Methane production was ~~much~~ higher using fresh material in both cases and cured material ~~resulted in a decrease in~~sometimes decreased CH_4 production (Table 2).

~~3.4~~ 3.4 Experiment 2b: Effect of cured versus fresh organic amendments on Fe^{2+} production

In Experiment 1b, we observed that curing also had an effect on the amount of Fe^{2+} produced. Hay was the only amendment that produced significantly more Fe^{2+} and

~~the use of~~ L ~~saw~~produced a significant reduction in Fe^{2+} (Figure ~~1~~3). In Experiment 2 we used biosolids (B) and manure (M) that had been cured at least 3 months. Whether the material had been cured had a strong influence on Fe^{2+} production and Fe^{2+} was higher using fresh material in both cases (Figure ~~2).~~ 5).

~~3.4.1~~ 3.4.1 Spectral ~~analysis~~Analysis: Effect of ~~cured and uncured~~ organic ~~matter~~ amendments and soil type on CH_4 gas production

We observed differences in CH_4 and Fe reduction rates ~~in~~when using organic material that had been cured versus uncured ~~organic material~~. The fluorescent spectral signatures of the cured materials (B and M) were similar as were the signatures of fresh material (Supplemental Figure S7), so ~~. The fluorescent signatures varied due to~~ curing differentiated the materials more than ~~, but not due to~~ the source ~~material~~. The difference in signatures was indicative of higher concentrations of organic (humic) acids and lower nominal oxidation state in the cured materials. We considered other organic matter characterization methods such as the material's carbon to nitrogen ratio, but we did not find another reliable predictor of CH_4 and Fe^{2+} production other than curing.

3.5 Experiment 3: Effect of pH on a) CH_4 and b) Fe^{2+} production

The soil pH affected both CH_4 and Fe^{2+} production. In Experiment 1, we observed that ~~on~~ Fe^{2+} varied with pH in the SL soil ($p<0.001$; Supplemental Figure S8a), but there was little variation in the SCL ($p=0.45$; Supplemental Figure S8b). In order to isolate the effect of pH, we performed experiment 3 using a single substrate (H leachate) in the SL soil. Higher pH increased the CH_4 production rate in both Phase 1 and 2 (Table 3) and reduced the production of Fe^{2+} (Figure ~~3~~6).

3.6 Experiment 4: Leached versus unleached H and pH considerations

In Experiment 4 we measured Fe^{2+} produced from H, H leachate, and ~~the~~ H residuals (Figure 47). We expected the soluble fraction to be more labile and produce more Fe^{2+}; however, t~~T~~he H residuals (solid fraction) appeared to produce more Fe^{2+} than the leachate. A~~However, a~~s noted on the figure, ~~leaching also resulted in a change in~~separate leached fractions changed the system pH. Using the results from Experiment 2, we predict that ~~had the pHs been the same,~~at comparable pH there would have been no difference in Fe^{2+} production between H, H residuals, and leachate (Supplemental Figure ~~S9~~S9). Given the potentially strong influence of pH, ~~Therefore,~~ we re-evaluated the results from Experiment 2b, correcting for pH and confirmed that the organic material age accounts for differences in Fe^{2+} production (Supplemental Figure S10). Similarly, we considered whether pH may have affected the out-come of Experiment 1 ~~results~~. ~~However, a~~A MANOVA analysis of the Experiment 1 data (Supplemental Table ~~S4~~S3) indicated that pH and soil type had a small effect ($p=0.30$ and 0.81, respectively) compared to organic matter type and ~~dose~~loading rate ($p<0.0001$).

4 Discussion

Net CH_4 emissions are a primary factor that determines whether a wetland is a C sink or contributes to long term global warming (Neubauer and Verhoeven, 2019). Soil management practices, such as wetland restoration methods, can have a large impact on CH_4 production and total greenhouse gas emissions (Paustian et al., 2016). Our data indicate that organic amendments used in ~~mitigation~~ created or restored wetlands may ~~can~~ have a large influence on CH_4 production. Organic amendments that had been cured

(L and W) only slightly increased CH_4 emissions, ~~whereas~~ but fresh material (M and H) resulted in large increases (Tables 1a&b). This is consistent with field studies where comparable cured amendments (composted wood and yard waste), did not result in increased CH_4 emissions (Winton and Richardson, 2015), but straw (Ballantine et al., 2015) and peat bales (Green, 2014) increased CH_4 emissions. Organic material is commonly cured, or composted, to remove plant pathogens (Noble and Roberts, 2004) and to reduce the amount of cellulosic material (Hubbe et al., 2010), which competes for oxygen, contributing to phytotoxicity (Saidpullicino et al., 2007; Hu et al., 2011). Curing produces humic acids and increases the nominal oxidation state (NOSC) of C (Guo et al., 2019). When cured material is then subjected to anaerobic conditions, less CH_4 is produced (Yao and Conrad, 1999), which would make composted material more suitable in a wetland restoration context ~~to maintain an electron balance~~.

Following soil inundation, we observed two distinct gas production phases (Phase 1 and 2). This pattern is difficult to distinguish in unamended soils but has been reported previously (Yao and Conrad, 1999; Drake et al., 2009). Our ~~The~~ breakpoint (5 – 45 days Table 1b)) was similar to ~~from 5 – 36 days in a study by~~ Yao and Conrad (1999) (5 – 36 days)~~and 5 – 45 days in our study (Table 4.1)~~. The Phase 2 ~~CH_4 production~~ rates in unamended soils were also similar: 0.96 – 3.98 cm^3 CH_4^{-1} Kg soil^{-1} day ~~cm^3/Kg/day~~ in Yao and Conrad (1999) and 1.82 – 1.94 cm^3 CH_4^{-1} Kg soil^{-1} day in our study (Table ~~4.~~1b).

