

# Temporal patterns and ~~potential~~ drivers of CO<sub>2</sub> emission from dry sediments in a groyne field of a large river

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**Abstract.** River sediments falling dry at low water level are sources of CO<sub>2</sub> to the atmosphere. While the general relevance of CO<sub>2</sub> emissions from dry sediments has been acknowledged and some regulatory mechanisms identified, knowledge on mechanisms and temporal dynamics is still sparse. Using a combination of high frequency measurements and detailed studies we thus aimed to identify processes responsible for CO<sub>2</sub> emissions and to assess temporal dynamics of CO<sub>2</sub> emissions from dry sediments at a large German river.

CO<sub>2</sub> emissions were largely driven by microbial respiration in the sediment. Observed CO<sub>2</sub> fluxes could be explained by patterns and responses of sediment respiration rates measured in laboratory incubations. We exclude groundwater as a significant source of CO<sub>2</sub> because potential evaporation rates were too low to explain CO<sub>2</sub> fluxes by groundwater evaporation. Furthermore, CO<sub>2</sub> fluxes were not related to radon fluxes, which we used to trace groundwater derived degassing of CO<sub>2</sub>.

CO<sub>2</sub> emissions were strongly regulated by temperature resulting in large diurnal fluctuations of CO<sub>2</sub> emissions with emissions peaking during the day. The diurnal temperature – CO<sub>2</sub> flux relation exhibited a hysteresis which highlights the effect of transport processes in the sediment and makes it difficult to identify temperature dependence from simple linear regressions. The temperature response of CO<sub>2</sub> flux and sediment respiration rates in laboratory incubations was identical. Also deeper sediment layers apparently contributed to CO<sub>2</sub> emissions because the CO<sub>2</sub> flux was correlated with the thickness of the unsaturated zone, resulting in CO<sub>2</sub> fluxes increasing with distance to the local groundwater level and with distance to the river. Rain events lowered CO<sub>2</sub> emissions from dry river sediments probably by blocking CO<sub>2</sub> transport from deeper sediment layers to the atmosphere. Terrestrial vegetation growing on exposed

sediments ~~greatly~~<sup>largely</sup> increased respiratory sediment CO<sub>2</sub> emissions. We ~~conclude~~<sup>show</sup> that the regulation of CO<sub>2</sub> emissions from dry river sediments is complex. Diurnal measurements are mandatory and even CO<sub>2</sub> uptake in the dark by phototrophic micro-organisms has to be considered when assessing the impact of dry sediments on CO<sub>2</sub> emissions from rivers.

## 1 Introduction

### 1.1 CO<sub>2</sub> emissions from dry river sediments – significance

Streams and rivers are well known to play an important role in the global carbon cycle. The transport of continental carbon to the ocean is mainly regulated by rivers (Schlesinger and Melack 1981). Moreover, carbon in rivers undergoes transformation processes and can be stored by means of sedimentation and photosynthesis or released due to biological respiration (Battin et al. 2009). One distinctive feature of rivers is their frequently changing water levels. Climate change is expected to further increase the seasonal and the inter-annual variability of rivers and hydrological regimes (Bolpagni et al. 2019; Coppola et al. 2014). In Europe, more frequent and longer-lasting droughts are expected during summers, which lead to low water levels in ~~desiccation of smaller~~ streams and ~~low water levels in high order~~ rivers (Spinoni et al. 2018). Consequently, previously submerged river sediment will be exposed to the atmosphere and influenced by drying (Steward et al. 2012). It has been shown that these exposed sediments can emit high amounts of CO<sub>2</sub> (von Schiller et al. 2014) and may represent a globally relevant carbon source to the atmosphere (Marcé et al. 2019).

### 1.2 Regulation of CO<sub>2</sub> emissions from dry sediments

While the relevance of CO<sub>2</sub> emissions from dry river sediments has been acknowledged, only little is known about underlying mechanisms and temporal patterns. A recent study identified organic matter content and moisture as common drivers of CO<sub>2</sub> emissions from dry aquatic sediments (Keller et al. 2020). However, high variability prevents the prediction of CO<sub>2</sub> fluxes for particular sites. Case studies showed that CO<sub>2</sub> emissions are affected by temperature (Doering et al. 2011), emergent vegetation (Bolpagni et al. 2017), organic matter (Palmia et al. 2021), water content (Martinsen et al. 2019), or the frequency of

dry-wet cycles (Machado dos Santos Pinto et al. 2020). Although it is known that CO<sub>2</sub> emission from dry sediment may change with time, existing studies are based on single or few measurements. Few studies  
55 did address temporal variability of CO<sub>2</sub> emissions, but ~~Nothing~~ is yet known about short term dynamics of GHG emissions from dry aquatic sediments. Investigating temporal variability of CO<sub>2</sub> fluxes should provide information about the potential sources of emitted CO<sub>2</sub>. Knowing sources of emitted CO<sub>2</sub> from dry sediments is crucial to be able to model or scale up GHG emissions from these systems.  
~~Investigating temporal variability has the potential to gain knowledge about the sources of emitted CO<sub>2</sub> and will improve our ability to model or scale up GHG emissions from dry sediments.~~  
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### 1.3 Possible sources of CO<sub>2</sub>

Carbon emissions from desiccated sediments derive from a number of possible biotic and abiotic sources (Marcé et al. 2019). Microbial respiration is well known to contribute to CO<sub>2</sub> emissions (Weise et al. 2016), similar to soil respiration. Organic matter originating from organic particle sedimentation may be  
65 mineralized to CO<sub>2</sub> or CH<sub>4</sub>. It is typically observed that CH<sub>4</sub> emissions from dry sediments are low indicating that anaerobic mineralization plays a minor role (Marcé et al. 2019).

In contrast to respiration, abiotic processes are rarely taken into account as sources of CO<sub>2</sub> (Rey 2015). Yet, recent findings revealed a spatial variability of CO<sub>2</sub> fluxes from dry river sediments with highest fluxes near to the river ~~at groundwater seeping sites that raised the question, how abiotic processes might~~  
70 ~~contribute to observed CO<sub>2</sub> emissions. As a result,~~ (Mallast et al. 2020). As a possible explanation the authors hypothesized that at decreasing river water level a groundwater flow gradient towards the river would transport groundwater to the river (Peters et al. 2006). Groundwater is usually 10 to 100 fold super-saturated with CO<sub>2</sub> (Macpherson 2009). Near to the river the thickness of the unsaturated layer approaches zero and CO<sub>2</sub> rich groundwater reaches the surface sediment where CO<sub>2</sub> would eventually degas  
~~suggested that CO<sub>2</sub> evasion from groundwater is likely to contribute to CO<sub>2</sub> emissions from dry river sediments. This hypothesis is based on the mechanism that during periods of desiccation the water level of the river is falling, while the response of the groundwater level is delayed (Peters et al. 2006). Hence, a flow gradient towards the river is established, resulting in discharge hotspots of groundwater close to the river. Considering that groundwater is usually 10 to 100 fold over saturated with CO<sub>2</sub> (Macpherson 2009) the~~  
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80 ~~dissolved CO<sub>2</sub> can be expected to degas when reaching the sediment-atmosphere interface and probably adds to the CO<sub>2</sub>-emissions~~ (Mallast et al. 2020).

## 1.4 Aim of study

Given the uncertainty of the origin of CO<sub>2</sub> emitted from dry river sediments, in this study we aimed to test the hypothesis of (Mallast et al. 2020) that CO<sub>2</sub> emissions from dry sediments of larger rivers are  
85 driven by groundwater degassing. If groundwater was a significant source of CO<sub>2</sub> we hypothesize a only weak temperature dependence of CO<sub>2</sub> emissions. We applied a combination of automatic high frequency measurements and detailed studies using a variety of methods to identify the source of CO<sub>2</sub> emissions from dry sediments at a large German river and to understand their temporal dynamics and drivers.

## 2 Material and Methods

### 90 2.1 Study site

The study was conducted at the lowland part of river Elbe, one of the largest rivers in Central Europe with a discharge average of about 559 m<sup>3</sup> s<sup>-1</sup> at the city of Magdeburg (Weigold and Baborowski 2009). Near Magdeburg, the Middle Elbe can be characterized as a free-flowing lowland river with comparably large floodplains, only regulated by groyne fields. Such groyne fields are the dominant shore type along the  
95 German part of the river (Bussmann et al. 2022).

Hence, seasonal water level fluctuations are shaping the different habitats alongside the river, ranging from alluvial forests and pastures to sandy beaches (Scholten et al. 2005). The study site is located near the farm “Apfelwerder” at river km 314 in between two groins and is characterized by a slight slope from the river to the adjoining pasture (52.038398 N, 11.715495 E). Groynes extended about 50 m into the  
100 river and distance between groynes was 130±37 m. A sandy beach of about 2 to 5 m with sparse vegetation (*Persicaria lapathifolia*, *Rorippa amphibia*, *Polygonum aviculare*) could be found directly at the river, while the vegetation became denser with distance to the river (~~figure~~ Figure S1).

