



To what extent can soil moisture and soil contamination stresses affect greenhouse gas emissions? An attempt to calibrate a nitrification/denitrification model

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Abstract

Continental biogeochemical models are commonly used to prospect the effect of land use, exogenous organic matter input or climate change on soil greenhouse gas emission. However, they can still not be used to investigate the effect of soil contamination while it is known to affect several soil processes and to concern a large fraction of land surface. We implemented a commonly used model estimating soil nitrogen (N) emission, the DeNitrification DeComposition (DNDC) model, with a function taking into account soil copper (Cu) contamination in nitrate production modulation. Then, we aimed at using this model to predict N-N₂O, N-NO₂ NO_x and N-NH₄ emissions in the presence of contamination and in the context of changes in precipitations. For that, incubations of soils were performed at different soil moistures in order to mimic expected rainfall patterns during the next decades and in particular drought and excess of water. The effect of this double stress on soil nitrate production was studied using a bio-assay. Then, data of nitrate production obtained under each moisture treatment were used to parameterize the DNDC model and estimate soil N emission considering the various effect of Cu. Whatever the moisture preincubation, experimental results showed a N-NO₃ decreasing production when Cu was added but with different sharpness depending on soil moisture. The DNDC-Cu version we proposed was able to reproduce these observed Cu effects on soil nitrate concentration with $r^2 > 0.99$ and RMSE < 10% for all treatments in the DNDC-Cu calibration range (>40% of the water holding capacity) but showed poor performances for the dry treatments. We modelled a Cu-effect inducing an increase in N-NH₄ soil concentration and emissions due to a reduced nitrification activity, and therefore a decrease in N-NO₃, N-N₂O and N-NO_x concentrations and emissions. The effect of added Cu was larger on N-N₂ and N-N₂O emissions than on the other N species and larger for the soils incubated under constant than variable moisture.

Keywords: copper, DNDC modelling, rainfall pattern, ecotoxicology, soil function

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1. Introduction

The increase in atmospheric greenhouse gases [GHG] like CO₂, CH₄, or N₂O is expected to induce a global climate change with e.g. higher mean temperature or changes in precipitation patterns with projections of increased precipitations or droughts depending on regions (Knutti and Sedláček 2012). These modifications in rainfall patterns may impact soil moisture which is one of the main drivers of soil microbial activity (Moyano et al. 2013). Microbial communities ensure key activities supporting numerous ecosystem functions, such as those involved in nitrogen (N) cycle influencing N₂O emissions (Jones et al. 2014) and are at the origin of more than 80% of N₂O fluxes (IPCC 2019). In particular, nitrification/denitrification processes are largely controlled by the local (an-)oxic treatments and therefore by soil moisture, denitrification being the main source of soil N₂O emission for moist soils whereas dry soil N₂O emissions are mainly due to nitrification (Bateman and Baggs 2005). This strong dependency to local soil O₂ availability (Khalil et al. 2004), by playing on the realization of nitrification/denitrification reactions and N species diffusion (Conrad 1996; Schurgers et al. 2006), makes N soil fluxes dynamics particularly difficult to predict at larger scales. Despite this, some continental biogeochemical have shown improved predictions when N cycle is explicitly represented (Kesik et al. 2005; Butterbach-Bahl et al. 2009; Vuichard et al. 2018).

In addition to climate change, human activities introduce significant quantities of contaminants into the environment, such as trace elements (TE) which are persistent and can be toxic for soil biota (Bech et al. 1997; Giller et al. 2009). Indeed, the contamination of soils by TE has become a major concern at global scale (De Vleeschouwer et al. 2007; Khan et al. 2008) coming from atmospheric sources (Steinnes et al. 1997) or through the use of pesticides (Nicholson et al. 2003). In particular, TE contaminations are known to largely affect soil microorganisms (Giller et al. 2009) and their activities, such as nitrification/denitrification processes (Broos et al. 2007; Mertens et al. 2010). Therefore, the combined effect of climate change and of soil contamination may largely impact the emissions of NO_x and N₂O from soils (Holtan-Hartwig et al., 2002; Vásquez-Murrieta et al. 2006). However, the effect of the interactions between climate change and soil contamination on the GHG emissions is still poorly documented (Rillig et al. 2019; Zandalinas et al. 2021).

Despite recent progress, the Earth system models (ESMs) used to predict future climate change still don't take into account soil contamination effect on GHG emissions (Anav et al. 2013) whereas a large fraction of the soils are impacted by contaminants (Lado et al. 2008). Furthermore, soil biogeochemical models are often used to estimate loss or accumulation of N species (ammoniac NH₄ volatilization, nitrate NO₃ leaching - Giltrap et al. 2010) or they respective concentrations under scenarii of organic fertilizer amendments, but do not take into account the contamination which often occurs simultaneously (Wuana and Okieimen 2011). Thus, there is a growing need to provide continentals models combining ecotoxicological/contamination and climate change concerns. Among biogeochemical models DeNitrification DeCompostion (DNDC, Changsheng Li et al. 1992) is a relatively simple model handling both biogeochemistry of denitrification and microbial growth (Li et al. 2000), and on which Land Surface Model-soil N component -a part of ESMs- like ORCHIDEE are built (Vuichard et al. 2018).

For this purpose, this study combines in an innovative way experimental and modelling approaches in order to evaluate the impact of soil moisture on the sensitivity of nitrification to copper (Cu) toxicity and



80 consequently on N-GHG emissions. Cu was chosen as a model of soil contamination due to its relevance in
agricultural soils and available data in the literature. Soil incubations were run during two months by applying
chosen soil moisture from drought to water saturation. Then, a bioassay estimating NO_3 production was performed
under a gradient of Cu added by spiking. The experimental data were then used to calibrate a new model, DNDC-
Cu, able to predict NO_x and N_2O emissions with the implementation of new functions to consider the effect of Cu
concentration ($[\text{Cu}]$) on nitrification/denitrification processes. Our hypothesis is that the building of such a model
85 will permit to gain in understanding on the effect of a soil $[\text{Cu}]$ on NO_x and N_2O and NH_4 cycling in a climate
change context. Hence, data are also used here to discuss on knowledge gaps in such modelling approaches, and
to question the use of soil contamination data in climate change scenarios.

2. Materials and Methods

90 2.1 Soil sampling

The soil was sampled in January 2017 at the surface layer (0–20 cm) of a control plot at the Qualiagro experimental
site ($48^{\circ}87'N$, $1^{\circ}97'E$ - [https://www6.inrae.fr/valor-pro_eng/Experimental-devices/QualiAgro/QualiAgro-web-
site](https://www6.inrae.fr/valor-pro_eng/Experimental-devices/QualiAgro/QualiAgro-web-site)). The soil sample was immediately wet sieved at 5mm and shortly stored at 4°C until microcosm build-up.
Aliquots of this sieved soil were used to measure the initial water content in addition to the maximum water
95 holding capacity (WHC) for the further microcosm experiments. This site is located at Feucherolles near Paris,
France, and had been designed to evaluate urban compost fertility together with the monitoring of contaminant
inputs (Cambier et al. 2019). Soil is a luvisol with 15% clay, 78% silt and 7% sand, a pH of 6.9, organic carbon
(Corg) and total N contents at 10.5 ± 0.2 and $1.00 \pm 0.03 \text{ g kg}^{-1}$ soil, respectively, and with a CEC of 7.9 ± 0.8
 $\text{cmol}^+ \text{ kg}^{-1}$ soil. This soil is not contaminated with Cu, and basal $[\text{Cu}]$ measured by ICP-AES after HF-HClO_4
100 extraction was of 12 mg Cu.kg^{-1} soil.

2.2 Experimental setup

In order to evaluate the impact of soil moisture on the sensitivity of nitrification to Cu toxicity, we carried out a
two-step experiment. The first step consisted in 5 different WHC incubation during 45 days, and the second to a
3-day bioassay with spiked Cu gradient (Fig. 1).

