



Significant methane undersaturation during austral summer in the Ross Sea (Southern Ocean)

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Abstract. Dissolved methane (CH₄) was measured at 9 stations along a transect at 75°S in the Ross Sea during austral summer in January 2020. CH₄ undersaturation (mean: 82±20%) was found in the water column, with a mean air-sea CH₄ flux density of -0.58±0.48 μmol m⁻² day⁻¹, which suggests that the Ross Sea was a net sink for atmospheric CH₄ during the austral summer. Simple box-model calculations revealed that the CH₄ depletion should occur in the surface mixed layer because of CH₄ oxidation and advection of CH₄-poor waters. We propose that freshwater injection caused by sea-ice melting in summer dilutes CH₄ concentrations within the surface layer and thus increases its potential for atmospheric CH₄ uptake in the Ross Sea. Thus, we argue that both CH₄ consumption and sea-ice melting are important drivers of CH₄ undersaturation, which implies that the high-latitude area of the Southern Ocean is a sink for atmospheric CH₄. We estimated that the Southern Ocean (>65°S) takes up about 0.02% of the global CH₄ emissions and thus represents a minor sink for atmospheric CH₄.

1 Introduction

Methane (CH₄) is one of the most important greenhouse gases. The dry mole fractions of CH₄ in the atmosphere have increased continuously since the onset of the industrial revolution, contributing more than 20% of the anthropogenic radiative forcing in the lower atmosphere (IPCC, 2021; Saunio *et al.*, 2020). In addition, the ice-covered parts of the Antarctic are now known to be a reservoir of organic carbon (Prisco *et al.*, 2008) which might fuel CH₄ production, suggesting that the Antarctic ice sheet may be a neglected but important component of the global CH₄ budget (Wadham *et al.*, 2012). However, the very limited number of direct observations of CH₄ in the water column of the Southern Ocean



30 constrains our ability to assess oceanic CH₄ dynamics and emissions. Hence, more field data should be collected to decipher the role of the Southern Ocean in global CH₄ cycling.

The Ross Sea (Figure 1a), located along the Antarctic margin between Cape Adare and Cape Colbeck, is one of the largest consistently forming polynyas in the Southern Ocean (*Arrigo and Dijken*, 2004). The shelf is usually free of sea ice during summertime in the Antarctic (December to February). During winter, winds blow northward along the Ross Ice Shelf and form discrete coastal polynyas and areas of exposed surface water surrounded by dense multiyear ice (*Orsi and Wiederwohl*, 2009). Evidence for the presence of gas hydrates in the western Ross Sea was inferred from a bottom-simulating reflector, indicating a potential source of CH₄ under the seabed (*Geletti and Buseti*, 2011). However, only a few studies have addressed the distribution of CH₄ in the water column of the Southern Ocean, and these studies have rarely focused on the Ross Sea. In the 1970s, surface CH₄ concentrations were investigated at several locations in the vicinity of the Ross Sea, showing a sink for atmospheric CH₄ that was explained by the microbial consumption of CH₄ (*Lamontagne et al.*, 1974). The CH₄ distribution around the Antarctic Peninsula and in the Weddell Sea was investigated in the 1990s and 2000s, suggesting a net uptake of CH₄ in open ocean water, which was controlled by mixing between surface water and deep water in which microbial consumption decreased the CH₄ content (*Heeschen et al.*, 2004; *Tilbrook and Karl*, 1994). Moreover, CH₄ undersaturation was observed in high-latitude areas (>65°S) of the Southern Ocean, except for the circumpolar areas (Pacific sector of the Southern Ocean) where the surface water showed a minor source of atmospheric CH₄ due to local CH₄ release in the water column (*Yoshida et al.*, 2011; *Bui et al.*, 2018).

Sea ice can be a physical obstruction for air-sea exchange of CH₄, and the storage of CH₄ in ice is usually a minor source compared to brine and under-ice seawater (*Crabeck et al.*, 2014; *Zhou et al.*, 2014). Thus, areas such as the Arctic Ocean have been found to produce short-term pulses of CH₄ emissions during the melting season (*Thornton et al.*, 2016). In contrast, the CH₄-depleted regions of the Southern Ocean, such as the Ross Sea, may act as CH₄ sinks after freshwater injection due to ice melting, which is similar to the uptake of atmospheric CO₂ in the ice-free polar seas (*Rysgaard et al.*, 2011; *Sandrini et al.*, 2007). However, more observations are needed to understand the CH₄ variability in the Southern Ocean. Here, on the basis of our results, we propose (i) that the mixing of water masses controls the CH₄ distribution in the Ross Sea and (ii) that sea-ice melting enhances the ability of surface water to take up atmospheric CH₄ during summertime.

