# Sediment Quality Assessment in an industrialized Greek coastal marine area (West Saronikos Gulf)

3 Georgia Filippi<sup>4</sup>, Manos Dassenakis<sup>4</sup>, Vasiliki Paraskevopoulou, Konstantinos Lazogiannis<sup>2</sup>

<sup>4</sup> Laboratory of Environmental Chemistry Department of Chemistry, National and Kapodistrian University of Athens, Athens,
 5 15784, Greece

6 <sup>a</sup>Laboratory of Environmental Chemistry, Athens, 15784, Greece 7

8 *Correspondence to:* Georgia Filippi (mphilippi@chem.uoa.gr)

10 Abstract. Eight sediment cores from the coastal marine area of West Saronikos Gulf have been analyzed for their grain size 11 and geochemistry. The concentrations of eight metals (Al, Fe, Mn, Cu, Cr, Ni, Pb and Zn) along with total organic carbon 12 (TOC) and carbonate content were measured. In cores taken at the deeper stations (above 100m) the analyses were performed 13 only in the prevailing fine fraction ( $f < 63\mu m$ ) while in cores from shallow stations (below 100m) the analyses were performed 14 separately in both fractions fine and coarse ( $63\mu m < f < 1mm$ ). The cores are fairly homogeneous, in terms of carbonates and 15 the down-core variability of % TOC, is characterized by high surficial values that decrease with depth. Metals concentrations 16 from both geological origin (Al, Mn, Cr, Ni) and anthropogenic origin (Cu, Pb, Zn), are higher at the silt and clayin the muddy 17 fraction of sediments than the sand fraction fraction of sediments. The spatial distribution of Al, Fe, Mn, Cu, Pb and Zn in 18 surface sediments presents increasing concentrations from the northeast to the southwest part of the study area, from the 19 shallow to the deeper parts in contrast to Cr and Ni which are increased in the northern nearshore stations. Based on the vertical 20 distributions, the metal to Al ratios of Cu, Pb and Zn show a constant decrease over depth along most cores, indicating the 21 anthropogenic effects to surface sediments, while Fe/Al is constant. Spearman's correlation analysis performed among the fine 22 grain metal contents, demonstrated a strong positive correlation (r > 0.5, p < 0.05) between <u>Al</u>, Fe, Mn, Cu <del>and Cu</del>, Pb and-23 Zn. Moreover, increased The calculated enrichment factors were determined at the fine fraction (f < 63 µmindicate minimal to 24 moderate pollution.) of some sediments. The concentrations of Cr at most surface sediments are higher than the ERL value 25 (81 mg Kg<sup>-1</sup>) but below the ERM value (370 mg Kg<sup>-1</sup>) and the concentrations of Ni are always higher than the ERM value 26 (51.6 mg Kg<sup>-1</sup>). MoreoverIn contrast, the concentrations of Cu, Pb, Zn, at most surface sediments, are below ERL values. The 27 mean effects range medium quotients (mERMq) of surface sediments, based on the overall metal concentrations indicated that 28 the surface sediments of most cores, are moderately toxic. The levels of Cr, Ni, Mn and Zn at most stations are decreased in 29 2017, but the concentrations of Pb and Cu are increased in 2017, compared to a previous study of 2007. The concentrations of 30 Cu, Pb and Zn in the surface sediments of West Saronikos Gulf are lower than levels reported for Inner Saronikos Gulf, and 31 Elefsis BayGul and other polluted hot spot areas of Greecef, owing to a lower degree of urban and industrial development.the 32 smaller industrial zone at the western coast, compared to the numerous polluting activities at the east coast of Saronikos.

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#### 35 1 Introduction

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Sediment cores are one of the most easily accessed natural archives, used to evaluate and reconstruct historical pollution trends
 in aquatic environments. The cores provide data to characterize sediment physical properties and their geochemistry and
 composition. Vertical profiles of heavy metals can present sedimentation rate, changes in diagenetic processes and effects
 evolution of human pressures. Metals released into aquatic systems, undergo several processes, such as adsorption, photolysis,
 chemical oxidation and microbial degredation. Sedimentation depends on the contaminant physicochemical properties, the

42 sediment physical properties, the adsorption capabilities and the partitioning constant at the water-sediment interface. Trace 43 metals removed from the water column are adsorbed on particulate matter and eventually deposited on bottom sediments 44 (Bigus et al, 2014).

Sediments are repositories for metals such as chromium, lead, copper, nickel, zinc and manganese that present as discrete 45 46 compounds, ions held by cation-exchanging clays, bound to hydrated oxides of iron and manganese, or chelated by insoluble 47 humic substances. Solubilization of metals from sedimentary or suspended matter depends on the presence of complexing 48 agents. Metals that are held by suspended particles and sediments are less available than those in true solution (Manahan, 49 2011).

50 Saronikos Gulf (Greece) is a marine area of the Aegean Sea between the peninsulas of Attiki and Argolida. The environmental 51 interest in Saronikos gulf arises from the fact that it is the marine border of the most urbanized areas of Greece, i.e., the 52 country's capital (Athens), the industrial zones of Attiki (Elefsis, Thriasio, Sousaki) and the large port city of Piraeus. As a 53 result, there have been on going environmental monitoring and oceanographic studies of Saronikos since the 80's. There are 54 several published works focusing on the eastern part of Saronikos due to the presence of extended and intensive anthropogenic 55 activities (Scoullos, 1986; Pavlidou et al., 2004; Scoullos et al., 2007; Kontogiannis, 2010; Paraskevopoulou et al., 2014; 56 Pangiotoulias et al., 2017; Karageorgis et al., 2020; Prifti et al., 2022). In contrast, fewer studies have focused on North West 57 Saronikos despite the presence of a less extensive industrial zone hosting a major oil refinery and coastal touristic activities 58 along with a particular geological (volcanic/hydrothermal) background (Paraskevopoulou, 2009; Kelepertsis et al., 2001).

59 The Saronikos Gulf is situated at the central Aegean Sea (north cast Mediterranean) between 37e30'N-38e00'N and 24e01' E-60 23400' E. The length of its coastline is 270 km, the surface is 2.866 km<sup>2</sup> and the mean water depth 100 m. To the north, a 61 shallow (30 m depth) embayment is formed, known as Elefsis bay. The islands of Salamina and Aigina and the plateau between 62 them, divide the gulf into two basins: the western basin (Western Saronikos Gulf) with maximum depths of 220 m in the north 63 and 440 m in the south (Kontoviannis, 2010) and the eastern basin which has a smooth bathymetry with depths of 50-70 m to 64 the north (inner Saronikos Gulf) reaching 200 m to the southeast, from where the gulf opens to the Aegean Sea (outer Saronikos 65 Gulf). To the west, the narrow Isthmus of Corinth connects Korinthiakos Gulf in the Ionian Sea with Saronikos Gulf in

66 the Accean Sea.

67 The gulf is subjected to a strong seasonal eyele of heating and cooling, with air temperatures between 0~40° C, which causes 68 the formation of a seasonal pycnocline from May to November. In winter, the water column is homogenized down to 120 m. 69 However, in the western part, vertical mixing never reached the sea bottom (440 m) in the years after 1992 and dissolved 70 oxygen concentration has approached nearly anoxic conditions (D.O. < 1mL/L) (Paraskevopoulou et al., 2014). 71

The gulf is subjected to intense anthropogenic pressure, as it is the marine border of the eities of Athens and Piraeus with 3-4 72 million inhabitants. Moreover, several point and non-point pollution sources are present. One of the most important point 73 sources is the Athens/Pireaus wastewater treatment plant (WWTP) on the small island of Psittalia, one of the largest in Europe, 74 with a population equivalent (p.e.) coverage of 5.6 million p.e. Other point sources along the coasts include marinas, touristic 75 facilities, fish farms and the effluents of smaller towns and settlements (Paraskevopoulou et al., 2014).

76 The coastal marine area of the north part of West Saronikos Gulf is affected by a few types of industries established there 77 during the 1970's, that include an oil refinery unit at the center of Susaki area, a cable manufacturer, sova mills and sulfur, 78 fertilizers manufacturing for agricultural use and the activate wastewater treatment plant of Aghioi Theodoroi. Moreover, the

79 increased touristic activities in the nearby coastal villages, especially during the summer months (Kelepertsis et al., 2001;

80 Paraskevopoulou, 2009) are important point sources.

81 The Susaki area, which extends parallel to the northern coast of West Saronikos for about 8 Km, is known for its volcanic 82 activity which took place during Pliocene-Quaternary. Most of the volcanic materials were transported by fluvial processes

83 and deposited in the alluvial plains and coastal regions. The formations observed are peridotites and serpentinites, neogene 84 deposits and Quaternary deposits. As a result, elevated values of Cr, Ni, Co, Mn, Fe in the soils and sediments of this area can

85 be explained by the existence of the ultrabasic rocks (Kelepertsis et al., 2001).

86 The main aim of this work is to assess the levels and the distribution of several heavy metals (Al, Fe, Mn, Pb, Zn, Ni, Cr, Cu) 87 in the sediment cores of West Saronikos, in order to discern between the relative contribution of geological and anthropogenic 88 origins of heavy metals and to identify the major sources of metal pollution. The second aim is to determine the evolution of 89 marine pollution in the area by comparing the results with those of a similar study ten years ago, conducted at the Laboratory 90 of Environmental Chemistry (of the Department of Chemistry, of the National and Kapodistrian University of Athens). The 91 last aim of this work is to assess and highlight the differences between the concentrations of heavy metals in the surface 92 sediments of the-West Saronikos in comparison to sediments in studies conducted in East Saronikos, Elefsis bay and other 93 areas of Greece.Gulf and the east part of the gulf.

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#### 96 2 Study area Materials and methods

97 The Saronikos Gulf is situated at the central Aegean Sea (north east Mediterranean) between 37°30'N-38°00'N and 24°01' E-98 23000' E formed between the peninsulas of Attiki and Argolida. The length of its coastline is 270 km, the surface is 2.866 km<sup>2</sup> 99 and the mean water depth 100 m. To the north, a shallow (30 m depth) embayment is formed, known as Elefsis bay. The .00 islands of Salamina and Aigina and the plateau between them, divide the gulf into two basins: the western basin (Western 101 Saronikos Gulf) with maximum depths of 220 m in the north and 440 m in the south (Kontoyiannis, 2010) and the eastern .02 basin which has a smooth bathymetry with depths of 50-70 m to the north (Iiner Saronikos Gulf) reaching 200 m to the .03 southeast, from where the gulf opens to the Aegean Sea (Oouter Saronikos Gulf). To the west, the narrow Isthmus of Corinth 104 connects Korinthiakos Gulf in the Ionian Sea with Saronikos Gulf in the Aegean Sea (Kontoyiannis, 2010; Paraskevopoulou, 105 2014).

The gulf is subjected to a strong seasonal cycle of heating and cooling, with air temperatures between -0 - 40 ° C, which causes
 the formation of a seasonal pycnocline from May to November. In winter, the water column is homogenized down to 120 m.
 However, in the western part, vertical mixing never reached the sea bottom (440 m) in the years after 1992 and dissolved
 oxygen concentration has approached nearly anoxic conditions (D.O. < 1mL/L) (Paraskevopoulou et al., 2014).</li>

.10 A few circulation studies have been conducted in Saronikos in order to discern the possible movement of the treated wastewater .11 effluent discharged at 65m south of Psittalia island in the eastern basin. The circulation is reported as strongly dependent on 112 local winds with predominant northerly direction throughout the year. However, westerly and southerly winds may also occur 113 in fall, winter and spring. Under the predominant northerly wind regime in summer and fall during the presence of the seasonal .14 pycnocline there is a distinctly different two-layer circulation above and below 60m. In the upper layer there is a general .15 eastward anticyclonic (clockwise) flow from the west to the east basin. Below 60m the flow is reversed from the northeast .16 through the Salamina-Aigina passage to the southwest towards the deeper part of the west basin following a cyclonic 117 (counterclockwise) path. During winter when the water column is fully mixed down to 90m the general flow is anticyclonic .18 from the west to east. Nearshore on the northwest Sousaki coast the currents reported are directed from the north to the south. .19 In spring during the pycnocline formation there is no continuous flow structure spanning both the west and east basins. In the 120 west during spring there is a rather strong north to south flow (SoHelME, 2005; Kontoyiannis, 2010). 121

# The gulf\_Saronikos gulf, as a whole, is subjected to intense anthropogenic pressure, as it is the marine border of the cities of Athens and Piraeus with 3-4 million inhabitants. Moreover, sSeveral point and non-point pollution sources are present mainly

0 on the northeastern coasts, One of the most important point sources is the Athens/Pireaus wastewater treatment plant (WWTP)

on the small island of Psittalia, one of among the largest in Europe, with a population equivalent (p.e.) coverage of 5.6 million.
 p.e. Other point sources along the coasts include the port of Piraeus, marinas, touristic facilities, fish farms and the effluents of smaller towns and settlements. Finally, pollution pressure arises from increased marine traffic, since Piraeus is one of the largest and busiest Mediterranean ports, and from the heavy vehicle traffic and the heating systems in the extended urban areas (Paraskevopoulou et al., 2014).
 The coastal marine area of the north part of West Saronikos Gulf is affected by a few types of industries established there

.31 during the 1970's, that include a majorn oil refinery unit at the center of Susaki area, a cable manufacturer, soya mills, and 132 sulfur /5 fertilizers manufacturing for agricultural use and the increased sewage load from nearby coastal villages due to the 133 summer tourist season (Kelepertsis et al., 2001; Paraskevopoulou, 2009). The settlements on the coast of West Saronikos, to 134 the best of our knowledge, are not yet connected to wastewater treatment facilities and a projected activate wastewater .35 treatment plant of in Aghioi Theodoroi is under construction. Moreover, the increased touristic activities in the nearby coastal 136 villages, especially during the summer months (Kelepertsis et al., 2001; Paraskevopoulou, 2009) are important point sources. 137 The Susaki area, which extends parallel to the northern coast of West Saronikos for about 8 Km, is known for its volcanic .38 activity which took place during Pliocene-Quaternary. Most of the volcanic materials were transported by fluvial processes .39 and deposited in the alluvial plains and coastal regions. The formations observed are peridotites and serpentinites, neogene .40 deposits and Quaternary deposits. As a result, elevated values of Cr, Ni, Co, Mn, Fe are found in the soils and sediments of L41 this area and can be explained by the existence of the ultrabasic rocks (Kelepertsis et al., 2001). The southwest deeper part of .42 West Saronikos (Epidavros basin) could be affected by the transport of organic and inorganic pollutants from the eastern basin .43 due to the periodical northeast to southwest water flow below the pycnocline (Psyllidou-Giouranovits and Pavlidou, 1998; L44 Kontoyannis, 2010).

### 45 <u>3. Materials and methods</u>

Eight <u>short</u> sediment cores (12-32 cm) were obtained at corresponding <u>number of</u> stations with varying depths (50-420 m) in the area of West Saronikos Gulf <u>using-with the use of</u> a box corer. The sampling <u>took was conducted place oin</u> 18 October 2017 with the Greek Oceanographic vessel RV *Aegaeo*. The <u>location of</u>-West Saronikos Gulf study area and the specific <u>location station</u> of stations are presented in Fig. 1.

