



1 **Assessing global-scale organic matter reactivity patterns in marine**
2 **sediments using a lognormal reactive continuum model**

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24 **Abstract**

25 Organic matter (OM) degradation in marine sediments is largely controlled by its reactivity
26 and profoundly affects the global carbon cycle. Yet, there is currently no general
27 framework that can constrain OM reactivity on a global scale. In this study, we propose a
28 reactive continuum model based on a lognormal distribution (*l*-RCM) that is fully
29 described by the mean μ and standard deviation σ of the sedimentary OM reactivity
30 distribution. We use the *l*-RCM to inversely determine μ and σ at 123 sites across the global
31 ocean. The results find that the apparent OM reactivity ($\langle k \rangle = \mu \cdot \exp(\sigma^2/2)$) decreases with
32 decreasing sedimentation rate (ω) and show that OM reactivity is more than three orders
33 of magnitude higher in shelf than that in abyssal regions. Despite the general global trends,
34 higher than expected OM reactivity is observed in certain deeper ocean regions, such as
35 the Eastern-Western Coastal Equatorial Pacific and the Arabian Sea, emphasizing the
36 complex control of the depositional environment (e.g., OM flux, oxygen content in the
37 water column) on benthic OM reactivity. Notably, the *l*-RCM can also highlight the
38 variability of OM reactivity in these regions. Based on inverse modeling results in our
39 database, we establish the significant statistical relationships between $\langle k \rangle$ and ω , and
40 further map the global OM reactivity distribution.

41

42 **1 Introduction**

43 Marine sediments act as the ultimate sink for organic carbon. The size and reactivity of the
44 benthic organic matter (OM) reservoir is a critical component of the global carbon cycle
45 (Arndt et al., 2013). In particular, the reactivity of benthic OM imposes a substantial control
46 on the magnitude of benthic carbon sequestration over geological timescales, the recycling



47 of inorganic carbon and nutrients, the dissolution and precipitation of carbonates, the
48 production of methane, and the activity of the seafloor biosphere (Dickens et al., 2004;
49 Boudreau, 1992). Decades of research have shown that OM reactivity is controlled by both
50 the nature of the OM (origin, composition and degradation state), as well as its
51 environmental and depositional conditions (e.g., redox conditions, sedimentation rate,
52 mineral protection, microbial community composition and biological mixing) (Burdige,
53 2007; Egger et al., 2018; Larowe et al., 2020a). However, due to the complex and dynamic
54 nature of the main controls on OM reactivity, the specific relative significance of these
55 controlling factors remains poorly quantified. Consequently, OM degradation models
56 generally do not explicitly describe the influence of environmental and depositional factors
57 on OM reactivity and its evolution but rather apply simplified parametrizations (Pika et al.,
58 2021). Over the past decades, several models have been developed and successfully used
59 to quantify OM degradation in marine sediments. They can be broadly divided into two
60 groups: discrete models, such as the (multi) *G* model (Berner, 1964; Jørgensen, 1978), and
61 continuum models, such as the reactive continuum model (RCM) and the power model
62 (Boudreau and Ruddick, 1991; Middelburg, 1989).

63 Discrete models divide the bulk OM pool into a number of discrete fractions, each with
64 its own constant reactivity. The 1-*G* model is the earliest OM degradation model. It is based
65 on the assumption that OM degrades according to first order dynamics with a single
66 constant degradation rate constant (Berner, 1964). The multi-*G* model, on the other hand,
67 divides OM into several fractions, and each fraction is degraded according to a first-order
68 rate with a fraction-specific reactivity (Jørgensen, 1978). Although multi-*G* models
69 successfully fit observed OM degradation dynamics when comprehensive data sets are



70 available, their application on a global scale is complicated by the need to partition the OM
71 reactivity into a finite number of fractions and define their reactivities. A multi-*G* model
72 with *n* discrete OM fractions requires constraining $2n-1$ parameters and is, thus, over-
73 parametrized (Jørgensen, 1978). Nevertheless, because of its mathematical simplicity and
74 wide use, multi-*G* models have been used in a range of diagenetic models designed for the
75 global/regional scale (e.g., CANDI, MEDIA, MEDUSA, and OMEN SED) (Boudreau,
76 1996; Meysman et al., 2003; Munhoven, 2007; Pika et al., 2021). Constraining the $2n-1$
77 OM degradation model parameters for these global-scale applications is not
78 straightforward. Early strategies for constraining the reactivity of OM on a global scale
79 have focused on deriving empirical relationships between OM reactivity and single, easily
80 observable characteristics of the depositional environment (water depth, sedimentation
81 rate, or OM flux) (Arndt et al., 2013). However, no statistically significant link between
82 OM reactivity and depositional environment could be established ($R^2 < 0.1$) after compiling
83 published multi-*G* models parameters across a wide range of depositional environments,
84 model complexities as well sediment depths/ burial time scales (Arndt et al., 2013).