~~There are several known contributors to thi~~There are several explanations that could account for the observed~~s~~ gas production pattern. One is the lag period required to re-establish populations of methanogenic archaea, which becomeare likely dormant under oxic conditions and doubling times for regrowth can be on the order of days (Jabłoński et al., 2015). In our study, B had the earliest onset ofshift to Phase 2 CH_4 production (Table 1b), possibly due to elevated levels of dormant methanogens present from anaerobic digestion. Another cause forThe the two-phase gas production is thecould also be due to depletion of bioavailable ironFe-oxides, which suppressthus relieving the competition between Fe reducers and methanogens methanogenesis (Megonigal et al., 2004). However, someThe Our of our data seemed to contradict this modelwere mixed, with some treatments showaving evidence of competition by for slower Fe reducerstion, and but in others cases we did not see competitionnot. In treatment Figure 4a shows that the trial with the amendment M1, (for example,) fit the expected pattern—ferrous iron Fe in the supernatant plateaued at about the same time as the breakpoint (Figure 4b), after which methaneCH_4 production increased. In contrastFigure 4b, in shows that with W3 soluble ironFe continued to be produced well after the breakpoint, and the amount of bioavailable ironFe used during the course of the incubation was less than 28 ± 4% (Figure 4b, Supplemental Table S2). We also looked at

In addition to quantifying Fe oxide concentrations, the CO_2:CH_4 ratios can be indicative of interactions between methanogens and other reducers (Bridgham et al. 2013). As with iron-oxide utilization, weIf Fe reduction or other reduction stops during Phase 2, we would expect the CO_2:CH_4 ratio to be near 1:1 (after the methane breakpoint Bridgham et al. (2013). However, we observed notable exceptions. (also discussed in Bridgham et al. 2013). The SCL L1 trial treatment had a ratio of 73:1 after the

~~breakpoint~~in Phase 2 (Table 1b), yet still had the characteristic shift to higher overall gas production (4.67x). Other treatments also had higher CO_2:CH_4 ratios: ~~r trials (~~L3, L6, W1, B1, C, and W1-3 in the SL soil (Table 1b)~~) showed similar unexpected behaviours, but to a lesser degree~~. ~~Therefore, there are likely underlying mechanisms that contributes to the breakpoint other than depletion of iron-oxides and a shift to methanogenesis. One possible explanation put forth in other work is that redox dynamics can be controlled by the development of microsites (Yang et al., 2017).~~ Our mixed observations may have been due to microsite formation. In high producing microcosms,~~In our~~ ~~experiments~~ microsite development may have been disrupted by gas ebullition, which was ~~severe~~substantial~~tainal~~ enough in H amended trials to cause effervescence. Amendments with low gas production and limited gas ebullition (e.g. L, W and C) continued to produce Fe^{2+} after the breakpoint, ~~however,~~ possibly because methanogens were active in undisturbed microsites, as described in ~~(~~Yang et al. (~~,~~2017).

The increased gas production from organic amendments was more pronounced in SL compared to SCL, where there was 2.4x higher CH_4 and 2.6x higher gas production (Figures ~~4.~~2~~1~~a & b). ~~There are few studies that compare CH_4 production rates by soil type.~~ We observed a more pronounced effect than a recent rice field study where there was more ~~methane~~CH_4 from SL soils versus SCL, ~~;~~although in that study results were not statistically significant (Kim et al., 2018). Yagi and Minami (1990) observed that compost (approximate ~~dose~~loading rate the same as our 1x treatment) increased respiration rates by 1.8x in a SCL versus a loam soil. Maietta, Hondula, et al. (2020) observed that respiration rates were higher in a sandy loam soil compared to a silty clay, with and without 3.3% & 23% wetland hay amendments. Thus, we might conclude that in general coarser grained (sandy) soil textures emit more ~~methane~~CH_4; however, there are a number of investigations where this was not the case (Yagi and Minami, 1990; Glissmann and Conrad, 2002). Other factors may have contributed. In our experiment the SCL had 7.6x dithionite extractable Fe, and 4.6x as much %C (Supplemental Table S1), so additional studies would be needed to isolate texture as the controlling factor.

We considered the gas production from H microcosms separately because they followed a different pattern than the other amendments ~~(Table 4.1)~~, but the pattern was similar to other studies using hay (Glissmann and Conrad, 2002) and wetland hay (Maietta et al., 2020b). Our study adds to these findings by observing that H produced very low CH_4 in the water column (after floating) compared to being mixed with soil (Table 1c). This may merit further study because if this is generally true, applying fresh organic matter as a mulch, rather than mixed into the soil, could greatly reduce the adverse consequence of increased CH_4 emissions.

Reduction of Fe-oxides occurs in saturated soils in the presence of an organic substrate and is a key biogeochemical process in wetland soils. With sufficient time, hydric soils may develop redoximorphic features from Fe reduction; however, studies have not shown lasting redoximorphic development due to organic amendments (Gray, 2010; Ott et al., 2020). Organizations responsible for constructing mitigation wetlands have an interest in documenting Fe reduction prior to redoximorphic feature development as evidence soils that are hydric. Some mitigation wetland practitioners experience challenges meeting hydric soil testing standards. Although reports in the scientific literature are rare, there are examples of sites meeting vegetation and hydrology wetland indicators, but not hydric soils (Berkowitz et al., 2014). Both the soils we tested produced sufficient Fe^{2+} and would have passed hydric soils tests, so a soil ~~without the aid of an~~ amendment would not be needed.

