Supplemental Material

Atmospheric Deposition of Reactive Nitrogen to a Deciduous Forest in the Southern Appalachian Mountains

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S1. Comparison of Measurement Methods for Wet Deposition and Air Concentrations Concentrations of inorganic N in precipitation



Figure S1. Comparison of NH4⁺ (top) and NO3⁻ (bottom) concentrations (as N) in weekly precipitation samples
 between SANDS glass precipitation sampler and standard weekly NADP/NTN polyethylene bucket.

Air concentrations of HNO₃ and NH₃

HNO₃ and NH₃ were measured by several methods during SANDS (Table 2), including continuous analyzers (MARGA and TD-PC-CL) and time integrated methods (URG denuder and CASTNET). Here we briefly compared the various methods, including those that were collocated at NC25/COW137 or operated concurrently at NC25/COW137 and the eddy flux tower (EFT) (Table 1). URG denuder/filter pack results were used as reference when available. Comparison to the collocated URG denuder (3- to 4-hour integration) on the EFT showed the MARGA underestimated HNO₃ by approximately 22%, though the two methods were highly correlated and showed no offset (i.e., intercept = 0) (Figure S2). This underestimate could be caused by HNO₃ sorption to the MARGA inlet,

15 which is a 30 cm length of 1.27 cm O.D. PFA tubing. Conversely, TD-PC-CL HNO₃ measured at NC25/COW137

2

was 23% higher than the URG denuder on the EFT, again showing good correlation and no offset. TD-PC-CL compared well (slope = 0.86) with CASTNET weekly HNO₃ concentrations (August 2015 to August 2016), particularly when considering the low weekly average concentrations observed at Coweeta.



5 Figure S2. Comparison of HNO₃ and NH₃ measurements.

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Comparison of hourly HNO₃ concentration measured by TD-PC-CL and MARGA during spring and summer 2016 is presented in Figure S3. MARGA measurements reflect a measurement heigh of 37.5m during spring and 43.5m during summer. The methods tracked very well temporally, showing diurnal patterns with a peak during the mid-day and a minimum at night. However, the MARGA consistently measured lower mid-day peak HNO₃ concentrations during

- both intensives. Similar to the MARGA underestimation relative to the URG denuder, this pattern may reflect loss of HNO₃ to the MARGA inlet. The disagreement is most pronounced during spring when relative humidity was higher, potentially causing more extensive sorption on tubing surfaces. Though less likely, overestimation by the TD-PC-CL method may occur if NOy species other than HNO₃ are scrubbed the KCl denuder (Section 2.2.2). Finally, the
- 15 possibility that the difference between methods partly reflects real spatial differences cannot be ruled out. The MARGA system was deployed approximately 7.5 m (spring) and 13.5 m (summer) above the forest canopy approximately 300 m to the southeast of the TD-PC-CL. The TD-PC-CL system was located in an open grassy area near the Coweeta offices and sampled from a height of 8 m above ground. Differences in dry deposition rates to the

forest (higher) versus grassy area (lower) and potentially higher NO₂ concentrations (HNO₃ precursor) near the Coweeta offices may result in higher daytime HNO₃ concentrations at the TD-PC-CL location.



Figure S3. Time series of measured hourly HNO₃ by MARGA and TD-PC-CL during spring and summer of 2016.

Ammonia measured by the MARGA showed good correlation with the collocated URG denuder and no systematic offset but underestimated by ~29% on average (Figure S2). Similar to HNO₃, this underestimation is attributed to NH_3 loss to the MARGA inlet tubing. The bi-weekly integration period of the AMoN sample, and subsequent small

10 sample size, precludes comparison to the seasonal intensive URG and MARGA measurements.

Air concentrations of PM

Comparison of particulate SO_4^{2-} and NH_4^+ concentrations measured by Tisch Hi-Vol $PM_{2.5}$ and CASTNET samplers is presented in Figure S4. 24-hr Tisch Hi-Vol measurement data were averaged to match the weekly CASTNET measurements. Data points with completeness less than 75% were exclude from comparison. The methods compared

well overall, exhibiting high correlation and offsets near zero. CASTNET concentrations were ~ 15% higher, on average, than Hi-Vol concentrations. Both methods (not shown) measured very low concentrations of NO_3^- during the periods of comparison. While typically found mostly in the fine mode, differences in SO_4^{2-} and NH_4^+ may be partly attributed to measurement of different particle sizes. CASTNET does not have a size selective inlet while the Hi-Vol

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measures $PM_{2.5}$. Thus, CASTNET samples some portion of the coarse particle fraction. The observation that particulate NO_3^- by CASTNET sampler was higher than the Hi-Vol (average of 0.14 and 0.018 µg m⁻³, respectively) also suggests the collection of some coarse nitrate by CASTNET. Differences in NH_4^+ and NO_3^- may also relate to the flow rate of the two samplers. The significantly higher flow rate of the Hi-Vol sampler (230 L min⁻¹) may promote greater volatilization of NH_4NO_3 than CASTNET (1.5 L min⁻¹).