105 Five microcosms were built up with about 5g of sampled soil. Three of them were set up with a constant
moisture corresponding to 30%, 60% and 90% of their WHC in order to span respectively limiting, optimal, and
saturating conditions for the microbial activities. These three samples will be called thereafter “30%, 60% and
90%”, respectively. Their water contents were verified by weighting every two days and water added if necessary.
The two other microcosms were incubated in order to simulate two kinds of drought and dry-rewetting cycles.
110 One, thereafter called “Drought” (or DO), started with one week at 60% WHC and then the soil was left for 3
weeks without added water to mimic a dry period until 10% of the WHC before rewetting at the initial 60% WHC.
The other, thereafter called “Dry-rewetting” (or DR) encountered alternatives cycles of one-week dry period (10%
of the WHC) followed by one-week near-saturation period (90% WHC). The moisture states of microcosms were
performed by air-drying and controlled by weighting.



115 At the end of the incubation period, we performed a nitrification bioassay in triplicate using incubated
soils and following an adaptation of the method proposed by Petersen et al. (2012). Soils were first spiked with a
gradient of increasing Cu^{2+} in the presence of an excess of NH_4^+ and the resulting potential nitrification activity
(PNA) measured. The microcosms incubated at constant moisture were kept at their moisture level (30, 60 or 90%
of WHC) whereas those incubated at variable moisture were set at 60% WHC. The NO_3^- production rates were
120 measured in soil slurries over a short-term aerobic incubation, for each Cu added concentration. Briefly, 1 mL of
Cu solution at different concentrations were added in soil slurries (soil solution 1:12) to reach final added [Cu] of
50, 100, 250, 500, 750, 1000 and 2000 mg Cu.kg soil⁻¹ (final soil [Cu] of 62, 112, 262, 512, 762, 1012 and 2012
mg Cu.kg soil⁻¹). The pH was adjusted to 7. Then, microcosms were incubated on a rotary shaker (150rpm) under
aerobic conditions at 25°C until 72h. After 0, 24 and 72h of incubation, 2 ml aliquots of 3g were transferred in
125 Eppendorf vials and centrifuged. The supernatants were collected and stored in microplates at - 20 °C until
analyses of NO_3^- and NO_2^- by colorimetric determinations, following the reduction of NO_3^- in NO_2^- by
vanadium(III) and then the detection of NO_2^- by the acidic Griess reaction (Miranda et al. 2001). Finally, PNA (μg
N- NO_3^- g⁻¹ soil h⁻¹) was calculated on the basis of N- NO_3^- + N- NO_2^- concentrations measured at different time
steps. In our bioassay, $[\text{NO}_2^-]$ were negligible and PNA was thus calculated following Eq. (1), by checking the
130 linear production rate of NO_3^- between 2 h, 24 h and 72h:

$$(1) PNA = \frac{[\text{NO}_3^-]_{T_{\text{final}}} - [\text{NO}_3^-]_{T_{\text{initial}}}}{T_{\text{final}} - T_{\text{initial}}} \times V_S \div W$$

with V_S : Volume of solution

W : Weight of fresh soil (approximately 5 g)

T : Time of incubation.

135 2.3 Nitrification/denitrification model

Nitrification and denitrification processes are represented following the DNDC model proposed by Changsheng
Li et al. (1992) and Li et al. (2000). In this study, we used a simplified version adapted by Zaehle and Friend
(2010) initially calibrated for soil WHC >40%, that we intended here to test for 30% of WHC. This simplified
version needs less boundary data but keeps a mechanistic description of the main processes. Modelled N species
140 are expressed in amount of N, i.e. N- NH_4 , N- NO_3 , N- NO_x and N- N_2O . To be able to represent both nitrification
and denitrification processes occurring in aerobic and anaerobic sites, the soil is split into aerobic and anaerobic
fractions based on an empirical relationship linking O_2 consumption to soil respiration. In aerobic microsites,
nitrification takes places following Eq. (2):

$$(2) \text{Nitrification} = f(\text{SWC}) \times f(\text{temp}) \times f(\text{pH}) \times k_{\text{Nit}} \times (1 - \text{anvf}) \times \text{NH}_4$$

145 with N- NH_4 being the stock of ammonium (in gN m⁻²), $(1 - \text{anvf})$ the aerobic fraction of the soil described
thereafter in Eq. (21), k_{Nit} the nitrification rate (day⁻¹), $f(\text{SWC})$, $f(\text{temp})$ and $f(\text{pH})$ three rate modifiers representing
the effect of soil water content (m³ m⁻³), temperature (K) and pH as scalar respectively. They are described by the
following Eq. (3), (4) and (5):

$$(3) f(\text{SWC}) = 0.0243 + 0.9975 \times \text{SWC} + 5.6358 \times \text{SWC}^2 + 17.651 \times \text{SWC}^3 + 12.904 \times \text{SWC}^4$$



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$$(4) f(temp) = 0.0233 + 0.3094 \times temp + 0.2234 \times temp^2 + 0.1566 \times temp^3 + 0.0272 \times temp^4$$

$$(5) f(pH) = 1.2314 + 0.7347 \times pH + 0.0604 \times pH^2$$

155 The N-NH₄ nitrified is transformed into N-N₂O, N-NO or N-NO₃ due to bacterial processes and chemonitrification following Eq. (6),(7) and (8):

$$(6) \text{Nitrification}_{N_2O} = ftv \times SWC \times k_{\text{Nitrif}_{N_2O}} \times \text{Nitrification}$$

$$(7) \text{Nitrification}_{NO} = ftv \times SWC \times k_{\text{Nitrif}_{NO}} \times \text{Nitrification} + 496950 \times e^{-1.62 \times pH} \times e^{-31494/(temp \times R)} \times \text{Nitrification}$$

160 (8) $\text{Nitrification}_{NO_3} = \text{Nitrification} - \text{Nitrification}_{NO} - \text{Nitrification}_{N_2O}$

with $k_{\text{Nitrif}_{NO}}$ and $k_{\text{Nitrif}_{N_2O}}$ two fixed rates (d⁻¹), ftv a rate modifier controlled by temperature and given in Eq. (9) and R the ideal gas constant.

$$(9) ftv = 2.72^{(34.6 - \frac{9615}{temp})}$$

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Then, the N-NO₃ produced during the nitrification process enters the denitrification module where it is reduced sequentially into N-NO_x, N-N₂O or N-N₂ following Eq. (10) to (12):

$$(10) \text{Denitrification}_{NOx} = anv f \times \left(\frac{\mu_{NO_3}}{0.401} + 0.09 \times \frac{NO_3}{N_{tot}} \right) \times B$$

170 (11) $\text{Denitrification}_{N_2O} = anv f \times \left(\frac{\mu_{NOx}}{0.428} + 0.035 \times \frac{NOx}{N_{tot}} \right) \times B$

$$(12) \text{Denitrification}_{N_2} = anv f \times \left(\frac{\mu_{N_2O}}{0.151} + 0.079 \times \frac{N_2O}{N_{tot}} \right) \times B$$

The anaerobic fraction $anvf$ is described following Eq. (13):

$$(13) anv f = 0.85 \times \left(1 - \frac{p_{soilO_2}}{p_{airO_2}} \right)$$

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with p_{airO_2} , p_{soilO_2} being the partial pressure in the air and in the soil respectively. p_{soilO_2} is calculated following Eq. (14)

$$(14) \frac{\partial p_{soilO_2}}{\partial t} = p_{soilO_2} - k \times SOC \times f_{Cu}$$

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with SOC being the soil organic carbon stock (gC m^{-2}), k the decomposition rate, $p_{O_2, resp}$ the O_2 partial pressure related to the respiration, and f_{Cu} the effect of Cu on CO_2 emissions as define in Eq.(15), following (Sereni et al. 2021; Eq. (5):

$$(15) f_{Cu\ CO_2} = \exp(-0.1 - 0.1 \times \log(Cu) + 0.12 \times pH)$$

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The relative growth rate of N- NO_3 , N- NO_x and N- N_2O denitrifiers are described respectively by μ_{NO_3} , μ_{NO_x} , μ_{N_2O} following Eq. (16), (17) and (18).

$$(16) \mu_{NO_3} = \frac{0.67 \times f_{denit}(temp) \times f_{denit_NO_3}(pH) \times NO_3}{NO_3 + 166}$$

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$$(17) \mu_{NO_x} = \frac{0.34 \times f_{denit}(temp) \times f_{denit_NO_x}(pH) \times NO_x}{NO_x + 166}$$

$$(18) \mu_{N_2O} = \frac{0.34 \times f_{denit}(temp) \times f_{denit_N_2O}(pH) \times N_2O}{N_2O + 166}$$

with $f_{denit}(temp)$, $f_{denit_NO_3}(pH)$, $f_{denit_NO_x}(pH)$, $f_{denit_N_2O}(pH)$ being rates modifiers depending on air temperature and soil pH described in Eq. (19) to (22).