55 2 Materials and Methods

2.1 Hydrographic conditions and water mass classification

Five water masses were identified in the Ross Sea according to their temperature/salinity properties (Figure 2a; Table S1) and the definitions used in previous studies (*Narayanan et al.*, 2019; *Orsi and Wiederwohl*, 2009; *Schodlok et al.*, 2016; *Zoccarato et al.*, 2016). Antarctic Surface Water (AASW) carried by coastal currents (blue arrow in Figure 1a) flows westward along the outer ice shelf, occupying the surface layer (<100 m) of the water column. It is a cold, fresh, and oxygen-rich layer due to melting sea ice at the surface and sufficient contact with the atmosphere. Shelf Water (SW) is formed



during winter when brine rejection occurs due to sea-ice formation, causing the local surface water to become dense and sink to the bottom along the slope. A thermohaline mechanism drives the overturning circulation of SW in the ice cavity; therefore, SW experiences extra cooling from the ice cavity and becomes Ice Shelf Water (ISW) when flowing off the ice sheet (green arrow in Figure 1a), which has a below-freezing point temperature (Lewis and Perkin, 1986). Warmer, saltier, and oxygen-poor Circumpolar Deep Water (CDW) carried by the Antarctic Circumpolar Current intrudes into the edge of the Ross Sea. Afterward, the Antarctic Slope Current (red arrow in Figure 1a) brings it to be mixed with AASW, which then reaches farther south and becomes Modified Circumpolar Deep Water (MCDW) (red dashed arrow in Figure 1a) (Orsi and Wiederwohl, 2009). In addition, Modified Shelf Water (MSW) is formed when near-freezing SW is vertically mixed with relatively warm MCDW, which is the classic mechanism to produce Antarctic Bottom Water (Assmann and Timmermann, 2005).

The relative abundance of a specific water mass at each station was estimated by the percentage of local volume following the description by Orsi and Wiederwohl (2009). Briefly, the spatial distribution of the potential temperature (θ)-salinity diagram provides the thickness of a certain water mass. Assuming all stations had the same sectional area, the volume of each water mass at a specific station was simply determined by the depth at which the water mass was encountered (Table S2). The volumetric percentages of each water mass (Table S3) were then calculated from the total volume divided by the volume of a specific water mass at a specific station.

2.2 Sampling and analysis

Seawater samples for CH₄ analysis were collected onboard R/V “Xuelong 2” during the 36th Chinese Antarctic Research Expedition (CHINARE) to the Ross Sea from 3–6 January 2020. The CH₄ distribution was measured at nine out of ten stations on a transect along 75°S from 164°E–182°E. The samples were collected with a rosette water sampler equipped with Niskin bottles, following the sampling strategies described by Zhan *et al.* (2018). Briefly, seawater was transferred to 250 mL borosilicate glass bottles with standard taper stoppers (Corning PYREX[®], USA), which were sealed with Apiezon grease (Sigma-Aldrich[®], USA) and stored in the dark at 4 °C after 180 μL of saturated HgCl₂(aq) solution was added. Samples were analyzed immediately after shipment to the home laboratory (storage time < six months). Hydrographic data were collected with a SBE-911plus conductivity-temperature-depth (CTD) unit (Sea-Bird, USA) that measured the salinity (± 0.002 psu), temperature (± 0.001 °C), and pressure (0.015% of full-scale range). Triplicate or duplicate subsamples were dispensed to 20 mL glass vials prior to analysis. CH₄ was extracted from the subsamples by a purge-and-trap method (Text S1; Table S4–S5; Figure S1), and its dry mole fraction (water vapor was removed by a Nafion[™] dryer) was determined with a gas chromatograph (Agilent 7890A) equipped with a flame ionization detector (Zhang *et al.*, 2004). CH₄ saturation (Sat, %) in seawater is given by the following equation:

$$\text{Sat} = \frac{C_{\text{obs}}}{C_{\text{eq}}} \cdot 100\% \quad (1)$$



where C_{obs} is the observed concentration of CH_4 in seawater and C_{eq} is the concentration in equilibrium with the atmosphere calculated from the solubility coefficient (Wiesenburg and Guinasso, 1979) with the *in-situ* temperature (1.9–1.4 °C) and salinity (33.68–34.84) at the time of sampling and the atmospheric CH_4 dry mole fraction of 1.82 ppm (monthly mean during the sampling period at the South Pole Observatory, available from NOAA Global Monitoring Laboratory, Dlugokencky *et al.*, 2020). Using a contemporaneous atmospheric CH_4 mole fraction for all water depths led to an underestimation of less than 1% for the equilibrium concentration of dissolved CH_4 in view of the residence time of about 4 years of the Ross Sea shelf water (Trumbore *et al.*, 1991). The overall mean analytical error of C_{obs} was $\pm 0.3 \text{ nmol L}^{-1}$, with a corresponding mean error of $\pm 10\%$ for the CH_4 saturations.

2.3 Sampling and analysis

The flux density (F_{ase} , $\mu\text{mol m}^{-2} \text{ day}^{-1}$) of CH_4 across the air-sea interface was estimated by the following equation:

$$F_{ase} = k_w \cdot (C_{obs} - C_{eq}) \quad (2)$$

where k_w is the gas transfer coefficient (m day^{-1}) and C_{obs} is the measurement from the uppermost sampling depth (2–5 m). We used an empirical model to determine k_w , which is a function of the molecular diffusivity of CH_4 in water (Jähne *et al.*, 1987), the kinematic viscosity of water, and the wind speed (Wanninkhof, 2014). The molecular diffusivity and kinematic viscosity, in turn, were computed as functions of the temperature and salinity. We used the average of the daily wind speeds over the Ross Sea during the sampling period ($5.8 \pm 0.1 \text{ m s}^{-1}$) which was obtained from the National Centers for Environmental Prediction (NCEP) and the National Center for Atmospheric Research (NCAR) reanalysis data (<http://www.psl.noaa.gov/data/gridded/data.ncep.reanalysis.derived.html>). This regional mean wind speed was in the same range ($5.2\text{--}6.3 \text{ m s}^{-1}$) as that reported from the Weddell Sea (Tilbrook and Karl, 1994) but lower than the 30-year January average (6.8 m s^{-1}) from the Southern Ocean south of 50°S (Bui *et al.*, 2018).

