Quantification of potential methane emissions associated with organic matter amendments following oxic soil inundation

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Abstract

Methane (CH_4) emissions are a potent contributor to global warming and wetlands can be a significant CH_4 source. In a microcosm study, we evaluated how the practice of amending soils with organic matter as part of wetland restoration projects may affect CH_4 production potential. Organic amendments including hay, manure, biosolids, composted yard waste, and wood mulch were evaluated at three different levels. Using 1-liter glass microcosms, we measured the production of biogenic gases over 60 days in two soils designated by texture: a sandy loam (SL) and a sandy clay loam (SCL). Fresh organic amendments increased CH_4 production, leading to potentially higher global warming potential and wetland C loss, and CH_4 production was more pronounced in the SL. We observed biogenic gas production in two sequential steady state phases: Phase 1 produced some CH_4 but was mostly carbon dioxide (CO_2) followed by Phase 2, two to six weeks later, with higher total gas and nearly equal amounts of CH_4 and CO_2. If this is generally true in soils, it may be appropriate to report CH_4 emissions in the context of inundation duration. The CH_4 from the SCL soil ranged from 0.003 – 0.8 cm^3 Kg^{-1} day in Phase 1 to 0.75 – 28 cm^3 Kg^{-1} day in Phase 2, and from the SL range from 0.03 – 16 cm^3 Kg^{-1} day in Phase 1 to 1.8 – 64 cm^3 Kg^{-1} day in Phase 2. Adding fresh organic matter

(e.g., hay) increased ferrous iron (Fe^{2+}) concentrations whereas in some cases composted organic matter decreased both Fe^{2+} concentrations and CH_4 production. Methanogenesis normally increases following the depletion of reduceable Fe; however, we observed instances where this was not the case, suggesting other biogeochemical mechanisms contributed to the shift in gas production.

Keywords Methane emissions, mitigation wetlands, organic amendments

1 Introduction

The ecological benefits of wetlands are well documented, including their role as carbon (C) sinks to stabilize global climate (Mitsch et al., 2015). Driven in part by this ecological contribution, from 1970 to 2015 human-made wetlands have increased 233% (Darrah et al., 2019). Between 2004 and 2009 the United States saw a net gain of 16,670 hectares of freshwater wetlands: 360,820 hectares of new wetlands to offset 344,140 hectares of existing (presumably C-sink) wetlands that were destroyed (Dahl, 2011). Although created or restored wetlands may effectively sequester C, it may take hundreds of years to offset their radiative forcing due to methane (CH_4) emissions (Neubauer, 2014). With such a large number of human-made wetlands, and their potential to increase global warming, it is vital to consider factors that may contribute to CH_4 emissions.

Organic amendments such as straw, wood mulch, manure, and biosolids, mixed into the soil, are thought to accelerate C storage by enhancing the conversion of plant-derived compounds to microbial residues (Richardson et al., 2016). Microbial residues, largely aliphatic-C from cell membrane lipids, can accumulate in soil and are not directly accessible by methanogens (Chen et al., 2018). Plants contribute both above and

48 belowground organic matter (OM). Belowground plant materials are preferentially
49 converted to soil organic carbon (SOC) (Mazzilli et al., 2015). In saturated soils root
50 residues of wetland plants contain suberin and cutin (Watanabe et al., 2013), which
51 persist, reducing biogenic gas production (Mikutta et al., 2006). Before contributing to
52 SOC, standing litter in natural wetlands is partially decomposed by fungi (Kuehn et al.,
53 2011), and further decomposed by aerobic bacteria (Yarwood, 2018). Allochthonous
54 organic amendments are derived from above-ground material, but they have not been
55 subjected to wetland biogeochemical processes. Studies suggest these materials are less
56 amenable to soil C stabilization compared to natural plant inputs and may increase CH_4
57 production (Scott et al., 2020). In addition to increasing CH_4 production directly, organic
58 amendments may cause SOC priming that produces additional CH_4 (Nottingham et al.,
59 2009), and can lead to an increase in iron (Fe) reduction and toxicity (Saaltink et al.,
60 2017).
61 Iron oxides play multiple roles in anoxic soils, being both an electron acceptor for
62 organic C metabolism (Straub et al., 2001), and a stabilizing agent for SOC on mineral
63 surfaces (Lehmann and Kleber, 2015). As a metabolite, Fe reduction competes with CH_4
64 production (Huang et al., 2009) and can facilitate sulfur recycling (which also competes
65 with CH_4 production) in freshwater sediments (Hansel et al., 2015). However, recent
66 literature suggests the relationship of Fe reduction and methanogenesis is more complex.
67 Some methanogens appear capable of switching between methanogenesis and Fe
68 reduction (Sivan et al., 2016). In cultures with *Methanosarcina acetivorans*, adding Fe
69 oxides increased methane production (Ferry, 2020), presumably by the utilization of a

70 metabolic pathway where electron flow is bifurcated with some electrons going toward
71 Fe reduction to increase energy yield (Zhuang et al., 2015; Prakash et al., 2019). In
72 systems that are near pH neutral, Fe reduction does not necessarily have an energetic
73 competitive advantage over CH_4 production (Bethke et al., 2011). In addition to
74 influencing metabolic pathways, metal-oxide surfaces can stabilize organic matter,
75 making it less bioavailable, which can affect both Fe reduction (Poggenburg et al., 2018),
76 C mineralization (Amendola et al., 2018; Lalonde et al., 2012) and production of CH_4.
77 We carried out a lab experiment using organic amendments commonly used in
78 wetland restoration (biosolids (Bloom®) - B, manure - M, composted yard waste
79 (LeafGro®) - L, wood chips - W, and hay - H) and measured how they affected CH_4
80 production and Fe reduction. One-liter (1-L) glass jar microcosms were incubated with
81 two different soils collected from sites where freshwater wetlands were recently created.
82 The microcosms were kept under anaerobic conditions to compare the ability of these
83 substrates to support anaerobic metabolism. We hypothesized that organic amendments
84 would stimulate dissimilatory Fe-reduction in soils (measured as soluble ferrous iron,
85 Fe^{2+}). Further, we hypothesized that amendments promoting Fe reduction would limit
86 methanogenesis. We also tested differences between cured (i.e., aged/composted) and
87 uncured (fresh) organic amendments and hypothesized that uncured amendments would
88 increase Fe reduction due to the presence of more labile, soluble, compounds. In the
89 United States organic amendments are often required in mitigation wetlands, that is,
90 wetlands created or restored to offset wetland losses; however, there has not been a

systematic evaluation of whether or not amendments promote hydric soil conditions (Fe reduction), lead to Fe toxicity (from Fe reduction), or increase CH_4 production.

2 Materials and methods

2.1 Microcosm setup

Saturated incubations were established using soil from two recent mitigation wetlands located in Maryland, USA. The first site (76°50'40.35"W, 38°47'5.41"N) was most recently a horse pasture and will be referred to as SCL denoting the texture (sandy clay loam). The second site (75°47'40.20"W, 39°1'52.42"N) was most recently a corn/soy farm with tile drains and was likely a wetland prior to conversion to farmland. The second site will be referred to as SL (sandy loam). Both sites had been recently graded to establish wetland topography, so the upper portion of the soils, where soil samples were collected, were mixed endo- and umbr-aquic horizons but with no ped structure. Soil was collected from these recently constructed surface horizons to a depth of 15 cm, a typical depth for mixing-in organic amendments, sieved (2mm) and homogenized prior to use. Additional soil information is shown on Supplemental Table S1.