#### 151 <u>Table 1. Coordinates, water column depth and core length at each sampling station.</u>

Station	Latitude N (dec.minutes)	Longtitude E (dec.minutes)	Depth (m)	Core length (cm)
MOT13A	<u>37º 54.602</u>	23º 03.184	<u>50</u>	<u>12</u>
MOT16A	<u>37º 53.995</u>	23º 03.080	<u>100</u>	<u>32</u>
UN5	<u>37º 53.459</u>	23º 04.393	<u>140</u>	<u>32</u>
<u>MOT16</u>	37º 54.179	23° 05. 312	<u>85</u>	<u>20</u>
UN6	37º 53.455	23º 10. 857	<u>193</u>	<u>26</u>
UN6A	<u>37º 51.610</u>	23º 15.932	<u>165</u>	<u>24</u>
<u>UN4</u>	<u>37º 57.057</u>	23º 20.331	<u>79</u>	22
<u>UN11</u>	<u>37º 38.800</u>	23º 15.338	<u>420</u>	<u>32</u>

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Stations MOT13A, MOT16A, UN5, MOT16, UN6 (near the Susaki area) and UN4 (Megara basin), at the north-western part,
 are affected by the coastal industrial zone, urbanization and touristic activities. The offshore station UN6A, at the middle of
 Megara basin, is probably less disturbed by anthropogenic activities. Finally, station UN11, in Epidavros basin at the southwest
 part of the study area, is influenced by trawling and aquaculture and potential transport of organic and inorganic pollutants
 from eastern Saronikos due to periodical circulation patterns in the water layers below 60m.



Figure 1: Location-Study area of West of Saronikos Gulf and sampling stations (<u>The map was constructed using Arc Map with</u>
 bathymetry representation data from the European Marine Observation and Data Network-EMODnet: https://portal.emodnet bathymetry.eu/) and © Google Mapland cover data from Corine 2000/s-2019). Stations MOT13A, MOT16A, UN5, MOT16, UN6,
 UN6A and UN4 locate at the northwest part of Saronikos Gulf and station UN11 at the southwest part.

168 Stations MOT13A, MOT16A, UN5, MOT16, UN6 (near the Susaki area) and UN4 (Megara basin), at the northwestern part,

169 are affected by the coastal industrial zone, urbanization and touristic activities. The offshore station UN6A, at the middle of

Megara basin, is probably less affected by anthropogenic activities. Finally, station UN11 locates at the southwest part of the
 gulf, at Epidavros basin and is affected by trawling and aquaculture.

172 The cores were stored frozen immediately after sampling until analysis. Subsequently, they were and cut in 1cm layers down 173 to the top 10cm of each core and 2cm layers layers of 1-2cm for the top 10cm and of 2cm below 10 cm after that. -and-The 174 separated layers they were stored frozen until analysisfurther processing.- The initial step of analysis is removal of water .75 content with the use of a Lab Conco freeze-dryer. The separated layers were then freeze dried. Subsequently, The grain size .76 treatment-analysis via dry sieving was performed using the 1mm and 63µm Retsch stainless steel sieves. The included dry .77 sieving for the separation of the gravel (>1mm), sand (> 63 µm) and silt and clay muddy fraction (<63 µm) fractions were 178 separated for the calculation of the respective percentages from the sand fraction (> 63µm) (Tsoutsia et al., 2013), using Retsch .79 stainless steel 'Test Sieves'. The percentage of both fractions (sand and silt clayto the total sediment) was calculated. The 180 gravel fraction was discarded. The percentage of tTotal organic carbon, and cacarbonates and the concentrations of heavy 181 metals were determined separately in both sand and fine-muddy sediment, when the fraction percentage was more than 10 % 182 of the total sediment or in the prevailing fraction only (above 90%). Table 1 presents the coordinates, and the depth of each .83 sampling station and along with the corresponding core length of sampling stations.

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#### L85 Table 1. The location and depth of sampling stations and the length of each core.

Station	<del>Latitude N</del> <del>(dec.minutes)</del>	Longtitude E (dec.minutes)	<del>Depth (m)</del>	Length of Cores (cm)
MOT13A	<del>37° 54.602</del>	<del>23° 03.184</del>	<del>50</del>	<del>12</del>
MOT16A	<del>37<sup>0</sup> 53.995</del>	<del>23°-03.080</del>	<del>100</del>	<del>32</del>
UN5	<del>37° 53.459</del>	<del>23° 04.393</del>	<del>140</del>	<del>32</del>
MOT16	<del>37° 54.179</del>	23° 05. 312	85	<del>20</del>
UNG	<del>37° 53.455</del>	23° 10. 857	<del>193</del>	<del>26</del>
UNGA	<del>37° 51.610</del>	23° 15.932	<del>165</del>	<del>2</del> 4
UN4	<del>37° 57.057</del>	<del>23° 20.331</del>	<del>79</del>	22
UN11	<del>37º 38.800</del>	23º 15.338	<del>420</del>	<del>32</del>

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The total organic carbon (TOC) content was measured using the standard Walkley method (Walkley, 1947) method as modified
by Jackson (Jackson, 1958) and Loring & and Rantala (Loring and Rantala, 1992), which is based on the exothermic reaction
(oxidation) of the sediment with potassium dichromate (K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>) and concentrated sulfuric acid (H<sub>2</sub>SO<sub>4</sub>), followed by backtitration with ferrous ammonium sulfate (FeSO<sub>4</sub>) and ferroine indicator.

191 The carbonate content was determined by calculating the weight difference of the sample before and after the strong

effervescence caused by adding hydrogen chloride (HCl) 6 M to the sediment causing an (exothermic reaction followed by
HCl gas and CO<sub>2</sub> emission (Loring and Rantala, 1992). ), a method modified from Loring and Rantala (Loring and Rantala, 1992).
194 1992).

The total metal contents were extracted via complete dissolution of sediment samples with an acid mixture of HNO<sub>3</sub>-HClO<sub>4</sub>-HF (ISO-14869-1:2000) (Peña-Icart et al., 2011). Then, the total metal concentrations were determined by Flame or Graphite Furnace Atomic Absorption Spectroscopy (FAAS-Varian SpectrAA-200 and GFAAS- Varian SpectrAA 640Z) (Skoog et al., 1981). In order to evaluate the precision and accuracy of the method for total metal analysis <u>a</u>\_certified reference <u>material</u> materials (ISE 921, 80MS, (PACS-3, ) from Wepal, Quasimeme and NRC-CNRC) wasere carried through the analytical procedure along with the sediment samples in every digestion batch and 1 or 2 random layers of each core were also analyzed in duplicate. Accuracy was calculated as % recovery (percentage ratio of the measured to the certified value). The precision

#### Μορφοποίησε: Αγγλικά (Ηνωμένων Πολιτειών)

202 was evaluated using the % RSD-Relative Standard Deviation (percent ratio of the standard deviation to the average 203 concentration of the replicates) calculated for each metal by each of the duplicate measurements (repeatability estimation) and 204 the multiple measurements of the reference material (reproducibility estimation).by replicate analysis (n=3) of the reference 205 materials under reproducibility conditions (different days of digestion and measurement) and the % RSD-Relative Standard 206 Deviation (percent ratio of the standard deviation to the average concentration of the replicates) was calculated for each metal. 207 The quality data for the total metal method are presented in the Appendix (Table A1). The precision and show that all recoveries 208 generally fall into the ranges 3-10% RSD and 80-120 % recovery, depending on analyte level. Such results of analytical 209 performance are anticipated for multistage analysis of solid samples and are roughly are between the recommended by the US 210 Environmental Protection PAgency (US EPA) and the Association of Official Analytical Chemists (AOAC) (US EPA, 1996; 211 AOAC International, 2016)ranges (75-125%). For every core the %RSD for each metal was calculated as an indication of the downcore variability and the results are Moreover, the ranges of % RSD for each metal at the collected cores are presented at 212 213 in Fig. A1 - A15 (Appendix A). 214 All-Most statistical treatment of data and the vertical distribution graphs were was performed by Microsoft Excel 2010.

215 Moreover, the graphs with the vertical distributions were plotted with Microsoft EXCEL 2010 and tThe horizontal distributions 216 of metals were visualized with the software package Ocean Data View (ODV) 2017. The software IBM-SPSS Statistics 2020 217 was used for statistical comparisons between stations regarding the trace metal concentrations and Spearman correlation 218 analysis to identify significant relationships between different heavy metals, total organic carbon and carbonates. The 219 comparison between stations was done by performing parametric (One-Way Anova) and non-parametric (Kruskal Wallis, 220 Kolmogorov-Smirnof) tests between the mean values and standard deviations calculated for each metal and each core by the 221 results of the top 5 cm. The comparison between two groups of values, e.g. a variable between two cores or concentrations of 222 variables above-below a certain layer in a single core was either done by two-sided t-test or by the non-parametric equivalents 223 Mann-Whitney and Kolmogorov-Smirnov tests. In all comparative tests statistical results were calculated at the 95% 24 confidence level (p<0.05). A Spearman correlation analysis that was performed with the statistical software IBM SPSS 225 Statistics 2020-was used to identify the significant relationship between different heavy metals, total organic carbon and 226 carbonates.

#### 227 <u>4</u>3 Results

#### 228 <u>43</u>.1 Geochemical\_-results

229 Table 2 summarizes the main findings from the determination of geochemical parameters. The grain size in cores from stations 230 MOT16A, UN5, UN6, UN6A, UN11, is dominated by elay and siltmud-(f < 63µm), while the percentage of sand fraction 231 (63µm < f < 1mm) is generally low and below 10% in most core layers.lower than 10 %, which can be attributed to the depth 232 of these stations (above 100 m). On the other hand, the grain size in cores MOT13A, MOT16, UN4 is dominated by the 233 percentages of both sand (39-66%), which can be explained by the shallow depth of these stations (depth below 100 m) and 234 their proximity to the northwestern coast. However, and the percentages of clay and siltmud (15-50%) in cores MOT13A, MOT16, UN4 are significantnot negligible (above 10 %). The sediments at stations MOT13A and MOT16 are coarser with 235 236 average mud percentages 25 and 38% respectively, while at station UN4 especially in the top 10cm the average mud percentage 237 is slightly increased (43%).

As a result, the percentages of total organic carbon (TOC) andrd carbonates as well asand the concentrations of heavy metals
 were determined in the <u>fine-mud</u> fraction (f-<-63μm) of sediments <u>in in cores MOT16A</u>, UN5, UN6, UN6A and UN11 and
 <u>both separately in the sand and fine-mud</u> fraction of sediments in cores MOT13A, MOT16 and UN4.

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242 Table 2. Summary statistics of variables measured grain size percentages, TOC and carbonate content along the collected cores.

Station	% s	and	% <u>mud (</u> s	ilt and clay)	% 1	ГОС	% 0	CO3 <sup>2-</sup>
	Min	Max	Min	Max	Min	Max	Min	Max
MOT13A	51	66	15	38	0.45	0.92	26	28
MOT16A	1	23	77	99	0.10	0.93	22	23
UN5	1	14	86	99	0.44	1.12	20	23
MOT16	47	65	29	45	0.33	1.28	20	22
UN6	1	32	68	99	0.57	2.44	19	23
UN6A	1	22	78	99	0.33	1.32	22	25
UN4	39	60	20	50	0.51	0.75	29	33
UN11	0	15	74	100	0.77	2.35	15	19

Apart from small variations, the cores are fairly homogeneous, in terms of carbonates. The high percentages of % CO322 in cores MOT13A and UN4 are associated with the coarse\_- grained samples and the abundant presence of shell fragments. Moreover, the down-core variability of % TOC at the collected cores, is characterized by high surficial values that decrease with depth. Figure 2 presents the vertical distribution of % TOC in selected cores. Comparing between stations the lowest TOC values in the top 5cm are measured in the coarser sediments of MOT13A (average 0.57%) and the highest in the deepest station UN11 (average 1.74%). There was no statistical difference between TOC in the remaining cores and the average values for the top 5cm are the following: UN4 (0.68%), MOT16A (0.69%), MOT16 (0.96%), UN5 (0.96%), UN6A (1.09%) and UN6 (1.26%). The distribution at core UN4 refers to the fine sediment fraction (f < 63  $\mu$ m).





<sup>255</sup> 256 257 258 259

Figure 2: Vertical distribution of % TOC at <u>selected</u> cores <del>UN5, UN6A, UN4, UN11. [The distribution at core UN4<u>MOT16 profile</u> refers depicts to the</del> fine sediment fraction (f< 63 μm) <u>TOC content]</u>.

The vertical TOC distribution in core MOT16 is differentiated than the corresponding in the adjacent station MOT13A. Below 6cm the levels of TOC are similar in both stations. However, in the top 6cm of core MOT16 average TOC in the total sediment is approximately 1% while below 6cm the corresponding average is 0,42%. This is caused by a striking increase in the TOC content of the muddy sediments, with an average of 2.4% TOC at the top 6cm approximately 5-fold higher than the corresponding in the deeper layers (0.5%). At the same time the TOC variability in the sandy fraction of the top 6cm of MOT16 is minimal, ranging between 0.3% and 0.4%. The vertical TOC distributions of the dominantly muddy cores (MOT16A, UN5, UN6, UN6A, UN11) present high surficial content and gradual increase with depth.

Table 3 presents TOC, carbonate and heavy metal contents in <u>the two fractions (sand-mud) at</u> the shallow coarse-grained cores

MOT13A, MOT16, UN4. It is apparent that the % TOC of the silt and clay sediment fractioncontent of the muddy material is

269 higher than the corresponding content of the sandy fraction as anticipated. The % CaCO<sub>3</sub> content is increased of in the sandy 270 material of cores MOT13A, UN4fraction is increased compared to that of the fine\_fraction sediment in cores MOT13A, UN4 271 and is approximately equal in the two sediment fractions of core MOT16. 272 The concentrations of Al, Cr, Cu, Mn, Pb, Ni and Zn are higher in the silt-claymuddy sediments of MOT13A and UN4 than 273 the corresponding values in sandy sediments. The same applies to Al, Mn, Cu, and Zn in MOT16. The high content of Al at 274 the fine fraction of sediments indicates that Al is predominantly associated with aluminosilicate minerals and occurs mostly in 275 the clay minerals. Generally, fine sediments tend to have relatively high trace element concentrations, due to the surface 276 adsorption and ionic attraction. Especially, the so-called anthropogenic trace metals (Cu, Pb, Zn) are normally bound within 277 or sorbed by the elay mineral fraction of sediments (Barjy et al., 2020; Karageorgis et al., 2005).-Unlike other metals the Fe 278 content is more or less similar in both sediment fractions of cores MOT13A and UN4. The sediments of core MOT16 appear 279 to be different with higher concentrations of Fe, Cr, Ni and Pb in the coarse-grained fraction. 280 The vertical distributions of the studied metals in mg kg<sup>-1</sup> along the collected cores are presented in Figures A1-A15 (Appendix 281 A). In cases of coarse-grained cores MOT13A, MOT16, UN4, the concentrations in the total sediment (calculated by the 282 corresponding values in both fractions) are depicted. Table A2 (Appendix A) presents the ratios of eight heavy metals to Al in 83 the surface and deeper sediment layer of the collected cores. The ratios presented for the cores MOT13A, MOT16, UN4 are 284 calculated by the results in the corresponding fine-grained sediments. The vertical profiles of metal to aluminum ratios (in fine 285 grained sediments) along the collected cores, are also given in Figures A1-A15 (Appendix A).