2.4 CH_4 budget in the mixed layer

In order to roughly estimate the CH_4 production/consumption in the mixed layer, we performed a mass balance calculation of CH_4 fluxes into and out of the surface mixed layer. The mixed layer was defined as the water depth where a potential density difference of 0.125 kg m^{-3} and a temperature difference of $0.5 \text{ }^\circ\text{C}$ in comparison with the ocean surface was computed (de Boyer Montégut *et al.*, 2004). The CH_4 fluxes in the surface mixed layer should be balanced and can be expressed by the following equation:

$$-F_{ase} + F_{vd} + F_{ad} + F_x = 0 \quad (3)$$

where F_{ase} ($\mu\text{mol m}^{-2} \text{ day}^{-1}$) is the air-sea exchange of CH_4 across the ocean/atmosphere interface obtained from equation (2). The F_{vd} ($\mu\text{mol m}^{-2} \text{ day}^{-1}$) is the vertical (diapycnal) diffusion of CH_4 , which can be obtained from Fick's First law: $F_{vd} = K_z \cdot \frac{dc}{dh}$, h is depth (m), d/d_h is the vertical CH_4 gradient measured at each station, K_z ($\text{m}^2 \text{ s}^{-1}$) is the diapycnal diffusivity. We used a K_z of $10^{-4} \text{ m}^2 \text{ s}^{-1}$ which was suggested to be representative for the Southern Ocean (Mashayek *et al.*,



2017). F_{ad} ($\mu\text{mol m}^{-2} \text{day}^{-1}$) is the lateral CH_4 transport (advection) into/out of the mixed layer which we assumed in a first
125 approximation to be negligible. F_x ($\mu\text{mol m}^{-2} \text{day}^{-1}$) is the unknown in-situ source/sink (such as production/oxidation) in the
mixed layer, which was estimated by the balance of other flux terms. Positive values represent a CH_4 source in the mixed
layer while negative values represent a CH_4 sink in the mixed layer.

3 Results

3.1 Section distribution of temperature, salinity, and CH_4

130 The water column was stratified within the upper 50 m, with density differences ($\Delta\sigma$) of 0.185–0.522 kg m^{-3} among
all stations. Other than the influence of wind, temperature differences ($0.04 \text{ }^\circ\text{C m}^{-1}$) contributed partly to the stratification of
waters in the western part of the study area (R1–R3) (Figure 1b). In contrast, the westward intrusion of warm AASW could
have partly contributed to the water stratification in the middle and eastern region of the study area. For example, an increase
in temperature ($0.01 \text{ }^\circ\text{C m}^{-1}$, 0–25 m) contributed the most to the stratified water at station R5 and the halocline (0.01 psu m^{-1} ,
135 0–25 m) probably drove the strong stratification at the stations R9. Below 50 m, uniform water ($<0.0001 \text{ }^\circ\text{C m}^{-1}$ and <0.0002
 psu m^{-1}) was found over the west of Mawson Bank with slightly heavier water in the trough. Lateral transport of water
masses by the slope current occurred in the edge of the Ross Sea, where a southward intrusion of saline MCDW was
observed during our cruise (a decrease in MCDW contribution from station R10 (300 m) to R7 (100 m), Figure 2b).
Consequently, the water column was vertically classified into three layers (AASW, SW/MSW, and MCDW) according to
140 their hydrographic features.

Generally, the seawater was undersaturated in CH_4 (mean \pm std: $82\pm 20\%$), with a relatively high mean CH_4 saturation
($88\pm 19\%$) near the coast (stations R1–R5) compared with that in the offshore areas ($74\pm 19\%$, stations R6–R10). In the west,
waters in the upper 50 m had the highest CH_4 saturation, with a mean of $95\pm 20\%$. Below 50 m, CH_4 saturation decreased to
 $86\pm 18\%$ and $84\pm 12\%$ at depths of 50–400 m and below 400 m, respectively. In the east, station R10 (where AASW
145 dominated the water column) had the lowest CH_4 saturation ($63\pm 8\%$) compared to those at the other stations. The highest
 CH_4 saturation (147%) was found at station R7 at 200 m together with an anomalous warm-water mass ($-1.0 \text{ }^\circ\text{C}$ compared
with an ambient temperature of $-1.7 \text{ }^\circ\text{C}$) at the same depth.

3.2 Water mass distribution

SW was the dominant water mass among the study areas west of 175°E , accounting for 60–90% of the total water
150 volume (Figure 2b). It formed near Victoria Land, and its contribution decreased to the east. With its high salinity, SW
would be expected to be trapped in deep troughs ($>400 \text{ m}$) and not flow out to the Ross Sea Basin ($<2\%$ east of the Joides
Trough). In contrast, MSW increased from $\sim 5\%$ in the west to 20–50% in the east. Correspondingly, AASW and MCDW
were distributed in the eastern part of the shelf, with the largest proportions of AASW and MCDW at station R10 (65% for
AASW and 34% for MCDW) and the lowest proportions at stations R1–R3 (0–6% for AASW and 0–5% for MCDW).



