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23	Keywords: Carbon cycle; DIC; POC; isotopic fractionation; Precambrian analogues,	 Supprimé: analogs

Biogeochemical processes captured by carbon isotopes in

Abstract. Redox-stratified water columns are a prevalent feature of Earth history, and ongoing environmental changes tend to promote a resurgence of such settings. Studying modern redox-stratified environments has improved our understanding of biogeochemical processes and element cycling in such water columns. These settings are associated with peculiar carbon biogeochemical cycling owing to a layered distribution of biological processes in relation to oxidant availability. These processes and their sedimentological expression, notably inferred from paired organic matter-carbonate isotopic compositions of carbon, can vary from one stratified environment to another, because metabolisms from different biogeochemical layers are diverse and may differently imprint the sedimentological record. This variability can arise from numerous physico-chemical parameters, such as sedimentation rate, pH, trace element availability, or basin physiography. Changes in the organic/inorganic carbon sources and mass balance can further complicate the isotopic message in these systems. Better understanding of these multifaceted carbon isotope signals requires further evaluation of the transfer function from the processes occurring in redox-stratified water columns to sediments. We therefore characterized and compared the isotopic signatures of carbonate, organic matter, and dissolved inorganic carbon (DIC) reservoirs at different depths in the water column and upper sediments of four stratified Mexican lakes that follow a gradient of alkalinity/salinity. Comparing these systems shows strong diversity in both the water column and sedimentary carbon isotope signals. Differences in inorganic carbon isotope signatures arise primarily from the size of the DIC reservoirs, buffering the expression of redox-dependent biological processes as alkalinity increases. Combining this isotopic dataset with physico-chemical parameters of the water columns allows us to identify the activity of oxygenic photosynthesis and aerobic respiration in the four lakes studied, while anoxygenic photosynthesis is evidenced in two of them. However, sedimentary organic matter does not originate from the same water column layers in the four lakes, highlighting the ecological variability that can stem from distinct stratified water columns and how it is transferred to the sedimentary record. The least alkaline lake shows higher isotopic variability, and signatures typical of methanogenesis in the sediment porewaters. This metabolism, however, does not leave diagnostic isotopic signatures in the sedimentary archives (organic matter and carbonates), underlining the fact that even when alkalinity does not strongly buffer the inorganic carbon reservoir, a comprehensive picture of the active biogeochemical carbon cycling is not necessarily transferred to the geological record.

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Supprimé: Abstract. The carbon cycle is central to the evolution of biogeochemical processes at the surface of the Earth. Understanding the interplay between the C cycle and physico-chemical and biological conditions in ancient times is challenging because of major differences between the modern and the ancient Earth. Notably, the atmosphere was less oxidizing and the oceans were redox-stratified. Here, we characterized and compared the C cycles in four redoxstratified alkaline lakes from Mexico based on the concentrations and isotopic compositions of dissolved inorganic and particulate organic C (DIC and POC). Measurements were performed in both the water columns and surficial bottom sediments with the assessment of their physico-chemical parameters (conductivity, temperature, O₂, Chl. a, turbidity,). The four lakes exhibit a range of DIC concentrations from 7 to 35 mM, following a gradient of alkalinity/salinity. The DIC and sedimentary carbonates isotopic compositions ($\delta^{13}C_{DIC}$, $\delta^{13}C_{Carb}$) also varied correlatively with alkalinity increasing from -4.1 to +2.0 % and -1.5 to +4.7 ‰, respectively. The porewaters $\delta^{13}C_{\rm DIC}$ reaches up to ~10 ‰ in the sediment of one of the lakes. The δ^{13} C_{POC} varies from -29.0 to -23.5 % in both the water columns and sediments of the four lakes. The depth profiles of δ13CPOC, [POC] and C:N of organic matter shows very similar variations among three of the lakes located in the same area. From the inter-comparison of these datasets in four different systems, we identify the impact of external abiotic factors such as the hydrological regime and inorganic C sources which control the alkalinity, carbonate isotopic signatures, and stratification of some of the physico-chemical parameters. We identify the presence of oxygenic photosynthesis and aerobic respiration metabolisms in the four lakes as well as of methanogenesis in the one with extreme porewater $\delta^{13}C_{\text{DIC}}$. Anoxygenic photosynthesis and/or chemoautotrophy are also recognized in two of the lakes, but their POC and DIC signatures can be equivocal. Finally, we find that geochemical signatures of OC in the surficial sediments do not always record the same part of the stratified water column and can be altered by early diagenesis, whereas recently deposited carbonates are more consistently recording the lakes oxycline isotopic compositions. Overall, this work highlights how the integration of datasets from multiple environments facilitates our understanding of the processes affecting the C cycle in redox-stratified systems, while showing the versatility of these processes and how they are recorded in the sedimentary

1. INTRODUCTION

The carbon cycle and biogeochemical conditions prevailing at the surface of the Earth are intimately bound through biological (e.g. photosynthesis) and geological processes (e.g. volcanic degassing and silicate weathering). The analysis of carbon isotopes in organic matter and carbonates ($\delta^{13}C_{org}$ and $\delta^{13}C_{carb}$) in the rock record, has been used to reconstruct the evolution of the biosphere and the oxygenation of the Earth's surface (e.g. Hayes et al., 1989; Karhu and Holland, 1996; Schidlowski, 2001). Coupling $\delta^{13}C_{org}-\delta^{13}C_{carb}$ has frequently been used to infer the burial rate of organic C, and thus the redox balance of the atmosphere and hydrosphere (e.g. Karhu and Holland, 1996; Aharon, 2005; Krissansen-Totton et al., 2015; Mason et al., 2017). It has also been used to deduce the presence of metabolisms like anoxygenic chemoautotrophic or methanotrophic bacteria (e.g. Hayes et al., 1999; Bekker et al., 2008; Krissansen-Totton et al., 2015). Coupling δ^{13} Corg- δ^{13} Ccarb has also been used to discuss oceans stratification and its effect on inorganic and organic C geochemical signatures in sediments (e.g. Logan et al., 1995; Aharon, 2005; Bekker et al., 2008; Ader et al., 2009). Stratification favors the expression and recording of different layers of the water column, with potentially very distinct isotopic signatures. As the oceans were redox-stratified during, most of the Earth's history (Lyons et al., 2014; Havig et al., 2015; Satkoski et al., 2015), processes affecting the C cycle were likely different from those occurring in most modern, well-oxygenated environments. This change of conditions could impact the $\delta^{13}C_{org}$ signal at various scales, from changes in diversity and relative abundance of microbial carbon and energy metabolism (e.g. Wang et al., 2016; Iñiguez et al., 2020; Hurley et al., 2021), to larger ecological interactions (e.g. Jiao et al., 2010; Close and Henderson, 2020; Klawonn et al., 2021) and global C dynamics (e.g. Ridgwell and Arndt, 2015; Ussiri and Lal, 2017).

Modern stratified lakes have been used as analogues of ancient redox-stratified systems to better understand the C cycle in the sedimentary isotopic record (e.g. Lehmann et al., 2004; Posth et al., 2017; Fulton et al., 2018). Several number of recent studies have investigated the C cycle in modern stratified water columns (e.g. Crowe et al., 2011; Kuntz et al., 2015; Posth et al., 2017; Schiff et al., 2017; Havig et al., 2018; Cadeau et al., 2020; Saini et al., 2021; Petrash et al., 2022), where many bio-geo-physico-chemical parameters can be directly measured, together with the main C reservoirs. However investigations of such Precambrian analogues do not necessarily include sediment data, and generally focus on a single environment without integrating views from several systems.

In this study, we measured the concentrations and isotopic compositions of dissolved inorganic carbon (DIC) and particulate organic carbon (POC) throughout the water column of four modern redox-stratified alkaline crater lakes, located in the Trans-Mexican Volcanic Belt (Ferrari et al., 2012). We also measured the concentrations and isotopic compositions of the sedimentary organic carbon and carbonates as well as porewater DIC from surficial sediments (~ 10 cm) at the bottom of the lakes. The four lakes share similar geological and climatic contexts but have distinct solution chemistries along a marked alkalinity-salinity gradient (Zeyen et al., 2021) — as well as distinct planktonic communities (Iniesto et al., 2022). We therefore seek to evaluate how these environmental and ecological differences are recorded in the C isotope signatures in the water columns (DIC-POC) and sedimentary archives (organic matter-carbonates). The four lakes are closed lakes in endorheic basins (Alcocer, 2021; Zeyen et al., 2021), which facilitates the identification of external environmental constraints (e.g. evaporation, C sources) and their influence on processes occurring within the water columns. Depth profiles of the main physico-chemical parameters together with trace and major elements concentrations were measured to pinpoint the dominant biogeochemical processes occurring in the water columns and link them to specific C isotopes signatures.

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First, we constrain the main DIC sources and external controls on the lakes' alkalinity. Next, we describe the influence of the inter-lake alkalinity gradient on the inorganic C cycle and stratification of the lakes, and how it is recorded in surficial sediments. Then, by combining POC and DIC data, we identify the sources of organic C to the lakes by describing the main autotrophic reactions occurring in the water columns (e.g. oxygenic and anoxygenic photosynthesis). Finally, we discuss the fate of POC, either recycled (e.g. via methanogenesis) or deposited in the sediments, and how all these processes are recorded (or not) in surficial sediments.

2. SETTING / CONTEXT

2.1. Geology

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The four lakes studied here are volcanic maars formed after phreatic, magmatic, and phreatomagmatic explosions, related to volcanic activity in the Trans-Mexican Volcanic Belt (TMVB, Fig. 1). The TMVB originates from the subduction of the Rivera and Cocos plates beneath the North America plate, resulting in a long (~1000 km) and wide (90–230 km) Neogene volcanic arc spreading across central Mexico (Ferrari et al., 2012). The TMVB harbors a large variety of monogenetic scoria cones and phreatomagmatic vents (maars and tuff-cones) as well as stratovolcanoes, calderas, and domes (Carrasco-Núñez et al., 2007; Ferrari et al., 2012; Siebe et al., 2014). Maar crater formation usually occurs when ascending magma meets, water-saturated substrates, leading to successive explosions and the excavation of older units (Lorenz, 1986; Carrasco-Núñez et al., 2007; Siebe et al., 2012; Chako Tchamabé et al., 2020).

The first lake, La Alberca de los Espinos (1985 masl), is located at the margin of the Zacapu tectonic lacustrine basin in the Michoacán-Guanajuato Volcanic Field (MGVF) in the central-western part of the TMVB (Fig. 1). It lies on andesitic basement rocks and was dated at ~25 ± 2 ka (Siebe et al., 2012, 2014). The other three lakes (La Preciosa, Atexcac and Alchichica) are all in the same area (~ 50 km²) of the Serdan-Oriental Basin (SOB) in the easternmost part of the TMVB (Fig. 1). The SOB is a closed intra-montane basin at high altitude (~2300 m), surrounded by the Los Humeros caldera to the north and the Cofre de Perote-Citlatépel volcanic range to the east. The basement is composed of folded and faulted Cretaceous limestones and shales, covered by an desitic to basaltic lava flows (Carrasco-Núñez et al., 2007; Armienta et al., 2008; Chako Tchamabé et al., 2020). The Alchichica and Atexcac craters was dated at ~ 6-13 ± 5-6 ka (Chako Tchamabé et al., 2020) and 330 ± 80 ka (Carrasco-Núñez et al., 2007), respectively (Table 1). The age of La Preciosa is not known.