Table 4 summarizes the concentrations of eight heavy metals in the surface and deeper sediment layer of the collected cores.
 For the sake of direct comparison in the cases of cores MOT13A, MOT16, UN4 the fine grained concentrations are presented in Table 4. The sediments at the deeper parts of West Saronikos (cores UN6, UN6A and UN11) present elevated concentrations of Al, Fe, Pb and Zn. The sediments of UN11 are also particularly enriched in Mn. On the contrary the values of Cr and Ni are elevated in cores MOT13A, MOT16 and MOT16A.

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Table 3. <u>Concentrations of metals (in mg Kg<sup>-1</sup>), Percentage of organic and inorganic carbon and concentrations of metals incontent</u>
 mg Kg<sup>-4</sup> in the two sediment fractions of coarse-grained cores (MOT13A, MOT16, UN4).

Variable/Core	MC	DT13A	М	OT16	U	N4
	fine fraction	coarse fraction	fine fraction	coarse fraction	fine fraction	coarse fraction
% TOC	0.64-2.80	0.26-0.45	0.37-2.70	0.15-0.46	0.65-0.94	0.43-0.58
% CaCO₃	22-23	27-30	21-23	18-22	26-29	32-36
AI	10561-18387	6615-9226	21677-28939	10610-18538	23271-34246	10449-19765
Cr	390-651	333-486	322-374	306-517	113-133	71.6-115
Ni	293-411	220-302	314-573	424-697	139-187	78.0-140
Fe	19878-21153	17430-20643	24130-31694	32265-36080	14149-17126	13501-17931
Mn	429-476	326-386	471-530	435-484	317-366	174-238
Cu	17.1-18.8	8.9-11.8	16.7-22.6	6.7-14.8	15.1-23.6	9.7-13.4
Pb	18.0-30.7	14.8-21.0	9.1-28.0	8.3-31.9	11.2-31.2	9.3-27.3
Zn	39.3-51.8	31.0-45.6	39.8-53.7	31.5-50.9	38.9-59.2	30.3-50.3

The vertical distributions of the study metals in mg kg<sup>+</sup> along the collected cores present at Fig. A1–A15 (Appendix A). In
 cases of coarse cores MOT13A, MOT16, UN4, the concentrations of the total sediment (both fractions) are depicted.
 Table 4 summarizes the concentrations of eight heavy metals in the surface and deeper sediment layer of the collected cores.
 The concentrations are measured at the fine sediment (fraction (f < 63µm) of cores MOT13A, MOT16, UN4.</li>

The sediments at the deeper parts of West Saronikos (cores UN6, UN6A and UN11) present elevated concentrations of Al, Fe,
 Pb and Zn especially. The sediments of UN11 are also particularly enriched in Mn, which is attributed to the prevalence of the
 silt clay sediment fraction and the suboxic waters at depths higher above 200 m (Kontoyiannis, 2010). The elevated values of

805 Cr and Ni in cores MOT13A, MOT16 and MOT16A can be explained by the existence of the ultrabasic rocks of the Susaki 806

area, in which, these metals are predominant (Kelepertsis et al., 2001).

308 309 Table 4. Concentrations in mg Kg<sup>-1</sup> of the study-metals in the surface and the deeper sediment layer of cores. The concentrations at coarse-grained cores MOT13A, MOT16, UN4 are measured at refer to the fine sediment fraction ( $f < 63\mu m$ ). 310

Core	Layer (cm)	Al	Cr	Ni	Fe	Mn	Cu	Pb	Zn
MOT13A	0-1	10561	651	411	20988	476	18.2	26.1	49.6
	10-12	14673	407	383	21153	444	18.1	22.5	44.6
MOT16A	0-1	27009	280	375	23315	578	22.6	20.4	48.3
	30-32	30070	256	382	26179	562	22.1	28.1	44.1
UN5	0-1	32705	199	305	25534	635	27.4	42.7	74.5
	30-32	37885	223	320	29170	520	26.0	24.7	57.5
MOT16	0-1	21677	369	377	25816	530	22.6	28.0	53.7
	18-20	23807	348	401	29752	482	16.9	9.1	40.4
UN6	0-1	43264	142	253	27301	954	36.7	52.9	92.1
	24-26	44547	153	274	29345	707	28.5	32.5	62.4
UN6A	0-1	39314	146	187	21838	570	26.3	38.4	73.8
	22-24	41303	161	209	23827	513	25.4	38.7	58.7
UN4	0-1	34246	132	162	15762	358	23.6	26.1	52.1
	20-22	31706	113	160	17112	365	15.1	12.0	40.3
UN11	0-1	54626	142	230	32177	3925	49.7	63.9	110
	30-32	48186	163	217	31573	1459	37.3	35.1	71.4

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812 The distribution of elements comprising the terrigenous phase of the sediments is best represented by Al, which is held almost 813 exclusively in terrigenous aluminosilicates (Karageorgis et al., 2005). Table A1 (appendix A) presents the ratios of eight heavy 814 metals to Al in the surface and deeper sediment layer of the collected cores. The ratios in coarse cores MOT13A, MOT16, 815 UN4 are calculated for the fine sediment fraction. The vertical profiles of ratios along the collected cores, are given in Fig. 816 A1=A15 (Appendix A), where the ratios in coarse cores MOT13A, MOT16, UN4 refer to the fine sediment fraction (f <63 µ). 817 The vertical distributions of Al (Fig. A1 (appendix A)) for the total sediment (both fractions) present minimal variation. The 818 down-core variability of Al is lower than 10 %, in the finer sediments (cores-MOT16A, UN5, UN6, UN6A and UN11) and 819 between 10-15 % in the coarser sediments (cores MOT13A, MOT16 and UN4). In the surface layer (0-1cm) of cores MOT13A 820 and UN5 there is a sharp decrease of Al content. Furthermore, -the average aluminum content (16709mgKg<sup>-1</sup>) at the top 6cm 321 of core MOT16 is statistically lower (two-sided t-test, p<0,05) than the corresponding average content in the layers below 822 (20477mgKg-1). On the contrary, while at the top 12cm of core UN11 aluminum is increased (average 52714mgKg<sup>-1</sup>) is 323 increased (two-sided t-test, p<0,05) compared to the deeper layers (average 48621 mgKg<sup>-1</sup>). The other so-called lithogenic 824 metals (Cr, Ni, Fe, Mn) generally present uniform vertical profiles with minimal variability, mostly below 10 % (Figures A2-825 A9). However, Cr, Ni, Fe and Mn in the surface sediments of MOT13A present an increasing trend, that remains pronounced 326 after normalization to Al in the muddy fraction at the top 0-1cm and is concurrent with the 2-fold decrease of Al content. The 827 vertical profiles of Cr and normalized Cr/Al in UN11 indicate a decrease of Cr in the upper sediment layers. This decrease is 828 confirmed statistically and the average Cr (148mgKg<sup>-1</sup>) and Cr/Al (28,0) values of the top 12cm are lower than the 329 corresponding in the deeper layers (155 mgKg<sup>-1</sup>and 32,0, respectively). The same trend of decrease in the top 0-1cm is seen in 830 the normalized profiles of Fe, Ni and Mn in the fine-grained sediment fraction of core UN4. The average concentrations of Fe, 331 Mn, and Ni in the top 12cm of UN11 are statistically higher than the corresponding in the deeper layers. The most pronounced 332 difference is calculated for Mn, where the average upper layer content is 2396 mgKg<sup>-1</sup> while the deeper layer content is 1527 833 mgKg<sup>-1</sup>. The same applies to Mn/Al ratio in UN11 but not to the ratios of Fe/Al and Ni/Al which are seemingly lower in the 834 upper layers like Cr/Al but statistically equal. The down-core variability of Mn in all stations, except UN4, is typical of shelf 335 sediments, with high surficial Mn concentrations and Mn/Al ratios that diminish with depth 336 Similarly, the so-called lithogenic metals (Cr, Ni, Fe, Mn) present uniform vertical profiles with minimal variability (mostly 837 below 10 %). The trend for increase of Cr, Ni, Fe and Mn in the surface sediments of MOT13A remains pronounced after

338 normalization to Al in silt and clay fraction at the top 0-1cm. The normalized vertical profile of Cr in UN11 indicates a decrease

339	of Cr through time in the upper sediment layers. The same trend is seen in the normalized profiles of Ni and Mn in fine
340	sediment fraction of core UN4. The down-core variability of Mn in all stations, except UN4, is typical of shelf sediments, with
341	high surficial Mn concentrations that diminish with depth to background values, as reducing conditions develop, These
342	variations are largely independent of lithological or carbonate content fluctuations, being dependent solely upon the respiration
343	of organic carbon (Karageorgis et al., 2005).
344	The concentrations of Cu, Pb and Zn as well as the normalized profiles (Fig. A10-A15) show a constant decrease over depth
345	to background levels, which can be attributed to increased inputs by anthropogenic activities in recent time (Karageorgis et al.,
346	2005)Figure 3 presents selected vertical profiles of Mn and Pb along core UN11_and those of Cr, Fe, Cu, Zn at the

847 eoresstations UN6A, MOT16A, UN4, UN6, respectively. Figure 4 presents selected vertical distributions of metal to Al ratios 348 at cores MOT13A, UN5, MOT16A, MOT16 and UN11. The concentrations and ratios were calculated presented in Figures 3 and 4 refer to at the muddy fine-fraction (f<63µm) of the sediments in the cases of the coarse-grained cores (MOT13A, 849 350 MOT16, UN4).







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Core UN6A	Core UN11	Core MOT16A	Core UN4	Core UN11	Core UN6
Cr (mg Kg <sup>-1</sup> )	Mn (mg Kg <sup>-1</sup> )	Fe (mg Kg <sup>-1</sup> )	Cu (mg Kg <sup>-1</sup> )	Pb (mg Kg <sup>-1</sup> )	Zn (mg Kg <sup>-1</sup> )
140 160	1000 6000	23000 26000	10 20 30	2045 70	50 75 100
0 5 15 20 25	0 5 10 15 20 7 30  35	0 5 10 15 20 25 30 35	0	0	0

354 355 356 357 Figure 3: Vertical profiles of Cr, Fe, Cu, Zn at the fine muddy fraction (f < 63µm)sediments of cores UN6A, MOT16A, UN4, UN6 respectively and Mn, Pb the vertical distributions of Mn and Pb along core UN11.

\$58 Figure 4 presents selected vertical distributions of ratios to Al at cores MOT13A, UN5, MOT16A, MOT16 and UN11 of West

859 Saronikos Gulf. The ratios were calculated at the fine fraction (f < 63 µm) of the sediments. Based on the vertical distributions,

860 Fe to Al ratios are constant with depth of the collected cores (Nolting et al., 1999). Ration of Cu, Pb and Zn to Al show a

861 constant decrease over depth along the collected cores, because the surface sediments are affected much higher by 362 anthropogenic activities than the deeper sediments (Karageorgis et al., 2005).







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Figure 4: Vertical distributions of ratios to Al at cores MOT13A, UN5, MOT16A, MOT16 and UN11 of the-northwest Saronikos Gulf (profiles for MOT13A and MOT16 depict - The ratios were calculated at for the fine muddy fraction) (f < 63 µm) of the sediments of cores MOT13A and MOT16.

869 The concentrations of Cu, Pb and Zn as well as the normalized profiles (Fig. A10-A15) show a constant decrease over depth 870 to background levels. That tendency is less pronounced with no statistical difference between upper and deeper sediment layers 871 in very few cases and particularly for Cu (MOT13A, MOT16, MOT16A), Cu/Al (MOT13A), Pb (MOT13A, MOT16A), Pb/Al 872 (MOT16A) and Zn (MOT16A). In all remaining cores the concentrations of Cu, Pb and Zn and the ratios to aluminum in the

373 upper sediment layers (above 10cm) are statistically higher than the corresponding in the deeper layers.

#### 874 34.2 Horizontal distributions

were used.

375 Figure 5 presents the horizontal distributions of heavy metals in the surface sediments (0-1 cm) of the study area. In cases of

coarse surface sediments of MOT13A, MOT16, UN4, the concentrations of total sediment fraction (f-sand and mud) <- 1mm)

378	The concentrations of Al, Fe, Mn, Cu, Pb and Zn are plotted increased from the northeast to the southwest area of West
379	Saronikos Gulf. The high content of Mn at the surface sediment of core UN11 can be explained by the prevalence of silt and
380	clay sediment fraction and the suboxic waters at the depth of 420 m (Ozturk, 1995). The waters of West Saronikos Gulf at
381	depths higher than 200 m, are suboxic (Kontoyiannis, 2010) and as a result, the slow diffusion of dissolved oxygen from the
382	more oxidizing overlying waters and the upward diffusion of dissolved Mn (II) from the pore water of anoxic surface sediments
383	to the sediment/water interface (Ozturk, 1995), cause the oxidation of dissolved Mn (II) and its precipitation as Mn (IV) oxides
384	(Pohl and Hennings, 1999).
385	On the other hand, the concentrations of Fe, Cr and Ni at the north part are-appear higher than those at the south area, which

can be explained by the existence of the ultrabasic rocks in the soils of the region and the natural weathering and transport to
 the coastal marine environment (Kelepertsis et al., 2001).

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AAI (mg Kg<sup>-1</sup>) Cr (mg Kg<sup>-1</sup>) 38°N 37.9 37.9 37.8 37.8 37.7 37.7°N 37.6 37.6°N 22.8°E 23.6°E 23°E 23.2'E 23.4' Ni (mg Kg<sup>-1</sup>) 23.2"E 23.4"E Fe (mg Kg<sup>-1</sup>) 23.6°E 23°E



Figure 5: The horizontal distributions of heavy metals in the surface sediments of West Saronikos Gulf. In cases of coarse surface sediments of stations MOT13A, MOT16, UN4, the concentrations of the total sediment fraction (sand and mud) [<1mm] were used.