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$$(19) f_{denit}(temp) = 2^{(temp-22.5)/10}$$

$$(20) f_{denit_NO_3}(pH) = 1 - \frac{1}{1 + e^{(4.25 \times pH)/0.5}}$$

$$(21) f_{denit_NO_x}(pH) = 1 - \frac{1}{1 + e^{(5.25 \times pH)}}$$

$$(22) f_{denit_N_2O}(pH) = 1 - \frac{1}{1 + e^{(6.25 \times pH)/1.5}}$$

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The denitrifier biomass dynamic B (kg m^{-2}) is described following Eq. (23).

$$(23) \frac{\partial B}{\partial t} = (anvf \times (\mu_{NO_3} + \mu_{NO_x} + \mu_{N_2O}) - 3.82 \times 10^{-3}) \times B$$

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Finally, all the gaseous forms of mineral N are emitted into the atmosphere. We wrote the adapted model in R, with a the time step of the model of 30 minutes and most of the parameters were kept to the original values of Changsheng Li et al. 1992; Li et al. 2000) except k_{nit} which was modified to 0.1743 instead of 0.2 to better fit the data from the control. Furthermore, the amounts of N- NH_4 fixed to the clay were reduced to 0 as the bioassay was performed in excess of N- NH_4 (see 2.2.0).

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We used measures of N species at the end of preincubation period as initial values of N species for DNDC (Table 1a and Fig. 2.). To estimate the anaerobic volume fraction during the 3 days bio-assay, we used a C mineralization rate k (eq 14) determined on the basis of measurements performed on the same soil (Annabi et al. 2007) and ran DNDC for a 45 days equilibrium period. We then extracted the initial anaerobic volume fraction and partial O_2 pressure.

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2.4 Statistical analysis



The dose-response curves of PNA to Cu gradient were plotted and tested with linear, quadratic or cubic functions as fitting models. Our aim was to find, if possible, a similar modelling fit function for all moisture incubation treatments. Thus, for each moisture treatment, the two best functions of fit were selected through AIC and R² criteria, and compared with ANOVA. After selection of a common type of functions, the permutability of the different functions parameters was tested with the Chow test (gap package which tested the regression 1 on the basis of the samples 2 and vice-versa). If the p-value exceeds its critical values, regressions cannot be considered equal (Zhao 2007)).

To estimate the effect of [Cu] and soil moisture on the different variables measured, we used nonparametric Kruskal-Wallis test. The fits between the model and the data were measured using root mean square error (RMSE, Eq. (24)):

$$(24) RMSE = \sqrt{\frac{1}{N} \sum_{i=1}^N (X_i - Y_i)^2}$$

where i is the number of observations (1 to N), X is the predicted value and Y is the observed value. RMSE was decomposed in standard bias (Eq. (25)), non-unity slope (Eq. (26)) and lack of correlation (Eq. (27)) component following Gauch et al. (2003), with \bar{X} and \bar{Y} the mean modelled and observed values, b the slope of the least squares regression of Y on X and r^2 the square of the correlation:

$$(25) : SB = (\bar{X} - \bar{Y})^2$$

$$(26) NU = (1 - b)^2 \times \sum \frac{x_i^2}{N}$$

$$(27) LC = (1 - r^2) \times \sum \frac{y_i^2}{N}$$

All the analysis were done with R 3.2.3 (R Core Team, 2015).

3 Results

3.1. Effect of Cu on potential nitrification activity (PNA): statistical model selection

The soil N species measured at the end of the soil pre-incubation in each soil moisture treatment were used to initialise the DNDC model (Table 1). Two anomalous points leading to anomalous calculated N-NO₂ values were excluded from the experimental results because of technical problems during measurements (the C replicates in the DR and DO cases).

The bioassay experiments performed at the end of the soil pre-incubation allowed us to determine the rate of nitrate production as a function of soil [Cu] for each soil moisture (Fig. 1). In all cases, the PNA was found to decrease with the increase in soil [Cu] but at different rates depending on the moisture treatment. Based on AIC (suppl. Table 1) values, we first selected the model that better fitted the data. For 30 and 60% of WHC, a quadratic model was found to provide the better compromise between the number of parameters and the prediction capacity for incubation. For 90% WHC, no significant difference was found between the cubic and the quadratic models



250 (ANOVA, $p.v=0.07$). For DR, no significant difference was found between linear and quadratic models (supplementary Tables 1a and 1b) whereas for DO the cubic model provided a substantially better fit than the quadratic model (AIC and adj. R2 score, suppl table 1). Finally, we found that the quadratic model fitted correctly all the sets of data, allowing to be homogeneous across the moisture's incubation treatments (Fig. 2b). The quadratic function was thus chosen to quantify the Cu effect on PNA also for the DO treatment.

255 The parameters of the 5 quadratic functions (one for each moisture treatment) were found different from each other, except between 60 and 90% WHC ($p.v=0.001$, Chow test). A single function was thus used to adjust PNA to soil [Cu] at 60 and 90% WHC but with different intercepts for these two WHC treatments (suppl. Table 2 and Fig. 2).

260 The final 4 quadratic equations (Eq. (28) for 30% WHC, Eq. (29) for 60 and 90% WHC, Eq. (30) for DR, and Eq. (31) for DO, Fig. 2) were then added in the DNDC model, allowing to adjust the Eq. (2) which regulates the nitrate production to soil Cu contents:

$$(28) F_{Cu30} = 0.782 - 0.000451 \times Cu + 9.49 \times 10^{-8} \times Cu^2$$

$$(29) F_{Cu60/90} = b - 0.000342 \times Cu + 4.30 \times 10^{-8} \times Cu^2$$

with $b= 0.795$ for 60% WHC and $b= 0.0.79629$ for 90 % WHC

$$(30) F_{CuDR} = 0.552 - 0.000164 \times Cu + 6.09 \times 10^{-8} \times Cu^2$$

265 $(31) F_{CuDO} = 0.625 - 0.000192 \times Cu + 2.82 \times 10^{-8} \times Cu^2$

3.2. Modelling soil nitrate concentrations in Cu contaminated treatments using a DNDC-Cu model.

3.2.a. Set up of the DNDC-Cu model

270 The DNDC model was originally constructed to model both C and N soil cycles. The relative proportion of nitrification and denitrification processes thus depends on soil aerobic fraction determined both by soil C respiration and soil moisture (Eqs. (13) and (14)). Before any addition of Cu function in DNDC, we estimated this soil aerobic fraction, arising from C mineralisation and the aerobic fraction. Therefore, we used previous data from 366 days incubations made on the same uncontaminated soil (Annabi et al. 2007) to first fit a C mineralisation
275 coefficient rate, k . The resulting k coefficient ($k = 1.234 \text{ e}^{-4} \text{ gC.m}^{-2} \cdot 30\text{min}^{-1}$) was introduced in the DNDC model and forced to equilibrium (45 days) without soil Cu contamination effect. This provided us a basal aerobic volume fraction for each soil moisture through Eq. (13) corresponding to 0.00352 at 30%, 0.006167 at 60% (and DR/DO to which bio assays were performed at 60% WHC) and 0.02705 at 90% of the WHC and partial O_2 pressure to 211.4 hPa at 30% WHC, 210.7 hPa at 60% WHC, DR and DO and 205.4 hPa at 90% WHC. These values were
280 then used to initiate our DNDC-Cu version. We then ran the DNDC-Cu version for a 3-day modelisation. The constant rate of C mineralisation, k , was adjusted to take into account Cu through the eq 13 and the eq 28-31 adjusted $N\text{-NO}_3$ production rate (Fig. 1) to Cu.