We observed that fresh organic matter resulted in increased Fe^{2+} compared to cured organic matter (Figure ~~4.~~3), likely due to the presence of labile carbon, allowing access to more crystalline Fe-oxides (Lentini et al., 2012). ~~Fresh material such as hay has been promoted as a soil amendment in wetland construction (Melvin 2003).~~ In some soils, Fe-reducing bacteria using fresh organic matter amendments could access crystalline Fe making it more bioavailable. However, without an anoxic/oxic cycle, increased Fe^{2+} production could lead to Fe^{2+} toxicity and ferrolysis (Kirk, 2004), similar to the way fresh organic matter leads to SOC priming (Blagodatsky et al., 2010). Ferrolysis occurs when bioavailable Fe-oxides are reduced to Fe^{2+} and are subject to hydraulic transport. We observed that cured amendments, like L, lowered Fe^{2+} concentrations (Figure 3~~4.1~~), ~~likely due to combination of factors including displacement of the Fe-bearing soil by the amendment~~ possibly due to the presence of humic acids that are generated during curing (Guo et al., 2019). Humic acids often contain insufficient biogeochemical energy to drive dissimilatory Fe reduction (Keiluweit et al., 2017), chelate Fe^{2+}, removing it from the liquid phase (Catrouillet et al., 2014), and create insoluble precipitates (Shimizu et al., 2013).

Regulating Fe^{2+} production, through the selection of the appropriate OM amendment, could influence the growth of wetland plants. For example, rice growth may be stimulated under low Fe^{2+} doses of 1 mg/L (Müller et al., 2015), but higher doses can produce detrimental Fe plaque (Pereira et al., 2014). Some native wetland species are adapted to high Fe^{2+} concentrations. *Juncus effusus* growth is stimulated at 25 mg/L Fe^{2+} (Deng et al., 2009). North American native reed *Phragmites australis* ssp. *americanus* was stimulated at 11 mg/L Fe^{2+} from ferrous sulfate (Willson et al., 2017), but the invasive Eurasian lineage of *Phragmites australis* seedling growth was inhibited by Fe^{2+} as low as 1 mg/L (Batty, 2003). Soils high in free Fe^{2+} adversely affected *P. australis* growth by creating an Fe-oxide plaque on roots (Saaltink et al., 2017). Therefore, promoting Fe reduction could have a beneficial effect on native wetland plant growth while limiting invasive species seedling recruitment or growth. In our experiment, fresh organic amendments increase Fe production, but had the detrimental side effect of high CH_4 generation.

Our results show that pH has a significant effect on both the production of Fe^{2+} (Figure 4.3) and CH_4 (Table 4.3). Between pH 5.6 and 6.6, the lower pH produced more Fe^{2+} and less CH_4, consistent with thermodynamic predictions (Ye et al., 2012). HHydrogenotrophic methanogens can maximize CH_4 production at pH 5 (Bräuer et al., 2004). In rice paddy soils, CH_4 emissions had a clear peak at pH 7, but almost none below pH 5.5 (Wang et al., 1993). The strong effect of pH underscores the need to take this parameter into account when interpreting data from experiments evaluating Fe-reduction and methanogenesis. Attempting to control the pH of soils could potentially introduce confounding effects. We used an MES buffer with 10x the quantity we estimated from a soil titration and still saw shifts in the pH after incubation. With a high residual soil acidity, the amount of buffer needed to control soil pH may increase the ionic strength to a level that could influence cellular sorption to mineral and Fe-oxide surfaces (Mills et al., 1994) as well as enzyme activity (Leprince and Quiquampoix, 1996).

5 Implications

In our experiment, we saw observed that organic amendments can increase CH_4 production, particularly after extended anaerobic periods. We quantified methaneCH_4 production potential from several organic amendments, and in a separate field experiment (unnot yet published) show that these results are useful in predicting field methaneCH_4 production. There is mounting concern that CH_4 from mitigation restored and created wetlands may result in net global warming for decades to centuries (Neubauer, 2014). Our results suggest that not only do organic amendments increase CH_4 gas production overall, but uncured amendments can also decrease the time it takes before there is a large increase in both total gas production and CH_4. Methane production is not constant and dramatically increases after several weeks. Because of this, it may be beneficial to report wetland methaneCH_4 data along with inundation duration, which can strongly affect CH_4 (Hondula et al., 2021). ITherefore, it may be possible to limit CH_4 in many wetland settings, particularly mitigation wetlands where hydrology by designing systems is part of the design: swith shorter flooding or saturationinundation periodsdurations, with alternating with drier conditions. This , strategy has been proposed for rice paddy fields (Souza, 2021). Our lab study demonstrates the potential for significant CH_4 emissions, but in a real system, methanotrophic activity could attenuate CH_4 some of the emissions

(Chowdhury and Dick, 2013); however, this would not decrease the overall C loss from soils, it only changes the pathway. If organic amendments are to be used, cured amendments may be preferrable because they are not as prone to high CH_4 generation and may attenuate Fe^{2+} toxicity. Amendments that lower the soil pH increase~~s~~ Fe reduction and limit~~s~~ methanogenesis (Marquart et al., 2019). When deciding whether or not ~~the~~ to use ~~of~~ organic amendments for wetland mitigation ~~is beneficial, or necessary,~~ consideration should be given to whether or not the material has been cured, the ~~material~~ pH, the soil texture, and expected hydroperiod.