Microcosm experiments were conducted in 1-L glass straight-sided wide-mouth food canning jars. Each microcosm had a total of 600cc of solid material and was filled with water for a total volume of 660cc. The volumes needed to be precise in order to facilitate headspace and liquid sampling and to allow space for soil expansion. When amendments were added, an equal volume of soil needed to be removed so the total volume of solid material was a constant 600cc. At the start of the experiment, the headspace was purged with nitrogen gas. The incubation temperature was 20ºC. Jar lids

had precision drilled holes fitted with grey butyl rubber stoppers, making it possible to non-destructively remove the overlying liquid (for Fe and pH analyses) using a 7.5 cm needle. Since the head-space pressure increased due to biogenic gas production, atmospheric pressure was re-established during gas sampling events by piercing the septa with a 24-gauge needle connected to a 50mL gas-tight syringe. This procedure allowed us to record the total volume of gas produced and collect gas samples (0.01 - 1000 μL) under atmospheric pressure (Supplemental Figure S1). A small coating of silicone applied to stoppers after piercing prevented leaks. All microcosm trials were run with three replicates except where noted.

2.2 Microcosm Experiments

2.2.1 Experiment 1

We measured CH_4 and Fe^{2+} production with various organic amendments, including composted yard waste (L), composted wood chips (W), class 1 biosolids - (B), manure (M), and hay (H) at three treatment levels: 8.8% (v/v), 26%, and 53%, in two soils, a SL and a SCL. We used horse M for the SCL incubations and cow M for the SL incubations. This matched the wetland mitigation conditions at each field location. The treatment levels reflect the Maryland Department of Environment (MDE) recommendation for wetland restoration (60 cubic yards per acre assuming a 6" mixing depth) = 1x, 3x, and 6x the MDE recommended level. All amendments were sieved to 5mm. Hay was chopped with a Wiley mill, blended, or cut with scissors until it could easily pass a 5mm sieve.

2.2.2 Experiment 2

We measured CH_4 and Fe^{2+} production using cured (aged) and uncured (fresh) organic materials. We used two amendments, B and M. The two cured materials were from the same two sources as the fresh material but had been cured for a minimum of 3 months. We added the same amount of amendment to each microcosm based on OM content. Each amendment was evaluated for OM by loss-on-ignition (LOI) (550°C for 2h). Based on the percent OM we adjusted the amount of amendment so the final loading rate was 20g OM/ 600 cm^3 soil. The microcosm setup was the same as Experiment 1 except we used the same volume of soil (600 cm^3) in all microcosms. These microcosms were incubated for 13 days and sampled periodically for Fe^{2+} and biogenic gases.

2.2.3 Experiment 3

We measured a) CH_4 and b) Fe^{2+} production as a function of pH. We used H leachate as a substrate (McMahon et al., 2005). We leached 5.63 g H with 125 cm^3 cold de-ionized water, shaking horizontally at 5°C for 24 hours. The leachate was filtered to 20 μm and immediately placed into jars with 600 cm^3 SL soil and incubated for 22 days. The pH was adjusted to target levels of 5.6, 6.1, and 6.6 using a non-substrate buffer: 2-(N-morpholino) ethanesulfonic acid (MES). To determine the necessary concentration of MES, we titrated SL (pH 5.8) to our maximum desired pH (6.6). We determined that the buffering capacity of the soils corresponded to ~ 2 mN in the 125 cm^3 of liquid (leachate volume), so we prepared microcosms using 125 cm^3 of 20 mN MES buffer.

2.2.4 Experiment 4

We measured Fe^{2+} production using leached H as a substrate (as in Experiment 3) but compared these finding to those with unleached H, and the H residuals.

157 **2.3 Soil, Liquid, and Gas Analyses**

158 Prior to the start of the experiments, we analyzed the SL and SCL for soil texture,
159 percent soil C, and extractable Fe (Supplemental Table S1). Soil texture was determined
160 by adding 50 g soil to a 1000 ml cylinder with 0.5% hexametaphosphate. Sand settled
161 after 1 minute and silt after 24 hours. Soil moisture content was determined as weight
162 loss of approximately 5 g of soil dried at 105^{o}C for 48 hours. We determined percent soil
163 C using thermal combustion at 950°C on a LECO CHN-2000 analyzer (LECO Corp., St.
164 Joseph, MI). Iron extractions were performed sequentially with 1 M hydroxylamine
165 hydrochloride (HHCL) in 25% v/v acetic acid; 50 g / 1 sodium dithionite in solution 0.35
166 M ace-tic acid / 0.2 M sodium citrate buffered to pH 4.8; 0.2 M ammonium oxalate / 0.17
167 M oxalic acid (pH 3.2) (Poulton and Canfield, 2005). The HHCL extraction targets
168 bioavailable iron, primarily ferrihydrite and lepidocrocite. Dithionite also includes more
169 crystalline iron oxide forms, hematite and goethite. Oxalate includes the bioavailable iron
170 oxides and magnetite.

171 Throughout the experiments we measured Fe^{2+}, pH, and biogenic gases in the
172 headspace. In some cases, Fe^{2+} and pH were measured only at the end of the incubation.
173 Using a 3" needle, we extracted 0.3 - 1 cm^3 (for Fe^{2+}) and 1 cm^3 (for pH) of the
174 supernatant liquid to avoid disturbing soil in the jars. Samples of liquid supernatant were
175 removed during gas sampling, when atmospheric pressure was maintained, to avoid loss
176 of biogenic gases and atmospheric contamination. For the final sample point the jar
177 contents were thoroughly mixed prior to sampling to include pore water and gases.
178 Ferrous iron in supernatant liquid was measured with a HACH DR4000

spectrophotometer. The spectrophotometer was also used to measure Fe in the Fe-oxide extractions. Prior to analysis, extracted Fe-oxides were reduced by adding thioglycolic acid. To confirm the spectrophotometer accuracy, a subset of samples was also analyzed on a PerkinElmer PinAAcle 900T atomic absorption spectrometer. An Orion 9142BN electrode was used to determine pH.

Gas samples were collected in 12 cm^3 N-purged exetainer vials and analyzed by injecting 5 cm^3 into a Varian Model 450-GC gas chromatograph. Since sample volume was typically 1 cm^3 or less, 5 cm^3 nitrogen gas was added to the vials immediately prior to analysis for CO_2 and CH_4, and measured concentrations were corrected for dilution and prior headspace gas concentrations. We also performed fluorescent spectral scans on dissolved organic matter that was extracted from organic materials with 1:10 solid (weight) / deionized water (volume) for 24 hours and filtered to 0.45 μm (Fischer et al. 2020). After diluting samples, emission spectra were recorded using an Aqualog fluorometer (Horiba Scientific; Edison, NJ).

2.4 Data analysis

Unless otherwise noted, statistical determinations were done using ANOVA in R or SAS. The Fe^{2+} concentrations were evaluated using contrasts for each of the amendments compared to the control using the R multcomp package. The gas curves were modelled as piecewise, bimodal linear functions using the R "Segmented" package (Muggeo, 2008). Breakpoints were determined using the total gas curves but, in some cases, Segmented could not identify a breakpoint in the total gas curve, so CH_4 curves were used as noted in Supplemental Figures S2 & S3. Gas curves from H amendments

did not fit a piecewise model and were modelled as sigmoidal functions using the SSgompertz function in R. However, SSgompertz is sensitive to data scatter, particularly at the beginning and end of the curve, so the gas curves for H6x in the SL were fitted with a power function in Excel.

3 Results

We present results from four separate experiments, summarized in Table 1. In Experiment 1, we evaluated Fe and CH_4 production by varying OM type and dose, and soil type (SL vs SCL). In Experiment 2 we controlled other factors and compared composted versus fresh OM. In Experiment 3 we characterized the effects of pH. In Experiment 4 we compared iron reduction from the soluble and particulate fraction of fresh hay, and the results were used to emphasize the pH effect.

3.1 Experiment 1a: Effect of organic amendments and soil type on CH_4 gas production

Gas production occurred in two distinct steady-state gas production periods, which we identified as Phase 1, and then after a breakpoint, Phase 2 (Figure 1) with individual gas curves are shown in Supplemental Figures S2 (SCL) and S3 (SL). Some CH_4 was produced almost immediately upon inundation (Table 2a), but after the breakpoint (40 days in both the SL and SCL soils), there is a large increase in CH_4 as well as an average 4.7x ± 1.9 increase in total gas production (Table 2b). One of our amendments, H, did not fit the linear bimodal pattern, so we reported rates separately on Table 2c.

