



26 these sources into biogenic VOC emission models, which in turn requires extensive research
27 on these sources to understand the conditions and factors that influence VOC emissions. The
28 decomposition of leaf litter, accompanied by the release of VOCs, is a very complex process
29 that depends on a number of biological, chemical and physical environmental factors, but
30 little information is currently available on the role each plays. Equally limited is information
31 on the chemical composition and emission rates of VOCs from these sources.

32 The review focuses on the main gaps in our knowledge of the sources of biogenic VOCs
33 under the forest canopy, and we are confident that filling them will make a significant
34 contribution to solving such an important task as closing the global organic carbon budget.

35

36 **Keywords:** leaf litter; living ground cover of forests; decomposition; litter-decomposing
37 microorganisms; volatile organic compounds; chemical composition; emission rate

38

39 **1. Introduction**

40 Over the last four decades, the emission of non-methane volatile organic compounds
41 (VOCs) into the atmosphere has attracted considerable attention, mainly because their
42 presence in the air significantly affects chemical processes in the boundary layer. Terrestrial
43 living vegetation is the main source of atmospheric VOCs (Isidorov, 1990; Fehsenfeld et al.,
44 1992; Guenther et al., 1995; Navarro et al., 2014; Sindelarova et al., 2014). The majority of
45 biogenic VOCs are reactive in the troposphere and therefore have a greater impact on the
46 atmospheric chemistry than comparable amounts of organic pollutants from industrial and
47 motor vehicle emissions (Bell and Ellis, 2004). In the atmosphere, VOCs undergo gas-phase
48 oxidation (Atkinson and Arey, 2003) or photo-stimulated oxidation on the surface of solid
49 atmospheric aerosols (Isidorov, 1992; Isidorov et al., 1997). Their reactions with $\cdot\text{OH}$ radicals
50 and NO_x result in the production of secondary photooxidants, such as ozone, H_2O_2 , ROOH



51 and peroxyacetyl nitrate (PAN). Furthermore, the oxidation of biogenic VOCs initiates the
52 production of secondary atmospheric aerosols, which influence the radiative budget of the
53 troposphere (Makkonen et al., 2012; Kulmala et al., 2001, 2013; Faiola et al., 2014;
54 Kourtchev et al., 2016; Lee et al., 2019). An increase in the tropospheric ozone concentration
55 can also affect the climate by perturbing the Earth's radiation budget as O₃ is the third-most
56 important greenhouse gas (Intergovernmental Panel on Climate Change, IPCC, 2013; Kim et
57 al., 2015).

58 Exposure to elevated surface ozone concentration has detrimental impacts on human
59 health, natural vegetation and crop yield (Chuwah et al., 2015; Zhou et al., 2017; De Marco et
60 al., 2019). In particular, ground-level ozone pollution has become one of the top issues in
61 China and India. According to Feng et al. (2015), throughout China current and future O₃
62 levels will induce a wheat yield loss of 6.4–14.9% and 14.8–23.0%, respectively. Moreover,
63 economic losses from wheat and rice in India attributable to ozone are estimated at \$6.5
64 billion per year (Sharma et al., 2019). For these reasons, tropospheric ozone production,
65 temporal variations in its concentrations over different regions and examinations of future
66 near-surface air quality changes have been the subjects of numerous recent investigations and
67 publications (Vingarzan, 2004; Curci et al., 2009; Derwent et al., 2015; Dolwick et al., 2015;
68 Gaudel et al., 2015; Kim et al., 2015; Santurtún et al., 2015; Zong et al., 2018; Masiol et al.,
69 2019; Strode et al., 2019; Archibald et al., 2020; Zhang et al., 2021).

70 To examine the contribution of biogenic organic volatiles to ozone and other
71 secondary pollutant formations, various chemistry transport models have been developed,
72 with VOC and NO_x emissions as the input data. Accurate and reliable emission inventories
73 are essential for good model performance; any underestimation of VOCs in the emissions
74 inventory could cause an underestimation of ozone levels by photochemical models (Bell and
75 Ellis, 2004; Mogensen et al., 2015; Zhou et al., 2017). However, it is well known that the



76 estimation of biogenic VOC emissions is still characterised by considerable uncertainty, with
77 a factor of 3–5 for key species such as isoprene (Curci et al, 2009), although research in this
78 area has been continuously conducted for many years (Guenther et al., 2006; Karl et al.,
79 2009). This uncertainty can be partially connected with systematic errors in the VOC budget,
80 e.g. inadequate emission factors or omission of significant biological sources. This applies
81 both to VOC emissions in general and to individual compounds such as methyl chloride: for
82 instance, Bahlmann et al. (2019) recently presented evidence of the existence of a 'lost' source
83 of this important component of the atmosphere. Recent work on the determination of the total
84 reactivity of hydroxyl radicals, TOHRE in different types of forests also indicate the existence
85 of unaccounted for sinks of this radical (Di Carlo et al., 2004; Sinha et al., 2010; Nölscher et
86 al., 2013; Yang et al., 2016; Praplan et al., 2020), that is, the presence of unaccounted sources
87 of reactive biogenic VOCs in the atmosphere. This implies the need to identify and
88 quantitatively characterize sources of unknown reactive VOCs in the atmosphere (Hellén et
89 al., 2021).

90 Emissions from leaf litter and non-arboreal vegetation under the forest canopy – such
91 as mosses, lichens, grasses and small shrubs as well as the dead remains of grass in vast
92 steppes (steppe mat), prairies and savannahs – belong to this category of 'lost' or 'omitted'
93 natural VOC sources (Isidorov, 1992; 1994; Isidorov et al., 1994; Tang et al., 2019). The
94 potential importance of this source of biogenic VOCs evidently results from its gigantic scale:
95 for land ecosystems, the litter mass is estimated to be 6×10^{16} g C (Bolin, 1983; Zlotin and
96 Bazilevich, 1993) and the global estimate of litter production is in the range $(9-10) \times 10^{16}$ g
97 dry matter (Matthews, 1997). According to Liu et al. (2003) the annual leaf litter production
98 in the boreal forests of Eurasia is $(1.49 \pm 0.61) \cdot 10^{15}$ g·yr⁻¹, with a total aboveground litter
99 production of $(2.07 \pm 0.85) \cdot 10^{15}$ g·yr⁻¹. Depending on different factors, the litter layer can
100 develop to significant thicknesses, exemplified by the ~40 mm reported from 20 year



101 unburned *Eucalyptus marginata* forest in Western Australia or the ~60 mm reported under
102 *Pinus pinaster* in France. For five types of forest in seven southern states of the USA, it was
103 reported an average litter depth of ~37 mm. Larger litter thicknesses have also been reported,
104 including a mean of 86 mm under *Cryptomeria japonica* in Japan (Dunkerley, 2015 and ref.
105 in it).

106 Dead plant material contains substantial amounts of volatile organic compounds. A
107 recent study of the chemical composition of green, senescent leaves and partially decomposed
108 silver birch (*Betula pendula*) leaves in southern Finland showed that although the
109 concentration of many secondary metabolites is significantly reduced, almost all of them are
110 found in both senescent leaves and fresh litter (Paaso et al., 2017). Furthermore, Wood et al.
111 (1995) did not detect statistically significant differences in monoterpene yields in green,
112 yellow or newly fallen brown leaves of the hardwood tree *Umbellularia californica* and
113 concluded that there is a lack of catabolism of monoterpenes prior to leaf senescence. This
114 further suggests that senescing leaves enter the soil environment with essentially the same
115 monoterpene yields that they had on the tree. However, such a situation is not common: in
116 one study, the total monoterpene content was higher in the fresh needles of *Pinus monophylla*
117 ($5.6 \pm 2.2 \text{ mg}\cdot\text{g}^{-1}$ extracted air dry weight, d.w.) than in senescent needles ($3.6 \pm 1.8 \text{ mg}\cdot\text{g}^{-1}$
118 d.w.), regardless of whether they were still attached to the tree or formed the freshest layer of
119 the understorey litter (Wilt et al., 1993). In another study, samples of Scots pine (*Pinus*
120 *sylvestris* L.) needles and needle litters collected in low-productive forests in eastern Siberia
121 contained, respectively, 7.9 ± 1.2 and $3.9 \pm 0.2 \text{ mg}\cdot\text{g}^{-1}$ (d.w.) of essential oils, mainly
122 consisting of volatile monoterpene hydrocarbons (Stepen' et al., 1987). About 15% of the total
123 store of essential oil in these forests is contained in litter, with a biomass of 12–15 t·ha⁻¹.
124 According to Klemmedson et al. (1990), litter needle biomass comprises 20% of the live
125 foliar biomass in Canadian ponderosa pine forest. More productive forests of the Northern



126 Hemisphere can accumulate enormous amounts of 'dead' organic material. For example, total
127 litter mass in a 100-year-old maritime pine (*P. pinaster*) forest in southern France was
128 $43.1 \pm 12.0 \text{ t}\cdot\text{ha}^{-1}$, and the depth of the needle compartment of the floor averaged 70 mm (Kurz
129 et al., 2000). The chemical composition of the litter was not determined by the authors, but it
130 can be supposed that the fallen needles in this maritime pine forest contain considerable
131 amounts of terpenes. Under the influence of external factors (both abiogenic and biogenic),
132 decomposition of the tissues and the cell structures of the needles takes place, resulting in the
133 formation of volatile compounds and their evolution into the gas phase.

134 The first authors of this review encountered this phenomenon in 1976 while studying
135 the composition of VOCs in the atmosphere of a city (Leningrad, former USSR), using gas
136 chromatography-mass spectrometry (GC-MS). Air samples were collected in a small public
137 garden at the end of October, i.e., when the vegetation period had finished. For this reason, it
138 was an absolute surprise to identify α -pinene, camphene and limonene in the air samples
139 (their peaks were among the more intensive peaks on the chromatograms) and six other less
140 abundant monoterpene hydrocarbons (Ioffe et al., 1977). In later investigations, it was
141 determined that fresh leaf litter of black poplar (*Populus nigra*) was the source of terpenes at
142 this sampling site. This was the second surprise, because the author's 1976 summer GC-MS
143 investigations of the volatiles emitted by the living foliage of this plant had demonstrated the
144 intensive evolution of isoprene into the gas phase, but the complete absence of any terpene
145 hydrocarbons among other volatiles. However, the author's attention was drawn to leaf litter
146 as a source of atmospheric VOCs, as reported by Zimmerman et al. (1978), who were the first
147 to pay attention to leaf litter and report the magnitude of the VOC release rate from this
148 source. According to their estimation, this averages $162 \mu\text{g m}^{-2}\cdot\text{h}^{-1}$ at 30°C . In spite of this
149 publication being cited more than 300 times (i.e., well known to the atmospheric chemistry
150 community), this natural source of reactive VOCs has remained unaccounted for and



151 unquantified (it was 'a hitherto unrecognised atmospheric source', Warneke et al., 1999). The
152 same is true for another component of forest ecosystems, namely living soil cover.

153 The purpose of this review is to draw the closer attention of atmospheric chemists to
154 the VOC emissions from these important components of terrestrial natural ecosystems. To our
155 mind, careful estimation of steppe- and forest floor-derived VOCs will help improve the
156 reliability of estimates of biogenic emissions so that they can be incorporated into chemistry
157 transport models as well as improve our understanding of land–atmosphere exchanges and
158 interactions.

159 In this review, we focus only on non-methane VOC emissions from boreal and
160 temperate natural ecosystems of the Northern Hemisphere. The vegetation of boreal and sub-
161 boreal forests (whose area represents approximately $16,780 \cdot 10^3 \text{ km}^2$ or about 11.2% of the
162 terrestrial area of our globe) accumulates a substantial proportion of the planet's living matter:
163 about $700 \cdot 10^6 \text{ t (d.w.)}$. Biomass in different types of forests varies from 10,000 to 40,000
164 $\text{t} \cdot \text{km}^{-2}$. Biomass production in coniferous forests in the north taiga zone is equal to ca. 450
165 $\text{t} \cdot \text{km}^{-2}$, whereas in the mixed and broad-leaved forests in the south taiga zone it reaches 800
166 and 900 $\text{t} \cdot \text{km}^{-2}$, respectively (Dobrovolsky, 1998).

167 Dying parts of plants fall off and form litter, the quantity of which is not proportional
168 to vegetation biomass. For example, south taiga forests with considerable biomass (more than
169 $30,000 \text{ t} \cdot \text{km}^{-2} \text{ d.w.}$) deposit approximately $500 \text{ t} \cdot \text{km}^{-2} \text{ (d.w.)}$ of litter in the soil yearly,
170 whereas meadow steppe vegetation with significantly lower amounts of biomass (ca. $2,500$
171 $\text{t} \cdot \text{km}^{-2}$) forms nearly $1,500 \text{ t} \cdot \text{km}^{-2}$ of litter (Dobrovolsky, 1998).