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2.2. Climate and limnology

La Alberca is a freshwater lake (0.6 psu) with a temperate to semi-humid climate (Rendon-Lopez, 2008; Sigala et al., 2017). In contrast, lakes from the SOB experience a similar temperate to semi-arid climate (Armienta et al., 2008; Sigala et al., 2017). The current climate of the SOB is dominated by dry conditions, reflected by higher evaporation than precipitation fluxes in Lake Alchichica (~ 1686 vs 392 mm/year; Alcocer, 2021). In La Preciosa, Atexcac, and Alchichica, this trend is reflected by a drop in water level evidenced by the emersion of microbialite deposits (Fig. S1; Zeyen et al., 2021). This evaporation-dominated climate strongly contributes to the relatively high salinity values in these lakes (1.2–7.9 psu) ranging from sub- to hyposaline.

(Zeven et al., 2021) Déplacé vers le haut [1]: - as well as distinct planktonic communities (Iniesto et al., 2022). Supprimé: Overall, their inter-comparison via the same methodology allows to assess the effects of specific physicochemical and biological parameters on the C cycle. Moreover, they all correspond to closed lakes in endorheic basins Mis en forme : Couleur de police : Noir Supprimé: in relation with Supprimé: -Supprimé: (~1000 and Supprimé: -Supprimé: , respectively Supprimé: phreatomagmatic vents (maars and Supprimé: magmas meet Supprimé: Three of the studied Déplacé (insertion) [2] Supprimé: Alchichica Supprimé: La Preciosa Supprimé: located Supprimé: a restricted Supprimé: in Supprimé: in Supprimé: mainly Supprimé: Supprimé: Supprimé: formations of Supprimé: were Supprimé: back to Supprimé: ka, respectively (Table 1; Carrasco-Núñez et a Supprimé: The fourth lake, La Alberca de los Espinos, is Déplacé vers le haut [2]: on andesitic basement rocks and Supprimé: Due Supprimé: their geographical proximity Supprimé: present Supprimé: as Supprimé: In Alchichica, Atexcac and Supprimé: their Supprimé: in these lakes Supprimé: achievement of

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The four lakes are warm monomictic; they are stratified for about nine months of the year, mixing only when thermal stratification breaks down in the cold of winter (Armienta et al., 2008). They are all closed lakes located in an "endorheic" basin (Alcocer, 2021; Zeyen et al., 2021), meaning that they have no inflow, outflow, or connection to other basins through surficial waters such as streams. The only water input is from precipitation and groundwater inflow (quantified for Lake Alchichica; (Alcocer, 2021 and references therein).

The four lakes are alkaline (pH ~ 9) but cover a broad range of chemical compositions (including alkalinity, salinity, and Mg/Ca ratio), interpreted as reflecting different concentration stages of an initial alkaline dilute water (Table 1; Zeyen et al., 2021). Variations in concentration stages may be due to differences in climate and more generally, different hydrological regimes. Microbialite deposits are found in all four lakes (Gérard et al., 2013; Saghaï et al., 2016; Iniesto et al., 2021a, 2021b; Zeyen et al., 2021), and increase in abundance from lower to higher alkalinity conditions (Zeyen et al., 2021).

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3. METHOD

3.1. Sample Collection

The sediment core from Lake La Preciosa was collected in May 2016. All other samples were collected in May 2019. The depth profiles of several physico-chemical parameters were measured in the water columns of the four lakes using an YSI Exo 2 multi-parameter probe: temperature, pH, ORP (oxidation reduction potential), conductivity, O_2 , chlorophyll a, phycocyanin, and turbidity. Precisions for these measurements were $0.01\,^{\circ}$ C, $0.1\,^{\circ}$ pH unit, $20\,^{\circ}$ mV, $0.001\,^{\circ}$ mS/cm, $0.1\,^{\circ}$ mg/L, $0.01\,^{\circ}$ μg/L, $0.01\,^{\circ}$ μg/L and $2\%\,^{\circ}$ FTU unit, respectively. The ORP signal was not calibrated before each profile and is thus used to discuss relative variations over a depth profile. Measurements of the aforementioned parameters served to pinpoint depths of interest for further chemical and isotopic analyses, notably around the redoxcline of the lakes. Water samples were collected with a Niskin bottle. Particulate matter was collected on pre-combusted ($2\,^{\circ}$ h at $490\,^{\circ}$ C) and weighted glass fiber filters (Whatman GF/F, $0.7\,^{\circ}$ µm) and analyzed for particulate organic carbon (POC), major and trace elements. Between $1.5\,^{\circ}$ and $5\,^{\circ}$ L of lake water was filtered before the GF/F filters became clogged. The processed solution was filtered again at $0.22\,^{\circ}$ µm with Filtropur S filters (pre-rinsed with lake water filtered at $0.7\,^{\circ}$ µm) for analyses of dissolved inorganic carbon (DIC), and major, minor, and trace ions.

Sediment cores were collected using a 90 mm Uwitec corer close to the deepest point of each lake's water column (Table 1), where anoxic conditions prevail almost all year long. Cores measured between 20 and 85 cm in length. Slices of about 2-3 cm were cut under anoxic conditions, using a glove bag filled with N_2 (anoxia was monitored using a WTW3630 equipped with a FDO O_2 optode). Interstitial porewater was drained out of the core slices using Rhizons in the glove bag. Sediments were transported back to the laboratory within aluminized foils (Protpack, UK). Sediments were then fully dried in a laboratory anoxic N_2 -filled glove box.

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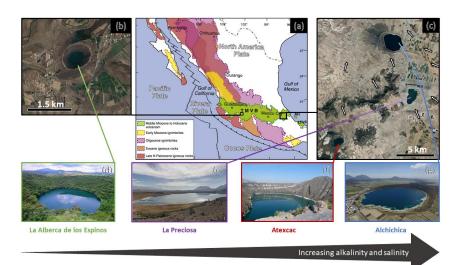


Figure 1. Geographical location and photographs of the <u>four</u> crater lakes. (a) Geological map from Ferrari et al. (2012) <u>with black squares showing</u> the location of the four studied lakes <u>within</u> the <u>Trans</u> Mexican <u>Volcanic Belt</u> (TMVB). (b, c) Close up Google Earth views of <u>La</u> Alberca de los Espinos and the Serdan-Oriental Basin (SOB). The white arrows represent the approximate groundwater flow path (based on Silva-Aguilera, 2019) (d-g) <u>Photographs</u> of the four lakes (d from Google Image ['enamoratedemexicowebsite'], e from Google Earth street view, and g from Google Engagen').

<u>Lake</u>	General location	Sampling location	Elevation (masl)
Alchichica	Serdan Oriental Basin,	19°24'51,5" N;	2220
Alchichica	eastern TMVB	97°24'09,9" W	<u>2320</u>
Atovos	Serdan Oriental Basin,	19°20'2.2" N;	2260
<u>Atexcac</u>	eastern TMVB	97°26'59.3" W	<u>2360</u>
La Preciosa	Serdan Oriental Basin,	19°22'18.1" N;	2220
La Preciosa	eastern TMVB	97°23'14.4" W	<u>2330</u>
<u>La Alberca de</u>	Zacapu Basin, MGVF,	19°54'23.9" N;	1005
los Espinos	central TMVB	101°46'07.8" W	<u>1985</u>

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<u>Lake</u>	Lake Basement	<u>Age</u>	Max Depth (m)	Alkalinity (mmoles/L)	Salinity (psu)	рH
<u>Alchichica</u>	limestone, basalts	<u>6-13 ± 5-6 ka</u>	<u>63</u>	<u>~35</u>	<u>7.9</u>	9.22
<u>Atexcac</u>	limestone, andesites, basalts	330 ± 80 ka	<u>39</u>	<u>~26</u>	<u>7.4</u>	<u>8.85</u>
La Preciosa	limestone, basalts	<u>Pleistocene</u>	<u>46</u>	<u>~13.5</u>	1.15	9.01
<u>La Alberca de</u> <u>los Espinos</u>	andesite xenoliths	<u>25 ± 2 ka</u>	<u>30</u>	<u>~7</u>	0.6	9.14

Table 1. General information about the <u>lakes</u> studied, Abbreviations: TMVB: Trans-Mexican <u>Volcanic Belt</u>; MGVF: Michoaćan-Guanajuato <u>Volcanic Field; mast</u>; meters above sea level. *NB*: Sampling <u>took place</u> in May 2019, except for La <u>Preciosa</u> sediments, sampled in May 2016.

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3.2. Dissolved inorganic carbon (DIC) concentration and isotope measurements

Twelve \underline{ml}_{\bullet} of the 0.7-µm filtered lake water \underline{was} filtered at 0.22-µm directly into hermetic Exetainer® tubes to avoid exchange between DIC and atmospheric CO₂. The DIC concentrations and isotopic compositions were measured at the Institut de Physique du Globe de Paris (IPGP, France), using an Analytical Precision 2003 GC-IRMS, running under He-continuous flow, following the protocol described by Assayag et al. (2006). A given volume of the solution was extracted from the Exetainer® tube with a syringe, while the same volume of helium was introduced to maintain stable pressure and atmospheric-CO₂-free conditions within the sample tubes. The collected sample was inserted into another Exetainer® tube pre-filled with a few drops of 100% phosphoric acid (H₃PO₄) and pre-flushed with He gas. Under acidic conditions, DIC quantitatively converts to gaseous and aqueous CO₂, which equilibrates overnight within the He-filled head space of the tube. Quantification and isotopic analyses of released gaseous CO₂ were then carried out by GC-IRMS using internal standards of known composition that were prepared and analyzed via the same protocol. Each measurement represented an average of four injections in the mass spectrometer. Chemical preparation and IRMS analysis were duplicated for all the samples. The δ^{13} CDIC reproducibility calculated for the 65 samples was better than ± 0.2 %, including internal and external reproducibility. Standard deviation for [DIC] was 0.6 ± 0.9 mmol/L on average.

Specific DIC speciation, i.e., $CO_{2(aq)}$, HCO_3^- and CO_3^{2-} activities, was computed using Phreeqe with the full dissolved chemical composition of each sample as an input. It should be noted that these results are <u>calculated from</u> theoretical chemical equilibria and do not necessarily take into account local kinetic effects, which, for example, could lead to local exhaustion of $CO_{2(aq)}$ where intense photosynthesis occurs.