Gas production varied by soil texture. In general, the SL soil produced 2.6 times as much total gas (Figure 2a) and 2.4 times as much CH_4 as the SCL (Figure 2b). In the

223 SCL soil, CH_4 production in Phase 1 was 0.003 cm^3 CH_4^{-1} Kg $soil^{-1}$ day and with
224 amendments increased to as much as 0.8 cm^3 CH_4^{-1} Kg $soil^{-1}$ day (Table 2a). In Phase 2
225 1.9 cm^3 CH_4^{-1} Kg $soil^{-1}$ day was produced in control soils and with amendments
226 increased to as much as 28 cm^3 CH_4^{-1} Kg $soil^{-1}$ day (Table 2b). In the SL soil,
227 amendments increased the rate from 0.04 to 16 cm^3 CH_4^{-1} Kg $soil^{-1}$ day Phase 1 and from
228 1.8 to 64 cm^3 CH_4^{-1} Kg $soil^{-1}$ day in Phase 2.

229 Gas production rates generally increased with amendment loading rate (Table 2a
230 & b), as expected. With the exception of L in the SL, all amendments reduced the time
231 required to transition from Phase 1 to Phase 2 (i.e. the breakpoint). Biosolids caused the
232 largest shift, decreasing the breakpoint to as little as 5 days. While amendments generally
233 increased CH_4 production there were exceptions. Low loading rates of cured amendments
234 (L and W) had lower CH_4 production rates than unamended soil: L1 in Phase 1 in both
235 soils; L3 in the SL; L3 in the SCL (Phase 2 only); and W1 in the SCL (Phase 2).
236 Biosolids (B1) also lowered CH_4 production rates in the SL soil (Phase 1) (Table 2a). We
237 examined the normalized CH_4 production rates (per g C in soil), but in most cases results
238 were not statistically different at $p < 0.05$ (Supplemental Figure S4). The general trends
239 indicate uncured amendments (e.g. B and M) produce more methane per unit carbon than
240 cured amendments (L).

241 Using fresh H, biogenic gas production followed a sinusoidal pattern and we
242 reported maximum CH_4 production rate at the inflection point (Table 2c). Hay was prone
243 to floating at higher loading rates and was present in the water column above the surface
244 (not in contact with soil). In the instances where this occurred (H3 and H6 in the SCL),

245 there was a decrease in overall gas production rate and very low CH_4 – much lower than
246 unamended soils (Table 2c and Supplemental Figure S2z). Floating also occurred in one
247 replicated for H6 in SL – the pattern is shown on Supplemental Figures S2&3z, but not
248 used in the average reported value (Table 2c).

249 **3.2 Experiment 1b: Effect of organic amendments and soil type on Fe^{2+}**

250 The type and loading rate of organic amendments affected total soluble Fe^{2+}
251 production, compared to the unamended control, in a limited number of cases (Figure 3,
252 Supplemental Table S2). In the SL soil, L caused a decrease ($p < 0.05$) in supernatant
253 Fe^{2+} concentrations whereas H increased supernatant Fe^{2+} in both soils ($p < 0.05$). In a
254 separate set of experiments, we documented the relationship between supernatant Fe and
255 pore water Fe (Supplemental Figure S5). Soil type affected the amount of soluble Fe^{2+}
256 produced ($p < 0.05$). We did not see a difference in Fe^{2+} in the unamended microcosms
257 even though the SCL had 2.2x the amount of hydrochloramine hydrochloride extractable
258 Fe (FeHHCl) compared to the SL and had 7.6x more dithionite extractable Fe
259 (Supplemental Table S1). Of the FeHHCl in soil, 19% or less in the SCL and 61% or less
260 in the SL was reduced to Fe^{2+}. Hay was an exception, where up to 155 % of the FeHHCl
261 in the SCL and 236 % in the SL was reduced to Fe^{2+} (Supplemental Table S2). During
262 the SL soil incubations, aqueous Fe^{2+} was measured simultaneous to CH_4 production. In
263 the H and M treatments, there was a marked increase in CH_4 production when Fe^{2+}
264 became asymptotic. However, with the other amendments, Fe^{2+} production continued or
265 even increased during periods of high CH_4 production. Figure 4 shows two examples that
266 highlight this pattern and the complete set of curves is in Supplemental Figure S6.

3.3 Experiment 2a: Effect of cured versus fresh organic amendments on CH_4 gas production

In Experiment 1a, it appeared that curing may have had an effect on CH_4 production. Fresh H produced the most CH_4. The H1 trials had maximum production rates of 18.2 and 27.8 cm^3 CH_4^{-1} Kg $soil^{-1}$ day in the SCL and SL soils, respectively (Table 2c). The H3 and H6 loading rates would likely have been higher had some portion of the H not floated. The M6 trials produced the most CH_4 at 27.7 and 64.0 cm^3 CH_4^{-1} Kg $soil^{-1}$ day in the SCL and SL soils, respectively. Of the amendments used, M was cured the least (after fresh H, which was uncured). LeafGro, a commercial composted yard waste, was cured the most and produced very little CH_4, in some cases less than the controls. Since we could not specify precisely how long the organic material had been cured, we conducted a separate experiment with organic materials of known curing periods (at least 90 days), using B and M. Rather than use the same volumetric quantities, we used the same loading rate based on OM content. The results confirmed that curing has a strong influence on CH_4 production. Methane production was higher using fresh material in both cases and cured material sometimes decreased CH_4 production (Table 3).

3.4 Experiment 2b: Effect of cured versus fresh organic amendments on Fe^{2+} production

In Experiment 1b, we observed that curing also had an effect on the amount of Fe^{2+} produced. Hay was the only amendment that produced significantly more Fe^{2+} and L produced a significant reduction in Fe^{2+} (Figure 3). In Experiment 2 we used biosolids (B) and manure (M) that had been cured at least 3 months. Whether the material had been cured had a strong influence on Fe^{2+} production and Fe^{2+} was higher using fresh material in both cases (Figure 5).

3.4.1 Spectral Analysis: Effect of organic amendments and soil type on CH_4 gas production

We observed differences in CH_4 and Fe reduction rates when using organic material that had been cured versus uncured. The fluorescent spectral signatures of the cured materials (B and M) were similar as were the signatures of fresh material (Supplemental Figure S7), so curing differentiated the materials more than the source. The difference in signatures was indicative of higher concentrations of organic (humic) acids and lower nominal oxidation state in the cured materials. We considered other organic matter characterization methods such as the material's carbon to nitrogen ratio, but we did not find another reliable predictor of CH_4 and Fe^{2+} production other than curing.

3.5 Experiment 3: Effect of pH on a) CH_4 and b) Fe^{2+} production

The soil pH affected both CH_4 and Fe^{2+} production. In Experiment 1, we observed that Fe^{2+} varied with pH in the SL soil ($p<0.001$; Supplemental Figure S8a), but there was little variation in the SCL ($p=0.45$; Supplemental Figure S8b). In order to isolate the effect of pH, we performed experiment 3 using a single substrate (H leachate) in the SL soil. Higher pH increased the CH_4 production rate in both Phase 1 and 2 (Table 4) and reduced the production of Fe^{2+} (Figure 6).

3.6 Experiment 4: Leached versus unleached H and pH considerations

In Experiment 4 we measured Fe^{2+} produced from H, H leachate, and H residuals (Figure 7). We expected the soluble fraction to be more labile and produce more Fe^{2+}; however, the H residuals (solid fraction) appeared to produce more Fe^{2+} than the leachate. As noted on the figure, separate leached fractions changed the system pH. Using the results from Experiment 2, we predict that at comparable pH there would have been

312 no difference in Fe^{2+} production between H, H residuals, and leachate (Supplemental
313 Figure S9). Given the potentially strong influence of pH, we re-evaluated the results from
314 Experiment 2b, correcting for pH and confirmed that the organic material age accounts
315 for differences in Fe^{2+} production (Supplemental Figure S10). Similarly, we considered
316 whether pH may have affected the out-come of Experiment 1. A MANOVA analysis of
317 the Experiment 1 data (Supplemental Table S3) indicated that pH and soil type had a
318 small effect ($p=0.30$ and 0.81, respectively) compared to organic matter type and loading
319 rate ($p<0.0001$).