172 Leaf litter is defined here as *the uppermost layer of the remains of dead leaves on the*
173 *soil surface of a forest, meadow or steppe, decomposed to varying degrees.* Living soil cover
174 (LSC), as we understand it, refers to *forest floor vegetation formed by understorey dwarf*
175 *shrubs, mosses, lichens, herbs and ferns.* As a rule, in boreal and temperate regions, litterfall



176 is associated with the coming of cold, autumnal weather. However, we should also take into
177 account the transitory but intensive spring falling of bud scales, petals and flowers. This
178 material is characterised by significantly higher N and P levels but has lower concentrations
179 of lignin and cellulose compared to leaf litter (Wang et al., 2016). For this reason, fallen
180 flowers decompose much faster (within 1–1.5 months) than leaves. Many bud scales (for
181 example, protecting buds of different poplar species) are covered with a resinous or gummy
182 substance, mainly consisting of volatile terpenoids. Defoliation of deciduous trees can also
183 occur during the summer, as in July 2015 in the north-eastern region of Poland and in the
184 summer months of 2018 in Germany. Intensive summer litterfall (more than 15% of the
185 foliage store) is the result of an acute soil water deficit. Besides, many evergreen coniferous
186 trees shed their leaves all year round.

187 The following chapters will give an overview of the main processes governing litter
188 decomposition and litter-derived VOC emissions, their chemical composition and the
189 participation of litter-destroying microorganisms in their formation, available information
190 regarding the emission rates of separate VOCs. The Supplementary material to this article
191 provides data on litter and living soil biomass stocks in boreal and temperate forests of the
192 Northern Hemisphere that are useful for estimating the magnitude of VOC emissions.

193

194 **II. Biotic and abiotic controls of leaf litter decomposition**

195 The evolution of volatile organic compounds from dead plant tissues mostly takes place after
196 their more or less advanced decomposition. Two groups of factors controlling this process in
197 terrestrial ecosystems can be distinguished: abiotic and biotic ones (Swift et al., 1979; Berg,
198 2014; Krishna and Mohan, 2017). The latter group includes complex interactions between
199 litter-inhabiting saprotrophic (invertebrate and vertebrate) animals and microorganisms.



200 Leaf-eating insects can influence decomposition processes by modifying the quality
201 of litter. Many studies have demonstrated that soil fauna such as isopods, gastropods,
202 arthropods and other detritophages significantly affect the decomposition rate (Coulis et al.,
203 2013; David, 2014; Frouz et al., 2015; Hassall et al., 1987; Kozlov et al., 2016; Lukowski et
204 al., 2021; Pokhylenko et al., 2020; Wang et al., 2015; Zimmer et al., 2003). Macroarthropods
205 are believed to enhance the decomposition rate of leaf litter and increase VOC emissions both
206 directly and indirectly. Their direct contribution includes the fragmenting of dead plant
207 material and increasing the surface area available for microbial colonisation, while their
208 indirect contribution consists in ingesting saprotrophic litter-colonising microbiota and
209 egesting their faeces in surface-increased substrate, thereby enhancing microbial activity
210 (Zimmer et al., 2003).

211 Further decomposition of crushed litter is performed by microorganisms, bacteria,
212 actinomycetes and fungi. The last constitute a major component of soil biota and represent the
213 primary decomposers in most environments. Indeed, fungi produce the enzymes (including
214 extracellular enzymes) necessary to decompose the lignocellulose matrix of leaf litter and
215 wood (Cox et al., 2001; Fioretto et al, 2007). Participating in the decomposition of organic
216 matter of plant origin, these saprotrophic organisms produce their own, inherent metabolites;
217 as part of these secondary products are volatile organic compounds, they can be transferred
218 into the gas phase (Svendsen et al., 2018).

219 These microbial-zoological interactions stimulate intense abiotic processes such as
220 the evaporation of volatiles and the leaching of water-soluble compounds. Another abiotic
221 process aiding the evolution of volatiles from plant tissues is their mechanical destruction,
222 brought about by the freezing–thawing of dead organic material (Porchikalov and Karelin,
223 2015). In some studies thermo- and photochemical reactions caused by direct solar radiation



224 are described as abiotic destruction processes; their role is much more pronounced in forest-
225 steppe and steppe landscapes than under the forest canopy.

226 The relative proportion of biotic and abiotic processes varies greatly, primarily
227 depending on climatic conditions, yet there is only scarce corresponding information available
228 in the literature (Kravchenko et al., 2019). According to Bazilevich et al. (1993), leaching
229 predominates among other abiotic factors in humid north taiga forests. To assess the role of
230 various natural factors in the decomposition of plant fall, these authors conducted field
231 experiments in different regions of Russia. The general conclusion was that there are two
232 principal factors limiting the intensity of organic matter destruction: in tundra-forest and north
233 taiga zones the key factor is a lack of heat, while in semi-arid steppe ecosystems it is a lack of
234 moisture. These authors' experiments, while supporting the phenomenon of the biological
235 decomposition of dead organic material, also highlight the important role of abiotic factors in
236 decomposition. The authors conducted experiments in two neighbouring ecosystems in the
237 central European part of Russia, finding that the proportions of biotic and abiotic factors differ
238 in forest and meadow steppe (Table 1).

239 The more pronounced distinction was the increased share of litter decomposition
240 under the impact of solar radiation in the open landscape. Litter-bag experiments with five
241 degrees of solar light intensity (controlled by special screens) showed that the decomposition
242 rate of dead aspen and grass leaves declines gradually with decreases in solar radiation
243 intensity and short wavelength radiation. Bosco et al. (2016) found additive effects of
244 ultraviolet (UV) radiation and soil water on litter mass loss. Recent laboratory experiments
245 have shown that the photodegradation of plant litter driven by UV radiation is important in
246 influencing mass loss, nutrient release and the carbon balance in a broad range of terrestrial
247 ecosystems (Austin et al., 2016; Li et al., 2016). This applies especially to dryland



248 ecosystems, which cover nearly 45% of the Earth's land area and account for large
249 proportions of terrestrial net primary production (Hewins et al., 2019).

250 Karl et al. (2005) did not see a significant influence of photosynthetically active
251 radiation (PAR) on emissions of VOCs from senescing grass crops. However, shortwave
252 radiation can stimulate different types of photochemical processes. UV-induced C₂–C₅
253 hydrocarbon emissions have been measured by Derendorp et al. (2011a,b) from the leaf litter
254 of some plant species, with the influence of experimental conditions determined. In the
255 absence of oxygen, no emissions of C₂–C₅ hydrocarbons were observed. When the litter was
256 placed in humid air, the emission rates approximately tripled compared with the emissions
257 from leaf litter in dry air. According to the authors, UV-induced hydrocarbon emissions from
258 leaf litter may have a small influence on atmospheric chemistry at the local scale, but do not
259 contribute significantly to their global budgets. However, it seems that this conclusion is
260 premature, because our knowledge about the influence of shortwave solar radiation on the
261 decomposition of dead plant material and VOC emissions is very limited.

262 Further field and laboratory measurements are needed to investigate the impact of
263 UV-B radiation (280–310 nm), together with UV-A (315–400 nm) and visible radiation
264 (400–700 nm). It is well worth remembering that senescent leaf matter photoproduces carbon
265 monoxide, which has an important effect on atmospheric composition (Tarr et al., 1995;
266 Derendorp et al., 2011c). In Tarr et al.'s (1995) study, leaf litter photoproduced CO at rates
267 that ranged from 1.3 to 5.4 times higher per unit area than did living leaves. According to
268 Bornman et al. (2015), UV radiation can also stimulate VOC emissions from both living plant
269 and leaf litter, although the magnitude, rates and spatial patterns of these (supposed)
270 emissions remain unknown at present.

271 On the other hand, there are experimental confirmations of the *de novo* synthesis of
272 partially oxidised volatile organic compounds in dead plant material under the action of the



273 heat component of solar radiation. Warneke et al. (1999) conducted experiments involving the
274 alternate 'roasting' (up to 60–100 °C) and wetting of dead beech leaves, demonstrating
275 multiple increases in the emission rates of oxidised VOCs such as methanol, acetone,
276 acetaldehyde, ethanol and butan-2-one. Moreover, these authors showed that even at room
277 temperature, considerable amounts of these volatiles are released over a period of days and
278 weeks. According to the authors' previous estimation, the decay of 1 g of leaf matter produces
279 at least 100 µg of acetone and 300–500 µg of methanol. Global annual emissions of acetone
280 and methanol from dead leaf matter may be at least 6–8 Tg and 18–40 Tg, respectively.
281 Considering the temperature dependence of the emission rates of the VOCs, Warneke et al.
282 (1999) predicted the largest emission rates to be in the late spring and early autumn.

283 Evidently, similar thermochemical processes can take place more intensively in
284 felted dead grass litter in semi-arid steppe ecosystems as well as on the outskirts of forests and
285 in open-canopy forests. The summer soil surface temperature can reach 60–80 °C in these
286 landscapes.

287 Paradoxically, the abiotic strengthening of VOC emissions can also be stimulated by the
288 process exactly opposite to 'roasting', namely freezing. Frequent freezing and thawing events,
289 especially characteristic in the last two decades for high-latitude regions, can positively
290 influence the decomposition rate (Porchikalov and Karelin, 2015; Jiang et al., 2016). Some
291 authors have presented the idea that a source of reactive VOCs are the wounding processes
292 that occur after leaves undergo freeze–thaw damage. Fukuj and Doskey (1998) measured
293 VOC emissions from a grassland site before and after frost events and reported morning frost-
294 enhanced emissions of acetaldehyde, acetone, methanol and ethanol. Fall et al. (2001)
295 detected the emissions of low-molecular-weight oxygenated VOCs during leaf drying,
296 senescence and following freeze–thaw damage. Leaves of beech, larch, clover, bluegrass and
297 fern wounded in this way released 1-penten-3-ol, 2- and 3-methylbutanal, 2(*E*)-penten-1-ol, 1-



298 penten-3-one, hexanal and some other minor compounds (supposedly, isomeric pentenals)
299 into the gas phase. Moreover, freeze–thawed C₅ and C₆ compounds were detected by GC-
300 PTR-MS and PTR-MS techniques in mid-November 1999 at the Sonnblick Observatory
301 (Austrian Alps, 3106 m a.s.l.). After a hard freeze had occurred in the central alpine valleys of
302 Austria, unprecedented levels of highly reactive C₅ and C₆ biogenic VOCs were registered in
303 ambient air transported from these valleys: isoprene, 2- and 3-methylbutanal, 1-penten-3-ol,
304 1-penten-3-one, hexanal, 3(*Z*)- and 2(*E*)-hexenal. These were the same VOCs seen in the
305 freeze–thaw wounding experiments. Fall et al. (2001) speculated that VOCs were released
306 from freeze-damaged local vegetation, larch and fir trees. This seems doubtful because living
307 frost-hardy trees are quite well protected from such kinds of wounding. However, saturated by
308 rainwater, dead plant matter (for example fallen larch and fir needles) is deprived of this
309 biological protection. After autumn rains, the litter can remain damp for a long time (Wood et
310 al., 1995). In frosty weather, the crystallisation of surplus water takes place, damaging cell
311 and tissue structures. Besides, freezing has a destructive effect on the resin canals in the litter
312 (Wu et al., 2010). All these forms of damage can increase the release and evolution of
313 volatiles.

314 At present, it is not possible to determine how various time or spatial features affect the
315 proportions of abiotic and biogenic processes in the emissions of VOCs from dead plant
316 material. According to some authors (Zhang et al., 2008; Gray et al., 2010), abiotic factors
317 play an additional role compared to litter microbial degradation and VOC emissions on a
318 global scale. The predominance of the microbiological production of volatile organics has
319 been demonstrated by Leff and Fierer (2008), who discovered a correlation between
320 respiration, microbial biomass and level of VOC emissions. However, the information
321 presented in this chapter allows us to conclude that substantial quantities of low molecular
322 weight C₅–C₆ VOCs can be formed as the result of the abiotic decomposition of plant litter.



323 Due to its importance in atmospheric chemistry, it can be concluded that much more research
324 should be devoted to this natural source of atmospheric VOCs. In particular, UV radiation,
325 heat and frost must be regarded as important triggers for seasonal emissions of these VOCs by
326 plant litter.

327 The second conclusion is related to the qualitative composition of leaf litter
328 emissions: one can expect among these emissions not only plant-derived compounds, but also
329 VOCs of microbial and animal origin.