In order to corroborate the above plots and produce solid statistical comparisons for the metal concentrations between stations
 the average values and standard deviation from the results of the top 5 cm were calculated and examined used the IBM-SPSS

Statistics 2020 software. In most cases of metals and stations the 5 value groups followed the normal distribution therefore the
 parametric One-Way Anova test was first applied. But since some cases did not follow the normal distribution non-parametric

15 parametric One-way Anova test was inst applied. But since some cases durinot follow die normal distribution non-parametric tests (Kruskal-Wallis for multiple groups and Mann-Whitney / Kolmogorov-Smirnov for two groups) were also applied to

- 417 corroborate the statistical comparison outcome. The results summary is the following:
- 418 AI: UN11 >> UN6 = UN6A > UN5 > MOT16A > UN4 = MOT16 >> MOT13A
- **419** Fe: UN11 = MOT16 > UN6 = UN5 > MOT16A > MOT13A = UN6A > UN4**419**Fe: UN11 = MOT16 > UN6 = UN5 > MOT16A > MOT13A = UN6A > UN4
- 420 Mn: UN11 >> UN6 > MOT16A = UN6A = UN5 = MOT16 > MOT13A > UN4
- 421 Cr: MOT13A = MOT16 > MOT16A > UN5 > UN6A = UN11 = UN6 > UN4
- **422** <u>Ni: MOT16 > MOT16A > UN5 = MOT13A = UN6 > UN11 > UN6A > UN4</u>
- 423 <u>Cu: UN11 > UN6 = UN5 > UN6A > MOT 16A > UN4 = MOT16 > MOT13A</u>
- 424 <u>Pb: UN11 = UN6 > UN5 = UN6A > UN4 = MOT16 > MOT 16A = MOT13A</u>
- **425** <u>Zn: UN11 > UN6 > UN5 = UN6A > MOT16 = UN4 = MOT 16A = MOT13A.</u>

# 426 <u>5</u>4 Discussion

427 <u>54.1 Geochemical findings and Ee</u>lement interrelations

428 The spatial grain size distribution falls into an expected pattern with coarser sediments closer to the coastline (stations 129 MOT13A, MOT16, UN4) and finer sediments in the more remote and deeper stations MOT16A, UN5, UN6, UN6A, UN11. 130 The observed spatial variability of TOC with higher content in the predominantly muddy cores is expected since fine-grained 431 sediments are known to contain elevated levels of organic material compared to coarse-grained sediments (Salomons & 132 Forstner, 1984). The pronounced increase of TOC in the muddy sediments of station MOT16 (top 6cm) could be attributed to 133 the treated wastewater effluents of the refinery which are discharged in the general vicinity of this station (Paraskevopoulou; 34 2009). The two deep stations UN6 (200m at Megara basin) and UN11 (440m at Epidavros basin) present the highest TOC 135 content. It has been stipulated that suspended matter from eastern Saronikos basin, which is affected by more pronounced 436 polluting activities is transported below the thermocline to the west basin (Psyllidou-Giouranovits and Pavlidou, 1998; 437 Kontoyannis, 2010). This could explain the gradual increase of TOC content in the more recently deposited upper layers of 138 sediment. 139 Fine-grained sediments are known to contain abundant geochemical phases, such as clay minerals, organic material and Fe-140 Mn oxy-hydroxides with high affinity for trace metals due to increased surface adsorption and ionic attraction. Thus, the 41 elevated levels of Al and the so-called anthropogenic metals (Cu, Pb, Zn) in the muddy sediments of the study area can be 42 attributed to the predominant occurrence of aluminum-rich aluminosilicate and clay minerals in fine-grained sediments and

their concurrent efficacy to bind or sorb trace metals (Barjy et al., 2020; Karageorgis et al., 2005; Salomons & Forstner, 1984).
In contrast Cr and Ni which are also regarded as lithogenic elements along with Al, present a different distribution and are
elevated in the coarser near-shore sediments of cores MOT13A, MOT16 and MOT16A. This can be attributed to the existence
of ultrabasic rocks in the coastal Susaki area, in which, these metals are abundant (Kelepertsis et al., 2001). In the case of Fe
and Mn the lower contents at station UN4 can be attributed to the coarser sediments and the different geological setting with
absence of ultrabasic rocks.

The minimal depth variations observed at the vertical distributions of major (Al) and lithogenic elements (Fe, Cr, Ni) in the
 study area are anticipated because of their terrigenous origin (Nolting et al., 1999; Karageorgis et al., 2005;), however some
 of the discrepancies identified in specific cores cannot be readily explained due to lack of data for other elements such as Ti,
 Si and Ca.

453 The down-core variability of Mn in all stations, except UN4, is typical of shelf sediments, with high surficial Mn concentrations 154 that diminish with depth, as reducing conditions develop. These variations are largely independent of lithological or carbonate 155 content fluctuations, and are attributed solely on the respiration of organic carbon and the redox-cycle of Mn (Karageorgis et 156 al., 2005; Sundby, 2006). The Mn enrichment is far more pronounced in the surface sediments of station UN11 and subtle in 157 the other muddy cores. In station UN11 the bottom waters have been hypoxic or near anoxic since the mid 90's (Kontoyiannis, 158 2010; Paraskevopoulou et al., 2014). A possible explanation is that the slow diffusion of dissolved oxygen from the more 159 oxidizing overlying waters and the upward diffusion of dissolved Mn (II) from the pore water of anoxic surface sediments to 160 the sediment/water interface (Ozturk, 1995), cause the oxidation of dissolved Mn (II) and its precipitation as Mn (IV) oxides 461 (Pohl and Hennings, 1999). The Mn enrichment in the surface layers of UN11 resembles concentrations in sediments from 162 suboxic parts of the Black Sea (Kiratli & Ergin, 1996; Chen et al., 2022). 163 The normalized profiles of Cu, Pb and Zn exhibit statistically higher metal ratios in the upper layers of the cores, which is a

typical indication of the effect of modern pollution sources to recently deposited sediments in contrast to pre-industrial
 deposition of the deeper layers (Karageorgis. et al., 2005).

The concentrations of Al, Fe, Mn, Cu, Pb and Zn are increased from the northeast to the southwest area of West Saronikos
Gulf generally following the distribution of muddy sediments. The opposite increase of Cr and Ni to the north of the study
area is attributed to the ultrabasic geological substrate of the Susaki coastal area. The exception of increased Fe concentrations
in one of the northern sandy cores (MOT16) can also be attributed to the geology of the coastal region (Kelepertsis et al.,
2001).

471 Spearman's correlation analysis was carried out to determine the relationships between heavy metals and percentages of total 472 organic carbon (TOC) and carbonates in sediments of the collected cores. The concentrations of metals and the percentages of 173 organic and inorganic carbon used for the analysis correspond to the fine fraction in the sediments of stations MOT13A, 174 MOT16, UN4. Spearman's correlation coefficients are presented in Table A3 (appendix A) and Fig. A16 (appendix A). 175 Al is highly correlated (r > 0.5, p < 0.05) with Fe, Mn, Cu, Pb and Zn, which probably indicates an association between these 176 metals in the form of metal-clay complexes of continental origin (Barjy et al., 2020). On the other hand, there is a negative 177 correlation of Al with Cr and Ni. 178 Cr is highly correlated with Ni, but both of them show negative correlation with Cu, Pb and Zn, which can be attributed to 179 their different origin (Barjy et al., 2020) and poor correlation with Mn. Cr shows poor correlation with Fe, as well. The strong 180 correlation between Cr and Ni can be observed at sediments of the northwest part, too. 181 Fe, Mn, Cu, Pb and Zn show positive correlation with each other. Cu, Pb and Zn are highly correlated with each other (r > 0.5, 182 p < 0.05), which can be observed at sediments of the northwest part, too, suggesting that they have a common origin and 183 identical behavior during transport in the marine environment (Barjy et al., 2020). 184 The % TOC content presents moderate correlation with Al, Cu, Pb and Zn and negative correlation with Cr, Ni and % 85 carbonates. Moreover, it shows poor correlation with Fe and Mn. Finally, the percentage of carbonates content presents 186 negative correlation with all metals. 487 The high content of Al at the fine fraction of sediments indicates that Al is predominantly associated with aluminosilicate 188 minerals and occurs mostly in the clay minerals. Generally, fine sediments tend to have relatively high trace element 189 concentrations, due to the surface adsorption and ionic attraction. Especially, the so-called anthropogenic trace metals (Cu, Pb, 190 Zn) are normally bound within or sorbed by the clay mineral fraction of sediments (Barjy et al., 2020; Karageorgis et al., 191 2005). 192 Spearman's correlation analysis was carried out to determine the relationships between heavy metals and percentages of total 93 organic carbon (TOC) and carbonates in sediments of the collected cores. The concentrations of metals and the percentages of 194 organic and inorganic carbon refer to the fine fraction of core sediments MOT13A, MOT16, UN4. Spearman's correlation 195 coefficients are presented in Table A2 (appendix A) and Fig. A16 (appendix A). 196 Al is highly correlated (r > 0.5, p < 0.05) with Fe, Mn, Cu, Pb and Zn, which probably indicates an association between these 197 metals in the form of metal-clay complexes of continental origin (Barjy et al., 2020). On the other hand, there is a negative 198 correlation of Al with Cr and Ni-199 Cr is highly correlated with Ni, but both of them show negative correlation with Cu, Pb and Zn, which can be attributed to 500 their different origin (Bariv et al., 2020) and poor correlation with Mn. Cr shows bad correlation with Fe. too. The strong 501 correlation between Cr and Ni can be observed at sediments of the northwest part, too. 502 Fe, Mn, Cu, Pb and Zn show positive correlation with each other. Cu, Pb and Zn are high correlated with each other (r > 0.5, 503 p < 0.05), which can be observed at sediments of the northwest part, too, suggesting that they have a common origin and 504 identical behaviour during transport in the marine environment (Barjy et al., 2020). Zn is highly correlated with Cu and Pb 505 also at sediments of core UN11 at the south area. 506 The % TOC content presents moderate correlation with Al. Cu. Pb and Zn and negative correlation with Cr. Ni and the % 507 carbonates. Moreover, it shows poor correlation with Fe and Mn. Finally, the percentage of carbonates content presents 508 negative correlation with all metals. 509 54.2 Enrichment Factors 510 The Enrichment Factors (EF) are used to distinguish between metals originating from anthropogenic activities and from natural 511 processes, assessing the degree of anthropogenic effect. Equation (1) was used for the calculations of EFs, where  $C_x$  is the 512 concentration of the analyzed metal and CEN is the concentration of the normalizing element. Al was used as the reference

513 element.

514  $EF = (C_x/C_{EN})_{sample}/(C_x/C_{EN})_{background}$ 

515 (1)

516 In Table 5, the categories of contamination according to the Enrichment Factor are presented (Diamantopoulou et al., 2019;

517 Sutherland, 2000). In general, EFs use concentrations normalized to Al to account for the heterogeneity of the samples due to

518 differences in texture and organic content (Gredilla et al., 2015).

519

## 520 Table 5. The categories of <u>infection pollution</u> according to the Enrichment Factor.

EF	Contamination Degree
< 2	Depletion to minimal enrichment- no or minimal pollution
2 to 5	Moderate enrichment- moderate pollution
5 to 20	Significant enrichment- significant pollution
20 to 40	Very high enrichment- very strong pollution
>40	Extreme enrichment- extreme pollution

522Table 6 shows-presents the Enrichment Factors of the surface sediments (0-1 cm) of the study area that were calculated523according based on to the measured concentrations of heavy metals. The EFs were calculated at the fine fraction ( $f < 63\mu m$ )524of sediments of cores MOT13A, MOT16, UN4. Most metals present minimal to moderate enrichment in almost all the cores525analysed. Moderate enrichment is found for Cr, Ni, Mn and Pb in core MOT13A, Mn in UN11, and finally for Pb in UN5 and526MOT16.

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521

Table 6. Enrichment Factors of the surface sediments (0-1cm) of the study area. In cases of stations MOT13, MOT16, UN4, the EFs
 were calculated at the fine surface sediment fraction (f < 63μm).</li>

Core	EF Cr	EF Ni	EF Fe	EF Mn	EF Cu	EF Pb	EF Zn	Ī
MOT 13 A	3.09	2.07	1.92	2.07	1.94	2.24	2.14	Ì
MOT 16A	1.36	1.22	1.10	1.27	1.27	0.90	1.36	
UN 5	1.20	1.28	1.17	1.64	1.42	2.32	1.74	
MOT 16	1.28	1.14	1.05	1.33	1.61	3.71	1.60	
UN6	0.98	0.98	0.99	1.43	1.36	1.73	1.57	
UN 6A	1.00	0.99	1.01	1.23	1.15	1.09	1.39	
UN 4	1.00	0.87	0.79	0.84	1.34	1.86	1.11	
UN11	0.68	0.82	0.79	2.09	1.04	1.42	1.20	

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#### 532 <u>5</u>4.3 Sediment Quality Guidelines

Sediment Quality Guidelines (SQG) of effects range low (ERL) and effects range median (ERM) are used to assess the level of toxicity of metals in the surface sediments. Metal concentrations below the ERL value, indicate that effects on biota are rarely -observed. Concentrations above the ERL but below the ERM, occasionally affect the biota and concentrations above the ERM frequently affect the biota. The ERL and ERM guideline values for trace metals (ppmmfKg<sup>-1</sup>, dry wt) and percent incidence of biological effects in concentrations ranges defined by the two values are presented in Table 7 (Long et al., 1995).

Metal		ERM (mg kg <sup>-1</sup> )	Percent incidence of effects			
	ERL (mg kg -)		<erl< th=""><th>ERL-ERM</th><th>&gt;ERM</th></erl<>	ERL-ERM	>ERM	
Cr	81	370	2.9	21.1	95.0	
Cu	34	270	9.4	29.1	83.7	
Pb	46.7	218	8.0	35.8	90.2	
Ni	20.9	51.6	1.9	16.7	16.9	
Zn	150	410	6.1	47.0	69.8	

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In this study, the concentrations of heavy metals at the surface sediments were compared with the ERL and ERM criteria. In cases of For cores MOT13A, MOT16, UN4, the concentrations of in the total sediment fraction (f-< 1mm) were used. for the comparison with ERL and ERM criteria. The concentrations of Cr in the surface sediments of cores UN4, UN5, MOT16A,
UN6, UN6A and UN11 are higher than the ERL value (81 mg Kg<sup>-1</sup>) but below the ERM value (370 mg Kg<sup>-1</sup>) (Hahladakis et al., 2012) and the values at surface sediments of MOT13A and MOT16 are higher than the ERM value. The concentrations of
Ni at the surface sediments of the collected cores are higher than the ERM value (51.6 mg Kg<sup>-1</sup>) and as a result, they frequently
affect the biota (Long et al., 1995; Hahladakis et al., 2012).

The concentrations of Cu and Pb at surface sediments of stations MOT13A, MOT16A, UN4, UN5, MOT16 and UN6A, are below the ERL values (34 mg Kg<sup>-1</sup> for Cu and 46.7 mg Kg<sup>-1</sup> for Pb), which indicates that effects on biota are rarely observed. On the other hand, the concentrations at surface sediments of cores UN6 and UN11 are higher than the ERL values but below the ERM values (270 mg Kg<sup>-1</sup> for Cu and 218 mg Kg<sup>-1</sup> for Pb), which means that they can occasionally affect the biota (Hahladakis et al., 2012)\_\_\_T The concentrations of Zn are below the ERL (150 mg Kg<sup>-1</sup>) value and the ERM value (410 mg Kg<sup>-1</sup>), which indicates that effects on biota are rarely observed (Long et al., 1995; Hahladakis et al., 2012).

#### 556 <u>54.4 Mean effects range medium quotients</u>

The mean effects range medium quotient (mERMq) is an index that is used to evaluate the possible biological effects of the coupled toxicity of all heavy metals in the surface sediments (Gredilla et al., 2015). -Briefly, mERMq's were calculated by dividing the average concentration of each metal at the top 9cm, by its respective ERM (effects range median), to obtain the corresponding sediment quality guideline quotient (ERMq). Following this, mERMq's for each core were obtained as the average of ERMqs previously calculated. ERMqs indicates the pollutant concentration above which effects are expected to be frequent and have been only defined for very toxic elements (Gredilla et al., 2015).