85 Continuum models are an alternative to discrete models. They assume that OM
86 compounds are continuously distributed over a wide range of reactivities. The degradation
87 rate can be described as the sum of an infinite number of discrete fractions, each degraded
88 according to first-order kinetics (Boudreau and Ruddick, 1991), as

$$89 \quad G(t) = \int_0^{\infty} G(0) \cdot g(k, 0) \cdot e^{-kt} dk \quad (1)$$

90 where $G(t)$ is OM content at time t , $G(0)$ is OM content at the sediment-water interface
91 (SWI), k is the first-order degradation rate constant, and $g(k,0)$ is the initial reactivity
92 distribution of OM at the SWI. Boudreau and Ruddick (Boudreau and Ruddick, 1991),



93 following Aris (Aris, 1968) and Ho *et al.* (Ho and Aris, 1987), proposed to use a Gamma
94 distribution (γ -RCM) due to its mathematical properties and its ability to capture the
95 observed dynamics:

$$96 \quad g(k, 0) = \frac{a^v \cdot k^{v-1} \cdot e^{-ak}}{\Gamma(v)} \quad (2)$$

97 where a is the average age of the OM at the SWI, v is the shape parameter, and $\Gamma(v)$ is the
98 Gamma function. In addition, Middelburg (Middelburg, 1989) empirically derived a power
99 law from a large data compilation of measured OM reactivity, which is mathematically
100 equivalent to the γ -RCM. The advantage of the continuum models over the discrete models
101 is that they merely require constraining two free parameters to capture the widely observed
102 continuous decrease in OM reactivity with degradation time/depth. Recently, γ -RCM has
103 been used to inversely determine the free γ -RCM parameters, and thus benthic OM
104 reactivity, from observed POC and sulfate depth profiles across a wide range of different
105 depositional environments (Freitas *et al.*, 2021). Although results revealed broad global
106 patterns, no significant statistical relationship ($R^2 < 0.46$) between the parameters (a and v)
107 of the γ -RCM (Arndt *et al.*, 2013) and characteristics of the depositional environment could
108 be found, and constraining OM degradation model parameters on the global scale thus
109 remains difficult.

110 Here, we present an RCM based on a lognormal distribution (l -RCM) (Forney and
111 Rothman, 2012b):

$$112 \quad g(k, 0) = \frac{1}{k \cdot \sigma \cdot \sqrt{2\pi}} \cdot e^{-(\ln k - \ln \mu)^2 / (2\sigma^2)} \quad (3)$$

113 where $\ln \mu$ is the mean of $\ln k$, and σ^2 is the variance of $\ln k$ (Fig. 1C). The lognormal
114 distribution is commonly observed in nature (e.g., the radioactivity of elements in the crust,
115 the incubation period of infectious diseases, ecological species abundance) (Limpert *et al.*,



116 2001), and the rates of ocean primary production and biological carbon export also follow
117 a lognormal distribution (Cael et al., 2018). In addition, inverse modeling of OM
118 degradation data from 27 different types of litter, including leaves, wood, grass, and wheat,
119 distributed across North America, ranging from the Alaskan tundra to the Panamanian
120 rainforests, showed that OM reactivity followed a lognormal distribution (Forney and
121 Rothman, 2012b). In this study, we use the *l*-RCM to simulate sedimentary OM profiles
122 compiled from 123 sites covering the global depositional conditions. On this basis, we
123 establish empirical relationships between OM reactivity versus sedimentation rate and
124 further map the distribution of global sedimentary OM reactivity at the SWI.

125 **2 Materials and methods**

126 **2.1 OM degradation model approach**

127 The $g(k,0)$ we used in equation (1) is the lognormal distribution (equation (3)). Because of
128 the tail of $g(k,0)$, the mean degradation rate or the apparent rate of the bulk OM ($\langle k \rangle$) is
129 greater than the median μ , as follows:

$$130 \quad \langle k \rangle = \int_0^{\infty} k \cdot g(k,0) dk = \mu \cdot e^{\sigma^2/2} \quad (4)$$

131 **2.2 Inverse model approach**

132 Here, we used 123 published datasets of OM depth profiles across a wide range of different
133 depositional environments that have been sourced from published literature (Middelburg,
134 1989; Arndt et al., 2013; Middelburg et al., 1997) and the IODP database (Supplementary
135 Fig. S1) to inversely determine the μ and σ parameters. We also analyzed a small number
136 ($n=12$) of laboratory experiment data on OM degradation (Middelburg, 1989), as well as
137 OM degradation data ($n=16$) from terrestrial soils (Katsev and Crowe, 2015). We followed
138 the inverse modeling approach by Forney *et al.* (Forney and Rothman, 2012a) to identify



139 the best-fitting parameters μ and σ based on the Newton method. We further improved this
140 inverse method to reduce the influence of the initial value through nonlinear polynomial
141 fitting using the least-squares method (Motulsky and Christopoulos, 2004).