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Figure S4. Comparison of particulate SO₄²⁻ and NH₄⁺ concentrations measured by co-located Tisch Hi-Vol PM_{2.5} and CASNET samplers.

Passive HNO3 sampler

15 In order to quantify ambient HNO₃ concentrations with the passive samplers, a calibration was conducted by comparing the passive sampler with a collocated CASTNET filter pack sampler at Screwdriver Knob. A median

"effective" passive sampling rate was calculated over the full collocation period. The calibrated passive measurement is compared to the collocated filter pack measurement in Figure S5. Average(median) concentrations of the data shown in the graph are 0.09(0.1) and $0.1(0.1) \mu g m^{-3}$ for the passive and filter pack, respectively. The sampling rate and correction for elevation (pressure) were applied to the passive HNO₃ samplers deployed across the basin (Figure 1, Table 1, Figure 13).



Figure S5. Comparison of passive and CASTNET HNO₃ used to derive an effective passive sampling rate.

S2. TD-PC-CL Instrument

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The thermal decomposition, photolytic conversion, chemiluminescence (TD-PC-CL) instrument at Coweeta used a 10 cm long 0.635 cm O.D. PFA Teflon inlet line, a 0.5 L PFA ballast tank, three 30 cm long by 0.5 cm diameter quartz converter tubes and three model BLC2 second generation blue LED converters (Air Quality Design, Inc., Wheat Ridge, CO) to sample inlet air, thermally convert total peroxy nitrates (Σ PNs) and total alkyl nitrates (Σ ANs) to NO₂

- 15 and then photolyze NO_2 to NO. The inlet was mounted approximately 8 meters above ground level and 5 m horizontal distance from the equipment shelter. Ambient air was drawn through a single inlet line into the ballast tank where flow was split into three separate channels. The purpose of the ballast tank was to provide approximately 16 seconds residence time to the system in order to smooth the signal as the analyzer switches between channels. Channel 1 air (baseline) flowed into a quartz converter tube maintained at shelter temperature. From there, the sample entered the
- 20 pre-reactor channel of a Thermo-Environmental Model 42i NO-NOx analyzer and the ambient NO₂ plus analyzer background signal was measured. In normal operation, the pre-reactor in the Model 42i was used to measure signal from instrument background plus non-NO interferences. For this application, the O₃ feed to the pre-reactor was

disconnected so that it could function as an independent measurement channel. The baseline signal from Channel 1 measured instrumental noise, plus chemical interferences, plus NO and NO₂.

Channel 2 air was directed through a quartz converter tube heated to 180 °C. Air from the converter then flowed through 10 cm of 0.635 cm O.D. stainless steel tubing to cool the sample, then into a second blue LED converter and

- 5 finally the NO channel of the NO-NOx analyzer. Channel 2 thus measures baseline plus ΣPNs. Channel 3 air followed an identical path, except the quartz converter was heated to 360°C, then into the blue LED and the NOx channel of the analyzer. Channel 3 measured baseline plus ΣPNs plus ΣANs. Prior to deployment, the NO-NOx detector was calibrated with NIST-traceable NO (Scott-Marrin, Riverside, CA) and
- conversion factors for the blue LEDs were determined with NO₂ (Scott-Marrin). Challenges with isopropylnitrate
 (Scott-Marrin) concentrations between 2 and 40 parts per billion (ppb) showed no response on either Channel 1 or
 Channel 2 and 96±2 % recovery on Channel 3. Post-deployment challenges with isoproylnitrate showed slightly lower
 recoveries on Channel 3 (i.e., 94 ±3 %).

In the field, detector responses and converter efficiencies were checked with NO and NO_2 , respectively, every three days. Zero air zeros and dynamic zeros (heaters turned off) were performed every 3 days and 7 days, respectively.

15 Both types of zeros indicated 1-sigma detection limits of 0.018-0.027 ppb. Data processing involved acquisition of signals from all three channels with 1-minute time resolution, followed by averaging to 5-minute intervals and adjustment of each channel for NO₂ response. PNs were then calculated as the difference between adjusted Channel 2 and adjusted Channel 3, and ANs were calculated as the difference between adjusted Channel 3 and adjusted Channel 2.