330

331 **III. Qualitative composition of plant litter and LSC emissions**

332

333 Over the last few decades, the chemical composition of volatile emissions from some
334 thousands of plant species – including practically all forest-forming trees of the Northern
335 Hemisphere – has been investigated. For many hundreds of species, the emission rates of
336 principal biogenic VOCs, isoprenes and monoterpenes have also been determined
337 (Kesselmeier and Staudt, 1999). The available information on the chemical composition of
338 litter-derived compounds is much poorer. This chapter provides an overview of the current
339 knowledge, which is based on only a few publications.

340 Table 2 contains a list of 164 volatile organic compounds detected in emissions from the
341 leaf litter of 30 plants, including 25 arboreal species. The largest group is made up of
342 hydrocarbons (79 compounds), including mono- and sesquiterpenes (63 compounds), which
343 are more active in atmospheric processes. Isoprene, ethanol, acetaldehyde, acetone and
344 methanol belong to the more frequently detected litter-derived components. With the
345 exception of isoprene, these short-chain volatiles have relatively long atmospheric lifetimes
346 and reach concentrations in the ppb range.



347 It is plausible that α -pinene is widespread within litter-derived VOCs; however, the
348 accurate identification of this important monoterpene is impossible in certain cases. In some
349 of the investigations cited in Table 2, instead of 'traditional' GC-MS analyses, a relatively
350 novel and, in many aspects, convenient proton transfer reaction mass spectrometry (PTR-MS)
351 technique was used (Gray et al., 2010; Ramirez et al., 2010; Stoy et al., 2021). This method
352 has the undoubted advantage of being a tool for the simultaneous online registration of VOC
353 emissions. However, PTR-MS has limited identification possibilities as only a single
354 protonated ion, with a molecular weight corresponding to that of the neutral analyte molecule
355 plus 1 amu, $[M+1]^+$, is used as the sole analytical parameter. This hinders the precise
356 identification of isomers and the distinguishing of isobaric compounds, for example $C_{10}H_{16}$
357 monoterpenes and the numerous aromatic $C_9H_{12}O$ compounds, nor-terpenoids and furans with
358 a molecular weight of 136 amu. For this reason, in some cases, the identification of
359 monoterpenes (Gray et al., 2010; Ramirez et al., 2010; Greenberg et al., 2012) or isomeric
360 pentenols, methyl butanols and pentenals (Fall et al., 2001) has to be recognised as tentative.
361 The speciation (i.e., determination of the exact chemical structure) of VOCs is highly
362 important for atmospheric chemistry because even closely related compounds differ in their
363 reactivity (Atkinson and Arey, 2003). In recent years, a more advanced version of the method,
364 a Proton Transfer Reaction Time-of-Flight Mass Spectrometer (PTR-TOF-MS), has become
365 more frequently used, which makes it possible to determine the mass of a registered ion with
366 an accuracy of 0.01 amu (Viros et al., 2020; Crocker 2021). This significantly increased the
367 identification capability of the method, however, the much more time- and work-consuming
368 GC-MS method (which uses two independent analytical parameters for identification, full
369 mass spectrum and chromatographic retention index) maintains its significance.

370 Another component of forest floor cover worthy of attention as a source of atmospheric
371 VOCs is non-arboreal vegetation. Small shrubs, grasses, lichens and mosses considerably



372 contribute to the overall green biomass of many types of Northern Hemisphere forests
373 (Isidorov, 1994; Isidorov et al., 1994; Jagodzinski et al., 2016) and play an important role in
374 the photosynthetic production of boreal forests (Kolari et al., 2006). These plants are adapted
375 to low photosynthetically active radiation (PAR) and begin photosynthesis as early as at a
376 PAR level above $3 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ (Morén and Lindroth, 2000). In boreal forests, different
377 feathermoss, peatmoss, and foliose lichen species often form a continuous layer on the forest
378 floor (Hageman and Moroni, 2015). Moreover, a significant fraction of the whole Earth's
379 land surface is dominated by bryophytes (Lindo and Gonzalez, 2010). Hence, mosses, lichens
380 and other forest floor plants can make up a principal part of total green biomass and VOC
381 emissions in forests, at least in subarctic and temperate regions of the Northern Hemisphere
382 (Isidorov, 1992; Isidorov et al., 1999; Mäki et al., 2017).

383 According to investigations performed in North European Russia, the LSC biomass in
384 20-year-old pine forests is (on average) $1,200\text{--}2,500 \text{ t}\cdot\text{km}^{-2}$, increasing to $7,600\text{--}8,100 \text{ t}\cdot\text{km}^{-2}$
385 in mature forests; at the same time interval, pine needle biomass decreased from $5,600\text{--}6,700$
386 $\text{t}\cdot\text{km}^{-2}$ to $3,200\text{--}4,700 \text{ t}\cdot\text{km}^{-2}$ (Zyabchenko, 1984). The LSC composition of young (20–40-
387 year-old) pine forests in northern Russia is, as a rule, is dominated by grasses and ferns
388 (*Melampyrum sylvaticum*, *M. pratense*, *Maianthemum bifolium*, *Avenella flexuosa*, *Dryopteris*
389 *flix-max*, etc.) and small shrubs (*Vaccinium vitis-idaea*, *V. myrtillus*, *Calluna vulgaris*, *Ledum*
390 *palustre*, *Empetrum nigrum*, *Erica* sp., etc.), but in the oldest forests, they are replaced by
391 green mosses and lichens (*Pleurozium schreberi*, *Politrichum commune*, *Licopodium*
392 *clavatum*, *Cetraria islandica*, *Sphagnum* sp., etc.). The share of LSC in the current accretion
393 of biomass in pine forests reaches 30% in north taiga and 50% in middle taiga zones
394 (Zyabchenko, 1984). On a level with arboreal vegetation, LSC forms a litter mass which can
395 reach $1.8\text{--}2.3 \text{ t}\cdot\text{ha}^{-1}$. In spite of its importance as a component of forest biomass, LSC has not
396 been studied sufficiently as a source of atmospheric VOCs. Table 3 presents a list of volatile



397 organics registered in gas emissions from a limited number of these plants. It contains 68 C₂-
398 C₁₄ compounds belonging to different classes, mainly hydrocarbons.

399

400 **IV. VOCs from litter-decomposing microbes**

401 Not only the chemical composition of leaf litter-derived VOCs but also how it changes during
402 litter decomposition is of great interest. Such changes have been registered in a laboratory via
403 the use of head-space solid phase microextraction (HS-SPME) combined with GC-MS
404 (Isidorov et al., 2003) and since in long-term litterbag field experiments (Isidorov et al.,
405 2010). This study determined that during the decomposition of Scots pine (*Pinus sylvestris*)
406 and Norway spruce (*Picea abies*) litter, substantial changes in the number of compounds
407 emitted into the gas phase take place (Fig. 1).

408 Over the entire observation period, mono- and sesquiterpene hydrocarbons were the
409 main components of volatile emissions, but the ratio of these VOC groups varied greatly (Fig.
410 2). During decomposition, new compounds appeared in the composition of VOCs, mainly
411 represented by monoterpenoids, the oxidation products of α - and β -pinene (pinocarvone,
412 pinocampon, myrtenal, myrtenol, *cis*-3-pinanone, verbenone) and aliphatic alcohols,
413 carbonyls and esters (Isidorov et al., 2010). Kainulainen and Hopolainen (2002) have also
414 reported the emission of verbenone and verbenol in decomposing pine needle litter. Most of
415 these 'additional' compounds were either not detected in the volatiles emitted by living pine
416 and spruce needles and in their essential oils or were present as minor and trace components.
417 For example, substantial quantities of acetone, butan-2-one, hexan-2-one, hexan-3-one, octan-
418 3-one and octan-4-one, as well as C₂-C₅ aliphatic alcohols and C₅-C₉ aldehydes, were only
419 registered in volatile emissions from pine and spruce needle litter after an experimental period
420 of four weeks. It has been proposed that all these volatiles are secondary products of litter-
421 destroying microorganisms (Isidorov et al., 2003). Therefore, the observed emission of VOCs



422 from leaf litter are the blend of volatiles stored in dead plant tissues, such as monoterpene
423 hydrocarbons and metabolites newly enzymatically synthesised by saprotrophic microbial
424 communities. In particular, C₃–C₈ carbonyl compounds, lower alcohols and their esters are the
425 main products of the enzymatic oxidative decomposition of unsaturated aliphatic acids
426 (Wurzenberg and Grosch, 1984).

427 It has long been known that diverse microorganisms grown on various nutritive artificial
428 media emit a broad range of VOCs into the gas phase (Norrman, 1969; Babich and Stotzky,
429 1974; Sunesson et al., 1995; Nilsson et al., 1996). Although the most labile primary and
430 secondary metabolites, such as mono- and oligosaccharides, amino acids and other organic
431 acids, are withdrawn by trees during leaf senescence, falling foliage still contains considerable
432 amounts of both volatile and non-volatile but easily degradable substances that can be
433 substrates for microorganisms (Tian et al., 2000; Isidorov et al., 2005; 2010). In Gray et al.'s
434 (2010) study, the cell-soluble fraction of the litter from 12 plant species was in the range of
435 28.9–74.6% of the litter carbon fraction. This material and more stable biopolymers
436 (cellulose, hemicelluloses, lignocellulose and lignin) can be used by various bacteria and
437 fungi. It can be assumed that these groups of microorganisms participate in the assimilation of
438 diverse substrates during the different stages of litter destruction; moreover, the composition
439 of microbial VOCs is species-specific. It is known that the relationship between bacteria and
440 fungi is characterized by antagonism. In most of the fungal cultures studied to date, the
441 biosynthesis of metabolites with bactericidal properties has been found, while many types of
442 bacteria demonstrate fungicidal effects. These competitive relationships can manifest
443 themselves in the progressive decomposition of litter and determine the composition of VOCs
444 of microbial origin.

445 During leaf litter decomposition, microbial succession takes place, defined as *a directed*
446 *change in the relative abundance and spatial pattern of species comprising communities*



447 (Frankland, 1998). Already senescent intact leaves are colonised by the associations of plant
448 pathogen fungi from genera such as *Venturia*, *Mycosphaerella* and *Tubakia* (Frankland,
449 1998). Undecomposed fallen leaves act as a carbon source for these 'weak pathogens' and
450 primary saprotrophs, belonging to bacteria and/or fungi (Bonanomi et al., 2017). After use by
451 the primary colonisers of simple sugars, oligosaccharides, amino acids and other easy-
452 assimilating, low-molecular-weight substances, litter is naturally colonised by other microbes,
453 fungi and cellulolytic bacteria (López-Mondéjar et al., 2016; Starke et al., 2016; Shao et al.,
454 2017; Štursová et al., 2020; Tennakoon et al., 2021). It is believed that more recalcitrant
455 lignified hemicelluloses and lignin are destroyed only by basidiomycetes, fungi capable of
456 producing extracellular lignolytic enzymes that catalyse the decomposition of acid-
457 unhydrolysable residue (Hobbie et al., 2012; Kirker et al., 2020). Individual litter-destroying
458 fungi show different preferences with respect to litter type and duration of occurrence
459 (Štursová et al., 2020). The successional nature of fungal decomposition of the herbaceous
460 litter of the Longleaf Pine savanna ecosystem was noted by Lodato et al. (2021). Sequencing
461 of fungi isolated by the authors from the litter of *Schizachyrium scoparium* and *Schizachyrium*
462 *tenerum* revealed the participation in the decomposition of several phylotypes, among which
463 representatives of Ascomycota (*Dothideomycetes*) and Basidiomycota (*Agaricomycetes*)
464 prevailed.

465 However, some groups of bacteria, which also have the ability to synthesize enzymes
466 suitable for these purposes, may be involved in the decomposition of recalcitrant biopolymers.
467 In particular, high lignolytic activity has been found in *Staphylococcus saprophyticus*,
468 *Pseudomonas aeruginosa* and certain bacteria from the genus *Bacillus* (Nahrowi et al., 2018).
469 According to Frankland (1998), fungal activity had almost ceased after 5 to 6 years when
470 Pokhymycelial bacteria capable of degrading resistant substrates. Due to their slow growth,
471 these microorganisms are unable to compete with non-mycelial bacteria for readily available



472 substances, in the assimilation of which actinobacteria are actively involved (Mahajan et al.
473 2016). Three-year observations of litter-associated microbial organisms in northern Japan, in a
474 pioneer forest dominated by birch and a climax forest dominated by oak, confirmed that the
475 succession of microorganisms in the litter occurs from fungi to bacteria. In addition, in both
476 forest types, Gram-negative bacteria were replaced by Gram-positive bacteria (Otaki and
477 Tsuyuzaki, 2019). Noteworthy is the observation of the role of microbes associated with
478 deciduous litter in the rotting of aboveground wood: contact with litter leads to an increase in
479 the moisture content of the wood and its visually observed faster decomposition (Kirker et al.,
480 2020).

