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Atmospheric Deposition of Reactive Nitrogen to a Deciduous Forest in the Southern Appalachian Mountains

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Abstract. Assessing nutrient critical load exceedances requires complete and accurate atmospheric deposition budgets for reactive nitrogen (N_r). The exceedance is the total amount of N_r deposited to the ecosystem in excess of the critical load, which is the amount of N_r input below which harmful effects do not occur. Total deposition includes all forms of N_r (i.e., organic and inorganic) deposited to the ecosystem by wet and dry pathways. Here we present results from the Southern Appalachian Nitrogen Deposition Study (SANDS), in which a combination of measurements and field-scale modeling were used to develop a complete annual N_r deposition budget for a deciduous forest at the Coweeta Hydrologic Laboratory. Wet deposition of ammonium, nitrate, nitrite, and bulk organic N were measured directly. The dry deposited N_r fraction was estimated using a bidirectional resistance-based model driven with speciated measurements of N_r air concentrations (e.g., ammonia, ammonium aerosol, nitric acid, nitrate aerosol, bulk organic N in aerosol, total alkyl nitrates, and total peroxy nitrates), micrometeorology, canopy structure, and biogeochemistry. Total annual deposition was ~ 6.6 kg N ha^{-1} yr^{-1}, which is on the upper end of N_r critical load estimates recently developed for similar ecosystems in nearby Great Smoky Mountains National Park. Of the total (wet + dry) budget, 50.7% was contributed by reduced forms of N_r (NH_x = ammonia + ammonium), with oxidized and organic forms contributing ~ 41.6% and 7.7%, respectively. Our results indicate that reductions in NH_x deposition would be needed to achieve the lowest estimates (~ 3.0 kg N ha^{-1} yr^{-1}) of N_r critical loads in southern Appalachian forests.

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1 Introduction

Prior to the Industrial Revolution, Earth's ecosystems received reactive nitrogen (N_r) deposition rates of ~0.5 kg ha^{-1} yr^{-1} (Holland et al., 1999). Since the 19th century, anthropogenic activities, both industrial and agricultural, have resulted in unprecedented quantities of N_r being released into the atmosphere, subsequently altering biogeochemical cycles (Neff et al., 2002a,b; Ollinger et al., 2002; Bragazza et al., 2006; Doney et al., 2007; Galloway et al., 2008; Boonstra et al., 2017). Excessive atmospheric deposition of N_r to terrestrial ecosystems may lead to soil and aquatic acidification, nutrient imbalance and enrichment, plant damage and microbial community changes as well as loss of biodiversity (Bobbink et al., 1998; Lohse et al., 2008; Simkin et al., 2016). Nitrogen deposition rates in many areas now exceed 10 kg ha^{-1} $yr^{-1}$and may double current rates by the year 2050 in some regions (Galloway et al., 2008).

The amount of N_r deposition below which significant harmful effects do not occur is known as the critical load (Nilsson and Grennfelt, 1988). Critical loads can be quantified using empirical relationships between ecosystem N input and ecosystem response (Pardo et al., 2011; Root et al., 2015), or mass balance type biogeochemical models (Lynch et al., 2017; McNulty et al., 2007). For the southern Appalachian Mountains, simple mass balance approaches yield critical loads similar to those derived from empirical approaches for forest health and biogeochemical responses. In a recent study employing a mass balance model for the Great Smoky Mountains National Park, Pardo et al. (2018) quantified critical loads for spruce-fir, beech, and mixed deciduous forests in the range of 2.8 to 7 kg N ha^{-1} yr^{-1}, with the highest value corresponding to a high-elevation spruce-fir site experiencing disturbance-induced regrowth. Accurate and complete deposition budgets (i.e., including all forms of N) are required to quantify the amount of N input to ecosystems in excess of the critical load (i.e., the critical load exceedance).

Estimates of N deposition for critical load assessments can be derived from gridded chemical transport models (CTMs) (Ellis et al., 2013; Lee et al., 2016; Simkin et al., 2016; Clark et al., 2018; Makar et al., 2018), measurement-model fusion (MMF) techniques that combine measurements with CTM output (Schwede and Lear, 2014; Nanus et al., 2017; McDonnell et al., 2018; U.S. EPA, 2019a), or inferential modeling with site specific measurements (Flechard et al., 2011; Li et al., 2016). While these approaches reflect the state-of-the-science and are widely used, they collectively suffer from incompleteness of the N deposition budget (and are therefore biased low) (Walker et al., 2019a). Monitoring networks for wet deposition (NADP/NTN) and air concentrations of N_r (CASTNET) focus only on inorganic species, excluding organic forms of N which account for ~ 25% of total N in wet deposition on average (Jickels et al., 2013). Due to difficulties in sampling (Walker et al., 2012) and the inability to fully speciate the wide range of constituents (Neff et al., 2002a; Altieri et al., 2009, 2012; Cape et al., 2011; Samy et al., 2013, Chen et al., 2018), organic N is not routinely monitored. Hence, deposition of organic N remains uncertain, and thus N deposition budgets developed from network monitoring data and CTMs remain incomplete.

Current N deposition estimates also have a relatively high degree of uncertainty in the estimation of dry deposition. While wet deposition is routinely measured, direct measurements of dry N_r deposition (i.e., flux measurements) in North America are relatively few (Walker et al., 2020). Estimates of dry deposition for ecosystem assessments is therefore derived from models (Schwede and Lear, 2014; Li et al., 2016; Lee et al., 2016). Of the inorganic N species, NH_3 is the most important contributor to dry deposition in many areas (Walker et al., 2019b) but also the most

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uncertain (Flechard et al., 2011) due to the bi-directionality of surface-atmosphere exchange. A paucity of flux measurements (Walker et al., 2020) precludes bias correction of dry deposition in CTMs and MMF techniques, making dry deposition much more uncertain relative to wet deposition.

The Coweeta study site represents southern Appalachian Mountain forests, highly diverse and productive ecosystems that provide a variety of ecosystem services, including a source of surface drinking water (Caldwell et al., 2014). While deposition of oxidized N to forests in the southeastern U.S. has declined in response to the Clean Air Act, montane ecosystems continue to receive high rates of deposition due to elevation-induced precipitation gradients (Weathers et al., 2006; Knoepp et al., 2008). Southern Appalachian forests continue to show signs of sensitivity to N deposition. For example, litterfall N fluxes and foliar N concentrations at Coweeta have steadily increased over the past two decades (Knoepp et al., 2018). Highly spatially variable meteorological patterns typical of complex terrain are difficult to model (Lehner and Rotach, 2018), leading to uncertainties in precipitation amounts and wet deposition (Zhang et al., 2018) as well as the micrometeorological processes that govern dry deposition. Estimates of deposition from gridded CTMs in mountainous terrain therefore contain a higher degree of uncertainty relative to low-elevation ecosystems. For these reasons, a better understanding of total N deposition in southern Appalachian forests is needed.

This study investigates the N deposition budget in a remote montane forest in the southeastern U.S. We combine measurements of wet deposition, speciated air concentrations of N_r, micrometeorology, biogeochemistry and forest canopy structure with in-situ inferential dry deposition modeling to develop an annual, speciated, total N deposition budget, including net and component NH_3 fluxes as well as dry and wet organic N deposition. Seasonal and annual total N deposition fluxes are presented in the context of long-term trends in air concentrations and wet deposition of inorganic N species. Spatial representativeness is characterized using measurements of air concentrations of the primary inorganic N species and previous wet deposition measurements along an elevation gradient across the topographically complex forested basin.

2 Methods

2.1 Site description

The study was conducted at the USDA Forest Service Coweeta Hydrologic Laboratory, a 2,185-ha experimental forest in southwestern North Carolina, USA (35°3' N, 83°25' W) near the southern end of the Appalachian Mountain chain. Topography is complex, with elevations ranging from 675 to 1592 m within the Coweeta Basin. Mean annual temperature and precipitation are 12.9 °C and 1795 mm, respectively. Dominant overstory species are *Liriodendron tulipifera*, *Quercus alba*, *Betula lenta*, and *Acer rubrum*, which comprise 24%, 17%, 11%, and 8% of the basal area, respectively, in the low-elevation forests where the study was conducted (Oishi et al., 2018). The dominant understory woody shrub species is *Rhododendron maximum* (evergreen), which comprises 13% of the basal area (Oishi et al., 2018). Species composition in the vicinity of the eddy flux tower (EFT), further described below, is detailed in Table S1. Canopy height surrounding the EFT is ~ 30 m.

The Coweeta Basin has been a long-term monitoring site for atmospheric chemistry and deposition since the late 1970s. Weekly wet deposition of ammonium (NH_4^+) and nitrate (NO_3^-), along with sulfate (SO_4^{2-}), chloride, and base

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cations have been measured as part of the NADP/NTN (Site NC25, https://nadp.slh.wisc.edu/networks/national-trends-network/) since 1978. Weekly integrated air concentrations of particulate NH_4^+, NO_3^-, SO_4^{2-}, chloride, and base cations, as well as nitric acid (HNO_3) and sulfur dioxide (SO_2), have been measured by CASTNET (Site COW137, https://www.epa.gov/castnet) since 1987. Since 2011, bi-weekly integrated air concentrations of ammonia (NH_3) have been measured by the NADP Ammonia Monitoring Network (AMoN Site NC25, https://nadp.slh.wisc.edu/networks/ammonia-monitoring-network/). Here we use these datasets to place our study results into historical context, supplement the more intensive atmospheric chemistry measurements described below, and as inputs for inferential modeling of dry deposition. The long-term NADP and CASTNET measurements are collected in the lower part of the basin, indicated as NC25/COW137 in Figure 1.

2.2 Southern Appalachian Nitrogen Deposition Study

Building on the long-term NADP and CASTNET measurements described above, the Southern Appalachian Nitrogen Deposition Study (SANDS) was conducted in 2015 and 2016 to better understand the atmospheric chemistry and deposition of reactive nitrogen at Coweeta. Intensive measurement campaigns were conducted from May 21–June 9, 2015; August 6–25, 2015; September 9–26, 2015; April 19–May 11, 2016; and July 13–August 3, 2016. A subset of measurements was conducted continuously between May 2015 to August 2016. As described below, time-resolved and time-integrated measurement techniques were used to characterize organic N in the gas phase, in particulate matter, and in wet deposition; the temporal variability of air concentrations of gas-phase oxidized and reduced forms of N; and the spatial variability of atmospheric N concentrations across the Coweeta Basin. Vertical profiles of air concentrations were measured within the forest canopy to examine source/sink processes and measurements of soil and vegetation chemistry were conducted to characterize NH_3 emission potentials. Measurements were combined with NADP and CASTNET data to develop seasonal and annual total N deposition budgets employing inferential modeling for the dry deposition component. Vertical concentration profile and biogeochemical measurements were used to inform the parameterization of NH_3 bi-directional exchange. Sampling locations are described in Figure 1 and Table 1. Measurement details are summarized in Table 2.

2.2.1 Wet deposition

Additional wet deposition measurements were conducted adjacent to the NTN NC25 sampler to quantify the contribution of bulk water-soluble organic N (WSON) to water-soluble total nitrogen (WSTN) in precipitation. Weekly precipitation samples were collected in a modified wet-only sampler with a borosilicate glass funnel and amber glass bottle (Walker et al., 2012), shielded from sunlight, and maintained in the field under continuous refrigeration to maintain the stability of ON until retrieval (Walker et al., 2012). Samples were sent to the NADP Central Analytical Laboratory on ice for analysis of NH_4^+, NO_3^-, NO_2^-, and WSTN as described by Walker et al. (2012). WSON concentration was calculated as:

$$WSON = WSTN - (NH_4^+ + NO_3^- + NO_2^-) \quad (1)$$

The method detection limit for WSON is 10 μg N L^{-1} (Walker et al., 2012). These measurements were collected continuously from February 2015 to August 2016.

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During the spring of 2015, thymol was added to the precipitation collection bottle as a biocide to inhibit organic nitrogen loss in the sample should the refrigerated collector malfunction or lose power. Thymol negatively affected the precision of the total nitrogen measurement, and its use was discontinued in fall of 2015. Ultimately, there were no issues with the refrigerated collector, and the thymol-containing samples were excluded from the analysis presented herein. However, this data loss resulted in a gap from August 2015 to mid-October 2015 of the 12-month period for which the total deposition budget is developed. The data gap comprised eight weekly periods in which precipitation occurred. For this period, the NH_4^+ and NO_3^- concentrations from the collocated NADP/NTN NC25 sampler were used. Based on the SANDS measurements, the ON concentration during this period was estimated by assuming that $NH_4^+ + NO_3^-$ contribute 89% of total nitrogen in rainfall, with WSON representing the balance (11%). For the annual budget, weekly concentrations were combined with measured precipitation depth to calculate weekly deposition (kg N ha^{-1}). Comparison between SANDS and NTN concentrations of NH_4^+ and NO_3^- showed very good agreement (Supplemental Section S1, Figure S1).

2.2.2 Air concentrations

Hourly concentrations of NOy, HNO_3, total gas-phase peroxy nitrates (ΣPN), and total gas-phase alkyl nitrates (ΣAN) were measured continuously from August 2015 to August 2016 at the height of 8 m adjacent to the COW137 CASTNET tower (Figure 1, Tables 1 and 2). NOy and HNO_3 were measured using a modified model 42S NO-NO_2-NOx analyzer; the NOy technique is described in detail by Williams et al. (1998). Briefly, total oxidized reactive nitrogen (NOy) is converted to NO using a molybdenum catalyst heated to 325 ºC. On a second channel, a metal denuder coated with potassium chloride (KCl) is used to remove HNO_3 before passing through a second molybdenum converter heated to 325 ºC. The difference between the total NOy measurement and the HNO_3-scrubbed NOy measurement is interpreted as HNO_3. Here we refer to the method as denuder difference chemiluminescence (DD-CL).

Total peroxynitrates (ΣPNs) and total alkylnitrates (ΣANs) were measured using a modification of the technique described by Day et al. (2002), in which PNs and ANs are thermally decomposed to NO_2 (plus an organic radical), followed by measurement of the incremental NO_2 above ambient background for each decomposition step. Day et al. (2002) quantified NO_2 via laser induced fluorescence, while photolytic conversion to NO and quantification of the resulting NO by NO-O_3 chemiluminescence is used in the current study. Here we refer to the method as thermal decomposition, photolytic conversion, chemiluminescence (TD-PC-CL). A single chemiluminescence analyzer was used for NOy, HNO_3, ΣPN, and ΣAN measurements. Additional detail on the instrument and associated QA/QC procedures is included in Supplemental Section S2.