Scientific name	Common name	Basal area		Leaf area i	ndex
		$(m^2 ha^{-1})$		(m ² m ⁻²)	
Betula lenta L.	Black (sweet) birch	3.19	10.90%	1.05	22.60%
Liriodedron tulipifera L.	Tulip (yellow) poplar	6.94	23.80%	0.81	17.50%
Quercus alba L.	White oak	5.09	17.50%	0.65	14.00%
Rhododendron maximum L.	Great (rosebay) rhododendron	4.39	15.10%	0.6	12.90%
Acer rubrum L.	Red maple	2.18	7.50%	0.44	9.50%
Nyssa sylvatica Marsh.	Blackgum	2.06	7.10%	0.33	7.10%
Oxydendrum arboreum L. (DC.)	Sourwood	1.89	6.50%	0.27	5.80%
Carya spp.	Hickory species	0.8	2.70%	0.13	2.80%
Fagus grandifolia Ehrh.	American beech	0.62	2.10%	0.1	2.20%
Quercus velutina Lam.	Black oak	0.46	1.60%	0.08	1.70%
Cornus florida L.	Flowering dogwood	0.32	1.10%	0.05	1.10%
Kalmia latifolia L.	Mountain laurel	0.25	0.90%	0.04	0.90%
Carpinus caroliniana Walter	American hornbeam	0.19	0.70%	0.03	0.60%
Quercus rubra L.	Red oak	0.17	0.60%	0.02	0.40%
Fraxinus americana L.	White ash	0.14	0.50%	0.02	0.40%
Tsuga canadensis L.	Eastern hemlock	0.31	1.10%	0.01	0.20%
Pinus strobus L.	Eastern white pine	0.08	0.30%	0.01	0.20%
Magnolia fraseri Walter	Mountain (Fraser) magnolia	0.05	0.20%	< 0.01	<0.2%
Ilex opaca Aiton	American holly	0.02	0.10%	< 0.01	<0.2%
TOTAL		29.14		4.64	

Table S1. Species list, basal area, and maximum leaf area index, based on allometric equations from four 25 × 25 m plots near the base of the EFT, sorted by leaf area index (Oishi et al., 2018).

S3. Above-canopy Gradient Flux Measurements

During the 2016 summer intensive a total of 19 vertical concentration profiles for reactive nitrogen compounds were measured during daytime (typically 0800 – 1700) using a glass annular denuder/filter pack (URG Corporation, Chapel Hill, NC) system as described in Section 2.2.2. The above canopy concentration gradients for HNO₃ and NH₃ were obtained from the measurements at 43 m and 34.6 m and the deposition fluxes were calculated by the MBR method.

10 There were three and two profiles for HNO₃ and NH₃, respectively, in which the measurements at 34.6 m were missing or of poor quality. In these cases, the measurements at a lower level (32.0 m) were used to calculate the concentration gradient, although they may suffer larger uncertainty due to greater influence of roughness sublayer (canopy height = 30 m). There are one and four concentration profiles for HNO₃ and NH₃, respectively, exhibiting counter-gradient phenomenon (i.e., emission) above canopy. The sample durations of the concentration profiles were typically 3 or 4

15 hours. The hourly K_t values were first calculated from the hourly heat flux and temperature gradient measurements and then averaged for the 3-4 hour sample period after omitting obvious outliers (points > 3 scaled median absolute deviation (MAD) away from the median).

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Resistance component	Formulation
Aerodynamic resistance	$R_a = u/u_*^2$
Leaf boundary layer resistance	$R_{bl} = \frac{v}{u_* D_c} \left(\frac{u_* l}{v LAI^2}\right)^{1/3}$
Stomatal resistance	$R_{s} = \frac{r_{s,\min}}{LAIf_{PAR}f_{T}f_{vpd}f_{w}}\frac{D_{H_{2}O}}{D_{c}} + \frac{1}{LAI(H^{*}/3000+100f_{0})}$
Cuticular resistances	$R_{cut} = \left[LAI \left(\frac{f_{wetleaf}}{R_{cut,wet}} + \frac{1 - f_{wetleaf}}{R_{cut,dry}} \right) \right]^{-1}$
	For NH ₃ , $R_{cut,dry} = R_{cut,\min} e^{\alpha_{cut}(100-RH)}$
	For other reactive N compounds, $R_{cut,dry} = R_{cut0} a_0 / f_{rel}$
	$R_{cut,wet} = R_{mt} + R_{mt} / (W_L H^*)$
	$R_{mt} = \left(k_{mt}r\right)^{-1}$
Total ground resistance	$R_g = R_{inc} + R_{bg} + R_{soil}$
In-canopy aerodynamic resistance	$R_{inc} = R_a \left(e^{0.5LAI} - 1 \right)$
Ground boundary layer resistance	$R_{bg} = \frac{\nu/D_c - \ln(\delta_0/z_1)}{ku_{*g}}$
Soil resistance	For NH ₃ , $R_{soil} = L_{dry}/D$
	For other reactive N compounds,
	$R_{soil} = \left(\frac{f_{wetsoil}}{R_{soil,wet}} + \frac{1 - f_{wetsoil}}{R_{soil,dry}}\right)^{-1}$
	$R_{soil,dry} = R_{soil0} a_0 / f_{rel}$
	$R_{soil,wet} = R_{mt} + R_{mt} / (W_L H^*)$
Note. $u =$ mean wind speed: $u_* =$ friction v	velocity: $v =$ kinematic viscosity of air: $D_c =$ molecular diffusivity of a

Table S2. Summary of resistance formulas implemented in STAGE.

