



1	Methane accumulation affected by particulate organic carbon in upper Yangtze
2	deep valley dammed cascade reservoirs, China
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39 Abstract

40 Methane (CH_4) emissions from freshwaters to the atmosphere have a profound impact on global atmospheric greenhouse gas (GHG) concentrations. Anthropogenic 41 42 footprints such as dam construction and reservoir operation significantly changed the fate and transport of CH₄ in freshwaters. The type of particulate organic carbon (POC) 43 in reservoirs is a critical factor controlling CH₄ production and emissions. However, 44 45 little is known of how reservoir operation mediates the distribution of POC and regulates CH₄ accumulation in cascade hydroelectric reservoirs. Here, spatial and 46 temporal variations in POC and CH4 were explored in the Xiluodu (XLD) and 47 Xiangjiaba (XJB) reservoirs which are deep valley dammed cascade reservoirs located 48 in the main channel of the upper Yangtze River. Based on the δ^{13} C-POC and N/C mole 49 ratios of particulate organic matter, the results of multi-endmember stable isotope 50 51 mixing models by a Bayesian model show that terrestrial POC and autochthonous POC accounted for approximately $56 \pm 19\%$ and $42 \pm 19\%$ (SD, n=181) of POC, respectively. 52 CH₄ concentrations and δ^{13} C-CH₄ in the cascade reservoirs were potentially influenced 53 54 by CH₄ oxidation. Together with other physicochemical parameters and structural 55 equation model, these results suggested that the input of terrestrial POC was dominantly influenced by water level variations and flow regulation due to reservoir operation. The 56 cumulative effect of POC caused by cascade reservoirs was not apparent at a bimonthly 57 58 scale. Terrestrial POC was more likely to dominate CH₄ accumulation in cascade reservoirs under reservoir operation. 59





60 1 Introduction

61 Methane (CH4) is widely recognized as the second most important greenhouse gas after carbon dioxide (CO_2) (Saunois et al., 2020). The latest data shows that global 62 atmospheric CH4 was 1888.5 ppb in July 2021 (Dlugokencky, 2021). The annual 63 increase in global atmospheric CH₄ between 2007 and 2020 fell within the range of 7.99 64 ppb·yr⁻¹ to 14.81 ppb·yr⁻¹ (Dlugokencky, 2021). Global CH₄ emissions were estimated 65 to be up to 579 Tg. yr⁻¹ from 2008 to 2017 (Saunois et al., 2020). There is a very high 66 67 level of confidence that the atmospheric CH₄ increase during the Industrial Era was caused by anthropogenic activities, which caused approximately 60% CH₄ emissions 68 (Ciais et al., 2014; Saunois et al., 2016; Saunois et al., 2020). However, not all the 69 70 sources of global CH₄ emissions are explicitly and well explained. Although half of the global methane emissions come from the aquatic ecosystems (Rosentreter et al., 2021), 71 72 a large proportion of the uncertainties in the global CH₄ budget arise from freshwater systems. The production of CH₄ in lakes and reservoirs is an important process in the 73 global methane cycle. This is partly because freshwater systems are closely linked to 74 and manipulated by anthropogenic activities, e.g., hydrological process regulation, 75 geomorphological alternation, large inputs of organic carbon, and nutrients from 76 surrounding communities. Anthropogenic footprints significantly change the fate and 77 transport of CH₄ in freshwaters. 78

79 Dam construction and reservoir impoundments are widely accepted as important 80 anthropogenic activities that significantly change the sink and sources of CH₄ in the 81 freshwater systems. In general, the net change in CH₄ emissions in reservoirs is 82 primarily contributed by the decomposition of organic matter (OM), e.g., soil organic 83 carbon and vegetation cover, due to flooding. Reservoir CH4 emissions may reach the highest levels immediately after impoundment and exponentially decline with aging 84 85 (Abril et al., 2005). Second, the reduced flow velocity and increased hydraulic retention time (HRT) in the reservoir accumulates terrestrial OM from the upstream watershed to 86 the reservoir bottom, supporting the development of anoxic habitats at the reservoir 87





bottom for intensive methanogenesis. Site-specific hydro-morphological characteristics, reservoir thermal regimes, and external OM loads from the upstream watershed supported consistent reservoir CH₄ emissions. This evidence supported a wide acceptance that reservoirs consistently emit CH₄ that exceeds those of upstream reaches of the same river and, on average, from natural lakes (Stanley et al., 2016).