Hourly concentrations of NH_3 and HNO_3 were measured on the eddy flux tower (EFT, Figure 1, Tables 1 and 2) at two heights above the canopy (34 m and 37.5 m above ground during spring, 2016; 34 m and 43.5 m above ground during summer, 2016) using the Monitor for Aerosols and Gases in Ambient Air (MARGA, Metrohm Applikon B.V., the Netherlands). Details and principles of the MARGA system have been previously described (Rumsey and Walker, 2016; Chen et al., 2017). Briefly, the MARGA 2S consisted of two sampler boxes positioned on the tower and a detector box located in a climate-controlled enclosure at the base of the tower. Sample boxes comprised an inlet of

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1.27 cm outer diameter 30 cm long PFA Teflon tubing with no particle size selection, through which air flow was mass controlled at ~ 16.7 L min^{-1}, a wet rotating denuder (WRD) for collection of soluble gases, and a steam jet aerosol collector (SJAC). Liquid sample from the WRD and SJAC is continuously drawn from the sample boxes down the tower to the analytical box for analysis Ion Chromatography (IC) on an hourly basis at the detector unit located in a climate-controlled enclosure at the base of the tower. At the beginning and end of the measurement intensive, multi-level liquid NO_3^- and NH_4^+ standards were introduced at the WRD and SJAC, with airflow turned off, to assess the analytical accuracy of the NH_3 and HNO_3 measurement. MARGA measurements were conducted during the spring and summer of 2016 intensives. Comparisons of continuous and time-integrated methods for HNO_3 and NH_3 are summarized in Supplemental Section S1 (Figures S2 and S3).

Concentrations of NH_3, HNO_3, SO_2, NH_4^+, NO_3^-, and SO_4^{2-} in air were measured concurrently on the EFT at 10 heights from just above the forest floor (0.5 m above ground) to several meters above the canopy (upper height of 37 m above ground during spring 2016, 43.5 m above ground during summer, 2016) using a glass annular denuder/filter pack (URG Corporation, Chapel Hill, NC) system. The sampling assembly included a 1% Na_2CO_3 coated denuder for collection of acid gases followed by a 1% H_3PO_3 coated denuder for collection of NH_3, and a filter pack containing a primary Teflon filter for collection of aerosol and a backup Nylon filter (47 mm, Pall Corp, Port Washington, NY) to collect HNO_3 liberated by dissociation of NH_4NO_3 on the primary filter. Inlets were Teflon coated glass impactors with a nominal 2.5 μm aerodynamic diameter cutpoint (URG Corporation, Chapel Hill, NC). Sample durations were typically 3 or 4 h at a flow rate of ~ 16.7 L min^{-1}. Flow rates were controlled by critical orifice and were verified before and after each sampling period with a NIST traceable primary standard flow meter (Bios DryCal DC-Lite flowmeter, Mesa Laboratories, Inc., Lakewood, CO).

Denuders and filters were extracted with 10 mL of deionized water and analyzed by ion chromatography (IC, Dionex model ICS-2100, Thermo Scientific, Waltham, MA). Extracts were analyzed for cations using Dionex IonPac 2-mm CG12 guard and CS12 analytical column**s**; separations were conducted using 20 mM methanesulfonic acid (MSA) as eluent at a flow rate of 0.25 mL min^{-1}. Anions were analyzed (IonPac 2-mm AG23 guard column, AS23 analytical columns) using an isocratic eluent mix of carbonate/bicarbonate (4.5/0.8mM) at a flow rate of 0.25 mL min^{-1}. Multi-point (≥5) calibrations were conducted using a mixture prepared from individual inorganic standards (Inorganic Ventures, Christiansburg, VA). A mid-level accuracy check standard was prepared from certified standards mix (AccuStandard, New Haven, CT) for quality assurance/quality control. Profile measurements were conducted during each of the five SANDS intensives.

Bulk organic nitrogen in aerosols was measured using a high-volume (Hi-Vol) Tisch TE-1000 (Tisch Environmental, Cleves, OH) dual cyclone $PM_{2.5}$ sampler operated at a flow rate of 230 L min^{-1}. The unit was deployed at ground level adjacent to the COW137 CASTNET tower and collected 24 h (started at 7 A.M. local time) integrated samples on pre-baked (550 °C for 12 h) quartz fiber (QF) filters (90-mm, Pall Corp, Port Washington, NY). Field blanks were collected the same way except being loaded in the sampler without the pump switched on. A QF punch (1.5 cm^2) from each sample was extracted with DI water (18.2 MΩ·cm, Milli-Q Reference system, Millipore, Burlington, MA) in an ultrasonic bath for 45 min. The sample extract was filtered through a 0.2-μm pore size PTFE membrane syringe filter (Iso-disc, Sigma Aldrich, St. Louis, MO) before subsequent analyses.

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Water-soluble total N (WSTN) concentrations were measured using a high-temperature catalytic combustion and chemiluminescence method that included a total organic carbon analyzer (TOC-VCSH) combined with a total nitrogen module (TNM-1) (Shimadzu Scientific Instruments, Columbia, MD). Briefly, the TN module converts all nitrogen compounds to NO at 720 ºC in a combustion chamber, and NO is quantified by NO_2 chemiluminescence through reaction with ozone. A 5-point calibration was conducted with KNO_3 standard solution for each batch of samples. Before and after each batch of samples being analyzed, a suite of quality assurance check analysis including lab DI and accuracy check standard were conducted to ensure accuracy and precision. Inorganic species (NH_4^+, NO_3^-, NO_2^-) were analyzed by IC as described above, and WSON was calculated according to equation (1). Comparisons of time HiVol and CASTNET PM measurements are summarized in Supplemental Section S1 (Figure S4).

To evaluate the spatial distribution of gaseous N across the Coweeta Basin, additional passive sampling of HNO_3 and NH_3 was conducted across an elevation gradient for the full year of 2015 (Figure 1, Tables 1 and 2). Samplers were deployed for two-week periods at the height of 10 m above ground on an aluminum tilt-down tower. NH_3 measurements followed AMoN methods. HNO_3 was collected on 47-mm Nylon filters (Nylasorb, Pall Corp, Port Washington, NY) as described by Bytnerowicz et al. (2005). NH_3 and HNO_3 sampler preparation and analysis was performed by the NADP Central Analytical Laboratory and CASTNET laboratories, respectively. Field calibration of the passive HNO_3 measurements (Supplemental Figure S5) was based on comparison with a collocated CASTNET sampler at Screwdriver Knob site (Figure 1, Tables 1 and 2), which also operated for the full year of 2015. Air concentrations derived from passive samplers were corrected for temperature and pressure.

2.2.3 Micrometeorology

Site characteristics of micrometeorology and ecosystem fluxes of water and carbon dioxide have been previously described (Novick et al., 2013; 2014; Oishi et al., 2018). Three-dimensional wind components were measured by sonic anemometer (Model 81000, R.M. Young Company, Traverse City, MI) above the forest canopy on the EFT. Momentum and kinematic heat fluxes were determined by eddy covariance (EC) from sonic data. For EC calculations, raw 10-Hz sonic data were processed into hourly averages after block average detrending and 2D coordinate rotation (Novick et al., 2013). Air temperature and relative humidity were measured at the top of the tower (EC155, Campbell Scientific, Logan, UT) and 2/3 canopy height (HMP-45, Vaisala, Helsinki, Finland). Photosynthetically active radiation (PAR; LI-190, LI-COR Biosciences, Lincoln, NE) and upward and downward, shortwave and longwave radiation (CNR 4, Kipp & Zonen, Delft, The Netherlands) were measured at the top of the tower. Surface wetness was measured in the canopy crown, understory and at the ground using leaf wetness sensors (Model 237, Campbell Scientific). Soil volumetric water content (VWC) averaged over 0–30 cm depth was measured in four locations around the tower using time domain reflectometry probes (CS616, Campbell Scientific). Soil temperture was measured at four depths (5, 20, 35, and 55 cm) in two locations near the tower using thermistors. For missing data, linear interpolation was used to fill short gaps (1–4 h). Longer gaps were filled by substitution using the average hourly diel profile calculated for each month. Micrometeorological data were used for inferential dry deposition modeling as described below.

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2.2.4 Biogeochemistry

Ammonia emission potentials (Γ) and compensation points for live vegetation, leaf litter on the forest floor, and soil were estimated from measurements of NH_4^+ and pH in the leaf and litter tissue and soil pore water (Massad et al., 2010). Green leaves were collected from 18 species (Table S1) within the flux footprint of the tower and other locations in the Coweeta Basin. Leaf litter was collected along transects extending to the northeast and southwest (i.e., predominant wind directions) of the flux tower ~100 m. Litter included a composite of intact leaves and leaf fragments and excluded the more decomposed material at the top of the organic soil layer. Approximately 5 g of leaf tissue was ground in liquid nitrogen using a mortar and pestle and small coffee grinder, then extracted with 20 mL of deionized water. pH was determined directly on the extracts (Oakton pH 2100 meter, Mettler Toledo InLab Micro electrode). The $[NH_4^+]$ in the extracts, which reflects the bulk tissue concentration, was determined by ion chromatography as described above for denuder measurements either directly or, for samples with high organic content, after separation of the NH_4^+ from the solution as NH_3 using headspace equilibration. For the headspace method, 5 mL of tissue extract was added to a 250-mL high-density polyethylene jar containing two ALPHA passive samplers (Center for Ecology and Hydrology; Tang et al., 2001), without the diffusion barrier, affixed to the interior of the lid. The jar was sealed, and 5 mL of 0.3 N NaOH was added to the extract via a septum. The NH_3 liberated from the liquid extract into the headspace was collected by the passive diffusion samplers over a period of 48 h, after which the passive sampler was extracted with 10 mL of deionized water. Extracts were then analyzed by ion chromatography as described above.

Emission potentials of the vegetation (Γ_s) and litter (Γ_l) were estimated from measured concentrations of $[H^+]$ (M) and $[NH_4^+]$ (μg g^{-1} tissue fresh weight) in the bulk tissue as:

$$\Gamma_{s,l} = \frac{[NH_4^+] \times (5.56E^{-5}) \times LD}{[H^+]} \quad (2)$$

where LD is leaf density (kg L^{-1} fresh tissue, equivalent to g cm^{-3} fresh tissue). In this case, LD for woody deciduous and woody evergreen species are 0.37 and 0.42 kg L^{-1}, respectively (Poorter et al., 2009). Emission potentials for litter, which consisted of a mix of in-tact or partial leaves and needles, assume an average density of 0.4 kg L^{-1}. The factor of $5.56E^{-5}$ in equation 2 is necessary to convert $[NH_4^+]$ from μg NH_4^+ g^{-1} tissue to mol NH_4^+ kg $tissue^{-1}$.

Soil chemistry was measured in 20 m x 20 m plots located in the vicinity of the tower. During 2010, soil NH_4^+ was determined on soil samples collected with PVC cores (5-cm diameter and 10-cm deep) in four locations (replicates) within each of four plots. Samples were collected bi-monthly during the growing season. NH_4^+ was extracted within two hours of collection using 5 g of sieved (<5 mm) soil in 20 mL of 2M potassium chloride followed by colorimetric analysis (Astoria2 autoanalyzer, Astoria-Pacific International; Coweeta Hydrologic Laboratory, 2016). Soil pH was measured on 0–10 cm samples collected in three plots in winter of 2013 (each sample representing a composite of 20–25 2.5-cm diameter soil cores). Soil pH was determined directly after mixing 5 g soil with 10 mL 0.01M calcium chloride (Coweeta Hydrologic Laboratory, 2016). Soil emission potential (Γ_{soil}) (unitless) was estimated directly from measured molar concentrations of $[H^+]$ and $[NH_4^+]$ as:

$$\Gamma_g = \frac{[NH_4^+]}{[H^+]} \quad (3)$$

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2.2.5 Above-canopy flux measurements

Above-canopy fluxes of NH_3 and HNO_3 were quantified from measurements of vertical concentration gradients conducted during the final (summer 2016) intensive when the tower was at maximum height and the greatest vertical separation of concentration measurements was achieved. Fluxes were determined using the modified Bowen-ratio (MBR) method (Meyers et al., 1996) as:

$$F = -K_c(z)\frac{dC}{dz}\,, \tag{4}$$

where K_c and dC/dz are the eddy diffusivity and vertical concentration gradient of the chemical species of interest. The value of K_c for trace gases was assumed equivalent to the eddy diffusivity of heat (K_t), calculated as:

$$K_c = K_t = -\overline{w't'}\frac{\Delta z}{\Delta t}\,, \tag{5}$$

where $\overline{w't'}$ is the kinematic surface heat flux measured above the canopy (43.5 m above ground level), Δt is the air temperature difference between the levels at 43 m and 35 m above ground level, and Δz is the height interval of air temperature measurements. Air temperature was measured using aspirated thermocouples, and Δt was corrected for a small bias between the sensors determined by collocated comparison. Concentration gradients were determined from URG annular denuder measurements at 34.6 m and 43 m above ground level, as described above in section 2.2.2. Given the complexity of the topography, no attempt was made to correct for potential roughness sub-layer effects on either the eddy diffusivity or concentration gradients, which should be acknowledged as a source of uncertainty in the calculated HNO_3 and NH_3 fluxes. Additional detail on the gradient measurements is included in Supplemental Section S3.

2.2.6 Seasonal and annual deposition budget

Speciated seasonal and annual total nitrogen deposition budgets were developed for the period August 2015 to August 2016. The wet deposited components, including NH_4^+, NO_3^-, and total WSON, were directly measured. Speciated dry deposition was estimated by combining measured air concentrations, micrometeorology, biogeochemistry and canopy physical characteristics within a box version of the Surface Tiled Aerosol and Gaseous Exchange (STAGE) model, which is an option in the Community Multi-scale Air Quality Model (CMAQ) version 5.3 (Appel et al., 2021). The STAGE model treats the deposition of gases and particles separately. The air-surface exchange of gases is parameterized as a gradient process and is used for both bidirectional exchange and dry deposition following the widely used resistance model of Nemitz et al. (2001) and Massad et al. (2010):

$$F = -f_{veg}\frac{\chi_a(z)-\chi_{z_0}}{R_a} - \left(1-f_{veg}\right)\frac{\chi_a(z)-\chi_g}{R_a+R_g}\,, \tag{6}$$

where F is the net flux above the canopy (a negative value represents a net deposition flux and a positive value represents a net emission flux), $\chi_a(z)$ is the ambient concentration at a reference height (z), χ_{z0} is the concentration at height d (displacement height) + z_0 (roughness length), χ_g is the ground layer compensation point, R_a is the aerodynamic resistance between z and $d + z_0$ is R_a, R_g is the total ground resistance including in-canopy aerodynamic resistance (R_{inc}), ground boundary layer resistance (R_{bg}), and soil resistance (R_{soil}) ($R_g = R_{inc} + R_{bg} + R_{soil}$), and f_{veg} is

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the vegetation coverage fraction. The ground is fully covered by vegetation at our forested site, and f_{veg} is therefore set to 1.