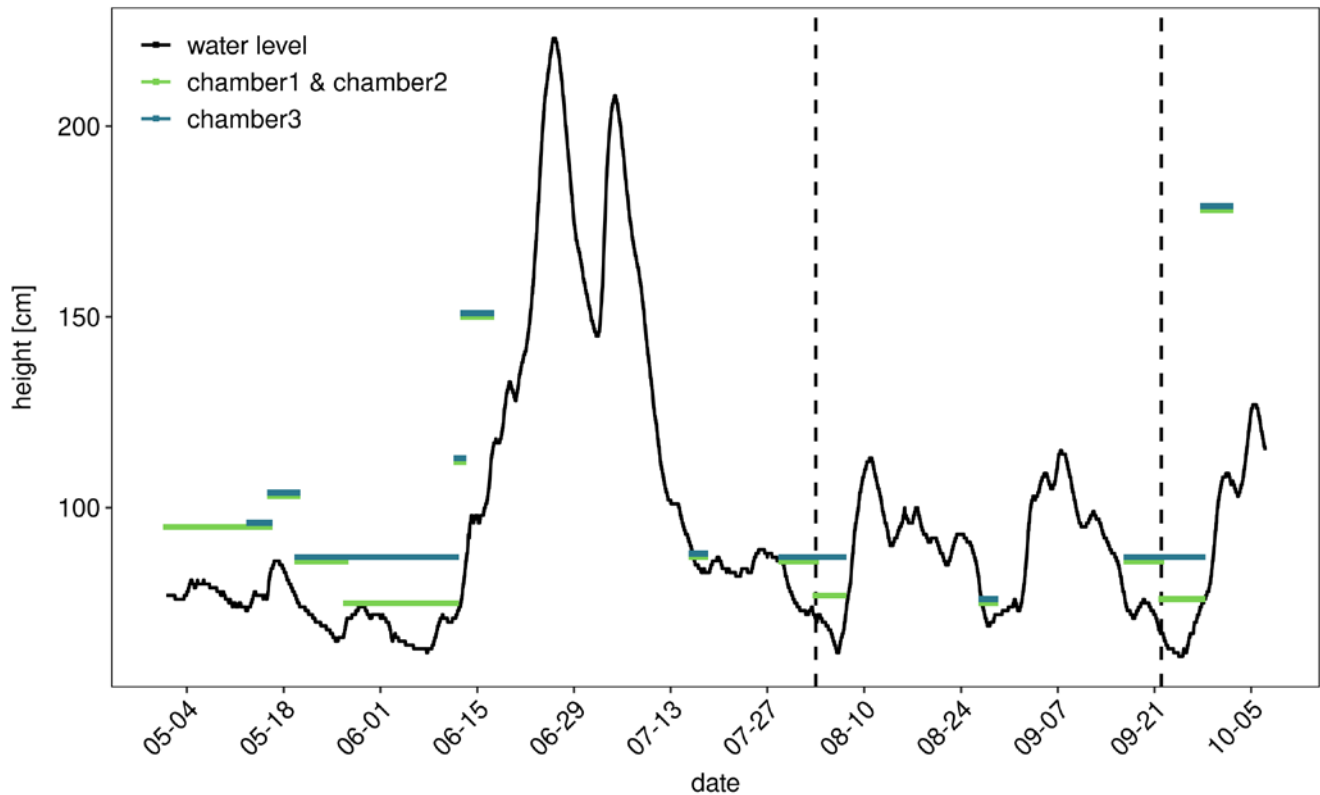
## 2.2 High frequency measurements

### 2.2.1 Automatic flux chambers, water table levels, and environmental data

105 To cover the temporal dynamics of CO<sub>2</sub> fluxes three opaque automatic chambers (CFLUX-1 Automated  
Soil CO<sub>2</sub> Flux System, PP Systems, Amesbury, Massachusetts, USA), were installed (Figure S1a). The  
chambers measured CO<sub>2</sub> fluxes once every hour. Each flux measurement lasted 5 minutes and between  
flux measurements the chambers were open for 55 min~~The chambers measured hourly CO<sub>2</sub> fluxes.  
Between flux measurements the chambers were open.~~ CO<sub>2</sub> fluxes were calculated from the linear increase  
110 of CO<sub>2</sub> during a closure time of 5 minutes. Each chamber was equipped with a soil moisture and  
temperature probe (Stevens HydraProbe, Stevens Water Monitoring Systems, Portland, Oregon, USA).  
Due to fluctuating water levels over the summer of 2020 (Figure 1), it was not possible to measure CO<sub>2</sub>  
fluxes from the sediment continuously over the whole measurement period. The chambers were set up in  
the periods from May 1st to June 10th, from August 3<sup>rd</sup> to August 6<sup>th</sup>, and from September 17th to  
115 September 26<sup>th</sup>; moreover, also during deployment they needed to be moved occasionally. Automatic flux  
chamber data were discarded when the collar was flooded or the sand was washed away by waves  
removed, which resulted in CO<sub>2</sub> concentrations fluctuating around ambient concentration~~Automatic flux  
chamber data were quality checked and data when the collar was flooded or the sand was washed away  
by waves removed.~~ The final dataset contained 3128 flux measurements.

120 To assure reproducibility and comparability of the automatically measured data, we installed the  
chambers at defined heights above the gauge “Magdeburg Strombrücke”. Therefore, the distance to the  
river and the height over water level were determined once, along a transect. Out of these parameters, a  
slope was calculated and afterward used to position the automatic chambers in the field. Positions, where  
the automatic chambers were placed were 75, 85 and 95 cm above zero point of the gauge “Magdeburg  
125 Strombrücke” (located 13 km downstream of the study side, zero point of gauge = 39.885 m above mean  
sea level (WSV 2020)). The thickness of the unsaturated sediment was calculated as the difference  
between the height above zero gauge for each chamber and the actual river level. Weather data from the  
German Weather Service were obtained for the monitoring station Magdeburg 15 km from Apfelwerder  
(DWD 2020).

12 km - ft. Land



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Figure 1: Water level of the Elbe River at gauge “Magdeburg Strombrücke” (12 km downstream) in summer 2020. Colored lines indicate positioning of automatic flux chambers. For example a horizontal line at 95 cm means that a particular chamber was located at the water line when the gauge recorded a water level of 95 cm. Vertical dotted lines indicate intensive sampling campaigns.

### 2.2.2 High resolution sediment respiration flux transects

135 To investigate spatial variability, On 11 occasions between May and September transects of sediment respiration were additionally measured with a portable soil respiration system (EGM-5 Portable CO<sub>2</sub> Gas Analyser + SRC, PP Systems, Amesbury, Massachusetts, USA) equipped with the same soil moisture and temperature probe as the automatic chambers. On each occasion 12 flux measurements along a 15m long transect from the water upslope were recorded. The opaque chamber was placed on vegetation free spots

140 to make sure that sediment respiration was measured. At each measuring spot we took note whether plants were growing nearby.

## 2.3 Detailed sampling campaigns

To closer investigate the mechanisms behind the CO<sub>2</sub> flux two intensive measurement campaigns were carried out on August 4<sup>th</sup>, 2020, and September 23<sup>rd</sup>, 2020.

### 145 2.3.1 Manual chamber measurements

To quantify ~~During these campaigns~~ CO<sub>2</sub> fluxes at different distance to the river and also check for CH<sub>4</sub> emissions, ~~capturing both biotic and abiotic flux components, from the dry river sediments were measured with a~~ manual chamber measurements were done in 1 m steps away from the flowing water, along a transect which was characterized by an uphill slope of ~ 11.5 %. Collars (39 cm diameter) were installed at 4 sites along the transect a day in advance to minimize disturbance during measurements (~~Fig.~~ Figure S1b). For flux measurements an opaque chamber ( $V = 0.0239 \text{ m}^3$ ,  $A = 0.1195 \text{ m}^2$ ) equipped with a pressure vent tube was placed on a collar. ~~The change of concentrations in the chamber was monitored~~ for every 30 s for ~ 5 minutes, with a multicomponent FTIR gas analyser (DX4000, Gaset Technologies GmbH, Helsinki, Finland). The FTIR gas analyser continuously measures CO<sub>2</sub>, CH<sub>4</sub>, and Nitrous oxide (N<sub>2</sub>O) with an accuracy of  $\pm 4 \text{ ppm CO}_2$  ~~<2% of the measuring range per zero-point calibration interval~~,  $\pm 0.1 \text{ ppm CH}_4$  and  $\text{N}_2\text{O}$  (Gaset Technologies GmbH 2018). Hence, the detection limit of the CO<sub>2</sub> flux was ~2 mmol m<sup>-2</sup> d<sup>-1</sup>, while the CH<sub>4</sub> flux was detectable if above 0.12 mmol m<sup>-2</sup> d<sup>-1</sup>, and N<sub>2</sub>O if above 0.2 mmol m<sup>-2</sup> d<sup>-1</sup>. Fluxes were calculated from the linear increase of the respective gas mixing ratio (Gomez-Gener et al. 2015) with time using the R package glimr (Keller 2020).

### 160 2.3.2 Rn sediment efflux measurements

To assess groundwater degassing <sup>222</sup>Rn measurements were performed. The geogenic gas <sup>222</sup>Rn is a commonly used natural tracer for groundwater influence in aquatic systems and is additionally known as a useful tool to trace the origins of CO<sub>2</sub> (Cook and Herczeg 2000). Therefore, <sup>222</sup>Rn concentrations and fluxes were measured with a portable radon detector (RAD7 Radon Detector, DURRIDGE, Billerica, Massachusetts, USA) to determine the groundwater influence on CO<sub>2</sub> fluxes from dry river sediments. The measurements of the RAD7 are based on electrostatic collection of alpha-emitters with spectral analysis. Measuring with the “Normal” mode counts decays of both Polonium decay products of <sup>222</sup>Rn

(<sup>218</sup>Po, <sup>214</sup>Po). The counts were measured over one hour and averaged, with a standard deviation of one sigma and expresses as decays per second [Bq]. The measurement range lies between 4 – 750000 Bq m<sup>-3</sup> with an accuracy of ± 5 %.

The <sup>222</sup>Rn concentration in 300 mL ~~water~~ samples from ground water (2.3.3) and the river was measured with the Wat250 mode. In addition, soil <sup>222</sup>Rn emissions were estimated with closed chamber measurements with the RAD7 over 3h (one Rn measurement per hour). Assuming that groundwater is the main source of CO<sub>2</sub> and that <sup>222</sup>Rn moves at the same mass flow as CO<sub>2</sub> (Megonigal et al. 2020), the same spatial dependence of CO<sub>2</sub> and <sup>222</sup>Rn fluxes would be expected in case of groundwater being the major source of CO<sub>2</sub>. For this reason, <sup>222</sup>Rn chamber measurements were performed simultaneously at two different positions, one with low and one with high CO<sub>2</sub> flux. We used two chambers of different size and corrected <sup>222</sup>Rn flux measurements [Bq m<sup>-3</sup> d<sup>-1</sup>] for different chamber geometry by multiplying with the volume [m<sup>3</sup>] and dividing by the area [m<sup>2</sup>] of the chamber to get the <sup>222</sup>Rn flux [Bq m<sup>-2</sup> d<sup>-1</sup>].

### 180 2.3.3 Water + sediment sampling

For groundwater sampling piezometers with a diameter of 2.7 cm and a length of 100 cm were installed next to each collar (Figure S1b) a day before the sampling campaign.

To determine the thickness of the unsaturated zone, the water level in the piezometers was measured with an electric contact gauge. *In situ* parameters pH, conductivity, temperature, and O<sub>2</sub> saturation were measured in the piezometers and the river with a multiparameter probe (WTW® MultiLine® Multi 3630 IDS, Xylem, Rye Brook, New York, USA). To analyze dissolved CO<sub>2</sub> and CH<sub>4</sub> concentrations, water samples were taken from the piezometers and the river using a syringe. Atmospheric air was added, with a headspace ratio of 1:1. After shaking for 2 minutes the headspace was transferred to 12 ml evacuated Exetainers (Labco Exetainers®, Labco Limited, Lampeter, UK) and stored till further analysis in the laboratory. Air samples were taken for headspace correction. Water samples for chemical analysis were collected in crimp vials without a headspace, stored at 4 °C and later analyzed in the laboratory.

Soil samples from the 0-5 cm layer were taken around each collar for incubation experiments. Samples were filled into plastic bags, stored at 4°C, and were analysed in the laboratory within a week.



### 2.3.4 Potential CO<sub>2</sub> production in laboratory incubations of sediment

195 Incubation experiments were set up to analyze the potential microbial respiration in dry river sediments under controlled conditions. For this purpose, fresh soil samples (25 g wet weight), taken along the transect were incubated in ~ 130 ml vials in replicates of four at 19.5 °C. To determine the temperature dependence of microbial respiration, 4 replicate samples of 25 g were incubated at 4, 12, 19.5, 28 and 35 °C. From each vial, 4 to 5 gas samples were taken over an incubation period of 2 to 3 days by a Pressure-Lok® syringe (Pressure-Lok® glass syringe, Valco Instruments, Waterbury, Houston, USA) and analyzed by gas chromatography for CO<sub>2</sub>. Respiration rates were calculated from the linear increase of the CO<sub>2</sub> content in the incubation vials divided by dry sediment weight.

