Ideas and perspectives: Emerging contours of a dynamic exogenous kerogen cycle

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Abstract. Growing evidence points to the dynamic role that kerogen is playing on Earth's surface in controlling atmospheric
chemistry over geologic time. Although quantitative constraints on the weathering of kerogen remain loose, its changing
weathering behavior modulated by the activity of glaciers suggests that this largest pool of reduced carbon on Earth may have
10 played a key part in atmospheric CO_2 variability across recent glacial-interglacial cycles and beyond. This work enunciates
the possibility of kerogen oxidation as a major driver of atmospheric CO_2 increase in the wake of glacial episodes. This
hypothesis of centennial and millennial-timescale ~~-~~relevance for this chemical weathering pathway is substantiated by several
lines of independent evidence synthesized in this contribution including the timing of atmospheric CO_2 increase, atmospheric
CO_2 isotope composition (^{13}C and ^{14}C), ~~seawater osmium record,~~ kerogen oxidation kinetics, observations of kerogen reburial,
15 and modeling results. The author hypothesizes that the deglaciation of kerogen-rich lithologies in western Canada contributed
~~in a major way~~ to the characteristic deglacial increase in atmospheric CO_2, which reached an inflection point ≤300 years after
the Laurentide Ice Sheet retreated into the kerogen-poor Canadian Shield. To quantitatively constrain the contribution of
kerogen oxidation to CO_2 rise at glacial terminations, systematic studies on CO_2 fluxes emanating from the weathering of
different lithologies, oxidation kinetics of kerogen along glacial chronosequences, and high-resolution temporal changes in the
20 aerial extent of glacially exposed lithological units are needed.

1 Introduction

Over geologic timescales, atmospheric CO_2 is controlled by the combined effects of chemical weathering of silicates and
carbonates and the organic carbon cycle (Berner et al., 1983; Berner, 1990; Torres et al., 2014). Organic carbon in the form of
kerogen[1] comprises around 15 million PgC, which encompasses over 99.9% of reduced carbon present on Earth. Tectonic
25 uplift and denudation subjects 150 PgC/kyr of kerogen to weathering on Earth's surface (Hedges and Oades, 1997) thereby
facilitating entry of this geologically ancient carbon into the atmosphere. Upon oxidation of kerogen, O_2 is consumed and CO_2
is released to the atmosphere. In reverse, biospheric organic carbon burial in marine sediments removes carbon from Earth's

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[1] In this work, kerogen is used as an umbrella term for all rock-derived forms of reduced carbon including soluble, insoluble,
rock disseminated, rock forming, solid and liquid forms, as well as fossil palynomorphs, biogenic and abiogenic graphite.

surface thereby drawing down atmospheric CO_2 and increasing O_2 over geological timescales (Galy et al., 2008). Therefore, kerogen weathering and sedimentary organic matter burial in the ocean play compensatory roles in governing atmospheric chemistry (Fig. 1) with kerogen oxidation considered important for atmospheric chemistry over million-year timescales (e.g., Petsch, 2014; Bolton et al., 2006).

However, the decay of kerogen on Earth's surface is incomplete (e.g., Hemingway et al., 2018; Leythaeuser, 1973), with physical erosion followed by riverine transport (Galy et al., 2015) and reburial in lacustrine and marine settings (e.g., Blattmann et al., 2018a; Blattmann et al., 2019b; Cui et al., 2016; Sackett et al., 1974; Sparkes et al., 2020). The operation of this "simple cycle besides the more complicated common circulation of carbon", enunciated by Sauramo (1938), begs the questions (i) what is the reburial efficiency of kerogen, (ii) what is the weathering efficiency of kerogen, (iii) what are their controlling factors, and (iv) how do reburial and weathering efficiency vary over geologic time and space? The answers presented will suggest that the exogeneous kerogen cycle behaves dynamically with this contribution hypothesizing a connection between kerogen oxidation and millennial/centennial-scale atmospheric CO_2 increases during glacial terminations.

2 Carbon isotopes and contradictions?

Driven mainly by orbital cyclicity, glaciers have rhythmically waxed and waned across Earth's surface together with changes in atmospheric chemistry, including the greenhouse gas CO_2 (Hays et al., 1976: Barnola et al., 1987). The amount and carbon isotope composition of atmospheric CO_2 (Fig. 2) depend on (1) the terrestrial biospheric organic carbon pool, (2) the dissolved inorganic carbon (DIC) pools residing in the surface and deep ocean, (3) exchanges between the atmosphere and the terrestrial biosphere and ocean, and (4) the export of organic matter and carbonate from the surface to deep waters and sediments of the oceans (Sigman and Boyle, 2000). During deglaciation, an increasingly large terrestrial biospheric carbon pool (Bird et al., 1994; Shackleton, 1977; Shackleton et al., 1983) is inferred to have controlled an increase in the stable carbon isotope composition of ~~dissolved inorganic carbon~~DIC in ocean waters. ~~However, this line of reasoning is complicated by various carbon pools changing in size at the same time as stable carbon isotope fractionation occurs, as carbon is exchanged between pools such as the terrestrial biosphere and pedosphere (see also Zeng, 2003, 2007).~~Reconstructed stable carbon isotope composition of DIC stems primarily from foraminifera which may also include bias from vital effects (e.g., Erez, 1978; Spero et al., 1997; Lea et al., 1999; see also Schmittner et al., 2017). Unlike the global, nearly unison rhythm of the glacial-interglacial marine oxygen isotope record, the global deglacial increase in carbon isotopes shows a notable exception: For much of the North Atlantic, the Holocene stable carbon isotope values of DIC are lighter than those of the Last Glacial Maximum (e.g., Peterson et al., 2020; Broecker and McGee, 2013; Bouttes et al., 2020). This is notable because the northernmost Atlantic is the locus of major downwelling driving global thermohaline circulation (de Carvalho Ferreira and Kerr, 2017).

In contrast to DIC of the oceans, atmospheric carbon isotope composition of CO_2 directly measured from ice core recovered CO_2 reflects a well-mixed, global signal. ~~In addition, d~~During times of most rapid CO_2 increase during transitions from glacial to interglacial periods, negative stable carbon isotope shifts in atmospheric CO_2 occurred (Fig. 3; Smith et al., 1999; Schmitt

60   et al., 2012). This is a strong indicator that, in conjunction with ocean-air exchange, respired organic matter acted as a direct source of carbon to the atmosphere (Bauska et al., 2016). ~~Terrestrial o~~Organic ~~carbon~~matter can be categorized into ~~comprises the sum of~~ biospheric and ~~surface exposed~~ lithogenic (i.e., kerogen) forms ~~of organic carbon~~, with both exhibiting similar ^{13}C signatures yet modern and dead ^{14}C signatures, respectively (e.g., Lewan, 1986; Meyers, 1994). ~~Additionally, t~~The constraints imposed by radiocarbon ~~clearly~~ indicate there must have been a source of carbon to the atmosphere and ocean depleted in or

65   devoid of radiocarbon (Broecker and Clark 2010; Hain et al., 2014; Rafter et al., 2019; Zhao et al., 2018), thereby limiting the potential contributions from a modern biospheric organic carbon source. Other studies have proposed that carbon sourced from ~~dissolved inorganic carbon from the~~ deep ocean DIC was the predominant source for carbon transferred to the atmosphere during glacial terminations (e.g., Hain et al., 2014). However, this hypothesis appears inconsistent with the negative fluctuation observed in the ^{13}C fingerprint of atmospheric CO_2 (see discussion in Broecker and McGee, 2013).