481 A propelling power and predominant regulator of the above microbial succession is the
482 change of litter quality, its chemical composition (Bray et al., 2012; Ge et al., 2017).
483 Therefore, it might be expected that the successive assimilation of different substrates by
484 different groups of microbes is accompanied by changes in the chemical composition of the
485 microbial emission of volatiles. Unfortunately, temporal changes in microbial VOC emission
486 profiles during litter decomposition remain uninvestigated. Recent single studies in this
487 direction are cited below.

488 During the course of a 690-day litterbag experiment in decaying pine (*P. sylvestris*) and
489 spruce (*P. abies*) needle litters, 92 fungi species were identified: 54 species in pine and 73 in
490 spruce litter (Isidorov et al., 2016). The composition of the fungal communities was
491 substantially specific, as only 38% of the species were found in both litter types. At the early
492 stage (30 days of the experiment), the typical primary colonisers from the orders of
493 *Capnodiales*, *Xylariales*, *Dothideales*, *Pleosporales*, *Eurotiales*, *Thelebolales* and
494 *Hypocreales* dominated. After 7 months, none of the primary colonisers were found in pine
495 litter, and only two (*Cladosporium herbarum* and *Pseudeurotium ovale*) were detected in
496 spruce needles. The observed disappearance of these fungi can be linked to the sharp decrease



497 in simple carbohydrates in the early stages of decomposition during the decay of leaf litter
498 (Kainulainen and Hopolainen, 2002; Osono and Takeda, 2005; Isidorov et al., 2010) due to
499 both use by microbes and leaching.

500 At the middle stage (280 days), *Penicillium purpurogenum* dominated in pine litter and
501 *Chloridium botryoideum* in spruce litter, whereas the late stage (490–690 days) was
502 characterised by the absolute predominance of *Trichoderma* species (*T. polysporum*, *T.*
503 *koningii* and *T. viride*). These filamentous fungi are producers of large amounts of cellulolytic
504 enzymes. Remarkably, in the late stage of the experiment, the recurrence of some primary
505 colonisers was also observed, which can be explained by the action of cellulolytic fungi,
506 producing readily available sugars released from the hemicelluloses. The recurrence of the
507 primary colonisers after approximately 350 days of *Pinus pinea* needle incubation was
508 previously documented by Virzo De Santo et al. (2002). Further degradation of more
509 recalcitrant lignified materials demands the participation of lignolytic fungi.

510 It can be noted that unlike the case of VOC emissions from living plants, there is little
511 knowledge regarding such emissions from litter-destroying microbes. The composition of
512 VOCs emitted into the gas phase by selected fungi isolated from Scots pine and Norway
513 spruce needles was determined after 490 days of incubation in one study (Isidorov et al.,
514 2016). Based on the results of HS-SPME/GC-MS analyses, as many as 75 C₂–C₁₅ volatile
515 organic compounds were identified. Table 4 presents the group composition of these VOCs
516 and the principal representatives of separate groups. As can be seen, VOC emissions are
517 species-specific: grown on the same cultivation media, the fungi differ in the qualitative and
518 quantitative composition of the volatiles produced. Even a representative of the same genus
519 (for example, *Trichoderma polysporum* and *T. koningi*, or *Penicillium minioluteum* and *P.*
520 *purpurogenum*) demonstrates considerable distinctions in VOC emissions. Striking



521 differences have also been observed by Nilsson et al. (1996): only three of 58 registered
522 VOCs were common for four *Penicillium* species.

523 Even less is known about VOC production by litter-destroying bacteria. However, the
524 emission of photochemical significant organic volatiles, including isoprene, by bacilli and
525 actinomycetes was discovered (Kuzma et al., 1997; Fall and Copley, 2000; Wilkins, 1996;
526 Wagner et al., 2000; Schöller et al., 2002). At present, the important role of bacteria in leaf
527 litter and coarse woody debris destruction is well documented (Kim et al., 2014; Urbanová et
528 al., 2015; López-Mondéjar et al., 2016; Mahajan et al., 2016; Purahong et al., 2016;
529 Bonanomi et al., 2017; Dolan et al., 2017; Hu et al., 2017). However, further studies are
530 required to understand their role in the litter-derived emission of VOCs into the atmosphere.

531 In terrestrial environments, algae represent the main component of the microbial flora.
532 Forest litter is also the place of the active development of microscopic algae, which form a
533 specific element of the living soil cover (Maltsev et al. 2017a,b). They participate in all
534 biogenic processes in leaf litter, together with other organisms. However, data on the algae of
535 leaf litter are scanty. Maltsev et al. (2017a) have recently studied the structural and dynamic
536 parameters of the algal communities of the anthropogenic and natural forests in the steppe
537 zone of Ukraine and found that the composition of communities varies by season and forest-
538 forming tree species. The authors identified 119 algae species belonging to six divisions:
539 *Chlorophyta* (68 species), *Xanthophyta* (21 species), *Cyanoprocarvota* (17 species),
540 *Bacillariophyta* (seven species), *Eustigmatophyta* and *Charophyta* (three species each). With
541 respect to species number and abundance, coniferous forest litter was predominated by green
542 (*Chlorophyta*) and yellow-green (*Xanthophyta*) algae, whereas leaf forest litter (*Q. robur* and
543 *Robinia pseudoacacia* plantations) was characterised by the significant participation of
544 oxygenic photosynthesising cyanobacteria *Cyanoprocarvota*. In the available literature, we



545 were unable to find information on the emission of VOCs by this group of photosynthetic
546 organisms of the soil cover.

547

548 **V. VOC emission rates: laboratory and field measurements**

549

550 In the past three and a half decades, many measurements of VOC emission rates have been
551 carried out both for the living foliage of individual plants and for various plant communities.
552 In particular, in a review covering the period from 1979 to 1997, 314 values of the emission
553 rates of isoprene and monoterpenes published in 52 articles are given. Another 126 values of
554 the emission rates of 39 non-terpene compounds from more than 50 plant species were
555 published from 1992 to 1999 (Kesselmeier and Staudt, 1999). Among the plants listed in this
556 review, only one species from the genus *Ericaceae*, which is part of the living soil cover of
557 boreal forests, is mentioned: blueberry (*Vaccinium uliginosum*). This shrub is one of the weak
558 sources of isoprene (emission rate $0.009 \mu\text{g g}^{-1} \text{h}^{-1}$). However, two other species of shrub from
559 the same genus – cassandra (*Chamaedaphne calyculata*) and Labrador tea (*Ledum*
560 *groenlandicum*) – are strong monoterpene emitters, with a release rate of more than $3 \mu\text{gC g}^{-1}$
561 h^{-1} (Isebrands et al., 1999).

562 Also limited are data in the literature on rates of VOC emissions from forest leaf litter in
563 natural environments, or even laboratory experiments. However, the few observations made
564 confirm the importance of the contribution of the forest floor to total VOC emissions with
565 regard to some forest types (Pettersson, 1988; Janson, 1993; Isidorov et al., 1994; Stepanov,
566 1999; Hayward et al., 2001). In addition to these early studies, several investigations have
567 been carried out in recent years as part of the SMEAR II (Station for Measuring Relationships
568 between Forest Ecosystems and the Atmosphere) research programme (Hari and Kulmala,
569 2005).



570 The information presented in Table 5 is remarkable in that almost all measurements
571 were carried out in boreal coniferous forests or in mixed forests with a predominance of
572 conifers. Meanwhile, there is no reason to believe that the decomposition of litter of small- or
573 broad-leaved species prevailing in the more southern forests contributes less to the total
574 emissions of biogenic VOCs than the northern, predominantly coniferous forests. However,
575 we were unable to find any publications on VOC emissions from deciduous forests. Besides,
576 plants of live soil cover were considered as a source of VOCs, along with litter, only in some
577 studies (Mäki et al., 2017; 2019a,b). It is also noteworthy that all but two studies were
578 conducted in forests of the Eastern Hemisphere (Hayward et al., 2001; Greenberg et al.,
579 2012).

580 Although this review focuses on the problems associated with VOC emissions from the
581 forest floor of the boreal and temperate zones of the Northern Hemisphere, Table 5 provides
582 the only information we know of these processes in tropical forests (Drewer et al., 2021). The
583 2-year exploration was carried out in a logged tropical forest and an oil palm plantation in
584 Malaysia on the island of Borneo. The amount of litter on the soil determined the level of
585 monoterpene fluxes, of which the main ones, as in the case of the floor of northern forests,
586 were α - and β -pinene as well as limonene (all other measured monoterpenoids, such as 3-
587 carene, camphene and eucalyptol, were emitted at lower rates). Fluxes from oilseed
588 plantations, which were practically devoid of litter, were low and increased only where litter
589 was present.

590 The results of the field experiments carried out to date and their conclusions can be
591 contradictory even within the same working group. For instance, Janson (1993) recorded a
592 fairly high rate (up to $580 \mu\text{gC m}^{-2} \text{h}^{-1}$) of terpene emission by litter in a mature pine forest in
593 southern Sweden, with significantly higher values in autumn than in summer. According to
594 the author's estimate based on these measurements, the flux of terpenes from the pine forest



595 litter accounted for 30% of the crown emissions. However, subsequent measurements (Janson
596 et al., 1999) in coniferous forests and wetlands in summer and early autumn yielded
597 considerably lower emission rates of monoterpenes ($19\text{--}90 \mu\text{g C m}^{-2} \text{ h}^{-1}$); it was concluded
598 that the flux of biogenic VOCs from the forest floor only accounts for a few percent of the
599 total forest flow. According to data obtained at the same time in western Russia (Stepanov,
600 1999), the emission rates of monoterpenes by spruce litter were higher ($21\text{--}402 \mu\text{g m}^{-2} \text{ h}^{-1}$),
601 and the highest values were recorded not in July–August at air temperatures of $22\text{--}25^\circ\text{C}$ but
602 in the second half of September at temperatures of $6\text{--}8^\circ\text{C}$. In both summer and autumn,
603 temporarily increased emission levels were observed immediately after rain events. The
604 strong effect of rain on VOC emissions from litter has been confirmed in a recent study
605 (Crocker, 2021). In the author's opinion, a short-term increase in emission by leaf litter of
606 *Clitoria fairchildiana* after its moistening occurs in accordance with Henry's law. The
607 concentration profile of monoterpenes, measured by Stepanov (1999) under the canopy of an
608 80-year-old spruce forest in summer in calm weather, showed the presence of two maxima,
609 one of which was located approximately at half the height of the crown tree, and the other
610 near the surface covered with coniferous litter at a height of 0.8 m.

611 In another study, monoterpene emissions from the boreal pine forest floor in Finland
612 varied from 0 to $373 \mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ (Hellén et al., 2006), that is, its upper value was consistent
613 with that obtained by Stepanov (1999). According to the authors, emissions of terpenes from
614 the forest floor can be a significant source for the atmosphere, especially in spring.

615 Kivimäenpää et al. (2018) conducted a latitudinal experiment in Finland, involving a
616 1,000-km North–South transect, and concluded that emissions from the floor of pine forests
617 have a noticeable additive effect on total VOC emissions. The authors of an earlier study
618 (Räisänen et al., 2009), who measured the fluxes of monoterpenes in a pine forest in eastern
619 Finland, were inclined to the same conclusion. According to their estimates, the total flux of



620 monoterpenes in June–September was 502 mg m⁻², whereas the flux from the forest canopy
621 was only 374 mg m⁻². According to the authors, the difference of more than 25% can be
622 partially explained by the emission of terpenes from the forest floor.

623 The results of recent measurements – carried out mainly by Scandinavian researchers
624 (Mäki et al., 2017, 2019; Kivimäenpää et al., 2018; Wang et al., 2018) – and their conclusions
625 about the important role of VOC emissions from forest litter (primarily photochemically
626 active terpenes and sesquiterpenes with high potential for secondary aerosol formation) are at
627 odds with the earlier conclusion (Janson et al., 1999; Greenberg et al., 2012) that they only
628 slightly contribute to the total ecosystem flux of VOCs. To clarify the nature of such
629 discrepancies, it is necessary to continue research with a wider coverage of geographical and
630 climatic conditions and, first of all, to include forests formed by deciduous trees in the
631 research programme.