To evaluate the temperature response of the microbial respiration in the sediment the Q10 temperature coefficient and the activation energy (E<sub>a</sub>) was calculated (Dell et al. 2011). The activation energy was calculated as the slope of Arrhenius plots as described in Gillooly et al. (2001).

To compare respiration data from lab incubations to CO<sub>2</sub> fluxes measured in the field rates rates of respiration per gram dry weight [μmol g-dw d<sup>-1</sup>] were converted to fluxes by multiplying with sediment bulk density [g-dw cm<sup>-3</sup>] and the thickness of the reactive sediment layer which we assume to be equal to the thickness of the unsaturated zone [cm].

## 210 2.4 Analytics

CO<sub>2</sub> and CH<sub>4</sub> concentrations in gas samples were measured with a gas chromatograph (GC) (SRI 8610C, SRI Instruments Europe, Bad Honnef, Germany). The GC was equipped with a flame ionization detector and a methanizer which allowed simultaneous measurement of CO<sub>2</sub> and CH<sub>4</sub> with an accuracy of < 5 %. Dissolved gas concentrations were calculated using temperature dependent Henry coefficients (UNESCO/IHA 2010). Because the carbonate system in the headspace vial may change during headspace equilibration ~~and~~ CO<sub>2</sub> concentrations were corrected for alkalinity as described in Koschorreck et al. (2021).

To analyze dissolved inorganic carbon (DIC) and dissolved organic carbon (DOC) water samples were filtered with a glass microfiber filter (Whatman GF/F). DIC and DOC concentrations were analyzed based on high-temperature oxidation and NDIR-Detection (DIMATOC® 2000, DIMATEC Analysentechnik,

Essen, Germany). The alkalinity of the water samples was determined by titration with HCl to pH of 4.3. To determine the concentration of the cations  $K^+$ ,  $Na^+$ ,  $Ca^{2+}$  and  $Mg^{2+}$  the water samples were filtered with a 0.45  $\mu m$  syringe filter, acidified with  $HNO_3$  and analyzed with an ICP OES (Optima 7300 DV, Perkin Elmer, USA). The Anion concentrations of  $SO_4^{2-}$  and  $Cl^-$  were measured with ion chromatography (Dionex-ICS 6000, Thermo Fisher Scientific, Waltham, Massachusetts, USA).

Soil samples were analyzed to determine soil moisture content, bulk density, and organic matter from weight loss after ~~loss after~~ drying for at least 2 days to constant weight at 105 °C and loss on ignition (LOI) at 550°C, respectively. Sediment texture was determined by the FAO method (FAO 2020).~~To analyze  $\delta^{13}C$  signatures of the organic matter, the sand and the organic matter in the sediment samples needed to be separated. This was done by suspending samples in ultrapure water followed by decanting and drying at 45 °C for 24 h. The dried and homogenized samples were analysed for carbon and nitrogen content and  $\delta^{13}C$  and  $\delta^{15}N$  signatures with an Element Analyzer connected to a mass spectrometer (EA IsoLink™ IRMS System, Thermo Fisher Scientific, Waltham, Massachusetts, USA).~~

## 2.5 Statistics

CO<sub>2</sub> flux data sets from manual and automatic measurements were visually checked for normality distribution with Q-Q-plots.~~The CO<sub>2</sub> fluxes from the manual measurements were summarized by date and tested with univariate ANOVA and with the post hoc Tukey's test paired t tests whether samples originate from the sample distribution and if groups differ significantly from each other. Additionally, the~~Data were summarized by distance to the river and tested with a one-sample t-test to determine if measured fluxes differed significantly from zero.

Spearman rank correlation was used to identify relationships between environmental variables and the observed CO<sub>2</sub> flux, and to identify the strength and direction of these relations (Leyer and Wesche 2007). Additionally, representative periods and single days were selected from automatic measurements to analyze patterns, hidden by the temporal variability of the data. The measured environmental variables of sediment temperature, sediment moisture, thickness of the unsaturated zone, organic matter content, and precipitation were used for correlation analysis. Water level and climate data were averaged over 1 hour. Linear mixed-effects models (lme) were applied to predict the influence of the environmental variables

on the CO<sub>2</sub> flux at the study site for variables for which a linear relationship with the CO<sub>2</sub> flux was presumed. Model selection was done by removing predictors and comparing conditional R<sup>2</sup> values of different models. To apply simple linear regression models and lme, assumptions of normality and homoscedasticity were visually checked with diagnostic plots, including residuals vs. fitted and Q-Q-plot. Flux data were log transformed for lme analysis. Because of occasional small negative fluxes we shifted all fluxes to positive values by adding 121 mmol m<sup>-2</sup> d<sup>-1</sup> prior to transformation (120 was is the value of the largest negative flux). Statistical analysis was performed using R (R-Core-Team 2016).

~~To evaluate the temperature response of the CO<sub>2</sub> flux as well as of microbial respiration in the sediment the Q10 temperature coefficient and the activation energy (Ea) was calculated. The activation energy was calculated as the slope of Arrhenius plots as described in Gillooly et al. (2001).~~

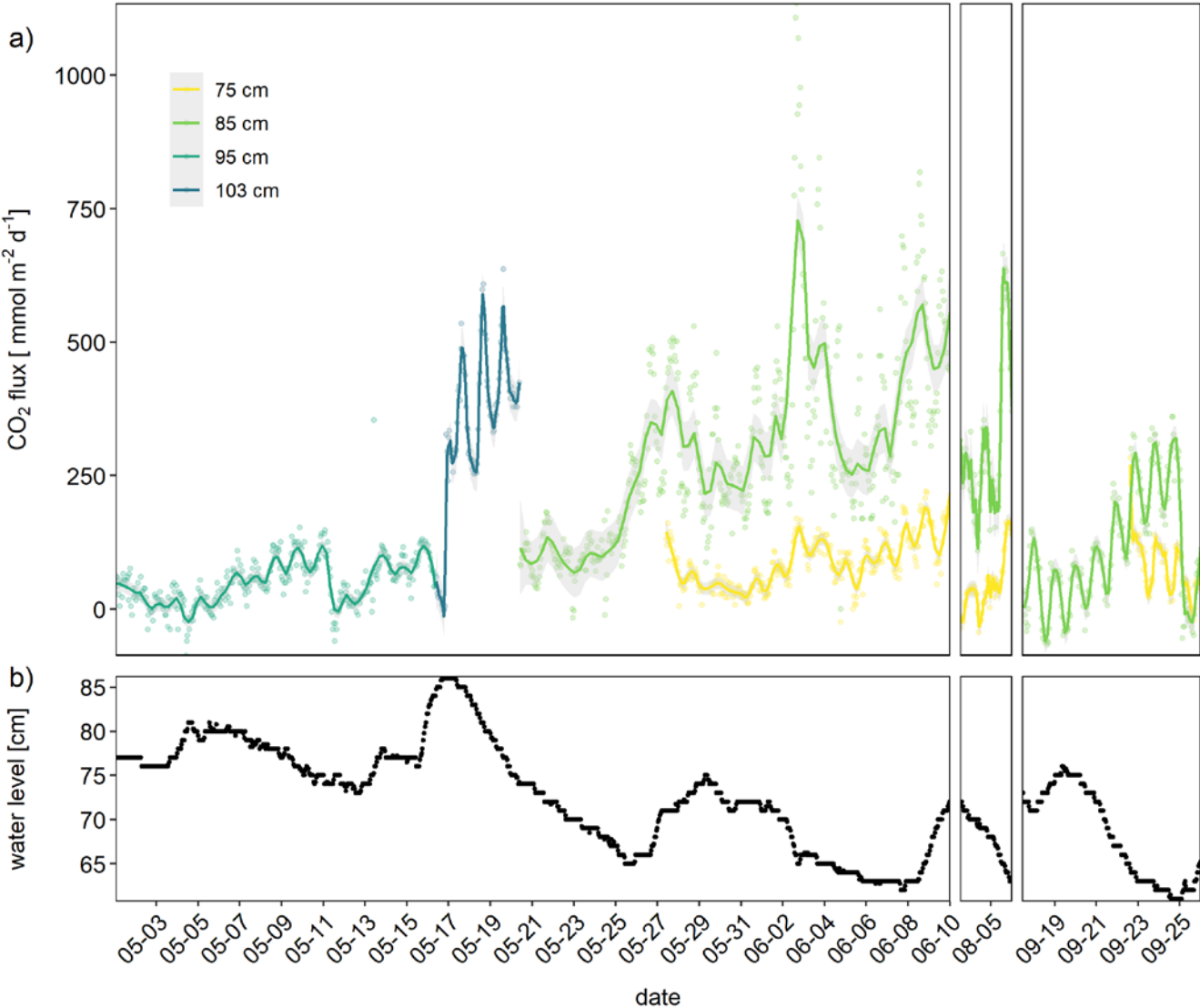
## 3 Results

### 3.1 Long term data

The river showed a typical summer discharge situation with a water level mostly below 1 m, interrupted by a high discharge event at the end of June (Figure 1). Considerable areas of dry sediments only emerged during 6 weeks in early summer, and short periods in the first week of August and in September. CO<sub>2</sub> fluxes measured during these periods showed high diurnal and seasonal fluctuations (Figure 2). Fluxes fluctuated over 3 orders of magnitude between -120 and 1135 mmol m<sup>-2</sup> d<sup>-1</sup> with a median of 98 and a mean $\pm$ SD of 149 $\pm$ 155 mmol m<sup>-2</sup> d<sup>-1</sup>. Fluxes fluctuated in a narrow range below 200 mmol m<sup>-2</sup> d<sup>-1</sup> during the first phase of the investigations in May<sup>i</sup>. Due to rising water level at May 17<sup>th</sup> we moved the chambers higher up where we measured both higher fluxes and larger diurnal amplitudes. When the water level decreased after May 20<sup>th</sup> we moved the chamber down to freshly emerged sediment. There, CO<sub>2</sub> fluxes were similar to the fluxes measured 10 cm higher during the first half of May and tended to increase with increasing time since drying. Negative fluxes were observed in 193 out of 3128 flux measurements (=6% of all fluxes). Negative fluxes were observed especially during the beginning of the measurement period and at sites near to the water. Interestingly, negative fluxes nearly exclusively occurred during the day

between 10:00 and 18:00, peaking in the afternoon (Figure S2). Chambers installed closer to the water measured lower and less variable fluxes than chambers installed higher upslope.

275 Fluxes showed considerable short-term variability. Variability was not constant during the investigated period but especially high after June. Clear diurnal patterns were observed during the entire study but most pronounced in September.