The quantity χ_{z0} is related to the canopy (χ_c) and ground compensation points (χ_g) according to:

$$\chi_{z_0} = \frac{\left(\frac{\chi_a(z)}{R_a} + \frac{\chi_c}{R_{bl}} + \frac{\chi_g}{R_g}\right)}{\left(\frac{1}{R_a} + \frac{1}{R_{bl}} + \frac{1}{R_g}\right)} \quad , \tag{7}$$

where R_{bl} is the leaf boundary layer resistance. χ_c follows Nemitz et al. (2001) but is modified to account for a cuticular compensation point (χ_{cut}):

$$\chi_c = \frac{\chi_a(z)\left(R_a R_{bl}\right)^{-1} + \chi_s\left[\left(R_a R_s\right)^{-1} + \left(R_{bl} R_s\right)^{-1} + \left(R_g R_s\right)^{-1}\right] + \chi_{cut}\left[\left(R_a R_{cut}\right)^{-1} + \left(R_{bl} R_{cut}\right)^{-1} + \left(R_g R_{cut}\right)^{-1}\right] + \chi_g\left(R_{bl} R_g\right)^{-1}}{\left(R_a R_{bl}\right)^{-1} + \left(R_a R_s\right)^{-1} + \left(R_a R_{cut}\right)^{-1} + \left(R_{b1} R_g\right)^{-1} + \left(R_{bl} R_s\right)^{-1} + \left(R_{bl} R_{cut}\right)^{-1} + \left(R_g R_s\right)^{-1} + \left(R_g R_{cut}\right)^{-1}} \quad , \tag{8}$$

where χ_s is the leaf stomatal compensation point, and R_s and R_{cut} are the stomatal and cuticular resistances, respectively. The stomatal, cuticular, and ground compensation points (χ_s, χ_{cut}, χ_g) are described according to Nemitz et al. (2000a) as a function of temperature (T) and the emission potentials (Γ):

$$\chi_{s,cut,g} = \frac{161512}{T} 10^{\frac{-4507.11}{T}} \Gamma_{s,cut,g} \quad . \tag{9}$$

Γ_{cut} is set to 0 in this study, and thus, there is only deposition to leaf cuticles. For unidirectional exchange of gases other than NH_3, Γ_s and Γ_g are also set to 0. In the case of NH_3, the foliage and ground layers may act as a source or sink depending on the ratio of the ambient concentration to the respective compartment compensation point (Husted and Schjoerring, 1995). Here values of Γ for NH_3 are derived from measurements of live vegetation, litter, and soil chemistry as described above. Values used in the base model simulation are described in Section 3.6, and the sensitivity of modeled NH_3 fluxes to Γ is discussed in Supplemental Section S5.

Formulas for each resistance component are summarized in Supplemental Table S2. The resistances are largely estimated following Massad et al. (2010) with the following exceptions. The value of R_s is based on the Noah (Chen and Dudhia, 2001) or P-X land surface schemes (Pleim and Xiu, 1995) in the Weather Research and Forecasting (WRF) model, and in this study, the P-X scheme is used. Deposition to wetted cuticular and ground surfaces considers the bulk accommodation coefficient, following Fahey et al. (2017), and can be a limiting factor for highly soluble compounds. The value of R_{inc} follows Shuttleworth and Wallace (1985) as does Massad et al. (2010), but uses the canopy momentum attenuation parameterization from Yi (2008) and in-canopy eddy diffusivity following Harman and Finnigan (2007). This parameterization is similar to Bash et al. (2010), and detailed descriptions of model resistance can be found in the references mentioned above.

Dry deposition (F) of aerosol nitrogen (NH_4^+ and NO_3^-) is estimated as the product of the measured concentration (C) and the STAGE modeled dry deposition velocity (V_d):

$$F = -V_d(z) \times C(z) \, . \tag{10}$$

Aerosol dry deposition processes include gravitational settling, Brownian diffusion, surface impaction, and rebound. Similar to gases, STAGE calculates the averaged V_d for particles by summing the V_d over vegetative or non-vegetative

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surfaces, weighted by vegetation cover fraction which is = 1 (full coverage) at Coweeta. V_d for a particle with aerodynamic diameter D_p is calculated as:

$$V_d(D_p) = \frac{V_g}{1-exp[-V_g(R_a+R_{bp})]} \quad , \tag{11}$$

where R_{bp} is the boundary layer resistance for particles, and the gravitational settling velocity (V_g) is calculated as:

$$V_g = \frac{\rho D_p^2 g}{18\mu} C_c \, , \tag{12}$$

where ρ is the density of the aerosol, g is the acceleration of gravity, μ is the air dynamic viscosity, and C_c is the Cunningham slip correction factor. The turbulent transport processes are considered similar for gas and aerosol, and R_a can be formulated based on the similarity theory relationships. Unlike deposition of gases, the boundary layer resistance usually dominates the aerosol deposition process as Brownian diffusion is much slower for particles than molecular diffusion is for gases (Pleim and Ran, 2011). Thus, R_{bp} depends on the collection efficiency of the surface and can be determined as:

$$R_{bp} = \left[F_f u_* \left(S_c^{-\frac{2}{3}} + E_{im}\right)\right]^{-1} , \tag{13}$$

where u_* is the friction velocity, and S_c is the Schmidt number for particles. The quantity E_{im} represents the collection efficiency by impaction and follows Slinn (1982) for vegetative canopies and Giorgi (1986) for smooth (non-vegetative) surfaces. The quantity F_f is an empirical correction factor to account for increased deposition in convective conditions, parameterized as:

$$F_f = V_{fac}\left(1 + 0.24\frac{{w_*}^2}{{u_*}^2}\right), \tag{14}$$

where V_{fac} is an empirical constant representing the enhanced effects over vegetation canopies. For vegetative canopies, V_{fac} is equal to the one-sided leaf area index (LAI) with a minimum value of one, and for non-vegetative surface, a value of one is used. The quantity w_* is the convective velocity scale (Deardorff velocity), defined as:

$$w_* = \left(\frac{g}{T_v} z_i \overline{w't'}\right)^{\frac{1}{3}} , \tag{15}$$

where T_v is virtual air temperature, z_i is average depth of the mixed layer, and $\overline{w't'}$ is the measured kinematic surface heat flux.

A bulk V_d for $PM_{2.5}$ is obtained by integrating size-resolved V_d according to the particle size distribution. The size distribution profiles for NH_4^+ and NO_3^- are from measurements at eight Canadian rural forest sites (Zhang et al., 2008) and the size distribution for particulate organic nitrogen is estimated as an average of that for NH_4^+ and NO_3^-. Model sensitivities of particle nitrogen fluxes to assumed size distributions are discussed in Supplemental Section S5.

The STAGE model is extracted from the CMAQ v5.3 and executed in a one-dimensional mode. The prescribed surface parameters (e.g., z_0, d) were modified according to the site conditions. The continuous LAI data were extracted from the MODerate resolution Imaging Spectroradiometer (MODIS) global LAI product (MCD15A2H), which is generated daily at a 500-m spatial resolution, and each data point covers an 8-d period. The MODIS LAI (Supplemental Figure S6) was adjusted using in-situ canopy measurements as described in Supplemental Section S4. Hourly meteorological

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measurements, including air temperature, relative humidity, u_*, atmospheric pressure, precipitation rate, global radiation, and soil temperature/moisture, were used to drive STAGE. The Obukhov length, which is defined as:

$$L = -\frac{u_*^3 T_v}{(kg\overline{w't'})} \quad , \tag{16}$$

where k is the von Karman constant, was also calculated from micrometeorological measurements.

2.2.7 Air concentrations for dry deposition modeling

Air concentration data used for dry deposition modeling are summarized in Table 3. Hourly measurements of HNO_3 by DD_CL and ΣAN and ΣPN by TD-PC-CL were conducted for a full year and were therefore used directly for modeling. Over the 12-month sampling period, 18%, 22%, and 22% of hourly HNO_3, ΣAN, and ΣPN concentrations were missing or invalid, respectively. Missing data were replaced with the corresponding hour from the median diel profile comprised of days with > 75% completeness. In order to convert measured ΣAN and ΣPN concentrations from ppb to μg N m^{-3}, surrogate formulas of nitrooxy-butanol ($C_4H_{10}NO_4$) and PAN ($C_2H_3NO_5$) were assumed for ΣAN and ΣPN, respectively. The CMAQ V5.2.1 output for Coweeta showed that these were the most abundant individual species in their corresponding organonitrate groups.

Continuous NH_3 concentrations were only measured during the last two intensives. Biweekly AMoN NH_3 measurements, with corrections for travel blanks and atmospheric pressure, were used to establish a continuous 12-month time series of air concentration for annual deposition modeling. Ammonia concentrations are known to exhibit pronounced diel patterns, even in remote areas (Wentworth et al., 2016). Variability in air concentration interacts with diel cycles in surface wetness, turbulence, and other factors to influence diel patterns in air-surface exchange rates. To incorporate this interaction, the diel concentration pattern determined during spring and summer 2016 by MARGA NH_3 measurements (Supplemental Figure S7) was imposed on the bi-weekly AMoN NH_3 concentration. The hourly profile of NH_3 concentrations was normalized by the corresponding overall mean concentration to produce a normalized mean diel concentration profile. This profile was then applied to each biweekly AMoN air concentration, temporally scaling the NH_3 concentration by time of day while maintaining the measured biweekly AMoN concentration. Gap filling of AMoN data was not required. Comparisons of NH_3 measurements are briefly discussed in Supplemental Section S1.

Hi-Vol measurements of speciated particulate N were only conducted during intensive periods to assess the relative contributions of inorganic and organic fractions to total water-soluble N. The CASTNET particulate NH_4^+ and NO_3^- were used to provide a continuous 12-month time series of air concentration for annual deposition modeling. Concentrations of Hi-Vol and CASTNET measurements were shown to be comparable (Supplemental Section S1). For the annual time series, particulate organic nitrogen (PON) concentration was estimated based on speciated measurements during intensives, which showed that inorganic N accounts for ~88% of WSTN on average. Weekly average PON concentration was estimated from the weekly CASTNET measurements assuming NH_4^+ + NO_3^- represents 88% of total particulate nitrogen and PON represents the balance (12%). Weekly concentrations were then expressed at the hourly time scale for modeling. Gap filling of CASTNET data was not required.

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Components of the atmospheric reactive N budget that are not routinely measured at Coweeta and were not directly measured during SANDS include NO, NO_2, HONO, and N_2O_5. At the continental scale, regional model simulations suggest that NO, HONO and N_2O_5 make minor contributions to the total dry deposition of reactive N (~ 2%), though the contribution of NO_2 is larger (~ 6%) (Walker et al., 2020). While NO, HONO, and N_2O_5 have been excluded from our modeling analysis, we have included an estimate of NO_2 concentration, from which dry deposition is estimated, based on measured NO_y and CMAQ (V5.2.1) derived estimates of the ratio of NO_2 to NOy at Coweeta. The "other" fraction of NO_y (i.e., $NO_y - HNO_3 - \Sigma PN - \Sigma AN$) measured at Coweeta represents between 47% (summer) and 76% (winter) of total NO_y on a seasonal basis. This "other" fraction includes NO, NO_2, HONO, N_2O_5, and some NO_3^- but is likely dominated by NO_2. The measured diel profile of "other" NO_y (Supplemental Figure S8) concentration shows the typical pattern indicative of morning and evening modes related to mobile NOx emissions. Winds at Coweeta are from the east/northeast during the morning, which is the direction of local residences, the town of Otto, NC, and U.S. Highway 23. Winds are from the west/southwest during the evening, which is the direction of the Nantahala National Forest. Consistent with the diel profile of "other" NOy, a much larger morning peak in NO_2 is therefore expected. To estimate the concentration of NO_2 from the measured "other" NO_y, we examined the ratio of NO_2 to the quantity NOy – HNO_3 – PANS – NTR (e.g., "other" NO_y) simulated by CMAQ (V5.2.1) for the Coweeta site over the year 2015, where PANS represents total peroxy nitrates, and NTR represents other organic nitrates. Relative to the measured NOy species, PANS and NTR are assumed to represent ΣPN and ΣAN, respectively. The ratio of CMAQ estimated NO_2 to "other" NOy ranges from 0.51 during summer to 0.60 during winter. These seasonal factors were applied to the measured "other" NOy to estimate the hourly NO_2 concentration. Gap filling procedures for hourly "other" NOy follow those for HNO_3, ΣPN, and ΣAN described above. Details of CMAQ V5.2.1 can be found in Supplemental Table S3.

3 Results and discussion

3.1 Long-term trends in atmospheric N at Coweeta

Emissions of oxidized nitrogen (NOx) and sulfur (SOx) have declined significantly in the eastern U.S. in response to the 1990 Clean Air Act Amendments (Figure 2, Supplemental Figure S9). Trends data from U.S EPA's National Emissions Inventory (NEI) indicate a nationwide decline of 74% and 46% for SO_x and NOx emissions from the early 1990s to 2010s, respectively, comparing 1990–1994 to 2010–2014 annual averages (U.S. EPA, 2014). By contrast, NH_3 emissions have been reported as relatively unchanged or slightly increasing for the same periods (Ellis et al., 2013; Paulot and Jacob, 2013; Xing et al., 2013), depending on the location and region of the U.S. Declining NOx and SOx emissions have resulted in decreasing trends in air concentrations of HNO_3 and SO_2 at Coweeta between the1990s and 2010s (Figures 2 and S9). Concentrations noticeably began to decline in 2008, the timeline of which likely indicates the effect of full implementation of the 2006 Tier 2 Gasoline Sulfur Program, as well as the enactment of the Clean Air Interstate Rule (CAIR), both of which aimed to further reduce NOx and SOx emissions (Sickles and

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Shadwick, 2015; LaCount et al., 2021). Compared to other species, NH_3 concentrations have only been measured at Coweeta for a relatively short period of time.

Atmospheric NH_3 reacts with acidic sulfate to form ammonium sulfate ($(NH_4)_2SO_4$) or bi-sulfate ($(NH_4)HSO_4$) aerosol. Under favorable thermodynamic conditions (low temperature and high RH), NH_3 in excess of acidic sulfate will react with HNO_3 to form ammonium nitrate aerosol (NH_4NO_3). Concentrations of SO_4^{2-} at Coweeta have tracked SO_2, and subsequently, NH_4^+ concentrations have declined substantially relative to early 1990s levels (Figure 2). However, concentrations of NO_3^- aerosol, which are relatively low at Coweeta, have not followed trends in SO_4^{2-} and NH_4^+ (Figure 2). Previous studies at other U.S. sites have also reported non-proportional changes in $PM_{2.5}$ mass in response to SO_2 and NOx control strategies (Blanchard and Hidy, 2005; Sickles and Shadwick, 2015). Non-linear reductions or increases of particulate NO_3^- with coincident SO_2 and NOx emission reductions relate to the thermodynamic equilibrium of the SO_4^{2-}- NO_3^-- NH_4^+-HNO_3-NH_3 aerosol system. As ambient SO_4^{2-} concentrations decline, the capacity for NH_4^+ formation (i.e., neutralization) also decreases, leaving additional NH_3 in the gas phase. Amounts of NH_3 in excess of acidic SO_4^{2-} can subsequently react with HNO_3 to form NH_4NO_3, confounding the relationship between NOx emission reductions and atmospheric NO_3^- concentrations.

The long-term trend in NO_3^- wet deposition at Coweeta (Figure 3) has tracked the downward trend in ambient HNO_3 concentration. Wet deposition of NH_4^+, however, shows no apparent trend, in contrast to the decline in NH_4^+ aerosol concentration. This pattern may relate to the combined effects of changes in regional NH_3 emissions, aerosol chemistry, and dry deposition rates on the long-term trend in atmospheric NH_3 concentrations. As noted above, declines in SO_2 emissions and SO_4^{2-} aerosol result in less conversion of NH_3 to NH_4^+ aerosol, leaving more NH_3 in the gas phase. Furthermore, reduced air concentrations of acidic species such as SO_2 and HNO_3 result in lower dry deposition rates and subsequently less acidic deposition surfaces, which in turn reduces the deposition velocity (i.e., increases the atmospheric lifetime) of NH_3 (Sutton et al., 2003). In addition to changes in emissions, these two processes are thought to be at least partly responsible for the increases in NH_3 air concentrations that have been observed in some locations across the U.S. (Butler et al., 2015; Yu et al., 2018; Yao and Zhang, 2019). While there is no discernable trend in NH_3 air concentrations over the relatively short period of record at Coweeta, a decline in wet deposition of NH_4^+ aerosol may have been offset to some extent by increased wet deposition of NH_3 gas (Asman et al., 1995), which is highly soluble, resulting in an overall lack of trend in NH_4^+ wet deposition at Coweeta over time. Similar to other areas in the U.S. (Li et al., 2016), the downward trend in NO_3^- wet deposition has led to an increase in the relative contribution of reduced forms of N (i.e., $NH_x = NH_3 + NH_4^+$) to inorganic wet N deposition at Coweeta (NH_4^+:NO_3^-, Figure 3).