Figures

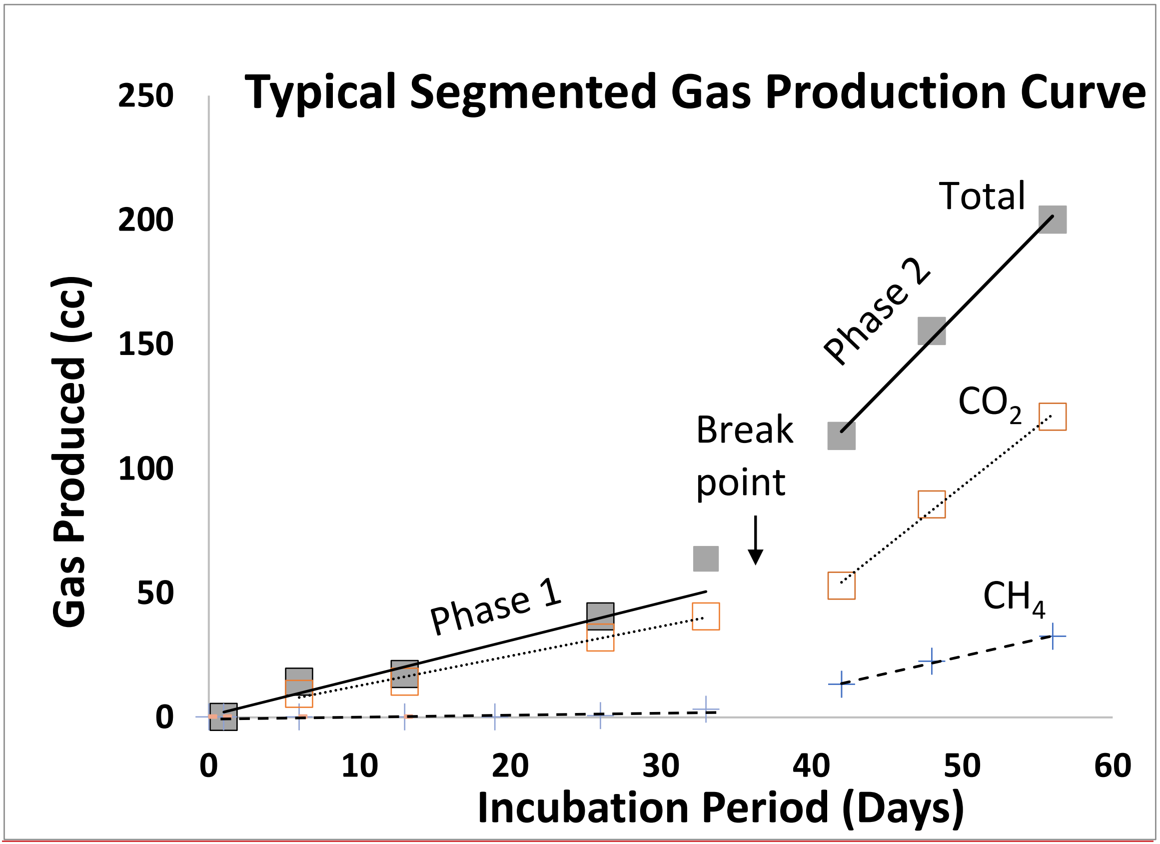


Fig. 1 – Typical gas production in saturated soils amended with organic matter (All Experiments).

Gases were best modeled using a segmented linear function. After a breakpoint the average total gas production increased by a factor of ~5 whereas there is a sharp (>> 5x) increase in methane production. Data presented is from the manure (1x) amended trials in sandy clay loam soil. Note that hay amended trials exhibited a typical sinusoidal pattern shown in Supplemental Figures S2&3h,i,j).

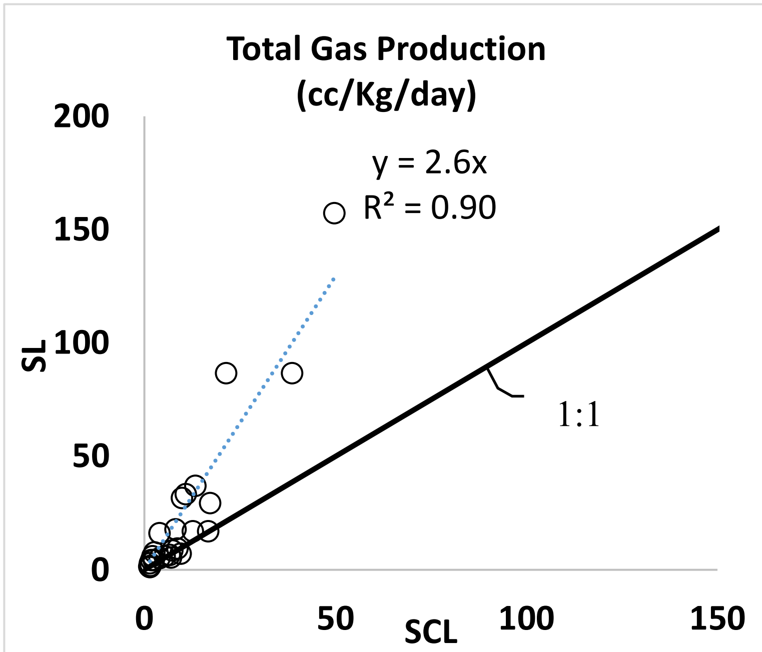


Fig. 2a. Experiment 1. Total biogenic gas production rate in the SL soil versus the SCL mesocosms. The SL mesocosms had, on average, 2.6 times higher gas production than the SCL.

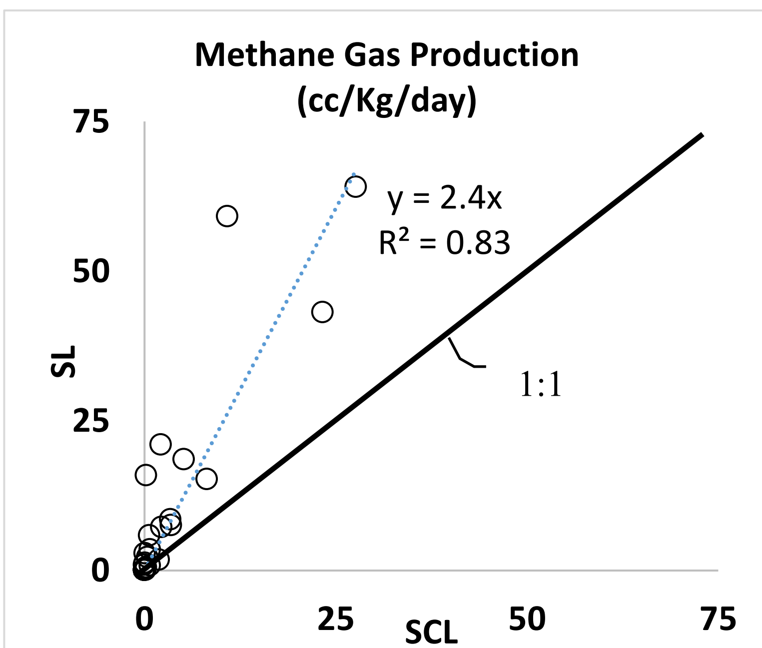


Fig. 2b. Experiment 1. Biogenic methane gas production rate in the SL soil versus the SCL mesocosms. The SL mesocosms had, on average, 2.4 times higher methane gas production than the SCL.