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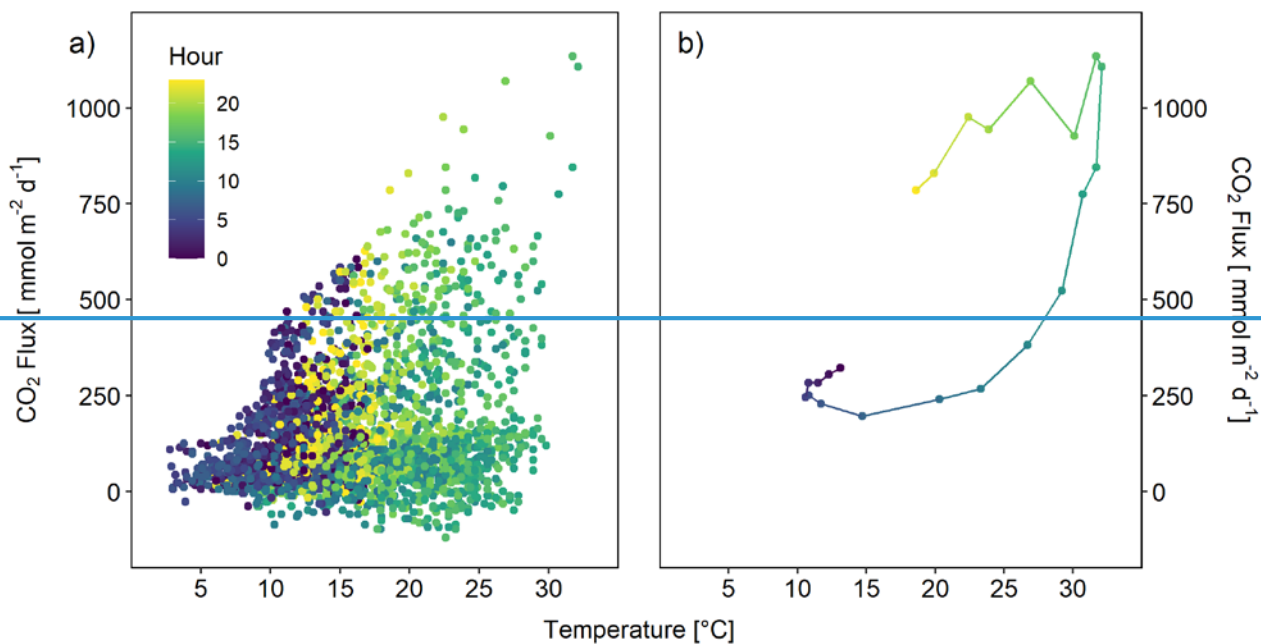
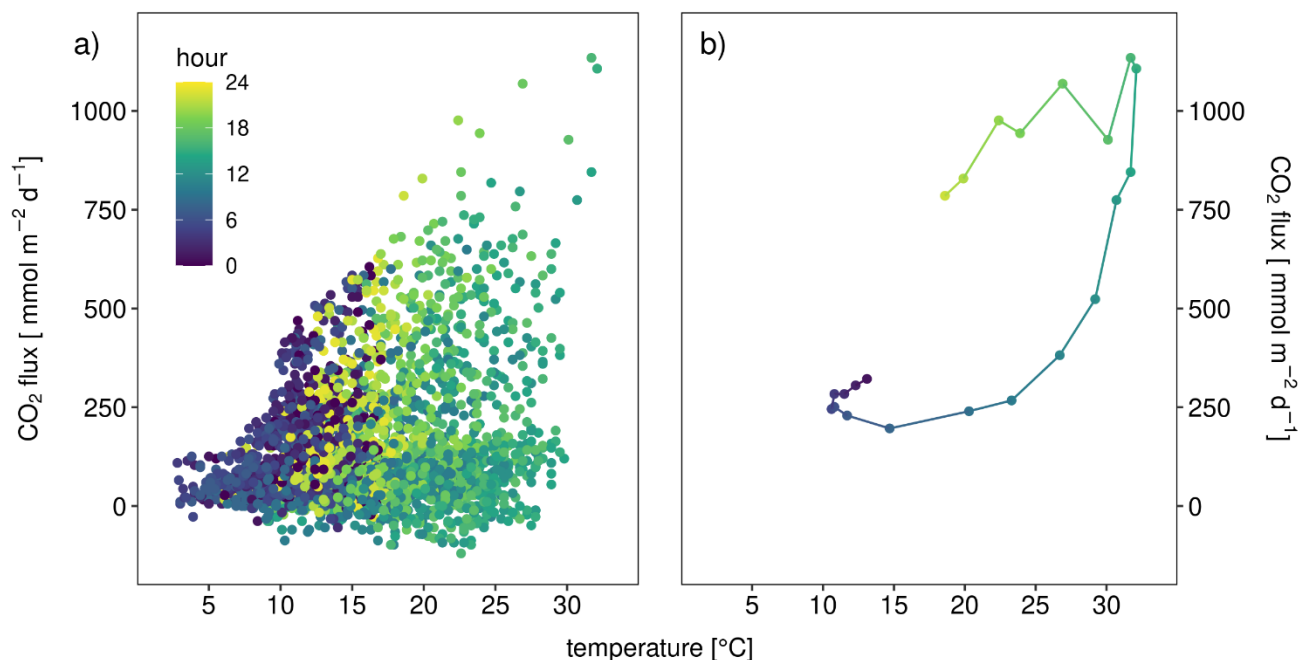
**Figure 2:** CO<sub>2</sub> fluxes in mmol m<sup>-2</sup> d<sup>-1</sup> measured with automatic chambers (a) and corresponding water level of the river measured 13 km downstream at the gauge “Magdeburg Strombrücke” (b). Colors indicate the position of the chamber relative to the gauge

(=elevation of the site), either 75, 85, 95, or 103 cm above current level. Lines indicate smoothed data  $\pm$  SD using LOESS smoother with span 0.1~~the geom\_smooth function in R (method="loess", span=0.1)~~. The grey areas indicate confidence intervals.

### 285 3.1.1 Regulatory factors ~~HF~~: sediment moisture, temperature, water level, climate

The observed diurnal pattern with higher CO<sub>2</sub> fluxes during the day suggested a temperature regulation of the flux. The CO<sub>2</sub> flux was indeed weakly (spearman  $p < 0.05$ ) correlated with the thickness of the unsaturated zone ( $R^2 = 0.31$ ), sediment temperature ( $R^2 = 0.19$ , Figure 3a) and moisture ( $R^2 = -0.19$ ), as well as precipitation ( $R^2 = -0.12$ ). A mixed effect linear model with site were the chamber was placed as random  
290 factor and temperature and thickness of the unsaturated zone as fixed factors explained 0.61 % of the variability. Adding moisture did not further improve the lme ([Table S1](#)).

The temperature response of the CO<sub>2</sub> flux was not very clear, however, if all data were plotted together (Figure 3a) but if data from single days were plotted a clear pattern emerged (Figure 3b and Figure S3). The temperature response of the flux was affected by the time of day resulting in typical hysteresis curves.  
295 Warming during the day resulted in exponentially increasing fluxes. However, fluxes stayed high despite cooling started in the afternoon – the temperature response of the flux was clearly delayed. From the CO<sub>2</sub> flux-temperature relation (Figure 3a) an activation energy of 0.56 eV (37 kJ mol<sup>-1</sup>) could be calculated which corresponds to a Q<sub>10</sub> of 1.7 between 10 and 20°C.



**Figure 3:** CO<sub>2</sub> flux in mmol m<sup>-2</sup> d<sup>-1</sup> (obtained from 3 automatic chambers) depending on sediment temperature (a) all data (b) only data from June 2<sup>nd</sup> as an example for hysteretic response to temperature. Color indicate hour of measurement.

A closer look at data from one week in September revealed how temperature, thickness of the unsaturated zone, and precipitation interacted in regulating the flux (Figure 4). Temperature drove the very clear

305 diurnal amplitude but the absolute level of the flux was higher with increasing thickness of the unsaturated  
 zone (which was accompanied by sediment drying). A single precipitation event at September 25<sup>th</sup>  
 resulted in a sudden increase in sediment moisture which was accompanied by a clear drop of the CO<sub>2</sub>  
 flux. If only data for the period shown in Figure 4 were considered a linear model containing sediment  
 temperature and moisture and the interaction between temperature and moisture explained 46 % of the  
 310 variance.