However, the role of dams and reservoirs in the global CH4 budget has been 93 94 received challenges in recent years. Previous studies reported that global reservoirs were either CH₄ neutral or CH₄ sources and that reservoir productivity and temperature 95 are better predictors than reservoir age (Deemer et al., 2016). Reservoirs may reduce 96 CH₄ and CO₂ emissions in downstream floodplain wetlands caused by upstream organic 97 carbon (OC) transport and sedimentation (Muller, 2019). Sediment loads and nutrient 98 enrichments were the primary and secondary driving factors that regulate CH₄ 99 formation in navigable river impoundments (Wilkinson et al., 2019). While river 100 101 impoundments did not account for all hotspots of CH₄ formation in their study, they implied that sediment management would offset CH4 emissions from damming rivers 102 (Wilkinson et al., 2019). The largest share of CH₄ emissions may be due to OC retention 103 and mineralization in the reservoir (Shi et al., 2017). Recent research showed that 104 105 imbalanced stoichiometric sedimentation mediated CH₄ accumulation in the mid-part of China's Three Gorges Reservoir (Li et al., 2020a). However, the findings supported that 106 107 reservoir operation strategy significantly impacts the patterns of CH₄ accumulation. 108 Growing research has highlighted the importance of reservoir hydrology and sediment 109 (or OC) dynamics contributing to CH₄ accumulation and emissions. If reservoir 110 operation could be proven to have an evident link with CH4 dynamics and emissions, 111 there would be best practices for the hydropower industry to reduce excessive CH₄ emissions, which is meaningful for mitigating the global warming potential of 112 hydropower. However, the question of how reservoir operation mediates particulate 113 114 organic carbon (POC) transport and finally impacts CH4 dynamics and emissions from reservoir appears to be sophisticated and site-specific and has not been well addressed. 115

116 The study was extended from a single reservoir to a cascade system in the upper





- 117 Yangtze River to explore dissolved CH₄ accumulation under reservoir operation. The
- 118 research objectives are (1) to determine the input and accumulation of POC that
- 119 influence dissolved CH₄ accumulation under the physiochemical parameters caused by
- 120 cascade reservoir operation and (2) to explain how the different sources of POC regulate
- 121 the dissolved CH₄ accumulation in deep river-valley dammed reservoirs.

122 **2 Methods**

123 2.1 Site description and sampling campaign

124 The Xiluodu (XLD) reservoir and Xiangjiaba (XJB) reservoir are two deep 125 river-valley dammed cascade reservoirs located along the main channel of the upper 126 Yangtze River, which is frequently referred to as the "Jinsha River" (Figure 1). Both the XLD and XJB hydro-projects serve as hydroelectricity production facilities. They also 127 128 perform significant seasonal water level adjustments for flood control as their partial 129 services (Figure 2). The initial impoundment of both reservoirs started in July 2013. XJB finished its impoundment in September 2013, while XLD finished its full 130 131 impoundment one year later. The parameters of the XLD and XJB reservoirs are shown 132 in Table S1 (Li et al., 2017b).

133 The sampling campaign was conducted every other month between January 2018 and January 2019. Fifteen sampling sites were located along the main channel of both 134 reservoirs, L3 to L1 in the XLD reservoir, B10 to B1 in the XJB reservoir and X2 to X1 135 downstream of the XJB reservoir (Figure 1). Because it was not possible to cover all the 136 137 sampling sites and finish sampling work in one day, each sampling event was limited 138 between the dates of the 10th and 15th of that month. The sampling time of a day was 139 controlled between 8:00 AM and 6:00 PM. In particular, the inconvenience of local 140 transportation in such deep valley areas along the XLD reservoir limited sampling to 40 km upstream of the XLD dam, encompassing sampling sites L3 to L1. Downstream of 141 142 the XJB, to avoid tributary disturbance to collected samples, the sampling work was 143 limited within 10 km downstream of the XJB dam, encompassing sampling sites X2 and 144 X1.

145 10 L water samples were collected at 0.5 m below the water surface and





approximately 2 m above the sediment layer at each sampling site. However, in the river reach right below both dams, where sites B10 to B8 and X2 to X1 were located, the water column was well mixed and only surface water samples were collected. These samples were then treated as representative of both surface and bottom samples in such a fully mixed water column. The water temperature, dissolved oxygen (DO) and pH were measured *in situ* by a calibrated YSI® Pro 2030 probe (YSI Inc., Ohio, USA).

152 Due to the different operation schemes of each reservoir, HRT was the fundamental key variable structuring the aquatic ecosystem of both reservoirs and was 153 distinctive between the XLD and XJB reservoirs. A detailed calculation of HRT is 154 155 provided in the Supporting Information (Section S1.1). Considering 2018 as an example (Figure S1), the annual average cumulative HRT in the XLD reservoir was 32.9 ± 20.6 156 days (mean \pm SD), with the 1st and 3rd quartiles between 14.7 days and 46.4 days. In 157 the XJB reservoir, the corresponding data collected in 2018 was 15.8 ± 8.6 days, with 158 159 1st and 3rd quartiles between 6.9 days and 23.4 days. The spatial and temporal 160 variations in the cumulative HRT in the XLD and XJB reservoirs also showed that the cumulative HRT was positively correlated with the water level and negatively 161 correlated with the flow and shortest in the flood season (Figure S2). 162









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Figure 1. The Xiluodu (XLD) and Xiangjiaba (XJB) cascade reservoirs. A) the map of the two reservoirs and sketch of sampling sites; B) vertical profiles of the cascade reservoirs and location of the sampling sites.