142 **2.3 Global upscaling of sedimentation rate**

143 The inversely determined μ , σ couples of all investigated marine sites were then used in a
144 linear regression method to derive the empirical relationships between OM parameters μ ,
145 σ , $\langle k \rangle$ and the local sedimentation rates (ω). A correction factor was applied to account for
146 the skewness bias inherent in the back conversion from a log-log transformed linear
147 regression model to arithmetic units. The newly derived empirical relationships between
148 $\langle k \rangle$ and ω (Fig. 2) were then used to calculate global maps of OM reactivity at the SWI on
149 a $1^\circ \times 1^\circ$ grid cell of the world ocean. At each grid point, ω were estimated based on the
150 empirical relationship between ω (ω in cm yr^{-1}) and the water depth (z in m) (equation (5)),
151 derived from 260 observations on the global continental shelves (Burwicz et al., 2011),
152 complemented here by an extra 360 sites including abyss regions (data from Arndt *et al.*
153 (Arndt et al., 2013), Egger *et al.* (Egger et al., 2018)).

$$154 \quad \omega(z) = \frac{0.1}{1 + \left(\frac{z}{200}\right)^{3.3}} + \frac{0.001}{1 + \left(\frac{z}{4500}\right)^{11.4}} \quad (5)$$

155 **2.4 Regional OM reactivity distribution at the SWI**

156 Unfortunately, the OM depth profile has not been measured and compiled for the global
157 seafloor at a suitable spatial resolution. The empirical equations we obtained for the
158 depositional environment (ω) and OM reactivity can be used to assess the distribution of
159 OM reactivity on a regional and global scale. In addition, the continuous distribution of
160 OM reactivity over the reactivity spectrum, $F_i(k,0)$, provides crucial quantitative
161 information on the degradation process and its evolution during burial.



$$162 \quad F_i(k, 0) = \sum_{i=1}^n \frac{G_i(0) \cdot g_i(k, 0)}{\sum_{i=1}^n G_i(0)} \quad (6)$$

163 where $G_i(0)$ is the content of OM at the SWI of each grid cell, and $g_i(k, 0)$ is its reactivity
164 distribution. Referring to Seiter *et al.* (Seiter *et al.*, 2004), we divided the ocean into 30
165 regions based on the distribution of OM content in global marine surface sediments (Fig.
166 3). Using surface OM data at 5600 sites (Supplementary Fig. S11) (Seiter *et al.*, 2004) and
167 the relationship between water depth, μ , σ , and ω (Supplementary Fig. S9 and S10), we
168 estimated the distribution of OM reactivity at the SWI, $F_i(k, 0)$, in the 30 regions we divided
169 (Supplementary Fig. S12).

170 We use a multi- G approximation method to calculate $\langle k_i \rangle$ in each region (Bradley *et al.*,
171 2020; Larowe *et al.*, 2020b; Freitas *et al.*, 2021). According to $F_i(k, 0)$, the bulk OM at the
172 SWI is divided into 1000 distinct fractions, and $\langle k_i \rangle$ was calculated as:

$$173 \quad \langle k \rangle = \int_0^{\infty} k \cdot F_i(k, 0) dk = \sum_{i=1}^{1000} f_i \cdot k_i \quad (7)$$

174 where f_i is the initial proportion of total OM in fraction i . The solution of f_i can be found in
175 Supplementary equation (S3)–(S4).