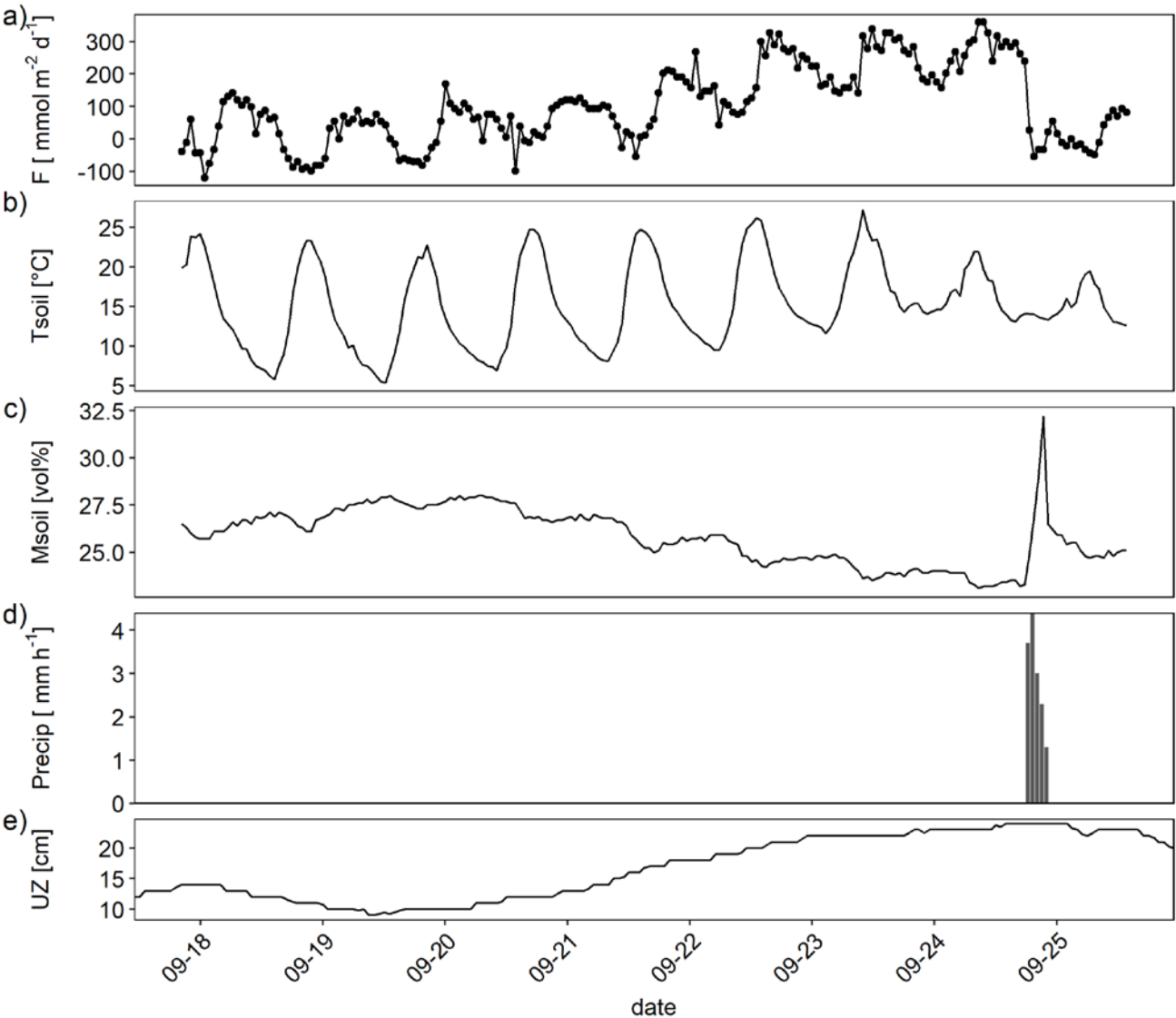
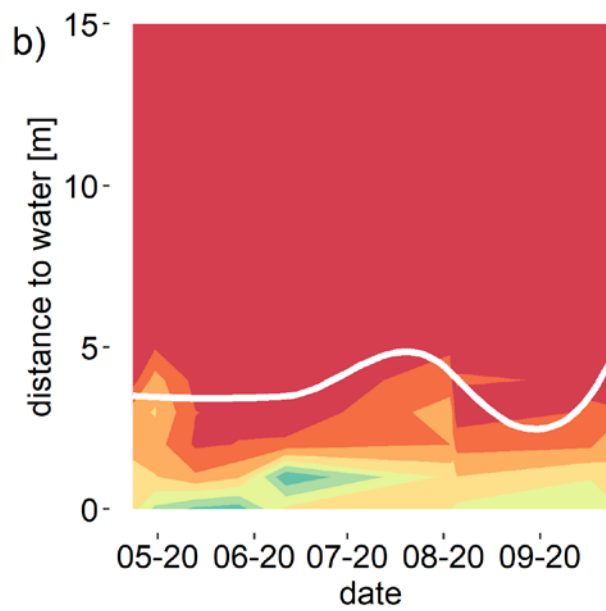
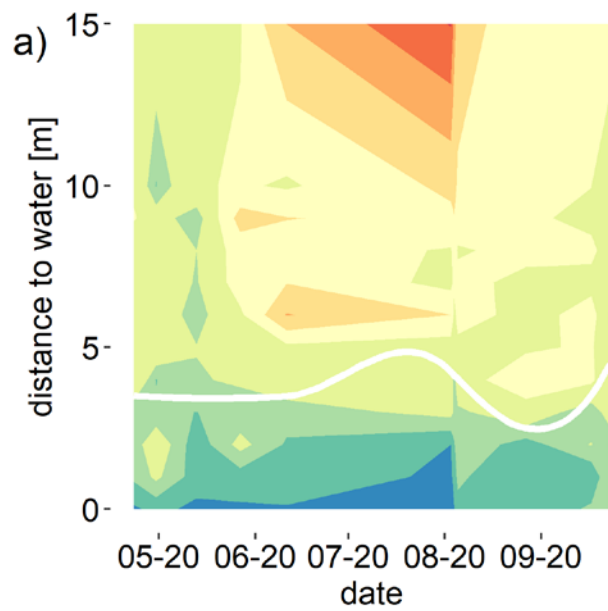


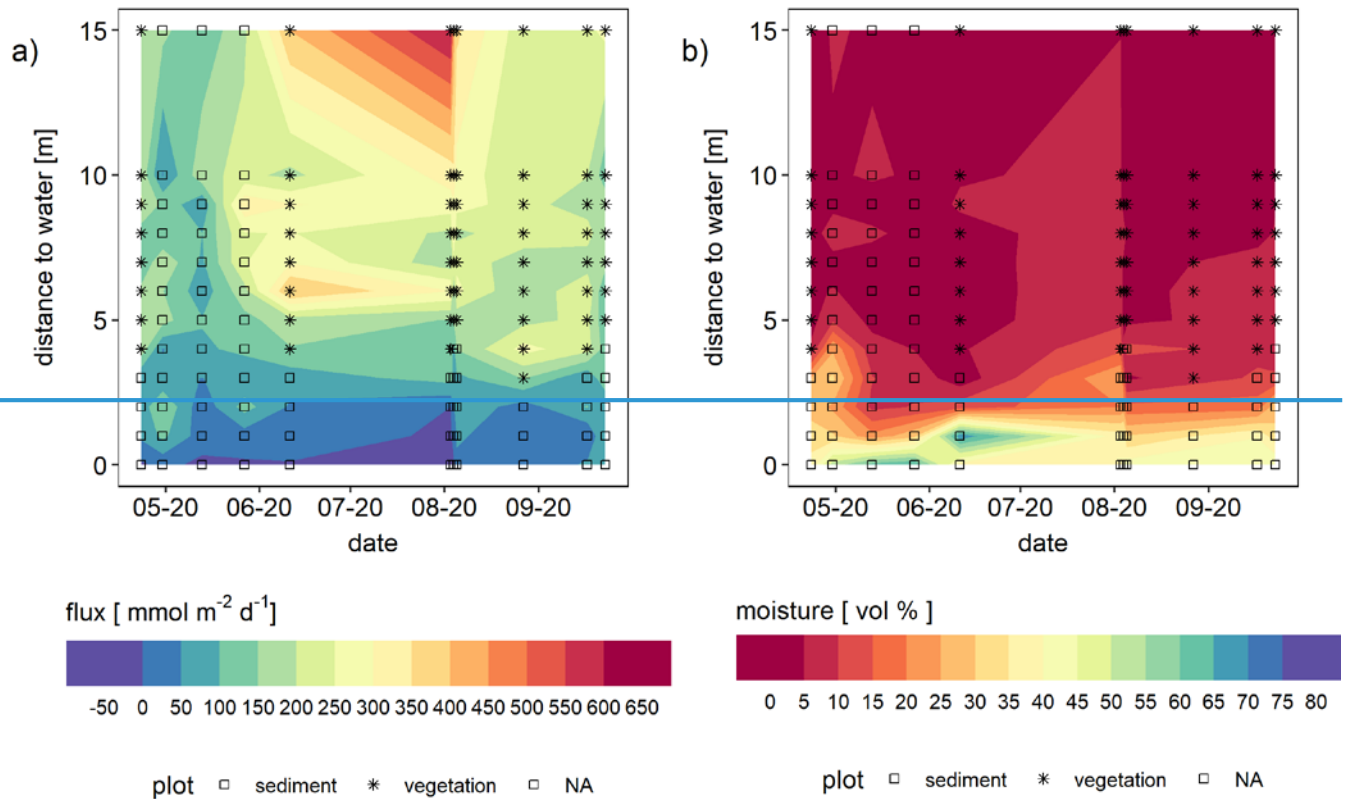
Figure 4: Example high frequency dataset showing (a) CO<sub>2</sub> flux (F) measured by an automatic chambers (b) sediment temperature (T<sub>soil</sub>; 0-5 cm depth) (c) sediment moisture (M<sub>soil</sub>; 0-5 cm depth) (d) precipitation (Precip) recorded in hourly resolution and (e) thickness of the unsaturated zone (UZ; distance between water table level and ground surface).

### 3.1.2 Spatial gradient of CO<sub>2</sub> flux

Manual ~~As seen from the automatic~~ chamber measurements at different distance to the water revealed, ~~we observed~~ a spatial gradient of the CO<sub>2</sub> flux. CO<sub>2</sub> fluxes were lowest near to the water line where sediment moisture was highest (Figure 5) and fluxes increased with distance to the water. This was also visible in the automatic chamber data when chambers were placed at different distance to the water (compare Figure 2). The chamber which was placed nearer to the water recorded consistently lower fluxes. This is also consistent with the observed positive correlation between CO<sub>2</sub> flux and the thickness of the unsaturated zone.







325 **Figure 5: CO<sub>2</sub> flux (mmol m<sup>-2</sup> d<sup>-1</sup>) as measured with a manual chamber (a) and sediment moisture (vol-%) measured with a probe in 0-5 cm depth (b) depending on the distance to the water. The white lines indicates the “plant line”. The areas below this line was free of vegetation Dots represent points of measurement and symbols indicate presence or absence of plants near to the chamber (never plants inside chambers).**

330 We also observed higher CO<sub>2</sub> fluxes in the vicinity of plants. Plants were consistently found from about 3 m from the water uphill. Fluxes above this “plant line” (indicated by the white linesymbols in Figure 5) tended to be higher than fluxes from the vegetation free area nearer to the water.

In sum, our field based measurements provide strong evidence that respiration in the sediment was the major driver of the observed CO<sub>2</sub> flux~~Taken this together there was strong evidence from field based flux measurements that respiration in the sediment was the major driver of the CO<sub>2</sub> flux.~~ To further support  
335 this conclusion detailed investigations were carried out.

### 3.2 Detailed investigations

The sediment pore water was quite similar to river water with respect to electric conductivity and dissolved solutes including DIC (Table 1). The water level difference between the wells and the river was below the detection limit – the hydraulic gradient was virtually zero during our sampling campaigns. The shallow hydraulic gradient and the similar chemistry suggest a large influence of river water on the sediment pore water. In contrast, concentrations of dissolved gases were quite different with high concentrations of CO<sub>2</sub> and CH<sub>4</sub>, and low concentrations of O<sub>2</sub> in the pore water. Pore water concentrations of CO<sub>2</sub> increased with distance to the river while CH<sub>4</sub> concentrations tended to be highest near to the river. In August the river water was slightly under saturated with respect to CO<sub>2</sub>. The sediment was poor in organic matter (LOI < 1%) ~~and~~ ~~Sediment texture as determined by the FAO method (FAO 2020)~~ was loamy sand. GHG emissions were dominated by CO<sub>2</sub> while CH<sub>4</sub> fluxes were low and N<sub>2</sub>O fluxes always below the detection limit (Table 1).

Table 1: Sediment, groundwater, and river water properties at the 2 sampling campaigns

parameter		unit	August 4 <sup>th</sup> 20204.8.2020					September 23 <sup>th</sup> 23.9.2020				
distance	to	m	river	1	3	5	6	river	1	2	3	4
river												
CO <sub>2</sub> flux		mmol m <sup>-2</sup> d <sup>-1</sup>	-3	33	87	153	153	36	103	49	142	126
CH <sub>4</sub> flux		mmol m <sup>-2</sup> d <sup>-1</sup>	0.7	3.4	0	0	0.6	6	0.5	0	0	-0.6
unsaturated		cm		10	31	62	78		9	19	32	36.5
zone												
moisture		[vol %]		30	13	25	12		30	25	-	9
organic		[% LOI]		0.78	0.39	1.11	0.94		0.85	0.97	-	0.52
matter in sediment												
CH <sub>4</sub>		μmol L <sup>-1</sup>	0.3	18	11	11	6	2.5	189	186	212	70
CO <sub>2</sub>		μmol L <sup>-1</sup>	13.3	610	883	1960	3681	32	1193	899	1118	1024
DIC		mg L <sup>-1</sup>	42	23	48	49	50	24	70	64	64	55

alkalinity	mg L <sup>-1</sup>	1.9	3.5	3.5	3.6	3.1	1.9	4.5	4.8	5.3	4.7
DOC	mg L <sup>-1</sup>	13.1	6.9	9.3	12	13.5	6.31	9.4	9.9	11.5	11
SO <sub>4</sub> <sup>2-</sup>	mg L <sup>-1</sup>	79	44	71	67	74	79	7.3	20	31	92
pH		8.3	7.2	6.8	6.6	6.6	8	7.2	7.3	7.2	7
conductivity	μS cm <sup>-1</sup>	640	610	658	640	1563	601	696	655	647	640
O <sub>2</sub>	mg L <sup>-1</sup>	9.1	0.8	1.1	1.9	2	9.3	3.4	2.5	4	4

Diffusive fluxes from the river were calculated from concentrations using the gas transfer coefficient from  
350 Matoušů et al. (2019).