3.2.b. DNDC-Cu model validation

285 A validation of our DNDC-Cu model was made by comparing experimental data of soil measured after 1 and 3
days of incubation with model outputs. A better fit was provided for 60 and 90% of WHC which are in the range
of DNDC calibration compared to 30 %WHC where the nitrate production is largely underestimated (more than
twice after 3 days of incubation, Fig. 3a). The regression slopes between modelled and measured soil [nitrate] for
60 and 90% WHC were respectively 0.94 and 0.91 ($R^2=0.99$ in both cases, Fig. 3a.) whereas for 30% WHC the
290 regression slope was 1.21 ($R^2=0.92$) (Fig. 3a). For DR, the soil nitrate stocks were overestimated at 762 mgCu.kg
soil⁻¹ soil and underestimated at 2012 mgCu.kg soil⁻¹ (Fig. 3b) (respectively 389.7 gN.m⁻² and 310.5 gN.m⁻² mean
modeled nitrate against 375.3 and 290.6 gN.m⁻² mean measured nitrate) but overall modelling adequately fitted
the data with a regression slope at 0.95 and $R^2=0.99$ too.

Considering all incubations treatments, RMSE was about 57.3 as a mean (46.4 gN-NO₃.m⁻² standard error)
295 for a mean soil nitrate measured at 390 gN-NO₃.m⁻² (69 gN-NO₃.m⁻² standard error) after 3 days of incubation.
However, for the 30% WHC, RMSE was 139.9 thus 3.7 times more than for the other treatments (Suppl. Fig. 1).
Despite the reduction of nitrate production rate from 0.2000 to 0.1753 gN.hour⁻¹ (see material and methods), soil
nitrate stock was still slightly overestimated in the 90% WHC as shown by the largest lack of correlation in this
case compared to the 60% WHC treatment (Fig. 3a, suppl. Fig. 1). Lack of correlation was small for all tested
300 moisture incubation treatments (mean $\sqrt{LC} = 23.0$, standard error =5.4 which is roughly 1/20 of the produced
nitrate in 3 days in uncontaminated treatment). Results showed that our DNDC-Cu version was able to reproduce
the variability observed in Cu contaminated soils except for the 30% WHC treatment where soil nitrate stocks
were largely underestimated. The following results thus focused on the use of DNDC-Cu for DR, DO, 60 and 90%
of WHC treatments to predict soil N emissions.

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3.3 Use of DNDC-Cu to predict N fluxes in contaminated soils.

3.3.a. Effect of soil [Cu] on soil N stocks.

The soil Cu function we included in the DNDC-Cu model modified specifically the default nitrification equation
in complement to pH, soil moisture and O₂ availability (Eq (2.)). In the presence of low [Cu] (12-512 mgCu.kg
310 soil⁻¹), the predicted N-NO₃ soil stocks were found equivalent between 60 and DO and, to a less extend, DR
treatments (Suppl. Fig. 2). When soil [Cu] increased, soil [N-NO₃] decreased but with different rates depending
on the moisture of incubations (eq 28-31). Up to 548 mgCu.kg soil⁻¹, we modelled the lowest N-NO₃ stocks in DR
incubated soils. Above it, N-NO₃ soil stocks were the smallest for the 60% WHC treatment as a result of the
sharpest decrease in N-NO₃ production due to soil [Cu]. N-NO₃ soil stock for incubation at 90% WHC were the
315 highest for soil [Cu] below 1432 mgCu.kg soil⁻¹. Between 1432 and 2000 mgCu.kg soil⁻¹, N-NO₃ soil stocks were
similar for 90% WHC, DR and DO (Suppl. Fig. 2).

The decrease in the nitrification rates with the increase in soil [Cu] resulted in an increase in soil N-NH₄
stocks. Our model also predicted largest N-NH₄ stocks in the DR and DO soils than in the 60 and 90% WHC soils



320 (Suppl. Fig.3). The variations in N-NH₄ and N-NO₃ stocks were however different across soil moistures so that at the highest [Cu], the N-NH₄ soil stocks were modelled similar between all moisture treatments whereas the N-NO₃ stocks were modelled smallest for the 60% WHC treatment. The N-NO₃/N-NH₄ stocks ratios thus varied between soil moistures with Cu levels whereas in the absence of Cu, N-NO₃/N-NH₄ ratios were similar among soil moisture treatments. The ratio of N-NO₃/N-NH₄ decreased faster for 60/90% WHC than for DR/DO with an increase in soil [Cu] (suppl Fig. 4; Table 2).

325 With the decrease in soil N-NO₃ stocks at high [Cu], we predicted a decrease in the growth of denitrifying bacteria (Eq. (13)). Consequently, the modelled denitrifying bacterial pool was reduced when soil [Cu] increases (Fig. 4). Whatever the soil [Cu], denitrification was however modelled roughly twice larger in the soils incubated at 90% WHC than in the other treatment as this moist treatment is defined as perfect condition for denitrifying bacteria in the DNDC model (Changsheng Li et al. 1992). Soils incubated at 60% WHC were modeled with the
330 lowest denitrifying bacterial pool. No difference between the DR and DO soils was found due to uncertainties in the modelled denitrifying bacterial pool which resulted from the different concentrations in N species used to initialize DNDC-Cu (Table 1). The soil N-N₂O stocks and dissolved N-NO_x being directly related to denitrifying bacteria and its followed similar trends than soil N-NO₃ stocks with a global decrease in soil stocks with an increase in soil [Cu] (table 2) and larger stocks at the wetter treatment.

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3.3.b. Estimation of soil N emissions under various moistures

Large differences are predicted in the N-NH₄, N-NO_x and N-N₂O fluxes between the 90% WHC soil and the 3 other soil moisture treatments (Fig. 5). For instance, we modelled a decrease comparable in N-NO_x emissions between DR/DO and 60-90% WHC for soil [Cu] about 112 mgCu.kg soil⁻¹ (2-3% respectively - Table 2a and 2b.)
340 but with the increase in soil [Cu], the variation of emissions between soil moisture became larger. Hence, around 2012 mgCu.kg soil⁻¹ we modelled more than 50% decrease in N-NO_x and 62% decrease in N-N₂O emission fluxes for soils at 60% WHC against only 40% decrease in N-NO_x and 54% in N-N₂O emission for soils previously exposed to DR (Tables 2a and b.). Thus, intensities of fluxes between two moisture treatments reversed with an increase in soil Cu contamination.

345 The smallest fluxes were predicted for the wetter treatment despite higher modelled N-N₂O stocks at 90% WHC whatever [Cu]. N-NH₄ fluxes were modelled higher for the DR soils than in the 60% WHC incubated for soil Cu higher than 1774 mgCu.kg soil⁻¹ and smallest below. The emissions of N-NH₄ in the DO treatment were predicted to be higher than those of the DR treatment for soil Cu higher than 1290 mgCu.kg soil⁻¹ and smallest below (Fig. 5a). In the studied range of added Cu, N-NO_x fluxes predicted by the model are largest from 60% WHC
350 to DO, DR and 90% WHC (Fig. 5(b.)) for moderate Cu input (~ below 1380 mgCu.kg soil⁻¹). The decrease in N-NO_x emission with the increase in soil [Cu] was however steeper for soils incubated at 60% WHC. Hence, at 2012 mgCu.kg soil⁻¹ N-NO_x fluxes in soil incubated at 60% WHC were similar to those in the soils incubated under drought treatment (Fig. 5 (b.)). N-N₂O fluxes had similar trends than N-NO_x for moderate Cu inputs but fluxes were still largest from 60% WHC to DO, DR and 90% WHC (Fig. 5 (c.)).