specific gas; l = length scale over which the viscous sub-layers are permitted, which is a typical leaf width; LAI = leaf area index; $r_{s,min} =$ minimum leaf stomatal resistance for water vapor; $f_{PAR} =$ environmental stress function of radiation; $f_T =$ environmental stress function of temperature; $f_{vpd} =$ environmental stress function of humidity; $f_w =$ environmental stress functions of leaf water potential; $D_{H2O} =$ molecular diffusivities for water vapor; $H^* =$ effective Henry's Law constant; $f_0 =$ reactive factor; $f_{wetleaf} =$ fraction of wet canopy leaf; $R_{cut,min} =$ minimum cuticular resistance of NH₃; $a_{cut} =$ an empirical factor; RH = relative humidity; $R_{cut0} =$ reference value for cuticular resistance; $a_0 =$ a constant value; $f_{rel} =$ relative reactivity; $W_L =$ liquid water content fraction; $k_{mt} =$ mass transfer coefficient of a

specific gas; r = droplet radius; $\delta_0 =$ distance above ground where the eddy diffusivity is equal to molecular diffusivity; $z_1 =$ upper height of logarithmic profile that forms above ground; $u_{*g} =$ friction velocity at ground level; $L_{dry} =$ soil dry layer thickness; D = gas diffusivity within soil; $f_{wetsoil} =$ fraction of wet soil surface; $R_{soil0} =$ reference value for soil resistance.

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S4. Canopy Physical Characteristics

Canopy total leaf area (m² m⁻²) was estimated along a 370 m transect during the spring and summer 2016 intensives using a wide-angle canopy imager (CID Bio-science CI-110 Plant Canopy Imager) at an average spacing of 28 m

- 10 between 15 image locations (Breda, 2003). The transect origin was randomly selected within the Coweeta EFT footprint, directed from the northeast to the southwest of the tower. The CI-110 (Software v1.1.71) estimates plant area index (PAI) through a gap-fraction inversion procedure (Campbell and Norman, 1998). PAI is a whole tree index, including branches, stem and leaf components of a tree. To avoid sampling and optical errors associated with view angles close to zenith and the horizon, a 30° 60° zenith angle range was selected for analysis (Leblanc et al., 2005).
- 15 Early morning and late evening collections ensured diffuse light conditions necessary for consistent exposure across the entire image. A species-specific woody-to-total (W:T) correction was applied to the *in situ* PAI to arrive at leaf area index (LAI) (Iiames et al., 2008). The CI-110 LAI estimates were then calibrated to LAI estimated using treespecific allometric equations developed through destructive sampling in the Coweeta basin (Martin et al., 1998). These equations were then applied to forest stand measurements (species, density, size class) within the tower footprint to
- 20 estimate LAI (Oishi et al., 2018; Table S1). The allometry based estimate of peak summer LAI (4.64 m² m⁻²) shown in Table S1 was used to adjust both the CID measurements and the MODIS LAI estimates as described below. The MODerate resolution Imaging Spectroradiometer (MODIS) global LAI product (MCD15A2H) was used to develop a continuous time series of LAI for deposition modeling. MODIS LAI estimates are generated daily at a 500 m spatial resolution and each data point covers an 8-day period. The primary MODIS algorithm solves a 3-
- 25 dimensional radiative-transfer model using atmosphere-corrected MODIS spectral surface reflectance and biome identification (Myneni et al., 2002). LAI for the eddy flux tower (EFT, Figure 1, Table 1) location was estimated from values of the surrounding four grid points by using inverse distance weighted interpolation. Raw MODIS data were corrected for MODIS QC, including CloudState, Confidence Score, Snow_Ice, Aerosol, Cirrus, Internal_CloudMask, and Cloud_Shadow flags then
- 30 smoothed and gapfilled. After these processing steps, the summer maximum MODIS LAI at the tower site was 5.62 m² m⁻², which is higher than the allometry base estimate of 4.64 m² m⁻² (Oishi et al., 2018) described above. A ratio of 0.8256 was applied to scale the summer MODIS LAI to the allometry based estimate. The minimum MODIS LAI at the tower site was 0.5 m² m⁻², which is close to the allometry based LAI estimate for *Rhododendron maximum L*. (0.6 m² m⁻²) shown in Table S1 and was therefore not adjusted. The daily time series of MODIS LAI used for
- 35 deposition modeling is shown in Figure S6 along with the spring and summer CID transect measurements of total LAI described above. The CID measurements are included to illustrate variability in LAI across the landscape surrounding the tower.



Figure S6. Time series of MODIS LAI used for deposition modeling and CID LAI measurements collected in 2016.



5 Figure S7. Diel profile of NH₃ concentration measured by the MARGA. Data from spring and summer 2016 intensives are combined. Observations represent median hourly concentration and bars represent interquartile range (~ 65 to 75 observations in each hourly bin).



Figure S8. Diel profile of measured "Other" NOy (i.e., $NO_y - HNO_3 - \Sigma PN - \Sigma AN$). Observations represent median hourly concentration and bars represent interquartile range.

Table S3. Summary of options used in the Community Multi-scale Air Quality (CMAQ V5.2.1) model and coupled Weather Research and Forecasting (WRF) meteorological model.