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3.3. Particulate organic carbon and nitrogen (POC / PON) $\,$

Particulate organic matter from the lake water columns was collected on GF/F filters, dried at room temperature and ground in a ball mill before and after decarbonation. Decarbonation was performed with 12N HCl vapors in a desiccator for 48 h. Aliquots of dry decarbonated samples (25 - 70 mg) were weighed in tin capsules. The POC and PON contents and $\delta^{13}C_{POC}$ were determined at the Laboratoire Biogéosciences (Dijon, France) using a Vario MICRO cube elemental analyzer (Elementar, Hanau, Germany) coupled in continuous flow mode with an IsoPrime IRMS (Isoprime, Manchester, UK). The USGS 40 and IAEA 600 certified materials used for calibration showed reproducibility better than 0.15 % for $\delta^{13}C$. External reproducibility based on triplicate analyses of samples (n=23) was 0.1 % on average for $\delta^{13}C_{POC}$ (1SD). External reproducibility for POC and PON concentrations was ρ .001 and 0.005 mmol/L on average, respectively (i.e. 3 and 7 % of measured concentrations).

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3.4. Geochemical characterizations of the sediments

Sedimentary organic carbon (SOC), sedimentary organic nitrogen (SON), and their isotopic compositions were measured on carbonate-free residues of the first 12 cm of the sediment cores, produced after overnight 1N HCl digestion. Plant debris (mainly found in La Alberca and Atexcac) was identified upon initial sediment grinding in an agate mortar and analyzed separately. Aliquots of dried decarbonated samples (\sim 4-70 mg) were weighed in tin capsules. The SOC and SON contents and δ^{13} C were determined at the Laboratoire Biogéosciences (Dijon) using

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a Vario MICRO cube elemental analyzer (Elementar GmbH, Hanau, Germany) coupled in continuous flow mode with an IsoPrime IRMS (Isoprime, Manchester, UK). The USGS 40 and IAEA 600 certified materials used for calibration had a reproducibility better than 0.2 % for $\delta^{13}C_{SOC}$. Sample analyses (n=67) were at least duplicated and showed an average external reproducibility of 0.1 % for δ¹³C (1SD). External reproducibility for SOC and SON contents was 0.1 and 0.03 wt. %, respectively. Carbon isotope compositions of sedimentary carbonates were analyzed at the Laboratoire Biogéosciences (Dijon) using a ThermoScientificTM Delta V PlusTM IRMS coupled with a Kiel VI carbonate preparation device. External reproducibility was assessed by multiple measurements of NBS19 standard and was better than \pm 0.1 % (2 σ). Total carbonate concentration was determined by mass balance after decarbonatation for SOC analysis. Mineralogical assemblages of sediments were determined on bulk powders by X-Ray diffraction (XRD) at the Laboratoire Biogéosciences (Dijon). Samples were ground in an agate mortar. Diffractograms were obtained with a Bruker D8 Endeavor diffractometer with CuKα radiation and LynxEye XE-T detector, under 40 kV and 25 mA intensity. Mineral identification was based on COD ("Crystallography Open Database") and BGMN databases. Mineral abundances were estimated by Rietveld refinement analysis implemented in the Profex software. Solid sulfide concentrations were determined on dry bulk sediments from La Alberca Lake after a wet chemical extraction using a boiling acidic Cr(II)-solution as detailed in Gröger et al. (2009).

3.5. Major and trace elements concentrations

Dissolved and particulate matter elemental compositions were measured at the Pôle Spectrométrie Océan (Plouzané, France) by inductively coupled plasma_atomic_emission_spectroscopy (ICP-AES, Horiba Jobin) for major elements and by high_resolution_ICP_mass spectrometry using an Element XR (HR-ICP-MS, Thermo Fisher Scientific) for trace elements. Major element measurement reproducibility based on internal multi-elemental solution was better than 5%. Trace elements were analyzed by a standard-sample bracketing method and calibrated with a multi-elemental solution. Analytical precision for trace elements was generally better than 5%. Dissolved sulfate concentrations were analyzed by ion chromatography at the IPGP (Paris, France) with uncertainty lower than 5%.

4. RESULTS

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4.1. Lake La Alberca de los Espinos

Stratification of the water column was well defined in La Alberca de los Espinos (Fig. 2). Temperature was higher than in the other lakes (decreasing from ~23 °C at the surface to 16.5 °C at depth). Dissolved O₂ was oversaturated at the lake surface (118 %, *i.e.*, 7.9 mg/L), rapidly decreasing to 0 between ~ 5 and 12 m, while the oxidation reduction potential (ORP) only decreased below 17 m depth. The offset between O₂ exhaustion and ORP decrease can be explained by the presence of other oxidant species and/or extended chlorophyll a peaks (supplementary text 1). Conductivity decreased from 1.20 to 1.17 mS/cm at 16 m before increasing to 1.27 mS/cm at 26 m (salinity

Déplacé (insertion) [6] Déplacé (insertion) [7] Déplacé (insertion) [8] Supprimé: Supprimé: -mass Supprimé: Supprimé: -Supprimé: -Supprimé: an Déplacé vers le bas [9]: <#>Lake Alchichica¶ <#>The water column of Lake Alchichica showed a pronounced stratification compared to previous years at the same period (Fig. 2, Fig. S2; Lugo et al., 2000; Adame et al., 2008; Macek et al., 2020). Supprimé: <#>The water temperature varied from about 20 °C at the surface to 15.5 °C at a 30 m depth and below. Déplacé vers le bas [10]: <#> pH remained constant at **Supprimé:** <#> 2).¶ Déplacé vers le bas [11]: <#> The $\delta^{13}C_{DIC}$ decreased Supprimé: <#>2). Calculated pCO2 was about three time Déplacé vers le bas [12]: <#>Lake Atexcac¶ **Supprimé:** <#>Temperature was about 20.6 °C at the Déplacé vers le bas [13]: <#> Turbidity showed a Supprimé: <#>Finally, pH remained around 8.85 Déplacé vers le bas [14]: $<\#>\delta^{13}C_{POC}$ showed minimum Supprimé: <#>Dissolved sulfate concentration was Déplacé vers le bas [15]: <#>¶ Mis en forme : Non souligné Supprimé: <#>The temperature varied from about 20 °C Déplacé vers le bas [16]: <#>Dissolved O₂ was Mis en forme: Police: Non Italique **Supprimé:** <#>8.4 mg/L) and rapidly decreased to 0 Déplacé vers le bas [17]: <#> and recorded the highest Supprimé: <#>Finally, pH showed a small decrease from Déplacé vers le bas [18]: <#>2).¶ Supprimé: <#>2). Calculated pCO_2 at the surface Déplacé vers le bas [19]: <#>). The $\delta^{13}C_{DIC}$ decreased Déplacé vers le bas [20]: <#>POC concentration **Déplacé vers le bas [21]:** <#> Porewaters from the 2016 **Supprimé:** $<\#>\delta^{13}C_{POC}$ increased downward from \sim -27 (Supprimé: was also well defined Supprimé: evolving

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747 between 0.58 and 0.64 psu). Chlorophyll a (Chl a) averaged 3.1 µg/L, and showed a profile with at least three Supprimé: La Alberca had relatively high chlorophyll a 748 distinctive peaks, (i) between 6 and 9.5 m, (ii) around 12.5 m and (iii) between 16 and 19 m, all reaching ~4 µg/L. Supprimé: on average) but 749 The turbidity profile showed a pronounced increase from 16 to 19 m. The pH profile showed important variation. **Supprimé:** all reaching approximately 4 μg/L. They were [750 Supprimé: at from 9.15 at the lake surface to 8.75 between 6.5 and 10 m, further decreasing to 7.5 between 16 and 26 m. Based Supprimé: 751 on the temperature profiles, epi-, meta- and hypolimnion layers of Lake La Alberca de los Espinos in May 2019 Supprimé: relatively 752 broadly extended from 0-5, 5-12 and 12-30 m, respectively (Fig. 2). The conductivity and pH profiles, however, Supprimé: variations 753 show that different conditions prevail at the top and bottom of the hypolimnion. Supprimé: can be defined as extending 754 Dissolved inorganic carbon (DIC) concentration progressively increased from 6.8 mM at 5 m to 8.7 mM at 26 m. Supprimé: 755 Supprimé: Though we notice via the The pCO2 calculated for surface waters was near equilibrium with atmospheric pCO2atm2 but strongly increased Supprimé: 756 with depth, up to ~ 40 times the pCO_{2atm} (Table S2). The δ^{13} C_{DIC} first decreased from about -2.5 % to -4.1 % Supprimé: Calculated 757 between 5 and 10 m, before increasing again up to -2 ‰ at 25 m. Particulate organic carbon (POC) concentrations Supprimé: at the 758 reached minimum values of 0.02 mM at 10 m but rose to maximum values in the hypolimnion (0.06 mM). The Supprimé: were 759 C:N molar ratio of particulate organic matter (POM) progressively decreased from 8.5 at the surface to less than, Supprimé: S1 760 6.5 in the hypolimnion. The $\delta^{13}C_{POC}$ had minimum values at 10 and 17 m (-28.3 and -29.2%, respectively). Above Supprimé: at 761 and below these depths, $\delta^{13}C_{POC}$ averaged -26.4 \pm 0.5 \%. Supprimé: and then increased back. Supprimé: increased back 762 Dissolved sulfates as measured by chromatography were only detectable at 5 m with a low concentration of 12 µM, Supprimé: (~ 763 while total dissolved S measured by ICP-AES, showed values in the hypolimnion higher than in the upper layers Supprimé: 05 764 (~ 10.3 ys. 7.4 μ M, Table S4). Dissolved Mn concentrations decreased from 1.5 to 0.5 μ M between 5 and 10 m, Supprimé: below 765 then increased to 2 µM at 25 m. Aqueous Fe was only detectable at 25 m with a concentration of 0.23 µM Supprimé: 766 (Table S4). In parallel, particulate S concentrations increased with depth, with a marked increase from 0.1 to Supprimé: was around 767 $0.6 \, \mu M$ between 20 and 25m. This was spatially correlated with a 25-fold increase in particulate Fe (from 0.2 to Mis en forme 768 5.97 µM). Particulate Mn showed a peak between 17 and 20 m around 1 µM, contrasting with values lower than Supprimé: S2 769 $0.15 \mu M$ in the rest of the water column (Fig. 2, Table <u>S5</u>). Supprimé: S2 Supprimé: S3 770 In the first centimeters of sediments, DIC concentration in the porewater varied between ~ 11 and 12 mM and Supprimé: the 771 δ¹³C_{DIC} varied between +8 and +10 ‰ (Figs. 3, 4). Surficial sedimentary carbonates corresponded to calcite and Supprimé: porewater 772 had a $\delta^{13}C$ around -1.5 ‰. Sedimentary organic matter had a $\delta^{13}C_{SOC}$ increasing from \sim -29.4 to -25.5 ‰ and a Supprimé: and 773 C:N molar ratio varying between 11.6 and 14.3 (Figs. 3, 4; Table S3). Supprimé: ~ 11 and 12 mM and between Supprimé: ‰, respectively 774 Supprimé: globally 775 4.2. Lake La Preciosa Supprimé: 3 776 Lake La Preciosa was also stratified at the time of sample collection (Fig. 2). Temperature decreased from ~ 20 °C Mis en forme 777 at the surface to 16 °C at 15m depth. Conductivity showed the same trend with values between 2.24 and Supprimé: Lake 778 2.22 mS/cm (salinity around 1.15 psu). Dissolved O2 was oversaturated at the lake surface (120 %, j.e., 8.4 mg/L), Mis en forme 779 rapidly decreasing to 0 between ~ 8 and 14 m, while the ORP decreased right below 16 m. Chl a concentration Déplacé (insertion) [15] 780 averaged 3 µg/L, and recorded the highest peak compared to the other lakes (about 9 µg/L at 10 m) before Mis en forme 781 decreasing to 0.7 µg/L below 15 m. Turbidity showed a large peak between 16 and 19 m. The pH showed a small Déplacé (insertion) [16]

decrease from 9 to 8.8 between the surface and 15 m depth. Based on the temperature profiles, epi-, meta- and

hypolimnion layers of La Preciosa in May 2019 broadly extended from 0-6, 6-15 and 15-46 m, respectively

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(Fig. 2).