3.2 Wet deposition

Of the various forms of N in precipitation, ammonium was the most abundant inorganic species, contributing 47.0% of WSTN in weekly samples (N = 52), on average, followed by NO_3^- (41.7%, Figure 4, Supplemental Table S4). The contribution of NO_2^- was negligible. Organic compounds (WSON) contributed 11% of WSTN, on average, which is within the range of values (3% to 33%) reported for other locations in the U.S. (Scudlark et al., 1998; Whitall and Paerl, 2001; Keene at al., 2002; Beem et al., 2010; Walker et al., 2012; Benedict et al., 2013). While concentrations

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of N compounds were generally higher during warm months, a seasonal pattern in the percent contribution of WSON to WSTN was not apparent. In a previous study at Coweeta (1994–1996), Knoepp et al. (2008) found that organic nitrogen contributed 21% of total nitrogen in bulk (wet + dry) deposition samples. Differences between Knoepp et al. (2008) and SANDS results may be related to interannual variability or trends in rainfall composition over the intervening two decades (e.g., Figure 3) or differences in collection method (wet only versus bulk deposition) or analytical techniques used for total N analysis (persulfate/UV digestion (Walker et al., 2012) versus total Kjeldahl N (Knoepp et al., 2008)).

3.3 Air concentrations of oxidized N

The oxidized fraction of reactive nitrogen (NOy) comprises a mixture of gaseous and particulate inorganic (NO, NO_2, N_2O_5, HONO, HNO_3, NO_3^-) and organic compounds. Owing to its large deposition velocity and typical atmospheric concentration, HNO_3 is the primary contributor to dry deposition of inorganic oxidized N (Walker et al., 2020). Much less is known about the dry deposition of oxidized organic nitrogen compounds (Walker et al., 2020). Peroxy nitrates (PNs) and alkyl and multifunctional nitrates (ANs) are formed during the photochemical oxidation of volatile organic compounds (VOCs) in the presence of NOx (NOx = NO + NO_2). While PNs exist in the gas phase, ANs can exist in the gas or particle phase and can be the dominant chemical sink for NO_x in high biogenic VOC (BVOC)/low NO_x environments (Farmer et al., 2008; Browne and Cohen, 2012; Paulot et al., 2012; Browne et al., 2013). Unlike PNs, ANs can also form at night via nitrate radical-induced oxidation of VOC. Further, PNs and ANs have been shown to contribute significantly to the total NO_y budget in geographically diverse rural and forested environments (e.g., Trainer et al., 1993; Nouaime et al., 1998; Farmer et al., 2008; Browne et al., 2013; Toma et al., 2019). Flux measurements at Blodgett Forest, CA, showed that PN dry deposition contributed 4–19% of total N deposition at the site (Wolfe et al., 2009). Chemical transport modeling with current representation of the atmospheric oxidized nitrogen system suggests that PNs and ANs together contribute ~ 6% of total N deposition and ~ 12% of dry N deposition at the U.S. continental scale, compared to ~ 21% and 34% for HNO_3 and ~ 6% and 9% for particulate NO_3^- (Walker et al., 2020).

The annual average concentration of NOy was 1.00 ppb, with the highest seasonal average concentration in the winter (1.32 ppb) and lowest in the summer (0.64 ppb) (Figure 5, Table S5). The nearest rural NOy monitoring site is 85 km to the northwest at Look Rock in the Great Smoky Mountains National Park, where the annual concentration was 1.5 ppb over the same period (NPS, 2020). Similar to Coweeta, NOy concentrations at Look Rock are typically lowest during summer and highest in winter, though the seasonal cycle exhibits some interannual variability. The annual mean concentrations of HNO_3, ΣPN, ΣAN determined by TD-PC-CL were 0.14, 0.1, and 0.09 ppb, respectively (Figure 5). HNO_3 and ΣPN concentrations peaked in spring, coincident with the seasonal peak in O_3 concentration, while concentrations of ΣAN were similar in spring and summer. Diel patterns of HNO_3, ΣPN, and ΣAN peaked during the day as expected for photochemical products. However, of the organic compounds, the ratio of peak daytime to minimum nighttime concentrations (Figure 5) was much smaller for ΣAN (2.3) compared to ΣPN (3.9), possibly indicative of the additional nighttime formation of AN.

Annually, HNO_3 (12.8%), ΣPN (12.2%), and ΣAN (12.7%) contributed approximately the same proportions of the NOy budget (Figure 5, Table S6). Their collective contribution (NO_z = HNO_3 + ΣPN + ΣAN) to total NOy peaked

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during the summer (52.9%) and reached a minimum during winter (24.2%). The contributions of ΣPN (16.7%) and ΣAN (20.0%) exceeded HNO_3 (16.2%) during summer when total NOy concentrations were lowest. Our results fall within the range of NOy budgets reported for other rural forested sites, in which ΣPN and ΣAN contribute ~ 8–40% (Nouaime et al., 1998; Farmer et al., 2008; Browne et al., 2013; Toma et al., 2019) and 10–22% (Day et al., 2003; Farmer et al., 2008; Browne et al., 2013) of NOy, respectively.

To put the SANDS period into context with longer-term variability of oxidized N concentrations at Coweeta, CASTNET HNO_3, and NO_3^- measurements for the period 2015–2020 are summarized in Figure 6 along with the SANDS period. We note here that NH_4NO_3 volatility on the CASTNET Teflon filter can result in positive and negative biases in HNO_3 and NO_3^-, respectively, with larger biases expected under warmer conditions (Lavery et al., 2009). Studies have shown the total NO_3^- (TNO_3 = HNO_3 and NO_3^-) to be conserved, though some portion of the NO_3^- collected by the CASTNET open-faced filter may be contributed by coarse particles. The partitioning of TNO_3 between gas and particulate phases is important, given the much larger deposition velocity of HNO_3 than NO_3^-. The CASTNET measurements reflect relatively low concentrations of both HNO_3 and NO_3^-, with HNO_3 exceeding NO_3^- during all seasons. Particulate NO_3^- concentrations are highest during cooler months, as expected, and negligible during the summer, a pattern that is consistent with observations from other networks across the Southeast (Kim et al., 2015). Additionally, TNO_3 is primarily in the gas phase even during winter. Seasonal mean concentrations during the SANDS period fall within the interquartile range (IQR) of the 6-year period between 2015 and 2020, with SANDS annual and 6-year averages being very similar (Figure 6). Seasonal and annual mean HNO_3 concentrations agreed closely with the CASTNET measurements (Figures 6 and Supplemental Section S1).

3.4 Air concentrations of reduced N

Reduced forms of nitrogen (NH_x) represent another important component of the inorganic dry N deposition budget. At the continental scale, NH_3 dry deposition contributes ~ 20% of total N deposition and ~ 32% of dry N deposition, whereas the contributions from NH_4^+ aerosol are ~ 4% and ~ 6%, respectively (Walker et al., 2020). Similar to oxidized forms of N, the partitioning of mass between the gas (NH_3) and particulate (NH_4^+) phases affects the dry deposition rate of NH_x, given the larger deposition velocity of NH_3 relative to NH_4^+.

During 2015–2020, with 2020 being the most recent full year of AMoN data, concentrations of NH_3 and NH_4^+ were similar, though with slightly more mass in the particulate phase (53.8%, Figure 7). Both species displayed a seasonal pattern of lowest concentrations in the winter and higher concentrations during warm months. NH_3 concentrations peaked in summer, reflecting the temperature dependence of regional agricultural and biogenic emissions. NH_4^+ concentrations followed the seasonal cycle in SO_4^{2-} concentrations, which were also similar in spring and summer and minimum in winter at Coweeta. The seasonal cycle of NH_3/NH_4^+ partitioning was driven more by NH_3 than NH_4^+, the former exhibiting more seasonal variability. Hourly measurements conducted during spring and summer 2016 showed that NH_3 also displayed significant diel variability (Supplemental Figure S7), reaching a maximum around mid-day and minimum overnight. Seasonal mean concentrations during the SANDS period fall within the IQR of the 6-year period between 2015 and 2020, with SANDS annual and 6-year averages being very similar. Concentrations of NH_x were higher relative to TNO_3 during SANDS and over the longer term at Coweeta (Figures 6 and 7).

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3.5 Aerosol N composition

Ammonium was the most abundant inorganic species, contributing 86.8% of WSTN (N = 103) on average (Figure 8, Table S7). The contributions of NO_3^- and NO_2^- were negligible. Organic compounds (WSON) contributed 11.6% of WSTN, which is very similar to precipitation. Our study-wide average of %WSON is slightly lower than measurements at other North American forest sites, including Duke Forest, North Carolina (~33%, Lin et al., 2010) and Rocky Mountain National Park (14–21%) (Benedict et al., 2012), but is within the global range of 10–39% (Cape et al., 2011). Similar to precipitation chemistry, there was no seasonal pattern in the percent contribution of WSON to WSTN in $PM_{2.5}$. Hi-Vol measurements of inorganic PM components compared well, overall, with collocated CASTNET measurements (Supplemental Section S1).

3.6 Biogeochemistry

Estimates of NH_3 emission potentials (Γ) for the ground and vegetation are needed to calculate compensation points (χ) and fluxes in STAGE. Measurements of pH, NH_4^+ and corresponding Γ of the leaves (Γ_s) and litter (Γ_l) are summarized in Figure 9 and Supplemental Tables S8 and S9. Measurements of Γ_s are divided into green leaves collected during the growing season (spring and summer) and senescent leaves collected in October. NH_3 emission potentials (Γ) for green leaves (Γ_s) ranged from zero to 4070 with a median value (35.8) (Table S8) corresponding to a compensation point of χ_s = 0.25 μg NH_3 m^{-3} at 25 °C. Large intra-species variability of tissue pH and NH_4^+ were observed (Table S9) and separating by crown versus understory species did not reveal distinct differences between groups. Given the variability of the observations, the median Γ_s was used for STAGE simulations. Senescence marks the translocation of N in leaves to storage tissues (Schneider et al., 1996). Along with a decline in photosynthetic activity, degradation of chlorophyll, and other metabolic changes, glutamine synthetase (GS) activity also declines (Pearson et al., 2002). Glutamine synthetase catalyzes assimilation of NH_4^+ into glutamine and is therefore important in regulating the pool of NH_4^+ available for exchange as NH_3 between the leaf and atmosphere and remobilizing organic N for storage during senescence. A decline in GS activity can thus result in increased leaf NH_4^+ concentrations (Pearson et al., 2002; Wang et al., 2011). Senescent leaves were similar to green leaves with respect to median tissue pH but showed distinctly higher concentrations of tissue NH_4^+. Median Γ_s was correspondingly higher (113), equivalent to χ_s = 0.8 μg NH_3 m^{-3} at 25 °C. For STAGE modeling, the median Γ for senescent leaves was used for Γ_s during the fall.

Leaf litter on the soil surface has been shown to be a source of NH_3 to the atmosphere in both natural and agricultural ecosystems (Nemitz et al., 2000b; David et al., 2009; Hansen et al., 2013). As litter decomposes, mineralization of organic N is a source of NH_4^+, some of which is lost to the overlying air as NH_3. Litter NH_4^+ concentrations were similar to green leaves but lower than senescent leaves (Figure 9). However, the pH was higher than both green and senescent leaves. The resulting median Γ_l (69.3) was larger than green leaves but smaller than senescent leaves, equivalent to χ_l = 0.49 μg NH_3 m^{-3} at 25 °C. Litter Γ was much larger than that of the underlying soil. Average (0–10 cm soil depth) soil pH (4.18) and NH_4^+ (1.21 mg N kg^{-1}) correspond to Γ_{soil} = 10 at a soil mass wetness of 0.1 g g^{-1}, equivalent to χ_{soil} = 0.07 μg NH_3 m^{-3} at 25 °C. This very low Γ_{soil} results from the low pH of the shallow soil.

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Vertical profiles of air concentrations within and above the canopy were measured to investigate patterns of air-surface exchange with specific ecosystem compartments (i.e., canopy crown, understory, and ground). A detailed analysis of bi-directional N fluxes is forthcoming; thus, we limit the discussion of these data to NH_3 in the context of interpreting patterns observed in the biogeochemical emission potentials and their prescription in the STAGE model. Nitric acid, NH_4^+, and NO_3^- showed expected decreasing concentrations from above the canopy to the forest floor, indicative of deposition. While NH_3 profiles showed patterns of deposition to the crown and understory, concentrations near the forest floor indicated both emissions and deposition (Figure 10). Of the 76 daytime profiles measured, 40% showed decreases toward the forest floor, and 60% showed increasing concentration from approximately the lower understory (~ 5 m above ground) to the forest floor. The former pattern is interpreted as deposition to the forest floor, and the latter is interpreted as emission. Thus, the mean profile suggests a source of NH_3 at the ground. The very low Γ_{soil} suggests that emission from the soil is unlikely given such a low pH. The leaf litter layer, which indicates a much higher emission potential (Γ_l) than the soil, is a more likely source of NH_3. This hypothesis is consistent with Hansen et al. (2013; 2017), in which emissions of NH_3 from a beech (*Fagus sylvatica*) forest after leaf fall were attributed to the decomposition of new litter. Similar to our site, the underlying soil also had low pH (4–5). Given our observations, we used Γ_l (median = 69.3, Table S8) rather than Γ_{soil} as the ground emission potential (Γ_g) in STAGE.

3.7 N deposition budget

Total annual N deposition for the period August 2015 – August 2016 was 6.6 kg N ha^{-1} (Figure 11). Over this period, wet deposition contributed 61.4% of total N deposition, of which NH_4^+ was the primary component (29.9%). Wet deposition of organic N contributed 5.5% of the total N deposition budget. Dry deposition accounted for 38.7% of total deposition, of which NH_3 was the primary contributor (19.7%). Reduced forms of inorganic N were the largest contributor to the budget (50.7%), with oxidized inorganic and organic N contributing 41.6% and 7.7% of total N deposition, respectively. Dry deposition of organic N made a small contribution (2.2%) to the total deposition budget. Ammonia is the most important contributor to the dry deposition budget (51%) and differs from the other species in that it is exchanged bidirectionally between the ground, canopy and atmosphere. Seasonal net canopy-scale and component fluxes are shown in Supplemental Figure S10. The mean net flux (F_{net}) is downward (i.e., deposition) during all seasons, generally following the seasonal pattern of the atmospheric NH_3 concentration. The cuticular flux (F_{cut}), which is unidirectional in STAGE, is the dominant deposition pathway, and ranged from -89.0 ng N m^{-2} s^{-1} (deposition) to near zero, and accounted for 72%, 81%, 96%, and 98% of the net flux in winter, spring, summer and fall, respectively. The stomatal flux (F_s) is bidirectional, ranging from -4.2 ng N m^{-2} s^{-1} (deposition) to 2.4 ng N m^{-2} s^{-1} (emission), with the largest fluxes occurring during warmer periods of the growing season when the stomatal resistance is lowest. Low LAI and large stomatal resistance in winter and fall and offsetting bidirectional fluxes in spring and summer result in a relatively small mean stomatal deposition flux (F_s) across seasons. On an annual scale, the ground flux (F_g) makes a larger contribution (7.2%) than F_s (2.4%) to F_{net}. F_g is also bidirectional, ranging from -4.3 ng N m^{-2} s^{-1} (deposition) to 0.4 ng N m^{-2} s^{-1} (emission). Fluxes are largest during spring as atmospheric NH_3 begins to increase with warmer temperatures but before peak LAI is reached, after which the denser canopy increases

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the in-canopy aerodynamic (R_{inc}) and air-side ground boundary-layer resistances (R_{bg}) (Table S2), thereby decreasing F_g.