70   This leaves us with some apparent contradictions. If there really was a source of oxidized organic matter to the atmosphere, how can this be reconciled with the changes in marine DIC, which shows a negative shift for the North Atlantic and positive shifts for the rest of the world's oceans? Towards answering this question, Ciais et al. (2012) proposed that with a release of 700 PgC from an inert terrestrial carbon pool, the growth of the terrestrial biosphere from glacial to interglacial was only around 300 PgC of increase rather than the 300-700 PgC increase reported by most other studies. This is possible as during

75   the redistribution of carbon from inorganic to organic pools with various carbon pools changing in size, isotope fractionation occurs with increased selectivity for lighter carbon at higher atmospheric CO_2 concentrations (Farquhar et al., 1989; Broecker and McGee, 2013; Zeng, 2003). Additionally the negative shift observed in the North Atlantic was modeled by Crichton et al. (2016) showing this is explainable by the marine uptake and subduction of light carbon released to the atmosphere by terrestrial organic matter oxidation (hypothesized as permafrost in their case). Furthermore, Crichton et al. (2016) model the positive

80   carbon isotope shift in DIC observed in the South Atlantic demonstrating that contemporaneous positive and negative shifts in different ocean sectors are reconcilable with the release of isotopically light carbon from land. ~~Terrestrial organic carbon comprises the sum of biospheric and surface exposed lithogenic (i.e., kerogen) forms of organic carbon, with both exhibiting similar ^{13}C signatures yet modern and dead ^{14}C signatures, respectively (e.g., Lewan, 1986; Meyers, 1994).~~ Complementing the widely supported hypotheses of a~~n strictly~~ increasing terrestrial biospheric organic carbon pool and major changes in ocean-

85   atmosphere CO_2 exchange (e.g., Shackleton, 1977; Lindgren et al., 2018; Sigman and Boyle, 2000), the author will ~~argue that the ^{13}C and ^{14}C lines of evidence~~ suggest major release of CO_2 to the atmosphere via kerogen oxidation during deglaciation.

3 Kerogen and glaciers – Dynamic modulators of the global carbon cycle?

Chemical weathering of silicates and carbonates as well as the burial and oxidation of organic matter exert fundamental control over atmospheric chemistry on Earth over geologic timescales (Berner et al., 1983; Berner, 1990: Hilton and West, 2020;

90   Petsch, 2014). On modern-day Earth, carbon sources~~release~~ and sinks from mineral ~~based~~ weathering (i.e., silicates and carbonates via carbonic and sulfuric acids) and biogeochemical fluxes (i.e., organic matter burial and kerogen oxidation) stand

in fine balance to one another with contrasting concoctions of these under different tectonosedimentary settings (e.g., Bufe et al., 2021; Hilton and West, 2020; Blattmann et al., 2019a; Horan et al., 2019). Deglaciation likely modulated silicate and carbonate weathering resulting in enhanced (temporary) drawdown of atmospheric CO_2 (e.g., Tranter, 1996; Gibbs and Kump, 1996; Munhoven and François 1996; c.f., Schachtman et al., 2019). Glaciers have been invoked as agents for accelerating chemical weathering of carbonate and silicate minerals by increasing sediment yield and creating high surface area, reactive substrate with high surface area (Torres et al., 2017; Vance et al., 2009; Yu et al., 2021), with carbonate weathering constituting a source of CO_2 to the atmosphere when sulfuric acid, stemming from oxidized sulfide minerals, is involved (c.f., Torres et al., 2014; Kölling et al., 2019). In the following, the idea of glaciers producing reactive substrate susceptible to chemical weathering is reviewed and explored for kerogen.

Evidence accumulated from over a century of scientific studies supports the idea that the reburial of kerogen was more extensive during cold interludes in Earth history where glacial erosion and ice rafting was widespread (Blattmann et al., 2018b). This general pattern of kerogen reburial is also observed spatially today with more efficient reburial in high latitude glaciated regions than in low latitude sedimentary systems (Cui et al., 2016). This enhanced reburial efficiency strengthens the short-circuiting of the exogenous kerogen cycle by keeping this ancient, reduced carbon locked away. However, under glaciofluvial conditions, the entry of kerogen-bound carbon into surficial carbon pools is promoted (Horan, 2018), with glacial meltwater releasing dissolved organic matter from kerogen, which is readily degraded and consumed by microbes (Hood et al., 2009; Schillawski and Petsch, 2008; Sharp et al., 1999). Additionally, frost shattering, together with the retreat of glaciers, exposes finely ground, high surface area, reactive sediment, thereby accelerating oxidation and the release of kerogen-derived CO_2 to the atmosphere (Fischer and Gaupp, 2005; Horan et al., 2017). The initially strong input of kerogen-derived CO_2 would die down as the availability of glacially ground, reactive kerogen declines into an interglacial period. Enhanced degradation of kerogen in the wake of glacial episodes is consistent with observations from areas of ongoing deglaciation (Horan et al., 2017). Contemporaneous perturbations in CO_2 fluxes are expected to result for silicate and carbonate weathering via carbonic and sulfuric acids (Vance et al., 2009; Torres et al., 2014). However, kerogen oxidation, owing to its faster weathering kinetics and direct conversion to CO_2 leads to considerable losses in kerogen content over decadal timescales (Horan et al., 2017 and references therein; Fischer et al., 2007). This is a process by which CO_2 can be injected directly into the atmosphere and impact glacial-interglacial cycles (Fig. 2).

In the following, quantitative arguments are presented: Of the approximately 150 PgC/kyr of kerogen reaching the Earth's surface (F_L; Hedges and Oades, 1997), $43^{+61}_{-25} \frac{PgC}{km^2 yr}$ is currently exported by rivers to oceans (F_{L-O}; Galy et al., 2015; Copard et al., 2007), indicating that the modern-day reburial efficiency of this carbon lies in the ballpark of 30% (10-70%). Normalized to global land area ($A_L = 149{,}000{,}000\ km^2$), the flux of carbon due to oxidation of kerogen to the atmosphere (F_{L-A}) can be calculated from:

$$F_{L-A} = \frac{F_L - F_{L-O}}{A_L} \qquad \text{(Equation 1)}$$

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Here, equation 1 yields a modern-day global F_{L-A} of $0.72^{+0.17}_{-0.41} \frac{MgC}{km^2 yr}$. Reports of local and regional CO_2 fluxes from kerogen oxidation are based on disparate study sites and methods (Table 1). From this, reported CO_2 fluxes stemming from oxidation of rock disseminated forms of kerogen and across larger catchment areas under aerobic conditions span over two orders of magnitude ranging from 0.3 to 64 $\frac{MgC}{km^2 yr}$. The only quantities available for catchments of ongoing deglaciation stem from the

130   Southern Alps of New Zealand, which are one to two orders of magnitude faster than those of the average Earth surface (ranging between 9-50 $\frac{MgC}{km^2 yr}$; Horan et al., 2017). Taking this flux range as a working estimate, how would kerogen oxidation translate into changes in atmospheric CO_2 concentration? In Zeng's (2003) modeling work postulating soil oxidation upon retreat of overriding glaciers, the effect of continuous inputs of organic matter-derived carbon directly to the atmosphere are calculated: 550 PgC oxidized lead to 60 PgC increase in atmospheric CO_2 (30 ppm increase). Similarly, modeling by Simmons