172 2.2 Measurement of CH₄ and CO₂ concentrations and other environmental

173 parameters

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CH₄ and CO₂ concentrations in the water phase were analyzed by the headspace 174 approach (Unesco/Iha, 2010). Equipped with a Picarro[®] G2201-*i* Isotopic Analyzer 175 (California, USA), our standard operation procedure was upgraded to the CH4 and CO2 176 177 headspace approach in 2017. Water samples were carefully collected by a 300 mL 178 polypropylene syringe barrel and sealed underwater by its matching piston. Then, surplus water and possible air were ejected out carefully until the water sampling was 179 180 200 mL in the syringe. Then, 100 mL of highly purified nitrogen was sucked into the 181 syringe and shaken immediately on-site for 2 minutes for gas exchange. Then, 100 mL gas samples were injected into pre-purified air sampling bags (0.3 L, HEDE tech, 182 Dalian, China) for further analysis by Picarro® G2201-i Isotopic Analyzer under HP 183 mode. Triplicates were performed for quality control. 184

185 For dissolved nutrients, water samples were filtered through pretreated (combusted





at 450 °C for 4 h in a muffle furnace and weighed after cooling) Whatman[®] GF/F glass
fiber membranes (Whatman[®], UK). Dissolved organic carbon (DOC) was then analyzed
by a Shimadzu[®] TOC-V TOC analyzer (Shimadzu, Japan). Dissolved total nitrogen
(DTN) and total phosphorus (DTP) were analyzed by the standard method (Wef and
Apha, 2005).

For particulate matter, the chlorophyll a (Chl-a) concentration was measured 191 192 spectrophotometrically after 90% cool acetone extraction. All residues on the GF/F 193 glass fiber membranes were dried at 65 °C for 48 h and reweighed. The concentration of total particulate matter (TPM) in the water column was the quotient of the mass 194 195 difference between the two weights and the volume of water samples used infiltration. The 65 °C dried fresh residues were also used for elemental composition analysis of C, 196 and N. POC, particulate organic nitrogen (PON), δ^{13} C-POC and δ^{15} N-PON were 197 measured by a Thermo Fisher ® Flash H T Elemental Analyzer for Isotope Ratio MS 198 (Thermo Fisher Scientific, MA, USA). The standard reference materials were Vienna 199 Pee Dee Belemnite for carbon and atmospheric N2 for nitrogen. Triplicates were also 200 201 performed for quality control.

202 2.3 Application of stable isotope mixing models

Fits Stable Isotope Mixing Models (SIMMs) were meant as a longer-term replacement to the previously widely used package SIAR (Stable Isotope Analysis in R) (Parnell et al., 2010). SIMMs were used to infer dietary proportions of organisms consuming various food sources from observations on the stable isotope values taken from the organisms' tissue samples. The proportional contribution of different sources of OM to a mixture was estimated by using stable isotope mixing models " SIMMs ", which are based on Bayesian methods (Parnell et al., 2010; Parnell et al., 2013).

The newest Simmr package of stable isotope mixing models was implemented in the R program (Version: 4.0.3) to estimate the contribution of different sources of POC. The contributions of different endmembers to POC were calculated by the δ^{13} C, δ^{15} N and N/C molar ratios of particulate organic matter (POM). To estimate the contribution





- of POC sources, the endmember values of C3 and C4 plants were set from references (Kendall et al., 2001; Wu et al., 2007; Jiang and Ji, 2013; Wang et al., 2014; Chen et al., 2018b; Deng et al., 2018; Xuan et al., 2019; Ru et al., 2020), and the endmember values of soil and plankton from the upper Yangtze River were sampled and analyzed in the field sampling campaign (Table 1). The 50% quantile values were the predicted values of the model. The contribution of terrestrial POC was obtained by C3 and C4 plants, coastal soil, and plankton as the contribution of autochthonous POC.
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Table 1. δ¹³C, δ¹⁵N, N/C molar ratios of particulate organic matter (POM) from different endmembers

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Endmember	δ ¹³ C-POM (‰)	δ ¹⁵ N-POM (‰)	N/C
C3 plant	$\textbf{-28.9} \pm 1.6$	1.8 ± 3.7	0.038 ± 0.019
C4 plant	$\textbf{-13.4} \pm 1.0$	2.4 ± 5.7	0.038 ± 0.019
Soil	$\textbf{-21.0}\pm2.0$	4.6 ± 1.4	0.134 ± 0.022
Plankton	$\textbf{-26.8}\pm0.6$	6.6 ± 1.8	0.184 ± 0.011

223 2.4 Data analysis

The structural equation model was performed using the Lavaan package in the R program to obtain the effects of different variables by verifying the theoretical model. Since the original data difficultly satisfied the normal distribution tested by Kolmogorov–Smirnov test, the maximum likelihood relaxation algorithm (MLR) was used as the estimator.

Data analyses and plotting (Kolmogorov–Smirnov test, linear regression, and oneway analysis of variance [ANOVA]) were performed using OriginPro® 2018 (OriginLab Corporation, MA, USA) and Statistical Product and Service Solutions (SPSS). In one-way ANOVA, differences between means and 50% quantile values were considered significant at p < 0.05. A detailed description of the mass balance approach of the POC and CH₄ in both reservoirs is provided in the Supporting Information (Section S1.1).