WRF Model options	
WRF version	V3.8
Land cover data set	NLCD 2011
Land-surface model	Pleim-Xiu
Data assimilation	Four-Dimensional Data Analysis (FDDA) with no nudging in the PBL
Microphysics module	Morrison double-moment
Radiation module	Rapid Radiative Transfer Model Global (RRTMG)
Convective parameterization	Kain-Fritsch
Lightning data assimilation	yes
MCIP post-processing	V4.3
CMAQ Model options	
CMAQ version	V5.2.1
Chemical mechanism	CB6r3
Aerosol module	CMAQ Aerosol Module version 6 (AE6)
Deposition module	M3DRY
Lightning NOx	yes
Bidirectional NH3 flux	yes
Emissions platform	2015fd
Boundary conditions	CMAQ v5.2 hemispheric model



Figure S9. Long-term CASTNET measurements of SO₂ (a) and SO₄²⁻ (b) air concentrations at site COW137.

		Mean	Median	S.D.	Q1	Q3	Min.	Max.
Concentration	$\mathrm{NH_{4}^{+}}$	204.6	130.1	263.5	66.0	297.0	16.0	1808.9
μg N L ⁻¹	NO ₃ -	147.9	133.9	109.2	71.7	176.0	15.7	611.4
	NO ₂ -	1.1	0.5	1.8	0.0	1.2	0.0	8.4
	WSON	51.6	25.7	94.8	10.8	53.6	0.1	559.6
%WSTN	$\mathbf{NH_4^+}$	47.0	46.8	8.3	41.0	51.9	26.9	73.9
	NO ₃ -	41.7	42.2	8.3	36.3	46.7	21.0	58.7
	NO ₂ -	0.3	0.2	0.5	0.0	0.4	0.0	2.5
	WSON	11.0	9.4	7.7	6.1	13.8	0.1	38.0

Table S4. Summary of NH₄⁺, NO₃⁻, NO₂⁻, and WSON concentrations in precipitation (μ g N L⁻¹) and % contribution to WSTN. *N* = 52 observations.

	Period	NOy	HNO ₃	ΣΡΝ	ΣΑΝ
Mean	Winter	1.318	0.115	0.080	0.074
	Spring	1.215	0.222	0.127	0.104
	Summer	0.643	0.127	0.101	0.104
	Fall	0.774	0.073	0.083	0.066
	Annual	1.002	0.140	0.099	0.089
Median	Winter	1.103	0.073	0.059	0.051
	Spring	1.095	0.130	0.098	0.086
	Summer	0.562	0.060	0.077	0.087
	Fall	0.635	0.036	0.059	0.047
	Annual	0.847	0.072	0.072	0.067
S.D.	Winter	0.950	0.118	0.070	0.075
	Spring	0.663	0.240	0.108	0.079
	Summer	0.410	0.145	0.085	0.076
	Fall	0.560	0.092	0.070	0.063
	Annual	0.739	0.173	0.088	0.076
Q1	Winter	0.723	0.031	0.033	0.027
	Spring	0.715	0.048	0.047	0.043
	Summer	0.307	0.024	0.043	0.046
	Fall	0.341	0.015	0.033	0.026
	Annual	0.486	0.027	0.038	0.035
Q3	Winter	1.640	0.165	0.107	0.092
	Spring	1.587	0.335	0.182	0.145
	Summer	0.897	0.195	0.137	0.146
	Fall	1.033	0.097	0.118	0.087
	Annual	1.323	0.196	0.137	0.123
Ν	Winter	2027	1909	1807	1797
	Spring	2130	2038	1858	1847
	Summer	2117	1966	2061	2035
	Fall	1477	1379	1233	1229
	Annual	7751	7292	6959	6908

Table S5. Summary of seasonal and annual measured NOy, HNO₃, Σ PN, Σ AN concentrations (ppb). S.D. represents 1 standard deviation, Q1 and Q3 represent 1st and 3rd quartiles (interquartile range), respectively, and *N* represents the number of observations.

Table S6. Mean seasonal and annual percent (%) contribution of measured HNO₃, Σ PN, Σ AN, and "Other" compounds to total NO₃. Other NO₃ is calculated at NO₃ – HNO₃ – Σ PN – Σ AN. Number of hourly observations (*N*) is also shown.

	Other NO _y	HNO ₃	ΣΡΝ	ΣΑΝ	Ν
Winter	75.8	8.6	8.1	7.5	1723
Spring	61.7	16.0	11.4	11.0	1777
Summer	47.1	16.2	16.7	20.0	1925
Fall	68.5	8.6	12.1	10.9	1160
Annual	62.3	12.8	12.2	12.7	6585

Table S7. Summary of NH_4^+ , NO_3^- , NO_2^- , and WSON concentrations in Hi-Vol PM_{2.5} samples and % contribution to WSTN. N = 103 observations.