320 **4 Discussion**

321 Net CH_4 emissions are a primary factor that determines whether a wetland is a C
322 sink or contributes to long term global warming (Neubauer and Verhoeven, 2019). Soil
323 management practices, such as wetland restoration methods, can have a large impact on
324 CH_4 production and total greenhouse gas emissions (Paustian et al., 2016). Our data
325 indicate that organic amendments used in created or restored wetlands may have a large
326 influence on CH_4 production. Organic amendments that had been cured (L and W) only
327 slightly increased CH_4 emissions, but fresh material (M and H) resulted in large increases
328 (Tables 1a&b). This is consistent with field studies where comparable cured amendments
329 (composted wood and yard waste), did not result in increased CH_4 emissions (Winton and
330 Richardson, 2015), but straw (Ballantine et al., 2015) and peat bales (Green, 2014)
331 increased CH_4 emissions. Organic material is commonly cured, or composted, to remove
332 plant pathogens (Noble and Roberts, 2004) and to reduce the amount of cellulosic
333 material (Hubbe et al., 2010), which competes for oxygen, contributing to phytotoxicity

334 (Saidpullicino et al., 2007; Hu et al., 2011). Curing produces humic acids and increases
335 the nominal oxidation state (NOSC) of C (Guo et al., 2019). When cured material is then
336 subjected to anaerobic conditions, less CH_4 is produced (Yao and Conrad, 1999), which
337 would make composted material more suitable in a wetland restoration context.
338 Following soil inundation, we observed two distinct gas production phases (Phase
339 1 and 2). This pattern is difficult to distinguish in unamended soils but has been reported
340 previously (Yao and Conrad, 1999; Drake et al., 2009). Our breakpoint (5 – 45 days
341 Table 2b) was similar to Yao and Conrad (1999) (5 – 36 days). The Phase 2 rates in
342 unamended soils were also similar: 0.96 – 3.98 cm^3 CH_4^{-1} Kg $soil^{-1}$ day in Yao and
343 Conrad (1999) and 1.82 – 1.94 cm^3 CH_4^{-1} Kg $soil^{-1}$ day in our study (Table 2b).
344 There are several explanations that could account for the observed gas production
345 pattern. One is the lag period required to re-establish populations of methanogenic
346 archaea, which are likely dormant under oxic conditions and regrowth can be on the order
347 of days (Jabłoński et al., 2015). In our study, B had the earliest shift to Phase 2 CH_4
348 production (Table 2b), possibly due to elevated levels of dormant methanogens present
349 from anaerobic digestion. The two-phase gas production could also be due to depletion
350 of bioavailable Fe-oxides, thus relieving the competition between Fe reducers and
351 methanogens (Megonigal et al., 2004). Our data were mixed, with some treatments
352 showing evidence of competition by Fe reducers, but in other cases we did not see
353 competition. In treatment M1, for example, ferrous Fe in the supernatant plateaued at
354 about the same time as the breakpoint (Figure 4b), after which CH_4 production increased.
355 In contrast, in W3 soluble Fe continued to be produced well after the breakpoint, and the

356 amount of bioavailable Fe used during the course of the incubation was less than 28 ± 4%
357 (Figure 4b, Supplemental Table S2). In addition to quantifying Fe oxide concentrations,
358 the CO_2:CH_4 ratios can be indicative of interactions between methanogens and other
359 reducers (Bridgham et al. 2013). If Fe reduction or other reduction stops during Phase 2,
360 we would expect the CO_2:CH_4 ratio to be near 1:1 (Bridgham et al. 2013). However, we
361 observed notable exceptions. The SCL L1 treatment had a ratio of 73:1 in Phase 2 (Table
362 2b), yet still had the characteristic shift to higher overall gas production (4.67x). Other
363 treatments also had higher CO_2:CH_4 ratios: L3, L6, W1, B1, C, and W1-3 in the SL soil
364 (Table 2b). Our mixed observations may have been due to microsite formation. In high
365 producing microcosms, microsite development may have been disrupted by gas
366 ebullition, which was substantial enough in H amended trials to cause effervescence.
367 Amendments with low gas production and limited gas ebullition (e.g. L, W and C)
368 continued to produce Fe^{2+} after the breakpoint, possibly because methanogens were
369 active in undisturbed microsites, as described in Yang et al. (2017).
370 The increased gas production from organic amendments was more pronounced in
371 SL compared to SCL, where there was 2.4x higher CH_4 and 2.6x higher gas production
372 (Figures 2a & b). We observed a more pronounced effect than a recent rice field study
373 where there was more CH_4 from SL soils versus SCL, although in that study results were
374 not statistically significant (Kim et al., 2018). Yagi and Minami (1990) observed that
375 compost (approximate loading rate the same as our 1x treatment) increased respiration
376 rates by 1.8x in a SCL versus a loam soil. Maietta, Hondula, et al. (2020) observed that
377 respiration rates were higher in a sandy loam soil compared to a silty clay, with and

without 3.3% & 23% wetland hay amendments. Thus, we might conclude that in general coarser grained (sandy) soil textures emit more CH_4; however, there are a number of investigations where this was not the case (Yagi and Minami, 1990; Glissmann and Conrad, 2002). Other factors may have contributed. In our experiment the SCL had 7.6x dithionite extractable Fe, and 4.6x as much %C (Supplemental Table S1), so additional studies would be needed to isolate texture as the controlling factor.

We considered the gas production from H microcosms separately because they followed a different pattern than the other amendments, but the pattern was similar to other studies using hay (Glissmann and Conrad, 2002) and wetland hay (Maietta et al., 2020b). Our study adds to these findings by observing that H produced very low CH_4 in the water column (after floating) compared to being mixed with soil (Table 2c). This may merit further study because if this is generally true, applying fresh organic matter as a mulch, rather than mixed into the soil, could greatly reduce the adverse consequence of increased CH_4 emissions.

Reduction of Fe-oxides occurs in saturated soils in the presence of an organic substrate and is a key biogeochemical process in wetland soils. With sufficient time, hydric soils may develop redoximorphic features from Fe reduction; however, studies have not shown lasting redoximorphic development due to organic amendments (Gray, 2010; Ott et al., 2020). Organizations responsible for constructing mitigation wetlands have an interest in documenting Fe reduction prior to redoximorphic feature development as evidence soils that are hydric. Some mitigation wetland practitioners experience challenges meeting hydric soil testing standards. Although reports in the scientific

400 literature are rare, there are examples of sites meeting vegetation and hydrology wetland
401 indicators, but not hydric soils (Berkowitz et al., 2014). Both the soils we tested produced
402 sufficient Fe^{2+} and would have passed hydric soils tests, so a soil amendment would not
403 be needed.

404 We observed that fresh organic matter resulted in increased Fe^{2+} compared to
405 cured organic matter (Figure 3), likely due to the presence of labile carbon, allowing
406 access to more crystalline Fe-oxides (Lentini et al., 2012). In some soils, Fe-reducing
407 bacteria using fresh organic matter amendments could access crystalline Fe making it
408 more bioavailable. However, without an anoxic/oxic cycle, increased Fe^{2+} production
409 could lead to Fe^{2+} toxicity and ferrolysis (Kirk, 2004), similar to the way fresh organic
410 matter leads to SOC priming (Blagodatsky et al., 2010). Ferrolysis occurs when
411 bioavailable Fe-oxides are reduced to Fe^{2+} and are subject to hydraulic transport. We
412 observed that cured amendments, like L, lowered Fe^{2+} concentrations (Figure 3), possibly
413 due to the presence of humic acids that are generated during curing (Guo et al., 2019).
414 Humic acids often contain insufficient biogeochemical energy to drive dissimilatory Fe
415 reduction (Keiluweit et al., 2017), chelate Fe^{2+}, removing it from the liquid phase
416 (Catrouillet et al., 2014), and create insoluble precipitates (Shimizu et al., 2013).