135   et al. (2016) also propose the oxidation of overridden soil organic carbon subsequent to deglaciation and calculate a 600 PgC release resulting in a 40-60 ppm increase in atmospheric CO_2 (c.f., also Ciais et al. (2012) who calculate the deglacial oxidation of 700 PgC of an "inert" terrestrial organic carbon pool). The idea of soil oxidation is mathematically analogous to kerogen oxidation in the hypothesis presented here. How much area of deglaciated terrain (A_D) would therefore be needed to release 550-600 PgC of kerogen-derived carbon? Using Zeng's (2003) and Simmons et al.'s (2016) carbon release estimates

140   ($550\ PgC \leq C_R \leq 600\ PgC$) and Horan et al.'s (2017) kerogen oxidation kinetics ($9 \frac{MgC}{km^2 yr} \leq F_{KO} \leq 50 \frac{MgC}{km^2 yr}$) over a 6,000-year time window (T_{G-I}, timespan of pronounced CO_2 increase over the glacial-interglacial transition; see Fig. 3) and subtracting global average baseline kerogen oxidation ($F_{L-A} \approx 0.72^{+0.17}_{-0.41} \frac{MgC}{km^2 yr}$; see Eq. 1):

$$A_D = \frac{C_R}{T_{G-I}*(F_{KO}-F_{L-A})}$$ (Equation 2)

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With calculated A_D yielding between 2 and 12 million km^2, in the most conservative estimate, 12 million km^2 of deglaciated terrain containing sedimentary and metasedimentary rocks and their debris (glacial flour accumulated over millennia of glaciation) would cover the source. In comparison, Canada covers approximately 10 million km^2, which was mostly covered by the Laurentide Ice Sheet during the Last Glacial Maximum. With this, a plausible scenario for releasing kerogen-derived

150   CO_2 to the atmosphere that could account for a 30-60 ppm rise during the glacial-interglacial transition encompassing an area equal to or less than the terrestrial extent of the Laurentide Ice Sheet is identified. Similar to the oxidation fluxes reported from deglaciating catchments (Horan et al., 2017) are those reported for the in-situ weathering of massive outcrop formations of marls and shales (e.g., Soulet et al., 2018; Littke et al., 1991; see Table 1). Kerogen oxidation fluxes appear higher for catchments showing larger glacial coverage (Horan et al., 2017) suggesting that the relative contribution of freshly exposed

155   and ground bedrock is highest for these areas and that kerogen oxidation fluxes progress over the course of deglaciation. The

CO_2 flux emanating from kerogen oxidation in soils formed from glacial till (Keller and Bacon, 1998) suggests that oxidation fluxes an order of magnitude greater than the global average can ~~be~~ sustain~~ed~~ for millennia after deglaciation. However, a systematic chronosequence understanding is currently lacking that allows conclusions to be drawn as a variety of factors (e.g., lithology, initial kerogen content, grain size) exert control over kerogen oxidation (c.f., Fischer and Gaupp, 2005; Bolton et al., 2006; Martínez and Escobar, 1995). The bedrock and river sediments of the Southern Alps are relatively lean in kerogen (Copard et al., 2007; Horan et al., 2017). Therefore, for deglaciating landscapes exposing glacially ground kerogen-rich lithologies, larger oxidation fluxes than those reported from the Southern Alps can be expected, increasing and decreasing the plausible ranges for F_{KO}~~G_K~~ and A_D, respectively (see next section).

Long-term accelerated decline in the radiocarbon concentration of atmospheric CO_2 parallel to an overall increase in CO_2 amount occurred since the Last Glacial Maximum (Reimer et al., 2013; Roth and Joos, 2013). Consistent with such a dilutional process, kerogen oxidation releases radiocarbon-dead CO_2 to the atmosphere. During the time period of greatest CO_2 increase from 17.5 until 11.5 kyr before present, the rate of decline in the concentration of ^{14}C in atmospheric CO_2 is greatest and averages about 35 ‰/kyr (Fig. 3; c.f., Broecker and Barker (2007)). With the release of 500 PgC (radiocarbon-dead) to the atmosphere over deglaciation, a drop in 100 to 200 ‰ $\Delta^{14}C$ of atmospheric CO_2 can be expected (see calculations by Zeng, 2007). Kerogen oxidation would also release isotopically light carbon to the atmosphere, which is also consistent with the atmospheric record (Fig. 3; Bauska et al., 2016). Finally, driven by orbital forcing (Hays et al., 1976), the response in atmospheric CO_2 faithfully echoing increasing global temperatures and diminishing glaciated terrain (Sigman and Boyle, 2000; Stips et al., 2016) goes hand in hand with continuous aerial exposure and enhanced oxidation of finely ground, reactive kerogen (Fig. 2).

4 A Canadian tale. And what about microbes?

Kerogen exposed on Earth's surface is distributed unevenly, with kerogen-rich surface lithologies extending across much of western Canada (see Fig. 2 in Copard et al., 2007). These areas experienced dramatic deglaciation across the 17.5-11.5 ka timeframe (Fig. 4; Dyke, 2004; Dalton et al., 2020). In addition to the widespread occurrence and high abundance of rock disseminated forms of kerogen, there are also notable surficial occurrences of hydrocarbon-rich lithologies including coal and oil sands within the deglaciated terrains of western Canada. Within the Province of Alberta, Andriashek and Pawlowicz (2002) report the widespread occurrence of reworked shale and bitumen in Quaternary till stemming from Cretaceous shales and the oil sands, with unoxidized and oxidized forms present, and the latter enhanced by aerial exposure across paleosurfaces. Today, bituminous erratic boulders are found strewn around the region over tens of thousands of square kilometers (Rutherford, 1928; Andriashek, 2018). As Andriashek (2018) points out, Rutherford (1928) states: "There are perhaps many more occurrences of bituminous sand within the glacial deposits of Alberta, but since they weather comparatively readily and become covered with soil, they are not likely to be detected unless by accident, …". This observation of rapid weathering of bitumen indicates that today's occurrences in glacial tills represent the tip of the iceberg of what once was present. Laboratory incubations simulating