### 3.2.1 CO<sub>2</sub> fluxes versus Rn fluxes

Groundwater contained more than one order of magnitude higher Rn concentrations than the river water  
(Table 2). As an indicator of groundwater evaporation and possible evasion of CO<sub>2</sub>, we measured the flux  
of radon out of the sediment, assuming groundwater as a major source. Rn fluxes were higher in  
355 September than in August although the Rn concentration in the groundwater was similar in both months  
(Table 2). The flux of radon out of the sediment was, however, not much different at two different  
distances to the river while the CO<sub>2</sub> flux differed by about one order of magnitude between the same sites.  
If groundwater was the source of CO<sub>2</sub>, we would expect Rn fluxes to be related to CO<sub>2</sub> evasion from  
groundwater; thus our data indicates that higher CO<sub>2</sub> fluxes were not ~~driven by higher~~  
360 ~~evaporation~~originating from groundwater.

**Table 2: Flux of radon measured as <sup>222</sup>Rn increase in static chambers compared to CO<sub>2</sub> flux measured in the same chambers, and radon concentration determined as detected activity (Bq m<sup>-3</sup>) in the groundwater sampled in wells directly beside the chambers as well as in the river water (0 m distance).**

date	Distance to river [m]	<sup>222</sup> Rn flux [Bq m <sup>-2</sup> d <sup>-1</sup> ]	CO <sub>2</sub> flux [mmol m <sup>-2</sup> d <sup>-1</sup> ]	<sup>222</sup> Rn in <del>GW</del> <u>water</u> [Bq m <sup>-3</sup> ]
<del>5.8.2020</del> <u>08-</u>	<u>0</u>			<u>327±109</u>
<u>05</u>	1	65	18±20	6090±418
	3	63	110±31	
	<u>0</u>			<u>532±135</u>

09-	1	174	7±41	6650±436
23.9.2020	4	205	169±36	

### 3.2.2 Sediment respiration rates

To check whether the observed CO<sub>2</sub> fluxes could be explained by microbial respiration in the sediment, laboratory incubations were carried out. Sediment respiration rates as measured in laboratory incubations were 0.9±0.45 μmol g<sup>-1</sup> d<sup>-1</sup> in August and 0.64 μmol g<sup>-1</sup> d<sup>-1</sup> in September with rates increasing with distance to the river. Potential CO<sub>2</sub> fluxes calculated from these rates were similar or higher than CO<sub>2</sub> fluxes measured *in situ* (Figure 6). Thus, sediment respiration was high enough to explain the observed CO<sub>2</sub> emissions.

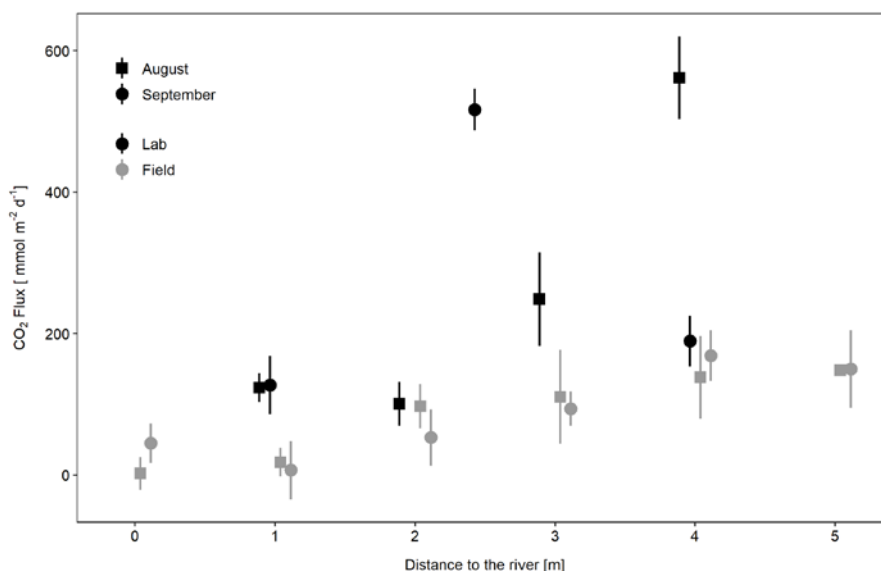


Figure 6: Potential CO<sub>2</sub> flux determined from laboratory incubations of sediment compared to *in situ* CO<sub>2</sub> fluxes depending on distance to the river. Potential fluxes per unit area were calculated from sediment respiration rates (mmol g<sup>-1</sup> dw d<sup>-1</sup>), the thickness of the unsaturated zone (cm), and the bulk density of the sediment (g dw cm<sup>-3</sup>).

### 3.2.3 Temperature dependence of sediment respiration

Sediment respiration increased exponentially with temperature (Figure 7) resulting in a Q<sub>10</sub> of 2.5. The calculated activation energy of 0.7 eV was similar to the activation energy calculated from the automatic chamber data. The comparison with the temperature response of the CO<sub>2</sub> flux measured by the automatic

chambers (line in Figure 7) visualizes the similar temperature response of sediment respiration and *in situ* fluxes.

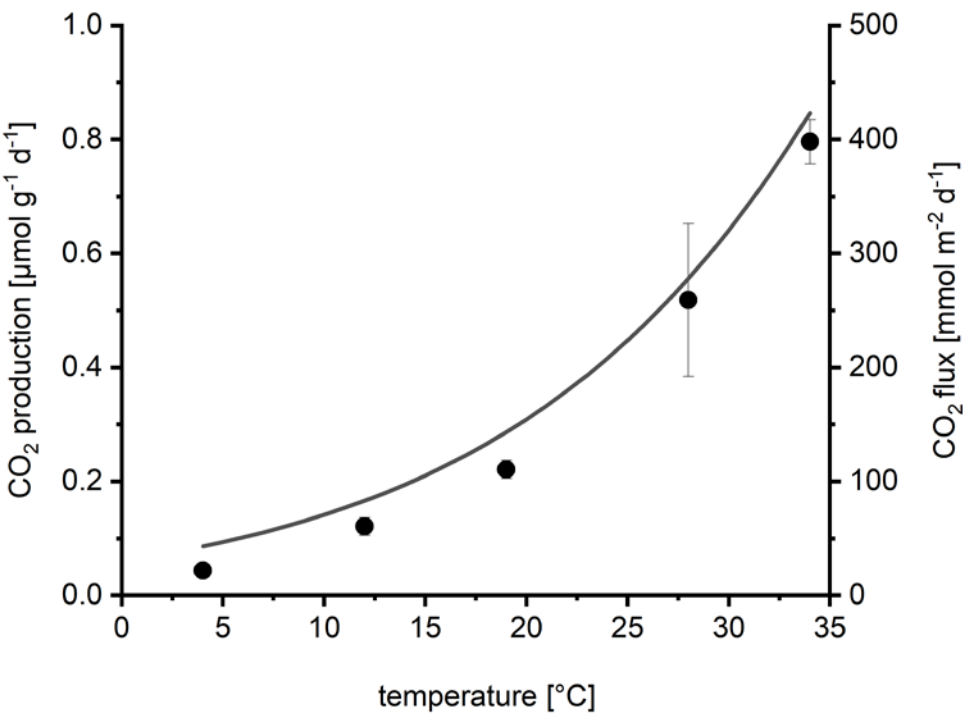


Figure 7: Temperature dependence of Sediment CO<sub>2</sub> production (=sediment respiration) in laboratory incubations depending on temperature (dots show mean±SD of 4 replicates). For comparison, the line shows the average temperature response of the CO<sub>2</sub> flux measured by automatic chambers, calculated by fitting the data from Figure 3a to the Arrhenius equation.

## 4 Discussion

### 4.1 Source of the CO<sub>2</sub>

Both our continuous data and detailed measurements show that the CO<sub>2</sub> emitted from dry Elbe sediments originated from respiration in the sediment rather than from groundwater ~~evasion~~. This conclusion is consistently supported by numerous evidence:

- The observed CO<sub>2</sub> fluxes could be fully explained by sediment respiration measured in laboratory incubations. From soil respiration measurements it is well known that basal respiration as measured in laboratory incubations cannot be equaled with soil CO<sub>2</sub> emissions (Reichstein et al.

2000). A major difference between both methods is the exclusion of root respiration in bottle incubations which would lead to an under estimation of total soil respiration in root free assays such as bottle incubations (Hanson et al. 2000). Thus, our sediment respiration rates measured in the laboratory are probably conservative estimates which even strengthens our argumentation.

- The temperature response of the CO<sub>2</sub> flux was very similar to the measured temperature response of sediment respiration and showed Q<sub>10</sub> values typical for biological processes (Yvon-Durocher et al. 2012) and soil respiration (Hamdi et al. 2013). Potential evaporation on the other hand depends on radiation, vapor pressure, and wind speed (Penman 1948) and only indirectly on surface temperature (Kidron and Kronenfeld 2016). The temperature dependence of evaporation of soils depends on a complex interaction of texture and soil moisture, and is not easy to predict (e.g. (Federer 2002)). The observed temperature dependence provides strong evidence for respiration being the primary driver of the CO<sub>2</sub> flux.
- CO<sub>2</sub> emissions increased with distance to the river. If groundwater evasion was a major source of CO<sub>2</sub> emissions we would expect higher emissions at lower sediment elevation where groundwater potentially exfiltrates into the sediment. If there was a hydraulic groundwater gradient towards the river this gradient should be steepest near to the river resulting in highest groundwater flux and potential outgassing near the river.
- The CO<sub>2</sub> flux was proportional to the volume of the unsaturated sediment. If CO<sub>2</sub> originated from groundwater emissions we would expect even a negative correlation because the transport of CO<sub>2</sub> from the groundwater surface to the sediment surface should be inhibited by a larger unsaturated zone.
- Higher CO<sub>2</sub> emissions were not accompanied by higher Rn emissions. Groundwater typically contains high Rn concentrations and Rn is a proven tracer to investigate groundwater input into surface waters (Cook and Herczeg 2000; Perkins et al. 2015). We observed emission of Rn from the sediments indicating some influence of groundwater on the sediments. Rn emission at different distance from the river were identical. Thus, the thickness of the unsaturated sediment did not affect Rn emissions, showing that the anoxic zone itself was probably not a source of Rn. Soil Rn concentrations are known to be affected by meteorological and soil physical conditions (Asher-

Bolinder et al. 1971). Similar Rn emissions, as observed in our study, are therefore an indication for similar sediment physical conditions. However, the magnitude of Rn emissions did not correspond to the magnitude of the CO<sub>2</sub> emissions, indicating that the CO<sub>2</sub> flux was independent from groundwater outgassing.