355 The N-N₂O emissions fluxes in the presence of Cu were predicted to be 4 times smallest in the 90% WHC treatment compared to the others, whereas N-N₂ emissions were larger at this wettest treatment (Fig. 5 (d.)). The



ratio of emitted N-N₂O per denitrification products (e.g. N-N₂O/ N-N₂O+ N-N₂) was hence smallest in the moistest soils, which means that the largest soils N-N₂O stocks in the case of 90% WHC had more chance to be transformed rather than emitted (Fig. 6).

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4 DISCUSSION

4.1. From laboratory experiment to soil N emission modelling

Thanks to our laboratory experiments, we were able to define a function modulating the soil N-NO₃ production rates in relation with soil [Cu] and depending on the soil moisture. Our results showed that soil nitrate decreases with an increase in soil [Cu].

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After implementing these Cu modulating functions into the DNDC-Cu model, we were able to reproduce the observed soil nitrate stock particularly for the soils incubated at 60 and 90% of WHC. The variability around the mean due to the Cu effect was also reproduced by our DNDC-Cu version at 30% of WHC despite strong underestimation of mean soil nitrate stocks due to model moisture-limit (Changsheng Li et al. 1992). In the case of the DR and DO incubated soils, the so-called “Cu function” also accounted for the effect of drought stress. In fact, our Cu functions were defined on the basis of soil nitrate production against the whole gradient of Cu thus also considering the control without Cu. However, the (double) stress effect was also well reproduced in nitrate production.

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4.2. From N-N₂O, N-N₂ and N-NO_x soil stocks to emissions

In the present study, we predicted highest soil N-N₂, N-N₂O and N-NO_x stocks in the moistest treatments as they are produced by the denitrifying bacteria expected to behave optimally at 90% WHC or after DR cycles (Changsheng Li et al. 1992; Homyak et al. 2017). However, N-N₂O and N-NO_x emissions were modelled higher in the driest soils, whereas numerous studies (Dobbie and Smith 2003; Xiong et al. 2007; Manzoni et al. 2012) reported high measured N-N₂O emissions with high moisture. In the present version of DNDC-Cu, the soil N emissions were directly controlled by their diffusion in soil, calculated on the basis of clay and soil moisture content. The diffusion of each species would hence be 11 times smaller under 90% WHC ($D_s=0.00357$) than under the 60% WHC treatment ($D_s=0.0306$) because the model described the diffusion as a whole and do not separated pores with or without water. Diffusion was hence slower in the water than in the air. Thus, the weighted mean diffusion was lower in the high moisture treatment. Without Cu soil nitrous stocks being roughly 1.6 times and soils N-N₂ stocks 11.1 larger under 90% WHC treatment than the other, the emission of N-N₂O were larger under driest treatment even if stocks were smaller.

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Several studies also reported flushing event with dry rewetting cycles which would enhance C mineralisation, known as the Birch effect (Birch 1958; Göransson et al. 2013), hence reducing soil O₂ concentration. Moreover, soil [O₂] is closely related to the pore size distribution, being of major importance in nitrification/denitrification

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control (Khalil et al. 2004) with a dominating nitrification for aggregates up to 0.25cm (Kremen et al. 2005). Pore size distribution under dry/rewet events is controlled by cracking, (des)aggregation (Denef et al. 2001; Cosentino et al. 2006) or gas displacement (Kemper et al. 1985) that we weren't able to take into account in the present study.

395 In DNDC, the calculation of denitrification rate and diffusion was based on a rough description of anaerobic zone with approximation of soil pore space distribution (Li et al. 2000; Blagodatsky et al. 2011). The soil pore space distribution approach has been demonstrated to be more generally applicable (Arah and Vinten 1995; Schurgers et al. 2006) whereas soil aggregates have been shown to control the extend of nitrification and denitrification (Kremen et al. 2005; Schlüter et al. 2018). However, if models have been proposed to take O₂ availability at the

400 aggregate size into account in the nitrous oxide production (Leffelaar 1988; Kremen et al. 2005), they also point out the difficulty in parametrization which need a large panel of soil measurements. Moreover, they are rarely transposable at the meso-and regional scale due to high spatial variations in soil structure (Butterbach-Bahl et al. 2013). The DNDC-Cu version we used here particularly pointed out the difficulty in dealing on biogeochemistry model with physical processes, with large discrepancies between modelled soils stocks and emissions. The

405 validation we performed focused on soil nitrates stocks and a second step to go further on would be the measure of gaseous species to ensure that emissions were also impacted by soils treatment. Moreover, we assumed here that soil [Cu] affected the C mineralisation with a decrease in soil O₂ production leading to an increase in denitrification and N-N₂O, N-NO_x. Nevertheless, the present DNDC-Cu version didn't take into account the retroaction between C and N cycles. Further research would thus be required to include Cu contamination into C

410 and N interacting cycles.

4.3. Expected ecological implications of soil Cu contamination

Based on nitrate production measurements, we modeled a decrease in denitrifying activities with an increase in soil [Cu] as a consequence of the decrease in soil nitrate stocks. However, the experiments performed here did not

415 allow us to determine if the soil Cu contamination rather affects nitrifying bacteria (e.g. decrease in nitrifying activity and in N-NO₃ production) or denitrifying bacteria (e.g. increase in denitrifying activities and N-NO₃ consumption). The effect of soil contamination on N-N₂O production is debated because i) species involved are not clearly identified (Wrage-Mönnig et al. 2018), ii) species richness is not necessary related to soil functions

420 (Ruyters et al. 2013) and iii) denitrifying communities could be differently sensitive than the nitrifying to soil contamination (Vásquez-Murrieta et al. 2006; Hund-Rinke and Simon 2008). Also, our modeling approach of N-N₂O, N-N₂ and N-NO_x production in the contaminated context could have been more constrained with measurement of denitrification rate to assess the effect of Cu on proportion of production and consumption of N-NO₃

425 Based on our simulations, the soil Cu contamination was expected to substantially modify the proportion of available N in soils with the increase in N-NH₄ stock at the expense of N-NO₃. N-NH₄ accumulation and the large expected decrease in N-NO₃/N-NH₄ ratio in contaminated soils (around 50% for the 60% WHC) may lead to shift in plant community structures with different preferences in N assimilation (Peacock et al. 2001; Cui and Song 2007). Therefore, Cu stress could not only have implications in microbial community patterns as a stressor,



430 but could also induce further shifts due to N species redistributions in soils.

4.4. Climate change could substantially modify contaminated soil N emission

435 It is well known that climate change and rainfall patterns could substantially modify the soil N balance and its
GHG emissions (Galloway et al. 2003, 2008; Butterbach-Bahl et al. 2013). Our results showed that increased Cu
contamination as well might affect soil N emissions with smallest emissions of N-NO_x and N-N₂O. These two
gases are of major importance in GHG mitigation with a warming potential per mass 300 and 40 times greater
than CO₂, respectively. Agricultural soils being the dominating source of N-N₂O (Beauchamp 1997; Signor and
Cerri 2013), even a limited decrease in their emissions could have major implication for climate. Based on our
modelling, the join effect of soil moisture and [Cu] was particularly important with larger differences in N-N₂O
440 and N-NO_x emissions between rainfall patterns at high [Cu] (3.3.b.). We (Sereni et al. in press) previously/also
showed that soil Cu contamination differently affect soils nitrification depending of primary soil moisture stress.
Here we showed that the N-N₂O and N-NO_x emission variations are significantly more sensitive to the combined
effect of Cu and precipitation regime than the nitrate stock. Based on these results, soil Cu inputs on moistest soils
would lead to a largest decrease in soil N-N₂O and N-NO_x emission compared to that on driest soils, and even
445 more than on soils submitted to abrupt and intense shifts in rainfall patterns as the DR and DO soils.