		Mean	Median	S.D.	Q1	Q3	Min.	Max.
Concentration	$\mathrm{NH_{4}^{+}}$	0.264	0.248	0.123	0.192	0.331	0.076	0.795
$\mu g \ N \ m^{-3}$	NO_3^-	0.004	0.003	0.004	0.002	0.005	0.000	0.020
	NO_2^-	0.001	0.001	0.001	0.000	0.001	0.000	0.007
	WSON	0.037	0.026	0.036	0.010	0.054	0.000	0.140
%WSTN	$\mathrm{NH_{4}^{+}}$	86.8	89.5	10.2	81.4	94.7	51.4	100.0
	NO ₃ -	1.4	0.9	1.4	0.6	1.8	0	7.0
	NO_2^-	0.3	0.2	0.4	0.1	0.3	0	3.6
	WSON	11.5	9.1	10.0	3.9	17.5	0	45.4

		Mean	S D	Median	P25	P75	Min	Max	N
		Wiedii	5.0.	Wiedian	125	175	IVIIII	Mux	
Green Leaves	pН	4.7	0.7	4.6	4.2	5.4	3.0	6.3	75
	$\mathbf{NH_4}^{+*}$	66.3	65.6	47.0	12.7	105.0	0.1	294.0	75
	Γ	267.0	630.4	35.8	10.3	171.0	0.0	4070.0	75
Senescent Leaves	pН	4.6	0.6	4.6	4.2	4.9	3.5	5.7	21
	$NH_4{}^{+\ast}$	346.2	338.0	186.0	72.7	637.0	23.5	1110.0	21
	Γ	784.5	1698.8	113.0	52.2	525.0	3.4	7370.0	21
Litter	pН	5.1	0.6	5.0	4.5	5.5	3.9	6.1	65
	$NH_4{}^{+\ast}$	44.6	24.4	36.8	26.4	58.2	11.5	104.7	65
	Г	314.7	464.7	69.3	26.0	494.2	4.9	2197.2	65

Table S8. Statistical summary of green leaf, senescent leaf and litter tissue chemistry along with Γ calculated from tissue chemistry.

* μ g NH₄⁺ g fresh tissue⁻¹

5 Table S9. Median statistics and number of observations of tissue chemistry for individual green leaf samples by species. Note single observations for Sassafras and Eastern Hemlock.

Scientific Name	Common Name	Ν	pН	$\mathrm{NH_4^{+}}^*$	Г
Acer rubrum L.	Red maple	9	4.2	156.0	46.4
Betula spp.	Birch species	1	3.9	7.1	1.0
Carya spp.	Hickory species	5	4.4	93.9	44.2
Fagus grandifolia Ehrh.	American beech	6	4.7	20.3	29.7
Ilex opaca Aiton	American holly	3	5.4	110.0	613.0
Kalmia latifolia L.	Mountain laurel	4	5.2	5.6	32.9
Liriodedron tulipifera L.	Tulip (yellow) poplar	7	5.7	129.0	1750.0
Magnolia fraseri Walter	Mountain (Fraser) magnolia	3	5.6	23.7	313.0
Oxydendrum arboreum L. (DC.)	Sourwood	4	3.2	6.4	0.3
Pinus strobus L.	Eastern white pine	3	4.1	21.7	6.8
Quercus alba L.	White oak	9	4.5	25.3	23.5
Quercus coccinea L.	Scarlet oak	4	4.3	37.5	13.4
Quercus prinus L.	Chestnut oak	3	5.4	139.0	719.0
Quercus velutina Lam.	Black oak	5	4.5	52.6	44.4
Rhododendron maximum L.	Great (rosebay) rhododendron	7	5.2	4.2	14.6
Sassafras albidum (Nutt.)	Sassafras	1	5.8	91.9	1090.0
Tsuga canadensis L.	Eastern hemlock	1	3.6	10.0	0.8

 $*NH_4^+$ (µg g fresh tissue⁻¹)

	Mean	S.D	Median	Q1	Q3
NH ₃	1.70	1.21	1.41	0.84	2.26
$\mathbf{NH_4^+}$	0.08	0.09	0.05	0.03	0.10
HNO ₃	3.13	1.99	2.75	1.63	4.22
NO ₃ -	0.07	0.07	0.04	0.02	0.08
NO_2	0.16	0.21	0.07	0.03	0.15
AN	0.52	0.57	0.29	0.21	0.56
PN	0.32	0.15	0.26	0.21	0.39
PON	0.07	0.08	0.04	0.02	0.08

Table S10. Statistical summary of hourly modeled deposition velocities (cm s⁻¹) for individual N compounds (N = 8784).

S5. Sensitivity of Dry Deposition to Model Parameterizations

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Sensitivity tests of dry deposition model parameterizations are summarized in Table S11. Leaf area index is a key model input, as fluxes of all species scale with the canopy surface area available for dry deposition. In our analysis we use LAI derived from MODIS for the annual dry deposition simulation (Supplemental Figure S6). Ground-based measurements of LAI along transects in the vicinity of the forest flux tower show large spatial variability, with individual values scattering around the MODIS estimate. The sensitivity of the modeled dry deposition to LAI is evaluated by adjusting LAI in the base case simulation by \pm 30%, equivalent to the mean of the ground-based observations \pm the relative standard deviation. Increasing LAI by 30% from the base case increases total dry deposition by 0.3 kg ha⁻¹ (+11.6%), increasing the contribution of dry to total deposition from 38.8% to 41.4%. Decreasing LAI by 30% reduces total dry N deposition by 0.35 kg ha⁻¹ (-13.7%), subsequently reducing the