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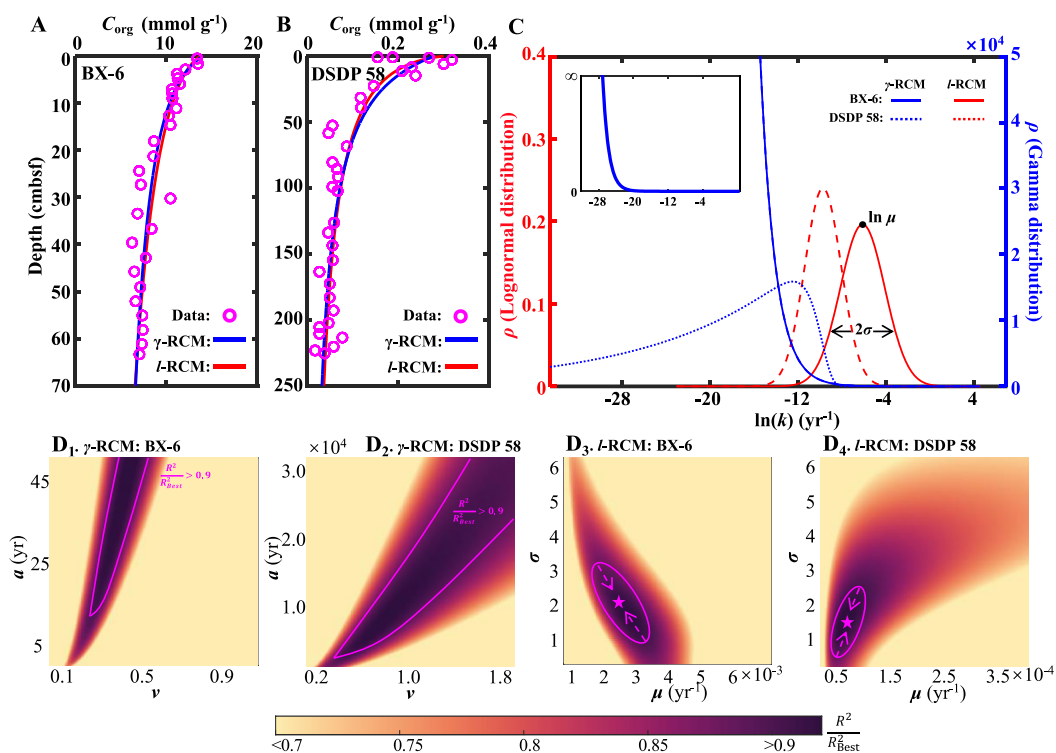
177 **3 Results and discussion**

178 **3.1 OM reactivity distribution based on the l -RCM and the γ -RCM**

179 First, we determined the best fit to the eight OM datasets reported by Westrich *et al.*
180 (Westrich and Berner, 1984), using the l -RCM and the γ -RCM. Both models fit the data
181 equally well, as illustrated by the high coefficient of determination for each fit ($R^2 > 0.9$,
182 Supplementary Fig. S3). However, the l -RCM and the γ -RCM differ in their ability to find
183 a unique solution and in their respective probability density functions of OM reactivity
184 ($\rho(k)$). For example, Fig. 1A and 1B show the best-fit OM profiles for two contrasting sites:



185 BX-6 in the shelf and DSDP 58 in the abyssal region. The inversely determined parameters
186 at the two sites are $\mu=2.23\times 10^{-3}\text{ yr}^{-1}$, $\sigma=2.03$ at BX-6, and $\mu=6.11\times 10^{-5}\text{ yr}^{-1}$, $\sigma=1.66$ at DSDP
187 58 by the *l*-RCM. At BX-6, the best-fitting parameters by the γ -RCM are $\nu=0.278$ and
188 $a=22.5$, and at DSDP 58, $\nu=1.08$ and $a=20224$. According to the parameter sensitivity
189 analysis (Fig. 1D), the R^2 of the fitted results remains greater than 0.9 when a and ν change
190 substantially simultaneously. As a result, different combinations of a and ν can fit the data
191 equally well. For example, simultaneously increasing ν and a ($\nu=0.5$ and $a=53$) at site BX-
192 6 or decreasing ν and a ($\nu=0.5$ and $a=4024$) at site DSDP 58 lead to a slight change in R^2 .
193 Freitas *et al.* (Freitas *et al.*, 2021) showed that adding additional observational data, such
194 as porewater depth profiles, can help find a unique solution. In contrast, the best-fit
195 parameters μ and σ are unique in the *l*-RCM, and even small changes in either parameter
196 can lead to abysmal fitting results (Fig. 1D). The second difference between the two models
197 concerns the shape of the probability distribution $\rho(k)$. Statistically, the features of the
198 Gamma distribution vary with the value of ν . It is divergent when $\nu<1$ and convergent when
199 $\nu>1$, making it difficult to visually compare the reactivity distributions of site BX-6 and
200 DSDP 58 (Fig. 1C). Compared to OM reactivity described by the γ -RCM, the *l*-RCM can
201 better distinguish OM reactivity distribution at different sites (Fig. 1C).



202

203 **Figure 1. Comparison of *l*-RCM and γ -RCM.** A, B, the fitting results of the *l*-RCM and
 204 the γ -RCM for site BX-6 and DSDP 58. C, OM reactivity distribution from *l*-RCM and γ -
 205 RCM. Top inset, Gamma distribution at site BX-6 at a larger y-axis. D, Distribution of
 206 R^2/R_{Best}^2 for parameter sensitivity analysis of the γ -RCM and the *l*-RCM at sites BX-6 and
 207 DSDP 58. The pink lines in the D₁ and D₂ denote the range that $R^2/R_{Best}^2 > 0.9$ in the γ -RCM.
 208 The R^2/R_{Best}^2 in the *l*-RCM converges as the pink arrows in the D₃ and D₄, ultimately
 209 reaching the best fitting results as the pink pentagrams.