CO$_2$ respiration from bituminous materials reveal fluxes that are markedly higher than those associated with oxidation of rock disseminated forms of kerogen (see Table 1). Chang and Berner (1998, 1999) report subaquatic bituminous coal oxidation (calculated >300 $\frac{MgC}{km^2 yr}$) releasing CO$_2$ at rates 1-2 orders of magnitude higher than those reported for rock disseminated kerogen, and 3 orders of magnitude greater than the average for Earth's surface. Microbes likely play key roles in assimilating and releasing this ancient organic carbon as CO$_2$ to the atmosphere (ZoBell, 1946; Hemingway et al., 2018) with microbial communities on the surface colonizing oil sand-derived bitumen under both summer and winter conditions (Wyndham and Costerton, 1981; Wong et al., 2015). Biodegradation is accelerated during summer months when temperatures of subaerially exposed outcrops of oil sands reach 60°C (Wong et al., 2015). Microbial degradation experiments on bitumen (Ait-Langomazino et al., 1991) reveal an even greater CO$_2$ release when extrapolated to natural systems (e.g., oil sands, see Table 1) exceeding 18 $\frac{MgC}{km^2 day}$, over one hundred times greater than the highest CO$_2$ flux released by rock disseminated kerogen oxidation. Other laboratory-based studies (e.g., Uribe-Alvarez et al., 2011) that investigated the oxidative decay of hydrocarbon fractions also suggest similarly high fluxes when scaled to natural systems, even though these studies were conducted over periods of only a few weeks. Such observations likely overestimate long-term natural fluxes over multiple years and decades, while the fluxes reported by Chang and Berner (1998, 1999) likely represent an underestimate due to sample storage for several years prior to analysis and the absence of microbial activity under their experimental conditions. Additionally, comparable quantities of kerogen-derived CO$_2$ can also be released under anaerobic conditions, which may become relevant under warming glaciers (e.g., Bertassoli Jr. et al., 2016; Rogieri Pelissari et al., 2021; see also Sharp et al., 1999; Sharp and Tranter, 2017). Overall, it is conceivable that super carbon source terrains (hypothetically, areas laced or covered with coal, bituminous materials, etc.) across western Canada could supply an overproportionate quantity of radiocarbon dead CO$_2$ to the atmosphere during glacial-interglacial transitions.

By 11.5 ka before present, the pronounced rise in atmospheric CO$_2$ had subsided. By this time, the Laurentide Ice Sheet had retreated beyond the easternmost reaches of the Western Canadian Sedimentary Basin and was exposing bedrock from the relatively kerogen-poor, highly metamorphosed Canadian Shield (Fig. 4; Dalton et al., 2020). The timing of the inflection point in the rise of CO$_2$ occurs ≤300 years after the entire western edge of the Laurentide Ice Sheet advanced onto the Canadian Shield suggests diminishing decay of kerogen and relaxation of the landscape. By this time, the rate of soil organic carbon sequestration in these paraglacial landscapes begins to pick up substantially (Harden et al., 1992). Lithologies of the Fennoscandian Shield and most of its adjacent landmasses (deglaciated eastern and southern sectors of the Fennoscandian Ice Sheet; see Stroeven et al., 2016 and Copard et al., 2007) also contain relatively low kerogen contents, therefore limiting its contributions to atmospheric CO$_2$ rise in the wake of Fennoscandian Ice Sheet retreat. Glacially ground graphite shed from these highly metamorphosed rocks is chemically recalcitrant and is mostly redeposited (Sauramo, 1938; Sackett et al., 1974; Sparkes et al., 2020). Geophysical arguments propose that the relatively narrow timing of the Laurentide Ice Sheet's isochrons retreating into the Canadian Shield is more than coincidence: the lithological base and its overlying glacial till, influence the dynamics of ice sheet retreat with the Western Canadian Sedimentary Basin representing a deformable base while the Canadian

Shield behaves rigidly providing an explanation for why the geological and ice sheet isochron map show similarities (Clark, 1994 and see Fig. 3 therein; Licciardi et al., 1998). At the same time in the context of the kerogen oxidation hypothesis, this offers an explanation for the well-defined inflection point in the rise of atmospheric CO_2 characteristic for glacial-interglacial transitions.

225 In the bigger picture, during Earth's recent glacial episodes, the Laurentide Ice Sheet was the most extensive element of the cryosphere that waxed and waned across the continents (Batchelor et al., 2019) and, in conjunction with its lithological underpinning (Copard et al., 2007), likely played the largest role in releasing kerogen-derived CO_2 to the atmosphere upon glacial retreat. However, estimates of CO_2 fluxes emanating from bedrock show a wide range (Table 1) and there is considerable uncertainty in our current state of knowledge: Dedicated biogeochemical weathering studies that provide

230 estimates of CO_2 fluxes from bedrock-derived kerogen under relevant environmental conditions and over appropriate timescales are lacking. Chronosequence studies of kerogen oxidation rates in deglaciated terrains are needed to provide constraints on time-integrated CO_2 release to the atmosphere. In tandem with this, quantification is needed for (temporary) kerogen reburial in subaerial and subaquatic terrestrial systems (e.g., moraines, lakes) on global and regional scales (e.g., Meybeck, 1993; Vonk et al., 2016; Blattmann et al., 2019b; Fox et al., 2020). Such information in conjunction with high

235 resolution reconstructions of changes in land ice extent and the lithologies of bedrock and glacial till (including data on kerogen content and its stable carbon isotope composition) ~~being~~ exposed by glacial retreat can, in theory, quantitatively disentangle the time-integrated contribution of kerogen-derived CO_2 to the atmosphere during glacial-interglacial transitions.

5 Tackling geologic deep time

While modern day kerogen weathering and reburial efficiencies are only loosely constrained, even less is known about how

240 they varied back in geologic time. Overarching controls on these processes include the mode of erosion and transport ranging from glacial to glaciofluvial to fluvial, remineralization intensity, controlled by continental margin type and geomorphology, and the intrinsic reactivity of the kerogen present locally (Blair and Aller, 2012; Blattmann et al., 2018b). Also important are the bedrock lithology and regolith composition which have been hypothesized to exert feedback on temporal patterns of glacial-interglacial cyclicity (e.g., Roy et al., 2004; Zeng, 2007). For the Paleocene-Eocene Thermal Maximum, an extreme

245 greenhouse episode, evidence increasingly suggests that enhanced remineralization and leaching of kerogen (Boucsein and Stein, 2009; Lyons et al., 2019), possibly enhanced by the activity of microbes (Hemingway et al., 2018; Petsch et al., 2001), increased the flux of carbon entering actively circulating pools on Earth's surface. Over Earth's history on 10^9 year timescales, the reburial efficiency of kerogen presumably varied as a function of atmospheric O_2 content, with lower O_2 contents tied to higher reburial efficiency (Daines et al., 2017). Kerogen is surmised to have acted as a major source of carbon to the atmosphere

250 and as an "antioxidant" during the early rise of atmospheric O_2 (Daines et al., 2017; Kump et al., 2011) and balancing O_2 levels over the Cenozoic (Galvez, 2020; see also Derry and France-Lanord (1996) for discussion on Paleogene). Across multiple

glacial-interglacial cycles, enhanced kerogen oxidation would also be consistent with declining atmospheric O_2 on 10^6 yr timescales (Stolper et al., 2016).

In order to understand changes in atmospheric chemistry through geologic time, in addition to comprehensively budgeting the effect of mineral chemical weathering (Blattmann et al., 2019a; Horan et al., 2019; Hilton et al., 2014), the changing efficiency of the oxidation and reburial of kerogen needs to be evaluated. Direct quantification of kerogen found reburied in sediments is often associated with considerable uncertainty owing to uncertainties in organic matter source apportionment (e.g., Lin et al., 2020; Blattmann et al., 2019b) and uneven spatial dispersal (e.g., Blattmann et al., 2018a; Cui et al., 2016). While radiocarbon was paramount for quantifying and establishing the importance of the reburial of kerogen in recent times (Blattmann et al., 2018b), its utility diminishes quickly for strata that pre-date the Last Glacial Maximum owing to its radioactive decay. However, associated with kerogen are a promising suite of trace elements and their respective isotope signatures including, among others, rhenium (Hilton et al., 2014; Horan et al., 2017), osmium (Georg et al., 2013; Ravizza and Esser, 1993), and iodine (Moran et al., 1998) as well as clay minerals (Blattmann and Liu, 2021), that can be exploited to trace sedimentary kerogen and its degradation. In the case of osmium, seawater records reveal isotopic shifts at the beginning of interglacial periods that are attributable to kerogen oxidation (Georg et al., 2013; compare discussions in Peucker-Ehrenbrink and Ravizza, 2020), consistent with the hypothesis presented here. More research constraining the exogenous kerogen cycle by quantification of reburied kerogen inputs (e.g., iodine isotopes, organic petrology) and kerogen oxidation recorded by chemical weathering proxies (e.g., osmium isotopes) is needed to put the presented hypothesis to the test.