In this study, Cr, Ni, Cu, Pb and Zn were considered in our calculations and the results are depicted in Table 8. In cases of cores MOT13A, MOT16, UN4, the concentrations of <u>the</u> total sediment fraction (f < 1mm) were used for the calculation of mERMq. Values of mERMq in the ranges of 0.0-0.1, 0.1-0.5, 0.5-1.5 and >1.5 correspond to the following probabilities of toxicity: 9 % (non-toxic), 21 % (slightly toxic), 49 % (moderately toxic) and 76 % (highly toxic), respectively (Gredilla et al., 2015). The mERMq values obtained for the sediments varied from 0.62 to 2.00, which means that the sediments are moderately or highly toxic.

The concentrations of Cr, Ni, Cu, Pb, Zn in sediments of cores MOT13A, UN5, UN6, UN6A, UN4 and UN11 are moderately
toxic and in sediments of MOT16A and MOT16 highly toxic. The concentrations of Cu, Pb, Zn in sediments of cores
MOT13A, MOT16A, MOT16, UN4 are non-toxic, with mERMq ranging frome 0.08 to -0.10 and those in sediments of UN5,
UN6, UN6A, UN11 are slightly toxic, with mERMq range 0.16-0.21.

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Table 8. MERMqs values calculated for the surface sediments (0-9 cm) of the collected cores of West Saronikos Gulf, by dividing
 the average concentration (mg Kg<sup>1</sup>) of each metal (Cr, Ni, Cu, Pb, Zn) by its respective ERM (mg Kg<sup>1</sup>). In cases of cores MOT13A,
 MOT16, UN4, the concentrations of total sediment fraction (f< 1mm) were used. for the calculation of mERMq.</li>

Core	mERMq (average)	toxicity of sediments
MOT13A	1.46	Moderately toxic
MOT16A	1.69	Highly toxic
UN5	1.46	Moderately toxic
MOT16	2.00	Highly toxic
UN6	1.21	Moderately toxic
UN6A	0.95	Moderately toxic
UN4	0.62	Moderately toxic
UN11	1.09	Moderately toxic

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#### 79 4<u>5</u>.5 Evolution of marine pollution

The total concentrations of eight heavy metals in the surface sediments were compared with those of a similar study ten years ago (Paraskevopoulou, 2009). In <u>cases of</u> cores MOT13A, MOT16, UN4, the concentrations of <u>the</u> total sediment fraction (<u>f</u> <-1mm) were used and the results are depicted in Fig.6. The levels of Cr, Ni <u>and</u>, Mn at most sediments, are decreased in 2017, compared to the study of 2007. On the other hand, the levels of Pb and Cu are increased in 2017, compared to the study of 2007. Moreover, the levels of Zn, at most sediments, are decreased in 2017 compared to the study of 2007.

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Figure 6: Levels of heavy metals in surface sediments of 2017 and 2007 and plotted with sediment quality guidelines. For the In cases of coarse sediments of stations MOT13A, MOT16, UN4, the concentrations of total sediment fraction (f< 1mm) were used.</li>

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# 599 <u>54.6</u> Comparison of metal concentrations in West Saronikos Gulf with other areas of Saronikos Gulf

The concentrations of heavy metals in the surface sediments of West Saronikos Gulf from the present study are compared with
 data from the other sub-areas of Saronikos. Specifically the data reviewed are: a) from surface sediments collected in Elefsis

602 Bay (EB), Inner Saronikos Gulf (ISG) and Outer Saronikos Gulf (OSG) during the same sampling of October 2017 and 503 analyzed in the Laboratory of Environmental Chemistry using the same methodologies (Xarlis; 2018; Vrettou, 2019) and b) 504 data from surface sediments in Elefsis Bay (EB), Inner Saronikos Gulf (ISG) and Outer Saronikos Gulf (OSG) from various 505 samplings conducted by the Hellenic Centre for Marine Research and analyzed by X-ray Fluorescence (Karageorgis et al., 606 2020; Karageorgis et al., 2020a). Summary results of this comparative review are provided in Table 9. The location of selected 507 stations from the other sub-areas of Saronikos are roughly given in Fig. A17 (Appendix). Furthermore, metal concentrations 508 in sediments from various areas of Greece analyzed by X-ray Fluorescence (Kanellopoulos et al., 2022) were also reviewed. 509 It is observed that high Al concentrations are recorded at West Saronikos as well as the Outer Gulf, related to the settling of 510 finer aluminosilicates in deeper waters. Moreover, high values of Al contents at Elefsis Bay (S1 and neighbouring stations, 611 Fig. A17) are attributed to the terrigenous inputs from ephemeral streams discharging into the bay (Karageorgis et al., 2020). 512 The concentrations of heavy metals in surface sediments of West Saronikos Gulf are compared with those measured at Elefsis 513 Bay (EB), Inner Saronikos Gulf (ISG) and Outer Saronikos Gulf (OSG) from the sampling of October 2017 (Panagopoulou, 514 2018; Vrettou, 2019; Xarlis, 2018). Figure 7 shows the location of sampling stations.

The station at Elefsis Bay with depth of 25 m locates near the Elefsis Port and station at Inner Saronikos Gulf with depth of 75 m near the Psittalia WWTP outfall. Moreover, there are two stations at Outer Saronikos Gulf. The first one, with depth of 85 m, locates near Vouliagmeni and the last one, with depth of 189 m, northwest of Sounio. The silt and clay fraction (f < 63µm) of surface sediments northwest of Sounio and near the Psittalia is higher than the sand fraction (63µm < f < 1mm). The sand fraction of the surface sediment near Vouliagmeni is higher than the silt and clay and similar to silt and clay of surface sediment of Elefsis Gulf. The total metal contents were extracted via complete dissolution of sediment samples with an acid mixture of HNO<sub>2</sub>-HE(ISO-14869-1:2000) (Peña Icart et al., 2011).

Cr and Ni contents show maxima in the northwestern stations (MOT13A, MOT16, MOT16A) offshore Susaki, due to the
 geological contribution from ophiolite complexes (Kelepertsis et al., 2001). Ni and Cr concentrations throughout the northwest
 area of Saronikos are higher than the mean background contents (117 mg Kg<sup>-1</sup> for Ni and 142 mg Kg<sup>-1</sup> for Cr) estimated from
 <u>sodiments sampled at various areas of Greece and similar to concentrations reported for Aliveri Bay and the Asopos river</u>
 basin, also related to occurrence of ultrabasic rocks (Kanellopoulos et al., 2022).

The concentrations of Mn and Fe are similar throughout the sub-areas of Saronikos Gulf, with the exception of Epidavros basin, where the maximum values are recorded due to the settling of finer aluminosilicates in deeper waters and possibly the implications of bottom water hypoxia/anoxia already discussed in previous sections (Kontoyannis, 2010; Karageorgis et al.,



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Figure 7: Map of Saronikos Gulf and the location of sediment sampling stations (© Google Earth 2019). The station at Elefsis Bay locates near the Elefsis Port and station at Inner Saronikos Gulf near the Psittalia WWTP outfall. Moreover, there are two stations at Outer Saronikos Gulf. The first one, locates near Vouliagmeni and the last one, northwest of Sounio. 637 The levels of Al of West Saronikos are comparable with those at the other areas and the low concentration of Al at station near 538 Vouliagment can be explained by the coarse surface sediment. The concentrations of Ni are decreased from the northwest to 539 the southeast part of Saronikos Gulf. High levels of Ni at the northwest area can be attributed to its geological origin 540 (Kelepertsis et al., 2001). The levels of Fe at West Saronikos are comparable with those at the other sediments. The low content 641 of Fe at station near Vouliagmeni may be attributed to its coarse sediment. The maximum values of AI, Fe and Mn at station 642 UN11 can be attributed to the fine surface sediment and the hypoxic conditions at waters of West Saronikos deeper than 200m. 643 Generally, the levels of Mn at West Saronikos are higher than those measured at Psittalia, Elefsis Gulf and station near 544 Vouliagmeni. The highest concentrations of Mn are observed at the deepest stations (UN6 and UN11 of West Saronikos and 545 station northwest of Sounio). 646 The levels of Cu, Pb, Zn at surface sediments of Psittalia and Elefsis Gulf are higher than those observed at the other sediments, 547 which can be explained by the numerous pollution sources in the marine environment and along the coast of Inner Saronikos 548

549 with those at Outer Saronikos Gulf but lower than those at the deep station UN11 and the concentrations of Pb at West 50 Saronikos are comparable with those at Outer Saronikos. The levels of Zn at the northwest part are higher than those observed 551 near Vouliagmeni, but lower than those at station UN11 and at the northwest area of Sounio. The results are depicted at Table 552 9. In cases of coarse surface sediments, the concentrations at total sediment fraction (f < 1 mm) were used.

Gulf and Elefsis Bay (Paraskevopoulou et al., 2014). Especially the concentrations of Cu at the northwest part are comparable

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554 555 Table 9. Comparison of heavy metals concentrations in mg Kg<sup>-1</sup> measured in this study with those in otherbetween areas of Saronikos Gulf.

Area / Stations	Al	Cr	Ni	Fe	Mn	Cu	Pb	Zn	References
West Saronikos									Present work
<u>MOT13A, MOT16,</u> <u>MOT16A</u>	<u>5697-</u> 27009	<u>280-552</u>	<u>344-484</u>	<u>20740-</u> 21191	<u>422-578</u>	<u>13.5-22.6</u>	20.0-30.3	<u>44.1-52.1</u>	
<u>UN5, UN6, UN6A</u>	<u>32705-</u> 43264	<u>142-199</u>	<u>187-305</u>	<u>21838-</u> 27301	<u>570-954</u>	<u>26.3-36.7</u>	<u>38.4-52.9</u>	73.8-92.1	
<u>UN4</u>	<u>22702</u>	<u>123</u>	<u>123</u>	<u>16682</u>	<u>270</u>	<u>17.5</u>	<u>24.5</u>	<u>43.6</u>	
<u>UN11</u>	54626	<u>142</u>	<u>230</u>	<u>32177</u>	<u>3925</u>	<u>49.7</u>	<u>63.9</u>	<u>110</u>	
Outer Saronikos	<u>14000-</u> <u>42000</u>	<u>81.5-133</u>	<u>29.5-106</u>	<u>6407-</u> 27139	<u>293-</u> 1159	<u>12.8-22.9</u>	<u>15.9-58.9</u>	<u>32.4-110</u>	Karageorgis et al., 2020a; Vrettou,2019
Inner Saronikos	<u>2900</u> <u>37700</u>	<u>30.7-184</u>	<u>9.6-87.1</u>	<u>3235-</u> 20795	<u>61.5-442</u>	7.9-68.5	<u>8.2-73.6</u>	<u>15.2-170</u>	<u>Karageorgis et al.,</u> 2020a; Vrettou, 2019; <u>Xarlis,2018</u>
Psitatalia (S7)	<u>26780</u>	÷	<u>83.2</u>	20623	<u>239</u>	<u>103</u>	<u>102</u>	<u>251</u>	Panagopoulou, 2018; Xarlis, 2018
Elefsis Bay	35000 61000	<u>108-176</u>	41.2-119	<u>29721-</u> <u>30499</u>	282-579	<u>32.1-137</u>	<u>61.0-171</u>	<u>188-521</u>	Karageorgis et al., 2020a; Xarlis,2018

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Station	Al	Ni	Fe	Mn	Cu	Pb	Zn	References
MOT13A	<del>5697</del>	<del>3</del> 44	<del>20740</del>	422	<del>13.5</del>	<del>20.0</del>	44 <del>.1</del>	Present work
MOT16A	<del>27009</del>	<del>375</del>	<del>23315</del>	<del>578</del>	<del>22.6</del>	<del>20.4</del>	4 <del>8.3</del>	Present work
UN5	<del>32705</del>	305	<del>25534</del>	635	<del>27.4</del>	42.7	74.5	Present work
MOT16	<del>16050</del>	484	<del>31191</del>	<del>503</del>	<del>16.9</del>	<del>30.3</del>	<del>52.1</del>	Present work
UN6	43264	253	<del>27301</del>	<del>954</del>	<del>36.7</del>	<del>52.9</del>	<del>92.1</del>	Present work
UN6A	<del>39314</del>	<del>187</del>	<del>21838</del>	<del>570</del>	<del>26.3</del>	<del>38.4</del>	<del>73.8</del>	Present work
UN4	<del>22702</del>	<del>123</del>	<del>16682</del>	270	<del>17.5</del>	24.5	4 <del>3.6</del>	Present work
<b>UN11</b>	<del>54626</del>	<del>230</del>	32177	3925	4 <del>9.7</del>	<del>63.9</del>	<del>110</del>	Present work
Deittelie	26790	02.2	20622	220	102	102	251	Panagopoulou, 2018;
PSittdild	20/00	00.2	20023	239	103	102	<del>231</del>	Xarlis, 2018
Elefsis Gulf	44853	<del>109</del>	<del>30499</del>	<del>394</del>	<del>132</del>	<del>141</del>	<del>368</del>	Panagopoulou, 2018; Xarlis, 2018
Vouliagmeni	4424	<del>13.4</del>	6407	243	<del>10.5</del>	<del>40.1</del>	27.4	Vrettou, 2019
<b>NWSounio</b>	44902	114	27139	<del>958</del>	30.7	<del>64.1</del>	<del>141</del>	Vrettou, 2019

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of Saronikos are reported lower. The levels at Epidavros basin (UN11) are the highest among the stations of West Saronikos,

661 Dassenakis et al., 2003). However, the West Saronikos levels of Cu, Pb and Zn remain lower than polluted stations of Inner 562 Saronikos, such as OS2 (near the port of Pireaus) and S7 (Psittalia, outfall of the Athens Waste Water Treatment Plant) 563 presented depicted in Gig. A17. Furthermore, Cu, Pb and Zn concentrations at Megara and Epidavros basins are similar to 564 those at surface sediments of Malliakos and Pagassitikos Gulf (Kanellopoulos et al., 2022) and the concentrations found in 665 most stations of Inner and Outer Saronikos (Karageorgis et al., 2020; Panagopoulou, 2018; Vrettou, 2019; Xarlis, 2018) but 566 higher than less affected island areas such as Chios Port, Milos and Andros (Kanellopoulos et al., 2022). In general the West 567 Saronikos sediments, affected by a relatively small industrial zone, are much less contaminated by Zn compared to the 568 concentrations well above 150 mg Kg -1 found in specific locations of Elefsis Bay, Inner Saronikos, Thessaloniki Bay, Ierissos 669 and Lavrio (Karageorgis et al., 2020, Kanellopoulos et al., 2022) with extensive pollution sources such as the Elefsis industrial 670 zone, Piraeus Port, the Athens Waste Water Treatment Plant outfall, major rivers of Northern Greece and current or historical 571 mining operations.

#### 674 <u>65</u> Conclusions

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The heavy metal pollution of West Saronikos Gulf has not been sufficiently studied, despite the scientific interest of this area, in contrast to the numerous studies of the eastern coast. The distribution of metals in the sediment samples of West Saronikos indicates that the area is enriched in metals from both geological and anthropogenic origins. The concentrations of all metals (Al, Mn, Cr, Ni, Cu, Pb, Zn) of fineim muddy sediments are higher than those measured in coarse-sandy sediments. The cores are fairly homogeneous, in terms of carbonates and the down-core variability of % TOC, is characterized by high surficial values that decrease with depth.