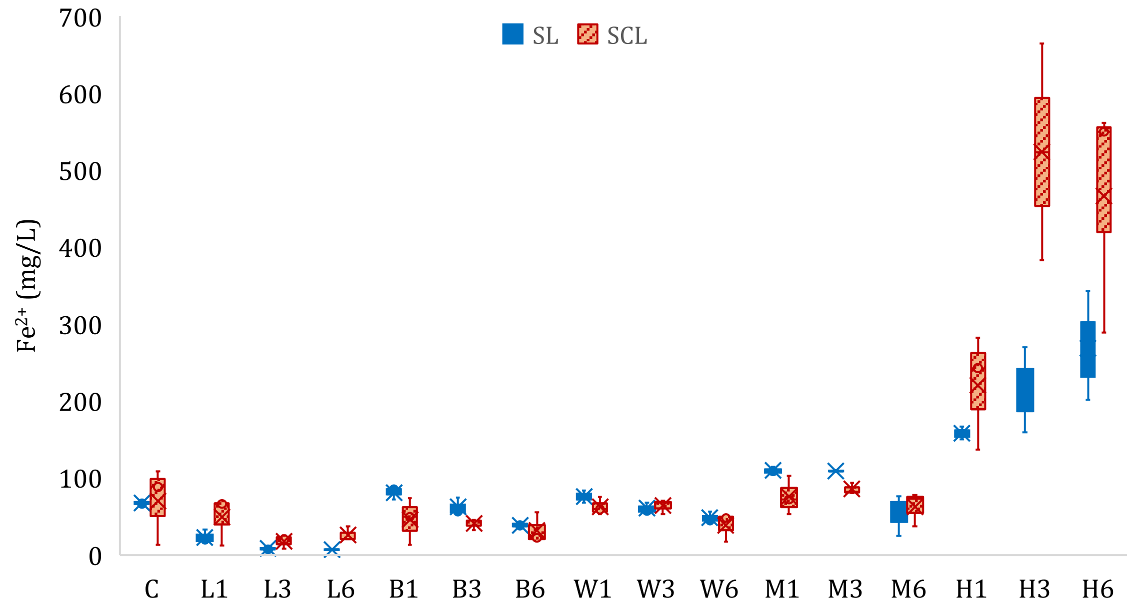
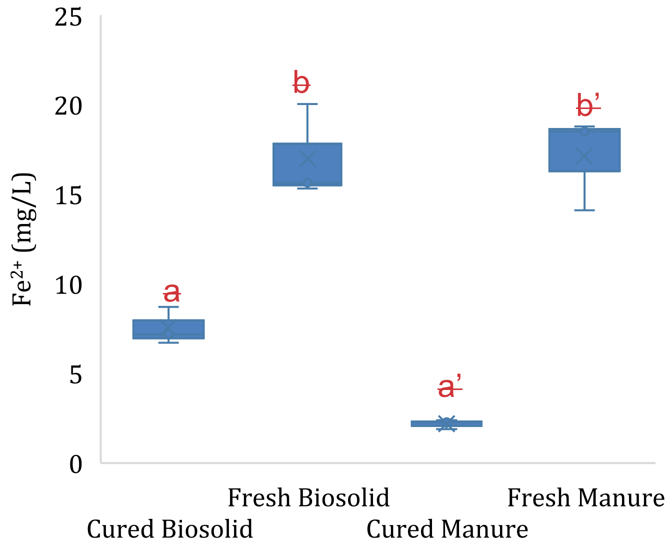
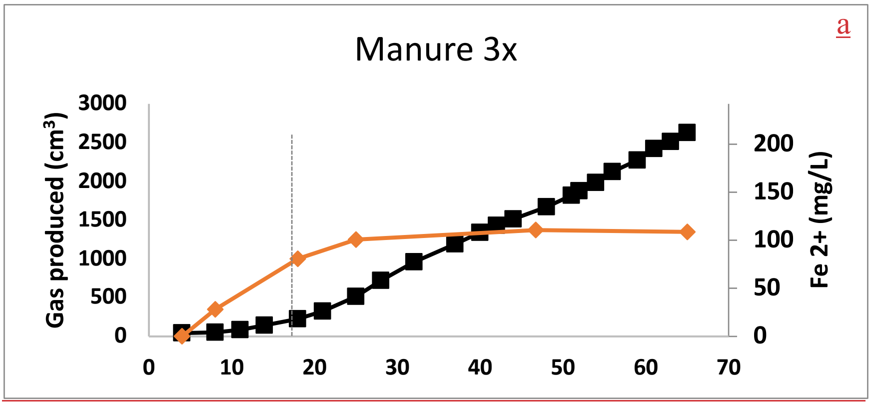


Fig. 3 – (Experiment 1b) Ferrous iron (Fe^{2+}) concentration in the liquid phase at the end of the incubation period in microcosms.