Nitric acid was the second largest component of dry deposition, contributing 37.1% of the total. While HNO_3 deposits more rapidly than NH_3 (Supplemental Table S10), the overall importance to the dry N budget is constrained by relatively low air concentrations at this remote forest site (< 0.2 ppb on average). Particulate species made much smaller contributions to the budget due to much lower deposition velocities (V_d = flux/air concentration) relative to their gaseous counterparts (Supplemental Table S10). For example, while NH_4^+ contributed more to the NH_x concentration budget than NH_3, (Figure 7), the NH_x flux budget was regulated by the much more rapid exchange of NH_3 between the forest and atmosphere relative to NH_4^+. A similar example was observed for oxidized N. While NO_2 represents an important fraction of the oxidized N concentration budget via its contribution to "Other NOy", NO_2 deposits much less rapidly than HNO_3 (Supplemental Table S10) thereby contributing a relatively small fraction (3.1%) of the dry N flux. Of the organic N species, AN contributed slightly more (3.0%) to dry N deposition than PN (2.3%) owing to a higher deposition velocity (Supplemental Table S10). Similar to particulate NH_4^+ and NO_3^-, PON made a small contribution to dry N deposition (0.4%) due to its low V_d (Supplemental Table S10). Reduced forms of N accounted for the majority of dry N deposition (53.9%), with oxidized inorganic and organic forms of N contributing 40.5% and 5.7%, respectively.

Total N deposition peaked during the summer (2.5 kg N ha^{-1}) and reached a minimum in the fall (1.0 kg N ha^{-1}) (Figure 12). Wet deposition exceeded dry deposition during all seasons. Seasonal variability in wet deposition was primarily driven by precipitation amount, whereas dry deposition was influenced by seasonality in air concentrations of the primary N_r species (Figures 6 and 7) and leaf area index (Figure S6). Ammonia fluxes followed the seasonal pattern of air concentration, peaking in the summer and reaching a minimum in winter. Concentrations and fluxes of HNO_3 peaked in the spring and reached a minimum in the fall. Deposition velocities, which can be thought of as the concentration-normalized flux, peaked during the summer and reached a minimum during winter for most N species. This pattern largely reflects the seasonal cycle in leaf area index, i.e., the total surface area of the forest canopy available for dry deposition. The seasonal pattern of V_d for HNO_3 differed slightly from the other species, peaking in spring and reaching a minimum in fall. In contrast to other N species, HNO_3 deposition is limited by turbulent transfer, the canopy (surface) resistance being zero. The pattern of HNO_3 V_d thus partially reflects seasonal patterns in wind speed and degree of turbulent mixing above the canopy.

3.8 Evaluation of the dry deposition model

While total uncertainty in the dry deposition budget cannot be rigorously quantified (Walker et al., 2019a), the sensitivity of the model to parameterizations and key inputs can elucidate important aspects of model uncertainty and inform a potential range of dry deposition estimates. Here we undertake such an exercise, evaluating several alternative modeling scenarios to assess the sensitivity of fluxes and total dry deposition to assumptions regarding LAI, NH_3 emission potentials ($\Gamma_{s,l}$), NH_3 cuticular resistance ($R_{cut,dry}$), and particle size distribution. We focus on NH_3, as it is the most important component of the dry deposition budget and more complex with regard to air-surface exchange processes than the other species. Sensitivity tests are summarized in Supplemental Section S5 and Table S11. Of the

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scenarios tested, increasing Γ_l and Γ_s within the range of observations and reducing $R_{cut,dry}$ within the variability reported by Massad et al. (2010) exerted the largest control over the dry deposition flux, establishing a range of total dry deposition from 2.1 (increasing $\Gamma_{s,l}$) to 3.0 (decreasing $R_{cut,dry}$) kg N ha^{-1} around the base value of 2.6 kg N ha^{-1}. The corresponding % contribution of NH_3 to total dry N deposition ranges from 40.6% to 57.1% (base = 51%) and the contribution of dry to total wet + dry deposition ranges from 34.4% to 42.1% (base = 38.7%). Our results point to the need for a better understanding of the processes of cuticular exchange and the importance of adequately characterizing the magnitude and variability of vegetation and litter emission potentials in forests.

Another method of evaluating model behavior is the comparison with measured V_d. During the final summer intensive, a small dataset (N = 19 observations) of V_d was determined from daytime measurements of vertical concentration gradients above the canopy using the MBR method. Measured V_d was compared to V_d derived from the STAGE model for overlapping periods, and the maximum possible V_{dmax} as $1/(R_a + R_b)$. Of the 19 MBR measurements, four NH_3 profiles exhibited emissions (6.8 to 22.4 ng NH_3 m^2 s^{-1}), which were not reproduced by STAGE. Analysis of the meteorological conditions during the MBR measurements suggests that emissions tend to occur during the warmest periods with lowest relative humidity. This would correspond to periods when R_{cut} and X_s are high and may indicate that the model is underestimating F_s emissions during these periods. Excluding the four emission periods, V_d estimated from MBR and STAGE agree reasonably well (Supplemental Figure S12). As is the case for STAGE, resistance-based models typically assume HNO_3 deposits at the rate of V_{dmax} (i.e., R_c = 0). As shown in Figure S12, fluxes measured during summer 2016 showed MBR V_d for HNO_3 larger than NH_3, as expected, but lower than V_{dmax}. This apparent non-zero R_c could result from a real non-zero R_c caused, for example, by equilibrium of HNO_3 and NO_3^- on leaf surfaces (Nemitz et al., 2004a). This pattern may also reflect the influence of flux divergence related to NH_4NO_3 evaporation in the canopy crown, which would reduce the magnitude of the downward vertical gradients, and therefore the measured V_d, of HNO_3 and NH_3 (Nemitz et al., 2004b). In this study, concentrations of NO_3^- (mean = 0.08 μg m^{-3}) were much lower than HNO_3 (mean = 0.47 μg m^{-3}), and NO_3^- gradients were therefore difficult to resolve, precluding a definitive explanation of HNO_3 $V_d < V_{dmax}$. Ignoring potentially significant uncertainties related to the measurement of chemical and temperature gradients within the roughness sublayer, our results suggest that periods of NH_3 emission during the day, particularly at higher air temperature and lower humidity, may be underestimated. Our results also reinforce the need for temporally extensive measurements of concentrations and fluxes of HNO_3, NH_3, and NO_3^- to examine exchange processes and uncertainties related to chemical flux divergence.

3.9 Spatial and temporal representativeness of deposition budget

The complexity of atmospheric flows in mountainous terrain influences the spatial variability of wet and dry deposition processes (Lehner and Rotach, 2018). As the deposition budget presented above is based on measurements from the lowest elevation portion of the Coweeta basin, the degree to which the budget is spatially representative must be consisdered. Potential effects on dry deposition were assessed by characterizing the magnitude and spatial variability of HNO_3 and NH_3 concentrations along an elevation gradient (Figure 1, Table 1) from the lower to upper portions of the Coweeta Basin. It should be noted that HNO_3 concentrations at NC25 were measured by CASTNET while HNO_3 passive samplers were used at the other locations. Concentrations are summarized in Figure 13, in which

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the sites are ordered left to right from lowest to highest elevation. Analysis of variance (ANOVA) rank analysis indicated that the differences in NH_3 across sites are statistically insignificant (p = 0.231). For HNO_3, site differences were statistically significant (p = 0.008) but primarily due to higher concentrations at a single site, Screwdriver Knob (SK). Screwdriver Knob is distinct from the other sites in that the measurement tower was situated on a relatively narrow exposed ridge. The measurements are therefore higher above the surrounding vegetation than at the other sites. With SK removed, differences among the other sites are statistically insignificant (p = 0.242). Overall, variability across sites, even including SK, is sufficiently small such that spatial variability of dry deposition across the basin would likely be driven more by variability in meteorology than air concentrations.

A quantitative assessment of the effects of air flow on dry deposition across the basin is not possible, but the analysis of Hicks (2008) illustrates the relevant effects in the context of the resistance analogy for V_d. Over flat homogeneous terrain, flux to the vegetation is driven by turbulent diffusion in the vertical direction above the canopy and horizontal advection is assumed to be zero. In the extreme case of air flow approaching a steep forested slope, flux to the vegetation is driven by horizontal advection into the canopy, and the aerodynamic resistance (R_a) becomes zero. Taking HNO_3 as an example under the typical assumption that the canopy resistance (R_c) = 0, V_d becomes limited by the quasi-laminar boundary layer resistance at the vegetation surfaces (R_b). For the less extreme case of a uniformly vegetated gentle hill, V_d for HNO_3 could be enhanced by a factor of $[1+(R_a/R_b)]^{1/2}$ (Hicks, 2008). Using median values of R_a and R_b from our modeling period, this would increase V_d for HNO_3 by a factor of ~ 1.3. For gases that have a significant R_c, enhancements will be smaller. Topographical relief across the Coweeta Basin may be gentle enough such that the flow separation described in the previous example is limited to certain areas and meteorological scenarios. However, as Hicks (2008) points out, flow complexity in mountainous areas has the overall effect of increasing V_d, with areal weighted fluxes being highly dependent on the topographical characteristics specific to the study area. Other effects related to katabatic flows (Novick et al., 2016) and diel patterns of hillside shading that drive temperature-related processes such as NH_3 compensation points introduce additional uncertainties.

The results of Knoepp et al. (2008) show that spatial patterns of wet deposition across the Coweeta Basin follow patterns of precipitation amount, which increase with elevation. In their study, bulk deposition of NH_4^+, NO_3^- and total organic nitrogen was measured from 1994–1996 at sites ranging in elevation from 788 to 1389 m. Annual precipitation depth and bulk deposition increased by 25% from the lowest to the highest elevation. This increase in precipitation with elevation is consistent with the 75-year analysis of Coweeta climatological data by Laseter et al. (2012), which showed annual precipitation amount at 1398 m was 32% greater than at 686 m. In our study, wet deposition was measured at the NC25 site at 686 m and therefore represents a lower wet deposition rate than would occur in higher elevation portions of the basin. An approximate 30% enhancement in both wet and dry (extreme case) deposition for the highest elevations within the basin would correspond to a total N deposition rate of 8.6 kg N ha^{-1} yr^{-1} based on our estimate of 6.6 kg N ha^{-1} yr^{-1} for the lower basin.

Regarding the temporal representativeness of the deposition budget calculated here, wet deposition of inorganic N (NO_3^- + NH_4^+) during our 12-month model period (3.69 kg N ha^{-1}) agrees well with the mean annual deposition rate measured at NTN site NC25 (3.72 kg N ha^{-1}) over the period 2015–2020, with 2020 being the most recent full year of reported observations. Air concentrations of NO_3^- and HNO_3 (Figure 6) as well as NH_3 and NH_4^+ (Figure 7) during

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our model period are also similar to the 6-year (2015–2020) mean concentrations measured by CASTNET and AMoN. In this context, our results are deemed temporally representative of the most recently available complete years of monitoring data.

4 Conclusions

Due to the success of the Clean Air Act, air concentrations and wet deposition of reactive N at Coweeta are the lowest observed since the beginning of routine monitoring in the late 1970s. However, even at historically low levels, our results show that N_r deposition remains highly ecologically relevant in the context of critical loads. Our estimate of total N_r deposition of 6.6 kg N ha^{-1} yr^{-1} is near the upper-end estimate of mass balance derived critical loads (2.8 to 7 kg N ha^{-1} yr^{-1}) recently reported for spruce-fir, beech, and mixed deciduous forests by Pardo et al. (2018) in nearby Great Smoky Mountains National Park. Our result also falls within the range of empirical critical loads of N for combined tree health and biogeochemical responses (3–8 kg N ha^{-1} $yr^{-1)}$ and changes in mycorrhizal fungi spore abundance, community structure and community composition (5–12 kg N ha^{-1} yr^{-1}) in eastern temperate forests (Pardo et al., 2011).

A key feature of the deposition budget derived for Coweeta is the predominance of reduced forms (NH_x) of inorganic nitrogen (50.1%) over oxidized inorganic N (41.6%). Reductions in deposition of NH_x will be needed to achieve the lower-end estimates of critical N loads (~ 3 kg N ha^{-1} $yr^{1)}$ for southern Appalachian forests. This presents a challenge, as emissions and air concentrations of NH_3 remain unregulated. Our results also show that organic forms of N make a non-trivial contribution (7.7%) to total N deposition, primarily via wet deposition. It is noted, however, that the gas-phase dry component of deposition only considers oxidized forms as alkyl and peroxy nitrates, excluding contributions from reduced (i.e., NH) organic compounds. While our results represent an advancement in accounting for organic dry N_r deposition in total N_r deposition, the application of new measurement technologies (Walker et al., 2019b) for broader chemical speciation of organic forms of dry N_r deposition is needed.

Our results underscore the need for long-term measurements of reactive chemical fluxes, and the coupling of atmospheric and biogeochemical measurements, to improve air-surface exchange models. Novel measurements that more directly elucidate the role of cuticular exchange of NH_3 and more temporally extensive measurements of leaf NH_3 emission potentials are particularly needed. For forest ecosystems, a physically representative parameterization for resistance to NH_3 diffusion through the leaf litter layer and more temporally extensive measurements of the litter NH_3 emission potential are also needed. Such long-term datasets are also required to assess the interactive effects of changing air quality and climate on both atmosphere-biosphere exchange and ecosystem response to deposition (e.g., Van Houtven et al., 2019). For sensitive ecosystems located in mountainous and other topographically complex landscapes, which includes much of the Class I wilderness area in the U.S., identification of locations suitable for micrometeorological flux measurements will be challenging. Novel flux measurement methods and application of in-situ models, including translation of measurements from more ideal to complex locations, will likely be needed.

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5. Ryan Daly: Investigation, formal analysis, validation
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Disclaimer

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Tables

Table 1. Sampling locations and type of sampler deployed.