6 Synthesis and outlook

In a nutshell, the hypothesis presented proposes the following: Less than 300 years after the Laurentide Ice Sheet retreated east past the eastern~~western~~most edges of the kerogen-rich Western Canadian Sedimentary Basin, atmospheric CO_2 levels stabilized and the rate of decrease in ^{14}C concentrations of CO_2 subsided (Figs. 3 and 4). This inflection point is mirrored in the lithologies of the Canadian Shield that were exposed during deglaciation, which contain relatively minor amounts of reactive kerogen. Within the context of the presented hypothesis, the coincidence in time of global trends in atmospheric chemistry with spatiotemporal patterns in the distribution of freshly exposed deglaciated terrain impregnated with oxidizable and biodegradable kerogen and bituminous materials suggests that a burst (or bursts) of respired CO_2 contributed to the characteristic deglacial increase in atmospheric CO_2. As soon as glacially ground shales and bituminous materials either exhausted their labile kerogen content and/or became buried by soil and vegetation taking hold on the deglaciated landscape, classically considered processes (land-ocean exchange, biospheric uptake, etc.) reassumed dominant control on fluctuations of atmospheric chemistry. As a corollary of this hypothesis: CO_2 rise within the envelope of glacial-interglacial cyclicity would primarily have responded as a slave to orbitally controlled glacial retreat across kerogen impregnated landscapes as a result of global temperature increase. CO_2 rise may therefore lead or lag global temperature depending on the spatiotemporal patterns of glacial retreat that exposing glacially ground, kerogen-rich or even bituminous parent material.

Various mechanisms have been proposed to explain CO_2 increases at glacial interglacial transitions including for example the solubility pump hypothesis, iron fertilization hypothesis, ocean circulation hypotheses and many more (see hypotheses and reviews by Martin, 1990; Broecker and Peng, 1993; Kohfeld and Ridgwell, 2009; Rapp, 2019). However, as a cogwheel operating under manifold feedbacks in the greater Earth system (Sigman and Boyle, 2000), continuous glacial retreat, and the oxidation of finely ground kerogen, provide a hypothesis consistent with a CO_2 increase in the wake of glacial episodes. This dilution of radiocarbon-dead CO_2 in the atmosphere may well have been complemented by other terrestrial sources such as the oxidation of subglacial paleosols and permafrost-bound organic carbon (Zeng, 2007; Simmons et al., 2016; Tesi et al., 2016; Crichton et al., 2016; Martens et al., 2020; Winterfeld et al., 2018; Lindgren et al., 2018; Köhler et al., 2014; Ciais et al., 2012) and by volcanic emissions triggered by deglacial unloading of the lithosphere (Roth and Joos, 2012). In addition to studies of weathering in glacial forefields and source-to-sink tracing of sedimentary kerogen, several lines of geochemical evidence including atmospheric carbon isotope composition (^{13}C and ^{14}C), which have thus far received contorted, partial explanations (Broecker and Clark, 2010; Schmitt et al., 2012; Broecker and McGee, 2013), glacial-interglacial changes in ^{14}C of DIC~~dissolved inorganic carbon~~ in seawater (Rafter et al., 2019, c.f., discussions therein), seawater osmium isotope changes, and long-term atmospheric O_2 content, conceptually go hand-in-hand with an opening of the exogenous kerogen cycle modulated by glacial activity. While geomagnetic variability and ocean ventilation together struggle to fully explain observed changes in atmospheric radiocarbon (Broecker and Barker, 2007; Cheng et al., 2018), the dilution of atmospheric CO_2 by accelerated oxidation of ancient terrestrial organic carbon at glacial terminations, in conjunction with other mechanisms including atmosphere-ocean gas exchange (e.g., Sigman et al., 2010; Marcott et al., 2014; Menviel et al., 2018; Martin, 1990; Sarntheim et al., 2013; Ai et al., 2020) appears as a simple and plausible explanation. In addition to kerogen oxidation at glacial terminations, the hypothesis presented carries other important implications, including the closure of the exogenous kerogen cycle during glacial periods (Fig. 2) potentially contributing to relatively high atmospheric ^{14}C signatures, a problem highlighted by Dinauer et al. (2020), as a reduced ^{14}C-free CO_2 flux would reduce the dilution of the atmosphere's cosmogenic [14]C. Overall, increased reburial efficiency of kerogen can account for several tens of PgC over millennial timescales, entirely bypassing actively circulating carbon pools on Earth's surface (Fig. 1). In contrast, increased oxidation efficiency of kerogen in the wake of glacial episodes that have built up stores of finely ground reactive substrate and exposed fresh weathering profiles can account for several hundreds of PgC over millennial timescales released into actively circulating pools. Owing to increased bedrock exhumation over the Ice Ages (Herman et al., 2013; Herman et al., 2015) the dynamism of the exogenous kerogen cycle may have been intensified, with greater fluxes of detrital kerogen reburied in ocean sediments during glacial episodes and increased supplies of ground kerogen exposed to the elements in their wake.

While basic controls on kerogen reburial efficiency have emerged, its quantitative impact on atmospheric chemistry through geologic time remains conjectural. Mathematically analogous modeling results (Zeng, 2003; Simmons et al., 2016) suggest that kerogen oxidation could account for 30-60 ppm rise in atmospheric CO_2 over the course of the last deglaciation. However, there is a lack of data on kerogen weathering to provide sufficient quantitative constraint for testing this hypothesis. To work towards such a test, investigating kinetics of kerogen oxidation along glacial chronosequences with contrasting lithologies

would provide numerical input for Earth system models. Such information in conjunction with spatiotemporal changes in land ice extent (e.g., Dalton et al., 2020; Stroeven et al., 2016) integrating over areal changes of glacially exposed lithological units

320 (c.f., Copard et al., 2007) would constrain the quantitative impact of the exogenous kerogen cycle on atmospheric chemistry over glacial-interglacial cycles (e.g., see Fig. 2 in Kohfeld and Ridgwell, 2009). Additionally, the role of super carbon source terrains (e.g., surficial oil sands and coal) deserve special attention as biogeochemical weathering and associated CO_2 fluxes emanating from such areas conceivably contribute disproportionately, particularly upon deglaciation. Overall, the spatiotemporal deglaciation of contrasting source terrains across North America with the coeval progression of atmospheric

325 chemistry provide a strong incentive to explore the potential role of the exogenous kerogen cycle in glacial-interglacial patterns. From a greenhouse perspective, further study of pivotal episodes such as the Paleocene-Eocene Thermal Maximum under this lens may provide ~~an~~geological outlooks ~~for geological processes~~ that are~~is~~ relevant today. In the context of our warming world, once critical thresholds are breached (Steffen et al., 2018), enhanced opening of the exogenous kerogen cycle may entrain the Earth system onto a new trajectory influencing the carbon cycle and climate for millennia to come.