236 **3 Results**

237 **3.1 Physical limnology and major environmental parameters**

The water temperature ranged from 12.45 °C to 25.9 °C and 13.6 °C to 238 26.3 °C in the XLD and XJB reservoirs, respectively. The thermal stratification 239 patterns of both reservoirs are different (Figure S3A). Thermal stratification developed 240 241 in the XLD reservoir in spring, e.g., March, and persisted throughout the whole summer. Metalimnion existed approximately 60 m below the water surface. With the increase in 242 243 water level due to reservoir impoundment, together with the decrease in air temperature, thermal stratification gradually disappeared in November. In the XJB reservoir, weak 244 245 thermal stratification was initiated in March, which supported its spring algal blooms (Figure S3C). However, because the upstream reservoir increased its discharge to 246 prepare abundant reservoir capacity before the summer flood season, the XJB reservoir 247 was fully mixed in May. Hypolimnion in the XJB reservoir was limited to 248 249 approximately 150 m below the water surface in May and approached to 180 m below 250 the water surface in July as the discharge of the XLD reservoir apparently increased. In September, the XJB reservoir was fully mixed. 251

252 The DO in all samples varied from 7.06 to 16.08 mg/L with a mean value of 9.23 \pm 0.91 mg/L (SD, n=181). The DO of surface water was significantly higher than that of 253 bottom water in the XJB reservoir due to algal blooms in March. Additionally, the DO 254 of the XLD reservoir in the flood season was slightly lower than that in the dry season 255 (Figure S3B). Although both reservoirs were impounded in less than a decade, the 256 trophic status of both reservoirs is meso-oligotrophic. The chlorophyll a (Chl-a) 257 258 concentrations varied widely, with values of 0.01-35.02 mg/m3. The maximum Chl-a concentration in surface water during the study was found in the XJB reservoir. The 259 diatom bloom initiated in late February and continued to April in the XJB reservoir. The 260 Chl-a concentration in the XLD reservoir was below 15 mg/m³ throughout the year 261 262 (Figure S3C). The bloom-forming period of the XLD reservoir was approximately the same as that of the XJB reservoir. 263





264 The DOC concentrations in all samples varied from 0.04 to 6.13 mg/L with an average value of 2.37 ± 1.24 mg/L (SD, n=175). The maximum DOC concentration was 265 found between May and July, with a mean value of 3.94 ± 0.69 mg/L (SD, n=52), and 266 the minimum average value was found in January (1.24 ± 0.75 mg/L, SD, n=45, Figure 267 268 S3D). There was no obvious spatial difference in DOC concentrations between the two reservoirs, which indicates that the DOC concentrations may be significantly affected 269 270 by the input of upstream. In all the sampling sites, the TN and TP in the water column were 146.78 \pm 64.6 μ M (SD, n=182) and 2.51 \pm 1.12 μ M (SD, n=156), respectively. 271 The mole ratio of TN: TP, as an indicator of nutrient limitation, fell in a range between 272 42.93 (1st quartile) and 91.84 (3rd quartile), indicating heavily P limitation for primary 273 producers in both reservoirs. 274

275 **3.2 Particulate organic matter and stable isotopic signatures**

276 The TPM concentrations in all sampling sites of both reservoirs varied from 1.10 277 to 38.80 mg/L with a mean of 5.74 ± 5.03 mg/L (SD, n=182), and the average maximum 278 TPM concentration was found in July and September with a value of 10.78 ± 4.34 mg/L (SD, n=52). The POC concentrations in all water samples varied from 0.03 to 2.81 mg/L 279 with a mean of 0.35 ± 0.39 mg/L (SD, n=182, Figure 3). The PON concentrations in all 280 samples ranged from 0.01 to 0.21 mg/L, with a mean of 0.04 ± 0.03 mg/L, and the N/C 281 282 mole ratios varied from 0.019 to 0.273, with an average value of 0.116 \pm 0.032 (SD, n=182, Figure 3). The maximum POC and PON concentrations and minimum N/C mole 283 ratios were found in March, with average values of 0.98 ± 0.64 mg/L, 0.07 ± 0.05 mg/L 284 and 0.079 ± 0.052 , respectively (SD, n=26, Figure 3). Meanwhile, the maximum POC 285 286 and PON concentrations in the surface and bottom water were found before the XJB dam, and the N/C mole ratios from upstream were lower than those downstream during 287 this period. In particular, the values in the surface water were different from those in the 288 bottom water, which may be due to the lower flow and temperature stratification (Figure 289 3A, 3B). 290









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Figure 3. Wavelet figure showing the spatial and temporal distributions in particulate organic carbon (POC), particulate organic nitrogen (PON), and N/C mole ratios of particulate organic matter of the surface (A) and bottom (B) water from the XLD and XJB reservoirs.

295 The stable isotopic ratios and N/C mole ratios of POC and PON reveal that POM originated from different sources in the XLD and XJB reservoirs. Most δ^{13} C-POC 296 values in all samples showed no significant differences during the year ranging between 297 -13.0% and -35.3%, with a mean of $-26.1 \pm 1.8\%$ (SD, n=181). However, δ^{13} C-POC 298 in March showed obvious differences that were higher than other values ($-24.3 \pm 3.3\%$), 299 SD, n=26). δ^{15} N-PON varied from -1.5‰ and 17.5‰ with a mean of 6.5 ± 2.7‰ (SD, 300 n=182) which displayed irregularly high values in January 2018 ($8.9 \pm 4\%$, SD, n=26). 301 This may be influenced by the inputs of excrement or other anthropogenic activities 302 during this period. 303

As shown in Figure 4A to Figure 4C, most POM in the XLD and XJB reservoirs may have been derived from plankton. However, the δ^{15} N-PON, which may be affected by trophic levels in the two reservoirs displayed large fluctuations (Chen et al., 2018a).