210

211 3.2 Global distribution of OM reactivity

212 The inverse determination of the *l*-RCM parameters μ and σ across the wide range of
 213 different depositional environments allows quantitative insights into OM reactivity and
 214 provides essential information on the main environmental controls on OM reactivity. Fig.
 215 2A illustrates the inversely determined μ - σ for all 123 depth profiles of marine sediment



216 POC investigated in this study and compares them with inversely determined parameters
217 from published soil and laboratory incubation data. It highlights the large inter- and
218 intraregional variability of best-fit μ (10^{-6} – 10^2 yr⁻¹) and σ (0.2–6). However, despite the
219 large variability, it also reveals broad global patterns in μ and σ and thus OM reactivity
220 ($\langle k \rangle$) (equation (4)). Notably, best-fit μ - σ couples form environmental clusters along a μ
221 gradient, with the highest μ being determined for laboratory degradation experiments of
222 fresh phytoplankton (Garber, 1984; Westrich and Berner, 1984) ($\mu=10^0$ – 10^2 yr⁻¹), followed
223 by soil incubation under natural (Katsev and Crowe, 2015), yet still idealized conditions
224 ($\mu=10^0$ – 10^1 yr⁻¹), while OM degraded in marine sediments generally reveals lower
225 inversely determined $\mu < 10^0$ yr⁻¹. The higher μ values determined for soil OM seemingly
226 contradict the widely accepted notion that soil OM is generally less reactive than marine
227 OM (Larowe et al., 2020a; Zonneveld et al., 2010). However, this apparent contradiction
228 can be explained by the idealized conditions of the incubation experiments (e.g., only one
229 type of material, some of which had nitrogen added), as well as the degradation state of the
230 investigated OM. Although soil OM is structurally less reactive (Zonneveld et al., 2010;
231 Hedges and Keil, 1995), the soil incubation experiments were conducted with initially
232 undegraded material. In contrast, OM deposited in marine sediments consists of a complex
233 mixture of OM from autochthonous and allochthonous sources that is altered to various
234 degrees during transit from its source to the sediment (Hewson et al., 2012).

235 In addition to the difference between incubation data and field observations, Fig. 2A also
236 reveals a three order of magnitude decrease in inversely determined μ for OM from the
237 shelf (10^{-3} – 10^{-1} yr⁻¹) to the slope (10^{-4} – 10^{-3} yr⁻¹), and ultimately abyssal regions ($< 10^{-4}$ yr⁻¹).
238 In addition, shelf and slope regions also generally reveal a larger σ (1–3), while abyssal



239 regions display a narrower σ range (0.5–1). This observed progressive decrease in μ and σ
240 from the shelf to the abyssal ocean confirms previously observed patterns (Arndt et al.,
241 2013; Freitas et al., 2021; Zonneveld et al., 2010) and reflects the interaction between OM
242 structure (or its source) and the degree of alteration/pre-processing as OM transits from its
243 original source to the ultimate sedimentary sink. In the dynamic shelf regions, highly
244 variable OM loads from different sources, including *in-situ* produced marine OM, laterally
245 transported, pre-processed terrestrial or marine OM, are often physically protected from
246 further erosion/deposition cycles due to high suspended sediment loads (Arndt et al., 2013;
247 Larowe et al., 2020a). As a result, benthic OM is composed of a complex mixture of fresh
248 and pre-aged compounds of highly variable (hence larger σ of the initial distribution), yet
249 generally higher reactivity. On the upper and mid-continental slopes, intensive lateral
250 and/or vertical transport processes or the abrupt relocation of sediment result in similar
251 complex mixtures of OM (hence similar σ of the initial distribution) (Larowe et al., 2020a).
252 However, transport timescales are often longer due to the greater water depths and distance
253 to land. The deposited OM is generally more degraded and thus less reactive than in shelf
254 environments. In contrast, benthic OM in abyssal regions is mainly derived from marine
255 production (Rowe and Staresinic, 1979; Larowe et al., 2020a). During its slow settling
256 through the water column, highly reactive OM compounds are rapidly degraded, and only
257 the less reactive compounds persist and settle onto the sediment (Dunne et al., 2007). The
258 values of μ and σ in the abyssal regions are thus significantly smaller than in the shelf and
259 slope regions. Especially in the NE-Pacific (NEPAC) region, deeper water depth
260 (>5000m), relatively low OM content (~0.2wt.%), and the old OM age (>10⁴ years) result
261 in comparably lower μ and σ values (Fig. 2A) and, thus, extremely low benthic OM