155 Station R9 was likely located in the most heterogeneity region, which consisted of near equal contributions of AASW, MSW, and MCDW. A small fraction of ISW (10–19%) was found only near the ice sheet, where supercooled water could be formed in the cavity below the ice sheet.

3.3 Sea ice distribution and surface CH₄

160 During the austral summer of 2019/2020, the sea ice in the Ross Sea began to melt in December near the western coast of the Ross Ice Shelf (Figure 3a). Afterward, the ice-covered areas decreased rapidly in January, and the western part of the Ross Sea became an ice-free area in February. We found that during our sampling period (January 2020), the Pennell Bank (stations R6–R10) had become an ice-free area (which may have already endured a 30-day ice-free period by then, Figure 3), while sea ice was still present at the Mawson Bank. Lower CH₄ saturation at the surface (76±13%) was found at the Pennell Bank while near-equilibrium saturation (91±8%) was found at the Mawson Bank (stations R1–R5, ice-covered area:
165 50±30%).

Different endpoints were chosen to quantify the factors that controlled CH₄ undersaturation (Figure 3e). Considering the water stratification and relatively slow gas exchange with the atmosphere (4 yr residence time for SW, Trumbore *et al.*, 1991), we assumed that the CH₄ saturation of 84% (mean saturations as derived from our measurements below 400 m water depth) in deep waters (>400 m) could represent the endpoint in the SW in early summer. In addition, the sediment-
170 influenced water at station R7 was assumed to be a CH₄ source. We found that mixing among AASW, SW, and CDW was responsible for the CH₄ undersaturation in the shelf area. A total decrease of 50% ice-covered area drove a decrease in surface CH₄ saturation by an average of 22% (from 91±8% to 76±13%).

3.4 Box-model calculation

175 The surface mixed layer was 25 m at stations R1–R6 (in the west) and 50 m at R7–R10 (in the east) and therefore we calculated two box models for comparison. At stations R1–R6, the air-sea exchange accounts for a mean uptake of 0.37 μmol m⁻² day⁻¹ from the atmosphere, the mean vertical diffusion rates ranged from -0.21 to 0.62 μmol m⁻² day⁻¹, with a mean value of 0.17 μmol m⁻² day⁻¹. If we assume that the lateral transport of CH₄ is zero, the F_x was estimated as -0.54 μmol m⁻² day⁻¹. For stations R7–R10, because we only have vertical profile CH₄ data at stations R9 and R10, we calculated the F_x to be -0.83 μmol m⁻² day⁻¹ for station R9 and -1.52 μmol m⁻² day⁻¹ for station R10, respectively, which was 2–3 times higher than that at
180 stations R1–R6.

4 Discussion

4.1 Role of water mass mixing on CH₄ distribution

The fate of CH₄ in the Ross Sea mainly results from mixing, air-sea exchange, and microbial consumption. Previous studies have indicated that microbial oxidation of CH₄ during deep water transport results in CH₄ undersaturation and causes



185 CH₄-depleted surface waters after an upwelling of the deep water in the Southern Ocean (Tilbrook and Karl, 1994; Heeschen
et al., 2004). Hence, the CH₄-poor CDW may play important role in contributing to the lowest CH₄ saturation at the edge of
the Ross Sea. MCDW, which is fed by CDW, contributes to CH₄ undersaturation by mixing with MSW (Figure 3e). In
addition, AASW, entrained by the Coastal Current (Figure 1a) and flowing progressively westward along the Ross Sea shelf,
results in a decreasing CH₄ gradient from west to east in the upper 200 m (Figure 2c). Within these depths, negative (positive)
190 correlations between CH₄ saturation and the percentage of AASW (SW+MSW) are found at all stations (Figure S2). Hence,
after the effect of microbial oxidation of CH₄, mixing between different water masses determines the distribution of CH₄
undersaturation in the Ross Sea. In contrast, the oversaturation of CH₄ (106±5%) found in this study, including the near-
equilibrium saturation (e.g., stations R1–R5), may originate from surface water that is sufficiently exposed to the air in
summer and subsequently concentrated and transported due to ice formation in winter.

195 At the surface (<100 m), AASW is the main source for water undersaturated in CH₄ in the Ross Sea, which is probably
caused by dilution with melted sea ice. During austral winter (June–August), sea-ice formation in the Ross Sea converts
local upper waters (i.e., AASW and MCDW) into SW with a high density, which in turn sinks along the continental slope
and produces a well-mixed water column (Oris and Wiederwohl, 2009). Hence, the dissolved CH₄ in surface water during
winter would be transported to deep water and involved in long-term cycles that would be modified by mixing, oxidation,
200 and supplementation from sedimentary sources. When sea ice melts in summer, seawater with undersaturated CH₄
concentrations then continues to be diluted by melting freshwater, which in turn leads to a continuous decrease in the CH₄
saturation within the surface layers; this situation is counteracted by the uptake of CH₄ from the overlying atmosphere. For
example, we found that the CH₄ saturation was decreased by 22% at the Pennell Bank (due to the complete melting of sea ice)
compared to that at the Mawson Bank (due to slow changes in sea-ice cover and uptake of CH₄ from the atmosphere) in the
205 Ross Sea (Figure 3d). As the ice-free areas increase, CH₄ saturation changes dynamically in the seawater due to mixing
between water masses and/or exchange with the atmosphere (Figure 3e). Thus, the magnitude of sea-ice melting may
determine the degree of CH₄ saturation in the surface water and strengthen the capability of the Ross Sea to take up CH₄
from the atmosphere.