307	Thus, the N/C mole ratios of POM and $\delta^{13}\text{C-POC}$ can better constrain the contribution
308	of different sources of POC in this study. As the average $\delta^{13}\text{C-POC}$ (–26.1‰) and N/C
309	mole ratios of POM (0.116) were similar to those of plankton, with values of -26.8%
310	and 0.184 respectively (Table 1), POC in the XLD and XJB reservoirs may have been
311	primarily from plankton (42 \pm 19%) and secondarily, from C3 plants (34 \pm 16%) and
312	less from soil (13 \pm 11%) and C4 plants (8 \pm 9%) (SD, n=181) (Figure 4D). Previous
313	studies have reported that particulate matter derived from erosion of soil may have low
314	OC adsorption due to the low clay content of surface soils in the Jinsha River Basin
315	(Wu et al., 2020), which agrees with the results of this study. Conversely, the POC/Chl-a
316	threshold ratio of 300 indicates the dominance of phytoplankton in the POC pool
317	(Suzuki et al., 2014; Kang et al., 2019; De Castro Bueno et al., 2020), and more than
318	half of the samples in the surface water were classified with the origin of phytoplankton
319	dominance (Figure S4), which is similar to the results of stable isotope mixing models.
320	POC can be divided into autochthonous and terrestrial POC. Autochthonous POC
321	mostly originates from plankton, yet terrestrial POC derives from plants and soil in the
322	catchment (Guillemette et al., 2013; Chen et al., 2018b; Tittel et al., 2019). The
323	concentrations of autochthonous and terrestrial POC provided in the Supporting
324	Information (Section S1.2, Figure S5) were estimated by the contributions of plankton,
325	soil, C3 and C4 plants.









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- Figure 4. A) N/C mole ratios vs. δ^{13} C-POC, B) δ^{15} N-PON vs. δ^{13} C-POC, C) N/C mole ratios vs. δ^{15} N-PON in particulate organic matter (POM) and endmembers (C3 and C4 plants, soil, plankton) (the dotted area in A, B and C was from literature (Kendall et al., 2001)); D) the contribution of endmembers in the XLD and XJB reservoirs.

331 **3.3** Spatial and temporal variations in CH₄ and CO₂ concentrations, δ^{13} C-CH₄

The annual CH₄ and CO₂ concentrations in all water samples from the two cascade reservoirs varied from 0.011 to 0.133 µmol/L and 0.016 to 0.257 mmol/L, with average values of 0.039 \pm 0.024 µmol/L and 0.046 \pm 0.025 µmol/L (SD, n=182), respectively. The annual average CH₄ concentration observed in the study was lower than those in global rivers and streams (1.45 \pm 7.98 µmol/L, SD, n=1439) (Stanley et al., 2016). In XLD reservoir (L3–L1), annual CH₄ fell in a range between 0.014 µmol/L and





338 0.052 μ mol/L, with a mean of 0.026 \pm 0.008 μ mol/L during the study (SD, n=42) 339 (Figure 5). Annual CO_2 in XLD reservoir fell in a range between 0.033 mmol/L and 0.092 mmol/L with a mean of 0.046 \pm 0.017 mmol/L (SD, n=42) (Figure 5). In XJB 340 reservoir (B10-B1), annual CH4 fell in a range between 0.012 µmol/L and 0.133 341 μ mol/L with a mean of 0.044 \pm 0.026 μ mol/L (SD, n=126). Correspondingly, annual 342 CO₂ in the XJB reservoir fell in a range between 0.016 mmol/L and 0.257 mmol/L with 343 344 a mean of 0.047 \pm 0.028 mmol/L. Downstream of the XJB reservoir (X1–X2) had a range of CH₄ between 0.011 μ mol/L and 0.064 μ mol/L with a mean of 0.033 \pm 0.016 345 μ mol/L (SD, n=14). The water column CO₂ concentration downstream of the XJB 346 reservoir was between 0.030 mmol/L and 0.077 mmol/L with a mean of 0.041 \pm 0.013 347 mmol/L (SD, n=14). The longitudinal gradient of CH₄ concentration from upstream 348 349 XLD reservoir to downstream XJB reservoir was evident, and the mid-part of the XJB reservoir (B6-B4) showed a relatively high level of CH4 concentration among all 350 samples. Comparatively, the longitudinal gradient of CO₂ concentration along the 351 cascade reservoir was not apparent (Figure 5). The mole ratios between CH₄ and CO₂ 352 353 (CH₄/CO₂) were among the highest in the middle of the XJB reservoir. For temporal 354 variations, the highest CH₄ concentration at the dam site of the XLD reservoir was 355 shown in the flood season. However, the peak values of CH₄ concentrations at both the surface and bottom water of the XJB reservoir were observed during the flood season. 356 357 However, there was no significant difference in CH₄ concentrations between the surface 358 and bottom water. The lowest CO₂ concentration was found before the XJB dam in 359 March, and the CO₂ concentrations in surface water were significantly lower than those 360 in bottom water (Figure 5A, 5B).