632 Table 5 also includes chamber measurements of mono- and sesquiterpene fluxes from
633 wet scrubland in northern Sweden, including an experiment that adds birch litter to the
634 chambers (Faubert et al. 2010). The relatively low emissions of terpenoids were doubled in
635 response to an air temperature increment of only 1.9–2.5 °C. This suggests that the predicted
636 warming of the climate in the subarctic and arctic regions is likely to lead to a significant
637 increase in the flux of reactive VOCs from plants of the living ground cover of the tundra.
638 The same conclusion was reached by the authors who studied the release of terpenes by plants
639 of the living soil cover of the high and low Arctic heaths and subarctic peatland (Lindwall et
640 al., 2015).

641 In addition to these field observations, the results of the small number of laboratory
642 measurements of the rates of VOC release by the litter of some species of woody and shrub
643 plants have been published. For example, the rates of terpene emission by fresh litter of
644 European larch as well as fresh and decaying litter of Norway spruce (*P. abies*) and Scots pine



645 (*P. sylvestris*) have been reported (Table 6). At the early stages of decomposition (2.5–5.5
646 months from the beginning of the experiment), the rates of terpene emission from pine and
647 spruce litter exceeded those observed for freshly fallen needles. Over the entire observation
648 period, in the case of spruce litter, camphene and α -pinene were released into the gas phase at
649 the highest rates, whereas in the case of pine litter, the main VOCs were α -pinene and 3-
650 carene.

651 In a recent study, the results of laboratory measurements of the rates of VOC release by
652 fresh (not decomposed) deciduous litter of 16 species of woody and shrub plants typical of the
653 Mediterranean region are presented (Viros et al., 2020). These species are divided by the
654 authors into two groups depending on the presence of terpene storage structures. Eighty-seven
655 VOCs were identified in the isolations of all the examined plants by GC-MS. The emission
656 rates of various VOCs at a temperature of 30°C ranged greatly. In the case of nine plants of
657 the terpene-storing species (*Eucalyptus globulus*, *Juniperus oxycedrus*, *Pinus pinea*, *P.*
658 *halepensis*, *Rosmarinus officinalis*, *Thymus vulgaris*, *Cistus albidus*, *C. salviifolius*, *Cotinus*
659 *corrygia*), the total emissions of monoterpenes, sesquiterpenes and non-terpene compounds
660 were in the range of < 0.01–4.62, 0.04–0.92 and < 0.01–0.23 $\mu\text{g g}^{-1} \text{h}^{-1}$, respectively (Table 7).
661 Emissions of seven non-terpene-storing species via litter (*Acer monspesulanum*, *Erica*
662 *arborea*, *Quercus ilex*, *Q. coccifera*, *Q. pubescens*, *Q. suber*, *Ulex parviflorus*) consisted
663 mainly of linear alkanes and carbonyl compounds and ranged from 0.01 to 0.61 $\mu\text{g g}^{-1} \text{h}^{-1}$
664 (Table 8).

665 A 15-month litter bag experiment was conducted to study the composition of VOCs
666 released by the decaying pine needles of *Pinus halepensis*, one of the main coniferous trees in
667 the Mediterranean (Viros et al., 2021). Material for laboratory studies was sampled every 3
668 months, and VOC emissions were determined both online (using the PTR-ToF-MS
669 instrument) and offline (GC-MS). During the observation period, 58 compounds were



670 registered in the composition of the fallen needle excretions. The maximum emission of 9.18
671 $\mu\text{g g}^{-1} \text{h}^{-1}$ was observed 3 months after the beginning of the experiment, with the main
672 components being α -pinene, terpinolene, 3-carene, limonene, sabinene and myrcene, whereas
673 the main sesquiterpenes were β -caryophyllene, α -humulene and copaene. In addition, at this
674 stage, the emission of highly volatile non-terpene compounds took place, with the main ones
675 being methanol, acetic acid and acetone. The maximum emission of oxidised monoterpenes
676 and sesquiterpenes occurred between the third and sixth months of the experiment.
677 Comparison of data on VOC emissions from green and decaying fallen needles *P. halepensis*
678 led the authors of the cited work to the conclusion that the contribution of litter to the total
679 emission in forests from this pine species (their area in the Western Mediterranean is about
680 3.5 million hectares) can be extremely high.

681 The data presented in this section indicate that VOC emissions from the forest floor are
682 not negligible. Moreover, during the spring and autumn seasons, the forest floor can be the
683 main source of these reactive components. However, to establish its effect on air quality,
684 additional studies are needed to determine the emission factors for different ecosystems and
685 biomes.

686 VI. Conclusions

687 It is recognised that a critical challenge for atmospheric chemistry is closing the
688 atmospheric organic carbon budget. This review shows that deciduous litter and plants of
689 living forest soil cover are an unaccounted source of large amounts of VOCs, for many of
690 which a high potential for the formation of ozone and other atmospheric photooxidants and/or
691 secondary aerosols has been postulated. The results of many studies indicate that these
692 elements of forest ecosystems are not negligible sources of atmospheric VOCs, especially in
693 spring and autumn. Based on their origin, they are a mixture of a variable composition of



694 products released as a result of various insufficiently studied physical and biological
695 processes.

696 There are serious knowledge gaps that prevent a quantitative assessment of the role of
697 VOC emissions from the forest floor in biospheric processes. These include the lack of
698 information on the composition and emission rates of litter VOCs in various forest types,
699 primarily those formed by deciduous tree species: the overwhelming majority of the
700 measurements were made in the coniferous forests of the northern belt. This applies to an
701 even greater extent to the dead parts of herbaceous plants of steppe and related ecosystems,
702 although the biomass of grass felt per unit area in them can significantly exceed that of
703 deciduous litter in forests. The participation of soil cover plants in the balance of these
704 compounds has hardly been studied, and it is unclear whether they serve as sources of VOCs
705 or as their absorbers (Mäki et al., 2017; Stoy et al., 2021). In some pioneering works, an
706 important influence on the rate of decomposition of litter of solar radiation, including in the
707 visible part of the spectrum, has been shown (Austin et al., 2016; Keiser et al., 2021).
708 However, the chemical composition of the resulting compounds, both volatile and easily
709 accessible for microbiological assimilation, has not been studied. Insufficient research has
710 focused on the impact of global changes on litter decomposition and VOC emissions by
711 ecosystems. There is also a lack of data on both litter and plants of the living ground cover of
712 forests in different geographic zones necessary for an inventory of VOC emissions on a
713 regional and global scale.

714 The study of the decomposition of leaf litter, the chemical composition and emission
715 rates of volatile organic compounds is of interest not only from the point of view of
716 atmospheric chemistry. Biogenic VOCs are involved in competitive interactions as both
717 allelochemicals and as neighbour detection signals (Kegge and Pierik, 2009). Volatile organic
718 compounds from leaf litter decomposition alter soil microbial communities and carbon



719 dynamics (McBride et al., 2020). However, it should be noted that this is not the only process
720 linking deciduous litter and soil biota. When litter decomposes, both labile and more stable
721 forms of dissolved organic matter are formed, significantly affecting the binding and
722 consumption of carbon in the soil (Hensgens et al., 2021); however, the chemical composition
723 of compounds leached from the litter and the very scale of this phenomenon remain
724 unexplored.

725 This review is limited to boreal and temperate natural ecosystems of the Northern
726 Hemisphere because of the almost complete absence of literature data on VOC emissions
727 from forest soil cover from other botanical-geographical zones. In particular, only a few
728 works are devoted to VOC emissions in subtropical and tropical forests (Drewer et al., 2021;
729 Crocker, 2021). We would like our review to serve as a stimulus for expanding the geography
730 of research and for a deeper study of the role of both 'dead' and living forest soil cover in the
731 release of organic compounds into the environment, which play an important role in many
732 ecological processes under the forest canopy and in the global atmosphere.

733

734 **Supplementary material related to this article is available online at:**

735 <http://www.biogeosciences.net/.....>

736

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742

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1305 **Table 1.** Rate of decomposition of dead plant parts (%) during the warm season in two
1306 ecosystems of the forest-steppe zone of Russia (Bazilevich et al., 1993).

Factors	Oak forest	Meadow steppe
Warm season		
Abiotic	14	72
Biotic		
– microorganisms	10	11
– mesofauna	30	9
– macrofauna	46	8
Whole year		
Abiotic	15	30
Biotic		
– microorganisms	45	55
– meso- and macrofauna	40	14

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1322 **Table 2.** Organic compounds in volatile emissions of leaves litter. Plant species: 1 - oriental beech (*Fagus*
 1323 *orientalis*); 2 - sequoia (*Sequoiadendron giganteum*); 3 - cherry prinsepia (*Prinsepia sinenes*); 4 - aspen (*Populus*
 1324 *tremula*); 5 - English oak (*Quercus robur*); 6 - maple (*Acer rubrum*); 7 - pine (*Pinus taeda*); 8 - poplar (*Populus*
 1325 *balsamifera*); 9 - birch (*Betula pendula*); 10 - willow (*Salix* sp.); 11 - common beech (*Fagus sylvatica*); 12 -
 1326 broad beech fern (*Phegopteris hexagonoptera*); 13 - northern red oak (*Q. rubra*); 14 - bur oak (*Q. macrocarpa*);
 1327 15 - lodgepole pine (*P. contorta*); 16 - ponderosa pine (*P. ponderosa*); 17 - eastern cottonwood (*Populus*
 1328 *deltoides*); 18 - quaking aspen (*Populus tremuloides*); 19 - green ash (*Fraxinus pennsylvanica*); 20 - great laurel
 1329 (*Rhododendron maximum*); 21 - eucalyptus (*Eucalyptus* sp.); 22 - spotted knapweed (*Centaurea maculosa*); 23 -
 1330 miscanth (*Miscanthus* sp.); 24 - wheatgrass (*Thinopyrum intermedium*); 25 - Scots pine (*P. sylvestris*); 26 -
 1331 Norway spruce (*Picea abies*); 27 - European larch (*Larix decidue*); 28 - clover (*Trifolium repens*); 29 - bluegrass
 1332 (*Poa pratensis*); 30 - European silver fir (*Abies alba*).
 1333 References: **a** - Derendorp et al. (2011); **b** - Isidorov et al. (2002); **c** - Ramirez et al. (2010); **d** - Warneke et al.
 1334 (1999); **e** - Fall et al. (2001); **f** - Gray et al. (2010); **h** - Isidorov et al. (2003); **i** - Isidorov et al. (2010); **j** -
 1335 Isidorov et al. (2005); **k** - Kesselmeier and Hubert (2002).
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Compound	Plant	Ref.
Aliphatic hydrocarbons		
Ethane	1,2	a
Ethylene	1-3	a
Propane	1-3	a
2-Butene	4,5	b
Butin	6,7	c
Pentane	1-3	a
Isoprene	4,5,8-24	b,d-f
Hexane	4,8	b
Heptane	4,5,8	b
Octane	4,5,8,9	b
Nonane	4,5,8-10	b
Decane	4,5,8-10	b
Undecane	4,5,8-10	b
Dodecane	4,5,8	b
Tridecane	4	b
Aromatic hydrocarbons		
Benzene	4,5,10,25,26	b,h,i
Toluene	4-10,13-24	b,f
Ethyl benzene	4,5,8-10	b
<i>p</i> -Xylene	4,5,9	b
<i>m</i> -Xylene	5,8,9	b
<i>o</i> -Xylene	5,9	b
Cumene	8	b
<i>n</i> -Propyl benzene	5	b
Mesitylene	5	b
<i>p</i> -Cymene	4,5,8-10,25,26	b, h,i
<i>p</i> -Cymenene	13	h
Monoterpene hydrocarbons		
Santene	26	h,i
Triicyclene	4,5,9,25,26	b, h,i
α -Thujene	25,26	g
α -Pinene	4,5,8-11,25-27	b,d, h-j
Camphene	4,5,8-10,25-27	b, h-j
Verbenene	25,26	h,i
Sabinene	25,26	i
β -Pinene	4,5,10,11,25-27	b,d,h-j
Myrcene	8,25-27	b,h-j
α -Phellandrene	9,25,26	b,i
3-Carene	4,10,25,26	b,h,i
α -Terpinene	8,25,26	b,i