4.2 CH₄ removal in the mixed layer

210 The surface mixed layer plays an important role in connecting the atmosphere and the deep ocean, determining the
CH₄ cycling in the Ross Sea. Our box-model results indicate that the changes in the CH₄ concentrations of the mixed layer
were influenced by air-sea exchange (40–70%) in the west and increased its contribution (>90%) in the east. In contrast, the
vertical diffusion of CH₄ was a source (upward flux) to the mixed layer in the west (except for station R1) but changed to a
sink (downward flux) in the east. The sea ice distribution may responsible for the differences in the contribution of F_{ase} and
215 F_{vd} . In the east, the sea ice was melting or incompletely melting during our sampling period. The physical mixing between
surface water and injected freshwater causes (1) a rapid decline in CH₄ concentration at the surface and thus resulting in an
uptake of CH₄ from the atmosphere and (2) leads vertical diffusion of CH₄ from deeper waters into the mixed layer. In the



east, after completely melting of sea ice and expose to the air for a 30-day ice-free period (Sect. 3.3), CH₄ concentrations in the mixed layer exceeded the CH₄ concentrations in deeper waters due to continuously uptake of CH₄ from the atmosphere and thus CH₄ began to diffuse into the deeper waters.

The continuous uptake of CH₄ in the surface water (from the initiation of sea-ice melting to a month after the ice-free period) suggests the presence of CH₄ removal process(es) in the mixed layer, which is reflected by the negative values of F_x calculated from our box model. A previous study reported a microbial oxidation rate of CH₄ for the Weddell Sea (Southern Ocean) to range from 0.34 to 1.03 $\mu\text{mol m}^{-2} \text{day}^{-1}$, which is comparable to our results. Hence, we argue that CH₄ oxidation is an important driver to reinforce the CH₄ undersaturation in the Ross Sea. In addition, the CH₄-consumption in the mixed layer were 2-3 times higher in the east than that in the west. Recalling that the contribution of the CDW was decreased from the east to the west (Figure 2b) with a relatively low CH₄ (Figure 3e), we would expect a stronger advection of CH₄-poor waters in the east may contribute to the higher rate of CH₄ depletion. Though the CH₄ removal rates were roughly estimated, the compared results among stations showed some regulations and suggest the presence of considerable CH₄ consumption in the mixed layer, which is the prerequisite of atmospheric CH₄ uptake in the Ross Sea.

4.3 CH₄ uptake in the Southern Ocean

Unlike most oceanic areas, where supersaturation of CH₄ is found in the surface water, the Southern Ocean is characterized by significant undersaturation of CH₄ in the surface layer, which will result in a net take up of CH₄ from the atmosphere, similar to the case of carbon dioxide in the Ross Sea (Rysgaard *et al.*, 2011; Sandrini *et al.*, 2007). Almost all previous studies of CH₄ in the Southern Ocean have shown different degrees of CH₄ undersaturation in the surface layer (Table 1), except for the circumpolar area near the Adélie Coast, which is influenced by local CH₄ production (Yoshida *et al.*, 2011). We calculated a mean CH₄ flux density of $-0.54 \pm 0.48 \mu\text{mol m}^{-2} \text{day}^{-1}$ (with a range from -1.56 to $-0.13 \mu\text{mol m}^{-2} \text{day}^{-1}$), which indicates that the Ross Sea was a sink for atmospheric CH₄ at the time of sampling, in line with previous studies (Table 1). Despite the increase in the dry mole fractions of atmospheric CH₄ during the past decades, our results indeed show that the high-latitude regions of the Southern Ocean still maintain the potential to take up atmospheric CH₄.

The overall CH₄ uptake by the Ross Sea (extrapolated to a surface area of $9.6 \times 10^5 \text{ km}^2$) was 0.001 Tg CH₄ during the austral summer (December–February). Thus, the Ross Sea counteracted 0.01–0.02% of the global oceanic CH₄ release of 6–12 Tg CH₄ yr⁻¹ (Weber *et al.*, 2019). Extrapolating this result to the high-latitude region (>65°S) of the Southern Ocean (Table 1), which accounts for ~6% of the global ocean surface area (www.ngdc.noaa.gov/mgg/global/etopo1_ocean_volumes.html), a total of 0.02 Tg CH₄ (accounting for 0.2–0.3% of the oceanic CH₄ release) was calculated to be taken up during the three ice-free months by the Southern Ocean surface seawater after sea-ice melting in summer (Figure 3a–c). Notably, however, the variability in wind speed could substantially affect this very rough estimate. For example, it is suggested that moderate winds may contribute to the slow equilibration, where surface seawater is undersaturated with CH₄ (Heeschen *et al.*, 2004). Here, we use the daily average wind speed (NCEP product), which has previously been shown to be underestimated (Borges *et al.*, 2018), possibly resulting in a conservative



estimate of the CH₄ flux densities. Moreover, the Ross Sea was covered by a large amount of sea ice in January 2020 (Figure 3b). Therefore, if 50% sea-ice melting would decrease CH₄ saturation by 22%, we would expect an increase of 44% in the CH₄ flux density (from the atmosphere to surface water) after the total disappearance of sea ice. This means that high-latitude regions are capable of take up atmospheric CH₄ during periods of sea-ice decline and could enhance CH₄ uptake in the Southern Ocean when sea-ice retreats even farther in the future (Figure S3). However, more studies are needed to understand the extent and variability of the CH₄ uptake and how ongoing human activities affect future uptake.