Stable carbon isotopic signatures of CH₄ in the water column could help to explore and identify CH₄ production and transformation in reservoirs (Whiticar and Faber, 1986; Lima, 2005; Templeton et al., 2006). δ^{13} C-CH₄ in all sampling water of both reservoirs ranged from -8.3‰ to -54.0‰ with a mean of -32.7 ± 9.0‰ (SD, n=156, Figure 5). δ^{13} C-CH₄ in all water samples showed obvious temporal and spatial variations, and maximum values were found before and under the two dams. Lower





- δ^{13} C-CH₄ values were observed in March and July, where the peak value of CH₄ concentrations was found. However, most δ^{13} C-CH₄ in both surface and bottom water
- 369 from the two reservoirs displayed no significant differences except for that in March



370 from the XJB reservoir (Figure 5A, 5B).

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Figure 5. Wavelet figure showing the spatial and temporal distributions in CH₄, CO₂ and δ¹³C-CH₄ of the surface (A) and bottom (B) water from the XLD and XJB reservoirs.

374 In freshwaters, a large fraction of CH4 in the aquatic food network exists as a carbon source which was oxidized by bacteria, can reduce CH4 entering the water by 375 diffusion (Jones and Grey, 2011; Taipale et al., 2011; Frossard et al., 2015; Sawakuchi et 376 al., 2016; Matoušů et al., 2019; Thottathil et al., 2019; Saarela et al., 2020), which may 377 change CH₄ dynamics in water. Previous studies reported that the methane oxidizing 378 379 bacteria (MOB) can survive at oxic-anoxic interfaces in freshwater systems, and CH₄ oxidation was frequently found in environments with high CH4 and DO concentrations 380 (Bagnoud et al., 2020; Reis et al., 2020; Bai et al., 2021). Moreover, the δ^{13} C values of 381





biogenic methane during methanogenesis were very low, such as methanogenesis by acetate fermentation in freshwaters, and the δ^{13} C-CH₄ was then gradually elevated during CH₄ oxidation due to ¹³C enrichment in residual CH₄ (Whiticar and Faber, 1986; Conrad, 1999; Bastviken et al., 2002; Lima, 2005). The magnitude of δ^{13} C-CH₄ by MOB significantly depends on environmental conditions, so the fractionation factor (*a*) caused by microbial CH₄ oxidation shows a large range of 1.003 to 1.039 (Templeton et al., 2006).

The study area was divided into three sections along longitudinal gradients, 389 B1-B4/L1-L3 as lacustrine regions (before the dam), B5-B7 as transition regions, and 390 X1-X2/B8-B10 as riverine reaches. In this study, there was no apparent anoxic water 391 392 column in the lacustrine regions (upstream of the dam) of the XJB reservoir. Therefore, 393 the oxic surface sediment and large water depth could have mitigated the release of CH₄ bubbles potentially produced by the deep sediment, which was likely to lead to low CH4 394 accumulation in water before the dam. The most δ^{13} C-CH₄ in the water column were 395 above -50% of methanogenesis by acetate fermentation, which indicated that CH₄ was 396 dominantly oxidized in the two reservoirs, especially δ^{13} C-CH₄ in the lacustrine region, 397 398 had higher values and was significantly different from riverine reaches and transition 399 regions (Figure S6). Therefore, the CH_4 accumulation in the water column may be influenced by MOB. Significantly, the most CH₄ oxidation process from B6 (the hotspot 400 401 of CH₄) to B1 (before dam) in the XJB reservoir can also be presented by Rayleigh 402 fractionation, for example, CH₄ concentrations and δ^{13} C-CH₄ in the March, July, 403 September 2018 and January 2019 satisfied the fractionation caused by microbial CH₄ 404 oxidation, and α ranged from 1.01 to 1.035 (Figure S7). The CH₄ oxidation process was 405 interestingly discovered along with the river flow, but the δ^{13} C-CH₄ in vertical water showed no significant differences. Only the δ^{13} C-CH₄ in bottom water from the XJB 406 reservoir was significantly lower than that in surface water in March, yet there was no 407 408 significant difference in CH₄ concentrations (Figure 5). Although recent studies have indicated that the δ^{13} C values of CH₄ produced in oxic waters are usually less negative 409 than those of CH₄ accumulation in an anaerobic environment (Hartmann et al., 2020), 410





411 more evidence is needed to support this process in this study.

412 **4 Discussion**

413 4.1 Cascade damming effect on input and accumulation of POC

The distribution of POC in a river can be significantly influenced by dams. At first, 414 damming increased losses of river connectivity and reduced water flow in the main 415 channel, which possibly decreased the transport of POC after damming due to the long 416 HRT, further resulting in bulk accumulation and burial of POC in reservoirs (Ulseth and 417 418 Hall Jr, 2015; Li et al., 2017a; Almeida et al., 2019; Li et al., 2020a; Wang et al., 2020). 419 Additionally, dam construction promotes the intensive water-level fluctuations to cause landslides and debris flows and indirectly influence land types, which may increase the 420 terrestrial POC in the reservoir (Yao et al., 2006; Luo et al., 2016; Iqbal et al., 2018; 421 422 Zorzal-Almeida et al., 2018).