The Cr and Ni concentrations at the northwest part of the study area are higher than those measured at the southwest area and their values are very stable with depth of most sediment cores, which can be <u>attributed explained by the ophiolite background</u> (Kelepertsis et al., 2001)to the geological background of the adjacent coast. Al, Fe and Mn are increased from the northeast to the southwest part of the study area. The concentrations of Al and Fe <u>are increasedincrease</u> with depth <u>of in</u> most cores, while the values of Mn <del>are</del> decreased with depth. Generally, concentrations of Fe and Mn at surface sediments are affected by oxic and hypoxic conditions <u>and the settling of finer aluminosilicates</u>.

The horizontal vertical distributions of Cu, Pb and Zn present a constant decrease over depth along most cores, which can be attributed to their anthropogenic origin. Moreover, their levels at most sediments are higher than those measured ten years ago. Finally, the Cu, Pb, Zn concentrations in West Saronikos Gulf surface sediments are comparable with those at Outer Saronikos Gulf and lower than those from Inner Saronikos Gulf, and Elefsis Bay and other pollution hot spots of Greece, which can be attributed to the smaller industrial zone and sparce urbanized settlements of the west-West Saronikos coast, in comparison to the numerous anthropogenic activities at the east coast.

The concentrations of metals that are measured higher than the ERL values and the indication for moderately or highly toxic sediments by the calculation of mERMq signify that more research is required, in order to investigate probable effects on the marine ecosystem. Continuous monitoring, updating of the results of the present <u>study workcomplemented with detailed</u> geochemical analysis for major elements and minerals identification as well as core dating, metal speciation and study of bioaccumulation should be conducted, to assess the impacts of heavy metal pollution on the marine environment of West Saronikos Gulf.

599 <u>76</u> Appendices

# 700 Appendix A

# Table A1. Quality control data for total metal analyses. 702

Quality parameter	AI	<u>Cr</u>	<u>Cu</u>	Fe	Mn	Ni	<u>Pb</u>	Zn
Daily duplicates %RSD (range)	0.8-4.6	1.2-6.2	0.4-6.0	0.4-5.1	0.2-6.6	0.8-9.2	<u>1.1-8.2</u>	0.5-4.0
Daily duplicates %RSD (average)	<u>2.3</u>	<u>3.6</u>	<u>3.4</u>	<u>1.3</u>	<u>1.5</u>	<u>3.4</u>	<u>3.8</u>	<u>1.9</u>
%RSD for PACS reproducibility (n=15)	<u>6.2</u>	<u>8.3</u>	<u>3.2</u>	2.6	<u>7.4</u>	<u>9.7</u>	<u>6.2</u>	<u>4.0</u>
Accuracy (% Recovery range)	<u>85-106</u>	84-114	<u>87-96</u>	86-95	<u>85-113</u>	<u>85-115</u>	86-109	<u>86-99</u>
Accuracy (%Recovery average)	<u>96</u>	<u>96</u>	<u>91</u>	<u>89</u>	<u>99</u>	<u>101</u>	<u>95</u>	<u>91</u>

Table A24. The ratios of eight heavy metals to Al in surface <u>and sediments-deeper layer and ss</u>ediments of the depth of the collected cores. In cases of coarse-<u>grained</u> sediment cores MOT13A, MOT16, UN4, the ratios in fine sediment fraction ( $f < 63 \mu m$ ) are <u>measuredcalculated</u>.

Core	Layer (cm)	Cr 10 <sup>4</sup> Al <sup>-1</sup>	Ni 10 <sup>4</sup> Al <sup>-1</sup>	Fe Al <sup>-1</sup>	Mn 10 <sup>4</sup> Al <sup>-1</sup>	Cu 10 <sup>4</sup> Al <sup>-1</sup>	Pb 10 <sup>4</sup> Al <sup>-1</sup>	Zn 10 <sup>4</sup> Al <sup>-1</sup>
MOT 13 A	0-1	616	389	1.99	451	17.3	24.7	46.9
	10-12	277	261	1.44	303	12.4	15.4	30.4
MOT 16A	0-1	104	139	0.86	214	8.35	7.55	17.9
	30-32	85.0	127	0.87	187	7.35	9.35	14.7
UN 5	0-1	61.0	93.1	0.78	194	8.39	13.1	22.8
	30-32	58.9	84.5	0.77	137	6.85	6.52	15.2
MOT 16	0-1	170	174	1.19	244	10.4	12.9	24.8
	18-20	146	168	1.25	203	7.11	3.81	17.0
UN6	0-1	32.7	58.4	0.63	220	8.48	12.2	21.3
	24-26	34.2	61.6	0.66	159	6.41	7.29	14.0
UN 6A	0-1	37.0	47.5	0.56	145	6.7	9.76	18.8
	22-24	39.0	50.6	0.58	124	6.14	9.38	14.2
UN 4	0-1	38.6	47.4	0.46	104	6.9	7.63	15.2
	20-22	35.7	50.5	0.54	115	4.75	3.80	12.7
UN11	0-1	26.1	42.0	0.59	719	9.1	11.7	20.2
	30-32	33.8	45.1	0.66	303	7.75	7.29	14.8

UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
Al (Γιτο το τ	γραφήματος kg <sup>-1</sup> )	Al (mg kg <sup>-1</sup> )	Al (mg kg <sup>-1</sup> )	Al (mg kg-1)	Al (mg kg⁻¹)	Al (mg kg <sup>-1</sup> )	Al (mg kg <sup>-1</sup> )
1500020000	5000 10000	15.000 21.000	26000 31000	32000 37000	39000 44000	36000 40000 44000	45000 55000
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25		20	35	35	30	25	35
Min 16469	Min 5697	Min 16050	Min 26732	Min 32705	Min 39715	Min 38023	Min 45069
Max 23792	Max 11493	Max 23214	Max 30997	Max 38855	Max 44547	Max 42405	Max 58437
% RSD 10	% RSD 15	% pcp 12	0/ DOD 5	0/ DCD /	0/ non n	0/ 000 0	N/ DOD C
		26 6 517 1 5	% KND 5	26 8 51 7 4	I % RND R	1 % R\$D3	1 % 850.6
UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
UN4 Al (mg kg <sup>-1</sup> )	MOT13A Al (mg kg <sup>-1</sup> )	MOT16 Al (mg kg <sup>-1</sup> )	MOT16A Al (mg kg <sup>-1</sup> )	UN5 Al (mg kg <sup>-1</sup> )	UN6 Al (mg kg <sup>-1</sup> )	UN6A Al (mg kg <sup>-1</sup> )	UN11 Al (mg kg <sup>-1</sup> )
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000	MOT16A Al (mg kg <sup>-1</sup> ) 26000 31000	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000	UN6 Al (mg kg <sup>1</sup> ) 39000 44000	UN6A Al (mg kg <sup>1</sup> ) 36000 40000 44000	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0	MOT16A Al (mg kg <sup>1</sup> ) 26000 31000 0	UN5 Al (mg kg <sup>1</sup> ) 32000 37000 0	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0
UN4 Al (mg kg <sup>1</sup> ) 15000 25000 0	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 -	MOT16A Al (mg kg <sup>-1</sup> ) 26000 31000 0	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0
UN4 Al (mg kg <sup>1</sup> ) 15000 25000 0	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 2	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 - 4	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 5	UN5 Al (mg kg <sup>1</sup> ) 32000 37000 0 5	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 5	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0	UN11 Al (mg kg <sup>1</sup> ) 45000 55000 0 5
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5	MOT13A Al (mg kg <sup>1</sup> ) 5000 10000 0 2 2	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 - 4 -	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 5 10	UN5 Al (mg kg <sup>1</sup> ) 32000 37000 0 5 	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 -	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 2 4 4	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 - 4 6	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 10	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 5 10	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 10
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 -	MOT13A Al (mg kg <sup>1</sup> ) 5000 10000 0 2 4 4	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 10 15	UN5 Al (mg kg <sup>1</sup> ) 32000 37000 0 5 	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 -	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 5	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 10 15
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5	MOT13A Al (mg kg <sup>1</sup> ) 5000 10000 0 2 4 6 6	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 - 4 - 6 - 8 - 10	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 10 15	UN5 Al (mg kg <sup>1</sup> ) 32000 37000 0 5 - 10 - 15	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 10 15	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 -	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 10 15
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	MOT13A Al (mg kg <sup>1</sup> ) 5000 10000 0 2 - 4 - 6 -	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 - 4 - 6 - 8 - 10 - 12 -	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 - 10 - 15 - 20 -	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 - 10 - 15 - 20 -	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 10 15	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 5 -	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 5 5 5 5 5 5 7 5 7 5 7 5 7	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 2 - 4 - 6 - 8 -	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 - 10 - 15 - 20 -	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 10 15 20	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 5 -	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 6 15	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 2 - 4 - 6 - 8 - 10 -	MOTI6 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 - 10 - 15 - 20 - 25 -	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 5 - 10 -	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 10 15 20 25
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 6 10 15 15 20 20	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 2 - 4 - 6 - 8 - 10 -	MOT16 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 5 10 15 20 25 25	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 - 10 - 15 - 20 - 20 - 25	UN6A Al (mg kg <sup>-1</sup> ) 36000 4000 44000 0 5 5 - 10 - 15 - 20	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 10 15 20 25
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 5 6 7 7 8 8 15 15 20 20	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 2 - 4 - 6 - 8 - 10 - 12 - 12 - 12 - 10 - 10 - 10 - - - - - - - - - - - - -	MOTI6 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 10 15 20 25 20 25 30	UN5 Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 5 10 10 15 20 25 25 30	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 - 10 - 15 - 20 - 25 -	UN6A Al (mg kg <sup>-1</sup> ) 36000 4000 44000 0 5 - 10 - 15 - 20 -	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 6 10 5 5 7 7 8 8 15 15 10 10 10 10 10 10 10 10 10 10 10 10 10	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 - 2 - 4 - 6 - 8 - 10 - 12	MOTI6 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 	UNS Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 5 10 10 15 20 25 25 30 35	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 	UN6A Al (mg kg <sup>-1</sup> ) 36000 44000 0 5 5 10 10 15 20 20	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	MOT13A Al (mg kg <sup>-1</sup> ) 5000 10000 0 - 2 - 4 - 6 - 8 - 10 - 12 - Min 5697	MOTI6 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 	UNS Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 5 10 15 15 20 25 30 35 30 35 Min 32705	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 - 10 - 15 - 20 - 25 - 30 Min 39715	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 5 - 10 - 15 - 20 - 25 - Min 38023	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 
UN4 Al (mg kg <sup>-1</sup> ) 15000 25000 0 5 5 5 5 5 5 5 6 10 5 5 7 7 8 8 10 10 7 8 8 10 10 10 10 10 10 10 10 10 10 10 10 10	MOT13A Al (mg kg <sup>1</sup> ) 5000 10000 0 2 - 4 - 6 - 8 - 10 - 12 Min 5697 Max 11493	MOTI6 Al (mg kg <sup>-1</sup> ) 15.000 21.000 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 - Min 16050 Max 23214	MOTI6A Al (mg kg <sup>-1</sup> ) 26000 31000 0 5 5 10 15 20 25 30 35 Min 26732 Max 30997	UNS Al (mg kg <sup>-1</sup> ) 32000 37000 0 5 5 10 15 20 20 25 30 35 Min 32705 Max 38855	UN6 Al (mg kg <sup>-1</sup> ) 39000 44000 0 5 5 10 15 20 25 30 Min 39715 Max 44547	UN6A Al (mg kg <sup>-1</sup> ) 36000 40000 44000 0 5 5 - 10 - 25 - 25 - 25 - 25 - 25 - Min 38023 Max 42405	UN11 Al (mg kg <sup>-1</sup> ) 45000 55000 0 5 



713 Figure A1: Vertical distributions of Al in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4 714 refer to the total sediment fraction (f < 1 mm).

716 717 Figure A2: Vertical distributions of Cr in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4 718 refer to the total sediment fraction (f-< 1 mm).

30

35

Min 184

Max 236

% RSD 6

30

35

Min 247

Max 293

% RSD 4

20

25

Min 133

Max 153

% RSD 4

20

25

Min 141

Max 161

% RSD 3

30

35

Min 135

Max 163

% RSD 5

Περιοχή σχεδίασης

20

25

Min 89.1

Max 123

% RSD 9

16

18

20

Min 313

Max 434

% RSD 11

12

Min 357

Max 552

% RSD 13

	UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
	Cr Al-1	Cr Al-1	Cr Al <sup>-1</sup>	Cr Al-1	Cr Al <sup>-1</sup>	Cr Al-1	Cr Al-1	Cr Al-1
	10,0 60,0 0 5 5 10 10 15 20 20 25	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	80 100 120 0 5 - 10 - 20 - 20 - 25 - 30 - 25 - - - - - - - - - - - - -	30 50 70 0 5 - 10 - 15 - 20 - 25 - 30 - - - - - - - - - - - - -	30,0 40,0 0 5 5 - 10 - 15 - 20 - 25 -	30,0 40,0 0 5 5 - 10 - 20 - 20 -	20,0 40,0 0 5 - 10 - 15 - 20 - 25 - 30 - 25 -
'19	23 -	12 -	20 -	33 -	55 -	30 -	23 -	55 -
	UN4	MOT13A	MOT16	MOT16A	UNS	UN6	UN6A	UN11

Figure A3: Vertical distributions of Cr Al<sup>-1</sup> × ( $10^{4}10^{4}$ ) in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are calculated at the fine sediment fraction ( $f < 63 \mu m$ ).

UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
Ni (mg kg-1)	Ni (mg kg <sup>-1</sup> )	Ni (mg kg <sup>-1</sup> )	Ni (mg kg⁻¹)	Ni(mg kg <sup>-1</sup> )	Ni (mg kg <sup>-1</sup> )	Ni (mg kg <sup>-1</sup> )	Ni (mg Kg <sup>-1</sup> )
0	240 290 340	350 450 550 0 2 -	350 380 410 0 5 -	270 310 350	240 270 300		200 225 250
	4 6 8 10 12	4 - 6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 - 20 - 20 - 20 - 20 - 20 - 20 - 2		10 - 15 - 20 - 25 - 30 - 35 -		5 - 1 10 - 15 - 20 - 20 - 25 - 20 - 20	10 - 15 - 20 - 25 - 30 - 35 - 35 - 35 - 35 - 35 - 35 - 3
25 J Min 110	Min 247	20 - Min 394	35 J Min 356	35 - Min 305	25 J Min 249	25 J Min 182	35 J Min 209
Max 146	Max 344	Max 552	Max 411	Max 345	Max 292	Max 230	Max 233
Max 146	Max 344 MOT13A	Max 552 MOT16	Max 411 MOT16A	Max 345 UN5	Max 292 UN6	Max 230 UN6A	Max 233 UN11
Max 146 UN4 Ni (mg kg <sup>-1</sup> )	Max 344 MOT13A Ni (mg kg <sup>-1</sup> )	Max 552 MOT16 Ni (mg kg <sup>-1</sup> )	Max 411 MOT16A Ni (mg kg <sup>-1</sup> )	Max 345 UN5 Ni (mg kg <sup>-1</sup> )	Max 292 UN6 Ni (mg Kg <sup>-1</sup> )	Max 230 UN6A Ni (mg kg-1)	Max 233 UN11 Ni (mg Kg <sup>-1</sup> )
Max 146 UN4 Ni (mg kg <sup>-1</sup> ) 105 125 145 5 5 5 5 6 10 6 10 5 5	Max 344 MOT13A Ni (mg kg <sup>-1</sup> ) 240 290 340 0 	Max 552 MOT16 Ni (mg kg*1) 350 450 550 0 2 4 6 8 10	Max 411 MOT16A Ni (mg kg*) 350 380 410 0 5 10 15	Max 345 UN5 Ni (mg kg <sup>-1</sup> ) 270 310 350 0 5 - 10 - 15 - 15 - - - - - - - - - - - - -	Max 292 UN6 Ni (mg Kg <sup>-1</sup> ) 240 290 5	Max 230 UN6A Ni (mg kg <sup>-1</sup> ) 170 200 230 0 5 5 -	Max 233 UN11 Ni (mg Kg <sup>-1</sup> ) 200 225 250 0 5 5 10 15
Max 146	Max 344 MOT13A Ni (mg kg*) 240 290 340 0 4 4 6 8 10 12 12	Max 552 MOT16 Ni (mg kg <sup>-1</sup> ) 350 450 550 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 - - 20 - - - - - - - - - - - - -	Max 411 MOT16A Ni (mg kg *) 350 380 410 0 5 5 10 5 20 25 20 25 30	Max 345 UN5 Ni (mg kg <sup>-1</sup> ) 270 310 350 0 5 - 10 - 15 - 20 - 25 - 30 - 25	Max 292 UN6 Ni (mg Kg <sup>-1</sup> ) 240 290 0 5 - 10 - 15 - 20 - 25 - - - - - - - - - - - - -	Max 230	Max 233 UN11 Ni (mg Kg*) 200 225 250 0 
Max 146 UN4 Ni (mg kg <sup>-1</sup> ) 105 125 145 5 	Max 344 MOT13A Ni (mg kg <sup>-1</sup> ) 240 290 340 0 4 - - - - - - - - - - - - -	Max 552 MOT16 Ni (mg kg <sup>-1</sup> ) 350 450 550 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 - Min (mg kg <sup>-1</sup> ) 350 450 550 0 - - - - - - - - - - - - - - - - -	Max 411 MOT16A Ni (mg kg*) 350 380 410 0 5 5 10 15 20 25 30 35 35 5 5 0 10 15 20 15 15 20 15 15 20 15 15 20 15 15 15 15 15 15 15 15 15 15 15 15 15	Max 345 UN5 Ni (mg kg <sup>-1</sup> ) 270 310 350 0 5 - 10 - 20 - 25 - 30 - 35 - - - - - - - - - - - - -	Max 292 UN6 Ni (mg Kg <sup>-1</sup> ) 240 290 0 5 - - - - - - - - - - - - -	Max 230 UN6A Ni (mg kg <sup>-1</sup> ) 170 200 230 0 5 - 10 - 15 - 20 - 25 - Min 182	Max 233 UN11 Ni (mg Kg*) 200 225 250 0 5 - 10 - - - - - - - - - - - - -

725 726 Figure A4: Vertical distributions of Ni in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4 refer to the total sediment fraction (f< 1 mm).

	UN4	MOTI2A	MOT16	MOT16A	UN5	UN6	UN6A	10111
	UN4 Ni Al-1 40,0 60,0 80,0 0 5 5 10 10 10 15 15 10 10 15 15 10	MOT13A NI AF1 100 500 0 2 - 1 4 - 1 6 - 1 8 - 1 10 - 1	MOT16 Ni Al <sup>11</sup> 100 300 2 4 6 - 10 - 12 - 14 - 16 -	MOTIGA NI Al <sup>11</sup> 100 150 0 5 - 10 - 15 - 20 - 25 - 30 -	UNS NI Al-1 70 100 0 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	UN6 Ni AF1 40,0 80,0 5 - 10 - 15 - 20 - 25 -	UN6A Ni Al <sup>11</sup> 40,0 70,0 0 5 5 - 10 - 15 - 20 -	UN11 Ni Al <sup>-1</sup> 30,0 60,0 0 5 - 10 - 15 - 20 - 25 - 20 - 25 - 30 -
	25	12	$\begin{bmatrix} 18\\20 \end{bmatrix}$	35	35 _	30 」	25	35
/27	UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
	Ni Al <sup>-1</sup> 40,0 60,0 80,0 0 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	Ni Al <sup>-1</sup> 100 2 - 4 - 6 - 8 - 10 - 12 -	Ν Περιοχή γραα 100 2 4 6 - 4 - 10 - 10 - 10 - 10 - - - - - - - - - - - - -	Ni Al <sup>-1</sup> 100 150 5 - 10 - 15 - 20 - 25 - 30 - 35 -	Ni Al <sup>-1</sup> 70 100 0 5 5 - 10 - 15 - 20 - 25 - 30 - 35	Ni Al <sup>-1</sup> 40 80 0 5 5 10 - 15 - 20 - 25 - 30	Ni Al <sup>-1</sup> 40 70 0 5 5 10 10 15 20 25	Ni Al <sup>-1</sup> 30 60 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - - - - - - - - - - - - -

729 730 Figure A5: Vertical distributions of Ni Al<sup>-1</sup>  $\leq$  (10<sup>4</sup>) in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are calculated at the fine sediment fraction ( $\pounds$  < 63 µm).



Figure A6: Vertical distributions of Fe in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4
 refer to the total sediment fraction (f < 1 mm).</li>

	UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
	Fe Al <sup>-1</sup> 0,20 0,70 0 5 5 5 10 10 15 15 20 25	Fe Al <sup>13</sup> 1,0 3,0 0	Fe Al <sup>4</sup> 0,5 1,5 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 - 14 - 18 - 10 -	Fe Al <sup>14</sup> 0,70 1,00 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - - - - - - - - - - - - -	Fe Al <sup>-1</sup> 0,60 1,00 5 - 4 10 - 4 15 - 4 20 - 4 30 - 4 35 - 4 10 - 4 1	Fe Al <sup>14</sup> 0,50 1,00 0 5 5 - - - - - - - - - - - - -	Fe Al <sup>14</sup> 0,20 1,00 0 1,00 5 - 1,00 10 - 1,00 1,00 1,00 1,00 1,00 1,00 1,00 1,00	Fe Al <sup>4</sup> 0.40 1,00 0 $+$ 5 $+$ 10 $+$ 15 $+$ 20 $+$ 30 $+$ 35 $+$ 30 $+$ 35 $+$ 30 $+$ 35 $+$ 30 $+$ 35 $+$ 36 $+$ 37
35	UNA	MOTION	MOTIS	MOTICA	UNS	UNG	UN6A	
	Fe Al <sup>-1</sup> 0.20 0.70 0 5 5 5 10 10 5 5	Fe Al <sup>-1</sup> 1.0 3.0 0 2 - 4 - 6 - 8 -	Fe Al <sup>-1</sup> 0.5 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - 14	Fe Al <sup>-1</sup> 0.70 1.00 0 5 5 10 15 20 -	Fe Al <sup>-1</sup> 0.60 1.00 5 - 10 - 15 - 20 - 25 -	Fe Al <sup>-1</sup> 0.50 1.00 0 5 - 10 - 15 - 20 -	Fe Al <sup>-1</sup> 0.20 1.00 0	UN11 Fe Al <sup>1</sup> 0,40 1.00 0 5 - 10 - 15 - 20 - 20 - 25 -

736 737 Figure A7: Vertical distributions of Fe Al<sup>-1</sup> in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are calculated at the fine sediment fraction ( $f < 63 \ \mu m$ ).



UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
Mn (mg kg <sup>-1</sup> )	Mn (mg kg <sup>.1</sup> )						
0						450 510 570	
5 -	4	6 -	10 -	10 -	5 -	5 -	10 -
	6 -	8 -	15	15 -	10 -	10 -	
15 -	8 -	12 -	25 -	25	15 -	15 -	25 -
20 -	10 - +	16 -	30 -	30 -	20 -	20 -	30 -
25		20 ]	35 -	35 _	25	25 ]	35 _
Min 251	Min 361	Min 456	Min 526	Min 448	Min 669	Min 476	Min 1458
Max 290	Max 422	Max 503	Max 578	Max 635	Max 954	Max 570	Max 3925
% RSD 5	% RSD 5	% RSD 3	% RSD 3	% RSD 8	% RSD 11	% RSD 4	% RSD 35

740 741 742 Figure A8: Vertical distributions of Mn in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4 refer to the total sediment fraction (f-< 1 mm).

	UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
	Mn Al⁻¹	Mn Al-1	Mn Al <sup>-1</sup>	Mn Al <sup>-1</sup>	Mn Al <sup>-1</sup>	Mn Al <sup>-1</sup>	Mn Al⁻¹	Mn Al <sup>-1</sup>
	100 130 160 0 5 5 10 6 15 20 25	200 500 0 4 4 6 7 10 12	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	150 200 250 0 5 10 15 20 20 25 30 30 35	100 150 200 0 5 10 10 20 20 25 30 35	100 200 300 0 5 5 10 10 20 20 20 30	100 125 150 0 5 5 - 10 - 15 - 20 - 25	200 500 800 0 5 - 10 - 15 - 20 - 25 - 30 - 35 -
13	UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
	Mn Al <sup>4</sup> 100 130 160 0 5 - 10 - 10 - 20 - 25 - 25	Mn Al <sup>4</sup> 200 500 0 - 4 - 6 - 8 - 10 - 12 - -	$\begin{array}{c} \text{Mn Al}^{14} \\ 100 & 200 & 300 \\ 0 & & & \\ 2 & - & & \\ 4 & - & & \\ 4 & - & & \\ 6 & - & & \\ 6 & - & & \\ 10 & - & & \\ 12 & - & & \\ 12 & - & & \\ 12 & - & & \\ 14 & - & & \\ 16 & - & & \\ 18 & - & & \\ 20 & - & & \\ \end{array}$	Mn Al <sup>1</sup> 150 200 250 0 5 - 10 - 15 - 20 - 25 - 30 - 35 -	Mn Al <sup>-1</sup> 100 150 200 0 5 - 5 - 10 - 15 - 10 -	Mn Al <sup>-4</sup>	Mn Al <sup>-1</sup> 100 125 150 0 - 5 - 10 - 10 - 15 - 20 - 25 - - - - - - - - - - - - -	Mn AI <sup>4</sup> 200 500 800 5

Figure A9: Vertical distributions of Mn Al<sup>-1</sup>× (10<sup>4</sup>) in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are calculated at the fine sediment fraction ( $f < 63 \mu m$ ).

UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
Cu (mg kg <sup>-1</sup> ) 10 15 20 0	Cu (mg kg <sup>-1</sup> ) 10 13 16 0 2 -	Cu (mg kg <sup>-1</sup> ) 10 14 18 0	Cu (mg kg <sup>1</sup> ) 20 23 26 0 5 -	Cu (mg kg <sup>-1</sup> ) 20 30 40 0 5	Cu (mg kg <sup>-1</sup> ) 20 30 40 0	Cu (mg kg <sup>-1</sup> ) 20 25 30 0	Cu (mg kg <sup>-1</sup> ) 30 40 50 0 5 -
	6 -	6 - 8 - 10 - 12 -	10 -	10 -	10 -	5 - 10 - 15 -	10 -
20 -		14 - 16 - 18 -	25 - 30 - 35	25 -	20 -	20 -	25 -
Min 11.5	Min 11.1	Min 11.7	Min 20.7	Min 24.3	Min 26.5	Min 24.2	Min 35.8
Max 18.3	Max 13.6	Max 17.0	Max 23.2	Max 31.0	Max 36.7	Max 28.6	Max 49.7
% RSD 14	% RSD 7	% RSD 10	% RSD 2	% RSD 6	% RSD 9	% RSD 5	% RSD 8

UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
Cu (mg kg <sup>-1</sup> ) 10,0 15,0 20,0	Cu (mg kg <sup>.1</sup> ) 10,0 13,0 16,0	Cu (mg kg <sup>-1</sup> ) 10,0 14,0 18,0	Cu (mg kg <sup>.1</sup> ) 20,0 23,0 26,0	Cu (mg kg <sup>-1</sup> ) 20,0 30,0 40,0	Cu (mg Kg <sup>-1</sup> ) 20 30 40	Cu (mg kg <sup>-1</sup> ) 20,0 25,0 30,0	Cu (mg kg <sup>-1</sup> ) 30,0 40,0 50,0
0 5 5 10 10 10 15 20 20	0	0 2 4 4 6 - 8 - 10 - 12 - 14 - 18 - - - - - - - - - - - - -	0 5 - 10 - 15 - 20 - 25 - 30 -	$ \begin{array}{c} 0 \\ 5 \\ 5 \\ 10 \\ 20 \\ - \\ 30 \\ - \\ 30 \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ - \\ -$	0 5 - 10 - 15 - 20 - 25 -	0	0 5 5 - 10 - 20 - 25 - 30 -
25 ]		20 ]	35	35	30 ]	25 」	35 ]
Min 11.5	Min 11.1	Min 11.7	Min 20.7	Min 24.3	Min 26.5	Min 24.2	Min 35.8
Max 18.3 % RSD 14	Max 13.6 % RSD 7	Max 17.0 % RSD 10	Max 23.2 % RSD 2	Max 31.0 % RSD 6	Max 36.7 % RSD 9	Max 28.6 % RSD 5	Max 49.7 % RSD 8

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Figure A10.: Vertical distributions of Cu in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4 refer to the total sediment fraction (f< 1 mm).