- 15 contribution of dry to total deposition from 38.8% to 35.3%. As the dominant deposition pathway for NH₃, it is important to understand the sensitivity of the model results to the cuticular resistance parameterization (R_{cut} , Table S2). The cuticular resistance for NH₃ is typically parameterized as a function of LAI, surface wetness, and the amount of NH₃ dissolved in water residing on the cuticle surface (Pleim et al., 2013) or its pH (van Hove et al., 1989). In STAGE, R_{cut} is specified for wet periods (i.e., $R_{cut,wet}$) (i.e., macroscale
- 20 wetness including rain and dew) and periods that are considered dry ($R_{cut,dry}$) but will include microscale wetness (i.e., thin layers on the cuticle surface) at high relative humidity (RH) (Table S2). Here we restrict our analysis to the role of $R_{cut,dry}$, as it dominates (78% of hourly periods) R_{cut} . R_{cut} cannot be measured directly, rather it is typically inferred from night-time canopy-scale NH₃ flux measurements under the assumption that the stomatal flux pathway is closed and the ground flux is negligible (Massad et al., 2010). Such datasets show a clear, generally non-linear, relationship
- 25 with RH, indicating a reduction in $R_{cut,dry}$ as thin water layers form on the cuticle surface (Massad et al., 2010) at high RH. The minimum cuticular resistance ($R_{cut,min}$) demonstrates a relationship with pH of the cuticle surface water, parameterized as the ratio of total acid to NH₃ in the atmosphere (i.e., acid ratio, Nemitz et al., 2001). Here we assume an acid ratio of 1, yielding $R_{cut,min} = 31.5$ (Massad et al., 2010). An empirical factor (α_{cut}) defines the form of the exponential relationship between $R_{cut,dry}$ and RH, thus exerting important control on the dynamics of F_{cut} .

Based on metanalysis of existing datasets, Massad et al. (2010) separate α_{cut} by ecosystem type based on the expectation that the factors controlling the relationship between RH and formation of microscale water layers on the cuticle, such as hygroscopicity and aerosol uptake, will differ by plant species (Massad et al., 2010). For forests, the mean and standard deviation of α_{cut} reported by Massad et al. (2010) are 0.0318 and 0.0179, respectively. In our

- 5 analysis, the sensitivity of F_{net} to $R_{cut,dry}$ is assessed by varying α_{cut} by ± 0.0179 , equivalent to a change of $\pm 56\%$. Increasing α_{cut} increases $R_{cut,dry}$, thereby decreasing NH₃ dry deposition by 0.25 kg N ha⁻¹ (-19.0%) relative to the base case. Total dry N deposition is reduced by 10%, subsequently reducing the contribution of dry to total deposition from 38.8% to 36.4%. Decreasing α_{cut} decreases $R_{cut,dry}$, thereby increasing NH₃ dry deposition by 0.39 kg N ha⁻¹ (29.8%) relative to the base case. Total dry N deposition increases by 14.7%, subsequently increasing the contribution
- 10 of dry to total deposition from 38.8% to 42.1%.

Emission potentials (Γ) of the ground and vegetation are key inputs to the model, as they govern the surface compensations points and subsequently the direction and magnitude of the component (e.g., ground and canopy) and net canopy-scale fluxes. As described above, parametrization of the leaf and litter emission potentials based on bulk tissue chemistry contains uncertainty. While the magnitude of the uncertainty is not known, the sensitivity of the net

- 15 canopy-scale flux to Γ can be assessed by varying the litter (Γ_1) and stomatal (Γ_s) emission potentials together and individually within the IQR of the observations (Table S8) assuming that uncertainty is ultimately less than naturally observed variability. Simultaneously reducing Γ_1 and Γ_s to their corresponding 25th percentiles increases net NH₃ dry deposition by 0.04 kg N ha⁻¹ (+3.1%) relative to the base case. Simultaneously increasing Γ_1 and Γ_s to their corresponding 75th percentiles has a larger impact, decreasing NH₃ dry deposition by 0.44 kg N ha⁻¹ (-33.6%) relative
- to the base case. Total dry N deposition is reduced by 17.4%, subsequently reducing the contribution of dry to total deposition from 38.8% to 34.4%. Adjusting Γ_1 and Γ_s individually within their respective IQR while holding the other constant reveals a similar pattern. Larger responses are observed by increasing rather than decreasing Γ_1 and Γ_s . Individually increasing Γ_1 to the 75th percentile reduces NH₃ dry deposition by 23.7%, while increasing Γ_s reduces NH₃ dry deposition by 10.7%. This response reflects differences in the magnitude and variability of the measured Γ_1

25 and $\Gamma_{s.}$

The ground flux (F_g) is controlled by Γ_g and the total ground resistance (R_g), which is the sum of R_{inc} , R_{bg} and soil (R_{soil}) resistances (Table S2). For NH₃, R_{soil} is a function of the length of the dry soil layer through which NH₃ originating from the soil solution must diffuse to the atmosphere (Sakaguchi and Zeng, 2009). As noted above, analysis of the soil and litter chemistry along with in-canopy profiles of NH₃ air concentration suggests that it is more

