

Responses to the Editor

Comments:

Dear authors,

many thank for your comprehensive revisions and my sincere apologies for the delay in coming to a decision. I am satisfied with your revisions, but would ask you to add the explanation how a flux rate of Ammonia is derived from Eq. 1 in the responses to the main manuscript. The ms is than ready for publication.

Best wishes,
Sönke Zaehle

Reponses:

We thank the editor's constructive feedback.

Per the editor's suggestions, we added the following explanation on how the NH₃ flux rate is calculated in lines 263–277, **Section 2.2**:

“The actual soil NH₃ to be emitted ($F_{\text{soil,act}}$; g-N m⁻³) from each soil layer is then determined **as:**

$$F_{\text{soil,act}} = \min(F_{\text{soil,pot}}, [\text{NH}_4^+]_{\text{available}}) \quad (1)$$

where $F_{\text{soil,pot}}$ is from Eq. (1) and $[\text{NH}_4^+]_{\text{available}}$ (g-N m⁻³) is the concentration of available soil NH₄⁺ in the soil layer. The model distributes available soil NH₄⁺ to all competing processes, **namely, NH₃ emission, plant uptake, microbial immobilization, and nitrification,** according to their relative demands (individual potential flux to sum of all four potential fluxes) without bias toward any process (Lawrence et al., 2019). The **column-level actual soil NH₃ prone to emission (F_{soil} ; g-N m⁻²)** is then computed as the sum of the product of $F_{\text{soil,act}}$ and **the soil layer thickness (in meters) across all vertical layers.**

Finally, assuming such NH₃ is released to the atmosphere at a constant rate over a model time step size ($\Delta t = 1800$ s in this study), our model estimates the NH₃ emission flux (\dot{F}_{soil} ; g-N m⁻² s⁻¹) as:

$$\dot{F}_{\text{soil}} = \frac{F_{\text{soil}}}{\Delta t} \quad (2)$$