Microcosms receiving different organic amendment types and levels in Sandy Clay Loam (SCL) and Sandy Loam (SL) soils. C = no amendment control, L = LeafGro (yard waste), B = biosolids, W = wood chips, M = manure, H = hay. Numbers signify treatment level (1, 3, or 6 times amount of organic matter equivalent to 60 yd^3 / acre -1 to a depth of 6 inches). Different lower-case letters signify differences ($p < 0.05$) based on contrasts compared to C and brackets signify all results in the bracketed group were not statistically different. Hay increased total Fe^{2+} production compared to the C in both soils, and L decreased total Fe^{2+} production compared to C (SL only)).


25
20
15
10
5
0
Fe2+ (mg/L)
a
b
a'
b'
Fresh Biosolid
Fresh Manure
Cured Biosolid
Cured Manure

~~Fig. 2~~



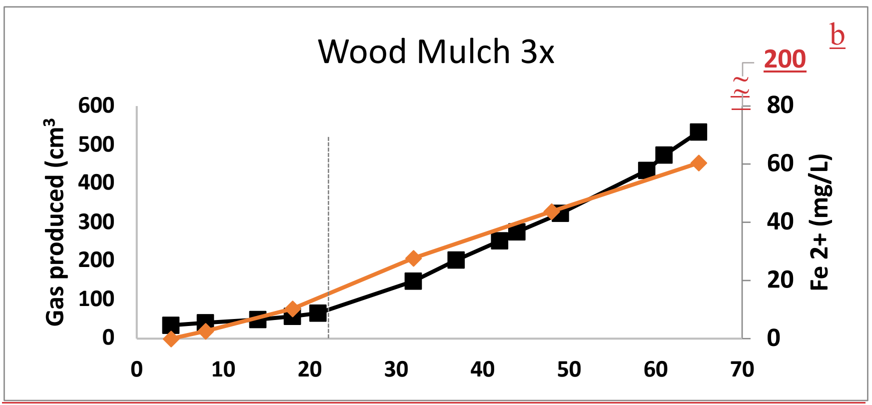


Fig. 4 – (Experiment 1b) Ferrous iron (Fe^{2+}) and methane (CH_4) in selected microcosms. Depletion of Fe coincided with the breakpoint (dashed line) with manure, but not with wood mulch. Other examples of this pattern are shown in Supplemental Figure S6. The maximum value on the secondary x-axis is the maximum expected Fe^{2+} concentration based on the HHCL extraction.

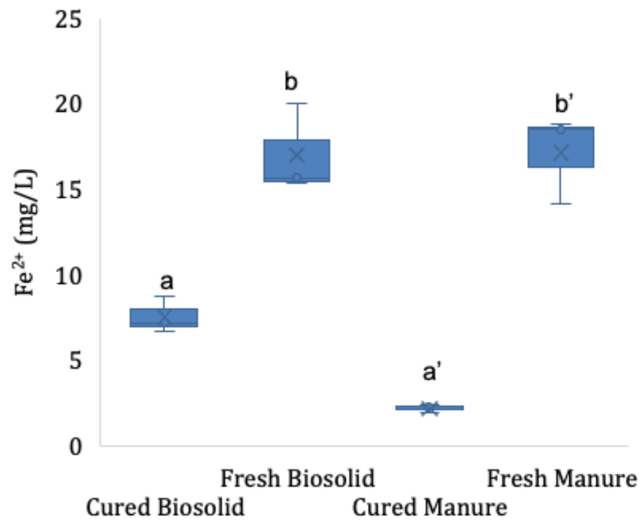


Fig. 5 – (Experiment 2b) Ferrous iron (Fe^{2+}) concentration in the liquid phase at the end of the incubation period (13 days).

Incubation was carried out ~~with cured and uncured biosolids (B) and manure (M)~~ in sandy loam ~~SL~~ soil. Different l~~L~~etters indicate a difference at $p<0.001$.

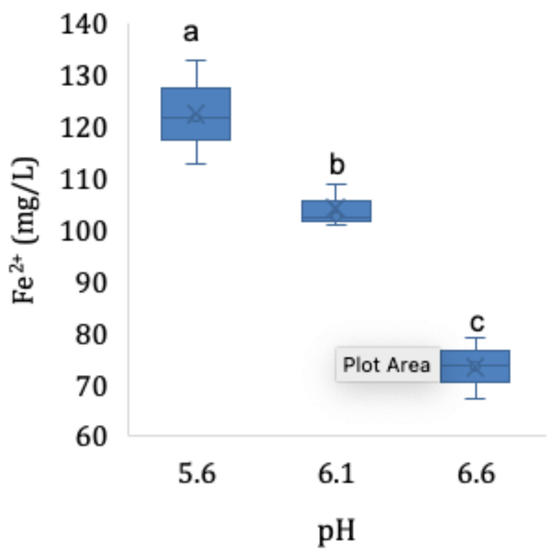


Fig. 6 – (Experiment 3) Ferrous iron (Fe^{2+}) concentration in the liquid phase with varied pH in ~~of~~ microcosms receiving hay ~~H~~ in Sandy Loam soils.

Different ~~L~~letters indicate a difference at $p<0.05$.

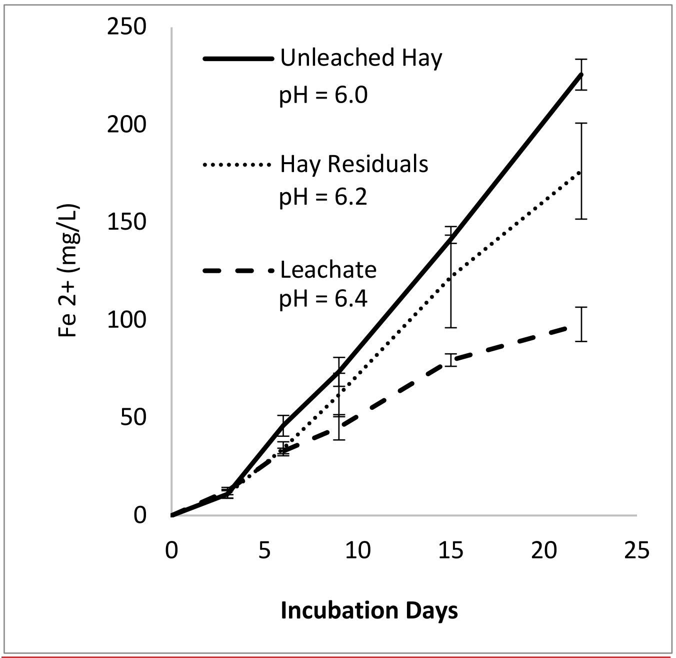


Fig. 17 – (Experiment 4) Ferrous iron (Fe^{2+}) concentration in the liquid phase with hay ~~varied in of microcosms receiving H in Sandy Loam soils. Letters indicate a difference at $p<0.05$~~as substrate.