Site code	Latitude(N)	Longitude(W)	Elevation(m)	Sampler type
NC25/COW137	35.0605	83.4305	686	AMoN (NC25), CASTNET (COW137), Tisch $PM_{2.5}$*, DD-CL, TD-PC-CL, NTN (NC25), EPA precipitation
EFT[a]	35.0591	83.4274	690	MARGA*, URG*, Passive NH_3 and HNO_3, micrometeorology
WS18	35.0512	83.4337	806	Passive NH_3 and HNO_3
SK[b]	35.0482	83.4542	986	Passive NH_3 and HNO_3, CASTNET (COW005)
CS28	35.0466	83.4650	1189	Passive NH_3 and HNO_3
CS77	35.0303	83.4604	1425	Passive NH_3 and HNO_3

[a]-Eddy flux tower; [b]-Screwdriver Knob; * Tisch $PM_{2.5}$, MARGA, and URG denuder/filter pack samplers were deployed only during intensive sampling periods.

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Table 2. Details of intensive and long-term atmospheric measurements at Coweeta.

Sampler name	Operating periods	Measured species	Resolution	Height (m)*
DD-CL, TD-PCCL	August 2015-August 2016	HNO_3, NOy, ΣAN[1], ΣPN[2]	Hourly	8
MARGA	Spring, summer 2016 intensives	HNO_3, NH_3, NO_3^-, SO_4^{2-}, NH_4^+	Hourly	~40
URG denuder/filter	All 5 intensives 2015-2016	HNO_3, NH_3, NO_2^-, NO_3^-, SO_4^{2-}, NH_4^+	3 or 4-hour integrated	~40
Tisch $PM_{2.5}$	All 5 intensives 2015-2016	NO_2^-, NO_3^-, SO_4^{2-}, NH_4^+, WSTN	24hr integrated	~1
CASTNET (COW137)	Long-term	HNO_3, NO_3^-, SO_4^{2-}, NH_4^+, Cl^-, base cations	Weekly integrated	10
AMoN (NC25)	Long-term	NH_3	Bi-weekly passive	2
Passive HNO_3, NH_3	2015	HNO_3, NH_3	Bi-weekly	10 (~40 on EFT)
CASTNET (COW005)	2015	HNO_3, NO_3^-, SO_4^{2-}, NH_4^+, Cl^-, base cations	Weekly integrated	10
NADP/NTN	Long-term	NO_3^-, NH_4^+, SO_4^{2-}, Cl^-, H^+, base cations	Weekly accumulated	Ground
EPA precipitation	February 2015-August 2016	NO_2^-, NO_3^-, SO_4^{2-}, NH_4^+, WSTN	Weekly accumulated	Ground

*Above ground; [1]Total alkylnitrates; [2]Total peroxynitrates

Table 3. Summary of air concentration data sources for STAGE dry deposition modeling.

Chemical Species	Data Source	Details
NH_3	Measurement	AMoN measurement with diel profile imposed
HNO_3	Measurement	Continuous DD-CL
ΣPN	Measurement	Continuous TD-PC-CL. Assume molecular weight (MW = 121.05) of PAN ($C_2H_3NO_5$)
ΣAN	Measurement	Continuous TD-PC-CL. Assume molecular weight (MW = 136.13) of nitrooxy-butanol ($C_4H_{10}NO_4$)
NH_4^+	Measurement	CASTNET
NO_3^-	Measurement	CASTNET
PON	Estimated based on measured NH_4^+ + NO_3^+	Based on intensive direct measurements, assume PON represents 12% of total PON + NH_4^+ + NO_3^+
NO_2	Estimated based on measured NO_y	Based on ratio of NO_2/NO_y simulated by CMAQ V5.2.1 at Coweeta

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Figures

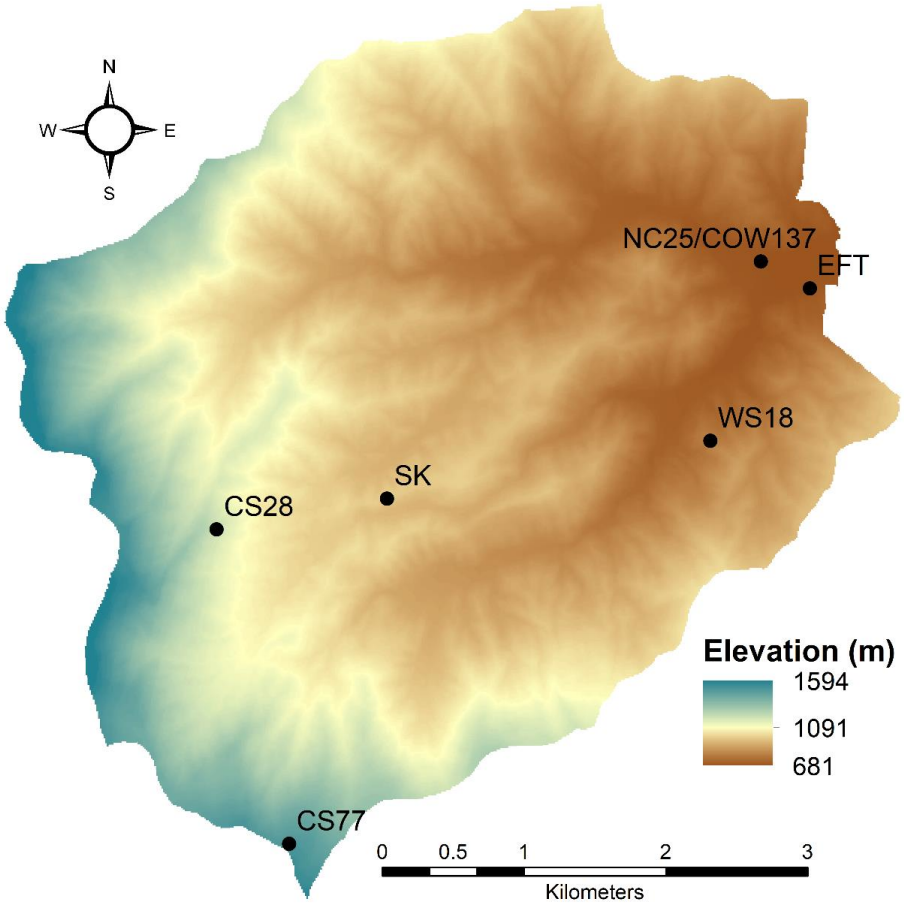

Figure 1. Elevation map of Coweeta Basin with sampling sites in Table 1 indicated.

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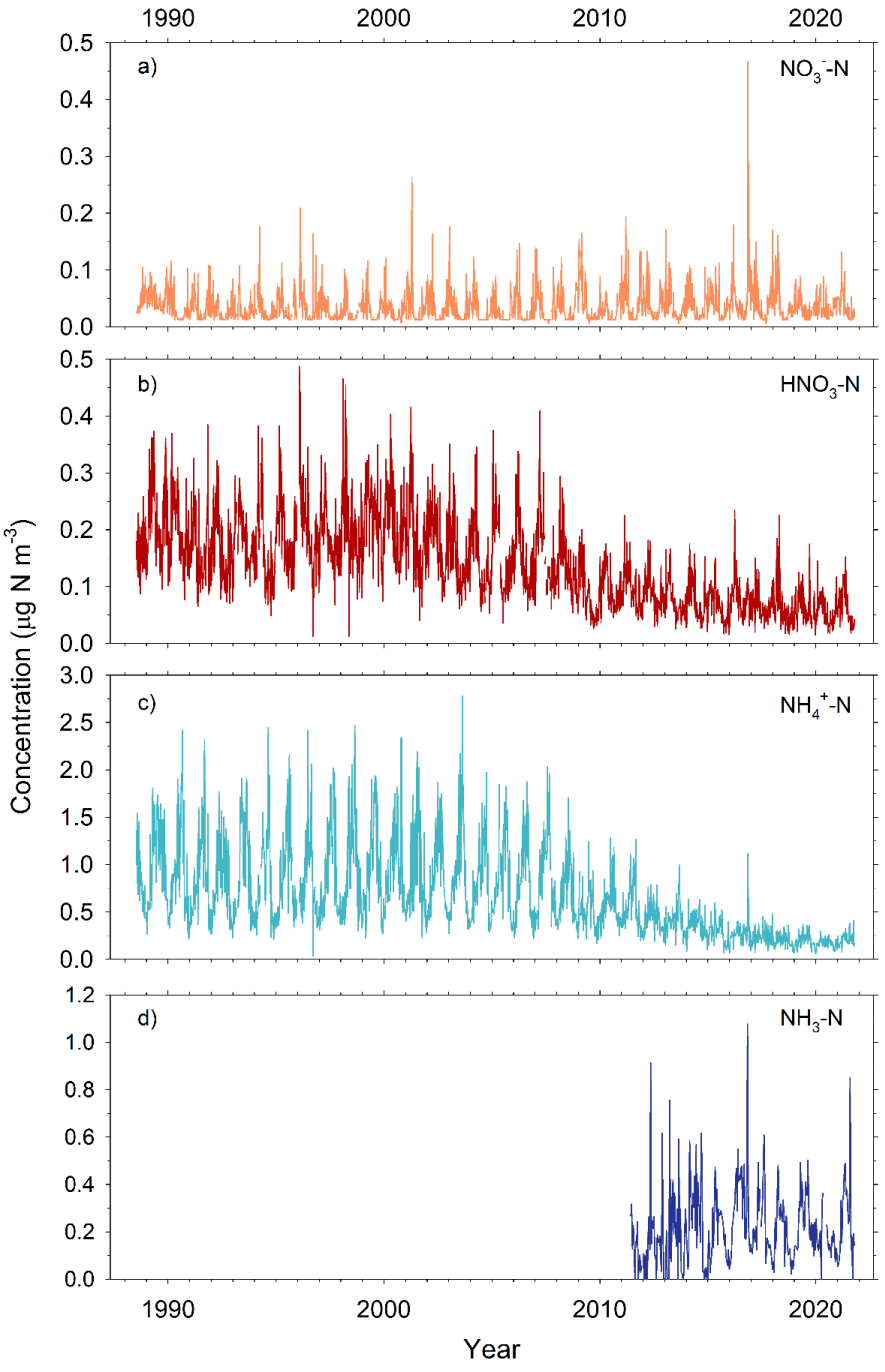

Figure 2. Long-term CASTNET (weekly, Site COW137) and AMoN (biweekly, Site NC25) air concentrations (as N) of NO_3^- (a), HNO_3 (b), NH_4^+ (c) and NH_3 (d).

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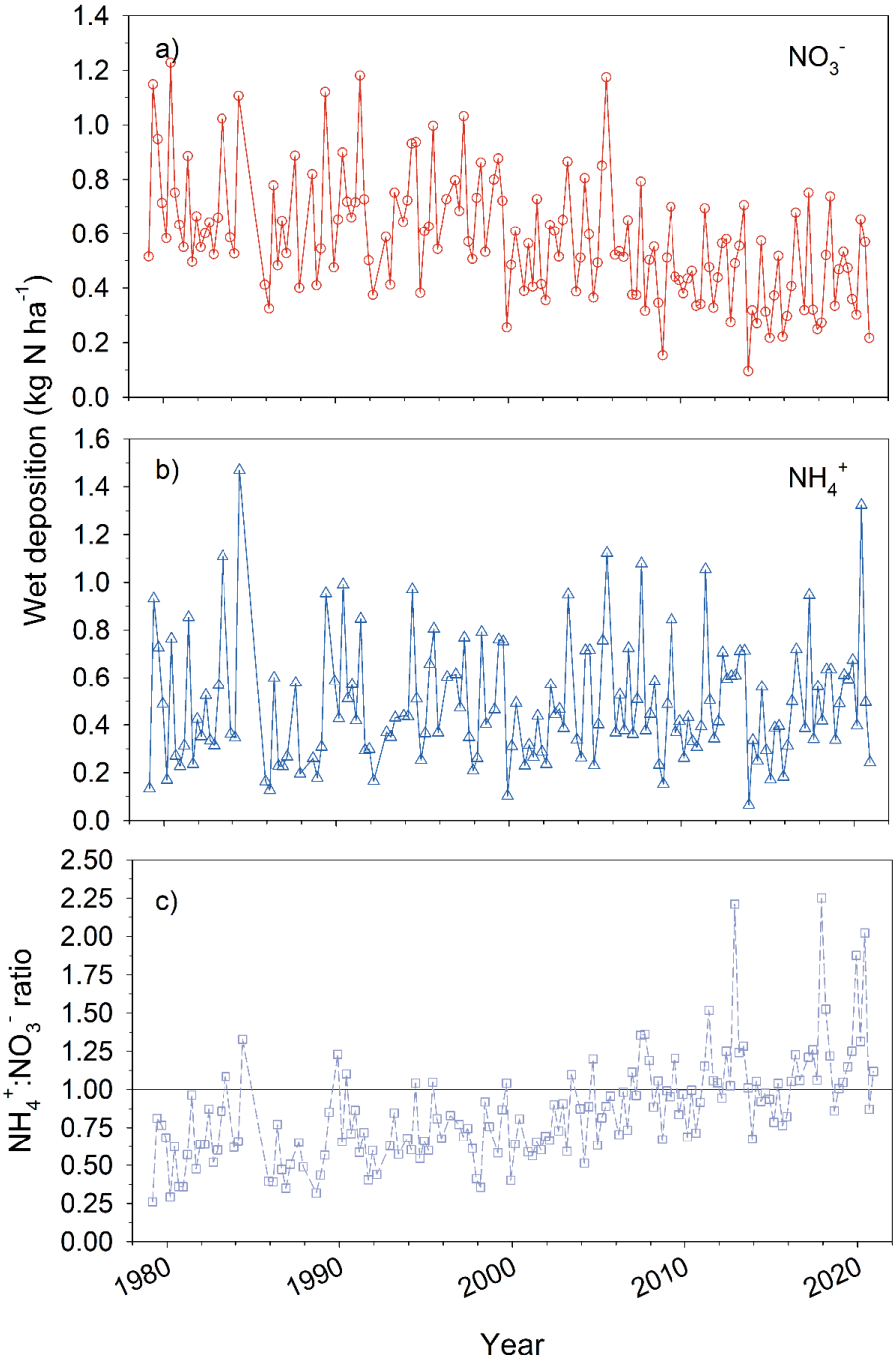

Figure 3. Long-term NTN NC25 measurements of seasonal NH_4^+ (a) and NO_3^- (b) wet deposition along with the ratio of NH_4^+ to NO_3^- as nitrogen and 1:1 reference line (c).