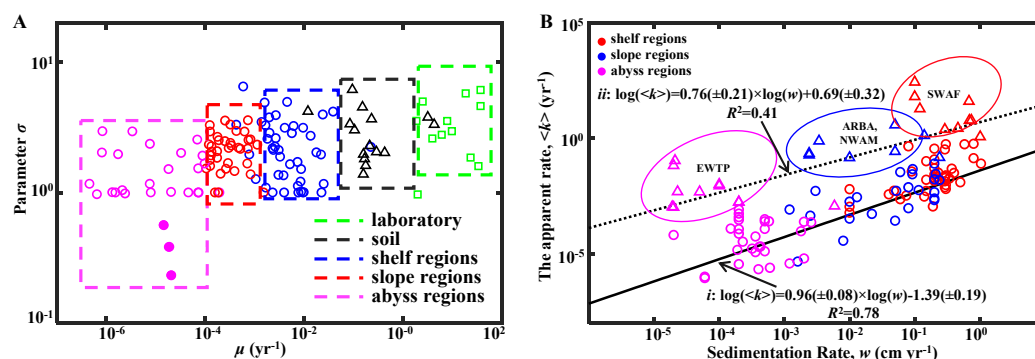
262 reactivity (Müller and Suess, 1979). The decrease of μ and σ from the shelf to abyssal
263 regions reveals a decline in reactivity during lateral transport of OM, where μ mainly
264 controls the overall reactivity and σ indicates the coverage of the main component of OM.

265 Fig. 2B shows the global trend of OM reactivity ($\langle k \rangle$) with sedimentation rate (ω),
266 which is a widely observed and comparably easy to measure proxy for local depositional
267 conditions with sizable global data sets or empirical formulas available (Burwicz et al.,
268 2011). In addition, Fig. 2B further reveals a general decreasing trend in $\langle k \rangle$ vs ω , and a
269 good regression ($R^2=0.78$) for the general sea regions (shelf (<200m), slope (200–2000m),
270 and abyss (>2000m)). However, the general trend is superimposed by a large variability
271 and apparent reactivity $\langle k \rangle$ in specific environments, notably deviating from this generally
272 observed trend. More specifically, higher μ and σ values and, thus, higher OM reactivities
273 occur in the Eastern-Western Coastal Equatorial Pacific (EWEP), Southwestern-Africa
274 continental margin (SWAF), Northwestern-America continental margin (NWAM), and the
275 Arabian Sea (ARBS) regions. These results are completely consistent with prior
276 observations and model results (Arndt et al., 2013) and can be directly linked to the
277 prevailing depositional conditions. High benthic OM reactivities have previously been
278 reported for depositional environments that are characterized by a dominance of marine
279 algal OM (e.g., EWEP) (Hammond et al., 1996), strong lateral transport processes (e.g.,
280 SWAF, NWAM)(Arndt et al., 2013). Furthermore, the reactivity of sedimentary OM is
281 considerably influenced by oxygen content or more precisely, by oxygen exposure time in
282 the water column and the bottom water (Aller, 1994; Hartnett et al., 1998; Hedges and Keil,
283 1995; Mollenhauer et al., 2003; Zonneveld et al., 2010). Lower oxygen concentrations, as
284 present in these regions in the form of pronounced oxygen minimum zones, will slow down



285 the degradation of OM both in the water column and at the sediment surface. This enables
 286 the burial of more reactive OM into the sediments and thus results in the occurrence of
 287 high sedimentary OM reactivity in these regions despite great water depth (Arndt et al.,
 288 2013; Bogus et al., 2012; Ingole et al., 2010; Luff et al., 2000; Volz et al., 2018). Our
 289 approach predicts higher OM reactivity in the Arabian Sea (ARBS) with deep water depth,
 290 that is consistent with the observation of one of the most pronounced ocean oxygen
 291 minimum zones in this region due to strong coastal upwelling or a high export rate of
 292 plankton-derived OM into the underlying water column. The *l*-RCM captures the broad
 293 patterns of OM reactivity across the global seafloor even better than previous models and
 294 also provides statistically more significant relationships between OM reactivity ($\langle k \rangle$) and
 295 sedimentation rate (ω) than inversely determined parameters of γ -RCM ($R^2 < 0.46$) and
 296 discrete models ($R^2 < 0.1$) (Arndt et al., 2013). Although the relationship between $\langle k \rangle$ and
 297 ω for special regions (e.g., EWEP, SWAF, NWAM, and ARBS) is less quality, it still
 298 provides an excellent first-order predictor and a step forward in assessing broad global
 299 patterns in OM reactivity.