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The DIC concentration was constant throughout the water column at 13.3 mM, with an exception at 12.5 m, where it decreased to 11.5 mM (Fig. 3, Table S1). Calculated pCO₂ at the surface represented about two times the atmospheric pCO_{2atm} (Table S2). The $\delta^{13}C_{DIC}$ decreased from about 0.5 % to -0.36 % between the surface and the hypolimnion. The POC concentration decreased from ~0.06 mM in the epi-/metalimnion to 0.02 mM in the hypolimnion. Similarly, (C:N)_{POM} decreased from ~11.2 in the epi-/metalimnion to 7.6 in the hypolimnion. The $\delta^{13}C_{POC}$ increased downward from ~ -27 to -25 % to with a peak to -23.5 % at 15 m.

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In the first 10 cm of sediments, $\delta^{13}C_{SOC}$ values increased downwards from ~ -25.5 to -23.2 % and C:N molar ratio from 9.8 to 11 (Figs. 3, 4; Table S3). Carbonates corresponded to aragonite and calcite and had a bulk C isotope composition averaging 2.6 % (Table S3), Porewaters from the 2016 La Preciosa core were not retrieved.

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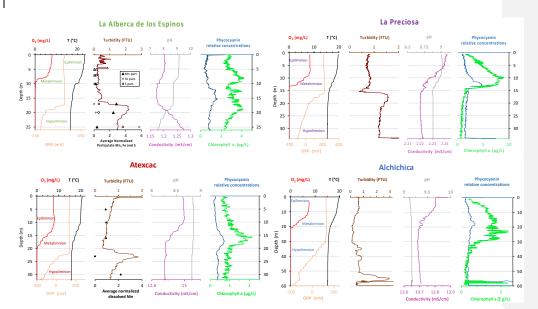


Figure 2. Physico-chemical parameters depth profiles in La Alberca de los Espinos , La Preciosa, Atexcac and Alchichica in May 2019 including: dissolved oxygen concentration (mg/L), water temperature (°C), oxidation-reduction potential (ORP, mV), turbidity (Formazin Turbidity Unit), pH, conductivity (mS/cm), phycocyanin and chlorophyll a pigments (μg/L). Absolute values for phycocyanin concentrations were not determined; only relative variations are represented (with increasing concentrations to the right). Discrete concentration values of dissolved Mn in Atexcac and particulate Mn, Fe and S in La Alberca, normalized by their respective average are represented. Epi-, meta- and hypo-limnion layers are depicted for each lake according to temperature profiles. The three layers closely corresponded to oxygen-rich, oxygen-poor and intermediate zones (except in La Preciosa where the oxycline was slightly thinner than the thermocline layer, ~5 vs. 8 m).

4.3. Lake Atexcac

Stratification of the Lake Atexcac water column was also very well defined (Fig. 2). Temperature decreased from ~ 20.6 °C at the surface to reach 16 °C below 20 m. Conductivity showed the same trend with values between 13 and 12.8 mS/cm near the surface (salinity around 7.4 psu). Dissolved O₂ was slightly oversaturated at the lake

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837	surface (115 % or 7.6 mg/L), rapidly decreasing to 0 mg/L between ~ 10 and 20 m, while ORP signal decreased
838	below a depth of 22 m. Chl a averaged 1 μ g/L and showed a narrow peak centered at around 16 m, reaching
839	~2 μg/L Turbidity showed a pronounced increase below 20 m, peaking at 23.3 m and returning to surface values
840	at 26 m. The pH remained around 8.85 throughout the water column. Based on the temperature profiles, the epi-,
841	meta- and hypolimnion of Atexcac in May 2019 broadly extended from 0-10, 10-20 and 20-39 m, respectively
842	(Fig. 2).
843	The DIC concentration was around 26 mM throughout the water column, except at 23 m where it decreased to
844	$\underline{24.2 \text{ mM}}$ (Fig. 3, Table S1). Calculated pCO ₂ was about five times higher than the atmospheric pCO _{2atm}
845	(Table S2). The $\delta^{13}C_{\underline{DIC}}$ was stable around 0.4 % in the epi-/metalimnion, but increased to 0.9 % at 23 m and
846	$\underline{\text{reached }0.2\%}$ minimum values at the bottom of the lake. The POC concentration was ~ 0.05 mM in the epi-
847	/metalimnion, decreasing to 0.02 mM in the hypolimnion. The C:N molar ratio of POM showed the same depth
848	profile, decreasing from ~9.6 in the epi-/metalimnion to 6.6 in the hypolimnion (Fig. 3). The $\delta^{13}C_{POC}$ showed
849	minimum values in the epi-/metalimnion (-29.3 % at 16 m) and increased to -26.5 % in the hypolimnion.
850	$\underline{\text{Dissolved sulfate concentration was relatively stable at} \sim 2.51 \text{ mM throughout the water column but increased to}$
851	$\underline{2.64}$ mM at 23 m. Dissolved Mn concentration was constant at 1 μ M down to 16 m before dropping to 0 at 23 m.,
852	and increasing again to 2.35 μM at 30 m (Fig. 2; Table S4). Similar depth profiles were found for other heavy
853	elements as well, including Cu, Sr, Ba or Pb among others.
854	In the first 12 cm of sediments, DIC concentration in the porewater varied between ~ 21 and 26 mM, and $\delta^{13}C_{DIC}$
855	was around 0 ‰. Carbonates corresponded to aragonite and calcite and had a bulk C isotope composition between
856	$\underline{2.1}$ and $\underline{2.6}$ % (Table S3). Sedimentary organic matter had a $\delta^{13}C_{\underline{SOC}}$ averaging $\underline{-26.8 \pm 0.1}$ % and a C:N molar
857	ratio increasing from 8 to 10 (Figs. 3, 4; Table S3).

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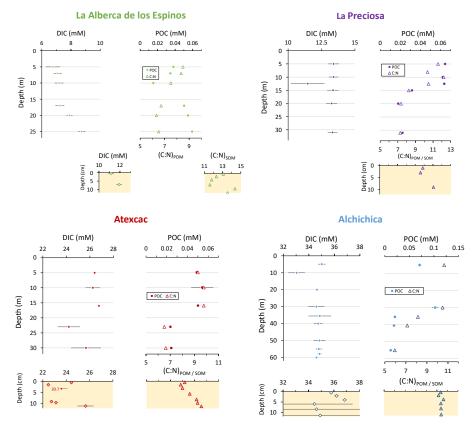


Figure 3. Concentrations in mmol/L (mM) of DIC, DOC, POC and sum of all three reservoirs, C:N molar ratios of POM as a function of depth in the water columns, as well as DIC concentrations in the surficial sediment porewaters and C:N molar ratios of sedimentary OM. Porewaters from La Preciosa's 2016 core were not retrieved.

4.4. Lake Alchichica

The water column of Lake Alchichica showed a pronounced stratification compared to previous years at the same period (Fig. 2, Fig. S2; Lugo et al., 2000; Adame et al., 2008; Macek et al., 2020). Temperature decreased from ~ 20 °C at the surface to 15.5 °C at depths below 30 m. Conductivity showed the same trend with values between around 13.8 mS/cm (salinity decreasing from 7.9 to 7.8 psu). Dissolved O₂ was slightly oversaturated at the lake surface (112 % or 7.5 mg/L), rapidly decreasing to 0 mg/L between ~ 10 and 20 m. The ORP followed a similar trend but decreasing below 30 m only. The offset between O₂ exhaustion and decrease of the ORP can be explained by the presence of other oxidant species and/or extended Chl a peaks (supplementary text 1). Chl a averaged 2 µg/L, with a broad peak extending from ~ 7 to 29 m (averaging 4 µg/L and showing a narrow 6 µg/L maximum values at 23 m. Then, it decreased to minimum values of ~ 0.5 µg/L in the lower water column. The pH remained

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constant at ~9.2 over the whole water column. Based on the temperature profiles, the epi-, meta- and hypolimnion layers of Lake Alchichica in May 2019 extended from 0-10, 10-20 and 20-63 m, respectively (Fig. 2).

The DIC concentration was around 34.8 mM throughout the water column, except at 10 m where it decreased to 33 mM (Fig. 3; Table S1). Calculated pCO₂ was about three times higher than the atmospheric pCO_{2atm} (Table S2)_{χ} The δ^{13} C_{DIC} decreased from 2 to ~ 1.5 ‰ between 5 and 60 m depth (Fig. 4; Table S1). The POC concentration was ~ 0.09 mM in the epi-/metallimnion, decreasing to 0.02 mM in the hypolimnion. The δ^{13} C_{POC} increased from ~26.5 ‰ in the top 30 m to ~24.1 ‰ at 55 m. The C:N molar ratio of POM showed a similar profile with values around 10.5 down to 30 m, progressively decreasing towards 5.9 at 55 m (Fig. 3; Table S1).

In the first 12 cm of sediments, porewater DIC had a concentration of ~ 35.5 mM and $\delta^{13}C_{DIC}$ decreased from 0.4 to -0.5 ‰. Solid carbonates were contained within several phases (aragonite, hydromagnesite, huntite and calcite) and had a bulk C isotope composition around 4.6 ‰ (Table S3). Sedimentary organic matter had a $\delta^{13}C_{SOC}$ increasing from -25.7 to -24.5 ‰ and a constant C:N molar ratio slightly higher than 10 (Figs. 3, 4; Table S3).

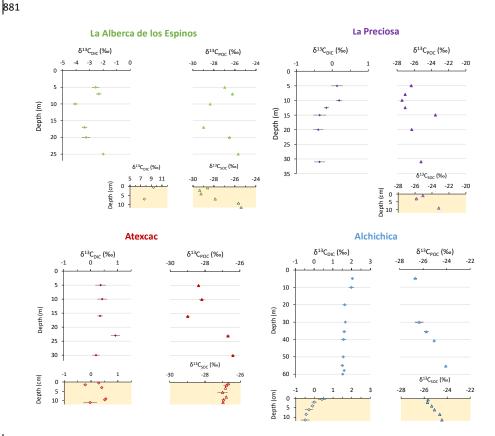


Figure 4. Isotopic compositions of DIC and POC reservoirs as a function of depth in the water columns, as well as isotopic compositions of the porewater-DIC and total organic carbon from the surficial sediments.