β -Phellandrene	25,27	h,j
Limonene	4,5,8,9,25-27	b,h-j
β -(Z)-Ocimene	25,26	h,i
β -(E)-Ocimene	9,25,26	h,i
γ -Terpinene	25,26	h,i
<i>p</i> -Mentha-2,4(8)-diene	25	i
Terpinolene	25,26	h,i
Monoterpenoids		
1,8-Cineol, dehydro-	8,9	b
1,8-Cineol	5,26	h
Camphor	25,26	h,i
Fenchone	4	b
Linalool	25,26	h
Camphene hydrate	26	i
Limonene oxide	5	b
β -Terpineol	26	h
Borneol	26	h,i
<i>cis</i> -3-Pinanone	25	h
Terpin-4-ol	25,26	h
α -Terpineol	25,26	h,i
Myrtenol	25,26	i
<i>p</i> -Cymen-8-ol	25	h
Verbenone	25	h,i
<i>cis</i> -Carveol	25	h
Myrtanol	26	h
Linalyl acetate	26	h
α -Terpenyl acetate	25,26	i
Bornyl acetate	25,26	b, i
Sesquiterpene hydrocarbons		
δ -Elemene	26	h
α -Cubebene	25,26	h
α -Longipinene	26	h,i
Longicyclene	26	h
Ylangene	25	i
α -Copaene	25,26	h,i
β -Bourbonene	25,26	h,i
β -Cubebene	25	h,i
β -Elemene	25	h,i
D-Longifolene	25,26	i
(Z)-Caryophyllene	25	i
α -Gurjunene	25	h
β -Caryophyllene	25,26	h,i
Aromadendrene	25,26	i
β -Humulene	25,26	h
α -Humulene	25,26	h,i
γ -Muurolene	25,26	h,i
Germacrene D	25	i
β -Selinene	25	h,i
α -Selinene	25	h,i
α -Muurolene	25,26	h,i
γ -Cadinene	25,26	h,i
Cadina-1,4-diene	25,26	h,i
α -Calacorene	25	i
Sesquiterpenoids		
Spathulenol	25	h
Caryophyllene oxide	25	h,i
<i>epi</i> - α -Cadinol	25	h
α -Muurolol	25	h



τ -Cadinol	25	i
α -Cadinol	25	h,i
Carbonyl compounds		
Acetaldehyde	11,13-26	d,f,h
α -Methylacroleine	4,5	b
Propanal	13-24	f
Isobutanal	5	b
2-Methyl butanal	28	e
3-Methyl butanal	8,28,29	b,e
Pentanal	4,8,26	b,i
Hexanal	25,26,28	e,i
2-Hexenal	5,8,10	b
3-(Z)-Hexenal	11,29	e
Nonanal	25,26	i
Benzaldehyde	4,5,8-10,26	b,i
Furfural	4	b
Anis aldehyde	26	h
Vanillin	6,7	c
Acetone	5-7,10,11,13-26	b-d,f-i
Butan-2-one	5,10,11,13-26	b,d,f-i
1-Penten-3-one	12,28-30	e
2-Pentanone	25	i
3-Hexanone	25	i
4-Octanone	25,26	i
3-Octanone	25,26	i
6-Methyl-5-hepten-2-one	25,26	i
2-Decanone	25	i
Alcohols		
Methanol	6,7,11,13-24	c,d,f
Ethanol	4,5,8-11,13-26	b,d,f,i
1-Propanol	11,13-24	d,f
1-Butanol	4,5,8,13-24	b,f
2-Methyl-1-butanol	25,25	h
3-Methyl-1-butanol	4,25,26	b,h
1-Penten-3-ol	12,28-30	e
2-(Z)-penten-1-ol	12,28-30	e
1-Pentanol	25	h
2-Ethyl-1-hexanol	25,26	i
Phenol	25,26	i
Benzyl alcohol	25,26	i
Esters		
Methyl formate	8	b
Methyl acetate	4	b
Ethyl acetate	4,5,10,25	b,h
Methyl tiglate	4	b
Butyl acetate	5,8,10,25,26	b,i
Ethyl butanoate	25,26	i
Allyl butyrate	4	b
Isopentyl acetate	4,10	b
Isopentyl butanoate	25	i
sec-Pentyl acetate	4	b
Allyl isopentanoate	4	b
Hexenyl acetate	28	e
Ethyl 3-methylbutanoate	7	c
Isopentyl butanoate	25	i
Sulfur and chlorine containing compounds		
Methyl mercaptane	8	a



Ethyl mercaptane	8	a
Dimethyl sulfide	11	k
Diethyl sulfide	4	a
Ethyl isopropyl sulfide	4	a
Methyl chloride	1-4	a,b
Dichloromethane	4,25,26	b,i
Trichloromethane	4,26	b,i
Trichloroethylene	25,26	i
Tetrachloromethane	4,5,8	a
Other VOCs		
Formic acid	13-24	f
Acetic acid	5,8,13-24	b,f
Diethyl ether	4,9,10	b
2-Methylfuran	5,9,10,25,26	b,i
3-Methylfuran	25,26	i
Thymol, methyl ether	25	i
<i>o</i> -Cresol	5	b
Anisol	8	b
Methoxy phenol	8	b

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Table 3. VOCs composition of some LS plants (forests and bog forests). Plants: 1 - lichen (*Cetraria islandica*); 2 - heather (*Calluna vulgaris*); 3 - moss (*Pleurozium schreberi*); 4 - club moss (*Lycopodium clavatum*); 5 - red bilberry (*Vaccinium vitis-idaea*); 6 - bilberry (*Vaccinium myrtillus*); 7 - moss (*Polytrichum commune*); 8 - marsh tea (*Ledum palustre*); 9 - male fern (*Dryopteris filix-max*); 10 - soft tree fern (*Dicksonia antarctica*); 11 - Japanese beech fern (*Thelypteris decursive-pinnata*); 12 - southern shield fern (*Thelypteris kunthii*); 13 - cottongrass (*Eriophorum* sp.); 14 - red raspberry (*Rubus idaeus*); 15 - Labrador tea (*Rhododendron groenlandicum*); 16 - bog heather (*Erica tetralix*); 17 - sphagnum (*Sphagnum fuscum*); 18 - thick stemmed wood fern (*Dryopteris crassirhizoma*); 19 - marsh andromeda (*Andromeda polifolia*).
 Ref.: a - Isidorov et al. (1994); b - Isidorov et al. (1992); c - Isidorov et al. (1985); d - Tingey et al. (1987); e - Drewitt et al. (1998); f - Helmig et al. (1999); g - Mochizuki et al. (2014); h - Rinnan et al. (2005).

Compound	Plant	Ref.
Aliphatic hydrocarbons		
Ethane	1,3,5	a
Propane	2-4	a
Butene	1-4	a
2-Methylbutane	1,2,4-9	a-c
Pentane	1-4	a
Pentene	5-7,9	b
Penta-1,3-diene	7,9	c
Isoprene	1,2,4-15,17,18	a-g
2,3-Dimethylbutane	17	h
2,3-Dimethylbutadiene	8	b
Heptane	1	a
Decane	1	a
Undecane	1,2	a
Aromatic hydrocarbons		
Benzene	1-3	a
Toluene	1-3	a
<i>p</i> -Cymene	1,2,14-16	a,f
<i>p</i> -Cymenene	15	f
Monoterpene hydrocarbons		
Bornylene	5	b
Tricyclene	7	b
α -Thujene	5,15,16	b,f
α -Pinene	2-14,16,19	a-f,h



β -Pinene	1,7,8,15,16	a-c,f
α -Fenchene	5,15	b,f
Camphene	2,5,7,8,15,17	a-c,f
Myrcene	2,7,8,14-16	a-c,f
3-Carene	7,8,16	b,c,f
α -Phellandrene	5,15,16	b,f
Limonene	1,7,8,13-16,19	a-c,f,h
Alloocimene	8	b,c
<i>trans</i> -Ocimene	15	f
α -Terpinene	15,16	f
γ -Terpinene	15	f
Terpinolene	8,14	b,c,f
Isobornyl acetate	15,16	f
Carbonyls		
Propanal	9	b,c
Butanal	9	b,c
Isobutenal	7,9	c
α -Methylacrolein	7,9	b
Hexanal	9	c
Octanal	12	f
Benzaldehyde	2,3,16	a,f
Acetone	1,2,5-9,14	a-c,f
Butan-2-one	9	b,c
2-Methylbutan-3-one	8	c
3-Methyl-3-buten-2-one	9	b
Pentan-2-one	5-7,9	b,c
Pentan-3-one	7	b,c
6-Methyl-5-hepten-2-one	12	f
5-Methyl-3-hepten-2-one	17	f
Octan-3-one	7,9	c
1-Octen-3-one	7,9	b
Furans		
2-Methyl furan	5-9,12	b,c,f
3-Methyl furan	6	b,c
Ethyl furan	5-7,9,12	b,c,f
Vinyl furan	8	b,c
Hexyl furan	8	b
Furan, 3-(4-methyl-3-pentenyl)-	8	c
Alcohols		
Ethanol	1,2,4-7,9	a-c
Butan-1-ol	1	a
<i>cis</i> -3-Hexen-1-ol	5,6,8,14,19	b,c,f,h
6-Methyl-5-hepten-2-ol	12	f
Other		
Diethyl ether	1-4	a
Ethyl acetate	1,2,5-7	a-c
3-Hexen-1-ol acetate	5,6,12,16	b,c,f
Methyl chloride	1,4	a
Chloroform	7	c
Dimethyl sulfide	9	b

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Table 4
 Relative composition (% of TIC) of volatile compounds emission from "fresh" pine and spruce litter and isolated from litter fungi after 490 days of the needle incubation)

Compound	Pine litter	Spruce litter	<i>Trichoderma polysporum</i>		<i>Trichoderma koningii</i>	<i>Mortirella isabellina</i>	<i>Penicillium purpurogenum</i>	<i>Penicillium minioluteum</i>	<i>Absidia glauca</i>	<i>Thielavia terricola</i>	Non-spore forming
			Pine	Spruce	Spruce	spruce	spruce	pine	pine	spruce	
Monoterpenes	8.0	10.2	69.8	22.5	34.8	72.2	1.1	4.0	44.1	67.9	56.3
- α -pinene	3.5	4.6	17.0	5.9	15.1	10.1	0.5	0.1	4.1	14.6	8.4
- camphene	0.2	0.5	0.7	-	trace	0.3	trace	-	-	-	0.1
- myrcene	-	trace	26.4	trace	8.2	24.5	-	0.1	1.8	-	3.2
- 3-carene	2.8	3.2	trace	8.7	9.9	1.5	0.6	3.4	33.1	36.9	37.1
- limonene	1.0	1.0	13.3	5.8	0.6	26.1	trace	trace	4.6	10.0	5.5
- terpinolene	-	-	6.1	-	0.2	4.6	-	0.4	0.5	0.2	-
Monoterpenoids	-	5.0	3.6	6.6	3.2	0.1	-	-	8.0	-	11.2
- linalool	-	1.2	-	-	1.2	-	-	-	-	-	-
- menthone	-	-	2.3	6.6	0.8	-	-	-	-	-	8.1
- menthol	-	-	0.7	2.1	1.2	-	-	-	-	-	2.5
- verbenone	-	-	-	0.1	0.1	-	-	-	-	-	0.5
Sesquiterpenes	2.1	1.3	16.5	0.3	19.0	9.0	-	-	-	-	0.4
- longifolene	2.1	1.3	-	-	1.5	-	-	-	-	-	-
- β -caryophyllene	-	-	15.2	0.3	14.2	9.0	-	-	-	-	0.4
Alcohols	13.2	7.2	2.8	51.8	39.5	3.6	8.4	-	0.8	16.2	10.5
- ethanol	-	-	-	25.0	22.5	2.8	trace	-	-	16.2	4.3
- isopentanol	-	-	0.8	8.5	12.6	0.7	trace	-	-	-	2.7
- 1-octen-3-ol	-	-	0.2	0.1	9.2	0.1	8.4	trace	-	-	-
Carbonyls	34.6	38.9	2.5	9.7	13.9	8.8	82.5	50.3	41.3	8.8	12.9
- acetaldehyde	-	-	0.5	0.1	2.0	-	-	-	trace	-	-
- acetone	26.8	36.8	2.0	9.7	6.9	8.8	82.5	46.0	18.7	8.8	12.9
- 2-butanone	-	-	-	trace	-	-	-	2.8	22.6	-	-
- 3-octanone	-	-	trace	trace	2.4	-	trace	trace	-	-	-
Esters	trace	-	-	1.2	1.5	0.9	-	-	-	0.8	4.8
Furans	-	-	-	2.5	3.0	-	3.6	6.1	-	2.0	0.3
Chlorohydrocarbons	12.5	4.9	-	trace	-	0.2	0.9	3.9	-	-	-
- chloroform	-	-	-	trace	-	0.2	0.9	3.9	-	-	-
Aromatic hydrocarbons	19.5	10.5	3.0	0.7	3.8	4.4	0.2	5.7	0.6	2.3	1.9
- benzene	7.0	3.1	-	-	2.1	0.2	-	0.9	-	-	-
- toluene	7.3	5.2	0.2	trace	0.7	0.5	0.1	1.1	0.4	1.6	0.8
- xylenes	1.0	1.2	0.1	0.2	trace	0.2	0.1	0.7	-	0.1	0.1
- <i>p</i> -cymene	0.7	0.5	2.7	0.5	0.5	3.5	-	2.6	0.2	0.3	1.0
Alkanes	12.3	8.3	0.7	0.7	0.4	0.9	1.1	14.3	-	1.6	0.7