5. CONCLUSION

450 In the present study, we aimed at combining ecotoxicological experiments and biogeochemical modelling focusing
on the effect of soil Cu contamination on soil N emission under different soil moisture treatments. Based on a 3-
day bioassay measuring soil N-NO₃ over time, we were able to adjust the DNDC model to take into account the
Cu effect on soil N emission. Experiments were performed under different moisture treatments to create a DNDC-
Cu model taking account the effect of an exposure to a constant moisture (30, 60 or 90% WHC) or to a single long
drought period (DO) or several dry rewetting cycle (DR). The DNDC-Cu version we proposed was able to
455 reproduce the observed Cu effect on soil nitrate stock with $r^2 > 0.99$ and $RMSE < 10\%$ for all treatments in the
DNDC calibration range (>40% WHC).

We modelled a Cu effect inducing a decrease in denitrifying bacterial pool leading to an increase in N-NH₄
soil stocks at the expense of N-NO₃, N-N₂O and N-NO_x stocks. Emissions of N-NH₄ were expected to slightly
increase with soil Cu contamination whereas those of N-N₂O and N-NO_x are expected to decrease. We also
460 showed that the effect of soil Cu contamination was different among moisture treatment and N species. In fact, we
modelled that the largest [Cu] (2012mg Cu.kg soil⁻¹) provoked a decrease in soil nitrate stocks from -28% in the
DR case to -44% in the 60% WHC whereas N-N₂O emission were expected to decrease up to 63% in the 90%
WHC (-62% in the 60% WHC case, -54% in the DO case). Our results tended to show that the amount of N-N₂O
emitted from denitrification would decrease with an increase in soil [Cu] and from 60% WHC to DR, DO and
465 90% WHC, so that less N-N₂O produced would be converted to N-N₂. This result points out two main difficulties



470 in biogeochemical modelling: i) the difficulty to take into account hydrological dynamics (produced N-NO₃ and N-NH₄ could be expected to leach) and soil structures at different spatial scale (denitrification is estimated based on rough estimation on anaerobic soil volume which also controlled emissions rates through diffusion processes) and ii) linking soil function to microbial dynamics, in particular in this case were only the N-NO₃ stock was measured (without dealing between production and consumption for instance). Despite these two main points of uncertainty, the combination of incubations and of modelisation we conducted here emphasize the need to account for soil contamination when dealing with soil GHG emission modelling and climate change, as both contamination and rainfall patterns affect in a different way the soil N-NO_x and N-N₂O emissions.

475 Data availability: Data are available at <https://doi.org/10.15454/ZUKN90>

Authors Contributions: the authors contributed as follows:

Laura Sereni: Methodology, Formal analysis, Data processing, Writing original draft.

Bertrand Guenet: Methodology, conceptualization, writing review and editing, supervision

Charlotte Blasi: Experimentations and draft initialization

480 Olivier Crouzet: Methodology, conceptualization, writing review and editing, supervision

Jean-Christophe Lata: writing review and editing, supervision

Isabelle Lamy: Methodology Conceptualization, Writing review and editing, Supervision Project administration, Funding acquisition.

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Table 1: N species measured in the soils at the end of pre-incubation period further used to initialise the DNDC model, mean modelled N-NO₃ stocks and mean emissions of N-NH₄, N-N₂, N-N₂O, N-NO_x

635 modelled without Cu. 90 = 90 % WHC; 60= 60%WHC; DO =Dry only; DR =Dry Rewet condition during pre-incubation. A, B and C are replicates.

Ech	Measured ($\mu\text{g.g soil}^{-1}$)			Modelled ($\text{gN.m}^{-2}\text{h}^{-1}$ for emissions, gN.m^{-2} for stocks)					
	N-NH ₄	N-NO ₂	N-NO ₃	N-NH ₄ emissions	N-N ₂ emissions	N-N ₂ O emissions	N-NO _x emissions	N-NO ₃ stocks	N-NO ₃ /N-NH ₄ stocks
30_A	4.3	0.1	15.3	NA	NA	NA	NA	NA	NA
30_B	4.0	0.2	14.4						
30_C	4.5	0.2	14.3						
60_A	6.9	0.1	18.8	$2.28 \cdot 10^{-10}$	$2.26 \cdot 10^{-7}$	$1.3 \cdot 10^{-4}$	$1.3 \cdot 10^{-3}$	456.3	0.21
60_B	6.9	0.2	18.8						
60_C	6.7	0.2	18.7						
90_A	8.2	0.2	23.6	$2.64 \cdot 10^{-11}$	$6.21 \cdot 10^{-7}$	$2.7 \cdot 10^{-5}$	$2.7 \cdot 10^{-4}$	509.8	0.24
90_B	12.6	0.9	24.0						
90_C	8.8	0.2	24.2						
DO_A	5.4	0.2	26.1	$2.35 \cdot 10^{-10}$	$4.3 \cdot 10^{-7}$	$1.1 \cdot 10^{-4}$	$1.1 \cdot 10^{-3}$	432.0	0.19
DO_B	5.9	0.3	29.8						
DO_C	7.4	0.9	26.4						
DR_A	3.7	0.2	28.4	$2.36 \cdot 10^{-10}$	$3.72 \cdot 10^{-7}$	$9.4 \cdot 10^{-5}$	$1.1 \cdot 10^{-3}$	454.5	0.21
DR_B	3.4	0.2	29.8						
DR_C	5.0	0.3	29.9						

Table 2: Percentage of variation in soil N-NO₃ stocks, soil N-NO₃/N-NH₄ stocks, N-NH₄, N-N₂, N-NO_x and N-

640 N₂O emissions in response to soil [Cu] in the various incubation conditions for a 3-day modelisation.

a.

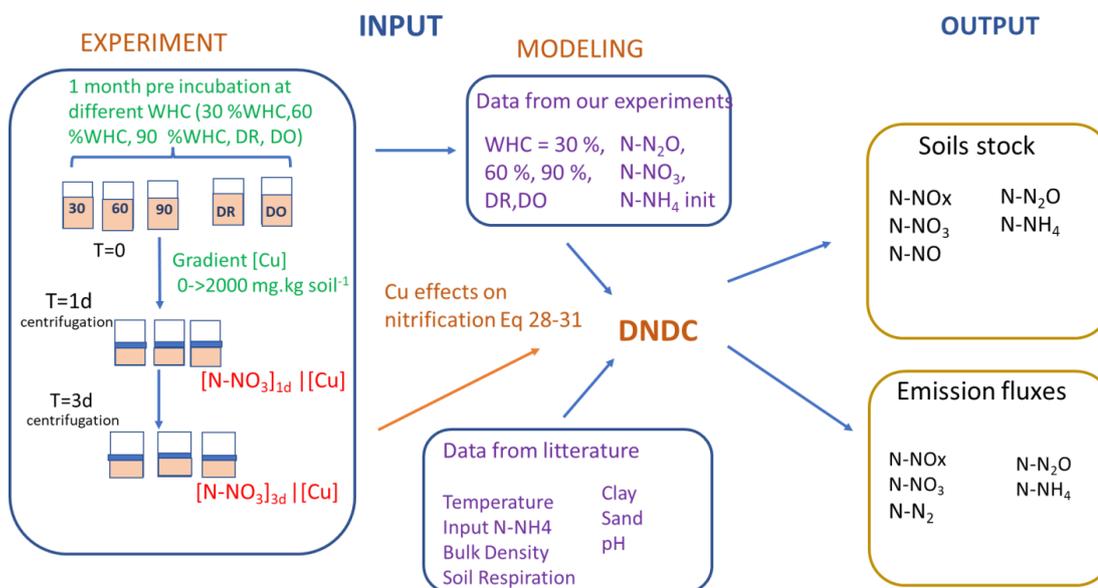
Moisture condition	Added Cu (mgCu.kg soil ⁻¹)	N-NO ₃ soils stocks	Emission N-NH ₄	Emission N-N ₂	Emission N-NO _x	Emission N-N ₂ O	Soil stocks N-NO ₃ /N-NH ₄
60	0	0.0	0.0	0.0	0.0	0.0	0.0
60	50	-1.3	0.3	-17.9	-3.5	-2.1	-1.5
60	100	-2.6	0.6	-24.4	-5.5	-4.1	-3.2
60	250	-6.7	1.5	-35.0	-10.5	-9.8	-8.0
60	500	-13.3	2.9	-45.6	-17.8	-19.0	-15.7