417 Regulating Fe^{2+} production, through the selection of the appropriate OM
418 amendment, could influence the growth of wetland plants. For example, rice growth may
419 be stimulated under low Fe^{2+} doses of 1 mg/L (Müller et al., 2015), but higher doses can
420 produce detrimental Fe plaque (Pereira et al., 2014). Some native wetland species are
421 adapted to high Fe^{2+} concentrations. *Juncus effusus* growth is stimulated at 25 mg/L Fe^{2+}

422 (Deng et al., 2009). North American native reed *Phragmites australis* ssp. *americanus*
423 was stimulated at 11 mg/L Fe^{2+} from ferrous sulfate (Willson et al., 2017), but the
424 invasive Eurasian lineage of *Phragmites australis* seedling growth was inhibited by Fe^{2+}
425 as low as 1 mg/L (Batty, 2003). Soils high in free Fe^{2+} adversely affected *P. australis*
426 growth by creating an Fe-oxide plaque on roots (Saaltink et al., 2017).

427 Our results show that pH has a significant effect on both the production of Fe^{2+}
428 (Figure 3) and CH_4 (Table 3). Between pH 5.6 and 6.6, the lower pH produced more Fe^{2+}
429 and less CH_4, consistent with thermodynamic predictions (Ye et al., 2012).
430 Hydrogenotrophic methanogens can maximize CH_4 production at pH 5 (Bräuer et al.,
431 2004). In rice paddy soils, CH_4 emissions had a clear peak at pH 7, but almost none
432 below pH 5.5 (Wang et al., 1993). The strong effect of pH underscores the need to take
433 this parameter into account when interpreting data from experiments evaluating Fe-
434 reduction and methanogenesis. Attempting to control the pH of soils could potentially
435 introduce confounding effects. We used an MES buffer with 10x the quantity we
436 estimated from a soil titration and still saw shifts in the pH after incubation. With a high
437 residual soil acidity, the amount of buffer needed to control soil pH may increase the
438 ionic strength to a level that could influence cellular sorption to mineral and Fe-oxide
439 surfaces (Mills et al., 1994) as well as enzyme activity (Leprince and Quiquampoix,
440 1996).

441 **5 Implications**

442 In our experiment, we observed that organic amendments can increase CH_4
443 production, particularly after extended anaerobic periods. We quantified CH_4 production

potential from several organic amendments, and in a separate field experiment (unpublished) show that these results are useful in predicting field CH_4 production. There is mounting concern that CH_4 from restored and created wetlands may result in net global warming for decades to centuries (Neubauer, 2014). Our results suggest that not only do organic amendments increase CH_4 gas production overall, but uncured amendments can also decrease the time it takes before there is a large increase in both total gas production and CH_4. Methane production is not constant and dramatically increases after several weeks. Because of this, it may be beneficial to report wetland CH_4 data along with inundation duration, which can strongly affect CH_4 (Hondula et al., 2021). It may be possible to limit CH_4 in many wetland settings, particularly mitigation wetlands where hydrology is part of the design: shorter flooding or inundation durations with alternating drier conditions. This strategy has been proposed for rice paddy fields (Souza, 2021). Our lab study demonstrates the potential for significant CH_4 emissions, but in a real system, methanotrophic activity could attenuate some of the emissions (Chowdhury and Dick, 2013); however, this would not decrease the overall C loss from soils, it only changes the pathway. If organic amendments are to be used, cured amendments may be preferrable because they are not as prone to high CH_4 generation and may attenuate Fe^{2+} toxicity. Amendments that lower the soil pH increase Fe reduction and limit methanogenesis (Marquart et al., 2019). When deciding whether or not to use organic amendments for wetland mitigation consideration should be given to whether or not the material has been cured, the pH, the soil texture, and expected hydroperiod.

Figures

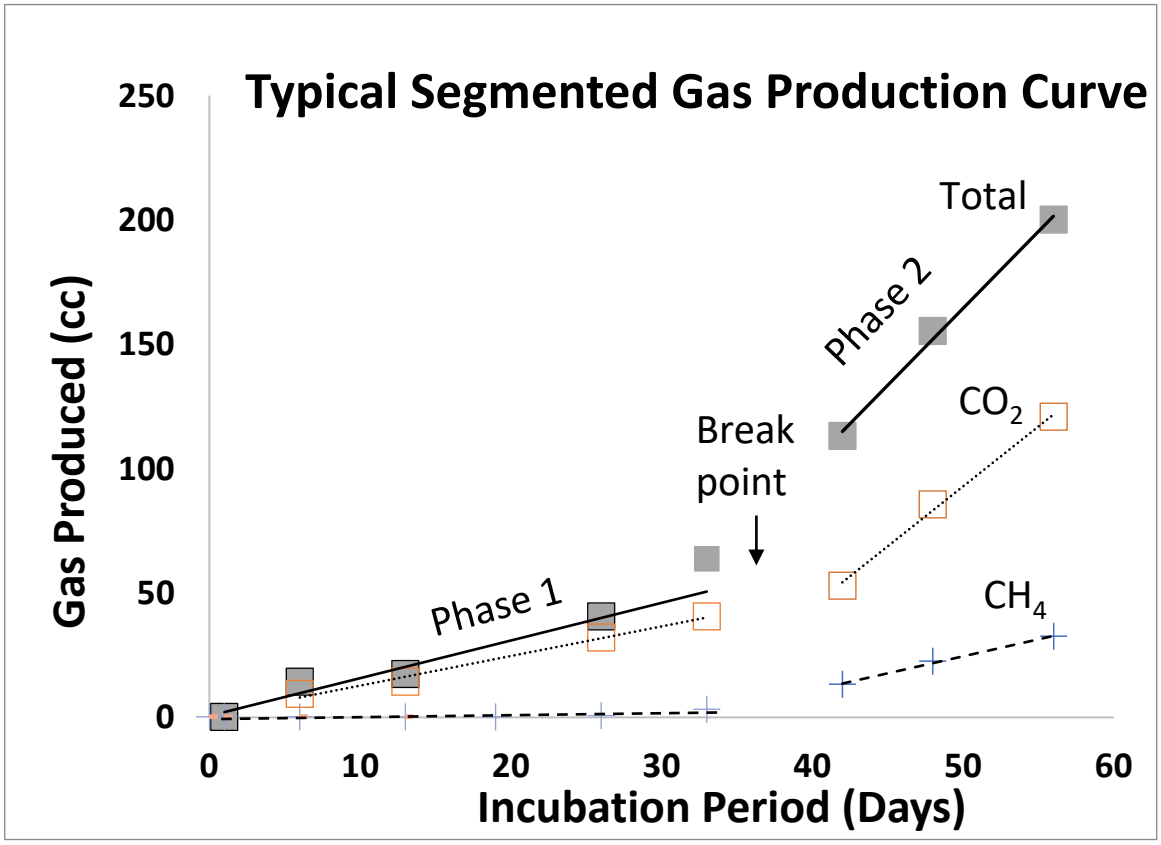


Fig. 1 – Typical gas production in saturated soils amended with organic matter (All Experiments).

Gases were best modeled using a segmented linear function. After a breakpoint the average total gas production increased by a factor of ~5 whereas there is a sharp (>> 5x) increase in methane production. Data presented is from the manure (1x) amended trials in sandy clay loam soil. Note that hay amended trials exhibited a typical sinusoidal pattern shown in Supplemental Figures S2&3h,i,j).