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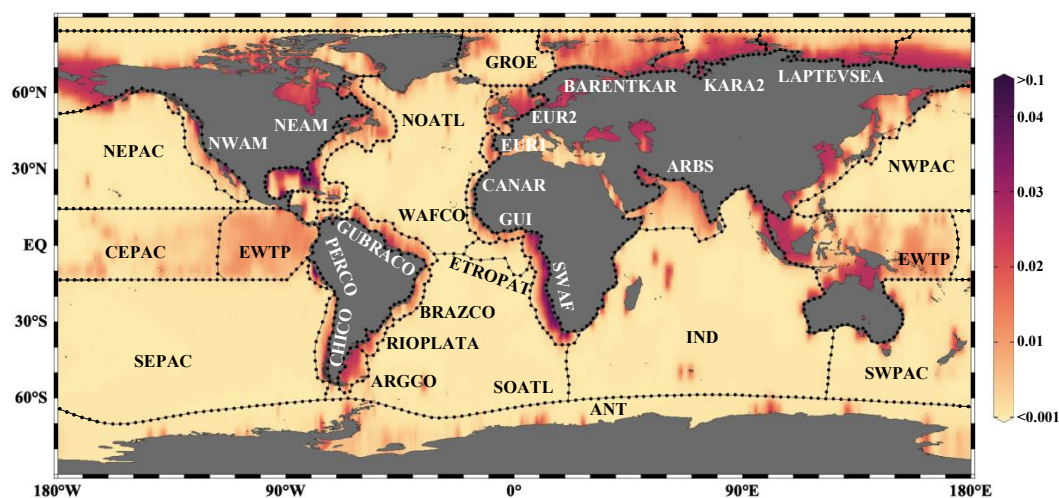
302 **Figure 2. Global distribution of OM reactivity.** **A**, Distribution of parameters σ and μ
303 in different regions. Pink solid circles denote fitting results of sites in the NEPAC with
304 extremely low OM reactivity. **B**, Log-log plot of ω and $\langle k \rangle$. The solid black line (*i*) denotes
305 linear regression for shelf, slope, and abyss regions. The black dotted line (*ii*) denotes linear
306 regression for high OM reactivity regions, including the EWTP, ARBS, NWAM, and
307 SWAF regions.

308

309 **3.3 Implication of Global OM reactivity patterns**

310 Considering the geographic differences in depositional environments and in order to
311 describe the global distribution of sedimentary OM reactivity in more detail, we divided
312 the global ocean into 30 different regions using 5600 single measured data of OM content
313 in global surface sediment (<5 cm sediment depth) and the previously used combined
314 qualitative and quantitative geostatistical methods (Seiter et al., 2004). Based on the
315 empirical $\langle k \rangle$ – ω relationships in Fig. 2B (*i* for the general water depth-related regions, *ii*
316 for the specific regions (EWTP, ARBS, NWAM, and SWAF)), and the water depth– ω
317 relationship (equation (5)), we finally derived, to our knowledge, the world’s first map of
318 the apparent reactivity of sedimentary OM distribution in the global ocean (Fig. 3).

319



320

321 **Figure 3. Distribution of $\langle k \rangle$ (yr⁻¹) at the SWI in the global ocean with 1°x1°**
322 **resolution and the 30 regions we divided of the global ocean. The abbreviations of the**
323 **30 regions we divided are defined in Table 1.**

324

325 Given OM degradation is closely related to its reactivity, this map has potentially
326 important applications for quantitative evaluation of the global OM degradation in
327 sediments and its related carbon budget. Previous work estimated the global budget of OM
328 degradation by the *G*-model or the γ -RCM from 1.314 Pg C yr⁻¹ to 3.127 Pg C yr⁻¹, with
329 ~85% occurring in the shelf regions (Jørgensen, 1983; Larowe et al., 2020b; Middelburg
330 et al., 1997; Smith and Hollibaugh, 1993). However, the γ -RCM describes the global-scale
331 OM reactivity patterns by a single empirical equation of parameters a - ω and parameter v
332 in a smaller range (0.1–0.2) (Larowe et al., 2020b; Arndt et al., 2013; Bradley et al., 2020)
333 and poor statistical relationship between the *G*-model parameters and depositional
334 environment, thus leading to underestimating OM degradation in special regions,
335 particularly in the EWTP, with large area and higher OM reactivity (Table 1, Fig. 3). In
336 addition, OM reactivity governs benthic biogeochemical cycling and exchange (Freitas et



337 al., 2021), and the dissolved inorganic carbon (DIC) generation via benthic OM
 338 degradation has a significant impact on ocean acidification and atmospheric CO₂ levels
 339 (Krumins et al., 2013). Therefore, the *I*-RCM can be further used to calculate the budget of
 340 OM degradation at regional or global scales and assess the significance of the sedimentary
 341 carbon cycle on the hydrosphere and atmosphere.