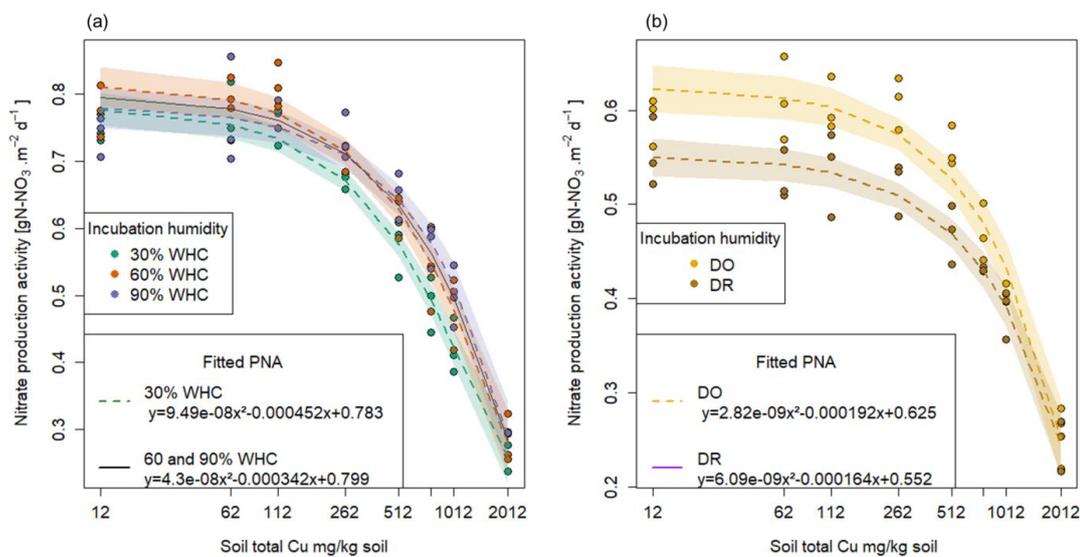
60	750	-19.5	4.3	-53.4	-24.5	-27.7	-22.8
60	1000	-25.4	5.5	-59.8	-30.6	-35.8	-29.3
60	2000	-44.5	9.7	-78.0	-50.5	-62.3	-49.4
90	0	0.0	0.0	0.0	0.0	0.0	0.0
90	50	-1.0	0.3	-16.4	-6.7	-3.1	-1.2
90	100	-2.2	0.6	-22.4	-9.4	-5.3	-2.7
90	250	-6.0	1.5	-32.3	-14.5	-11.1	-7.3
90	500	-12.1	3.0	-42.7	-20.8	-20.1	-14.7
90	750	-18.0	4.5	-50.7	-26.1	-28.4	-21.5
90	1000	-23.6	5.8	-57.4	-30.8	-36.2	-27.8
90	2000	-41.8	10.3	-76.4	-46.0	-61.6	-47.2

b.

Moisture condition	Added Cu (mgCu.kg soil ⁻¹)	NO ₃ soil stocks	Emission NH ₄	Emission N ₂	Emission NOx	Emission N ₂ O	Stocks NO ₃ /NH ₄
DO	0	0.0	0.0	0.0	0.0	0.0	0.0
DO	50	-0.7	0.2	-17.7	-3.2	-1.7	-0.8
DO	100	-1.5	0.3	-23.9	-4.8	-3.2	-1.8
DO	250	-3.9	0.8	-33.5	-8.4	-7.6	-4.7
DO	500	-8.1	1.7	-42.8	-13.6	-14.8	-9.6
DO	750	-12.3	2.6	-49.8	-18.4	-22.1	-14.5
DO	1000	-16.5	3.5	-55.8	-23.1	-29.3	-19.3
DO	2000	-33.3	7.0	-75.7	-41.6	-58.3	-37.7
DR	0	0.0	0.0	0.0	0.0	0.0	0.0
DR	50	-0.6	0.1	-17.6	-3.6	-1.6	-0.7
DR	100	-1.3	0.3	-23.8	-5.3	-3.1	-1.6
DR	250	-3.5	0.7	-33.3	-9.1	-7.3	-4.2
DR	500	-7.2	1.4	-42.4	-14.2	-14.3	-8.6
DR	750	-10.9	2.2	-49.1	-19.0	-21.2	-12.8
DR	1000	-14.5	2.9	-54.8	-23.5	-27.9	-16.9
DR	2000	-28.6	5.7	-73.2	-40.7	-54.1	-32.5

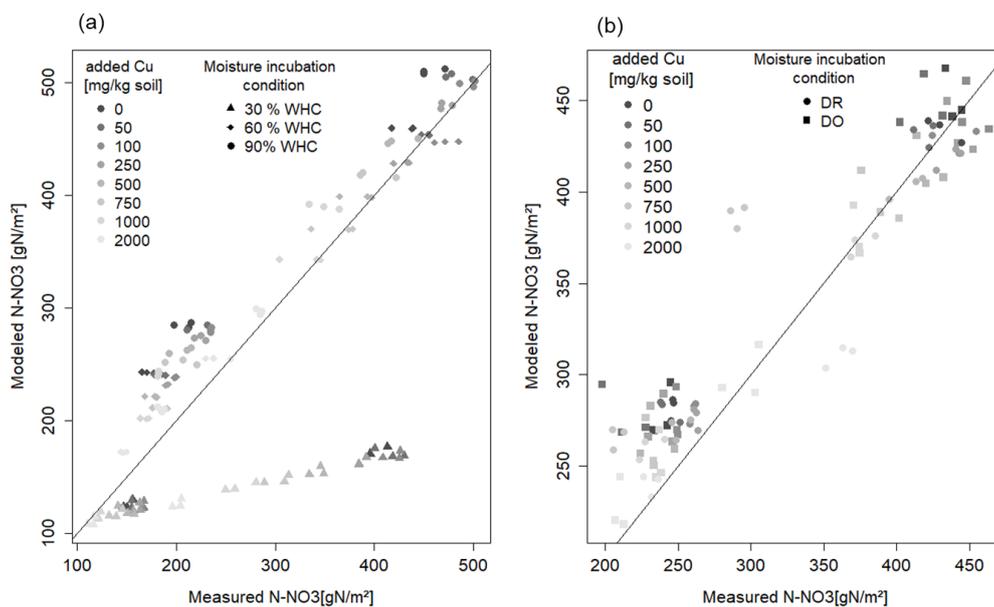


650 **Fig. 1:** Schematic representation of experimental and modelling procedures. Left refers to the experimental part
 and center to right to the modelling part. Soils were first pre-incubated 5 weeks at different WHC. N-NO_2 , N-NO_3
 and N-NH_4 soil concentrations were then measured after this preincubation, and values were used to initialise
 DNDC. A bioassay was then applied on soil aliquots. Copper (Cu) was spiked at 0, 50, 100, 250, 500, 750, 1000,
 2000 $\text{mg Cu.kg soil}^{-1}$ of soil to reach concentrations of 12, 62, 112, 262, 512, 762, 1012 and 2012 $\text{mg Cu.kg soil}^{-1}$
 655 $^{-1}$ and left for incubation. After 1 and 3 days of incubation, N-NO_3 production was measured in the supernatant.
 N-NO_3 productions against [Cu] gradients were used to define the functions of eq. 28 to 31 in §3.1. Soil respiration
 values were extracted from the curve C_i of Fig 1 in Annabi et al. (2007).