Figure A11: Vertical distributions of Cu Al<sup>-1</sup> × (10<sup>4</sup>) in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are calculated at the fine sediment fraction (f-< 63  $\mu m).$ 

UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN1
Pb (mg kg <sup>-1</sup> )	Pb (mg kg <sup>-1</sup> )	Pb (mg kg <sup>-1</sup> )	Pb (mg kg <sup>-1</sup> )	Pb (mg kg <sup>-1</sup> )	Pb (mg kg <sup>-1</sup> )	Pb (mg kg <sup>-1</sup> )	Pb (m
				20 40 60 0			20 50
10 - E t t t t t t t t t t t t t t t t t t	4 - 6 - 8 - 10 -	6 - 8 - 10 - 12 - 14 - 16 - 18 -	10 - 15 - 20 - 25 - 30 -	10 - 15 - 20 - 25 - 30 -	10 -	10	10 - 15 - 20 - 25 - 30 -
25	12 _	20 _	35 -	35	25 -	25	35 -
Min 9.8	Min 11.5	Min 9.1	Min 17.5	Min 20.6	Min 24.0	Min 29.8	Min 31.9
Max 28.4	Max 23.3	Max 30.3	Max 28.1	Max 43.8	Max 54.0	Max 47.3	Max 63.9
	1						-
UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN
UN4 Pb (mg kg <sup>-1</sup> )	MOT13A Pb (mg kg <sup>-1</sup> )	MOT16 Pb (mg kg <sup>-1</sup> )	MOT16A Pb (mg kg <sup>-1</sup> )	UN5 Pb (mg kg <sup>-1</sup> )	UN6 Pb (mg Kg <sup>-1</sup> )	UN6A Pb (mg kg <sup>-1</sup> )	UN Pb (m
UN4 Pb (mg kg*1) 8,0 18,0 28,0 0 5 - 5	MOT13A Pb (mg kg 1) 10,0 18,0 26,0 0 2 4 4	MOT16 Pb (mg kg <sup>1</sup> ) 8,0 23,0 38,0 0 2 - 4 - 6 -	MOT16A Pb (mg kg <sup>3</sup> ) 12,0 22,0 32,0 0 5 - 10 -	UN5 Pb (mg kg <sup>-1</sup> ) 20,0 40,0 60,0 0 5 - 10 - - - - - - - - - - - - -	UN6 Pb (mg Kg <sup>-1</sup> ) 20 40 60 0	UN6A Pb (mg kg <sup>1</sup> ) 20,0 35,0 50,0 0 	UN Pb (m 20 5 0 5 - 10 -
UN4 Pb (mg kg*1) 8,0 18,0 28,0 0 5 - 10 15 - 10 15 - 10 15 - 10 15 - 10 15 - - 15 - - - - - - - - - - - - -	MOT13A Pb (mg kg <sup>-1</sup> ) 10,0 18,0 26,0 0 	MOT16 Pb (mg kg <sup>-1</sup> ) 8,0 23,0 38,0 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - - 14 - - - - - - - - - - - - -	MOT16A Pb (mg kg <sup>-1</sup> ) 12,0 22,0 32,0 0 5 - 10 - 15 - 20 - 25 - - - - - - - - - - - - -	UN5 Pb (mg kg*) 20,0 40,0 60,0 0 5 - 10 - 15 - 20 - - - - - - - - - - - - -	UN6 Pb (mg Kg <sup>-1</sup> ) 20 40 60 0 5 - 10 15 - 20 - 20 - 10 20 - 10 - 20 - 10 - 20 - 10 - 20 - 10 - 1	UN6A Pb (mg kg <sup>-1</sup> ) 20,0 35,0 50,0 0 	UN Pb (m 20 5m 5 - 1 10 - 1 15 - 1 20 -
UN4 Pb (mg kg *1) 8,0 18,0 28,0 0 5 5 10 5 20 25 20 25 20	MOT13A Pb (mg kg <sup>-1</sup> ) 10,0 18,0 26,0 0 	MOT16 Pb (mg kg <sup>1</sup> ) 8,0 23,0 38,0 0 2 - 4 - 6 - 8 - 10 - 11 - 16 - 18 - 20 - 20 - 18 - 20 - 20 - 18 - 20 - 18 - 20 - 10 -	MOT16A Pb (mg kg <sup>-1</sup> ) 12,0 22,0 32,0 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - - - - - - - - - - - - -	UN5 Pb (mg kg*) 20,0 40,0 60,0 0 5 - 10 - 20 - - - - - - - - - - - - -	UN6 Pb (mg Kg <sup>-1</sup> ) 20 40 60 0 5 5 10 15 20 - 25 - 30	UN6A Pb (mg kg <sup>-1</sup> ) 20,0 35,0 50,0 0 	UN Pb (m 20 5 0 5 - 10 - 15 - 20 - 20 - 10 - 20 - 10 - 15 - 20 - 10 - 15 - 20 - 10 - 15 - 10 -
UN4 Pb (mg kg *1) 8,0 18,0 28,0 0 5 5 10 5 20 25 20 25 Min 9.8	MOT13A Pb (mg kg <sup>-1</sup> ) 10,0 18,0 26,0 0 	MOT16 Pb (mg kg <sup>1</sup> ) 8,0 23,0 38,0 0 2 - 4 - 6 - 8 - 10 - 14 - 16 - 18 - 20 - Min 9.1	MOT16A Pb (mg kg <sup>-1</sup> ) 12,0 22,0 32,0 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - Win 17.5	UN5 Pb (mg kg*) 20,0 40,0 60,0 0 5 - 10 - 15 - 20 - - - - - - - - - - - - -	UN6 Pb (mg Kg <sup>-1</sup> ) 20 40 60 0 5 - 10 - 15 - 20 - 25 - 30 Min 24.0	UN6A Pb (mg kg <sup>-1</sup> ) 20,0 35,0 50,0 0 	UN Pb (m 20 5 5 - 10 - 15 - 20 - 30 - 35 - Min 31.9
UN4 Pb (mg kg*1) 8,0 18,0 28,0 0 5 - 10 - 5 - 20 - 25 - - - - - - - - - - - - -	MOT13A Pb (mg kg <sup>-1</sup> ) 10,0 18,0 26,0 0 	MOT16 Pb (mg kg <sup>1</sup> ) 8,0 23,0 38,0 0 2 - 4 - 6 - 8 - 10 - 14 - 16 - 18 - 20 - Min 9.1 Max 30.3	MOT16A Pb (mg kg <sup>-1</sup> ) 12,0 22,0 32,0 0 5 - 10 - 15 - 20 - 25 - 30 - 35 Min 17.5 Max 28.1	UN5 Pb (mg kg*) 20,0 40,0 60,0 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - Min 20.6 Max 43.8	UN6 Pb (mg Kg <sup>-1</sup> ) 20 40 60 0 5 5 10 15 20 - 25 - 30 Min 24.0 Max 54.0	UN6A Pb (mg kg <sup>-1</sup> ) 20,0 35,0 50,0 0 	UN Pb (m 20 50 0 - 10 - 15 - 20 - 25 - 30 - 35 - Min 31.9 Max 63.9

Figure A12: Vertical distributions of Pb in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4 refer to the total sediment fraction (f-< 1 mm).



Figure A13-: Vertical distributions of Pb Al<sup>-1</sup>×(10<sup>4</sup>) in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are calculated at the fine sediment fraction ( $f < 63 \mu m$ ).

	UN4	MOT13A	MOT16	MOT16A	UN5	UN6	UN6A	UN11
	Zn (mg kg <sup>4</sup> ) 30.0 45.0 60.0	Zn (mg kg <sup>-1</sup> ) 30.0 42.0 54.0 0 2 -	Zn (mg kg <sup>-1</sup> ) 30.0 45.0 60.0 0 2	Zn (mg kg <sup>-1</sup> ) 40.0 45.0 50.0 0 5 -	Zn (mg kg <sup>-1</sup> ) 40 70 100 0	Zn (mg kg <sup>3</sup> ) 50 75 100 0 1	Zn (mg kg <sup>-1</sup> ) 50 75 100 0	Zn (mg kg <sup>-1</sup> ) 60 85 110 0
	10 - E t a 15 - 20 - 25 - Min 33,6 Max 64 7	4 - 6 - 8 - 10 - 12 - 12 - 12 - 12 - 12 - 12 - 12	6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 - 14 - 16 - 18 - 14 - 14 - 16 - 18 - 14 - 16 - 16 - 18 - 16 - 16 - 16 - 16 - 16	10 - 15 - 20 - 25 - 30 - 35 - Min 42,1 May 48 3	10 - 15 - 20 - 25 - 30 - 35 - Min 56.4 May 77 9	10 - 1 15 - 1 20 - 1 20 - 1 Min 58.5 May 021	10 - 15 - 20 - 25 - Min 54.4 May 75 8	10 - 15 - 20 - 25 - 30 - 35 - Min 70.8 May 110
3						·····		Wiax 110
,	UN4 Zn (mg kg*) 30,0 45,0 60,0 0 5 5 10 20 25 20 25 Min 33,6 Max 54,7 % RSD 17	MOT13A Zn (mg kg *) 30,0 42,0 54,0 0 - 2 - 4 - 6 - 8 - 10 - 12 Min 32,8 Max 47,0 % RSD 12	MOT16 Zn (mg kg <sup>4</sup> ) 30,0 45,0 60,0 0 2 - 4 - 6 - 8 - 10 - 12 - 14 - 16 - 18 - 20 Min 35,9 Max 52,1 % R5D 14	MOT16A Zn (mg kg*) 40,0 45,0 50,0 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - Min 42,1 Max 48,3 % RSD 4	UN5 Zn (mg kg <sup>-1</sup> ) 40 70 100 0 5 - 10 - 15 - 20 - 25 - 30 - 35 - Min 56,4 Max 77,9 % RSD 11	UN6 Zn (mg Kg <sup>-1</sup> ) 50 100 0 	UN6A Zn (mg kg <sup>-1</sup> ) 50 75 100 0 5 - 5 - 10 - 15 - 20 - 15 - 20 - 15 - 20 - 16 - 20 - 16 - 20 - 25 - 20 - 25 - 20 - 25 - 20 - 25 - 20 - 25 - 25 - 25 - 20 - 25	UN11 Zn (mg kg <sup>-1</sup> ) 60 85 110 0 

Figure A14: Vertical distributions of Zn in mg kg <sup>-1</sup> in sediment cores. The concentrations in coarse cores MOT13A, MOT16, UN4
 refer to the total sediment fraction (f< 1 mm).</li>



Figure A15: Vertical distributions of Zn Al<sup>-1</sup>×(10<sup>4</sup>) in sediment cores. The ratios in coarse cores MOT13A, MOT16, UN4 are
 calculated at the fine sediment fraction (f< 63 μm).</li>

**Correlations** TOC <u>%</u> Spearman's Al Cr Ni Fe Mn Cu Pb Zn CaCO<sub>3</sub> rho 1.000 Al -0.521\*\* Cr 1.000 -0.453\*\* 0.841\*\* Ni 1.000 0.624\*\* Fe 0.081  $0.179^{*}$ 1.000 0.735\*\* Mn -0.108 0.029 0.694\*\* 1.000 Cu 0.924\*\* -0.419\*\* -0.342\*\* 0.633\*\* 0.746\*\* 1.000 Pb 0.676\*\* -0.440\*\* -0.433\*\* 0.244\*\* 0.397\*\* 0.779\*\* 1.000  $0.790^{**}$ -0.452\*\* 0.479\*\* 0.882\*\* -0.441\*\*  $0.459^{**}$  $0.894^{**}$ 1.000 Zn 0.244\*\* TOC -0.198\* -0.272\*\* 0.096 <u>0.155</u> 0.351\*\* 0.428\*\* 0.483\*\* 1.000 -0.222\*\* -0.472\*\* -0.597\* % CaCO3 -0.215\* -0.766 -0.566\* -0.358\* -0.516\*\* -0.282\*\* 1.000 \*\*. Correlation is significant at the 0.01 level (2-tailed). \*. Correlation is significant at the 0.05 level (2-tailed). 774 <del>Table A2. Spearman's correlation coefficient matrix for A1 (mg Kg<sup>-t</sup>), Cr (mg Kg<sup>-t</sup>), Ni (mg Kg<sup>-t</sup>), Fe (mg Kg<sup>-t</sup>), Mn (mg Kg<sup>-t</sup>), Cu (mg Kg<sup>-t</sup>), Pb (mg Kg-1), Zn (mg Kg<sup>-t</sup>), TOC (% Total Organic Carbon), carbonates (% CaCO<sub>3</sub>) (N=140 sediment samples),</del> **Correlations** Cr Fe Ph Zn TOC 0/\_ Spearman's ANi Mn Cu rho CaCO<sub>2</sub> Al 1.000 Cr 0.521\*\* 1.000 -0.453\*\* -0.841\*\* 1.000Ni 0.624\*\* 0.081 0.179\* 1.000Fe <del>0.735\*\*\*</del> <del>0.694\*\*</del> Mn -0.108 0.029 1.000<del>0.924\*\*</del> <del>-0.419\*\*</del> <del>-0.342\*\*\*</del> <del>0.633\*\*</del> <del>Cu</del> 0.746\*\* 1.000 ₽₽ 0.676<sup>\*\*</sup> <del>-0.440<sup>\*\*\*</sup></del> <del>-0.433\*\*\*</del> 0.244<sup>\*\*</sup> 0.397\*\* 0.779<sup>\*\*</sup> 1.000

Μορφοποίησε: Αγγλικά (Ηνωμένων Πολιτειών)

#### Table A3. Spearman's correlation coefficient matrix for Al (mg Kg<sup>-1</sup>), Cr (mg Kg<sup>-1</sup>), Ni (mg Kg<sup>-1</sup>), Fe (mg Kg<sup>-1</sup>), Mn (mg Kg<sup>-1</sup>), Cu (mg Kg<sup>-1</sup>), Pb (mg Kg-1), Zn (mg Kg<sup>-1</sup>), TOC (% Total Organic Carbon), carbonates (% CaCO<sub>3</sub>) (N=140 sediment samples).

771 772

773

775 776

777

Zn

TOC

% CaCO

0.790<sup>\*\*</sup>

0.244\*\*

<u>-0.472\*\*</u>

0.452\*\*

-0.198<sup>\*</sup>

-0.215\*

\*\*. Correlation is significant at the 0.01 level (2 tailed). \*. Correlation is significant at the 0.05 level (2-tailed).

-0.441\*\*\*

0.272\*\*

-0.222<sup>\*\*</sup>

0.459\*\*

<u>-0.766\*\*</u>

-0.096

0.479\*\*

0.155

<del>-0.597\*\*</del>

0.882\*\*\*

0.351\*\*

<del>-0.566\*\*</del>

0.894\*\*

0.428\*\*

<del>-0.358\*\*</del>

1.000

0.483\*\*

-0.<u>516\*\*</u>

1.000

<del>-0.282\*\*\*</del>

 $\frac{1.000}{1.000}$ 



Figure A16: Correlations of heavy metals for the core samples of West Saronikos Gulf.





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Figure A17: Map of Saronikos with stations in Elefsis Bay (red circles), Inner Saronikos (yellow circles) and Outer Saronikos (green
 circles) from which metal concentrations in sediments were review for the comparative discussion to the present study results in
 West Saronikos (black circles).

808 7 Data availability

809 Datasets and their sources are fully detailed in the manuscript.

811 8 Team list

812 Not applicable.

813

810

## 814 9 Author contribution

Georgia Filippi and Vasiliki Paraskevopoulou conducted the chemical analyses in the Laboratory of Environmental Chemistry
of the National and Kapodistrian University of Athens. Georgia Filippi wrote the paper, with contributions and reviews from
all co-authors. Konstantinos Lazogiannis constructed the map of the study area and contributed to compilation and
improvements of figures. Manos Dassenakis was the supervisor of the laboratory work and this article.

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#### 820 10 Competing interests

821 The authors declare that they have no conflict of interest.

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823 11 Special issue statement

824 The statement on a corresponding special issue will be included by Copernicus, if applicable.

#### 826 12 Acknowledgements

We are grateful to the Hellenic Center for Marine Research (HCMR) and our colleagues Prof. S. Poulos, Dr Aik. Karditsa,and Dr F. Botsou for the assistance during the sampling. The research -was funded by the European Union (European Social

829 Fund) and National Funds (Hellenic General Secretariat for Research and Technology) in the framework of the project

ARISTEIA I, 640 "Integrated Study of Trace Metals Biogeochemistry in the Coastal Marine Environment", within the

831	"Lifelong Learning Programme". This project gave us the opportunity to reach the heavy metal pollution of West Saronikos
832	Gulf. This paper summarizes the results of this study.
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