5 Conclusions

Our measurements in January 2020 show significant undersaturation of CH₄ in the water column during austral summer in the Ross Sea. We find that the CH₄ oxidation may be the prerequisite of CH₄ undersaturation and that sea-ice melting is likely to enhance the degree of surface CH₄ undersaturation, which increases the capability of surface water in the Ross Sea to take up CH₄.

The distribution and air-sea flux of CH₄ has been studied in coastal seawaters but have rarely focused on the high-latitude area of the Southern Ocean. Our measurement of CH₄ in the water column of the Ross Sea shelf demonstrate, for the first time after 1972 (*Lamontagne et al.*, 1974), that the Ross Sea was a sink for atmospheric CH₄ at the time of our sampling in January 2020, which underlines the significance of the role of the Southern Ocean for both the regional and global atmospheric CH₄ budget. We anticipate that ongoing studies to be conducted during summer and winter, combined with a large spatial coverage, will lead to improvements in our understanding of the underlying mechanisms and processes as well as the magnitude of the CH₄ undersaturation in the high-latitude regions of the global ocean.

Appendix A

Supporting information Text S1, Fig. S1–S3, and Table S1–S5.

Appendix B

Dataset.

Data Availability

Data from this study are available at the National Arctic and Antarctic Data Center (NAADC) (DOI: 10.11856/SNS.D.2021.005.v0) via the link:

<https://www.chinare.org.cn/en/metadata/172181e0-9bbd-4069-998e-ddeef3b6abe8>

To obtain data from NAADC, a registered account is necessary due to the data management policy required by government and national funders. Please follow the instructions (<https://www.chinare.org.cn/en/help/using-help>) to obtain the free license and access the data. For convenience, data are also provided in the Appendix B.



280 The CH₄ flux density in Table 1 and the definition of water masses in Table S1 were extracted from published literature as indicated in the tables.

Author Contributions

W.Y., L.Z., and Y.L. designed the study with input from other coauthors; H.H. provided the hydrographic data; W.W. and J.Z. conducted the experiments; J.L. and M.W. analyzed the results. W.Y. drafted the manuscript with input from H.W.B.,
285 D.L.A.M., and all other authors.

Competing interests

The authors declare that they have no conflict of interest.

Acknowledgements

The authors wish to thank our colleagues from the Third Institute of Oceanography, MNR, China, particularly Yuanhui
290 Zhang and Derong Zhao, and the crew of R/V “Xuelong 2” for help with the sampling. We also thank colleagues from the Polar Research Institute of China for providing hydrographic data. This study was supported by the Scientific Research Foundation of the Third Institute of Oceanography, MNR (HaiSanKe2020004, 2019033, and 2018032), the Fujian Provincial Natural Science Foundation of China (2020J01102 and 2019J05147), the National Natural Science Foundation of China (42006040 and 41906193), and the Ministry of Natural Resources of the PRC (IRASCC2020–2022). We thank two
295 anonymous reviewers for their constructive comments which helped to improve the manuscript significantly.

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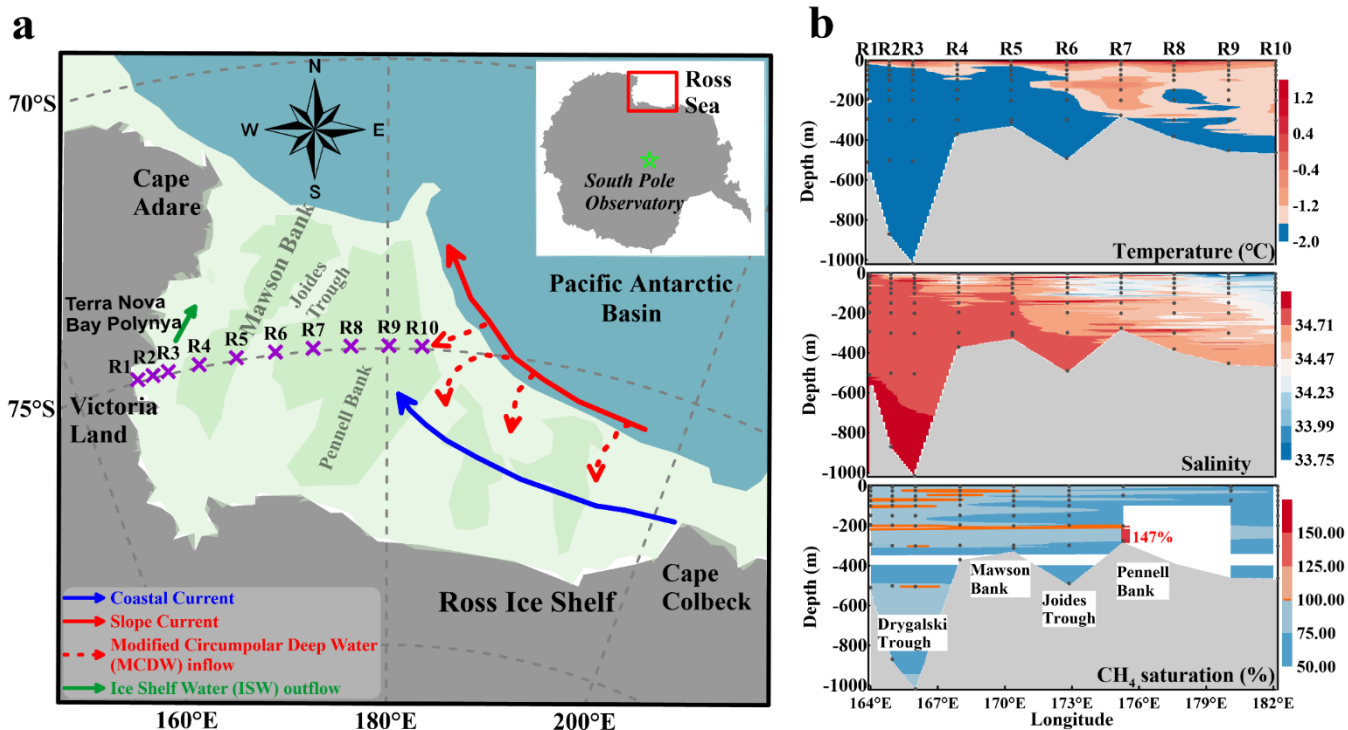