- 425 • As we did not see hydraulic gradients indicative of larger groundwater inflow at our location of study, unrealistic high evaporation rates would be necessary to explain the observed CO<sub>2</sub> flux with groundwater evasion. Groundwater degassing is relevant in situation when groundwater is pumped to the surface (Wood and Hyndman 2017) or seeps into surface waters (Duvert et al. 2018). In rivers it might be relevant at seep sites which probably especially occur after fast water  
430 level drops and at extremely low water level.

Taken together our data consistently show that the observed CO<sub>2</sub> emissions originated from respiratory CO<sub>2</sub> production in the sediment. After having identified the primary source of CO<sub>2</sub> we now look on the regulators of the magnitude of the CO<sub>2</sub> emissions.

## 4.2 Regulation of CO<sub>2</sub> emissions

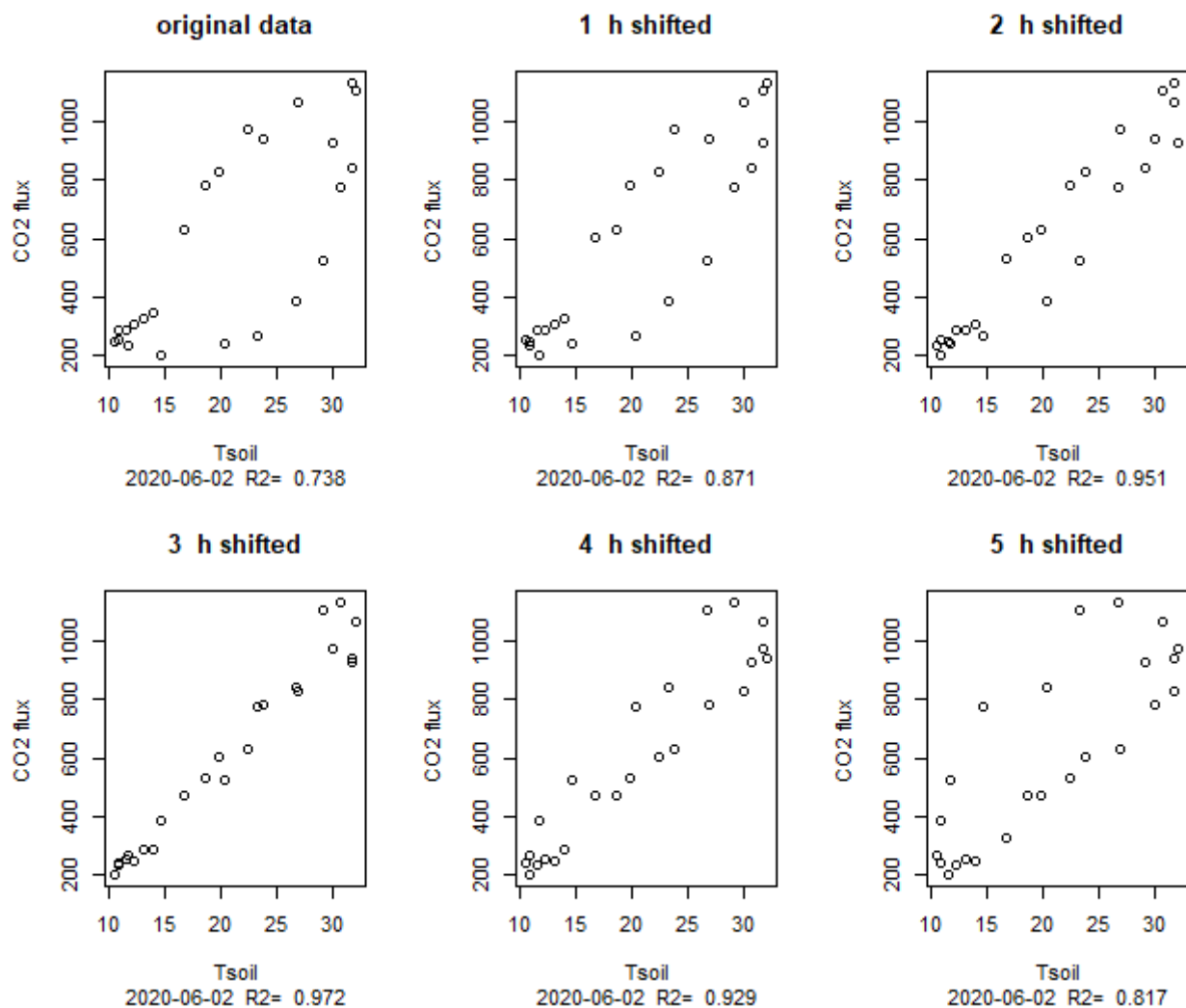
435 Temperature is a master variable regulating several biogeochemical processes. Our temperature dependence ( $Q_{10} = 2.5$ ,  $E_a$  0.7 eV) is in line with the temperature response of numerous ecological processes. A meta analysis of 63 studies of temperature dependence of soil respiration revealed a mean  $Q_{10}$  of 2.6 (Hamdi et al. 2013). Diverse types of ecosystems have an activation energy of respiration of 0.65 eV (Yvon-Durocher et al. 2012) which is very similar to our study.

440 Temperature was an important regulator not only because of the temperature dependence of sediment respiration but also because the diurnal temperature amplitude was quite large. Sediment temperature not only ranged between 2.8 and 32°C during the study period but the complete temperature amplitude of about 20°C could be observed during single days (Figure 4). The large diurnal amplitude at these sites is favored by a lack of shadow and the fast heating of the sand which can lead to temperatures easily  
445 exceeding 40°C (Mallast et al. 2020).

Although the temperature dependence of the CO<sub>2</sub> flux is evident, it was not easily visible in flux versus temperature plots which show a large scatter (Figure 3a). Only when looking at single days a typical



hysteresis pattern (Figure 3b) became apparent. Such hysteresis curves have frequently been observed in high frequency datasets of soil respiration (e.g. (Riveros-Iregui et al. 2007)). They originate from a phase lag between temperature and CO<sub>2</sub> flux and can be explained by different transport of heat and CO<sub>2</sub> in soils (Phillips et al. 2011) or by variable C supply from plants (Oikawa et al. 2014). The rotation direction as well as the shape of the ellipsoid depends on the vertical profile of temperature and activity in the soils as well as on the depth where soil temperature was measured. We measured temperature in 5 cm depth and obtained anticlockwise hysteresis which means that CO<sub>2</sub> emissions were delayed relative to temperature measurements. A plausible explanation is that a large part of the CO<sub>2</sub> was produced in deeper sediment layers where the daily temperature maximum was reached later. This is consistent with the observed positive correlation between CO<sub>2</sub> flux and the thickness of the unsaturated zone. Theoretically the effect could also be caused by delayed outgassing of CO<sub>2</sub> from deeper sediment layers due to CO<sub>2</sub> transport limitation. However model calculations had shown that this mechanism was less relevant for shaping diurnal hysteresis in soils (Phillips et al. 2011). We quantified the delay by shifting flux and sediment-temperature data against each other (Figure 8Figure-S3). By correlating the flux with the temperature 3 hours before we obtained the best linear correlation ( $R^2=0.97$ ) for the data in Figure 3b). However, the time shift which produced the best linear fit differed between days (min=0, max=10, mean±SD = 4.8±3.7 h) with a median of 4 hours and no apparent differences between sites. Also the  $R^2$  of the best fit differed between 0.2 and 0.97. Thus, the hysteresis pattern obviously depended on the day of measurement and it is not possible to derive a general relation which then could be used to analyze temperature-flux relations of time-shift corrected data.



**Figure 8: Hysteresis loop for June 2<sup>nd</sup> (same data as in Figure 3b) with flux data shifted for various hours. 3 hours shifted means that the flux at 10:00 was correlated with the temperature at 7:00.**

Wetting of dry soils typically triggers a pulse of CO<sub>2</sub> production (Birch 1958). However, in our case wetting events caused by rainfall reduced the CO<sub>2</sub> flux as exemplified in Figure 4. This shows that CO<sub>2</sub> production in the sediment was not water limited and/or that the CO<sub>2</sub> flux was rather transport limited when rain water blocked gas filled pores (Asher-Bolinder et al. 1971). At sediment moisture around 30% in sandy sediments as measured in our study microbial activity in the sediment is probably not water

stressed and consequently not stimulated by wetting. Thus, it is probable that the reduced CO<sub>2</sub> flux after rain events was caused by physical blocking of soil pores. This is consistent with the observed long term increase of the CO<sub>2</sub> flux with decreasing moisture. Direct mechanistic dependence, however, is difficult to show because moisture also correlates with the thickness of the unsaturated zone (=water level of the river relative to the sediment surface). This is why adding moisture to our mixed model only marginally increased the predictive power of the statistical model.

The thickness of the unsaturated zone was a strong predictor of the CO<sub>2</sub> flux. The entire unsaturated zone obviously contributed to the CO<sub>2</sub> flux. This is plausible because the intermediate sediment moisture both favored microbial processes and enabled gas exchange through gas filled pores. This may also explain high CO<sub>2</sub> fluxes in situations with extremely high sediment surface temperature (Mallast et al. 2020). Even if under such conditions CO<sub>2</sub> production is inhibited at the surface respiration in deeper layers may maintain high CO<sub>2</sub> emissions.

The occurrence of vegetation, although excluded from our chamber measurements and restricted to the vicinity of the chambers, obviously is a game-changer, largely stimulating sediment CO<sub>2</sub> emissions. From our data we cannot fully distinguish whether higher fluxes near plants were caused by the plants or only by distance to the water (which is equivalent to the thickness of the unsaturated zone). However, the thickness of the unsaturated zone increased continuously while the “plant line” represents a sudden change of conditions. Our data show a consistent high CO<sub>2</sub> flux above the “plant line”. It is well known that root respiration may contribute about 50% to soil respiration (Hanson et al. 2000) and soil respiration is typically correlated with root biomass (Tufekcioglu et al. 2001). Thus, as we did not use trenched collars to exclude roots from chamber fluxes, it is highly probable that plants contributed to the elevated CO<sub>2</sub> emissions through root respiration or provision of root exudates above the “plant line”. Higher sediment CO<sub>2</sub> emissions, however, do not mean net CO<sub>2</sub> emissions from the ecosystem since the vegetation growing on the dry sediments also fixes carbon and can even turn exposed sediments into a carbon sink (Bolpagni et al. 2017). To assess the effect of emerging vegetation on the overall carbon cycle of dry sediments other methods like plant biomass determination or flux measurements including photosynthesis in transparent chambers are necessary.