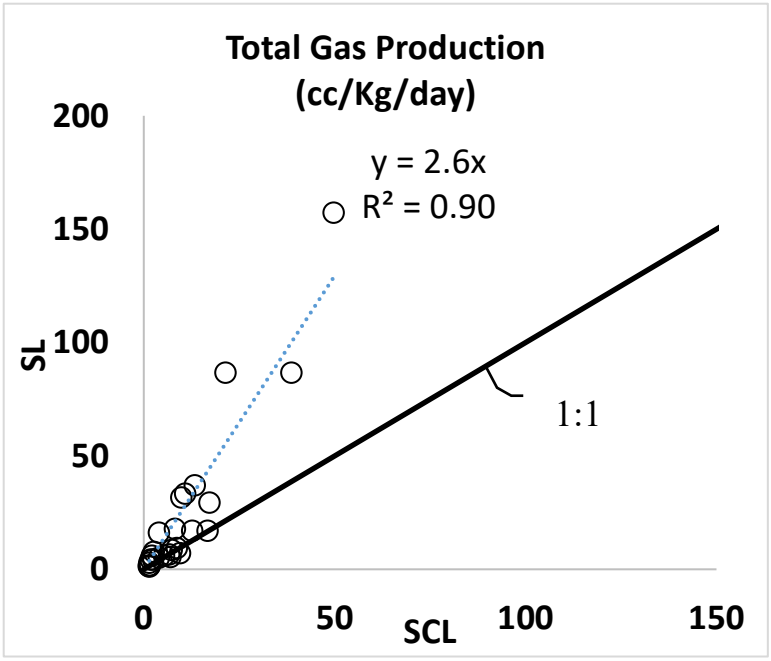


Fig. 2a. Experiment 1. Total biogenic gas production rate in the SL soil versus the SCL mesocosms. The SL mesocosms had, on average, 2.6 times higher gas production than the SCL.

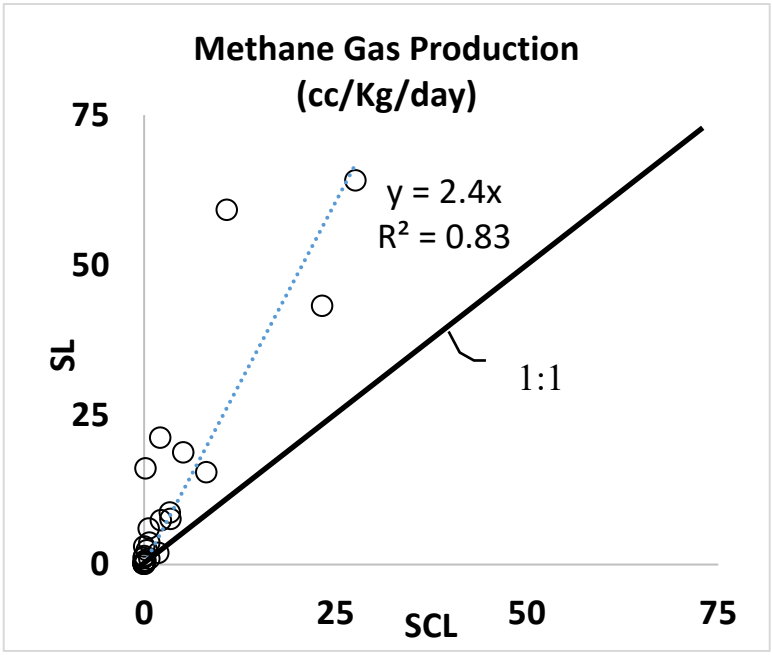


Fig. 2b. Experiment 1. Biogenic methane gas production rate in the SL soil versus the SCL mesocosms. The SL mesocosms had, on average, 2.4 times higher methane gas production than the SCL.

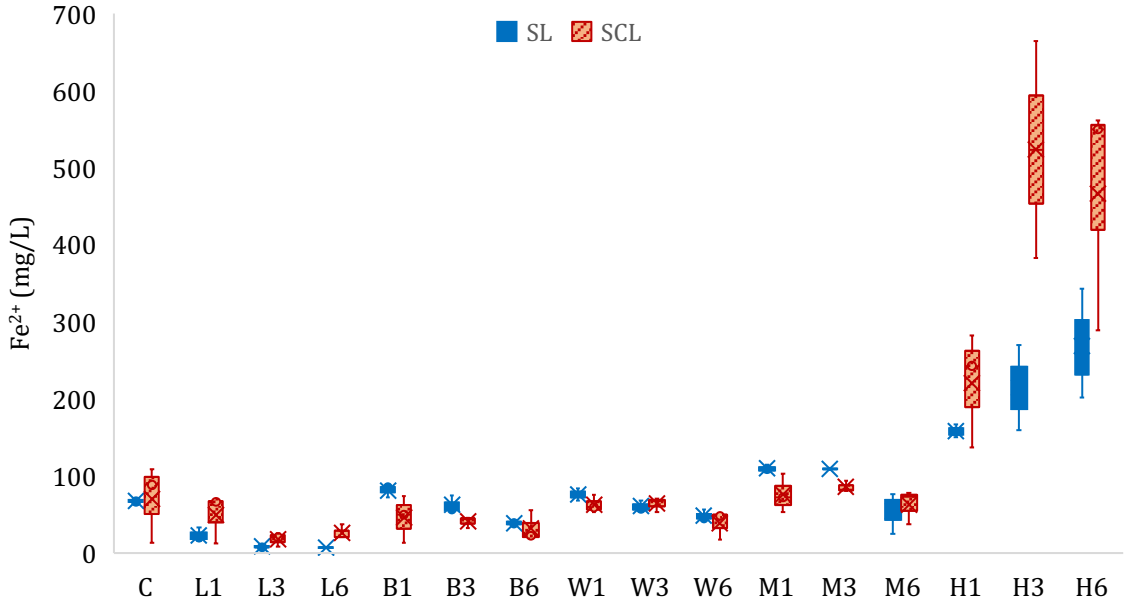
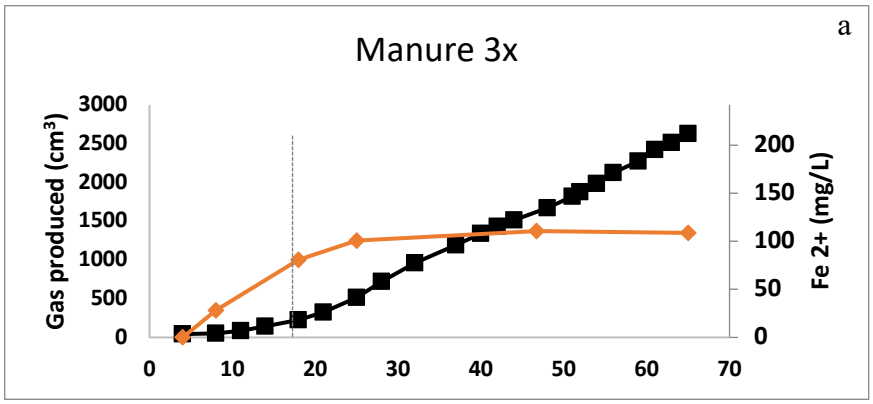


Fig. 3 – (Experiment 1b) Ferrous iron (Fe^{2+}) concentration in the liquid phase at the end of the incubation period.

Microcosms receiving different organic amendment types and levels in Sandy Clay Loam (SCL) and Sandy Loam (SL) soils. C = no amendment control, L = LeafGro (yard waste), B = biosolids, W = wood chips, M = manure, H = hay. Numbers signify treatment level (1, 3, or 6 times amount of organic matter equivalent to 60 yd^3 / acre to a depth of 6 inches). Different lower-case letters signify differences ($p < 0.05$) based on contrasts compared to C and brackets signify all results in the bracketed group were not statistically different. H increased total Fe^{2+} production compared to the C in both soils, and L decreased total Fe^{2+} production compared to C (SL only).