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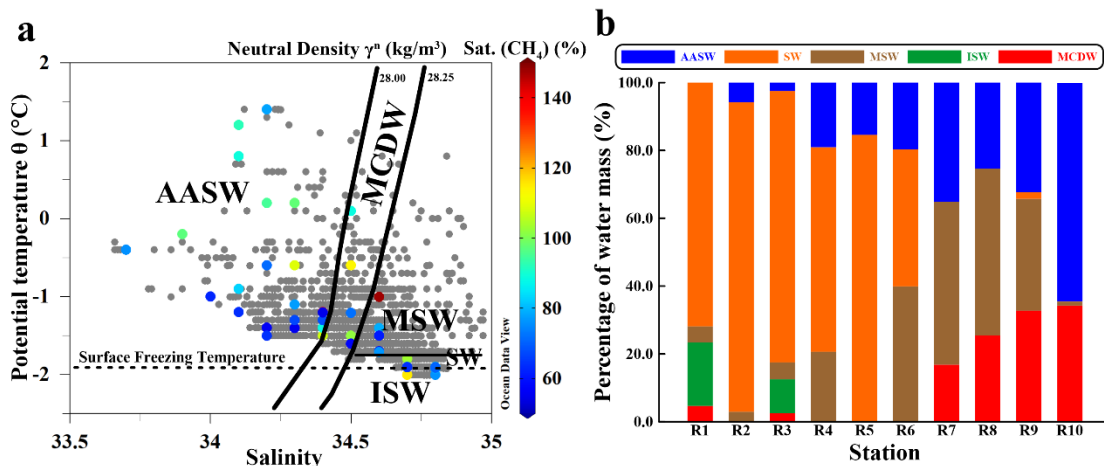
Table 1. Summary of saturation and air-sea flux densities of CH₄ in the Southern Ocean (60–75°S).

Study area	Date	Latitude	Atmospheric mol fraction, ppm	Saturation %	Flux density $\mu\text{mol m}^{-2} \text{day}^{-1}$	References
Ross Sea	Dec. 1972	70–77°S	1.36	64	-1.1*	Lamontagne <i>et al.</i> (1973)
Antarctic Peninsula (offshore)	Dec. 1986–Mar. 1987	62–64°S	1.51	87	-0.35	Tilbrook and Karl (1994)
Weddell Sea	Mar. –May 1998	60°S	1.69	75 – 94	-0.5	Heeschen <i>et al.</i> (2004)
Adélie Coast	Dec. 2001–Feb. 2002	60–66°S	1.71	87 – 130	-1.0–1.2*	Yoshida <i>et al.</i> (2011)
Indian/Pacific Ocean sectors	Dec. 2012–Feb. 2013	60–65°S	1.75	88–105	-3.9–0.5	Bui <i>et al.</i> (2018)
Ross Sea	Jan. 2020	75°S	1.82	86	-0.58	This study

420 *Values were not given but recalculated by using parameters from the specific reference. The atmospheric mixing ratio and NCEP reanalysis wind speed taken from the sampling year were used in the flux estimation.