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Table 5. Field measurements of VOC emission from forest floor

Method	Forest/sampling site type	Time of experiments	Emission rates	References
Static flow chamber technique. Sampling in sorption tubes with Tenax TA. GC-MS analysis of VOCs after thermal desorption.	VOCs emission from pine forest floor on two plots in the 140-year-old stand of Scots pine in the southern Sweden.	VII and X.1992	The terpene emission rate in July normalized to 20°C was 21 $\mu\text{gC m}^{-2} \text{h}^{-1}$. In October, the emission rate at the two test sites was 104 and 580 $\mu\text{gC m}^{-2} \text{h}^{-1}$. According to the author's estimates, the emission from the bottom of a pine forest reaches 30% of the emission from tree crowns.	Janson, 1993
Static flow chamber technique. VOC	Middle taiga pine (<i>P. sylvestris</i>) forest in	VII–VIII. 1993	Depending on the temperature (16–28°C)	Isidorov et al., 1994



concentration in dual-bed sorption tubes with Tenax GC and Carbotrap. GC-MS analysis of VOCs after thermal desorption	Karelia (62° 07' N, 33° 56' E) with soil cover with green mosses (<i>Pleurocium schreberi</i> , <i>Politrichum commune</i>), lichens (<i>Cetraria islandica</i> , <i>Cladonia rangiferina</i>), shrubs (<i>Calluna vulgaris</i>) and club moss.		within 6.1–40.5 $\mu\text{gC g}^{-1} \text{h}^{-1}$ (based on the dry weight of pine litter and living plants inside the static chamber).	
Static chamber technique. VOC concentration on Tenax TA; whole air sampling of light hydrocarbons; sample analysis by GC-FID.	Southern boreal zone in Sweden. Mixed pine (<i>P. sylvestris</i>) and spruce (<i>P. abies</i>) forest, about 50 years of age, and a <i>Sphagnum</i> fen.	VI–IX.1995	The monoterpene flux from forest floor made up only a few percent of total forest flux, reaching 50 $\mu\text{gC m}^{-2} \text{h}^{-1}$. The highest flux was observed from the moss plots. Small emission of ethene and propane was registered. <i>Sphagnum</i> fen emitted isoprene at the average rate 55±45 $\mu\text{gC m}^{-2} \text{h}^{-1}$ in June and 408±240 $\mu\text{gC m}^{-2} \text{h}^{-1}$ in August. Monoterpene fluxes were 19±9 $\mu\text{gC m}^{-2} \text{h}^{-1}$ in June and 90±60 $\mu\text{gC m}^{-2} \text{h}^{-1}$ in August.	Janson et al., 1999
Static chamber technique. VOC concentration in sorption tubes with Tenax TA. GC-MS analysis of VOCs after thermal desorption.	Leningrad region (59° 57' N, 29° 51' E). Coniferous forest with a predominance of Norway spruce (<i>Picea abies</i>). The soil cover is formed by litter of spruce needles and green mosses.	VII–IX.1997	The monoterpenes fluxes from fallen spruce needles at 6–25 °C was 20.9–402.0 $\mu\text{g m}^{-2} \text{h}^{-1}$. The share of α -pinene in the total emission of terpenes is 82%. The highest emission rates were recorded shortly after rainfall.	Stepanov, 1999
PTR-MS system was used for simultaneous monitoring of the VOCs concentration.	Semi-natural experiment on VOC emission by decaying beech leaves	IV.1997	The observed emission rate of methanol, acetaldehyde and acetone at 19 °C was 1.0, 0.23 and 0.1 $\mu\text{g g}^{-1}(\text{dw}) \text{h}^{-1}$, respectively	Warneke et al., 1999
Dynamic soil enclosure technique. Air samples were collected on dual-bed adsorption trap (Tenax TA + Carbotrap). GC-MS analysis of VOCs after thermal desorption.	Sitka spruce (<i>Picea sitchensis</i>) commercial plantation (54° 20' N, 3° 2' W). The trees on the plantation are 35 years old.	VII.1999	The average rate of terpene emission from the undisturbed forest floor was 33.6 $\mu\text{g m}^{-2} \text{h}^{-1}$ (30 °C). Limonene and α -pinene dominated the emission of monoterpenes; camphene, β -pinene and myrcene were also present in significant quantities. According to authors, the surface needle litter was the most likely source of soil emissions to the	Hayward et al., 2001



Static chamber technique. VOC concentration in sorption tubes with Tenax TA-Carbopack B adsorbent. GC-MS analysis of VOCs after thermal desorption.	Boreal wetland and a Scots pine forests at the SMEAR II station (61° 51' N, 24° 17' E) in southern Finland. Forest is dominated by <i>P. sylvestris</i> with admixture of aspen (<i>P. tremula</i>) and birches (<i>B. pendula</i> and <i>B. pubescens</i>). The ground vegetation consists mainly of shrubs. Northern ribbed fen is dominated by mosses (<i>Sphagnum</i> sp.), sedges and Rannoch-rush.	VI–X.2014; IV–VI, VIII.2015	atmosphere. Isoprene was main hydrocarbon emitted by the wetland with the mean emission potential 224 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ for the whole season. Monoterpene emission from boreal forest floor varied from 0 to 373 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ (0–0.8 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ of β -caryophyllene). Comparison with Scots pine emission showed that forest floor may be an important monoterpene source, especially in spring	Hellén et al., 2006
Dynamic PVC soil cuvette system. VOCs released by soil and litter were concentrated in sorption tubes with Carbotrap C, Carbotrap B, and Carbosieve S-II separated by quartz wool. GC-MS analysis after thermal desorption.	Mediterranean mountains environment in the Prades region of Southern Catalonia dominated by evergreen oak <i>Quercus ilex</i>	Spring 2003–spring 2004	The measured monoterpene emission in summer was $1.67\pm 1.57 \mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ normalised to 30° C.	Asensio et al., 2007
The air was sampled using a transparent polycarbonate chamber placed on an aluminum collar. VOCs were concentrated in sorption tubes filled with a combination of Tenax TA and Carbopack B adsorbents. GC-MS analysis of VOCs after thermal desorption	Wet dwarf shrub heath in northern Sweden (68° 21' N, 18° 49' E; 385 m a. s. l.). The vascular vegetation consist mainly of the sedge (<i>Carex vaginata</i>) and the evergreen shrubs <i>Empetrum hermaphroditum</i> and <i>Andromeda polifolia</i> . Additionally, an experiment was carried out to study the effect of the addition of <i>Betula pubescens</i> ssp. <i>czerepanovii</i> litter.		The measured monoterpene hydrocarbons emission rate varied from 1.5 to 9.8 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$. The relatively low emissions of terpenoids were doubled in response to an air temperature increment of only 1.9–2.5°C, while litter addition had a minor influence.	Faubert et al., 2010
Steady-state chamber system. VOC concentration in sorption tubes with Tenax-Carbopack-B adsorbent. GC-MS analysis of VOCs after thermal desorption.	Scots pine forest floor at the SMEAR II station in southern Finland.	IV–XI.2008	Emission rates: $\text{C}_{10}\text{H}_{16}$ – 5.04 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$; $\text{C}_{15}\text{H}_{24}$ – 0.045 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$; C_5H_8 – 0.05 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$. Seasonal variations were recorded with a maximum in early summer and autumn.	Aaltonen et al., 2011
Static enclosure technique. PTR-MS system was used for simultaneous monitoring of	The Manitou Experimental Forests (39° 6' N, 150° 5'30" W) with mixed-age ponderosa pine up to	VI–VII.2010	Litter fluxes made only a minor contribution to the above-canopy flux for the VOCs measured. In the case of methanol,	Greenberg et al., 2012



methanol, acetaldehyde, acetone and propanal, acetic acid, mono- and sesquiterpenes emission.	100 years old and a sparse vegetation surface cover of grasses, sage and forbs.		acetaldehyde, acetone, and the sum of monoterpenes, the average daily flux is 1.3, 0.3, 0.3, and 3.1 $\mu\text{g m}^{-2} \text{h}^{-1}$, respectively. The litter emission clearly follow the trend in temperature; the dependence of litter moisture is of lesser importance, however, VOC emissions increased immediately after the rain.	
BVOC emissions were measured during 24-hour periods in the field using a push-pull chamber technique and collection of volatiles in adsorbent cartridges followed by analysis with GC-MS	Five arctic vegetation communities were examined: high arctic (74°30' N, 21°00' W) heaths dominated by <i>Salix arctica</i> and <i>Cassiope tetragona</i> , low arctic (69°14' N, 53°32' W) heaths dominated by <i>Salix glauca</i> and <i>Betula nana</i> and a subarctic peatland (67°22' N, 26°38' E) dominated by the moss <i>Warnstorfia exannulata</i> and the sedge <i>Eriophorum russeolum</i> .	The subarctic site was measured on 23–24 July and 5–6 August 2008; The high and low arctic sites were measured on June and July 2013	The total monoterpene fluxes in high Arctic <i>Salix</i> -heath and high Arctic <i>Cassiope</i> -heath were 0.01 and 7 $\mu\text{g} \cdot \text{m}^{-2} \cdot \text{h}^{-1}$, respectively. Warming by open top chambers increased the emission rates both in the high and low arctic sites, forewarning higher emissions in a future warmer climate in the Arctic.	Lindwall et al., 2015
Static chamber technique. Concentration of VOCs in sorption tubes filled with Tenax TA and Carboxpack B. GC-MS analysis of VOCs after thermal desorption.	55-year old Scots pine forest in southern Finland with <i>Sorbus aucuparia</i> , <i>B. pendula</i> , <i>Picea abies</i> grow below canopy. LSC was formed by shrubs (<i>Vaccinium</i> sp. and <i>Caluna vulgaris</i>), mosses (<i>P. schreberi</i> , <i>Dicranum</i> sp., <i>H. splendens</i>), and graces (<i>Deschampsia flexuosa</i> , <i>Melampyrum sylvaticum</i>).	IV–X.2015	Monoterpenes flux range from the forest floor was 0.40–221.0 $\mu\text{g} \cdot \text{m}^{-2} \cdot \text{h}^{-1}$ (average 23 $\mu\text{g} \cdot \text{m}^{-2} \cdot \text{h}^{-1}$); the most dominant compounds: α -pinene and 3-carene. Sesquiterpene flux range was 0.01–10.9 $\mu\text{g} \cdot \text{m}^{-2} \cdot \text{h}^{-1}$; the most abundant waere β -caryophyllene and aromadendrene. Isoprene flux range was 0.005–99.8 $\mu\text{g} \cdot \text{m}^{-2} \cdot \text{h}^{-1}$	Mäki et al., 2017
Static chamber technique. Concentration on sorption tubes(Tenax TA). GC-MS analysis after thermal desorption.	VOC emissions from forest floor were studied in a latitudinal (over a 1000 km South-North transect) provenance experiments. The provenances studied were Saarema (58° 22' N), Korpilahti (62° 0' N), Suomussalmi (65 ° 10') and Muonio	VIII.2010	The main compounds emitted by forest floor were α -pinene, 3-carene, and camphene. Average unstandardised VOC emission rate from forest floor was 50 $\mu\text{g} \cdot \text{m}^{-2} \cdot \text{h}^{-1}$. Total emission rates were 65% higher in Korpilahti than in Muonio. High emission rates were explained by the high	Kivimäenpää et al., 2018



	(67° 56' N).		amount of decomposing needle litter and low moss coverage. α -Pinene and 3-carene contributed a major fraction of the emissions from the forest floor.	
Dynamic soil chambers. . Concentration on sorption tubes(Tenax TA + Carborgraph 1). GC-MS analysis after thermal desorption.	Norway spruce and Scots pine 120yr old forests in Sweden.	VI–X.2015	The VOC emissions from the forest floor ranged from 0.4 to 66.6 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$. The emission rates peaked in October when the air temperature was $<10^\circ\text{C}$. However, monoterpene emission rate increased exponentially with air temperature inside the soil chambers. It was concluded that needle litter may be an important source of VOC emission.	Wang et al., 2018
Glass steady-state flow-through chambers. . Concentration on sorption tubes (Tenax TA + Carbopack B). GC-MS analysis after thermal desorption.	(1) 60 yr old <i>P. sylvestris</i> and <i>P. abies</i> forests in Finland (boreal, 61° 51' N, 24° 17' E); (2) <i>P. sylvestris</i> , <i>P. abies</i> , <i>Betula</i> sp. forests Estonia (hemiboreal, 58° 25' N, 27° 46' E); (3) 100 yr old <i>P. abies</i> forest; SMEAR II rural background site.	V–X. 2017; IV–VI.2018	The total terpene fluxes were higher from the Scots pine forest floor (boreal 23 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$) and from the mixed forest floor (hemiboreal, 32 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$) compared to the Norway spruce forest floor in both boreal (12 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ and hemiboreal (9 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$) climates. Due to higher litterfall production, the forest floor seems to be a greater source of monoterpenoids and sesquiterpenes in the hemiboreal mixed stand dominated by Scots pine compared to the boreal Scots pine stand.	Mäki et al., 2019a
Static flux chambers. Concentration on sorption tubes (Tenax TA + Carbopack B). GC-MS analysis after thermal desorption.	Measurement of monoterpene emissions every 2 months over 2 years from logged tropical forests and oil palm plantation floor in Malaysian Borneo (4°49' N, 116°54' E).	I. 2015–XI. 2016	The main emitted monoterpenes were α - and β -pinene and limonene. Mean α -pinene fluxes were around 2.5-3.5 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ from the forest floor with occasional fluxes exceeding 100 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$. Fluxes from the oil palm plantation were on average 0.5-2.9 $\mu\text{g}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$ and only	Drewer et al., 2021



			higher when litter was present. No seasonal trends could be identified for all monoterpenes (the climate is wet tropical with a wet season typically from October to February and a dry season typically from March to September).	
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1377 **Table 6.** Laboratory measurements of emission rates of monoterpene hydrocarbons from
 1378 fresh and decaying litter of Norway spruce and Scots pine (Isidorov et al., 2010) as well as
 1379 from fresh litter of European larch, *Larix decidue* (Isidorov et al., 2005).