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Competing interests

335 The author declares that he has no conflict of interest.

Author contributions

TMB conceived of and wrote this contribution.

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Table 1: Fluxes of kerogen oxidation normalized to one year from onsite (soil and outcrops), catchment-wide studies (river), and laboratory-based incubation studies ordered approximately from small to large CO₂ release fluxes. For calculations, see online supplemental.

Area/ Material	Oxidation flux [MgC/km²/yr]	Comments	References
Andes	≥0.3	Kerogen oxidation in the Madeira floodplain based on radiocarbon isotope mass balancing of sedimentary organic carbon. Oxidation flux normalized to catchment area reported in Clark et al., (2017).	Bouchez et al., (2010)
	~1.2	Kerogen oxidation in the Madeira floodplain based on radiocarbon isotope mass balancing of sedimentary organic carbon refining the result by Bouchez et al. (2010).	Clark et al., (2017)
Mackenzie River Basin	$0.45^{+0.19}_{-0.11}$	Rhenium-based estimate integrating over catchment and riverine in-situ weathering.	Horan et al., (2019)
Global average	$0.72^{+0.17}_{-0.41}$	Calculated global average rate of kerogen oxidation on land. See Equation 1.	See text.
Soil, glacial till (Canada)	4.2	Holocene-averaged flux of onsite kerogen oxidation based on CO₂ evolution and carbon isotopes.	Keller and Bacon (1998)
Taiwan	6.1-18.6	Radiocarbon biogeochemistry reveals microbially-mediated weathering of kerogen onsite in soils.	Hemingway et al., (2018)
	12±6	Sedimentary organic carbon mass balancing with radiocarbon integrating over catchment and riverine in-situ weathering	Hilton et al., (2011)
	7.4-13.0	Rhenium-based estimate integrating over catchment and riverine in-situ weathering.	Hilton et al., (2014)
Shale (Posidonia, Germany)	11-16	Onsite kerogen oxidation based on Holocene scenario described in reference using mass loss of kerogen and estimated erosion rates. Weathering depth of 5 m and organic carbon to organic matter conversion factor of 1.5 assumed.	Littke et al., (1991)
Southern Alps, New Zealand	9-50	Rhenium-based estimate integrating over catchment and riverine in-situ weathering. Oxidation fluxes reported for 4 catchments with minimum and maximum reported here.	Horan et al., (2017)
Tar sandstone (Brazil)	~54	Laboratory-based experiments over 1 year duration under water saturated and anoxic conditions at 20°C. Results scaled with density taken as 2000 kg/m³ and 1 m weathering depth.	Rogieri Pelissari et al., (2021)
Shale (Jurassic marl, France)	61-64	Field-based observations spanning over minutes to up to 101 days using zeolite traps.	Soulet et al., (2018)
Bituminous coal	320-530	Laboratory-based subaquatic oxidation experiments at 24°C for air-saturated water over 179-442 days duration. Carbon release calculated assuming 30-50% of oxygen reacts to form CO₂, an average specific surface area of coal of 1 m²/g, average density of 1400 kg/m³, and 1 m weathering depth.	Chang and Berner (1998, 1999)
Shale (Brazil)	80-1050	Laboratory-based experiments over 75 days duration under water saturated, dark, and anoxic conditions at 25°C. Results scaled with density taken as 2200 kg/m³ and 1 m weathering depth.	Bertassoli Jr. et al., (2016)
Oil sands (Bitumen)	6800-13000	Laboratory-based experiments of 100 days duration for biodegradation of bitumen with different microbial cultures. Results scaled to oil sands with density taken as 2000 kg/m³ with 10% bitumen content, and 1 m weathering depth.	Ait-Langomazino et al., (1991)
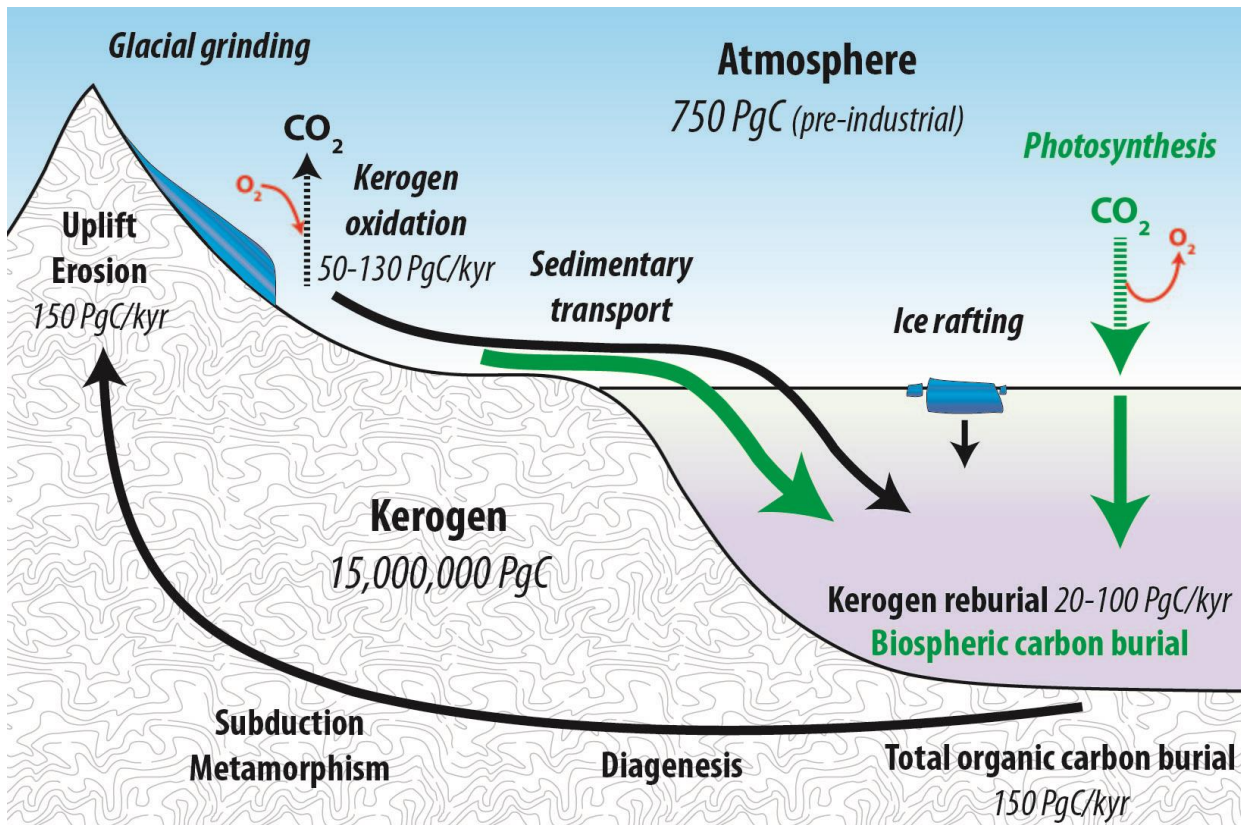
Figure 1: Organic carbon cycle with the flow of kerogen (black solid lines) and the flow of biospheric carbon (green solid lines) showing the fixation of atmospheric CO₂ by both terrestrial and marine primary productivity. The combined flux of reworked kerogen and biospheric carbon into ocean sediments constitutes the total organic carbon burial into the endogenous kerogen pool (Galy et al., 2015; Hedges and Oades, 1997).