Based on the N/C mole ratios and δ^{13} C-POC values, POC mostly originated from 423 424 terrestrial POC in the XLD and XJB reservoirs. Water temperature was the only factor 425 that exhibits an obvious positive effect on autochthonous POC (Figure S8), which was mainly caused by the outbreak of algal blooms in spring to increase autochthonous POC 426 427 in summer (Figure S5). The water level of the XJB reservoir and water flow exhibit a 428 significantly positive effect on terrestrial POC (Figure S8). Due to intensive water-level 429 fluctuations, the high-water level in January was modulated to a low water level in 430 March by reservoir scheduling operation, which led to an increase in the input of terrestrial POC in March (Figure 2C, 2D, S5). Additionally, terrestrial POC in July 431 under high water flow displayed an increasing trend that was possibly affected by runoff 432 erosion. The water level and water flow caused by reservoir scheduling operation were 433 primary factors for the input of terrestrial POC. Meanwhile, due to the low clay content 434 of surface soil in the river basin, the particulate matter derived from soil erosion may 435 have a low OC content (Yu et al., 2011; Wu et al., 2020). This study showed that the 436 437 proportion of POC in TPM during the flood season was lower than that in the dry season in cascade reservoirs (Figure S9). These results imply that particulate matter 438





- from the river basin was less composed of OC, which suggests that a small amount ofterrestrial POC possibly enters the XLD and XJB reservoirs during the flood season.
- 441 The DOC/POC ratios with a global average value of 1-1.2 displayed an exponentially negative correlation with the deposition of particulate matter in major 442 443 rivers (Ran et al., 2013). In this study, the DOC/POC ratios in March from the XLD and XJB reservoirs with an average value of 3.96 were significantly lower than those in the 444 445 other months (Figure 6) and close to the Long Chuan River in the Jinsha River Basin (Lu et al., 2012), which suggests that the sedimentation of POC in March was 446 significant. However, the average DOC/POC ratio in the flood period was higher than 447 10, which indicates that the accumulation of POC in the flood period was very low. 448 449 Because there are no large tributaries in the XJB reservoir, the accumulation of POC from upstream and riverbanks were primarily reflected by the mass balance calculation 450 of POC (Figure 7A). According to the mass balance results of POC, the accumulation of 451 POC was mainly reflected in March (Figure 7A), while the difference in POC flux in 452 July and September during the flood period was not evident. These results also indicate 453 that the accumulation of POC in the XJB reservoir was limited during the flood season. 454 Therefore, the cumulative effect of POC at a bimonthly scale was not significant in the 455 456 XLD and XJB reservoirs.







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Figure 6. Box plot showing the distribution of DOC/POC ratios in sampling month from the XLD and XJB reservoirs (the asterisks, black squares and middle black lines represented outliers, mean values and medians).



⁴⁶² 463

461

XJB reservoir.

464 4.2 CH₄ accumulation influenced by terrestrial POC in deep river-valley dammed

465 cascade reservoirs

466 CH₄ production and emissions in a freshwater system can be affected by 467 parameters such as geography, hydrology, OM, water temperature, and final electron





468 receptors, yet POC was the most important factor for CH₄ accumulation in rivers 469 (Baulch et al., 2011; Crawford et al., 2014; Stanley et al., 2016). The degradation of 470 different sources of POC in freshwaters was different. Autochthonous POC has been found to have a high bioavailability, yet terrestrial POC was already partly degraded and 471 472 less available for microorganisms (Guillemette et al., 2013). Autochthonous POC can be 473 the main reason for promoting CH₄ production in freshwaters, such as Lake Fuxian and 474 Lake Diamond in the USA (West et al., 2012; Li et al., 2020b). Due to terrestrial POC accounting for large proportions in sediment, terrestrial POC can also have a potential 475 promotion effect on CH₄ accumulation, such as the William H. Harsha reservoir in the 476 USA (Berberich et al., 2020) and the Congo River (Upstill-Goddard et al., 2017). 477 Therefore, the river reservoirs affected by dam interception have complicated 478 479 hydrological characteristics of lakes and rivers, which results in different impacts of autochthonous and terrestrial POC on CH4 in reservoirs. 480

481 Young reservoirs are frequently believed to be significant CH₄ emitters due to the decomposition of flooded OC (Prairie et al., 2018). The structural equation model can 482 be used to explore the relationship between different sources of POC and CH₄ by 483 various physicochemical factors (Figure S8). In terms of exogenous variables, the 484 485 hydropower production of the XJB reservoir, water temperature and flow show positive correlations, and the water level of the XJB reservoir displayed negative correlations 486 487 with the hydropower production of the XJB reservoir, water flow, and temperature. 488 Additionally, CH₄ accumulation in deep river-valley dammed cascade reservoirs was positively affected by terrestrial POC, water flow, and temperature rather than 489 490 autochthonous POC, hydropower production, and the water level of the XJB reservoir 491 (Figure S8). However, this path model cannot illuminate the relationship among CH₄ 492 $(R^2=0.22)$, autochthonous $(R^2=0.17)$ and terrestrial POC $(R^2=0.11)$ in the two reservoirs. 493 This implies that the causality among variables in the temporal dimension is not 494 synchronous, and more factors affecting relationships among CH₄ concentration and autochthonous and terrestrial POC in the spatial dimension need to be considered. 495

496 The dam constructions hindered the water velocity, resulting in increased HRT





497 (Wang et al., 2020). The relationships among HRT, CH₄ concentration, autochthonous 498 and terrestrial POC are discussed below. The terrestrial POC and CH₄ concentrations in 499 the XLD and XJB reservoirs present single peaks with HRT (Figure 8A, 8C), and autochthonous POC displays a positive correlation with HRT (Figure 8B). HRT as an 500 exogenous variable in the structural equation model, was also used to explain the spatial 501 relationship among endogenous variables such as CH4 and autochthonous and terrestrial 502 503 POC (Figure 9). The structural equation model shows a significant positive effect between HRT and autochthonous POC, which can be explained by an increase in 504 phytoplankton photosynthesis (Crawford et al., 2016). 505









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Figure 8. Polynomial and linear fit relationship between POC, CH₄ and HRT in the spatial dimension from the XLD and XJB reservoirs (the blue and red area represented confidence bands and confidence level for curves). The mean values and 50% quantiles of these variables were considered.