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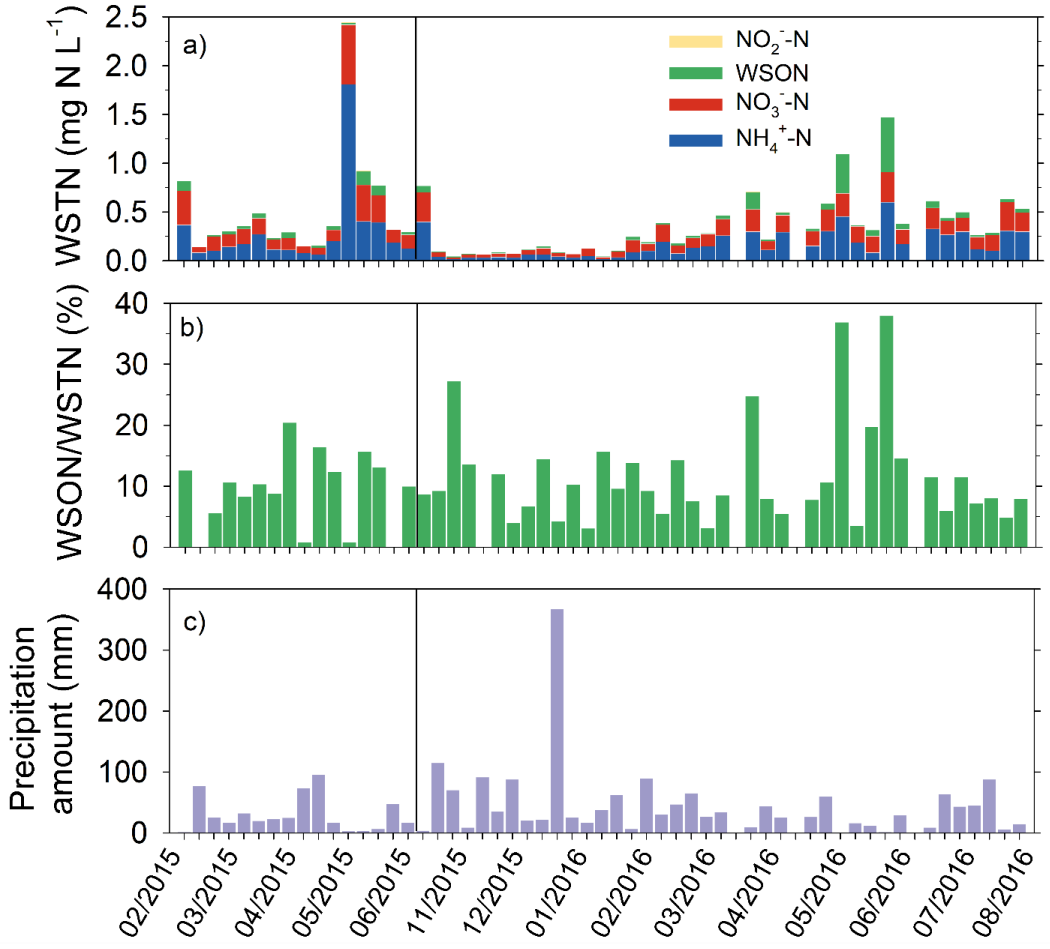

Figure 4. Concentrations of nitrogen species in weekly precipitation samples (a), percentage contribution of WSON to WSTN in precipitation (b), and precipitation amount (c). Vertical line marks discontinuity due to missing data from 06/01/2015 to 10/19/2015.

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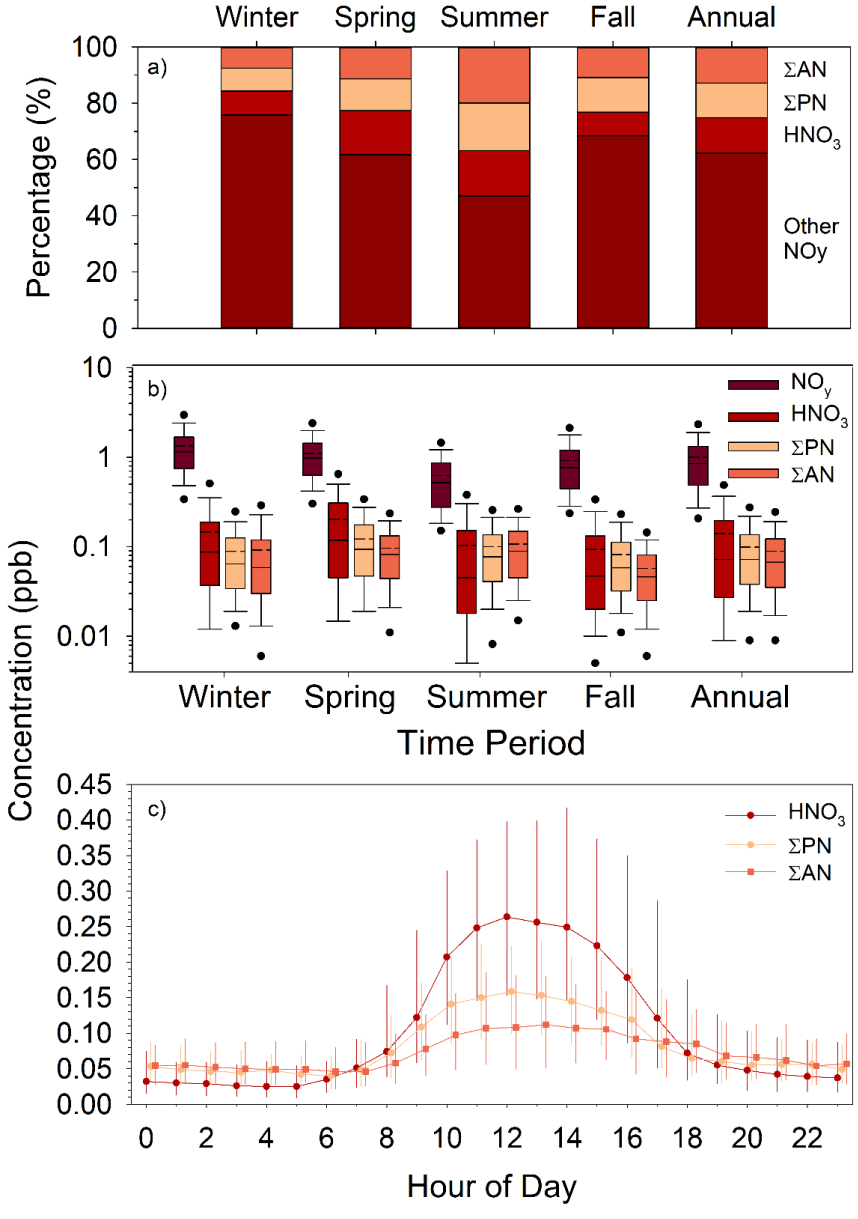

Figure 5. Seasonal and annual % contribution of HNO_3, ΣPN, ΣAN, and other compounds to total NOy (a); seasonal and annual boxplots of NOy, HNO_3, ΣPN and ΣAN where solid and dashed lines inside box represent median and mean, respectively; top and bottom of box represent 75th and 25th percentiles; whiskers represent 10th and 90th percentiles, and dots represent 5th and 95th percentiles (b); and diel profiles of HNO_3, ΣPN, and ΣAN where observations represent median hourly concentration and bars represent the interquartile range (c). "Other NOy" is calculated as NOy – HNO_3 – ΣPN – ΣAN which, while primarily comprised of NOx, includes N_2O_5, HONO, NO_3^-, and possibly other organics.

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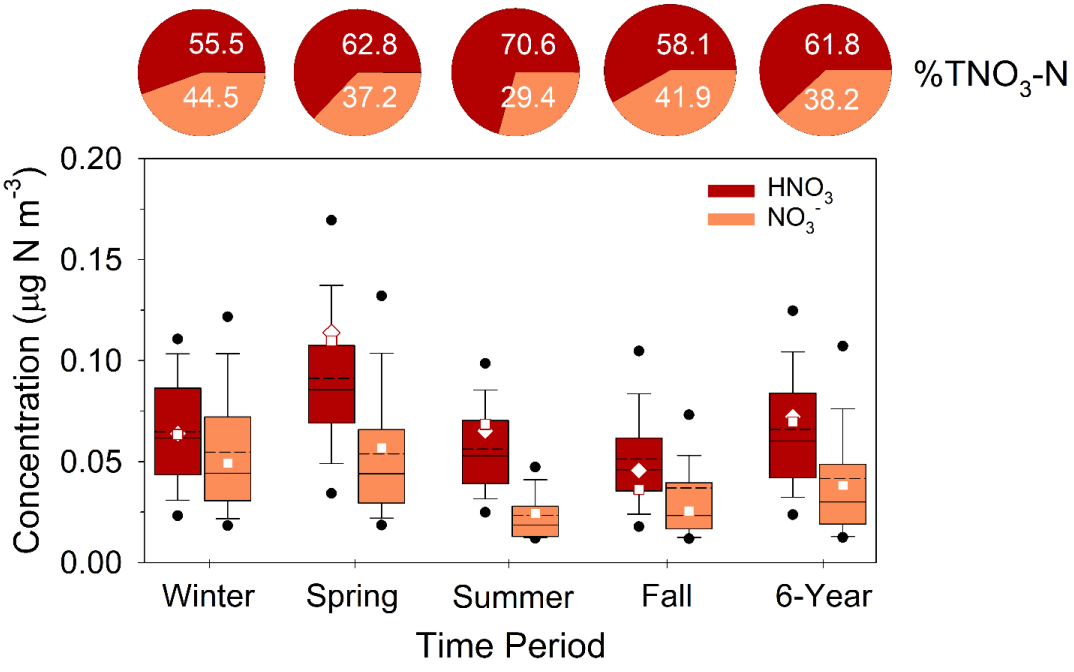

Figure 6. Summary of CASTNET HNO_3 and NO_3^- concentrations (as N) from 2015-2020 during winter, spring, summer and fall. Solid and dashed lines inside box represent median and mean, respectively. Top and bottom of box represent 75th and 25th percentiles. Whiskers represent 10th and 90th percentiles and dots represent 5th and 95th percentiles. "6-Year" represents the statistics for the entire 6-Year period. Squares and diamonds represent the seasonal and annual mean CASTNET (HNO_3 and NO_3^-) and continuous DD-CL HNO_3 for the August 2015 – August 2016 modeling period, respectively. Pie charts represent average % contribution of HNO_3 and NO_3^- to total NO_3^- (HNO_3 + NO_3^-) expressed as nitrogen.

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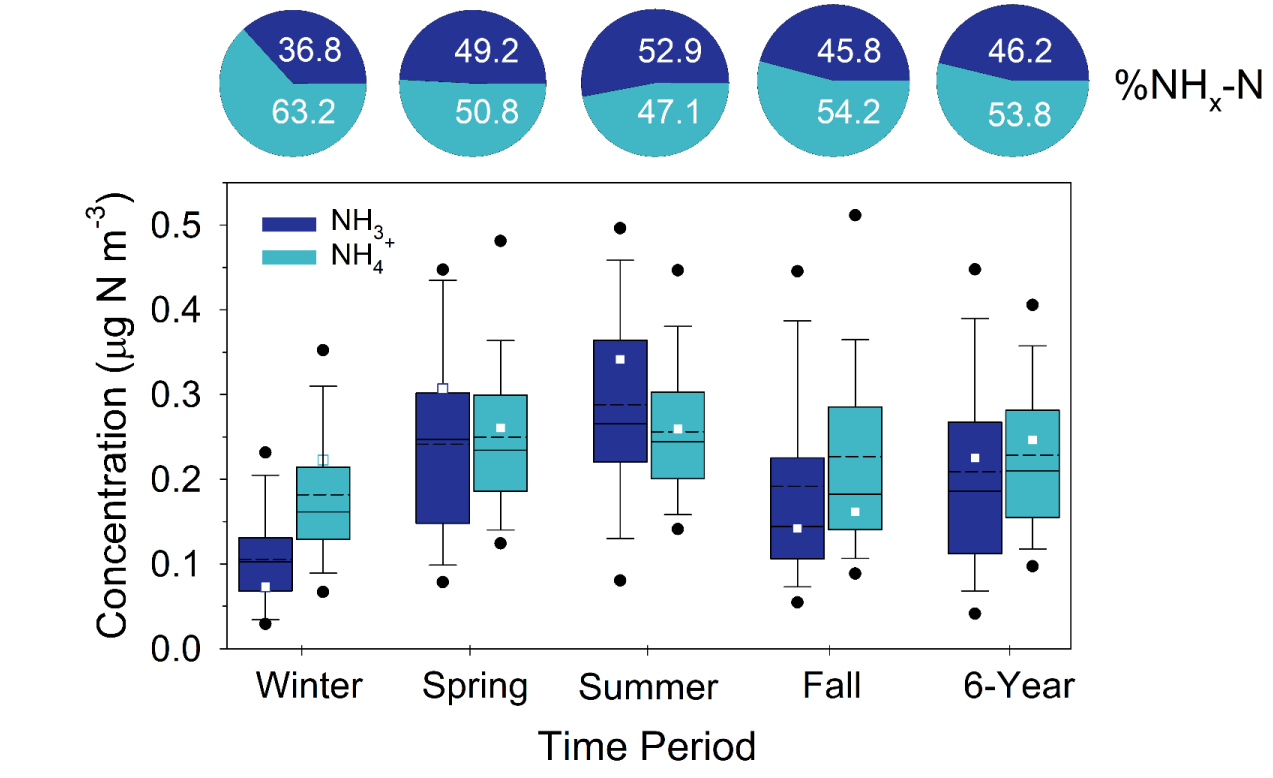

Figure 7. Summary of AMoN NH_3 and CASTNET NH_4^+ concentrations (as N) from 2015-2020 during winter, spring, summer and fall. Solid and dashed lines inside box represent median and mean, respectively. Top and bottom of box represent 75th and 25th percentiles. Whiskers represent 10th and 90th percentiles and dots represent 5th and 95th percentiles. "6-Year" represents the statistics for the entire 6-Year period. Squares represent the seasonal and annual mean concentration for the August 2015 – August 2016 modeling period. Pie charts represent average % contribution of NH_3 and NH_4^+ to total NH_x (NH_3+NH_4^+) expressed as nitrogen. AMoN concentrations were adjusted by subtracting the mean travel blank for the 6-year period, equivalent to 0.09 µg N m^{-3}, and applying a factor of 0.92 to account for the effect of higher elevation/lower pressure at Coweeta on the sampler diffusion coefficient.

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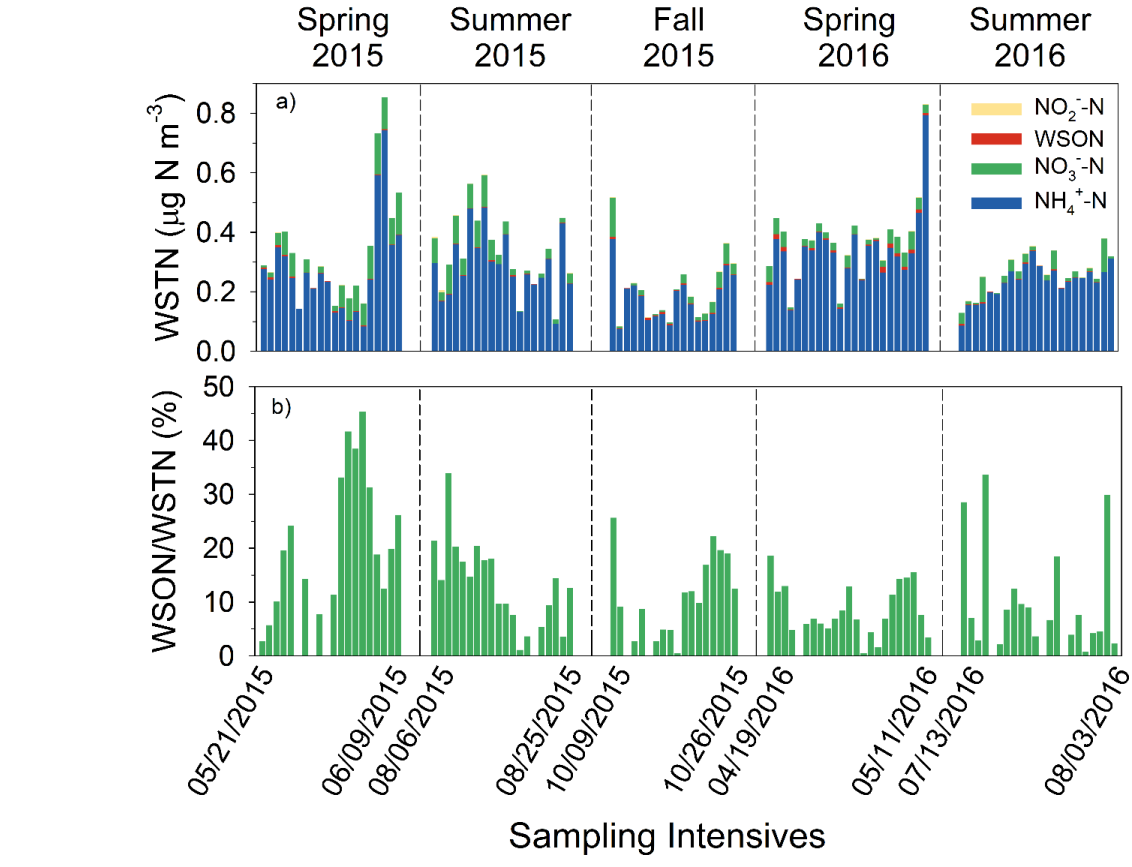

Figure 8. Contributions of N species to WSTN in 24-hour Hi-Vol $PM_{2.5}$ samples during seasonal SANDS intensives (a) along with percentage of WSON to WSTN (b).