342

343

Table 1 Abbreviations of regions in this paper, and their area, mean OM content in surface sediment (<5 cm), and <k>.

Abbreviation	Region	Seafloor area ^a	Mean OM ^b	<k> ^c
		(×10 ¹² m ²)	(wt.%)	(yr ⁻¹)
SWAF	Southwestern-Africa continental margin	0.8	2.5	0.50936
NWAM	Northwestern-America continental margin	0.5	1.7	0.13596
ARBS	Arabian Sea	3.9	1.4	0.10302
EWTP	Eastern-Western Coastal Equatorial Pacific	35.1	1.2	0.00173
ANT	South Polar Sea	24.2	0.3	0.02557
ARGCO	Argentina continental margin	0.8	0.3	0.00020
BARENTKAR	Barents Sea and Kara Sea	1.7	1.1	0.02413
BRAZCO	Brazil continental margin	1.1	0.5	0.02662
CANAR	Canaries	0.6	0.6	0.01073
CEPAC	Central Equatorial Pacific	30.9	0.3	0.00004
CHICO	Chile continental margin	0.8	1.5	0.01620
ETROPAT	Eastern tropical Atlantic	1.8	0.7	0.00828
EUR1	Northern-European continental margin	2.3	0.8	0.03132
EUR2	Southern-European continental margin	1.7	0.3	0.02176
GROE	Northern Nordic Sea	4.8	0.7	0.00445
GUBRACO	Southeastern-America continental margin	1.6	0.4	0.02462
GUI	Gulf of Guinea	0.3	1.1	0.02678
INA	Indian Ocean deep sea	54.9	0.4	0.00012
KARA2	Kara Sea	1.6	1.2	0.03685
LAPTEVSEA	Laptev Sea	2.1	0.9	0.05052
NEAM	Northeastern-America continental margin	5.2	0.9	0.02907
NEPAC	Northeastern-Pacific	24.9	0.4	0.00011
NOATL	Northern Atlantic	30.3	0.4	0.00014
NWPAC	Northwestern-Pacific	15.6	0.6	0.00007
PERCO	Peru continental margin	0.6	4.8	0.01244



RIOPLATA	Rio de la Plata mouth	0.7	0.8	0.00778
SEPAC	Southeastern-Pacific	49.6	0.5	0.00018
SOATL	Southern Atlantic	39.3	0.4	0.00013
SWPAC	Southwestern-Pacific	22.3	0.8	0.00017
WAFCO	Western-Africa continental margin	0.9	0.6	0.00017

^aThe total ocean area covered in this study equals 3.60×10^8 km², corresponding to ~99% of the total ocean. ^bMean OM data were collected from Seiter *et al.* (Seiter *et al.*, 2004). ^cSee equation (7) for the solution of $\langle k \rangle$ in each region.

344

345 **4 Conclusions**

346 Compared with the previous OM degradation models, the *l*-RCM not only well fits OM
347 depth-profiles, but also better represents the distribution of OM reactivity by the parameters
348 μ and σ . We use the *l*-RCM to inversely determine μ and σ at 123 sites across the global
349 ocean, including the shelf, slope and abyss regions. Our results show that the apparent OM
350 reactivity ($\langle k \rangle = \mu \cdot \exp(\sigma^2/2)$) decreases with decreasing sedimentation rate (ω), and OM
351 reactivity is more than three orders of magnitude higher in shelf than that in abyssal regions.
352 Due to the complex depositional environments, such as oxygen minimum zones caused by
353 strong coastal upwelling or a high export rate of plankton-derived OM into the underlying
354 water column, OM reactivity is higher than predicted in some specific regions (e.g., the
355 NWAM, SWAF, ARBS, and EWTP regions), yet the *l*-RCM can still capture OM
356 reactivity features in these regions. Based on two empirical relationships between the OM
357 reactivity ($\langle k \rangle$) and sedimentation rate (ω), we obtained the global OM reactivity
358 distribution patterns and finally mapped the global OM reactivity distribution for the first
359 time. Overall, our *l*-RCM can potentially be used to calculate the budget of OM degradation
360 at regional or global scales and assess the significance of the sedimentary carbon cycle on
361 the hydrosphere and atmosphere.

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527 **Author contributions**

528 S.X. and B.L. designed the study and performed the research with S.A., S.K., and Z.W.;
529 All authors discussed the results and commented on the manuscript.

530 **Competing interests**

531 The authors declare that they have no competing interests.