- 30 appropriate to set the ground emission potential to that of the litter rather than the much more acidic underlying soil. We do so acknowledging that the physical process by which NH₃ diffuses from the litter layer to the atmosphere will differ from diffusion through the soil dry surface layer. However, while studies have investigated the potential role of leaf litter in NH₃ air-surface exchange above forests (Hanson et al., 2013; 2015), a parameterization for a litter layer resistance (R_{litter}) for forests has not been developed. Thus, we retain the current parameterization for R_{soil} and set Γ_g
- 35 = Γ_1 to calculate compensation points and F_g in STAGE. To test the potential implications of substituting R_{litter} for R_{soil} , we assess the impact to F_{net} of changing R_{soil} by a factor of ±2. Doubling R_{soil} or reducing it by half has a minor

effect on NH₃ dry deposition equivalent to $\pm \sim 0.04$ kg N ha⁻¹ ($\sim \pm 2.5\%$). This might be expected given that R_g depends not only on R_{soil} but also R_{inc} and R_{bg} . On average, R_{inc} , R_{bg} , and R_{soil} comprise 9.7%, 41.5%, and 48.8% of Rg (= $R_{inc} + R_{bg} + R_{soil}$) for NH₃. While development and implementation of a mechanistically representative R_{litter} to replace R_{soil} for forests is a necessary long-term goal, it is unlikely that it would dramatically alter the results presented

5 here.

The final sensitivity scenarios relate to the assumption of the particle size distribution in the calculation of NH_4^+ , NO_3^- , and PON fluxes. Three size distributions based on the data of Zhang et al. (2008) are shown in Supplemental Figure S11. The base STAGE model run assumes profile 1, which corresponds to the size distribution measured at clean sites of Zhang et al. (2008). Two more profiles (profile 2 and 3) were selected from the polluted sites for the purpose

10 of sensitivity tests. Adopting profile 2 shifts the distribution to smaller sizes, increasing particulate V_d and subsequently increasing total dry N deposition but only by a very small amount (0.06 kg N ha⁻¹ or 2.5%). Adopting profile 3 shifts the distribution to larger sizes, decreasing particulate V_d and subsequently decreasing total dry N deposition by an even smaller amount (<1%). Given the already much smaller V_d of particles relative to gases, the model results are relatively insensitive to assumptions of particle size distribution assumed in the V_d parameterization.

	NH ₃ dry	Dry	Total	NH ₃ dry	Drv
	kg N ha ⁻¹	kg N ha ⁻¹	kg N ha ⁻¹	% of Total Dry	% of Total Wet + Dry
Base	1.31	2.58	6.64	50.1	38.8
LAI+30%	1.49	2.88	6.94	51.9	41.4
LAI-30%	1.09	2.22	6.29	48.9	35.3
R _{cut,dry} +	1.06	2.32	6.40	45.5	36.4
R _{cut,dry} -	1.70	2.96	7.03	57.1	42.1
$\Gamma_{s,l}=P_{25}$	1.35	2.62	6.69	51.6	39.2
$\Gamma_{s,l}=P_{75}$	0.87	2.13	6.21	40.6	34.4
$\Gamma_s = \mathbf{P}_{50}, \Gamma_1 = \mathbf{P}_{25}$	1.33	2.60	6.67	51.2	39.0
$\Gamma_s = P_{50}, \Gamma_l = P_{75}$	1.00	2.27	6.34	44.2	35.8
$\Gamma_{\rm s}=P_{25}, \Gamma_{\rm l}=P_{50}$	1.33	2.60	6.67	51.2	39.0
$\Gamma_s = P_{75}, \Gamma_l = P_{50}$	1.17	2.44	6.51	48.0	37.5
R _{soil} x2	1.28	2.55	6.61	50.2	38.5
R _{soil} x0.5	1.35	2.61	6.68	51.4	39.1
PSD = Profile 2		2.64	6.71		39.4
PSD = Profile 3		2.57	6.64		38.7

Table S11. Results of model sensitivity testing. Annual NH₃ dry deposition, total dry deposition, and total deposition (kg N ha⁻¹) are reported, along with contribution (%) of NH₃ dry deposition to total deposition and contribution of dry deposition to total deposition, for combinations of alternative model parameterizations and inputs relative to the base model scenario.







Figure S11. Particle size distributions used in the calculation of NH_{4^+} , NO_3^- and PON fluxes (prf1) and alternative profiles used to test model sensitivity (prf2 and prf3). Particle diameter (Dp) and corresponding fraction (%) of total particulate mass are shown. Note that the mass median aerodynamic diameter (MMAD) and geometric standard deviation (GSD) used to generate the size distributions for NH_{4^+} and NO_3^- were from Table 5 of Zhang et al. (2008). PON was not measured at Zhang et al. (2008) and the values used here were taken as the mean of NH_{4^+} and NO_3^- .



Figure S12. Box-plots of deposition velocity (V_d) estimated by modified Bowen ratio (MBR), STAGE model, and maximum V_d as $1/(R_a + R_b)$.

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