Tables

Table 1a – (Experiment 1a[AB2] – Phase 1). Carbon dioxide (CO_2), methane (CH_4) and total gas ~~data for incubations~~production. ~~of different o~~Organic amendment types: B (bioso~~i~~lids), M (manure), L (composted yard waste), W (composted wood chips) and levels (1 = 60 yd^3 / acre equivalent: 3 = 180; 6 = 360) in silty clay loam (SCL) and sandy loam (SL) soils. Instances where organic amendments did not increase CH_4 production are bolded. Note: CO_2 : CH_4 ratios are based on calculated gas production rates, not total gas produced[SAY3].

			CO_2		CH_4		Total Gas		
Soil	Treatment	Soil (g)	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	CO_2:CH_4
SCL	Control	621.63	0.97	1.56	0.002	0.003	0.99	1.59	520.0
SCL	B1	425.24	1.53	3.61	0.08	0.18	4.13	9.70	20.1
SCL	B3	544.53	1.50	2.76	0.44	0.80	3.85	7.06	3.5
SCL	B6	468.02	2.09	4.46	0.06	0.13	3.53	7.55	34.3
SCL	M1	583.40	0.74	1.27	0.02	0.04	1.33	2.27	31.8
SCL	M3	495.56	1.79	3.61	0.32	0.64	2.05	4.13	5.6
SCL	M6	394.39	1.49	3.77	0.12	0.30	4.35	11.03	12.6
SCL	L1	586.46	0.83	1.42	0.001	**0.001**	0.85	1.45	1420.0
SCL	L3	516.34	0.89	1.72	0.01	0.01	0.91	1.77	172.0
SCL	L6	410.17	0.67	1.63	0.04	0.09	0.80	1.95	18.1
SCL	W1	593.36	1.00	1.68	0.01	0.01	0.92	1.56	168.0
SCL	W3	539.61	0.98	1.81	0.10	0.19	1.39	2.58	9.5
SCL	W6	457.42	1.03	2.25	0.11	0.24	1.29	2.81	9.4
SL	Control	634.60	0.50	0.79	0.03	0.04	0.56	0.88	19.8
SL	B1	606.80	1.25	2.06	0.02	**0.04**	4.13	6.80	51.5
SL	B3	551.50	1.57	2.84	0.44	0.79	2.92	5.29	3.6
SL	B6	467.87	2.08	4.44	0.59	1.27	3.81	8.15	3.5
SL	M1	619.92	2.62	4.22	0.58	0.93	3.49	5.63	4.5
SL	M3	588.37	4.48	7.61	3.44	5.85	9.42	16.02	1.3
SL	M6	540.93	8.63	15.95	8.59	15.87	17.92	33.13	1.0
SL	L1	600.10	0.35	0.58	0.02	**0.03**	0.73	1.22	19.3
SL	L3	530.30	0.61	1.15	0.02	**0.03**	0.78	1.46	38.3
SL	L6	425.87	0.62	1.47	0.11	0.26	1.66	3.89	5.7
SL	W1	603.27	0.98	1.62	0.06	0.10	1.55	2.56	16.2
SL	W3	538.77	1.42	2.64	0.20	0.36	2.14	3.98	7.3
SL	W6	442.57	3.05	6.88	0.24	0.54	3.23	7.31	12.7

Table 1b – (Experiment 1a[AB4] – Phase 2). Carbon dioxide (CO_2), methane (CH_4) and total gas production ~~pre and post breakpoint. Phase 2~~, and the Phase 1 : Phase 2 breakpoint. Organic amendment types: B (biosolids), M (manure), L (composted yard waste), W (composted wood chips) and levels (1 = 60 yd^3 / acre equivalent: 3 = 180; 6 = 360) in silty clay loam (SCL) and sandy loam (SL) soils. Instances where organic amendments did not increase CH_4 production are bolded. Note: r^2 values represent the combined best fit curve, using triplicate samples, for Phase 1 (Table 1a) and Phase 2.