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5. DISCUSSION

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5.1. Inorganic Carbon: origins and implications of the alkalinity/DIC gradient

5.1.1 Sources of DIC and origin of the inter-lake alkalinity gradient

Salinity and DIC concentration gradually increase from La Alberca de los Espinos (0.6 psu, 7 mM) to Alchichica (7.9 psu, 35 mM), while La Preciosa (1.15 psu, 13 mM) and Atexcac (7.44 psu, 26 mM) have intermediate values (Table 1 and S1). This trend matches the alkalinity gradient (with values of ~ 8, 15, 32 and 47 meq/L, Fig. S3a) previously described for these lakes (Zeyen et al., 2021), consistent with the fact that alkalinity is mainly composed of HCO₃⁻ and CO₃² ions in most natural waters. This alkalinity gradient may result from different concentration stages of an initial dilute alkaline water (Zeyen et al., 2021), ultimately controlled by differences in hydrological regime between the four lakes. In the SOB, the weathering of basaltic/andesitic bedrock (Armienta et al., 2008; <u>Carrasco-Núñez et al., 2007; Lelli et al., 2021) and Cretaceous limestone (with $\delta^{13}C \approx 0 \pm 1$ %; Gonzales-Partida</u> et al., 1993; Armstrong-Altrin et al., 2011) favors the inflow of more alkaline and DIC-concentrated groundwater than in La Alberca, which lies on an essentially basaltic basement (Rendon-Lopez, 2008; Siebe et al., 2014; Zeyen et al., 2021). The SOB is currently experiencing higher rates of evaporation than precipitation (Alcocer, 2021), which may play an important role in concentrating solutes and decreasing the water level in La Preciosa Atexcac, and Alchichica (Anderson and Stedmon, 2007; Zeyen et al., 2021). Substantial "sub-fossil" microbialite deposits emerge well above the current water level in lakes Atexcac and Alchichica, confirming this fall in water level (~15 m for Atexcac, and ~5 m for Alchichica). Scattered patches of microbialites emerge at La Preciosa (suggesting a water level decrease of ~6 m). By contrast, emerged microbialites are virtually absent in Lake La Alberca de los Espinos (Fig. S1).

Additional local parameters, such as <u>variable</u> groundwater paths and fluxes (Furian et al., 2013; Mercedes-Martín et al., 2019; Milesi et al., 2020; Zeyen et al., 2021) most likely play a role in explaining <u>some</u> of the <u>variation</u> in DIC concentration between lakes. La Preciosa's water composition significantly differs from that of <u>Atexcac</u>, and Alchichica, despite a similar geological context and climate (all <u>are</u> located within 50 km², Fig. 1). <u>Groundwater</u> in the SOB area <u>becomes</u> more saline as <u>it flows</u> towards the center of the basin and through the crater lakes (Silva Aguilera, 2019; Alcocer, 2021). Since <u>groundwater flows</u> through La Preciosa first, <u>it becomes</u> more concentrated as <u>it enters</u> Alchichica (Silva Aguilera, 2019; Alcocer, 2021; Lelli et al., 2021). <u>Different regimes of volcanic CO2</u> degassing into these crater lakes may also contribute to <u>variation in the C mass balance and δ¹³C_{DIC} values between the four lakes. Near the lakes from the SOB area, geothermal fluids derived from meteoric waters <u>have been shown</u> to interact with deep volcanic fluids as well as the calcareous basement rocks (Peiffer et al., 2018; Lelli et al., 2021). In the water column of La Alberca, *δ*¹³C_{total} averages -4.8 % (Havas et al., submitted). This <u>isotopic composition</u> is very similar to signatures of mantle-CO₂ (Javoy et al., 1986; Mason et al., 2017), which could buffer the overall C <u>isotope</u> composition of this lake La Alberca is located on top of a likely active normal fault (Siebe et al., 2012), favoring the ascent of volcanic gases.</u>

Differences in the remineralization rate of organic carbon (OC) could also contribute to the heterogeneous DIC content among the lakes. However, assuming that all OC from the lakes ultimately remineralized into DIC, it would still represent only a small proportion of the total carbon (9 % for La Alberca, ~5 % for La Preciosa and Alchichica, and 16 % for Atexcac, Havas et al., submitted). From an isotopic mass balance perspective, Lake La

Supprimé: Table 2. Concentrations and isotopic compositions of dissolved inorganic carbon (DIC) and particulate organic carbon (POC) and C:N molar ratios of particulate organic matter (POM).

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Supprimé: Salinity and DIC gradually increase from Lake La Alberca de los Espinos (~0.6 psu and 7 mM) to Alchichica (~7.9 psu and 35 mM), while lakes La Preciosa and Atexcac have intermediate values of 1.15 and 7.44 psu and 13 and 26 mM, respectively (Table 1 and 2). This matches a gradient of alkalinity (with values of ~ 8, 15, 32 and 47 meq/L, Fig. S3a) previously described for these lakes (Zeyen et al., 2021), consistently with the fact that alkalinity is mostly composed of HCO3- and CO32- ions in most natural waters. This alkalinity gradient may result from different concentration stages of an initial dilute alkaline water (Zeyen et al., 2021), those different concentration stages being ultimately controlled by the different hydrological regimes of the lakes. First, the weathering of Cretaceous limestone in the SOB (with a δ^{13} C of approximately 0 ± 1 %; Gonzales-Partida et al., 1993; Armstrong-Altrin et al., 2011) together with basaltic/andesitic bedrock (Armienta et al., 2008; Carrasco-Núñez et al., 2007; Lelli et al., 2021) favors the inflow of more alkaline and DIC-concentrated groundwaters than in Lake La Alberca which lies on a essentially basaltic basement (Rendon-Lopez, 2008; Siebe et al., 2014; Zeyen et al., 2021). {"citationID":"wxjCvLQB","properties":{"formattedCitation" "(Rendon-Lopez, 2008; Siebe et al., 2014; Zeyen et al., 2021)", "plainCitation": "(Rendon-Lopez, 2008; Siebe et al., 2014; Zeyen et al.,

2021)", "noteIndex":0}, "citationItems":[{"id":427, "uris":["htt p://zotero.org/users/6690596/items/HSE8CNXR"], "uri":["htt p://zotero.org/users/6690596/items/HSE8CNXR"], "itemData" :["id":427, "type":"article", "title":"Limnologia fisica del lago crater los Espinos, Municipio de Jiménez

Michoacan", "URL": "https://www.google.com/url?sa=t&rct=j &q=&esrc=s&source=web&cd=&ved=2ahUKEwjx1YXR26z 2AhVPzRoKHbDrCQ0QFnoECAUQAQ&url=http%3A%2F %2Fbibliotecavirtual.dgb.umich.mx%3A8083%2Fxmlui%2F handle%2FDGB_UMICH%2F5774&usg=A0vVaw35MpX

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Supprimé: may partially explain the striking increases of P_{CO2} and [DIC] and decrease of pH observed at depth (Table S1; Figs. 2 and 3). Moreover, this lake

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Alberca exhibits more negative $\delta^{13}C_{DIC}$ (and $\delta^{13}C_{Carb}$), slightly closer to OC signatures, whereas the $\delta^{13}C_{DIC}$ of the three SOB lakes lie very far from OC isotopic signatures (Fig. 4). Dense vegetation surrounds La Alberca (Fig. \$1), making it the only lake in this study where OC respiration could be a significant source of inorganic C to the water column (potentially influencing the P_{CO2} , [DIC] and pH profiles described above).

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In summary, a combination of very local and external environmental factors generates the contrasting water chemistries of the lakes, notably a gradient in their alkalinity/[DIC]. This variability stems from the exact nature of the basement rocks, the distinct groundwater flow paths feeding the lakes, differences in evaporation rates, and potentially different volcanic-CO₂ degassing regimes.

5.1.2 Influence of alkalinity on physico-chemical stratification in the four lakes

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Stratified water columns can sustain strong physico-chemical gradients, where a wide range of biogeochemical reactions impacting the C cycle can take place (e.g. Jézéquel et al., 2016). In the four lakes studied here, the evolution of pH with depth exemplifies the interplay between the alkalinity gradient, the physico-chemical stratification of the lakes, and their respective C cycle. The pH shows a stratified profile in La Alberca and La Preciosa, but remains constant in Atexcac and Alchichica. The decline in pH at the oxycline of La Preciosa is associated with the decrease in POC and chlorophyll a concentrations and δ^{13} Cpic values, reflecting the impact of oxygen respiration (j.e. carbon remineralization) at this depth (Figs. 2-4). In La Alberca, the surface waters are markedly more alkaline than the bottom waters, with a two-step decrease in pH occurring at 8 m and 17 m (with a total drop of 1.5 pH unit). As in La Preciosa, this likely results from high OM respiration, although input of volcanic acidic gases (e.g. dissolved CO₂ with δ^{13} C ~-5 %) might also contribute to the pH decrease in the bottom waters, reflected by negative δ^{13} Cpic signatures and an increase of [DIC] and conductivity in the hypolimnion (Figs. 2 and 4). By contrast, while the same evidence for oxygen respiration ([POC], chlorophyll a) can be detected in the other two lakes, it does not impact their pH profile in a similar way (Fig. 2). This result suggests that the acidity generated by these reactions is buffered by the much higher alkalinity measured in these two lakes.

<u>vis particularly important considering the critical interplay between pH and biogeochemical reactions affecting the C cycle (e.g. Soetaert et al., 2007).</u>

External forcings such as lake hydrology and fluid sources thus impact the alkalinity buffering capacity of these lakes and influence the vertical pH profile of the water columns. This

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5.1.3 Sinks of DIC along the alkalinity gradient

Interplay between pH and sources of alkalinity/DIC in the lakes also has a strong impact on their C storage capacity as it can result in different fluxes of the C sinks (inorganic and organic C precipitation / sedimentation, CO_2 degassing).