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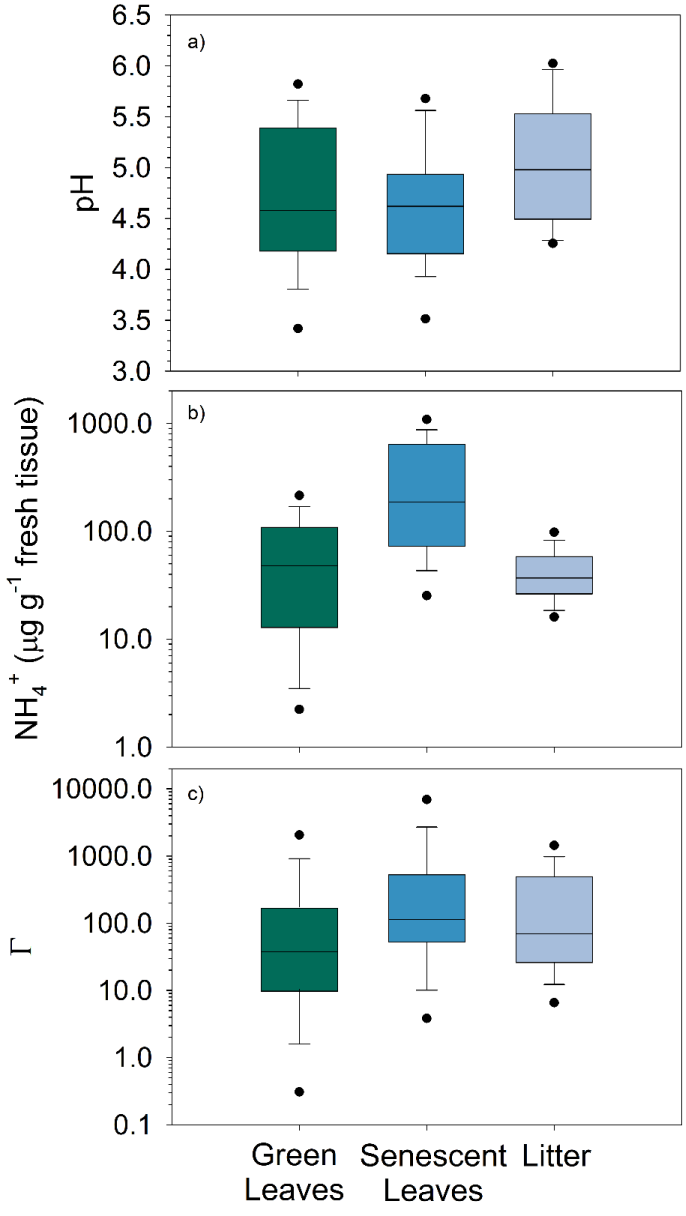

Figure 9. Boxplots of pH (a), NH_4^+ (b) concentration (µg g^{-1} fresh tissue), and equivalent emission potential (Γ) (c) in tissue of green leaves, senescent leaves, and litter on the forest floor. Solid line inside box represents median. Top and bottom of box represent 75th and 25th percentiles. Whiskers represent 10th and 90th percentiles and dots represent 5th and 95th percentiles.

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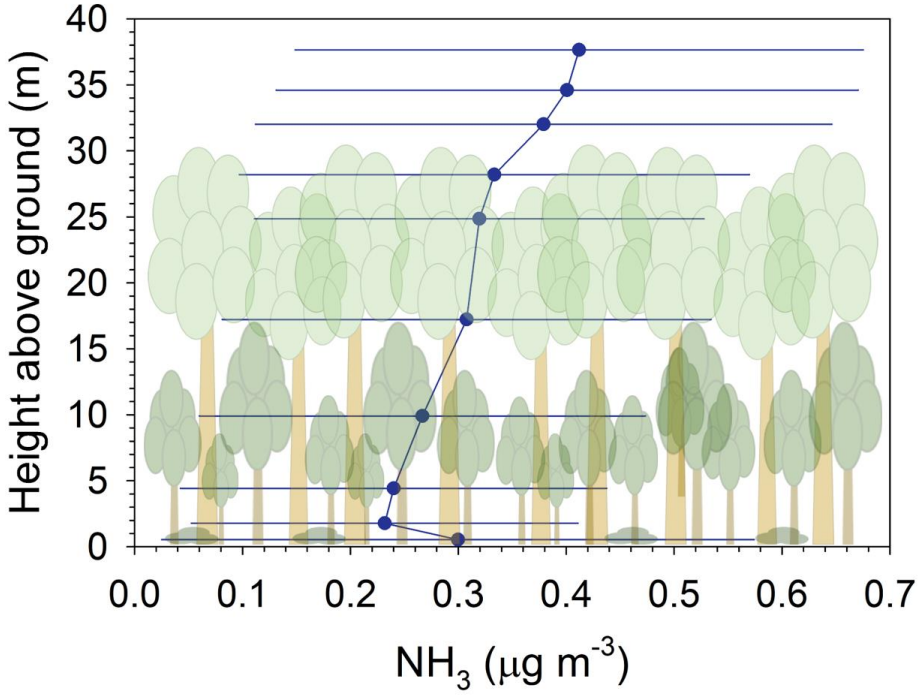

Figure 10. Vertical concentration profiles of NH_3. Mean (filled circle) and standard deviation (bars) of concentrations are shown for *N* = 76 daytime profiles.

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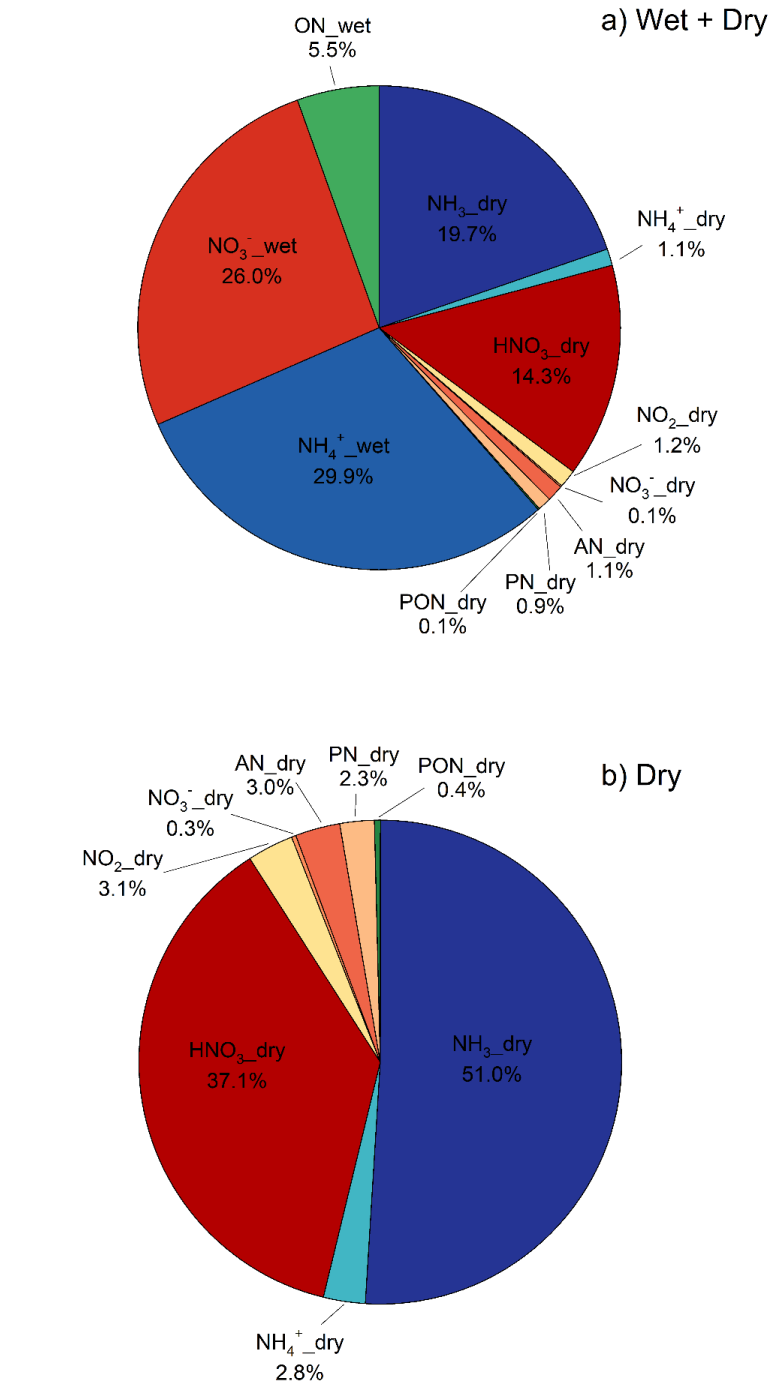

Figure 11. Speciated annual total (wet and dry) (a) and dry (b) deposition showing percent contribution of individual components.

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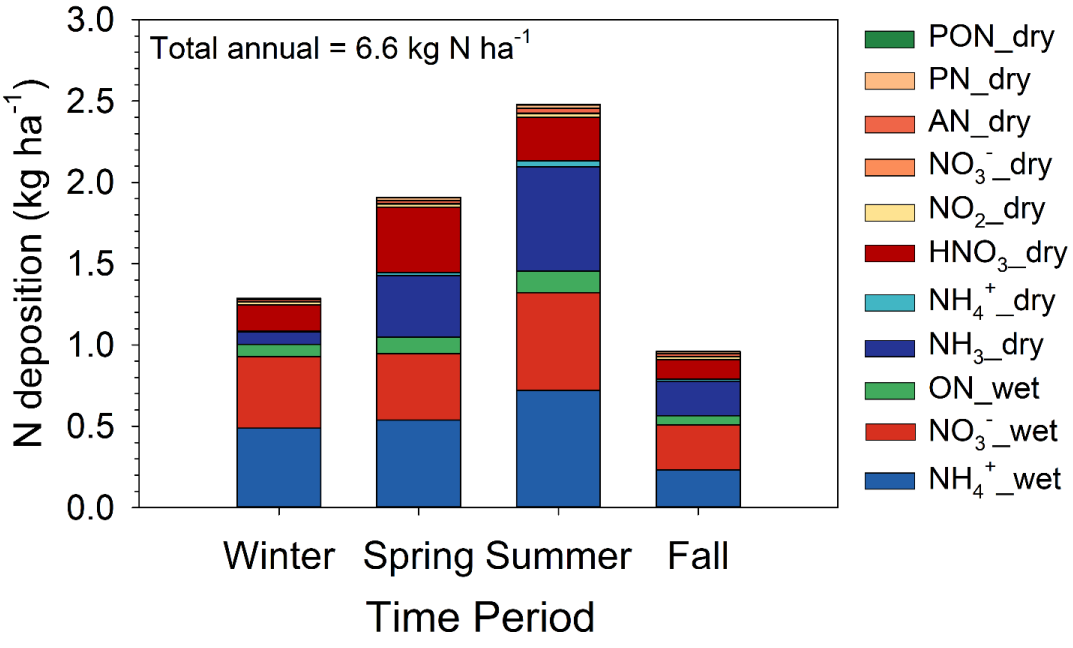

Figure 12. Seasonal speciated deposition budget. Nr species are listed in the legend as defined in the text, along with indication of the deposition pathway (dry or wet).

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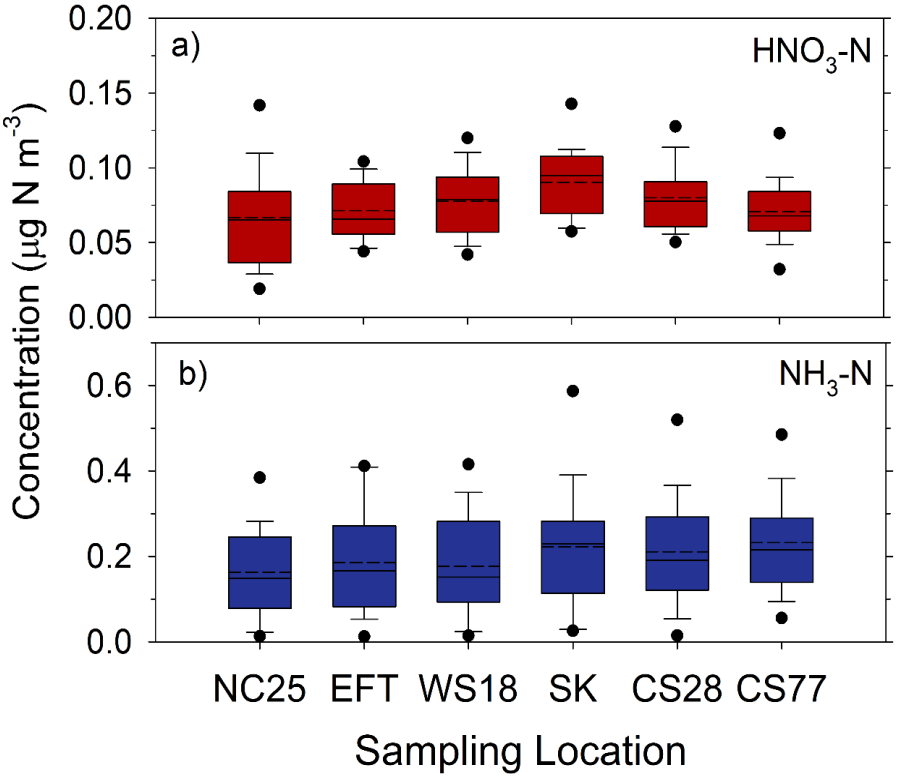

Figure 13. Concentrations (as N) of HNO_3 (a) and NH_3 (b) measured at different elevations, increasing from left to right (see Figure 1 and Table 1), across the Coweeta Basin. Solid and dash lines inside box represent median and mean, respectively. Top and bottom of box represent 75th and 25th percentiles. Whiskers represent 90th and 10th percentiles and dots represent 95th and 5th percentiles.