		CO_2		CH_4		Total Gas					
Soil	Treatment	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	CO_2:CH_4	Break Point	r^2	Ph 2: Ph1
SCL	Control	2.06	3.31	1.20	1.94	2.54	4.09	1.7	40.0 ± 4.5	0.959	2.57
SCL	B1	5.58	13.13	1.47	3.45	5.49	12.91	3.8	29.3 ± 1.9	0.987	1.33
SCL	B3	3.74	6.86	4.45	8.17	9.48	17.40	0.8	20.1 ± 3.4	0.974	2.46
SCL	B6	7.42	15.85	10.90	23.29	18.20	38.89	0.7	10.3 ± 2.4	0.994	5.15
SCL	M1	2.26	3.88	1.29	2.22	5.82	9.97	1.7	40.2 ± 2.1	0.997	4.39
SCL	M3	4.64	9.37	5.39	10.89	10.69	21.58	0.9	20.8 ± 0.8	0.997	5.23
SCL	M6	5.85	14.83	10.91	27.67	19.69	49.93	0.5	22.1 ± 3.2	0.956	4.53
SCL	L1	3.85	6.57	0.05	**0.090**	3.96	6.76	73.0	32.2 ± 1.6	0.966	4.67
SCL	L3	4.21	8.16	0.39	**0.75**	4.54	8.79	10.9	32.0 ± 2.2	0.983	4.97
SCL	L6	5.90	14.39	0.92	2.24	6.95	16.95	6.4	32.0 ± 3.7	0.923	8.68
SCL	W1	1.56	2.63	0.27	**0.460**	3.22	5.42	5.7	34.0 ± 3.7	0.986	3.48
SCL	W3	1.93	3.58	1.90	3.52	4.51	8.35	1.0	24.2 ± 3.1	0.989	3.23
SCL	W6	2.19	4.79	2.36	5.15	6.22	13.60	0.9	13.0 ± 2.4	0.981	4.84
SL	Control	1.00	1.58	1.16	1.82	3.11	4.91	0.9	40.0 ± 3.2	0.957	5.55
SL	B1	4.44	7.31	5.16	8.50	10.19	16.79	0.9	8.6 ± 3.0	0.880	2.47
SL	B3	8.76	15.89	8.42	15.28	16.12	29.23	1.0	4.7 ± 1.8	0.989	5.53
SL	B6	12.61	26.96	20.15	43.07	40.39	86.33	0.6	9.1 ± 1.2	0.992	10.59
SL	M1	8.64	13.93	13.03	21.02	19.41	31.30	0.7	16.7 ± 0.7	0.998	5.56
SL	M3	15.23	25.88	34.77	59.10	50.79	86.33	0.4	17.2 ± 1.5	0.992	5.39
SL	M6	29.50	54.53	34.62	64.00	84.92	156.98	0.9	29.4 ± 1.4	0.974	4.74
SL	L1	1.35	2.24	1.71	2.85	3.76	6.26	0.8	38.3 ± 1.2	0.992	5.12
SL	L3	2.27	4.27	1.86	3.50	4.82	9.09	1.2	40.5 ± 2.0	0.977	6.22
SL	L6	4.25	9.99	3.07	7.21	7.15	16.78	1.4	44.8 ± 1.3	0.988	4.31
SL	W1	2.10	3.48	1.32	2.19	3.47	5.76	1.6	25.6 ± 7.6	0.762	2.25
SL	W3	6.58	12.22	4.05	7.51	9.46	17.56	1.6	23.2 ± 2.3	0.974	4.41
SL	W6	10.10	22.83	8.23	18.60	16.22	36.65	1.2	23.2 ± 1.1	0.991	5.02
									AVERAGE		4.7
									STDEV		1.9

Table 1c – Experiment 1a. Carbon dioxide (CO_2), methane (CH_4) and total gas production with hay (H) amendment. H amended trials fit a sigmoidal, not segmented, pattern, and therefore there was no breakpoint and we present p values for the sigmoidal fit, except H6 SL rates where we used a power function in Excel and report the r^2 value. Gas production rates (cm 3 gas Kg soil -1 day -1) represent maximum at the inflection point. ~~T~~The amendment ~~amendment hay (H)~~ floated to the surface in the SCL H3 and H6 trials, which resulted in unusually low CH_4 production rates.~~ in SCL.~~

Sigmoidal curve values			CO_2			CH_4			Total Gas			
Soil	Treatment	Soil (g)	cm^3/day	cm^3/Kg/day	p	cm^3/day	cm^3/Kg/day	p	cm^3/day	cm^3/Kg/day	p	CO_2:CH_4
SCL	H1	573.03	9.70	16.93	2.0E-16	10.40	18.15	0.164	~~18.40~~37.1	64.75~~32.11~~	1.3E-12	0.93
SCL	H3	477.85	7.50	15.70	3.0E-14	0.02	0.04	0.933	9.90	20.72	7.8E-6	393
SCL	H6	334.20	6.60	19.75	0.019	0.09	0.27	0.921	6.70	20.05	9.6E-13	73
SL	H1	582.57	8.90	15.28	5.5E-14	16.20	27.81	0.283	18.40	31.58	2.9E-4	0.55
SL	H3	478.00	20.80	43.51	1.8E-13	12.20	25.52	0.636	36.80	76.99	0.0093	1.7
SL	H6	321.13	~~14.70~~50.71	158.0~~45.78~~	0.93(r^2)	~~13.20~~77.7	242.1~~41.10~~	0.69(r^2)	~~35.60~~79.79	~~110.86~~ 248.47	0.74(r^2)	0.65~~1.1~~

Table 2 – (Experiment 2a). Methane gas data for incubations with fresh and cured organic matter in sandy loam soil~~SL (Experiment 1)~~.

Control data (*) from Experiment 1a (Table 1a) included for reference. Different l~~L~~etters indicate a difference at $p<0.001$.

Treatment	Phase 1 Methane (cm^3/Kg/day)	Phase 2 Methane (~~cc~~cm^3/Kg/day)
Control*~~[a]~~	0.04	1.8
Cured Biosolids[a]	0.003	0.37
Fresh Biosolids[b]	3.29	17.48
Cured Manure[a’]	0.22	5.4
Fresh Manure[b’]	3.85	42.36

Table 3 – (Experiment 3). Methane gas data versus pH of microcosms.

Microcosms receiving hayH in Sandy Loam soils (Experiment 3). Different lLetters indicate a difference at $p<0.001$.

pH	Phase 1 CH_4 (cm^3/Kg/day)	Phase 2 CH_4 (cm^3/Kg/day)
5.6[a]	0.44	10.6
6.1[b]	1.0	13.0
6.6[c]	1.8	13.8

Declarations

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Conflicts of interest/Competing interest

Authors declare no conflict of interest

Availability of data and material

Significant data detail is available in the supplementary materials. Additional raw data available upon request.

Code availability

None

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