Alkaline pH can store large quantities of DIC because it favors the presence of HCO_3^- and CO_3^{2-} species over $H_2CO_3^*$ (the intermediate species between gaseous $CO_{2(g)}$ and bi-/carbonate ions, defined here as the sum of H_2CO_3 and $CO_{2(gq)}$). Carbonate and bicarbonate ions represent over 99% of total DIC in the four lakes (Table S2). In La Alberca de los Espinos, the lake with the lowest DIC, the surface water pCO_2 is slightly lower than atmospheric pCO_{2stm} (Table S2). By contrast, large amounts of CO_2 degas at the surface of the SOB lakes, as indicated by their

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1275	elevated surface water pCO ₂ , from 2 to 5 times higher than atmospheric pCO _{2atm} (Table S2). These observations	
1276	are consistent with the notion that higher DIC concentrations favor CO ₂ degassing through higher pCO ₂ (e.g.	
1277	Duarte et al., 2008). Although La Alberca and Alchichica (the two endmembers of the alkalinity gradient) have	
1278	the same surface water pH, CO ₂ degassing is three times higher at Alchichica, for a given value of gas transfer	
1279	<u>velocity.</u>	
1280	amounts of carbonate deposits (Zeyen et al., 2021). The occurrence of microbialites increases along the alkalinity	Déplacé (insertion) [27]
1281	gradient, with limited presence at La Alberca, and more massive deposits at Atexcac and Alchichica (Zeyen et al.,	
Ì	Another important C sink for these lakes is the precipitation of carbonate minerals, found in the microbialites	
	and lake sediments. Lake alkalinity and resulting mineral saturation index greatly influence the amount of C	
	precipitated from the lake waters. Although the four lakes are supersaturated with aragonite, calcite and the precursor phase monohydrocalcite, they present highly contrasted	
1282	2021; Fig. S1). Similarly, surficial sediments contain only 16 wt. % for La Alberca, but from 40 to 62 wt. %	Déplacé (insertion) [28]
1283	carbonates for the SOB lakes (Table S3). Thus, the SOB lakes seem to bury more C than La Alberca de los Espinos.	
1284	Nonetheless, the data from May 2019 indicate that La Alberca was the only one of the four lakes with a pCO ₂	
1285	slightly lower than atmospheric pCO _{2atm} , thus representing a net sink of C ₂ Classifying the three other lakes as net	Déplacé (insertion) [29]
1286	C sources or sinks - notably in order to see the influence of their respective position in the alkalinity gradient -	
1287	will require a more detailed description of C in- and out-fluxes since they all store and emit significant amounts	
1288	of C (as organic and inorganic C deposits and via CO ₂ degassing, respectively). However, this is out of the scope	
1289	of the present study.	Mis en forme : Police :Gras
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1291	5.1.4 Isotopic signatures of inorganic C in the four lakes ($\delta^{13}C_{DIC}$ and $\underline{\delta^{13}C_{Carbonates}}$)	
1292	δ^{13} C _{DIC} in La Alberca is consistent with influence of remineralized OC and/or volcanic CO ₂ . The δ^{13} C _{DIC} in the	Déplacé (insertion) [30]
1293	SOB lakes suggests groundwater δ ¹³ C _{DIC} values resulting from the dissolution of the Cretaceous limestone	- spines (messass) [ce]
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	The DIC isotopic composition of the lakes (between ~ -3 and +2 ‰ on average; Table S1) is consistent with the DIC sources described above. The lower	
1294	<u>basement.</u>	
1295	_for lakes with a pH around 9 (Bade et al., 2004), where DIC is dominated by HCO ₃ . However, the pH values of	Déplacé (insertion) [31]
1296	the four lakes studied here are too similar to explain the significant difference between their δ ¹³ C _{DIC} (Fig. 4;	Deplace (insertion) [31]
	By controlling DIC speciation ($H_2CO_3/CO_{2(aq)}$, HCO_3 , CO_3^{2-}), pH also strongly influences $\delta^{13}C_{DIC}$. Indeed, there is a temperature-dependent fractionation of up to 10 % between the different DIC species (Emrich et al., 1970;	
	Mook et al., 1974; Bade et al., 2004; Table S6). The Mexican lakes present $\delta^{13}C_{DIC}$ values that are common	
1297	$p=4.2 \times 10^{-3}$ for La Preciosa and Atexcac, which have the closest $\delta^{\bar{1}3} C_{DIC}$). Part of the variability of $\delta^{13} C_{DIC}$ among	
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	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate	
1299	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by	
1300	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by increasing lake pCO ₂ and primary productivity. This bolsters CO ₂ degassing and organic C burial, which have low	
1300 1301	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by increasing lake pCO ₂ and primary productivity. This bolsters CO ₂ degassing and organic C burial, which have low $\delta^{13}C$ compared to DIC (e.g. Li and Ku, 1997; Talbot, 1990). Accordingly, the pCO ₂ of La Alberca is lower than	Déplacé (insertion) [32]
1300 1301 1302	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by increasing lake pCO ₂ and primary productivity. This bolsters CO ₂ degassing and organic C burial, which have low $\delta^{13}C$ compared to DIC (e.g. Li and Ku, 1997; Talbot, 1990). Accordingly, the pCO ₂ of La Alberca is lower than that of the other lakes (Table S2). The $\delta^{13}C_{DIC}$ in lakes with lower DIC concentrations is expected to be more easily	Déplacé (insertion) [32]
1300 1301 1302 1303	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by increasing lake pCO ₂ and primary productivity. This bolsters CO ₂ degassing and organic C burial, which have low $\delta^{13}C$ compared to DIC (e.g. Li and Ku, 1997; Talbot, 1990). Accordingly, the pCO ₂ of La Alberca is lower than that of the other lakes (Table S2). The $\delta^{13}C_{DIC}$ in lakes with lower DIC concentrations is expected to be more easily influenced by exchanges with other carbon reservoirs, such as organic carbon (through photosynthesis/respiration),	
1300 1301 1302 1303 1304	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by increasing lake pCO ₂ and primary productivity. This bolsters CO ₂ degassing and organic C burial, which have low $\delta^{13}C$ compared to DIC (e.g. Li and Ku, 1997; Talbot, 1990). Accordingly, the pCO ₂ of La Alberca is lower than that of the other lakes (Table S2). The $\delta^{13}C_{DIC}$ in lakes with lower DIC concentrations is expected to be more easily influenced by exchanges with other carbon reservoirs, such as organic carbon (through photosynthesis/respiration), or other DIC sources (e.g., depleted volcanic CO ₂ or groundwater DIC) compared with buffered, high DIC lakes	Déplacé (insertion) [32] Déplacé (insertion) [33]
1300 1301 1302 1303	the lakes may result from their distinct evaporation stages, as the mean $\delta^{13}C_{DIC}$ values of the lakes broadly correlate with their salinity/alkalinity (Fig. S3b). Evaporation generally increases the $\delta^{13}C_{DIC}$ of residual waters by increasing lake pCO ₂ and primary productivity. This bolsters CO ₂ degassing and organic C burial, which have low $\delta^{13}C$ compared to DIC (e.g. Li and Ku, 1997; Talbot, 1990). Accordingly, the pCO ₂ of La Alberca is lower than that of the other lakes (Table S2). The $\delta^{13}C_{DIC}$ in lakes with lower DIC concentrations is expected to be more easily influenced by exchanges with other carbon reservoirs, such as organic carbon (through photosynthesis/respiration),	

the four lakes, likely reflecting organic and/or volcanic C influence and thus higher responsiveness to biogeochemical processes of the inorganic C reservoir. By contrast, the three SOB lakes exhibit $\delta^{13}C_{DIC}$ with less internal variability, with a maximum amplitude of 0.7 % within a single water column.

 $\delta^{13}C_{Carb}$ also follows and reflects the alkalinity gradient, with the lowest $\delta^{13}C_{Carb}$ found in the surficial sediments of La Alberca (~ -1.5 %), intermediate values in La Preciosa and Atexaca (~2.5 %), and the highest values in

Surficial sedimentary carbonates are in isotopic equilibrium with the $\delta^{13}C_{DIC}$ of the water columns, within the uncertainty of $\delta^{13}C_{DIC}$ measurement, and more specifically with the $\delta^{13}C_{DIC}$ values at the oxycline/thermocline of the lakes (Tables S6 and S7). This is estimated by correcting the carbonates C isotope composition ($\delta^{13}C_{Carb}$) by the fractionation value between DIC and the different carbonate mineralogies (supplementary text S2). Therefore, the

Alchichica (~+4.6 %) (Table S3).

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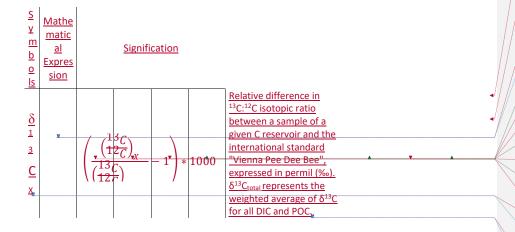
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In summary, although all four lakes present the same general structure and environmental conditions (i.e. tropical alkaline stratified crater lakes), external and local factors (e.g. hydrology, fluid sources, and stratification characteristics) result in contrasting water chemistry compositions, which have a critical impact on the physicochemical depth profiles of each lake and their biogeochemical carbon cycle functioning. These external factors represent a first-order control on the size, isotopic composition, and responsiveness to biogeochemical processes of the inorganic C reservoir. Lakes with the highest alkalinity/DIC content will poorly record internal biological processes. Interestingly, C storage in mineral carbonates seems to be significant in watersheds where carbonate deposits pre-exist in the geological substratum (here, the Cretaceous limestone basement), providing more alkaline and C-rich sources.



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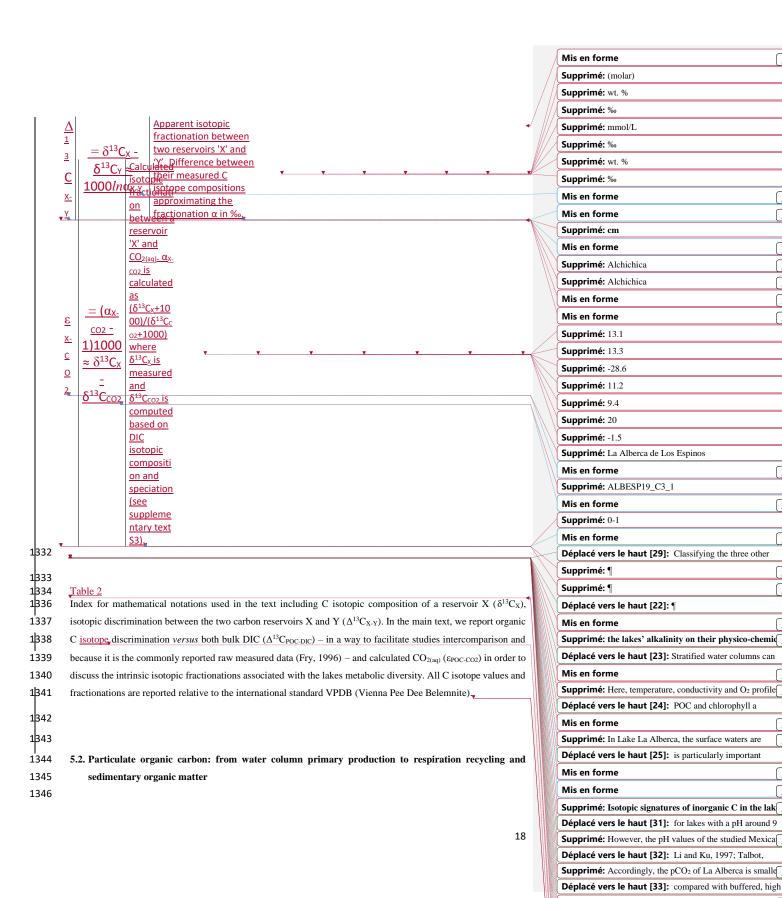
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5.2.1. Particulate organic C sources

Primary productivity by oxygenic photosynthesis in the upper water column

All four crater lakes are endorheic basins, with no surface water inflow or outflow. The organic carbon sources are therefore predominantly autochthonous, resulting mainly from planktonic autotrophic C fixation. This is supported by C:N ratios of POM that ranged from 6 to 12 in the four lakes close to the phytoplankton Redfield but far from land plant ratios. Abundant vegetation covers the crater walls of La Alberca and to a lesser extent those of Atexcac; some plant debris was observed and sampled from the sediment cores of these two lakes. They have high C:N ratios (between 24 and 68), typical of plant tissues and significantly higher than those of the bulk organic matter in the surficial sediments (between 8 and 13) and water column (between 6 and 12) (Fig. 3). Thus, the allochthonous organic carbon in these two lakes does not significantly contribute to their bulk organic signal.