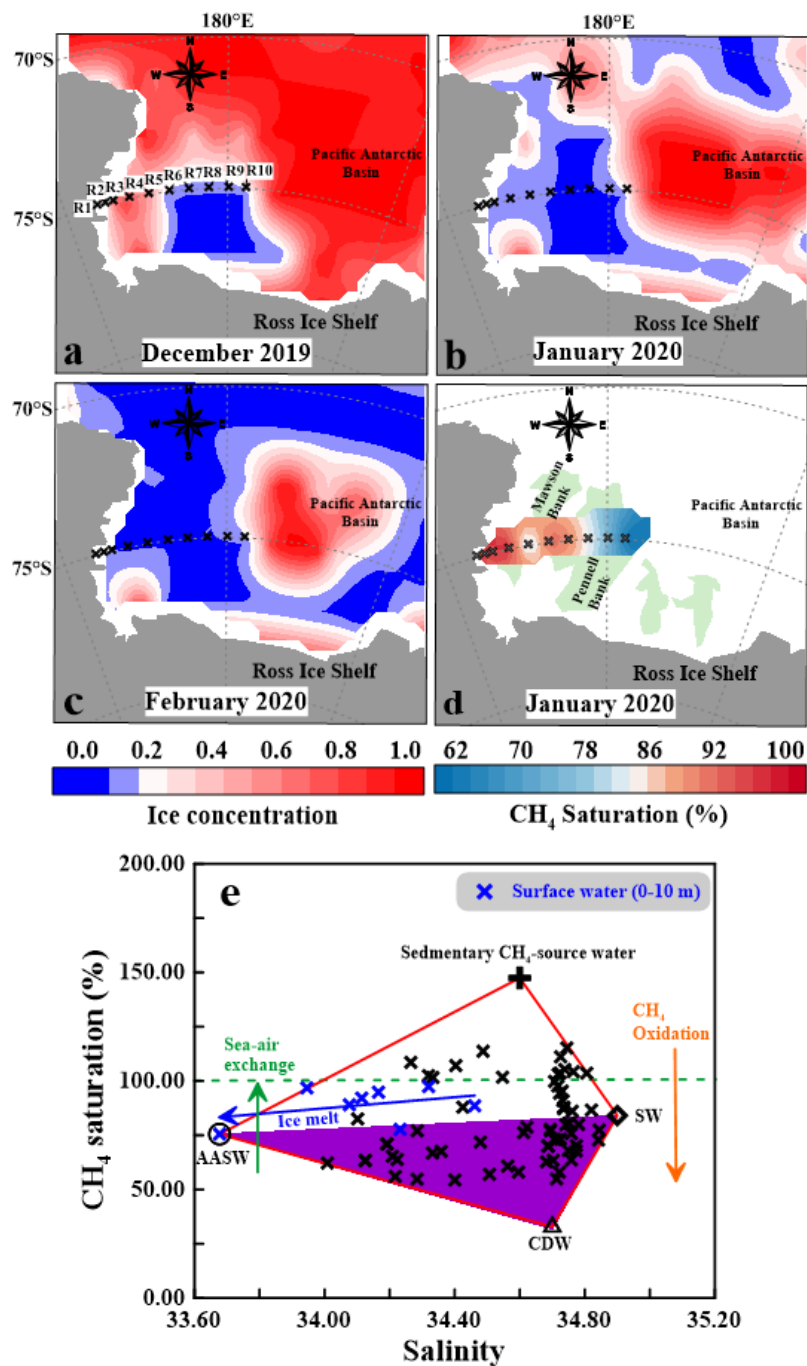
425 **Figure 1: Study location and distribution of sampling sites along the transect in the Ross Sea.** a) Map of the study area in
 the Ross Sea with an inset indicating its location in the Southern Ocean. Sampling stations are marked by purple crosses with
 the station number. The arrows represent schematic flows of currents during this study (Stewart, 2018). The light green
 shaded area represents the Ross Sea shelf, with a mean water depth of less than 500 m. The blue shaded area represents the
 outer shelf area, with a mean water depth greater than 1000 m (Marsay et al., 2017; Padman et al., 2009). Major banks are
 430 labeled on the map. The green star indicates the location of the South Pole Observatory. b) Vertical distributions of
 temperature, salinity, and CH₄ saturation along the transect in the Ross Sea. The contour line of 100% CH₄ saturation
 (equilibrium with the atmosphere) is shown by the orange line. The source point of CH₄ released from sediment is
 marked with a red contour at station R7 (see text). The white boxes indicate that no CH₄ data were available.



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Figure 2: Distribution of the main water masses in the Ross Sea during the austral summer of 2020. a) Potential temperature θ -salinity scatter plot for the sampling stations. Water masses are classified as previously described (Narayanan *et al.*, 2019; Orsi and Wiederwohl, 2009; Williams *et al.*, 2016; Schodlok *et al.*, 2016; Zoccarato *et al.*, 2016). Solid lines show the 28.00 and 28.25 kg m^{-3} neutral density γ^n surfaces (Jackett and McDougall, 1997), which are used to define Antarctic Surface Water (AASW, lower bound) and to separate the Circumpolar Deep Water (CDW) and modified Circumpolar Deep Water (MCDW) from the Antarctic Bottom Water. The dashed horizontal line shows the surface freezing temperature of seawater. Dots are highly overlapping in the range of Shelf Water (SW). b) The proportions of water masses at each station.



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Figure 3: Distribution of ice concentrations and the correlation between salinity and CH_4 saturation. a-c) Changes in monthly mean ice concentrations. d) Spatial distribution of surface CH_4 saturation values in the Ross Sea during the sampling period (January 2020). Pennell and Mawson Banks are marked with light green shading. Sea ice data were obtained from the Physical Sciences Laboratory, NOAA



450 (<http://www.psl.noaa.gov/data/gridded/data.ncep.reanalysis.derived.html>). e) Data and model simulations of factors
influencing CH₄ undersaturation on the Ross Sea shelf. Arrows represent controlling processes (blue for ice melt, green for
air-sea exchange, orange for microbial oxidation, and red for water mixing). The green dashed line represents CH₄ in
equilibrium with respect to the atmosphere. The open circle represents the endpoint of AASW that was obtained from station
R9 with the lowest salinity; the plus represents the endpoint of sediment-influenced water that was obtained from the near-
455 bottom water (200 m) at station R7; the diamond represents the endpoint of SW, which was the mean value obtained from
stations R1–R3 at depths greater than 400 m; the triangle represents the endpoint of CDW that was obtained from *Heeschen
et al.* (2004). The purple shading indicates the mixing of MCDW and MSW.