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Figure 9. Structural equation model among the mean values (left) and 50% quantiles (right) of
 HRT, autochthonous POC, terrestrial POC and CH₄ in the spatial dimension from the XLD
 and XJB reservoirs.

Previous investigations have shown that autochthonous POC in lake reservoirs 515 with depths greater than 100 m was completely mineralized before carbon burial 516 (Steinsberger et al., 2020). Autochthonous POC can be rapidly used to produce CH₄, 517 while the decomposition of terrestrial POC was slower (Grasset et al., 2018; Isidorova 518 et al., 2019). Autochthonous POC can be easily oxidized in deep water columns before 519 520 entering anaerobic environments or less distributed in surface sediments. Additionally, surface sediments often do not have good stability and an anaerobic environment, which 521 522 results in autochthonous POC being more likely to be converted to CO₂ (Isidorova et al.,





523 2019). Therefore, it was inferred that autochthonous POC displays no significant524 relationship with CH₄ in this study.

525 The CH₄ concentrations in the two reservoirs were positively influenced by terrestrial POC (Figure 8, 9). The structural equation model can well explain the 526 relationship with autochthonous POC ($R^2=0.33$ for mean value, $R^2=0.35$ for 50% 527 quantile) and CH₄ (R²=0.53 for mean value, R²=0.67 for 50% quantile), except for 528 terrestrial POC (R²=0.13 for mean value, R²=0.41 for 50% quantile). However, 529 terrestrial POC and CH₄ concentrations display a positive correlation in the spatial 530 dimension (Figure S10). The POC concentrations exhibit an exponential relationship 531 with the contribution of terrestrial POC, which may imply that terrestrial POC was a 532 major input for POC in the XLD and XJB reservoirs (Figure 10). Although the 533 autochthonous POC and CH₄/CO₂ mole ratios in March show a significantly positive 534 relationship mainly due to the outbreak of algal blooms, the most positive relationship 535 between terrestrial POC and CH₄/CO₂ mole ratios also indicates that the input of 536 terrestrial POC in the flood season is a crucial reason for CH₄ production (Figure S11). 537



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Figure 10. Polynomial fit relationship between the contribution of terrestrial POC and POC
 concentrations in the XLD and XJB reservoirs.

541 The mass balance of CH₄ in the XJB reservoir was related to the estimation of CH₄





542 production. The annual CH₄ production showed an obvious single peak, and the highest 543 peak value was found in July (Figure 7B). However, there was low accumulation and no significant increasing trend of terrestrial POC during the flood season influenced by 544 cascade damming, which indicates that no large amount of fresh terrestrial POC was 545 546 decomposed by bacteria to produce more CH₄. Water temperature significantly affects the respiration rate of bacteria and then impacts the degradation of OM. Therefore, CH₄ 547 548 production in the flood season may be affected by higher water temperatures, which was supported by the structural equation model discussed above (Figure S8) and 549 consistent with previous studies (Delsontro et al., 2010; Beaulieu et al., 2014). However, 550 the difference in temperature from the XLD to XJB reservoirs in July and September 551 was not evident (Figure S3), which means that the CH4 accumulation in deep 552 553 river-valley dammed cascade reservoirs is limitedly influenced by water temperature. These results suggest that terrestrial POC was the key factor in CH4 accumulation in the 554 555 XLD and XJB reservoirs.

However, there are numerous factors influence carbon migration and transformation in damming rivers. More work needs to be done in the future, such as the contribution of resuspended sediment POC to terrestrial POC. As complicated processes of CH₄ accumulation are influenced by the input of terrestrial POC, more modeling factors need to be considered.

561 **5 Conclusions**

562 POC in the XLD and XJB reservoirs mainly originates from terrestrial POC (56%). 563 CH₄ oxidation potentially influences the variations in CH₄ concentrations and δ^{13} C-CH₄ 564 in cascade reservoirs. Water level variations and flow regulation caused by reservoir 565 operation are primary factors for the input of terrestrial POC. The cumulative effect of POC at a bimonthly scale was not significant in the XLD and XJB reservoirs. Terrestrial 566 567 POC displays more persistent impacts on CH₄ accumulation. This study provides a scientific basis for revealing the major reason and mechanisms of CH₄ accumulation in 568 high-energy-density reservoirs under reservoir scheduling operation and helps to further 569





570 understand biogeochemical cycles of carbon in river-reservoir systems.

571 Author contributions

572 YZ and YS contributed equally to this work, processed the data, drew the figures,

573 designed the research framework and wrote the manuscript. ZL designed the study

574 protocol, supervised the study, wrote and revised the manuscript. SG, LL, BZ and YQ

575 carried out the field sampling and analyzed the samples.

576 Competing interests

577 The authors declare that they have no conflict of interest.

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