### 505 4.3 CO<sub>2</sub> uptake by the sediment

We frequently observed CO<sub>2</sub> uptake by the sediment, although there were no plants and no light in our chamber. This is known from other studies and has been attributed to inorganic processes (Ma et al. 2013; Marcé et al. 2019). In our case the observed CO<sub>2</sub> uptake could also be explained by the interaction of the sediment with river water. During May and June the river was under saturated with CO<sub>2</sub> (Fig. Figure S4).  
510 The groundwater chemistry data show a gradient of concentrations increasing with distance to the river. This shows that the sediment pore water near to the river was affected by river water. Interestingly, negative fluxes were nearly exclusively observed during the daylight hours. A plausible explanation would be that ship induced wave action might have triggered occasional river water intrusion and CO<sub>2</sub> uptake by the sediment (Hofmann et al. 2010). This mechanism, however, cannot explain negative fluxes  
515 in September when the river was over saturated with CO<sub>2</sub> (Fig. Figure S4).  
Dark CO<sub>2</sub> uptake could theoretically be caused by chemoautotrophic micro-organisms like nitrifiers. However, chemoautotrophic CO<sub>2</sub> uptake should not be stimulated by light and is thus not consistent with our observation of nearly exclusive CO<sub>2</sub> uptake during the day.  
A straightforward explanation for negative CO<sub>2</sub> fluxes during the day is CO<sub>2</sub> uptake by phototrophic  
520 organisms. Algae and cyanobacteria are well known to have active carbon concentrating mechanisms (CCM) which allow CO<sub>2</sub> uptake also in the dark (Giordano et al. 2005). Phototrophs living at the surface of dry sediments are facing a harsh environment with high salinity in thin water films covering particles and high irradiation and temperature – all factors favoring the activation of CCMs (Beardall and Giordano 2002). Dark CO<sub>2</sub> uptake is a common observation in <sup>14</sup>CO<sub>2</sub> uptake measurements and known to depend  
525 on pre-darkness light conditions (Legendre et al. 1983). In pure cultures it has been shown that CO<sub>2</sub> uptake by algae may proceed for more than an hour in darkness (Goldman and Dennett 1986; Ohmori et al. 1984). Thus, it is highly plausible that the observed CO<sub>2</sub> uptake by dry sediments was caused by photosynthetic algae and/or cyanobacteria. Future studies including chlorophyll analysis of sediments or the application of specific inhibitors may clarify the mechanism behind CO<sub>2</sub> uptake in exposed river  
530 sediments.

## 4.4 Implications

Photosynthetic uptake of CO<sub>2</sub> in the dark would have consequences for the interpretation of dark chamber measurements. If a chamber is placed on the sediment photosynthetic CO<sub>2</sub> uptake may proceed for an unknown period of time. The fact that no net uptake was observed in the night shows that the capability of dark CO<sub>2</sub> uptake could not be sustained for periods longer than one hour, which is consistent with pure culture observations (Goldman and Dennett 1986). However, flux measurements are usually performed within a few minutes making it highly probable that they include eventual photosynthetic CO<sub>2</sub> uptake. Comparison of transparent and opaque chamber measurements are sometimes used to detect photosynthesis of algae. Our results imply that such interpretation have to be treated with care because photosynthetic CO<sub>2</sub> uptake may proceed during dark flux measurements.

Our median CO<sub>2</sub> flux of 98 mmol m<sup>-2</sup> d<sup>-1</sup> would result in annual emissions of 429 g C m<sup>-2</sup> y<sup>-1</sup> which is in the range of fluxes typical for temperate ecosystems (Doering et al. 2011) and similar to fluxes reported for dry Elbe sediments (Mallast et al. 2020), but high compared to the gravel bed of an alpine river (38 mmol m<sup>-2</sup> d<sup>-1</sup>, (Doering et al. 2011)), and low compared to exposed sediments of Mediterranean streams (781 mmol m<sup>-2</sup> d<sup>-1</sup>, Gómez-Gener et al. (2016)). Although our observations thus fit the reported range, these differences as well as the large variation of fluxes observed in our high frequency measurements (-120 to 1135 mmol m<sup>-2</sup> d<sup>-1</sup> – this range is larger than the range of typical fluxes for all kinds of terrestrial ecosystems as compiled by (Doering et al. 2011)) implies that care must be taken when upscaling fluxes for certain ecosystems but also for larger scales.

The observed hysteresis obscures flux-temperature relations if measurements were only performed at one time during the day. Thus, temperature regulation of dry sediment CO<sub>2</sub> emissions might be more relevant and more complex than identified in a recent study (Keller et al. 2020).

Our high frequency measurements show that standard measuring protocols are probably under estimating CO<sub>2</sub> emissions from dry sediments because high fluxes in the night resulting from a delayed temperature response are not considered. The median flux measured between normal working hours (8:00 – 18:00) was 87 mmol m<sup>-2</sup> d<sup>-1</sup> compared to 98 mmol m<sup>-2</sup> d<sup>-1</sup> if all data were considered. Thus, only measuring during daytime would lead to a flux under estimation of 11%. We therefore recommend to assess temporal

shifts in flux-temperature responses in order to obtain better estimates for upscaling based on a representative choice of flux data.

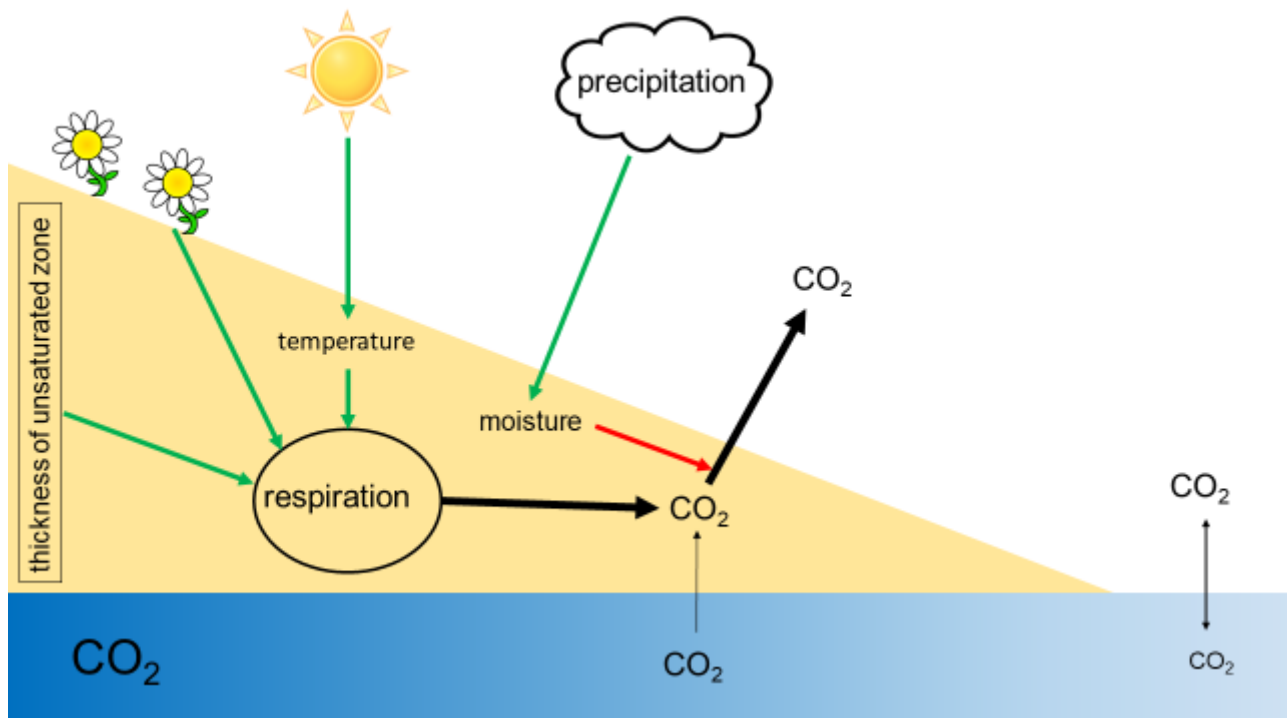
560 Our results are partly contradicting results from (Mallast et al. 2020) who observed highest CO<sub>2</sub> emissions near to the waterline. The two studies, however, are not directly comparable because the previous study by (Mallast et al. 2020) was carried out under extreme drought conditions. Under such conditions deeper lying sediments which tend to be higher in organic matter and less sandy were exposed to the atmosphere. Such conditions should favor CO<sub>2</sub> emissions (Keller et al. 2020). Furthermore the very dry conditions  
565 (<10% sediment moisture) under the extreme drought might have inhibited microbial processes in the sandy sediment. While the drivers of CO<sub>2</sub> emissions from dry sediments are known, their complex interaction makes it difficult to predict CO<sub>2</sub> emissions under a given situation.

The observed relation between CO<sub>2</sub> flux and distance to the river, however, might facilitate upscaling of CO<sub>2</sub> emissions from dry river sediments. The width of the dry sediment zone can be extracted from  
570 satellite images or aerial photographs. The observed consistent spatial pattern also implies that the CO<sub>2</sub> flux was probably not much affected by time after exposure. Thus, combining few diurnal datasets of CO<sub>2</sub> flux and lateral transects with seasonal data of the width of the dry sediments zone along a river is a promising approach to quantify total CO<sub>2</sub> emissions from such systems.

## 5 Conclusions

575 We could clearly show that CO<sub>2</sub> emissions from dry river sediments under the given conditions here were primarily driven by respiration in the sediment. Thus, existing knowledge about soil respiration might also apply to dry river sediments.

We could further show that CO<sub>2</sub> emissions were regulated by temperature and the thickness of the unsaturated zone (Figure 8). The observed hysteresis effect clearly show that simple correlations between  
580 environmental parameters and CO<sub>2</sub> emissions from sediments may be too simplistic to study regulatory mechanisms. Positively spoken the analysis of such hysteresis relations may allow conclusions about underlying mechanisms (Musolff et al. 2021).



585 **Figure 9: Scheme of processes and drivers of CO<sub>2</sub> fluxes from dry river sediments. Green arrows indicate positive, red arrows negative effects.**

Our data show that the occurrence of terrestrial vegetation has a large and not yet assessed impact on the carbon cycle of dry sediments. To assess the effect of vegetation not only ecosystem production has to be quantified but also the fate of plant biomass upon re-flooding. While it is clear that CO<sub>2</sub> emissions from dry river sediments are relevant the exact quantification of the effect of low river levels on the river carbon cycle remains challenging. Short term temporal dynamics-variation is very high and probably equally relevant as seasonal variability. Any attempt to quantify annual GHG emissions or the relevance of dry river sediments for carbon cycling needs to address temporal dynamics of CO<sub>2</sub> emissions from dry river sediments.

## 6 Data availability

595 The high frequency dataset is supplied as a supplement.

## 7 Supplement link

## 8 Author contribution:

MK initiated the study and prepared the manuscript with contributions from all co-authors. LT and MK performed measurements. All authors planned measurements and discussed the results.

## 600 9 Competing interests

The authors declare that they have no conflict of interest.

## 10 Acknowledgements

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