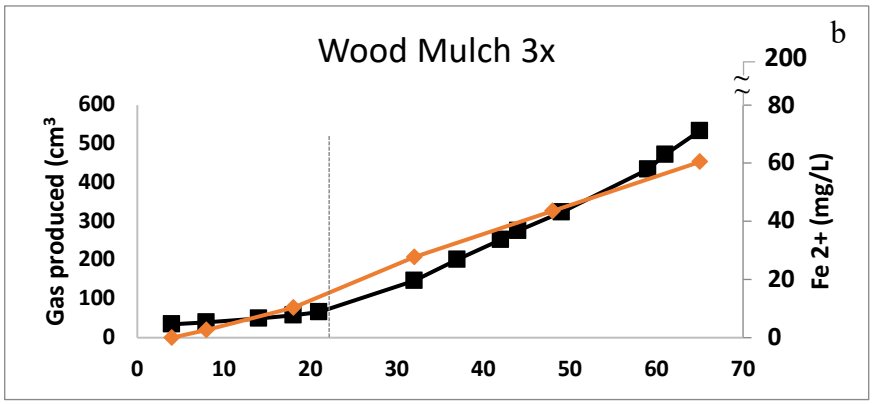
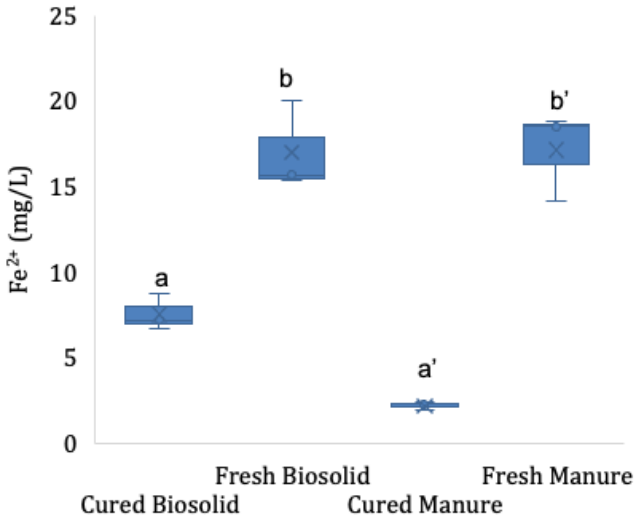


Fig. 4 – (Experiment 1b) Ferrous iron (Fe^{2+}) and methane (CH_4) in selected microcosms. Depletion of Fe coincided with the breakpoint (dashed line) with manure, but not with wood mulch. Other examples of this pattern are shown in Supplemental Figure S6. The maximum value on the secondary x-axis is the maximum expected Fe^{2+} concentration based on the HHCL extraction.



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514

515 Fig. 5 – (Experiment 2b) Ferrous iron (Fe^{2+}) concentration in the liquid phase at the end

516 of the incubation period (13 days).

517 Incubation was carried out in sandy loam soil. Different letters indicate a difference at

518 $p<0.001$.

519

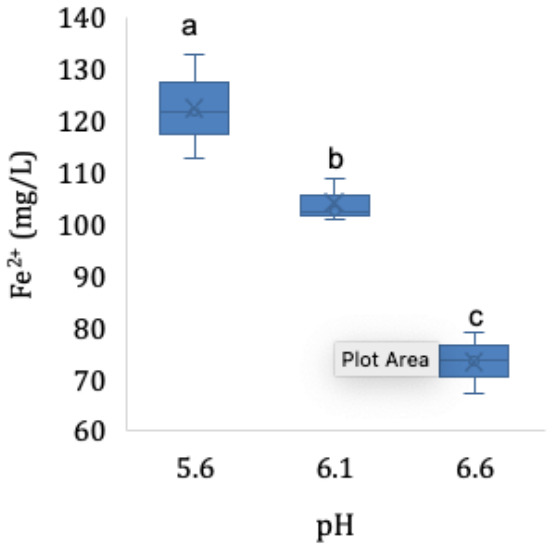

Fig. 6 – (Experiment 3) Ferrous iron (Fe^{2+}) concentration in the liquid phase with varied pH in microcosms receiving hay in Sandy Loam soils.

Different letters indicate a difference at $p<0.05$.

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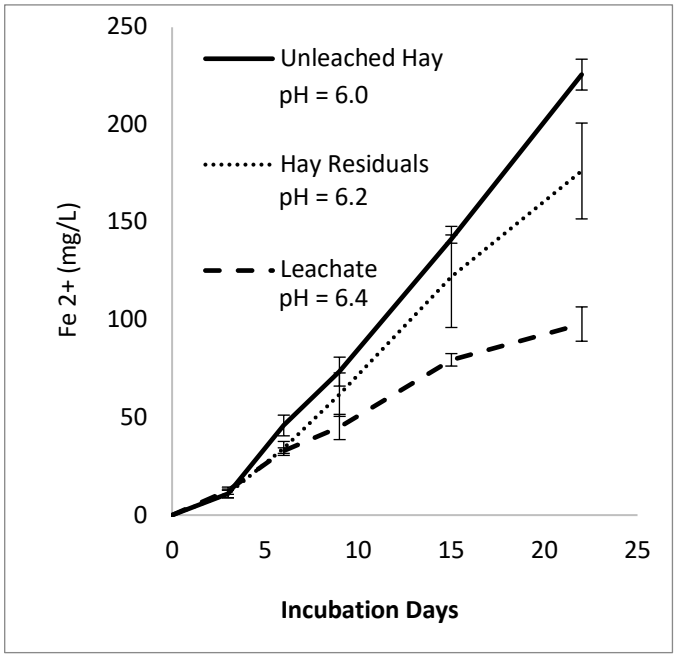

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530

531 Fig. 7 – (Experiment 4) Ferrous iron (Fe^{2+}) concentration in the liquid phase with hay as

532 substrate.

533

534

Tables

Table 1 – Summary of Results

Treatment	Effect		
	Iron Reduction	Methane	Breakpoint
Organic Matter	↔	↑	↓
Increased Dose	↘	↑	↓
Composting/Curing	↓	↓	↑
Decreased pH	↑	↓	N/A
SL vs SCL	↓	↑	N/A
Soluble vs. particulate OM	↔	N/A	N/A

Breakpoint = time to increased methane production

SL = Sandy Loam, SCL = Sandy Clay Loam

↔ No change, ↑ increase, ↓ decrease, ↘ slight decreasing trend

Table 2a – (Experiment 1a – Phase 1). Carbon dioxide (CO_2), methane (CH_4) and total gas production. Organic amendment types: B (biosolids), M (manure), L (composted yard waste), W (composted wood chips) and levels (1 = 60 yd^3 / acre equivalent: 3 = 180; 6 = 360) in silty clay loam (SCL) and sandy loam (SL) soils. Instances where organic amendments did not increase CH_4 production are bolded. Note: CO_2 : CH_4 ratios are based on calculated gas production rates, not total gas produced.

			CO_2		CH_4		Total Gas		
Soil	Treatment	Soil (g)	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	CO_2:CH_4
SCL	Control	621.63	0.97	1.56	0.002	0.003	0.99	1.59	520.0
SCL	B1	425.24	1.53	3.61	0.08	0.18	4.13	9.70	20.1
SCL	B3	544.53	1.50	2.76	0.44	0.80	3.85	7.06	3.5
SCL	B6	468.02	2.09	4.46	0.06	0.13	3.53	7.55	34.3
SCL	M1	583.40	0.74	1.27	0.02	0.04	1.33	2.27	31.8
SCL	M3	495.56	1.79	3.61	0.32	0.64	2.05	4.13	5.6
SCL	M6	394.39	1.49	3.77	0.12	0.30	4.35	11.03	12.6
SCL	L1	586.46	0.83	1.42	0.001	**0.001**	0.85	1.45	1420.0
SCL	L3	516.34	0.89	1.72	0.01	0.01	0.91	1.77	172.0
SCL	L6	410.17	0.67	1.63	0.04	0.09	0.80	1.95	18.1
SCL	W1	593.36	1.00	1.68	0.01	0.01	0.92	1.56	168.0
SCL	W3	539.61	0.98	1.81	0.10	0.19	1.39	2.58	9.5
SCL	W6	457.42	1.03	2.25	0.11	0.24	1.29	2.81	9.4
SL	Control	634.60	0.50	0.79	0.03	0.04	0.56	0.88	19.8
SL	B1	606.80	1.25	2.06	0.02	**0.04**	4.13	6.80	51.5
SL	B3	551.50	1.57	2.84	0.44	0.79	2.92	5.29	3.6
SL	B6	467.87	2.08	4.44	0.59	1.27	3.81	8.15	3.5
SL	M1	619.92	2.62	4.22	0.58	0.93	3.49	5.63	4.5
SL	M3	588.37	4.48	7.61	3.44	5.85	9.42	16.02	1.3
SL	M6	540.93	8.63	15.95	8.59	15.87	17.92	33.13	1.0
SL	L1	600.10	0.35	0.58	0.02	**0.03**	0.73	1.22	19.3
SL	L3	530.30	0.61	1.15	0.02	**0.03**	0.78	1.46	38.3
SL	L6	425.87	0.62	1.47	0.11	0.26	1.66	3.89	5.7
SL	W1	603.27	0.98	1.62	0.06	0.10	1.55	2.56	16.2
SL	W3	538.77	1.42	2.64	0.20	0.36	2.14	3.98	7.3
SL	W6	442.57	3.05	6.88	0.24	0.54	3.23	7.31	12.7