The importance of planktonic autotrophic C fixation as a major source of POC in the four lakes is further supported by the assessment of the isotopic discrimination between DIC and organic biomass, expressed as $\Delta^{13}C_{POC-DIC}$ and $\epsilon_{POC-CO2}$ (Table 2). The $\Delta^{13}C_{POC-DIC}$ varies between ~ -29 and -23 % (corresponding to $\epsilon_{POC-CO2}$ between ~ -19 and -13 %), throughout the four water columns, within the typical range of planktonic oxygenic phototrophs (Pardue et al., 1976; Sirevag et al., 1977; Thomas et al., 2019). Yet these values exhibit variability – both within a single water column (up to 4.5 %) and among the four lakes (up to 6 %, Figs. 4 and 5). This variability may reflect several abiotic and biotic factors.

Notably, lower DIC availability in La Alberca and La Preciosa probably makes the carboxylation step less limiting

during photosynthesis (e.g. O'Leary, 1988; Descolas-Gros and Fontungne, 1990; Fry, 1996), decreasing [EPOC-CO2]

in these lakes (between 14,5 and 17.7, at the peak of Chl. a) compared with Atexcac, and Alchichica (Fig. 5a;

between 17,5 and 19.2 ‰). Lower $CO_{2(aq)}$ availability and/or higher reaction rates result in transport-limited rather than carboxylation-limited fixation, with smaller C isotope fractionation between POC and DIC (Pardue et al., 1976; Zohary et al., 1994; Fry, 1996; Close and Henderson, 2020). The isotopic fractionation associated with diffusion is much smaller than with carboxylation, and a higher proportion of the DIC entering the cells is converted into organic biomass (e.g. Fogel and Cifuentes, 1993). We consistently notice a correlation among the lakes between $a(CO_2)_{(aq)}$ (or [DIC]) and $|\varepsilon_{POC-CO2}|$ at depths where oxygenic photosynthetic peaks (Fig. 6). Furthermore, La Alberca and La Preciosa are considered less oligotrophic than the two other lakes (Lugo et al., 1993; Vilaclara et al., 1993; Havas et al., submitted), with higher chlorophyll a contents and thus smaller $|\varepsilon_{POC-CO2}|$ (Fig. 5). Higher water temperatures in La Alberca de los Espinos (by ~ 3 °C) could also partly contribute to a

smaller | \(\varepsilon_{OC-CO2}\)| in this lake (Sackett et al., 1965; Pardue et al., 1976; Descolas-Gros and Fontungne, 1990).

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Unlike $\delta^{13}C_{DIC}$, organic carbon isotope, signatures do not evolve linearly with the alkalinity/salinity gradient, suggesting other lake- and microbial-specific controls on these signatures. These controls include: diffusive or active uptake mechanisms, specific carbon fixation pathways, the fraction of intracellular inorganic carbon released out of the cells, cell size and geometry (Werne and Hollander, 2004 and references therein) and remineralization efficiency. Moreover, an increasing amount of isotopic data has evidenced a significant variability of the isotopic fractionation achieved by different purified RuBisCO enzymes ($\epsilon_{RuBisCO}$, Iñiguez et al., 2020), and even by a single RuBisCO form (Thomas et al., 2019). Thus, caution should be paid to the interpretation of the origin of small isotopic variations of the biomass in distinct environmental contexts because RuBisCO alone can be an important source of this variability (Thomas et al., 2019).

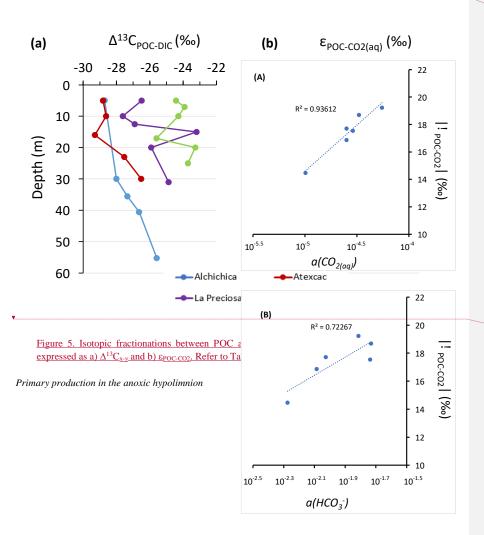


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Lastly, at 23 m in Atexcac and 17 m in La Alberca, we find a striking turbidity peak precisely where the redox potential and the concentration of dissolved Mn drops (Fig. 2). In Atexcac, the concentration in dissolved metals, such as Cu, Pb, or Co also drops at 23 m (Fig. S4). In La Alberca, a peak of particulate Mn concentration is detected at 15 m (Fig. 2; data unavailable for Atexcac). This is most likely explained by the precipitation of Mn mineral particles, where reduced bottom waters meet oxidative conditions prevailing in the upper waters. These oxidized Mn phases can be used as electron acceptors during chemoautotrophy (Havig et al., 2015; Knossow et al., 2015; Henkel et al., 2019; van Vliet et al., 2021). Even at a low particle density, such phases can catalyze abiotic oxidation of sulfide to sulfur compounds, which in turn can be used and further oxidized to sulfate by phototrophic or chemoautotrophic sulfur-oxidizing bacteria (van Vliet et al., 2021). This is also consistent with the small increase in [SO4²] observed at 23 m in Atexcac (Table S4).

In summary, combined POC and DIC data allowed us to recognize the most representative autotrophic metabolisms in the Mexican lakes. The upper water columns are all dominated by oxygenic photosynthesis. Lower in the water columns, anoxygenic photosynthesis and/or chemoautotrophy were found to have a noticeable impact on POC and DIC reservoirs in La Alberca and Atexcac only. Their activity was associated with metal elements cycling. More specifically in La Alberca, the anoxygenic phototrophs correspond to GSB.

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Figure 6.

Cross plots of DIC species activities *versus* absolute values of calculated C isotopic fractionations between POC and CO₂ at depths of peak oxygenic photosynthesis where data was available (5 and 30 m for Alchichica, 16 m for Atexcac, 10 and 12.5 m for La Preciosa and 7 m for La Alberca). (A) Dissolved CO_{2(aq)} activity and (B) bicarbonate activity as functions of $|\epsilon_{POC-CO2}|$ in $|\epsilon_{POC-CO2}|$ in |

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Aerobic respiration at the oxycline

At the oxycline of stratified water bodies, aerobic respiration of OM by heterotrophic organisms favors the transition from oxygenated upper layers to anoxic bottom waters. In the water column of the four lakes, $\Delta^{13}C_{POC}$. DIC (and EPOC-CO2) show increasing values in the hypolimnion, and especially below the chlorophyll a peaks (Figs. 2 and 5). This trend also correlates with increasing $\delta^{13}C_{POC}$, decreasing (C:N)_{POM} ratios as well as decreasing POC concentrations except in La Alberca (Figs. 3 and 4). Decreasing POC concentrations near the oxycline and redoxcline are consistent with the fact that part of the upper primary production is degraded deeper in the water columns and/or that there is less primary production in the anoxic bottom waters. Increased, $\delta^{13}C_{POC}$ in the hypolimnion of the lakes is consistent with heterotrophic activity and points out that POC at these depths could mainly record secondary production rather than being a residue of sinking degraded OM formed by primary production. Heterotrophic, bacteria preferentially grow on available 13C-enriched amino acids and sugars, thus becoming more enriched than their C source (Williams and Gordon, 1970; Hayes et al., 1989; Zohary et al., 1994; Briones et al., 1998; Lehmann et al., 2002; Jiao et al., 2010; Close and Henderson, 2020). The decrease in C:N ratios in the POM also reinforces this conclusion since secondary heterotrophic bacteria biomass generally have C:N between 4 and 5 (Lehmann et al., 2002), whereas residual degraded OM from primary producers would carry higher C:N signatures (van Mooy et al., 2002; Buchan et al., 2014). These latter signatures are not recorded by POM in the lower water columns of the lakes (Fig. 3).

5.2.2. Sinks of particulate organic carbon: respiration and sedimentation

The $\delta^{13}C_{DIC}$ signatures in La Preciosa and Alchichica are consistent with the mineralization of OM as they exhibit lower values below the oxycline than in surficial waters (Figs. 2 and 4). Similarly to what is observed in several other water bodies and notably stratified water columns such as the Black Sea (e.g. Fry et al., 1991), surface photosynthesis increases $\delta^{13}C_{DIC}$ by fixing light DIC, while respiration transfers light OC back to the DIC pool at depth. Such a decrease $\frac{in}{2}\delta^{13}C_{DIC}$ can also be seen in the oxycline of Lake La Alberca between 7 and 10 m.

Influence of methanogenesis in Lake La Alberca de los Espinos

La Alberca shows the least saline/alkaline water column and most peculiar geochemical depth profiles among the four lakes. Notably, its [DIC] and $\delta^{13}C_{DIC}$ (the lowest of the studied lakes) increase from the lower metalimnion to the hypolimnion, and further into the first cm of sediment porewaters, with $\delta^{13}C_{DIC}$ reaching almost 10 % (Figs. 3; 4). The calculated CO₂ partial pressure (P_{CO2}) increases downward from slightly less than $1x P_{CO2,atm}$ near the lake surface up to almost 40x at the bottom of the lake (Table S2).

While the increase of [POC] at depth may contribute to the observed $\delta^{13}C_{DIC}$ increase, by mass balance, it should also lower the [DIC] instead of increasing it. Similarly, the sinking of POC at depth followed by its remineralization into DIC cannot explain these observations since it would lower the $\delta^{13}C_{DIC}$ in the hypolimnion (Fig. 4). Overall, these observations require that a significant source of inorganic ^{13}C -rich carbon fuels the bottom waters of La Alberca de los Espinos. The source of heavy carbon most likely results from methanogenesis, which consumes organic carbon in the sediments and produces ^{13}C -depleted methane and ^{13}C -rich carbon dioxide diffusing upward in the water column (i.e. acetoclastic methanogenesis, dominant in lacustrine contexts, Whiticar

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et al., 1986). Methanogenesis, as an "alternative" OM remineralization pathway would be favored in La Alberca, because it is relatively rich in OM (notably with high [DOC], Havas et al., submitted), and depleted in SO42-(Wittkop et al., 2014; Birgel et al., 2015; Cadeau et al., 2020) compared with the three other Mexican lakes. Based on the δ^{13} Csoc and porewater δ^{13} Cbic, we can tentatively calculate the methane isotopic signature in La Alberca (see supplementary text S5). The <u>resulting</u> $\delta^{13}C_{CH4}$ in the first 10 cm of sediments is between -59 and -57 ‰, which is consistent with the range of isotopic composition of methane after biogenic methanogenesis (Whiticar et al., 1986).