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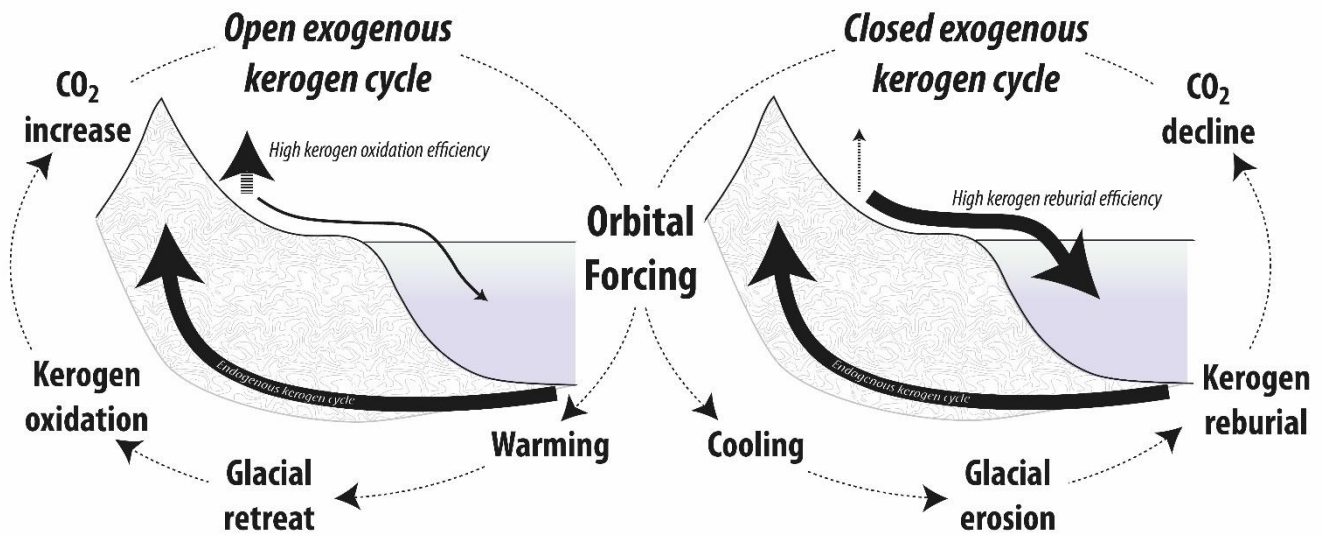
Figure 2: Conceptual hypothesis overview: Changes in kerogen reburial efficiency and its effect on the reentry of ancient carbon into surficial carbon pools as a function of overall climate state. During glacial times, kerogen reburial is promoted by the activity of glaciers and ice sheets with relatively little oxidation of this carbon during its transit across Earth's surface characteristic of a "closed" exogenous kerogen cycle. During glacial terminations and interglacials, the oxidation of kerogen is more efficient leading to the exhalation of this carbon to the atmosphere characteristic of an "open" exogenous kerogen cycle.
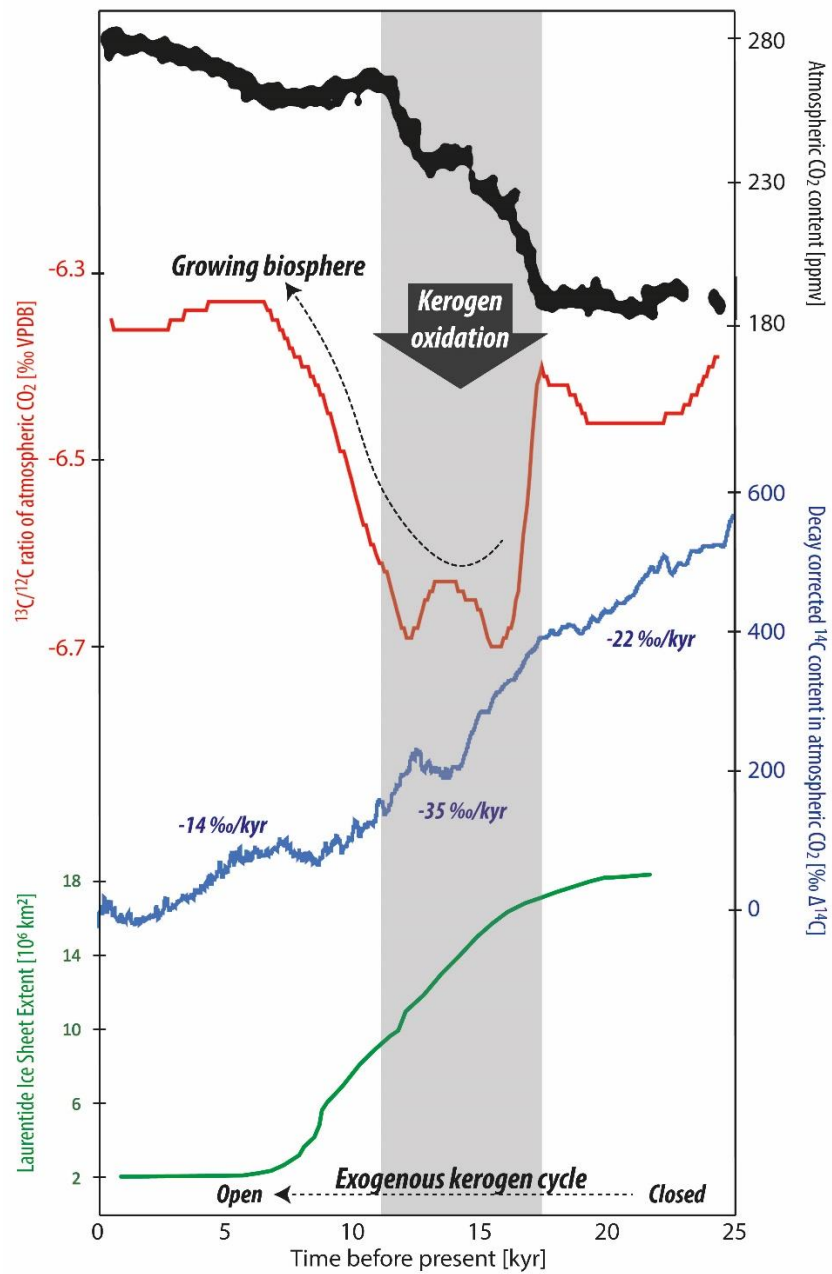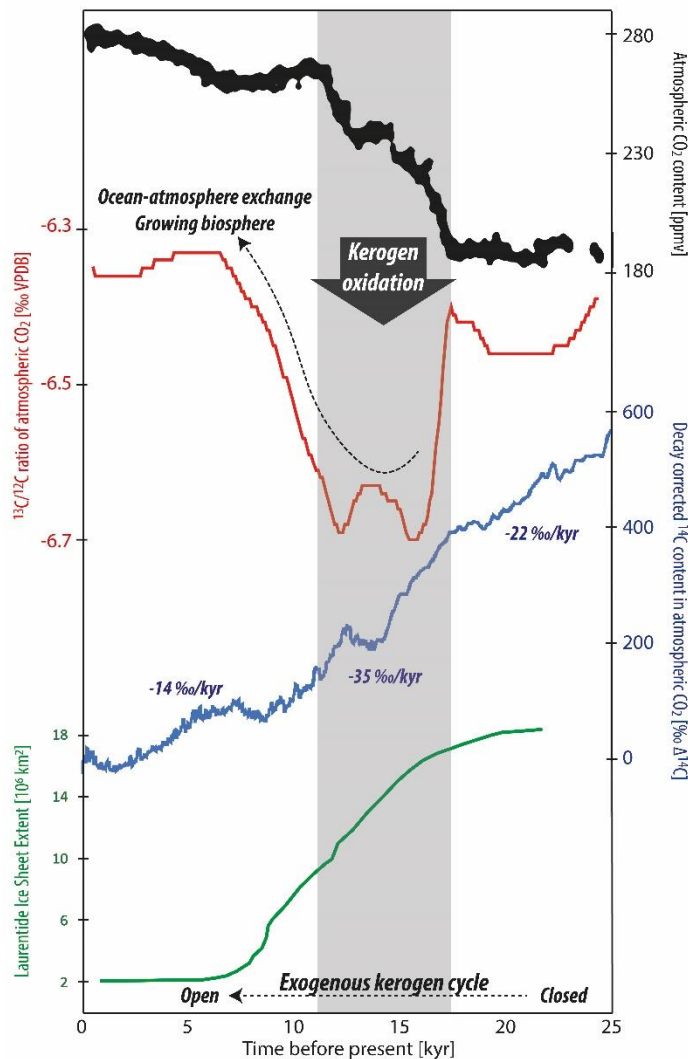
Figure 3: Evolution of atmospheric CO₂ and its carbon isotope composition starting from the Last Glacial Maximum with redrawn data from Schmitt et al. (2012) and ¹⁴C data from Reimer et al. (2013). The area of the Laurentide Ice Sheet (including Greenland) is plotted after Dalton et al. (2020). The gray time envelope indicates the greatest rate of atmospheric CO₂ increase coinciding with a pronounced negative pulse in stable carbon isotope composition and accelerated decrease in radiocarbon concentration. In addition to the hypothesized role of kerogen oxidation, the superimposed effects of ocean-a~~tmosphere~~ir gasCO₂ exchange, a growing terrestrial biosphere, ~~changes in radiocarbon production in the atmosphere,~~ etc. all contribute to the evolution of these parameters.