Table 2b – (Experiment 1a – Phase 2). Carbon dioxide (CO_2), methane (CH_4) and total gas production, and the Phase 1 : Phase 2 breakpoint. Organic amendment types: B (biosolids), M (manure), L (composted yard waste), W (composted wood chips) and levels (1 = 60 yd^3 / acre equivalent: 3 = 180; 6 = 360) in silty clay loam (SCL) and sandy loam (SL) soils. Instances where organic amendments did not increase CH_4 production are bolded. Note: r^2 values represent the combined best fit curve, using triplicate samples, for Phase 1 (Table 1a) and Phase 2.

		CO_2		CH_4		Total Gas					
Soil	Treatment	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	cm^3/day	cm^3/Kg/day	CO_2:CH_4	Break Point	r^2	Ph 2: Ph1
SCL	Control	2.06	3.31	1.20	1.94	2.54	4.09	1.7	40.0 ± 4.5	0.959	2.57
SCL	B1	5.58	13.13	1.47	3.45	5.49	12.91	3.8	29.3 ± 1.9	0.987	1.33
SCL	B3	3.74	6.86	4.45	8.17	9.48	17.40	0.8	20.1 ± 3.4	0.974	2.46
SCL	B6	7.42	15.85	10.90	23.29	18.20	38.89	0.7	10.3 ± 2.4	0.994	5.15
SCL	M1	2.26	3.88	1.29	2.22	5.82	9.97	1.7	40.2 ± 2.1	0.997	4.39
SCL	M3	4.64	9.37	5.39	10.89	10.69	21.58	0.9	20.8 ± 0.8	0.997	5.23
SCL	M6	5.85	14.83	10.91	27.67	19.69	49.93	0.5	22.1 ± 3.2	0.956	4.53
SCL	L1	3.85	6.57	0.05	**0.090**	3.96	6.76	73.0	32.2 ± 1.6	0.966	4.67
SCL	L3	4.21	8.16	0.39	**0.75**	4.54	8.79	10.9	32.0 ± 2.2	0.983	4.97
SCL	L6	5.90	14.39	0.92	2.24	6.95	16.95	6.4	32.0 ± 3.7	0.923	8.68
SCL	W1	1.56	2.63	0.27	**0.460**	3.22	5.42	5.7	34.0 ± 3.7	0.986	3.48
SCL	W3	1.93	3.58	1.90	3.52	4.51	8.35	1.0	24.2 ± 3.1	0.989	3.23
SCL	W6	2.19	4.79	2.36	5.15	6.22	13.60	0.9	13.0 ± 2.4	0.981	4.84
SL	Control	1.00	1.58	1.16	1.82	3.11	4.91	0.9	40.0 ± 3.2	0.957	5.55
SL	B1	4.44	7.31	5.16	8.50	10.19	16.79	0.9	8.6 ± 3.0	0.880	2.47
SL	B3	8.76	15.89	8.42	15.28	16.12	29.23	1.0	4.7 ± 1.8	0.989	5.53
SL	B6	12.61	26.96	20.15	43.07	40.39	86.33	0.6	9.1 ± 1.2	0.992	10.59
SL	M1	8.64	13.93	13.03	21.02	19.41	31.30	0.7	16.7 ± 0.7	0.998	5.56
SL	M3	15.23	25.88	34.77	59.10	50.79	86.33	0.4	17.2 ± 1.5	0.992	5.39
SL	M6	29.50	54.53	34.62	64.00	84.92	156.98	0.9	29.4 ± 1.4	0.974	4.74
SL	L1	1.35	2.24	1.71	2.85	3.76	6.26	0.8	38.3 ± 1.2	0.992	5.12
SL	L3	2.27	4.27	1.86	3.50	4.82	9.09	1.2	40.5 ± 2.0	0.977	6.22
SL	L6	4.25	9.99	3.07	7.21	7.15	16.78	1.4	44.8 ± 1.3	0.988	4.31
SL	W1	2.10	3.48	1.32	2.19	3.47	5.76	1.6	25.6 ± 7.6	0.762	2.25
SL	W3	6.58	12.22	4.05	7.51	9.46	17.56	1.6	23.2 ± 2.3	0.974	4.41
SL	W6	10.10	22.83	8.23	18.60	16.22	36.65	1.2	23.2 ± 1.1	0.991	5.02
									AVERAGE		4.7
									STDEV		1.9

Table 2c – Experiment 1a. Carbon dioxide (CO_2), methane (CH_4) and total gas production with hay (H) amendment. H amended trials fit a sigmoidal, not segmented, pattern, and therefore there was no breakpoint and we present p values for the sigmoidal fit, except H6 SL rates where we used a power function in Excel and report the r^2 value. Gas production rates (cm 3 gas Kg soil -1 day -1) represent maximum at the inflection point. The amendment floated to the surface in the SCL H3 and H6 trials, which resulted in unusually low CH_4 production rates.

Sigmoidal curve values			CO_2			CH_4			Total Gas			
Soil	Treatment	Soil (g)	cm^3/day	cm^3/Kg/day	p	cm^3/day	cm^3/Kg/day	p	cm^3/day	cm^3/Kg/day	p	CO_2:CH_4
SCL	H1	573.03	9.70	16.93	2.0E-16	10.40	18.15	0.164	37.1	64.75	1.3E-12	0.93
SCL	H3	477.85	7.50	15.70	3.0E-14	0.02	0.04	0.933	9.90	20.72	7.8E-6	393
SCL	H6	334.20	6.60	19.75	0.019	0.09	0.27	0.921	6.70	20.05	9.6E-13	73
SL	H1	582.57	8.90	15.28	5.5E-14	16.20	27.81	0.283	18.40	31.58	2.9E-4	0.55
SL	H3	478.00	20.80	43.51	1.8E-13	12.20	25.52	0.636	36.80	76.99	0.0093	1.7
SL	H6	321.13	50.71	158.0	0.93(r^2)	77.7	242.1	0.69(r^2)	79.79	248.47	0.74(r^2)	0.65

Table 3 – (Experiment 2a). Methane gas data for incubations with fresh and cured organic matter in sandy loam soil.

Control data (*) from Experiment 1a (Table 2a) included for reference. Different letters indicate a difference at $p<0.001$.

	Phase 1	Phase 2
Treatment	Methane (cm^3/Kg/day)	Methane (cm^3/Kg/day)
Control*	0.04	1.8
Cured Biosolids[a]	0.003	0.37
Fresh Biosolids[b]	3.29	17.48
Cured Manure[a']	0.22	5.4
Fresh Manure[b']	3.85	42.36

Table 4 – (Experiment 3). Methane gas data versus pH.

Microcosms receiving hay in Sandy Loam soils (Experiment 3). Different letters indicate a difference at $p<0.001$.

pH	Phase 1 CH_4 (cm^3/Kg/day)	Phase 2 CH_4 (cm^3/Kg/day)
5.6[a]	0.44	10.6
6.1[b]	1.0	13.0
6.6[c]	1.8	13.8

Declarations

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Conflicts of interest/Competing interest

Authors declare no conflict of interest

Availability of data and material

Significant data detail is available in the supplementary materials. Additional raw data available upon request.

Code availability

None

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