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Upward diffusing methane may be either (i) partly lost from the lake's surface (i.e. escaping the system) by degassing or (ii) totally retained in the water column by complete oxidation (either abiotically by oxygenated surface waters or biologically by methanotrophic organisms). The oxidation of CH4 in the water column should lead to the formation of 13C-depleted carbon dioxide that would mix back with the lake DIC (and notably with heavy methanogenic CO2 produced at depth) and/or ¹³C-depleted biomass (as POC or SOC) if it occurs through, methanotrophy. Thus, the net effect of combined methanogenesis and methane oxidation is expected to (i) generate a $\delta^{13}C_{DIC}$ gradient from high to low values between the sediment porewaters and the oxycline as proposed elsewhere (Assayag et al., 2008; Wittkop et al., 2014) and (ii) progressively lower sedimentary $\delta^{13}C_{SOC}$ in the case of methanotrophy. Abiotic oxidation of methane by dioxygen is consistent with the observation, that $\delta^{13}C_{DIC}$ decreases from porewaters (~+10 ‰) to the oxycline (-4 ‰), reaching minimum values where dissolved-O2 starts to appear (Fig. 2). Microbial anaerobic oxidation of methane (AOM) could occur at the 17 m depth through Mnoxide reduction (Cai et al., 2021; Cheng et al., 2021) and possibly bacterial sulfate-reduction closer to the watersediment interface, as inferred for the surficial sediments of meromictic Lake Cadagno (Posth et al., 2017). Indeed, we observe a net increase of particulate Fe and S concentrations at a depth of 25 m and a peak of solid sulfide minerals in the surficial sediments (Fig. S5). However, $\delta^{13}C_{SOC}$ and $\delta^{13}C_{POC}$ are far from calculated $\delta^{13}C_{CH4}$, suggesting that AOM is not a major process in the bottom lake waters and surface sediments (Lehmann et al., 2004) and thus that methanotrophy is not the main CH₄ oxidation pathway in Lake La Alberca.

Alternatively, if some portion of the methane escaped oxidation and degassed out of the lake, $\delta^{13}C_{DIC}$ would likely be driven to extreme positive values with time (Gu et al., 2004; Hassan, 2014; Birgel et al., 2015; Cadeau et al., 2020). This is not consistent with the average $\delta^{13}C_{DIC}$ in La Alberca (~ -3 \(\frac{\psi}{20}\); Fig. 4), unless an additional counterbalancing source of DIC to this lake exists. This source of DIC could be volcanic CO2-degassing (see section 5.1.1). Such a contribution may maintain the lake's average δ¹³C_{total} close to a mantle isotopic signature and notably away from extreme positive values if CH4-escape dominated. It is also possible that volcanic CO2 degassing is coupled to methanogenesis by CO2 reduction in addition to the acetoclastic type described above.

Although volcanic CO2 could be an important source in the C mass balance of Lake La Alberca, we note that it cannot explain the very positive $\delta^{13}C_{DIC}$ in the sediment porewaters alone, thus bolstering the identification of methanogenesis. Importantly, this methane cycle is cryptic to the sediment record, as it is evidenced in the dissolved inorganic C phase, but not in the sedimentary organic matter or carbonates. This is a consequence of the lake's stratified nature, where the location of carbonate precipitation and methane production is decoupled.

Transfer of OM from the water column to the surficial sediments

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The OC content in the first 12 cm of the sediment cores from the four lakes ranges from 1 to 13 wt. % (Table S3). This appears to be relatively elevated considering the predominantly autochthonous nature, of OC and the, oligotrophic conditions in these lakes (Alcocer et al., 2014; Havas et al., submitted). In Lake Alchichica, the recent OC burial flux in the sediment was estimated to represent between 15 and 26 g.yr⁻¹.m⁻² (Alcocer et al., 2014). This is within the range of values observed for small lakes around the world (Mulholland and Elwood, 1982; Dean and Gorham, 1998; Mendonça et al., 2017), though most of them receive allochthonous OM inputs. Different factors can favor the preservation of OM including lower respiration and oxidation rates due to anoxic bottom waters and scarce benthic biota and/or high sedimentation rates (Alcocer et al., 2014). Anaerobic respiration clearly occurs in the four lakes to some extent, as detailed for La Alberca, and as seen in the surficial sediment data of the other lakes as well (decreasing δ13C_{DIC} in Alchichica, increasing C:N ratio in Atexcac and La Preciosa; Table S3). Nonetheless, the anoxic conditions prevailing in the hypolimnion most of the year are significantly more favorable to OM preservation than oxic conditions (Sobek et al., 2009; Kuntz et al., 2015). While the yearly mixing oxidizes most of the water column during the winter, it also generates a bloom of diatoms which fosters OM production (through shuttling up of bio essential nutrient such as N and Si) and development of anoxia (e.g. Adame et al., 2008). In Alchichica, the large size of some of the phytoplankton was also suggested to favor OM preservation (Adame et al., 2008; Ardiles et al., 2011). Because bacterial, sulfate reduction (BSR) is a major remineralization pathway in SO₄-rich environments (e.g. Jørgensen, 1982). the low sulfate content in La Alberca, probably favors the preservation of high TOC in the sediments. Even though, appreciable BSR rates may occur in this lake (see discussion above and Fig. S5), similarly to other sulfate-poor environments due to rapid S-cycling (e.g. Vuillemin et al., 2016; Friese et al., 2021). Again, a complete mass-balance of these lakes C fluxes will be required to estimate their net C emission or sequestration behavior.

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Although the nature and geochemical signatures of the OM that deposits in the sediments may vary throughout the year, it is interesting to infer from what part(s) of the water column surficial sedimentary OM comes during the stratified seasons. In the three lakes from the SOB, $\delta^{13}C_{SOC}$ and $(C:N)_{SOM}$ signatures of the surficial sedimentary. OM lie somewhere between POM signatures from the upper water column and from the hypolimnion (Figs. 3, 4). More precisely, in Alchichica, the most surficial $\delta^{13}C_{SOC}$ and $(C:N)_{SOM}$ signatures (-25.7 ‰ and 10.4, respectively) are much closer to values recorded in the upper water column (~ -26.5 ‰ and 10.5, respectively), implying that the upper oxygenic photosynthesis production is primarily recorded. It is consistent with previous studies suggesting that most of the phytoplankton biomass being exported is composed of diatoms (Ardiles et al., 2011). In Lake Atexcac, however, $\delta^{13}C_{SOC}$ and $(C:N)_{SOM}$ signatures (~ -26.8 ‰ and 8, respectively) are closer to values recorded in the hypolimnion (~ -26.5 ‰ and 6.5, respectively) suggesting that SOM records mostly the anaerobic primary production.

In La Alberca, surficial δ^{13} Csoc is markedly more negative (by ~ 2 to 3 ‰) than the deepest and shallowest water column values (Fig. 4), but close to what is recorded at the redoxcline depth of 17 m. However, the (C:N)_{SOM} values are much higher than what is measured anywhere in the water column, which is consistent with OM remineralization by sulfate-reduction and methanogenesis in the sediments of this lake. Therefore, OM biogeochemical signatures in the surficial sediments of La Alberca could be strongly influenced by early diagenesis occurring at the water-sediment interface – despite favorable conditions for OM preservation.

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methanogenesis/methanotrophy are recorded in the surficial sediments porewaters (notably seen through extremely positive $\delta^{13}C_{DIC}$) but not in the solid sediments that show neither very negative $\delta^{13}C_{SOC}$ nor positive $\delta^{13}C_{carbonates}$ in the first 10 cm.

Overall, this suggests that OM depositing at the bottom of these stratified lakes does, not always record geochemical signatures from the same layers of the water columns and can be modified by very early diagenesis. It does not necessarily record the signatures of primary production by oxygenic photosynthesis from the upper column. For example, in Lake Atexcac, sedimentary OM records primary production by anoxygenic photosynthesis, even though POC concentration is highest in the upper water column. This highlights the diversity of geochemical signatures that can stem from continental environments despite their geographical, geological, and climatic proximity. A deeper understanding of the OM transfer process from water column to sediment will require more detailed analyses and comparison of the different OM pigments and molecules and could have strong implications for the interpretation of the fossil record in deep anoxic time.

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6. CONCLUSIONS AND SUMMARY

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The carbon cycles of four stratified alkaline crater lakes were described and compared based on the concentration and isotopic compositions of DIC and POC in the water columns and surficial (~10 cm) sedimentary carbonates and organic carbon. Overall our study shows the wide diversity of geochemical signatures found in continental stratified environments despite similar geological and climatic contexts. We identify different regimes of C cycling in the four lakes due to different biogeochemical reactions related to slight environmental and ecological variations. In more detail, we show that:

- External abiotic factors, such as the hydrological regime and the inorganic C sources in the lakes, control their alkalinity and thus, the buffering capacity of their waters. In turn, these differences in buffering capacity constrain variations in pH along the stratified water columns as well as the inorganic C isotope. signatures recorded in the water columns and sediments of the lakes. The $\delta^{13}C_{carb}$ reflects the abiotic factors generating the alkalinity gradient, but it is poorly representative of biological processes in lakes with high alkalinity. The external environmental factors further impact the C mass balance of the lakes with probable consequences on their net C-emitting or -sequestering status.
- Based on POC and DIC concentrations and isotopic compositions, combined with physico-chemical parameters, we are able to identify the activity of oxygenic photosynthesis and aerobic respiration in the four lakes studied. Anoxygenic photosynthesis and/or chemoautotrophy are also evidenced in two of the lakes, but their POC and DIC signatures can be equivocal.
- Methanogenesis is evidenced in the surficial sediments of the OM-rich Lake La Alberca de los Espinos and influences the geochemical signatures lower in the water column. However, it is recorded only in analyses of porewater dissolved species, but not imprinted in the sedimentary archives (OM and
- The SOM geochemical signatures of these stratified lakes do not all record the same "biogeochemical layers" of the water column (e.g. anaerobic vs. aerobic metabolisms), and, in some cases, can be greatly modified by early diagenesis,

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2355 2356 2357 2358 2359	Author Contributions RH and CT designed the study in a project directed by PLG, KB and CT. CT, MI, DJ, DM, RT, PLG and KE collected the samples on the field. RH carried out the measurements for C data; DJ the physico-chemical paramete probe measurements and EM provided data for trace and major elements. RH and CT analyzed the data. RH wrote the manuscript with important contributions of all co-authors.
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2361 2362	Competing Interests The authors declare that they have no conflict of interest.
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2364	Disclaimer
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