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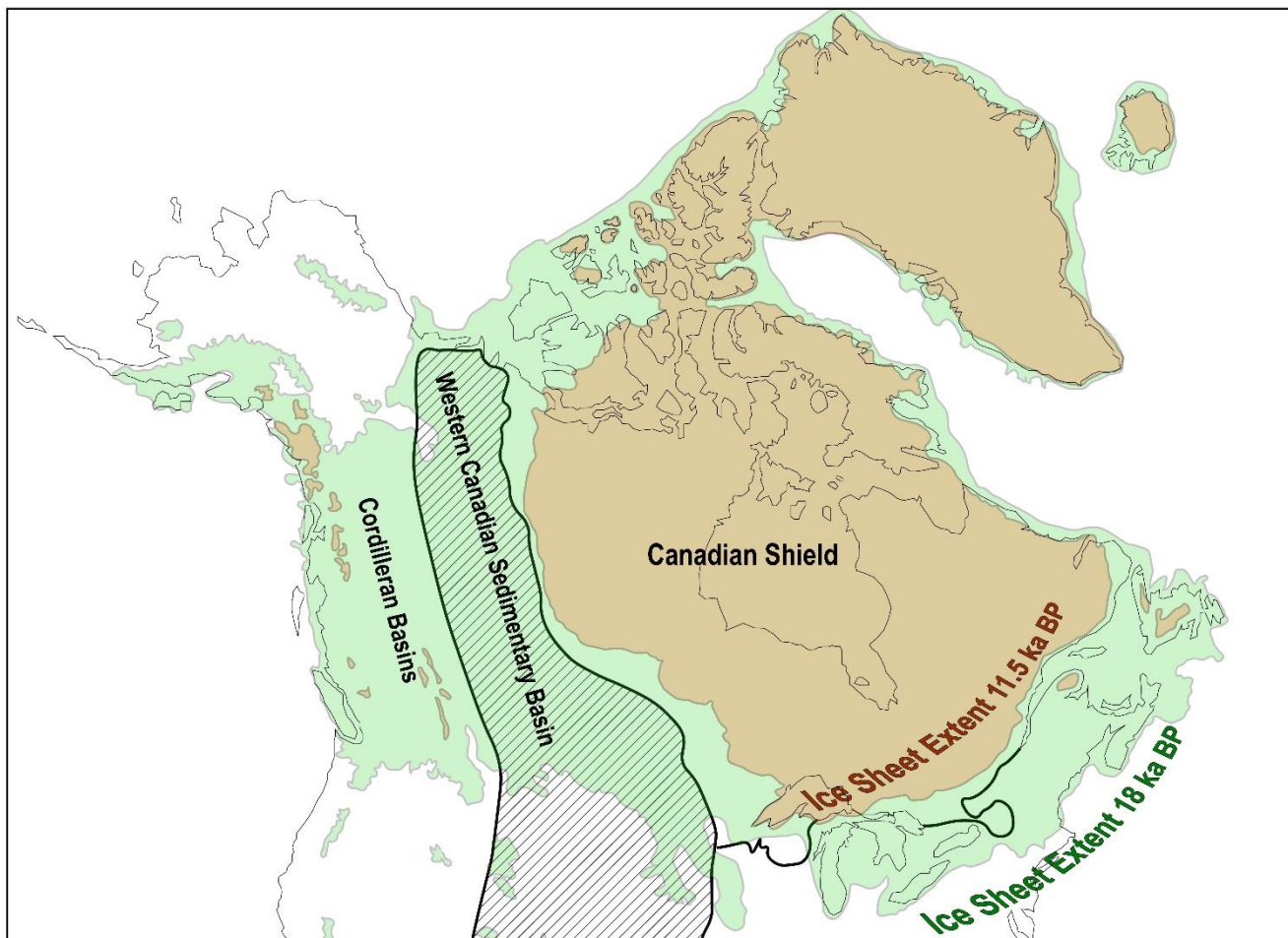
Figure 4: Laurentide Ice sheet extents bracketing the time span of most pronounced glacial-interglacial CO$_2$ increase (redrawn after Dalton et al., 2020). The Western Canadian Sedimentary Basin (hatched area, with extension of related units into the United States) and the Cordilleran Basins are to the west of the Canadian Shield (simplified after Miall and Blakey, 2019). The southern border of the Canadian shield is drawn with mainly Phanerozoic sedimentary units to the south (simplified after Reed et al., 2004). The
660 **surficial kerogen content in the Canadian Shield is generally low, while high kerogen contents are present along the axes of the Cordilleran and Western Canadian Sedimentary Basins (see Fig. 2 in Copard et al., 2007). ≤300 years after glacial retreat extended into the Canadian Shield, the rise in atmospheric CO$_2$ subsided (compare with Fig. 3). Here, kerogen oxidation in western Canada is hypothesized to have acted as a major source of CO$_2$ to the atmosphere in the direct wake of glacial retreat.**