Terpene	Emission rate at 20°C (µg·g ⁻¹ ·h ⁻¹ , dw) vs. decomposition time (days)										Larch litter*
	Spruce litter					Pine litter					
	0	77	165	282	490	0	77	165	282	490	
α-Pinene	0.06	0.56	0.44	0.16	0.09	1.15	5.13	1.01	0.18	0.06	0.50
β-Pinene	0.02	0.07	0.06	trace	trace	0.09	0.24	0.04	0.01	trace	0.18
Camphene	0.10	0.63	0.76	0.28	0.10	0.18	0.47	0.15	0.03	0.04	0.10
3-Carene	-	0.03	0.02	0.02	trace	0.53	1.65	0.42	0.10	0.10	-
Limonene	0.02	0.16	trace	0.05	0.02	0.01	0.03	0.02	trace	trace	0.11
Terpinolene	0.01	-	-	-	-	0.01	0.02	-	-	-	-
Myrcene	-	-	-	-	-	-	-	-	-	-	0.05
β-Phellandrene	-	-	-	-	-	-	-	-	-	-	0.12
Total	0.20	1.46	1.28	0.51	0.21	1.96	7.54	1.64	0.32	0.20	1.08

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* Emission at 22°C.



1390 **Table 7.** Emission rates of volatile organic compounds ($\text{ng g}^{-1} \text{h}^{-1}$) from litter of nine terpene
 1391 storing species. Values are mean \pm SE ($n=3$ to 10) (Viros et al., 2020).
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Compound	<i>Cistus albidus</i>	<i>Cistus salviifolius</i>	<i>Cotinus coggygria</i>	<i>Eucalyptus globulus</i>	<i>Juniperus oxycedrus</i>	<i>Pinus halepensis</i>	<i>Pinus pinea</i>	<i>Rosmarinus officinalis</i>
α -thujene	-	-	-	-	-	4.35 \pm 1.11	-	-
α -pinene	2.99 \pm 0.21	-	181.64 \pm 43.11	4.26 \pm 1.86	6.17 \pm 2.14	61.63 \pm 10.61	24.85 \pm 18.61	197.91 \pm 30
camphene	-	-	27.37 \pm 4.44	-	-	-	1.34 \pm 0.66	155.84 \pm 25.51
sabinene	-	-	-	-	3.19 \pm 0.3	36 \pm 9.3	-	-
β -pinene	-	-	17.38 \pm 4.46	2.85 \pm 0.58	-	-	15.75 \pm 0.58	31.03 \pm 6.98
myrcene	-	-	250.3 \pm 69.87	1.39 \pm 0.3	5.46 \pm 0.65	137.67 \pm 24.79	-	-
α -phellandrene	-	-	-	3.47 \pm 0.65	0.67 \pm 0.17	-	13.68 \pm 1.01	-
δ -3-carene	-	-	0.84 \pm 0.2	-	-	5.7 \pm 1.35	7.53 \pm 2.5	-
α -terpinene	-	-	1.69 \pm 0.85	-	-	-	-	-
limonene	142.94 \pm 28.14	5.15 \pm 1.14	1591.51 \pm 315.27	2.78 \pm 1.01	14.21 \pm 0.09	128.95 \pm 21.37	226.98 \pm 5.3	-
eucalyptol	-	-	-	82.85 \pm 68.64	-	-	-	-
<i>cis</i> - β -ocimene	-	-	686.08 \pm 217.7	-	2.95 \pm 0.15	3.07 \pm 0.41	3.16 \pm 0.05	-
<i>trans</i> - β -ocimene	-	-	148.33 \pm 49.28	-	0.99 \pm 0.13	5.89 \pm 1.72	1.49 \pm 0.07	-
γ -terpinene	-	-	6.71 \pm 2.49	-	2.49 \pm 0.13	14.14 \pm 2.5	1.26 \pm 0.14	-
<i>trans</i> -4-thujanol	-	-	-	-	-	-	-	-
α -terpinolene	-	2.75 \pm 0.68	77.93 \pm 20.61	-	5.43 \pm 0.18	42.58 \pm 11.77	3.58 \pm 2.41	-
linalool	-	5.02 \pm 2.07	-	-	2.03 \pm 0.22	-	1.71 \pm 0.08	-
α -campholenal	-	-	-	-	1.04 \pm 0.15	-	-	-
<i>allo</i> -ocimene	-	-	72.56 \pm 22.92	-	-	-	-	-
<i>cis</i> - <i>p</i> -mentha-2,8-dien-1-ol	-	-	-	-	-	-	-	-

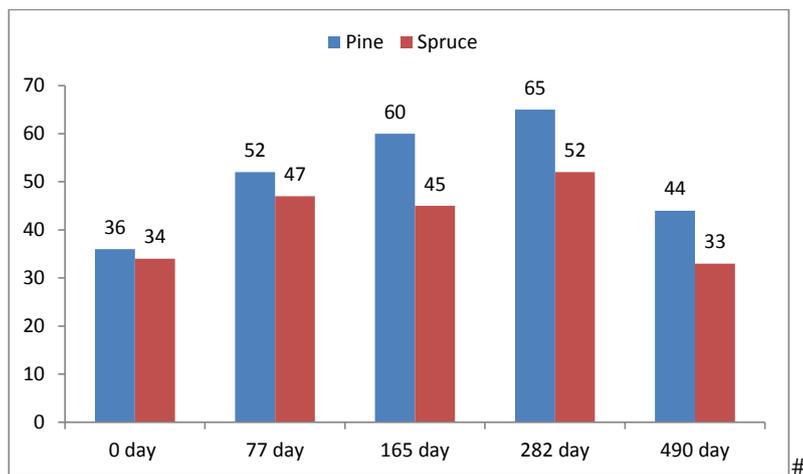
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1404 **Table 8.** Emission rates of compounds ($\text{ng g}^{-1} \text{h}^{-1}$) from litter of seven non-terpene storing
 1405 species. Values are mean \pm SE (n=3 to 10) (Viros et al., 2020).

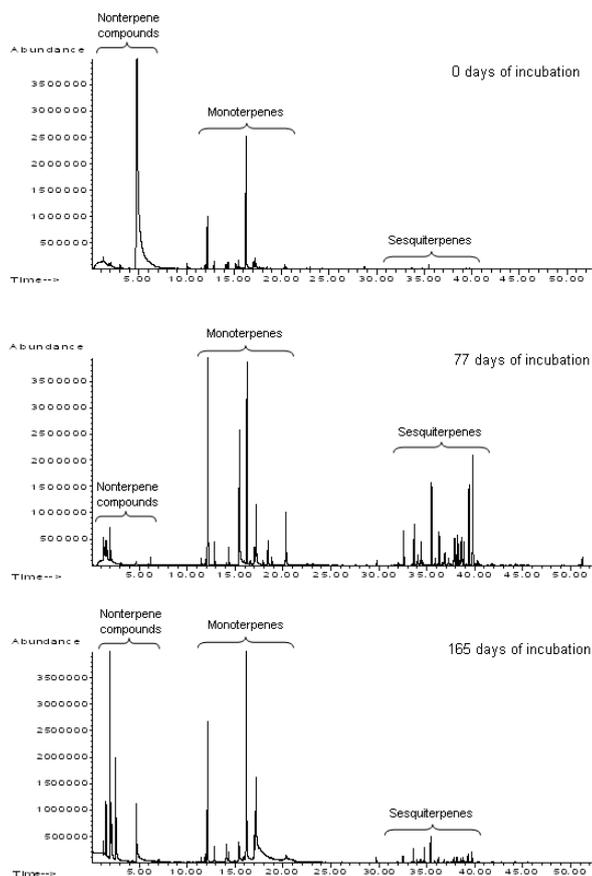
Compound	<i>Acer monspessulanum</i>	<i>Erica arborea</i>	<i>Quercus coccifera</i>	<i>Quercus ilex</i>	<i>Quercus pubescens</i>	<i>Quercus suber</i>	<i>Ulex parviflorus</i>
Aliphatic carbonyls							
- benzaldehyde	-	4.06 ± 0.22	-	-	-	-	-
- 6-methyl-5-hepten-2-one	0.59 ± 0.20	10.61 ± 0.55	6.08 ± 2.70	0.9 ± 0.54	-	2.45 ± 0.49	-
- octanal	-	-	-	-	-	0.20 ± 0.31	0.84 ± 0.02
- nonanal	2.76 ± 0.18	11.85 ± 0.39	27.32 ± 3.90	1.48 ± 0.74	55.37 ± 13.61	3.84 ± 0.84	4.65 ± 0.29
- decanal	0.54 ± 0.06	1.31 ± 0.86	32.19 ± 5.3	-	52.7 ± 10.87	1.05 ± 0.38	1.96 ± 0.08
- 2,4-heptadienal. (E,E)	-	2.16 ± 0.58	-	-	-	-	-
Alkane and alkene							
- undecane	-	-	2.32 ± 0.45	-	6.56 ± 3.52	-	-
- dodecane	1.67 ± 0.10	4.66 ± 0.39	4.97 ± 0.77	5.19 ± 2.62	6.86 ± 3.96	9.49 ± 1.78	1.03 ± 0.01
- tridecane	0.32 ± 0.14	1.59 ± 0.1	26.03 ± 5.68	0.98 ± 0.49	47.08 ± 22.19	2.88 ± 0.40	0.91 ± 0.07
- 1-tetradecene	-	0.55 ± 0.28	-	-	-	-	-
- tetradecane	0.52 ± 0.21	-	87.21 ± 7.09	-	163.19 ± 50.51	-	1.48 ± 0.16
- 1-pentadecene	-	-	-	-	29.62 ± 9.91	-	-
- pentadecane	-	-	38.42 ± 10.26	-	217.48 ± 93.11	-	-
- hexadecane	-	0.54 ± 0.27	17.45 ± 1.99	-	26.42 ± 6.04	-	1.52 ± 0.02
Other							
- 3-methyl-2(5H)-furanone	-	-	82.59 ± 10.63	-	-	-	-
- phenyl ethyl alcohol	-	3.58 ± 0.56	-	-	-	-	1.05 ± 0.11
Total emission, $\mu\text{g g}^{-1} \text{h}^{-1}$	0.01 ± 0.01	0.04 ± 0.01	0.32 ± 0.03	0.01 ± 0.01	0.61 ± 0.16	0.02 ± 0.01	0.01 ± 0.01

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Fig. 1. Changes in the number of different VOCs registered in emissions from decomposing Scots pine and Norway spruce needles (Isidorov et al., 2010).



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1437 **Fig. 2.** Changes in the profiles of volatile emissions of pine litter at the first stages of

1438 decomposition in natural conditions. HS-SPME/GC-MS chromatograms from pine needle

1439 litter emissions concentrated on polydimethylsiloxane (PDMS) fibre (Isidorov et al., 2010).

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