660 **Fig. 2:** Fitted functions of potential nitrifying activities against soil [Cu] for each moisture incubation condition.
Points are the measured nitrate production and lines the fitted quadratic function with their 95% confidence interval.
(a). Constant moisture treatments: green is for 30% WHC, red for 60% WHC and purple for 90% WHC. The black
line is the common fitting function used for 60 and 90% WHC incubation conditions. (b). Variable moisture
treatments: brown is for Dry-rewetting (DR) and yellow for dry only (DO).

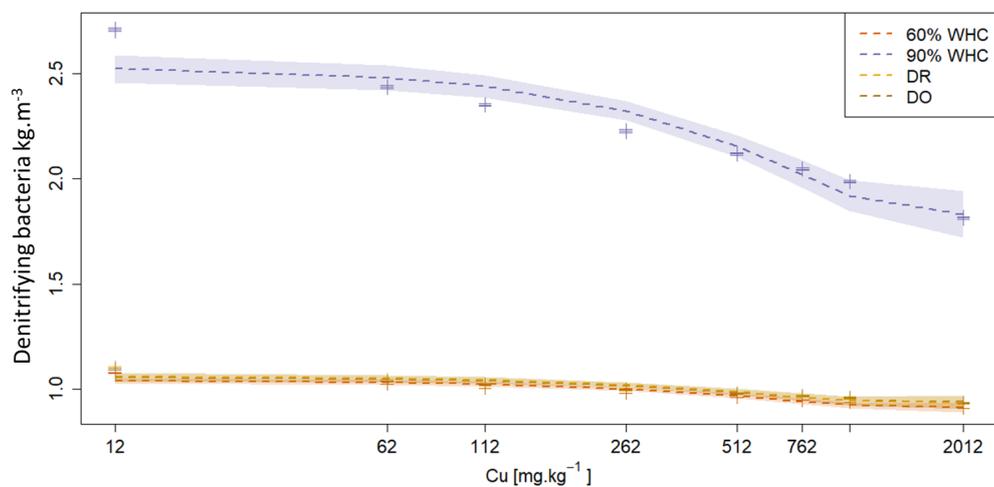
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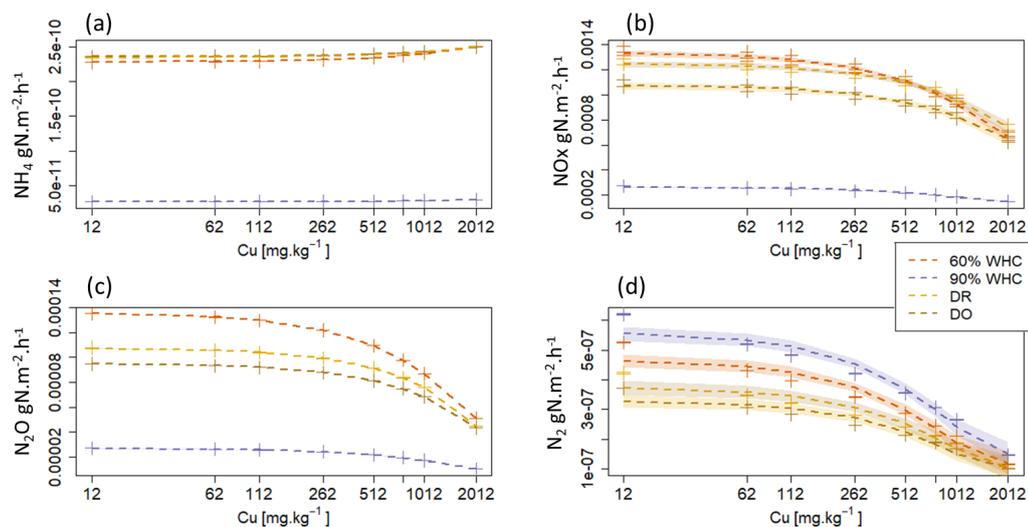
Fig 3: Comparison of modeled against measured soil [nitrate] incubated in different moisture. (a) = 3 incubations under constant humidity. (b) = Dry-rewetting (DR) and Dry only (DO) conditions. For 30% WHC, Model=1.84 * Measure and R2=0.93; for 60% WHC Model=0.93 *measure. R2=0.99; for 90% WHC Model=0.90 * measure. R2=0.99 ; for Dry -rewetting (DR) model = 0.96*measure. R2=0.98; for Dry only

675 (DO)Model=0.95*measure. R2=0.99

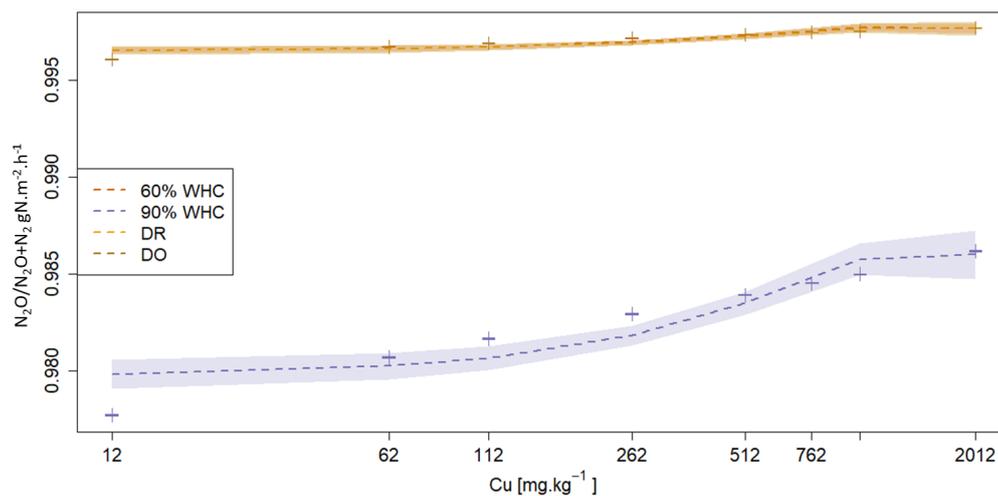


680 Fig 4. Modelled soil denitrifying bacterial pool after 3 days (gN.m⁻³ soil) for the 4 moisture treatments. Purple is
for 90% WHC, red for 60% WHC, brown for dry rewetting (DR) and yellow for dry only (DO). Red, brown and
yellow curves being superposed. Pools were modelled for 12, 62, 112, 262, 512, 762, 1012 and 2012 mg Cu,kg
soil⁻¹ as represented by cross. Quadratic fits were used for representation.

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690 Fig 5: Modelled N emission fluxes at 3 days in $\text{gN.m}^{-2}.30\text{min}$ under the different moisture conditions. a.) N- NH_4 emission fluxes. b.) N- NO_x emission fluxes c.) N- N_2O emission fluxes and d.) N- N_2 emission fluxes. Purple is for 90% WHC, red for 60% WHC, brown for dry rewetting (DR) and yellow for dry only (DO). Fluxes were modelled for 12, 62, 112, 262, 512, 762, 1012 and 2012 mg Cu.kg^{-1} soil $^{-1}$ as represented by cross. Quadratic fits were used for representation.



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Fig. 6: Proportion of N_2O emitted arising from the denitrification calculated as $N_2O/(N_2O+N_2)$ modelled fluxes in response to soil Cu concentration for the various moisture conditions. Red is for 60 % WHC, purple is for 90% WHC, yellow for the DR and brown for the DO. Red, yellow and brown curves are superposed. Fluxes were modelled for 12, 62, 112, 262, 512, 762, 1012 and 2012 $mg \cdot kg^{-1}$ as represented by cross. Quadratic fits were used for representation.

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